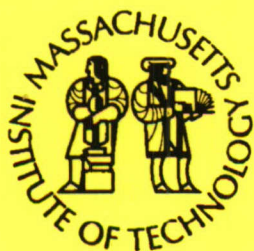


MIT/WHOI 2007-23

**Massachusetts Institute of Technology
Woods Hole Oceanographic Institution**



**Joint Program
in Oceanography/
Applied Ocean Science
and Engineering**



DOCTORAL DISSERTATION

Geochemical and Rheological Constraints on the
Dynamics of the Oceanic Upper Mantle

by

Jessica Mendelsohn Warren

September 2007

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DOCTORAL DISSERTATION

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Geochemical and Rheological Constraints on the Dynamics of the Oceanic Upper Mantle

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Submitted to the MIT/WHOI Joint Program in Marine Geology and Geophysics
in partial fulfillment of the requirements for the degree of

Doctor of Philosophy

at the

MASSACHUSETTS INSTITUTE OF TECHNOLOGY

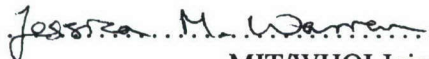
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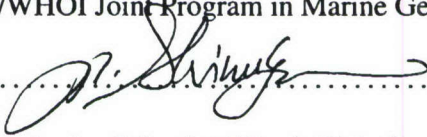
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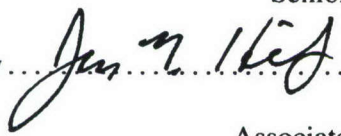
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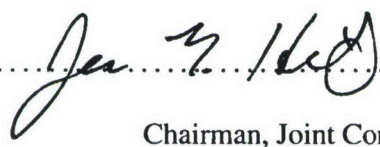
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Abstract

I provide constraints on mantle convection through observations of the rheology and composition of the oceanic upper mantle. Convection cannot be directly observed, yet is a fundamental part of the plate tectonic cycle. Relative motion among plates is accommodated by localized deformation at their boundaries. I demonstrate that in the ductile regime, strain localization occurs when different mineral phases are mixed together, limiting grain annealing. Upper mantle flow is by dislocation creep, resulting in seismic anisotropy due to mineral alignment. I use a shear zone in the Josephine Peridotite to quantify the relationship between mineral orientation and shear strain, providing an improved framework for the interpretation of seismic anisotropy. The upper mantle is generally assumed to be homogeneous in composition. From detailed isotopic and chemical analyses of abyssal peridotites from the Southwest Indian Ridge, I show that the mantle is heterogeneous at a range of length-scales. Abyssal peridotites recovered at ocean ridges are generally interpreted as the depleted residues of melt extraction. I find that melt-rock reaction is a significant part of the melt extraction process, modifying the composition of the lithospheric mantle. The generation of heterogeneous lithosphere provides a source for asthenospheric heterogeneity, via subduction and mantle convection.

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Chapter 1

Introduction

This thesis provides constraints on mantle convection through observations of the rheology and composition of the oceanic upper mantle. Mantle convection is a fundamental part of the plate tectonic cycle, but cannot be directly observed. Instead, it must be studied by indirect methods, such as seismic observations, modeling, and sampling of peridotites, the depleted residue of mantle melting that produces oceanic crust. In this thesis, I present two investigations into the rheology of peridotites (Chapters 2 and 3) and two investigations into the geochemistry of peridotites (Chapters 4 and 5). From analysis of olivine microstructures, I address the origins of localized ductile deformation in the lithosphere and the response of olivine orientation to flow in the upper mantle. I then present datasets of peridotite chemical and isotopic compositions and use these to constrain mantle composition – both lithospheric and asthenospheric – and ridge melting processes.

1.1 Rheological Constraints

The oceanic crust and upper mantle, collectively known as the oceanic lithosphere, form two-thirds of the rigid plates that make up the outer layer of the Earth. Relative motion among plates is accommodated at their boundaries, resulting in zones of localized defor-

mation, such as transform faults. At the surface, strain accumulation results in brittle failure on these faults, producing earthquakes. At depth below a fault, increasing pressure and temperature result in a transition from brittle to ductile accommodation of strain. In Chapter 2, I present observations from peridotite mylonites – ultra-fine grained rocks collected at oceanic fracture zones – for the mechanisms involved in ductile shear localization.

Strain localization in the ductile regime is promoted by the transition from grain-size insensitive to grain-size sensitive creep, which at small grain size results in a lower viscosity and thus higher strain rate. This transition requires grain size reduction, which must be permanent for strain localization to be permanent. In Chapter 2, I use electron backscatter diffraction (EBSD) analysis of a peridotite mylonite to examine olivine grain orientation and size, and the distribution of secondary phases. The observations place constraints on the dominant creep mechanisms in peridotites during strain localization and the process by which grain size is limited. In addition, the conditions of mylonite deformation are estimated by comparison to laboratory data for olivine, to provide quantitative constraints on conditions at the base of the seismogenic zone.

In the convecting mantle, movement is accommodated by creep mechanisms. In the upper mantle, the observation of seismic anisotropy indicates that creep has produced mineral alignment, leading to the conclusion that olivine is deforming by dislocation creep (Hess, 1964). In Chapter 3, I provide quantitative constraints on the relationship between olivine orientation and shear strain. These observations are critical for quantifying the relationship between the kinetics of deformation and the direction and magnitude of seismic anisotropy. Experimental investigations (e.g., Nicolas et al., 1973) into the relationship between olivine LPO and shear strain have provided a framework for LPO evolution models. However, experiments are conducted on pure olivine aggregates at higher stresses and faster strain rates than occur within the Earth. Using EBSD, I determine the evolution of olivine lattice preferred orientation as a function of shear strain in naturally deformed peridotites from the

Josephine Peridotite in southwest Oregon. Our observations from natural samples provide an improved framework for the interpretation of seismic anisotropy.

1.2 Geochemical Constraints

The oceanic lithosphere is produced by pressure-release melting of the convecting asthenospheric mantle when it upwells beneath ocean ridges. This lithosphere eventually cools and becomes negatively buoyant, sinking back into the asthenosphere at subduction zones. The extent to which subducted slabs are re-mixed into the convecting mantle and eventually exposed again at ridges is the subject of much debate. In general, the upper mantle is treated as compositionally homogeneous, due to the relative homogeneity of MORBs (mid-ocean ridge basalts) with respect to ocean island basalts (Zindler and Hart, 1986). The composition of the upper mantle is constrained from (1) the average MORB isotopic composition, which reflects continental crust extraction over the past 3 Gy, and (2) the composition of residual abyssal peridotites. Workman and Hart (2005) used these constraints to construct a model average composition for the upper mantle, referred to as DMM (Depleted MORB Mantle). I use abyssal peridotites in Chapters 4 and 5 to assess the extent to which the upper mantle deviates from this average composition.

In Chapter 4, I present detailed isotopic and trace element analyses of clinopyroxenes and orthopyroxenes from abyssal peridotites, to constrain isotopic heterogeneity in the asthenospheric mantle. MORBs, as mixtures of melts sourced over relatively large areas, cannot preserve heterogeneities at the shortest length-scales. In addition, by mass balance, the isotopic composition of MORBs are dominated by the least chemically depleted component in the melt. However, despite these constraints, significant variations in MORB composition have been observed (e.g., Dupré and Allègre, 1983). In addition, while a general consensus exists as to the presence of recycled crustal components in the mantle, the debate in recent years has centered around the “veined mantle hypothesis” (e.g.,

Allègre and Turcotte, 1986). In this model, the upper mantle contains eclogite and pyroxenite veins that are direct remnants of stretched and thinned subducted oceanic crust. I test this hypothesis in Chapter 4 by examining the isotopic and trace element compositions of pyroxenite-veined abyssal peridotites.

The composition of peridotites collected at ocean ridges reflects the processes of melting and melt extraction, in addition to initial source composition. In Chapter 5, I assess the role of melt-rock reaction in modifying the composition of the lithospheric mantle at ridges. Abyssal peridotites, collected at all major ocean ridges, are generally interpreted as the depleted residues of near-fractional melting (Johnson et al., 1990). Melt is extracted from peridotites by focused flow in dunite channels, which produces the observed chemical disequilibrium between peridotites and MORBs (e.g., Kelemen et al., 1995). This interpretation minimizes the role of other melt-rock reaction processes in modifying peridotite composition. In Chapter 5, I demonstrate that abyssal peridotites encompass a larger compositional range than previously recognized and that much of this variation is the result of a variety of melt-rock reaction mechanisms. Heterogeneity of the lithospheric mantle is an important component of the convection cycle, as the lithosphere is eventually recycled back into the convecting mantle at subduction zones. The time-integrated effect of recycling heterogeneous lithospheric mantle into the asthenosphere will be increased heterogeneity within the Earth.

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Chapter 2

Grain Size Sensitive Deformation Mechanisms in Naturally Deformed Peridotites*

Abstract

Microstructural analyses of peridotite mylonites from the oceanic lithosphere indicate that shear localization results from the combined effects of grain size reduction, grain boundary sliding and second phase pinning during deformation. The pinning effect, combined with experimental flow laws for olivine, suggests that a permanent transition from dislocation creep processes to diffusion creep occurs. This rheological transition provides a mechanism for long term weakening of the lithosphere for rocks deforming in the brittle-ductile regime. In addition, our results support the hypothesis that a transition to diffusion creep promotes the randomization of pre-existing lattice preferred orientations, which would reduce seismic anisotropy.

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2.1 Introduction

The observation of fine grained mylonites in shear zones indicates that grain size sensitive creep processes promote strain localization (e.g., Kirby, 1985; Tullis et al., 1990; Drury et al., 1991; Jaroslow et al., 1996; Jin et al., 1998; Newman et al., 1999; Jiang et al., 2000). For the sample in Fig. 2.1, microstructural observations show that ductile deformation resulted in three orders of magnitude grain size reduction. Peridotite mylonites are commonly sampled from oceanic fracture zones (Jaroslow et al., 1996). The geologic provenance of these mylonites suggests that they formed in the ductile region of the lithosphere along the down-dip extension of transform faults. We describe fabrics in a peridotite mylonite from the Shaka Fracture Zone that support extrapolation of experimentally determined rheological data (Hirth and Kohlstedt, 2003) to geologic conditions appropriate for the oceanic lithosphere. We conclude that grain boundary sliding and grain boundary pinning lead to permanent grain size reduction and strain localization.

Previous studies of peridotite mylonites in orogenic massifs and ophiolites have interpreted grain size reduction in mylonites as the result of (i) fluid addition (e.g., Vissers et al., 1995), (ii) reaction with melt (e.g., Dijkstra et al., 2002), and (iii) the breakdown reaction of spinel to plagioclase with decreasing pressure (e.g., Furusho and Kyuichi, 1999; Newman et al., 1999); see Drury et al. (1991) for a review of earlier literature. Vissers et al. (1995) interpreted the microstructures in Erro-Tobbio lherzolite mylonites as resulting from a wet olivine rheology with reaction-related grain size reduction. Dijkstra et al. (2002) concluded that a melt-present reaction promoted grain size sensitive deformation in shear zone mylonites from the Othris peridotite massif. In the Turon de Técoüère peridotite shear zone, Newman et al. (1999) suggested that grain size reduction resulted from the spinel to plagioclase phase transition reaction.

Several lines of evidence indicate that the Shaka Fracture Zone mylonites deformed in the absence of water and melt, and that breakdown reactions were not involved. We find

no syn-deformational hydrous phases such as amphibole, talc or serpentine. The mantle at fracture zones is expected to be dehydrated and depleted due to earlier on-axis melting and melt extraction (Hirth and Kohlstedt, 1996). In addition to the lack of evidence for melt or water involvement in the mylonite deformation, we find no indication of reaction enhancement via the breakdown of spinel to plagioclase. The rare earth element concentrations measured by ion microprobe in a clinopyroxene porphyroclast have the typical depleted compositions found in other abyssal peridotites (Johnson et al., 1990), with no Eu anomaly, which is a sensitive indicator of plagioclase formation (Warren et al., 2003).

2.2 Methods

To study deformation mechanisms in mylonites, we used electron backscatter diffraction (EBSD) to analyze olivine and orthopyroxene crystal orientations in thin sections of mylonitized peridotites cut parallel to lineation and perpendicular to foliation. We made orientation measurements down to a 1 μm grain size, an order of magnitude smaller size than that accessible with a U-stage. In addition, EBSD permits identification and quantification of the distribution of secondary phases. Data were gathered using an electron backscatter detector attached to a JEOL 840 SEM. Portions of thin sections were mapped for lattice orientation in 1, 2, or 4 micron steps (depending on the average grain size of the region being mapped), using a rasterized beam across the sample surface. EBSD patterns were processed and analyzed using HKL Technology's Channel 5 software package.

Orientation maps in Figs. 2.2-2.4 contain from 52% to 72% indexed data. Non-indexed (white) pixels represent points with a mean angular deviation (MAD) number $\geq 1^\circ$, which result from surface roughness and computer mis-indexing. The MAD number quantifies the mismatch between lattice planes in the calculated orientation and those quantified from the digitized bands of the diffraction pattern. Raw data, consisting of all indexed points with a MAD number $< 1^\circ$, are shown as maps and pole figures for coarser and finer grained regions

in Fig. 2.2. Processed data, for which wild spikes have been removed and all pixels with zero solutions have been extrapolated, are also shown in Fig. 2.2. Wild spikes are single pixels (i) which are misoriented by $>10^\circ$ from the average orientation of the surrounding eight pixels and (ii) for which the maximum misorientation between any two of the surrounding pixels is $<10^\circ$. All pixels which are wild spikes or have a MAD number $>1^\circ$ are replaced with zero, or null, solutions. In the processed data, pixels with zero solutions were replaced with the most common neighbor orientation. This method of processing the data minimizes the overcounting of individual grains when extracting one orientation point per grain for the construction of pole figures. However, it under-represents the smallest grains, which are close to the sampling step size and are filtered out with noise in the data.

Pole figures, shown in Figs. 2.2-2.4, were calculated using one point per grain, with all data sets containing ≥ 180 grains, which has been demonstrated to provide statistically robust results (Ben Ismaïl and Mainprice, 1998). Grain boundaries were defined by misorientations $\geq 10^\circ$ between adjacent points and subgrains by $2^\circ - 10^\circ$ misorientations. Pole figures of raw and processed data in Fig. 2.2 demonstrate that in both the finer and coarser grained areas, the processing affects the strength of the lattice preferred orientation (LPO), but does not alter the overall LPO pattern. There are two main differences between the raw and processed data: (1) the removal of small grains, points with no solutions and wild spikes, which decreases the randomness of the data, leading to a stronger LPO and (2) extrapolation of the data, which further reduces the number of small grains but increases the randomness of the data, as groups of adjacent small grains with the same orientation are replaced by individual large grains.

Using an alternative processing method, shown in Fig. 2.2I-J, data are extrapolated prior to the removal of wild spikes. This provides a better representation of the distribution of the pyroxene and spinel grains in the finest grained areas of the sample. However, as we wish to minimize potential bias in the olivine fabrics, we show data processed by the first

method only in Figs. 2.3 and 2.4. These issues are inherent in analyzing such fine grained material, which has a grain size close to the minimum step size.

2.3 Results

Microstructural observations of the mylonites indicate that grain size reduction leads to progressive localization of deformation into fine-grained bands. The influence of grain size on rheology can be seen in the optical-scale microstructures in Fig. 2.1. The mylonite consists of anastomosing bands of coarser ($\leq 100 \mu\text{m}$) and finer ($\leq 10 \mu\text{m}$) grained olivine, with distributed grains of orthopyroxene, clinopyroxene and spinel. Large, round to elongate porphyroclasts of olivine, orthopyroxene, clinopyroxene and spinel, ranging in size from 0.1 mm to 5 mm, are preserved in the fine grained matrix. These microstructures indicate that the large porphyroclast grains behaved as hard inclusions during deformation. In some cases, these inclusions are relatively coarse-grained aggregates of olivine (Fig. 2.1). Kinematic indicators, such as vergence of small folds and asymmetric recrystallization trails around porphyroclasts, indicate the relative sense of shear for the mylonite.

Olivine orientations in coarser and finer grained regions of the mylonite are shown in maps and pole figures in Figs. 2.3 and 2.4. The data are plotted with respect to the banding and lineation orientations, which we assume are equivalent to the shear plane for this high strain rock. Data are presented for two coarser grained regions of the mylonite in Fig. 2.3, with the pole figures divided into 1-10 μm and 10-100 μm size grains for each dataset. In the first region (Fig. 2.3A-C), a [100] maximum is observed in the plane of the banding and sub-parallel to the lineation, while a [001] maximum is observed perpendicular to the banding. In some areas, the LPO is significantly stronger among 10-100 μm grains than among 1-10 μm grains. In the other coarse grained region, a pole figure is only shown for the 10-100 μm size grains (Fig. 2.3E), due to the large step size at which this dataset was collected. The [100] maximum is not oriented parallel to the banding. A variable LPO

was also found by limited universal stage work on coarser grained regions in this sample (Jaroslow et al., 1996).

In the finer grained regions, shown in Fig. 2.4, a decrease in LPO strength is observed in all three areas measured. Almost no LPO is observed in some regions of the fine-grained bands (Fig. 2.4A-F). Locally, in the fine grained region shown in Fig. 2.4E-F, we find a weak and dispersed “ghost” of the LPO observed in the coarser-grained region in Fig. 2.3A-C.

The variation of LPO and grain size correlates with the secondary phase distribution. Pyroxenes and spinel are inhomogeneously distributed in the mylonite, with small grains of these phases predominant in regions with a smaller olivine grain size. Based on the very fine grain size, the raw data (Fig. 2.2A,E), and the alternative method of processing the data (Fig. 2.2I), the number of small orthopyroxene grains is greater than that shown in Figs. 2.3 and 2.4. The EBSD orientation maps highlight the larger number of secondary phases in finer grained areas relative to coarser grained areas (compare Fig. 2.2B to Fig. 2.2F). The finer grained areas have a smaller range of grain sizes and a more random distribution of secondary phases. In contrast, the coarser grained regions have a larger range of olivine grain sizes. Locally, small grains of secondary phases are mixed with small grains of olivine at the edges of larger olivine grains.

Figs. 2.2F and 2.2I provide lower and upper bounds on the amount of secondary phases pinning the grain boundaries in the finest grained areas. We have also examined back-scattered electron images and X-ray maps of coarse and fine grained areas. These confirm our observation of a greater concentration of secondary phases in the finest grained areas. Similarly, high resolution EBSD and X-ray maps of Michibayashi and Mainprice (2003, 2004) show a greater accumulation of secondary phases in finer grained regions of the peridotite mylonite in their study.

2.4 Discussion

2.4.1 Fabric variations with grain size

The variation in olivine fabric with grain size in the mylonite provides an opportunity to constrain grain size sensitive deformation mechanisms in the lithosphere. Three observations in particular can be exploited to constrain the deformation mechanisms: (i) randomization of a pre-existing LPO in the finest grained regions of the mylonite; (ii) observation of an LPO in coarser grained regions; (iii) the preservation of porphyroclasts.

The randomization of olivine LPO in the finest grained regions of the mylonite indicates a change in deformation mechanism to diffusion creep. Two observations suggest that the randomization of LPO occurred as deformation localized into fine grained regions. First, a weak LPO is observed in one fine grained area (Fig. 2.4F) which is similar to the LPO in a coarser grained area (Fig. 2.3B-C). Second, the LPO among 1-10 μm grains in a coarser grained area is similar, but weaker, than that in the 10-100 μm size fraction (Fig. 2.3B-C). We conclude that the randomization of LPO is related to grain rotations that occur during the grain boundary switching process, which is required during diffusion creep (e.g., Raj and Ashby, 1971).

The LPO in the coarser grained regions indicates that a dislocation process accommodated deformation during the early stages of mylonitization. At the same time, the preservation of large, sometimes equant, porphyroclasts implies that the rheology is grain size sensitive. If the rheology was not grain size sensitive, then all grains should be deformed to the same degree and large, relatively undeformed porphyroclasts would not be preserved. Based on comparison to experimental studies, we suggest that deformation in the coarser grained bands occurred by dislocation accommodated grain boundary sliding (DisGBS). DisGBS is a grain size sensitive creep mechanism that was identified experimentally on the basis of rheologic data on olivine aggregates deformed near the transition between disloca-

tion creep and diffusion creep (Hirth and Kohlstedt, 1995, 2003). Of particular significance for our analysis is that deformation by DisGBS has been shown to produce LPOs in calcite (Schmid et al., 1987; Rutter et al., 1994) and ice (Durham et al., 2001; Goldsby and Kohlstedt, 2002).

During DisGBS, strain is dominantly accommodated by the relative movement of grains (i.e. grain boundary sliding) and is limited by either this process or the movement of dislocations within grains on the easy slip system (Goldsby and Kohlstedt, 2001; Hirth and Kohlstedt, 2003). These two processes act in series, as deformation is not accommodated independently by either grain boundary sliding or easy slip. Thus the strain rate for DisGBS ($\dot{\epsilon}_{DisGBS}$) is limited by the slowest of the two mechanisms through the relationship:

$$\dot{\epsilon}_{DisGBS} = \left(\frac{1}{\dot{\epsilon}_{GBS}} + \frac{1}{\dot{\epsilon}_{Easy}} \right)^{-1} \quad (2.1)$$

where $\dot{\epsilon}_{GBS}$ is the strain rate for grain boundary sliding (GBS) and $\dot{\epsilon}_{Easy}$ is the strain rate for easy slip. The grain size sensitivity of DisGBS results from the influence of grain size on the GBS component of the flow law.

There is considerable confusion regarding the terminology used for grain size dependent deformation mechanisms. For example, deformation in both the diffusion creep and DisGBS regimes involves grain boundary sliding, with relative grain movement accommodated either by diffusion or dislocation processes, respectively. Furthermore, the term superplasticity is often used when referring to deformation accommodated by GBS (see discussion in Goldsby and Kohlstedt, 2001)). To clarify our terminology, we refer to the serial deformation process (Eq. 2.1) as DisGBS and the grain boundary sliding component of the process as GBS.

2.4.2 Deformation mechanism maps

We use olivine deformation mechanism maps to constrain the deformation conditions of the mylonites (Fig. 2.5A). The total strain rate ($\dot{\epsilon}_{Total}$) during deformation is contoured in stress versus grain size space, using the constitutive law (Hirth and Kohlstedt, 2003):

$$\dot{\epsilon}_{Total} = \dot{\epsilon}_{Dif} + \dot{\epsilon}_{Dis} + \dot{\epsilon}_{LTP} + \dot{\epsilon}_{DisGBS} \quad (2.2)$$

where $\dot{\epsilon}_{Dif}$ is the strain rate for diffusion creep, $\dot{\epsilon}_{Dis}$ for dislocation creep, and $\dot{\epsilon}_{LTP}$ for low-temperature plasticity (LTP). The flow law formulation and experimental data for LTP are from Goetze (1978) and Evans and Goetze (1979). These four deformation mechanisms are independent, so the mechanism with the fastest strain rate controls the rheology.

The boundaries for the fields of diffusion creep, dislocation creep, and DisGBS are calculated using a compilation of olivine experimental data from Hirth and Kohlstedt (2003). The flow laws for melt- and water-free aggregates in these regimes have the form:

$$\dot{\epsilon} = A\sigma^n d^{-m} \exp\left(-\frac{E + PV}{RT}\right) \quad (2.3)$$

where A is a constant, σ is the differential stress, n is the stress exponent, d is the grain size, p is the grain size exponent, E is the activation energy, P is the pressure, V is the activation volume, R is the gas constant, and T is the absolute temperature. For the mylonite, d is in the range of 1-100 μm . The deformation temperature is constrained from thermometry to lie in the range 600-750°C (Jaroslow et al., 1996). We use a nominal pressure of 400 MPa for these calculations. The pressure of deformation is not well constrained, however the PV term in Eq. 2.3 is negligible for deformation in the shallow oceanic lithosphere.

Application of the experimental olivine data to the conditions of mylonite formation involves a significant extrapolation of temperature, but not of grain size or stress. Therefore, the greatest uncertainty in the map in Fig. 2.5A is in the activation energies for the defor-

mation mechanisms. We calculated the error range for the field boundaries due to error in these activation energies. The activation energies for DisGBS are the most uncertain. For the grain boundary sliding component, the activation energy is assumed to be the same as that for the easy slip system, which is constrained experimentally at high temperature (Bai et al., 1991). However, results from Carter and Ave'Lallemant (1970) on the experimental deformation of dunites indicate that the easy slip system in olivine changes at low temperature. Despite uncertainty in the extrapolation to low temperature, the range of estimated mylonite deformation conditions lies within the DisGBS field, as shown in Fig. 2.5A.

To emphasize the conditions where DisGBS is a significant deformation mechanism, and its significance for LPO development, we plot strain rate versus grain size for a stress of 300 MPa in Fig. 2.5B. This plot represents a constant stress section through Fig. 2.5A. The total strain rate is dominated by DisGBS at grain sizes in the range ~ 15 -700 μm . Furthermore, the strain rate for DisGBS is at least an order of magnitude greater than that for dislocation creep at grain sizes $< \sim 300$ μm . The strain rate for diffusion creep becomes greater than that for DisGBS at grain sizes $< \sim 15$ μm . For comparison, at a grain size of ~ 15 μm , the strain rate for diffusion creep is approximately three orders of magnitude greater than that for dislocation creep. Assuming that LPOs form when a dislocation process accommodates a significant component of the total strain – and recalling that our microstructural observations indicate a randomization of LPO when grain size decreases below ~ 10 μm – we conclude that the LPOs observed in coarser grained regions of the mylonite resulted from DisGBS.

Our results suggest that an LPO is maintained during DisGBS when the easy slip component is dominant during deformation, though rate limited by the GBS component. Grain boundary sliding, whether diffusion or dislocation accommodated, tends to randomize LPOs. Prior and co-workers have demonstrated that a grain boundary sliding process leads to an increased dispersion in the misorientation between grains, which weakens an initial

single crystal LPO (Jiang et al., 2000; Bestmann and Prior, 2003; Storey and Prior, 2005). Our observation of a randomized LPO during diffusion creep supports the suggestion by Bestmann and Prior (2003) that a grain boundary sliding process weakens LPOs.

2.4.3 Recrystallization and grain boundary pinning

To understand how grain size sensitive creep promotes strain localization, we consider the processes that control the evolution of grain size during mylonite deformation. As deformation continues to lower temperature, and therefore higher stress, grain size decreases as a result of dynamic recrystallization. Empirical relationships between stress and grain size have been determined for olivine and this recrystallization piezometer (Karato et al., 1980; Van der Wal et al., 1993) is shown in Fig. 2.5A. Dynamic recrystallization is driven by gradients in dislocation density, which promote subgrain rotation and grain boundary migration. In the diffusion creep field, the applicability of the empirical piezometer is problematic owing to a potential lack of driving force for recrystallization. This insight motivated the field boundary hypothesis (De Bresser et al., 1998, 2001), in which the piezometer is defined by the boundary between the dislocation creep and diffusion creep fields. Thus, the piezometric relationship may be temperature dependent due to differences in the activation energy for the creep processes. However, Drury (2005) emphasized the lack of evidence for temperature dependence in the empirical relationships for olivine, based on the observation of statistically equivalent piezometric relationships for samples deformed at 1100-1300°C (Van der Wal et al., 1993) and at 1500°C (Karato et al., 1980). Thus, while the field boundary hypothesis has merit, the higher temperature experimental data indicate that a recrystallization piezometer applies within the dislocation creep regime.

The magnitude of grain size reduction in the mylonite can be explained by either the empirical piezometric relationship or a modified field boundary hypothesis. As illustrated in Fig. 2.5A, the grain size at which we see evidence for LPO randomization (i.e., ~ 10

μm) falls near the boundary between diffusion creep and DisGBS at geologic strain rates, consistent with a modified field boundary hypothesis in which the piezometer is defined by the diffusion creep/DisGBS field boundary, rather than the diffusion creep/dislocation creep field boundary. At the same time, the empirical piezometric relationship falls in the DisGBS field (within uncertainty), indicating that the driving force for recrystallization arises from the easy slip component of the DisGBS mechanism. In either case, we emphasize that the extreme grain size reduction observed in the mylonites is difficult to explain without the DisGBS mechanism.

Grain boundary pinning can lead to permanent grain size reduction and a transition to diffusion creep. Without a mechanism to limit grain growth, grain size reduction may not promote a transition to diffusion creep in single phase materials, due to a dynamic balance between grain growth and recrystallization (De Bresser et al., 1998, 2001). However, we suggest that the translation of grains during DisGBS by grain boundary sliding (and phase boundary sliding) results in the mixing of different mineral phases, which eventually inhibits grain growth. In the mylonite, the randomization of LPO in the fine grained regions indicates that the presence of pyroxenes and spinel results in the pinning of olivine grain boundaries, leading to a transition to diffusion creep. Previous studies have emphasized the role of grain size pinning, resulting from fluid addition, melt crystallization or metamorphic reactions, in promoting grain size sensitive creep processes (Vissers et al., 1995; Furusho and Kyuichi, 1999; Newman et al., 1999; Dijkstra et al., 2002).

2.4.4 Strain localization

To explore the implications of grain size sensitive creep mechanisms for strain localization in the oceanic lithosphere, in Fig. 2.6 we compare the viscosities of a fine grained shear zone and the surrounding coarse grained mantle. Based on our analysis of microstructural data and experimental flow laws, we assume that deformation in the shear zone occurs at

a stress and grain size on the boundary between the DisGBS and diffusion creep fields on the deformation mechanism map, which leads to the relationships:

$$\dot{\epsilon}_{DisGBS} = \dot{\epsilon}_{Dif} \quad (2.4)$$

$$\dot{\epsilon}_{SZ} = \dot{\epsilon}_{DisGBS} + \dot{\epsilon}_{Dif} = 2\dot{\epsilon}_{DisGBS} = 2\dot{\epsilon}_{Dif} \quad (2.5)$$

where $\dot{\epsilon}_{SZ}$ is the shear zone strain rate.

The effective viscosity (η) of the shear zone in Fig. 2.6 is calculated using Eqs. 2.3 and 2.5, the relationship $\eta = \sigma / \dot{\epsilon}_{SZ}$ and assuming $\dot{\epsilon}_{SZ} = 10^{-12} \text{s}^{-1}$. The calculations are made for both the GBS and the easy slip components of DisGBS, as shown in Fig. 2.6. At temperatures greater than $\sim 700^\circ\text{C}$, viscosity in the shear zone is controlled by the GBS component of the DisGBS flow law and this component may dominate, within error, at lower temperatures as well.

The shear zone viscosity at 700°C is approximately four orders of magnitude lower than outside the shear zone (Fig. 2.6). Outside the shear zone, we assume (i) that dislocation creep accommodates deformation of the coarse-grained mantle and (ii) that stress is limited by the strength of the shear zone. The effective viscosity for the shear zone remains lower than the coarse grained mantle until a temperature of $\sim 950^\circ\text{C}$, at which point deformation is no longer localized. Where grain size pinning results in a transition to diffusion creep, the viscosity contrast would be even larger than that illustrated in Fig. 2.6.

A convenient flow law for deformation in mantle shear zones – which accounts for grain size dependence – can be derived by solving Eq. 2.4 for grain size and substituting into Eq. 2.5:

$$\dot{\epsilon}_{SZ} = 2 \frac{A_{DisGBS}^3}{A_{Dif}^2} \sigma^{8.5} \exp\left(\frac{2H_{Dif} - 3H_{DisGBS}}{RT}\right) \quad (2.6)$$

where $H = E + PV$. Using the values for these parameters from Hirth and Kohlstedt

(2003), Eq. 2.6 can then be written as:

$$\dot{\epsilon}_{SZ} = 2.4 \times 10^{-7} \sigma^{8.5} \exp\left(\frac{-4.7 \times 10^5}{RT}\right) \quad (2.7)$$

for σ in MPa. This relationship is applicable for shear zone deformation at temperatures $< 950^\circ\text{C}$ and lithospheric depths, where the PV term is negligible. We reiterate that the largest uncertainty is related to the activation energy, as shown in Fig. 2.5A. At higher temperature, and therefore lower stress, errors associated with extrapolation in stress and grain size should also be considered.

2.4.5 Relationship of mylonites to earthquake processes

A combination of geophysical, experimental and geochemical observations indicate that the transition from brittle to ductile processes occurs at a temperature of $\sim 600^\circ\text{C}$ in the oceanic lithosphere. Seismological studies of the depth of earthquakes along transform faults indicate that seismicity is confined to depths in the oceanic lithosphere with temperatures $< 600^\circ\text{C}$ (e.g., Engeln et al., 1986; Abercrombie and Ekström, 2001). Extrapolation of experimental data for olivine aggregates indicates that the 600°C isotherm also marks the transition from stable to unstable frictional sliding (Boettcher et al., 2007). Finally, Jaroslow et al. (1996) estimated that the minimum temperature for mylonite deformation is $\sim 600^\circ\text{C}$, based on olivine-spinel geothermometry.

In Fig. 2.7, we compare the estimated strain rate for mylonite deformation to the strain rates for different geologic processes. The stress and strain rate of mylonite deformation are estimated using the microstructural observations combined with the olivine flow laws. From the recrystallization piezometer (Karato et al., 1980; Van der Wal et al., 1993), the estimated differential stress for a recrystallized grain size of $10 \mu\text{m}$ is 240 MPa and of $3 \mu\text{m}$ is 580 MPa. At $600\text{--}700^\circ\text{C}$ and these stress and grain size conditions, the estimated strain

rate for mylonite deformation is $10^{-14} - 10^{-10} \text{s}^{-1}$. The strain rate for post-seismic slip in Fig. 2.7 is calculated as:

$$\dot{\epsilon} = \left(\frac{\text{slip}}{\text{width}} \right) \times \left(\frac{1}{\Delta t} \right) \quad (2.8)$$

where Δt is the duration of post-seismic slip. The estimated strain rate for post-seismic slip is $10^{-10} - 10^{-5} \text{s}^{-1}$, for $\text{slip} < 1 \text{ m}$, $\text{width} \sim 1 \text{ m to } < 1 \text{ km}$, and $\Delta t \approx 1 - 100 \text{ days}$. Therefore, the estimated strain rate for mylonites overlaps the low end of post-seismic slip and the high end of tectonic strain rates (i.e., $10^{-15} - 10^{-12} \text{s}^{-1}$). Our results place constraints on the conditions of deformation at the base of the seismogenic zone in the oceanic lithosphere, which may be used to interpret geodetic data for post-seismic slip (e.g., Montési and Hirth, 2003).

2.4.6 Mantle anisotropy

The microstructural observations of the mylonite demonstrate that the transition to diffusion creep results in a randomization of pre-existing LPOs. Evidence that pre-existing anisotropy may be destroyed by the transition to diffusion creep is important for understanding mantle anisotropy. In the upper mantle, seismic anisotropy reflects high temperature flow by dislocation creep (Nicolas and Christensen, 1987; Nishimura and Forsyth, 1989; Mainprice and Silver, 1993; Jung and Karato, 2001). However, with increasing depth and temperature in the mantle, the dominant mechanism of olivine deformation may change from dislocation creep to diffusion creep (Karato and Wu, 1993). Our results indicate that such a change can lead to a decrease in mantle anisotropy, if enough strain is accumulated to randomize the fabric. We emphasize that in the case of shear zones, we do not expect fabric randomization to have a detectable effect on regional mantle anisotropy. However, following the suggestion of Karato and Wu (1993) that a transition to diffusion creep may occur at depth in the upper mantle, our results imply that this would lead to an isotropic layer in the lower portion of the upper mantle.

2.5 Conclusions

Peridotite mylonites demonstrate that three orders of magnitude grain size reduction occurs during localized deformation in the oceanic lithosphere. Microstructural analyses of the mylonite fabric suggest that deformation of aggregates with a grain size of 10-100 μm occurred by DisGBS and produces an LPO. Grain boundary pinning, due to the mixing of pyroxenes and spinel among olivine grains during DisGBS, resulted in permanent grain size reduction. Reduction in grain size down to 1-10 μm , combined with second phase pinning, led to a transition to diffusion creep and resulted in a randomization of the pre-existing LPO.

Our microstructural observations are consistent with extrapolation of experimental olivine flow laws to temperatures of 600-800°C during mylonite deformation. The empirical recrystallization piezometer plots within the field of DisGBS, providing a driving force for grain size reduction. Alternatively, grain size reduction may have followed a modified version of the field boundary hypothesis (De Bresser et al., 1998, 2001). At temperatures below $\sim 950^\circ\text{C}$, deformation in the DisGBS and diffusion creep fields occurs at lower stress and viscosity than in the dislocation creep field, providing a mechanism for weakening and strain localization in the oceanic lithosphere at transform faults.

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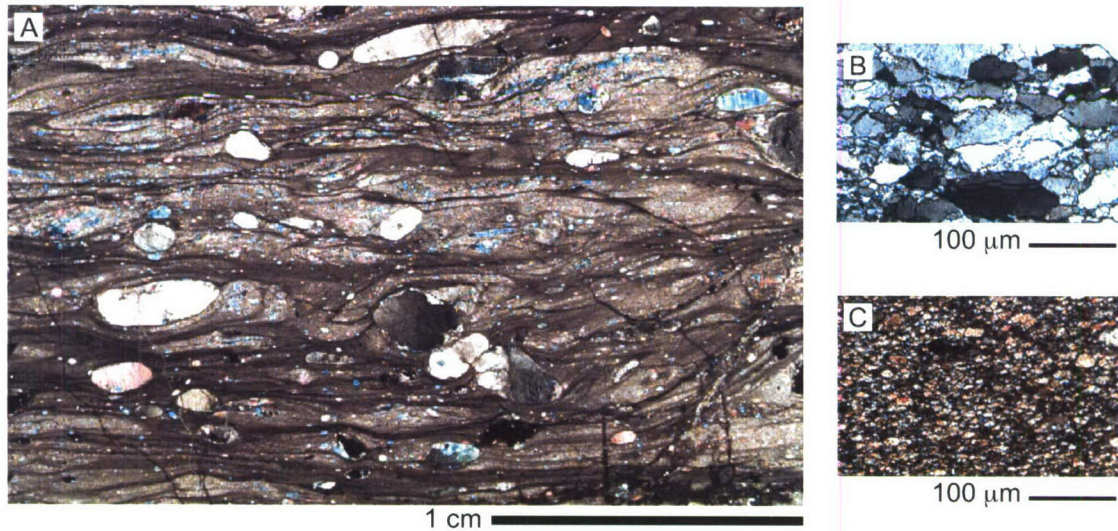


Figure 2.1: **(A)** Photomicrograph of ultra-mylonite peridotite sample AII-107-61-83 from the Shaka Fracture Zone on the Southwest Indian Ridge. Grain size sensitive deformation has resulted in the formation of anastomosing bands of coarser ($<100\ \mu\text{m}$) and finer ($<10\ \mu\text{m}$) grained olivine, with distributed grains of orthopyroxene, clinopyroxene and spinel. Large porphyroclasts of olivine, orthopyroxene, clinopyroxene, spinel and coarser grained olivine aggregates remain and appear to have behaved as hard inclusions during deformation. **(B)** Photomicrograph of a coarser grained area of the ultra-mylonite, with a grain size range of $1\text{-}100\ \mu\text{m}$. **(C)** Photomicrograph of a finer grained area, with an olivine grain size of $1\text{-}10\ \mu\text{m}$.

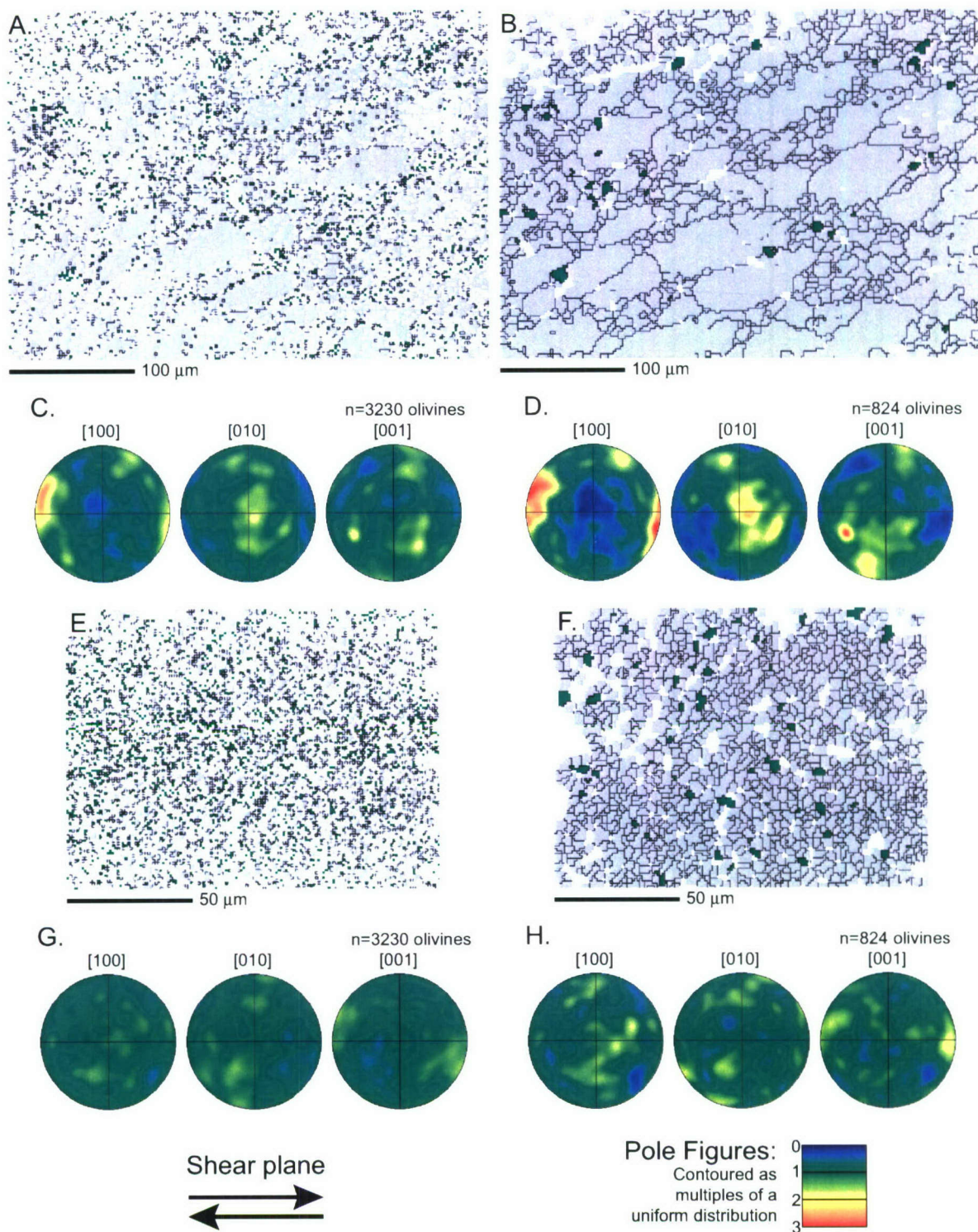


Figure 2.2: Maps of raw data collected in a coarser grained area (A) and a finer grained area (E) of the ultra-mylonite. (Continued on next page.)

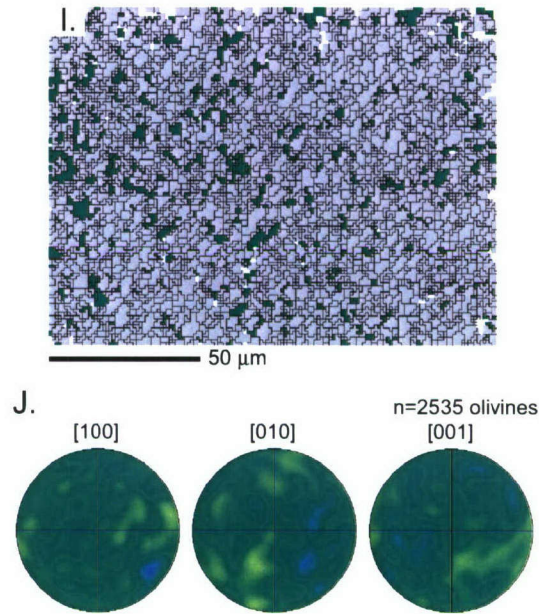


Figure 2.2: (Continued) The coarser grained area was collected at a $2\ \mu\text{m}$ step size with a total of 64% of the area indexed; the finer grained area was collected at a $1\ \mu\text{m}$ step size with 52% indexed. Each grey point represents an olivine orientation collected by EBSD and green points represent secondary phases (orthopyroxene, clinopyroxene, and spinel) for which orientations were measured. White points have a MAD number $\geq 1^\circ$. Black lines indicate grain boundaries, defined by a 10° misorientation between adjacent points. Maps of the processed data, for which wild spikes have been removed and data extrapolated, are shown in (B) and (F). The map in (I) of the fine grained area shows an alternative method of processing the data, in which the data is extrapolated prior to removing wild spikes. Pole figures of one orientation point per olivine grain are shown in (C),(D),(G),(H), and (J). All pole figures are equal area lower hemisphere projections, contoured using a 15° half width and as multiples of a uniform distribution (MUD) to a maximum of 3 MUD. Contouring of 1 MUD indicates no significant difference from a uniform distribution and corresponds to the absence of an LPO. The shear direction is parallel to the horizontal axis in the pole figures.

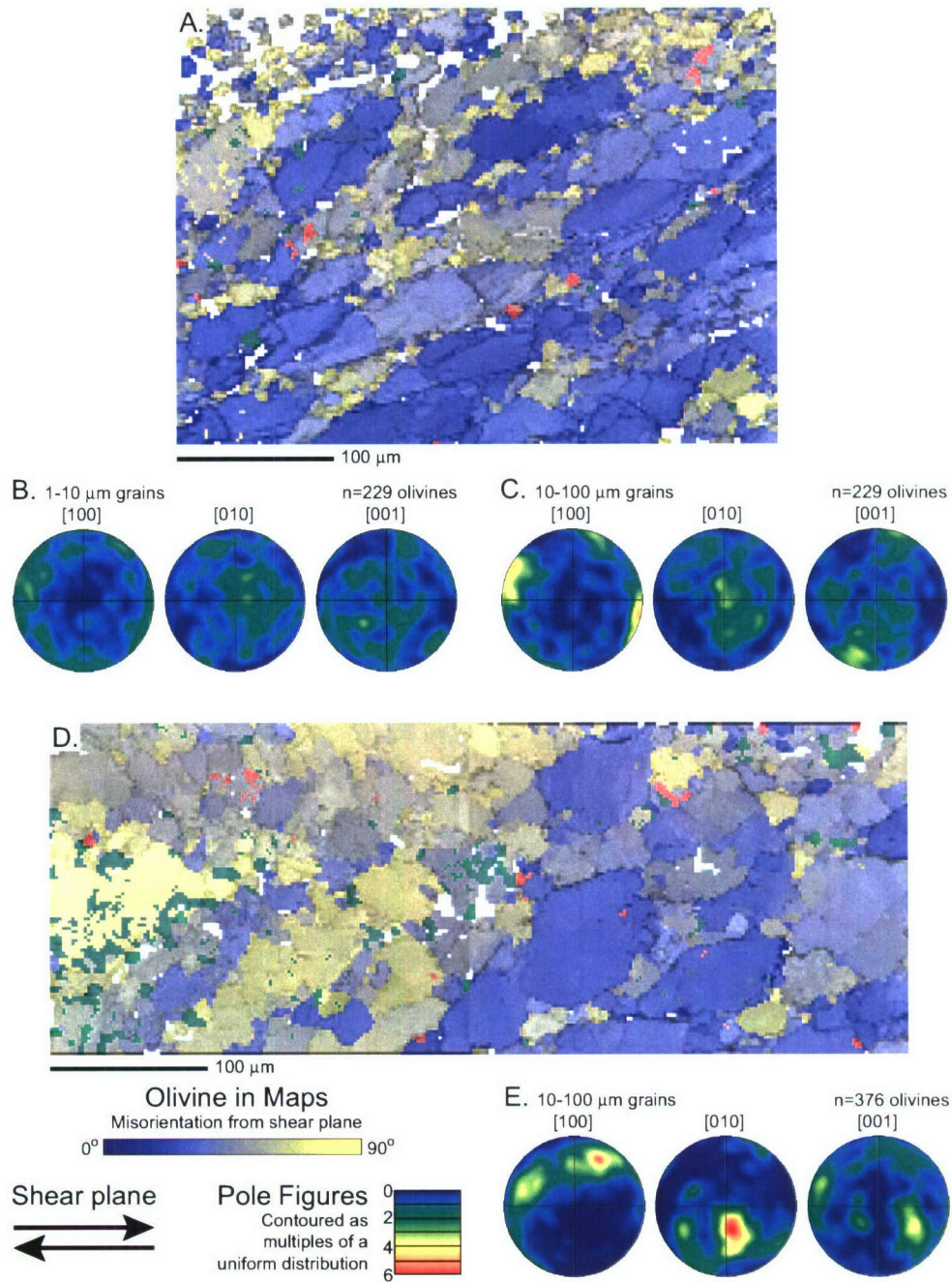


Figure 2.3: Orientation maps collected by EBSD for two coarser grained areas of the mylonite (A, D). The area in (A) is also shown in Fig. 2.2; the area in (D) was collected at a 4 μm step size, with 72% of the area indexed. Olivine is shaded from blue to yellow, with the color range representing the degree of misorientation of each pixel from the shear plane. Dark green corresponds to orthopyroxene, light green to clinopyroxene, red to spinel, and white areas have no data. Pole figures of one orientation per olivine for the area in (A) are shown for 1-10 μm grains in (B) and for 10-100 μm grains in (C). For the area in (D), data are only plotted for 10-100 μm grains in (E), due to the large step size of the map. Contouring in all pole figures is to a maximum of 6 MUD. The shear direction is parallel to the horizontal axis in the pole figures.

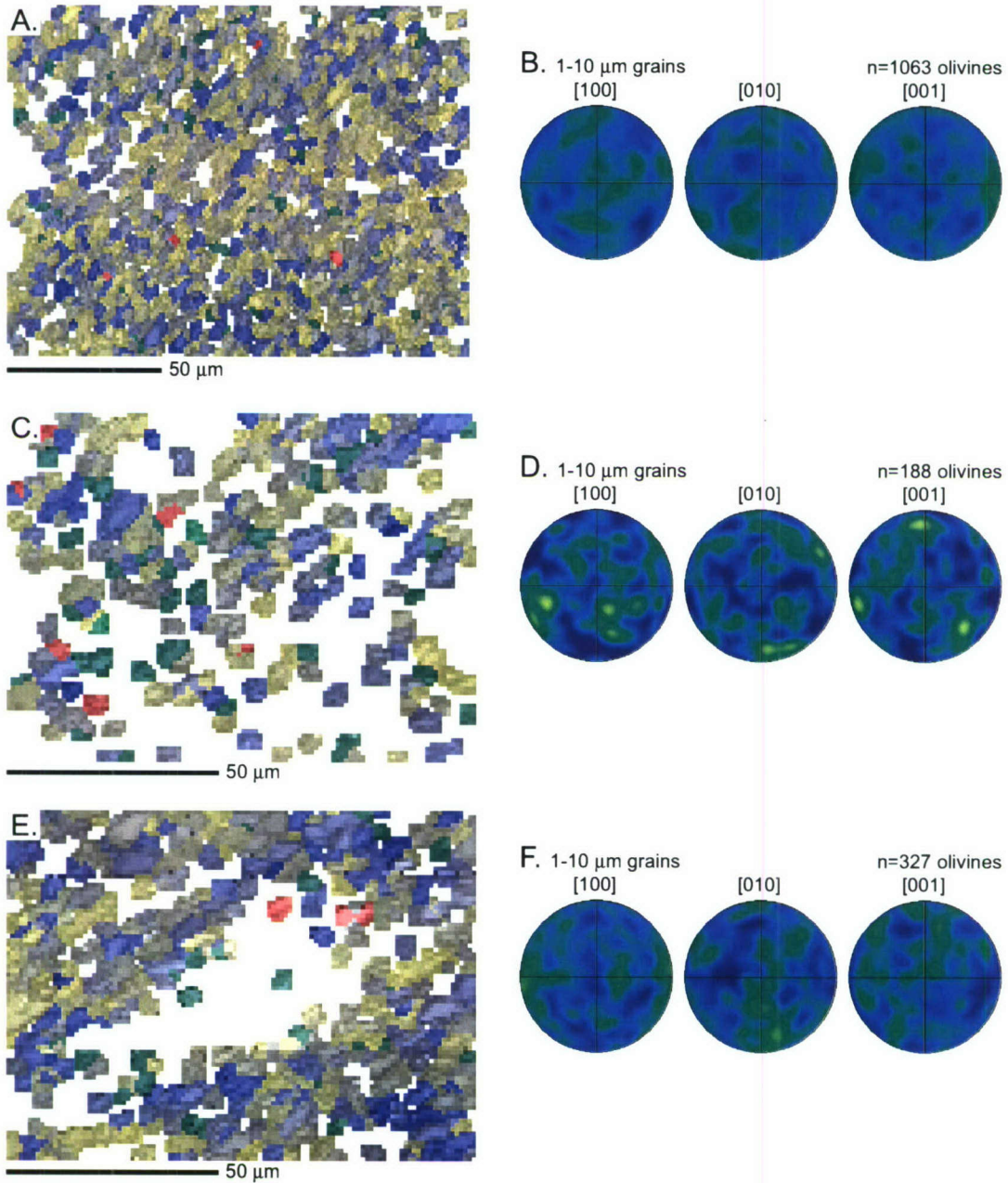


Figure 2.4: Orientation maps for three finer grained areas of the mylonite (A, C, E), using the same color scheme as in Fig. 2.3. The area in (A) is also shown in Fig. 2.2; the area in (C) was collected at a $1\ \mu\text{m}$ step size, with 57% indexed; the area in (E) was also collected at a $1\ \mu\text{m}$ step size, with 58% of the area indexed. Pole figures of one orientation per olivine are shown (B, D, F) for the 3 areas, all of which have grain sizes of 1-10 μm . The contour scale is the same as in Fig. 2.3, to a maximum level of 6 MUD. In the first area, the actual maximum is 2 MUD; in the second and third areas it is 3 MUD.

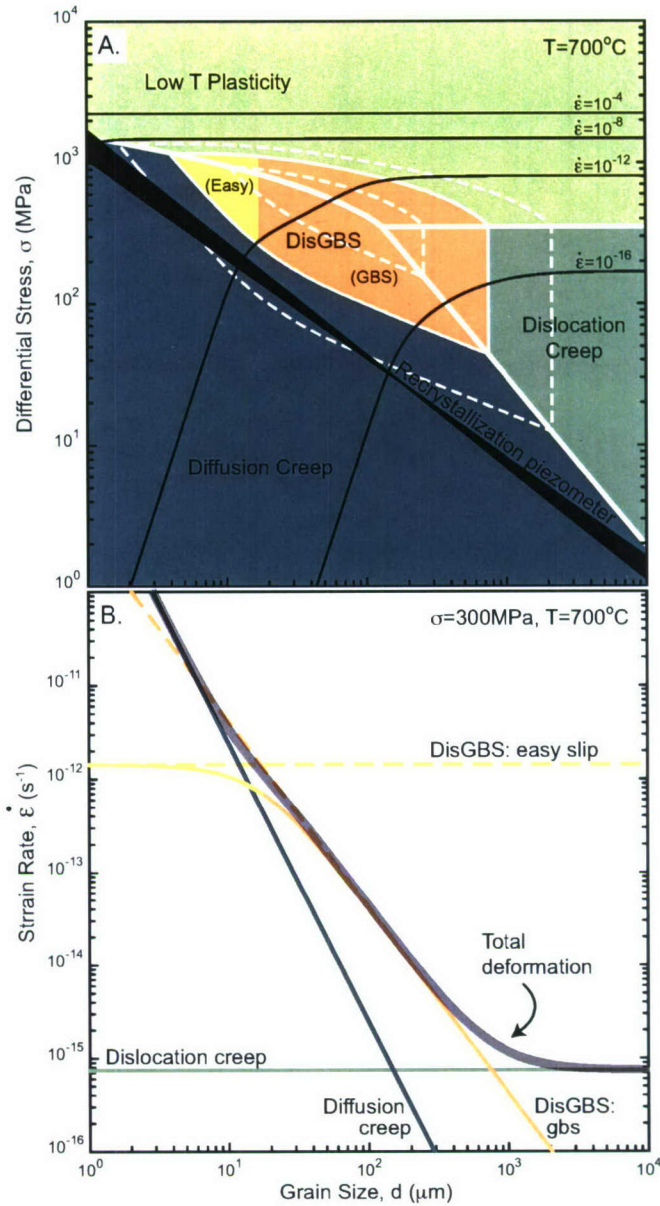


Figure 2.5: (A) An olivine deformation mechanism map, on axes of differential stress versus grain size, contoured for strain rate, plotted using the flow laws for dry, melt free olivine at 700°C and 12 km depth (Goetze, 1978; Evans and Goetze, 1979; Hirth and Kohlstedt, 2003). The black recrystallization piezometer field delimits the grain size predicted by the empirical piezometric relationship for olivine deformed under dry conditions (Karato et al., 1980; Van der Wal et al., 1993). The thick white boundaries between deformation fields indicate field boundaries in the absence of DisGBS. The thin white boundaries delineate the DisGBS field and the dashed boundaries on either side represent uncertainty in the location of the DisGBS boundaries due to uncertainty in its activation energy. The change in shading within the DisGBS field indicates the transition from the regime where creep on the easy slip system limits the strain rate to the regime where GBS limits the strain rate. (B) The variation of total and individual strain rates with grain size, at a differential stress of 300 MPa, and at 700°C and 12 km depth. The dashed lines represent the individual components of DisGBS.

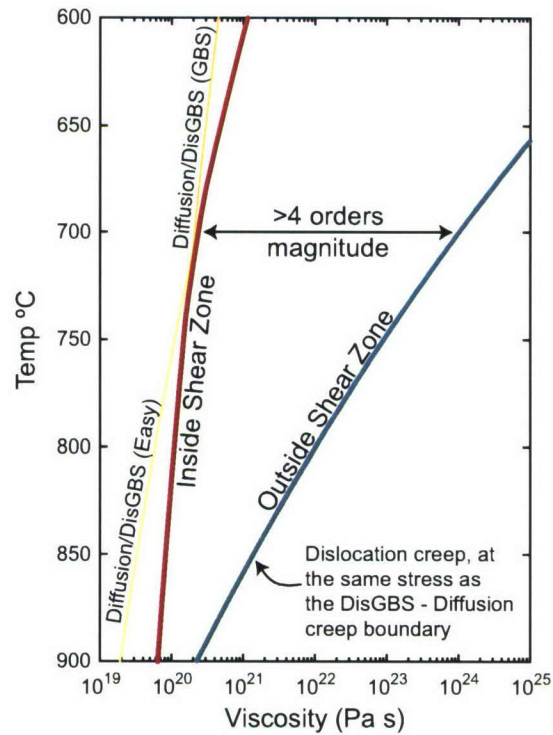


Figure 2.6: The variation of effective viscosity with temperature, for deformation within a shear zone compared to that outside a shear zone. Within the shear zone, viscosity is defined by deformation at the DisGBS-diffusion creep boundary at a total strain rate of 10^{-12} s^{-1} . Deformation outside the shear zone is by dislocation creep, calculated at the same stress as within the shear zone.

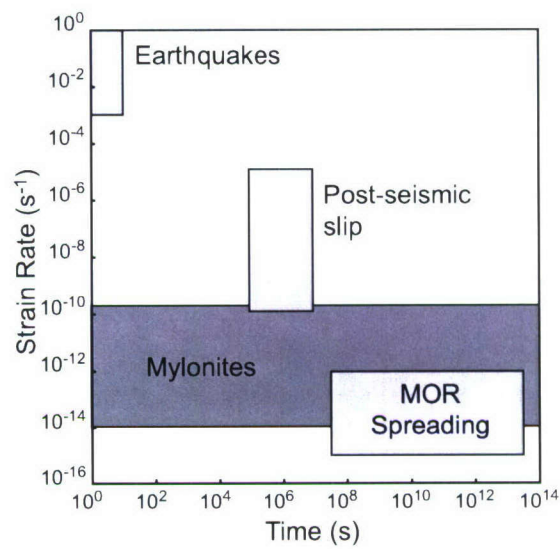


Figure 2.7: Duration versus estimated strain rate for the mylonite and for earthquakes, post-seismic slip and plate spreading. Earthquakes last 1-10 seconds, post-seismic slip lasts 1-100 days, and spreading events from 1 yr - 1 My.

Chapter 3

A Natural Example of Olivine LPO Variation With Shear Strain

Abstract

The variation of olivine lattice preferred orientation (LPO) as a function of shear strain is important for models that relate seismic anisotropy to kinematics of deformation in the mantle. We present results on the evolution of olivine orientation in natural samples as a function of increasing shear strain, for a shear zone in the Josephine Peridotite in southwest Oregon. We find that the LPO in harzburgites re-orient from a pre-existing LPO outside the shear zone to a new LPO with the olivine [100] maxima aligned sub-parallel to the shear direction between 168% and 258% shear strain. Fabric strengths, quantified using both the J- and M- indexes, do not increase with increasing shear strain. Both (010)[100] and (001)[100] are active in olivine at high strain, suggesting that re-alignment of olivine grains is accommodated by plastic flow during dislocation creep. Unlike experimental observations, our natural samples do not have a secondary LPO peak. We conclude that recrystallization by subgrain rotation dominates over grain boundary migration during fabric realignment in natural samples. The occurrence of girdle patterns among the [010] and [100] axes of harzburgites, but not dunites, suggests that secondary phases affect olivine deformation. We suggest that secondary phases limit olivine grain growth, promoting slip by grain boundary sliding. The activation of grain boundary sliding allows for enhanced activation of the two easiest olivine slip systems, resulting in [010] and [001] girdle patterns. Overall, our results provide an improved framework for calibration of LPO evolution models.

3.1 Introduction

Understanding olivine orientation as a function of shear strain is critical for quantifying relationships between the kinematics of deformation and the direction and magnitude of seismic anisotropy. For example, constraining the variation of olivine lattice preferred orientation (LPO) produced during simple shear is key to interpreting seismic anisotropy in terms of upper mantle convection (Hess, 1964; Nicolas and Christensen, 1987; Ribe, 1992; Mainprice and Silver, 1993; Blackman and Kendall, 2002). The relationships among olivine deformation, LPO development and seismic anisotropy have been examined experimentally (Nicolas et al., 1973; Zhang and Karato, 1995; Bystricky et al., 2000). Observations from these experiments have been used to place constraints on models (e.g., Wenk and Tomé, 1999; Tommasi et al., 2000; Kaminski and Ribe, 2001) that predict LPO development and thus upper mantle seismic anisotropy. Application of these models to deformation in the earth is improved by comparison of experimental results to rocks deformed under natural conditions, i.e., at lower stress and strain rate than can be achieved in laboratory experiments. To constrain deformation in the upper mantle, we analyzed the evolution of olivine LPO as a function of shear strain in naturally deformed peridotites from a shear zone in the Josephine Peridotite in southwest Oregon.

Mantle anisotropy results from ductile flow in the asthenosphere by dislocation creep, which produces alignment of elastically anisotropic minerals. Olivine and orthopyroxene, the dominant mineral phases in the upper mantle, have orthorhombic symmetry and are anisotropic (V_p anisotropies of 22% and 16%, respectively; Nicolas and Christensen, 1987). At upper mantle pressure and temperature conditions, they deform by dislocation creep, resulting in an LPO. Deformation is principally accommodated by slip on (010)[100] and (001)[100] in olivine and on (100)[001] in orthopyroxene.

Zhang and Karato (1995) carried out simple shear experiments on olivine aggregates at 1200°C and 1300°C over a range of shear strains to investigate olivine fabric evolution

during deformation. They found that the originally random fabric of their aggregates developed an LPO with the [100] maxima parallel to the flow direction by a shear strain of $\sim 150\%$, as had previously been suggested experimentally by Nicolas et al. (1973). While the Nicolas et al. (1973) experiments were performed in an axial geometry, bubbles in olivine grains aligned with the flow direction at high strain and were interpreted as having deformed by simple shear. Bystricky et al. (2000) demonstrated that the [100] alignment persists to high shear strains (500%).

The initial theoretical treatments of olivine LPO assumed that olivine grain orientations are controlled by finite strain (e.g., McKenzie, 1979). As slip on (010)[100] has the lowest activation energy (Bai et al., 1991), the olivine [100] axis was predicted to align with the finite strain ellipsoid (McKenzie, 1979; Ribe, 1992). However, experimental results (Nicolas et al., 1973; Zhang and Karato, 1995; Bystricky et al., 2000) indicate that only at low strains ($< 100\%$) does alignment of the olivine [100] axis coincide with the orientation of the finite strain ellipsoid. This alignment may be more a coincidence than an indication of control on the fabric by the strain geometry. The viscoplastic self-consistent model (VPSC; Lebensohn and Tomé, 1993; Tommasi et al., 2000) approaches alignment of the olivine [100] maxima with the flow direction, but does not achieve alignment due to the lack of dynamic crystallization in the model. Models that include dynamic recrystallization (e.g., Wenk and Tomé, 1999; Kaminski and Ribe, 2001) are able to produce alignment with the shear direction. For example, the DRex model (Kaminski and Ribe, 2001, 2002) achieves a good fit to the experimental data and is parameterized to predict the time-scale for LPO evolution.

We present data from peridotite samples that provide a guide for the extrapolation of experimental results (Nicolas et al., 1973; Zhang and Karato, 1995; Bystricky et al., 2000) to natural conditions. Studies of deformation in naturally deformed peridotites are often hindered by the lack of a well-defined finite strain marker. The Josephine Peridotite is

ideal for the analysis of fabric evolution with shear strain as it has a pre-existing foliation, defined by variations in pyroxene content, which provide a passive strain marker, as shown in Fig. 3.1. In addition, variations in pyroxene content permit assessment of the effects of secondary phases on olivine LPO development.

The Josephine Peridotite in southwestern Oregon is the mantle section of a ~ 150 Ma ophiolite from a fore-arc or back-arc setting (Dick, 1976; Harper, 1984; Kelemen and Dick, 1995). The peridotite is predominantly composed of harzburgite, with pyroxene-rich layers in some localities (Dick and Sinton, 1979). A series of shear zones, described by Loney and Himmelberg (1976) and Kelemen and Dick (1995), outcrop over a distance of 300 m in the Fresno Bench area of the Josephine Peridotite. The shear zones are defined by the sub-vertical to vertical transposition of the originally sub-horizontal lithological layering and by highly lineated pyroxene textures at the centers of the transposed layers (Fig. 3.1). The shear zones vary in width from ~ 1 m to 60 m and exhibit right lateral displacement with a component of NW-down vertical movement (Kelemen and Dick, 1995). Foliations at the shear zone centers strike $035\text{--}045^\circ$, with a maximum dip of 90° in the highest strain shear zones.

Maximum temperatures during deformation are constrained by syn-deformational magmatic features. Some of the shear zones cut or are cut by dunites, pyroxenites or gabbroic segregations, implying that temperatures during deformation may have been upwards of $\sim 1200^\circ\text{C}$. The lower temperature limit during deformation is constrained by geothermometry of coexisting pyroxene neoblast pairs in deformed harzburgites. Harding (1988) estimated a temperature range of $900\text{--}1100^\circ$, while Himmelberg and Loney (1973) estimated a temperature of $\sim 1000^\circ$, both from two pyroxene thermometry.

3.2 Methods

We analyzed olivine fabrics in harzburgites from the widest of the Josephine shear zones, shown in Fig. 3.1. The shear plane is approximately vertical, based on observations of how it cuts across topography along strike and the similar orientation of nearby shear zones with higher strains (Kelemen and Dick, 1995). Based on our field observations and those of Kelemen and Dick (1995), the shear plane is oriented at 035°/90° with a lineation plunge of 50°NE. Harzburgite and inter-layered dunite samples were collected on a NW-SE transect across the shear zone; the pyroxene layer orientation was measured wherever possible. In the geographic reference frame, the pyroxene layers dip 10°SW outside of the shear zone and reach a maximum dip of 75°SW at the shear zone center (Table 3.1).

A kinematic cross section of the shear zone is shown in Fig. 3.2A, oriented, with the X-axis defined as being parallel to the shear direction and the Z-axis defined normal to the shear plane. This X-Z frame of reference is used for the remainder of the figures. For the cross section, the field data are rotated and projected onto the plane 305°/50°NE, which lies perpendicular to the shear plane. In this kinematic reference frame, the pyroxene layers are at an angle of 78° to the shear plane outside of the shear zone and are rotated to an angle of 10° at the center of the shear zone (Table 3.1).

Strain across the shear zone is calculated from the change in pyroxene layer orientation in the kinematic reference frame, shown on the stereonet in Fig. 3.2B. Following the method of Ramsay and Graham (1970), shear strain, γ , is given by:

$$\gamma = \cot(\alpha') - \cot(\alpha) \quad (3.1)$$

where α is the initial angle of the pyroxene layering with respect to the shear plane and α' is the deflected angle, as shown in Fig. 3.2C. If the pyroxene layering was initially perpendicular to the shear plane ($\alpha=90$), shear strain would be calculated from the cotangent of

the deflection angle. However, as shown in Fig. 3.2 and Table 3.1, the layering is initially at an angle of 78° to the shear plane.

A maximum shear strain of 525% is reached at the center of the shear zone. The shear zone is 50-60 m wide, with a total displacement across the shear zone of 60 m, based on the area under a distance versus strain curve (Ramsay and Graham, 1970). The deflection of the pyroxene layers as a passive strain marker is not the same as the evolution of the finite strain ellipsoid with increasing strain, as shown in Fig. 3.2C. In addition, when the passive strain marker is not initially perpendicular to the shear plane, it is only coincident with the finite strain ellipsoid at high strain, as indicated in Table 3.1.

For fabric analyses, the Josephine samples were cut on the plane $305^\circ/50^\circ\text{NE}$. Thin sections were prepared with one edge parallel to 305° , so that all fabric data can be oriented with the X-axis parallel to the shear direction and the Z-axis normal to the shear plane. Grain size in two samples, shown in Fig. 3.3, was measured by the line intercept method (Underwood, 1970). We applied a geometric correction factor of 1.75 (Pickering, 1976), which is consistent with the Van der Wal (1993) olivine piezometric data. Average grain size was calculated using the geometric mean, as the grain size distribution is approximately log-normal and thus the geometric mean represents the size with the greatest frequency (Underwood, 1970).

Olivine LPOs were measured on polished thin sections using a JEOL 840 SEM with an electron backscatter diffraction (EBSD) detector and HKL Technology's Channel 5 software package. Thin sections were prepared for analysis by polishing with $0.03\mu\text{m}$ colloidal silica for >2 hours. To limit charging during EBSD analysis, thin sections were coated with gold, then polished for one minute to remove gold from grain surfaces, while leaving gold along cracks and grain boundaries. Samples were mapped for orientations and mineral phases at 40x magnification and 40-100 μm step sizes. Between 24 and 48 overlapping maps were made per thin section and these were combined into a single image using the

Channel 5 program MapStitcher.

EBSD maps (Fig. 3.4) have $\sim 50\%$ indexed data, following rejection of all points with a mean angular deviation (MAD) number $\geq 1^\circ$. The MAD number quantifies the mismatch between lattice planes in a calculated orientation and lattice planes quantified from the digitized diffraction pattern bands. The MAD number provides an indication of data quality, with high numbers resulting from surface roughness and computer mis-indexing. Data were further processed by removing wild spikes and replacing these, and points with zero solutions, with the most common neighbor orientation. Wild spikes are single pixels (i) which are misoriented by $> 10^\circ$ from the average orientation of the surrounding eight pixels and (ii) for which the maximum misorientation between any two of the surrounding eight pixels is $< 10^\circ$. See Warren and Hirth (2006) for a more detailed discussion of our EBSD data processing techniques.

Pole figures and inverse pole figures, shown in Figs. 3.4-3.6, are calculated using one point per grain. Pole figures are equal area lower hemisphere projections and inverse pole figures are equal area upper hemisphere projections. All datasets contain > 200 grains, as Ben Ismaïl and Mainprice (1998) have demonstrated that > 100 grain are necessary to provide robust estimates of fabric pattern and strength. Grain boundaries are defined by misorientations $\geq 10^\circ$ between adjacent points and subgrains by 2° - 10° misorientations.

3.3 Results

From analyses of nine samples across the Josephine shear zone, we find that the olivine [100] maxima, initially oriented at 62° to the shear plane, is aligned parallel to the shear direction at the center of the shear zone. To visually demonstrate the change in olivine orientation with strain, EBSD orientation maps and inverse pole figures of a low strain and a high strain sample are shown in Fig. 3.4. Olivine is colored as a function of the angle between the [100] axis and the shear plane. In the 65% strain sample, the majority of

grains are mid-blue in color, corresponding to a relatively high angle to the shear plane. In contrast, many grains in the 525% strain sample are dark blue, indicating alignment with the shear plane.

The inverse pole figures in Fig. 3.4B show the orientation of individual grains parallel (X) and perpendicular (Z) to the shear plane. At 65% strain, while considerable scatter exists in the distribution, the maximum density of points in the X-section is oriented 37° to [100]. In the Z-section, the maximum density is close to [001] with a low density around [010], suggesting that (001) is better aligned as the slip plane during the initial realignment of the fabric. At 525% strain, the highest density of points in the X-section is around [100]. In the Z-section, points cluster around [010] with scatter towards [001], indicating that both (010) and (001) are well oriented as slip planes. The Z-section also indicates that the same grain cannot be well oriented for slip on both (010) and (001).

Pole figures of olivine orientation are shown in Fig. 3.5 for the harzburgites and Fig. 3.6 for the dunite. Outside of the shear zone, the peridotite has a pre-existing LPO, with the olivine [100] maxima sub-parallel to the pre-existing foliation (also shown on the map cross section in Fig. 3.2A). In samples with shear strains up to 168%, the olivine [100] maxima remains inclined to the shear plane, with only a moderate rotation away from the original LPO (Fig. 3.5). Between a shear strain of 168% and 258%, the olivine LPO rotates so that the [100] maxima is sub-parallel to the shear plane. At higher shear strains, the [100] maxima remains sub-parallel to the shear plane.

The behavior of olivine [010] and [001] axes with increasing strain is more variable than the [100] axis (Fig. 3.5). Outside of the shear zone, (010) planes are sub-parallel to the pyroxene layering, suggesting that (010) was the dominant slip plane during the previous deformation event. At low strain, (001) is sub-parallel to the local layering, suggesting that (001) is initially the dominant slip plane during the fabric realignment. However, at high strain, (010) is sub-parallel to the transposed pyroxene layering. In addition, at high strain,

(010) and (001) in the harzburgites exhibit girdles, whereas in the dunite they approximate a single maxima (Fig. 3.6).

Grain size and shape remain relatively constant during deformation, despite the change in olivine LPO. Photomicrographs and grain size distributions of two samples, one at low strain and the other at high strain, are shown in Fig. 3.3. At low strain, pyroxenes are slightly elongated, with their long axes approximately aligned with the pyroxene layering and the olivine [100] maxima. Olivine grains are generally equant, with an aspect ratio (X:Z) of 1.1. At high strain, both orthopyroxenes and olivines are equant, with an olivine aspect ratio of 1.2. The lack of stretched grains indicates significant recrystallization occurred during deformation. The mean grain size is 0.57 mm in the low strain sample and 0.54 mm in the high strain sample. The grain size distributions in Fig. 3.3B are approximately log-normal, with recrystallization resulting in deviations from the log-normal distribution at small grain sizes (<0.5 mm).

The variation of the angle of the olivine [100] maxima to the shear plane with strain is compared to experimental results and models in Fig. 3.7. The angle of the olivine [100] maxima relative to the shear plane was determined using eigenvector analysis provided by the program PFch5.app (courtesy of D. Mainprice). The first eigenvector of the orientation tensor represents the mean direction of a crystal axis and is called the principal axis (Woodcock, 1977). We assume that this principal axis is more representative of the average [100] orientation than the location of the maximum density of data on the pole figure. In Fig. 3.7, we use the principal axis orientation to plot the variation of olivine [100] maxima with shear strain. In comparison to experiments, the Josephine samples are observed to require higher strain to align with the shear direction.

The strength of an LPO can be quantified using either the J-index (Bunge, 1982; Mainprice and Silver, 1993) or the M-index (Skemer et al., 2005), both of which are plotted for our samples in Fig. 3.8 as a function of shear strain. In addition, we plot the published

J-index values for the experimental datasets and models. Both indexes quantify overall fabric strength of all three olivine axes combined. The M-index quantifies the deviation of the uncorrelated misorientation angle distribution from a random crystal misorientation distribution (Skemer et al., 2005). Uncorrelated misorientation angles represent the angular difference in orientation (i.e. misorientation) between random grain pairs (i.e., not necessarily adjacent). The M-index varies between 0 for a random fabric and 1 for a single crystal. The J-index is a dimensionless characterization of the orientation distribution function (ODF) of crystal orientations as specified by Euler angles. It describes the distribution of Euler angle rotations away from a single crystal orientation, varying between 1 for a random LPO and infinity for a single crystal. In practice, the J-index has a maximum value of 250, as the ODF is truncated at degree 22. For our J-index calculations, we used the program SuperJctf.app (courtesy of D. Mainprice) with a 10° Gaussian half-width, data clustered in 1° bins and combined even and odd spherical harmonics.

In the Josephine samples, neither the M-index nor the J-index demonstrate a significant increase in fabric strength with strain. The J-index is relatively constant as a function of shear strain and typically in the range 5-8. The M-index initially increases in strength but then is relatively constant with an average value of 0.14. The only exception is the 386% strain harzburgite, which has a visibly weaker fabric in the pole figure (Fig. 3.5) and the lowest J- and M-index values.

3.4 Discussion

Our results on olivine LPO evolution during simple shear extend observations of LPO variations to lower stresses and strain rates than available from experimental datasets (Zhang and Karato, 1995; Bystricky et al., 2000). While our observations broadly agree with the experimental observations, our results suggest that a pre-existing LPO influences the strain necessary for LPO alignment with the shear direction. In addition, the pre-existing LPO

and the presence of secondary phases affect the behavior of olivine slip systems during deformation.

The orientation of the olivine [100] maxima as a function of shear strain in the Josephine shear zone is compared to the experimental datasets and models in Fig. 3.7. The experiments and models initially have a random fabric, whereas the [100] maxima in the Josephine harzburgites initially lies at an angle of 62° to the shear direction and has a J-index of 6.2. Also in the natural samples, the [100] maxima does not align with the shear direction until $\sim 250\%$ strain, whereas alignment occurs before 200% strain in the Bystricky experiments and $\sim 150\%$ strain in the Zhang and Karato experiments. Below, we compare our results in more detail to LPO evolution models and then discuss the effects of a pre-existing LPO, grain size and secondary phases on the behavior of olivine during deformation in the upper mantle.

3.4.1 Comparison to LPO evolution models

As a tool for predicting and interpreting seismic anisotropy, various theoretical models predict olivine LPO evolution during deformation (for a review, see Tommasi et al., 2000). The evolution of the olivine [100] axis with strain is shown for four models in Fig. 3.7. Two are end-member models that assume the olivine LPO follows either the shear direction or the finite strain ellipsoid (McKenzie, 1979; Ribe, 1992). The experimental datasets (Nicolas et al., 1973; Zhang and Karato, 1995; Bystricky et al., 2000) demonstrated that these end-member models do not accurately predict the evolution of olivine LPOs with shear strain, and the Josephine data agree with this conclusion. The best fit of the VPSC (Tommasi et al., 2000) and DRex (Kaminski and Ribe, 2001) models to the Zhang and Karato experiments are also shown in Fig. 3.7.

The VPSC model is a parameterization of the deformation of an olivine aggregate that treats each grain as an inhomogeneity embedded in a homogeneous effective medium

(Lebensohn and Tomé, 1993). The average stress and strain rate for all grains is constrained to be consistent with the equivalent macroscopic magnitudes. A reasonable match of VPSC pole figures to experimental pole figures is achieved by relaxing the strain compatibility requirement, but the evolution of the olivine LPO does not occur at low enough shear strain to be consistent with either the experiments or the natural samples. The VPSC line shown in Fig. 3.7 is for a model run to 350% shear strain with a dimensionless strain compatibility value of $\alpha=100$. Linear extrapolation to higher strain suggests that the [100] maxima will only align with the shear direction at $\sim 1000\%$ shear strain. However, the VPSC model is not well constrained at $>100\%$ strain, as it does not account for complexities associated with highly deformed grains (Blackman et al., 2002) or recrystallization.

To obtain a better match to experimental data, Wenk and Tomé (1999) and Kaminski and Ribe (2001) developed models that include dynamic recrystallization, which has been demonstrated to occur at high strain and temperature (Carter, 1976; Karato, 1988). In both models, recrystallization is treated as a balance of grain boundary migration (relatively undeformed grains replace highly deformed grains) and grain nucleation (highly deformed grains nucleate strain-free subgrains with the same orientation). DRex (Kaminski and Ribe, 2001) predicts the deformation of an olivine aggregate by defining a local velocity gradient tensor for each grain and a macroscopic velocity gradient tensor. A good fit to the experimental data is provided by optimizing the dimensionless grain boundary migration (M^*) and grain nucleation (λ^*) parameters. For $M^*=200$ and $\lambda^*=5$, the [100] maxima aligns with the flow direction by 100% strain and pole figures are in good agreement with the Zhang and Karato experiments. DRex is a computationally simpler model than VPSC, but relies on two adjustable parameters, whereas VPSC has only one. In addition, the relative stress necessary for activation of different slip systems is specified in all models based on experimental data (e.g., Bai et al., 1991).

In detail, the Josephine shear zone data do not agree with predictions from either VPSC

or DRex, which were both optimized to fit the Zhang and Karato experiments (Fig. 3.7). The transition to a shear aligned fabric in the Josephine harzburgites occurs at significantly lower strain than predicted by VPSC, due to the absence of dynamic recrystallization in VPSC. DRex has a similar rate of change of LPO with strain, when compared to the Josephine samples. However, the change occurs at lower shear strain in DRex, probably due to the initially random LPO in DRex. The predicted DRex pole figures have more pronounced maxima than the Josephine samples, which is likely a result of the absence of secondary phases in DRex (discussed below). A more recent version of DRex allows for incorporation of orthopyroxene (Kaminski et al., 2004), but the effect of orthopyroxene on pole figure patterns has not yet been explored.

3.4.2 Active slip systems and the pre-existing LPO

We suggest that the high strain necessary for [100] alignment in the Josephine samples is related to the high initial angle of the [100] maxima to the shear plane. The olivine slip system (001)[100] dominates at low strain, which probably relates to the initial orientation of (001) with respect to the principle compressive stress, σ_1 . If σ_1 is assumed to be at 45° to the shear plane and in the plane of deformation, then the [100] maxima is calculated to have an angle of 78° to σ_1 and to lie outside of the plane of deformation. The olivine [010] and [001] maxima are calculated to have initial angles of 41° and 53° to σ_1 , respectively. Hence both (010) and (001) are well aligned as slip planes, but [100] is poorly aligned as the slip direction. Movement along any other glide direction apart from [100] in olivine requires higher stress or high water contents (e.g., Jung and Karato, 2001). Re-alignment of the [100] maxima into the plane of deformation and parallel to the shear direction requires dislocation creep and recrystallization. We conclude that the initial angle of the [100] maxima affects the amount of strain necessary for re-alignment. For example, if the olivine [100] maxima was initially at $<45^\circ$ to σ_1 and the shear direction, we anticipate that only a

small amount of strain would be necessary for alignment with the shear direction.

The behavior of the olivine [010] and [001] axes with increasing strain suggests that the dominant slip system in olivine varied during deformation. The orientation of the [100] maxima does not vary significantly until it changes between 161% and 258% strain (Fig. 3.5). In contrast, (010) is well aligned for slip outside the shear zone (in the previous kinematic reference frame), whereas at low strain (001) is better aligned. Our observations suggest that from 65% to 131% strain, the dominant slip system is (001)[100], instead of the “typical” (010)[100] system. At strains $>168\%$, both slip systems are equally active, indicated by variable amounts of girdling in the [010] and [001] pole figures (Fig. 3.5).

Variation of the dominant olivine slip plane between (010) and (001) suggests that re-alignment of the LPO is accommodated by rotation during slip on these two planes, combined with recrystallization. In contrast, in the Zhang and Karato experiments, [010] and [100] are observed to rotate around the perimeter of the pole figure with increasing strain, while [001] is generally inactive in all experiments (Zhang et al., 2000). Their results suggest that alignment of [100] with the shear direction is accomplished predominantly by grain rotation and slip on (010)[100]. In contrast, in the Josephine samples, the fact that [100] and [010] do not rotate around the perimeter of the pole figure indicates that LPO re-alignment is not accomplished simply by rigid grain rotation. Instead, we suggest that the LPO re-alignment occurs by internal rotation within the olivine crystal during dislocation creep on (010)[100] and (001)[100].

The presence of a pre-existing LPO is particularly relevant to modeling anisotropy during corner flow beneath ridge axes and subduction zones. The rate at which LPO re-aligns has been included in more recent versions of DRex (Kaminski and Ribe, 2002; Kaminski et al., 2004) and in recent mantle flow models that predict anisotropy (e.g., Conrad et al., 2007). Our results can be used in these models to better constrain the amount of strain necessary for LPO alignment. While the upwelling mantle may initially have no fabric, during

active upwelling it will develop an LPO with the olivine [100] axis vertically aligned (e.g., Blackman et al., 1996). After corner flow, the LPO will be poorly oriented with respect to σ_1 , requiring more strain for the [100] axis to align horizontally than if it formed from a random fabric. Alternatively, as the mantle undergoes corner flow, the change in olivine orientation may be able to track the change in σ_1 orientation, depending on the rate at which LPO re-aligns versus the rate of plate spreading.

3.4.3 LPO Strength

Fabric strength, or the amount to which minerals align, is another important parameter for constraining seismic anisotropy. In Fig. 3.8, the strengths of Josephine LPOs are compared to the experimental datasets and model predictions. The Zhang and Karato experiments have similar fabric strengths to the Josephine samples at 1200°C, but at 1300°C the experiments reach stronger fabrics at higher strain. Comparison of the pole figures for the high temperature experiments (Zhang et al., 2000) to the Josephine samples reveals that the strengths of the [100] peaks are similar, but that the experimental samples have much stronger [010] and [001] fabrics. Hence, the rapid increase in J-index with shear strain observed in the experiments is a result of alignment of the [010] and [001] axes. In the Josephine samples, [010] and [001] tend to have girdled patterns, leading to weaker J-indexes.

The high strain torsion experiments of Bystricky et al. (2000) have increasing J-indexes with strain, but at a lower rate than observed in the Zhang and Karato experiments. The maximum J-index value among the Bystricky experimental samples is similar to the maximum value among the Josephine natural samples. However, the Bystricky experiments suggest that fabric strength will continue to increase with increasing strain, whereas the Josephine samples do not. These differences may be due to a combination of an initially random fabric and a lack of secondary phases in the experiments.

Both DRex and VPSC predict rapidly increasing fabric strength with shear strain. While the models initially have a random fabric and hence the fabric strength must increase at low strain when an LPO forms, the continued increase in the models does not match our observations. In the limit of infinite strain, the models would reach a single crystal fabric that has not been observed in natural samples. We suggest that this discrepancy is due to the fact that the models do not account for all processes occurring during deformation. For example, the presence of secondary phases in peridotites is one limitation on fabric strength, as it promotes slip on multiple systems.

The M-index is a more recently developed quantification of fabric strength than the J-index (Skemer et al., 2005) and has not been calculated for the majority of published data. Both indexes follow the same general trend with shear strain in Fig. 3.8. In Fig. 3.9, we compare the M-index to the J-index for the Josephine samples. The two indexes cannot be directly related as they are based on different parameterizations of crystal orientation. A linear least squares regression through the dataset produces a reasonable correlation between the J-index and M-index, with a correlation coefficient of 0.7. Improved understanding of these indexes of fabric strength requires more published datasets to include values for both the J- and M-indexes.

Overall, we observe lower LPO intensities than predicted by the VPSC and DRex models. The match is better for the experimental datasets, with the exception of the high temperature, high strain experiments of Zhang and Karato (1995). These differences indicate that the models do not replicate all aspects of the natural environment. However, seismic properties are only weakly dependent on LPO intensity (e.g., Tommasi et al., 2000). For the prediction of seismic anisotropy, the rate at which olivine aligns with the shear direction is more important than the fabric strength which is produced.

3.4.4 Grain size variation and recrystallization

Grain size reduction does not occur during the transition from low strain to high strain in the Josephine samples. Both low and high strain samples have a grain size of 0.5-0.6 mm and nearly equant grains (Fig. 3.3). The absence of stretched grains indicates that dynamic recrystallization occurred during deformation. This further supports the observation from the olivine pole figures (Fig. 3.5) that the LPO realignment was not the result of the rigid rotation of olivine grains. Our observations are in contrast to the experiments (Zhang and Karato, 1995; Zhang et al., 2000; Bystricky et al., 2000), which at high strain have stretched and elongated grains. We suggest that these differences are related to the higher differential stress used for the experiments, and the difference between initial and steady-state grain size in the experiments.

The similar grain size at low and high strain in the Josephine samples indicates relatively constant stress during deformation. In addition, stress must be continuous across the shear zone. The olivine piezometer (Karato et al., 1980; Van der Wal et al., 1993) relates dynamically recrystallized grain size to stress. As shown in Fig. 3.10, the piezometer indicates that for a grain size of 0.5-0.6 mm, differential stress during deformation was ~ 10 MPa. The experimental datasets are also plotted in Fig. 3.10, for which the stress was measured during the experimental run (Zhang et al., 2000; Bystricky et al., 2000). The grain size estimates from Zhang et al. (2000) have been adjusted to the same geometrical correction factor as the Van der Wal piezometric dataset (Van der Wal, 1993) and show reasonable agreement with the piezometer. Differential stress estimated from the piezometer can be used to estimate the strain rate during deformation of the Josephine peridotite. Using the olivine flow law for dislocation creep (Bai et al., 1991; Hirth and Kohlstedt, 2003) at a temperature of 1100°C, we estimate a strain rate of 10^{-12} s^{-1} and a viscosity of 10^{19} Pa s .

In the experimental datasets (Zhang and Karato, 1995; Bystricky et al., 2000), a second

LPO maxima – [100] aligned perpendicular to σ_1 – is observed. Lee et al. (2002) relate this peak to grains that are poorly oriented for slip and thus have low dislocation densities. These grains grow by grain boundary migration at the expense of well-oriented grains that have developed high dislocation densities. The subset of grains with [100] aligned with σ_1 is not observed in the Josephine samples (Figs. 3.5 and 3.6). The important role of grain boundary migration in the experiments is likely due to the higher differential stress, which results a larger driving force for grain boundary migration. In contrast, at low stress, we suggest that subgrain rotation is a more important process for fabric development than grain boundary migration.

In the DRex model, the parameter M^* controls grain boundary mobility during LPO formation. At $M^*=0$, the LPO predicted by DRex never aligns with the shear plane, but instead aligns with the finite strain ellipsoid. For $M^*>0$, the olivine [100] maxima aligns with the shear plane, with decreasing amounts of shear strain necessary for alignment as M^* increases. The best fit of DRex to the 1300°C Zhang and Karato experimental data is achieved when $M^*=200$, as shown in Fig. 3.7. The strain at which [100] aligns in the Josephine samples is consistent with a M^* value of ~ 50 . However, Kaminski and Ribe (2001) also found a second [100] maxima when $M^*=50$, observed in the experimental samples but not the Josephine.

None of the current set of models for LPO formation accurately replicate the microstructural observations from the Zhang and Karato experiments. In the experiments, the secondary [100] maxima results from the growth of grains that have low dislocation densities and are poorly aligned for slip. In the DRex model (Kaminski and Ribe, 2001), the secondary peak is produced by the formation of recrystallized grains perpendicular to σ_1 that are well oriented for slip. In the Wenk and Tomé (1999) model, the secondary maxima occurs when grain boundary migration dominates over grain nucleation, similar to the Zhang and Karato experiments. However, the primary [100] maxima is formed by

nucleation of grains with low dislocation densities, in contrast to the Zhang and Karato experiments (e.g., Lee et al., 2002). In the VPSC model, which does not include dynamic recrystallization, the secondary [100] maxima is not observed, but alignment of the [100] maxima with the shear direction also does not occur. Intriguingly, the model of Etchecopar and Vasseur (1987) produces a single [100] maxima aligned with the shear plane. However, this model does not accurately treat strain compatibility, as it includes only one olivine slip system and assumes recrystallization of all grains to minimize grain misfits.

3.4.5 Effect of secondary phases

In the Josephine samples, girdle patterns are observed among the high strain harzburgites, whereas the high strain dunite has point maxima (Figs. 3.5-3.6). In the VPSC model, girdled patterns are produced by relaxing the strain compatibility requirement (e.g., $\alpha=100$) and equating the critical resolved shear stress for slip on (010)[100] and (001)[100]. Similarly, DRex produces a girdle pattern by allowing these two slip systems to have equal softness, whereas a point maxima is only produced when slip on (001)[100] is twice as hard as slip on (010)[100]. These observations suggest that the secondary phases (pyroxenes and spinel) in the Josephine harzburgites provide a mechanism for promoting dominant slip on both (010)[100] and (001)[100], by relaxing the strain compatibility requirement.

We suggest that strain compatibility in the harzburgites is effectively relaxed by a reduced grain size that leads to a component of dislocation-accommodated grain boundary sliding (DisGBS). Secondary phases limit olivine grain growth in the harzburgites, as has been observed in mylonites (Warren and Hirth, 2006) and in Oman Ophiolite harzburgites and dunites (Braun, 2004). At a smaller grain size, some deformation is by DisGBS, which accommodates deformation by the movement of dislocations within crystals and by grain boundary sliding (Hirth and Kohlstedt, 2003; Drury, 2005; Warren and Hirth, 2006). Following (Braun, 2004), we conclude that grain boundary sliding during DisGBS relaxes the

requirement for slip on the hardest olivine slip system, (010)[001], as otherwise required by the von Mises criterion (von Mises, 1928). Consequently, a significant component of slip is accommodated by (001)[100], which has only a slightly larger critical resolved shear stress with respect to the (010)[100] slip system (Bai et al., 1991). In contrast, deformation in the dunite is by dislocation creep due to a slightly larger grain size and thus activity of the hard slip system (010)[001] limits activity on (001)[100].

Evidence for relaxation of the strain compatibility requirements is also observed in the Bystricky et al. (2000) torsion experiments, which have girdled pole figure patterns. The high stress and strain rate of the experiments results in dynamic recrystallization to a grain size of 3-4 μm , within the field for dominant deformation by DisGBS (Hirth and Kohlstedt, 2003; Drury, 2005; Warren and Hirth, 2006). In this situation, secondary phases are not necessary to limit grain size during the experiment.

3.5 Conclusions

At low strain in a shear zone in the Josephine Peridotite, the olivine LPO is at an angle of $\sim 50^\circ$ to the shear direction, whereas at $>258\%$ strain, the olivine [100] maxima is aligned with the shear direction. Our results on olivine LPO evolution during shear are consistent with the conclusion from experimental data (Zhang and Karato, 1995; Bystricky et al., 2000) that olivine LPO aligns with shear direction during deformation. However, the naturally deformed samples require higher shear strains for alignment to occur and fabric strength does not increase with shear strain. Consequently, the Josephine shear zone data do not agree in detail with predictions from polycrystal plasticity models (Wenk and Tomé, 1999; Tommasi et al., 2000; Kaminski and Ribe, 2001), which have been fit to the experimental datasets.

In comparison to the experimental datasets and the model predictions, the natural samples have (i) a pre-existing LPO, (ii) no significant variation in LPO strength, (iii) secondary

phases in addition to olivine, and (iv) recrystallization without grain size reduction at high strain. The higher strain necessary for LPO alignment in the Josephine samples appears to be related to the orientation of the pre-existing LPO. The realignment of the olivine grains is not simply a passive rotation, due to the observation of slip on both (010)[100] and (001)[100] during the fabric realignment. We suggest that the initial activity of (001)[100] is due to the orientation of the pre-existing LPO, whereas at high strain, activation of (001)[100] is facilitated by the presence of additional phases in the peridotite. We also suggest that secondary phases pin olivine grain size, leading to deformation by DisGBS and similar amounts of activity on two olivine slip systems. The lack of significant grain size reduction in the Josephine samples indicates that stress was relatively constant during deformation. In addition, from comparison to the models and from thin section observation, we conclude that recrystallization by subgrain rotation dominates over grain boundary migration in the natural setting. Our results extend the observations of olivine LPO evolution to much lower stresses and strain rates than are accessible experimentally and indicate the importance of a pre-existing fabric and secondary phases for the evolution of anisotropy with shear strain.

3.6 Bibliography

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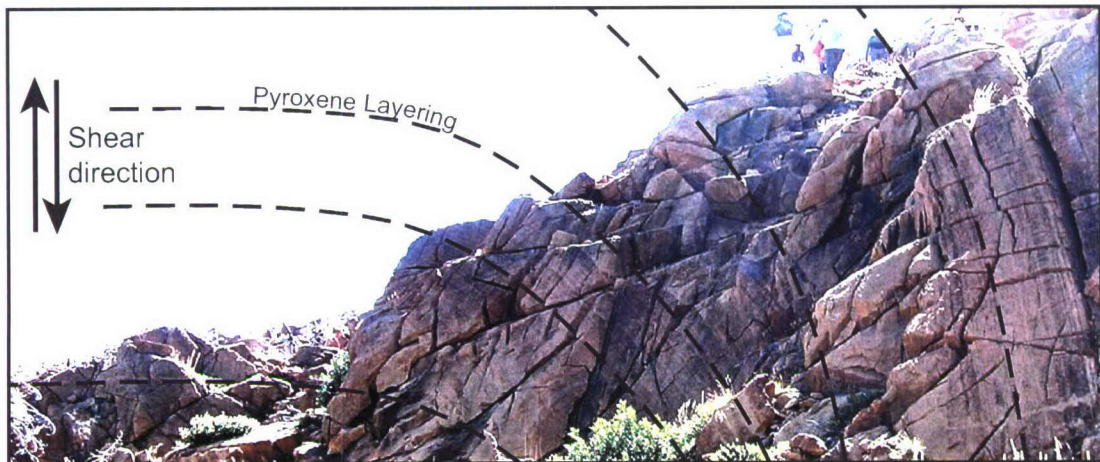


Figure 3.1: Photo of deformed layers in a Josephine shear zone, with the trace of the pyroxene layers outlined. Deflection of the regional pyroxene layering by right lateral shear provides a passive marker of strain.

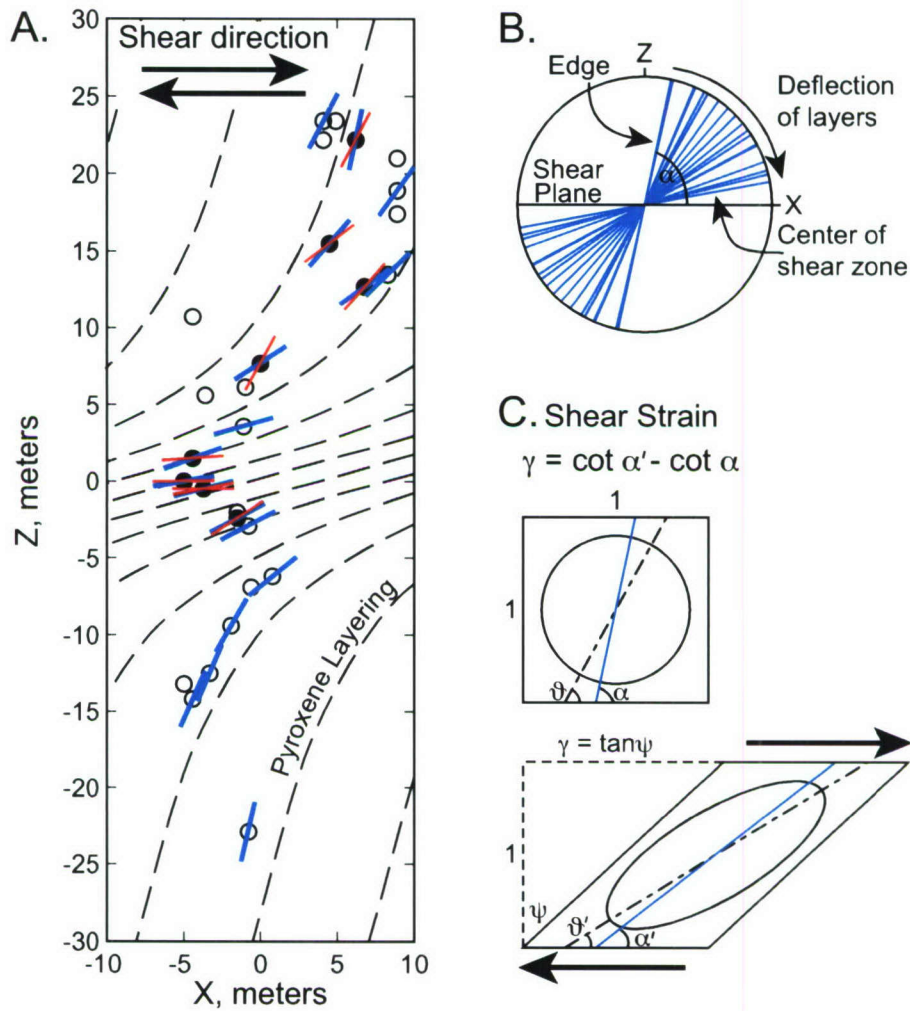


Figure 3.2: **(A)** X-Z cross-section of the shear zone constructed from field data of sample locations and the strike and dip of pyroxene layers. This map is in the kinematic reference frame, perpendicular to the shear plane and parallel to the shear lineation, represented by the plane $305^\circ/50^\circ$. Circles indicate sample locations, with analyzed samples indicated by filled circles. The pyroxene layer orientations are shown by blue lines and the orientations of olivine [100] maxima (from EBSD data) by red lines. **(B)** Stereonet of the variation of pyroxene layer orientations with respect to the shear plane. To represent the true deflection of a passive strain marker by shear deformation, the data have been rotated and projected onto the plane perpendicular to the shear plane, as in the map cross-section. The angle α is the initial angle of the pyroxene layering outside the shear zone. **(C)** The geometrical relationship of shear strain, γ , to the orientation of a marker layer, which initially lies at an angle, α , to the shear plane and is deflected to a smaller angle, α' . The orientation of the finite strain ellipsoid long axis, represented by deflection through the angle θ , is not coincident with the marker layer. Diagram adapted from Ramsay and Graham (1970).

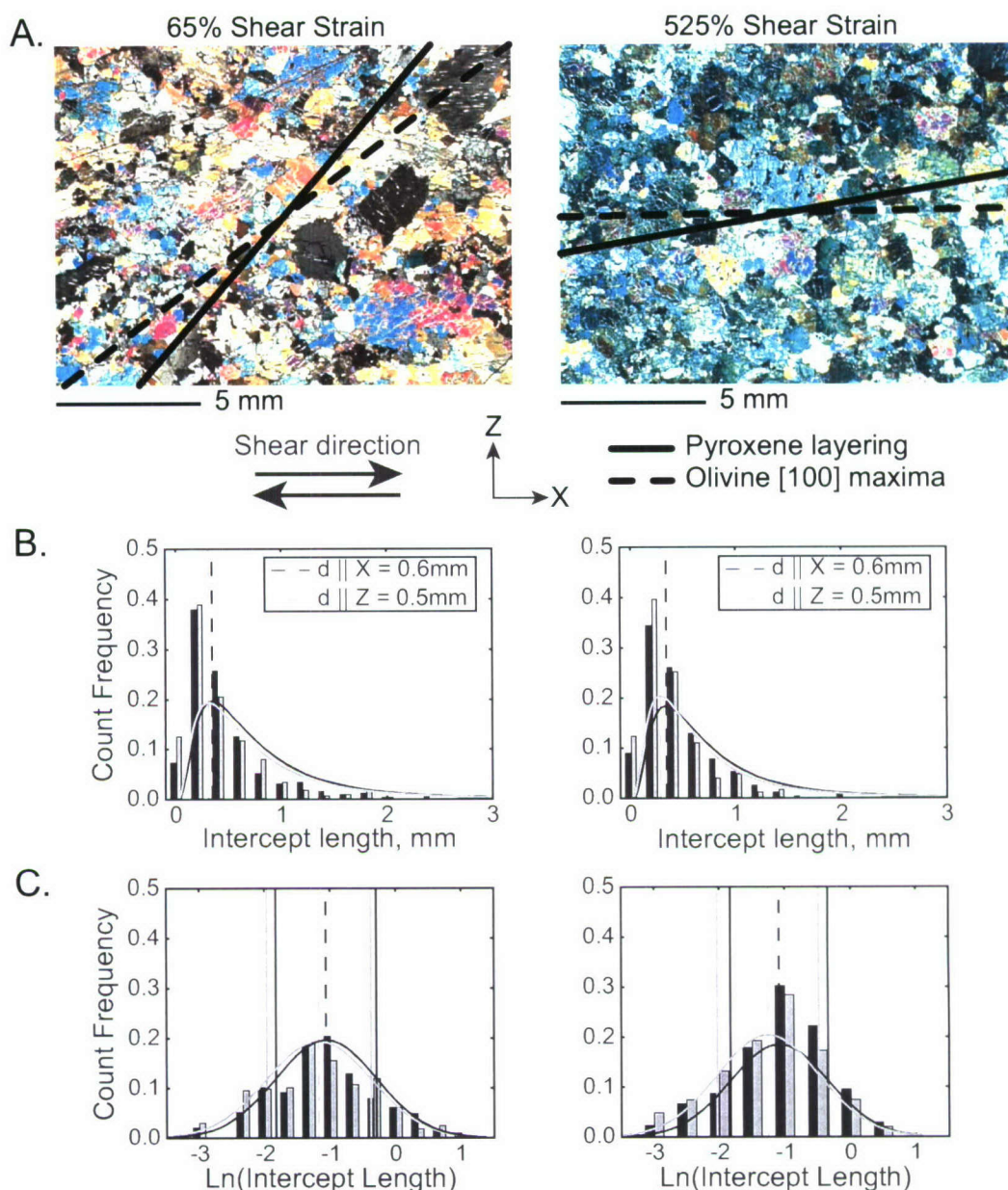


Figure 3.3: **(A)** Cross-polarized photomicrographs of low strain (65%) and high strain (525%) harzburgites. Solid lines indicate field-measured orientations of pyroxene layers and dashed lines indicate EBSD-determined orientations of olivine [100] maxima. Note that the high strain sample has a higher degree of alteration, especially among pyroxenes. **(B)** Histograms of grain intercept lengths parallel (black) and perpendicular (grey) to the shear direction. Dashed lines indicate the geometric mean intercept length and solid curves indicate the theoretical log-normal distribution. The average grain size (d) is calculated from intercept length using a geometric conversion factor of 1.75 (Pickering, 1976). **(C)** Log-normal histograms of intercept length, with geometric mean intercept length (dashed lines), 1σ deviations (solid lines) and theoretical distributions (solid curves) also shown.

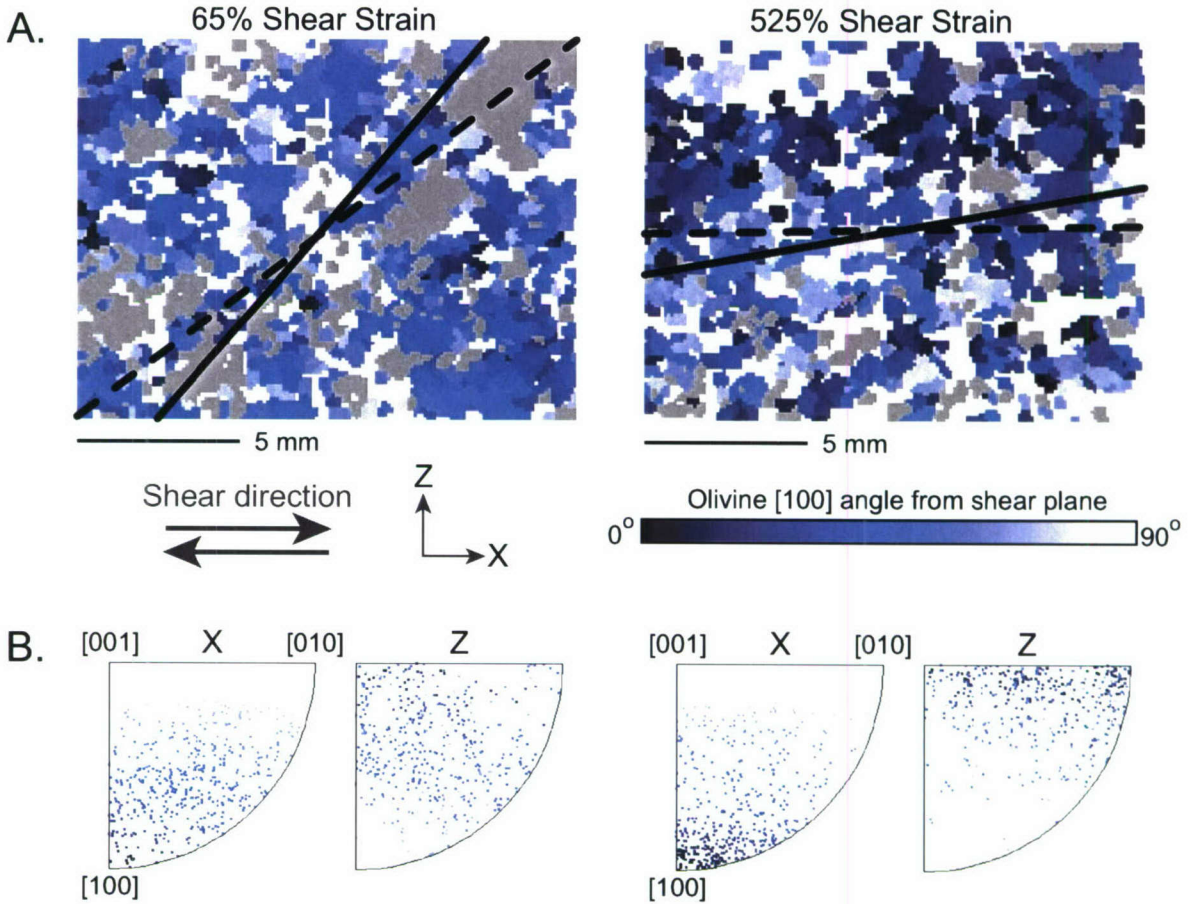


Figure 3.4: **(A)** EBSD maps of the same sample areas as shown in Figure 3.3. Grey indicates secondary phases (orthopyroxene, clinopyroxene, and spinel) and white areas have no data. Olivine is shaded as a function of the orientation of the [100] axis from the shear plane. Dark grains are aligned closely with the shear plane whereas light grains are at a high angle to the shear plane. The solid line is the field-measured orientation of the pyroxene layers and the dashed line is the EBSD-determined orientations of the olivine [100] maxima. **(B)** Inverse pole figures for olivine, demonstrating orientations parallel (X) and perpendicular (Z) to the shear plane. At low strain, grains are oriented with their axes at an angle to both the X and Z directions. At high strain, the majority of grains are oriented with [100] parallel to X and either [010] or [001] parallel to Z.

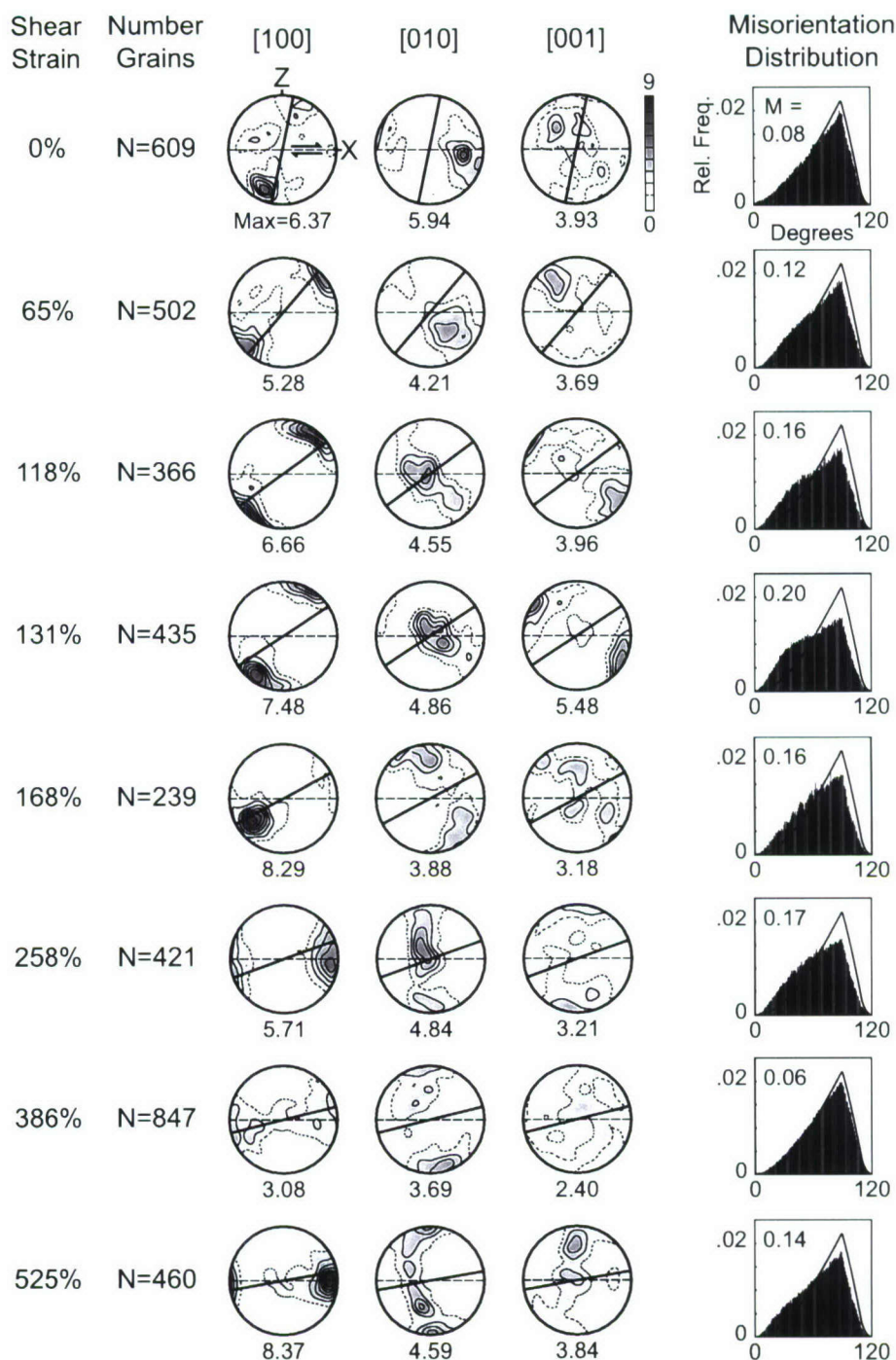


Figure 3.5: Olivine pole figures for harzburgites, with the shear plane (dashed line) parallel to X and pyroxene layering indicated by a solid line. Contouring is multiples of a uniform distribution (MUD), with a dashed line at 1 MUD. Maximum MUD values are identified below each individual pole figure. Misorientation distributions are for uncorrelated angles, with M-index values indicated. The solid line is the theoretical orthorhombic random distribution (Grimmer, 1979).

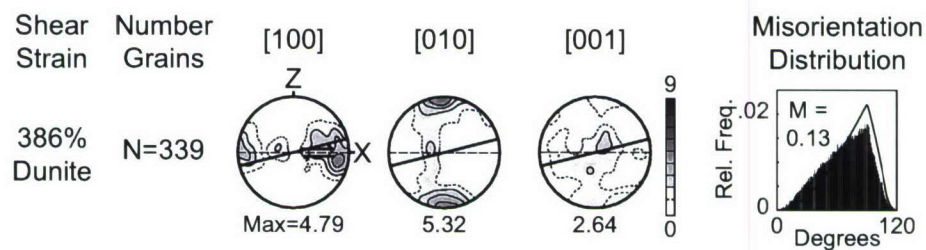


Figure 3.6: Olivine pole figure and misorientation distribution for a high strain dunite sample. As in Figure 3.5, the pole figure is oriented with the shear plane (dashed line) parallel to X, the pyroxene layering indicated by a solid line, and contour from 0 to 9 MUD. In contrast to the high strain harzburgites, the dunite has more pronounced [010] and [100] maxima.

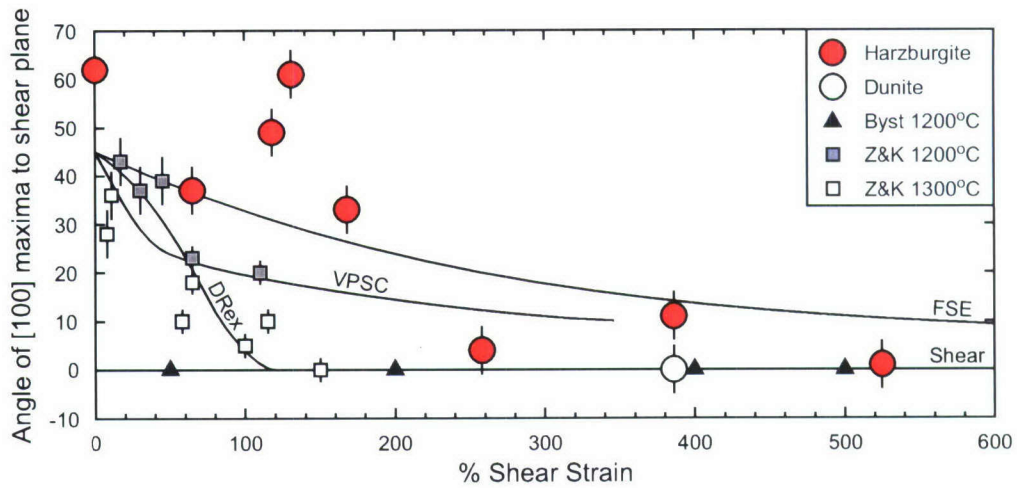


Figure 3.7: Angle of the olivine [100] maxima to the shear plane as a function of shear strain in the Josephine peridotites, experiments and models. The Josephine harzburgites are shown as filled circles and the dunite as an open circle. The models and experiments initially have a random fabric, represented by an average angle of 45° to the shear direction. The experimental data are from Bystricky et al. (2000) and Zhang and Karato (1995). The simplest models are FSE, which follows the finite strain ellipsoid and Shear, which follows the shear direction. VPSC is the best fit ($\alpha=100$) of the viscoplastic self-consistent model (Tommasi et al., 2000) to the experiments. DRex is the best fit ($M^*=200$) of the dynamic recrystallization model to the experiments (Kaminski and Ribe, 2001). Similar results to DRex were reported by Wenk and Tomé (1999) using a VPSC model which included recrystallization.

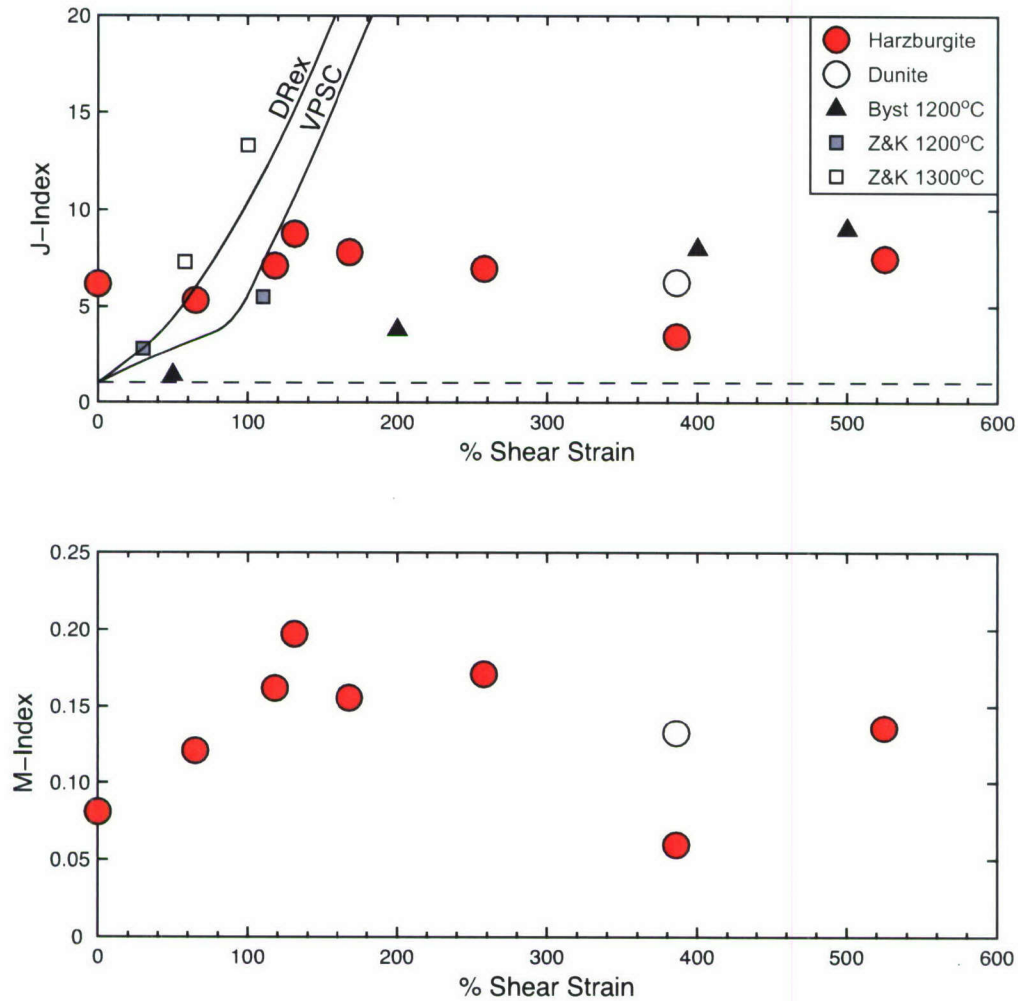


Figure 3.8: **(A)** Variation in the J-index as a function of shear strain. Dashed line indicates the theoretical lower limit (i.e. a random fabric) for the J-index. The results for the Josephine harzburgites are shown as filled circles and the dunite as an open circle. Also shown are the Bystricky et al. (2000) high strain experiments, the Zhang and Karato (1995) experiments (from the J-index calculation by Tommasi et al., 2000), the VPSC model ($\alpha=100$; Tommasi et al., 2000) and the DRex model ($M^*=200$; Kaminski and Ribe, 2001). **(B)** Variation in the M-index as a function of shear strain in the Josephine samples. The M-index varies between 0 for a random fabric and 1 for a single crystal fabric (Skemer et al., 2005).

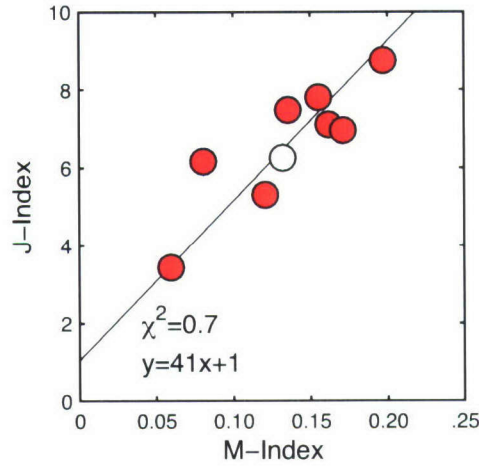


Figure 3.9: Variation of the J-index versus the M-index for the Josephine samples. Filled circles are harzburgites and the open circle is the dunite. The line is a minimum least squares regression through the dataset.

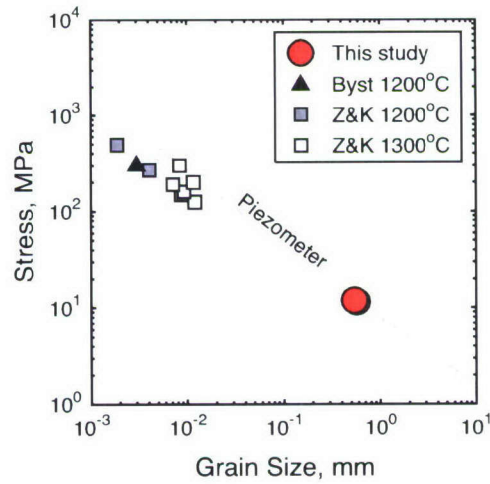


Figure 3.10: The olivine piezometer – the variation of stress with grain size – as determined by experimental data (Karato et al., 1980; Van der Wal et al., 1993). The piezometer is used to estimate differential stress in the shear zone from the grain size of the Josephine samples at 65% and 525% shear strain. Also shown are the Zhang and Karato experiments, from the analysis by Zhang et al. (2000), and the Bystricky et al. (2000) experiments. The Zhang and Karato dataset has been re-adjusted to a geometric correction factor of 1.75 (Van der Wal, 1993), for consistency with our results and the piezometer.

Table 3.1: Sample locations, strikes and dips, and the results of strain and fabric analyses.

Sample ^a		Dist. (m)		Field Strike/Dip	Rot&Proj ^b Strike/Dip	Shear Strain	Angle ^c		[100]	Fabric Strength	
		X	Z				α	θ		J-index	M-index
3923J01	H	6.2	22.2	245/10	192/90	0	78°	n/a	62°	6.2	0.08
3923J02	D	4.9	23.4								
3923J03	H	8.9	18.9	210/30	216/90	51%	54°	38°			
3923J04	D	8.9	17.4								
3923J05	D	8.9	21.0								
3923J06	H	4.1	22.2								
3923J07	D	4.1	23.4	200/25	208/90	32%	62°	40°			
3923J08	D	8.3	13.5								
3923J09	H	8.3	13.5	210/40	226/90	81%	44°	34°			
3923J10	H	4.5	15.5								
3923J11	H	4.5	15.5	210/35	221/90	65%	49°	36°	37°	5.3	0.12
3923J12	D	6.8	12.7								
3923J13	H	6.8	12.7	215/47	234/90	118%	36°	30°	49°	7.1	0.16
3923J14	H	0.0	7.7	210/52	237/90	131%	33°	28°	61°	8.8	0.20
3924J01	H	-3.6	5.6								
3924J02	D	-4.4	10.7								
3924J03a	H	-1.1	3.6	215/70	254/90	337%	16°	15°			
3924J03b	D	-1.1	3.6	215/70	254/90	337%	16°	15°			
3924J04	D	-1.0	6.1								
3924J05	D	-4.4	1.5								
3924J06	H	-4.4	1.5	215/65	250/90	258%	20°	19°	4°	7.0	0.17
3924J07	D	-5.0	0.0								
3924J08	H	-5.0	0.0	217/65	260/90	525%	10°	10°	1°	7.5	0.14
3924J09a	H	-3.7	-0.5	218/65	256/90	386%	14°	14°	11°	3.4	0.06
3924J09b	D	-3.7	-0.5	218/65	256/90	386%	14°	14°	0°	6.3	0.13
3924J10	H	-1.5	-2.4	214/56	242/90	168%	28°	25°	33°	7.8	0.16
3924J11	D	-1.5	-2.0								
3924J12	D	-0.8	-2.9	215/55	242/90	165%	28°	25°			
3924J13	H	0.8	-6.2	215/43	231/90	100%	39°	32°			
3924J14	D	-0.6	-6.9								
3924J15	H	-1.9	-9.4	213/24	210/90	36%	60°	40°			
3924J16	H	-3.3	-12.5	214/18	203/90	21%	67°	42°			
3924J17	D	-5.0	-13.2								
3924J18	H	-4.4	-14.2	228/18	204/90	23%	66°	42°			
3924J19	H	-0.8	-22.9	230/10	193/90	2%	77°	45°			

^a Sample lithology - harzburgite (H) or dunite (D).

^b Data have been rotated and projected onto the plane 305/50.

^c Angle from shear plane to pyroxene foliation (α), finite strain ellipsoid (θ), and olivine [100] maxima.

Chapter 4

An Assessment of Mantle Heterogeneity and Depletion Based on Abyssal Peridotite Isotope and Trace Element Compositions

Abstract

Abstract

The size and scale of heterogeneity and the full extent of depletion of the asthenospheric mantle cannot be fully constrained by basalt chemistry. We present detailed isotope and trace element analyses of pyroxene mineral separates, from veined and unveined abyssal peridotites, that demonstrate greater variability of the mantle on smaller length-scales than has previously been observed. The isotopic range within one dredge from the Southwest Indian Ridge (SWIR) covers 46% of the entire Indian Ocean basalt range for $^{143}\text{Nd}/^{144}\text{Nd}$. Radiogenic ingrowth during upwelling and exposure on the seafloor are shown to be negligible, producing $<0.01\%$ change in Nd isotopic composition. Cpx and Opx pairs from the same sample are generally in Sr but not Nd isotopic equilibrium. This may reflect Opx rare earth element disequilibrium due to slow diffusion following recent melt-rock reaction. Pyroxenite veins in abyssal peridotites are found to have overlapping compositions with their host peridotites. Combined with other compositional characteristics, we suggest that they formed recently by melt crystallization, rather than as direct remnants of recycled oceanic crust.

The average isotopic composition of peridotites is found to be similar to that of basalts, though peridotites extend to more depleted compositions. These depleted compositions should be accompanied by depletions in trace element concentrations. As basalts are mixtures of melts, they will be dominated by the least depleted peridotite source and the signature of the most depleted peridotite will not be preserved in the melt.

Abyssal peridotite Cpx Nd isotopic compositions and Sm/Nd ratios extend to higher values than can be reconciled with two stage or continuous depletion models of mantle isotopic evolution. We suggest that high Sm/Nd ratios were produced during previous episodes of melt extraction that are not included in current models of DMM formation. We conclude that the scale of mantle heterogeneity and extent of melt depletion observed among abyssal peridotites requires a reassessment of current models for mantle evolution and MORB production, which assume that peridotites in the asthenospheric mantle are uniform in composition.

4.1 Introduction

The ratios of radiogenic isotopes in basalts from mid-ocean ridges (MORB) and ocean islands (OIB) are highly variable, demonstrating that chemical heterogeneity in the mantle has persisted for long periods of time and indicating that convective mixing of the mantle is relatively inefficient at removing heterogeneities (Zindler and Hart, 1986; van Keken et al., 2002). Ranges in isotopic compositions for individual systems (e.g., a given OIB suite or a specific ridge segment) have led to much debate over the form and size of the distinct mantle reservoirs involved. A still significant gap exists between observational geochemistry and numerical models of mantle dynamics (e.g., van Keken et al., 2002), partly because geochemical observations of the length-scales and amplitudes of chemical and isotopic heterogeneities in the mantle are inadequate. Basalts, as mixtures of melt fractions derived from large areas of the mantle that have undergone variable degrees of melting, effectively “smooth-out” heterogeneities. Nevertheless, basalts from individual dredges at some mid-ocean ridge localities reveal significant variations in isotopic compositions and suggest that high-amplitude heterogeneities exist over length-scales <2 km (Dosso et al., 1999). Large variations in Pb isotopic compositions among olivine-hosted melt inclusions from individual MORB and OIB samples (Saal et al., 1998; Shimizu and Layne, 2003; Kobayashi et al., 2004; Shimizu et al., 2005; MacLennan, 2006) provide additional clear evidence for the existence of higher amplitude heterogeneities on smaller length-scales than hitherto

considered.

The depleted MORB source mantle (DMM) is one of the most important geochemical reservoirs and its compositional heterogeneities are of fundamental importance to geochemistry and mantle dynamics. In this chapter, we use abyssal peridotites — depleted solid residues of ocean ridge melting — to assess the amplitudes and length-scales of chemical and isotopic heterogeneities in DMM. Due to severe hydrothermal alteration and the geochemically depleted character of abyssal peridotites, radiogenic isotope data has previously been collected in only four studies (Snow et al., 1994; Kempton and Stephens, 1997; Salters and Dick, 2002; Cipriani et al., 2004). These studies attempted to reconcile the isotopic compositions of peridotites to those of basalts from the same region, over >100 km length-scales.

The present study is different from previous abyssal peridotite studies in two important ways. First, we specifically focus on lithologically-mixed samples from the ultra-slow spreading Southwest Indian Ridge (SWIR), including some of the very few peridotites recovered from a ridge axis. As shown previously (Sleep, 1975; Reid and Jackson, 1981; Bown and White, 1994; Dick et al., 2003), at slow upwelling rates, conductive cooling limits mantle melting. Thus, peridotites exposed on the seafloor are expected to have experienced limited degrees of melting and represent the closest approach to asthenospheric mantle (DMM) composition. Second, we describe geochemical variations in peridotite samples from individual dredges. Sampling length-scales of individual dredges can be estimated from dredge wire tension records and the dredges in this study represent length-scales of <1 km. At this length-scale, in an upwelling and partially melting system, mantle peridotites experience the same pressure and temperature conditions. Any chemical and isotopic variations observed among peridotites from a single dredge must be due to either initial (pre-melting) heterogeneities or recent melt-rock reaction.

This chapter uses detailed trace element and isotope analyses of peridotites from indi-

vidual dredges to constrain the spatial scale of heterogeneities in the mantle, the extent of depletion of the mantle and the role of melt-rock reaction during ridge processes. We report results from two areas of the Southwest Indian Ridge: the Oblique 9°-16°E Segment and the Atlantis II Fracture Zone (Fig. 4.1). Basalts from the Oblique Segment have already been studied in detail by Standish (2006), providing a MORB framework for comparison to the peridotite systematics. The local geodynamic setting of both localities has important implications as to the degree of mantle melting. Along the Oblique Segment, the obliquity of the angle of spreading to the direction of plate motion (Fig. 4.1) results in reduced degrees of melting. The effective full spreading rate varies along the segment from 7.5 to 14 mm/yr, at the lowest end of the spectrum of global ridge spreading rates. Peridotites from the Atlantis II Fracture Zone are also expected to have undergone low degrees of melting, due to an ultra-slow spreading rate combined with a transform fault.

Both the Oblique Segment and the Atlantis II Fracture Zone have been extensively sampled. Fig. 4.1 shows the locations and lithologies of dredges from each area and Table 4.1 lists the locations of the four specific dredges used in this study. We focus on two on-axis Oblique Segment dredges from the 2003 *RV Melville* cruise Vancouver Expedition Leg 7 to 9°-25°E on the SWIR. This area was also sampled during the 1976 *ARA Islas Orcadas* cruise IO11/76, the 1981 *SA Agulhas* cruise AG22, the 1986 *RV Polar Stern* cruise PS86, and the 2001 *RV Knorr* cruise Kn162. Basalts from these cruises have been analyzed by le Roex et al. (1983, 1992), Mahoney et al. (1992), Janney et al. (2005), Standish (2006), and Standish et al. (submitted). The Atlantis II Fracture Zone has been sampled during the 1976 *RV Atlantis II* cruise AII93-5, the 1986 *RV Conrad* cruise RC27-9, and the 1998 *RRS James Clark Ross* cruise JR31. In addition, Atlantis Bank, an uplifted gabbro massif located along the transform fault and formed by detachment faulting between 9.5 and 13 Ma (Dick et al., 1991), has been the focus of Ocean Drilling Program (ODP) Leg 118 in 1987 and Leg 176 in 2000 and dives using the manned submersible Shinkai 6500 in

1998 and 2001 and the remotely operated vehicle Kaiko in 2000. Atlantis II Fracture Zone basalts have been analyzed by Mahoney et al. (1989), Dick et al. (1991), Kempton et al. (1991), Snow (1993) and Coogan et al. (2004). In addition, 1.5 km of gabbro from Hole 735B — drilled on Atlantis Bank during ODP Legs 118 and 176 — have been analyzed for isotopic compositions by Kempton et al. (1991), Hart et al. (1999) and Holm (2002). Finally, the regional scale isotopic variations for peridotites from the Oblique Segment and the Atlantis II Fracture Zone, presented by Salters and Dick (2002), provide a framework for our local-scale datasets.

4.2 Methods

Analysis of modal compositions, trace element concentrations, and Nd and Sr isotopes were carried out on sixteen samples: eleven peridotites, three pyroxenite veined peridotites, and two pyroxenites. Cpx mineral separates were analyzed for isotopes in all samples and Opx for isotopes in six samples. In the veined samples, Cpx from the vein and matrix were measured separately. Cpx from three peridotites were successfully analyzed for Pb isotopes. The only basalt recovered in any of the same dredges as the peridotites — a glass fragment in a breccia from dredge Van7-96 — was also analyzed. Trace element and isotope analyses were carried out at the Pheasant Memorial Laboratory at the Institute for Study of the Earth's Interior in Japan.

4.2.1 Sample selection and characteristics

Peridotites were analyzed from five dredges, as listed in Table 4.1, with a focus on two dredges: Van7-85 (herein after referred to as Dredge 85), containing typically depleted peridotites, and Van7-96 (referred to as Dredge 96), containing pyroxenite-veined peridotites. Also analyzed were one peridotite from Dredge PS86-6, for interlab-comparison

to Salters and Dick (2002), and one peridotite from Dredge Van7-86, a more altered peridotite dredge adjacent to Dredge 85, for determination of the effects of alteration on Sr isotopic composition. From the Atlantis II Fracture Zone, we analyzed sample RC27-9-6-2 (abbreviated to 6-2 in some figures), which contains a clinopyroxenite vein.

Dredge 85 is from an amagmatic region of the Oblique Supersegment (Fig. 4.1) and contains 65 kg peridotite, 11 kg dunite, and 2 kg diabase. The peridotites are “typical” in appearance (Fig. 4.2A), lacking veins and with modal Cpx contents ranging from 2% to 6%. As shown in Fig. 4.5, they have trace element concentrations characteristic of depleted abyssal peridotites (e.g., Johnson et al., 1990). Two harzburgites (Van7-85-27 and Van7-85-30) and three lherzolites (Van7-85-42, Van7-85-47, and Van7-85-49), representative of the modal Cpx range of the dredge, were chosen for isotopic analyses.

Dredge 96 consists predominantly of lherzolites and harzburgites with pyroxenite veins (Fig. 4.2B-F). The dredge is from the inside-corner high of the ridge with the Shaka Fracture Zone (Fig. 4.1) and consists of 36 kg peridotite, 44 kg pyroxenite-veined peridotites, 3 kg pyroxenites, 39 kg polymict breccias and 8 kg diabase. Pyroxenite veins range in width from 1 to 10 cm and either cross-cut larger peridotites samples (Fig. 4.2C) or occur as individual samples with thin peridotite skins (Fig. 4.2E). The veins themselves are cross-cut by narrower veins (Fig. 4.2E) predominantly composed of altered plagioclase and olivine. The absence of plagioclase in the coarser-grained regions of the pyroxenites and the presence of Opx, which does not appear with plagioclase on the low pressure liquidus of basalt (Stolper, 1980), indicate that plagioclase is related to a second, shallow level melt intrusion event. In addition, rims of Cpx grains have Eu anomalies whereas Cpx cores do not (Warren, unpublished ion probe data), further evidence for the later crystallization of plagioclase. Samples selected for analysis from this dredge include pyroxenites with little associated peridotite (Van7-96-09 and Van7-96-16), peridotites with pyroxenite veins (Van7-96-19 and Van 7-96-21), vein-free peridotites (Van7-96-25, Van7-96-28, Van7-96-

35 and Van7-96-38) and basalt from a breccia (Van7-96-68).

4.2.2 Modal analyses

All samples were point counted for mineral modes under an optical microscope using large, 51x75 mm thin sections and a grid spacing of 1 mm (Table 4.2). For the peridotites, a minimum of 1700 points were counted. The pyroxenite modes, due to limited material availability, have fewer points and hence larger associated errors. Images of the thin sections, produced by point counting, are shown in Fig. 4.3.

4.2.3 Chemical analyses

Mineral separates of Opx and Cpx were obtained by lightly crushing thin slices of the peridotite samples, handpicking under a binocular microscope and then re-picking to ensure purity of the concentrate. All grains were visually free of inclusions and uniform in color, except for a less-pure separate of Van7-96-16, included for comparison to a more pure separate of the same sample. Unleached concentrate weights were in the range 25-900 mg. Concentrates were leached ultrasonically in HCl overnight and rinsed in water. Further light crushing in a SiN mortar and pestle exposed cleavage alteration surfaces. Concentrates were then leached ultrasonically for 5 minutes in HF, rinsed with water, leached ultrasonically twice more with HCL for 8 hours each, rinsed with water and finally leached ultrasonically with HNO₃ for 5 minutes. Following a final rinsing with water, samples were dried on a hot plate in a draft chamber with clean air. After leaching, 15-700 mg material remained per mineral separate. Prior to dissolution for high field strength elements (HFSE) and isotope analyses, concentrates were crushed to powder in a SiN mortar and pestle.

ICP-MS analyses of trace elements

Samples were analyzed by ICP-MS for 28 trace element concentrations, of which the HFSE were prepared and measured independently of the other trace elements. In addition, blanks and the external rock standard JB-2 (tholeiitic basalt, Geological Survey of Japan) were prepared with the samples. For analyses of Li, Be, Rb, Sr, Y, Cs, Ba, rare earth elements (REEs), Pb, Th, and U, from 2 to 10 mg of weighed sample was spiked with ^{149}Sm and dissolved in two stages following the method of Makishima and Nakamura (1997). In the first stage, samples were decomposed in HClO_4 and HF using a combination of ultrasonic bath and hotplate heating for several days, followed by stepwise drying. In the second stage, the samples were heated overnight after addition of HCl, followed by stepwise drying. Finally, the samples were dissolved in HNO_3 to a dilution factor of 1000 prior to analysis.

Analyses of HFSE – Zr, Nb, Hf, and Ta – used weighed sample amounts of 3-12 mg and a mixed ^{91}Zr - ^{179}Hf spike. Full recovery of HFSE following dissolution requires dissolution in HF, however – in Ca-rich materials – HF leads to the precipitation of fluoride compounds containing HFSE. To suppress the coprecipitation of HFSE, a solution of Al metal dissolved in HF and HCl was added prior to dissolution. The amount of Al solution added per sample was calculated to bring the $\text{Al}/(\text{Al}+\text{Ca})$ ratio of the pyroxene to 0.9, which has been shown to effectively limit HFSE coprecipitation in fluoride compounds (Tanaka et al., 2003). Following the teflon bomb method of Tanaka et al. (2003), samples were dissolved in HF at 245°C for 48 hours. After dissolution, mannitol was added to sample solutions, which were then placed overnight in ultrasonic baths before being dried at 80°C in a clean air system. Prior to analysis, samples were dissolved in HF to a dilution factor of 800, centrifuged and transferred to new containers to remove any remaining fluorides from the solutions.

Trace elements were analyzed using an Agilent 7500cs Q-pole type ICP-MS, following

the procedures of Makishima and Nakamura (1997, 2006) and Makishima et al. (1999). Samples were introduced via fluid injection using a micro-concentric nebulizer. Every two samples were bracketed by analysis of the JB-2 unspiked standard. Data reduction followed the isotope dilution internal standardization method of Makishima and Nakamura (2006), using ^{149}Sm as the internal standard. For HFSE, the internal standard ^{91}Zr was used for Zr and Nb and the internal standard ^{179}Hf was used for Hf and Ta. The 2σ reproducibility of the rock standard JB-2 is 1.4-6.6%, blanks were < 0.01 ppb and detection limits in the range 0.1-10 ppb. The elements Rb, Cs, and Ta were below detection limits for the samples in this study and are not reported here.

TIMS analyses for Sm-Nd and Rb-Sr

Analyses of Sm-Nd and Rb-Sr isotope ratios and concentrations by TIMS followed the techniques of Nakamura et al. (2003). Sample amounts ranged between 9 and 240 mg (Table 4.4), with the amount used determined by the Nd and Sr concentrations of the sample. Sample powders were spiked with ^{84}Sr , ^{87}Rb , ^{150}Nd and ^{149}Sm prior to dissolution in HClO_4 , HF, HNO_3 , and HCl using a combination of ultrasonic bath and heating. The dissolved minerals were stepwise dried using closed-system evaporators to limit contamination; this evaporation system was used in all subsequent drying steps. A combination of HClO_4 addition and additional stepwise drying were used to completely decompose insoluble fluorides. The first separation step for Rb, Sr and REE used 1 ml columns with Bio-Rad AG-50WX10 cation exchange resin and HCl (Yoshikawa and Nakamura, 1993; Nakamura et al., 2003). Rb and Sr were separated using a second cation exchange column with 0.5 ml of Muromac AG-50WX8 cation exchange resin and HCl. Nd and Sm were separated in a second column with 0.3 ml of Muromac AG-50WX8 resin, but using α -hydroxyisobutyric acid for elution.

Sr and Nd isotopic compositions and Rb, Sr, Nd and Sm concentrations were deter-

mined on a Finnigan-MAT262 solid-source thermal ionization mass spectrometer equipped with 5 Faraday cups. In addition, some Sr analyses were run on a Finnigan-Triton mass spectrometer with 9 Faraday cups. Analytical procedures followed Nakamura et al. (2003). Sr was dissolved in HNO₃ and loaded with a Ta₂O₅ activator on trapezoid shaped single W ionization filaments. Repeat analyses of the NBS 987 standard on the MAT262 gave a value of 0.710183 ± 0.000023 ($2\sigma_m$, n=10) and on the Triton a value of 0.710271 ± 0.000012 ($2\sigma_m$, n=8). Data are fractionation corrected to $^{86}\text{Sr}/^{88}\text{Sr} = 0.1194$ and reported relative to $^{87}\text{Sr}/^{86}\text{Sr} = 0.710250$ for the NBS 987 standard (Faure and Mensing, 2005). Rb concentrations were measured using a Ta-Re double filament with Rb dissolved in HNO₃ prior to loading on the Ta filament. The 2σ analytical reproducibility for 1 pg of Rb by this method is better than 1%, based on 10 replicate analyses (Nakamura et al., 2003).

Nd isotope analyses use Re double filaments, with Nd dissolved in HNO₃ prior to loading onto the Re ionization filament. Data were collected relative to the internal PML Nd standard, which has a value of $^{143}\text{Nd}/^{144}\text{Nd} = 0.511769 \pm 0.000011$ ($2\sigma_m$, n=5) with respect to 0.511892 ± 0.000007 ($2\sigma_m$, n=8) for the La Jolla Nd standard. Data are fractionation corrected to $^{146}\text{Nd}/^{144}\text{Nd} = 0.7219$ and reported relative to $^{143}\text{Nd}/^{144}\text{Nd} = 0.511858$ for the La Jolla standard (Lugmair and Carlson, 1978). Determination of Sm concentrations followed the same Ta-Re double filament method used for Rb, with a 2σ analytical reproducibility of 1% for 1 pg Sm, based on 10 replicate analyses (Nakamura et al., 2003).

TIMS analyses for Pb

For Pb analyses, 50-600 mg of weighed mineral separates were dissolved in equal amounts of HBr and HF, with HBr used to suppress fluoride coprecipitation of Pb. Samples were heated and placed in ultrasonic baths for 3-7 days, and then centrifuged before transferring to new containers. The material remaining in the old containers was further treated with HBr and heating or ultrasonic bath for an additional day to promote fluoride disso-

lution and Pb removal from fluorides. In none of the samples were fluorides completely dissolved. Samples were dried in closed-system evaporators and diluted with HBr prior to column loading. Following the method of Kuritani and Nakamura (2002), a column with 0.1 ml of Bio-Rad AG-1X8 anion exchange resin was used to remove most elements apart from Pb, which was eluted in H₂O. A second column, with 0.01 ml Bio-Rad AG-1X8 resin, was used to remove trace contaminants from the Pb fraction, which was again eluted using H₂O.

Following column separation, Pb concentrations were estimated from Pb count rates on the Agilent 7500cs ICP-MS. Of the 22 Cpx and Opx samples for which Pb was separated, only 3 Cpx samples yielded >1 ng Pb, sufficient for measurement of isotope ratios. The low Pb yields are probably due to the large amount of fluoride coprecipitation during dissolution, which was not adequately suppressed by HBr addition. Three basalt samples – one from dredge Van7-96 and two repeat analyses (Van7-92-03 and Kn162-61-71) from Standish (2006) – were also prepared and these all yielded >5 ng Pb.

Analytical procedures for the two Cpx samples followed the two double spikes method of Kuritani and Nakamura (2003). A double spike enriched in ²⁰⁷Pb and ²⁰⁴Pb was added to a quarter of the sample and a double spike enriched in ²⁰⁵Pb and ²⁰⁴Pb was added to the remainder of the sample. For the basalts, analytical procedures followed the normal double spike method (e.g., Hamelin et al., 1985; Galer, 1999), by adding a spike enriched in ²⁰⁷Pb and ²⁰⁴Pb to a quarter of the sample. The dried samples were dissolved in silicic acid and loaded on single Re filaments. All samples were analyzed on a Finnigan-MAT261 mass spectrometer equipped with 6 Faraday cups. Replicate analyses of the NBS 981 standard gave values of ²⁰⁶Pb/²⁰⁴Pb = 16.942 ± 0.002 (2σ_m), ²⁰⁷Pb/²⁰⁴Pb = 15.500 ± 0.002 (2σ_m) and ²⁰⁸Pb/²⁰⁴Pb = 36.727 ± 0.004 (2σ_m). Data are reported relative to the values of NBS 981 from Todt et al. (1996), for consistency with the Standish (2006) Oblique Segment basalt dataset.

4.3 Results

4.3.1 Data quality

Analyses of abyssal peridotites are difficult due to the low concentrations of relevant trace elements and the effects of seawater alteration. In Fig. 4.4, we compare concentrations of Sr, Nd and Sm measured by ICP-MS and by TIMS. In general, the data plot either on the 1:1 line or towards slightly higher values for the TIMS results, but with no systematic offset as a function of concentration. The average difference between ICP-MS and TIMS concentrations is 14%. An average of 3 mg Cpx and 7 mg Opx were used for the ICP-MS analyses and blank amounts were < 10 pg. For TIMS analyses, average sample weights were 70 mg for Cpx and 150 mg Opx, which should produce more precise results. However, the use of larger sample volumes required larger acid volumes for dissolution, leading to increased blanks. Blank amounts for TIMS analyses were < 1 ng and typically in the range 1-50 pg. The competing effects of increased precision and increased blank amounts with sample volume are the likely cause of the non-systematic variation between ICP-MS and TIMS concentrations.

To assess the importance of mineral separate purity, we analyzed two Cpx mineral separates from pyroxenite sample Van7-96-16. While one separate consisted of optically pure grains, the other consisted of Cpx which were not optically pure and contained inclusions of both black and white grains. The REE compositions of both mineral separates are plotted in Fig. 4.5B, with the impure separate shown as a dashed line. The pure Cpx separate is indistinguishable from the other pyroxenites. The impure separate has low REE concentrations with a positive Eu anomaly, both indicative of analysis of Cpx and plagioclase combined. The Nd isotope ratios for these two separates are within error, while the pure separate has a higher Sr isotope ratio than the impure separate. Hence, the purity of the separate appears to affect trace element concentrations more than Nd isotopic ratio.

To test the precision of our analyses and for interlab comparison, we repeated the analysis of PS86-6-36 from Salters and Dick (2002). As indicated in Table 4.4, our values for $^{143}\text{Nd}/^{144}\text{Nd}$ and $^{87}\text{Sr}/^{86}\text{Sr}$ agree within error with the Salters and Dick values. For Pb isotopes, we ran replicate analyses of two basalt samples from Standish (2006). $^{206}\text{Pb}/^{204}\text{Pb}$ agree within error, but our samples have $^{207}\text{Pb}/^{204}\text{Pb}$ and $^{208}\text{Pb}/^{204}\text{Pb}$ offset to higher values (Fig. 4.7), despite normalization of all data to the same standard values for NBS 981. To check internal reproducibility, we ran a duplicate analysis of Van7-85-47, following the entire procedure starting from a split of the same powdered mineral separate. Table 4.4 shows that the duplicate analyses agree within error for $^{143}\text{Nd}/^{144}\text{Nd}$, but not for $^{87}\text{Sr}/^{86}\text{Sr}$ (0.702286 vs 0.702176). The discrepancy between Sr isotope ratios may be due to heterogeneous Sr distribution due to alteration phases not completely removed by leaching. We interpret the lower $^{87}\text{Sr}/^{86}\text{Sr}$ as the more representative value as seawater alteration increases Sr isotopic ratios.

Abyssal peridotites are generally severely altered by hydrothermal processes at the ridge and our samples are no exception. The majority of olivine is either serpentinized or altered to brownish orange clay minerals. Opx and Cpx are typically much fresher than olivine, but alteration occurs occasionally along rims and cleavage planes or as amphibole pseudomorphs. The alteration process results in isotopic exchange between minerals and fluid, and the precipitation of fluid-derived carbonates. Nd isotopic exchange is negligible due to low Nd concentrations in seawater, but Sr isotopic exchange and carbonate precipitation shift the $^{87}\text{Sr}/^{86}\text{Sr}$ of peridotite towards 0.709, the isotopic composition of seawater. In order to eliminate the seawater Sr signature, we followed a multi-day, multi-stage leaching procedure using HCl, HF, and HNO_3 , with an intermediate step of additional crushing in a SiN mortar, to remove alteration phases from the rims and cleavage planes of pyroxene grains. This procedure was effective in significantly reducing the seawater alteration signature based on the following three observations. (1) With a few exceptions, data plot on the

same Nd-Sr isotope trend defined by basalts from the Oblique Segment (Standish, 2006), as illustrated in Fig. 4.6. (2) Coexisting Cpx and Opx have similar $^{87}\text{Sr}/^{86}\text{Sr}$ ratios. As Opx is more susceptible to hydrothermal alteration due to its lower Sr concentrations, this result strongly supports our contention that alteration affects were minimized by our analytical technique. (3) Our results include a $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of 0.702096 in Cpx from Van7-96-38, the most depleted value measured to date in abyssal peridotites and among the lowest measured in any basalt, thus attesting to the effectiveness of our leaching procedure.

Despite extensive leaching, the two most visibly altered samples have $^{87}\text{Sr}/^{86}\text{Sr} > 0.704$ (Van7-85-30 and Van7-86-27) and do not plot on the basalt Nd-Sr isotope trend (Fig. 4.6). We conclude that pristine $^{87}\text{Sr}/^{86}\text{Sr}$ values were not recovered for these samples. The only sample in which a high ratio does not appear to be due to visible alteration is RC27-9-6-2, which is fresh for an abyssal peridotite. This sample is unusual in various ways, including the greater isotopic and trace element enrichment in the peridotite matrix than in the clinopyroxenite vein. However, until $^{87}\text{Sr}/^{86}\text{Sr}$ can be measured *in situ*, the origin of $^{87}\text{Sr}/^{86}\text{Sr}$ enrichment in this sample cannot be conclusively identified. We have not, however, found any evidence for the “orphan” $^{87}\text{Sr}/^{86}\text{Sr}$ component identified in this sample by Snow et al. (1993).

4.3.2 Modes and petrography

The peridotites in this study are typical of abyssal peridotites (Dick, 1989), ranging in modal Cpx content from 2% to 13%. Mineral major element compositions (Appendix 4.A) are also typical of abyssal peridotites. The variations of spinel Cr# ($\text{Cr}/(\text{Cr}+\text{Al})$), Mg# ($\text{Mg}/(\text{Mg}+\text{Fe})$) and TiO_2 suggest variable degrees of melting and, among veined peridotites, melt refertilization (Fig. 4.A1). The most unusual feature among Cpx major element compositions is the relatively low Cr_2O_3 among Dredge 96 pyroxenites whereas the clinopyroxenite vein in RC27-9-6-2 has relatively high Cr_2O_3 (Fig. 4.A2).

Dredge 96 pyroxenites consist of 25-39% Opx and 24-57% Cpx, with minor amounts of olivine, hence they are properly classified as websterites and Ol-websterites. The boundaries of narrower veins are difficult to distinguish in thin section, as veins are sometimes the width of a single pyroxene porphyroclast (Fig. 4.2D). Veins of finer grained Ol-Plag-Cpx cross-cut the coarser grained pyroxenites or forms rims around spinel trails. As discussed earlier, we interpret the Ol-Plag-Cpx assemblage to be a late-stage gabbroic melt which crystallized at shallow depths.

The clinopyroxenite vein in RC27-9-6-2 is composed of 82% large, bright green, Cr-rich Cpx. Plagioclase is present in trace amounts (0.01-0.2%) in the vein and matrix. It may either be the result of spinel breakdown or related to a late stage 0.5 mm wide magmatic vein that occurs in the peridotite matrix, unconnected to the clinopyroxenite vein. The clinopyroxenite vein boundaries are sharply defined by a transition from large Cpx grains, with associated small olivine, Cpx and Opx grains, to the coarse grained olivine matrix (Fig. 4.2G-H).

4.3.3 Trace elements

Peridotite Cpx in this study cover the range of trace element compositions previously observed in abyssal peridotite Cpx (e.g., Johnson et al., 1990). Trace element data for Cpx and Opx are presented in Table 4.3 and REE patterns, normalized to primitive upper mantle (PUM; McDonough and Sun, 1995), are plotted in Fig. 4.5. For comparison, the abyssal peridotite Cpx datasets from Johnson et al. (1990) and Johnson and Dick (1992) are also shown, using only samples away from hotspots. Samples from Dredge 85 and the Dredge 86 sample have typical depleted REE patterns for abyssal peridotites, with Cpx falling in the middle of the range of Cpx compositions from Johnson et al. (1990). In contrast, the Dredge 96 samples and the Atlantis II Fracture Zone sample have higher overall REE concentrations and less depleted patterns.

Both the peridotites and pyroxenites from Dredge 96 have enriched compositions, with almost flat patterns, except for variable depletion in the light REE (LREE), which range over an order of magnitude. The most notable feature of the Dredge 96 samples is that the pyroxenites do not have distinct trace element compositions from the peridotites. The basalt from Dredge 96 has a LREE enriched composition, with only slightly higher heavy REE (HREE) concentrations than the peridotites. The sample from dredge PS86-6 is also relatively undepleted, with a similar composition to the most LREE depleted of the Dredge 96 samples.

The Atlantis II Fracture Zone sample RC27-9-6-2 has a distinctive REE pattern that is nearly flat and relatively depleted in HREE and enriched in LREE, with respect to the field for residual abyssal peridotite Cpx. In addition, the REE pattern of the matrix peridotite Cpx crosses that of the vein Cpx. The matrix Cpx is almost identical to PUM in composition, while vein Cpx is less enriched in LREE but more enriched in HREE. In comparison to other peridotite Cpx from the fracture zone (Johnson and Dick, 1992), LREE in RC27-9-6-2 are more enriched but HREE are less enriched.

4.3.4 Isotopes

Results of Nd, Sr and Pb isotope analyses are presented in Tables 4.4 and 4.5, revealing variations both within and between dredges and down to the sub-sample scale. In Fig. 4.6, we plot the variation in $^{143}\text{Nd}/^{144}\text{Nd}$ against the variation in $^{87}\text{Sr}/^{86}\text{Sr}$ for both localities. Pb isotope data are shown for the three Cpx separates successfully analyzed in Fig. 4.7. In addition, we plot published peridotite Cpx compositions for the same localities (Snow et al., 1994; Salters and Dick, 2002) and related basalts and gabbros from a compilation of published data (see Fig. 4.6 caption for references).

The Oblique Segment

On the Oblique Segment, Dredge 85 samples are isotopically depleted, except for alteration-related enrichment of Sr in some samples. One Dredge 85 sample is more depleted in both Nd and Sr than the estimated isotopic composition of DMM, which is defined by the average composition of MORBs (Su and Langmuir, 2003). This sample is also more depleted than depleted-DMM (D-DMM), defined as the isotopic composition that is 2σ depleted from average MORB (Su and Langmuir, 2003; Workman and Hart, 2005). Similarly, in Dredge 96, Van7-96-38 is more depleted than DMM for both Nd and Sr, and has among the lowest $^{87}\text{Sr}/^{86}\text{Sr}$ value ever measured in MORBs or abyssal peridotites.

The isotopic composition of Dredge 96 ranges from the extremely depleted composition of Van7-96-38 (unveined peridotite) to the enriched compositions of Van7-96-16 (pyroxenite) and Van7-96-35 (unveined peridotite). The pyroxenites and peridotites do not form two distinct isotopic groups, though the unveined peridotites extend to more depleted compositions than the pyroxenites. The basalt from Dredge 96 plots towards the depleted end of the peridotite trend, close to DMM.

In Fig. 4.6A, we also plot the composition of Bouvet Island basalts, located to the west of the Oblique Segment near the Bouvet Triple Junction (Fig. 4.1). Bouvet Island is the present-day location of Bouvet hotspot, which passed along the African plate section of the Shaka Fracture Zone at $\sim 15\text{-}25$ Ma (Hartnady and le Roex, 1985). Basalts and peridotites from the fracture zone (le Roex et al., 1983; Snow, 1993) have similar Nd isotopic compositions to Bouvet basalts (Fig. 4.6A). The most enriched Dredge 96 samples fall within the range of Bouvet basalt compositions. As a whole, Dredge 96 peridotites, Oblique basalts, and other Oblique peridotites form a trend between Bouvet Hotspot and DMM, with some peridotite $^{87}\text{Sr}/^{86}\text{Sr}$ exceptions due to seawater alteration. In contrast, peridotites from the ridge near Bouvet Island (Snow et al., 1994) have depleted compositions similar to DMM.

The Pb isotopic variation of the two Dredge 96 peridotites, one veined and the other

unveined, show significant variation in $^{206}\text{Pb}/^{204}\text{Pb}$ and $^{208}\text{Pb}/^{204}\text{Pb}$, but less variation in $^{207}\text{Pb}/^{204}\text{Pb}$ (Fig. 4.7). Van7-96-28 (unveined peridotite) is similar in composition to Bouvet basalts. Van7-96-21 (veined peridotite) and the Dredge 96 basalt have less radiogenic compositions, with the basalt close in composition to DMM. All three samples are offset in $^{207}\text{Pb}/^{204}\text{Pb}$ from the trend of the Oblique basalts from Standish (2006), whereas our replicate analyses of two basalts from the Standish (2006) dataset plot on the trends (Table 4.5, Fig. 4.7).

Atlantis II Fracture Zone

The peridotite matrix of RC27-9-6-2 is enriched in $^{143}\text{Nd}/^{144}\text{Nd}$, while the clinopyroxene vein is, surprisingly, less isotopically enriched. Sr isotopes are enriched, also with greater enrichment in the peridotite than in the vein. While seawater alteration cannot be ruled out as the cause of $^{87}\text{Sr}/^{86}\text{Sr}$ enrichment, an isotopic range from 0.704 to 0.705 on a scale of ~ 10 cm is unlikely for seawater alteration. The isotopic composition of RC27-9-6-2 contrasts with previously analyzed Atlantis II Fracture Zone peridotites (Snow et al., 1994; Salters and Dick, 2002), which extend from $^{143}\text{Nd}/^{144}\text{Nd}$ compositions similar to DMM towards values more depleted than D-DMM.

In Fig. 4.6B we plot three types of MORB data from the Atlantis II Fracture Zone region for comparison to the peridotites: samples from (i) the Atlantis II Fracture Zone, (ii) a transform volcano at the northern ridge-transform intersection, and (iii) the ridge segments on either side of the fracture zone. While generally depleted in composition, all three types of basalt have at least one sample with $^{143}\text{Nd}/^{144}\text{Nd}$ similar to RC27-9-6-2. The gabbros from Hole 735B on Atlantis Bank are all depleted in composition, encompassing the range of depleted MORBs from the area. In terms of Pb isotopes, basalts and gabbros from the fracture zone are depleted, ranging between DMM and D-DMM in composition. Cpx from the peridotite matrix of RC27-9-6-2 is more depleted in $^{206}\text{Pb}/^{204}\text{Pb}$ than the

basalts and gabbros but similar in terms of $^{207}\text{Pb}/^{204}\text{Pb}$ and $^{208}\text{Pb}/^{204}\text{Pb}$.

4.4 Discussion

4.4.1 Partition coefficients

In abyssal peridotites, determination of Cpx-Opx partition coefficients is important for constraining the occurrence of mineral-mineral equilibrium during melt-rock reaction and subsolidus re-equilibration in the upwelling mantle. In Table 4.6, we present the first set of Cpx-Opx partition coefficients for abyssal peridotites from bulk mineral analyses. In addition, listed in Table 4.6 are the dominant valence state of each element in pyroxene and their ionic radii (Shannon, 1976) in Cpx and Opx. The ionic radii provide a measure of the relative compatibility of each element in Cpx and Opx and we use these radii to order the elements in the spider diagrams in Fig. 4.8. In Cpx, trace elements vary between VI-fold and VIII-fold coordination, depending on whether they occupy the M1 or M2 site, respectively, whereas elements in Opx are in VI-fold coordination in both the M1 and M2 sites. Thus elements have the same ionic radii in Cpx and Opx if they occupy the M1 site in Cpx, which is the case for cations such as Zr and Ti that have relatively small cationic radii. Ordering elements in this way is more relevant to inter-mineral behavior than the commonly used element ordering from Hofmann (1988), which is based on the partitioning of elements into melt.

The range of partition coefficients from Witt-Eickschen and O'Neill (2005) and from Hellebrand et al. (2005) are shown on the spider diagrams in Fig. 4.8. The Witt-Eickschen and O'Neill (2005) dataset consists of ICP-MS analyses of mineral separates from peridotite xenoliths thought to have equilibrated over a temperature range of 900-1250°C. Their results constrain the change in partition coefficients with temperature and composition combined. These are the main variables controlling partitioning and they cannot be

separated for pyroxenes with an immiscibility gap. Our results overlap the high temperature end of the Witt-Eickschen and O'Neill dataset and require only a small amount of extrapolation to reach a temperature relevant to mantle melting processes ($\sim 1350^{\circ}\text{C}$). Despite this extrapolation, our partition coefficients have the advantage over the Witt-Eickschen and O'Neill coefficients of being determined for pyroxene compositions relevant to the upper oceanic mantle.

The second background dataset in Fig. 4.8 is from Hellebrand et al. (2005) and consists of *in situ* ion probe analyses of abyssal peridotite pyroxenes. The Hellebrand et al. partition coefficients extend over a range of up to two orders of magnitude, with a larger range among the more incompatible elements. Most pyroxenes in abyssal peridotites exsolve a second pyroxene phase during their ascent to the seafloor, which redistributes the high temperature trace element budget within the mineral. To account for exsolution lamellae, Hellebrand et al. (2005) applied correction factors to their partition coefficients, with larger corrections applied to the more incompatible elements. Application of their correction factors shifts their partition coefficients to higher temperature values, but does not reduce the partition coefficient range for a given element. We suggest that the large range is due to the difficulty in analyzing Opx, which typically exsolves closely spaced, narrow Cpx lamellae. Combined with the lower trace element concentrations of Opx with respect to Cpx, the spread of the Hellebrand et al. dataset towards lower partition coefficients with increasing incompatibility is probably due to overlap of the ion beam onto Cpx lamellae during Opx analysis.

In general, the partition coefficients that we measured vary smoothly as a function of ionic radius. The breaks in slope on the spidergram correspond to large changes in radius, changes in cation valence state, or changes in site occupancy (e.g., between Li and Zr, cations switch from M2 to M1 site occupancy). Only Pb, the most incompatible element, exhibits a large range of partition coefficients, which may be due to measurement

uncertainty or sulfide control of the Pb system.

The smooth variation of Cpx/Opx partition coefficients in Fig. 4.8 requires the distribution of trace elements in Cpx to be homogeneous. Diffusion in Cpx is faster for elements with smaller cationic radii (Van Orman et al., 2001), hence disequilibrium in Cpx would fractionate partition coefficients, which we do not observe. Models of REE diffusion in Cpx at conditions relevant to mid-ocean ridges have shown that Cpx equilibrates on a time-scale of <1 My for grains <5 mm in diameter (Van Orman et al., 2001; Cherniak and Liang, 2007). In addition, no core to rim variations are observed among Cpx based on ion probe analyses (Warren, unpublished data). The only exception is RC27-9-6-2, which has core to rim REE variations (Warren, unpublished data), coarser grain sizes and a shallower slope among LREE partition coefficients (Fig. 4.8).

In contrast to Cpx, Opx is typically coarse grained and has lower diffusion rates that are independent of cationic radius (Cherniak and Liang, 2007). Modeling by Cherniak and Liang has demonstrated that significant core to rim variations can form in Opx because of the slow diffusion rates. Opx disequilibrium changes the absolute value of the partition coefficients without changing the shape of the partition coefficient pattern. Hence, Opx in Dredge 96 and RC27-9-6-2 could represent disequilibrium, leading to overestimates of the partition coefficients with respect to their closure temperatures.

4.4.2 Isotopic disequilibrium between Cpx and Opx

Cpx and Opx from the same peridotite samples do not have the same $^{143}\text{Nd}/^{144}\text{Nd}$ ratios, but they generally have the same $^{87}\text{Sr}/^{86}\text{Sr}$. In addition, Cpx from the matrix and vein of the same sample do not have the same $^{143}\text{Nd}/^{144}\text{Nd}$ ratios. Isotopic disequilibrium can result from radiogenic ingrowth in minerals with the same initial isotopic composition or it can be produced by kinetic processes. We test for radiogenic ingrowth by fitting Sm-Nd isochrons for the six Cpx-Opx pairs of measurements, plus three subsets of matrix-vein

Cpx measurements, as shown in Fig. 4.9. We do not determine isochrons for the Rb-Sr system, as Rb concentrations are extremely low in pyroxenes (Table 4.4) and significant isotopic disequilibrium is not observed, with the exception of RC27-9-6-2.

In Fig. 4.9, the mineral “isochrons” are plotted using a linear least squares regression through Cpx-Opx pairs and through Cpx-Opx-Cpx pairs among veined samples. Both the pyroxenites and peridotites that we analyzed have inconsistent ages, ranging from -98 Ma to 156 Ma. Addition of vein Cpx to the calculation emphasizes the inconsistency of the ages, as Van7-96-21 changes from 46 Ma to 274 Ma and RC27-9-6-2 from 68 Ma to 223 Ma. Isochrons regressed through samples from the same dredge also do not demonstrate a coherent set of ages. Hence, the isotopic disequilibrium of the SWIR peridotites is not the result of radiogenic ingrowth and must be related to kinetic processes in the peridotites.

We suggest that the grain-scale Nd isotopic disequilibrium between Cpx and Opx is due to incomplete diffusive equilibration of Opx during melt-rock reaction at the ridge. The general disequilibrium between veins and matrices and among samples from the same dredge suggest variable amounts of melt-rock reaction during melt extraction. As discussed in the previous section, the absence of REE fractionation among our measured partition coefficients requires Cpx to be equilibrated. In contrast, diffusion in Opx does not fractionate REE, so Opx equilibrium is not required by the partition coefficient data. In addition, experimental and model results from Cherniak and Liang (2007) indicate that REE diffusion in Opx is slow and equilibration requires time scales of > 20 My. Sr diffusion is faster than REE diffusion and Cpx-Opx pairs have $^{87}\text{Sr}/^{86}\text{Sr}$ close to or at isotopic equilibrium. Hence, we conclude that the observed isotopic disequilibrium between Cpx and Opx represents incomplete Opx equilibration following melt-rock reaction at the ridge axis.

4.4.3 Radiogenic ingrowth

Peridotites from slow spreading ridges spend a considerable amount of time between passing through their closure temperatures and exposure on the seafloor. The isotopic range among samples may be smaller than measured because of radiogenic ingrowth during this time. To determine the extent to which ingrowth has affected our measured $^{143}\text{Nd}/^{144}\text{Nd}$ compositions, we assess (i) the amount of ingrowth since a sample was at the 1200°C isotherm and (ii) the time necessary to produce the observed isotopic range among samples from the same location. Ingrowth of $^{87}\text{Sr}/^{86}\text{Sr}$ is not addressed because Rb concentrations in peridotite pyroxenes are extremely low.

In order to age-correct peridotites, we calculate the time since the peridotites left the 1200°C isotherm below the ridge axis to their exposure at variable distances from the ridge axis. We select the 1200°C isotherm as a maximum closure temperature for REE diffusion in pyroxenes (Van Orman et al., 2002), to estimate a maximum upwelling time and thus a maximum value for the amount of ingrowth. The ingrowth time, t_I , is calculated as a combination of the upwelling time, t_U , and the seafloor spreading time, t_S . The value of t_S is determined by dividing the measured distance, d , between the dredge location and the ridge axis by the effective half-spreading rate, v_H .

To determine t_U , we first use the results of a numerical model for the thermal structure of the ridge axis (Montési et al., 2006) to determine the depth, L_{1200}^* , to the 1200°C isotherm:

$$L_{1200}^* = 5 \frac{\kappa}{v_H} \quad (4.1)$$

where κ is the thermal conductivity ($10^{-6} \text{m}^2 \text{s}^{-1}$). The solution for corner flow beneath the ridge axis (Batchelor, 1967; McKenzie, 1969) provides the relationship between upwelling velocity, v_U , and spreading rate:

$$v_U = \frac{2}{\pi} v_H \quad (4.2)$$

leading to an expression for t_U :

$$t_U = \frac{5\pi}{2} \frac{\kappa}{v_H^2} \quad (4.3)$$

Thus, the time available for ingrowth is:

$$t_I = t_U + t_S = \frac{5\pi}{2} \frac{\kappa}{v_H^2} + \frac{d}{v_H} \quad (4.4)$$

For Dredge 85 and 96 samples, t_U is ~ 7 My and t_S is 0.5 My for Dredge 85 and 4 My for Dredge 96. On the Oblique Segment, at locations where the ridge has a high angle of obliquity to the spreading direction, t_U approaches 18 My. For ridges with half spreading rates > 10 mm/yr and no obliquity, $t_U \leq 1$ My.

In Fig. 4.10A, we plot $^{143}\text{Nd}/^{144}\text{Nd}$ corrected for t_I (the “age” corrected value) versus the measured $^{143}\text{Nd}/^{144}\text{Nd}$ value. Samples plot on or to the right of the 1:1 line, with a maximum change of 0.01% among our samples. The field between DMM and D-DMM is shaded to demonstrate that the age correction does not shift depleted peridotites back to DMM values. In addition, in Fig. 4.10A, we plot age corrected values calculated for the global abyssal peridotite Cpx dataset (Snow, 1993; Snow et al., 1994; Salters and Dick, 2002; Cipriani et al., 2004). The maximum change globally in $^{143}\text{Nd}/^{144}\text{Nd}$ composition is 0.04%, even among samples that are ~ 20 Ma. The small amount of ingrowth in the global dataset indicates that ingrowth can be ignored when interpreting peridotite isotopic data.

The amount of ^{143}Nd ingrowth is a function of time and the Sm/Nd ratio, the effects of which are best understood using ϵ_{Nd} notation, as the absolute change in $^{143}\text{Nd}/^{144}\text{Nd}$ is small owing to the long half-life of ^{147}Sm . The difference between the measured, M , and corrected, C , values of ϵ_{Nd} is:

$$\Delta\epsilon_{Nd} = \epsilon_{Nd}^M - \epsilon_{Nd}^C = \frac{\left(\frac{^{143}\text{Nd}}{^{144}\text{Nd}}\right)_M - \left(\frac{^{143}\text{Nd}}{^{144}\text{Nd}}\right)_C}{\left(\frac{^{143}\text{Nd}}{^{144}\text{Nd}}\right)_{CHUR}} \times 10^4 \quad (4.5)$$

where *CHUR* is the chondritic uniform reservoir, which has a $^{143}\text{Nd}/^{144}\text{Nd}$ value of 0.512638. The amount of the age correction for variable Sm/Nd ratios can be calculated independently of $^{143}\text{Nd}/^{144}\text{Nd}$:

$$\Delta\epsilon_{Nd} = \frac{\lambda t_I c \frac{Sm}{Nd}}{\left(\frac{^{143}Nd}{^{144}Nd}\right)_{CHUR}} \times 10^4 \quad (4.6)$$

where *c* is the conversion factor of 0.6049 between Sm/Nd and $^{147}\text{Sm}/^{144}\text{Nd}$.

The results of calculations using Equations 4.5 and 4.6 are shown in Fig. 4.10B. For samples in our study, the ingrowth time is generally ≤ 10 Ma and Sm/Nd ratios are < 2 . Samples from the global dataset that have longer ingrowth times are either far from the ridge axis or have very low effective spreading rates. Samples with the highest amounts of ingrowth ($\Delta\epsilon_{Nd} \approx 3-4$) have both high Sm/Nd ratios and long ingrowth times. Overall, the amount of ingrowth is generally < 2 ϵ units. From Fig. 4.10, the isotopic range among abyssal peridotites is clearly related to long term mantle heterogeneity and not created in the time since melting beneath the ridge axis.

To assess whether the range of isotopic compositions observed within a single dredge can be produced following the recent melting event, by different amounts of ingrowth due to variable Sm/Nd ratios, we calculate the time necessary for ingrowth of the measured range of $^{143}\text{Nd}/^{144}\text{Nd}$. Using the parent and daughter ratios for a pair of samples (*A* and *B*) and assuming the same initial $^{143}\text{Nd}/^{144}\text{Nd}$ ratio, then the time, *T*, for radiogenic ingrowth is:

$$T = \frac{1}{\lambda} \ln \left[\frac{\left(\frac{^{143}Nd}{^{144}Nd}\right)_A - \left(\frac{^{143}Nd}{^{144}Nd}\right)_B}{\left(\frac{^{147}Sm}{^{144}Nd}\right)_A - \left(\frac{^{147}Sm}{^{144}Nd}\right)_B} + 1 \right] \quad (4.7)$$

where λ is the ^{147}Sm to ^{144}Nd decay constant ($6.54 \times 10^{-12} \text{y}^{-1}$).

From Equation 4.7, we calculate that the range of isotopic compositions observed among the SWIR peridotites is too large to result from recent radiogenic ingrowth in the peridotites following melt extraction under the ridge axis. Comparing each pair of samples analyzed from Dredge 85, we calculate an average ingrowth time of 230 My. For Dredge

96, the average ingrowth time among all samples is 520 My, among peridotites it is 510 My, and among pyroxenites it is 240 My. All of these times are much longer than the time-scale of mid-ocean ridge processes and should not be considered meaningful model ages. The above sets of calculations indicate that the isotopic heterogeneity in the mantle pre-dates the current melting and melt-rock reaction processes under the ridge axis.

4.4.4 Origin of the pyroxenite veins

Coexisting pyroxenites and peridotites in Dredge 96 were selected for analysis to address the possibility of pyroxenites as direct evidence for the “veined mantle hypothesis”. Although a consensus exists as to the presence of recycled crustal components in the mantle, based on isotopic heterogeneities among OIBs (e.g., White and Hofmann, 1982; Allègre and Turcotte, 1986; Zindler and Hart, 1986; Hofmann, 1997), no consensus exists as to the form that recycled crust takes in the mantle following millions of years of convection. The occurrence of pyroxenites in peridotite massifs led Allègre and Turcotte (1986) to suggest the marble-cake mantle hypothesis, in which recycled crust has been stretched and boudinaged to length-scales below meaningful thermodynamic integrity and absorbed into the peridotitic lithology. While numerous studies have documented mineralogical, chemical and isotopic variations among pyroxenites (e.g., Frey and Prinz, 1978; Polvé and Allègre, 1980; Zindler et al., 1983; Hamelin and Allègre, 1988; Piccardo et al., 1988; McDonough and Frey, 1989; Mukasa et al., 1991; Shervais and Mukasa, 1991; Pearson et al., 1993; Python and Ceuleneer, 2003), these pyroxenites have all been identified as representing recent magmatic addition to the host peridotite, rather than ancient recycled crust older than the host. Only studies which include $\delta^{18}\text{O}$ analyses have made a compelling case for recycled ancient crust in the mantle as the original source of present-day pyroxenites or eclogites in massifs and xenoliths (e.g., Pearson et al., 1993; Jacob et al., 1994; Barth et al., 2001). The presence of recycled crust in the asthenospheric mantle has also been invoked

to explain isotopic variations in MORB (e.g., Allègre et al., 1984; Kellogg et al., 2002; Salters and Dick, 2002) and to reconcile geochemical garnet signatures in MORB and the thickness of oceanic crust (Hirschmann and Stolper, 1996).

Pieces of recycled crust in the form of pyroxenites or eclogites should possess certain characteristics. (1) Mineralogically, they are most likely garnet bearing or they should have the signature of garnet breakdown, such as high HREE concentrations in Cpx (Hart and Dunn, 1993; Vannucci et al., 1993). (2) Major element mineral compositions are expected to have lower Mg# than peridotite counterparts, even if the precursor eclogites underwent melting in a subduction zone. For example, experimental data at 2-3 GPa show that Cpx in residual eclogites have Mg# ranging from 0.60 to 0.76, depending on the degree of melting (Rapp et al., 1999; Pertermann and Hirschmann, 2003a,b). Coexisting garnets have even lower Mg# (0.48 - 0.63) and the garnet breakdown reaction will decrease the Mg# of the recrystallized “low-pressure” Cpx. Eclogitic Cpx are also characterized by a high abundance of the jadeite component. For example, Pertermann and Hirschmann (2003b) report 14-28 mol% jadeite with 2-4 wt% Na₂O, and 14-17 wt% Al₂O₃. (3) The expected Nd isotopic composition of recycled oceanic crust depends on the time since its formation. As $(\text{Sm/Nd})_{\text{MORB}} < (\text{Sm/Nd})_{\text{Source}}$, radiogenic growth of $^{143}\text{Nd}/^{144}\text{Nd}$ is decelerated in oceanic crust relative to its source mantle. Oceanic crust of 1 Ga with an average MORB Sm/Nd ratio of 0.3356 (Hofmann, 1988) would have $^{143}\text{Nd}/^{144}\text{Nd} = 0.512829$, assuming that it formed from a depleted mantle source that differentiated from bulk silicate earth (BSE) at 3 Ga. Recycled crust of a greater age would have a less radiogenic Nd isotopic composition because the depleted mantle would have evolved for less time before crustal formation. Thus, coexisting pyroxenites and peridotites should have distinct $^{143}\text{Nd}/^{144}\text{Nd}$ compositions, with the latter having a more radiogenic (i.e. depleted) composition.

Our observations of Dredge 96 pyroxenites and the RC27-9-6-2 clinopyroxenite do not fit any of the expected characteristics for direct samples of recycled crust. No garnet or ev-

idence for garnet precursors, such as garnet-like REE patterns in pyroxenes, are observed in the pyroxenites and the pyroxenite REE patterns are identical to peridotite REE patterns. Spinel-pyroxene symplectites, a common product of garnet breakdown in peridotite massifs, are not observed. The major element compositions of Cpx, Opx, olivine and spinel in Dredge 96 pyroxenites (Tables 5.4-5.7) are identical to their peridotite counterparts. For example, the Mg# of Cpx and Opx range from 89.1% to 90.7%, much greater than typical eclogitic values. The abundance of the jadeite component in Cpx is extremely low, with barely measurable $\text{Na}_2\text{O} < 1 \text{ wt\%}$ and 5.1-7.7 wt% Al_2O_3 , significantly lower than the range cited above. Overall, pyroxene compositions are peridotitic rather than eclogitic.

The third characteristic of our pyroxenites that demonstrates that they are not direct remnants of recycled oceanic crust are the pairs of Cpx isotope analyses from pyroxenite veins and host peridotites (Van7-96-19, Van7-96-21, RC27-9-6-2; Table 4.4). For each pair, the pyroxenite and peridotite either have identical Nd isotope ratios or the pyroxenite is more depleted than the peridotite, neither of which corresponds to the expected composition of mantle containing recycled ancient oceanic crust. The two largest pyroxenite samples (Van7-96-09 and Van7-96-16) are both coarse-grained (5-10 mm) and have distinct isotopic compositions (Table 4.4, Fig. 4.6). Pyroxenite Van7-96-16 defines the enriched end of the isotopic spectrum, with both $^{143}\text{Nd}/^{144}\text{Nd}$ and $^{87}\text{Sr}/^{86}\text{Sr}$ compositions indistinguishable from Bouvet OIBs (e.g., O'Nions et al., 1977; Kurz et al., 1998) and the most enriched Oblique Segment MORBs (Standish, 2006). Based on these observations, we conclude that the pyroxenites in this study are not direct samples of recycled oceanic crust, though ultimately the enriched endmember of their isotopic range may have been derived from a recycled crustal component within Bouvet plume.

The range of isotopic compositions and trace element abundance patterns of Dredge 96 peridotites are best explained by reaction with and addition of melt fractions derived from enriched Bouvet material. This material could either be veined or unveined mantle,

either from Bouvet plume or mantle metasomatized by the Bouvet plume during its passage along the Shaka Fracture Zone at 15-25 Ma. The fact that the pyroxenites share chemical and isotopic compositions with associated peridotites suggests that near-equilibrium conditions prevailed at temperatures close to but slightly above the peridotite solidus. It is conceivable that melts from the Bouvet-generated source migrated through partially melting asthenospheric peridotite and locally reacted with pyroxenes. This resulted in a wide array of reaction products on the sub-kilometer scale, ranging from the almost non-reacted (peridotite Van7-96-38, which has a very depleted isotopic signature) to the extensively reacted (pyroxenite Van7-96-16).

The existing literature on abyssal pyroxenites is limited to descriptions of less than ten samples combined from the EPR (Constantin et al., 1995), MAR (Serri et al., 1988; Juteau et al., 1990; Kempton and Stephens, 1997) and SWIR (Dantas et al., 2007). All these occurrences describe pyroxenites of depleted compositions similar to adjacent peridotites, suggesting that, along with the results of this study, the bulk of the abyssal pyroxenites do not provide direct evidence for the veined mantle hypothesis. Instead, they indicate variable amounts of melt crystallization in the mantle, starting at variable depths.

4.4.5 Relationship between peridotites and basalts

Previous studies of radiogenic isotopes in abyssal peridotites have attempted to reconcile the isotopic composition of peridotites with that of basalts, to determine whether they complement basalt compositions or whether they indicate the existence of an additional mantle component. Snow et al. (1994) and Kempton and Stephens (1997) found that abyssal peridotites have similar isotopic compositions to MORBs and are from the same, depleted mantle source. Salters and Dick (2002) looked in detail at the Oblique Segment and the Atlantis II Fracture Zone and concluded that peridotites are isotopically more depleted in comparison to associated basalts. They advocated for the existence of an enriched pyrox-

enite component in the mantle that is removed by melting and sampled by basalts, but not preserved in the peridotites. Cipriani et al. (2004) analyzed basalts and peridotites from the Vema Lithospheric Section on the MAR and found that the age-corrected averages for the two lithologies are not statistically different.

In this study, we have found large isotopic variations among peridotites and pyroxenites from one dredge (Dredge 96, Fig. 4.6). They cover 46% of the Nd isotopic range of Indian Ocean MORB and encompass the Nd and Sr isotopic range of basalts from the Oblique Segment (Standish, 2006). Also noticeable is the fact that basalts from Bouvet hotspot plot at the “enriched” end of the Oblique peridotite and basalt spectrums and that the hotspot was proximal to the Oblique Segment at 15-25 Ma (Fig. 4.1). These observations suggest a genetic link between Dredge 96 enriched peridotites and pyroxenites and the Bouvet hotspot, through melt-rock reaction and veining of mantle peridotites by Bouvet-derived melts at ~20 Ma. Bouvet-modified mantle then upwelled beneath the Oblique ridge axis, producing melts from the metasomatized mantle with chemically and isotopically enriched signatures. These melts subsequently migrated through and reacted with the depleted mantle, resulting in the chemical and isotopic variations observed among both basalts and peridotites at the present-day ridge axis.

A genetic link between Bouvet source mantle and the Oblique Segment is supported by the Pb isotopic composition of Cpx, which have similar radiogenic compositions to Bouvet basalts (Table 4.5 and Fig. 4.7). Peridotite Cpx also have similar Pb isotopic compositions to primary sulfide (monosulfide solid solution) grains in peridotite Van7-96-28, as determined by *in situ* analyses with the Misasa multi-collector SIMS 1270 (two grain average: $^{207}\text{Pb}/^{206}\text{Pb} = 0.7907$ and $^{208}\text{Pb}/^{206}\text{Pb} = 1.9984$, compared to 0.7947 and 1.9955 for Cpx, Table 4.5). As uranium is absent in sulfide, $\mu (^{238}\text{U}/^{204}\text{Pb}) = 0$, whereas $\mu = 17.8$ in Cpx (Table 4.3). Hence, the apparent isotopic equilibrium between Cpx and sulfide in the peridotite must be recent, suggesting that the radiogenic Pb signature of both Cpx and

sulfide are from migration of a Bouvet-derived melt through the mantle. As Van7-96-28 is among the relatively enriched samples from Dredge 96, both in terms of Nd and Sr isotopic ratios and REE abundance patterns (Fig. 4.5), this supports the Pb isotope argument for recent Bouvet-derived melt metasomatism of the mantle.

The variations in Nd and Sr isotopic compositions of peridotites and basalts along the Oblique Segment and the comparison to Bouvet Island basalts are shown in Fig. 4.11. Similar Bouvet-derived metasomatism of the mantle may have occurred in other parts of the Oblique Segment, as the five samples from dredge PS86-6, reported by Salters and Dick (2002), range from $^{143}\text{Nd}/^{144}\text{Nd}$ of 0.513202 (more depleted than DMM) to 0.512985 (similar to Bouvet basalts). Among the basalts, samples from the Narrowgate region, far to the east of the hotspot track, have the greatest overlap with the Bouvet isotopic range (Standish, 2006). From the overlapping ranges of the basalts and peridotites, we conclude that all mantle isotopic components observed in regional basalts are found in the peridotites and that the existence of an enriched component specifically in the form of pyroxenite veins is not required to explain our dataset. Basalts do not extend to the extremely depleted compositions of the peridotites, because of mass balance constraints. Peridotites with depleted $^{143}\text{Nd}/^{144}\text{Nd}$ compositions (such as Dredge 85) must be derived from high Sm/Nd ratios and thus relatively low Nd concentrations, with respect to enriched peridotites. Hence, the contribution of depleted peridotites to the melt mixtures that form basalts is small compared to the contribution from enriched compositions.

Atlantis II Fracture Zone is not associated with any hotspot track, but exhibits isotopic enrichment in some basalts. This enrichment is matched by the Nd isotopic enrichment in the matrix of peridotite RC27-9-6-2. As shown in Fig. 4.12, the isotopic enrichment in basalts does not occur specifically on one side of the fracture zone or at a specific location along the fracture zone. The transform volcano, located adjacent to the sampling location of peridotite RC27-9-6-2 (Fig. 4.1), ranges in composition from DMM ($^{143}\text{Nd}/^{144}\text{Nd} =$

0.51313) to an enriched composition ($^{143}\text{Nd}/^{144}\text{Nd} = 0.51298$). RC27-9-6-2 ranges from this enriched composition in the peridotite matrix ($^{143}\text{Nd}/^{144}\text{Nd} = 0.512915$) to a near-DMM composition in the clinopyroxenite vein ($^{143}\text{Nd}/^{144}\text{Nd} = 0.513099$). Hence, in one peridotite sample we observe the entire isotopic range of basalts from the transform volcano. The randomly distributed isotopic enrichment of the Atlantis II Fracture Zone suggests that isotopic heterogeneities are ubiquitous and exist on a sub-kilometer scale. This enriched component may only be recognizable at low degrees of melting, such as at fracture zones.

As observed in this study, chemical and isotopic heterogeneities could be both ancient and modern, and could occur on the sub-kilometer scale. Isotopic enrichment or depletion requires long term enrichment or depletion in trace element concentrations, so the enrichment signature will dominate over the depletion signature in melts produced from a heterogeneous mantle. If this picture is generally true for the mantle beneath mid-ocean ridges, basalts should carry slightly more enriched (or less depleted) isotopic signatures than their associated peridotites. In Fig. 4.13, we demonstrate this by plotting histograms of basalt and peridotite Nd isotopic compositions for the Oblique Segment, the Atlantis II Fracture Zone, and the global ridge system. It is particularly noticeable that the global peridotite average ($^{143}\text{Nd}/^{144}\text{Nd} = 0.513167$) is more depleted than the MORB-derived estimate of DMM (0.51313; Workman and Hart, 2005), and that there is a significant population of peridotites with much more depleted Nd isotopic compositions (extending to 0.51366) than D-DMM (0.51326). This suggests that the asthenospheric mantle is indeed much more depleted than generally thought and the implications of this suggestion are discussed below.

4.4.6 Variation of isotopic composition with peridotite melt depletion

Prior to considering the implications of ultra-depleted compositions and mantle heterogeneity, we assess the extent to which peridotites exhibit long term trace element depletion

associated with isotopic depletion. Peridotites in a heterogeneous mantle should demonstrate correlations between isotope depletion and indicators of degree of melting. In contrast, recent melting at the ridge axis will deplete peridotite concentrations without changing the isotopic composition.

In Fig. 4.14, the variations of Cpx Nd isotopic composition with indicators of degree of melting – modal Cpx, spinel Cr#, and Cpx (Ce/Yb)_N – suggest a possible link between mineralogical, chemical and isotopic signatures of depletion. However, as all peridotites have been affected by recent melting processes, considerable scatter exists for the global data set. When samples from Dredge 85 and Dredge 96 are considered by themselves, reasonable correlations between indicators of major and trace element depletion and isotopic depletion are observed. This suggests that at least some of the variability in isotopic composition of the peridotites reflects ancient melting events.

Dredge 85 peridotites, which have more radiogenic ¹⁴³Nd/¹⁴⁴Nd, are more depleted in modal Cpx, with Cpx having lower (Ce/Yb)_N and (Nd/Sm)_N than Dredge 96 peridotites. Among Dredge 96 peridotites, samples with more depleted isotopic signatures have more depleted trace element ratios. Incorporation of available data for abyssal peridotites from the SWIR (Snow et al., 1994; Salters and Dick, 2002) follows the same general trend. These correlations are broad, with correlation coefficients <0.3, but no correlation should be present if the SWIR peridotites formed as simple melt extraction residues from a uniform source mantle. Moreover, variable degrees of melting and melt-rock reaction during the recent melting process will have obscured pre-existing trends between major and trace elements and isotopes. The negative correlation between Nd isotopes and the Cr# of co-existing spinel rules out a simple relationship between isotopic and chemical depletion, as spinel Cr# should increase with increasing melt depletion. These observations suggest a combination of pre-existing heterogeneity, recent melting and in some regions, production of enriched Cpx by melt-rock reaction with an exotic component (e.g., Bouvet plume).

To examine in more detail the relationship between enriched and depleted sources among peridotites, we plot the variation of $^{143}\text{Nd}/^{144}\text{Nd}$ as a function of inverse Nd concentration in Fig. 4.15. Mixing between enriched and depleted sources with different daughter isotope concentrations results in a linear relationship on an inverse concentration diagram (Faure and Mensing, 2005). As shown in Fig. 4.15, a linear regression of Oblique Segment data produces only a minimal correlation and regression of the Atlantis II Fracture Zone dataset produce an anti-correlation. The lack of a mixing relationship on the large scale emphasizes both the complicated nature of melt-rock reaction, the effects of melt removal (e.g. the extremely low Nd concentration samples from the fracture zone in Fig. 4.15), and the heterogeneous nature of the mantle.

On the local scale, the three analyzed samples from dredge RC27-9-6 appear to demonstrate mixing between enriched and depleted sources, with a correlation coefficient of 0.8 (Fig. 4.15). Cpx from the vein in RC27-9-6-2 plots on the line between matrix Cpx and a second, depleted sample from the dredge. Hence, the vein is a mixture of the isotopically and chemically enriched peridotite in which it sits and melt from a more isotopically and chemically depleted peridotite. This depleted peridotite is more depleted than DMM, but the products of mixing – recorded in the vein and a third peridotite from the dredge – are chemically and isotopically similar to DMM (Fig. 4.15).

The major element characteristics of Dredge 85 peridotites suggest low degrees of melting, whereas the trace and isotopic composition suggests extreme depletion. Spinel Cr#’s are low and the dredge is located on an obliquely spreading ridge section (Fig. 4.1) with little evidence for significant mantle melting and thin to absent volcanic crust (Dick et al., 2003; Standish et al., submitted). However, peridotite trace element concentrations and Nd isotope ratios are considerably more depleted than DMM. Also worth noting is the fact that these samples are unlikely to have been affected by trapped melt, unless this melt was ultra-depleted, as metasomatism would have a considerable effect on samples with

low trace element concentrations. Dredge 85 peridotites contain no evidence for late melt impregnation and gabbroic rocks are absent in the region. The isotopic, and to some extent the trace element, composition of these peridotites must have been established prior to melt extraction at the SWIR. Hence, the mantle beneath the SWIR has contained depleted peridotites for a long period of time.

Heterogeneously depleted mantle is implied by comparison of the MORB-based estimate of DMM (Su and Langmuir, 2003; Workman and Hart, 2005) to Cpx compositions in Fig. 4.14. DMM and D-DMM are both less depleted than abyssal peridotites, not just in terms of isotopic composition, but also with respect to Cpx mode, LREE/HREE ratios (e.g., Ce/Yb), and Sp Cr#. As will be discussed below, the present-day Sm/Nd ratio of DMM (on the basis of the continuous depletion model) is too low relative to the ratios measured in depleted abyssal peridotites such as those from Dredge 85. This discrepancy cannot be reconciled as an effect of melt extraction, because predicted degrees of melting are too large given the lack of crust (basalts and gabbros) in the region of Dredge 85. Therefore, we suggest that the correlations in Fig. 4.14, although too broad for detailed discussion, carry important information as to the geochemical prehistory of the asthenospheric mantle. We caution against the prevailing assumption that the entire range of chemical variations observed among abyssal peridotites were generated by melt extraction beneath the current set of ocean ridges.

4.4.7 Implications for mantle composition

Results from the present study, combined with the data of Salters and Dick (2002) and Cipriani et al. (2004), suggest that the asthenospheric mantle (DMM) is isotopically more depleted than previously considered (e.g., Workman and Hart, 2005). For example, Figs. 4.6 and 4.11-4.12 demonstrate that the most depleted SWIR peridotites are more depleted than D-DMM of Workman and Hart (2005), with $^{87}\text{Sr}/^{86}\text{Sr}$ as low as 0.702096 and

$^{143}\text{Nd}/^{144}\text{Nd}$ up to 0.513417. Since radiogenic ingrowth of $^{143}\text{Nd}/^{144}\text{Nd}$ following melt extraction beneath the SWIR was found to be small, the observed isotopic signature requires a long-term depletion of Nd relative to Sm. Such long-term LREE depletion suggests that there exists, in the asthenospheric mantle, more strongly LREE-depleted regions than suggested by MORBs. These concentration variations are less apparent in surveys of MORB compositions because the Nd mass balance for aggregated melts formed by melt fractions derived from variously depleted source peridotites is dominated by the least depleted end-member. Hence, MORBs are skewed towards more isotopically enriched compositions than peridotites (Fig. 4.13).

The presence of ultra-depleted mantle requires a reassessment of the history of mantle depletion, to produce higher Sm/Nd and lower Rb/Sr than current estimates for DMM. The continuous transport, melt depletion model of Workman and Hart (2005) predicts the evolution of a parent/daughter ratio in DMM based on the isotopic ratios of DMM and BSE and the parent/daughter ratio of BSE. In this model, material is continuously removed from the mantle from the time at which depletion begins (t). Workman and Hart estimate t to be 3 Ga, as a best estimate for the start of significant continental crustal growth (Taylor and McLennan, 1995). The parent/daughter ratios of BSE and DMM are related by a transport coefficient, k :

$$k = \frac{1}{t} \ln \left[\frac{\left(\frac{^{147}\text{Sm}}{^{144}\text{Nd}} \right)_{BSE}^0}{\left(\frac{^{147}\text{Sm}}{^{144}\text{Nd}} \right)_{DMM}^0} \right] \quad (4.8)$$

where the subscript 0 indicates present-day values. The evolution of the isotopic composition of DMM is dependent on k and t :

$$\left(\frac{^{143}\text{Nd}}{^{144}\text{Nd}} \right)_{DMM}^0 = \left(\frac{^{143}\text{Nd}}{^{144}\text{Nd}} \right)_{BSE}^t + \frac{\lambda \left(\frac{^{147}\text{Sm}}{^{144}\text{Nd}} \right)_{BSE}^t [1 - e^{-(\lambda+k)t}]}{\lambda + k} \quad (4.9)$$

In Fig. 4.16A, we calculate the variation of $^{143}\text{Nd}/^{144}\text{Nd}$ as a function of bulk peri-

dotite Sm/Nd ratio, for different initial depletion times. In addition, we show a traditional two-stage evolution model (cf. Workman and Hart, 2005), in which all depletion occurs at initial time t . We plot abyssal peridotite data in Fig. 4.16A, using the age corrected Cpx $^{143}\text{Nd}/^{144}\text{Nd}$ ratios and assuming that these are representative of the whole rock. Bulk Sm/Nd ratios are reconstructed from Cpx concentrations and the bulk partition coefficient calculated from mineral/melt partition coefficients and the peridotite mode (Equations 1-2 in Workman and Hart, 2005). The abyssal peridotite data trend away from the 3 Ga depletion line towards higher Sm/Nd ratios. In comparison to the range among abyssal peridotites, the difference between the continuous and two stage depletion models is relatively small.

To explain the peridotite data in the context of DMM evolution, we consider the effects of melting, which produces high Sm/Nd ratios without changing the $^{143}\text{Nd}/^{144}\text{Nd}$ ratio. In Fig. 4.16B, we plot Sm/Nd ratios produced by different degrees of modal fractional melting from an initial DMM Sm/Nd ratio of 0.411 (Workman and Hart, 2005). We use modal melting instead of non-modal, for consistency with Workman and Hart (2005). While the simple melting model appears to explain the Sm/Nd peridotite range, the predicted $\sim 10\%$ melting for Dredge 85 does not correspond to observations of the dredge location. As discussed previously, Dredge 85 is from an amagmatic section of the ridge (Fig. 4.1) and the combined slow spreading rate and obliquity to the plate direction implies minimal melting during mantle upwelling. Using a more sophisticated melting model might reduce the amount of melting predicted for Dredge 85, but this dredge would still appear to have undergone higher degrees of melting than dredges associated with significant crustal production. A second difficulty with using any melting model to explain observed Sm/Nd variations is that peridotites from the same dredge should undergo the same degree of melting, as they have experienced the same temperature regime. Hence, even if variation in degree of melting could explain the Sm/Nd ratios between dredges, it cannot explain the few percent

variation implied by Sm/Nd ranges within dredges. The Sm/Nd ratios of peridotites, while invariably affected to some degree by recent melting, must largely be inherited signatures that predate recent ridge processes.

In another attempt to explain peridotite isotopic and trace element depletions, we modify the continuous depletion model by removing material at different times and calculating the present day ratio of this material. In Fig. 4.17, we plot the evolution of BSE and DMM Nd isotopic compositions with time and with Sm/Nd ratio. We then calculate the evolution of material removed at various times from DMM, which is found to evolve to compositions between present-day BSE and DMM. As observed in Fig. 4.17, the limitation of this model is that it cannot produce compositions outside the BSE-DMM range, whereas peridotites extend to much higher Sm/Nd ratios. Our results indicate that the evolution of the asthenosphere is more complicated than implied by depletion models constrained by BSE and DMM alone, particularly when the Sm/Nd ratio of DMM is determined from the present-day $^{143}\text{Nd}/^{144}\text{Nd}$ ratio of DMM. A more accurate model of asthenosphere evolution needs to explicitly include the effect of melting on Sm/Nd ratios at various times in the past. In addition, the effects of material return at subduction zones need to be included, to explain the enriched compositions of peridotites such as RC27-9-6-2, which plot on the enriched side of BSE in Fig. 4.17.

If depleted peridotites such as Dredge 85 form a volumetrically significant component of the asthenospheric mantle and if enriched components such as RC27-9-6-2 are relatively ubiquitous in the mantle, then estimates of melting systematics based on MORB compositions need to be reassessed. Workman and Hart (2005) concluded that MORB is produced by 6% aggregated fractional melting of DMM, resulting in a Nd concentration in MORB of 8.37 ppm. Alternatively, the upper mantle can be considered a mixture of two (or more) components, for example the E-DMM and D-DMM calculated by Workman and Hart (2005). A 6% melt of each of these lithologies, mixed in proportions of 55% D-

DMM and 45% E-DMM, produces a MORB-like melt with 8.37 ppm Nd and $^{143}\text{Nd}/^{144}\text{Nd} = 0.51312$. This calculation assumes similar solidi for both end-members, which is fine for variable modes but not for variable mineral compositions (Pickering-Witter and Johnston, 2000; Schwab and Johnston, 2001).

Based on the above discussion of depletion models, the Nd concentrations of depleted peridotites are likely to be considerably more depleted than D-DMM. Thus, depleted peridotites such as in Dredge 85 would contribute a relatively small amount to the observed trace and isotopic composition of MORBs. In the context of the veined mantle model discussed earlier, this implies that if pyroxenites are present in the mantle, they would dominate the melt isotopic signature and MORBs should have a more enriched isotopic signature than is observed. Hence, we propose that the upper mantle consists of variably depleted and enriched peridotites and that these variations in isotopic, trace, major and modal compositions, when properly incorporated into melt production and extraction models, can explain many of the observed variations in MORB composition.

4.5 Conclusions

Analyses of peridotites from the SWIR demonstrate greater isotopic and trace element variability on a smaller lengthscale than has previously been observed. We find that peridotites extend to more depleted compositions than estimates of DMM, with $^{87}\text{Sr}/^{86}\text{Sr}$ as low as 0.702096 and $^{143}\text{Nd}/^{144}\text{Nd}$ as high as 0.513335 among our samples. Other peridotites are relatively enriched and in one SWIR dredge we find 46% of the Nd isotopic range of Indian Ocean MORB. On the basis of our analyses, we reach several conclusions:

1. Cpx and Opx in abyssal peridotites are in Nd isotopic disequilibrium and the disequilibrium is not the result of recent radiogenic ingrowth. In all the peridotites for which both Cpx and Opx were measured, evidence for recent melt-rock reaction is

found. Hence, we suggest that isotopic disequilibrium reflects Opx REE disequilibrium, as Opx is coarse grained and has slower diffusion rates than Cpx. In addition, if the disequilibrium originated in Cpx, fractionated partition coefficient patterns and core-to-rim variations in trace elements should be observed, but are not.

2. Peridotites spend ~ 1 -20 My from the time that they pass through their closure temperature to the time that they are collected at variable distances from the ridge axis. However, radiogenic ingrowth during this time is negligible, with $<0.01\%$ change among our samples, despite slow upwelling rates along the Oblique Segment (partly due to low Nd/Sm ratios in many samples). Thus, the observed isotopic range of abyssal peridotites and the extremely depleted compositions among some peridotites reflects the composition of the asthenospheric mantle.
3. Pyroxenite veins in both Dredge 96 and RC27-9-6-2 formed by recent crystallization of melt in a conductively cooling mantle. Veins do not have the predicted compositions of recycled oceanic crust. For example, in sample RC27-9-6-2, the clinopyroxenite vein is more depleted than the host peridotite. In Dredge 96, the major, trace and isotopic compositions of the veins overlap those of the coexisting peridotites. The isotopic compositions of Dredge 96 samples indicate melt-rock reaction with Bouvet Hotspot material, which could have originated as recycled oceanic crust, but the samples do not provide direct evidence for the “veined mantle hypothesis”.
4. The same range of isotopic compositions is found among peridotites as among basalts. The average composition of peridotites is skewed towards slightly more depleted values, as the mass balance of aggregated melts formed by melt fractions from variably depleted sources is dominated by the least-depleted component. No missing (vein) component is required to reconcile MORB isotopic compositions with peridotite compositions. In particular, peridotite RC27-9-6-2 provides evidence for enriched

mantle unrelated to any hotspot.

5. Isotopic depletion requires long term chemical depletion. Peridotites show some evidence of a correlation between $^{143}\text{Nd}/^{144}\text{Nd}$ and indicators of melt depletion. However, peridotites have also undergone recent melt extraction, which complicates identification of long term chemical depletion trends.
6. Portions of the mantle are significantly more depleted than predictions of DMM composition derived from MORB isotopic ratios. The degree of melting suggested by the trace element composition of Dredge 85 peridotites is higher than the amount expected in this region of ultra-slow spreading. The isotopic and trace element compositions of peridotites cannot be reconciled with either a two stage model or a continuous depletion model of mantle isotopic evolution. In addition to recent melt extraction, the combination of high $^{143}\text{Nd}/^{144}\text{Nd}$ and high Sm/Nd requires ancient melt extraction episodes that are not accounted for by these models.

We conclude that the asthenospheric mantle is heterogeneous in terms of isotopes, major and trace elements, and modal composition. In some regions, the mantle is more depleted than current MORB-based estimates of DMM. The scale of heterogeneity is relatively small, probably down to < 1 km. These observations require a reassessment of current models of MORB production.

Appendix 4.A Major elements

Major element compositions of all primary silicate phases in the peridotites and pyroxenites were determined by electron microprobe, using a JEOL JXA-733 at the Massachusetts Institute of Technology. In Tables 5.4-5.8, mineral compositions are presented as averages of 5 or more spot analyses per grain. For pyroxenes, to average out the effects of exsolution, a 10 μ m defocussed beam was used and data were collected by line traverses on grain cores and rims. In addition, for all Cpx and some Opx, multiple grains were analyzed per sample and the total number of points measured are listed in Table 4.A2.

Fig. 4.A1 shows the variation of spinel Cr# with spinel Mg# and with TiO₂. During melting, spinel Cr# increases and Mg# decreases. TiO₂ concentrations greater than DMM (\sim 0.2 wt%), are indicative of melt refertilization in the peridotites. Spinel in Dredge 96 and RC27-9-6-2 indicate variable degrees of melting and melt fractional crystallization. In contrast, Dredge 85 peridotites have low TiO₂, suggesting that their compositions reflect only melting and melt extraction. In addition, Dredge 96 pyroxenite spinels typically have plagioclase rims, due to the second melt infiltration event and are thus not necessarily representative of processes that occurred during pyroxenite formation.

The variations of major elements in Cpx are shown in Fig. 4.A2. Dredge 96 peridotites and pyroxenites have unusually low Cr₂O₃, Mg# and high Na₂O, in comparison to other peridotites and pyroxenites. In contrast, RC27-9-6-2 has high Cr₂O₃, as evidenced by the bright green color of the Cpx versus the grey color of Dredge 96 Cpx. FeO and MgO in RC27-9-6-2 form a trend towards high concentrations, corresponding to a Mg# range of 89.8-92. This variation does not correspond to clinopyroxenite versus peridotite lithology or to distance from the clinopyroxenite vein.

4.6 Bibliography

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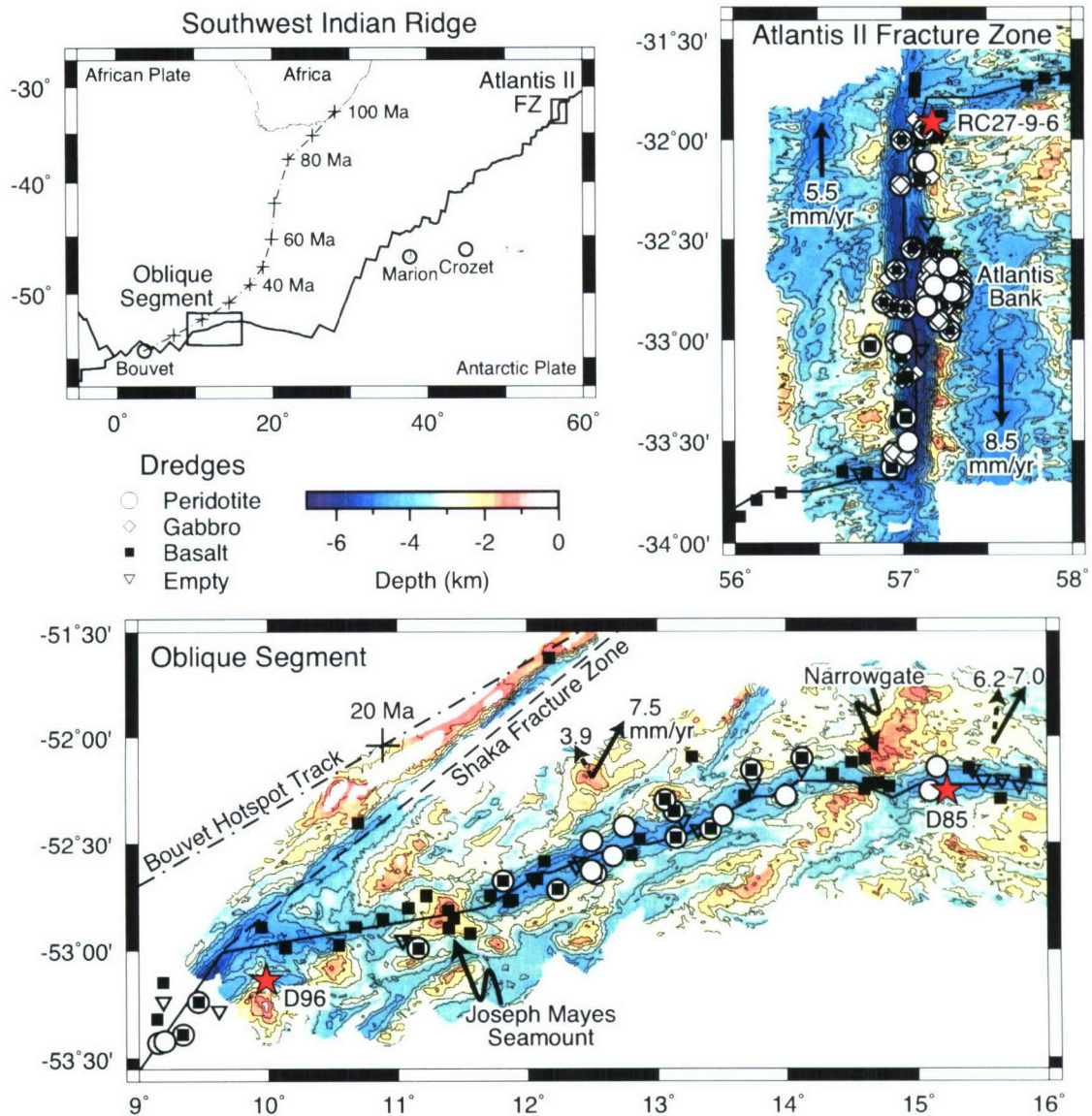


Figure 4.1: Maps of the Southwest Indian Ridge, Oblique Segment and Atlantis II Fracture Zone. Bouvet, Marion and Crozet are hotspots associated with the SWIR. The dot-dashed line indicates the path of Bouvet hotspot, calculated by Hartnady and le Roex (1985) using finite reconstruction poles from Morgan (1983). Bouvet hotspot passed along the trace of the Shaka Fracture Zone (dashed line) at the eastern end of the Oblique Segment from 15-25 Ma. No hotspot tracks are associated with the Atlantis II Fracture Zone. Atlantis Bank is an uplifted gabbroic massif located under the large cluster of dredges along the Atlantis II Fracture Zone. Solid arrows indicate the direction of plate motion. On the Oblique Segment, dashed arrows indicate the obliquity of the spreading direction to the plate direction.

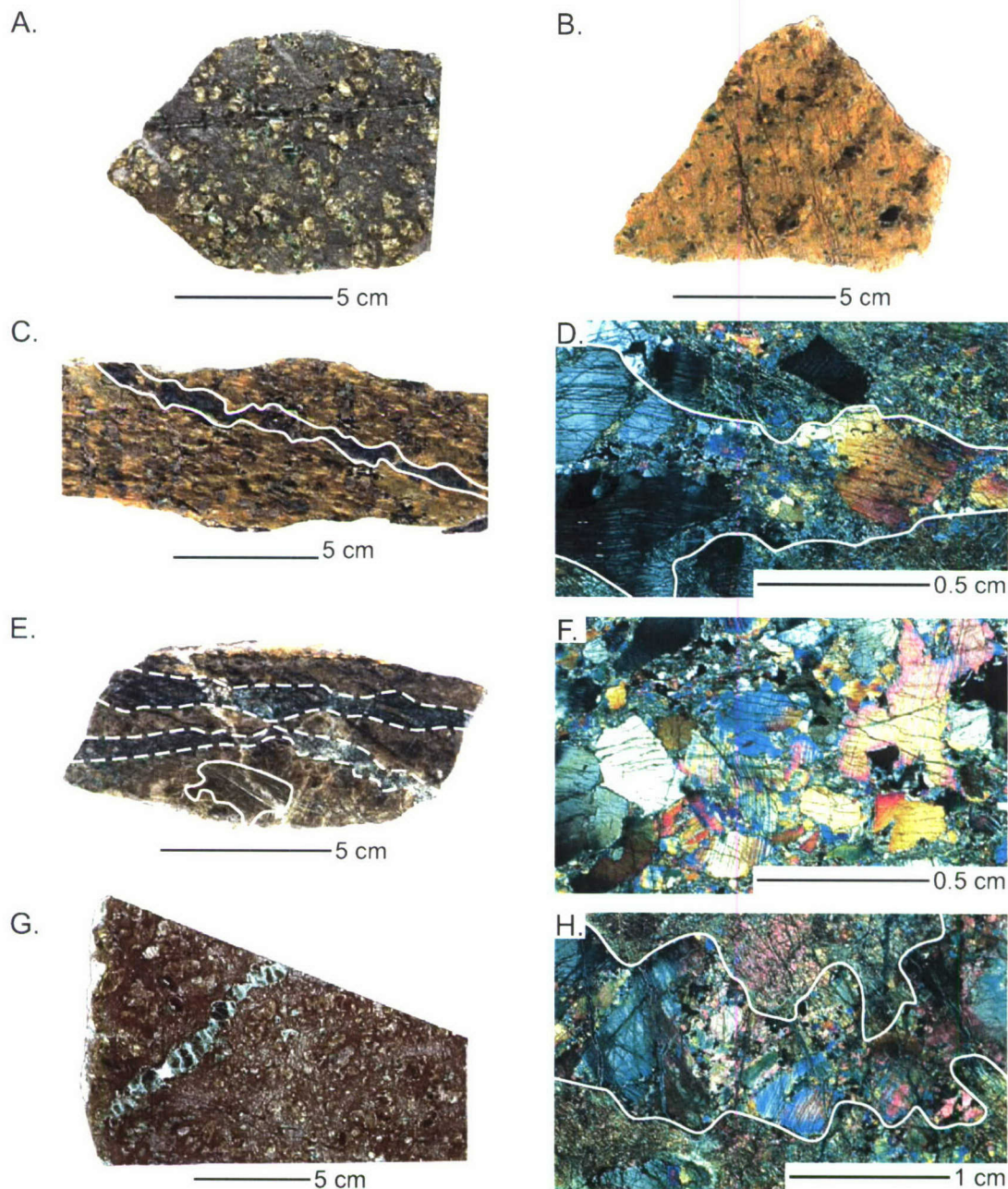


Figure 4.2: **A.** Depleted lherzolite, Van7-85-49, with 5% Cpx. **B.** Enriched peridotite, Van7-96-28, with 5% Cpx. Apart from the different alteration style, this peridotite has no visual characteristics that distinguish it from the Dredge 85 peridotite. **C.** Pyroxenite veined peridotite, Van7-96-21, with the vein outlined in white. **D.** Crossed polarized photomicrograph of the same sample. **E.** Pyroxenite, Van7-96-16, with a thin, altered, orange peridotite skin at the top. A particularly large Cpx grain is outlined in white. The pyroxenite is cross-cut by two later-stage veins of fine-grained Ol-Plag-Cpx, outlined with dashed lines. **F.** Crossed polarized photomicrograph of a pyroxenite, Van7-96-09, consisting of coarse-grained Cpx and Opx with minor olivine. The fine-grained, cross-cutting assemblage of Ol-Plag-Cpx is from the later-stage melt infiltration event. **G.** RC27-9-6-2, the clinopyroxenite veined lherzolite. **H.** Crossed polarized photomicrograph of the same sample, with the vein outlined in white.

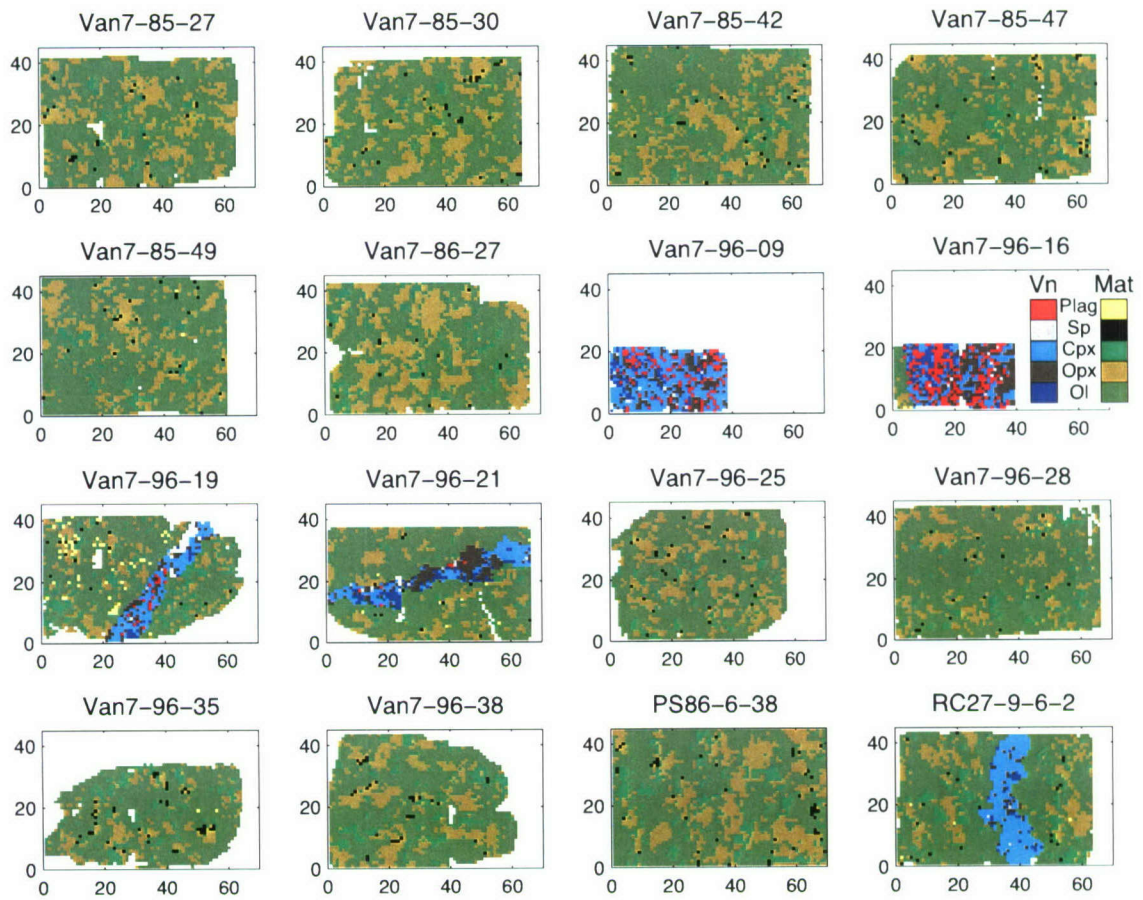


Figure 4.3: Mineral mode maps of peridotite and pyroxenite samples, created by point counting thin sections on a 1 mm grid. “Mat” refers to peridotite matrix and “Vn” to pyroxenite vein.

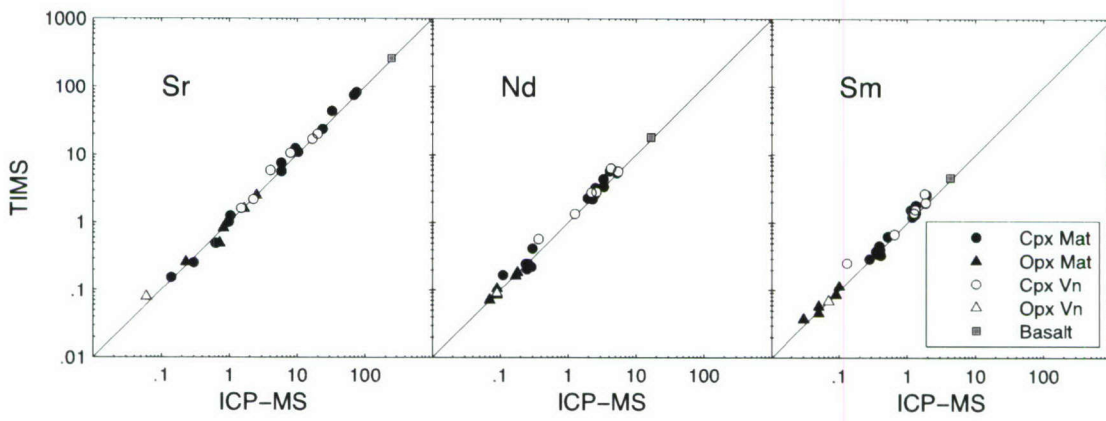


Figure 4.4: Comparison of ICPMS and TIMS data for Sr, Nd and Sm, in ppm. Solid line is a 1:1 correlation. Error bars are smaller than symbols.

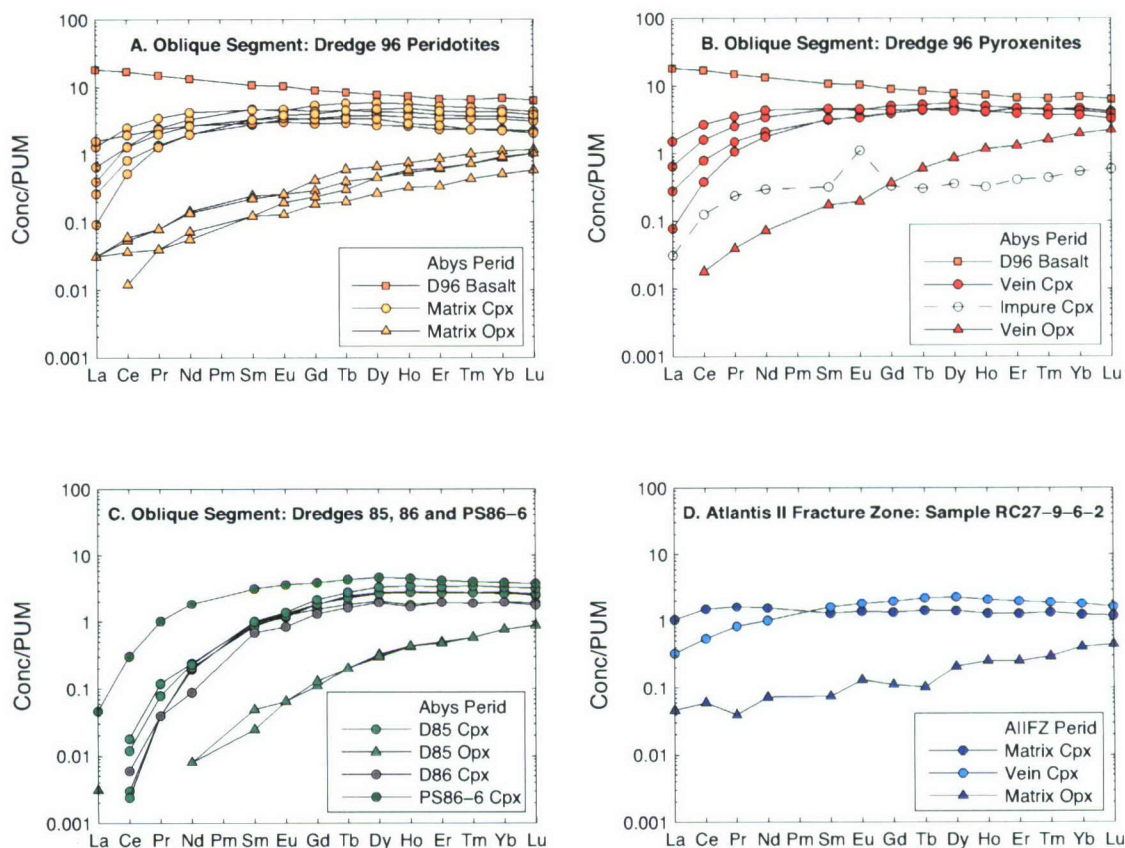


Figure 4.5: ICPMS rare earth element data for Cpx (circles) and Opx (triangles), normalized to PUM (McDonough and Sun, 1995). In **A-C**, the grey shaded field is the range for abyssal peridotite Cpx from Johnson et al. (1990). In **D**, the grey shaded field is the range for peridotite Cpx from the Atlantis II Fracture Zone (Johnson and Dick, 1992).

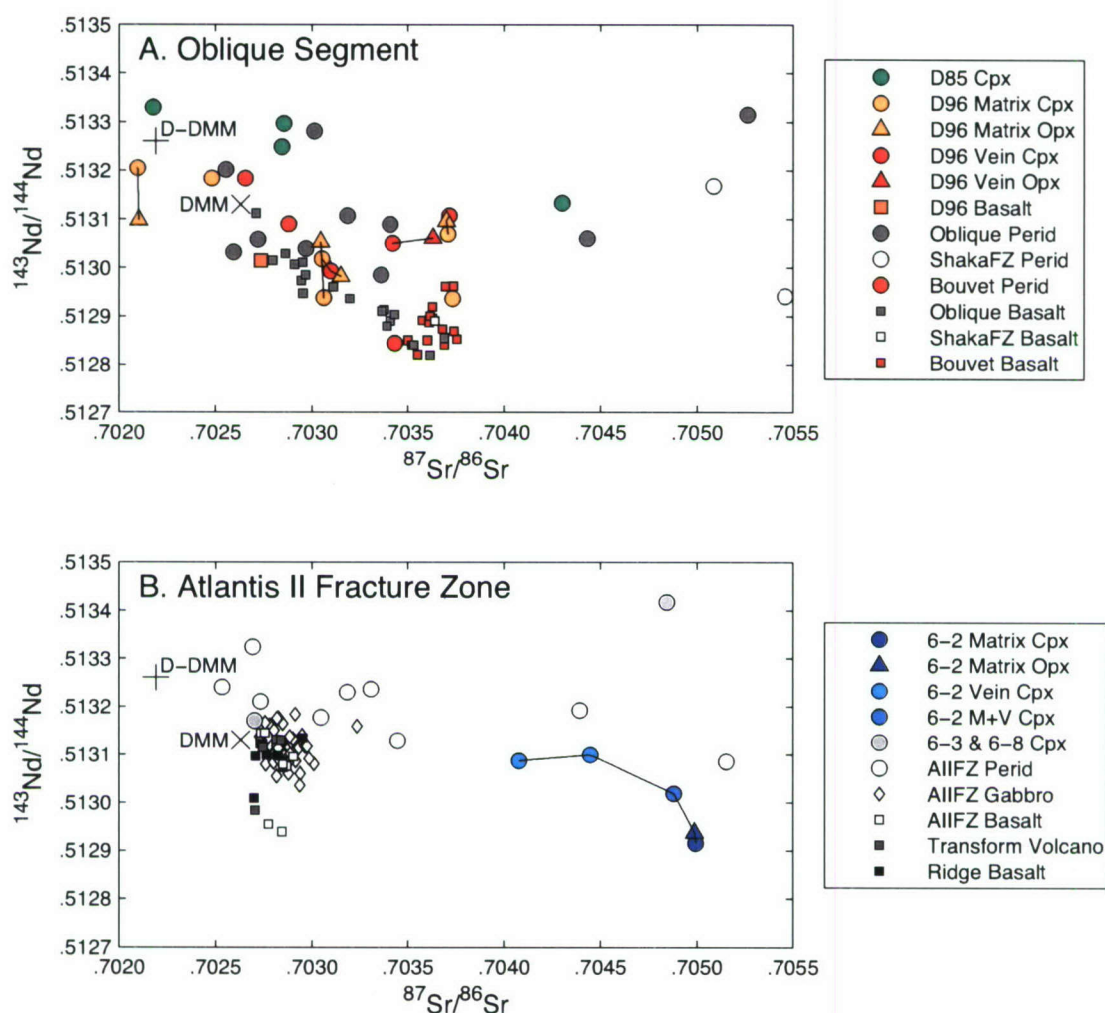


Figure 4.6: Nd and Sr isotopic variations in peridotites (from this study and literature), basalts and gabbros. All data are corrected to the standard values used in this study. Peridotite data from other studies are for Cpx mineral separates, except the Shaka Fracture Zone samples, which are whole rock analyses of the fresh cores of mylonitized peridotites. The Cpx referred to as [RC27-9-]6-2 M+V is a combined matrix and vein Cpx analysis (Snow et al., 1994) and the Cpx referred to as 6-3 & 6-8 are two samples from the same dredge as RC27-9-6-2.

Data sources: DMM from Su and Langmuir (2003); D-DMM from Workman and Hart (2005). **A.** Oblique Segment peridotites from Snow et al. (1994) and Salters and Dick (2002); Shaka Fracture Zone peridotites from Snow (1993); Bouvet regional peridotites from Snow et al. (1994); Oblique Segment basalts from le Roex et al. (1983, 1992), Mahoney et al. (1992), Janney et al. (2005) and Standish (2006); Shaka Fracture Zone basalts from le Roex et al. (1983); Bouvet basalts from O'Nions and Pankhurst (1974), O'Nions et al. (1977), Sun (1980), Kurz et al. (1998), Prestvik et al. (1999), Janney et al. (2005) and Barry et al. (2006). **B.** Atlantis II Fracture Zone peridotite from Snow et al. (1994) and Salters and Dick (2002); gabbro, all from Hole 735B on Atlantis Bank, from Kempton et al. (1991), Hart et al. (1999) and Holm (2002); fracture zone and transform volcano basalts from Mahoney et al. (1989) and Snow (1993); basalts from adjacent ridge segments from Hamelin and Allègre (1985), Snow (1993) and Meyzen et al. (2005).

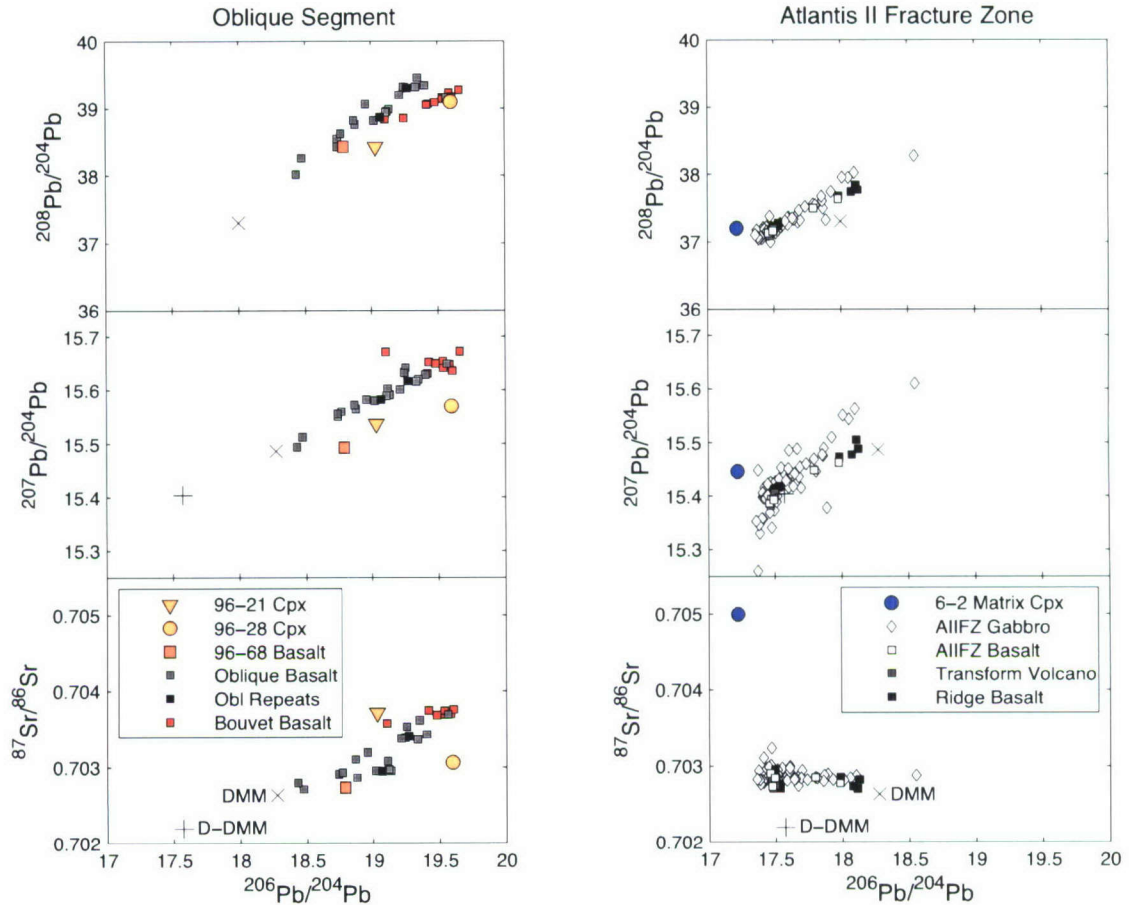


Figure 4.7: Variation of $^{206}\text{Pb}/^{204}\text{Pb}$ with $^{208}\text{Pb}/^{204}\text{Pb}$, $^{207}\text{Pb}/^{204}\text{Pb}$ and $^{87}\text{Sr}/^{86}\text{Sr}$ in peridotite Cpx and basalt glasses from this study. Also shown are related basalts and gabbros and the estimated compositions of DMM and D-DMM; see Figure 4.6 for data references. The two black squares among Oblique Segment basalts are repeat analyses of samples analyzed by Standish (2006), which plot within the general trend between DMM and Bouvet, unlike the Dredge 96 samples.

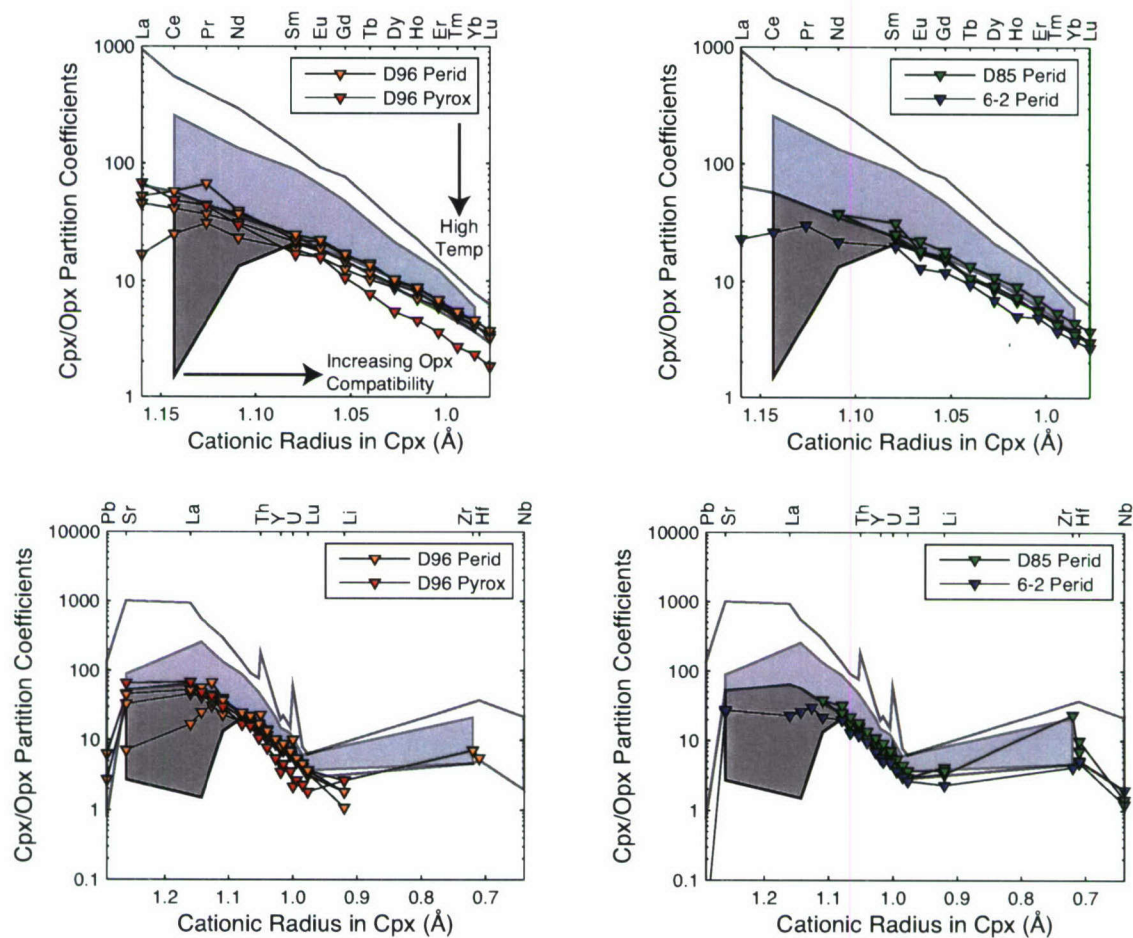


Figure 4.8: Cpx/OpX partition coefficients in abyssal peridotites, plus one pyroxenite, plotted as a function of cationic radius in Cpx. For the elements Pb to Li, this corresponds to VIII-fold coordination in the M2 sites whereas for the elements Zr-Nb, this corresponds to VI-fold coordination in the M1 site. REE variations are plotted in the top panels and the full set of trace elements in the bottom panels. The light grey field is the partition coefficient range from Witt-Eickschen and O'Neill (2005) for xenoliths equilibrated over a range of temperatures from 900°C (high values) to 1250°C (low values). The dark grey field is the partition coefficient range for Hellebrand et al. (2005) from *in situ* analyses of abyssal peridotite pyroxenes.

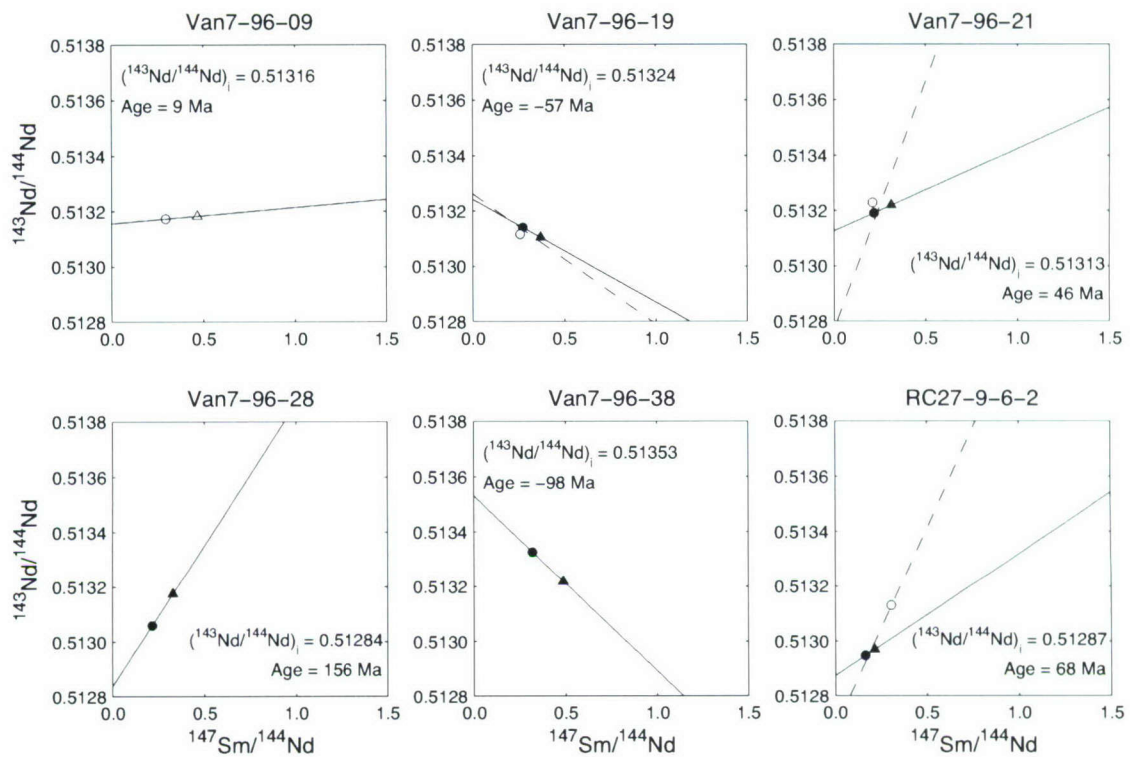


Figure 4.9: Nd isochrons for samples with analyses of more than one mineral phase. Circles are Cpx and triangles are Opx; filled symbols are peridotite minerals and open symbols are vein minerals. Solid line is a linear least squares regression through Cpx and Opx mineral pairs; dashed line is a regression through matrix Cpx, matrix Opx and vein Cpx. Initial $^{143}\text{Nd}/^{144}\text{Nd}$ and ages are for Cpx-Opx pairs only.

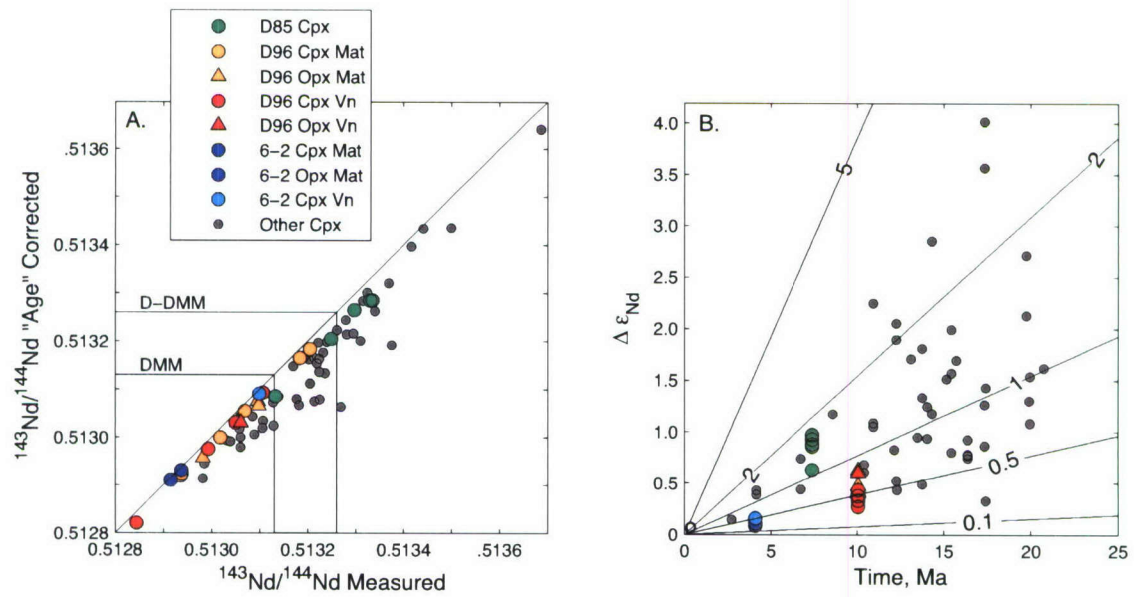


Figure 4.10: **A.** "Age" corrected $^{143}\text{Nd}/^{144}\text{Nd}$ compared to measured $^{143}\text{Nd}/^{144}\text{Nd}$. The solid line is a 1:1 correlation line. Samples either plot on the line or towards higher measured values, due to ingrowth in the time since the sample left the 1200°C isotherm. Data for other Cpx are from Snow (1993), Snow et al. (1994), Salters and Dick (2002), and Cipriani et al. (2004). **B.** The difference in ϵ_{Nd} between the measured and corrected $^{143}\text{Nd}/^{144}\text{Nd}$ values, as a function of time since the sample left the 1200°C isotherm. Contours are for constant Sm/Nd ratios.

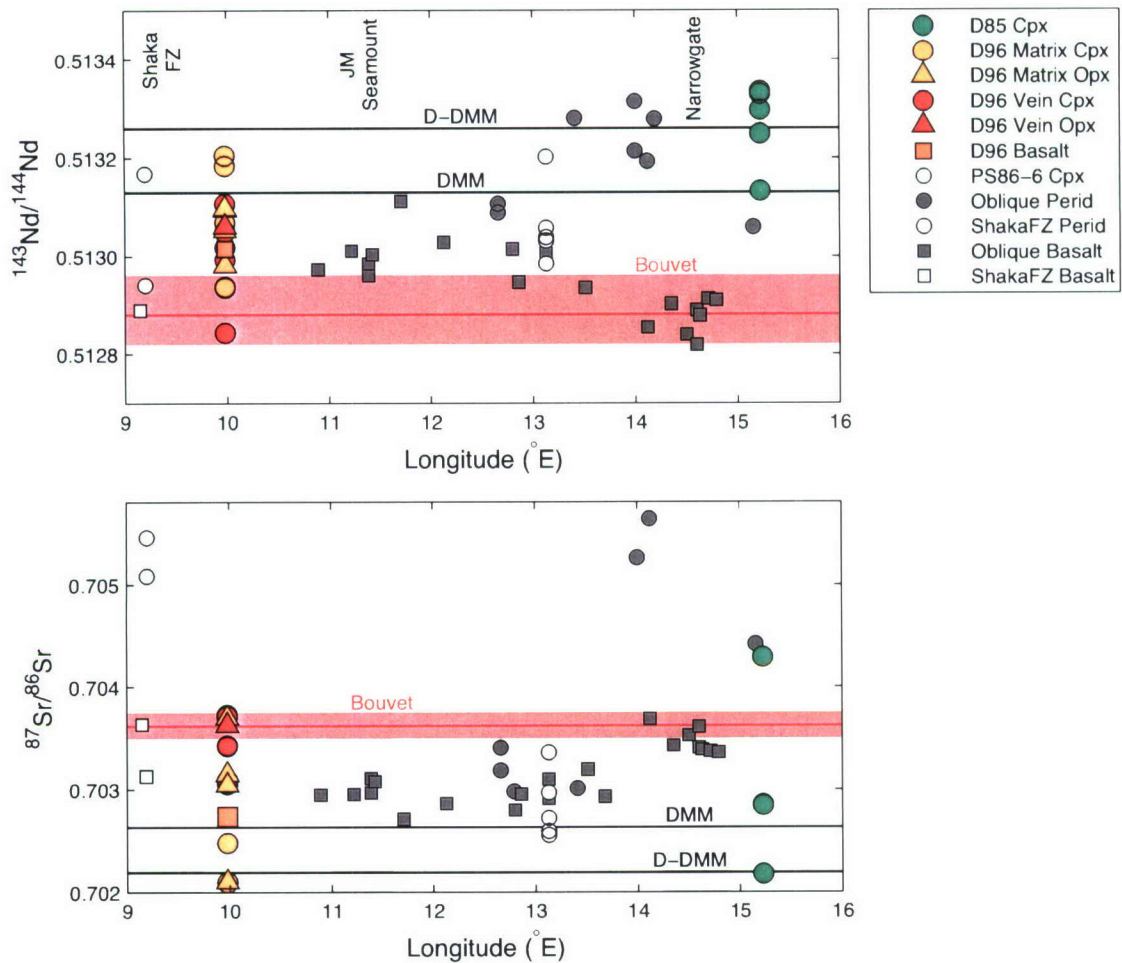


Figure 4.11: Variation of Nd and Sr isotopes along the SWIR Oblique Segment. Joseph Mayes Seamount and Narrowgate are segments where spreading is approximately orthogonal, volcanic activity occurs and significant basaltic crust is present. These features are generally absent along the rest of the Oblique Segment. Data for other peridotite Cpx and basalts are from references in Figure 4.6.

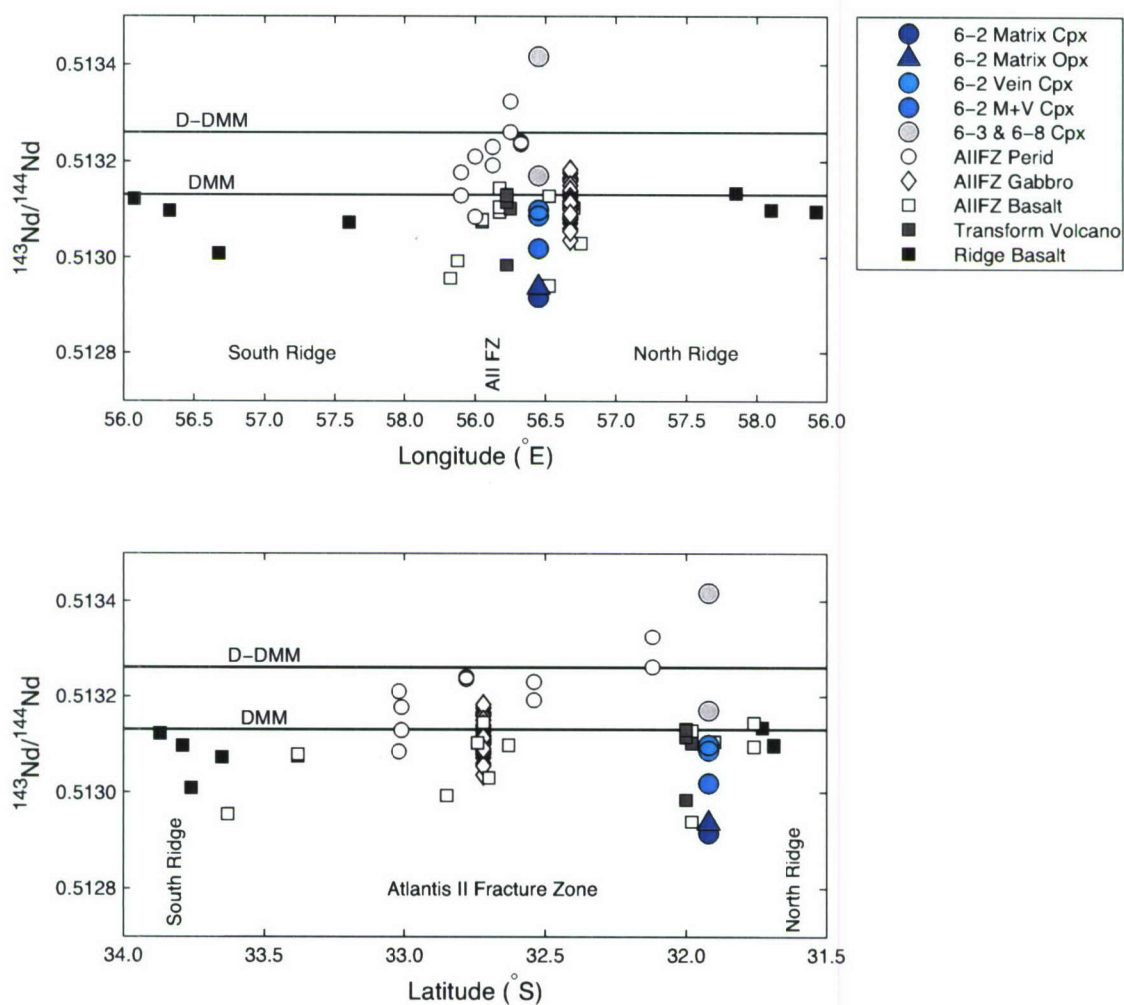


Figure 4.12: Variation of Nd across (longitude, top panel) and along (latitude, bottom panel) the Atlantis II Fracture Zone. No systematic variation is observed in either direction. Cpx 6-3 & 6-8 refer to two samples from the same dredge as RC27-9-6-2. Data for other peridotite Cpx, gabbros, and basalts are from references in Figure 4.6, with additional basalt data from Coogan et al. (2004).

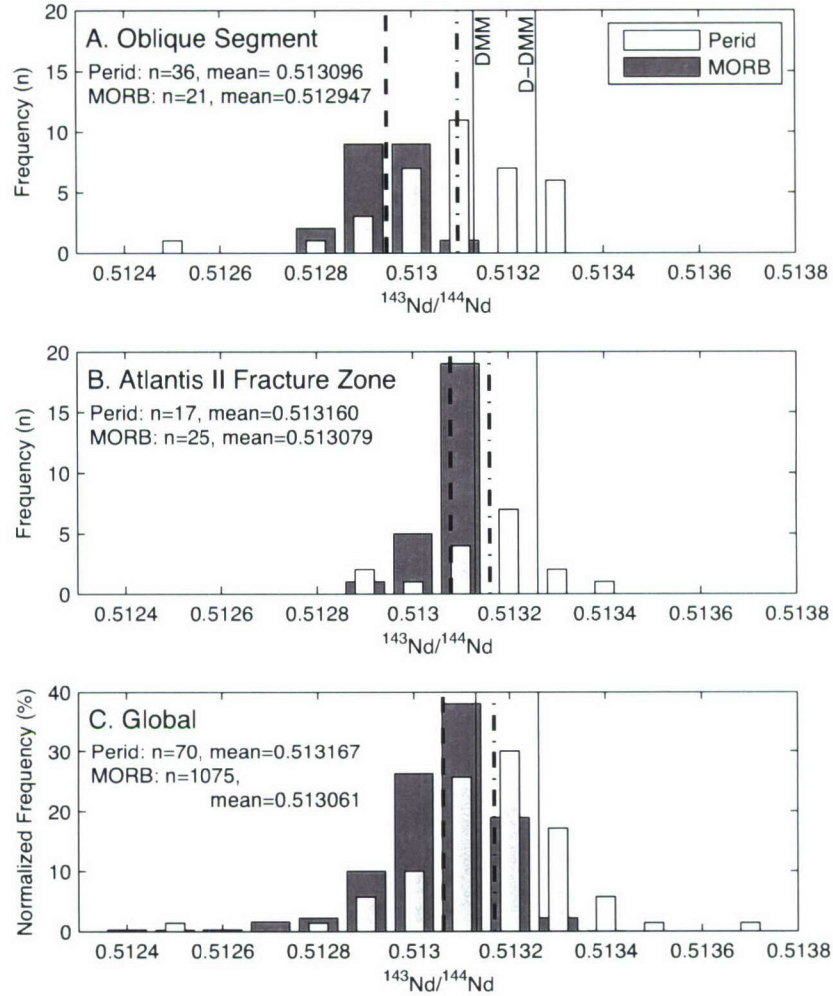


Figure 4.13: Histograms of peridotite and basalt Nd isotopic ratios for **(A)** the Oblique Segment, **(B)** the Atlantis II Fracture Zone, and **(C)** global abyssal peridotites and MORBs. Mean values for the peridotites are indicated by the dot-dashed lines and the basalts by the dashed lines. Note that Oblique Segment basalts and peridotites are shifted to lower average $^{143}\text{Nd}/^{144}\text{Nd}$ due to the influence of Bouvet Hotspot. Data in A and B are from this study and references in Figure 4.6; global peridotite data in C also include data from Kempton and Stephens (1997) and Cipriani et al. (2004); global MORB is a compilation from PetDB (<http://www.petdb.org>, Lehnert et al., 2000).

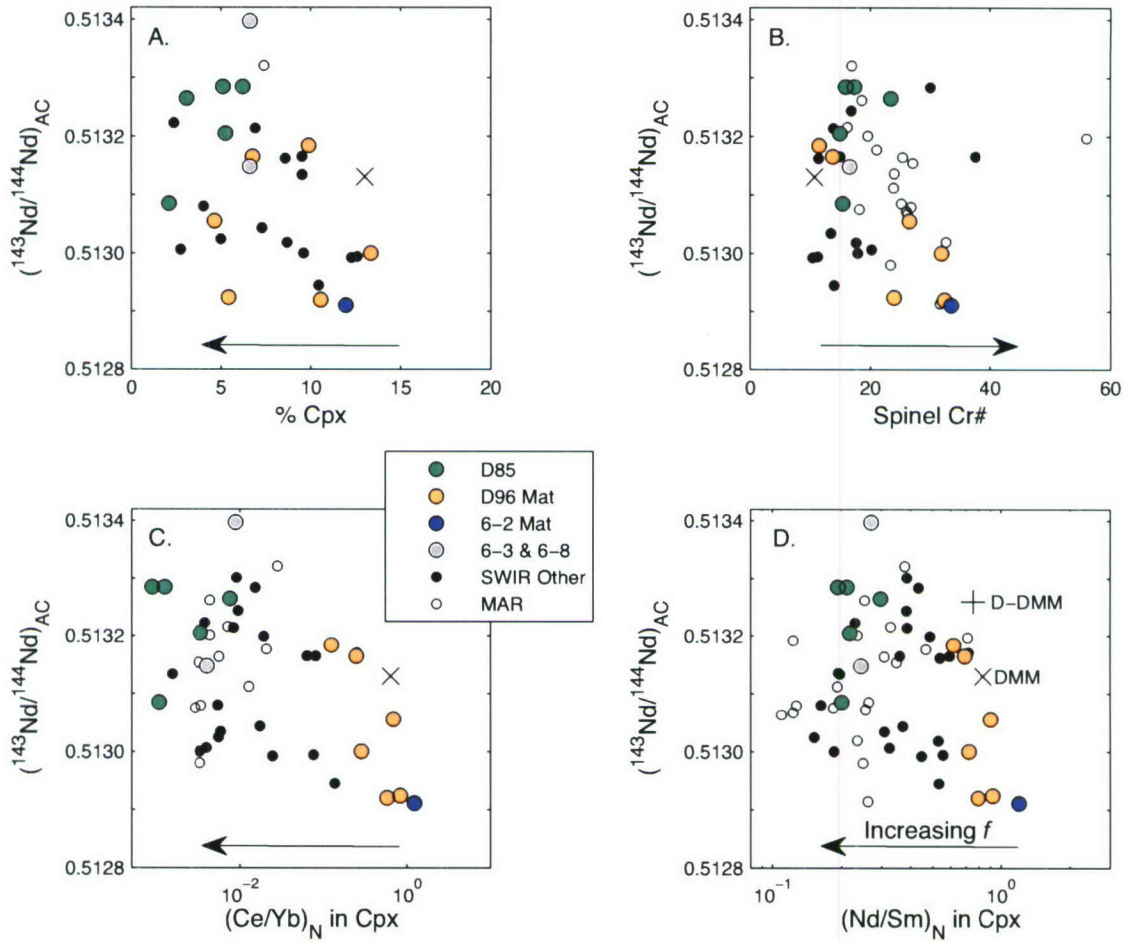


Figure 4.14: The variation of abyssal peridotite Cpx $^{143}\text{Nd}/^{144}\text{Nd}$ as a function of modal Cpx, spinel Cr#, Ce/Yb, and Nd/Sm. $^{143}\text{Nd}/^{144}\text{Nd}$ values are age corrected to the 1200°C isotherm, Cr# is calculated as $\text{Cr}/(\text{Cr}+\text{Al}) \times 100$, and the N subscript for REE ratios indicates normalization to PUM. Modal Cpx, Ce/Yb in Cpx and spinel Cr# are traditionally used as indices of degree of peridotite melting, with the first two parameters decreasing with increasing melting and the third parameter increasing during melting. Arrows indicate the direction of change in the compositional parameters with increasing degree of recent melting (f). Data for other SWIR Cpx are from this study and references in Figure 4.6; data for MAR Cpx are from Cipriani et al. (2004) and Brunelli et al. (2006); values for DMM and D-DMM are from Su and Langmuir (2003) and Workman and Hart (2005).

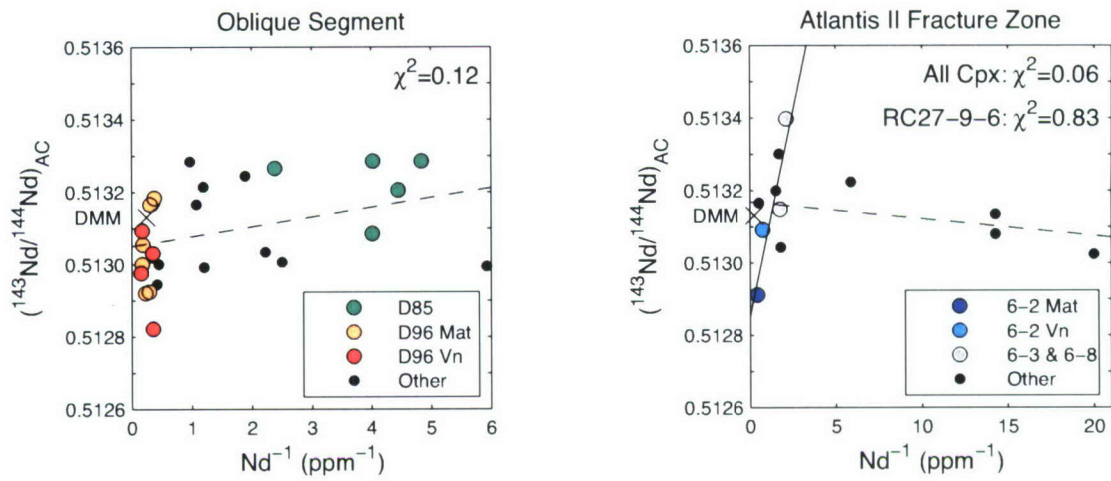


Figure 4.15: Variation of Nd isotopic ratio as a function of inverse Nd concentration for the Oblique Segment and the Atlantis II Fracture Zone. $^{143}\text{Nd}/^{144}\text{Nd}$ values are age corrected to the 1200°C isotherm and the location of DMM is shown for comparison to observed compositions. Dashed lines are regressions through the datasets for each locality and the χ^2 values indicate the lack of a significant correlation. The solid line is a regression through the three samples from dredge RC27-9-6, which yield a good correlation. Data for other SWIR Cpx are from this study and references in Figure 4.6.

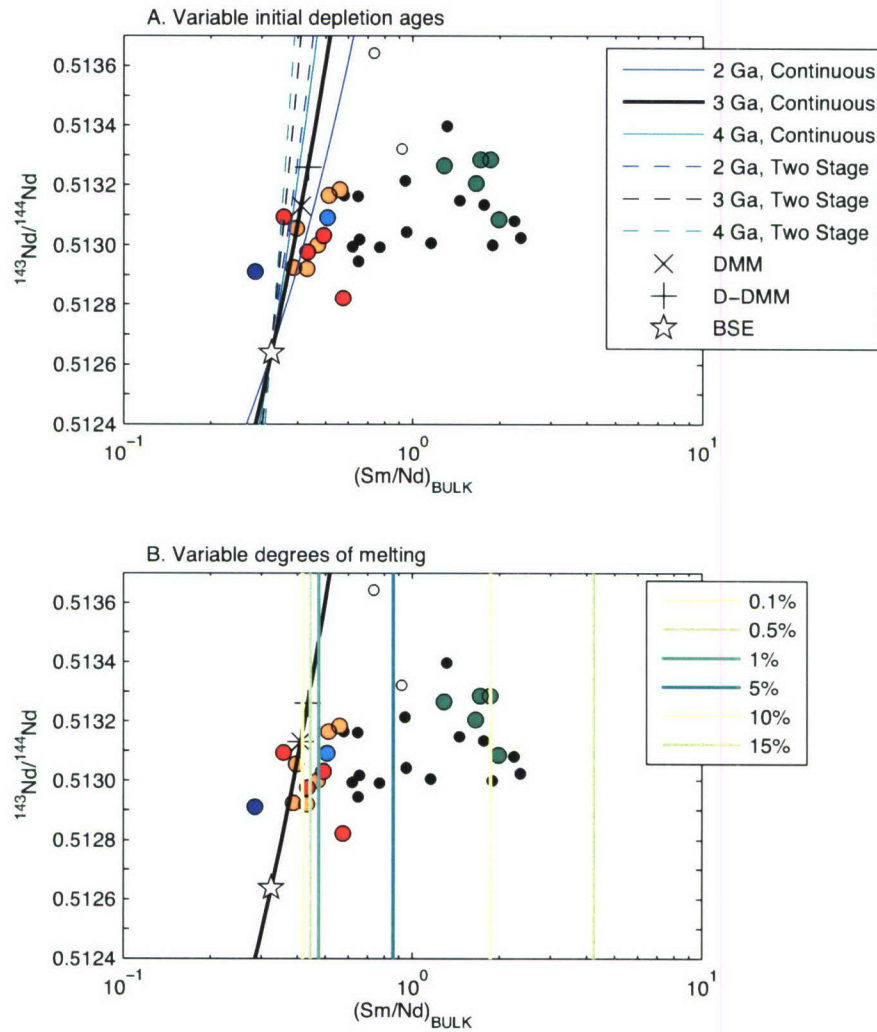


Figure 4.16: **A.** Predicted present-day mantle composition based on the continuous (solid lines) and two-stage (dashed lines) depletion models and the composition of BSE. The preferred model of Workman and Hart (2005) is the solid black line for continuous depletion at 3 Ga and their estimates of DMM and D-DMM lie on this line. Reconstructed peridotite bulk Sm/Nd and $^{143}\text{Nd}/^{144}\text{Nd}$ from this study and the literature (see previous figures for symbols and references) extend over a larger range of Sm/Nd compositions than predicted by the models. **B.** The effect of modal fractional melting on the Sm/Nd ratio of DMM. The horizontal spread in the Sm/Nd ratios of the peridotites can be explained by variable degrees of melting from an initial DMM composition. However, the highest degree of melting is predicted for Dredge 85 peridotites (green circles), which come from a section of the SWIR that lacks crust and thus cannot have undergone $\sim 10\%$ melting. Peridotite bulk Sm/Nd compositions are reconstructed from Cpx concentrations and modal analyses. Peridotite Nd ratios are age corrected values for Cpx, assumed to be representative of the whole rock.

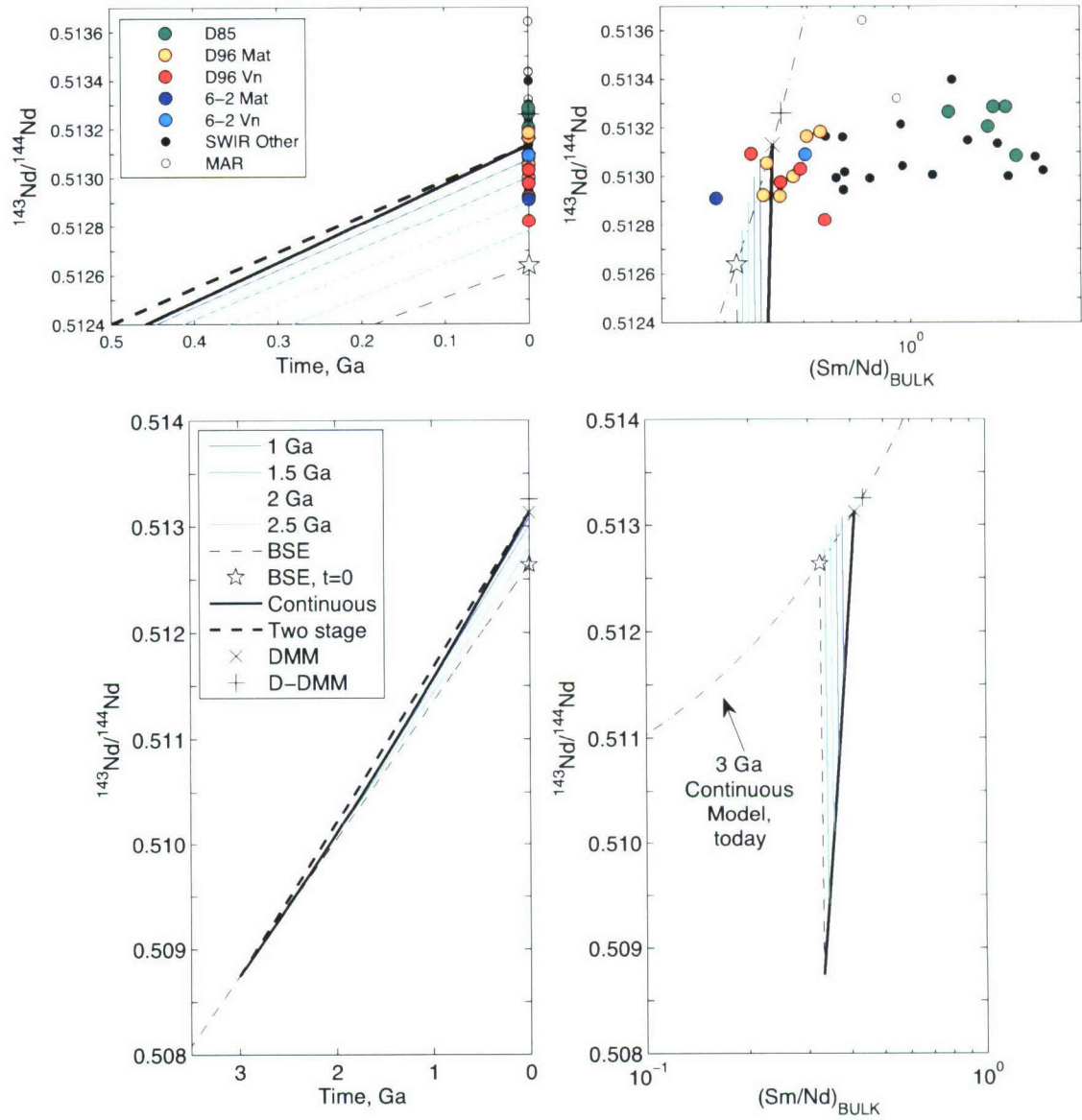


Figure 4.17: Models of DMM isotopic and trace element evolution through time, following the two stage model, the continuous depletion model and the continuous depletion model with periodic removal, all assuming initial depletion of BSE at 3 Ga. The continuous depletion and two stage models follow slightly different trajectories, but are constrained to produce the same present-day $^{143}\text{Nd}/^{144}\text{Nd}$ DMM ratio. Material removed at different times from the continuously depleting mantle produces a range of present-day compositions between DMM and BSE. The dot-dashed line is the present-day variation of $^{143}\text{Nd}/^{144}\text{Nd}$ and Sm/Nd ratios for the continuous depletion model (for depletion beginning at 3 Ga) and is the same as the solid black line in Figure 4.16.

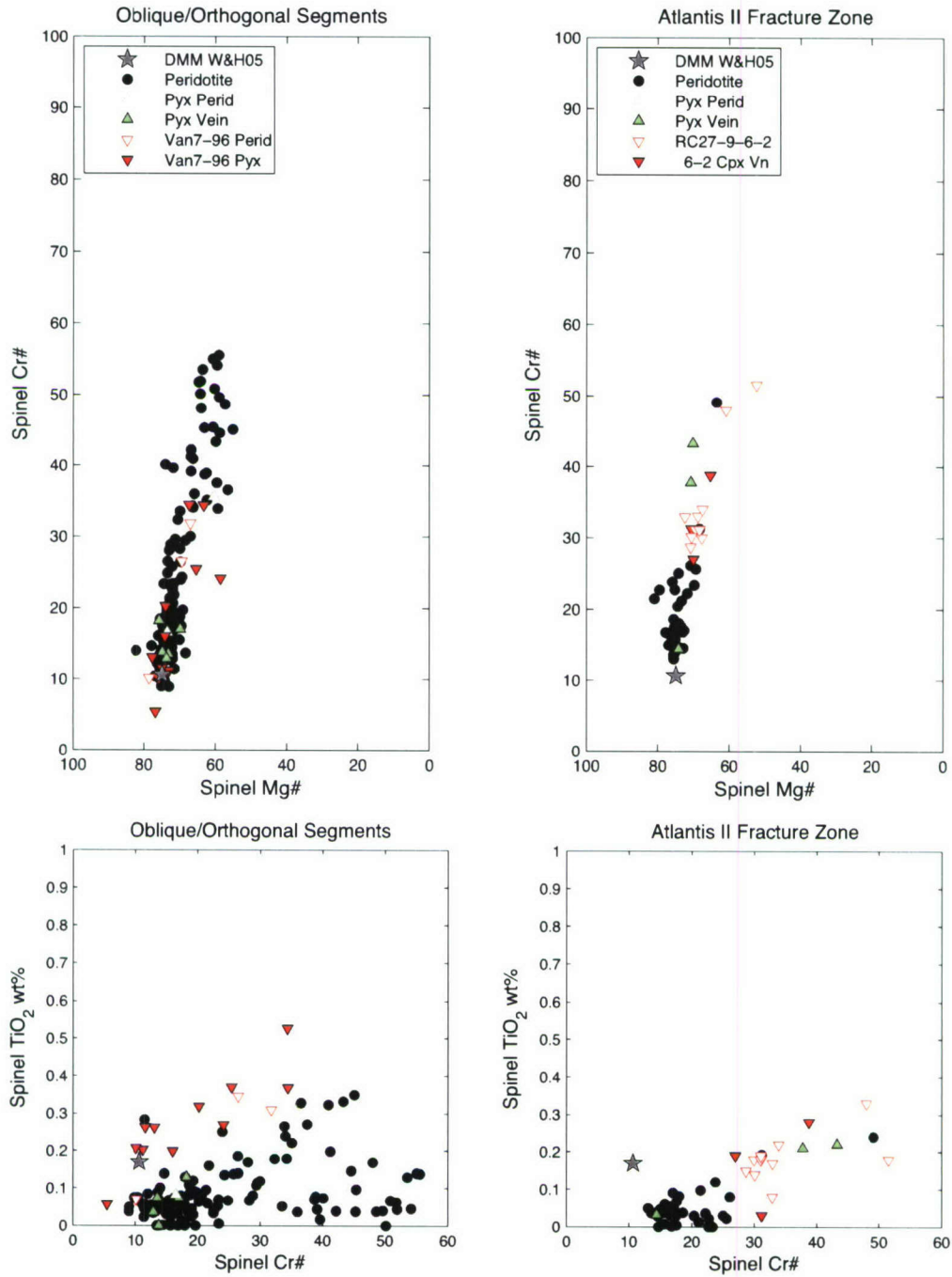


Figure 4.A1: Variations in spinel Cr#, Mg# and TiO₂ in peridotites and pyroxenites from this study. Cr# is calculated as $\text{Cr}/(\text{Cr}+\text{Al}) \times 100$ and Mg# as $\text{Mg}/(\text{Mg}+\text{Fe}) \times 100$. Also shown are data for spatially associated peridotites and pyroxenites, from the dataset in Chapter 5. The composition of spinel in DMM is from Workman and Hart (2005).

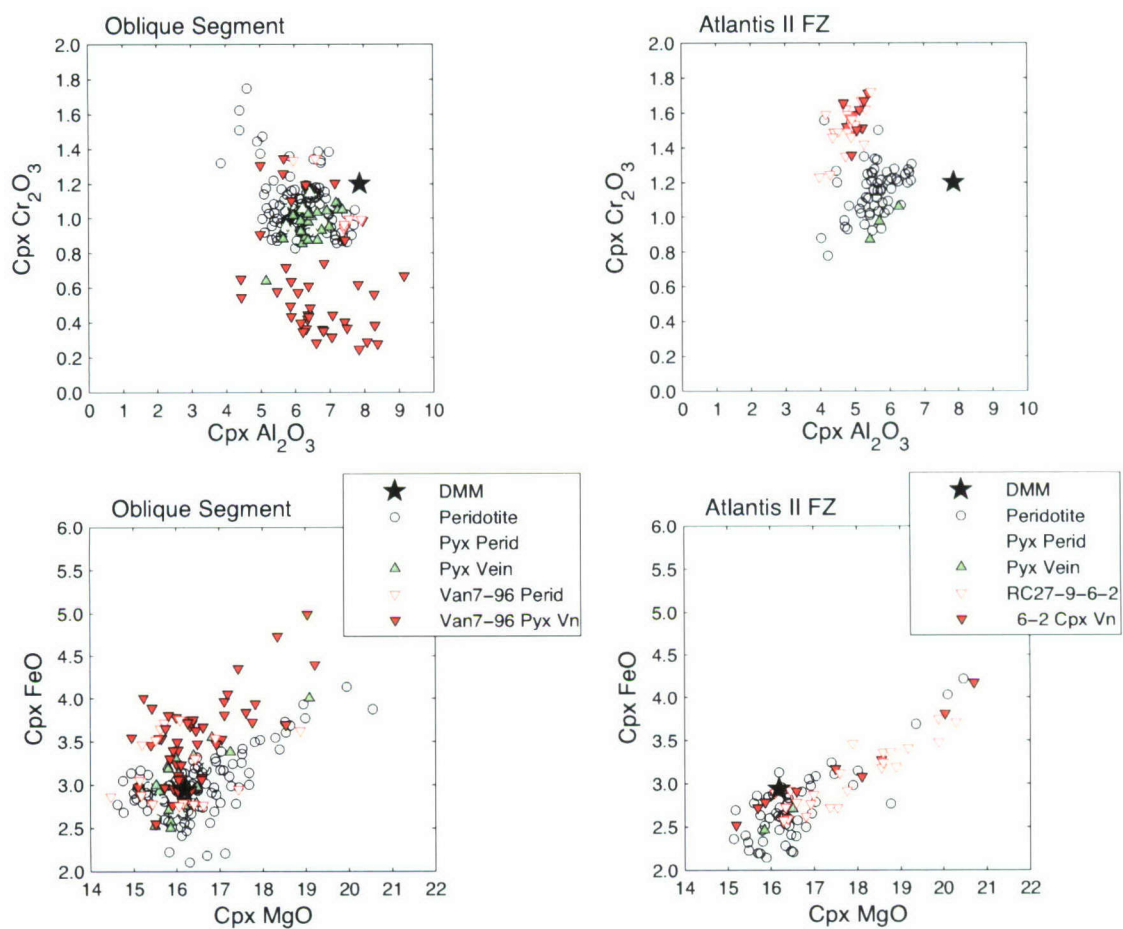


Figure 4.A2: Variations Cpx major element compositions among peridotites and pyroxenites from this study and spatially associated samples (from the dataset in Chapter 5). The composition of Cpx in DMM is from Workman and Hart (2005).

Table 4.1: Dredge locations

Dredge	Contents	Wt kg	On/Off Bottom			FSR ^a mm/yr	Angle ^a	ESR ^a mm/yr	Location
			Lat °S	Long °E	Depth m				
Oblique Supersegment, SWIR									
Van7-85	Peridotite, dunite, diabase	90	52.25 52.27	15.23 15.23	4190 3547	14.22	32.5°	11.99	S wall of axial trough, east of Narrowgate
Van7-86	Dunite, peridotite	146	52.14 52.13	15.16 15.15	3765 3128	14.21	32.6°	11.97	N wall of axial trough, east of Narrowgate
Van7-96	Cataclastites, perid, pyrox, diabase	81	53.14 53.15	9.98 9.97	2970 3527	14.06	28.5°	12.36	SW wall of axial trough at Shaka FZ intersection
PS86-6	Peridotite, basalt, gabbro veined perid	187	52.44 52.33	13.13 13.15	4509 3073	14.14	51.4°	8.82	N wall of axial trough, midway along segment
Atlantis II Fracture Zone, SWIR									
RC27-9-6	Peridotite, dunite	37	31.92 31.93	57.18 57.18	4010 3500	14.00 ^b	0.0°	14.00	E wall of transform, at ridge intersection

^a The effective spreading rate (ESR) is calculated from the full spreading rate (FSR) and angle of obliquity, following the method of Abelson and Agnon (1997).

^b Due to asymmetrical spreading, the half spreading rate is 8.5 mm/yr to the south (Hosford et al., 2003).

Table 4.2: Peridotite modal compositions

Sample	Lithology	Oliv	Opx	Cpx	Spin	Plag	Sum	Points
Van7-85-27	Harzburgite	68.6	27.5	3.1	0.9	0.0	100.0	2506
Van7-85-30	Harzburgite	67.4	29.2	2.1	1.2	0.2	100.0	2483
Van7-85-42	Lherzolite	67.1	25.6	6.2	1.0	0.0	100.0	2825
Van7-85-47	Lherzolite	68.7	24.7	5.1	1.5	0.0	100.0	2594
Van7-85-49	Lherzolite	71.6	22.2	5.3	0.9	0.1	100.0	2607
Van7-86-27	Lherzolite	55.8	34.0	9.6	0.6	0.0	100.0	2566
Van7-96-09V	Pyx Vein	9.3	25.8	48.5	1.9	14.6	100.0	756
Van7-96-16V	Pyx Vein	11.6	34.0	24.4	1.6	28.4	100.0	697
Van7-96-19M	Lherz w/ Pyx Vein	64.8	17.9	13.3	0.8	3.1	100.0	1883
Van7-96-19V	Pyx Vein in Lherz	7.2	26.4	57.2	2.8	6.3	100.0	318
Van7-96-21M	Harz w/ Pyx Vein	76.2	18.1	4.6	0.6	0.4	100.0	1894
Van7-96-21V	Pyx Vein in Harz	19.7	39.3	37.5	2.2	1.3	100.0	461
Van7-96-25	Lherzolite	66.2	25.6	6.8	1.4	0.0	100.0	2215
Van7-96-28	Lherzolite	76.7	17.2	5.4	0.7	0.0	100.0	2738
Van7-96-35	Lherzolite	70.4	16.7	10.5	2.3	0.1	100.0	1765
Van7-96-38	Lherzolite	64.2	24.1	9.9	1.6	0.2	100.0	2278
PS86-6-38	Lherzolite	56.7	29.0	12.6	1.7	0.0	100.0	3124
RC27-9-6-2M	Lherz w/ Cpx Vein	67.5	19.5	11.9	0.9	0.2	100.0	2194
RC27-9-6-2V	Cpx Vein in Lherz	7.6	9.2	82.6	0.7	0.0	100.0	436

Table 4.3: Trace elements Cont.

Cruise Dredge Sample Phase	Van7		Van7		Van7		Van7		Van7		Van7		Van7		Van7		Van7		PS86		RC27-9		RC27-9		RC27-9	
	96	19V	96	21M	96	21V	96	25	96	28	96	28	96	35	96	38	96	68	6	38	6	2M	6	2M	6	2V
	Cpx	Cpx	Cpx	Opx	Cpx	Cpx	Cpx	Cpx	Cpx	Cpx	Cpx	Cpx	Cpx	Cpx	Cpx	Cpx	Cpx	Basalt	Cpx	Cpx	Cpx	Cpx	Cpx	Cpx	Cpx	Cpx
Ba	1.12																									
Pb																										
Sr	7.94		0.005	0.002	0.005	0.005	9.38	0.016	0.016	74.86	0.016	0.003	0.003	32.44	10.33	0.004	105.21	1.073	5.87	68.81	0.012	2.51	0.029	0.029	0.208	16.77
La	0.417		0.851	0.019	0.984	0.170	0.170	1.026	0.020	1.026	0.020	0.020	0.020	0.433	0.064	0.023	248.91	11.77	0.025	0.668	0.029	0.029	0.029	0.029	0.208	0.208
Ce	2.693		4.196	0.102	4.486	1.380	1.380	3.292	0.057	3.292	0.057	0.057	0.057	2.184	0.882	0.018	28.40	28.40	0.508	2.503	0.096	0.096	0.096	0.096	0.911	0.911
Pr	0.643		0.878	0.024	0.901	0.351	0.351	0.596	0.009	0.596	0.009	0.009	0.009	0.508	0.327	0.007	3.77	3.77	0.255	0.411	0.014	0.014	0.014	0.014	0.212	0.212
Nd	4.29		5.27	0.17	5.47	2.52	2.52	3.37	0.09	3.37	0.09	0.09	0.09	3.32	2.49	0.07	16.63	16.63	2.30	1.94	0.09	0.09	0.09	0.09	1.26	1.26
Sm	1.81		1.87	0.09	1.87	1.13	1.13	1.19	0.05	1.19	0.05	0.05	0.05	1.34	1.34	0.05	4.32	4.32	1.30	0.52	0.03	0.03	0.03	0.03	0.65	0.65
Eu	0.67		0.72	0.04	0.70	0.51	0.51	0.47	0.02	0.47	0.02	0.02	0.02	0.54	0.59	0.03	1.60	1.60	0.57	0.21	0.02	0.02	0.02	0.02	0.28	0.28
Gd	2.75		2.37	0.16	2.37	1.86	1.86	1.57	0.10	1.57	0.10	0.10	0.10	1.82	2.18	0.13	4.86	4.86	2.17	0.72	0.06	0.06	0.06	0.06	1.07	1.07
Th	0.0013			0.0003	0.0057			0.0177	0.0008	0.0177	0.0008	0.0008	0.0008	0.0183		0.03	1.42	1.42	0.44	0.14	0.0051	0.0051	0.0051	0.0051	0.0023	0.0023
Tb	0.52		0.45	0.04	0.44	0.37	0.37	0.29	0.02	0.29	0.02	0.02	0.02	0.34	0.45	0.03	0.83	0.83	0.44	0.14	0.01	0.01	0.01	0.01	0.22	0.22
Dy	3.76		2.88	0.31	2.85	2.57	2.57	1.83	0.18	1.83	0.18	0.18	0.18	2.26	3.17	0.31	5.20	5.20	3.21	0.94	0.14	0.14	0.14	0.14	1.55	1.55
Ho	0.79		0.64	0.08	0.63	0.57	0.57	0.39	0.05	0.39	0.05	0.05	0.05	0.46	0.72	0.08	1.10	1.10	0.70	0.20	0.04	0.04	0.04	0.04	0.33	0.33
Y	21.55		18.15	2.35	17.85	15.40	15.40	11.36	1.41	11.36	1.41	1.41	1.41	12.36	20.68	2.57	31.54	31.54	19.84	5.49	1.08	1.08	1.08	1.08	9.04	9.04
Er	2.04		1.70	0.27	1.70	1.50	1.50	1.04	0.15	1.04	0.15	0.15	0.15	1.18	1.92	0.28	2.92	2.92	1.88	0.55	0.11	0.11	0.11	0.11	0.87	0.87
U	0.0005			0.0001				0.0046	0.0005	0.0046	0.0005	0.0005	0.0005	0.0064		0.05	0.42	0.42	0.28	0.0026	0.0026	0.0026	0.0026	0.0010	0.0010	
Tm	0.31		0.25	0.05	0.25	0.23	0.23	0.16	0.03	0.16	0.03	0.03	0.03	0.16	0.29	0.05	0.44	0.44	0.28	0.09	0.02	0.02	0.02	0.02	0.13	0.13
Yb	1.87		1.61	0.39	1.63	1.46	1.46	1.04	0.23	1.04	0.23	0.23	0.23	1.00	1.87	0.41	3.01	3.01	1.75	0.54	0.18	0.18	0.18	0.18	0.80	0.80
Lu	0.26		0.23	0.07	0.22	0.21	0.21	0.15	0.04	0.15	0.04	0.04	0.04	0.14	0.26	0.07	0.43	0.43	0.26	0.08	0.03	0.03	0.03	0.03	0.11	0.11
Li	2.11		1.70	0.94	1.52	2.64	2.64	1.47	0.81	1.47	0.81	0.81	0.81	1.88	4.36	1.64	6.14	6.14	2.75	14.32	6.19	6.19	6.19	6.19	12.80	12.80
Zr	35.31		<i>n.m.</i>	4.28	<i>n.m.</i>	<i>n.m.</i>	<i>n.m.</i>	<i>n.m.</i>	1.65	<i>n.m.</i>	1.65	1.65	1.65	<i>n.m.</i>	14.28	2.01	<i>n.m.</i>	<i>n.m.</i>	7.93	3.64	0.87	0.87	0.87	0.87	4.53	4.53
Hf	1.14		<i>n.m.</i>	0.16	<i>n.m.</i>	<i>n.m.</i>	<i>n.m.</i>	<i>n.m.</i>	0.06	<i>n.m.</i>	<i>n.m.</i>	0.06	0.06	<i>n.m.</i>	0.64	0.12	<i>n.m.</i>	<i>n.m.</i>	0.46	0.14	0.03	0.03	0.03	0.03	0.22	0.22
Nb																										
Be	0.014		<i>n.m.</i>	0.001	<i>n.m.</i>	<i>n.m.</i>	<i>n.m.</i>	<i>n.m.</i>	0.013	<i>n.m.</i>	<i>n.m.</i>	0.013	0.013	0.020	0.024		0.86	0.86	0.002	0.028	0.014	0.014	0.014	0.014	0.037	0.037

Table 4.4: Nd and Sr isotopic compositions of peridotite clinopyroxenes and orthopyroxenes

Sample ^a	Phase	Wt (mg)	$^{143}\text{Nd}/^{144}\text{Nd}^{b,c}$	Sm (ppm) ^e	Nd (ppm) ^e	$^{87}\text{Sr}/^{86}\text{Sr}^{b,c}$	Rb (ppm) ^e	Sr (ppm) ^e
Van7-85-27	Cpx	41.47	0.513297(25)	0.4651(5)	0.4199(1)	0.702855(25)	0.00652(1)	1.0098(1)
Van7-85-30	Cpx	15.41	0.513133(100)	0.4069(7)	0.2490(1)	0.704301(27)	0.00120(1)	0.2585(1)
Van7-85-42	Cpx	190.99	0.513335(27)	0.3522(8)	0.2065(1)		0.00113(1)	
Van7-85-47	Cpx	149.60	0.513322(27)	0.3363(4)	0.2128(1)	0.702286(22)	0.00151(1)	0.1558(1)
Van7-85-47R ^d	Cpx	139.74	0.513330(24)	0.3849(2)	0.2482(1)	0.702176(25)	0.00029(1)	0.1542(1)
Van7-85-49	Cpx	43.30	0.513249(25)	0.3387(4)	0.2247(1)	0.702844(24)	0.00050(1)	0.4971(1)
Van7-86-27	Cpx	105.32	0.513060(31)	0.2983(2)	0.1687(1)	0.704431(24)	0.02403(1)	1.2541(1)
Van7-96-09V	Cpx	19.65	0.513050(24)	1.4008(12)	2.8771(2)	0.703421(24)	0.00065(1)	5.8877(1)
Van7-96-09V	Opx	204.86	0.513061(29)	0.0696(1)	0.0899(1)	0.703631(25)	0.00028(1)	0.0787(1)
Van7-96-16V	Cpx	29.44	0.512844(24)	1.581(1)	2.8256(2)	0.703432(24)	0.01558(2)	2.2144(1)
Van7-96-16V	CpxII ^e	37.25	0.512798(24)	0.2590(6)	0.5812(1)	0.703282(24)	0.02724(4)	1.6223(1)
Van7-96-19M	Cpx	9.89	0.513018(26)	2.635(2)	5.801(1)	0.703052(14)	0.00722(2)	7.5357(2)
Van7-96-19M	Opx	79.30	0.512982(27)	0.1139(2)	0.1854(1)	0.703151(15)	0.01372(1)	0.8226(1)
Van7-96-19V	Cpx	9.49	0.512993(27)	2.723(6)	6.349(2)	0.703097(14)	0.03113(6)	10.5837(3)
Van7-96-21M	Cpx	241.45	0.513069(22)	1.978(5)	5.4349(2)	0.703709(24)	0.00033(1)	23.8869(6)
Van7-96-21M	Opx	100.80	0.513096(25)	0.086(3)	0.1653(1)	0.703701(24)	0.00009(1)	0.5021(1)
Van7-96-21V	Cpx	66.42	0.513107(23)	2.007(2)	5.7292(2)	0.703715(25)	0.00230(1)	20.1384(4)
Van7-96-25	Cpx	12.89	0.513184(27)	1.555(1)	3.2773(7)	0.702480(13)	0.00052(1)	12.3843(3)
Van7-96-28	Cpx	191.22	0.512938(23)	1.224(1)	3.4344(2)	0.703061(24)	0.00025(1)	82.843(3)
Van7-96-28	Opx	137.78	0.513053(34)	0.0459(1)	0.0845(1)	0.703046(24)	0.00008(1)	1.6014(1)
Van7-96-35	Cpx	10.15	0.512936(27)	1.831(2)	4.4269(9)	0.703732(15)	0.00110(1)	43.607(2)
Van7-96-38	Cpx	91.37	0.513205(22)	1.411(1)	2.6592(1)	0.702096(24)	0.00567(1)	10.8928(2)
Van7-96-38	Opx	224.64	0.513098(29)	0.0578(1)	0.0715(1)	0.702102(24)	0.00232(1)	0.2634(1)
Van7-96-68	Basalt	21.85	0.513015(22)	4.709(5)	18.1274(8)	0.702735(24)	9.89(2)	261.022(9)
PS86-6-38	Cpx	11.91	0.513032(27)	1.319(2)	2.2341(1)	0.702593(13)		5.6929(1)
PS86-6-38S ^f	Cpx		0.513044(10)			0.702598(13)		
RC27-9-6-2M	Cpx	19.53	0.512915(27)	0.6340(8)	2.3170(6)	0.704993(13)	0.00429(1)	74.815(1)
RC27-9-6-2M	Opx	170.89	0.512937(43)	0.0375(1)	0.1053(1)	0.704987(14)	0.00194(1)	2.5309(1)
RC27-9-6-2V	Cpx	21.61	0.513099(27)	0.685(2)	1.3526(2)	0.704447(14)	0.00176(1)	16.9512(5)

^a M indicates the peridotite matrix of a veined sample and V indicates a pyroxenite vein.

^b Data are fractionation corrected to $^{146}\text{Nd}/^{144}\text{Nd} = 0.7219$ and $^{86}\text{Sr}/^{88}\text{Sr} = 0.1194$ and reported relative to La Jolla $^{143}\text{Nd}/^{144}\text{Nd} = 0.511857$ and NBS987 $^{87}\text{Sr}/^{86}\text{Sr} = 0.710271$.

^c Number in parenthesis indicates the 2σ error in the last digits.

^d Repeat analysis.

^e Less pure Cpx separate: non-optically pure grains contain small black inclusions and white alteration material.

^f Analysis by Salters and Dick (2002), renormalized to above standard values and included for interlab comparison.

Table 4.5: Pb isotopic compositions of peridotite Cpx and basalt glasses

Sample	Phase	$^{206}\text{Pb}/^{204}\text{Pb}^a$	2σ	$^{207}\text{Pb}/^{204}\text{Pb}^a$	2σ	$^{208}\text{Pb}/^{204}\text{Pb}^a$	2σ
Van7-96-21M	Cpx	19.032	0.015	15.537	0.012	38.423	0.025
Van7-96-28	Cpx	19.598	0.016	15.570	0.012	39.096	0.026
RC27-9-6-2M	Cpx	17.221	0.002	15.446	0.002	37.201	0.005
Van7-96-68	Basalt	18.789	0.015	15.493	0.012	38.426	0.025
Van7-92-03	Basalt	19.068	0.002	15.582	0.002	38.866	0.006
Van7-92-03S ^b	Basalt	19.065	0.002	15.601	0.001	38.893	0.004
Kn162-61-71	Basalt	19.270	0.002	15.617	0.002	39.297	0.006
Kn162-61-71S ^b	Basalt	19.288	0.001	15.642	0.001	39.368	0.002

^a All data are normalized to the NBS981 values from Todt et al. (1996): $^{206}\text{Pb}/^{204}\text{Pb}=16.936$, $^{207}\text{Pb}/^{204}\text{Pb}=15.489$ and $^{208}\text{Pb}/^{204}\text{Pb}=36.701$.

^b Values from Standish (2006), for interlaboratory comparison.

Table 4.6: Cpx/Opx partition coefficients.

	Van7 85 42	Van7 85 47	Van7 85 49	Van7 96 09V	Van7 96 19M	Van7 96 21M	Van7 96 28	Van7 96 38	RC27-9 6 2V	Valence	Cpx IR ^a (Å)	Opx IR ^a (Å)
Pb			0.02			2.73	6.34			2	1.29	1.19
Sr			25.80	66.38	7.13	32.81	45.39	44.88	27.43	2	1.26	1.18
La				68.39	16.79	45.78	52.60		22.94	3	1.16	1.03
Ce				48.43	25.03	41.34	57.72	48.02	26.05	3	1.14	1.01
Pr				43.61	31.01	36.94	67.34	44.26	30.05	3	1.13	0.99
Nd	37.94	37.37		29.90	23.21	31.22	39.44	37.31	21.68	3	1.11	0.98
Sm	23.42	31.55	25.24	16.87	18.33	21.35	22.15	24.71	19.88	3	1.08	0.96
Eu	17.33	18.18	22.13	15.96	15.87	18.70	21.64	22.20	12.94	3	1.07	0.95
Gd	15.16	16.04	18.02	10.64	12.53	14.68	16.34	17.00	11.99	3	1.05	0.94
Th							23.24			4	1.05	0.94
Tb	10.66	10.70	13.53	7.73	10.27	11.73	14.19	13.59	9.54	3	1.04	0.92
Dy	8.53	9.19	10.98	5.44	8.75	9.32	10.00	10.27	6.95	3	1.03	0.91
Ho	6.88	7.33	9.05	4.55	7.09	8.04	8.63	8.78	5.63	3	1.02	0.90
Y	6.61	6.67	8.16	3.50	7.42	7.72	8.08	8.06	5.07	3	1.02	0.90
Er	5.71	5.52	7.12	3.57	6.04	6.28	6.82	6.88	4.92	3	1.00	0.89
U				2.17			10.03			4	1.00	0.89
Tm	4.41	4.24	5.41	2.68	4.87	5.01	5.14	5.46	3.71	3	0.99	0.88
Yb	3.69	3.47	4.43	2.30	4.12	4.14	4.63	4.57	3.07	3	0.99	0.87
Lu	2.97	2.97	3.69	1.82	3.39	3.20	3.68	3.69	2.66	3	0.98	0.86
Li	4.15	3.32	3.74	2.66	1.07	1.81	1.82	2.65	2.32	1	0.92	0.76
Zr		23.47						7.12	4.17	4	0.72	0.72
Hf	9.91	4.87	7.07					5.58	5.18	4	0.71	0.71
Nb	1.43	1.22							1.98	5	0.64	0.64

^a Ionic radii from Shannon (1976).

Table 4.A1: Olivine major element concentrations (wt %)

Sample	Pts	SiO ₂	TiO ₂	Al ₂ O ₃	Cr ₂ O ₃	FeO	MnO	MgO	CaO	NiO	Total	Mg#
Van7-85-27	5	40.06	0.01	0.01	0.00	8.87	0.16	49.23	0.05	0.33	98.72	90.8
Van7-85-30	4	40.17	0.01	0.00	0.01	9.28	0.15	48.78	0.04	0.34	98.80	90.4
Van7-85-42	6	40.03	0.01	0.00	0.01	9.57	0.13	49.36	0.04	0.35	99.50	90.2
Van7-85-47	6	40.34	0.00	0.00	0.01	9.76	0.16	49.48	0.04	0.33	100.12	90.0
Van7-85-49	6	40.49	0.01	0.00	0.00	9.57	0.15	48.91	0.04	0.40	99.57	90.1
Van7-86-27	6	40.26	0.00	0.00	0.00	9.91	0.14	49.00	0.01	0.35	99.68	89.8
Van7-96-09V	6	40.30	0.00	0.01	0.01	9.21	0.11	50.44	0.02	0.45	100.55	90.7
Van7-96-16V	5	40.18	0.02	0.01	0.01	10.89	0.22	47.75	0.03	0.37	99.50	88.7
Van7-96-19M	11	40.13	0.01	0.00	0.00	10.27	0.15	48.96	0.04	0.40	99.95	89.5
Van7-96-19V	6	40.15	0.00	0.00	0.00	10.51	0.14	48.47	0.02	0.38	99.67	89.2
Van7-96-21M	6	40.16	0.01	0.01	0.00	9.69	0.16	49.85	0.03	0.35	100.27	90.2
Van7-96-21V	6	40.17	0.02	0.01	0.00	9.76	0.16	50.03	0.02	0.34	100.51	90.1
Van7-96-25	6	40.09	0.00	0.01	0.00	10.10	0.12	48.87	0.06	0.39	99.64	89.6
Van7-96-28	6	40.42	0.00	0.01	0.00	9.18	0.13	49.33	0.03	0.36	99.47	90.5
Van7-96-35	6	40.62	0.00	0.01	0.00	8.66	0.11	50.34	0.03	0.43	100.19	91.2
Van7-96-38	6	40.41	0.01	0.02	0.01	9.80	0.15	49.81	0.01	0.35	100.55	90.1
PS86-6-38	6	40.50	0.01	0.01	0.00	9.95	0.11	48.88	0.02	0.41	99.89	89.8
RC27-9-6-2M	76	40.42	0.00	0.00	0.02	9.64	0.13	49.85	0.02	0.34	100.41	90.2
RC27-9-6-2V	20	40.30	0.00	0.00	0.01	9.67	0.14	49.40	0.03	0.33	99.88	90.1

Table 4.A2: Pyroxene major element concentrations (wt %)

Sample	Pts	SiO ₂	TiO ₂	Al ₂ O ₃	Cr ₂ O ₃	FeO	MnO	MgO	CaO	Na ₂ O	Total	Mg#	Cr#
Orthopyroxene													
Van7-85-27	28	55.02	0.09	3.99	0.66	5.74	0.11	31.81	1.80	0.04	100.74	90.8	10.0
Van7-85-30	17	55.42	0.06	3.91	0.51	6.11	0.12	32.11	1.56	0.02	101.33	90.4	8.0
Van7-85-42	30	55.08	0.08	5.20	0.72	6.23	0.12	31.50	2.03	0.03	102.45	90.0	8.5
Van7-85-47	21	54.76	0.10	5.24	0.76	6.18	0.14	31.01	2.55	0.05	107.02	89.9	8.9
Van7-85-49	16	54.59	0.04	4.48	0.64	6.17	0.10	31.99	1.72	0.01	101.91	90.2	8.8
Van7-86-27	17	54.85	0.08	4.61	0.69	6.16	0.13	31.64	1.88	0.05	106.68	90.1	9.1
Van7-96-09V	18	54.17	0.15	4.69	0.28	6.49	0.18	31.11	1.69	0.02	102.32	89.5	3.8
Van7-96-16V	21	54.45	0.23	5.03	0.57	6.94	0.15	31.79	1.28	0.04	101.45	89.1	7.0
Van7-96-19M	19	54.31	0.21	4.18		6.49	0.11	32.88	1.40	0.00	101.37	90.0	
Van7-96-19V	9	55.42	0.21	2.74		6.95	0.11	32.92	1.32	0.00	101.06	89.4	
Van7-96-21M	27	53.92	0.15	6.02	0.62	6.42	0.14	31.30	1.14	0.04	101.16	89.7	6.4
Van7-96-21V	35	55.09	0.19	3.97	0.54	6.25	0.14	31.99	1.25	0.03	101.89	90.1	8.3
Van7-96-25	17	53.46	0.11	5.58		6.68	0.11	31.50	1.48	0.04	100.11	89.4	
Van7-96-28	14	53.75	0.14	5.38		5.99	0.10	31.38	2.44	0.04	106.34	90.3	
Van7-96-35	20	55.21	0.10	3.91	0.92	5.55	0.12	32.02	1.59	0.08	100.45	91.1	13.7
Van7-96-38	44	53.64	0.14	6.36	0.57	6.45	0.13	30.49	1.36	0.06	100.61	89.4	5.6
PS86-6-38	16	53.78	0.14	6.12	0.61	6.33	0.15	30.71	1.76	0.08	102.01	89.6	6.3
RC27-9-6-2M	79	55.67	0.08	3.16	0.76	6.30	0.11	32.75	1.48	0.02	100.33	90.3	13.9
RC27-9-6-2V	10	56.10	0.09	2.29	0.62	6.54	0.18	33.38	0.81	0.02	100.03	90.1	15.4
Clinopyroxene													
Van7-85-27	33	51.64	0.23	5.38	1.22	2.83	0.09	16.63	20.84	0.54	99.40	91.3	13.2
Van7-85-30	54	51.01	0.22	6.22	1.09	3.01	0.09	16.30	21.58	0.37	99.89	90.6	10.5
Van7-85-42	31	51.65	0.22	6.28	1.08	3.27	0.10	17.06	21.09	0.36	101.11	90.3	10.4
Van7-85-47	46	51.54	0.22	6.31	1.07	3.21	0.10	17.08	20.88	0.37	100.78	90.5	10.2
Van7-85-49	29	51.34	0.17	6.08	1.08	3.23	0.07	17.48	20.27	0.29	100.02	90.6	10.7
Van7-86-27	62	51.64	0.24	5.80	1.03	2.98	0.09	16.84	21.59	0.29	100.51	91.0	10.6
Van7-96-09V	72	50.51	0.44	6.75	0.60	3.41	0.12	16.46	20.51	0.51	99.33	89.6	5.6
Van7-96-16V	90	51.03	0.74	6.50	0.51	3.43	0.13	16.44	21.22	0.46	100.45	89.5	5.0
Van7-96-19M	66	50.84	0.63	6.26	1.34	2.92	0.08	16.52	21.40	0.62	100.61	91.0	12.5
Van7-96-19V	63	51.16	0.58	6.02	1.26	3.02	0.08	16.11	21.35	0.56	100.14	90.5	12.3
Van7-96-21M	39	50.37	0.55	7.58	0.97	3.19	0.09	15.74	20.96	0.73	100.19	89.8	7.9
Van7-96-21V	27	50.70	0.49	7.70	0.96	3.57	0.10	16.77	19.01	0.84	100.16	89.3	7.8
Van7-96-25	53	51.08	0.36	7.06	1.05	3.35	0.07	14.79	20.18	1.31	99.24	88.7	9.1
Van7-96-28	55	51.22	0.35	6.75	1.32	3.21	0.07	15.97	19.96	1.05	99.89	89.9	11.6
Van7-96-35	55	52.17	0.32	5.33	1.69	2.58	0.08	16.02	20.27	1.38	99.85	91.7	17.6
Van7-96-38	44	51.05	0.45	7.83	0.87	3.56	0.09	15.98	18.68	1.20	99.72	88.9	7.0
PS86-6-38	33	51.20	0.41	7.23	0.93	3.12	0.12	15.37	20.96	1.09	100.43	89.8	7.9
RC27-9-6-2M	197	52.27	0.17	4.76	1.50	3.13	0.09	18.21	19.70	0.63	100.46	91.2	17.5
RC27-9-6-2V	113	51.98	0.23	5.08	1.59	3.28	0.11	17.96	19.66	0.61	100.50	90.7	17.3

Table 4.A3: Spinel major element concentrations (wt %)

Sample	Pts	TiO ₂	Al ₂ O ₃	Cr ₂ O ₃	Fe ₂ O ₃	FeO	MnO	MgO	CaO	NiO	ZnO	Total	Mg#	Cr#
Van7-85-27	6	0.01	47.18	21.50	0.51	12.28	0.10	17.44	0.00	0.18	0.04	99.23	71.7	23.4
Van7-85-30	6	0.06	53.52	14.56	0.43	12.18	0.08	18.16	0.01	0.27	0.17	99.44	72.6	15.4
Van7-85-42	6	0.06	53.70	15.15	0.72	11.69	0.11	18.70	0.00	0.30	0.21	100.65	74.0	15.9
Van7-85-47	6	0.07	52.47	16.48	0.94	11.69	0.09	18.65	0.01	0.29	0.18	100.86	74.0	17.4
Van7-85-49	6	0.03	54.14	14.19	0.63	11.49	0.12	18.61	0.00	0.37	0.16	99.76	74.3	14.9
Van7-86-27	6	0.00	51.80	16.87	0.00	12.69	0.08	17.75	0.00	0.12	0.00	99.30	71.4	17.9
Van7-96-09V	6	0.37	36.73	28.76	4.02	13.62	0.14	15.76	0.02	0.21	0.21	99.87	67.3	34.4
Van7-96-16V	6	0.21	58.19	9.73	0.78	11.04	0.16	19.50	0.00	0.43	0.12	100.17	75.9	10.1
Van7-96-19M	5	0.31	39.70	27.66	2.01	13.95	0.20	15.84	0.00	0.24	0.01	99.99	66.9	31.9
Van7-96-19V	6	0.53	37.78	29.47	1.84	15.43	0.22	14.90	0.00	0.22	0.02	100.46	63.2	34.4
Van7-96-21M	5	0.35	43.08	23.20	2.40	13.08	0.16	16.62	0.02	0.25	0.33	99.52	69.4	26.5
Van7-96-21V	12	0.29	51.40	15.26	2.45	10.79	0.13	19.01	0.01	0.31	0.21	99.88	75.8	16.6
Van7-96-25	5	0.07	53.35	12.68	2.81	12.28	0.14	18.13	0.00	0.38	0.00	99.89	72.5	13.8
Van7-96-28	5	0.25	45.92	21.56	1.37	13.15	0.16	16.97	0.01	0.25	0.02	99.76	69.7	24.0
Van7-96-35	5	0.18	40.15	28.60	1.56	12.51	0.18	16.82	0.00	0.16	0.00	100.16	70.5	32.3
Van7-96-38	6	0.28	56.04	10.85	1.58	11.20	0.08	19.20	0.01	0.33	0.18	99.74	75.3	11.5
PS86-6-38	3	0.05	57.31	10.84	0.58	11.15	0.10	19.23	0.00	0.33	0.00	99.58	75.4	11.3
RC27-9-6-2M	50	0.20	38.46	28.82	3.63	13.55	0.27	15.99	0.07	0.31		101.31	67.8	33.5
RC27-9-6-2V	16	0.12	37.04	30.09	3.30	14.02	0.21	15.63	0.03	0.20		100.73	66.5	35.3

Table 4.A4: Plagioclase major element concentrations (wt %)

Sample	Pts	SiO ₂	Al ₂ O ₃	FeO	MgO	CaO	Na ₂ O	K ₂ O	Total	An
Van7-96-09V	11	47.95	33.34	0.12	0.01	15.78	2.28	0.00	99.48	79.2
Van7-96-16V	12	46.40	34.88	0.14	0.01	17.71	1.33	0.01	100.47	88.0
Van7-96-19M	5	51.23	30.37	0.12	0.37	12.87	4.24	0.00	99.20	62.6
Van7-96-19V	5	51.86	30.27	0.11	0.00	12.53	4.57	0.01	99.34	60.2
Van7-96-21V	15	51.53	31.40	0.10	0.01	13.24	3.77	0.00	100.06	66.0
RC27-9-6-2M	2	54.15	29.79	0.14	0.03	11.87	5.03	0.09	101.10	56.3
RC27-9-6-2V	6	49.8	32.60	0.12	0.05	15.24	3.00	0.00	100.81	73.7

Chapter 5

Variations in Abyssal Peridotite Composition: Implications for Oceanic Upper Mantle Composition and Processes

Abstract

Abyssal peridotites are typically assumed to be the residues of near-fractional melting of a uniform initial source composition. We present new data on 67 veined and unveined peridotites to assess the extent to which abyssal peridotite compositions conform to this model. We find that peridotite Cpx trace element concentrations vary by 2-4 orders of magnitude at length-scales ranging from the grain-scale to the ridge segment-scale. We assess estimates of mantle degree of melting based on Cpx trace elements and spinel Cr#. We demonstrate that variations in predicted degree of melting occur on too small a scale to be the result of variation in mantle thermal structure. Instead, we find that peridotite compositions reflect melt-rock reaction and a heterogeneous initial source composition. Melt-rock reaction during peridotite melting is demonstrated to be an important process during the creation of oceanic lithosphere. Via subduction, this provides a mechanism for the creation of a heterogeneous source mantle. The magnitude of chemical variations in abyssal peridotites is found to be scale independent.

5.1 Introduction

The composition of the upper mantle is commonly assumed to be homogeneous when using MORB (Mid-Ocean Ridge Basalt) compositions to determine the degree of melting and variation in mantle potential temperature of a ridge segment. This composition is represented by DMM (Depleted MORB Mantle), a model average composition, most recently derived from the average global isotopic composition of MORB and the depletion trend among abyssal peridotites (Workman and Hart, 2005). This composition produces a good fit to parental normal MORB composition – as predicted from global MORB compositions (Su and Langmuir, 2003) – by 6% aggregated fractional melting. In this chapter, we use abyssal peridotites to assess the extent to which mantle composition deviates from the predicted average composition and we assess the role of melt-rock reaction in modifying mantle composition.

Abyssal peridotites are samples of the oceanic upper mantle and are thought to be the depleted residues of MORB formation. The pioneering work by Johnson et al. (1990) concluded that trace element variations observed in abyssal peridotite Cpx (clinopyroxene) were produced by near-fractional melting of a uniform initial upper mantle composition. This study has become a benchmark for our understanding of melting processes beneath ocean ridges. Two important assumptions were made by Johnson et al. (1990): (1) the mantle has a uniform initial composition and (2) all chemical variations are due to melting. Subsequent studies have attempted to use abyssal peridotites to constrain degrees and/or depths of melting (e.g., Johnson et al., 1990; Bonatti et al., 1993; Hellebrand et al., 2001), based on these assumptions. However, some studies of MORBs and abyssal peridotites have questioned these assumptions. For example, Elthon (1992) argued that major element variations in abyssal peridotites are better fit by a model of mixing between depleted peridotites and basaltic melts, rather than variable degrees of melt extraction. Niu et al. (1997) concluded that the FeO-MgO variations in abyssal peridotites show strong effects of olivine

addition, attributable to precipitation of olivine as a result of melt-rock reaction during melt migration.

Large isotopic heterogeneities have been observed among basalts from individual dredges and individual ridge segments (e.g., Dupré and Allègre, 1983; Hamelin et al., 1985; Dosso et al., 1999), suggesting the existence of long-term chemical (parent/daughter ratio) heterogeneities in the MORB source upper mantle. In Chapter 4, we documented isotopic heterogeneities among peridotites from the Southwest Indian Ridge (SWIR) and argued that the upper mantle beneath the SWIR was heterogeneously depleted prior to upwelling at the ridge. Evidence for large chemical heterogeneities in the form of pre-melting depletions have also been suggested for the easternmost end of the SWIR. In this region, ultra-slow spreading rates and thin crust imply a low degree of melting (Cannat et al., 1999; Sauter et al., 2004), yet basalt compositions indicate a depleted harzburgitic source (Meyzen et al., 2003) and spatially associated peridotites are harzburgites (Seyler et al., 2004).

Available data for abyssal peridotites obtained from all major mid-ocean ridges (Fig. 5.1) provide a global context for the evaluation of geochemical variations. Fig. 5.2 illustrates the variations of several peridotite indicators of melt extraction, as a function of spreading rate. Each panel has model reference trends for residues of melt extraction from DMM (Workman and Hart, 2005), based on the predicted maximum degree of melting as a function of spreading rate, including the effect of conductive cooling on ridge thermal structure following the method of Bown and White (1994). The variation in Yb was calculated using the non-modal fractional melting formulation (Johnson et al., 1990) with melting reactions from Kinzler (1997) and mineral/melt partition coefficients compiled by Kelemen et al. (2003). Variation in modal Cpx was calculated using the lever-rule to predict changes in phase proportions during melting. The variation of spinel Cr# with degree of melting was predicted from the empirical correlation of spinel Cr# with heavy rare earth elements (HREE) in Cpx (Hellebrand et al., 2001).

Two features are evident from Fig. 5.2: first, the amount of variation is greater at slow and ultra-slow spreading rates than at fast spreading rates. Second, the most depleted peridotite compositions – indicated by low modal Cpx, low Yb_N in Cpx and high spinel Cr# – occur at all ridges, independent of spreading rate. These features are contrary to predictions derived from prevalent views of upper mantle composition and melting processes beneath mid-ocean ridges. Instead, these features suggest that (1) parts of the upper mantle have previously been depleted, as discussed in Chapter 4, and (2) large chemical variations observed at slow to ultra-slow spreading ridges involve melt-rock interaction mechanisms which have not yet been considered in detail in the context of abyssal peridotites. A comprehensive abyssal peridotite trace element dataset at various sampling length-scales needs to be established, so that processes operating beneath ridges can be more critically examined and mechanisms responsible for creating chemical heterogeneities in abyssal peridotites can be quantitatively understood.

This chapter presents ion probe Cpx trace element abundance data obtained from 67 veined and unveined peridotite samples from a ridge segment and a fracture zone on the SWIR. This dataset includes a detailed evaluation of sample-scale and dredge-scale variations in peridotite compositions, to establish a database for evaluating the magnitude of local-scale chemical variations. We emphasize the importance of any chemical variations at length-scales less than individual dredges, as peridotites at these length-scales would have experienced the same temperature regime during mantle upwelling and theoretically should have undergone the same degree of melting. Any chemical variations at the local-scale must be due to either initial (pre-melting) heterogeneities or melt-rock reaction at the ridge. The data from this study, combined with the global peridotite dataset, are used to assess the influence of melt-rock reaction and initial chemical heterogeneities of the upper mantle on the composition of abyssal peridotites exposed at mid-ocean ridges.

5.2 Methods

The SWIR Oblique and Orthogonal Segments (9° – 25° E) and the Atlantis II Fracture Zone are the focus areas for this study. Both areas have been extensively sampled, as shown in Fig. 5.3, and peridotites with a range of characteristics were recovered. For the 9° – 25° E region, 5 Oblique Segment dredges, 2 Orthogonal Segment dredges, and 1 dredge from the Shaka Fracture Zone were analyzed. From Atlantis II Fracture Zone, 1 dredge and 1 dive were analyzed from the ridge-transform intersection (RTI) and 1 dredge and 5 dives from Atlantis Bank. Both locations on Atlantis II are uplifted massifs of lower crust and upper mantle, often termed oceanic core complexes (Dick et al., 1991; Baines et al., 2003).

In total, 67 samples were analyzed, of which 47 are unveined peridotites, 11 are veined peridotites, 2 are veined dunites and 7 are various pyroxenites varieties. Dredge and dive locations, sample descriptions and modal mineralogy are presented in Tables 5.2 and 5.3. Modes were determined by point counting on a 1 mm grid, with a minimum of 1700 grains, except veins, for which sample material was limited.

Major element compositions of all primary phases were determined *in situ* on polished thin sections by wavelength dispersive spectroscopy using the JEOL JXA-733 Superprobes at the Massachusetts Institute of Technology (Tables 5.4-5.9). The electron microprobes were operated at 15kV accelerating potential with a 10nA beam current and calibrated using a standard set of silicate minerals. Olivine, spinel, plagioclase and additional minor phases were analyzed using a focussed beam, with 6 points analyzed per grain. For pyroxenes, all porphyroclasts in abyssal peridotites exhibit exsolution of a second pyroxene phase, due to the relatively slow exhumation rate of mantle upwelling that results in phase re-equilibration during cooling. To average out the effects of exsolution and thus recover the high temperature mineral composition, pyroxenes were analyzed using a beam defocused to a spot size of 10 μ m and line transects of ≥ 10 points with 10 μ m steps. Data were reduced using modified matrix correction factors (Bence and Albee, 1968; Albee and

Ray, 1970). Analyses were discarded if their totals were <98.5 wt% or >101.5 wt%, or if their stoichiometry varied by $>1.5\%$. Elements occurring at >10 wt% levels have reproducibilities within 2%, elements at <10 wt% levels have reproducibilities within 5% and elements at <1 wt% level have reproducibilities within 30%, based on repeat analyses of a basaltic glass by Gaetani and Grove (1998).

Abundances of Na, Ti, Cr, Sr, Y, Zr (referred to collectively as TE) and REE (La, Ce, Nd, Sm, Eu, Dy, Er, Yb) were determined using the Cameca IMS-3f at the Woods Hole Oceanographic Institution Northeast National Ion Microprobe Facility (Tables 5.10-5.13). Samples were ionized using a primary beam of O^- with an accelerating voltage of -8.30 kV and a current of 6-7 nA. The beam was focussed to a ~ 10 μm spot diameter for REEs and a ~ 4 μm spot for TE. The energy filtering technique of Shimizu et al. (1978) was used to suppress molecular ion interferences, with a high voltage offset of 60 V for REE and 90 V for TE.

Working curves, which relate secondary ion intensity to concentration, were determined using Cpx from KH1 (Kilbourne Hole peridotite, Irving and Frey, 1984) as the REE standard and basalt glass KL2G (Kilauea tholeiite, Jochum et al., 2000) as the TE standard. Standards were typically measured 2 or more times at the start of a session and occasionally at the middle or end of a session. The average value of all standard analyses for a session were used to calculate session-specific working curves, with the deviation among repeat measurements providing a estimate of analytical reproducibility (Tables 5.14 and 5.15).

Data were collected in 5 count cycles with count times of 30 s for REE and 5-20 s for TE. For REE, ^{30}Si was measured at the start and end of each analysis, with the time interpolated average used to calculate REE/Si ratios, for determination of concentrations from the working curves. For trace elements, ^{28}Si was measured during each cycle, with the average value after all cycles used to determine TE concentrations. Traditionally, within-cycle

ratios of REE/Dy and TE/Si are used to account for temporal variability in peak intensities (e.g., Johnson et al., 1990; Johnson and Dick, 1992). Due to random background noise that produced occasional extreme high counts, the ratio scheme was abandoned. However, the error introduced by temporal variability in peak intensity is insignificant compared to primary beam instability, which cannot be monitored on the IMS-3f.

To remove the effects of random background noise, the 5-cycle raw count data were averaged by element. Low count rates among REEs are best treated statistically by a Poisson distribution, with the standard deviation given by $\sqrt{\bar{x}}$, where \bar{x} is the count average. Any individual REE count greater than $3\sqrt{\bar{x}}$ away from the *median* count value was discarded. After recalculating \bar{x} , any count more than $2\sqrt{\bar{x}}$ from the *average* count value was discarded. For TE, which have higher count rates, any count greater than 2 standard deviations from the *median* count value was discarded. After a final re-averaging of counts, REE and TE ratios to Si were calculated and concentrations derived from working curves.

5.3 Ion probe data quality

Trace element analyses in abyssal peridotites are difficult due to their low concentrations, particularly for REE. We assess data quality from (1) repeated standard measurements, (2) measurements of an additional standard, (3) sample count statistics, (4) sample duplicate analyses, (5) REE pattern shape, and (6) comparison to ICP-MS and TIMS bulk mineral separate analyses. In addition, we compare ion probe data to electron probe data for Na₂O, TiO₂ and Cr₂O₃. Overall, the smoothness of REE pattern shape and the orders of magnitude variation in trace element concentrations means that ion probe data are significant, despite their relatively large errors.

Variations in standard concentrations from repeat measurements provide an estimate of analytical reproducibility. In Tables 5.14 and 5.15, we calculate the average % deviation of standard measurements for each analytical session. The variation for REEs is 6-10% and

for TEs is 3-5%. We also measured REEs in an independent standard, KLB1 (Iherzolite, Takahashi, 1986), with a reproducibility of 9%-17% among 21 analyses over 15 months (Table 5.10).

Analytical error can also be estimated from the count statistics of each analysis and by comparison of duplicate analyses (Table 5.10). The statistical counting error for each analysis is the square root of the total counts for an element. The average % error in concentration based on the counting error ranges from 0.2-21% (Table 5.10), with LREE having the largest errors. The error based on the average difference between duplicate analyses is also given in Table 5.10 and shown on a comparison plot in Fig. 5.4. In general, samples plot on a 1:1 correspondence line over 3-4 orders of magnitude, with an average deviation for Cpx ranging from 7% to 37%. While errors are relatively large for REE, a final check on data quality is provided by the shape of the REE pattern, which should vary smoothly, with the exception of Eu due to its behavior as both a 2+ and 3+ cation. Analyses with irregular patterns that were not reproduced by repeat analysis were discarded and are not reported here.

In Fig. 5.5, ion probe data for 16 samples are compared to TIMS and ICP-MS data (from Chapter 4) for mineral separates. The ion probe data have been averaged by sample, or by vein and matrix occurrence, and are shown with a horizontal bar representing the range of measured values. In general, Cpx data plot on a 1:1 line, except at low concentrations for Sr, Zr and Ce/Yb. Opx data have more scatter, but only one ion probe analysis is available for each sample where Opx was measured as a mineral separate. An accurate comparison to bulk mineral separate data cannot be made, especially if – as suggested in Chapter 4 – some Opx are zoned. At low concentrations, the scatter in the comparison of ion probe *in situ* data to bulk mineral separate data probably indicates the minimum ion probe detection limit for these elements. The scatter is not the result of the repartitioning of elements following Cpx/Opx exsolution, as such a process would result in a systematic

offset of ion probe data from TIMS and ICP-MS data at all concentration levels. As we demonstrated in Chapter 4, Cpx have $\geq 90\%$ of the Cpx-Opx trace element budget and the effect of exsolution on trace element concentrations is relatively minor.

Elements measured by both ion probe and electron probe are compared in Fig. 5.6. Na_2O in abyssal peridotite phases has not previously been reported by ion probe. In this study, we used tholeiitic basalt KL2G as a standard (Jochum et al., 2000) and 5 s count times. In general, Na_2O , TiO_2 and Cr_2O_3 all show reasonable agreement between the two analytical methods. Only for plagioclase, where Na_2O is present at higher concentrations, do the data fall consistently off the 1:1 line towards higher electron probe concentrations. We suggest that because the KL2G standard has 2.27 wt% Na_2O , whereas Na_2O in plagioclase extends to 6 wt%, the ion probe data are unreliable at concentrations $\gtrsim 2$ wt%.

The electron probe analyses in Fig. 5.6 are the average of multiple points located near an individual ion probe analysis. The standard deviation of electron probe analyses, shown in Fig. 5.6, is considerably larger than the estimated error for ion probe analyses. In particular, Cpx have large standard deviations when measured by electron probe, due to the inclusion of exsolution lamellae in the analysis. In addition, long count times were not used during electron probe analyses, resulting in limited precision at the < 1 wt% level. Hence, we conclude that ion probe data provide a better estimate of Na, Ti and Cr concentrations, with the exception of Na in plagioclase.

5.4 Results

Mineral major element data for the peridotites are presented in Tables 5.4-5.9 and trace element data in Tables 5.10-5.13. Included in the tables are additional data for samples RC27-9-6-2 and RC27-9-6-5 from Lee (1997), with ion probe trace element data reprocessed following the above method. The major element variations among spinel and Cpx are presented in Figs. 5.7, 5.8 and 5.9. In Figs. 5.10 and 5.11, we plot the trace element

variations in individual Cpx by location. In Appendix 5.A, each individual REE analysis is plotted by sample.

The peridotites in this study can be classified into four general groups - “typical” residual peridotites, gabbro veined and plagioclase peridotites (abbreviated to gabbro/plag peridotites), pyroxenite veined peridotites and cryptically metasomatized peridotites (Table 5.3). Of the samples analyzed, 41 are classified as residual peridotites, due to the absence of veins and trace element compositions in the abyssal peridotite range of Johnson et al. (1990). Plagioclase, if present, does not account for $>.05\%$ of the modal mineralogy and is assumed to be the result of spinel plus two pyroxene breakdown to plagioclase plus olivine. The gabbro/plag peridotite group consists of 4 peridotites with gabbro veins, 1 dunite with a central gabbro vein and surrounding harzburgite, and 3 peridotites with significant amounts of matrix plagioclase. The group of 15 pyroxenite peridotites consists of 7 peridotites with pyroxenite veins, 1 peridotite with a clinopyroxenite vein, 6 pyroxenite veins and 1 orthopyroxenite vein.

The three cryptically metasomatized peridotites have no veins and look like other residual peridotites. However, they are classified as a separate group on the basis of 1-3 orders of magnitude variation in trace element concentrations at the thin section and grain scales (Fig. 5.12). These variations are in contrast to the generally uniform composition of Cpx cores, rims and interstitial (<1 mm diameter) grains in the residual peridotites, as shown in Fig. 5.14. The variations in trace element concentrations in the cryptic peridotites is not limited to interstitial grains, as shown in the photomicrograph of a zoned Cpx porphyroclast in Fig. 5.15.

Spinel compositions for the four peridotite lithologies are shown in Figs. 5.7 and 5.8. Spinel in Fig. 5.7 plot on the general melting trend of $Cr\# = Cr/(Cr+Al)$ increasing from $\sim 10\%$ to $\sim 60\%$ during melting, while $Mg\# = Mg/(Mg+Fe)$ decreases over a more limited range of $\sim 80\%$ - 60% . The decrease in spinel $Mg\#$ reflects re-equilibration of Mg^{2+} and

Fe^{2+} between olivine and spinel with decreasing temperature during upwelling (Dick and Bullen, 1984). Samples which plot off the trend towards lower Mg#’s, such as the gabbro/plag peridotites, have been refertilized by differentiated melts. TiO_2 is incompatible in spinel, so values > 0.2 wt% (the approximate composition of spinel in DMM; Workman and Hart, 2005) indicate melt refertilization of the peridotite (Dick and Bullen, 1984). In Fig. 5.8, gabbro/plag peridotites, dunites and some pyroxenites have high TiO_2 concentrations.

Cpx major element compositions have small variations among the different lithological groups (Fig. 5.9). Depleted peridotites and pyroxenite peridotites, with some exceptions, have $\sim 6\text{--}8$ wt% Al_2O_3 and 1 wt% Cr_2O_3 . The exceptions are dredge Van7-96 pyroxenites, which have similar Al_2O_3 but 0.4 wt% Cr_2O_3 , and the clinopyroxenite-veined sample RC27-9-6-2, which has 1.5 wt% Cr_2O_3 . These two pyroxenite lithologies, described in detail in Chapter 4, appear to be different from the majority of other abyssal pyroxenites. Gabbro/plag peridotites generally have similar Cr_2O_3 to residual peridotites, but Al_2O_3 extends to 2 wt%. Cpx in some gabbro veins are low in Cr_2O_3 (< 0.3 wt%). Finally, the cryptically metasomatized samples have similar Cr_2O_3 compositions to the residual peridotites but a relatively large range of 2.5–7 wt% Al_2O_3 .

Peridotites from the four different lithological groups have overlapping trace element compositions (Figs. 5.10 and 5.11). In Fig. 5.13, we plot histograms of Ce/Yb using the global dataset, to demonstrate the similar compositional range of all four peridotite lithologies. Trace element concentrations range from depleted to relatively enriched, with compositions similar to DMM. Variations in the most incompatible trace elements extend over 3 orders of magnitude in the depleted, pyroxenite and cryptic peridotites, while the gabbro/plag peridotites extend to 4 orders of magnitude. In comparison, the Johnson and Dick (1992) dataset from the Atlantis II Fracture Zone covers 1.5 orders of magnitude for LREE (Fig. 5.10).

5.5 Discussion

5.5.1 Justification for the use of Cpx to interpret whole-rock processes

In attempting to extract geochemical information about melt-rock reaction and source composition from abyssal peridotites, we assume that Cpx trace element concentrations accurately reflect the whole rock budget. This assumption is necessary given the highly altered nature of abyssal peridotites, but can be tested using the mineral modes and Cpx/Opx partitioning data presented in Chapter 4, Tables 4.2 and 4.3. We do not consider the role of olivine and spinel on the trace element budget, as these are generally negligible on the basis of mineral structure and mineral/melt partition coefficients (with some exceptions, e.g., spinel can contain significant amounts of TiO_2). We calculate that Cpx contains $>90\%$ of the whole rock budget for LREE and Sr in lherzolites (i.e., for $\text{Cpx} > 5\%$), whereas in harzburgites, Cpx carries $\sim 70\%$. For HREEs, Cpx in lherzolites represents $\sim 50\%$ of the budget, whereas in harzburgites it represents $\sim 20\%$. Overall, these calculations suggest that Cpx adequately record LREE/HREE fractionation during melting. However, whole-rocks provide a more sensitive record of the process, as decreasing modal Cpx with increasing degree of melting results in a decrease in the contribution of Cpx to whole-rock HREE. Thus, in more depleted peridotites, the LREE/HREE whole-rock ratio is lower than the ratio measured in Cpx.

Trace element abundances measured in Cpx have been modified by cooling and subsolidus re-equilibration between Cpx and Opx. The magnitude of this effect can be assessed from pyroxene phase relations, represented by the temperature-dependent pyroxene miscibility gap at appropriate pressures and Cpx/Opx trace element partition coefficients as a function of temperature. Witt-Eickschen and O'Neill (2005) demonstrated that Cpx/Opx trace element partitioning co-varies with pyroxene closure temperature in natural peridotites, in the direction of increasing partition coefficients with increasing temperature.

If the Witt-Eickschen and O'Neill (2005) conclusion on the temperature effect on Cpx/Opx trace element partitioning is correct, $D_{Cpx/Opx}^{La}$ varies from ~ 60 at 1250°C to ~ 1000 at 900°C . In contrast, $D_{Cpx/Opx}^{Yb}$ varies from 3 to 7 over the same temperature interval. Extrapolating the Witt-Eickschen and O'Neill (2005) trends to 1350°C , according to a $\ln(D)$ versus $1/T$ relationship, we calculate $D_{Cpx/Opx}^{La} \sim 30$ and $D_{Cpx/Opx}^{Yb} \sim 2.5$. Lindsley (1980) showed that the Opx side of the miscibility gap is considerably steeper than the Cpx side and that at 1.5 GPa, cooling from 1350°C to 1250°C will change the Cpx/Opx ratio by $\sim 15\%$, for a fixed Opx-rich bulk composition. Given these conditions, mass balance calculations demonstrate that the concentration of La measured in 1250°C Cpx underestimates La in 1350°C Cpx by about 8%, whereas Yb is overestimated by $\sim 7\%$. These values are both within analytical error, indicating that, to a first approximation, the effect of the sub-solidus re-distribution of trace elements between Cpx and Opx can be ignored. Hence, Cpx trace element concentrations provide a reasonably accurate picture of whole rock systematics.

5.5.2 Mechanisms of melt-rock reaction

Melting and melt extraction in upwelling peridotites beneath ocean ridges proceeds via the interaction of minerals with the melt percolating around them. This process is one of several mechanisms often referred to as melt-rock reaction. At the site of melt generation, melt first forms at grain junctions when the solidus of the system is overstepped. Melt mass increases via reaction with minerals as it wets grain boundaries and begins to migrate upward. Experimental and theoretical studies of dissolution-precipitation mechanisms (e.g., Iwamori, 1992; Liang, 2003; Lo Cascio et al., 2004; Morgan and Liang, 2005) provide constraints on the grain-scale kinetics of this process.

Melt extraction by fractional or near-fractional melting results in residual peridotites that are depleted in modal Cpx and incompatible trace elements. As degree of melting

increases, so does the degree of depletion. Melt migration is faster than the upward flow of residual peridotites, creating conditions for reactive porous flow. Through this process, depleted peridotites can be chemically and mineralogically modified by reaction with enriched small-degree melts, producing significant changes in Cpx trace element compositions. The general aspects of focused reactive porous flow and its importance for the extraction of melt from mid-ocean ridges have been developed by Dick et al. (1994), Kelemen et al. (1997) and colleagues. This approach is particularly successful in explaining the formation of dunite channels in the uppermost part of the mantle beneath the EPR (Dick and Natland, 1996) and the Oman Ophiolite (Kelemen et al., 1995). Open system melting models, such as those of Johnson and Dick (1992) and Ozawa and Shimizu (1995) were also developed in this context. Further development of the reactive porous flow approach has been made in the form of dissolution channel models (e.g., Spiegelman and Kenyon, 1992; Kelemen et al., 1995; Spiegelman and Kelemen, 2003), which predict the range of trace element depletion patterns expected for abyssal peridotites at large sampling length-scales.

Descriptions and models of the interactions of peridotites with migrating melts and fluids are not restricted to the ridge environment and have a long history in studies of mantle metasomatism of the continental lithosphere. Navon and Stolper (1987) and Bodinier et al. (1990) were among the first to attempt to explain melt-rock reaction in a quantitative fashion. Particularly interesting and relevant to this study is the chromatographic fractionation of trace elements predicted by their models. As melt percolates around mantle minerals, fractionation of trace elements occurs because of the differences among mineral-melt partition coefficients for different elements. When an enriched small-degree melt enters a column of depleted peridotite, REE patterns of Cpx in the peridotite will be modified successively, beginning with an increase in the more incompatible LREE and progressing towards less incompatible HREE. This process can result in Cpx with “spoon-shaped” and

“sharply inflected” REE patterns, as observed by Takazawa et al. (1992) in depleted lherzolites from the Horoman peridotite massif and by Sen et al. (1993) in xenoliths from Oahu. These early models concentrated on explaining variations among peridotite Cpx trace element patterns and did not include the modal changes that occur during melt-rock reaction. More generalized models that include mineralogical reactions have since been developed (Godard et al., 1995; Vernières et al., 1997).

In general, the nature of melt-rock reaction depends on whether melt mass is increasing (i.e., melting), constant, or decreasing (i.e., crystallization). The effect of decreasing melt mass on lithospheric mantle composition is important in the context of both ridge and subduction processes. The lithospheric mantle that is transported back into the asthenosphere may have a substantially different and more variable composition than surveys of depleted abyssal peridotites alone imply. Fig. 5.1 demonstrates that veined peridotites are found from the entire spectrum of spreading rates and at most localities where depleted abyssal peridotites are collected. Cannat (1996) argued that incomplete melt extraction (or melt entrapment) is important during lithosphere formation at lower spreading rates. Sleep and Barth (1997) used thermal modeling to demonstrate that melts can freeze directly in the mantle due to cooling from the surface and that this effect occurs at higher spreading rates than those at which conductive cooling limits the extent of melting. Lizarralde et al. (2004) presented seismic evidence for the retainment of a gabbroic phase in the mantle at low spreading rates.

The thermal regime for peridotites in this study is constrained by ultra-slow spreading rates and/or fracture zone proximity - i.e., the degree of melting is generally limited due to conductive cooling from above. Under these conditions, migrating melt is also cooled and melt extraction is expected to be incomplete. In systems where melt mass is decreasing, reaction of melt with the peridotite can refertilize a depleted peridotite. The nature of refertilization and its effect on peridotites will depend on several variables: melt/rock ratio, melt

composition, degree of peridotite depletion, pressure and temperature. Clear evidence for incomplete melt extraction at high melt/rock ratios over a range of pressures is recognized in the form of pyroxenite and gabbro veins in peridotites (Dick, 1989; Cannat et al., 1992; Constantin et al., 1995; Seyler and Bonatti, 1997; Dantas et al., 2007). Refertilization by low volume melts percolating along grain boundaries results in more cryptic forms of refertilization, the signature of which is difficult to detect. However, recent detailed studies by Seyler and co-workers (Seyler et al., 2001, 2004; Brunelli et al., 2006; Seyler et al., 2007) have documented that many residual peridotites have been weakly refertilized, with small enrichments in the most incompatible trace elements, interstitial Cpx veinlets and trace metasomatic phases.

In Fig. 5.2, we emphasized the range of compositions among unveined abyssal peridotites. In Fig. 5.13, we use the same global dataset to compare residual peridotite compositions to the compositions of veined and cryptically metasomatized peridotites. Depleted peridotites have a mean normalized Ce/Yb ratio of ~ 0.01 , but range from ~ 0.001 –10. Gabbro/plag peridotites and pyroxenite-veined peridotites span close to the same range as residual peridotites. The cryptically metasomatized peridotites – represented by individual analyses from only 3 samples – also span a similar range. The Ce/Yb range among residual peridotites probably reflects melt-rock reaction, in addition to melt depletion.

5.5.3 Length-scales of peridotite variations and their origins

We observe orders of magnitude variation in abyssal peridotite trace element concentrations at length-scales ranging from single grains to ridge segments. At the smallest length-scales, kinetics and cooling rate are important, as volume diffusion at high temperature rapidly equilibrates compositional variations, even in the absence of melt (Hofmann and Hart, 1978). Thus, variations at the dredge and outcrop scale often reflect melt-rock reaction processes over a range of temperatures. For example, among Oblique/Orthogonal

Segment dredges and the two uplifted massifs on Atlantis II Fracture Zone, a variety of vein compositions are observed. At the ridge segment scale, the Oblique Segment has large variations among residual peridotites, which we will discuss in the context of melt-rock reaction and pre-existing heterogeneities.

Porphyroclast and interstitial grain variations

Cpx in the majority of peridotites are present as both porphyroclasts and interstitial grains. Interstitial grains are <1 mm diameter and lack exsolution lamellae, whereas porphyroclast grains are typically 2-5 mm diameter, with exsolution lamellae. The morphology of some interstitial grains suggests that they formed by crystallization of melt on grain boundaries, but the majority are more equant in shape. To investigate the possibilities of mineral zoning and of crystallization from melt, we analyzed porphyroclast cores and rims and interstitial grains wherever possible. Core, rim and interstitial compositions among vein-free peridotites are compared in Fig. 5.14. With the exception of the cryptically metasomatized samples, no systematic difference exists between porphyroclast core and rim or between porphyroclast and interstitial grain. While some analyses do not show an exact correspondence, the majority of these occur at low concentrations where error is probably larger than estimated (due to low count rates).

During mantle melting, various studies have attempted to constrain whether minerals remain in equilibrium with melt or whether diffusion limits equilibrium (Qin, 1992; Iwamori, 1992, 1993; Van Orman et al., 2002; Cherniak and Liang, 2007). The lack of clear evidence for systematic incompatible element depletion of Cpx grain rims with respect to cores indicates that disequilibrium mantle melting does not occur with respect to Cpx. The approximation $x = \sqrt{D_i t}$, where x is distance, D_i is the diffusion coefficient for an element i , and t is time, can be used to assess diffusive homogenization time-scales. For Ce and Yb in Cpx at 1.5 GPa and 1350°C, using diffusion coefficients from Van Orman

et al. (2001) and a grain diameter of 2 mm, ~ 1 My is required for Ce equilibration and ~ 0.1 My for Yb equilibration. For these time-scales and at relevant ridge spreading rates, cooling rates for mantle upwelling along an adiabat are not fast enough to limit diffusive re-equilibration during melting. In contrast, significant core-to-rim enrichment of Ce in Cpx is observed in the cryptically metasomatized peridotites, suggesting that metasomatism occurred at shorter time-scales, as discussed below.

Sample- and grain- scale variations

The three cryptically metasomatized peridotites represent the smallest length-scale at which compositional variation has been observed in abyssal peridotites. These samples are unveined protogranular lherzolites. Samples 6K-465-2 and RC27-9-6-5 are from the northern ridge-transform intersection of the Atlantis II Fracture Zone and 6K-458-3 is from Atlantis Bank. In these samples, Cpx grains with typical, depleted trace element concentrations and those with enriched concentrations, similar to or greater than DMM, occur within 1-2 cm of each other. Fig. 5.12 shows REE patterns for these samples, including the zoned “transitional” Cpx in 6K-465-2 and RC27-9-6-2. Transitional Cpx are grains with gradients in trace element concentrations, from depleted cores to enriched rims, which are found spatially between depleted Cpx and enriched Cpx. The samples have 2-3 orders of magnitude range in LREE, with transitional grains covering 2 orders of magnitude alone. In sample 6K-465-2, the transitional grain, shown in Fig. 5.15, is 2 mm in the longest dimension and has variations in Ce from 0.1 ppm up to 2 ppm (Cpx 6 in Table 5.10). The lowest trace element concentrations occur at Point 4 (Fig. 5.15), suggesting that this point is close to the true center of the grain in three dimensions. The REE variations in the cryptic peridotites are similar to the theoretical predictions of chromatographic melt-rock reaction models, for example by Vernières et al. (1997). However, given the limited number of analyses for our samples and the lack of spatial information on the exact shape of the zoning

front, detailed modeling is precluded at this point.

Sample 6K-465-2 can be used to constrain the time-scales of metasomatic processes, as it has the greatest number of analyses. Of 28 points analyzed in 10 Cpx grains distributed over a 3x4 cm area of the thin section (Table 5.10), 21 points have uniform, low Sr concentrations ($\text{Sr} = 1.18 \pm 0.24$ ppm). In contrast, Ce varies by a factor of almost 400 (0.02-7.99 ppm). As Cpx-melt partition coefficients for these elements are similar ($K_d^{\text{Ce}} = 0.09$, $K_d^{\text{Sr}} = 0.13$; Hart and Dunn, 1993), the observed decoupling is unusual and probably reflects diffusion kinetics. Sr appears to have remained constant, while Ce increased 400-fold during the metasomatism, resulting in large negative Sr anomalies on a spidergram. We interpret this seemingly anomalous behavior as a result of cryptic metasomatism followed by diffusive homogenization of Sr.

At 1350°C, the diffusion coefficient for Sr is $4.6 \times 10^{-16} \text{ m}^2/\text{s}$ (Sneeringer et al., 1984), and that for Ce is $9.9 \times 10^{-20} \text{ m}^2/\text{s}$ (Van Orman et al., 2001). This implies that Sr in a 2 mm grain will homogenize in ~ 300 yr, while a crystal the size of the analyzed region of the thin section will homogenize in ~ 0.1 My. For a crystal aggregate, the equilibration time-scale across the same area is shorter, as diffusion along grain boundaries is faster than volume diffusion across a grain. In contrast, homogenization of Ce in a 2 mm Cpx grain takes ~ 1 My at 1350°C. Thus, cryptic metasomatism produces large variations in Ce and Sr, on a time-scale that is considerably shorter than the time-scale associated with near-fractional melting (e.g., > 0.1 My but < 1 My).

The upwelling rate of the SWIR at the Atlantis II Fracture Zone is 5.4 mm/yr, based on corner flow at a half-spreading rate of 8.5 mm/yr (Hosford et al., 2003). At this upwelling rate, if melt-rock reaction occurred at 1350°C, which corresponds to a depth of ~ 26 km, Cpx grains would have time for Ce homogenization. Thus, melt-rock reaction must have occurred at a lower temperature. At Atlantis Bank, gabbro intrusion occurred on-axis to relatively shallow depths (Dick et al., 2000; John et al., 2004). This agrees with the re-

quirement for cryptic metasomatism to occur at relatively low temperature ($<1200^{\circ}\text{C}$) and suggests that metasomatism occurred in peridotite wallrock adjacent to a gabbro intrusion.

Trace amounts of plagioclase, intergrown with spinel, olivine and pyroxenes, is present in sample 6K-465-2. We suggest that plagioclase is the result of the breakdown reaction of spinel, Opx and Cpx to form plagioclase and olivine. Plagioclase crystallization appears to post-date the cryptic metasomatism event, as both enriched and depleted Cpx have negative Eu anomalies on their rims but not their cores (Fig. 5.12). The spinel/plagioclase transition occurs around $\sim 15\text{-}20$ km depth, but this is highly dependent on bulk composition and temperature. In particular, the presence of Cr promotes the persistence of spinel (O'Neill, 1981), while low Na suppresses plagioclase formation. Spinel is locally enriched in Cr ($\text{Cr}\#=47$ near plagioclase, versus $\text{Cr}\#=22$ away from plagioclase, Table 5.8), while plagioclase has very low Na concentrations (An_{90-93} , Table 5.8), suggesting a relatively shallow depth for plagioclase formation. Plagioclase also has variable Sr concentrations, ranging from 1 ppm to 60 ppm (Table 5.11), with the lower concentrations suggestive of partitioning with Cpx, not with a melt.

Sr concentrations in 6K-465-2 are sporadically high in some Cpx rims (up to 6 ppm) and some interstitial plagioclase (up to 60 ppm). This implies a final stage of melt-rock reaction, which led to crystallization of a small amount of melt along grain boundaries. If true, a first metasomatic event occurred deeper than the depth of stability for anorthitic plagioclase, whereas the second melt infiltrated at a shallower depth. Our observations are compatible with the first event occurring, for example, in the range $\sim 1000\text{-}1200^{\circ}\text{C}$, equivalent to a depth of $\sim 11\text{-}19$ km or an upwelling time to the seafloor of 2-3 My. This first metasomatic event was followed by plagioclase breakdown around 10-15 km depth, before a second episode of metasomatism. Our interpretation coincides with observations of a complex magmatic history at Atlantis Bank, involving several cycles of melt intrusion, including late-stage, off-axis events (John et al., 2004). While 6K-465-2 is located at the present-

day inside corner high, the processes that are recorded at Atlantis Bank are assumed to be equivalent to the processes that are currently occurring at the ridge-transform intersection. The multiple episodes of melt-rock reaction recorded in the peridotites demonstrate the complexity of the path of melt through the upwelling mantle.

Massif-scale variations on the Atlantis II Fracture Zone

In Fig. 5.16, we plot sample averaged Cpx REEs for dredges and dives from the Atlantis II Fracture Zone ridge-transform intersection and Atlantis Bank. Individual analyses for the cryptically metasomatized peridotites are shown as sample averages to emphasize the variation among other peridotites. The large variations in the cryptically metasomatized peridotites (Fig. 5.12) are mirrored in the range of compositions observed among all samples at the Atlantis II Fracture Zone localities. In contrast, the range of depleted Atlantis II Fracture Zone peridotite compositions from the full length of the active transform fault (Johnson and Dick, 1992) is much more limited (Fig. 5.16) and residual peridotites at Atlantis Bank extend over most of this range.

Johnson and Dick (1992) related the variation in peridotite composition along the fracture zone to variations in degree of melting. Our observations suggest that their sample suite is not representative of the range of peridotite characteristics along the fracture zone. In particular, we note the role of melt-rock reaction in modifying peridotite composition at both localities that we have sampled in detail. In addition, in Chapter 4 we presented evidence for source heterogeneity beneath the fracture zone. Clinopyroxenite-veined peridotite RC27-9-6-2 has an enriched isotopic composition, implying that it originated from a long-term mantle heterogeneity. As noted in Chapter 4, the sample is depleted in all REE with respect to DMM and the vein is more depleted than the host peridotite, suggesting that the vein post-dates the initial isotopic heterogeneity of the sample.

Both the ridge-transform intersection and Atlantis Bank are uplifted oceanic massifs

of lower crust and upper mantle. At Atlantis Bank, gabbro is the dominant lithology and peridotites have been collected in close proximity to the gabbros (Dick et al., 2000). In addition, the gabbros at Atlantis Bank are cut by many late small ferrogabbro intrusions (Dick et al., 2000). The most enriched gabbro vein compositions at Atlantis Bank are considerably more enriched than Atlantis II basalts and are probably compatible with derivation from an evolved melt similar to the one that produced the ferrogabbros. In peridotite 6K-465-3, one of the cross-cutting veins contains plagioclase, kaersutite, rutile, ilmenite, and apatite. In addition, clinopyroxene in the peridotite matrix have unusual exsolution consisting of olivine, plagioclase and spinel, as discussed in Appendix 5.D. Gabbro veins and pyroxenite veins in Atlantis II Fracture Zone peridotites also extend to relatively depleted compositions, suggesting that crystallizing vein melts had a variety of compositions.

Dredge-scale variations on the Oblique/Orthogonal Segments

We analyzed 6 dredges in detail from the Oblique and Orthogonal Segments, to assess the degree of within- and between-dredge variability. Samples analyzed were chosen to represent the range of macroscopic dredge characteristics, so we assume that we have accurately sampled the compositional range of each dredge. REE patterns in Fig. 5.17 demonstrate that within-dredge variations are relatively small but that considerable differences exist between dredges. Dredge tracks are typically 1 km in length, though wire tension records indicate that sampling is discontinuous and often dominated by one section. Dredge Kn162-47 is unusual in containing both pyroxenites and plagioclase peridotites. In addition, the dredge contains peridotite mylonites, suggesting that samples come from different, fault-juxtaposed, mantle domains. In contrast, most dredges are characterized by samples with a single set of characteristics, such as vein-free residual peridotites (PS86-6, Van7-85 and Van7-86), pyroxenite veins (Van7-96) or peridotites with gabbro veins and matrix plagioclase (Van7-78).

Among the three vein-free peridotite dredges, Van7-85 and Van7-86 have “typical” depleted compositions, with samples plotting in the middle of the residual peridotite field from Johnson et al. (1990). One sample, Van7-86-37, deviates from the general trend, with elevated LREE and a negative Eu anomaly. These observations suggest melt refertilization and plagioclase crystallization, but alteration of the sample is too great for fresh plagioclase to be identified. The third unveined peridotite dredge, PS86-6, is less depleted in LREE than dredges Van7-85 and Van7-86, and has high modal Cpx (8-12%, Table 5.3). These peridotites could either have undergone a relatively low degree of melting or they could have been refertilized following a more typical amount of melt extraction. As discussed in Chapter 4, these peridotites are isotopically heterogeneous and may have been influenced by proximity of Bouvet hotspot along the Shaka Fracture Zone at 20 Ma (Fig. 5.3).

Dredge Van7-96 contains peridotites with pyroxenite veins that are relatively undepleted with respect to DMM. Variations in trace element concentrations among these samples are greater than the observed variations among depleted dredges. As discussed in detail in Chapter 4, the pyroxenite veins and host peridotites have a range of isotopic compositions between DMM and Bouvet hotspot, suggesting that they represent hotspot-modified mantle. A negative correlation between $^{143}\text{Nd}/^{144}\text{Nd}$ and Ce/Yb (Chapter 4, Fig. 4.14) further suggests that the isotopic and LREE enriched signatures are due to interaction with the Bouvet hotspot. The isotopically most depleted peridotite (Van7-96-38: $^{143}\text{Nd}/^{144}\text{Nd} = 0.513205$, $^{87}\text{Sr}/^{86}\text{Sr} = 0.702096$) is the most depleted in LREE (Fig. 5.17). In addition the pyroxenites have a secondary, cross-cutting gabbroic vein assemblage, suggesting multiple stages of melt-rock reaction over a range of pressures. Dredge 96 has an effective spreading rate of 12 mm/yr and is located adjacent to Shaka Fracture Zone, implying minimal melting during upwelling, yet even at this locality melt-rock reaction occurred.

The gabbro veined peridotites in dredge Van7-78 do not exhibit a large compositional range and have concentrations similar to DMM, with a slight fanning out of LREE among

different samples. The veins are not enriched with respect to their host peridotites (see the REE plots in Appendix 5.A for detailed within-sample comparisons). With respect to the gabbro-veined peridotites from Atlantis II Fracture Zone (Fig. 5.16), these veins and their host peridotites have compositions in the middle of the gabbro range. Variations among these samples are probably due to variable amounts of melt refertilization.

Dredge Kn162-47 contains residual peridotites, pyroxenite-veined peridotites, and a plagioclase peridotite. All three lithologies have typical depleted compositions, with the exception of Kn162-47-09. The depleted composition of the pyroxenite and plagioclase peridotites implies that they formed from unaggregated melts in equilibrium with the surrounding peridotite.

Sample Kn162-47-09 has an unusual, sinusoidal shaped REE pattern, accompanied by large negative Ti and Zr anomalies on a spidergram. Together with its Nd isotopic composition ($\epsilon_{Nd}=-2.8$) reported by Salters and Dick (2002), the geochemical characteristics of this sample, highly unusual for abyssal peridotites, resemble continental lithosphere. Cpx from a spinel harzburgite xenolith from the Premier Mine, South Africa (PHN 5266, Boyd et al., 1999) has an almost identical spidergram (N. Shimizu, unpublished data). In addition, Opx and Cpx have high Mg# – 91.7 for Opx and 93.1 for Cpx, compared to global average values for residual peridotites of 91 for both Opx and Cpx. Cr_2O_3 is also high (0.82 wt% for Opx and 1.55 wt% for Cpx; cf. 0.57 - 1.08 wt%), whereas Al_2O_3 is low (3.29 wt% for Opx and 4.32 wt% for Cpx; cf. 5.20 - 7.83 wt%). These mineral compositions are consistent with the Kaapvaal spinel peridotite data (Boyd et al., 1999).

The alteration features and general appearance of sample Kn162-47-09 are identical to other peridotite samples from the same dredge, indicating that this peridotite was part of the oceanic lithosphere when it was emplaced on the seafloor (and not ice-rafted debris). On the basis of this sample, we suggest that rocks from the continental lithosphere can maintain mineralogical and geochemical integrity throughout the process of entrainment

into the convecting mantle, surviving the melt-rock reaction process during upwelling beneath ocean ridges. We note that in the global abyssal peridotite dataset, 6 other samples, including 3 from the Gakkel Ridge, have similar REE patterns, but no isotopic data are available. Recent studies by Goldstein et al. (2006) and Michael et al. (2006) postulated that continental lithosphere is involved in magma genesis at the Gakkel Ridge, on the basis of MORB geochemistry.

Overall, dredges along the Oblique/Orthogonal Segments indicate variations in the occurrence, amount and depth of melt-rock reaction. However, the compositional range among Oblique/Orthogonal samples is small with respect to Atlantis II Fracture Zone samples. The larger variations at smaller length-scales at Atlantis II imply that the nature of melt-mantle interaction is different between these localities. At Atlantis Bank, magmatism occurred for a long time, ultimately intruding into relatively cool lithosphere (John et al., 2004). In contrast, spreading along the Oblique Segment is ultra-slow and large gabbro bodies, such as Atlantis Bank, are not observed. In fact, small gabbro intrusions are also absent from peridotite dredges along the Oblique Segment. Therefore, melt-mantle interaction at the Oblique Segment was likely more limited in spatial and temporal extent, than at faster upwelling rates.

5.5.4 Estimates of degree of melting

In previous sections, we have suggested that many abyssal peridotites have undergone melt-rock reaction during melt extraction. We have also suggested that the mantle may not be homogenous in composition, as commonly assumed (e.g., Johnson et al., 1990). To assess the role of source heterogeneity on peridotite composition, we use peridotite compositions to calculate degrees of melting along the Oblique Segment. We then compare these calculations to predictions from a thermal model of the Oblique Segment by Montési et al. (2006).

The data that we use to assess Oblique Segment source composition are Cpx trace elements from 11 dredges and spinel major elements from 23 dredges, for residual peridotites only. Trace elements have been determined in Cpx from 38 residual peridotites (this study; H. J. B. Dick unpublished data; Johnson et al., 1990; Salters and Dick, 2002). For spinel, using additional unpublished data from H. J. B. Dick, data are available from 132 samples, with multiple samples analyzed per dredge, except for two dredges. In Fig. 5.18, spinel Cr#, Mg# and TiO₂ are shown for this expanded dataset of Oblique Segment residual peridotites. Oblique Segment spinels are offset to slightly lower Mg# at a given Cr#, in comparison to the global dataset. As Mg# reflects thermal re-equilibration between olivine and spinel (Dick and Bullen, 1984), the offset may reflect the ultra-slow upwelling rate along the Oblique Segment.

In Fig. 5.19, the variation of Cpx Ce/Yb and spinel Cr# are shown as a function of longitude along the ~ 500 km long Oblique Segment. Both indices of peridotite depletion show large variations within individual dredges and among closely spaced dredges. For example, at 13°-13.5°E (a distance of ~55 km), spinel Cr# covers the entire global range, from ~10%-60%. This includes dredge PS86-6, for which the Cpx Nd isotopic range is 30% of the Indian Ocean MORB range (Chapter 4). Pre-melting chemical variation in the mantle is expected to accompany this isotopic variability.

To demonstrate the implications of the large range in peridotite composition in Fig. 5.19, we estimate degree of melting for the peridotites, assuming an initial DMM composition. In calculating degree of melting from spinel compositions, we first filtered the dataset for high TiO₂ (Fig. 5.18), using the composition of DMM spinel (0.17 wt%, Workman and Hart, 2005), to remove spinels that have been affected by melt infiltration. Degree of melting is calculated from the empirical correlation of Cr# with Cpx HREE, based on a non-modal fractional melting model for HREE (Hellebrand et al., 2001, see Appendix 5.B for a discussion of the calibration). For Cpx trace elements, we also use fractional melt-

ing, to provide a direct comparison to the predictions from spinel Cr#. Degree of melting for Ce and Yb are calculated using the melting reaction from Kinzler (1997) and partition coefficients compiled by Kelemen et al. (2003). These parameters are similar to those used by Hellebrand et al. (2001) to calibrate the Cr# relationship to melting.

The results of the fractional melting calculations are shown in Fig. 5.19. If the maximum Cr# in a dredge represents the maximum degree of melting, then the average maximum degree of melting based on spinel composition is 9%. The within-dredge range in spinel Cr# translates to an average within-dredge melting range of 5%. However, at the dredge length-scale (<1 km), this range cannot result from degree of melting and must reflect either initial heterogeneity or melt-rock reaction. Ce in Cpx and Yb in Cpx predict 5% and 7% melting, respectively, with an average per dredge difference between these two estimates of 3%. Brunelli et al. (2006) suggest, on the basis of trace element modeling, that the majority of residual peridotites have undergone a small amount of refertilization by melts that are close to being in equilibrium with the peridotite. These melts result in a larger increase in LREE than in HREE. If true, Ce is expected to underestimate degree of melting while Yb is expected to provide a more representative estimate. However, Ce does not consistently provide the lower estimate of melting, which may be a signal of initial source heterogeneity in relative REE abundances. The differences in average degree of melting among the three estimates (Cr#, Ce, and Yb) could be related to the simplified melting model, but the range within and between dredges is significant and model independent.

In Fig. 5.19, we also plot the predicted variation in degree of melting from a thermal model by Montési et al. (2006). This 3-D model of thermal structure combines the effect of conductive cooling on ridge thermal structure with the specific geometry of the Oblique Segment. As discussed by Dick et al. (2003) and Standish et al. (submitted), the Oblique Segment consists of three long, obliquely spreading, amagmatic segments separated by two short, orthogonally spreading, volcanic centers (Joseph Mayes Seamount and Narrowgate,

Fig. 5.3). The model predicts an overall maximum degree of melting $<12\%$, with reduced degrees of melting along the amagmatic segments and enhanced melting at the two volcanic centers. To a first order, the model agrees with bathymetry, gravity, and basalt chemistry, except for the 15° - 16° E segment between Narrowgate and the Orthogonal Segment. In this region, the model predicts a higher degree of melting than expected from geological observations (L. Montési, personal communication, 2007).

We focus on the $\sim 12^{\circ}$ - 14° E section in Fig. 5.19, where the model is robust and abundant observations are available from the peridotites. The thermal model predicts ~ 4 - 6% melting, with a continuous decrease towards the segment center. Predicted degree of melting varies between 2-6% for Ce in Cpx and 2-9% for Yb in Cpx. Given the simplicity of the trace element model and the precision of the thermal model, these estimates agree within error. However, the trace element data exhibit variability at shorter length-scales than predicted by the thermal model. Spinel Cr# exhibits even greater variability and predicts considerably higher degrees of melting. Using the maximum Cr#, the predicted range for degree of melting is 3-18%, whereas using the minimum Cr#, the range is 0-15%.

The thermal model of Montési et al. (2006) demonstrates the length-scale over which temperature is expected to produce variations in degree of melting. The shorter length-scale variations in peridotite composition along the Oblique Segment must reflect either compositional heterogeneity or melt-rock reaction. From the Cpx trace element data, we suggest that within-dredge variability is dominated by melt-rock reaction, but that the offset among dredges dominantly reflects source heterogeneity. However, the within-dredge variation among spinel Cr# cannot result from melt-rock reaction, as spinel TiO_2 and, to some extent, Mg#, should also be affected (for example, compare the composition of dunites to peridotites in Figs. 5.7 and 5.8). Therefore, we suggest that Cpx trace elements may be decoupled from spinel major elements, reflecting recent melt-rock interaction. In contrast, spinel Cr# variations, unaccompanied by TiO_2 variations, may reflect pre-existing major

element heterogeneity of the mantle.

5.5.5 Towards an understanding of mantle initial composition and regional variability

We interpret the large range in Cr# at the single dredge scale and the variation in Cpx trace elements among dredges as indicators of initial source heterogeneity. Our working model is that heterogeneous lithosphere is created at mid-ocean ridges by large-scale variations in degree of melting and the small-scale effects of melt-rock reaction. This lithosphere is eventually returned to the convecting mantle at subduction zones. Consequently, the mantle exposed at the ridge axis today is one whose composition over time has become increasingly heterogeneous. The important question then becomes, how variable is mantle bulk composition?

The issue of initial heterogeneity has important implications for the interpretation of peridotite and MORB compositions. Assuming a homogeneous source, basalts and peridotites can be used to calculate asthenospheric mantle composition (e.g., Workman and Hart, 2005). This source, DMM, is then combined with the local composition of basalts or peridotites to estimate degree of mantle melting. Variations in degree of melting are used to reflect variations in mantle thermal structure. However, if the mantle is heterogeneous, variations in predicted degree of melting could be a result of chemical variations, not thermal variations. With a heterogeneous source, degree of melting and extent of melt-rock reaction are difficult to determine for a given abyssal peridotite composition.

The issue of local-scale variability in abyssal peridotite and MORB compositions has been noted before (Dick et al., 1984; Klein and Langmuir, 1989). Dick et al. (1984) suggested that averaging peridotite modal compositions on a regional scale defines separate melting trends among peridotites, interpreting this as indicative of regional differences in mantle composition. They also observed modal variations within-dredges and among

dredge from the same locality, which they attributed to variations in degree of melt extraction for the same degree of melting. Klein and Langmuir (1989) suggested that local variations in MORB composition result from variable sampling of instantaneous melts throughout the melt column, but they could not rule out major element source heterogeneity as an origin.

In this study, we have documented variations in peridotite composition at the scale of a grain, a sample, a dredge, a massif, a fracture zone and a ridge segment. We use these observations in Fig. 5.20 to assess the length-scales of compositional variability at ridges. In Fig. 5.20, variations in peridotite composition are observed to extend over similar ranges in Cpx Ce/Yb at all length-scales. The modal data from Dick et al. (1984) suggests a similar picture, with peridotite modes varying from the ridge-scale down to the dredge-scale. Data from this study and Hellebrand et al. (2002) suggest that spinel Cr# is also heterogeneous at all lengthscales. Thus, we suggest that the composition of the mantle is heterogeneous over a range of length-scales.

5.6 Conclusions

Abyssal peridotites are typically interpreted as the residues of variable degrees of melting. However, the range of abyssal peridotite compositions, shown in Fig. 5.13, cannot result from melt depletion alone. In this chapter, we considered two additional origins for peridotite compositional variations: melt-rock reaction and initial heterogeneity. The occurrence of melt-rock reaction is readily observed in veined samples. In vein-free samples, distinguishing between melt-rock reaction, degree of melting and initial heterogeneity requires careful analysis. Peridotite compositions are frequently used to predict the degree of melting at a ridge (e.g., Johnson and Dick, 1992; Hellebrand et al., 2001; Le Mée et al., 2004). If source heterogeneity and melt-rock reaction are significant, as we suggest they are, then peridotites can only be used with great care to determine melting histories.

Melting and melt extraction in the oceanic upper mantle are often assumed to be two distinct processes, such that melt and mantle do not significantly interact following melt formation. This view is based on the composition of basalts, which require melt transport out of equilibrium with the mantle, and by the occurrence of dunites, which provide pathways for disequilibrium melt extraction (e.g., Dick et al., 1994; Kelemen et al., 1995). Consideration of these processes has often resulted in the exclusion of other mechanisms from the interpretation of abyssal peridotite compositions. However, the compositional and lithological variation among abyssal peridotites (Fig. 5.13) and the global distribution of these lithologies (Fig. 5.1) indicates that interaction of melt with the mantle is important at ridges.

We conclude that at least one of the major processes for creating large chemical variations in abyssal peridotites is melt-rock reaction and refertilization, in addition to melt extraction by focussed flow. Abyssal peridotites have compositions that vary over 3 orders of magnitude in Ce/Yb, as demonstrated in Fig. 5.13. The dataset for 67 peridotites presented in this study indicates that refertilization of peridotites occurs at many length-scales and to varying extents. Consequently, when the oceanic lithosphere is eventually recycled back into the mantle, it introduces a greater range of heterogeneity than expected from a simple melt extraction model for mid-ocean ridge melting.

On the basis of significant small-scale variations in peridotite compositions from the Oblique Segment, we suggest that the mantle is compositionally heterogeneous. If Oblique Segment peridotites are interpreted entirely in terms of degree of melting, spinel Cr# and Cpx trace element concentrations predict different degrees of melting and variations on a shorter length-scale than predicted by the thermal modeling by Montési et al. (2006). We suggest that some of the dredge-scale variation in peridotite trace elements is related to melt-rock reaction processes. However, the large range in spinel Cr# is not accompanied by increasing TiO₂, indicating that the range does not reflect melt infiltration. Consequently,

we conclude that the peridotites were compositionally heterogeneous before the onset of melting, as degree of melting does not vary at the length-scale of a dredge.

Our interpretations in this chapter are based on data obtained at small sampling length-scales. This study expands the observational length-scales for variations in abyssal peridotite compositions down to the single grain ($\sim 1\text{-}5$ mm) and single rock ($\sim 10\text{-}50$ cm) scales and up to the single dredge (< 1 km) and single ridge segment (~ 500 km) scales. We demonstrate that the ranges and nature of chemical variations are similar at all the length-scales: the magnitude of chemical variation in the mantle is scale independent.

Appendix 5.A

Plots of REE variations in individual samples

In Fig. 5.A1, REE plots of each individual analysis for Cpx, Opx and Plag are shown by sample. For Oblique Segment peridotites, the background field from Johnson et al. (1990) for residual peridotites away from hotspots is included for comparison. For Atlantis II Fracture Zone peridotites, the peridotite field from Johnson and Dick (1992) is used instead, to provide a direct comparison to the previous assessment of Atlantis II variability.

Appendix 5.B

Spinel Cr# and predictions of degree of melting

Hellebrand et al. (2001) proposed that the co-variation of spinel Cr# with Cpx HREE (Dy, Er and Yb) reflects variation in degree of melting and used this to calibrate a degree of melting relationship for spinel Cr#. In Fig. 5.19, we used this relationship to derive an estimate of the variation in degree of melting along the Oblique Segment. However, the Cr# estimates for degree of melting do not agree with the estimate from Cpx trace elements. In addition, a considerable range in spinel Cr# is observed within dredges, which does not reflect the degree of recent melting.

In Fig. 5.B1, we plot the co-variation of spinel Cr#, Cpx Yb_N and modal Cpx, also an indicator of degree of melting. Using the full global dataset, the correlation coefficient between Cr# and Cpx Yb is 0.57, versus 0.75 reported by Hellebrand et al. (2001). Similar decreases in correlation coefficients are observed for Dy and Er. Oblique Segment peridotites are offset to slightly lower Cr#'s with respect to the global dataset. However, the correlation is worse if these samples are excluded.

We suggest that the correlation of spinel Cr# with Cpx HREE is only loosely applicable.

Considerable within-dredge variation in spinel Cr# is observed at the Oblique Segment. If the average spinel Cr# per dredge is used to estimate degree of melting, instead of the highest Cr#, the estimate is still higher than trace element and thermal model estimates (Fig. 5.19). The extent of sample-scale variation in spinel Cr# has not been thoroughly assessed, though Hellebrand et al. (2002) found a Cr# range of 26-49 in spinels from a single Central Indian Ridge peridotite. These observations imply that spinel Cr# should only be used with caution in estimating degree of melting from peridotite composition.

Appendix 5.C

Armalcolite in Dredge 96

The mineral armalcolite ($\text{MgFeTi}_2\text{O}_5$) was identified in pyroxenites from dredge Van7-96 (Fig. 5.C1). Armalcolite is present as small grains at the tips of spinel grains, which are themselves surrounded by plagioclase within secondary gabbro veins (Fig. 5.C1). The approximate composition of the armalcolite is 71 wt% TiO_2 , 12% FeO , 5% Cr_2O_3 , 5% CaO , 3% MgO , and 1% Al_2O_3 . An unidentified trace element oxide must also be present, on the basis of the 97 wt% total and comparison to xenolith armalcolites (Haggerty, 1991). Further work is necessary to constrain the extent of the armalcolite occurrence in dredge Van7-96. We have not yet determined whether the armalcolite originated with the peridotite, the pyroxenites veins or the secondary gabbro veins that crosscut the pyroxenites.

To our knowledge, the armalcolite in dredge Van7-96 is the first observation of this mineral in abyssal peridotites, though it has been identified in continental peridotite xenoliths (e.g., Haggerty, 1987; Ionov et al., 1999) and in harzburgite xenoliths from the Kerguelen islands (Grégoire et al., 2000). Its presence in xenoliths is associated with melt metasomatic assemblages (Haggerty, 1991). We suggest that armalcolite formation in mantle assemblages is related to the breakdown of high-Ti spinel at high temperature to low-Ti spinel

and a pseudobrookite phase at low temperature (Muan et al., 1971). Armalcolite is a solid solution between ferropseudobrookite and karrooite and can undergo further breakdown to ilmenite solid solution and rutile with decreasing temperature or decreasing MgO (Lindsley et al., 1974; Xirouchakis et al., 2001). Both rutile and ilmenite have been observed in another SWIR abyssal peridotite in this study, 6K-465-3, suggesting that low volume, highly TiO₂ enriched melts are not entirely uncommon during late-stage melt-rock reaction at the ridge axis.

Appendix 5.D

Notes on gabbro-veined peridotite 6K-465-3

Sample 6K-465-3 from the Atlantis II Fracture Zone contains two melt veins of distinct compositions and matrix Cpx with unusual exsolution. One of the melt veins is gabbroic, consisting of plagioclase and Cpx with accessory ilmenite. Cpx porphyroclast cores have been altered to actinolite by high-temperature hydrothermal alteration. The second vein contains kaersutite, plagioclase, ilmenite, rutile, and apatite (Fig. 5.D1), indicating crystallization from a highly differentiated melt. In the peridotite matrix, Cpx porphyroclasts have blebby exsolution of plagioclase, olivine, and spinel, as shown in Fig. 5.D2. Opx does not appear to be present as an exsolution phase in Cpx. Major and trace element compositions of these accessory mineral phases and the unusual exsolution are presented in Table 5.D1.

5.7 Bibliography

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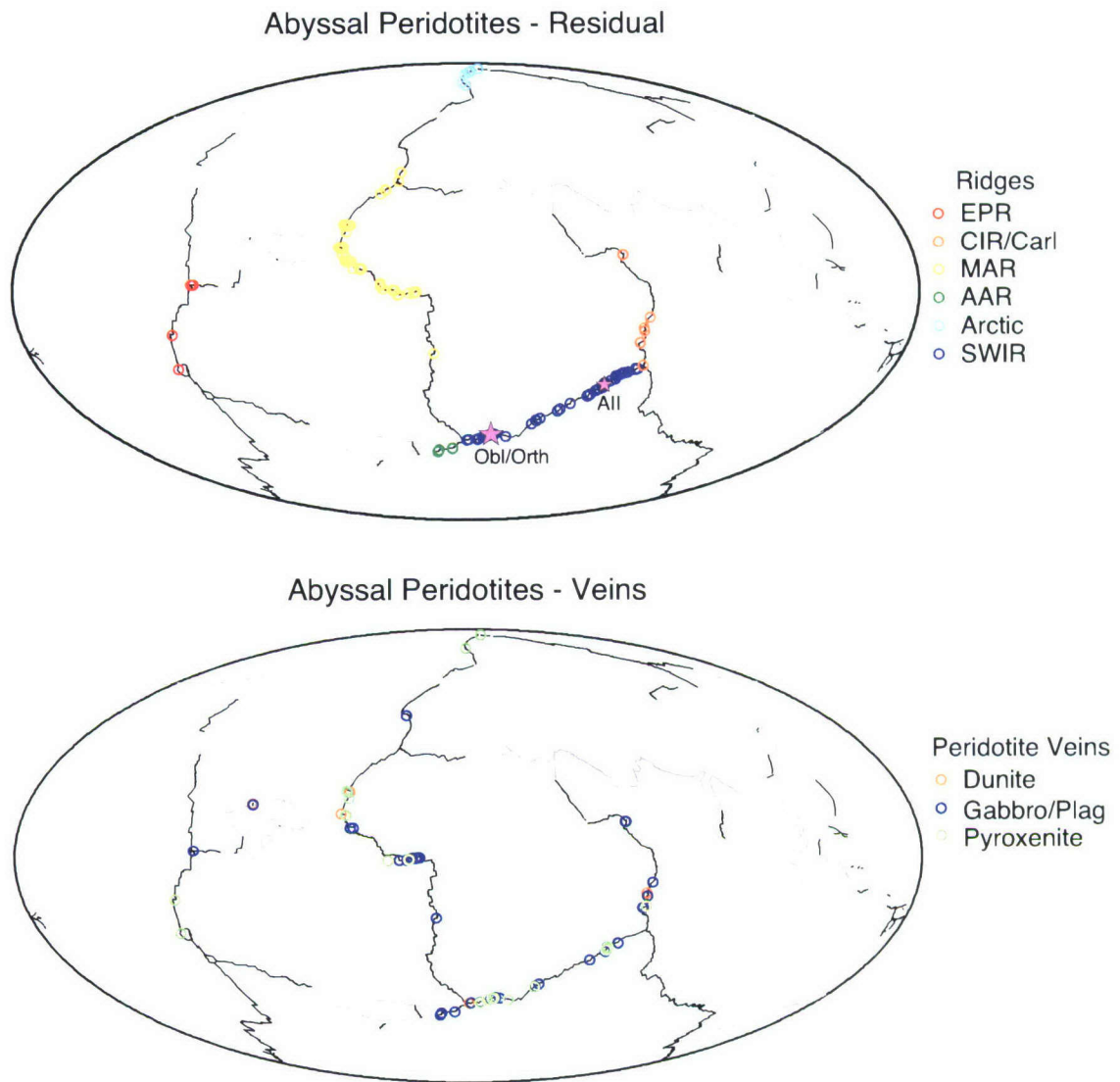


Figure 5.1: Global distribution of depleted (top) and veined (bottom) abyssal peridotites. Peridotite locations are for peridotites with published modal, Cpx trace element, or spinel Cr# analyses; sources are listed by location in Table 5.1. Depleted peridotites have been filtered for any peridotites with veins or $>.05\%$ plagioclase and are colored according to ridge location: East Pacific Rise (EPR), Mid-Cayman Rise (MCR), Central Indian Ridge/Carlsberg Ridge (CIR/Carl), Mid-Atlantic Ridge (MAR), American-Antarctic Ridge (AAR), Gakkel Ridge/Lena Trough/Knipovitch Ridge (Arctic) and the Southwest Indian Ridge (SWIR). Locations of the two focus areas for this study, the Oblique and Orthogonal Segments and the Atlantis II Fracture Zone are indicated by purple stars. Veined peridotites are colored according to their dominant vein lithology. Gabbro/Plag refers to both gabbro veins in peridotites and peridotites with $>.05\%$ matrix plagioclase.

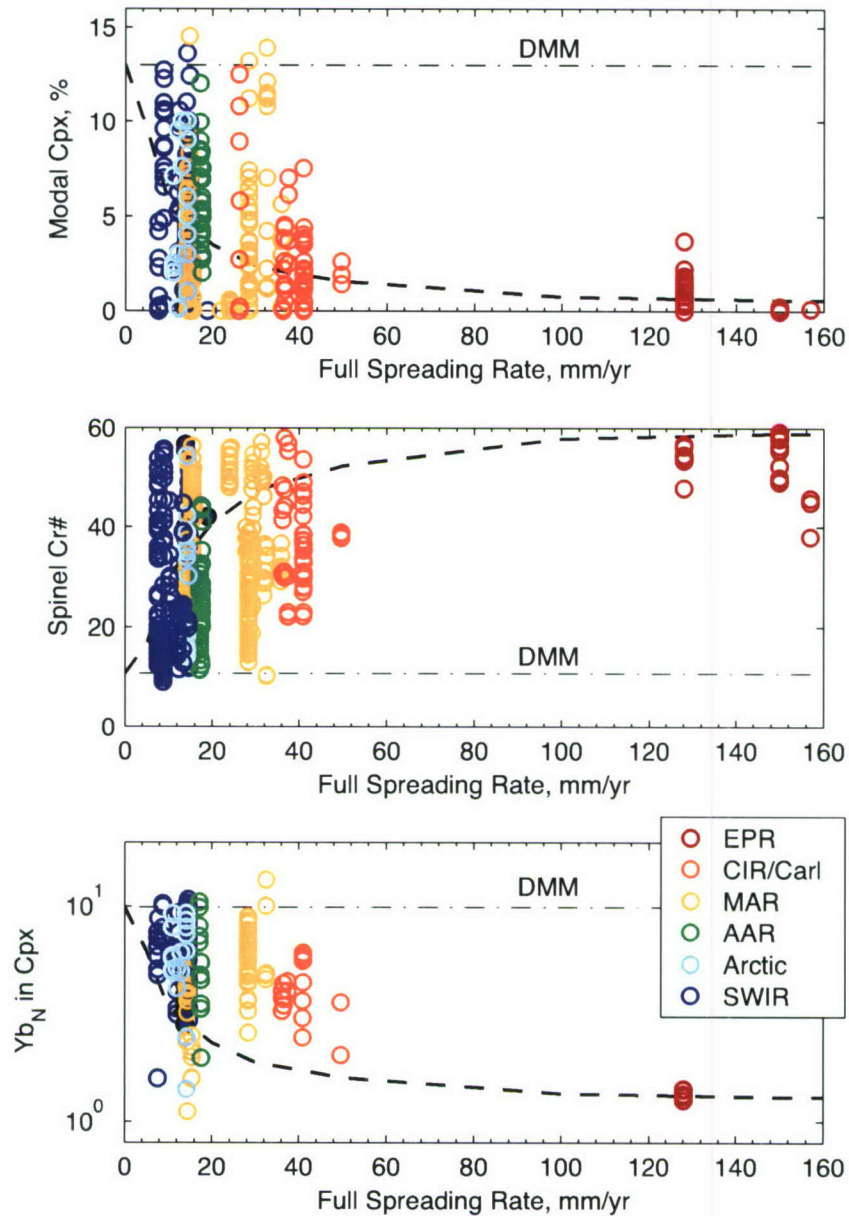


Figure 5.2: Variations in peridotite compositions as a function of spreading rate. Data are from this study and references in Table 5.1. Dashed lines are the predicted variations in composition based on (i) the degree of melting predicted by a simple plate model with conductive cooling (Reid and Jackson, 1981; Bown and White, 1994) and (ii) non-modal fractional melting using parameters and equations from Johnson et al. (1990), Kinzler (1997), Hellebrand et al. (2001), Kelemen et al. (2003) and Workman and Hart (2005). Yb is normalized to chondrite (Anders and Grevesse, 1989), Cr# is calculated as $\text{Cr}/(\text{Cr} + \text{Al})$ and the composition of DMM is from Workman and Hart (2005).

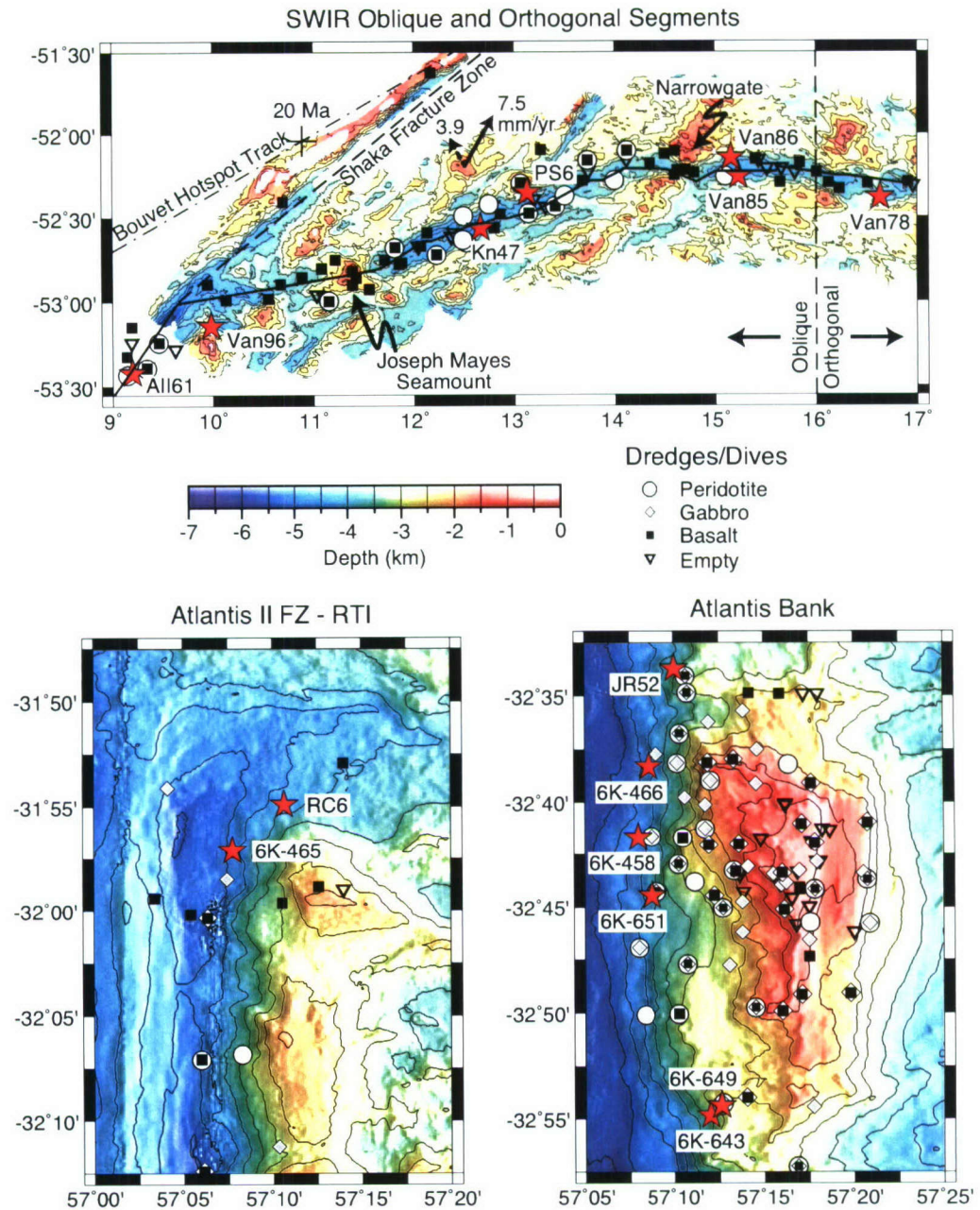


Figure 5.3: Maps of the SWIR study areas. Shown at top are the Oblique Segment and western end of the Orthogonal Segment, along with the Bouvet hotspot path Hartnady and le Roex (1985). At bottom, two localities on Atlantis II Fracture Zone are shown — the northern ridge-transform intersection (RTI) and Atlantis Bank. The fracture zone is parallel to the western edge of both maps, with Atlantis Bank located on the active portion of the transform fault at 11-12 Ma (John et al., 2004). Red stars indicate sample locations for this study. Dredges are labeled with a letter abbreviation to indicate the cruise (see Table 5.2). Dives are indicated by the prefix 6K, to indicate sampling by the submersible Shinkai 6500.

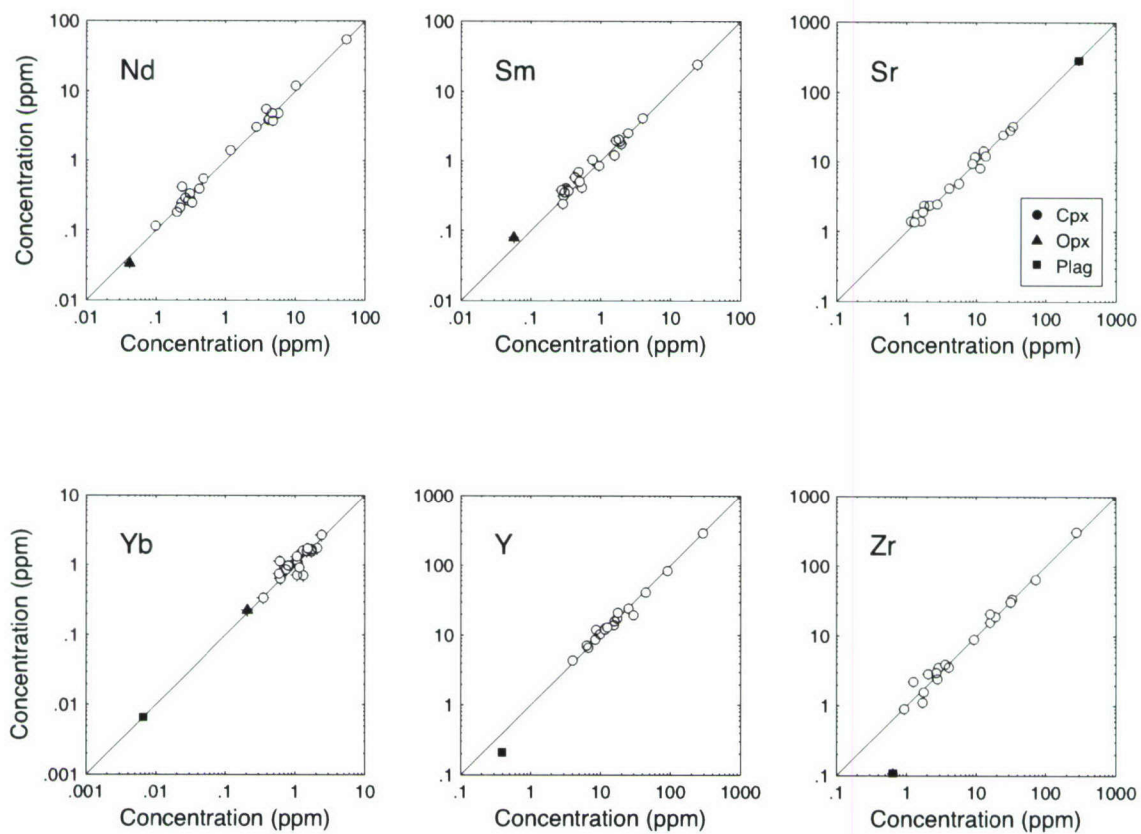


Figure 5.4: Comparison of duplicate ion probe concentration analyses for Cpx, plus one Opx and one Plag analysis. Error bars are generally smaller than the symbol size.

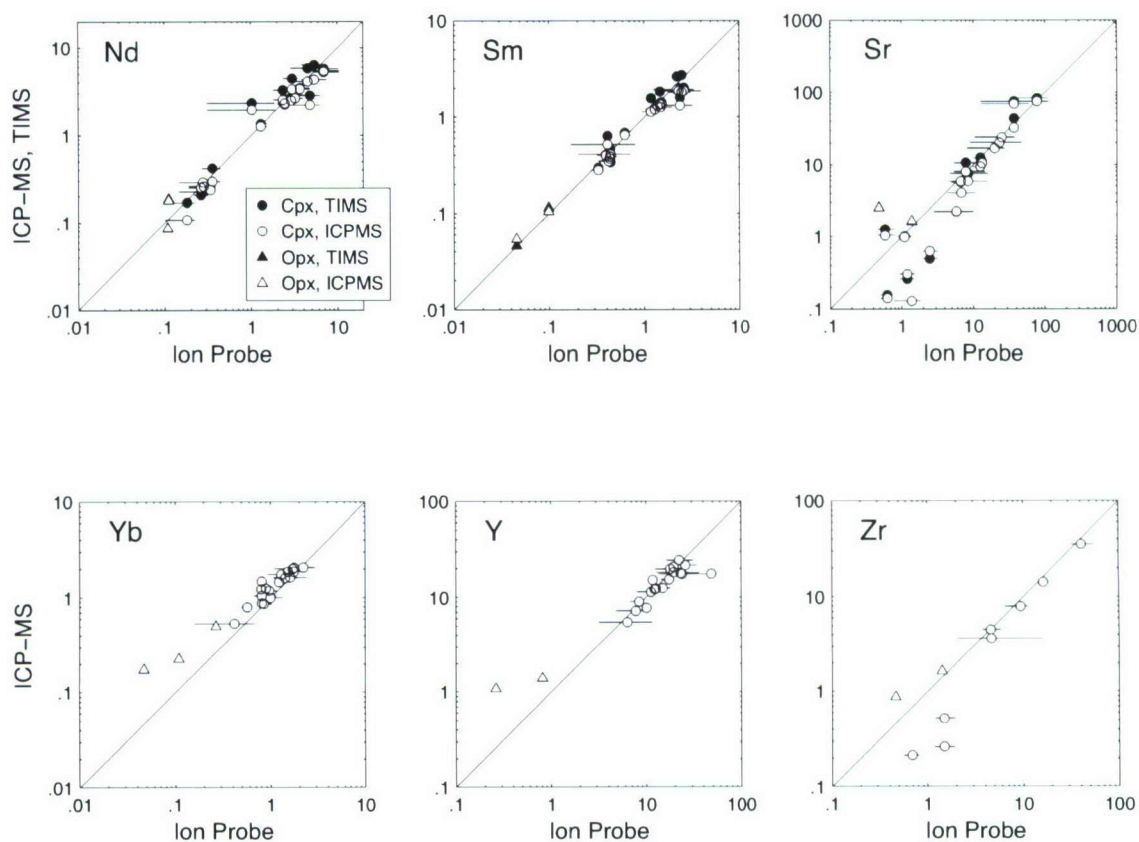


Figure 5.5: Comparison of ion probe concentration data (in ppm) with ICP-MS and TIMS data for Cpx and Opx. The ICP-MS and TIMS data were determined for mineral separates and the error is generally smaller than the symbol size. The ion probe Cpx data are averages of several *in situ* analyses per sample and the horizontal bar indicates the compositional range of the sample. For the three Opx analyses, the ion probe data is not representative of the bulk Opx composition as only one point was measured per sample.

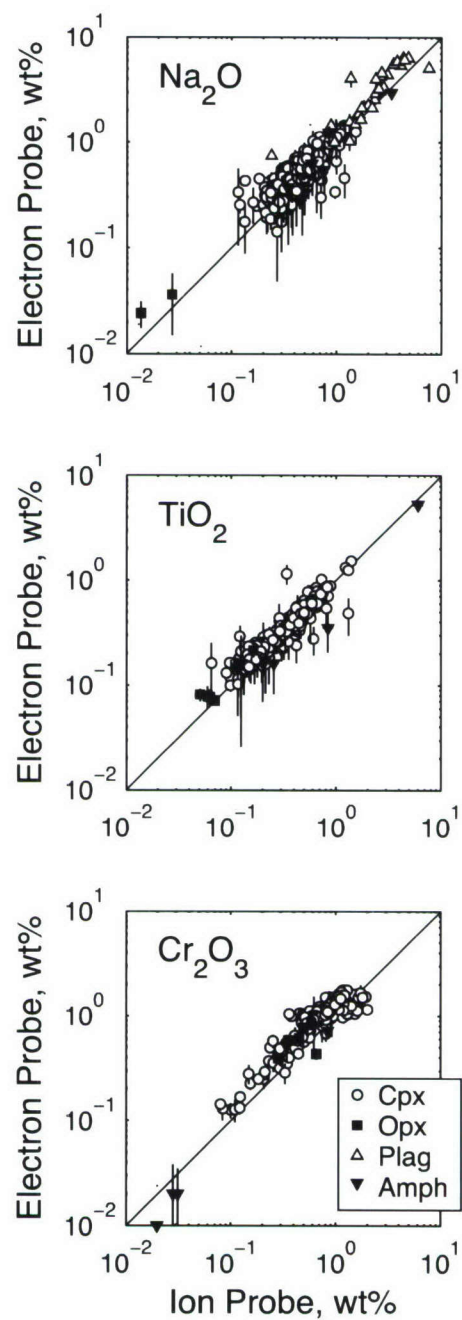


Figure 5.6: Comparison of Na₂O TiO₂ and Cr₂O₃ concentrations, in wt%, measured by ion probe and electron probe, for Cpx, Opx, Plag, and amphibole. Error bars for ion probe analyses are calculated based on from the average percent deviation of duplicate Cpx analyses (Table 5.10) and are generally smaller than the symbol size. Electron probe analyses are the averages of multiple analyses and vertical bars are the standard deviations. Variations are largest for pyroxenes, due to exsolution effects.

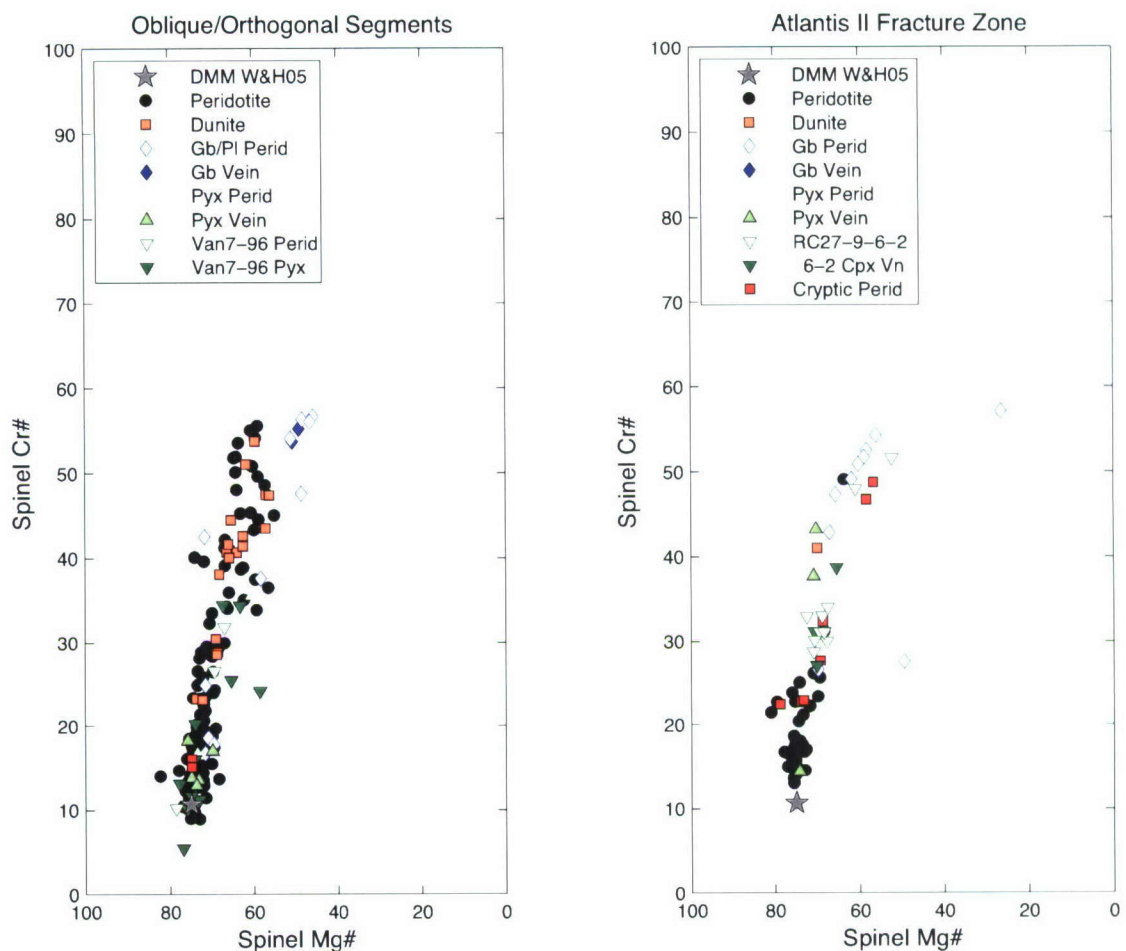


Figure 5.7: Variation of Spinel Cr# with Mg# in peridotites, veined peridotites, and dunites. The range of the residual peridotites is typical for abyssal peridotites (see Figure 5.18 for a comparison). Additional spinel data for these localities are from the peridotite compilation in Table 5.1 and unpublished data from H. J. B. Dick.

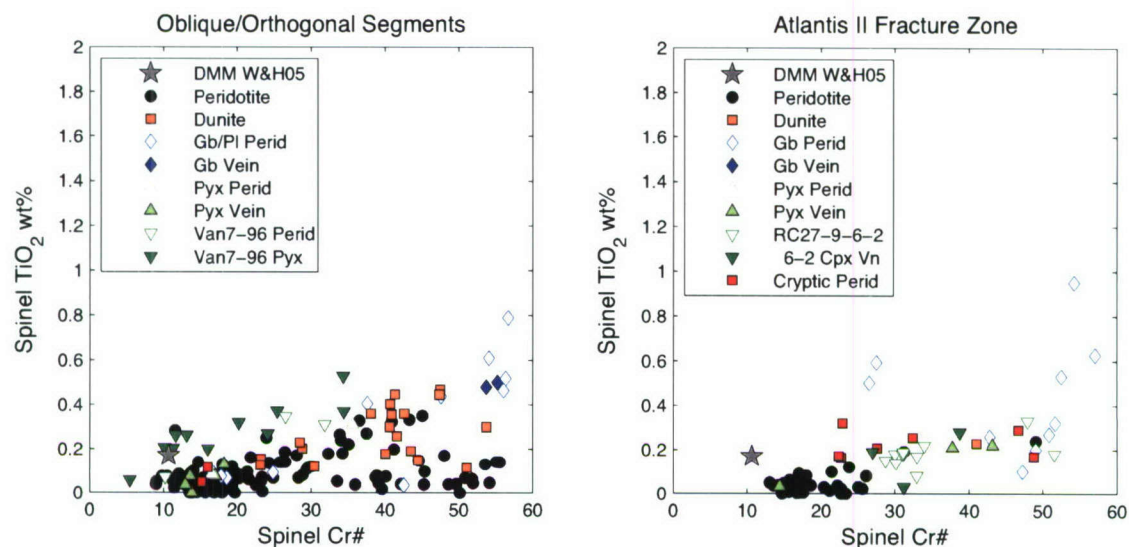


Figure 5.8: Variation of Spinel TiO₂ with Cr# among veined and unveined peridotites and dunites. TiO₂ concentrations greater than 0.2 wt% (e.g., greater than DMM) are indicative of melt crystallization (Dick and Bullen, 1984).

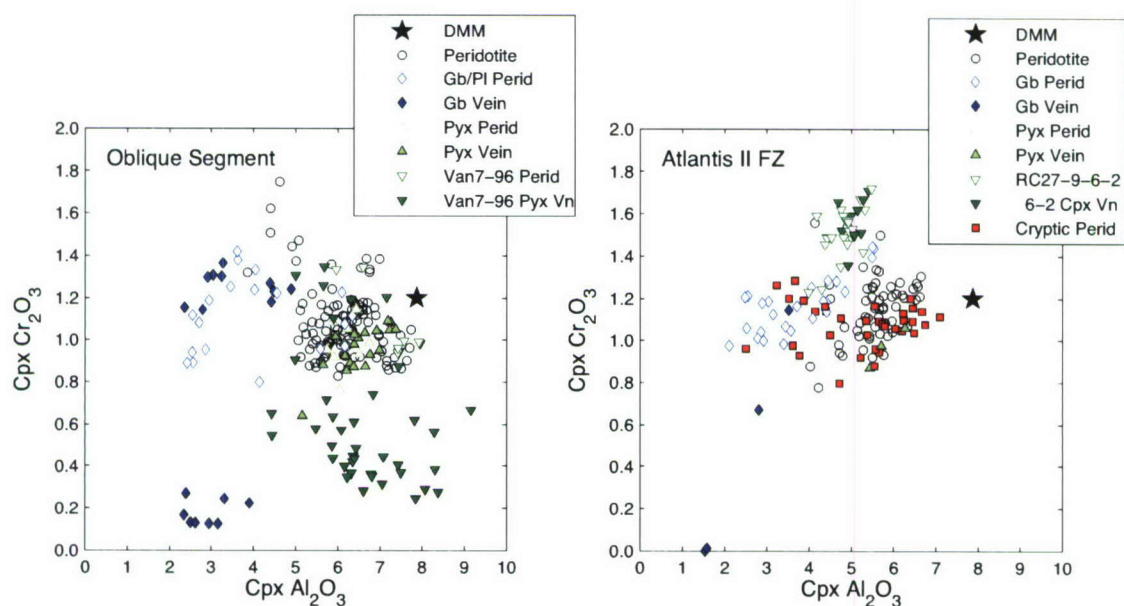


Figure 5.9: Variations of Cr₂O₃ with Al₂O₃ in Cpx from veined and unveined peridotites. The pyroxenite-veined peridotites from dredge Van7-96 and clinopyroxenite-veined peridotite sample RC27-9-6-2 were the focus of Chapter 4. Additional data for these localities are from Johnson et al. (1990), Johnson and Dick (1992) and H. J. B. Dick unpublished data.

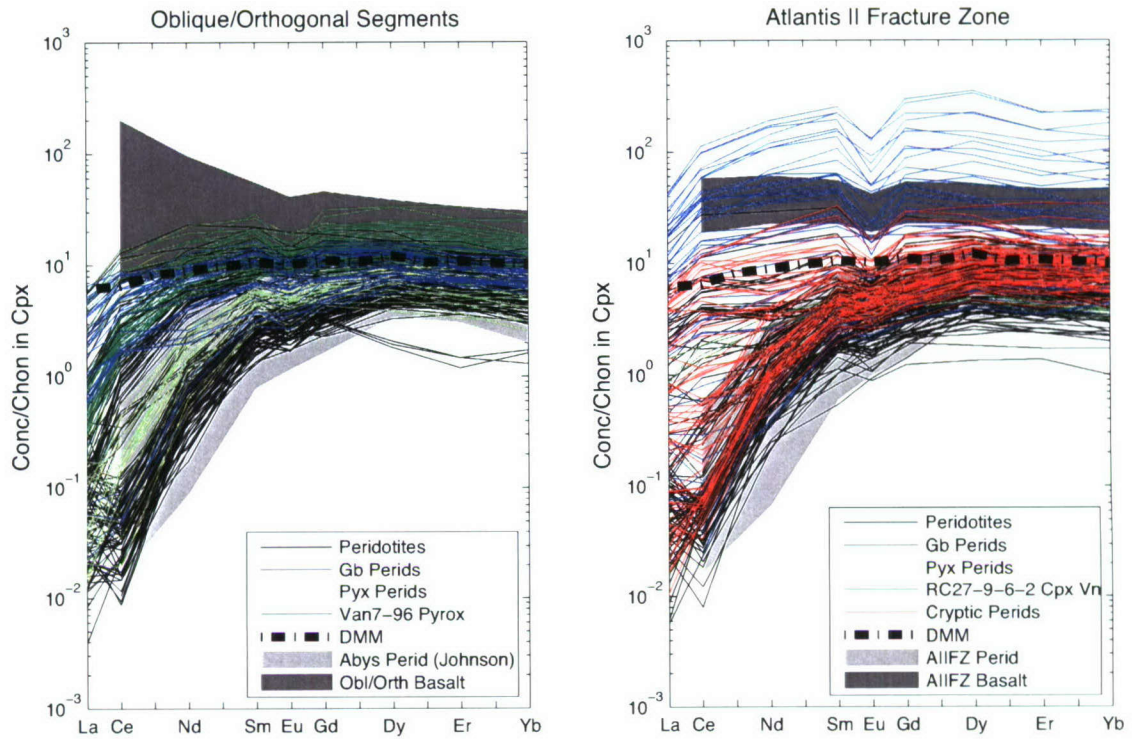


Figure 5.10: Variations in REEs among individual Cpx from veined and unveined peridotites. For the Oblique/Orthogonal Segments, the light grey field is abyssal peridotites away from hotspots (Johnson et al., 1990) and the dark grey field is Oblique/Orthogonal Segment basalts (Standish, 2006). For the Atlantis II Fracture Zone, the background fields are peridotites (light grey) and basalts (dark grey) along the fracture zone (Johnson and Dick, 1992).

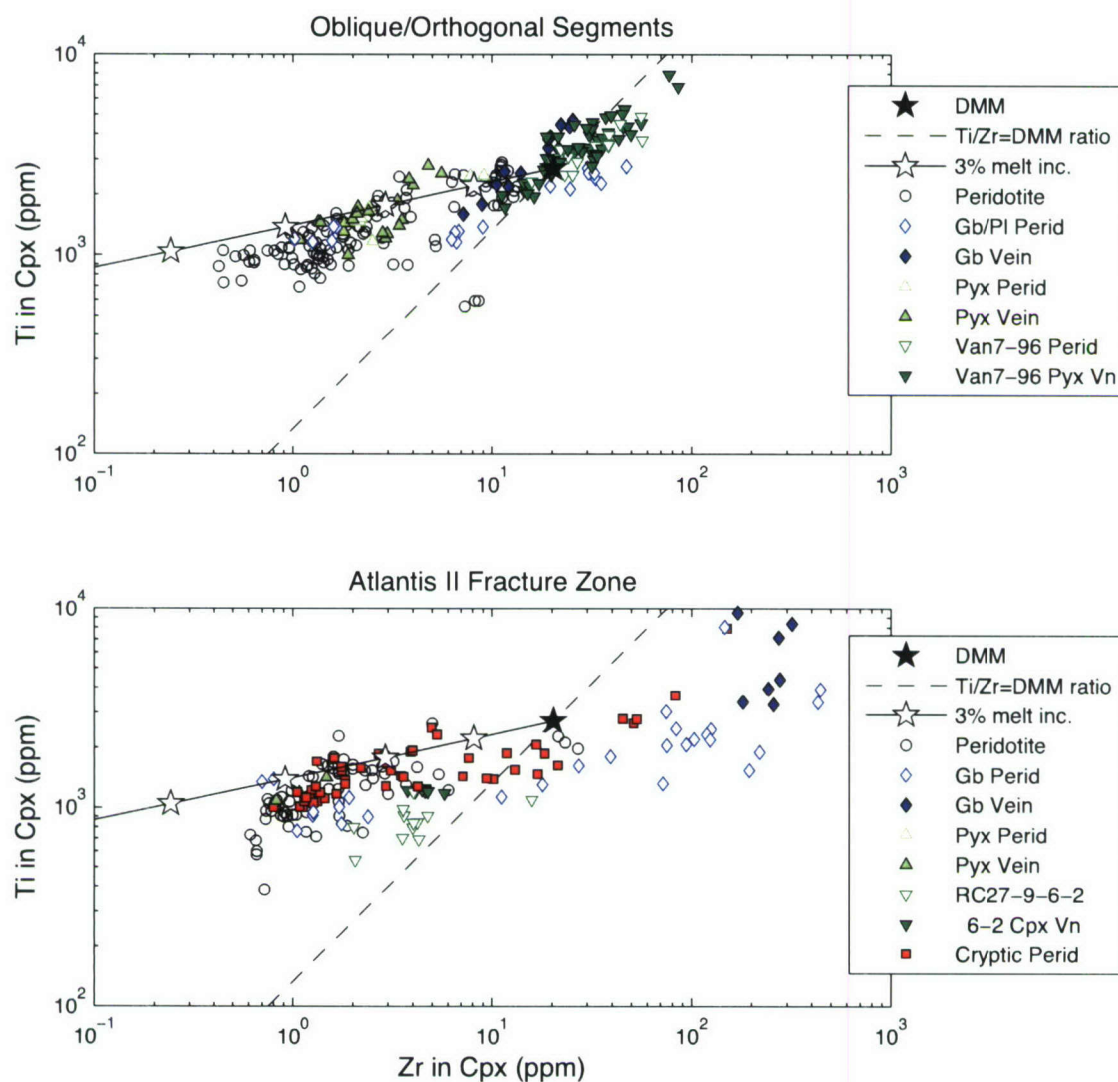


Figure 5.11: Co-variations of Ti and Zr in peridotite Cpx from the Oblique and Orthogonal Segments and the Atlantis II Fracture Zone. The melting trend, labeled for 3% fractional melting increments, was calculated using melt modes from Kinzler (1997) and partition coefficients from the compilation in Kelemen et al. (2003). DMM is from Workman and Hart (2005) and the dashed line indicates the variation of Ti with Zr at a constant DMM Ti:Zr ratio.

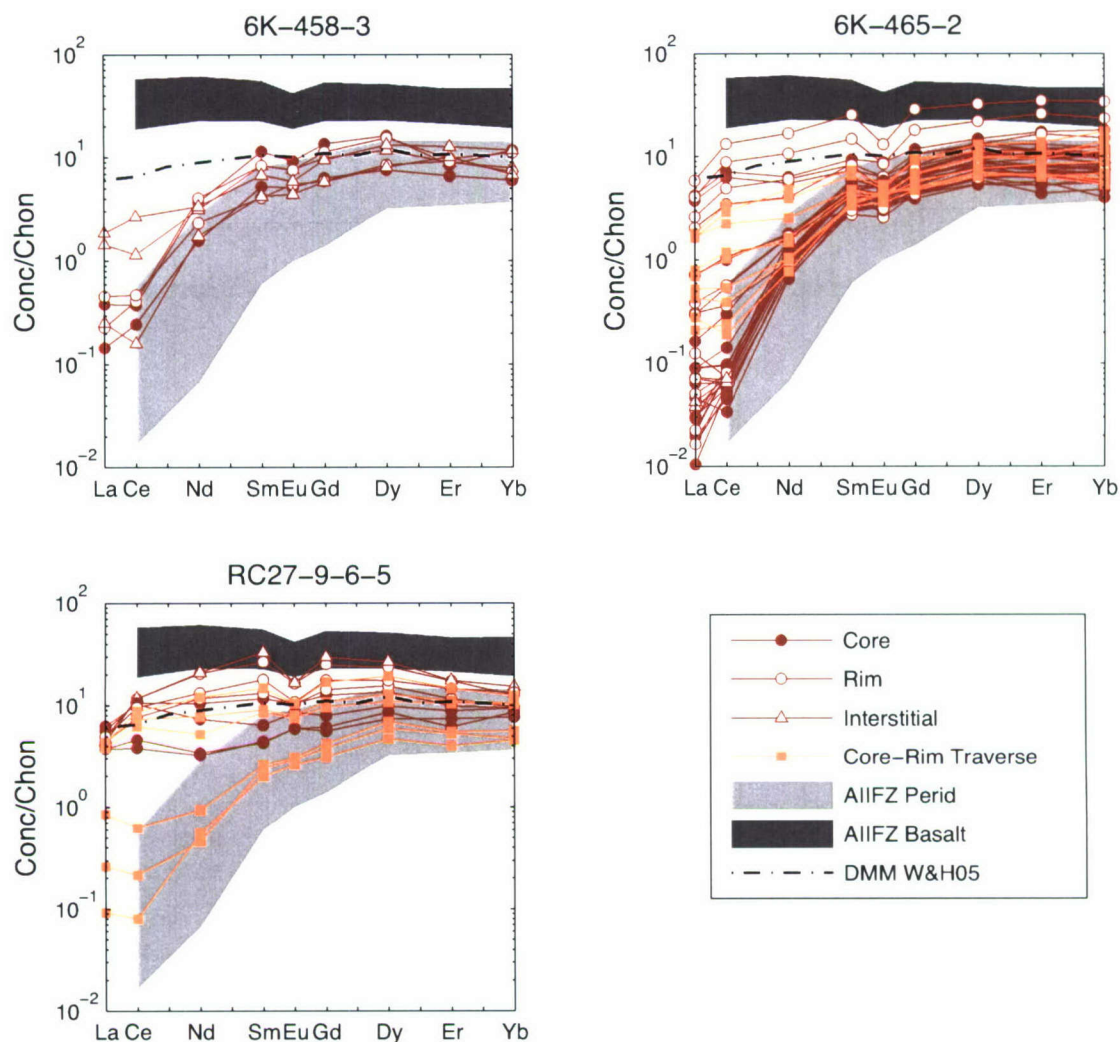


Figure 5.12: Variation of REEs in cryptically metasomatized peridotites from Atlantis II Fracture Zone. 6K-458-3 is from Atlantis Bank and the other two samples are from the ridge-transform intersection (Figure 5.3). The core-rim traverses in two of the plots are for individual grains. In 6K-465-2, this grain is located between depleted and enriched grains and is shown in Figure 5.15. In the other two samples, the location of the boundary between depleted and enriched compositions has not been fully mapped out. Background fields are from Johnson and Dick (1992).

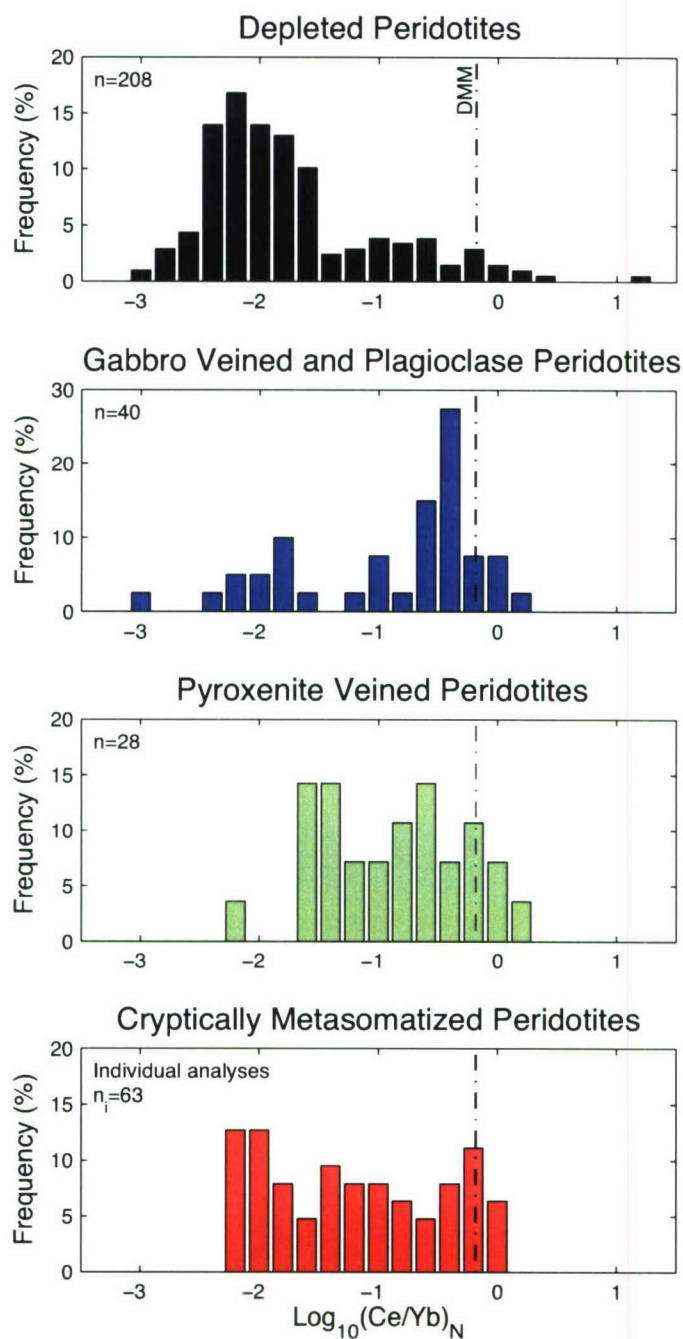


Figure 5.13: Histograms of Ce/Yb in Cpx, for the global abyssal peridotite dataset. Histograms are based on sample average compositions, with the exception of the cryptically metasomatized peridotites. For these three samples, individual analyses were used due to their large within-sample variations. The composition of DMM is from Workman and Hart (2005) and peridotite compositions are from this study, H. J. B. Dick unpublished data, and the compilation in Table 5.1.

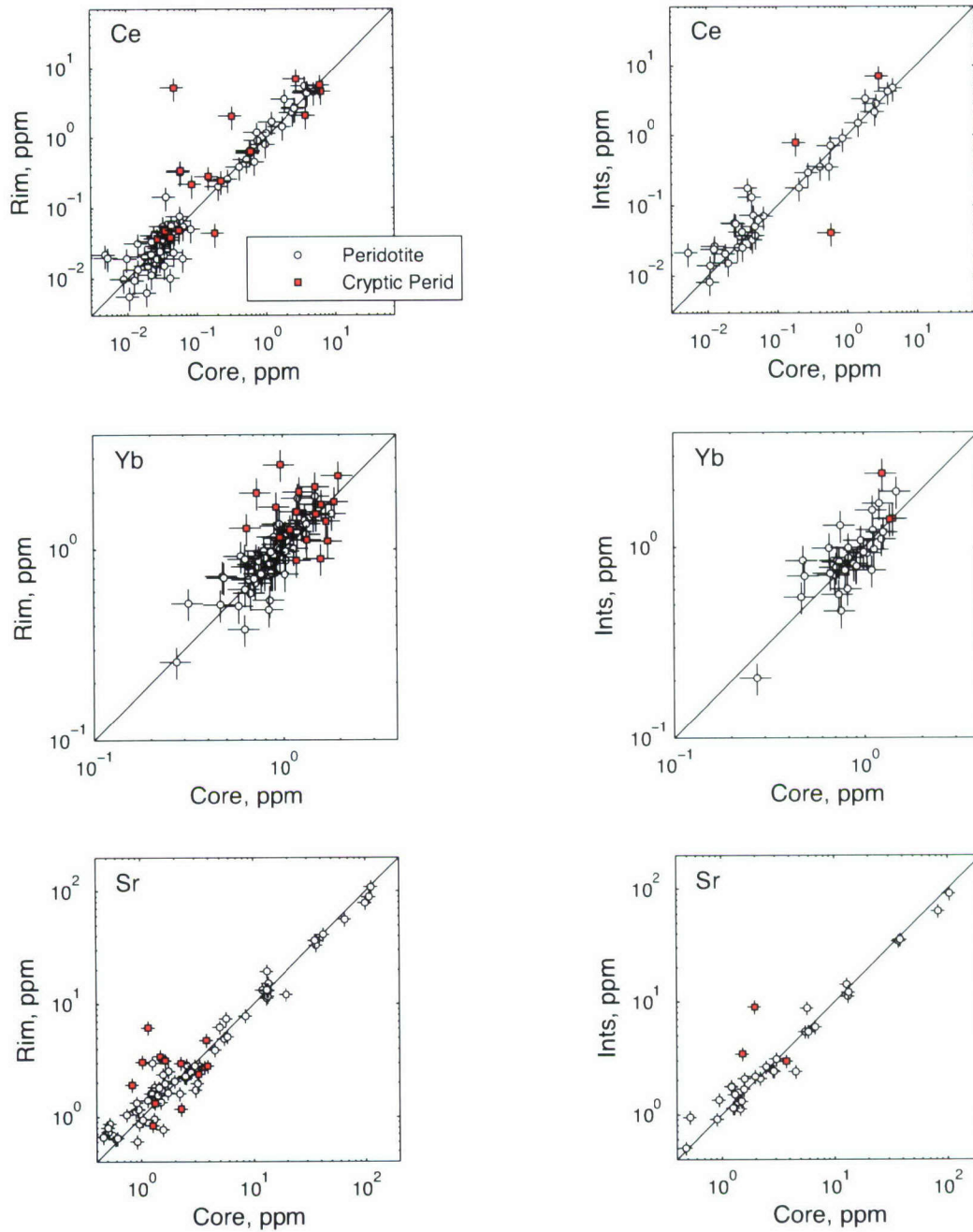


Figure 5.14: Comparisons of Cpx porphyroclast core and rim analyses and porphyroclast core and interstitial grain analyses, for vein-free peridotites. Cryptic refers to the 3 cryptically metasomatized peridotites. Interstitial grains are <1 mm diameter and typically have no exsolution lamellae. Error bars are based on the percent error among duplicate analyses (Table 5.10).

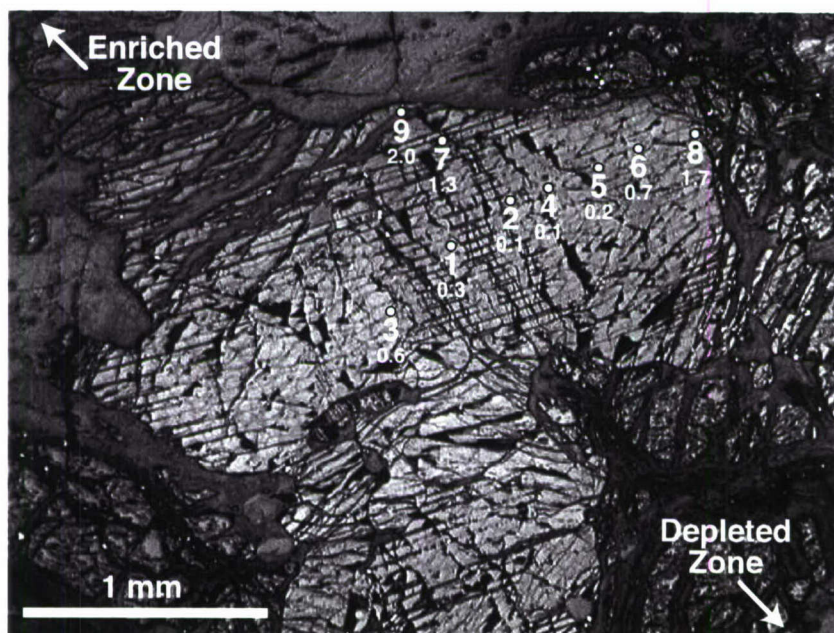


Figure 5.15: Photomicrograph of compositionally zoned Cpx6 from sample 6K-465-2. Numbering of analytical points corresponds to the numbering in Tables 5.6 and 5.10. Smaller numbers are Ce concentrations in ppm. The lowest trace element concentrations are centered around Point 4. Point 3, while appearing to be near the grain center, has high concentrations, suggesting that in the third dimension it is near the grain rim.

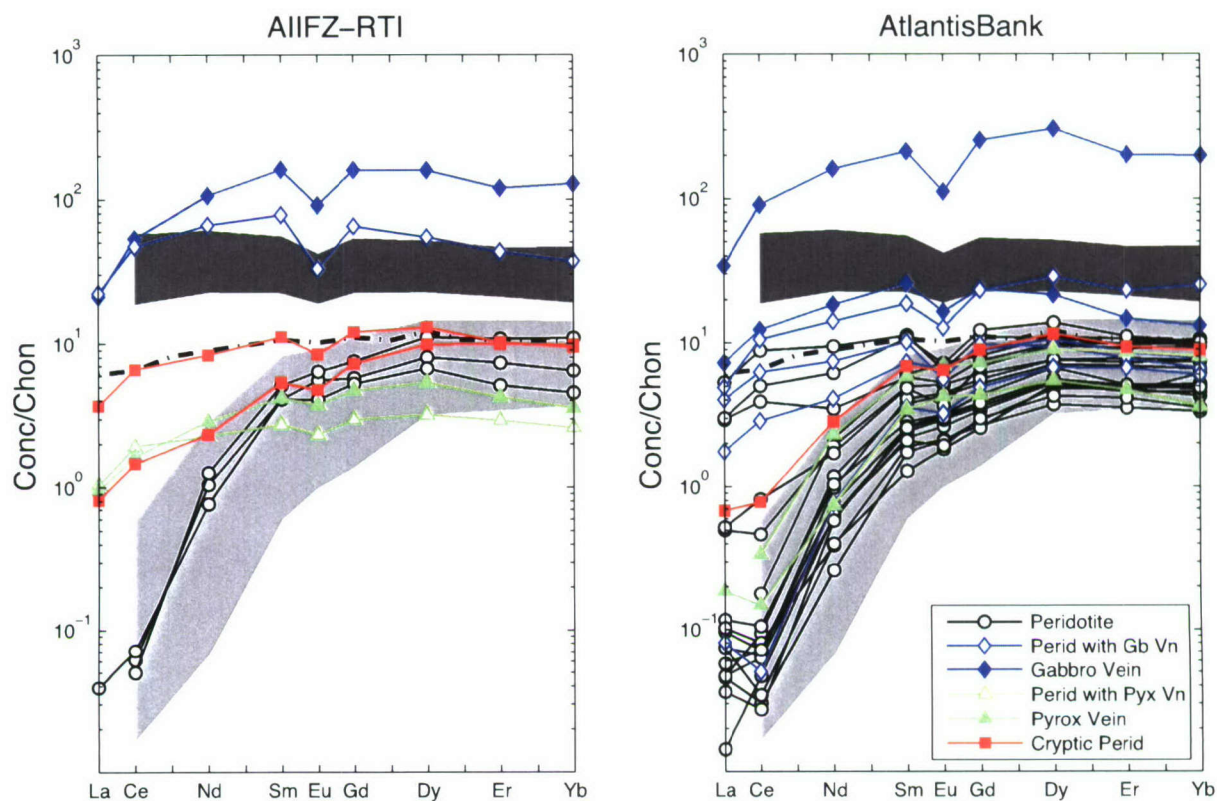


Figure 5.16: Atlantis II Fracture Zone REE variations at the ridge-transform intersection (RTI) and on Atlantis Bank. Peridotites, shown as sample averages, have large compositional variations at both locations. The sample averages for the chromatographic peridotites (red), when plotted as individual analyses (Figure 5.12) extend over almost the entire range of other peridotites from these locations. Background fields are for fracture zone peridotites (light grey) and basalts (dark grey) from Johnson and Dick (1992). Additional individual samples for the two locations are from Johnson (1990) and unpublished data from H. J. B. Dick.

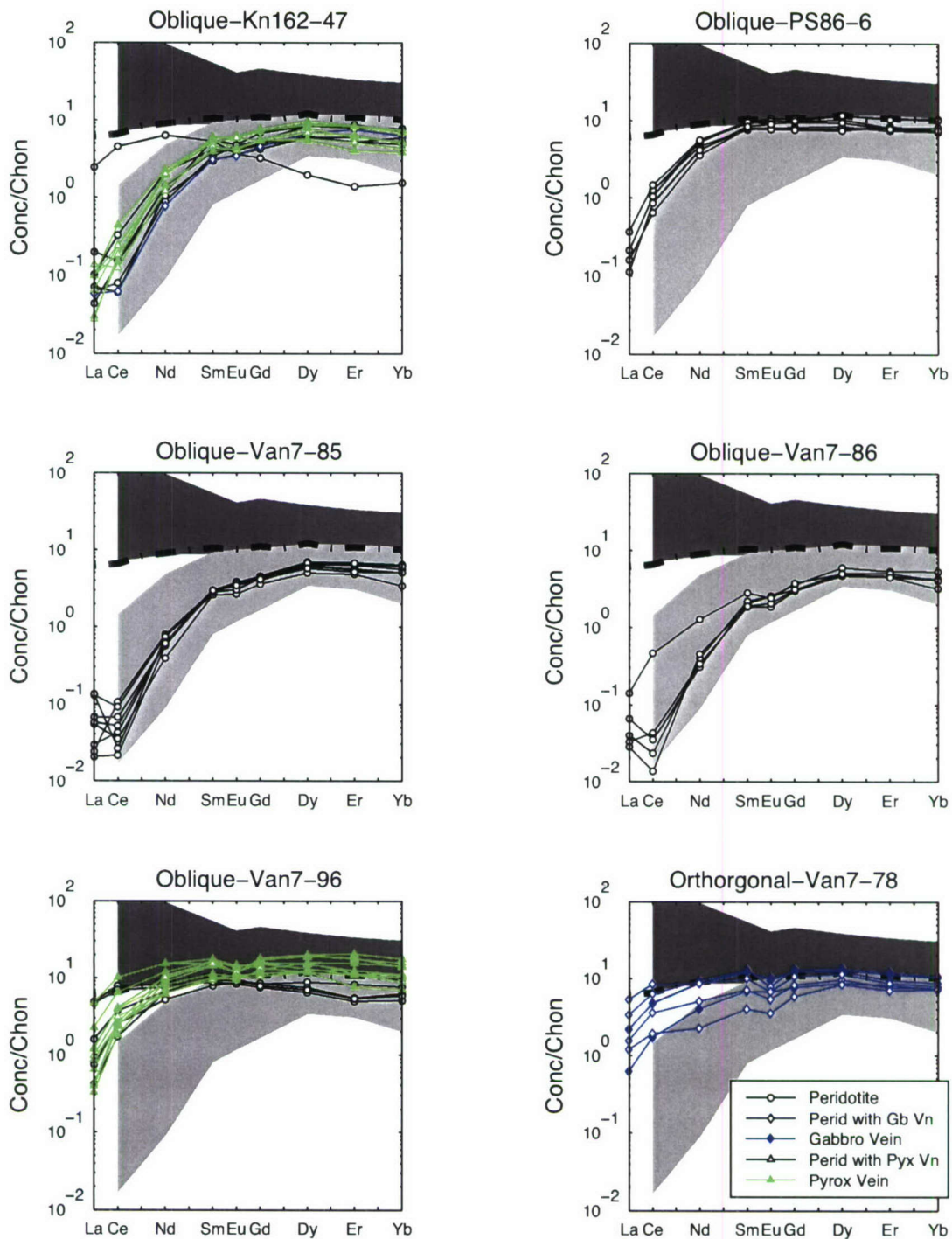


Figure 5.17: Variation in REEs for dredges from the Oblique and Orthogonal Segments. Additional data for PS86-6 from Johnson et al. (1990) and unpublished data from H. J. B. Dick. Background fields are for abyssal peridotites (light grey, Johnson et al., 1990) and Oblique/Orthogonal Segment basalts (dark grey, Standish, 2006).

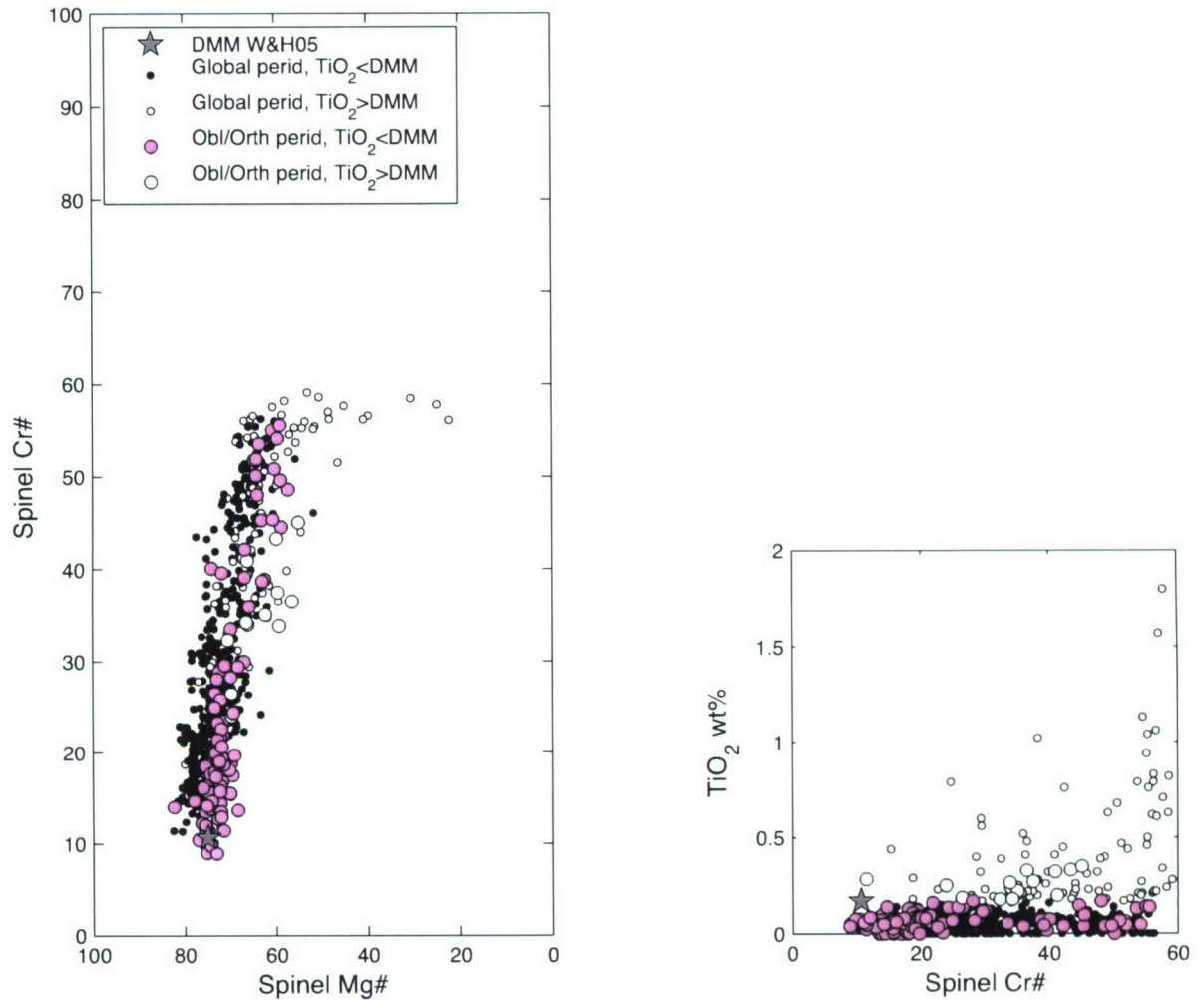


Figure 5.18: Variation of spinel compositions in the global peridotite dataset (small symbols) and the Oblique/Orthogonal Segments (large symbols). Spinel with TiO₂ concentrations greater than DMM spinel (0.17 wt% Workman and Hart, 2005) are identified by unfilled symbols. These spinels reflect melt addition to the peridotite (Dick and Bullen, 1984).

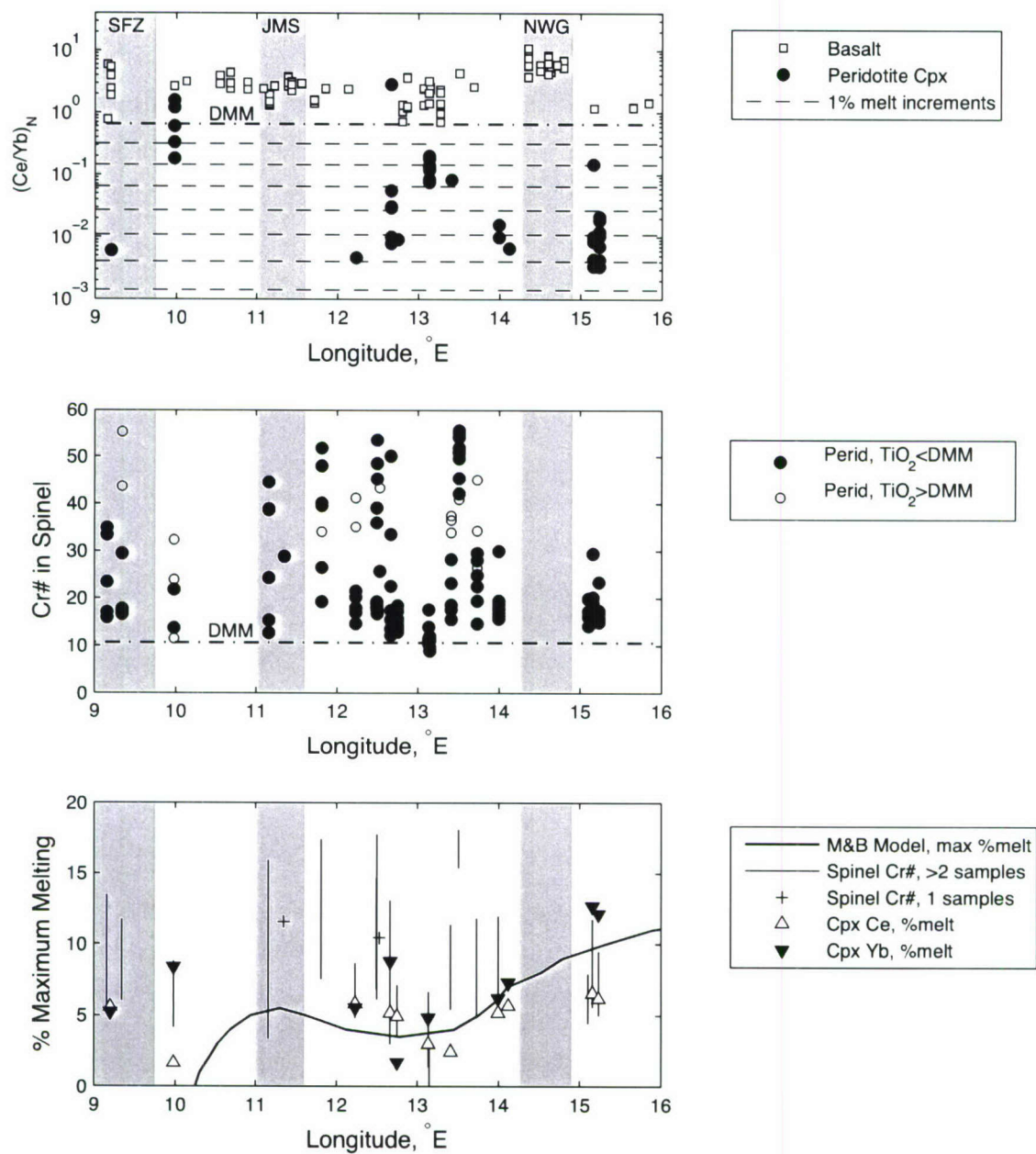


Figure 5.19: Variations along the Oblique Segment in trace element compositions, Spinel Cr#, and predicted degree of melting. Degree of melting from Ce, Yb and Ce/Yb in Cpx are calculated for fractional melting using parameters discussed in the text. Degree of melting from spinel Cr# is calculated using the empirical relationship from Hellebrand et al. (2001). The predicted maximum degree of melting line is from the model by Montési et al. (2006), which accounts for variations in ridge thermal structure due to changes in ridge geometry along the Oblique Segment. The grey shaded areas indicate the location of the Shaka Fracture Zone (SFZ), Joseph-Mayes Seamount (JMS), and Narrowgate (NWG). Basalt data are from Standish (2006).

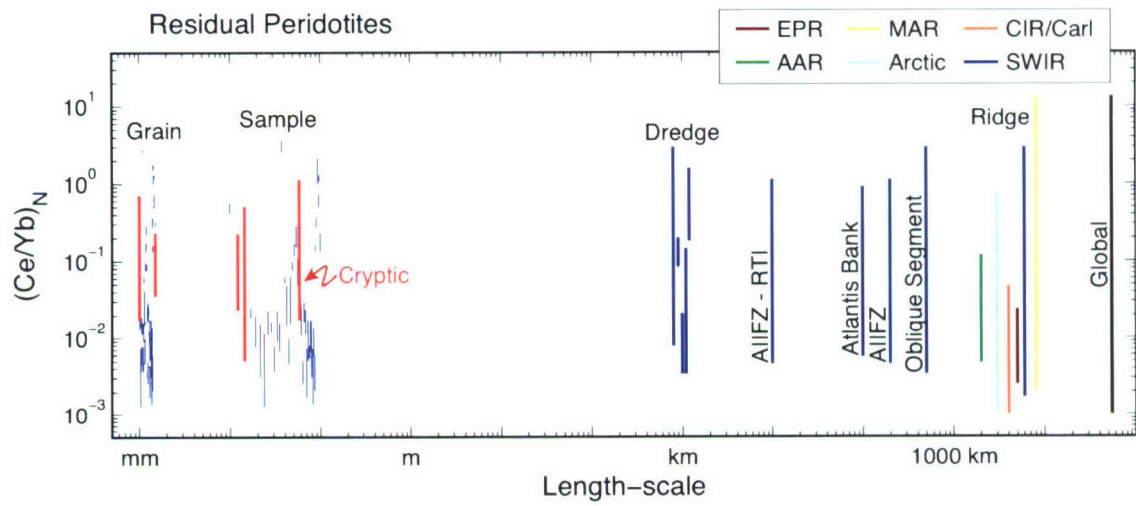


Figure 5.20: Length-scales of residual peridotite variability, based on the Ce/Yb range among Cpx. Data for the smallest length-scales are from this study; at the ridge-scale, data are from the compilation summarized in Table 5.1.

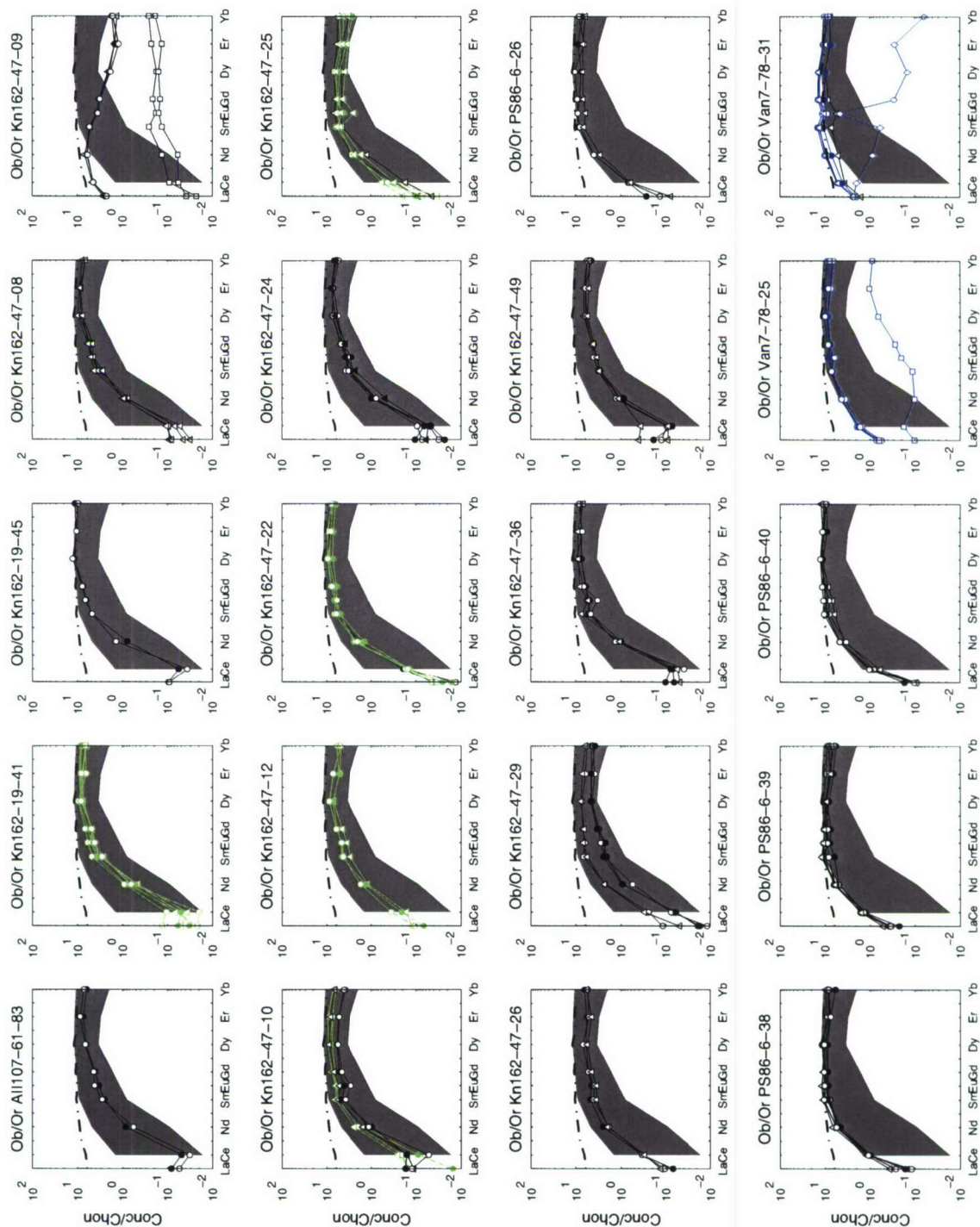


Figure 5.A1: REE plots of each analysis, by sample. See last page of figure for legend.



Figure 5.A1: (Continued)

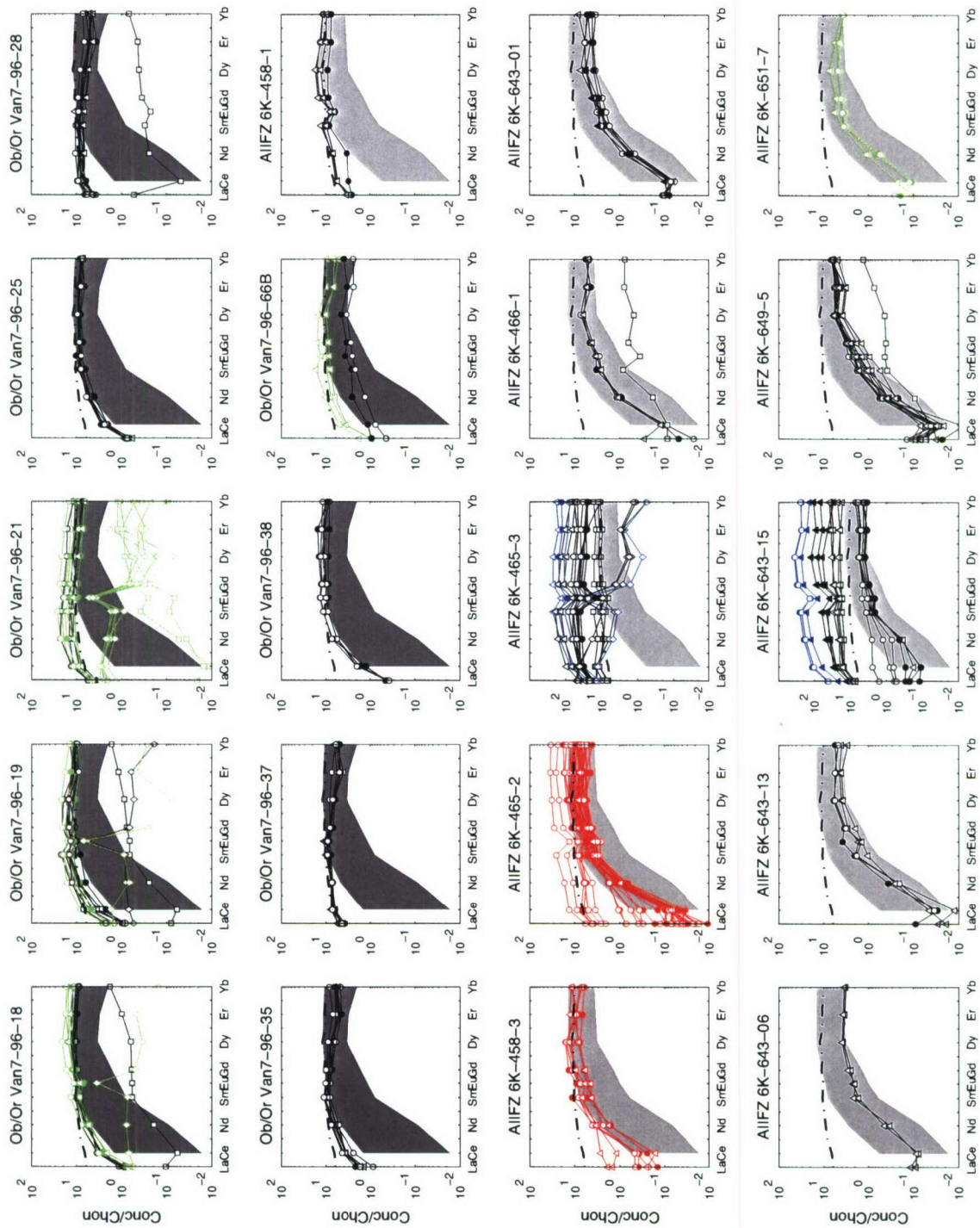


Figure 5.A1: (Continued)

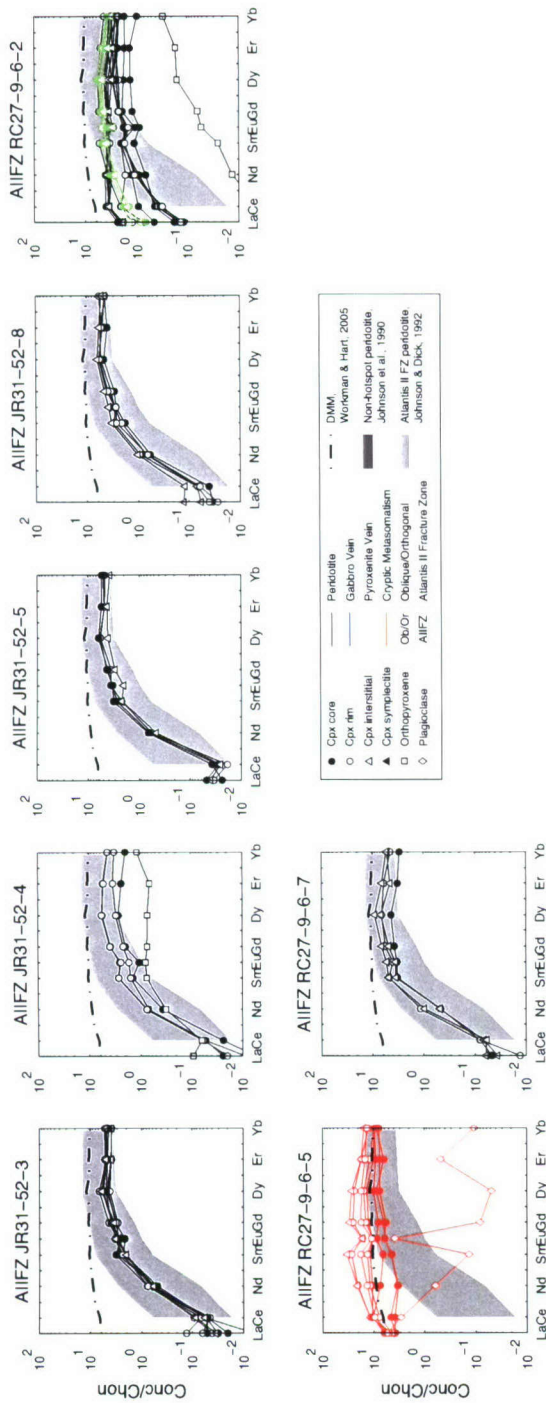


Figure 5.A1: (Continued)

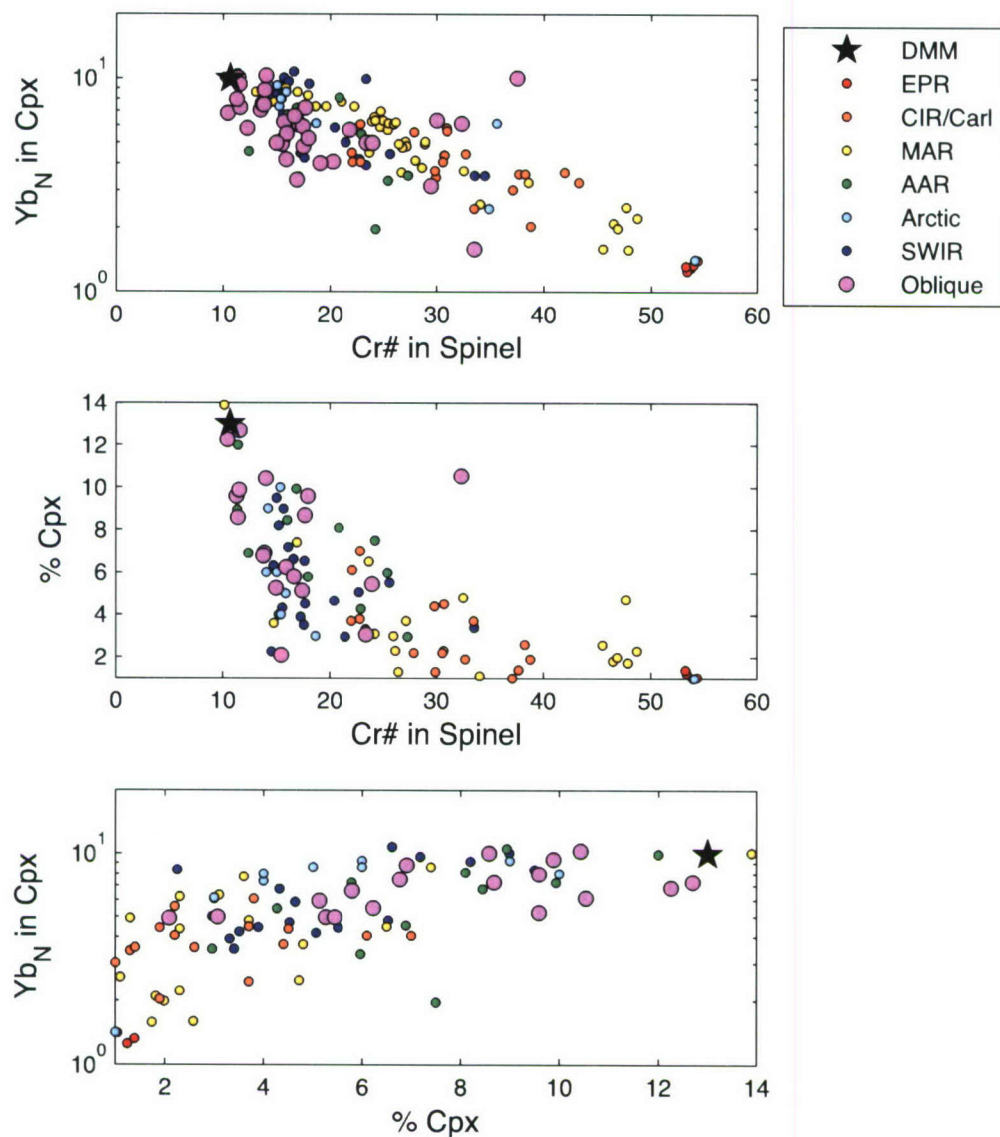


Figure 5.B1: Co-variation of Spinel Cr#, Yb_N in Cpx and modal Cpx. The variation of spinel Cr# with Cpx HREE (Dy, Er and Yb) has been used to calibrate a degree of melting relationship for spinel Cr#. As modal Cpx is also an indicator of degree of melting in peridotite, we compare the co-variation among all three parameters.

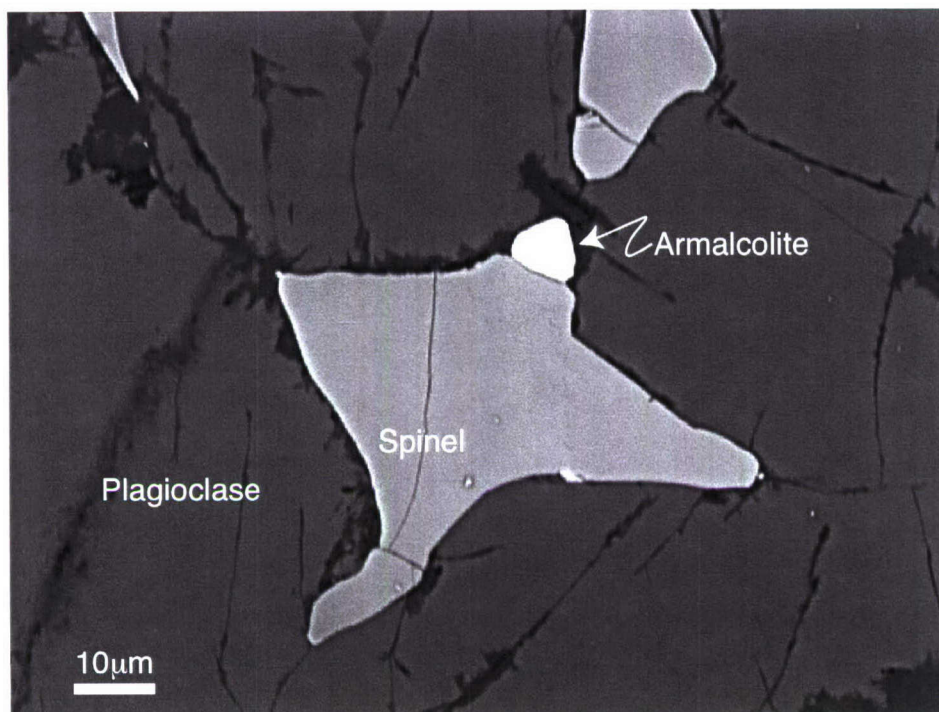


Figure 5.C1: Armalcolite, associated with spinel and plagioclase in pyroxenite vein (Van7-96-14).

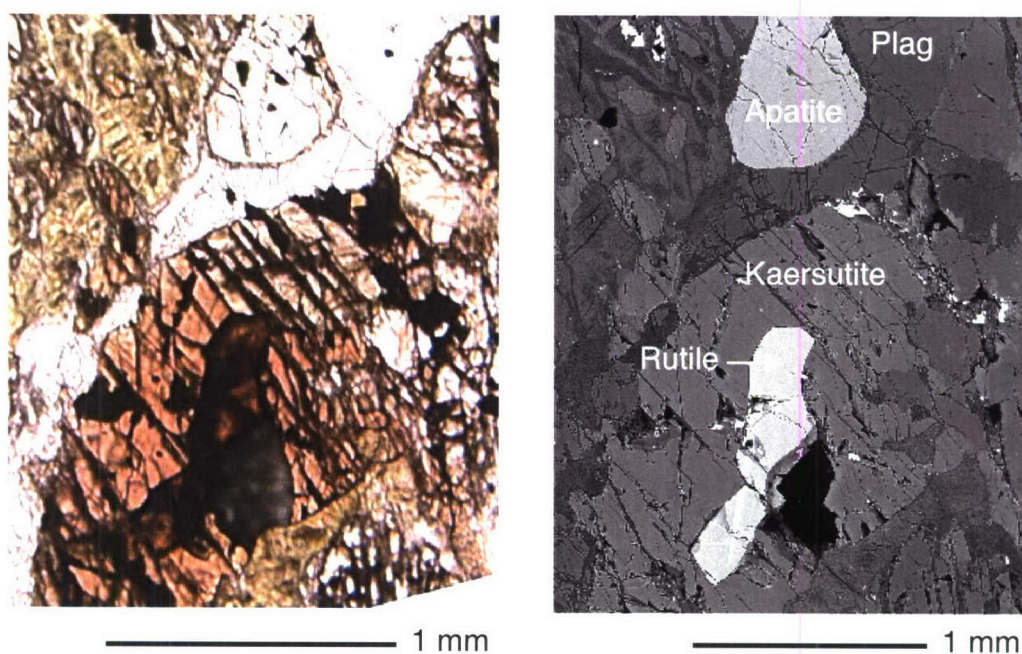


Figure 5.D1: Photomicrograph and backscattered electron image of a vein in sample 6K-465-3 from Atlantis II Fracture Zone. Compositions of kaersutite, apatite, rutile and plagioclase in the vein are given in Table 5.D1. Bright spots in the backscattered image are remnants of the gold coat from ion probe analyses.

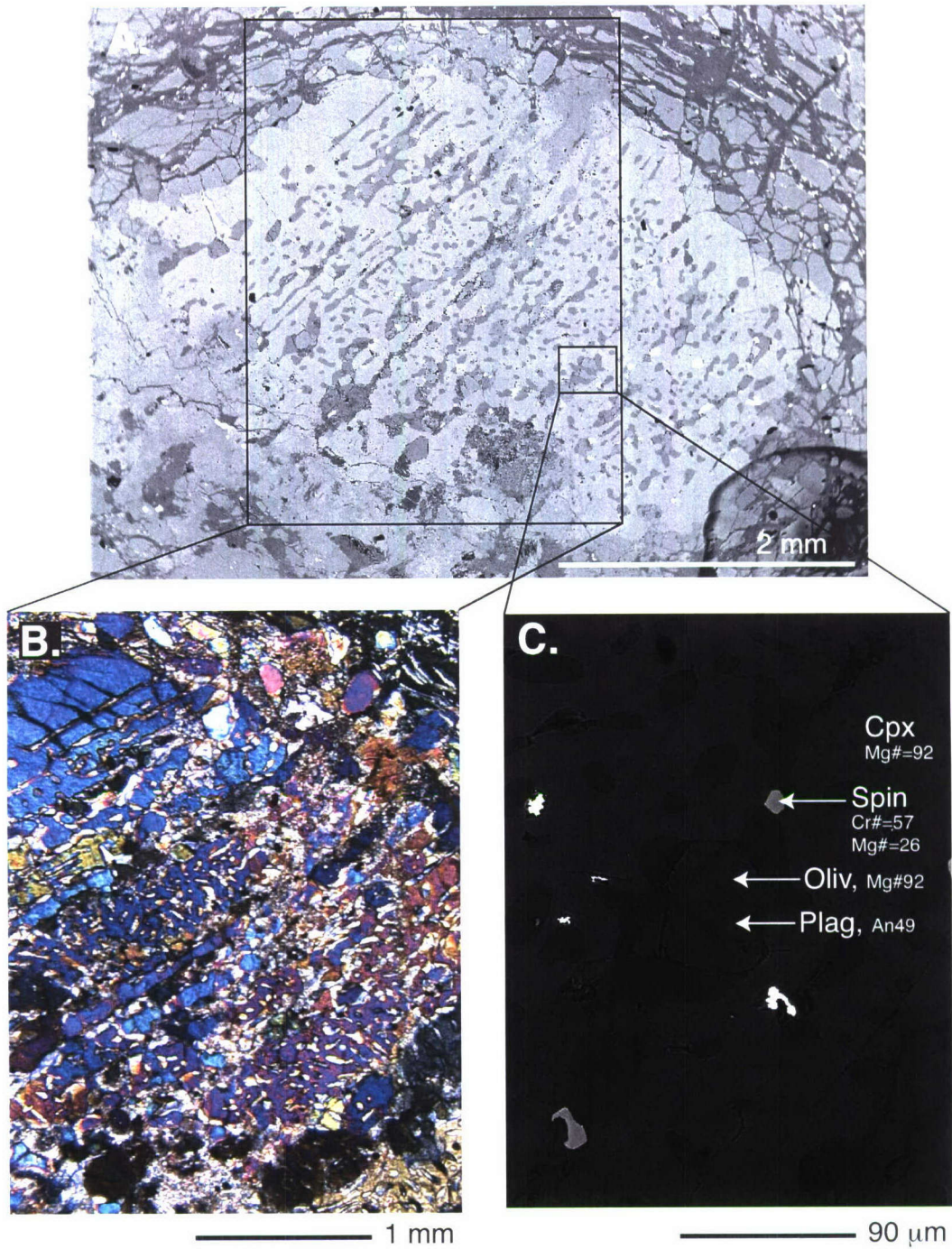


Figure 5.D2: Photomicrograph and backscattered electron images of unusual exsolution in gabbro-veined sample 6K-465-3 from Atlantis II Fracture Zone. Major and trace element analyses of Cpx12 and exsolved plagioclase, spinel and olivine are presented in Table 5.D1, along with analyses for similar occurrences in three other Cpx porphyroclasts. Bright spots in the backscattered images are remnants of the gold coat from ion probe analyses.

Table 5.1: Summary of global peridotite compilation.

Ridge	Location	F.S.R. ^a	Lithology	Reference
East Pacific Rise				
EPR	Terevaka FZ	157	Harz, Pl dun, Gabbro vein, Cpx vein	Constantin et al. (1995)
EPR	Garrett FZ	150	Harz, Pl perid, Dun, Pl dun, Wehlite, Gabbro vein	Niu and Hékinian (1997), Constantin (1999)
EPR	Hess Deep	128	Harz, Dun, Pl dun	Dick and Natland (1996)
Mid-Cayman Rise				
MCR	Cayman Trough	71-73	Lherz, Harz, Pl lherz, Pl harz, Dun	Dick et al. (1984), H. J. B. Dick (unpublished)
Central Indian Ridge				
CIR	Green Rock Hill	50	Harz	Hellebrand et al. (2002)
CIR	Marie Celeste FZ	41	Harz, Pl harz, Dun, Ol-webst	Dick et al. (1984), Hellebrand et al. (2002)
CIR	Argo FZ	38	Lherz, Harz, Pl harz	Dick et al. (1984), Hellebrand et al. (2002)
CIR	Vema II FZ	37	Harz, Gabbro vein	Dick et al. (1984), Hellebrand et al. (2002)
CIR	Rift at 12°S, 66°E	36	Harz, Dun	Dick et al. (1984), Hellebrand et al. (2002)
Carlsberg Ridge				
Carl	Owen FZ	26	Lherz, Harz, Pl lherz, Pl harz, Dun	Hamlyn and Bonatti (1980)
Mid-Atlantic Ridge				
MAR	22°S FZ	36	Lherz, Harz, Pl harz	Dick et al. (1984)
MAR	Romanche FZ	33	Lherz, Harz, Pl lherz, Pl harz, Pl Ol-webst	Dick et al. (1984), Bonatti et al. (1993), Seyler and Bonatti (1997), Tartarotti et al. (2002)
MAR	St Paul FZ	32	Perid	Bonatti et al. (1992)
MAR	2°-4°N Area	31	Perid	Bonatti et al. (1992)
MAR	Doldrums FZ	29	Perid	Bonatti et al. (1992)
MAR	Mercurius FZ	28	Perid	Bonatti et al. (1992)
MAR	Vema FZ	28	Lherz, Harz, Pl perid, Gabbro vein	Prinz et al. (1976), Cannat and Seyler (1995), Seyler and Bonatti (1997), Brunelli et al. (2006)
MAR	43°N Area	24	Harz	Dick et al. (1984)
MAR	Kurchatov FZ	24	Harz	Dick and Bullen (1984)
MAR	Gibbs FZ	23	Lherz, Gabbro vein	Seyler and Bonatti (1997)
MAR	Oceanographer FZ	22	Harz	Michael and Bonatti (1985)
MAR	Marathon FZ	16	Perid	Bonatti et al. (1992)
MAR	Rift at 21°N, 45°W	15	Lherz, Harz, Dun, Ol-webst	Fujii (1990), Juteau et al. (1990), Komor et al. (1990)
MAR	15°20'N Area	15	Harz, Dun	Bonatti et al. (1992), Seyler et al. (2007), H. J. B. Dick (unpublished)
MAR	Famous FZ	14	Harz	Dick and Bullen (1984)
MAR	Kane FZ	14	Lherz, Harz, Pl harz, Pl Ol-webst	Dick et al. (1984), Michael and Bonatti (1985), Lugué et al. (2003)
MAR	MARK Area	14	Lherz, Harz, Dun, Ol-webst	Ross and Elthon (1997), Stephens (1997), Kempton and Stephens (1997)
American Antarctic Ridge				
AAR	Bullard FZ	17	Lherz, Harz, Pl lherz, Pl harz	Dick (1989), Johnson et al. (1990)
AAR	Vulcan FZ	17	Lherz, Harz, Pl lherz	Dick (1989), Johnson et al. (1990)
Knipovitch Ridge				
Knip	Spitzbergen FZ	14	Perid	Jaroslow et al. (1996)
Lena Trough				
LT	Lena Trough	14	Lherz, Harz	Hellebrand and Snow (2003)
Gakkel Ridge				
GAK	Gakkel Ridge	11-13	Lherz, Harz, Ol-webst	Hellebrand et al. (2005)

^a Full spreading rate in mm/yr (or km/Ma).

Table 5.1: Global compilation Cont.

Ridge	Location	F.S.R. ^a	Lithology	Reference
Southwest Indian Ridge				
SWIR	Speiss Ridge	19	Harz, Dun	H. J. B. Dick (unpublished)
SWIR	Andrew Bain FZ	15	Lherz, Harz, Pl harz, Dun, Webst	Johnson et al. (1990)
SWIR	Discovery II FZ	15	Lherz, Harz	Dick et al. (1984), Johnson et al. (1990)
SWIR	Gallieni FZ	15	Lherz, Harz, Pl harz	Dick et al. (1984), Jaroslow et al. (1996)
SWIR	Indomed FZ	15	Lherz	Johnson et al. (1990)
SWIR	Melville FZ	15	Lherz, Harz, Pl lherz	Dick et al. (1984)
SWIR	Prince Edward FZ	15	Lherz, Harz, Pl harz	Johnson et al. (1990), Jaroslow et al. (1996)
SWIR	SWIR 52°-68°E	14-15	Lherz, Harz	Seyler et al. (2003), Seyler et al. (2004)
SWIR	Atlantis II FZ	14	Lherz, Harz, Pl perid, Dun, Webst, Ol-webst, Cpx vein, Gabbro vein	Johnson and Dick (1992), Jaroslow et al. (1996), Lee (1997), Salters and Dick (2002), Kumagai et al. (2003), Coogan et al. (2004), H. J. B. Dick (unpublished), This study
SWIR	Bouvet FZ	14	Harz, Pl harz, Dun	Dick et al. (1984), Johnson et al. (1990)
SWIR	Islas Orcadas FZ	14	Lherz, Harz, Pl lherz, Pl harz, Ol-webst	Dick (1989), Johnson et al. (1990), Snow (1993)
SWIR	Orthogonal Seg.	14	Harz, Pl perid, Dun, Ol-webst, Gabbro vein	This Study
SWIR	Rift at 26°S, 67°E	14	Lherz	Dick and Bullen (1984), Dick et al. (1984)
SWIR	Rift at 54°S, 7°E	14	Harz	Dick et al. (1984)
SWIR	Shaka FZ	14	Perid	Johnson et al. (1990), Snow (1993), Jaroslow et al. (1996), This Study
SWIR	Oblique Segment	8-14	Lherz, Harz, Pl perid, Dun, Webst, Ol-webst, Gabbro vein	Johnson et al. (1990), Salters and Dick (2002), Dantas et al. (2007), H. J. B. Dick (unpublished), This study

Table 5.2: Dredge locations

Dredge/Dive	Dredge description	Wt kg	Lat °S	Long °E	Depth (m)	FSR ^a (mm/yr)	Location
SWIR 9°–25°E							
R/V Atlantis II, 1980							
AII107-61	Peridotites, perid mylonites	48	-53.42	9.20	4063	14	Shaka FZ
R/V Polarstern, 1986							
PS86-6	Peridotite, basalt, gabbro veined perid	187	-52.35	13.13	3332	9	Oblique
R/V Knorr, 2000							
Kn162-19	Basalt, peridotite, pyroxenite, diabase	81	-53.05	21.38	3992	14	Orthogonal
Kn162-47	Peridotite, pyroxenites, plag perid	144	-52.56	12.66	4361	8	Oblique
R/V Melville, 2003							
Van7-78	Peridotite, dunite, gabbros/veined perid, plag perid, basalt	192	-52.38	16.64	3901	14	Orthogonal
Van7-85	Peridotite, dunite, diabase	90	-52.25	15.23	4190	12	Oblique
Van7-86	Dunite, peridotite	146	-52.14	15.16	3738	12	Oblique
Van7-96	Perid/pyrox/bas cataclastites, perid, pyrox, diabase	81	-53.14	9.98	3134	12	Oblique
ATLANTIS II FRACTURE ZONE							
R/V Conrad, 1986							
RC27-9-6	Peridotite, dunite	37	-31.92	57.18	3930	14	AIIFZ RTI ^b
R.R.S. James Clark Ross, 1998							
JR31-52	Peridotite, diabase, gabbro	125	-32.56	57.17	3200	14	Atlantis Bank
R/V Yokosuka, Shinkai 6500 submersible, 1998							
6K-458-1		3	-32.70	57.13	4609	14	Atlantis Bank
6K-458-3		8	-32.70	57.13	4609	14	Atlantis Bank
6K-465-2		10	-31.95	57.13	4988	14	AIIFZ RTI
6K-465-3		6	-31.95	57.13	4899	14	AIIFZ RTI
6K-466-1		3	-32.64	57.14	4714	14	Atlantis Bank
R/V Yokosuka, Shinkai 6500 submersible, 2001							
6K-643-1		9	-32.91	57.20	3396	14	Atlantis Bank
6K-643-6		2	-32.91	57.21	2974	14	Atlantis Bank
6K-643-13		13	-32.91	57.21	2712	14	Atlantis Bank
6K-643-15		13	-32.91	57.21	2605	14	Atlantis Bank
6K-649-5		3	-32.91	57.21	2564	14	Atlantis Bank
6K-651-7		2	-32.74	57.15	4088	14	Atlantis Bank

^a Full spreading rate. For Oblique Segment dredges, the effective full spreading rate is given, calculated following the method of Abelson and Agnon (1997) and using the measured angle of obliquity of the ridge axis to the plate spreading direction. For the Atlantis II Fracture Zone, the half spreading rate to the south is 8.5 mm/yr, due to asymmetrical spreading (Hosford et al., 2003).

^b Ridge-transform intersection.

Table 5.3: Sample descriptions and modal analyses

Sample	Location	Description	Ol	Opx	Cpx	Sp	Pl	Sum	Pts ^a	
SWIR 9°–25°E										
All107-61-83	Shaka FZ	Lherzolite mylonite	72.4	19.8	5.8	1.4	0.0	99.4 ^b	3124	
PS86-6-26	Oblique	Lherzolite	60.7	26.2	12.3	0.9	0.0	100.0 ^c		
PS86-6-38	Oblique	Lherzolite	59.5	29.9	9.6	1.0	0.0	100.0		
PS86-6-39	Oblique	Lherzolite	69.7	20.9	8.7	0.8	0.0	100.0 ^c		
PS86-6-40	Oblique	Lherzolite	66.7	23.4	8.6	1.4	0.0	100.0 ^c		
Kn162-19-41	Orthogonal	Olivine websterite	82.4	16.6	0.4	0.7	0.0	100.0 ^d	2132	
Kn162-19-45	Orthogonal	Harzburgite								
Kn162-47-08	Oblique	Harzburgite	4.0	50.0	40.0	6.0	0.0	100.0 ^e		
Kn162-47-09	Oblique	Lherzolite								
Kn162-47-10	Oblique	Websterite vein in peridotite, mylonite								
Kn162-47-12	Oblique	Websterite vein in peridotite, mylonite								
Kn162-47-22	Oblique	Ol-websterite vein in harzburgite	12.0	35.0	45.0	8.0	0.0	100.0 ^e	2506	
Kn162-47-24	Oblique	Plag peridotite	0.0	32.0	65.0	3.0	0.0	100.0 ^e		
Kn162-47-25	Oblique	Websterite vein in peridotite								
Kn162-47-26	Oblique	Harzburgite	68.6	27.5	3.1	0.9	0.0	100.0		
Kn162-47-29	Oblique	Peridotite mylonite								
Kn162-47-36	Oblique	Harzburgite mylonite								
Kn162-47-49	Oblique	Peridotite protomylonite								
Van7-78-25	Orthogonal	Dunite with gabbro vein								
Van7-78-31	Orthogonal	Plag peridotite with gabbro vein	67.4	29.2	2.1	1.2	0.2	100.0		
Van7-78-36	Orthogonal	Plag peridotite with gabbro vein								
Van7-78-40	Orthogonal	Plag harzburgite								
Van7-78-41	Orthogonal	Plag harzburgite, protomylonite								
Van7-85-24	Oblique	Lherzolite								
Van7-85-27	Oblique	Harzburgite	67.1	25.6	6.2	1.0	0.0	100.0	2825	
Van7-85-30	Oblique	Harzburgite, trace Plag								
Van7-85-32	Oblique	Harzburgite								
Van7-85-39	Oblique	Lherzolite								
Van7-85-42	Oblique	Lherzolite								
Van7-85-47	Oblique	Lherzolite	68.7	24.7	5.1	1.5	0.0	100.0	2594	
Van7-85-49	Oblique	Lherzolite, trace Plag								
Van7-86-25	Oblique	Lherzolite								
Van7-86-27	Oblique	Lherzolite								
Van7-86-28	Oblique	Lherzolite								
Van7-86-30	Oblique	Harzburgite	55.8	34.0	9.6	0.6	0.0	100.0	2566	
Van7-86-37	Oblique	Harzburgite								
Van7-96-09	Oblique	Pyroxenite vein								
Van7-96-14	Oblique	Pyroxenite vein, with thin peridotite rim								
Van7-96-15	Oblique	Pyroxenite vein, with thin peridotite rim	11.6	34.0	24.4	1.6	28.4	100.0	697	
Van7-96-16	Oblique	Pyroxenite vein, with thin peridotite rim								
Van7-96-18	Oblique	Lherzolite with pyroxenite vein								
Van7-96-19 (Matrix)	Oblique	Lherzolite with pyroxenite vein								
Van7-96-19 (Vein)	Oblique	Pyroxenite vein in lherzolite								
Van7-96-21 (Matrix)	Oblique	Harzburgite with pyroxenite vein	7.2	26.4	57.2	2.8	6.3	100.0	318	
Van7-96-21 (Vein)	Oblique	Pyroxenite vein in harzburgite	76.2	18.1	4.6	0.6	0.4	100.0	1894	
Van7-96-25	Oblique	Lherzolite	19.7	39.3	37.5	2.2	1.3	100.0	461	
Van7-96-28	Oblique	Lherzolite, trace Plag	66.2	25.6	6.8	1.4	0.0	100.0	2215	
Van7-96-35	Oblique	Lherzolite, trace Plag	76.7	17.2	5.4	0.7	0.0	100.0	2738	
			70.4	16.7	10.5	2.3	0.1	100.0	1765	

Table 5.3: Sample descriptions Cont.

Sample	Location	Description	OI	Opx	Cpx	Sp	Pl	Sum	Pts ^a
Van7-96-37	Oblique	Peridotite							
Van7-96-38	Oblique	Lherzolite, trace Plag	64.2	24.1	9.9	1.6	0.2	100.0	2278
Van7-96-66B	Oblique	Subangular clast of orthopyroxenite in breccia							
Atlantis II Fracture Zone									
RC27-9-6-2 (Matrix)	AIIFZ RTI	Lherzolite with clinopyroxenite vein, trace Plag	67.5	19.5	11.9	0.9	0.2	100.0	2194
RC27-9-6-2 (Vein)	AIIFZ RTI	Clinopyroxenite vein in lherzolite, trace Plag	7.6	9.2	82.6	0.7	0.0	100.0	436
RC27-9-6-5	AIIFZ RTI	Lherzolite, cryptic metasomatism, trace Plag							
RC27-9-6-7	AIIFZ RTI	Lherzolite							
JR31-52-3	Atlantis Bank	Harzburgite	57.6	37.5	3.9	1.0	0.0	100.0	2056
JR31-52-4	Atlantis Bank	Lherzolite							
JR31-52-5	Atlantis Bank	Harzburgite	66.8	27.3	4.5	1.4	0.0	100.0	2258
JR31-52-8	Atlantis Bank	Lherzolite	62.8	29.9	6.5	0.8	0.0	100.0	1975
6K-458-1	Atlantis Bank	Peridotite							
6K-458-3	Atlantis Bank	Peridotite, cryptic metasomatism							
6K-465-2	AIIFZ RTI	Lherzolite, cryptic metasomatism, trace Plag	71.0	22.0	5.9	1.1	0.0	100.0	1926
6K-465-3	AIIFZ RTI	Lherz, with gabbro vein and kaersutite/oxide vein	76.2	13.4	8.6	1.0	0.8	100.0	2028
6K-466-1	Atlantis Bank	Harzburgite	74.8	21.2	3.0	1.1	0.0	100.0	1489
6K-643-01	Atlantis Bank	Lherzolite							
6K-643-06	Atlantis Bank	Harzburgite							
6K-643-13	Atlantis Bank	Lherzolite	68.7	24.9	5.1	0.7	0.0	99.4	1566
6K-643-15	Atlantis Bank	Lherzolite/dunite/gabbro vein							
6K-649-5	Atlantis Bank	Lherzolite							
6K-651-7	Atlantis Bank	Olivine websterite							

^a Points counts on 51x75 mm thin sections using 1 mm grid.

^b Snow (1993).

^c H. J. B. Dick, unpublished data.

^d Y. Ohara, unpublished data.

^e Dantas et al. (2007).

Table 5.4: Major element composition of olivines, in wt%.

Sample	Grn ^a	Pts	SiO ₂	±	FeO	±	MnO	±	MgO	±	NiO	±	Total	Mg#	Notes ^b
6K-458-1	1	6	40.55	0.26	9.22	0.18	0.08	0.03	49.64	0.29	0.34	0.04	99.87	90.6	
6K-465-2	1	6	40.53	0.28	9.63	0.11	0.14	0.02	50.00	0.36	0.32	0.03	100.68	90.3	Dep
6K-465-2	2	6	41.91	0.28	9.39	0.09			49.32	0.19			100.67	90.3	Dep
6K-465-3	1	9	39.45	0.14	10.52	0.13	0.15	0.02	48.50	0.16	0.35	0.02	99.03	89.2	NrMat
6K-465-3	2	6	38.02	0.15	22.94	0.30	0.43	0.04	39.16	0.09	0.28	0.04	100.86	75.3	NrMat
JR31-52-3	1	7	40.10	0.08	9.45	0.12	0.14	0.02	49.48	0.26	0.40	0.03	99.61	90.3	
JR31-52-5	1	6	40.47	0.12	9.43	0.15	0.16	0.02	49.06	0.09	0.35	0.03	99.51	90.3	
JR31-52-8	1	6	40.22	0.09	9.59	0.24	0.14	0.02	49.04	0.14	0.37	0.02	99.38	90.1	
Kn162-19-41	1	6	40.58	0.21	9.32	0.14	0.13	0.02	48.91	0.25	0.31	0.03	99.29	90.3	PyxVn
Kn162-47-22	1	6	40.79	0.16	10.66	0.14	0.11	0.02	48.75	0.16	0.38	0.03	100.75	89.1	PyxVn
Kn162-47-25	1	4	40.51	0.18	9.25	0.10	0.12	0.01	48.98	0.29	0.32	0.04	99.22	90.4	NrMat
Kn162-47-26	1	6	41.04	0.21	9.69	0.51	0.14	0.02	48.60	0.06	0.19	0.01	99.69	89.9	
Kn162-47-29	1	6	41.20	0.22	8.83	0.35	0.11	0.02	49.87	0.17	0.21	0.03	100.23	91.0	
Kn162-47-36	1	6	40.62	0.14	10.10	0.45	0.14	0.02	49.02	0.37	0.40	0.03	100.31	89.6	
Kn162-47-49	1	6	41.03	0.09	10.08	0.18	0.12	0.03	48.24	0.26	0.25	0.02	99.76	89.5	
PS86-6-26	1	6	41.33	0.34	9.29	0.22	0.10	0.02	49.11	0.40	0.19	0.03	100.05	90.4	
PS86-6-38	1	6	40.50	0.16	9.95	0.20	0.11	0.02	48.88	0.26	0.41	0.02	99.89	89.8	
PS86-6-39	1	6	40.61	0.18	9.15	0.31	0.11	0.03	49.60	0.26	0.39	0.04	99.92	90.6	
PS86-6-40	1	6	40.42	0.27	9.71	0.14	0.10	0.04	48.85	0.35	0.37	0.05	99.47	90.0	
RC27-9-6-2	L1	5	40.40	0.30	9.91	0.10	0.13	0.02	49.46	0.41	0.33	0.01	100.26	89.9	FarMat
RC27-9-6-2	L6	5	40.40	0.10	9.74	0.14	0.12	0.02	50.21	0.12	0.34	0.03	100.86	90.2	FarMat
RC27-9-6-2	L7	5	40.40	0.20	9.71	0.10	0.13	0.01	50.17	0.22	0.33	0.02	100.79	90.2	FarMat
RC27-9-6-2	L8	5	40.50	0.10	9.58	0.09	0.11	0.02	49.77	0.28	0.32	0.01	100.30	90.3	FarMat
RC27-9-6-2	L9	5	40.40	0.20	9.60	0.23	0.14	0.01	49.75	0.17	0.34	0.01	100.27	90.2	FarMat
RC27-9-6-2	L12	5	40.50	0.10	9.51	0.10	0.12	0.02	49.44	0.17	0.33	0.02	99.94	90.3	NrMat
RC27-9-6-2	L13	5	40.50	0.10	9.60	0.16	0.14	0.02	49.28	0.14	0.33	0.02	99.90	90.1	NrMat
RC27-9-6-2	L14	5	40.60	0.20	9.74	0.17	0.13	0.00	49.60	0.29	0.34	0.02	100.45	90.1	NrMat
RC27-9-6-2	L15	6	40.30	0.20	9.71	0.17	0.14	0.02	49.97	0.09	0.32	0.02	100.50	90.2	NrMat
RC27-9-6-2	L16	5	40.20	0.20	9.60	0.12	0.13	0.01	50.02	0.22	0.35	0.03	100.33	90.3	NrMat
RC27-9-6-2	L17	5	40.40	0.30	9.76	0.18	0.15	0.01	49.47	0.27	0.32	0.02	100.14	90.0	CpxVn
RC27-9-6-2	L18	5	40.30	0.20	9.64	0.18	0.13	0.01	49.47	0.29	0.34	0.01	99.92	90.1	CpxVn
RC27-9-6-2	L19	5	40.40	0.20	9.55	0.12	0.13	0.02	49.04	0.15	0.35	0.02	99.51	90.1	CpxVn
RC27-9-6-2	L20	5	40.10	0.10	9.74	0.11	0.13	0.01	49.62	0.33	0.32	0.01	99.96	90.1	AdjVn
RC27-9-6-2	L22	5	40.20	0.30	9.56	0.17	0.13	0.01	49.85	0.27	0.33	0.02	100.11	90.3	NrMat
RC27-9-6-2	L23	5	40.50	0.30	9.64	0.11	0.12	0.02	50.33	0.56	0.35	0.02	100.98	90.3	NrMat
RC27-9-6-2	L24	5	40.40	0.10	9.51	0.14	0.13	0.01	49.98	0.45	0.36	0.02	100.40	90.4	NrMat
RC27-9-6-2	L25	5	40.50	0.20	9.60	0.10	0.14	0.02	50.05	0.16	0.34	0.02	100.66	90.3	NrMat
RC27-9-6-2	L26	5	40.50	0.50	9.56	0.20	0.13	0.01	49.89	0.52	0.33	0.03	100.44	90.3	FarMat
RC27-9-6-5	L1	6	40.50	0.02	10.32	0.07	0.11	0.01	49.43	0.07	0.35	0.02	100.76	89.5	
RC27-9-6-5	L2	6	40.70	0.30	9.85	0.11	0.11	0.02	49.98	0.31	0.39	0.02	101.12	90.0	
RC27-9-6-5	1	5	40.76	0.17	9.93	0.26	0.13	0.04	49.16	0.31	0.34	0.03	100.36	89.8	Enr
RC27-9-6-7	1	6	41.08	0.18	9.60	0.06	0.14	0.01	49.33	0.61	0.31	0.02	100.50	90.2	
Van7-78-25	1	6	40.12	0.06	10.19	0.31	0.20	0.01	48.57	0.16	0.28	0.02	99.41	89.5	GbVn
Van7-78-31	1	6	40.26	0.08	9.82	0.17	0.16	0.02	48.57	0.20	0.31	0.03	99.16	89.8	NrMat
Van7-78-31	2	5	39.90	0.11	10.61	0.15	0.17	0.02	48.29	0.08	0.29	0.03	99.33	89.0	AdjVn
Van7-78-36	2	6	39.71	0.32	11.57	0.08	0.20	0.02	47.27	0.25	0.31	0.03	99.13	87.9	AdjVn
Van7-78-36	11	6	40.41	0.11	9.75	0.20	0.16	0.02	48.80	0.16	0.32	0.03	99.48	89.9	FarMat
Van7-78-40	1	6	40.15	0.29	9.09	0.13	0.15	0.02	49.70	0.22	0.32	0.02	99.48	90.7	
Van7-78-41	1	6	40.35	0.15	10.22	0.16	0.17	0.02	48.43	0.14	0.31	0.04	99.53	89.4	
Van7-85-24	1	6	40.12	0.11	9.31	0.05	0.15	0.02	48.87	0.16	0.34	0.02	98.85	90.3	
Van7-85-27	1	2	40.01	0.04	8.92	0.11	0.17	0.02	49.30	0.25	0.31	0.02	98.77	90.8	
Van7-85-27	2	3	40.11	0.11	8.82	0.22	0.15	0.02	49.16	0.08	0.35	0.03	98.67	90.9	
Van7-85-30	1	4	40.17	0.10	9.28	0.08	0.15	0.03	48.78	0.24	0.34	0.04	98.80	90.4	
Van7-85-32	1	6	40.24	0.10	9.52	0.13	0.14	0.02	49.26	0.32	0.30	0.03	99.51	90.2	
Van7-85-39	1	5	40.19	0.13	9.47	0.15	0.16	0.03	49.07	0.39	0.37	0.02	99.29	90.2	
Van7-85-42	1	6	40.03	0.26	9.57	0.14	0.13	0.02	49.36	0.27	0.35	0.02	99.50	90.2	
Van7-85-47	1	6	40.34	0.26	9.76	0.18	0.16	0.03	49.48	0.12	0.33	0.03	100.12	90.0	
Van7-85-49	1	6	40.49	0.11	9.57	0.16	0.15	0.01	48.91	0.18	0.40	0.03	99.57	90.1	
Van7-86-25	1	6	41.14	0.47	9.02	0.23	0.10	0.02	49.17	0.52	0.22	0.02	99.70	90.7	
Van7-86-27	1	6	40.26	0.26	9.91	0.10	0.14	0.02	49.00	0.27	0.35	0.05	99.68	89.8	
Van7-86-28	1	6	40.29	0.20	9.69	0.21	0.15	0.02	48.77	0.40	0.39	0.03	99.30	90.0	

Table 5.4: Olivine major elements Cont.

Sample	Gm ^a	Pts	SiO ₂	±	FeO	±	MnO	±	MgO	±	NiO	±	Total	Mg#	Notes ^b
Van7-86-30	1	4	40.49	0.18	9.11	0.26	0.14	0.01	49.24	0.26	0.41	0.04	99.43	90.6	
Van7-86-37	1	3	40.19	0.11	9.32	0.07	0.15	0.03	48.53	0.09	0.35	0.02	98.65	90.3	
Van7-96-09	1	6	40.30	0.09	9.21	0.17	0.11	0.01	50.44	0.16	0.45	0.02	100.55	90.7	PyxVn
Van7-96-14	1	7	40.00	0.16	11.64	0.15	0.20	0.01	48.38	0.07	0.40	0.03	100.71	88.1	AdjVn
Van7-96-14	2	6	39.75	0.17	12.19	0.32	0.24	0.03	47.78	0.13	0.43	0.01	100.45	87.5	PyxVn
Van7-96-15	1	6	39.88	0.18	10.98	0.15	0.22	0.02	48.08	0.13	0.37	0.03	99.61	88.6	PyxVn
Van7-96-16	2	5	40.18	0.19	10.89	0.18	0.22	0.02	47.75	0.13	0.37	0.02	99.50	88.7	PyxVn
Van7-96-18	1	5	40.27	0.21	10.49	0.19	0.14	0.03	48.84	0.17	0.38	0.03	100.19	89.2	NrMat
Van7-96-18	2	6	40.02	0.11	10.56	0.33	0.16	0.02	48.51	0.21	0.39	0.02	99.68	89.1	PyxVn
Van7-96-19	1	6	40.20	0.12	10.11	0.29	0.13	0.02	49.51	0.28	0.40	0.02	100.40	89.7	NrMat
Van7-96-19	2	5	40.07	0.15	10.43	0.40	0.16	0.03	48.40	0.22	0.40	0.02	99.51	89.2	PyxVn
Van7-96-19	3	6	40.15	0.14	10.51	0.16	0.14	0.01	48.47	0.33	0.38	0.05	99.67	89.2	PyxVn
Van7-96-21	2	6	40.17	0.13	9.76	0.21	0.16	0.03	50.03	0.24	0.34	0.03	100.51	90.1	PyxVn
Van7-96-21	3	6	40.16	0.23	9.69	0.11	0.16	0.03	49.85	0.12	0.35	0.04	100.27	90.2	AdjVn
Van7-96-25	1	6	40.09	0.16	10.10	0.32	0.12	0.02	48.87	0.24	0.39	0.04	99.64	89.6	
Van7-96-28	1	6	40.42	0.09	9.18	0.15	0.13	0.03	49.33	0.39	0.36	0.03	99.47	90.5	
Van7-96-35	1	6	40.62	0.14	8.66	0.21	0.11	0.02	50.34	0.23	0.43	0.04	100.19	91.2	
Van7-96-37	1	6	40.40	0.43	9.22	0.25	0.10	0.01	49.39	0.12	0.37	0.03	99.53	90.5	
Van7-96-38	1	6	40.41	0.12	9.80	0.20	0.15	0.02	49.81	0.14	0.35	0.04	100.55	90.1	

^a Grain number; L preceding a grain number indicates Lee (1997) data for dredge RC27-9-6 samples.

^b Abbreviations for lithologies (blank indicates residual peridotite): Dep, Trans and Enr are depleted, transitional and enriched zones in cryptically metasomatized sample; GbVn refers to a gabbroic vein, PyxVn to a pyroxenite vein, CpxVn to a clinopyroxenite vein, and OpxVn to an orthopyroxenite vein; AdjVn is adjacent to a vein, NrMat is <5 cm from a vein and FarMat is >5 cm from a vein.

Table 5.5: Major element composition of Opx, in wt%.

Analysis	Gm ^a Pts	SiO ₂	±	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	FeO	±	MnO	±	MgO	±	CaO	±	Na ₂ O	±	Total	Notes ^b
6K-458-1	1C	53.01	1.81	0.10	0.04	4.17	0.89	1.38	0.81	6.43	1.08	0.21	0.07	32.27	1.15	1.24	1.26	0.07	0.07	98.88	
6K-458-1	1R	54.82	0.44	0.09	0.02	5.05	0.14	0.85	0.03	6.05	0.15	0.14	0.01	32.43	0.51	1.29	0.71	0.02	0.02	100.74	
6K-458-3	1C	54.17	0.29	0.09	0.03	5.64	0.24	0.75	0.03	6.12	0.23	0.12	0.02	30.56	0.88	1.97	1.04	0.01	0.01	99.42	
6K-458-3	1C	54.77	0.24	0.08	0.02	4.91	0.11	0.62	0.02	6.23	0.16	0.12	0.03	31.18	0.46	1.32	0.65	0.00	0.00	99.24	
6K-465-2	1C	55.44	0.29	0.09	0.01	4.17	0.16	0.64	0.01	6.32	0.12	0.12	0.01	32.18	0.37	1.58	0.41	0.03	0.01	100.57	Dep
6K-465-2	1R	55.78	0.34	0.09	0.01	3.34	0.46	0.61	0.02	6.35	0.10	0.12	0.02	33.04	0.45	1.19	0.20	0.03	0.01	100.54	Dep
6K-465-2	2C	53.42	0.30	0.07	0.02	5.51	0.21	0.81	0.05	6.12	0.29	0.13	0.03	31.24	0.53	1.77	0.84	0.00	0.00	99.07	Dep
6K-643-3	2C	54.25	0.05	0.12	0.01	4.37	0.22	0.65	0.01	6.58	0.09	0.12	0.00	31.51	0.07	1.78	0.24	0.02	0.02	99.41	NrMat
6K-643-3	2R	54.81	0.37	0.18	0.07	3.24	0.69	0.68	0.06	6.80	0.00	0.13	0.01	31.44	0.56	1.64	0.09	0.13	0.12	99.05	NrMat
6K-466-1	1C	54.70	0.50	0.08	0.03	4.57	0.15	0.77	0.03	5.80	0.29	0.14	0.05	31.80	0.91	2.37	1.22	0.04	0.04	100.27	
6K-466-1	1R	55.77	0.30	0.07	0.01	3.98	0.07	0.60	0.03	5.92	0.13	0.14	0.02	32.74	0.36	1.36	0.45	0.02	0.02	100.59	
6K-466-1	2I	54.19	0.21	0.05	0.01	5.07	0.10	0.90	0.04	5.97	0.16	0.13	0.02	31.15	0.56	2.18	0.55	0.02	0.01	99.65	
6K-643-13	1C	54.90	0.46	0.05	0.02	3.98	0.27	0.67	0.05	6.01	0.30	0.13	0.02	32.05	1.29	1.92	1.75	0.01	0.01	99.71	
6K-643-13	1R	54.48	0.18	0.05	0.01	4.97	0.08	0.68	0.02	6.57	0.16	0.12	0.03	31.38	0.50	0.90	0.57	0.00	0.00	99.14	PyxVn
6K-651-7	1C	54.28	0.32	0.07	0.02	5.31	0.09	0.73	0.02	6.37	0.17	0.12	0.02	31.03	0.56	1.40	0.76	0.00	0.01	99.32	PyxVn
6K-651-7	1R	54.36	0.29	0.06	0.02	5.45	0.13	0.90	0.03	5.97	0.17	0.13	0.02	30.94	0.47	1.96	0.56	0.06	0.03	99.84	
JR31-52-03	1C	55.17	0.36	0.05	0.01	4.29	0.29	0.58	0.07	6.01	0.19	0.13	0.02	32.00	0.39	1.21	0.40	0.02	0.01	99.45	
JR31-52-03	1R	54.25	0.25	0.08	0.02	5.25	0.09	0.71	0.08	5.97	0.20	0.12	0.03	31.24	0.62	1.80	0.79	0.06	0.03	99.47	
JR31-52-04	1R	54.46	0.12	0.08	0.01	5.18	0.07	0.79	0.02	5.91	0.23	0.12	0.02	31.07	0.96	2.12	1.28	0.06	0.04	99.80	
JR31-52-04	2I	55.77	0.28	0.08	0.01	3.63	0.20	0.43	0.05	6.12	0.11	0.13	0.02	32.81	0.19	0.89	0.08	0.02	0.01	99.89	
JR31-52-05	1C	54.17	0.49	0.08	0.01	5.38	0.15	0.81	0.04	5.84	0.16	0.14	0.03	30.86	0.68	2.30	0.89	0.05	0.03	99.63	
JR31-52-05	1R	54.59	0.40	0.05	0.02	5.15	0.28	0.76	0.04	5.93	0.18	0.08	0.02	31.78	0.42	1.20	0.37	0.04	0.01	99.58	
JR31-52-08	1C	54.45	0.42	0.09	0.02	5.16	0.24	0.80	0.05	5.93	0.18	0.11	0.03	30.88	0.72	2.30	1.00	0.06	0.03	99.76	
JR31-52-08	1R	55.82	0.51	0.08	0.01	3.68	0.31	0.42	0.09	6.16	0.16	0.12	0.02	32.64	0.34	0.83	0.26	0.02	0.02	99.76	
JR31-52-08	3I	56.00	0.20	0.05	0.01	3.94	0.11	0.57	0.02	6.10	0.10	0.12	0.02	32.37	0.19	1.32	0.17	0.05	0.01	100.53	PyxVn
Kn162-19-41	1C	55.24	0.19	0.07	0.02	4.47	0.14	0.53	0.03	5.83	0.22	0.12	0.02	32.13	0.23	1.55	0.31	0.00	0.01	99.96	PyxVn
Kn162-19-41	1R	54.88	0.34	0.06	0.02	4.68	0.28	0.54	0.04	6.26	0.11	0.13	0.03	32.26	0.39	0.94	0.39	0.00	0.00	99.76	
Kn162-47-02	1I	56.70	0.16	0.00	0.01	2.51	0.07	0.97	0.04	4.89	0.11	0.13	0.02	34.06	0.19	1.32	0.30	0.00	0.02	100.59	
Kn162-47-08	1C	53.77	0.28	0.14	0.03	5.94	0.18	0.74	0.04	6.22	0.20	0.13	0.02	30.54	0.81	2.95	1.17	0.00	0.01	100.43	
Kn162-47-08	1R	54.53	0.48	0.09	0.01	4.79	0.64	0.57	0.09	6.42	0.16	0.15	0.02	31.16	0.48	1.24	0.27	0.03	0.02	98.99	
Kn162-47-09	1C	56.02	0.34	0.06	0.02	3.57	0.22	0.96	0.04	5.31	0.22	0.11	0.02	32.66	0.79	1.84	0.91	0.09	0.05	100.63	
Kn162-47-09	1R	56.56	0.43	0.06	0.01	3.33	0.14	0.76	0.04	5.30	0.13	0.12	0.02	33.30	0.30	1.26	0.29	0.07	0.03	100.75	
Kn162-47-09	2C	55.41	0.27	0.07	0.01	3.33	0.20	0.88	0.06	5.23	0.17	0.13	0.02	32.02	0.63	1.81	0.84	0.04	0.04	98.93	
Kn162-47-09	2R	55.55	0.15	0.06	0.01	3.08	0.07	0.76	0.03	5.34	0.09	0.13	0.02	32.64	0.32	1.27	0.17	0.04	0.02	98.88	
Kn162-47-09	4C	56.51	0.59	0.05	0.01	3.12	0.08	0.74	0.03	5.32	0.17	0.10	0.02	33.48	0.49	1.35	0.44	0.09	0.02	100.75	
Kn162-47-10	1C	53.98	0.18	0.08	0.01	5.59	0.18	0.70	0.02	6.31	0.20	0.13	0.02	30.94	0.47	1.78	0.43	0.01	0.01	99.52	PyxVn
Kn162-47-10	1R	54.40	0.53	0.09	0.04	4.50	0.23	0.52	0.05	6.17	0.75	0.13	0.02	30.64	2.67	2.97	3.70	0.03	0.07	99.46	PyxVn
Kn162-47-10	2C	54.38	0.70	0.07	0.02	4.89	0.47	0.63	0.05	6.31	0.16	0.14	0.01	31.34	0.64	1.41	0.61	0.01	0.03	99.18	FarMat
Kn162-47-10	2R	55.59	0.24	0.08	0.02	3.10	0.15	0.29	0.02	6.59	0.12	0.15	0.02	32.58	0.14	0.97	0.04	0.00	0.00	99.35	FarMat

Table 5.5: Opx major elements Cont.

Analysis	Gm ^a	Pts	SiO ₂	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	FeO	±	MnO	±	MgO	±	CaO	±	Na ₂ O	±	Total	Notes ^b	
Kn162-47-12	IC	9	53.82	0.32	0.09	0.03	5.71	0.26	0.75	0.04	6.26	0.20	0.14	0.03	30.97	0.94	1.78	1.19	0.01	0.02	99.54	PyxVn
Kn162-47-12	IR	9	54.36	0.32	0.09	0.03	4.86	0.38	0.64	0.06	6.22	0.27	0.15	0.02	31.10	0.95	2.13	1.55	0.01	0.02	99.56	PyxVn
Kn162-47-22	IC	10	53.85	0.66	0.09	0.01	5.61	0.06	0.68	0.02	6.89	0.10	0.15	0.02	31.36	0.24	1.73	0.30	0.00	0.00	100.36	PyxVn
Kn162-47-22	IR	10	53.87	0.29	0.07	0.01	5.72	0.09	0.70	0.02	6.89	0.10	0.14	0.01	31.97	0.36	1.31	0.24	0.00	0.00	100.68	PyxVn
Kn162-47-24	IC	10	54.36	0.47	0.06	0.02	5.41	0.64	0.83	0.04	6.32	0.19	0.11	0.02	30.82	0.75	2.29	0.80	0.02	0.02	100.22	
Kn162-47-24	IR	10	54.98	0.70	0.05	0.03	5.08	0.61	0.75	0.07	6.47	0.28	0.10	0.03	31.34	1.26	1.72	1.17	0.02	0.02	100.50	
Kn162-47-25	IC	10	53.69	0.46	0.16	0.03	5.35	0.12	0.66	0.03	6.50	0.37	0.14	0.02	30.38	1.31	2.43	1.79	0.03	0.06	99.35	PyxVn
Kn162-47-25	IR	8	54.22	0.27	0.11	0.02	5.00	0.08	0.61	0.02	6.70	0.11	0.14	0.02	31.55	0.27	1.16	0.08	0.00	0.00	99.49	PyxVn
Kn162-47-26	IC	9	54.50	0.11	0.07	0.02	5.46	0.21	0.64	0.03	6.41	0.11	0.12	0.02	31.11	0.27	1.67	0.39	0.04	0.03	100.03	
Kn162-47-26	IR	9	54.94	0.20	0.03	0.01	4.95	0.08	0.59	0.02	6.39	0.12	0.11	0.02	32.22	0.43	1.12	0.16	0.01	0.01	100.35	
Kn162-47-29	IC	10	54.21	0.30	0.09	0.02	6.01	0.24	0.79	0.04	5.56	0.32	0.08	0.03	31.04	1.12	2.48	1.52	0.04	0.04	100.31	
Kn162-47-29	IR	10	54.91	0.24	0.08	0.01	5.50	0.15	0.65	0.04	5.74	0.10	0.08	0.02	31.98	0.29	1.49	0.30	0.03	0.02	100.44	
Kn162-47-36	IC	10	54.53	0.43	0.06	0.01	5.24	0.18	0.74	0.03	6.66	0.21	0.10	0.03	31.13	0.37	2.01	0.35	0.04	0.05	100.49	
Kn162-47-36	IR	9	55.09	0.51	0.06	0.01	4.85	0.43	0.58	0.10	6.71	0.16	0.11	0.03	31.87	0.45	1.48	0.25	0.02	0.03	100.77	
Kn162-47-49	IC	10	54.65	0.52	0.08	0.03	5.03	0.55	0.70	0.08	6.34	0.37	0.10	0.02	30.65	1.15	2.35	1.77	0.03	0.03	99.91	
Kn162-47-49	IR	9	55.85	0.51	0.04	0.03	3.25	0.80	0.42	0.08	6.77	0.34	0.11	0.02	32.20	0.87	0.89	0.90	0.05	0.04	99.58	
PS86-6-26	IC	10	54.56	0.79	0.11	0.08	5.88	0.28	0.56	0.05	5.99	0.47	0.08	0.04	30.99	2.13	2.03	2.66	0.08	0.15	100.27	
PS86-6-26	IR	14	54.95	0.42	0.09	0.02	5.33	0.22	0.43	0.03	6.27	0.18	0.10	0.02	31.86	0.36	1.16	0.23	0.02	0.02	100.21	
PS86-6-38	IC	8	53.76	0.32	0.14	0.03	6.22	0.18	0.61	0.02	6.37	0.16	0.15	0.05	30.82	0.37	1.54	0.45	0.06	0.04	99.67	
PS86-6-38	IR	8	53.80	0.28	0.14	0.04	6.02	0.15	0.61	0.03	6.28	0.27	0.16	0.05	30.60	0.99	1.97	1.29	0.10	0.08	99.68	
PS86-6-39	IC	9	55.30	0.29	0.10	0.02	4.22	0.12	0.57	0.04	6.19	0.09	0.13	0.03	32.12	0.15	1.30	0.13	0.04	0.04	99.96	
PS86-6-39	IR	8	55.67	0.65	0.10	0.02	3.86	0.08	0.51	0.03	6.25	0.20	0.15	0.03	32.63	0.26	1.10	0.08	0.05	0.03	100.31	
PS86-6-40	IC	11	54.40	0.57	0.14	0.03	4.94	0.72	0.48	0.06	6.44	0.22	0.11	0.02	31.14	0.57	1.31	0.45	0.06	0.03	99.02	
PS86-6-40	IR	10	54.39	0.26	0.13	0.02	5.16	0.09	0.50	0.02	6.40	0.18	0.15	0.03	31.47	0.43	1.27	0.10	0.05	0.04	99.52	
RC27-9-6-2	L1C	10	56.2	0.2	0.07	0.02	2.06	0.09	0.67	0.05	6.23	0.10	0.08	0.03	33.07	0.4	1.46	0.46	0.00	0.01	99.84	FarMat
RC27-9-6-2	L4I	9	55.4	0.3	0.07	0.03	3.96	0.19	1.00	0.04	6.13	0.25	0.15	0.02	32.50	0.8	1.92	1.0	0.01	0.03	100.71	NrMat
RC27-9-6-2	L6I	10	56.1	0.3	0.09	0.02	2.29	0.27	0.62	0.04	6.54	0.12	0.18	0.02	33.38	0.5	0.81	0.3	0.04	0.02	99.93	CpxVn
RC27-9-6-2	L7C	10	54.9	0.2	0.09	0.02	3.87	0.12	0.92	0.05	6.80	0.12	0.08	0.02	32.41	0.4	1.63	0.4	0.05	0.03	99.37	NrMat
RC27-9-6-2	L8R	10	56.4	0.6	0.20	0.03	1.94	0.09	0.62	0.03	6.22	0.19	0.17	0.01	34.00	0.6	1.15	0.76	0.02	0.03	101.15	FarMat
RC27-9-6-2	L9C	10	55.2	0.2	0.08	0.01	3.57	0.09	0.81	0.02	6.07	0.13	0.07	0.02	32.21	0.3	1.88	0.3	0.00	0.01	100.70	FarMat
RC27-9-6-2	L9R	10	55.9	0.4	0.05	0.02	2.58	0.39	0.46	0.08	6.32	0.09	0.08	0.03	33.24	0.4	0.69	0.3	0.02	0.02	100.03	FarMat
RC27-9-6-2	L3C	10	55.34	0.47	0.06	0.01	4.14	0.10	0.98	0.03	6.14	0.11	0.13	0.02	31.58	0.32	2.36	0.22	0.06	0.03	100.78	FarMat
RC27-9-6-2	L3R	10	56.00	0.34	0.04	0.01	3.12	0.18	0.59	0.07	6.48	0.07	0.14	0.01	33.02	0.41	0.77	0.18	0.02	0.02	100.19	FarMat
RC27-9-6-5	IC	9	55.02	0.52	0.09	0.02	3.82	0.55	0.68	0.04	6.30	0.14	0.09	0.02	31.30	0.60	1.56	0.71	0.04	0.03	98.91	Trans
RC27-9-6-5	IR	10	56.31	0.45	0.13	0.02	2.40	0.19	0.61	0.03	6.54	0.15	0.10	0.02	32.63	0.36	0.88	0.36	0.06	0.08	99.66	Trans
RC27-9-6-5	5I	7	55.77	0.83	0.08	0.01	3.30	0.32	0.70	0.04	6.49	0.02	0.14	0.02	32.39	0.24	1.10	0.31	0.01	0.01	99.99	Enr
RC27-9-6-5	L1C	10	53.9	0.3	0.09	0.03	5.71	0.13	0.76	0.04	6.59	0.50	0.11	0.02	31.11	1.5	2.29	2.1	0.02	0.02	100.58	
RC27-9-6-7	IC	10	54.16	0.27	0.08	0.02	5.25	0.40	0.72	0.04	6.16	0.16	0.14	0.02	31.13	0.65	1.86	0.58	0.01	0.02	99.53	
RC27-9-6-7	IR	10	55.13	0.27	0.06	0.01	4.22	0.15	0.38	0.02	6.46	0.12	0.14	0.02	32.88	0.27	0.59	0.11	0.00	0.01	99.84	
Van7-78-25	3I	7	56.85	0.36	0.15	0.02	1.81	0.19	0.58	0.07	6.84	0.10	0.18	0.02	33.02	0.09	0.88	0.08	0.00	0.00	100.31	GbVn
Van7-78-31	IC	12	57.12	0.31	0.13	0.02	1.54	0.11	0.43	0.05	6.40	0.18	0.15	0.03	33.20	0.48	1.13	0.74	0.01	0.01	100.11	NrMat

Table 5.5: Opx major elements Cont.

Analysis	Gm ^a	Pts	SiO ₂	±	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	FeO	±	MnO	±	MgO	±	CaO	±	Na ₂ O	±	Total	Notes ^b
Van7-78-31	IR	13	57.32	0.26	0.12	0.01	1.52	0.13	0.39	0.04	6.55	0.10	0.16	0.02	33.38	0.17	0.69	0.10	0.01	0.02	100.13	NrMat
Van7-78-31	2C	7	56.06	0.18	0.14	0.02	1.90	0.03	0.61	0.01	7.10	0.12	0.13	0.03	31.95	0.25	1.01	0.35	0.03	0.02	98.95	AdjVn
Van7-78-31	2R	10	56.61	0.10	0.17	0.02	1.20	0.06	0.31	0.03	7.16	0.10	0.15	0.01	32.62	0.20	0.60	0.10	0.02	0.02	98.84	AdjVn
Van7-78-36	1C	13	55.35	1.10	0.17	0.03	2.73	0.39	0.99	0.66	7.83	0.41	0.17	0.03	31.78	0.65	1.37	0.70	0.04	0.03	100.43	AdjVn
Van7-78-36	IR	12	56.49	0.33	0.22	0.01	1.47	0.16	0.37	0.06	7.89	0.14	0.19	0.02	32.99	0.13	0.69	0.13	0.01	0.02	100.31	AdjVn
Van7-78-36	11C	13	56.86	0.88	0.16	0.02	1.79	0.28	0.54	0.03	6.72	0.28	0.14	0.06	33.38	1.16	1.28	1.17	0.03	0.02	100.90	FarMat
Van7-78-36	11R	12	56.65	0.50	0.17	0.02	1.92	0.11	0.51	0.05	6.61	0.22	0.14	0.03	33.24	0.52	1.23	0.61	0.02	0.01	100.49	FarMat
Van7-78-40	1C	11	56.13	0.25	0.10	0.01	2.90	0.18	0.72	0.06	6.03	0.24	0.14	0.02	32.98	0.44	1.19	0.55	0.01	0.01	100.20	
Van7-78-40	IR	12	57.35	0.26	0.12	0.02	1.35	0.18	0.34	0.06	6.11	0.12	0.16	0.02	34.25	0.14	0.49	0.07	0.01	0.01	100.18	
Van7-78-41	1C	12	56.67	0.31	0.09	0.02	1.82	0.19	0.54	0.04	7.02	0.18	0.15	0.02	33.26	0.38	0.98	0.47	0.01	0.01	100.53	
Van7-78-41	IR	13	56.40	0.23	0.09	0.02	2.38	0.14	0.64	0.04	6.92	0.22	0.15	0.02	32.90	0.50	1.08	0.68	0.02	0.02	100.57	
Van7-85-24	1C	8	54.69	0.47	0.06	0.01	4.78	0.23	0.69	0.03	6.23	0.14	0.14	0.02	30.99	0.49	1.76	0.41	0.04	0.03	99.36	
Van7-85-24	IR	7	54.26	0.14	0.05	0.01	5.14	0.07	0.71	0.02	6.38	0.12	0.12	0.03	31.03	0.13	1.18	0.03	0.04	0.02	98.90	
Van7-85-27	2C	11	54.86	0.35	0.09	0.02	3.98	0.22	0.70	0.03	5.84	0.11	0.12	0.01	31.91	0.35	1.75	0.36	0.03	0.02	99.27	
Van7-85-27	2R	10	55.02	0.22	0.10	0.01	4.10	0.24	0.68	0.02	5.61	0.21	0.12	0.01	31.36	0.61	2.21	0.80	0.05	0.03	99.24	
Van7-85-27	3I	7	55.19	0.14	0.08	0.02	3.88	0.05	0.60	0.04	5.77	0.09	0.11	0.02	32.16	0.25	1.45	0.22	0.04	0.02	99.27	
Van7-85-30	1C	9	54.88	0.29	0.07	0.01	4.51	0.19	0.61	0.01	6.04	0.18	0.11	0.01	31.49	0.40	1.90	0.46	0.02	0.02	99.64	
Van7-85-30	IR	8	55.97	0.22	0.06	0.01	3.30	0.31	0.40	0.05	6.18	0.12	0.12	0.03	32.72	0.33	1.23	0.42	0.01	0.01	99.98	
Van7-85-32	3C	11	54.42	0.25	0.07	0.02	5.06	0.12	0.68	0.03	6.13	0.21	0.13	0.01	31.74	0.40	1.64	0.59	0.01	0.01	99.87	
Van7-85-32	3R	8	55.33	0.42	0.08	0.01	4.02	0.13	0.52	0.02	6.19	0.16	0.13	0.01	31.49	0.32	1.40	0.15	0.04	0.02	99.18	
Van7-85-39	1C	9	53.97	0.23	0.07	0.01	4.84	0.11	0.66	0.02	6.23	0.18	0.10	0.01	31.68	0.44	1.81	0.52	0.00	0.01	99.35	
Van7-85-39	IR	10	54.47	0.30	0.05	0.01	4.84	0.22	0.67	0.04	6.29	0.18	0.14	0.02	31.75	0.36	1.51	0.27	0.02	0.02	99.74	
Van7-85-42	1C	8	55.15	0.22	0.08	0.01	5.09	0.18	0.72	0.02	6.33	0.12	0.10	0.01	31.83	0.18	1.72	0.24	0.02	0.02	101.04	
Van7-85-42	3C	11	54.87	0.27	0.08	0.01	5.26	0.12	0.73	0.02	6.21	0.17	0.12	0.02	31.49	0.40	2.17	0.56	0.03	0.02	100.95	
Van7-85-42	3R	11	55.20	0.51	0.08	0.01	5.25	0.29	0.71	0.03	6.15	0.17	0.13	0.03	31.17	0.37	2.21	0.44	0.04	0.03	100.94	
Van7-85-47	1C	11	54.43	0.90	0.11	0.06	5.45	0.27	0.83	0.08	5.94	0.94	0.13	0.01	30.12	3.87	3.65	5.26	0.08	0.09	100.74	
Van7-85-47	IR	10	55.10	0.30	0.08	0.02	5.03	0.09	0.69	0.02	6.42	0.15	0.14	0.02	31.90	0.17	1.46	0.29	0.03	0.01	100.85	
Van7-85-49	1C	10	54.06	0.51	0.06	0.02	5.30	0.25	0.78	0.03	6.21	0.17	0.09	0.02	31.43	0.58	2.15	0.65	0.00	0.00	100.09	
Van7-85-49	IR	6	55.12	0.30	0.03	0.02	3.66	0.54	0.50	0.11	6.14	0.18	0.12	0.02	32.56	0.52	1.28	0.38	0.02	0.02	99.42	
Van7-86-25	1C	10	54.96	0.15	0.04	0.01	4.71	0.12	0.79	0.02	5.94	0.09	0.11	0.02	31.57	0.41	1.68	0.35	0.03	0.02	99.82	
Van7-86-25	IR	10	54.67	0.28	0.04	0.01	4.97	0.18	0.83	0.04	6.01	0.10	0.10	0.02	31.32	0.38	1.65	0.27	0.02	0.01	99.59	
Van7-86-27	1C	11	54.31	0.78	0.10	0.06	5.38	0.14	0.82	0.07	6.14	0.93	0.14	0.03	30.88	3.61	2.53	5.02	0.05	0.07	100.34	
Van7-86-27	IR	6	55.40	0.52	0.07	0.02	3.84	0.44	0.56	0.08	6.19	0.25	0.12	0.02	32.40	0.97	1.24	0.23	0.05	0.06	99.86	
Van7-86-28	1C	6	54.14	0.24	0.06	0.01	4.98	0.20	0.77	0.03	6.25	0.16	0.07	0.02	31.52	0.47	1.26	0.44	0.00	0.00	99.05	
Van7-86-28	IR	8	54.38	0.23	0.06	0.01	4.50	0.26	0.71	0.06	6.10	0.31	0.08	0.02	31.33	0.41	1.81	0.50	0.00	0.00	98.99	
Van7-86-30	1C	9	55.18	0.15	0.07	0.02	4.40	0.05	0.68	0.02	6.03	0.08	0.08	0.03	32.50	0.27	1.61	0.33	0.00	0.00	100.54	
Van7-86-30	IR	8	55.41	0.45	0.05	0.01	3.54	0.16	0.52	0.01	5.99	0.15	0.09	0.01	32.47	0.51	1.47	0.22	0.00	0.00	99.56	
Van7-86-37	1C	7	54.73	0.24	0.07	0.01	3.88	0.14	0.96	0.04	5.79	0.15	0.08	0.03	31.35	0.53	2.14	0.36	0.00	0.00	99.00	
Van7-86-37	IR	8	55.19	0.39	0.06	0.01	3.27	0.21	0.77	0.07	5.81	0.08	0.09	0.03	32.38	0.34	1.60	0.21	0.00	0.00	99.14	
Van7-86-37	3C	9	55.01	0.22	0.07	0.01	3.56	0.06	0.82	0.02	5.80	0.10	0.07	0.01	31.74	0.39	2.04	0.37	0.00	0.01	99.11	
Van7-86-37	3R	6	55.27	0.29	0.08	0.01	3.38	0.24	0.76	0.04	5.81	0.20	0.09	0.03	32.17	0.38	1.72	0.43	0.00	0.01	99.29	

Table 5.5: Opx major elements Cont.

Analysis	Gm ^a	Pts	SiO ₂	±	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	FeO	±	MnO	±	MgO	±	CaO	±	Na ₂ O	±	Total	Notes ^b
Van7-96-09	2C	10	53.01	0.46	0.16	0.03	6.33	0.58	0.30	0.02	6.51	0.16	0.16	0.02	30.54	0.58	1.62	0.90	0.02	0.02	98.66	PyxVn
Van7-96-09	2R	8	55.33	0.45	0.14	0.04	3.04	0.40	0.25	0.04	6.48	0.29	0.20	0.01	31.69	1.36	1.76	1.76	0.02	0.03	98.89	PyxVn
Van7-96-14	1C	12	54.53	0.25	0.22	0.04	4.86	0.12	0.38	0.02	7.26	0.35	0.16	0.02	30.48	0.95	1.95	1.41	0.06	0.04	99.90	PyxVn
Van7-96-14	1R	12	55.81	0.27	0.26	0.02	2.66	0.30	0.41	0.02	7.63	0.17	0.18	0.03	32.02	0.23	0.90	0.17	0.03	0.02	99.91	PyxVn
Van7-96-14	2C	17	55.39	0.58	0.30	0.03	3.49	0.68	0.31	0.01	7.03	0.18	0.18	0.03	31.62	0.59	1.30	0.56	0.05	0.02	99.67	PyxVn
Van7-96-14	2R	16	55.78	0.39	0.34	0.02	2.84	0.30	0.45	0.04	6.98	0.09	0.20	0.02	32.11	0.22	0.88	0.23	0.05	0.02	99.62	PyxVn
Van7-96-15	1C	14	54.14	0.16	0.22	0.03	5.26	0.08	0.26	0.01	7.50	0.37	0.17	0.02	30.83	0.61	1.96	0.82	0.04	0.03	100.39	PyxVn
Van7-96-15	1R	12	55.63	0.20	0.24	0.02	3.26	0.22	0.27	0.02	7.70	0.18	0.19	0.02	32.04	0.19	1.19	0.28	0.02	0.02	100.52	PyxVn
Van7-96-16	2C	11	54.19	0.12	0.23	0.02	5.52	0.04	0.55	0.02	6.84	0.16	0.15	0.02	31.21	0.27	1.71	0.47	0.05	0.02	100.46	PyxVn
Van7-96-16	2R	10	54.71	0.19	0.24	0.02	4.54	0.13	0.58	0.02	7.03	0.24	0.16	0.02	32.37	0.13	0.86	0.05	0.03	0.02	100.51	PyxVn
Van7-96-18	1C	7	54.57	0.24	0.13	0.01	5.35	0.10			6.99	0.08	0.15	0.03	32.29	0.52	1.53	0.16	0.01	0.01	101.01	NrMat
Van7-96-18	1R	7	54.40	0.25	0.14	0.02	5.20	0.09			7.00	0.25	0.13	0.03	32.50	0.42	1.39	0.26	0.00	0.01	100.77	NrMat
Van7-96-18	2C	10	53.99	0.35	0.17	0.02	4.65	0.11			7.25	0.11	0.16	0.02	31.96	0.26	1.58	0.33	0.00	0.00	99.75	PyxVn
Van7-96-18	2R	10	55.81	0.32	0.16	0.02	2.19	0.21			7.11	0.10	0.16	0.03	33.21	0.43	1.26	0.14	0.00	0.00	99.91	PyxVn
Van7-96-19	1C	9	53.51	0.40	0.20	0.04	5.45	0.11			6.41	0.22	0.11	0.02	32.19	0.36	1.62	0.55	0.01	0.02	99.50	NrMat
Van7-96-19	1R	10	55.11	0.59	0.21	0.03	2.92	0.38			6.57	0.13	0.12	0.03	33.58	0.39	1.18	0.29	0.00	0.00	99.70	NrMat
Van7-96-19	2I	9	55.42	0.36	0.21	0.03	2.74	0.22			6.95	0.07	0.11	0.02	32.92	0.54	1.32	0.15	0.00	0.00	99.67	PyxVn
Van7-96-21	1C	10	53.79	0.37	0.18	0.05	5.80	0.36	0.63	0.02	6.09	0.32	0.13	0.02	30.80	0.70	1.83	0.87	0.06	0.02	99.31	PyxVn
Van7-96-21	1R	14	55.17	0.89	0.18	0.03	3.83	1.03	0.59	0.04	6.39	0.33	0.13	0.02	32.02	0.65	1.17	0.43	0.04	0.04	99.52	PyxVn
Van7-96-21	2C	12	53.76	0.25	0.15	0.03	6.34	0.19	0.61	0.03	6.37	0.16	0.14	0.02	31.09	0.46	1.22	0.51	0.06	0.03	99.74	AdjVn
Van7-96-21	2R	15	54.08	0.31	0.14	0.02	5.71	0.10	0.62	0.02	6.47	0.14	0.15	0.02	31.51	0.28	1.07	0.26	0.03	0.01	99.78	AdjVn
Van7-96-21	3I	11	56.31	0.35	0.22	0.03	2.28	0.19	0.39	0.04	6.28	0.09	0.15	0.03	33.16	0.26	0.75	0.18	0.01	0.01	99.54	PyxVn
Van7-96-25	1C	9	53.24	0.29	0.11	0.02	5.81	0.11			6.71	0.18	0.09	0.02	31.18	0.41	1.65	0.49	0.06	0.05	98.86	
Van7-96-25	1R	8	53.69	0.16	0.11	0.02	5.35	0.08			6.65	0.07	0.12	0.03	31.82	0.26	1.31	0.15	0.03	0.02	99.08	
Van7-96-28	1C	8	53.63	1.32	0.17	0.18	5.35	1.07			5.78	0.75	0.09	0.02	30.66	3.81	3.38	4.66	0.07	0.17	99.13	
Van7-96-28	1R	6	53.87	0.44	0.11	0.03	5.41	0.31			6.20	0.18	0.11	0.03	32.10	0.73	1.50	0.72	0.01	0.01	99.31	
Van7-96-35	1C	10	54.88	0.27	0.11	0.01	4.16	0.07	1.05	0.01	5.47	0.18	0.12	0.04	31.60	0.41	1.92	0.21	0.13	0.04	99.44	
Van7-96-35	1R	10	55.53	0.15	0.10	0.01	3.66	0.14	0.80	0.01	5.63	0.14	0.11	0.02	32.44	0.17	1.27	0.05	0.03	0.02	99.57	
Van7-96-37	1C	10	54.49	0.27	0.11	0.02	4.82	0.09	0.86	0.04	6.06	0.11	0.14	0.02	31.22	0.47	1.68	0.27	0.03	0.04	99.42	
Van7-96-37	1R	9	54.95	0.30	0.10	0.02	4.43	0.09	0.77	0.03	6.09	0.16	0.17	0.04	31.79	0.26	1.41	0.34	0.01	0.02	99.72	
Van7-96-38	1C	12	53.31	0.21	0.15	0.03	6.88	0.09	0.61	0.02	6.38	0.17	0.14	0.02	30.04	0.57	1.56	0.61	0.10	0.06	99.19	
Van7-96-38	1R	10	53.45	0.19	0.12	0.03	6.47	0.10	0.59	0.02	6.61	0.19	0.12	0.02	30.68	0.24	1.09	0.26	0.03	0.02	99.15	
Van7-96-38	2C	10	53.86	0.23	0.16	0.03	6.21	0.24	0.57	0.02	6.19	0.15	0.12	0.01	30.28	0.45	1.74	0.67	0.11	0.05	99.26	
Van7-96-38	2R	12	53.96	0.25	0.11	0.01	5.89	0.21	0.50	0.02	6.63	0.12	0.14	0.03	30.96	0.22	1.03	0.21	0.02	0.02	99.24	
Van7-96-66B	1C	10	53.18	0.20	0.21	0.03	6.84	0.10	0.41	0.02	6.80	0.13	0.16	0.03	30.26	0.27	1.76	0.49	0.05	0.04	99.65	OpxVn
Van7-96-66B	1R	8	53.83	0.57	0.20	0.05	5.58	0.59	0.38	0.05	6.89	0.13	0.18	0.03	31.39	0.49	1.09	0.48	0.01	0.02	99.56	OpxVn

^a Grain number; core (C), rim (R) or interstitial (I). Interstitial grains are < 1 mm diameter. L preceding a grain number indicates Lee (1997) data for dredge RC27-9-6 samples.

^b Abbreviations for lithologies (blank indicates residual peridotite): Dep. Trans and Enr are depleted, transitional and enriched zones in cryptically metasomatized samples; GbVn refers to a gabbroic vein, PyxVn to a pyroxenite vein, CpxVn to a clinopyroxenite vein, and OpxVn to an orthopyroxenite vein; AdjVn is adjacent to a vein, NrMat is <5 cm from a vein and FarMat is >5 cm from a vein.

Table 5.6: Major element composition of Cpx, in wt%.

Analysis	Gm ^a	Pts	SiO ₂	±	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	FeO	±	MnO	±	MgO	±	CaO	±	Na ₂ O	±	Total	Notes ^b
6K-458-1	1C	9	51.38	0.76	0.23	0.02	5.69	1.19	1.50	0.13	2.20	0.11	0.10	0.02	15.70	0.38	23.12	0.29	0.53	0.04	100.44	
6K-458-1	1R	8	52.36	1.81	0.26	0.08	5.28	1.34	1.35	0.26	2.77	1.42	0.10	0.03	18.77	0.48	19.69	8.07	0.48	0.20	101.05	
6K-458-1	2I	9	51.92	0.59	0.33	0.03	5.37	0.72	1.25	0.05	2.22	0.13	0.10	0.03	16.47	0.30	22.77	0.88	0.50	0.07	100.93	
6K-458-1	3I	9	52.43	0.48	0.31	0.05	4.14	0.60	1.56	0.13	2.21	0.13	0.10	0.03	16.51	0.26	23.24	0.19	0.48	0.03	100.98	
6K-458-3	1I	10	52.09	0.34	0.32	0.02	4.71	0.22	0.80	0.02	2.51	0.15	0.09	0.01	15.97	0.15	22.54	0.55	0.41	0.04	99.43	
6K-458-3	2I	12	52.21	0.28	0.27	0.01	5.66	0.19	0.95	0.05	2.53	0.18	0.09	0.02	15.66	0.39	22.73	0.69	0.47	0.05	100.57	
6K-458-3	3C	13	51.43	1.36	0.24	0.08	7.09	1.05	1.11	0.21	3.21	1.54	0.10	0.03	17.30	5.37	19.68	7.70	0.44	0.16	100.60	
6K-458-3	3R	8	52.78	0.38	0.45	0.05	3.77	0.42	0.93	0.08	2.22	0.07	0.07	0.01	16.34	0.42	23.76	0.38	0.36	0.03	100.67	
6K-458-3	4C	9	51.41	0.68	0.20	0.05	6.48	0.17	1.04	0.06	3.57	0.87	0.11	0.02	18.22	3.15	18.06	4.09	0.42	0.14	99.50	
6K-458-3	4R	8	50.96	0.62	0.27	0.04	6.20	0.55	1.05	0.03	2.36	0.18	0.07	0.01	15.58	0.86	22.51	0.93	0.54	0.18	99.55	
6K-458-3	5I	11	51.69	0.43	0.32	0.04	5.22	0.42	0.92	0.06	2.24	0.12	0.06	0.02	15.47	0.42	23.07	0.31	0.44	0.05	99.42	
6K-458-3	6I	10	51.96	0.31	0.43	0.02	4.50	0.66	1.03	0.03	2.21	0.09	0.10	0.01	15.49	0.42	23.29	0.26	0.47	0.06	99.46	
6K-465-2	1C	8	50.64	0.59	0.23	0.04	6.24	0.06	1.10	0.05	3.13	0.68	0.11	0.02	17.38	2.89	20.57	3.81	0.30	0.09	99.69	Dep
6K-465-2	1R	13	51.99	0.56	0.30	0.02	3.66	0.38	1.28	0.05	2.49	0.17	0.05	0.02	16.64	0.51	23.02	0.81	0.26	0.05	99.71	Dep
6K-465-2	2C	10	51.18	0.39	0.21	0.03	6.45	0.06	1.09	0.04	3.25	0.55	0.03	0.03	17.10	2.25	20.63	3.06	0.30	0.06	100.23	Dep
6K-465-2	2R	10	51.72	0.52	0.24	0.03	5.57	0.42	0.96	0.06	3.07	0.47	0.04	0.02	17.05	1.83	21.50	2.58	0.31	0.06	100.46	Dep
6K-465-2	3C	8	51.25	0.15	0.25	0.03	5.38	0.52	1.03	0.02	2.90	0.13	0.06	0.02	16.40	0.23	22.98	0.63	0.19	0.06	100.45	Dep
6K-465-2	3R	8	51.33	0.48	0.33	0.02	4.75	0.21	1.11	0.03	2.81	0.18	0.07	0.02	16.59	0.63	22.95	0.89	0.22	0.05	100.16	Dep
6K-465-2	4C	10	51.91	0.34	0.26	0.04	4.15	0.40	1.14	0.04	2.67	0.16	0.07	0.01	16.92	0.27	23.01	0.47	0.19	0.04	100.30	Dep
6K-465-2	4R	10	51.60	0.65	0.30	0.03	4.40	0.51	1.14	0.03	2.80	0.12	0.06	0.02	16.62	0.54	22.90	0.70	0.18	0.05	99.99	Dep
6K-465-2	5C	10	50.45	0.43	0.24	0.03	6.41	0.56	1.20	0.03	3.01	0.09	0.10	0.02	15.98	0.30	22.37	0.33	0.41	0.06	100.17	Enr
6K-465-2	6C	9	51.38	0.54	0.18	0.04	6.04	0.34	1.06	0.04	3.78	0.45	0.11	0.01	19.24	2.14	18.03	2.79	0.39	0.05	100.20	Trans
6K-465-2	6R	9	51.15	1.11	0.26	0.06	5.35	0.26	1.10	0.13	2.98	1.18	0.10	0.03	16.84	4.98	21.66	6.63	0.41	0.15	99.85	Trans
6K-465-2	8C	15	51.34	0.49	0.20	0.02	6.46	0.16	1.15	0.03	3.51	0.15	0.10	0.01	18.15	0.43	19.43	0.45	0.38	0.05	100.72	Dep
6K-465-2	8R	18	51.08	1.23	0.26	0.08	5.55	0.31	1.16	0.13	2.84	1.37	0.10	0.02	15.97	5.63	22.89	7.71	0.39	0.20	100.25	Dep
6K-465-2	9C	8	52.59	0.57	0.25	0.03	3.23	0.52	1.26	0.03	2.56	0.08	0.10	0.02	17.01	0.49	23.08	0.24	0.42	0.05	100.51	Enr
6K-465-2	9R	10	52.37	0.75	0.31	0.05	3.52	0.85	1.20	0.11	2.55	0.07	0.11	0.02	16.82	0.39	23.04	0.43	0.40	0.05	100.29	Enr
6K-465-3	5R	10	53.33	0.19	0.63	0.04	1.54	0.06	0.00	0.00	6.97	0.29	0.32	0.03	16.41	0.49	20.25	0.77	0.44	0.03	99.87	GbVn
6K-465-3	6C	8	52.37	0.26	0.57	0.05	1.58	0.05	0.01	0.01	7.90	0.47	0.31	0.02	15.50	0.44	21.25	0.91	0.39	0.06	99.89	GbVn
6K-465-3	8I	14	52.40	0.38	0.69	0.10	3.57	0.58	1.05	0.05	2.87	0.09	0.11	0.02	16.14	0.11	22.25	0.45	0.60	0.06	99.67	NrMat
6K-465-3	9I	9	52.46	0.77	0.69	0.14	3.44	0.65	1.07	0.04	2.83	0.08	0.11	0.03	16.25	0.40	22.20	0.74	0.66	0.14	99.69	NrMat
6K-465-3	10C	8	53.24	0.37	0.42	0.03	2.55	0.13	1.21	0.03	2.72	0.07	0.12	0.03	16.25	0.15	22.53	0.16	0.57	0.03	99.61	NrMat
6K-465-3	10R	12	53.35	0.24	0.38	0.03	2.48	0.15	1.21	0.04	2.89	0.11	0.11	0.02	16.50	0.10	22.36	0.18	0.57	0.03	99.84	NrMat
6K-465-3	11I	13	53.32	0.59	0.48	0.06	2.52	0.26	1.06	0.08	2.91	0.22	0.13	0.04	16.86	0.46	21.94	0.69	0.58	0.04	99.80	NrMat
6K-465-3	12R	14	53.21	0.36	0.46	0.07	3.40	2.04	0.98	0.10	2.73	0.22	0.07	0.02	16.51	1.48	22.22	1.23	0.76	0.45	100.36	NrMat
6K-465-3	13C	12	52.77	0.64	0.50	0.17	3.04	0.57	1.19	0.07	2.85	0.13	0.08	0.02	16.44	0.27	22.08	0.60	0.62	0.18	99.56	NrMat
6K-465-3	13R	8	52.76	0.47	0.55	0.09	2.89	0.40	1.18	0.04	2.87	0.12	0.08	0.03	16.46	0.16	22.15	0.51	0.62	0.06	99.56	NrMat
6K-465-3	15R	9	53.07	0.36	0.37	0.06	3.15	0.76	1.12	0.07	2.48	0.18	0.06	0.02	16.27	0.52	22.50	0.48	0.58	0.16	99.61	NrMat
6K-466-1	1C	10	51.97	0.29	0.26	0.02	5.55	0.10	1.16	0.03	2.51	0.34	0.04	0.02	16.42	1.16	21.88	1.49	0.64	0.05	100.44	
6K-466-1	1R	10	51.94	0.29	0.29	0.02	5.34	0.33	1.19	0.06	2.14	0.06	0.04	0.02	15.89	0.22	23.21	0.22	0.58	0.05	100.64	

Table 5.6: Cpx major elements Cont.

Analysis	Gm ^a	Pts	SiO ₂	TiO ₂	Al ₂ O ₃	Cr ₂ O ₃	FeO	MnO	MgO	CaO	Na ₂ O	Total	Notes ^b						
6K-466-1	3I	13	52.16	0.93	5.18	0.36	1.02	0.18	2.98	1.05	0.13	0.03	17.90	5.84	0.70	0.14	100.05		
6K-643-13	1I	8	51.78	0.30	5.14	0.15	1.05	0.04	2.79	0.20	0.10	0.02	17.03	0.47	0.21	0.90	0.63	100.20	
6K-643-13	2C	10	51.39	0.94	6.15	0.22	1.28	0.12	4.03	1.12	0.12	0.02	20.10	4.98	16.47	6.97	0.18	99.84	
6K-643-13	2R	10	51.43	0.43	5.76	0.07	1.21	0.03	3.11	0.35	0.11	0.02	17.47	1.62	20.44	2.17	0.23	0.04	99.90
6K-651-7	1I	11	51.39	0.57	5.71	0.28	0.97	0.05	2.71	0.83	0.08	0.02	16.51	2.52	21.66	3.45	0.26	0.04	99.50
6K-651-7	2C	11	50.84	0.27	6.27	0.09	1.06	0.05	2.90	0.62	0.09	0.01	16.10	1.83	21.35	2.64	0.28	0.07	99.10
6K-651-7	2R	12	51.38	0.29	5.44	0.21	0.87	0.03	2.46	0.24	0.07	0.01	15.85	0.99	22.63	1.29	0.27	0.05	99.22
JR31-52-03	4C	12	51.13	0.31	6.46	0.10	1.25	0.04	2.66	0.22	0.08	0.01	16.12	0.80	21.46	0.88	0.61	0.05	99.95
JR31-52-03	4R	10	51.25	0.32	6.25	0.08	1.20	0.05	2.60	0.13	0.08	0.02	16.40	0.63	21.27	0.74	0.60	0.03	99.82
JR31-52-03	5C	12	51.73	0.59	6.53	0.16	1.20	0.11	3.69	1.06	0.09	0.02	19.35	3.94	16.93	5.34	0.47	0.18	100.12
JR31-52-03	5R	12	50.81	0.39	6.65	0.10	1.30	0.04	2.60	0.20	0.07	0.03	15.97	0.77	21.54	1.02	0.59	0.05	99.72
JR31-52-03	6I	12	51.21	0.21	6.54	0.06	1.27	0.04	2.63	0.13	0.08	0.03	15.75	0.39	21.66	0.51	0.58	0.05	99.93
JR31-52-03	7C	11	51.30	0.49	6.06	0.08	1.22	0.02	2.62	0.19	0.12	0.02	16.27	0.76	21.20	0.98	0.60	0.04	99.59
JR31-52-03	7R	10	51.57	0.39	5.43	0.26	1.02	0.06	2.41	0.17	0.09	0.02	16.47	0.34	21.88	0.46	0.56	0.05	99.62
JR31-52-03	8I	9	51.81	0.34	5.56	0.10	1.07	0.04	2.65	0.36	0.09	0.02	16.94	1.23	21.14	1.66	0.58	0.05	100.04
JR31-52-03	9I	11	51.10	0.27	5.80	0.11	1.08	0.03	2.57	0.14	0.07	0.03	16.62	0.47	21.25	0.69	0.61	0.02	99.31
JR31-52-04	4C	11	51.82	0.20	5.96	0.25	1.21	0.05	2.23	0.11	0.07	0.02	15.49	0.13	22.28	0.17	0.63	0.04	99.89
JR31-52-04	4R	11	51.84	0.32	5.75	0.52	1.15	0.10	2.32	0.08	0.07	0.01	15.47	0.22	22.55	0.19	0.55	0.04	99.92
JR31-52-05	1C	10	51.62	1.14	6.62	0.28	1.21	0.16	3.15	1.68	0.11	0.03	18.01	6.21	18.67	8.30	0.49	0.22	100.05
JR31-52-05	1R	10	52.19	0.29	5.28	0.54	0.96	0.08	2.19	0.08	0.09	0.02	15.75	0.46	23.15	0.27	0.42	0.07	100.23
JR31-52-05	3C	9	51.79	0.36	5.68	0.22	1.03	0.05	2.28	0.14	0.09	0.01	16.33	0.62	22.06	0.60	0.55	0.06	100.00
JR31-52-05	5I	11	51.78	0.34	5.98	0.18	1.04	0.04	2.64	0.26	0.04	0.02	16.31	1.07	21.46	1.25	0.56	0.04	99.99
JR31-52-05	6I	8	51.57	0.37	5.96	0.09	1.09	0.03	2.51	0.06	0.04	0.03	15.86	0.22	22.06	0.35	0.56	0.04	99.86
JR31-52-08	1C	10	51.18	0.28	6.21	0.11	1.22	0.05	2.77	0.38	0.12	0.02	16.38	1.37	20.87	1.92	0.56	0.08	99.49
JR31-52-08	1R	20	51.55	0.45	6.06	0.38	1.16	0.10	2.60	0.45	0.10	0.02	16.39	1.09	21.21	1.71	0.57	0.07	99.85
JR31-52-08	3I	9	51.71	0.15	5.64	0.10	1.11	0.04	2.62	0.27	0.08	0.02	16.17	0.98	21.21	1.35	0.66	0.07	99.38
JR31-52-08	4R	9	51.91	0.25	5.43	0.28	1.01	0.10	2.46	0.08	0.09	0.02	16.19	0.35	21.80	0.40	0.62	0.06	99.74
Kn162-19-41	1C	5	50.85	0.45	6.38	0.20	0.87	0.01	2.53	0.14	0.11	0.02	15.50	0.42	23.12	0.50	0.24	0.02	99.86
Kn162-19-41	IR	8	51.35	0.48	6.21	0.15	0.85	0.05	2.71	0.54	0.09	0.02	15.82	2.10	22.56	2.68	0.23	0.04	100.06
Kn162-19-41	2I	10	51.09	0.27	6.29	0.37	0.89	0.05	2.57	0.14	0.10	0.02	15.87	0.41	22.43	0.65	0.25	0.03	99.76
Kn162-19-41	3I	9	50.73	0.35	6.37	0.46	0.87	0.07	2.76	0.17	0.10	0.04	16.18	0.66	22.38	0.79	0.22	0.03	99.90
Kn162-19-41	4C	7	50.49	2.08	6.99	2.77	0.95	0.25	2.98	0.57	0.11	0.02	16.48	0.99	21.66	1.67	0.21	0.01	100.09
Kn162-19-41	4R	10	51.46	0.35	5.65	0.28	0.88	0.04	2.50	0.19	0.09	0.02	15.86	0.70	23.00	0.83	0.21	0.03	99.90
Kn162-47-08	1I	10	50.60	0.46	6.97	0.11	1.03	0.03	2.99	0.15	0.12	0.02	15.81	0.33	21.80	0.44	0.51	0.06	100.21
Kn162-47-08	2I	10	51.38	0.38	6.17	0.10	0.92	0.03	3.01	0.19	0.11	0.02	16.18	0.66	21.51	0.78	0.53	0.06	100.12
Kn162-47-08	3I	15	51.92	0.39	6.05	0.27	0.87	0.07	3.05	0.12	0.07	0.02	16.13	0.30	21.57	0.38	0.66	0.05	100.66
Kn162-47-08	5I	11	51.95	0.61	6.13	0.16	0.93	0.05	3.27	0.58	0.07	0.02	16.92	1.95	20.32	2.60	0.62	0.10	100.53
Kn162-47-09	2I	12	53.20	0.37	4.40	0.17	1.51	0.05	2.18	0.16	0.08	0.02	16.71	0.69	21.70	0.77	0.90	0.04	100.80
Kn162-47-09	3C	8	52.89	0.67	4.62	0.11	1.75	0.04	2.11	0.08	0.07	0.02	16.31	0.30	21.90	0.30	0.98	0.07	100.73
Kn162-47-09	3R	6	53.03	0.58	4.41	0.43	1.62	0.13	2.23	0.26	0.08	0.02	15.83	0.88	21.75	1.00	0.81	0.07	99.85
Kn162-47-09	6I	7	53.36	0.24	3.86	0.06	1.32	0.03	2.21	0.15	0.08	0.02	17.13	0.71	21.68	0.69	0.82	0.06	100.54

Table 5.6: Cpx major elements Cont.

Analysis	Gm ^a	Pts	SiO ₂	±	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	FeO	±	MnO	±	MgO	±	CaO	±	Na ₂ O	±	Total	Notes ^b
Kn162-47-10	1C	10	51.45	1.22	0.20	0.08	6.65	0.61	0.87	0.14	4.01	1.60	0.13	0.04	19.09	6.13	16.85	8.28	0.29	0.16	99.54	PyxVn
Kn162-47-10	1R	10	50.86	0.32	0.22	0.03	6.77	0.30	0.93	0.05	2.92	0.07	0.10	0.01	15.60	0.67	21.77	0.76	0.43	0.03	99.60	PyxVn
Kn162-47-10	2I	7	51.38	0.46	0.24	0.04	5.16	0.67	0.64	0.12	2.89	0.17	0.10	0.02	16.22	0.39	22.26	0.32	0.32	0.04	99.21	PyxVn
Kn162-47-10	5I	10	51.09	0.36	0.23	0.02	6.05	0.48	0.76	0.08	3.01	0.43	0.11	0.03	16.12	0.84	21.68	1.02	0.37	0.05	99.43	NrMat
Kn162-47-10	6C	7	50.98	0.18	0.27	0.02	6.22	0.35	1.05	0.07	3.08	0.41	0.12	0.06	15.57	0.28	21.91	0.30	0.49	0.06	99.70	FarMat
Kn162-47-10	6R	7	51.30	0.37	0.29	0.02	5.74	0.42	0.96	0.09	2.83	0.21	0.10	0.02	15.57	0.30	22.16	0.15	0.55	0.02	99.50	FarMat
Kn162-47-10	7I	4	51.32	0.27	0.26	0.05	5.90	0.39	0.92	0.06	2.72	0.23	0.08	0.04	15.96	0.18	21.99	0.33			99.19	FarMat
Kn162-47-12	1I	10	51.24	0.90	0.25	0.04	5.94	0.80	1.02	0.15	2.79	0.79	0.09	0.02	16.46	2.61	21.62	3.26	0.41	0.10	99.82	PyxVn
Kn162-47-12	2C	10	51.06	0.82	0.23	0.05	6.16	0.20	0.99	0.07	3.38	0.94	0.11	0.03	17.25	3.40	19.62	4.64	0.38	0.10	99.18	PyxVn
Kn162-47-12	2R	8	50.92	0.27	0.26	0.02	6.17	0.11	0.92	0.02	2.96	0.29	0.10	0.02	16.08	0.93	21.89	1.49	0.44	0.05	99.73	PyxVn
Kn162-47-22	1C	9	51.26	0.47	0.22	0.02	6.75	0.24	0.99	0.03	3.06	0.11	0.09	0.02	15.50	0.30	22.20	0.28	0.51	0.06	100.57	AdjVn
Kn162-47-22	1R	8	51.82	0.39	0.24	0.04	6.51	0.15	0.93	0.04	3.24	0.74	0.10	0.03	16.72	2.27	20.78	3.06	0.48	0.10	100.82	AdjVn
Kn162-47-22	2C	8	50.84	0.28	0.27	0.02	7.26	0.13	1.08	0.03	3.21	0.33	0.10	0.03	15.80	0.79	21.56	1.04	0.49	0.04	100.61	PyxVn
Kn162-47-22	2R	9	50.85	0.23	0.29	0.02	7.19	0.14	1.09	0.04	3.07	0.27	0.11	0.03	16.10	0.91	21.55	1.08	0.47	0.07	100.73	PyxVn
Kn162-47-22	3C	10	50.92	0.31	0.24	0.03	7.23	0.24	1.05	0.05	3.55	0.47	0.12	0.01	16.82	1.51	20.17	2.06	0.36	0.08	100.47	PyxVn
Kn162-47-22	3R	9	50.64	0.63	0.26	0.02	6.92	0.24	1.04	0.05	3.13	0.12	0.10	0.03	16.05	0.42	21.63	0.51	0.40	0.03	100.17	PyxVn
Kn162-47-22	4I	10	50.66	0.26	0.26	0.02	7.37	0.20	1.05	0.03	3.31	0.18	0.10	0.03	15.98	0.71	21.39	0.90	0.38	0.06	100.49	PyxVn
Kn162-47-24	1C	6	50.83	1.40	0.19	0.08	6.17	0.58	1.07	0.20	3.65	1.84	0.11	0.03	18.25	6.65	18.54	8.88	0.40	0.21	99.21	
Kn162-47-24	1R	7	51.64	1.07	0.21	0.06	4.14	1.59	0.80	0.24	2.50	0.35	0.10	0.03	16.39	0.57	23.05	1.20	0.39	0.23	99.22	
Kn162-47-24	2C	9	50.98	0.78	0.19	0.05	6.35	0.29	1.11	0.08	3.55	1.30	0.11	0.04	18.24	4.51	18.38	6.16	0.38	0.16	99.29	
Kn162-47-24	2R	10	51.53	1.00	0.22	0.06	5.59	0.56	0.96	0.15	3.01	1.15	0.08	0.03	17.67	4.38	20.42	5.94	0.33	0.10	99.79	
Kn162-47-24	3S	10	51.12	0.48	0.25	0.04	6.25	0.04	0.97	0.07	3.04	0.77	0.09	0.03	16.79	2.60	20.66	3.54	0.40	0.10	99.51	
Kn162-47-25	1I	7	51.07	0.22	0.43	0.03	6.43	0.05	1.15	0.01	2.62	0.16	0.11	0.03	15.49	0.32	21.36	0.47	0.79	0.05	99.45	NrMat
Kn162-47-25	2I	10	51.39	0.41	0.44	0.02	5.88	0.09	1.09	0.02	2.70	0.11	0.10	0.02	16.06	0.44	21.01	0.51	0.74	0.04	99.41	NrMat
Kn162-47-25	11C	10	51.20	0.43	0.40	0.02	6.38	0.19	0.98	0.02	3.34	0.35	0.13	0.02	16.40	1.07	20.75	1.39	0.57	0.05	100.13	PyxVn
Kn162-47-25	11R	8	51.00	0.18	0.44	0.02	6.48	0.10	1.01	0.03	3.18	0.18	0.12	0.02	15.81	0.57	21.18	0.61	0.59	0.05	99.82	PyxVn
Kn162-47-25	13I	8	50.81	0.15	0.43	0.04	6.39	0.25	1.03	0.03	3.00	0.25	0.12	0.03	15.53	0.99	21.76	1.35	0.60	0.08	99.65	PyxVn
Kn162-47-25	14I	10	50.66	0.66	0.43	0.02	6.65	0.19	1.03	0.02	2.93	0.10	0.12	0.03	15.17	0.48	22.02	0.41	0.57	0.05	99.58	PyxVn
Kn162-47-26	1C	9	50.70	0.50	0.23	0.02	6.97	0.17	1.02	0.03	2.88	0.12	0.10	0.02	15.49	0.31	21.82	0.42	0.47	0.04	99.68	
Kn162-47-26	1R	4	50.61	0.30	0.30	0.04	6.35	0.36	1.05	0.02	2.70	0.09	0.07	0.03	15.32	0.45	22.44	0.61	0.53	0.07	99.36	
Kn162-47-26	2I	7	51.82	0.34	0.24	0.02	5.49	0.07	0.87	0.06	2.82	0.19	0.09	0.02	16.23	0.56	21.83	0.76	0.48	0.05	99.87	
Kn162-47-26	3I	9	51.60	0.33	0.22	0.01	6.00	0.34	0.83	0.02	2.95	0.17	0.08	0.02	16.49	0.30	21.34	0.44	0.44	0.03	99.97	
Kn162-47-29	1C	8	51.44	0.66	0.23	0.06	6.90	0.48	1.12	0.08	3.00	1.14	0.07	0.02	17.68	3.90	19.25	5.21	0.37	0.12	100.05	
Kn162-47-29	1R	10	52.00	0.57	0.23	0.06	6.28	0.29	0.98	0.06	2.78	1.13	0.07	0.04	17.38	4.32	20.29	5.90	0.33	0.10	100.35	
Kn162-47-29	2R	10	51.73	0.20	0.26	0.02	6.21	0.18	0.97	0.04	2.56	0.47	0.07	0.01	16.18	1.35	21.74	1.84	0.35	0.05	100.06	
Kn162-47-29	3I	9	52.06	0.40	0.26	0.02	5.72	0.13	0.94	0.03	2.47	0.16	0.06	0.01	16.23	0.68	22.04	0.91	0.32	0.02	100.08	
Kn162-47-36	1C	9	51.38	0.28	0.22	0.03	6.60	0.16	1.12	0.05	3.52	0.75	0.12	0.02	17.94	2.64	19.63	3.48	0.36	0.13	100.90	
Kn162-47-36	1R	7	51.01	0.29	0.25	0.03	6.25	0.46	1.05	0.08	3.00	0.17	0.11	0.04	15.85	0.24	22.70	0.21	0.37	0.06	100.60	
Kn162-47-36	2C	9	51.71	1.02	0.21	0.08	5.87	1.00	1.00	0.16	3.77	1.67	0.11	0.03	18.98	5.55	18.42	7.61	0.32	0.18	100.40	
Kn162-47-36	2R	8	51.12	0.51	0.30	0.04	5.70	0.43	0.92	0.08	2.99	0.35	0.09	0.01	16.21	0.37	22.52	0.92	0.38	0.10	100.25	
Kn162-47-36	3I	10	51.24	0.60	0.26	0.02	6.07	0.22	0.98	0.04	3.13	0.15	0.11	0.03	16.19	0.32	22.18	0.28	0.42	0.06	100.58	

Table 5.6: Cpx major elements Cont.

Analysis	Gm ^a	Pts	SiO ₂	TiO ₂	Al ₂ O ₃	Cr ₂ O ₃	FeO	MnO	MgO	CaO	Na ₂ O	Total	Notes ^b
Kn162-47-49	II	8	51.19	0.73	5.59	0.33	0.90	0.08	2.56	0.20	0.06	0.02	99.77
Kn162-47-49	2C	6	51.85	1.23	5.41	0.38	1.03	0.15	3.50	2.09	0.10	0.06	99.87
Kn162-47-49	2R	9	51.03	0.46	5.99	0.58	1.05	0.05	2.67	0.23	0.06	0.01	99.69
Kn162-47-49	3I	10	51.21	0.39	6.17	0.20	1.04	0.03	2.74	0.46	0.06	0.02	99.80
PS86-6-26	1C	9	51.39	0.52	7.37	0.39	0.87	0.07	3.14	0.52	0.08	0.03	99.53
PS86-6-26	1R	9	51.03	0.60	7.16	0.25	0.88	0.04	2.68	0.17	0.07	0.02	99.80
PS86-6-26	2I	10	51.25	0.51	7.21	0.16	0.86	0.03	2.88	0.33	0.08	0.02	99.85
PS86-6-26	3C	10	51.29	0.34	7.64	0.16	0.90	0.06	3.28	0.73	0.08	0.02	99.72
PS86-6-26	3R	10	51.31	0.30	7.50	0.13	0.86	0.01	2.84	0.20	0.07	0.02	99.98
PS86-6-38	1C	6	50.77	0.29	7.70	0.14	0.96	0.03	3.14	0.18	0.14	0.02	100.46
PS86-6-38	1R	7	51.54	0.54	7.04	0.36	0.95	0.04	2.98	0.22	0.11	0.03	100.99
PS86-6-38	2I	6	51.68	0.75	7.16	0.18	0.90	0.03	3.32	0.44	0.13	0.04	101.03
PS86-6-38	3C	7	51.19	0.42	6.92	0.11	0.92	0.02	3.05	0.10	0.09	0.02	99.74
PS86-6-38	3R	7	50.82	0.31	7.34	0.10	0.92	0.04	3.12	0.16	0.13	0.03	99.95
PS86-6-39	1C	7	51.42	0.19	6.99	0.20	1.38	0.03	2.90	0.11	0.10	0.03	100.23
PS86-6-39	1R	8	51.36	0.51	6.76	0.16	1.33	0.05	2.94	0.24	0.10	0.03	99.69
PS86-6-39	2I	9	51.60	0.38	6.32	0.09	1.19	0.03	2.84	0.20	0.13	0.03	100.04
PS86-6-39	3C	7	51.35	0.41	6.68	0.21	1.39	0.04	2.82	0.25	0.10	0.03	99.84
PS86-6-39	3R	7	51.21	0.26	6.75	0.16	1.32	0.04	2.98	0.35	0.11	0.02	99.80
PS86-6-39	4I	8	51.45	0.23	6.54	0.11	1.34	0.02	2.79	0.20	0.09	0.03	99.73
PS86-6-40	II	13	52.24	0.56	6.39	0.28	0.92	0.05	3.19	0.66	0.07	0.02	100.56
PS86-6-40	2C	13	51.00	0.47	7.74	0.29	1.05	0.03	3.17	0.23	0.09	0.02	100.09
PS86-6-40	2R	5	50.96	0.13	7.48	0.15	1.02	0.05	2.78	0.05	0.11	0.02	100.43
PS86-6-40	3C	5	51.12	0.47	7.07	0.12	0.99	0.04	2.91	0.16	0.13	0.03	100.12
PS86-6-40	3R	7	51.30	0.34	6.53	0.29	0.92	0.05	2.96	0.29	0.13	0.05	100.09
RC27-6-9-2	L1C	20	52.3	0.4	4.76	0.10	1.62	0.04	3.41	0.16	0.12	0.02	FarMat
RC27-6-9-2	L1R	10	51.8	1.0	4.18	0.16	1.59	0.17	2.62	1.12	0.11	0.03	FarMat
RC27-6-9-2	L2C	10	52.1	0.5	4.85	0.19	1.59	0.11	2.92	0.67	0.10	0.02	FarMat
RC27-6-9-2	L2R	10	51.9	1.9	4.99	0.205	1.49	0.24	2.72	0.70	0.08	0.03	FarMat
RC27-9-6-2	L3R	10	52.0	0.4	4.78	0.16	1.52	0.09	3.08	0.36	0.12	0.03	CpxVn
RC27-9-6-2	L4I	20	52.3	1.0	4.69	0.32	1.66	0.32	3.28	1.17	0.13	0.03	CpxVn
RC27-9-6-2	L5I	10	51.5	0.2	4.99	0.09	1.59	0.04	2.72	0.17	0.11	0.02	CpxVn
RC27-9-6-2	L6C	10	51.9	1.2	4.82	0.31	1.49	0.18	3.18	1.11	0.10	0.02	CpxVn
RC27-9-6-2	L7C	20	52.2	0.4	4.30	0.16	1.25	0.05	2.72	0.44	0.11	0.03	NrMat
RC27-9-6-2	L11C	11	52.3	1.0	4.81	0.27	1.59	0.19	3.24	0.99	0.10	0.03	NrMat
RC27-9-6-2	L11R	6	52.1	0.2	4.41	0.20	1.49	0.02	2.59	0.06	0.09	0.02	FarMat
RC27-9-6-2	L14C	5	52.5	1.3	5.21	0.19	1.67	0.27	3.75	1.70	0.09	0.02	FarMat
RC27-9-6-2	L15C	5	52.1	1.0	5.40	0.13	1.71	0.15	3.17	1.20	0.10	0.02	NrMat
RC27-9-6-2	L16C	10	52.2	0.6	5.23	0.18	1.51	0.11	4.17	0.73	0.14	0.02	CpxVn
RC27-9-6-2	L16R	10	52.5	0.6	5.05	0.18	1.50	0.14	3.81	0.67	0.13	0.02	CpxVn
RC27-9-6-2	L18I	5	51.7	0.3	5.48	0.18	1.72	0.06	2.89	0.12	0.10	0.02	AdjVn

Table 5.6: Cpx major elements Cont.

Analysis	Gm ^a	Pts	SiO ₂	±	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	FeO	±	MnO	±	MgO	±	CaO	±	Na ₂ O	±	Total	Notes ^b
RC27-9-6-2	L19R	5	52.0	0.9	0.25	0.05	5.32	0.22	1.62	0.12	3.46	1.05	0.09	0.03	17.90	3.40	19.80	4.40	0.61	0.17	101.05	AdjVn
RC27-9-6-2	L20I	5	52.4	0.7	0.23	0.06	5.07	0.38	1.57	0.20	3.36	0.90	0.10	0.02	18.60	3.50	19.50	4.40	0.43	0.27	101.19	NrMat
RC27-9-6-2	L21C	5	52.6	1.2	0.17	0.07	4.94	0.32	1.50	0.27	3.48	1.58	0.09	0.03	19.90	6.90	18.00	8.80	0.56	0.28	101.31	NrMat
RC27-9-6-2	L21R	5	52.2	0.3	0.24	0.01	4.95	0.20	1.52	0.04	2.57	0.20	0.07	0.01	16.30	0.70	22.90	1.20	0.67	0.03	101.42	NrMat
RC27-9-6-2	L22C	10	53.0	0.9	0.10	0.12	4.91	0.26	1.46	0.08	4.26	1.07	0.11	0.03	22.50	0.05	14.40	0.10	0.48	0.01	101.22	NrMat
RC27-9-6-2	L22R	5	51.6	1.5	0.23	0.07	5.28	0.43	1.42	0.30	3.12	1.90	0.09	0.03	17.60	7.50	20.50	9.60	0.70	0.33	100.54	NrMat
RC27-9-6-2	L24C	5	52.3	0.4	0.14	0.07	4.88	0.43	1.57	0.33	3.71	1.75	0.11	0.04	20.30	7.10	15.50	9.30	0.49	0.36	99.00	NrMat
RC27-9-6-2	L24R	5	52.3	0.5	0.19	0.07	4.89	0.51	1.57	0.15	3.19	1.17	0.09	0.04	18.90	4.70	19.00	5.80	0.56	0.25	100.69	NrMat
RC27-9-6-2	L27C	5	53.3	1.9	0.14	0.09	4.75	0.32	1.35	0.33	4.22	2.23	0.09	0.04	22.90	9.60	14.20	12.30	0.43	0.39	101.38	NrMat
RC27-9-6-2	L27R	5	52.2	0.4	0.23	0.01	5.03	0.05	1.53	0.04	2.78	0.45	0.07	0.02	16.60	1.40	22.00	2.60	0.64	0.11	101.08	NrMat
RC27-9-6-2	L2C	10	52.21	1.12	0.12	0.04	4.93	0.28	1.56	0.20	3.37	1.16	0.11	0.03	18.77	4.59	18.20	6.20	0.66	0.25	99.93	FarMat
RC27-9-6-2	L2R	10	52.62	1.08	0.14	0.03	4.43	0.53	1.49	0.23	2.76	0.87	0.10	0.02	16.93	3.75	21.62	4.65	0.59	0.17	100.67	FarMat
RC27-9-6-2	L3I	7	51.91	0.28	0.14	0.01	4.53	0.06	1.49	0.04	2.92	0.21	0.08	0.03	16.46	0.57	21.02	0.68	0.72	0.06	99.27	FarMat
RC27-9-6-2	L4I	11	52.03	0.35	0.15	0.01	4.38	0.08	1.46	0.06	2.87	0.45	0.09	0.02	17.00	1.50	20.53	1.94	0.72	0.08	99.21	FarMat
RC27-9-6-2	L5I	7	52.41	0.12	0.16	0.01	3.99	0.06	1.23	0.03	2.73	0.20	0.09	0.01	16.34	0.55	21.72	0.73	0.67	0.06	99.33	FarMat
RC27-9-6-2	L21C	9	52.69	1.46	0.15	0.08	4.92	0.52	1.36	0.27	4.59	1.77	0.13	0.03	22.70	6.47	12.91	8.71	0.50	0.29	99.94	Cpx Vn
RC27-9-6-2	L21R	10	51.58	0.17	0.21	0.02	5.28	0.09	1.67	0.04	2.52	0.08	0.10	0.01	15.19	0.47	22.82	0.25	0.85	0.05	100.22	Cpx Vn
RC27-9-6-2	L22C	9	51.46	0.35	0.22	0.01	5.15	0.09	1.62	0.03	2.78	0.17	0.09	0.02	15.87	0.51	22.34	0.54	0.73	0.07	100.26	Cpx Vn
RC27-9-6-2	L23I	10	51.43	0.44	0.21	0.03	5.07	0.40	1.50	0.08	2.91	0.76	0.10	0.02	16.60	1.98	21.55	2.96	0.77	0.16	100.14	Cpx Vn
RC27-9-6-5	L1C	8	51.00	1.13	0.16	0.08	6.75	0.16	1.08	0.16	3.96	1.95	0.12	0.03	19.09	6.28	16.99	8.82	0.46	0.17	99.60	Enr
RC27-9-6-5	L1R	5	51.27	0.81	0.27	0.08	4.38	1.76	1.16	0.02	2.66	0.35	0.10	0.03	16.35	0.83	22.65	1.35	0.44	0.08	99.28	Enr
RC27-9-6-5	L2C	8	51.26	1.08	0.17	0.07	6.67	0.19	1.14	0.13	3.71	1.63	0.11	0.04	18.34	5.21	18.74	7.56	0.36	0.15	100.51	Trans
RC27-9-6-5	L2R	9	51.44	0.71	0.49	0.13	5.65	0.35	1.09	0.04	3.14	0.36	0.09	0.02	17.15	1.22	21.39	1.54	0.42	0.06	100.86	Trans
RC27-9-6-5	L3I	6	52.40	0.80	0.49	0.19	2.50	0.80	0.96	0.33	2.23	0.12	0.08	0.02	17.19	0.60	23.37	0.52	0.45	0.16	99.68	Enr
RC27-9-6-5	L4C	13	51.57	0.25	0.31	0.04	5.73	0.11	1.08	0.03	3.04	0.45	0.04	0.02	16.25	1.31	21.76	1.89	0.50	0.05	100.28	Enr
RC27-9-6-5	L4R	10	52.24	0.71	0.47	0.05	3.61	0.89	0.98	0.09	2.80	0.29	0.10	0.01	16.41	0.82	22.22	0.87	0.39	0.06	99.23	Enr
RC27-9-6-5	L5C	8	51.76	1.33	0.34	0.11	5.79	0.57	1.07	0.15	3.91	1.75	0.11	0.04	19.36	5.68	17.42	7.94	0.60	0.27	100.35	Enr
RC27-9-6-5	L5R	11	52.45	1.08	0.55	0.15	3.87	0.84	1.19	0.20	2.58	0.49	0.04	0.03	17.39	2.14	21.28	2.57	0.71	0.25	100.06	Enr
RC27-9-6-5	L11C	10	51.3	0.6	0.24	0.03	6.23	1.06	1.13	0.08	3.11	0.29	0.09	0.02	16.30	0.87	22.06	1.67	0.39	0.23	100.85	
RC27-9-6-5	L11R	10	51.8	0.6	0.25	0.03	5.55	0.89	0.88	0.08	2.82	0.44	0.07	0.02	16.46	1.10	22.64	1.73	0.35	0.10	100.82	
RC27-9-6-7	L1C	10	51.59	1.43	0.16	0.09	6.33	0.27	1.07	0.18	4.21	2.05	0.14	0.02	20.46	7.42	15.56	10.04	0.33	0.23	99.85	
RC27-9-6-7	L1R	10	51.34	0.95	0.22	0.05	5.86	0.53	1.02	0.14	3.05	0.93	0.11	0.02	16.88	3.37	21.22	4.40	0.35	0.09	100.05	
RC27-9-6-7	L3I	10	51.49	0.38	0.23	0.03	5.76	0.12	0.93	0.06	3.09	0.56	0.08	0.02	17.04	2.04	21.26	2.57	0.39	0.09	100.27	
RC27-9-6-7	L5I	7	50.83	0.99	0.25	0.04	5.50	0.93	0.92	0.17	2.78	0.22	0.09	0.02	16.39	0.55	22.59	1.03	0.41	0.09	99.78	
Van7-78-25	L1C	14	52.72	1.05	0.32	0.08	4.48	0.34	1.23	0.17	3.26	1.28	0.13	0.03	18.41	5.06	19.82	6.89	0.40	0.15	100.78	GbVn
Van7-78-25	L1R	13	53.42	0.84	0.37	0.07	3.25	0.56	1.30	0.12	2.77	0.64	0.10	0.02	17.79	2.71	21.53	3.39	0.41	0.09	100.93	GbVn
Van7-78-25	L2C	13	53.32	0.32	0.32	0.02	3.04	0.07	1.31	0.04	2.51	0.22	0.11	0.02	16.99	0.70	22.67	1.03	0.41	0.03	100.67	GbVn
Van7-78-25	L2R	10	53.91	0.14	0.33	0.03	2.35	0.10	1.15	0.07	2.53	0.10	0.10	0.03	17.10	0.10	23.14	0.31	0.37	0.03	100.98	GbVn
Van7-78-25	L3C	9	52.21	0.53	0.33	0.02	4.89	0.10	1.24	0.04	2.66	0.21	0.12	0.03	16.57	0.74	22.40	0.91	0.43	0.05	100.85	GbVn
Van7-78-25	L3R	11	53.53	0.23	0.35	0.03	2.91	0.16	1.30	0.05	2.32	0.09	0.11	0.02	16.94	0.11	23.00	0.27	0.45	0.03	100.89	GbVn
Van7-78-25	L4C	9	52.89	1.13	0.35	0.10	4.39	0.33	1.27	0.19	3.27	1.43	0.12	0.02	18.69	5.02	19.38	6.91	0.38	0.13	100.74	GbVn

Table 5.6: Cpx major elements Cont.

Analysis	Gm ^a	Pts	SiO ₂	TiO ₂	Al ₂ O ₃	Cr ₂ O ₃	FeO	MnO	MgO	CaO	Na ₂ O	Total	Notes ^b									
Van7-78-25	4R	11	53.20	1.32	0.44	0.10	3.27	0.47	1.36	0.12	0.04	18.75	5.06	20.26	6.91	0.38	0.14	100.81	GbVn			
Van7-78-25	5C	10	52.42	0.33	0.34	0.03	4.42	0.37	1.18	0.06	2.88	0.53	1.54	22.00	2.28	0.39	0.07	100.64	GbVn			
Van7-78-25	5R	13	53.64	0.44	0.34	0.05	2.79	0.36	1.14	0.09	2.44	0.15	0.11	0.03	17.12	0.34	22.83	0.40	0.41	0.05	100.82	GbVn
Van7-78-31	1I	14	53.06	0.77	0.47	0.03	2.95	0.24	1.19	0.17	2.59	0.09	0.10	0.03	16.51	0.44	23.12	0.30	0.46	0.05	100.45	NrMat
Van7-78-31	2C	10	52.47	0.41	0.74	0.08	3.16	0.39	1.13	0.02	4.12	0.81	0.14	0.04	17.76	2.27	20.76	3.16	0.40	0.05	99.67	GbVn
Van7-78-31	2R	10	52.48	0.57	0.81	0.09	2.95	0.31	1.13	0.03	3.93	0.73	0.13	0.03	17.74	2.15	21.10	2.86	0.37	0.08	99.63	GbVn
Van7-78-31	3R	11	52.75	0.50	0.75	0.06	2.49	1.10	1.13	0.02	3.67	0.24	0.12	0.02	16.52	0.87	22.89	0.34	0.33	0.04	99.66	GbVn
Van7-78-31	4I	12	52.79	0.20	0.80	0.03	2.34	0.13	1.17	0.01	4.22	0.15	0.13	0.02	16.74	0.13	22.39	0.19	0.39	0.05	99.95	GbVn
Van7-78-31	5I	13	52.55	1.05	0.71	0.06	2.61	1.49	1.13	0.02	3.75	0.27	0.13	0.02	16.69	0.54	22.58	0.58	0.35	0.03	99.50	GbVn
Van7-78-36	10C	12	52.48	0.15	0.49	0.03	3.63	0.13	1.38	0.03	2.77	0.15	0.06	0.02	16.52	0.08	23.10	0.19	0.44	0.04	100.88	FarMat
Van7-78-36	11I	12	53.28	0.23	0.45	0.04	2.70	0.10	1.08	0.03	2.58	0.12	0.08	0.02	17.04	0.27	23.44	0.32	0.41	0.03	101.06	FarMat
Van7-78-36	12C	12	52.51	0.20	0.53	0.03	3.62	0.13	1.42	0.03	2.68	0.12	0.08	0.02	16.48	0.15	23.28	0.22	0.47	0.05	101.06	FarMat
Van7-78-36	12R	11	53.08	0.24	0.44	0.02	2.54	0.06	1.12	0.03	2.61	0.06	0.09	0.01	17.13	0.09	23.55	0.14	0.40	0.03	100.95	FarMat
Van7-78-36	1C	11	51.99	0.73	0.66	0.10	3.90	0.34	1.23	0.03	3.79	0.18	0.11	0.02	16.42	0.57	22.70	0.77	0.49	0.16	100.29	GbVn
Van7-78-36	1R	12	52.37	0.15	0.77	0.12	3.31	0.17	1.25	0.05	3.59	0.16	0.13	0.01	16.28	0.53	23.21	0.49	0.45	0.05	100.36	GbVn
Van7-78-36	3I	11	52.86	0.18	0.69	0.04	2.86	0.06	0.95	0.02	3.20	0.10	0.09	0.02	16.75	0.14	22.78	0.24	0.51	0.03	100.69	AdjVn
Van7-78-36	4I	13	52.98	0.27	0.71	0.07	2.39	0.20	1.27	0.02	3.65	0.14	0.13	0.03	17.02	0.42	22.70	0.50	0.39	0.04	100.26	GbVn
Van7-78-40	1C	13	51.12	0.41	0.30	0.03	6.10	0.38	1.23	0.04	2.92	0.06	0.08	0.02	15.69	0.28	22.86	0.55	0.43	0.12	100.74	
Van7-78-40	1R	12	52.02	0.21	0.52	0.02	4.56	0.24	1.22	0.03	2.77	0.06	0.10	0.02	16.19	0.27	22.98	0.33	0.40	0.04	100.75	
Van7-78-40	2I	10	52.27	0.21	0.56	0.02	4.03	0.09	1.24	0.02	2.71	0.07	0.10	0.02	16.52	0.13	22.94	0.22	0.40	0.03	100.76	
Van7-78-40	3I	12	53.51	0.18	0.45	0.01	2.56	0.09	0.89	0.03	2.49	0.12	0.09	0.02	17.36	0.16	22.91	0.26	0.39	0.03	100.66	
Van7-78-41	1C	12	52.24	0.51	0.25	0.03	4.05	0.16	1.33	0.07	2.98	0.47	0.09	0.02	16.78	1.78	22.33	2.34	0.36	0.05	100.41	
Van7-78-41	1R	13	53.40	0.27	0.25	0.01	2.53	0.15	0.94	0.07	2.57	0.07	0.08	0.02	16.99	0.11	23.48	0.15	0.31	0.04	100.56	
Van7-78-41	2C	12	52.53	0.18	0.24	0.02	3.46	0.07	1.25	0.04	3.05	0.12	0.10	0.02	16.94	0.34	22.41	0.51	0.32	0.03	100.29	
Van7-78-41	2R	13	53.36	0.34	0.23	0.01	2.42	0.15	0.89	0.06	2.74	0.10	0.10	0.02	17.24	0.15	23.04	0.16	0.31	0.04	100.34	
Van7-85-05	1I	8	49.79	0.26	1.32	0.07	5.02	0.20	0.79	0.07	5.52	0.13	0.13	0.03	15.46	0.25	21.38	0.18	0.34	0.06	99.75	
Van7-85-05	2I	6	49.22	0.31	1.35	0.06	4.48	0.27	0.96	0.04	5.46	0.20	0.11	0.01	15.60	0.30	21.39	0.25	0.35	0.06	98.92	
Van7-85-24	1C	12	50.52	0.52	0.22	0.04	6.59	0.51	1.14	0.07	2.97	0.50	0.08	0.03	16.47	1.89	21.21	2.43	0.37	0.08	99.56	
Van7-85-24	1R	10	51.25	0.29	0.23	0.02	5.33	0.26	0.88	0.03	2.67	0.18	0.09	0.02	16.34	0.54	22.19	0.54	0.33	0.02	99.32	
Van7-85-24	2C	11	51.15	0.60	0.20	0.03	6.19	0.16	1.09	0.06	3.26	0.74	0.09	0.02	17.51	2.76	19.87	3.79	0.36	0.10	99.72	
Van7-85-24	2R	13	51.27	0.36	0.22	0.03	5.76	0.52	0.99	0.07	3.02	0.68	0.10	0.02	16.68	2.12	21.65	2.98	0.34	0.07	100.02	
Van7-85-24	3C	14	51.22	0.97	0.20	0.05	6.34	0.31	1.09	0.10	3.69	1.16	0.10	0.03	18.64	4.81	18.27	6.49	0.30	0.13	99.83	
Van7-85-24	3R	8	50.74	0.43	0.23	0.02	6.57	0.25	1.16	0.03	3.21	0.47	0.09	0.03	16.51	1.68	21.14	2.27	0.34	0.05	99.99	
Van7-85-24	4I	11	50.66	0.45	0.25	0.02	6.36	0.17	1.18	0.03	2.93	0.29	0.08	0.02	16.00	0.83	22.05	1.15	0.39	0.04	99.90	
Van7-85-24	5C	12	50.77	0.54	0.21	0.03	6.49	0.14	1.17	0.05	3.17	0.63	0.10	0.02	16.70	2.35	20.87	3.24	0.33	0.06	99.81	
Van7-85-24	5R	14	50.69	0.26	0.23	0.02	6.71	0.18	1.18	0.04	3.06	0.28	0.09	0.02	16.00	0.86	21.51	1.18	0.37	0.06	99.83	
Van7-85-27	1C	10	51.28	0.36	0.25	0.01	5.39	0.11	1.22	0.03	2.68	0.19	0.08	0.02	16.15	0.37	21.68	0.50	0.56	0.05	99.30	
Van7-85-27	1R	6	51.70	0.20	0.22	0.02	5.17	0.14	1.17	0.02	2.57	0.24	0.10	0.01	15.81	0.43	22.10	0.46	0.57	0.03	99.42	
Van7-85-27	2C	10	51.30	0.70	0.20	0.05	5.85	0.09	1.36	0.11	3.41	1.03	0.11	0.02	18.39	4.34	18.16	5.80	0.48	0.17	99.27	
Van7-85-27	2R	7	52.28	0.66	0.24	0.01	5.11	0.26	1.14	0.09	2.65	0.14	0.08	0.01	16.16	0.41	21.40	0.55	0.56	0.05	99.61	
Van7-85-30	1C	10	51.01	0.43	0.20	0.02	6.60	0.20	1.18	0.04	3.20	0.38	0.10	0.02	16.65	1.39	20.64	1.89	0.35	0.05	99.94	

Table 5.6: Cpx major elements Cont.

Analysis	Gm ^a	Pts	SiO ₂	±	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	FeO	±	MnO	±	MgO	±	CaO	±	Na ₂ O	±	Total	Notes ^b
Van7-85-30	IR	11	50.93	0.21	0.22	0.01	6.18	0.15	1.07	0.03	2.88	0.26	0.09	0.02	16.11	0.96	21.95	1.31	0.35	0.04	99.80	
Van7-85-30	3I	14	50.86	0.49	0.23	0.02	6.18	0.23	1.11	0.03	2.88	0.42	0.08	0.02	16.13	1.26	21.96	1.86	0.42	0.06	99.85	
Van7-85-30	4C	8	51.06	0.52	0.21	0.02	6.38	0.13	1.12	0.04	3.13	0.53	0.08	0.03	16.40	1.64	21.31	2.30	0.35	0.05	100.05	
Van7-85-30	4R	11	51.21	0.48	0.23	0.01	5.77	0.18	0.98	0.04	2.94	0.23	0.08	0.02	16.22	0.54	22.02	0.79	0.37	0.15	99.84	
Van7-85-32	3C	13	51.12	0.76	0.16	0.06	6.72	0.38	1.08	0.08	3.93	1.78	0.12	0.03	18.95	5.42	17.95	8.12	0.28	0.14	100.32	
Van7-85-32	3R	8	50.97	0.45	0.22	0.02	6.34	0.29	1.04	0.02	2.81	0.18	0.11	0.03	15.72	0.40	22.82	0.66	0.33	0.03	100.35	
Van7-85-39	3C	8	50.55	0.80	0.19	0.03	6.45	0.31	1.09	0.07	3.36	0.63	0.08	0.02	17.53	2.95	19.73	3.93	0.28	0.08	99.25	
Van7-85-39	3R	7	50.89	0.37	0.23	0.02	5.46	0.29	0.89	0.07	2.75	0.12	0.06	0.02	16.06	0.21	22.44	0.40	0.34	0.07	99.11	
Van7-85-39	4I	9	50.54	0.20	0.21	0.08	5.82	0.08	0.99	0.03	2.85	0.21	0.07	0.03	16.12	0.58	22.13	0.74	0.30	0.06	99.02	
Van7-85-42	IC	3	51.53	0.67	0.21	0.01	6.40	0.11	1.17	0.03	3.39	0.51	0.10	0.03	17.04	0.84	20.96	1.92	0.36	0.11	101.15	
Van7-85-42	IR	6	51.38	0.71	0.26	0.04	6.69	0.55	1.14	0.07	3.15	0.46	0.09	0.02	16.60	1.81	21.60	2.20	0.35	0.03	101.26	
Van7-85-42	2C	10	51.94	0.44	0.19	0.02	6.12	0.21	1.05	0.04	3.42	0.55	0.11	0.01	17.54	2.46	20.19	3.08	0.34	0.07	100.89	
Van7-85-42	2R	6	52.16	0.29	0.22	0.01	5.67	0.15	0.91	0.04	3.10	0.07	0.09	0.01	17.23	0.23	21.54	0.41	0.36	0.04	101.28	
Van7-85-42	3I	6	51.27	0.57	0.23	0.02	6.52	0.21	1.14	0.03	3.31	0.56	0.09	0.01	16.90	2.01	21.16	2.84	0.36	0.05	100.97	
Van7-85-47	2C	8	51.43	0.35	0.22	0.02	6.44	0.23	1.12	0.05	2.92	0.49	0.11	0.02	16.06	1.68	22.04	2.43	0.40	0.06	100.74	
Van7-85-47	2R	13	51.73	0.43	0.24	0.02	5.99	0.23	1.00	0.03	2.95	0.29	0.09	0.02	16.45	1.17	22.05	1.45	0.39	0.07	100.88	
Van7-85-47	3C	15	51.66	1.25	0.20	0.07	6.87	0.28	1.15	0.10	3.73	1.61	0.10	0.02	18.52	6.02	18.46	8.62	0.33	0.15	101.01	
Van7-85-47	3R	10	51.34	0.54	0.23	0.03	5.94	0.32	1.02	0.05	3.24	0.84	0.09	0.03	17.31	3.24	20.97	4.12	0.36	0.07	100.50	
Van7-85-49	3I	9	51.07	0.37	0.18	0.02	6.12	0.09	1.12	0.02	3.17	0.43	0.06	0.02	17.09	1.69	20.52	2.08	0.32	0.06	99.65	
Van7-85-49	4C	10	51.42	0.78	0.16	0.04	6.40	0.15	1.16	0.09	3.61	1.18	0.08	0.02	18.54	4.30	18.68	6.00	0.25	0.10	100.30	
Van7-85-49	4R	10	51.52	0.74	0.18	0.02	5.73	0.22	0.97	0.02	2.91	0.37	0.06	0.02	16.82	1.35	21.62	1.64	0.31	0.06	100.11	
Van7-86-25	IC	7	51.81	0.72	0.14	0.02	5.86	0.72	1.24	0.13	2.60	0.23	0.07	0.02	15.72	0.78	22.26	0.53	0.33	0.04	100.03	
Van7-86-25	IR	10	51.78	0.49	0.15	0.03	5.61	0.24	1.17	0.04	2.54	0.35	0.07	0.02	16.31	1.02	21.99	1.26	0.36	0.04	99.99	
Van7-86-25	2I	9	52.56	0.87	0.15	0.02	5.06	0.14	1.00	0.04	2.74	0.18	0.07	0.02	16.61	0.62	21.70	0.68	0.33	0.03	100.20	
Van7-86-25	3I	9	51.93	0.45	0.14	0.01	5.25	0.06	1.03	0.03	2.97	0.22	0.07	0.02	16.87	0.59	21.13	0.84	0.33	0.04	99.72	
Van7-86-25	4C	10	52.61	1.32	0.10	0.05	5.78	0.40	1.14	0.20	3.87	1.61	0.08	0.02	20.55	6.56	15.81	8.91	0.24	0.15	100.17	
Van7-86-25	4R	9	52.23	0.31	0.16	0.02	5.18	0.26	1.06	0.05	2.41	0.28	0.07	0.01	16.12	0.50	22.60	0.71	0.32	0.05	100.15	
Van7-86-27	IC	12	51.65	0.82	0.21	0.05	6.33	0.21	1.15	0.10	3.55	1.19	0.11	0.02	18.29	4.25	19.04	5.95	0.27	0.10	100.60	
Van7-86-27	IR	10	51.47	0.13	0.24	0.04	5.75	0.11	0.97	0.02	2.99	0.47	0.08	0.04	16.36	0.07	22.29	0.29	0.30	0.00	100.44	
Van7-86-27	2I	17	51.89	1.92	0.22	0.03	5.60	2.12	0.97	0.55	2.99	0.48	0.09	0.02	17.27	0.91	21.21	2.08	0.34	0.24	100.57	
Van7-86-27	3C	10	51.42	0.34	0.23	0.02	5.93	0.10	1.06	0.02	2.88	0.24	0.10	0.02	16.20	0.61	22.33	0.85	0.30	0.03	100.45	
Van7-86-27	3R	13	51.77	0.55	0.29	0.08	5.40	0.34	0.98	0.06	2.51	0.28	0.08	0.02	16.10	0.78	23.10	1.10	0.25	0.04	100.49	
Van7-86-28	IC	10	51.31	0.73	0.14	0.03	6.14	0.15	1.11	0.09	4.14	1.46	0.05	0.03	19.95	4.75	16.41	6.79	0.14	0.09	99.39	
Van7-86-28	IR	9	50.93	0.46	0.19	0.02	5.72	0.17	1.04	0.04	2.93	0.38	0.05	0.02	16.50	1.41	21.63	1.97	0.22	0.05	99.21	
Van7-86-28	2I	9	51.15	0.24	0.20	0.02	5.33	0.25	0.96	0.09	2.75	0.41	0.04	0.02	16.53	0.84	21.94	1.29	0.21	0.04	99.11	
Van7-86-28	3C	10	50.74	0.26	0.18	0.01	5.99	0.16	1.17	0.02	3.01	0.28	0.06	0.02	16.37	0.72	21.31	0.96	0.19	0.04	99.03	
Van7-86-28	3R	10	51.13	0.34	0.19	0.02	5.70	0.31	1.11	0.05	2.61	0.19	0.09	0.02	15.54	0.30	22.37	0.41	0.33	0.04	99.05	
Van7-86-28	4I	9	51.25	0.62	0.19	0.03	5.48	0.54	1.00	0.06	3.17	0.85	0.04	0.02	17.02	2.18	21.00	3.54	0.22	0.07	99.37	
Van7-86-30	IC	7	50.85	0.63	0.17	0.02	5.90	0.09	1.16	0.04	3.14	0.46	0.04	0.02	17.68	2.19	19.64	3.02	0.24	0.07	98.83	
Van7-86-30	IR	8	51.37	0.68	0.18	0.02	5.39	0.21	1.08	0.04	2.79	0.21	0.05	0.01	17.02	1.15	21.00	1.81	0.28	0.02	99.15	
Van7-86-30	2I	10	51.63	0.56	0.20	0.03	5.56	0.07	1.08	0.04	2.89	0.45	0.05	0.03	16.92	1.64	21.73	2.18	0.28	0.07	100.33	

Table 5.6: Cpx major elements Cont.

Analysis	Gm ^a	Pts	SiO ₂	±	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	FeO	±	MnO	±	MgO	±	CaO	±	Na ₂ O	±	Total	Notes ^b
Van7-86-30	3I	9	52.18	0.42	0.21	0.01	5.12	0.21	0.92	0.05	2.69	0.24	0.05	0.02	16.88	0.60	22.03	0.79	0.27	0.04	100.36	
Van7-86-37	1C	10	52.44	0.47	0.15	0.02	5.07	0.19	1.47	0.08	2.97	0.47	0.09	0.03	17.44	1.92	19.90	2.35	0.57	0.08	100.10	
Van7-86-37	1R	6	51.53	0.20	0.18	0.01	4.92	0.13	1.44	0.03	2.79	0.09	0.05	0.02	16.37	0.22	21.23	0.37	0.53	0.05	99.03	
Van7-86-37	2I	9	52.02	0.27	0.17	0.02	5.01	0.13	1.37	0.05	2.70	0.23	0.07	0.02	16.27	0.62	21.60	0.70	0.55	0.03	99.77	
Van7-96-09	1C	10	50.27	0.54	0.37	0.05	7.43	0.27	0.40	0.03	3.81	0.69	0.12	0.03	17.12	3.06	19.29	4.00	0.47	0.12	99.28	PyxVn
Van7-96-09	1R	10	50.56	0.28	0.44	0.04	6.35	0.21	0.42	0.02	3.40	0.30	0.13	0.03	16.04	1.28	21.23	1.66	0.47	0.06	99.05	PyxVn
Van7-96-09	2C	11	49.71	0.29	0.37	0.03	8.30	0.15	0.38	0.02	3.53	0.26	0.14	0.01	15.65	1.30	20.69	1.73	0.58	0.06	99.36	PyxVn
Van7-96-09	2R	10	50.50	0.32	0.50	0.08	6.36	0.47	0.45	0.04	3.24	0.22	0.14	0.02	16.00	0.70	21.57	0.87	0.49	0.04	99.26	PyxVn
Van7-96-09	3C	10	50.84	0.97	0.35	0.11	7.44	0.33	0.87	0.10	3.70	1.11	0.10	0.02	18.54	5.32	17.17	6.85	0.52	0.22	99.53	PyxVn
Van7-96-09	3R	10	50.77	0.25	0.53	0.08	5.90	0.22	1.10	0.02	2.97	0.26	0.10	0.02	15.73	0.84	21.81	1.13	0.60	0.04	99.51	PyxVn
Van7-96-09	4I	11	50.93	0.40	0.56	0.04	5.48	0.28	0.58	0.03	3.23	0.11	0.12	0.02	16.11	0.44	21.84	0.47	0.46	0.03	99.30	PyxVn
Van7-96-14	1C	12	51.16	0.62	0.72	0.10	6.39	0.28	0.43	0.02	3.96	0.72	0.14	0.02	17.11	2.48	20.08	3.29	0.51	0.10	100.51	PyxVn
Van7-96-14	1R	12	52.03	0.22	0.86	0.04	4.44	0.27	0.55	0.04	3.11	0.09	0.12	0.02	16.05	0.16	22.87	0.22	0.52	0.04	100.54	PyxVn
Van7-96-14	2C	13	50.72	0.80	0.52	0.13	7.84	0.28	0.25	0.03	4.35	1.31	0.15	0.02	17.44	4.25	18.78	5.88	0.48	0.16	100.54	PyxVn
Van7-96-14	2R	19	51.38	0.32	0.74	0.05	5.86	0.38	0.50	0.03	3.69	0.41	0.14	0.02	16.32	1.38	21.24	1.87	0.55	0.07	100.42	PyxVn
Van7-96-14	5C	12	50.85	0.44	0.77	0.13	6.43	0.38	0.48	0.06	3.71	0.62	0.15	0.02	16.37	1.98	20.36	2.71	0.56	0.09	99.69	PyxVn
Van7-96-14	5R	15	50.75	1.32	1.16	0.25	6.08	0.94	0.57	0.08	3.63	1.16	0.13	0.03	16.45	4.18	20.22	5.48	0.58	0.17	99.59	PyxVn
Van7-96-14	6C	19	50.01	0.63	0.52	0.09	8.07	0.79	0.29	0.03	3.78	0.74	0.14	0.02	16.00	2.70	20.72	3.77	0.44	0.09	99.98	PyxVn
Van7-96-14	6R	16	50.45	0.21	0.70	0.06	6.78	0.22	0.36	0.04	3.54	0.19	0.13	0.02	15.60	0.50	22.00	0.73	0.42	0.03	99.98	PyxVn
Van7-96-14	7C	16	50.69	1.39	0.54	0.18	7.50	0.90	0.37	0.04	3.94	1.33	0.15	0.04	17.83	5.40	18.40	7.16	0.45	0.17	99.87	PyxVn
Van7-96-14	7R	20	50.90	1.00	0.56	0.14	6.81	0.95	0.35	0.06	3.84	1.30	0.14	0.03	17.62	4.25	19.12	6.07	0.43	0.13	99.77	PyxVn
Van7-96-15	1I	10	51.26	0.57	1.24	0.08	4.99	0.26	0.91	0.04	3.09	0.14	0.13	0.02	16.60	0.55	21.75	0.67	0.53	0.06	100.50	PyxVn
Van7-96-15	2I	12	52.24	0.43	0.89	0.08	4.43	0.50	0.65	0.05	3.06	0.15	0.13	0.02	16.58	0.28	22.05	0.33	0.49	0.03	100.53	PyxVn
Van7-96-15	3C	10	50.74	0.64	0.57	0.09	7.05	0.64	0.32	0.04	3.75	0.50	0.14	0.02	16.28	1.83	21.40	2.45	0.40	0.06	100.66	PyxVn
Van7-96-15	3R	11	50.99	0.42	0.58	0.05	6.32	0.19	0.37	0.02	3.75	0.60	0.12	0.03	16.39	1.85	21.77	2.70	0.43	0.08	100.72	PyxVn
Van7-96-15	4C	12	50.20	1.55	0.44	0.16	8.38	0.98	0.28	0.06	4.72	1.77	0.14	0.03	18.35	6.10	17.92	8.45	0.39	0.20	100.82	PyxVn
Van7-96-15	4R	12	51.14	1.22	0.68	0.16	6.22	1.06	0.35	0.05	4.06	1.30	0.12	0.03	17.19	4.30	20.75	5.73	0.39	0.13	100.89	PyxVn
Van7-96-16	1C	13	50.08	0.52	0.56	0.06	8.28	0.16	0.56	0.04	3.46	0.50	0.14	0.02	15.41	1.78	21.38	2.44	0.54	0.09	100.41	PyxVn
Van7-96-16	1R	13	51.08	0.51	1.01	0.12	5.73	0.50	0.72	0.06	2.76	0.15	0.10	0.02	15.91	0.29	22.74	0.48	0.46	0.03	100.52	PyxVn
Van7-96-16	2I	14	51.29	0.59	0.76	0.12	6.38	0.39	0.61	0.03	3.53	0.61	0.12	0.04	17.06	1.95	20.35	2.68	0.45	0.08	100.55	PyxVn
Van7-96-16	3I	10	51.36	0.63	0.65	0.08	6.16	0.84	0.40	0.04	3.30	0.28	0.13	0.03	15.85	0.53	22.18	1.02	0.49	0.10	100.51	PyxVn
Van7-96-16	4C	11	51.68	1.53	0.43	0.18	6.61	1.12	0.28	0.06	4.40	1.62	0.14	0.04	19.21	5.66	17.29	7.96	0.35	0.16	100.38	PyxVn
Van7-96-16	4R	12	51.12	0.41	0.71	0.03	5.88	0.37	0.44	0.03	3.50	0.15	0.12	0.02	16.02	0.27	22.28	0.64	0.36	0.03	100.42	PyxVn
Van7-96-16	5C	8	50.83	0.61	0.63	0.06	7.08	0.75	0.44	0.04	3.40	0.44	0.12	0.02	15.93	1.65	21.80	2.39	0.44	0.07	100.68	PyxVn
Van7-96-16	5R	9	50.81	0.65	1.18	0.11	5.88	0.25	0.64	0.03	3.08	0.17	0.13	0.03	16.13	0.66	21.76	1.06	0.57	0.06	100.17	PyxVn
Van7-96-18	1C	9	50.08	0.43	0.47	0.04	7.78	0.34			3.47	0.27	0.08	0.02	15.20	0.98	21.36	1.07	1.07	0.09	99.52	NrMat
Van7-96-18	1R	10	50.36	0.29	0.48	0.04	7.23	0.28			3.71	0.31	0.09	0.02	15.73	1.29	21.10	1.44	0.96	0.11	99.67	NrMat
Van7-96-18	2I	9	50.09	0.39	0.55	0.05	7.73	0.32			3.52	0.16	0.09	0.01	15.53	0.53	21.02	0.46	1.04	0.10	99.58	NrMat
Van7-96-18	3I	9	50.05	0.32	0.47	0.03	7.64	0.15			3.75	0.37	0.11	0.03	16.09	1.62	20.02	1.78	1.06	0.14	99.19	NrMat
Van7-96-18	4I	9	50.58	0.22	0.50	0.02	7.62	0.12			3.65	0.15	0.10	0.02	15.63	0.49	20.56	0.67	1.22	0.07	99.87	NrMat
Van7-96-18	6C	10	49.99	1.08	0.32	0.11	8.30	0.25			4.99	1.48	0.15	0.02	19.04	5.71	16.54	7.65	0.45	0.25	99.78	PyxVn

Table 5.6: Cpx major elements Cont.

Analysis	Gm ^a	Pts	SiO ₂	±	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	FeO	±	MnO	±	MgO	±	CaO	±	Na ₂ O	±	Total	Notes ^b
Van7-96-18	6R	10	50.20	0.32	0.64	0.08	5.91	0.29			3.66	0.14	0.11	0.01	15.75	0.35	22.44	0.43	0.55	0.04	99.27	PyxVn
Van7-96-18	7C	10	49.37	0.22	0.41	0.02	7.98	0.11			3.89	0.15	0.10	0.02	15.44	0.59	21.71	0.63	0.51	0.04	99.42	PyxVn
Van7-96-18	7R	10	49.93	0.25	0.63	0.04	6.83	0.24			3.81	0.14	0.10	0.03	15.82	0.28	21.82	0.44	0.54	0.04	99.48	PyxVn
Van7-96-18	8I	8	51.23	0.60	0.50	0.03	5.35	0.08			3.47	0.33	0.10	0.02	16.49	0.79	21.86	0.91	0.49	0.06	99.49	PyxVn
Van7-96-19	1C	7	50.44	1.04	0.47	0.12	7.29	0.31			3.63	1.26	0.10	0.02	18.88	4.89	17.68	6.17	0.53	0.24	99.02	NrMat
Van7-96-19	1R	5	49.94	0.18	0.78	0.02	6.70	0.09			2.79	0.18	0.07	0.03	16.16	0.28	21.77	0.39	0.66	0.08	98.88	NrMat
Van7-96-19	2C	9	50.39	0.41	0.60	0.04	6.83	0.27			2.76	0.28	0.09	0.02	16.06	1.29	21.64	1.54	0.73	0.09	99.11	NrMat
Van7-96-19	2R	8	51.18	0.73	0.76	0.06	5.37	0.85			2.77	0.20	0.09	0.02	16.41	0.48	22.65	0.88	0.55	0.11	99.79	NrMat
Van7-96-19	3I	10	51.39	0.22	0.71	0.05	5.83	0.26			2.76	0.34	0.08	0.01	16.63	1.41	22.24	1.80	0.68	0.08	100.34	NrMat
Van7-96-19	4I	10	51.44	0.68	0.65	0.06	5.50	0.30			2.95	0.56	0.07	0.01	17.44	2.25	21.32	2.84	0.59	0.11	99.99	NrMat
Van7-96-19	5C	9	51.08	0.48	0.61	0.08	6.31	0.35	1.19	0.06	3.06	0.68	0.07	0.01	16.06	1.89	21.06	2.84	0.53	0.08	99.96	PyxVn
Van7-96-19	5R	10	51.15	0.32	0.82	0.04	5.66	0.49	1.26	0.03	2.55	0.13	0.07	0.02	15.53	0.28	22.27	0.56	0.56	0.04	99.86	PyxVn
Van7-96-19	6I	10	51.83	0.23	0.59	0.03	5.00	0.24	1.30	0.05	2.96	0.20	0.08	0.01	15.94	0.59	21.66	0.73	0.55	0.04	99.90	PyxVn
Van7-96-19	7C	9	50.88	0.29	0.48	0.05	6.70	0.08			3.73	0.52	0.09	0.02	17.76	1.60	19.31	2.48	0.55	0.07	99.51	PyxVn
Van7-96-19	7R	7	51.32	0.46	0.58	0.03	5.63	0.30			2.94	0.11	0.08	0.01	16.35	0.28	22.23	0.35	0.62	0.05	99.76	PyxVn
Van7-96-19	8C	10	50.54	0.27	0.50	0.03	7.16	0.13	1.20	0.04	2.98	0.22	0.08	0.02	15.10	0.64	21.81	0.83	0.56	0.06	99.93	PyxVn
Van7-96-19	8R	8	51.33	0.50	0.51	0.05	5.68	0.23	1.35	0.05	2.93	0.37	0.09	0.02	16.04	1.20	21.07	1.47	0.57	0.04	99.59	PyxVn
Van7-96-19	9C	9	50.81	0.64	0.46	0.05	6.60	0.44	1.34	0.05	2.88	0.16	0.08	0.02	15.17	0.55	21.87	0.60	0.62	0.04	99.83	AdjVn
Van7-96-19	9R	8	51.09	0.32	0.63	0.06	5.96	0.46	1.33	0.07	2.78	0.16	0.07	0.02	15.45	0.38	22.07	0.42	0.60	0.05	99.99	AdjVn
Van7-96-21	1C	14	50.72	0.73	0.50	0.10	7.45	0.31	0.95	0.06	3.47	0.81	0.10	0.02	16.93	3.32	19.43	4.44	0.65	0.16	100.20	PyxVn
Van7-96-21	1R	13	50.68	0.78	0.48	0.09	7.95	0.22	0.98	0.08	3.67	0.92	0.10	0.02	16.62	3.39	18.60	4.27	1.02	0.27	100.12	PyxVn
Van7-96-21	4C	8	50.57	0.28	0.50	0.06	7.57	0.32	1.00	0.03	3.30	0.34	0.10	0.01	16.44	1.24	20.19	1.49	0.67	0.06	100.34	AdjVn
Van7-96-21	4R	11	50.47	0.96	0.54	0.19	7.42	0.38	0.94	0.09	3.54	1.30	0.10	0.03	16.91	4.96	19.51	6.65	0.68	0.24	100.11	AdjVn
Van7-96-21	5C	10	50.08	0.37	0.55	0.06	7.92	0.28	0.99	0.03	3.05	0.37	0.09	0.02	15.14	1.46	21.42	1.78	0.82	0.08	100.07	AdjVn
Van7-96-21	5R	10	50.34	0.46	0.60	0.07	7.43	0.25	0.96	0.03	2.87	0.17	0.09	0.02	14.47	0.24	22.73	0.24	0.77	0.06	100.25	AdjVn
Van7-96-25	1C	6	51.39	0.34	0.36	0.02	6.96	0.26			3.32	0.23	0.08	0.01	15.11	0.81	20.51	0.77	1.44	0.10	99.18	
Van7-96-25	2I	8	51.27	0.44	0.35	0.02	6.72	0.31	0.98	0.03	3.16	0.18	0.06	0.02	14.83	0.69	20.21	0.80	1.34	0.05	98.92	
Van7-96-25	3I	10	51.15	0.26	0.37	0.02	7.08	0.14	1.01	0.03	3.39	0.11	0.07	0.02	14.90	0.44	19.89	0.54	1.31	0.09	99.17	
Van7-96-25	4I	9	50.89	0.26	0.36	0.02	7.26	0.16	1.12	0.04	3.35	0.28	0.06	0.02	14.55	0.84	20.27	1.05	1.29	0.11	99.14	
Van7-96-25	5C	10	51.05	0.36	0.33	0.03	7.32	0.20	1.05	0.05	3.51	0.40	0.08	0.01	14.89	1.37	19.42	1.69	1.29	0.15	98.93	
Van7-96-25	5R	10	50.72	0.19	0.40	0.04	7.05	0.21	1.08	0.05	3.39	0.22	0.06	0.02	14.43	0.88	20.77	1.05	1.16	0.10	99.05	
Van7-96-28	3C	10	51.23	0.54	0.30	0.06	6.67	0.21	1.36	0.10	3.26	0.85	0.08	0.02	16.21	3.48	18.99	4.32	1.03	0.26	99.13	
Van7-96-28	3R	8	51.18	0.94	0.37	0.09	6.79	0.52			3.50	1.26	0.08	0.02	16.88	4.45	19.17	5.64	0.96	0.37	98.93	
Van7-96-28	4I	10	51.18	0.37	0.36	0.04	6.65	0.13			3.24	0.37	0.07	0.02	16.13	1.39	20.68	1.67	0.97	0.09	99.30	
Van7-96-28	5C	8	51.19	0.46	0.37	0.05	7.22	0.39			3.23	0.45	0.06	0.02	16.06	1.47	20.24	1.87	1.11	0.10	99.51	
Van7-96-28	5R	9	50.97	0.18	0.36	0.04	6.75	0.14	1.34	0.02	2.89	0.21	0.06	0.03	14.55	0.68	21.22	0.76	1.08	0.09	99.23	
Van7-96-28	6I	10	51.59	0.52	0.32	0.03	6.42	0.31	1.27	0.06	3.13	0.56	0.06	0.01	15.96	2.35	19.43	2.69	1.13	0.17	99.32	
Van7-96-35	3C	8	52.10	0.44	0.31	0.02	5.37	0.12	1.73		2.63	0.20	0.09	0.02	16.03	1.04	20.33	1.17	1.38	0.07	99.98	
Van7-96-35	3R	9	51.90	0.33	0.33	0.02	5.46	0.09	1.73	0.04	2.61	0.21	0.09	0.02	15.91	1.01	20.28	0.99	1.39	0.08	99.72	
Van7-96-35	4I	10	52.17	0.26	0.34	0.01	5.29	0.07	1.68	0.03	2.46	0.09	0.08	0.01	15.72	0.21	20.94	0.27	1.39	0.04	100.06	
Van7-96-35	5I	10	52.33	0.18	0.33	0.02	5.13	0.10	1.60	0.03	2.56	0.10	0.08	0.01	15.97	0.31	20.48	0.42	1.35	0.06	99.82	

Table 5.6: Cpx major elements Cont.

Analysis	Gm ^a	Pts	SiO ₂	±	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	FeO	±	MnO	±	MgO	±	CaO	±	Na ₂ O	±	Total	Notes ^b
Van7-96-35	6C	8	52.24	0.45	0.27	0.03	5.48	0.18	1.75	0.11	2.83	0.45	0.10	0.02	17.18	2.32	18.68	2.79	1.33	0.20	99.85	
Van7-96-35	6R	10	52.29	0.12	0.36	0.03	5.23	0.06	1.67	0.02	2.38	0.05	0.07	0.02	15.30	0.50	20.93	0.39	1.45	0.10	99.68	
Van7-96-37	1C	10	51.37	0.52	0.30	0.03	6.54	0.26	1.56	0.08	3.25	0.72	0.10	0.03	15.73	2.58	19.43	3.05	1.29	0.25	99.57	
Van7-96-37	1R	9	51.28	0.52	0.33	0.06	6.29	0.16	1.53	0.10	2.96	0.60	0.09	0.02	15.96	2.42	20.51	3.08	1.02	0.18	99.99	
Van7-96-37	2I	6	51.38	0.45	0.32	0.02	6.51	0.14	1.48	0.06	3.34	0.47	0.11	0.04	16.01	1.80	19.93	2.13	0.94	0.13	100.01	
Van7-96-37	4I	6	51.53	0.80	0.37	0.04	6.50	0.24	1.44	0.06	2.85	0.41	0.10	0.03	15.00	0.91	20.90	1.30	1.35	0.07	100.05	
Van7-96-37	5C	9	51.93	0.77	0.32	0.06	6.70	0.22	1.56	0.13	3.56	0.89	0.14	0.04	17.53	3.47	18.00	4.19	1.25	0.29	101.00	
Van7-96-37	5R	10	51.56	0.56	0.33	0.06	6.30	0.20	1.45	0.10	3.24	0.62	0.13	0.04	16.47	2.45	19.29	3.07	1.17	0.19	99.94	
Van7-96-38	1C	11	50.81	0.42	0.44	0.05	8.16	0.26	0.92	0.06	3.44	0.81	0.08	0.04	15.64	2.94	19.01	3.56	1.32	0.26	99.82	
Van7-96-38	1R	13	51.00	0.43	0.48	0.07	7.88	0.44	0.89	0.07	3.40	0.88	0.09	0.02	15.62	2.54	19.09	3.33	1.22	0.24	99.67	
Van7-96-38	2C	10	51.25	0.89	0.39	0.14	7.62	0.53	0.85	0.11	4.04	1.70	0.11	0.03	17.68	5.52	16.49	7.18	1.12	0.44	99.55	
Van7-96-38	2R	10	51.13	0.17	0.49	0.02	7.66	0.10	0.84	0.02	3.35	0.16	0.10	0.02	14.98	0.54	20.14	0.58	1.15	0.09	99.84	
Van7-96-66B	1I	10	48.24	1.32	0.87	0.10	9.15	1.64	0.67	0.14	4.01	0.74	0.15	0.06	15.24	1.97	21.00	2.89	0.56	0.11	99.88	Opx Vn
Van7-96-66B	2I	10	49.53	0.50	0.74	0.04	7.82	0.26	0.62	0.04	3.55	0.34	0.12	0.03	14.95	0.81	22.12	1.10	0.58	0.08	100.03	Opx Vn
Van7-96-66B	3I	10	50.10	0.47	0.86	0.08	6.84	0.28	0.74	0.06	3.72	0.67	0.13	0.04	16.26	1.88	20.99	2.56	0.55	0.08	100.19	Opx Vn
Van7-96-66B	5C	10	50.92	0.99	0.17	0.05	7.92	0.62	0.13	0.03	4.52	1.39	0.15	0.03	15.81	4.27	20.28	5.70	0.65	0.19	100.55	Breccia
Van7-96-66B	5R	8	49.94	0.58	0.18	0.01	9.13	0.56	0.14	0.02	4.47	0.33	0.12	0.03	14.89	1.39	20.84	1.96	0.70	0.06	100.39	Breccia

^a Grain number; core (C), rim (R), interstitial (I) or symplectite (S). Interstitial grains are < 1 mm diameter. L preceding a grain number indicates Lee (1997) data for dredge RC27-9-6 samples.

^b Abbreviations for lithologies (blank indicates residual peridotite): Dep. Trans and Enr are depleted, transitional and enriched zones in cryptically metasomatized samples; GbVn refers to a gabbroic vein, PyxVn to a pyroxenite vein, CpxVn to a clinopyroxenite vein, and OpxVn to an orthopyroxenite vein; AdjVn is adjacent to a vein, NrMat is < 5 cm from a vein and FarMat is > 5 cm from a vein.

Table 5.7: Major element composition of spinels, in wt%.

Sample	Gm ^a	Pts	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	Fe ₂ O ₃	±	FeO	±	MnO	±	MgO	±	NiO	±	ZnO	±	Total	Notes ^b
6K-458-1	1	5	0.07	0.06	48.28	1.62	21.93	1.91	0.30	0.27	11.30	0.31	0.12	0.04	18.49	0.33	0.27	0.04	0.00	0.00	100.80	
6K-458-3	1	6	0.05	0.01	55.41	0.41	14.72	0.52	0.00	0.00	11.33	0.15	0.15	0.03	18.90	0.19	0.32	0.02	0.12	0.04	101.06	
6K-465-2	1	6	0.17	0.01	46.54	0.27	20.02	0.22	4.06	0.58	9.34	0.56	0.08	0.02	19.50	0.32	0.27	0.02			99.99	Dep
6K-465-2	2	6	0.29	0.02	28.89	1.05	37.78	0.59	2.55	0.56	16.49	0.49	0.25	0.04	12.89	0.17	0.11	0.03			99.34	Dep
6K-465-2	3	6	0.26	0.03	39.47	1.12	28.23	1.18	1.58	0.33	13.18	0.26	0.17	0.02	16.16	0.26	0.18	0.02			99.25	Dep
6K-465-3	1	4	0.59	0.03	38.38	0.31	21.70	0.46	5.79	0.37	20.50	0.18	0.28	0.04	11.24	0.07	0.20	0.03			98.72	NrMat
6K-465-3	4	7	0.50	0.13	43.85	4.56	23.59	4.27	1.37	0.42	13.09	1.32	0.14	0.04	16.92	1.22	0.27	0.04			99.82	NrMat
6K-466-1	1	6	0.10	0.02	48.21	0.16	19.61	0.16	2.97	0.31	8.47	0.15	0.09	0.02	20.17	0.14	0.29	0.02			99.92	
6K-643-13	1	4	0.02	0.03	47.53	0.31	20.85	0.14	1.18	0.19	10.78	0.14	0.11	0.03	18.36	0.03	0.33	0.01	0.00	0.00	99.17	
6K-651-7	1	6	0.03	0.01	55.60	0.39	13.95	0.29	0.00	0.00	11.49	0.43	0.14	0.01	18.50	0.29	0.30	0.04	0.23	0.03	100.27	PyxVn
JR31-52-3	1	6	0.02	0.02	53.01	0.44	16.48	0.27	0.36	0.31	11.03	0.29	0.15	0.03	19.01	0.13	0.31	0.03			100.37	
JR31-52-4	2	6	0.09	0.11	52.57	0.18	16.07	0.31	0.00	0.00	12.13	0.26	0.12	0.03	18.08	0.14	0.28	0.03			99.37	
JR31-52-5	1	6	0.04	0.01	51.41	1.23	17.53	1.09	0.86	0.50	10.90	0.39	0.13	0.01	18.90	0.25	0.30	0.03			100.09	
JR31-52-8	1	6	0.07	0.02	52.46	0.17	16.73	0.18	0.19	0.20	11.07	0.36	0.11	0.03	18.86	0.15	0.31	0.03			99.81	
Kn162-19-41	1	2	0.04	0.00	56.08	0.73	12.38	0.13	0.04	0.06	11.74	0.70	0.16	0.00	18.48	0.59	0.21	0.05	0.00	0.00	99.17	PyxVn
Kn162-47-02	2	6	0.00	0.00	28.67	0.95	42.95	0.97	0.00	0.00	13.95	0.68	0.15	0.04	14.05	0.25	0.03	0.01	0.17	0.04	100.00	
Kn162-47-08	1	6	0.00	0.00	55.59	0.33	12.90	0.23	0.80	0.24	11.07	0.21	0.09	0.02	19.20	0.18	0.26	0.02	0.10	0.05	100.01	
Kn162-47-09	1	6	0.05	0.01	39.67	0.63	29.82	0.72	0.33	0.15	12.63	0.15	0.17	0.02	16.47	0.10	0.17	0.02	0.16	0.04	99.48	
Kn162-47-10	1	6	0.07	0.02	51.56	0.42	15.78	0.39	0.69	0.36	13.23	0.43	0.15	0.02	17.24	0.32	0.25	0.04	0.01	0.01	99.04	PyxVn
Kn162-47-10	2	6	0.08	0.01	52.58	0.26	15.55	0.12	0.80	0.42	11.82	0.31	0.19	0.03	18.35	0.30	0.24	0.03	0.37	0.11	100.00	FarMat
Kn162-47-12	1	6	0.07	0.02	54.36	0.34	12.70	0.46	1.31	0.13	12.00	0.15	0.13	0.01	18.36	0.12	0.26	0.03	0.01	0.01	99.21	PyxVn
Kn162-47-22	1	5	0.00	0.00	55.82	0.30	13.30	0.09	0.25	0.32	11.36	0.31	0.10	0.03	19.03	0.28	0.31	0.02	0.02	0.02	100.19	PyxVn
Kn162-47-24	1	4	0.09	0.01	51.34	0.41	15.40	0.43	1.50	0.26	12.53	0.36	0.15	0.02	17.69	0.26	0.22	0.03	0.04	0.03	99.02	
Kn162-47-24	3	5	0.07	0.01	50.43	0.45	16.31	0.31	1.27	0.40	13.34	0.52	0.15	0.02	17.04	0.39	0.21	0.03	0.09	0.08	99.07	
Kn162-47-25	1	3	0.13	0.02	51.42	0.28	17.08	0.20	0.44	0.38	10.79	0.31	0.13	0.02	18.90	0.22	0.21	0.01	0.00	0.00	99.13	NrMat
Kn162-47-26	1	6	0.04	0.01	57.12	0.27	11.85	0.18	0.23	0.23	10.84	0.40	0.12	0.02	19.59	0.25	0.20	0.02	0.25	0.11	100.27	
Kn162-47-29	1	6	0.08	0.01	52.66	0.21	16.61	0.26	0.38	0.24	11.34	0.26	0.11	0.02	18.87	0.20	0.18	0.04	0.19	0.11	100.45	
Kn162-47-36	1	6	0.09	0.01	55.68	0.89	13.13	0.67	0.01	0.01	14.03	0.47	0.12	0.02	17.00	0.45	0.37	0.03	0.24	0.03	100.73	
Kn162-47-49	1	6	0.03	0.01	52.38	0.97	14.32	0.62	1.17	0.40	13.20	0.48	0.13	0.02	17.33	0.35	0.19	0.02	0.51	0.06	99.34	
PS86-6-26	1	6	0.07	0.02	57.73	0.39	10.01	0.19	0.95	0.68	10.51	0.17	0.10	0.02	19.71	0.21	0.19	0.03	0.26	0.09	99.55	
PS86-6-38	1	3	0.05	0.02	57.31	0.10	10.84	0.12	0.58	0.51	11.15	0.15	0.10	0.01	19.23	0.06	0.33	0.01	0.00	0.00	99.58	
PS86-6-39	1	6	0.06	0.03	51.89	0.33	16.58	0.32	0.85	0.21	11.43	0.19	0.13	0.03	18.55	0.12	0.28	0.02	0.00	0.00	99.77	
PS86-6-40	1	6	0.02	0.02	56.85	0.32	10.90	0.19	0.72	0.30	11.36	0.16	0.11	0.01	18.97	0.09	0.31	0.03	0.00	0.00	99.24	
RC27-9-6-2	L1	5	0.18	0.01	25.83	0.75	41.01	1.35	3.14		19.06	0.41	0.28	0.03	11.71	0.20	0.09	0.02			101.30	FarMat
RC27-9-6-2	L3	5	0.33	0.01	28.13	0.39	38.67	0.46	4.33		15.63	0.23	0.34	0.01	13.64	0.17	0.27	0.02			101.41	FarMat
RC27-9-6-2	L4	5	0.17	0.02	38.57	1.29	28.31	1.40	4.26		13.15	0.64	0.28	0.05	16.22	0.42	0.33	0.02			101.38	FarMat
RC27-9-6-2	L6	5	0.18	0.01	40.18	0.68	27.11	0.72	3.55		13.44	0.12	0.27	0.03	16.24	0.28	0.32	0.04			101.36	NrMat
RC27-9-6-2	L9	5	0.22	0.00	38.28	1.33	29.37	1.37	3.23		13.67	0.65	0.29	0.03	15.89	0.65	0.33	0.03			101.35	NrMat
RC27-9-6-2	L10	6	0.03	0.02	40.64	3.78	27.45	3.56	3.02		12.57	1.34	0.13	0.03	16.91	1.04	0.15	0.04			100.93	CpxVn
RC27-9-6-2	L11	5	0.19	0.01	43.67	0.89	24.08	0.88	2.56		12.87	0.33	0.26	0.02	16.88	0.22	0.36	0.02			100.93	CpxVn

Table 5.7: Spinel major elements Cont.

Sample	Gm ^a	Pts	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	Fe ₂ O ₃	±	FeO	±	MnO	±	MgO	±	NiO	±	ZnO	±	Total	Notes ^b
RC27-9-6-2	L12	5	0.28	0.02	34.66	0.58	32.68	0.83	3.76		14.49	0.37	0.18	0.01	15.29	0.18	0.14	0.03			101.49	CpxVn
RC27-9-6-2	L14	5	0.15	0.01	42.11	1.33	25.29	1.05	3.36		12.49	0.41	0.28	0.02	16.95	0.36	0.35	0.03			101.05	NrMat
RC27-9-6-2	L15	5	0.18	0.01	41.19	0.61	26.31	0.42	2.96		13.73	0.48	0.27	0.02	16.10	0.12	0.33	0.03			101.15	NrMat
RC27-9-6-2	L17	5	0.18	0.01	40.02	0.71	26.89	1.03	4.04		12.99	0.16	0.28	0.02	16.48	0.15	0.33	0.03			101.27	NrMat
RC27-9-6-2	L18	5	0.14	0.01	41.13	0.82	26.43	0.80	3.55		12.54	0.18	0.26	0.02	16.86	0.11	0.38	0.03			101.38	NrMat
RC27-9-6-2	L20	5	0.19	0.02	40.31	1.10	27.13	1.09	3.27		13.41	0.20	0.28	0.03	16.21	0.24	0.34	0.03			101.24	NrMat
RC27-9-6-5	1	6	0.21	0.03	42.99	3.91	24.44	3.71	2.07	0.66	13.16	0.96	0.17	0.05	16.57	0.96	0.30	0.03	0.18	0.07	100.12	Dep
RC27-9-6-5	2	6	0.17	0.03	27.59	1.14	39.19	0.61	2.71	0.39	17.01	0.30	0.23	0.03	12.43	0.28	0.19	0.03	0.05	0.10	99.66	Dep
RC27-9-6-5	L1	6	0.32	0.09	46.56	2.30	20.61	2.26	2.43		11.63	0.60	0.25	0.02	17.87	0.44	0.41	0.03			100.14	
RC27-9-6-7	1	3	0.00	0.01	49.49	2.90	15.73	0.46	3.74	2.34	11.29	1.03	0.24	0.26	18.12	1.03	0.33	0.03	0.00	0.00	99.03	
RC27-9-6-7	2	6	0.00	0.00	51.79	0.76	15.58	0.57	1.67	0.44	11.95	0.31	0.07	0.03	18.15	0.38	0.31	0.02	0.00	0.00	99.57	
Van7-78-25	1	6	0.50	0.02	24.16	0.40	44.38	0.63	0.58	0.20	19.69	0.20	0.35	0.03	10.69	0.11	0.07	0.02	0.13	0.03	100.56	GbVn
Van7-78-25	2	6	0.48	0.04	25.01	0.45	43.23	0.48	0.92	0.41	19.11	0.21	0.34	0.02	11.05	0.08	0.08	0.02	0.19	0.05	100.50	GbVn
Van7-78-31	2	7	0.52	0.02	22.02	0.68	42.35	0.76	4.63	0.58	19.67	0.43	0.29	0.03	10.36	0.34	0.09	0.01	0.19	0.04	100.19	AdjVn
Van7-78-36	11	6	0.61	0.01	23.46	1.07	41.17	1.03	3.65	0.15	18.82	0.23	0.18	0.02	11.07	0.27	0.00	0.00	0.00	0.00	98.95	FarMat
Van7-78-40	1	4	0.79	0.02	21.51	0.62	41.94	0.46	5.30	0.16	20.79	0.28	0.29	0.01	9.87	0.28	0.11	0.01	0.29	0.07	100.89	
Van7-78-41	1	6	0.46	0.02	21.75	0.78	41.35	0.96	5.95	0.33	20.31	0.13	0.28	0.03	9.95	0.17	0.05	0.03	0.04	0.03	100.15	
Van7-85-24	1	6	0.00	0.00	53.87	0.67	14.90	0.52	0.00	0.00	12.18	0.43	0.06	0.01	18.09	0.23	0.20	0.03	0.04	0.05	99.34	
Van7-85-27	2	6	0.01	0.01	47.18	0.95	21.50	1.14	0.51	0.18	12.28	0.32	0.10	0.02	17.44	0.25	0.18	0.01	0.04	0.04	99.23	
Van7-85-30	1	6	0.06	0.01	53.52	0.37	14.56	0.34	0.43	0.14	12.18	0.17	0.08	0.02	18.16	0.13	0.27	0.01	0.17	0.04	99.44	
Van7-85-32	2	6	0.00	0.00	52.56	0.46	15.89	0.34	0.85	0.41	11.46	0.24	0.01	0.01	18.66	0.09	0.21	0.02	0.00	0.00	99.64	
Van7-85-39	1	6	0.05	0.01	52.19	0.19	15.99	0.18	1.49	0.28	11.72	0.20	0.11	0.03	18.53	0.17	0.32	0.03	0.13	0.05	100.57	
Van7-85-42	1	6	0.06	0.01	53.70	0.77	15.15	0.49	0.72	0.19	11.69	0.26	0.11	0.02	18.70	0.28	0.30	0.03	0.21	0.03	100.65	
Van7-85-47	1	6	0.07	0.02	52.47	0.17	16.48	0.20	0.94	0.05	11.69	0.25	0.09	0.02	18.65	0.11	0.29	0.01	0.18	0.02	100.86	
Van7-85-49	1	6	0.03	0.01	54.14	0.47	14.19	0.12	0.63	0.48	11.49	0.43	0.12	0.01	18.61	0.37	0.37	0.05	0.16	0.03	99.76	
Van7-86-25	1	5	0.06	0.01	50.17	0.26	18.97	0.21	0.31	0.35	11.76	0.29	0.15	0.04	18.30	0.25	0.19	0.02	0.27	0.03	100.23	
Van7-86-27	1	6	0.00	0.00	51.80	0.90	16.87	0.94	0.00	0.00	12.69	0.23	0.08	0.03	17.75	0.30	0.12	0.04	0.00	0.00	99.30	
Van7-86-28	1	6	0.03	0.01	54.00	0.39	15.17	0.38	0.41	0.37	11.67	0.13	0.13	0.03	18.71	0.09	0.33	0.03	0.18	0.05	100.65	
Van7-86-30	1	6	0.05	0.01	50.81	0.44	17.83	0.47	0.89	0.34	11.64	0.24	0.14	0.02	18.35	0.13	0.26	0.04	0.18	0.04	100.18	
Van7-86-37	1	6	0.11	0.01	42.46	0.76	26.40	0.44	0.88	0.52	13.47	1.03	0.20	0.03	16.33	0.64	0.23	0.02	0.22	0.05	100.37	
Van7-96-09	1	6	0.37	0.06	36.73	2.21	28.76	2.19	4.02	0.33	13.62	0.71	0.14	0.02	15.76	0.62	0.21	0.02	0.21	0.04	99.87	PyxVn
Van7-96-14	3	6	0.20	0.08	55.79	5.29	10.51	4.77	2.04	0.40	12.18	1.07	0.14	0.04	18.46	1.26	0.30	0.04	0.20	0.06	99.85	PyxVn
Van7-96-14	4	5	0.37	0.06	43.21	1.55	21.99	1.36	2.94	0.46	14.65	0.56	0.18	0.04	15.53	0.49	0.24	0.02	0.19	0.04	99.39	PyxVn
Van7-96-14	5	6	0.27	0.09	44.32	0.96	21.02	0.87	2.41	0.19	17.47	2.12	0.37	0.11	13.86	1.54	0.24	0.03	0.36	0.08	100.37	PyxVn
Van7-96-15	1	6	0.26	0.03	56.15	0.42	10.99	0.30	1.53	0.16	11.35	0.13	0.17	0.02	19.11	0.14	0.40	0.03	0.18	0.03	100.17	PyxVn
Van7-96-16	1	6	0.21	0.03	58.19	0.22	9.73	0.20	0.78	0.24	11.04	0.19	0.16	0.03	19.50	0.11	0.43	0.02	0.12	0.03	100.17	PyxVn
Van7-96-18	1	5	0.07	0.04	56.84	0.37	9.62	0.50	3.27	0.42	9.84	0.17	0.12	0.02	20.14	0.29	0.41	0.03	0.00	0.00	100.33	NrMat
Van7-96-18	2	6	0.20	0.08	51.39	1.42	14.62	1.12	3.47	0.48	11.52	0.37	0.14	0.02	18.54	0.41	0.43	0.03	0.00	0.00	100.34	PyxVn
Van7-96-19	1	5	0.31	0.21	39.70	0.84	27.66	0.98	2.01	0.55	13.95	0.60	0.20	0.03	15.84	0.29	0.24	0.03	0.01	0.01	99.99	NrMat
Van7-96-19	3	6	0.53	0.19	37.78	1.67	29.47	1.27	1.84	0.42	15.43	0.97	0.22	0.04	14.90	0.70	0.22	0.02	0.02	0.02	100.46	PyxVn
Van7-96-21	1	6	0.26	0.26	54.38	1.01	12.21	1.10	2.39	0.25	10.08	0.43	0.11	0.02	19.70	0.24	0.37	0.02	0.17	0.03	99.69	PyxVn

Table 5.7: Spinel major elements Cont.

Sample	Grm ^a	Pts	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	Fe ₂ O ₃	±	FeO	±	MnO	±	MgO	±	NiO	±	ZnO	±	Total	Notes ^b
Ván7-96-21	2	6	0.32	0.08	48.43	0.85	18.31	0.66	2.51	0.40	11.49	0.15	0.15	0.03	18.31	0.21	0.26	0.03	0.25	0.05	100.07	PyxVn
Ván7-96-21	4	5	0.35	0.05	43.08	1.61	23.20	1.58	2.40	0.23	13.08	0.45	0.16	0.03	16.62	0.41	0.25	0.05	0.33	0.04	99.52	AdjVn
Ván7-96-25	1	5	0.07	0.03	53.35	0.21	12.68	0.08	2.81	0.27	12.28	0.20	0.14	0.02	18.13	0.17	0.38	0.04	0.00	0.00	99.89	
Ván7-96-28	1	5	0.25	0.04	45.92	0.99	21.56	0.61	1.37	0.25	13.15	0.30	0.16	0.02	16.97	0.38	0.25	0.04	0.02	0.03	99.76	
Ván7-96-35	1	5	0.18	0.02	40.15	1.62	28.60	1.49	1.56	0.19	12.51	0.26	0.18	0.02	16.82	0.29	0.16	0.03	0.00	0.00	100.16	
Ván7-96-37	1	4	0.16	0.02	47.74	0.87	19.88	0.61	1.81	0.48	12.45	0.28	0.14	0.03	17.58	0.24	0.22	0.02	0.00	0.00	99.99	
Ván7-96-38	1	6	0.28	0.50	56.04	0.87	10.85	0.85	1.58	0.81	11.20	0.46	0.08	0.02	19.20	0.25	0.33	0.03	0.18	0.04	99.74	
Ván7-96-66B	1	5	0.06	0.02	61.60	0.38	5.26	0.37	1.66	0.19	10.67	0.13	0.10	0.01	19.87	0.08	0.58	0.05	0.00	0.00	99.80	OpxVn

^a Grain number; L preceding a grain number indicates Lee (1997) data for dredge RC27-9-6 samples.^b Abbreviations for lithologies (blank indicates residual peridotite): Dep, Trans and Enr are depleted, transitional and enriched zones in cryptically metasomatized samples; GbVn refers to a gabbroic vein, PyxVn to a pyroxenite vein, CpxVn to a clinopyroxenite vein, and OpxVn to an orthopyroxenite vein; AdjVn is adjacent to a vein, NrMat is <5 cm from a vein and FarMat is >5 cm from a vein.

Table 5.8: Major element composition of plagioclase, in wt%.

Analysis	Grn ^a	Pts	SiO ₂	±	Al ₂ O ₃	±	FeO	±	CaO	±	Na ₂ O	±	Total	An	Notes ^b
6K-465-2	1I	6	45.34	0.32	35.20	0.38	0.11	0.02	18.00	0.35	1.02	0.19	99.72	90.7	Dep
6K-465-2	2I	6	44.75	0.14	35.35	0.15	0.13	0.04	18.42	0.11	0.74	0.04	99.44	93.2	Dep
6K-465-2	3I	7	45.29	0.36	35.02	0.35	0.15	0.04	18.04	0.21	1.06	0.07	99.60	90.4	Dep
6K-465-3	1aI	3	55.36	0.29	29.34	0.45	0.16	0.07	11.17	0.32	5.09	0.31	101.21	54.5	NrMat
6K-465-3	3I	3	57.69	0.69	27.80	0.27	0.26	0.04	8.91	0.68	6.31	0.15	101.05	43.7	GbVn
6K-465-3	4aI	26	55.11	0.72	28.38	0.51	0.05	0.03	10.38	0.51	5.43	0.26	99.44	51.2	NrMat
6K-465-3	4bI	8	55.25	0.32	28.20	0.23	0.06	0.02	10.22	0.23	5.55	0.23	99.34	50.3	NrMat
RC27-9-6-2	L1I	1	54.7	0.0	29.33	0.01	0.09	0.26	11.30	0.01	5.23	0.02	101.34	54.1	FarMat
RC27-9-6-2	L2I	1	53.6	0.0	30.25	0.01	0.19	0.20	12.43	0.01	4.83	0.02	101.62	58.5	NrMat
RC27-9-6-2	L3I	6	49.8	0.8	32.60	0.33	0.12	0.03	15.24	0.63	3.00	0.38	103.05	73.7	CpxVn
RC27-9-6-5	L1I	6	47.2	0.4	33.62	0.28	0.10	0.03	17.02	0.18	1.84	0.19	101.03	83.6	
RC27-9-6-5	1I	6	47.19	0.32	34.64	0.40	0.11	0.04	16.75	0.30	1.74	0.21	100.44	84.1	
Van7-78-25	1I	6	47.88	0.25	34.05	0.17	0.13	0.03	16.25	0.14	1.98	0.14	100.32	81.9	
Van7-78-31	1I	6	48.93	0.14	32.94	0.10	0.17	0.01	15.41	0.11	2.55	0.11	100.06	76.8	NrMat
Van7-78-31	2I	6	50.38	0.36	31.96	0.31	0.13	0.01	14.19	0.29	3.34	0.29	100.05	69.9	GbVn
Van7-78-36	11I	6	47.29	0.27	34.33	0.25	0.13	0.04	17.09	0.31	1.67	0.21	100.53	84.8	FarMat
Van7-78-36	1C	4	52.61	0.41	30.74	0.29	0.09	0.01	12.80	0.27	4.38	0.16	100.72	61.5	GbVn
Van7-78-36	2I	6	53.10	0.28	30.53	0.22	0.11	0.02	12.49	0.24	4.56	0.35	100.89	59.9	GbVn
Van7-78-40	1I	6	47.94	0.43	33.26	0.75	0.19	0.04	16.17	0.51	2.03	0.14	100.04	81.3	
Van7-78-41	1I	5	46.03	0.18	35.16	0.10	0.18	0.02	18.16	0.13	1.09	0.13	100.63	90.2	
Van7-96-09	1I	5	49.75	0.25	31.91	0.22	0.11	0.02	14.29	0.19	3.11	0.12	99.18	71.7	PyxVn
Van7-96-09	2I	6	46.15	0.53	34.77	0.48	0.13	0.01	17.27	0.20	1.46	0.15	99.79	86.7	PyxVn
Van7-96-14	1I	6	47.79	0.46	33.96	0.39	0.04	0.02	16.14	0.14	2.17	0.09	100.10	80.5	PyxVn
Van7-96-14	3C	3	46.92	0.13	34.09	0.07	0.05	0.01	16.77	0.02	1.87	0.07	99.70	83.2	PyxVn
Van7-96-14	3R	3	46.10	0.61	34.82	0.42	0.06	0.03	17.31	0.47	1.57	0.30	99.87	85.9	PyxVn
Van7-96-14	4C	4	47.62	0.38	33.64	0.17	0.08	0.01	16.01	0.12	2.25	0.10	99.62	79.7	PyxVn
Van7-96-14	4R	2	47.52	0.08	33.60	0.02	0.08	0.02	16.04	0.13	2.13	0.03	99.40	80.5	PyxVn
Van7-96-14	7	4	45.85	0.25	34.82	0.33	0.12	0.01	17.49	0.28	1.40	0.14	99.72	87.3	PyxVn
Van7-96-15	2I	6	47.94	0.36	33.66	0.19	0.11	0.04	16.46	0.29	2.13	0.12	100.32	81.0	PyxVn
Van7-96-15	3I	6	46.74	0.31	34.62	0.31	0.13	0.03	17.52	0.19	1.50	0.12	100.51	86.5	PyxVn
Van7-96-15	4I	4	46.95	0.36	35.47	0.08	0.13	0.04	17.44	0.20	1.54	0.15	101.53	86.2	PyxVn
Van7-96-16	1I	6	47.07	0.17	34.41	0.19	0.12	0.03	17.14	0.03	1.66	0.12	100.41	85.1	PyxVn
Van7-96-16	3I	6	45.72	0.28	35.35	0.13	0.16	0.04	18.28	0.17	1.00	0.09	100.53	91.0	PyxVn
Van7-96-18	1I	6	49.54	0.47	32.95	0.24	0.16	0.03	14.88	0.26	2.94	0.16	100.47	73.7	PyxVn
Van7-96-19	1I	5	51.23	0.29	30.37	0.35	0.12	0.05	12.87	0.36	4.24	0.15	99.20	62.6	NrMat
Van7-96-19	3I	5	51.86	0.37	30.27	0.26	0.11	0.02	12.53	0.24	4.57	0.14	99.34	60.2	PyxVn
Van7-96-21	1I	5	50.16	2.06	32.21	1.19	0.09	0.03	14.17	1.41	3.23	0.86	99.86	70.8	PyxVn
Van7-96-21	2I	5	51.94	1.72	31.00	0.70	0.09	0.02	12.90	1.07	4.00	0.70	99.94	64.1	PyxVn
Van7-96-21	3I	5	52.50	2.15	31.00	1.09	0.13	0.05	12.64	1.47	4.08	0.71	100.37	63.1	PyxVn

^a Grain number and core (C), rim (R) or interstitial (I). Interstitial grains are < 1 mm diameter. L preceding a grain number indicates Lee (1997) data for dredge RC27-9-6 samples.

^b Abbreviations for lithologies (blank indicates residual peridotite): Dep, Trans and Enr are depleted, transitional and enriched zones in cryptically metasomatized samples; GbVn refers to a gabbroic vein, PyxVn to a pyroxenite vein and CpxVn to a clinopyroxenite vein; NrMat is <5 cm from a vein and FarMat is >5 cm from a vein.

Table 5.9: Major element compositions of amphiboles, in wt%.

Analysis	Grn ^a	Pts	SiO ₂	±	TiO ₂	±	Al ₂ O ₃	±	Cr ₂ O ₃	±	FeO	±	MgO	±	CaO	±	Na ₂ O	±	K ₂ O	±	Total	Notes ^b
6K-465-3	A2C	4	54.84	1.20	0.35	0.15	2.27	0.83	0.02	0.02	5.93	0.49	20.47	0.51	12.07	0.12	0.62	0.17	0.05		96.73	GbVn
6K-465-3	A3C	2	56.67	0.09	0.17	0.02	1.02	0.02	0.02	0.02	5.17	0.04	21.40	0.04	12.35	0.02	0.37	0.01	0.03		97.32	GbVn
6K-465-3	A5C	7	55.56	0.80	0.16	0.08	1.58	0.54	0.01	0.01	6.51	0.29	20.22	0.33	12.27	0.15	0.54	0.15	0.03	0.01	97.04	GbVn
6K-465-3	A8C	10	52.22	0.87	0.16	0.13	6.13	1.07	0.51	0.14	3.94	0.27	21.12	1.39	10.86	2.30	1.23	0.18			96.29	NrMat
RC27-9-6-2	PL11	1	45.1	0.0	2.03	0.02	12.69	0.01	1.01	0.08	3.75	0.03	18.28	0.01	12.20	0.01	2.94	0.02	0.59	0.04	98.63	FarMat
RC27-9-6-2	PL21	1	44.8	0.0	2.12	0.02	13.61	0.01	1.47	0.06	4.13	0.03	18.18	0.01	12.30	0.01	3.19	0.02	0.50	0.05	100.34	FarMat
RC27-9-6-2	PL31	1	45.0	0.0	2.26	0.02	12.74	0.01	1.04	0.07	3.71	0.03	18.28	0.01	12.53	0.01	3.09	0.02	0.47	0.04	99.19	FarMat
RC27-9-6-2	PL41	1	44.8	0.0	1.75	0.02	12.68	0.01	0.97	0.08	3.73	0.03	18.60	0.01	12.52	0.01	3.11	0.02	0.44	0.05	98.68	FarMat
RC27-9-6-2	PL51	1	44.8	0.0	1.85	0.02	12.56	0.01	0.95	0.08	4.08	0.03	18.40	0.01	12.44	0.01	3.13	0.02	0.46	0.05	98.75	FarMat
RC27-9-6-2	PL61	1	44.4	0.0	1.59	0.02	12.33	0.01	1.46	0.06	3.82	0.03	18.15	0.01	12.59	0.01	2.99	0.02	0.46	0.05	97.81	FarMat

^a Amphibole type, grain number and core (C) or interstitial (I) analysis. Low temperature metamorphic amphiboles are actinolite (A) and high temperature igneous amphiboles are pargasite (P). L indicates data from Lee (1997).

^b Abbreviations for lithologies: GbVn indicates a gabbroic vein and FarMat indicates >5 cm from a clinopyroxenite vein.

Table 5.10: Trace element concentrations in Cpx by ion probe, in ppm.

Sample	Grain ^a	Type ^b	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^c
KH1 ^d	Cpx		1.22	5.39	5.56	2.12	0.75	2.30	1.12	0.99							
KL2G ^e	Basalt										2.27	15200	290	362.00	27	160.00	
KLBI, Ave, n=21	Cpx		0.22	1.86	3.29	1.70	0.70	2.83	1.67	1.49							
KLBI, Std Dev	Cpx		0.04	0.21	0.38	0.25	0.07	0.25	0.18	0.18							
KLBI, % Error	Cpx		17	11	11	15	10	8.9	11	12							
Count % Error ^f	Cpx		21	13	10	10	8.2	4.9	6.2	5.5	0.3	0.6	0.2	5.9	2.2	7.0	
Duplicate % Error ^g	Cpx		32	37	16	17	17	15	18	19	7.6	11	6.8	14	9.4	17	
6K-458-1	Cpx1	Core	0.59	1.86	1.51	1.28	0.38	2.18	1.24	1.21	0.56	1470	7430	3.05	18.36	5.39	
6K-458-1	Cpx1	Rim	0.73	3.55	3.35	1.78	0.47	3.83	2.16	1.85	0.48	2122	8246	1.72	22.13	23.22	
6K-458-1	Cpx2	Ints	0.66	3.33	3.22	1.95	0.39	4.10	2.06	1.86	0.44	1979	6520	3.11	21.82	26.93	
6K-458-1	Cpx3	Ints	0.79	3.28	3.01	1.46	0.48	3.09	1.58	1.53	0.57	2274	8182	3.08	27.37	21.43	
6K-458-3	Cpx2	Ints	0.33	0.69	1.41	1.25	0.46	2.82	1.43	1.26	0.40	1908	5112	1.84	19.67	3.90	
6K-458-3	Cpx3	Core	0.09	0.22	1.66	1.72	0.55	3.89	1.35	1.88	0.39	2514	6894	3.81	21.41	4.96	
6K-458-3	Cpx3	Rim	0.05	0.25	1.83	1.25	0.45	3.72	1.56	1.77	0.37	2316	5660	4.76	23.93	5.28	
6K-458-3	Cpx4	Core	0.03	0.15	0.71	0.78	0.31	1.82	1.06	0.96	0.44	1374	6040	3.66	13.97	1.83	
6K-458-3	Cpx4	Rim	0.11	0.28	1.05	0.61	0.33	2.04	1.47	1.16	0.49	1541	5624	2.75	15.64	3.11	
6K-458-3	Cpx5	Ints	0.06	0.10	0.78	0.63	0.27	1.93	1.62	1.09	0.45	1934	5006	4.10	16.11	3.98	
6K-458-3	Cpx6	Ints	0.43	1.60	1.51	1.01	0.35	3.18	2.03	1.88							
6K-465-2	Cpx1	Core1	0.01	0.04	0.47	0.70	0.33	2.60	1.53	1.60	0.33	1312	5485	3.90	15.78	1.83	Dep
6K-465-2	Cpx1	Core2	0.04	0.18	0.67	0.85	0.36	2.68	1.83	1.59							Dep
6K-465-2	Cpx1	Rim1	0.01	0.05	0.71	0.90	0.30	3.31	1.82	1.72	0.31	1858	5802	2.81	20.02	2.69	Dep
6K-465-2	Cpx1	Rim2	0.03	0.04	0.53	0.73	0.21	1.82	1.23	0.89							Dep
6K-465-2	Cpx2	Core1	0.00	0.03	0.37	0.47	0.20	1.64	1.13	1.10	0.22	1003	4291	1.16	6.63	0.80	Dep
6K-465-2	Cpx2	Core2	0.00	0.03	0.31	0.44	0.20	1.29	0.84	0.90	0.28	1169		1.59	12.22	1.66	Dep
6K-465-2	Cpx2	Core3	0.00	0.05	0.40	0.44	0.22	1.44	0.71	0.77	0.27	1183		1.08	11.40	1.30	Dep
6K-465-2	Cpx2	Core4	0.00	0.03	0.32	0.57	0.22	1.51	0.83	1.00							Dep
6K-465-2	Cpx2	Core5	0.01	0.02	0.30	0.49	0.27	1.77	1.09	1.07							Dep
6K-465-2	Cpx2	Rim	0.00	0.04	0.47	0.65	0.17	2.16	1.17	1.26	0.26	1584		6.17	16.97	2.19	Dep
6K-465-2	Cpx3	Core	0.02	0.04	0.41	0.48	0.24	1.97	1.01	1.19	0.22	1073	4669	1.32	10.13	1.33	Dep
6K-465-2	Cpx3	Rim	0.02	0.04	0.50	0.75	0.26	2.64	1.60	1.57	0.24	1774	4506	1.33	14.95	1.61	Dep
6K-465-2	Cpx4	Core	0.01	0.03	0.40	0.56	0.29	2.54	1.63	1.73	0.29	1681	6007	1.47	23.18	2.93	Dep
6K-465-2	Cpx4	Rim	0.01	0.05	0.45	0.41	0.16	1.68	0.98	1.11	0.30	1825	6264	3.42	21.05	2.88	Dep
6K-465-2	Cpx5	Core1	0.86	3.77	2.48	1.10	0.38	2.69	1.92	1.98	0.52	1437	5768	2.27	27.20	7.13	Enr
6K-465-2	Cpx5	Core2	0.97	4.29	2.86	1.41	0.53	3.59	2.78	2.87	0.36	1283		1.27	22.06	2.92	Enr
6K-465-2	Cpx5	Rim	0.47	2.09	1.82	1.27	0.38	3.17	2.65	2.42	0.17	1877		1.17	38.52	11.86	Enr
6K-465-2	Cpx6	Core1	0.12	0.32	0.47	0.50	0.22	1.63	1.08	0.97	0.29	1069		1.02	13.14	1.19	Trans
6K-465-2	Cpx6	Mdpt2	0.05	0.14	0.45	0.52	0.26	1.69	1.02	1.09	0.37	1143		1.28	13.21	1.21	Trans
6K-465-2	Cpx6	Mdpt3	0.16	0.62	0.73	0.69	0.26	2.34	1.53	1.83	0.37	1442	5376	3.21	22.45	3.49	Trans
6K-465-2	Cpx6	Mdpt4	0.06	0.11	0.35	0.56	0.26	1.64	1.02	0.86	0.30	1001		0.98	11.30	1.12	Trans

Table 5.10: Cpx trace elements, Cont.

Sample	Grain ^a	Type ^b	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^c
6K-465-2	Cpx6	Mdpt5	0.10	0.23	0.47	0.52	0.28	1.54	1.09	0.91	0.30	1053		1.01	13.07	1.27	Trans
6K-465-2	Cpx6	Mdpt6	0.18	0.71	0.67	0.64	0.25	1.83	1.34	1.42	0.29	1108		1.13	14.55	1.45	Trans
6K-465-2	Cpx6	Mdpt7	0.38	1.34	1.14	0.72	0.25	2.39	1.63	2.24	0.41	1426	5305	1.72	26.16	3.58	Trans
6K-465-2	Cpx6	Rim8	0.38	1.73	2.17	1.10	0.32	2.71	1.77	1.97	0.27	1233		0.93	21.25	4.69	Trans
6K-465-2	Cpx6	Rim9	0.41	2.05	1.77	0.99	0.30	3.12	2.21	2.78	0.48	1774	6731	3.06	30.18	7.65	Trans
6K-465-2	Cpx7	Core	0.17	0.60	0.81	0.73	0.35	2.40	1.93	1.70							Enr
6K-465-2	Cpx7	Rim	0.16	0.65	0.77	0.48	0.29	2.06	1.55	1.40							Enr
6K-465-2	Cpx8	Core1	0.02	0.08	0.50	0.69	0.33	2.05	1.28	1.19	0.43	1222		0.92	14.25	1.24	Dep
6K-465-2	Cpx8	Core2	0.02	0.06	0.48	0.52	0.18	1.30	0.92	0.64	0.42	1186		1.26	13.19	1.37	Dep
6K-465-2	Cpx8	Core3	0.02	0.06	0.39	0.49	0.26	1.48	1.02	0.92							Dep
6K-465-2	Cpx8	Rim1	0.07	0.22	0.44	0.45	0.20	1.50	1.10	0.87							Dep
6K-465-2	Cpx8	Rim2	0.09	0.33	0.75	0.58	0.28	2.23	1.70	1.29	0.38	1528		0.83	18.63	1.76	Dep
6K-465-2	Cpx8	Rim3	0.09	0.34	0.76	0.72	0.30	2.41	1.71	1.67	0.40	1597		0.80	18.87	1.75	Dep
6K-465-2	Cpx9	Rim1	0.60	2.98	2.72	1.24	0.28	3.15	2.11	1.95	0.40	1396	5186	2.69	28.75	10.15	Enr
6K-465-2	Cpx9	Rim2	0.94	5.33	4.84	2.23	0.51	5.31	4.17	3.76	0.33	1398		1.03	31.12	9.39	Enr
6K-465-2	Cpx9	Rim3	1.34	7.99	7.57	3.83	0.79	7.78	5.58	5.46	0.37	2075		1.41	42.81	16.63	Enr
6K-465-2	Cpx10	Ints	0.01	0.04	0.49	0.78	0.24	3.08	1.82	1.50	0.34	1705	4662	3.45	9.89	1.32	Dep
6K-465-2	Cpx11	Core1	0.01	0.06	0.54	0.81	0.29	2.05	1.31	1.35							Dep
6K-465-2	Cpx11	Core2	0.01	0.05	0.50	0.66	0.23	2.21	1.31	1.24							Dep
6K-465-2	Cpx11	Rim	0.01	0.05	0.44	0.57	0.19	1.95	1.16	1.12							Dep
6K-465-2	Cpx12	Ints	0.01	0.04	0.42	0.52	0.23	1.94	1.24	1.29							Dep
6K-465-3	Cpx4	Ints	6.54	42.87	63.93	33.37	7.93	52.70	24.87	28.87	0.58	3955	290	19.87	232.07	242.43	GbVn
6K-465-3	Cpx5	Rim	2.68	18.53	30.67	16.08	3.05	25.63	13.34	12.42	0.72	3291	246	14.75	420.08	257.52	GbVn
6K-465-3	Cpx6	Core	5.43	34.90	49.02	23.04	5.50	36.59	19.51	20.54	0.47	3390	66	17.11	225.33	181.01	GbVn
6K-465-3	Cpx7	Core	3.97	21.72	23.82	9.14	2.10	9.52	4.07	1.78	1.09	2464	4608	11.83	72.03	125.56	NrMat
6K-465-3	Cpx7	Rim2	4.10	23.02	28.08	12.62	2.97	14.74	7.79	9.07	0.49	2316	3416	3.96	95.91	119.51	NrMat
6K-465-3	Cpx8	Ints	7.02	41.01	48.90	20.53	2.94	22.43	10.57	8.65	0.47	3925	4741	4.01	154.29	444.41	NrMat
6K-465-3	Cpx9	Ints	9.67	58.39	74.46	29.17	3.89	32.42	16.40	11.78	0.50	3371	4720	3.80	161.97	431.24	NrMat
6K-465-3	Cpx10	Core	4.44	23.40	18.79	7.05	0.71	8.57	4.45	3.85	0.49	2480	5873	3.71	50.80	83.98	NrMat
6K-465-3	Cpx10	Rim	4.02	22.33	19.68	7.75	0.90	8.64	4.02	3.97	0.49	2047	5306	1.31	47.56	75.40	NrMat
6K-465-3	Cpx12	Core1	5.08	22.81	18.57	6.58	1.40	7.40	4.82	4.18	0.56	2185	4739	2.73	69.58	124.38	NrMat
6K-465-3	Cpx12	Rim	4.87	26.78	22.62	9.49	1.44	9.09	6.08	5.32							NrMat
6K-466-1	Cpx1	Core	0.01	0.06	0.41	0.49	0.20	1.57	0.70	0.91	0.46	1294	9529	1.22	10.77	2.31	
6K-466-1	Cpx1	Rim	0.01	0.05	0.52	0.39	0.17	1.75	0.89	0.75	0.41	1400	8847	1.61	12.37	2.53	
6K-466-1	Cpx3	Ints	0.06	0.07	0.47	0.38	0.20	1.50	0.85	0.79	0.40	1480	8195	1.77	11.77	2.89	
6K-466-1	Cpx3D	Ints									0.39	1628	8259	2.38	12.75	3.59	
6K-643-01	Cpx1	Ints	0.02	0.04	0.23	0.29	0.15	1.74	0.93	1.30	0.20	913	6261	1.15	9.90	0.94	
6K-643-01	Cpx1D	Ints	0.02	0.02	0.25	0.32	0.26	2.36	1.26	1.60	0.18	1032	6518	1.41	10.39	0.90	
6K-643-01	Cpx2	Core	0.02	0.04	0.26	0.25	0.18	1.37	0.94	0.85	0.21	579	5425	1.27	5.30	0.66	
6K-643-01	Cpx2	Rim	0.02	0.04	0.20	0.25	0.12	0.83	0.61	0.54	0.34	918	5538	1.39	8.98	1.00	
6K-643-01	Cpx4	Core	0.03	0.04	0.25	0.33	0.15	0.92	0.65	0.65	0.42	809	5284	1.33	9.26	1.87	

Table 5.10: Cpx trace elements, Cont.

Sample	Grain ^a	Type ^b	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^c
6K-643-01	Cpx4	Rim	0.02	0.04	0.39	0.39	0.19	1.35	0.95	0.81	0.35	714	4664	1.63	7.01	1.31	
6K-643-01	Cpx8	Symp	0.02	0.06	0.31	0.45	0.16	1.34	0.71	0.80	0.29	905	5245	1.41	8.71	1.71	
6K-643-01	Cpx8D	Symp	0.02	0.06	0.28	0.54	0.17	1.49	0.83	0.92	0.31	1261	4916	1.78	12.00	1.12	
6K-643-06	Cpx5	Ints	0.03	0.05	0.20	0.25	0.13	0.86	0.54	0.51	0.44	1233	9453	2.39	8.26	1.69	
6K-643-06	Cpx6	Ints	0.02	0.05	0.15	0.32	0.12	0.92	0.58	0.55	0.48	1223	10072	1.81	8.37	6.04	
6K-643-13	Cpx1	Ints	0.01	0.01	0.10	0.30	0.10	0.97	0.71	0.63	0.30	961	5195	1.90	9.95	0.73	
6K-643-13	Cpx2	Core	0.02	0.02	0.17	0.31	0.23	1.42	0.82	0.88	0.13	602	5945	1.25	5.23	0.66	
6K-643-13	Cpx2	Rim	0.02	0.02	0.09	0.18	0.09	1.33	0.89	0.75	0.24	877	6416	1.60	9.60	0.73	
6K-643-13	Cpx3	Ints	0.00	0.03	0.11	0.25	0.08	0.87	0.63	0.46	0.28	1047	3633	1.68	8.53	0.76	
6K-643-15	Cpx1	Core	9.62	67.80	86.11	37.97	7.55	84.70	36.23	35.82	0.55	8375	5575	20.85	428.80	318.73	GbVn
6K-643-15	Cpx1	Rim	9.12	59.61	77.20	33.49	7.86	80.46	35.36	38.06	0.57	7132	5853	25.66	342.46	273.51	GbVn
6K-643-15	Cpx2	Symp	4.87	35.34	53.63	24.10	4.70	54.37	25.09	21.68	0.46	4393	7116	9.41	291.10	276.88	GbVn
6K-643-15	Cpx2D	Symp	4.58	37.56	54.23	24.24	4.87	61.34	26.05	25.41	0.48	4194	7275	11.85	286.94	306.50	GbVn
6K-643-15	Cpx3	Symp	2.22	14.48	15.77	6.98	1.37	19.53	9.62	11.05	0.50	1899	7709	2.90	202.00	219.88	NrMat
6K-643-15	Cpx4	Symp	2.85	17.44	16.75	6.41	1.27	14.30	6.82	4.68	0.47	1316	7317	2.12	90.97	72.18	NrMat
6K-643-15	Cpx4D	Symp									0.44	1229	6696	2.41	83.05	64.57	NrMat
6K-643-15	Cpx5	Symp	2.28	14.57	20.97	9.33	2.22	29.86	14.34	16.91	0.41	1535	6811	5.39	202.21	195.06	NrMat
6K-643-15	Cpx6	Core	0.05	0.07	0.17	0.33	0.14	0.94	0.55	0.69	0.48	760	6496	1.06	7.47	1.05	FarMat
6K-643-15	Cpx6	Rim	0.14	0.48	0.56	0.42	0.19	1.42	0.94	1.15	0.52	1116	6228	6.48	11.04	1.91	FarMat
6K-643-15	Cpx7	Core	0.03	0.06	0.17	0.38	0.17	1.13	0.70	0.71	0.42	912	7277	1.38	8.72	1.26	FarMat
6K-643-15	Cpx7	Rim	0.12	0.37	0.31	0.39	0.20	1.34	0.85	1.01	0.44	1003	6380	6.62	9.95	1.71	FarMat
6K-643-15	Cpx8	Ints	0.06	0.10	0.15	0.37	0.14	0.95	0.57	0.70	0.61	946	6623	5.61	8.42	1.27	FarMat
6K-643-15	Cpx8D	Ints									0.52	1001	6890	4.93	8.63	2.25	FarMat
6K-643-15	Cpx9	Core	0.07	0.17	0.20	0.31	0.12	1.00	0.62	0.55	0.54	827	6428	1.08	7.34	1.75	FarMat
6K-643-15	Cpx9	Rim	0.36	1.46	1.13	0.55	0.29	1.36	0.81	0.97	0.59	896	5954	26.15	9.44	2.38	FarMat
6K-643-15	Cpx13	Ints4									0.59	1618	6323	9.65	68.95	27.27	FarMat
6K-643-15	Cpx13	Ints5	1.71	8.77	6.66	2.55	1.06	4.56	3.14	3.62	0.67	1124	6303	8.21	37.49	11.16	FarMat
6K-643-15	Cpx13	Ints6	1.60	9.48	7.79	3.20	0.98	5.16	3.02	2.67	0.54	1296	6007	9.28	39.35	17.90	FarMat
6K-643-15	Cpx13	Ints7	2.36	11.48	9.34	4.40	1.29	7.35	4.68	5.63							FarMat
6K-643-15	Cpx13	Ints8	1.92	9.63	8.64	3.57	1.19	7.85	5.45	6.16							FarMat
6K-649-5	Cpx1	Core	0.01	0.00	0.22	0.26	0.13	1.13	0.85	0.96	0.26	683	6554	2.20	9.12	0.65	
6K-649-5	Cpx1	Rim	0.03	0.02	0.25	0.27	0.17	1.33	0.92	0.91	0.27	802	5774	1.60	9.62	0.95	
6K-649-5	Cpx2	Ints	0.03	0.04	0.15	0.23	0.13	1.42	0.66	0.96	0.17	947	11618	1.26	13.05	1.17	
6K-649-5	Cpx4	Ints	0.02	0.03	0.23	0.25	0.12	1.28	0.85	1.04							
6K-649-5	Cpx5	Core	0.02	0.02	0.10	0.15	0.09	1.17	0.67	0.95							
6K-649-5	Cpx5	Rim	0.02	0.02	0.22	0.17	0.13	1.40	0.77	1.06	0.30	756	4771	1.28	8.10	1.17	
6K-649-5	Cpx6	Ints	0.01	0.02	0.16	0.12	0.07	0.94	0.53	1.04							
6K-649-5	Cpx7	Ints	0.02	0.01	0.11	0.08	0.06	0.87	0.49	0.76							
6K-649-5	Cpx8	Ints	0.01	0.02	0.18	0.24	0.08	1.32	0.84	1.01							
6K-649-5	Cpx9	Ints	0.02	0.02	0.17	0.15	0.11	1.35	0.71	1.01	0.33	748	4813	3.71	8.20	2.24	
6K-651-7	Cpx1	Ints	0.02	0.13	0.48	0.53	0.33	1.70	0.85	0.61	0.35	1186	5080	2.59	9.58	1.08	PyxVn

Table 5.10: Cpx trace elements, Cont.

Sample	Grain ^a	Type ^b	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^c
6K-651-7	Cpx1D	Ints	0.03	0.12	0.55	0.41	0.31	1.77	1.18	1.13							
6K-651-7	Cpx2	Core	0.04	0.06	0.29	0.50	0.21	1.11	0.73	0.55	0.33	1083	5110	3.19	9.19	0.83	PyxVn
6K-651-7	Cpx2	Rim	0.06	0.07	0.24	0.51	0.22	1.11	0.68	0.58	0.39	1416	4373	2.89	9.56	1.48	PyxVn
AIH107-61-83	Cpx1	Core	0.02	0.03	0.38	0.38	0.20	1.56	1.19	1.01	0.64	965	4207	0.74	13.59	0.99	PyxVn
AIH107-61-83	Cpx1	Rim	0.01	0.02	0.26	0.43	0.25	1.62	1.42	1.15	0.65	926	4659	1.04	13.96	0.60	
JR31-52-03	Cpx4	Core	0.01	0.03	0.27	0.38	0.19	1.36	0.77	0.78	0.53	912	10956	0.96	11.03	0.87	
JR31-52-03	Cpx4	Rim	0.01	0.06	0.26	0.43	0.17	1.61	0.77	0.79	0.51	941	10004	0.86	11.15	0.91	
JR31-52-03	Cpx5	Core	0.01	0.03	0.22	0.34	0.13	1.09	0.62	0.63							
JR31-52-03	Cpx5	Rim	0.01	0.03	0.26	0.31	0.18	1.01	0.63	0.62							
JR31-52-03	Cpx5D	Rim	0.01	0.05	0.29	0.39	0.17	1.18	0.65	0.63							
JR31-52-03	Cpx6	Ints	0.01	0.05	0.30	0.31	0.17	1.15	0.64	0.61							
JR31-52-03	Cpx7	Core	0.00	0.03	0.35	0.47	0.16	1.19	0.74	0.71	0.50	904	10097	0.93	10.13	0.90	
JR31-52-03	Cpx7	Rim	0.03	0.04	0.34	0.36	0.20	1.41	0.83	0.84	0.49	1071	8866	1.15	10.40	0.96	
JR31-52-03	Cpx9	Ints	0.00	0.03	0.26	0.30	0.19	1.43	0.80	0.81	0.48	1018	8834	1.35	10.20	0.82	
JR31-52-04	Cpx3	Core	0.00	0.01	0.16	0.21	0.06	0.65	0.38	0.32							
JR31-52-04	Cpx3	Rim	0.00	0.03	0.31	0.23	0.10	0.72	0.55	0.52							
JR31-52-04	Cpx4	Core	0.01	0.03	0.33	0.40	0.15	1.34	0.86	0.71	0.39	1052	10374	0.89	12.14	0.81	
JR31-52-04	Cpx4	Rim	0.00	0.03	0.34	0.40	0.15	1.38	0.84	0.70	0.30	1426	13683	1.12	16.93	1.26	
JR31-52-05	Cpx1	Core	0.01	0.02	0.30	0.49	0.20	1.43	0.77	0.76	0.43	920	10969	0.59	12.06	1.04	
JR31-52-05	Cpx1	Rim	0.01	0.01	0.26	0.34	0.22	1.39	0.77	0.82	0.29	1126	7950	0.64	12.58	1.36	
JR31-52-05	Cpx3	Core	0.01	0.02	0.28	0.42	0.21	1.52	0.89	0.88	0.35	1126	10428	0.91	14.01	0.88	
JR31-52-05	Cpx5	Ints	0.01	0.02	0.21	0.33	0.12	1.00	0.69	0.61	0.39	1002	8800	0.91	11.27	0.85	
JR31-52-08	Cpx1	Core	0.01	0.02	0.25	0.27	0.16	1.12	0.64	0.71	0.37	730	10371	0.52	7.62	0.61	
JR31-52-08	Cpx1	Rim	0.01	0.04	0.31	0.32	0.18	1.29	0.86	0.70	0.47	1097	9310	0.87	10.98	0.75	
JR31-52-08	Cpx1D	Rim	0.01	0.04	0.34	0.41	0.21	1.48	0.77	0.82							
JR31-52-08	Cpx3	Ints	0.01	0.04	0.41	0.44	0.17	1.39	0.79	0.72	0.50	1213	11035	0.99	12.42	0.99	
JR31-52-08	Cpx4	Rim	0.01	0.04	0.30	0.40	0.16	1.36	0.83	0.92	0.42	959	5842	2.02	8.68	0.78	
JR31-52-08	Cpx5	Ints	0.03	0.07	0.46	0.49	0.23	1.52	0.92	0.85	0.48	1086	10099	0.91	10.48	0.83	
Kn162-19-41	Cpx1	Core	0.01	0.03	0.39	0.56	0.30	2.27	1.22	1.10	0.28	1310	3637	2.07	13.25	1.80	PyxVn
Kn162-19-41	Cpx1	Rim	0.03	0.06	0.41	0.71	0.36	2.32	1.03	1.12	0.26	1003	3251	2.76	11.32	1.89	PyxVn
Kn162-19-41	Cpx2	Ints	0.01	0.04	0.31	0.40	0.28	2.09	1.47	1.41	0.20	1445	3755	3.91	13.42	1.81	PyxVn
Kn162-19-41	Cpx3	Ints	0.01	0.02	0.21	0.46	0.31	1.82	1.27	1.01	0.24	1466	4119	1.63	12.37	1.34	PyxVn
Kn162-19-41	Cpx4	Core	0.01	0.03	0.23	0.48	0.26	1.92	1.41	1.31	0.24	1190	3887	1.37	12.64	1.08	PyxVn
Kn162-19-41	Cpx4D	Core	0.00	0.01	0.42	0.69	0.19	1.54	1.12	0.71							PyxVn
Kn162-19-41	Cpx4	Rim	0.00	0.01	0.29	0.43	0.25	1.98	1.31	1.07	0.25	1457	4197	1.66	14.26	1.38	PyxVn
Kn162-19-41	Cpx4D	Rim	0.01	0.06	0.26	0.58	0.21	1.69	0.96	0.71							PyxVn
Kn162-19-45	Cpx1	Core	0.02	0.03	0.35	0.72	0.38	2.81	1.67	1.82	0.27	1338	6102	1.74	12.70	0.94	
Kn162-19-45	Cpx1	Rim	0.02	0.02	0.63	0.71	0.40	3.06	1.66	1.53	0.24	1455	5215	2.53	12.17	1.80	
Kn162-47-08	Cpx1	Ints	0.01	0.04	0.34	0.47	0.28	2.10	1.32	1.04	0.67	1849	4951	1.46	14.18	2.63	
Kn162-47-08	Cpx2	Ints		0.03	0.39	0.52	0.28	2.34	1.43	1.23	0.74	2456	5225	1.94	16.29	3.42	
Kn162-47-08	Cpx3	Ints	0.01	0.03	0.39	0.52	0.28	2.34	1.43	1.23	0.52	1640	3961	1.46	14.24	2.86	

Table 5.10: Cpx trace elements, Cont.

Sample	Grain ^a	Type ^b	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^c
Kn162-47-08	Cpx4	Ints	0.02	0.06	0.45	0.41	0.26	1.85	1.40	1.25	0.45	1970	3788	1.60	15.48	2.67	
Kn162-47-08	Cpx5	Ints	0.02	0.06	0.42	0.64	0.30	1.84	1.38	1.05	0.49	1740	3548	1.41	14.11	3.02	
Kn162-47-08	Cpx5D	Ints	0.62	2.81	3.10	0.76	0.22	0.50	0.24	0.21	0.80	556	6931	34.37	2.93	7.28	
Kn162-47-09	Cpx2	Ints	0.56	2.65	2.75	0.82	0.22	0.47	0.23	0.27	0.80	594	8034	36.92	3.33	8.54	
Kn162-47-09	Cpx3	Core	0.51	2.66	2.70	0.81	0.21	0.44	0.19	0.26	0.76	594	7411	36.35	3.06	8.19	
Kn162-47-09	Cpx3	Rim	0.00	0.05	0.66	0.86	0.30	1.84	1.21	1.02	0.38	1216	4203	1.92	13.58	2.91	Pyx Vn
Kn162-47-10	Cpx1	Core	0.04	0.12	0.88	0.85	0.37	2.09	1.22	1.09	0.36	1869	3750	2.02	15.64	3.35	Pyx Vn
Kn162-47-10	Cpx1	Rim	0.03	0.15	1.08	0.96	0.38	2.24	1.53	1.18	0.30	1280	2807	2.19	12.71	2.98	Pyx Vn
Kn162-47-10	Cpx2	Ints	0.03	0.10	0.68	0.82	0.31	2.13	1.32	1.07	0.36	1169	3616	2.33	13.04	2.50	NrMat
Kn162-47-10	Cpx5	Ints	0.03	0.10	0.44	0.69	0.24	1.42	0.87	0.63	0.46	1472	5328	1.70	12.69	2.10	FarMat
Kn162-47-10	Cpx6	Core	0.04	0.10	0.44	0.51	0.57	1.36	0.86	0.66	0.46	1471	4772	1.90	12.98	2.15	FarMat
Kn162-47-10	Cpx6	Rim	0.03	0.03	0.51	0.57	0.18	1.71	0.85	0.97	0.45	1498	4459	3.06	14.43	3.53	Pyx Vn
Kn162-47-12	Cpx1	Ints	0.03	0.10	0.55	0.44	0.21	1.71	0.85	0.97	0.45	1498	4459	3.06	14.43	3.53	Pyx Vn
Kn162-47-12	Cpx2	Core	0.02	0.13	0.61	0.69	0.29	1.79	0.83	0.78	0.45	1293	4837	3.57	13.91	2.81	Pyx Vn
Kn162-47-12	Cpx2	Rim	0.02	0.21	0.78	0.64	0.28	2.10	1.16	0.78	0.37	1403	3953	3.07	14.89	3.42	Pyx Vn
Kn162-47-22	Cpx1	Core	0.01	0.12	0.89	0.94	0.34	2.32	1.32	1.16	0.51	1530	4996	1.72	14.83	2.19	Adj Vn
Kn162-47-22	Cpx1	Rim	0.00	0.10	0.84	0.87	0.38	2.30	1.37	1.22	0.46	1718	4593	3.02	14.35	2.40	Adj Vn
Kn162-47-22	Cpx2	Core	0.01	0.09	1.11	0.76	0.41	2.14	1.31	1.16	0.50	1651	5173	1.89	14.14	2.40	Pyx Vn
Kn162-47-22	Cpx2	Rim	0.01	0.10	1.00	1.01	0.38	2.32	1.53	1.25	0.46	1820	5217	1.71	15.85	2.80	Pyx Vn
Kn162-47-22	Cpx2D	Rim	0.00	0.09	0.65	0.76	0.37	1.77	1.16	1.03	0.51	1509	5277	1.92	15.99	2.47	Pyx Vn
Kn162-47-22	Cpx3	Core	0.01	0.09	0.92	0.91	0.35	2.28	1.30	1.24	0.48	1738	4912	2.23	14.73	2.12	Pyx Vn
Kn162-47-22	Cpx3	Rim	0.01	0.09	0.92	0.91	0.35	2.28	1.30	1.24	0.48	1738	4912	2.23	14.73	2.12	Pyx Vn
Kn162-47-22	Cpx4	Ints	0.02	0.04	0.35	0.43	0.16	1.55	1.25	0.95	0.53	1612	5572	2.29	14.90	2.10	Pyx Vn
Kn162-47-24	Cpx1	Core	0.02	0.04	0.35	0.43	0.16	1.55	1.25	0.95	0.38	1174	5766	1.20	12.23	1.56	
Kn162-47-24	Cpx1	Rim	0.01	0.03	0.37	0.57	0.20	1.34	1.14	0.87	0.42	1371	5107	1.55	13.45	1.66	
Kn162-47-24	Cpx2	Core	0.01	0.03	0.44	0.47	0.23	1.81	1.13	1.07	0.38	1163	5765	1.53	12.66	1.25	
Kn162-47-24	Cpx2	Rim	0.02	0.06	0.35	0.48	0.20	1.72	1.18	1.01	0.39	1382	5132	1.81	14.09	1.60	
Kn162-47-24	Cpx3	Symp	0.01	0.03	0.22	0.34	0.23	1.54	1.25	1.04	0.32	1201	4659	1.41	12.53	1.02	
Kn162-47-25	Cpx1	Ints	0.02	0.18	1.25	0.87	0.40	1.65	0.93	0.87	0.67	2498	5883	2.76	15.06	9.09	NrMat
Kn162-47-25	Cpx2	Ints	0.01	0.10	0.55	0.67	0.29	1.11	0.72	0.49	0.77	2480	5837	2.25	12.98	7.76	NrMat
Kn162-47-25	Cpx11	Core	0.02	0.28	1.07	0.93	0.25	1.24	0.58	0.41	0.47	2227	5144	3.38	12.10	4.02	Pyx Vn
Kn162-47-25	Cpx11	Rim	0.02	0.26	1.22	0.74	0.35	1.60	0.77	0.84	0.34	2800	5928	3.81	13.86	4.78	Pyx Vn
Kn162-47-25	Cpx13	Ints	0.01	0.18	0.74	0.72	0.15	0.93	0.53	0.50	0.49	2400	5356	3.16	12.69	3.83	Pyx Vn
Kn162-47-25	Cpx14	Ints	0.04	0.35	1.06	0.88	0.28	1.44	0.70	0.68	0.36	2563	6003	3.90	15.00	5.56	Pyx Vn
Kn162-47-26	Cpx1	Core	0.02	0.21	1.13	0.72	0.27	1.42	0.92	1.04	0.63	1220	4072	5.49	9.89	2.81	
Kn162-47-26	Cpx1	Rim	0.02	0.20	0.84	0.61	0.24	1.68	0.83	0.87	0.64	1307	4157	4.92	9.56	2.31	
Kn162-47-26	Cpx2	Ints	0.03	0.18	0.89	0.53	0.21	1.35	0.72	0.93	0.56	1389	4077	5.12	10.08	2.12	
Kn162-47-26	Cpx3	Ints	0.03	0.18	0.89	0.53	0.21	1.35	0.72	0.93	0.56	1389	4077	5.12	10.08	2.12	
Kn162-47-29	Cpx1	Core	0.00	0.04	0.41	0.31	0.15	1.10	0.71	0.72	0.46	1819	6489	2.54	16.57	3.14	
Kn162-47-29	Cpx1	Rim	0.00	0.05	0.25	0.35	0.17	1.05	0.61	0.67	0.36	1923	5591	2.87	19.02	3.70	
Kn162-47-29	Cpx2	Core	0.00	0.04	0.42	0.37	0.13	1.13	0.73	0.60	0.37	1548	5815	2.42	15.88	3.90	

Table 5.10: Cpx trace elements, Cont.

Sample	Grain ^a	Type ^b	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^c
Kn162-47-29	Cpx2	Rim	0.03	0.15	1.01	1.02	0.37	1.78	0.97	0.92	0.34	1714	5106	2.24	18.39	3.62	
Kn162-47-29	Cpx3	Ints	0.01	0.18	1.05	0.85	0.41	1.86	1.15	0.99	0.33	2108	5556	2.64	18.16	3.72	
Kn162-47-36	Cpx1	Core	0.01	0.04	0.45	0.70	0.34	1.82	1.24	1.21	0.45	1237	4350	1.57	13.87	1.79	
Kn162-47-36	Cpx1	Rim									0.40	1419	3108	2.36	14.82	1.73	
Kn162-47-36	Cpx2	Core	0.02	0.05	0.55	0.80	0.30	1.78	1.21	1.27	0.47	1157	4760	1.62	13.52	1.31	
Kn162-47-36	Cpx2	Rim		0.02	0.59	0.70	0.20	2.28	1.15	1.35	0.48	1550	3835	1.96	15.85	2.33	
Kn162-47-36	Cpx3	Ints	0.01	0.03	0.47	1.11	0.32	2.12	1.43	1.10	0.54	1565	3685	2.07	14.15	2.63	
Kn162-47-49	Cpx1	Ints	0.09	0.21	0.45	0.43	0.21	1.58	1.06	0.89	0.26	1665	5020	2.43	13.31	2.48	
Kn162-47-49	Cpx2	Core	0.04	0.04	0.38	0.43	0.21	1.34	0.97	0.72	0.71	1384	6436	4.52	13.64	2.16	
Kn162-47-49	Cpx2	Rim	0.03	0.06	0.57	0.47	0.24	1.46	0.97	0.89	0.35	1279	3367	3.89	9.08	2.25	
Kn162-47-49	Cpx3	Ints	0.02	0.05	0.49	0.45	0.20	1.31	0.81	0.80	0.39	1409	5924	2.36	13.65	1.86	
PS86-6-26	Cpx1	Core	0.06	0.41	1.76	1.37	0.55	1.81	1.41	1.13	0.97	2500	4619	5.86	19.04	7.18	
PS86-6-26	Cpx1	Rim	0.03	0.39	1.77	1.22	0.54	2.54	1.03	1.36	0.83	2526	4006	5.10	15.92	6.50	
PS86-6-26	Cpx2	Ints	0.02	0.36	1.29	1.00	0.36	1.67	1.10	0.98	0.81	2315	4217	5.42	15.77	6.27	
PS86-6-38	Cpx1	Core	0.04	0.58	2.37	1.16	0.59	2.32	1.25	1.28	0.69	1741	4361	5.00	12.45	6.44	
PS86-6-38	Cpx1	Rim	0.06	0.60	2.43	1.19	0.59	2.80	1.31	1.36	0.91	2505	4641	6.20	17.87	10.44	
PS86-6-38	Cpx2	Ints	0.08	0.72	2.98	1.64	0.71	2.67	1.47	1.56	0.79	2093	4666	5.95	15.68	7.91	
PS86-6-38	Cpx3	Core	0.03	0.59	2.04	1.33	0.51	2.11	1.22	0.93	1.00	2823	4720	8.48	22.72	11.09	
PS86-6-38	Cpx3	Rim	0.03	0.66	2.53	1.58	0.61	2.48	1.19	1.33	0.96	2898	4520	7.83	21.06	11.17	
PS86-6-39	Cpx1	Core	0.05	0.76	2.21	0.92	0.46	2.16	1.20	1.32	1.02	1764	7291	11.98	14.82	10.21	
PS86-6-39	Cpx1	Rim	0.08	0.92	2.87	1.55	0.64	2.61	1.53	1.42	1.11	1933	6746	13.28	15.60	11.22	
PS86-6-39	Cpx2	Ints	0.09	0.88	2.18	1.51	0.54	2.40	1.22	1.34	1.08	1949	6689	13.22	15.36	12.77	
PS86-6-39	Cpx3	Core	0.09	0.98	2.54	1.70	0.50	1.96	1.03	0.92	1.01	1770	7042	13.60	15.41	10.72	
PS86-6-39	Cpx3	Rim	0.08	0.81	2.67	1.24	0.52	2.18	1.31	0.99	1.03	2088	5874	15.19	16.25	12.86	
PS86-6-39	Cpx4	Ints	0.12	0.94	2.67	1.78	0.65	2.18	1.18	1.12	1.16	2198	7165	15.19	16.79	13.33	
PS86-6-40	Cpx1	Ints	0.02	0.35	1.63	0.91	0.40	2.74	1.78	1.95	1.00	2324	4327	11.18	17.46	9.29	
PS86-6-40	Cpx1D	Ints									0.89	2281	4285	8.11	17.58	8.95	
PS86-6-40	Cpx2	Core	0.02	0.53	2.10	1.32	0.58	2.83	1.69	1.51	0.82	2195	6089	13.18	17.84	9.27	
PS86-6-40	Cpx2	Rim	0.02	0.49	1.50	1.19	0.48	2.97	1.67	1.72	0.78	2436	3504	19.44	17.40	11.15	
PS86-6-40	Cpx3	Core	0.04	0.58	2.05	1.37	0.59	2.67	1.42	1.46	0.97	2300	4761	13.10	18.86	10.19	
PS86-6-40	Cpx3	Rim	0.02	0.62	2.09	1.55	0.64	3.03	1.80	1.51	0.95	2424	4325	11.24	17.95	10.31	
RC27-6-9-2	CpxL1	Core	0.10	0.57	0.57	0.31	0.07	0.44	0.27	0.27	0.45	543	11399	24.39	4.00	2.06	FarMat
RC27-6-9-2	CpxL1D	Core									0.48	663	12280	24.88	4.37	2.89	FarMat
RC27-6-9-2	CpxL1	Rim	0.17	1.25	1.67	0.80	0.17	1.36	0.80	0.68	0.46	1090	10234	12.59	11.56	15.80	FarMat
RC27-6-9-2	CpxL1D	Rim									0.52	1062	11640	14.37	12.21	15.84	FarMat
RC27-6-9-2	CpxL2	Core	0.04	0.27	0.31	0.17	0.05	0.32	0.22	0.16	0.49	689	10333	21.12	3.18	4.30	FarMat
RC27-6-9-2	CpxL2	Rim	0.28	1.21	0.75	0.31	0.09	0.54	0.42	0.36	0.99	796	11316	27.59	4.19	2.02	FarMat
RC27-9-6-2	CpxL3	Rim									0.43	1196	13293	9.99	7.55	4.12	CpxVn
RC27-9-6-2	CpxL4	Ints									0.59	1217	11804	8.13	6.67	3.76	CpxVn
RC27-9-6-2	CpxL6	Core	0.03	0.24	0.41	0.29	0.10	0.61	0.38	0.35	0.64	911	12342	33.74	6.34	3.61	NrMat
RC27-9-6-2	CpxL6D	Core	0.04	0.19	0.39	0.24	0.09	0.56	0.36	0.34	0.56	918	11505	33.12	7.11	3.97	NrMat

Table 5.10: Cpx trace elements, Cont.

Sample	Grain ^a	Type ^b	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^c
RC27-9-6-2	CpxL6	Rim	0.03	0.23	0.48	0.27	0.09	0.70	0.41	0.36	0.54	976	10439	34.19	6.61	3.58	NrMat
RC27-9-6-2	CpxL7	Core	0.03	0.20	0.47	0.26	0.11	0.72	0.37	0.39							NrMat
RC27-9-6-2	CpxL7	Rim	0.03	0.19	0.50	0.26	0.09	0.71	0.46	0.41	0.50	1191	8951	30.72	6.73	4.04	NrMat
RC27-9-6-2	CpxL7D	Rim									0.46	1068	8796	28.59	6.66	3.63	NrMat
RC27-9-6-2	Cpx12	Core	0.55	2.41	1.87	0.51	0.22	0.95	0.55	0.47	0.60	700	8101	59.83	6.00	3.55	FarMat
RC27-9-6-2	Cpx12	Rim	0.49	2.21	1.52	0.50	0.23	1.04	0.60	0.55	0.50	906	6002	53.15	6.94	4.75	FarMat
RC27-9-6-2	Cpx13	Ints	0.49	2.06	1.64	0.54	0.19	1.04	0.56	0.54	0.52	790	5507	47.13	6.27	4.00	FarMat
RC27-9-6-2	Cpx14	Ints	0.43	2.10	1.40	0.55	0.24	0.86	0.47	0.49	0.42	842	7618	47.72	6.93	4.23	FarMat
RC27-9-6-2	Cpx15	Ints	0.43	2.10	1.74	0.59	0.18	0.73	0.58	0.38	0.43	837	5418	53.95	7.18	4.03	FarMat
RC27-9-6-2	Cpx21	Core	0.15	0.83	1.24	0.60	0.21	1.38	0.79	0.60	0.65	1202	7961	26.76	9.40	4.62	CpxVn
RC27-9-6-2	Cpx21	Rim	0.32	1.10	1.25	0.57	0.28	1.29	0.70	0.60	0.63	1177	8264	23.32	9.12	5.77	CpxVn
RC27-9-6-2	Cpx22	Core	0.20	1.10	1.51	0.72	0.24	1.34	0.65	0.59	0.48	1226	7351	26.11	8.93	4.46	CpxVn
RC27-9-6-2	Cpx23	Ints	0.20	0.92	1.19	0.61	0.17	1.13	0.58	0.50	0.53	1198	6508	24.68	8.84	4.76	CpxVn
RC27-9-6-5	Cpx1	Core1	0.86	2.77	1.54	0.64	0.36	1.77	1.17	1.23	0.29	1273	5768	1.63	18.74	4.23	Enr
RC27-9-6-5	Cpx1	Core2	0.86	2.29	1.45	0.66	0.36	2.04	0.98	1.36	0.48	1194	6389	2.26	14.35	1.05	Enr
RC27-9-6-5	Cpx1	Rim	1.12	6.93	9.11	4.01	0.98	5.65	2.75	2.00	0.24	3659	4979	3.15	42.21	82.90	Enr
RC27-9-6-5	Cpx2	Core1	0.02	0.05	0.25	0.30	0.16	1.12	0.64	0.73	0.36	1007	5579	0.83	11.61	1.08	Trans
RC27-9-6-5	Cpx2	Mdpt2									0.30	1070	6809	1.73	13.86	1.14	Trans
RC27-9-6-5	Cpx2	Mdpt3	0.06	0.13	0.20	0.36	0.17	1.43	0.84	0.76	0.39	1121	5698	1.28	13.07	1.17	Trans
RC27-9-6-5	Cpx2	Mdpt4	0.19	0.37	0.42	0.38	0.18	1.66	0.88	0.90	0.36	1270	7237	1.91	16.11	1.30	Trans
RC27-9-6-5	Cpx2	Mdpt5	0.88	3.74	2.35	1.27	0.45	2.59	1.67	1.63	0.39	1544	6811	2.40	26.81	13.06	Trans
RC27-9-6-5	Cpx2	Mdpt6	1.00	4.40	3.51	1.35	0.48	3.06	1.76	1.66	0.41	1873	6857	2.16	29.80	18.35	Trans
RC27-9-6-5	Cpx2	Rim7	1.00	5.21	5.37	2.20	0.62	4.57	2.32	1.98	0.31	2792	4956	1.92	31.24	45.18	Trans
RC27-9-6-5	Cpx3	Ints	1.00	7.00	9.36	4.96	0.99	6.44	2.78	2.43	1.19	7912	6282	9.01	58.17	150.43	Enr
RC27-9-6-5	Cpx4	Core	1.31	6.33	4.68	1.74	0.62	3.24	1.56	1.51	0.55	1471	5683	2.26	23.67	16.90	Enr
RC27-9-6-5	Cpx4	Rim	1.00	4.58	5.23	1.97	0.61	3.69	1.83	1.53	0.35	2646	5562	2.97	30.66	51.25	Enr
RC27-9-6-5	Cpx5	Core	1.45	6.11	3.30	0.96	0.51	2.35	1.41	1.49	0.77	1632	6555	3.24	25.19	21.34	Enr
RC27-9-6-5	Cpx5	Rim	1.14	5.69	5.93	2.65	0.65	4.16	2.32	2.12	0.50	2776	8399	2.38	37.76	53.11	Enr
RC27-9-6-7	Cpx1	Core	0.01	0.05	0.21	0.49	0.20	1.04	0.52	0.48	0.12	387	4695	1.31	3.37	0.72	
RC27-9-6-7	Cpx1	Rim	0.00	0.05	0.52	0.72	0.24	1.59	0.96	0.72	0.31	1235	4952	1.82	12.74	1.52	
RC27-9-6-7	Cpx3	Ints	0.01	0.04	0.22	0.52	0.22	1.65	0.73	0.83	0.29	1196	4220	1.39	11.13	1.26	
RC27-9-6-7	Cpx5	Ints	0.01	0.04	0.44	0.73	0.32	2.18	1.06	0.87	0.43	1515	4490	1.63	12.59	1.70	
Van7-78-25	Cpx1	Core	0.16	1.06	1.77	0.98	0.40	1.96	1.16	1.12	0.31	1599	5585	4.12	12.05	7.16	GbVn
Van7-78-25	Cpx1	Rim	0.16	1.14	1.87	1.01	0.44	2.13	1.42	1.37	0.35	2199	6357	4.87	16.85	12.08	GbVn
Van7-78-25	Cpx2	Core	0.13	1.03	1.92	1.04	0.37	1.97	1.22	1.00	0.21	1791	7086	6.03	17.38	8.88	GbVn
Van7-78-25	Cpx2	Rim	0.12	0.88	1.54	1.06	0.36	2.00	1.13	0.99	0.21	2563	7061	4.80	15.91	13.93	GbVn
Van7-78-25	Cpx3	Core	0.14	1.05	1.88	0.98	0.49	2.03	1.59	1.24	0.14	2618	9659	10.40	19.87	11.55	GbVn
Van7-78-25	Cpx3	Rim	0.14	0.97	1.88	1.05	0.36	2.33	1.31	1.22	0.18	2256	8181	5.15	14.57	10.54	GbVn
Van7-78-31	Cpx1	Ints	0.36	2.15	2.26	1.06	0.32	2.30	1.21	1.21	0.43	2700	6479	3.49	22.09	30.03	NrMat
Van7-78-31	Cpx2	Core	0.46	2.34	3.10	1.57	0.49	2.27	1.15	1.18	0.34	3885	686	18.22	21.57	19.63	GbVn
Van7-78-31	Cpx2	Rim	0.53	3.09	4.44	2.14	0.66	3.30	1.72	1.61	0.34	4690	754	14.73	26.16	25.38	GbVn

Table 5.10: Cpx trace elements, Cont.

Sample	Grain ^a	Type ^b	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^c
Van7-78-31	Cpx3	Rim	0.53	3.02	3.95	2.00	0.52	3.02	1.69	1.29	0.36	4366	837	16.66	25.25	24.44	GbVn
Van7-78-31	Cpx4	Ints	0.46	2.99	4.17	1.93	0.60	3.24	1.52	1.49	0.40	4460	847	17.46	23.82	22.13	GbVn
Van7-78-36	Cpx1	Core	0.55	3.08	3.31	1.48	0.59	2.44	1.53	1.41	0.49	3418	1056	11.49	20.80	19.12	GbVn
Van7-78-36	Cpx1	Rim	0.53	2.99	4.09	2.01	0.61	3.49	1.97	1.78	0.25	3388	1417	12.65	20.74	21.55	GbVn
Van7-78-36	Cpx3	Ints	0.80	4.06	5.75	2.46	0.68	3.84	2.25	2.06	0.31	3813	4763	11.79	26.90	39.85	AdjVn
Van7-78-36	Cpx3D	Ints	0.78	4.14	4.84	2.53	0.55	3.59	1.91	1.73	0.26	3777	1419	13.47	29.70	19.07	AdjVn
Van7-78-36	Cpx4	Ints	0.42	2.39	4.13	1.97	0.65	3.26	1.94	1.72	0.26	3777	1419	13.47	29.70	19.07	GbVn
Van7-78-36	Cpx4D	Ints	0.37	2.37	3.89	1.73	0.62	3.13	1.75	1.55	0.24	3412	1419	12.14	19.61	19.29	GbVn
Van7-78-36	Cpx10	Ints	0.81	3.74	4.23	1.86	0.46	3.21	1.66	1.73	0.36	2397	6696	2.73	25.15	33.03	FarMat
Van7-78-36	Cpx10D	Ints	0.71	3.57	3.99	1.88	0.44	3.54	2.00	1.66	0.33	2493	5438	2.50	24.37	33.89	FarMat
Van7-78-36	Cpx11	Ints	0.78	3.67	3.46	1.64	0.37	2.98	1.80	1.39	0.36	2122	4895	2.32	20.36	24.57	FarMat
Van7-78-36	Cpx12	Core	0.87	4.21	4.28	2.18	0.48	3.67	2.31	1.88	0.37	2757	6612	2.60	26.54	47.12	FarMat
Van7-78-36	Cpx12	Rim	0.60	3.13	3.44	1.52	0.36	2.61	1.63	1.44	0.29	2267	3880	2.48	22.03	35.18	FarMat
Van7-78-40	Cpx1	Core	1.49	5.33	2.77	0.95	0.33	1.82	1.11	1.08	0.35	1374	6095	3.96	15.88	8.95	
Van7-78-40	Cpx1D	Core	1.59	6.30	3.06	0.85	0.37	1.85	1.01	1.08	0.31	2554	4806	3.28	20.49	30.45	
Van7-78-40	Cpx1	Rim	1.12	4.90	4.19	1.65	0.50	3.11	1.48	1.38	0.31	2554	4806	3.28	20.49	30.45	
Van7-78-40	Cpx2	Ints	1.18	5.10	4.59	1.69	0.42	3.04	1.50	1.34	0.33	2556	6012	3.40	20.52	32.52	
Van7-78-40	Cpx3	Ints	1.06	4.87	3.96	1.73	0.37	3.01	1.45	1.25	0.28	2197	4050	1.97	18.51	19.58	
Van7-78-41	Cpx1	Core	0.37	1.55	1.18	0.76	0.25	2.40	1.22	1.48	0.28	1307	7225	1.11	20.85	6.79	
Van7-78-41	Cpx1D	Core	0.47	1.79	1.38	1.04	0.31	2.72	1.66	1.57	0.21	1159	3505	0.65	16.88	6.61	
Van7-78-41	Cpx1	Rim	0.24	0.99	0.92	0.60	0.16	1.83	1.04	0.90	0.21	1159	3505	0.65	16.88	6.61	
Van7-78-41	Cpx2	Core	0.25	1.02	1.03	0.57	0.22	1.98	1.01	1.13	0.25	1293	6406	1.07	16.05	6.47	
Van7-78-41	Cpx2	Rim	0.24	1.03	0.90	0.49	0.23	1.82	1.19	1.05	0.26	1192	4507	1.04	14.97	6.24	
Van7-85-05	Cpx1	Ints	1.98	8.57	9.57	3.13	1.20	4.67	2.78	2.01	0.36	7380	3913	25.86	27.80	52.09	
Van7-85-24	Cpx1	Core	0.00	0.01	0.27	0.37	0.22	1.80	1.17	1.03	0.35	1197	5728	1.73	14.99	1.02	
Van7-85-24	Cpx1	Rim	0.00	0.01	0.22	0.47	0.24	1.83	1.22	1.17	0.27	1110	4326	1.63	12.86	1.21	
Van7-85-24	Cpx2	Core	0.00	0.01	0.19	0.38	0.16	1.34	0.85	0.92	0.31	1012	5290	1.40	12.47	0.84	
Van7-85-24	Cpx2	Rim	0.01	0.01	0.23	0.44	0.17	1.73	1.06	0.95	0.31	1208	5047	1.55	14.24	1.03	
Van7-85-24	Cpx3	Core	0.01	0.01	0.27	0.45	0.17	1.49	1.04	0.94	0.31	1091	5528	1.26	12.03	0.88	
Van7-85-24	Cpx3	Rim	0.01	0.02	0.28	0.53	0.21	1.65	1.05	1.02	0.97	1302	5062	2.99	11.96	1.13	
Van7-85-24	Cpx4	Ints	0.00	0.01	0.33	0.52	0.25	1.74	1.17	1.08	0.34	1272	5959	1.12	13.96	0.97	
Van7-85-27	Cpx1	Core	0.01	0.06	0.34	0.44	0.19	1.52	0.86	0.85	0.47	1024	5549	0.94	9.48	1.27	
Van7-85-27	Cpx1	Rim	0.01	0.07	0.37	0.36	0.20	1.65	0.88	0.84	0.56	1120	5275	1.16	11.48	2.08	
Van7-85-27	Cpx2	Core	0.00	0.08	0.27	0.51	0.16	1.38	0.79	0.74	0.30	851	6215	0.91	9.26	1.10	
Van7-85-27	Cpx2	Rim	0.00	0.05	0.44	0.44	0.17	1.52	0.93	0.81	0.50	957	5608	1.33	10.18	1.34	
Van7-85-30	Cpx1	Core	0.04	0.03	0.36	0.32	0.18	1.39	0.97	0.74	0.22	898	5663	1.03	10.49	1.01	
Van7-85-30	Cpx1	Rim	0.02	0.03	0.31	0.34	0.22	1.38	0.91	0.81	0.32	1274	4910	0.94	14.19	1.60	
Van7-85-30	Cpx3	Ints	0.04	0.07	0.29	0.47	0.22	1.59	0.89	0.84	0.35	1514	6295	1.15	15.20	1.66	
Van7-85-30	Cpx4	Core	0.03	0.06	0.36	0.36	0.25	1.46	0.88	0.75	0.28	1366	8097	1.50	17.92	1.24	
Van7-85-30	Cpx4	Rim	0.03	0.08	0.36	0.48	0.27	1.45	0.85	0.87	0.27	1400	6452	1.35	16.95	1.42	
Van7-85-32	Cpx3	Core	0.01	0.03	0.23	0.40	0.13	1.20	0.76	0.58	0.22	918	4706	0.92	9.78	0.64	

Table 5.10: Cpx trace elements, Cont.

Sample	Grain ^a	Type ^b	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^c
Van7-85-32	Cpx3	Rim		0.02	0.13	0.37	0.19	1.19	0.78	0.51	0.25	1051	4809	0.60	11.66	0.61	
Van7-85-39	Cpx1	Core	0.01	0.02	0.32	0.50	0.27	1.49	0.93	1.05	0.43	1276	5391	1.30	12.56	1.78	
Van7-85-39	Cpx1D	Core	0.01	0.02	0.25	0.50	0.23	1.63	1.11	1.15	0.44	1195	5064	1.37	13.03	1.59	
Van7-85-39	Cpx1	Rim									0.45	1313	5039	0.95	12.89	1.48	
Van7-85-39	Cpx3	Core	0.01	0.02	0.39	0.47	0.16	1.59	1.00	1.15	0.38	1091	4668	1.14	12.99	0.95	
Van7-85-39	Cpx3	Rim		0.01	0.32	0.38	0.20	1.71	1.31	1.01	0.31	1078	4001	1.40	12.46	0.96	
Van7-85-39	Cpx4	Ints	0.08	0.02	0.30	0.43	0.27	1.67	0.80	0.76	0.41	1224	4247	1.75	11.62	1.19	
Van7-85-42	Cpx1	Core	0.03	0.06	0.28	0.50	0.23	1.44	1.05	0.96	0.30	1155	7085	1.44	13.91	1.94	
Van7-85-42	Cpx1	Rim	0.01	0.02	0.22	0.46	0.23	1.64	1.01	1.00	0.38	1616	7389	1.80	15.48	1.65	
Van7-85-42	Cpx2	Core	0.02	0.03	0.24	0.38	0.21	1.55	0.96	0.80	0.33	1003	4641	1.56	10.59	1.32	
Van7-85-42	Cpx2	Rim	0.02	0.04	0.25	0.42	0.20	1.67	1.11	0.90	0.27	1077	3888	0.77	10.35	1.37	
Van7-85-42	Cpx3	Ints	0.01	0.05	0.32	0.40	0.24	1.42	0.93	0.82	0.31	1251	6314	1.31	13.13	1.21	
Van7-85-47	Cpx2	Core	0.01	0.02	0.25	0.33	0.18	1.46	0.91	0.89	0.34	998	5182	0.61	12.53	0.56	
Van7-85-47	Cpx2	Rim	0.01	0.03	0.37	0.49	0.20	2.25	1.37	1.16	0.32	1053	4084	0.65	12.84	0.81	
Van7-85-47	Cpx3	Core	0.01	0.03	0.32	0.49	0.18	1.37	0.90	0.89	0.34	942	5793	0.51	11.64	0.64	
Van7-85-47	Cpx3	Rim	0.01	0.02	0.19	0.37	0.17	1.41	1.10	0.93	0.37	1051	4728	0.73	12.17	0.74	
Van7-85-49	Cpx1	Core	0.01	0.02	0.15	0.23	0.14	1.17	0.71	0.86	0.37	1119	5775	2.50	12.42	1.45	
Van7-85-49	Cpx1	Rim	0.01	0.02	0.20	0.20	0.13	1.15	0.68	0.75	0.38	951	5271	2.22	11.11	1.49	
Van7-85-49	Cpx3	Ints	0.02	0.06	0.36	0.63	0.30	1.66	1.04	0.88	0.42	1062	5227	2.44	11.86	1.90	
Van7-85-49	Cpx4	Core	0.01	0.03	0.31	0.44	0.22	1.45	0.80	0.80	0.37	892	5445	3.17	11.48	1.19	
Van7-85-49	Cpx4	Rim	0.01	0.03	0.36	0.72	0.25	1.47	0.87	0.75	0.35	982	4037	1.97	11.84	1.41	
Van7-86-25	Cpx1	Core	0.02	0.01	0.14	0.36	0.20	1.51	0.82	0.84	0.33	767	6545	2.47	10.00	1.37	
Van7-86-25	Cpx1	Rim	0.02	0.01	0.18	0.24	0.15	0.79	0.58	0.48	0.36	900	6101	2.29	9.84	3.18	
Van7-86-25	Cpx2	Ints		0.02	0.10	0.30	0.12	1.05	0.65	0.59	0.40	872	4932	2.37	8.76	1.17	
Van7-86-25	Cpx2D	Ints	0.01	0.02	0.11	0.35	0.12	1.28	0.86	0.75		811	4990	2.45	8.34	1.28	
Van7-86-25	Cpx3	Ints	0.03	0.03	0.18	0.35	0.15	1.00	0.63	0.54	0.34	811	4990	2.45	8.34	1.28	
Van7-86-25	Cpx4	Core	0.00	0.04	0.10	0.27	0.14	1.04	0.76	0.63	0.30	696	6037	3.33	8.36	1.08	
Van7-86-25	Cpx4	Rim	0.00	0.01	0.13	0.33	0.17	1.35	0.89	0.89	0.34	918	5803	2.81	9.87	1.42	
Van7-86-27	Cpx1	Core	0.01	0.01	0.21	0.29	0.15	1.34	0.83	1.02	0.16	745	4897	0.50	7.55	0.55	
Van7-86-27	Cpx1	Rim	0.00	0.02	0.22	0.35	0.17	1.33	0.82	0.74	0.21	1056	3495	0.80	8.88	0.45	
Van7-86-27	Cpx1D	Rim	0.01	0.02	0.21	0.37	0.15	1.42	0.98	0.86		984	4230	0.50	9.25	0.51	
Van7-86-27	Cpx2	Ints	0.03	0.02	0.20	0.28	0.17	1.57	0.89	0.79	0.35	984	4230	0.50	9.25	0.51	
Van7-86-27	Cpx2D	Ints	0.01	0.02	0.18	0.38	0.17	1.45	0.93	0.97		880	4780	0.46	8.38	0.42	
Van7-86-27	Cpx3	Core	0.00		0.17	0.37	0.15	1.46	0.82	0.80	0.24	880	4780	0.46	8.38	0.42	
Van7-86-27	Cpx3	Rim	0.00	0.01	0.10	0.37	0.13	1.43	0.92	0.90	0.12	729	2600	0.66	4.83	0.45	
Van7-86-28	Cpx1	Core	0.01	0.03	0.11	0.29	0.10	1.22	0.79	0.67	0.27	899	5018	2.77	8.30	0.91	
Van7-86-28	Cpx1	Rim	0.01	0.02	0.15	0.33	0.10	1.09	0.61	0.59	0.23	858	2485	2.46	8.25	1.25	
Van7-86-28	Cpx2	Ints	0.00	0.02	0.08	0.33	0.10	1.22	0.87	0.81	0.21	1158	4698	2.41	10.92	1.26	
Van7-86-28	Cpx3	Core	0.01	0.03	0.20	0.25	0.13	1.10	0.73	0.66	0.24	990	5315	3.00	9.31	1.36	
Van7-86-28	Cpx3	Rim	0.00	0.02	0.20	0.28	0.10	1.15	0.77	0.69	0.24	1061	5124	2.82	9.40	1.53	
Van7-86-28	Cpx4	Ints	0.00	0.03	0.17	0.19	0.13	1.10	0.67	0.65							

Table 5.10: Cpx trace elements, Cont.

Sample	Grain ^a	Type ^b	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^c
Van7-86-30	Cpx1	Core	0.00	0.01	0.13	0.20	0.10	0.97	0.57	0.49	0.26	887	5579	1.98	9.89	1.90	
Van7-86-30	Cpx1	Rim	0.01	0.01	0.26	0.25	0.15	1.29	0.89	0.71	0.31	898	4946	2.06	8.71	1.55	
Van7-86-30	Cpx2	Ints	0.01	0.01	0.21	0.39	0.12	1.13	0.88	0.68	0.28	1090	4923	1.82	10.22	1.66	
Van7-86-30	Cpx3	Ints	0.01	0.01	0.22	0.30	0.13	1.46	0.85	0.73	0.29	1147	4601	2.51	11.20	2.84	
Van7-86-37	Cpx1	Core	0.04	0.28	0.54	0.40	0.15	1.17	0.66	0.47	0.41	895	7107	5.72	8.58	3.76	
Van7-86-37	Cpx1	Rim	0.02	0.26	0.62	0.41	0.15	1.10	0.77	0.51	0.43	1109	6938	7.41	10.28	5.20	
Van7-86-37	Cpx2	Ints	0.04	0.30	0.56	0.46	0.14	1.30	0.75	0.55	0.55	1195	7206	8.78	10.19	5.15	
Van7-96-09	Cpx1	Core	0.21	1.55	3.16	1.31	0.60	2.91	2.05	1.83	0.39	2098	1919	6.31	24.02	14.43	PyxVn
Van7-96-09	Cpx1	Rim	0.19	1.15	2.47	1.33	0.56	2.31	1.76	1.50	0.38	2292	2030	5.72	22.39	14.37	PyxVn
Van7-96-09	Cpx2	Core	0.19	1.43	2.61	1.20	0.52	2.55	1.65	1.58	0.52	2212	1972	8.44	22.91	15.38	PyxVn
Van7-96-09	Cpx2	Rim	0.17	1.55	3.65	1.87	0.77	3.99	3.16	2.18	0.46	2759	2311	4.39	27.80	19.63	PyxVn
Van7-96-09	Cpx3	Core	0.18	1.55	2.96	1.10	0.58	2.13	1.33	1.09	0.47	1957	4322	10.78	13.65	16.21	PyxVn
Van7-96-09	Cpx3	Rim	0.20	1.41	3.17	1.40	0.62	2.91	1.69	1.40	0.52	2511	4570	5.56	17.22	19.75	PyxVn
Van7-96-09	Cpx4	Ints	0.26	1.67	3.96	1.77	0.75	3.62	2.30	2.17	0.44	3931	2988	7.29	35.52	30.78	PyxVn
Van7-96-09	Cpx5	Ints	0.22	1.46	3.75	1.98	0.83	3.58	2.06	1.89	0.46	2822	2809	5.82	27.26	20.40	PyxVn
Van7-96-14	Cpx1	Core	0.07	0.75	3.92	1.94	0.60	4.06	2.73	2.11	0.45	3881	1948	2.01	31.53	21.41	PyxVn
Van7-96-14	Cpx1	Rim	0.04	0.87	3.42	2.23	0.65	4.51	2.64	2.13	0.37	3876	2326	3.76	27.11	18.79	PyxVn
Van7-96-14	Cpx2	Core	0.08	0.86	3.12	1.57	0.63	3.53	2.53	2.43	0.47	3040	1259	3.06	32.49	18.71	PyxVn
Van7-96-14	Cpx2	Rim	0.08	1.17	3.99	2.04	0.76	4.74	3.24	2.72	0.41	4461	1626	1.93	39.44	25.88	PyxVn
Van7-96-14	Cpx5	Core	0.09	1.21	3.56	2.25	0.70	5.29	4.09	3.92	0.35	3245	2016	1.84	42.68	25.77	PyxVn
Van7-96-14	Cpx5	Rim	0.07	1.34	5.20	3.39	0.58	7.97	5.32	4.43	0.54	2026	1731	3.87	16.08	15.06	PyxVn
Van7-96-14	Cpx6	Core	0.05	1.04	3.01	1.49	0.69	3.09	2.33	2.30	0.34	2655	1451	2.88	27.44	18.19	PyxVn
Van7-96-14	Cpx6	Rim	0.08	1.41	4.75	2.48	0.88	5.53	4.10	2.90	0.39	4578	1704	1.96	40.21	31.85	PyxVn
Van7-96-14	Cpx7	Core	0.10	1.04	3.55	1.76	0.83	3.57	2.48	2.41	0.36	2584	1590	2.90	27.33	19.85	PyxVn
Van7-96-14	Cpx7	Rim	0.08	1.20	5.32	2.67	0.76	5.37	3.42	2.84	0.54	3749	1741	2.72	45.21	33.79	PyxVn
Van7-96-15	Cpx1	Ints	0.12	1.73	7.08	4.36	0.81	8.07	4.09	4.08	0.53	7894	5528	4.14	79.65	77.00	PyxVn
Van7-96-15	Cpx2	Ints	0.10	1.70	5.28	2.71	0.74	5.51	3.30	2.60	0.46	5317	3640	3.72	62.35	46.14	PyxVn
Van7-96-15	Cpx3	Core	0.05	1.10	3.04	1.59	0.63	2.92	1.97	1.96	0.43	3413	2011	5.65	50.39	30.74	PyxVn
Van7-96-15	Cpx3	Rim	0.09	1.65	4.21	2.42	0.84	4.62	2.71	2.56	0.42	3847	2174	3.98	50.42	35.74	PyxVn
Van7-96-15	Cpx4	Core	0.10	1.03	2.78	1.45	0.51	3.19	2.23	1.88	0.28	1731	1018	7.74	19.08	11.59	PyxVn
Van7-96-15	Cpx4	Rim	0.08	1.18	3.39	2.15	0.63	4.31	2.62	1.95	0.38	4254	2365	4.05	44.05	31.68	PyxVn
Van7-96-15	Cpx4D	Rim									0.41	3948	1928	4.19	41.14	31.03	PyxVn
Van7-96-16	Cpx1	Core	0.08	1.02	2.90	1.35	0.55	2.77	1.85	1.56	0.50	3281	3427	7.39	37.03	28.17	PyxVn
Van7-96-16	Cpx1	Rim	0.07	1.58	5.52	2.98	0.74	5.97	3.49	2.88	0.46	4351	3904	3.81	53.97	45.49	PyxVn
Van7-96-16	Cpx2	Ints	0.10	1.53	6.12	2.95	0.71	5.52	3.36	2.46	0.43	4357	3833	2.83	55.74	47.82	PyxVn
Van7-96-16	Cpx3	Ints	0.11	1.94	6.24	3.20	0.96	5.74	4.00	2.92	0.55	4077	2458	6.02	52.16	38.18	PyxVn
Van7-96-16	Cpx4	Core	0.10	1.48	3.73	1.75	0.61	3.42	2.18	1.86	0.42	3437	2261	10.06	38.25	27.10	PyxVn
Van7-96-16	Cpx4	Rim	0.11	1.72	4.08	2.11	0.57	3.89	2.50	1.59	0.32	5045	3015	4.51	53.17	45.16	PyxVn
Van7-96-18	Cpx1	Core	0.22	2.11	3.60	1.61	0.59	2.40	1.55	1.41	1.11	3078	4985	16.28	28.89	24.08	NrMat
Van7-96-18	Cpx1	Rim	0.26	2.11	2.26	1.40	0.69	3.25	2.01	1.91	0.99	3047	4629	14.56	24.49	21.55	NrMat
Van7-96-18	Cpx2	Ints	0.23	1.78	2.69	1.52	0.64	2.78	1.77	1.65	1.02	3342	4104	11.27	23.81	22.06	NrMat

Table 5.10: Cpx trace elements, Cont.

Sample	Grain ^a	Type ^b	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^c
Van7-96-18	Cpx3	Ints	0.22	1.69	2.65	1.19	0.49	2.60	1.66	1.53	0.87	2312	3659	8.68	17.80	15.80	NrMat
Van7-96-18	Cpx3D	Ints									0.86	2923	3380	9.48	21.17	20.97	NrMat
Van7-96-18	Cpx4	Ints	0.21	1.66	2.95	1.35	0.56	3.06	1.76	1.48	1.11	2980	4186	12.16	22.90	19.66	NrMat
Van7-96-18	Cpx6	Core	0.13	1.39	2.26	1.31	0.42	2.94	2.22	1.97	0.50	1977	2047	4.82	20.95	11.19	PyxVn
Van7-96-18	Cpx6	Rim	0.14	1.30	2.56	1.36	0.53	3.75	2.46	2.41	0.60	2975	2989	3.66	28.58	20.29	PyxVn
Van7-96-18	Cpx7	Core	0.18	1.55	3.03	1.59	0.65	3.65	2.81	2.49	0.50	2272	2336	3.75	22.64	17.16	PyxVn
Van7-96-18	Cpx7	Rim	0.16	1.98	3.55	2.06	0.70	5.07	2.77	2.78	0.54	3367	2978	2.41	28.46	23.99	PyxVn
Van7-96-18	Cpx8	Ints	0.14	1.25	2.39	1.26	0.51	3.67	2.20	1.90	0.47	2910	3704	2.39	29.62	21.86	PyxVn
Van7-96-19	Cpx1	Core	0.22	2.18	3.79	2.12	0.59	2.83	1.49	1.53	0.54	2905	5532	10.54	17.28	26.75	NrMat
Van7-96-19	Cpx1	Rim	0.19	1.68	4.09	1.76	0.70	3.78	2.07	1.74	0.58	3941	6070	6.74	22.70	38.56	NrMat
Van7-96-19	Cpx2	Core	0.23	1.35	2.85	1.25	0.60	3.47	1.79	1.66	0.73	3493	4998	9.33	25.89	33.00	NrMat
Van7-96-19	Cpx2	Rim	0.33	3.15	6.12	3.32	0.86	4.75	1.92	2.23	0.49	4479	5371	5.57	31.13	43.86	NrMat
Van7-96-19	Cpx3	Ints	0.22	2.10	4.27	2.19	0.81	3.97	1.83	1.81	0.51	3716	5721	6.20	18.12	56.53	NrMat
Van7-96-19	Cpx4	Ints	0.21	2.54	4.90	2.66	0.67	3.76	2.05	1.86	0.47	3588	5837	4.63	23.80	37.75	NrMat
Van7-96-19	Cpx5	Core	0.62	4.13	5.62	3.06	0.86	3.72	2.20	1.92	0.47	3989	5974	5.73	33.23	49.67	PyxVn
Van7-96-19	Cpx5	Rim	0.58	4.99	7.45	3.02	0.94	5.17	2.92	2.30	0.51	4526	6200	5.41	34.21	55.80	PyxVn
Van7-96-19	Cpx6	Ints	0.41	3.33	4.71	2.19	0.81	3.38	1.73	1.54	0.46	2997	6183	7.04	19.49	32.35	PyxVn
Van7-96-19	Cpx8	Core	0.46	2.89	3.82	1.81	0.68	2.61	1.40	1.51	0.46	2774	5648	14.56	21.44	31.64	PyxVn
Van7-96-19	Cpx8D	Core	0.57	4.08	5.57	2.06	0.91	3.45	1.80	1.72							PyxVn
Van7-96-19	Cpx8	Rim	0.53	4.17	5.12	2.35	0.78	3.33	1.69	1.50	0.53	3380	6941	6.27	21.08	35.55	PyxVn
Van7-96-19	Cpx9	Rim	0.52	4.27	5.71	2.16	0.81	3.60	1.44	1.56	0.56	3489	7218	15.83	16.35	38.33	AdjVn
Van7-96-21	Cpx1	Core	0.92	4.60	4.64	1.65	0.57	1.96	1.15	1.07	1.00	3204	4756	47.27	17.32	33.22	PyxVn
Van7-96-21	Cpx1D	Core	1.36	3.42	4.87	1.96	0.84	2.79	1.45	1.31							PyxVn
Van7-96-21	Cpx1	Rim	1.32	8.19	8.76	3.14	1.14	3.54	2.14	2.02	0.60	3195	4319	23.71	19.61	33.01	PyxVn
Van7-96-21	Cpx4	Core	0.84	4.45	5.46	2.03	0.74	2.27	1.37	1.20	0.48	2500	4455	27.98	12.76	23.09	AdjVn
Van7-96-21	Cpx4	Rim	1.29	7.74	10.33	3.29	1.18	3.94	2.58	1.89	0.63	4884	4597	10.40	27.49	55.82	AdjVn
Van7-96-21	Cpx5	Core	1.01	5.09	5.34	1.75	0.73	2.21	1.40	1.09	0.53	2504	4460	24.44	14.19	25.32	AdjVn
Van7-96-21	Cpx5	Rim	1.26	6.99	6.48	2.14	0.89	2.92	1.66	1.52	0.68	3534	4424	37.47	23.09	38.24	AdjVn
Van7-96-21	Cpx6	Ints	1.10	4.58	5.31	1.81	0.69	2.20	1.16	1.08	0.47	3780	4264	8.89	21.50	43.21	PyxVn
Van7-96-21	Cpx7	Core	1.04	6.45	7.41	2.68	0.84	3.28	1.68	1.62	0.50	3122	4555	18.48	21.21	33.64	PyxVn
Van7-96-21	Cpx7	Rim	1.01	7.07	10.19	3.98	1.15	5.11	2.95	2.39	0.81	6902	4580	17.96	35.84	85.62	PyxVn
Van7-96-21	Cpx7D	Rim	1.06	8.23	11.90	4.20	1.16	5.93	2.97	2.67							PyxVn
Van7-96-21	Cpx8	Ints	1.15	5.50	5.37	2.48	0.72	2.79	1.47	1.50							PyxVn
Van7-96-25	Cpx1	Core	0.15	1.22	1.79	0.95	0.48	1.94	1.05	1.20	1.18	2272	5938	13.29	18.38	15.35	
Van7-96-25	Cpx1	Rim	0.17	1.70	2.43	1.25	0.51	2.21	1.49	1.24	1.10	2692	5397	13.06	19.38	19.36	
Van7-96-25	Cpx2	Ints	0.13	1.54	2.26	1.11	0.56	2.03	1.45	1.25	1.28	2463	4974	13.83	17.19	17.21	
Van7-96-25	Cpx3	Ints	0.18	1.68	2.28	1.38	0.50	2.13	1.44	1.22	1.24	2393	5464	12.87	17.19	16.10	
Van7-96-25	Cpx4	Ints	0.20	1.28	2.29	1.12	0.51	2.05	1.18	1.13	0.93	2576	5538	9.76	14.38	16.41	
Van7-96-25	Cpx5	Core	0.19	1.71	2.56	1.24	0.62	2.14	1.33	1.33	1.08	2693	5919	13.39	18.43	18.09	
Van7-96-25	Cpx5	Rim	0.17	1.45	2.71	1.19	0.47	2.29	1.27	1.20	1.10	2680	5785	11.67	17.34	17.70	
Van7-96-28	Cpx1	Core	1.32	4.94	3.66	1.31	0.51	1.60	0.86	0.76	1.79	2805	6171	112.95	12.32	19.69	

Table 5.10: Cpx trace elements, Cont.

Sample	Grain ^a	Type ^b	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^c
Van7-96-28	Cpx1	Rim	0.85	4.53	3.24	1.28	0.48	1.46	0.78	0.74	1.68	2308	5863	108.53	11.51	16.39	
Van7-96-28	Cpx2	Core	1.48	5.71	3.66	1.49	0.47	1.57	0.83	0.73							
Van7-96-28	Cpx3	Core	0.97	3.99	3.15	1.08	0.45	1.31	0.57	0.70	0.75	1812	6802	65.88	7.98	14.69	
Van7-96-28	Cpx3	Rim	0.98	4.64	3.74	1.42	0.47	1.39	0.84	0.70	0.86	2769	7288	55.97	14.61	22.52	
Van7-96-28	Cpx4	Ints	1.10	5.46	4.04	1.32	0.65	1.76	0.72	0.84	0.66	2416	5887	38.54	13.63	22.25	
Van7-96-28	Cpx5	Core	0.95	3.68	3.80	1.24	0.53	1.62	0.81	0.99	0.92	1785	6960	70.83	8.89	14.12	
Van7-96-28	Cpx5	Rim	1.32	5.48	4.76	1.57	0.54	1.99	1.06	1.15							
Van7-96-28	Cpx5D	Rim	1.10	4.24	3.75	1.21	0.62	1.74	0.97	0.91	1.13	2129	6734	89.19	9.82	16.90	
Van7-96-28	Cpx6	Ints	1.05	4.03	2.97	1.03	0.39	1.23	0.68	0.67	1.43	1584	8281	42.31	11.75	18.43	
Van7-96-35	Cpx1	Core									1.33	1604	7836	41.40	11.26	19.28	
Van7-96-35	Cpx1	Rim	0.20	1.43	2.61	1.34	0.49	1.46	0.72	0.76	1.67	1841	8344	34.33	12.55	22.17	
Van7-96-35	Cpx2	Ints									1.14	1915	7862	36.54	13.77	23.11	
Van7-96-35	Cpx3	Core	0.40	2.44	3.35	1.29	0.48	1.62	0.95	1.04	1.14	1915	7862	36.54	13.77	23.11	
Van7-96-35	Cpx3	Rim	0.38	2.28	2.36	1.23	0.49	1.89	0.80	0.98	1.24	2047	7892	33.12	13.84	26.06	
Van7-96-35	Cpx4	Ints	0.32	2.09	2.76	1.37	0.55	1.90	0.83	1.04	1.17	1903	7800	32.39	12.55	23.25	
Van7-96-35	Cpx5	Ints	0.38	2.18	2.97	1.62	0.55	1.56	0.96	0.85	1.21	1822	8156	39.99	12.79	23.16	
Van7-96-35	Cpx6	Core	0.49	2.60	3.24	1.68	0.56	1.93	0.72	0.96	1.16	1506	8777	35.62	12.10	17.46	
Van7-96-35	Cpx6	Rim	0.39	2.62	3.46	1.69	0.62	2.16	1.13	1.35	1.31	1958	7801	36.44	13.30	22.63	
Van7-96-37	Cpx1	Core	0.88	3.84	3.51	1.36	0.53	1.59	0.82	0.79	1.20	1915	8046	99.97	10.74	18.50	
Van7-96-37	Cpx1	Rim	1.21	4.41	3.55	1.21	0.62	1.97	1.19	0.98	1.08	1951	7398	79.00	10.72	19.68	
Van7-96-37	Cpx2	Ints	1.23	4.68	3.37	1.52	0.67	1.67	0.91	1.05	1.07	2045	8006	85.14	10.84	20.80	
Van7-96-37	Cpx4	Ints	1.00	3.73	3.38	1.32	0.60	1.61	0.65	0.93	1.30	2183	7616	97.75	10.37	20.05	
Van7-96-37	Cpx5	Core	1.06	3.97	3.21	1.17	0.50	1.50	0.76	0.86	1.53	1776	8350	108.54	10.94	18.63	
Van7-96-37	Cpx5	Rim	1.01	4.30	3.31	1.41	0.55	1.71	0.80	0.97	1.10	1748	7605	89.47	9.94	17.67	
Van7-96-38	Cpx1	Core	0.09	1.01	2.56	1.34	0.53	2.12	1.34	1.34	1.08	2372	4230	13.52	17.93	14.32	
Van7-96-38	Cpx1	Rim	0.10	1.16	2.56	1.34	0.60	2.57	1.82	1.35	1.09	2775	3961	13.21	18.84	16.40	
Van7-96-38	Cpx2	Core	0.11	0.74	3.21	1.74	0.82	3.44	2.44	1.49	1.01	2663	3989	13.17	20.10	15.57	
Van7-96-38	Cpx2	Rim	0.09	1.20	3.31	1.72	0.80	3.13	2.01	1.90	0.86	2985	3613	13.29	20.72	18.62	
Van7-96-66B	Cpx1	Ints	0.22	2.20	3.45	1.41	0.61	2.67	1.06	1.05	0.53	4853	3370	4.84	18.08	37.13	Opx Vn
Van7-96-66B	Cpx2	Ints	0.21	2.03	2.80	1.18	0.51	2.35	1.06	0.84	0.61	4292	3310	3.96	17.85	29.83	Opx Vn
Van7-96-66B	Cpx3	Ints	0.41	2.97	4.64	2.47	0.77	3.49	1.57	1.51	0.56	4944	3765	7.19	23.30	39.50	Opx Vn
Van7-96-66B	Cpx5	Core	0.22	0.68	1.13	0.39	0.22	1.08	0.53	0.63	0.68	1074	575	19.64	8.31	4.45	Breccia
Van7-96-66B	Cpx5	Rim	0.10	0.46	0.59	0.32	0.15	0.81	0.37	0.38	0.52	1045	546	12.09	7.72	4.22	Breccia

^a D following a grain number indicates duplicate analysis; L preceding a grain number indicates reprocessed data for sample RC27-9-6-2, from Lee (1997).^b Ints refers to interstitial grain, < 1 mm diameter; Symp refers to symplectite of Cpx with Sp \pm Opx; Numbered analyses and Mdpt (midpoint between core and rim) refer to multiple analyses of grains in cryptically metasomatized samples.^c Abbreviations for lithologies: Dep. Trans and Enr are depleted, transitional and enriched zones in cryptically metasomatized samples; GbVn refers to a gabbroic vein, PyxVn to a pyroxenite vein, CpxVn to a clinopyroxenite vein, and OpxVn to an orthopyroxenite vein; AdjVn is adjacent to a vein. NrMat is < 5 cm from a vein and FarMat is > 5 cm from a vein.^d Cpx from Kilbourne Hole (Irving and Frey, 1984), used as a standard for REEs.^e Kilauela tholeiitic basalt glass (Jochum et al., 2000), used as a standard for the other trace elements.^f Percent error based on average counting error among all analyses. ^g Percent error based on average difference among duplicate analyses.

Table 5.1.1: Trace element concentrations in plagioclase by ion probe, in ppm.

Sample	Grain	Type ^a	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^b
Count % Error	Plag		8	5	10	23	7	28	22	31	0	3	12	3	14	21	
6K-465-2	Plag1	Ints									1.33	13033		57.8	0.95	2.18	Dep
6K-465-2	Plag2	Ints									0.24	2419		38.6	0.35	1.36	Dep
6K-465-2	Plag3	Ints									0.94	52		2.5	0.33	0.41	Dep
6K-465-2	Plag4	Ints									1.02	52		1.4	0.18	0.33	Dep
6K-465-2	Plag5	Ints									1.02	60		31.8	0.29	0.37	Dep
6K-465-2	Plag6	Ints									0.96	52		1.3	0.34	0.46	Dep
6K-465-3	Plag1a	Ints									7.60	49	11.70	17.3	5.15	8.31	NrMat
6K-465-3	Plag1b	Ints	3.29	8.53	2.74	0.65	0.56	0.36	0.43	0.09	7.44	95	6.84	22.3	6.48	9.72	NrMat
6K-465-3	Plag3	Ints	3.18	7.43	2.13	0.52	2.28	0.17	0.36	0.09	4.34	245	4.11	325.5	1.26	0.71	GbVn
6K-465-3	Plag4a	Ints	11.65	27.07	8.26	1.39	2.46	0.47	0.43	0.18	4.07	250	1.61	50.5	1.23	0.25	NrMat
6K-465-3	Plag4b	Ints	10.69	24.47	8.26	1.38	2.33	0.41	0.47	0.21	3.49	247	1.51	54.1	1.40	0.36	NrMat
RC27-9-6-5	Plag1	Ints	1.23	1.73	0.27	0.02	0.23	0.01	0.08	0.02	1.59	36	2.58	6.5	0.06	0.12	Dep
Van7-78-31	Plag1	Ints									2.32	87	1.49	81.5	0.48	0.10	NrMat
Van7-78-31	Plag2	Ints	0.52	1.15	0.38	0.09	0.27	0.03	0.04	0.01	2.82	174	1.82	371.1	0.49	0.30	GbVn
Van7-78-36	Plag11	Ints	0.98	1.56	0.47	0.08	0.23	0.04	0.03	0.01	1.51	69	2.79	47.4	0.21	0.33	FarMat
Van7-78-36	Plag1	Core	0.42	0.93	0.30	0.05	0.26	0.03	0.04	0.01	3.06	184	2.97	260.7	0.31	0.36	GbVn
Van7-78-36	Plag2	Ints	0.38	0.85	0.32	0.08	0.28	0.02	0.04	0.01	2.67	205	0.49	297.9	0.39	0.64	GbVn
Van7-78-36	Plag2	IntsR									2.94	201	0.91	286.9	0.21	0.11	GbVn
Van7-78-41	Plag1	Ints									0.85	29	1.45	9.9	0.20	0.62	
Van7-96-09	Plag1	Ints	0.17	0.52	0.38	0.15	0.34	0.07	0.07	0.06	2.58	121	2.45	58.4	0.37	0.42	PyxVn
Van7-96-09	Plag2	Ints	0.19	0.55	0.34	0.08	0.17	0.01	0.03	0.01	0.88	62	1.20	70.9	0.28	0.49	PyxVn
Van7-96-14	Plag1	Ints	0.08	0.30	0.45	0.16	0.34	0.25	0.10	0.07	1.83	121	1.58	31.0	0.25	0.41	PyxVn
Van7-96-14	Plag3	Core	0.06	0.38	0.59	0.15	0.41	0.09	0.08	0.04	1.64	162	1.05	22.1	0.56	0.36	PyxVn
Van7-96-14	Plag3	Rim	0.05	0.38	0.52	0.27	0.38	0.10	0.08	0.12	1.33	112	1.40	146.3	0.46	0.50	PyxVn
Van7-96-14	Plag4	Core	0.11	0.61	0.55	0.21	0.38	0.17	0.11	0.13	2.01	244	1.33	22.2	0.56	0.30	PyxVn
Van7-96-14	Plag4	Rim	0.07	0.49	0.44	0.21	0.31	0.14	0.16	0.04	1.74	207	1.53	15.7	0.57	0.30	PyxVn
Van7-96-14	Plag7	Ints	0.06	0.36	0.52	0.14	0.38	0.07	0.08	0.03							PyxVn
Van7-96-15	Plag2	Ints	0.08	0.57	0.54	0.09	0.41	0.05	0.09	0.04	2.19	215	1.66	33.0	0.75	0.39	PyxVn
Van7-96-15	Plag3	Ints	0.07	0.36	0.31	0.09	0.27	0.05	0.10	0.03	1.51	99	1.77	24.5	0.53	0.36	PyxVn
Van7-96-15	Plag4	Ints	0.06	0.39	0.67	0.27	0.26	0.44	0.29	0.14							PyxVn
Van7-96-16	Plag1	Ints	0.04	0.37	0.63	0.27	0.24	0.07	0.08	0.15	1.71	128	1.49	150.9	0.65	0.29	PyxVn
Van7-96-16	Plag3	Ints	0.07	0.53	0.48	0.06	0.23	0.04	0.06	0.02	0.95	59	1.15	36.7	0.24	0.22	PyxVn
Van7-96-18	Plag1	Ints	0.12	0.41	0.35	0.11	0.21	0.08	0.08	0.02	2.36	170	6.39	30.9	0.45	0.88	PyxVn
Van7-96-19	Plag1	Ints	0.12	0.42	0.36	0.13	0.35	0.13	0.10	0.03	2.63	173	7.58	29.9	0.33	0.64	NrMat
Van7-96-19	Plag3	Ints	0.33	1.03	0.31	0.13	0.42	0.02	0.07	0.04	2.69	141	4.54	145.6	0.25	0.48	PyxVn
Van7-96-21	Plag1	Ints	0.55	1.68	0.91	0.13	0.31	0.14	0.05	0.09	2.53	194	0.93	106.2	0.49	0.27	PyxVn
Van7-96-21	Plag2	Ints	0.58	1.66	0.86	0.13	0.28	0.04	0.29	0.06	2.36	188	1.06	84.4	0.45	0.48	PyxVn
Van7-96-21	Plag3	Ints	0.69	1.75	0.78	0.20	0.32	0.04	0.07	0.02	1.36	176	0.77	75.6	0.25	0.21	PyxVn

Table 5.1.1: Plagioclase trace elements Cont.

Sample	Grain	Type ^a	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^b
Van7-96-21	Plag4	Ints	0.46	1.61	0.92	0.26	0.22	0.10	0.07	0.04							PyxVn
Van7-96-21	Plag5	Ints	0.47	1.81	1.00	0.22	0.26	0.07	0.06	0.01							PyxVn
Van7-96-21	Plag6	Symp	0.30	1.24	0.63	0.20	0.26	0.02	0.16	0.08							PyxVn

^a Ints is an interstitial grain; Symp is a symplectite with Opx; R is a duplicate analysis.

^b Dep refers to the depleted zone of a cryptically metasomatized sample; GbVn to a gabbroic vein; PyxVn to a pyroxenite vein; NrMat is <5 cm from a vein and FarMat is >5 cm from a vein.

Table 5.12: Trace element concentrations in Opx by ion probe, in ppm.

Sample	Grain ^a	Type ^b	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^c
Count % Error	Opx		34	22	30	32	26	14	16	11	1	1	0	11	8	15	
6K-465-3	Opx1	Core1	1.512	6.921	4.051	1.169	0.399	2.504	1.503	1.760							NrMat
6K-465-3	Opx1	Core2	2.132	9.378	5.734	2.560	0.532	3.844	2.307	2.673							NrMat
6K-465-3	Opx1	Core3	1.640	6.970	3.689	1.622	0.487	2.845	1.852	1.991							NrMat
6K-466-1	Opx2	Ints	0.020	0.051	0.081	0.126	0.021	0.115	0.125	0.122							
6K-649-5	Opx3	Symp	0.012	0.024	0.042	0.057	0.026	0.104	0.119	0.209							
JR31-52-04	Opx2	Ints	0.021	0.037	0.141	0.113	0.048	0.177	0.106	0.187	0.014	304	4485	0.44	0.77	0.13	
Kn162-47-9	Opx1	Core	0.005	0.054	0.060	0.038	0.011	0.041	0.035	0.041							
Kn162-47-9	Opx4	Core	0.009	0.034	0.026	0.020	0.009	0.037	0.021	0.029							
RC27-9-6-2	OpxL7	Core	0.001	0.003	0.006	0.004	0.003	0.038	0.027	0.047	0.014	295	4283	0.48	0.26	0.46	AdjVn
Van7-78-25	Opx3	Ints	0.023	0.105	0.045	0.017	0.012	0.155	0.158	0.137	0.042	708	2432	1.27	1.50	0.72	GbVn
Van7-78-36	Opx3	Ints									0.008	1123	1494	0.73	0.93	1.06	NrMat
Van7-85-27	Opx2	Core									0.029	319	3034	0.33	0.91	0.19	
Van7-85-27	Opx3	Ints	0.002	0.007	0.021		0.007	0.131	0.134	0.157	0.027	358	3085	0.33	1.02	0.29	
Van7-86-37	Opx3	Core		0.009	0.017	0.036	0.013	0.143	0.102	0.163	0.051	417	4016	1.84	1.27	1.07	
Van7-86-37	Opx3	Rim	0.001	0.009	0.014	0.010		0.103	0.087	0.112	0.035	375	3389	1.70	0.82	1.10	
Van7-96-18	Opx3	ExLm	0.024	0.034	0.087	0.087	0.035	0.149	0.159	0.291							NrMat
Van7-96-19	Opx2	Ints	0.018	0.035	0.111	0.099	0.040	0.212	0.190	0.267							NrMat
Van7-96-21	Opx3	Ints	0.000	0.009	0.016	0.013	0.004	0.150	0.196	0.194	0.013	979	1946	0.44	1.73	3.98	PyxVn
Van7-96-21	Opx4	Rim	0.005	0.007	0.024	0.058	0.015	0.206	0.140	0.171							PyxVn
Van7-96-28	Opx5	Ints	0.120	0.028	0.108	0.045	0.014	0.098	0.069	0.109	0.075	711	2169	1.38	0.81	1.41	

^a L preceding a grain number indicates reprocessed data for sample RC27-9-6-2, from Lee (1997).

^b Ints refers to interstitial grain, <1 mm diameter; Symp refers to symplectite of Cpx with Sp and Opx; ExLm refers to exsolution lamellae in Cpx.

^c Dep, Trans and Enr are depleted, transitional and enriched zones in cryptically metasomatized samples; GbVn refers to a gabbroic vein, PyxVn to a pyroxenite vein, CpxVn to a clinopyroxenite vein, and OpxVn to an orthopyroxenite vein; AdjVn is adjacent to a vein, NrMat is <5 cm from a vein and FarMat is >5 cm from vein.

Table 5.13: Ion probe trace element concentrations in alteration phases, in ppm.

Sample	Grain ^a	Type	La	Ce	Nd	Sm	Eu	Dy	Er	Yb	Na ₂ O	Ti	Cr	Sr	Y	Zr	Notes ^b
6K-465-3	Actin2	Core	1.18	9.80	19.69	10.56	2.13	17.30	8.79	8.17	0.58	5024	214	10.88	371.31	311.60	GbVn
6K-465-3	Actin2	Rim	2.73	20.13	40.84	23.06	4.33	39.15	20.49	18.47	0.60	2084	222	8.83	313.77	177.35	GbVn
6K-465-3	Actin3	Core	2.28	19.01	35.00	20.24	4.50	33.41	17.84	16.71	0.34	1261	193	6.53	160.52	80.15	GbVn
6K-465-3	Actin3	Rim	2.10	17.44	31.42	16.48	3.76	29.78	15.43	14.81	0.76	2496	196	9.98	369.62	235.58	GbVn
6K-465-3	Actin4	Core	1.12	9.85	18.89	11.09	1.98	19.26	9.45	9.56	0.30	1588	204	6.14	155.43	82.48	GbVn
6K-465-3	Actin4	Rim	2.06	14.90	28.78	16.44	3.21	29.35	14.70	14.82	0.79	2759	208	11.01	373.55	262.25	GbVn
6K-465-3	Actin5	Core	5.01	38.69	69.99	35.72	7.94	52.37	24.78	26.41	0.75	1532	135	6.21	326.75	298.62	GbVn
6K-465-3	Actin8	Core	4.79	23.94	21.79	7.34	2.73	8.45	5.05	5.38	0.85	751	2350	4.53	39.02	49.69	NrMat
JR31-52-04	Amph5	Core	0.27	0.26	0.49	0.77	0.20	1.40	1.05	0.96	0.54	921	7545	2.20	4.46	0.24	
JR31-52-04	Amph5	Rim	0.08	0.08	0.37	0.39	0.12	0.93	0.59	0.61							
Van7-85-32	Amph1	Rim	0.18	0.40	0.24	0.25	0.14	1.30	0.78	0.84	1.78	1235	6063	8.32	11.01	1.28	
Van7-85-32	Amph2	Core	0.62	1.79	0.72	0.43	0.23	1.67	1.15	1.00	1.36	1004	5179	8.40	12.66	0.98	
Van7-85-32	Amph2	Rim	0.31	0.88	0.34	0.34	0.14	1.52	1.01	0.88	1.41	1251	4916	8.01	8.34	0.91	
RC27-9-6-2	SerpL1		0.01	0.05	0.12	0.08	0.02	0.15	0.08	0.09							InCpxL6
RC27-9-6-2	SerpL2		0.01	0.03	0.10	0.05	0.02	0.12	0.06	0.07							InCpxL6
RC27-9-6-2	SerpL3		0.001	0.005	0.013	0.013	0.001	0.010	0.007	0.010							AfterOliv
RC27-9-6-2	SerpL4		0.002	0.004	0.007	0.005	0.003	0.007	0.013	0.006							AfterOliv
RC27-9-6-2	SerpL5		0.003	0.007	0.011	0.012	0.005	0.010	0.011	0.009							AfterOpx

^a Actinolite (Actin), undifferentiated amphibole (Amph), and serpentine (Serp); L preceding a grain number indicates reprocessed data from Lee (1997).

^b GbVn is a gabbroic vein; NrMat is <5 cm from a vein.

Table 5.14: Percent deviation among repeat measurements of the KH1 standard

Date (ymd)	n ^a	La	Ce	Nd	Sm	Eu	Dy	Er	Yb
2001/08/21	2	6	8	12	3	5	11	3	14
2001/08/30	2	2	1	3	2	2	1	3	0
2002/01/15	2	6	2	2	7	4	0	18	4
2002/01/22	2	3	3	2	9	4	5	4	3
2002/01/24	2	2	3	6	6	1	3	2	6
2004/01/27	2	4	3	1	5	7	4	13	4
2003/04/9-11	4	30	11	10	12	12	14	12	13
2003/11/13	2	1	7	4	18	5	14	3	6
2004/02/20	2	3	2	4	5	8	6	13	5
2004/04/13	2	6	10	3	1	2	5	1	8
2004/07/21	2	2	7	1	2	2	12	2	2
2004/07/22	4	7	4	6	10	2	7	4	6
2004/07/23	2	4	1	3	10	9	11	13	6
2004/08/14	3	3	1	2	8	3	9	11	16
2005/01/29	2	3	3	10	5	3	4	0	2
2005/02/07	3	2	4	2	5	3	5	5	3
2005/02/08	2	14	4	0	1	2	2	9	9
2005/03/30	3	7	4	5	8	4	5	12	14
2005/05/31	4	11	10	8	6	17	10	9	24
2005/06/01	2	5	9	17	14	3	15	12	1
2005/06/02	4	7	7	3	3	5	6	9	9
2005/06/28	2	4	2	7	6	4	1	5	7
2005/06/29	3	26	15	3	20	15	22	22	14
2005/06/30	3	8	2	4	5	11	4	6	6
2006/01/10	6	3	6	9	10	9	14	20	16
2006/01/11	8	10	8	6	9	10	15	16	13
2006/01/12	8	7	7	8	10	11	15	11	18
2006/01/13	7	11	9	6	10	5	10	13	7
2006/02/17	2	30	1	10	3	16	13	2	0
2006/06/21	8	7	10	12	8	12	6	13	17
2006/09/25	9	9	7	9	10	10	14	16	14
2006/09/26	6	12	9	10	15	10	12	14	10
2006/09/27	4	14	4	11	4	10	9	26	19
2006/09/29	10	12	7	10	25	20	11	23	18
Average		8	6	6	8	7	9	10	9

^a Number of analysis.

Table 5.15: Percent deviation among daily repeat analyses of the KL2G standard

Date (ymd)	n ^a	Na2O	Ti	Cr	Sr	Y	Zr
2001/08/22	2	12	7	7	8	4	3
2002/01/28	2	4	2		3	1	0
2002/01/29	2	2	2		2	3	2
2003/12/13	2	1	0	1	1	2	2
2004/02/12	4	17	9	11	10	11	10
2004/02/17	2	7	6	7	7	4	7
2004/04/14	2	1	0	0	1	3	2
2004/05/07	3	5	5	4	5	4	5
2004/09/17	3	6	2	5	4	5	6
2004/11/26	2	4	0	1	0	1	0
2005/01/29	2	2	2	1	3	0	2
2005/02/09	3	2	1	3	2	2	3
2005/03/28	4	6	3	5	6	5	4
2005/03/29	3	1	4	2	3	4	5
2005/07/01	4	6	0	2	1	3	3
2005/07/29	5	9	4	9	7	8	5
2006/01/09	4	2	3	1	3	5	3
2006/02/16	6	7	3	7	9	6	7
2006/02/17	4	3	1	1	2	3	3
2006/09/28	4	9	4	8	7	8	8
Average		5	3	4	4	4	4

^a Number of analysis.

Table 5.D1: Major and trace element compositions of unusual phases in Sample 6K-465-3.

Grain ^a Type ^b Notes ^c	Ilm1 Core GbVn	Ilm1 Rim GbVn	Apat2 Ints GbVn	Kaer1 Ints KtVn	PlagK1A Ints KtVn	PlagK1B Ints KtVn	RutK1 Ints KtVn	ApatK1 Ints KtVn	Cpx7 Rim1 NrMat	Plag7 ExLm NrMat
Major element oxides (Wt%)										
Pts	2	3	2	6	8	8	6	5		
SiO ₂	0.00	0.00	0.18	42.97	56.65	56.33	0.00	0.11		
TiO ₂	53.79	53.90		5.27			98.09			
Al ₂ O ₃	0.00	0.00	0.00	11.78	26.97	27.26	0.01	0.01		
Cr ₂ O ₃	0.10	0.08		0.71			0.74			
Fe ₂ O ₃	0.00	0.00					0.00			
FeO	40.53	41.34	0.13	4.20	0.07	0.09	0.09	0.08		
MnO	0.68	1.34	0.06	0.05			0.01	0.02		
MgO	3.82	2.07	0.04	16.62	0.00	0.35	0.00	0.11		
NiO	0.05	0.02					0.03			
ZnO										
CaO	0.00	0.00	56.01	11.77	8.87	9.16	0.06	55.99		
Na ₂ O			0.01	2.97	6.34	6.08		0.00		
K ₂ O			0.00	0.51	0.16	0.06		0.00		
P ₂ O ₅			42.14					42.14		
Cl			0.44					0.71		
F			1.94					1.41		
Total	98.97	98.76	100.95	96.84	99.07	99.33	99.05	100.57		
Plag An%					43.21	45.29				
Trace elements (ppm)										
La			57051	6.91	2.64	2.29	246	93224	4.33	3.85
Ce			225756	40.99	5.20	4.76	633	299119	25.52	8.16
Nd			196323	49.06	1.15	1.58	78	199727	28.28	2.64
Sm			67559	19.16	0.14	0.12		56328	9.57	0.72
Eu			17484	4.51	1.15	0.95	12	10851	2.62	1.63
Dy			60326	26.35	0.08	0.07	15	46743	12.38	0.36
Er			26201	13.80	0.09	0.10	10	19338	7.10	0.49
Yb			23713	10.61	0.03	0.03	58	16184	7.50	0.28
Na ₂ O				3.35	4.83	4.45			0.47	4.37
Ti				36269	256	229			2067	428
Cr			204	5730	1.25	1.29		575	3501	5100
Sr			83355	31.02	89.29	76.86	3210	86244	4.60	68.32
Y			485488	156	0.27	0.60	1049	362111	70.91	1.64
Zr			11675	270	0.14	0.75		9155	94.31	1.79

^a Grain number and mineral phase: ilmenite (Ilm), apatite (Apat), kaersutite (Kaer), plagioclase (Plag), clinopyroxene (Cpx) and rutile (Rut). Vertical lines separate different mineral groupings.

^b ExLm indicates exsolution lamellae in Cpx.

^c GbVn refers to a gabbroic vein, KtVn to a kaersutite vein that also contains Plag, Rut and Apat, and NrMat to the peridotite matrix <5 cm from the gabbroic vein.

Table 5.D1: 6K-465-3 Cont.

Grain	Cpx12	Plag12	Oliv12	Spin12	Cpx14	Plag14C	Cpx14	Plag14R	Cpx15	Plag15
Type	Core2	ExLm	ExLm	ExLm	Core	ExLm	Rim	ExLm	Core	ExLm
Notes	NrMat	NrMat	NrMat	NrMat	NrMat	NrMat	NrMat	NrMat	NrMat	NrMat
Major element oxides (Wt%)										
Pts	19	6	7	1	17	1	8	1	18	2
SiO2	53.10	55.27	40.78	0.04	51.89	55.21	53.44	55.33	53.16	53.26
TiO2	0.42			0.63	0.42		0.37		0.33	
Al2O3	2.86	27.71	0.00	21.26	3.26	27.20	2.11	27.91	2.78	29.29
Cr2O3	1.04			42.17	2.18		0.98		1.01	
Fe2O3				1.97						
FeO	2.72	0.16	11.42	26.52	3.10	0.13	2.63	0.10	2.64	0.17
MnO	0.08			0.39	0.08		0.07		0.07	
MgO	16.56	0.04	46.77	5.36	16.35	0.86	16.54	0.02	16.52	0.01
NiO				0.12						
ZnO				0.43						
CaO	22.24	10.10	0.06	0.33	21.59	9.65	22.78	9.94	22.72	11.82
Na2O	0.63	5.86			0.59	5.68	0.49	5.92	0.49	4.80
K2O										
P2O5										
Cl										
F										
Total	99.66	99.16	99.04	99.21	99.47	98.73	99.41	99.22	99.73	99.35
Plag An%		48.80				48.42		48.13		57.67
Trace elements (ppm)										
La	3.92	4.96								
Ce	18.87	9.76								
Nd	15.68	2.71								
Sm	5.81	0.78								
Eu	0.92	1.03								
Dy	6.16	0.56								
Er	4.37	0.46								
Yb	3.84	0.33								
Na2O	0.53	3.77	0.12							
Ti	2194	343	54							
Cr	5235	413	220							
Sr	1.51	34.38	1.83							
Y	58.05	5.29	0.16							
Zr	103	8.82	0.32							

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16. Abstract (Limit: 200 words) I provide constraints on mantle convection through observations of the rheology and composition of the oceanic upper mantle. Convection cannot be directly observed, yet is a fundamental part of the plate tectonic cycle. Relative motion among plates is accommodated by localized deformation at their boundaries. I demonstrate that in the ductile regime, strain localization occurs when different mineral phases are mixed together, limiting grain annealing. Upper mantle flow is by dislocation creep, resulting in seismic anisotropy due to mineral alignment. I use a shear zone in the Josephine Peridotite to quantify the relationship between mineral orientation and shear strain, providing an improved framework for the interpretation of seismic anisotropy. The upper mantle is generally assumed to be homogeneous in composition. From detailed isotopic and chemical analyses of abyssal peridotites from the Southwest Indian Ridge, I show that the mantle is heterogeneous at a range of length-scales. Abyssal peridotites recovered at ocean ridges are generally interpreted as the depleted residues of melt extraction. I find that melt-rock reaction is a significant part of the melt extraction process, modifying the composition of the lithospheric mantle. The generation of heterogeneous lithosphere provides a source for asthenospheric heterogeneity, via subduction and mantle convection.			
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