

**AD-A229 548**

2

**OFFICE OF NAVAL RESEARCH**

Grant N00014-90-J-1193

**TECHNICAL REPORT No. 36**

**Thermoelectric Phenomena in Metals Under Large Temperature Gradients**

by

**A. N. Grigorenko, P. I. Nikitin, Daniel A. Jelski and Thomas F. George**

Prepared for publication

in

**Journal of Applied Physics (Communications)**

Departments of Chemistry and Physics  
State University of New York at Buffalo  
Buffalo, New York 14260

December 1990

Reproduction in whole or in part is permitted for any purpose of the  
United States Government.

This document has been approved for public release and sale;  
its distribution is unlimited.

**S DTIC  
ELECTE  
DEC 20 1990  
E D**

01 1 2 3 4 5 6 7 8 9

REPORT DOCUMENTATION PAGE

Form Approved  
OMB No. 0704-0188

1a. REPORT SECURITY CLASSIFICATION Unclassified		1b. RESTRICTIVE MARKINGS	
2a. SECURITY CLASSIFICATION AUTHORITY		3. DISTRIBUTION/AVAILABILITY OF REPORT Approved for public release; distribution unlimited	
2b. DECLASSIFICATION/DOWNGRADING SCHEDULE			
4. PERFORMING ORGANIZATION REPORT NUMBER(S) UBUFALLO/DC/90/TR-36		5. MONITORING ORGANIZATION REPORT NUMBER(S)	
6a. NAME OF PERFORMING ORGANIZATION Depts. Chemistry & Physics State University of New York	6b. OFFICE SYMBOL (If applicable)	7a. NAME OF MONITORING ORGANIZATION	
6c. ADDRESS (City, State, and ZIP Code) Fronczak Hall, Amherst Campus Buffalo, New York 14260		7b. ADDRESS (City, State, and ZIP Code) Chemistry Program 800 N. Quincy Street Arlington, Virginia 22217	
8a. NAME OF FUNDING/SPONSORING ORGANIZATION Office of Naval Research	8b. OFFICE SYMBOL (If applicable)	9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER Grant N00014-90-J-1193	
8c. ADDRESS (City, State, and ZIP Code) Chemistry Program 800 N. Quincy Street Arlington, Virginia 22217		10. SOURCE OF FUNDING NUMBERS	
		PROGRAM ELEMENT NO.	PROJECT NO.
		TASK NO.	WORK UNIT ACCESSION NO.
11. TITLE (Include Security Classification) Thermoelectric Phenomena in Metals Under Large Temperature Gradients			
12. PERSONAL AUTHOR(S) A. N. Grigorenko, P. I. Nikitin, Daniel A. Jelski and Thomas F. George			
13a. TYPE OF REPORT	13b. TIME COVERED FROM _____ TO _____	14. DATE OF REPORT (Year, Month, Day)	15. PAGE COUNT 9
16. SUPPLEMENTARY NOTATION Prepared for publication in the <i>Journal of Applied Physics (Communications)</i>			
17. COSATI CODES		18. SUBJECT TERMS (Continue on reverse if necessary and identify by block number)	
FIELD	GROUP	THERMOELECTRIC PHENOMENA	
		METALS	
		LARGE TEMPERATURE GRADIENTS	
		NONLINEAR CONTRIBUTIONS	
		PARABOLIC DISPERSION LAW	
		MEASURABLE EFFECTS.	
19. ABSTRACT (Continue on reverse if necessary and identify by block number) Nonlinear contributions to thermoelectricity are studied when large temperature gradients are present in metals. A theory is presented to account for these phenomena in the case of monovalent metals obeying a parabolic dispersion law. Simple experiments are proposed in which nonlinear terms are relevant and produce measurable effects.			
20. DISTRIBUTION/AVAILABILITY OF ABSTRACT <input checked="" type="checkbox"/> UNCLASSIFIED/UNLIMITED <input checked="" type="checkbox"/> SAME AS RPT <input type="checkbox"/> DTIC USERS		21. ABSTRACT SECURITY CLASSIFICATION Unclassified	
22a. NAME OF RESPONSIBLE INDIVIDUAL Dr. David L. Nelson		22b. TELEPHONE (Include Area Code) (202) 696-4410	22c. OFFICE SYMBOL

THERMOELECTRIC PHENOMENA IN METALS UNDER LARGE TEMPERATURE GRADIENTS

A. N. Grigorenko and P. I. Nikitin  
General Physics Institute, USSR Academy of Sciences  
38 Vavilov Street, 117942 Moscow, USSR

Daniel A. Jelski\*  
Department of Chemistry  
State University of New York, College at Fredonia  
Fredonia, New York 14063

Thomas F. George  
Departments of Chemistry and Physics & Astronomy  
239 Fronczak Hall  
State University of New York at Buffalo  
Buffalo, New York 14260

ABSTRACT

Nonlinear contributions to thermoelectricity are studied when large temperature gradients are present in metals. A theory is presented to account for these phenomena in the case of monovalent metals obeying a parabolic dispersion law. Simple experiments are proposed in which nonlinear terms are relevant and produce measurable effects.

\* Author to whom correspondence should be addressed.

Accession For	
NTIS GRA&I	<input checked="" type="checkbox"/>
DTIC TAB	<input type="checkbox"/>
Unannounced	<input type="checkbox"/>
Justification	
By _____	
Distribution/	
Availability Codes	
Dist	Avail and/or Special
A-1	



The conventional theory of thermoelectric phenomena is based on the concept of local equilibrium and small deviations from global thermal equilibrium. In this case, only linear relations between flows and thermodynamic forces should be considered, and linear nonequilibrium thermodynamics is valid. Such considerations account for the well-known Thomson and Seebeck effects, described in many textbooks.<sup>1,2</sup>

However, the conventional theory of thermoelectricity excludes the so-called Benedicks' effect in metals.<sup>3</sup> This effect is a thermoelectric phenomenon in which a potential is developed between two points at the same temperature, but separated by nonzero temperature gradients as described below. It is much smaller than the Thomson effect, in which a potential is developed between two points at different temperatures, and initial experiments demonstrating Benedicks' effect were therefore inconclusive.<sup>4</sup> Similar observations for semiconductors are less ambiguous.<sup>5</sup>

It has been hitherto assumed that large temperature gradients cannot be produced in metals because of the high thermal diffusivity. But modern achievements in short-pulse laser generation and thin-film technology force a change in this perspective. In recent experiments<sup>6</sup> high voltages were measured when a pulsed laser is used to heat a thin metal film deposited on a grating, creating temperature gradients as high as  $10^6$  K/cm. The complexities of these experiments are beyond the scope of this paper, but suffice it to say that Benedicks' effect is one possible explanation of the observed phenomenon. In this view, the potential is produced by nonlinear thermoelectric effects caused by differential laser heating. This possibility prompts the present work, which is the first theoretical account of nonlinear thermoelectric effects in metals.

Throughout, we assume local thermal equilibrium, without which temperature would be undefined. If we define a dimensionless parameter  $\omega$  as a measure of the deviation from local equilibrium, then for the experiment just described,  $\omega$  can be estimated as  $\omega = \frac{\ell_0 \nabla T}{T} \approx 0.1$ , where  $\ell_0$  is the electron free path. For simplicity, we consider metals with one parabolic conduction band, e.g., for Ag,  $\ell_0 = 10^{-5}$  cm (see Ref. 7, p. 268), and for the value of  $\nabla T \approx 3 \times 10^6$  K/cm, we get  $\omega \approx 0.1$ . Thus  $\omega$  is not necessarily small, and it is no longer sufficient to consider only first-order

terms. On this scale, nonlinear terms are essential and can produce measurable experimental effects. For simplicity, we consider metals with one parabolic conduction band.

It is well known<sup>7</sup> that for a bulk metal with a conductivity  $\sigma$ , linear non-equilibrium thermodynamics yields an electric current  $j$  as

$$j = \sigma(E_{\text{eff}} - \alpha \nabla T) \quad (1)$$

where  $E_{\text{eff}}$  is the effective electric field and  $\alpha$  is the absolute thermoelectric coefficient. The  $E_{\text{eff}}$  can be written as

$$E_{\text{eff}} = E + \nabla(\zeta/e) \quad (2)$$

where  $E$  is the external electric field and  $\zeta$  is the chemical potential. When temperature gradients are sufficiently large, then Eq. (1) is insufficient, and nonlinear terms must be taken into account:

$$j = \sigma(E_{\text{eff}} - \alpha \nabla T - \alpha_1 \nabla(\Delta T) - \alpha_2 \nabla T(\Delta T) - \alpha_3 \nabla(\nabla T)^2 - \alpha_4 (\nabla T)^3 \dots) \quad (3)$$

It should be noted that the concept of local equilibrium is still valid because otherwise a local temperature  $T(x)$  is undefined. This implies that  $\omega < 1$ , as described in above.

Our task is to evaluate the thermoelectric coefficients  $\alpha_i$ , for  $i = 1, 2, \dots$ . For this purpose, we use the conventional theory of electron transport in metals,<sup>8</sup> keeping terms up to third order. Let  $n_k$  be the electron density in  $k$ -state. For a stationary state in the absence of external fields, the Boltzmann equation reduces to

$$(\mathbf{v}_k \cdot \nabla) n_k = \left. \frac{\partial n_k}{\partial t} \right|_{\text{coll}} \quad (4)$$

To simply evaluate the collision term on the rhs, we use the conventional  $\tau$ -approximation<sup>8</sup> and neglect the effect of phonon drag, assuming that the temperature is sufficiently high. By expanding  $n_k$  in a series of temperature gradients, we obtain

$$n_k = n_k^0 + g_k^1 + g_k^2 + g_k^3 + \dots \quad (5)$$

where

$$n_k^0 = \frac{1}{\exp\left(\frac{\epsilon_k - \zeta}{T}\right) + 1} \quad (6)$$

and where  $\epsilon_k$  is the electron energy in the  $k$ -state. Substituting Eq. (5) into Eq. (4) and comparing the terms of same order in both parts of the expression, we get

$$g_k^1 = -\tau(\epsilon_k)(v_k \nabla) n_k^0 \quad (7)$$

$$g_k^2 = -\tau(\epsilon_k)(v_k \nabla) g_k^1 \quad (8)$$

$$g_k^3 = -\tau(\epsilon_k)(v_k \nabla) g_k^2 \quad (9)$$

We then derive an expression for  $j$ , valid for large  $\omega$ , as

$$j = e \sum_k (g_k^1 v_k + g_k^2 v_k + g_k^3 v_k) \quad (10)$$

The first term in Eq. (10) is familiar from linear thermoelectric theory,<sup>7,8</sup> and yields  $j = -\sigma \alpha \nabla T$ . The second term vanishes because of symmetry after integrating over  $k$ -space. The third term is of interest here.

In the simplest case of a spherically-symmetric Fermi surface, the integration over  $k$ -space yields<sup>8</sup>

$$j^i = -e \int \frac{dS d\epsilon_k}{4\pi^3 \hbar v_k} v_k^i (\tau v_k^j \nabla_j) (\tau v_k^l \nabla_l) (\tau v_k^m \nabla_m) n_k^0 \quad (11)$$

At high temperatures, the electron free path is limited by electron-phonon scattering,<sup>7</sup> and  $\tau$  is inversely proportional to  $T$ . Then performing simple, but somewhat cumbersome, calculations, we obtain

$$\alpha_1/T = \alpha_2/3 = \alpha_3/3 = \frac{14\pi^2 k_b^2 \tau(\mu)^2}{5me}, \quad \alpha_4 = 0 \quad (12)$$

where  $\tau(\mu)$  refers to the scattering time at the Fermi energy  $\mu$ ,  $k_b$  is the Boltzmann constant, and where terms to order  $(k_b T/\mu)^2$  are retained.

We now evaluate Benedicks' effect in metals for the following sawtooth geometry. Suppose two points, A and B, are at a temperature  $T_0$ , and are separated by a point C at temperature  $T_1$ . Let the temperature gradient on the segment (AC) be twice the gradient along the segment (CB). This system may be approximated by a triangle-like temperature function, in which case the problem is analytically solvable.

As usually assumed in the stationary state, there is no current, and therefore  $j = 0$ . From Eq. (3) the electric field can be written as

$$E = -\nabla(\zeta/e) + \alpha \nabla T + \alpha_1 \nabla(\Delta T) + \alpha_2 \nabla T(\Delta T) + \alpha_3 \nabla(\nabla T)^2 \quad (13)$$

Then the voltage arising between points A and B is

$$V_{AB} = \int_A^B E \, dx = \int_A^B \left( \frac{d^2 \alpha_1}{2dT^2} - \frac{d}{dT} (\alpha_2 + 2\alpha_3) \right) T_x'^3 \, dx \quad (14)$$

where A and B are assumed to lie along the x-axis separated by distance  $\Lambda$ , and  $T' = dT/dx$ .

From (12) it follows that  $\alpha_1 \sim 1/T$  and  $\alpha_2 \sim 1/T^2$ . Then

$$V_{AB} = \frac{28(\pi k_b \tau_0 T_0)^2}{5me} \int_A^B \frac{T^3}{T^3} dx \approx 40 \alpha_0 \frac{(T_1 - T_0)^3 (T_0 + T_1)}{T_1^2 T_0} \frac{\rho_0^2}{\Lambda^2} \quad (15)$$

where the 0 subscript refers to values at  $T_0$ , and where the evaluation of the integral depends explicitly on the geometry described above. If the gradients along (AC) and (CB) are equal, then the integral vanishes and the effect disappears. If  $T_0 = 300$  K,  $T_1 = 600$  K,  $\Lambda = 10 \mu\text{m}$ ,  $\rho_0 = 10^{-5}$  cm and  $\alpha_0 = 10^{-6}$  V/K, then  $V_{AB} \approx 2 \mu\text{V}$ . This very small potential can be measured as follows. A thin, flat metal film is irradiated by short-pulsed light with spatially modulated intensity, the modulation stretching over  $N = 10^4$  periods, and each period assuming the sawtooth geometry. The modulation must be accomplished by masking, since interfering laser beams will always produce symmetric gradients, and hence no effect. Summing the voltage over  $10^4$  periods yields  $V = NV_{AB} \approx 20$  mV, which can be easily measured by conventional stroboscopic methods.

Experiments with laser irradiation of metal gratings<sup>6,9</sup> provides another interesting example where large temperature gradients arise. In these experiments, thin transition or semi-metal films are evaporated onto gratings and are illuminated by pulsed laser light. Unexpectedly high voltages of about 1 V are measured along the gratings. One possible reason for this phenomenon may be Benedicks' effect caused by a laser-induced, periodic but asymmetric temperature distribution similar to that just described and evaluated for monovalent metals.

Calculations show that for transition and semi-metals, nonlinear contributions to thermoelectricity are considerably larger because of overlapping conduction bands. In this case, an inverted conduction band may be regarded as a trap,<sup>7</sup> and even linear thermoelectric coefficients are one or two orders of magnitude greater than for monovalent metals. The nonlinear terms increase because of the greater length required for inter-band transmission of electrons, in particular, because of longer transmission times from d- to s-zones for transition metals. Detailed calculations of this will be given elsewhere.

We further remark that as a consequence of Eq. (3), we can see that  $j_x$  depends not only on  $\partial T/\partial x$  but also on the Laplacian. Under laser irradiation of metal gratings, in addition to the periodic gradients in the x-direction,  $10^6$  K/cm along



the film, high-temperature gradients arise in the z-direction. The value of such gradients can easily be estimated by  $\partial T/\partial z \approx (T_1 - T_0)/\delta \approx 10^8$  K/cm, where the skin depth  $\delta$  is about  $0.05 \mu\text{m}$  in the present case. For such gradients, the dimensionless parameter  $\omega$  assumes a value greater than 1, and we are clearly in the domain of nonequilibrium thermodynamics. Any description of electron transport in this regime will require a new physical approach. Further, in the 2-dimensional case  $\nabla \times \mathbf{E}$  is not necessarily zero in Eq. (13), and for this reason, the  $j = 0$  condition cannot be assumed. The cold bulk substrate will serve to complete the circuit. Thus a 2-dimensional treatment of thermoelectric phenomena is significantly more complex, and this very interesting problem will be the subject of further discussion.

In conclusion, we have calculated the nonlinear contributions to thermoelectricity arising from large temperature gradients in a monovalent metal. Such contributions play an important role in various transport processes, of which one example is Benedicks' effect. We have evaluated Benedicks' effect, forbidden in the linear theory, but which may explain interesting experimental phenomena. Practical applications could include very small and convenient sensors of laser radiation parameters.<sup>10</sup> This theory can also be extended to double-band metals such as Ni, Ti, and Bi, where the effects could be significantly greater, and also to situations where a magnetic field is present.

#### Acknowledgments

TFG acknowledges support by the Office of Naval Research. Acknowledgment is made by DAJ to the donors of the Petroleum Research Fund, administered by the American Chemical Society, for partial support of this research. DAJ also acknowledges that this research was partially supported by a grant from the Research Corporation. Finally, DAJ thanks the Research Foundation of the State University of New York for an Equipment Matching Grant. All authors would like to thank the Fredonia campus, including the Department of Chemistry and, especially, Andrea Domst, for hospitality shown to the Soviet authors during their visit.

**References**

1. C. Kittel, Introduction to Solid State Physics, 3rd Ed. (Wiley, New York, 1966).
2. N. W. Ashcroft and N. D. Mermin, Solid State Physics (Saunders, Philadelphia, 1976).
3. C. Benedicks, Ergeb. Exakt. Naturwiss. 8, 26 (1929).
4. R. R. Heikes and R. W. Ure, Jr., Thermoelectricity: Science and Engineering (Interscience, New York, 1961), p. 311.
5. J. Tauc, Czech. J. Phys. 6, 108 (1956).
6. V. I. Konov, P. I. Nikitin, D. G. Satjukov and S. A. Uglov, Laser Surface Microprocessing, ed. by V. I. Konov, B. S. Luk'yanchuk and I. W. Boyd, Proc. Soc. Photo-Opt. Instrum. Eng., in press.
7. N. F. Mott and H. Jones, The Theory of the Properties of Metals and Alloys (Oxford, Oxford, 1956), p. 310 ff.
8. J. M. Ziman, Electrons and Phonons (Oxford, Oxford, 1960).
9. V. I. Konov, P. I. Nikitin and S. A. Uglov, (unpublished).
10. V. I. Konov, P. I. Nikitin, D. G. Satjukov and S. A. Uglov, Sensors and Actuators A 22, 498 (1990).

FY90 Abstracts Distribution List for Solid State & Surface Chemistry

Professor John Baldeschwieler  
Department of Chemistry  
California Inst. of Technology  
Pasadena, CA 91125

Professor John Eyler  
Department of Chemistry  
University of Florida  
Gainesville, FL 32611

Dr. Sylvia Johnson  
SRI International  
333 Ravenswood Avenue  
Menlo Park, CA 94025

Professor Paul Barbara  
Department of Chemistry  
University of Minnesota  
Minneapolis, MN 55455-0431

Professor James Garvey  
Department of Chemistry  
State University of New York  
Buffalo, NY 14214

Dr. Zakya Kafafi  
Code 6551  
Naval Research Laboratory  
Washington, DC 20375-5000

Dr. Duncan Brown  
Advanced Technology Materials  
520-D Danury Rd.  
New Milford, CT 06776

Professor Steven George  
Department of Chemistry  
Stanford University  
Stanford, CA 94305

Professor Larry Kesmodel  
Department of Physics  
Indiana University  
Bloomington, IN 47403

Professor Stanley Bruckenstein  
Department of Chemistry  
State University of New York  
Buffalo, NY 14214

Professor Tom George  
Dept. of Chemistry and Physics  
State University of New York  
Buffalo, NY 14260

Professor Max Lagally  
Dept. Metal. & Min. Engineering  
University of Wisconsin  
Madison, WI 53706

Professor Carolyn Cassady  
Department of Chemistry  
Miami University  
Oxford, OH 45056

Dr. Robert Hamers  
IBM T.J. Watson Research Center  
P.O. Box 218  
Yorktown Heights, N Y 10598

Dr. Stephen Lieberman  
Code 522  
Naval Ocean Systems Center  
San Diego, CA 92152

Professor R.P.H. Chang  
Dept. Matls. Sci. & Engineering  
Northwestern University  
Evanston, IL 60208

Professor Charles Harris  
Department of Chemistry  
University of California  
Berkeley, CA 94720

Professor M.C. Lin  
Department of Chemistry  
Emory University  
Atlanta, GA 30322

Professor Frank DiSalvo  
Department of Chemistry  
Cornell University  
Ithaca, NY 14853

Professor John Hemminger  
Department of Chemistry  
University of California  
Irvine, CA 92717

Professor Fred McLafferty  
Department of Chemistry  
Cornell University  
Ithaca, NY 14853-1301

Dr. James Duncan  
Federal Systems Division  
Eastman Kodak Company  
Rochester, NY 14650-2156

Professor Leonard Interrante  
Department of Chemistry  
Rensselaer Polytechnic Institute  
Troy, NY 12181

Professor Horia Metiu  
Department of Chemistry  
University of California  
Santa Barbara, CA 93106

Professor Arthur Ellis  
Department of Chemistry  
University of Wisconsin  
Madison, WI 53706

Professor Roald Hoffmann  
Department of Chemistry  
Cornell University  
Ithaca, NY 14853

Professor Larry Miller  
Department of Chemistry  
University of Minnesota  
Minneapolis, MN 55455-0431

Professor Mustafa El-Sayed  
Department of Chemistry  
University of California  
Los Angeles, CA 90024

Professor Eugene Irene  
Department of Chemistry  
University of North Carolina  
Chapel Hill, NC 27514

Professor George Morrison  
Department of Chemistry  
Cornell University  
Ithaca, NY 14853

Professor Daniel Neumark  
Department of Chemistry  
University of California  
Berkeley, CA 94720

Professor Robert netten  
Department of Chemistry  
University of California  
Los Angeles, CA 90024

Professor David Ramaker  
Department of Chemistry  
George Washington University  
Washington, DC 20052

Professor R. Stanley Williams  
Department of Chemistry  
University of California  
Los Angeles, CA 90024

Dr. Gary Rubloff  
IBM T.J. Watson Research Center  
P.O. Box 218  
Yorktown Heights, NY 10598

Professor Nicholas Winograd  
Department of Chemistry  
Pennsylvania State University  
University Park, PA 16902

Professor Richard Smalley  
Department of Chemistry  
Rice University  
P.O. Box 1892  
Houston, TX 77251

Professor Aaron Wold  
Department of Chemistry  
Brown University  
Providence, RI 02912

Professor Gerald Stringfellow  
Dept. of Matls. Sci.  
& Engineering  
University of Utah  
Salt Lake City, UT 84112

Professor Vicki Wysocki  
Department of Chemistry  
Virginia Commonwealth University  
Richmond, VA 23284-2006

Professor Galen Stucky  
Department of Chemistry  
University of California  
Santa Barbara, CA 93106

Professor John Yates  
Department of Chemistry  
University of Pittsburg  
Pittsburg, PA 15260

Professor H. Tachikawa  
Department of Chemistry  
Jackson State University  
Jackson, MI 39217-0510

Professor William Unertl  
Lab. for Surface Sci.  
& Technology  
University of Maine  
Orono, ME 04469

Dr. Terrell Vanderah  
Code 3854  
Naval Weapons Center  
China Lake, CA 93555

Professor John Weaver  
Dept. of Chem. Eng. & Mat. Sci.  
University of Minnesota  
Minneapolis, MN 55455

Professor Brad Weiner  
Department of Chemistry  
University of Puerto Rico  
Rio Piedras, Puerto Rico 00931