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## **Demonstration of Corrosion-Resistant Coatings for Air-Conditioning Coils and Fins**

Final Report on Project F10-AR01

Alfred D. Beitelman and Susan A Drozd

June 2015



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# **Demonstration of Corrosion-Resistant Coatings for Air-Conditioning Coils and Fins**

Final Report on Project F10-AR01

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Final report

Approved for public release; distribution is unlimited.

Prepared for Office of the Secretary of Defense (OUSD(AT&L))  
Washington, DC 20301-3090

Under Project F10-AR01, "Evaluation of Corrosion Resistant Coatings for Air  
Conditioning Coils in Hawaii"

## Abstract

Department of Defense military installations operate air-conditioning equipment in humid and coastal locations that produce high levels of corrosion. Installation experience has shown that severe corrosion has reduced expected equipment life cycles by up to 50%. This project for the Office of the Secretary of Defense Corrosion Prevention and Control Program demonstrated and evaluated the performance characteristics of two corrosion prevention coatings on condenser coils in highly corrosive environments. This work assessed the capabilities and advantages of an aluminum-impregnated polyurethane coating and a flexible epoxy polymer applied to two separate condenser units at Schofield Barracks, Hawaii. The work also included fabricating and exposing test coupons of various substrates, each coated with the two coating systems and a third coating commonly used in less-corrosive environments. These coupons were exposed both at the Schofield Barracks site and in a laboratory-accelerated exposure cabinet. The work recommends that copper rather than aluminum components be specified in air conditioners operating in highly corrosive environments. A copper substrate protected with a newer type of corrosion coating is projected to extend the equipment's expected life cycle from 7–10 years to 25–30 years. The return on investment for the project is 3.89.

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## Preface

This demonstration was performed for the Office of the Secretary of Defense (OSD) under Department of Defense (DoD) Corrosion Control and Prevention Project F10AR01, “Evaluation of Corrosion Resistant Coatings for Air Conditioning Coils in Hawaii.” The proponent was the U.S. Army Office of the Assistant Chief of Staff for Installation Management (ACSIM), and the stakeholder was the U.S. Army Installation Management Command (IMCOM). The technical monitors were Daniel J. Dunmire (OUSD (AT&L) Corrosion) and David N. Purcell (DAIM-ODF).

The work was performed by the Materials Branch (CEERD-CF-M), Facilities Division (CF), U. S. Army Engineer and Structures Research and Development Center – Construction Engineering Research Laboratory (ERDC-CERL), Champaign, IL. Significant portions of this work were performed by Christopher Olaes, Dave Butler, and Larry Clark from Mandaree Enterprise Corporation, Warner Robins, Georgia. At the time of publication, Vicki L. Van Blaricum was Chief CEERD-CF-M, L. Michael Golish was Chief, CEERD-CF, and Kurt Kinnevan, CEERD-CV-T, was the Technical Director for Adaptive and Resilient Installations. The Deputy Director of ERDC-CERL was Dr. Kirankumar Topudurti, and the Director was Dr. Ilker Adiguzel.

The following individuals are gratefully acknowledge for their contributions to this project:

- Mr. Fred Ching, Directorate of Public Works, Schofield Barracks, Hawaii
- Mr. Greg Lee of Island Wide A/C Service, Honolulu, Hawaii

The Acting Commander of ERDC was LTC John T. Tucker III, and the Director was Dr. Jeffery P. Holland.

## Executive Summary

Department of Defense military installations operate air-conditioning (AC) equipment all over the world, including highly corrosive environments where their service life is shortened by damage from corrosion. Two newer coating technologies were selected and applied to the cooling vanes of air-conditioner condensers with the expectation they would better resist corrosion on common AC substrates of aluminum than the traditional, baked, phenolic coating systems which are brittle and easily damaged.

This project demonstrated and evaluated, over a 24-month period, the ability and cost effectiveness of two of the newer coatings to protect building air-conditioning condenser evaporator coils and fins from corrosion. Exposure coupons made of various substrates received various coatings and were placed at the sites of the condensers in a highly corrosive environment at Schofield Barracks, Hawaii. Additional coupons with the same combinations of substrates and coatings were exposed in a laboratory's accelerated-weathering test cabinets for 12 months, according to industry standards. Based on the test-cabinet results, copper will provide a longer service life than aluminum, and the service life of equipment can be further extended by the addition of a coating system. Equipment protected by one of the newer coatings is predicted to have a service life 3–4 times longer than equipment with traditional coatings. The return on investment ratio was calculated to be 3.89 over a 30-year period.

Based on the test cabinet results, it was recommended that HVAC components in highly corrosive environments be made of copper and coated with a manufacturer's standard coating to extend the unit's service life. It is recommended that the HVAC condensers and on-site coupons under this test be periodically revisited over the next 30 years, and the corrosion-protection performance of the two coating technologies be re-evaluated. Additional inspections will provide additional data to more accurately predict the life cycle of each coating used in this application. Recommendations were also made to update related military guidance documents to reflect findings of this effort for corrosion protection.



## Unit Conversion Factors

Multiply	By	To Obtain
inches	0.0254	Meters
microns	1.0 E-06	Meters
mils	0.0254	Millimeters
tons (2,000 pounds, mass)	907.1847	kilograms

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# **1 Introduction**

## **1.1 Problem statement**

Facilities located in highly corrosive environments operated by the Army and Department of Defense (DoD) are vulnerable to corrosion degradation of their air-conditioning (AC) coils and fins.

For example, the warm, humid island climate of Hawaii creates a high demand for AC; meanwhile, Hawaii's ocean coastal location creates a high-chloride salt environment that is highly corrosive to the aluminum components of AC systems as well as to the steel components of the associated machinery and housing. According to the Department of Public Works at Schofield Barracks, Hawaii, an AC system is scoped for a 15–20 year service life, but the actual average service life in the highly corrosive environment is closer to 7–10 years. The DoD's reported annual cost of all heating ventilation and cooling (HVAC)-related corrosion was \$1.67 million or 21% of the total costs of facility corrosion (Herzberg et al. 2010). Extending the service life of HVAC units could reduce and control the costs of AC maintenance, repair, and replacement due to corrosion.

Coatings have been previously applied to the cooling vanes of AC condensers installed at DoD installations in Hawaii; however, the traditional, baked phenolic coating systems have presented a number of application and performance problems including: (a) phenolic coatings are brittle and easily damaged; (b) the dip or flow-coat methods used in their application often result in an uneven thickness of coating material; and (c) unevenness in coating material can bridge the narrow space between cooling fins and thereby reduce heat transfer efficiency and leave exposed bare metal behind the filled spaces (Meister 2000, 583).

Under this demonstration-evaluation program, two types of selected coatings were applied and evaluated on a fixed facility's AC units at Schofield Barracks on the island of Oahu, Hawaii. There was a need to determine the life expectancy and efficiency for AC coils and fins coated with the two alternative coatings when compared to uncoated and phenolic-coated metal parts commonly used in less-corrosive environments.

## **1.2 Objective**

This project investigated whether the life cycle of standard AC parts could be increased in a highly corrosive environment by applying two new coating technologies or by specifying a different substrate than the commonly used aluminum AC components. This project demonstrated and evaluated two newer coating technologies to protect building AC condenser coils and fins from corrosion. Additionally, a service-life prediction for the demonstrated technology and various substrates was compared with that of typical phenolic-coated units and uncoated units.

## **1.3 Approach**

Two aging AC condenser units at Schofield Barracks, Hawaii, were replaced with similar units in which standard aluminum condenser coils and fins were coated with the candidate coating technologies. The two technologies used were Coating B—an aluminum-impregnated polyurethane coating (PoluAl XT from Blygold International), and Coating E—a flexible epoxy polymer coating (ElectroFin from Luvata). Each coating was applied by an authorized professional according to each manufacturer's standard. The new condenser units were monitored for 2 years to observe the performance of the new coatings. Data from the coated condensers was supplemented by data from exposure panels having 12 coupons of various substrates prepared either with standard phenolic coating, each of the candidate coating technologies, or uncoated. Identical coupons were exposed both at the two field sites and in a laboratory's accelerated-exposure test cabinets.

## 2 Technical Investigation

### 2.1 Technology overview

An AC coil is a heat exchanger that permits controlled transfer of thermal energy between different media such as the ambient air and a volatile fluid coolant (e.g., refrigerant) inside the coil. The coil's fins enhance the efficiency of the heat transfer to the air. (A standard AC condenser and evaporator coils consist of serpentine tubing surrounded by and attached to thin metal fins.) The tubing and fins may be of aluminum or copper, and they may or may not be coated. A compressor pumps the refrigerant between the evaporator and the condenser elements within a closed system. The liquid refrigerant evaporates inside the indoor evaporator coil as it absorbs heat from the indoor environment, reducing the temperature in the conditioned space inside the facility. The heated refrigerant gas is pumped outdoors to the condenser where it releases heat through the heat-exchange tubing and fins, condensing back into its liquid form for cycling back into the interior space. Figure 1 shows a condenser unit's interior layout.

Figure 1. Interior layout of a Carrier-brand condenser, shown with coated fins.



Two Carrier-brand AC condensers (30-ton and 40-ton) were procured by Island Wide Air Conditioning Services of Honolulu, Hawaii, for installation at the Schofield Barracks Bowling Center. These units and their applied coatings (described in Section 2.2; example shown in Figure 1) were evaluated as replacements for pre-existing condensers which were removed during the project. However, the facility's HVAC systems were not completely replaced under this project, and the original air-handling systems were not involved with the project equipment replacement. Application of the newer coatings was done by electrocoating or a spray process to overcome the problems associated with the dip or flow-coat methods fused for phenolic coatings. As stated previously, the older processes often produced uneven thickness of coating which could fill spaces between cooling fins, reducing heat-transfer efficiency and leaving exposed bare metal in areas behind the filled space.

After installation, the condenser systems were monitored along with coupons mounted on sample racks that were attached to each of the condensers. These field-exposed systems and coupons were monitored for 2 years. Additional coupons of substrates and coatings identical to the field-exposed coupons were exposed in a laboratory's accelerated-testing chambers and monitored for 1 year.

## **2.2 Field work**

The 30-ton condenser system was shipped to International Air Conditioning Coatings Inc., a licensed Blygold applicator in Waipahu, Hawaii, for Coating B to be applied. Prior to coating, the condenser coils were disassembled and rinsed by using a high-pressure warm-water spray system and then inspected for bent fins to ensure adequate spacing. The coils were then degreased with a pH-neutral detergent, to remove any manufacturing oils and soiling, and then rinsed again. A chromate-free conversion layer was applied to the coil by using an air-assisted airless spray gun to achieve total coverage and penetration. Upon curing of the primer, the aluminum-impregnated polyurethane topcoat (Coating B) was applied by using air-assisted airless spray equipment to ensure total penetration and coverage without bridging or significantly affecting the heat-transfer ability of the coil. The total dry film thickness of the top coat was between 20–25 microns. After curing, the coils were reassembled into the condenser unit and delivered to Schofield Barracks.

The 40-ton condenser unit was received by Island Wide Air Conditioning Services with Coating E preapplied by the manufacturer's factory in Texas. Coating E is a water-based, flexible, epoxy polymer coating process engineered specifically for HVAC/R heat-transfer coils. The electrocoating process performed by Carrier consisted of submerging the metallic coils in a paint/water bath where an electrical charge was used to deposit paint onto the coils.

The removal and replacement operation for each condenser was conducted during two separate weeks to avoid interruption in the bowling center's operation and to minimize impacts on the air-conditioning system's load while one unit was down. Prior to removing each existing condenser, power was disconnected from the unit and refrigerant liquids were vacuum-drained from the system. All copper plumbing connected to the interior units of each AC system was cut away to allow for the existing condenser's removal. A crane was used to lift the existing unit over the block retaining wall. Once the existing condenser was removed, the area was cleaned of all debris and new cushioned footings were installed for the new condenser. The crane was used again to lift the new condenser over the retaining wall and position it on the footings. New copper plumbing along with new refrigerant ball valves were installed and connected to the interior systems. Upon completion of the plumbing installation, new R-22 refrigerant was pumped through the system and primed. Electrical power was connected to the unit, and it was commissioned. This process was repeated for the second unit. Figure 2 and Figure 3 show the location of one of the units and its installation. The two condensers were located adjacent to each other outside the Schofield Barracks Bowling Center, as shown in Figure 4.

Figure 2. A condenser area cleaned of debris, with new footings in place.



Figure 3. Positioning one of two new condenser units outside Schofield Barracks Bowling Center.





Figure 4. Two units in side-by-side configuration as installed at Schofield Barracks Bowling Center.



## 2.3 Commissioning and monitoring

### 2.3.1 Field operation and monitoring

An aluminum exposure rack was fabricated for placement at each of the two AC condenser locations (Figure 5). Each rack was designed to isolate 12 coupons that were 1 in. wide x 2 in. long x 1/8 in. thick from any galvanic effects of other metals. Each rack contained four copper coupons, four 1010 steel coupons, and four 2024-T3 aluminum coupons. A single coupon of each material was coated by Carrier with Coating E. Similarly, a single coupon of each material was coated by International Air Conditioning Coatings with Coating B, and IFM Effector, Inc. of Exton, Pennsylvania, coated coupons of each material with the phenolic coating commonly used by Carrier for condensers in less-corrosive environments. The remaining coupons were left bare. Each coupon used in this program was printed with a distinctive code indicating: (a) the substrate, (b) the coating, and (c) a consecutive number for the system. The coupons and each condenser unit were photographed and assessed at 12 months and 24 months after installation. Additional photographs of the on-site coupon racks are shown in Section 3.2 and the appendix of this report. Observations of blistering

and rusting were recorded according to industry rating standards (ASTM 2009, 2012), as shown by Table 1 in Chapter 3.

Figure 5. Aluminum coupon rack was fabricated to hold 12 coupons and mounted at each of the two condenser sites.



### 2.3.2 Laboratory-accelerated exposure testing and monitoring

Additional coupons, comprised of substrates and coatings identical to the field-exposed coupons, were prepared and exposed in a laboratory's accelerated-exposure cabinets. Testing was conducted according to ASTM standard D 5894 (ASTM 2010). According to this standard, specimens were exposed to alternating periods of 1 week in a fluorescent UV/condensation chamber followed by 1 week in a cyclic salt fog/dry chamber, throughout the 52-week period for this portion of the project. The exposure chambers used were manufactured by the Q-Lab Corp, Cleveland, Ohio; the chambers included a Q-Fog Model Q-FOG-SSP-600 and a UV chamber Model QUV-SE. Observations of blistering and rusting were recorded according to industry rating standards (ASTM 2009, 2012), as shown by Table 4 in Chapter 3.

An anomaly occurred in the preparation of the exposure coupons. The initial set of coupons with phenolic coating had a coating thickness typically in the area of 3.5–4.0 mils. This thickness was significantly in excess of the 2.5–3.0 mils recommended for condenser coatings in UFGS 23 81 00.00

20, “Unitary Air Conditioning Equipment” (DoD 2009, par. 2.6.1, “Phenolic Coatings”). As a result, a new set of coupons was prepared with the proper prescribed coating thickness. This anomaly resulted in a 12-week delay in the exposure of the proper phenolic-coated coupons (i.e., data was recorded at 8 weeks and not again until 20 weeks). The appendix to this report gives additional notes about the process, and the anomaly did not affect the conclusions given in Chapter 5.

## 3 Discussion

### 3.1 Metrics

The results of the project were assessed against a performance goal of minimal degradation of the coating and minimal corrosion of the condenser fins over the performance evaluation period. The exposure coupons and condenser units were visually assessed using criteria from ASTM D714, “Standard Test Method for Evaluating Degree of Blistering of Paints” (ASTM 2009) and ASTM D610 “Standard Test Method for Evaluating Degree of Rusting on Painted Steel Surfaces” (ASTM 2012). Although ASTM D610 is intended to be used for steel surfaces, the criteria therein were used for the evaluation of visual corrosion on the copper and aluminum coupons.

### 3.2 Results

#### 3.2.1 On-site data

The two racks of field-exposed coupons were photographed and evaluated annually for a 24-month period in accordance with ASTM D714 and with ASTM D610 (ASTM 2009, 2012). Alpha-numeric results of the evaluations are shown in Table 1 (with explanation of rating standards in Table 2 and Table 3). Photographic evidence is shown in Figure 6–Figure 10.

The following observations were made from the evaluation data:

- The phenolic coating and Coating B failed to protect the steel substrates at both of the on-site locations.
- Rust and blistering were most prominent on the edges of each coupon.
- Blistering on the aluminum and copper coupons was substantially less.
- The phenolic coating adhered better to the aluminum and copper coupons than did Coating B and Coating A.
- The uncoated coupons showed severe signs of corrosion on 100% of the material within 12 months of exposure.
- The bare aluminum coupon on Rack 2 began to crack and separate into layers from exfoliation corrosion, causing the coupon to break away from the nylon fastener (Figure 7).
- The uncoated aluminum surface of the condenser exterior showed significant corrosion after 24 months of exposure (Figure 8).

- The condenser fins coated with Coating B and Coating E showed no visible signs of corrosion or coating failures after 24 months of exposure (Figure 9 and Figure 10).

Additional photographs of on-site coupons and condenser fins are shown in the appendix of this report.

**Table 1. On-site coupons evaluations according to two industry standards, after 12 months and 24 months of exposure.<sup>1</sup>**

Coupon*	ASTM D610		ASTM D714	
	(12-mo.)	(24-mo.)	(12-mo.)	(24-mo.)
AE-1	10	10	8F	6F
CE-1	10	10	8F	8F
SE-1	10	10	2D	2D
AB-1	10	10	10	10
CB-1	10	10	10	10
SB-1	4S	2G	4M	2M
AP-1	10	10	10	10
CP-1	10	10	10	8F
SP-1	8S	5S	10	10
AR-1	0	0	N/A	N/A
CR-1	0	0	N/A	N/A
SR-1	0	0	N/A	N/A
AE-2	10	10	8F	8F
CE-2	10	10	8F	8F
SE-2	10	10	2D	2D
AB-2	10	10	10	2D
CB-2	10	10	10	10
SB-2	4S	3G	4M	4D
AP-2	10	10	10	10
CP-2	10	10	10	10
SP-2	6S	4S	10	8F
AR-2	0	0	n/a	n/a
CR-2	0	0	n/a	n/a
SR-2	0	0	n/a	n/a
Key to coupon codes: 1 <sup>st</sup> Character = Substrate (A=Aluminum, C=Copper, S=Steel) 2 <sup>nd</sup> Character = Coating (B=Blygod, E= ElectroFin; P=phenolic; R=uncoated) 3 <sup>rd</sup> Character = Rack No. (1,2)				

<sup>1</sup> Note: key to coupon codes is given at end of this table, and key ratings standards are given in Table 2 and Table 3 that follow.

**Table 2. Rust ratings according to ASTM D610-07, “Standard Test Method for Evaluating Degree of Rusting on Painted Steel Surfaces.”**

Scale Rating	Percentage of Surface Found Rusted	Expression of Rust Found (S=spots; G=general; P=pinpoint)
10	≤0.01	(none)
9	>0.01 to 0.03	9S, 9G, 9P
8	>0.03 to 0.1	8S, 8G, 8P
7	>0.1 to 0.3	7S, 7G, 7P
6	>0.3 to 1.0	6S, 6G, 6P
5	>1.0 to 3.0	5S, 5G, 5P
4	>3.0 to 10.0	4S, 4G, 4P,
3	>10.0 to 16.0	3S, 3G, 3P
2	>16.0 to 33.0	2S, 2G, 2P
1	>33.0 to 50.0	1S, 1G, 1P
0	>50.0	(none)

**Table 3. Blister ratings according to ASTM D714-02, “Standard Test Method for Evaluating Degree of Blistering of Paints.”**

Scale	Description of Reference Standards
Size	Reference standards have been selected for four steps as to size on a numerical scale from 10 to 0, in which No. 10 represents no blistering. Blistering standard No. 8 represents the smallest size blister easily seen by the unaided eye. Blistering standards Nos. 6, 4, and 2 represent progressively larger sizes.
Frequency	Reference standards have been selected for four steps in frequency at each step in size, designated as follows:  D = dense MD = medium dense M = medium F = few



Figure 6. Rack-1 coupons after 24 months of on-site field exposure.

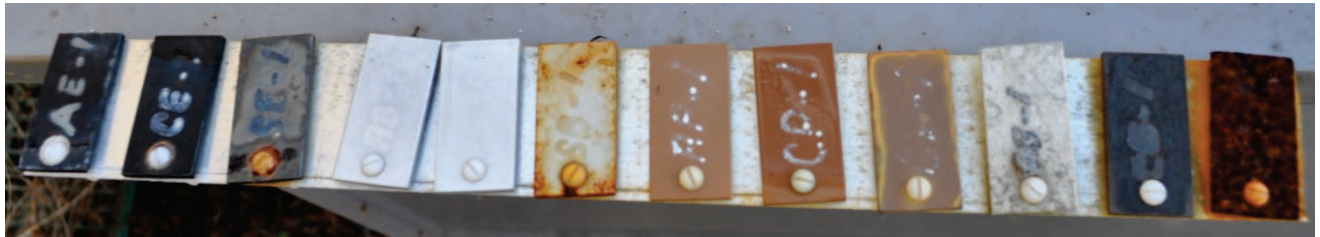


Figure 7. Rack-2 coupons after 24 months of on-site field exposure.

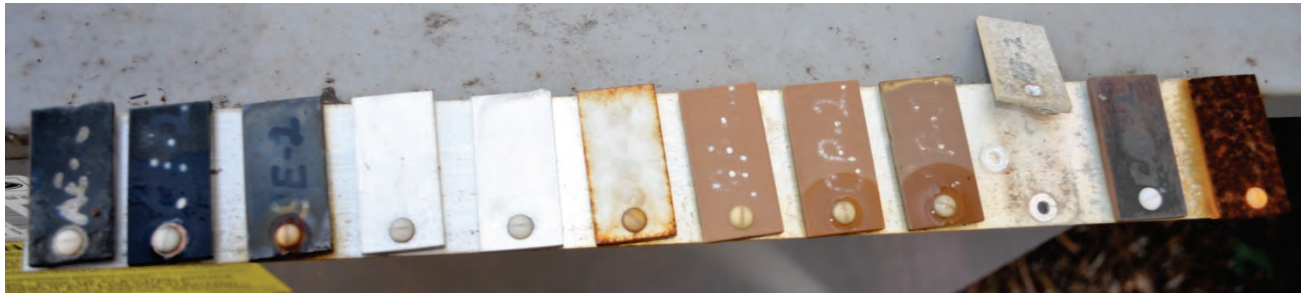


Figure 8. Exfoliation of a bare aluminum coupon after 24 months of on-site field exposure.



Figure 9. HVAC condenser fins with Coating E after on-site exposure for 24 months.

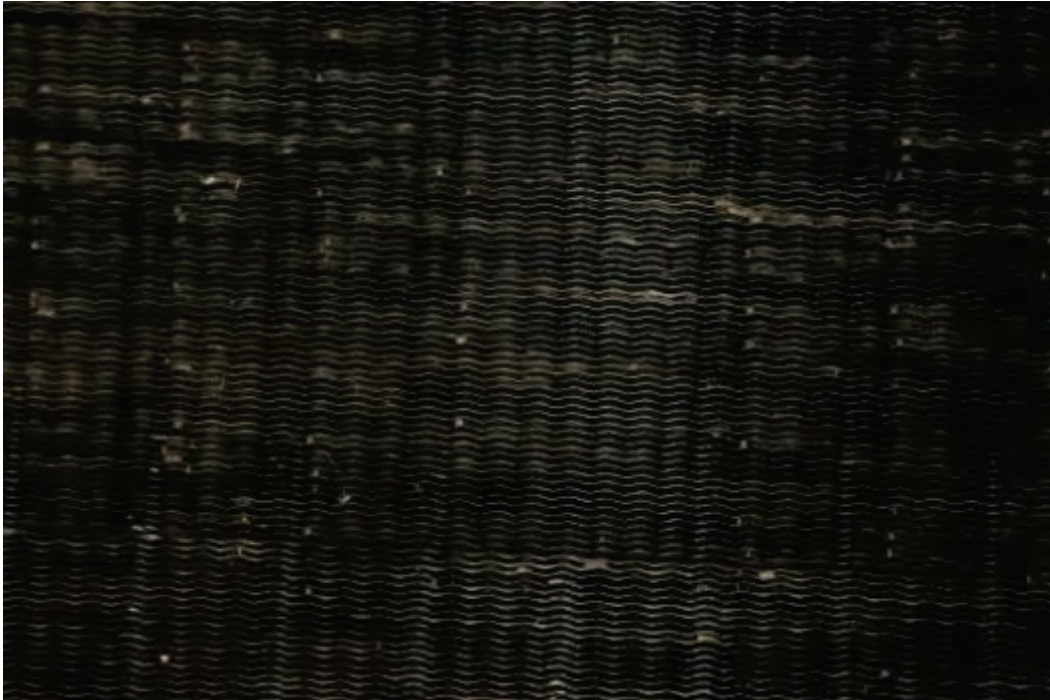
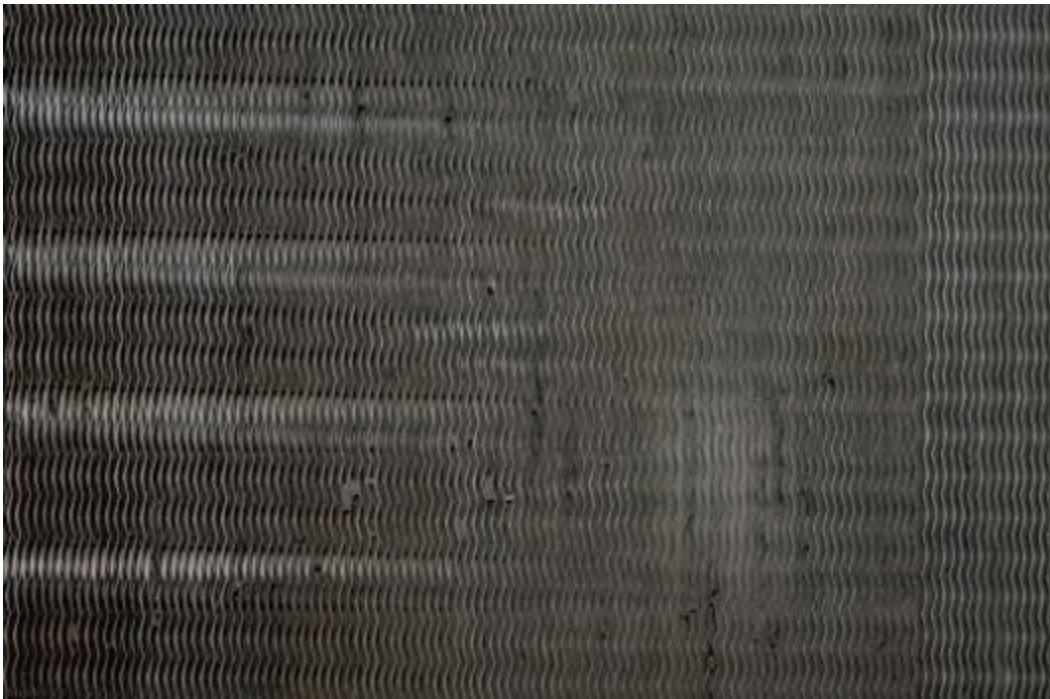


Figure 10. HVAC condenser fins with Coating B after on-site exposure for 24 months.



### 3.2.2 Laboratory data

As previously stated, the accelerated-weathering coupon test standard required that all coupons be moved weekly between different exposure



chambers (ASTM 2009). For the accelerated weathering tests, two from each set of four coated coupons were scored (scratched) to bare metal so the ability of the coating to resist undercutting could be evaluated. The accelerated-weathering coupon test for this project occurred over 52 weeks. Each time the coupons were transferred, a determination was made regarding the need to document changes in the condition of the coatings. Documentation recorded at 8, 12, 20, 32, 42, and 52 weeks for the initial set of coupons is shown in Table 4. The replacement coupons coated with the proper thickness of phenolic (thinner coating) were documented at 8, 20, 30, and 40 weeks (see Table 4). Photographic documentation of the coupons after 20 weeks of exposure is shown in Figure 11. During the course of testing, coupons were removed from testing once the corrosion became severe (with significant metal loss). These coupons included the aluminum and steel panels with Coatings E and B, and the bare aluminum and epoxy panels. The remaining coupons at the conclusion of testing are shown in Figure 12.

Review of accelerated-weathering data in Table 4 shows the copper substrate to be the superior substrate. Uncoated, it quickly developed a green copper oxide patina that progressively grew thicker, eventually flaking off the surface at 32-weeks. Coupons that were scored developed a minor amount of oxides in the score, but the oxide development seemed to seal the score from further corrosion. Clearly, copper is the superior substrate, but the exposure did not indicate any difference in the performance of the various coatings applied to it because all coatings performed similarly well on the copper substrate.

Aluminum is a common fin material used in AC condensers that are not exposed to highly corrosive atmospheres. Because aluminum costs considerably less than copper, it was hoped that a high-quality coating (such as the demonstrated technologies) would protect aluminum from corrosion, even in more corrosive atmospheres. Clearly, each of the coatings did provide a level of protection. Coating B provided the lowest level of protection, with ratings of “0” for both corrosion and blistering in as little as 12 weeks. The scored panels coated with the Coating E epoxy suffered the same level of failure in the same 12-week length of time, but the unscored coupons had a slightly better rating at the 20-week rating period. The phenolic coatings provided the highest level of protection on aluminum. On unscored coupons, the excessively thick phenolic coatings provided a high level of protection at the end of the test period, and the standard-thickness

phenolic coating was still providing a good level of protection after 40 weeks of exposure. On coupons where the phenolic was scored, the substrate began to deteriorate at the score early in the test; this deterioration eventually resulted in exfoliation under the coating. As the aluminum deterioration progressed under the coating, it was often difficult to determine if there was actual blistering associated with the coating, and this difficulty perhaps resulted in some anomalies in the recorded blistering values. Whether the deterioration was recorded as blistering or general deterioration, the anomalies did not affect the overall assessment of each coating.

Steel is never used for condenser fins, but the material is used elsewhere in the condenser framework and housing. On this substrate, the excessively thick phenolic coating provided the highest level of protection, with the standard thickness still providing performance similar to or slightly superior to that of Coating E. The steel coupons with Coating B had total substrate failure in as little as 12 weeks, with exfoliation of the scored panels in 32 weeks.

Figure 11. Accelerated-weathering coupons after 20 weeks of exposure.

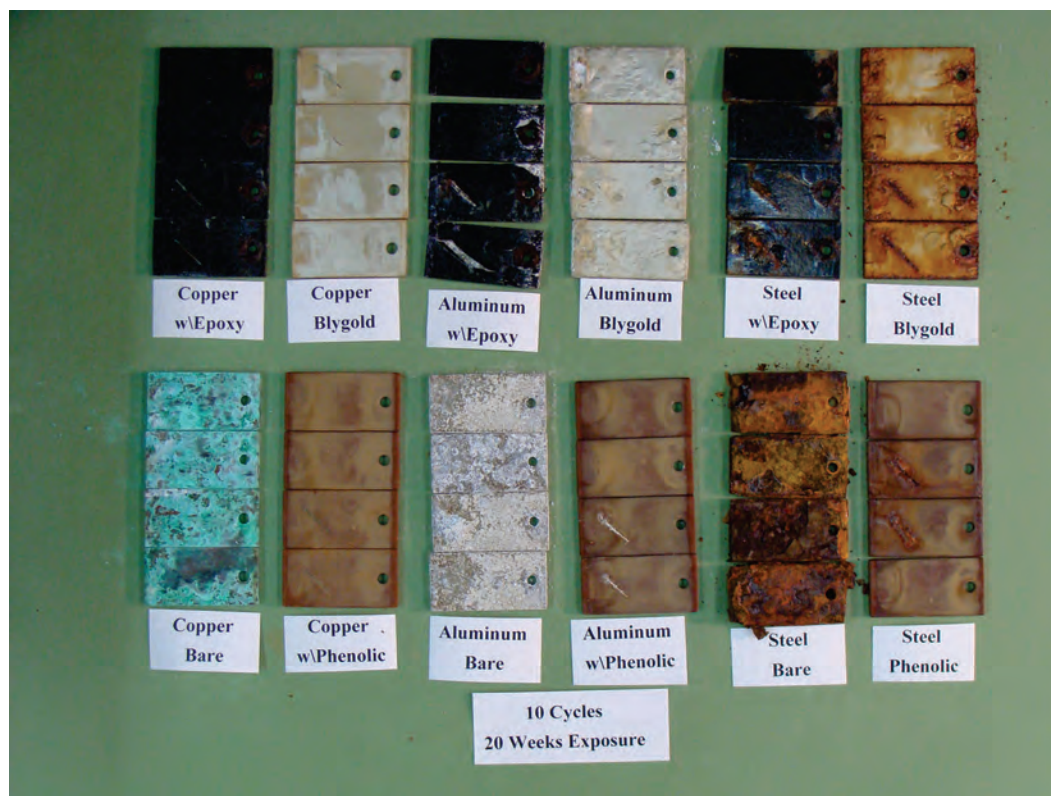


Figure 12. Accelerated-weathering coupons remaining at conclusion of test after 52 weeks of exposure.

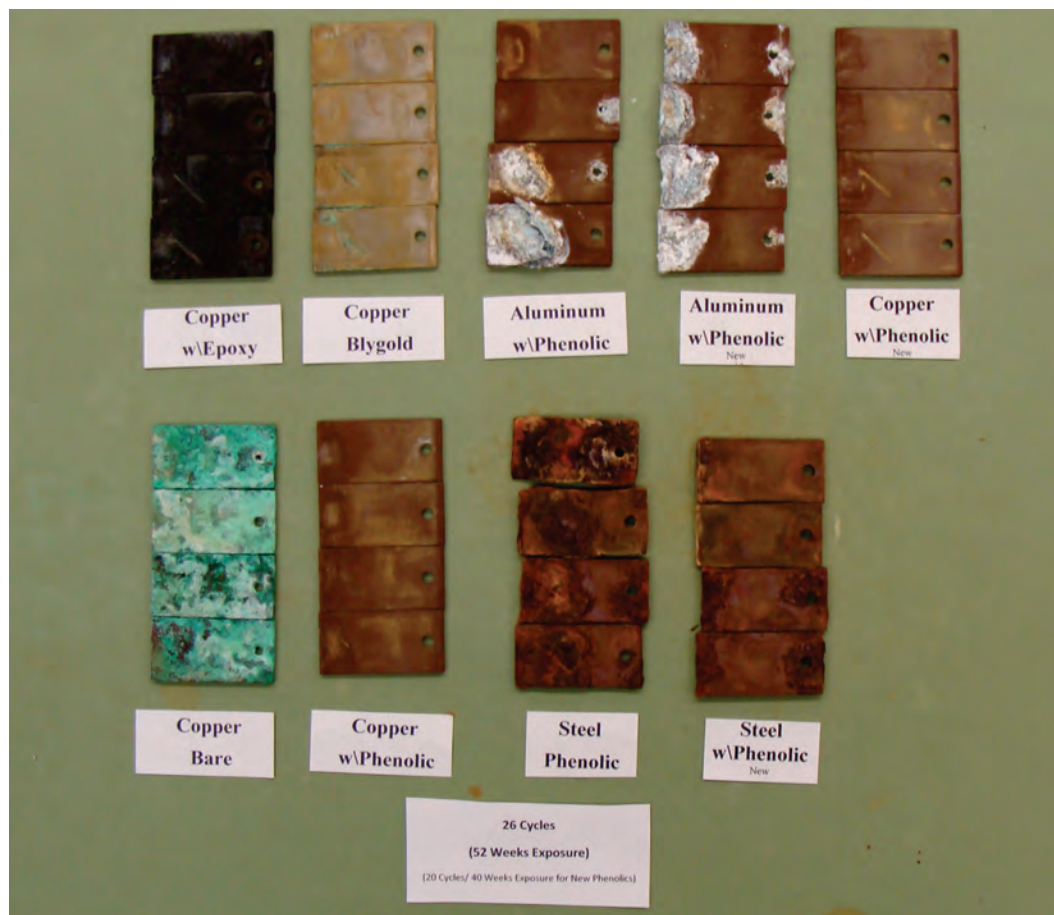


Table 4. Alpha-numeric evaluation indicators of accelerated-weathering coupons at end of various observation periods according to ASTM D610 and D714.<sup>2</sup>

Coupon I.D.†	Scored?	8 Weeks		12 Weeks		20 Weeks		32 Weeks		42 Weeks		52 Weeks	
		D610	D714	D610	D714	D610	D714	D610	D714	D610	D714	D610	D714
CE-19		10	10	10	10	10	10	10	10	10	10	10	10
CE-20		10	10	10	10	10	10	10	10	10	10	10	10
CE-21	Yes	10	10	10	10	10	10	10	10	10	10	10	10
CE-22	Yes	10	10	10	10	10	10	10	10	10	10	10	10
CR-19		Green oxide		Green oxide		Green oxide		Green oxide		Green oxide		Green oxide	
CR-20		Green oxide		Green oxide		Green oxide		Green oxide		Green oxide		Green oxide	
CR-21		Green oxide		Green oxide		Green oxide		Green oxide		Green oxide		Green oxide	
CR-22		Green oxide		Green oxide		Green oxide		Green oxide		Green oxide		Green oxide	
CB-19	Yes	10	10	10	10	10	10	10	10	10	10	10	10
CB-20	Yes	10	10	10	10	10	10	10	10	10	10	10	10
CB-21		10	10	10	10	10	10	10	10	10	10	10	10
CB-22		10	10	10	10	10	10	10	10	10	10	10	10
CP-19		10	10	10	10	10	10	10	10	10	10	10	10
CP-20		10	10	10	10	10	10	10	10	10	10	10	10
CP-21	Yes	10	10	10	10	10	10	10	10	10	10	10	10
CP-22	Yes	10	10	10	10	10	10	10	10	10	10	10	10
AE-19		10	10	10	1D	4S	4D*						
AE-20		10	10	7S	4D	0G	2D*						
AE-21	Yes	9S	4F	0G	0D	Heavy exfoliation*							
AE-22	Yes	10	1F	0G	0D	Heavy exfoliation*							
AR-19		White corrosion		White corrosion		Heavy exfoliation*							
AR-20		White corrosion		White corrosion		Heavy exfoliation*							

<sup>2</sup> Note: key to coupon codes is given at end of this table, and keys to ratings standards were given in Table 2 and Table 3.

Coupon I.D.‡	Scored?	8 Weeks		12 Weeks		20 Weeks		32 Weeks		42 Weeks		52 Weeks	
		D610	D714	D610	D714	D610	D714	D610	D714	D610	D714	D610	D714
AR-21		White corrosion		White corrosion		Heavy exfoliation*							
AR-22		White corrosion		White corrosion		Heavy exfoliation*							
AB-19		10	1F	0G	0D	Heavy exfoliation*							
AB-20		10	1F	0G	0D	Heavy exfoliation*							
AB-21	Yes	7S	2F	0G	0D	Heavy exfoliation*							
AB-22	Yes	7S	2F	0G	0D	Heavy exfoliation*							
AP-19		10	10	10	10	10	10	10	10	10	10	10	10
AP-20		10	10	10	10	10	10	10	10	10	10	10	10
AP-21	Yes	10	10	8S	10	8S	10	3S	10	0G	0D	0G	0D
AP-22	Yes	10	10	8S	10	8S	10	3S	10	0G	0D	0G	0D
SE-19		10	10	10	7F	10	5MD	10	3MD	0G	0D*		
SE-20		10	10	10	10	10	10	3G	10	0G	0D*		
SE-21	Yes	10	10	3S	10	6S	7F	0G	0D	0G	0D*		
SE-22	Yes	10	4F	2S	4F	7S	7F	0G	0D	0G	0D*		
SR-19		100% rusted		Heavy exfoliation*									
SR-20		100% rusted		Heavy exfoliation*									
SR-21		100% rusted		Heavy exfoliation*									
SR-22		100% rusted		Heavy exfoliation*									
SB-19		8S	2F	0G	0D	0G	0D	0G	0D	Heavy exfoliation*			
SB-20		8S	2F	0G	0D	0G	0D	0G	0D	Heavy exfoliation*			
SB-21	Yes	8S	2F	0G	0D	0G	0D	Some exfoliation		Heavy exfoliation*			
SB-22	Yes	8S	2F	0G	0D	0G	0D	Some exfoliation		Heavy exfoliation*			
SP-19		10	10	10	10	10	10	10	10	8S	10	Heavy exfoliation*	
SP-20		10	10	10	10	10	10	10	10	10	10	Heavy exfoliation*	
SP-21	Yes	7S	10	7S	4F	7S	3F	7S	3F	6S	3F	Heavy exfoliation*	
SP-22	Yes	7S	10	7S	6F	7S	3F	7S	3F	6S	3F	Heavy exfoliation*	

Coupon I.D.‡	Scored?	8 Weeks		12 Weeks		20 Weeks		32 Weeks		42 Weeks		52 Weeks					
		D610	D714	D610	D714	D610	D714	D610	D714	D610	D714	D610	D714				
Phenolic at standard thickness																	
Coupon ID	Score	8 Weeks		(12-week delay to recoat coupons at standard thickness)		20 Weeks		30 Weeks		40 Weeks		NOTE: due to 12-week delay, 40 weeks of exposure equaled end of project for this coupon.					
CP-4		10	10	n/a	na/	10	10	10	10	10	10						
CP-5		10	10	n/a	na/	10	10	10	10	10	10						
CP-7	Yes	10	10	n/a	na/	10	10	10	10	10	10						
CP-9	Yes	10	10	n/a	na/	10	10	10	10	10	10						
AP-3		10	10	n/a	na/	8S	10	8S	10	8S	10						
AP-4		10	10	n/a	na/	8S	10	6S	10	6S	10						
AP-6	Yes	10	10	n/a	na/	8S	10	6S	10	1G	10						
AP-12	Yes	10	10	n/a	na/	8S	10	6S	10	1G	10						
SP-5		10	10	n/a	na/	8S	10	4S	6M	0G	0D						
SP-7		10	10	n/a	na/	8S	10	4S	6MD	0G	0D						
SP-9	Yes	10	10	n/a	na/	5S	10	2S	6M	0G	0D						
SP-12	Yes	10	10	n/a	na/	3S	10	2S	6MD	0G	0D						
* Green corrosion products heavy enough to fall off the coupon																	
* Coupon removed from test																	
‡ Coupon code identifications (refer to Table 2 and Table 3																	
Table 2 for keys to rust and blister ratings):																	
1 <sup>st</sup> Character = Base Matl.				A=Aluminum C=Copper S=Steel				2 <sup>nd</sup> Charac. = Coating				B = Coating B E = Coating E P = phenolic R = bare				3 <sup>rd</sup> Charac. = Rack No. (1,2)	

### 3.3 Lessons learned

Prior to the initiation of this project, it was known that copper condenser coils provided the highest level of corrosion resistance for condensers. It was suggested that using less-costly aluminum with a good coating system could provide a cost-effective alternative to copper. However, based on coupon performance in the accelerated testing portion of this project, this suggestion does not appear to be the case. At the end of 1 year (52 weeks) of accelerated testing, every coating system applied to the copper substrate had perfect protection. On field-exposed coupons exposed for 2 years, the copper substrate coated with phenolic and Coating B were beginning to exhibit minor blistering, and the performance of coating system was generally good. All of the coating systems applied to aluminum had significant coating failures with associated substrate deterioration. These findings clearly support copper as the material of choice for condensers in aggressive corrosion environments.

In accelerated testing, the bare copper developed a green copper oxide that increased in thickness and eventually became so thick that it began falling off the coupons at the 32-week observation point. Obviously, this process resulted in a loss of substrate thickness and probably would result in a loss of heat-exchanger fin material and overall efficiency for a condenser. All of the coatings tested prevented this patina from occurring. This set of findings clearly supports the use of a coating system on copper, but it does not identify any of the coatings as being superior.

Within the 2-year evaluation period, there was a clear difference in the corrosion-protection capability of the two coatings on various substrates when comparing coupons subjected to accelerated-exposure cabinet testing. The field evaluation of the two coating systems on the condenser coils and fins, however, did not reveal any notable differences between the two coatings after 2 years of exposure (see Figure 9 and Figure 10 plus photos of condenser fins in this report's appendix). As a result, it is suggested that any future work in this area should be conducted first on coupons in accelerated-exposure testing cabinets. Data developed in the accelerated cabinets may then be used as the basis for field evaluations on actual condenser units, but such field exposures must be allowed to continue for extended times in order to generate desired field data. The 2-year field exposure period used in this work was insufficient to show any performance differences between the two coating systems.

## 4 Economic Analysis

This analysis was performed in accordance with Office of Management and Budget (OMB) Circular No. A-94, *Guidelines and Discount Rates for Benefit-Cost Analysis of Federal Programs* (OMB 1994).

### 4.1 Costs and assumptions

**Demonstration Project Costs.** The total cost for the demonstration project was \$670,000, with details shown in Table 5.

Table 5. Detail for \$670,000 project demonstration costs/funding.

Item	OSD	Army
Labor	\$105,000	\$105,000
Contracts	\$180,000	\$150,000
Travel	\$30,000	\$30,000
Report	\$35,000	\$35,000
<b>Total</b>	<b>\$350,000</b>	<b>\$320,000</b>

Two alternatives have been analyzed and details are given below. The first alternative is a traditional phenolic coating system on the condenser units. The second alternative is a corrosion-resistant coil coating such as Blygold (Coating B) or ElectroFin E-coat (Coating E). The test units selected were the new 30-ton and 40-ton condensers installed at Schofield Barracks.

**Alternative 1 (Baseline Case).** According to information provided by Schofield Barracks, the costs associated with the installation of the two condenser units are shown in Table 6.

Table 6. Baseline Case, costs and expected service life.

Item	Total
Labor (83.5 hr @\$80/hr)	\$6,680
Condensers (2)	\$43,735
Materials	\$27,189
<b>TOTAL</b>	<b>\$77,604</b>
<b>Service Life</b>	<b>7 years</b>



A total of 70 condensers are included in this analysis. The installation replaces 10 condensers per year (5 x \$77,604—cost for two) for a cost of \$388,020 annually. Note that a condenser's expected service life in Alternative 1 is 7 years.

**Alternative 2 (Demonstrated Technology).** In calculating the costs of the alternative coating systems demonstrated in this project (Table 7), the costs for Demonstrated Technology are the same as for Alternative 1 except for the added cost of the alternative coating systems. The additional cost is \$315 for Coating E and \$2,400 for Coating B. Note that a condenser's service life in Alternative 2 is extended to 30 years.

Table 7. Demonstrated Technology, costs and expected service life.

Item	Total
Labor (83.5 hr @\$80/hr)	\$6,680
Condensers – 2	\$43,735
Cost of coatings – 2	\$2,715
Materials to install	\$26,874
<b>TOTAL</b>	<b>\$80,004</b>
<b>Service Life</b>	<b>30 years</b>

In the first year and using project funding, two condensers were installed with a new coating. The other eight requiring replacement were installed with the standard phenolic system. In years two through seven, 10 additional condensers are to be installed with the new coating system each year. In year eight, the last six of the existing condensers with phenolic system are upgraded. Over a 30-year period, a condenser unit coated with either Coating B or Coating E is expected not to require replacement.

## 4.2 Projected return on investment (ROI)

A 7% discount rate is used for the ROI calculation, consistent with CPC program guidance (OMB Circular A-94). As previously stated, over a 30-year period, a condenser unit coated with either Coating B or Coating E is expected not to require replacement. However, a condenser unit installed with a standard coating, as described in Alternative 1, is expected to be replaced 4 times over a 30-year period.

The projected ROI is 3.89 over 30 years. The calculation is based on a required CPC project investment of \$670,000. A summary of the ROI analysis is shown in Table 8.

Table 8. ROI analysis.

Investment Required					670,000		
Return on Investment Ratio					3.89	Percent	389%
Net Present Value of Costs and Benefits/Savings					2,211,731	4,814,824	2,603,093
A Future Year	B Baseline Costs	C Baseline Benefits / Savings	D New System Costs	E New System Benefits / Savings	F Present Value of Costs	G Present Value of Savings	H Total Present Value
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30	388,020		310,416		290,115	362,643	72,529
	388,020		400,020		349,377	338,897	-10,481
	388,020		400,020		326,536	316,741	-9,796
	388,020		400,020		305,175	296,020	-9,155
	388,020		400,020		285,214	276,658	-8,556
	388,020		400,020		266,533	258,538	-7,996
	388,020		400,020		249,092	241,620	-7,472
	388,020		240,012		139,687	225,828	86,141
	388,020					211,044	211,044
	388,020					197,231	197,231
	388,020					184,348	184,348
	388,020					172,281	172,281
	388,020					161,028	161,028
	388,020					150,474	150,474
	388,020					140,618	140,618
	388,020					131,422	131,422
	388,020					122,847	122,847
	388,020					114,815	114,815
	388,020					107,288	107,288
	388,020					100,264	100,264
	388,020					93,707	93,707
	388,020					87,576	87,576
	388,020					81,833	81,833
	388,020					76,479	76,479
	388,020					71,473	71,473
	388,020					66,817	66,817
	388,020					62,432	62,432
	388,020					58,358	58,358
	388,020					54,556	54,556
	388,020					50,986	50,986

## **5 Conclusions and Recommendations**

### **5.1 Conclusions**

In corrosive environments, copper far outperforms aluminum and steel. The coating systems tested did not serve to protect these alternate metals sufficiently to make those coated alternative metals competitive with bare copper. The work also demonstrated that corrosion of copper will be delayed by the application of a coating system. While the tests of the coated copper coupons were not continued until failure, it was seen that the application of a coating to the copper parts doubled the time until the onset of significant corrosion and metal loss. A doubling of the performance life would easily justify the added cost of a coating system.

Highly corrosive environments like Hawaii create a problem for DoD facilities and vehicles. Coatings like phenolic, Blygold, and E-coat provide a method to mitigate corrosion and extend the service life of DoD assets like HVAC condensers. However, the application methods for Blygold and E-Coat show advantages over the dip or flow-coat methods used to apply phenolic coatings. This study validates that any coating provides a level of corrosion protection that is cost beneficial to the service life of HVAC condensers. Additionally, the protected fins of the condenser allow the unit to maintain efficient heat transfer and to reduce the strain on mechanical components. The sample coupons with phenolic and E-coat (Coating E) in this experiment exhibited less blistering and coating failures than the Blygold coating. Over a 30-year period, it was calculated that the ROI ratio of coating HVAC condensers in Hawaii was 3.89.

### **5.2 Recommendations**

Based on the accelerated weathering data, it is recommended that condensers in corrosive environments be manufactured with copper coils and fins that are coated. The data does not identify any coating as being superior on copper, so it is suggested the manufacturer's standard coating system be accepted.

It is recommended that the HVAC condensers and on-site coupons under this test be periodically revisited over the next 30 years, and the corrosion-protection performance of the two coating technologies be re-evaluated.

Additional inspections will provide additional data to more accurately predict the life cycle of each coating used in this application.

### **5.2.1 Applicability**

This work's findings are applicable for finned-type heat exchangers in extremely corrosive environments. Applicable heat exchangers include condensers used in AC applications as well as other heat recovery and dissipation applications that employ lightweight heat-transfer surfaces. Guidance documents typically contain a note to the designer suggesting that the project location be researched to determine the environmental effects on finned tube coils and providing options. Where the environmental effects are severe, the option of using factory-coated copper fins should be encouraged.

### **5.2.2 Implementation**

Guidance documents should be revised to provide guidance for selecting coated copper condenser coils and fins in highly corrosive environments. The following documents should be revised:

- In Federal Specification OO-A-374C (DoD 2011) edit the third sentence in Paragraph 6.1.10, to read as follows:

“In coastal areas, condenser should be made of copper tube and copper fins and coated with the manufacturer’s standard coating. Copper items should be insulated for protection against galvanic corrosion from ferrous metal parts.”

- In UFGS 23 64 00.00 10 (DoD 2008), add the following to the end of the Note in Paragraph 2.6:

“Copper tubes with copper fins coated with the manufacturer’s standard coating should be specified for the most severely corrosive locations.”

- In UFGS 23 81 00.00 20 (DoD 2009):
  - Paragraph 2.2.8 a—add the following sentence to the NOTE: “Copper tubes with copper fins coated with the manufacturer’s standard coating should be specified for the most severely corrosive locations.”

- Paragraph 2.3.2—add the following sentence to the NOTE: “Copper tubes with copper fins coated with the manufacturer’s standard coating should be specified for the most severely corrosive locations.”
- Paragraph 2.4.7—add the following sentence to the NOTE: “Copper tubes with copper fins coated with the manufacturer’s standard coating should be specified for the most severely corrosive locations.”
- Paragraph 2.6—add the following sentence to the second NOTE: “Copper tubes with copper fins coated with the manufacturer’s standard coating should be specified for the most severely corrosive locations.”

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- Herzber, Eric F., Amelia Kelly, and Norman T. O'Meara. 2010. *The Annual Cost of Corrosion for the Department of Defense Facilities and Infrastructure: 2007–2008 Update*. Report DL907T2. Tysons, VA: LMI.
- Meister, John J. 2000. *Polymer Modification Principles, Techniques and Applications*. Boca Raton, FL: CRC Press
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## Appendix: Annual Assessment, Photo Documentation

Photos of test coupons were made before the project period began (time zero). Then, on-site visual assessments of coupons and fins coated with each of the two subject coating technologies occurred at approximately 12 and 24 months after installation. In addition, on-site coupons with phenolic coating and on-site bare coupons were also assessed at the same time periods.

Photographic documentation of the state of all coupons was recorded during each visual inspection. These photos are arranged by coating and elapsed time. Coating E comparisons are shown in Figures A1–A3 (Rack 1), Figures A13–A15 (Rack 2) and Figures A26–28 (condenser fins). Coating B comparisons are shown in Figures A4–A6 (Rack 1), Figures A16–A18 (Rack 2) and Figures A29–A31 (condenser fins). Phenolic coating comparisons are shown in Figures A7–A9 (Rack 1) and Figures A19–A21 (Rack 2). Bare coupon comparisons are shown in Figures A10–A12 (Rack 1) and Figures A22–A25 (Rack 2).

For convenience, the key to codes painted on the coupons and shown in the figures that follow is provided again in this appendix (Table A1).

Table A1. Explanation of coupon codes and numbers.

Code Location/Reference	Abbreviation	Meaning
First alpha character = BASE MATERIAL	A	Aluminum
	C	Copper
	S	Steel
Second alpha character = COATING	E	Coating E (ElectroFin)
	B	Coating B (Bygold)
	P	Phenolic coating
	R	Bare (no coating)
Third character (numeric) = RACK NUMBER	1	Rack 1
	2	Rack 2

## Coating E comparisons (Rack 1)

Figure A1. Rack 1, Coating E coupons at time zero.



Figure A2. Rack 1, Coating E coupons at 12 months.



Figure A3. Rack 1, Coating E coupons at 24 months.





## Coating B comparisons (Rack 1)

Figure A4. Rack 1, Coating B coupons at time zero.



Figure A5. Rack 1, Coating B coupons at 12 months.

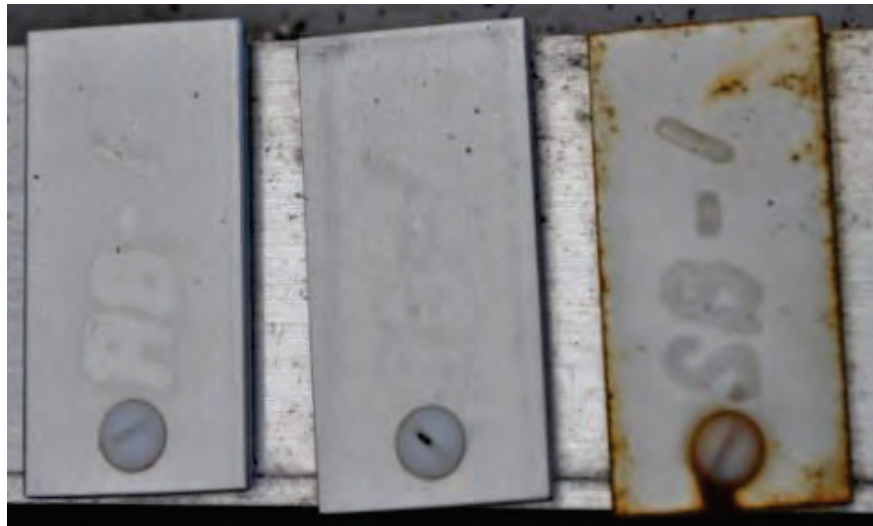


Figure A6. Rack 1, Coating B coupons at 24 months.



## Phenolic-coated coupon comparisons (Rack 1)

Figure A7. Rack 1, phenolic-coated coupons at time zero.



Figure A8. Rack 1, phenolic-coated coupons at 12 months.



Figure A9. Rack 1, phenolic-coated coupons at 24 months.



## Bare coupon comparisons (Rack 1)

Figure A10. Rack 1, bare coupons at time zero.



Figure A11. Rack 1, bare coupons at 12 months.

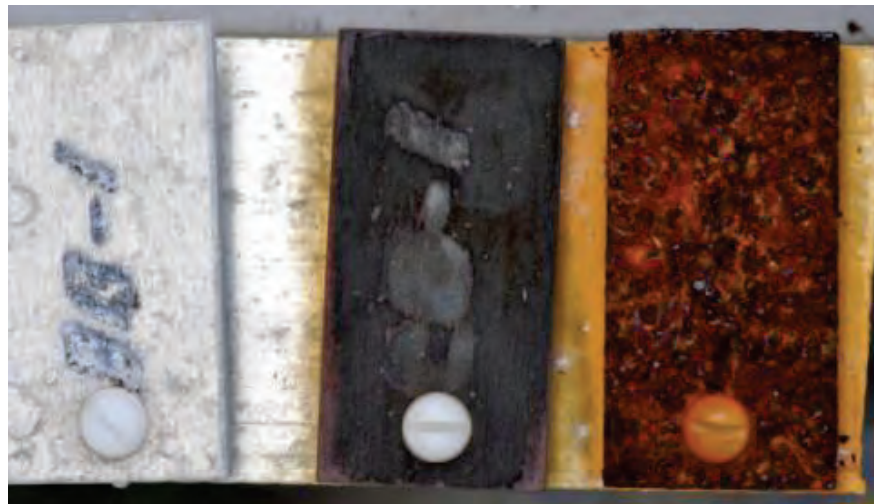


Figure A12. Rack 1, bare coupons at 24 months.





## Coating E coupon comparisons (Rack 2)

Figure A13. Rack 2, Coating E coupons at time zero.



Figure A14. Rack 2, Coating E coupons at 12 months.



Figure A15. Rack 2, Coating E coupons at 24 months.



## Coating B coupon comparisons (Rack 2)

Figure A16. Rack 2, Coating B coupons at time zero.



Figure A17. Rack 2, Coating B coupons at 12 months.



Figure A18. Rack 2, Coating B coupons at 24 months.



## Phenolic-coated coupon comparisons (Rack 2)

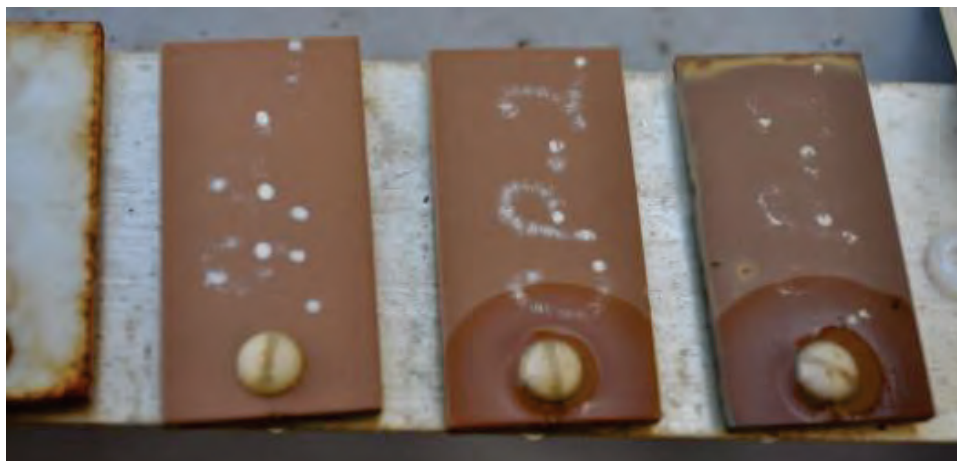
Figure A19. Rack 2, phenolic-coated coupons at time zero.



Figure A20. Rack 2, phenolic-coated coupons at 12 months.



Figure A21. Rack 2, phenolic-coated coupons at 24 months.





## Bare coupon comparisons (Rack 2)

A22. Rack 2, bare coupons at time zero.



A 23. Rack 2, bare coupons at 12 months.



*(see next page)*

Figure A24. Rack 2, bare coupons at 24 months.  
Note that aluminum coupon (at far left) has become detached.



Figure A25. Rack 2, bare coupons at 24 months (photo 2).  
Note that this photo was taken after aluminum coupon was repositioned.





## Coating E: condenser fin comparisons

Figure A26. Coating E condenser fins at time zero.



Figure A27. Coating E condenser fins at 12 months.

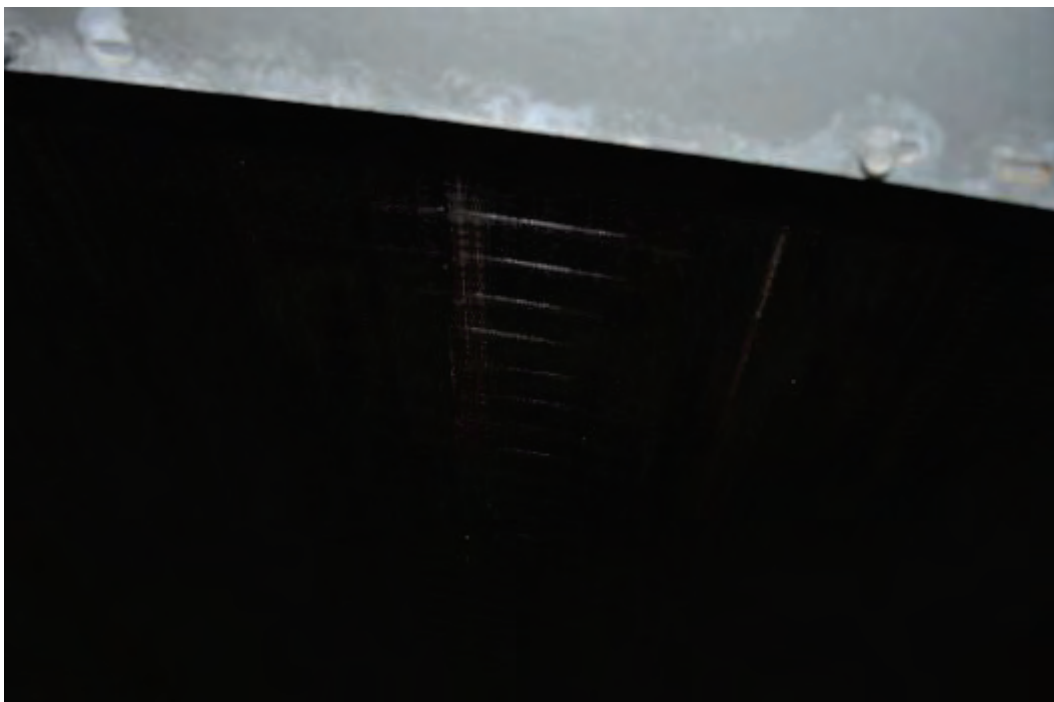
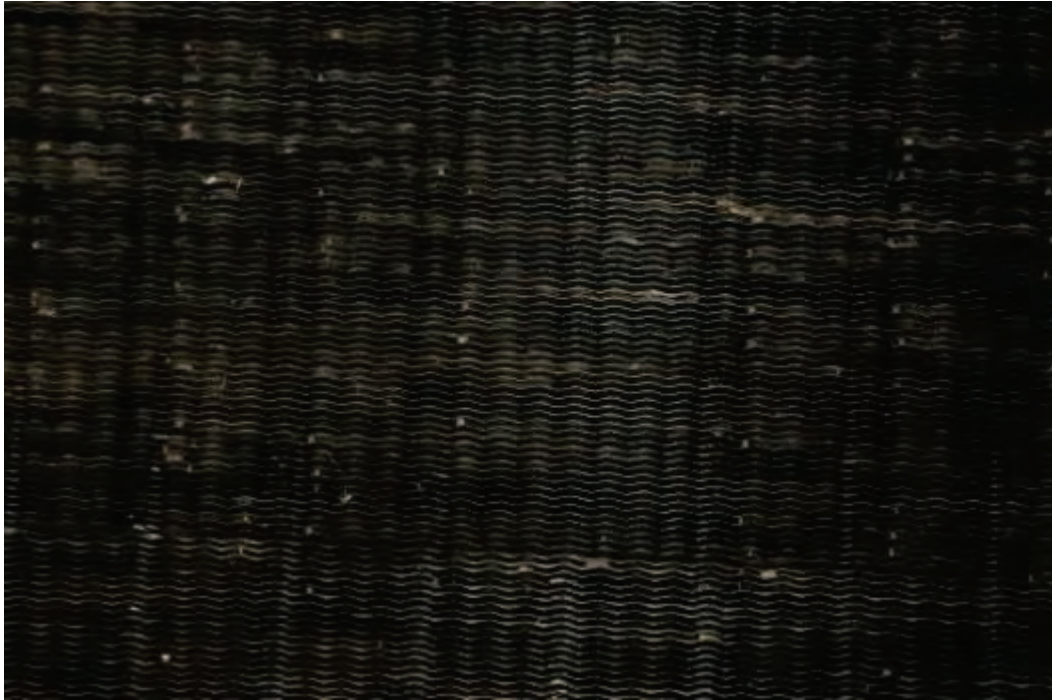


Figure A28. Coating E condenser fins at 24 months.



### Coating B: condenser fin comparisons

Figure A29. Coating B condenser fins at time zero.

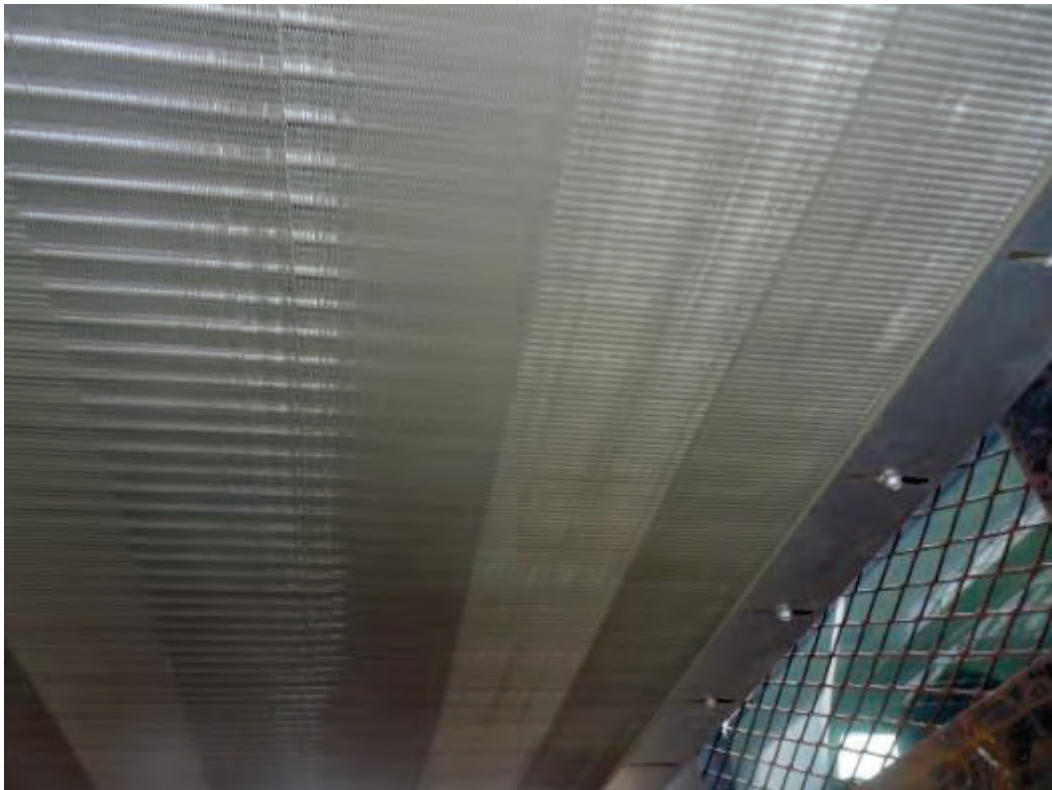
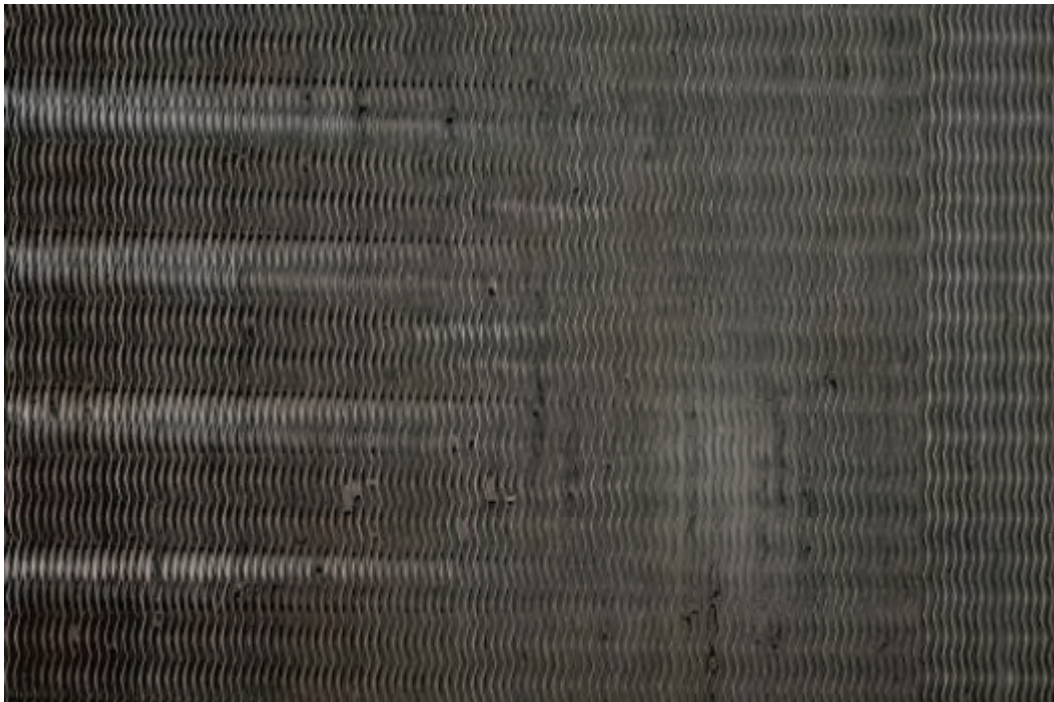




Figure A30. Coating B condenser fins at 12 months.



Figure A31. Coating B condenser fins at 24 months.



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1. REPORT DATE (DD-MM-YYYY) June 2015		2. REPORT TYPE Final		3. DATES COVERED (From - To)	
4. TITLE AND SUBTITLE Demonstration of Corrosion-Resistant Coatings for Air-Conditioning Coils and Fins: Final Report on Project F10-AR01				5a. CONTRACT NUMBER	
				5b. GRANT NUMBER	
				5c. PROGRAM ELEMENT NUMBER Corrosion Prevention and Control	
6. AUTHOR(S) Alfred D. Beitelman and Susan A Drozdz				5d. PROJECT NUMBER F10-AR01	
				5e. TASK NUMBER	
				5f. WORK UNIT NUMBER	
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) US Army Engineer Research and Development Center Construction Engineering Research Laboratory P.O. Box 9005 Champaign, IL 61826-9005				8. PERFORMING ORGANIZATION REPORT NUMBER ERDC/CERL TR-15-12	
9. SPONSORING / MONITORING AGENCY NAME(S) AND ADDRESS(ES) Office of the Secretary of Defense (OUSD(AT&L)) 3090 Defense Pentagon Washington, DC 20301-3090				10. SPONSOR/MONITOR'S ACRONYM(S) OSD	
				11. SPONSOR/MONITOR'S REPORT NUMBER(S)	
12. DISTRIBUTION / AVAILABILITY STATEMENT Approved for public release; distribution is unlimited.					
13. SUPPLEMENTARY NOTES					
14. ABSTRACT  Department of Defense military installations operate air-conditioning equipment in humid and coastal locations that produce high levels of corrosion. Installation experience has shown that severe corrosion has reduced expected equipment life cycles by up to 50%. This project for the Office of the Secretary of Defense Corrosion Prevention and Control Program demonstrated and evaluated the performance characteristics of two corrosion prevention coatings on condenser coils in highly corrosive environments. This work assessed the capabilities and advantages of an aluminum-impregnated polyurethane coating and a flexible epoxy polymer applied to two separate condenser units at Schofield Barracks, Hawaii. The work also included fabricating and exposing test coupons of various substrates, each coated with the two coating systems and a third coating commonly used in less-corrosive environments. These coupons were exposed both at the Schofield Barracks site and in a laboratory-accelerated exposure cabinet. The work recommends that copper rather than aluminum components be specified in air conditioners operating in highly corrosive environments. A copper substrate protected with a newer type of corrosion coating is projected to extend the equipment's expected life cycle from 7-10 years to 25-30 years. The return on investment for the project is 3.89.					
15. SUBJECT TERMS corrosion prevention, coatings, HVAC condenser coils, evaluation, performance testing, Schofield Barracks, HI					
16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF ABSTRACT  UU	18. NUMBER OF PAGES  52	19a. NAME OF RESPONSIBLE PERSON
a. REPORT Unclassified	b. ABSTRACT Unclassified	c. THIS PAGE Unclassified			19b. TELEPHONE NUMBER (include area code)