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DIVISION B

NATIONAL DEFENSE RESEARCH COMMITTEE NDRC-21

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OFFICE OF SCIENTIFIC RESEARCH AND DEVELOPMENT

OSRD No. 382

Serial No. 176

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Division B
NATIONAL DEFENSE RESEARCH COMMITTEE
of the
OFFICE OF SCIENTIFIC RESEARCH AND DEVELOPMENT

Report on "The Development of Oil Incendiary Bombs"

by

R. P. Russell, Executive Vice-President
Standard Oil Development Company

OSRD No. 382

Serial No. 176

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Date: February 7, 1942

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Serial No. 176

Division B
NATIONAL DEFENSE RESEARCH COMMITTEE
of the
OFFICE OF SCIENTIFIC RESEARCH AND DEVELOPMENT
Section B-7

Report on "The Development of Oil Incendiary Bombs" (CWS-21)

Endorsement (1) From R. P. Russell, Member Section B-7-D, to E. P. Stevenson, Chairman, Section B-7. Forwarding report.

(2) From E. P. Stevenson, Chairman, Section B-7 to Roger Adams, Chairman, Division B. Forwarding Report and noting:

"Four pound bombs containing gasoline gels appear to be as effective as four pound magnesium bombs for setting fire to wooden structures, such as attics, and markedly superior to four pound thermate bombs. Rubber gels and soap gels are equally good in non-explosive bombs."

(3) Twenty-three copies forwarded to Dr. Irvin Stewart, Secretary of the National Defense Research Committee, as Progress Report under Contract B-204, OEMsr-183 with Standard Oil Development Company.

Roger Adams, Chairman
by Harris M. Chadwell
Technical Aide

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STANDARD OIL DEVELOPMENT COMPANY

C O N F I D E N T I A L

QUARTERLY PROGRESS REPORT ON
"THE DEVELOPMENT OF OIL INCENDIARY BOMBS"

to the

National Defense Research Committee

of the

Office of Scientific Research and Development

Quarterly Progress Report No. 1

S.O.D. Project No. 20907

N.D.R.C. Project No. 199

Contract N.D.R.C.-OEMsr-183

January 20, 1942

I. Abstract

A quarterly progress report outlining the Standard Oil Development Company's work under N.D.R.C. Project No. 199, Contract OEMsr-183 on the development of air-borne incendiary bombs to provide a substitute for the present 4-lb. magnesium bomb may be summarized as follows:

A. The preponderance of official information is to the effect that most incendiaries come to rest in attics and hence many or most fires started by such incendiaries initially were attic fires. In view of this information primary attention has been paid to developing mixtures and methods of application which would cause destructive attic fires. It has been assumed temporarily that a technique successful in this direction would also be satisfactory as regards ability to fire other structures as well. It has been found that attic structures can be destructively burned with the minimum amount of fuel if the incendiary is placed near or discharged into the eaves line where the roof lath and rafters serve as kindling.

B. Petroleum mixtures of 10 to 12 pounds per sq.in. Reid vapor pressure, corresponding to average motor gasoline, give satisfactory ignition characteristics down to -40°F. and can be handled in light weight containers (with proper allowance for outage) up to 150°F.

C. Light petroleum fuels, such as pentane or lighter and unthickened gasoline give rapid, flash burns which are ineffective. Heavy oils, such as gas oil or fuel oil give long burning but too low flames and are difficult to ignite at low temperatures. Gasolines or naphthas having final boiling points of from 350 to 500°F. when properly thickened with rubber or soaps, do not flash burn nor bleed unduly, are long burning, are adhesive and cohesive, and give the most effective destruction for a given weight of fuel.

D. Studies of various thickening agents have shown that (1) if rubber is used, less latex than crepe or smoked sheet is required for a given degree of thickening, (2) crepe rubber is simplest in application since strips of crepe may be immersed in the gasoline and thickening is accomplished without heating or stirring, (3) cold sett gasoline soaps duplicate the burning characteristics of these gasoline-rubber gums, maintain their consistency reasonably well from 150° to -40°F., do not bleed unduly or flash burn, and can be compounded by a simple technique which it is believed can be readily applied in the field.

The most satisfactory gasoline soap, Formula 122, has the following composition:

Stearic acid (or hydrofol)	3.5 wt.%
Rosin	1.8 "
Cottonseed oil (or castor oil)	5.0 "
Caustic soda	1.1 "
Water	2.2 "
Motor gasoline	88.4 "
	100.0 wt.%

E. Burning tests have been carried out in several attic structures in which destructiveness and completeness of any fire are taken as the criteria. Various attic structures have shown the same general conclusions:

1. With a given weight of fuel located at a given distance from the eaves, thickened gasoline is equal to magnesium and is distinctly superior to thermate.
2. Gasoline-rubber gums and gasoline soaps are equally effective.
3. Destructive fires are not started by discharging discreet particles into and adhering onto the underside of the roof, near the ridge pole.
4. A given weight of fuel which causes a destructive fire when fires on a single discreet particle may not start a destructive fire if separated into several smaller particles.
5. Gasoline soaps discharged from the end of a 3 inch diameter bomb into the eaves line may cause destructive fires with as little as 2 pounds of fuel.

F. Preliminary designs indicate that a bomb of approximately 4 pounds total weight will contain approximately 2 pounds of soap thickened gasoline. It is believed that such a bomb will be nearly as effective as the 4-lb. magnesium bomb (1.2 lb. of magnesium) and distinctly superior to the present standard 4-lb. thermate bomb.

G. Further work is planned including the testing of 4-lb. gasoline soap bombs by projecting them from a mortar through typical roof structures to permit final data for design purposes.

II. Introduction

A. Purpose of the Report

The purpose of this report is (a) to outline the progress of the Standard Oil Development Company's work for N.D.R.C. to date (i.e. from October 20, 1941 to January 20, 1942) on development of air-borne incendiary bombs and (b) to present Development's tentative conclusions based on data obtained from Professor L. F. Fieser of Harvard, from Mr. S. W. Adey of the British Petroleum (Warfare) Board, and from Development's own investigations.

B. Statement of the Problem

The primary problem as understood by the Standard Oil Development Company is the development of an air-borne oil type incendiary bomb approaching as nearly as possible the effectiveness of the present 4-lb. magnesium incendiary for firing either (a) debris produced from ordinary wood and brick structures by the action of demolition bombs, or (b) ordinary light construction such as wood or brick walled dwellings, shops, warehouses, and light factory construction.

The problem is understood to be further sub-divided into (a) the earliest possible development of an immediate substitute for the 4-lb. magnesium bomb and replacement for the thermate bomb until late 1942, at which time adequate supplies

of magnesium are expected to become available, and (b) development, if possible, over whatever development period is necessary, of an oil type incendiary bomb as a satisfactory long-term replacement of or supplement to the 4-lb. magnesium incendiary bomb.

The preponderance of official information made available thus far is to the effect that most incendiaries come to rest in attics and therefore most of the fires in dwellings which have been started by incendiaries have initially been attic fires. Firing of attic structures, therefore, has been taken as the first objective on the assumption that a bomb of maximum effectiveness for causing destructive attic fires could be made equally effective for firing other parts of the structure or debris as well. It has also been assumed that the attics have been cleared of rubbish as a preliminary air-raid precaution and therefore auxiliary kindling material is absent.

III. Fuel Quality

As shown in Table 1, hydrocarbon fuels have greater calorific values than an equal weight or volume of thermite or magnesium and hence the work thus far has dealt principally with various petroleum mixtures.

A. Ignition Characteristics of Hydrocarbon Fuels

An important requirement of a hydrocarbon fuel for use in an incendiary bomb is that it be readily ignitable under all temperature conditions to which it may be subjected. Hydrocarbon fuels can be ignited and will continue to burn if the fuel temperature is above its fire point. Kerosene, for example, has a fire point slightly above 110°F. and therefore is not suitable for ignition at a room temperature of 60°F. Since the bombs are carried at high altitudes for considerable periods of time, it is quite probable that the fuel temperature in the bomb will approach that of the surrounding atmosphere (see Appendix I).

Temperatures at altitudes reached in bomber flights are as low as -40°F. and therefore the fuel should have a fire point below this temperature. The volatility of the fuel controls the fire point, which is not readily determined directly with precision at low temperatures and accordingly the Reid vapor pressure at 100°F. in pounds per square inch has been adopted as the criterion for the suitability of a fuel for low temperature ignition.

The relationship of Reid vapor pressure at 100°F. in lb./sq.in. to the flash and fire point is given in Table 2, for synthetic blends of butane and kerosene. The best information on the required R.V.P., however, is determined by filling a bomb with the fuel in question, cooling to -40 or -60°F. and exploding the bomb. Such tests (see Appendix IV) indicate that a fuel is required with a R.V.P. of 10 to 12 lb./sq.in. at 100°F. Winter grade motor gasoline is the most generally available fuel for meeting this requirement. Aviation gasoline (7 lb./sq.in. R.V.P.) is not considered to be suitable.

B. Time-Temperature Relationships for Burning Fuels

In order to evaluate the various fuels employed in this work with regard to flame duration and maximum temperature, a test apparatus was constructed which essentially comprised eight thermocouples suspended by means of

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Table 1

Heat of Combustion of Various Fuels

	Basis-Pure Fuel		Basis-4 Lb. Bomb		
	Btu/Lb.	M Btu/cu.ft.	Btu/Bomb	Btu/Lb. of Bomb	M Btu/cu.ft. of Bomb
Thermite	1,500	244*	2,340	624	72
Aluminum	13,000	2210			
Magnesium	10,780	1158	14,115	3600	470
Pine Wood	9,200	310			
Coal	14,000	840			
Fuel Oil-12°API	18,400	1130			
Motor Gasoline	18,800	880	39,600***.9900		652**

* Based on packing at 4000 #/sq.in.

** Assumed to be tailless to be equipped with streamers.

*** Est. at 2.2 lb. gasoline-soap per bomb.

Table 2

IGNITION QUALITIESRELATIONSHIP OF REID VAPOR PRESSURE TO FLASH AND FIRE
OF UNTHICKENED FUELS

		Reid Vapor Pressure	Flash* Point	Fire* Point
25% n-butane, 75% Kero.(Varsol No.1)		16 lb./sq.in.	-40°F.	-40°F.
20%	" 80%	" " 15.2	-40	-40
15%	" 85%	" " 12.5	-40	-40
10%	" 90%	" " 8.0	-40	-40
5%	" 95%	" " 4.8	-40	+20
0%	" 100%	" " 0	+105	+115
Motor Benzol (90% Benzene)		3.5	+5**	+24**
Light P.C. Naphtha		8.0	-40	-40

* Open cup flash and fire test.

** Solid benzene present and temperatures may be in error.

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Figure 2 - FUEL BURNING RATE-TEMPERATURE RELATIONSHIPS FOR THICKENED PAPER

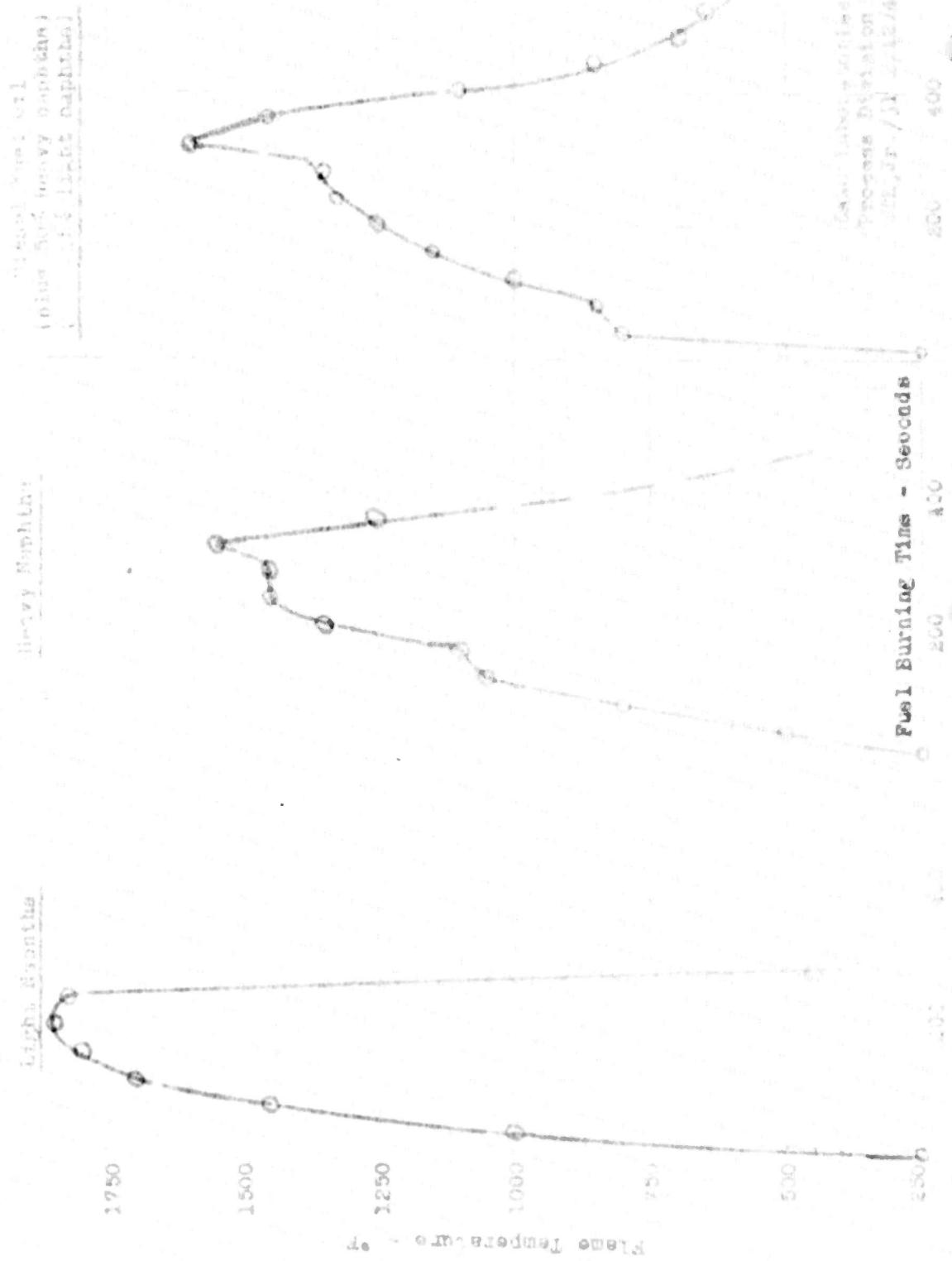


Figure 3

TIME - TEMPERATURE RELATIONSHIPS FOR LATEX-THICKENED FUELS

2000

4% Latex in Light Naptha

4% Latex in Heavy Naptha

4% Latex in Diesel Fuel Oil

Flame Temperature - °F.

1750

1500
1250

1000

750

500

250

200

400

200

400

200

200

400

Fuel Burning Time - Seconds

ESO Laboratories
Process Division
WTK,Jr./J1 1/12/42

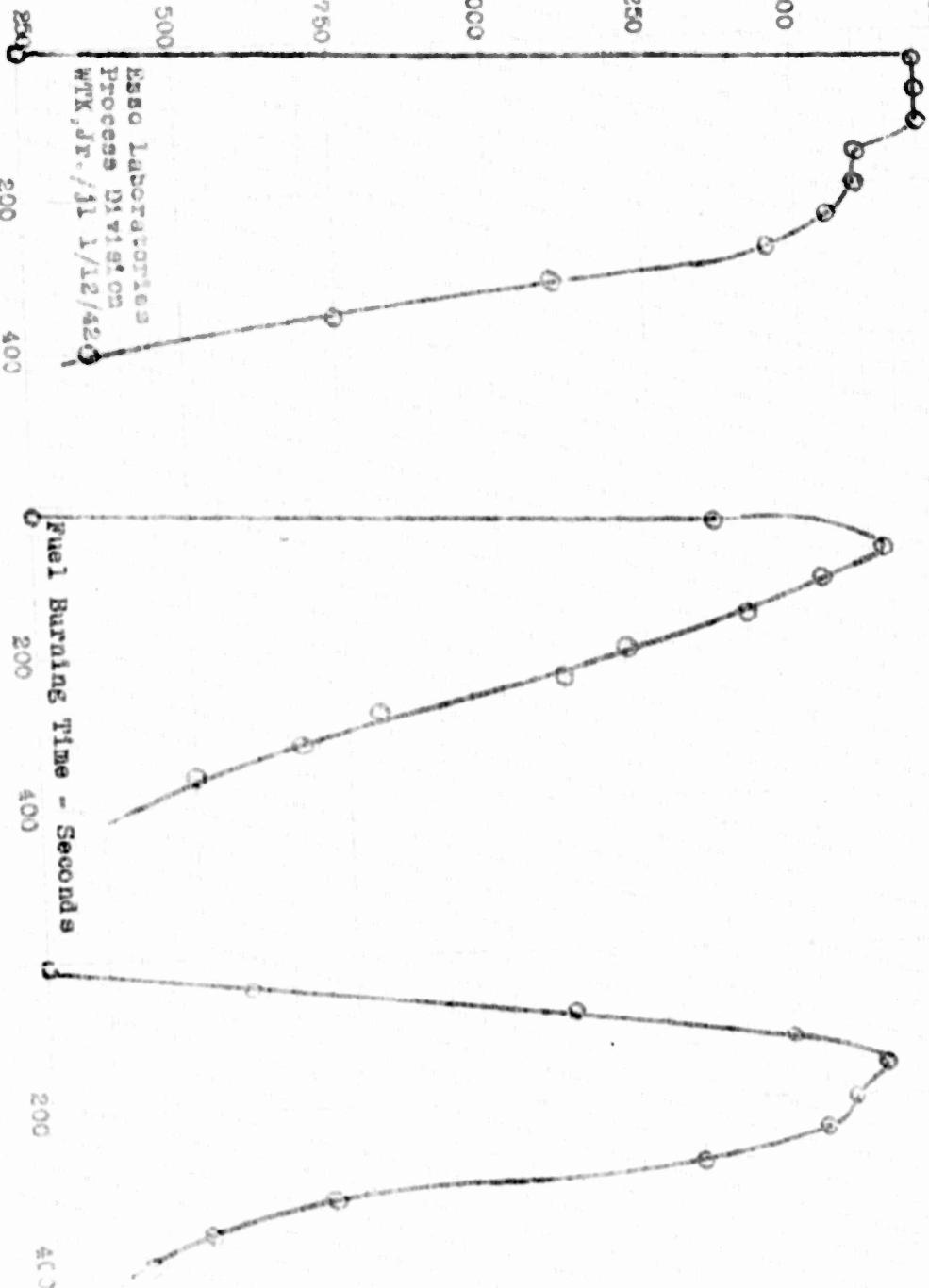
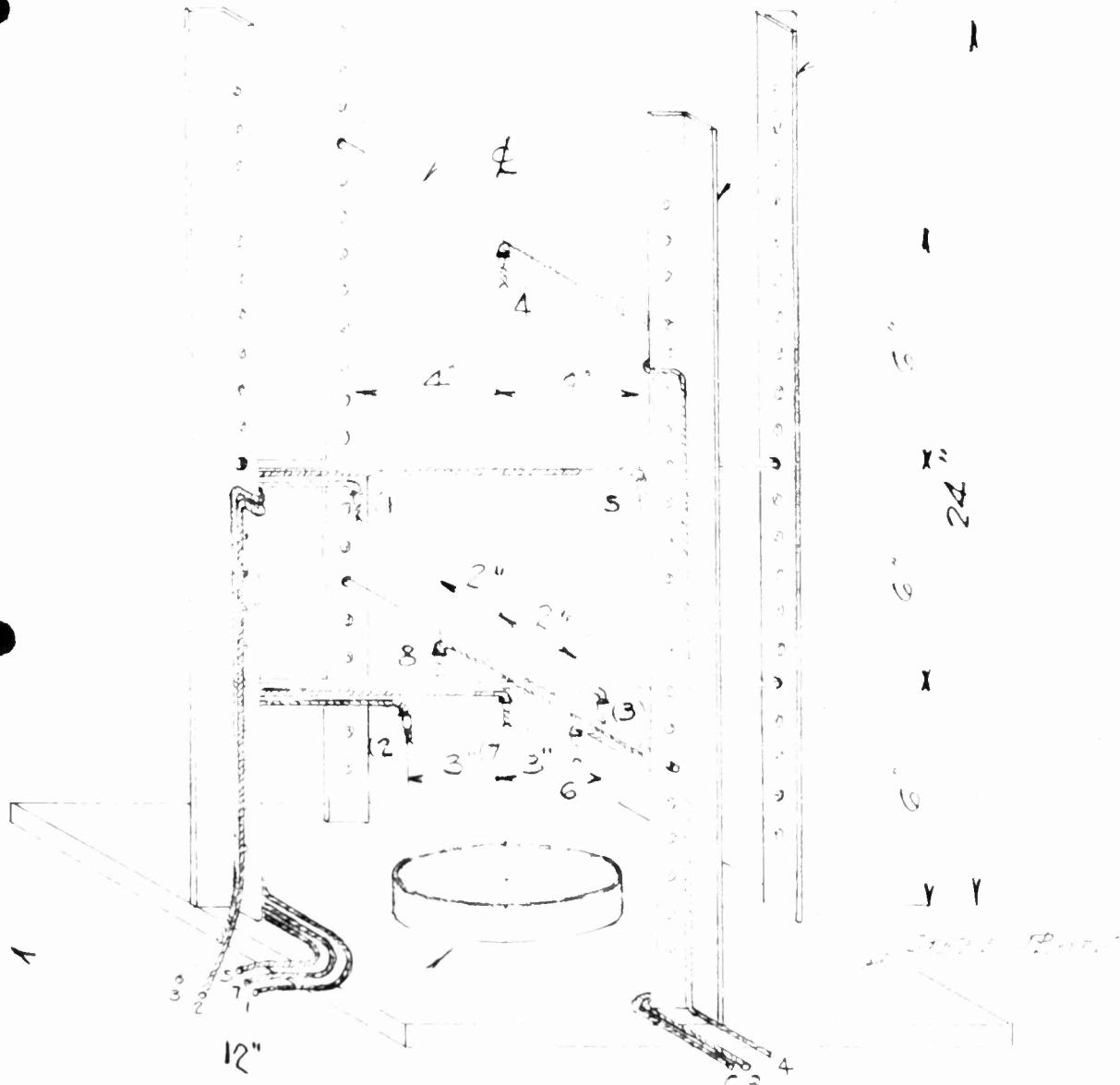


FIG 1



APPARATUS FOR MEASURING
TIME-TEMPERATURE-RELATIONSHIPS
OF BURNING FUELS

a steel framework at varying distances above an open dish containing the burning fuel (see Figure 1). A cylindrical shield mounted on legs protected the flames from stray drafts. The dimensions of the steel framework were identical with those of the wooden test structure used by Professor L. F. Ficser at Harvard University. Temperatures were recorded by photographing meters connected with the thermocouples.

The lighter hydrocarbon fuels, such as gasoline, gave appreciably higher flame temperatures for a shorter duration than did the heavier oils, such as Diesel fuel. This was accompanied in the case of the lighter fuels by a large flame volume, whereas with the heavy fuels the flames tended to remain small and low. In Figure 2 are shown typical data obtained when burning (1) a light naphtha (120-280°F.B.P.) fraction, (2) a heavy naphtha (300-400°F.B.P.) fraction, and (3) a Diesel fuel oil (400-600°F.B.P.) blended with 50% gasoline for ignition purposes. The relatively high temperatures (1800°F.) and short burning time (250 seconds) associated with the light naphtha, and the long burning time (500+ seconds) and relatively low temperatures (1300°F.) of the Diesel oil, as well as the intermediate characteristics of the heavy naphtha fraction are clearly evident.

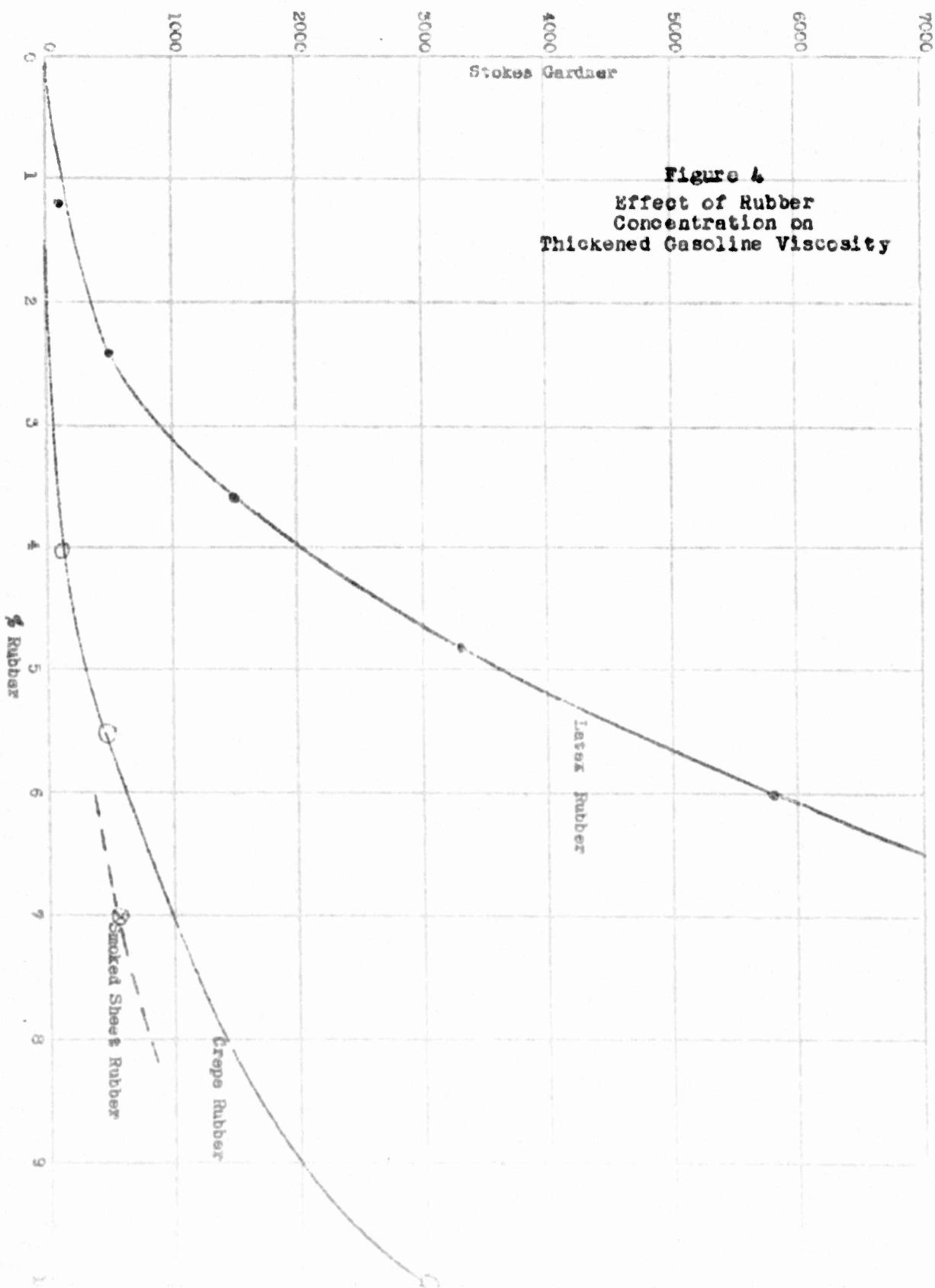
In an attempt to increase the burning time of the light naphtha fraction without sacrificing any of its high temperature advantage, various materials, such as latex, crepe, and smoked sheet rubber, were used as thickening agents. Figure 3 shows the data obtained when burning (1) 4% latex in light naphtha, (2) 4% latex in heavy naphtha, and (3) 4% latex in Diesel fuel oil. By comparison with Figure 1, it will be seen that the addition of 4% latex increased the active burning time of the light naphtha by 20-30%, with little decrease in flame temperature. The effect of the latex on the Diesel fuel was quite different from its effect on the light naphtha; the heavier oils were made to burn faster, probably due to the wick action of the gel structure. However, it is significant that these heavier fuels did not produce as hot flames as the light naphtha, even though their burning rate was considerably increased. On the basis of these data and similar tests employing crepe and smoked sheet rubber, and bearing in mind supply considerations as well as the fuel characteristics necessary for ignition, a petroleum fraction boiling from about 100-400°F. (regular motor gasoline) has been selected as the best base stock for petroleum incendiary bombs.

IV. Fuel Thickeners

Early burning tests had shown that thickened gasolines had good ignition characteristics and gave long burning, high temperature flames. Thickened fuels also possessed the advantage of being cohesive, thus preventing flash firing and shattering when ejected from a bomb.

A. Rubber Gums

The British had investigated the merits of benzene thickened with rubber and further work has been done on rubber gums at Harvard. On the basis of this early work, two standard gums were prepared by dispersing smoked sheet rubber in benzene and naphtha for burning studies.



Work was next undertaken on the use of latex since latex gums required less agitation for solution, and less rubber was required for a given consistency. Latex gels benzene on merely stirring into solution. However, in more paraffinic solvents, an acid is required as a coagulant. Oleic acid was first used usually being neutralized to utilize the soap as a further thickening agent. It was found, however, that less acetic acid was required, so this was used in the later latex work. The processing was as follows: the warm (70°F.) gasoline was stirred rapidly with a propeller while the latex was added. The acid was then added, and stirring continued until thickening had ceased, not over 5 minutes being required. The alkali (in 3%-1/3% solution) was then added, and the batch stirred slowly until homogeneous. The alkali caused a great increase in viscosity. This produced a satisfactory gum when the following conditions were observed:

1. The temperature must not be too low, as this slows coagulation.
2. The time of stirring must be limited, and the speed not too great, to prevent permanent depolymerization of the rubber.
3. A moderate (10%) concentration of aromatics is required to speed coagulation.
4. The gasoline must be doctor "sweet" as reactive sulfur compounds cause partial loss of viscosity in a few days.

As is shown in Figure 4, the latex blends require much less dry rubber than those made from processed rubber. Like all rubber blends, they separate slightly at -40°F., especially those coagulated with oleic acid. Those containing sodium acetate cause slight surface rusting of steel, in spite of the free alkali present.

At this time, the advantages of crepe rubber were disclosed by the work at Harvard. Because of the porous nature of this form of rubber it disperses readily on merely standing in the solvent. As a result of its high molecular weight, a 5.5% blend is equal in viscosity to 7% smoked sheet. The investigation of this material disclosed several pertinent facts. Swelling is much more rapid in aromatic fuels than in paraffinic ones. Temperature is an important factor, swelling being extremely slow below 40°F. It is difficult to obtain homogeneous mixtures by swelling small percentages of crepe rubber in fuel, but homogeneous mixtures may be made by cutting back concentrated mixtures in a kettle with paddles. The product is intermediate in structure between latex and smoked sheet blends, having some of the gel-like body of the former.

The shortage of rubber closed the above lines of investigation, and tests were made on reclaimed rubber. This material proved entirely unsatisfactory, since it is composed of broken down rubber and vulcanized material, the former having little or no thickening power, and the latter does not disperse. In view of the inadequate supply of synthetic polymers, no investigation has been made of these thickeners.

B. Soap Thickened Gasolines

The first soaps considered for use as thickeners to replace rubber were the stringy aluminum ones. Aluminum naphthenate, oleate and stearate were tested, but proved very difficult to manufacture. All required compounding at about 250°F., though when the soap content was reduced to about 20% they could be mixed at about 150°F. This mixing required the use of pressure equipment, or the addition of considerable heavy oil. The alternative method of cold ball milling would be very cumbersome. The soap must be made by double decomposition and should be well washed for satisfactory results. All of these soaps have been found unreliable and difficult to use in grease making. In addition, the temperature susceptibility proved worse than that of rubber blends causing the gum to melt and flash on ignition and to become very hard at -40°F. For these reasons, the investigation of aluminum soaps was dropped.

Calcium soaps were also considered though the product was not at all like a rubber blend. It is an old process to gel oils with a mixture of lime and rosin oil to produce "Sett Greases", and this could be done with gasoline. However, this type of soap would be very thixotropic, and the gel completely atomized on explosion. About 25% of non-fuel material would be required.

Tests were also made on batches prepared by cutting back ordinary cup greases (made from the cooked soap of pork fat) to a heavy oil consistency. These products were found to be too fluid, and gave a flash burn on explosion.

A process was developed for making a sett type of grease from fatty acids. This type was much less thixotropic than the rosin oil mixtures, but more difficult to compound. A gel was made with 15% preformed soap present, and the necessary lime was dispersed in it. This grease was then dispersed in gasoline by stirring, after which the fatty acid was added. Setting took place in about two hours. The non-fuel content was about 20%. This type will be investigated further.

The tar-lime gels used by the British have not been investigated, because the vertical gas retort coal tar required is not produced in sufficient quantities in this country.

The melted type of sodium soap gel was studied by several investigators and showed several advantages; it was easy to prepare, was of low non-fuel content, and melted sharply to a thin fluid so that it could be poured into cases. Such a material, however, was a brittle gel which tended to bleed gasoline, and required heating to about 130°F. during compounding. This required either pressure equipment or the use of a high flash, low-vapor pressure fuel. Large amounts of alcohol were also required to lower the melting point, and no readily available substitute was found. This gel was also found to shatter badly on explosion, giving a flash burn unless a binder was added. For these reasons work on this type has been suspended.

The most satisfactory soap thickener was a cold compounded sodium soap gel (Formula 122) made from a mixture of stearic and rosin acids, plasticized with a fatty oil. It had a grease-like texture, was only moderately thixotropic, and had excellent temperature characteristics.

The processing was as follows: the rosin, stearic acid, fatty oil and about 6% heavy naphtha were melted together at about 200°F. This base was a soft solid, melting at about 80°F. About 15% was dissolved in a gasoline for the final compounding, and the temperature adjusted to 90-110°F. About 4% of 53-1/3% aqueous caustic soda was then added, and dispersed by shaking or stirring. The mixture set in a few minutes to a semi-solid and hardened gradually in the next 24 hours. Many modifications of this formula have been tested, in an effort to increase the time of set, to widen the mixing temperature range, and to lower the melting point of the base. On the basis of these tests, the following conclusions were made:

1. The best grade of stearic acid for this purpose was Hydrofol, a hydrogenated fish oil acid.
2. A good grade of lump rosin was required.
3. The best proportions were two of stearic to one rosin.
4. Castor oil was the best plasticizer, though other fatty oils could be used.
5. Increasing the excess of caustic soda improved the stability slightly.
6. Replacing a large part of the caustic soda with borax gave a longer setting time, and may have broadened the mixing temperature range.

Further work is planned to determine (a) the optimum ratio of reactive ingredients, (b) the ability to substitute various fatty acids and fatty oils on the basis of availability, (c) a means for slowing down the setting time to permit easy field filling, and (d) methods for adjusting the mixing temperature to permit a wider temperature range than 90-110°F.

Formula 122 represents a solidified or thickened gasoline with the following desirable features: (a) simple and easy compounding to permit filling in the field so that empty and unarmed bomb cases can be transported; (b) the gasoline retains its good ignition characteristics even down to -40°F.; (c) the cohesive characteristics are greatly improved being about equal to that obtained with rubber gums; and (d) the burning quality of the gasoline equals that of any thickener studied. Formula 122 has the following composition:

Stearic acid (or hydrofol)	3.5	wt.%
Rosin	1.8	"
Cottonseed oil (or castor oil)	3.0	"
Caustic soda	1.1	"
Water	2.2	"
Motor gasoline	88.4	"
	100.0	wt.%

Table 3

AVAILABILITY OF INGREDIENTS REQUIRED
BY FORMULA 122

Material	Million Pounds Per Year Produced in U. S. A.				Million Pounds per Year Required to Produce 9,000,000 4-lb.* Bombs/Month
	1935	1937	1939	1940	
Hydrofol	--	--	--	4.8	{ 7.5
Steric Acid	25	31	35	41	
Rosin (Gum)	800	850	--	650	{ 3.8
Rosin (Wood)	300	420	--	450	
Cottonseed Oil	--	1678	1390	--	{ 6.5
Castor Oil	--	52	76	--	
Fish Oil (Total)	--	163	210	--	
Crustic Soda			2000		2.4

*) Based on 2-lb. of Formula 122 per 4 lb.-Bomb.

A suggested method for compounding and filling Formula 122 into bomb cases in the field is described in Appendix II.

The availability of the ingredients required for Formula 122 is presented in Table 3, and appears to permit production of 9,000,000 bombs (4-lb.) per month.

A discussion of physical testing and the results of investigating various gasoline thickeners is presented in Appendix I.

V. Burning Tests in Attic Structures

A. Experimental Equipment and Procedure

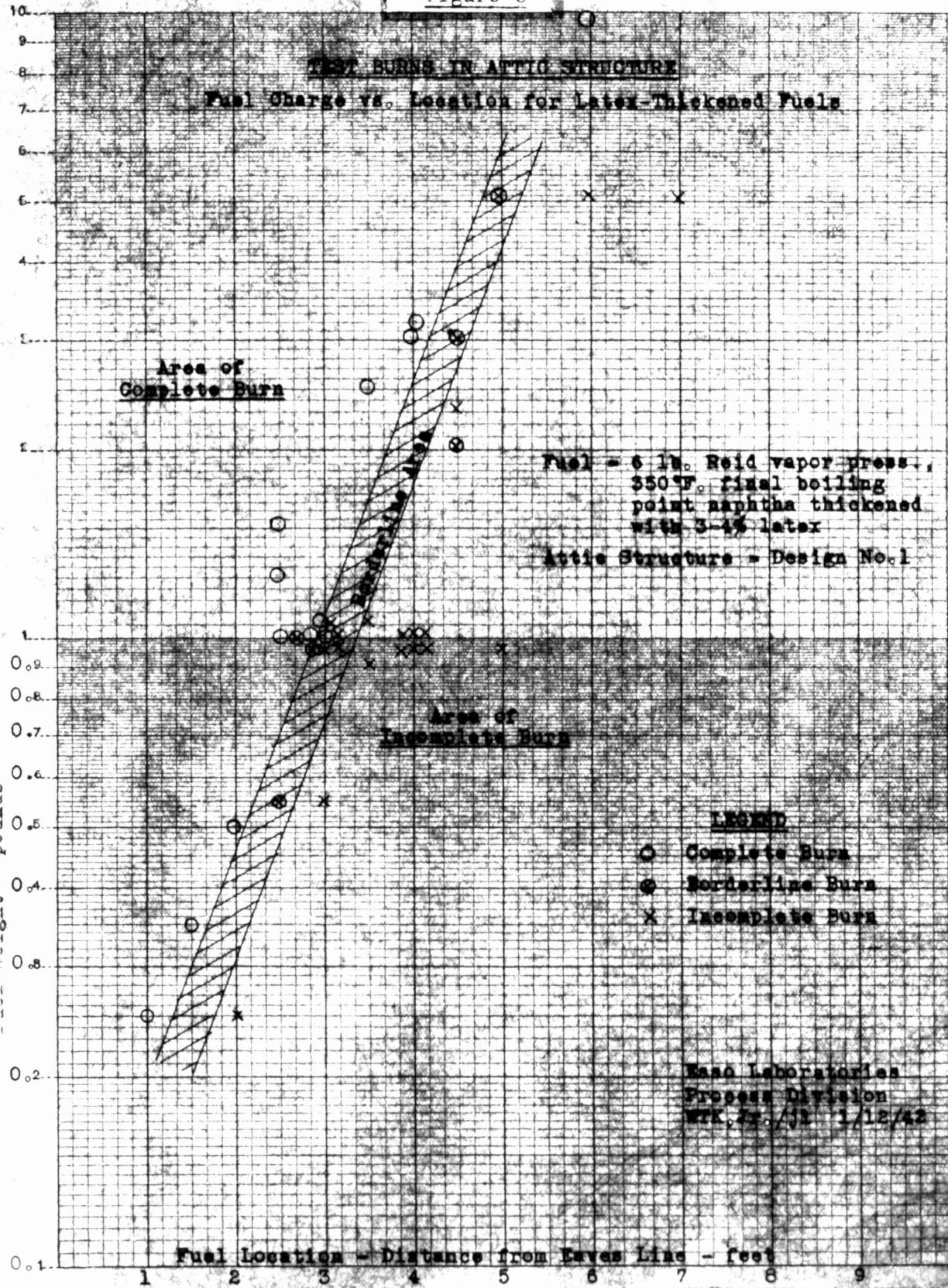
In order to evaluate the burning characteristics of various fuels, wooden test structures were constructed simulating a section of an attic. Two general types of attic structure were employed -- Design No. 1 ("American" type) was built with 2x6 joists and 2x4 rafters on 16 inch centers, and wooden boards underneath the joists; Design No. 2 ("European" type) was constructed with 2x6 joists and rafters on 24 inch centers, and a plastered ceiling underneath the joists. A non-combustible roof of transite or sheet metal, simulating corrugated tile, with regular shingle lath on 6-7/8 inch centers were employed. Drawings of the test structures are shown in Appendix Figures A and B.

The majority of tests were made on this small structure, open on three sides, representing that section of an attic roof and floor within 6 ft. of the eaves lines. Some tests were made on a double structure formed by placing two single structures together in order to obtain the effect of a complete roof. Large scale tests were conducted in a structure closely approximating in size a typical attic (actual dimensions - 28x14x7 ft.). A replaceable wooden center section 10 ft. long was actually employed for test purposes; 9 ft. long metal end sections were attached merely to provide an air volume approximately equal to that of a normal attic. Test burnings in this large-scale structure were generally carried out with all doors closed in order to resemble the actual conditions existing in a typical attic.

The procedure employed for making test burns consisted of placing the fuel charge on the flooring at a given distance from the eaves, lighting the fuel, and at the completion of the burn determining the extent to which the flames had damaged the test structure. Adequate ventilation free from drafts was maintained. In some cases, the fuel charge was flung or smeared against the roof laths and rafters, in thick instances a portion of the burning fuel usually dropped to the floor and continued burning there. Tests were made in which the fuel charge was blown from a 3 inch diameter bomb casing towards the test structure. For the purposes of these tests, a "destructive" burn was defined as one in which either (1) the flames were definitely increasing at the end of ten minutes burning time, or (2) the structure was burned to such an extent that it collapsed.

Burning tests involving magnesium and thermite were generally carried out employing a double thickness (1-5/4 inches) of floor board. It was

Figure 5



KEUFFEL & ESBER CO., N.Y. NO. 350-60
Semi-Logarithmic, 2 cycles, 60 divisions

recognized that this test method measured solely the destructive effect of these incendiaries in the attic structure. However, it was found that the burning properties of the magnesium and thermite bombs in the attic test structures were considerably aided by the double wooden flooring. No attempt has yet been made to determine the effectiveness of various incendiaries as regards burning through several floors of a building.

B. Burning Tests Employing Rubber Gums

At the outset of work on this problem the supply of the various types of rubber was not critically limited; consequently, a large number of tests were made using rubber gums made from latex, crepe rubber, and smoked sheet rubber added to various fuels. Detailed data for these tests are presented in Appendix III.

The relationship existing between size of fuel charge and its distance from the eaves for petroleum gums containing 3-4% by weight of latex (equivalent to 1.8-2.4 wt.% rubber) is shown in Figure 5. No appreciable deviation from this curve was noticed even when using such different fuels as pentane and diesel fuel oil, although normal motor gasoline appeared to possess the optimum boiling range. This curve clearly demonstrates the advantage gained by placing the fuel charge as near as possible to the eaves as, for example, from a tail-ejection bomb.

A significant difference was noted in the behavior of fuel charges below one pound in size and those above two pounds. When using the larger fuel charges it was found that the degree of destructive burning was considerably greater than with the smaller charges. Thus, it was observed that in the majority of cases a destructive burn resulting from a one pound charge was confined to one or possibly two sections of the floor and roof; whereas, a destructive burn caused by a 3 pound charge generally led to a collapse of the test structure.

The burning characteristics of petroleum gums made with crepe and smoked sheet rubber were essentially identical with those of latex gums, as evidenced by the data presented in Appendix III. Latex gums were much less tacky than those made from crepe or smoked sheet rubber, and consequently, did not adhere to roof surfaces as well as the latter.

In an attempt to evaluate the property of adhesivity as applied to gum incendiaries, numerous tests were made employing latex, crepe, and smoked sheet rubber gums in which all or a portion of the fuel charge was scattered or thrown against the roof. In no instances, regardless of the fuel distribution between roof and floor, was there a destructive burn, although had the fuel charge been placed on the floor at an equal distance from the eaves, in most cases a destructive burn resulted. These data indicated, therefore, that the maximum effectiveness of an incendiary material in the attic was realized when placing that material on the floor instead of spreading the fuel over the roof and floor. From this it follows that adhesivity is not a controlling factor in developing suitable petroleum incendiaries. Preliminary tests (Appendix III, Runs H-41, 51, and 52) also have shown that the maximum effectiveness of a given weight of material resulted when that material was held together as a single large particle rather than segregating it into several smaller units.

Figure 7

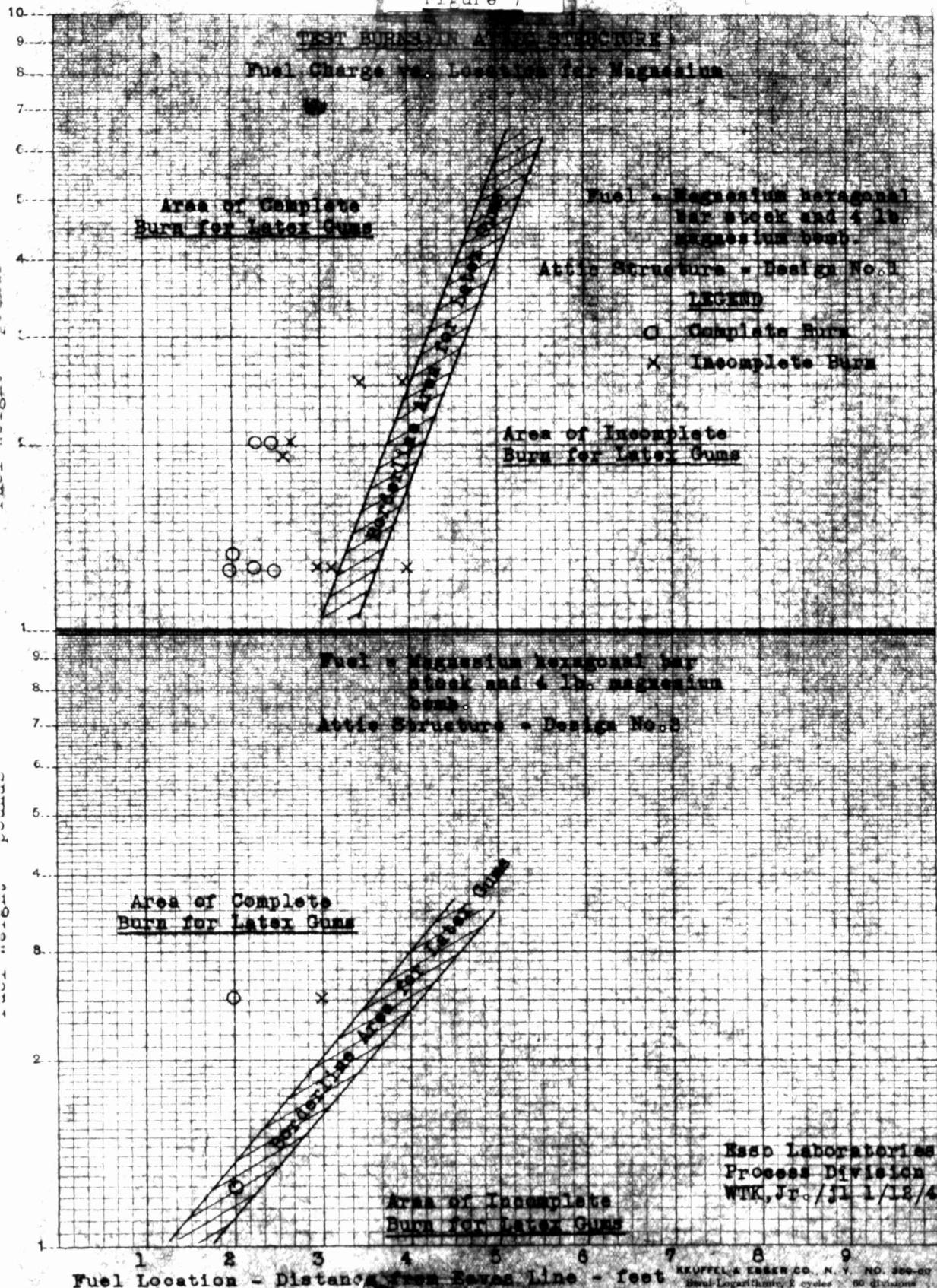
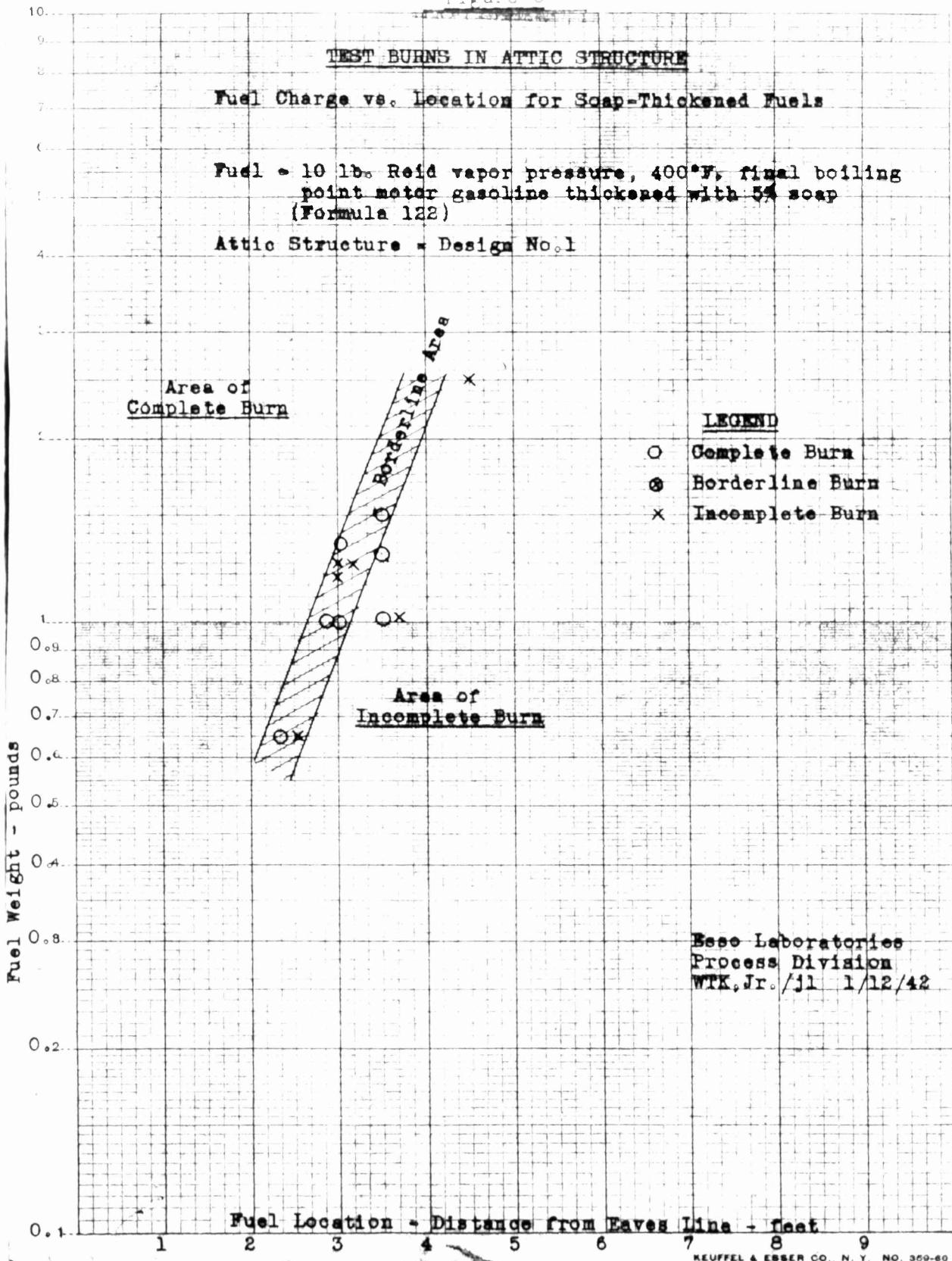


Figure 6



C. Burning Tests with Gasoline Soaps

The limited supply of rubber in the United States and its increasingly difficult procurement after December 7, 1941, necessitated the development of a non-rubber thickening agent. Experimental work was undertaken, therefore, on the use of various sodium stearate-eleate-resin soaps in regular motor gasoline. Detailed data from the burning tests of these various soap-thickened gasolines are presented in Appendix III.

The relationship existing between weight of fuel charge and its distance from the eaves for gasoline soaps is shown in Figure 6. In general, gasoline soaps were equivalent to latex gums, regardless of whether the soap was produced as a relatively hard gel or as a soft gel similar in consistency to apple-sauce. However, a significant difference in burning characteristics was noticed between rubber gums and the gasoline soaps. Whereas the rubber-thickened gasolines gave a fast burn with large flames from the start, the soap-thickened gasolines usually burned slowly with moderately small flames for about one minute after the start of burning. Thus, in order to obtain the same degree of destructive burning, the active burning time of the gasoline soaps generally needed to be of about 4-7 minutes duration, as compared with the 3-4 minutes active burning time for the rubber gums.

It was expected from the nature of these soap-thickened gasolines that there would possibly be a greater tendency to "bleed" gasoline or undergo syneresis than in the case of the rubber-thickened gasolines. Consequently, an extensive program was undertaken to investigate the effect of various binding agents, such as sisal, excelsior, and cotton waste, on the cohesion of the gasoline-soap gels. Detailed data from the burning tests together with a few ejection tests of these soaps with binders are presented in Appendix III. In general, the addition of binding material to soap-thickened gasolines increased the active burning time of the fuel and simultaneously reduced the flame volume. Cotton waste was the most effective in this respect, and sisal the least effective. However, numerous tests in which gasoline soaps containing sisal were flung against the roof of the test attic structure indicated that sisal imparted nearly equally effective cohesive properties to the soap gel as did the cotton waste. As reported in the following section, ejection tests from 3 inch bombs have shown the desirability of including 1-2% of binder in order to hold the fuel charge together during firing. Recognizing that binders may be difficult to incorporate in field compounding, further tests are being conducted to determine if binders are necessary.

D. Burning Tests with Magnesium

In order to compare gasoline soaps with the present 4-lb. U.S. and British magnesium bomb, tests were made in the standard attic structure with magnesium bombs and with hexagonal magnesium bar stock (ignited by means of a welding torch). Detailed data for these tests are reported in Appendix III, and are summarized in Figure 7.

The results show that magnesium on wooden flooring (Design No. 1) was nearly as effective as soap-thickened gasoline when used in quantities of

approximately one pound, but that magnesium was not as effective as gasoline soap when employed in amounts greater than 2 pounds. It was found that magnesium damaged the wooden floor of the test structure much more intensively but less extensively than petroleum fuels. However, although the magnesium bomb burned through 1-3/4 inches of wooden floor, it failed to burn through 1/2 inch thick plastered attic floors simulating the ceiling of a room below the attic. When using a plastered floor (Design No. 2) magnesium appeared to be even less effective. These tests indicate that thickened gasolines are at least equal to magnesium.

E. Burning Tests with Thermite

Detailed data obtained from several test burns employing the regular U.S. Army thermite incendiary bomb are presented in Appendix III. The conclusion reached regarding the burning characteristics of thermite was that it appeared to be quite ineffective in starting a destructive burn in the attic test structure, although it burned through either 1-3/4 inches of wooden flooring or 1/2 inch of plaster. This was illustrated by Run H-173, in which the thermite bomb was placed directly against the eaves board of the test structure. Some localized damage was done, but no self-propagating fire resulted. This indicated that the incendiary effect of thermite is almost entirely downward and very localized, as compared to the relatively wide-spread, upward effect of oil incendiaries. These tests indicate that in the attic structure the thermite bomb is quite inferior to either the magnesium bomb or the thickened gasoline.

VI. Bomb Ejection Tests

The technique of carrying out explosion tests by ejecting thickened fuels from bombs, has gone through several changes since the start of the project. The original test bomb consisted of a 5-lb. pail at the bottom of which was placed a powder chamber (4" I.D. x 1/4" high) with a lacquered cardboard diaphragm to keep the powder and fuel separated. A 6 in. length of 1/4 in. pipe extended from the powder chamber to the outside of the pail and a safety fuse was inserted through this pipe. The pail was approximately 6 in. diameter by 5-1/2 in. high and held 3 lbs. of fuel. The bomb was placed in a vertical position, was exploded usually with 20 gms. of black powder, sporting grade, and the degree of fuel scattering and ignition was noted. Usually the fuel was scattered in small pieces over a radius of 20 ft. and the ignition was excellent down to -40°F. if a fuel of 10 lb./sq.in. or higher R.V.P. was used.

A markedly different result as to scattering was obtained (Appendix IV, exp. 14) if the bomb was placed on its side. The fuel could be heaved a considerable distance without excessive shattering. Some improvement in fuel cohesion was obtained by adding cotton waste, excelsior, sisal, etc., as a binder to the fuel, although the necessity for binders has not been firmly established.

Latex blends proved to be very difficult to ignite when ejected from the 3 in. diameter bombs with the powder chamber located at the bottom. Coarse magnesium turnings and coal particles mixed into the powder did not

TABLE 4
COMPARISON OF THERMAL EFFICIENCIES

Type Unit	Gross Wt. of Unit	Gross Wt. of Fuel	Per Cent Fuel to Gross Wt.	BTU Per Unit	BTU Per Lb. of Unit	Sectional Density
4-lb. Mg.	3.91 lb.	1.86 lb.	47.6%	14,115	3,600	1.58
4-lb. Thermite	3.75	1.55	41.3	2,340	624	1.41
4-lb. SOD	5.5	2.00	36.4	36,000	6,540	.778
STD. Fuze 20 Ga. Case						
4-lb. SOD New Fuze	4.48	2.00	44.7	36,000	8,030	.634
20 Ga. Case						
4-lb. SOD STD. Fuze	4.64	2.00	43.2	36,000	7,760	.656
28 Ga. Case						
4-lb. SOD New Fuze	4.0	2.20	55.1	39,600	9,900	.566
28 Ga. Case						

BTU/Lb. Mg. = 10,780
BTU/Lb. Thermite = 1,500
BTU/Lb. SOD = 18,000

Sectional Density of German 1 Kilo Mg. Bomb = .700

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help ignition. Later it was found that a long narrow tube (3/8 in. radio spaghetti tubing) completely filled with powder, placed coaxially in the bomb, produced good ignition. Formula 122 fuels are not as difficult to ignite as the latex fuels and the importance of the coaxial powder tube is not as great with the gasoline soaps.

The results of all the explosion tests are given in Appendix IV and may be summarized as follows:

1. Gasoline with a Reid vapor pressure of at least 10 lb./sq.in. at 100°F. is required to insure ignition at the low temperatures encountered at high altitudes (-40°F.).
2. Hard setting gasoline greases are not suitable because they shatter excessively and flash-burn in mid-air.
3. Rubber gels (5.5% crepe) and soft non-bleeding gasoline soap (Formula 122) give the best results because they do not shatter unduly and give desirable burning characteristics.
4. Binders such as sisal and excelsior while important in the scattering type of bomb, are not quite as important in tail-ejection bombs where the scattering of the fuel is reduced. However, sufficient work has not yet been done to show whether binders can be excluded.

VII. Design and Testing of Small Petroleum Incendiary Bombs

Test data have shown that the most effective method of burning the test attic structures consisted in placing the fuel charge as near the eaves line as possible. In order to assure the placing of the fuel near the eaves, a bomb has been constructed in which the fuel is ejected from the tail after the bomb has come to rest. The advantages of having the incendiary material thrown against surrounding objects are obvious.

Preliminary design studies have shown the desirability of using a bomb casing of minimum wall thickness in order to obtain the maximum fuel content. However, it is understood that the Army Air Corps would be opposed to case thicknesses less than 20 gage (.035 in.). It is also understood that any change in nose fuse design over those now approved for the 4-lb. thermite or the M-46 and M-47 (M 108) bombs might result in considerable delay in obtaining final approval for a new incendiary. Consequently, two bomb designs are now under development, both being of the tail ejection type approximately 15 in. long and 3 in. diameter with a cloth streamer to help control ballistic and terminal velocity characteristics. The first design utilizes a 20 gage case with the standard thermite nose fuse (see Appendix Figure C) with an estimated gross weight of 5.5 lb. and containing 2.0 lb.

of fuel (Formula 122). The second design will incorporate the minimum wall thickness consistent with good impact resistance together with a new light weight fuze design (see Appendix Figures D and E). Tests are under way to determine this minimum case thickness, but if 28 gage material can be used, then this bomb would have an estimated gross weight of 4.0 lb. and would contain 2.2 lb. of fuel. This latter design lends itself to methods now used in can production. As shown in Table 4, the 28 gage case and new fuze design improve the ratio of fuel to gross weight, and either design of oil bomb shows an advantage over the magnesium and thermite bombs when compared on a basis of calorific value per bomb or per unit weight of bomb.

Firing tests have shown that the can seam developed by Continental Can Co. is suitable for use in so far as withstanding the powder explosion is concerned. This type of seam is limited to 28 gage or thinner metal. If 20 gage metal is required, the bomb casing may be made of seamless, lap-welded or butt-welded tubing.

Further tests are under way wherein the 20 gage and 28 gage cases will be fired from a mortar into a roof structure to determine the impact resistance and ignition and burning characteristics obtained with both the thick and thin cases. Tests will also be conducted with the standard and new fuzes.

According to present information, these bombs will be carried in a bomb crib 15 inches diameter by 61 inches long with a maximum permissible weight of 120 pounds. The proposed 3 inch diameter bomb will thus permit a bundle of 19 bombs which will weigh approximately 100 pounds, although this procedure will leave about three-fourths of the crib empty. Weight limitations aside, four bundles (76 bombs) with 15-1/4 inch maximum bomb length or three bundles (57 bombs) with 20 inch maximum bomb length could be carried.

Present development is concerned with a bomb of approximately 4 lb. gross weight to be approximately equal in effectiveness to the 4-lb. magnesium bomb. Additional data are needed to establish the optimum size of oil bomb and tests are in progress on various types of structures to determine the most effective bomb size, which may be larger than 4 lb.

VIII. Future Work

1. A mortar has been built from which bombs will be fired through typical roof construction in order to determine (1) the necessary casing and nose strength, (2) velocities corresponding to roof penetration alone and roof plus plastered ceiling penetration, (3) time delay necessary to insure fuel discharge below the roof, and (4) the reliability of fuze design.

2. Further gasoline soap work is under way to increase the gelling time during compounding, to decrease the temperature limitations required for mixing the soap and gasoline, and to provide the widest range in substitution of ingredients.

3. Several other attic designs will be studied to determine the optimum fuel charge required for an effective small sized oil bomb.

IX. Conclusions

The Development staff has reached the following highly tentative conclusions:

a) It is believed that a 4-lb. air-borne oil incendiary bomb more effective (per ton of bombs released) than the 4-lb. thermite bomb and possibly as effective as the 4-lb. magnesium bomb for firing light structures and perhaps debris will result provided the 4-lb. oil incendiary possesses the following characteristics:

- 1) A bomb case approximately 3 inches in diameter with a black powder bursting charge which will cause total ejection of the bomb contents as nearly as possible as a single discreet particle for a distance of approximately 20 feet from the point at which the bomb comes to rest.
- 2) Ignition of all particles assured through use of an ignition tube which will bring the bomb contents during ejection into contact with part of the flame produced by the black powder bursting charge.
- b) A bomb designed to incorporate the principles expressed above appears to lend itself to mass production with available ingredients and materials. Such a bomb can be made to meet the safety requirements of the Army Air Corps and Chemical Warfare Service in so far as these safety requirements have been disclosed.

APPENDIX I
GASOLINE THICKENERS

Physical Tests

The viscosity was at first determined by the falling ball method recommended at Harvard. In this test, a 5/32" steel ball is timed for a fall of 10 cm. This method fails on latex blends, since the slight rigidity of the blend suspends the ball indefinitely. The Gardner Mobilometer was, therefore, adopted for control tests. This consists of a tube 20 cm. long and 3.5 cm. I.D., in which moves a loose-fitting perforated piston which may be loaded. For this purpose the piston disk had four 1/4" holes. The load was so adjusted that the time of fall was approximately 100 seconds. The viscosity was calculated as "stokes = 0.250 x load x time", the constant having been determined by tests on known oils. It was found that the curve Stokes vs. time was a straight line on log-log paper, so that any two loads and times served to determine the viscosity at 100 seconds. The viscosity of such blends varies greatly with the flow rate, and the 100 seconds time represents a rate similar to running down a vertical wall.

The mobilometer is not suitable for the soap-thickened products, so the A.S.T.M. grease penetrometer was used. This instrument uses a double-pitch cone weighing 150 grams. The distance is tenth-millimeters penetrated in 5 seconds is reported as the penetration.

The apparent densities of latex and soap blends have been determined. The latex blend, which contained some air, was 0.729/g/cc., and the soap mixture (Formula 122) was 0.750/g/cc.

Vapor pressure tests were made on a few batches, using a modified Reid vapor pressure bomb, and the results were found to agree perfectly with those on the unthickened fuel. Bleeding and shrinkage tests were made by visual observation. Stability was judged by penetrations after storage, shaking one hour in a laboratory shaker, and freezing to -40°F.

The heat transfer characteristics of Batch 42 (Latex blend) were studied by cooling tests in an air bath at 0°F. The tests were of forced and natural convection of the cooling medium over a typical bomb case. The heat transmission was of the unsteady state type. The data was correlated by use of Gurney-Lurie charts which gave integrated forms of the differential cooling once time-temperature-position ratios were established.

The tests showed that cooling was relatively rapid and that in normal bombing operations of two or three hours duration at high altitudes would result in cooling the fuel bombs to the temperature of the surroundings. This means that ignition of the incendiary fuel must occur at temperatures in the range of -40°F., requiring the fuel to have an adequate vapor pressure at this temperature to give unfailing ignition. This stipulation demands use of a fuel similar to a 400° E.P. motor gasoline of at least 10 lb. Reid vapor pressure.

Physical constants determined are:

$$C_p = 0.61 \text{ Btu/lb} \times ^\circ\text{F.}$$

$$= 45.5 \text{ Btu/ft}^3$$

$$k = 0.17 \text{ Btu/hr.} \times \text{ft}^2 \times ^\circ\text{F.} \times \text{ft}^{-1}$$

$h_t = 1.80 \text{ Btu/hr.} \times \text{ft}^2 \times ^\circ\text{F.}$ (natural convection,
as anticipated in operation with bomb bay closed).

Appendix I

1. Rubber Batches

A. Smoked Sheet Rubber

Batch No.	% Rubber	Fuel	Viscosity	Remarks		
				% Acid	% Alkali	Remarks
12	7	Motor Benzene	315 Sec. Ball	-	-	Bell mill 36 hours
13	7	P.C. Naphtha	560 Stokes Gardner	-	-	Bell mill 36 hours
65	7	10% Benzene 90% P.C. Naphtha	340 Sec. Ball 933 Stokes Gardner	-	-	Soaked 3 days
<u>B. Latex</u>						
Batch No.	% Rubber	Fuel	Viscosity			Remarks
4	6.5	Motor Benzene	0	0	1200 Sec. Ball	Stirred 1 hour.
5	5.25	Motor Benzene	2% Oleic	0.3% NaOH	-	Stirred 2 hours.
7	3.8	Summer Gasoline	1.8% Oleic	0.28% NaOH	-	Stirred 10 minutes.
8	6.5	Summer Gasoline	1.0% Oleic	0	-	-
10	6.5	Motor Benzene	0	0	-	-
11	2.75	Summer Gasoline	1.0% Oleic	0	-	Stirred 15 minutes.
14	4.9	P.C. Naphtha	1.8% Oleic	0.23% NaOH	-	-
15	4.9	P.C. Naphtha	6.0% Oleic	0.23% NaOH	-	Put NaOH in Latex.
						Very poor.
16	6.5	P.C. Naphtha	1.8% Oleic	0.23% NaOH	-	Solid at 0° F.
17	5.5	P.C. Naphtha	1.8% Tallow	0.23% NaOH	-	Lake #16.
18	7	P.C. Naphtha	2% Oleic	NH ₄ OH	-	Very corrosive.
20	5.25	54 Naphtha	1.8% Oleic	2.7% Ca(OH) ₂	-	Heated to 150° F.
21	5.25	54 Naphtha	1.8% Stearic	1.6% Ca(OH) ₂	-	Heated to 110° F.
24	6.0	Motor Benzene	1.8% Oleic	NH ₄ OH	-	New Latex (used hereafter)
25	5.5	#5 & #6 Naphtha	2% Oleic	0.10% NaOH	-	10% Sandus
26	5.1	#5 & #6 Naphtha	0.5% Acetic	0.5% NaOH	-	For viscosity curve
26A	6.0	Motor Benzene	1.0% Acetic	0.8% NaOH	-	

P.3

Batch No.	% Rubber	Fuel	% Acid	% Alkali	Viscosity	Remarks
25B	4.0	Motor Benzene	0.8% Acetie	0.6% NaOH	3300 Stokes Gardner	For viscosity curve.
26C	3.6	Motor Benzene	0.6% Acetie	0.6% NaOH	1500 Stokes Gardner	For viscosity curve.
25D	2.4	Motor Benzene	0.4% Acetie	0.6% NaOH	500 Stokes Gardner	For viscosity curve.
26E	1.2	Motor Benzene	0.2% Acetie	0.6% NaOH	97 Stokes Gardner	For viscosity curve.
26F	1.8	Motor Benzene	0.3% Acetie	0.6% NaOH	320 Stokes Gardner	For viscosity curve.
27	2.4	P.C. Naphtha	0.6% Acetie	0.6% NaOH	400 Stokes Gardner	Bleeds at -40° F. Mixed at 45° F.
28	2.4	P.C. Naphtha	0.4% Acetie	0.6% NaOH	300 Stokes Gardner	Mixed at 45° F. 25 lb. batch.
29	2.4	P.C. Naphtha	0.6% Acetic	0.6% NaOH	295 Stokes Gardner	Mixed at 55° F. 25 lb. batch.
29A	2.4	P.C. Naphtha	0.6% Acetic	0.6% NaOH	420 Stokes Gardner	Laboratory batch.
29B	2.4	P.C. Naphtha + 10% Benzene	0.6% Acetic	0.6% NaOH	510 Stokes Gardner	Laboratory batch.
30	3.0	Pentene + 10% Benzene	0.75% Acetie	0.75% NaOH	-	-
31	2.4	50% P.C. Naphtha 50% /3 Naphtha	0.6% Acetic	0.6% NaOH	670 Stokes Gardner	Very thin.
32	2.4	Sour P.C. Naphtha	0.6% Acetie	0.6% NaOH	750 Stokes Gardner	-
33	2.4	Sour P.C. Naphtha + 10% Benzene	0.6% Acetie	0.6% NaOH	640 Stokes Gardner (610 one month later)	Mixed at 65° F. 25 lb. batch.
34	2.4	P.C. Naphtha + 10% Benzene	0.6% Acetie	0.6% NaOH	625 Stokes Gardner	Mixed at 71° F. 25 lb. batch.
35	0.4	P.C. Naphtha + 10% Benzene	0.6% Acetie	0.6% NaOH	-	-
37	2.4	Sour P.C. Naphtha 95% Sour P.C. Naphtha + 5%	0.6% Acetie	0.6% NaOH	570 Stokes Gardner	To study effect of aromatics.
38	2.4	Motor Benzene	0.6% Acetie	0.6% NaOH	500 Stokes Gardner	-
39	2.4	90% Sour P.C. Naphtha + 10% Motor Benzene	0.6% Acetie	0.6% NaOH	600 Stokes Gardner	-
40	2.4	90% Sour P.C. Naphtha + 20% Motor Benzene P.C. Naphtha + 10% Benzene	0.6% Acetie	0.6% NaOH	650 Stokes Gardner	-
41	2.4	-	0.6% Acetie	0.6% NaOH	800 Stokes Gardner	25 lb. batch.

<u>Batch No.</u>	<u>% Rubber</u>	<u>Fuel</u>	<u>% Acid</u>	<u>% Alkali</u>	<u>Viscosity</u>	<u>Remarks</u>	
42	2.4	Sour P.C.Naphtha + 10% Benzene	0.6% Acetic	0.6% NaOH	620 Stokes Gardner (275 two weeks later)	Loss of viscosity due to sourness.	
43	2.4	Diesel Fuel Oil	0.6% Acetic	0.6% NaOH	430 Stokes Gardner	26 lb batch.	
44	2.4	Pentane + 10% Lube Oil + 10% Benzene	1.5% Acetic	2.0% NaOH	-	Mixed at 40°F.	
45	2.4	Pentane + 10% Lube Oil + 15% Benzene	1.3% Acetic	1.2% NaOH	-	Mixed at 50°F.	
46	2.4	#3 Naphtha + 6% P.C. Naphtha	0.6% Acetic	0.6% NaOH	400 Stokes Gardner	25 lb. batch.	
47	2.4	Sour P.C.Naphtha + 10% Benzene	0.6% Acetic	0.6% NaOH	570 Stokes Gardner	25 lb. batch.	
48	2.4	Winter Gasoline	0.6% Acetic	0.6% NaOH	580 Stokes Gardner	25 lb. batch.	
49	2.4	Kerosene + 10% #3 Naphtha	0.6% Acetic	0.6% NaOH	576 Stokes Gardner	25 lb. batch.	
50	2.4	Medium Fuel Oil	0.6% Acetic	0.6% NaOH	450 Stokes Gardner	25 lb. batch.	
51	2.4	+ 10% #5 Naphtha	Sour P.C.Naphtha + 10% Benzene	0.6% Acetic	0.6% NaOH	230 Stokes Gardner	75 lb. batch.
52	2.4	Sour P.C.Naphtha + 10% Benzene	0.6% Acetic	0.6% NaOH	350 Stokes Gardner (250 five days later)	50 lb. batch.	
67	2.1	Sour P.C.Naphtha + 10% Benzene	0.6% Acetic	0.6% NaOH	410 Stokes Gardner	50 lb. batch.	
74	2.4	Winter Gasoline	0.6% Acetic	0.6% NaOH	-	25 lb. batch.	
217	2.1	Sour P.C.Naphtha + 10% Benzene	0.6% Acetic	0.6% NaOH	-	25 lb. batch.	
<u>C. Crepe Rubber</u>							
<u>Batch No.</u>	<u>% Rubber</u>	<u>Fuel</u>	<u>Time</u>	<u>Viscosity</u>	<u>Remarks</u>		
57	5.5	Sour P.C.Naphtha	77	24 hours	-	Grainy	
58	5.5	Pure Benzene	77	7 hours	-	Smooth	
59	5.5	Sour P.C.Naphtha + 10% Benzene	77	24 hours	-	Grainy	
62	5.5	Pentane	55	2 weeks	400 Stokes Gardner	Not dispersed	
64	5.5	Sour P.C.Naphtha	77	24 hours	50 lb. batch	50 lb. batch	

<u>Batch No.</u>	<u>% Rubber</u>	<u>Fuel</u>	<u>Temp. °F.</u>	<u>Time</u>	<u>Viscosity</u>	<u>Remarks</u>
65	5.5	Sour P.C.Naphtha Pure Benzene	-10° F.	2 weeks	-	Not dispersed
66	5.5	Sour P.C.Naphtha	77	8 hours	450 Stokes Gardner	25 lb. batch
68	10	Pure Benzene	77	3 days	3000 Stokes Gardner	Very smooth
69	5	Sour P.C.Naphtha	77	3 days	-	Not homogeneous
70	2	Sour P.C.Naphtha	77	3 days	-	Not homogeneous
71	1	Sour P.C.Naphtha	77	3 days	-	Naphtha added after 1 hour - did not disperse
72	5.5	22% Benzene + 75% Sour P.C.Naphtha	77	3 days	-	75 lb. batch
75	5.5	Winter Gasoline	77	24 hours	"	Cut back from 10%
76	2	Winter Gasoline + 20% Sour P.C.Naphtha	77	-	Too thin	Cut back from 10%
77	4	Winter Gasoline + 40% Sour P.C.Naphtha	77	-	98 Stokes Gardner	Cut back from 5.5%.
79	2.9	Winter Gasoline	77	-	-	9.1% cotton waste.
80	3.3	Winter Gasoline	77	-	-	Cut back from 5.5%.
81	1.5	Winter Gasoline	77	-	-	9.1% cotton waste.
82	10	Winter Gasoline	77	20 hours	-	Cut back from 5.5%.
126	5.5	Winter Gasoline	70	-	-	Slightly lumpy. Cut back from 10%.

D. Reclaimed Rubber

<u>Batch No.</u>	<u>% Rubber</u>	<u>Fuel</u>	<u>Type</u>	<u>Fuel</u>	<u>Remarks</u>
3	8	Benzene	Black Reclaimed	Benzene	Did not disperse - swelled only.
3A	16	Benzene	Black Reclaimed	Benzene	Did not disperse - swelled only.
5	12	Benzene	Tube Reclaimed	Benzene	Did not disperse - swelled only.
36	6	Benzene	Light Reclaimed Emulsion	-	Dispersed to give a non-stringy gel.

2. Soap Batches

A. Aluminum Soaps

<u>Batch No.</u>	<u>% Soap</u>	<u>Type</u>	<u>Fuel</u>	<u>Manufacture</u>	<u>Viscosity</u>	<u>Remarks</u>
19	11.4	Naphthenate	54 Naphtha P.C. Naphtha + 10% Kerosene	Bell Mill 40 hrs. Heated to 225° F., Added Winter Gasoline as it cooled.	40 Sec. Bell	Paint Drier Soap Contains 14 Glycerol
45	10	Stearate		-	-	
120	8	Naphthenate	Kerosene	Heated to 200° F.	Too thin	Paint Drier Soap
121	10	Naphthenate	Kerosene + Winter Gasoline	Heated to 250° F.	1630 Strokes Gardner	Grease Making Soap
123	6.2	Oleate	+ 15% Kerosene	Heated to 300° F., Added Winter Gasoline as it cooled.	-	From Cottontseed oil
134	8	Naphthenate	Winter Gasoline + 3% Lube Oil	Heated to 175° F., Added Winter Gasoline as it cooled.	-	
135	9	Oleate	Winter Gasoline + 3% Lube Oil	Heated to 190° F., Added Winter Gasoline as it cooled.	-	From Cottontseed oil

B. Calcium Soaps

<u>Batch No.</u>	<u>% Soap</u>	<u>Type</u>	<u>Fuel</u>	<u>Penetration</u>	<u>Remarks</u>
1	19	Rosin Oil	Motor Benzene	350	Spatters badly.
109	6.2	Hog fat	Winter Gasoline + 30% Lube Oil	Fluid	From #3 Cup Grease
110	9.5	Rosin Oil	Kerosene	-	Did not gel.
111	11	Stearic Acid	Kerosene	-	Did not gel.
119	6	Hog fat	Winter Gasoline + 25% Lube Oil	Fluid	From #3 Cup Grease
127	6	Hog fat	Winter Gasoline + 25% Lube Oil	Fluid	From #3 Cup Grease

Sample No.	Zeta	Fuel	Penetration		Remarks
			From #3 Cup Grease	To #3 Cup Grease	
128	4	Motor Gasoline + 15% Lube Oil	270	222	Used cup grease to suspend line.
137	12	Oleic acid	270	-	#3A not set without any grease present
139	12	Maleic Acid	-	-	Used Batch 137 to re- place cup grease
145	12	Maleic acid	219	-	
C. Molted Codium Strips					
Sample No.	Zeta	Fuel	Penetration	Remarks	
128	5	Isopropanol	5 Impractical	Yerocene + 30%	Open cooked
135	5.5	Isopropanol	4	D.G. Ethylate + P.C. Lubrizol + 5% Petroleum	Open cooked
140	5.9	Isopropanol	5	P.G. Ethylate + 5% Xerocene	Open cooked
142	6.2	Castor oil	10 ³	P.C. Lubrizol + 10% Kerosene	Open cooked
73	7.0	Isopropanol	100	Winter Gasoline	146
88	7.4	Kerosene	100	Winter Gasoline	138
94	7.4	Castor oil	1.0	Winter Gasoline	130
95	7.6	Castor oil	1.0	Winter Gasoline	150
95	7.6	Castor oil	1.0	Kerosene	169
97	7.6	Castor oil	1.0	Kerosene	160
97	7.6	Castor oil	1.0	Kerosene	160
98	7.6	Castor oil	1.0	Castor oil	175
99	7.6	Castor oil	1.0	Kerosene	190
100	7.6	Castor oil	1.0	Water	190
101	7.6	Castor oil	1.0	Kerosene	218
102	7.6	Cotton	1.0	Kerosene	160
102	7.6	Hydrocol	1.0	Kerosene	160
102	7.6	Cotton	1.0	Kerosene	175
103	7.6	Hydrocol	1.0	Kerosene	190
104	7.6	Cotton	1.0	Kerosene	200
105	7.6	Oleic oil	1.0	Kerosene	132
106	7.6	Oleic oil	1.0	Castor oil	400+

<u>Batch No.</u>	<u>% Acid</u>	<u>% Resin</u>	<u>% Plasticizer</u>	<u>Fuel</u>	<u>Temp. •F.</u>	<u>Penetration</u>	<u>Remarks</u>
95	2.4	Hydrofol	1.2	2.4 Fish Oil	Kerosene	210	187
97	2.4	Hydrofol	1.2	2.4 Tasseed Oil	Kerosene	195	100
98	2.4	Hydrofol	1.2	5 Glycerol	Kerosene	160	165
99	2.4	Hydrofol	1.2	4.8 Castor Oil	Kerosene	230	157
100	2.4	Hydrofol	1.2	5.4 Methanol	Kerosene	150	141
101	2.4	Hydrofol	1.2	1.8%Castor Oil	Kerosene	193	155
113	2.4	Hydrofol	1.2	2.4 Castor Oil	Kerosene	198	-
114	2.4	Hydrofol	1.2	3.2 Isopropanol	Winter Gasoline	137	-
115	2.4	Hydrofol	1.2	3.2 Isopropanol	Winter Gasoline	137	-
116	2.4	Hydrofol	1.2	3.2 Isopropanol	Winter Gasoline	132	180
124	2.4	Hydrofol	1.2	3.2 Isopropanol	Winter Gasoline	132	162
125	2.4	Hydrofol	1.2	3.2 Isopropanol	Winter Gasoline	132	Cooked at 15 p.s.i.

D. Cold Made Sodium Soaps

Benzo Formula

Batch No.	% Acid	% Rosin	% Plasticizer	Temp. °F.	Penetration	Bleeding	Stability	Remarks
122	3.5 Hydrofol	1.8 W.G. (Wood)	3.0 Castor Oil	100	240	Trace	Excellent	Added fillers.
136	3.5 Hydrofol	1.8 W.G. (Wood)	3.0 Castor Oil	100	-	Trace	Excellent	Added fillers.
141	3.5 Hydrofol	1.8 W.G. (Wood)	3.0 Castor Oil	100	-	Trace	Excellent	Added fillers.
142	3.5 Hydrofol	1.8 W.G. (Wood)	3.0 Castor Oil	100	-	Trace	Excellent	2% Chopped Sisal added before alkali
149	3.5 Hydrofol	1.8 W.G. (Wood)	3.0 Castor Oil	100	-	Trace	Excellent	-
152	3.5 Hydrofol	1.8 W.G. (Wood)	3.0 Castor Oil	100	-	Trace	Excellent	1% Long Sisal
167	3.5 Hydrofol	1.8 W.G.	3.0 Castor Oil	100	-	Trace	Excellent	No filler.

I. Effect of Changing Acid Type

Batch No.	Acid	Penetration	Bleeding	Stability	Shrinkage	Reporting	Remarks
150	Stearic Cotton	207	Medium	-	-	-	Orally
160	-	-	-	-	-	-	Did not set

II. Effect of Changing Rosin Grade

Batch No.	Rosin	Penetration	Bleeding	Stability	Shrinkage	Reporting	Remarks
146	"B" Grade W.G. (Gum)	199	Bad	-	-	-	Separates
153	-	Trace	Excellent	Slight	-	Good	Equal to basic formula

III. Effect of Acid-Rosin Ratio

Batch No.	% Hydrofol	% Rosin	Penetration	Bleeding	Stability	Shrinkage	Reworking	Remarks
156	4.5	1.0	254	Trace	Good	Normal	Good	Sets a little slower
161	4.9	0.5	205	Trace	Good	Normal	Good	Separates
163	0.5	4.8	-	Bad	-	-	-	Sets a little slower
164	1.8	3.5	248	Medium	Poor	-	Bad	

IV. Effect of Fatty Oil Changes

Batch No.	% Fatty Oil	Penetration	Bleeding	Stability	Shrinkage	Reworking	Remarks
103	None	-	25%	Poor	-	-	Lower soap content
104	1 Castor Oil	-	Medium	Fair	-	-	Lower soap content
105	2.4 Castor Oil	-	Slight	-	-	-	Lower soap content
149	3 Fish Oil	195	Medium	Good	Medium	Difficult	Sets slower
154	3 Cottonseed Oil*	254	Trace	Good	Good	Good	Smooth
157	1 Fish Oil	258	Trace	Good	Medium	Good	Smooth
162	6 Castor Oil	251	None	Good	Little	Excellent	Smooth
170	3 Sperm Oil	223	Medium	-	-	-	Smooth
171	3 Refined Cottonseed Oil	196	Medium	-	-	-	Smooth
172	3 Refined Rapeseed Oil	225	Slight	-	-	-	Smooth
178	1 Castor Oil	350	Bad	-	-	-	Smooth
183	3 Castor Oil	Soft	-	-	-	-	Smooth
	1.5 Fish Oil	-	-	-	-	-	Smooth

* Heat treated

V. Effect of Alkali Content

<u>Batch No.</u>	<u>% NaOH</u>	<u>Water</u>	<u>Penetration</u>	<u>Bleeding</u>	<u>Stability</u>	<u>Shrinkage</u>	<u>Reworking</u>	<u>Remarks</u>
132	1.25	2.5	240	Trace	Excellent	Slight	Excellent	Basic Formula
165	1.07	2.14	280	Trace	Excellent	Slight	Excellent	Sets normally
166	0.89	1.76	255	Medium	Good	Medium	Poor	Rubbery
168	0.72	1.44	226	Fair	Fair	Medium	Poor	Sets slower
180	1.79	3.58	"	None	Good	Slight	Good	Sets normally

VI. Effect of Borax

<u>Batch No.</u>	<u>% Borax</u>	<u>% NaOH</u>	<u>Penetration</u>	<u>Bleeding</u>	<u>Stability</u>	<u>Shrinkage</u>	<u>Reworking</u>	<u>Remarks</u>
184	6.0	0	"	=	=	-	-	Sets very slowly
185	3.0	0.65	240	Medium	-	Slight	Medium	Sets in 1 hour
186	1.2	0.28	270	Bal	-	Slight	Good	Sets normally

VII. Effect of Mixing Temperature

<u>Batch No.</u>	<u>Temperature °F.</u>	<u>Penetration</u>	<u>Bleeding</u>	<u>Stability</u>	<u>Shrinkage</u>	<u>Reworking</u>	<u>Remarks</u>
129	40	"	Bad	-	-	-	Separates
130	85	220	Medium	Fair	-	Good	-
131	105	197	None	Good	-	Good	Best

APPENDIX II

SUGGESTED METHOD FOR COMPOUNDING AND
FILLING FORMULA 122 IN THE FIELD

In places where rubber is available, it may be used as a satisfactory thickening agent. The crepe rubber is cut or torn into strips about one inch wide, and 5.5-6% is stuffed into the empty bomb cases (10 oz. in an M-46 case, or 2 lb. 3 oz. in an M-47 case). The cases are then filled to 90% of their volume with motor gasoline, capped and allowed to stand in a warm place (not less than 70°F., preferably 100°F.) for two days.

Where it is desired to use the gasoline soap, two ready-made supplies must be available, namely, Base "A" described below, and caustic soda. Base "A" can be prepared in domestic plants using the following formula:

Hydrogenated fish oil acid - 23.5%
Wood or gum rosin, W.G. Grade - 12.1%
Castor oil, No. 3 Grade - 20.1%
Heavy naphtha, 100°F. Flash - 44.3%

In the field 45 gallons of gasoline are measured into an open-head drum and warmed to 95°F. by means of a steam coil or electric heaters. To this is added 50 lb. (7-1/3 gallons) of Base "A", which should be stirred before measuring. To this is then added 4 lb. of caustic soda (lye) dissolved in one gallon of water. The mixture is well stirred with a paddle during the addition of the lye, and for a few minutes thereafter. The gasoline will thicken rapidly. It is then ready to fill into bomb cases, using large grease guns or ordinary dippers. Occasional stirring will keep the grease soft enough to pour for about one hour. (Note--Usual precautions required for handling ethyl gasolines should be observed).

Field supplies required are:

52 gallon open-head drums of Base "A".
4 lb. cans of lye (caustic soda).
Open-head 52 gallon mixing drum.
Heater coil and steam supply or electric heater
and current.
5 gallon measure, graduated in 1/8 gallons.
Grease gun or large funnel and dipper.

If desired, the mixing can be done directly in the bomb cases, by shaking well after addition of Base "A" and again after addition of lye solution. This is recommended only for the M-46 or M-47 case, which is of a convenient size.

Appendix III
Tee Buns - Arctic Structures
Latex Gum Without Binder

Run No.	wt. of Charge (lbs.)	Distance from Leaves (ft.)	Position from Burn (ft.)	Poss. (ft.)	Burning Time (minutes)	Active Total (sq. ft.)	Floor joists Harbor Lath(s) (ft.)	Left Cent. Right	Remarks
H-10	0.25	3.0	No	Batch 29	-	0	0	0	Charge flung against roof; 75% fall to floor.
H-16	0.25	2.0	No	"	3.0	7	1	1	Very marked low point at 6 min.
H-17	0.25	1.0	Yes	"	2.0	10+	5	4	Very marked low point at 6 min.
H-20	0.75	2.5	Yes	Batch 30	1.2	10+	4	4	Very marked low point at 1 min; low point at 6 min.
H-25	0.60	5.0	No	Batch 32	"	"	5	5	Same as H-20.
H-35	0.55	3.0	No	Batch 34	"	"	0	0	Charge flung against roof; 75% fall to floor.
H-36	0.55	2.6	Broke	"	3.0	7+	2	2	Charge flung against roof; 75% fall to floor.
H-42	0.50	2.0	Yes	"	3.0	6+	2	5	Charge flung against roof; 60% fall to floor.
H-48	0.35	5.0	No	Batch 34	4.5	10	4	0	Flame volume seemed unusually small.
H-49	0.55	4.0	No	Batch 39	3.5	7+	2	5	Small smoky flame.
H-50	1.00	4.0	No	Batch 34	1.5	5	2	2	Volatile draft conditions during test.
H-52	1.00	4.0	No	"	4.0	8	2	0	Flame licked roof sporadically.
H-53	0.85	4.0	No	(B)	5.0	10	4	0	Variable draft conditions during test.
H-53	1.00	4.0	No	Batch 41	4.0	6	2	4	Flame volume seemed unusually small.
H-42	0.95	4.0	No	Batch 55	3.0	5	3	4	Heat intensity lower than with naphtha gums.
H-30	0.80	3.5	No	Batch 44	4.0	7	4	4	All runs made on Design #1 structure.
H-39	1.05	3.5	No	Batch 41	4.0	6	2	4	Fuel composition data shown on attached sheet.
H-42	0.95	3.0	No	Batch 54	4.0	10	4	4	(A) Numbers in parentheses refer to number of laths burned through.
H-52	0.95	3.0	Broke	(B)	5.0	10	6	6	(B) 45% latex in 50% diesel fuel oil, 25% heavy naphtha, 25% light naphtha.
H-35	1.00	3.0	Yes	Batch 47	"	4.5	"	"	(E)

Footnotes:

(A) Fuel composition data shown on attached sheet.
 (B) Numbers in parentheses refer to number of laths burned through.
 (C) 45% latex in 50% diesel fuel oil, 25% heavy naphtha, 25% light naphtha.

Eso Laboratories
Process Division
WTK, Jr., 31 1/2/42

TEST BURNS - AVTC STRUCTURE
LATEX GUM WITH & BUTYL

Run No.	Wt. of Charge from Latex Burn (lbs.)	Distance Positive Burn (ft.)	Fuel (A)	Burning Time Minutes Active Total	Floor Joists Rafters (sq. ft.)	Lathe(A) Left Cent. Right	Amount Charred or Burned			Remarks
							Floor	Joists	Rafters	
H-38	1.00	3.0	No	Batch 41	4.0	8	2	4	3	0 0 0
H-39	1.00	3.0	Yes	"	4.0	10+	Complete destruction of all three sections (roof and floor) in 10 minutes.			Variable draft conditions during test.
H-40	1.00	3.0	No	"	4.0	5	2	5	0 0 0	Variable draft conditions during test. Flame's directed into eaves from start of burn.
H-41	1.00	3.0	Yes	Batch 55	4.0	10+	Complete destruction of all three sections (roof and floor) in 10 minutes.			50% fuel unburned after 5 mins.
H-73	1.25	2.5	Yes	Batch 67	4.0	30+	Fuel very fluid - spread over 2 sq.ft. area.			Pellet very fluid - spread over 2 sq.ft. area.
H-73	1.25	2.5	Yes	"	4.0	30+	Complete destruction of all three sections (roof and floor) in 30 minutes.			
H-5	2.00	4.5	Brdle	Batch 20	5.0	10+	4	6	14	- - -
H-32	2.50	3.5	Yes	Batch 47	5.0	10+	6	8	20	8 14(12) -
H-21	5.00	4.5	Brdle	Batch 54	5.5	20+	8	8	16	3 6 5
H-22	5.00	4.0	Yes	"	4.5	9+	Complete destruction of all three sections (roof and floor) in 9 minutes.			Structure collapsed.
H-24	3.00	4.0	Yes	batch 45	5.0	10+	Complete destruction of all three sections (roof and floor) in 10 minutes.			
H-25	3.00	4.0	No	Batch 54	3.0	10	12	12	8	3 4 5
H-6	5.00	7.0	No	Batch 29	5.0	14+	3	8	0	0 0 0
H-7	5.00	6.0	No	"	5.0	8+	4	6	2	0 4 0
H-9	5.00	5.0	Brdle	"	5.0	10+	5	8	15	4 5(5) 4
H-26	9.65	6.0	Yes	Batch 34	"	8+	Large amount of floor afire seemed to be deciding factor in propagating burn.			

Footnotes: All runs made on Design #1 structure.

(B) Fuel composition data shown on attached sheet.

(A) Numbers in parentheses refer to number of lathe burned through.

See Laboratories
Process Division
W.R./JL 1/2/42

TEST BURNS - ATTIC STRUCTURE

Run No.	Wt. of Charge (lbs.)	Distance from Eaves (ft.)	Positive Burn (ft.)	Fuel	Burning Time				Amount Charged or Burned				Remarks	
					Minutes	Active Total	Floor Joists (ft.)	Rafter Rafters (ft.)	Laths (ft.)	Lat. Cont.	Height	Rafter Type		
(sq. ft.)					(sq. ft.)	(ft.)	(ft.)	(ft.)	(ft.)	(ft.)				
H-55	0.95	4.0	No	Battie	67	3.5	7	1.5	1	0	0	2x6 fir	Flame spreading hindered by floor construction.	
H-54	1.00	3.0	No	"	"	3.0	7	2.5	5	2	0	5	0	Initial large flame; but rapid decrease.
H-50	1.05	3.0	No	"	"	3.5	13	5.0	7	10	0	11	0	2x6 w.pine
H-55	1.00	2.5	No	"	"	3.5	7	5.0	6	8	0	11	0	2x6 fir
H-57	1.00	2.5	No	"	"	3.5	8	5.0	2	2	0	4	0	2x4 w.pine
H-56	1.00	2.0	No	"	"	3.5	15	4.0	6	8	0	15(2)	0	2x6 fir
H-55	1.00	1.5	No	"	"	3.0	12	5.0	4	8	0	11(2)	0	Center section ablaze for 3 mins.
H-50	1.00	1.0	No	"	"	3.0	15	5.0	6	12	0	14(1)	0	Blto above.
H-58	1.00	1.0	Yes	"	"	3.5	17	5.0	6	15	10(5)	2x4 w.pine	Roof completely afire at 7 mins.	
H-57	2.95	5.0	Brdle	"	"	6.0	11	5.0	6	8	0	10(2)	0	2x6 fir
H-59	3.00	4.0	Yes	"	"	6.5	10	8.0	10	20	0	14(1)	0	Char deeper than with 1 pound charges.
H-69	-	5.06	-	5.5	-	Yes	"	4.0	-	7.0	6	18	-	2x4 w.pine
H-61	-	1.00	5.5	-	-	-	-	4.0	-	1.0	1	0	-	Structure collapsed after 10 mins.
H-83	1.00	2.0	No	"	"	3.0	7	4.0	4	8	0	15	0	Floor laths only slightly charred.
H-82	1.00	1.0	No	"	"	2.5	-	4.0	5	10	0	-	-	2x6 fir
H-85	1.00	1.0	No	"	"	2.5	-	4.0	5	10	-	-	-	2x4 w.pine
H-84	2.50	3.0	Brdle	"	"	4.0	10	6.0	6	18	12	15(6)	12	2x6 fir
H-85	2.50	2.0	Yes	"	"	4.0	10+	10.0	10	50	17(1)	17(15)	17(2)	Tremendous flames. Ridgepole charred for 6 ft.
H-85	2.95	3.0	No	"	"	4.0	-	-	-	-	-	-	-	4 floor laths burned thru; plaster still adhered to floor structure.
H-64	5.00	3.0	No	"	"	2.0	5	-	-	-	-	-	-	Charge spread over 1 sq.ft. area. Ditto above.
H-66	5.00	3.0	No	"	"	4.5	14	-	-	-	-	-	-	Floor laths burned thru over 6 sq.ft. area. Plaster adhered but could be easily loosened.

Last 9 runs(H-61 thru H-66) made on design #2.
 (B) Fuel composition data shown on attached sheet.
 (A) Numbers in parentheses refer to number of lots burned through.

Appendix III
TEST BURNS - ATTIC STRUCTURES

APPENDIX III - P. 4

Run No.	Wt. of Charge (lbs.)	Distance from Eaves (ft.)	Positive Burn	Fuel (%)	Burning Time Minutes	Amount Charred or Burned			Remarks		
						Poor Joists (sq. ft.)	Rafter (ft.)	Lathe (A)			
H-47	0.95	3.0	No	Batch 41	4.0	6	2	5	0	0	10% charge on roof; 90% on floor.
H-48	1.00	5.0	No	Batch 55	3.5	5	5	4	12	0	20% charge on roof; 80% on floor. Tremendous flames. Roof ablaze at 1 min., out at 2.5 mins.
H-51	1.00	2.8	No	"	1.0	3	12	18	15	0	Roof ablaze at 1 min., out at 2.5 mins.
H-52	1.05	2.5	No	"	1.0	5	12	18	15	0	Roof ablaze at 1 min., out at 2.5 mins.
H-174	1.00	5.0	Brdle	Batch 117	2.5	5	6	8	12	0	Mass of flames penetrated to eaves in 1 min., but died down sharply after fuel was exhausted.
H-177	1.00	5.0	No	"	2.5	5	2	4	8	5	Old wood (not planed) from dismantled house. Flames did not reach eaves.
H-178	1.00	2.8	Brdle	"	2.5	10	6	8	12	5	New wood. Flames did not reach eaves.
H-179	1.00	2.8	Brdle	"	2.5	8	4	6	12	6	Old wood. Eaves afire at 1.5 mins.
H-180	1.00	2.5	Yes	"	15*	5	6	10	6	12(4)	New wood. Flames nearly reached eaves.
H-181	1.00	2.5	Yes	"	15*	5	6	14	8(5)	7	Old wood. Eaves afire at 1.5 mins. Low point at 4 mins. Deep char.
H-186	1.50	2.5	Yes	"	3.5	10+	6	8	20	9	Used structure with 20 in. of board to form a peaked roof. No evidence that such construction helped flames.
H-199	2.30	4.5	No	"	3.5	11	3	6	6	0	All runs made on Design #1 structure. Runs H-47, 48, 51 & 52 made on double structure formed by placing two open-end structures together.
											(*) Fuel composition data shown on attached sheet.
											(A) Numbers in parentheses refer to number of lathes burned through.

Footnotes:

All runs made on Design #1 structure. Runs H-47, 48, 51 & 52 made on double structure formed by placing two open-end structures together.

(*) Fuel composition data shown on attached sheet.

(A) Numbers in parentheses refer to number of lathes burned through.

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WPA, Jr./JL 1/2/42

TEST BURNS = ATTIC STRUCTURE

Smoked Sheet Rubber® Gums without Binder

Run No.	Wt. of Charge (lbs.)	Distance from Eaves (ft.)	Positive Burn	Fuel (X)	Batch 15	4.0	4.5	Burning Time Minutes	Amount Charred or Burned			Remarks
									Floor (sq. ft.)	Joints (sq. ft.)	Rafter (sq. ft.)	
									Left	Cent.	Right	
H-2	1.00	4.0	No									
H-5	1.00	5.0	Yes	"	4.0	10*	5	8	12	9	12	0
H-1	0.95	2.0	Yes	Batch 12	"	10*	4	8	14	9	9	9
H-11	0.25	8.0	No	Batch 15	"	"	0	0	4	0	6	0
H-12	0.25	8.0	No	"	"	"	0	0	4	0	8	0
H-46	0.55	2.0	No	Batch 12	3.0	6	1	1	5	0	7	2
H-25	1.00	4.0	No	Batch 15	3.0	6	0	0	4	0	5	0
H-29	1.00	4.0	No	Batch 12	3.0	8	0	0	"	0	-	0
H-45	1.00	3.5	No	"	1.0	5	2	0	0	0	0	0
H-44	1.00	3.5	No	"	1.5	3	0	0	15	4	10	9
H-45	1.00	3.0	No	"	2.5	8	5	4	0	6	2	35% charge smeared on roof; 65% placed on floor.

Footnotes: All runs made on Design #1 structure. Runs H-43, 44, 45 & 46 made on double structure formed by placing two open-end structures together.
 (X) Fuel composition data shown on attached sheet.

Esso Laboratories
 Process Division
 WTK, Jr./JL 1/2/42

TEST BURNS - ATTIC STRUCTURE

Run No.	Wt. of Charge (lbs.)	Distance from Eaves (ft.)	Positive Burn	Fuel (#)	Burning Time Minutes	Amount Charred or Burned			Remarks		
						Floor joists Rafters (sq. ft.)	Lathe(A) (ft.)	Left Cent. Height			
H-49	1.00	6.0	No	Batch 58	3.0	5	0	0	Charge smeared around ridgepole; 1/3 charge dropped to floor.		
H-151	2.30	-	Yes	Batch 75	4.0	10*	20	22	Charge blown into eaves from 3" bomb casing(20 grams black powder). Moderate low point at 5 min.		
H-184	1.50	4.5	No	"	3.0	6	1.5	4	Tremendous flames but little damage.		
H-191	1.50	4.5	No	"	4.0	10*	1	2	0	0	0
H-185	1.50	2.5	Brdls	"	3.0	8	6	11	Eaves afire at 2 min.; tremendous flames while fuel lasted.		
H-192	1.50	2.0	Yes	"	-	10*	Complete destruction of attic(roof and floor) in approximately 10 minutes.				
HH-197	2.30	7.0	No	"	5.0	10	4	8	Charge blown onto ridgepole from 3" bomb casing (20 grams black powder).		
HH-92	5.95	3.5	Yes	"	2.0	10*	35	55	---	104(10) ---	

Footnotes: First 7 runs - H49 to HH-197 made on Design #1 structure; HH-92 on Design #2.
See also Runs HH-89, 90 and 91 for crepe rubber used with cotton waste.

(#) Fuel composition on data shown on attached sheet.

(A) Numbers in parentheses refer to number of lathe burned through.

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ATK, Jr./JL 1/2/42

APPENDIX III
TEST BURNS - ATTIC STRUCTURES

Appendix III - P. 7

Run No.	Wt. of Charge (lbs.)	Distance from Eaves (ft.)	Positive Burn	Fuel (k)	Duration min Active Total	Floor joists (ft.)	Roof joists (ft.)	Left. Center. Right Laths(1)	Burned through floor (1-5/4") ; joists (1/2")		
									80, ft.	100, ft.	Structure
H-71 1.25	4.0	No Hex.	7.0	30	1.5 4	C	0	0	0	0	Design #1 Charred through floor (1-5/4"); joists (1/2")
H-73 1.25	3.0	No "	10.0	20+	1.5 4	O	0	0	0	0	" ditto above.
H-132 1.25	3.0	No	8.0	20+	2.0 6	O	0	0	0	0	" ditto above.
H-76 1.25	2.5	Yes	"	-	30+	Complete destruction of entire attic(roof and floor) after 30 minutes. Burn aided by torch and oil flames nearby. (Design #1)	"	"	"	"	
H-75 1.25	2.3	Yes	"	-	30+	3 6	12	0	14(11)	5	Design #1
H-72 1.25	2.0	Yes	"	-	20+	Complete destruction of entire attic(roof and floor) after 30 minutes. Burn considerably aided by torch. (Design #1)	"	"	"	"	
H-74 1.25	2.0	Yes	"	4.0	20	2 4	8	0	12(2)	5	Design #1 High flames at start - rapid decrease.
H-80 2.50	4.0	No	"	4.0	15+	3 8	12	0	10	9	"
H-81 2.50	3.5	No	"	6.0	20+	4 4	12	2	11(3)	2	"
H-155 1.25	2.0	Yes	"	4.0	10+	4 4	16	0	14(3)	7	Design #2 Floor laths charred thru. Ridgepole slightly burned.
H-197 2.50	3.0	No	"	5.0	12+	6 6	14	2	17	2	"
H-198 2.50	2.0	Yes	"	6.0	10+	6 6	20	0	15(6)	17(17)	"
H-197 4.00	2.5	No	Bomb	8.0	15+	2 4	4	0	9	4	Design #3 Plaster fell through on touching.
H-199 4.00	2.5	Edle	"	8.0	15+	4 8	10	5	7(5)	4	"
HH-195 4.00	2.5	Yes	"	6.0	10+	4 8	14	7	14(5)	6	"
H-200 4.00	2.5	No	"	6.0	11	1.5 3	0	0	0	0	"

Footnotes: First 9 runs(H-31) on Design #1; last run (H-200) on Design #2 structure; next 3 runs(H-86 thru H-89) on Design #1 except plastered floor.

(R) Horizontal Mg bar stuck 1-1/2" dia. Fired by orgo-acetylene torch or regular Army Mg bomb.

(A) Numbers in parentheses refer to number of laths burned through.

Test Laboratories Process Division W.H.Jr./31/2/42

TEST BURNS - ATTIC STRUCTURE

Run No.	Wt. of Charge (lbs.)	Distance from Eaves (ft.)	Positive Burn	Fuel (%)	Burning Time Minutes	Amount Charred or Burned			Remarks		
						Floor (sq. ft.)	Joists (ft.)	Rafters (ft.)			
H-77	2.00	2.5	No	Grenade	0.3	5	1	2	0	0	0
H-148	4.00	2.5	No	Bomb	1.5	5+	4	2	0	0	0
H-176	4.00	2.5	No	n	1.5	-	1	2	0	0	Sputtered considerably. Charred through 1-1/2" floor.
H-187	4.00	2.5	No	n	0.5	10+	1	2	0	0	Charred through 1-3/4" floor.
H-194	4.00	2.5	No	n	0.5	10+	2	4	0	0	Charred through 1-3/4" floor.
H-173	4.00	0.0	No	n	0.9	15+	1	2	0	0	Flame dying rapidly at 15 min. Low point at 1-1/2 mins.
H-201	4.00	2.5	No	n	1.0	5	0.5	0	0	0	At 1 min. burned 3 small holes in ceiling 1/8-1/4" dia. Molten material running through formed stalactites.

Footnotes: First 6 runs (H-77 to H-173) made on Design #1 structures; H-201 on Design #1 except plastered floor.

(ft) Regular Army thermite hand-grenade or air-borne incendiary bomb.

(A) Numbers in parentheses refer to number of laths burned through.

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W.M., Jr., 1/2/42

APPENDIX III
TEST BURNS - ATTIC STRUCTURE

Gasoline Soap Without Binder

Appendix III - P. 9

Run No.	Wt. of Charge (lbs.)	Distance from Burn Leaves (ft.)	Positive Burn (ft.)	Fuel (lb.)	Burning Time Minutes	Amount Charred or Burned			Remarks
						Floor (sq. ft.)	Joists (ft.)	Laths (ft.)	
H-114	0.65	2.5	No	Batch 115	2.0	5	2	4	Charge flung against roof; 95% dropped to floor in 1 ft. circle.
H-98	0.65	2.5	Yes	Batch 72E	2.5	10*	4	5	Slow starting. Hard gel retarded flames.
H-115	0.65	2.0	No	Batch 115	1.0	2	4	5	Charge flung against roof; splattered over entire structure.
H-105	1.00	3.5	No	Batch 85	4.5	6	2	4	Rapid extinction after fuel was exhausted.
H-108	1.00	3.5	Brdle	Batch 112	3.5	6	2	4	More rapid burning than Batch 85. Softer gel.
H-106	1.00	5.0	Yes	Batch 85	4.5	12*	12	25	14(4) 14(9) 14(4)
H-109	1.00	3.0	Yes	Batch 112	3.5	10*	6	9	Eaves afire after 2 mins., rapidly spreading flames.
H-139	1.25	3.0	No	Batch 125	2.0	4	5	11	Section ignited after 1 min. Low point at 4 minutes.
H-140	1.25	3.0	No	"	2.0	6	4	6	Flash burn.
H-103	1.30	3.5	Brdle	Batch 72E	3.5	8	6	8	Flash burn.
H-119	1.35	5.0	Brdle	Batch 115	3.0	10*	5	8	14
H-107	1.50	3.5	Yes	Batch 85	6.0	10*	8	10	14(4) 14(6) 14
H-156	2.30	3.5	No	Batch 134	4.0	5.5	9	11	Eaves afire after 4 mins.; low point after 7 minutes.
H-187	2.30	3.5	Yes	Batch 125	3.5	10*	8	10	14(4) 14(6) 14
H-126	2.50	4.5	Brdle	Batch 112	6.0	10*	9	12	Flames reached eaves just as fuel was exhausted.
H-182	2.30	-	Yes	Batch 167	2.0	10*	2	4	Charge blown into eaves from 5' back casting(20 g. black powder). Only 50% of charge went into eaves. Moderately low point.
H-137	1.20	3.0	No	Batch 115	3.5	10	2	4	at 4, 8 and 10 minutes.
HH-112	1.00	2.5	No	Batch 112	4.0	8	1.5	2	Plastered floor retarded spread of flames.
HH-115	2.00	3.0	No	"	5.0	7	7	16	Eaves afire only after 5 minutes, charred slightly.
HH-127	2.50	5.5	No	"	4.0	6	5	12	Plastered floor retarded spread of flames.
HH-128	2.5*	2.5	Yes	"	5.0	12	6	14	Eaves afire at 2 mins. Transite roofing exploded, blowing out large parts of the flame.
HH-129	4.00	5.5	Yes	Batch 118	8.0	10*	10	20	Eaves afire at 4 mins. Ditto above.

Footnotes: First 16 runs (H-114 to H-182) made on Design #1 structure; Run H-157, Design #1 except plastered floor; last 5 runs(HH-112 to HH-129) Design #2.
 (*) Fuel composition data shown on attached sheet.
 (A) Numbers in parentheses refer to number of laths burned through.

WTR, Jr., /31 1/2/42

Appendix III
**TESTS ON AEROSOL CONSTRUCTION
 Enclosed Shrine in Steel-enforced Gypsum**

Appendix III - p. 10

Run No.	Wt. of Charge (lbs.)	Distance from Burn Eyes (ft.)	Positive Burn	Fuel (lb.)	Burning Time Minutes Active Total	Amount Charred or Burned				Percent Binder	Remarks
						Floor joists (sq. ft.)	Joists Rafters (ft.)	Laths (ft.)	Left Cent. Right		
H-39 1.00	2.6	No	Batch 85	5.5	10+	1.5	2	0	0	0	10
H-110 1.00	5.0	No	Batch 112	8.0	10+	3	4	0	0	0	10
H-100 1.00	2.5	No	Batch 85	3.0	5+	1.5	2	0	0	0	10
H-111 1.00	2.5	Yes	Batch 112	6.0	10+	6	8	10	-	10(8)	8
H-151 2.00	2.5	No	Batch 85	4.0	10+	4	4	8	2	6	4
H-118 0.65	2.0	Brod's	Batch 115	1.5	10+	5	4	8	-	11(1)	4
H-119 0.65	2.0	Yes	"	2.5	10-	4	5	10	7	9(5)	7
H-122 1.00	3.0	No	Batch 116	10	10+	0	0	5	0	0(6)	6(5)
H-117 1.35	3.0	Yes	Batch 115	2.5	10+	6	8	12	5	14(9)	-
H-125 2.50	4.5	No	Batch 116	6.0	10+	3	4	2	0	1.5	
H-186 2.30	5.5	No	Batch 141	10	10+	2	3	0	0	0	2
H-169 2.50	3.5	No	Batch 140	7.0	10+	2	3	0	0	0	2
H-170 2.30	3.0	Yes	"	7.0	15+	Complete destruction. Leaves fire at 7 minutes. Floor sections afire at 14 minutes.				All roof section afire at 10 minutes. All	
H-171 2.50	3.0	Yes	Batch 141	10	20+	5	8	18	6(2) 14(9)	7(3)	2

Footnotes:
 (K) All runs made on Design #1 structure.
 (L) Fuel composition on data shown on attached sheet.
 (J) Numbers in parentheses ref. to number of laths burned through.

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TEST BURNS - ATTIC STRUCTURE

Sisal and Moss Binder in Various Fuels

Run No.	Wt. of Charge (lbs.)	Distance from Hayes (ft.)	Positive Burn	Fuel (lb.)	Burning Time Minutes	Amount Charred or Burned Active Total (sq. ft.)	Floor Joists Rafters Laths (A)			Percent Binder	Remarks
							(ft.)	(ft.)	Left. Cent. Right		
H-133	1.00	3.0	No	Batch 118	4.5	8	1.5	2	0	0	0
H-131	1.30	3.0	No	"	2.0	7	3	5	8	2	10
H-132	0.65	2.5	Yes	"	2.0	10+	6	8	9	11(6)	6
H-130	1.30	3.5	No	"	3.0	8	2	4	2	0	0
H-141	1.25	3.0	Yes	Batch 125	3.5	10+	6	8	12	8	12(7)
H-149	1.25	2.5	Yes	"	3.5	10+	4	5	9	6	12(7)
H-142	2.30	3.5	No	Batch 127	3.5	12	4	6	12	6	12
H-143	2.30	3.5	No	"	4.0	15+	3	5	8	0	10
H-144	2.30	3.5	Brdle	"	4.5	15+	4	6	12	7	11(5)
H-150	2.00	2.5	Yes	Batch 122	5.0	10+	5	6	16	6	14(9)
H-157	2.30	3.5	No	"	6.0	12+	2	3	0	0	0
H-158	2.30	2.5	Yes	"	8.0						
H-161	2.30	3.5	No	Batch 136	5.0	10+	1	2	0	0	0
H-160	2.30	3.0	Yes	"	5.0	25					
H-162	1.25	3.0	No	"	5.5	10+	1.5	3	0	0	0
H-163	1.25	2.5	No	"	4.0	20+	2.5	4	1	0	2
H-164	1.25	2.0	Yes	"	5.5	15+	6	8	24	12(6)	14(12)12(7)
H-168	2.50	3.5	No	Batch 135	7.0	10+	2	3	0	0	0

Footnotes: All runs made on design #1 structure.
 (R) Fuel composition on data shown on attached sheet.

(A) Numbers in parentheses refer to number of lathe burned thru.

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(Moss binder used only in Runs H-131 & 133)

Moss binder retarded flames too much.

Charge flung against roof. Moss binder not as retentive of fuel as sisal. (H-130).

Charge flung against roof - seemingly held together as well as with 1.5% binder. (H-130).

1 ft. circle.

Binder retarded flames for optimum burning time.

Fuel too fluid to be retained by binder.

Increased percentage binder retarded flames somewhat.

However, fuel burned too rapidly.

Complete destruction of all three sections (roof and floor) in approximately 10 mins. Hayes afire at 2 mins.; right section afire at 7 mins.; left section at 10 mins. Fuel & binder same as H-137.

Complete destruction of all three sections (roof and floor) in 20 minutes. Hayes afire at 7 mins., low point at 15 mins. Fuel and binder same as H-131.

Complete destruction of all three sections (roof and floor) in approximatly 10 mins. Fuel unburned after 10 mins.

25% fuel unburned after 10 mins. Chopped sisal employed instead of usual long-fiber sisal.

25% fuel unburned after 10 mins. Chopped sisal.

TEST BURNS - ATTIC STRUCTURE

Notes:

Fuel composition in samples shown on attached sheet

Design #1 except plastered floors
First 9 runs(H-165 thru H-198) on Design #1 structure; H-165 thru H-198 on attached screen.

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Appendix III
TEST BURNS - ATTIC STRUCTURE

Appendix III - D. 15

Run No.	Wt. of Charges from Eaves (lbs.)	Distance from Burn (ft.)	Positive Burn	Fuel (z)	Burning Time Minutes	Amount Charred or Burned Active Total (sq. ft.)	Bottom Paste Binder in Various Fuses			Percent Binder	Remarks		
							Floor Joists	Rafters	Laths (A)				
H-54	0.70	3.5	No	P.C.Naph.	5	5	6	6	0	3	0	Fuel leaked through floor.	
H-55	0.85	4.0	No	"	5	5	4	0	0	0	0	Ditto above.	
H-94	1.10	3.5	No	Eco Base	2.5	7+	2	5	0	0	0	Floor cracks sealed with putty - no fuel leaked thru.	
H-95	1.00	2.5	Yes	"	5.0	12+	4.5	7	12	6	9(5) 6	Ditto above.	
H-65	1.70	3.5	No	"	3.0	10+	1	1	0	0	0	Fuel leaked through floor.	
H-102	1.00	5.5	No	Batch 85	4.5	10+	1.6	2	0	0	0	Flash burn at start, followed by low flame for long period.	
H-124	1.00	5.0	No	Batch 118	5.0	10+	1.6	2.5	0	0	0	Small amount of fuel left unburned.	
H-120	1.00	2.5	Yes	Batch 112	10+	4	6	10	7	14(7) 14(12)	10.0	Eaves afire at 2 minutes, no low point.	
H-125	1.00	2.5	Yes	Batch 116	5.0	10+	5	6	11	8	10(6) 8	Eaves afire at 4.5 minutes.	
H-148	2.00	3.5	No	Batch 123	2.0	20+	3	6	4	0	4	Flash burn at start, followed by low flame for long period.	
H-104	2.00	3.0	Yes	Batch 85	8.0	12+	6	8	12	14(8) -	10.0	Eaves afire at 4.5 minutes.	
H-145	2.30	5.5	No	Batch 127	2.5	20+	2	4	2	0	5	0	6.0
H-121	0.90	3.0	No	Eco Base	10+	1	2	0	0	0	0	Flash burn.	
HR-89	1.05	2.0	No	Batch 81	10+	0.5	0	0	0	0	0	Flash burn at start, followed by low flame for long period; 40% fuel unburned after 10 minutes.	
HE-90	2.10	2.0	No	"	1.5	11+	1	1	2	0	4(1) 0	9.1	
HE-91	3.00	2.0	No	"	2.0	14+	4	3.5	8	2	10(5) 0	9.1	

Footnotes: First 12 runs (H-54 to H-145) made on Design #1 structure; Run H-121, Design #1 except plastered floor; last 3 runs on Design #2.
 (X) Fuel composition data shown on attached sheet.
 (A) Numbers in parentheses refer to number of laths burned through.

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TEST BURNS - ATTIC STRUCTURE
Cotton Bags With Various Fuels

Run No.	Wt. of Charge (lbs.)	Distance from Eaves (ft.)	Positive Burn	Fuel (lb.)	Burning Time Minutes	Amount Charred or Burned			Remarks
						Floor joists (sq. ft.)	Rafters (sq. ft.)	Laths	
H-50	1.10	4.0	No	Batch 3A	.10+	2	3	0	Small fire. Major portion of fuel leaked thru floor.
H-90	1.25	5.5	No	Kerosene	.1	4	4	6	Perforated cardboard mailing tube used to distract bag. Flash burn.
H-97	1.30	3.5	No	Batch 4A	2	0	0	0	Double bag used to contain fuel.
H-152	1.20	3.0	No	Batch 127	7	10+	1.5	2	Fuel drained from bag very slowly.
H-155	1.40	5.0	No	Batch 127A	6	10+	2	4	Ditto above.
H-154	1.85	3.0	No	Batch 127B	5	6	2	4	Ditto above.
H-158	1.40	3.0	No	Batch 58A	9	10+	2	3	Ditto above.
H-159	1.80	3.0	No	Batch 58B	6	10+	2	5	Ditto above.
H-236	1.90	5.0	No	Batch 4A	2	5	6	9	Flash burn effect. Plastered floor helped direct fuel toward eaves.
H-155	1.25	3.5	No	Batch 4A	3	10	3	5	Fuel spreading retarded by plaster ridges on floor.
H-236	1.20	3.0	No	Batch 4B	2.5	8	3.5	6	Increased proportions of kerosene in fuel charge did not appreciably lengthen burning time.

Footnote: First 8 runs on Design #1; last 3 runs on Design #1 except plastered floor.
 (#) Fuel composition on date shown on attached sheet.

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COMPOSITION OF FUELS USED IN TEST BURNS

Appendix III

Appendix III - P. 15

A. Straight Fuels									
P.C.	Naphtha	"	8 pound Reid vapor pressure, 360°F. E.P. gasoline.	"	"	"	"	"	"
Baseo Base	"	"	40% heavy naphtha, 300-400°F. B.P.)	10% light naphtha, 110-265°F. B.P.)	winter grade.				
Batch 3A	"	"	60% baseo base, 50% Diesel fuel oil, 395-660°F. B.P.)	"	"				
Batch 4A	"	"	"	"	"				
Batch 4B	"	"	40% " CC ₂ "	"	"				
B. Crepe Rubber Blends	Batch 5A		Base Stock	Grease Rubber, wt. %					
"	58A		Benzol	5.5					
"	56B		Easeo Base	0.12					
"	75		"	0.04					
"	81		"	5.5					
C. Smoked Sheet Rubber Blends	Batch 12		Base Stock	Smoked Sheet Rubber, wt. %					
"	12		Benzol	7.0					
D. Latex Blends (L)	Batch 30		Base Stock	Latex, wt. %					
"	34		Portane	5.0					
"	41		P.C. Naph.	4.0					
"	45		"	4.0					
"	44		Diesel Fuel Oil	4.0					
"	47		Portane	4.0					
"	50		Light Naph.	4.0					
"	51		Heavy Naph.	4.0					
"	65		Heating Oil	4.0					
"	67		P.C. Naph.	5.0					
"	117		P.C. Naph.	5.5					
E. Gasoline Soap Blends	Batch 5E		Base Stock	Composition = wt. %					
"	93		P.C. Naph.	1.0					
"	112		Easeo Base	2.5					
"	115		"	3.5					
"	116		"	2.4					
"	118		"	2.6					
"	122		"	3.5					
"	125		"	2.5					
"	127		"	2.5					
"	127A		"	30% Cup Grease(10% calcium-hog fat soap; 25 H ₂ O, 75% lube oil distillate).					
"	127B		"	10% Ditto above.					
"	128		"	25% Ditto above.					
"	129		"	65% Aluminum Naphthenate, 10% Heavy Naphtha, 32% Lube Oil Distillate.					
"	135		"	10% Aluminum Oleate, 8% Heavy Naphtha, 32% Lube Oil Distillate.					
"	136		"	3.5					
"	140		"	3.5					
"	141		"	5.5					
"	152		"	3.5					
"	157		"	3.5					
"	209		"	3.5					

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Appendix IV
Explosion Tests

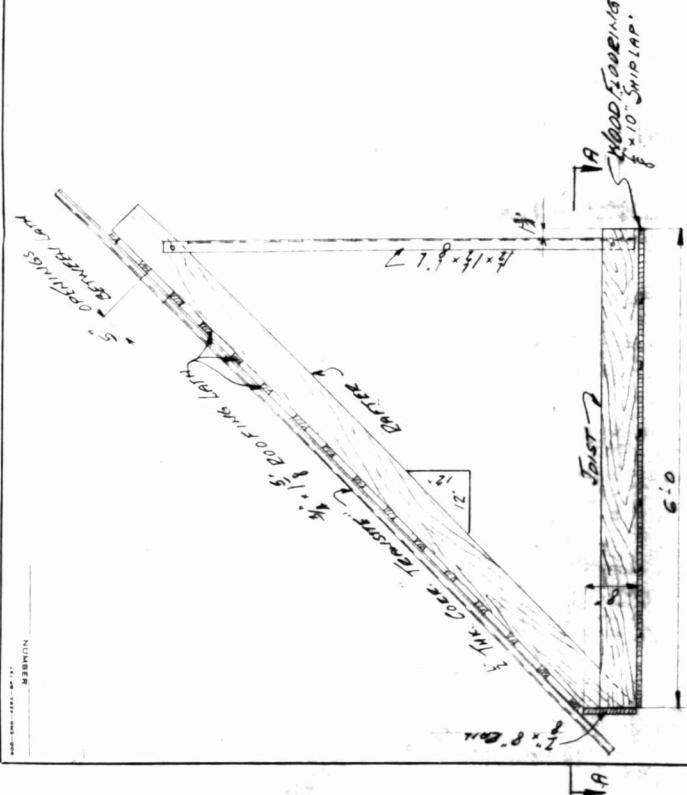
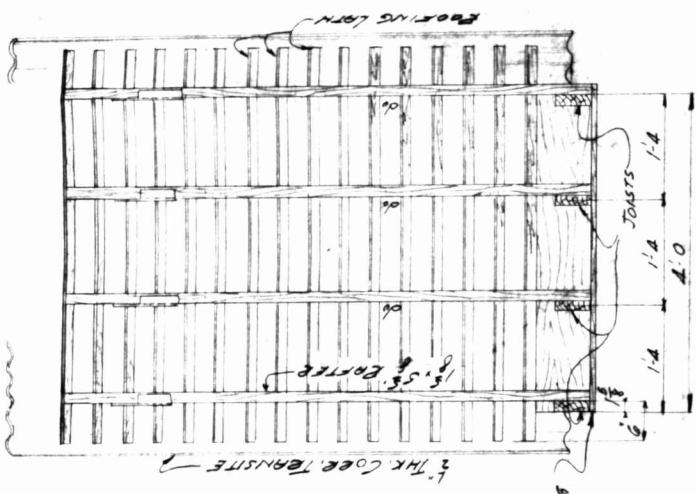
No.	Case	Powder	Fuel	Temp. •F.	Ignitability	Remarks
1	5# pail	20 gms. in bottom pan	3# 2.4% rubber latex	60	Powder chamber in base satisfactory.	
2	5# pail	30 gms. in bottom pan	3# 2.4% rubber latex	60	Cheat on No. 1. 90% in 1 blob.	
3	5# pail	30 gms. in bottom pan	3# batch 27, 2.4% rubber latex in P.C. Naphtha	-40	Poor	
4	5# pail	20 gms. in bottom pan	3# batch 27, 2.4% rubber latex in P.C. Naphtha	+28	Good	90% in 1 blob.
5	5# pail	30 gms. in bottom pan	3# batch 28, 2.6% rubber latex in P.C. Naphtha	+50	Good	50% in 1 blob - rest in small pieces.
6	5# pail	30 gms. in bottom pan	3# batch 28, 2.6% rubber latex in P.C. Naphtha	+50	Good	Whole charge finely divided.
7	-	-	-	-	-	
8	5# pail	20 gms. in bottom pan	3# batch 28, 2.6% rubber latex in P.C. Naphtha	+60	Good	Lit with match - flame not different from P.C. Naphtha.
9	5# pail, lid soldered on	20 gms. in bottom pan	3# batch 28, 2.6% rubber latex in P.C. Naphtha	+60	Good	Divided into many small pieces.
10	5# pail	20 gms. in bottle at bottom	3# batch 28	+60	Good	Did not eject well - half charge left in can.
11	5# pail	30 gms. in bottle at bottom	3# batch 28	+60	Good	Poor ejection - central bottom charge of powder will not do.
12	5# pail	20 gms. in bottom pan	3# batch 28 in 3 1/2 paper bags	+60	Good	Bags thrown out without breaking.
13	5# pail	20 gms. in bottom pan	3# batch 35, 2.4% rubber latex	+60	Poor	Fuel was much thicker than usual but many small pieces were thrown out.
14	5# pail	30 gms. in bottom pan	3# batch 35, 2.4% rubber latex	+60	Good	Bomb was laid on its side, fuel thrown 25 ft. horizontally in one go.
15	5# pail	30 gms. in bottom pan	3# batch 35, 2.4% rubber latex	+60	Good	Waste balls controlled size of fuel gods quite satisfactorily.
16	5# pail	30 gms. in bottom pan	3# batch 35, 2.4% rubber latex in 2 paper bags	+60	Good	Bags ripped apart.

No.	Temp. °F.	Type	Igni- tion temp. °C.	Remarks
17	5°	Ball	20 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Good Held as 1 large glob.
18	17°	Center ball	20 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Good Smaller over 5° radius.
19	5°	Ball	20 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Good 5 blots, none ignited. Reid 5.7%.
20	5°	Ball	20 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Good Thrown out in 1 blob. Reid 6.7%.
21	3°	Pipe	30 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Bad Thrown out in many small pieces.
22	5°	Pail	100 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Good Thrown out in 6 balls - 5 ignited.
23	5°	Pail	100 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Good Waste too tightly packed.
24	5°	Pail	100 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Good
25	5°	Pail	100 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Bad
26	5°	Pail	100 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Good
27	5°	Pipe	100 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Good
28	3°	Pipe	15 gms. In pan 3" dia. 1" center tube 1" height + 3	Bad Magnesium turnings did not help ignition.
29	5°	Pail	20 gms. In pan 3" dia. 1" center tube 1" height + 4	Good No ignition in tube.
30	5°	Pail	20 gms. In pan 3" dia. 1" center tube 1" height + 4	Good Blown into many small pieces.
31	3°	Pipe	30 gms. In pan + 3 grams. Mg turnings	Good Excellent performance.
32	3°	Pipe	30 gms. In pan 3" dia. 1" center tube 1" height + 4	Bad No sign of big turnings igniting.
33	5°	Pail	30 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Good Distribution good.
34	5°	Pail	30 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Bad Winter gasoline - 10.1% Reid vapor pressure. -71°F. too cold for 13% Reid vapor pressure gasoline.
35	5°	Pail	30 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Good Flash burn in mid air.
36	3°	Pail	20 gms. In bottom pan 3" dia. 1" center tube 1" height + 4	Good

No.	Cane	Powder	Fuel	Temp., °F.	Ignition	Remarks
37	5 ¹ / ₂ pail	30 gms. In bottom pan	3 ¹ / ₂ batch 66. 5.5% crepe In benzene	•60	Good	Splatter in many small pieces.
38	5 ¹ / ₂ pail	30 gms. In bottom pan	3 ¹ / ₂ , 4 cotton bags containing rubber gel	•60	Good	Good shot size control.
39	5 ¹ / ₂ pail	30 gms. In bottom pan	3 ¹ / ₂ batch 49. 2.4% rubber latex	•42	Good	Powder was sealed with copper wire and was much better protected from naphtha.
40	5 ¹ / ₂ pail	20 gms. In bottom pan	3 ¹ / ₂ batch 49	-42	Good	Electricians lacquer used to seal powder and great improvement noticed. Balls scattered 20 ft.
41	5 ¹ / ₂ pail	20 gms. In bottom pan	3 ¹ / ₂ P.C. Naphtha on 4 waste balls	•60	Good	
42	5 ¹ / ₂ pail	20 gms. In bottom pan	3 ¹ / ₂ batch 34. 2% water in 4 waste balls	•60	Poor	Packed too tight.
43	5 ¹ / ₂ pail	20 gms. In bottom pan	3 ¹ / ₂ batch 55. thin latex in 4 waste balls	•60	Good	
44	5 ¹ / ₂ pail	20 gms. In bottom pan	3 ¹ / ₂ 5.5% crepe 12 P.C. Naphtha	•60	Good	All small shots.
45	6" x 10" - 4 sears	30 gms. In bottom pan	5.5% crepe rubber 14 P.C. - Naphtha	•60	Good	Did not scatter much, need
46	5 ¹ / ₂ pail	15 gms. In bottom pan	Batch 735 - 5 balls of rubberless gels - N.P. 120° F.	•60	Good	Preliminary test of rubberless gels seems promising.
47	5 ¹ / ₂ pail split seas	20 gms. In bottom pan	3 ¹ / ₂ 5.5% crepe 14 P.C. Naphtha	•60	Good	Lid wired on, sides split but needs more powder to throw far.
48	3" pipe	20 gms. In bottom pan	2-1/2" batch 48. 5.5% crepe + 20 gms. coal	•60	Poor	Cool did not help ignition.
49	3" pipe	10 gms. In bottom pan	2-1/2" batch 48	•60	Poor	
50	3" pipe	5 gms. In bottom pan	2-1/2" batch 48	•60	Poor	
51	5 ¹ / ₂ pail with side seas	40 gms. In bottom pan	3 ¹ / ₂ batch 48	•60	Good	Splattered all over - extra sides seem did not help control.
52	5 ¹ / ₂ pail	30 gms. In bottom pan	3 ¹ / ₂ 2% crepe in winter gasoline - 3 wads of excelsior	•60	Good	Worked fairly well.
53	5 ¹ / ₂ pail	30 gms. In bottom pan	2 ¹ / ₂ batch 88 - solid gel	•60	Good	Shattered and flash burnt.

No.	Case	Powder	Fuel	Temp. °F.	Igni- tive	Remarks
54	5# pall	20 gms. in bottom pan	3 $\frac{1}{2}$ batch 88 - 4 excelsior balls	+60	Good	4 separate balls of fire formed. Waste holds this solid gel together.
55	5# pall	20 gms. in bottom pan	3 $\frac{1}{2}$ batch 88 - 3 waste balls	+60	Good	
56	5" x 24" H-46	75 gms. in center tube	10 $\frac{1}{2}$ batch 43	+60	Fair	
57	3" pipe	12 gms. in spaghetti tube	2-1/2 $\frac{1}{2}$ batch 48 - 5.5% crepe in Winter gasoline	+60	Good	First time ignition secured in 3" pipe bomb.
58	5" x 24" H-46	70 gms. in center tube	10 $\frac{1}{2}$ Winter gasoline in 1 $\frac{1}{4}$ waste	+60	Good	Too much gasoline wasted but made good fire.
59	3" pipe	17 gms. in spaghetti tube	2-1/2 $\frac{1}{2}$ hard gel	+60	Good	Flash burn.
60	3" pipe	15 gms. in center tube - bottom igniticia	2-1/2 $\frac{1}{2}$ 5.5% crepe in gasoline	+60	Good	Blew out in 1 blob.
61	3" pipe	15 gms. in center tube	2-1/2 $\frac{1}{2}$ gasoline in waste	+60	Good	Blew out in 1 pieces.
62	3" pipe	17 gms. in center tube of glass	2-1/2 $\frac{1}{2}$ hard soap	+60	Good	Glass tube shattered and ejection was poor.
63	3" pipe	17 gms. in fibre tube	2-1/2 $\frac{1}{2}$ batch 116 in extensor	+60	Good	Flash burn.
64	3" pipe	17 gms. in central fibre tube	2-1/2 $\frac{1}{2}$ cup grease cut back with gasoline	+60	Good	Waste burns too long.
65	3# pall	20 gms. in bottom pan	3 $\frac{1}{2}$ hard soap in 0.6% 51gel	+60	Good	Flash burn.
66	3# pall	20 gms. in bottom pan	3 $\frac{1}{2}$ hard soap in 0.6% 51gel	+60	Good	Flash burn.
67	3" pipe	20 gms. in fibre tube	2-1/2 $\frac{1}{2}$ soap in 0.6% 51gel	+60	Good	Poor performance - too large a fibre tube used.
68	3" pipe	9 gms. in small fibre tube	2-1/2 $\frac{1}{2}$ 5.5% crepe in Winter gasoline	+60	Poor	Fibre tube did not burn along length.
69	3" pipe	15 gms. in fibre tube	2-1/2 $\frac{1}{2}$ 5.5% crepe in Winter gasoline	+60	Good	Horizontal - blown to wall at 15'. Performance excellent.
70	3" pipe	20 gms. powder in large tube	2-1/2 $\frac{1}{2}$ 5.5% crepe in Winter gasoline	+60	Poor	Same as 69.
71	3" pipe	17 gms. powder in medium tube	2-1/2 $\frac{1}{2}$ 5.5% crepe in Winter gasoline	+60	Good	Excellent performance against wall.

No.	Case	Powder	Fuel	Temp. °F.	Ignition	Remarks
72	3" pipe	17 gms. powder	2-1/2% batch 122	+60	Good	Flash burn with no filler.
73	3" pipe	17 gms. powder	2-1/2% Al naphthalene	+60	Good	Melted and gave flash burn.
74	3" pipe	17 gms. powder	2-1/2% formula 122 in paper cylinder	+60	Cool	Flash burn.
75	3" pipe	17 gms. powder	2-1/2% formula 122, 2% Sisal	-10	Failed to eject charge.	
76	3" pipe	17 gms. powder	2-1/2% formula 122, 1% Sisal	-10	Good	Performance good.
77	3" pipe	17 gms. powder	2-1/2% formula 122, 3-1/3% chopped Sisal	+60	Good	Excellent performance.
78	3" pipe	20 gms. powder	2-1/2% formula 122, 1% chopped Sisal	+60	Good	Fair performance.
79	3" pipe	20 gms. powder	2-1/2% formula 122, 2% chopped Sisal	+60	Good	Excellent performance at 15 ft.
80	3" pipe	20 gms.	2-1/2% formula 122, 2% short Sisal	-40	Good	Performance good.
81	3" tube with cork	20 gms.	2-1/2% formula 122, 2% short Sisal	+60	Good	Split setup.
82	3" tube with cork	20 gms.	2-1/2% formula 122, 2% short Sisal	+60	Good	Soldered joint held.
83	3" tube	20 gms.	2-1/2% 5.5% crepe directed at ridge	+60	Good	Showed ridges was not the place to direct charge.
84	3" tube	20 gms.	2-1/2% formula 122	+60	Good	No explosive burn in end.
85	3" tube	20 gms.	2-1/2% formula 122	+60	Good	In roof structure aimed at roof.
86	3" tube	20 gms.	2-1/2% formula 122, 1% Sisal	+60	Good	Into eaves of roof - worked excellently.
87	3" tube	20 gms.	2-1/2% formula 122, 2% Sisal	+60	Good	Laid on side in open - case travelled farther than fuel.
88	3" tube	20 gms.	2-1/2% 5.5% crepe in winter gasoline	-60	Poor	Ignition was very poor - little splattering.
89	3" tube	20 gms.	2-1/2% 5.5% crepe in winter gasoline	+60	Good	Performance excellent.
90	3" tube	20 gms.	2-1/4% formula 122	-60	Fair	
91	3" tube	20 gms.	2-1/4% formula 122	+60	Good	Performance excellent.
92	3" tube	20 gms.	2-1/4% formula 122, 1% Sisal	-60	Fair	
93	3" tube	20 gms.	2-1/4% formula 122, 1% Sisal	+60	Fair	Performance excellent.

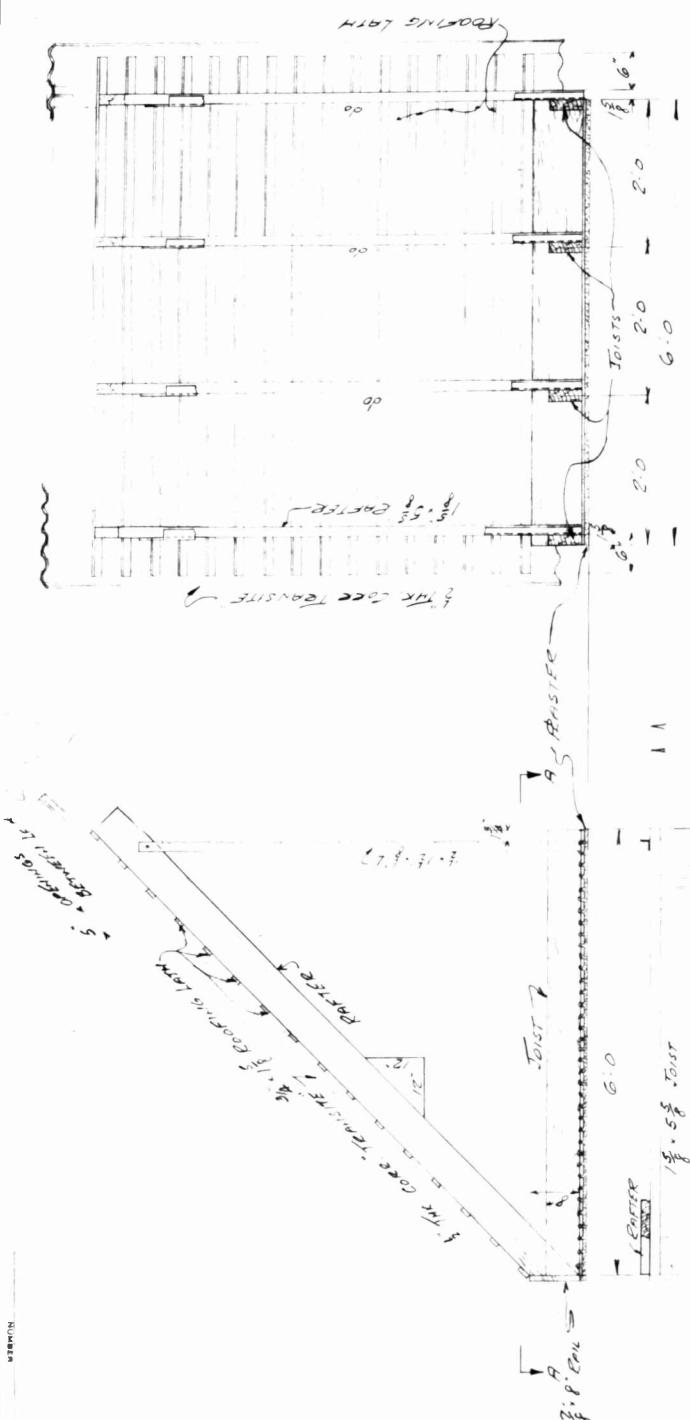


Appendix Figure A
Attic Test Structure
Design No.1

Design No. 1
P D R C 199

BUREAU
COUNTRY OR STATE
JOB
PLACE

STANDARD OIL DEVELOPMENT CO.
GENERAL ENGINEERING DEPT.
ELIZABETH, N. J.



DESIGN No. 2
P D R C 199
TYPICAL SECTION X-X
INCLINED
SCALE: 3'-0"

REVISIONS		DATE	DRAWN BY	CHECKED BY	APPROVED BY	REV.	CROSS	CIRC.	SHEET	DRAWER
NO.	DESCRIPTION	DATE	CHECKED	APPROVED	REV.	CROSS	CIRC.	SHEET	DRAWER	
1										
2										
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7										
8										
9										
10										

STANDARD OIL DEVELOPMENT CO.	
GENERAL ENGINEERING DEPT.	
DESIGNER	CONTRACTOR
DATE	DATE
LOCATION	LOCATION
NAME	NAME
COMP.	COMP.
DESIGN	DESIGN
DRWGS.	DRWGS.
20907	20907
10	10

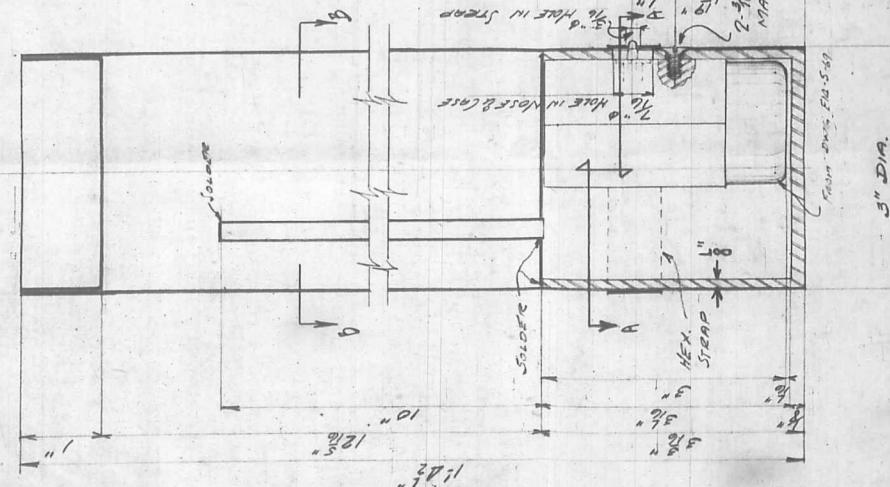
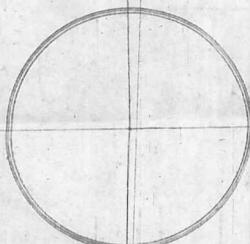
SCALE	DRAWING	DRWGS.
1/2 INCH = 2'-0"	20907	10

USING = .015 (.72 GR) MATERIAL FOR BODY

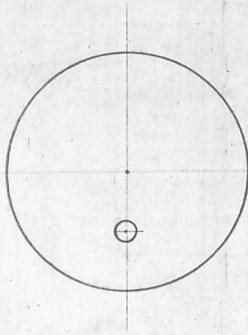
TOTAL WT. CONTAINER	-	2.019 "
" " FUSE	-	.500 "
TOTAL WT.	=	2.519 "

CONTAINER + FUSE	-	2.519 "
LIQUID CHARGE	-	2.000 "
POWDER "	-	.120 "
TOTAL GROSS	-	2.639 "

SECTIONAL DENSITY = .655



SECTION B-B



Appendix Measure C
PETROLEUM INCENDIARY BOMB
20 Gauge Body and Standard Fuse

USING = .035 (.70 GR) MATERIAL FOR BODY

TOTAL WT. CONTAINER	-	2.079 "
" " FUSE	-	.500 "
TOTAL WEIGHT	=	2.579 "

CONTAINER + FUSE =	2.579 "
LIQUID CHARGE =	2.000 "
POWDER CHARGE =	.120 "
TOTAL GROSS	2.699 "

APPENDIX



SECTION A-A
SECTIONAL DENSITY = .776

REVISIONS

DATE	BY	REV.
1/16/49	SA AGENT	C.R.D.
"	2 1/2 hrs	C.R.D.
"	2 hrs	A.M.C.
"	2 hrs	S.C.

56-10

PLACE: STANDARD OIL DEVELOPMENT CO.
GENERAL ENGINEERING DEPT.
ELIZABETH, N.J.

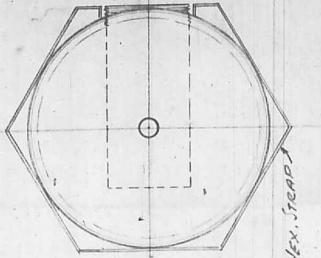
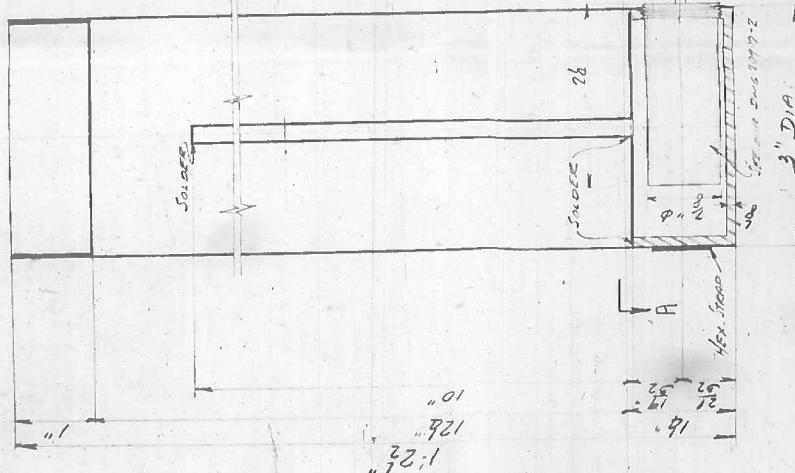
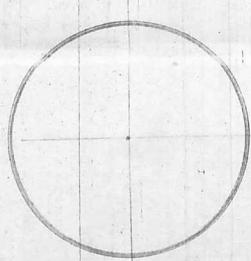
DATE: 1/16/49
SHEET: 1 OF 1
SCALE: DRAWER /

Using - .015" (2½ GA.) MATERIAL FOR BODY
 TOTAL WT. CONTAINER = 1.43 "

$$\begin{array}{rcl} \text{FUSE} & = & .23 \\ \text{TOTAL WT.} & = & 1.66 \end{array}$$

$$\begin{array}{rcl} \text{CONTAINER + FUSE} & = & 1.66 \\ \text{LIQUID CHARGE} & = & 2.20 \\ \text{POWDER} & = & .10 \\ \text{TOTAL GROSS} & = & 4.00 \end{array}$$

SECTIONAL DENSITY = .566



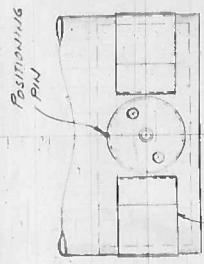
SECTION A-A

USING - .035" (20 GA.) MATERIAL FOR BODY
 TOTAL WT. CONTAINER 2.105 "

$$\begin{array}{rcl} \text{" FUSE} & = & .250 \\ \text{TOTAL HEIGHT} & = & 2.355 \end{array}$$

$$\begin{array}{rcl} \text{CONTAINER + FUSE} & = & 2.355 \\ \text{LIQUID CHARGE} & = & 2.000 \\ \text{POWDER} & = & .120 \\ \text{TOTAL GROSS} & = & 4.475 \end{array}$$

SECTIONAL DENSITY = .633

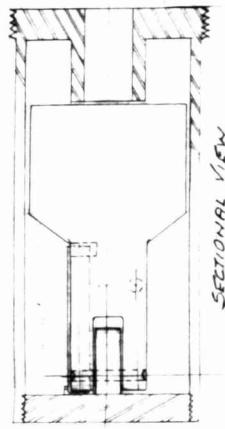


HEX. STRAP

REVISIONS		DATE	BY	DATE	BY	DRAWER
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1						7
REV.						
CHD.						
S.C.						

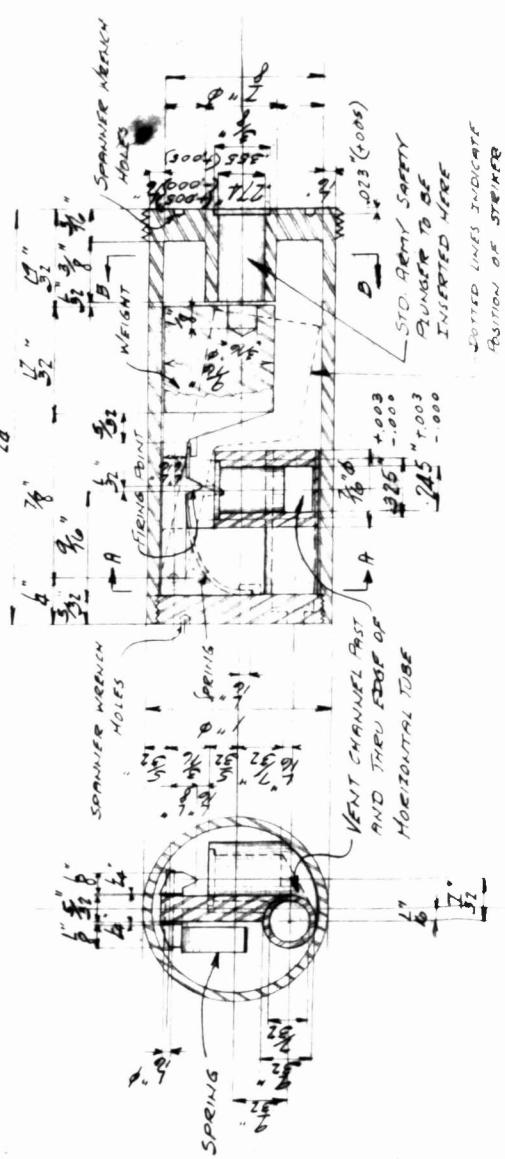
COUNTRY OR STATE		NAME	DATE	DRAWER
PLACE	CO.	DESIGN.	GENERAL ENGINEERING DEPT.	
ELIZABETH, N. J.	STANDARD OIL DEVELOPMENT CO.			

FIG 11



SECTIONAL VIEW

26



SECTION A-A

Scallop 2 =

Ber.

Section 8-8 Exo. 11

APPENDIX FIGURE E
DATA'S OF NEW PUZZLE

REEL - C

1 332

A.T.I.

3 1 6 0 8

TITLE: The Development of Oil Incendiary Bombs

AUTHOR(S): Russell, R. P.

ORIGINATING AGENCY: Standard Oil Development Co.

PUBLISHED BY: Office of Scientific Research and Development, NDRC, Div. 11

ATI- 31608

REVISION

0010. AGC

176

PUBLISHING

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DATU	DOC. CLASS.	COUNTRY	LANGUAGE	PAGES	ILLUSTRATIONS
Feb '42	Conf'd'l	U.S.	Eng.	59	drwgs, tables, graphs

ABSTRACT:

A quarterly report is given on the development of oil incendiary bombs. Burning tests have been carried out in several attic structures in which the destructiveness and completeness of any fire are taken as criteria, and special attention has been paid to developing mixtures and methods of application which would cause destructive attic fires. It has been found that attic structures can be destructively burned with the minimum amount of fuel if the incendiary is placed near or discharged into the eaves line where roof lath and rafters serve as kindling. Results show that with a given weight of fuel located at a given distance from the eaves, thickened gasoline is equal to magnesium and is distinctly superior to thermate. Gasoline-rubber gums and gasoline soaps are equally effective in non-explosive bombs.

DISTRIBUTION: Copies of this report obtainable from Air Documents Division; Attn: MCIDXD

DIVISION: Ordnance and Armament (22)

SUBJECT HEADINGS: Bombs, Incendiary - Development

SECTION: Chemicals and Incendiaries (11)

(17440.5)

AD-B810 770

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Wright-Patterson Air Force Base
Dayton, Ohio

390369

(18) XC

(19) OSRD

(23) *Incendiary bombs

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Classification: calcined

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AUTH: OSRD Tech 3

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