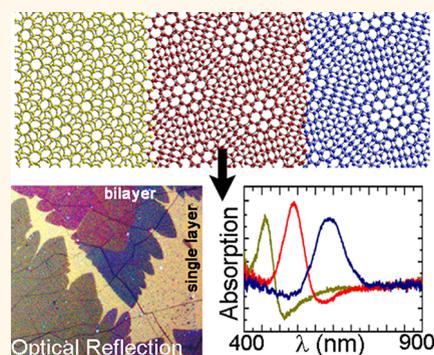


Electronic Hybridization of Large-Area Stacked Graphene Films

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ABSTRACT Direct, tunable coupling between individually assembled graphene layers is a next step toward designer two-dimensional (2D) crystal systems, with relevance for fundamental studies and technological applications. Here we describe the fabrication and characterization of large-area ($> \text{cm}^2$), coupled bilayer graphene on SiO_2/Si substrates. Stacking two graphene films leads to direct electronic interactions between layers, where the resulting film properties are determined by the local twist angle. Polycrystalline bilayer films have a “stained-glass window” appearance explained by the emergence of a narrow absorption band in the visible spectrum that depends on twist angle. Direct measurement of layer orientation *via* electron diffraction, together with Raman and optical spectroscopy, confirms the persistence of clean interfaces over large areas. Finally, we demonstrate that interlayer coupling can be reversibly turned off through chemical modification, enabling optical-based chemical detection schemes. Together, these results suggest that 2D crystals can be individually assembled to form electronically coupled systems suitable for large-scale applications.



KEYWORDS: twisted bilayer graphene · interlayer coupling · absorption · functionalization

The discovery and development of graphene^{1,2} has drawn attention to a class of materials broadly defined as two-dimensional (2D) crystals.³ As the field matures in understanding and exploiting single 2D crystals, new research is emerging to build up materials from individual layers forming “designer” thin films.^{4–7} While simple material combinations may be realized through direct growth,^{8,9} many others will require the physical stacking of individual layers. However, the top-down synthesis of multilayer structures is often plagued by surface contamination, which limits direct van der Waals contact between layers.^{10,11} In the simplest case—the stacking of two graphene layers—new properties arise when clean interfaces are realized.^{12–14} It is already clear that bilayer graphene is a diverse material system because the film properties vary as a function of relative orientation of one layer to the other. Such twisted graphene systems,¹⁵ whose theoretical^{16–18} and experimental^{11–14,19–22} properties are now being intensely considered, will guide our development of other synthetic van der Waals films.

In this work, we describe the top-down synthesis and characterization of coupled

bilayer graphene films that exhibit intimate contact over macroscopic areas ($> \text{cm}^2$). The observation of widespread “colored” domains in these stacked films reveals that coupling between layers results in new properties not intrinsic to the individual components, an effect not observed previously in large-area artificial bilayer films on SiO_2/Si substrates. To determine if these colored domains are an intrinsic feature of the film, we use Raman spectroscopy, which has proven to be useful in quantifying various degrees of interaction between graphene layers.^{11,13,19,23,24} Interlayer hybridization in twisted bilayer graphene (TBG) results in measurable changes in the intensity, position, and shape of the characteristic G ($\sim 1600 \text{ cm}^{-1}$) and 2D ($\sim 2700 \text{ cm}^{-1}$) Raman peaks; as such, a strong one-to-one correlation between Raman intensity and twist angle has been established.^{11,13} In addition to changes in the Raman response, the hybridization of the Dirac cones in TBG²¹ results in changes of the optical conductivity, including the emergence of an absorption peak^{11,20} in the relatively wavelength-independent spectrum of single-layer or Bernal bilayer graphene.^{25,26} In this work,

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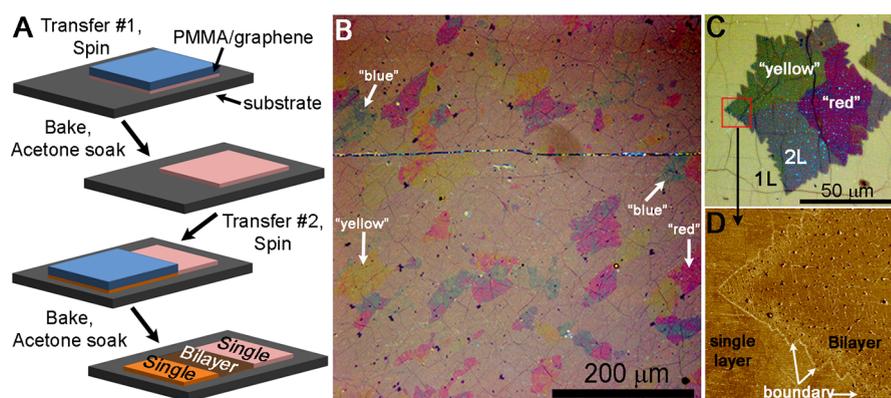


Figure 1. Fabrication and characterization of two stacked graphene films. (A) Schematic of the process flow used in forming bilayer graphene from single-layer CVD graphene films. (B) Optical microscope image of a TBG film. The “red”, “yellow”, and “blue” domains are labeled. (C) Sample in which the top (second) graphene layer is only a partial layer, forming single-layer (1L) and bilayer (2L) regions. (D) AFM phase image ($15 \times 15 \mu\text{m}^2$) showing the boundary between the single-layer and bilayer region highlighted by the red box in (C).

optical spectroscopy reveals the emergence of an adsorption peak in the visible spectrum for TBG domains with twist angles between ~ 10 and 16° , which we have independently confirmed using low-energy electron diffraction. Analogous to the rapid development of graphene research provided by simply “visualizing” graphene,^{27,28} the visualization of specific twist orientations with an optical microscope should further enable the rapid study of TBG systems. As such, we use optical microscopy to demonstrate that interlayer coupling can be effectively switched “on” and “off” through chemical functionalization of the top surface.

Graphene films were grown *via* low-pressure chemical vapor deposition (CVD) in Cu foil enclosures²⁹ and transferred onto $\text{SiO}_2(100 \text{ nm})/\text{Si}$ substrates using conventional wet-chemistry techniques,^{30–32} including the use of a poly(methyl methacrylate) (PMMA) protective layer³³ and APS copper etchant 100 (Transene Company, Inc.). Before etching Cu, we oxygenate the Transene etchant, which we find results in fewer carbonaceous residues after PMMA removal. Immediately after transferring the PMMA/graphene film from a H_2O bath, the substrate is spun at 2000 rpm then 4000 rpm to remove bulk water (Figure 1A). The sample is subsequently heated on a hot plate ($T = 85^\circ\text{C}$, 3 min; $T = 150^\circ\text{C}$, 20 min) and then submerged in an acetone bath, rinsed in acetone and isopropyl alcohol, and dried with N_2 . Bilayer samples are generated by sequentially transferring a second CVD graphene layer onto the graphene/ SiO_2/Si substrate and repeating the steps described above. The polycrystalline nature of the starting CVD graphene results in bilayers with regions of varying twist angle and is referred to here as twisted bilayer graphene.

RESULTS AND DISCUSSION

Optical inspection of as-fabricated TBG samples reveals expected features such as wrinkles and folds,^{34,35} together with an unexpected patchwork of colored

domains that vary in size and shape across the sample. Figure 1B shows an optical microscope image (Olympus DP25 CCD camera) of a TBG film with distinct regions that appear red, yellow, and blue. No filters are necessary to see these colored features (see Supporting Information). Notably, subsequent thermal annealing (e.g., 400°C in Ar/H_2)³⁶ or additional solvent cleaning does not change the extent of these colored regions, suggesting that their presence is not related to extrinsic processing residues. Figure 1C shows a higher resolution image of a different TBG sample in which the second (top) graphene layer is noncontinuous. Here we find that the colored domains are only observed in the bilayer regions and not the neighboring single-layer regions. In addition to these macroscopic colored features, small isolated islands are observed (bluish “dots” Figure 1C; discussed below), together with the wrinkles/folds mentioned earlier.

Atomic force microscope (AFM) imaging at the boundary between a bilayer and single-layer region shows an increased level of roughness in the bilayer region (Figure 1E). This increased roughness is in the form of small isolated islands that have typical footprints of less than $1 \mu\text{m}^2$ and height of ~ 10 – 40 nm and are separated by relatively smooth areas. Similar islands have been observed elsewhere when multiple graphene layers are individually assembled,^{10,13} where the largest islands align well with the bluish dots observed optically (Figure 1C). Cross-sectional imaging of stacked graphene layers confirms the presence of locally trapped amorphous material in such samples, separated by pristine regions in direct van der Waals contact.¹⁰ The ability of trapped (interlayer) hydrocarbons/adsorbates to segregate over micrometer-scale distances gives rise to these isolated islands.¹⁰ We note that using a similar sample fabrication procedure as described above, we have measured coupling between atomically flat epitaxial graphene on SiC and transferred CVD graphene films,^{21,37}

supporting the presence of clean interface regions between such isolated islands.

The direct correlation between the Raman spectral response and layer orientation in TBG films^{11,13} provides a rapid means to determine if interlayer coupling is present, as well as to coarsely identify twist angle in coupled bilayers. Arguably the most dramatic feature in TBG Raman spectra is a substantial enhancement ($>20\times$) in the G peak intensity when the laser photon energy (E_{photon}) is approximately equal to the energy level where the Dirac cones overlap/hybridize. Because the position of band overlap is dependent on twist angle (θ), the orientation at which this maximum Raman enhancement occurs is dubbed the critical angle (θ_c), with a critical energy E_{θ_c} . Conveniently, this G peak enhancement occurs in a somewhat narrow range—when θ is within a few degrees (about $\pm 2^\circ$) of θ_c or when E_{photon} is within a few hundred meV of E_{θ_c} . When outside of this window, the Raman spectra can only indicate if $\theta < \theta_c$ or $\theta > \theta_c$.^{11,13}

Unlike two noninteracting stacked graphene sheets, the Raman spectra for our TBG samples show a rich variation in peak intensities and shapes as would be expected for two interacting layers with distinct twist angles. Figure 2 shows Raman maps and spectra from TBG films in Figure 1, which contain red, yellow, and blue domains. Figure 2A,B shows a map of the G/2D peak ratio (integrated intensity) for the TBG film in Figure 1C, measured at two different wavelengths (488 and 532 nm). Close inspection of the map reveals six unique intensity ratios in the TBG region (labeled). Notably, the colored domains in the optical image align very well with the domains imaged by Raman. This result, together with many other Raman/optical image comparisons (not shown), provides strong support for the colored domains being related to an intrinsic property of the film and being uniquely dependent on the twist angle between the graphene layers.

Individual Raman spectra (Figure 2C,D) from the samples shown in Figure 1B,C can be understood in the frameworks recently presented by Havener *et al.*¹¹ and Kim *et al.*¹³ The use of two different photon energies allows us to order the spectra in terms of increasing twist angle through analyzing the G/2D peak ratios and peak positions. A strong G peak enhancement occurs at the yellow domain using $E_{\text{photon}} = 2.54$ eV (488 nm) and at the red domain using $E_{\text{photon}} = 2.33$ eV (532 nm). If we assume E_{photon} here is equal to E_{θ_c} , this corresponds to $\theta_{\text{yellow}} \approx 14.5^\circ$ and $\theta_{\text{red}} \approx 12.5^\circ$.¹¹ Comparing histograms of the normalized integrated G peak intensity (A_G) for many spectra within these same regions sheds light on the closeness of E_{photon} to E_{θ_c} , as well as on the angular distribution or deviation in θ (Figure 2E,F). For example, at the resonance condition ($E_{\text{photon}} = E_{\theta_c}$) for $\theta_c = 12.5^\circ$, the intensity of A_G decreases by 50% within approximately $\pm 1^\circ$

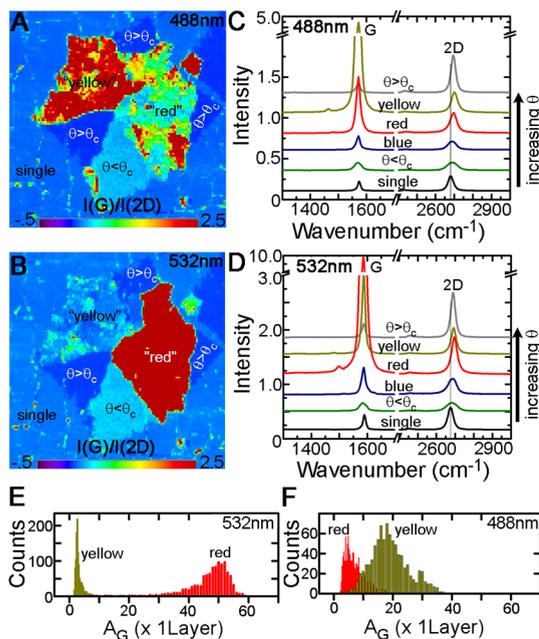


Figure 2. Raman mapping and spectra of TBG domains using a (A,C) 488 nm (2.54 eV) and (B,D) 532 nm (2.33 eV) laser. The map is of the TBG region shown in Figure 1C. (C,D) Averaged spectra from different regions of the sample as labeled. The same areas are analyzed for both wavelengths. The spectra are offset for clarity and are arranged in order of increasing twist angle. The G and 2D peaks are labeled. (E,F) Histogram of the normalized integrated G peak intensity (A_G) taken from individual spectra within the yellow and red regions shown in (A) and (B) at each wavelength (labeled). (E) For the red distribution: mean = 46.7, median = 48.1, range = 46.2. (F) For the yellow distribution: mean = 18.6, median = 17.8, range = 37.3.

and by 80% within approximately $\pm 2.5^\circ$ twist variation.¹¹ If the twist angle of the red domain equals 12.5° with a small twist deviation ($\pm 1^\circ$), the histogram at $\lambda = 532$ nm should appear like that in Figure 2E, having a high mean value and left skew. On the other hand, if θ (E_{photon}) differs by even 1° (150 meV) from θ_c (E_{θ_c}), then the A_G intensity distribution will have a lower mean value (by up to $\sim 50\%$) and have a more symmetric or right skew as observed with the yellow domain for $\lambda = 488$ nm (Figure 2F). Together, this implies $\theta_{\text{red}} \approx 12.5 \pm 1^\circ$ and that θ_{yellow} is slightly larger than 14.5° . Finally, for the blue domain, the slight increase in the G/2D ratio and upshift of the 2D peak from $\lambda = 488$ to 532 nm indicates it is by several degrees smaller than θ_{red} .

The presence of colored TBG domains suggests there are unique differences in the absorption spectrum that are dependent on twist angle. The electronic hybridization in TBG leads to changes in the optical transition matrix elements and joint density of states (JDOS), which can strongly affect the absorption properties.^{11,20} Figure 3 shows the measured and calculated optical contrast spectra for a red, yellow, blue, and large-angle bilayer domain ($\theta > \theta_c$ for $E_{\text{photon}} = 2.54$ eV), together with a single-layer region. The contrast spectrum is defined as $C(\lambda) = [R_0(\lambda) - R(\lambda)]/R_0(\lambda)$, where

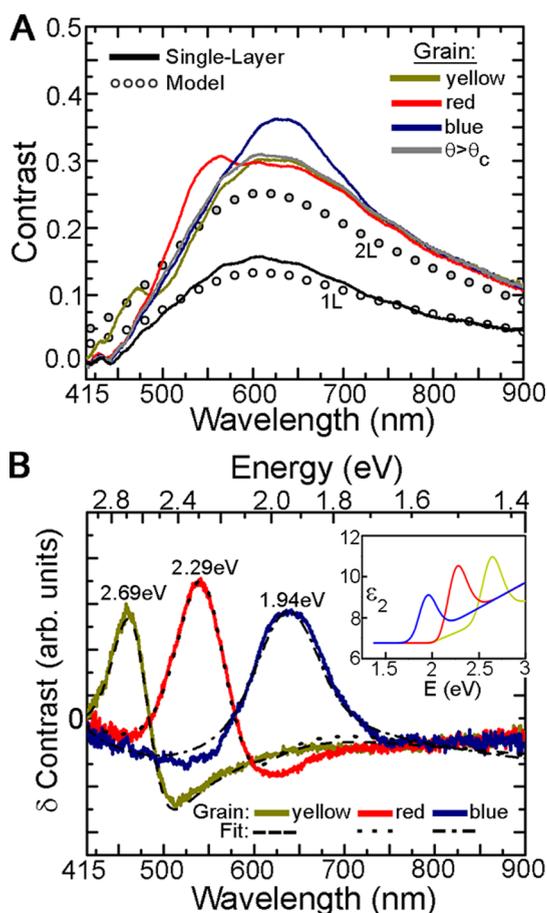


Figure 3. Averaged contrast spectra ($C(\lambda)$) for TBG domains. (A) Measured $C(\lambda)$ for a red, yellow, blue, and large-angle ($\theta > \theta_c$) domain, together with a single-layer region. All spectra are referenced to the $\text{SiO}_2(100 \text{ nm})/\text{Si}$ substrate. The calculated contrast for single-layer (1L) and two-layer (2L) graphene is included. The red, yellow, and $\theta > \theta_c$ domain were measured from Figure 1C. (B) Contrast difference of the yellow, red, and blue domains with respect to the large-angle domain, together with the best-fit curve calculated using $\epsilon_2(\omega)$ shown in the inset.

$R_0(\lambda)$ is the reflection spectra from the substrate (SiO_2/Si) and $R(\lambda)$ is reflection from the film plus substrate^{28,38} (see Supporting Information). For single-layer graphene, we find a reasonably good match between the measured and calculated $C(\lambda)$ using an index of refraction for graphene of $n_g = 2.6 - 1.3i$ ²⁸ and thickness of 0.34 nm, together with literature values of $n(\lambda)$ for SiO_2 and Si.³⁹ For two-layer graphene, the measured $C(\lambda)$ (TBG domains) and calculated $C(\lambda)$ (Bernal stacking) do not agree well. In particular, $C(\lambda)$ for the colored TBG domains is not the smooth envelope shape, but instead there is a shoulder feature unique to each domain.

Since contrast here is approximately proportional to absorption,^{11,20} the difference between two contrast spectra highlights relative absorption features of one domain to another. Hence we define a contrast difference spectrum, $\delta C_i(\lambda) = C_i(\lambda) - C_{\theta > \theta_c}(\lambda)$, where the subscript i indicates yellow, red, or blue. We chose a large-angle

domain as a common reference since its properties in visible wavelengths are close to that of two independent graphene layers. Figure 3B compares $\delta C(\lambda)$ among the red, yellow, and blue domains. For each curve, there is a distinct peak (E_{peak}), indicating enhanced relative absorption at specific regions of the visible spectrum.

Given knowledge of the substrate (see Supporting Information), it should, in principle, be possible to model the complex dielectric function $\epsilon(\omega) = \epsilon_1(\omega) - i\epsilon_2(\omega)$ of TBG that produces the strong features in the contrast difference data ($\epsilon_2(\omega)$ is directly related to the intrinsic material absorption and includes contributions from the TBG hybridization states). To carry out the modeling, we used a commercial software package able to model the reflectance from layered stacks.⁴⁰ We were able to obtain rough agreement by using the time-honored approach of modeling narrow absorptions with Gaussian or Lorentzian oscillators but found that the fits improved significantly when we employed a more flexible model that, in a Kramers–Kronig consistent fashion, reproduces the van Hove singularity (vHs) and linear density of states expected for the TBG absorption.^{11,41} Our fits to $\delta C(\lambda)$ for each domain angle are shown with dashed/dotted lines in Figure 3B, and the corresponding $\epsilon_2(\lambda)$ values are included in the inset of Figure 3B. While qualitative in nature, the fit reproduces the small dip after the main peak, as well as the increasing tail at lower wavelengths, indicating that the singularity plus linear shape of the JDOS is a good approximation for the presence and shape of the absorption feature. In addition, the fwhm of the absorption feature is approximately 0.25 eV, close to that predicted by DFT calculations.¹¹ Assuming here that E_{peak} equals E_{θ_c} , this corresponds to $\theta_{\text{blue}} \approx 10.5^\circ$, $\theta_{\text{red}} \approx 12.5^\circ$, $\theta_{\text{yellow}} \approx 15.5^\circ$,¹¹ in excellent agreement with the Raman results and analysis in Figure 2.

To directly quantify interlayer twist angles and variations within one layer's in-plane orientation, we conduct microdiffraction experiments using low-energy electron microscopy/diffraction (LEEM/LEED). Unlike LEED of TBG on SiC ,³⁷ the LEED spots of graphene and TBG on SiO_2 are diffuse (Figure 4A,B) due to the nanometer-scale substrate roughness of SiO_2 .^{42,43} Since diffracting electrons from the bottom layer in TBG on SiO_2 are greatly attenuated in these samples, we presume that the orientation of the diffraction pattern represents primarily that of the top surface layer. As such, we only measure one family of diffraction spots on either TBG or single-layer graphene (Figure 4A,B).

Mapping diffraction patterns allows direct correlation between the angular orientation and spatial extent of TBG domains. Figure 4C shows such a false color map of LEED angular orientation. The measurement was carried out by translating the partial TBG film (shown in Figure 4D) with respect to the electron beam ($5 \mu\text{m}$ diameter) over $250 \times 200 \mu\text{m}^2$ (50×40 data

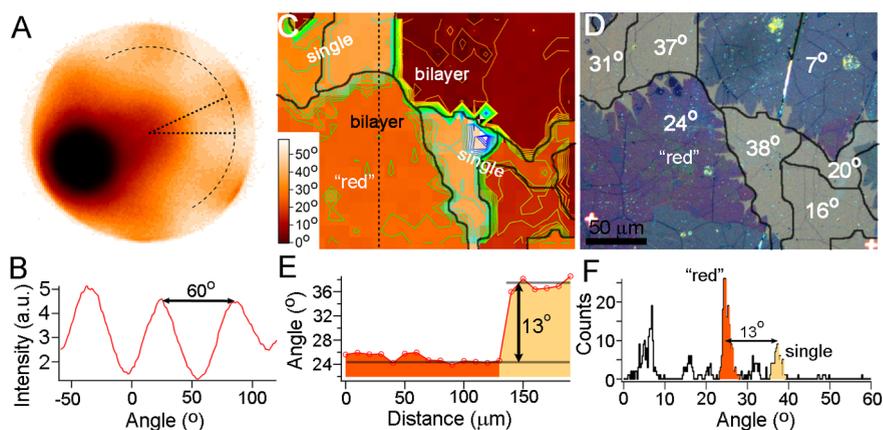


Figure 4. LEED analysis of TBG films. (A) Typical LEED pattern of a TBG film measured using $5\ \mu\text{m}$ diameter area selective aperture and an electron energy of 55 eV. (B) Intensity profile along the dotted arc in (A). (C) False color map of the LEED angular orientations of a TBG film. (D) Optical microscope images of the TBG film measured in (C). The numbers indicate the approximate LEED orientations. (E) Vertical line scans taken at the black dash line in (C). (F) Histogram from the angular contour plot in (C). The red and neighboring single-layer domain from (C) are specified in (F).

points). We specify the orientation of the diffraction pattern from the angle defined in Figure 4A, based on the intensity profile along the dotted arc (Figure 4B). Distinct angle-specific domains (labeled) in both the single-layer and bilayer regions become apparent as highlighted by the black outlines in Figure 4C,D. Cross-sectional line scans across TBG and the neighboring single-layer regions allows us to quantify the relative twist angle between the upper and lower layers (Figure 4E). A histogram of all diffraction spots also reveals the relative LEED orientation, as well as the deviation in graphene's in-plane orientation within a domain. Each peak (or cluster) in Figure 4F represents a domain within Figure 4C and shows a typical domain angular variation of $\pm 1^\circ$. Using this same process to determine twist angles of other colored domains (see Supporting Information), we find $\theta_{\text{blue}} = 11 \pm 1^\circ$, $\theta_{\text{red}} = 13 \pm 1^\circ$, and $\theta_{\text{yellow}} = 15 \pm 1^\circ$, in good agreement with Raman and optical spectroscopy measurements despite the large angular uncertainties due to broadened diffraction patterns.

The macroscopic appearance of primarily three distinct, colored TBG domains is striking since the energy of the vHs feature continuously varies with twist angle.¹² As such, this may be an indication that the fabrication process leads to preferred TBG orientations. A larger optical microscope survey does show that some TBG domains appear to have distinct, mixed color phases (e.g., Figure 1B and Figure 5B). We are currently conducting higher spatial and spectral resolution imaging to determine if there are statistically significant differences in the areal coverage of specific TBG domains. It can be said that the observed mixed colored domains are at least in part due to the measured angular (or twist) deviation within each domain, where a few degree in-plane rotations can result in a 200–300 meV shift in the absorption feature (Figure 3), as well as E_{θ_c} . As such, it is most common to find mixing

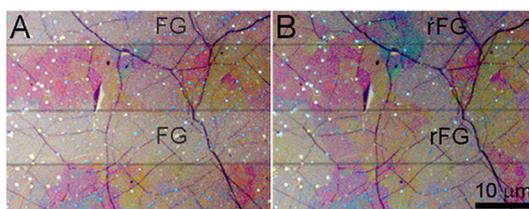


Figure 5. Local addition (A) and removal (B) of fluorine on the top surface of TBG films. Lines help guide the eye where TBG was fluorinated (labeled "FG") and reduced (labeled "rFG").

of blue ($\theta_{\text{blue}} = 11 \pm 1^\circ$) and red ($\theta_{\text{red}} = 13 \pm 1^\circ$), or red ($\theta_{\text{red}} = 13 \pm 1^\circ$) and yellow ($\theta_{\text{yellow}} = 15 \pm 1^\circ$) domains, but not the blue ($\theta_{\text{blue}} = 11 \pm 1^\circ$) and yellow ($\theta_{\text{yellow}} = 15 \pm 1^\circ$) domains, which have an angular separation larger than the angular deviation of domains in these samples. Overall improvements in TBG domain size could be greatly facilitated through the use of highly orientated graphene films as recently demonstrated on Au-foil substrates,⁴⁴ while local twist angle variations may be reduced through further optimizing the transfer process, including the use of smoother starting surfaces and low surface energy solvents during transfer and drying.⁴⁵

Finally, we demonstrate lithographic control of interlayer coupling by selective fluorination of the top graphene layer using XeF_2 gas.⁴⁶ Following fluorination, even for very short XeF_2 exposures (e.g., 1 s, 1 Torr XeF_2), color contrast is fully quenched in TBG domains with twist angles between 10 and 16° . This is seen in Figure 5A following fluorination in a pattern defined by a PMMA mask. Upon fluorine desorption ($T = 175^\circ\text{C}$ for 1 h in flowing Ar), the color contrast returns, demonstrating the sensitivity of interlayer coupling to covalent functionalization and the usefulness of optical characterization as a probe for interlayer hybridization. The reappearance of interlayer coupling after defluorination is noteworthy since defects are often introduced during the adsorption/desorption process,⁴⁶

highlighting the overall robust nature of coupling. Fluorine adsorption changes the electronic properties of graphene by reducing the charge in the conducting π orbitals, by introducing scattering centers, and by opening band gaps.⁴⁶ Fluorine also structurally deforms the graphene skeleton as bond angles shift due to transiting sp^2 to sp^3 carbon bonding, which is confirmed by the emergence of the D peak in Raman spectroscopy^{46,47} (not shown). Thus, decoupling *via* mechanical separation or *via* quenching of the hybridized electronic state are promising routes for optical-based chemical detection using TBG films with twist angles between 10 and 16°. In addition to chemical or mechanical routes, it is likely that electrostatic gating will provide tunable control over interlayer coupling and band structure, similar to that reported for Bernal (AB-stacked) bilayer graphene.^{48,49}

CONCLUSION

In summary, we show that interlayer coupling between two large-area, individually stacked graphene

films is possible and that interlayer coupling enables the direct visualization of specific bilayer orientations on SiO₂/Si substrates. The diffusion of interlayer adsorbates into localized islands facilitates the formation of atomically clean regions that can electronically hybridize. The polycrystalline nature of the initial CVD graphene leads to bilayer films whose properties depend on the relative twist angle between layers. The interlayer coupling results in the emergence of a relatively narrow absorption peak in the visible range for twist angles between 10 and 16°, where the twist angle variations within a single domain are found to be approximately $\pm 1^\circ$ in these samples. By selectively functionalizing the top surface of these bilayer films, the interlayer coupling is effectively quenched, opening possibilities of optical-based chemical detection applications. We believe the results presented here will facilitate the formation of various large-area homo- and heterogeneous bilayer systems, where interlayer hybridization can result in exploitable electronic or optical properties.

METHODS

Raman measurements were performed using a confocal geometry. Dichroic beam splitters were used to reflect single-mode 488 or 532 nm laser light onto the excitation/detection optical axis. A 100 \times microscope objective (NA = 0.65) focused the laser (spot $\approx 0.4 \mu\text{m}$) onto the sample and gathered Raman scattered light for detection. The Raman scattered photons were dispersed in a half-meter Acton Sp-2500 spectrometer and were detected using a Princeton Instruments CCD array (Spec-10:400BR back-thinned, deep-depleted array).

Optical spectroscopy measurements were performed with an inverted microscope (Nikon TE2000) coupled directly through a side port to an imaging spectrometer (Princeton Instruments MicroSpec 300). The spectrometer was equipped with an 8 μm pixel CCD (Andor model 885) and entrance slit assembly that could be temporarily moved aside for imaging. Spatial registration between images and spectra was assured because the spectrometer grating (50 groove/mm) could be exchanged under computer control with a mirror for direct imaging. The sample was illuminated through the 20 \times objective (NA = 0.45) with a quartz-tungsten-halogen lamp. The Koehler illuminator was apertured to restrict the incident rays to be within $\pm 10^\circ$ of the sample normal. Reflected light was collected by the same objective. For optical modeling of the contrast difference spectra, to reproduce the expected van Hove singularity and linear density of states in TBG, we used the software's⁴⁰ built-in functions psemi0 and psemi1.^{40,50,51}

To slightly improve the diffraction contrast during LEED measurements, we first deposited a self-assembled monolayer of hexamethyldisilazane (HMDS) on SiO₂⁵² before TBG deposition, in order to reduce surface roughness. We then acquired the LEED images *via* a 5 μm illumination aperture selecting the single graphene layer and then the TBG domains (see Supporting Information). In order to ensure that both single-layer and bilayer regions have uniform orientations in the field of view, dark-field LEEM images were acquired, as shown in Supporting Information Figure S4. Fluorination experiments were carried out using a Xactix XeF₂ etching tool. In Figure 5A, the sample was exposed in pulse mode with the following parameters: 10 cycles, 30 s/cycle, 1 Torr XeF₂, 35 Torr N₂.

Conflict of Interest: The authors declare no competing financial interest.

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Supporting Information Available: Additional data related to optical microscope imaging, LEED/LEEM measurements and optical contrast modeling are provided. This material is available free of charge *via* the Internet at <http://pubs.acs.org>.

REFERENCES AND NOTES

- Geim, A. K.; Novoselov, K. S. The Rise of Graphene. *Nat. Mater.* **2007**, *6*, 183–191.
- Geim, A. K. Graphene: Status and Prospects. *Science* **2009**, *324*, 1530–1534.
- Novoselov, K. S.; Jiang, D.; Schedin, F.; Booth, T. J.; Khotkevich, V. V.; Morozov, S. V.; Geim, A. K. Two-Dimensional Atomic Crystals. *Proc. Natl. Acad. Sci. U.S.A.* **2005**, *102*, 10451–10453.
- Ponomarenko, L. A.; Geim, A. K.; Zhukov, A. A.; Jalil, R.; Morozov, S. V.; Novoselov, K. S.; Grigorieva, I. V.; Hill, E. H.; Cheianov, V. V.; Fal'ko, V. I.; *et al.* Tunable Metal-Insulator Transition in Double-Layer Graphene Heterostructures. *Nat. Phys.* **2011**, *7*, 958–961.
- Britnell, L.; Gorbachev, R. V.; Jalil, R.; Belle, B. D.; Schedin, F.; Mishchenko, A.; Georgiou, T.; Katsnelson, M. I.; Eaves, L.; Morozov, S. V.; *et al.* Field-Effect Tunneling Transistor Based on Vertical Graphene Heterostructures. *Science* **2012**, *335*, 947–950.
- Gao, G.; Gao, W.; Cannuccia, E.; Taha-Tijerina, J.; Balicas, L.; Mathkar, A.; Narayanan, T. N.; Liu, Z.; Gupta, B. K.; Peng, J.

- et al. Artificially Stacked Atomic Layers: Toward New van der Waals Solids. *Nano Lett.* **2012**, *12*, 3518–3525.
7. Zhalalutdinov, M. K.; Robinson, J. T.; Junkermeier, C. E.; Culbertson, J. C.; Reinecke, T. L.; Stine, R.; Sheehan, P. E.; Houston, B. H.; Snow, E. S. Engineering Graphene Mechanical Systems. *Nano Lett.* **2012**, *12*, 4212–4218.
 8. Liu, Z.; Song, L.; Zhao, S.; Huang, J.; Ma, L.; Zhang, J.; Lou, J.; Ajayan, P. M. Direct Growth of Graphene/Hexagonal Boron Nitride Stacked Layers. *Nano Lett.* **2011**, *11*, 2032–2037.
 9. Shi, Y.; Zhou, W.; Lu, A.-Y.; Fang, W.; Lee, Y.-H.; Hsu, A. L.; Kim, S. M.; Kim, K. K.; Yang, H. Y.; Li, L.-J.; et al. van der Waals Epitaxy of MoS₂ Layers Using Graphene as Growth Templates. *Nano Lett.* **2012**, *12*, 2784–2791.
 10. Haigh, S. J.; Gholinia, A.; Jalil, R.; Romani, S.; Britnell, L.; Elias, D. C.; Novoselov, K. S.; Ponomarenko, L. A.; Geim, A. K.; Gorbachev, R. Cross-Sectional Imaging of Individual Layers and Buried Interfaces of Graphene-Based Heterostructures and Superlattices. *Nat. Mater.* **2012**, *11*, 764–767.
 11. Havener, R. W.; Zhuang, H.; Brown, L.; Hennig, R. G.; Park, J. Angle-Resolved Raman Imaging of Interlayer Rotations and Interactions in Twisted Bilayer Graphene. *Nano Lett.* **2012**, *12*, 3162–3167.
 12. Li, G.; Luican, A.; Lopes dos Santos, J. M. B.; Castro Neto, A. H.; Reina, A.; Kong, J.; Andrei, E. Y. Observation of Van Hove Singularities in Twisted Graphene Layers. *Nat. Phys.* **2010**, *6*, 109–113.
 13. Kim, K.; Koh, S.; Tan, L. Z.; Regan, W.; Yuk, J. M.; Chatterjee, E.; Crommie, M. F.; Cohen, M. L.; Louie, S. G.; Zettl, A. Raman Spectroscopy Study of Rotated Double-Layer Graphene: Misorientation-Angle Dependence of Electronic Structure. *Phys. Rev. Lett.* **2012**, *108*, 246103.
 14. Luican, A.; Li, G.; Reina, A.; Kong, J.; Nair, R. R.; Novoselov, K. S.; Geim, A. K.; Andrei, E. Y. Single-Layer Behavior and Its Breakdown in Twisted Graphene Layers. *Phys. Rev. Lett.* **2011**, *106*, 126802.
 15. MacDonald, A. H.; Bistritzer, R. Materials Science: Graphene Moiré Mystery Solved? *Nature* **2011**, *474*, 453–454.
 16. Lopes dos Santos, J. M. B.; Peres, N. M. R.; Castro Neto, A. H. Graphene Bilayer with a Twist: Electronic Structure. *Phys. Rev. Lett.* **2007**, *99*, 256802.
 17. Trambly de Laissardière, G.; Mayou, D.; Magaud, L. Localization of Dirac Electrons in Rotated Graphene Bilayers. *Nano Lett.* **2010**, *10*, 804–808.
 18. Mele, E. J. Band Symmetries and Singularities in Twisted Multilayer Graphene. *Phys. Rev. B* **2011**, *84*, 235439.
 19. Ni, Z.; Liu, L.; Wang, Y.; Zheng, Z.; Li, L.-J.; Yu, T.; Shen, Z. G-Band Raman Double Resonance in Twisted Bilayer Graphene: Evidence of Band Splitting and Folding. *Phys. Rev. B* **2009**, *80*, 125404.
 20. Wang, Y.; Ni, Z.; Liu, L.; Liu, Y.; Cong, C.; Yu, T.; Wang, X.; Shen, D.; Shen, Z. Stacking-Dependent Optical Conductivity of Bilayer Graphene. *ACS Nano* **2010**, *4*, 4074–4080.
 21. Ohta, T.; Robinson, J. T.; Feibelman, P. J.; Bostwick, A.; Rotenberg, E.; Beechem, T. E. Evidence for Interlayer Coupling and Moiré Periodic Potentials in Twisted Bilayer Graphene. *Phys. Rev. Lett.* **2012**, *109*, 186807.
 22. Zhan, D.; Yan, J.; Lai, L.; Ni, Z.; Liu, L.; Shen, Z. Engineering the Electronic Structure of Graphene. *Adv. Mater.* **2012**, *24*, 4055–4069.
 23. Righi, A.; Costa, S. D.; Chacham, H.; Fantini, C.; Venezuela, P.; Magnuson, C.; Colombo, L.; Bacsá, W. S.; Ruoff, R. S.; Pimenta, M. A. Graphene Moiré Patterns Observed by Umlapp Double-Resonance Raman Scattering. *Phys. Rev. B* **2011**, *84*, 241409.
 24. Ferrari, A. C.; Meyer, J. C.; Scardaci, V.; Casiraghi, C.; Lazzeri, M.; Mauri, F.; Piscanec, S.; Jiang, D.; Novoselov, K. S.; Roth, S.; et al. Raman Spectrum of Graphene and Graphene Layers. *Phys. Rev. Lett.* **2006**, *97*, 187401.
 25. Yang, L.; Deslippe, J.; Park, C.-H.; Cohen, M. L.; Louie, S. G. Excitonic Effects on the Optical Response of Graphene and Bilayer Graphene. *Phys. Rev. Lett.* **2009**, *103*, 186802.
 26. Mak, K. F.; Sfeir, M. Y.; Wu, Y.; Lui, C. H.; Misewich, J. A.; Heinz, T. F. Measurement of the Optical Conductivity of Graphene. *Phys. Rev. Lett.* **2008**, *101*, 196405.
 27. Novoselov, K. S.; Geim, A. K.; Morozov, S. V.; Jiang, D.; Zhang, Y.; Dubonos, S. V.; Grigorieva, I. V.; Firsov, A. A. Electric Field Effect in Atomically Thin Carbon Films. *Science* **2004**, *306*, 666–669.
 28. Blake, P.; Hill, E. W.; Neto, A. H. C.; Novoselov, K. S.; Jiang, D.; Yang, R.; Booth, T. J.; Geim, A. K. Making Graphene Visible. *Appl. Phys. Lett.* **2007**, *91*, 063124-3.
 29. Li, X.; Magnuson, C. W.; Venugopal, A.; Tromp, R. M.; Hannon, J. B.; Vogel, E. M.; Colombo, L.; Ruoff, R. S. Large-Area Graphene Single Crystals Grown by Low-Pressure Chemical Vapor Deposition of Methane on Copper. *J. Am. Chem. Soc.* **2011**, *133*, 2816–2819.
 30. Kim, K. S.; Zhao, Y.; Jang, H.; Lee, S. Y.; Kim, J. M.; Kim, K. S.; Ahn, J.-H.; Kim, P.; Choi, J.-Y.; Hong, B. H. Large-Scale Pattern Growth of Graphene Films for Stretchable Transparent Electrodes. *Nature* **2009**, *457*, 706–710.
 31. Reina, A.; Jia, X.; Ho, J.; Nezich, D.; Son, H.; Bulovic, V.; Dresselhaus, M. S.; Kong, J. Large Area, Few-Layer Graphene Films on Arbitrary Substrates by Chemical Vapor Deposition. *Nano Lett.* **2009**, *9*, 30–35.
 32. Bae, S.; Kim, H.; Lee, Y.; Xu, X.; Park, J.-S.; Zheng, Y.; Balakrishnan, J.; Lei, T.; Ri Kim, H.; Song, Y.; et al. Roll-to-Roll Production of 30-Inch Graphene Films for Transparent Electrodes. *Nat. Nanotechnol.* **2010**, *5*, 574–578.
 33. 450 PMMA A2 spun at 2000 rpm and baked at 85°C for 5 minutes.
 34. Kim, K.; Lee, Z.; Malone, B. D.; Chan, K. T.; Alemán, B.; Regan, W.; Gannett, W.; Crommie, M. F.; Cohen, M. L.; Zettl, A. Multiply Folded Graphene. *Phys. Rev. B* **2011**, *83*, 245433.
 35. Zhu, W.; Low, T.; Perebeinos, V.; Bol, A. A.; Zhu, Y.; Yan, H.; Tersoff, J.; Avouris, P. Structure and Electronic Transport in Graphene Wrinkles. *Nano Lett.* **2012**, *12*, 3431–3436.
 36. Ishigami, M.; Chen, J. H.; Cullen, W. G.; Fuhrer, M. S.; Williams, E. D. Atomic Structure of Graphene on SiO₂. *Nano Lett.* **2007**, *7*, 1643–1648.
 37. Ohta, T.; Beechem, T. E.; Robinson, J. T.; Kellogg, G. L. Long-Range Atomic Ordering and Variable Interlayer Interactions in Two Overlapping Graphene Lattices with Stacking Misorientations. *Phys. Rev. B* **2012**, *85*, 075415.
 38. Ni, Z. H.; Wang, H. M.; Kasim, J.; Fan, H. M.; Yu, T.; Wu, Y. H.; Feng, Y. P.; Shen, Z. X. Graphene Thickness Determination Using Reflection and Contrast Spectroscopy. *Nano Lett.* **2007**, *7*, 2758–2763.
 39. Palik, E. D. *Handbook of Optical Constants of Solids*; Academic Press: New York, 1998.
 40. VASE32, *Spectroscopic Ellipsometry Data Acquisition and Analysis*, J. A. Woollam Co.
 41. Sato, K.; Saito, R.; Cong, C.; Yu, T.; Dresselhaus, M. S. Zone Folding Effect in Raman G-Band Intensity of Twisted Bilayer Graphene. *Phys. Rev. B* **2012**, *86*, 125414.
 42. Knox, K. R.; Wang, S.; Morgante, A.; Cvetko, D.; Locatelli, A.; Mendes, T. O.; Niño, M. A.; Kim, P.; Osgood, R. M., Jr. Spectromicroscopy of Single and Multilayer Graphene Supported by a Weakly Interacting Substrate. *Phys. Rev. B* **2008**, *78*, 201408.
 43. Locatelli, A.; Knox, K. R.; Cvetko, D.; Mendes, T. O.; Niño, M. A.; Wang, S.; Yilmaz, M. B.; Kim, P.; Osgood, R. M.; Morgante, A. Corrugation in Exfoliated Graphene: An Electron Microscopy and Diffraction Study. *ACS Nano* **2010**, *4*, 4879–4889.
 44. Wofford, J., M.; Starodub, E.; Walter, A. L.; Nie, S.; Bostwick, A.; Bartelt, N. C.; Thürmer, K.; Rotenberg, E.; McCarty, K. F.; Dubon, O. D. Extraordinary Epitaxial Alignment of Graphene Islands on Au(111). *New J. Phys.* **2012**, *14*, 053008.
 45. Chan, J.; Venugopal, A.; Pirkle, A.; McDonnell, S.; Hinojos, D.; Magnuson, C. W.; Ruoff, R. S.; Colombo, L.; Wallace, R. M.; Vogel, E. M. Reducing Extrinsic Performance-Limiting Factors in Graphene Grown by Chemical Vapor Deposition. *ACS Nano* **2012**, *6*, 3224–3229.
 46. Robinson, J. T.; Burgess, J. S.; Junkermeier, C. E.; Badescu, S. C.; Reinecke, T. L.; Perkins, F. K.; Zhalalutdinov, M. K.; Baldwin, J. W.; Culbertson, J. C.; Sheehan, P. E.; et al. Properties of Fluorinated Graphene Films. *Nano Lett.* **2010**, *10*, 3001–3005.
 47. Ferrari, A. C.; Robertson, J. Interpretation of Raman Spectra of Disordered and Amorphous Carbon. *Phys. Rev. B* **2000**, *61*, 14095–14107.

48. Ohta, T.; Bostwick, A.; Seyller, T.; Horn, K.; Rotenberg, E. Controlling the Electronic Structure of Bilayer Graphene. *Science* **2006**, *313*, 951–954.
49. Zhang, Y.; Tang, T.-T.; Girit, C.; Hao, Z.; Martin, M. C.; Zettl, A.; Crommie, M. F.; Shen, Y. R.; Wang, F. Direct Observation of a Widely Tunable Bandgap in Bilayer Graphene. *Nature* **2009**, *459*, 820–823.
50. Herzinger, C. M.; Johs, B. D. Dielectric Function Parametric Model, and Method of Use. U.S. Patent 5796983, 1998
51. Herzinger, C. M.; Johs, B.; McGahan, W. A.; Woollam, J. A.; Paulson, W. Ellipsometric Determination of Optical Constants for Silicon and Thermally Grown Silicon Dioxide via a Multi-Sample, Multi-Wavelength, Multi-Angle Investigation. *J. Appl. Phys.* **1998**, *83*, 3323–3336.
52. Lafkioti, M.; Krauss, B.; Lohmann, T.; Zschieschang, U.; Klauk, H.; Klitzing, K. v.; Smet, J. H. Graphene on a Hydrophobic Substrate: Doping Reduction and Hysteresis Suppression under Ambient Conditions. *Nano Lett.* **2010**, *10*, 1149–1153.