

A CONTINUING SYSTEMATIC PROGRAM ON TABLES OF THERMOPHYSICAL PROPERTIES OF MATERIALS

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Thermophysical Properties Research Center Purdue University

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FOREWORD

This report was prepared by the Thermophysical Properties Research Center (TPRC), Purdue University, West Lafayette, Indiana, under USAF Contract No. F33615-68-1229, "Scientific Documentation and Preparation of Data Tables on Thermophysical Properties." This continuing program is performed under Project No. 8975, Materials Information Analysis Center, Task No. 897502, Thermophysical Properties Research Center. The work was administered under the direction of the Air Force Materials Laboratory, Air Force Systems Command, Wright-Patterson Air Force Base, Ohio, with Mr. John H. Charlesworth (MAAM) acting as Air Force project engineer.

This report covers work conducted from 1 January 1969 to 31 December 1969. Dr. Y.S. Touloukian has been the principal investigator and program director. The report was submitted by the author in March 1970.

All the staff of TPRC's Administration, Data Tables Division, Scientific Documentation Division, Engineering Graphics Unit, and Technical Typing Unit have contributed to this program. In particular, the following members of TPRC's professional staff have contributed greatly to the respective projects; Dr. R.W. Powell (senior investigator), Mr. K. Y. Wu, and Mrs. E. K. C. Lee to the tables of thermal conductivity of solids; Dr. P. E. Liley (senior investigator) to the tables of thermal conductivity of fluids; Dr. T. Makita (senior investigator), Dr. Y. Tanaka, Mr. K. Ueda, Miss K. Inoue, and Miss E. Hata to the tables of specific heat of fluids; Dr. D. P. DeWitt (senior investigator), Mr. M.C. Muinzer, Mr. R.S. Hernicz, Dr. J. J. Hsia, Mrs. S. L. Miller, and Mr. K. F. Sohn to the tables of thermal radiative properties; Dr. M.C. Nicolaou (senior investigator) to the tables of thermal diffusivity; Dr. P. Hestermans (senior investigator) and Dr. S. C. Saxena (senior investigator) to the tables of viscosity: Dr. R. E. Taylor (senior investigator), Mr. P. L. Wang, Mr. B. M. Whitcomb, Mr. S. N. Vo, and Mr. M. Nalbantyan to the tables of thermal expansion; Mr. W.H. Shafer to the technical inquiry services; Prof. F.E. Davis to the computerized information storage and retrieval; Mrs. J.K. Gerritsen, Mrs. N.Y. Moore, Mrs. B. M. Schick, Mrs. C. A. Pelikan, Mrs. V. Ramdas, Mr. G. Kvakovszky, Mrs. S. A. O'Kane, Mrs. K. Singleton, and Mrs. M.R. Troyer to the extensive documentary work so necessary to make the tables projects possible.

This technical report has been reviewed and is approved,

EDWARD DUGGER, Chief Materials Information Branch Material Support Division

Air Force Materials Laboratory

ABSTRACT

This technical report covers work in a continuing systematic program on the thermophysical properties of materials involving the literature search, acquisition, codification, and organization, and data extraction, compilation, evaluation, correlation, analysis, and synthesis, the preparation of "intermediate tables" presenting the total available experimental information, and the final preparation of internally consistent tables of "best data" referred to as "Tables of Recommended Reference Values." The work reported on consists of both data tables projects and scientific documentation efforts. The data tables projects are on the thermal conductivity, specific heat, thermal radiative properties (emittance, reflectance, absorptance, transmittance), thermal diffusivity, and thermal linear and volumetric expansion of elements, ferrous and nonferrous alloys, intermetallic, semiconducting, and nonmetallic compounds, cermets, ceramics, mixtures, composites, systems, polymers, etc., and on the thermal conductivity, specific heat, and viscosity of fluids and fluid mixtures. Property data are presented in both tabular and graphical forms, with accompanying tables giving specifications and characterizations of the test specimens for the data. The resulting data tables are disseminated at large though the 13-volume TPRC DATA SERIES published commercially. This report does not contain the completed thousands of data sheets, but does reproduce in the Appendix, the Table of Contents and the Grouping of Materials and List of Figures and Tables for each of the first seven volumes (which contain over 8000 pages) of the TPRC DATA SERIES to show the scope of their coverage. In scientific documentation, the scope is broader. TPRC covers all materials and maintains cognizance over sixteen thermophysical properties (six more than mentioned above). There are now 55 700 references in TPRC's automated Information Storage and Retrieval System. The resulting information on research literature is disseminated through the THERMOPHYSICAL PROPERTIES RESEARCH LITERATURE RETRIEVAL GUIDE, published commercially.

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SECTION I

INTRODUCTION

The knowledge of the properties of materials is of great importance in modern science and engineering. The present obstacles and limitations in many technological developments are often a direct result of the paucity of knowledge of the properties of materials. The ever increasing amount of activities in property determinations in recent years attests to the seriousness of the situation with the result that the volume of research literature has increased many fold. Despite the widespread efforts, the present level of research on thermophysical properties of materials still falls short of existing needs and anticipated future demands. However, what is even more disturbing is the fact that engineers across the nation are using no more than a small fraction of the information already existent, either because its existence is not generally known to them or because it is in a form not directly usable, and such information remains buried in the world's technical literature.

The Thermophysical Properties Research Center (TPRC) has been pioneering to provide a remedy to this disturbing situation by conducting a continuing systematic program for more than a decade to dig the buried treasures out of the world's enormous volume of literature and put them in a form directly useful to engineers and scientists at large. This program on the thermophysical properties of materials involves the literature search, acquisition, codification, and organization, and data extraction, compilation, evaluation, correlation, analysis, and synthesis, the preparation of "intermediate tables" presenting the total available experimental information, and the final preparation of internally consistent tables of "best data" referred to as "Table of Recommended Reference Values."

The data tables projects of this program include the thermal conductivity, specific heat, thermal radiative properties (emittance, reflectance, absorptance, transmittance), thermal diffusivity, and thermal linear and volumetric expansion of elements, ferrous and nonferrous alloys, intermetallic, semiconducting, and nonmetallic compounds, cermets, ceramics, composites, mixtures, systems, polymers, etc., and the thermal conductivity, specific heat, and viscosity of fluids and fluid mixtures. Property data are presented in both tabular and graphical forms, with accompanying tables giving specifications and characterizations of the test specimens for the data. The resulting data tables are disseminated at large through the 13-volume TPRC DATA SERIES published commercially.

In scientific documentation, the scope is broader. TPRC covers all materials and maintains cognizance over the following sixteen thermophysical properties: thermal conductivity, accommodation coefficient, thermal contact resistance, thermal diffusivity, specific heat, viscosity, emittance, reflectance, absorptance, transmittance, absorptance to emittance ratio, Prandt number, diffusion coefficient, thermal linear expansion coefficient, thermal volumetric expansion coefficient, and surface tension. The resulting information on research literature is disseminated through the THERMOPHYSICAL PROPERTIES RESEARCH LITERATURE RETRIEVAL GUIDE, published commercially.

The overall activities of TPRC are divided into four areas: namely, (1) Scientific Documentation, (2) Generation of Data Tables, (3) Experimental Research, and (4) Theoretical Research. The activities discussed in this report cover the first two areas.

The literature search, information processing, and data processing for preparing the Retrieval Guide and the TPRC Data Series and the status of the various active projects are outlined in the following sections. It should be noted that part of the projects discussed below are not funded under this contract.

SECTION II

SCIENTIFIC DOCUMENTATION

1. LITERATURE SEARCH AND INFORMATION PROCESSING

The Scientific Documentation Division of TPRC provides comprehensive and authoritative source information on the thermophysical properties of all matter through continuing and systematic literature search, acquisition, codification, and organization of the existing information in the world literature, and putting the information in TPRC's automated Information Storage and Retrieval System.

From 1957 to 1964 TPRC has searched the world literature primarily through sixteen abstracting journals. This search, covering the publication period from January 1920 to June 1964, has involved the scanning of approximately 33 400 000 abstracts out of approximately 81 000 000 abstracts reported by these journals. Out of the 33 400 000 abstracts scanned, only 52 500 (0.16%) were considered pertinent. Subsequent examination and checking of the 52 500 abstracts and the original papers revealed a large number of duplications between the various abstracting journals and nearly 9900 irrelevant ones, leaving a net total of 28 800

research documents obtained from these sources. In addition to abstracts, 4900 documents had come to TPRC's attention from other sources making a total of 33 700 documents up to June 1964. These 33 700 references are covered in the revised second edition of the Retrieval Guide published in October 1967.

Subsequently, in preparation for future Retrieval Guide publications, an additional 22 000 reference entries have been made. Thus, as of 31 December 1969 there are 55 700 references identified for TPRC's Information Storage and Retrieval System. The above figures give an insight as to the magnitude of the effort involved in a thorough search of world knowledge even in a relatively specialized field. The overall current statistical data for TPRC's scientific documentation are presented in Table I.

When the retrospective search of the world literature, primarily through the medium of abstracting journals, was completed early in 1965. TPRC reviewed its procedure of using abstracting journals for the identification of current literature on thermophysical properties research. It was recognized that continued use of abstracting journals for research awareness would represent, at best, one to two years delay in identifying and procuring such literature, with the result that bibliographic searches provided by TPRC could not be on a reasonably current basis. Hence, beginning in 1965 TPRC has subscribed to some 80 highly selected scientific and technical journals to obtain current research articles. To complete the literature search, the Chemical Abstracts continue to be scanned to cover the open literature and several government abstracting journals such as the Scientific and Technical Aerospace Reports, U. S. Government Research and Development Reports, etc. are scanned to cover the government report literature. In addition to these basic sources, TPRC searches certain specialized sources such as Masters Theses in the Pure and Applied Sciences, Doctoral Dissertation Abstracts, and a number of special compendia. Of particular note here are the two overseas Branches of TPRC at Brussels, Belgium, and Kobe, Japan, who serve a very important input function for European and Far Eastern literature*. Furthermore, TPRC continues to develop its cooperative working arrangements on the exchange of research results and scientific information with major national and international institutions engaged in thermophysical properties research. As a result of this policy, TPRC is now able to keep abreast of current research with an average time

^{*}These TPRC overseas Branches are not supported under this contract.

TABLE I. STATISTICAL SUMMARY OF SCIENTIFIC DOCUMENTATION ACCOMPLISHMENTS

(As of 31 December 1969)

	<u>In 1969</u>	Total at End
Documents identified	4 261	55 700
Documents received	10 747	56 628
Documents coded	6 605	47 651
Reference entries on all properties	16 804	183 135

Reference Entries by Property (Since publication of Retrieval Guide, 2nd edition)

	Referen (In 1969)	ce Entries (Total at End)	% of File (At End)
Thermal conductivity	3 371	8 989	20. 51
Specific heat	2 389	6 299	14.37
Diffusion coefficient	1 385	4 905	11.19
Viscosity	1 848	4 731	10.79
Thermal linear expansion coefficient	2 526	4 202	9.59
Reflectance	1 307	3 511	8.01
Surface tension	1 252	3 205	7.31
Transmittance	875	3 067	7.00
Emittance	481	1 470	3,35
Absorptance	347	1 133	2.59
Thermal diffusivity	302	841	1.92
Thermal volumetric expansion coeffic	cient 200	593	1.35
Accommodation coefficient	241	374	0.85
Thermal contact resistance	192	282	0.64
Praedil number	73	146	0,33
Absorptance to emittance ratio	15	82	0.19
Total	16 804	43 830*	100%

There are 139 305 reference entries (see Table II) in Book 2 of Retrieval Guide, 2nd Edition, making a grand total of 183 135 reference entries as of 31 December 1969.

lag from document publication to acquisition not to exceed six months. Besides being used as input to the Data Tables Projects at TPRC, the information acquired under the documentary phase of this program is intended to render technical inquiry services to governmental, industrial, and academic laboratories in answering questions on thermophysical properties.

A statistical study completed in late 1968 showed that some 330 of the total over 3600 scientific and technical journals cited to date yielded more than 85 percent of the articles. The top-yielding 50 journals subscribed by TPRC plus all the government reports, Ph.D. dissertations, and masters theses received by TPRC yielded over 60 percent of the current holdings.

It would be desirable to replace the manual literature searching and scanning by computer searching and scanning if it could be workable so as to reduce the time lag in document identification. However, it was found that, due to the prevailing inaccurate assignment of key words to articles, the computer could pick out less than 50 percent of the pertinent articles from the literature and the other 50 percent or more would be lost. Furthermore, the majority of the articles picked out by the computer would still have to be screened manually for being valid and useful for the documentation file. In view of the above, it appears that manual processing must continue for document identification and screening.

The problem of procuring research documents from the open literature is beginning to assume major proportions especially in the case of foreign literature and special publications of limited distribution. Therefore, TPRC's specialized holdings constitute a unique national asset and are assuming increasing importance for rapid access to the world literature on thermophysics and thermophysical properties. It is TPRC's experience that literature retrieval programs which yield only bibliographies as their end product are becoming increasingly less useful because of the time lapse involved in procuring the cited documents. To remedy this situation, TPRC has supplemented its long-standing practice of submitting bibliographic responses to literature search requests with copies of the actual documents in the form of standard microfiche. The conversion of TPRC's hard copy document holdings into microfiche form was completed during 1967.

2. RETRIEVAL GUIDE

The comprehensive second edition of the THERMOPHYSICAL PROPERTIES RESEARCH LITERATURE RETRIEVAL GUIDE was published in October 1967 by the Plenum Press.

This three-book volume represents the printout of a special computer program and provides quick access to the world literature on thirteen thermophysical properties of all matter. This definitive work contains a complete coverage of the world literature published from January 1920 (in some cases earlier) to June 1964 on thirteen thermophysical properties. Its 2800 pages (in 3 books) report 45 116 materials, citing 33 700 references representing 26 562 authors and 3600 separate scientific and technical journals and books in addition to Government reports. Its substance and property coverages are listed in Table II. The contents of the three books of the Retrieval Guide are as follows:

Book 1: Primarily constitutes TPRC's classified Directory of Substances in which information on the thirteen thermophysical properties are reported. Book 1 also contains three other major chapters which greatly enhance its usefulness. These consist of: (1) Guide to TPRC's Substance Classification Procedure and Numerical Codes; (2) Dictionary of Synonyms and Trade Names with a Listing of Cross References; and (3) Index to Mixtures.

Book 2: Contains the classified code entries and publication year of each reference for each property of each material. The classified code entries cover the following:

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Phys. State: 1-Solid; 2-Liquid; 3-Gas; 4-Semi-solid; 5-Powder; 6-Suspensoid; 7-Sintered; 8-Solid-Gas system;
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9-Solid-Liquid system.

Subject: 1-Theoretical; 2-Experimental; 3-Theo. and Exp.; 4-Property values; 5-Theo. and Prop. val.; 6-Exp. and Prop. val.; 7-Theo., Exp., and Prop. val.; 8-Survey, Review, Compendia, or Bibliography.

Language: 1-Eng.; 2-Fr.; 3-Ger.; 4-Dutch; 5-It.; 6-Jap.; 7-Rus.; 8-Span.; 9-Other.

Temperature: 1-Low, 0 to 75 K; 2-Normal, 75 to 1275 K; 3-High, 1275 K and up; 4-(Low + Normal); 5-(Normal + High); 6-(Low + Normal + High); 7-Not specific.

Book 3: Part A provides bibliographic citations for the 33 700 references covering scientific and technical journals in addition to university dissertations and

TABLE II. SUBSTANCE AND PROPERTY COVERAGES OF RETRIEVAL GUIDE*

SUBSTANCE COVERAGE - All Mate	er	PROPERTY COVERAGE - Transport and Thermodynamic Properties Encountered in Heat and Mass Transfer Calculations	
Elements and chemical compounds Ferrous and nonferrous alloys	9 030 9 970	Thermal conductivity (including accommodation coefficient and thermal contact resistance)	31 050
Mixtures Systems, composites, etc.	13 396 1 643	Specific heat	28 030
Polymers, rubbers, etc. Refractories	$\begin{array}{c} 2\ 600 \\ 961 \end{array}$	Viscosity (Newtonian and non-Newtonian; including fluidity)	46 870
Glasses Natural products Minerals	$egin{array}{c} 1\ 109 \\ 1\ 100 \\ 662 \end{array}$	Thermal radiative properties (emittance, reflectance, absorptance, transmittance,	
Paints, surface finishes, coatings	2 632	and optical constants)	9 400
Slags, scales, aggregates, cermets,		Diffusion coefficient	21 720
fuels, lubricants, fibers, fabrics, pharmaceuticals, insulations,		Thermal diffusivity	1 705
building materials, residues, etc. General	1 967 46	Prandtl number	540
Total number of substances	45 116	Total number of reference entries	139 305

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This storehouse of information has come from 33 700 references representing 26 562 authors and 3600 separate scientific and technical journals and books in addition to sources of governmental and industrial reports (e.g., Defense Documentation Center, Clearinghouse for Federal Scientific and Technical Information, Atomic Energy Commission, National Aeronautics and Space Administration, research centers, and the like).

technical reports of governmental agencies, industrial organizations, and research centers and laboratories. Part B contains an index to names of contributing authors.

Effective January 1967, the scope of property coverage by TPRC was increased to include the coefficients of linear and volumetric thermal expansion and surface tension. Furthermore, each thermophysical property is coded separately instead of the past practice of coding several related properties in groups. For instance, the thermal conductivity, accommodation coefficient, and thermal contact resistance were formerly all coded under the property thermal conductivity. They are now coded separately as individual properties. Similarly, the five entries under thermal radiative properties are now listed separately as individual properties. Thus, effective since January 1967, TPRC maintains cognizance over sixteen thermophysical properties for all materials. They are:

- A. Thermal conductivity
- B. Accommodation coefficient
- C. Thermal contact resistance
- D. Thermal diffusivity
 E. Specific heat at constant pressure
- F. Viscosity G. Emittance
- H. Reflectance
- I. Absorptance
- J. Transmittance
- K. Absorptance to emittance ratio
- L. Prandtl number
 M. Diffusion coefficient
- N. Thermal linear expansion coefficient
- O. Thermal volumetric expansion coefficient
- P. Surface tension

The code entries are also revised as follows:

Physical State:

A-Solid (including porous); B-Liquid; C-Gas; D-Powder (loose or cold-pressed - including coarser granular particles), for example, sand; E-Sintered (including hotpressed); F-Suspensoid (including fine emulsions), for example, petroaltum, blood, milk, paint, mud, clay, paste, gel, etc.; G-Powder - fluid system (including coarser granular particles), for example, sand + liquid, fluidized bed, glass spheres suspended in glycerin, etc.; H-Solid - fluid system, for example, contact in gaseous atmosphere, etc.; I-Liquid - gas system.

Subject:

A-Theoretical (about properties) - Original theory or a sufficient amount of modification of existing theory. A theoretical discussion. Analytical development for the prediction of results; B-Experimental - Description, discussion or theory about experimental technique or apparatus for the determination of property values; C-Property Values - Calculated or measured data with no theory or experimental techniques discussed. Limited data compendia, correlations and nomographs are included here; D-Theoretical and Experimental - Combination of codes A and B; E-Theoretical and Property Values - Combination of codes A and C; F-Experimental and Property Values - Combination of codes B and C; G-Theoretical, Experimental and Property Values - Combination of codes A, B and C; H-Survey or Review - Critical or informative survey or review on a broad scale about experimental techniques or theories; I-Data Collections - Compilation of data on a broad scale (critical or uncritical). Handbooks, data books, encyclopedias, etc.

Language:

A-Eng.; B-Fr.; C-Ger.; D-Dutch; E-It.; F-Jap.;

G-Rus.; H-Span.; I-Other.

Temperature:

A-Low \leq -198 C, \leq 135 R, \leq -325 F, \leq 75 K; B-Normal, \leq 1000 C, \leq 2292 R, \leq 1832 F, \leq 1273 K; C-High, >1000 C, >2292 R, >1832 F, >1273 K; D-Low + Normal; E-Normal + High; F-Low + Normal + High; G-Undefined.

3. AUTOMATION AND COMPUTERIZED INFORMATION STORAGE AND RETRIEVAL SYSTEM

With the installation of the new CDC 6500 computer facility at Purdue University in summar 1967, TPRC's long needed capability of full mechanization in its search and retrieval operations has been fulfilled. As a result, TPRC now has an automated bibliographic search capability to respond to specific inquiries or to process standing requests for a continuing bibliographic service tailored to meet demands for specific technical profiles of individual engineers, scientists, corporations, laboratories, or governmental agencies.

During 1969, TPRC will continue to generate, quarterly, miniature "Retrieval Guide" for each property in order to serve its own inhouse programs as well as subscribers at nominal costs. For the future, it is considered to generate a series of Retrieval Guides for individual properties for formal publication so as to parallel the property-oriented 13-volume TPRC Data Series.

SECTION III

DATA TABLES PROJECTS

1. DATA PROCESSING PROCEDURES

As a result of the comprehensive search of literature in the scientific documentation phase of this program described above, the original research papers of interest to TPRC are uncovered. These papers are procured, from which the data are extracted, scrutinized, organized, converted to be in uniform units, and homogeneously tabulated and plotted, in the form of "intermediate tables" presenting the total available experimental information, as the first stage toward the preparation of internally consistent tables of "best data" referred to as "Tables of Recommended Reference Values." TPRC senior staff then review this information and give the organized data final critical evaluation. At this second stage, the experimental data are analyzed, correlated, and synthesized. This two-stage data processing is found by TPRC to be the most logical approach lending itself to greater effectiveness in bringing to the user the results of this type of painstaking research in the shortest possible time.

The detailed procedures which TPRC follows in data compilation and in data analysis and synthesis are not necessarily a matter of established routines and do vary from property to property and from one group of materials to another. There are certain principles which must be followed, however. For example, the data should be extracted directly from their original sources to ensure freedom from errors of transcription. The characterization of the test specimen should be specified as clearly as possible so as to fully identify the material tested. Especially for solids, the source of the material, method of fabrication, treatment and heat history of the specimen and the measuring method and conditions should be noted. If a comparative measuring method is used, the material used as comparative standard should be cited. The accuracy and precision of the data reported should be separately denoted. The complete reference to the original work should always be cited with the data. Whenever some of the above criteria cannot be satisfied because of absence of necessary information in the original work, an attempt is made to contact the author, if possible. In the cases where data cannot be adequately evaluated by TPRC due to lack of required information, such data are appropriately "flagged."

TPRC's Gerber Data Point Reader has been in full operation to accurately and speedily read data points off graphs presented in the research literature. Whenever the graph is too small to give accurate readings, an attempt is made to contact the author, if possible, for original data in tabular form.

In connection with its activities in data processing, TPRC has established, through experience, appropriate procedures of operational practice which lend to good organization of work, uniform recording and filing, and other procedures of "good housekeeping," thus assuring ready tractability of original records of processed data as permanent working records for reference at any time in the future.

The procedure in data analysis and synthesis involves critical evaluation of the validity of available data and related information, resolution and reconciliation of disagreements of conflicting data, correlation of data in terms of various affecting parameters (sometimes in reduced form), comparison of the resulting data with theoretical predictions or with results derived from semi-theoretical relationships or from generalized empirical correlations, etc. Furthermore, thermodynamic principles and semi-empirical techniques are employed to fill in gaps and to extrapolate existing data so that the resulting recommended values are internally consistent and cover as wide a range of the controlling parameters as possible.

2. ACTIVE PROJECTS

TPRC data tables projects include the thermal conductivity, specific heat, thermal radiative properties (emittance, reflectance, absorptance, transmittance), thermal diffusivity, and thermal linear and volumetric expansion of elements, ferrous and nonferrous alloys, intermetallic, semiconducting, and nonmetallic compounds, cermets, ceramics, composites, mixtures, systems, polymers, etc., and the thermal conductivity, specific heat, and viscosity of fluids and fluid mixtures. Table III gives a statistical summary of the accomplishments in 1969 of all the data tables projects. Part of the projects are, however, not funded under this contract.

The resulting data tables are to be disseminated at large through the 13-volume TPRC DATA SERIES, as described in more detail later in the next subsection. A summary of the statistical data on the completed Volumes 1 to 7 of the TPRC DATA SERIES is given in Table IV. The property data are presented in both tabular and graphical forms, with accompanying tables giving specifications and characterizations of the test specimens for the data.

TABLE III. STATISTICAL SUMMARY OF ACCOMPLISHMENTS OF ACTIVE DATA TABLES PROJECTS

(As of 31 December 1969)

	<u>In 1969</u>	Total at End
No. of documents processed	6892	16 094
No. of documents accepted as data sources	2259	7 311
No. of materials compiled	1073	5 437
No. of data sets compiled	9642	38 270

TABLE IV. SUMMARY OF STATISTICAL DATA ON VOLUMES 1 TO 7 OF "TPRC DATA SERIES"

		No. of Pages	No. of Materials	No. of Data Sets	No. of References to Data Sources*
Volume 1.	Thermal Conductivity – Metallic Elements and Alloys	1600	892	5539	1013
Volume 2.	Thermal Conductivity - Nonmetallic Solids	1300	812	4627	598
Volume 3.	Thermal Conductivity – Nonmetallic Liquids and Gases	700	170	1505	725
Volume 4.	Specific Heat - Metallic Elements and Alloys	820	322	1186	428
Volume 5.	Specific Heat – Nonmetallic Solids	1730	550	1009	457
Volume 6.	Specific Heat - Nonmetallic Liquids and Gases	380	56	863	595
Volume 7.	Thermal Radiative Properties - Metallic Elements and Alloys	1650	242	5184	371

These are the references to data sources only, not including those references to the text on the theory, estimation, and measurement of the respective thermophysical properties.

The following brief summaries will serve to characterize each of the data tables projects.

a. Thermal Conductivity

Thermal conductivity constitutes Volumes 1, 2 and 3 of the TPRC DATA SERIES. The data compilation and critical evaluation and the generation of recommended reference values for the thermal conductivity of the elements is totally completed and is being maintained on a current basis. For the other groups of materials the work is finished to various degrees depending upon the importance of the various groups of materials. It has been a standing practice to try to finish the data processing for the most current research documents first and then to work backwards on the older documents of earlier years. Because of the immense amount of new research documents generated every year and because of the limited funding level and man-year effort, it is impossible at the present level of operation to process completely all the research documents for all the materials.

In the Retrieval Guide, second edition, which contains 33 700 references, there are 7329 references on thermal conductivity, i.e., 21.7 percent (neglecting the relatively small number of references on accommodation coefficient and thermal contact resistance). The present rate of document input into TPRC's Information Storage and Retrieval System is about 6000 per year. If the past ratio remains approximately the same, as indicated by the relative file size (20.51%) for thermal conductivity shown in Table I, there are over 1200 new documents per year on thermal conductivity entering TPRC System. Therefore, it becomes a too arduous task even just to maintain current by processing the most recent documents, 1200 per year. A similar situation is present for most of the other projects.

The thermal conductivity of metals, alloys, and intermetallic compounds, which are organized into seven groups, is included in Volume 1 of the TPRC DATA SERIES. Besides the presentation of experimental data, recommended values are given for the elements and for a few alloys. Volume 2 presents data for thirty groups of nonmetallic solids. Recommended values are given for the elements and a few oxides. Volume 3 contains the critically evaluated and recommended reference values for 58 fluids and also the compiled data for 112 systems of gas mixtures which are organized into four groups. The pure fluids are organized into three groups; for the elements recommended values are given for solid, saturated liquid, saturated vapor, and gaseous states while for the other two groups of pure fluids, recommended values are given for saturated liquid and gaseous states. See the Appendix for complete lists of the materials covered.

Since all the three volumes on the thermal conductivity are completed, in addition to updating and upgrading the existing tables and compiling data for new materials, the future efforts will also include the critical analysis of the data for alloys of Volume 1 and refractory oxides of Volume 2 and the extension of the work on Volume 3 to the thermal conductivity of an increasing number of refrigerants and to the high-pressure effect on the thermal conductivity of liquids and gases.

b. Specific Heat

Specific heat constitutes Volumes 4, 5 and 6 of the TPRC DATA SERIES. Tables on the specific heat of the elements and of all the important alloys, intermetallics, compounds, cermets, and mixtures have been prepared. Data on the specific heat of metallic elements and alloys and nonmetallic solids are compiled here at TPRC while the work on nonmetallic liquids and gases is done at TPRC's Kobe Affiliate in Japan.

The specific heat of the metals and alloys, which are organized into four groups, is included in Volume 4. Volume 5 presents data for twenty-four groups of nonmetallic solids. Volume 6 contains the critically evaluated and recommended reference values for 56 fluids organized into four groups, and the recommended values are given for both liquid and gaseous states.

Besides updating and upgrading the existing tables and compiling data for new materials, the future efforts will also include the critical analysis of the data for the elements of Volumes 4 and 5 and the extension of the work on Volume 6 to the specific heat of an increasing number of refrigerants and to the high-pressure effect on the specific heat of liquids and gases.

c. Thermal Radiative Properties (Emittance, Reflectance, Absorptance, and Transmittance)

This group of properties constitutes Volumes 7, 8 and 9 of the TPRC DATA SERIES. The radiative properties of metals and alloys are presented in the completed Volume 7. Volume 8 will contain the data for many groups of nonmetallic solids, while Volume 9 will present the data for coatings of all types, especially those particularly suitable for thermal control and for high temperature applications.

The present tables are organized in a way that is much different from that in the original TPRC Data Book. It is due to the establishment of a better scheme for the designation and categorization of the subproperties. According to the new scheme, by applying the proper geometric conditions (angular, normal, hemispherical) and wavelength conditions (spectral, total, integrated, solar) to the four prime properties, there are althogether thirty-three subproperties for any one material. The wavelength range covers from 500 Å to 1000 μ m, which encompasses the thermal portion of the spectrum, and special attention is given to solar spectrum conditions.

In addition to the presentation of original data, the data for many materials have been analyzed and "analyzed data graphs" are presented in parallel to give the user an evaluated review of the available data and where possible to recommend values under specific environmental conditions.

Since Volume 7 is now completed, the major efforts are concentrated on the processing of data on nonmetallic solids and coatings. A new classification scheme for coatings has been developed after considerable study by the TPRC staff in consultation with several national experts. The coatings are now classified into conversion-diffusion coatings, contact coatings, pigmented coatings (vitreous enamels and paints), and uncharacterized coatings (commercial designations).

Although in the Retrieval Guide, second edition, there are only 2829 references on thermal radiative properties (8.4 percent of the 33 700 references), in recent years the number of new research documents on the thermal radiative properties has increased steadily and rapidly. Presently, 21 percent of the TPRC annual input are related to thermal radiative properties, and a half of these documents contains information on coatings.

d. Thermal Diffusivity

Thermal diffusivity will constitute Volume 10 of the TPRC DATA SERIES. The work on this property has been greatly accelerated since 1968. All the previous tables have been completely updated and revised and new research documents are processed to extend the scope of coverage. The recommended values for the thermal diffusivity of the elements are being generated also. Volume 10 of the TPRC DATA SERIES will be published in late 1970 and will contain the thermal diffusivity of elements, alloys, and many groups of nonmetallic solids.

e. Viscosity

Viscosity will constitute Volume 11 of the TPRC DATA SERIES. The work on this property had been suspended from 1964 to 1966. Starting early 1967 this work was reactivated in TPRC's European Branch at the Belgian Institute for High Pressure, Brussels, Belgium.

Data compilation for the same 56 fluids as covered by Volume 6 of the TPRC DATA SERIES and also for a number of refrigerants is well under way. Critical evaluation of the compiled data so as to generate the recommended reference values is also in progress. A computational procedure for generating the recommended values from experimental data has been fully established. This Volume 11 will be published in 1971.

f. Thermal Expansion (Linear and Volumetric)

Thermal expansion will constitute Volumes 12 and 13 of the TPRC DATA SERIES. This project was initiated in mid-1968 and has since been in full progress. The work includes both linear and volumetric expansion and covers elements, alloys, intermetallic compounds, and various groups of nonmetallic solids. The two volumes are planned to be published in 1972.

3. TPRC DATA SERIES

Synthesis of existing fragments of knowledge is as important as so-called original observation. The availability of adequate standard reference data tables is essential to national progress, economy, and defense. To this end TPRC has contributed greatly through the generation and dissemination of data tables on thermophysical properties.

The 13-volume TPRC Data Series brings together the most comprehensive and authoritative compilations of existing numerical data on thermophysical properties of materials. Recommended reference values of particular properties for particular materials are also often included. This new TPRC Data Series has evolved from the old three-volume loose-leaf 11" x 17" size TPRC Data Book, which was disseminated by TPRC from 1960 to 1967 and is well known nationally and, indeed, internationally. As of January 1967, the old TPRC Data Book contains 3322 data sheets, reporting 11 425 test specimens and citing 3424 references.

In view of the old TPRC Data Book's continuing rapid growth since 1960 and the extensive physical proportions it had assumed and in order to eliminate the cumbersome merging of supplements and the associated high cost of dissemination, it was decided in 1967 to discontinue the procedure of publication in loose-leaf format and its semi-annual dissemination by TPRC. Instead, the old Data Book has been restructured into volumes by properties and extensively updated, upgraded, and enlarged to become this new TPRC Data Series in 13 volumes.

One of the major improvements of the new TPRC Data Series over the old Data Book is that each volume of the Data Series includes also a comprehensive text on the theory, estimation, and measurement of the property and an overall material index, in addition to the most comprehensive compilation of numerical data. The text provides a general background and review of the theory and physics of the property considered in that volume, the methods for estimating the property values when experimental data are not available, and the experimental methods and apparatus for measuring the property. Such a background information enables the user to properly and fully utilize the compiled data and also enhance the usefulness of the data themselves. The material index lists alphabetically all the materials contained in that volume and in the companion volumes on the same property together with their respective volume numbers and page numbers. Many commercial designations of materials are cross-indexed with their previous designations and synonyms are cross-indexed with each other. Thus it enables the user to quickly locate the data for any material contained in the volume or in the companion volumes and to completely retrieve the desired data.

The numerical data for each material (except for the pure fluids) are presented in a full-page figure and also tabulated in a data table, accompanied by a specification table providing for every set of data a concise description of the sample characterization, as well as the method of measurement and test conditions. For the pure fluids of Volumes 3, 6 and 11, a critical discussion is given for each fluid in place of the specification table for the samples and data table for the raw data, which are presented through a departure plot, since all data have been critically reviewed and recommended values are presented.

Table V gives an indication of the structure, scope, and publication schedule of the TPRC Data Series. The first seven volumes of the Data Series have been completed and are in the publisher's hands for printing. They will soon be available through Plenum Press, New York, in the form of formal hard-bound volumes, 9-1/4" x 11-1/4" in size. It is contemplated that work on Volumes 8, 9 and 10 will be completed in late 1970. Table V also reflects the substantial increase in scope of Volumes 12 and 13, which are now anticipated to comprise 1000 pages each instead of 500 pages projected a year ago. The Table of Contents and Grouping of Materials and List of Figures and Tables of the completed first seven volumes are given in the Appendix.

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TABLE V. PUBLICATION SCHEDULE FOR TPRC DATA SERIES*

		1970	1971	1972	1973	1974	1975	1976	1977	1978
Volume 1.	Thermal Conductivity - Metallic Elements and Alloys	First Ed. 1600				Second Edition				
Volume 2.	Thermal Conductivity - Nonmetallic Solids	First Ed. 1300				Second Edition				
Volume 3.	Thermal Conductivity - Nonmetallic Liquids and Gases	First Ed. 700				Second Edition				
Volume 4.	Specific Heat – Metallic Elements and Alloys	First Ed. 820					Second Edition			
Volume 5,	Specific Heat - Nonmetallic Solids	First Ed. 1730					Second Edition			
Volume 6.	Specific Heat – Nonmetallic Liquids and Gases	First Ed. 380					Second Edition			
Volume 7.	Thermal Radiative Properties - Metallic Elements and Alloys	First Ed. 1650						Second Edition		
Volume 8.	Thermal Radiative Properties - Nonmetallic Solids	First Ed. 880						Second Edition		
Volume 9.	Thermal Radiative Properties - Coatings	First Ed. 1690						Second Edition		
Volume 10.	Thermal Diffusivity	First Ed. 500							Second Edition	
Volume 11.	Viscosity		First Ed. 600						Second Edition	
Volume 12.	Thermal Expansion - Metallic Elements and Alloys			First Ed. 1000						Second Edition
Volume 13.	Thermal Expansion - Nonmetallic Solids			First Ed. 1000						Second Edition

^{*} Figures given under First Ed. represent estimated numbers of pages for the volumes. After the second edition, subsequent editions of each volume are to be released at intervals of five years.

As more volumes are readied for publication, an increasing number of the data tables and a number of the complete volumes will be in the form of critically evaluated and analyzed reference data. In fact, in the years ahead an increasingly larger portion of TPRC's total effort will be directed toward data analysis and synthesis along with data compilation. The former is a slow and painstaking task seldom fully appreciated by those who have not been involved with the generation or use of such information.

The continuing and systematic nature of this program makes the TPRC Data Series unique, quite distinctive from other data sources or handbooks, in that its coverage is constantly updated to maintain current and constantly upgraded to include more and more critically evaluated and analyzed standard reference data. Furthermore, for those users who have need for the most up-to-date information, TPRC will provide specific inquiry services or one may subscribe to the automatic data update plan tailored to meet a specific technical profile of an engineer, scientist, corporation, or laboratory. Thus, the user can always get the "last word" from the Data Series generator, TPRC.

The above-outlined procedure for data dissemination closely parallels the concept which TPRC has followed during the past twelve years for the dissemination of bibliographic information. In other words, TPRC publishes its major accomplishments in formal volumes through commercial channels while it directly disseminates up-to-date information to maintain its publications and audience on a current basis.

SECTION IV

TECHNICAL INQUIRY SERVICES

TPRC's contributions in information services continue to be primarily in the nature of technical advisory and consulting, data recommendations and predictions, and special bibliographic and data searches. During 1969, 436 inquiries (nearly nine per week) have been responded by TPRC covering various categories. Gratis responses to inquiries were given on an increasingly selective basis due to limited funds and time available for this purpose. Figure 1 gives an indication of the growth and distribution of these queries over the years. Figure 2 shows the detailed geographical distribution of the inquiries responded in 1969. As shown in Figure 2, these inquiries are from 36 states and District of Columbia and 21 foreign countries.

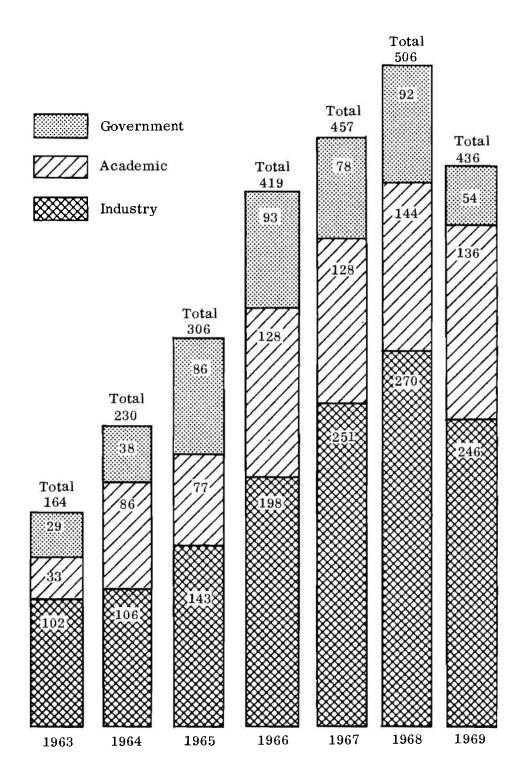


Figure 1. Growth and Distribution of TPRC's Inquiry Responses over the Years

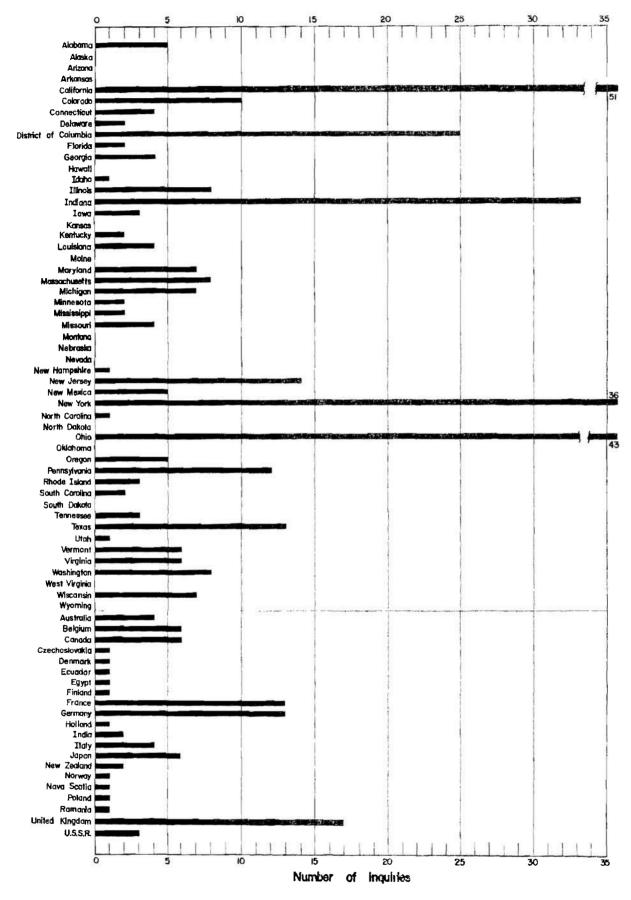


Figure 2. Geographical Distribution of Technical Inquiries for 1969

While Figure 1 indicates nearly a 14 percent decrease in the number of queries responded in 1969 over the preceding year, the time spent in answering these queries was 55 percent greater. It is of interest to note that, on the average, TPRC devoted approximately 72 hours to respond to a question requiring data analysis, 4.2 hours to answer a technical question, and 1.3 hours to respond to a query requesting literature. The average time spent per inquiry was 3.1 hours. These figures are substantially larger than the corresponding figures for 1968, primarily because TPRC devoted large block of time in 1969 to a few data analysis inquiries from the Air Force and NASA.

It is TPRC's aim and expectation that as our data books and other outputs are disseminated to the scientific and technical community on a worldwide basis through commercial publishers, the number of general technical inquiries will drop further. On the other hand, a larger number of engineers and scientists are becoming increasingly aware of TPRC's activities, thus reflecting a continuing incrase in inquiries concerning TPRC's services and capabilities.

SECTION V

OTHER PUBLICATIONS AND RELATED WORK NOT UNDER THIS CONTRACT

1. THERMOPHYSICAL PROPERTIES OF HIGH TEMPERATURE SOLID MATERIALS

"Thermophysical Properties of High Temperature Solid Materials,"
Y. S. Touloukian (Editor), Vol. 1 to 6, MacMillan Company, New York, 8479 pp.,
1967.

This major encyclopedic reference work consists of 6 volumes comprising 9 books (since Volumes 2, 4 and 6 consist of 2 books each) with a total of 8479 pages reporting data from 2447 references for 14 381 specimens in 1375 material groups. This work is available from the MacMillan Company as individual volumes as well as a complete set. The general contents of the respective volumes are as follows:

Volume 1 - Elements

Volume 2 - Nonferrous Alloys

Part I - Nonferrous Binary Alloys

Part II - Nonferrous Multiple Alloys

Volume 3 - Ferrous Alloys

Volume 4 - Oxides and Their Solutions and Mixtures

Part I - Simple Oxygen Compounds and Their Mixtures

Part II - Solutions and Their Mixtures of Simple Oxygen Compounds, Including Glasses and Ceramic Materials

Volume 5 - Nonoxides and Their Solutions and Mixtures, Including Miscellaneous Ceramic Materials

Volume 6 - Intermetallics, Cermets, Polymers, and Composite Systems

Part I - Intermetallics

Part II - Cermets, Polymers, and Composite Systems

The twelve specific properties covered in each volume are: density, melting point, heat of fusion, heat of vaporization, heat of sublimations, electrical resistivity, specific heat at constant pressure, thermal conductivity, thermal diffusivity, thermal linear expansion, thermal radiative properties (absorptance, emittance, reflectance, and transmittance), and vapor pressure.

2. NATIONAL STANDARD REFERENCE DATA SERIES

As a component of the National Standard Reference Data System - National Bureau of Standards (NSRDS-NBS), TPRC has assumed its responsibility for the generation of Standard Reference Data on the thermophysical properties of substances and materials to meet the national needs. As a result, TPRC generates a series of reports giving recommended values of thermophysical properties of definable substances, materials, or systems. While all numerical data find their way in TPRC Data Series, these research reports discuss in detail the considerations involved in arriving at the recommended reference values with a full assessment of the experimental data and theoretical guidelines on which the critical evaluation is based. The reports are normally released by TPRC first in the form of a "preliminary report" and distributed internationally to a number of experts for review and criticism. The formal reports are then published under NSRDS auspices and released in the National Standard Reference Data Series.

TPRC's activity in this area during the first two years (1 June 1964 to 30 June 1966) was concentrated on the generation of critical tables of the thermal conductivity of specifically selected materials. This results in the following two reports:

(1) "Thermal Conductivity of Selected Materials," R. W. Powell, C. Y. Ho, and P. E. Liley, National Standard Reference Data Series - National Bureau of Standards NSRDS-NBS 8, 1-168, 1966.

A total of 29 substances and materials are studied. These are the metals aluminum, copper, gold, Armco iron, iron, manganin, mercury, platinum, platinum-rhodium (40%) alloy, silver, tungsten; the nonmetallic solids aluminum oxide, beryllium oxide, Corning code 7740 glass, diamond, magnesium oxide, Pyroceram brand glass-ceramic code 9606, quartz, thorium dioxide, titanium dioxide; and the fluids argon, carbon tetrachloride, diphenyl, helium, nitrogen, m-terphenyl, p-terphenyl, toluene, and water.

(2) "Thermal Conductivity of Selected Materials, Part 2," C. Y. Ho, R. W. Powell, and P. E. Liley, National Standard Reference Data Series - National Bureau of Standards NSRDS-NBS 16, 1-146, 1968.

A total of 16 substances and materials are studied. These are the metals cadmium, chromium, lead, magnesium, molybdenum, nickel, niobium, tantalum, tin, titanium, zinc, zirconium; the nonmetal graphite (5 species); and the fluids acetone, ammonia, and methane.

In 1967 and 1968 efforts were concentrated on the generation of critical tables of the thermal conductivity of the elements which are not covered in the above two reports. Efforts were also made to expand the tables for the elements which are in liquid or gaseous state at normal temperature and pressure to include recommended values for the solid and saturated vapor states also. This results in the following "preliminary report":

(3) "Standard Reference Data on the Thermal Conductivity of Selected Materials, Part 3," C. Y. Ho, R. W. Powell, and P. E. Liley, Thermophysical Properties Research Center Final Report on NSRDS-NBS Contract CST-1346, 1-435, 1968.

A total of 68 elements are studied. These are antimony, argon, arsenic, barium, beryllium, bismuth, boron, bromine, calcium, cerium, cesium, chlorine, cobalt, dysprosium, erbium, europium, fluorine, gadolinium, gallium, germanium, hafnium, helium, holmium, hydrogen (including deuterium and tritium), indium, iodine, iridium, krypton, lanthanum, lithium, lutetium, manganese, neodymium, neon, neptunium, nitrogen, osmium, oxygen, palladium, phosphorus, rhodium, rubidium, ruthenium, samarium, scandium, selenium, silicon, sodium, strontium, sulfur, technetium, tellurium, terbium, thallium, thorium, thulium, uranium, vanadium, xenon, ytterbium, and yttrium. For most of the nonmetallic elements which

are liquid or gaseous at normal temperature and pressure, recommended values are given for solid, saturated liquid, saturated vapor, and atmospheric-pressure gaseous states. For the other elements, recommended values are provided mainly for the solid state, though values for both solid and liquid states are given for 13 elements.

After five years of diligent effort, we have completed the critical evaluation, analysis and synthesis of the available data for the thermal conductivity of the elements across the periodic table. This five-year effort is now in the process of final review and updating, and it is anticipated that a definitive work on the program entitled "Thermal Conductivity of the Elements" will become available in 1970 under the auspices of the National Bureau of Standards, Office of Standard Reference Data.

3. MASTERS THESES IN THE PURE AND APPLIED SCIENCES

"Masters Theses in the Pure and Applied Sciences Accepted by Colleges and Universities of the United States," Beth M. Schick (Editor), Volumes I to XIII.

This work provides a unique source of information on current academic research and serves as a barometer on research trends.

This annual publication has evolved from a rather limited attempt to report masters thesis titles in five disciplines to include all the pure and applied sciences (except mathematics and the life sciences). In Volume III, an experiment was made to report doctoral dissertations as well, but this was discontinued in subsequent volumes. Because of the gradual development of this publication it has undergone some modifications of its title in the first three volumes; however, the work is now recognized as listed above. The volumes contain a table of contents and an index to the universities and colleges. A brief statistical summary of coverage is given below.

Masters Theses in the Pure and Applied Sciences

	Thesis Year	Contributing Institutions	Titles Reported
Volume I Part I Part II	1955 1956	93 93	1002 1027
Volume II	1957	154	1727
Volume III* Part I	1958	139	3736
Volume IV	1959	162	4984
Volume V	1960	183	5708
Volume VI	1961	186	5911
Volume VII	1962	186	$\boldsymbol{6321}$
Volume VIII	1963	175	6505
Volume IX	1964	174	6 940
Volume X	1965	170	7310
Volume XI	1966	173	7099
Volume XII	1967	167	6909
Volume XIII**	1968	174	7802

^{*}Part II of Volume III includes doctoral dissertations for 1956-57 academic year, citing 2846 titles from 103 universities.

SECTION VI

CONCLUSIONS

This technical report has covered those phases of TPRC's activities which are either fully or partly funded under this contract for the period 1 January to 31 December 1969, and has also described briefly a number of related activities not under this contract. The results of this program are disseminated at large through two major publications: the Thermophysical Properties Research Literature Retrieval Guide and the TPRC Data Series.

^{**}Effective with Volume XIII, "Masters Theses in the Pure and Applied Sciences" appears under a new cover design and represents a turning point in the twelve-year publication of this unique reference work as the Thermophysical Properties Research Center (TPRC) and University Microfilms Library Services, Ann Arbor, Michigan, join forces in its publication. As in the past, TPRC will gather the thesis titles and will prepare the manuscript for the volume while the University Microfilms will print and disseminate this work on a worldwide basis. Back volumes of the "Masters Theses in the Pure and Applied Sciences" may also be purchased from University Microfilms.

The total efforts at TPRC are centered on the single goal of the advancement of knowledge concerning all aspects of thermophysics and thermophysical properties of materials. Comprehensiveness of treatment, compatible with the highest scientific and engineering standards, forms the basic guiding philosophy of performance. To the extent that TPRC's activities constitute a continuing systematic program, the planning of its activities is geared to both short range as well as long range goals, thus leading to effective utilization of both intellectual and fiscal resources.

APPENDIX

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Volume 2.	Thermal Conductivity - Nonmetallic Solids
	Contents Grouping of Materials and List of Figures and Tables
Volume 3.	Thermal Conductivity - Nonmetallic Liquids and Gases Contents
	Grouping of Materials and List of Figures and Tables
Volume 4.	Specific Heat - Metallic Elements and Alloys Contents
	Grouping of Materials and List of Figures and Tables
Volume 5.	Specific Heat - Nonmetallic Solids Contents
	Grouping of Materials and List of Figures and Tables
Volume 6.	Specific Heat - Nonmetallic Liquids and Gases Contents
	Grouping of Materials and List of Figures and Tables
Volume 7.	Thermal Radiative Properties - Metallic Elements and Alloys Contents
	Grouping of Materials and List of Figures and Tables

VOLUME 1. THERMAL CONDUCTIVITY - METALLIC ELEMENTS AND ALLOYS

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GROUPING OF MATERIALS AND LIST OF FIGURES AND TABLES

1. ELEMENTS

Figure and/or Table No.	Name	Symbol	Page No
1*	Aluminum	A1 ,	1
2*	Antimony	Sb	10
3	Arsenic	Ав	15
4*	Beryllium	Ве	18
5*	Bismuth	Ві	25
6*	Boron	в	41
7*	Cadmium	Cd	45
8*	Cerium	Ce	50
Sh.	Cesium	Св	54
10*	Chromium	Cr	60
11*	Cobalt	Co	64
12*	Copper	Cu	68
13*	Dysprosium	Dy	82
14*	Erbium	Er	
15	Europium	£u	90
16*	Gadolin <u>i</u> um	Gd	93
17*	Gallium	Ga	97
18 [*]	Germanium	Ge	108
19 th	Gold	Aŭ	132
20*	Hafnium	нт	138
21*	Holmium	Но	142
22*	Indium	In	146
23*	lridium	Ir	152
24*	Iron	Fe	156
25*	Lanthanum	La	171
26*	Lead	Рь	175
27*	Lithium	ы	192
28*	Lutetium	Lu	198
29+	Magnesium	Mg	202
30*	Manganese	Mn	208
31*	Mercury	нд	$\dots \dots $
32*	Molybdenum	Мо	222
33*	Neodymium	Nd	230
34	Neptunium	Np	234
35*	Nickel	Ni	$\dots \dots 237$
3 <i>6</i> *	Niobium	Nb	245
37*	Osmium	Os	254
38*	Palladium	Pd	258
39*	Platinum	Pt	262

^{*} Number marked with an asterisk indicates that recommended values are also reported for this material on separate figure and table of the same number followed by the letter R.

1. ELEMENTS (continued)

Figure and/or Table No.	Name	Symbol													Page No.
40 ∜	Plutonium	i'u													270
41*	Potassium	ĸ	,	•											274
42*	Prascodymium	Pr				•	٠					٠			281
43	Promethium	Рm													285
44*	Rhenium	Ro													288
45*	Rhodium	Rh							•					٠	292
46*	Rubidium	Rb						•							296
47**	Ruthenium	iłu		•											300
48**	Samarium	Sm											٠		305
49°	Scandium	Su			•								2		309
50*	Selenium	Se													313
51 [#]	Silicon	Si													326
52 [‡]	Silver	Ag													340
53*	Sodium	Na													349
54 *	Tantaium	TH													355
55	Technetium	Te										•			363
56*	Teliurium	Te											٠		366
5 7 *	Terbium	Tb													372
58*	Thallium	TI								٠		٠			376
59*	Thorium	Th													381
60 *	Thuitum	Tm													385
61*	Tín	Sn													269
62*	Titanium	Ti													410
63*	Tungsten	w	•												415
64 *	Uranium	U												٠	429
65*	Vanadium	V													441
66 *	Ytterbium	Yb													446
67*	Yttrium	Y													449
68*	Zine	Zn									•				453
69*	Zirconium	Zr													46 i

2. NONFERROUS BINARY ALLOYS

Figure and/or Table No.	Name	Formula								Page No.
70	Atuminum + Antimony	Al + Sb								469
71	Aluminum + Copper	Al + Cu								470
72	Aluminum + Iron	Ai + Fe								474
73	Aluminum + Magnesium	Ai + Mg								477
74	Aiuminum + Silicon	Al + Si								480
75	Aiuminum + Tin	Al + Sn								483
76	Aiuminum + Uranium	Al + U								484
77	Aiuminum + Zinc	Al + Zn								487
78	Antimony + Aiuminum	Sb + Al								488
79	Antimony + Bismuth	Sb+Bi		_						489

Number marked with an asterisk indicates that recommended values are also reported for this material on separate figure and table of the same number followed by the letter R.

Figure and/or Table No.	Name	Formul	А															P	uge No.
80	Antimony + Cadmium	Sb + Co	i.																492
81	Antimony + Copper	Sb + Cu	1																495
82	Antimony + Lead	Sb + Pt																	496
83	Antimony + Tin	Sb + Sn	١,																497
84	Beryllium + Aluminum	Be + A	1.																498
85	Beryllium + Magnesium	Be + M	g .																499
86	Biemuth + Antimony	Bi + Sb	, ,											. 1	ì				502
87	Bismuth + Cadmium	Bi + Co	1 .																505
88	Bismuth + Lead	Bi + Pt)																508
89	Bismuth + Tin	Bi + Sn	١.				•												511
90	Cadmium + Antimony	Cd + St																	514
91	Cadmium + Bismuth	Cd + B	i.																517
92	Cadmium + Thallium	Cd + T	1.																520
93	Cadmium + Tin	Cd + Sr	ı .																521
94	Cadmium + Zinc	Cd + 2	n,																524
95	Chromium + Nickel	Cr + N	i.																525
96	Cobalt + Carbon	Co + C																	526
97	Cobalt + Chromium	Co + C	r .																527
98	Cobalt + Nickel	Co + N	i.																528
99	Copper + Aluminum	Cu + A	1 .									-							530
100	Copper + Antimony	Cu + St	٠.																534
101	Copper + Arsenic	Cu + A	B.		•														535
102	Copper + Beryllium	Cu + B	e .						•										538
103	Copper + Cadmium	Cu + C	ď.		•	•	•	•	•	•				•					541
104	Copper + Chromium	Cu + C	r .	•		•			•		•	•			•			•	542
105	Copper + Cobalt	Cu + C	0			•							•			•		•	545
106	Copper + Gold	Cu + A	u	•		•			•			•	•				•		549
107	Copper + Iron	Cu + F	e	•	•		•		٠		•		•	•					551
108	Copper + Lead	Cu + P	b	•	•	•	•	٠			•		•					•	554
109	Copper + Manganese	Cu + M	n	•	•	•		•	•	•		•	•		•			•	557
110	Copper + Nickel	Cu + N	i	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	561
111	Copper + Palladium	Cu + P		•	•	•	•	•	•	•	•	•	•	•	•	٠	•	•	568
112	Copper + Phosphorus	Cu + P		•	•	•	•	•	•			•	•	•	•	•	•	•	571
113	Copper + Platinum	Cu + P						•				•	•	٠	•	•	•	٠	574
114	Copper + Silicon	Cu + Si						•								•	•	•	575
115	Copper + Silver	Cu + A	_		•	•	•	•	•					•	•	•	•	•	578
116	Copper + Tellurium	Cu + T		•	•	•	•	•	•	•	•	•	•	٠	•	•	•	•	581
117	Copper + Tin	Cu + Si		•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	584
118	Copper + Zinc	Cu + Z		•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	588
119	Germanium + Silicon	Ge + S		•	•	•			•	•	•	•	•	•	•	•	•	•	597
120	Gold + Cadmium	Au + C								•	•	•	•	•	•	•	•	•	600
121	Gold + Coholt	Au + C									•				•				606 603
122	Gold + Copper	Au + C							•	•	•	•	•	•	•	•	•	•	609
123	Gold + Copper	Au + C					•	•	•	•	•	•	•	•	•	•	•	•	614
124	Gold + Palladium Gold + Platinum	Au + P		•	•	•	•	•	•	•	•	•	•	•	*				617
125				•	•	•	•	•	•	•	•	•	•	•	•		•		
126	Gold + Silver	Au + A	6	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	620

Figure and/or Table No.	Name	Formula															P	age No.
127	Gold + Zinc	Au + Zn				•												623
128	Hafnium + Zirconium	Hf + Zr															•	624
129	Indium + Lead	ln + Pb				•				•	٠	•				•		627
130	Indium + Thallium	In + Tl	•									•				•	•	630
131	Indium + Tin	ln + Sn		•												•	٠	634
132	Lead + Antimony	Pb + Sb				•	•											637
133	Lead + Blamuth	Pb + Bi				•		٠								•	•	640
134	Lead + Indium	Pb + In	•	•		•	•				•			•		٠	•	643
135	Lead + Silver	Pb + Ag		•				•	•							•	•	646
136	Lead + Thallium	Pb + Tl	•		•	•					•							649
137	Lead + Tin	Pb+Sn	•	•	•			•				•		•	•	•	•	652
138	Lithium + Sodium	Li + Na				•		•							•	•		655
139	Magnesium + Aluminum	Mg + Al																658
140	Magnesium + Cadmlum	Mg + Cd												•				661
141	Magnesium + Calcium	Mg + Ca																662
142	Magnesium + Cerium	Mg + Ce																663
143	Magnesium + Copper	Mg + Cu															•	666
144	Magnesium + Manganese	Mg + Mn																669
145	Magnesium + Nickel	Mg + NI																672
146	Magnesium + Silicon	Mg + Si																675
147	Magnesium + Silver	Mg + Ag																678
148	Magnesium + Tln	Mg + Sn							. •									679
149	Magnesium + Zinc	Mg + Zn																680
150	Mangunese + Copper	Mn + Cu																683
151	Manganese + Iron	Mn + Fe																684
152	Manganese + Nickel	Mn + NI																685
153	Mcreury + Sodium	Hg + Na																686
154	Molybdenum + Iron	Mo + Fe																690
155	Molybdenum + Titanium	Mo + Tl																691
156	Molybdenum + Tungsten	Mo + W																694
157	Nickel + Chromium	NI + Cr																697
158	Nickel + Cobalt	NI + Co																700
159	Nickel + Copper	NI + Cu																703
160	Nickel + Iron	NI + Fe																707
161	Nickei + Manganese	Ni + Mn																710
162	Niobium + Uranium	Nb + U															' ,	713
163	Nlobium + Zirconium	Nb + Zr																716
164	Palladium + Copper	Pd + Cu																720
165	Palladium + Gold	Pd + Au																723
166	Palladium + Piatinum	Pd + Pt		٠									•	•				726
167	Palladium + Silver	Pd + Ag															•	727
168	Platinum + Copper	Pt + Cu																730
109	Platinum + Gold	Pt + Au																7.33
170	Platinum + Irldium	Pt + lr																734
171	Platinum + Palladium	Pt + Pd																737
172*	Platinum + Rhodium	Pt + Rh						•										738

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Figure and/or Table No.	Name	Formula	Page No.
173	Platinum + Ruthenium	Pt + Ru , , , , , , , , , , ,	. 743
174	Platinum + Silver	Pt + Ag	· 745
175	Plutonium + Aluminum	Pu + Al	. 746
176	Plutonium + Iron	Pu + Fe	. 747
177	Potassium + Sodium	K + Na	. 748
178	Rubidium + Cesium	Rb + Cs	- 751
179	Selenium + Bromine	Se + Br	. 754
180	Selenium + Cadmium	Su + Cd	. 755
181	Selenium + Chlorine	Se + Cl	. 756
182	Selenium + Iodine	Se + 1	. 757
183	Selenium + Thallium	Se + Tl	. 758
184	Silicon + Germanium	Se + Ge	. 761
185	Silicon + Iron	Si + Fe	. 764
186	Silver + Antimony	Ag + Sb	. 767
187	Silver + Cadmlum	Ag + Cd	. 770
188	Silver + Copper	Ag + Cu	. 773
189	Silver + Gold	Ag + Au	. 774
190	Silver + Indium	Ag + In	. 777
191	Silver + Lead	Ag + Pb	. 780
192	Silver + Manganese	Ag + Mn	. 783
193	Silver + Palladium	Ag + Pd	- 786
194	Silver + Platinum	Ag + Pt	- 790
195	Silver + Tin	Ag + Sn	. 791
196	Silver + Zinc	Ag + Zn	. 792
197	Sodium + Mercury	Na + Hg	. 795
198	Sodium + Potassium	Na + K	. 798
199	Tantalum + Nioblum	Ta + Nb	. 801
200	Tantalum + Tungsten	Ta + W	. 802
201	Tellurium + Selenium	Te + Se	. 805
202	Tellurium + Thallium	Te + Tl	. 808
203	Thallium + Cadmium	Tl + Cd	. 811
204	Thallium + Indium	Tl + In	. 812
205	Thallium + Lead	Tl + Pb	. 815
206	Thallium + Tellurium	Tl + Te	. 818
207	Thallium + Tin	Ti + Sn	. 821
208	Thorium + Uranium	Th + U	. 622
209	Tin + Aluminum	8n + Al	. 823
210	Tin + Antimony	Sn + Sb	. 824
211	Tin + Bismuth	Sn + Bi	. 827
212	Tin + Cadmium	Sn + Cd	. 830
213	Tin + Copper	Sn + Cu	. 833
214	Tin + Indium	Sn + In	. 834
215	Tin + Lead	Sn + Pb	. 839
216	Tin + Mercury	Sn + itg	. 842
217	Tin + Silver	Sn + Ag	. 845
218	Tin + Thallium	Sn + Tl	. н46

Figure and/or

	Figure and/or Table No.	Name	Formula	Page No.
	219	Tin + Zinc	Sn + Zn	. 847
	220	Titanium + Aluminum	Ti + Al	. 848
	221	Titanium + Manganese	Ti + Mn	. 849
	222	Titanium + Oxygen	Ti + O	. 852
	223	Tungsten + Rhenium	W + Re	. 855
	224	Uranium + Aluminum	U + A1	. 858
	225	Uranium + Chromium	U + Cr	. 859
	226	Uranium + Iron	U + Fe	. 862
	227	Uranium + Magneaium	U + Mg	. 863
	228	Uranium + Molybdenum	U + Mo	864
	229	Uranium + Niobium	U + Nb	. 867
	230	Uranium + Silicon	U + Si	. 868
	231	Uranium + Zirconium	U + Zr	. 871
	232	Vanadium + Iron	V + Fe	. 874
	233	Vanadium + Yttrium	V + Y	. 877
	234	Zinc + Aluminum	Zn + Ai	. 880
	235	Zinc + Cadmium	Zn + Cd	. 881
	236	Zirconium + Aiuminum	Zr + Al ,	. 882
	237	Zirconium + Hafnium	Zr + Hf	. 883
	238	Zirconium + Niobium	Zr + Nb	. 886
	239	Zirconium + Tin	Zr + Sn	. 887
	240	Zirconium + Titanium	Zr + Ti	. 890
	241	Zirconium + Uranium	Zr + U	. 891
3.	NONFERROUS	MULTIPLE ALLOYS		
	242	Aluminum + Copper + ΣX_i	$Ai + Cu + \Sigma X_i$. 895
	243	Aluminum + Iron + ΣX_i	$Ai + Fe + \Sigma X_i$. 905
	244	Aluminum + Magnesium + ΣX_1	$Ai + Mg + \Sigma X_{i}$. 908
	245	Aluminum + Manganese + ΣX_i	$Ai + Mn + \Sigma X_i$. 91 <u>1</u>
	246	Aluminum + Nickei + ΣX_i	$Ai + Ni + \sum_{i} \dots \dots$. 914
	247	Aluminum + Silicon + ΣX_1	$Ai + Bi + \sum X_i $	
	248	Aluminum + Zinc + ΣX_i	$Ai + 2n + \Sigma X_i$. 922
	249	Aluminum + ΣX_1	$Ai + \Sigma X_i$	
	250	Antimony + Beryllium + \sum_{i}	Sb + Be + ΣX_i	
	251	Beryilium + Fluorine + ΣX_i	Be + \mathbf{F} + ΣX_1	
	252	Beryilium + Magnesium + ∑Xi	Be + Mg + ΣX_i	
	253	Bismuth + Cadmium + ΣX_i	$Bi + Cd + \Sigma X_1 \dots \dots$	
	254	Bismuth + Lead + ΣX ₁	$B_1 + P_5 + \Sigma X_1 \dots \dots \dots$	
	255	Cadmium + Bismuth + ∑X ₁	$Cd + Bi + \sum_{i} \dots \dots$	
	25G	Chromium + Iron + DX ₁	$Cr + Fe + \Sigma X_i$	
	257	Cobalt + Chromium + ΣX_i	$Co + Cr + \Sigma X_i$,	
	258	Cobalt + Iron + ΣX_i	$Co + Fe + \Sigma X_i$	
	259	Cobalt + Nickel + DX ₁	$Co + Ni + \Sigma X_i \dots \dots \dots \dots$	
	260	Copper + Aluminum + ΣX_i	$Cu + Al + \Sigma X_1 \dots \dots \dots \dots$. 952

3. NONFERROUS MULTIPLE ALLOYS (continued)

Figure and/or Table No.	Name	Formula	Page No.
261	Copper + Beryllium + ΣX_i	Cu + Be + ΣX_i	. 955
262	Copper + Cadmium + ∑Xi	$Cu + Cd + \Sigma X_i$	
263	Copper + Cobsit + ∑X,	$Cu + Co + \Sigma X_1 \dots \dots \dots$	
264	Copper + Iron + ΣX_i	$Cu + Fe + \Sigma X_i$	
265	Copper + Lead + ∑X,	$Cu + Pb + \Sigma X_1 \dots \dots \dots$	
266*	Copper + Manganese + ∑Xi	$Cu + Mn + \Sigma X_i$	*
267	Copper + Nickel + EX;	$Cu + Ni + \Sigma X_i$,	. 969
268	Copper + Silicon + ΣX,	$Cu + Si + \Sigma X_i$	
269	Copper + Tin + $\sum X_1$	$Cu + Sn + \Sigma X_1 \dots \dots \dots$	
270	Copper + Zinc + ΣX_i	$Cu + Zn + \Sigma X_1$	
271	Copper + Zirconium + ∑X _i	Cu + Zr + ΣX ₁	
272	Lanthanum + Neodymium + ∑X _i	La + Nd + ΣX _i	
273	Lead + Antimony + ∑X ₁	$Pb + Sb + \Sigma x_i$	
274	Lithium + Boron + ΣX_i	Li + B + ΣX ₁	
275	Lithium + Sodium + ΣX_1	Li + Na + $\sum_{i=1}^{n}$	
276	Magnesium + Aluminum + ΣX_i	$M_g + Al + \Sigma X_l$	
277	Magnesium + Cerium + ΣX,	Mg + Ce + ΣX ₁	
278	Magnesium + Cobalt + ΣX ₁	$Mg + Co + \Sigma X_i$	
279	Magnesium + Copper + ΣX,	Mg + Cu + ΣX _i	
280	Magnesium + Nickel + ∑X _i	$Mg + Ni + \Sigma X_i$	
281	Manganese + lron + ΣX _i	Mn + Fc + ΣΧ ₁	
282	Manganese + Silicon + ∑X _i	Mn + Si + ΣX ₁	
283	Molybdenum + Iron + ∑X,	$Mo + Fe + \Sigma X_1 \dots \dots \dots$	
284	Nickel + Aluminum + ΣX,	$Ni + Al + \Sigma X_1 \dots \dots$	
285*	Nickei + Chromium + ΣX ₁	$Ni + Cr + \Sigma X_i$	
286	Nickel + Cobalt + ∑X	$Ni + Co + \Sigma X_i$	
287	Nickei + Copper + DX	$Ni + Cu + \Sigma X_i$	
288	NickeI + Iron + ∑X ₁	$Ni + Fe + \Sigma X_i$	
289	Nickei + Manganese + ∑X;	$Ni + Mn + \Sigma X_1$	
290	Nickel + Molybdenum + ΣX _i	Ni + Mo + ΣΧ ₁	
291	Nickei + ∑X ₁	NI + ΣX ₁	
292	Niobium + Molybdenum + ∑X _i	Nb + Mo + ΣX_i	. 1046
293	Niobium + Tantalum + ΣX _i	Nb + Ta + ΣX_i ,	. 1049
294	Niobium + Titanium + ΣX_i	$Nb + Ti + \Sigma X_1 \dots \dots$. 1052
295	Niobium + Tungsten + ΣX_i	Nb + W + ΣX_1	. 1055
296	Silver + Cadmium + ΣX_i	$Ag + Cd + \Sigma X_i$. 1058
297	Sliver + ∑X _i	$Ag + \Sigma X_i$. 1061
298	Tantalum + Niobium + ΣX _i	$Ta + Nb + \Sigma X_i$. 1062
299	Tantalum + Tungsten + $\sum X_i$	$Ta + W + \Sigma X_1 \dots \dots \dots \dots$. 1065
300	Tellurium + Arsenic + EX	Te + A_6 + ΣX_i	. 1068
301	Tin + Antimony + ΣX _i	$\operatorname{Sn} + \operatorname{Sb} + \Sigma X_1 \dots \dots \dots \dots$. 1069
302	Tin + Copper + ΣX _i	$\operatorname{Sn} + \operatorname{Cu} + \Sigma X_{i}$. 1072
303	Titanium + Aluminum + ΣX_i	$Ti + Al + \Sigma X_i$. 1973
304	Titanium + Chromium + ΣX_i	$Ti + Cr + \Sigma X_1$, 1077
305	Titanium + Iron + EX	$Ti + Fe + \Sigma X_i$. 1080
306	Titanium + Manganese + ∑Xi	$Ti + Mn + \Sigma X_i$, 1083

Number marked with an asteriak indicates that recommended values are also reported for this material on separate figure and table of the same number followed by the letter R.

3. NONFERROUS MULTIPLE ALLOYS (continued)

	Figure and/or Table No.	Name	Formula		Page No.
	307	Titanium + Vanadium + ΣX _i	$Ti + V + \Sigma X_i$. 1086
	308	Titanium + ΣX_i	$Ti + \Sigma X_1$. 1089
	309	Tungsten + Iron + ΣX_i	$W + Fe + \Sigma X_i$. 1090
	310	Tungsten + Nickei + DX	$W + Ni + \Sigma X_i$. 1091
	311	Uranium + Molybdenum + ΣX_i	$U + Mo + \Sigma X_i$.		, 1094
	312	Uranium + Zirconium + ΣX _i	$U + Zr + \Sigma X_i$. 1097
	313	Zinc + Aluminum + ΣX ₁	$Zn + Ai + \Sigma X_i$. 1098
	314	Zinc + Lead + ΣX _i	$Zn + Pb + \Sigma X_i$. 1099
	315	Zirconium + Aiuminum + £X	$Zr + Al + \Sigma X_i$, 1100
	316	Zirconium + Hafnium + £Xi	$Zr + Hf + \Sigma X_i$. 1101
	317	Zirconium + Molybdenum + ΣX_i			
	318	Zirconium + Tantalum + ΣX_i			
	319	Zirconium + Tin + ΣX ₁			. 1108
	320	Zirconium + Uranium + ΣX;			. 1111
	321	Zirconium + ΣX _i			. 1112
41	FERROUS AL	STEELS			
	322	Iron + Carbon + ΣX_i	Fe + C + ΣX_1	Group 1	
	323	iron + Carbon + ΣX_i	Fe + C + ΣX_i	Group II	. 1124
	B. CAST IRO	DNS			
	324	Iron + Carbon + ΣX_i	$Fe + C + \Sigma X_1$	Group I	- 1125
	325	Iron + Carbon + ΣX_i	Fe + C + ΣX _i	Group II	. 1132
	C. ALLOY S	TEELS			
	326	Iron + Aluminum + $\sum X_i$	Fe + Al + ΣX _i	Group I	. 1142
	327	Iron + Aluminum + ΣX_i	Fe + Al + ΣX_1	Group II , , , , , , ,	. 1145
	328	Iron + Chromium + ΣX_1	Fe + Cr + ΣX_1	Group I ,	· 1148
	329	Iron + Chromium + ΣX_1	Fe + Cr + ΣX_i	Group II	. 1152
	330	Iron + Chromium + Nickel + ΣX_i	Fe + Cr + Ni + ΣX_1	Group 1 ,	. 1160
	331*	Iron + Chromium + Nickei + ΣX_1	Fe + Cr + Ni + ΣX_i	Group Il	. 1164
	332	Iron + Cobait + ΣX_i	Fe + Co + ΣX_i	Group II	. 1176
	333	Iron + Copper + ΣX_{i}	Fe + Cu + ΣX_1	Group I	. 1179
	334	Iron + Mangunese + ∑Xi	Fe + Mn + ΣX_1	Group I	. 1182
	235	Iron + Manganese + ΣX ₁	Fe + Mn + ΣX_i	Group II	. 1191
	336	Iron + Molybdenum + ∑Xi	Fe + Mo + ΣX _i	Group II	1194
	337	Iron + Nickel + ΣX_i	Fe + Ni + ΣX_1	Group I	. 1197
	338	Iron + Nickei + ΣX ₁	Fe + Ni + ΣX_i	Group II	. 1202
	339	iron + Nickei + Chromium + $\Sigma X_{\hat{I}}$	Fe : $NI + Cr + \Sigma X_1$	Group 1	. 1209
	340	Iron + Niekei + Chromium + ΣX_i	Fe + Ni + Cr + ΣX_i	Group Ii	. 1212
*					

^{*}Number murked with an asterisk indicates that recommended values are also reported for this material on squarate figure and table of the same number followed by the letter R.

4. FERROUS ALLOYS (continued)

Figure and/ Table No.		Formula								
C. ALLOY	STEELS (continued)									
341	Iron + Phosphorus + ∑X _i	Fe + P + Σx_i	Group I			1216				
342	Iron + Silicon + ΣX _i	Fe + Si + $\sum X_i$	Group I			1217				
343	Iron + Silicon + ΣX_i	Fe + Si + ∑X _i	Group II			1221				
344	Iron + Titanium + ΣX _i	Fe + Ti + ΣX_i	Group I			1225				
345	Iron + Tungsten + ΣX_i	Fe + W + Σx_i	Group I			1226				
346	Iron + Tungsten + ΣX_i	$Fe + W + \Sigma X_1$	Group II		•	1229				

5. INTERMETALLIC COMPOUNDS

Figure and/or Table No.	Formula																									i	Page No.
347	Sb ₂ Te ₃																							٠			1241
348	As ₁ Te ₃																										1244
349	Ba ₁ Pb																										1245
350	Ba ₂ Sn																										1246
351	Be _x Nb _y																										1247
352	Be _x Ta _y																										1250
353	Be _x U _y																										1253
354	Be ₁₂ Zr																										1256
355	Bi ₂ Te ₃																										1257
356	$\mathbf{B_xSi_y}$																						٠				1262
357	CdSb																				•						1264
358	CdTe				•																						1267
359	$Ca_{\chi}Pb_{\gamma}$																								•		1270
360	Ca ₂ Sn							•						•													1273
361	CoSi	•	•					•						•				•	•		•		•		•		1274
362	CuSbSe ₂		•		•	•					•				•			•			•						1275
363	Cu ₂ Se ₂				•							•			•	•		•			•						1276
364	GaAs	•						•	•		•	•			•		•						•				1277
366	GeTe	•	•		•	•					•	•			•	•	•	•	•	•	•				•		1280
366	Au _x Cu _y	•		•	•		•		•	•		•			•	•	•	•	•	•	•				•	•	1281
367	HfB ₂	•		•											•	•		•	•		•				•		1284
368	InSb	•			•	•	•		•		•	•		•	•	•		•	•		•				•		1287
369	Inās	•	•	•	•	•			•	•	•	•	•		•	•	•	•	•	•	•	•		•	•	•	1292
370	In ₂ Se ₂	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	1295
371	In ₂ Te ₂	•	•	•	•	•	•	•	•	•	•	•	•		•	•	•	•	•	•	•	•	•	٠	•	•	1298
372	LaSe	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	٠	•	•	•	•	•	•	•		•	1301
373	LaTe	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	1304
374	PbTe		•	•		•	•		•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	1307
376	Mg ₂ Sb ₂			•	٠		•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	1310
376	MgzGe		•	•		•	•	•	•	•		•	•	•	•	•	•	•	•	•	•			٠	•	•	1311
377	Mg ₂ Si		•	•		•	•	•		•				•		•	•	٠	•	•	•	•	•	•	•	•	1314
378	Mg ₂ Sn					•	•			•			•			•	•	•	•	•				•	•		1317

5. INTERMETALLIC COMPOUNDS (continued)

Figure and/or Table No.	Formula	Page No.
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380	HgTe	1321
381	MoSi ₂	1324
382	NiSb . ,	1327
383	Re ₂ As ₇	1330
384	RexGev	1331
385	ReSe ₂	1332
386	AgSbTe2	1335
387	AgCu	1338
388	Ag ₂ Se	1339
389	Ag _x Te _y	1342
390	Sr ₂ Si	1343
391	Sr ₂ Sn	1344
392	TaB ₂	1345
393	TaGe2	1348
394	Tl ₂ Pb	1349
395	SnSe ₂	1352
396	SnTe	1355
397	TiB ₂	1358
398	TiNi	1361
399	W ₃ As ₇	1364
400	WB	1365
401	WSe ₂	1368
402	wsi ₂	1369
403	WTe ₂	1370
404	ZnSe	1371
405	ZnSiAs ₂	1374
406	ZrB	1375
	RMETALLIC COMPOUNDS	
407	Sb ₂ Se ₂ + Ag ₂ Se + PbSe	1379
408	Sb ₂ Te ₃ + Bi ₂ Te ₃	1380
409 410	Sb ₂ Tc ₃ + In ₂ Tc ₃	1386
411	Bi ₂ Tc ₃ + Sb ₂ Te ₃	1388
411	$Bi_2Te_3 + Sb_2Te_3 + Sb_2Se_3$	1392
413	$Bi_2Te_3 + Bi_2Se_3 \qquad \dots \qquad \dots \qquad \dots$	1393
414	Cd ₃ As ₂ + Zn ₃ As ₂	1396
414	CdSb + ZnSb	1397
416	Out Book of the	1400
417		1401
418	To Month Ou To A Au To	1403
419	NoTe + CdTe	1406
420		1407
440	AgSbTe ₂ + SnTe	1410

6. MIXTURES OF INTERMETALLIC COMPOUNDS (continued)

	Table No.	Formula																									Puge No
	421	SnTe + AgSh	Τe	2 -							•		4														1411
	422	ZnSb + CdSb)	٠	•	٠	•	•	•	•	•	•	٠	•	•	•	•	•	•	•	•	•	•	•	٠	•	1412
7,	MISCELLANEOU	s alloys and m	ίχ	TU	JRI	s																					
	423	Bi ₂ Te ₃ + Te																									1415
	424	Be + BeO																									1416
	425	$Cr + Al_2O_3$																									1419
	426																										1420
	427	GaAs + GaP																									1423
	428	InAs + InP																					٠				1426
	429	$Mo + ThO_2$																									1429
	430	Na + Na ₂ O						,																			1432
	431	TiNi + Cu																									1433
	432	TiN1 + Ni																									1436
	433	W + ThO2																									1439
	434	U + UOz								•																	1442
	435	$Zr + ZrO_2$																									1444

VOLUME 2. THERMAL CONDUCTIVITY - NONMETALLIC SOLIDS

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MATERIAL INDEX

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GROUPING OF MATERIALS AND LIST OF FIGURES AND TABLES

1. ELEMENTS

Figure and/or Table No.	Name	Symbol	l												Page No.
1*	Boron	В													1
2	Carbon	C										•			5
3*	Carbon (Diamond)	С													9
	Carbon (Graphite)	C													
4	AGOT Graphite									•					13
5*	ATJ Graphite														20
6	AWG Graphite														24
7*	Pyrolytic Graphite														30
8	SA-25 Graphite			•		•				•					42
9*	875S Graphite								•						45
10*	890S Graphite						-				•				49
11	Miscellaneous Graphite								•						53
12*	Iodine	1													83
13%	Phosphorus	P											٠		86
14*	Sulfur	S													89

2. SINGLE OXIDES

Figure and/or Table No.	Name	Formula		P	age No.
15*	Aluminum Oxide (Sapphire)	Al_2O_3	٠		93
16*	Aluminum Oxide	Al ₂ O ₃ , , , , , , , , , , , , , , , , , , ,			98
17	Barium Oxide	BaO			120
18*	Beryllium Oxide	BeO			123
19	Boron Oxide	B ₂ O ₃			138
20	Calcium Oxide	CaO		•	141
21	Cerium Dioxide	CeO ₂			144
22	(di)Copper Oxide	Cu ₂ O			147
23	Hafnium Dioxide	HfO ₂			150
24	Indium Oxide	InO			153
25	(tri)Iron Tetraoxide	Fe ₃ O ₄			154
26	Lithium Oxide	Li ₂ O			157
27*	Magnesium Oxide	MgO			158
28	Manganese Oxide	MnO			168
29	(tri)Manganese Tetraoxide	Mn ₃ O ₄			170
30	Nickel Oxide	NiO			171
31*	Silicon Dioxide (Crystalline)	SiO_2			174
32*	Silicon Dioxide (Fused)	SiO ₂			183
33	Strontium Oxide	SrO			194
34*	Thorium Dioxide	${ m ThO}_2$, , , ,			195

Number marked with an asterisk indicates that recommended values are also reported for this material on separate figure and table of the same number followed by the letter R.

2. SINGLE OXIDES (continued)

	Figure and/or Table No.	Name	Formula													F	age No.
	35	Tin Dioxide	SnO_2														199
	36*	Titanium Dioxide	TiO ₂							-			,				202
	37	Tungsten Trioxide	wo,														209
	38	Uranium Dioxide	UO ₂														210
	39	(tri)Uranium Octoxide	U ₃ O ₈									٠					2:37
	40	Yttrium Oxide	Y ₂ O ₃														240
	41	Zinc Oxide	ZnO														243
	42	Zirconium Dioxide	ZrO ₂	•	•	•	•	•	•	•	•	•	•	٠	•	•	246
3.	OXIDE COMPO	ounds															
	43	Aluminum Fluosilicate (Topaz)	2AiFO ∙ SiO ₂														251
	44	Aluminum Silicate (Mullite)	3Al ₂ O ₃ · 2SiO ₂								-						254
	45	Barium Metatitanate	BaTiO ₃														257
	46	Barium Dititanate	BaO · 2TiO2														260
	47	Bismuth Stannate	$Bi_2(SnO_3)_3$														261
	48	Calcium Stannate	CaSnO ₃														264
	49	Calcium Metatitanate	CaTiO ₃														267
	50	Calcium Tungstate	CaWO ₄														270
	51	(tri)Cobalt Strontium Metatitanate	Co ₂ SrTiO ₃														271
	52	Cobalt Zinc Ferrate	Co(Zn)Fe ₂ O ₄														272
	53	Forsterite	Mg2SiO4														275
	54	Garnet	$[M_{1}^{II}M_{1}^{III}(SiO_{4})_{3}]$														278
	55	Lead Metatitanate	PbTiO ₃														279
	56	Lead Zirconate	PbZrO,														282
	57	Magnesium Aluminate	MgO · Al ₂ O ₃														283
	58	Magnesium Aluminate	MgO · 3. 5Al ₂ O ₃														286
	59	Magnesium Stannate	MgSnO ₃														289
	60	Manganese Ferrate	MnFe ₂ O ₄														292
	61	Manganese Zinc Ferrate	Mn(Zn)Fe ₂ O ₄														295
	62	Nickel Zinc Ferrate	$Ni(Zn)Fe_2O_4$														298
	63	Sodium Tungsten Bronze	Na _x WO ₃														108
	64	Strontium Metatitanate	SrTiO3														304
	65	Strontium Zirconate	SrZrO3														307

YaAlaO₁₂

 $ZnFe_2O_4$

ZrSiO4

 $Y_3Fc_2(FeO_4)_3$

308

311

314

317

.

Yttrium Aluminate

Zirconium Orthosilicate

Yttrium Ferrate

Zinc Ferrate

66

67

68

69

Number marked with an asterisk indicates that recommended values are also reported for this material on separate figure and table of the same number followed by the letter R.

4. BINARY MIXTURES OF SINGLE OXIDES AND/OR OXIDE COMPOUNDS

Figure and/or Table No.	Name		Page No.
70	Aluminum Oxide + Aluminum Silicate	$Al_2O_3 + 3Al_2O_3 \cdot 2SiO_2$	321
71	Aluminum Oxide + (di)Chromium Trioxide	$Al_2O_3 + Cr_2O_3$	324
72	Aluminum Oxide + (di)Manganese Trioxide	$Al_2O_3 + Mn_2O_3$	327
73	Aluminum Oxide + Silicon Dioxide	$Al_2O_3 + SiO_2$	328
74	Aluminum Oxide + Zirconium Dioxide	$Al_2O_3 + ZrO_2$	331
75	Aluminum Silicate + Aluminum Oxide	$3Al_2O_3 \cdot 2SiO_2 + Al_2O_3$	334
76	Barium Oxide + Strontium Oxide	BaO + SrO	337
77	Barium Metatitanate + Calcium Metatitanate	BaTiO ₃ + CaTiO ₃	340
78	Barium Metatitanate + Magnesium Zirconate	$BaTiO_3 + MgZrO_3$	343
79	Barium Metatitanate + Manganese Niobate	$BaTiO_3 + Mn_2Nb_2O_7$	344
80	Beryllium Oxide + Uranium Dioxide	BeO + UO ₂	347
81	Cerium Dioxide + Magnesium Oxide	$CeO_2 + MgO$	350
82	Cerium Dioxide + Uranium Dioxide	$CeO_2 + UO_2$	353
83	Gadolinium Oxide + Samarium Oxide	Gd ₂ O ₃ + Sm ₂ O ₃	356
84	Lead Oxide + Silicon Dioxide	PbO + SiO ₂	359
85	Magnesium Aluminate + Magnesium Oxide	$MgO \cdot Al_2O_3 + MgO$	362
86	Magnesium Aluminate + Silicon Dioxide	$MgO \cdot Al_2O_3 + SiO_2$	365
87	Magnesium Aluminate + (di)Sodium Oxide	$MgO \cdot Al_2O_3 + Na_2O$	368
88	Magnesium Oxide + Beryllium Oxide	MgO + BeO	371
89	Magnesium Oxide + Clay	MgO + Clay	374
90	Magnesium Oxide + Magnesium Aluminate	$MgO + MgO \cdot Al_2O_3$	375
91	Magnesium Oxide + Magnesium Orthosilicate	$MgO + 2MgO \cdot SiO_2$	378
92	Magnesium Oxide + Niekel Oxide	MgO + NiO	38 1
93	Magnesium Oxide + Silicon Dioxide	MgO + SiO ₂	384
94	Magnesium Oxide + Tin Dioxide	MgO + SnO ₂	387
95	Magnesium Oxide + Uranium Dloxide	MgO + UO ₂	390
96	Magnesium Oxide + Zinc Oxide	MgO + ZnO	391
97	Magnesium Orthosilicate + Magnesium Oxide	$2MgO \cdot SiO_2 + MgO$	394
98	(di)Manganese Trioxide + Aluminum Oxide	$Mn_2O_3 + Al_2O_3$	397
99	(di)Manganese Trioxide + Magnesium Oxide	$Mn_2O_3 + MgO$	398
100	(di)Manganese Trioxide + Silicon Dioxide	Mn ₂ O ₃ + SiO ₂	399
101	Silicon Dioxide + Aluminum Oxide	$SiO_2 + Al_3O_3$	402
102	Silicon Dioxide + Calcium Oxide	SiO ₂ + CaO	407
103	Silicon Dioxide + (di)Iron Trioxide	$SiO_2 + Fe_2O_3$	410
104	Thorium Dioxlde + Uranium Dioxide	$ThO_2 + UO_2$	413
105	Tin Dloxide + Magnesium Oxide	SnO ₂ + MgO	416
106	Tin Dioxide + Zinc Oxide	$SnO_2 + ZnO$	419
107	Tungsten Trioxide + Zinc Oxide	WO ₃ + ZnO	422
108	Uranium Dioxide + Beryllium Oxide	UO ₂ + BeO	423
109	Uranium Dioxide + Calcium Oxide	UO ₂ + CaO	426
110	Uranium Dioxide + (di)Niobium Pentoxide	UO ₂ + Nb ₂ O ₆	427
111	Uranium Dioxide + Yttrium Oxide	$UO_2 + Y_2O_1$	428
112	Uranium Dioxide + Zirconium Dioxide	UO ₂ + ZrO ₂	429
113	Yttrium Oxide + Uranium Dloxide	Y ₂ O ₃ + UO ₂	432
114	Zinc Oxide + Magnesium Oxide	ZnO + MgO	435
115	Zine Oxide + Tin Dioxide	$ZnO + SnO_2$	438
		=	

4. BINARY MIXTURES OF SINGLE OXIDE AND/OR OXIDE COMPOUNDS (continued)

Figure and/or Table No.	. Name	Formula	Page No.
116	Zirconium Dioxide + Aluminum Oxide	$ZrO_2 + Al_2O_3$. 441
117	Zirconium Dioxide + Calcium Oxide	ZrO ₂ + CaO	. 442
118	Zirconium Dioxide + Magnesium Oxide	$ZrO_2 + MgO$. 446
119	Zirconium Dioxide + Yttrium Oxide	$ZrO_2 + Y_2O_3$. 449
5. MULTIPLE M	MEXTURES OF SINGLE OXIDES AND/OR OXIDE COM	POUNDS	
120	Aluminum Oxide + Silicon Dioxide + ΣX_i	$Al_2O_3 + SiO_2 + \Sigma X_1$. 453
121	Aluminum Oxide + Titanium Dioxide + $\sum X_i$	$Al_2O_3 + TiO_2 + \Sigma X_1$	456
122	Barium Oxide + Silicon Dioxide + ΣX_i	BaO + SiO ₂ + ΣX_i	457
123	Barium Oxide + Strontium Oxide + ΣX_i	BaO + SrO + ΣX_i	. 460
124	Beryllium Oxide + Aluminum Oxide + ΣX_i	$BeO + Al_2O_3 + \Sigma X_1 . . .$. 461
125	Beryllium Oxide + Magnesium Oxide + ΣX_i	BeO + MgO + ΣX_i	. 464
126	Beryllium Oxide + Thorium Dioxide + ΣX_1	BeO + ThO ₂ + ΣX_1	467
127	Beryllium Oxide + Zirconium Dioxide + ΣX_1	BeO + $ZrO_2 + \Sigma X_1$. 470
128	(di)Chromium Trioxide + Magnesium Oxide + ΣX_1	$Cr_2O_3 + MgO + \Sigma X_i$	473
129	Lead Oxide + Silicon Dioxide + ΣX_i	PbO + SiO ₂ + ΣX_i	- 474
130	Magnesium Oxide + Calcium Oxide + ΣX ₁	$MgO + CaO + \Sigma X_i$	• 477
131	Magnesium Oxide + (di)Chromium Trioxide + ΣX _i	$MgO + Cr_2O_3 + \Sigma X_i$,	- 480
132	Magnesium Oxide + (di)Iron Trioxide + ΣΧ ₁	$MgO + Fe_2O_3 + \Sigma X_i$,	. 483
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135	Silicon Dioxide + Barium Oxide + EX	$SiO_2 + BaO + \Sigma X_1$. 495
136	Silicon Dioxide + Boron Oxide + EX	$SiO_2 + B_2O_3 + \Sigma X_1$. 498
137	Silicon Dioxide + Calcium Oxide + $\sum X_i$	$SiO_2 + CaO + \Sigma X_i$. 501
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140	Silicon Dioxide + (di)Sodium Oxide + ΣX_i	$SiO_2 + Na_2O + \Sigma X_1$	
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143	Strontium Oxide + Titanium Dioxide + EXi	$SrO + TiO_2 + \Sigma X_i$	- 517
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145	Tin Dioxide + Magnesium Oxide + ΣX_i	$SnO_2 + MgO + \Sigma X_1$	• 523
146	Tin Dioxide + Zinc Oxide + ΣX_i	$SnO_2 + ZnO + \Sigma X_1$. 524
147	Zinc Oxide + Strontium Oxide + ΣX_i	$ZnO + SrO + \Sigma X_i$	• 527
148	Zinc Oxide + Tin Dioxide + ΣX_i	$ZnO + SnO_2 + \Sigma X_1$	• 528
149	Zirconium Dioxide + Caleium Oxide + ΣX_i	$ZrO_1 + CaO + \Sigma X_1$	- 531
150	Zirconium Dioxide + Silicon Dioxide + ΣX _i	$ZrO_2 + SiO_2 + \Sigma X_i$	
161	Zirconium Dioxide + Yttrium Oxide + ΣX_1	$ZrO_2 + Y_2O_3 + \Sigma X_1$	

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	152	(tetra)Boron Carbide + Sodium Me	tasilica	te				B	; +	Na	ųO	٠ s	iO ₂							541
	153	Graphite + Thorium Dioxide						C i	T	hO	1									544
	154	Graphite + Uranium Dioxide						C 4	U	02										547
	155	Magnesium Oxide + Tale						Mg	0 4	Н	M	53(S	iO,)4						550
	156	Silicon Carbide + Silicon Dioxide						SiC	+	Si),									553
	157	Silicon Carbide + Silicon Dioxide +	ΣX_1					SiC	+	Sic	+ در	Σ	x,							554
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	229	Strontium Metatitanate + Cobalt	SrTiO ₃ + Co		. 7	22
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	259	Silicon Carbi	de + Gr	aphite	3				SiC	+ C	: .											789
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	320	Diphenyl					C	6H	ьCе	Н															•			•	១ឧប
	321	Diphenyi Ox	ide				((الها	5)2(υ						•	•			•		•		•	•				990
	322	Diplieny lam:	ine				(C	eH	5) ₂ 1	IIN					•	•	•		•	•	•		•	•			•	•	991
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	324	Naphthalene	;				C	10l	ie				•	•		•	•	•		•	•	•	•	•	•	•	•	•	995
	325	Naphthol					C	2 ₁₀ £	I ₇ O	Ηl			•	•		•	•	٠	٠	٠	•		•	•	•	•	•	•	99 8
	326	Nitrophenol					N	102	C_{6}	I ₄ C	H		•	٠	•	•	•	•	•	•	•	•	•	٠	٠	٠	٠	٠	1001
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GROUPING OF MATERIALS AND LIST OF FIGURES AND TABLES

1. ELEMENTS

	Figure and/or Table No.	Nume	Symbol	Physical State*									£	Page No.
	1	Argon	A	S, L, V, G										1
	2	Bromine	Br ₂	-, L, V, G										13
	3	Chlorine	Cl ₂	-, L, V, G						•	•			17
	4	Deuterium	D ₂	-, L, V, G						٠		•	•	21
	5	Fluorine	F ₂	-, L, V, G									-	26
	6	Helium	He	S, L, '-, G					•			•	•	29
	7	Hydrogen	H ₂	S, L, V, G									•	41
	8	Krypton	Kr	S, L, V, G										50
	9	Neon	Ne	S, L, V, G										56
	10	Nitrogen	N ₂	S, L, V, G										64
	11	Oxygen	O ₂	-, L, V, G										76
	12	Radon	Rn	-, L, V, G				•						64
	13	Tritium	Tą	-, L, ~, -	 •							•		87
	14	Xenon	Xe	S, L, V, G	 •	•		•	٠	•	•	•	٠	88
2.	15 16 17 18 19 20 21 22 23 24	Ammonia Boron Trifluoride Hydrogen Chloride Hydrogen lodide Hydrogen Sulfide Nitric Oxide Nitrogen Peroxide Nitrous Oxide Sulfur Dioxide Water	NH ₃ BF ₃ HC1 HI H ₂ S NO NO ₂ N ₂ O SO ₂ H ₂ O	-, L, G, - -, -, G, - -, -, G, - -, -, G, - -, -, G, - -, L, G, - -, L, G, - -, L, G, -										95 99 101 103 104 106 108 114 116 120
3.	ORGANIC COM	POUNDS												
	25	Acetone	(CH ₃) ₂ CO	-, L,G, -		•				•				129
	26	Acetylene	СНСН	-, -, G, -		٠		•	•		•	•	•	133
	27	Benzene	Celie	-, L, G, -	 •		•	•	•	•	•	•		135
	28	i-Butane	i-C ₄ H ₁₀	-, -, G, -		•		•	•	•				139
	29	n-Butane	n-C4H46	-, L, G, -								•		141
	30	Carbon Dioxide	CO2	-, L, G, -	 •	•	•		•		•	٠	•	145

^{*}S = solid, L = saturated liquid, V = saturated vapor, G = gas.

3. ORGANIC COMPOUNDS (continued)

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31	Carbon Monoxide	CO	L, G												151
32	Carbon Tetrachloride	CCL	$\mathbf{L}_{\mathbf{r}}\mathbf{G}$												156
33	Chloroform	СИСІ <mark>3</mark>	L, G												161
34	n-Decane	C ₁₀ H ₂₂	L, G												164
35	Ethane	C ₂ H _€	-, G												167
36	Ethyl Alcohol	C₂H₅OH	L, G												169
37	Ethylene	CH ₂ CH ₂	L, G												173
38	Ethylene Glycol	СН-ОНСН-ОН	L, -												177
39	Ethyl Ether	$(C_2H_5)_{\overline{a}}O$	L, G												179
40	Freon 11	Cl _f CF	L, G		•										183
41	Freon 12	Cl ₂ CF ₂	L, G												187
42	Freon 13	CICF3	-, G		•			•							191
43	Freon 21	Cl ₂ CHF	L, G												193
44	Freon 22	CICHF ₂	L, G					•		•			•	•	197
45	Freon 113	CCI_FCCIF1	L, G	•	•			•		•	•			•	201
16	Freon 114	CCIF ₂ CCIF ₂	L, G	•	•			•	•			•	•		205
47	Glycerol	сн₄онснонсн₄он	L, -	•	•			•		•	•			•	209
48	n-Heptane	C _T H _M	L, G											•	211
49	n-Hexane	C _e H _M	L, G					•		•				٠	214
50	Methane	CH₄	L,G	•				•	•	•					218
51	Methyl Alcohol	сн₃он	L, G	٠	•										223
52	Methyl Chloride	CH ₂ CI	L, G							•					227
53	n-Nonane	C ₀ H ₂₀	L, G		•										230
54	n-Octane	C _a H ₁₈	L, G				, ,	•						•	233
55	n-Pentane	C ₈ H ₁₂	L, G	•			,								236
56	Propane	C ₃ H ₈	~, G	•				•	•	•					240
57	Toluene	C ₄ H ₈ CH ₃	L, G	•		,		•	•	•	•	•	-		242
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59	Argon and Neon	Ar and Ne	-, a					u				,		,	258
60	Argon and Krypton	Ar and Kr	-, G						٠						263

4.

58	Argon and Helium	Ar and He	-,G	•						•	251
59	Argon and Neon	Ar and Ne	-, a						,		258
60	Argon and Krypton	Ar and Kr	-, G					٠			263
61	Argon and Xenon	Ar and Xe	-, G					,			267
62	Helium and Neon	He and Ne	-, G					٠			271
63	Helium and Krypton	He and Kr	-, G								276
64	Helium and Xenon	He and Xe	-, G							٠	280
65	Krypton and Neon	Kr and Ne	-,G .		٠						284
66	Krypton and Xenon	Kr and Xe	-,G	٠						•	288
67	Neon and Xenon	Ne and Xa	G								291

L = saturated liquid, G = gas.

4. BINARY SYSTEMS (continued)

B. Monatomic - Nonpolar Polyatomic Systems

Figure and/or Table No.	Name	Formula	Physical State*	Page No.
68	Argon and Benzene	Ar and CoHe	$G \cdot \cdot \cdot \cdot$	 295
69	Argon and Carbon Dioxide	Ar and CO1	G	 297
70	Argon and Deuterium	Ar and Da	G	 299
71	Argon and Hydrogen	Ar and Ha	G	 301
72	Argon and Methane	Ar and CH4	G	 304
73	Argon and Nitrogen	Ar and Na	G	 306
74	Argon and Oxygen	Ar and O2	G	 311
7 5	Argon and Propane	Ar and C ₃ H ₈	G	 316
76	Helium and Air	He and Air	G	 318
77	Helium and n-Butane	He and C4H	G	 320
78	Helium and Carbon Dioxide	He and CO2	G	 322
79	Helium and Cyclopropane	He and C ₃ H ₆	G	 325
80	Helium and Deuterium	He and D ₂	G	 327
81	Helium and Ethane	lle and C2H4	G	 329
82	Helium and Ethylene	ile and C2H4	G	 331
83	Helium and Hydrogen	He and H ₂	G	 333
84	Helium and Methane	He and CH4	G	 338
85	Helium and Nitrogen	He and N ₂	G	 340
86	Helium and Oxygen	He and O2	G	 343
87	Helium and Propane	He and CaHe	G	 345
88	Helium and Propylene	He and C ₃ H ₆	G	 347
89	Krypton and Deuterium	Kr and D ₂	G	 349
90	Krypton and Hydrogen	Kr and H ₂	G	 351
91	Krypton and Nitrogen	Kr and N ₂	G	 354
92	Krypton and Oxygen	Kr and O	G	 356
93	Neon and Carbon Dioxide	Ne and CO2	G	 358
94	Neon and Deuterium	Ne and D2	G	 360
95	Neon and Hydrogen	Ne and H ₂	G	 362
96	Neon and Nitrogen	Ne and N2	G	 365
97	Neon and Oxygen	Ne and O2	G	 368
98	Xenon and Deuterium	Xe and D2	G	 371
99	Xenon and iiydrogen	Xe and H ₂	G	 374
100	Xenon and Nitrogen	Xe and N ₂	G	 377
101	Xenon and Oxygen	Xe and O ₂	G	 379
C. Nonpolar Po	olyatomic - Nonpolar Polyatomic Syst	ems		
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103	Air and Carbon Monoxide	Air and CO	G	 383
104	Air and Methane	Air and CH4	G	 385
105	Benzene and Hexane	C ₆ H ₆ and C ₆ H ₁₄	G	 387
106	Carbon Dioxide and Ethylene	CO2 and C2H4	Q	 389
107	Carbon Dioxide and Hydrogen	CO ₁ and H ₁	G	 391
108	Carbon Dioxide and Nitrogen	CO ₂ and N ₂	G	 396
109	Carbon Dioxide and Oxygen	CO ₂ and O ₂	G	 401
110	Carbon Dioxide and Propane	CO ₂ and C ₂ H ₂	G	 403

G = gas

4. BINARY SYSTEMS (continued)

C. Nonpolar Polyatomic - Nonpolar Polyatomic Systems (continued)

D ₂ and H ₂ G	405 407 410 413 415 417 419
113 Deuterium and Nitrogen D ₂ and N ₂ G	410 413 415 417
	413 415 417
	415 417
114 Ethylene and Hydrogen C ₂ H ₄ and H ₂ G	417
115 Ethylene and Methane C ₂ H ₄ and CH ₄ G	
116 Ethylene and Nitrogen C ₂ H ₄ and N ₂ G · · · · ·	419
117 Hydrogen and Nitrogen H ₂ and N ₂ G	
118 Hydrogen and Nitrous Oxide H ₂ and N ₂ O G	427
119 Hydrogen and Oxygen H ₂ and O ₂ G	429
120 Methane and Propane CH4 and C3H8 G	432
121 Nitrogen and Oxygen N ₂ and O ₂ G	434
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139 Ethunol and Methyl Formate C ₂ H ₆ O and C ₂ H ₄ O ₂ G	474
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141 Helium-Argon-Xenon He-Ar-Xo G	479
142 Holium-Krypton-Xenon He-Kr-Xe G	480
143 Helium-Argon-Krypton He-Ar-Kr G	481
144 lielium-Neon-Xenon lie-Ne-Xe G	482
145 Argon-Krypton-Xenon Ar-Kr-Xe G	483

G = gas.

5.

5. TERNARY SYSTEMS (continued)

B.	Monatomic	and	Nonvolar	Polyatomic	Systems
----	-----------	-----	----------	------------	---------

	ure and/or able No.	Name	Formula	Physical State*							1	'age No.
	146	Helium-Oxygen-Methane	He-O ₂ -CH ₄	G								484
	147	Argon-Oxygen-Methane	Ar-Oz-CH4	G								485
	148	Helium-Argon-Nitrogen	He-Ar-N ₂	G	-							486
	149	Helium-Nitrogen-Methane	He-N2-CII4	G								467
	150	Argon-Krypton-Deuterium	Ar-Kr-D2	G								488
	151	Helium-Neon-Deuterium	He-Ne-D ₂	G								489
	152	Neon-Argon-Douterium	Ne-Ar-D2	G								490
	153	Neon-Krypton-Deuterium	Ne-Kr-D ₂	G								491
	154	Neon-Hydrogen-Oxygen	Ne-H ₂ -O ₂	G								492
	155	Argun-Hydrogen-Nitrogen	Ar-H ₂ -N ₂	G						٠.		493
	156	Neon-Hydrogen-Nitrogen	Ne-H ₂ -N ₂	G								494
	157	Neon-Nitrogen-Oxygen	Ne-N ₂ -O ₂	G								495
	158	Argon-Krypton-Hydrogen	Ar-Kr-H ₂	G	•	•	•	•			•	496
c.	Nonpolar Po	lyatomic Systems										
	159	Nitrogen-Oxygen-Carbon Dioxide	N2-O2-CO2	G								497
	160	Hydrogen-Nitrogen-Oxygen	H ₂ -N ₂ -O ₂	G		•	٠			٠	•	498
D.	Nonpolar and	1 Polar Systems										
	161	Argon-Propane-Ethanol	Ar-C ₂ H ₆ -C ₂ H ₆ O	G		•						499
	162	Hydrogen-Nitrogen-Ammonia	H ₂ -N ₂ -NH ₃	G	•	•	٠	٠	•	٠	•	500
. QUA	ATERNARY S	YSTE MS										
A.	Monatomic 8	lystems										
	163	Neon-Argon-Krypton-Xenon	Ne-Ar-Kr-Xe	G				•				504
В.	Monatomic a	and Nonpolar Polyatomic Systems										
	164	Argon-Krypton-Xenon-Hydrogen	Ar-Kr-Xe-Ha	G								505
	165	Argon-Krypton-Xenon-Deuterium	Ar-Kr-Xe-D2	G								506
	166	Argon-Hydrogen-Deuterium-Nitrogen	Ar-H ₂ -D ₂ -N ₂	G								507
	167	Argon-Hydrogen-Nitrogen-Oxygen	$Ar-H_2-N_2-O_2$	G								508
	168	Neon-Argon-Hydrogen-Nitrogen	Ne-Ar-H ₂ -N ₂	G								509
	169	Argon-Xenon-Hydrogen-Deuterlum	Ar-Xe-H ₂ -D ₂	G	•				•			510
. MU	LTICOMPON	ent systems										
	170	Alr		G								512

^{*}G = gas.

VOLUME 4. SPECIFIC HEAT - METALLIC ELEMENTS AND ALLOYS

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GROUPING OF MATERIALS AND LIST OF FIGURES AND TABLES

1. ELEMENTS

Figure and/or Table No.	Name	Symbol																			Page No.
1	Aluminum	Al																			1
2	Antimony	Sb																			6
3	Arsenic	As																			ម
4	Barium	Ва								•											13
5	Beryllium	Be													•						16
6	Bismuth	Bi							•						•						21
7	Boron	В								•											25
8	Cadmium	Cd																			29
9	Calcium	Ca												•			•				32
10	Cerium	Ce	•						•												36
11	Cesium	Cs																•			40
12	Chromium	Cr																			44
13	Cobalt	Co																			48
14	Copper	Cu																			51
15	Dysprosium	Dy																			62
16	Erbium	Er																			65
17	Europium	Eu																	,		68
18	Gadolinium	Gd																			72
19	Gallium	Gu																			75
20	Germanium	Ge																			79
21	Gold	Au																			833
22	Hafnium	Hſ																			87
23	Holmium	Но																			90
24	Indium	In																			95
25	Iridium	Ir											•								99
26	lron	Fe									4					•					102
27	Lanthanum	La																			110
28	Lead	Pb																			113
29	Lithium	Li	•	•					•						•	•					117
30	Lutetium	Lu	•												•						121
31	Magnesium	Mg	•	•	•	•		•	•	•	•	•	•	•	•	•	•	•		•	124
32	Manganese	Mn					•			•	•		•	•	•	•					127
33	Mercury	lig	•	•	•		•	•	•	•	•	•	•			•	٠	•	•		131
34	Molybdenum	Mo		•			•	•							•	•					135
35	Neodym i um	Nd		•			•								٠			-			140
36	Neptunium	Nр											•		•				•	•	143
37	Nickel	Ni								•	•					•			•		146
38	Niobium	Nb					-			•											153
39	Osmium	Ou																			157
40	Palladium	Pd																			160

1. ELEMENTS (continued)

Figure and/or Table No.	Name	Symbol																	Page No.
41	Platinum	Pt																	163
42	Plutonium	Pu																	167
43	Potassium	ĸ																	171
44	Praseodymium	\mathbf{Pr}								•				•					177
45	Rhenium	Re		•			•	•			•		•	•	•	•		•	181
46	Rhodium	Rh					•								•		•	•	184
47	Rubidium	Rb			•		•	•					•	•	•	•	•	•	187
48	Ruthenium	Ru			•		٠				•			•	•	•		•	190
49	Samarium	Sm		•		•	•	•			•				•	•	•		193
50	Scandium	Se		•	•			•				•				•	•		198
51	Selenium	Se												•					201
52	Silicon	Si												•				•	204
53	Silver	Ag																	208
54	Sodium	Na																	213
55	Strontium	Sr			•		•				•								218
56	Tantalum	Ta		•														-	221
57	Tellurium	Te																	229
58	Terbium	Tb																	232
59	Thallium	Tl										•		•			•		237
60	Thorium	Th												•					242
61	Thulium	Tm		•		•	•		•	•	•								245
62	Tin	Sn		•													•		249
63	Titanium	Ti					•												257
64	Tungsten	w					•									•			263
65	Uranium	ប																	268
66	Vanadium	v	•	•									•	•					271
67	Ytterblum	YЪ																•	274
68	Yttrium	Y																	278
69	Zinc	Ζŋ		•				. •		•									281
70	Zirconium	Zr		•	•		•	•			•		•	•	•		•		287

(. .

2. NONFERROUS BINARY ALLOYS

Figure and/or Table No.	Name	Formula										Page No.
71	Bismuth + Lead	Bi + Pb										291
72	Cadmium + Magnesium, MgCd	Mg + Cd										294
73	Cadmium + Magnesium, Mg Cd	Mga + Cd										297
74	Cadmium + Magnesium, MgCd ₃	Mg + Cd ₂		•								300
75	Chromium + Aluminum	Cr + Al	•						•			304
76	Chromium + Iron	Cr + Fe										307
7 7	Chromium + Manganese	Cr + Mn										311
78	Cobalt + Dysprosium	Co + Dy										314
79	Cobalt + Iron	Co + Fe										317
80	Cobalt + Nickel	Co + NI										320
81	Copper + Aluminum	Cu + Al			•					•		323

2. NONFERROUS BINARY ALLOYS (continued)

Figure and/or Table No.	Name	Formula													Page No.
82	Copper + Gallium	Cu + Ga													327
83	Copper + Iron	Cu + Fe													331
84	Copper + Magnesium	Cu + Mg													335
85	Copper + Manganesc	Cu + Mn													338
86	Copper + Nickel	Cu + Ni													341
87	Copper + Zinc	Cu + Zn								•					346
88	Gold + Nickel	Au + Ni	•												353
89	Hasnium + Zirconium	Hf + Zr													356
90	Indium + Tin	In + Sn				•		•							359
91	Lead + Tin	Pb + Sn						•	•						362
92	Lithium + Magnesium	Li + Mg									•				366
93	Magnesium + Silicon	Mg + Si				•					•				369
94	Manganese + Aluminum	Mn + Al							•				•		372
95	Manganese + Copper	Mn + Cu									•			-	377
96	Manganese + Nickel	Mn + Ni							•	,					380
97	Molybdenum + Titanium	Mo + Ti													383
98	Molybdenum + Tungsten	Mo + W													386
99	Nickel + Aluminum	Ni + Al													389
100	Nickel + Chromlum	Ni + Cr													392
103	Nickel + Copper	Ni + Cu													398
102	Nickel + Iron	Ni + Fe													403
103	Nickel + Magnesium	Ni + Mg													407
104	Nickel + Manganese	Ni + Mn													410
105	Nickel + Silicon	Ni + Si													413
106	Nickel + Tungsten	Ni + W													416
107	Nickel + Zinc	Ni + Zn													419
108	Niobium + Zirconium	Nb + Zr													422
109	Pallodium + Silver	Pd + Ag													425
110	Potassium + Sodjum	K + Na													428
111	Sodium + Polassium	Na + K													431
112	Tantalum + Tungsten	Ta + W													434
113	Thallium + Lead, PbTi2	Tl + Pb													437
114	Tin + Bismuth	Sn + Bi													440
115	Tín + Indium	Sn + In										•			443
116	Tin + Lead	Sn + Pb													440
117	Tin + Magnesium	Sn + Mg								•					449
118	Titanium + Manganese	Ti + Mn													453
119	Titanium + Molybdenum	Ti + Mo													4 56
120	Tungsten + Cobait	₩ + Co			•						•				459
121	Tungsten + 1ron	W + Fe								٠		٠			462
122	Vanadium + Aluminum	V + Al													465
123	Vanadium + Antimony	V + Sb													468
124	Vanadium + Iron	V + Fo													471
125	Vanadium + Tin	V + Sn													474
126	Vanadium + Titanium	V + Ti											٠		477
127	Zinc + Copper	Zn + Cu					-								460
128	Zinc + Magnesium	Zn + Mg													483

2. NONFERROUS BINARY ALLOYS (continued)

	Figure and/or Table No.	Name	Formula	Page No.
	129	Zine + Zirconium	Zn + Zr	. 486
	130	Zirconium + Indium	Zr + In	. 489
	131	Zirconium + Iron	Zr + Fe ,	. 492
	132	Zirconium + Niobium	Zr + Nb	. 495
	133	Zirconium + Silver	Zr + Ag	. 498
	134	Zirconium + Tin	Zr + Sn	. 501
	135	Zirconium + Titanium	Zr + Ti	. 504
	136	Zirconium + Uranium	Zr + U	. 507
3.	NONFERROUS	MULTIPLE ALLOYS		
	137	Aluminum + Copper + ∑X	Ai + Cu + ΣX ₁	. 511
	138	Aluminum + Zinc + ΣX_1	$Ai + Z_n + \Sigma x_i$. 514
	139	Chromium + Aluminum + ∑X ₁	$Cr + Ai + \sum X_i$	
	140	Chromium + Iron + ∑X ₄	$\operatorname{Cr} + \operatorname{Fe} + \Sigma X_1 \dots \dots \dots \dots$	
	141	Cobalt + Chromium + ΣX_1	$Co + Cr + \sum_{i=1}^{n} \dots \dots \dots \dots \dots$. 523
	142	Copper + Chromium + \(\Sigma\),	$Cu + Cr + \Sigma X_1$. 526
	143	Copper + Magnesium + Aluminum, MgCu _{2-X} Ai _X	Cu + Mg + Ai	. 529
	144	Copper + Magnesium + Silicon, MgCu _{2-X} Si _X	Cu + Mg + Si	. 532
	145	Magnesium + Aluminum + ∑X ₁	$Mg + Ai + \Sigma X_1 \dots \dots$. 535
	146	Magnesium + Thorium + ∑X ₄	$M_g + Th + \sum_i X_i$. 538
	147	Magnesium + Zinc + ΣX_i	$Mg + Zn + \Sigma X_1 \dots \dots \dots$. 541
	148	Molybdenum + Titanium + ΣX_1	Mo + Ti + ΣX_i	
	149	Neptunium + Calcium + ΣX_1	$Np + Ca + \sum X_1 \dots \dots \dots$. 547
	150	Nickel + Chromium + ΣX_1	$Ni + Cr + \Sigma X_i$. 550
	100	Meker Contonium Cas	(9≤Cr≤11)	. 550
	151	Nickei + Chromium + $\sum X_i$	$\begin{array}{c} Ni + Cr + \sum X_i \\ (15 \le Cr \le 16) \end{array}$. 553
	152	Nickei + Chromium + ΣX_i	$\begin{array}{lll} Ni + Cr + \sum X_i & \dots & \dots & \dots & \dots \\ (18 \le Cr \le 20) & & & & & & & & & & & & & & & & & & &$. 556
	153	Nickel + Chromium + ΣX_{i}	Ni + Cr + ΣX_i	. 559
	154	Nickei + Copper + ∑X _i	$Ni + Cu + \Sigma X_1$. 562
	155	Nickel + Iron + ΣX	Ni + Fe + ΣX_1	
	156	Nickol + Manganese + ∑X	$Ni + Mn + \Sigma X_1$	
	157	Nickei + Molybdenum + ΣX ₁	Ni + Mo + ΣX ₁	
	158	Niobium + Iron + ∑X _i	Nb + Fe + ΣX_1	
	159	Niobium + Molybdenum + ∑X,	Nb + Mo + ΣX_1	
	160	Niobium + Tanialum + ∑X ₁	Nb + Ta + ΣX_1	
	161	Niobium + Titanium + ∑X ₁	Nb + Ti + ΣX_i	
	162	Niobium + Tungsten + ∑X	Nb + W + ΣX ₁	
	163	Plutonium + Cerium + XX,	Pu + Ce + ΣX _i	
	164	Tantalum + Niobium + ΣX ₄	Ta + Nb + ΣX _j	
	165	Tantalum + Tungsten + XX	$Ta + W + \Sigma X_1 \dots \dots$	
	166	Titanium + Aluminum + ΣX_1	$Ti + Ai + \Sigma X_1$	
	167	Titanium + Chromium + ΣX_i	$Ti + Cr + \Sigma X_i$	

3. NONFERROUS MULTIPLE ALLOYS (continued)

	Figure and/or Table No.	Name	1	Form	ula														I	Page No
	168	Titanium + Iron + Cobalt	Ti	+ Fe	+ Co															604
	169	Titanium + Vanadium + ΣX ₄	Ti	+ V	+ ΣX ₁															607
	170	Zirconium + Iron + EX,			+ ΣX															610
	171	Zirconium + Hafnium + XX			+ ΣX															613
	172	Zirconium + Uranium + XX			+ ΣX ₁	_														616
		•			-															
١.	FERROUS AL	LOYS																		
	A. CARBON	STEELS GROUP I																		
	173	Iron + Carbon + ΣX _i		+ C C< 2,	+ ΣX ₁ 00)	•	•	•	•	•	•	•	٠	•	•	•	•	•	•	619
	B. CARBON	STEELS GROUP II																		
	174	Iron + Carbon + ∑X _i	Fe (C	+ C C<2.	+ ΣX ₁ 00)	•		•	•	•	•	•	•	•	•		•	•	•	623
	C. ALLOY S	STEELS GROUP I																		
	175	Iron + Aluminum		Fe +	Al ,															626
	176	Iron + Antimony		Fe +	Sb .															629
	177	Iron + Chromlum		Fe + 8≤Cı	Cr < 25)		•	1	•		•	•	•	•				•	•	632
	178	Iron + Chromium		Fe +	Cr < 50)		•	•								•		•	•	635
	179	Iron + Chromium + Manganese			r + Mr	١.														638
	180	Iron + Cobalt + ∑X _i			o + ΣX															641
	181	Iron + Copper + ΣX _i			u + ΣΧ															644
	182	Iron + Manganese + EX,			n + Σ															647
	183	Iron + Manganese + ΣX ₁	Fe	+ M	n + Σ⊅ (n≤50)															650
	184	Iron + Manganese + Carbon	,		n + C		_													655
	185	Iron + Nickel + ΣX_i			ı + Σx															660
	186	Iron + Nickei + Carbon		+ N1			•	•	Ī	Ċ	į		Ī				•		·	665
	187	Iron + Silicon + ΣX_i			. + ΣX	. •	•	•	•	•	٠	٠	٠	٠	•	•	•	•	Ċ	668
	198	Will be a second of the second	10	Fe +		١.	•	•	•	٠	•	•	•	•	•	•	•	•	Ċ	672
	189	Iron + Tin Iron + Titanium		Fe +		•	•				•						•		į	675
				•		·	·	·	·	·	•									
	D. ALLOY 8	TEELS GROUP II																		
	190	AlSI 420, Iron + Chromium + D	x _i	Fe 1	- Cr +	ΣX	ı ·						•						•	678
	191	AlSi 430, Iron + Chromium + Σ	$\mathbf{x_i}$	Fo ·	Cr +	ΣΧ	į ·			•				•		•	•	•	•	681
	192	AlSI 446, Iron + Chromium + Σ	x,	Fe t	cr+	ΣX	1 •			•	•	•	•					•	•	684
	193	lron + Chromlum + Manganese ΣΧ ₁	+	Fo 1	+ Cr + Cr < 5.	Mn 0)	+	ΣX	٠.	•			•		•				•	687

4. FERROUS ALLOYS (continued)

D. ALLOY STEELS GROUP II (continued)

Figure and/or Table No.	Name		Page No
194	Iron + Chromium + Manganese + ΣX_1	Fe + Cr + Mn + ΣX _i	. 690
195	AlSi 301, Iron + Chromium + Nickel + ΣΧ _i	Fe + Cr + Ni + ΣX _i	. 693
196	17-7 PH, Iron + Chromium + Nickel + ΣX _i	Fe + Cr + Ni + ΣX _i	. 696
197	Iron + Chromium + Nickel + ΣX_i	Fe + Cr + Ni + $\sum X_i$	
198	AISI 305, Iron + Chromium + Nickel + ΣΧ _i	Fe + Cr + Ni + ΣX _i	. 702
199	AISI 310, Iron + Chromium + Nickel + ΣΧ _i	Fe + Cr + Ni + ΣX _i	
200	AISI 316, Iron + Chromium + Nickel + ΣΧ _i	Fe + Cr + Ni + ΣX _i	
201	AISI 347, Iron + Chromium + Nickel + ΣΧ ₁	Fe + Cr + Ni + ΣX _i	. 711
202	Crucible HMN, Iron + Chromium + Nickei + EXi	Fe + Cr + Ni + ΣX _i	- 714
203	Iron + Chromium + Nickel + ΣX_i	Fe + Cr + Ni + ΣX_1	. 717
204	EI 257, Iron + Chromium + Nickel + ΣΧ _i	Fe + Cr + Ni + ΣX _i	. 720
205	Iron + Manganese + ΣX _i	Fe + Mn + ΣX ₁	- 723
206	Iron + Nickel + ΣX _i	Fe + Ni + ΣX_i	. 726
207	Iron + Nickei + Chromium + ΣX_i	Fe + Ni + Cr + ΣX_i	
208	Iron + Silicon + ΣX_i	Fe + Si + ΣX _i	
209	Iron + Titanium + ΣX_i	Fe + Ti + ΣX_i	- 735
210	Iron + Tungsten + ΣX_i	Fe + W + ΣX_i	. 738

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14.	TELLURIDES	717
15.	BROMIDES	759
16.	CHLORIDES	785
17,	FLUORIDES	915
18.	HYDRIDES	1033
19.	NITRIDES	1075
20.	CARBONATES	1109
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GROUPING OF MATERIALS AND LIST OF FIGURES AND TABLES

1. ELEMENTS

	Figure and/or Table No.	Name	Symbol																		Page No.
	1	Boron	В																		1
	2A	Carbon (Diamond)	С					•		•											4
	2B	Carbon (Graphite)	C																		9
	3	Iodine	I																		15
	4	Phosphorous	P																		18
	5	Sulfur	S		•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	21
2.	OXIDES		Formula																		
	6	Aluminum Oxide	Al_2O_3							٠.											25
	7	Diantimony Tetraoxide	Sb_2O_4																		30
	8	Diantimony Pentaoxide	$\mathrm{Sb}_{2}\mathrm{O}_{5}$		٠																33
	9	Arsenic Sesquioxide	As ₂ O ₃																		36
	10	Diarsenic Pentaoxide	As ₂ O ₅																		39
	11	Barium Oxide	BaO																		42
	12	Beryllium Oxide	BeO																		45
	13	Bismuth Sesquioxide	Bi_2O_3																		48
	14	Boron Sesquioxide	B_2O_3																		51
	15	Cadmium Oxide	CdO																		54
	16	Calcium Oxide	CaO																		57
	17	Cerium Dioxide	CeO ₂															*			60
	18	Cerium Sesquioxide	Ce ₂ O ₃																		64
	19	Chromium Sesquioxide	Cr_2O_3																		67
	20	Cobalt Monoxide	CoO																		70
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48	Molybdenum Dioxide	MoO ₂					٠								•			160
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74	Titanium Sesquioxide	Ti ₂ O ₃	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	250
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	164	Triiron Tetraseienide	Fe ₃ Se ₄																536
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	167	Nickel Selenide (Nonstoichiometric) Nl _x Se																545
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	173	Chromium Silicide	CrSi						•						•				665
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	175	Cobalt Silicide	CoSi								•	•	•		•				571
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	178	Pentairon Trisilicide	Fe ₂ Si ₃		580
	179	Triiron Silicide	Fe ₃ Si		563
	180	Trimanganese Silicide	Mn ₃ Si		586
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	182	Molybdenum Disilicide	MoSi ₂ ,		592
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	192	Triuranium Distlicide + Triuranium Monostlicide	U ₂ Si ₂ + U ₂ Si		622
	193	Trivanadium Silicide	V ₃ Si		625
	194	Vanadium Disilicide	VSi ₂		628
	195	Pentayanadium Trisilicide	V _p Si _g		631
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	198	Diarsenic Trisulfide	As ₂ S ₃		641
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	200	Dibismuth Trisulfide	Bi ₂ S ₂		647
	201	Cadmium Sulfide	CdS		650
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	209	Iron Sulfide	Fe5		674
	210	lron Disulfide	FeS ₂		677
	211	Lead Sulfide	PbS		681
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	214	Molybdenum Disulfide	MoS ₂		690
	216	Nickel Sulfide	Nis		693
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	220	Strontium Sulfide	SrS	7	708
	221	Strontium Disulfide	SrS_2	7	711
	222	Zinc Sulfide	ZnS	7	714
14.	TELLURIDES				
	223	Dibismuth Tritelluride	Bi ₂ Te ₃	7	717
	224	Cadmium Telluride	CdTe	7	720
	225	Digallium Tritelluride	Ga_2Te_3	, . 7	723
	226	Iron Telluride (Nonsteichiometric)	Fe _x Te		726
	227	lron Ditelluride	FeTe2	7	729
	228	Manganous Telluride	MnTc		732
	229	Nickel Telluride (Nonstoichiometric)	NiTe _x	7	735
	230	Nickel Ditelluride	NiTe ₂		738
	231	Palladium Telluride	PdTe	7	741
	232	Palladium Ditelluride	PdTe ₂	7	744
	233	Platinum Telluride	PtTe		747
	234	Platinum Ditelluride	PtTe ₂	7	750
	235	Disilver Telluride	Ag ₂ Te	7	753
	236	Silver Telluride (Nonstoichiometric)	Ag _X Te	7	756
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	237	Cadmium Dibromide	CdBr ₂	7	759
	238	Copper Bromide	CuBr		762
	239	Potassium Bromide	КВг	7	765
	240	Rubidium Bromide	RbBr	7	769
	241	Sodium Bromide	NaBr	7	772
	242	Strontium Bromide	SrBr		775
	243	Titanium Tribromide	TiBr ₁	7	778
	244	Titanium Tetrabromide	TiBr4		781
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	245	Barium Dichloride	BaCl ₂	7	785
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	247	Cadmium Dichloride	CdCl ₂ , , , , , , , , , , , , , , , , , , ,	7	791
	248	Calcium Dichloride	CaCl ₂		794
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	253	Cobalt Dichloride Hexahydrate	CoCl ₂ · 6H ₂ O													809
	254	Copper Dichloride	CuCl ₂			•	•									812
	255	Copper Dichloride Dihydrate	CuCl ₂ · 2H ₂ O													815
	256	Dysproslum Trichloride Hexahydrate	DyCl ₃ · 6H ₂ O													818
	257	Erblum Trichloride Hexabydrate	ErCl ₃ · 6H ₂ O													822
	258	Gadolinium Trichloride Hexahydrate	GdCl ₃ · 6H ₂ O													826
	259	Holmium Trichloride Hexshydrate	HoCl ₃ · 6H ₂ O													829
	260	Iron Dichloride	FeCl ₂													832
	261	Lithium Chloride	Lici													835
	262	Magnesium Dichloride	MgCl ₂													838
	263	Magnesium Dichloride Monohydrate	MgCl ₂ · H ₂ O											٠		841
	264	Magneslum Dichloride Dihydrate	MgCl ₂ · 2H ₂ O													844
	265	Magnesium Dichloride Tetrahydrate	MgCl ₂ · 4H ₂ O													847
	266	Magnesium Dichloride Hexabydrate	MgCl ₂ · 6H ₂ O													850
	267	Manganese Dichloride	MnCl.													853
	268	Manganous Dichloride Tetrahydrate	MnCl ₂ · 4H ₂ O	Ī												856
	269	Neodymium Trichloride Hexabydrate	NdCla · 6H2O										·			859
	270	Nickel Dichloride	NICl,											•	·	863
	271	Nickel Dichloride Hexabydrate	NICI2 · 6112O											•	•	866
	272	Phosphorus Trichloride	PCl ₂	•		•			•		•	•	•	•	•	869
	273	Potassium Chloride	KCI	•	•	•	•	•	•	٠	•	•	٠	•	•	872
	274	Rhenium Trichloride	ReCl	•	•	•	•	•	•	•	•	•	•	•	•	878
	275	Silicon Tetrachioride	SICI4	•	•	•	•	•	•	•	•	٠	•	•	٠	881
	276	Silver Chloride	AgCl	•	•	•	•	•	•	•	٠	•	•	•	•	HH4
	277	Sodium Chloride	NaCl	٠	•	•	•	•	•	•	•	•	•	٠	•	887
	278	Strontium Dichloride	SrCl ₂	•	•	•	•	•	•	•	•	•	•	•	٠	890
	279	Titanium Trichloride	TiCl ₁	•		•	•	•	•	•	•	•	•	٠	•	893
	290	Uranium Trichloride	UCI,			•	•	•	•	•	•	•	•	•	•	896
	281	Uranium Tetrachloride			•	•	•	•	•	•	•	•	•	•	•	899
	282	Vanadium Dichloride	VCl ₂	•	•	٠	٠	•	•	•	•	٠	•	•	•	902
	283	Vanadium Trichloride	VCl ₁	•	•	٠	•	•	•	•	•	•	•	•	•	905
	284	Zinc Dichloride	_	•	•	•	•	•	•	•	•	٠	•	•	•	908
	285	Zirconium Tetrachloride	ZnCl ₂ ZrCl ₄	•	•	•	•	•	•	•	•	٠	•	•	٠	911
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	286	Aluminum Trifluorlde	Alf ₃													915
	287	Barium Difluorlde	BaF ₂		•							•				518
	288	Beryllium Difluorlde	BeF ₂													921
	289	Calcium Difluoride	CaF ₂													924
	290	Cerlum Trifluoride	CeF,													527
	291	Cesium Monohydrogen Difluoride	CellF ₂		-											931
	292	Cobait Diffuoride	CoF ₂													534
	293	Hafnium Tetrafluoride	HfF4			-										537
	294	Iron Difluoride	FeF ₂													940
	295	Lithium Fluoride	LIF													943

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	298	Lithlum Monohydrogen Difluoride	LiHF ₂			•		•		٠,			•					953
	299	Magneslum Difluoride	Mg F ₂				•		•				•					956
	300	Manganese Difluorlde	MnF ₂								•							959
	301	Molybdenum Hexafluoride	MoF ₈															962
	302	Nickel Fluosilicate Hexahydrate, A	NiSiF $_8$.	6ŀ	120													966
	303	Nickel Fluositicate Hexahydrate, B	NiSiF€ •	6 H	ι _τ Ο				•									970
	304	Nickel Difluoride	NiF ₂															973
	30 5	Niobium Pentafluoride	NbF ₆															976
	306	Potassium Fluoride	KF															979
	307	Potassium Hydrogen Difluoride	KHF2															982
	308	Rubidium Fluoride	Rb F															985
	309	Rubidium Monohydrogen Difluoride	RbHF ₂															988
	310	Sllicon Tetrafluoride	Si F4															991
	311	Sodium Fluoride	NaF															994
	312	Trisodium Aluminum Hexafluoride	Na,AlF															997
	313	Sodium Monohydrogen Difluorlde	NaHF ₂															1000
	314	Strontlum Difluoride	SrF ₂															1003
	315	Thallium Monohydrogen Difluoride	TIHF2															1006
	316	Thorlum Tetrafluoride	Th F4													,		1009
	317	Titanium Tetrafluoride	TlF4															1012
	318	Uranium Tetrafluoride	UF4															1015
	319	Uranium Hexafluoride	UF.															1018
	320	Vanadium Trifluoride	VF ₈								,							1021
	321	Xenon Tetrafluoride	XeF4															1024
	322	Zinc Difluorlde	Zn F ₂															1027
	323	Zirconium Tetrafluoride	ZrF4	•		•	•	•	•	•	•			•	•	•	•	1030
18.	HYDRIDES																	
	324	Germanium Tetrahydride	Geli₄									,						1033
	3 2 5	Lithium llydride	Lih															1036
	326	Ditantalum Hydride	Ta ₂ H															1040
	327	Titanium Hydride (Nonstoichlometrie)	TIH _X															1044
	328	Titanium Dihydride	TlH ₂															1047
	329	Uranium Trihydride	UH ₃															1050
	330	Vanadium Hydride (Nonstoichiometric)	VH _x															1053
	331	Yttrium Dihydride	YH2											,				1056
	33 2	Yttrium Trihydride	YH3															1059
	333	Yttrium Dideutride	YD2															1062
	334	Yttrium Trideutride	YD3															1066
	335	Zirconium Hydrlde (Nonstoichiometric)	Zril _x				٠											1069
	336	Zirconium Dihydride	ZrH ₂		•													1072

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19. NITRIDES

	Figure and/or Table No.	Name	Formula Page No.
	337	Aluminum Nitride	Aln 1075
	338	Boron Nitride	BN
	339	Hafnium Nitride	HfN 1081
	340	Trimagnesium Dinitride	Mg ₂ N ₂ 1094
	341	Silicon Nitride	SiN
	342	Tantalum Nitride	TaN 1090
	343	Titanium Nitride	TiN 1093
	344	Uranium Nitride	UN 1096
	345	Uranium Nitride (Nonstoichiometric)	UN _X 1099
	346	Vanadium Nitride	VN 1103
	347	Zirconium Nitride	ZrN ,
20.	CARBONATES		
	348	Barium Carbonate	BaCO ₃
	349	Calcium Carbonate	CaCO ₁
	350	Calcium Magnesium Dicarbonate	CaMg(CO ₃) ₂
	351	Dilithium Carbonate	Li ₂ CO ₃
	352	Manganese Carbonate	MnCO ₃
	353	Dipotassium Carbonate	K ₂ CO ₃
	354	Distiver Carbonate	Ag ₂ CO ₃
	355	Disodium Carbonate	Na ₂ CO ₂
	356	Sodium Bicarbonate	NaliCO3
	357	Strontium Carbonate	SrCO ₃
21.	NITRATES and	NITRITES	
	358	Barium Dinitrate	Ba(NO ₃) ₂
	359	Gadolinium Trinitrate ilexahydrate	Gd(NO ₃) ₃ · 6H ₂ O
	360	Potassium Nitrate	KNO ₁
	361	Silver Nitrite	AgNO ₂
	362	Sodium Nitrate	NaNO ₃ 1151
	363	Strontium Nitrate	SrNO ₃
	364	Thailium Nitrate	TINO;
22.	SULFATES		
	365	Dialuminum Trisulfate	Al ₂ (SO ₂) ₃
	366	Diaiuminum Trisulfate liexahydrate	$Al_2(SO_4)_1 \cdot 6H_2O$
	367	Diammonium Sulfate	(Nil ₄) ₂ SO ₄
	:168	Ammonium Aluminum Disulfate	NII4AI(SO4)2
	369	Ammonium Aluminum Disulfate Dodecahydrate	NII4AI(SO4)2 · 12II2O · · · · · · · · · 1173

22. SULFATES (continued)

	Figure and/or Table No.	Name	Formula	Page No.
	370	Barium Sulfate	BaSO4	. 1176
	371	Beryllium Sulfate	BeSO ₄	. 1179
	372	Calcium Sulfate	CaSO ₄	. 1182
	37 3	Calcium Sulfate Hemihydrate	Caro' · Th'o · · · · · · ·	. 1185
	374	Calcium Sulfate Dihydrate	CaSO ₄ · 2H ₂ O	. 1188
	375	Cesium Aluminum Disulfate Dodecahydrate	CsAl(SO ₄) ₂ ·12H ₂ O	. 1191
	376	Cobalt Sulfate Heptahydrate	CoSO ₄ · 7H ₂ O	. 1194
	377	Dieuroplum Trisulfate Octahydrate	Eu ₂ (SO ₄) ₃ · 8H ₂ O	. 1197
	378	Iron Sulfate Heptahydrate	FeSO ₄ · 7H ₂ O	. 1200
	379	Dimercury Sulfate	Hg ₂ SO ₄	. 1203
	380	Nickel Sulfate Hexahydrate	NiSO4 · 6H ₂ O	. 1206
	381	Dipotassium Sulfate	K_2SO_4	. 1209
	382	Potassium Aluminum Disulfate	$KAi(SO_4)_2$. 1212
	383	Potassium Aluminum Disulfate Dodecahydrate	$KAl(SO_4)_2 \cdot 12H_2O$. 1215
	384	Disodium Sulfate	Na ₂ SO ₄	. 1218
	385	Disodium Sulfate Decahydrate	Na ₂ SO ₄ · 10H ₂ O	. 1221
	:186	Zinc Sulfate Heptahydrate	ZnSO4 · 7H2O	. 1224
23.	GLASSES and C	Aluminosilicate Glass	$SiO_2 + Ai_2O_3 + \Sigma X_i$. 1221
	387	Aluminosilicate Glass	• • •	
	388	Borosllicate Giass	$SiO_2 + B_2O_3 + \Sigma X_i$	
	389	High Silica Glass	$SiO_2 + \Sigma X_i$	
	390	Pyroceram		
	391	Soda Lime Glass	$SiO_2 + Na_2O + \Sigma X_i$,	
	392	Beryllium + Beryllium Oxide, Cermet	Be + BeO	
	393 394	Beryllium Oxide + Beryllium, Cermet Beryllium Oxide + Beryllium + Molybdenum,	BeO + Be	. 1246
	65	Cermet	BeO + Be + Mo	•
	395	Beryllium Oxide + Molybdenum, Cermet	BeO + Mo	. 1252
	396	Beryllium Oxide + Molybdenum Dodecaberyllide, Cermet	BeO + МоВе ₁₂	. 1255
	397 398	Beryllium Oxide + Niobium Dodecaberyilide, Cermet Bowyllium Oxide + Tantalum Dodecaberyilide	BeO + NbBe ₁₂	. 1258
	399	Beryllium Oxide + Tantalum Dodecaberyllide, Cermet Beryllium Oxide + Titanium Dodecaberyllide,	BcO + TaBe ₁₂	. 1261
	400	Cermet Beryllium Oxide + Zirconium 13-Beryllide,	BeO + TiBe ₁₂	. 1264
	401	Cermet Boron Nitride + Diboron Trioxide + ΣX_i	BeO + ZrBe ₁₃	. 1267
	401	Cermet	BN + B_2O_3 + ΣX_i	
	402	Boron Nitride + Carbon, Cermet	BN + C	
	403	Carbon + Silicon Carbide, Cermet	C + SiC	. 1276
	404	Silicon Carbide + Carbon + ΣX_i , Cermet	$\operatorname{SiC} + \operatorname{C} + \Sigma X_{i} \dots \dots \dots$	
	405	Tungsten Carbide + Cobalt, Cermet	WC + Co ,	. 1282
	406	Zirconium Dioxide + Titanium, Cermet	ZrO ₂ + Ti	. 1285

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24, OXYGEN COMPOUNDS

Figure and/or Table No.	Name	Formu l a							Page No.
407	Dialuminum Silicon Pentaoxide	Al ₂ SiO ₅		٠					1289
408	Hexaaluminum Disilicon 13-Oxide	Al ₆ Si ₂ O ₁₃							1292
409	Dialuminum Disilicon Heptaoxide Dihydrate	Al ₂ Sl ₂ O ₇ · 2H ₂ O .							1295
410	Dialuminum Titanium Pentaoxide	Al ₂ TiO ₅							1298
411	Barium Silicon Trioxide	BaSiO ₂							1301
412	Dibarium Silicon Tetraoxide	Ba ₂ SiO ₄							1304
413	Barium Disilicon Pentaoxide	BaSi ₂ O _i							1307
414	Dibarium Trisilicon Octaoxide	Ba ₂ Si ₂ O ₁							1310
415	Barium Titanium Trioxide	BaTiO ₃							1313
416	Dibarium Titanium Tetraoxide	Ba ₂ TiO ₄ ,							1316
417	Barium Uranium Tetraoxide	BaUO							1319
418	Barlum Zirconium Trioxide	BaZrO ₁							1322
419	Beryllium Dialuminum Tetraoxide	BeAl ₂ O ₄							1325
420	Diberyllium Silicon Tetraoxide	Be ₂ SiO ₄							1329
421	Calcium Dialuminum Tetraoxide	CaAl ₂ O ₄					_		1332
422	Culcium Tetraaluminum lleptaoxide	CaAl ₄ O ₁	_			-		_	1335
423	Tricalcium Dialuminum Hexaoxide	Ca ₂ Al ₂ O ₂							1338
424	Dodecacalcium 14-Aluminum 33-Oxide	Ca ₁₂ Al ₁₄ O ₂₂					•	•	1341
425	Calcium Diboron Tetraoxide	CaB ₂ O ₄					•	•	1344
426	Calcium Tetraboron lieptaoxide	CaB ₁ O ₇							1347
427	Dicalclum Diboron Pentaoxide	Ca ₂ B ₂ O ₅		•	•	•	•	•	1350
428	Triculcium Diboron Ilexaoxide	Ca ₂ B ₂ O ₆		•	•	•	•	•	1353
429	Calcium Diiron Tetraoxide	CaFe _z O ₄		•	•	•	•	٠	1956
430	Dicalcium Difron Pentaoxide	Ca ₂ Fc ₂ O ₃	•	•	•	•	•	•	1359
431	Calcium Molybdenum Tetraoxide	CaMoO ₄	•	•	•	•	•	•	1362
432	Calcium Silicon Trioxide	Ca SiO ₃					•	•	1365
433	Dicalclum Silicon Tetraoxide	Ca ₂ SiO ₄						•	1368
434	Tricalcium Silicon Pentaoxide	Ca ₃ SiO ₅							1371
435	Tricalcium Disilicon Heptaoxide	Ca ₂ Si ₂ O ₇							1374
436	Calcium Titanium Trioxide	CaTiO ₃		•	•	•	•	٠	1377
437	Tricalcium Dititanium Heptaoxide	Ca ₃ Ti ₂ O ₇			•	•	•	•	1380
438	Calcium Tungsten Tetraoxide	5 1110					•		1383
439	Calcium Uranium Tetraoxide	CaUO					•		1386
440	Calcium Divanadium Hexaoxide	CaV ₂ O ₆							1389
441	Dicalclum Divanadium Heptauxide	$Ca_2V_2O_1$							1392
442	Tricalcium Divanadium Octaoxide	Ca,V,O,							1395
443	Calcium Zirconium Trioxide	CaZrO ₃							1398
444	Dicalclum Dialuminum Silicon Heptaoxide	Ca ₂ Al ₂ SiO ₁							1401
445	Calcium Dialuminum Digilicon Octaoxide	CaAl ₂ Si ₂ O ₂							1404
446	Calcium Dialuminum Disilicon Octaoxide Dihydrate	CaAl ₂ Si ₂ O ₄ · 2H ₂ O							1407
447	Dicalcium Tetraaluminum Octasilicon 24-Oxido ileptahydrate	Cu ₂ Al ₄ Si ₄ O ₂₄ · 7H ₂ O							1410
448	Calcium Magnesium Distlicon Hexaoxide	CaMgSi2Og							1413
449	Dicalcium Magnesium Disilicon Heptaoxide	Ca2MgSi2O7							1416
450	Tricalcium Magnesium Disilicon Octaoxide	Ca ₂ MgSi ₂ O ₈							1419
451	Dicalcium Pentamagnesium Octasilicon 23-Oxide Monohydrate	Ca ₂ Mg ₅ Sl ₄ O ₂₃ - il ₂ O							1422

24. OXYGEN COMPOUNDS (continued)

Figure and/or Table No.	Name	Formula								Page No.
452	Cobalt Diiron Tetraoxide	CoFe ₂ O ₄ .								1425
453	Cobalt Iron Tetrapxide (Nonstoichiometric)	Co _x Fe _y O ₄ .								1428
454	Cobalt Tungsten Tetraoxide	CoWO4								1431
455	Copper Iron Tetraoxide (Nonstoichiometric)	Cu _x Fe _y O ₄ .								1434
456	Copper Diiron Tetraoxide	CuFe ₁ O ₄ .								1437
457	Trierbium Pentagaillum Dodecaoxide (Garnet)	Er ₃ Ga ₅ O ₁₂								1440
458	Iron Dialuminum Tetraoxide	FeAl ₂ O ₄ .								1443
459	Iron Dichromium Tetraoxide									1446
460	Iron Dicobalt Tetraoxide									1449
461	Diiron Silicon Tetraoxide									1452
462	Iron Titanium Trioxide									1455
463	Lead Molybdenum Tetraoxide									1458
464	Lead Tungsten Tetraoxide	14.1110								1461
465	Lithium Aluminum Dioxide								•	1464
466	Lithium Iron Dioxide									1467
467	Lithium Iron Tetraoxide (Nonstoichiometric)	Li _x Fe _y O ₄ .								1470
468	Dilithium Titanium Trioxide	•								1473
469	Lithium Zinc Iron Tetraoxide (Nonstoichiometric)									1476
470	Magnesium Dialuminum Tetraoxide	- .								1479
471	Magnesium Dichromium Tetraoxide	•								1482
472	and the same of th	=								1485
	Magnesium Diiron Tetraoxide	MgFe ₂ O ₄ .								
473		Mg _X Fe _y O ₄ .								1488
474	Magnesium Molybdenum Tetraoxide	MgMoO4 ,								1491
475	Magnesium Silicon Trioxide	•								1494
476	Dimagnesium Silicon Tetraoxide	Mg ₂ SiO ₄ .		•	•	•	•	٠	•	1497
477	Trimagnesium Tetrasilicon Undecaoxide Monohydrate	Mg ₃ Si ₄ O ₁₁ · H ₂ O) .							1500
478	Dimagnesium Tetraaluminum Pentasilicon 18-Oxide	Mg ₂ Al ₄ Si ₆ O ₁₈								1503
479	Magnesium Titanium Trioxide	- •		•	٠	•	•	•	•	1506
480	Magnesium Dititanium Pentauxide	MgTi ₂ O ₆ .		•	٠	•	٠	٠	٠	1509
481	Diniagnesium Titanium Tetraoxide	Mg_2TiO_4 .		•	•	٠	•		•	1512
482	Magnesium Tungsten Tetraoxide			•	•	•	•	•	•	1515
483	Magnesium Diyanadium Hexaoxide	$MgV_{2}O_{g}$.		٠	•	٠	٠	•	٠	1518
484	Dimagnesium Divanadium Heptaoxide	$Mg_2V_2O_7$.		•	•	•	•	•	•	1521
485	Manganese Silicon Trioxide	=								1524
486	Trineodymlum Pentagailium Dodecaoxido (Garnot)									1527
487		NiFe ₂ O ₄ .								1530
488	Nickel Iron Tctraoxide (Nonstoichiometric)	Nix Fuy O4 .								1533
489	Nickel Zinc Diiron Tetraoxide (Nonstoichiometric)									1536
490		KAl ₃ Si ₃ O ₁₁ .		•	•	•	•	•		1540
491	Potassium Trialuminum Trisilicon Undecaoxide Monohydrate	KAl ₃ Si ₃ O ₁₁ · Il ₂	ο.							1543
492	Silicon Dioxide + Dialuminum Trioxide + ΣX_i	SIO2 + Al2O3 +	ΣX_j							1546
493	Sodium Aluminum Dioxide	NaAlO ₂ .								1649
494	Sodium Boron DloxIde									1652
495	Disodium Tetraboron Heptaoxide	NiizB4O7 .								1666
		•								

24. OXYGEN COMPOUNDS (continued)

Figure and/or Table No.	Name	Formula										Page No.
496	Sodium Iron Dioxide	Na FeO2										1560
497	Disodium Molybdenum Tetraoxide	Na ₂ MoO ₄										1563
498	Disodium Dimolybdenum Heptaoxide	$Na_2Mo_2O_7$										1566
499	Disadium Silicon Trioxide	Na ₂ SiO ₃	٠									1569
500	Disodium Disilicon Pentaoxide	Na ₂ Si ₂ O ₅										1572
501	Disodium Teilurium Tetraoxide	Na ₂ TeO ₄			•							1575
502	Disodium Titanium Trioxide	Na ₂ TiO ₃										1578
503	Disodium Dititanium Pentaoxide	$Na_2Ti_2O_5$		•		•					•	1581
504	Disodium Trititanium Heptaoxide	$Na_2Ti_3O_7$						•	•	•		1584
505	Disodium Tungsten Tetraoxide	Na ₂ WO ₄			•						•	1587
506	Disodium Ditungsten Heptaoxlde	Na ₂ W ₂ O ₇		•		•		•	•	•	•	1590
507	Sodium Vanadium Trioxide	NaVO ₃				•	•		•	•		1593
508	Trisodium Vanadium Tetraoxide	Na ₃ VO ₄	•				•			•		1596
509	Tetrasodium Divanadium Heptaoxide	$Na_4V_2O_7$	•				٠		•	•		1599
510	Sodium Aluminum Trisilicon Octaoxide	NaAiSi ₃ O ₀	•			•		•	•			1602
511	Strontium Silicon Trioxide	SrSiO ₃		•								1605
512	Distrontium Sliicon Tetraoxide	Sr ₂ SiO ₄										1608
513	Strontium Titanium Trioxide	SrTiO3	•									1611
514	Distrontium Titanium Tetraoxide	Sr ₂ TiO ₄				•		•	•	•	•	1614
515	Strontium Zirconium Trioxide	SrZrO3	•	,			•			•	•	1617
516	Triytterbium Pentagallium Dodecaoxide(Garnet)	$Yb_{1}Ga_{1}O_{12}$								•		1620
517	Triyttrium Pentagailium Dodecaoxide(Garnet)	$Y_3Ga_6O_{12}$					•		•			1623
518	Zinc Diiron Tetraoxide	Zn Fe ₂ O ₄						•		•	•	1626
519	Dizinc Silicon Tetraoxide	Zn ₂ SiO ₄						•				1629
520	Dizine Titanium Tetraoxide	Zn ₂ TiO ₄		•		٠	•	•	•	•	•	1632
521	Zirconium Silicon Tetraoxide	ZrSiO4										16:15

VOLUME 6. SPECIFIC HEAT - NONMETALLIC LIQUIDS AND GASES

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GROUPING OF MATERIALS AND LIST OF FIGURES AND TABLES

1. ELEMENTS

	Figure and/or Table No.	Name	Symbol	Physicai State*									I	Page No.
	1	Argon	A	L, G										1
	2	Bromine	Br ₂	L, G								•	•	7
	3	Chlorine	Cl2	L, G										11
	4	Deuterium	D ₂	L, G										15
	5	Figorine	F ₂	L, G										19
	6	Helium	He	-, G										23
	7	Hydrogen	H ₂	L, G										26
	8	Krypton	Kr	L, G										34
	9	Neon	Ne	-, G										37
	10	Nitrogen	N ₂	L, G										39
	11	Oxygen	0,	L, G										48
	12	Xenon	Xe	L, G	 •	•	•		•	•	•	•		57
2.	INORGANIC CO	M POUNDS												
	13	Ammonia	NH ₃	L, G										61
	14	Boron Trifluoride	BF3	L, G										67
	15	Hydrogen Chloride	нсі	L, G										72
	16	Hydrogen lodide	HI	-, G										76
	17	Hydrogen Sulfide	H ₂ S	L, G										78
	18	Nitric Oxide	NO	L, G										83
	19	Nitrogen Peroxide	NO ₂	L, -						,				90
	20	Nitrous Oxide	O _z N	L, G										92
	21	Sulfur Dioxide	so ₂	L, G										97
	22	Water	H ₂ O	L, G	 •	•	•	•	•	•	•	•	•	102
3,	ORGANIC COMI	POUNDS												
	23	Acetone	(CH ₃) ₂ CO	L, G										113
	24	Acetylene	СИСН	-, G										117
	25	Benzene	C _e H _e	L, G									•	121
	26	i-Butane	i-C4H10	L, G										129
	27	n-Butane	n-C ₄ H ₁₀	L, G								•		136
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L = liquid, G = gas.

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Note: Figure number with "A" Indicates analyzed data graph.
*No figure

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This technical report covers work in a continuing systematic program on the thermophysical properties of materials involving the literature search, acquisition, codification, and organization and data extraction, compilation, evaluation, correlation, analysis, and synthesis, the preparation of "intermediate tables" presenting the total available experimental information, and the final preparation of internally consistent tables of "best data" referred to as "Tables of Recommended Reference Values." The work reported on consists of both data tables projects and scientific documentation efforts. The data tables projects are on the thermal conductivity, specific heat, thermal radiative properties (emittance, reflectance, absorptance, transmittance), thermal diffusivity, and thermal linear and volumetric expansion of elements, ferrous and nonferrous alloys, intermetallic, semiconducting, and nonmetallic compounds, cermets, ceramics, mixtures, composites, systems, polymers, etc., and on the thermal conductivity, specific heat, and viscosity of fluids and fluid mixtures. Property data are presented in both tabular and graphical forms, with accompany ing tables giving specifications and characterizations of the test specimens for the data. The resulting data tables are disseminated at large through the 13-volume TPRC DATA SERIES published commercially. This report does not contain the completed thousands of data sheets, but does reproduce in the Appendix, the Table of Contents and the Grouping of Materials and List of Figures and Tables for each of the first 7 volumes (which contain over 8000 pages) of the TPRC DATA SER IES to show the scope of their coverage. In scientific documentation, the scope is broader. TPRC covers all materials and maintains cognizance over 16 thermophysical properties (six more than mentioned above). There are now 55 700 references in TPRC's automated Information Storage and Retrieval System. The resulting information on research literature is disseminated through the THERMOPHYSICAL PROPERTIES RESEARCH LITERATURE RETRIEVAL GUIDE, published commercially.

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