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14. ABSTRACT Time-resolved force sensing and intensified charge-coupled device (ICCD) imaging techniques were applied to the study of the force generation mechanism for laser ablation of liquids. A Transversely Excited at Atmospheric pressure (TEA) CO_2 laser operated at 10.6 µm, 300 ns pulse width, and 9 J pulse energy was used to ablate liquids contained in various aluminum and glass vessels. Net imparted impulse and coupling coefficient were derived from the force sensor data and relevant results will be presented for various container designs and liquids used. ICCD imaging was used in conjunction with the dynamic force techniques to examine dependencies on absorption depth, irradiance, surface curvature, and container geometry. ICCD imaging was also used to determine whether surface or volume absorption should be preferable for laser propulsion using liquid propellants. Finally, ballistic experiments were conducted in order to verify the dynamic force data and lend additional evidence as to the predominant methods of force generation.						
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Ablation of Liquids for Laser Propulsion with TEA CO₂ Laser (Preprint)

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Scott Anderson *Audio Arrangement*



REDEROILLING

Outline



LISER CISION PROPERTY

- 1. Introduction
- 2. Force
- 3. ICCD Ballistics
- 4. ICCD Imaging
- 5. Results and Discussion
- 6. Conclusions



THE RESEARCH LUB



- I_{sp} up to 50 s

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Motivation

 (1) What is the physical mechanism generating thrust during the laser ablation of a liquid?

- Plasma formation
- Vaporization
- Explosive boiling / Phase explosion
- Cavity collapse / splashing

(2) How do the ablation mechanisms for surface and volumeabsorbing liquids differ?

(3) How does C_m depend on:

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- Container geometry
- Liquid surface curvature
- Absorption depth





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Piezoelectric Force Sensors

Small Force Sensor

- Maximum Force: 9.786 N
- Linearity: < 1%</p>

LSER CSOT

- Sensitivity: 526.6 mV/N
- Discharge time constant: > 1.0 s
- Rise Time: 5 μs
- Resolution: 10⁻⁴ N

PCB-209C01



- Maximum Force: 444.8 N
- Linearity: < 0.4%
- Sensitivity: 11.96 mV/N
- Discharge time constant: > 500 s
- Rise Time: 8 µs
- Resolution: 10⁻³ N



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Coupling Coefficients and Peak Force

Peak Force (N)		Hexane	Ethanol	Water	
Cylinder	Flat	3.4 ± 0.4	5.2 ± 0.6	4.1 ± 0.4	
	Concave	3.1 ± 0.2	6.0 ± 0.1	5.7 ± 0.5	
Cone	Flat	3.6 ± 0.3	5 ± 1	4.1 ± 0.4	
	Concave	3.7 ± 0.3	5.8 ± 0.4	5.3 ± 0.4	

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C _m (dyne/W)		Hexane	Ethanol	Water
Cylinder	Flat	22 ± 3	50 ± 3	43 ± 4
	Concave	22 ± 3	56 ± 4	60 ± 6
Cone	Flat	23 ± 1	49 ± 6	47 ± 5
	Concave	24 ± 3	56 ± 3	56 ± 6

(1.6 x 10⁷ W/cm², 0.4 J, Force Sensor)

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THE RESEARCH LINGS





ICCD Imaging

7 mm 10 L10 шт 100 ns gate ethanol t=100µus

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Information on: Initial plume velocities Cavity growth rates Characteristic physical processes Timeline of processes





ICCD Ballistics Setup

- Gated CCD technique
- Single laser shot per image
- Highly repeatable •

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Composite sequences •

- 382 x 574 pixel images •
- >5 ns gate width (exposure) •

PUBLE RESENTED LURA

5 ns - 83 ms delay •



Coupling Coefficients, dyne/W

(Water, 3 mm² spot size)

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Energy (J)	Irradiance (W/cm ²)	Small Force Sensor	Large Force Sensor	Ballistics
0.4	4 x 10 ⁷	110 ± 14	-	98 ± 9
1.2	1 x 10 ⁸	44 ± 24	53 ± 21	37 ± 5
3.6	5 x 10 ⁸	_	14 ± 3	14 ± 2



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Volume Absorption (Hexane)

• 0.4 J

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- $4 \times 10^7 \text{ W/cm}^2$
- No plasma
- No cavity
- No plume
- Boiling



- 1.1 J
- $1 \ge 10^8 \text{ W/cm}^2$
- Plasma
- Surface cavity
- Vapor plume



HORTE RESEARCH LIBOR



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Liquid Mass Loss

- All data using cylinder, flat surface, 0.4 J, 7 x 10^7 W/cm²
- Increased cavity growth during initial 100 µs
- <u>ICCD imaging:</u> Mass (mg) removed during initial 100 µs
- <u>Scientific Balance:</u> Total mass (mg) removed in entire process

Liquid	Hexane	Ethanol	Isopropanol	Water
0-100 µs	0	2 ± 1	1 ± 1	1 ± 1
Total Process	4 ± 1	51 ± 6	51 ± 2	51 ± 5
Ratio	0 %	4 %	2 %	2 %

Mass Loss (mg)

ListR 15105

THE RESERVILLE

Analysis of Initial Velocities Center plume front height h Δh paired with known Δt $\lim_{t\to 0} \frac{\Delta h}{\Delta t} = v_0$ 14.0 12.0 Hexane, 1.2.3 Ethanol, 0.4.3 Water, 0.4.3 Unitial Velocities Surface absorbing liquids



height (mm)

Surface absorbing liquids transonic – supersonic initial velocities observed

Volume absorbing liquids: subsonic to transonic velocities observed



Specific Impulse and Internal Efficiency

(1.6 x 10⁷ W/cm², 0.4 J)

Specific Impulse (s)	Hexane	Ethanol	Water	
from v ₀ /g	-	42 ± 1	84 ± 3	
1 st 100 µs	-	10 ± 5	20 ± 10	
Total process	2.3 ± 0.6	0.40 ± 0.05	0.34 ± 0.05	

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 $\approx \frac{u_e}{d}$

W

g

Hexane	Ethanol	Water	
_	10.3 ± 0.7	18 ± 2	
_	3 ± 1	4 ± 2	
0.24 ± 0.08	0.10 ± 0.01	0.07 ± 0.01	
	-	$-$ 10.3 \pm 0.7	

PUBLE RESEARCH LUNCH





Conclusions

- A series of experiments with time-resolved force sensors and ICCD imaging were conducted on liquids.
- 2) 2 major physical ablation processes are observed: vaporization and splashing. In some cases plasma formation was also achieved.
- 3) The major source of thrust generation in the laser ablation of liquids is vaporization.
- ICCD imaging shows vaporization occurs from 0 to 100 μs after the laser pulse. Splashing is initiated after about 100 μs.
- 5) Force generation occurs during the vaporization regime.

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- 6) The peak force was observed \sim 40-50 µs after the laser pulse.
- 7) Ballistics experiments corroborate the impulse measurements with force sensors.
- 8) Surface absorbing liquids show higher C_m (~50-150 dyne/W) than volume absorbing liquids (~10-50 dyne/W).

Conclusions (continued)

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- 9) C_m is dependent on surface geometry for surface absorbing liquids. Changing geometry does not affect C_m for volume absorbing liquids.
- 10) C_m is dependent on container geometry for volume absorbing liquids.
 Changing container geometry does not affect C_m for surface absorbing liquids.
- 11) The majority of mass loss occurs due to splashing (>95%) rather than vaporization (<5%).
- 12) Momentum coupling was observed to be about 3 times more sensitive to changes in the surface curvature for surface absorbing liquids than to changes in the container geometry for volume absorbing liquids. This is additional evidence in favor of a dominant vaporization mechanism.









Challenges of Using Piezoelectric Force Sensors

- Measurements in an accelerating frame
 - Distortion*

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- Solution: Sensor at rest
- Rise time limits detection speed
 - Plasma processes $\sim 1 \ \mu s$
 - Liquid vaporization $\sim 100~\mu s$
 - Cavity Collapse ~ 1 ms
 - Liquid splashing $\sim 10 \text{ ms}$
 - (Force sensors: $5 \ \mu s$ rise time)
- Natural frequencies
 - Solution: Fourier analysis

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Liquids

LISTR LISTON PROPUSION



	Hexane	Ethanol	Isopropanol	Acetone	Water
Chemical Formula	C ₆ H ₁₄	C ₂ H ₅ OH	C ₃ H ₇ OH	C ₃ H ₆ O	H ₂ O
Absorption Coefficient (cm ⁻¹)	~0	17	67	~100	~3300
Absorption Depth (µm)	large	574	149	78	3
Density (g/ml)	0.66	0.79	0.785	0.790	1.0
Molecular Weight (g/mol)	86.18	46.07	60.10	58.08	18.02
Enthalpy of Vaporization (kJ/mol)	28.85	38.6	39.85	29.1	43.99
Surface Tension (dyne/cm)	18	22.3	21.7	23.7	73.05
Viscosity (mPa-s)	0.3	1.07	2	0.306	1.002





