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## Interactions Between Structure and Processing that Control Moisture Uptake in High-Performance Polycyanurates

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24 March 2015

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# Outline: Basic Studies of Moisture Uptake in Cyanate Ester Networks



- Background / Motivation
- SOTA Theories of Moisture Uptake in Thermosetting Networks
- New Tools and New Discoveries
- Unresolved Issues and Ways to Address Them

**Acknowledgement: Air Force Office of Scientific Research; AMG Group Members**





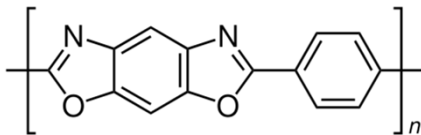
# Importance of Moisture Uptake in Composite Component Performance



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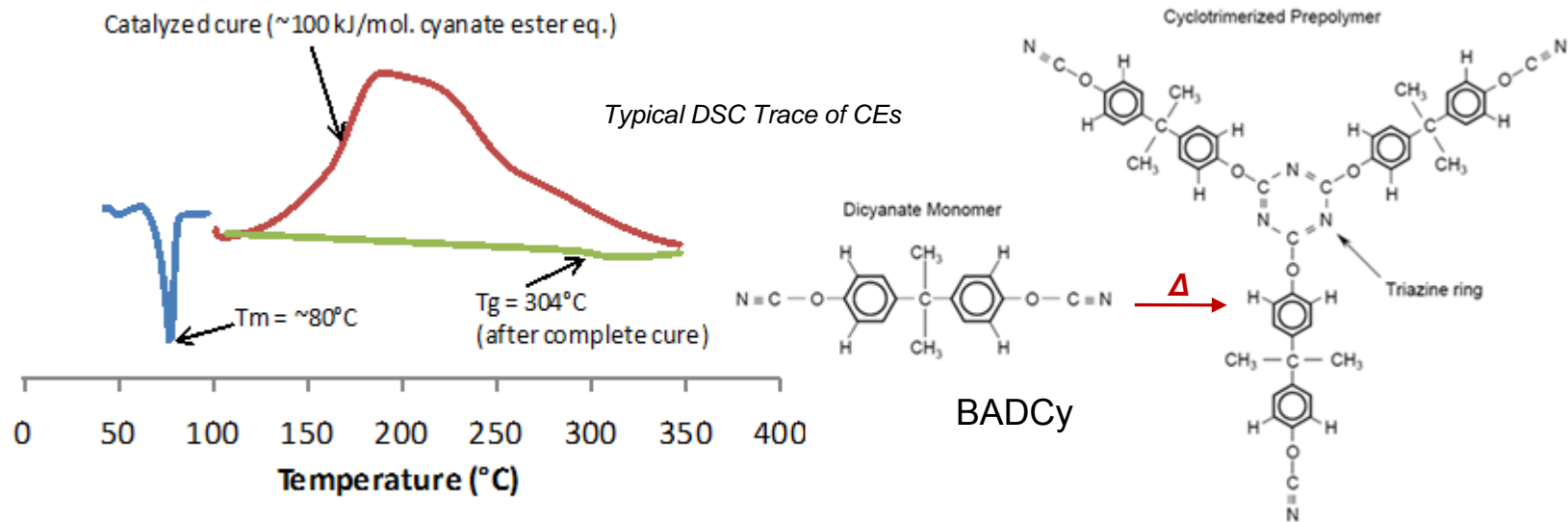


U.S. Navy photo by Mass Communication  
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- Water can add significantly to launch or take-off weight (3% water in composite resins = about 50 lbs of extra weight on an large SRM)
- Items with high water content can fail catastrophically when suddenly heated
- Long-term exposure to water can facilitate many mechanisms of chemical degradation, necessitating substantial “knock down” factors in design allowables
- Though more stable than epoxy resins, cyanate esters can degrade on long-term exposure to hot water



# Model High-Temperature Thermosetting Polymers: Cyanate Esters



- Glass transition temperatures at full cure of  $200 - 400^\circ\text{C}$
- Uncured resins exist as low-melting solids, or low to moderate viscosity liquids, making them ideal for processes such as filament winding
- Broad compatibility with co-monomers, thermoplastic tougheners, or nanoparticles for control of physical and mechanical characteristics
- Single species reaction chemistry is “cleaner” than epoxy resin and well-understood; enables development of superior predictive models for failure; readily catalyzed to cure at reasonable temperatures





# Cyanate Esters: Next-Generation High-Performance Composite Resin

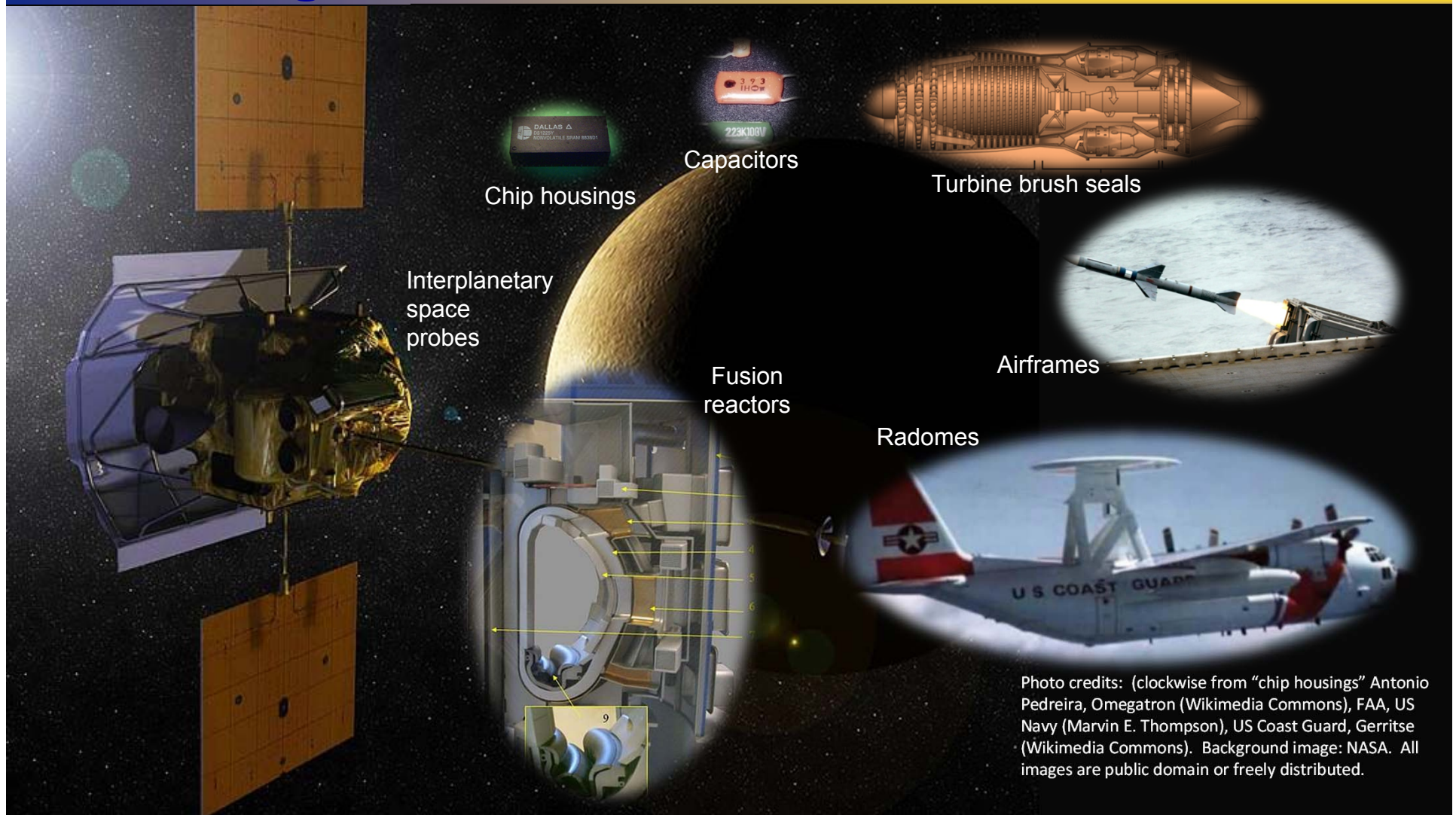


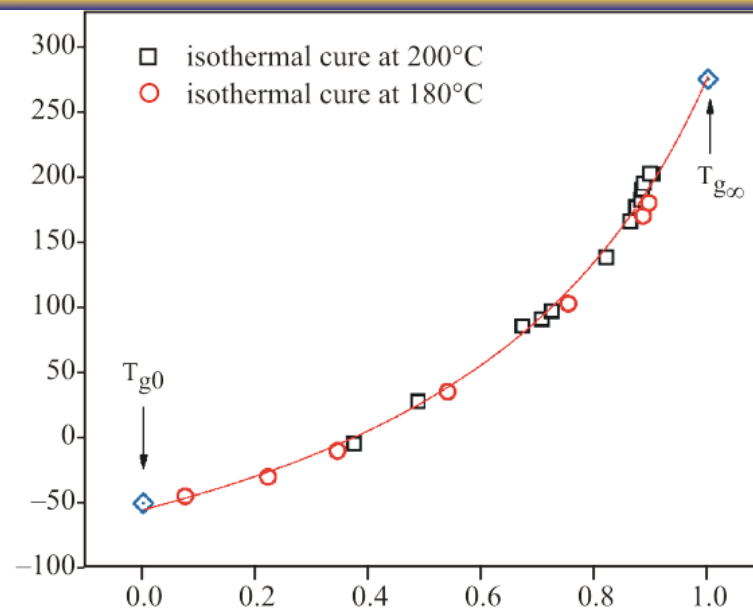
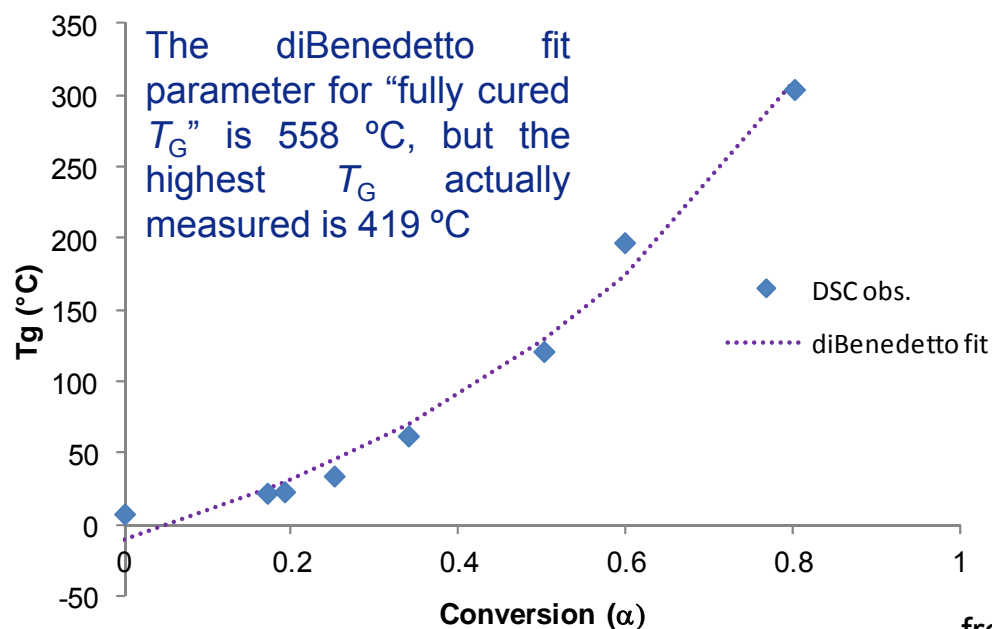
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- Many opportunities for technical transition beyond SRM cases ...

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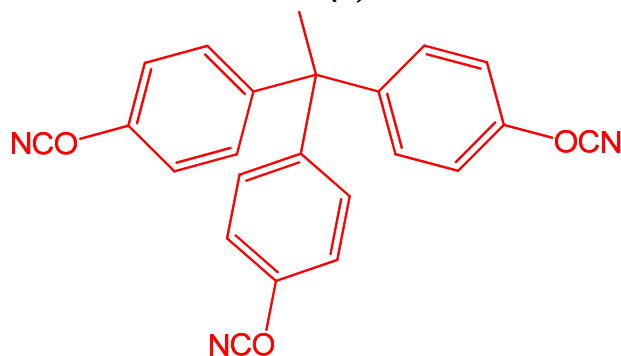


# Cyanate Ester Networks: Defined by Composition and Conversion

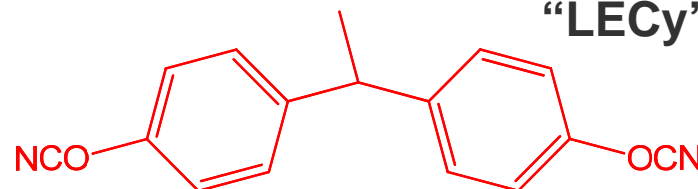


from X. Sheng, M. Akinc, and M. R. Kessler, *J. Therm. Anal. Calorim.* **2008**, 93, 77-85 for EX-1510

“ESR255”



“LECy”



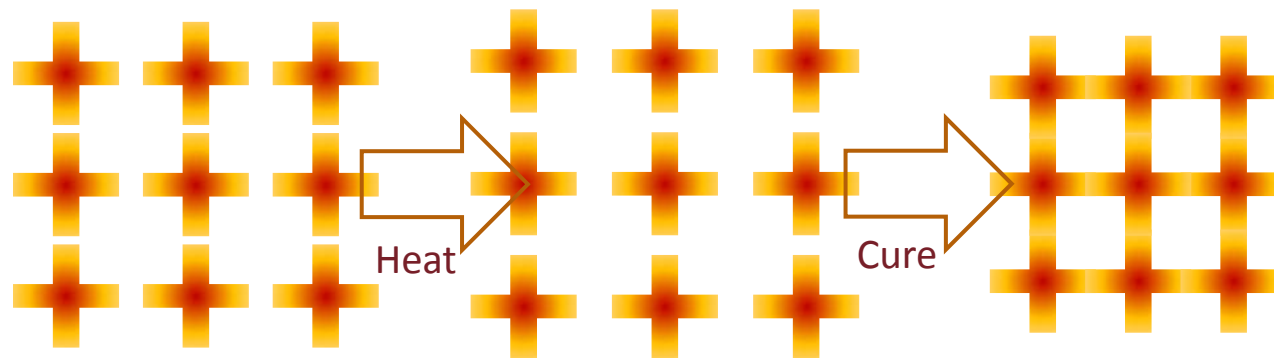
- In addition to glass transition temperature, many properties, such as density and water uptake, are mainly functions of conversion and monomer type



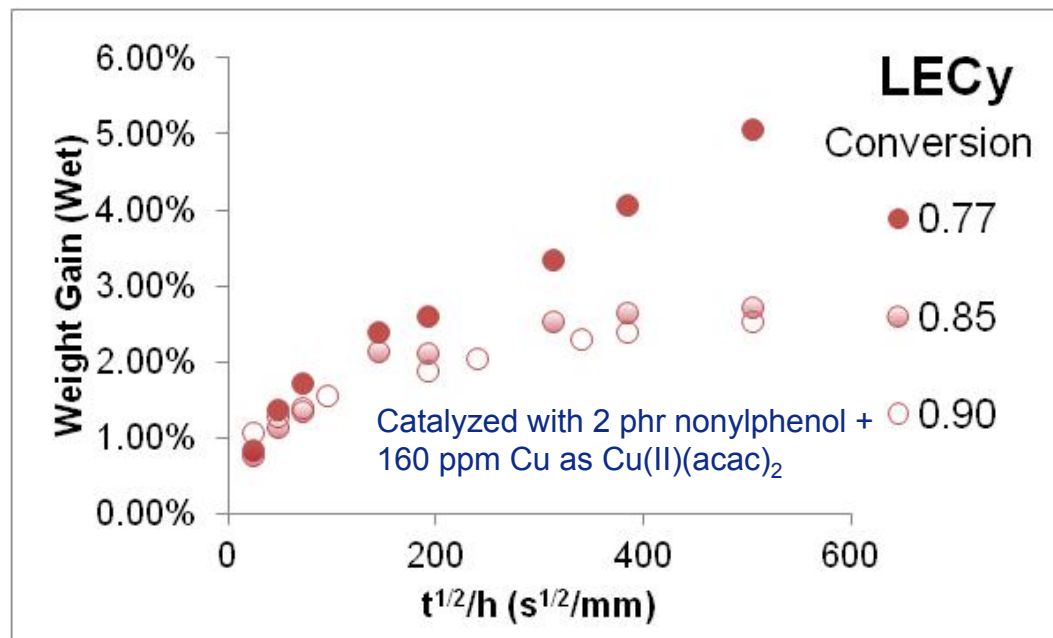
# Basic Understanding of Moisture Uptake in Highly Cross-linked Networks



## Main Stage Thermal Cure



- Cure results in:
  - *Net Shrinkage*
  - *Less permeability*
  - *Higher modulus*
  - *Brittleness*



- Increasing conversion joins together more "loose ends" in the network, eliminating free space where water can be absorbed, therefore water uptake is expected to decrease with increasing conversion

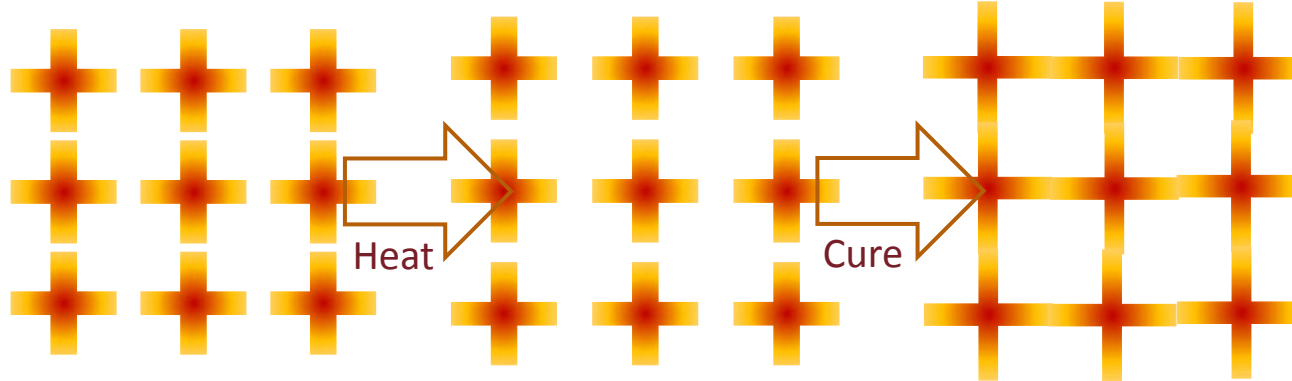




# Proposed Role of Vitrification in Controlling Moisture Uptake

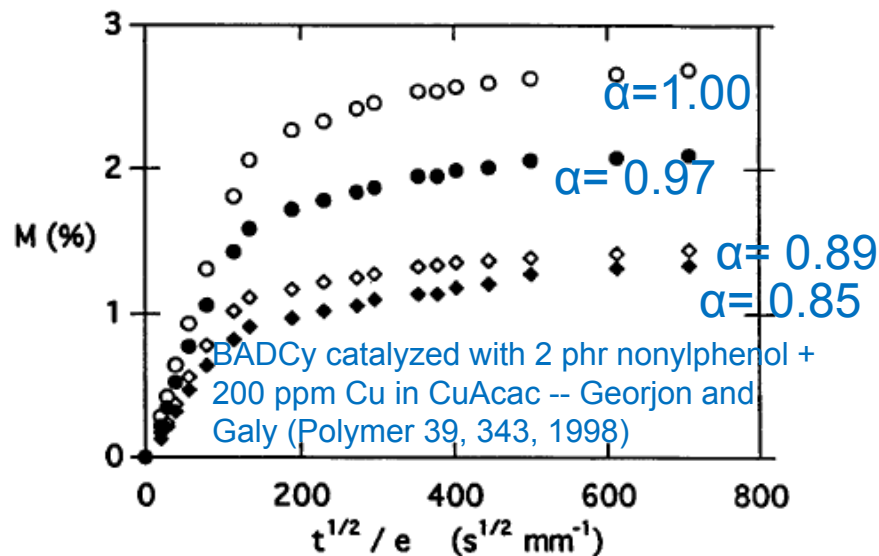


## "Vitreous Cure"



- Cure results in:

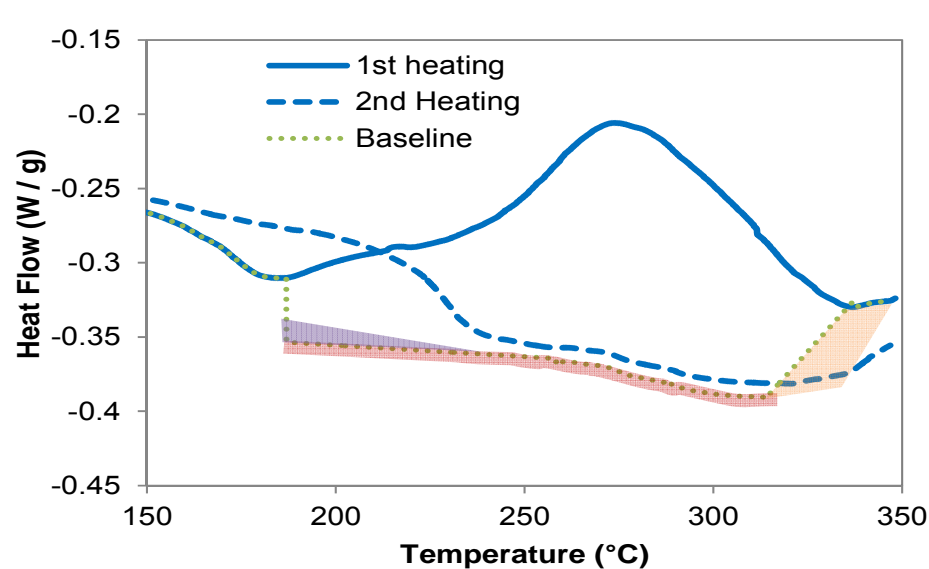
- *Net Expansion*
- *Higher permeability*
- *Lower modulus*
- *Toughness*



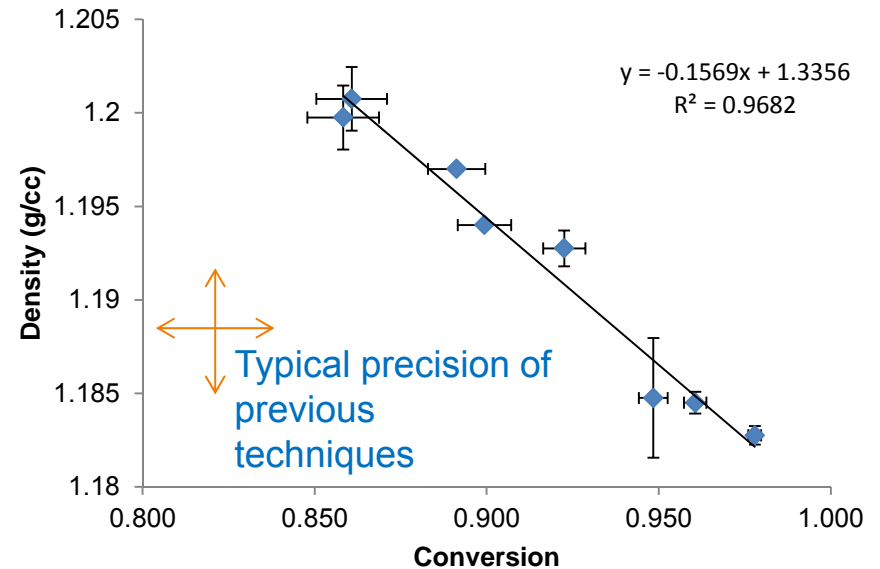
- Increasing conversions join "loose ends", but in the process they "freeze" the individual molecular centers-of-mass in place, resulting in exactly the opposite of the effects predicted by the traditional picture.



# New Tools for Understanding Network Structure



Typical computed DSC baseline with estimated zones of uncertainty



BADCy catalyzed with 2 phr nonylphenol + 160 ppm Cu in Cu(II)Acac:  
 $dV_m/dn_{tr} = 46 \pm 4$  cc/mol vs. previously determined value (2012) of  $37 \pm 15$  cc/mol

- New DSC method allows for more precise and objective assessment of conversion, combined with  $T_G$  assessment for determination of conversion to within 0.01.
- New density technique adapted for use by Mr. Michael Ford allows for 100x faster assessment with high precision.

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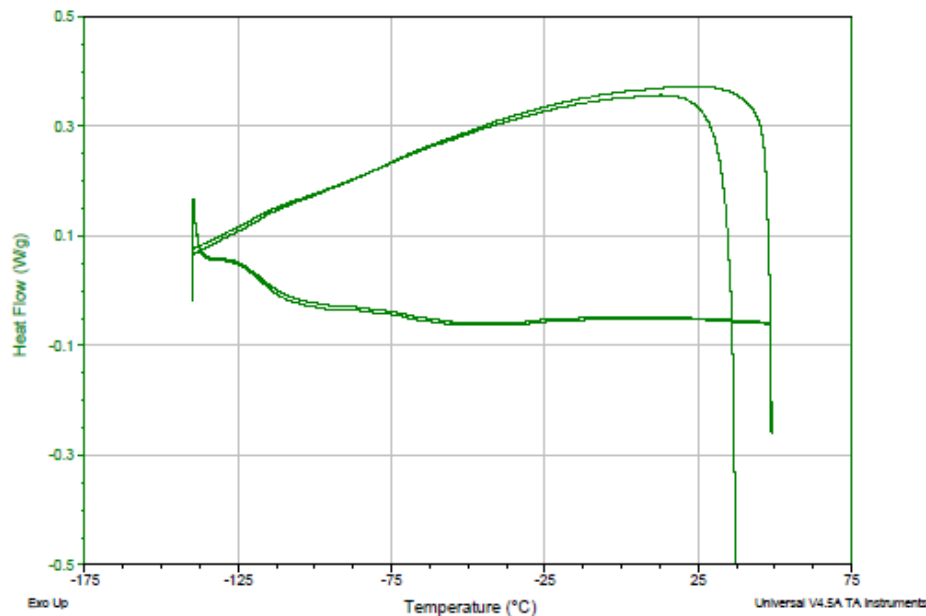
# DSC Shows No Freezing of Water in Polycyanurate Networks



Sample: LECY 96hr waterbott cryo rerun  
Size: 5.3000 mg

DSC

File: C:\...Iwet LECY Cryo experiment-rerun.001  
Operator: Kevin  
Run Date: 17-Dec-2013 14:25  
Instrument: DSC Q200 V24.10 Build 122



- In accordance with properties of water in cyanurate networks (i.e. higher miscibility at lower temperatures, no clustering)
- Recent NASA study of cryogenic toughness of cyanate ester / carbon fiber composites showed that residual stress minimization (by minimization of cure temperature) was the most important factor in achieving good toughness at cryogenic temperatures



# Moisture Uptake Is Likely a Minor Influence on Cryogenic Toughness



| Material  | B - Thickness |       | W - Height |       | a - Notch<br>+ Crack Length |       | P <sub>Q</sub> - Applied Load |         | K <sub>q</sub> (K <sub>1c</sub> ) |                         |
|---|---------------|-------|------------|-------|-----------------------------|-------|-------------------------------|---------|-----------------------------------|-------------------------|
| Specimen ID   | (in)          | (mm)  | (in)       | (mm)  | (in)                        | (mm)  | (lbf)                         | (N)     | (psi-in <sup>1/2</sup> )          | (MPa-m <sup>1/2</sup> ) |
| Catalyzed LECy Panel Cured 1-6-14 Water Soaked-Tested 4-16-14 |               |       |            |       |                             |       |                               |         |                                   |                         |
| STL300-1  | 0.1429        | 3.630 | 0.3920     | 9.957 | 0.1895                      | 4.813 | 24.24                         | 107.8   | 2,739                             | 3.01                    |
| STL300-2  | 0.1456        | 3.698 | 0.4270     | 10.85 | 0.1985                      | 5.042 | 27.70                         | 123.2   | 2,782                             | 3.06                    |
| STL300-3  | 0.1452        | 3.688 | 0.4280     | 10.87 | 0.2150                      | 5.461 | 19.89                         | 88.47   | 2,247                             | 2.47                    |
| STL300-4  | 0.1508        | 3.830 | 0.4640     | 11.79 | 0.2195                      | 5.575 | 27.76                         | 123.5   | 2,647                             | 2.91                    |
| STL300-5  | 0.1423        | 3.614 | 0.4310     | 10.95 | 0.2115                      | 5.372 | 27.67                         | 123.1   | 3,064                             | 3.37                    |
|   |               |       |            |       |                             |       |                               | Average | 2,696                             | 2.96                    |
| Catalyzed LECy (1-10-14) Tested 4-17-14                       |               |       |            |       |                             |       |                               |         |                                   |                         |
| STL300-6  | 0.1413        | 3.589 | 0.3575     | 9.081 | 0.1815                      | 4.610 | 22.30                         | 99.19   | 2,881                             | 3.17                    |
| STL300-7  | 0.1472        | 3.739 | 0.3335     | 8.471 | 0.1810                      | 4.597 | 19.89                         | 88.47   | 2,868                             | 3.15                    |
| STL300-8  | 0.1424        | 3.617 | 0.3545     | 9.004 | 0.1625                      | 4.128 | 21.75                         | 96.74   | 2,404                             | 2.64                    |
| STL300-9  | 0.1408        | 3.576 | 0.3575     | 9.081 | 0.1710                      | 4.343 | 27.20                         | 120.99  | 3,216                             | 3.53                    |
| STL300-10   | 0.1425        | 3.620 | 0.3520     | 8.941 | 0.1605                      | 4.077 | 19.82                         | 88.16   | 2,181                             | 2.40                    |
|   |               |       |            |       |                             |       |                               | Average | 2,710                             | 2.98                    |

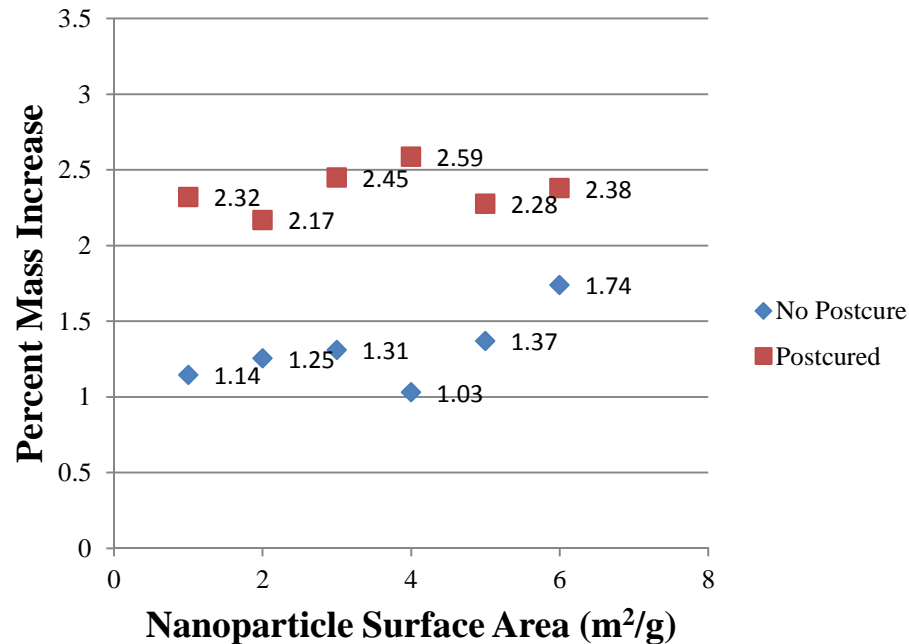
- LECy samples were soaked for 28 days at room temperature so as to maximize water uptake (1.5 wt%) while minimizing carbamate formation ( $\Delta V/V \sim 0.2\%$ ).
- No significant difference between exposed and control samples in fracture toughness at RT; same glassy state dynamics should apply at cryogenic temperatures
- The water in cyanate ester networks appears to be neither clustered nor strongly bound



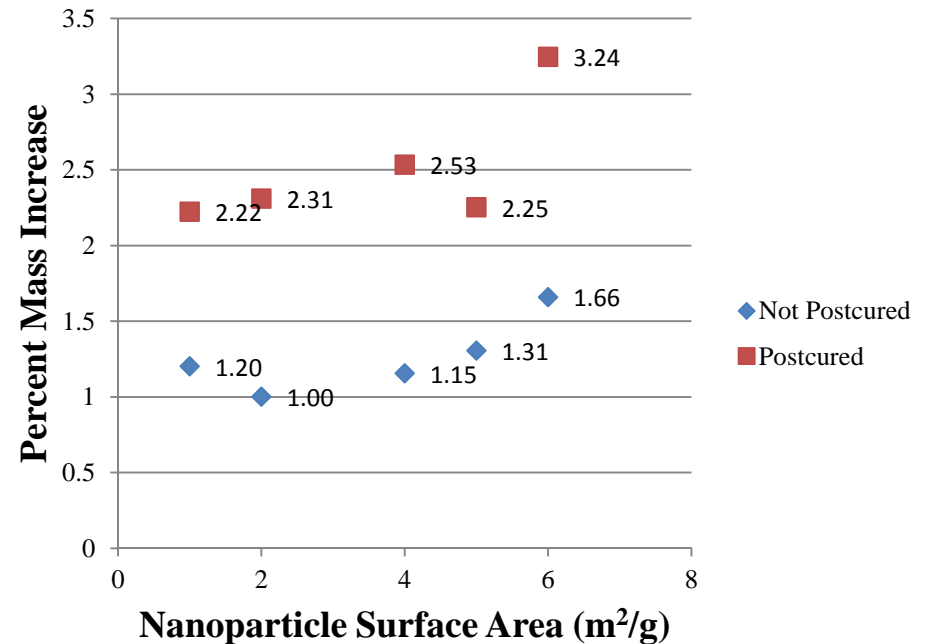
# Nanoscale Reinforcement and Interphases



**-OH Modified Surface**



**Octyl Modified Surface**

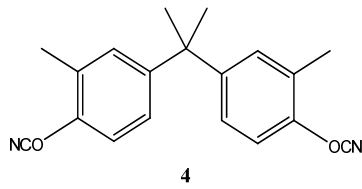
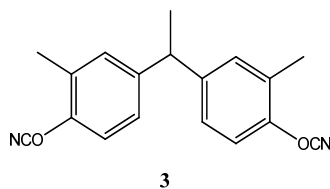
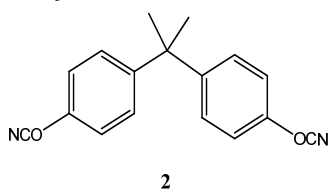
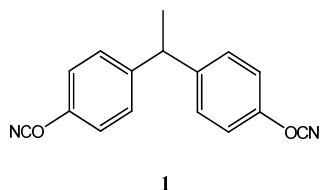
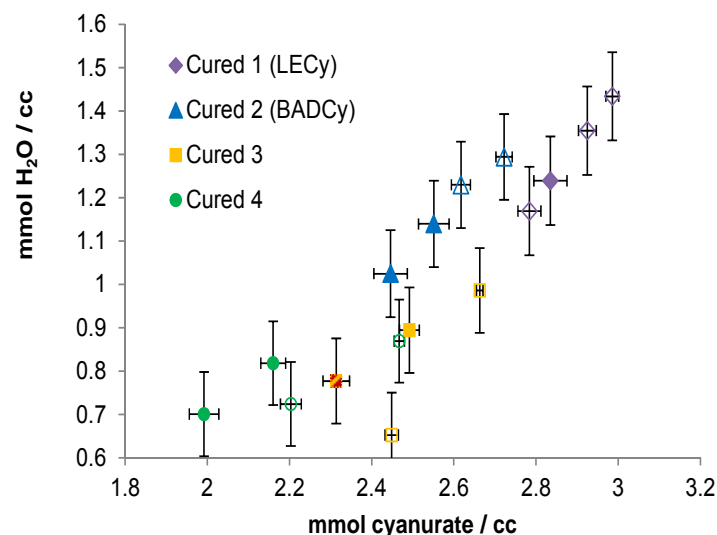
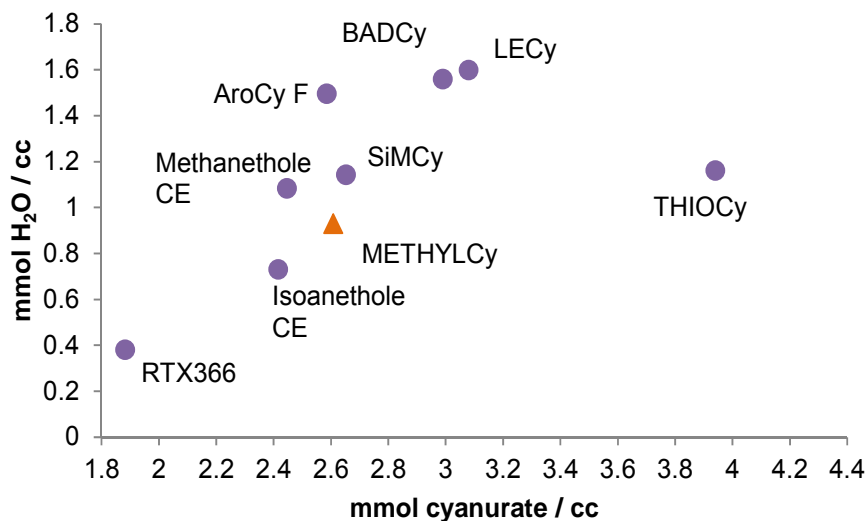


- A slight amount of water does accumulate at hydrophilic interphases in cyanate esters, but other effects such as poor bonding and damage tend to overwhelm these effects
- Nanoscale reinforcements such as graphene oxide with extreme water transport characteristics can alter the water uptake significantly.
- Generally speaking, blends and co-networks tend to follow linear rules of mixtures





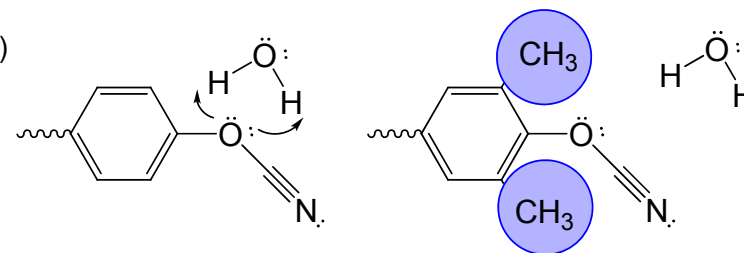
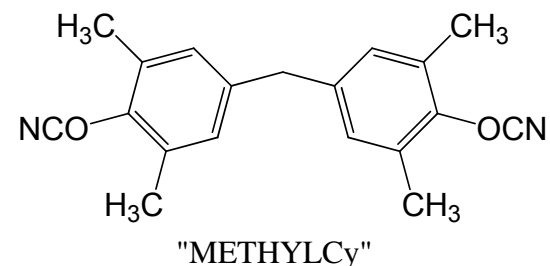
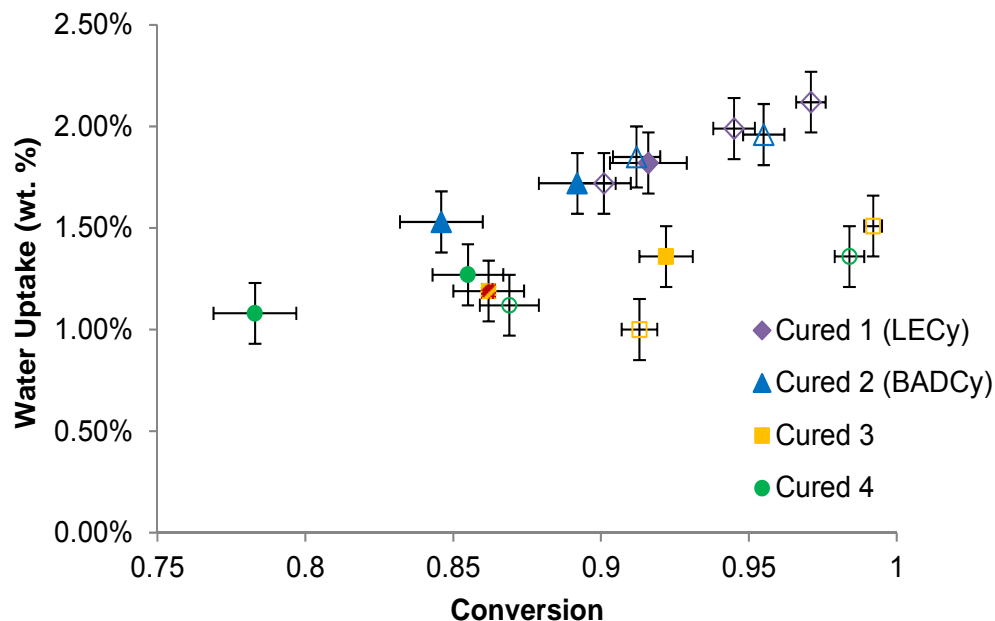
# Effect of Cyanurate Density on Moisture Uptake



- Each monomer has a different water uptake as a function of cyanurate density, although all show an increasing trend
- The free volume / cyanurate density relationships are similar for all monomers



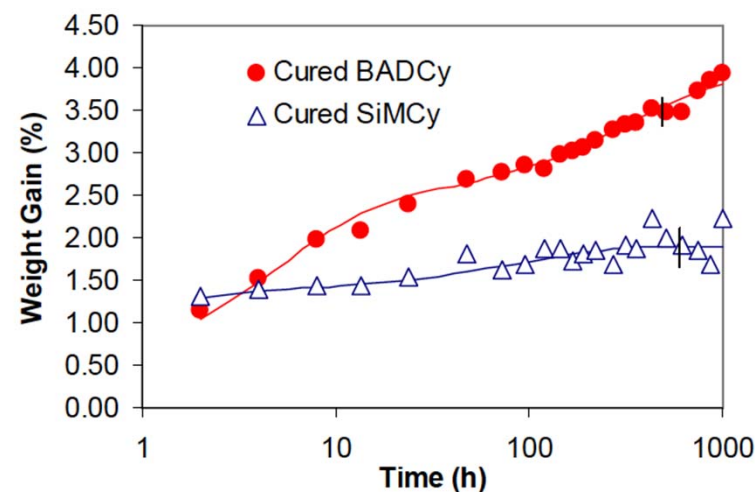
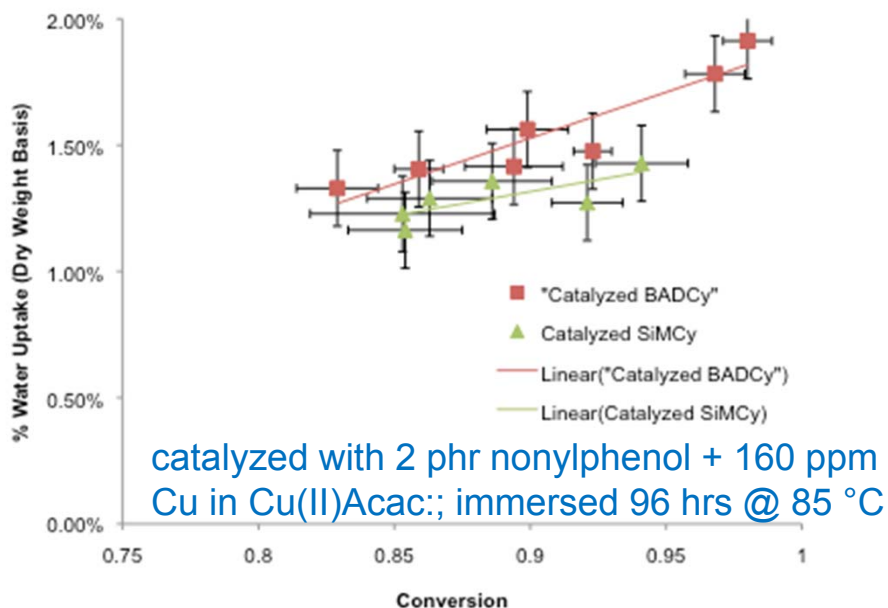
# Effect of Methylation on Moisture Uptake



- Addition of a methyl group near the cyanurate oxygen causes a significant reduction in the tendency of moisture uptake to increase at high conversions
- Addition of a methyl group far from the cyanurate oxygen has no effect on water uptake as a function of conversion
- Methyl groups near the cyanurate oxygen block the favored sites for water uptake
- Vitrification appears not to influence the results



# Effect of Silicon Substitution on Moisture Uptake



No catalyst; from Guenther et al., *Macromolecules* **2006** 39, 6046.

- Substitution of a silicon atom for a carbon atom in di(cyanate ester) network segments reduces moisture uptake by up to 50%
- In the corresponding tri(cyanate ester) networks, the same substitution increases moisture uptake by ~100% (i.e. effect is not intrinsic property of Si atom)
- Differences in vitrification also do not explain these effects, as they cause little change in moisture uptake



# Summary: Basic Studies of Moisture Uptake in Cyanate Ester Networks



- Many aspects of moisture uptake (and its minimization) in thermosetting polymer networks have been clarified through recent research efforts at AFRL/RQRP
- New tools and techniques for quantifying structure and properties of thermosetting networks have allowed for significant new insights into structure-property relationships in thermosetting networks
- Several unresolved issues relating to our understanding of moisture uptake in cyanate ester networks remain; approaches based on molecular modeling may provide insights leading to significant application payoffs

# QUESTIONS?



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