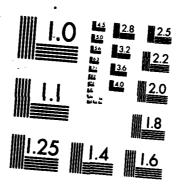
AD-A173 455 SILYLATED PHOSPHINES AND PHOSPHORANIHINES(U) TEXAS
CHRISTIAN UNIV FORT MORTH DEPT OF CHEMISTRY
R H MEILSON ET AL. 10 OCT 86 TCU/DC/TR-86-02
UNCLASSIFIED N80914-79-C-0632
F/G 7/3
NL



MICROCOPY RESOLUTION TEST CHART NATIONAL BUREAU OF STANDARDS-1963-A

UNCLASS:	(12)
AD-A173 455 CUMENTATION PAGE	
1a REPORT SEI	16. RESTRICTIVE MARKINGS
L'nclassified 2a. SECURITY CLASSIFICATION AUTHORITY	3. DISTRIBUTION / AVAILABILITY OF REPORT
2b. DECLASSIFICATION / DOWNGRADING SCHEDULE	Approved for public release; distribution unlimited.
4. PERFORMING ORGANIZATION REPORT NUMBER(S)	5. MONITORING ORGANIZATION REPORT NUMBER(S)
6a. NAME OF PERFORMING ORGANIZATION Department of Chemistry (If applicable)	7a. NAME OF MONITORING ORGANIZATION Office of Naval Research
6c. ADDRESS (City, State, and ZIP Code) Texas Christian University Fort Worth, Texas 76129	7b. ADDRESS (City, State, and ZIP Code) 800 N. Quincy St. Arlington, VA 22217
8a. NAME OF FUNDING/SPONSORING ORGANIZATION Office of Naval Research	9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER NOO014-79-C-0632
8c. ADDRESS (City, State, and ZIP Code)	10. SOURCE OF FUNDING NUMBERS
see block 7b	PROGRAM PROJECT TASK WORK UNIT NO. NO. ACCESSION NO.
11. TITLE (Include Security Classification) Silylated Phosphines and Phosphoranimines	
12 PERSONAL AUTHOR(S) R.H. Neilson, et al.	
13a. TYPE OF REPORT 13b. TIME COVERED FROM TO	14. DATE OF REPORT (Year, Month, Day) 15. PAGE COUNT
16. SUPPLEMENTARY NOTATION The view, opinions and/or findings contained in this report are those of the author(s) and should not be construed as an official Department of the Navy position, policy, or decision, unless so designated by other documentation.	
FIELD GROUP SUB-GROUP phosphoran	Continue on reverse if necessary and identify by block number) hine sliylaminophosphine imine methylenephosphine
19. ABSTRACT (Continue on reverse if necessary and identify by block number) The N-silylphosphoranimines [e.g.,	
Megsin=P(OCH2CF3)Meg] undergo a variety of reactions including deprotonation/substitution of the tions including deprotonation, and condensation with	
reactive phosphines. Some new 1,2-addition and oct 23 1986 oxidation reactions of the P=C systems, R2NP=CHR, R2NP(=NR)=CHR, and R2NP=CR-CH=CHR (R = SiMe3) are also reported. Lywords:	
76-10-21-055	
20. DISTRIBUTION / AVAILABILITY OF ABSTRACT	21. ABSTRACT SECURITY CLASSIFICATION
	Unclassified 22b. TELEPHONE (Include Area Code) 22c. OFFICE SYMBOL
DD FORM 1473, 84 MAR 83 APR edition may be used u	35.001111 (00.3311 (00.110)14 (01.111)3 1 1.005

UNCLASSIFIED 8.6 10 21 059

OFFICE OF NAVAL RESEARCH

Contract NOO014-79-C-0632

R&T Code 413a001---01 Replaces Old Task # 056-123

Technical Report No. TCU/DC/TR-86-02

Silylated Phosphines and Phosphoranimines

bу

RIH. Neilson, B.A. Boyd, D.A. DuBois, R. Hani, G.M. Scheide, J.T. Shore, and U.G. Wettermark

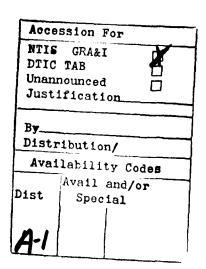
Prepared for Publication

in

Phosphorus and Sulfur

Texas Christian University Department of Chemistry Fort Worth, Texas 76129

October 10, 1986



Reproduction in whole or in part is permitted for any purpose of the United States Government.

* This document has been approved for public release and sale; its distribution is unlimited.



SILYLATED PHOSPHINES AND PHOSPHORANIMINES

ROBERT H. NEILSON*, BRUCE A. BOYD, DONN A. DUBOIS, RAHIM HANI, GARY M. SCHEIDE, JOHN T. SHORE, AND URSZULA G. WETTERMARK

Department of Chemistry, Texas Christian University, Fort Worth, Texas 76129, USA

Abstract The N-silylphosphoranimines [e.g., Me_3SiN=P(OCH_2CF_3)Me_2] undergo a variety of reactions including deprotonation/substitution of the P-Me groups, transilylation, and condensation with reactive phosphines. Some new 1,2-addition and oxidation reactions of the P=C systems, $R_2NP=CHR$, $R_2NP(=NR)=CHR$, and $R_2NP=CR-CH=CHR$ (R = SiMe_3) are also reported.

The chemistry of compounds containing Si-N-P and/or Si-C-P linkages is usually markedly different from that of the nonsilylated analogues. On the one hand, the Si-N or Si-C bonds may function as reactive sites so that processes such as transilylation and condensation reactions via silane elimination are commonly found. Alternatively, the steric and electronic properties of the silyl groups can impart both kinetic stability and interesting chemical reactivity to a variety of low-coordinate phosphorus systems. We report here several illustrative examples of these principles.

REACTIONS OF N-SILYLPHOSPHORANIMINES

Various P-functionalized N-silylphosphoranimines are useful precursors to cyclic and polymeric phosphazenes. Recently, the synthetic utility of such compounds has

been greatly expanded through two general types of reactions. First, deprotonation/silylation of the P-methyl group followed by a Peterson olefination affords a series of vinyl derivatives (eq 1). Some of these

 $R_2C(0) = Me_2C(0)$, PhC(0)H, PhC(0)Me, $PhC(0)CF_3$, $PhC(0)CH=CH_2$, $Me_2C=CH-C(0)-CH=CMe_2$

"monomers" undergo condensation polymerization to yield poly(phosphazenes) bearing substituted C=C side groups.

Second, we have studied a broad spectrum of Si-N bond cleavage reactions. For example, transilylation processes yield a series of novel N-silylphosphoran-imines (eq 2) bearing reactive substituents on silicon as well as some bis(phosphoranimino)silanes and -siloxanes (eq 3). The related reactions involving cleavage of the Si-N bond in $Me_3SiN=P(X)Me_2$ [X = Cl, Br, OCH_2CF_3 , and $N(SiMe_3)_2$] by reactive phosphines, e.g., Ph_2PY (Y = Cl, OCH_2CF_3), are generally more complex.

$$\begin{array}{ccc} \text{CF}_3\text{CH}_2\text{O} & \text{RSiMe}_2\text{Cl} & \text{CF}_3\text{CH}_2\text{O} \\ \text{Me}_3\text{SiN} = \text{PMe}_2 & \longrightarrow & \text{RMe}_2\text{SiN} = \text{PMe}_2 \end{array} \tag{2}$$

 $R = Ph. CH = CH_2, C1, (CH_2)_3CN, CH_2CH_2C(O)Me,$ $SiMe_2(CH_2)_2SiMe_2C1$

$$\begin{array}{ccc}
 & CF_3CH_2O & ECI_2 & CF_3CH_2O \\
2 & Me_3SiN=PMe_2 & & [Me_2P=N]_2E
\end{array}$$
(3)

$$E = SiMe_2$$
, $SiMe_2(CH_2)_2SiMe_2$, $Me_2Si(OSiMe_2)_n$
 $n = 1-5$

These reactions occur with elimination of Me_3SiY and, depending upon the choice of reactants, lead to various open-chain ionic species (e.g., A and B).

Surprisingly, the permethylated analoque of A (X = $0\text{CH}_2\text{CF}_3$) is produced in high yield in the "transilylation" of Me₂SiHCl with Me₃SiN=P(OCH₂CF₃)Me₂.

REACTIONS OF SILYLATED P=C COMPOUNDS

In this area our recent work has included a comparative study of the reactivity of the P=C bonds in the analogous P(III) and P(V) compounds, (Me₃Si)₂NP=CHSiMe₃ and (Me₃Si)₂NP(=NSiMe₃)=CHSiMe₃, respectively. In general, the 1,2-addition of polar reagents to the P=C bond occurs more readily in the P(V) system. For example, while the former does not react with PhPCl₂, the latter undergoes Si-N cleavage and ring formation (eq 4) subsequent to the 1,2-addition process. Some reagents, on the other hand, react smoothly with both substrates. Cyclopentadiene, for example, affords a Diels-Alder adduct in each case.

$$(Me_3Si)_2N-P \xrightarrow{PhPCl_2} NSiMe_3 \\ CHSiMe_3 - Me_3SiCl & H-C-P \\ Me_2Si & Ph$$

A stable, acyclic 1-phosphadiene has also been prepared (eq 5) and we have begun to investigate its derivative chemistry. Interestingly, simple oxidation reactions (e.g., with sulfur or Me_3SiN_3) are accompanied by isomerization to four-membered ring products (eq 6). In the sulfur reaction, a three-membered disulfide ring, $(\text{Me}_3\text{Si})_2\text{NP}(\text{S})-\text{S-C}(\text{SiMe}_3)\text{CH=CHSiMe}_3$ is also produced.

$$(Me_{3}Si)_{2}NP - C - SiMe_{3}$$

$$H - C$$

$$Me_{3}Si$$

$$Me_{3}Si$$

$$H - C - H$$

$$Me_{3}Si)_{2}N - P - C$$

$$H - C - C$$

$$Me_{3}Si)_{2}N - P - C$$

$$H - C - C$$

$$Me_{3}Si)_{2}N - P - C$$

$$H - C - C$$

$$Me_{3}Si)_{3}$$

$$E = NSiMe_{3}$$

$$E = NSiMe_{3}$$

$$(6)$$

ACKNOWLEDGEMENTS

This research is generously supported by the U.S. Army Research Office, the U.S. Office of Naval Research, and the Robert A. Welch Foundation.

\$55550 | Perrent Course of the CONTROL DESCRIPTION OF SERVICE CONTROL CONTROL