

AD-A237 880



OFFICE OF NAVAL RESEARCH

Grant # N0001489J1848

R&T Code 413u002

DTIC
ELECTE
JULI 0 1991
S C D

Technical Report No. 4

Infrared and Raman Spectra of C_{60} and C_{70}
Solid Films at Room Temperature

by

R.J. Meilunas, R.P.H. Chang,
S. Liu, M. M. Kappes

Prepared for publication

in the

Applied Physics Letters

Northwestern University
Department of Materials
Science and Engineering
and Dept. of Chemistry
Evanston, IL 60208

July 1, 1991

Reproduction in whole or in part is permitted for any
purpose of the United States Government

This document has been approved for public release and sale;
its distribution is unlimited

91-04309



91 7 8 010

REPORT DOCUMENTATION PAGE				Form Approved OMB No. 0704-0188	
1a REPORT SECURITY CLASSIFICATION Unclassified		1b RESTRICTIVE MARKINGS			
2a SECURITY CLASSIFICATION AUTHORITY		3 DISTRIBUTION/AVAILABILITY OF REPORT Approved for public release; distribution unlimited			
2b DECLASSIFICATION/DOWNGRADING SCHEDULE					
4 PERFORMING ORGANIZATION REPORT NUMBER(S) Technical Report number 4		5 MONITORING ORGANIZATION REPORT NUMBER(S)			
6a NAME OF PERFORMING ORGANIZATION Northwestern University R.P.H. Chang		6b OFFICE SYMBOL (if applicable)	7a. NAME OF MONITORING ORGANIZATION Office of Naval Research		
6c. ADDRESS (City, State, and ZIP Code) Materials Science & Engineering Dept. Evanston, IL 60208		7b ADDRESS (City, State, and ZIP Code) Chemistry Division 800 N. Quincy Street Arlington, VA 22217			
8a. NAME OF FUNDING / SPONSORING ORGANIZATION Office of Naval Research		8b. OFFICE SYMBOL (if applicable)	9 PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER Grant # N0001489J1848		
8c. ADDRESS (City, State, and ZIP Code) Chemistry Division 800 N. Quincy Street Arlington, VA 22217		10 SOURCE OF FUNDING NUMBERS			
		PROGRAM ELEMENT NO	PROJECT NO	TASK NO	WORK UNIT ACCESSION NO
11 TITLE (Include Security Classification) Infrared and Raman Spectra of C ₆₀ and C ₇₀ Solid Films at Room Temperature					
12 PERSONAL AUTHOR(S) R.J. Meilunas, R.P.H. Chang, S. Liu, M.M. Kappes					
13a. TYPE OF REPORT Technical		13b TIME COVERED FROM _____ TO _____		14 DATE OF REPORT (Year, Month, Day) July 1, 1991	15 PAGE COUNT 10
16 SUPPLEMENTARY NOTATION					
17 COSATI CODES			18 SUBJECT TERMS (Continue on reverse if necessary and identify by block number)		
FIELD	GROUP	SUB-GROUP			
19 ABSTRACT (Continue on reverse if necessary and identify by block number) Infrared and high resolution Raman spectroscopic probes of thin films of C ₆₀ and C ₇₀ are presented and discussed in terms of previous measurements, semi-empirical calculations and plausible molecular geometries.					
20 DISTRIBUTION/AVAILABILITY OF ABSTRACT <input checked="" type="checkbox"/> UNCLASSIFIED/UNLIMITED <input type="checkbox"/> SAME AS RPT <input type="checkbox"/> DTIC USERS			21 ABSTRACT SECURITY CLASSIFICATION Unclassified		
22a NAME OF RESPONSIBLE INDIVIDUAL Dr. Mark Ross			22b TELEPHONE (Include Area Code) (703)696-4410	22c OFFICE SYMBOL	

Infrared and Raman Spectra of C₆₀ and C₇₀ Solid Films at Room Temperature

Ray Meilunas and R.P.H. Chang, Department of Materials Science and Engineering, Northwestern University, Evanston, IL 60208

Shengzhong Liu and Manfred M. Kappes, Department of Chemistry, Northwestern University, Evanston, IL 60208

Submitted for publication: May 16, 1991

Abstract:

Infra-red and high resolution Raman spectroscopic probes of thin films of C₆₀ and C₇₀ are presented and discussed in terms of previous measurements, semi-empirical calculations and plausible molecular geometries.

APPROVED FOR
DISSEMINATION
BY THE
DIRECTOR AND
INSTITUTIONAL
REPRESENTATIVE

by _____
Date _____
Availability Code _____
Avail number _____
Dist _____ Special _____

A-1

QUALITY INSPECTED

The development of simple methods for the gram scale synthesis of air stable hollow core all-carbon clusters (fullerenes)¹ has accessed a new class of solid materials whose properties are expected to hold many useful surprises. For example, preliminary studies on alkali doped C₆₀ have shown resulting materials to be molecular superconductors with transition temperatures in the 18-30 K range². Pure fullerenes are presently being studied in applications ranging from lubrication³ to the nucleation of diamond thin films⁴. An understanding of the molecular level processes accessed in such studies requires structural determination, as well as sensitive and selective quantitative analysis. Both can be achieved via vibrational spectroscopy.

First FT-IR absorption^{1,5-7} and Raman measurements^{8,9} have established that room temperature solid films of C₆₀ and C₇₀ manifest characteristic vibrational spectra differing strongly from those of other carbon modifications. The high symmetry of the C₆₀ truncated icosahedron (see figure 1) gives rise to only 4 IR and 10 Raman active modes¹⁰⁻¹². Their tentative assignment in IR and Raman measurements of C₆₀ thin films has provided strong evidence for the proposed molecular structure⁹. Interestingly, the associated Raman depolarization ratios suggest effectively random molecular orientation at room temperature - consistent with a plastic r.t. phase inferred from solid state ¹³C-NMR¹³. The supposed lower symmetry molecular structure of C₇₀¹⁴ would be expected to lead to much more complicated vibrational spectra^{15,16} and in fact 21 Raman and 16 IR features have so far been reported for this cluster⁹.

In this letter we present Raman and IR measurements on C₆₀ and C₇₀ thin films. Raman data were obtained at significantly greater resolution (9x) than in previous studies. Results encompass a confirmation of C₆₀ IR and Raman spectral assignment⁹ as well as the identification of multiple previously unresolved C₇₀ vibrational features, which are compared to computational prediction. Detailed analysis of the underlying vibrational dynamics at various film temperatures is ongoing and will be reported elsewhere. This study is part of an investigation into the vibrational properties of carbon species involved in the nucleation and growth of diamond thin films⁴.

C₆₀ and C₇₀ were obtained from carbon soot, prepared by graphite evaporation in a helium atmosphere^{1,17}. Fullerenes were solvent extracted and chromatographically separated¹⁸. Purity of resulting C₆₀ and C₇₀ fractions was confirmed by electron impact mass spectroscopy (see below). Thin films were generated by subliming the respective purified solids under high vacuum onto Si(100) or KBr substrates. C₆₀/C₇₀ were sublimed at 450/550^o C, respectively. At lower evaporator temperatures, the substrate was kept shuttered. Raman spectra of fullerene films on silicon substrates were taken in air at room temperature in a backscattering geometry using either the 514 nm or 488 nm lines of an Ar⁺ laser (20 mW). No significant excitation frequency dependence was observed. Both polarized and unpolarized

spectra were obtained. Instrument resolution was 1 cm^{-1} with absolute error $\pm 1\text{ cm}^{-1}$. IR measurements were performed on KBr substrates using an N_2 -purged Mattson FTIR at 2 cm^{-1} resolution.

Unpolarized Raman spectra of C_{60} and C_{70} are shown in Figures 1a and 1b, respectively and compared to semi-empirical computation¹⁵. Tables I and II contain peak positions, intensities and symmetry assignments. Also listed are depolarization ratios for bands having sufficient signal-to-noise to yield meaningful numbers. Figures 2a and 2b show the corresponding IR absorption spectra again compared to calculation. IR spectral features are also listed in Tables I and II.

(i) C_{60} . IR and Raman spectra are in good agreement with previous work⁹. Using more recent semi-empirical calculations for $I_h\text{C}_{60}$ vibrational fundamentals¹², our data confirm the proposed spectral assignment, including 1100 and 1250 cm^{-1} features which had only been tentatively identified before. All IR active (4) and Raman active (10) modes are now clearly resolved. Among several additional (weaker) IR features, contaminant lines at 805 , 1257 , 1539 cm^{-1} (IR) are not tabulated⁹. Remaining minor tabulated features reflect combination bands or matrix effects.

(ii) C_{70} . The proposed D_{5h} structure (see figure 1) encompasses 31 IR and 53 Raman active modes. IR absorption studies have so far resolved 16 features⁹. We confirm 13 of these. At the higher resolution and sensitivities accessed, 21 previously identified Raman lines give way to a total of 33 features, some with discernible (and likely further separable!) shoulders. While similarities are apparent between experiment and D_{5h} semi-empirical prediction¹⁵ for both IR and Raman spectra, assignment must await: (i) more highly resolved measurements on thin films of $^{13}\text{C}_{70}$ and $^{12}\text{C}_{70}$ at various temperatures and (ii) calculations of Raman cross sections. In the interim, Table II gives tentative vibrational symmetry assignments for a number of lines based on calculations for the D_{5h} structure¹⁵.

The isolation of C_{70}O , $\text{C}_{76}/\text{C}_{78}$, C_{84} and C_{92} from fullerene preparations has recently been reported¹⁹. The underlying "statistical synthesis" from graphite likely forms traces of even larger all carbon clusters. Their isolation and mass spectral characterization is ongoing⁸. It is likely that such larger species will share with C_{60} and C_{70} the hollow core structural motif comprising 12 five fold rings (to achieve closure) supplemented by hexagons (20 in the case of C_{60}). Given enough carbon atoms, the vibrational spectra of large fullerenes should approach those of (perturbed) microcrystalline graphite²⁰. Systematic Raman studies of these systems may be of great interest in unraveling the size dependent transition from molecular vibrational to bulk phonon behaviour.

This work was supported by NSF-MRL, ONR and DOE. M.K. acknowledges support through the Sloan Foundation fellowship program.

References

1. W. Krätschmer, L. Lamb, K. Fostiropoulos and D. Huffman, *Nature*, 347, 354 (1990).
2. K doping (18 K): A. Hebard, M. Rosseinsky, R. Haddon, D. Murphy, S. Glarum, T. Palstra, A. Ramirez and A. Kortan, *Nature*, 350, 600 (1991); Rb doping (30 K): R. Smalley, private communication to S.L.
3. A. Mackay, *Nature*, 347, 336 (1990).
4. R. Meilunas, S. Liu, M.Kappes and R.P.H. Chang, in preparation.
5. W. Krätschmer, K. Fostiropoulos and D. Huffman, *Chem. Phys. Lett.*, 170, 167 (1990).
6. D. Cox, S. Behal, M. Disko, S. Gorun, M. Greaney, C. Hsu, E. Kollin, J. Millar, J. Robbins, W. Robbins, R. Sherwood and P. Tindall, *J. Am. Chem. Soc.*, 113, 2940 (1991).
7. J. Hare, J. Dennis, H. Kroto, R. Taylor, A. Wahab Allaf, S. Balm and D. Walton, *J. Chem. Soc., Chem. Commun.*, 412 (1991).
8. D. Bethune, G. Meijer, W. Tang and H. Rosen, *Chem. Phys. Lett.*, 174, 219 (1990).
9. D. Bethune, G. Meijer, W. Tang, H. Rosen, W. Golden, H. Seki, C. Brown and M. de Vries, *Chem. Phys. Lett.*, 179, 181 (1991).
10. R. Stanton and M. Newton, *J. Am. Chem. Soc.*, 92, 2141 (1988).
11. W. Harter and D. Weeks, *J. Chem. Phys.*, 90, 4727 (1989).
12. D. Weeks and W. Harter, *J. Chem. Phys.*, 90, 4744 (1989).
13. C. Yannoni, R. Johnson, G. Meijer, D. Bethune and J. Salem, *J. Phys. Chem.*, 95, 9 (1991).
14. R. Johnson, G. Meijer, J. Salem and D. Bethune, *J. Am. Chem. Soc.*, 113, 3619 (1991).
15. D. Bakowies and W. Thiel, *Chem. Phys.*, 151, 309 (1991).
16. Z. Slanina, J. Rudzinski, M. Togasi and E. Osawa, *J. Mol. Struct.*, 202, 169 (1989).
17. R. Whetten, M. Alvarez, S. Anz, K. Schriver, R. Beck, F. Diederich, Y. Rubin, R. Ettl, C. Foote, A. Darmany and J. Arbogast, *Mat. Res. Soc. Symp. Proc.* 206, xxx (1991).
18. H. Aje, M. Alvarez, S. Anz, R. Beck, F. Diederich, K. Fostiropoulos, D. Huffman, W. Krätschmer, Y. Rubin, K. Schriver, D. Sensharma and R. Whetten, *J. Phys. Chem.*, 94, 8630 (1990).
19. F. Diederich, R. Ettl, Y. Rubin, R. Whetten, R. Beck, M. Alvarez, S. Anz, D. Sensharma, F. Wudl, K. Khemani and A. Koch, *Science*, 252, 548 (1991).
20. D. Knight and W. White, *J. Mater. Res.*, 4, 385 (1989).

Tables

I. C₆₀ IR and Raman line positions^a, relative intensities (unpolarized), Raman depolarization ratios and assignments.

Assignment	ν_{IR}	I_{IR}	ν_{RAMAN}	I_{RAMAN}	ρ^b
			267(S)	7	
H _g			273	15	.36
H _g			434	3	.47
			486(S)	3	
A _g			496	13.1	.07
F _{1u}	527	100			
F _{1u}	577	37			
H _g			710	4	.40
H _g			774	4	.46
H _g			1100	1	
F _{1u}	1183	23			
H _g			1250	1	
H _g			1426	13.1	.35
F _{1u}	1428	24			
			1464(S)	29.2	
A _g			1470	100	.08
H _g			1576	8.1	.39

^aPositions in wavenumbers. All features with signal to noise > 2 are listed (see figs. 1/2 and text). Typical relative intensity variations between measurements were $\pm 10\%$. Raman measurements were done on essentially pure C₆₀ and C₇₀ samples. The C₇₀ IR sample was slightly contaminated by C₆₀. ^bDepolarization ratios refer to $I_{\text{perpendicular}}/I_{\text{parallel}}$. Numbers are not tabulated for weak signals or incompletely resolved shoulders (s).

II. C₇₀ vibrational line positions, relative intensities, Raman depolarization ratios and tentative assignments^a.

Assignment (IR/Raman)	ν_{IR}	I_{IR}	ν_{RAMAN}	I_{RAMAN}	ρ
			224(S)	5	
			229	5	
			252(S)	24	
a ₁ '			260	31	.24
			400	7	
			413	17	
			436	3	
			457	10	
	458	16	508	4.3	
			521	5	
	535	47			
/a ₁ '	565	16	572	17	.12
	578	53			
	641	16			
e ₁ '/?	673	35	704	19	.4
			715	12	
e ₂ ' or e ₁ "			740	16	.75
e ₂ ' or e ₁ "			771	7	.7
e ₁ '	795	19			
			1053(S)	7	
a ₁ '			1063	26	.29
a ₂ "/	1133	13	1167	9	
/a ₁ '	1181	7	1187	69	.25
a ₁ '			1232	76	.2
e ₂ ' or e ₁ "			1259	10	.62
			1301	7	
e ₂ ' or e ₁ "			1316	12	.5
/e ₂ ' or e ₁ "	1320	7	1335	14	.54
			1371	24	.67
e ₁ '	1415	26			
e ₁ '/	1429	100	1439(S)	26	
			1443(S)	33	
a ₁ '			1449	71	.23
			1461(S)	26	
	1460	6.9	1463(S)	26	
a ₁ '			1471	52	.21
e ₂ ' or e ₁ "			1515	22	.59
a ₁ '			1569	100	.29

^aSee table I footnotes.

Figure Captions

Figure 1. Unpolarized Raman spectra of purified films of C_{60} and C_{70} on Si(100) substrate - see text and Tables. Superimposed are the calculated Raman active modes [MNDO frequencies scaled to C_{60} measurement using $0.872x$]¹⁵ for both the C_{60} I_h structure and the postulated C_{70} D_{5h} structure.

Figure 2. FTIR spectra of purified films of C_{60} and C_{70} on KBr substrates. Also shown are the semi-empirical computational predictions of ref. 15 upon which we base the symmetry assignments in Tables I and II.

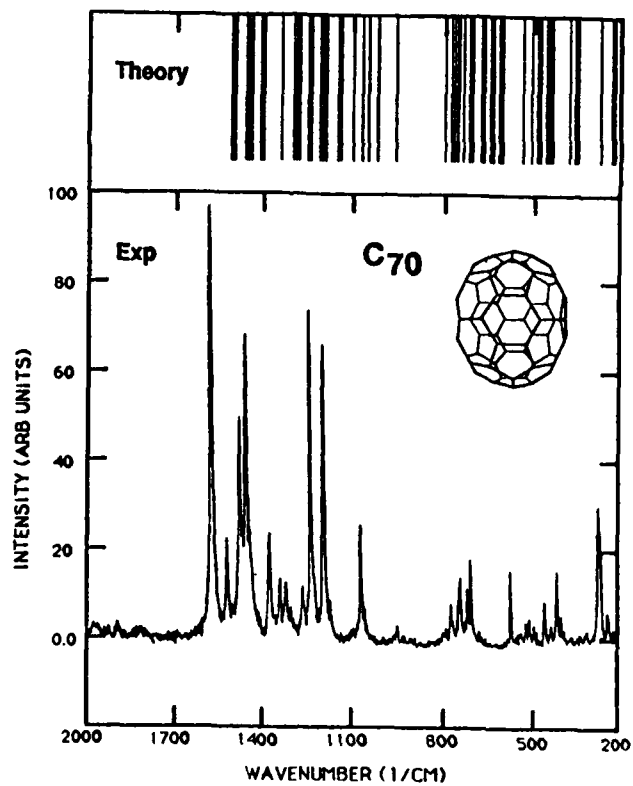
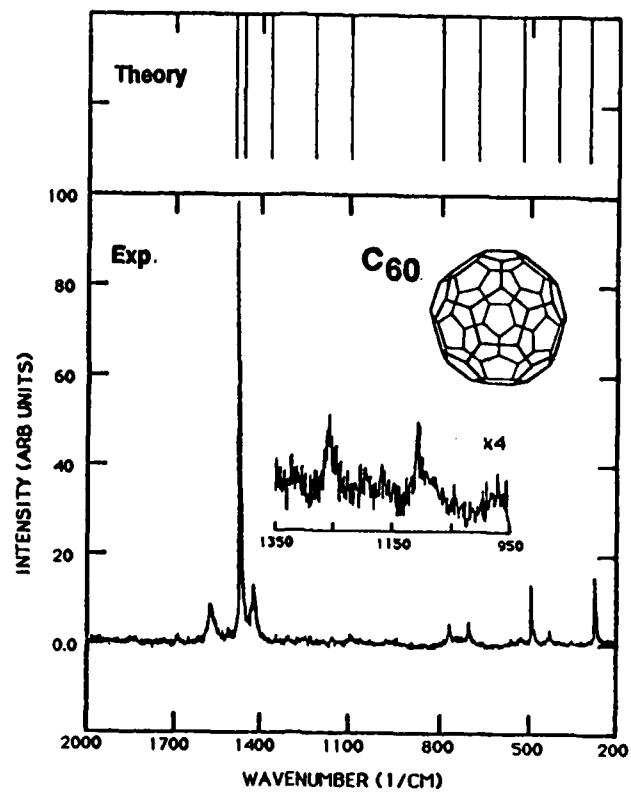


Figure 1

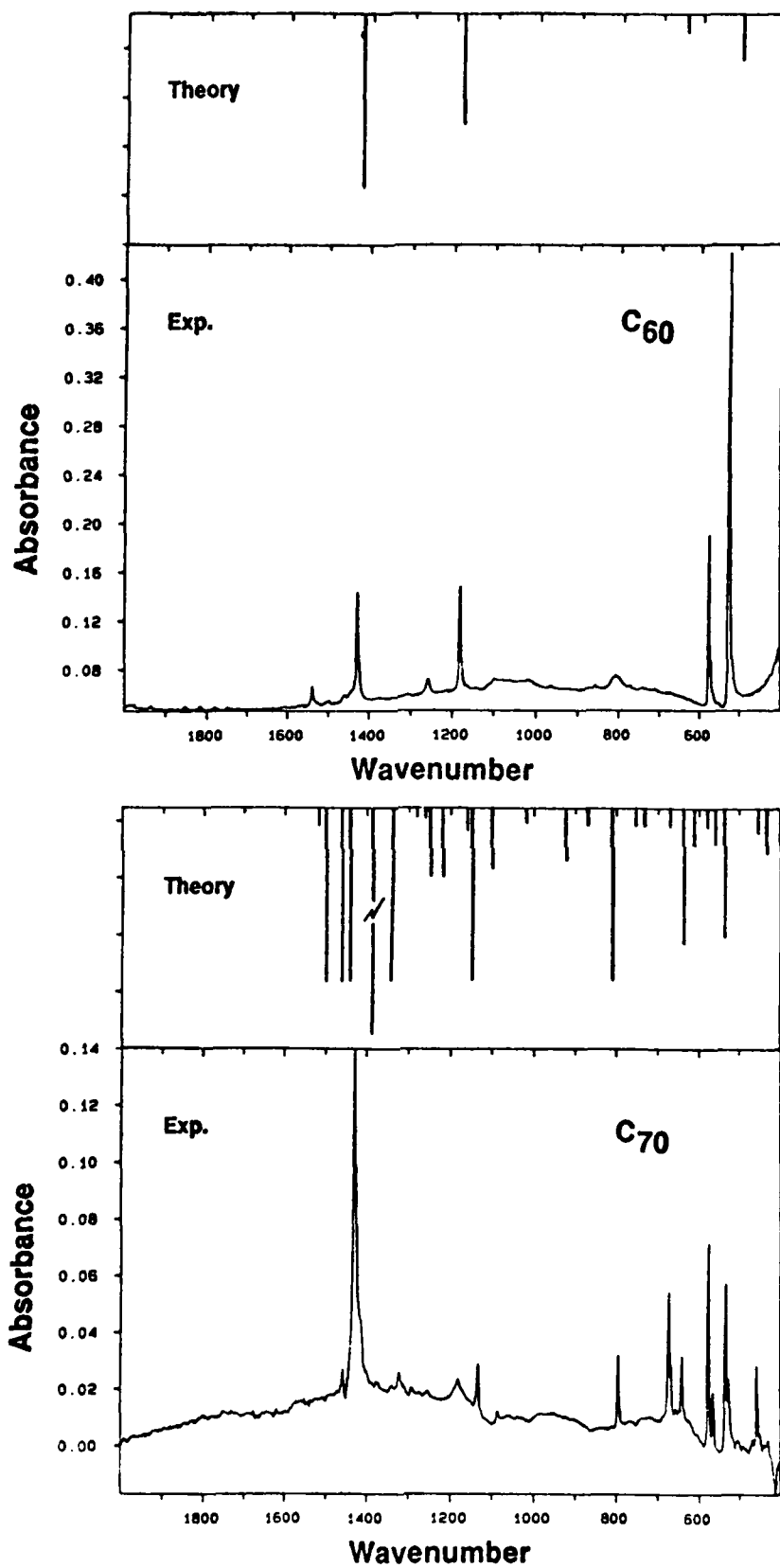


Figure 2

TECHNICAL REPORT DISTRIBUTION LIST - GENERAL

Office of Naval Research (2)
Chemistry Division, Code 1113
800 North Quincy Street
Arlington, Virginia 22217-5000

Dr. Richard W. Drisko (1)
Naval Civil Engineering
Laboratory
Code L52
Port Hueneme, CA 93043

Dr. James S. Murday (1)
Chemistry Division, Code 6100
Naval Research Laboratory
Washington, D.C. 20375-5000

Dr. Harold H. Singerman (1)
David Taylor Research Center
Code 283
Annapolis, MD 21402-5067

Dr. Robert Green, Director (1)
Chemistry Division, Code 385
Naval Weapons Center
China Lake, CA 93555-6001

Chief of Naval Research (1)
Special Assistant for Marine
Corps Matters
Code 00MC
800 North Quincy Street
Arlington, VA 22217-5000

Dr. Eugene C. Fischer (1)
Code 2840
David Taylor Research Center
Annapolis, MD 21402-5067

Defense Technical Information
Center (2)
Building 5, Cameron Station
Alexandria, VA 22314

Dr. Elek Lindner (1)
Naval Ocean Systems Center
Code 52
San Diego, CA 92152-5000

Commanding Officer (1)
Naval Weapons Support Center
Dr. Bernard E. Douda
Crane, Indiana 47522-5050

* Number of copies to forward