NOVEL PROCESSING OF BORON CARBIDE (B₄C): PLASMA SYNTHESIZED NANO POWDERS AND PRESSURELESS SINTERING FORMING OF COMPLEX SHAPES

James Campbell, Melissa Klusewitz, Jerry LaSalvia, and Ernest Chin U.S. Army Research Laboratory Aberdeen Proving Ground, MD 21005-5069

> Robert Speyer and Namtae Cho Georgia Institute of Technology Atlanta, GA 30332

Noel Vanier, Cheng-Hung Hung, Edward Abbott and Peter Votruba-Drzal PPG Industries 4325 Rosanna Drive Allison Park, Pennsylvania 15101

> William Coblenz and Toni Marcheaux DARPA, Defense Sciences Office 3701 N. Fairfax Drive Arlington, VA 22203-1714

ABSTRACT

Boron carbide, most often used in personnel armor systems, is rather difficult to form and is almost always densified from ceramic powders under heat and pressure. Recent developments by PPG and Georgia Tech are allowing for the full densification of nano-sized powders through a pressureless sintering route. PPG is manufacturing these powders in a plasma processing technique, and they are less expensive than traditionally produced powders. The powders are formed directly in the plasma, and they can be doped with densification aids. Pressureless sintering techniques developed by Georgia Tech offers a low-cost processing route for B₄C that was previously unavailable. Together both efforts will lead to boron carbide with shapes and sizes unavailable today and using a fine-grained domestically produced starting powder.

1. INTRODUCTION

Boron carbide is an important armor ceramic that is primarily used in personnel protection applications due to its very low density and its ability to defeat small arms threats. Boron carbide has a density of 2.5 g/cm³, compared to 3.2 g/cm^3 for silicon carbide (SiC) and 3.9 g/cm^3 for aluminum oxide (Al₂O₃).

Traditionally, boron carbide (B_4C) armor is manufactured through the hot pressing of ceramic powders, which requires the application of uniaxial pressure during the sintering (densification) process. This processing route allows for only very simple shapes to be formed, such as flat plates or those with some curvature, such as a Small Arms Protective Insert (SAPI) plate. Until recently, it has been very difficult to densify this ceramic to full density via a pressureless sintering route. Sintering would give the ability to make components with much more complicated geometries, such as conformable body armor and helmet liners. Georgia Institute of Technology (GIT) has developed a sintering approach with a follow-on hot isostatic (gas) pressing (HIP) to produce pressureless sintered, additive-free, B_4C components at full density.

Currently, almost all B_4C powders used in armor applications are imported due the high amounts of electricity required to manufacture the material. It is formed in an electric arc furnace as a large billet of material, which is then ground down into the powders used to make ceramic bodies. PPG offers an alternate approach to making the powders. The starting materials are reacted in a plasma and powders are formed directly. These powders are nano-sized and the process allows for the addition of sintering aids directly into the powder.

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2. TRADITIONAL B₄C POWDERS

Like SiC, B₄C is a man-made material; the ceramic powder is typically produced through the carbothermal reduction of boron oxide in an electric arc (Acheson) furnace, Figure 1. The end product is a large ingot of B₄C that must be pulverized and ground down into the desired grit and powder sizes. Both the high temperature furnace operation and the subsequent grinding operations make it a very energy-intensive approach to making B₄C powder; and this accounts for the high cost of the powder and the limited powder production capacity in the United States. After the B₄C powder is ground and sized, only a portion of the powder is of armor grade quality, since there is variability of the material in the ingot. The remainder must be used for other applications such as the abrasives industry. Nearly all B₄C powder is imported, with China becoming a major player in the B₄C powder business. Figure 2 shows the two main suppliers of boron carbide to the United States are China and Germany. It should be noted that these numbers are for all grades of powder, not just armor-grade material. But it is interesting to note that major military procurement programs for body armor were going on during this period of rapid increases in powder imports. Typically, armor-grade B₄C powder has a cost of approximately \$50/lb, while a sintered aluminum oxide might be \$5/lb for the finished tile.



Figure 1. A large B_4C ingot is manufactured in an Acheson furnace, and is subsequently ground into powder of various grades.



Figure 2. B₄C powder imports as a function of country and year

3. PPG NANOPOWDERS

PPG is producing their B_4C powders through a very different method, in which the ceramic powder is formed by directly by reacting the precursors in a gas. The raw materials are vaporized and atomized in the thermal plasma, along with any sintering aids. The basic reaction would be a boron precursor, such as boron oxide (B_2O_3), and a carbon source (carbon or a hydrocarbon) are vaporized and atomized in a thermal plasma. As the plasma cools as it moves through the reactor, B_4C powder is formed through the following reaction:

$$2B_2O_3 + 7C \rightarrow B_4C + 6CO$$

The B_4C powders are then collected. Depending on the residence time in the reactor, powders of nano-scale particle sizes are formed. Reactor parameters must be chosen such that the reaction has time to progress to completion but the time/temperature/pressure profile must minimize reversal of the reaction. The powders are collected in a filter system and the powder is then washed to remove any residual B_2O_3 that remains.

For this process, the powder is formed directly from the starting materials, and the final powder should have a more consistent chemistry and less contamination, since there are no grinding operations. Currently, the powders were manufactured in a laboratory scale plasma system and larger systems are planned. Cost estimates show that powders produced through this method would be less expensive than B₄C considerably powders manufactured through traditional methods. Since powders are manufactured directly, the costs associated with grinding disappear. It also eliminates contamination of the powder with the materials in grinding medium and grinding vessel. The high electric costs and pollution associated of the traditional electric arc furnace have limited the amount of domestic production of B₄C powders. A process such as the gas plasma system

demonstrated here will allow for the domestic production of B_4C powders. It is targeted that powders produced by this process may be less expensive when produced in large quantities, when compared to \$50/lb which is a common cost for armor-grade powder produced by traditional methods.

Initial powders produced by the process were very uniform and with a very fine grain size, but they showed great difficulty in sintering to high density, even attempts to hot press were unsuccessful. Therefore, sintering aids were added directly to the plasma. These sintering aids were added as oxides and sometimes more than one were added together. A large number of sintering aids and dopants were investigated and densification studies were conducted. The studies looked at densification rates throughout the sintering schedule and to determine which combination of furnace conditions (hold times, ramp rate, atmosphere, etc) would provide the highest densities for a particular powder system. The goal would be to produce sintered densities that would be high enough so that a post-sinter hot isostatic press (HIP) could be done to close the remaining porosity and allow for full density to be achieved. Georgia Tech conducted a number of dilatometry studies on these powders Figure 3 shows a typical densification experiment for one of the PPG compositions, where the temperature profile with a ramp rate to 2300°C is shown as well as the dimensional changes that were measured by the dilatometer. In the upper temperature ramp, rapid densification is apparent, which slows at the isothermal soaking. The nano-sized grains are very unstable at these temperatures; their melting temperatures are significantly lowered, resulting in near-instantaneous grain growth at the temperatures corresponding to the momentary expansions shown in the sintering curves. The final result often is a B₄C with micron-sized grains; typical to what is observed when conventionally produced powders are used.



Figure 3. Dilatometry showing dimensional changes over time for the shown furnace temperature profile

4. B₄C HOT PRESSING

After B₄C powder is produced, regardless of how it was manufactured, it has to densified to form a useful armor product. Traditionally, all B₄C is manufactured through hot pressing, in which the powder is densified at a high temperature with the concurrent application of uniaxial pressure via graphite dies. In the US, there are two main commercial manufacturers of boron carbide: Ceradyne, Costa Mesa, CA and BAE-Advanced Ceramics Division, Vista, CA. Verco Materials, Atlanta, GA is a start-up company from the same Georgia Tech research group that conducted the dilatometry studies on the PPG powders. They are manufacturing complex-shaped B₄C through pressureless sintering with a subsequent HIP step. They are producing armor tiles and extremity armor using conventional powders, and supplying them to a number of armor system suppliers and laboratories. Verco has also produced armor tiles using the PPG nanopowders, which were tested at ARL.

Traditional hot pressing is a batch process, where the "green" (undensified) powder compacts are formed through some method with the ceramic powder and loaded into a hot pressing die. The die and powder are brought up to the sintering temperature and a load is applied to the die. Due to the very high temperature required for sintering, graphite tooling and specialized furnaces are required. Many of these non-oxide ceramics require sintering temperatures in excess of 2000°C. As the size of the ceramic part increases, the load required for the desired pressure becomes quite large. Since body armor inserts are the most common B4C parts manufactured, the load frames in the hot pressing equipment needs to be quite large. To increase throughput, commercial vendors typically stack a number of plates together with machined graphite spacers and apply pressure to the entire stack. Although parts of simple geometry are most easily manufactured by hot pressing, the commercial vendors have introduced forming methods to allow for curved plates to be manufactured, such as a Small Arms Protective Insert (SAPI) plate. When armor tiles are manufactured by hot pressing, billets of the material are produced and the tiles ground and sectioned. B₄C is a very brittle material with low fracture toughness. Therefore, it is very difficult to machine and prone to edge chipping, which adds additional expensive to hot pressed B4C tiles.

Hot pressing allows for the use of coarser ceramic powder and less reliance on sintering aids to form a fully dense part. Therefore, the powders may be more expensive for sintering; however, the expendable furnace and any postdensification machining would add more expense to a hot pressed part. Overall, it is anticipated that it would be considerably less expensive to manufacture parts, such as a SAPI plate, using a pressureless sintered technique.



Figure 4. B_4C plates formed via hot pressing with a curved shape.

Commercial B_4C shows a large number of lenticular graphitic inclusions, Figure 5. Inclusions of this type are also observed in fracture surfaces; therefore suggesting that these are not polishing artifacts. It is difficult to ascertain how these inclusions are introduced into the material - in the starting powders, during green forming or during densification. Interestingly, the ballistic properties don't seem to be particularly sensitive to these inclusions. However, these inclusions do affect the mechanical properties of the materials and they act as crack initiation points in flexure testing.



Figure 5. SEM micrograph showing large lenticular graphitic inclusions in commercial hot pressed B_4C

5. B₄C PRESSURELESS SINTERING AND HIPING

Georgia Tech and Verco Materials have developed methods to manufacture B_4C without the aid of uniaxial

pressure during densification. They have been able to develop sintering profiles that allow for high densities to be achieved. By understanding when chemical reactions occur, when compounds breakdown and how materials out-gas, a sintering schedule can be developed that minimizes time at temperature ranges where unfavorable transformations occur. A differential dilatometer was used to develop these sintering schedules and understand when these reactions occur, similar to the data shown in They have attained closed-porosity Figure 2. microstructures which can then be successfully post-hot isostatically pressed to a zero-porosity state. HIPing involves heating the part in the presence of high-pressure inert gas which applies a hydrostatic pressure on the part which is shape preserving. HIPing closes the remaining porosity and theoretical density can be achieved. HIPing is done commercially in very large vessels and large number of parts can be densified at one time. Therefore, this step adds marginal cost to the overall process.

A major opportunity for pressureless sintering is parts of complex geometry, which cannot be produced by hot pressing. The undensified "green" bodies" for these parts are made by established ceramic processing methods, such as extrusion, injection molding, or slip casting. Figure 6 shows a throat and upward blast protection B_4C insert that Verco Materials produced via pressureless sintering. This processing route would allow for the manufacture of complex-shaped pieces for conformable body armor, such as shoulder inserts and neck protectors.



Figure 5. Complex shaped part of B_4C manufactured by Verco Material via pressureless sintering using conventional powders.

Summary

Boron Carbide is an important armor ceramic that is used primarily in personnel protection due to its low density. Pressureless sintering techniques developed by Verco Materials for B_4C would allow for lower costs and the ability to form conformable shapes. Traditional B_4C powders are all imported and are rather expensive. PPG is manufacturing B_4C powders through a different route. These powders are nano-sized and offer the opportunity to produce price-competive powders domestically. The PPG powders have been densified to full density by Verco Materials.

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A. General Information 1. Present Date 09/22/2008		2. Unclassified Title						
		NOVEL PROCESSING OF BORON CARBIDE (B4C): PLASMA SYNTHESIZ						
3. Author(s) James Campbell, Melissa Klusewitz et al.		4. Office Symbol(s) AMSRD-ARL-WM-MD		5. Telephone Nr(s) 4-0896				
6. Contractor generated No Yes		7. Type: Report Abstract Publication Presentation (speech, briefing, video clip, poster, etc) Book Book Chapter Web						
		8. Key Words boron carbide, nanopowders, sintering, body armor						
9. Distribution Statement (<i>required</i>) Is manuscript subject to export control?		Circle appropriate les	Circle appropriate letter and number. (see instructions for statement text) 10. Security					
		☆ B C D E F	Classification					
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