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#### VITREOUS ENAMEL DAMPING MATERIAL DEVELOPMENT

Binod Kumar University of Dayton Research Institute 300 College Park Dayton, Ohio 45469

November 1982

Final Report For Period October 1979 - July 1982

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REPORT DOCUMENTATION		READ INSTRUCTIONS BEFORE COMPLETING FORM
1. REPORT NUMBER		3. RECIPIENT'S CATALOG NUMBER
AFWAL-TR-82-4162	AD-4124 29	
4. TITLE (and Subsiste)  VITREOUS ENAMEL DAMPING		5. TYPE OF REPORT & PERIOD COVERED Final Technical Report October 1979 - July 1982
MATERIAL DEVELOPMENT		6. PERFORMING ORG. REPORT NUMBER UDR-TR-82-105
7. AUTHOR(a)	-	S. CONTRACT OR GRANT NUMBER(s)
Binod Kumar		F33615-79-C-5108
9. PERFORMING ORGANIZATION NAME AND ADDRES	is	10. PROGRAM ELEMENT, PROJECT, TASK AREA & WORK UNIT NUMBERS
University of Dayton	1	62102F
Research Institute Dayton, Ohio 45469	1	24180312
11. CONTROLLING OFFICE NAME AND ADDRESS		12. REPORT DATE
Materials Laboratory (AFWAL/MLLN)		November 1982
Air Force Wright Laboratories (AF		13. NUMBER OF PAGES
Wright-Patterson AFB, Ohio 45433		100
14. MONITORING AGENCY NAME & ADDRESS(If differently)	ant from Controlling Office)	15. SECURITY CLASS. (of this report)
		Unclassified
	!	15a. DECLASSIFICATION/DOWNGRADING SCHEDULE
16. DISTRIBUTION STATEMENT (of this Report)		

Approved for public release; distribution unlimited

17. DISTRIBUTION STATEMENT (of the electrect entered in Block 20, if different from Report)

18. SUPPLEMENTARY NOTES

19. KEY WORDS (Continue on reverse side if necessary and identify by block number)

Damping

Glass

Ename1

Measurements

**Vitreous** 

20. ABST RACT (Continue on reverse side if necessary and identify by block number)

This report describes the results of several experimental investigations pertaining to the effects of composition, viscosity, microstructure, and constraining layer on the damping properties of vitreous enamels. New vitreous enamels such as mixed alkali silicate, lead silicate, and two phase fluoride composition have been characterized. The mixed alkali and lead silicate compositions exhibits the characteristic, viscoelastic damping peak. temperatures and the peak intensities are compositions dependent.

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fluoride composition initially shows a typical vitreous enamel damping peak; however, after a heat treatment significant broadening of the damping peak is noted. It is believed that the broadening occurs due to a second phase which develops after the heat treatment. A relationship between the glass transition temperature and damping temperature have been developed. It has been determined that the viscosity in the damping range corresponds to approximately  $10^6 - 10^7$  poise. The viscosity range is favorable for processes like nucleation and crystallization. Effects of nucleation and crystallization on the damping properties have been studied and analyzed for two compositions. The nucleation and crystallization have an adverse effect on the loss factor. Damping properties of three constrained layer systems have been evaluated and analyzed. The general effect of the constraining layer is to broaden the damping peak.

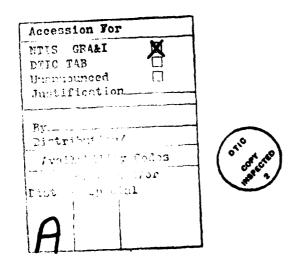
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#### FOREWORD

The work described in this report was performed at the University of Dayton Research Institute for the Metals Behavior Branch, Metals and Ceramics Division, Materials Laboratory, Air Force Wright Aeronautical Laboratories (AFWAL/MLLN), under Contract number F33615-79-C-5108, "High Frequency Fatigue of Turbine Blade Material." The contract was administered under the direction of AFWAL by Dr. D. I. G. Jones (MLLN). The program was conducted by the Aerospace Mechanics Group, University of Dayton Research Institute, Dayton, Ohio with Mr. Michael L. Drake as the principal investigator.

The investigations of the effects of composition, viscosity, microstructure and constraining layers on the damping properties of vitreous enamels were conducted by Dr. Binod Kumar. Experimental evaluation and data reduction was accomplished by Mr. S. Hilton, Mr. W. Goddard, Mr. D. Hopkins, and Mr. P. Graf. This work was performed during the period October 1979 to July 1982.



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## SECTION I INTRODUCTION

In order to control high cycle fatigue in vibrating structures at elevated temperatures, vitreous enamel coatings have been suggested and used [1,2,3]. The coatings dissipate vibratory energy due to their characteristic high mechanical loss in the viscoelastic range. The dissipation of vibratory energy by such treatment improves the performance, reliability, and reduces maintenance costs of the structure. At lower temperatures (<460°C), polymeric coatings are commonly used. Preliminary feasibility experiments and viscoelastic damping data for vitreous enamel coatings have been reported by Nashif [4] and Sridharan [5].

The University of Dayton has been engaged, in developing viscoelastic damping measurement techniques and vitreous enamel compositions for damping applications in aircraft engines for approximately 10 years, under contracts funded by the Air Force Materials Laboratory, Wright-Patterson Air Force Base. Significant developments have been made in the measurement and application techniques. A better understanding has developed in regard to the effects of composition (6) on the damping properties.

The primary objectives of the studies reported here were to further evaluate damping properties of new compositions, analyze previous data and develop a relationship between the structure and the damping properties, evaluate the constrained layer damping concept with the vitreous enamel, and analyze the difficulties associated with the utilization such as nucleation and crystallization of vitreous enamels as a vibration damping material.

Vitreous enamel or inorganic silicate glasses exhibit damping peaks at various temperatures. For a frequency of 1 Hz a peak occurs at about  $-30\,^{\circ}$ C, which is related to the alkali ion diffusion. A second peak at the same frequency occurs at about 200 $^{\circ}$ C. The damping ( $\eta$ ) of the first peak is approximately 0.001, whereas the damping of the second peak varies from 0.001 to 0.02, depending on the glass composition. For a single alkali glass, the

damping intensity (n) of the second peak is approximately 0.001. The addition of a second alkali causes a large increase in the intensity of this peak to as high as 0.02. The damping peaks are also frequency dependent. Increasing frequency increases the peak temperature. For example, a decade increase in frequency shifts the peak toward a higher temperature by approximately 15°C. These two peaks are the results of anelastic damping. Readers can refer to a recently published review paper (7) for the mechanisms and the related information concerning the two peaks.

A third peak of significantly higher damping intensity  $(Q^{-1} \simeq 1)$  than the first two occurs at a higher temperature. Associated with this damping peak is a large change in the complex modulus. A change of about 40 percent in the modulus can be expected. This peak is frequently referred to as the viscoelastic peak. This is the peak of primary interest in dissipating the vibratory energy at elevated temperature and will be the main topic of discussion in this report.

#### SECTION II

#### TECHNICAL PROCEDURES

#### 2.1 EXPERIMENTAL

#### 2.1.1 Glass Preparation

All of the compositions, except the standard commercial compositions, were formulated and synthesized from the analytical grade raw materials. Required amounts of oxide and carbonate raw materials were weighed and mixed in a V-blender for approximately one hour. The weighed material was then melted in a platinum crucible at appropriate melting temperatures. The molten glass was quenched by pouring into a container of cold water. The quenched glass was then dry ball milled, using Al<sub>2</sub>O<sub>3</sub> grinding media, for 24 hours. The milled powder was then screened to obtain various particle size splits. The test beams were coated with vitreous enamels of particle sizes passing through a 100-mesh screen but not through a 150-mesh screen.

## 2.1.2 Method of Coating Application

Prior to spraying, the metal substrate was sandblasted using 36-mesh silicon carbide grit and an air pressure of 75 psi. The powder material was applied to the metal substrate by plasma spraying. A Meteo 3MB plasma spray apparatus was operated at 400 amps DC, 75 volts DC at 30,000 watts setting with a gas flow of 15 scfh H<sub>2</sub> and 80 scfh Ar. A gas mixture of 84.2 percent Ar and 15.7 percent H<sub>2</sub> was used. The coated beam was then fired in a resistance-heated furnace for approximately three minutes or until the surface appeared to be smooth.

#### 2.1.3 Vibration Damping Measurements

An apparatus and technique was developed to accurately and reliably excite, and measure the response of a beam specimen at temperatures exceeding 1,000°C.

The specimen used was a cantilever beam coated on one side with an enamel or glass. A typical specimen is illustrated in

Figure 1. The specimen was clamped in the fixture as illustrated in Figure 2. The air cylinder insured that a constant clamping pressure was maintained on the root section of the specimen over the entire temperature range. This fixture also allowed for thermal expansion of the fixture and high temperature creep of the clamping bolts.

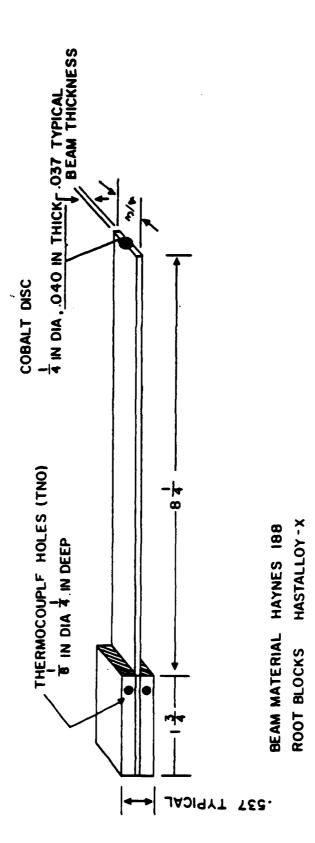
The force gage, mounted in series with the clamping bolt and the air cylinder, was used to measure the response of the specimen. The force gage was well removed from the high temperature environment.

An electromagnetic transducer was used to excite the specimen. This transducer was specially designed to operate from room temperature to at least 1,000°C. The design used a closed loop cooling system incorporating a modified room air conditioner. A block diagram of the apparatus and complete measuring system is shown in Figure 3, and Figure 4 is a schematic of the transducer cooling system.

The specimen and fixture were placed in the furnace and heated to the desired temperature, usually above the annealing temperature of the enamel. The cantilever beam specimen was excited at its free end by a sinusoidally varying magnetic force induced by the electromagnetic transducer. A high Curie temperature cobalt disc was attached to the end of the beam to allow for excitation of the nonmagnetic specimens. The frequency of oscillation was varied until a resonance was detected. At resonance the shear force in the beam reached a maximum and was measured by the dynamic force gage. The force gage measured the variation of the shear force at the root of the cantilever beam specimen.

#### 2.2 CALCULATION OF DAMPING PROPERTIES

The damping characteristics of the coatings were determined by measuring the vibration response of a composite cantilever beam at varying temperatures over the viscoelastic range. It is assumed that the enamel is a viscoelastic material; that is, the modulus of the enamel can be treated as a complex quantity



Typical Cantilever Beam Specimen for High Temperature Damping Material Evaluation. Figure 1.

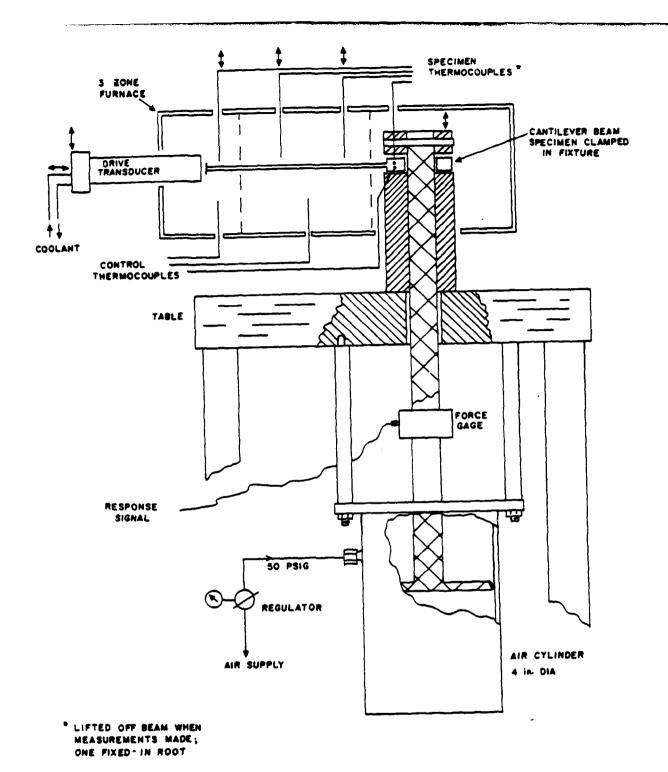


Figure 2. High Temperature Damping Test Apparatus.

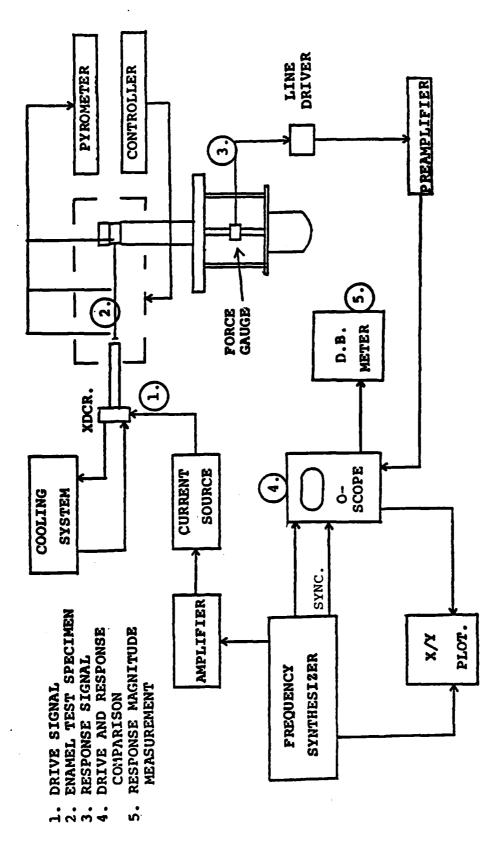
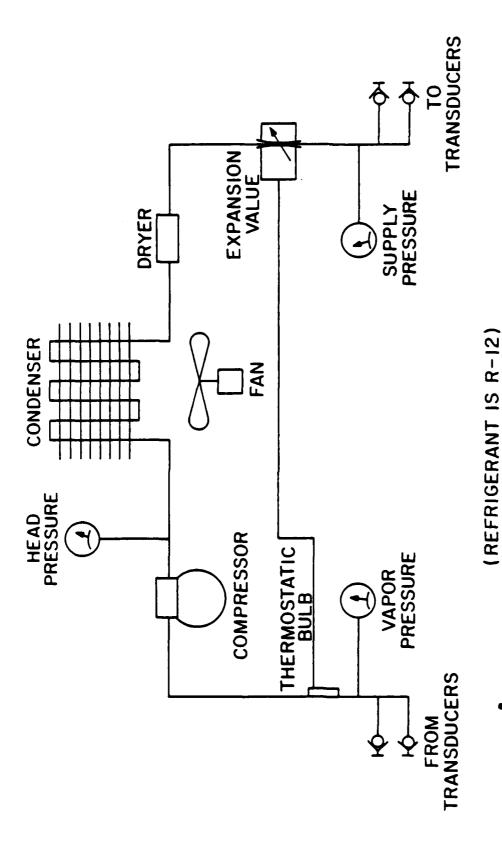


Figure 3. BLOCK DIAGRAM OF HIGH-TEMPERATURE TEST APPARATUS



Schematic of High Temperature Damping Test Apparatus Cooling System. Figure 4.

$$E_D^* = E_D^* + iE_D^* = E_D^* (1 + i tan \delta)$$

$$\eta_2 = tan \delta = E_D^* / E_D^*$$

where  $E_D^*$  is the storage or Young's Modulus of the enamel and  $\tan\delta$  is the ratio of the dissipative modulus,  $E_D^*$ , to the storage modulus. Consider the metal beam with an enamel coating on one side as shown in Figure 5.

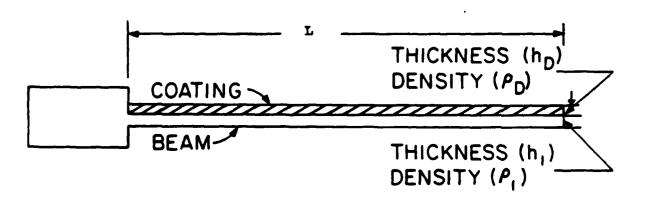


Figure 5. Coated Oberst Test Beam

The formulas developed by Oberst [8] and used by many other investigators were used to determine the damping properties of the enamel as a function of frequency and temperature. These formulas are:

$$(\omega_{n}/\omega_{1n})^{2}(1+h_{D}\rho_{D}/h_{1}\rho_{1}) = \frac{1+2(E_{D}/E_{1})(h_{D}/h_{1})A+(E_{D}/E_{1})^{2}(h_{D}/h_{1})^{4}}{1+(E_{D}/E_{1})(h_{D}/h_{1})}$$
(1)

and

$$\frac{\eta}{\eta_2} = \frac{(E_D/E_1) (h_D/h_1) [2A+2 (E_D/E_1) (h_D/h_1)^3 + (E_D/E_1)^2 (h_D/h_1)^4 -1]}{[1+(E_D/E_1) (h_D/h_1)][1+2A (E_D/E_1) (h_D/h_1) + (E_D/E_1)^2 (h_D/h_1)^4]} (2)$$

where:

$$A = 2 + 3(h_D/h_1) + 2(h_D/h_1)^2.$$
 (3)

 $\omega_{\rm n}$  = natural frequency of the nth mode of the composite beam,  $2\pi f_{\rm n}$ , rad/sec

 $\omega_{\text{ln}}$  = natural frequency of the nth mode of the metal beam,  $2\pi f_{\text{in}}$ , rad/sec

 $h_{\text{D}}$  = thickness of enamel coating applied to composite beam

 $h_1$  = thickness of metal beam

 $\rho_D$  = density of enamel coating

 $\rho_1$  = density of metal beam

 $E_{D}$  = real part of the modulus of enamel coating

E<sub>1</sub> = Young's modulus of metal beam

 $tan \delta_{p} = effective loss factor of composite beam (=n)$ 

 $tan \delta = loss factor of enamel coating (= <math>\eta_2$ )

The quantities of  $h_D$ ,  $h_1$ ,  $\rho_D$ , and  $\rho_1$  are known and are assumed to remain constant with temperature. The parameters,  $\omega_n$ ,  $\omega_{1n}$ , and  $\eta$  are experimentally measured. The value of  $\eta$  is determined from

$$\eta = \tan \delta_e = \frac{\Delta \omega_n}{\omega_n} = \frac{\Delta f_n}{f_n}$$
 (4)

where  $\Delta f_n$  is the bandwidth at the half-power points of the response peak for the nth mode. The value of  $E_1$  can be determined from the measured response of the uncoated metal beam using

$$\xi_n^4 = \mu_1 \omega_{1n}^2 L^4 / E_1 I_1$$
 (5)

where:

 $\xi_n^4$  = the eigen value corresponding to the nth mode and is a constant, determined by the boundary conditions

 $\mu_1 = \rho_1 bh_1 = the mass per unit length of the metal beam$ 

L = the length of the beam

 $I = \frac{1}{12} bh_1^3 =$ the second moment of area of the metal beam about its centerline.

The values of  $\xi_n^4$  for beam with classical boundary conditions are well known and can be found in reference[9]. Thus, from the measured resonant frequencies of the coated and bare beams and the measured composite loss factor, tan  $\delta_e$ , the damping properties of the enamel can be determined as a function of temperature and frequency.

The resonant frequencies and modal damping of five to six modes of the coated beam, covering a frequency range of 100 Hz to 1,500 Hz, can usually be measured for each temperature. Thus, the damping properties of the vitreous coating over a decade of frequency at a given temperature can be easily and quickly determined.

## SECTION III RESULTS AND DISCUSSION

During the development of high temperature vibration damping enamels, one of the primary goals has been to develop a better understanding of the effects of structure, composition, and physical properties on the damping behavior of vitreous enamels. In view of this goal, several experimental and theoretical investigations were conducted. Those investigations can be broadly classified as follows:

- Damping properties of new compositions;
- Relationship between the glass transition temperature and the damping temperature;
- Effect of nucleation and crystallization on the viscoelastic damping;
- Constrained layer damping with vitreous enamels.

#### 3.1 DAMPING PROPERTIES OF NEW COMPOSITIONS

## 3.1.1 Mixed Alkali Glasses

As indicated in Section I, mixed alkali glasses exhibit a characteristic damping peak at lower temperatures (100-200°C). It has also been noted that some commercial glasses containing mixed alkali exhibit intense and broad damping peaks. Therefore, it was hypothesized that there is a relationship between the viscoelastic damping and presence of mixed alkali in the composition. In order to verify the hypothesis, an experiment was designed in which alkali oxides were systematically varied. Properties of the glasses, i.e., the thermal expansion, damping behavior, and their thermal stability in the damping range were determined. This subsection describes the results of the experimental design on the mixed alkali glasses and also recommends the future course of the investigation.

Molar composition of the glasses and their properties are summarized in Table 1. Compositions numbered 1 through 6 have been obtained from the  $\text{Li}_2\text{O-K}_2\text{O-SiO}_2$  system. The coefficient of

thermal expansion shows a linear increase as Li<sub>2</sub>O is replaced by the K<sub>2</sub>O, which is also graphically shown in Figure 6. The coefficients of thermal expansion of Li<sub>2</sub>0.3SiO<sub>2</sub> and K<sub>2</sub>.3SiO<sub>2</sub> glasses are  $53 \times 10^{-7}$ /°C and  $140 \times 10^{-7}$ /°C respectively. There is a slight variation in  $T_{\alpha}$ ; however, the dialtomeric softening point exhibits a general decrease as the Li<sub>2</sub>O is replaced by These glasses also exhibited varying degrees of thermal stability. For example, the composition 1 becomes opaque after the powdered frit was applied and vitrified on the metallic substrate. Compositions 2 and 2A became opaque during the damping measurements. Compositions 3, 3A, and 4 remained unchanged after the damping measurements. General appearance of the glass compositions 2, 2A, 3, 3A, and 4 after damping measurements are shown in Figure 7. Compositions 5 and 6 could not be successfully applied on to the metallic substrate, possibly because of the extreme hygroscopic nature of these compositions.

All compositions from the system Li<sub>2</sub>O-Na<sub>2</sub>O-SiO<sub>2</sub> (7 through 11) exhibited unstable behavior during application and damping measurements and therefore were dropped experimental investigation.

Reduced Temperature Nomograms for compositions 2, 3, and 4 are shown in Figures 8, 9, and 10 respectively, with experimental and reduced data given in Tables 2, 3, and 4. Composition 2 shows a broad and relatively high loss factor as compared to compositions 3 and 4. In order to make a direct comparison of loss factor of the three compositions, Figure 11 exhibits a plot of the loss factor at 100 Hz versus temperature.

It is obvious that only a narrow compositional range of the mixed alkali series exhibit stable glasses whose damping properties could be satisfactorily determined. It appears that the damping properties of these stable compositions are superior than other multicomponent commercial compositions. However, further investigations are needed to substantiate any relationship between the presence of mixed alkali and viscoelastic damping.

TABLE 1
GLASS COMPOSITIONS AND THEIR PROPERTIES

Remarks	Opaque during application	Opaque after damping measurement	Stable compositions		Powder could not be	Vitrified	lall of these glasses	exhibited unstable	behavior during	application and damping measurements	
Dialtomeric Soft.Point	630°C	2°7E3	231°C	518°C	518°C						;
F	475	481	475	469	469						
α (× 10 <sup>7</sup> )	52.3	70.9	89.3	104.6	121.0	140.0					
Molar Composition	L120 · 3S102	$2K_20 \cdot 8L1_20 \cdot 3S10_2$	0.4K20 · 0.6L120 · 3S102	$0.6K_2^{\circ}0 \cdot 0.4Li_2^{\circ}0 \cdot 3Sio_2^{\circ}$	0.8K20 · 0.2L120 · 3S102	K <sub>2</sub> 0 · 3S10 <sub>2</sub>	$0.2$ Na $_2$ O $\cdot$ $0.8$ Li $_2$ O $\cdot$ $3$ SiO $_2$	0.4Na20 . 0.6Li20 . 3SiO2	0.6Na20 · 0.4Li20 · 3SiO2	0.8Na20 . 0.2Li20 . 3SiO,	Na <sub>2</sub> 0 · 3Sio <sub>2</sub>
	-	2 2A	34	4	2	9	7	8	6	10	11

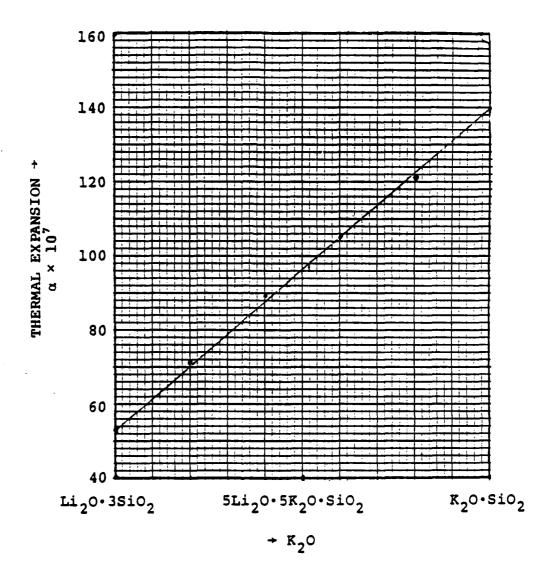


Figure 6. Effect of K<sub>2</sub>O Substitution for Li<sub>2</sub>O on the Thermal Expansion Coefficient.

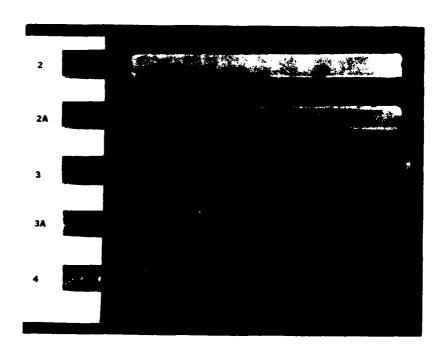
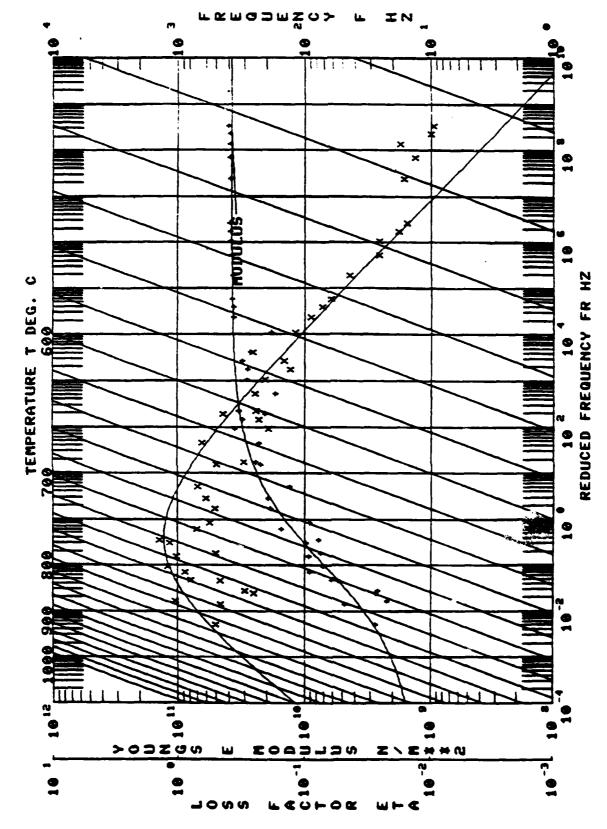


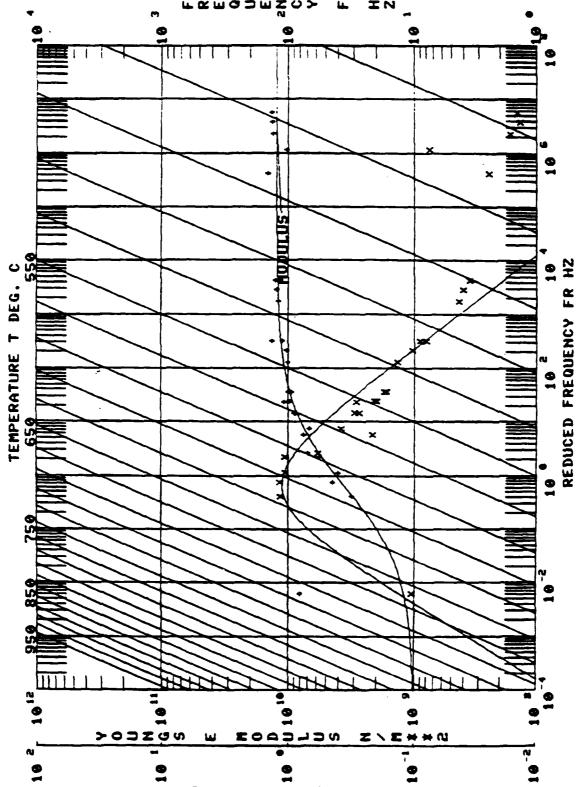
Figure 7. Appearance of the Glass Coatings After Damping Measurements. Numbers on the Left Indicate Glass Composition.



Reduced Frequency Nomogram Showing Loss Factor and Young's Modulus for Composition 2. 8 Figure

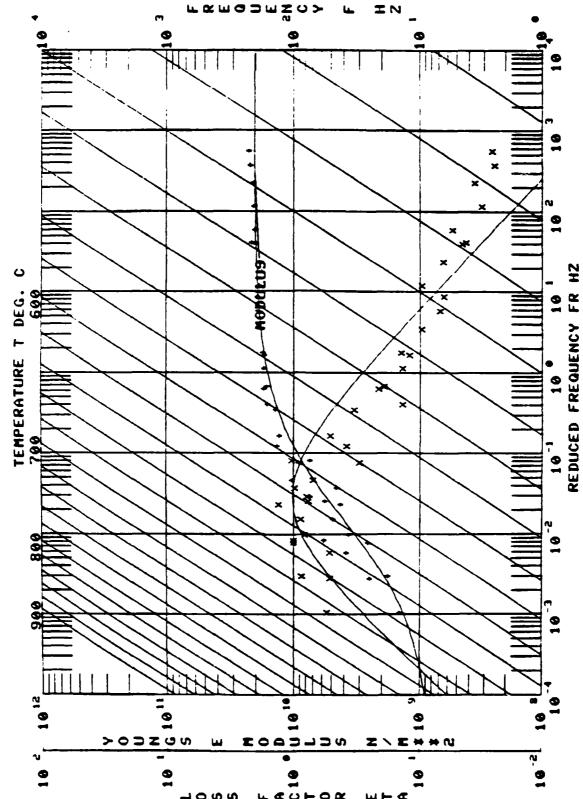
TABLE 2
EXPERIMENTAL AND REDUCED DATA OF COMPOSITION 2(AM2)

1.56. (2.5.



Reduced Frequency Nomogram Showing Loss Factor and Young's Modulus for Composition 3. ъ Figure

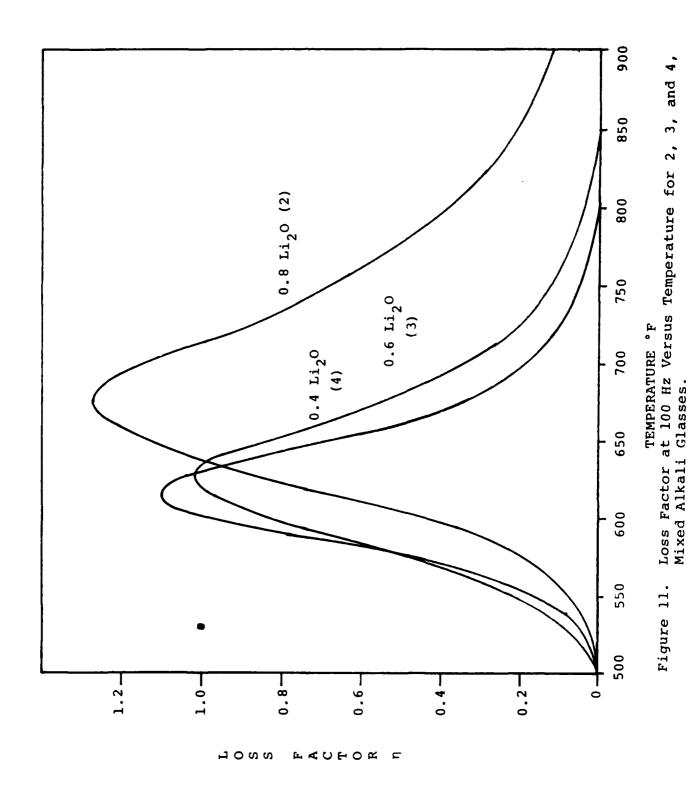
			EXPERI	EXPERIMENTAL	AND	REDI	REDUCED	DATA	OF COMPO	COMPOSITION 3	(AM3)	
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10. Reduced Temperature Nomogram Showing Loss Factor and Young's Modulus for Composition 4. Figure

TABLE 4
EXPERIMENTAL AND REDUCED DATA OF COMPOSITION 4 (AM4)

EXPER	EXPERIMENTAL COLE 118' MATERIAL :01-40-3, ANS. DATA CAMPER	00 <del></del>	, 72. 48% 5102, 22. 72% 20, 4.8% I E0	2.724(20,	1.8.F	(5)					
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į	MODULUS	LOSS TEM	TENP.	FPEO.	NOVE	BEAN MOD.	COMPOSITE BENN F LOSS FAC.	BEMM FREG. MZ		LOSS MOD.	
-	1.49922E+09	.5616	704.4		ri.	1.53666 +11	5000		-,	3.41892E+08	
'n	1.83826E+09	. 8352	704.	266.3	mi.	1.83128E+11	6.E.S.			62715E+09	
m·	.95186E+09	5256	784.4	000 000 000	÷u	1 . 26 43 56 + 11	0.10.		•	16618F+09	
<b>+</b> u	79476792 .	1868	676.7	9.00	์เณ่	1.35545E+11	6114			35729E+09	
D 44	. 64236£+03	1.0296	676.1	269.1	'n	1.35311E+11	- 020·		•	72067E+09	
~	5.039316+09	. 9061	676.7	532.7	÷ı	1.85314E+11	9968		•	53568E+09	
æ	.83131E+09	7876	676.7	9.555. 2.555. 6.655.	'nú	1.350166+11	2000		•	59655 + 69 6976 3F + 69	
<b>)</b> =	4.05233E+03	1.0249	6.5	97.8	ini	1.87434E+11	.0482		_	04-346-40	
=	4.351836+09	1.3469	6.25	272.9	ų.	1.885146+11	.0473		•	86136E+89	
2	1. 001 CAE+10	5172	0.00 0.00	248.5	÷u	1.88673E+11	.0573			66997.E+69	
23	9.400/5E+03	הליקמ הליקמ	5.10	, c	أما	1.90%576+11	.0472		-	32126	
<u> </u>	7.672785+03	1.0512		278.2	i'n	9.9E9E	.0620			<b>3€639€</b>	
9	1.33304€+19	.5217	621.1	557.7	÷	1.91662E+11	. 0562		_	954736	
2:	1.64943E+10	140	521.1	1396.7	ەف	1.913676+11	2004			121428 + 26 A	
2 Q	1.38505E+10	3308	593.3	236.7	im	3565E	.0341		•	10 30 W	
8	1.66192E+10	1973	600	566.8	÷1	3621E	.0227		•	279475+09	
ä	1.767038+10		503. 503. 503.	940.3 443.3	j, d	1.935958+11	9719		UN	177835+09	
32	1.76745E+10	2155	265.6	164.3	ini	1.96283E+11	0250		יטו	80924E+09	
21	1.866936-10	7 4 4 4	565.6	÷ 000	ų.	6323E	61.75		N -	.592678+09	
Ko	1.85846E+10	1866	200	2,7	÷ú	1.96630E+11	200		-	38496E+09	
12	2.00393€+16	.0658	565.6	1424.1	ó	7017	.0087			31S21E+09	
<b>8</b> 8	1.90916E+10	809.	<b>537</b> .0	8 8 8 8	ų.	22	. 0125 250			889796+09	
38	2.02999E+10	. 456 6466	537.8	7.050	ŕ	1.98676E+11	. 0061			32854E+08	
5	2.06922E+13	355	537.8	1434.7	6	1.99435E+11	.0075			15475E+09	
38	20364E+10 14064E+10	. 6454 4244	9.4 9.4 9.4	200	ย่า	2.01039E+11	.0001			0.00 LOC 1400 894.00 PF + 98	
i A	3	. 0365	516.0	585.4	i÷	2.01016E+11	. 0050			79811E+A8	
A	24580E+10	. 0253	510.0	9.179	S	9	. 6936			67084E+48	
ጸዩ	3 <b>0810E+1</b> 8	.0262	5.00 0.00 0.00	1453.0	<b>.</b>	ä٠	.000 8000			93810E+68 75250E+38	
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24	79782E+10	. 6387 6193	338.9	601.1 200.2	÷ď	2.188746+11	. 603.	587.			
<b>;</b> =	2.75254E+11	. 6123	388.0	1491.1	: 6	. 27751E	<b>8050</b>			2106	
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## 3.1.2 Fluoride Glass

This material was formulated and synthesized in our laboratory and is different from other materials in the sense that it develops two phase structure upon heat treatment. The beams were coated using the standard plasma spray technique. Damping data for the material before and after heat treatment are shown in Figures 12 and 13 respectively. Before heat treatment, the material exhibits a damping peak typical of many vitreous enamel compositions (Figure 12). After heat treatment, the peak becomes broader and the peak intensity decreases. This interesting damping behavior of the material after heat treatment believed to have resulted from the two phase structure.

### 3.1.3. Lead Silicate Compositions

Damping properties of several compositions from the Pb0- ${\rm Si0}_2$ - ${\rm R}_2$ 0 system were evaluated. Table 5 shows the compositions of the glasses as calculated from the batch ingredients. The table also shows general characteristics of the glasses. It is noted that the incorporation of alkali oxide ( ${\rm R}_2$ 0) in mixture rather than a single oxide helps in stabilizing the glass as evidenced by compositions III, V, VII, IX, X, and XI.

Loss factors as a function of temperature of the seven stable compositions from Table 5 are shown in Figures 14 through 20, with experimental and reduced datagiven in Tables 6,7,8,9,10,11,and 12.

TABLE 5
COMPOSITIONS (PERCENT WEIGHT) OF LEAD SILICATE GLASSES

	ı	II	III	IV	V	VI	VII	VIII	IX	х	XI
SiO <sub>2</sub>	49.02	45.9	44.3	49.02	44.3	44.3	43.9	43.9	44.3	44.3	44.3
PbO	40.02	45.9	44.3	49.02	44.3	44.3	43.9	43.9	44.3	44.3	44.3
к <sub>2</sub> 0	1.96		1.7		1.7		1.7	3.3	1.7	3.4	5.1
Na <sub>2</sub> O		8.2	8.0	1.96	9.7	11.4	7.2	7.2	6.3	6.3	6.3
Na <sub>2</sub> 0 Li <sub>2</sub> 0			1.7				3.3	1.7	3.4	1.7	
Remarks	Devitrified (Opacified)	Devitrified	Stable Glass	Devitrified	Stable Glass	Devitrified	Stable Glass				

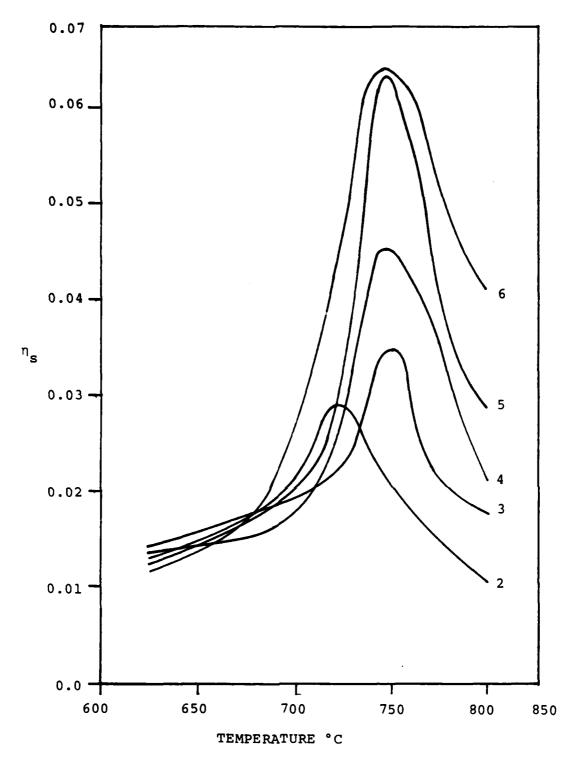
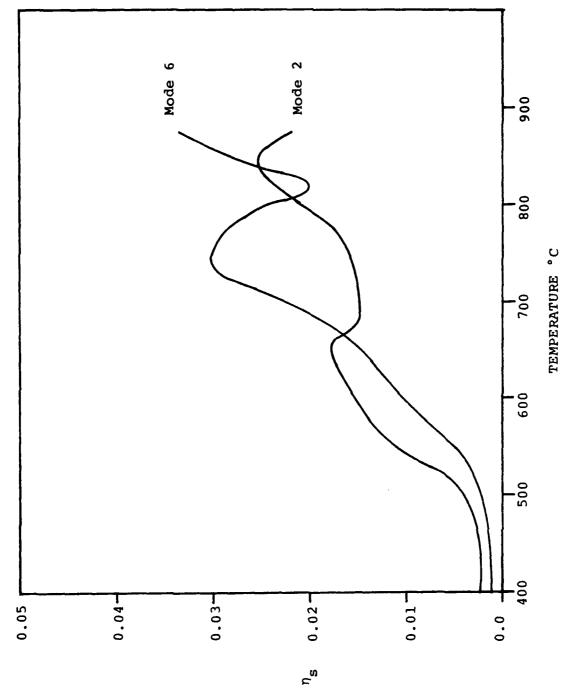


Figure 12. Modal Damping from Fluoride Materal (Number Superimposed on the Curves Indicate Mode Number)



Modal Damping From Fluoride Material After Heat Treatment at 800°C for 20 Hours. Figure 13.

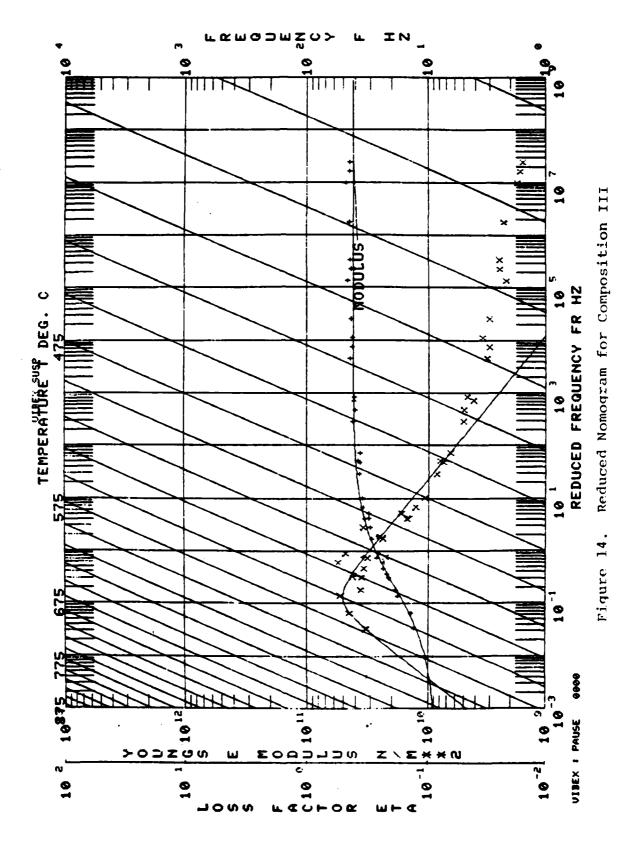
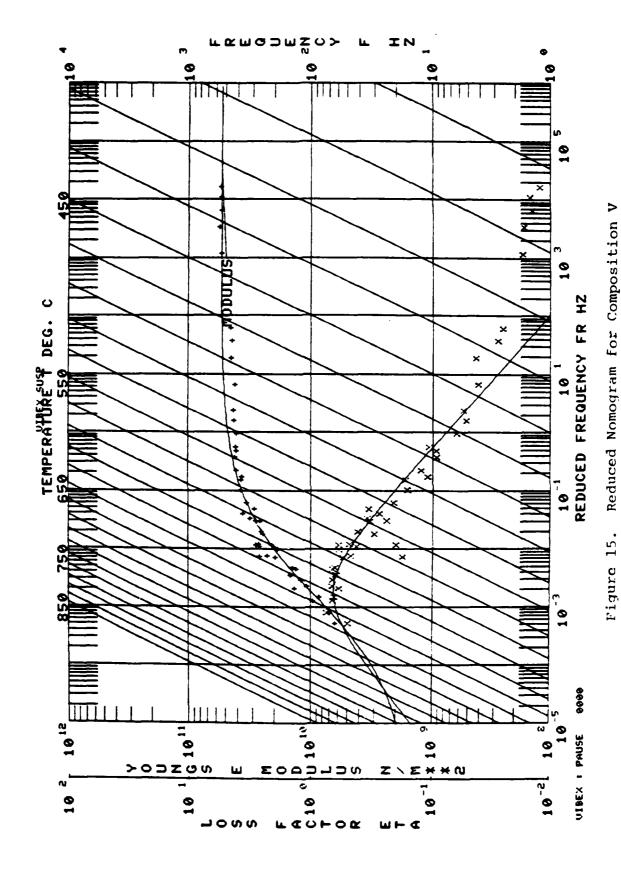


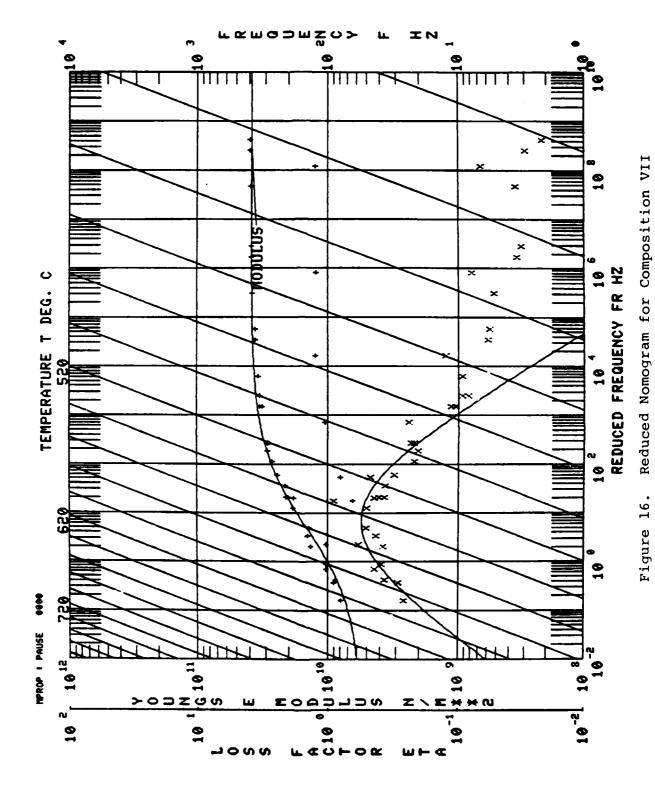
TABLE 6
EXPERIMENTAL AND REDUCED DATA OF COMPOSITION III (AMB3)

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7.000000000000000000000000000000000000	4.58722E+10
S	REPUC S



OTHER :N									
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7	3175+00	5,62	676.7		nj	1.82388E+11		.65	26791
8.79	944E+99	6603	676.7		m	1.317576+11	.0307	278.	S. 75069E+8
1.22	841E+10	.6752	676.7	534.	÷,	1.32121E+11	SET-0.	500	59477
43	353£+10	.6487	0	821	יינט	1. 232.CE+11	.847c	300	2997
2.68	527E+10	. 1815	676.7	13.7	٠.۵٠	1.24239E+11	. 9225	1353.	57431
7.10	368E+83	. 7621	200	36	ni (	1.83672E+11	9620.		5051
1.10	527E+19	.6879	662.0	, V	m.	1.830056+11	9336	ر ان ان	933
_	376E+10	.6739	56.0	538	÷1	1.533318+11	.040	χ. 	26936
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2	2.94458E+1	14171	9	918.0		1.96394E+11	0850.		52786	
C	2.37410E+1	6 . 2343	2	104.6	Ni ·	2.00320E+11	. 0363		7348	
I	3.25830£+1	. 1164	3	589.0		2.00473E+11	9619		50562	
15	3.39138E+1	6+68.	519.9	963.8		2.01329E+11	9166		21354	
91	3.85531E+1	. 4535	464.4	1.88.1		2.04049€+11	. 0102		2.06274E+09	
2	1.25365E+1	. ୯୫୫୨	454.4	231.3		2.03060E+11	2900.		1.01369€+09	
<b>=</b>	3.91979€+1	. e35e	į.	535.6		2.05345€+11	. 0067		1.37368E+09	
2	3.92175E+1	. 0324	424.4	986.1		2.06107E+11	. 0062		1.26269E+09	
8	6.57825E+8	9277	5	271.4		1.95959€+11	.0406		6.09547E+09	
<b>7</b>	Z.17593E+1	37.68	;	558.6 500	÷ 1	1.968576+11	.6467		8.86932F+09	
86	6. 47.363E+1	9326	35	933.6	<i>i</i>	1.976756+11	.6443		7.87936F+@9	
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ຂ	3.36265E+1	. 1073	9	581.5			.0137		3.613776+09	
ጽ	3.391386+1	₩. ₩348	510.0	963.8	'n	2.01329€+11	.0148		2.87536E+09	
<u></u>	3.49308E+1	9769.	<u>بن</u>	106.7			.0169		3.30378E+09	
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E (	3.65620E+1	0000	ည္တဲ့	588.6			.0110		2.19525E+09	
<b>7</b>	3.65133E+1	5850	eg.	974.9	'n		.0197		2.12672E+49	
S.	3.96355E+1	. 0363	ģ	108.8			.0000		1.43695E+03	
<b>%</b>	1.27669E+1	5390.	426.7	284.1			. 6052		8.74942E+08	
Ä	4.02447E+1	60000	9	599.5	÷ (		6800.		1.21945E+09	
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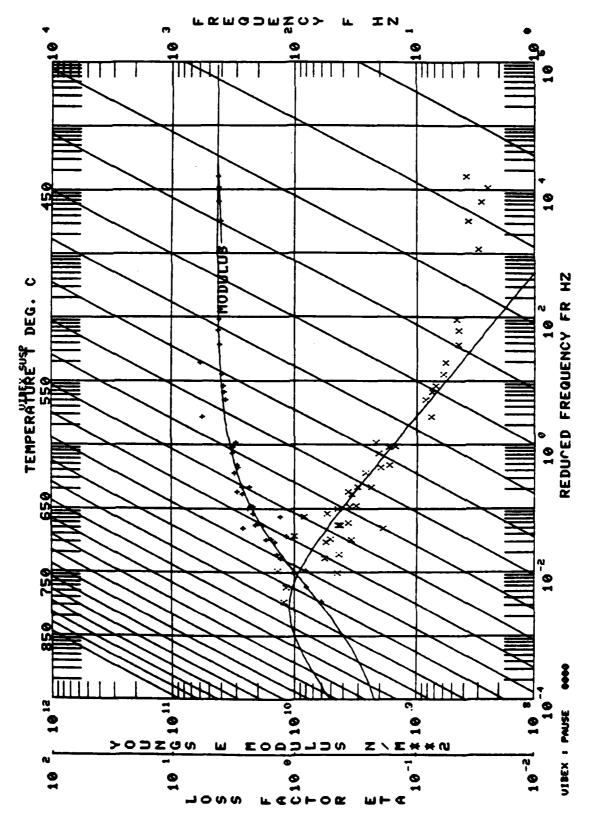


Figure 17. Reduced Nomogram for Composition VIII

TABLE 9
EXPERIMENTAL AND REDUCED DATA OF COMPOSITION VII (AMB8)

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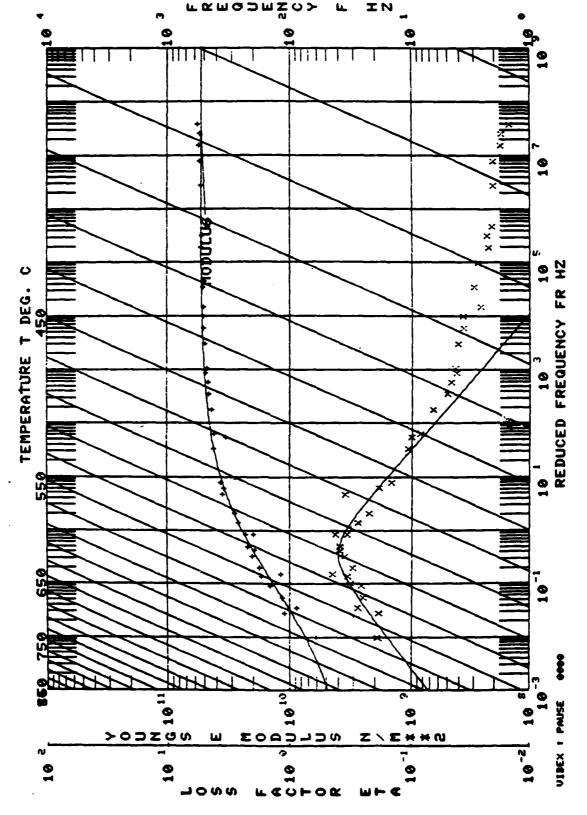


Figure 18. Reduced Nomogram for Composition IX

TABLE 10 EXPERIMENTAL AND REDUCED DATA OF COMPOSITION IX (AMB9)

EXPERIMENTAL COPE :201
MATERIAL :AMB9, MAETX
DATA SOURCES
MANUFACTURER :NO!

-	1:45	11E+98	4E+08	96+96	10E+03	SOE + 93	12E+08	31E+09	SE+99	37E+09	SF +MG	1F+38	2014	204	0040	20110			00100		100	115 + 69	96+09	SE+03	9E+09	.E+09	2E+09	6E+03	46+0.3	7.00 1.00 1.00 1.00 1.00 1.00 1.00 1.00	N 0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	454.00	3F + 30	₩ + 08	9E+99	1E+09	95+09	30 t L C C C C C C C C C C C C C C C C C C	20110	56+60	SE+09	7E+09	0E+00	5E+03	10 + 00 1 F + 00	4E+09	4E+09	36+09
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COCO MEST	H2	95.8	267.3	522.5	865.6	1295.0	95.5	266.7	524.1	362.0	1300.0	a v	9.44	7000	9	1.000	1	1000	900	200	3.795.1	3	629.0	529.0	876.0	1310.0	S:-2	5.1.5	530.5	9.00	1315.6	20.00	ייבי מיקי	283.0	1323.0	9.	274.0	9.00	9.00 700 1		V. 0E.S	893.5	1336.9	2000	0 (P. C. P. P. C. P.	9.00	1344.0	₹.66
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	•	1.690@7E+11	1.30654E+11	1.391196+11	1.39666E+11	1.91213€+11	1.99200E+11	1.89202E+11	1.90270E+11	1.90354E+11	1.926326+11	913076+11	883616+11	1 91738F+11	1 02 3076 411	1 034346+11		030566	11.300636.1	0.000000	1.950111111	1 . 93862E+11	1.939136+11	1.93845E+11	1.94521E+11	1.95668E+11	1.95010E+11	1.95640E+11	1.94946E+11	1.95410E+11	1.97164E+11	1.37036671	1.9.636611	1.97642F+11	1.99571E+11	1.99472E+11	1.99701E+11	.99827E	•	411516	.01616E	.02378E	.03786E	.03160E	•	04414E	2.059\$7E+11	•
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(100)		92.1	259.0	567.5	842.5	1251.2	95.4	259.7	£.665.	345.6	1266.3	0	900	-	0					7000	ָּיִנְיִי עייָע עייַע	23.5	263.0	- 18. - 18.	828.0	1233.1	93.9	204-20	521.9	800	1.1961	1.070	900	2.083	1321.3	96.6	272.3	7 C C C C C C C C C C C C C C C C C C C	שית המס	275.7	541.7	0.00 0.00	1349.5	CO	, y y	0.00	1362.1	100.0
Q d J h		535	635.0	635.8	638.0	635.e	635.e	621.1	621.1	621.1	1.15		100	100	7.69	7.7.		200	70	77.00	200	7.52.5	£79.4	579.4	579.4	579.4	565.6	565.6	565.6	50. 10. 10.	000 000 000 000	0.763	27.00	537.8	537.8	516.0	510.0	2.00	20.00 70.00	482.2	482.2	482.2	482.2	4.4.4	454.4	424.4	454.4	426.7
976	FACTOR	1935	1072	1970	1224	.2291	. 107A	. 1125	1223	1629	300				7766	1000	9.00	900	9000	2573.	7	E161.	5.53	.5338	.3077	.29%6.	. 2657	3800	3939	4.00	. A 350	- 6000	20.40	. 0205	.1320	.3113	. 2082	2011.	1.00	972	.0744	.0646	. <b>6</b> 598	97.00	0454	986	8256	.0434
	N/MKR2	43930F	51814	01326F	-28407E	.20661E	7.55\$71E+09	18431E	. 06538E	256035	351866	122011.		014364661	30000	1.3.5555.10	01.3010A+.7	30.00.00	1.5030 X 414	1.484166.10	1.705976+16	9.39576E-69	1.23596E+10	1.6219X+10	.73663E	2.20859E+10	. 0250cE	.25690E	1.955396+10	. C. 1.00 P. C.	.400/1E	436916	51046	.99624E	.14711E	.29711E	.76629E	301777	19176F	476136	.60431E	.96613E	.07779E	3.669368+10 4.663616410	97064	31095	4.45332E+10	3.89450E+10
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2.05574 2.06126 2.06126 2.08168 2.08368 1.0874 2.1074 1.2274 1.2275 1.1074 1.09
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800-00 80
24444 333742 533747 5937 5937 5937 5937 6937 6937 6937 6937 6937 6937 6937 6
50 4.19445E+10 51 4.23714E+10 52 4.53515E+10 54 4.7857E+10 55 4.58769E+10 55 4.88769E+10 55 1.14128E+10 59 1.14128E+10 59 1.06120E+10

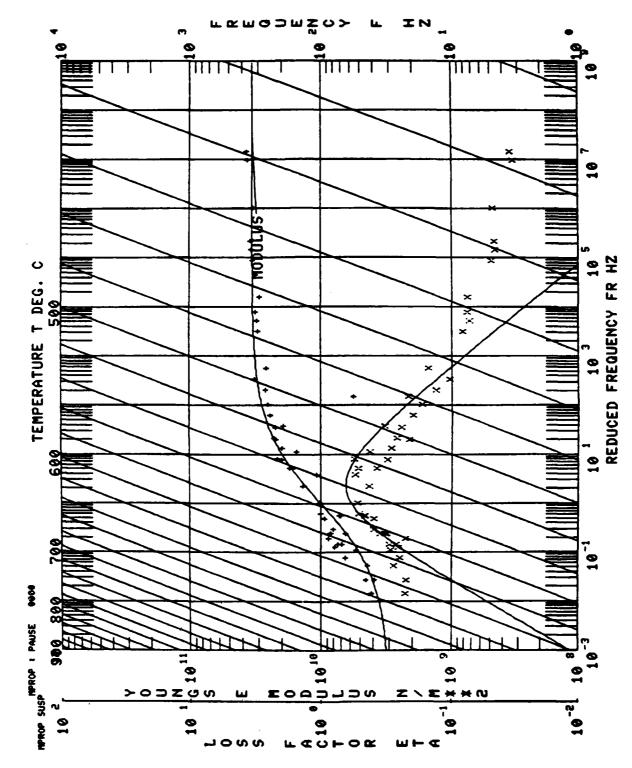


Figure 19. Reduced Nomogram for Composition X

TABLE 11

EXPERIMENTAL AND REDUCED DATA OF COMPOSITION X (AMB10)

MATERIAL :01-77-1; HAB10

DATA SOURCES

MANUFACTURER :UDRI BEAM COATED ONE SIDE

UDRI :UDRI
OTHER :NONE

LOSS MUD. N/B112 N/B112 6.21324E+09 5.561324E+09 3.42727E+09 9.39709E+09 1.65127E+09 5.5598E+09 6.51106E+09	3.64469 1.996666 1.99666666 1.996666666 1.696966669 1.8969676669 1.8796669 1.8796669 1.8796669 1.8796669 1.8796669 1.8796669 1.8796669 1.8796669 1.8796669 1.8796669 1.87966699 1.87966699 1.87966699	1. 255122 (0.92) 1. 25512 (0.92) 1. 25512 (0.92) 2. 3113 (0.92) 2. 3113 (0.92) 2. 3113 (0.92) 2. 3113 (0.92) 3. 35512 (0.92) 3	2.32864E+09
EE HT FPEO. H7 738.59 96.59 977.6 977.6 977.6 977.6 977.6	๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛		523.0
00800511E L055 FMC. 00173 00173 001576 00276 00161 00161 00161	99999999999999999999999999999999999999	a a a a a a a a a a a a a a a a a a a	.0072
E.M. McD. 1.52-6.11 1.52-6.26 1.52-6.26 1.52-6.26 1.52-6.26 1.4835.16 1.4835.16 1.4835.16 1.4835.16 1.531.55 1.531.	1.53515 1.53515 1.53515 1.553515	4.4.2.2.2.2.2.2.2.2.2.2.2.2.2.2.2.2.2.2	1.46706E+11
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TABLE 11 (continued)

50 8.61519E+09 .3292 621.1 547.1 5. 1.47196E+11 .0067 866.6 51 1.26069E+10 .1323 579.4 1286.6 6. 1.51041E+11 .0049 1308.0 52 8.25095E+09 .5200 621.1 1265.9 6. 1.43053E+11 .0131 1295.0 PEDIUC : PAUSE 0.000
.3292 621.1 547.1 5.1 .1323 579.4 1286.6 6.1 5.0 621.1 1265.9 6.1 1
.323 621.1 .1323 679.4 .5280 621.1
.323 621.1 .1323 679.4 .5280 621.1
. 3292 . 1323 . 5286 . 66
\$
50 8.61519E+09 51 1.26069E+10 52 8.25095E+09 REDUC : PAUSE AN

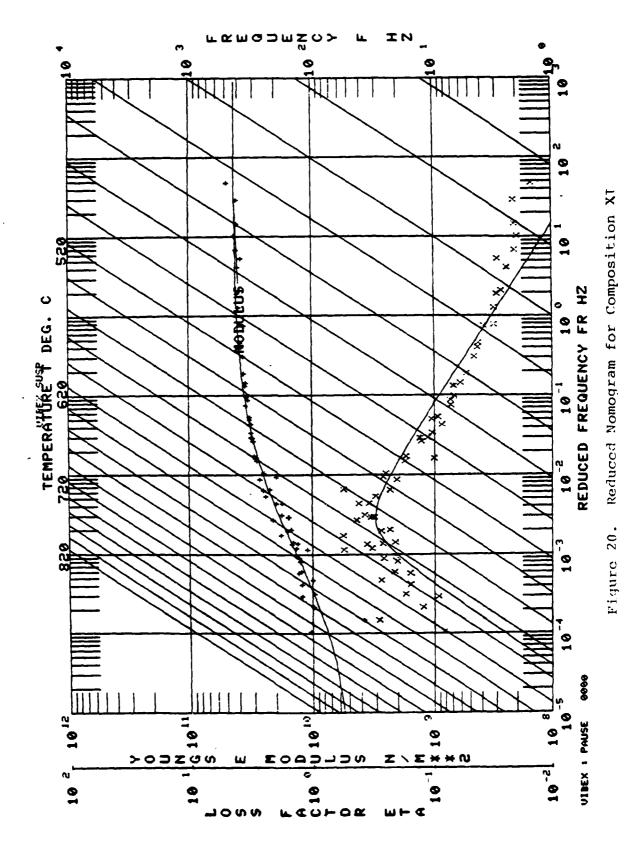


TABLE 12
EXPERIMENAL AND REDUCED DATA OF COMPOSITION XI (AMBI1)

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	5863E+19	Ė		9.00	'n	1.31998E+11		•	5.94763E+03	
	4970E+13	•	204.4	654°	<u>ښ</u> .	S2330E	.0040		1.19964E+09	
	3555E+19	•	4.4	2000	· u	1.56.34E+11	000		6.18633£+69	
	59436+13	•	104.4	1243.1		1.84764E+11	6016.		3.17506E+09	
٠,	5856E+03	•	690.6	90.06	പ്	1.83356E+11	.0041		1.11939€+09	
	4542E+10	•	690.6	255.9	m.	1.84889E+11	82.04·		2.01326E+09	
	38246+19	•	690.6	500.7	÷.	1.845828+11	6000.		2.70829E+09	
	49-56-16 . 03-16+16	•	900	7070	ri ce	1.86513F+11	5510		3.5.5.35.483	
•	0.2086+10	•	676.3	91.3	نمز	1.85175E+11	. 0045		1.284395+09	
_	5363E+15	•	676.7	256.7	<u>ښ</u>	1.85132€+11	5600.		2.75579€+09	
_	139946+10	•	676.7	584 0.00	÷.	1.85659E+11	.0157		4.57706E+09	
	2346E+10	•	֝֝֝֝֝֝֝֝֝֝֝֝֝ ֓֞֞֞֞֞֞֞֞֞֞֞֞֞֞֞֞֞֞֞֞֞֞֞֞	200	s	1.863938+11	2010		4.04.761.403	
	19386+11	•	- 0	0. CO	هٔ م	1.881025+11	2000		2.84206F+09	
_	3632E+10	366.1	0.00	259.6	im	1.87957E+11	.0184		5.47369E+09	
	2765E+10	0777	9	4.4.7	+1	88133E	78EO.		9.44760E+09	
•	\$083€+10	.3537	9.00	. 050 . 050 . 050 . 050	ហ់	1.89236E+11	.0230		7.07651E+09	
	4745E+10	5638	90	9.8/21 9.8/21	n is	1.986298+11			1.267128+18	
_	0911E+10	. 2687	90.09	257.9	in	1.86542E+11	.0120		3.51811E+09	
-	3186E+10	1976	662.8	509.7	+1	1.86450E+11	.0351		1.05532E+10	
	1136E+10	3372	662.8	9	Lin v	1.87921E+11	9182		5.43518E+09	
	2310E+10	186.5	621.1	100	ەم	1.085555 + 11	92150		6.380275+03	
•••	4994E+10	2344	621.1	521.1		9872E	0108		5.97622E+03	
	2557E+10	255.0	621.1	263.5	'n.	308816	. 6265		6.540338+09	
<b>u</b> -	26456+10	100 E	יינים איני בירנים איני	3.657 3.4.6	, n	1.936666+11 030876+11	2000 2000 2000 2000		3.00041F+60	
	3273E+10	.2074	593.3	268.2	imi	93934E	. 9175		S.66682E+09	
***	23036+10	.1733	593.3	523.6	÷	93651E	. 6159		5.23913E+09	
•••	14676+113	4000	7 7 7 7 7 7	90.0	ų,	1.946978+11			4.CU(44E+09	
	37826+13	3133	5.0.4	25. 25. 25.	نہو	1.949126+11	0837		7.63751E+09	
	94476+10	1300	579.4	269.6	mi-	1.942456+11	.6159		5.21147E+09	
-,,,,,	9162E+10	1921	4.000	222	÷	1.943366+11			3. (6143E+69	
, (-,	3492E+10	.0745	579.4	1322.1		1.96676E+11	- COS.		2.63415E+09	
	8679E+10	.2701	565.6	96.5	aic	1.99997E+11	. 0166 300		5.36575E+09	
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4-14	81725+19	. 0749	565.6	884.3	'n	1.96774E+11	9000		2.53203E+09	
44 U.B	5337E+19 953RE+10	. 6788 1924	565.6	1325.8	ώn	1.97,88E+11	. 66.4 14.4		2.51487E+69	
34-3	9201E+10	6690	537.8	27.5	in	1.301536+11	.0071		2.44207E+09	
4.14	6914E+19	. 6553	537.00	538.7	÷ı	1.97710E+11	990°	933.0	2.03000E+00	
- or or or	33 10E + 10 3893E + 16	. 04 ra	27.V	288. 200. 100.		1.387945+11	1000		1 15286E+69	
			?	3.677	ò	*********	B 1 2 2 2 .			

1.27406E+09	1.301535+09	1.24364E+09	1.45393E+09	1.08408E+09	1.01264E+09	8.97599€+08	8.75672E+08	1.15577E+09	8.85377E+08	3.10953E+08	7.66558E+08	1.69552E+03	2.18266E+09	
536.0	83 <b>3.</b> 0	1328.5	98.5	274.8	538.5	892.0	1334.0	38.7	276.0	541.6	587.0	273.4	97.7	
.0036	.0037	.0035	. Oë41	0600	3506.	.0025	+200.	. 0032	. A025	\$20 <b>@</b> .	. 6021	.0040	€90₽•	
1.99942E+11	2.00824E+11	2.021775+11	2.02050E+11	2.01311E+11	2.01512E+11	2.026376+11	2.03855E+11	2.041126+11	2.035776+11	2.041426+11	2.00372E+11	1.9976RE+11	1.93997E+11	
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543.6	2. 1.6°	1350.0	99.5	273.5	246.9	976.9	1358.3	G. 65.	<b>2</b> 80.4	550.5	915.8	276.7	93.5	
510.0	S10.9	510.0	482.2	482.2	482.5	482.2	482.3	424. <b>4</b>	4.4.4	<b>+2+.</b>	424.4	510.0	510.0	
. 0322	.0322	9365	4460.	2750.					. (213	. 0217	£310.	.0446	.0618	909
50 3.95221£+10	51 4.04168E+10	52 4.13828E+10	53 3.68508E+10	54 3.938796+10	55 4.09146E+10	56 4.18404E+10	57 4.35865E+10	58 3.87708E+10	59 4.15423€+10	68 4.20078E+10	61 5.01690E+1&	62 3.80219E+10	63 3.53224E+10	PEDUC : PAUSE & REPUC SUSP

## 3.1.4. Low Temperature Glasses

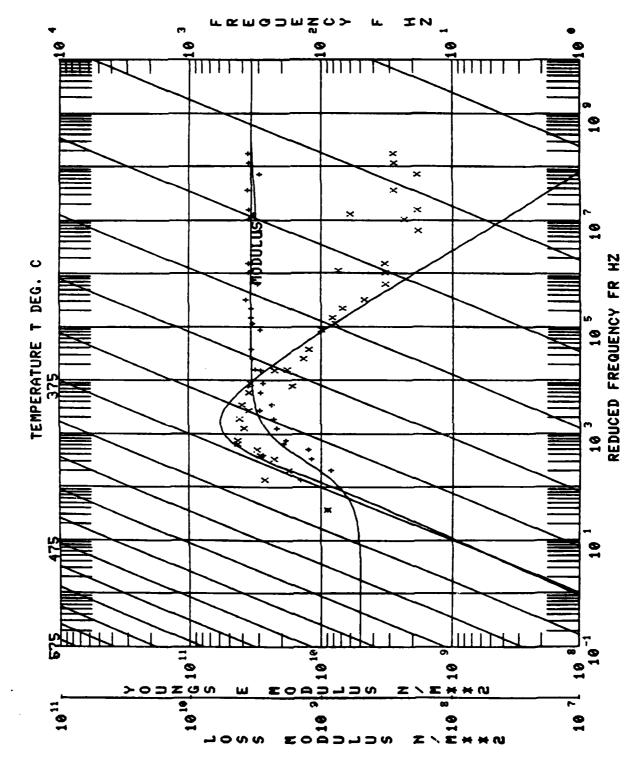
Six low melting materials were formulated for damping measurements. Compositions of these materials are shown in Table 13. Reduced nomograms for loss factor and loss modulus are shown in Figure 21 through 30, with experimental and reduced data in Tables 12,13,14,15, and 16. As expected, these compositions exhibit peak damping in the range of 600°F to 800°F (315°C to 427°C) All these compositions exhibited slight tendency towards devitrification; nonetheless, composition LT-3 appears to be the most stable composition. Contrary to expectation, composition LT-5 shows decreased loss factor and increased peak damping temperature as compared to LT-1 and LT-3. It is most probable that there has been a greater degree of devitrification in composition LT-5.

TABLE 13
COMPOSITIONS OF LOW TEMPERATURE GLASSES.

Ident. No.	PbO	PbF <sub>2</sub>	B <sub>2</sub> O <sub>3</sub>	2.0
LT-1	84.0		9.0	7.0
LT-2	80.4	3.6	9.0	7.0
LT-3	76.4	7.2	9.0	7.0
LT-4	73.2	10.8	9.0	7.0
LT-5	69.6	14.4	9.0	7.0
LT-6	66.0	18.0	9.0	7.0

The composition LT-2 exhibited flat loss factor vs frequency and temperature curves indicating that the composition has completely crystallized during the measurement.

These low temperature compositions were designed to cover the damping temperature range of 600°F to 800°F, which is normally considered to be the range of high temperature polymers.



Reduced Nomogram Displaying Loss Factor for LT-1. Figure 21.

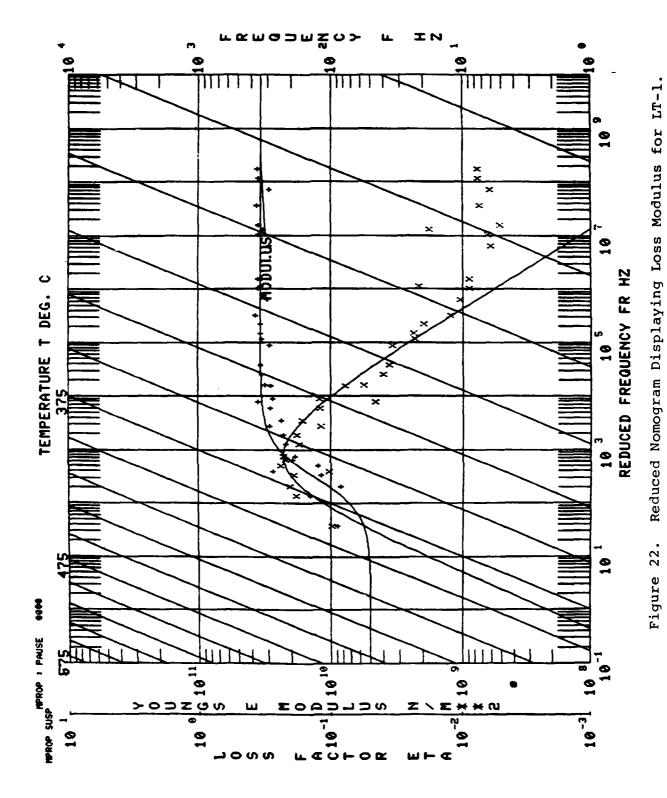
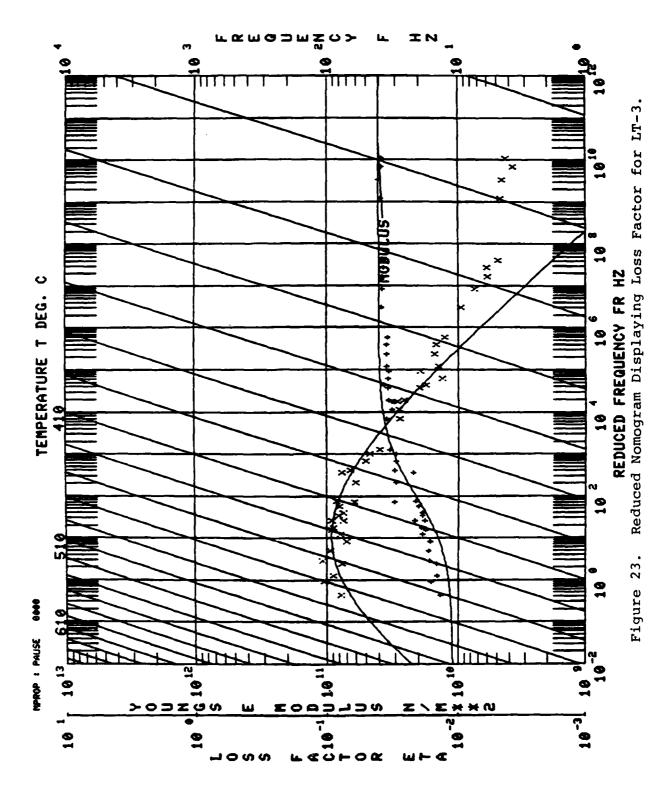


TABLE 14 EXPERIMENTAL AND REDUCED DATA OF COMPOSITION LT-1.

	FREG. HJDE BERN HOD. COMPOSITE EEAIT FEED. LOSS HOD. 1855.4 95.5 11123E11 5.00 96.1 2.	549.4 4. 2.1779£+11 .0022 556.7 1.
232 (Es		.0346 329.4
EXPERIMENTAL CODE: 232 MATERIAL: LT1; 015501 DATA SOURCE: UDRI: NONE OTHER: NONE	20	36 2.94098E+10



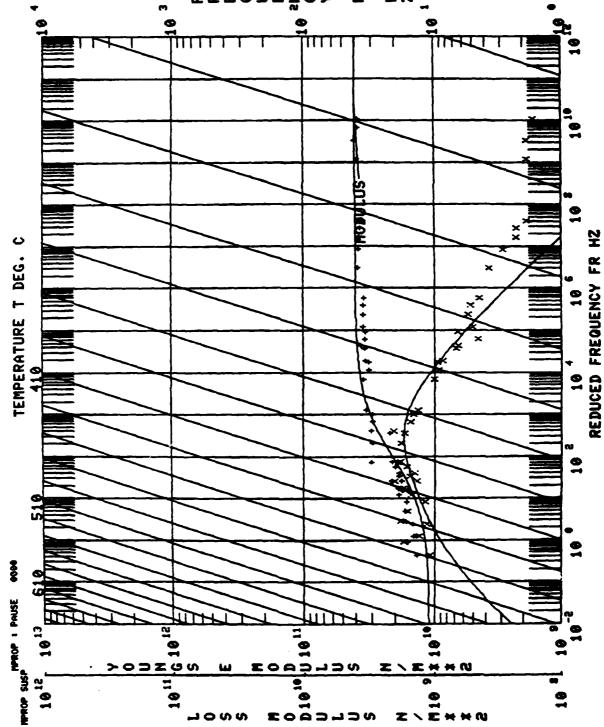
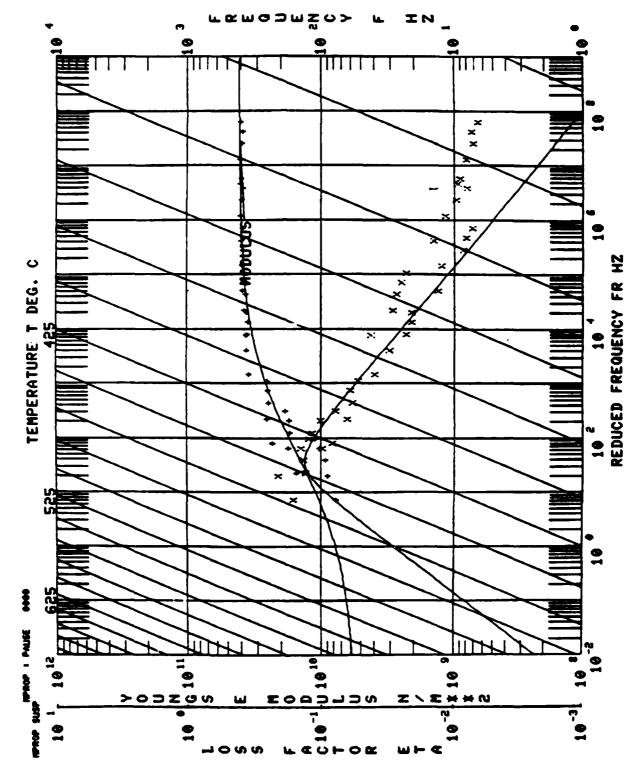


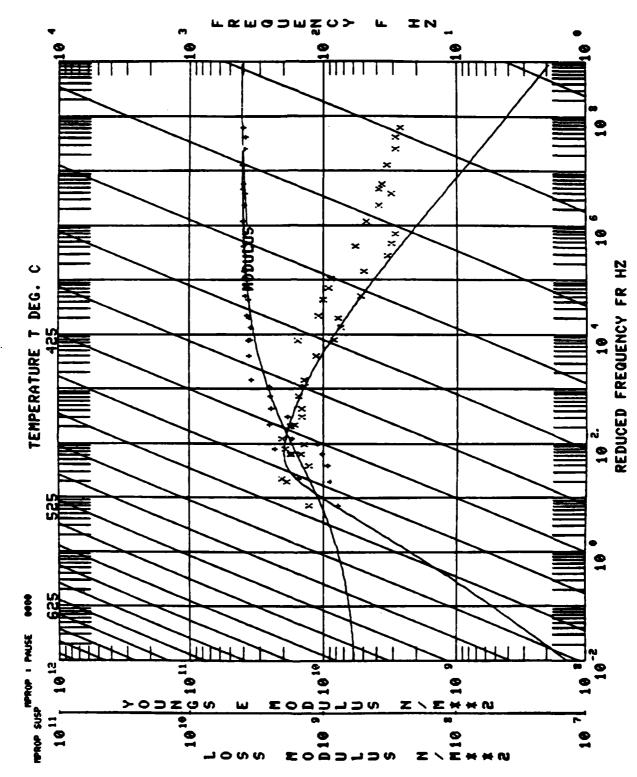
Figure 24.

Reduced Nomogram Displaying Loss Modulus for LT-3.

TABLE 15
EXPERIMENTAL AND REDUCED DATA OF COMPOSITION LT-3.
MATERIAL HUBLT 3
DATA SOUPCES
DATA SOUPCES
MANUFACTUREP HUBRI.01-84-01



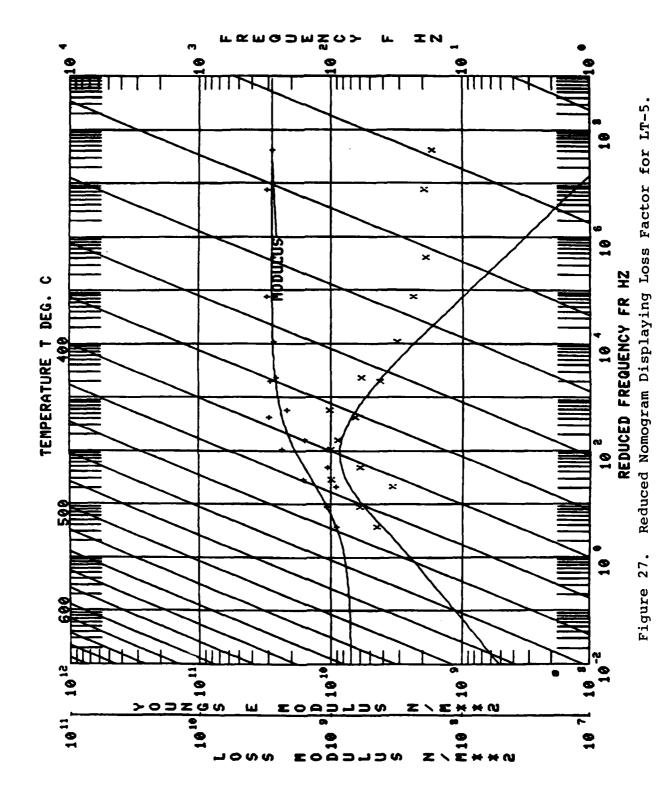
Nomogram for LT-4 with Modulus and Loss Factor Figure 25.

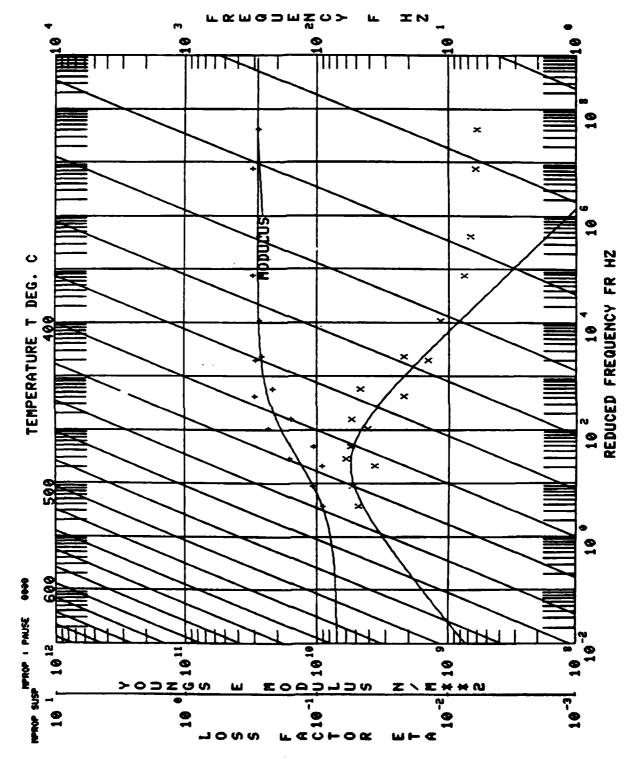


Nomogram for LT-4 with Loss Modulus and Modulus Figure 26.

TABLE 16
EXPERIMENTAL CODE :233
NATERIAL :UDLT4
DAT4 SOURCES
NAMEWINENTAL :UDLT4
DAT4 SOURCES
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	LOSS MOD.	1.30273E	1.92860E	1.31181E	I . Seeber	1.4645AF	A. C. C.	1.78128E	2.0770EE	1.86002E	1.48798F	1.973646	1 673155	102605	1.106006.1	1.55 ( Set 1	1.39436E4	1.4277364	1.155736	A 562575	7.5730KF	20105	20300	3000	302350	33000	312020	330166	2. 30419C	30000	302500	010 AFF	5.87158F	4.88877E	3.98491E	3.17725E	3.70727E	3.9369£	3.44242E+	2.94417E4	2.94592E+08	6. 1386 R	
	BEAM FREG.		274.2	539.5	9.00	1338.6	200	675.0	541.6	_	_					_	_		_	_		_	_		i٩	ė				344				2.00				=	7	~	912.5	9	
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	FREG. HZ	*	m	510.5	D	12/1.8	9.4	265.8	522.3	866.0	1298.0	3	1.00	ר בי	100	7000	1323.4	93.6	279.3	7.875	8	36.0				90.00	926	1 200 1		יינ ענייני ענייני	91710 01710	1270	191	233.2	2		2	161.6	284.0	556.9	922.2	1381.1	
	TENP.	154.4	4.4.4 4.4.4	454.4	7.7	424.4	9.0	440.6	443.6	9.07	140.6	4	1.75	7.00	- 1	40.0	426.7	308.0	308	305	800	900	) ( ) ( ) (	94		0.00°		900	1.1.0	:-			X	357.2	357.2	357.2	357.2	343.3			940. 0.000. 0.000.	343.3	
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:KUMAP 01-81-01 :3-9-82	MODULUS	8	60+36606P	44990E+09	.04695E+10	13203E+10	.5570ZE+10	. 78373E+10	75017E+16	76140E+18	87821F : 10	25,3716+10	CC 4005 1 14		91.36616	547914-10	59797E+10	52903€+10	6892F+10	614745+10	C6447F+16	£9427£+10	31.1555.10	01 20C11	9129556	011202021	41.716.44	014316 AG	41.30.100	87785F+18	BOIGEFILE	035676+10	044466+10	101676-10	97828E+10	913865+10	03746E+10	€1+3997.0	.14511E+10	\$799E+10	918598+10		
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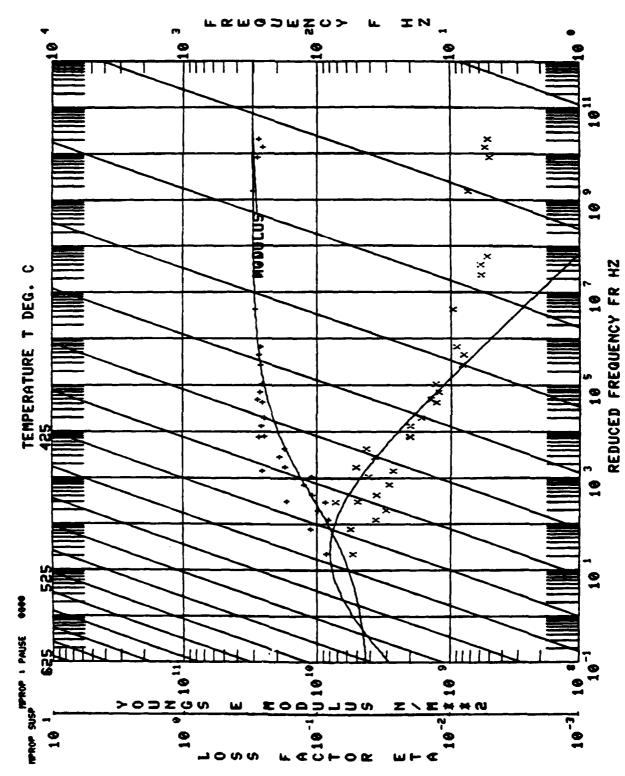




Reduced Nomogram Displaying Loss Modulus for LT-5. Figure 28.

TABLE 17
EXPERIMENTAL AND REDUCED DATA OF COMPOSITION LT-5.

LUSS MOD.	4.55626E+08	3.47235E+08	6.12404E+08	6.13204E+08	1.00954E+03	3.93560E+02	1.00301E+09	1.04189€+09	6.67731E+88	5.93974€+88	4.33212E+08	3.21465E+08	2.41754€+08	1.95143€+98	1.99325E+03	1.75032E+08		
BEAM FREG. HZ	98.9	542.5	99.0	644.0	99.S	5.45.5	<b>93.</b> ♥	547.5	9.66	549.0	18.0	5.055	100.3	853.0	106.7	555.5		
COMPOSITE	3500.	.0019	.0033	. 0033	500.	.0946	GT00.	1500.	.8031	. 0023	6500	.0015	.001	6000.	6000	8000		
EENN HOD. IV. NEB2	2.02129€+11	2.22420E+11	2.02948E+11	2.035416+11	2.03769E+11	2.04665E+11	2.045926+11	2.06169E+11	2.05416E+11	2.07340E+11	2.97111E+11	2.08434€+11	2.08313€+11	2.103325+11	2.09978E+11	2.12238E+11		
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FRES.	93.4	513.0	0.10	516.4	9.06	524.8	4	534.6	99.1	541.8	00	544.2	100.6	547.1	100.4	5-19.6		
TENP.	437.8	437.8	426.7	426.7	412.8	412.8	368.9	366	3.53E	325.0	371.1	371.1	M43.3	343.3	315.6	315.6		
LOSS	76F@	0756	5550	#564	E19	9558	9423	0474	A226	6223	412	9117	3,00	0000	.0063	6000	900	
MODULUS	0 22701F+80	0 302705 +30	1 12405F+10	1 ARECEF+10	#1+3695F9 1	CAPALF+10	C1+367785 C	2 100016+10	2 00757	5 56460F+10	2 07436+13	0 74774 C	41800E+10	2 81014F+10	3.14005E+10	2.83561E+10	: PAUSE	SUSP
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Nomogram for LT-6 with Modulus and Loss Factor Figure 29.

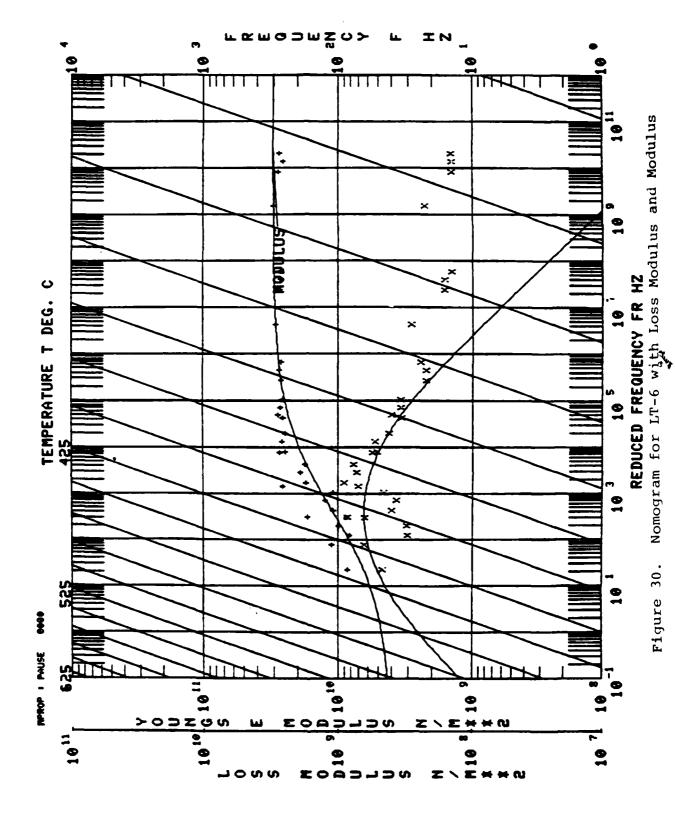


TABLE 18
EXPERIMENTAL CORE 1242
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DATH SOURCES
DAT

LOSS MOD. N/Mxr2	4.83861E+08	3.115985+08	3.14808E+08	6.5009E+08	6.56919F+08	4 403025400	1.0000000	3. 7. COC + 63	4.68364E+03	8.75523€+08	9.21947£+08	7.41892E+08	7.31170E+08	7.24359E+08	5.2000EE+03	5.44120E+08	4.24763€+08	5.67873E+03	3.43610E+08	3.45791E+08	3.47422E+03	4.09475E+68	2.24150E+08	2.25453E+08	2.47169E+08	2.94594£+98	1.64918E+08	1.65341E+08	1.45628E+08	2.31024E+08	1.45904E+08	1.45497E+08	1.47343E+08
BEAM FREG. HZ	98.5	542.0	804.0	1344.0	28.7	247.5	0.00	2000	1348.0	6.86 6.8	544.5	838.5	1351.5	99.1	546.0	901.0	1355.0	99.3	547.2	953.2	1358.0	₹.66	548.5	9.986 9.00	1361.0	a. 8	529.5	912.0	1367.0	100.1	552.5	920.5	1372.0
COMPOSITE EEAN F LOSS FAC. HZ	.0928	.0018	310%	.0037	. 0037	6600	1000	1000	9000	. 0042	. 0449	.0039	. 0241	.0036	.0026	. 0027	. 0021	8200.	. 0017	<b>.001</b>	. 6617	. 6020	.001	- 60 - 100	. 0012	. 0014	. 0008	8999	. 0007	.691	6000	000	.000
BENT MOD.	2.03235E+11	2.04332E+11	2.03496E+11	2. &6870E+11	2.042E3E+11	255755		C. 647446411	G. 68164E+11	2.04883E+11	2.06282E+11	2.05550E+11	2.09186E+11	2.05718€+11	2.07420E+11	2.066358+11	2.10271E+11	2.06550E+11	2.08333€+11	2.07706E+11	2.11203E+11	2.07174E+11	2.09324€+11	2.08996E+11	2.12137E+11	2.08635E+11	2.10853£+11	2.11773€+11	2. 14011E+11	2.09E91E+11	2.12388E+11	2.15739E+11	2.15580E+11
M¢₽E 110.	'n	÷	ŝ	Ġ	ก่	-	ŕL	'n	۰	က်	4	'n	ė	'n	÷	'n	ė	'n	÷	'n	ۇ.	તાં	÷	'n	ف	ai	÷	Į,	6	'n	+	v	9
FREG. H2	m		845.6	1256.A	93.8	A 10		100	1278.2	92.4	525.7	871.8	1304.2	9.7.00	537.2	883.8	1332.1	. 98 . 3	539.5	833.3	1337.7	98.7	241.4	396.5	1342.8	99.5	5.44.0	901.0	1348.3	29.7	546.3	905.4	1355.6
TEMP.	440.6	440.6	440.6	440.6	126.7	7.00		- 1	- 92	412.8	412.8	412.8	412.8	398.9	398.9	336.9	398.3	385. W	385.0	385.0	385.0	371.1	371.1	371.1	371.1	343.3	343.3	343.3	343.3	315.6	315.6	315.6	315.6
LOSS	. 0551	.0371	.0312	6735	27.7	3200	7070	2003	.0410	6550°	0550	.0330	0439	0275	. AE 07	. 4205	.4163	9550	. 0132	.0127	.0133	++10·	<b>9884</b>	1900.	C620.	8600.	9000	0000	1900.	. 0075	.0052	٠	5
MODULUS N.Mare	78336E+09	8 39629F+09	. 00331E+10	8.88374E+09	15128F+10	1.101E01.10	1.110011.1	1.282836+10	1.12158E+10	1.71992E+10	1.773048+10	1.95435E+10	1.77939E+10	61439E+10	517896+10	65885E+14	@1+36861S	753076+19	2.61265E+10	73205E+10	61213€+10	8493SE+10	67138E+10	75867E+10	67115E+10	96226E+10	74986E+10	73763€+10	71706E+10	96467E+10	82333£+1¢	2.60736E+10	79342E+18
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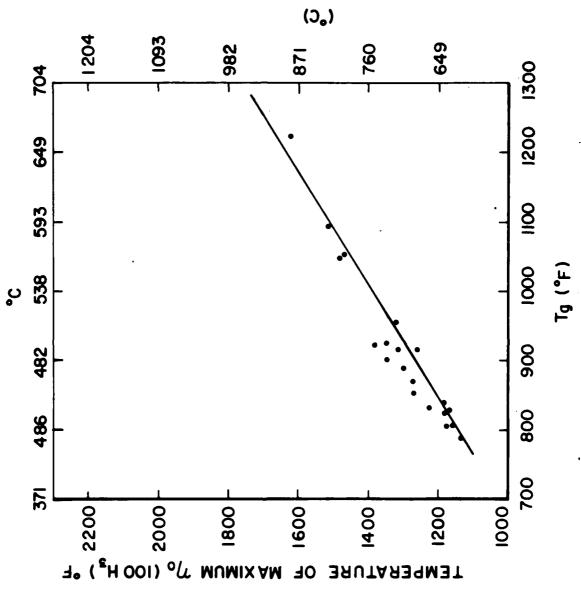
## 3.2 RELATIONSHIP BETWEEN THE DAMPING TEMPERATURE AND GLASS TRANSITION TEMPERATURE

Figure 31 shows a plot of the maximum loss factor at 100 Hz versus the glass transition temperature ( $T_g$ ) of test matrix glasses whose damping properties were reported earlier (10). The glass transition temperatures for the compositions are also shown in Table 19. It is obvious from the figure that there is a linear relationship between the loss factor peak temperature and  $T_g$ . Regression analysis of the data provided a relationship between the damping peak temperature,  $T_{\eta}$  at 100 Hz and the glass transition temperature as expressed by equation (6). The correlation coefficient for this relationship is 0.96.

$$T_{\eta} = 166.02 + 1.10T_{g} \tag{6}$$
 where:  $T_{\eta}$  and  $T_{g}$  are in °C.

The relationship is useful in predicting the damping peak temperature if the glass transition temperature is known. This suggests that the loss factor peak temperature  $\mathbf{T}_{\eta}$  is determined by the viscosity of the glass just as  $\mathbf{T}_{g}$  corresponds to a given viscosity (approximately  $10^{13}$  poise). Also, increasing  $\mathbf{T}_{g}$  indicates general stiffening of the glass structure and, therefore, one can expect a decrease in the  $\eta_{D}$  peak height. However, an attempt to correlate the  $\eta_{D}$  peak height with the glass transition temperature was not successful.

An estimate of the viscosity corresponding to the  $^{\eta}D$  peak temperature can be made because the viscosity-temperature relationship of the Corning 0010 glass is known. Figure 32 shows the viscosity-temperature relationship of the glass. The loss factor peak temperature for the glass at 100 Hz is 650°C, which corresponds to a viscosity of  $^{10}$  poise. Similarly, at 1000 Hz, the loss factor peak corresponds to a viscosity of  $^{10}$  poise. It is, therefore, noted that the observed damping occurs when the glass temperature is near the softening point (viscosity =  $^{10}$ . Also, the damping temperature is about 200°C above the glass transition temperature of the



Maximum Loss Factor ( $\eta_D)$  Temperature Versus the Glass Transition Temperature  $(\eta_g)$  . Figure 31.

Specimen Number	T <sub>g</sub> (°C)
м30	429
Ml	532
M36	601
M2	441
M19	408
M17	491
M29	583
M49	475
M32	568
M26	441
M12	496
M21	662
M35	466
M24	494
Mll	489
M22	511
M14	447
M48	421
M5	456
M27	482
M41	541
M31	466
M4	590
M45	491
M46	564
м6	447
м7	431

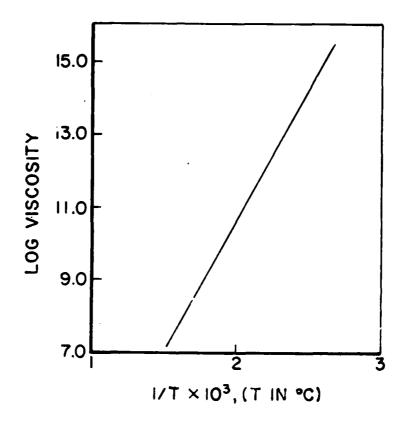


Figure 32. Log Viscosity Versus 1/T for Corning 0010 Glass (viscosity in poise).

Corning 0010 glass is approximately 430°C. At the damping temperature, the glass can be characterized as a viscous, rather than a viscoelastic, material.

## 3.3 EFFECT OF NUCLEATION AND CRYSTALLIZATION ON THE VISCOELASTIC DAMPING

In the previous section, it has been shown that the viscosity of the vitreous enamel ranges from  $10^6$  to  $10^7$  poise in the damping range. The viscosity range is favorable for processes like nucleation and crystallization. It has also been observed that when a vitreous enamel is aged in the damping range, generally there is a reduction in the loss factor  $(Q^{-1})$ . The structural mechanisms leading to the reduction are not understood. It has been hypothesized that the reductions is related to the devitrification, but no experimental data exist to support this hypothesis. In view of this lack of understanding of the effects of thermal aging on the damping properties, this work was undertaken.

Figure 33 illustrates a typical viscosity versus temperature relationship for silicate glasses. The viscoelastice damping range is indicated by a shaded area in the figure. Included schematically in the figure are nucleation and growth rate curves for silicate glasses. It should be noted that the damping range approximately coincides with the crystal growth range. A glass coating containing nucleating sites would be expected to crystallize in the damping range. This assumption can be justified if the real situation is examined. These vitreous enamel coatings are generally applied on superalloys substrates. Diffusion of the metallic ions into the coating can enhance the nucleation and growth processes. Prolonged exposure of the coating in the damping range would be expected to transform the vitreous enamel coating into a crystalline coating. A consequence of the transformation would be a significant change in the damping properties.

In view of the background provided above, two glass compositions were thermally aged in the damping temperature range.

Their damping properties and microstructural development were

characterized as a function of heat treatment time. Optical microscope, scanning electron microscope (SEM), and x-ray diffraction were the primary analytical tools for microstructural characterization.

Compositions of the two glasses designated as A and B with the heat treatment conditions are shown in Table 20. Composition A is  ${\rm Al_2O_3}$  free, whereas composition B contains 5.70 percent  ${\rm Al_2O_3}$ . Heat treatment temperatures approximately correspond to the respective peak damping temperatures. Heat treatment times were arbitrarily chosen.

TABLE 20 COMPOSTIONS IN WEIGHT PERCENT

	A	Heat Treatment Conditions	В	Heat Treatment Conditions
Sio <sub>2</sub>	74.65	at 760°C	70.40	at 815°C
Al <sub>2</sub> O <sub>3</sub>		I no heat treatment	5.70	I no heat treatment
Na <sub>2</sub> O	12.78	II 100 hours	12.05	II 112 hours
CaO	10.77	III 314 hours	10.15	III 309 hours
C00	1.80		1.70	

The loss factor of these glasses as a function of temperature and heat treatment are shown in Figures 34 and 35. The loss factor decreased with increasing heat treatment time near the peak damping for both compositions. Composition B, in general, has a lower loss factor than composition A. This can be explained on the basis of the  $\mathrm{Al}_2\mathrm{O}_3$  concentration in the glass. It has been shown that the addition of  $\mathrm{Al}_2\mathrm{O}_3$  decreases the loss factor (6) significantly.

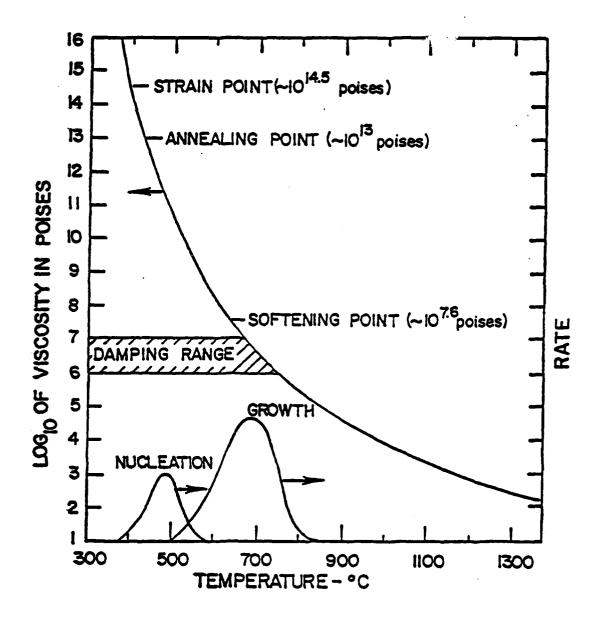


Figure 33. Typical Viscosity Curve for Glass Showing Common Reference Points.

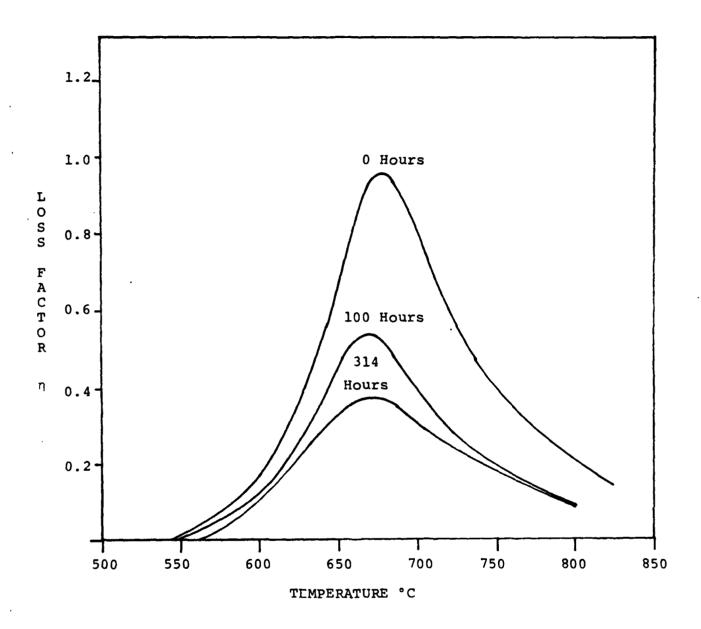
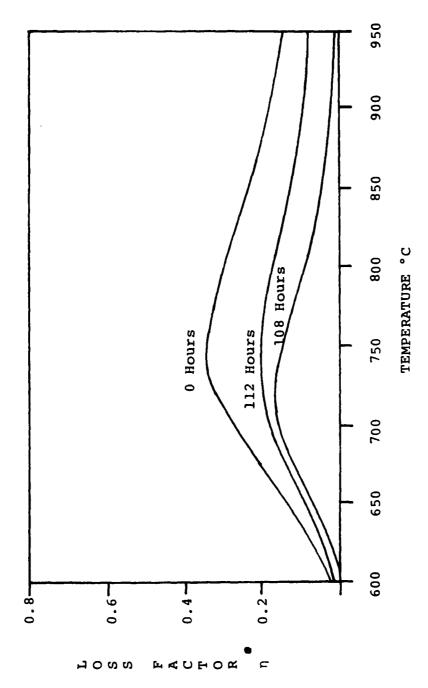


Figure 34. Loss Factor as a Function of Temperature and Heat Treatment Time at 760°C for Composition A.



Loss Factor as a Function of Temperature and Heat Treatment Time at 815°C for Composition B. Figure 35.

Figure 36 shows the optical micrographs of the enamel-metal interface of "as-fired" and "heat treated" (112 hours at 815°F or 435°C) specimens obtained from composition B. Both specimens exhibited seeds and bubbles at the interface. No crystalline phase or microstructure was present in the as-fired enamel coating. However, after the heat treatment, a crystalline phase in the vitreous enamel coating appeared. The crystalline phase was noted to originate from the seeds and bubbles in most instances, and appears in the micrograph as elongated needles. There was also a marked variation in the grain sized distribution of the metal at the interface. It appears that the metallic grains at the interface have been fragmented into smaller sizes due to chemical reaction with the vitreous enamel coating. The fragmentation of the metallic grain is much more pronounced after heat treatment (Figure 36b).

The evidence of crystallization from the optical microscopy and possibility of glass-metal interaction at the interface promoted further microscopic examination. Several specimens were examined under SEM and the observations are decribed in the following paragraphs.

Figures 37 shows the glass-metal interface for composition A after various heat treatments. All these micrographs have similar magnification. It should be noted that the fragmentation of the metallic grains at the interface increases with increasing heat treatment time. Also the degree of crystallization of the vitreous enamel coating increases with progressive heat treatment. No change in the size and distribution of the pores at the interface was noted.

Figure 38 shows morphology of the vitreous enamel coating "A" at varying magnifications after a heat treatment of 100 hours. The appearance of crystals of hexagonal habit can be noted in micrographs 38b and 38c.

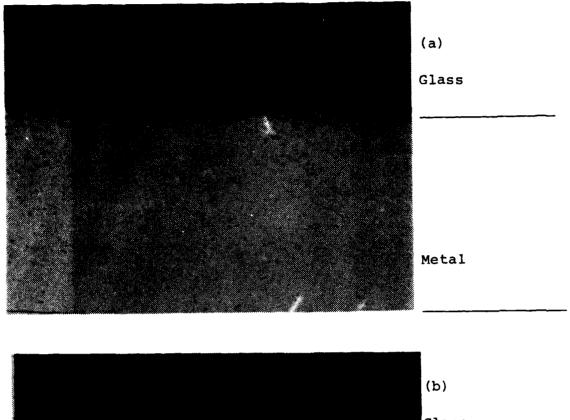
Figure 39 demonstrates the microstructure of the vitreous enamel coating "A" at varying magnifications after 314 hours of heat treatment. As expected, the degree of crystallinity has increased when compared to the 100 hour specimen (Figure 38a, 38b, 38c).

Figure 40 shows the metal-glass interface for composition "B" as a function of heat treatment time. Fragmentation of the metallic grains at the interface is similar to the one observed in Figure 37. There is no evidence of crystallinity in the vitreous enamel coating in the as-fired specimen (Figure 40a). After heat treatment the coating crystallized as evidenced by micrographs 40b and 40c.

Figure 41 shows the enamel coating for the composition "B" at various magnifications after heat treatment for 112 hours. All three micrographs show the existence of a predominately glassy phase. After heat treatment for 309 hours, there is a significant increase in the degree of crystallinity, as shown in Figure 42. However, in this case, characteristics of the crystals appear to be different from those of the composition "A" for similar heat treatment (Figure 39). The crystalline phase is present as elongated needles rather than in a hexagonal habit.

The crystalline phases for both compositions which appeared after heat treatment were identified by x-ray diffraction. Table 21 summarizes the x-ray diffraction results. After heat treatment, composition "A" developed  $\alpha$ -quartz and devitrite (Na<sub>2</sub>O·3CaO·6SiO<sub>2</sub>) crystals;  $\alpha$ -quartz being the major crystalline phase. The hexagonal crystals shown in Figures 38 and 39 can now be identified as  $\alpha$ -quartz. After heat treatments of composition "B",  $\alpha$ -cristobalite and devitrite (Na<sub>2</sub>O·3CaO·6SiO<sub>2</sub>) appear as crystalline phases;  $\alpha$ -cristobalite being the major crystalline phase. The elongated crystals observed in Figure 42 are therefore  $\alpha$ -cristobalite.

Prolonged heat treatment of the vitreous enamel coatings
"A" and "B" in the viscoelastic range progressively reduced the
loss factors as shown in Figures 2 and 3. Microstructural analysis
by optical microscopy, SEM, and x-ray diffraction revealed
crystallization of the coatings. The degree of crystallinity for
both compositions was related to the heat treatment time. It is
suggested that the crystallization of the glass coatings is
primarily responsible for the reduced loss factor. This suggestion
can further be justified by considering the fundamental origin of



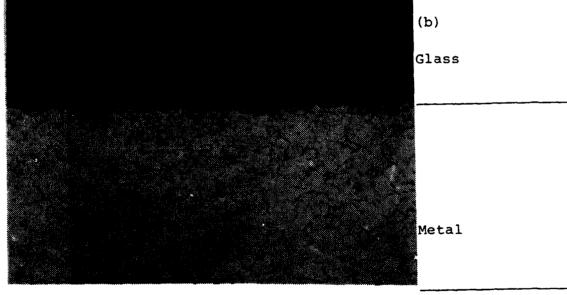
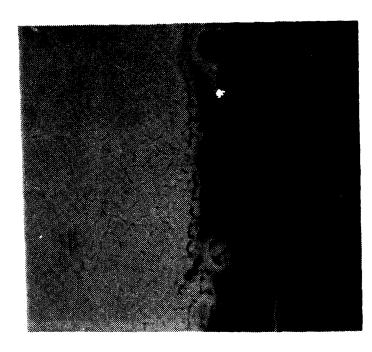
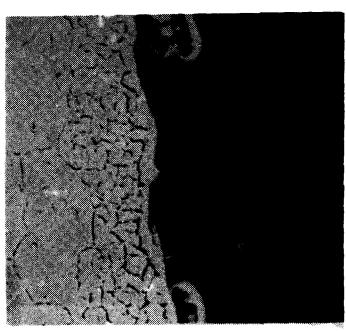


Figure 36. Optical Micrographs: (a) As Fired Specimen (150X); (b) Heat Treated for 112 Hours at 1,500°F (150X).

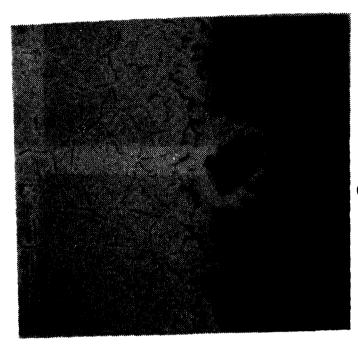


(a) As Fired (1000X)



(b) At 100 Hours (1000X)

Figure 37. Glass-Metal Interface for Composition "A" After Various Heat Treatments (continued).



(c) At 314 Hours (1000X)

Figure 37. Glass-Metal Interface for Composition "A" After Various Heat Treatments (concluded).

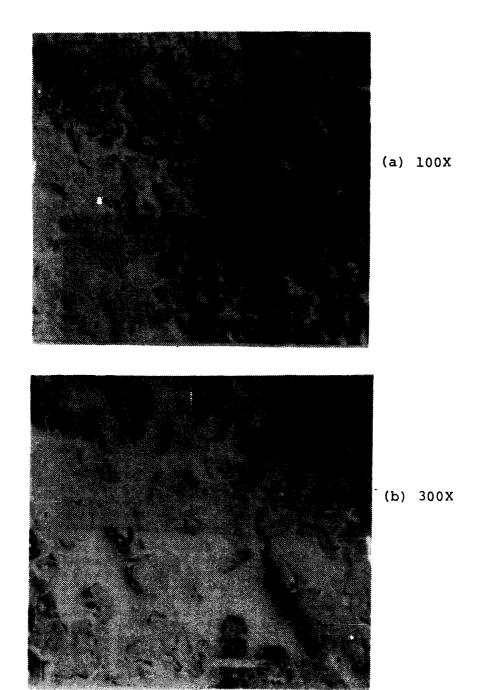


Figure 38. Morphology of Vitreous Enamel Coating "A" After Heat Treatment of 100 Hours (continued).

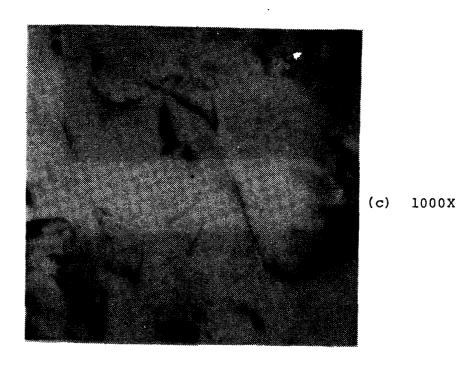
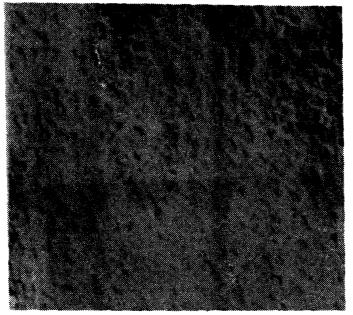
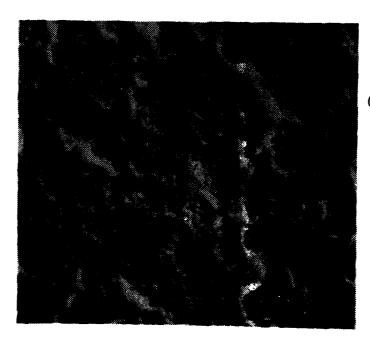


Figure 38. Morphology of Vitreous Enamel Coating "A" After Heat Treatment of 100 Hours (concluded).



(a) 100X



(b) 300X

Figure 39. Microstructure of Vitreous Enamel "A" After 314 Hours of Heat Treatment (continued).

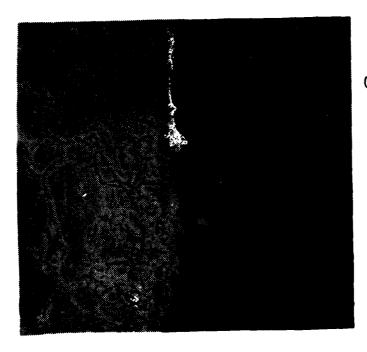


(c) 1000X

Figure 39. Microstructure of Vitreous Enamel "A" After 314 Hours of Heat Treatment (concluded).



(a) As Fired (1000X)



(b) At 112 Hours (1000X)

Figure 40. Metal-Glass Interface for Composition "B" as a Function of Heat Treatment Time (continued).

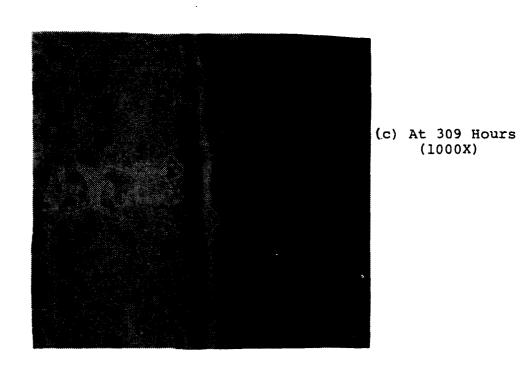


Figure 40. Metal-Glass Interface for Composition "B" as a Function of HEat Treatment Time (concluded).

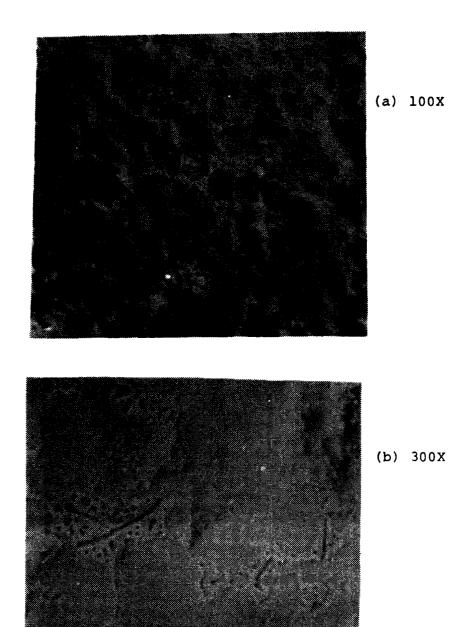


Figure 41. Enamel Coating "B" at Various Magnifications After Heat Treatment for 112 Hours (continued).

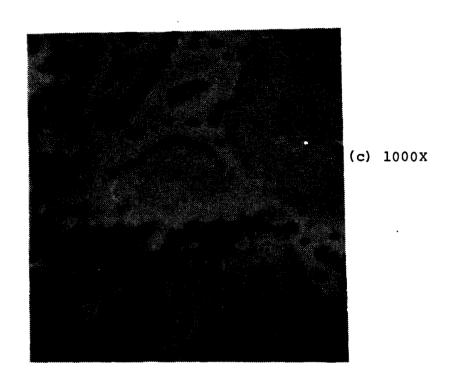
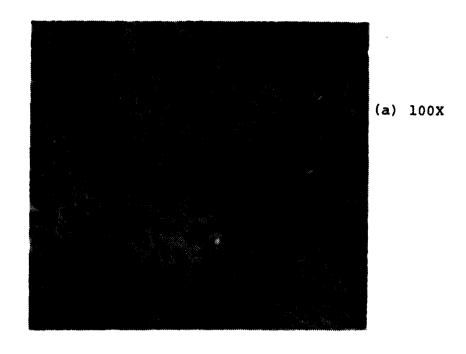


Figure 41. Enamel Coating "B" at Various Magnifications After Heat Treatment for 112 Hours (concluded).



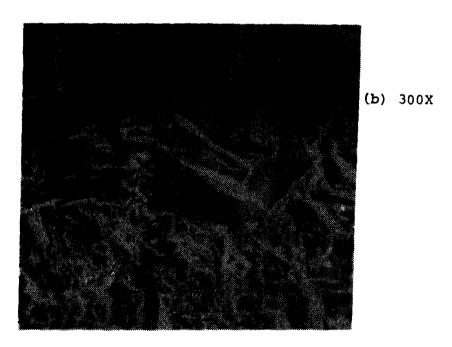


Figure 42. Enamel Coating "B" at Various Magnification After Heat Treatment for 309 Hours (continued).

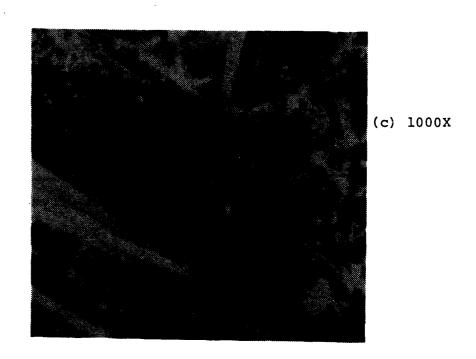


Figure 42. Enamel Coating "B" at Various Magnifications After Heat Treatment for 309 Hours (concluded).

TABLE 21
CRYSTALLINE PHASES AFTER VARIOUS HEAT TREATMENTS

Compositions	Hea	t Treatment	Crystalline Phases
A	I	as fired	no crystalline phase
	II	heat treated at 760°C for 100 hours	major - α-quartz minor - devitrite
	III	heat treated at 760°C for 314 hours	major - α-quartz minor - devitrite
В	I	as fired	no crystalline phase
	II	heat treated at 815°C for 112 hours	major - a-cristobalite minor - possibly devitrite and mullite
-	III	heat treated at 815°C for 309 hours	major - α-cristobalite minor - possibly devitrite

a-quartz - hexagonal
a-cristobalite - tetragonal

the viscoelastic damping. The viscoelastic damping originates from the vitreous nature of a material in the proper viscosity range. Any deviation from the vitreous nature would affect the damping properties. Crystallization of the coating is such a deviation where the amount of a vitreous material is progressively reduced. Number of relaxation units and sites which give rise to the damping are directly proportional to the amount of vitreous material in the coating. Crystallization reduces the number of relaxation units and sites and therefore the loss factor.

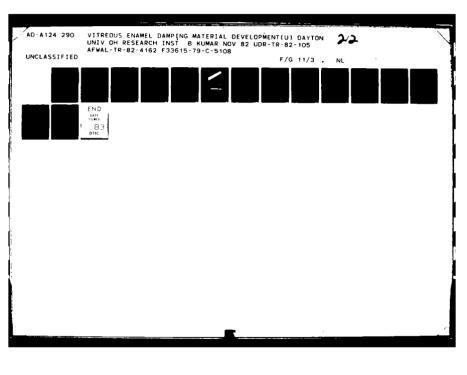
One would expect a shift in the loss factor peak temperatures of Figures 34 and 35 with progressive crystallization. With crystallization the viscosity of the residual glass would change and therefore shift the damping peak. No shift in the peak temperatures would indicate that the viscosity of the residual glass was similar to the parent glass. Compositions used in this investigation provide such a situation. During crystallization crystalline  $\mathrm{SiO}_2$  ( $\alpha$ -quartz and cristobalite were major phases for both compositions) precipitate thereby making the residual glass richer in  $\mathrm{Na}_2\mathrm{O}$  and  $\mathrm{CaO}$ .  $\mathrm{Na}_2\mathrm{O}$  would make the glass softer whereas  $\mathrm{CaO}$  has the opposite effect and the overall effect being neutral. No shift in the loss factor peak temperatures (Figures 33 and 34) suggest a neutral effect of the crystallization on the viscosity.

The general observation on the microstructural development suggests that increasing heat treatment or aging time leads to a progressive increase in the crystallinity of the vitreous enamel coating. There is also a significant microstructural change of the metallic substrate microstructure. These two processes independently or in combination alter the damping properties. However, it is well known that metals have much lower damping than the vitreous enamel at these temperatures, and any microstructural change in the metallic substrate would not be expected to influence the damping properties significantly. Therefore, the variations in the damping data (as shown in Figures 34 and 35) must be related to the microstructural change in the vitreous enamel coating. This leads to the conclusion that the reduction in damping capacity of

the enamel coating is related to the increasing degree of crystallinity. This does not mean that vitreous enamel coatings cannot be used as a vibration damping material. However, care must be taken when developing and utilizing enamel compositions in view of the fact that the conditions during the viscoelastic damping are favorable for nucleation and growth processes.

## 3.4. CONSTRAINED LAYER DAMPING WITH VITREOUS ENAMEL

For many years, vitreous enamels have been known to exhibit high loss modulus in the transition range. As a result, many of the research and application efforts to date that have utilized vitreous enamels as vibration damping materials have been directed toward free layer damping designs. In previous publications (6, 11, and 12), the viscosity temperature relationship of a vitreous enamel in the damping range and its influence on the damping properties had been discussed. Areas of concern have developed after consideration of viscosity within the damping range and practical application. Some areas of condern are; flow under gravity and centrifugal force, weathering and reaction with atmospheric gases, and volatilization of components from the vitreous enamel. All of these potential problems can be controlled by applying a protective layer over the damping material. A protective layer over an enamel damping layer has been used successfully in several applications (2, 3, and 13). In cases where a protective layer was needed, an interesting phenomenon was recorded. Enamel damping materials in a constrained layer damping design exhibited an increased temperature range of effective system damping in contrast to a free layer damping system. In addition to this gain, the handling of vitreous enamel constraining layer damping designs would be less cumbersome as the enamel could be coated onto a constraining layer of desired size and shape. The enamel could then be soldered or spot welded into place. Several constrained layer vitreous enamel treatments were evaluated both to better view their advantages and to determine the mechanism involved. discusses the results and mechanisms involved in the vitreous enamel constrained layer damping treatment.





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The damping material used in this experiment was Corning composition 1990\*. The 1990 was applied to: 1) test beams in partial and full coverage (see Figure 43); and 2) iron and stainless steel foils. All foil applications were done using the plasma spray technique. The coated constraining layers were welded onto cantilever beams. A photograph of the coated iron foil and the composite beam is shown in Figure 44. The damping of the four composite beams was determined using the standard resonant beam technique described in Reference . The experimental results are presented in Table 22. Modal loss factors were determined by the half power bandwidth method. Verification test measurements were carried out on the free layer and iron constrained specimens. Reproducibility was excellent at lower temperatures. At high temperatures (above 700°C), there were minor discrepancies which are believed to be related to the devitrification of the vitreous enamel. Oxidation of the iron foil was noted during the damping tests at elevated temperature. The stainless steel foil exhibited little or no oxidation under similar conditions.

The modal loss factor versus temperature plots for a free layer damping treatment, a constrained layer damping treatment, and the bare beam itself, are shown in Figure 45. The free layer treatment shows a loss factor of .018 at 480°C, whereas the constrained layer treatment exhibits a loss factor of .0318 at 710°C. The temperature range where the modal loss factor  $(\eta_{\rm S})$  is above .01 for the free layer treatment is 440°C to 525°C. The constrained layer treatment has  $\eta_{\rm S}$  of .01 from 475°C to 850°C where the experiment was terminated. The temperature range of effective damping for the constrained layer treatment is of significant interest. Although a broadening of the modal damping peak with the constrained layer polymer materials is anticipated, a peak broadening which extends several hundred degrees centigrade is unusual. Analysis of the origin of broadness will be discussed in detail in the following paragraphs.

Like the polymeric materials, vitreous enamels span three distinct modulus regions with increasing temperature. However, unlike the polymeric materials which have a rubbery region above

<sup>\*</sup>SiO<sub>2</sub>-41, Li<sub>2</sub>O-2, Na<sub>2</sub>O-5, K<sub>2</sub>O-12, PbO-40

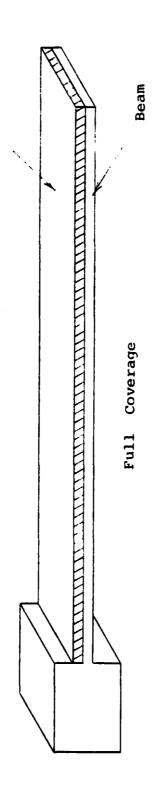
TABLE 22 EXPERIMENTAL TEST MATRIX

Beam Configuration	Damping Material Thickness	Constraining Layer Thickness	Maximum System n	Temperature Range Where $\eta_s \geq .01$
Free layer beam full coated	.031 cm		.025	427°-538°C
Free layer beam partially coated	.031 cm		.019	432°-524°C
Iron constrained layer beam	.023 cm	.013 cm	.032	480°->840°C
Stainless steel constrained layer beam	.028 cm	.013 cm	.055	470°->850°C

Tg, the vitreous enamel becomes a liquid when heated above Tg. A typical modulus and loss factor versus temperature plot for a viscoelastic material is illustrated in Figure 46. Region I is the design range for a free layer damping treatment. Region II is the design range for a constrained layer damping treatment. The mechanisms of damping for these two regions are well understood. In polymeric materials, Region III is the rubbery region and is the design region for tuned viscoelastic dampers. A proposed mechanism in enamels above Tg follows.

The broad damping peak of the constrained layer system shown in Figure 45 covers all three regions. The addition of a constraining layer to a free layer damping material should result in a decreased effectiveness of the damping system in the temperature region of Figure 46 as a result of the reduction in extensional deformation in the damping material. This is demonstrated in the test results in Figure 45. The free layer effectivenees is reduced, but the materila loss modulus is sufficiently high to contribute significant damping at approximately 475°C. As the temperature continues to increase, the material loss factor and modulus begin to decrease. Ordinarily, this would result in decreased system damping as shown in the free layer curve in Figure 45; however, a new phenomenon now takes control of the damping. As the modulus begins to drop, significant shear caused by the constraining layer begins to be established in the damping layer. As a result, the system damping continues to increase with temperature to a peak value. This damping trend continues throughout the temperature range of the experiment. A review of the characteristics of vitreous enamels above Tg sheds light on this phenomenon.

Most of the vitreous enamels behave like Newtonian liquid at temperature beyond the glass transition range. The static viscosity of the enamel decreases continuously as the temperature is increased. The static viscosity for the Corning 1990 glass in the temperature range of interest exhibits typical behavior and is shown in Figure 47. The rate of decrease is higher at lower temperatures than those at the high temperatures.



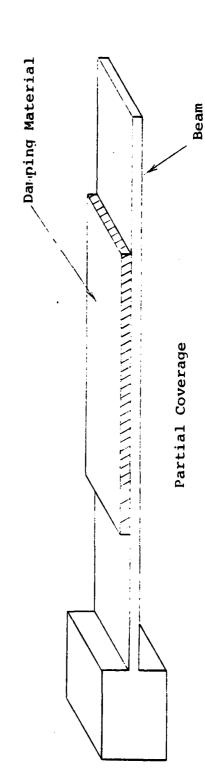


Figure 43 Test Beams for Corning Composition 1990.

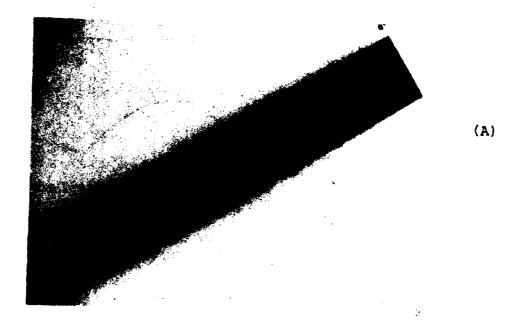




Figure 44. (A) Vitreous enamel coated iron foil. (B) Spot welded iron foil on beam.

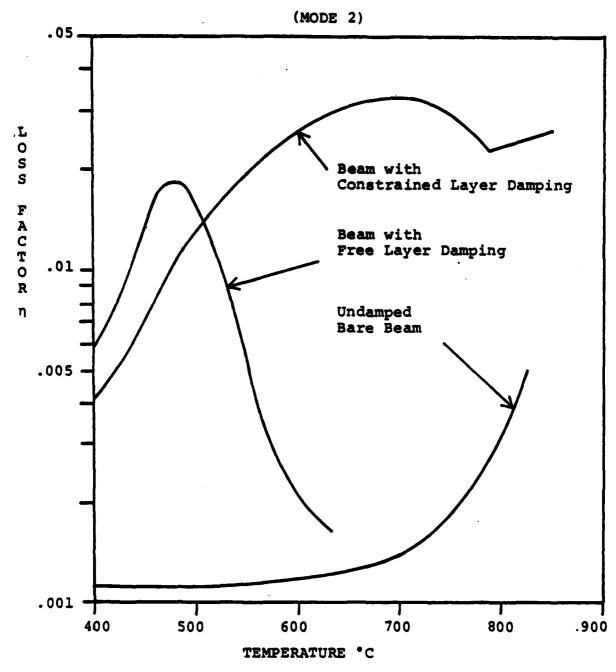


Figure 45. Loss Factor Versus Temperature Comparison for Undamped, Free Layer Damped, and Constrained Layer Damped.

TEMPERATURE

Figure 46. Typical Temperature Dependence of Polymer Viscoelasticity.

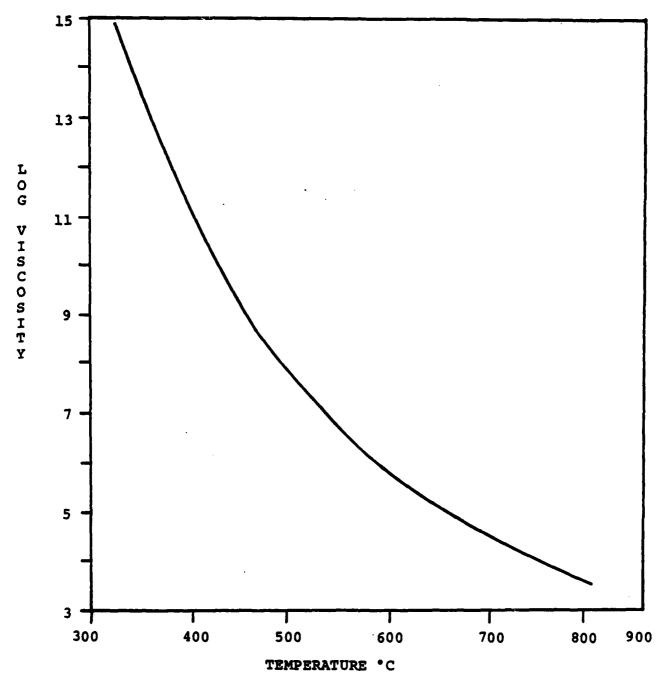


Figure 47. Static Viscosity Versus Temperature of Corning Glass 1990 Composition.

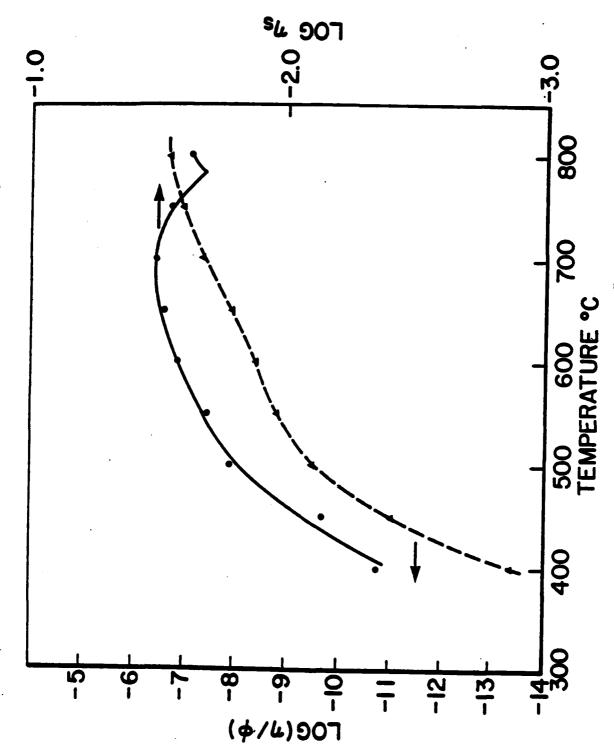
Plots of log  $\frac{\eta}{\phi}$  and experimentally determined log  $\eta_s$  are shown in Figure 48. There is a good correlation between the measured  $\eta_s$  and the calculated  $\frac{\eta}{\phi}$  for the constrained system at lower temperatures (up to 700°C). Table 23 gives values of  $\frac{\eta}{\phi}$  at various temperatures.

A minimum for the modal loss factor of the constrained layer system that is centered around 790°C can be seen in Figure 45. The second measurement of a similar constrained layer specimen did not reproduce the minimum. Only speculative explanation can be provided for this behavior as further characterization continues. Two possible explanations for the behavior are possible. Any defect arising from the spot welding of the enamel coated iron foil could contribute to the unexpected damping behavior. Also, it is known that the crystallization of the vitreous enamel coating decreases the loss factor (3). The temperature at which the minimum occurs might correspond to the liquidus temperature of the enamel. The rate and degree of crystallization of a vitreous material increase as the liquidus temperature is reached and the increased crystallization rate would adversely affect the loss factor. Beyond the liquidus temperature, the crystallites redissolve and the enamel becomes again a homogeneous liquid giving rise to an increased damping. It is hoped that further studies would clarify this high temperature behavior of the vitreous enamel.

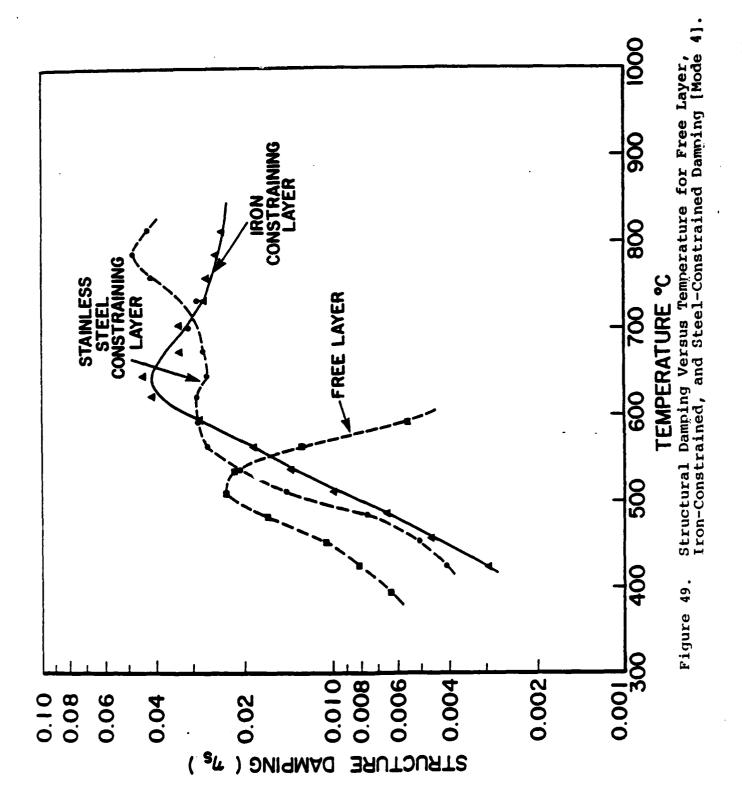
The damping effectiveness and the temperature profile of a constrained layer damping system are dependent on the damping material thickness and the constraining layer stiffness. These typical effects are shown for enamel constrained layer systems in Figure 49 for the iron-constrained and steel-constrained layer damping treatments. At lower temperatures (400-700°C), both the iron- and steel-constrained specimens exhibit similar damping trends which agree with the proposed theory. Nonetheless, at temperatures beyond 700°C, a difference in the damping behavior is noted. Whether to attribute this behavior to the differences in the elastic properties of the two constraining layers or to other conditions including crystallization of the vitreous enamel discussed earlier remains a matter for further investigation.

TABLE 23  $\frac{1}{6}$  AT VARIOUS TEMPERATURES

Temperature °C	$\frac{\eta}{\phi}$ (Poise <sup>-1</sup> )	log $\frac{1}{\eta}$
400	4.61 × 10 <sup>-14</sup>	-13.34
450	8.83 x 10 <sup>-12</sup>	-11.05
500	$3.19 \times 10^{-10}$	-9.50
550	$13.06 \times 10^{-10}$	-8.88
600	3.33 x 10 <sup>-9</sup>	-8.48
650	11.91 x 10 <sup>-9</sup>	-7.92
700	4.12 x 10 <sup>-8</sup>	-7.39
750	10.69 x 10 <sup>-8</sup>	-6.97
800	18.23 x 10 <sup>-8</sup>	-6.74



Shape of the Function Log  $\frac{\eta}{\phi}$  Versus Temperature. Figure 48.



## CONCLUSIONS

From the results presented in the previous section, the following conclusions can be drawn.

- (1) New compositions such as mixed alkali, lead silicate and low temperature glasses exhibited the characteristic damping properties in the viscoelastic range. The damping temperatures and the peak intensities were composition dependent. The fluoride composition initially showed a typical vitreous enamel damping peak; however, after a heat treatment significant broadening of the damping was noted. It was suggested that the broadening of the damping peak appears to be related to a precipitation phenomenon of a second phase during the heat treatment.
- (2) It has been shown that there is a linear relationship between the glass transition temperature and the damping temperature. This suggests that the damping temperature is directly related to the viscosity. The viscoelastic damping occurs at a temperature near the softening point and corresponds to a viscosity range of  $10^6 10^7$  poise.
- (3) The viscosity range  $10^6 10^7$  was found to be favorable for processes like nucleation and crystallization. Effects of nucleation and crystallization on the damping properties were studied for two soda-lime-silica type of composition. Progressive increase in the crystallinity decreased the loss factor of the two compositions and it was determined that the nucleation and crystallization have an adverse effect on the loss factor.
- (4) It has been demonstrated that the concept of constrained layer damping is feasible and promising. The constraining layer smeared the free layer damping peak in such a fashion that the structural damping increased from a low value to a peak with increasing temperature and remained at the peak value even at temperatures over 800°C.

## REFERENCES

- 1. "Development of High Temperature Vibration Damping Coatings," AiResearch Technical Report, U.S. Army Contract DA-44-009-AMC 838 (T), AiResearch Manufacturing Company of Arizona, Phoenix, February, 1966.
- 2. "High Temperature Damping System Development for the J-85-21 Afterburner Liner," University of Dayton Research Institute, UDR-TR-80-79, August, 1980.
- 3. "Evaluation of High Temperature Damping Applications to Increase Fatigue Life in Rotating Jet Engine Components," University of Dayton Research Institute, UDR-TR-80-82.
- 4. A. D. Nashif, "Enamel Coatings for High Temperature Damping Materials," Am. Ceram. Soc. Bull., 52 (12) 846-49, 1974.
- 5. P. I. Sridharan, "Damping in Porcelain Enamel Coatings," AFML-TR-74-191, 53 (12), 1976.
- 6. B. Kumar, C. A. Cannon, and G. A. Graves, Jr., "Effect of Composition on the Vibration Damping Properties of Glass (Vitreous Enamel) Coatings," Glass Technology, April, 1981.
- 7. W. A. Zdanieswski, G. E. Rindone, D. E. Day, "Review The internal friction of glasses," J. Mat. Sci., 14, 763-75 (1979).
- 8. H. Oberst, Acoustica (Akustiche Berhefte), 4, 181 (1952).
- 9. R. E. D. Bishop and D. C. Johnson, The Mechanics of Vibration, Cambridge Press, (1960).
- 10. "A Study to Determine the Effect of Glass Compositional Variations on Vibration Damping Properties," AFWAL-TR-80-4061.
- 11. B. Kumar, G. A. Graves, Jr. "Vitreous Enamel as a Vibration Damping Material," Amer. Ceram. Soc. Bull. 61 (4), 480-83, 1982.
- 12. B. Kumar, G. A. Graves, Jr., D. M. Hopkins, and M. L. Drake, "Nucleation, Crystallization, and Viscoelastic Damping in Vitreous Enamel," Advances in Ceramics, Vol. 4, Edited by J. H. Simmons, Dr. Uhlman, and G. H. Beall., Amer. Cer. Soc., 1982.
- D. I. G. Jones, and C. M. Cannon, "Control of Gas Turbine Stator Blade Vibrations by Means of Enamel Coatings," J. of Aircraft, 12 (4), 226-230, 1975.