

REPORT DOCUMENTATION PAGE			Form Approved OMB No. 0704-0188	
Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing the collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden, to Washington Headquarters Services, Directorate for Information Operations and Reports, 1215 Jefferson Davis Highway, Suite 1204, Arlington, VA 22202-4302, and to the Office of Management and Budget, Paperwork Reduction Project (0704-0188), Washington, DC 20503.				
1. AGENCY USE ONLY (Leave blank)	2. REPORT DATE May 31, 1995	3. REPORT TYPE AND DATES COVERED Annual Report 1 Jun 94 - 31 May 95		
4. TITLE AND SUBTITLE  Reactions of atmospheric cluster ions		5. FUNDING NUMBERS  F49620-93-1-0372 61103D 3484/XS		
6. AUTHOR(S)  Stephen R. Leone Veronica M. Bierbaum		8. PERFORMING ORGANIZATION REPORT NUMBER  AFOSR-IR-95-0400		
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES)  The Regents of the University of Colorado Campus Box 19 Boulder, CO 80309-0019		10. SPONSORING / MONITORING AGENCY REPORT NUMBER		
9. SPONSORING / MONITORING AGENCY NAME(S) AND ADDRESS(ES) AFOSR/MENL Building 410, Bolling AFB, DC 20332-6448  Dr. Michael R. Berman.		11. SUPPLEMENTARY NOTES		
12a. DISTRIBUTION / AVAILABILITY STATEMENT  Approved for Public Release; Distribution is unlimited.		12b. DISTRIBUTION CODE  <b>DTIC SELECTED JUN 27 1995</b>		
13. ABSTRACT (Maximum 200 words)  A flow-drift tube instrument has been employed to measure the mobilities of five cluster ion systems that have been detected in the earth's atmosphere. The mobilities of $\text{NO}^+(\text{H}_2\text{O})_{0-2}$ , $\text{H}_3\text{O}^+(\text{H}_2\text{O})_{0-3}$ , $\text{NH}_4^+(\text{NH}_3)_{0-3}$ , $\text{NH}_4^+(\text{CH}_3\text{CN})_{0-3}$ , and $\text{NO}^+(\text{CH}_3\text{CN})_{0-3}$ were determined as a function of field strength by the dual ion signal depletion method. For $\text{H}_3\text{O}^+(\text{H}_2\text{O})_{0-3}$ cluster ions, the reduced zero-field mobility decreases by almost equal amounts as a function of n, whereas for the $\text{NH}_4^+(\text{NH}_3)_{0-3}$ , $\text{NH}_4^+(\text{CH}_3\text{CN})_{0-3}$ , and $\text{NO}^+(\text{CH}_3\text{CN})_{0-3}$ cluster ions, the decrease in mobility with cluster size is more gradual for higher n. A simple geometrical model based on the effective cross-sectional area of the cluster ions is consistent with the observed n-dependence of the mobilities.				
14. SUBJECT TERMS  atmospheric cluster ions    hydronium ion    water flow-drift    ammonium ion    ammonia mobility    nitric oxide ion    acetonitrile			15. NUMBER OF PAGES 2	
17. SECURITY CLASSIFICATION OF REPORT UNCLASSIFIED			18. SECURITY CLASSIFICATION OF THIS PAGE UNCLASSIFIED	
19. SECURITY CLASSIFICATION OF ABSTRACT UNCLASSIFIED			16. PRICE CODE	
20. LIMITATION OF ABSTRACT				

Standard Form 298 (Rev. 2-89)

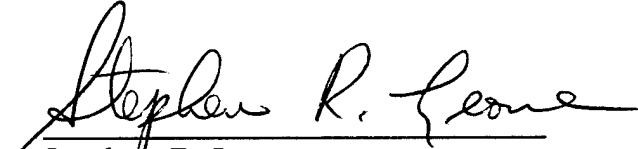
DTIC QUALITY INSPECTED 3

19950626 038

AASERT Evaluation Report  
F49620-93-1-0372  
May 31, 1995

Stephen R. Leone  
JILA, University of Colorado  
Boulder, CO 80309-0440

Parent Award No.	FA9620-92-J-0072
Amount of parent award prior to AASERT award:	\$520,192 (36 months)
Number of full time graduate students one year prior to AASERT award:	2.0
Amount of parent award after AASERT award:	\$501,000 (36 months)
Number of full time graduate students one year after AASERT award on parent grant:	1.25
Number of full time graduate students two years after AASERT award on parent grant	0
Amount of funding of AASERT award F49620-93-1-0372:	\$102,582 (36 months)
Continuous number of full time graduate students on AASERT award:	1.0

  
\_\_\_\_\_  
Stephen R. Leone

# "REACTIONS OF ATMOSPHERIC CLUSTER IONS"

## Annual Technical Report

June 1, 1994 - May 31, 1995

Cluster ions play an important role in the ion chemistry of the earth's atmosphere. For example, clusters with  $\text{NO}^+$ ,  $\text{H}_3\text{O}^+$  and  $\text{NH}_4^+$  as core ions, and with  $\text{H}_2\text{O}$ ,  $\text{NH}_3$  or  $\text{CH}_3\text{CN}$  as solvating ligands, among many others, have been detected in the earth's troposphere, stratosphere and ionosphere. Although the association and dissociation processes forming these ions are in general well-understood, their mobilities are largely unknown. These values are essential, not only in modeling atmospheric phenomena, but also in providing direct information about the ion-buffer gas interaction potential.

Our selected ion flow tube apparatus has been modified to include a well-defined flow drift region and instrumentation for ion modulation and data acquisition. Core ions are generated in an ion source, mass-selected and injected into the flow tube where they associate with added solvent molecules before entering the drift region. Two drift rings, at known separation, are simultaneously pulsed; the resulting ion depletions are detected with a quadrupole mass filter in a time-resolved manner, as a function of  $E/N$ . The measured zero-field mobilities for five cluster systems in helium are summarized in Table 1.

Table 1. Reduced zero-field mobilities ( $\text{cm}^2 \text{V}^{-1} \text{s}^{-1}$ ) of cluster ions at 300K

$\text{A}^+(\text{B})_n$	$n=0$	$n=1$	$n=2$	$n=3$
$\text{NO}^+(\text{H}_2\text{O})_n$	$22.4 \pm 0.5$	$16.8 \pm 0.5$	$12.9 \pm 0.3$	
$\text{H}_3\text{O}^+(\text{H}_2\text{O})_n$	$21.5 \pm 0.5$	$17.6 \pm 0.4$	$13.7 \pm 0.3$	$10.4 \pm 0.3$
$\text{NH}_4^+(\text{NH}_3)_n$	$22.1 \pm 0.6$	$15.7 \pm 0.3$	$11.4 \pm 0.3$	$10.2 \pm 0.3$
$\text{NH}_4^+(\text{CH}_3\text{CN})_n$	$22.1 \pm 0.6$	$12.3 \pm 0.5$	$6.8 \pm 0.2$	$6.7 \pm 0.2$
$\text{NO}^+(\text{CH}_3\text{CN})_n$	$22.4 \pm 0.5$	$12.3 \pm 0.3$	$7.9 \pm 0.5$	$7.9 \pm 0.6$

The observed size dependence of the cluster ion mobilities can be divided into two classes. For  $\text{H}_3\text{O}^+(\text{H}_2\text{O})_n$  with  $n=0-3$ , the mobilities decrease uniformly as the cluster size increases; this behavior reflects the systematic increase in the ionic cross section as water molecules add to the planar ionic structure. In contrast, for three other systems, the mobility decreases less dramatically for higher  $n$  species, and ions with two and three solvent molecules have similar mobilities. This behavior reflects a "tetrahedral" geometry where the effective ionic cross section increases only slightly during the filling of the first solvation shell. These experimental findings are consistent with simple molecular modeling. We are currently studying the mobilities of clusters with other core ions and solvent molecules, mixed cluster systems and the effect of other buffer gases.

<input checked="" type="checkbox"/>
<input type="checkbox"/>
<input type="checkbox"/>
ies
or
Dist
A-1