

14 PAT-APPL -

(BEST COPIES AVAILABLE)

Serial No. 148,428

Filing Date 9 May 1980

Inventor(s): Thomas G. Roberts

Title: Nuclear Activated CW Chemical Laser.

Classification: Class Subclass

9 PAT. APPL. *Filed 9 May 80
NOTICE

The Government owned invention described herein is available for licensing. Inquiries and requests for license information should be addressed to:

Chief, Intellectual Property Division
Office of The Judge Advocate General
Department of the Army
Room 2D 444, The Pentagon
Washington, D. C. 20310

AD D607591

DDC FILE COPY

Accession For	
NTIS GRA&I	<input checked="" type="checkbox"/>
DTIC TAB	<input type="checkbox"/>
Unannounced	<input type="checkbox"/>
Justification	
By _____	
Distribution/	
Availability Codes	
Dist	Avail and/or
	Special

LEVEL II
DTIC
ELECTE
S OCT 20 1980 D
F

DISTRIBUTION STATEMENT A
Approved for public release;

109900

LB

Name of Inventor: Thomas G. Roberts

Title of Invention: Nuclear Activated CW Chemical Laser

Serial No.: 148,428

Filing Date: 5-9-80

AMC Docket No.: 3537

NUCLEAR ACTIVATED CW CHEMICAL LASER

DEDICATORY CLAUSE

The invention described herein may be manufactured, used, and licensed by or for the Government for governmental purposes without the payment
5 to me of any royalties thereon.

BACKGROUND OF THE INVENTION

Current high power cw chemical lasers which use hypergolic gases require special equipment and special techniques for handling these highly reactive, corrosive, and toxic substances. In applications where these systems
10 need to be portable, these substances are nearly prohibitive. For example, in the hydrogen fluoride (HF) chemical lasers gaseous F_2 is usually used. These hypergolic gas problems also plague the high power supersonic chemical transfer laser. However, these lasers have been operated where the electrons in an electric arc have been used to heat a gas such as nitrogen (N_2) and the hot
15 N_2 is used to produce fluorine (F) atoms by thermally dissociating a gas such as sulfur-hexafluoride (SF_6) prior to mixing with hydrogen (H_2) or deuterium (D_2) to produce lasing by the reaction $F + D_2 \rightarrow DF^* + D$ where the DF^* is produced in an excited vibrational state. While this approach solves the problem of using F_2 , it introduces the need for an electric arc and its associated generator
20 and control equipment which can be large and heavy in addition to not being very efficient in the application. Also, when F_2 is used it is first burned with a little H_2 in the plenum to heat the mixture to a temperature where some of the extra F_2 is thermally dissociated. This creates some HF which acts as a deactivator when the D_2 is mixed in to produce the lasing. When
25 the electric arc is used heating the mixture is no problem, but with the arc the amount of diluent like N_2 is not easily varied for optimum laser operation.

High energy particles like electrons from the electrode space of flash X-ray machines have been used to operate high power chemical lasers in the pulsed mode by radiating mixtures such as $F_2/H_2/O_2$, SF_6/H_2 , $F_2/D_2/O_2$, SF_6/D_2 , and other mixtures. Neutron-sources like pulsed reactors have also been used to cause these mixtures to lase. High power chemical lasers can also be operated by introducing the proper mixtures into the cavity of conventional electron beam driven CO_2 laser devices. However, in each of these cases very large and elaborate auxiliary power supplies are necessary. In some cases very high voltages are required.

5

SUMMARY OF THE INVENTION

The nuclear activated chemical laser is a chemical laser or a chemical transfer laser in which the radiation from radioactive waste is used to produce the active atom concentration necessary for operation of the laser. These atoms are produced from an inert, noncorrosive, nontoxic, nonhypergolic gas such as SF_6 passing through an active atom generator and no auxiliary power source is needed. Neither is an initiator or igniter needed; the laser action starts when the gas source valves are opened and stops when the source valves are closed. This is accomplished by using the radiation products such as α , β , and γ rays from processed radioactive waste to dissociate the SF_6 in the plenum of the laser. The radioactive material is held in the passageways of a housing device similar to a heat exchanger. This exchanger device is located in the gas generator section of a chemical laser such as a chemical transfer laser. The SF_6 gas is passed through the exchanger device thereby allowing the radiation from the radioactive material to dissociate the SF_6 , producing a concentration of free F atoms. This active atom generator may then feed the nozzle bank of a supersonic cw chemical laser or of a supersonic chemical transfer laser. It may also be used to feed the mixing section of either a subsonic chemical laser or a subsonic chemical transfer laser.

10
15
20
25

BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 is a diagrammatic illustration of a preferred embodiment of the active atom generator and laser device.

Figure 2 is a perspective view of the exchanger device of the active atom generator section.

Figure 3 is a cutaway view of an alternative exchanger device for the active atom generator section.

DESCRIPTION OF THE PREFERRED EMBODIMENT

Referring to the drawings. In Figure 1 the laser includes an active atom generating section 10 that provides a flow which contains a relatively large concentration of F atoms to a nozzle section 12 mounted downstream. The active atom generator 10 includes a dispersing section 13, a heat exchanger 14 or radiator having a set of passageways through which an inert gas is passed and an output chamber 15.

Exchanger or radiator 14 may be similar in function to an automobile or other type radiator for cooling an engine. A fluid or gas contained in the radiator housing may be stationary or moving and directs radiation through the housing walls where it affects the inert gas passing through openings in the housing. Similarly, for processed solid radioactive waste disposed on the surfaces of the passageways through the housing, inert gas passing through the radiator is affected by radiation from the waste. Typical embodiments of the exchanger 14 are shown in Figures 2 and 3.

As shown in Figure 2, a radioactive material may be in the form of a solid material 16 afixed to the walls 17 which form the passageways of the exchanger. The number, spacing, and shape of the passageways depend on the properties of the particular radioactive material used and on the pressure of the inert gas which is to be passed through the passageways. The pressure also depends on the type of laser used, i.e., supersonic or subsonic, and on the details of the nozzle bank 12. The radioactive material 16, which eliminates the need for a power source or igniter, may be processed solid radioactive waste and processed waste which may be obtained from the evaporation of liquids afixed to surfaces of the passageways.

As shown in Figure 3, in an alternative embodiment, exchanger or radiator 14 also include protrusions or fins 19 in the passageways for reducing flow and prolonging exposure to radiation. Instead of solid waste, a liquid or gaseous waste 20 is in chamber 22. The walls of chamber 22 also being the walls 17 of the passageways through the radiator. Liquid waste processed to enrich the β and γ radiations may be inclosed in chamber 22. Also it may be possible to dispense with the passages and simply mix a radioactive gaseous waste with the SF_6 in the gas generator. However, this can introduce too much contamination into the flow and reduce or eliminate the possibility of producing a population inversion by the desired chemical reaction.

The radioactive waste material is processed to eliminate the unwanted waste components and to concentrate the desired waste components which are then deposited on the surface walls or located as desired in gaseous or liquid form. As shown in Figure 3 and typical also for Figure 2 radioactive decay 23 is shown emanating from the walls for affecting the SF_6 gas.

While the inert gas such as SF_6 is passing through the passageways the radiation from the radioactive material thereon dissociates the gas SF_6 , thus producing the desired F atom concentration. As shown in Figure 2, the passageways are arranged so that the gas passes straight through. However if the dwell time of the SF_6 within the passageways must be increased to prolong the effect of radiation on the SF_6 , the passageways can be shaped to offer more resistance to the flow as shown typically in Figure 3 where obstructions 19 slow the flow. In this application the SF_6 pressure may be quite high and the passage walls may be fairly close together. The average spacing between the walls is adjusted to be approximately the range of the dominate radiation in the SF_6 . The amount of radioactive material used is proportional to the concentration of F atoms desired. Also in the event that it is necessary to raise the temperature of the gas by an additional amount a little H_2 or D_2 may be added to the flow just downstream of the passageways. For this case the passages and the radiation from the radiative material are simply used as an initiator and sustainer for a "conventional" chemical laser, except that now it is not necessary for the device to be supplied with a dangerous substance such as F_2 . Therefore it is not necessary to transport or handle the dangerous gas, a substantial improvement for chemical lasers of this type. However, where practical it is better not to introduce H_2 or

or D_2 upstream of the nozzle bank. In those lasers where this additive is not necessary the flow supplied to the nozzle bank contains no undesirable deactivators like HF or DF.

After the flow leaves the active atom generator, downstream processes
5 are substantially the same as for prior art chemical transfer lasers. Typical of the structure and operation of such lasers is the "Supersonic Chemical Transfer Laser" of patent number 3,832,650 issued August 27, 1974 to Thomas G. Roberts. In the Background, patent number 3,832,650 discusses existing subsonic and supersonic flow in chemical transfer lasers. Also discussed is
10 the object of the invention, providing a laser in which the molecule to which energy is to be transferred is caused to be at a proper place during the proper time.

In the preferred embodiment of Figure 1, active atom generator 10 has an input dispersion member or nozzle 11 coupled to receive and disperse
15 the SF_6 gas received from a suitable source 24 via input conduit 26. The inert gas SF_6 is dispersed and passed through radiator 14 providing a mixture output affected by radiation from the passageway walls of radiator 14. The output from the radiator may comprise a mixture of F, F_2 , SF_4 , SF and other gases. A diluent source 27 of N_2 or He may be coupled to the input side of radiator
20 14 (not shown) or to the output side of radiator 14 via conduit 28. The combination of N_2 or He with the SF_6 or output gases causes a decrease in the gas temperature and may be introduced if the gas temperature is above that required for mixture in nozzle 12. Similarly, H_2 from source 29 may be introduced into the generator to cause part of the fluorine to be burned and thereby release energy as heat to increase
25 the gas temperature if needed. The gas mixture in the plenum chamber output 15 of generator 10 is thereby brought to a thermal equilibrium condition wherein a large concentration of fluorine atoms exist prior to expansion of the gas through nozzle section 12. Obviously, with adequate temperature control of the SF_6 and the radiator 14, there is no requirement for H_2 , N_2 , or He. A carbon dioxide
30 (CO_2) source 30 is shown coupled through conduit 32 to nozzle 12 and a deuterium (D_2) source 34 is coupled through conduit 36 to nozzle 12. The fluorine from generator

10 is thoroughly mixed with the CO_2 in the nozzle before the D_2 is injected. When the D_2 is injected into and diffuses into the mixture, lasing of the CO_2 occurs in lasing chamber or optical cavity 40. The spent gases are then coupled to an exhaust system 42 for removal. For exo-atmospheric operation, exhaust
5 system 42 is not required and may be omitted.

The particular operation and detail of nozzle 12 is well disclosed in the prior art teachings as evidenced in the nozzle section teachings of Patent Number 3,832,650, as well as teachings respecting various other chemical reaction transfers and the introduction of the respective chemicals to provide
10 lasing action.

While a particular embodiment and form of this invention has been illustrated, it will be obvious to those skilled in the art that modifications may be made without departing from the scope and spirit of the foregoing disclosure. For example, cooling fins may be incorporated in radiator 14, water may be circulated within chambers of
15 the radiator, or the SF_6 may itself be circulated in chambers 22. Similarly, for exo-atmospheric application appropriate pumps, valving, and standard fluid handling techniques are employed with all cryogenic fluid. Therefore, it is understood that the invention is limited only by the claims appended hereto.

claims not included

ABSTRACT OF THE DISCLOSURE

A cw chemical laser which uses processed radioactive waste to produce active atoms from a chemically inert gas before being mixed with another molecule such as hydrogen or deuterium is disclosed. This laser uses no toxic or corrosive fuels and does not require any electrical or other type of auxiliary power supply. The energy released by the radioactive material is used to produce the active atoms such as fluorine. This is accomplished by using the radiation products from processed radioactive waste to dissociate the inert gas in the plenum of the laser. The radioactive material is held in the passageway walls of a device similar to a heat exchanger. The exchanger device may be located in the gas generator section of a chemical laser. The inert gas is passed through the exchanger device and while passing through it the radiation from the radioactive material dissociates the gas, producing a concentration of free active atoms. This active atom generator then feeds the nozzle bank or mixing section of a laser to produce a lasing action.

7

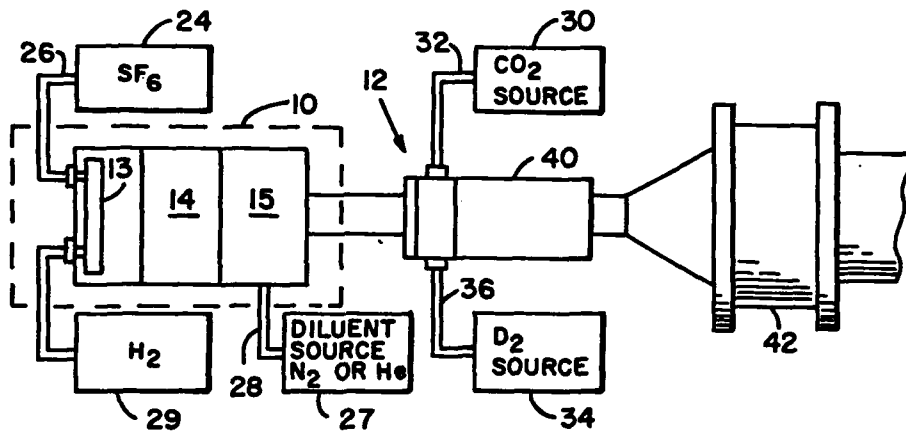


FIG. 1

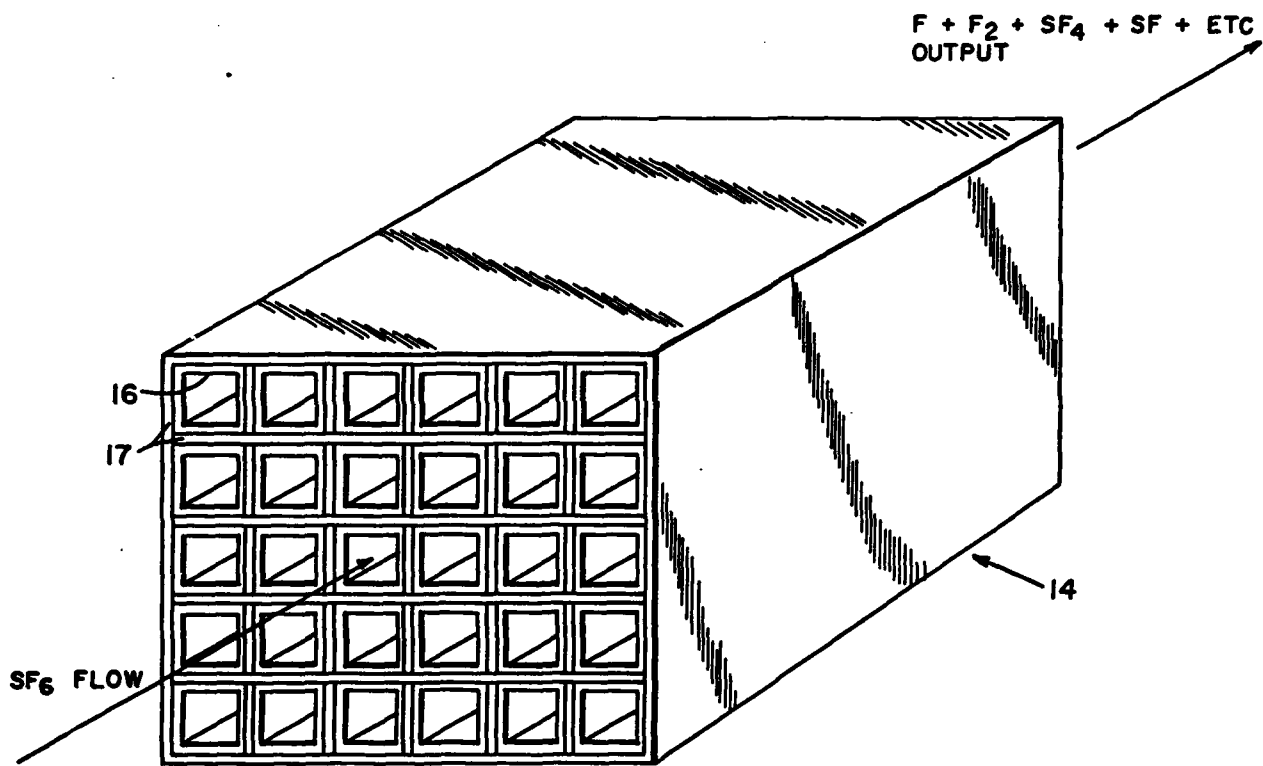


FIG. 2

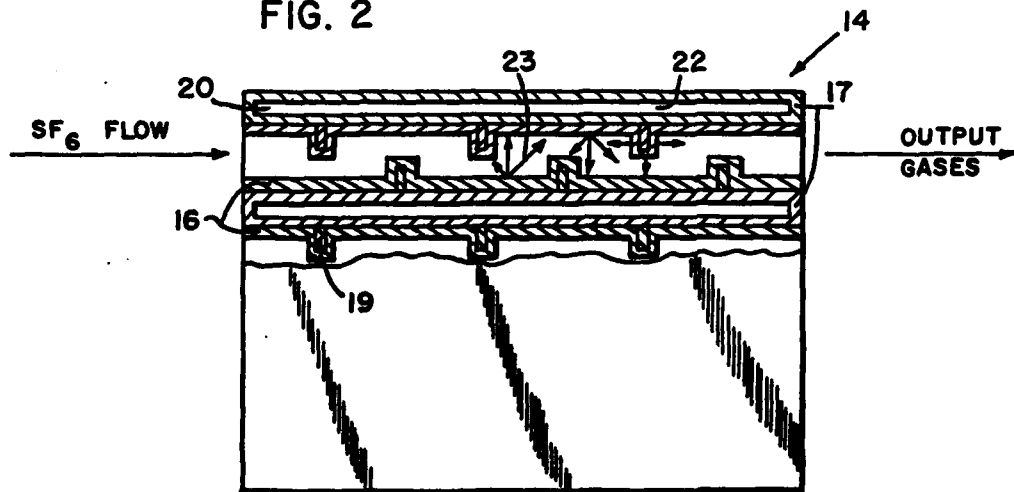


FIG. 3