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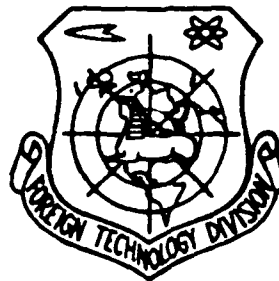
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CARBON/CARBON COMPOSITION ABLATION EFFECTS ON IONIZED BOUNDARY LAYERS

by

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TITLE: CARBON/CARBON COMPOSITION ABLATION EFFECTS ON IONIZED
BOUNDARY LAYERS

AUTHOR: Wei Shuru Wang Ce Wang Fuhan

Abstract A calculation method for carbon/carbon composition thermochemical ablation effects on ionized laminar and turbulent boundary layers is given. In the boundary layers 20 gas species are considered and all chemical reactions which occurred in them are assumed to be equilibrium. The new solving method for boundary layer equations and simple disposal for transport properties are applied. The calculation examples indicate that ablation has great effects on ionization properties of the boundary layers and that ionizations of the alkaline metals are not always dominant.

Key words ablation, ionization, boundary layer, laminar, turbulent.

V. SAMPLE CALCULATIONS AND ANALYSIS

We used Reference [11]'s sample calculations and sample calculations for other object forms as well as flight conditions to carry out calculations. A part of the results are as shown in Fig.1-3. From the results of calculations, it is possible to obtain:

(1) Combustion corrosion or ablation has a very great influence 414 on boundary layer ionization characteristics. At analogous or corresponding points (for example, wall surfaces), the ablation values for electron number densities are highest as compared to pure air values under the same conditions by 3-4 orders of magnitude^[9] (See Fig.1). The peak values in boundary layers are 1-2 orders of magnitude higher (Fig.2).

(2) The introduction of ablation products causes electron number density cross sections to tend toward saturation, that is to say, the high density regions get thicker, turbulent flow boundary layers in particular (Fig.2).

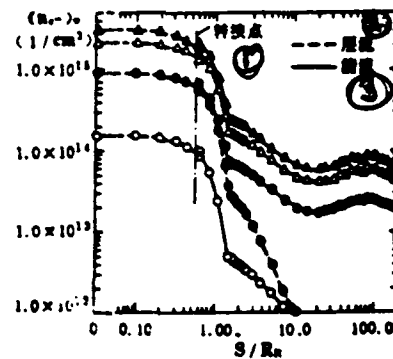


图1 烧蚀壁面电子数密度分布

● $\bar{N}_{Nec} = 100 \text{ ppm}$ $\Delta \bar{N}_{Nec} = 500 \text{ ppm}$
 $\blacktriangle \bar{N}_{Nec} = 1000 \text{ ppm}$ ○ 纯空气

Fig.1 Ablation Wall Surface Electron Number Density Distribution (1) Turning or Transition Point (2) Laminar Flow (3) Turbulent Flow (4) Pure Air

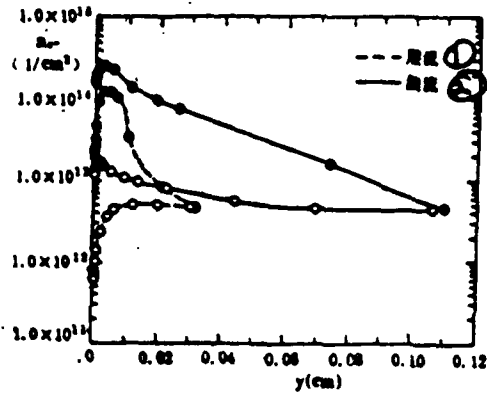


图2 边界层中电子数密度分布
 ● $\bar{K}_{N_{O_2}} = 100 \text{ ppm}$ ○ 纯空气 (3)
 $S/R_N = 2.35$ $P_s = 7.7499 \times 10^5 \text{ Pa}$
 $u_s = 2.8678 \times 10^3 \text{ m/s}$ $h_s = 4.9610 \times 10^6 \text{ J/kg}$
 $R_N = 1.27 \text{ cm}$

Fig.2 Distribution of Electron Number Density in Boundary Layers (1) Laminar Flow (2) Turbulent Flow (3) Pure Air

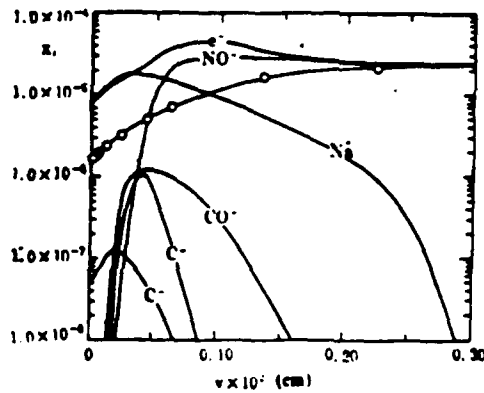


图 3(a)
 ○ 空气边界层的 x_s (1) $\bar{K}_{N_{O_2}} = 100 \text{ ppm}$
 $S/R_N = 0.00$ (驻点层流) $R_N = 1.27 \text{ cm}$
 $P_s = 1.5180 \times 10^7 \text{ Pa}$ $T_w = 4417.2 \text{ K}$
 $u_s = 0.00 \text{ m/s}$ $\dot{m}_w = 0.2789 \text{ g/cm}^2 \cdot \text{s}$
 $h_s = 9.0730 \times 10^6 \text{ J/kg}$

3(a) (1) Air Boundary Layer (2) Stationary Point Laminar Flow

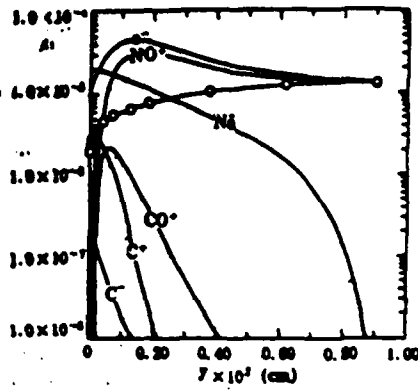


图 3(b):

① 空气边界层的 x_e ② 湍流

$\bar{K}_{Nec} = 100 \text{ ppm}$
 $R_N = 1.27 \text{ cm}$
 $T_w = 4427.0 \text{ K}$
 $\dot{m}_w = 0.8955 \text{ g/cm}^2 \cdot \text{s}$
 $P_e = 9.7110 \times 10^5 \text{ Pa}$
 $u_e = 1.2343 \times 10^3 \text{ m/s}$
 $h_e = 7.8430 \times 10^5 \text{ J/kg}$

Fig.3(b) (1) Air Boundary Layer (2) Turbulent Flow

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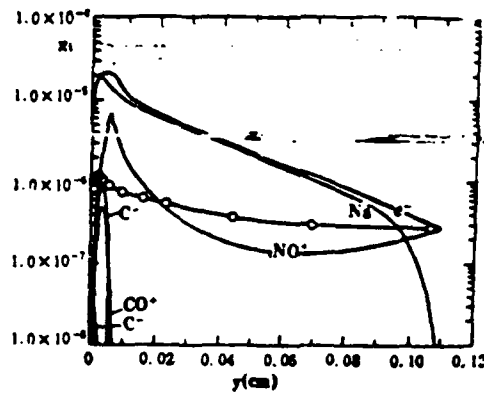


图 3(c):

① 空气边界层的 x_e ② 湍流

$\bar{K}_{Nec} = 100 \text{ ppm}$
 $R_N = 1.27 \text{ cm}$
 $T_w = 3831.9 \text{ K}$
 $\dot{m}_w = 0.1465 \text{ g/cm}^2 \cdot \text{s}$
 $P_e = 7.7499 \times 10^5 \text{ Pa}$
 $u_e = 2.8678 \times 10^3 \text{ m/s}$
 $h_e = 4.9610 \times 10^5 \text{ J/kg}$

Fig.3(c) (1) Air Boundary Layer (2) Turbulent Flow

(3) Within ablation wall surfaces and their adjacent thin layers, electrons, generally speaking, come mainly from ablation products, in particular, ions from alkali metals. However, in most of the depth of boundary layers, as far as whether air components or ablation products occupy the dominant position in ionization mechanisms

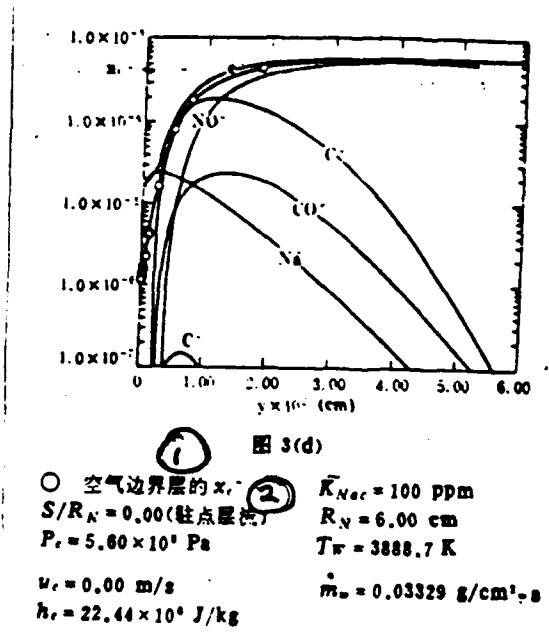


Fig.3(d) (1) Air Boundary Layer (2) Stationary Point Laminar Flow

Fig.3 Mol Fraction Cross Sections for Charged Components

is concerned, it is still decided by the conditions of flight and the locations at which the boundary layers find themselves. Generally speaking, stationary point laminar flows, even to include turning or transition points, hold relatively close to past turbulent flows, and air components are dominant (Fig.3(a), (b), and (d)). However, in the lower reaches of the flow, by contrast, ablation products hold the dominant position (Fig.3(c)). As far as the ionization component which occupies the dominant position is concerned, the rules or patterns for the transition from air to ablation products are somewhat similar to the rules or patterns for boundary layer transitions. However, the points of change are all in the reaches of the flow below the boundary layer transition points.

(4) Among ablation products, alkali metal ionization most certainly does not occupy the dominant position overall. Among the boundary layers of spherical tip regions when flying at relatively high altitudes, the relevant contributions of CO^+ --in particular C^+ --(this refers to $X_{\text{C}^+}/X_{\text{e}^-}$, and so on), at times, reach even as high as 70% (Fig.3(d)). These mechanisms have important effects on

raising electron number density peak values and increasing the thicknesses of high concentration regions.

REFERENCES

- [1] Blottner, F.G., *AIAA J.* 8, 2(1970), 193-205.
- [2] Huheade (phonetic), et.al.: "Tip Cover Plasma Analysis of Sphericoconical Long Thin Objects", *Acta Aerodynamica Sinica*, 1, 4(1983), 77-85
- [3] Wei Shuru, Wang Fuhan, and Wang Ce; "Ionized Air Equilibrium Boundary Layers", *Acta Aerodynamica Sinica*, 7, 2(1989), "183-191
- [4] Bartlett, E.P. & Kendall, R.M., NASA CR-1062 (1968).
- [5] Duo Lansì; "Hypersonic Viscous Flows", National Defense Industry Press, (1966), 193-203
- [6] Tong, H., et al, NASA N73-31835 (1973).
- [7] Wei Shuru, et.al.; "One Type of New Solution Method for Boundary Layer Equations", *Mechanics Journal*, 19, extra, (1987), 55-63
- [8] Putq, K.E. & Bartlett, E.P., *AIAA Paper* 72-91.
- [9] Wei Shuru, et.al.; "Ablation of Carbonized Materials in Ionized Air Environments", *Acta Aerodynamica Sinica*, 4, 4(1986), 352-360
- [10] Глушко, В.П. и др., Термодинамические свойства индивидуальных веществ, ТОМ I. ТОМ IV. «НАУКА», МОСКВА (1979-1982).
- [11] Anderson, L.W. & Morse, H.L., AD-734744. (1971).

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