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ENGINEERING THE ACTIVITY AND LIFETIME OF HETEROGENEOUS CATALYSTS FOR CARBON NANOTUBE GROWTH VIA SUBSTRATE ION BEAM BOMBARDMENT (POSTPRINT)

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Engineering the Activity and Lifetime of Heterogeneous Catalysts for Carbon Nanotube Growth via Substrate Ion Beam Bombardment

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Supporting Information

ABSTRACT: We demonstrate that argon ion bombardment of single crystal sapphire leads to the creation of substrates that support the growth of vertically aligned carbon nanotubes from iron catalysts with a density, height, and quality equivalent to those grown on conventional, disordered alumina supports. We quantify the evolution of the catalyst using a range of surface characterization techniques and demonstrate the ability to engineer and pattern the catalyst support through control of ion beam bombardment parameters.



KEYWORDS: Carbon nanotubes, heterogeneous catalyst, catalyst support, ion beam bombardment

C arbon nanotubes (CNTs) have demonstrated unequaled structural,¹⁻³ transport,^{1,4-6} and electronic/optoelectronic⁷ properties. These properties have enabled the application of CNTs as ultralightweight composites in aerospace applications⁸⁻¹⁰ and promise future applications such as electrical cables,¹ interconnects,¹¹ bio/chemical sensors,^{12,13} energy storage devices,¹⁴⁻¹⁶ field-emitting devices,¹ and high speed electronics.¹⁷⁻²⁰ The realization of these applications requires improvements in large-scale production of CNTs with sufficient control over yield,²¹⁻²⁴ density,^{11,25} and chirality.^{19,20} CNTs are grown via heterogeneous catalysis using a thin film of catalyst on a wide variety of catalyst supports. Films of transition metals like Fe, Co, and Ni generally form the most active catalyst nanoparticles for vertically aligned CNT (VA-CNT) growth.^{26,27} The catalytic activity of these nanoparticles on solid supports is affected by water vapor^{21,22,28} and carbon feedstock.²³ In addition, the catalytic activity is most efficient only on certain types of catalyst supports, specifically alumina or silica.²⁹⁻³⁶ Even for materials like alumina or silica, only certain types of alumina,^{29-31,37,38} quartz,³⁹ or sapphire⁴⁰ yield extended catalytic activity and hence longer VA-CNT growth.

Our previous study of alumina supports deposited by different methods clarified the role of the support in extending catalyst lifetime.^{29–31} It was shown that while single-crystal, c-cut sapphire (Al_2O_3) did not support VA-CNT growth,

substoichiometric alumina (AlO_x) deposited by magnetron sputtering, electron beam evaporation, and atomic layer deposition (ALD) showed high levels of catalytic activity and CNT nucleation, resulting in ~10–100 μ m tall VA-CNT films. The key microstructural element for extending catalytic activity was established to be the porosity of the alumina support. Deposition techniques that led to higher porosity for the alumina support were shown to reduce coarsening of the deposited catalyst by inhibiting Ostwald ripening.²⁸

Here we expand upon these insights and demonstrate that ion beam bombardment can be used to create catalyst supports with controllable degrees of catalytic activity and lifetime by generating a similar critical porosity and surface activity as that of conventional substrates.^{29–36} Ion beam bombardment on crystalline substrates is generally known to introduce porosity at the top surface.^{41–43} We have used this approach to convert an inactive catalyst support into a highly active support that results in a controlled level of vertically aligned growth using Fe as a catalyst. In this paper, we have chosen c-cut sapphire as the inactive catalyst support, as it shows insignificant VA-CNT growth^{29,40} and hence provides a nice platform to engineer the

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Figure 1. Ion beam bombarded sapphire catalyst supports grow tall vertically aligned carbon nanotube arrays. (a) Low- and (b) high-magnification SEM images showing ~0.8 mm vertically aligned carbon nanotube (VA-CNT) growth using ~1 nm Fe catalyst film on an ion beam damaged sapphire surface using Ar^+ ion at a dose of ~2.1 × 10²⁰ cm⁻² and accelerated at 5 kV. (c) Raman spectra taken from the top of grown VA-CNT shows the presence of G' band, a G/D ratio of ~1.4, and radial breathing modes (inset) in the range of 150–300 cm⁻¹. (d) Transmission electron microscopy image of grown CNT shows presence of single-wall, double-wall, and multiwall CNTs, along with amorphous carbon. In comparison to ion beam damaged sapphire, insignificant amount of CNTs are observed on pristine sapphire samples in the (e) low- and (f) high-magnification SEM images.

catalytic activity via ion beam bombardment. We utilize wellestablished characterization approaches to understand how the ion beam bombardment modifies the sapphire structure and improves the resultant catalytic activity. The method described herein expands the choice of catalyst supports, an important component for CNT growth, from a few types of alumina (determined by the deposition technique) to a continuum of engineered substrates.

To demonstrate the effectiveness of ion beam bombardment in growing VA-CNTs, we modified a c-cut sapphire substrate by bombarding it with 5 kV Ar⁺ ions at a dose of $\sim 2.1 \times 10^{20}$ cm⁻² (see section S1 of Supporting Information for calculation of ion dose), then deposited ~ 1 nm Fe film, and finally subjected the substrate to a CNT growth process, as elaborated in Methods. Figure 1a shows that the resulting VA-CNT films are dense and well aligned with similar characteristics [e.g., overall carpet height (~0.8 mm) and Raman G/D ratio (1.4)] as we achieve using standard ALD alumina supports.²⁹ In comparison, a pristine sapphire sample (with no ion beam bombardment) subjected to the same growth conditions yielded only a few CNTs on the substrate and no vertically aligned growth (Figure 1e,f). The degree of VA-CNT growth in terms of carpet height (Figure 1a) and density (Figure 1b) is comparable to that obtained when using ALD or sputter-deposited AlO_x and lateral gas flow, although it is less than that obtained from showerhead growth.²⁴ Raman spectra taken from the top of the carpet (Figure 1c) suggests the presence of disordered CNTs with strong D and G' bands, with intensities comparable to that of the G band. The low G/D ratio (~1.4) from the Raman spectra is also an indicator of the abundance of

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multiwalled CNTs in the samples,^{44,45} a result that was confirmed using transmission electron microscopy (TEM) (Figure 1d). In addition, single-walled CNTs at the top of the carpet were observed via radial breathing mode (RBM) signals in the range of $150-300 \text{ cm}^{-1}$.

Modification of the catalyst support via Ar⁺ ion bombardment provides a novel way to pattern the catalyst activity by selectively masking certain parts of the sapphire from ion beam bombardment. Figure 2 shows VA-CNT growth in unmasked



Figure 2. Demonstration of patterned VA-CNT growth via masked ion beam bombardment in selective regions of a sapphire substrate: (a) Low- and (b) high-magnification SEM images show VA-CNTs grown in patterned areas of a sapphire substrate, where ion beam damage was performed using a SiO_x-coated Mo-TEM grid as shadow mask, while Fe catalyst was deposited all over the substrate.

areas of a sapphire substrate, where the catalyst support was modified via Ar^+ ion damage through a SiO_x -coated Mo TEM grid that acted as a simple shadow mask. Despite the deposition of Fe catalyst (~1 nm Fe film) uniformly over the entire substrate, including the regions of pristine sapphire, only the

ion bombarded regions grew VA-CNTs. The approach, therefore, demonstrates the unique advantage of limiting growth only in Ar^+ ion damaged areas. In contrast, patterning the catalyst film deposition^{22,46} to achieve selective area growth suffers from potential loss of fidelity in the pattern transfer via catalyst migration into regions outside the patterned area.

The catalyst supports that are currently used for VA-CNT growth²⁹⁻³⁶ can be considered "static", that is, the deposition method largely fixes their structure and activity. In contrast, ion beam bombardment enables continuous control over the structure of the catalyst support through independent control of the ion beam bombardment parameters. We explored the response of the catalyst support to changes in the ion beam accelerating voltage, as well as the ion beam dose. Figure 3a,b shows the measured height of VA-CNT carpets (like the one shown in Figure 1a) grown on sapphire surfaces damaged with varying ion beam parameters. Increased degrees of Ar⁺ ion damage, caused by either increasing the accelerating voltage at a fixed damage dose (Figure 3a) or by increasing the damage dose at a fixed acceleration voltage (Figure 3b), results in higher carpet height. Observed variations in nanotube growth correlate nicely with the changes in catalyst morphology and also with the changes in both the structure and chemistry of the catalyst support.29-31

To monitor the changes in catalyst morphology, Ar^+ ion damaged sapphire substrates were coated with the same ~1 nm Fe catalyst film as used for nanotube growth and then annealed in H₂ ambient at 585 °C for 10 min. The resultant catalyst morphology replicates the condition at the onset of VA-CNT growth. Annealing of the catalyst film in an H₂ ambient induces dewetting and leads to the formation of iron nanoparticles on top of the engineered sapphire surfaces. Nanoparticles formed following a lower degree of Ar⁺ ion damage (Figure 4a and

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Figure 3. Control of VA-CNT growth and catalyst evolution via ion beam bombardment: Height of VA-CNT carpets grown on ion beam damaged sapphire substrates as a function of (a) Ar^+ ion dose at a fixed voltage and (b) accelerating voltage at a fixed dose. (c,d) Surface roughness and (e,f) nanoparticle density of the ripened catalyst on ion beam damaged sapphire substrates. These values were obtained via AFM (see Figure 4 and Supporting Information Figure S1 for example AFM images) after a sputter-deposited 1 nm Fe film on ion beam damaged sapphire was annealed in H₂ ambient at 585 °C for 10 min. For each ion beam bombardment conditions in (c-f), several (5–7) AFM topographies at different parts of the substrate were used to obtain the averages and standard deviations. Ion beam damage conditions for (c,e) and (d,f) are same as that for (a,b), respectively. Trends indicate that increasing damage, either via increase in dosage or increase in ion energy, results in similar improvement in VA-CNT growth and heterogeneous catalysis of Fe on sapphire.



Figure 4. AFM topography of Fe nanoparticles on sapphire surfaces showing reduced Ostwald ripening rate with increased degree of damage: Sapphire surfaces were damaged using Ar^+ ion at doses of (a) 7×10^{19} cm⁻², (b) 1.4×10^{20} cm⁻², and (c) 2.1×10^{20} cm⁻² at a fixed accelerating voltage of 6 kV. To form these nanoparticles on damaged sapphire surfaces, Fe film of ~1 nm was coated on the surfaces, which was followed by a 10 min hydrogen anneal at 585 °C. Catalyst particles show a reduction in size (with lower surface roughness) and increase in areal number density with increasing damage, as summarized in Figure 3c,e.

Supporting Information, Figure S1a) are larger and isolated, while those formed following higher degrees of Ar^+ ion damage are smaller and are more closely spaced (Figure 4b,c and Supporting Information, Figure S1b,c). Atomic force microscopy (AFM) measurements of the ripened catalyst particles on the engineered surfaces exposed to higher degrees of Ar^+ ion damage show lower surface roughness (Figure 3c,d) and higher particle number density (Figure 3e,f), both of which are known to enhance catalytic activity.²⁹ On the basis of our previous studies regarding catalyst morphology and carpet height,^{29–31,34,38} the results in Figure 3 demonstrate control over the catalytic activity by controlling the Ostwald ripening rate.

In order to investigate the influence of the microstructure and surface properties on catalyst activity, we have used crosssectional TEM (X-TEM) and X-ray photoelectron spectroscopy (XPS) to characterize the ion beam-damaged sapphire. The X-TEM image of an Ar⁺ ion engineered sapphire substrate (Figure 5) damaged with 3.8×10^{20} Ar⁺ ions/cm² at an



Figure 5. Cross-section TEM image of an ion beam damaged sapphire substrate: The substrate was bombarded with 3.8×10^{20} Ar⁺ ions/cm² at an acceleration voltage of 5 kV and then coated with a ~ 1 nm Fe film. TEM image shows the creation of amorphous and nanocrystalline interfacial layers on top of crystalline sapphire with a total damage depth of ~13 nm.

acceleration voltage of 5 kV shows the presence of ~13 nm thick damaged layer near the surface of the substrate that acts as the porous catalyst support. (Supporting Information Figure S2 illustrates how energy filtered maps of iron, carbon, and oxygen were used to identify different regions of Figure 5.) Ar^+ ion damage presumably results in epitaxial regrowth of an

amorphous layer, followed by a nanocrystalline interfacial layer between the amorphous and crystalline regions (see Figure 5). This process has been studied extensively in the context of amorphous layer formation during ion implantation of both silicon and sapphire substrates.⁴¹⁻⁴³ Supporting Information Figure S3 presents X-ray reflection (XRR) data measured using Cu K α radiation (wavelength, $\lambda = 0.154$ nm) on ion beam bombarded sapphire substrates prepared by varying Ar^+ dose from 1.4×10^{19} to 2.1×10^{20} ions/cm² at an acceleration voltage of 5 kV. Analysis of XRR data reveals variation of damage layer thickness from ~9.4 to ~12.8 nm with the increase in ion beam damage dose. The lowest damage depth of ~9.4 nm is comparable to the thickness of ALD alumina (~10 nm) routinely used for VA-CNT growth.^{29-31,37,38} This lowest ion beam damaged substrate resulted ~0.4 mm tall VA-CNT carpet after completion of the catalyst annealing and the CNT growth phases. Experimental intricacy to produce a reliable beam current with shorter ion exposure inhibited us from determining the threshold dosage required for the initiation of VA-CNT growth.

Finally, XPS is used to characterize surface chemistry of the catalyst support following ion beam bombardment. Measurements of surface chemistry show good correlation with the catalyst activity. As has been recently shown,³¹ deviations from Al₂O₃ stoichiometry at the surface can be characterized by measuring the oxygen-to-aluminum (O/Al) ratio at the surface, while the surface active sites can be detected by monitoring the increase in surface-active oxygen species relative to stable oxygen. Figure 6a,b presents the ratios of the O 1s ($O_2^$ component) peak area to the Al 2p peak area and the full width at half-maximum (FWHM) for the O 1s peak, respectively, for sapphire surfaces that were damaged at different degrees of Ar⁺ ion bombardment by varying the acceleration voltage. Figure 6c-e shows how the O 1s spectra obtained for different sapphire surfaces are decomposed into contributions from OH⁻ (with a peak at ~532.1 \pm 0.2 eV) and O₂⁻ (with a peak at \sim 530.9 ± 0.2 eV), which is important for the correct extraction of the O/Al ratio and the FWHM.⁴⁷ The values obtained for the O/Al ratio and FWHM of our engineered catalyst supports are consistent with similar values (dotted lines in Figure 6a,b) recently measured for ALD-AlO_x 31,47 which is routinely used as catalyst support for VA-CNT growth. An increase in both the O/Al atomic ratio and the O 1s peak width with the increase in ion beam damage suggests a hydroxyl enrichment on the engineered sapphire surface, which can be correlated with the higher degree of catalytic activity³¹ and longer lifetime of the Fe

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Figure 6. Comparison of ALD alumina and ion beam damaged sapphire surfaces via XPS shows a controlled increase in surface activity with increase ion beam damage: (a) the ratios of the O 1s (O_2^- component) peak area to the Al 2p peak area (O/Al ratio) and (b) FWHM for the O 1s peak for engineered sapphire surfaces with increasing degrees of Ar⁺ ion bombardment. To calculate the O/Al ratio, each O 1s spectra is decomposed into contributions from OH⁻ and O_2^- . Such decomposition is shown for the O 1s spectra obtained at the surface of (c) pristine sapphire and Ar⁺ ion damaged sapphire at (d) 3 and (e) 5 kV accelerating voltage. These XPS results demonstrate the ability of ion beam damage to engineer the catalyst support surface over a wide range of conditions.

catalyst nanoparticles formed on the damaged substrate surface. We can thus conclude that the Ar^+ ion bombardment makes the sapphire surface nonstoichiometric and enriched in hydroxyl groups and introduces the disorder that has been shown to enhance catalytic activity and lifetime.

In conclusion, a new approach⁴⁸ has been developed to transform a relatively inert sapphire substrate into a highly active catalyst support that enables supergrowth of carbon nanotubes. The resultant supports grow nanotubes with a density, height, and quality equivalent to those obtained from conventional, vapor-deposited alumina-based catalyst supports. The correlation between the observed formation of surfaces that are disordered, nonstoichiometric, and enriched in hydroxyl groups is consistent with and further validates our understanding of the roles Ostwald ripening, coarsening, and surface chemistry as the defining mechanisms for nanotube growth.²⁸⁻³¹ The contrast between the vertically aligned CNT growth on areas bombarded with Ar⁺ ions and the lack of growth on the pristine substrate highlights the sharp difference in catalyst dynamics between an atomically perfect surface and an intentionally disrupted one. The ion beam bombardment method of modifying catalyst supports enables one to engineer the degree of catalyst activation through control of the ion beam energy, dose, and spatial patterning. These results can be complimented by the rich literature concerning ion beam processing of surface to further understand, improve, and control nanotube growth for electronic and thermal applications. Finally, these results indicate that ion bombardment can be considered as yet another method in catalysis science to engineer supports to enhance both catalytic activity and lifetime with general implications for heterogeneous catalysis.^{49,50}

Methods. *Experiment.* For VA-CNT growth on crystalline c-cut sapphire substrates, the substrates were bombarded with

argon ions (Ar⁺) accelerated at 3-6 kV and at variable ion fluxes $(0.7-2.1 \times 10^{20} \text{ cm}^{-2})$. A thin film (~1 nm) of Fe was deposited via ion beam sputtering on the Ar⁺ ion-damaged samples as the source of catalyst for VA-CNT growth. The Fedeposited, Ar⁺ ion-damaged samples were annealed in hydrogen ambient (200 sccm flow) at 585 °C for 10 min to dewet the catalyst layer into discrete nanoparticles. The samples were then rapidly cooled down to room temperature in a hydrogen ambient. Some samples were removed from the reactor at this point to characterize the degree of catalyst evolution using atomic force microscopy (AFM). Other samples were immediately subjected to growth conditions at 760 °C for 30 min, at a water concentration of 50-80 ppm, and gas flow rates of 470, 100, and 25 sccm for argon, hydrogen, and ethylene, respectively. Grown nanotubes were analyzed using scanning electron microscopy (SEM), Raman spectroscopy, and transmission electron microscopy (TEM) for qualitative and quantitative analyses. In addition, surface properties of the ion beam-bombarded catalyst support were characterized using AFM and X-ray photoelectron spectroscopy (XPS), and the degree of Ar⁺ ion damage was characterized using crosssectional TEM. XPS analysis was performed using a Surface Science Instruments (SSI) M-probe equipped with an Al K α Xray source (operated at $\sim 4 \times 10^{-7}$ Pa base pressure) and collected data were analyzed using CASA XPS software using Shirley background subtraction. AFM was carried out in a Veeco Nanoscope Multimode instrument using MikroMasch HQ:NSC15/Al-BS (having tip radius ~8 nm, resonant frequency 265-410 kHz, and force constant 20-80 N/m). TEM studies were performed in a FEI Titan Environmental TEM at an accelerating voltage of 300 kV.

ASSOCIATED CONTENT

Supporting Information

The calculation procedure of ion beam damage dose, example AFM images analyzed to calculate surface roughness and nanoparticle density in Figure 3d,f, energy-filtered TEM images showing the amount of chemical species at different locations of Figure 5, and the analysis of ion beam damaged layers on the surface of c-cut sapphire using X-ray reflectivity measurements for different ion damage dose are presented. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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REFERENCES

(1) Behabtu, N.; Young, C. C.; Tsentalovich, D. E.; Kleinerman, O.; Wang, X.; Ma, A. W. K.; Bengio, E. A.; ter Waarbeek, R. F.; de Jong, J. J.; Hoogerwerf, R. E.; et al. Strong, Light, Multifunctional Fibers of Carbon Nanotubes with Ultrahigh Conductivity. Science 2013, 339, 182-186.

(2) Qu, L.; Dai, L.; Stone, M.; Xia, Z.; Wang, Z. L. Carbon nanotube arrays with strong shear binding-on and easy normal lifting-off. Science 2008. 322. 238-242.

(3) Wong, E. W.; Sheehan, P. E.; Lieber, C. M. Nanobeam mechanics: Elasticity, strength, and toughness of nanorods and nanotubes. Science 1997, 277, 1971-1975.

(4) Pop, E.; Mann, D.; Wang, Q.; Goodson, K. E.; Dai, H. J. Thermal Conductance of an Individual Single-Wall Carbon Nanotube above Room Temperature. Nano Lett. 2006, 6, 96-100.

(5) Zhou, X. J.; Park, J. Y.; Huang, S. M.; Liu, J.; McEuen, P. L. Band structure, phonon scattering, and the performance limit of singlewalled carbon nanotube transistors. Phys. Rev. Lett. 2005, 95.

(6) Yao, Z.; Kane, C. L.; Dekker, C. High-field electrical transport in single-wall carbon nanotubes. Phys. Rev. Lett. 2000, 84, 2941-2944.

(7) Avouris, P.; Freitag, M.; Perebeinos, V. Carbon-nanotube photonics and optoelectronics. Nat. Photonics 2008, 2, 341-350.

(8) Breuer, O.; Sundararaj, U. Big returns from small fibers: A review of polymer/carbon nanotube composites. Polym. Compos. 2004, 25, 630-645.

(9) De Rosa, I. M.; Sarasini, F.; Sarto, M. S.; Tamburrano, A. EMC impact of advanced carbon fiber/carbon nanotube reinforced composites for next-generation aerospace applications. IEEE Trans. Electromagn. Compat. 2008, 50, 556-563.

(10) Sandler, J. K. W.; Kirk, J. E.; Kinloch, I. A.; Shaffer, M. S. P.; Windle, A. H. Ultra-low electrical percolation threshold in carbonnanotube-epoxy composites. Polymer 2003, 44, 5893-5899.

(11) Esconjauregui, S.; Fouquet, M.; Bayer, B. C.; Ducati, C.; Smajda, R.; Hofmann, S.; Robertson, J. Growth of Ultrahigh Density Vertically Aligned Carbon Nanotube Forests for Interconnects. ACS Nano 2010, 4, 7431-7436.

(12) Pengfei, Q. F.; Vermesh, O.; Grecu, M.; Javey, A.; Wang, O.; Dai, H. J.; Peng, S.; Cho, K. J. Toward Large Arrays of Multiplex Functionalized Carbon Nanotube Sensors for Highly Sensitive and Selective Molecular Detection. Nano Lett. 2003, 3, 347-351.

(13) Collins, P. G.; Bradley, K.; Ishigami, M.; Zettl, A. Extreme oxygen sensitivity of electronic properties of carbon nanotubes. Science 2000, 287, 1801-1804.

(14) Cao, Z.; Wei, B. A perspective: carbon nanotube macro-films for energy storage. Energy Environ. Sci. 2013, 6, 3183-3201.

(15) Dillon, A. C. Carbon Nanotubes for Photoconversion and Electrical Energy Storage. Chem. Rev. 2010, 110, 6856-6872.

(16) Zhang, Q.; Huang, J.-Q.; Qian, W.-Z.; Zhang, Y.-Y.; Wei, F. The Road for Nanomaterials Industry: A Review of Carbon Nanotube Production, Post-Treatment, and Bulk Applications for Composites and Energy Storage. Small 2013, 9, 1237-1265.

(17) Avouris, P. Nanotube electronics - Electronics with carbon nanotubes. Phys. World 2007, 20, 40-45.

(18) Cao, Q.; Han, S.-j. Single-walled carbon nanotubes for highperformance electronics. Nanoscale 2013, 5, 8852-8863.

(19) Cao, Q.; Han, S.-j.; Tulevski, G. S.; Zhu, Y.; Lu, D. D.; Haensch, W. Arrays of single-walled carbon nanotubes with full surface coverage for high-performance electronics. Nat. Nanotechnol. 2013, 8, 180-186.

(20) Jin, S. H.; Dunham, S. N.; Song, J.; Xie, X.; Kim, J.-H.; Lu, C.; Islam, A.; Du, F.; Kim, J.; Felts, J.; et al. Using nanoscale thermocapillary flows to create arrays of purely semiconducting single-walled carbon nanotubes. Nat. Nanotechnol. 2013, 8, 347-355.

(21) Hasegawa, K.; Noda, S. Millimeter-Tall Single-Walled Carbon Nanotubes Rapidly Grown with and without Water. ACS Nano 2011, 5, 975-984.

(22) Hata, K.; Futaba, D. N.; Mizuno, K.; Namai, T.; Yumura, M.; Iijima, S. Water-assisted highly efficient synthesis of impurity-free single-waited carbon nanotubes. Science 2004, 306, 1362-1364.

(23) Kimura, H.; Goto, J.; Yasuda, S.; Sakurai, S.; Yumura, M.; Futaba, D. N.; Hata, K. Unexpectedly High Yield Carbon Nanotube Synthesis from Low-Activity Carbon Feedstocks at High Concentrations. ACS Nano 2013, 7, 3150-3157.

(24) Yasuda, S.; Futaba, D. N.; Yamada, T.; Satou, J.; Shibuya, A.; Takai, H.; Arakawa, K.; Yumura, M.; Hata, K. Improved and Large Area Single-Walled Carbon Nanotube Forest Growth by Controlling the Gas Flow Direction. ACS Nano 2009, 3, 4164-4170.

(25) Naeemi, A.; Meindl, J. D. Design and performance modeling for single-walled carbon nanotubes as local, semiglobal, and global interconnects in gigascale integrated systems. IEEE Trans. Electron Devices 2007, 54, 26-37.

(26) Hafner, J. H.; Bronikowski, M. J.; Azamian, B. R.; Nikolaev, P.; Rinzler, A. G.; Colbert, D. T.; Smith, K. A.; Smalley, R. E. Catalytic growth of single-wall carbon nanotubes from metal particles. Chem. Phys. Lett. 1998, 296, 195-202.

(27) Yuan, D.; Ding, L.; Chu, H.; Feng, Y.; McNicholas, T. P.; Liu, J. Horizontally Aligned Single-Walled Carbon Nanotube on Quartz from a Large Variety of Metal Catalysts. Nano Lett. 2008, 8, 2576-2579.

(28) Amama, P. B.; Pint, C. L.; McJilton, L.; Kim, S. M.; Stach, E. A.; Murray, P. T.; Hauge, R. H.; Maruyama, B. Role of Water in Super Growth of Single-Walled Carbon Nanotube Carpets. Nano Lett. 2009, 9.44-49.

(29) Amama, P. B.; Pint, C. L.; Kim, S. M.; McJilton, L.; Eyink, K. G.; Stach, E. A.; Hauge, R. H.; Maruyama, B. Influence of Alumina Type on the Evolution and Activity of Alumina-Supported Fe Catalysts in Single-Walled Carbon Nanotube Carpet Growth. ACS Nano 2010, 4, 895-904.

(30) Amama, P. B.; Pint, C. L.; Mirri, F.; Pasquali, M.; Hauge, R. H.; Maruyama, B. Catalyst-support interactions and their influence in water-assisted carbon nanotube carpet growth. Carbon 2012, 50, 2396-2406.

(31) Amama, P. B.; Putnam, S. A.; Barron, A. R.; Maruyama, B. Wetting behavior and activity of catalyst supports in carbon nanotube carpet growth. Nanoscale 2013, 5, 2642-2646.

(32) de los Arcos, T.; Garnier, M. G.; Oelhafen, P.; Mathys, D.; Seo, J. W.; Domingo, C.; Garci-Ramos, J. V.; Sanchez-Cortes, S. Strong influence of buffer layer type on carbon nanotube characteristics. Carbon 2004, 42, 187-190.

(33) Han, S. J.; Yu, T. K.; Park, J.; Koo, B.; Joo, J.; Hyeon, T.; Hong, S.; Im, J. Diameter-Controlled Synthesis of Discrete and Uniform-Sized Single-Walled Carbon Nanotubes Using Monodisperse Iron Oxide Nanoparticles Embedded in Zirconia Nanoparticle Arrays as Catalysts. J. Phys. Chem. B 2004, 108, 8091-8095.

(34) Kim, S. M.; Pint, C. L.; Amama, P. B.; Hauge, R. H.; Maruyama, B.; Stach, E. A. Catalyst and catalyst support morphology evolution in single-walled carbon nanotube supergrowth: Growth deceleration and termination. J. Mater. Res. 2010, 25, 1875-1885.

(35) Mattevi, C.; Wirth, C. T.; Hofmann, S.; Blume, R.; Cantoro, M.; Ducati, C.; Cepek, C.; Knop-Gericke, A.; Milne, S.; Castellarin-Cudia, C.; et al. In-Situ X-ray Photoelectron Spectroscopy Study of Catalyst-Support Interactions and Growth of Carbon Nanotube Forests. J. Phys. Chem. C 2008, 112, 12207-12213.

(36) Wang, B.; Yang, Y.; Li, L.-J.; Chen, Y. Effect of different catalyst supports on the (n,m) selective growth of single-walled carbon nanotube from Co-Mo catalyst. J. Mater. Sci. 2009, 44, 3285-3295.

(37) Kim, S. M.; Pint, C. L.; Amama, P. B.; Zakharov, D. N.; Hauge, R. H.; Maruyama, B.; Stach, E. A. Understanding Growth Termination of Single-Walled Carbon Nanotube Carpets by Documenting the Evolution of Catalyst Morphology with the Transmission Electron Microscope. Microsc. Microanal. 2009, 15, 1176-1177.

(38) Kim, S. M.; Pint, C. L.; Amama, P. B.; Zakharov, D. N.; Hauge, R. H.; Maruyama, B.; Stach, E. A. Evolution in Catalyst Morphology Leads to Carbon Nanotube Growth Termination. J. Phys. Chem. Lett. 2010, 1, 918-922.

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(39) Murakami, Y.; Chiashi, S.; Miyauchi, Y.; Hu, M. H.; Ogura, M.; Okubo, T.; Maruyama, S. Growth of vertically aligned single-walled carbon nanotube films on quartz substrates and their optical anisotropy. *Chem. Phys. Lett.* **2004**, 385, 298–303.

(40) Ohno, H.; Takagi, D.; Yamada, K.; Chiashi, S.; Tokura, A.; Homma, Y. Growth of vertically aligned single-walled carbon nanotubes on alumina and sapphire substrates. *Jpn. J. Appl. Phys.* **2008**, 47, 1956–1960.

(41) Linnros, J.; Svensson, B.; Holmen, G. Ion-beam-induced Epitaxial Regrowth of Amorphous Layers in Silicon on Sapphire. *Phys. Rev. B* **1984**, *30*, 3629–3638.

(42) Canut, B.; Benyagoub, A.; Marest, G.; Meftah, A.; Moncoffre, N.; Ramos, S. M. M.; Studer, F.; Thevenard, P.; Toulemonde, M. Swift-Uranium-ion-induced Damage in Sapphire. *Phys. Rev. B* **1995**, *51*, 12194–12201.

(43) Golecki, I.; Chapman, G. E.; Lau, S. S.; Tsaur, B. Y.; Mayer, J. W. Ion-beam Induced Epitaxy of Silicon. *Phys. Lett. A* **1979**, *71*, 267–269.

(44) Eswaraiah, V.; Sankaranarayanan, V.; Ramaprabhu, S. Inorganic nanotubes reinforced polyvinylidene fluoride composites as low-cost electromagnetic interference shielding materials. *Nanoscale Res. Lett.* **2011**, *6*, 137.

(45) Chakrabarti, S.; Kume, H.; Pan, L.; Nagasaka, T.; Nakayama, Y. Number of Walls Controlled Synthesis of Millimeter-Long Vertically Aligned Brushlike Carbon Nanotubes. *J. Phys. Chem. C* **2007**, *111*, 1929–1934.

(46) Yun, Y.; Shanov, V.; Tu, Y.; Subramaniam, S.; Schulz, M. J. Growth Mechanism of Long Aligned Multiwall Carbon Nanotube Arrays by Water-Assisted Chemical Vapor Deposition. *J. Phys. Chem. B* **2006**, *110*, 23920–23925.

(47) van den Brand, J.; Sloof, W. G.; Terryn, H.; de Wit, J. H. W. Correlation between hydroxyl fraction and O/Al atomic ratio as determined from XPS spectra of aluminium oxide layers. *Surf. Interface Anal.* **2004**, *36*, 81–88.

(48) Maruyama, B.; Sargent, G.; Islam, A. E. Methods of Enhancing Carbon Nanotube Growth on Substrates. Patent Pending.

(49) Lu, J.; Fu, B.; Kung, M. C.; Xiao, G.; Elam, J. W.; Kung, H. H.; Stair, P. C. Coking- and Sintering-Resistant Palladium Catalysts Achieved Through Atomic Layer Deposition. *Science* **2012**, 335, 1205–1208.

(50) Park, J. B.; Graciani, J.; Evans, J.; Stacchiola, D.; Ma, S.; Liu, P.; Nambu, A.; Fernandez Sanz, J.; Hrbek, J.; Rodriguez, J. A. High catalytic activity of $Au/CeO_x/TiO_2(110)$ controlled by the nature of the mixed-metal oxide at the nanometer level. *Proc. Natl. Acad. Sci. U.S.A.* **2009**, *106*, 4975–4980.

NOTE ADDED IN PROOF

A recent publication has also demonstrated the utility of oxygen plasma bombardment in aiding VA-CNT growth. See J. Yang, et al., J. Phys. Chem. C, Article ASAP, DOI: 10.1021/jp5022196.