UNCLASSIFIED

Defense Technical Information Center Compilation Part Notice

ADP012908

TITLE: The Preparation of Ordered Colloidal Magnetic Particles by Magnetophoretic Deposition

DISTRIBUTION: Approved for public release, distribution unlimited Availability: Hard copy only.

This paper is part of the following report:

TITLE: Nanostructures: Physics and Technology. 7th International Symposium. St. Petersburg, Russia, June 14-18, 1999 Proceedings

To order the complete compilation report, use: ADA407055

The component part is provided here to allow users access to individually authored sections of proceedings, annals, symposia, etc. However, the component should be considered within the context of the overall compilation report and not as a stand-alone technical report.

The following component part numbers comprise the compilation report: ADP012853 thru ADP013001

UNCLASSIFIED

The preparation of ordered colloidal magnetic particles by magnetophoretic deposition

Andrei Susha†‡, Dangsheng Su¶ and *Michael Giersig*‡ † Permanently address: Physico-Chemical Research Institute, Belarusian State University Leningradskaya 14, 220080 Minsk, Belarus ¶ Humboldt Universität zu Berlin, Institute of Physics, Invalidenstraße 110, 10115 Berlin, Germany ‡ Hahn-Meitner-Institut, Glienicker Straße 100, 14109 Berlin, Germany

Abstract. The preparation of ordered two-dimensional (2-D) magnetic nanoparticles using a magnetophoretic technique is reported. The quality of the ordering can be readily observed by electron microscope and the lattice constants determined by electron diffraction. Using image processing, it can be shown that the cobalt particles condense into hexagonal close packing and also that the crystallographic axes of the individual cobalt particles are randomly oriented. The equilibrium distance between the particles corresponds approximately to the size of the absorbed stabilisers and the strength of the magnetic field. The method is of general interest as a means of preparing monolayer films of nanosized magnetic particles such as cobalt or iron oxide.

1 Results and discussion

Cobalt nanoparticles were prepared by thermolysis of dicobaltoctacarbonyls in an organic carrier at 110 °C in presence of two different surfactants. The surfactants used were sodiumbis 2-(ethyl-hexyl) sulfosuccinate (Co I) and oleoylsarcosine (Co II). As a result ferrofluids (FF) of saturation magnetization of at least 20 mT were obtained. After the reaction non-stabilised particles were separated in an external magnetic field. After that, the stable ferrofluid was diluted with toluene in a ratio of 1:100. The composition of the particles (core and shell) was analysed by chemical and physical methods. The results corresponded to the magnetic properties of the core. Furthermore, magnetite particles were prepared by co-precipitation. The particles were modified by an inner surfactant layer of lauric acid and an outer layer of an ethoxylated alcohol. Aqueous base FF of a saturation magnetization by 100 mT were obtained. For observations all FF were highly diluted with the carrier.

Observation by high resolution transmission electron microscopy (HRTEM) (Fig. 1 and Fig. 2) showed Co particles well isolated and regularly dispersed in the surfactant with a narrow size distribution of about 8 nm \pm 6% (Co I) and 12 nm \pm 5% (Co II) in diameter. The sizes of the particles deduced from the analysis of the magnetic susceptibilities and magnetisation curves are consistent with those measured by HRTEM.

The distribution of the particle magnetic moments in ferrofluids reconstructed from the magnetization curves is in a good agreement with those computed from the particle sizes on HRTEM-images.

We have used the small nanoparticles for the creation of two dimensional arrays. The magnetic particles were ordered by drying a drop (5 μ L) on a carbon-coated grid in the presence of an external magnetic field of ca. 0.8 T, which was created by placing the grid between two magnets.

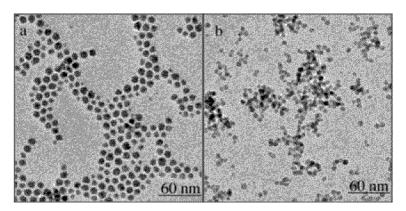


Fig. 1. Low magnification TEM micrograph of small Co particles in different solutions: (a) Co I; (b) Co II.

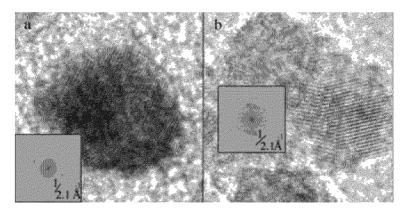


Fig. 2. HRTEM micrograph of two Co particles from the pictures Fig. 1 and Fig. 2 respectively.

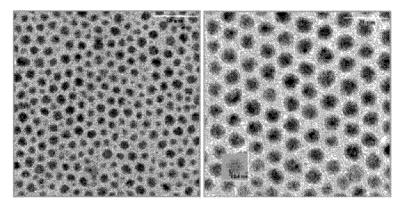


Fig. 3. TEM micrograph of 2-D ordering of Co nanoparticles by using an external magnetic field.

Figure 3 shows two TEM images of 2-D orderd Co particles at different magnetic field strengths.

References

- [1] G. Schmid Ed., Clusters and Colloids, from Theory to Application, V.C.H. Weinheim, 1994.
- [2] K. A. Eason, K. J. Klabunde, C. M. Sorensen and G. C. Hadjipanayis, *Polyhedron* 13, 1197 (1995).