

DEPARTMENT OF THE NAVY

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IN REPLY REFER TO:

Attorney Docket No. 82601 Date: 19 July 2002

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NON-CHROMATE METAL SURFACE ETCHING SOLUTIONS

TO WHOM IT MAY CONCERN:

BE IT KNOWN THAT (1) WAYNE C. TUCKER, (2) MARIA G. MEDEIROS, employees of the United States Government, citizens of the United States of America, and resident of (1) Exeter, County of Washington, State of Rhode Island, (2) Bristol, County of Bristol, Rhode Island; and, (3) RICHARD BROWN, citizen of the United Kingdom and resident of Wakefield, Washington County, Rhode Island have invented certain new and useful improvements entitled as set forth above of which the following is a specification:

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1	Attorney Docket No. 82601
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3	NON-CHROMATE METAL SURFACE ETCHING SOLUTIONS
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5	STATEMENT OF GOVERNMENT INTEREST
6	The invention described herein may be manufactured and
7	used by or for the Government of the United States of America for
8	Governmental purposes without the payment of any royalty thereon
9	or therefor.
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11	CROSS REFERENCE TO OTHER PATENTS APPLICATIONS.
12	This patent application is co-pending with one related
13	patent applications entitled NON-CHROMATE COVERSION COATING
14	(Attorney Docket No. 82602), by the same inventors as this
15	application.
16	
17	BACKGROUND OF THE INVENTION
18	(1) FIELD OF THE INVENTION
19	The present invention relates to a non-chromate metal
20	surface treating composition for increasing the adhesion of a
21	metal's surface to any one of a group of layers applied thereto,
22	such as corrosion resistant layers, and method of applying same.
23	More particularly, the present invention relates to a metal

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trioxide, is replaced with a titanate, namely sodium metatitanate
 or an oxide of titanium, namely, titanium dioxide, respectively.
 (2) DESCRIPTION OF THE PRIOR ART

It is known that solutions containing hexavalent chromium 4 can be used to treat the surface of a metal as etching agents to 5 increase the adhesion of layers which are subsequently applied 6 thereto, such as protective coatings. However, although 7 hexavalent chromium-containing solutions are efficient etching 8 agents, they are also highly toxic and adversely affect the 9 environment and human health. For this reason, many chromate-10 free solutions for treating metal surfaces have been proposed. 11

Thus, various non-chromate metal surface treatments, such as 12 disclosed in Tomlinson U.S. Patent No. 5,759,244, the disclosure 13 of which is incorporated by reference herein, have been disclosed 14 which can increase the adhesion of a metal's surface to a layer 15 subsequently applied thereto. Many of these metal treatments are 16 based on group IV-B metals such as titanium, zirconium and 17 hafnium. For example, U.S. Patent No. 5,868,872 to Karmaschek et 18 al discloses a chromium-free aqueous bath solution for non-rinse 19 treatment of aluminum and its alloys. The solution comprises 20 zirconium and titanium, orthophosphate, fluoride and a water-21 soluble or homogeneously water-dispersible organic film former. 22 When applied, the solution is dried on the surface of the 23 aluminum without rinsing. Similarly, U.S. Patent No. 5,897,716 24 to Reghi et al discloses a chemically and thermally stable 25

chromate-free aqueous liquid treatment for metals which increases
 the adhesion of protective layers to the metals' surfaces. The
 chromate-free aqueous liquid comprises components selected from
 the group consisting of H₂TiF₆, H₂ZrF₆, H₂HfF₆, H₂SiF₆, H₂GeF₆,
 H₂SnF₆, HBF₄, and mixtures thereof.

The shortcoming of conventional non-chromate metal surface 6 treatments, such as those described above, is that they cannot be 7 integrated into and employed in place of chromium-containing 8 compounds in current metal treatment solutions which otherwise 9 would contain chromium. As such, conventional non-chromate metal 10 surface treatments are usually so different from previously 11 employed chromate-containing metal surface treatments that 12 significant changes are required to be made in the metal treating 13 process and in the production of the metal surface treatment 14 These changes can amount substantial expenditures and itself. 15 usually require additional approval from Department of the Navy. 16 Thus, there is a need for "drop-in replacements" that can be 17 employed in place of chromium-containing compounds, such as 18 sodium dichromate, now used in conventional chromate-containing 19 "Drop-in replacement" refers to a metal treatment solutions. 20 compound that can be employed in a metal surface treatment 21 solution in lieu of a chromium-containing compound without 22 requiring any or substantial changes in the make-up of the metal 23 surface treatment process or metal surface treatment solution. 24

SUMMARY OF THE INVENTION

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2	It is a primary object of the invention to provide a non-
3	chromate metal surface treatment solution for increasing the
4	adhesion of a layer, such as a corrosion resistant layer, to a
5	metal's surface wherein the solution contains a titanate or
6	titanium dioxide in place of a chromium-containing compound.
7	It is a further primary object of the invention to provide a
8	"drop-in replacement" for a chromium-containing compound that can
9	be employed in a metal surface treatment solution which otherwise
10	would include chromium.
11	Another object of the invention is to provide a method of
12	increasing the adhesion of a metal's surface to a layer, such as
13	a protective or corrosion resistant layer, applied thereto.
14	The objects of the invention are accomplished by providing a
15	highly effective, non-chromate metal surface treatment solution
16	which includes a titanate, such as sodium metatitanate or
17	potassium titanate, or titanium dioxide in lieu of a chromium-
18	containing compound in a metal surface treatment solution that
19	otherwise would include chromium. More particularly, the
20	invention relates to a non-chromate metal surface etching
21	solution for etching metals, specifically, aluminum, aluminum
22	alloys, stainless steel, titanium and titanium alloys, to
23	increase the adhesion properties of a particular metal's surface.
24	The present invention is developed on the basis of findings
25	that the adhesion of a layer or coating, such as corrosion

resistant coating, to a metal's surface can be increased by 1 bathing a metal substrate in an aqueous solution which contains a 2 chromium-containing compound. Specifically, for example, it is 3 known that a solution containing distilled or deionized water, 4 sulfuric acid, seed aluminum and sodium dichromate dihydrate 5 creates a superb etching solution for aluminum and aluminum 6 alloys. It is further known that a solution containing chromium 7 trioxide and deionized water creates a superb etching solution 8 for stainless steel and titanium. It is believed that the 9 chromium-containing compound in each of the foregoing etching 10 solutions provides increased adhesion to the respective metal 11 surface by providing a contact surface chemistry and allowing for 12 ionic bonding. 13

Test results show that a metal tested without being treated 14 with an etching solution has poor durability and weak boundry 15 layer. For example, untreated aluminum has weak boundry layer 16 and weak oxides; untreated stainless steel has controlled surface 17 properties; and untreated titanium has controlled surface 18 properties. However, since personal exposure limits (PEL) for 19 chromates is 0.1 mg/m³ (milligram per cubic meter), chromate-20 containing etching solutions are not practical for use. Thus, 21 "drop-in replacements" for chromium-containing compounds are 22 needed for etching solutions that otherwise would contain 23 chromium. 24

Sodium metatitanate, potassium titanate and titanium dioxide 1 have been found to be well-suited as "drop-in replacements" for 2 chromium-containing compounds in conventional metal surface 3 etching solutions which typically include, in addition to sodium 4 dichromate, potassium dichromate or chromium trioxide, various 5 other less toxic or non-toxic components. The PEL of the 6 titanium compounds is 15 mg/m^3 , and thus, the solutions provide 7 highly effective, non-toxic, metal alternatives to solutions 8 which otherwise would include chromium-containing compounds. 9

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DESCRIPTION OF THE PREFERRED EMBODIMENT

The present invention will hereafter be described in detail
 with reference to the following embodiments.

The preferred embodiments of the present invention are non-14 chromate metal surface etching solutions for aluminum, aluminum 15 alloys, steel and titanium which include a titanate or titanium 16 dioxide in place of a chromium-containing compound in a metal 17 surface etching solution that otherwise would include chromium. 18 For example, it is known that a solution containing 1 liter of 19 distilled or deionized water, 300 grams of sulfuric acid, 60 20 grams of sodium dichromate dehydrate and 1.5 grams of seed 21 aluminum provides an excellent aluminum and aluminum alloy 22 etching solution. However, as explained above, such chromate 23 containing solutions pose serious health risks. 24

It has now been found that sodium dichromate dihydrate 1 present in the foregoing conventional aluminum and aluminum alloy 2 etching solution can be replaced with sodium metatitanate or 3 potassium titanate without having to alter the various other non-4 chromate constituents therein or the method of employing the 5 solution. Thus, an etching solution for aluminum and aluminum 6 alloys that otherwise would contain sodium dichromate dihydrate, 7 a highly toxic compound, can be rendered non-toxic. 8

In such cases, the aluminum or aluminum alloy to be etched 9 is first bathed in an etching solution comprising distilled or 10 deionized water in an amount ranging about 0.5 liter (L) to 1.5 11 L, sulfuric acid in an amount ranging from about 150 grams (g) to 12 450 g, sodium metatitanate or potassium titanate in an amount 13 ranging from about 10 g to about 150 g and bare aluminum in an 14 amount ranging from about 0 g to about 5 g. The aluminum or 15 aluminum alloy is immersed in the bath from about 5 minutes to 16 about 20 minutes while the etching solution is maintained at a 17 temperature of about 120° F to about 180° F. Immediately after 18 removing the aluminum or aluminum alloy from the bath, it is 19 rinsed by spraying it with tap water for about 5 minutes. This 20 is contrary to prior art methods for applying chromium-free 21 solution wherein the solution typically is not rinsed from the 22 metal but rather is allowed to dry thereon forming a polymer 23 layer. Thereafter, the aluminum or aluminum alloy is soaked in 24 deionized water and then dried at a temperature of about 120° F 25

to about 140° F. Bonding layers to the metal substrate is
performed within about 16 hours of drying.

Similarly, it has been found that titanium dioxide can 3 replace chromium trioxide in a metal surface etching solution for 4 stainless steel and titanium which otherwise typically includes 1 5 part by weight (pbw) chromium trioxide and 4 pbw deionized water. 6 More particularly, etching stainless steel typically requires two 7 baths which include two different solutions. For example, a 8 pretreatment bath or first bath for stainless steel which 9 includes a solution of 2.5 pbw sodium metasilicate, 1.1 pbw 10 tetrasodium pyrophosphate, 1.1 pbw sodium hydroxide, 0.3 pbw 11 nacconol and 95 pbw deionized water is required to clean the 12 stainless steel. A second bath is further required which 13 includes an etching solution containing 1 pwb of chromium 14 trioxide and 4 pbw of deionized water. The present invention 15 provides a "drop-in replacement" for chromium trioxide in the 16 foregoing steel and titanium etching solution. 17

Therefore, according to the present invention, stainless 18 steel to be etched is first immersed in a pretreatment bath 19 including sodium metasilicate in an amount ranging from about 1 20 pbw to 5 pbw, tetrasodium pyrophosphate in an amount ranging from 21 about 1 pbw to 4 pbw, sodium hydroxide in an amount ranging from 22 about 0.5 pbw to 2.0 pbw, nacconol in an amount ranging from 23 about 0.1 pbw to 1.0 pbw and deionized water in an amount ranging 24 from about 90 pbw to 95 pbw. The steel is immersed in the 25

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pretreatment solution for about 5 minutes to 15 minutes while the 1 solution is maintained at a temperature of about 120° F to about 2 Thereafter, the steel is rinsed throughly in water 180° F. 3 before being immersed in a second bath or etching bath which 4 includes titanium dioxide in an amount ranging from about 0.5 pbw 5 to about 6 pbw and deionized water in an amount ranging from 6 about 2 pbw to about 10 pbw. The steel is immersed in the 7 etching bath from about 10 minutes while the etching solution is 8 maintained at a temperature of about 140° F to about 190° F. The 9 stainless steel is then washed in cold running deionized water 10 and dried in a forced-draft oven at less than 140° F. Thus, like 11 the etching solution for aluminum and aluminum alloys described 12 above, the etching solution of the present invention for steel is 13 not dried thereon thereby forming a polymer layer on the surface 14 of the steel. Bonding to the stainless steel's surface is best 15 performed as soon as the metal's surface cools. 16

Etching titanium also requires that the metal be bathed in 17 two baths that include two different solutions. Typically, a 18 first bath containing 400 ml (milliliter) of 38% hydrochloric 19 acid, 40 ml of 85% phosphoric acid and 20 ml of 52% hydrofluoric 20 acid is required to clean and etch the surface to the titanium. 21 Thereafter, a second bath is employed which contains an etching 22 solution comprising 1 pbw chromium trioxide and 4 pbw deionized 23 24 water.

Therefore, according to the present invention, titanium to 1 be etched is first immersed in a first bath including a solution 2 comprising about 350 ml to about 450 ml of a 38% solution of 3 hydrochloric acid, about 35 ml to about 45 ml of a 85% solution 4 of phosphoric acid and about 10 ml to about 30 ml of a 52% 5 solution of hydroflouric acid. Thereafter, it is immersed in a 6 second bath or an etching bath, like the bath for stainless 7 steel, which includes titanium dioxide in an amount ranging from 8 about 0.5 pbw to about 6 pbw and deionized water in an amount 9 ranging from about 2 pbw to about 10 pbw. 10

More particularly, titanium to be etched employing the 11 foregoing solutions is first cleaned with a cloth wetted with 12 trichloroethane in order to degrease the surface. It is 13 preferred that wiping occurs in one direction only. This serves 14 to remove dirt. Thereafter, the titanium is immersed in the 15 first bath or pretreatment bath for about 5 minutes to about 15 16 minutes at a temperature of about 120° F to about 180° F. The 17 titanium is then rinsed thoroughly in water before being immersed 18 in the second bath or etching bath from about 5 minutes to about 19 20 minutes at a temperature of about 120° F to about 180° F. 20 Thereafter, the titanium is washed in cold running deionized 21 water before being dried in a forced-draft oven at 225 \pm 25° F \sim 22 for 1 hour. Again, the etching solution is not dried on the 23 surface of the metal. The bonding surfaces of the titanium are 24 primed within about 4 hours of etching. 25

While the preferred embodiment of the non-chromate metal 1 treatment solution and method of applying same has been described 2 in detail above, various modifications and variations of the 3 invention are possible in light of the above teaching. As an 4 example, the composition of the surface treatment mixtures and 5 the duration of treatments of various surfaces can be varied 6 without deviating from the scope of the invention. It is 7 therefore understood that within the scope of the appended claims 8 the invention may be practiced otherwise and above described. 9

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1 Attorney Docket No. 82601

2 NON-CHROMATE METAL SURFACE ETCHING SOLUTIONS 3 4 ABSTRACT OF THE DISCLOSURE 5 Non-chromate solutions for treating and/or etching metals, 6 particularly, aluminum, aluminum alloys, steel and titanium, and 7 method of applying same wherein the solutions include either a 8 titanate or titanium dioxide as a "drop-in replacement" for a 9 chromium-containing compound in a metal surface etching solution 10 that otherwise would contain chromium. 11