DEL-TR-3295

AD-A243 180

TECHNICAL REPORT BRL-TR-3295

BRL



LASER-BASED IGNITION OF H<sub>2</sub>/O<sub>2</sub> AND D<sub>2</sub>/O<sub>2</sub>
PREMIXED GASES NEAR 243 NM:
THE FIRST REPORT OF A DEUTERIUM ISOTOPE
WAVELENGTH EFFECT IN LASER IGNITION

BRAD E. FORCH ANDRZEJ W. MIZIOLEK

DECEMBER 1991



APPROVED POR PUBLIC RELEASE; DISTRIBUTION IS UNLIMITED.

U.S. ARMY LABORATORY COMMAND

BALLISTIC RESEARCH LABORATORY
ABERDEEN PROVING GROUND, MARYLAND

**91** 12 - - - - -

### NOTICES

Destroy this report when it is no longer needed. DO NOT return it to the originator.

Additional copies of this report may be obtained from the National Technical Information Service, U.S. Department of Commerce, 5285 Port Royal Road, Springfield, VA 22161.

The findings of this report are not to be construed as an official Department of the Army position, unless so designated by other authorized documents.

The use of trade names or manufacturers' names in this report does not constitute indorsement of any commercial product.

# **UNCLASSIFIED**

### Form Anoroved REPORT DOCUMENTATION PAGE OMB No. 0704-0188 Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, aid completing and reviewing the collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for inducing this burden to washington Headquarters Services. Directorate for information Operations and Reports, 1215 Jefferson Davis Highway, Suite 1204, Arlington, VA 22202-4302, and to the Office of Management and Budget. Paperwork Reduction Project (0704-0188), Weshington, DC 20503 2. REPORT DATE 3. REPORT TYPE AND DATES COVERED 1. AGENCY USE ONLY (Leave blank) December 1991 Final, Jan 90 - Dec 90 5. FUNDING NUMBERS 4. TITLE AND SUBTITLE Laser-Based Ignition of H<sub>2</sub>/O<sub>2</sub> and D<sub>2</sub>/O<sub>2</sub> Premixed Gases Near 243 nm: PR: AFOSR-MIPR-91-0004 The First Report of a Deuterium Isotope Wavelength Effect in Laser Ignition 6. AUTHOR(S) Brad E. Forch and Andrzej W. Miziolek B. PERFORMING ORGANIZATION 7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) REPORT NUMBER 9. SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS(ES) 10. SPONSORING / MONITORING AGENCY REPORT NUMBER U.S. Army Ballistic Research Laboratory BRL-TR-3295 ATTN: SLCBR-DD-T Aberdeen Proving Ground, MD 21005-5066 11. SUPPLEMENTARY NOTES 12b. DISTRIBUTION CODE 12a, DISTRIBUTION / AVAILABILITY STATEMENT Approved for public release; distribution is unlimited. 13. ABSTRACT (Maximum 200 words) We have investigated the use of a tunable laser system which operates in the ultraviolet (uv) to ignite premixed reactive gaseous flows of HJO2 and DJO2 at atmospheric pressure. The amount of incident laser energy (ILE) required to ignite the premixed flows as a function of laser excitation wavelength show two distinct minima. The spectral position of these minima correspond exactly to the location of the resonance, two-photon excitation wavelengths of atomic hydrogen and deuterium at 243.07 and 243.00 nm respectively. The relative spacing between these minima at the energy level of the 1S-2S two-photon excited transition is 22 cm<sup>-1</sup> which is in excellent agreement with the known value for H-D deuterium isotope shift (22.4 cm<sup>-1</sup>). We believe that this is both the first report of a sensitive wavelength dependence on the laser energy required to ignite these mixtures through resonant multiphoton excitation of H and D atoms (produced from H, and D, photolysis) and the first report of a deuterium isotope-wavelength-effect in laser ignition. Measurement of the ILE required for ignition vs. equivalence ratio (Φ) shows that the most efficient ignition occurred with ~0.55 mJ ILE at Φ=0.7 in the fuel lean region. 15. NUMBER OF PAGES 14. SUBJECT TERMS 28 16. PRICE CODE Laser; ignition; hydrogen; oxygen 20. LIMITATION OF ABSTRACT 17. SECURITY CLASSIFICATION 18. SECURITY CLASSIFICATION SECURITY CLASSIFICATION OF REPORT OF THIS PAGE OF ABSTRACT

UNCLASSIFIED

**UNCLASSIFIED** 

UNCLASSIFIED

INTENTIONALLY LEFT BLANK.

# TABLE OF CONTENTS

		Page
	LIST OF FIGURES	V
	ACKNOWLEDGMENTS	vii
1.	INTRODUCTION	1
2.	EXPERIMENTAL	3
3.	RESULTS AND DISCUSSION	5
3.1 3.2 3.3	Microplasma Formation	5 9 11
4.	CONCLUSION	14
<b>5</b> .	REFERENCES	17
	DISTRIBUTION LIST	<b>د</b> 1

Assession For		
988 <b>4</b> €	4	
Ofti Twb		
ភាពប្រជាធានគឺ	<b>(</b> )	
lastification.	· <del></del>	
e en constitución de la constitu		
8,		
Distribution/		
72 <b>4</b> 17513,514	1.P.C.	
वि∀सः । स≇र	i/er	
olet , epacial		
	:	
17.1	:	
: i		



INTENTIONALLY LEFT BLANK.

# LIST OF FIGURES

<b>Figure</b>		Page
1.	Schematic of Experimental Apparatus	4
2.	Simple Energy Level Diagram for Atomic Hydrogen. Schemes 1–5 Depict Various Laser-Based Multiphoton Processes for Excitation and Ionization of the H Atom	6
3.	Excitation Wavelength Scans for Microplasma Formation Near 243 nm With the Laser Beam Focused (ILE~1.2 mJ) Into Cold Flows of: (a) $H_2$ , $\lambda$ obs. = 656.3 nm and (b) $D_2$ , $\lambda$ obs. = 656.1 nm. The Wavelength Separation Between Peaks is 11 cm <sup>-1</sup> Which Corresponds to 22 cm <sup>-1</sup> at the Two-Photon Excitation Enerry Which Is Identical With the H-D Deuterium Isotope Shift	6
4.	Time Resolved Emission From (a), Scattered Laser Radiation and (b) H-Atom Emission at 656.3 nm From Micropiasmas Formed in H <sub>2</sub> Flows, Using Scheme 1. Figure 2, !LE = 1.2mJ	8
5.	Excitation Wavelength Scans for Microplasma Formation Near 243 nm Plotted as a Function of ILE: (a) 0.15 mJ, (b) 0.43 mJ, (c) 0.73 mJ, and (d) 1.4 mJ in at 300 K in an Atmospheric Pressure Flow	8
6.	Excitation Spectra of H <sup>+</sup> and D <sup>+</sup> Taken in a Time-of-Flight Mass Spectrometer Using Resonance Two-Photon Excitation and Three-Photon Ionization (2+1) of H and D Atoms Near 243 nm. The Wavelength Separation Between Peaks is 11 cm <sup>-1</sup> Which Corresponds to 22 cm <sup>-1</sup> at the Two-Photon Excitation Energy Which is Identical With the H-D Deutenum Isotope Shift	10
7.	ILE Necessary to Ignite Premixed Flows of: (a) H <sub>2</sub> /O <sub>2</sub> and (b) D <sub>2</sub> /O <sub>2</sub> , as a Function of Excitation Wavelength Near 243 nm. A Shift of +11 cm <sup>-1</sup> of Ignition Curve b Relative to Ignition Curve a Is Evident	12
8.	Dependence of the ILE Required to Ignite Premixed Flows of: (a) H <sub>2</sub> /O <sub>2</sub> and (b) D <sub>2</sub> /O <sub>2</sub> , as a Function of Equivalence Ratio. The Laser Was Sc <sup>2</sup> at the Peak of the Two-Photon Excitation Wavelength of: (a) H Atoms 27 243.07 and (b) D Atoms at 243.00	13

INTENTIONALLY LEFT BLANK.

## **ACKNOWLEDGMENTS**

This work was supported in part by the Air Force Office of Scientific Research (AFOSR), Directorate of Aerospace Sciences, Contract Numbers 88-0013, 89-0017 and in-house laboratory research funding (ILIR).

INTENTIONALLY LEFT BLANK.

#### 1. INTRODUCTION

The use of lasers to initiate combustion events in reactive gaseous mixtures has been the subject of recent investigations in many laboratories (Forch and Miziolek 1986, 1987; Syage et al. 1988; Lavid and Stevens 1985; Raffel, Warntaz, and Wolfrum 1985). Laser ignition has generally been achieved through photochemical initiation (combustion initiation through chainbranching chemical reactions), thermal heating (laser heating of gases), nonresonant spark formation (which results from gas breakdown by intense laser radiation), and through the resonant multiphoton photochemical formation of microplasmas. Laser-induced gas breakdown, which is defined as the point in which single ionization of the gas occurs, results from the absorption of many photons which leads to ionization, collision-induced cascade ionization, and spark formation. It has been shown, however, that the laser energy at the threshold of nonresonant gas breakdown is typically far in excess of the requisite minimum ignition energy such that the formation of a blast wave may lead to a detonation (Weinberg and Wilson 1971). Here, nonresonant gas breakdown refers to spark formation in a gas which is initially transparent (early on in the laser pulse) at a particular laser wavelength. At sufficiently high field strength and as the laser pulse increases in time, electrons can be generated from multiphoton ionization of the parent fuel and/or oxidizer molecule. Electrons can then absorb radiation from the intense laser field, ionize other gas molecules, and eventually lead to cascade ionization and gas breakdown. It naturally follows that the laser energy at the sharp gas breakdown threshold may indeed exceed the minimum energy required to ignite the gas mixture and thus the extraction of a minimum ignition energy from this experiment may be difficult. However, if the laser energy, threshold for the photoproduction of free electrons can be reduced (and thus electrons are produced early on in the laser pulse) then laser spark ignition may be a more controllable process. The efficient formation and characterization of laser produced sparks (microplasmas) as an ignition source is the subject of this paper.

The resonant multiphoton photochemical formation of microplasmas, which is an ignition means that we have been investigating at the Ballistic Research Laboratory, appears to be a more controllable means to generate laser-produced sparks than gas breakdown. Recent investigations (Forch and Miziolek 1986, 1987) have revealed the first example of a strong wavelength dependence in the amount of incident laser energy which was required to ignite a

premixed flow of H<sub>2</sub>/O<sub>2</sub> at atmospheric pressure. A tunable ultraviolet (UV) laser system which operates near 225.6 nm was found to induce photodissociation of the oxidizer component, O<sub>2</sub> or N<sub>2</sub>O, to produce oxygen (O) atoms in three ground-electronic spin-orbit split states 2p<sup>4</sup> <sup>3</sup>P<sub>2,1,0</sub>. It was found that the minima in a piot of incident laser energy (ILE) required for ignition vs. wavelength were located exactly at the same spectral positions as the oxygen atom two-photon allowed absorption transitions from the 2p<sup>4</sup> <sup>3</sup>P<sub>2,1,0</sub> states to the lowest excited state of the same symmetry (Alden et al. 1982; Miziolek and DeWilde 1984; Meier, Kohse-Hoinghaus, and Just 1986; Goldsmith 1983; Dagdigian, Forch, and Miziolek 1988). Subsequent detailed experimental investigations (Forch and Miziolek 1986, 1987) resulted in the formulation of a mechanism for this process which consists of three components: (1) the multiphoton photochemical formation of oxygen atoms; (2) resonant multiphoton ionization of these atoms to efficiently form free electrons in the laser focal volume early in the laser pulse; (3) and the controlled, resonance-enhanced formation of a microplasma using seed electrons which were generated in the previous process.

This ignition method appears to alleviate the problems associated with the sharp and uncontrolled ignition thresholds which are encountered in the extraction of minimum ignition energy measurements using desirable short-pulse lasers (10° sec) (where energy release occurs within a very short time, over a very small volume in free space and is not associated with catalytic and intrusive effects of electrode surfaces). Therefore, we began a detailed experimental investigation of the potential photochemical interaction of the UV laser and the fuel components of these premixed flows. In this paper, we describe a sensitive wavelength dependence on the laser energy required to ignite a premixed gaseous flow of H<sub>2</sub>/O<sub>2</sub> and D<sub>2</sub>O<sub>2</sub> through resonant multiphoton excitation of hydrogen (H) and deuterium (D) atoms near 243 nm. We show that there is a definitive excitation wavelength shift near 243 nm (11 cm<sup>-1</sup> at the single photon wavelength, 243 nm) for the resonant formation of microplasmas that corresponds exactly to H-D deuterium isotope shift of 22 cm<sup>-1</sup> at the two-photon excitation energy (two photons at 143 nm). (D is excited at a wavelength 11 cm<sup>-1</sup> to the blue of H and the 2S level of D is 22 cm<sup>-1</sup> higher in energy than the 2S level of H). Plots of ILE vs. equivalence ratio,  $\Phi$ , with the excitation wavelength of the laser tuned to either the H or Datom two-photon transition, shows a minimum at  $\Phi = 0.7$  in the fuel lean region at an ILE of ~0.55 mJ. Characterization of the photochemistry involved in microplasma formation was

made. A pressure threshold for microplasma formation was determined, and an estimate of the laser power dependence for the ionization process was made.

#### 2. EXPERIMENTAL

A schematic of the experimental apparatus that was used in this investigation is given in Figure 1. Tunable laser radiation near 243 nm was generated by using the second harmonic (532 nm) of a Q-switched Quanta-Ray Nd:YAG laser (DCR-2A,) to pump a Quanta-Ray Dye laser (PDL-1) which was operated at ~580 nm with a DCM dye. The dye laser beam was then frequency doubled using an angle-cut KDP crystal in the first stage of a servo-motor based tracking system, Quanta-Ray WEX-1; then the doubled radiation was frequency mixed with the 1.06l fundamental from the Nd:YAG in an angle-cut KDP crystal which was contained in the second-stage, WEX-1 module. The 243 nm laser radiation was separated from unwanted beams with two Pellin-Broca prisms (which were positioned to avoid net beam-steering), a broad-band pass filter and aperture. This configuration yielded 7-nsec (10 Hz), ~2.0 cm<sup>-1</sup> band width FWHM, 1.5-mJ tunable laser pulses.

The atmospheric pressure burner system consists of a water-cooled and argon-shrouded nozzie with a pinhole aperture (0.6 mm) that was mounted onto an x-y-z translation stage. Gas flows were controlled with Matheson (Model 629) flow meters or Tylan mass flow controllers (Model FC-280-V), which were calibrated by a GCA Precision Scientific wet test meter for  $H_2$ ,  $D_2$  and  $O_2$  flows up to 2 L/min. Linear flow velocities were in the  $10^3$  cm/sec range. The incident laser energy was measured just before a 50-mm focal length lens (the beam waist at the focus was about 50  $\mu$ ) with a Scientech (Model 38-0103) disc calorimeter-power-energy meter. The output laser energy could be varied by insertion of one (or more) dielectrically-coated partially transmitting filters in the beam and/or attenuation of the Nd:YAG amplifier stage gain. It was found that the beam characteristics (such as spatial and temporal profiles) were unaltered using this method.

Laser ignition measurements were made by flowing a carefully metered, homogeneous mixture of fuel and oxidizer through the burner. The beam was precisely focused ~0.5 mm above the pinhole orifice, then the laser energy was adjusted until a single pulse ignited the gases and a flame stabilized on the burner surface. The flame was immediately extinguished

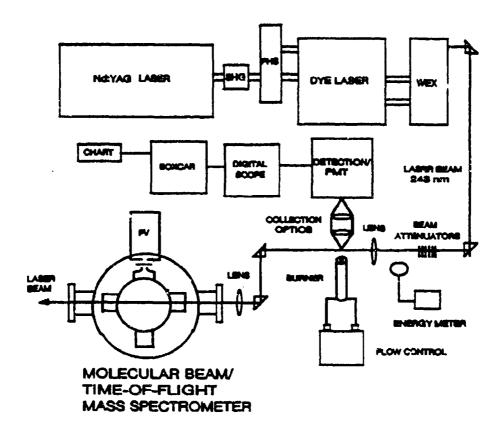


Figure 1. Schematic of Experimental Apparatus.

(fuel and oxidizer valves were closed) in order to maintain a constant burner temperature (as not to preheat the gases in succeeding measurements and to conserve the gases, especially deuterium) then the incident laser energy was measured. It was found that the measured values of ILE for ignition were independent of lateral positioning across the burner aperture (entrainment of room air in the Ai-shrouded, premixed gas jet was not occurring). Wavelength dependent ignition plots near 243 nm were generated by holding U constant and measuring ignition ILE as a function of excitation wavelength.

Excitation scans of the laser produced microplasma emission (from H or D atoms) (in cold gaseous flows or a variable pressure cell) were recorded by collecting emission (with a pair of lenses that were matched to the f/number of a 0.22-m McPherson model 180 monochromator-Hamamatsu R928 photomultiplier system). Emission signals were captured on a 500-MHz Hewlett-Packard model 54111D digital oscilloscope and/or Stanford Research Systems boxcar integrator and strip chart recorder.

Ion signals were detected in a R.M. Jordan time-of-flight mass spectrometer or using a platinum-tipped optogalvanic probe which was biased to detected free electrons liberated in the ionization process. The TOF system consists of a pulsed-molecular Learn valve, a skimmed differentially pumped laser ionization region, a 1.3-m drift tube and microchannel plate detector. Operation pressure is ~10<sup>-7</sup> Torr. The optogalvanic probe was used in atmospheric pressure gas samples. All gases were Matheson UHP and were used "as is" without further purification.

### 3. RESULTS AND DISCUSSION

3.1 Microplasma Formation. Figure 2 illustrates the single laser spectroscopic schemes which have been utilized for the detection of H atoms in flames, molecular beams, discharges, and other environments (Lucht et al. 1983; Alden et al. 1984; Goldsmith 1982; Goldsmith 1984; Tjossem and Cool 1983; Forch, Morris, and Miziolek 1990). The energy gap between the 1S ground electronic state and the 2S lowest excited electronic state (near-degenerate 2S<sub>121</sub> 2P<sub>121</sub> 2P<sub>32</sub> levels) is ~10 eV (Figure 2, scheme 1). Since tunable laser sources which operate in this regime are not readily available, two-photon absorption of laser radiation at 243.07 is required to excite the 1S-2S transition. Absorption of an additional third photon from the resonantly excited 2S state is sufficient to ionize an H atom since the three-photon excitation energy (3 x 5.1 eV) exceeds the 13.6 eV H atom ionization potential. (Note that absorption of an additional photon by an H atom in any one of the excited state electronic manifolds given in Figure 2, schemes 1-5, are sufficient to achieve ionization.) Detection of laser-induced fluorescence from resonance excitation of the n=2 level in flames and other environments is difficult since the emission at 122 nm is absorbed by flame and room gases. Therefore, we investigated the possibility of inducing multiphoton photolysis of H<sub>2</sub> in a cold atmospheric flow, followed by resonance enhanced (2 + 1) multiphoton ionization H atoms to form a laser-produced microplasma. Within the high temperature plasma environment, there exist collisional processes which may populate (for example) the n=3 or 4 levels where Balmer- $\beta$  (486.1 nm) or Balmer- $\alpha$  (656.3 nm) emissions could be detected (monitored by setting the monochromator at the wavelength of interest). Figure 3a gives an excitation wavelength scan where Balmer-β emission was monitored. The signal maximum is found precisely at the known two-photon resonance transition of atomic hydrogen, although the spectral width is much larger than that of a purely atomic transition. This observation is

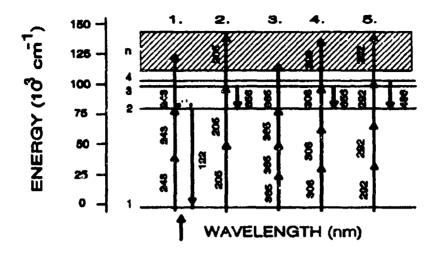


Figure 2. Simple Energy Level Diagram for Atomic Hydrogen. Schemes 1–5 Depict Various Laser-Based Multiphoton Processes for Excitation and Ionization of the H Atom.

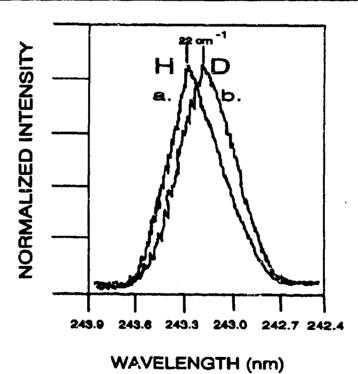


Figure 3. Excitation Wavelength Scans for Microplasma Formation Near 243 nm With the Laser Beam Focused (ILE~1.2 mJ) Into Cold Flows of: (a) H<sub>2</sub>, λ obs. = 656.3 nm and (b) D<sub>2</sub>, λ obs. = 656.1 nm. The Wavelength Separation Between Peaks Is 11 cm<sup>-1</sup> Which Corresponds to 22 cm<sup>-1</sup> at the Two-photon Excitation Energy Which Is Identical With the H-D Deuterium Isotope Shift.

clearly indicative of a much more complex, highly nonlinear phenomenon, i.e., the formation of a microplasma. Multiphoton absorption and ionization generates free electrons (priming or seed electrons) early in the laser pulse which initiate cascade ionization and the formation of a plasma which is heated up to a very high temperature by the inverse brehmsstrahlung effect.

The temporal profile of the emission, Figure 4b, is also indicative of microplasma formation. The lifetime of the emission (~40 ns) is much greater than that of the time-resolved scattered laser radiation (~7 nsec) (Figure 4a) or, for example, from resonance multiphoton laser induced fluorescence from H-atoms that we have detected using scheme 5, where the emission signal follows the temporal profile of the exciting laser pulse (Figure 4a). Although microplasma formation is initiated by atomic ionization, the lifetime of the emission is purely a property of the microplasma.

Figure 5a–d, gives the emission profiles as a function of four different laser pulse energies (0.15–1.5 mJ). The widths of the spectral profiles both increases and shows a linear dependence on laser pulse energy. Extrapolation of the measured linewidths to zero laser energy give a FWHM of ~6 cm<sup>-1</sup>, which is a factor of about six times greater than the measured spectral bandwidth of the laser at this wavelength. It is known, for example, that at pressures near atmospheric the radiation intensities required to cause breakdown by the cascade processes are sufficiently strong that optically induced Stark shifts (which is a linear function of laser energy) and broadening of the electronic levels of an atom occur and have a marked influence on transition probabilities (Tjossem and Cool 1983; Lambropoulos 1974). The signals certainly are severely AC Stark broadened (9 cm<sup>-1</sup>/GW cm<sup>2</sup>), however the widths are much greater than the expected natural linewidth of the atomic transition (<<1cni<sup>-1</sup>). This evidence further indicates that emission we observe is not simply an atomic property, but that of a much more complex event.

To further verify the importance of resonance enhancement in multiphoton photochemical formation of a microplasma, these experiments were repeated under identical conditions except that  $H_2$  was replaced with  $D_2$ . Figure 3b gives the normalized spectral profile of emission signal which was observed. Note that spectra (a) and (b) are essentially identical except that the emission signal in (b) reaches a maximum at a wavelength position which is  $\sim 11$  cm<sup>-1</sup> to the blue of (a) (the spacing at the two-photon energy is  $\sim 22$  cm<sup>-1</sup>). The

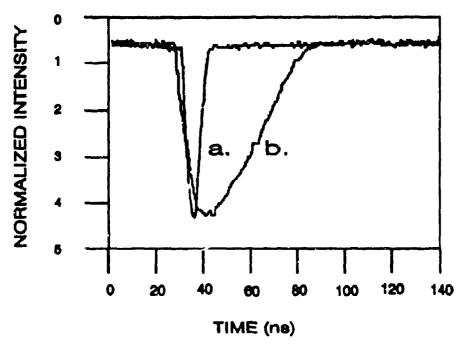


Figure 4. Time Resolved Emission From: (a) Scattered Laser Radiation and (b) H-Atom Emission at 656.3 nm From Microplasmas Formed in H<sub>2</sub> Flows, Using Scheme 1. Figure 2, ILE = 1.2mJ.

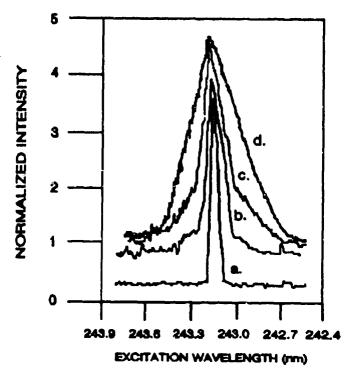


Figure 5. Excitation Wavelength Scans for Microplasma Formation Near 243 nm Plotted as a Function of ILE: (a) 0.15 mJ, (b) 0.43 mJ, (c) 0.73 mJ, and (d) 1.4 mJ in at 300 K in an Atmospheric Pressure Flow.

observation wavelength-bandpass of the monochromator was broadened and centered to equally catch both emissions at 656.3 nm and 656.1 nm for H and D, respectively. A simple calculation and check with a known value 19 shows that this separation is nearly identical with the H-D isotope shift (the D 2S level is 22.4 cm<sup>-1</sup> higher in energy than the H 2S level). This isotopic shift is indeed identical to that which has been observed in spectroscopic probes of H and D atoms in flames and other environments. These results strongly support our contention that microplasma formation is the result of initial photolysis of H<sub>2</sub> or D<sub>2</sub> to produce ground electronic state H and D atoms and subsequent multiphoton excitation and ionization are important in the efficient wavelength dependent formation of these microplasmas, just as had been observed previously for oxidizers, O<sub>2</sub> and N<sub>2</sub>O (Forch and Miziolek 1986, 1987). The photoproduction of free electrons early on in the temporal evolution of the laser pulse from resonant multiphoton ionization of H or D atoms is responsible for the efficient microplasma formation. When the laser is tuned off the resonance absorption wavelengths of either H or D atoms, then no microplasma formation is seen under these experimental conditions.

3.2 Laser Power Dependencies. In order to gain information on the laser power dependence for the photolysis and mulitphoton ionization processes, we performed a series of experiments in a well-controlled low-pressure (10<sup>-7</sup> Torr) environment using a molecular-beam time-of-flight mass spectrometer (Figure 1). Curiously, we were unable to detect any H\* or D\* from laser irradiation of molecular beams of H<sub>2</sub> or D<sub>2</sub> when the laser beam was focused into the ionization region of the spectrometer (with the tightest focusing (200-mm f.l. lens) that neometrical constraints would allow). We did, however, find a pressule threshold of ~70 Torr, ILE=1.2 mJ, for the onset of microplasma formation in a variable pressure flow cell. Comparison of these results suggests that a collisionally induced photodissociation process may be responsible for the production of ground state H and D atoms and subsequent microplasma formation, and, is currently under investigation. We were able to generate intense pulsed and effusive (ground state) H and D atom beams with a hot wire tungsten filament inserted before the skin med differentially pumped ionization region in the TOF-M 20. No ions were produced from the hot wire filament itself. Ions were detected only when the laser was tuned on resonance with H and D atoms. Excitation spectra (Figure 6) were recorded by mass gating either the H\* or D\* signals with a boxcar integrator, then scanning the laser through the two-photon resonance, three photon ionization transitions (2+1) (note

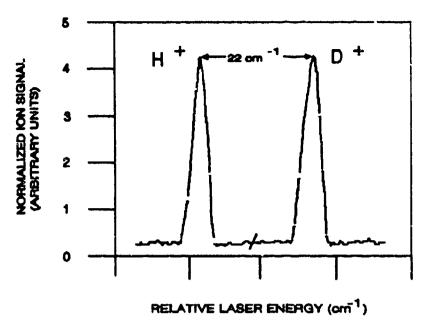


Figure 6. Excitation Spectra of H<sup>+</sup> and D<sup>+</sup> Taken in a Time-of-Flight Mass Spectrometer Using Resonance Two-Photon Excitation and Three-Photon Ionization (2+1) of H and D Atoms Near 243 nm. The Wavelength Separation Between Peaks Is 11 cm<sup>-1</sup> Which Corresponds to 22 cm<sup>-1</sup> at the Two-Photon Excitation Energy Which Is Identical With the H-D Deuterium Isotope Shift.

that there was virtually no signal off resonance of either excitation wavelength). Again, a wavelength spacing of 11 cm<sup>-1</sup> is evident, which corresponds to a deuterium isotope shift of 22 cm<sup>-1</sup> at the two-photon energy). Laser power dependencies for ion signal production were made assuming a signal dependence on laser intensity of I<sup>n</sup>, where n equals the number of photons absorbed. Values of n=2.75+/-0.24 and 2.63+/-0.36 were obtained for H<sup>+</sup> and D<sup>+</sup> production, respectively, which corresponds to two-photon resonance excitation and one photon ionization at low iaser energies (<0.5 mJ). These results (*e.s.* expected) indicate that the overall excitation and ionization was a three-photon process. The noninteger values are indicative of partial saturation of the initial two-photon absorption and ionization step. We could easily fully saturate the ionization process at higher laser energies (>8.4 GW/cm<sup>2</sup>/cm<sup>-1</sup>) and record laser energy dependencies of ~2 for H<sup>+</sup> and D<sup>+</sup>. Although the dissociation energy of H<sub>2</sub> is ~4.5 eV and the energy of a photon at 243 nm is ~5.2 eV, absorption of a single photon is insufficient for the photoproduction of an H atom from H<sub>2</sub> because of negligible single-photon absorption at this wavelength (the ionization potentials of H<sub>2</sub> and O<sub>2</sub> are 12.06 and 15.43, respectively). Laser power dependence measurements for the photolysis of cold

flows of H<sub>2</sub> at atmospheric pressure were made using the optogalvanic probe. The laser beam was focused ~0.5 mm from the 1-mm-diameter anode of the probe, which was inserted in the flowing gas, H<sub>2</sub>. Photolysis of H<sub>2</sub> produces ground state H atoms which are photoexcited and photoionized. The liberated electrons are then detected with the optogalvanic probe. A laser power dependence measurement gave a value of n=4.54+/-0.3. This result also suggests a two-photon dissociation of H<sub>2</sub> to produce H atoms. This reasoning is consistent with a two-photon dissociation of H<sub>2</sub> to produce ground state atoms, two-photon resonance excitation and a one photon ionization. These results, when compared to the time-of-flight data and variable pressure cell data, strongly suggest a collisional induced photodissociation process may occur.

This spectroscopic data gives valuable information on the laser power dependence for microplasma formation. The overall laser energy dependence for the microplasma formation must then be at least a five photon process and clearly is highly nonlinear. The absorption of two laser photons are required for the photoproduction of ground state atoms and the absorption of three additional laser photons are necessary for photoionization. In comparison, photoproduction of atomic oxygen from molecular oxygen photodissociation near 225.5 nm gave a two-photon dependence even though the single photon energy would be sufficient for photolysis (overall a five photon process) (Forch and Miziolek 1987).

3.3 Ignition Experiments. We then began a series of experiments on premixed flows of  $H_2/O_2$  and  $D_2/O_2$  at atmospheric pressure in order to investigate the possibility of using these microplasmas as ignition sources. Previous work ( orch and Miziolek 1986, 1987; Syage et al. 1988; Lewis von Elbe 1951) has shown that the minimum ignition energy for these reactive gaseous mixtures occurs in the fuel lean region ( $\Phi = 0.7$ –.8) and is characteristic of light, diffusive fuels such as  $H_2$ , which can replenish burned fuel early on in the evolution of the ignition kernel as it expands into a stabilized flame front. In our recent work on excimer laser (ArF, 193 nm) ignition of  $H_2/O_2$  we also found that the most efficient ignition was in the fuel lean region (Forch et al. 1989). In two separate, but identical experiments, premixed flows of either  $H_2/O_2$  or  $D_2/O_2$  which were held constant at  $\Phi = 0.7$  were produced. The laser wavelength was varied and the amount of ILE required to ignite either flow into a stabilized combustion was measured. Figures 7a and 7b give the wavelength dependence on the amount of ILE necessary to ignite premixed flows of H2/O2 and  $D_2/O_2$ , respectively. The

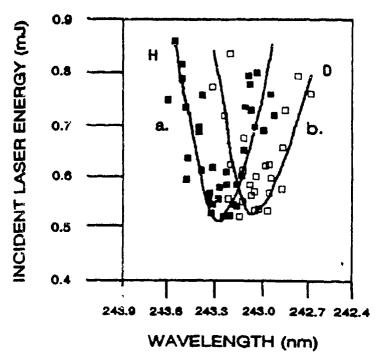


Figure 7. <u>ILE Necessary to Ignite Premixed Flows of: (a) H<sub>2</sub>/O<sub>2</sub> and (b) D<sub>2</sub>/O<sub>2</sub>, as a Function of Excitation Wavelength Near 243 nm. A Shift of +11 cm<sup>-1</sup> of Ignition Curve b Relative to Ignition Curve a Is Evident.</u>

ignition curves clearly show a strong dependence of the ILE on the laser wavelength. Two prominent minima are evident. A simple comparison with the microplasma data (as described above) show that they correspond exactly to the spectral locations of the H and D atom twophoton resonance excitation wavelengths near 243 nm. Apparently, the focused UV laser near 243 nm not only photodissociates H<sub>2</sub> or D<sub>2</sub> to yield ground state atoms, but also, when on resonance with H or D, requires the least amount of laser energy to ignite the gases into a stabilized combustion. We believe that the ignition of these reactive gases occurs through the resonant formation of a microplasma from multiphoton excitation of H and D atoms. The wavelength shift of +11 cm-1 of the  $D_2/O_2$  ignition curve relative to that of the  $H_2/O_2$  curve is clearly related to the H-D deuterium isotope shift and underscores the importance of resonance enhancement in the ignition event. We next measured the dependence of the ILE (at a fixed laser wavelength) on Φ. Figure 8a and b gives plots of incident laser energy required for ignition of premixed flows of H<sub>2</sub>/O<sub>2</sub> and D<sub>2</sub>/O<sub>2</sub> vs.  $\Phi$  when the laser is tuned to the two-photon resonance exitation wavelengths for H and D atoms, respectively. The observed minima in both curves occurs in the fuel lean region ( $\Phi = 0.7$ ) and viith ILE=0.53+/-0.07 mJ and 0.52+/-0.05 for  $H_2/O_2$  and  $D_2/O_2$ , respectively. These experiments support the three step mechanism which we have proposed which consists of the laser photoproduction of ground

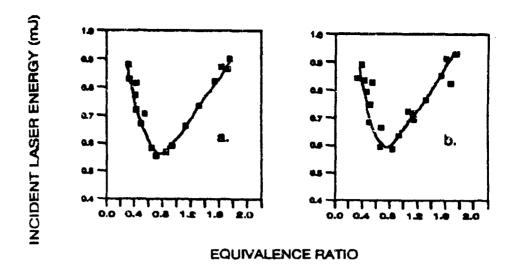


Figure 8. Dependence of the ILE Required to Ignite Premixed Flows of: (a) H<sub>2</sub>/O<sub>2</sub> and (b) D<sub>2</sub>/O<sub>2</sub>, as a Function of Equivalence Ratio. The Laser Was Set at the Peak of the Two-Photon Excitation Wavelength of: (a) H Atoms at 243.07 and (b) D Atoms at 243.00.

state atoms, resonant multiphoton ionization of the atoms to liberate free electrons early on in the laser pulse and the formation of a laser produced microplasma which serves as an ignition kernel. We have also found that microplasmas could be formed very easily through resonance multiphoton ionization, at very low laser energy, that were not intense enough to cause ignition. Thus, our method allows for precise control of laser energy in the ignition process. By comparison, we have investigated ignition of these gas mixtures using the fundamental wavelength of the same Nd:YAG laser (1064 nm) or its second harmonic (532 nm) under identical experimental conditions and found that: 1) a factor of greater than forty times more laser energy was required for ignition; 2) ignition curve plots of ILE vs.  $\Phi$ were flat (independent of  $\Phi$ ; 3) a relatively intense spark was produced. This behavior is consistent with our mechanism and reasoning. The 1064 nm and 532 nm beams are essentially initially transparent to the reactive gaseous mixture (there is no absorption of the laser light). As the laser pulse grows in intensity with time, the laser field strength reaches a point where multiphoton ionization of the fuel and/or oxidizer liberates free electrons in the laser focal volume and a very intense laser spark is produced (through nonresonant gas breakdown). Here the laser energy greatly exceeds the requisite minimum ignition energy.

Visual examination of the laser ignition sparks produced from resonant multiphoton ionization and nonresonant gas breakdown are differ dramatically the former being much less intense than the latter. This is only a qualitiative observation at present, but we have purchased a high-speed streak camera to follow the temporal evolution of the laser-produced microplasmas into an ignition kernel and stabilized combustion and also perform more detailed spectroscopic examination of the sparks.) In our experiments, the incident laser energy, which is required for ignition, has been measured. Attempts to measure the energy absorbed by the microplasmas (through time-resolved absorption measurements) have been very difficult. The microplasma is heated to a very high temperature and, of course, expands with time wherein radiation losses could be significant. Furthermore, scattering of laser radiation by the microplasma occurs and has been calculated to be about 10% (Syage et al. 1988). The microplasmas are formed during the laser pulse (ca. / nsec) through a highly nonlinear process (at least a five photon dependence). Thus, it was very difficult to obtain reliable absorption data although recently we have successfully performed time-resolved absorption measurements on C2H2/O2 and C2H2/air ignition experiments. A rough stimate of the absorbed laser energy (making these corrections) is about 0.12 mJ at  $\Phi = 0.7$  for  $H_2/O_2$  and  $D_2/O_2$ .

### 4. CONCLUSION

We have observed a strong wavelength dependence on the ignition of  $H_2/O_2$  and  $D_2/O_2$  premixed flows using a tunable laser near 243 nm. Furthermore, we have observed a wavelength shift in both microplasma formation in  $H_2$  and  $D_2$  gases and in the ignition of  $H_2/O_2$  and  $D_2/O_2$  flows. These results underscore the importance of two-photon atomic resonances in the microplasma formation process. We have demonstrated that our recent observations of a new laser ignition phenomenon that involves resonant multiphoton photochemical formation of microplasmas appears to be more general and applies to fuel molecules ( $H_2$  and  $D_2$ ) as well all exidizers ( $O_2$  and  $N_2O$ ). We believe that this is the first report of a sensitive wavelength dependence on the laser energy required to ignite these mixtures through resonant multiphoton excitation of H and D atoms (produced from  $H_2$  and  $D_2$  photolysis) and the first report of a deuterium isotope effect in laser ignition. Measurement of the ILE required for ignition vs. equivalence ratio ( $\Phi$ ) shows that the most efficient ignition occurred with 0.55 mJ iLE at  $\Phi = 0.7$  in the fuel lean region. Strong experimental evidence is given which shows that ignition occurs through the resonant formation of a laser-produced

microplasma in a well-defined volume. These new experimental results indicate that resonance enhancement in the formation of a microplasma is a well controlled ignition method which appears to alleviate the problems associated with the sharp thresholds encountered in the well-known laser-produced spark (gas breakdown) process. Currently, we are exploring other possible resonance effects for the purpose of activating or enhancing the combustion of other reactive systems.

INTENTIONALLY LEFT BLANK.

#### 5. REFERENCES

- Alden, M., H. Edner, P. Grafstrom, and S. Svanberg. Optics Communications. Vol. 42, p. 244, 1982.
- Alden, M., A. L. Schawlow, A. L. Svanberg, S. Wende, and P. L. Zhang. Optics Letters. Vol. 9, p. 211, 1984.
- Dagdigian, P. J., B. E. Forch, and A. W. Miziolek. <u>Chemical Physics Letters</u> Vol. 148, p. 299, 1988.
- Downey, S. W., and R. S. Hozack. Optics Letters. Vol. 14, p. 15, 1989.
- Forch, B. E., and A. W. Miziolek. Optics Letters. Vol. 11, p. 129, 1986.
- Forch, B. E., and A. W. Miziolek. <u>Combustion Science and Technology</u>. Vol. 52, p. 151, 1987.
- Forch, B. E., A. W. Miziolek, C. N. Merrow, J. B. Morris, and R. J. Locke. <u>Proceedings of the 1989 JANNAF Combustion Meeting</u>, The Chemical Propulsion Information Agency. Vol. 4, p. 113, October 1989.
- Forch, B. E., J. B. Morris, and A. W. Miziolek. <u>Laser-Based Approaches in Luminescence Spectroscopy</u>, invited book chapter, "Laser-Induced Fluorescence and Ionization Techniques in Combustion Diagnostics." Edited by T. Vo Dinh and D. Eastwood, STP 1066, p. 50, Philadelphia, 1990.
- Goldsmith, J. E. M. Optics Letters. Vol. 7, p. 437, 1982.
- Goldsmith, J. E. M. Journal of Chemical Physics. Vol. 78, p. 1610, 1983.
- Goldsmith, J. E. M. <u>Twentieth Symposium (International) on Combustion</u>, p. 1331, The Combustion Institute, 1984.
- Lambropoulos, P. Physics Review. Vol. 9, p. 1992, 1974.
- Lavid, M., and J. G. Stevens. Combustion and Flame. Vol. 60, p. 195, 1985.
- Lewis, B., and G. von Elbe. <u>Combustion, Flames and Explosions of Gases</u>. p. 390, Academic Press, Inc., 1951.
- Lucht, R. P., J. T. Salomon, G. B. King, D. W. Sweeney, and N. M. Laurendeau. Optics Letters. Vol. 8, p. 365, 1983.
- Meier, U., Th. Kohse-Hoinghaus and Just. Chemical Physics Letters. Vol. 126, p. 567, 1986.
- Miziolek, A. W., and M. A. DeWilde. Optics Letters. Vol. 9, p. 330, 1984.

- Raffel, B., B. Warntaz, and J. Wolfrum. Journal of Applied Physics. Vol. B37, p. 344, 1985.
- Striganov, A. R., and N. S. Sventitskii. <u>Tables of Spectral Lines of Neutral and Ionized Atoms.</u> p. 75, IFI/Plenum, 1978.
- Syage, J. A., E. W. Fournier, R. Rianda, and R. B. Cohen. <u>Journal of Applied Physics</u> Vol. 64, p. 1499, 1988.
- Tjossem, P. J. H., and T. A. Cool. Chemical Physics Letters. Vol. 100, p. 479, 1983.
- Weinberg, F. J., and J. R. Wilson. <u>Proceedings of Royal Society of London Series</u>. Vol. A321, p. 41, 1971.

No. of		No. of	
	Organization		Organization
			<del> </del>
2	Administrator	1	Commander
	Defense Technical Info Center		U.S. Army Missile Command
	ATTN: DTIC-DDA		ATTN: AMSMI-RD-CS-R (DOC)
	Cameron Station Alexandria, VA 22304-6145		Redstone Arsenal, AL 35898-5010
	Alexandria, VA 22304-0143	1	Commander
1	Commander	•	U.S. Army Tank-Automotive Command
•	U.S. Army Materiel Command		ATTN: ASQNC-TAC-DIT (Technical
	ATTN: ÁMCAM		Information Center)
	5001 Eisenhower Avenue		Warren, MI 48397-5000
	Alexandria, VA 22333-0001		
	Commendan	1	Director
1	Commander		U.S. Army TRADOC Analysis Command
	U.S. Army Laboratory Command ATTN: AMSLC-DL		ATTN: ATRC-WSR
	2800 Powder Mill Road		White Sands Missile Range, NM 88002-5502
	Adelphi, MD 20783-1145	1	Commandant
	• •		U.S. Army Field Artillery School
2	Commander		ATTN: ATSF-CSI
	U.S. Army Armament Research,		Ft. Sill, OK 73503-5000
	Development, and Engineering Center	( <del>************************************</del>	
	ATTN: SMCAR-IMI-I	(Class. only)1	Commandant
	Picatinny Arsenal, NJ 07806-5000		U.S. Army Infantry School ATTN: ATSH-CD (Security Mgr.)
2	Commander		Fort Benning, GA 31905-5660
	U.S. Army Armament Research,		, a 10 10 00 0000
	Development, and Engineering Center	(Unclass, only)1	Commandant
	ATTN: SMCAR-TDC		U.S. Army Infantry School
	Picatinny Arsenal, NJ 07806-5000		ATTN: A)SH-CD-CSO-OR
1	Director		Fort Berning, GA 31905-5660
•	Benet Weapons Laboratory	1	Air Force Armament Laboratory
	U.S. Army Armament Research,	•	ATTN: WL/MNOI
	Development, and Engineering Center		Eglin AFB, FL 32542-5000
	ATTN: SMCAR-CCB-TL		
	Watervliet, NY 12139-4050		Aberdeen Proving Ground
(Unclass, only)1	Commondo	•	D'. HOAMOAA
(Oncomes. Chay)	Commander U.S. Army Armament, Munitions	2	Dir, USAMSAA ATI'N: AMXSY-D
	and Chemical Command		AMXSY-MP, H. Cohen
	ATTN: AMSMC-IMF-L		AMAGE IN , II. COREII
	Rock Island, IL 61299-5000	1	Cdr, USATECOM
			ATTN: AMSTE-TC
	Director	_	
	U.S. Army Aviation Research	3	Cdr, CRDEC, AMCCOM
	and Technology Activity ATTN: SAVRT-R (Library)		ATTN: SMCCR-RSP-A
	M/S 219-3		SMCCR-MU SMCCR-MSI
	Ames Research Center		
	Moffett Field, CA 94035-1000	1	Dir, VLAMO
			ATTN: AMSLC-VL-D
		غاد خان	B: 201
		10	Dir, BRL
			ATTN: SLCBR-DD-T

### No. of Copies

### Organization

- 1 HQDA (SARD-TR, C.H. Church) WASH DC 20310-0103
- 4 Commander US Army Research Office

ATTN: R. Ghirardelli

D. Mann

R. Singleton

R. Shaw

P.O. Box 12211

Research Triangle Park, NC

27709-2211

2 Commander

US Army Armament Research,

Development, and Engineering Center

ATTN: SMCAR-AEE-B, D.S. Downs SMCAR-AEE, J.A. Lannon

Picatinny Arsenal, NJ 07806-5000

1 Commander

US Army Armament Research,

Development, and Engineering Center

ATTN: SMCAR-AEE-BR, L. Harris Picatinny Arsenal, NJ 07806-5000

2 Commander

US Army Missile Command

ATTN: AMSMI-RD-PR-E, A.R. Maykut

AMSMI-RD-PR-P, R. Betts

Redstone Arsenal, AL. 35898-5249

1 Office of Naval Research

Department of the Navy

ATTN: R.S. Miller, Code 432

800 N. Quincy Street

Arlington, VA 22217

Commander

Naval Air Systems Command

ATTN: J. Ramnarace, AJR-54111C

Washington, DC 20360

1 Commander

Naval Surface Warfare Center

ATTN:

J.L. East, Jr., G-23

Dahlgren, VA 22448-5000

No. of Copies

Organization

2 Commander

Naval Surface Warfare Center

ATTN: R. Bernecker, R-13

G.B. Wilmot, R-16

Silver Spring, MD 20903-5000

5 Commander

Naval Research Laboratory

ATTN:

M.C. Lin

J. McDonald

E. Oran

J. Shnur

R.J. Doyle, Code 6110

Washington, DC 20375

1 Commanding Officer

Naval Underwater Systems

Center Weapons Dept.

ATTN: R.S. Lazar/Code 36301

Newport, RI 02840

2 Commander

Naval Weapons Center

ATTN: T. Boggs, Code 388

T. Parr. Code 3895

China Lake, CA 93555-6001

1 Superintendent

Naval Postgraduate School

Dept. of Aeronautics

ATTN: D.W. Netzer

Monterey, CA 93940

3 AL/LSCF

ATTN: R. Corley

R. Geisler

J. Levine

Edwards AFB, CA 93523-5000

1 AL/MKPB

ATTN: B. Goshgarian

Edwards AFB, CA 93523-5000

1 AFOSR

ATTN:

J.M. Tishkoff

**Bolling Air Force Base** 

Washington, DC 20332

1 OSD/SDIO/IST

ATTN: L. Caveny

Pentagon

Washington, DC 20301-7100

# No. of <u>Organization</u>

- 1 Commandant USAFAS ATTN: ATSF-TSM-CN Fort Sill, OK 73503-5600
- 1 F.J. Seiier ATTN: S.A. Shackleford USAF Academy, CO 80840-6528
- University of Dayton Research Institute ATTN: D. Campbell AL/PAP Edwards AFB, CA 93523
- NASA

   Langley Research Center
   Langley Station
   ATTN: G.B. Northam/MS 168
   Hampton, VA 23365
- 4 National Bureau of Standards
  ATTN: J. Hastie
  M. Jacox
  T. Kashiwagi
  H. Semerjian
  US Department of Commerce
  Washington, DC 20234
- Aerojet Solid Propulsion Co.
   ATTN: P. Micheli
   Sacramento, GA 95813
- 1 Applied Combustion Technology, Inc. ATTN: A.M. Varney P.O. Box 607885 Orlando, FL 32860
- 2 Applied Mechanics Reviews The American Society of Mechanical Engineers ATTN: R.E. White A.B. Wenzel 345 E. 47th Street New York, NY 10017
- 1 Atlantic Research Corp. ATTN: M.K. King 5390 Cherokee Avenue Alexandria, VA 22314

# No. of Copies Organization

- 1 Atlantic Research Corp. ATTN: R.H.W. Waesche 7511 Wellington Road Gainesville, VA 22065
- 1 AVCO Everett Research Laboratory Division ATTN: D. Stickler 2385 Revere Beach Parkway Everett, MA 02149
- 1 Battelle
  ATTN: TACTEC Library, J. Huggins
  505 King Avenue
  Columbus, OH 43201-2693
- Cohen Professional Services
   ATTN: N.S. Cohen
   141 Channing Street
   Redlands, CA 92373
- Exxon Research & Eng. Co. ATTN: A. Dean Route 22E Annandale, NJ 08801
- 1 Ford Aerospace and Communications Corp. DIVAD Division Div. Hq., Irvine ATTN: D. Williams Main Street & Ford Road Newport Beach, CA 92663
- General Applied Science

   Laboratories, Inc.

   77 Raynor Avenue

   Ronkonkama, NY
   11779-6649
- 1 General Electric Ordnance Systems ATTN: J. Mandzy 100 Plastics Avenue Pittsfield, MA 01203
- 1 General Motors Rsch Labs Physical Chemistry Department ATTN: T. Sloane Warren, MI 48090-9055

### No. of Copies

### Organization

- 2 Hercules, Inc.
  Allegheny Ballistics Lab.
  ATTN: W.B. Walkup
  E.A. Yount
  P.O. Box 210
  Rocket Center, WV 26726
- 1 Aliant Techsystems, Inc.
  Marine Systems Group
  ATTN: D.E. Broden/
  MS MN5C-2000
  600 2nd Street NE
  Hopkins, MN 55343
- 1 Alliant Techsystems, Inc. ATTN: R.E. Tompkins MN38-3300 5700 Smetana Drive Minnetonka, MN 55343
- 1 IBM Corporation ATTN: A.C. Tam Research Division 5600 Cottle Road San Jose, CA 95193
- 1 IIT Research Institute ATTN: R.F. Remaly 10 West 35th Street Chicago, IL 60616
- 2 Director
  Lawrence Livermore
  National Laboratory
  ATTN: C. Westbrook
  M. Costantino
  P.O. Box 808
  Livermore, CA 94550
- Lockheed Missiles & Space Co. ATTN: George Lo
   3251 Hanover Street
   Dept. 52-35/B204/2
   Palo Alto, CA 94304
- 1 Director
  Los Alames National Lab
  ATTN: B. Nichols, T7, MS-B284
  P.O. Box 1663
  Los Alamos, NM 87545

### No. of Copies

### Organization

- National Science Foundation ATTN: A.B. Harvey Washington, DC 20550
- 1 Olin Ordnance ATTN: V. McDonald, Library P.O. Box 222 St. Marks, FL 32355-0222
- 1 Paul Gough Associates, Inc. ATTN: P.S. Gough 1048 South Street Portsmouth, NH 03801-5423
- 2 Princeton Combustion Research Laboratories, Inc. ATTN: M. Summerfield N.A. Messina 475 US Highway One Monmouth Junction, NJ 08852
- Hughes Aircraft Company ATTN: T.E. Ward 8433 Fallbrook Avenue Canoga Park, CA 91303
- Rockwell International Corp.
   Rocketdyne Division
   ATTN: J.E. Flanagan/HB02
   5633 Canoga Avenue
   Canoga Park, CA 91304
- 4 Director
  Sandia National Laboratories
  Division 8354
  ATTN: R. Cattolica
  S. Johnston
  P. Mattern
  D. Stephenson
  Livermore, CA 94550
- Science Applications, Inc. ATTN: R.B. Edelman
   23146 Cumorah Crest
   Woodland Hills, CA 91364
- 3 SRI International
  ATTN: G. Smith
  D. Crosley
  D. Golden
  333 Ravenswood Avenue
  Menlo Park, CA 94025

No. of	No. of
Copies Organization	Copies Organization
Copies Organization	Other Other Control
1 Stevens institute of Tech.	1 Brigham Young University
Davidson Laboratory	Dept. of Chemical Engineering
ATTN: R. McAlevy, III	ATTN: M.W. Beckstead
Hoboken, NJ 07030	Provo, UT 84058
1 Sverdrup Technology, Inc.	1 California Institute of Tech.
LERC Group	Jet Propulsion Laboratory
ATTN: R.J. Locke, MS SVR-2	ATTN: L. Strand/MS 512/102
2001 Aerospace Parkway	4800 Oak Grove Drive
Brook Park, OH 44142	Pasadena, CA 91109
1 Sverdrup Technology, Inc.	1 California Institute of
ATTN: J. Deur	Technology
2001 Aerospace Parkway	ATTN: F.E.C. Culick/
Brook Park, OH 44142	MC 301-46
	204 Karman Lab.
1 Thiokol Corporation	Pasadena, CA 91125
Elkton Division	
ATTN: S.F. Palopoli	<ol> <li>University of California</li> </ol>
P.O. Box 241	Los Alamos Scientific Lab.
Elkton, MD 21921	P.O. Box 1663, Mail Stop B216
	Los Alamos, NM 87545
3 Thiokol Corporation	
Wasatch Division	1 University of California,
ATTN: S.J. Bennett	Berkeley
P.O. Box 524	Chemistry Deparment
Brigham City, UT 84302	ATTN: C. Bradley Moore
	211 Lewis Hall
1 United Technologies Research Center	Berkeley, CA 94720
ATTN: A.C. Eckbreth	
East Hartford, CT 06108	1 University of California, San Diego
3 United Technologies Corp.	ATTN: F.A. Williams
United Technologies Corp.     Chemical Systems Division	AMES, B010
•	
ATTN: R.S. Brown T.D. Myers (2 copies)	La Jolla, CA 92093
P.O. Box 49028	2 University of California,
San Jose, CA 95161-9028	Santa Barbara
Jan 1036, UK 33101-3020	Quantum Institute
1 Universal Propulsion Company	ATTN: K Schofield

- Universal Propulsion Company ATTN: H.J. McSpadden Black Canyon Stage 1 Box 1140 Phoenix, AZ 85029
- Veritay Technology, Inc. ATTN: E.B. Fisher 4845 Millersport Highway P.O. Box 305 East Amherst, NY 14051-0305

- ATTN: K. Schofield M. Steinberg Santa Barbara, CA 93106
- University of Colorado at Boulder **Engineering Center** ATTN: J. Daily Campus Box 427 Boulder, CO 80309-0427

## Organization

- 2 University of Southern California Dept. of Chemistry ATTN: S. Benson: C. Wittig Los Angeles, CA 90007
- 1 Cornell University
  Department of Chemistry
  ATTN: T.A. Cool
  Baker Laboratory
  Ithaca, NY 14853
- 1 University of Delaware ATTN: T. Brill Chemistry Department Newark, DE 19711
- 1 University of Florida Dept. of Chemistry ATTN: J. Winefordner Gainesville, FL 32611
- 3 Georgia Institute of Technology School of Aerospace Engineering ATTN: E. Price W.C. Strahle B.T. Zinn Atlanta, GA 30332
- University of Illinois
   Dept. of Mech. Eng.
   ATTN: H. Krier
   144MEB, 1206 W. Green St.
   Urbana, IL 61801
- 1 Johns Hopkins University/APL Chemical Proputsion Information Agency ATTN: T.W. Christian Johns Hopkins Road Laurel, MD 20707
- University of Michigan
   Gas Dynamics Lab
   Aerospace Engineering Bldg.
   ATTN: G.M. Faeth
   Ann Arbor, MI 48109-2140

# No. of Copies

### Organization

- University of Minnesota
   Dept. of Mechanical
   Engineering
   ATTN: E. Fletcher
   Minneapolis, MN 55455
- 3 Pennsylvania State University
  Applied Research Laboratory
  ATTN: K.K. Kuo
  H. Palmer
  M. Micci
  University Park, PA 16802
- Pennsylvania State University
   Dept. of Mechanical Engineering
   ATTN: V. Yang
   University Park, PA 16802
- 1 Polytechnic Institute of NY Graduate Center ATTN: S. Lederman "oute 110 Farmingdale, NY 11735
- 2 Princeton Universit, Forrestal Campus Library ATTN: K. Brezinsky I. Giassman P.O. Box 710 Princeton, NJ 08540
- Purdue University
   School of Aeronautics
   and Astronautics
   ATTN: J.R. Osborn
   Grissom Hall
   West Lafayette, IN 47906
- Purdue University
   Department of Chemistry
   ATTN: E. Grant
   West Lafavette, IN 47906
- 2 Purdue University
  School of Mechanical
  Engineering
  ATTN: N.M. Laurendeau
  S.N.B. Murthy
  TSPC Chaifee Hall
  West Lafayette, IN 47906

## No. of Copies

### Organization

- Rensselaer Polytechnic Inst.
   Dept. of Chemical Engineering ATTN: A. Fontijn Troy, NY 12181
- 1 Stanford University
  Dept. of Mechanical
  Engineering
  ATTN: R. Hanson
  Stanford, CA 94305
- 1 University of Texas
  Dept. of Chemistry
  ATTN: W. Gardiner
  Austin, TX 78712
- University of Utah
   Dept. of Chemical Engineering
   ATTN: G. Flandro
   Salt Lake City, UT 84112
- Virginia Polytechnic Institute and State University ATTN: J.A. Schetz Blacksburg, VA 24061
- Freedman Associates
   ATTN: E. Freedman
   2411 Diana Road
   Baltimore, MD 21209-1525