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DATE	10	March	1964	
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ELECTROCHEMICAL

DEBURRING OF MOLYBDENUM,

ALUMINUM AND

STAINLESS STEEL

REPORT A478 SERIAL NO. 20

MCDONNELL

This report was prepared under Contract Number AF33(657)-11215 and BPSN: 63-6899-7381-738103. Additional information pertaining to any data contained herein may be obtained from the Directorate of Materials and Processes (ASRCEM-1), Aeronautical Systems Division, Air Force Systems Command, United States Air Force, Wright-Patterson Air Force Base, Ohio, or McDonnell Aircraft Corporation, St. Louis, Missouri

INDEX			
CODE	(Mo-4,	Alwt-8,	FeAH-2)(V1-c)

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FIRAL REPORT

(Mo-4, Alwt-8, FeAH-2) VI-c)

ELECTROCHEMICAL DEBURRING OF MOLYEDENUM, ALAMINUM AND STAINLESS STEEL

ABSTRACT

Sharp edges left after chemical milling and blanking are presently being mechanically removed at high cost. The sharp edges would be areas of high current density in electrochemical operations and could be easily removed in electrochemical solutions.

The criteria of a good electrochemical deburring solution were leveling characteristics and removal rate. Various solutions were evaluated for breaking sharp edges on molybdenum, among them being 25% by weight nitric acid, nitric-hydrofluoric acid solution for milling molybdenum and variations of Turco 105 steel etchant. These solutions were successful in a leveling action on sharp burrs, but would not round the side edges. A combination of nitric acid leveling and side edge burnishing produced the most promising results.

Aluminum and stainless steel deburring was evaluated in proprietary solutions, with good results obtained in polishing and breaking of sharp edges.

PREPARED BY James E.	Word APPROVED BY Senior Engineer, Materials and
Test Engineer	Senior Engineer, Materials and Manager Methods, Chemical Group
APPROVED BY WEILL	APPROVED BY

Chief, Structures Laboratory

Laboratory Project Engineer

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1. OBJECT

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To determine the electrochemical techniques and solutions required to remove sharp edges which result in chemical milling and blanking.

2. MATERIALS AND SOLUTIONS

(1) Molybdenum per MMS-184

(2) Aluminum 7178-T6 per MIL-A-9180

(3) Stainless Steel, 17-7 pH per MIL-S-25043, 321 per MIL-S-6721
(4) Electro-Gleam 55 (Mac Dermid Inc.)

- (5) Electropolish Bs (Mac Dermid Inc.) (6) Electropolish BA (Mac Dermid Inc.)
- (7) Sulfuric acid alcohol polish

One part concentrated sulphuric acid to seven parts methyl alcohol (95%)

(8) Mitric acid 50% by weight 25% by weight 75% by weight Concentrated.

(9) Chromic-Sulphuric acid 11.3% by volume, concentrated sulphuric acid, chromic acid .17 lb/gal, and remainder water.

(10) Nitric acid (42°Be') 45% by volume - hydrofluoric acid (70%) 3.5% by volume - remainder water.

(11) Turco 105 steel etchant

- (a) Fe(NO₃)3 9H₂0 370 gms FeCl₃ 6H₂0 - 236 gms H₃PO₄ 70% - 124 mls CrCl₃ .6H₂0 - 51 gms HCl 12.4N - 121 mls water to make 1 liter
- (b) Same as (a) adjusted to H+=6.0N and $NO_3=4.5N$
- (12) 4140 Steel etchant H+ = 6.5 N, NOZ 2 2.7N
- (13) Incomel 718 etchant $H+ = 3.0 \text{ N}, \text{ NO}_3 = 2.5 \text{ N}$

3. TEST PROCEDURE

Approximately 4 liters of each of the solutions listed in Section 2 were prepared. The specimens were first deburred with the mask left on. They were cleaned by immersion in Prosolve B for 5 minutes at 150°F, rinsed in hot water, then cold. A spot was bared on the specimen by scraping

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3. TEST PROCEDURE (CONT'D.)

away a portion of the mask. A clip was attached to the bare spot and the specimen lowered into the solution up to, but not including the clip. The specimen was deburred and then removed and rinsed in cold water. Examination of each specimen run was performed with the aid of a microscope.

The next step was to run a different set of specimens by removing the mask. The mask was soaked loose and removed in cold trichloroethylene. They were immersed in Prosolve B for 5 minutes at 150°F and rinsed in hot water, then cold. The specimen was clipped and immersed in the deburring solution up to, but not including the clip. It was then removed and rinsed in cold water. Examination was made for metal removed, leveling of the burr, and surface finish. At one time during the testing, production specimens were submitted for deburring. A racking procedure was devised to eliminate rack marks from the specimen. A clip was constructed in such a manner as to provide a point contact through the mask. The point contact was then remasked with KMER for protection. See Figure 2, page 45. Another method for racking (to reduce rack marks) was to spot-weld the alumel wire of a chromel-alumel thermocouple onto the edge of the specimen. The method was successful for a limited period of time until the etching action of the solution loosened the alumel wire, permitting the specimen to drop off.

4. TEST RESULTS

4.1 Nitric Acid

Deburring with and without maskant was performed in the following concentrations.

4.1.1 25% By Weight

Of all the solutions evaluated, this solution produced the best leveling. The best leveling occurred at 125°F and 8 amps per square inch. The time varies with the height of the burr. See Table 1, page 7 and Table 11, page 17, for parameter data. The one disadvantage of this solution was that it would not round the edges. In an effort to round the edges, approximately .020-inch of the mask was scraped around the edge and removed. This was not successful, in that the sides of the edges were undercut, producing more sharp edges. See Table 17, page 23, specimens 1 through 6 and specimen 14, in conjunction with Figure 1, Photographs of specimens 1 through 6, and 14, are presented on pages 25 and 26.

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4.1.2 50% by Weight

This solution produced leveling, but was not as good as the 25% by weight. Parameter data is shown in Table 2, page 8.

4.1.3 75% by Weight

There was some leveling produced, although the surface was pitted and irregular. The conduction of the solution was considerably low, which produced a slow deburring rate. See Table 3, page 9, for parameter data.

4.1.4 Concentrated

The conduction of the solution was too low to allow a reaction between the molybdenum and the solution, consequently no deburring was accomplished. See Table 3, page 9 for parameter data.

4.2 Chromic Sulphuric Acid

This solution is not satisfactory for deburring because of poor leveling action. A slight polish is produced, which is of little value. See Table 4, page 10, and Table 12, page 18, for parameter.

4.3 Alcohol - Sulphuric Acid

The inability of this solution to level the burr excludes it as a possible deburring solution. It also attacks and removes the maskant. See Table 5, page 11, and Table 13, page 19, for parameter data.

4.4 Mitric-Hydrofluoric Acid

Acceptable leveling action is produced by this solution; however, its one disadvantage is the high current density necessary for burr removal. See Table 6, page 12, for parameter data.

4.5 Electrogleam 55

There was very little leveling action produced by this solution. It also attacks the maskant. A high polish was obtained, which is of little value. See Table 7, page 13 and Table 9, page 15 for parameter data. Because the solution did have a rounding ability for sharp edges, the edge was leveled first in nitric acid and the mask completely removed. The Electrogleam 55 treatment was applied and the specimens exhibited a radius on the edges from .002-in. to .006-in. depending on current density. This method produced good results, but a satisfactory manner of racking on the bare specimen to eliminate rack marks was not achieved. See Table 17, page 23 in conjunction with Figure 1, photographs 7 through 13. The next step was to leave the mask on and rack with the point contact method, explained under section 3, but scraping away .020-in. of the maskant on the sides of

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4.5 Electrogleam 55 (Cont'd.)

the edge. This produced undercutting of the sides, which was not acceptable. See Table 17, on page 23, Specimen 27A through 29C in conjunction with Figure 1, photographs 27A through 29C on pages 33 - 35.

4.6 Electrogleam B8

This solution was not successful in leveling the burr, but did exhibit a rounding of sharp edges. See Table 8, page 14, and Table 10, page 16, for data. The same racking problem was encountered with electrogleam BS on the base specimen, as with Electrogleam 55, discussed under Section 4.5. The point contact method of racking was used with the mask left on. Approximately .020-in. of the maskant was scraped away from the sides of the edge. The specimen was leveled in nitric acid and the Electrogleam BS treatment applied. No satisfactory specimens were obtained because of undercutting of the sides of the edges. See Table 17 on page 23, and specimens 15 through 18C in conjunction with Figure 1, photographs 15 through 18C, pages 29 and 30.

4.7 Turco 105 Etchant for Steel

Turco 105 was successful in leveling the burr with current densities ranging from 5 amps per square inch to 20 amps per square inch. Rounding of the side edges was accomplished on some specimens, but was not repeatable. See Table 18, page 24, specimens 30 and 31; also see Table 17, page 23, specimens 19 and 20 in conjunction with Figure 1; photographs 19, 20, 30 and 31, pages 31 and 36.

In an effort to round the edges, approximately .020-in. of the maskant was scraped from the edges and Turco 105 treatment applied. This did not prove successful because of undercutting of the sides of the edges. See Table 17, page 23, specimens 21 through 26, in conjunction with Figure 1, and photographs 21 through 26. Burnishing of edges was tried by scraping the edges with a knife after Turco 105 treatment. This proved very successful and radii were obtained from .005-in. to .008-in. See Table 18, page 24, specimens 39 through 42, in conjunction with Figure 1 and photographs 39 through 42. Specimens were also performed without burnishing of the edges after Turco 105 treatment. This resulted in very sharp edges which were not acceptable. See Table 18, page 24, specimens 46 through 48, in conjunction with Figure 1, and photographs 46 through 48.

4.8 Variations of Turco 105 Steel Etchant

Attempts to improve leveling and rounding of edges by raising and lowering H+ and NO3 were unsuccessful. There was some tendency toward faster removal rates with the higher H+ and NO3 ion concentration, however, this was not substantial. Specimens were run by scraping the mask away at a 45° angle to the edge. This still produced sharp edges. See Table 18,

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4.8 Variations of Turco 105 Steel Etchant (Cont'd.)

page 24, specimens 46 through 48, 49 through 51, and 52 through 56, in conjunction with Figure 1, pages 41 - 44 and photographs 46 through 48, 49 through 51, and 52 through 56. Attempts to round the edges by burnishing with a knife after leveling were again successful, with radii ranging from .005-in. to .0058-in. See Table 18, page 24, specimens 32 through 34, 35 through 38, 43 through 45, in conjunction with Figure 1, pages 37, 38 and 40 and photographs 32 through 34, 35 through 38, 43 through 45.

5. ELECTROCHEMICAL DEBURRING OF STAINLESS STEEL AND ALUMINUM

Proprietary solution Electropolish BS produced a more satisfactory polish and rounding of edges than did the Electrogleam 55 on stainless steel 321. The Electropolish BS produced an RMS of 10 from a raw sample of RMS 60. The metal removal was much less from Electropolish BS than Electrogleam 55. See Tables 15 and 16, pages 21 and 22, in conjunction with Figure 4, page 47. Electropolish BA produced an RMS value of 25-30 from a raw sample of RMS 20-22. See Table 14, page 20, in conjunction with Figure 3, page 46.

6. CONCLUSIONS AND RECOMMENDATIONS

25% by weight nitric acid and Turco 105 steel etchant exhibit the greatest leveling characteristics. However, due to the high cost of make-up and maintenance of Turco 105 etchant, nitric acid would have the preference. Burnishing with a special tool or a pencil-type grinding wheel after leveling of the burr in nitric acid is recommended in rounding the edges to a .005-in, radius.

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							713	L Min
TABLE I ACID 25% BY WT Wolybdenum	COMMENTS	SOME LEVELING OF BURR. FORMS A PRECIPITATE WHICH INHIBITS CUTTING	ADJUSTED CATHODE TO ANODE AREA TO 10 to 1. NO NOTICEABLE DIFF	SOME LEVELING OF BURR. FORMS A PRECIPITATE WHICH INHIBITS CUTTING	SOME LEVELING OF BURR. REQUIRES DESMUTTING	VERY GOOD LEVELING OF BURR. REQUIRES DESMUTTING	GOOD LEVELING OF BURR. SOME PITTING.	DE COMPOSES ACID. NO DEBURRING
TRIC ACID	TEMP (*F)	06	9.6	06	100	125	100	90
N/TE	TIME (MIN.)	12	/3	ત	1,	61	ı	'n
	VOLTAGE	9	9	01	7	- 0	8	4
	CURRENT DENSITY (AMP/SQIN)	m	8	8	00	00	12	2.5
	AREA (Sq. IN)	90.	50.	50.	81.	90.	80	80.

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		FORMS B . T S	FORMS H. B. TS	PittiNG	P. TING	P. #1NG	SEVERE
molybdenum	ENTS	OF BURR. FO WHICH INHIBITS	OF BURR. FORM	OF BURB.	OF BURR	OF BURR.	9 F 8 U.R. P.
SPECIMENS of molybolenum	COMMENTS	1 E L! NG	SOME LEVELING A PRECIPITATE CUTTING	SOME LEVELING	LEVELING	LEVELING OF BURR.	LEVELING
		SOME LEVA PRECIP	SOME LE A PRECIP CUTTING	SOME	SOME	SOME	SOME L
MASKED	TEMP (OF)	8/	100	44	811	96	12.5
NITRIC WITH MA	TIME (MIN)	12	13	h	71	12	'n
	VOITAGE	1	2.5	1.5	e	01	7
	CURRENT VOITAGE T	1	80	4	8.3	3,0	2.5
	AREA (Sq IN)	Fo.	10.	10.	1.0.	10.	90.

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PIRAL REPORT PITTING P: # 1 10 PITING MASK MASK OF BURK. SEVERE BURR SEVERE REACTION SEVERE REACTION ENS Of molybdenum AHACKS ATTACKS OF BURR. 3100 SIOW COMMENTS E BURRING. OF CONCENTRATED COMMENTS DEBURRING MASK TEVELING LEVELING LEVELING MASK. SPEC.M RAPIDLY RAPIDLY. ATACKS ATTACKS 0 ATTAKS SOME SOME SOME 50 20 MASKED W TABL ACID (OF) TEMP (OF) 00 NITRIC WITH M 112 00 Do 00 00 00 NITRIC 1 × × E SE 10 01 0 0 0 CURRENT DENSITY VOITAGE (MMP/SQIN) CURRENT VOITAGE 0 20 01 0 0 2 0 2 0

3

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SQ 1N.)

AREA

110

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						MAT I	
CHROMIC - SULPHURIC ACID WITH MASKED SPECIMENS Of molybdenum	COMMENTS	NO LEVELING OF BURK	SLIGHT POLISH NO LEVELING OF BURR	SLIGHT POLISH. NO LEVELING OF BURR SOME PITTING	SLIGHT POLISH. NO LEVELING OF BUKK SOHE PITING	SlIGHT POLISH NO LEVELING OF BUCK SOME PITTING	SUGHT POLISH NO LEVELING OF BURR
	TEMP (OF)	011	011	47	35	76	76
	TIME	9	4	9	9	۸	'n
	398 TIOV	2.0	0	3.2	o Vi	0.9	0.8
		+	Da	Ŋ	Ŋ	5.5	15
	AREA CURRENT (SQ IN) DENSITY	.39	.39	.39	.39	.39	.39

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					PINAL	REPOR
MASKED SPECIMENS Of molybehum	COMMENTS	HIGH POLISH. NO LEVELING OF BURR. ATTACKS MASK	HIGH POLISH. NO LEVELING OF BURR, ATTACKS MASK	HIGH POLISH. NO LEVELING OF BURR. ATTACKS MASK	HIGH POLISH. NO LEVELING OF BURK. ATTACKS MASK, Add & 100 MLS. H2504. PECOMPOSES ALCOHO!	400 FD 200 MLS OF H2504
ALCOHOL W:TH MA	TEMP (OF)	0 &	08	80	80	0 %
4 3	TIME (MIN)	ر,	p)	01	0 M	6
	VoltAGE	3·S	4.5	e e	و	0.9
	CURRENT DENSIFY	7.7	r.	۶.٬	0 :	07
	AREA (SO.IN)	+1.	I.	7.	1.17	94

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THE MARKET LEVEL, NG DEBURRING ENS Of molybdenum DEBURRING AND LEVELING AND AND BURRING COMMENTS DEBUKKING HYDROFLOURIC DEBURRING PEVELING 4 MASKED GOOD G005 G005 Pook PooR ١ TABLE NITRIC (PF) 3 911 0 115 211 00 TINE T 0 1.5 1.0 n m VOITAGE 1.4 1.0 0 0 DENSITY AMP/SO IN CURRENT 04 2 20

0

0

100

60.

60.

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(SQ .IN)

80.

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		reservations are the			THE PERSON
ELECTROPOLISH BS	COMMENTS	SIJCHT LEVELING OF BUKR. ROUND EDGES NOT AS BRIGHT A POLISH AS ELECTROGIEAM SS. ATTACKS MASK	SIIGHT LEVELING OF BURR. ROUNDS EDGES	SLIGHT LEVELING OF BURK. ROUNDS EDGES. HIGH POLISH, ATTACKS MASK.	ROUNDS EDGES. HIGH POLISH.
ELECT WITH	TEMP (°F)	061	306	081	200
١	Time (M.N)	18	0/	0/	18
	VOITAGE	7	0/	۵/	51
	CURRENT DENSITY AME/59 IN	5.1	ત	4	m
	AREA (50.1N)	10.	04 0	90.	60.

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molybdenum	RMS	09	09	9
ELECTROGIEAM SS From molybdenum	REMOVED (IN/SIDE)	8000'	. 0005	5/00'
	TEMP (°F)	180-190	061-081	061-081
	TIME (MIN)	0/	0,	0/
	V01 + 4.6 E	0/	Ь	/2/
	CURRENT DENSITY (AMP/Sq. IN)	o o	8.0	7.6
	AREA (SQ.IN)	m	4	//

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PINL BEFORE ELECTROGLEAM 55 GREATER THAN ELECTRO GIEAM SS GREATER THAN ELECTROGIEAM SS GREATER THAN EIECTROPOLISH BS From molybdenum RMS METAL REMOVAL (IN/SIDE) 8000. .0005 2000, 700-000 200-205 302-202 TEMP (°F) 0 TABLE 0/ TIME (MIN) 0/ 0/ VOITAGE 5 0/ 1 DENSITY (AMP/SAIN) CURRENT 4 <u>ا.</u> م 0:1 AREA (SQ.IN) 4 d 4

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					PINAL REPORT
	SEMOVED from molybolenum	٠ ٢ ٧	GREATEL THAN ELECTROGLEAM SS	GREATEK THAN ELECTRO GLEAM SS	GREATER THAN ELECTRO GLEAM SS
	1% BY &1	METAL REMOVAL (IN/SIDE)	1000 ·	S400.	900.
	3LE 11 1C ACID 25 MASK REM	TEMP (°F)	72	241	ر ا
	TABLE NITRIC A	7 : ME (M: N)	0/	0/	07
	< 3	VOLTAGE	и ò	9.0	1.1
		CURBENT DENSITY (AMP/59 IN)	η	⊳	7.
*		/11 E A (50 1N)	~	તું	1.1

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molybdenum	ν Σ	GREATER THAN ELECTROGIEAN SS	GREATER THAN	GREATER THAN ELECTROGLEAM SS
C Acid	METAL REMOVAL (IN/S.DE)	\$ 600.	. 2002	700.
CHROMIC - SULPHURIC ACID WITH MASK REMOVED From molybdenum	76 M P	011	011	105
	T.ME	20	0 1	0 d
	VOITAGE	ls d	2.5	4.5
	DENSITY (AMP/SO IN)	7	d	∞
	AREA (SQ IN)	d d	d	0.4

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				PIRAL REPORT
TABLE 13 ALCOHOL - SULPHURIC ACID WITH MASK REMOVED From molybachum	RMS	GREATER THAN ELECTROGIEAM SS	GREATER THAN ELECTROGLEAM SS	GREATER THAN ELECTROGIEAM SS
C ACID OUED Fro	METAL REMOVAI	700'	-S 200 ·	£ 00 ·
SULPHURI ASK REM	TEMP (°F)	<i>⊙</i>	08	08
TABLE ALCOHOL - S WITH MI	TIME (MIN)	u 0	70	0 &
	VOLT A GE	ε V:	5.4	0.9
	CURRENT DE NSITY (AMP/SQ IN)	Ń	•	۲.
	AKE A (50. 14)	d T	o o	ત

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				PIRAL REPOR
	COMMENTS	HIGH POLISH ATTACKS MASK ROUNDS EDGES	HIGH POLISH ATTACKS MASK ROUNDS EDGES	HIGH POLISH ATTACKS MASK ROUNDS EDGES
	RMS	25	2 %	30
B A	METAL REMOVAL (IN/SIDE)	100.	\$100.	.000
	TEM P (°F)	180-190	061-081	061-081
ELECTROPOLISH ON ALUMINUM	7 1 ME (M1N)	70	20	02
E1	VOLTAGE	9	6	77
	CURRENT DENSITY (AMP/SQIN)	رة بح		м ∞ •
	AREA (50 1N)	7-	7	+

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	COMMENTS	HIEH POlisH Rounds EDGES	HICH POLISH ROUNDS EDGES	HIGH POLISH ROUNDS EDGES
	RMS	01	0,	0/
H BS 321	METAL REMOVAL (IN/SIDE)	100,	.0015	. s o o s
15 POLISH E LESS STE	TEMP (0F)	180-2-00	180-200	002 - 08/
TABLE 15 ELECTROPOLISH ON STAINLESS	7: ME (M:N)	Ŋ	٨	Ŋ
	10LTA GE	6	12	15
	CURRENT DENSITY (AMP/SQ IN)	3.3	5,0	O ' %
	AREA (SQ IN)	Ŋ	بع. ج	<i>Se</i> '.'

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	1	7		PINAL R
	COMMENTS	HIGH POLISH ROUNDS EDGES	HIGH POLISH ROUNDS EDGES	HIGH POLISH
	R M S	25	35	\$
55 321	METAL REMOVAL (IN/SIDE)	. 003	500.	. 00 5 5
LEA MES	TEMP (OF)	180-200	180-200	180-200
ELECTROGLEAM SS 321	VOITAGE	٥	6	61
Д	TIME (MIN)	Ŋ	Ŋ	И
	CURRENT DENSITY (AMP/SQIN)	"	15	61
	AREA (Sa. 1N)	1.12	200	.75

TABLE

NITRI	CAC	D TR	EATA	NENT					C AC	10
SPECIMEN A	REA (2 IN)	CURRENT	TIME (MIN)	TEMP (1F)	VOLTAGE		AREA (SQ IN)	CURRENT DENSITY (AMP/SQIM)	TIME (MIN)	Ţ
1 2 3 4	.114 .14 .18 .11	8 9 8 8 8 8 8 8	10 6 6 20 20 20 20	1/8-128 1/8-128 1/8-128 1/8-128 1/8-128 1/8-128 1/8-128			· 14 · 18 · 63 · 11	3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3	6 2 2,5 1	11 ê 11 8 11 8
NITR	RIC	ACID	TRE	ATME	NT		El	ECT	ROGLI	E AI
7 9 10 11 12 13	.11	ש מי שנט מי מינט	10 20 20 20 20 20	8- 28 8- 28 8- 28 8- 28 8- 28 8- 28	.1	-	342011.18	8 1 1 6546	/ C / C / C 2 O 2 O 2 O / C	12 12 12 12 12 12
NITE	210 /	ACID	TRE	ATME	NT		EL	ECTR	OPOL	158
15	.08	8000	20 20	118-128 118-128 118-128	1,2 1,2 1,5		, 16 . 16 . 34	3 2.5 3	30	33 33
18 A 18 B 18 C	. 22	8	20	118-128	، \$. 44	2.5	30	22
TURC	0 /	05 TR	EAT	MEN-	Γ		TI	JRCO	105	5
19 20 21 22 23 24 25 26	.16 .12 .10 .15 .3 2	20 20 20 20 20 20 20	\$ 7 5 5 5 5 5 7 0 2 0	145 ± 5 145 ± 5 145 ± 5 145 ± 5 145 ± 5	2		.16	20 20 20 23	1 1, 5 2 1	14
NIT	RIC	ACID	TRE	AT ME	NT	complete and the second	EL	ECTRO	GLEA	M
27A 27B 27C	. 10	2	20	118-128	12		.06	3.4	10	18
28 A 28 B 28 C	. 15	8	20	118-12	g 1.3		. 55	3, 2	20	18
29 A 29 3 29 C	. 07	8	20	118 - 128	7 1.0		. 05	3.2	30	r
						1	7			

-				NITRI	c Ac	10 TRE	EATME	NT		COMMENT
P	VOLTAGE	-	AREA (SQ IN)	CURRENT DENSITY (AMP/19 IN)	TIME (MIN)	TEMP (°F)	VOLTAG	E REMOVED	RADIUS (IN)	
28 28 28 28 28 28	.05 .05 .1 .1 .1		· 14 · 18 · 03 · 11	2 2 2	6 2 2,5	118-128 118-128 118-128 118-128	.1 .1 .1		NONE NONE NONE NONE NONE NONE	Mask was scraped from nitric acid treatme scraped from edge be Treatment on specin Figure 1 with picture specimen No.
E	N T		EL	_E C T R	06L1	EAM 5	5 TR	EATM	ENT	
2 2 2 2 2 2 2 2 2	.05 .1 .1 .05 .1		34 434 1.18	. 8 1 1.65 1.65 1.6	10 10 10 20 20 20	120-190 120-190 120-190 120-190 120-190 120-190	10 10 10 10	.0005	.002 .0016 .0012 .006 .006 .005	Mask was completely hitric Acid treatment ss treatment with specimen 13 where t scrayed from the ed treatments see Figure corresponding to specim
Ε	NT		EL	ECTRO	O PO L	ISH B	S TR	EATM	ENT	Mask was screped from
8 8 8 60	1,2 1,2 1,5		.16 .16 .34	3 2,5 3	30	220-230 220-230 220-230	15		NONE NONE NONE	withis Acid Treatment BS treatment. Specime respectively, raw sampl reueling, exectropolish i Figure 1 with pictures
8	, 3		. 44	2.5	30	250-530	15		NONE	Søftermen No.
1			TU	JRCO	105	TR	EAT	MENT		
Soundand	2 2 2 2 2 2 2 2 2		.16 .12 .10 .15	20 20 20 23	1 1, S 2 1	142 #2 142 #2 142 #2	. 2 2 2 4	. 601	NONE NONE NONE NONE NONE NONE NONE	Mask was scraped fro TREatments on specim Mask was completely r 25 and 26, Specimen Current. See Figure 1 Corresponding to Specime
N	T		EL	ECTRO	GLEAI	M 55	TREA	TMEN	VT	Mask was scraped fro Treatments. Specimen
28	1.2		. 06	3.4	10	150 - 190	10		NONE	raw samples. Specimen. Nitric acid Leveling. Spe are Electrogleam SS tre 27A, B, C. all regresent
28	1.3	8	.05	3, 2	20	180-190	10		NONE	corresponding to specin
78	1.0		. 05	3.2	30	180 - 190) 10		NONE	
			:							

	T	URCO	105	· T	REAT	MENT			COA
	SPECIME	(SQ. IN)	DENSITY	TIME (MIN)	TEMP (F)	VOITAG	E RADIUS		Etchant
	30	. 06	20	5 5			.003		NO opera See Figu Correspon
	41	40 S	TEEL	. ET	CHAN	T	-		Etchant
	32 33 34	• [] • [] • []	10 5 20		145 ± 5 145 ± 5 145 ± 5		.0058		a Knife on Edge Figure 1 to specim
-	IN	CONEI	L 718	E'	TCHAN	T			burnished
	35 36 37 38	.088 .088 .088 .088	10 5 20 10	10 15 5 30	145±5 145±5 145±6 75-85	٠5	.0054 .0054 .0046 NONE	-	A Knife Edge and with pict No. Etc. 2.7 No.
-				REA	TMEN	IT			a knife edg
V	39 40 41 42	RAW SPE	10 CIMEN F	15 OR 35	145±5 145±5 145±5 145±5 1460 40	, 5 0 4	.0058		burnished w deburring. corresponding
-	TUR	CO H	+= 6.0		N 03 = 4	ł. 5			a Knife Edge and mask s
	43 44 45	.192 .2 .2	5 10 20		145±5 145±5 145±5	1.5 3.2 6.0	NONE NONE		burnished wi deburring. S corresponding
	TUR	0 1	05 T	REAT	MENT				a Knife Ed a mask scra Figure 1 wit
The second secon	46 47 48	.12 .12 .12	5 10 20	10	145±5 145±5 145±5	. 5 1.5 2.0	NONE NONE		Figure 1 with
	INC	ONEL	718	ETCH	TANT				a Knife Edg
	4 9 50 51	.04	\$ /• 20		145 ± 5 145 ± 5 145 ± 5	, 25 2,0 3.0	NONE NONE		and mask figure I w te specimen
	4140 52 53 54	STE .07 .17	EL ET		VT 145±5 145±5 145±5	, \$ 3.0 5.0	NONE NONE NONE		a Knife Edg and mark
	55	RAW SPEC	CIMEN FOR	R 46	and 49 18, 50,51,58				Specimen N

TABLE 18

O 105 TREATMENT	COMMENT
CURRENT TIME TEMP VOITAGE RADIUS	The state of the s
20 5 145±5 2 .003 20 5 145±5 2 .003	NO operation performed on mack. See Figure 1 with pictures corresponding to specimen No.
STEEL ETCHANT	Etchant analyzed 3.0 Ht and 2.5 NO.
10 20 145±5 2 .0658 5 30 145±5 1 .005 20 10 145±5 3 :.0058	on Edge and mask scraped. SEE Figure 1 with gictures corresponding to specimen No. Specimens were
L 718 ETCHANT	burnished with a knife edge after deburring.
10 10 145±5 1 .0058 5 15 145±5 15 .0054 20 5 145±6 2.5 .1046 10 30 75-85 1.5 NONE	a knife edge was placed 45° or. Edge and mask scraped. See Figure 1 with pictures corresponding to specimen. No. Etchant Arabysid 6.5 Ht and 2.7 NO3. Specimens were burnished with a Knife edge after deburring.
105 TREATMENT 20 10 145±5 1 .0058 10 15 145±5 .5 .0058 ECIMEN FOR 35 thru 40 ECIMEN FOR 32 thru 34	a knife edge was placed 45° on edge and mark scraped. Specimens were burnished with a Knife edge after deburring. See Figure 1 with pictures corresponding to specimen No.
$H^{+}=6.0$ $N0_{3}^{-}=4.5$ 5 30 145±5 1.5 .007 10 15 145±5 3.2 NONE 20 8 145±5 6.0 NONE	a Knife edge was placed 45° on edge and mask scraped. Specimen 43 was burnished with a Knife edge after deburring. See Figure 1 with fictures corresponding to specimen No.
05 TREATMENT 5 10 145±5 .5 .008 10 10 145±5 1.5 NONE 20 5 145±5 2.0 NONE	a Knife Edge was placed 45° on Edge a mask scraped. No burnishing. SEE Figure 1 with pictures corresponding to specimen No. 8
718 ETCHANT \$ 20 145±5 .15 .003 10 15 145±5 2.0 NONE 20 10 145±5 3.0 NONE	a Knife Edge was Blaced 45° on Edge and mask scraped. Mo burnishing, SEE Figure 1 with pictures corresponding te specimen No.
EL ETCHANT \$ 40 145±5 S NONE 10 20 145±5 3.0 NONE 20 9 145±5 5.0 NONE CIMEN FOR 46 and 49. CIMEN FOR 47,48,50,51,52,53 and 54	a Knife edge was placed 45° on edge and mask scraped. No burnishing. See Figure 1 with pictures corresponding to specimen No.

COMMENT	
NO operation performed on mark. See Figure 1 with pictures corresponding to specimen No.	
Etchant analyzed. 3.0 Ht and 2.5 NOg A Knife Edge was placed 45° on Edge and mask scraped. See Figure 1 with pictures corresponding to specimen No. Specimens were burnished with a Knife edge after deburring.	
a knife edge was placed 45° on Edge and mask scraped. See Figure 1 with pictures corresponding to specimen No. Etchant Analyzed 6.5 Ht and 2.7 NOZ. Specimens were burnished with a knife edge after deburring.	
a knife edge was placed 45° on edge and mask scraped. Specimens were burnished with a knife edge after deburring. See Figure I with pictures corresponding to specimen No.	
a Knife edge was placed 45° on edge and mask scraped. Specimen 43 was burnished with a Knife edge after deburring. See Figure 1 with pictures corresponding to specimen No.	
a Knife Edge was placed 45° on Edge a mask scraped. No burnishing. SEE Figure 1 with pictures corresgonding to specimen No.	
a Knife Edge was placed 45° on Edge and mask scraped. <u>No</u> burnishing, see Figure 1 with pictures corresponding to specimen No.	
a Knife Edge was placed 45° on edge and mask scraped. No burnishing. See Figure 1 with pictures corresponding to specimen No.	



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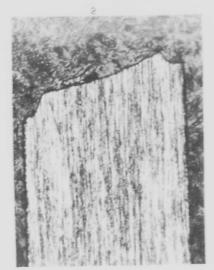
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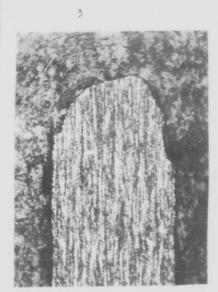
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FIGURE 1 - DEBURRED SPECIMENS - 25% MITRIC ACID

MAG. 75%









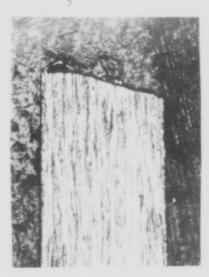
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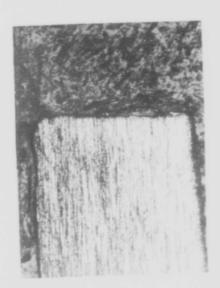
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FIGURE 1 - DEBURRED SPECIMENS - 25% NITRIC ACID

MAG. 75X







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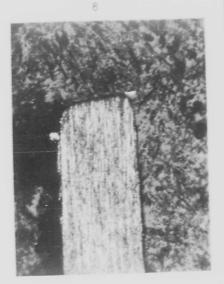
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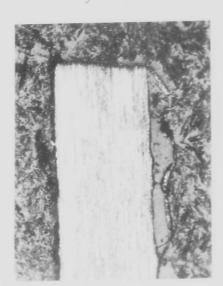
PIGURE 1 - DEBURRED SPECIMENS LEVELED IN 25% NITRIC ACID AND POLISHED IN ELECTROGLEAM 55. MAG. 75X

-77

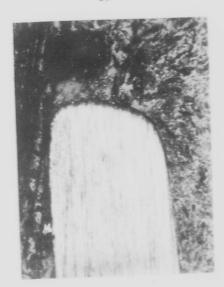




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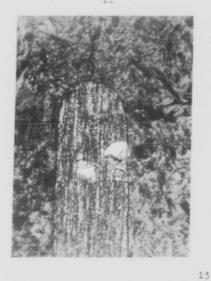


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FIGURE 1 - DEBURRED SPECIMENS LEVELED IN 25% NITRIC ACID AND POLISHED IN ELECTROGLEAM 55

MAG. 75X





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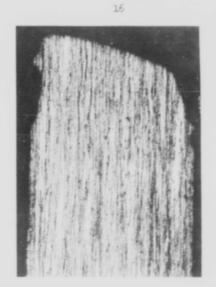
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MEVISED D4E-255991

FIGURE 1 - DEBURRED SPECIMENS LEVELED IN 25% NITRIC ACID AND POLISHED IN ELECTROPOLISH B.S. MAG. 75X

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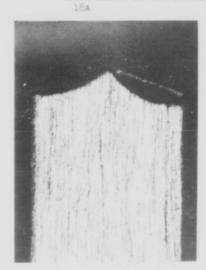
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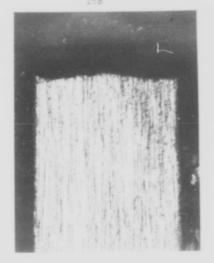
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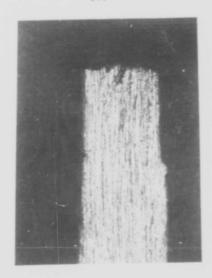
FIGURE 1 - DEBURRED SPECIMENS LEVELED IN 25% NITRIC ACID AND POLISHED IN ELECTROPOLISH B.S.

MAG. 75X





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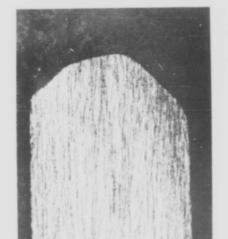
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FIGURE 1 - DEBURRED SPECIMENS. TURCO 105 ETCHANT

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MAG. 75%

19







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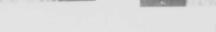
FIGURE 1 - DEBURRED SPECIMENS, TURCO 105 TREATMENT

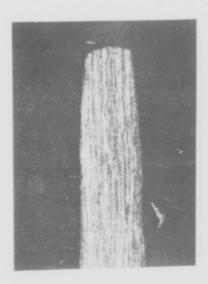
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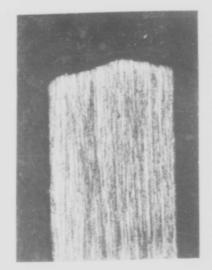
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MAG. 75X 24









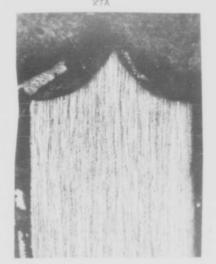
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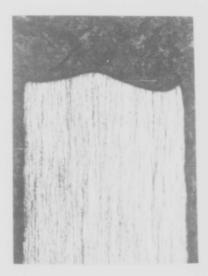
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FIGURE 1 - DEBURRED SPECIMENS, LEVELED IN 25% NITRIC ACID AND POLISHED IN ELECTROGLEAM 55.

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FIGURE 1 - DEBURRED SPECIMEN LEVELED IN 25% NITRIC AND POLISHED IN ELECTROGLEAM 55

28A

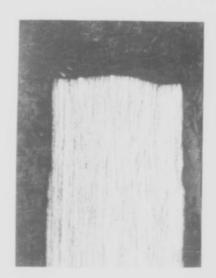


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MAG. 75X 28B



28C



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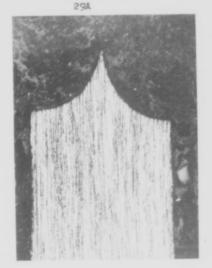
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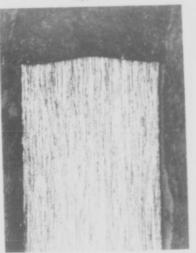
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FIGURE 1 - DEBURRED SPECIMENS LEVELED IN 25% NITRIC ACID AND POLISHED IN ELECTROGLEAM 55 29A

MAG. 75X

29B





29C



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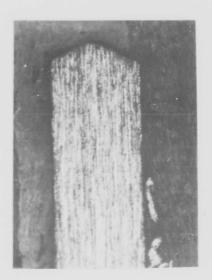
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FIGURE 1 - DEBURRED SPECIMENS TURCO 105 ETCHANT

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MAG. 75X

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FIGURE 1 - DEBURRED SPECIMENS - 4140 STEEL ETCHANT

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MAG. 75% 33







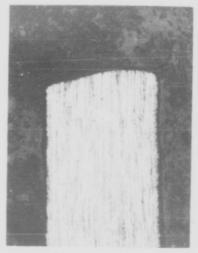
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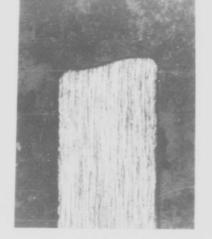
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FIGURE 1 - DEBURRED SPECIMENS INCONEL 718 ETCHANT

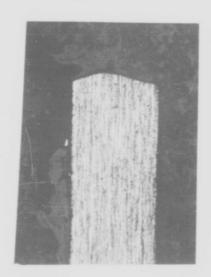
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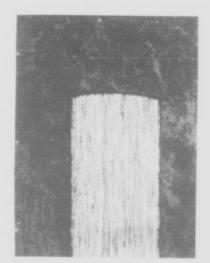
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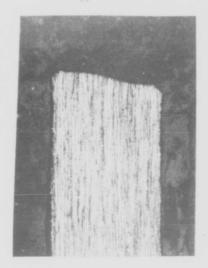
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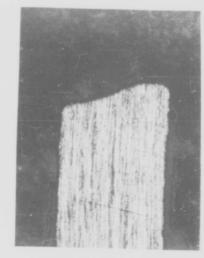
FIGURE 1 - DEBURRED SPECIMENS TURCO 105 ETCHANT

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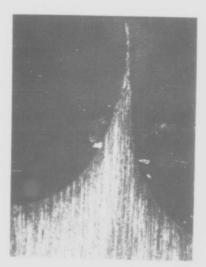
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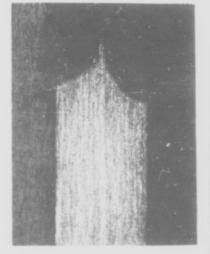






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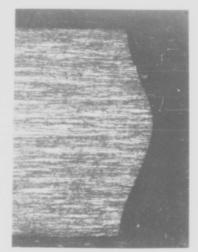
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FIGURE 1 - DEBURRED SPECIMENS. TURCO 105 ETCHANT
H+ = 6.0, NO₃" = 4.5

MAG. FINAL REPORT







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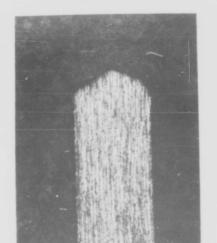
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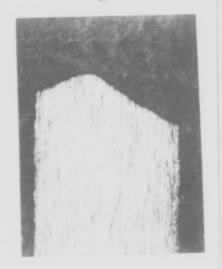
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FIGURE 1 - DEBURRED SPECIMENS - TURCO 105 ETCHANT

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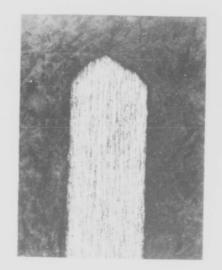


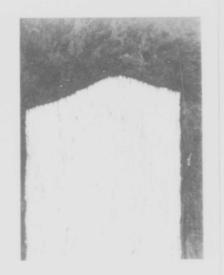


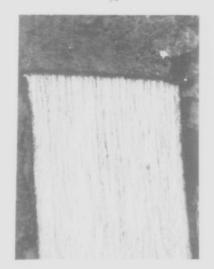
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FIGURE 1 - DEBURRED SPECIMENS - INCONEL 718 ETCHANT







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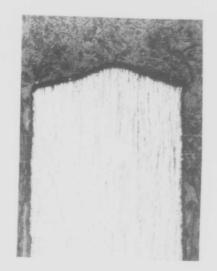
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FIGURE 1 - DEBURRED SPECIMENS - 4140 STEEL ETCHANT

53

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MAG. 75X

52







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FIGURE 1 - RAW SPECIMENS

MAG. 75X





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FIGURE 1

METHOD OF RACKING





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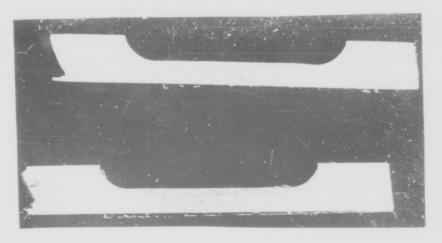
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FIGURE 3 - ELECTROPOLISH AND ELECTRO -DEBURRING OF ALUMINUM

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AFTER ELECTROPOLISH



BEFORE ELECTROPOLISH

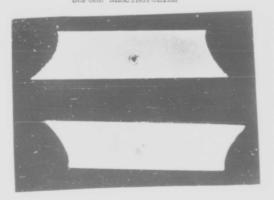
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FIGURE 4 - ELECTROPOLISH AND ELECTRO-DEBURRING OF STAINLESS FINAL REPORT STEEL

BEFORE ELECTROPOLISH



AFTER ELECTROPOLISH

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PAGE 1	TEST REC	QUEST					
TITLE Electrochemical Deburring of Molybdenum Aluminum							
and	Stainless Steel						
LABORA	Misc.						
	ARTS ON IBM : ON TPL NO.	None					
WORK REQUESTED							
OBJECTIVE (GIVE PURPOSE OF TEST, WORK AND DATA REQUIRED. INCLUDING SERVICE HISTORY AND BACKGROUND INFORMATION)							
1.0		Rn7 5-065					
To determine the electrochemical techniques and solutions required to remove sharp edges which r in chemical milling and blanking.							
2.0 HISTORY AND JUSTIFICATION							
	Sharp edges left after chemical milling a are presently being mechanically removed cost. During mechanical removal delamina occur, particularly with molybdenum and i Some parts are also complicated in geomet adding to the high cost of removal.	at high tion may ts alloys.					
	The sharp areas would be areas of high cu in electro-chemical operations and should reduced in electrochemical solutions. It necessary to test various solutions for b sharp edges and to select the solution mo for each material.	be easily is reaking					
3.0	O MATERIALS AND SOLUTIONS						
	 3.1 Molybdenum per MMS-184. 3.2 Aluminum 7178-T6 per MIL-A-9180. 3.3 Stainless Steel, 17-7PH per MIL-S-25 per MIL-S-6721. 	043, 321					
	NOTE: Chem-blanked parts of molybde chem-milled parts of aluminum less will be furnished by Dep	and stain-					
	3.4 Electro-Gleam 55 (MacDermid Inc.) 3.5 Electropolish BS (MacDermid Inc.) 3.6 Electropolish BA (MacDermid Inc.)						

PAGE _ REPORT_

Equipment 50 - REV B ADOL ACTUALT

11-62

Del 10. REFERENCES OR ENCLOSURES

NOTE: The above solutions will be furnished by Department 272.

- 3.7 Sulfuric acid alcohol electro polish.
- 3.8 Chromic-sulfuric acid electro polish.

4.0 PROCEDURE

- 4.1 Make up one gallon of each solution as follows:
 - 4.1.1 Solutions of paragraph 3.4 through 3.6 will be made up per vendors literature furnished by Dept. 272.
 - 4.1.2 Sulfuric acid-alcohol polish:

One part concentrated sulfuric acid to seven parts methyl alcohol (95%). Anode current density - 4 ASI. Cathode - Stainless Steel. Temperature - Room Temperature.

- 4.1.3 Nitric acid 50 percent (by weight). Temperature 80-105 F.
- 4.1.4 Chromic-Sulfuric Acid 11.3%, by volume, concentrated sulfuric acid, Chromic acid.17 lb./gal., and the remainder water. Cathode Stainless Steel.

4.2 Chemical Deburring Tests

4.2.1 Chem-deburr the sharp edges of the following masked materials in listed solutions as follows:

Solution	Materials		
Electro-Gleam 55	Molybdenum Stainless Steel Aluminum		
Electropolish BA	Aluminum		
Electropolish BS	Stainless Steel Molybdenum		
Paragraph 4.1.2 4.1.3 4.1.4	Molybdenum Molybdenum Molybdenum		

4.2.2 Immerse the part (anodic) in solution and apply amperage at the middle of the range suggested by vendor literature or rate specified in 4.1.2 for solutions of paragraph 4.1.2 through 4.1.4.

NOTE: Use copper clips for attachment of parts in MacDermid solutions and stainless clips for remainder of solutions. If clips are immersed in solution, a maskant shall be applied to prevent attack of clip.

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- 4.2.3 Vary the amperage, length of chem-deburring time, and concentration of solution until a minimum chem-deburr time requirement is obtained for breaking all sharp edges to an approximate 0.005 inch radius with minimum undercut.
- 4.2.4 Repeat the test as listed in paragraph 4.2 with the maskant removed. Chem-polish the surface until a maximum smoothness is obtained by varying the amperage and solution concentration.

5.0 RESULTS

- 5.1 Current density, temperature, concentration, time, voltage, and calculated area used to produce 0.005 inch radius of each metal, surface finish of unmasked chem-polish surfaces (RMS reading) and metal removal rate.
- 5.2 List any unusual characteristics of solutions noticed while chem-polishing which might produce detrimental effect in production such as temperature increase and effect of materials on maskant. Also, state effective life of solutions without additions.

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