### **OFFICE OF NAVAL RESEARCH**

Grant or Contract N00014-95-1-0302 PR# 97PR02146-00

Technical Report No. P287

Simulations of the In Situ Cyclic Voltammetry Dependent EPR Spectra and DC Conductivity

by

X.-L. Wei and A.J. Epstein

Prepared for Publication in

Synthetic Metals

The Ohio State University Department of Physics Columbus, OH

DTIC QUALITY INSPECTED 2

September 20, 1997

Reproduction in while in part is permitted for any purpose of the United States Government

This document has been approved for public release and sale; its distribution is unlimited.

This statement should also appear in item ten (10) of the Document Control Data DD Form 1473. Copies of the form available from cognizant or contract administrator.

# 19971015 026

1

REPORT DOCUMENTATION PAGE	Form Approved OMB No. 0704-0188
Public reporting burden for this collection of information isestimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, collection of information, including suggestions for reducing this burden, to Washington Headquarters Services, Directorate for information Operations and Rejports, 1215 Jefferson Davis Highway, Suite 1204, Arlington VA 22202-4302, and to the Office of Management information Project (0704-0188), Washington DC 20503	
and Budger, Paperwork Reduction Figure (1) 2. REPORT DATE 3. REPORT TY   1. AGENCY USE ONLY (Leave blank) 2. REPORT DATE 3. REPORT TY   9/20/97 Te	/PE AND DATES COVERED echnical
4. TITLE AND SUBTITLE 5. FUNDING NUMBERS Simulations of the In Situ Cyclic Voltammetry Dependent EPR Spectra N00014-95-1-0302 and DC Conductivity	
6. AUTHOR(S) XL. Wei, A.J. Epstein	
7. PERFORMING ORGANIZATION NAMES AND ADDRESS(ES)8. PERFORMINGDepartment of PhysicsP2The Ohio State University174 West 18th AvenueColumbus, OH43210-1106	NG ORGANIZATION REPORT NUMBER 287
9. SPONSORINGMONITORING AGENCY NAME(S) AND ADDRESS(ES)   10. SPONSORINGMONITORING AGENCY REPORT NUMBER     Office of Naval Research   800 N. Quincy Street     Arlington, VA 22217   22217	
11. SUPPLEMENTARY NOTES Prepared for publication in Synthetic Metals	
12a. DISTRIBUTION/AVAILABILITY STATEMENT   12b. DISTRIB     Reproduction in whole or in part is permitted for any purpose of the US Government.   12b. DISTRIB     This document has been approved for public relearse and sale; its distribution is unlimited.   12b. DISTRIB	UTION CODE
13. ABSTRACT (Maximum 200 words)	
In Situ cyclic voltammetry (CV) dependent EPR and DC conductivity of polyaniline have been reported by many groups. However, the variation of the EPR intensities at two half-wave potentials of the CV scan and the asymmetric CV dependent DC conductivity have remained to be fully accounted for. Here we present in detail a novel quasi-random oxidation model and the related simulation results to interpret the reported in situ experimental results. This model quantitatively describes many phenomena and physical properties found in polyaniline including the origin of the defect states, the variations of the in situ EPR signal during CV potential scan, the effects of the hydrolysis, and the pH-dependent DC conductivity data. The statistical nature of this model suggests its general applicability to the oxidation processes of other conducting polymers. The important roles of nearest neighbor Coulomb interaction and formation of a metallic polaron lattice in the computer modeling are evaluated and discussed	
14. SUBJECT TERMS	15. NUMBER OF PAGES
Polyaniline, Electron spin resonance spectra, Cyclic voltainmetry, DC concurrence spectra 2 Quasi-random oxidation model, Polaron lattice, Coulomb interaction, Computer simulation 2	
17. SECURITY CLASS. OF RPT 18. SECURITY CLASS OF THIS PG. 19. SECURITY CLASS Unclassified Unclassified Unclass	16. PRICE CODE OF ABSTRCT. 20. LIMITATION OF ABSTRACT ified Unlimited

## SIMULATIONS OF THE IN SITU CYCLIC VOLTAMMETRY DEPENDENT EPR SPECTRA AND DC CONDUCTIVITY

X.-L. Wei<sup>a</sup>, A. J. Epstein<sup>a, b, c</sup>

<sup>a</sup> Chemical Physics Program, <sup>b</sup> Department of Physics, and <sup>c</sup> Department of Chemistry The Ohio State University, Columbus, OH 43210-1106, USA

#### Abstract

In situ cyclic voltammetry (CV) dependent EPR and DC conductivity of polyaniline have been reported by many groups. However, the variation of the EPR intensities at two half-wave potentials of the CV scan and the asymmetric CV dependent DC conductivity have remained to be fully accounted for. Here we present in detail a novel quasi-random oxidation model and the related simulation results to interpret the reported in situ experimental results. This model quantitatively describes many phenomena and physical properties found in polyaniline including the origin of the defect states, the variations of the in situ EPR signal during CV potential scan, the effects of the hydrolysis, and the pH-dependent DC conductivity data. The statistical nature of this model suggests its general applicability to the oxidation processes of other conducting polymers. The important roles of nearest neighbor Coulomb interaction and formation of a metallic polaron lattice in the computer modeling are evaluated and discussed.

Keywords: Polyaniline, Electron spin resonance spectra, Cyclic voltammetry, D.c. conductivity, Quasi-random oxidation model, Polaron lattice, Coulomb interaction, Computer simulation.

#### Introduction

Polyaniline has been an exciting research field for magnetic, electric, and optical studies in both the theoretical and the applied areas. One of the highlights of these research efforts lay in the design and performance of many types of sophisticated in situ cyclic voltammetry (CV) dependent experiments, for example, the in situ CV/EPR [1-8], in situ CV/(DC conductivity) [5, 7-12], and in situ CV/optical studies [5, 8, 13-15]. However, many essential questions left as unexplained or remained to be fully accounted for until now. For example, the interpretations on the variation of the EPR intensities at two half-wave potentials of the CV scan and the asymmetric CV dependent DC conductivity have not yet been given until now. In addition, some confusion about conducting mechanisms [4, 7, 8] of emeraldine salt arose from these experiments and remained to be resolved. In this paper, a novel quasi-random oxidation model and the related simulation results are presented to account for the in situ CV dependent experimental results and to resolve the confusion in conducting mechanisms in conducting polymer materials.

#### Background

In situ CV/EPR and CV/(DC Conductivity) Spectra. In Figure 1, a typical set of reported in situ CV/EPR [2] and CV/(DC conductivity) [7] spectra are shown. We note the following features: (1) The DC conductivity does not rise until the EPR intensity begins to drop, which was suggested to originate from a competition between formation of a bipolaron lattice vs. a polaron lattice mechanism [7-9]; (2) the DC conductivity is asymmetric about the vertical axis cross its center of gravity; (3) the EPR peak maxima occur almost coincidentally at the same



Figure 1. A set of reported *in situ* CV/EPR [2] and CV/(DC conductivity) [7] spectra: All the curves are labeled in the legend box inside the figure.

Locations (slightly inward) with those of CV; (4) the two integrated EPR peaks are very different in their intensities (the ratio is about  $2 \sim 3$ ) as contrast to the integrated areas of the two CV peaks that are almost equal in size.

Quasi-random Oxidation Model. Several assumptions are made in this model [16]: (1) At the early stage of the oxidation process the polymer repeat units are oxidized to polarons; (2) Coulomb repulsion favors the configurations with the largest separation of charged polarons; (3) adjacent pairs of polarons when they do occur have zero magnetic susceptibility [17]; (4) Curie and Pauli susceptibility are separable and are directly related to the CV



Figure 2. The simulation results of the *in situ* DC conductivity as a function of CV potential scan. Various curves are labeled in the legend table: CV (---), DC conductivity (0) (after Kruszka *et al* [7]), and the number of Pauli spin lattice points for various  $P_2/P_1$  ratios (broken lines).

in situ EPR signal (Curie and Pauli) and DC conductivity (Pauli alone). The detailed discussion can be found elsewhere [19].

Computer simulations. Simulations [19] have been carried out for chain lengths up to 1000  $C_6N$  units. Hundreds (usually 500 or more) of sequential simulations were carried out and their results were averaged. The integrated intensity was normalized to the chain length used in the simulations. Then the normalized oxidation density distribution function was calculated and used in the simulations of CV-dependent spectra. The important contribution of polaron lattice to Pauli susceptibility is included in the simulations at the tenth the susceptibility of room temperature Curie spins [17].

#### **Results and Discussion**

The simulation results of CV-dependent EPR spin density converges for an assumed probability ratio of forming 2<sup>nd</sup> nearest neighbor sites (P<sub>2</sub>) to 1<sup>st</sup> nearest neighbor sites (P<sub>1</sub>) of P<sub>2</sub>/P<sub>1</sub> ~ 200, well agreeing the reported CV-dependent EPR spectra in all features described in previous section [17]. In Figure 1, it is seen that the simulation results of the number of Pauli spin carriers, defined as the number of Pauli lattice points, converge to an asymmetric asymptote with  $P_2/P_1 \sim 200$ , which fits well to the experimental DC conductivity data (labeled with "o"). This provides evidence for the importance of the polaron lattice formation for high conductivity. In addition, insight can also be drawn for conducting mechanism studies: (1) the DC conductivity is proportional to the number of Pauli spin carriers; (2) the conduction of polyaniline is thus via the polaron lattice. Therefore this result disagrees with the bipolaron conduction model for polyaniline samples.

It is noticed that the converged  $P_2/P_1$  ratio yields a dielectric constant ~ 10 [19], a reasonable averaged value of nonconducting and conducting powder samples [20]. Therefore the simulation result is also consistent with the experimental dielectric constant data of PANI. It is worthwhile to note that the presence of the defect states of the base forms of polyaniline and its derivatives is a natural outcome of the *quasi-random oxidation model*, being also a supporting evidence for our model.

#### Conclusions

A summary of the *quasi-random oxidation model* was presented. The related simulation results showed good agreement with the reported *in situ* CV-dependent spectra. The important role of Coulomb interaction and the formation of polaron lattice were shown.

#### Acknowledgment

This work has been supported in part by the US Office of Naval Research.

#### References

[1] J. F. Wolf, C. E. Forbes, S. Gould, and L. W. Shacklette, J. Electrochem. Soc. 136, 2887 (1989).

[2] S. H. Glarum and J. H. Marshall, J. Phys. Chem. 92, 4210 (1988).

[3] B. Villeret and M. Nechtschein, Phys. Rev. Lett. 63, 1285 (1989).

[4] J. Tang, R. D. Allendoerfer, and R. A. Osteryoung, J. Phys. Chem. 96, 3531 (1992).

[5] T. Ohsawa, T. Kabata, O. Kimura, M. Onoda, and

Katsumi Yoshino, Jpn J. of Applied Physics 28, 996 (1989).

[6] M. Lapkowski and E. M. Genies, J. Electroanal. Chem. 279, 157 (1990).

[7] J. Kruszka, M. Nechtschein, and C. Santier, Rev. Sci. Instrum. 62, 695 (1991).

[8] J. Lippe and R. Holze, Mol. Cryst. Liq. Cryst. 208, 99 (1991).

[9] W. W. Focke and G. E. Wnek, J. Electroanal. Chem. 256, 343 (1988).

[10] T. Ohsawa, O. Kimura, M. Onoda, and K. Yoshino, Synth. Met. 47, 151 (1992).

[11] L. Olmedo, I. Chanteloube, A. Germain, M. Petit, E. M. Genies, Synth. Met. 28, C165 (1989).

[12] M. Kaya, A. Kitani and K. Sasaki, Japan. Chem. Lett. 724, 147 (1986).

[13] H. Kuzmany and M. Bartonek, Europhys. Lett. 12, 167 (1990).

[14] A. Efremova, A. Regis, L. Arsov, Electrochim. Acta 39, 839 (1994).

[15] R. Jiang, S. Dong, and S. Song, J. Chem. Soc., Faraday Trans. 1 85, 1575 (1989).

[16] J.S. Miller and A.J. Epstein, Angew. Chem. Int. Ed. Engl. 33, 385 (1994).

[17] X.-L. Wei, Y.-Z. Wang, S. M. Long, C. Bobeczco, and A. J. Epstein, J. Am. Chem. Soc. **118**, 2545 (1996).

[18] H. Weiss, O. Pfefferkorn, G. Kotora, and D. Brian, J. Electrochem. Soc. 136, 3711 (1989).

[19] X.-L. Wei, Ph. D. Dissertation, The Ohio State University (1996).

[20] Z. H. Wang, A. Ray, A. G. MacDiarmid, and A. J. Epstein, Phys. Rev. B. 43, 4373 (1991); J. Joo, Z. Oblakowski, G. Du, J. P. Pouget, E. J. Oh, J. M. Wiesinger, Y. Min, A. G. MacDiarmid, and A. J. Epstein, Phys. Rev. B. 19, 2977 (1994).