SHORT COURSE Tools for Management of Chlorinated Solvent – Contaminated Sites

3 December 2009





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Report Documentation Page

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Short Course Agenda

8:30 AM	Welcome and Introduction	Hans Stroo		
8:40 AM	Source Zone Protocol for Remedy Selection of Chlorinated Solvents Released at DoD Facilities	Thomas Sale & Charles Newell		
9:50 AM	Break			
10:10 AM	Development of a Protocol and Screening Tool for Selection of DNAPL Source Area Remediation	Carmen Lebrón, Bernard Kueper, David Major, Julie Konzuk & Jason Gerhard		
11:50 AM	Lunch			
1:00 PM	Decision & Management Tools for DNAPL Sites: Optimization of Chlorinated Solvent Source and Plume Remediation Considering Uncertainty	Ronald Falta & Charles Newell		
2:20 PM	Emulsion Design Tool for Planning Aqueous Amendment Injections Systems	Robert Borden		
2:50 PM	Break			
3:10 PM	Permanganate Design Tool for Planning Aqueous Amendment Injection Systems	Robert Borden		
4:00 PM	Improved Field Evaluation of NAPL Dissolution and Source Longevity	Michael Kavanaugh, Mark Widdowson, Lloyd Stewart & Rula Deeb		
5:20 PM	Summary & Conclusion	Hans Stroo		





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Guide for Selecting Remedies for Subsurface Releases of Chlorinated Solvents ER-0530

Tom Sale, Chuck Newell,

Hans Stroo, Rob Hinchee, and Paul Johnson

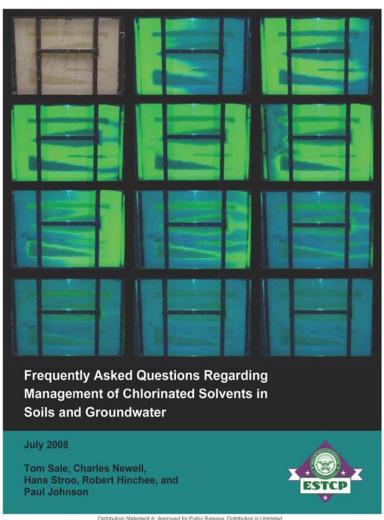






Frequently Asked Questions

- Provides quick access to key concepts and references for those who need to know more
- August 2008
- Google Chlorinated Solvents FAQs
- http://www.estcp.org/Technology/upload/ER-0530-FAQ.pdf

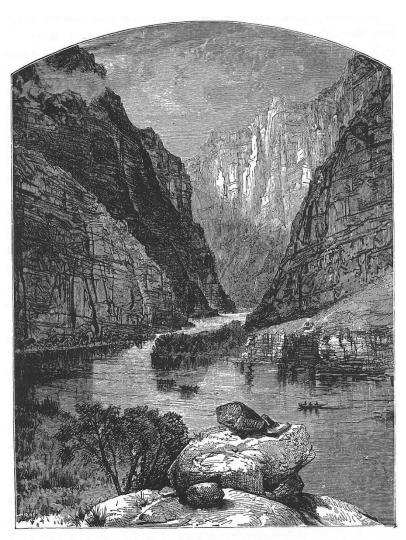


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Decision Guide

- Supports
 - Understanding site specific conditions,
 - Developing goals,
 - Selecting technologies, and
 - Packaging site remedies



GATE OF LODORE.



Decision Guide

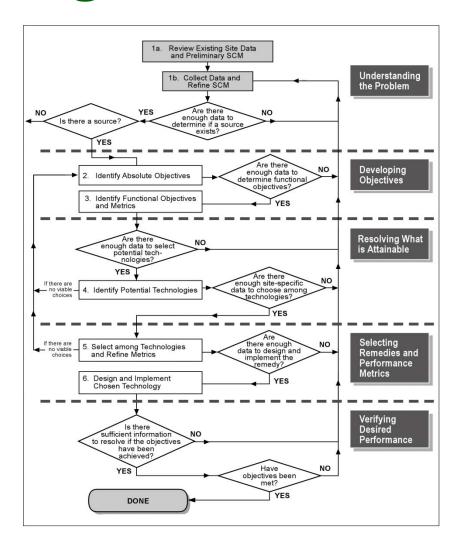
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- └--- **y** Packaging site remedies



GATE OF LODORE.



Following NRC 2005

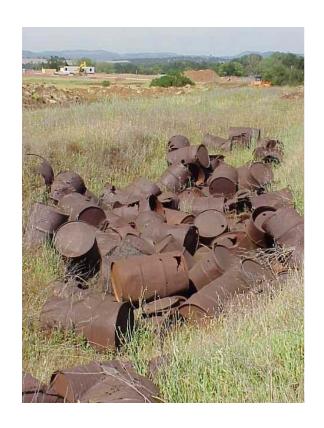




Understanding site specific conditions

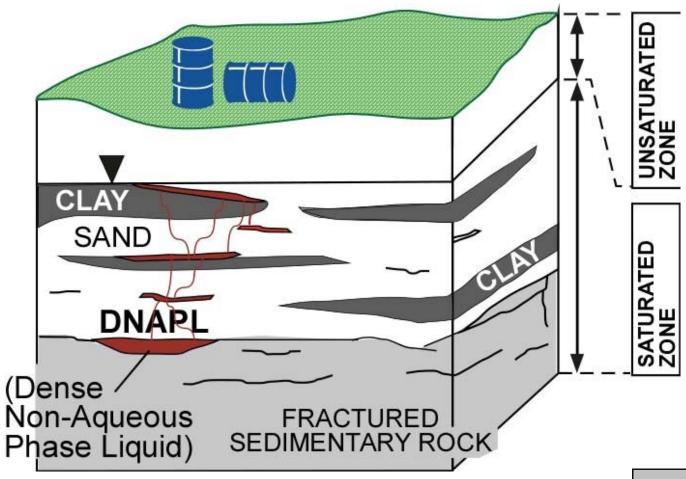


Inadvertent releases reflecting past practices...





Early Stage

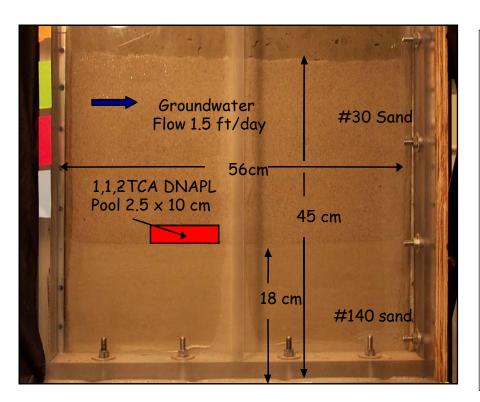


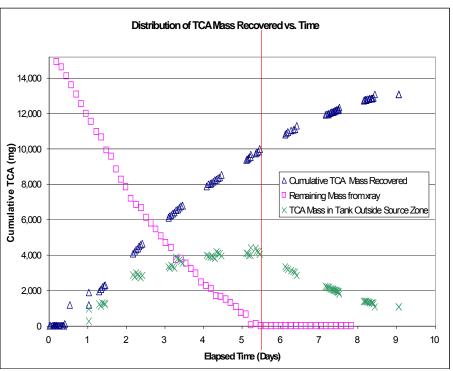
Sale et al., 2008



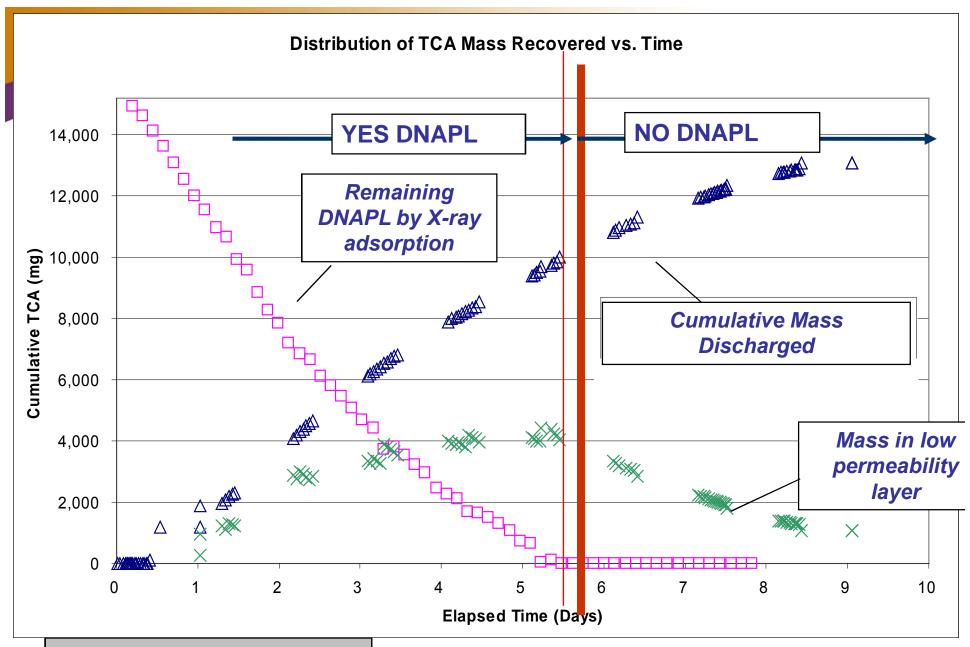
Two layer sand tank study

Colorado School of Mines (Tissa Illangasekare and Bart Wilkins)



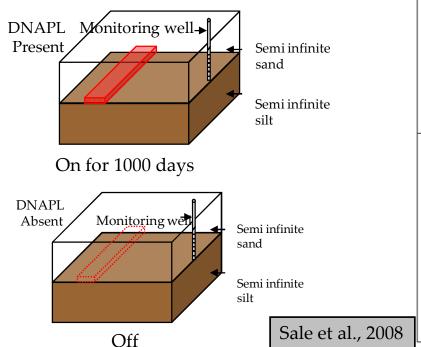


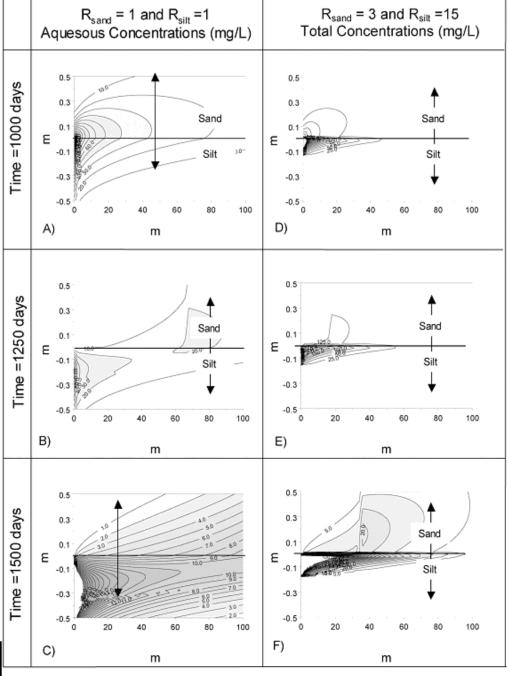
AFCEE Source Zone Initiative (2007)



AFCEE Source Zone Initiative (2007)

Aqueous and sorbed phases in transmissive and low permeability zone







Back Diffusion – The Movie

Lee Ann Doner – (2008) MS CSU





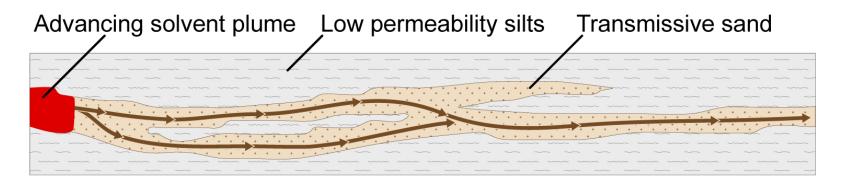
"Sandy aquifers"

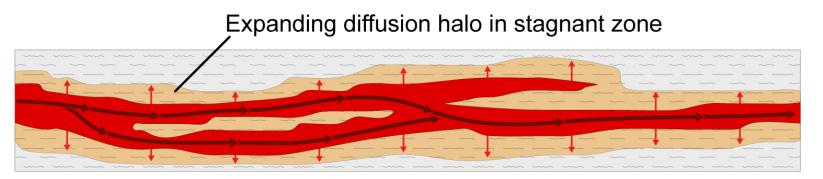


Image from Fred Payne / ARCADIS

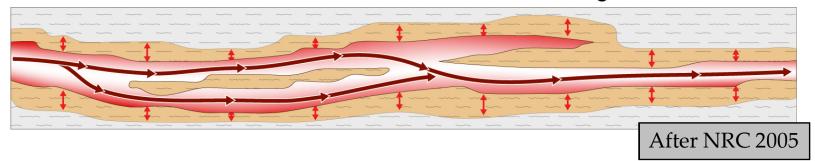


New Paradigm



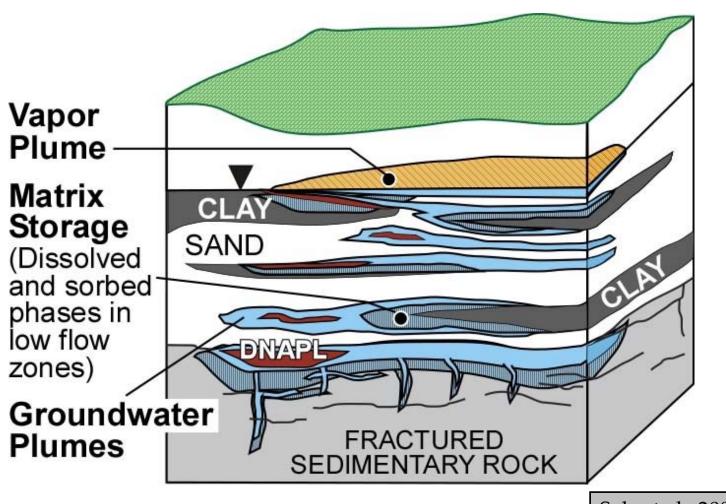


Simultaneous inward and outward diffusion in stagnant zones





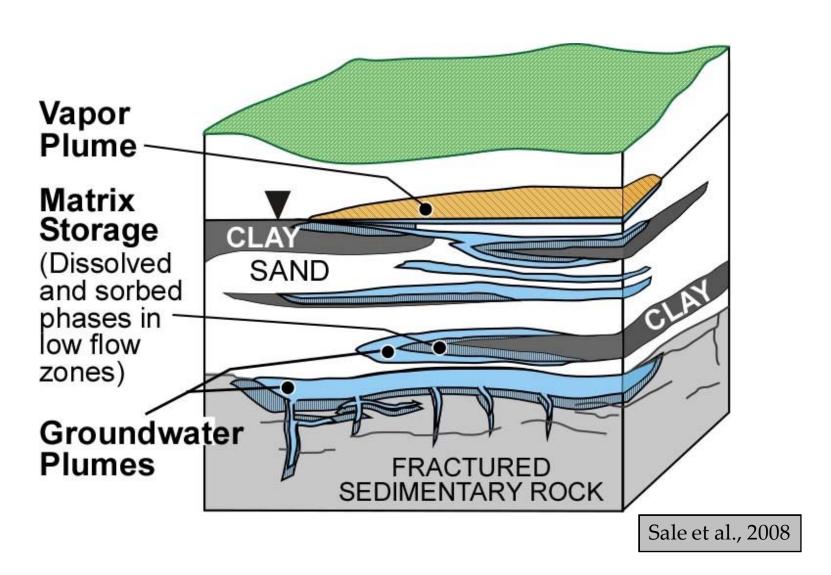
Middle Stage



Sale et al., 2008



Late Stage





The 14 Compartments Model

(a holistic perspective)

	Source Zone		Plume	
Phase/Zone	Low Permeability	Transmissive	Transmissive	Low Permeability
Vapor				
DNAPL			NA	NA
Aqueous				
Sorbed				

Sale et al., 2008



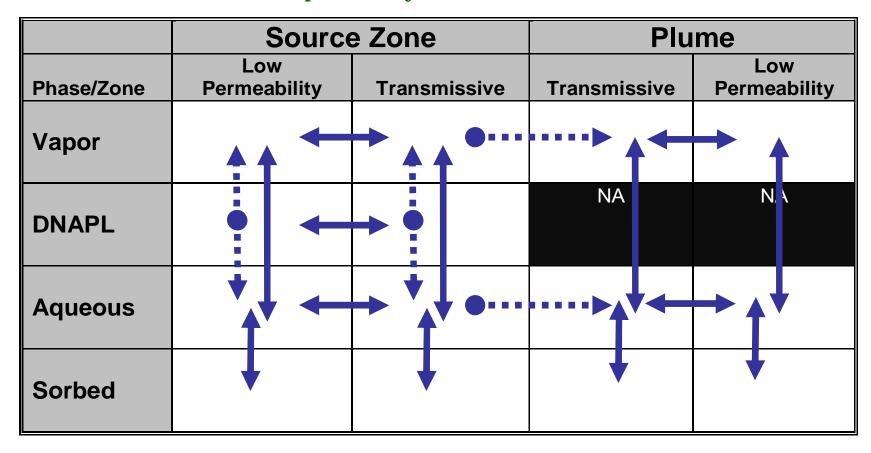
With interdependencies (Option 1)

	Source Zone		Plume	
Phase/Zone	Low Permeability	5 Transmissive	7 Transmissive	7 Low Permeability
Vapor				
DNAPL 3			NA	NA
Aqueous				
Sorbed				



With interdependencies (Option 2)

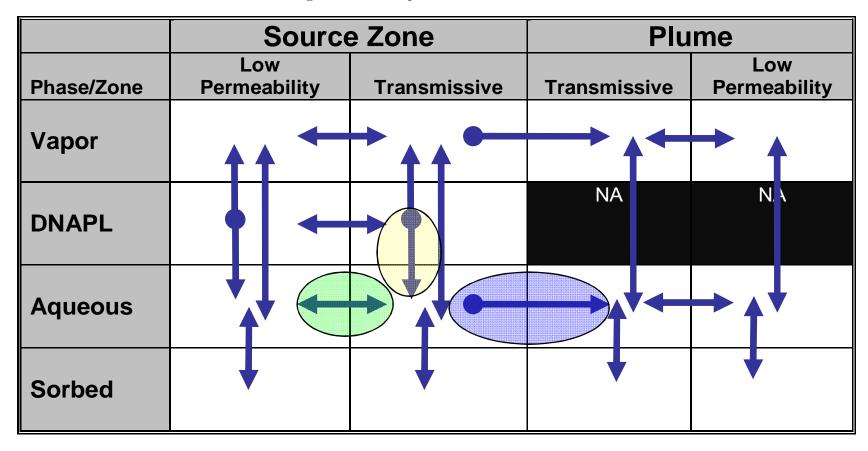
Lattice of 17 potentially relevant fluxes





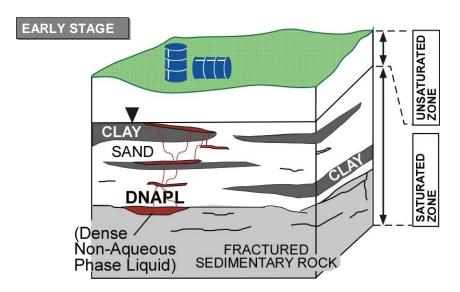
With interdependencies (Option 2)

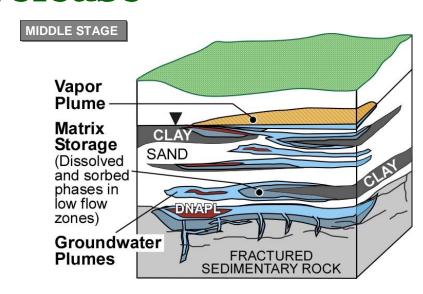
Lattice of 17 potentially relevant fluxes

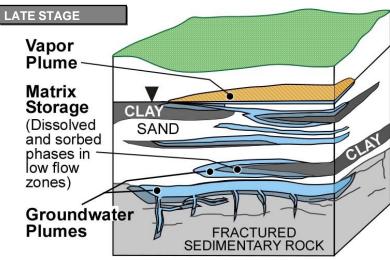




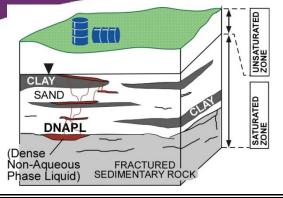
Mapping the evolution of a chlorinated solvent release









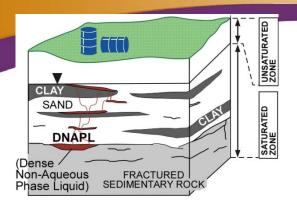


Early Stage

Gw. or equivalent gw. conc.

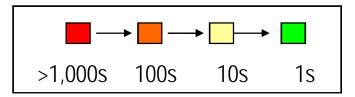


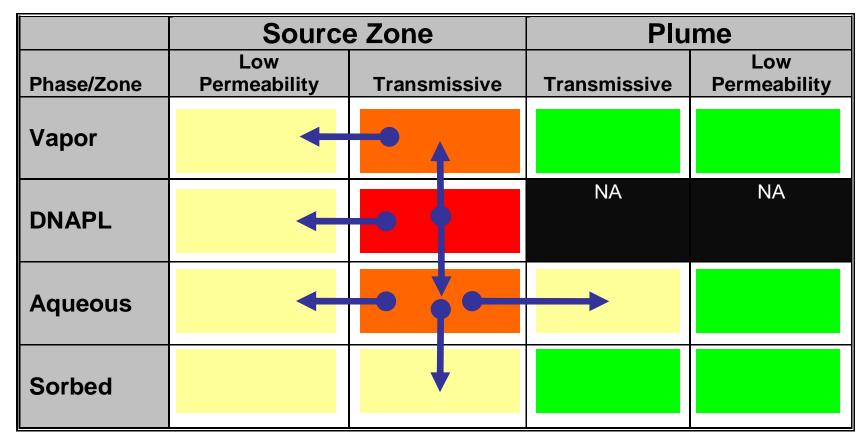
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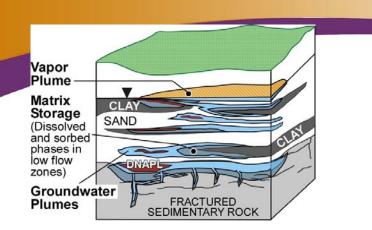




Early Stage



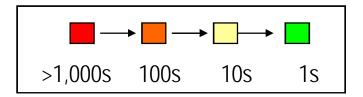


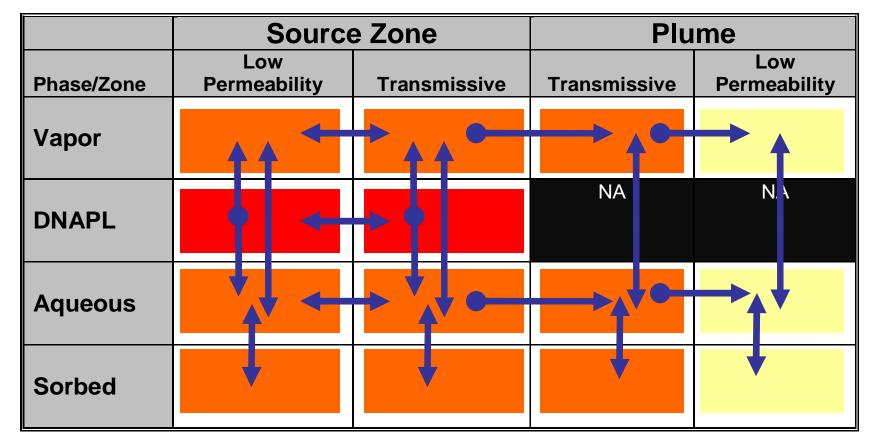


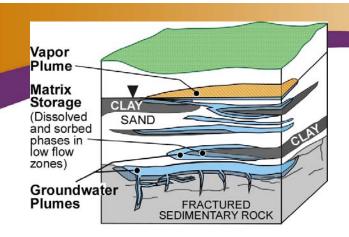




Middle Stage



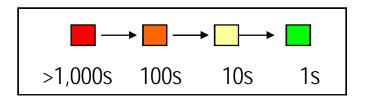


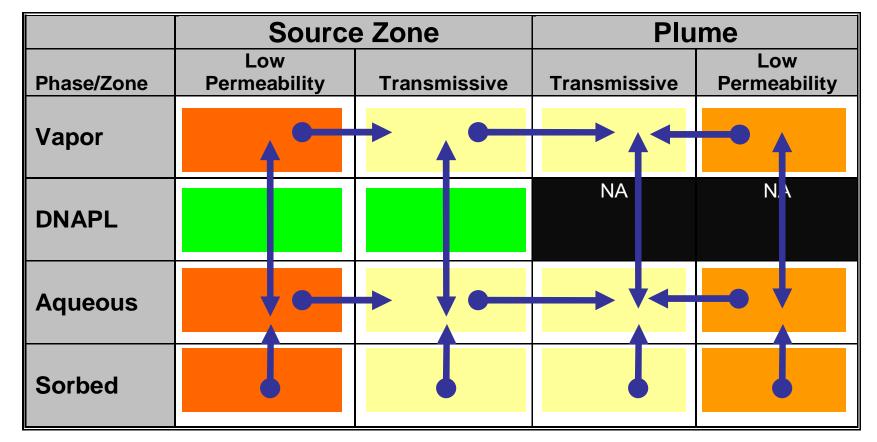






Late Stage

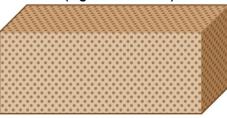




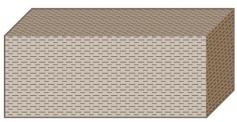


Type Setting

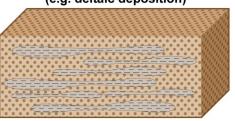
(I) Granular Media with Mild Heterogeneity and Moderate to High Permeability (e.g. eolian sands)



(II) Granular Media with Mild Heterogeneity and Low Permeability (e.g. lacustrine clay)



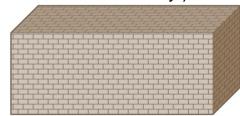
(III) Granular Media With Moderate to High Heterogeneity (e.g. deltaic deposition)



(IV) Fracture Media with Low Matrix Porosity (e.g. crystalline rock)

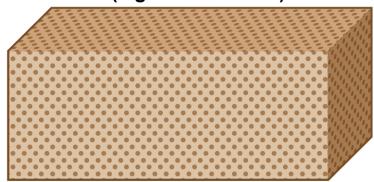


(V) Fracture Media with High Matrix Porosity (e.g. limestone, sandstone or fractured clays)



(I) Granular Media with Mild Heterogeneity and Moderate to High Permeability (e.g. eolian sands)





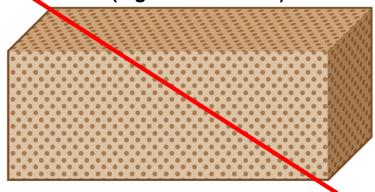




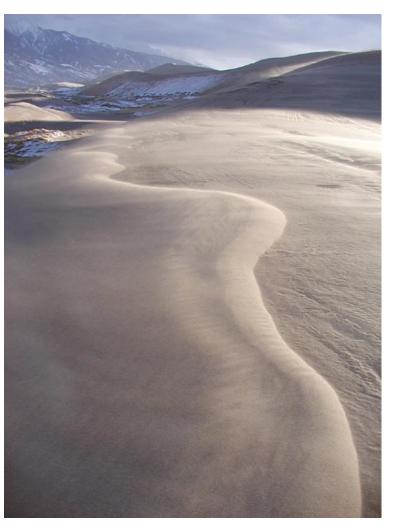
Great Sand Dunes National Park (Source http://www.nps.gov/grsa)

(I) Granular Media with Mild Heterogeneity and Moderate to High Permeability (e.g. eolian sands)





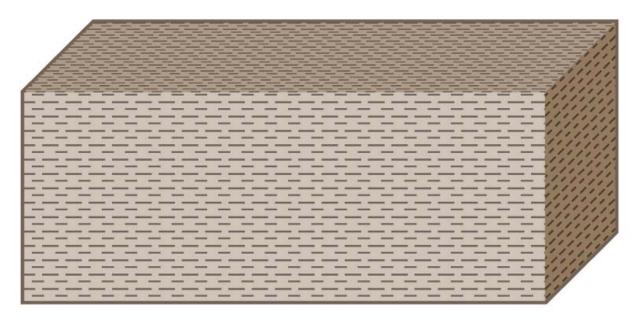




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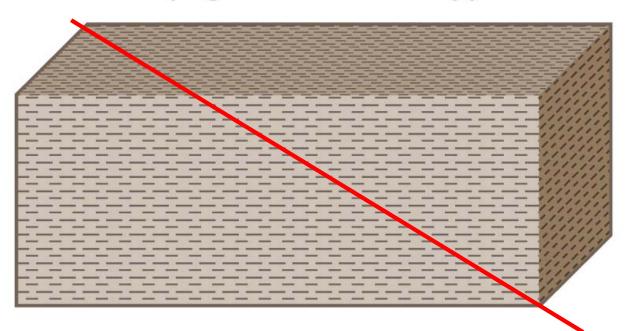


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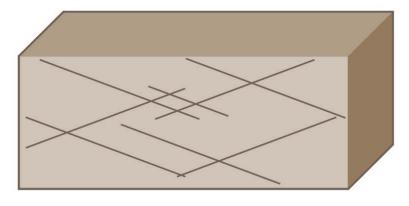
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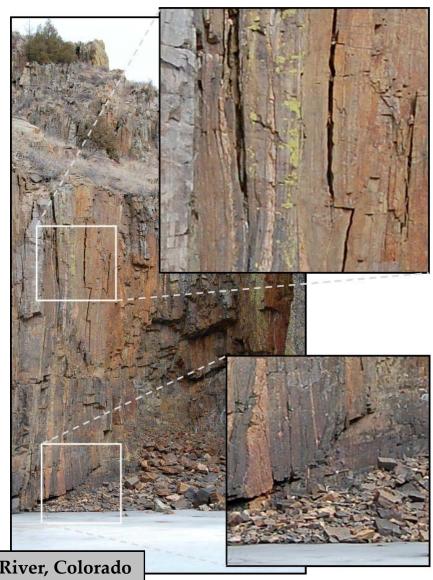






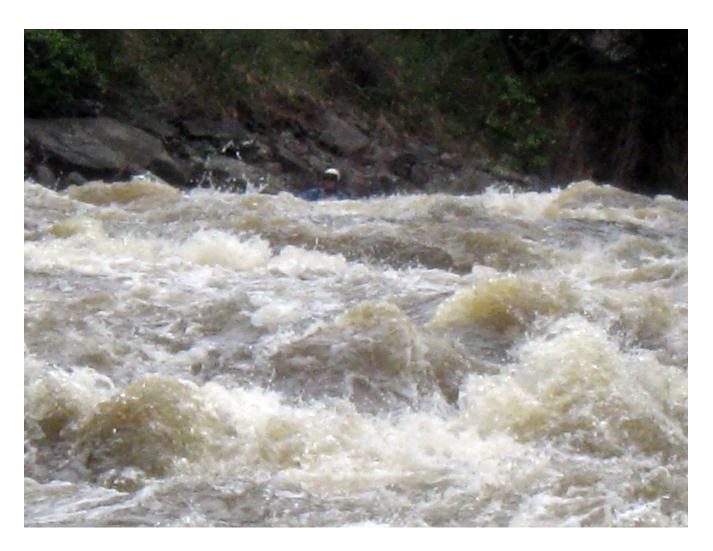
(IV) Fracture Media with Low Matrix Porosity (e.g. crystalline rock)





Cache La Poudre River, Colorado

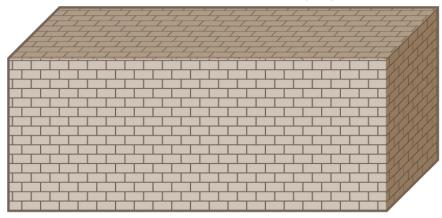




Cache La Poudre River, Colorado



(V) Fracture Media with High Matrix Porosity (e.g. limestone, sandstone or fractured clays)



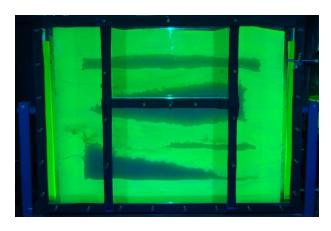


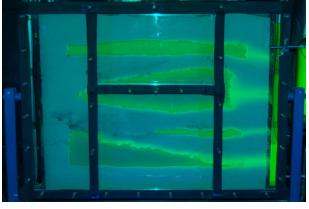
Bedding planes, joints, and vertical fractures in carbonate rock, Southern Ontario, Canada (Courtesy of Dr. Beth Parker)

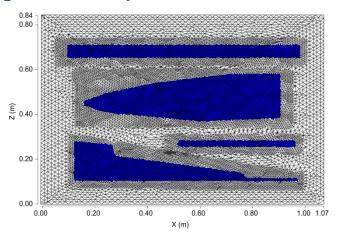


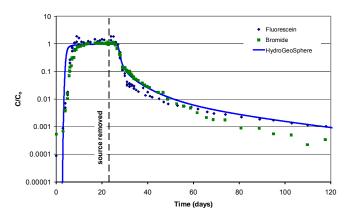
Modeling

With proper grid discretization and time-stepping constraints, standard finite element numerical models can be used to evaluate contaminants in low permeability zones.



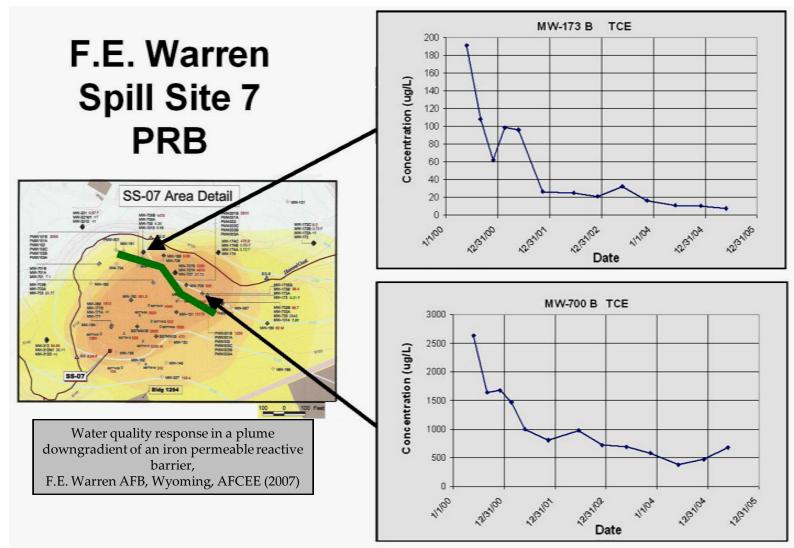


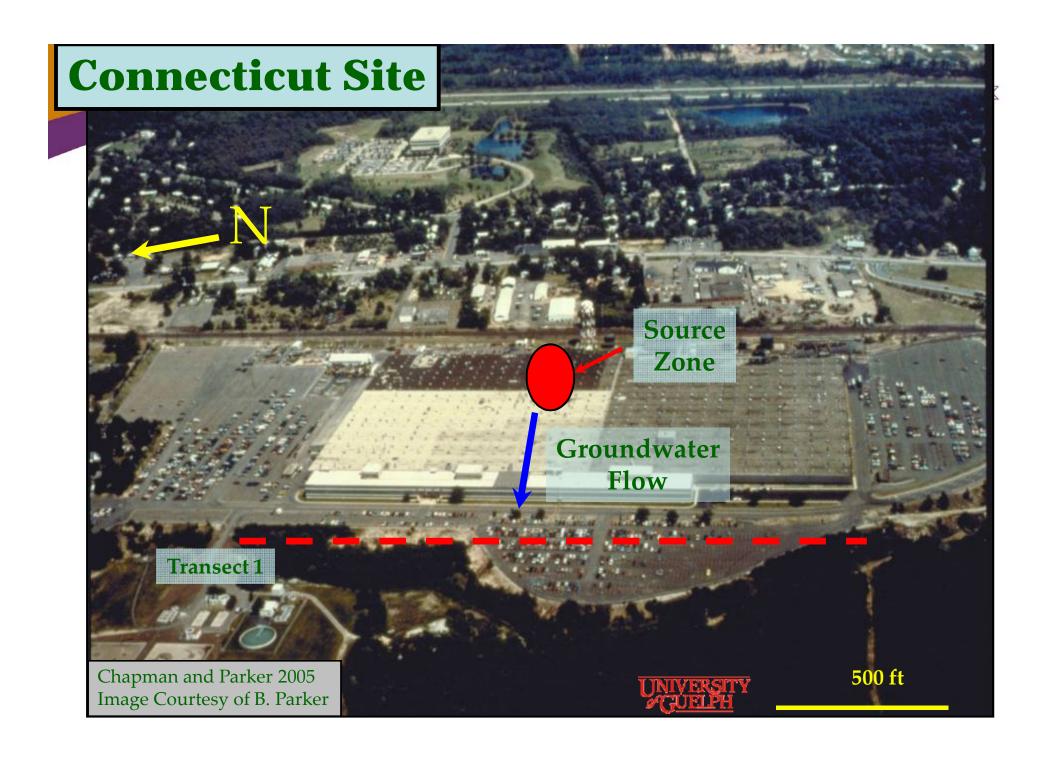






Field site showing impact of low-perm zones

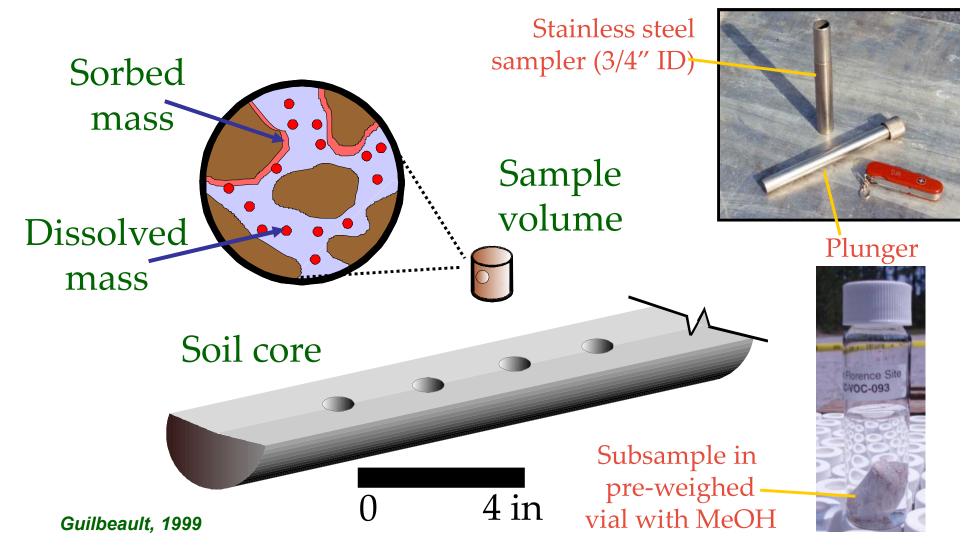


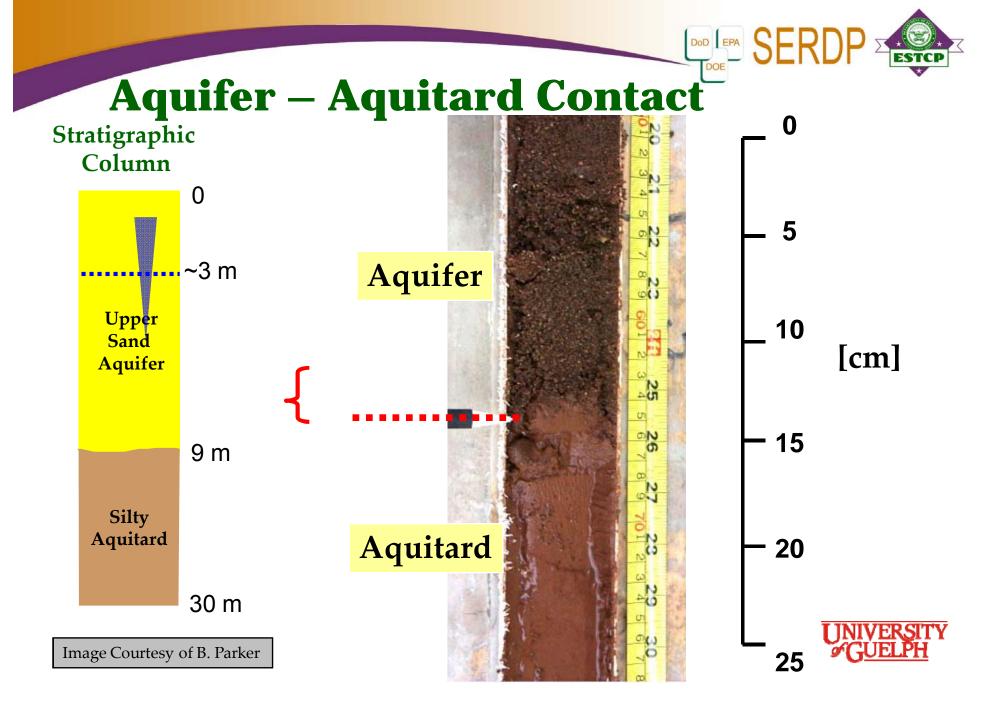






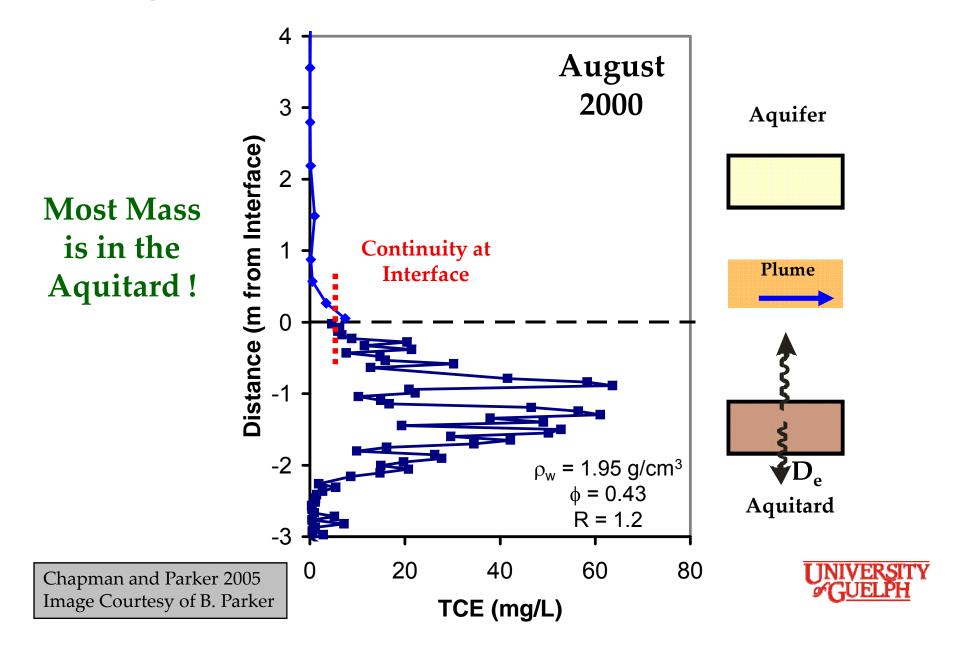
Technical Approach: Soil Core Subsampling (Task 2)



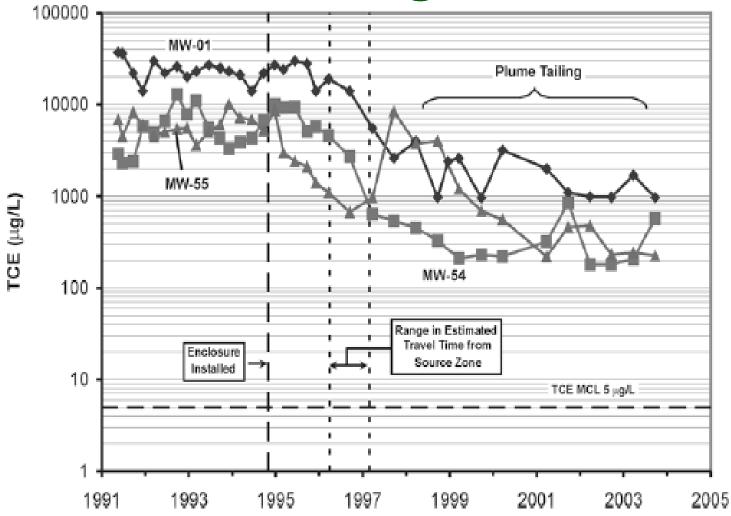


High-Resolution Data from Core





Concentration vs. Time from Monitoring Wells

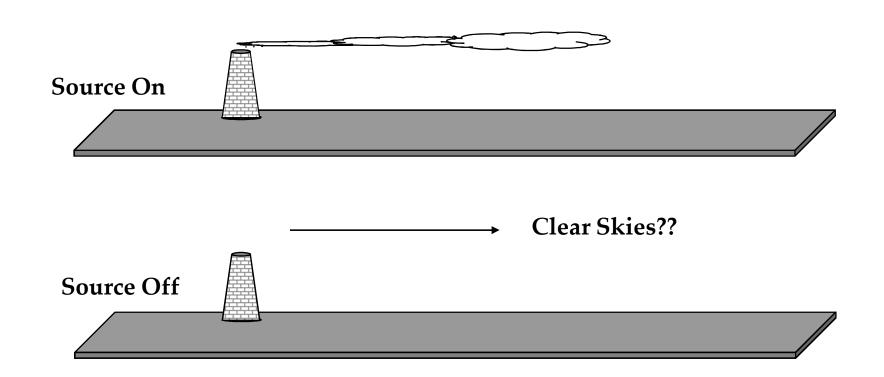


Source: Chapman and Parker, 2005 Copyright 2005 American Geophysical Union. Reproduced/modified by permission of AGU.

Key Concepts about L&D Plumes



What happens after the "source" is addressed?









Setting Objectives

Tom Sale, Chuck Newell

University Consortium for Field-Focused Groundwater Contamination Research University of Guelph, Ontario May 19-20, 2009



NRC (2005) observations regarding objectives

Failure to explicitly state remedial objectives appears to be a significant barrier ...

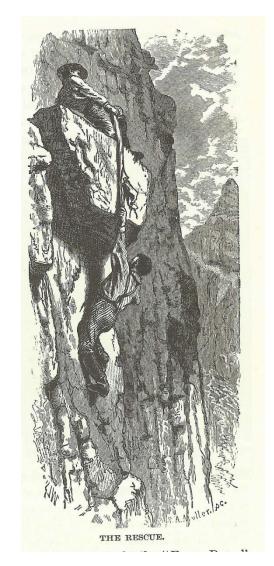
and

Vagueness of objectives for remedial projects can preclude effective decision making



Yoggi Berra

- "if you don't know where you are going you might end up someplace else"
- "if you don't know where you are going you might not get there"



John Wesley Powell, Exploration of the Colorado River and its Tributaries



Objectives (NRC 2005)

• Absolute - Broad

• Functional - Specific



Comments on goals

- Set by participating parties
- Reflects the values of the participants
- Site specific
- Different priorities for different participants
- Should not be dictated
- Should be SMART*
 - Specific
 - Measureable
 - Attainable
 - Relevant (or Realistic)
 - Timely

Should be BAV

- Beneficial
- Attainable
- Verifiable

^{*}Peter Drucker "The Practice of Management"



Shopping List - Absolute Objectives

- Protection of human health and the environment
- Conservation of natural resources
- Mitigate adverse community impacts
- Minimize the burden of past practices on future generations



Shopping List — Functional Objectives

- Risk
 - Human Health
 - Ecological receptors
 - Worker
- Extent
 - Limit expansion
 - Reduce footprint
- Reduce Longevity
 - Source
 - Plume
- Regulatory
 - Compliance
- Community
 - Beneficial land use
 - Avoidance of undue disruptions

- Economics
 - Practical costs
 - Limit economic interruptions
 - Sustain property value
- Sustainability
 - Net environmental benefit
 - Passive solutions
 - Effectiveness of combinations
- Resource Conservation
 - Limit future losses
 - Renovation of impacted resources
 - Protect habitat



	Protection of human health and the environment	Conservation of natural resources	Address adverse community impacts	Minimize the burden of past practices on future generations
Risk				
Prevent active adverse human exposure via groundwater or soil gas				
Prevent active ecological exposure via groundwater or soil gas				
Prevent adverse worker related exposures via soil, groundwater, and/or soil				
vapor				
Extent				
Prevent expansion of source zones and plumes				
Reduce the extent of source zones and plumes				
Longevity				
Reduce the period in which immobile contaminants in source zones will				
provide persistent releases to groundwater and/or soils gas.				
Reduce the period in which immobile contaminants in plume will provide				
persistent releases to groundwater and/or soils gas. Regulatory				
Comply with local, state, and federal regulations				
Community				
Address adverse (non-health) impacts to communities				
Land use				
Restore beneficial use of impacted lands				
Economic				
Select actions that have a practical near terms capital costs and minimal life				
cycle cost				
Avoid undue interruptions to communities, government, and industry activities				
Redress adverse impacts to property values				
Sustainability				
Select measures that have a net positive environmental benefit				
Progress to a state in which passive remedies will be sufficient to address				
residual impacts				
Enhance the effectiveness of complementary technologies				
Resource Conservation				
Limit future degradation of resources				
Restore impacted groundwater to standards needed for beneficial use				
Protect sensitive biological habitat				



The Perfect Remedy

Absolute Objectives

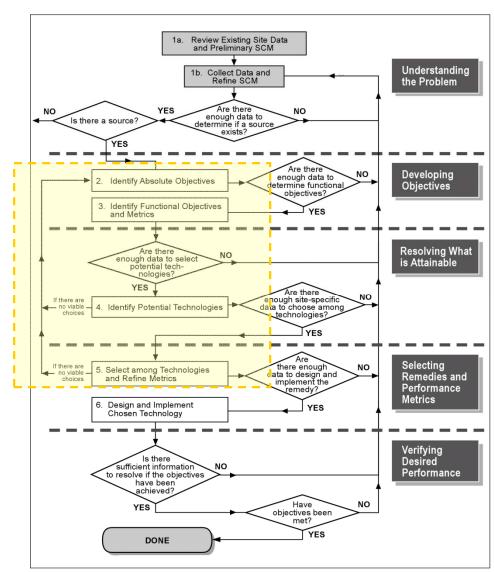
	Protection of human health and the environment	Conservation of natural resources	Address adverse community impacts	Minimize the burden of past practices on future generations
Risk				
Prevent active adverse human exposure via groundwater or soil gas				
Prevent active ecological exposure via groundwater or soil gas				
Prevent adverse worker related exposures via soil, groundwater, and/or soil vapor				
Extent				
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Reduce the extent of source zones and plumes				
Longevity				
Reduce the period in which immobile contaminants in source zones will				
provide persistent releases to groundwater and/or soils gas.				
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Restore impacted groundwater to standards needed for beneficial use				
Protect sensitive biological habitat				



Iterative Nature of Setting Goals

- Desired outcome
- Remedy Selection
- Prediction of outcome
- Comparison to goals

• ...



NRC 2005



Selecting technologies



What technologies do

- Treatment
 - Flux reduction
 - Longevity reduction
- Containment
 - Flux reduction



General classes of proven treatment technologies addressed include

- Physical Processes
- In Situ Chemical Oxidation
- In Situ Chemical Reduction
- In Situ Biological Reduction
- Thermal



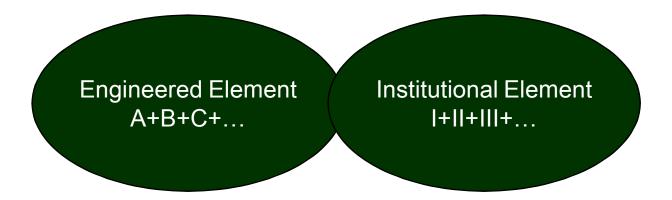


General classes of proven containment technologies addressed include

- Hydraulic Containment (Pump and Treat)
- Hydraulic barriers
- Coupled Hydraulic-Physical Containment
- In Situ Stabilization
- Permeable Reactive Barriers



Combined Remedies





OoM Rules of Thumb

- Well implemented in-situ remediation remedies are likely to reduce source zone groundwater concentrations by **about** *one order-of-magnitude* (90% reduction) from pre-treatment levels.
- One order-of-magnitude source reduction...
 - gives one order-of-magnitude improvement downgradient water quality.
- But with fast groundwater flow, low mass storage, and/or active attenuation...
 - potentially gives 2-3 orders-of-magnitude improvement downgradient over several years



Mapping technology performance using the 14 compartment model



Pump & Treat

	Source	e Zone	PI	um e		
Phase / Zone	Low Permeability	Transmissive	Transmissive	Low Permeability		
Vapor	Extraction of contamina	ated groundwater from tra vapor in the v		ly to have little effect on		
DNAPL	Depletion of aqueous	vapor in the vadose zone. DNAPL has the potential to be a long term source of aqueous phase Not Applicable				
Aqueous	phase from transmissive zones will cause slow release from low permeability zones	Pumping groundwater from the source zone will cause direct depletion of aqueous phase in transmissive zones	Pumping groundwater from the source zone will drive direct depletion of aqueous phase in transmissive zones			
Sorbed		transmissive zones will	ease of sorbed phase	zones in plumes		

How Does PUMP AND TREAT* Affect Contaminants in the 14 Different Compartments?

* (when used for treatment, not containment)

Key: Technology has this effect on contaminants in this compartment:

Direct depletion
Depletion but as a secondary effect
Limited secondary effect
Largely unaffected



Orders of Magnitude (OoM)

DEGREE OF CONTAMINATION								
Degree of Contamination	Level described by equivalent concentrations in water							
3 = Very High	1 – 10s (plus) mg/L in water							
2 = High	100 -1000 ug/L in water							
1 = Moderate	10-100 ug/L							
0 = Low	1-10 ug/L							

Anticipated Performance								
Description	Approximate Removal							
3 = Direct	> 90%							
2 = Secondary	90-10 %							
1 = Limited	< 10% -1%							
0 = Largely Unaffected	<1%							



Distribution of chlorinated solvents in a late stage Type IV setting

(Fractured rock with low matrix porosity)

	Source Z	ce Zone Plume			
Zone/Phase	Low Permeability	Transmissive	Transmissive	Low Permeability	
Vapor	0	0	0	0	
DNAPL	0	0			
Aquesous	0	1	1	0	
Sorbed	0	1	1	0	

Distribution of chlorinated solvents in a late stage Type 4 setting (Fractured Rock with Low Matrix Porosity)



Pump and treat in a late stage Type IV setting

	Source Zone				Source Zone Plume					me		
Zone/Phase	Low Permeal	v Permeability		Transmissive		sive	Low Permeab		. V			
Vapor	0	0	0	0	0	0		0	0			
DNAPL	1	0	1	0								
Aqueous	1	0	3	1	3	1		1	0			
Sorbed	1	0	2	1	2	1		1	0			

Maxium	12	Actual	10	Score	83



Screening pump and treat in a middle stage Type III setting.

Maximum 81

	Sour	ne		Plur	ne				
Zone/Phase	Low Permeability		hase Low Permeability Transmissive		ransmissive		sive	Low Permeability	
Vapor	0	2	0	2	0	1	0	1	
DNAPL	1	2	1	3					
Aqueous	1	2	3	3	3	2	1	1	
Sorbed	1	2	2	3	2	2	1	1	

Actual

36

Score

44



Outcome from pump and treat

	Sour	ne		Plur	me			
Zone/Phase	Low Permea	Low Permeability Transmissive		Transmiss	sive	Low Permeab	ility	
Vapor	0	2	0	2	0	1	0	1
DNAPL	1	2	1	3			-	
Aquesous	1	2	3	3	3	2	1	1
Sorbed	1	2	2	3	2	2	1	1

Maxium	81	Actual	36	Score	44
--------	----	--------	----	-------	----



	Sour		Plur	ne					
Zone/Phase	Low Permeability		Transmis	sive	Transmiss	sive	Low P	ermeabi	lity
Vapor	0	2	0	2	0	1		0	1
DNAPL	1	2	1	3			-		
Aquesous	1	2	(3	3) (3	2		1	1
Sorbed	1	2	2	<mark>ا</mark>	2	2		1	1



Source Excavation

	Source	e Zone	Plume			
Phase / Zone	Low Permeability Transmissive		Transmissive	Low Permeability		
Vapor			May reduce vadose zone vapor concentrations			
DNAPL		_	Not Applicable			
Aqueous	removed, and pro contamination shoul	entire source zones is perly backfilled no d remain in the source nes	Removal of the upgradient source should yield 1 to 3 order of magnitude improvements in downgradient water quality Depletion of the aqueous phase in transmissive zones will drive release of sorbed compounds. Note release of sorbed	1 1114		



Source excavation as a function of age

Source excavation in an early stage Type 3 setting.

	Source	ie	Plume						
Zone/Phase	Low Permeability		Transmissive		Transmissive		Low Permeability		
Vapor	3	1	3	2	1	0	1 0		
DNAPL	3	1	3	3					
Aqueous	3	1	3	3	2	1	1 0		
Sorbed	3	1	3	3	2	0	1 0		
	Maxium	48	Actual	47	Score	98			
			·						

Source excavation in a late stage Type 3 setting.

	Source Zone				Plume			
Zone/Phase	Low Permeability		Transmissive		Transmissive		Low Permeability	
Vapor	3	1	3	1	1	1	1	1
DNAPL	3	0	3	0				
Aquesous	3	2	3	1	2	1	1	2
Sorbed	3	2	3	1	2	1	1	2

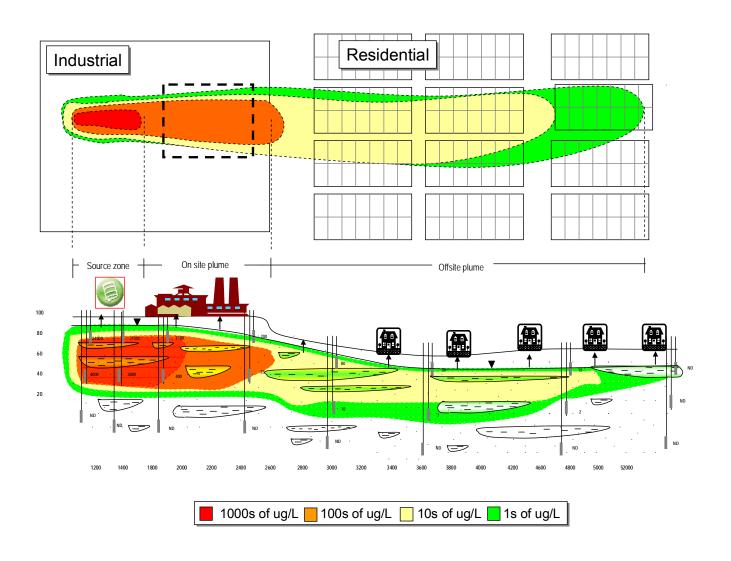
Maximum	48	Actual	34	Score	71	1
_						



Packaging Remedies



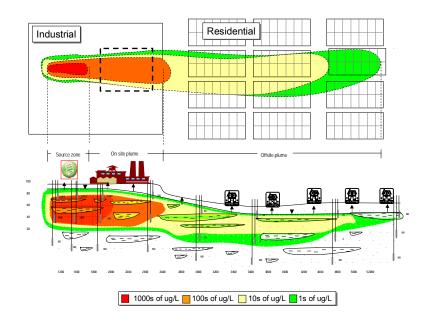
Example NO. 1





Site attributes

- 30 year old release of chlorinated solvent
- ~ 1 mile plume in a sandy aquifer
- 1000s of ug/L in the source area to 1s of ug/L at the end of the plume

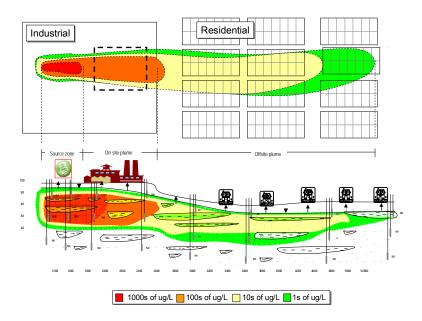


- No DNAPL observed in the source zone
- Stable plume with active degradation
- Lower permeability media (clays layers) are accumulating contaminant via inward diffusion
- Indoor air is a concern in the residential area



Drivers

- Home owners are concerned about health effects, property values, and disruptions in the neighborhood.
- Regionally, the community is committed to a clean environment while wanting to preserve jobs.

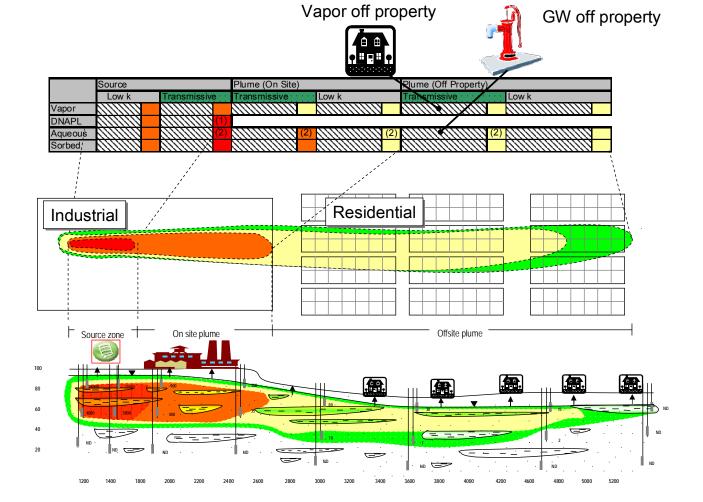


- Facility is committed to immediately addressing exposure pathways and meeting all other obligations with constraints of
 - a preference for actions with consequential benefits
 - economically feasibility
- Regulators support the interests of the community, provide technical support, and pursue compliance.



Before Treatment

- Setting
 - Middle stage
 - Type II
 - Cont. in low k zones
- With potential exposure via
 - Vapor
 - Groundwater
 - Onsite worker





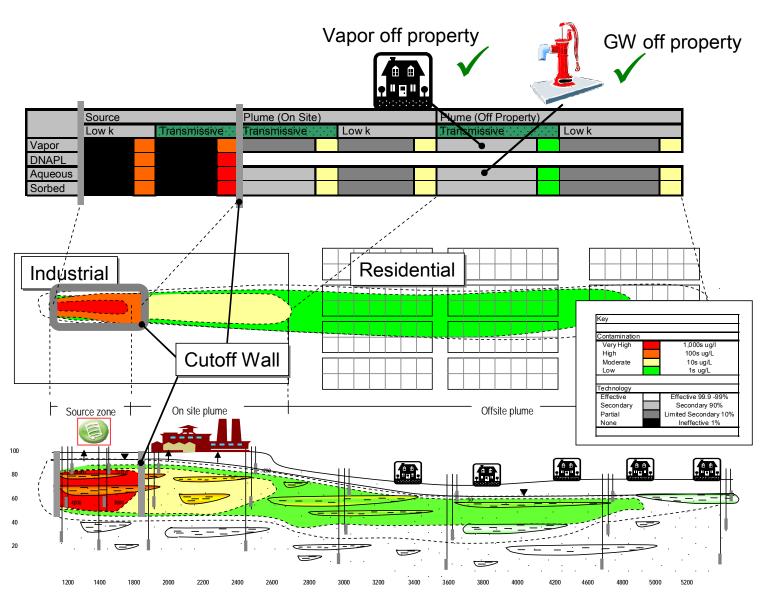
Consensus goals

Not Applicable

goale				
	Protection of human health and the environment	Conservation of natural resources	adverse community impacts	Minimize the burden of past practices on future generations
Risk				
Prevent active adverse human exposure via groundwater or soil gas				
Prevent active ecological exposure via groundwater or soil gas				
Prevent adverse worker related exposures via soil, groundwater, and/or soil vapor	•			
Extent				
Prevent expansion of source zones and plumes		,		
Reduce the extent of source zones and plumes				
Longevity				
Reduce the period in which immobile contaminants in source zones will provide persistent releases to groundwater and/or soils gas. Reduce the period in which immobile contaminants in plume will provide persistent releases to groundwater and/or soils gas.		Tob	e defined	1
Regulatory				
Comply with local, state, and federal regulations				
Community				
Address adverse (non-health) impacts to communities				
Land use				
Restore beneficial use of impacted lands				
Economic				
Select actions that have a practical near terms capital costs and minimal life cycle cost				
Avoid undue interruptions to communities, government, and industry activities				
Redress adverse impacts to property values				
Sustainability				
Select measures that have a net positive environmental benefit				
Progress to a state in which passive remedies will be sufficient to address residual impacts				
Enhance the effectiveness of complementary technologies				
Resource Conservation				
Limit future degradation of resources				
Restore impacted groundwater to standards needed for beneficial use				
Protect sensitive biological habitat				

Source containment with institutional controls for GW





Source Containment + GW Institutional Controls

good ok marginal no eff.

		good	OK IIIa	arginar no en
Current Conditions	Protection of human health and the environment	Conservation of natural resources	Address adverse community impacts	Minimize the burden of past practices on future generations
Risk				
Prevent active adverse human or ecological exposure via groundwater				
Prevent active adverse human or ecological exposure via soil gas				
Prevent adverse worker related exposures via soil, groundwater, and/or soil vapor				
Extent				
Prevent expansion of source zones and plumes				
Reduce the extent of source zones and plumes				
Longevity				
Reduce the period in which immobile contaminants in source zones will provide persistent releases to groundwater and/or soils gas. Reduce the period in which immobile contaminants in plume will provide				
persistent releases to groundwater and/or soils gas.				
Regulatory				
Comply with local, state, and federal regulations				
Community				
Address adverse (non-health) impacts to communities				
Land use				
Restore beneficial use of impacted lands				
Economic		···········		
Select actions that have a practical near terms capital costs and minimal life cycle cost				
Avoid undue interruptions to communities, government, and industry activities				
Redress adverse impacts to property values		3		
Sustainability				
Select measures that have a net positive environmental benefit				
Progress to a state in which passive remedies will be sufficient to address residual impacts				
Enhance the effectiveness of complementary technologies				
Resource Conservation				
Limit future degradation of resources				
Restore impacted groundwater to standards needed for beneficial use				
Protect sensitive biological habitat				



Other Examples

- Plume without natural attenuation
- Fracture rock without matrix porosity
- Fracture rock with matrix porosity



Closing



Key Points

- Holistic evaluation of all compartments
- The nature of the problem evolves with time
- Goals need to be SMART
- Single Technologies rarely address all compartments
- Many goals compete with each other
- Learning to value what is achievable and live with what remains



Discussion

- Value of compromise
 - Finding ways to go forward with what is beneficial, attainable, and verifiable
 - Learning to live with what will remain
- Alternatives to strict numerical standards
- Challenge of non-degradation policies
- Ways to break the log jam
- Time frames



Short Course Agenda

8:30 AM	Welcome and Introduction	Hans Stroo			
8:40 AM	Source Zone Protocol for Remedy Selection of Chlorinated Solvents Released at DoD Facilities	Thomas Sale & Charles Newell			
9:50 AM	Break				
10:10 AM	Development of a Protocol and Screening Tool for Selection of DNAPL Source Area Remediation	Carmen Lebrón, Bernard Kueper, David Major, Julie Konzuk & Jason Gerhard			
11:50 AM	Lunch				
1:00 PM	Decision & Management Tools for DNAPL Sites: Optimization of Chlorinated Solvent Source and Plume Remediation Considering Uncertainty	Ronald Falta & Charles Newell			
2:20 PM	Emulsion Design Tool for Planning Aqueous Amendment Injections Systems	Robert Borden			
2:50 PM	Break				
3:10 PM	Permanganate Design Tool for Planning Aqueous Amendment Injection Systems	Robert Borden			
4:00 PM	Improved Field Evaluation of NAPL Dissolution and Source Longevity	Michael Kavanaugh, Mark Widdowson, Lloyd Stewart & Rula Deeb			
5:20 PM	Summary & Conclusion	Hans Stroo			



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Development Of A Protocol And Screening Tool For Selection Of DNAPL Remedial Technologies ER-0424

Carmen A. Lebrón NAVFAC ESC

Dr. David Major Dr. Julie Konzuk

Geosyntec Consultants

Dr. Bernard KueperQueen's University



Dr. Jason Gerhard

University of Western Ontario





Seminar Outline:

Thursday, December 3, 2009				
Start	End	Topic		
10:25 AM	10:40 AM	Background, Objectives and Introduction to Screening Tool Development (Presented by Ms. Carmen A. Lebrón)		
10:40 AM	11:05 AM	Numerical Modeling: Simulations && Conclusions/Generalizations from Simulations (Presented by Dr. Bernard Kueper)		
11:05 AM	11:20 AM	Conclusions/Generalizations from Case Studies (Presented by Dr. Julie Konzuk)		
11:20 PM	11:30 AM	Screening Tool Demonstration (Presented by Dr. Julie Konzuk)		
11:30 PM	11:45 AM	Questions & Answers		



DNAPL Remediation Paradigm

- Uncertainties in DNAPL remediation technology selection:
 - How do different technologies perform in various geological/chemical environments?
 - What are reasonable expectations in terms of mass removal and concentration reductions?
 - What technology best meets our goals/needs?





Project Objectives

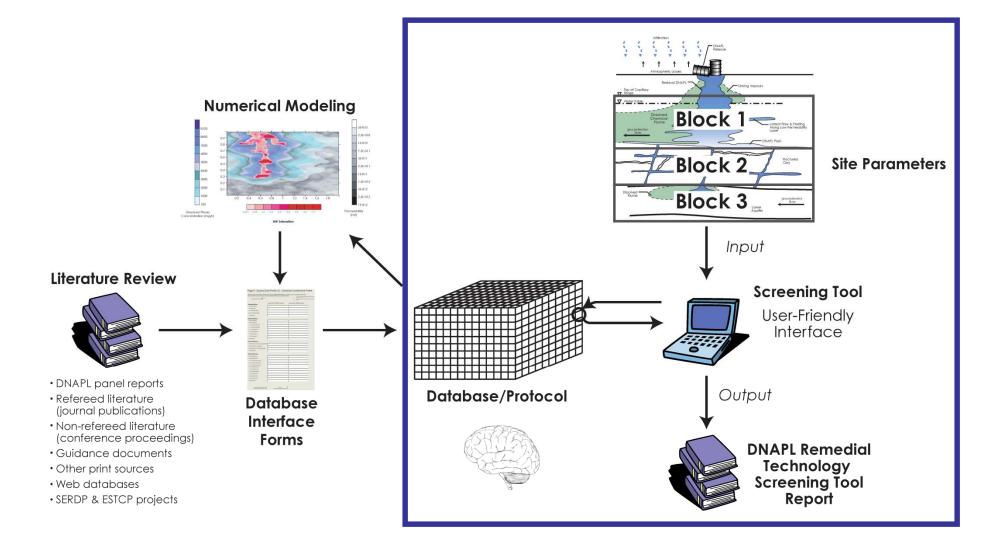
- Develop a screening tool that can be applied at DNAPL source zone sites to:
 - Reduce uncertainty in estimating remedial outcomes
 - Evaluate potential technology performance
 - Aid RPMs in technology selection based on desired performance metrics
- Screening tool developed using a modular approach, which allows for:
 - Incorporating other features in the future
 - Periodic updates of information in the screening tool database without reprogramming the screening tool



Expectation Management Tool

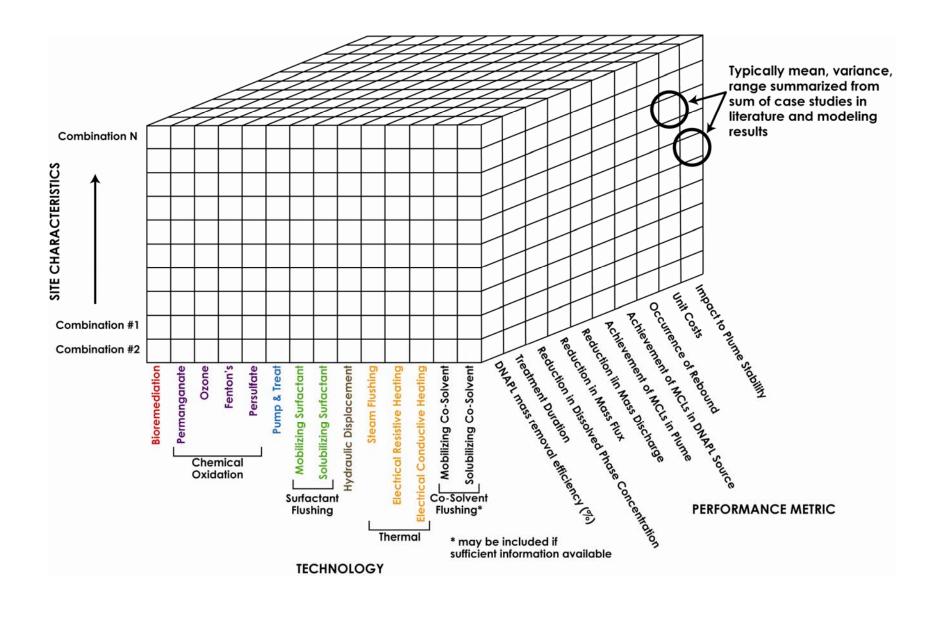


Technical Approach





The Matrix, a.k.a. Database

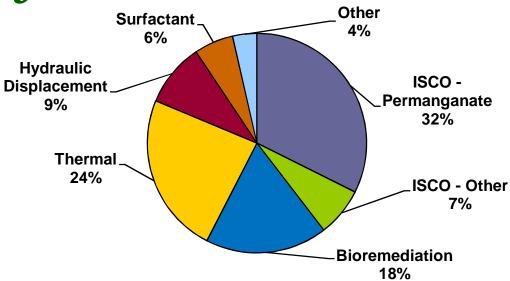


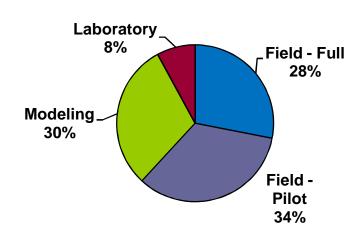


Case Study Collection

- Case studies entered into database to date:
 - 42 modeling case studies
 - 11 lab studies
 - 86 field case studies

 An additional 76 field case studies identified and >70 modeling case studies still to be entered into database

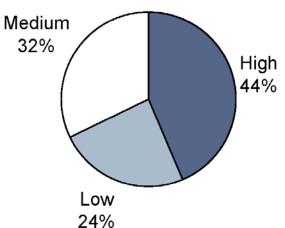






Case Study Quality Control

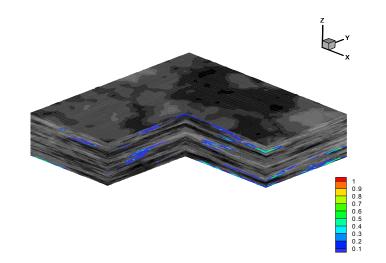
- Data Quality Rankings (DQRs) developed for each case study
 - Value between 1 (low) and 3 (high)
- Weighted average of ratings for:
 - Information source (low weighting)
 - Age of study (medium)
 - Methods used to characterize DNAPL (medium)
 - Completeness of pre-treatment data set (high)
 - Completeness of post-treatment data set (high)
- In screening tool, users can filter data based on DQRs





Modeling/Simulations

- ✓ Why Modeling?
- ✓ Allows us to:
- Simulate DNAPL releases in various geologic settings and create different architectures
- Compare technology performance
- Evaluate impact of various factors on tech performance
- Assess source removal long-term impacts on groundwater quality







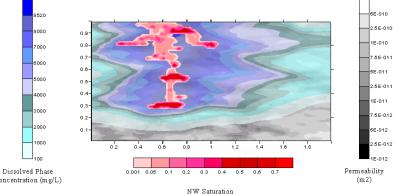
Modeling/Simulations

Step 1. Creating template sites

• Simulate a range of geological, hydrogeological,

and chemical environments

Simulate a range of DNAPL releases and architectures



Step 2. Modeling DNAPL Treatment

- Simulate treatment with selected technologies
- Metrics evaluated include DNAPL mass reduction, source zone concentration reduction, mass flux reduction, plume length



Simulations/Template Sites

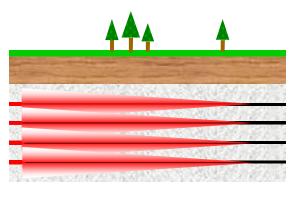
Porous Media Template Site Parameters

Template Site	DNAPL Type	DNAPL Release Volume	Hydraulic Conductivity	Soil Heterogeneity
Low Heterogeneity	TCE	7.57 m^3	10 ⁻³ cm/s	In k = 1
Low K	TCE	7.57 m ³	10 ⁻⁴ cm/s	In k = 2
Low DNAPL Volume	TCE	1.89 m ³	10 ⁻³ cm/s	In k = 2
Lower Density DNAPL	1,1,1-TCA	7.57 m ³	10 ⁻³ cm/s	In k = 2
Base Case	TCE	7.57 m ³	10 ⁻³ cm/s	In k = 2
Higher Density DNAPL	PCE	7.57 m ³	10 ⁻³ cm/s	In k = 2
High DNAPL Volume	TCE	18.9 m ³	10 ⁻³ cm/s	In k = 2
High K	TCE	7.57 m ³	10 ⁻² cm/s	In k = 2
High Heterogeneity	TCE	7.57 m ³	10 ⁻³ cm/s	In k = 4



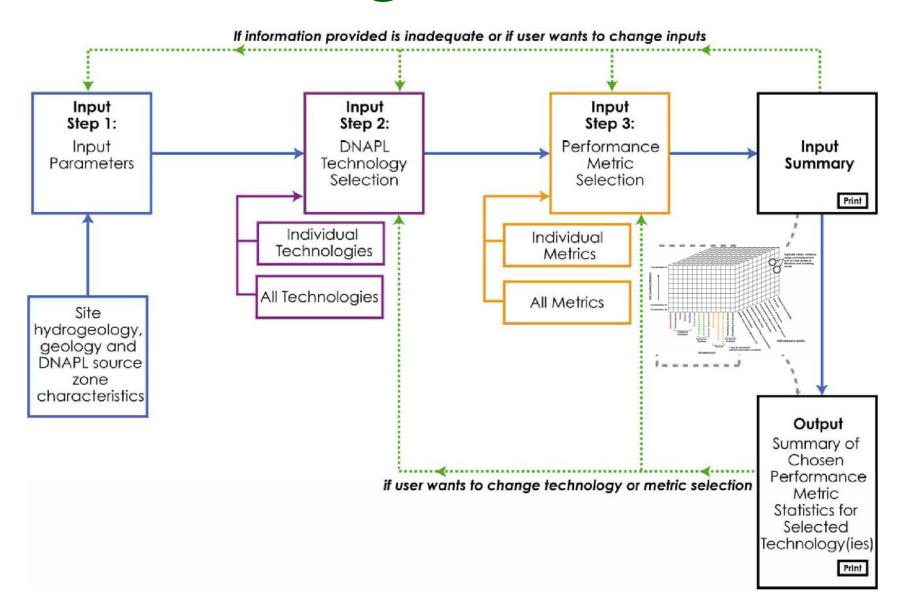
Simulations/Template Sites Fractured Clay Template Site Parameters DNAPL Fractured DNAPL

Template Site	DNAPL Type	Fracture Aperture	Matrix Porosity	Fraction Organic Carbon	Fracture Spacing
Low Organic Carbon	TCE	75 μm	30%	0.0015	1.0 m
Low Matrix Porosity	TCE	75 μm	15%	0.003	1.0 m
Low Fracture Aperture	TCE	37.5 μm	30%	0.003	1.0 m
Low Density DNAPL	1,1,1- TCA	-	-	-	-
Base Case	TCE	75 µm	30%	0.003	1.0 m
High Density DNAPL	PCE	75 µm	30%	0.003	1.0 m
High Fracture Aperture	TCE	150 µm	30%	0.003	1.0 m
High Matrix Porosity	TCE	75 μm	45%	0.003	1.0 m
High Organic Carbon	TCE	75 μm	30%	0.006	1.0 m





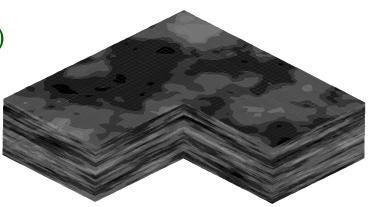
Screening Tool Structure

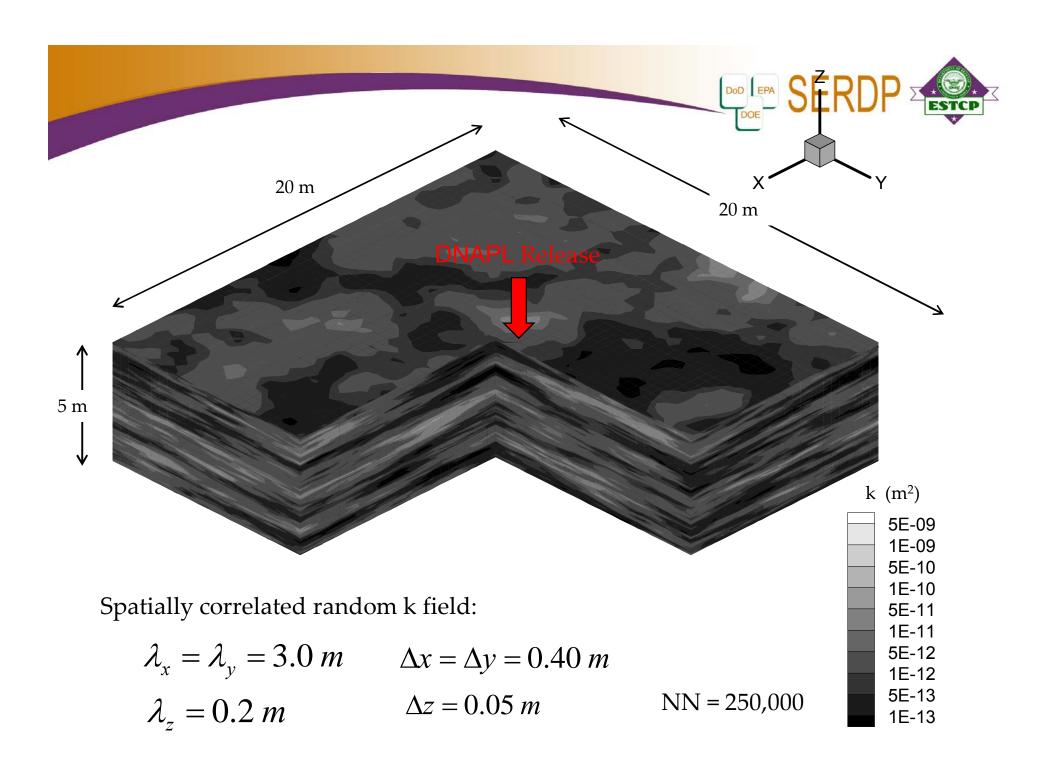




Numerical Modeling

- Numerical simulation of DNAPL source zone remediation in porous and fractured media
- Technologies considered:
 - Hydraulic Displacement (PM only)
 - Pump-and-Treat
 - In Situ Chemical Oxidation
 - Enhanced In Situ Bioremediation
 - Surfactant Flushing
- Technologies applied to 'Template Sites'

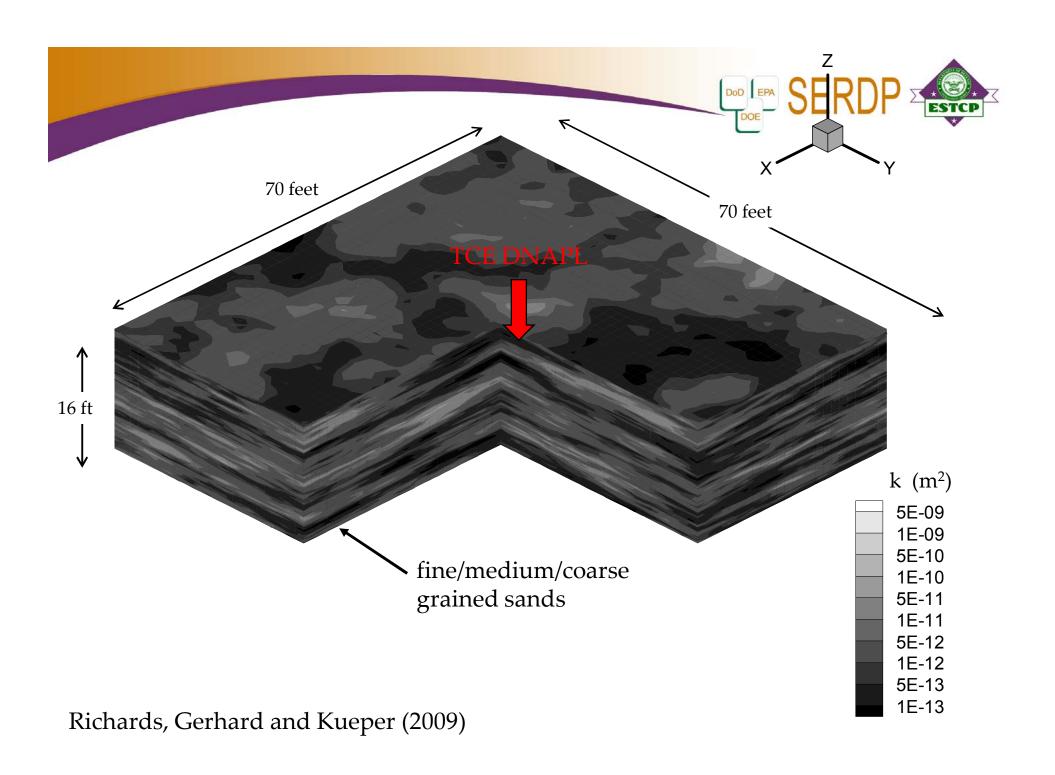


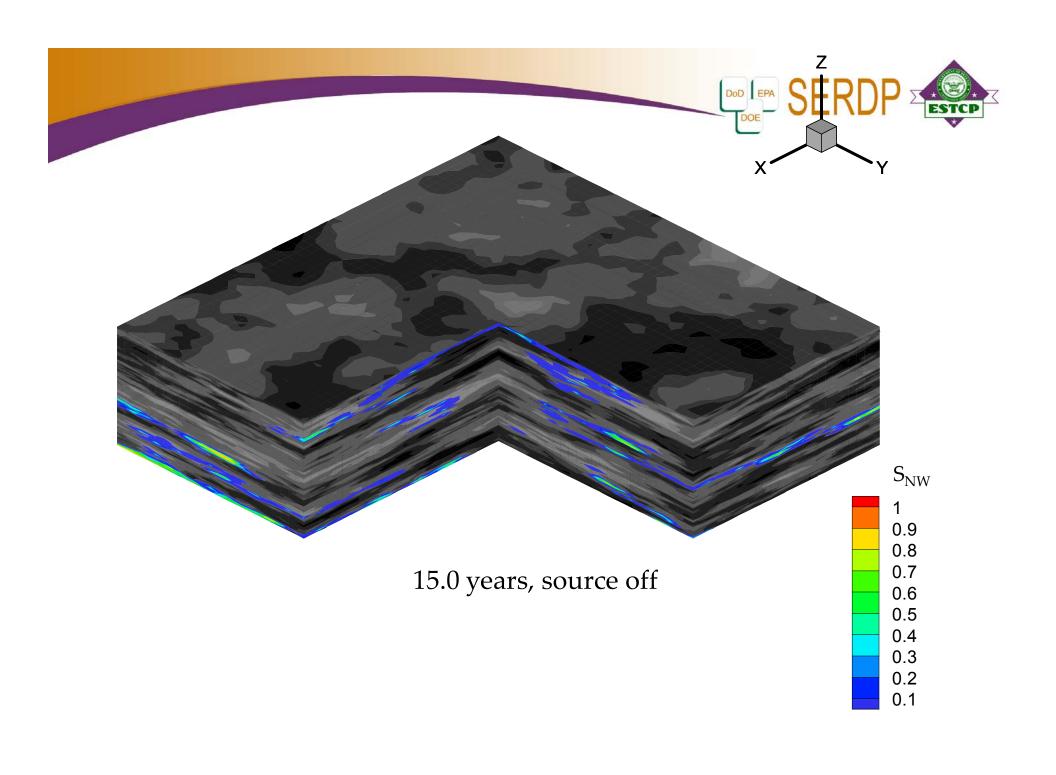




Porous Media Template Sites

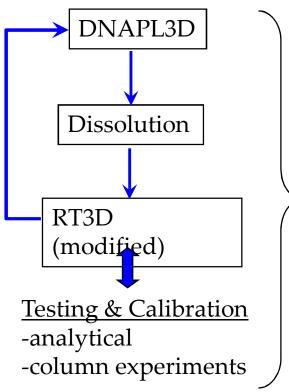
Template Site	DNAPL	Initial Mass (kg)	Initial Volume (m³)	Mean <i>k</i> (m²)	Variance In <i>k</i>
Base case	TCE	3520	2.41	3.03 10 ⁻¹²	1.74
High mean k	TCE	3496	2.39	3.02 10 ⁻¹¹	1.74
Low mean <i>k</i>	TCE	3535	2.42	3.04 10 ⁻¹³	1.74
Low heterogeneity	TCE	3355	2.30	1.87 10 ⁻¹²	0.87
High heterogeneity	TCE	3186	2.18	7.41 10 ⁻¹²	3.48
Small DNAPL volume (post HD)	TCE	785	0.54	3.03 10 ⁻¹²	1.74
Small DNAPL volume (pre HD)	TCE	803	0.55	3.03 10 ⁻¹²	1.74
Large DNAPL volume	TCE	7343	5.03	3.03 10 ⁻¹²	1.74
High density DNAPL	PCE	3871	2.37	3.03 10 ⁻¹²	1.74







Remediation Model Development & Application



DNAPL3D-RX (remediation model)

System Specific Characteristics

- -reaction kinetics
- -species and initial conditions
- -clogging effects



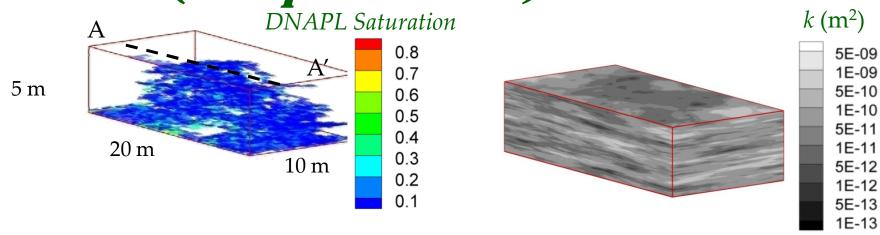
Field Scale Applications

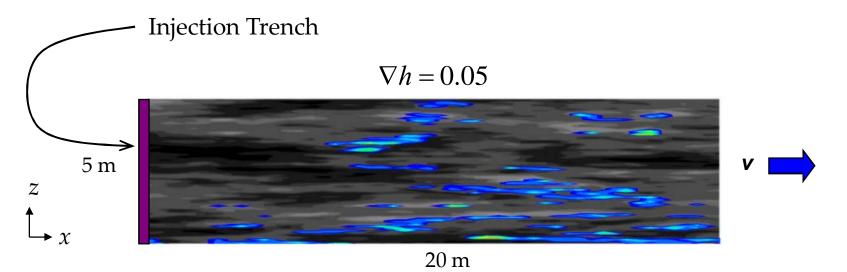
- 9 Template Sites (geology, TCE/PCE/111-TCA, DNAPL volume)
- 5 technologies (HD, P&T, ISCO, EISB, SEAR)





Base Case Modeling Domain (Template Site 1)





Cross-Section A-A'



In-Situ Chemical Oxidation (ISCO) with Potassium Permanganate

- Stoichiometry, kinetics and rate constants from literature
- 2nd order reactions for TCE/PCE and OAM with MnO₄-
- KMnO₄ injected at 2,500 mg/L
- Species specific diffusion coefficients (TCE and MnO₄⁻)
- OAM cross-correlated with *k* (negative)
- Pore clogging due to rind formation (West et al., 2008, AWR)
- Perfectly buffered system assumed
- Local equilibrium dissolution of DNAPL

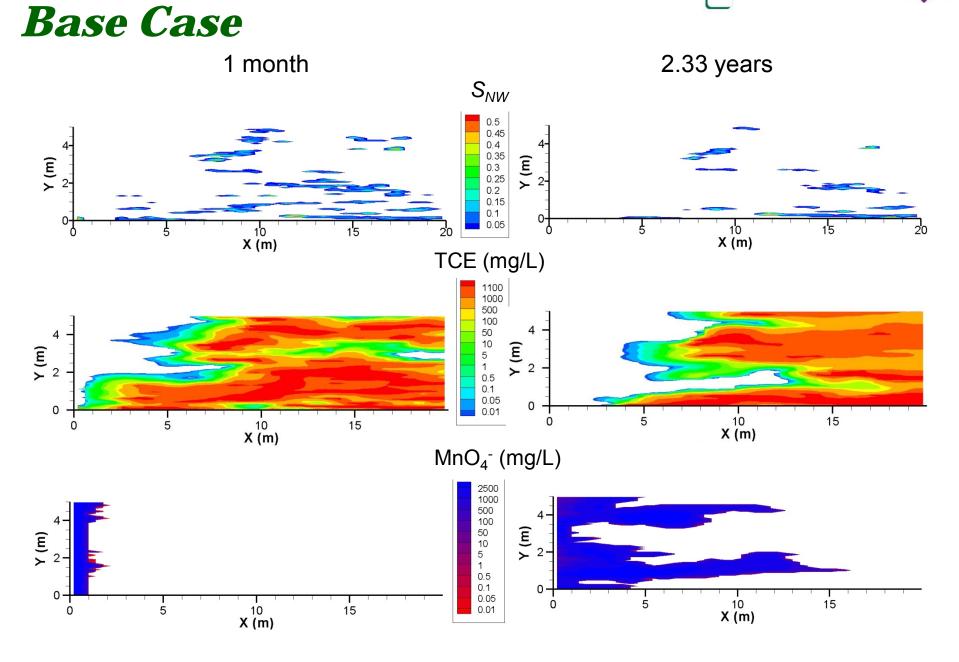


ISCO Simulations

Simulatio n	Description	Injection Duration (days)	$KMnO_4$ Breakthrough at Exit Face?
1	Base case	849	No
2	High mean k	83	No
3	Low mean <i>k</i>	3650	No
4	Low heterogeneity	1086	No
5	High heterogeneity	575	No
6a	Small DNAPL volume (post HD)	163	No
6b	Small DNAPL volume (pre HD)	166	No
7	Large DNAPL volume	2251	Yes
8	PCE DNAPL	724	No
9	Base case, no Rind	849	No
10	Base case, no NOD	849	Yes
11	Base case, no NOD & no Rind	849	Yes

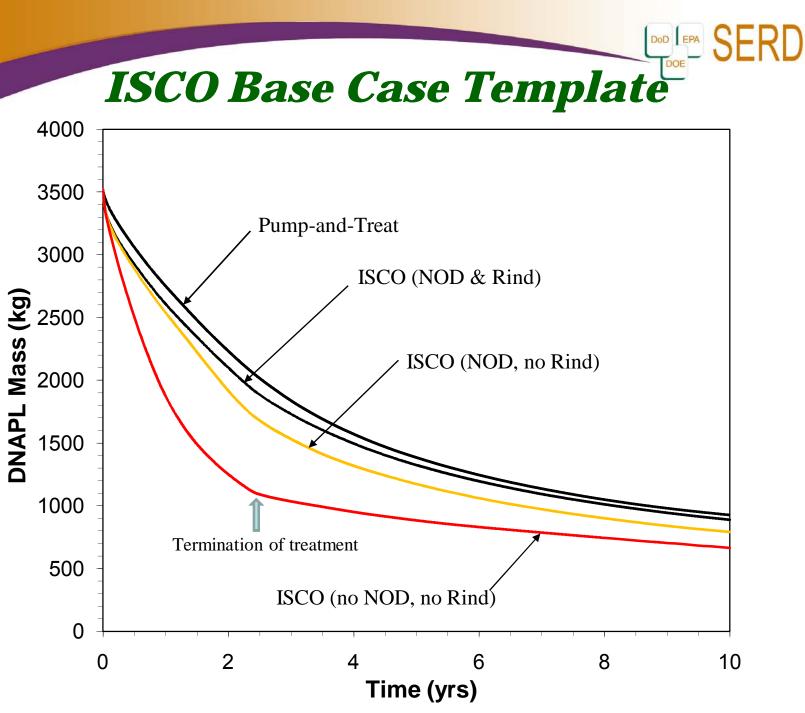
Comparison of ISCO Output - SERDP





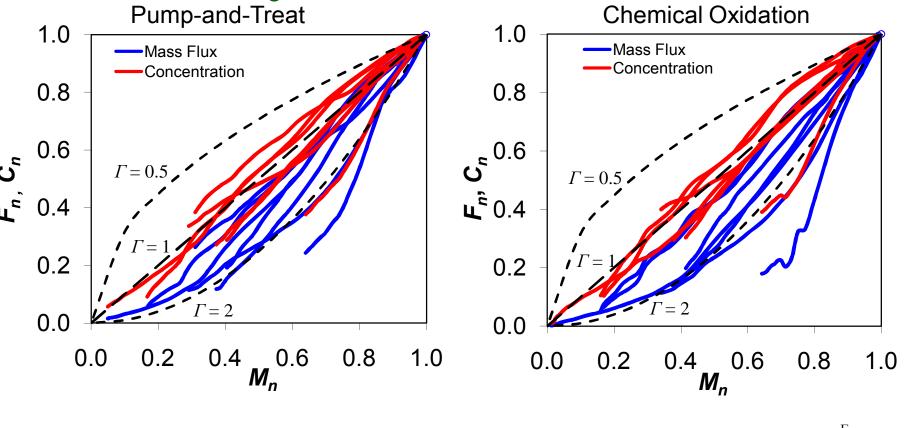








Pump-and-Treat vs Chemical Oxidation (Boundary Flux and Concentration)



$$C_n = \frac{C(t)}{C(t_0)} \qquad F_n = \frac{M_f(t)}{M_f(t_0)} \qquad M_n = \left(\frac{M_{DNAPL}(t)}{M_{DNAPL}(t_0)}\right) \qquad C_n = \frac{C(t)}{C(t_0)} = \left(\frac{M_{DNAPL}(t)}{M_{DNAPL}(t_0)}\right)^{\Gamma}$$



Enhanced In-Situ Bioremediation (EISB)

- Stochiometry, kinetics and rate constants from literature
- TCE (or PCE) degrades to cis-DCE
- Monod-type kinetics
- First-order decay of biomass
- Lactate injected @ 1 day/week for 2.5 years
- 3 biologic species: fermentors, dechlorinators, & methanogens (competitors)
- All microbes initially uniformly distributed
- Lactate converted to H₂ by fermentors
- H₂ consumed by both dechlorinators & methanogens
- Bioclogging due to dechlorinator & methanogen biomass

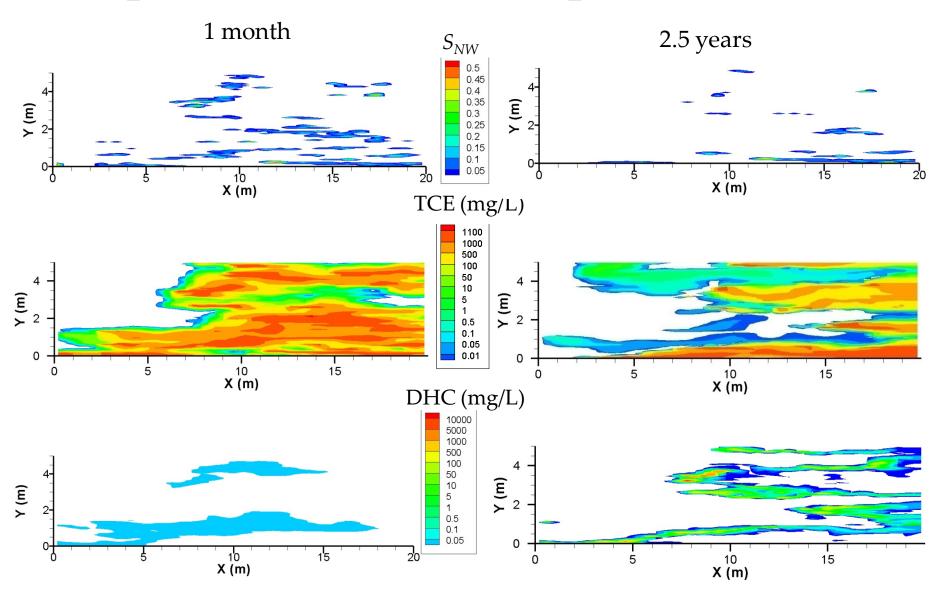


EISB Simulations

Simulation	Description	Lactate injection concentration (mg/L)		
1	Base case	39130		
2	High mean <i>k</i>	39130		
3	Low mean <i>k</i>	39130		
4	Low heterogeneity	39130		
5	High heterogeneity	39130		
6a	Small DNAPL volume (post HD)	39130		
6b	Small DNAPL volume (pre HD)	39130		
7	High DNAPL volume	39130		
8	PCE DNAPL	39130 or 7511		
BC1	Base case, no bioclogging	39130		
BC2 Base case, no competition		39130		
BC3	Base case, no bioclogging & no competition	39130		
PS1	Base case, 1 hour/day lactate pulse	134160		
PS2	Base case, 1 week/month lactate pulse	24113		

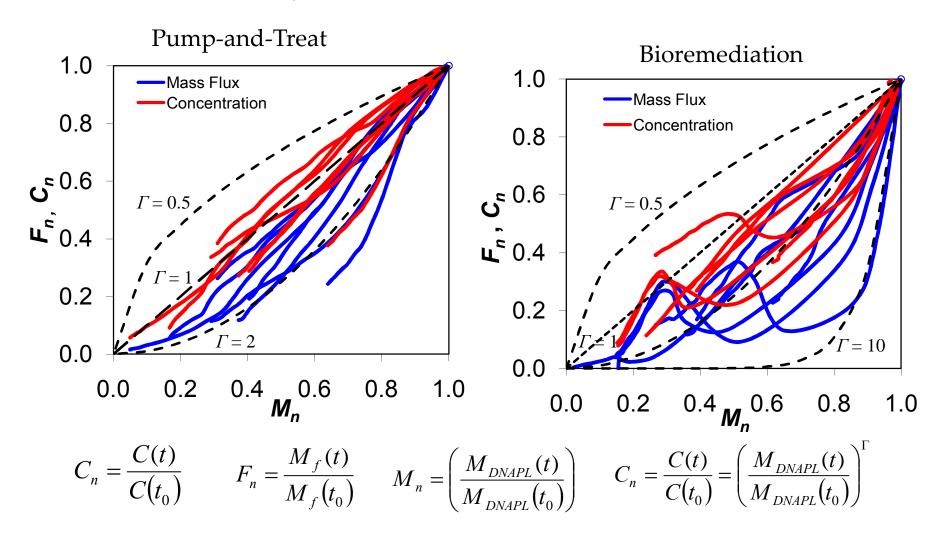


Comparison of EISB Output — Base Case





Pump-and-Treat v. Bioremediation (boundary flux and concentration)





Surfactant Enhanced Aquifer Remediation (SEAR)

- Dissolution kinetics and rate constants from literature
- Tween 80 injected at 40,000 mg/L for 22 days (base case)
- 3 species: TCE/PCE solute, Tween 80 micelles, and pseudo microemulsion
- Enhanced dissolution by linear driving function
- Interfacial tension reduction not simulated
- Model tested against published column experiments

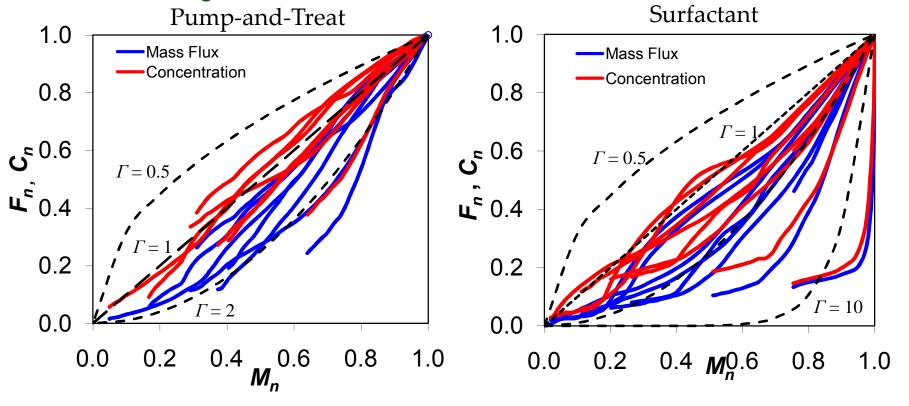


SEAR Simulations

Simulation	Template site	SEAR injection time (days)
1	Base case (TCE)	22
2	High mean k	2
3	Low mean k	223
4	Low heterogeneity	29
5	High heterogeneity	11
6a	Small DNAPL volume (post-HD)	5
6b	Small DNAPL volume (pre-HD)	5
7	Large DNAPL volume	48
8	PCE DNAPL	35



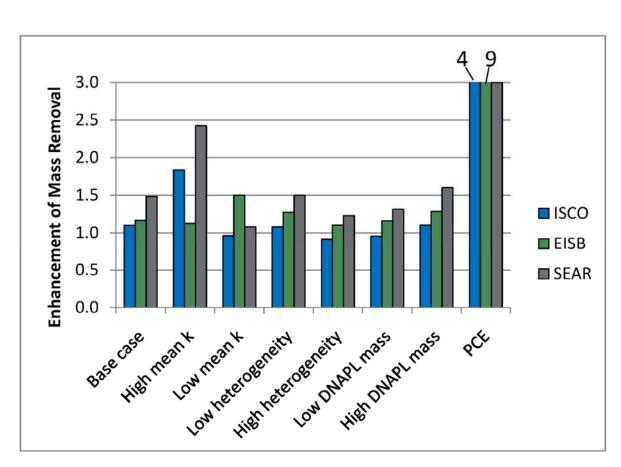
Pump-and-Treat vs SEAR (boundary flux and concentration)



$$C_n = \frac{C(t)}{C(t_0)} \qquad F_n = \frac{M_f(t)}{M_f(t_0)} \qquad M_n = \left(\frac{M_{DNAPL}(t)}{M_{DNAPL}(t_0)}\right) \qquad C_n = \frac{C(t)}{C(t_0)} = \left(\frac{M_{DNAPL}(t)}{M_{DNAPL}(t_0)}\right)^{\Gamma}$$

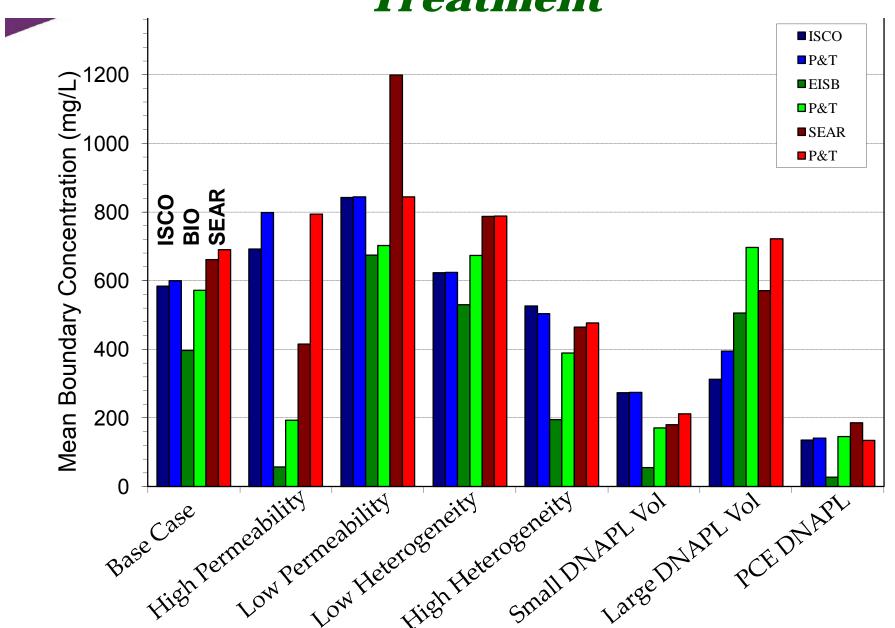


Comparative Enhancement of DNAPL Mass Removal at End of Treatment



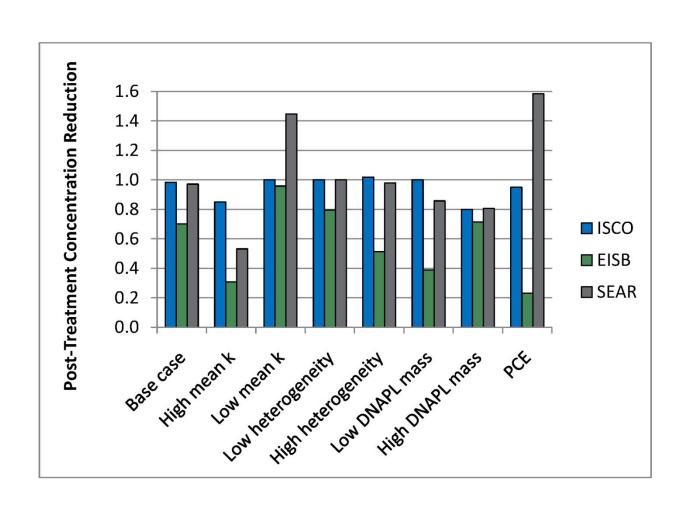
- Enhancement of mass removal greatest where dissolution is not a significant mass removal factor:
 - Lower solubility DNAPLs (PCE)
- ISCO (MnO4) only technology where incomplete treatment reduces mass removal efficiency below doing P&T

Mean Boundary Concentration at End of Treatment

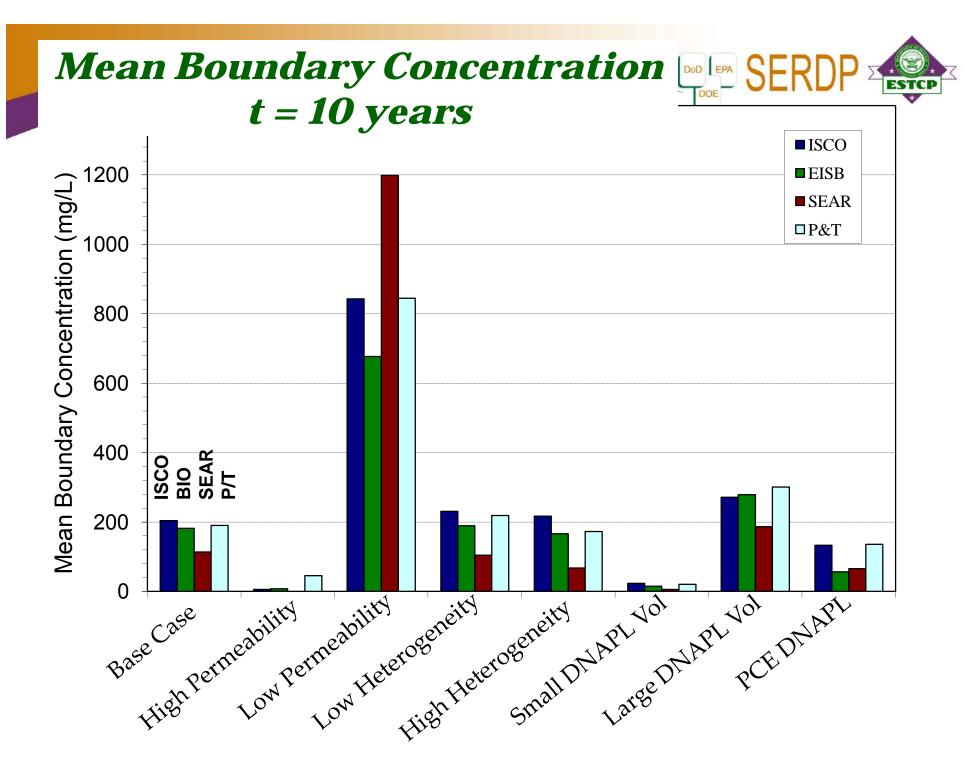




Concentration Reduction Enhancement Normalized to P&T — at End of Treatment

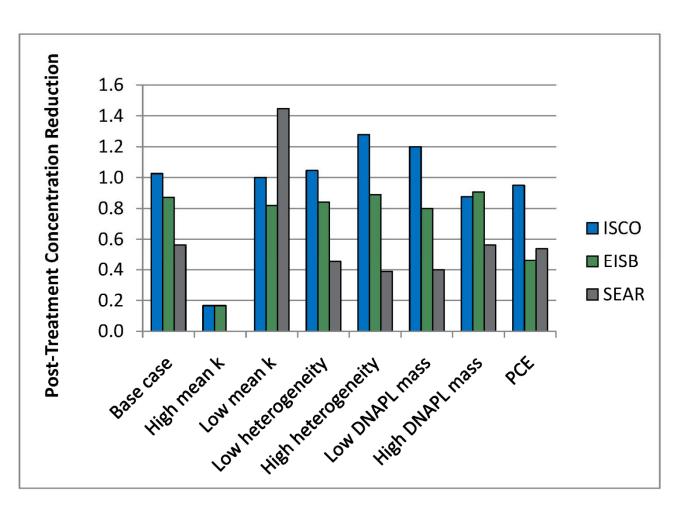


- Reduction factor >1 =
 concentrations higher
 than seen for
 dissolution only
- EISB has greatest enhancement in concentration reductions at treatment termination
- ISCO reductions were minimal





Concentration Reduction Enhancement Beyond P&T – After 10 Years



- SEAR enhancement of concentration reduction continues to improve, except for low permeability soils
- EISB concentrations still lower than P&T for all, but enhancement in reduction is reduced
- ISCO enhancement still minimal, and worsened in some cases



Conclusions – Porous Media Modeling

- Technology performance (DNAPL mass, flux and concentration reduction) is site specific (geology, DNAPL volume)
- Flux decreases faster than concentration
- Low permeability generally not conducive to injection technologies
- Important to arrive at accurate estimate of DNAPL mass to optimize design



Conclusions — Porous Media Modeling

- P/T & ISCO (MnO4) typically lead to near-linear reduction in concentration with mass removal, while EISB and SEAR have greater proportion of concentration reduction with DNAPL mass removal
- Partial mass removal will not achieve MCLs in groundwater concentrations
- Demand from natural organic matter can result in significantly more oxidant demand compared to stoichiometric DNAPL mass requirements
 - Cost issue





Case Study Trend Analysis

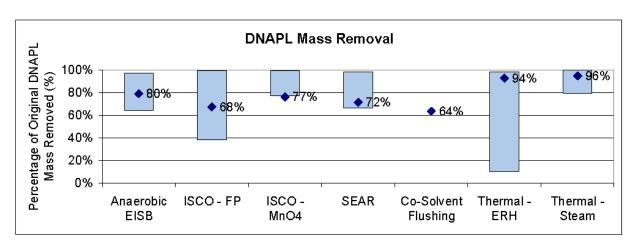
- Linear and non-linear multivariate regression used to evaluate influence of site parameters on technology performance:
 - Correlations between site parameters and performance metrics
 - Determine the 'key' site & technology parameters correlated to performance

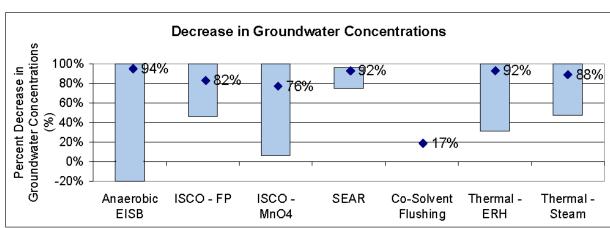
Performance Metric	Treatment Area	Saturated Thickness	Soil Heterogeneity	Pooled DNAPL	Electrode Spacing
Decrease in Groundwater Concentrations	Equally good performance in nearly all case studies			udies	
Decrease in Soil Concentrations					
Removal of DNAPL Mass					
Treatment Duration (ERH)		++			++++
Treatment Duration (Steam)	+++		++++		
Rebound of Groundwater Concentrations	ter Equally good performance in all case s		case studie	es	
Unit Cost (\$/m³)		+++		++	
Achievement of MCLS	No apparent influence from site parameters			s	

- indicates weakest negative correlation, ---- indicates strongest negative correlation
- + indicates weakest positive correlation, ++++ indicates strongest positive correlation



Technology Performance Comparison

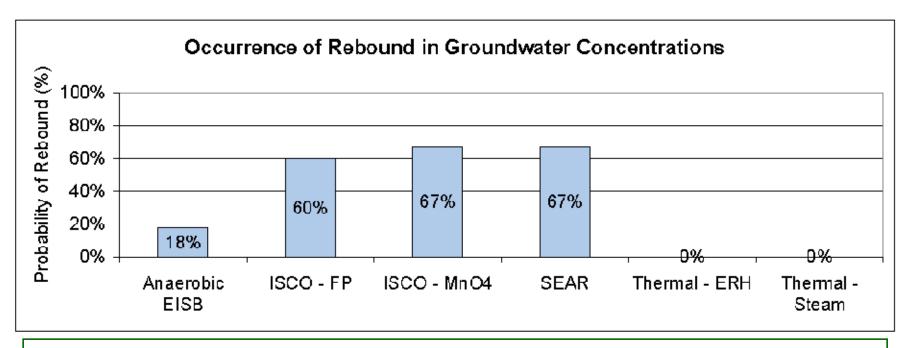




- Thermal typically better at removing DNAPL mass, other technologies more likely to have partial mass removal
- Temporary increases in flux/ concentrations may be seen during EISB due to increased dissolution of the DNAPL and production of more soluble daughter products



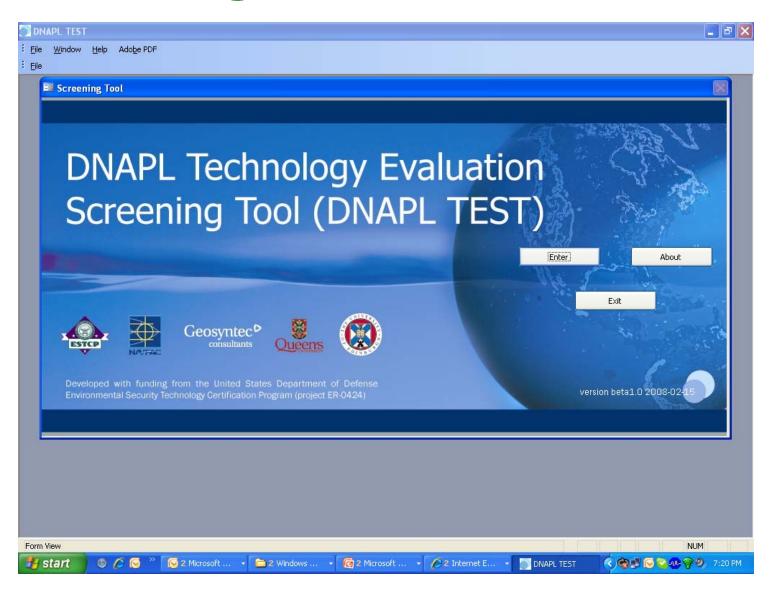
Long-Term Impact on Concentrations



- Extent of mass removal impacts long-term groundwater concentrations
 - Thermal most likely to have near complete mass removal, others more likely to be partial mass removal



Screening Tool Demonstration





Screening Tool Demonstration



General Analysis

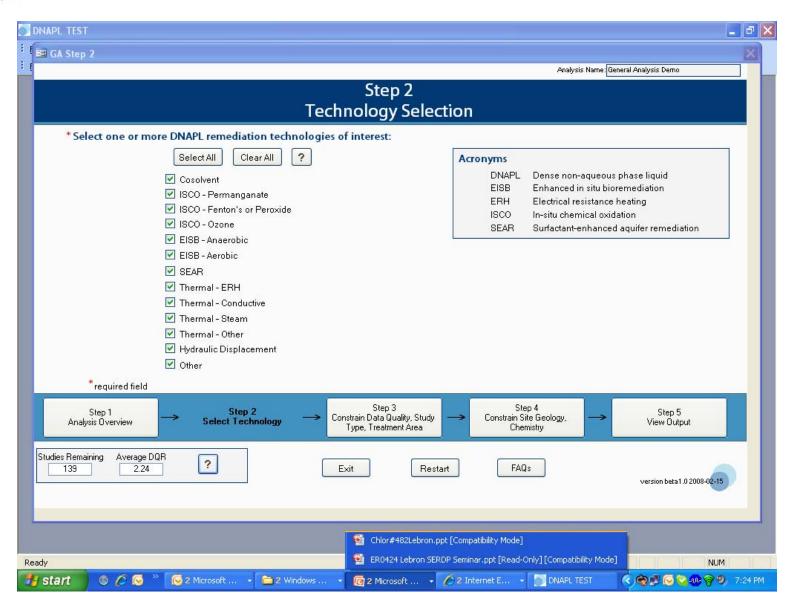
- General trends in tech performance
- Filter out various factors to narrow analysis, evaluate changes

Site-Specific Analysis

- User inputs site parameters of interest
- Tools searches for statistically similar case studies, and outputs technology performance info

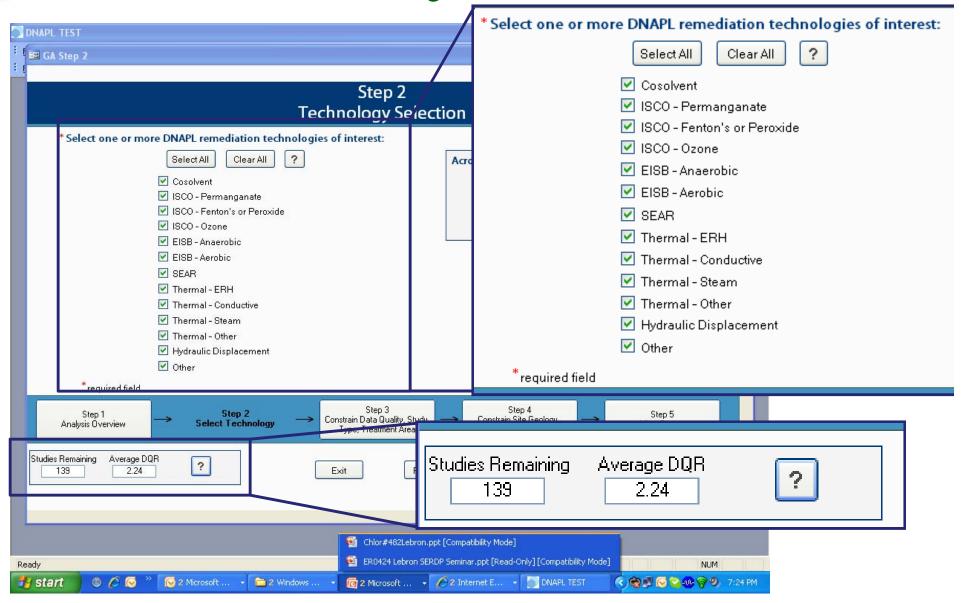




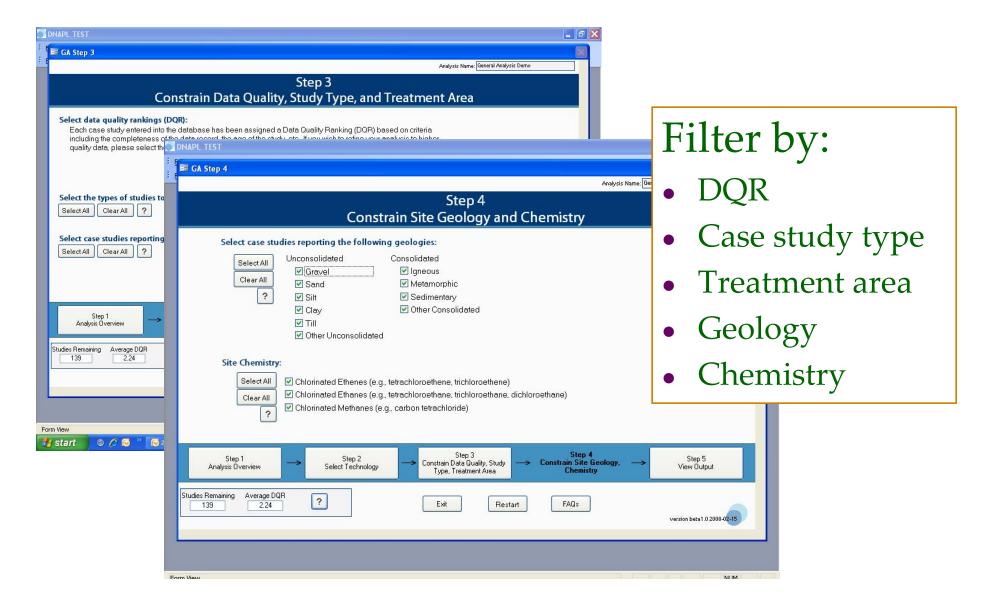






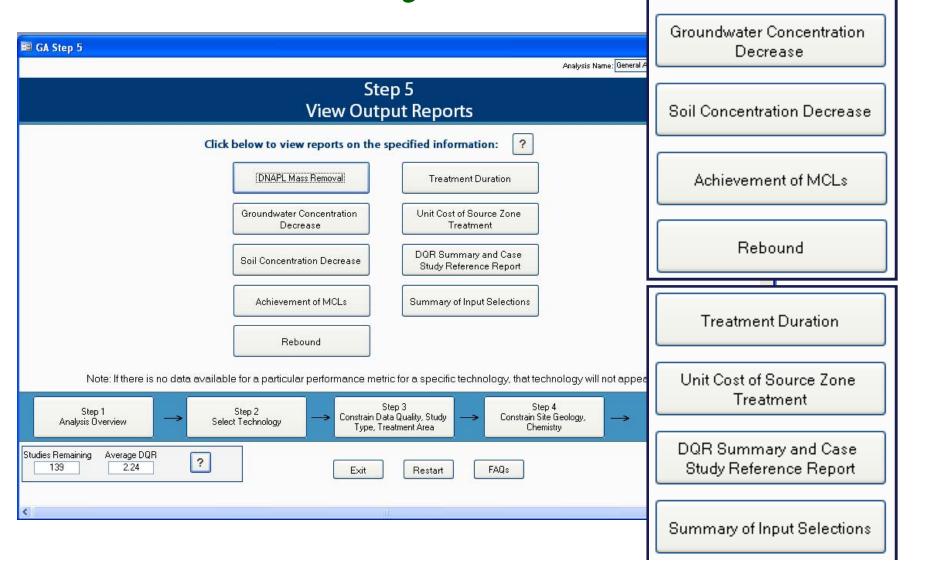




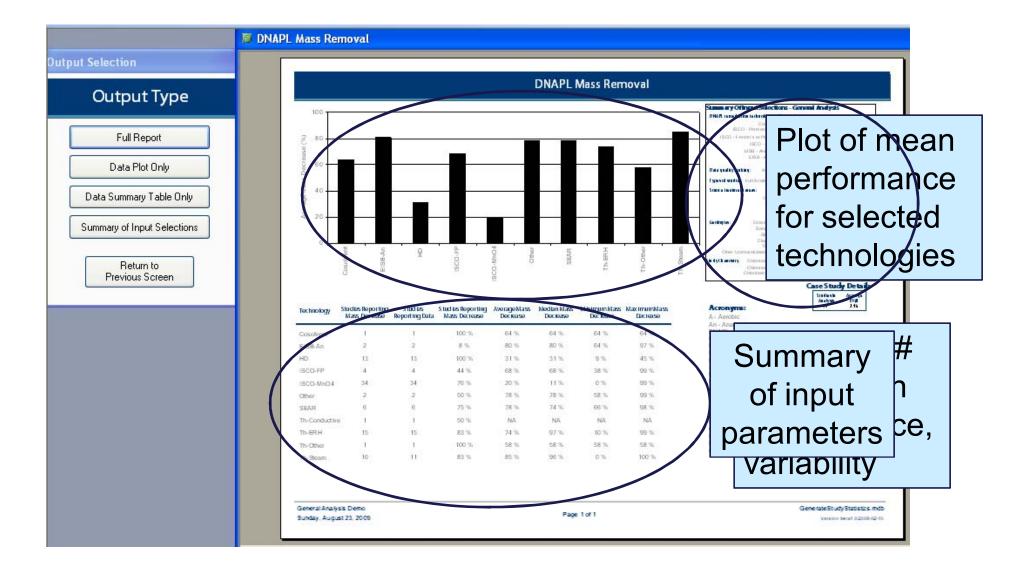




DNAPL Mass Removal







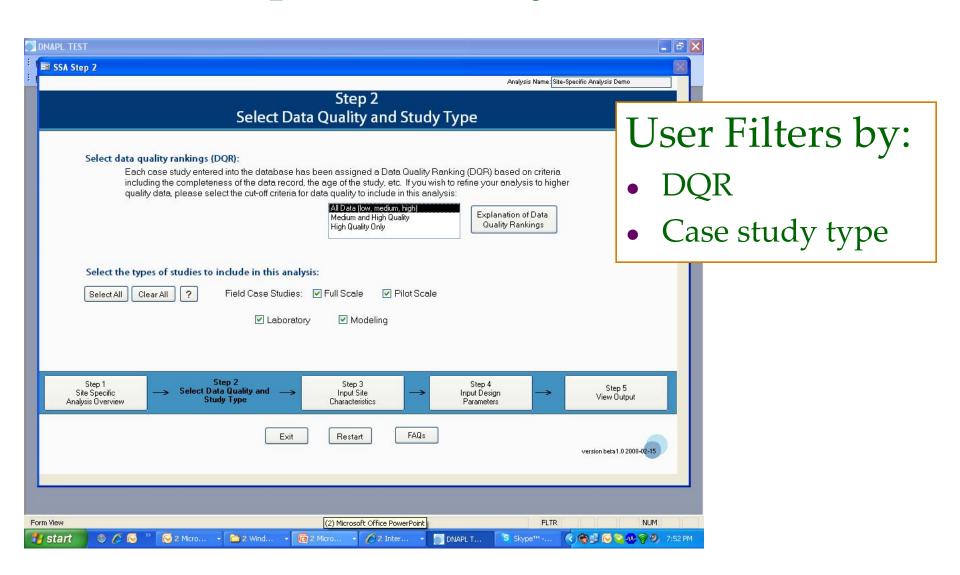


	DQR Summary and Case Study Reference General Analysis Demo					
Technology	Case Study Identifier	DQR	Reference Type	Citation		
Cosolvent		2.0				
	141	2.8	Journal	Jawitz JW, Sillan RK, Annable MD, Rao PSC, Warner K. In- Situ Alcohol Flushing of a DNAPL Source Zone at a Dry Cleaner Site. Envirionmental Science and Technology. 2000 34: 3722-3729.		
dividual ca study	se		Government	IRTC DNAPL Team Case Study Summary Report: Sages Dry Cleaners Jacksonville Florida. www.irtcweb.org/Documents//DNAPLs-3		
reference information		1.3	Conference	Lewis RF, Dooley MA, Johnson JC, and Murray WA. 1998. Sequential anaerobic/aerobic biodegradation of chlorinated solvents: Pilot-scale field demonstration. In Proceedings of the First International Conference on Remediation of Chlorinated and Recalcitrant Compounds, pp. 1-7, vol. C1-6 Physical, Chemical, and Thermal Technologies (Editors:		
EISB-An				Wickramanayake GB and Hinchee RE), Monterey CA, May 18-21.		
	19	2.1	ESTCP	Martin J and Sorenson K. Appendix E.1-Case study of enhanced bioremediation of a DNAPL source area: four years of data from Test Area North, INEEL. In: Principles		

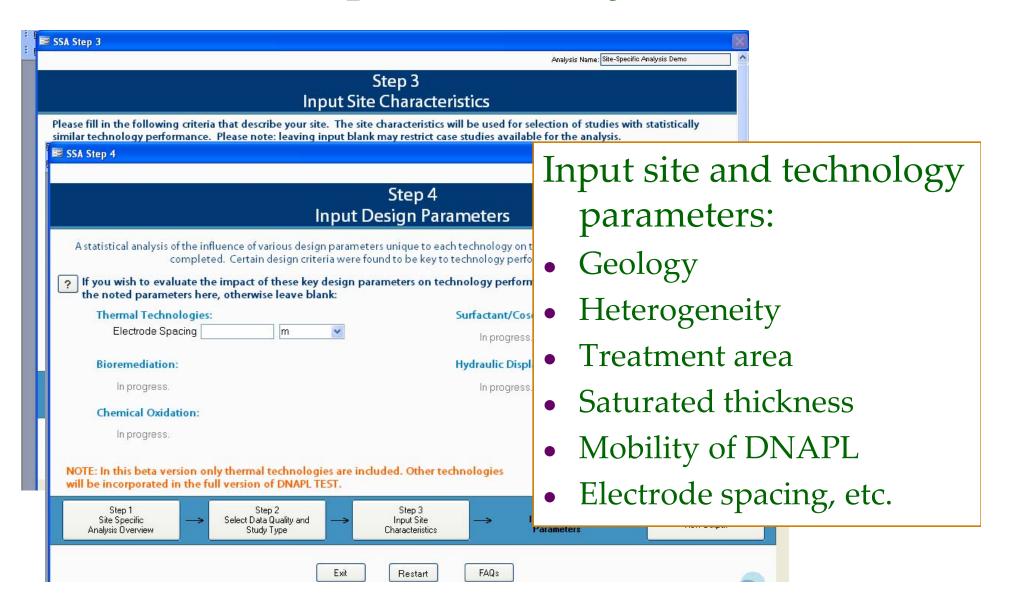


- Demo site characteristics:
 - Unconsolidated media, fine to medium-grained sand
 - 20,000 ft² DNAPL source area
 - Saturated aquifer thickness of 10 ft
 - DNAPL is present as both residual and pools
 - Moderate soil heterogeneity (3-5 order of magnitude variability)
- Technology of interest thermal technologies
 - Statistically "similar" case studies in database identified by DNAPL TEST using relationships identified as part of the linear and non-linear multi-variate analysis

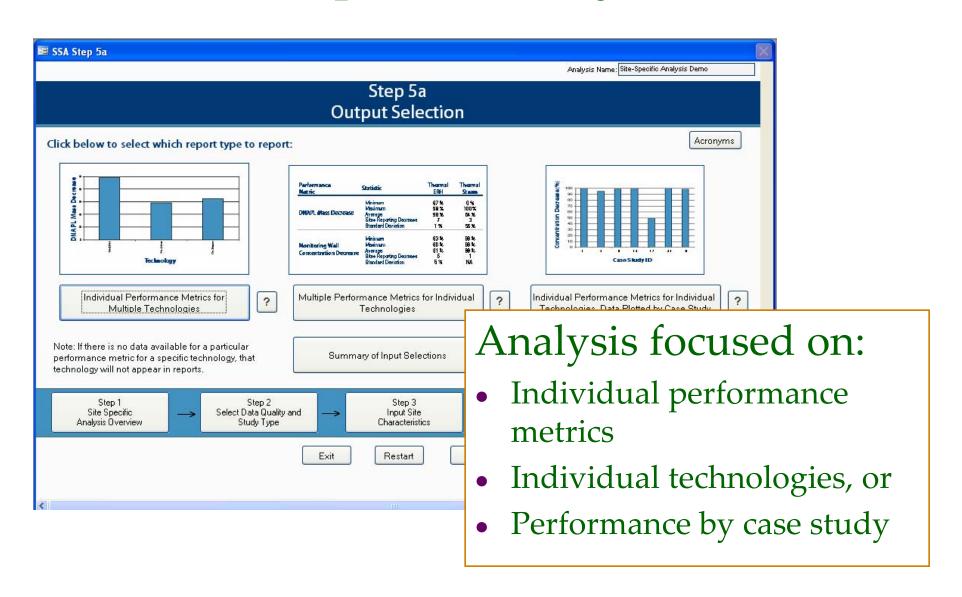




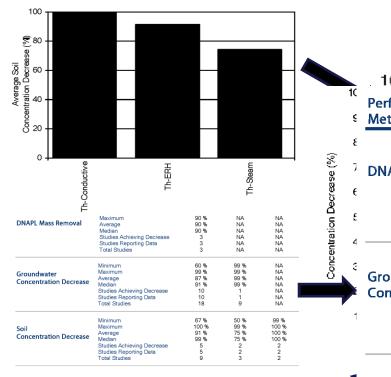




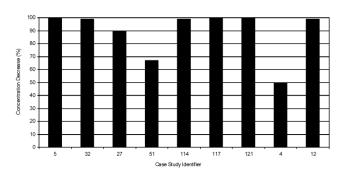








	100				
1C g	Performance Metric	Statistic	Thermal ERH	Thermal Steam	Thermal Conductive
8 7 €	DNAPL Mass Removal	Minimum Maximum Average Median Studies Achieving Decrease Studies Reporting Data Total Studies	90 % 90 % 90 % 90 % 3 3	NA NA NA NA NA NA	NA NA NA NA NA NA
4 23	Groundwater Concentration Decrease	Minimum Maximum Average Median Studies Achieving Decrease Studies Reporting Data Total Studies	60 % 99 % 87 % 91 % 10 10	99 % 99 % 99 % 99 % 1 1	NA NA NA NA NA NA
	Soil Concentration Decrease	Minimum Maximum Average Median Studies Achieving Decrease Studies Reporting Data Total Studies	67 % 100 % 91 % 99 % 5 5	50 % 99 % 75 % 75 % 2 2 3	99 % 100 % 100 % 100 % 2 2 2





Summary

- This project has resulted in the creation of one of the most comprehensive database on source treatment technologies
- The modeling has shown to be a powerful means to:
 - Understand what factors affect performance,
 - Allow us to develop case studies for various situations where there are no documented cases, and
 - Increase our knowledge on how these technologies work in different environments
- The tool is infinitely scalable:
 - We can add more data,
 - Run analysis that allows "filtering" of knowledge
 - eg, new information that indicates rind dissolution, which can be modeled
- Screening Tool available by Spring 2010
- Periodic updates as warranted
 - New Case Studies or technologies
 - Enhancements to technologies
 - Continued site monitoring; i.e., Rebound



Short Course Agenda

8:30 AM	Welcome and Introduction	Hans Stroo				
8:40 AM	Source Zone Protocol for Remedy Selection of Chlorinated Solvents Released at DoD Facilities	Thomas Sale & Charles Newell				
9:50 AM	Break					
10:10 AM	Development of a Protocol and Screening Tool for Selection of DNAPL Source Area Remediation	Carmen Lebrón, Bernard Kueper, David Major, Julie Konzuk & Jason Gerhard				
11:50 AM	Lunch					
1:00 PM	Decision & Management Tools for DNAPL Sites: Optimization of Chlorinated Solvent Source and Plume Remediation Considering Uncertainty	Ronald Falta & Charles Newell				
2:20 PM	Emulsion Design Tool for Planning Aqueous Amendment Injections Systems	Robert Borden				
2:50 PM	Break					
3:10 PM	Permanganate Design Tool for Planning Aqueous Amendment Injection Systems	Robert Borden				
4:00 PM	Improved Field Evaluation of NAPL Dissolution and Source Longevity	Michael Kavanaugh, Mark Widdowson, Lloyd Stewart & Rula Deeb				
5:20 PM	Summary & Conclusion	Hans Stroo				



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Decision & Management Tools for DNAPL Sites: Optimization of Chlorinated Solvent Source and Plume Remediation Considering Uncertainty

Presented by: Ronald Falta (Clemson University) and Charles Newell (GSI Environmental, Inc.)



ESTCP ER-0704

- Ronald W. Falta and Hailian Liang Clemson University
- Charles J. Newell and Shahla Farhat, GSI Environmental, Inc.
- P. Suresh Rao and Nandita Basu Purdue University



Model Objectives

Develop a practical analytical tool that allows site managers to:

- Quickly simulate changes in DNAPL source zones and dissolved plumes over time, with and without source remediation, source containment, and/or plume remediation
- Explore site management decisions in a probabilistic framework, so uncertainty becomes an integral part of the decision making process
- Compare the cost, risk, and performance of source treatment to plume treatment approaches



Key Concept: Sources

- Most dissolved plumes can be traced back to a concentrated "source" area, where the original release occurred.
- The source area is usually small compared to the plume footprint
- The source may contain DNAPL, or it may consist of high concentrations of dissolved solvents in low permeability zones
- The mass of contaminant in the source zone, and the mass discharge of contaminant out of the source zone play a central role in the evolution of dissolved plumes



Key Concept: Plumes

- Plumes are fed by the source, and move with the groundwater flow with some dispersion
- The dissolved contaminants may also adsorb or diffuse into aquifer materials
- The groundwater pore velocity (Darcy velocity divided by porosity) and the rate at which the chemical degrades play a central role the nature of the plume
- High velocities with low decay rates = large plumes
- Low velocities with high decay rates = small plumes



Questions to be addressed by Mass Balance Type Modeling

- What will happen if no action is taken?
- Will source remediation meet site goals? How effective must the source remediation be?
- Will enhanced biodegradation of the plume meet site goals? How effective (and long-lived) must the plume treatment be?
- Should I combine source and plume remediation? How much of each do I need before I get to transition to MNA?
- What is the remediation time-frame?
- What is a reasonable remediation objective?





Core Model: REMChlor

Source

Analytical model for source behavior

Flow Plume

> **Analytical model for** plume response

Mass balance model on source zone predicts discharge including effects of remediation

Couple Models At the Edge of the Source Zone to **Provide Contaminant** Discharge to Plume Model

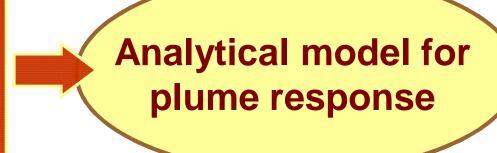
Plume model simulates mass balance based on advection, dispersion, retardation, and degradation reactions

plume remediation (but all with simple flow field)



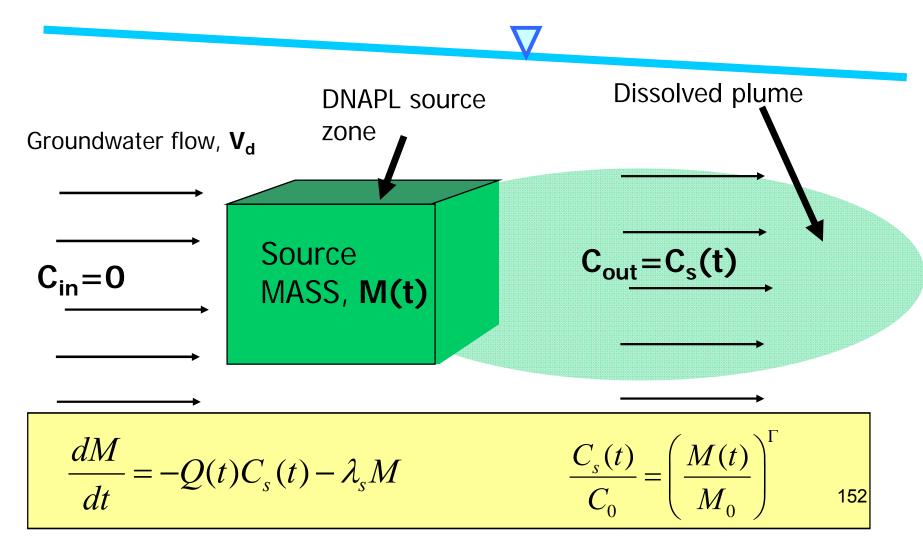
Source TermWorks in REMCHLOR

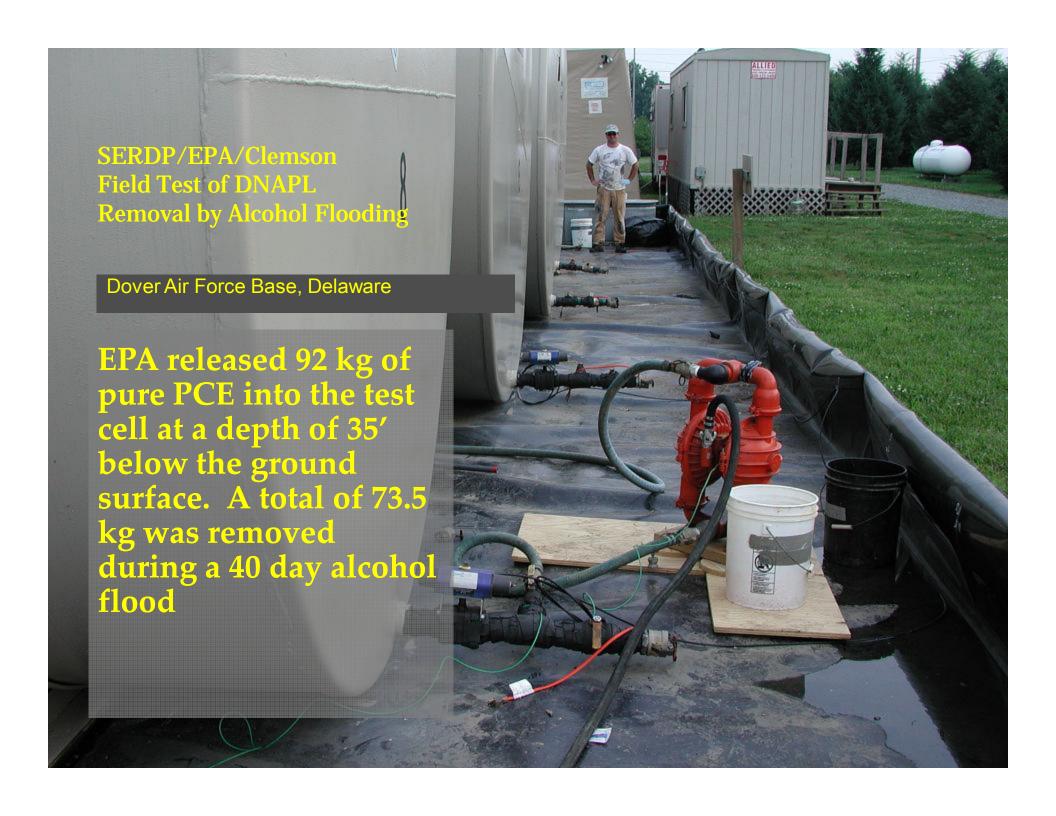
Analytical model for source behavior





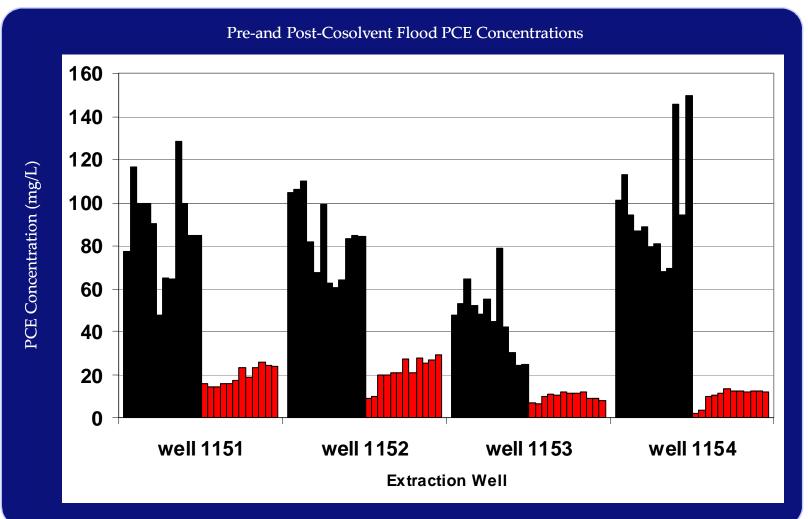
Source conceptual model: Mass is mainly removed by flushing. The discharging concentration (C_s) depends on the mass remaining in the source zone, (M)







80% Source Removal Resulted in 81% Reduction in Groundwater Concentration

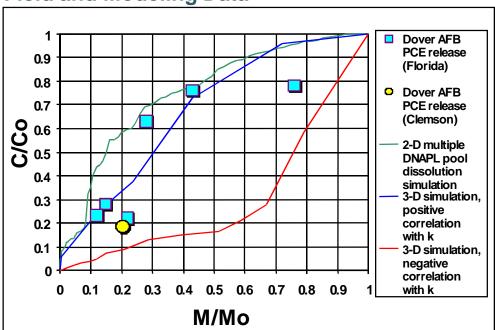




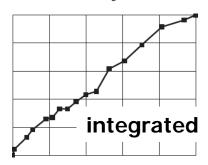
Source Mass Reduction Leads to Discharge Reduction

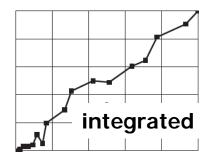
(Jawitz et al.)

Field and Modeling Data



Laboratory dissolution experiments

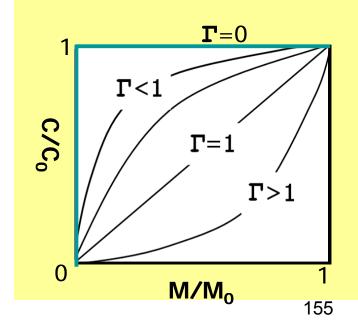




Power function model

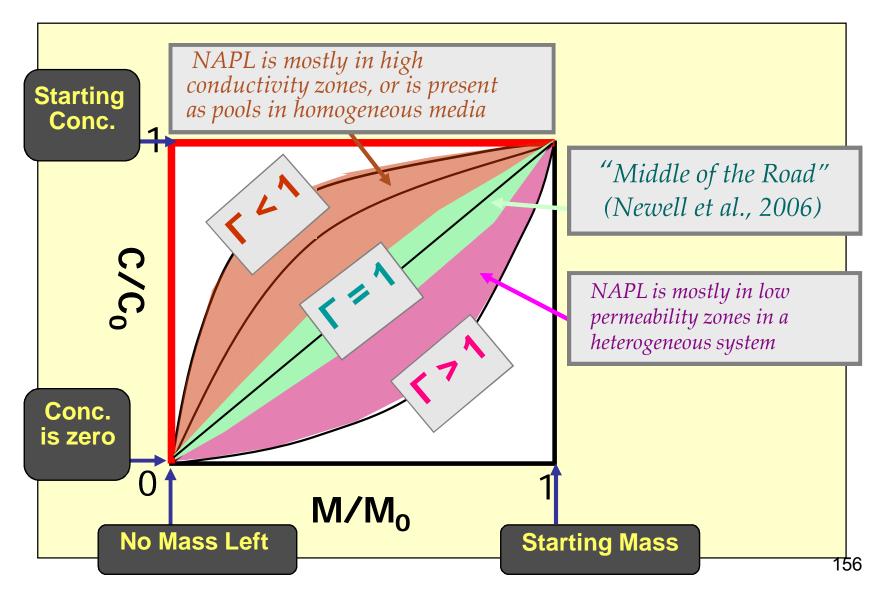
[Rao et al., 2001; Parker and Park, 2004; Zhu and Sykes, 2004]

$$\frac{C}{C_0} = \left(\frac{M}{M_0}\right)^1$$



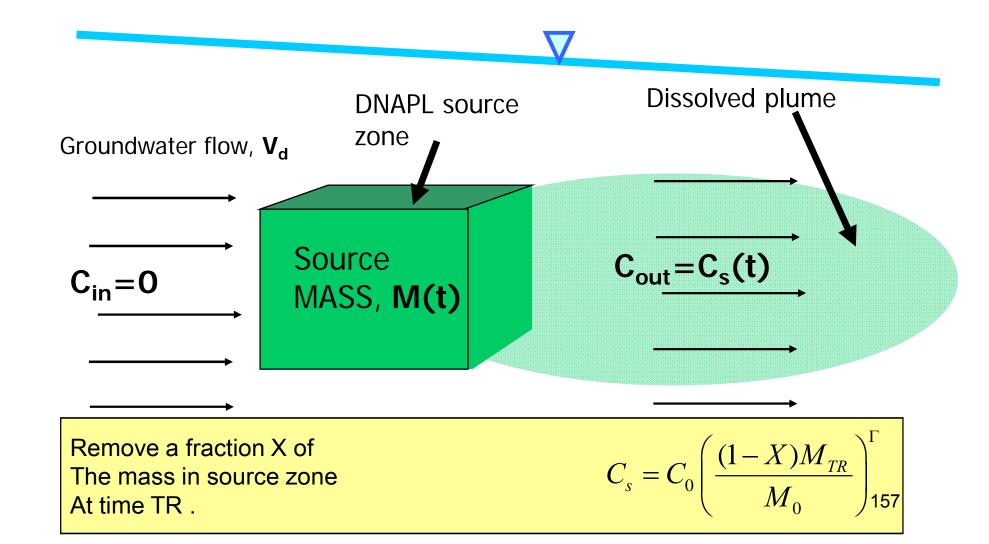


Source Power Function - What's That?



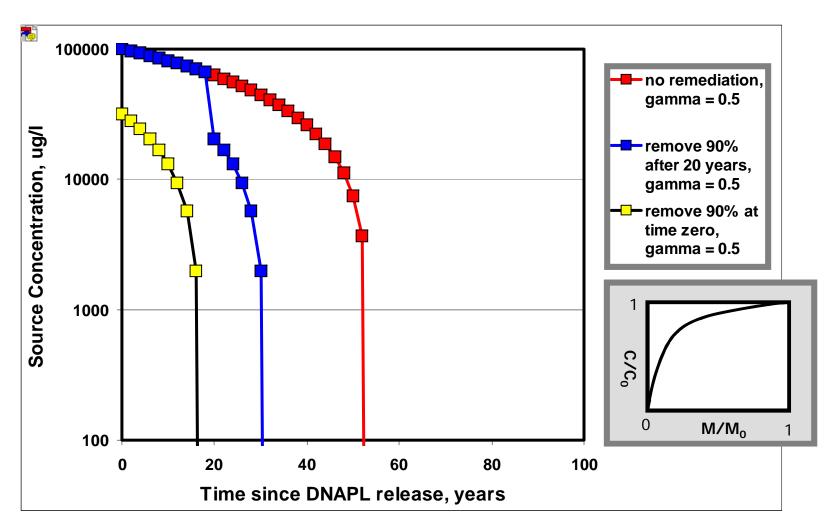
Source conceptual model: Remediation is simulated by removing a fraction of the source mass at the time of remediation





Source Behavior:

$$\Gamma = 0.5$$
, $M_0 = 1,620$ kg, $V = 20$ m/yr, $A = 10$ m x 3m, $C_0 = 100$ mg/l

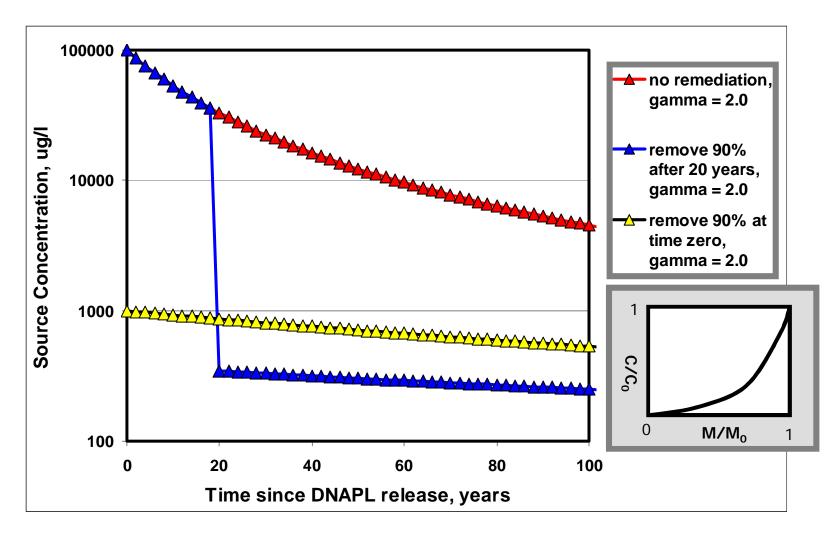


Source Behavior:



$$\Gamma = 2.0, M_0 = 1,620 \text{ kg},$$

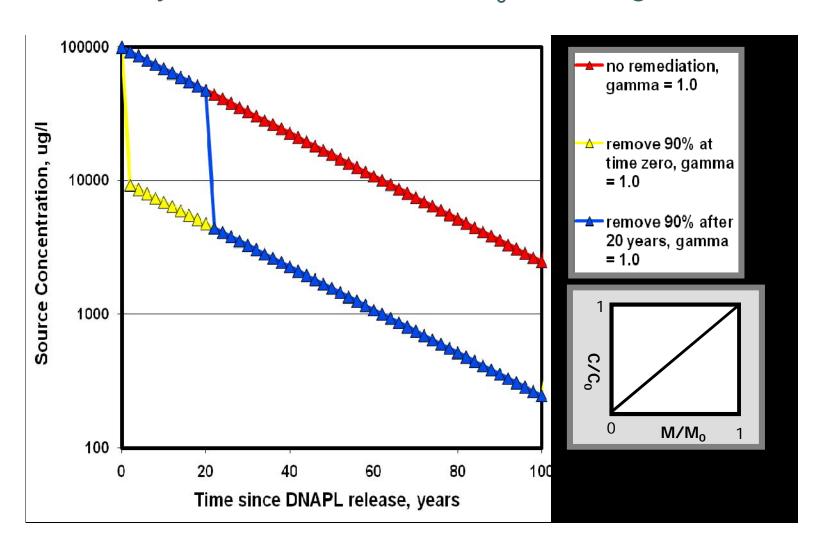
 $V = 20 \text{ m/yr}, A = 10 \text{m x 3m}, C_0 = 100 \text{ mg/l}$



Source Behavior:



$$\Gamma$$
 = 1.0, M_0 = 1620 kg, V = 20 m/yr, A = 10m x 3m, C_0 = 100 mg/l





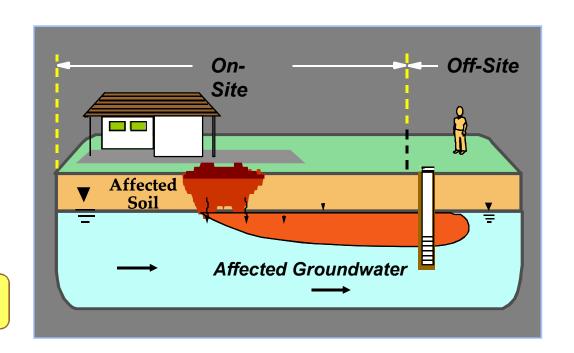
Explanation of How the Plume Works in REMCHLOR

Analytical model for source plume response plume response



Key Process in REMChlor

- Source Term
- Advection
- Dispersion
- Adsorption
- Biodegradation

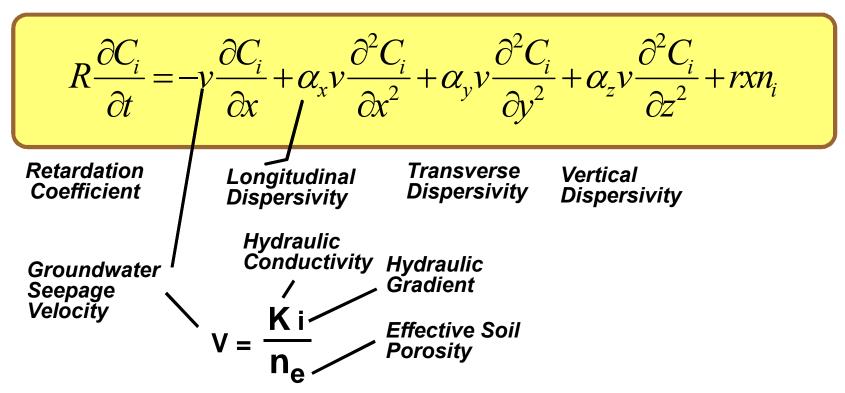




Key Mass Balance Equations - Plume

Plume equation solved for each species. Equations are linked through the chemical reaction terms

First-Order Decay reactions





Groundwater Transport Processes - Biodegradation

Indigenous micro-organisms are capable of degrading many contaminants.

Need electron donor and electron acceptor.

Fuels like benzene serve as electron donor. Oxygen, nitrate, sulfate, iron are electron acceptor.

Chlorinated solvents act as electron acceptor. Hydrogen/acetate serve as electron donor.

Biodegradation Decay Chain for Chlorinated Ethenes

Key footprints

cis-DCE

ethene or ethane

PCE Rapid; occurs under all anaerobic conditions **TCE** Rapid; occurs under all anaerobic conditions **Aerobic Oxidation** by Cometabolism cis-1,2-DCE **Direct Aerobic** Oxidation (?) Slower; sulfatereducing and methanogenic conditions Aerobic or Anaerobic Oxidatio VC Slower; sulfate reducing and methanogenic conditions only Aerobic Oxidation **Ethene**

Halorespiration (Reductive dechlorination)

(Adapted from RTDF, 1997)





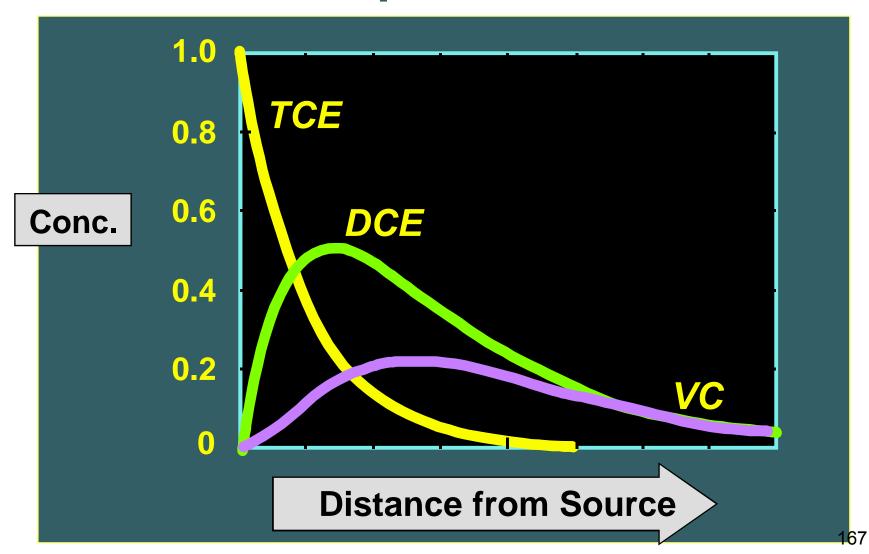
Sequential Reactions

$$Rate_{PCE} = - \lambda_1 C_{PCE}$$

Rate
$$_{TCE} = \lambda_1 y_1 C_{PCE} - \lambda_2 C_{TCE}$$



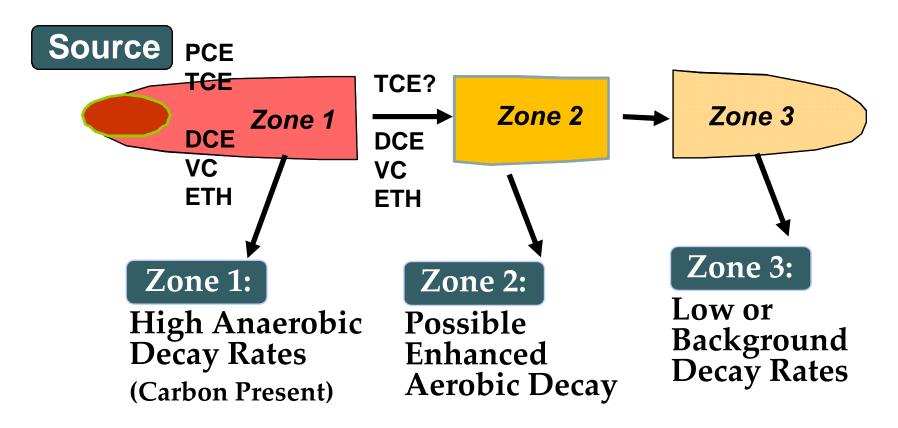
Results of Sequential Reactions





REMChlor Model: Other Features

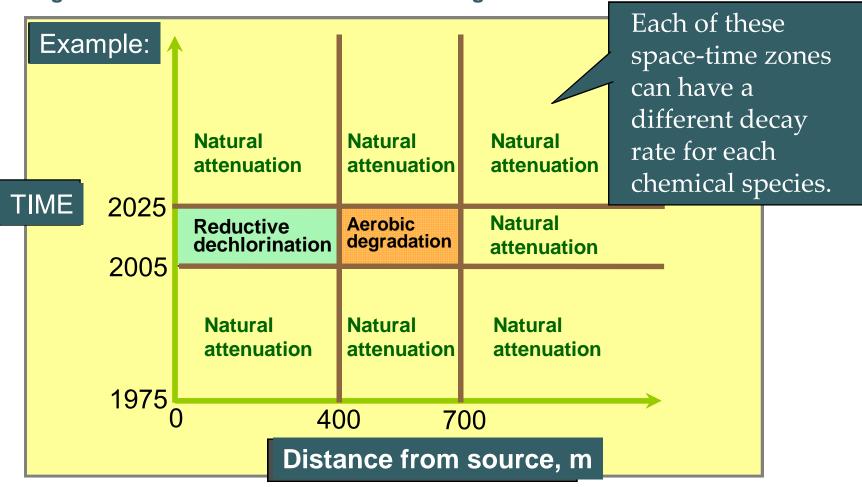
Three Reaction Zones for Mixed Sites



Plume Remediation Model



Divide space and time into "reaction zones", solve the coupled parent-daughter reactions for chlorinated solvent degradation in each zone





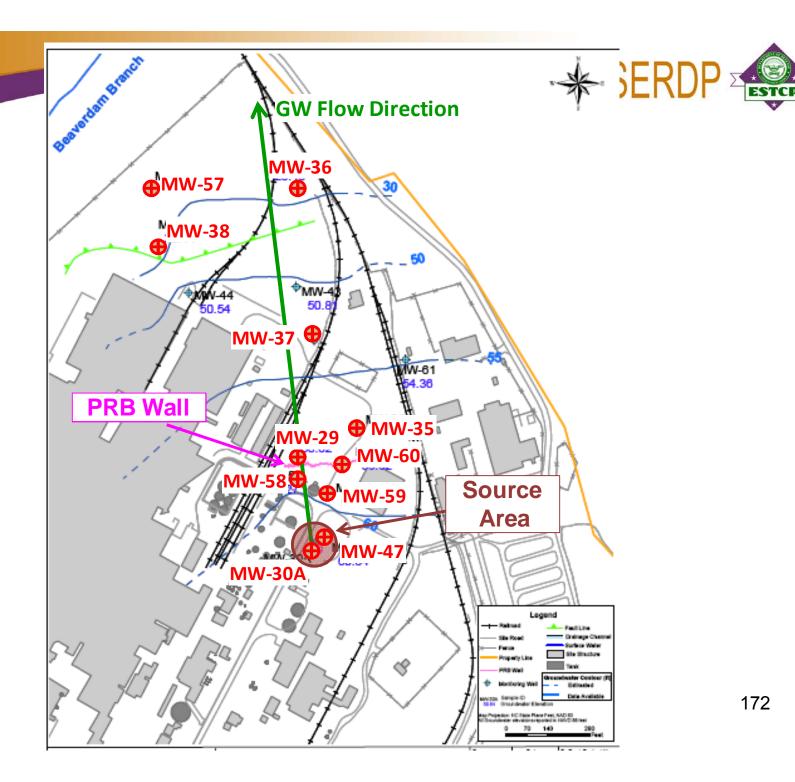
REMChlor Case Study: TCE plume at a manufacturing plant in North Carolina

- DuPont Kinston Plant in eastern NC, currently produces Dacron polyester resin and fibers
- TCE contamination of groundwater discovered in the late 1980's; ~ stable plume about 1250 ft long (380 m).
- Release date unknown, but before 1980.
- Plume is dominated by TCE; small amounts of cis-1,2-DCE are present and VC is essentially absent
- Groundwater velocity is slow, less than 100 ft/yr pore velocity



REMChlor Case Study: TCE plume at a manufacturing plant in North Carolina

- Source zone TCE mass estimated at 300 lbs (136 kg), source zone concentrations up to ~6,000 ug/l
- Source remediation took place in 1999, consisting of ZVI injection throughout the suspected source zone. Although source mass removal was reported as 95%, wells in the source zone have not seen large reductions in concentration.
- A 5 inch thick permeable reactive barrier (PRB) using ZVI was installed 290 ft downgradient of the source in 1999.



REMChlor Model Parameters for Transport/Natural Attenuation

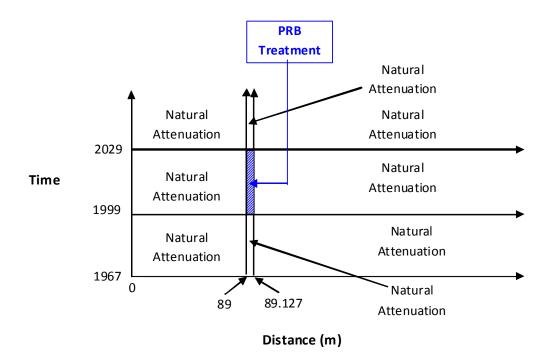


Parameter	Value	Comment
Initial Source Conc., C _o	6,000 ug/l	Estimated from source wells
Initial Source Mass, M _o	136 kg	From site reports; assume 1967 release date
Source function exponent, Γ	1	Estimated
Source Width, W	8m	From site reports
Source Depth, D	3.5m	From site reports
Darcy velocity, V	8m/yr	Calibrated; reports had estimated 1.5 to 4.6 m/yr
Porosity, φ	0.33	From site reports
Retardation Factor, R	2	Estimated
Longitudinal dispersivity, α_l	x/20	Calibrated
Transverse dispersivity, $\alpha_{\rm t}$	x/50	Calibrated
Vertical dispersivity, $\alpha_{ m v}$	x/1000	Estimated
TCE decay rate in plume, λ	0.125/yr	Calibrated (equal to $t_{1/2}$ of 5.5 yrs)

REMChlor Model Parameters for Source and Plume Remediaton

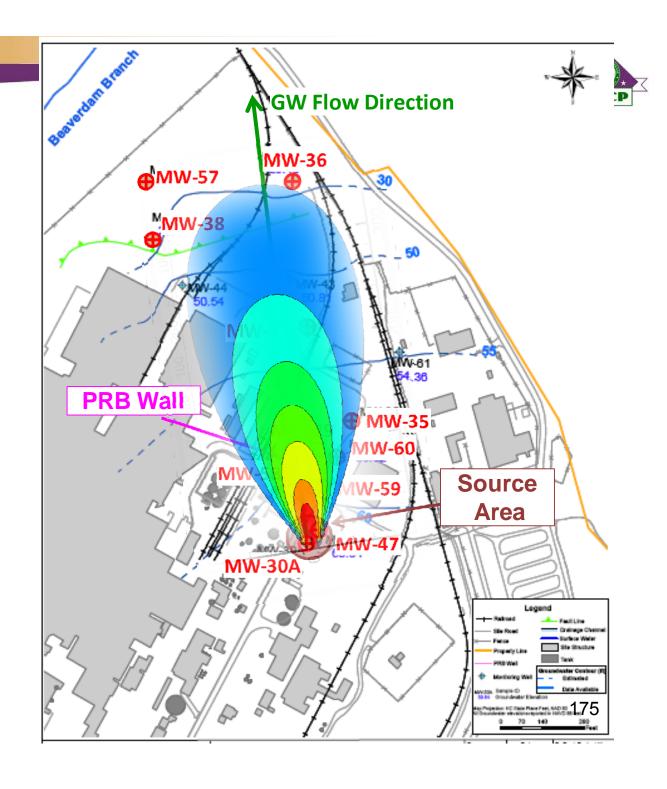


Parameter	Value	Comment
Fraction of source removed in 1999, X	95%	From site reports (but large uncertainty)
PRB wall thickness (after 1999)	0.127m (5")	From site reports
TCE decay rate in PRB	435/yr	Estimated from well data (equal to $t_{1/2}$ of 14 hours)



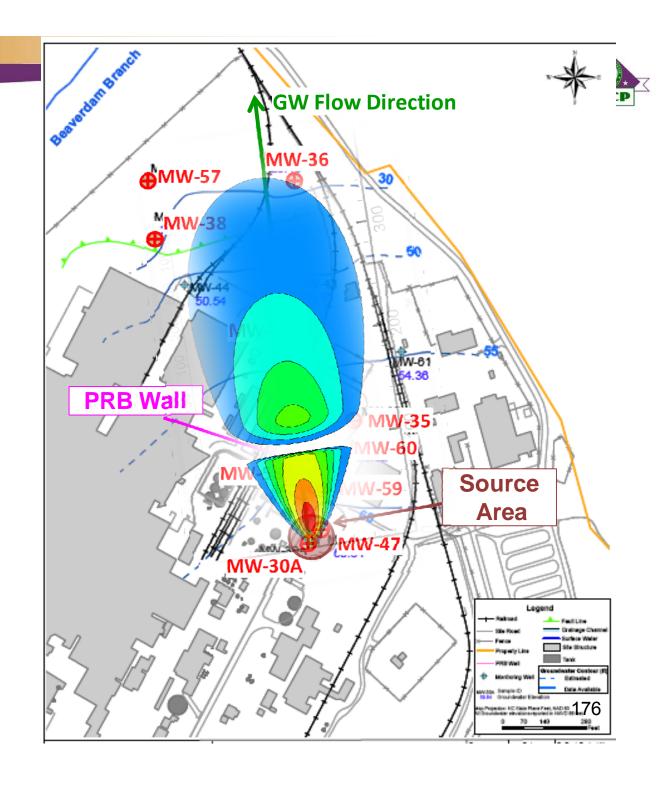
Simulated TCE concentrations In 1999 prior to source remediation or PRB wall installation

Contours at 5, 20, 50,100, 200, 500, and 1000 ug/l



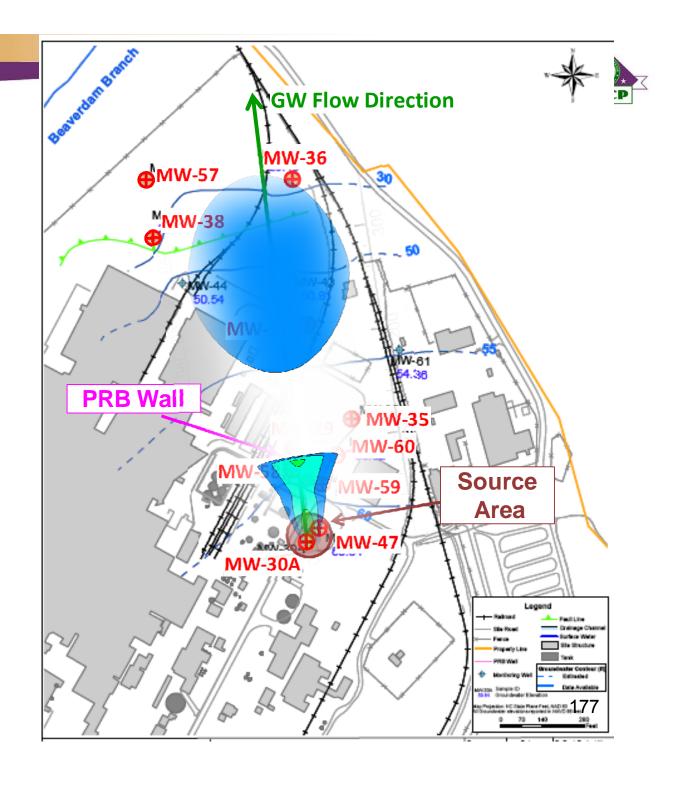
Simulated TCE concentrations In 2001, 2 years after source remediation and PRB wall installation

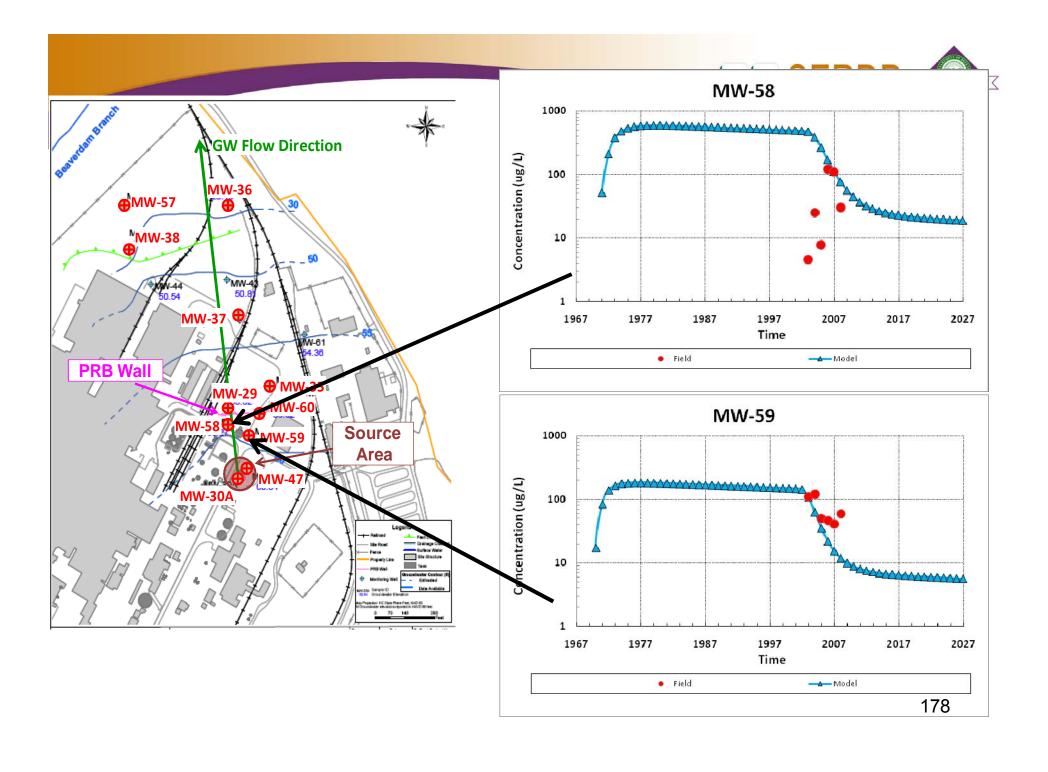
Contours at 5, 20, 50,100, 200, 500, and 1000 ug/l

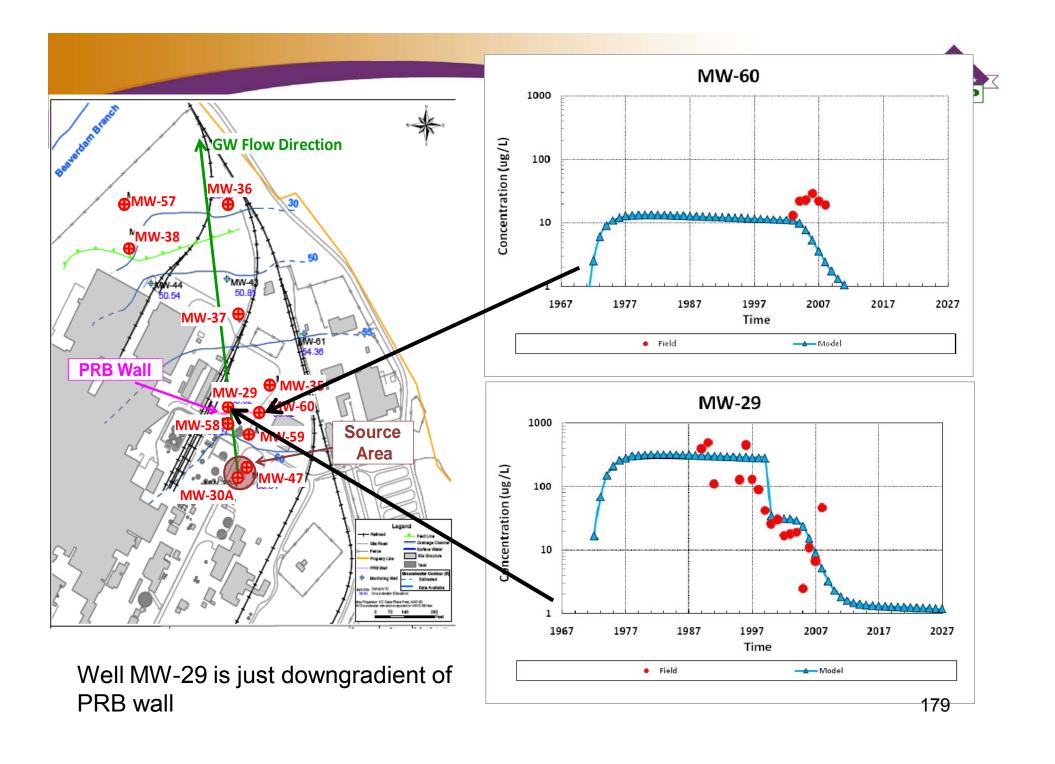


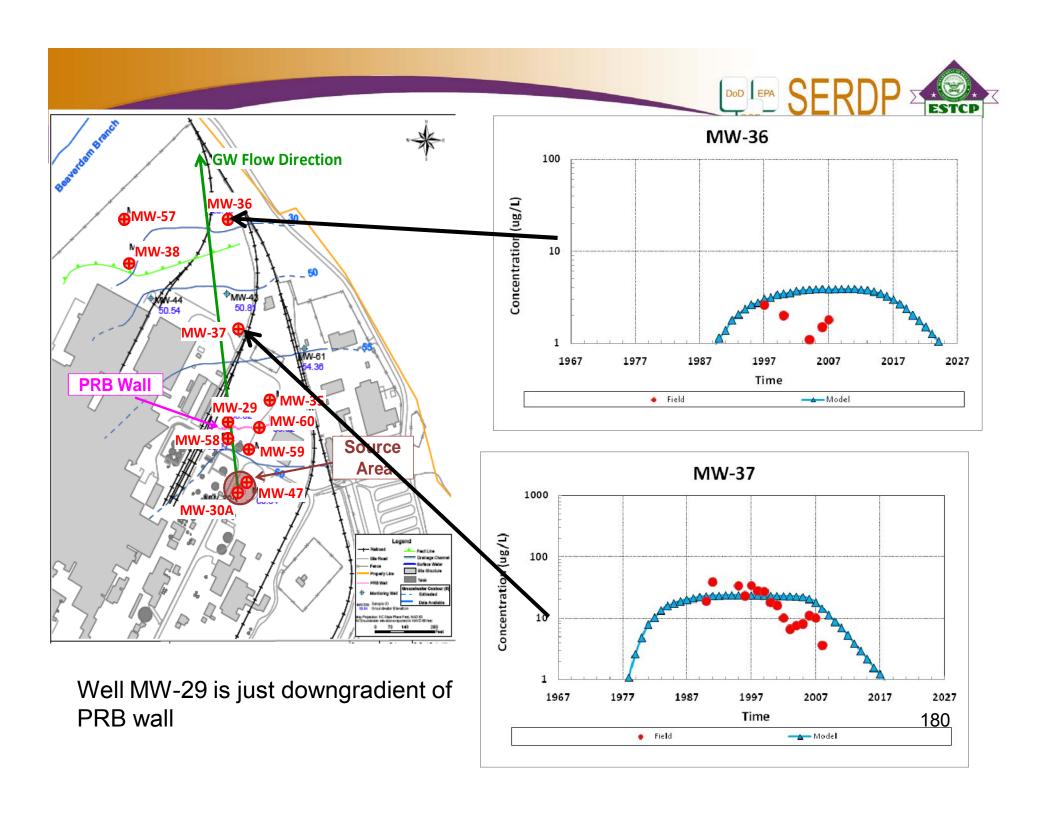
Simulated TCE concentrations In 2009, 10 years after source remediation and PRB wall installation

Contours at 5, 20, 50,100, 200, 500, and 1000 ug/l



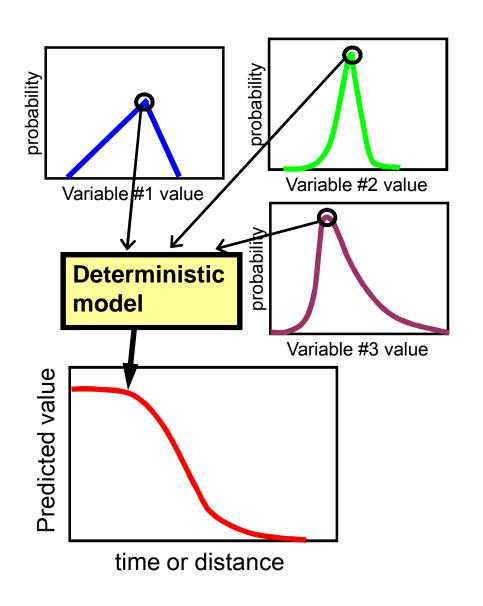


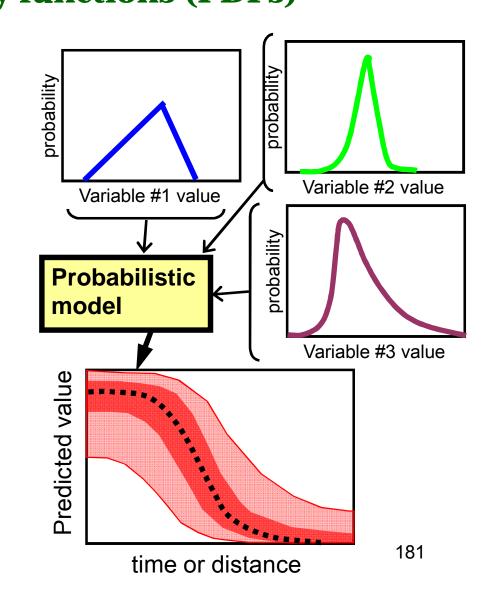




Probabilistic Simulation – treat input variables as uncertain parameters using probability density functions (PDFs)

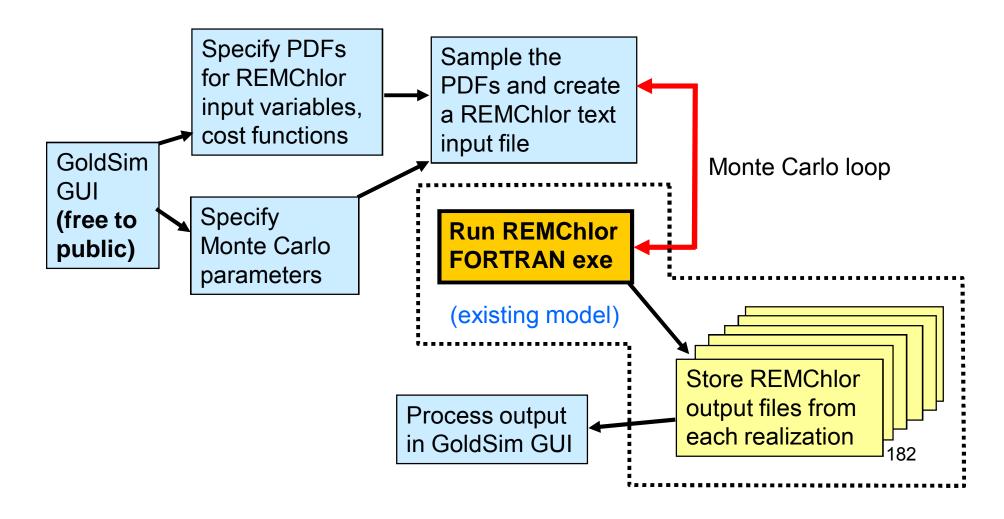








We have coupled the REMChlor FORTRAN code with the GoldSim probabilistic modeling software, and have produced graphical user interface using GoldSim. We now have >70 probabilistic variables

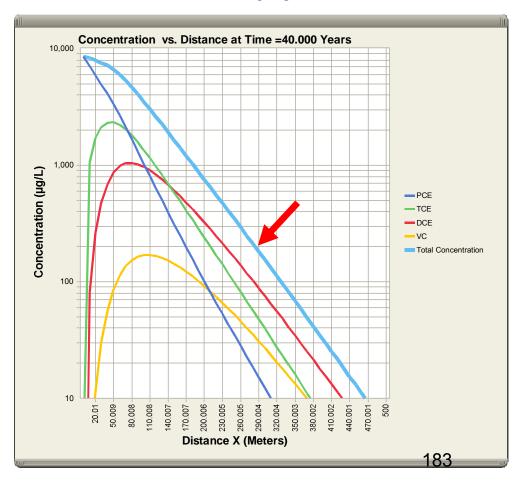




Deterministic REMChlor Example

- 1600 kg release of PCE in 1983
- Plume stabilized in 2001, but is not shrinking
- The 200 ug/l total CVOC contour extends out to 290m in 2008.
- In the next 15 years, the plume will only shrink by 5m without remediation

2023

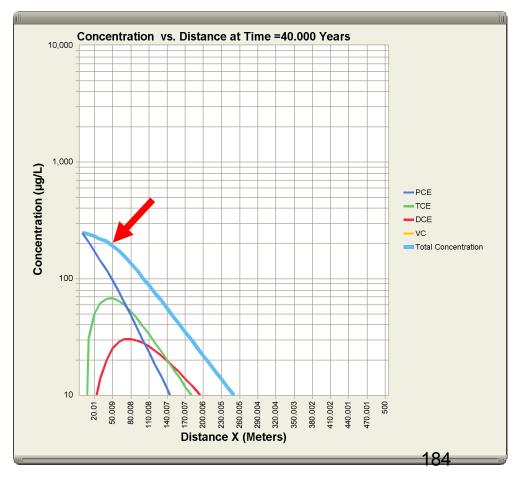




Deterministic REMChlor Example

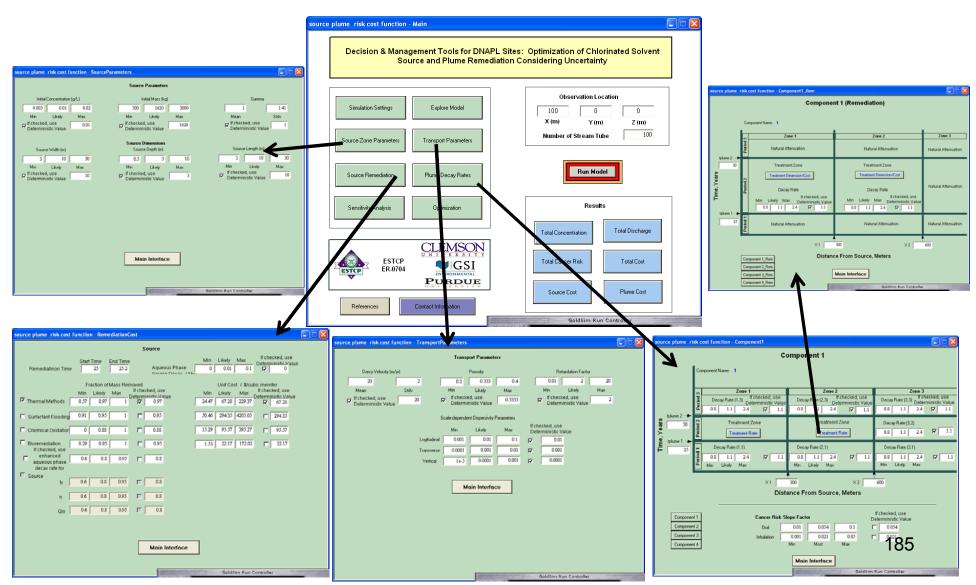
- Simulate a very effective thermal remediation of the source that removes 97% of the source mass this year
- Remediation goal is to shrink the 200 ug/l contour to less than 100m in 15 years
- Maximum plume extent is only 50m, so this remediation should work

2023



Setup REMChlor-GoldSim to SERDP run same problem

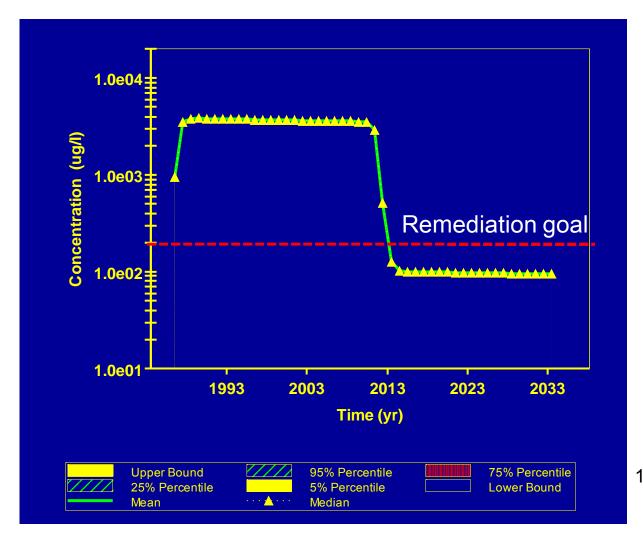






Setup REMChlor-GoldSim to run same problem – deterministic result is the same

Concentration versus time at compliance point located 100m downgradient from source.

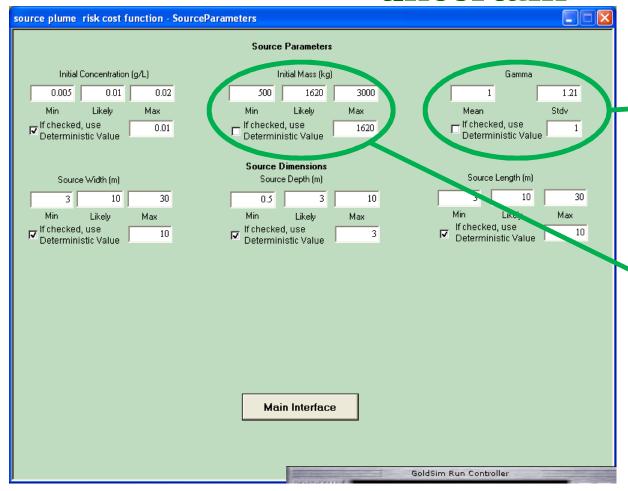


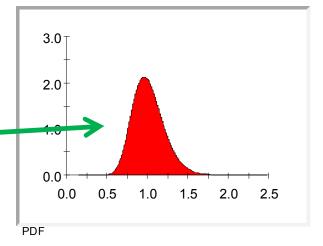


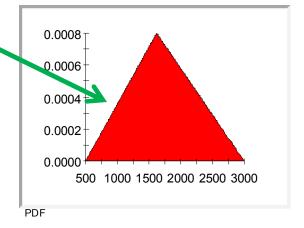


Setup REMChlor-GoldSim to run same problem – make some source parameters

uncertain



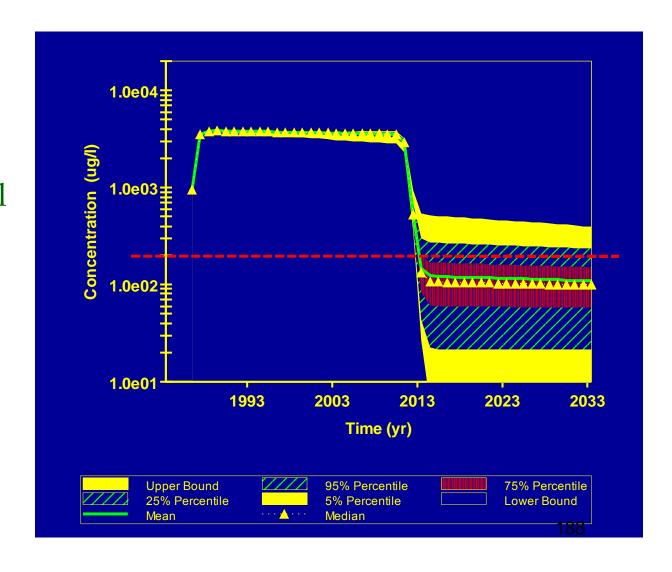






Probabilistic result

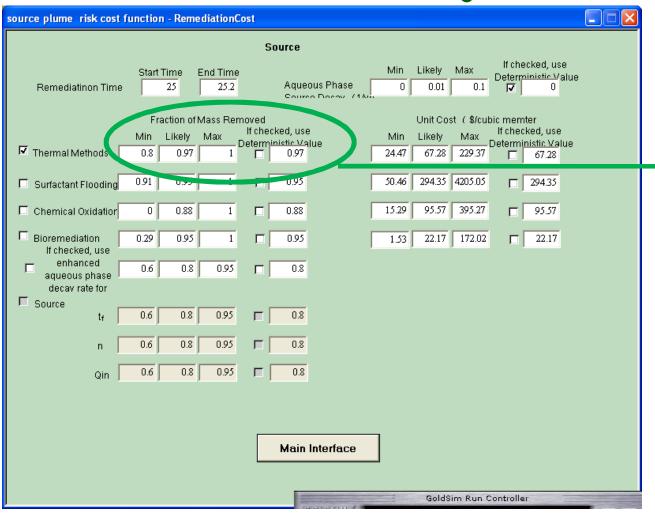
- Model predicts
 >75% chance of
 meeting
 concentration goal
- Upper bound concentration at 100m in 2023 is 460 ug/l.

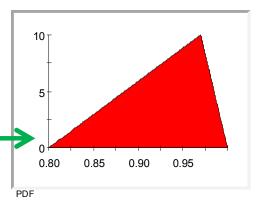






Add remediation efficiency uncertainty

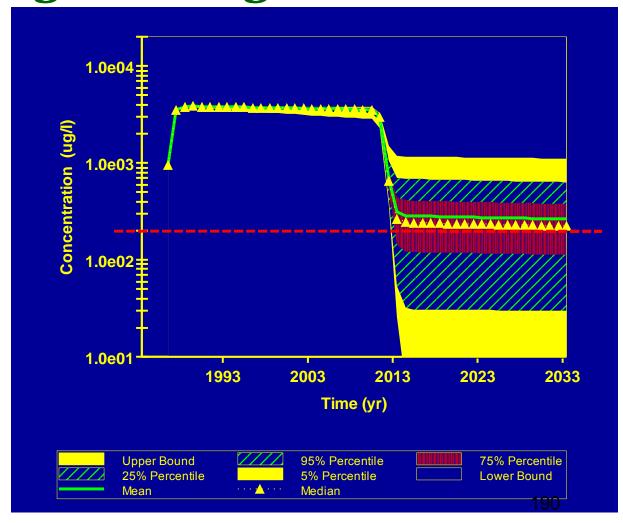






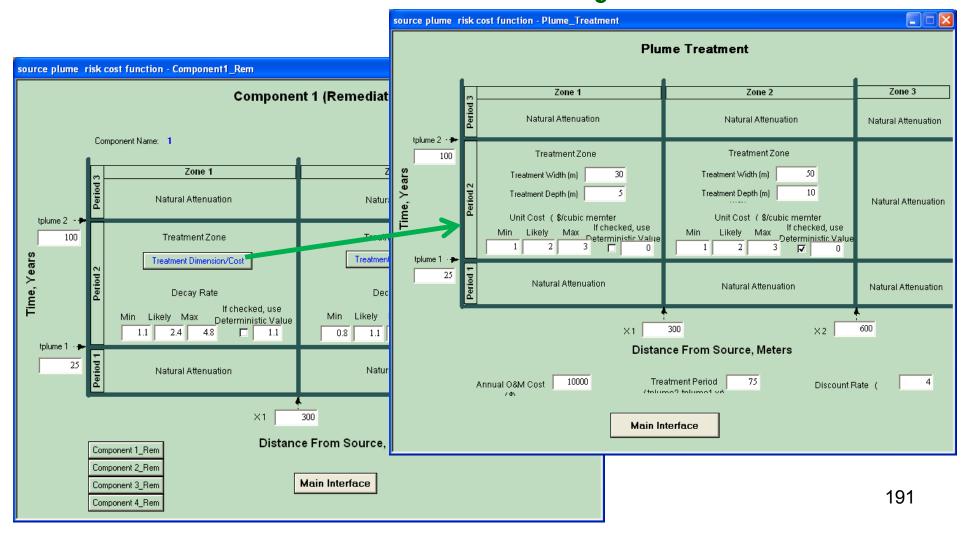
Model predicts possible failure of original design

- Remediation
 effort is
 predicted to
 meet goal only
 ~50% of the time
 given
 uncertainty
- Upper bound concentration at compliance point is 1130 ug/l





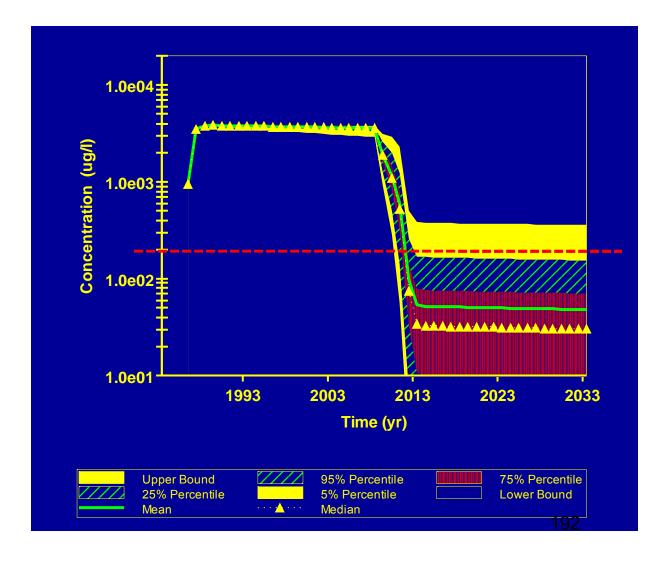
Add enhanced bioremedation of the plume in the first 300m, sustained indefinitely





New design appears to be robust

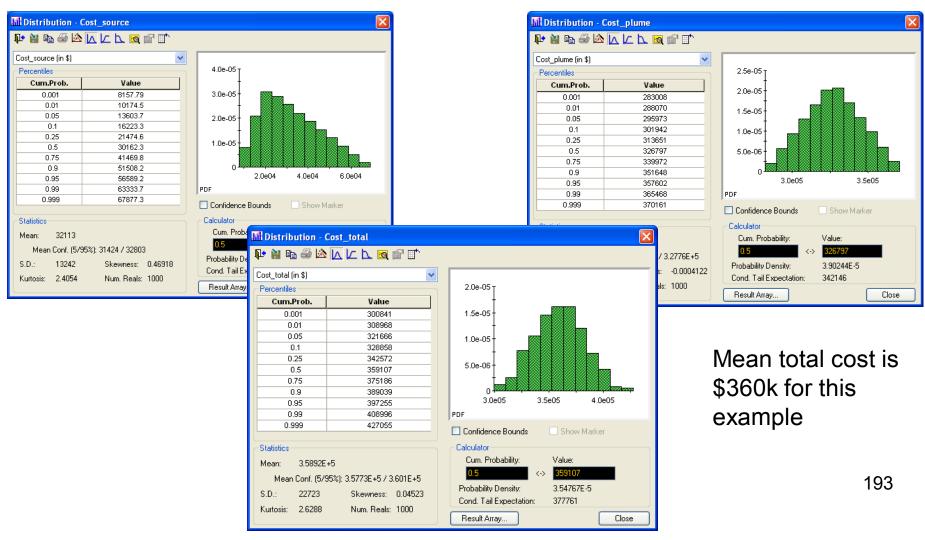
- Remediation
 will meet goal
 with >95%
 certainty
- Upper bound concentration at compliance point is 370 ug/l, which is less than a factor of 2 above the goal







Estimated cost of remediaton (using probabilistic cost functions)





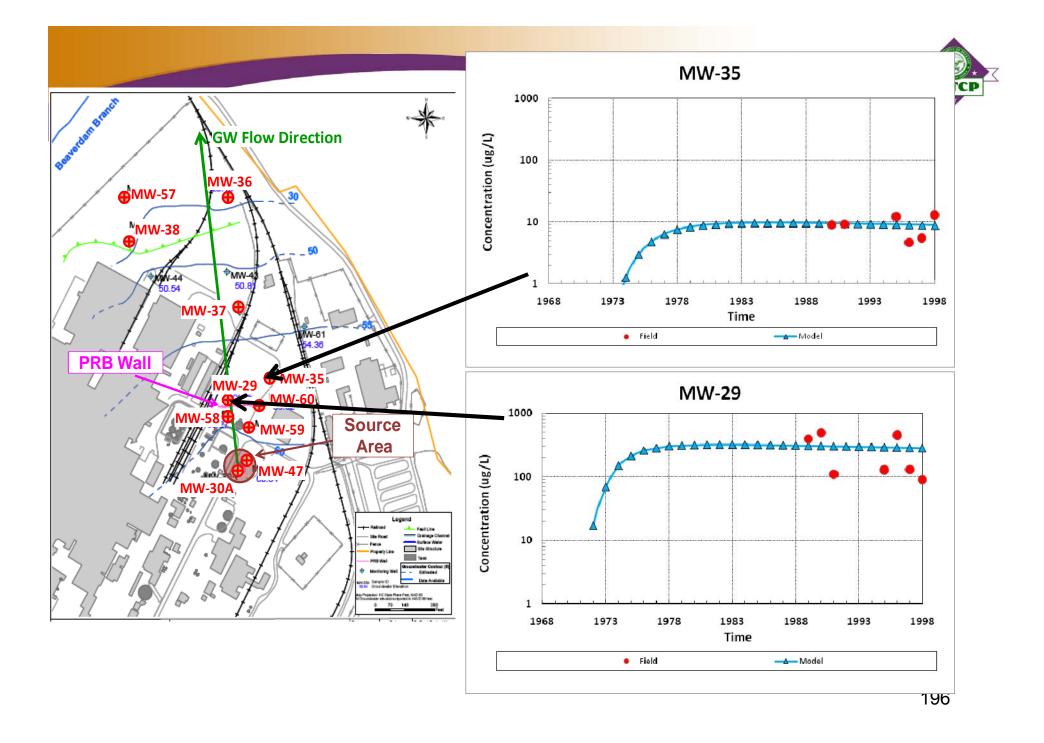
Now Revisit the Kinston, NC site

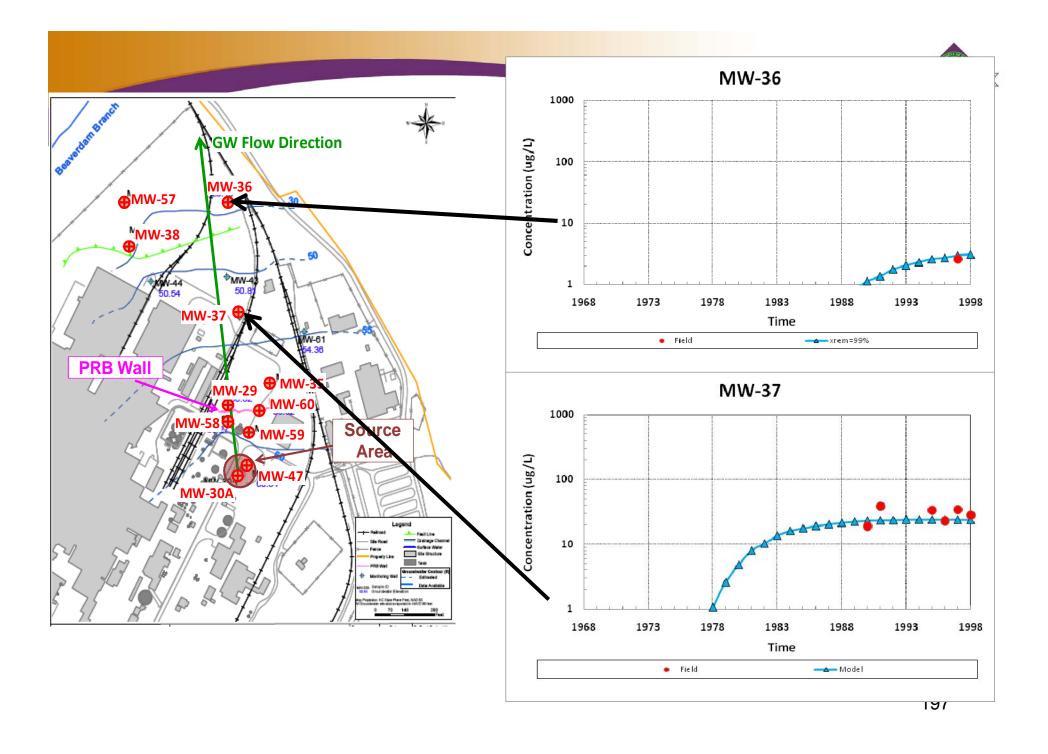
- The earlier example was a calibrated deterministic model
- What if we wanted to predict the response of the system to proposed remediation operations?
- Let's pretend that it is 1998. We have been monitoring this plume for 10 years, so we have some parameter estimates
- We'll use the probabilistic model to simulate the source and plume remediation considering uncertainty

Step 1: Calibrate transport model using pre-1998 data



Parameter	Value	Comment
Initial Source Conc., C _o	6,000 ug/l	Estimated from source wells
Initial Source Mass, M _o	136 kg	From site reports; assume 1967 release date
Source function exponent, Γ	1	Estimated
Source Width, W	8m	From site reports
Source Depth, D	3.5m	From site reports
Darcy velocity, V	8m/yr	Calibrated; reports had estimated 1.5 to 4.6 m/yr
Porosity, φ	0.33	From site reports
Retardation Factor, R	2	Estimated
Longitudinal dispersivity, α_l	x/20	Calibrated
Transverse dispersivity, $\alpha_{\rm t}$	x/50	Calibrated
Vertical dispersivity, $\alpha_{ m v}$	x/1000	Estimated
TCE decay rate in plume, λ	0.125/yr	Calibrated (equal to $t_{1/2}$ of 5.5 yrs)







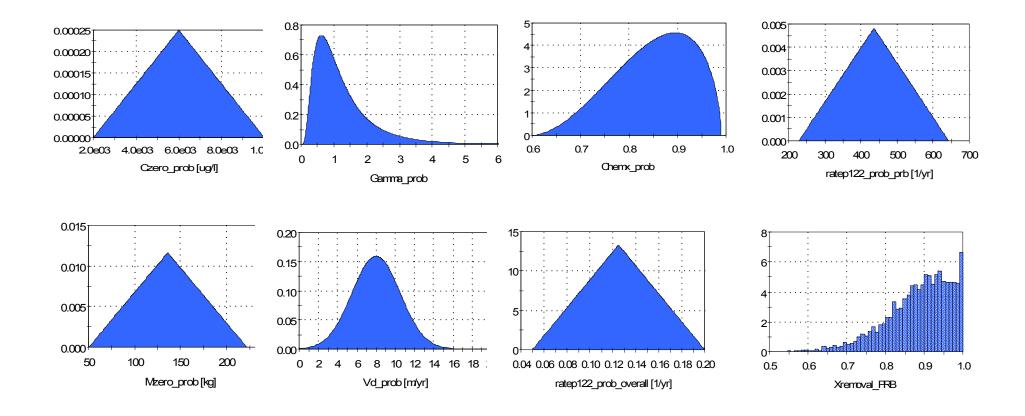
Add Probabilistic Inputs

Parameters	Distribution	Distribution Parameters	Reference
Initial source concentration (ug/l)	Triangular	most likely=6000, min=2000, max=10,000	estimated
Initial source mass (kg)	Triangular	most likely=136, min=50, max=222	estimated
Power function exponent	Log-normal	geo. Mean =1, geo stdv=2	
Darcy velocity (m/yr)	Normal	mean=8, stdv=2.5	
Overall plume natural attenuation rate for TCE (1/yr)	Triangular	most likely=0.125, min=0.05, max=0.2	
Fraction of source mass removal (%)	Beta	mean=0.85, stdv = 0.08, min=0.6, max=0.99	McGuire et al, 2006
PRB enhanced decay rate (1/yr)	Triangular	most likely=436, min=228, max=643	

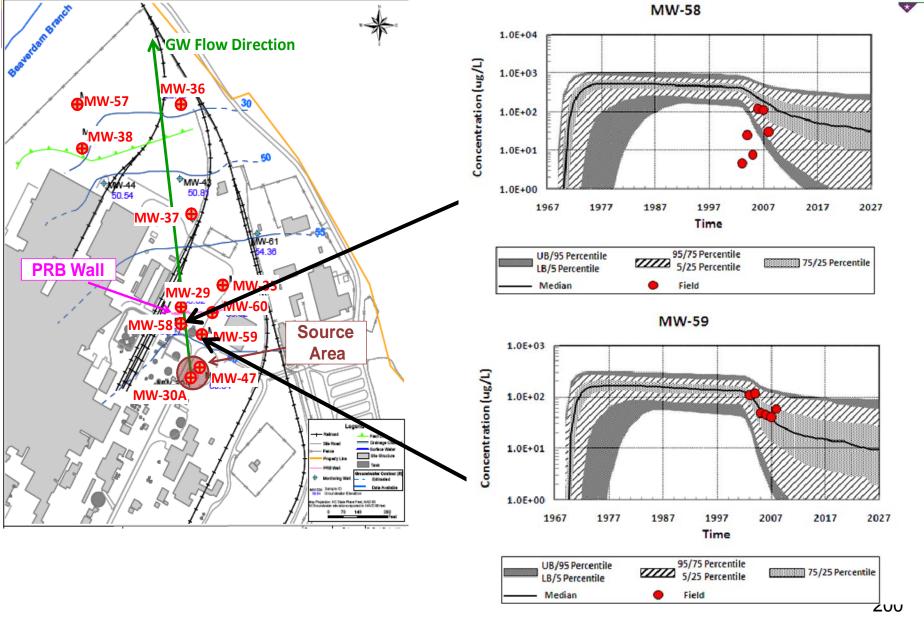


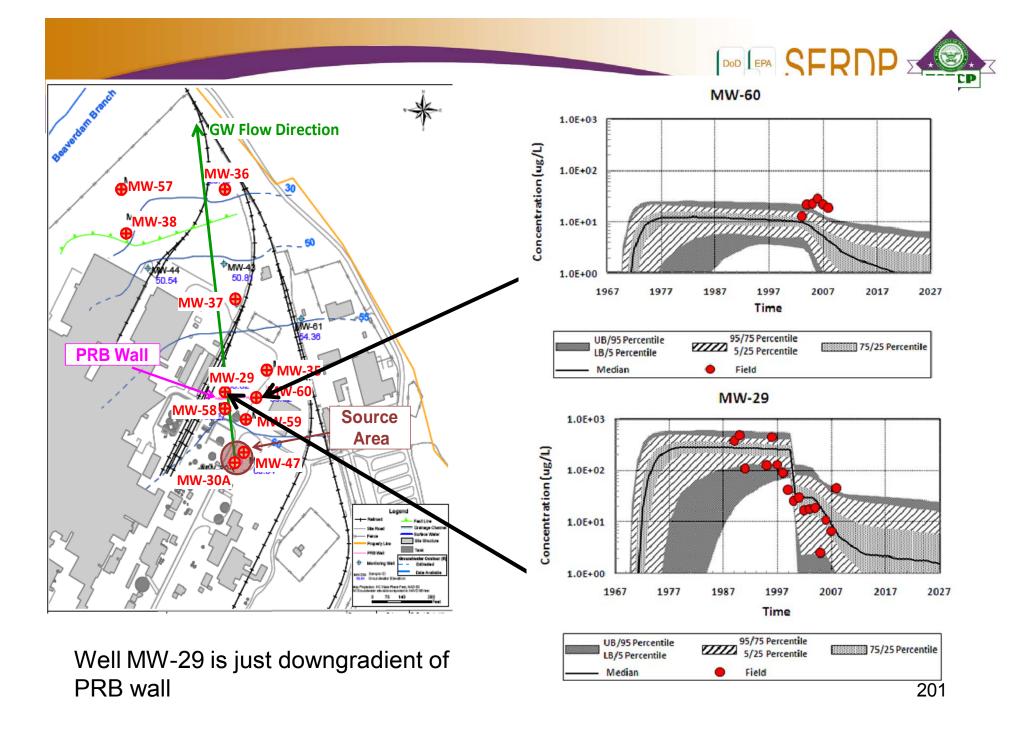


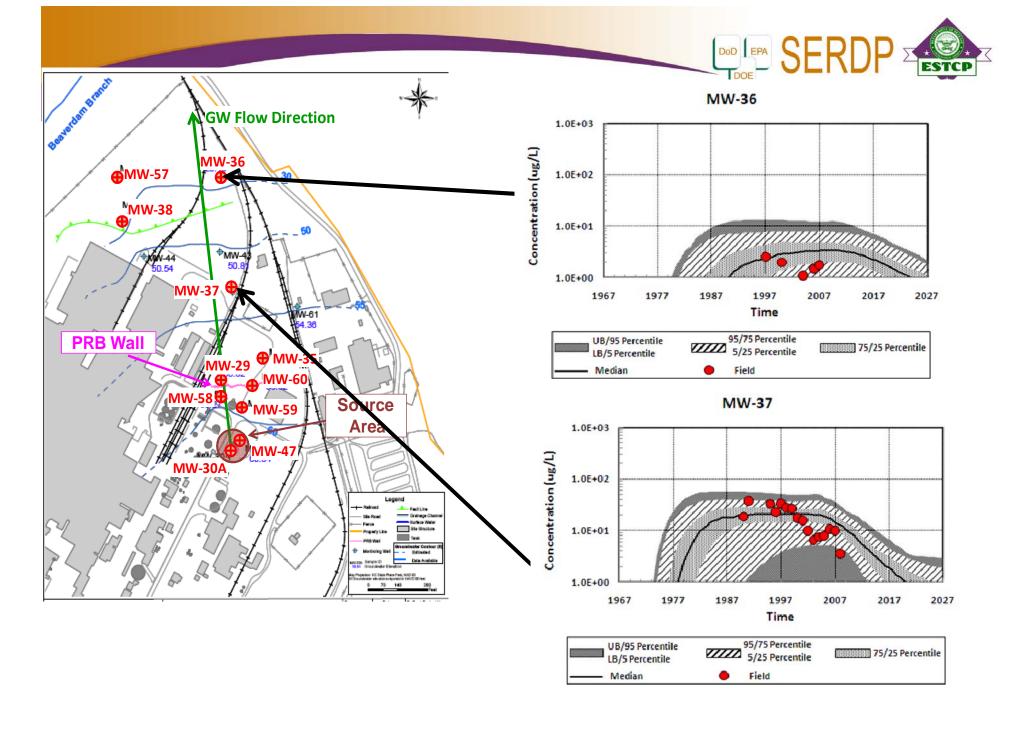
PDFs













How to get Probabilistic REMChlor

- See handout.
- Download dll and player files from Clemson FTP site;
- Download GoldSim Player executable from GoldSim site (free)



Short Course Agenda

8:30 AM	Welcome and Introduction	Hans Stroo	
8:40 AM	Source Zone Protocol for Remedy Selection of Chlorinated Solvents Released at DoD Facilities	Thomas Sale & Charles Newell	
9:50 AM	Break		
10:10 AM	Development of a Protocol and Screening Tool for Selection of DNAPL Source Area Remediation	Carmen Lebrón, Bernard Kueper, David Major, Julie Konzuk & Jason Gerhard	
11:50 AM	Lunch		
1:00 PM	Decision & Management Tools for DNAPL Sites: Optimization of Chlorinated Solvent Source and Plume Remediation Considering Uncertainty	Ronald Falta & Charles Newell	
2:20 PM	Emulsion Design Tool for Planning Aqueous Amendment Injections Systems	Robert Borden	
2:50 PM	Break		
3:10 PM	Permanganate Design Tool for Planning Aqueous Amendment Injection Systems	Robert Borden	
4:00 PM	Improved Field Evaluation of NAPL Dissolution and Source Longevity	Michael Kavanaugh, Mark Widdowson, Lloyd Stewart & Rula Deeb	
5:20 PM	Summary & Conclusion	Hans Stroo	



Planning and Design of Emulsified Oil Injection Systems



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M. Tony Lieberman



Aaron Weispfenning Matthew Clayton

NC STATE UNIVERSITY

Thomas Simpkin



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 - North Carolina State University
 (Robert Borden, Aaron Weispfenning, and Matt Clayton)
 - Solutions-IES (M. Tony Lieberman)
 - CH2M Hill (Tom Simpkin)
- Financial and technical support from ESTCP
- NCSU is not sponsoring or endorsing this presentation







Emulsified Oil Process

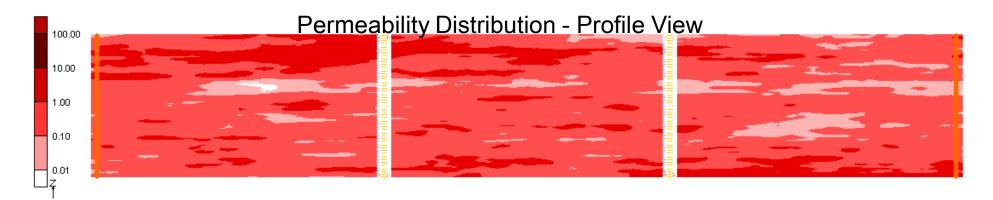
- Install temporary or permanent injection points
 - Grids or barriers
- Prepare and inject emulsion
- Inject water to distribute emulsion throughout treatment zone
- Oil droplets eventually stick to sediment surfaces
- Oil slowly ferments to H₂ and acetate
- H₂ and acetate drive anaerobic biodegradation processes
- Bioaugment if needed
- Monitor and wait

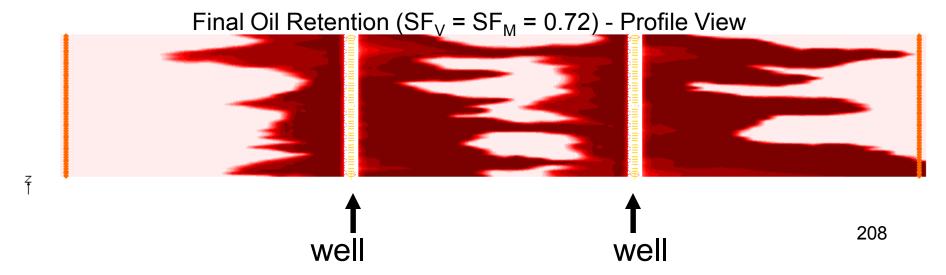


Numerical Modeling of Emulsified Oil Distribution



- MODFLOW/RT3D
- 3D heterogeneous aquifer







How to Improve Treatment?

- Good treatment requires good contact
- How to improve contact
 - Inject more emulsified oil → more \$\$\$
 - Inject more water to distribute oil → more \$\$\$
 - ◆ Install more closely spaced wells → more \$\$\$
- Problem: Which do I focus on?
- Solution: ESTCP Project ER-0626

Development of a Design Tool for Planning

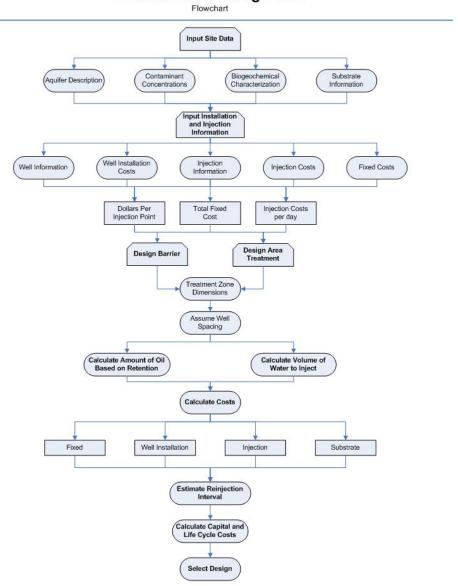
Aqueous Amendment Injection Systems

Injection System Design Tool – Injection Only

Input

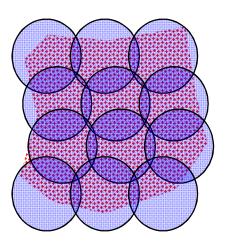
- Site Data
 - Aquifer characteristics
 - Contaminants
 - Biogeochemical data
- Costs
 - Fixed
 - Drilling
 - Substrate
 - Labor for injection
- Design Info
 - Treatment zone dimensions
 - Contact time
 - Design life
 - Scaling factors
- Output
 - Contact efficiency
 - Capital costs
 - Life cycle costs

Emulsified Oil Design Tool



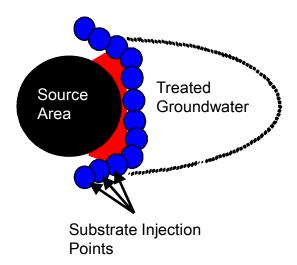
Injection System Layout





Source Area

- Grid of injection wells to 'saturate' source area
- Requires
 - More wells
 - More substrate
- Can displace contaminants
- Treatment most effective in high K zones

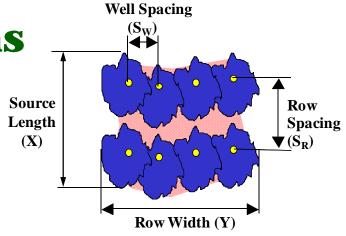


• Barrier(s)

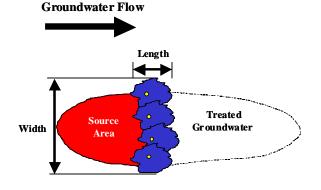
- Row of injection wells to intercept plume
- Lower cost
 - Fewer wells
 - Less substrate
- Lower potential for contaminant displacement
- Does not eliminate source

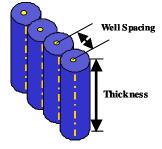
Treatment Zone Dimensions

- Width perpendicular to GW flow (Y)
 - Source width
 - Barrier width
- Length parallel to GW flow (X)
 - Source length
 - In barriers, provide enough contact time
- Vertical Thickness (Z)
 - Use effective thickness when visibly different units are present
- Wells arranged in rows perpendicular to flow
 - Wider spacing between rows to allow for downgradient drift
- S_W = spacing of wells within a row
- S_R = spacing of rows
 - Design tool has an allowable ratio of S_R to S_W = 1:1 or 2:1
- Design tool will help select optimum S_W











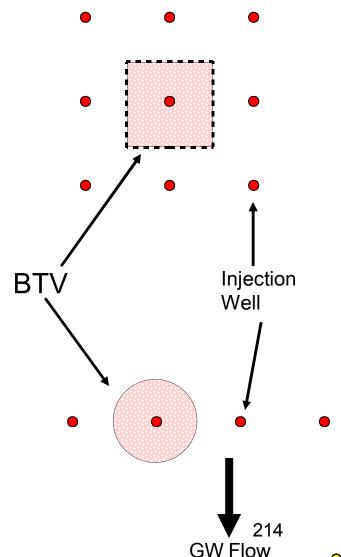
How to Design an Injection System

- Fluid Volume = BTV * n_e * SF_V BTV = Base Treatment Volume n_e = Effective Porosity (dimensionless) SF_V = Volume Scaling Factor (dimensionless) SF_V typically 0.2 to 0.6 for area treatment
- Oil Requirement = OR_M * BTV * ϱ_B * SF_M OR_M = Maximum Oil Retention (lb oil / lb soil) ϱ_B = Soil Bulk density (lb/ft³) SF_M = Mass Scaling Factor (dimensionless) SF_M typically 0.2 to 0.6 for area treatment

Base Treatment Volume (BTV)



- BTV = 'standard' volume around each well used to scale treatment quantities
- For Area Treatment
 BTV = volume of rectangular
 prism surrounding
 each well
 BTV = S_W*S_R* Z
- For Barrier Treatment BTV = volume of cylinder surrounding each well $BTV = \frac{1}{4} \pi S_W^2 * Z$





Oil Retention (OR) by Sediment

Oil retention is a function of

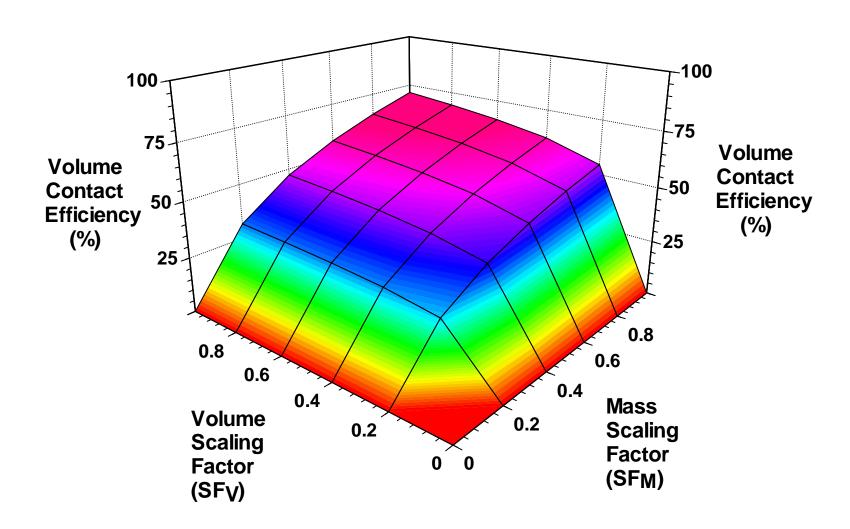
- Droplet size
 - Oil droplets should be smaller than sediment pores for easy transport
 - $\sim 1 \mu m$ easily pass through most pores (30 100 μm)
- 'Capacity' of soil to hold oil droplets
 - Silts and clays have more charged sites → hold more oil
- Surfactant type
 - Non-ionics typically have lower sorption
 - lonics have higher sorption (lecithin sorption is very high)
- Surface charge (zeta potential) of sediments and droplets
 - Most clays have a net negative charge
 - Negatively charged droplets will have lower retention



Maximum Oil Retention (OR_M)

Aquifer Material	Emulsion	Test Condition	Maximum Retention (g/g)
Blended sand (7% Silt+Clay)	Homemade	Column	0.0054
Blended sand (9% Silt+Clay)	Homemade	Column	0.0061
Blended sand (12% Silt+Clay)	Homemade	Column	0.0095
Alluvium (clayey sand)	EOS®	Column	0.0037
Low K, weathered rock (sandy clay)	EOS®	Field	0.003 (estimated)
High K, gravelly sand	EOS®	Field	0.0004 (estimated)

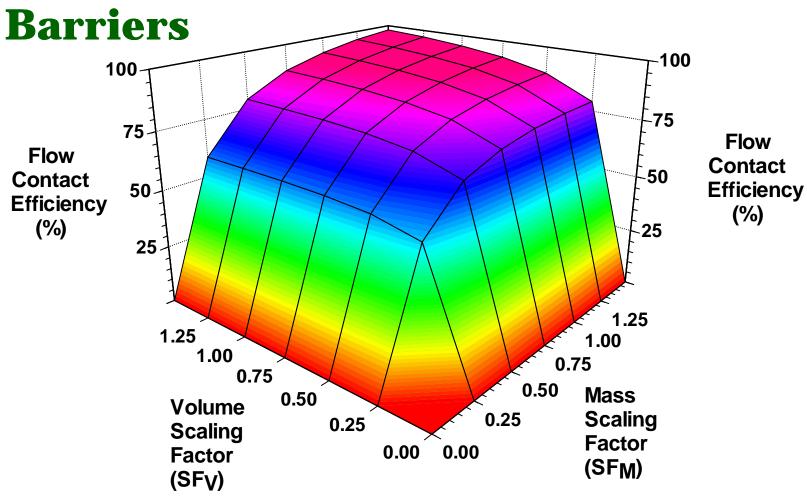
Volume Contact Efficiency for Area Treatment (Row Spacing = Well Spacing)



Clayton, M. H., and R. C. Borden, Numerical Modeling of Emulsified Oil Distribution in Heterogeneous Aquifers, Ground Water, 47(2): 246–258, 2009.



Flow Contact Efficiency for





Barrier Contact Time

- Contact time (C_t) between oil and contaminants
 - Provide 60 120 days for satisfactory chlorinated solvent removal
 - Use longer C_t for:
 - High sulfate loading
 - 'Unknown' high K layers that could cause short-circuiting through oil treated zone
 - High contaminant concentrations
 - High removal efficiency required
- Barrier length along flow direction (x) (length parallel to flow)

$$X = C_t * v$$

v = non-reactive transport velocity





How to Estimate Oil Reinjection Frequency

- Calculate oil required for biodegradation
 - Background Electron Acceptors
 - O₂, NO₃, SO₄
 - Contaminant to be treated
 - TCE, ClO₄, etc.
 - Organic carbon released to downgradient aquifer
 - Based on chemical composition of oil and microbiology
 - Typically assume average of 50-100 mg/L over project life for EOS®
 - Reduced compounds produced
 - Dissolved Fe, Mn, CH₄
- Oil Demand (D) is substrate consumed per volume of water that flows through each treatment row



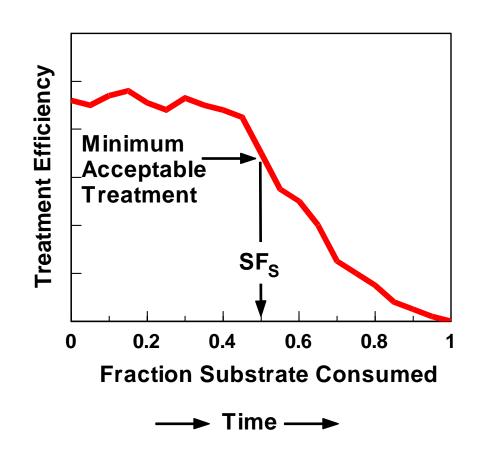


Treatment efficiency

- maximum when excess substrate is present
- will drop as substrate is consumed

Substrate Scaling Factor (SF_S)

- SF_S = fraction of substrate consumed when treatment < acceptable
- $SF_S \rightarrow time to reinject$
- SF_S typically varies from 0.3 to 0.6



Determine Injection Frequency for Barriers



- Theoretical life of single injection (T)
 - T = Oil Injected / (D * Q)
 - Water Flux (Q) = Y * Z * K * i
 - Y= Width perpendicular to flow
 - Z = Effective treatment zone thickness
 - K = Hydraulic conductivity
 - i = Hydraulic gradient
 - D = Oil Demand (mg of oil / L of water)
- Reinjection Interval (RI) = T * SF_S
 - Design tool has maximum allowable period between reinjections that will over-ride calculation



Design Tool Table of Contents

- Four sections
- Click on button to navigate to a page
- Each page has a button to go forward, backward, or back to table of contents
- Reset buttons reset all pages within a section
- Start with Aquifer Description

Emulsified Oil Design Tool

Version 33 - 2/13/2008

This tool is intended to assist engineers with the design of injection only systems for distributing emulsified oils for enhancing the anaerobic bioremediation of groundwater contaminants. More specifically, this tool allows users to evaluate the use of emulsified oils applied in barriers and area treatments. This design tool requires the user to provide all necessary information for site data and information for at least one installation and injection method. The model uses this information to evaluate the costs of various designs using different well spacings. Graphical representations of the effect of well spacing on project costs are generated. Users should have a good understanding of enhanced anaerobic bioremediation using emulsified oils before using this tool.

Table of Contents Site Data Installation and Injection Sarrier Treatmen. Area Treatment Injection Through Aquifer Description Design Information Design Information Direct Push Rods Contaminant DPT Vell Capital Cost Capital Cost Concentrations Installation Analysis Analysis: Well Installation by Biogeochemical Life Cycle Analysis Life Cycle Analysis Conventional Characterization Substrates and Installation and NPV for Selected NPV for Selected Reagants Injection Summary Design Design Reset Installation Summary of Summary of Reset Site Data Selected Design and Injection Selected Design Reset Barrier Reset Area Treatment Treatment



Aquifer Description

- Enter information in the cells outlined in red
- White cells outlined in black are for additional information and do not need to be completed

	Information on the physical characteristics and to calculate injection volumes				information	will later b
1	,					
1	Site Information					
э	Name					
b	Description (e.g., project number)					
С	Location					
		in the second				
2	Hydraulic Characteristics			_		
а	Depth to water table			ft	0.00	m
b	Depth to top of injection zone			ft	0.00	m
٥	Depth to bottom of injection zone			ft	0.00	m
d	Hydraulic Gradient			ft/ft	0	m/m
е	Hydraulic Conductivity			ft/day	0.00E+00	cm/s
f	Estimated Total Porosity					
g	Estimated Effective Porosity					
h	Seepage Velocity		#DIV/0!	ft/day	#DIV/0!	cm/s
	18 285 N		# 117701	tt/yr	#I)IV/(II	m/yr
3)	,					
3	Soil Characteristics					
а	Description of Soil Lithology					2 22
b	Bulk Density			lbs/f: ³	0.0	g/cm³
	Maximum Oil Retention by soil (see	e Table 4 2 in design				
	manual). This value has a critical i	mpact on cost and				
0	treatment performance.			lbs oil/lbs soil	0	kg oi/kg
	Return to Table of Contents	Go Back to Previous P	ane	Go Honwan	d to Next Pa	ace

Contaminant / Biogeochemical Characterization

- Enter concentrations for contaminants and background electron acceptors
- Additional
 contaminants can be
 included by
 specifying the
 concentration,
 molecular weight, and
 the electron
 equivalents per mole

common contaminants are listed below along with thin, in, and o allow the user to enter information on ad- user must enter the contaminant concentration. MW	ditional contaminan			
and the containing concentration, my	una e equivinoie.	MV	e- equiv/	e- equiv demand
	µg/L	(g/mole)	mole	(e- equiv/L)
Tetrachloroethene (PCE), C ₂ Cl ₄		155.8	8	
Trichloroethene (TCE), C ₂ HCl ₃		131.4	6	
cis-1 2-dichloroethene (c-DCE), C ₂ H ₂ Cl ₂		96.9	4	
Vinyl Chloride (VC), C₂H₂CI		62.5	2	
Carbon tetrachloride, CCl ₄		153.8	8	
Chloroform, CHCl₂		119.4	6	
sym-tetrachioroethane, C ₂ H ₂ Cl ₄	7	167.8	8	
1,1,1-Trichloroethane (TCA), CH ₂ CCl ₃		133.4	6	
1,1-Dichloroethane (DCA), CH2CHCl2		99.0	4	
Chloroethane, C₂H ₆ CI		64.9	2	
Perchlorate, CID ₄		99.4	8	
Hexavalent Chromium, Cr[VI]		52.0	3	

֓֟֝֟֝֟֟֝֟֝֟֝֟֝֟֝֟֝֟֟ ֓֓֓֓֟֞֓	ata - Biogeochemical Characterization							
	Information on the concentration of background electron acceptors is entered on this page. This information is used to calculate the number of electron equivalents (e- equiv) required to deplete these materials. The total e- equivalent is then calculated from the contaminant demand and the background electron acceptor demand. This value is later used to calculate the annual substrate demand.							
		mg/L or mg/Kg	MW (g/mole)	e- equiv/ mole	e- equiv demand (e- equiv/L)			
а	Background Dissolved Oxygen (mg/L)		32.0	4				
b	Background Nitrate (mg/L as N)		14.0	5				
С	Background Sulfate (mg/L)		96.1	8				
d	Estimated methane produced (mg/L)		16.0	8				
е	Soil Manganese Content (mg/Kg) (not used in calculation)							
f	Estimated Mn ²⁺ produced (mg/L)		54.9	2				
g	Soil Iron Content (mg/Kg) (not used in calculation)							
h	Estimated Fe ²⁺ produced (mg/L)		55.8	1				
i	pH (not used in calculation)							
j	Alkalinity (mg/L) (not used in calculation)				Total			
					e- equiv demand			
k	e- equiv demand from biogeochemical characterizeation	0.00E+00	e- equiv/L		(e- equiv/L) 0.00E+00			

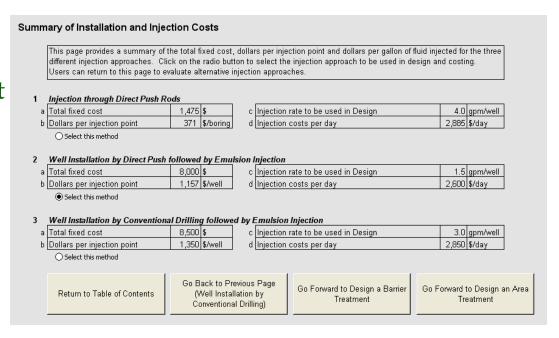
Data - Contaminant Concentrations





Well Installation Method

- Approach assumes temporary or permanent wells are installed using direct push equipment
- Multiple wells are manifolded together for emulsion injection
- Select the method on the Installation and Injection Summary page



Results of the analysis are broken into:

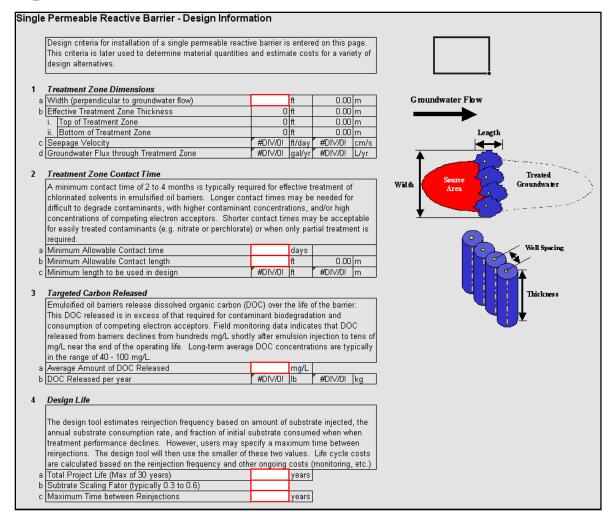
- Total fixed cost
- Dollars per injection point
- Injection rate
- Injection costs per day





Barrier Design Information

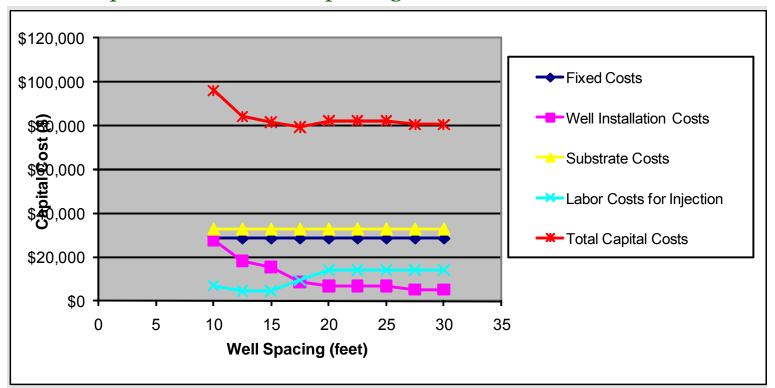
- User enters information on:
 - Treatment zone dimensions
 - Treatment zone contact time
 - Targeted carbon released
 - Design life
 - Mass, volume and substrate scaling factors
- Model calculates expected contact efficiency





Capital Cost Analysis

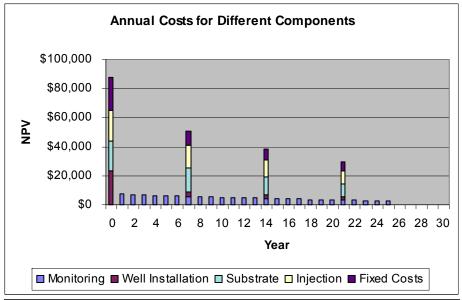
- Enter minimum well spacing and incremental increase
- Enter planning and engineering costs
- Look at capital cost vs well spacing

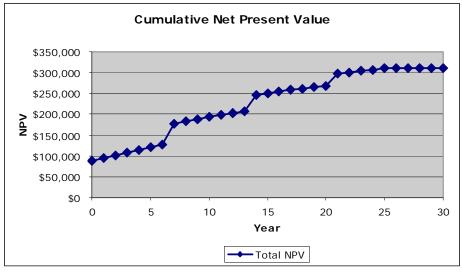




Life Cycle Cost Analysis

- Enter information on:
 - Annual interest rate
 - Engineering costs for each future event
 - Well replacement / rehabilitation for future injections
 - Annual monitoring and reporting costs
 - Results are presented as lifetime Net Present Value (NPV) vs well spacing
 - Select a design (15 ft) to see additional information





Print Out Design Summary

Area Treatment Using a Series of Barriers - Selected Design

This sheet shows a summary of the selected design that can be saved or printed before looking at alternative designs.

1 Site Information

а	Name	Example Site
b	Description (e.g., project number)	AFB
С	Location	Florida
d	Maximum Oil Retention	0.009 lbs oil/lbs soil

2 Treatment Design Criteria

а	Reinjection Interval	4	years
b	Timeframe in which all groundwater in targeted area		
	should theoretically flush through active treatment zones.	8	years

3 Well Layout

а	Well Spacing	13	ft	3.81 m	
b	Number of Wells per Row	3	wells/row		
С	Row Spacing	12.5	ft	3.81 m	
d	Number of Rows	7	rows		
е	Total Number of Wells	21	wells		

4 Logistics for Each Injection Event

а	Total Mass of Oil Injected	5,891	lbs	2,672	kg
b	Total Injection Volume	9,425	gallons	35,679	L
С	Total Injection Volume per well	449	gal/well	1,699	L/well
d	Estimated Injection Rate	1.0	gpm/well		
е	Number of wells injected simultaneously	10	wells		

5 Costs for Initial Installation and Injection

а	Fixed Costs (planning and installation)	\$28,570
b	Well Installation Costs	\$18,200
С	Injection Costs	\$7,050
d	Substrate Costs	\$32,725
е	Total Installation and Injection Costs	\$86,545

6 Costs for Future Injection Events

а	Fixed Costs (engineering and installation)	\$13,570
b	Well Rehabilitation and/or Installation Costs	\$4,550
С	Labor Cost for Injection	\$7,050
d	Substrate Costs	\$32,725
е	Total Installation and Injection Costs	\$57,895

7 Total Life Cycle Costs

а	Annual Interest Rate	5%
b	Monitoring and Reporting	\$64,632
	Total Injection Costs (fixed, well installation, labor for	
С	injection, and substrate)	\$173,361
d	Project Life NPV	\$237,993

8 Design Parameters

а	Volume Scaling Factor	0.5
b	Mass Scaling Factor	0.5
С	Estimated Contact Efficiency for Injection	40%

0	
,	



Additional Resources

- Software Download
 - http://docs.serdp-estcp.org/ (search for Design Tool)
 - http://www4.ncsu.edu/~rcborden/Design_Tool.html
- Manual
 - Emulsified Oil Design Tool USERS MANUAL
 - Tutorial included in Manual Appendix
- Websites
 - SERDP/ESTCP (www.serdp-estcp.org)
 - A Treatability Test for Evaluating the Potential Applicability of the Reductive Anaerobic Biological In Situ Treatment Technology to Remediate Chloroethenes"
 - "Protocol for Enhanced In Situ Bioremediation Using Emulsified Edible Oil"
 - AFCEE (http://www.afcee.brooks.af.mil/products/techtrans/)
 - "Principles and Practices of Enhanced Anaerobic Bioremediation of Chlorinated Solvents"
 - "Protocol for In Situ Bioremediation of Chlorinated Solvents Using Edible Oil"



Short Course Agenda

8:30 AM	Welcome and Introduction	Hans Stroo	
8:40 AM	Source Zone Protocol for Remedy Selection of Chlorinated Solvents Released at DoD Facilities	Thomas Sale & Charles Newell	
9:50 AM	Break		
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11:50 AM	Lunch		
1:00 PM	Decision & Management Tools for DNAPL Sites: Optimization of Chlorinated Solvent Source and Plume Remediation Considering Uncertainty	Ronald Falta & Charles Newell	
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Planning and Design of Permanganate Injection Systems



Bob Borden (rcborden@eos.ncsu.edu)

M. Tony Lieberman



Ki Young Cha



Thomas Simpkin







Acknowledgements

- Research conducted jointly by:
 - North Carolina State University
 - **Robert Borden**
 - Ki Young Cha
 - **Solutions-IES**
 - M. Tony Lieberman
 - **CH2M Hill**
 - Tom Simpkin
- Financial and technical support from ESTCP
- NCSU is not sponsoring or endorsing this presentation







ISCO using MnO₄

- Target Contaminants
 - Chlorinated Ethenes (PCE, TCE, DCE, VC)
 - ◆ RDX, HMX, TNT
- Not Effective for
 - Chlorinated ethanes (e.g., 1,1,1-TCA)
 - Carbon tetrachloride
 - Benzene, MTBE
- Injection Procedure
 - Install injection points
 - Prepare MnO₄ solution
 - Inject water to distribute MnO₄
 solution throughout treatment zone
- MnO₄ is consumed by
 - Natural Oxidant Demand (NOD)
 - Target contaminant





automatic KMnO₄ feed system





What is the Secret to making ISCO Work?

"Success is achieved by having enough oxidant in contact with the contaminant for a long enough period of time to react effectively"

ISCO Technology Practices Workshop Colorado School of Mines, March 2007

- Design Tool Performance Criteria
 - Reagent distributed throughout target zone
 - MnO₄ concentration > _____ mg/L after ____ days
 - Target MnO₄ Concentration ~ 100 to 1000 mg/L
 - Target contact time ~ 10 to 100 days





- Develop reaction kinetics to simulate
 MnO₄ consumption by NOD
- Implement model as:
 - RT3D
 - simple spreadsheet model (CDISCO)
- RT3D sensitivity analysis
 - 3-D heterogeneous aquifer
 - Range of injection volumes, MnO₄ loading and model parameters
- Use RT3D results to 'calibrate'
 CDISCO spreadsheet model





Modeling Approach

 Standard Advection – Dispersion Equations for Contaminant (C) and MnO₄ (M) transport

$$\frac{1}{R} \frac{\partial C}{\partial t} = \frac{\partial}{\partial x_i} \left(D_{ij} \frac{\partial C}{\partial x_j} \right) - \frac{\partial}{\partial x_i} (v_i C) - F(C, M)$$

$$\frac{\partial M}{\partial t} = \frac{\partial}{\partial x_i} \left(D_{ij} \frac{\partial M}{\partial x_j} \right) - \frac{\partial}{\partial x_i} (v_i M) - F(C, M, N_I, N_S)$$

- Reaction Kinetics
 - Instantaneous reaction between C and M
 - Instantaneous reaction between M and NOD_I
 - 2nd Order reaction between M and NOD_S (N_S)

$$\frac{dM}{dt} = -K_S M N_S \rho_B / n$$

- Equations coded into
 - RT3D reaction module
 - Spreadsheet as series of CSTRs
- Model assumes No NAPL present



RT3D Simulations

Simulate small part of large injection grid

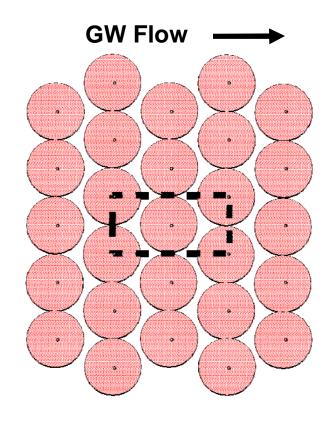
3-D Heterogeneous K distribution

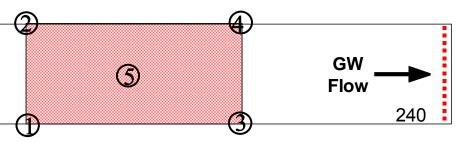
Low, medium and high heterogeneity

Vary

- Mass of MnO₄ injected
- Volume of water injected
- Well spacing
- Injection sequence
- NOD kinetic parameters

Examine contact efficiency in target zone





Typical Simulation Results for Stochastic Permeability Distribution SERDP



- 'Medium Heterogeneity'

Permeability Distribution - Plan View

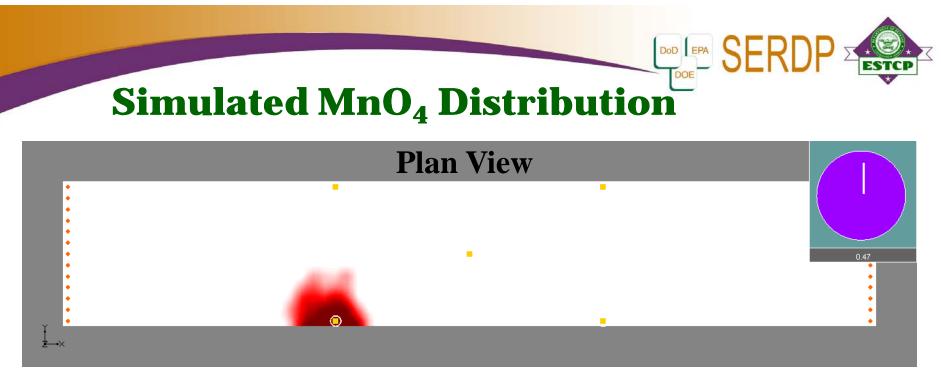


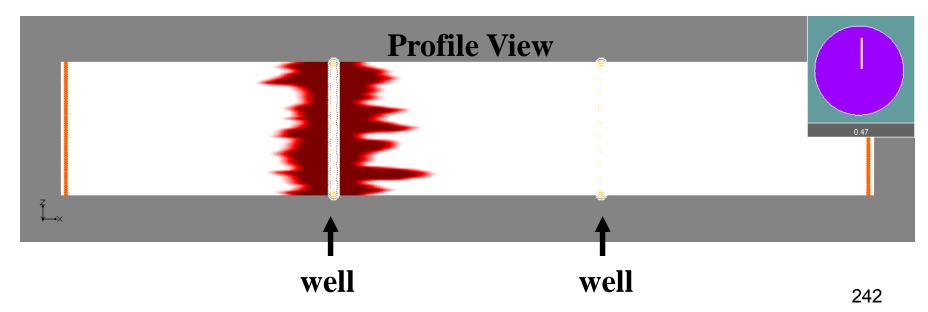
Permeability Distribution - Profile View



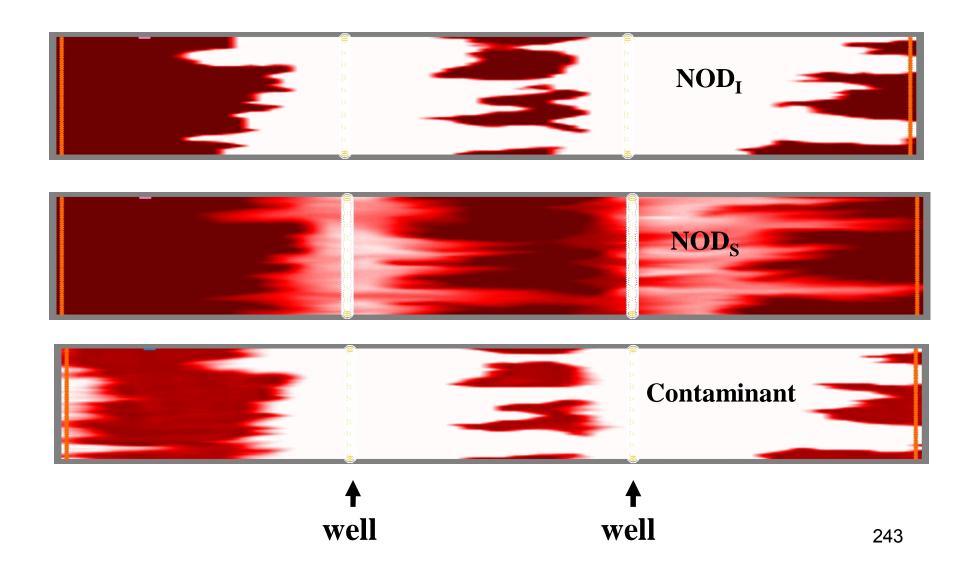








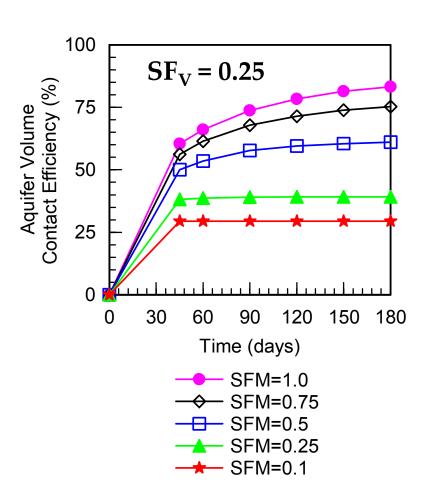
Simulation Results – Profile View – 180 Days after Injection





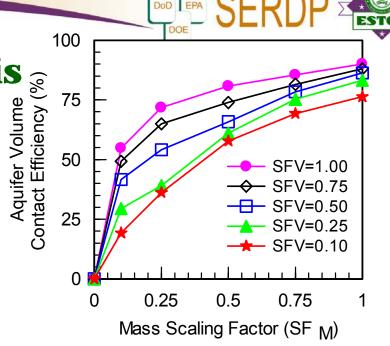
RT3D Sensitivity Analysis

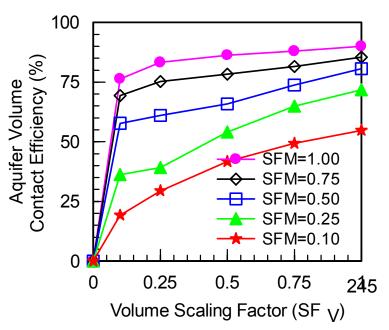
- Design Parameters
 - Mass scaling factor (SF_M)
 SF_M = MnO₄ applied / ultimate demand
 - Volume scaling factor (SF_V)
 SF_V = Volume water / pore volume
- Performance Measure
 - Aquifer Volume Contact Efficiency (E_V)
- Results
 - E_V increase with time for large SF_M
 - Downgradient drift of MnO₄
 - Diffusion into low K zones
 - E_V at 180 days will be used as primary performance measure



RT3D Sensitivity Analysis

- Effect of MnO₄ Mass Injected
 - Increasing SF_M (more MnO₄) increases contact efficiency
 - Caution: too much MnO₄ can cause downgradient release of MnO₄
- Effect Water Injection Volume
 - Increasing SF_M (more MnO₄)
 increases contact efficiency
 - For SF_M > 0.5, large injection volumes have less benefit



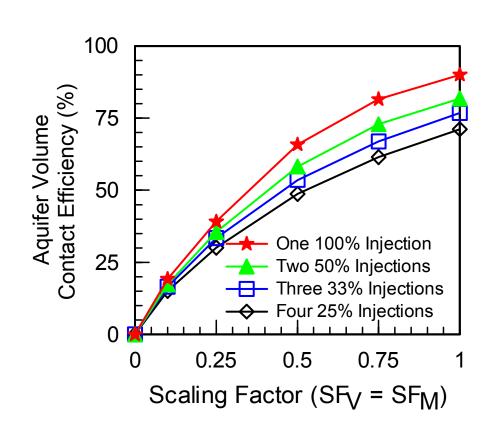






RT3D Sensitivity Analysis

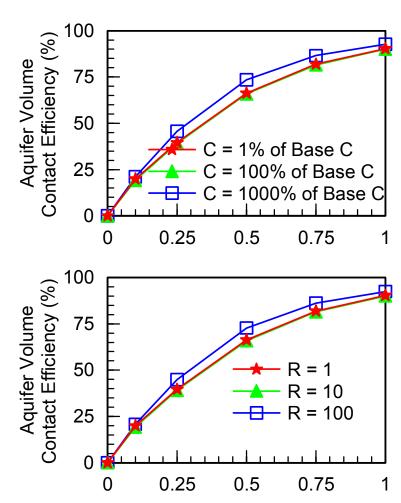
- Evaluate Impact of Multiple Injections
 - Total volume constant
 - Total MnO₄ constant
- Results
 - One large injection slightly more effective than four small injections
 - Four small injections much more effective than one small injection
 - Multiple injections has lower risk of downgradient migration





RT3D Sensitivity Analysis

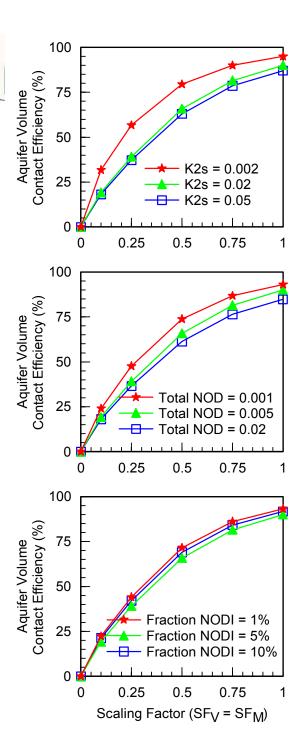
- Initial Contaminant Concentration
 - Minimal effect on E_V
 - Assumes you provide enough MnO₄
 - No NAPL in model
- Contaminant Retardation Factor
 - Minimal effect on E_V
 - Assumes you provide enough MnO₄



Scaling Factor ($SF_V = SF_M$)

RT3D Sensitivity Analysis

- NOD Kinetics
 - Slow NOD rate
 - Total NOD
 - Fraction NOD_I
- Results
 - Contact efficiency sensitive to both Total NOD and NOD kinetics
 - Cannot use simple design curves to estimate contact efficiency
 - Need 'simple' spreadsheet model for design

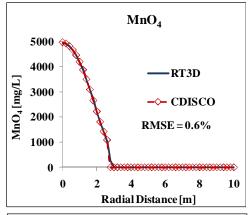


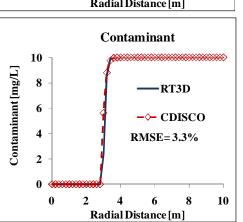


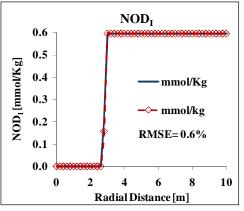


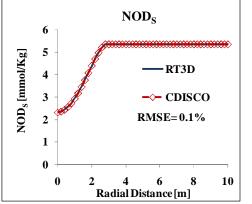
Spreadsheet Design Tool

- CDISCO –
 Conceptual Design of ISCO
 - MS Excel based Numerical Model
 - Developed jointly with ER-0623
- Mechanics
 - MnO₄ transport and consumption
 - Based on series of CSTRs
 - NOD kinetics identical to RT3D
 - Includes cost estimating tool to aid in comparing alternatives
- Model Validation
 - Results 'identical' to full RT3D for homogeneous aquifers











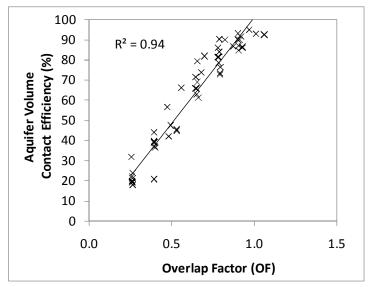
How to Design an Injection System

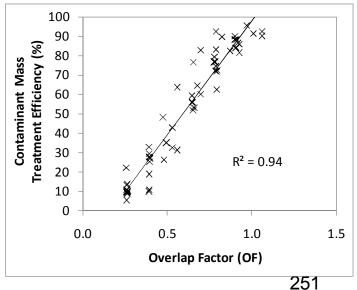
- 1. Enable Macros
- 2. Enter site data
- 3. Enter Design Criteria
 - a. Target MnO₄ concentration (typical $\sim 100 1000 \text{ mg/L}$)
 - ь. Target contact time (typical $\sim 10 100$ days)
 - c. Overlap Factor (OF)
- 4. Click 'calculate' (run MnO₄ transport model)
- 5. Enter cost data
- 6. Review cost summary
- 7. Revise design and repeat model run



Overlap Factor (OF)

- Overlap Factor (OF)
 - Well Spacing = 2*ROI / OF
 - ROI = radius of influence
- CDISCO calculates ROI
 - Minimum MnO₄ concentration after <u>days</u>
- User must pick OF
 - Currently, no guidance on correct OF
 - Increasing OF increases cost
- Comparison of RT3D and CDISCO
 - Obtain E_V and E_M from 3D heterogeneous simulations
 - Obtain ROI from CDISCO
- Conclusion
 - OF between 1.0 and 1.5 generates good results











Site Data

- Model run parameters
 - simulation duration
 - time step
- Hydrogeologic characteristics 2.
 - Permeability
 - **Porosity**
 - effective thickness
- NOD parameters 3.
 - **Total NOD**
 - Fraction instantaneous
 - Slow NOD rate coefficient
- Oxidant and contaminant info 4.
- Injection info 5.
 - Injection well diameter and design flow per well
 - Hours per day of injection and days of injection
- Design criteria 6.
 - Target oxidant concentration and contact time
 - Radius of influence overlap factor (OF)

I hadro no al cui a Chana eta vietia e		
Hydrogeologic Characteristics		
Top of Injection Interval	30	ft bgs
Bottom of Injection Interval	40.00	ft bgs
Aquifer Thickness	10	ft
Thickness of Mobile Zone (Z)	10.0000	ft
Porosity	0.20	L/L
Longitudinal Dispersivity	2.0000	ft
Hydraulic Conductivity (k)	50.00	ft/day
Depth to Water Table	15	ft
Soil and NOD Characteristics		
Bulk Density	1.60	Kg/L
NOD	1	g/Kg
Fraction Instantaneous	0.20	
Second Order Slow NOD Consumption Rate (Ks)	0.1000	L / mmol - d
Oxidants Information		
Name of Oxidant	Permanganate (MnO ₄ -)	
Molecular Weight of Oxidant	118.94	g/mol
Initial Oxidant Concentration	0.00	mg/L
	•	Z UZ

Cost Data

1. Categories

- expenses)

 Prime contractor

 (mobe, hourly labor,
- b. Subcontractor (mobe, hourly labor, expenses)
- Reagent, materials and equipment rental

2. Activities

- Fixed costs(design, permitting, etc.)
- b. Injection well or probe installation
- c. Reagent injection

Installation and Injection Costs for: Injection through Direct Push Probes

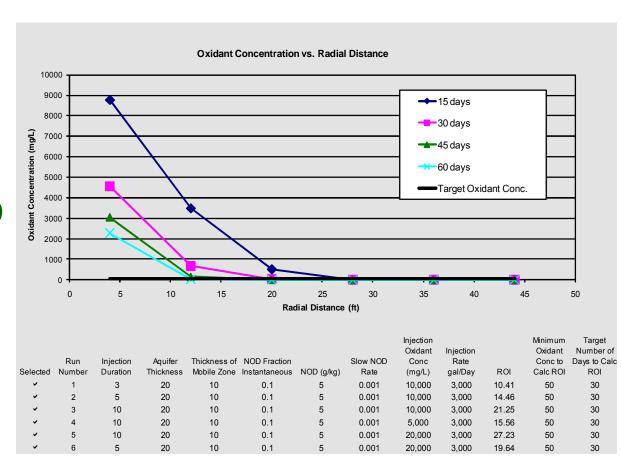
Information on the labor and materials required for ISCO injection by direct push injection (DPI) is entered on this page. Drilling and injection is assumed to be performed by a subcontract driller with supervision by the prime contractor. In this approach the oxidant is injected in a single operation where the DPI equipment drives the rod to the desired depth immediately followed by oxidant injection over an aquifer thickness equal to the injection screen length. The rod is moved to a different depth and the operation is repeated. Once injection is complete over the entire injection interval, the rod is removed, the boring grouted and the DPI equipment is shifted to a new location. DPI injections can be performed into a single probe or into multiple probes simultaneously.

1 Injection Information	•		
a Top of Injection Interval	30	ft	
b Bottom of Injection Interval	50	ft	
c Injection rate to be used in Design	3,000	gpd/probe	
d Number of probes injected simultaneously, or number of probes drilled and injected per day	5		
2 Fixed Costs	•		
a Prime contractor mobilization	500	\$	
b Subcontractor mobilization	2,000	\$	
c Water Supply	500	\$	
d Piping and other equipment for oxidant preparation and injection	2000	\$	
e Time required for equipment setup and removal	8	person - hr	
f Average labor rate for equipment setup and removal	100	\$/hr	
g Labor cost for setup and removal	800	\$	
h Total fixed cost	5,800	\$	
3 Prime Contractor Information and Daily Costs	0		
a Prime contractor personnel on-site each day of injection	1	person(s)	
b Average labor rate of prime contractor personnel	100	\$/hr	
c Hours billed per person per day	10	hr/person/day	
d Per Diem (e.g., meals, travel, vehicle rental, lodging)	200	\$/person/day	
e Additional costs (consumables, H&S, and monitoring equipment)	200	\$/day	
f Injection equipment rental costs (pumps, tanks, hoses, etc.)	200	\$/day	
9			
h Total daily cost for prime contractor	1,600		
4 Subcontractor Information and Daily Injection Costs	4,800		
a Drilling Equipment to be used			
b Daily cost for DPT equipment and operator	3000	\$/day	
f Additional material and IDW daily costs	200	\$/day	
g Total daily cost for subcontractor	3,200	\$/day	
5 Daily Costs for Injection using DPT Equipment			
a Injection costs per day	4.800	\$/day	



Permanganate Design Tool

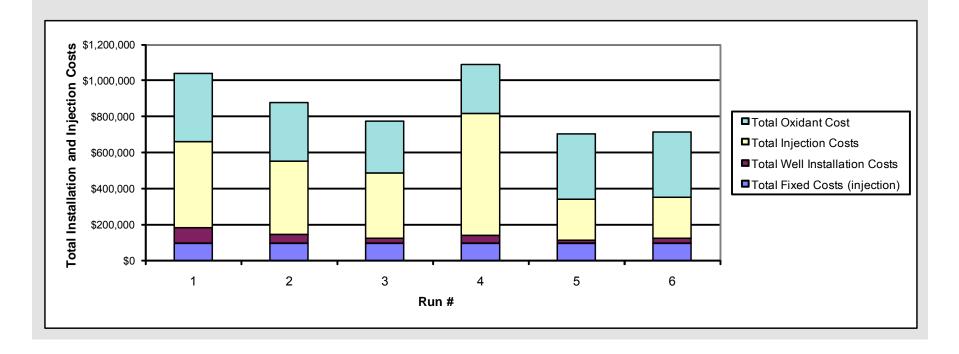
- Typical CDISCO Results
 - Generates graphs of MnO₄ conc. vs distance for different injection conditions
- Determines effective Radius of Influence (ROI) based on
 - Minimum MnO₄ Conc.
 - Contact Time
- Determines injection well spacing based on
 - ROI
 - Overlap factor







Design	Too	l Su	nma	DoD EPA DOE	SER	DP EST
Run	1	2	3	4	5	6
Total Fixed Costs (injection)	\$94,800	\$94,800	\$94,800	\$94,800	\$94,800	\$94,800
Total Well Installation Costs	\$85,667	\$47,700	\$25,367	\$41,000	\$18,667	\$29,833
Total Injection Costs	\$478,800	\$410,400	\$364,800	\$684,000	\$228,000	\$228,000
Total Oxidant Cost	\$378,547	\$324,469	\$288,417	\$270,391	\$360,521	\$360,521
Total Installation and Injection Costs	\$1,037,814	\$877,369	\$773,384	\$1,090,191	\$701,988	\$713,155
Number of probes or wells required	35	18	8	15	5	10
NOD (g/kg)	5	5	5	5	5	5
Injection Oxidant Concentration	10000	10000	10000	5000	20000	20000
Injection Oxidant Mass (lbs)	26288	22533	20029	18777	25036	25036
Injection Duration (days)	3	5	10	10	10	5
Volume Injected per Day (gal/d)	3000	3000	3000	3000	3000	3000
Thickness of Mobile/Target Thickness	0.5	0.5	0.5	0.5	0.5	0.5





Additional Resources

- Software Download
 - http://docs.serdp-estcp.org/ (search for Design Tool)
 - http://www4.ncsu.edu/~rcborden/Design_Tool.html
- Technical Report
 - Design Tool for Planning Permanganate Injection Systems
- Websites
 - SERDP/ESTCP (http://docs.serdp-estcp.org)
 - In Situ Chemical Oxidation Initiative
 - Decision Support Tools for In Situ Chemical Oxidation
 - ITRC (http://www.itrcweb.org/gd_ISCO.asp)
 Technical and Regulatory Guidance for In Situ Chemical Oxidation of
 Contaminated Soil and Groundwater, 2nd Ed.
 - USEPA,
 In-Situ Chemical Oxidation Engineering Issue
 http://www.epa.gov/ada/download/issue/600R06072.pdf



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5:20 PM	Summary & Conclusion	Hans Stroo			



Improved Field Evaluation of NAPL Dissolution and Source Longevity

Dr. Michael Kavanaugh Malcolm Pirnie, Inc.

Dr. Mark WiddowsonVirginia Tech

Dr. Rula DeebMalcolm Pirnie, Inc.

Dr. Lloyd "Bo" Stewart
Praxis Environmental
Technologies, Inc.



Project Team: ER-0833

- Malcolm Pirnie, Inc.
 - Michael Kavanaugh, Ph.D., P.E. (PI)
 - Rula Deeb, Ph.D. (Project manager)
 - Jennifer Nyman, Ph.D. (Deputy project manager)
- Praxis Environmental Technologies, Inc.
 - ◆ Lloyd "Bo" Stewart, Ph.D., P.E. (co-PI)
- Virginia Polytechnic Institute and State University
 - Mark Widdowson, Ph.D. (co-PI)
- GeoTrans, Inc.
 - Jim Mercer, Ph.D.



Acknowledgements

- Air Force (funding of the TEE pilot study)
 - Mr. Bill Lopp, AFCEE
- BEM (contractor at the site)



Decision-Making Tool for NAPL Source Zones

- Key challenge: Determining magnitude of NAPL source depletion needed to meet site Remedial Action Objectives (RAOs) at defined point of compliance.
- Technical challenges
 - Rate of LNAPL dissolution as function of time
 - Accurate prediction of transformation processes for chemicals of concern (e.g., benzene, naphthalene)
- Proposed approach
 - Field determination of pre and post remediation rates of dissolution based on field estimates of mass transfer coefficients
 - Application of SEAM3D fate and transport model combined with flow model to assess the potential effectiveness of source removal scenarios

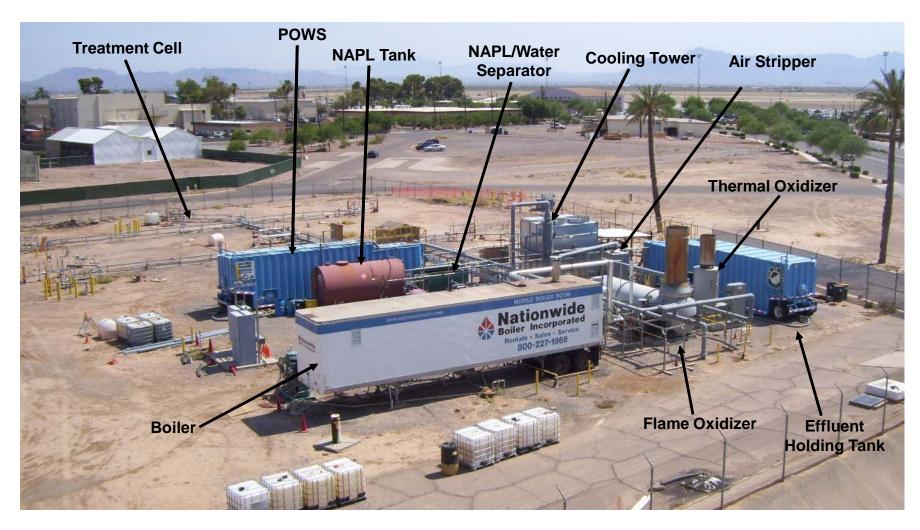


Demonstration Site

- Site ST012, former Williams AFB, AZ
- Multi-component NAPL source zone
 - JP-4 fuel, BTEX, naphthalene
- Variety of NAPL architectures
 - Extensive smear zone from rising water table; dispersed ganglia
 - Pooled NAPL below low permeability, semi-confining units
- Pilot test of Thermal Enhanced Extraction (TEE) by USAF
 - Duration: October 2008 May 2009
 - Mass transfer tests before and after TEE
 - Data interpretation and simulation of various source depletion options using SEAM3D



Demonstration Site



View facing northeast across Site ST012 (July 2, 2008)



Multi-Scale Measures of Mass Dissolution Rate

- Sweep the NAPL source zone with clean water
- Collect multi-scale data for water movement through the source zone
- Collect multi-scale data for concentrations and mass flux of target chemicals

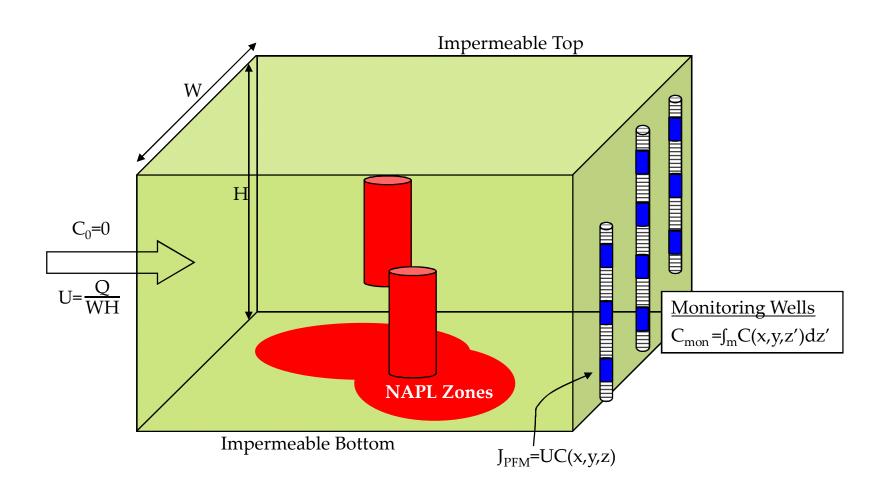


Field Technologies

- Conventional Monitoring Well Data
 - Concentration
 - Water level
 - NAPL thickness and recharge rate
- Integrated Pumping Test (IPT)
 - Modified to include water injection
 - Modified to include tracer test (e.g., bromide)
 - No new capital if Pump &Treat system is in place
- Passive Flux Meters (PFMs)
 - Vertically segmented within multiple monitoring wells



Conventional Data with PFMs Collected Downgradient





Nomenclature

W = width of flow cross-section through NAPL zone

H = height of flow cross-section through NAPL zone

Q = volumetric flow rate through cross-section

U = velocity of groundwater

 C_0 = ambient concentration entering NAPL zone

C(x,y,z) = concentration at position x,y,z

C_{ext} = concentration in extracted groundwater

 Q_{ext} = volumetric extraction rate

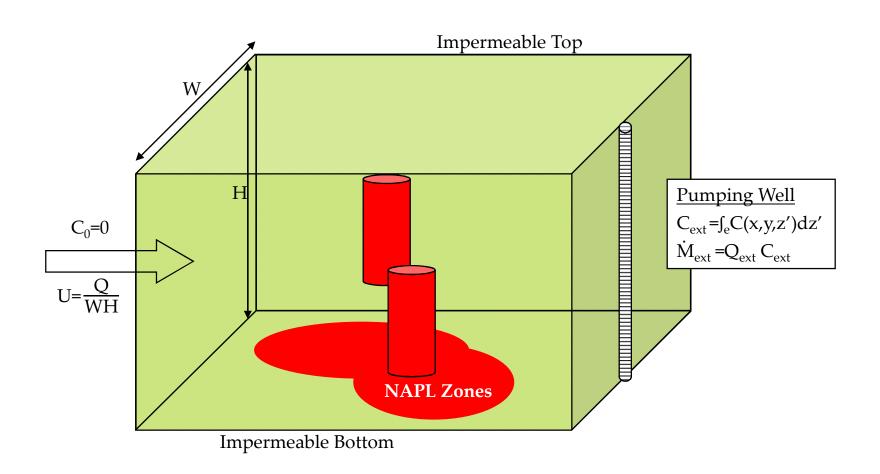
 M_{ext} = mass extraction rate of contaminant

 C_{mon} = concentration in monitoring well

 J_{PFM} = contaminant flux measured by PFM at x,y,z

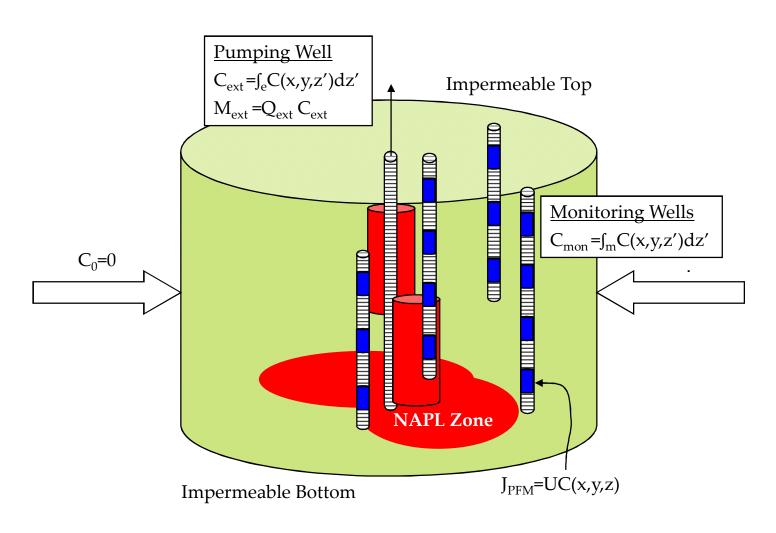


Mass Dissolution from IPT Collected Downgradient





Multi-Scale Mass Dissolution Measurements in Source Zone



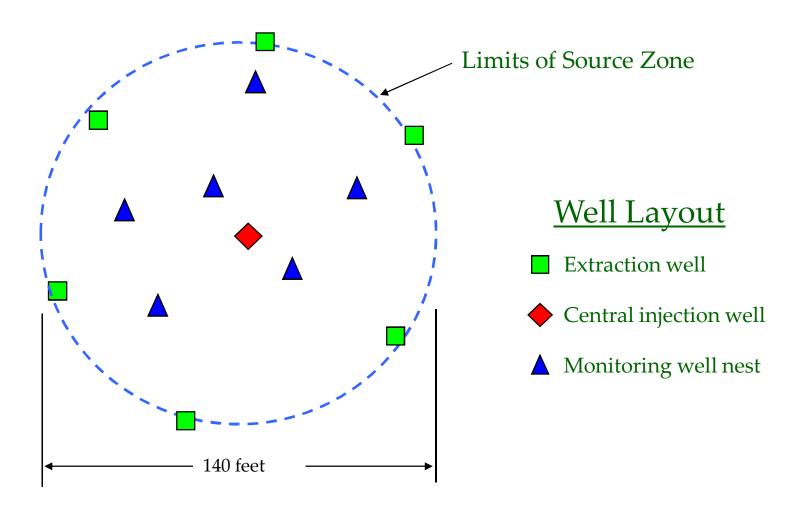


Integrated Pumping Test

- Water Injection to Sweep Source Zone
- Tracer Test to Define Flow Intervals
 - Bromide
 - Multi-Level Sensors
- Defines Mass Dissolution on a Large Scale
 - Imposed flow rate higher than ambient groundwater flow
 - Yields a maximum mass dissolution rate
 - Mass dissolution on the scale of the DNAPL source dimensions

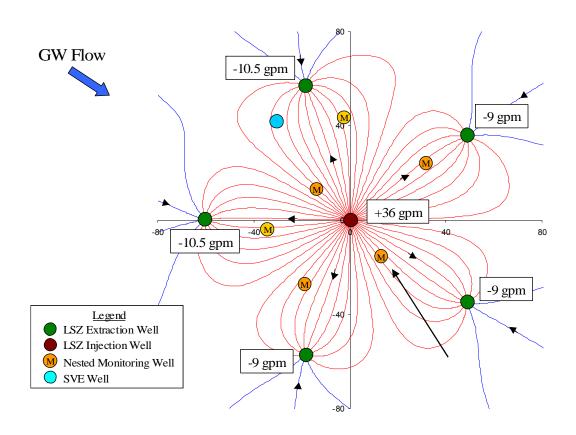


Plan View of Wells





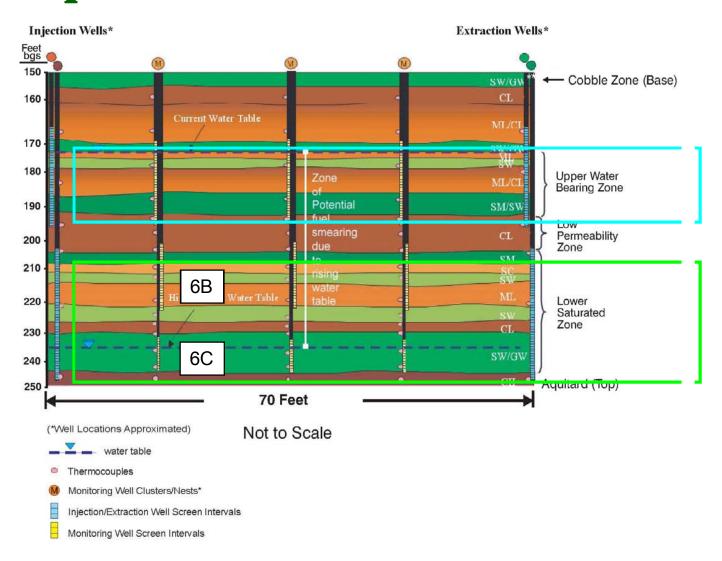
Idealized Streamlines during IPT



Streamlines Depicting Idealized Groundwater Flow During Mass Transfer Test

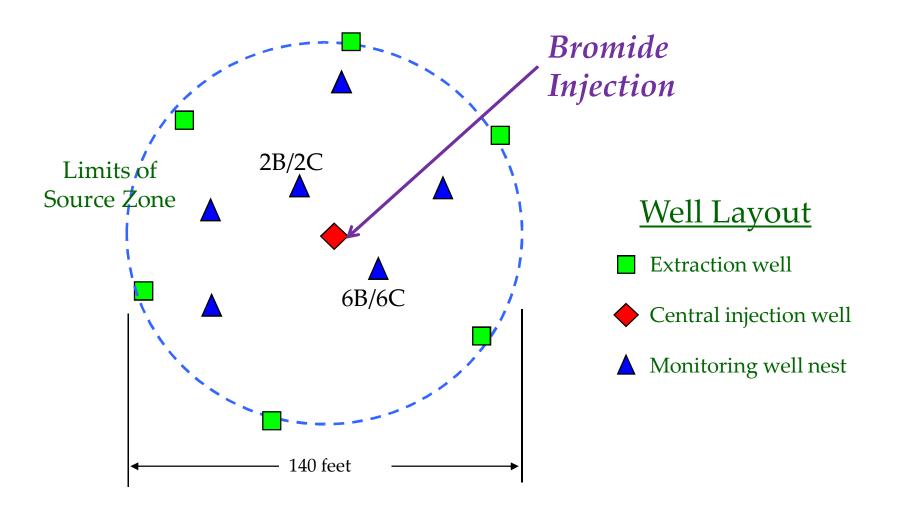


Conceptual Cross-Section



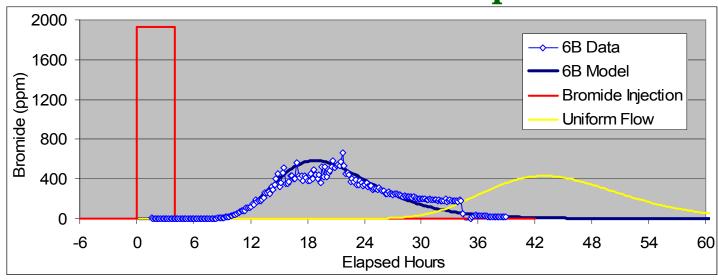


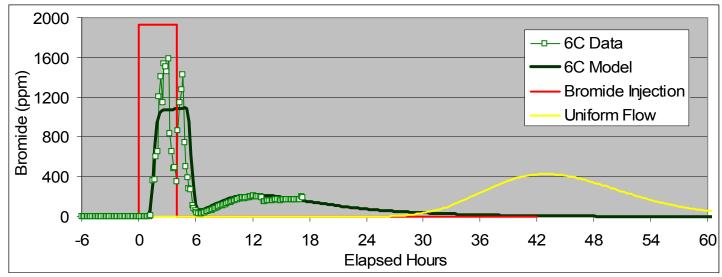
Tracer Test Layout





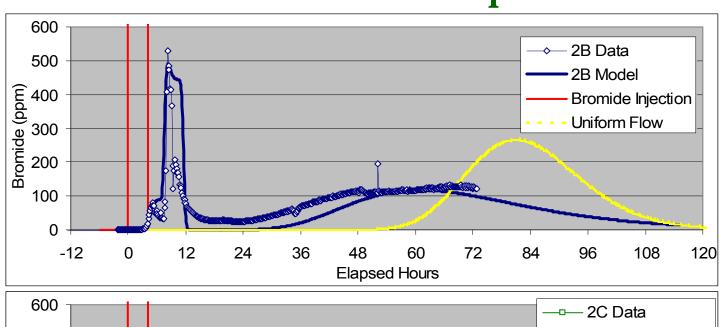
Bromide Tracer Responses

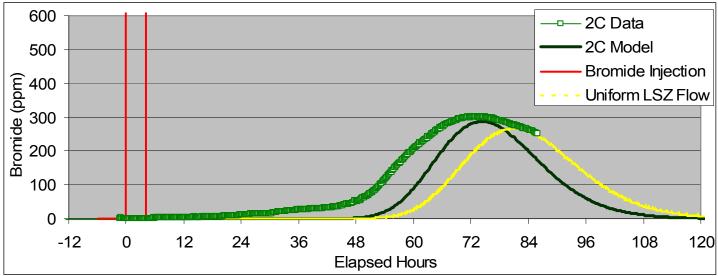






Bromide Tracer Responses







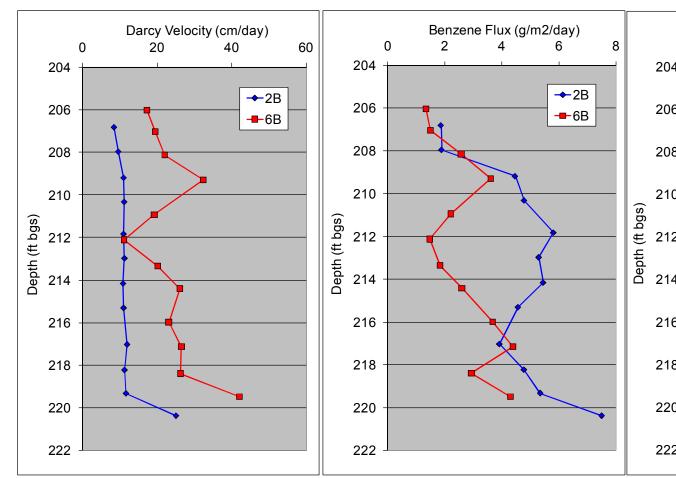
Passive Flux Meters (PFMs)

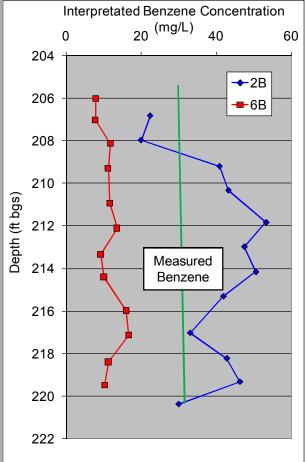
- PFMs are segmented nylon mesh tubes filled with a sorbent/tracer mixture
- Inserted into monitoring wells to passively intercept groundwater flow
- Permeable sorbent (e.g., GAC) retains dissolved contaminants
- Preloaded alcohol tracers are leached as groundwater flows through the PFM
- PFM provides vertical profiles of horizontal water and contaminant fluxes





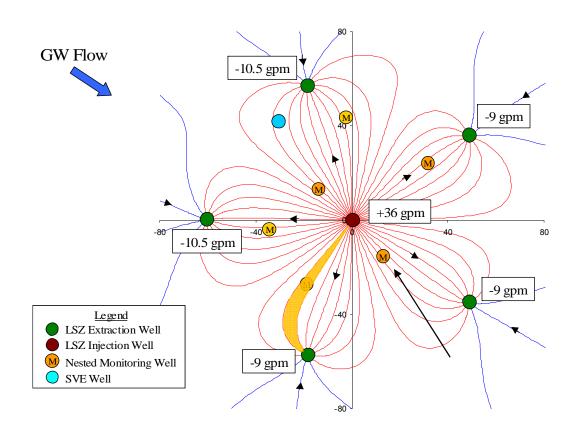
Passive Flux Meters







Idealized Streamlines during IPT



Streamlines Depicting Idealized Groundwater Flow During Mass Transfer Test



Interpretation of IPT

• Large-scale bulk mass transfer coefficient determined from the IPT (yields a maximum value):

$$M_{source,i}^{NAPL} = K_{i,IPT} C_i^{eq} V_{IPT} = Q C_{i,ext}$$

$$K_{i,IPT} = \frac{QC_{i,ext}}{C_i^{eq}V_{IPT}}$$

- M^{NAPL} = total mass extraction/dissolution rate of component i
- Q = total extraction rate
- C_i^{eq} = equilibrium aqueous concentration
- $C_{i,ext}$ = concentration of i in extracted groundwater
- $K_{i,IPT}$ = bulk mass transfer coefficient
- V_{IPT} = sweep volume



Interpretation of PFMs

• Streamtube-scale bulk mass transfer coefficient determined from the IPT and PFM:

$$K_{i,streamtube} = \frac{J_{i,PFM} A_{PFM}}{V_{streamtube} C_i^{eq}}$$

- $J_{i,PFM}$ = contaminant flux measured by the PFM
- A_{PFM} = streamtube cross-sectional area at the PFM
- V_{streamtube} = volume of soil flushed by clean water intersected by the PFM
- C_i^{eq} = equilibrium aqueous concentration
- ♦ K_{i,streamtube} = streamtube-scale bulk mass transfer coefficient



HOW DO WE USE THE MULTI-SCALE MASS DISSOLUTION MEASUREMENTS?

Modeling



Modeling Objectives

- Validate results of MMT data interpretation
 - Source zone parameters
- Predict post-TEE conditions
 - New equilibrium plume size and concentrations
- Quantify time of remediation estimates for source longevity in support of decision making
 - Remedial action work plan
 - Evaluate range of uncertainty
 - Additional mass removal scenarios



Tools and Steps

Calibrated Site Model

- •Groundwater flow (MODFLOW)
- •Solute transport and attenuation (SEAM3D)
 - ♦ NAPL source zone

Validate Source Zone Model

- Data
- ♦ Pre-TEE MTT
- ♦ Post-TEE MTT
- •SEAM3D NAPL Package
 - ♦ NAPL mass
 - Mass transfer rate

Decision-Making Tool

- •Updated site model
- •Plume and source longevity
 - Post-Tee conditions
 - Remediation timeframe



Calibrated Site Model

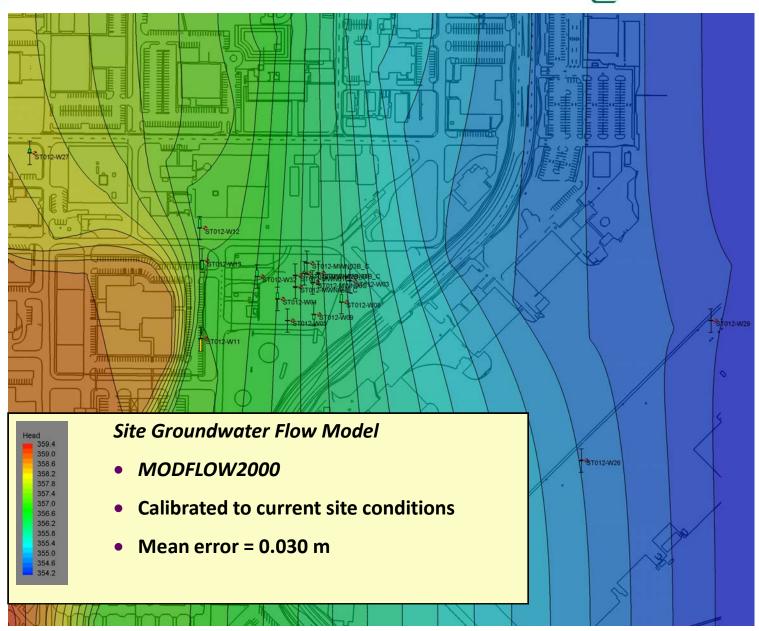
- Groundwater flow MODFLOW 2000
- Solute transport and attenuation SEAM3D
 - Physical transport
 - Biodegradation
 - Aerobic
 - Anaerobic
 - NAPL dissolution
 - Multi-component
 - Upscaled mass transport coefficient



MNA Modeling Objective

- Objective Simulate current site conditions, including historical data
 - PHC transport coupled to NAPL dissolution and aerobic/anaerobic biodegradation
- Approach
 - Construct and calibrate groundwater flow model to match observed historical water level data
 - Calibrate a solute transport model to historic PHC concentrations and TEAP/redox conditions







Model Variables

- Hydrocarbon Compounds NAPL
- Electron Acceptors (aq)
 - Oxygen
 - Nitrate
 - Sulfate
- Electron Acceptors (s)
 - Bioavailable Fe(III)
- End Products
 - Fe(II)
 - Sulfide
 - Methane



Solute Transport

Hydrocarbon Compounds: C_i

$$-\frac{\partial}{\partial x}(q_sC_i) + \frac{\partial}{\partial x}\left(\theta D\frac{\partial C_i}{\partial x}\right) + Q_sC_i^* - M_{snk,i}^{Bio} + M_{source,i}^{NAPL} = \theta R\frac{\partial C_i}{\partial t}$$

$$-\frac{\partial}{\partial x}(q_sC_i) + \frac{\partial}{\partial x}\left(\theta D\frac{\partial C_i}{\partial x}\right) + Q_sC_i^* - M_{snk,i}^{Bio} + M_{source,i}^{NAPL} = \theta R\frac{\partial C_i}{\partial t}$$

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$$-\frac{\partial}{\partial x}\left(\theta D\frac{\partial C_i}{\partial x}\right) + Q_sC_i^* - M_{snk,i}^{Bio} + M_{source,i}^{Aio} + M_{source,i}^{Aio} = \theta R\frac{\partial C_i}{\partial x}$$

$$-\frac{\partial}{\partial x}\left(\theta D\frac{\partial C_i}{\partial x}\right) + Q_sC_i^* - M_{snk,i}^{Aio} + M_{source,i}^{Aio} + M_{source,i}^{Aio} = \theta R\frac{\partial C_i}{\partial x}$$

$$-\frac{\partial}{\partial x}\left(\theta D\frac{\partial C_i}{\partial x}\right) + Q_sC_i^* - M_{snk,i}^{Aio} + M_{source,i}^{Aio} + M_{source,i}^{Aio} = \theta R\frac{\partial C_i}{\partial x}$$

$$-\frac{\partial}{\partial x}\left(\theta D\frac{\partial C_i}{\partial x}\right) + Q_sC_i^* - M_{snk,i}^{Aio} + M_{source,i}^{Aio} + M_{source,i}^{Aio} = \theta R\frac{\partial C_i}{\partial x}$$

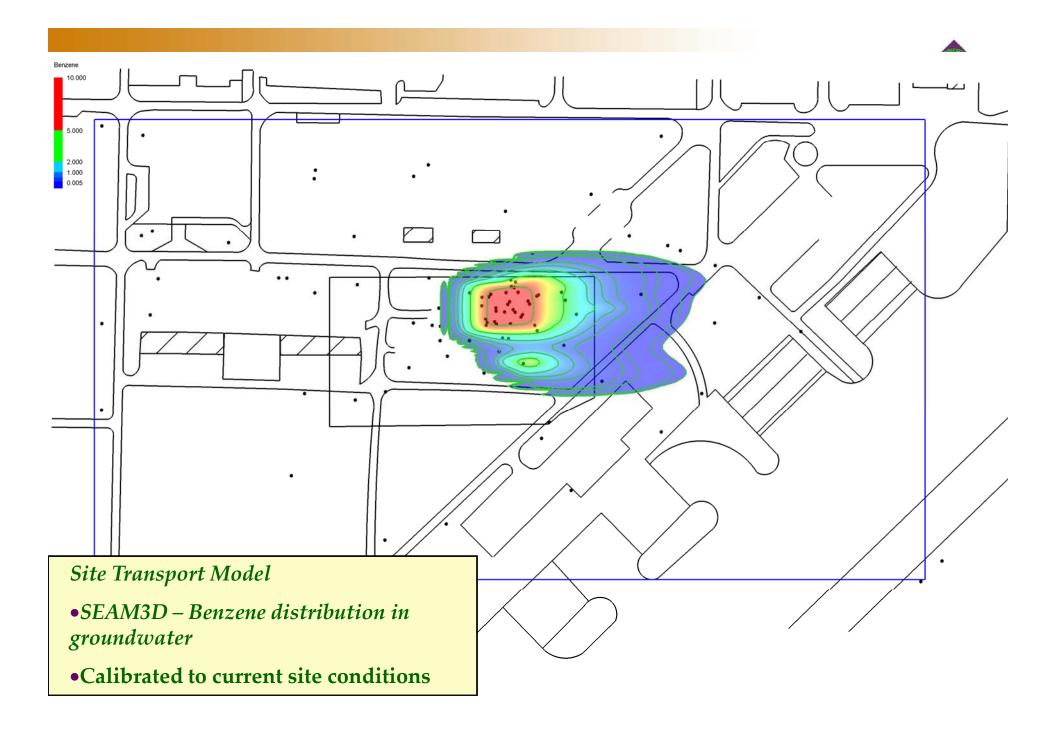
$$-\frac{\partial}{\partial x}\left(\theta D\frac{\partial C_i}{\partial x}\right) + Q_sC_i^* - M_{source,i}^{Aio} + M_{so$$

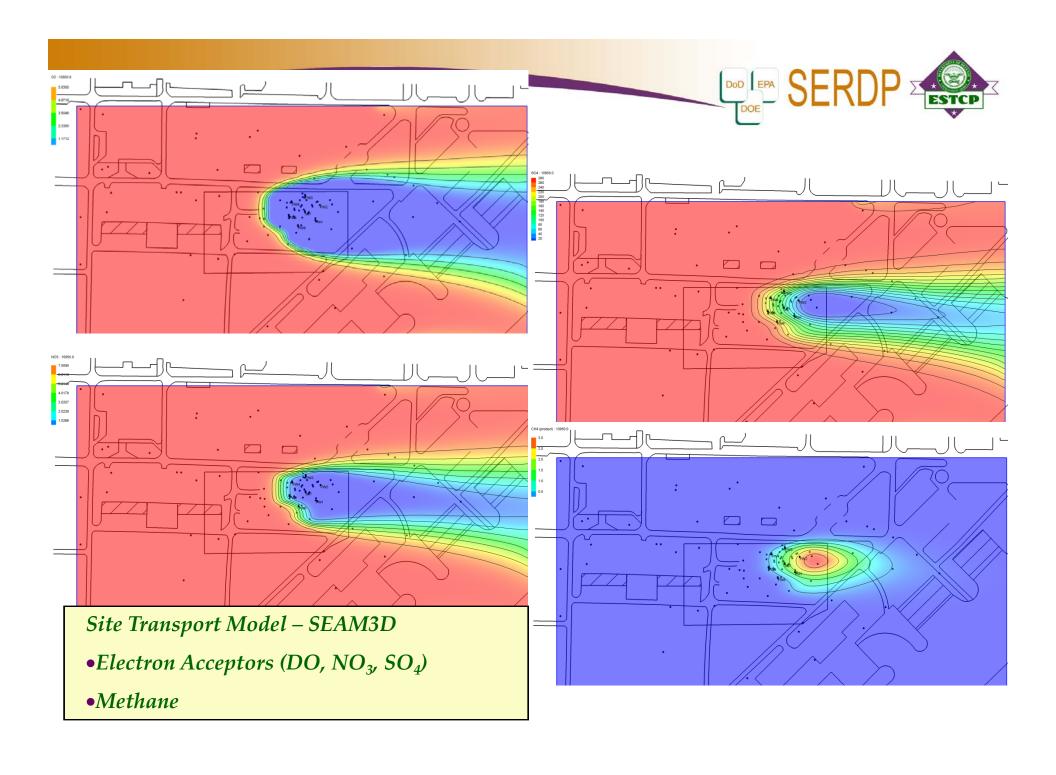


Biodegradation

- Hydrocarbon Biodegradation Sink Term
 - Sum of all applicable terminal electron-accepting processes (TEAPs)
 - Utilization rates for compound, i (TEAP-specific)

$$M_{snk,i}^{Bio} = \sum_{ea} v_{x,i,ea}^{\max} \left[\frac{C_i}{K_{x,i,ea}^{ed} + C_i} \right] \left[\frac{E_{ea}}{K_{x,le}^{ea} + E_{ea}} \right] I_{ea,li}$$

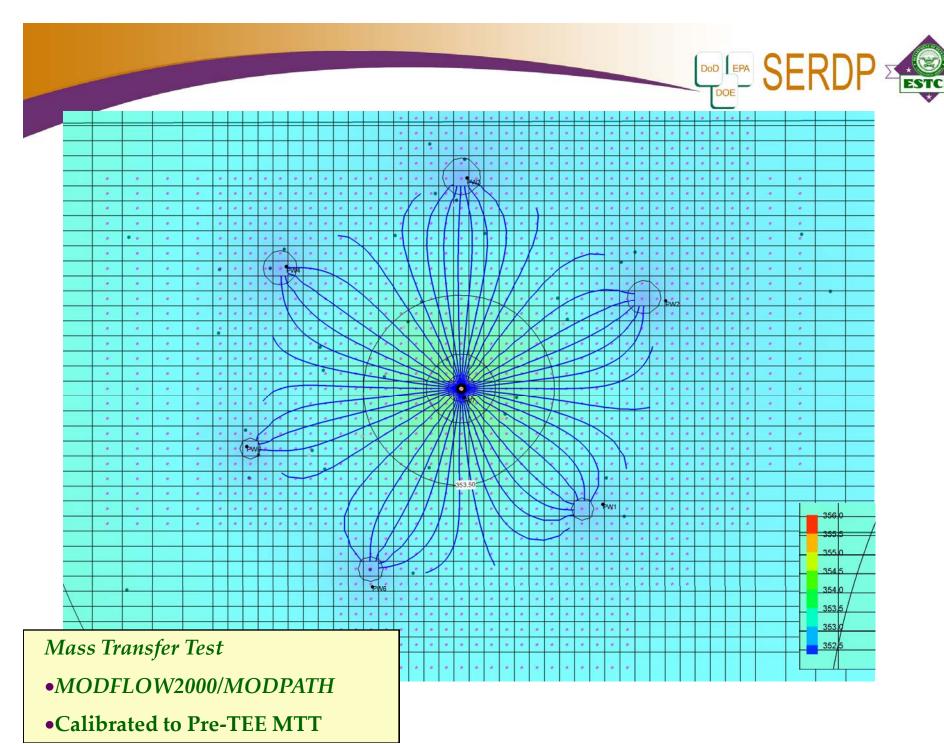






Model Validation - MTT

- Validation of the source zone model is accomplished by simulating the mass transfer test using MODFLOW and SEAM3D
- Steps
 - Improve resolution of model grid
 - Validate flow model
 - Injection/pumping data
 - Water level data
 - Refine NAPL mass estimates and mass transfer parameters
 - Estimates constrained by results of MTT data interpretation





Source Zone Model

• NAPL Dissolution – Hydrocarbon mass transfer is modeled using a first order mass transfer function:

$$M_{source,i}^{NAPL} = K(C_i^{eq} - C_i)$$

• K is a time-dependent mass transfer coefficient based on the upscaled mass transfer function

$$K(t) = k^{NAPL} \left(\frac{V}{V_o}\right)^{\Gamma}$$

- V = volume of NAPL
- \bullet k_{NAPL} = field-scale mass transfer coefficient
- G = upscaled mass transfer parameter
- C_i^{eq} = equilibrium aqueous concentration

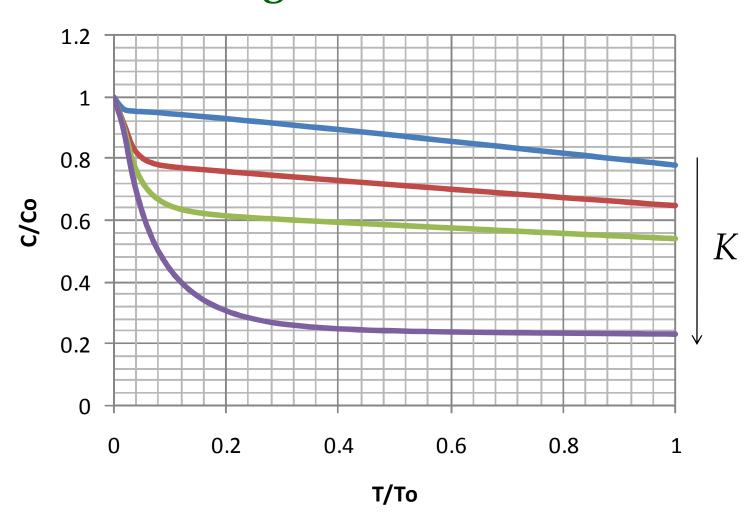


Source Model – Initial Parameter Estimates

- Calibrated STA model input parameters
- Pre-test monitoring well data contaminant concentrations
 - NAPL source components equilibrium concentrations
 - Composition of NAPL
- Results of MMT analysis
 - NAPL mass and distribution
 - Mass transfer rate

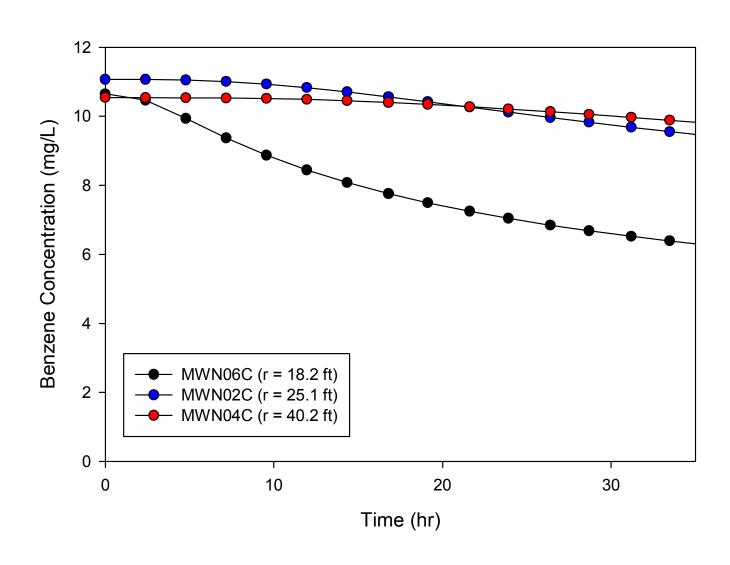


Breakthrough Curves (Pre-TEE)



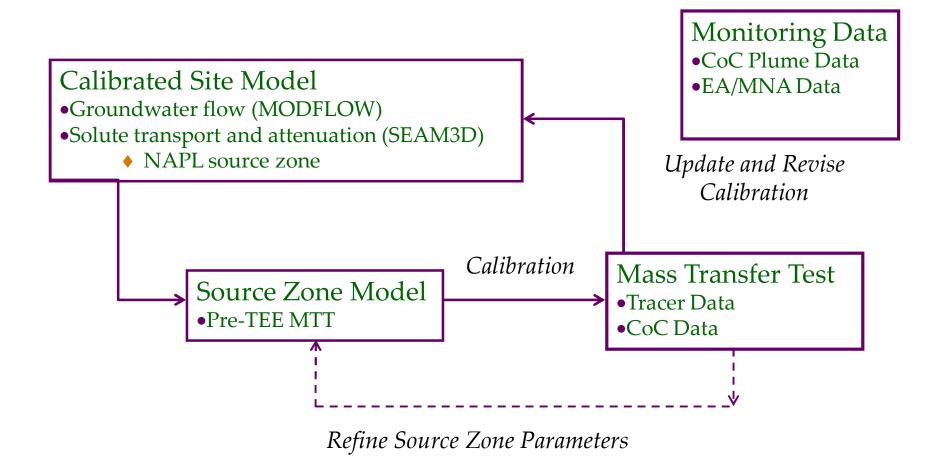


Breakthrough Curves (Pre-TEE)



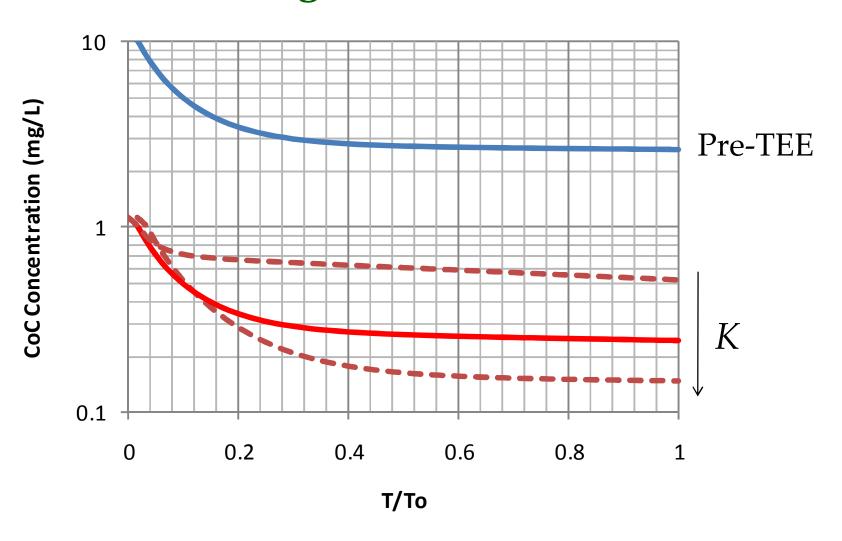


Source Model – Parameter Revision



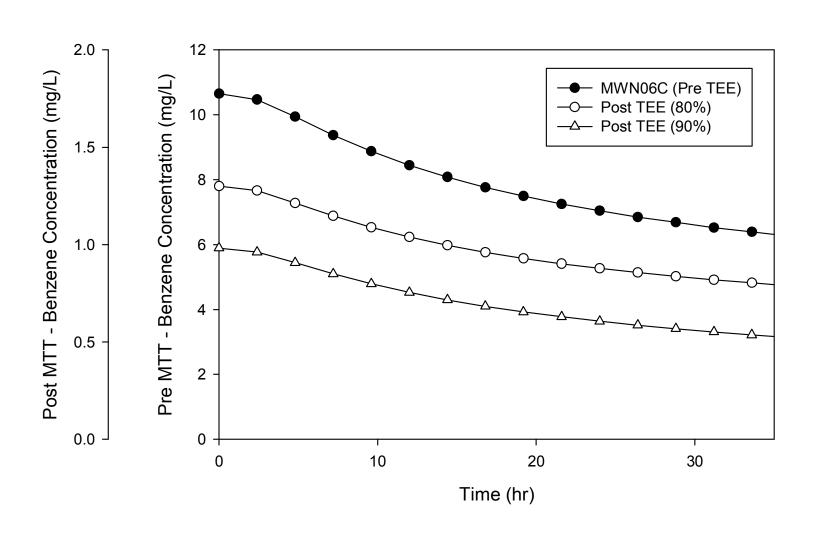


Breakthrough Curves (Post-TEE)



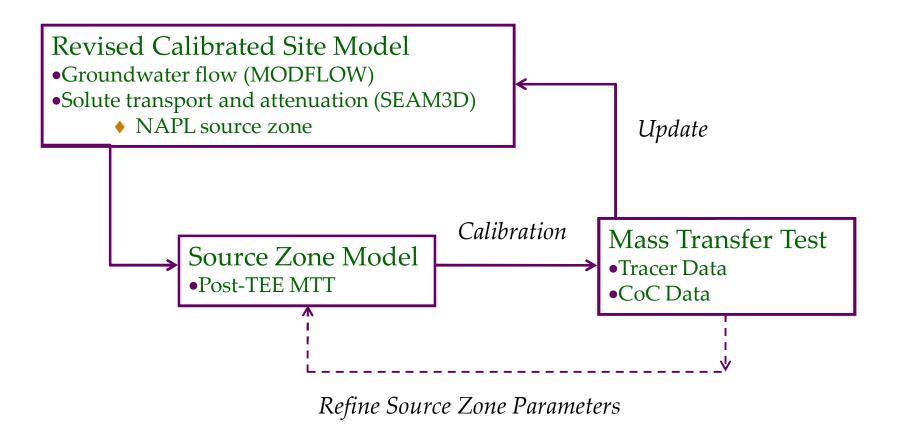


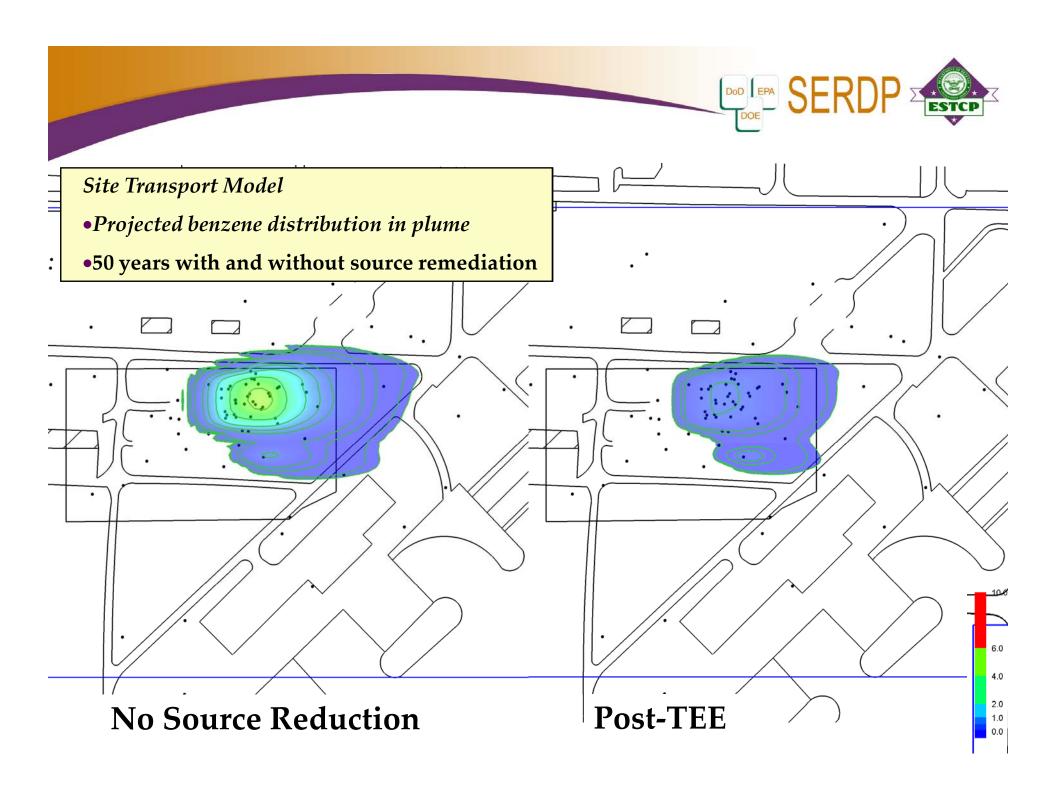
Breakthrough Curves (Post-TEE)

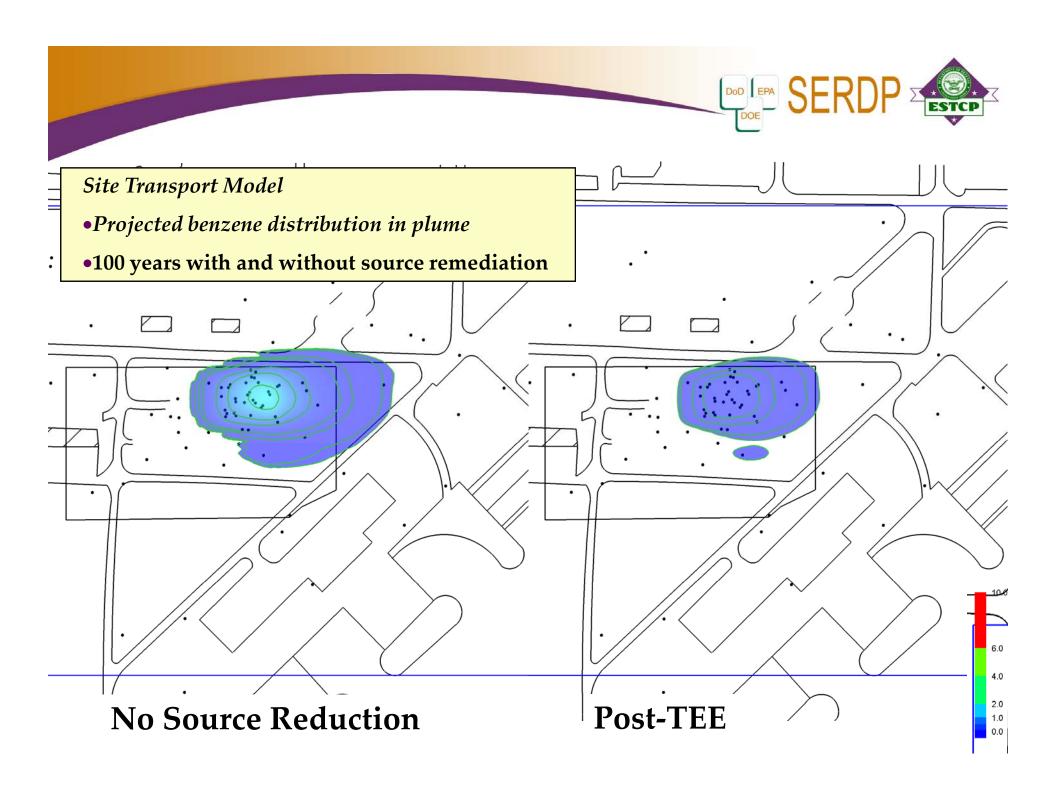




Source Model – Post Remediation









Conclusions: Decision-Making Tool for NAPL Source Zones

- Overview of tool
 - Combination of innovative field measurements and interpretation using a computational model
 - Measurement of mass dissolution rate from the source zone
 - Modeling source term to predict future mass dissolution rates and plume longevity
- Advantages
 - Testing and analytical tool for evaluating multiple scenarios for source zone reduction and plume longevity
 - Reduces uncertainty associated with remedial timeframe estimates – additional data collection constrains model input parameters that control source depletion and plume longevity



Conclusions: Decision-Making Tool for NAPL Source Zones (Continued)

Limitations

- Mass transfer coefficients may not be applicable across a site
 - Does the test accurately measure mass transfer from low permeability units?
- Model predictions are dependent on NAPL mass estimates that may vary widely within the source zone

Cost

- Application of this tool will require a monetary investment
- Cost saving may be realized by use of available test infrastructure



Short Course Agenda

8:30 AM	Welcome and Introduction	Hans Stroo
8:40 AM	Source Zone Protocol for Remedy Selection of Chlorinated Solvents Released at DoD Facilities	Thomas Sale & Charles Newell
9:50 AM	Break	
10:10 AM	Development of a Protocol and Screening Tool for Selection of DNAPL Source Area Remediation	Carmen Lebrón, Bernard Kueper, David Major, Julie Konzuk & Jason Gerhard
11:50 AM	Lunch	
1:00 PM	Decision & Management Tools for DNAPL Sites: Optimization of Chlorinated Solvent Source and Plume Remediation Considering Uncertainty	Ronald Falta & Charles Newell
2:20 PM	Emulsion Design Tool for Planning Aqueous Amendment Injections Systems	Robert Borden
2:20 PM	Break	
3:10 PM	Permanganate Design Tool for Planning Aqueous Amendment Injection Systems	Robert Borden
4:00 PM	Improved Field Evaluation of NAPL Dissolution and Source Longevity	Michael Kavanaugh, Mark Widdowson, Lloyd Stewart & Rula Deeb
5:20 PM	Summary & Conclusion	Hans Stroo