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14. ABSTRACT Mitochondria are the powerplants of the cell. They produce the ATP necessary for the neurons to engage in reactions geared toward their proper function. Mitochondria contain a series of enzymes, in a chain-like array, that pass electrons along this chain via proton motive force which is initiated by complex I, the first of this series of enzymes. Complex I deficiency is considered one of the hallmarks of Parkinson's Disease as it contributes greatly to the energy crisis in the neurons. In an earlier study, bypassing this complex I deficiency using D- β - hydroxybutyrate (D β HB) in the MPTP (1-methyl-4-phenyl-1,2,3,6-tetrahydropyridine) mouse model of PD, dopamine neurons in the substantia nigra pars compacta were protected. Our goal in this study is to assess the effects of D β HB analogues to ascertain if they are longer-acting compounds than the parent compound. Although obtaining the first and only drug at the moment was quite difficult (it took close to 10 months, we have now initiated our first experiment which is to determine the effective dose to use in future experiments.							
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Scientific Progress Report (W81XWH-10-1-0539: July 2012)

Pre-Clinical Testing of New Hydroxybutyrate Analogs

Introduction

Parkinson's disease (PD) is the second most common neurodegenerative disorder after Alzheimer's Disease (AD) (Fahn and Przedborski, 2009). It is characterized clinically by resting tremor, rigidity, akinesia, and postural instability (Fahn and Przedborski, 2009), These motor manifestations are attributed mainly, though not exclusively, to the loss of the dopaminergic neurons in the substantia nigra pars compacta (SNpc) and their dopaminergic terminals in the corpus striatum of the nigrostriatal pathway in the brain. Biochemically, PD presents with a profound loss of DA levels in the striatum, which, in part, accounts for the noted motor manifestations (Hornykiewicz and Kisk, 1987). Although current therapy includes dopaminergic agonists and cholinergic antagonists, the most reliable and most common therapy remains Levodopa (L-DOPA), a precursor for dopamine (DA) (Fahn and Przedborski, 2009).

PD is thought of as a multi-faceted disease (Reichmann and Jost, 2008), in that it is a disease not based on a single cause, but on many interacting causes. Recent evidence suggests that mitochondrial dysfunction may be one contributing factor (Fahn and Przedborski, 2009). Mitochondria are the powerhouses of the cell and, as such, produce the energy necessary for the cell to function. Movement of electrons through the mitochondrial electron transport chain (METC), a series of enzymes (complexes I, II, III, IV, V) starts with proton motive force at the complex I site, the first of the METC series of enzymes. Electrons move down the chain to eventually produce ATP. However, when complex I is compromised, as is reported in PD (Fukae et al, 2007) and demonstrated in the MPTP mouse model of PD (Cassarino et al, 1997), mitochondria become dysfunctional as mitochondrial membrane potential collapses, ATP production is reduced and protons no longer travel properly along the chain. Thus, the neuron experiences energy crisis and respiratory failure, which means that oxidative phosphorylation is compromised and there is an increase in the presence of the superoxide radical.

Part of PD neurochemistry is a deficit in complex I activity in the METC (Suzuki et al, 1997) and this neurochemical deficit can be reproduced using the MPTP mouse model of PD. Previously, we demonstrated that we could overcome the complex I deficit using β -hydroxybutyrate, a ketone body, in the acute MPTP mouse model (Tieu et al, 2003). Infusion of this compound partially protected dopamine neurons in the SNpc against the damaging effects of MPTP and improved the motor deficits elicited by MPTP. This was accomplished by enhancing oxidative phosphorylation through augmentation of complex II activity. The only drawback to the use of β -hydroxybutyrate is that it is short-

acting and has to be delivered via alzet pumps in order to maintain steady-state blood levels. Recently, we have received a new compound, based on the skeleton of β -hydroxybutyrate. This new compound, glyceryl tris(3-hydroxybutyrate) (G3HB) from Eastman Chemical Company, supposedly has a much longer half-life. Not much is known about the new compound and further, its chemical and toxicological properties have yet to be fully tested.

Body of Work

SA I: we will examine the effects of the beta-hydroxybutyrate analogs on mitochondrial function and HDAC activity *ex vivo* and compare the obtained results with those of mitochondrial exposure to D β HB.

SA II: we will assess the neuroprotective qualities of these beta-hydroxybutyrate analogs in an in vivo setting in the MPTP mouse model of neurodegeneration. SA II has four parts.

Part I, as we have done previously with DβHB, we will first initiate the infusion, via Alzet mini-osmotic pumps, of the compounds analogues of DβHB with longer half-lives and a suitability for single daily dosage . Twenty-four hours after pump implantation, MPTP will be administered in an acute regimen. Animals will be perfused at 2 and 7 days after MPTP administration and the integrity of the nigrostriatal system will be assessed. Mac-1 and GFAP immunostaining will be done to gauge inflammation as it relates to HDAC activity. Tyrosine hydroxylase and Nissl stained neurons will be counted and complex II histochemistry will be performed.

Part II, we will prepare a new set of implanted mice treated with MPTP and motor function will be assessed on the rotarod. We will them remove the pumps and use regional sections from these brains to measure nigrostriatal levels of dopamine, 3,4-dihydroxyphenylacetic acid (DOPAC) and homovanillic acid (HVA).

Part III, a third set of implanted animals will be prepared and used for brain and blood measurements of MPTP and MPP+ will be measured at 90 minutes after the last dose of MPTP in our acute regimen to examine MPTP metabolism in the presence of compoundc A and B. We will also measure D β HB levels in blood and brain at selected times during the infusion of compound A and compound B to demonstrate whether these compounds can act as replacements for D β HB. To test the suitability for single daily dosing,

Part IV will involve single daily dosing starting at twenty-four hours prior to the initiation of MPTP. Single daily dosing will continue for four additional days and mice will be sacrificed 7 days after the last dose of MPTP. Brains will be assessed as in parts I and II.

Key Accomplishments

In our hands, G3HB was first tested in C57bl/6 mice from Charles River, as this is the strain that we have tested from all of the Charles River breeding houses in the United States and found the the mice from Kingston NY gave significant cell death with less than 5% animal death. Thus, we have used these mice in all of our MPTP studies for the past 20 years. In our initial experiment with these mice, acute MPTP (18 mg/kg free base x 4 doses over 8 hours) and the highest concentrations of G3HB (0.8 and 1.6mm/kg/day) using Alzet pumps for the delivery of the G3HB. As noted in A., there was ~a 55% loss of TH+ neurons in the SNc of the C57bl/6 mice (green:saline vs red: MPTP) which was reversed to 70% of control (orange: 0.8mmol/kg/day G3HB + MPTP and to `91% of control (yellow 1.6mmol/kg/day + MPTP). Although there was a small loss in ventral tegmental area (VTA) TH+ neurons, unlike in the SNc, the TH+ neuronal loss in the VTA was not significant.



Figure A. TH+ neuron counts for the SNc and the VTA of saline (green), MPTP, 18 mg/kg ip x 4 (red), MPTP + 0.8 mmol/kg/day G3HB) (orange), MPTP + 1.6 mmol/kg/day; N/group range from 3 to 8.

Reportable Outcomes

Although the results with C57bl/6 are good, the problem here is that the ensuing experiments using these mice were not fruitful as all of the MPTP-treated and MPTP-G3HB treated mice died. We tried repeating our first study using 4 different concentrations of the G3HB with MPTP and all of the mice died. Thinking that this was some kind of anomaly, we repeated the study a third time with the same results. We are not sure why all of the animals died, but we have considered gene drift, a mix-up in the

mouse order, mice being shipped from the wrong breeding house. While we are still in the dark about the deaths, since we are examining other strains in another study, we tried two other strains, SJL and C3H that were used in another study. About 50% of the SJL mice died whereas none of the C3H mice died. Figure B shows the TH+ neuron counts for one side of the SNc. In these mice, we saw ~36% TH+ neuron loss which was ~90% corrected with G3HB, 1.6mmol/kg/day. Because of the cell death difference between the C57 and the C3H mice, we have pushed to dosing of MPTP to 20 mg/kg in these C3H mice. If this proves fruitful, we can perform our studies in the C3H mice, but we will investigate the problem with the C57bl/6 mice from Charles River labs.



Figure B. TH+ neuron loss in MPTP-G3HB treated mice. Saline, MPTP-G3HB, and MPTP-Salinee were pump-implanted. Just MPTP had no pump. N=4/group; no animal died.

Conclusions

From our studies thus far, we know that the provided compound G3HB is neuroprotective toward the TH+ SNpc neurons but that its use may be strain dependent.

References

Fahn S and Przedborsk S. Parkinsonism. In: Merritt's Neurology, 12th Edition, LP Rowland and TA Pedley, eds. Philadelphia, Waters Kluwer/Lippincott, Williams and Wilkins, pp 751-769 (2009).

Hornykiewicz O and Kish SJ. Biochemical Pathophysiology of Parkinson's Disease. In: Parkinson's Disease, M Yahr and KJ Bergmann, eds. New York: Raven Press, pp 19-34. (1987).

Reichmann H, Jost W. <u>Parkinson's disease - Many diseases with many faces.</u> J Neurol. 2008 Sep;255 Suppl 5:1.

Fukae J, Mizuno Y, Hattori N. <u>Mitochondrial dysfunction in Parkinson's disease.</u> Mitochondrion. 2007 Feb-Apr;7(1-2):58-62.

Cassarino DS, Fall CP, Swerdlow RH, Smith TS, Halvorsen EM, Miller SW, Parks JP, Parker WD Jr, Bennett JP Jr. Elevated reactive oxygen species and antioxidant enzyme activities in animal and cellular models of Parkinson's disease. Biochim Biophys Acta. 1997 1362(**1**):77-86.

Suzuki K, Mizuno Y, Yoshida M. Effects of **1**-methyl-4-phenyl-**1**,2,3,6-tetrahydropyridine (MPTP)-like compounds on mitochondrial respiration. Neurol. 1990;53:215-8.

C, Caspersen C, Teismann P, Wu DC, Yan SD, Naini A, Vila M, Jackson-Lewis V, Ramasamy R, Przedborski S.<u>D-beta-hydroxybutyrate rescues mitochondrial respiration</u> and mitigates features of Parkinson disease. J Clin Invest. 2003 Sep;112(6):892-901.

Appendices follow: Figures and Tables

Material Safety Data Sheets for Glycerol tris(3-hydroxybutyrate).

Eastman Chemical Company Eastman Research Division P.O. Box 1972 Kingsport, Tennessee 37662

EASTMAN

PURITY PROFILE*

25 March 2011

Product Name: Glyceryl tris(3-hydroxybutyrate)

Ship to:

Attn: Dr. Serge Przedborski/Dr. Vernice Jackson-Lewis Columbia Univeristy, MNC P&S 4-401 630 West 168th Street New York, NY 10032 Tel: (212)305-8689

Date shipped: 11 April 2011 Containers shipped: 1 Weight shipped: 80 g

Lot Number EX001250-031

Properties	Sample		Method(s)		
identity	consistent with structur	e	NMR	ΜW	350.36
wt% assay	>98%		wt% NMR		
GC assay					
glyceryl tris(3-hydroxybutyrate)	93.7% GC deri		GC derivatiza	atization area%	
glyceryl bis(3-hydroxybutyrate)	2.6%		GC derivatiza	tion area	%
recidual colvent (ethyl acetate)	<0.5%		Wt% NMR		

Neil W. Boaz Eastman Representative 423-229-8105 Email nwboaz@eastman.com

*This product is subject to ongoing development. The results provided in this Purity Profile were obtained by analyzing the Batch/Lot described and may or may not be representative of any past or future Batches/Lots. The methodology and/or techniques of analysis used to obtain these results may or may not be validated. The recipient should independently determine whether this product meets their specifications and is technically suitable for their intended purpose. For additional information regarding this product and its analysis, please contact your Eastman representative. This material is NOT for human consumption.

4/13/11

TMAN MATERIAL SAFETY DATA SHEET

Revision Date: 04/08/2011

MSDSUSA/ANSI/EN/150000072403/Version 2.0

1. CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

Product Name	Glyceryl tris(3-hydroxybutyrate)		
Product Identification Number(s)	33046-00, E3304601		
Manufacturer/Supplier	Eastman Chemical Company		
Wallulacturenouppher	200 South Wilcox Drive		
	Kingsport, TN 37660-5280		
	US		
	+14232292000		
MSDS Prenared by	Eastman Product Safety and Health		
Chemical Name	Not applicable		
Synonym(s)	985287		
Molecular Formula	Not applicable		
Molecular Weight	Not applicable		
Molecular Weight	research and development sample		
Product Use	assumed hazardous; not fully investigated		
OSHA Status			

For emergency health, safety, and environmental information, call 1-423-229-4511 or 1-423-229-2000.

For emergency transportation information, in the United States: call CHEMTREC at 800-424-9300 or call 423-229-2000.

2. COMPOSITION INFORMATION ON INGREDIENTS

(Typical composition is given, and it may vary. A certificate of analysis can be provided, if available.)

Weight %	
>90%	
<5%	
<1%	

Component Glyceryl tris(3-hydroxybutyrate) Glyceryl bis(3-hydroxybutyrate) ethyl acetate

CAS Registry No. 135413-30-8 Not assigned 141-78-6

3. HAZARDS IDENTIFICATION

THE PHYSICAL-CHEMICAL AND TOXICOLOGICAL PROPERTIES OF THIS MATERIAL HAVE NOT BEEN FULLY INVESTIGATED

Health - 2, Flammability - 1, Chemical Reactivity - 0 HMIS® Hazard Ratings:

HMIS® rating involves data interpretations that may vary from company to company. They are intended only for rapid, general identification of the magnitude of the specific hazard. To deal adequately with the safe handling of this material, all the information contained in this MSDS must be considered.

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exposure limits. If exposure limits have not been established, maintain airborne levels to an Respiratory Protection: If engineering controls do not maintain airborne concentrations below recommended exposure limits (where applicable) or to an acceptable level (in countries where exposure limits have not been established), an approved respirator must be worn. Respirator exposure limits have not been established), an approved respirator must be worn, respirator type: Air-purifying respirator with an appropriate, government approved (where applicable), air-purifying filter, particides or conjuster. Context health and participative professional or manufactures for type: Air-puniying respirator with an appropriate, government approved (where applicable), air-purifying filter, cartridge or canister. Contact health and safety professional or manufacturer for Eye Protection: wear satety glasses with side shields (or goggles). Skin Protection: Wear chemical-resistant gloves, footwear, and protective clothing appropriate for the risk of exposure. Contact health and safety professional or manufacturer for specific information. nsk of exposure. Contact nearn and salety professional of manufacturer for specific in Recommended Decontamination Facilities: Eye bath., Washing facilities., Safety shower. 9. PHYSICAL AND CHEMICAL PROPERTIES Physical Form: Viscous Liquid Color: Yellow Odor: Slight Specific Gravity: <1 Boiling Point: 200 °C 0.5 mm Hg Flash Point: > 93 °C (estimated) Thermal Decomposition Temperature: Thermal stability not tested. Low stability hazard expected Solubility in Water: Appreciable at normal operating temperatures. Not fully evaluated. Materials containing similar structural groups 10. STABILITY AND REACTIVITY Material reacts with Strong oxidizing agents. are normally stable. Stability: Will not occur. Incompatibility: Hazardous Polymerization: Acute toxicity data, if available, are listed below. Additional toxicity data may be available on request. 11. TOXICOLOGICAL INFORMATION

12. ECOLOGICAL INFORMATION

Acute toxicity data, if available, are listed below. Additional toxicity data may be available on request.

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This material has not been tested for environmental effects.

13. DISPOSAL CONSIDERATIONS

Dispose of waste and residues in accordance with local authority requirements. Incinerate. Since emptied containers retain product residue, follow label warnings even after container is emptied.

Important Note: Shipping descriptions may vary based on mode of transport, quantities, package size, and/or origin and destination. Consult your company's Hazardous Materials/Dangerous Goods expert for information specific to your situation. 14. TRANSPORT INFORMATION

specific to your situation.

DOT (USA)

Class not regulated

Sea - IMDG (International Maritime Dangerous Goods)

Class not regulated

Air - ICAO (International Civil Aviation Organization)

Class not regulated

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Revision Date: 04/08/2011 MSDSUSA/ANSI/EN/150000072403/Version 2.0

This product has been classified in accordance with hazard criteria of the Controlled Products Regulations and the MSDS contains all the information required by the Controlled Products Regulations.

WHMIS (Canada) Status: controlled

WHMIS (Canada) Hazard Classification: D/2/B

SARA 311-312 Hazard Classification(s): immediate (acute) health hazard

SARA 313:

1 . . .

Carcinogenicity Classification (components present at 0.1% or more): none, unless listed below

TSCA (US Toxic Substances Control Act): One or more components of this product are not listed on the TSCA inventory. In the USA, commercial industrial use is restricted to FDA-regulated applications.

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