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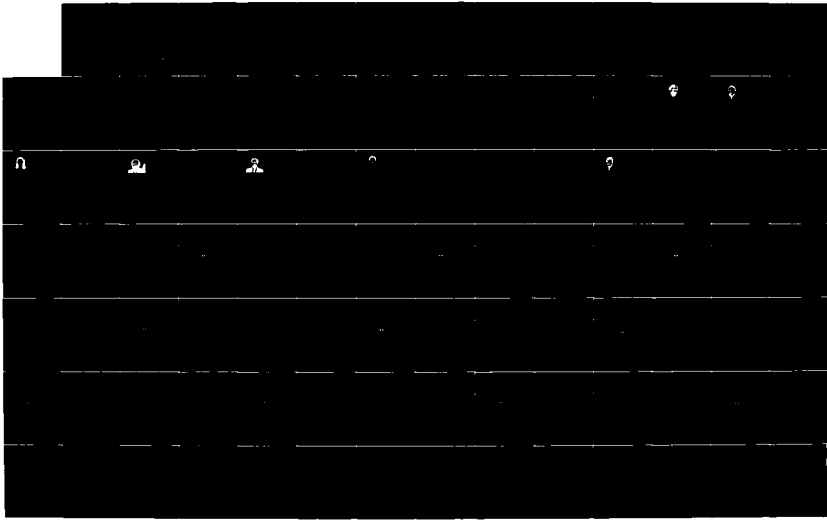
INSTALLATION RESTORATION PROGRAM PHASE 2
CONFIRMATION/QUANTIFICATION STAGE 1 VOLUME 2 APPENDICES
(U) WESTON (ROY F) INC WEST CHESTER PA JUN 86
F33615-80-D-4006

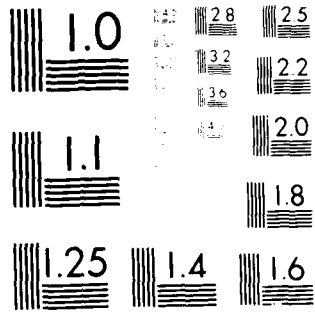
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AD-A184 581

**INSTALLATION RESTORATION PROGRAM
PHASE II - CONFIRMATION/QUANTIFICATION
STAGE 1**

VOLUME II - APPENDICES

FOR

**Mather Air Force Base
Sacramento, California**

PREPARED BY:

**Roy F. Weston, Inc.
West Chester, Pennsylvania 19380**

JUNE, 1986

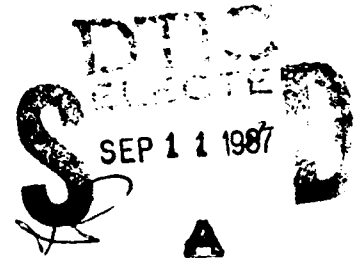
FINAL REPORT FOR PERIOD SEPTEMBER 1983 TO JUNE 1986

Approved for Public Release; distribution is unlimited

PREPARED FOR:

**HEADQUARTERS AIR TRAINING COMMAND
COMMAND SURGEON'S OFFICE (HQATC/SGPB)
BIOENVIRONMENTAL ENGINEERING DIVISION
RANDOLPH AIR FORCE BASE, TEXAS**

**UNITED STATES AIR FORCE
OCCUPATIONAL & ENVIRONMENTAL HEALTH LABORATORY (USAF OEHL)
TECHNICAL SERVICES DIVISION (TS)
BROOKS AIR FORCE BASE, TEXAS 78235-5501**



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INSTALLATION RESTORATION PROGRAM
PHASE II - CONFIRMATION/QUANTIFICATION

STAGE 1

VOLUME II

FINAL REPORT

FOR

Mather Air Force Base
Sacramento, California

USAF Air Training Command
Randolph Air Force Base, Texas

June, 1986

Prepared by

ROY F. WESTON, INC.
Weston Way
West Chester, Pennsylvania 19380

USAF Contract No. F33615-80-D-4006, Delivery Order 0026
Contractor Contract No. F33615-80-D-4006, Delivery Order No. 26

USAFOEHL Technical Program Manager - LTC Edward Barnes

Prepared for

UNITED STATES AIR FORCE
OCCUPATIONAL AND ENVIRONMENTAL HEALTH LABORATORY (USAFOEHL)
BROOKS AIR FORCE BASE, TEXAS 78235-5501

APPENDIX A

ACRONYMS, DEFINITIONS, NOMENCLATURE,
UNITS OF MEASUREMENT



A rectangular stamp containing a grid of cells. The bottom-left cell contains the handwritten text "A-125".

A-125			



APPENDIX A

ACRONYMS, DEFINITIONS, NOMENCLATURE, UNITS OF MEASUREMENT

ACW	Air Command and Warning
AFFF	Aqueous film forming foam
ASTM	American Society for Testing and Materials
ATC	Air Training Command
alluvium	Unconsolidated deposits laid down by relatively recent rivers
andesite	Crystalline volcanic rock type
aquifer	Zone beneath the earth's surface capable of producing water for a well
artesian	Groundwater condition in which pressure within an aquifer causes groundwater to rise in a well above the top of the aquifer, and sometimes above ground surface
AVGAS	Aviation gas (fuel)
BES	Bioenvironmental Engineering Services
B.G.S.	Below ground surface
breccia	A rock made up of highly angular coarse fragments
CDHS	California Department of Health Services (also referred to as DOHS)
CDWR	California Department of Water Resources
CERCLA	Comprehensive Environmental Response Compensation and Liability Act of 1980



cm/s	Centimeters per second
confined	An aquifer condition in which the more permeable aquifer materials are confined between two less permeable strata, and in which artesian pressures cause water to rise in wells to levels above the base of the upper confining stratum
connate water	Interstitial water trapped in sedimentary rock at the time it was deposited
Cretaceous	The third and last period of the Mesozoic Era, occurring approximately 144 to 66 million years ago
CRWQB	California Regional Water Quality Board
DEQPPM	Defense Environmental Quality Program Policy Memorandum
detritus	Material produced from disintegration or weathering of rocks that has been moved from its site of origin
DMN	Dimethylnitrosamine
DoD	Department of Defense
ephemeral	Describes a surface water body (stream or pond) which only has water in it during the season(s). Opposite of perennial
escarpment	A geomorphic feature represented by a steep slope or face at the edge of a highland
fault block	A mass of earth materials bounded on at least two sides by structural faults
feet/day	Feet per day

groundwater divide	A theoretical dividing line in the water table on each side of which the water table slopes away, forming a boundary between separate groundwater basins
GC	Gas chromatographic analytical instrument or method
gpm	Gallons per minute
gpd	Gallons per day
GPR	Ground-penetrating radar
HARM	Hazard Assessment Rating Methodology
HNu	A brand name for a volatile organic vapor photoionization detection meter
hydraulic conductivity	Ratio of flow velocity to driving force for viscous flow of water under saturated conditions in a porous medium, or volume of water moving through a unit area of aquifer under a unit hydraulic gradient
hydraulic gradient	Rate of change in pressure or hydraulic head in groundwater over a given distance of flow
igneous	Describes crystalline rocks formed by solidification from a molten magma either beneath the surface or on the surface
IRP	Installation Restoration Program
JP-4	Jet fuel
K	Common symbol for hydraulic conductivity
lens	A body of sediment or rock thick in the middle and thin at the edges
lenticular	Lens-shaped

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mafic	Pertaining to or composed primarily of "dark minerals" (iron magnesium silicates)
metamorphic	Describes rocks which have formed in the solid state in response to pronounced changes in temperature, pressure, and/or chemical environment
ug/g	Micrograms per gram (equal to mg/kg and equivalent to parts per million in solids)
ug/L	Micrograms per liter (equivalent to parts per billion in water)
mg/g	Milligrams per grams (equivalent to parts per thousand)
mg/L	Milligrams per liter (equivalent to parts per million in water)
mgd	Million gallons per day
MSL	Mean sea level datum
N	North
MAFB	Mather Air Force Base
O&G	Oil and grease
OEHL	Occupational and Environmental Health Laboratory
PCB	Polychlorinated biphenyl compound
perched	A saturated zone above the main saturated groundwater flow zone or aquifer, and separated from the main aquifer by a zone of low permeability
P.G.	Registered Professional Geologist
Ph.D.	Doctor of Philosophy degree
POL	Petroleum oil and lubricants

WESTON

potentiometric (piezometric) surface	Surface defined by the levels to which water will rise in wells penetrating a single aquifer, caused by hydrostatic pressure
ppb	Parts per billion (equivalent to ug/L in water)
ppm	Parts per million (equivalent to mg/L in water)
Quaternary	The last of two periods in the Cenozoic Era, subdivided in Pleistocene and Holocene (or Recent) epochs, occurring approximately 1.6 million years ago to the present
RCRA	Resource Conservation and Recovery Act of 1976
Recent	The second epoch of the Quaternary, including modern time and the period of time (approximately 10,000 years) since the last ice age (synonymous with Holocene)
SAC	Strategic Air Command
sedimentary	Describes rocks resulting from deposition of transported material that has accumulated in layers
semi-confined	An aquifer condition in which the confining strata above the aquifer are not laterally continuous
specific capacity	The sustained yield of a well divided by the drawdown in that well after a stabilized pumping condition is obtained (reported in gpm/foot)
specific yield	Volume of water yielded by gravity per unit volume of saturated aquifer materials (corresponds to storativity in an unconfined aquifer)
square feet/day	Square feet per day



storativity (storage coefficient)	Volume of water than an aquifer releases from or takes into storage per unit surface area of aquifer per unit change in hydraulic head
TCE	Trichloroethylene
terrigenous	Deposited in or on the earth's crust, not in a marine environment
tertiary	The first of two periods in the Cenozoic Era, occurring approximately 66 to 1.6 million years ago
TOC	Total organic carbon
transmissivity	The volume of water moving per unit time per unit width of a saturated layer under a unit hydraulic gradient
tuff-breccia	Indurated volcanic rock, formed from coarse angular volcanic fragments in a matrix of finer volcanic particles
unconfined	An aquifer in which the water table forms the upper boundary
unconsolidated sediments	Sediments that are uncemented and thus include interconnected void space (primary porosity) that allows storage and transmission of significant volumes of groundwater
USAF	United States Air Force
U.S. EPA	United States Environmental Protection Agency
VOA	Volatile organic and aromatic hydrocarbon compounds
water table	The level below which earth materials are saturated with water

APPENDIX B

SCOPE OF WORK

TASK ORDER 0026-02

STAGED

INSTALLATION RESTORATION PROGRAM

21 FEB 1984

Phase IIB Field Evaluation

Mather AFB, California*

I. Description of Work

The purpose of this task is to determine if environmental contamination has resulted from waste disposal practices at Mather AFB CA; to provide estimates of the magnitude and extent of contamination, should contamination be found; to identify potential environmental consequences of migrating pollutants; to identify any additional investigations and their attendant costs necessary to properly evaluate the magnitude, extent and direction of movement of discovered contaminants.

The presurvey report (mailed under separate cover) and Phase I IRP report (mailed under separate cover), incorporated background and description of the sites for this task. To accomplish the survey effort, the contractor shall take the following steps:

A. General

1. The areal extent of each site shall be determined by reviewing available aerial photos of the base, and by field reconnaissance.

2. Each location where surface water, sediment, or core samples are collected shall be marked with a permanent marker (where practical), and the location recorded on a project map for the site.

3. Three rounds of sampling shall be performed on the monitoring wells installed during this task. The rounds shall be performed at three different aquifer level conditions. Exact dates shall be determined by the contractor in the field.

B. AC&W Area

1. Install three downgradient monitoring wells at the AC&W site. Wells shall be installed according to procedures outlined in paragraph H below.

2. Collect one sample per well during each sampling round. Analyze the samples for VOC, TOC, oils and greases (IR method) and PCBs.

C. "7100" Area Disposal Site

1. Install three downgradient monitoring wells along the perimeter road west and south of the disposal site. The wells shall be installed according to procedures outlined in paragraph H below.

*Highlights of modification underscored

2. Collect one sample per well during each sampling round. Analyze samples for oils and greases (IR), TOC, VOC, phenol, cyanide; Cr, Pb, Cd, Ni, Ag.

D. West Ditch Area

1. Install two downgradient monitoring wells west of the ditch near the base perimeter. The wells shall be installed according to procedures outlined in paragraph H below.

2. Collect two sediment samples from the ditch, one north and one south of the west ditch skimmer.

3. Collect one sample per well during each sampling round. Analyze groundwater and sediment samples for oils and greases (IR), TOC, VOC, phenol, cyanide; Pb, Cr, Cd, Ni, Ag.

E. Northeast and East Base Perimeter

1. Install three monitoring wells along the northeast and east base perimeter. The wells shall be installed according to procedures outlined in paragraph H below.

2. Collect one sample per well during each sampling round. Analyze samples for dimethylnitrosamine (DMN); oil and greases (IR), TOC, VOC, Cr, Pb, Cd, Ni, Ag, DDT, Chlordane, 2,4-D.

F. Base Production Wells

1. All 15 base production wells shall be sampled one time, and analyses performed for TOC, VOC, and oils and greases (IR) on each sample.

2. In addition to the parameters specified in I.F.1. above, analyze specific well water samples for:

AC&W well

PCB

MB-1,2,3,4 (4 total)

DMN, Cr, Pb, Cd, Ni, Ag, DDT, Chlordane, 2,4-D

Engine Test Cell well

Phenol, Cyanide, Cr, Pb, Cd, Ni, Ag

G. Sampling and Analysis

Sampling, maximum holding time and preservation of samples shall strictly comply with the following references: Standard Methods for The Examination of Water and Wastewater, 15th Ed., (1980), pp 35-42; ASTM, Part 31, pp 72-82, (1976), Method D-3370; and Methods for Chemical Analysis of Waters and Wastes, EPA Manual 600/4-79-020, pp xiii to xix (1979). Minimum detection limit for analyses are shown in Attachment 1.

H. Well Installation and Cleanup

1. All wells installed during this survey shall be constructed of black iron materials, and shall be grout-sealed in accordance with State of

California requirements. Wells shall average 120 feet in depth and shall be logged in accordance with U.S. Army Toxic and Hazardous Materials Agency procedures (furnished under separate cover). Location and elevation of each well shall be surveyed, and recorded on the project map. Total well footage installed shall not exceed 1320 feet.

2. Each well installation shall be cleaned following the completion of the well. Drill cuttings shall be removed and the general area cleaned.

I. Data Review

Results of each round of sampling and analysis shall be tabulated in an Informal Technical Information report (Sequence 3 as reflected in Item VI below) and forwarded to USAF OEHL/CVT for review.

J. Report Preparation

1. A draft final report delineating the findings of this field investigation shall be prepared and forwarded to the USAF OEHL as specified in Item VI below. This report shall include a discussion of the regional hydrogeology, well logs of all project wells, data from water level surveys, water quality analysis results, available geohydrologic cross sections, groundwater surface and gradient vector maps, vertical and horizontal flow vectors and Laboratory quality assurance information. The report shall follow the USAF OEHL supplied format (mailed under separate cover).

2. Estimates shall be made of the magnitude, extent and direction of movement of contaminants discovered. Potential environmental consequences of discovered contamination must be identified. Where survey data are insufficient to properly determine or estimate the magnitude, extent and direction of movement of discovered contaminants, specific recommendations, fully justified, shall be made for additional efforts required to properly evaluate contamination migration and included in a separately bound appendix to the draft final report (see K below).

3. Specific requirements for future groundwater and surface water monitoring must be identified.

K. Cost Estimates

The contractor shall provide estimates for all additional work recommended to permit proper determination of contaminants. The recommendations provided shall include all efforts required to determine the magnitude and direction of movement of discovered contaminants along with an estimate of the time required to accomplish the proposed effort. This information shall be provided in a separately bound appendix to the draft final report.

L. Meetings

The contractor's project leader shall attend two meetings, to be held at Mather AFB to discuss project status at dates and times to be specified later. Each meeting shall last 12 hours.

II. Site Location and Dates:

Mather AFB CA
Building, Time &
Dates to be established

III. Base Support: None

IV. Government Furnished Property: None

V. Government Points of Contact:

- | | |
|--|--|
| 1. Dr Dee Ann Sanders
USAF OEHL/CVT
Brooks AFB TX 78235
(512) 536-2158
AV 240-2158 | 2. Capt Dennis Korycinski
USAF Hospital Mather/SGPB
Mather AFB CA 95655
(916) 364-2284
AV 828-2284 |
|--|--|

VI. In addition to sequence numbers 1, 5 and 11 listed in Atch 1 to the contract, which are applicable to all orders, the reference numbers below are applicable to this order. Also shown are data applicable to this order.

Sequence No.	Block 10	Block 11	Block 12	Block 13	Block 14
4	CNE/R	15 MAC	16 MAC		
3	AS REQ	12 MAC **	15 MAC **	<u>21 MAC</u>	*

*A minimum of two draft reports will be required. After incorporating Air Force comments concerning the first draft report, the contractor shall supply the USAF OEHL with a second draft report. The report shall be forwarded to the applicable regulatory agencies for their comments. Contractor shall supply the USAF OEHL with 25 copies of each draft report and 50 copies plus the original camera ready copy of the final report.

**Upon completion of analyses.

Attachment 1
Required Sample Detection Limits

<u>Parameter</u>	<u>Soil/Sediment</u>	<u>Water</u>
*Total Organic Carbon (TOC)	1.0 milligram/gram	1.0 milligram/L
Oils and Greases (IR Method 412.3)	100 micrograms/gram	10 micrograms/L
Phenol	1 microgram/gram	1.0 microgram/L
Cyanide	1 microgram/gram	10 micrograms/L
Lead	2 micrograms/gram	20 micrograms/L
Chromium	5 micrograms/gram	50 micrograms/L
Cadmium	1 microgram/gram	10 micrograms/L
Nickel	10 micrograms/gram	100 micrograms/L
Silver	1 microgram/gram	10 micrograms/L
PCB's	---	0.25 micrograms/L
Dimethylnitrosamine (DMN)	---	1.0 microgram/L
DDT isomers	---	0.02 micrograms/L
Chlordane	---	0.02 micrograms/L
2,4-D	---	0.06 microgram/L
Volatile Organic Compounds (VOC)	**	**

*Detection levels for TOC must be 3 times the noise level of the instrument. Laboratory distilled water must show no response; if it shows a response, corrections of positive results must be made.

**Detection limits for VOCs shall be as specified for those compounds listed in EPA Methods 601 and 602.

Method: Federal Register, Vol 44, No. 233, pp 69468-69473.

This method should be strictly followed including these items:

Item 1.4 - This method is recommended by EPA for use only by experienced residue analysts or under the close supervision of such qualified persons.

Item 2.2 - This is most important. If interferences are encountered (as in early peaks such as vinyl chloride), the method provides a secondary gas chromatographic column that will be helpful in resolving the compounds of interest from interferences. This must be done in the case of vinyl chloride and so noted in analysis report.

Items 3.3, 7.1-7.3 - These sections on interferences, contamination and QC should be strictly followed.

Items 8.3 - All samples must be analyzed within the recommended holding times. This must be followed without exception.

If questions are encountered about certain contaminants you may be asked to show both chromatograms used to rule out possible interferences.

Table 3: SUMMARY OF ANALYTICAL REQUIREMENTS

Analyte	Sampled Medium	AC&W	7100	W. Ditch	Perimeter	Base Wells	QA	Total Samples
VOA	Water	9	9	6	9	15	10	58
VOA	Soil	0	0	2	0	0	1	3
TOC	Water	9	9	6	9	15	10	58
O & G	Water	9	9	6	9	15	10	58
O & G	Soil	0	0	2	0	0	1	3
PCB	Water	9	0	0	0	1	2	12
Phenol	Water	0	9	6	0	1	3	19
Phenol	Soil	0	0	2	0	0	1	3
Cyanide	Water	0	9	6	0	1	2	18
Cyanide	Soil	0	0	2	0	0	1	3
Metals ¹	Water	0	9	6	9	5	6	35
Metals ¹	Soil	0	0	2	0	0	1	3
DMN ²	Water	0	0	0	9	4	3	16
Pesticides ³	Water	0	0	0	9	4	3	16

1. Metals - Cr, Pb, Cd, Ni, Ag
2. DMN = Dimethylnitrosamine
3. Pesticides - DDT, chlordane, 2,4-D

Revision No. 2 to Description of Work
INSTALLATION RESTORATION PROGRAM

STAGED

84 Feb 21

Phase IIB Field Evaluation

Mather AFB, California*

I. Description of Work

The purpose of this task is to determine if environmental contamination has resulted from waste disposal practices at Mather AFB CA; to provide estimates of the magnitude and extent of contamination, should contamination be found; to identify potential environmental consequences of migrating pollutants; to identify any additional investigations and their attendant costs necessary to properly evaluate the magnitude, extent and direction of movement of discovered contaminants.

The presurvey report (mailed under separate cover) and Phase I IRP report (mailed under separate cover), incorporated background and description of the sites for this task. To accomplish the survey effort, the contractor shall take the following steps:

A. General

1. The areal extent of each site shall be determined by reviewing available aerial photos of the base, and by field reconnaissance.

2. Each location where surface water, sediment, or core samples are collected shall be marked with a permanent marker (where practical), and the location recorded on a project map for the site.

3. Three rounds of sampling shall be performed on the monitoring wells installed during this task. The rounds shall be performed at three different aquifer level conditions. Exact dates shall be determined by the contractor in the field.

B. AO&W Area

1. Install three downgradient monitoring wells at the AO&W site. Wells shall be installed according to procedures outlined in paragraph H below.

2. Collect one sample per well during each sampling round. Analyze the samples for VOC, TOC, oils and greases (IR method) and PCBs.

C. "7100" Area Disposal Site

1. Install three downgradient monitoring wells along the perimeter road west and south of the disposal site. The wells shall be installed according to procedures outlined in paragraph H below.

*Highlight of modification underscored

2. Collect one sample per well during each sampling round. Analyze samples for oils and greases (IR), TOC, VOC, phenol, cyanide; Cr, Pb, Cd, Ni, Ag.

D. West Ditch Area

1. Install two downgradient monitoring wells west of the ditch near the base perimeter. The wells shall be installed according to procedures outlined in paragraph H below.

2. Collect two sediment samples from the ditch, one north and one south of the west ditch skimmer.

3. Collect one sample per well during each sampling round. Analyze groundwater and sediment samples for oils and greases (IR), TOC, VOC, phenol, cyanide; Pb, Cr, Cd, Ni, Ag.

E. Northeast and East Base Perimeter

1. Install three monitoring wells along the northeast and east base perimeter. The wells shall be installed according to procedures outlined in paragraph H below.

2. Collect one sample per well during each sampling round. Analyze samples for dimethylnitrosamine (DMN); oil and greases (IR), TOC, VOC, Cr, Pb, Cd, Ni, Ag, DDT, Chlordane, 2,4-D.

F. Base Production Wells

1. All 15 base production wells shall be sampled one time, and analyses performed for TOC, VOC, and oils and greases (IR) on each sample.

2. In addition to the parameters specified in I.F.1. above, analyze specific well water samples for:

AC&W well

PCB

MB-1,2,3,4 (4 total)

DMN, Cr, Pb, Cd, Ni, Ag, DDT, Chlordane, 2,4-D

Engine Test Cell well

Phenol, Cyanide, Cr, Pb, Cd, Ni, Ag

G. Sampling and Analysis

Sampling, maximum holding time and preservation of samples shall strictly comply with the following references: Standard Methods for The Examination of Water and Wastewater, 15th Ed., (1980), pp 35-42; ASTM, Part 31, pp 72-82, (1976), Method D-3370; and Methods for Chemical Analysis of Waters and Wastes, EPA Manual 600/4-79-020, pp xiii to xix (1979). Minimum detection limit for analyses are shown in Attachment 1.

H. Well Installation and Cleanup

1. All wells installed during this survey shall be constructed of black iron materials, and shall be grout-sealed in accordance with State of

California regular basis. Wells shall average 120 feet in depth and shall be logged in accordance with U.S. Army Toxic and Hazardous Materials Agency procedures (furnished under separate cover). Location and elevation of each well shall be surveyed, and recorded on the project map. Total well footage installed shall not exceed 1500 feet.

2. Each well installation shall be cleaned following the completion of the well. Drill cuttings shall be removed and the general area cleared.

I. Data Review

Results of each round of sampling and analysis shall be tabulated in an Internal Technical Information Report (See Item B as reflected in Item VI below) and forwarded to USAF CDR/OT for review.

J. Report Preparation

1. A draft final report delineating the findings of this field investigation shall be prepared and forwarded to the USAF CDR/OT as specified in Item VI below. This report shall include a discussion of the regional hydrogeology, well logs of all project wells, data from water level surveys, water quality analysis results, available geohydrologic cross sections, groundwater surface and gradient vector maps, vertical and horizontal flow vectors and laboratory quality assurance information. The report shall follow the USAF CDR/OT supplies format (furnished under separate cover).

2. Estimates shall be made of the magnitude, extent and direction of movement of contaminants discovered. Potential environmental consequences of discovered contamination must be identified. Where survey data are insufficient to properly determine or estimate the magnitude, extent and direction of movement of discovered contaminants, specific recommendations, fully justified, shall be made for additional efforts required to properly evaluate contamination migration and included in a separately bound appendix to the draft final report (see K below).

3. Specific requirements for future groundwater and surface water monitoring must be identified.

K. Cost Estimate

The contractor shall provide a cost estimate for all activities with respect to permit preparation and other administrative matters. The contractor shall also provide a cost estimate for the field investigation and identification of contaminants, including the cost of the time required to accomplish the proposed effort. This information shall be provided in a separately bound appendix to the draft final report.

L. Milestones

The contractor's project schedule shall be submitted to the USAF CDR/OT at 1500 hours on the project start date. The contractor shall submit a final report to the USAF CDR/OT within 120 days of the project start date.

II. Site Location and Dates:

Mather AFB CA
Building, Time &
Dates to be established

III. Base Support: None

IV. Government Furnished Property: None

V. Government Points of Contact:

- | | |
|--|--|
| 1. Dr Dee Ann Sanders
USAF OEHL/CVT
Brooks AFB TX 78235
(512) 536-2158
AV 240-2158 | 2. Capt Dennis Konyeinski
USAF Hospital Mather/SFPB
Mather AFB CA 95655
(916) 364-2284
AV 828-2284 |
|--|--|

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Sequence No.	Block 10	Block 11	Block 12	Block 13	Block 14
4	ONE/R	<u>84NOV26</u>	<u>84DEC31</u>	<u>85MAY27</u>	*
3	AS REQ	**	**		

*A minimum of two draft reports will be required. After incorporating Air Force comments concerning the first draft report, the contractor shall supply the USAF OEHL with a second draft report. The report shall be forwarded to the applicable regulatory agencies for their comments. Contractor shall supply the USAF OEHL with 25 copies of each draft report and 50 copies plus the original camera ready copy of the final report.

**Upon completion of analyses.

Attachment 1
Required Sample Detection Limits

<u>Parameter</u>	<u>Soil/Sediment</u>	<u>Water</u>
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Cadmium	1 microgram/gram	10 micrograms/L
Nickel	10 micrograms/gram	100 micrograms/L
Silver	1 microgram/gram	10 micrograms/L
PCB's	---	0.25 micrograms/L
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*Detection levels for TOC must be 3 times the noise level of the instrument. Laboratory distilled water must show no response; if it shows a response, corrections of positive results must be made.

**Detection limits for VOCs shall be as specified for those compounds listed in EPA Methods 601 and 602.

Method: Federal Register, Vol 44, No. 233, pp 69458-69473.

This method should be strictly followed including these items:

- Item 1.4 - This method is recommended by EPA for use only by experienced residue analysts or under the close supervision of such qualified persons.
- Item 2.2 - This is most important. If interferences are encountered (as in early peaks such as vinyl chloride), the method provides a secondary gas chromatographic column that will be helpful in resolving the compounds of interest from interferences. This must be done in the case of vinyl chloride and so noted in analysis report.
- Items 3.3, 7.1-7.3 - These sections on interferences, contamination and QC should be strictly followed.
- Item 8.3 - All samples must be analyzed within the recommended holding time. This must be followed without exception.

If questions are encountered about certain contaminants you may be asked to show both chromatograms used to rule out possible interferences.

APPENDIX C

BIOGRAPHIES OF KEY PERSONNEL



Peter J. Marks

Education

B S. Biology, Franklin and Marshall College (1963)

M.S. Environmental Engineering and Science, Drexel University (1965)

Employment History

1965-Present Weston

1963-1964 Lancaster County General Hospital
Research laboratory for analytical
methods development

Relevant Experience

Mr. Marks has 15 years experience in environmental laboratory and engineering activities as a Project Scientist, Project Engineer, Project Manager, and Vice President of Weston's environmental laboratory. He has analytical laboratory experience, supervision of source emission testing projects, and was the Project Manager on numerous source testing and ambient air monitoring projects, including a major contract with EPA for source sampling and analysis. He also has experience in field testing to determine efficiencies of control equipment, and chemical analysis of atmospheric emissions from various industries.

Mr. Marks was the Project Manager for a major corporate (65 plants) air testing contract (\$350,000 year). The plants included glass, wood, textiles, and asphalt production.

Mr. Marks' industry experience in source emissions testing includes fossil-fuel-fired steam generators, municipal incinerators, cement plants, nitric acid plants, petroleum refineries and petrochemical plants, iron and steel plants (basic

oxygen and electric arc furnaces), wet process phosphoric acid plants; superphosphoric acid plants; diammonium phosphate plants; triple superphosphate plants; granular triple superphosphate storage facilities; intermediate size steam boilers (10-250 x 10⁶ Btu); mercury plants; solvent degreasing facilities; steel foundries; synthetic organic chemical plants; pulp and paper mills; chlor-alkali plants; glass manufacturing facilities; stone crushing facilities; plastic plants; clay and ore processing operations.

Mr. Marks' air contaminant testing experience includes: particulates, NO_x, fluorides, SO₂, SO₃, H₂SO₄, chlorides, hydrocarbons, aldehydes, organic acids, total reduced sulfur, permanent gases, odor, mercury, particle size, resistivity, hydrogen sulfide, chloride, ozone, metals, sulfates, vinyl chloride, solvents, TSP, and asbestos.

His field instrumentation experience includes: Orsat apparatus, Teledyne combustible instrument, Lira nondispersive infrared instrument, Servomex oxygen analyzer, Lear-Siegler transmissometer, duPont sulfur dioxide monitoring instrumentation, Anderson cascade impactor, Omega pyrometer, Meloy ozone analyzer, thermoelectron SO₂ analyzer, RAC Hi Vol Samplers, and RAC Nutech control console.

Mr. Marks has the following laboratory instrumentation experience: infrared, ultraviolet, and atomic absorption spectrophotometry; dissolved oxygen analyzer, gas chromatography, and total oxygen demand and total organic carbon analyzers.

Mr. Marks is a member of the Air Pollution Control Association, the American Society for Testing and Materials, the Water Pollution Control Federation, and the Water Pollution Control Association of Pennsylvania.

Professional Profile



Frederick Bopp III, Ph.D., P.G.

Registration

Registered Professional Geologist in the State of Indiana

Fields of Competence

Groundwater resources evaluation; hydrogeologic evaluation of sanitary landfills and other waste disposal sites; detection and abatement of groundwater pollution; digital modeling of groundwater flow and solute transport; statistical analysis of geological and geochemical data; geochemical prospecting; estuarine geology and geochemistry; trace metal and aqueous geochemistry.

Experience Summary

Seven years experience in hydrogeology and geochemistry, involving such activities as: assessment of subsurface water and soil contamination; development of contamination profiles; evaluation of remediation actions for groundwater quality restoration; quantitative chemical analysis of water and soil; ore assay and ore body evaluation; drilling supervisor; hydrogeologic assessment; pollution detection and abatement; estuarine pollution analysis; application of flow and solute transport computer models; computer programming; project management; teaching environmental geology and geochemistry.

Credentials

B.A., Geology—Brown University (1966)
M.S., Geology—University of Delaware (1973)
Ph.D., Geology—University of Delaware (1979)
Sigma Xi, The Scientific Research Society of North America
Geological Society of America, Hydrology Division
National Water Well Association, Technical Division
American Association for the Advancement of Science
Estuarine Research Federation, Atlantic Estuarine Research Society

Employment History

1979-Present	WESTON
1977-1979	U.S. Army Corps of Engineers Waterways Experiment Station
1976-1977	University of South Florida Department of Geology
1970-1976	University of Delaware Department of Geology
1974-1976	Earth Quest Associates President and Principal Partner
1974 (Summer)	WESTON
1966-1970	United States Navy Commissioned Officer

Key Projects

Project manager on seven task orders for environmental assessment services at United States Air Force facilities in nine states.

Task manager for a Superfund site evaluation in Ohio.

Site manager for drum recovery operations in Pennsylvania and New Jersey.

Project manager for site assessments of oil and fuel spills in four states.

Project manager for closure plan development at a hazardous waste landfill in New Jersey.

Definition and abatement of groundwater contamination from chemical manufacturing in Delaware.

Flow and solute transport digital model of a heavily-pumped regional aquifer in southern New Jersey.

Definition and abatement of groundwater contamination from chemical manufacturing in the Denver area

Hydrogeologic impact assessment of on-land dredge spoil disposal in coastal North Carolina

Geochemical prospecting and ore body analysis in Arizona.

Professional Geologist

Definition and abatement of groundwater contamination from a hazardous waste site in northern New England.

Definition and abatement of groundwater contamination from plating and foundry wastes in eastern Pennsylvania.

Operational test and evaluation of new naval mine ordinances in southern Florida.

Publications

"Metals in Estuarine Sediments: Factor Analysis and Its Environmental Significance". *Science*, 214 (1981): 441-443.

"The Remobilization of Trace Metals from Suspended Sediments Entering the Delaware Estuary". Presented at the 27th Annual Meeting, Southeastern Section, Geological Society of America, Chattanooga, Tennessee, April 1978.

"Trace Metals in Delaware Bay Sediments and Oysters". Presented at the International Conference on Heavy Metals in the Environment, Toronto, Canada, October 1975.



Katherine A. Sheedy

Fields of Competence

Geologic investigation and site evaluation; environmental impact assessment; quantitative and qualitative groundwater analysis; design of groundwater monitoring systems.

Experience Summary

Nine years experience in geological investigations including environmental impact analysis in geology, groundwater, and soils; hydrogeologic investigations of hazardous waste sites; preparation and delivery of expert testimony; assessment and mitigation of low-level radioactive contamination of groundwater and soils; migration of low-level radioactive contamination of groundwater and soils; migration of radionuclides in groundwater; site stability in limestone terrains; development of evaluation criteria for site search and selection projects; pre-mine opening hydrologic investigations for surface and underground coal mines; development of clean-up strategies for hazardous and radioactive waste disposal sites; Environmental Impact Statement preparation and review; site suitability investigations of waste disposal facilities for industrial and residential developments.

Credentials

B.A.—Queens College, CUNY (1969)
M.S., Geology—University of Delaware (1975)
American Geophysical Union
Geological Society of America
National Water Well Association, Technical Division

Employment History

1974-Present WESTON
1972-1974 University of Delaware

Key Projects

Preparation of RCRA Part B permit application for facilities in the Midwest and on the West coast

Initial Assessment Studies to identify possible contamination resulting from past practices at military installations.

Assessment of groundwater contamination from a municipal landfill in the Atlantic Coastal Plain including aquifer simulation to determine migration 10, 20 and 30 years in the future.

Hydrogeologic assessment of a multi-source military installation. The project includes groundwater modeling for the installation and for areas outside the installation in conjunction with State and Federal agencies

Design of monitoring systems for a large industrial complex in Montana.

Assessment of regulatory requirements for hazardous waste lagoon closure in over forty states

Assessment and analysis of emerging trends in groundwater research as applied to the utility industry

Preparation of EPA Remedial Action Master Plans for five uncontrolled hazardous waste sites

Principal investigator for geology, soils and groundwater portion of an Environmental Impact Statement for the decontamination of a radioactive waste disposal site in Canonsburg, Pennsylvania

Project manager and principal investigator on clean-up of a site contaminated by pharmaceutical wastes in New Jersey.

Project manager and principal investigator for assistance in EIS preparation for five synthetic fuel plants in east-central United States.

Evaluation of environmental impact and operation of 23 municipal landfills in the Atlantic Coastal Plain

Hydrogeologic investigations at mine sites prior to, during and after mining operations in Illinois

Hydrogeologic investigations to determine site suitability for landfills, sewage sludge disposal, spray irrigation and industrial waste disposal

Principal investigator on a dredge material disposal site feasibility study for Interstate Division for Baltimore City. This project was conducted to evaluate the feasibility of specific sites for disposal of 5 million cubic yards of

Professional Profile

material dredged from the Fort McHenry Tunnel in Baltimore. The evaluation included examination of costs, engineering feasibility, site stability, impact on biology and groundwater and ultimate use of the site as an inner-city park.

Supervision of an investigation: to determine groundwater quality, delineate the extent of groundwater pollution and develop a groundwater-quality management program for a six-county area. Evaluated the adequacy of existing groundwater-quality standards and interacted with regulatory agencies.

Evaluation of groundwater quality, quantity and facilities: impact on groundwater for sites in semi-arctic environments and within the Columbia River Basin Project area.

Environmental assessment for a 200,000-BPCD refinery on a semi-arid island with extensive groundwater use in the West Indies.

Evaluation of structural stability problems in limestone solution area in Pennsylvania.

Supervision of a leachate collection system and groundwater monitoring program for an industrial landfill.

Investigation of potential sources of petroleum product found to be discharging through the subsurface, at the shore of Lake Erie.

Development of a state-of-the-art study and environmental analysis of the geothermal steam industry.

Publications

Sheedy, K. A., 1979, "Three-Phase Approach to Determination of Site Stability in Limestone". Presented at Association of Engineering Geologists 1979 Annual Meeting, Chicago, Illinois.

Sheedy, K. A., Schoenberger, R. J., Haderer, P., Dovey, R., 1979, "Solid Waste Disposal in the Coastal Plain: A Case Study." Presented at Association of Engineering Geologists 1979 Annual Meeting, Chicago, Illinois.

Sheedy, K. A., Leis, W., Thomas, A., 1980, "Land Use in Limestone Terrain, Problems and Case Study Solutions". In *Applied Geomorphology*, (The "Binghamton" symposia: 11) George Allen and Unwin, 1982.

Sheedy, K. A., Leis, W., Bopp, F., Anderson, J., "Use of Ground Penetrating Radar in Limestone Terrain". American Geographers Association, 1981.

Sheedy, K. A., "Methodology for the Selection of Low-Level Radioactive Waste Disposal Sites". American Nuclear Society, 1982.



Walter M. Leis, P.G.

Registration

Registered Professional Geologist in the States of Georgia (No. 440) and Indiana.

Fields of Competence

Detection and abatement of groundwater contamination; design of artificial recharge wells; deep well disposal; simulation of groundwater systems; hydrogeologic evaluation of hazardous waste sites and landfills; practical applications of geophysical surveys to hydrologic systems, site investigations, and borehole geophysical surveys. Geochemical studies of acid mine drainage and hazardous wastes.

Experience Summary

Sixteen years experience as field hydrogeologist, field supervisor, project director, research director. Six years research involving two consecutive projects. 1) application of geophysical techniques in evaluating groundwater supplies in fractured rock terrain in Delaware and Pennsylvania; 2) project director for an artificial recharge and deep well disposal study. Provided consultation for waste disposal and aquifer quality problems for coastal communities.

Developed geochemical sampling techniques for deep mine sampling. Evaluated synthetic and field hydrologic data for deep formulational analysis in coal field projects.

Earlier research experience involved developing techniques for mapping subsurface regional structures having interstate hydrologic significance, and defining ore bodies by geochemical prospecting.

Credentials

B.S., Biochemistry—Albright College (1966)

M.S., Hydrogeology—University of Delaware (1975)

Cooperative Program Environmental Engineering—University of Pennsylvania

Additional special course work in Geology and Hydrology, Franklin and Marshall College and Pennsylvania State University

Remote Sensing Data Processing Training, Goddard Space Center (1978)

OWRR Research Fellow, 1973

National Water Well Association, Technical Division.

Geological Society of America, Engineering Geological Division

Society of Economic Paleontologists and Mineralogists

Employment History

1974-Present	WESTON
1973-1974	University of Delaware Water Resources Center
1971-1973	University of Delaware
1967-1971	Pennsylvania Department of Environmental Resources

Key Projects

Definition of groundwater contamination from sanitary landfill leachate and recovery of contaminants to protect heavily used aquifer in Delaware.

Field design studies for artificial recharge and waste disposal wells.

Design and construction of hydrologic isolation systems for various class hazardous wastes.

Design and supervision of chemical and physical rehabilitation of groundwater collection systems in fractured rock and coastal plain areas.

Principal investigator for six projects involving subsurface migration of PCB's in New York, New Jersey, Pennsylvania, and Oklahoma.

Design and construction supervision of hydrocarbon recovery wells in Pennsylvania.

Professional Profile

Geochemical evaluation of coal mine pools in West Virginia.

Geochemistry of subsurface migration of toxic substances.

Principal investigator for eight projects involving migration of volatile chlorinated hydrocarbons in groundwater.

Mineable reserve evaluations for coal, sand and gravel, limestone, clay deposits, mine reclamation, and monitoring.

Design geophysical and remote sensing assessments of hazardous waste disposal areas.

Publications

Leis, W., and R.R. Jordan, 1974, "Geologic Control of Groundwater Movement in a Portion of the Delaware Piedmont", OWRR—DEL 20.

Leis, W., 1976, "Artificial Recharge for Coastal Sussex County, Delaware", University of Delaware Press, Water Resources Center.

Leis, W., D.R. Clark, and A. Thomas, 1976, "Control Program for Leachate Affecting a Multiple Aquifer System, Army Creek Landfill, New Castle County, Delaware", National Conference on Management and Disposal of Residue on Land.

Leis, W., W.F. Beers, J.M. Davidson, and G.D. Knowles, 1978, "Migration of PCB's by Groundwater Transport—A Case Study of Twelve Landfills & Dredge Disposal Sites on the Upper Hudson Valley, New York", Proceedings of the 1st Annual Conference of Applied Research & Practice on Municipal and Industrial Waste.

Leis, W., R.D. Moose, and W.F. Beers, "Critical Area Maps, a Regional Assessment for Karst Topography", Association of Engineering Geologists 1978 Annual Meeting.

Leis, W., and W.F. Beers, "Soil Isotherm Studies to Predict PCB Migration Within Groundwater", (Abstract) ASTM 1979 Annual Meeting, Philadelphia, Pennsylvania.

Thomas, A., and W. Lein, "Physical & Chemical Rehabilitation of Contaminant Recovery Wells", Association of Engineering Geologists 1978 Annual Meeting.

Leis, W., W.F. Beers, and F. Benenati, "Migration of PCB's from Landfills and Dredge Disposal Sites in the Upper Hudson River Valley", New York Academy of Science Symposium on PCB's in the Hudson River.

Leis, W., "Subsurface Reclamation by Counter Pumping Systems: Geologic and Geotechnical Aspects of Land Reclamation", ASCE/AEG 1979 Symposium.

Leis, W., and A. Metry, "Field Characterization of Leachate Quality", Water Pollution Control Federation 1979 Annual Meeting.

Leis, W., and A. Metry, "Multimedia Pathways of Contaminant Migration", Water Pollution Control Federation 1980 Annual Meeting.

Leis, W., and K. Sheedy, "Geophysical Location of Abandoned Waste Disposal Sites", 1980 National Conference on Management of Uncontrolled Hazardous Waste Sites.

Sheedy, K., and W. Leis, 1982, "Hydrogeological Assessment in Karst Environments (chapter)."



James S. Smith, Ph.D.

Fields of Competence

Analytical laboratory management; organic chemistry; mass spectrometry, GC/MS/DS, high and low resolution, chemical ionization and special techniques; gas chromatography including capillary column techniques, high performance liquid chromatography (HPLC), the uses of NMR, IR, UV, visible, inorganic analyses, electrochemical, thermal techniques and surface methodologies (SEM, ESCA, SIMS) to solve industrial problems, the development of quality control measures in analytical protocols, the testing of laboratory safety methodologies, innovation of new analytical techniques and methods to solve industrial product liability, production and environmental problems.

Experience Summary

Eleven years experience in the supervision of an analytical group involved in solving all types of industrial problems including environmental, product safety, production, research and development. The main emphasis was on the innovative development of analytical methods utilizing instrumental technologies. In-depth experience in the organic chemicals, inorganic chemicals, polymer, fiber, tire, solvent, fluorine chemicals, coke and coal tar industries. Numerous scientific presentations. Contributor to three Chemical Manufacturers Association Task Groups: Environmental Monitoring, Groundwater, and Hazardous Waste Response Center.

Taught general chemistry, analytical chemistry, organic chemistry, and instrumental analysis for four years at Eastern Michigan University and the University of Illinois.

Credentials

B.A., Chemistry—Williams College (1960)
Ph.D., Organic Chemistry—Iowa State University (1964)
Postdoctoral Organic Chemistry—University of Illinois (1966)
Postdoctoral Mass Spectroscopy—Cornell University (1969)

American Chemical Society
American Society for Testing Materials
American Society of Mass Spectroscopists

Employment History

1981-Present	WESTON
1969-1981	Allied Chemical Corporation Corporate Research Center
1966-1968	Eastern Michigan University Assistant Professor of Chemistry
1965-1966	University of Illinois

Key Projects

Directed analytical group for five years of intensive sampling and analysis of a toxic insecticide. Analyses involved soil, air, water, sludge, blood, bile, feces, urine, animal feed, and plant samples to detect the compound at the low parts-per-billion level. The project involved rapid development of new and accurate analytical methods.

Developed an instrumental analytical laboratory consisting of trace environmental analyses, gas chromatography, high performance liquid chromatography, mass spectrometry, surface analyses, X-ray photoelectron spectroscopy and nuclear magnetic resonance spectroscopy including the design and manufacture of instrument modifications, purchasing instruments, and hiring of key personnel.

Isolated, identified, and developed a method of analysis for a colored impurity on a bulk chemical product. Synthesized the colorant for proof of identification and as a standard for future analysis. Proved the mechanism of the development of the color from the packaging materials. Designed new specifications eliminating the problem.

Conducted corporate plant environmental laboratory QA/QC audits including the development of a corporate QA/QC manual.

Professional Profile

Provided an inexpensive and accurate method of analysis of lead for a manufacturing plant effluent. A published methodology in kit form was modified for plant personnel use to measure soluble and total lead in a waste stream without use of excessive manpower or capital. QA/QC procedures were included as well as the use of performance samples.

Supervision of analytical technological advances that lead to either patents and new products in the fields of coal tar chemicals, food packaging and transformer manufacturing.

Publications

- Smith, J., A. Weston, and C. Wezwick, "Tire Cord Emission Studies. Conclusion". The International Society of Industrial Yarn Manufacturers, Savannah, Georgia, 3-4 November 1977.
- Hanrahan, J., E. McCarthy, D. Richton, J. Smith, and A. Weston, "Identification of an Interfering Compound in the Determination of Dimethylnitrosamine by Gas Chromatography-Mass Spectrometry". 26th Annual Conference on Mass Spectrometry and Allied Topics, St. Louis, Missouri, 28 May to 2 June 1978.
- Brozowski, E., D. Jerolamon, D. Richton, D. Smith, J. Smith, and A. Weston, "Industrial Applications of Chemical Ionization with the Ammonium Ion". 26th Annual Conference on Mass Spectrometry and Allied Topics, St. Louis, Missouri, 28 May to 2 June 1978.
- Mueller, B.W., L. Palmer, G. Rebyak, and J. Smith, "Analysis of Alpha and Beta Naphthalene Sulfonic Acids by High Performance Liquid Chromatography". North Jersey A.C.A. Chromatography Discussion Group, Nutley, New Jersey, 14 March 1979.
- French, C., L. Palmer, and J. Smith, "Analysis of Polymer Oligomers by High Performance Liquid Chromatography". Middle Atlantic Regional A.C.S. Meeting, West Long Branch, New Jersey, 19-23 March 1979.
- Burkitt, D. and J. Smith, "A Simple Chromatographic Modification Providing for Rapid Interchange of Capillary and Packed Columns". Middle Atlantic Regional A.C.S. Meeting, West Long Branch, New Jersey, 19-23 March 1979.
- Brozowski, E., D. Jerolamon, D. Richton, D. Smith, and J. Smith, "A Convenient Method for the Evaporation of Solvent in the Priority Pollutant Program." Middle Atlantic Regional A.C.S. Meeting, West Long Branch, New Jersey, 19-23 March 1979.
- Mady, N., D. Smith, J. Smith, and C. Wezwick, "The Analysis of Kepone in Biological Samples". Proceedings of the 9th Materials Research Symposium, Gaithersburg, Maryland, 10-12 April 1978.
- Mueller, B., L. Palmer, and J. Smith, "A High Performance Liquid Chromatographic Method for the Analysis of Bis-phenol-A and Its Impurities". Middle Atlantic Regional A.C.S. Meeting, West Long Branch, New Jersey, 19-23 March 1979.
- Gabriel, M., J. Hanrahan, and J. Smith, "A Sensitive Method for the Quantitative Analysis of Pyridine at the Low PPM Level". Middle Atlantic Regional A.C.S. Meeting, West Long Branch, New Jersey, 19-23 March 1979.
- Burkitt, D., J. Hanrahan, and J. Smith, "Analysis of Hexachloroacetone and Hexafluoroacetone in Industrial Wastewater". Proceedings of the A.S.T.M. Committee D-19 Symposium, "The Measurement of Organic Pollutants in Water and Wastewater". Denver, Colorado, 19-20 June 1978.
- Brozowski, E., D. Burkitt, M. Gabriel, E. McCarthy, J. Hanrahan, and J. Smith, "A Simple, Sensitive Method for the Quantitative Analysis of Carbon Tetrachloride and Chloroform in Water at the Parts Per Billion Level". Proceedings of the 9th Materials Research Symposium, Gaithersburg, Maryland, 10-12 April 1978.



Theodore F. Them, Ph.D.

Fields of Competence

Inorganic and organic chemistry, instrumental analytical techniques, synthesis of organic chemicals, laboratory management, chemical research and education

Experience Summary

Nine years experience in inorganic and organic chemistry with strong synthetic organic and instrumental analytical background. Experienced researcher and teacher. Background in conceptualizing, founding, effecting, and administering a chemical consulting firm.

Credentials

M.S. Chemistry—University of New Mexico (1975)

Ph.D. Chemistry—University of New Mexico (1977)

American Chemical Society

The Society of Sigma Xi

Southwest Association of Forensic Scientists—Associate Member

Society of Applied Spectroscopy, Rio Grande Section

Employment History

1982-Present	WESTON
1981-1982	Bell Petroleum Services, Inc.
1982-1982	Bell Petroleum Laboratories
1977-1981	AnaChem, Inc. Co-Founder, Vice President
1975-1977	University of New Mexico

Practical Experience

Familiarity with use, maintenance, and operation of gas chromatographs with flame ionization, electron capture,

thermal conductivity, and photoionization detectors. Experience includes methods development, separation optimization, and data reduction.

Familiarity with use, maintenance, and operation of gas chromatograph/mass spectrometer/data system (GC/MS/DS) in separations and identifications of complex mixtures and molecules. Experience includes methods development, separation enhancement, packed and capillary column techniques, and data reduction.

Familiarity with use and operation of various infrared, nuclear magnetic resonance (NMR), atomic absorption (AA), and liquid chromatographic (LC) instrumentation.

Familiarity with use, maintenance, and operation of Tekmar Models LSC-2 and ALS purge/trap and liquid sample concentrator devices and associated gas chromatographic methods.

Familiarity with use, maintenance, and operation of Fisher Model 490 Coal Analyzer for analysis of moisture, volatiles and ash in coal.

Familiarity with use, maintenance, and operation of Fisher Sulfur Analyzer System for analysis of sulfur in coal and hydrocarbon fuels.

Familiarity with use, maintenance, and operation of Parr Adiabatic Bomb Calorimeter and associated Master Controller in calorimetric analysis of coal and coke, foodstuffs, and fuels.

Familiarity with use, maintenance, and operation of Fisher Models Titralyzer II (Fixed End Point) and Tritrimeter II automatic titration systems for analysis of water by pH or millivolt-sensitive methods.

Publications

Hazardous Properties and Environmental Effects of Materials Used in Solar Heating and Cooling (SHAC) Technologies: Interim Handbook, J.Q. Search (ed.), August 1978. Sandia Laboratories report Sand 78-0842, available from National Technical Information Service, Springfield, Virginia.

Professional Profile

"Isomerism in Complexes of Bidentate Ligands with Enantiotopic Donor Atoms". R.E. Tapscott, J.D. Mather, and T.F. Them, *Coordination Chemistry Reviews*, Vol. 19, Nos. 2/3, September 1979.

"Stereochemical Studies on Diastereomers of Tris (2,3-butanediamine)-Cobalt (III)". C.J. Hilleary, T.F. Them, R.E. Tapscott, *Inorganic Chemistry*, Vol. 19, No. 102, 1980.

"Staying Abreast of PCB Regulations: TESTING". R.M. Holland and T.F. Them, *Professional Trade Publication*, June 1980.

"Stereochemistry of Arsenic (III) and Antimony (III) 1,2-DihydroxyEcyohexane-1,2-dicarboxylates." D. Marcovich, E.N. Duesler, R.E. Tapscott, and T.F. Them, *Inorganic Chemistry*, 1982.

Alison L. Dunn

Fields of Competence

Groundwater flow system analysis and numerical modelling, groundwater contamination assessment and remediation, hydrogeologic evaluation of solid and hazardous waste sites, water supply and recovery well design and testing, monitor well network design and implementation, sampling of soil and water for conventional and hazardous chemical compounds.

Experience Summary

Three years experience as field hydrogeologist and project geologist in industrial and hazardous waste disposal site investigations including two Superfund sites and in inventories and assessments of various classes of injection wells. Three years of graduate research in hydraulic properties of shales and mudstones, watershed hydrology and coastal hydrogeology including practical applications of numerical groundwater flow models.

Credentials

B.A. Geology—Mount Holyoke College (1976)
M.S. Hydrogeology—University of Arizona (1981)
National Water Well Association, Technical Division
American Geophysical Union, Hydrology Division

Employment History

1984 Present	WESTON
1981-1984	SMC Martin Inc.
1976-1981	University of Arizona, Dept. of Hydrology, Environmental Research Laboratory and Office of Arid Land Studies
1978 (Summer)	Office of the State Geologist, Montpelier, VT

Key Projects

Field evaluation of potential groundwater contamination at an Air Force Base in California, including monitor

well installation and sample collection, analysis of hydrogeologic and chemical data.

Site assessment and remediation at an uncontrolled hazardous waste disposal site in New Jersey including field sampling of highly contaminated groundwater and soils, conceptual development of site remediation measures, and testing of remedial measures on a computer groundwater flow model.

Hydrogeologic investigation of a 50-acre site for impact of past electronic components manufacturing operations on ground and surface water.

Evaluation of the effect of placing an innovative top seal for closure of a 25-acre municipal landfill, including analysis of long-term hydrogeologic and geochemical conditions.

Site assessment and remediation at an uncontrolled hazardous waste disposal site in Ohio including a metal detector survey for buried drums, soil sampling, drilling and well construction, supervision, well logging, and data analysis.

Evaluation of surface seepage from a 3-acre wastewater lagoon, including water level monitoring and a detailed water budget.

Publications

Trichloroethylene Occurrence and Ground Water Restoration in Highly Anisotropic Bedrock: A Case Study. Co-author David L. Kraus in Proceedings of the Third National Symposium and Exposition on Aquifer Restoration and Groundwater Monitoring, National Water Well Association, Columbus, OH, 1983.

The Impact of Top Sealing on the Windham, Connecticut Landfill. Co-authors R.M. Schuller and W.W. Beck, Jr. in Proceedings of the 9th Annual U.S. EPA MERL/SHWRD Conference, 1983.

Leachate Quality Improvements after Top Sealing. Co-authors W.W. Beck, Jr. and G.H. Emrich in Proceedings of the 8th Annual U.S. EPA MERL/SHWRD Conference, 1982.

"Preliminary Assessment of the Hydrologic Environment of Klamath Marsh, Oregon." Co-authors M.E. Norvelle, S.L. Vierek, and S. Ince. *NADSAT Project Completion Report No. 31*, Office of Arid Land Studies, University of Arizona, 1981, 71 p.

"A Study of Salinity in Effluent Lakes, Puerto Penasco, Sonora, Mexico." *Hydrology and Water Resources in Arizona and the Southwest*, American Water Resources Association, Arizona Section, 1981.

"Analysis of a Saline Ground-Water Flow System in Puerto Penasco, Sonora, Mexico." Presented to the Cordilleran Section Meeting of the Geological Society of America, March, 1981.

"A Bibliography of Vermont Geology." Compiled with Charles A. Ratte and Diane Vanacek, Office of the State Geologist, Montpelier, Vermont, 1980.



Glenn R. Smart

Fields of Competence

Hydrogeologic investigations of potential hazardous waste sites and landfills; design and supervision of installation of groundwater monitoring programs; collection of field data and evaluation of potential environmental impact; management of hydrogeologic projects at hazardous waste sites

Experience Summary

Seven years of experience in various aspects of the water resource industry. Involvement in over 100 hazardous waste projects in sixteen states. Development of hazardous waste site preliminary assessments and full field investigations. Development of site safety plans for use during hazardous waste site evaluations. Fully trained in the use of respiratory protective equipment, emergency first aid procedures, site sampling protocols and chain-of-custody procedures, and general site safety programs. Frequent interaction with government and industrial clients. Provided expert testimony for superfund litigation.

Employed remote sensing techniques and on-site investigations to locate favorable sites for the development of groundwater supplies. Collected field data, compiled hydrologic and hydraulic input, prepared reports for flood insurance studies. Presented study results to federal, state and local authorities.

Credentials

B.S., Hydrology—University of New Hampshire (1977)
National Water Well Association, Technical Division
American Water Resource Association

Employment History

1984-Present	WESTON
1979-1984	Ecology and Environment, Inc.
1977-1979	Sverdrup & Parcel and Associates, Inc.

Key Projects

Project Manager for Superfund site hydrogeologic investigation to determine potential impact on local well water supplies

Project Manager for complete hydrogeologic investigation of Superfund site involving alleged contamination of municipal field

Project Manager for confidential industrial client. Project included hydrogeologic study to determine the groundwater quality beneath site slated for industrial development

Supervised a team of six field geologists and participated in collection of geologic data for nationwide mineral survey. Responsible for all planning, logistics, quality assurance and financial control of the team.

Designed shallow water table study to assess impact of past waste disposal practices of confidential client

Designed and supervised installation of numerous groundwater monitoring programs at hazardous waste sites.

Publications

Hagger, C.L.D., and G.R. Smart, "Drilling and Installation of Groundwater Monitoring Wells on Hazardous Waste Sites: Construction Specifications and Preparations for Non-ideal Field Conditions." Paper presented to Northeast Conference on the Impact of Waste Storage and Disposal on Groundwater Resources, Ithaca, New York, July, 1982.

Smart, G.R., "A Cost-Effective Approach to Monitoring Well Installation." Paper presented to Triangle Conference on Environmental Technology, University of North Carolina at Chapel Hill, North Carolina, April, 1983.

Smart, G.R., "Installation of Monitoring Wells at Hazardous Waste Sites." Paper presented to 1983 Spill Control and Hazardous Materials Conference, New Haven, Connecticut, 1983

Smart, G.R., "Design of Monitoring Well Systems to Meet RCRA Requirements." Presented at the HMCRI Waste Site Conference, Houston, Texas, March, 1984

Steven I. Michelson

Registration

Engineer-In-Training

Fields of Competence

Field investigations; groundwater resource evaluations; hydrogeologic investigations of landfills and potential water resource impacts; geologic mapping, regional and local structural and geomorphological analyses; microscopic identification of minerals; foundation and structural concrete design; surveying; analysis of soil stability and mechanics; small systems analysis and design, CPM generation, Fortran IV program design and analysis.

Employment History

1983-Present	WESTON
1982	Getty Refining and Marketing
1981	Geological Mapping and Interpretation Wyoming-Idaho Rockies

Credentials

B.S., Geology—Lehigh University (1982)
B.S., Civil Engineering—Lehigh University (1982)

Key Projects

Assisted in the evaluation of contaminant migration to a future Bedford, Massachusetts well-water site as part of a U.S. Air Force-sponsored project at Hanscom Field.

Technical supervision and participation in the scheduled operation and disassembly of pilot treatment plant. Conducted sampling and field studies in support of pilot treatment unit.

Participated in procedural design and operation of field sampling and analysis of a chemical waste impoundment

Assisted in design and evaluation for fresh water storage in Lincoln, New Hampshire. Organized written and plan specifications for contract bidding.

Assisted in the evaluation of the environmental impact of present landfill leachate and seepage

Conducted site design, evaluation and construction cost estimations for wastewater treatment plant in North Andover, Massachusetts.

Professional Profile

Fields of Competence

Sampling of groundwater, wastewater, soils and air; chain-of-custody protocols; operation, calibration and maintenance of laboratory and field sampling equipment and analytical equipment. Laboratory analysis of water and gas samples ranging from wet chemistry to automated instrumental methods.

Experience Summary

Seven years laboratory and field experience including environmental water sampling, soil sampling, and air pollution testing. Experience in developing and fabricating equipment for groundwater sampling. Analytical laboratory experience in water, wastewater, and air pollution. Process and quality control analysis for laboratory serving an agricultural chemicals complex.

Credentials

B.A. — California State College Stanislaus

Additional courses in Chemistry

Employment History

1983-Present	WESTON
1982-1983	J. R. Simplot Company
1978-1982	Occidental Chemical Company
1976-1978	Valley Nitrogen Producers

Key Projects

Occidental Chemical Company: Sampling of groundwater monitoring wells, domestic wells, and city-maintained wells for trace organic and inorganic analysis.

Inorganic analysis of water and soil.

Assisted in planning a new laboratory facility including startup and certification.

APPENDIX D

MONITOR WELL LOGS
AND CONSTRUCTION DIAGRAMS



SKETCH MAP

DRILLING LOG

WELL NUMBER MAFB-1 OWNER USAF
 LOCATION ACW Area ADDRESS Mather AFB

 TOTAL DEPTH 128'
 SURFACE ELEVATION _____ WATER LEVEL _____
 DRILLING COMPANY Stang DRILLING METHOD Mud Rotary DATE DRILLED: 3/7/84
 DRILLER J. Kirby HELPER _____
 LOG BY SIM/ALD

NOTES:

DEPTH FEET	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE SIZE	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
0					0-8" brown Gravelly topsoil
					8"-2' red CLAY and SILT some Sand and Gravel (dense)
					2'-6' red SILT & GRAVEL, little Clay, Sand, Cobbles
10					6'-15' Orange-red SILT and SAND, some gravel, little clay
					15-41' brown, GRAVEL and COBBLES, some SAND
					little Silt and Clay (very difficult drilling)
20					
30					
40					



DRILLING LOG

WELL NUMBER: MAFB-1 OWNER: _____
 LOCATION: _____ ADDRESS: _____
 _____ TOTAL DEPTH: _____
 SURFACE ELEVATION: _____ WATER LEVEL: _____
 DRILLING COMPANY: _____ DRILLING METHOD: _____ DATE DRILLED: _____
 DRILLER: _____ HELPER: _____
 LOG BY: _____

SKETCH MAP

NOTES:

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS**	DESCRIPTION / SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
40					41'-51' GRAVEL and COBBLES, layers of soft tan Silt and Clay less than 1' thick with little Sand (could be cemented)
50					
60					51'-76' Tan, CLAY, some Silt, little fine Sand, few bands of Gravel less than 1' thick
70					
80					76'-110' CLAY and SILT little fine Sand, trace Gravel (clean)



DRILLING LOG

WELL NUMBER: MAFB-1 OWNER: _____
 LOCATION: _____ ADDRESS: _____
 _____ TOTAL DEPTH: _____
 SURFACE ELEVATION: _____ WATER LEVEL: _____

DRILLING COMPANY: _____ DRILLING METHOD: _____ DATE DRILLED: _____
 DRILLER: _____ HELPER: _____

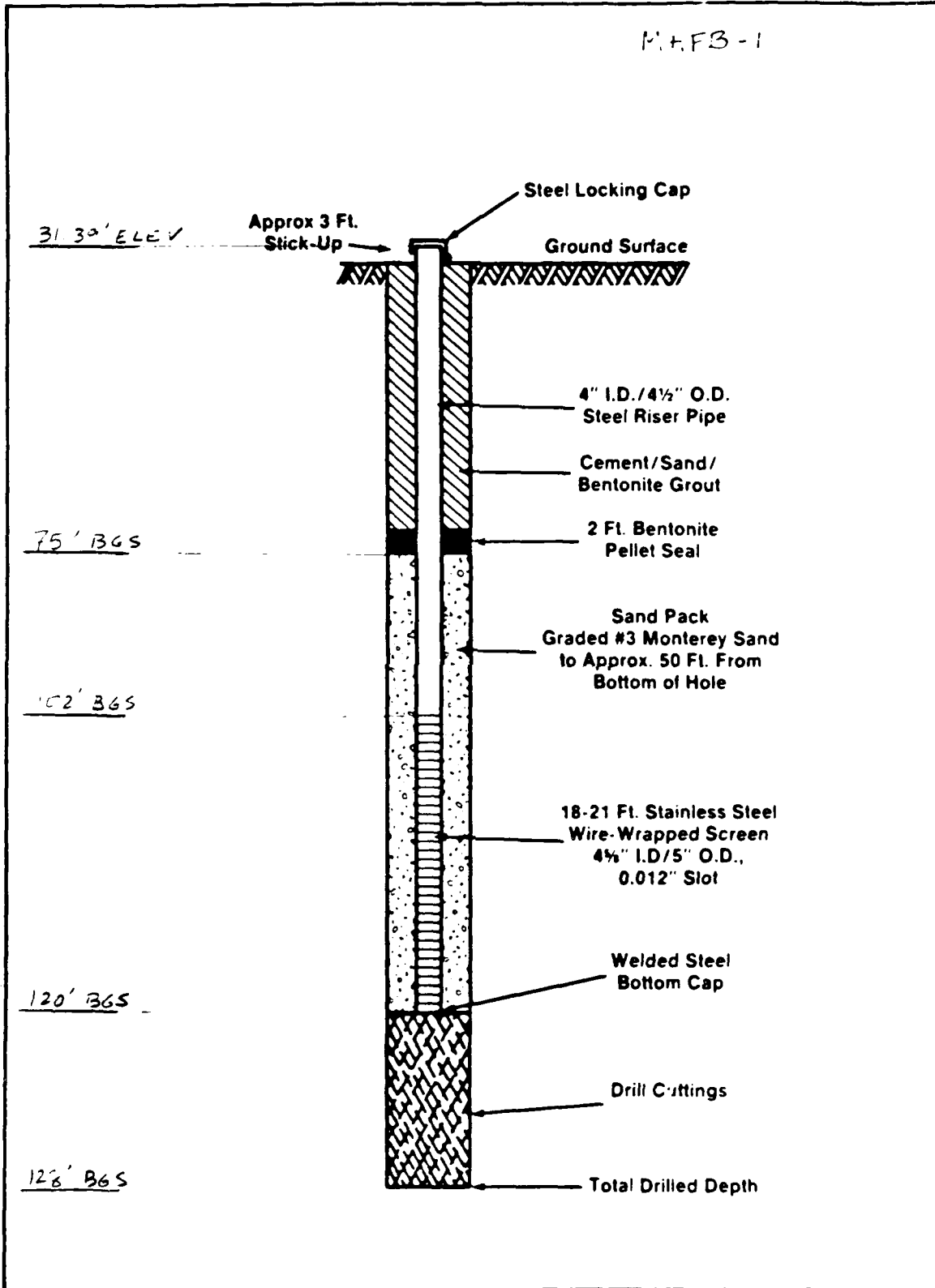
LOG BY: _____

SKETCH MAP

NOTES:

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
80					
90					
100					
110					110'-126' brown, SILT and coarse GRAVEL, some Clay and Sand (dense)
120					126'-128' brown, fine SAND and SILT, some Gravel
					128' End of Hole

M.F.B-1





DRILLING LOG

WELL NUMBER MAFB-2 OWNER USAF
 LOCATION ACW Area ADDRESS Mather AFB

 _____ TOTAL DEPTH 131'
 SURFACE ELEVATION _____ WATER LEVEL _____
 DRILLING COMPANY Stang DRILLING METHOD Mud Rotary DATE DRILLED 3/3/84
 DRILLER Cliff Patrick HELPER _____

LOG BY SIM/ALD

SKETCH MAP

NOTES

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
0					0-1½' red fine-medium SAND, some Silt/Clay
					1½-2' red CLAY, little Silt
					2-6' red fine to medium SAND and GRAVEL some Silt and Clay
10					
					6-15' red, GRAVEL and COBBLES little Silt and Clay (difficult drilling)
20					
30					
40					



DRILLING LOG

WELL NUMBER MAFB-2 OWNER _____
 LOCATION _____ ADDRESS _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION _____ WATER LEVEL _____
 DRILLING COMPANY _____ DRILLING METHOD _____ DATE DRILLED _____
 DRILLER _____ HELPER _____
 LOG BY _____

SKETCH MAP

NOTES

DEPTH FEET	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION SOIL CLASSIFICATION (COLOR TEXTURE STRUCTURES)
40					
					15-59' red-brown
					GRAVEL and COBBLES, clean
50					trace fine Sand and Silt
					(very difficult drilling)
60					59-65' brown, CLAY and SILT, some Gravel
					little Sand (easy drilling)
					65-66' COBBLES (difficult drilling)
					66-68' brown, CLAY, some Sand, little Silt and Gravel
					trace Cobbles
70					68-80 COBBLES, GRAVEL, SAND
					some Silt and Clay (difficult drilling)
80					



DRILLING LOG

WELL NUMBER MAFB-2 OWNER _____
 LOCATION _____ ADDRESS _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION _____ WATER LEVEL _____
 DRILLING COMPANY _____ DRILLING METHOD _____ DATE DRILLED _____
 DRILLER _____ HELPER _____
 LOG BY _____

SKETCH MAP

NOTES

DEPTH FEET	DIAPHRAGM NO.	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWN	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE STRUCTURES)
80					80-100' brown SILT and CLAY and SAND little Gravel gravel content decreasing with depth (smooth drilling)
90					
100					100-131' brown, fine SAND, SILT, CLAY little fine Gravel Gravel content increasing with depth
110					
120					131' End of Hole

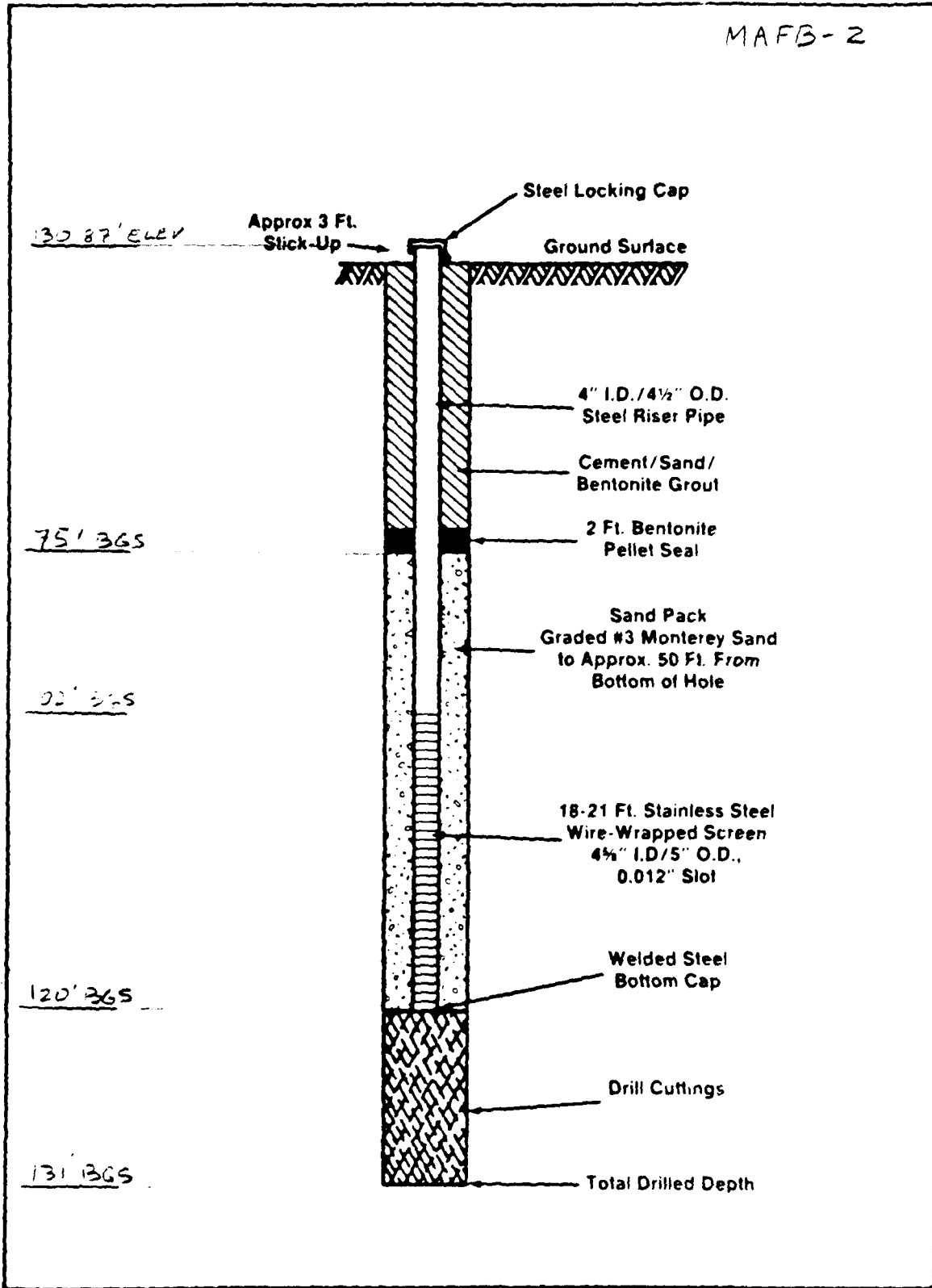
120

ASTM D1586

D-7

SHEET 3 OF 3

MAFB-2





DRILLING LOG

WELL NUMBER MAFB-3 OWNER USAF
LOCATION ACW Area ADDRESS Mather AFB

TOTAL DEPTH 136'
SURFACE ELEVATION: _____ WATER LEVEL _____
Mud

DRILLING COMPANY Stang DRILLING METHOD Rotary DATE DRILLED 3/8/84
DRILLER: J. Kirby HELPER _____

LOG BY SIM/ALD

DRAWING MAP

NOTES

DEPTH FEET	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS*	DESCRIPTION SOIL CLASSIFICATION
					(COLOR TEXTURE STRUCTURES)
0					0-3' red-brown TOPSOIL CLAY, some Sand and Gravel
					3-8' red, SILT and COBBLES little Sand local green mottling
10					8-24' COBBLES and GRAVEL little Silt, Clay, and Sand (difficult drilling)
20					24-30' brown, fine SAND little Silt, trace fine Gravel (asy drilling)
30					30-36' COBBLES and GRAVEL little Silt, Clay, and Sand (difficult drilling)
40					36-42' brown-grav fine SAND and SILT, little Clay trace fine Gravel (hard pan)



DRILLING LOG

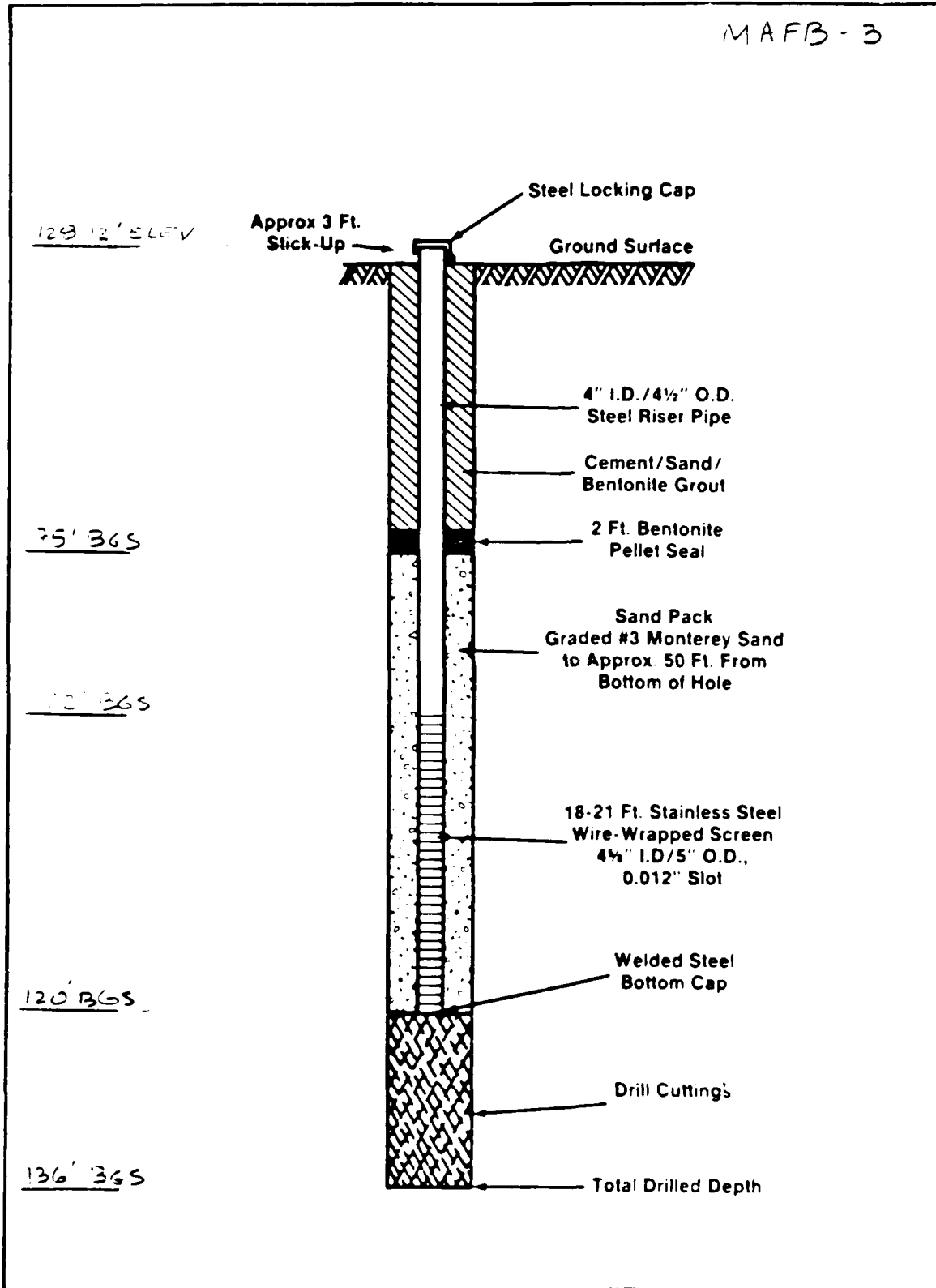
WELL NUMBER MAFB-3 OWNER _____
 LOCATION _____ ADDRESS _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION _____ WATER LEVEL _____
 DRILLING COMPANY _____ DRILLING METHOD _____ DATE DRILLED _____
 DRILLER _____ HELPER _____
 LOG BY _____

SKETCH MAP

NOTES

DEPTH FEET	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION SOIL CLASSIFICATION
					(COLOR TEXTURE STRUCTURES)
40					42-44' COBBLES and GRAVEL and SAND little Silt and Clay (difficult drilling)
50					44-66' brown-gray SILT and CLAY trace fine Gravel little fine Sand Gravel content increasing with depth (easy drilling)
60					66-75' SAND - (gradual from above, SAND content increasing with depth) little Gravel little Clay, Silt (smooth, easy drilling)
70					
80					
90					
30					

MAFB-3





DRILLING LOG

WELL NUMBER MAFB-4 OWNER USAF
 LOCATION NE Perimeter near Mather Lake ADDRESS Mather AFB
 TOTAL DEPTH 166'
 SURFACE ELEVATION _____ WATER LEVEL: _____
 Mud
 DRILLING COMPANY Stang DRILLING METHOD Rotary DATE DRILLED 2/29/84
 DRILLER Cliff Patrick HELPER: _____
 LOG BY SIM/ALD

SKETCH MAP

NOTES

DEPTH FEET	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
0					0-3" brown top soil
					3-5" green SILT, little fine Sand
10					5"-18' red, GRAVEL some Silt
20					18-42' brown to tan SILT little Clay trace fine Sand
30					
40					



DRILLING LOG

WELL NUMBER MAFB-4 OWNER _____
 LOCATION _____ ADDRESS: _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION _____ WATER LEVEL: _____
 DRILLING COMPANY _____ DRILLING METHOD _____ DATE DRILLED _____
 DRILLER _____ HELPER: _____
 LOG BY _____

SKETCH MAP

NOTES:

DEPTH FEET	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS*	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
40					42-45' tan SILT, little Gravel and Clay trace fine Sand
50					45-58' tan SAND and GRAVEL, little cobbles trace Silt and Clay few zones of SILT and fine SAND, little Gravel 1'-1½' thick
60					58-78' fine to coarse SAND and fine GRAVEL little Silt trace Clay (red and Cobbles)
70					
80					



DRILLING LOG

WELL NUMBER: MAFB-4 OWNER _____
 LOCATION: _____ ADDRESS _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION: _____ WATER LEVEL: _____
 DRILLING COMPANY: _____ DRILLING METHOD: _____ DATE DRILLED: _____
 DRILLER: _____ HELPER: _____
 LOG BY _____

SKETCH MAP

NOTES:

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
80					
					78-91' SAND and fine GRAVEL little Silt
90					91-99' coarser material GRAVEL and COBBLES trace Silt and Clay few zones of no cobbles - Silt/Clay - about 1' thick (difficult drilling)
100					99-105' CLAY and SILT trace Gravel, Cobbles, Sand (easy drilling)
110					
120					



DRILLING LOG

WELL NUMBER MAFB-4 OWNER: _____
 LOCATION _____ ADDRESS: _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION: _____ WATER LEVEL: _____
 DRILLING COMPANY: _____ DRILLING METHOD: _____ DATE DRILLED: _____
 DRILLER: _____ HELPER: _____
 LOG BY: _____

SKETCH MAP

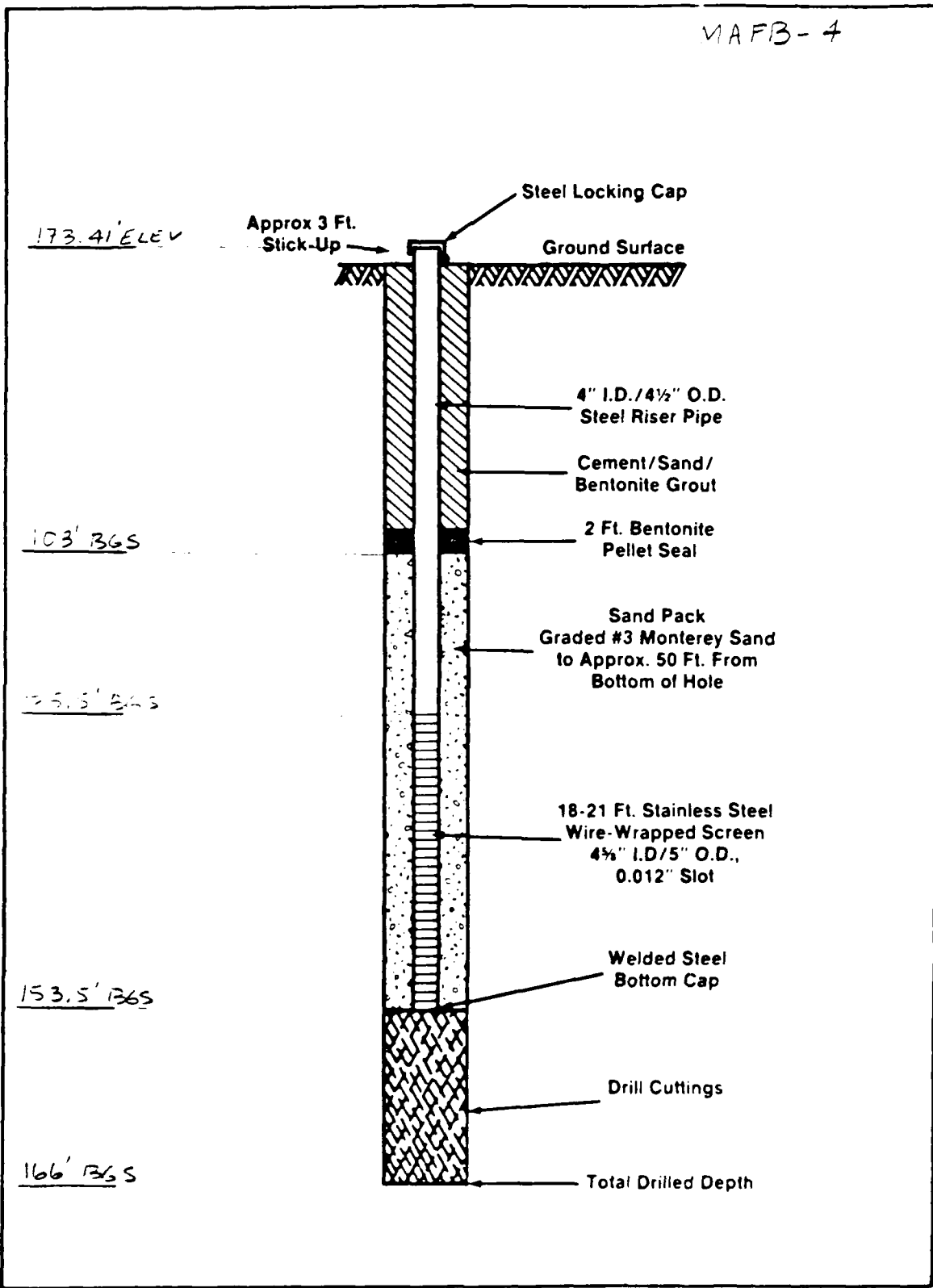
NOTES

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS*	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
120					
130					105'-146' tan, brown CLAY little fine Sand, Gravel and Silt
140					146-166' blue-gray CLAY little fine Sand and Gravel
150					
160					166' End of Hole

ASTM D1586

D-16

SHEET 4 OF 4





SKETCH MAP

NOTES:

DRILLING LOG

WELL NUMBER: MAFB-5 OWNER: USAF
 LOCATION: NE Perimeter near Folsom Canal ADDRESS: Mather AFB
 TOTAL DEPTH: 146'
 SURFACE ELEVATION: _____ WATER LEVEL: _____
 DRILLING COMPANY: Stang DRILLING METHOD: Mud Rotary DATE DRILLED: 3/1/84
 DRILLER: Cliff Patrick HELPER: _____
 LOG BY: SIM/ALD

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS*	DESCRIPTION: SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
0					0-1' red-brown CLAY, little Sand, Gravel, Cobbles
					1-1½' red, stratified Hardpan, SILT, little Gravel, Cobbles, wood fibers and roots (cemented)
					1½-2½' red, SILT, some Gravel and Cobbles, Little Clay (slightly unconsolidated)
10					2½-5' red SILT, little Gravel (massive)
					5-6' green SILT, little Gravel (mottled)
					6-9' red, SILT and CLAY, little Sand and Gravel
					9-16' Cobbles (difficult drilling)
20					16-32' red CLAY and SILT
					little Sand and Gravel
					Gravel and Cobble
					content increasing with depth
30					32-40' GRAVEL and COBBLES (difficult drilling)
40					



DRILLING LOG

WELL NUMBER MAFB-5 OWNER: _____
 LOCATION: _____ ADDRESS: _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION: _____ WATER LEVEL: _____
 DRILLING COMPANY _____ DRILLING METHOD: _____ DATE DRILLED: _____
 DRILLER: _____ HELPER: _____
 LOG BY: _____

SKETCH MAP

NOTES:

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
40					40-48' red tan, SILT and CLAY
					some Gravel, little Cobbles
50					48-60' light brown, clean SILT and CLAY
					little Gravel
60					60-68' light brown, CLAY and SILT
					little Sand
					Some Cobbles (difficult drilling)
70					68-90' brown, SAND and GRAVEL
					some Cobbles
					trace Silt and Clay (difficult drilling)
80					

* ASTM D1586



DRILLING LOG

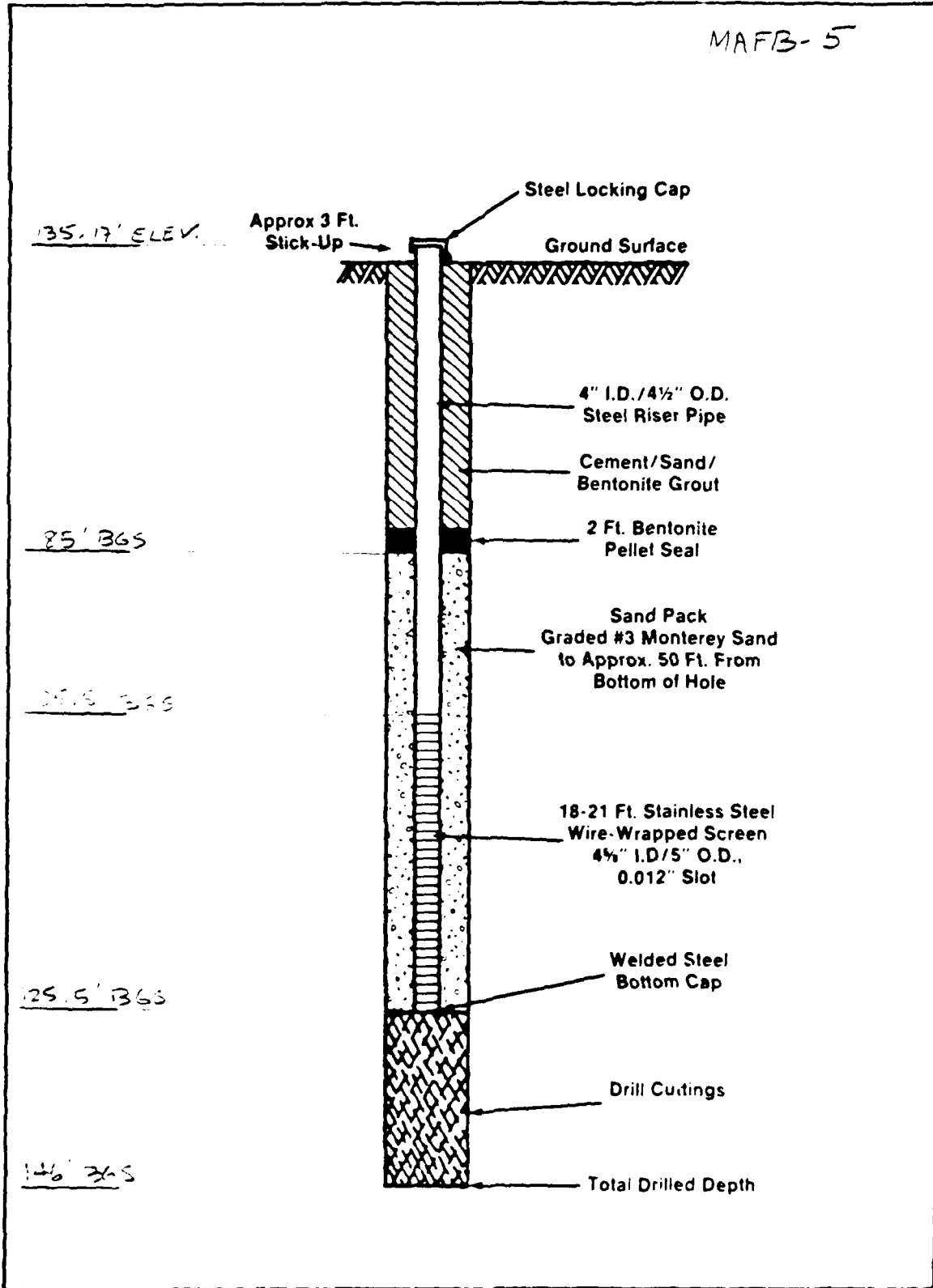
WELL NUMBER MAFB-5 OWNER: _____
 LOCATION _____ ADDRESS: _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION _____ WATER LEVEL: _____
 DRILLING COMPANY: _____ DRILLING METHOD: _____ DATE DRILLED: _____
 DRILLER: _____ HELPER: _____
 LOG BY _____

SKETCH MAP

NOTES:

DEPTH FEET	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION / SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
80					
90					90-93' brown, SAND and GRAVEL, little clay (losing circulation)
					93-98' brown COBBLES, little Gravel, some Clay (probably in a water zone)
100					98-146' brown, CLAY and SILT little Sand and Gravel (easy drilling, dense)
120					146' End of Hole

MAFB-5





SKETCH MAP

DRILLING LOG

WELL NUMBER MAFB-6 OWNER: USAF
 LOCATION: NE Perimeter on ADDRESS: Mather AFB
Main Base
 TOTAL DEPTH 128'
 SURFACE ELEVATION: _____ WATER LEVEL: _____
 DRILLING COMPANY: Stang DRILLING METHOD: Rotary DATE DRILLED: 3/12/84
 DRILLER: J. Kirby HELPER: _____
 LOG BY SIM/ALD

NOTES:

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE STRUCTURES)
0					0-1 1/2' red-brown TOPSOIL
					silt, some coarse Gravel
					1 1/2-8' SILT, some Clay and coarse Gravel
					layers of coarse Gravel, some Clay
10					little Sand
					8-34' COBBLES and GRAVEL
					LITTLE Sand, Silt, and Clay (difficult drilling,
20					may have some cobble cementation)
					34-87' brown, SILT
					some clay, little Sand and fine Gravel
					grading to tan CLAY
					some Silt, little Sand and Gravel
40					



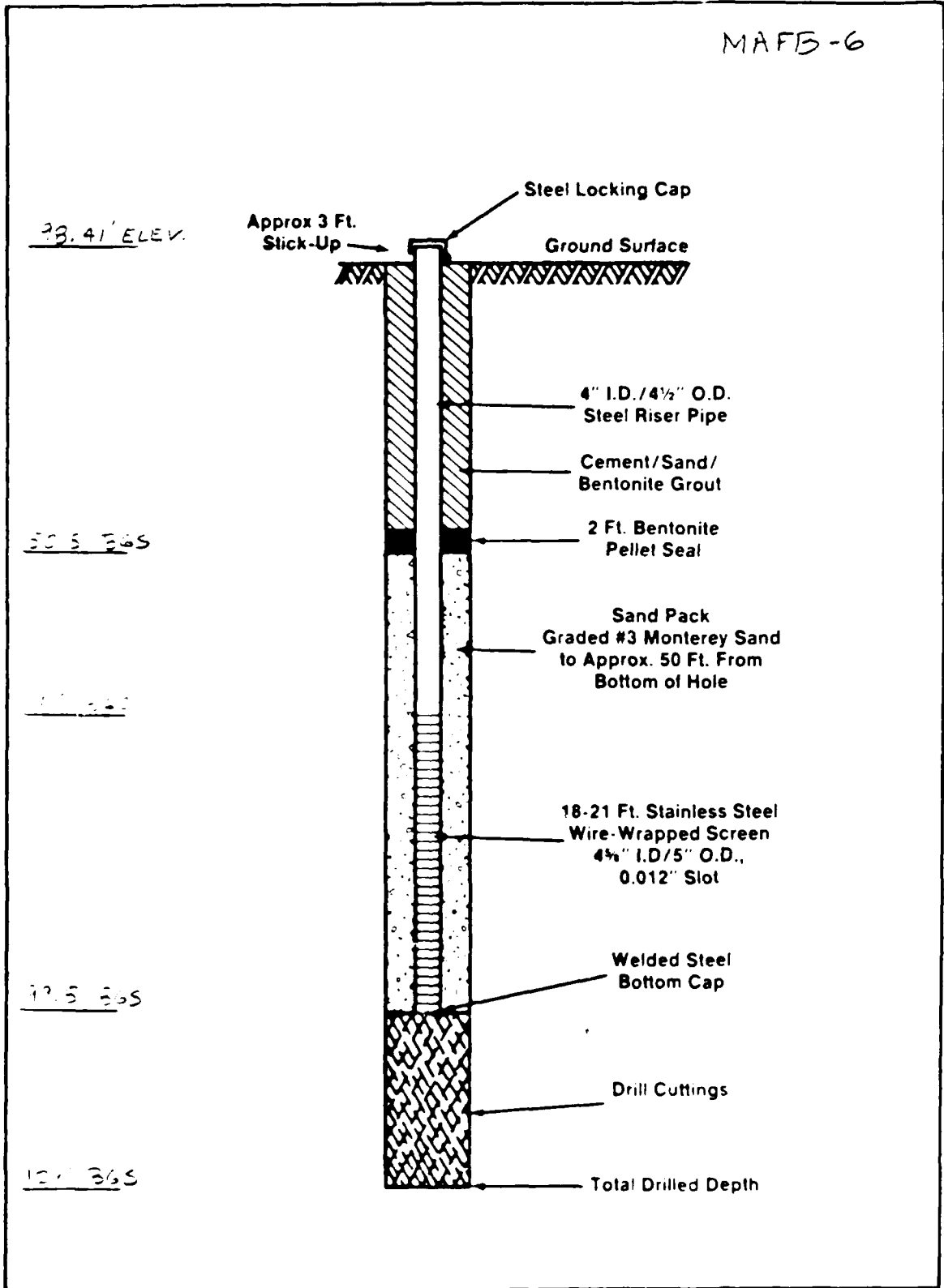
DRILLING LOG

WELL NUMBER MAFB-6 OWNER _____
 LOCATION _____ ADDRESS _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION _____ WATER LEVEL _____
 DRILLING COMPANY _____ DRILLING METHOD _____ DATE DRILLED _____
 DRILLER _____ HELPER _____
 LOG BY: _____

SKETCH MAP

NOTES:

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
80					
90					
100					105-128' GRAVEL
					little Sand
					little Silt and Clay
110					
120					128' End of Hole





DRILLING LOG

WELL NUMBER MAFB-7 OWNER: USAF
 LOCATION 7100 Area ADDRESS: Mather AFB
 _____ TOTAL DEPTH 128'
 SURFACE ELEVATION: _____ WATER LEVEL: _____
 DRILLING COMPANY Stang DRILLING METHOD: Rotary DATE DRILLED: 3/15/84
 DRILLER: J. Kirby HELPER: _____
 LOG BY SIM/ALD

SKETCH MAP

NOTES

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION SOIL CLASSIFICATION
					(COLOR TEXTURE STRUCTURES)
0					0-1' coarse GRAVEL and COBBLES, little Sand and Silt (road base, little demolition debris, asphalt)
					1-4' brown-red, SILT and CLAY, little fine Sand trace fine Gravel (very dense)
10					4-4½' cemented COBBLES & GRAVEL - rounded Clay & Silt matrix
					4½-6' COBBLES, some Gravel, little Silt & Clay matrix (compacted, stratified)
20					6-27' COBBLES and GRAVEL little Sand, Clay, and Silt (difficult drilling)
30					27-40' brownish-gray fine to medium SAND little Silt and Clay, trace Gravel (easy drilling) - organic chemical odor below 30'- HNU: 0-1.5 ppm with probe near drilling mud effluent
40					



DRILLING LOG

WELL NUMBER MAFB-7 OWNER _____
 LOCATION _____ ADDRESS _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION _____ WATER LEVEL _____
 DRILLING COMPANY _____ DRILLING METHOD _____ DATE DRILLED _____
 DRILLER _____ HELPER _____
 LOG BY _____

SKETCH MAP

NOTES

DEPTH FEET	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
40					40-42' CLAY & GRAVEL
					some Silt and little Sand
					42-78' Tan SILT
					some Clay
50					little Sand and Gravel
					grading to
					medium to coarse SAND
					Some Silt and Gravel
					little Clay
60					grading to
					fine to medium SAND and SILT
					little Clay
					(strong chemical odor, solvent or
					fuel, below 50')
70					
80					



DRILLING LOG

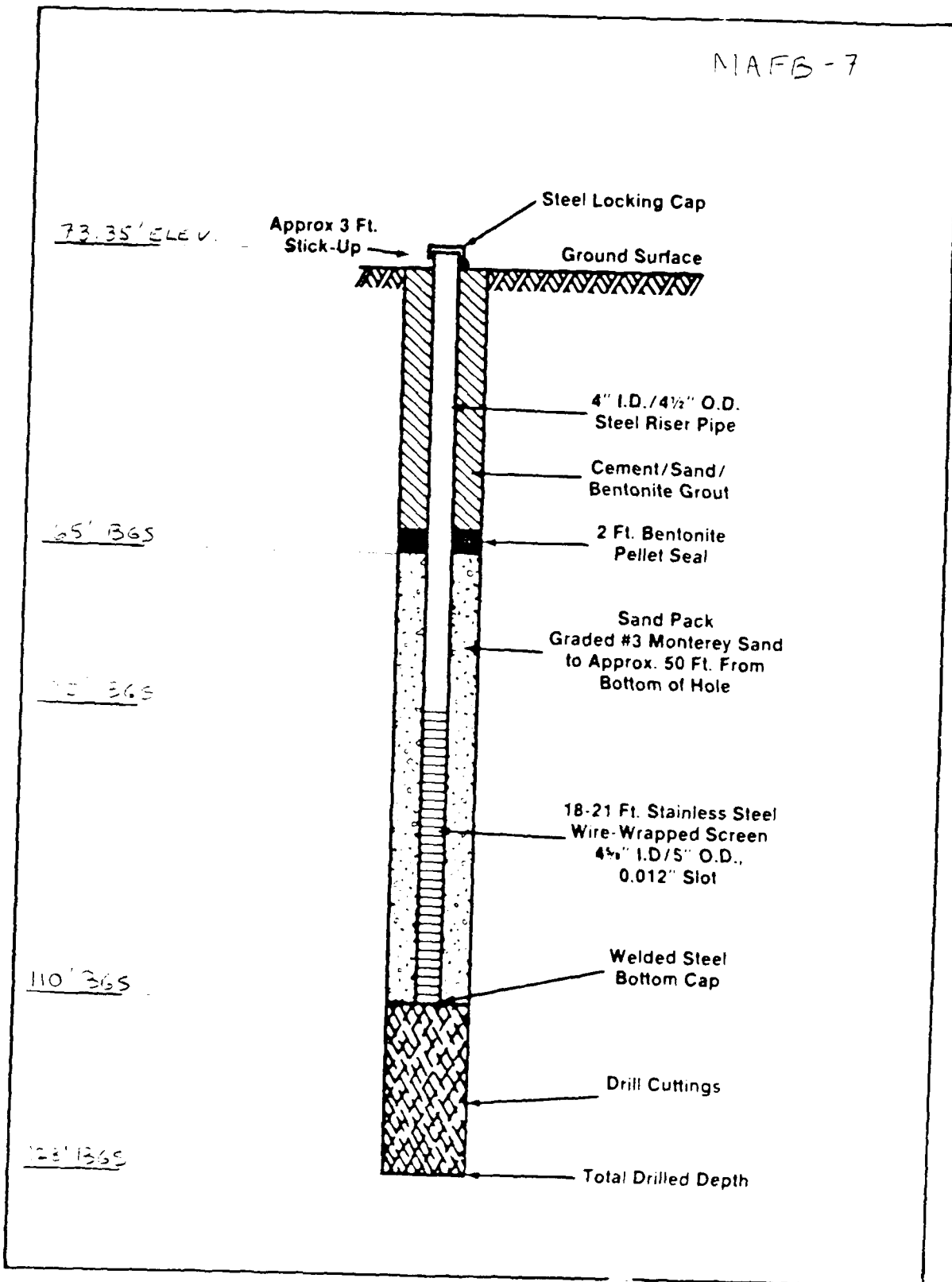
WELL NUMBER: MAFB-7 OWNER: _____
 LOCATION: _____ ADDRESS: _____
 _____ TOTAL DEPTH: _____
 SURFACE ELEVATION: _____ WATER LEVEL: _____
 DRILLING COMPANY: _____ DRILLING METHOD: _____ DATE DRILLED: _____
 DRILLER: _____ HELPER: _____
 LOG BY: _____

SKETCH MAP

NOTES:

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
80					78-106' black medium GRAVEL
					some fine to medium Sand
					tan silt/Clay matrix
					Gravel content increases with
90					depth (smooth drilling)
100					106-115' brown-gray, stiff, CLAY and SILT
					some fine to medium Sand
					little fine Gravel (easy drilling)
110					115-128' tan, soft, SILT
					and fine to medium SAND
					Some clay
					little gravel
120					becomes denser with depth

NAFB-7





SKETCH MAP

DRILLING LOG

WELL NUMBER MAFB-8 OWNER: USAF
 LOCATION: 7100 Area ADDRESS: Mather AFB
 _____ TOTAL DEPTH 128'
 SURFACE ELEVATION _____ WATER LEVEL: _____
 Mud
 DRILLING COMPANY Stang DRILLING METHOD: Rotary DATE DRILLED: 3/17/84
 DRILLER J. Kirby HELPER: _____
 LOG BY SIM/ALD

NOTES:

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS*	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
0					0-8" TOP SOIL
					8"-2 1/2' red brown, fine SAND and SILT little clay roots
					2 1/2'-3' Old pavement zone or asphalt fill
10					3-8' tan, fine, SAND, some clay little Silt
					8-18' fine COBBLES and GRAVEL little Sand trace Silt (moderate drilling difficulty)
20					18-36' COBBLES and GRAVEL little Sand trace Silt (difficult drilling)
30					36-60' brown CLAY some Silt, trace fine Sand occasional zones of gravelly clay and tan Silt, some fine to medium Sand
40					



DRILLING LOG

WELL NUMBER MAFB-8 OWNER: _____
 LOCATION _____ ADDRESS _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION: _____ WATER LEVEL: _____
 DRILLING COMPANY: _____ DRILLING METHOD: _____ DATE DRILLED _____
 DRILLER: _____ HELPER: _____
 LOG BY _____

SKETCH MAP

NOTES

DEPTH FEET	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS *	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
40					
50					
60					60-94' red-brown SILT some Clay fine-medium Sand
70					
80					



DRILLING LOG

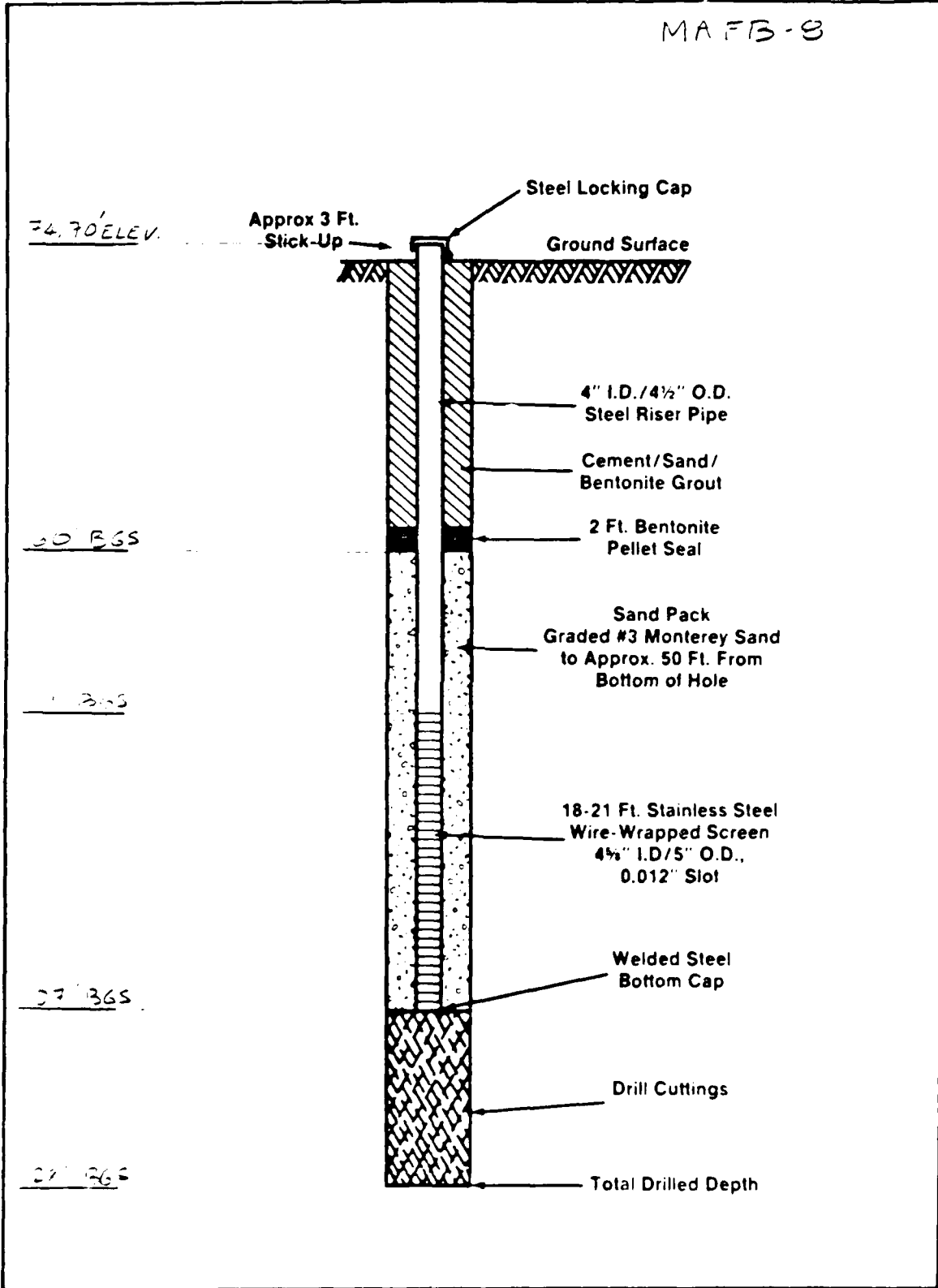
WELL NUMBER MAFB-8 OWNER: _____
 LOCATION _____ ADDRESS: _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION _____ WATER LEVEL: _____
 DRILLING COMPANY: _____ DRILLING METHOD: _____ DATE DRILLED: _____
 DRILLER _____ HELPER: _____
 LOG BY _____

SKETCH MAP

NOTES

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE STRUCTURES)
30					
90					94-107' fine to medium SAND some Gravel and tan Silt little Clay interlavered with CLAY some medium black Gravel
100					
110					107-128' coarse GRAVEL little Sand, Silt and Clay
					128' End of Hole

MAFB-8





DRILLING LOG

WELL NUMBER MAFB-9 OWNER USAF
 LOCATION 7100 Area ADDRESS Mather AFB
 _____ TOTAL DEPTH 128'
 SURFACE ELEVATION _____ WATER LEVEL _____
 DRILLING COMPANY Stang DRILLING METHOD Rotary DATE DRILLED 3/20/84
 DRILLER J. Kirby MUD _____ HELPER _____
 LOG BY SIM/ALD

SKETCH MAP

NOTES

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS	DESCRIPTION, SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
0					0-1' red TOPSOIL
10					1-15' COBBLES and GRAVEL matrix-red, fine to coarse SAND and SILT little Clay
20					15-45' COBBLES and GRAVEL some Sand little Silt, trace Clay (may have some cementation of cobbles to 21').
30					
40					



DRILLING LOG

WELL NUMBER MAFB-9 OWNER _____
 LOCATION _____ ADDRESS _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION _____ WATER LEVEL _____
 DRILLING COMPANY _____ DRILLING METHOD _____ DATE DRILLED _____
 DRILLER _____ HELPER _____
 LOG BY _____

SKETCH MAP

NOTES

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS*	DESCRIPTION - SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
40					
					45-48' tan, SILT some Clay and fine Sand (soft)
50					48-69' tan, SILT some Clay and fine Sand little Gravel becomes coarser with Depth
60					some coarse Sand and Gravel
70					69-96' red-brown SILT and fine to medium SAND little Clay and Gravel
80					

TABLE M-11586

D-35



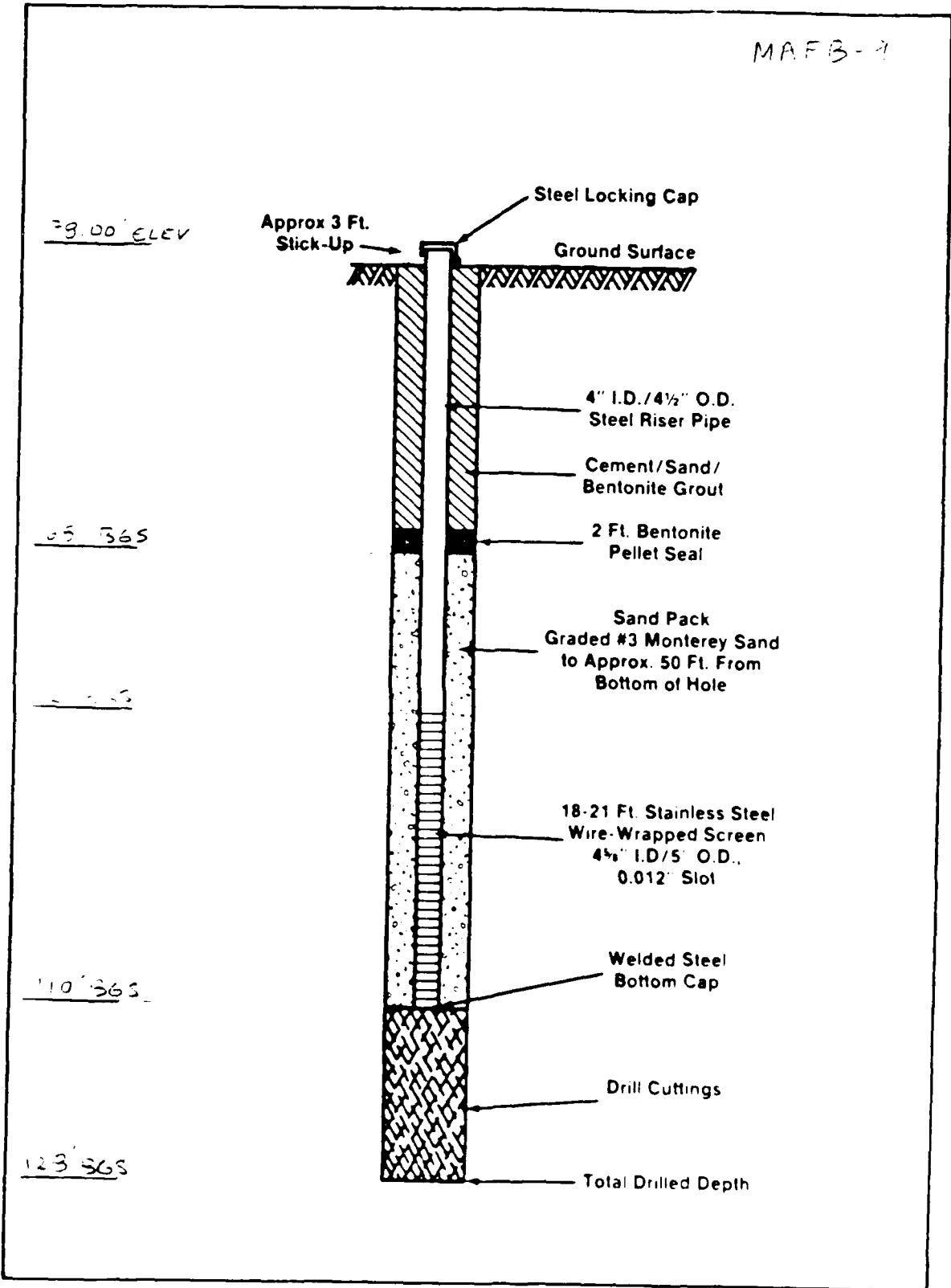
DRILLING LOG

WELL NUMBER: MAFB-9 OWNER: _____
 LOCATION: _____ ADDRESS: _____
 _____ TOTAL DEPTH: _____
 SURFACE ELEVATION: _____ WATER LEVEL: _____
 DRILLING COMPANY: _____ DRILLING METHOD: _____ DATE DRILLED: _____
 DRILLER: _____ HELPER: _____
 LOG BY: _____

SKETCH MAP

NOTES

DEPTH (FEET)	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS*	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
80					
90					96-128' tan, Silt and fine SAND
					little Clay and Gravel
					some Gravel zones
					approx. 6" thick
100					
110					
120					128' End of hole





DRILLING LOG

WELL NUMBER MAFB-10 OWNER USAF
 LOCATION West Ditch ADDRESS Mather AFB
 _____ TOTAL DEPTH 120'
 SURFACE ELEVATION _____ WATER LEVEL _____
 DRILLING COMPANY Stard DRILLING METHOD Rotary DATE DRILLED 8/31/84
 DRILLER J. Kirby HELPER _____
 LOG BY SJM/ALD

SKETCH MAP

NOTES

DEPTH FEET	CORRECTION	CORRECTION	CORRECTION	CORRECTION	CORRECTION	CORRECTION	CORRECTION	CORRECTION	DESCRIPTION SOIL CLASSIFICATION
									COLOR TEXTURE STRUCTURES
0-1'									TOPSOIL
1-3'									red, fine SAND and SILT little clay
3-9'									red, Silt/Clay and fine SAND brittle hardpan
9-18'									COBBLES and GRAVEL matrix-red, fine SAND and SILT little Clay (difficult drilling)
18-42'									COBBLES and GRAVEL-black (mafic) one foot thick zones of brown, dense, fine SAND/SILT little Clay (difficult drilling)



SKETCH MAP

DRILLING LOG

WELL NUMBER MAFB-10 OWNER _____
 LOCATION _____ ADDRESS _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION _____ WATER LEVEL _____
 DRILLING COMPANY _____ DRILLING METHOD _____ DATE DRILLED _____
 DRILLER _____ HELPER _____
 LOG BY _____

NOTES

DEPTH FEET	CORRECTION FEET	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOW COUNT	DESCRIPTION SOIL CLASSIFICATION
					COLOR TEXTURE STRUCTURES
40					42-45' brown, CLAY, little Silt trace fine Sand grading to little Gravel (moderately dense, easy drilling)
					45-50 brown, CLAY little Silt, little Gravel trace fine Sand (easy drilling)
50					50-60' tan, GRAVEL and medium to coarse SAND, matrix - silt/clay, trace fine Sand (difficult drilling)
60					60-63' brown-black, fine SAND, little Silt trace Clay and Gravel (easy drilling)
					63-67' tan, fine to coarse SAND and SILT, trace fine Gravel and Clay (easy drilling)
70					67-77' dark brown, fine to medium SAND and GRAVEL little Silt, trace Clay (easy drilling)
80					



DRILLING LOG

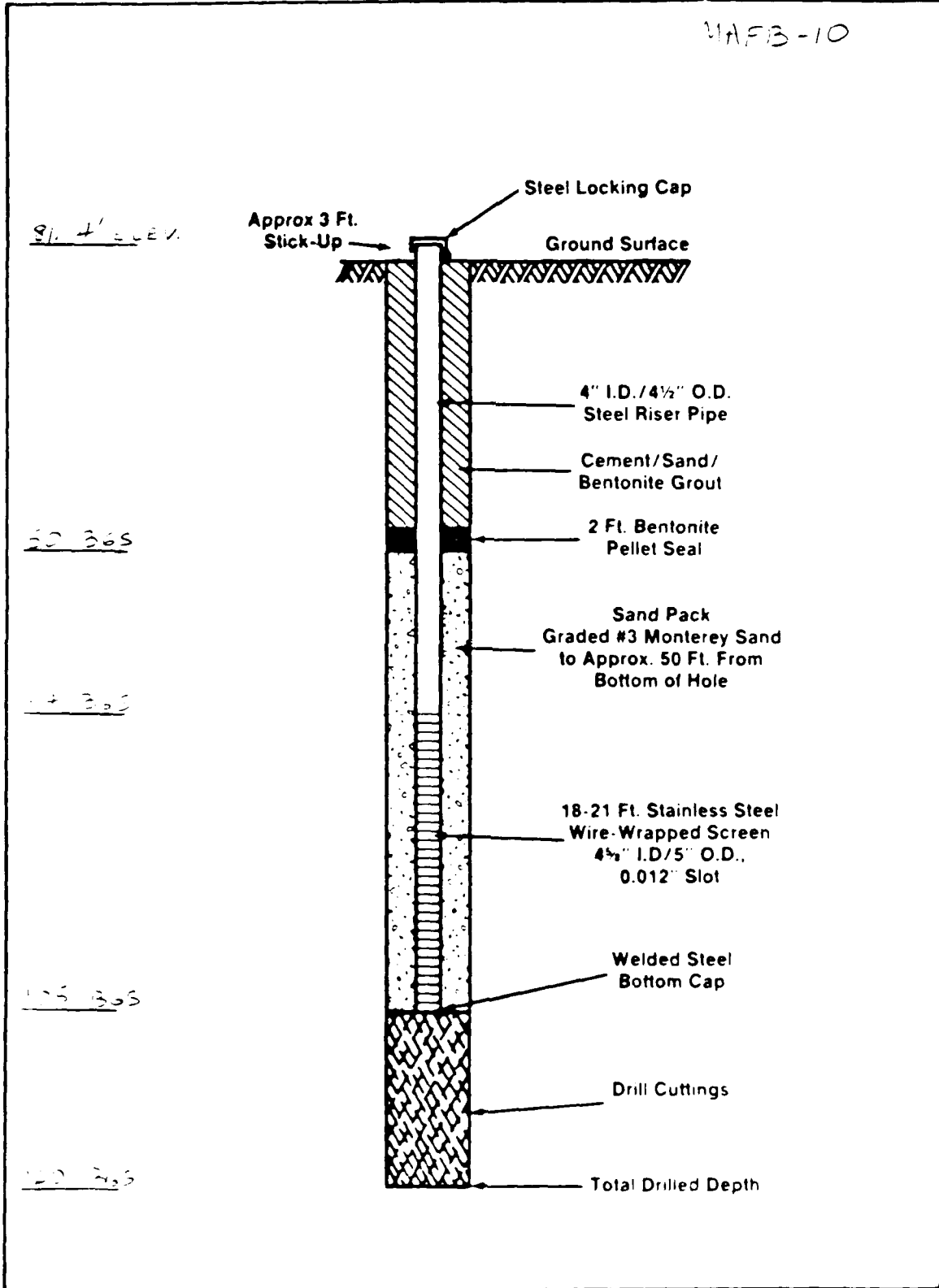
WELL NUMBER MAFB-10 OWNER _____
 LOCATION _____ ADDRESS _____
 _____ TOTAL DEPTH _____
 SURFACE ELEVATION _____ WATER LEVEL _____
 DRILLING COMPANY _____ DRILLING METHOD _____ DATE DRILLED _____
 DRILLER _____ HELPER _____
 LOG BY _____

SKETCH MAP

NOTES

DEPTH FEET	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOW'S	DESCRIPTION SOIL CLASSIFICATION (COLOR TEXTURE STRUCTURES)
80					77-120' brown SILT, CLAY and fine SAND lenses of medium to coarse SAND (easy drilling)
85					
90					
95					
100					
105					
110					
115					
120					120' End of hole

MAFB-10





DRILLING LOG

WELL NUMBER MAFB-11 OWNER USAF
 LOCATION West Ditch ADDRESS Mather AFB
 _____ TOTAL DEPTH 118'
 SURFACE ELEVATION _____ WATER LEVEL _____
 DRILLING COMPANY Stang DRILLING METHOD Rotary DATE DRILLED 3/22/84
 DRILLER J. Kirby HELPER _____
 LOG BY SIM/ALD

SKETCH MAP

NOTES

DEPTH FEET	HAPLIC (0)	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOW'S	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
0					0-6" red-brown TOPSOIL
					6"-3' red, fine SAND and SILT, little Clay
					3-6' red, SILT, little Clay and fine Sand, trace Gravel
					6-42' COBBLES and GRAVEL
					red, Silt and fine Sand matrix
					little Boulders
					Gravel is black, mostly mafic
					(At 19' bit ground on boulder for 105 minutes before breaking through)



DRILLING LOG

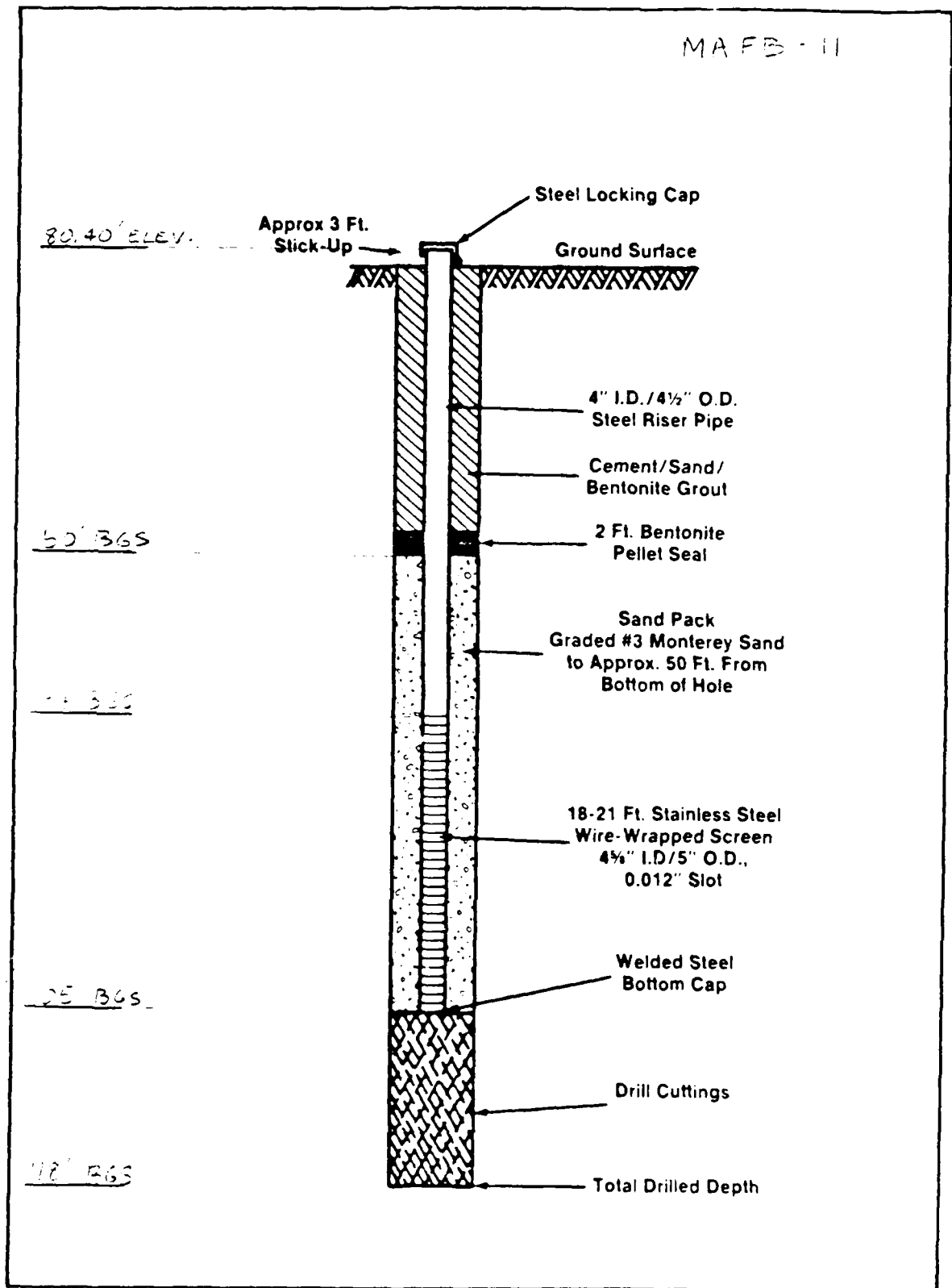
WELL NUMBER MAFB-11 OWNER _____
LOCATION _____ ADDRESS _____

TOTAL DEPTH _____
SURFACE ELEVATION _____ WATER LEVEL _____
DRILLING COMPANY _____ DRILLING METHOD _____ DATE DRILLED: _____
DRILLER _____ HELPER _____
LOG BY _____

SKETCH MAP

NOTES:

DEPTH - FEET	GRAPHIC LOG	SAMPLE NUMBER	SAMPLE TYPE	SAMPLE BLOWS*	DESCRIPTION SOIL CLASSIFICATION (COLOR, TEXTURE, STRUCTURES)
40					
41					
42					
43					
44					
45					
46					42-113 tan. SILT and fine SAND
47					little clay
48					
49					Occasional seams of Gravel
50					
51					
52					
53					
54					
55					
56					
57					
58					
59					
60					
61					



APPENDIX E

AQUIFER RECOVERY TESTS

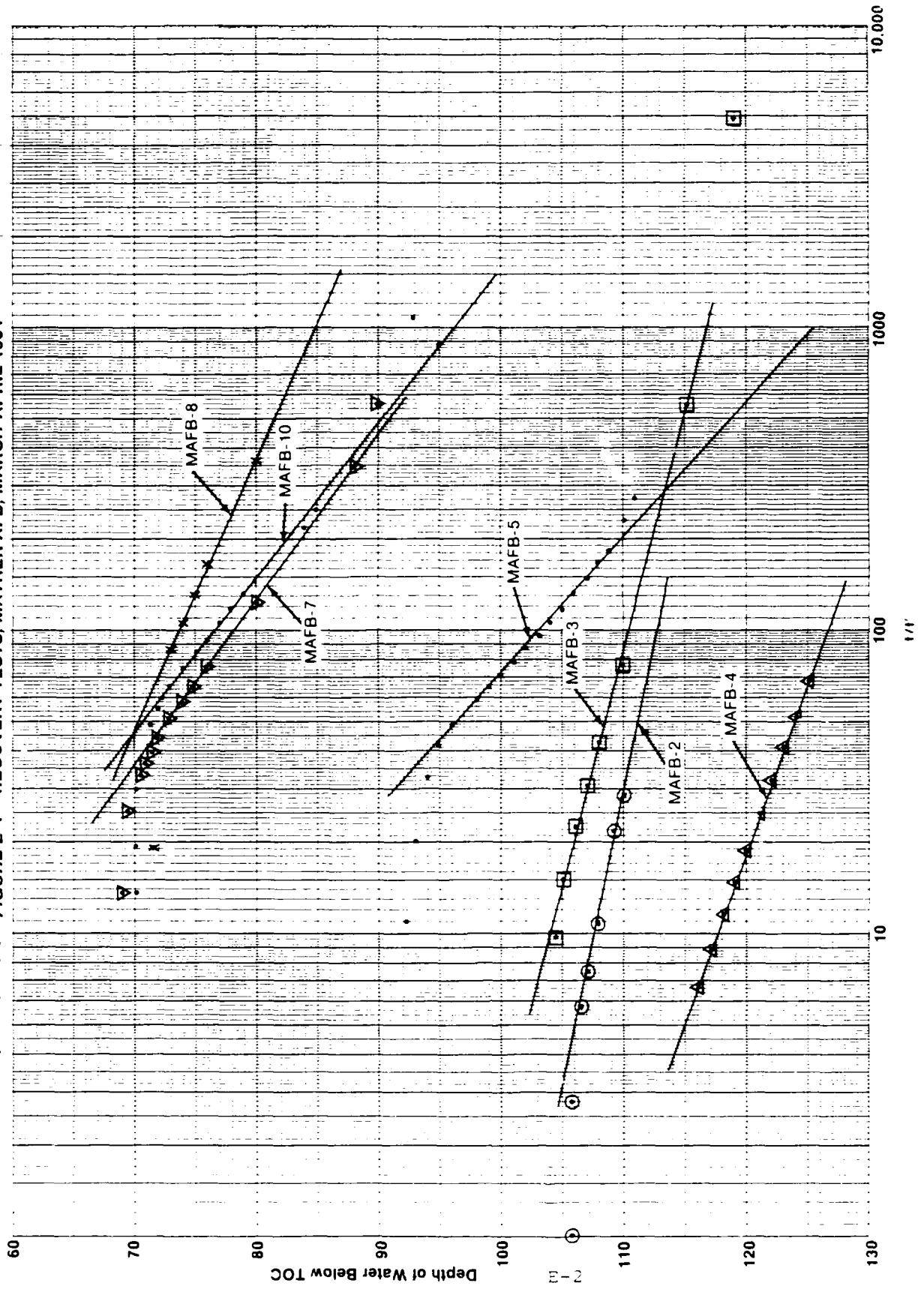
TABLE E-1

SUMMARY OF RECOVERY TEST RESULTS

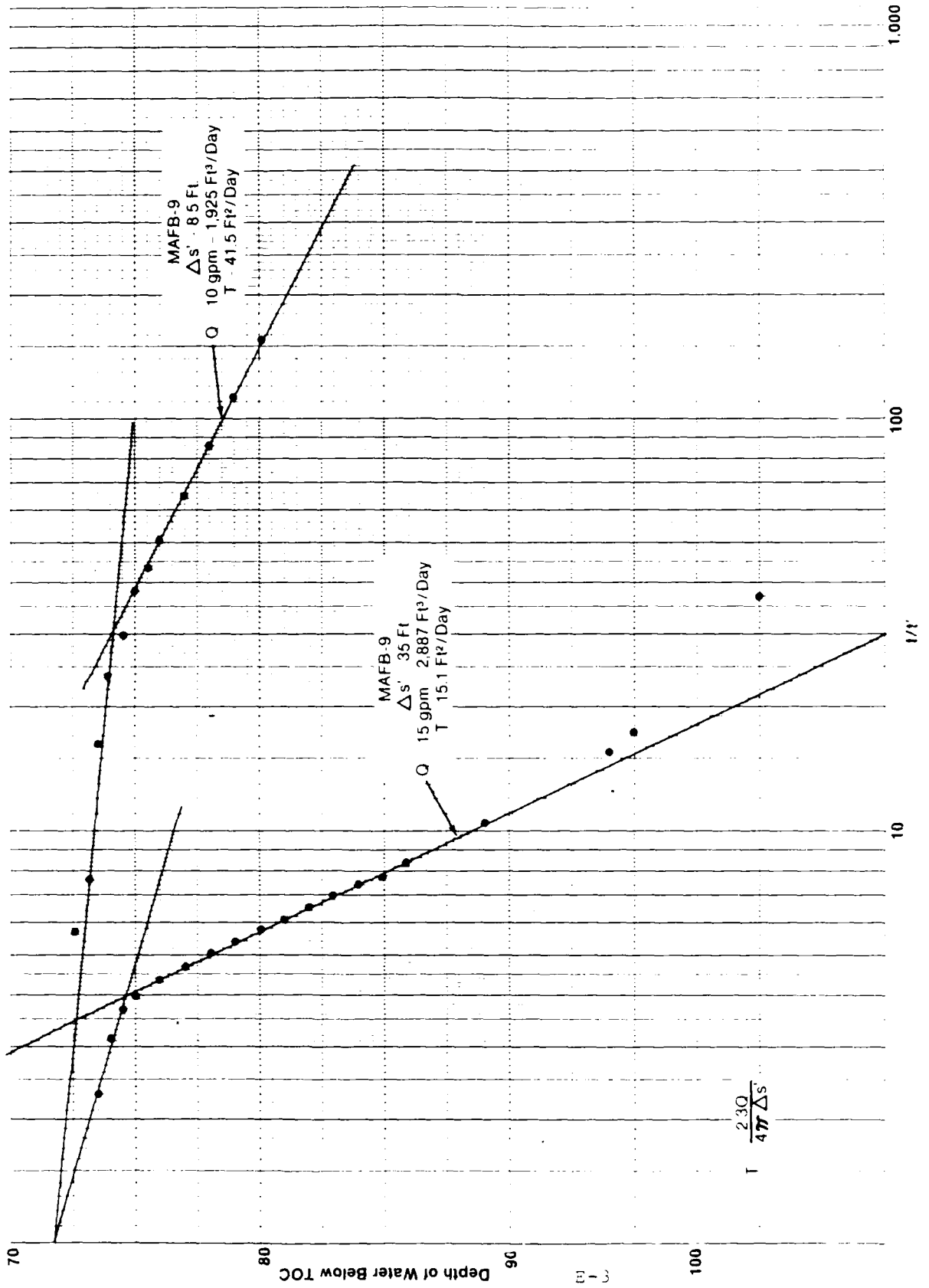
WELL NUMBER	RESIDUAL DRAWDOWN Over One Log Cycle, $\Delta s'$ (feet)	PUMPING RATE, Q_3 (gpm)	(ft ³ /day)	TRANSMISSIVITY T (1) (ft ² /day)
2	5.2	5	962	33.87
3	6.5	7.3	1,405	39.57
4	9.0	12	2,310	46.97
5	22.2	6	1,155	9.52
7	23.1	12	2,310	18.30
8	11.2	25	4,812	78.64
9	8.5	10	1,925	41.45
9	35	15	2,887	15.10
10	19.6	20	3,850	35.95

(1) Formula : $T = \frac{2.3 Q}{4 \pi \Delta s'}$

FIGURE E-1 RECOVERY TESTS, MATHER AFB, MARCH-APRIL 1984



RECOVERY TESTS, MATHER AFB, APRIL 1984



APPENDIX F

FIELD SAMPLING AND QA/QC PLAN

SAMPLING AND QA/QC PLAN

Base Well Pumping

The following procedures will be followed in sampling the 15 base production wells:

1. Ascertain whether the well is pumping or not. If it has not been pumped during the last 2 hours, treat it as a static well and go to Step 3).
2. If the well is pumping, measure the pumping water level and record that level, the pumping rate, and the cumulative reading (total gallons pumped) on the flow meter. Run the sampling tap just long enough to clear any standing water from the line and begin sampling.
3. If the well is static, measure the static water level (SWL). Compare this to the total well depth (see Table 1) to obtain the height of the column of standing water in the well. Convert this to a value in gallons. Multiply by 3 to obtain 3 well volumes.
4. Turn the pump on, and use the flow meter to obtain a flow rate. When the flow rate has stabilized, calculate the time necessary to pump three well volumes.
5. Run the pump for the total time necessary to pump three volumes, if possible. If there are physical constraints in the pumping or distribution system, make a note of the SWL, well volume, pumping rate, and total time pumped. At the end of this time, run the sampling tap long enough to clear standing water and begin sampling following Steps 6 through 9 below.

Monitor Well Pumping and Sampling

In general, ground-water samples will be collected from pump discharge after pumping 3 well volumes from each well. Prior to initiation of well pumping, however, a small volume of sample will be collected from the top of the water column for analysis of oils and greases. The parameters to be sampled for and appropriate containers are described in separate attachments.

All monitor wells will be pumped using an electric submersible pump connected to a Teflon[®] discharge line. The pump and discharge line will have been completely decontaminated (including purging with a detergent, nitric acid and distilled water) prior to first use at the site.

Between monitor wells, the equipment will be decontaminated by flushing the inside and hosing the outside with approximately 50 gallons of potable water from a Base source (use the same source throughout the sampling). At the end, approximately 5 to 10 gallons of dilute nitric acid solution (mixed in Base water) will be flushed through the pump and line, followed by 5 to 10 gallons of distilled or deionized water.

The following procedures will be followed in sampling each well:

1. Measure the SWL with reference to the measuring point marked on the top of the casing.
2. Lower a bottom-loading Teflon[®] bailer slowly down to the air-water interface, and draw off the top 6 inches, approximately, of the water column. Transfer this sample into the container for oil and grease analysis until it is three-quarters full.
3. Calculate the volume of standing water in the well.
4. Lower the pump and begin pumping. Record the pumping rate and total time to pump at least 3 well volumes.
5. At the end of this time, decrease the pumping rate, if necessary, and begin sampling.

6. Collect grab samples for immediate measurement of temperature, pH and specific conductance.
7. Gently fill each sample container from the pump line, taking care to avoid aeration or turbulence of the sample water. All containers should be filled completely (taking care not to spill preservatives if they are pre-dosed) except for the bottles for oil and grease analysis, which are to be only three-quarters full.
8. Filter 750 to 1000 ml of sample using a field filtering apparatus with as little exposure to air as possible. Transfer the filtered sample to an appropriate container and add sufficient nitric acid to lower pH of sample below 2 (approximately 2 ml).
9. Wrap the sample containers in protective packaging and pack with ice in a thermal chest to insure cooling to 4 °C.

Soil Sampling

Fully decontaminate the trowel or spatula used before collecting samples. Work from area of less suspected contamination (downstream from the oil skimmer to the area of greater suspected contamination (upstream).

1. Collect samples for VOA analysis in 30 ml vials, taking care to leave as little void space as possible and to completely crimp the Teflon-lined cap.
2. Collect another liter of sample in a separate amber glass jar with Teflon[®]-lined cap for the remaining analyses.

QC Samples

Approximately 20 percent additional samples will be collected for the purpose of validating field and analytical techniques. These will include 1 field blank and 2 duplicate ground-water samples.

The field blanks will consist of distilled water collected using methods and equipment the same or as close as possible to those used in actual sample collection: e.g. distilled water will be pumped from a clean glass jar through the pump

and line to obtain the field blank associated with ground-water sampling and distilled water will be poured into the closing sampler and from there into sample containers to obtain the field blank associated with surface-water sampling. Duplicates will be collected as separate samples, not splits of a single sample.

In addition, a trip bag of distilled water in two 30 ml glass vials with Teflon^R-lined septa will accompany each ice chest during sampling, and will be returned un-opened with the ice chest as it is shipped back to the laboratory.

Container Preparation

Another consideration in this, or any analytical project is that of sample container preparation. Accordingly, all appropriate sample bottles shall be cleaned in a manner mandated by the U.S. EPA to insure maximal cleanliness (and minimal contamination) before the containers go to the field. Sufficient bottles to accommodate both laboratory and field requirements will be necessary.

Chain-of-Custody

Since they document the history of samples, chain-of-custody procedures are a crucial part of a sampling/analysis program. Chain-of-custody documentation enables identification and tracking of a sample from collection to analysis to reporting.

WESTON's chain-of-custody program necessitates the use of EPA-approved sample labels, secure custody, and attendant record keeping. Depending on the client's requirements, WESTON also offers container sealing during unattended transportation of samples.

In essence, WESTON considers a sample in custody if it: is in a WESTON employee's physical possession; it is in view of that WESTON employee; is secured by that WESTON employee to prevent tampering; or is secured by that WESTON employee in an area that is restricted to authorized personnel.

Each time a sample is relinquished from one analyst to another, or from one location to another, WESTON's analytical personnel are required to make appropriate entries. Personnel-specific initials are used as identifiers of analysts, as are location codes for various locations (refrigerators, extraction areas, analytical areas, etc.).

Quality Assurance Plan

WESTON's analytical services enforces a rigid QA/QC program toward maintenance of validity and reliability of all analytical data. The Laboratory QA/QC Manual outlines the specifics of the QA/QC plan. This plan is patterned after the EPA Handbook for Analytical Quality Control in Water and Wastewater Laboratories (EPA-600/4-79-019, March 1979), augmented by general applicable experience and interaction with the QA/QC plan of the U.S. Army Toxic and Hazardous Materials Agency (USATHAMA). All methods and procedures followed by WESTON are either USEPA or ASTM-approved. Any variations from such procedures, regardless of cause, are documented by the responsible analyst(s) and are documentable and literature-traceable.



BY AND DATE 4-10-89 DIV _____ SHEET _____ OF _____
 CHKD BY _____ DATE _____ DEPT _____ W.O. NO. 0628-05-26
 PROJECT _____
 SUBJECT SUMMARY OF WELL SPECS - HATHER AFB BASE WELLS

WELL NAME	FACILITY #	TOTAL DEPTH (FEET BELOW G.S.)	PERFORATED INTERVAL	ORIGINAL YIELD (GPM)	CURRENT CAPACITY (GPM)
GC-1	8867	561	302-462	2400	-
GC-2	8880	403	-	-	-
FH-1	14418	531	262-517	1500	550
FH-2	14147	400	348-400	1100	635
FH-3	14988	500	280-500	1020	600
FH-4	14987	500	205-500	825	575
FH-5	17757	550	452-550	1100	955
FH-6	16000	499	246-500	3400	-
B-1	3476	532	262-532	1225	365
B-2	3795	584	186-404	1400	616
B-3	2795	501	294-500	1,000	600
B-4	2930	500	246-422	1,200	490
ACW	10151	250	200-220	140	?
K-9	18005	250	-	50	..
JTC	7098	~250	-	-	-

AD-A184 581

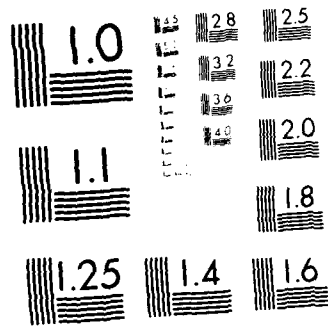
INSTALLATION RESTORATION PROGRAM PHASE 2
CONFIRMATION/QUANTIFICATION STAGE 1 VOLUME 2 APPENDICES
(U) WESTON (ROY F) INC WEST CHESTER PA JUN 86
F33615-80-D-4006 F/G 24/4

2/5

UNCLASSIFIED

NL





MICROCOPY RESOLUTION TEST CHART
NATIONAL BUREAU OF STANDARDS-1963-A

ATTACHMENT 3: Analytical Methods and Required Holding Times
(W. O. #0623-05-26)

<u>Parameter</u>	<u>EPA Method (Reference (1))</u>	<u>Sample Holding Time</u>
1. Water Samples		
VOA	601 and 602	14 days
TOC	415.1	28 days
Oils & Greases	413.2	28 days
Phenols	420.1	28 days
Cyanides	335.2	14 days
Lead	239.2	6 months
Chromium	ICP Optical Emission Spectrometric Method	6 months
Cadmium	213.2	6 months
Nickel	ICP Optical Emission Spectrometric Method	6 months
Silver	272.2	6 months
Dimethylnitrosamine (DMN)		7 days before extrac., 30 days thereafter
PCB	625	"
DDT isomers	608	"
Chlordane	608	"
2,4-D	600/4-81-053	"
2. Soil Samples		
VOA	5020	14 days
Oils & Greases	Modifield 3540 (Soxhlet Extraction)	28 days
Phenols	420.1	28 days
Cyanide	9010	14 days
Metals (Pb, Cr, Cd, Ni, Ag)	ICP Optical Emission Spectrometric Method	6 months

(1) It is assumed that the methods listed will meet the required detection limits given on the next page. Please notify me if any modification is made to meet the requirements.

Attachment 3
Required Sample Detection Limits

<u>Parameter</u>	<u>Soil/Sediment</u>	<u>Water</u>
*Total Organic Carbon (TOC)	1.0 milligram/gram	1.0 milligram/L
Oils and Greases (IR Method 412.3)	100 micrograms/gram	10 micrograms/L
Phenol	1 microgram/gram	1.0 microgram/L
Cyanide	1 microgram/gram	10 micrograms/L
Lead	2 micrograms/gram	20 micrograms/L
Chromium	5 micrograms/gram	50 micrograms/L
Calcium	1 microgram/gram	10 micrograms/L
Nickel	10 micrograms/gram	100 micrograms/L
Silver	1 microgram/gram	10 micrograms/L
PCB's	---	0.25 micrograms/L
Dimethylnitrosamine (DMN)	---	1.0 microgram/L
DDT isomers	---	0.02 micrograms/L
Chlordane	---	0.02 micrograms/L
2,4-D	---	0.06 microgram/L
Volatile Organic Compounds (VOC)	**	**

*Detection levels for TOC must be 3 times the noise level of the instrument. Laboratory distilled water must show no response; if it shows a response, corrections of positive results must be made.

**Detection limits for VOCs shall be as specified for those compounds listed EPA Methods 601 and 602.

Method: Federal Register, Vol 44, No. 233, pp 69458-69473.

This method should be strictly followed including these items:

Item 1.4 - This method is recommended by EPA for use only by experienced residue analysts or under the close supervision of such qualified persons.

Item 2.2 - This is most important. If interferences are encountered (as in early peaks such as vinyl chloride), the method provides a secondary gas chromatographic column that will be helpful in resolving the compounds of interest from interferences. This must be done in the case of vinyl chloride and so noted in analysis report.

Items 3.3, 7.1-7.3 - These sections on interferences, contamination and QC should be strictly followed.

Item 8.3 - All samples must be analyzed within the recommended holding time. This must be followed without exception.

If questions are encountered about certain contaminants you may be asked to show both chromatograms used to rule out possible interferences.

APPENDIX G

FIELD SAMPLE LOG SHEETS

ROUND 85-1



GROUNDWATER WELL LOG SHEET

Site Mather AFB Samplers Baylor/Holes

Field # _____

Sample I.D. FB-1 Date 5-31-85 Time 1600

TOTAL WELL DEPTH _____ FT.

S.W.L. _____ FT.

WATER COLUMN _____ FT.

WATER VOLUME _____ G.

X3= _____ G.

FLOW RATE _____ G.P.M.

PUMPING TIME REQUIRED _____ MINS.

ACTUAL PUMPING TIME _____ MINS.

BAILED _____

SAMPLES TAKEN:

VCA
TCC
Oil: Grease
PCB

_____ °C (Immediate)

16.9 Conductivity
(umhos)

6.2 pH

18.4 °C (at pH Cond.
Readings)

Duplicate Taken: Yes No

I.D. Assigned _____

Field # _____

SAMPLES TAKEN:

Field Blank samples
taken from the well
supplies samples
35000 with some reading
from conductivity to 1000.

_____ °C (Immediate)

_____ Conductivity
(umhos)

_____ pH

_____ °C (at pH Cond.
Readings)



GROUNDWATER WELL LOG SHEET

Site W-10-AB Samplers D.B./J

Field # _____

Sample I.D. 10-2 Date 5-30-55 Time 1:55

TOTAL WELL DEPTH _____ FT.

S.W.L. _____ FT.

WATER COLUMN _____ FT.

WATER VOLUME _____ G.

X3= _____ G.

FLOW RATE _____ G.P.M.

PUMPING TIME REQUIRED _____ MINS.

ACTUAL PUMPING TIME _____ MINS.

BAILED _____

SAMPLES TAKEN:

10-2
10-2
10-2
10-2
10-2

_____ °C (Immediate)

_____ Conductivity (umhos)

_____ pH

_____ °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned _____

Field # _____

SAMPLES TAKEN:

_____ °C (Immediate)

_____ Conductivity (umhos)

_____ pH

_____ °C (at pH Cond. Readings)

10-2
10-2
10-2
10-2



GROUNDWATER WELL LOG SHEET

Site Marine AFB Samplers DB/T

Field # _____

Sample I.D. FB-3 Date 5-30-85 Time 12:25

TOTAL WELL DEPTH _____ FT.

S.W.L. _____ FT.

WATER COLUMN _____ FT.

WATER VOLUME _____ G.

X3= _____ G.

FLOW RATE _____ G.P.M.

PUMPING TIME REQUIRED _____ MINS.

ACTUAL PUMPING TIME _____ MINS.

BAILED _____

SAMPLES TAKEN:

Metals 20, 30, 40, 50, 60

18 °C (Immediate)

11,700 Conductivity (umhos)

8.1 pH

15.3 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned _____

Field # _____

SAMPLES TAKEN:

Water at 10, 20, 30, 40, 50, 60
depths, 100, 150, 200, 250, 300
depths, 350, 400, 450, 500

1.
Building 7021

_____ °C (Immediate)

_____ Conductivity (umhos)

_____ pH

_____ °C (at pH Cond. Readings)

WESTERN

PRODUCTION WELL

Site Walter AFB

Samplers VB/DJ

Field # _____

Sample I.D. FH-1 ✓

Date 6-3-85

Time 0855

STATIC _____

TOTAL WELL DEPTH 531 FT.

S.W.L. _____ FT.

WATER COLUMN _____ FT.

WATER VOLUME _____ G.

X3= _____ G.

FLOW RATE _____ G.P.M.

PUMPING TIME REQUIRED _____ MINS.

ACTUAL PUMPING TIME _____ MINS.

SAMPLES TAKEN:

VOA (601 + 602)

TOC

CFE

22 °C (Immediate)

168 Conductivity (umhos)

7.3 pH

19.1 °C (at pH Cond. Readings)

1 1/2" screen

OPERATING X

WATER LEVEL 170 FT.

PUMPING RATE 1050 ~~540~~ G.P.M.

TOTAL GALLONS PUMPED 269362 G.

DUPLICATE TAKEN YES NO

I.D. ASSIGNED _____

FIELD # _____

21.2 °C

29 CONDUCTIVITY (umhos)

7.4 pH

SAMPLES TAKEN:

WESTERN

PRODUCTION WELL

Site Marker AFB Samplers DF/BJ
 Field # _____ 6-3-85
 Sample I.D. PH-2V Date ~~5-25-85~~ Time 1040

STATIC _____
 TOTAL WELL DEPTH 400 FT.
 S.W.L. _____ FT.
 WATER COLUMN _____ FT.
 WATER VOLUME _____ G.
 X3= _____ G.
 FLOW RATE _____ G.P.M.
 PUMPING TIME REQUIRED _____ MINS.
 ACTUAL PUMPING TIME _____ MINS.

12" screen
 OPERATING _____
 WATER LEVEL _____ FT.
 PUMPING RATE 800 G.P.M.
 TOTAL GALLONS PUMPED 144,569,000

SAMPLES TAKEN:

V3A (601+602)
 TCC
 C+G

71 °C (Immediate)
113 Conductivity (umhos)
7.3 pH
21.0 °C (at pH Cond. Readings)

DUPLICATE TAKEN YES NO
 I.D. ASSIGNED _____
 FIELD # _____
 _____ °C
 _____ CONDUCTIVITY (umhos)
 _____ pH

SAMPLES TAKEN:

WESTON

PRODUCTION WELL

Site Mather AFB Samplers DB/DT
 Field # _____
 Sample I.D. FH-3 Date 6-3-85 Time 1000
~~5-2-85~~

STATIC _____
 TOTAL WELL DEPTH 500 FT.
 S.W.L. _____ FT.
 WATER COLUMN _____ FT.
 WATER VOLUME _____ G.
 X3= _____ G.
 FLOW RATE _____ G.P.M.
 PUMPING TIME REQUIRED _____ MINS.
 ACTUAL PUMPING TIME _____ MINS.

SAMPLES TAKEN:
 VOA (G01 + G02)
 TOC
 C+G

*Samples were obtained
 directly from the well -
 no intermediate storage
 samples were obtained*

1.7" Screen
 OPERATING _____
 WATER LEVEL 170 FT.
 PUMPING RATE 1120 G.P.M.
 TOTAL GALLONS PUMPED 7793.6

_____ °C (Immediate)
 _____ 310 Conductivity (umhos)
 _____ 7.2 pH
 _____ 22.3 °C (at pH Cond. Readings)

DUPLICATE TAKEN YES NO
 I.D. ASSIGNED FH-30
 FIELD # _____
 _____ 21.7 °C
 _____ 212 CONDUCTIVITY (umhos)
 _____ 7.3 pH

SAMPLES TAKEN:

WESTON

PRODUCTION WELL

Site Mather AFB Samplers DR/ST
Field # _____
Sample I.D. FH-6 ✓ Date 6-3-85 Time 1320
~~5-30-85~~

STATIC _____
TOTAL WELL DEPTH 500 FT.
S.W.L. _____ FT.
WATER COLUMN _____ FT.
WATER VOLUME _____ G.
X3= _____ G.
FLOW RATE _____ G.P.M.
PUMPING TIME REQUIRED _____ MINS.
ACTUAL PUMPING TIME _____ MINS.

12" screen
OPERATING _____
WATER LEVEL _____ FT.
PUMPING RATE 1600 G.P.M.
TOTAL GALLONS PUMPED 145849 G.

SAMPLES TAKEN:

VEA (601-602)
TOC
DTG

21.5 °C (Immediate)
129 Conductivity (umhos)
7.45 pH
21.2 °C (at pH Cond. Readings)

DUPLICATE TAKEN YES NO
I.D. ASSIGNED _____
FIELD # _____
_____ °C
_____ CONDUCTIVITY (umhos)
_____ pH

SAMPLES TAKEN:

Probe very oily when
trying to take WC.



PRODUCTION WELL

Site Mather AFB

Samplers DB/DJ

Field # _____

Sample I.D. B-1

Date 6-3-85 Time 1505

STATIC _____

TOTAL WELL DEPTH 532 FT.

S.W.L. _____ FT.

WATER COLUMN _____ FT.

WATER VOLUME _____ G.

X3= _____ G.

FLOW RATE _____ G.P.M.

PUMPING TIME REQUIRED _____ MINS.

ACTUAL PUMPING TIME _____ MINS.

SAMPLES TAKEN:

VIA

TCC

OTC

Metals

DMN

Pest/Herbic.

12"
OPERATING _____

WATER LEVEL 120 FT.

PUMPING RATE 8 G.P.M.

TOTAL GALLONS PUMPED 11340 G.

51 °C (Immediate)

_____ Conductivity (umhos)

_____ pH

_____ °C (at pH Cond. Readings)

DUPLICATE TAKEN YES NO

I.D. ASSIGNED _____

FIELD # _____

_____ °C

_____ CONDUCTIVITY (umhos)

_____ pH

SAMPLES TAKEN:

WESTERN

PRODUCTION WELL

Site Maxwell AFB

Samplers DB/DJT

Field # _____

Sample I.D. B-2

Date 6-3
~~5-30~~ 85 Time _____

STATIC _____

TOTAL WELL DEPTH 584 FT.

S.W.L. _____ FT.

WATER COLUMN _____ FT.

WATER VOLUME _____ G.
X3= _____ G.

FLOW RATE _____ G.P.M.

PUMPING TIME REQUIRED _____ MINS.

ACTUAL PUMPING TIME _____ MINS.

SAMPLES TAKEN:

VDA
 TCC
 C+G
 Metals
 DMN
 Pect/Ker'ic

12" screen
 OPERATING _____

WATER LEVEL 175' FT.

PUMPING RATE 1140 G.P.M.

TOTAL GALLONS PUMPED 304,150,000

21.5 °C (Immediate)

_____ Conductivity (umhos)

_____ pH

_____ °C (at pH Cond. Readings)

DUPLICATE TAKEN YES NO

I.D. ASSIGNED _____

FIELD # _____

_____ °C

_____ CONDUCTIVITY (umhos)

_____ pH

SAMPLES TAKEN:

WESTON

PRODUCTION WELL

Site MANAGER AFB Samplers LE/DJ
Field # _____
Sample I.D. B-3 Date 6-3-85 Time _____

STATIC _____
TOTAL WELL DEPTH 501 FT.
S.W.L. 90 FT.
WATER COLUMN _____ FT.
WATER VOLUME _____ G.
X3= _____ G.
FLOW RATE _____ G.P.M.
PUMPING TIME REQUIRED _____ MINS.
ACTUAL PUMPING TIME _____ MINS.

12" SAMPLE
OPERATING _____ °C (Immediate)
WATER LEVEL 110 FT. _____ Conductivity (umhos)
PUMPING RATE 590 G.P.M. _____ pH
TOTAL GALLONS PUMPED 696 058 070 G. _____ °C (at pH Cond. Readings)

DUPLICATE TAKEN YES NO
I.D. ASSIGNED _____
FIELD # _____
_____ °C
_____ CONDUCTIVITY (umhos)
_____ pH

SAMPLES TAKEN:
VOA
TCC
C+G
Metals
DMN
Pest/Herb

SAMPLES TAKEN:

WESTON

PRODUCTION WELL

Site Mather AFB Samplers DPB DJ
 Field # _____
 Sample I.D. B-4 Date 6-3-85 Time _____

STATIC _____
 TOTAL WELL DEPTH 500 FT.
 S.W.L. 90 FT.
 WATER COLUMN _____ FT.
 WATER VOLUME _____ G.
 X3= _____ G.
 FLOW RATE _____ G.P.M.
 PUMPING TIME REQUIRED _____ MINS.
 ACTUAL PUMPING TIME _____ MINS.

SAMPLES TAKEN:
VDA (501+601) (3 ea.)
TRC
OTG
Metals
DMN
Pest/Herb (3)

1/2" screen
 OPERATING _____
 WATER LEVEL _____ FT.
 PUMPING RATE _____ G.P.M.
 TOTAL GALLONS PUMPED _____ G.

_____ °C (Immediate)
 _____ Conductivity (umhos)
 _____ pH
 _____ °C (at pH Cond. Readings)

DUPLICATE TAKEN YES NO
 I.D. ASSIGNED B-4
 FIELD # _____
 _____ °C
 _____ CONDUCTIVITY (umhos)
 _____ pH

SAMPLES TAKEN:
VDA
TRC
OTG
Metals
DMN
Pest/Herb

WESTON

PRODUCTION WELL

Site North of AFB Samplers T.B.H.
Field # _____
Sample I.D. JTC Date 6-3-85 Time 1505

STATIC _____
TOTAL WELL DEPTH 200 FT.
S.W.L. _____ FT.
WATER COLUMN _____ FT.
WATER VOLUME _____ G.
X3= _____ G.
FLOW RATE _____ G.P.M.
PUMPING TIME REQUIRED _____ MINS.
ACTUAL PUMPING TIME _____ MINS.

OPERATING _____ °C (Immediate)
WATER LEVEL _____ FT.
PUMPING RATE _____ G.P.M.
TOTAL GALLONS PUMPED _____ G.

SAMPLES TAKEN:

VOA (601-602)
TOC
O+G
Phenolics
Cyanide
Metals

_____ °C (Immediate)
_____ Conductivity (umhos)
_____ pH
_____ °C (at pH Cond. Readings)

DUPLICATE TAKEN YES NO
I.D. ASSIGNED JTC-1
FIELD # _____
_____ °C
_____ CONDUCTIVITY (umhos)
_____ pH

SAMPLES TAKEN:

Phenolics
Cyanide

WESTON

PRODUCTION WELL

Site Matmor AFB Samplers JB/DJ
Field # _____
Sample I.D. K-9 ✓ Date 6-3-85 Time 1240

STATIC _____
TOTAL WELL DEPTH 300 FT.
S.W.L. _____ FT.
WATER COLUMN _____ FT.
WATER VOLUME _____ G.
X3= _____ G.
FLOW RATE _____ G.P.M.
PUMPING TIME REQUIRED _____ MINS.
ACTUAL PUMPING TIME _____ MINS.

SAMPLES TAKEN:
VCA
TOC
Oil + Grease

_____ °C (Immediate) 22
_____ Conductivity (umhos)
_____ pH 1.54
_____ °C (at pH Cond. Readings)

OPERATING ✓
WATER LEVEL 110 FT.
PUMPING RATE 30 G.P.M.
TOTAL GALLONS PUMPED 2249, 060

DUPLICATE TAKEN YES NO
I.D. ASSIGNED _____
FIELD # _____
_____ °C
_____ CONDUCTIVITY (umhos)
_____ pH

SAMPLES TAKEN:

WESTON

PRODUCTION WELL

Site MARSH AFB Samplers LR/ITJ
 Field # _____
 Sample I.D. GC-1 Date 6-3-85 Time _____
~~5-13-85~~

STATIC _____ FT.
 TOTAL WELL DEPTH 562 FT.
 S.W.L. _____ FT.
 WATER COLUMN _____ FT.
 WATER VOLUME _____ G.
 X3= _____ G.
 FLOW RATE _____ G.P.M.
 PUMPING TIME REQUIRED _____ MINS.
 ACTUAL PUMPING TIME _____ MINS.

SAMPLES TAKEN:
VQA (GC1 & GC2)
TRC
C+L

_____ °C (Immediate)
45 Conductivity (umhos)
7.4 pH
20.0 °C (at pH Cond. Readings)

15" screen
 OPERATING _____
 WATER LEVEL _____ FT.
 PUMPING RATE 3 G.P.M.
 TOTAL GALLONS PUMPED _____ G.

DUPLICATE TAKEN YES NO
 I.D. ASSIGNED _____
 FIELD # _____
 _____ °C
 _____ CONDUCTIVITY (umhos)
 _____ pH

SAMPLES TAKEN:

WESTON

PRODUCTION WELL

Site Mudflat #2 Samplers DB-1
 Field # _____
 Sample I.D. 60-2 ✓ Date 6-3-85 Time 12:15

STATIC _____
 TOTAL WELL DEPTH 403 FT.
 S.W.L. _____ FT.
 WATER COLUMN _____ FT.
 WATER VOLUME _____ G.
 X3= _____ G.
 FLOW RATE _____ G.P.M.
 PUMPING TIME REQUIRED _____ MINS.
 ACTUAL PUMPING TIME _____ MINS.

SAMPLES TAKEN:
 VCA (60-1, 60-2)
 P.R.
 C.I. (60-2)

_____ °C (Immediate)
 _____ Conductivity (umhos)
 _____ pH
 _____ °C (at pH Cond. Readings)

12' SECTION
 OPERATING X
 WATER LEVEL 80 FT.
 PUMPING RATE 5.0 G.P.M.
 TOTAL GALLONS PUMPED _____ G.

DUPLICATE TAKEN YES NO
 I.D. ASSIGNED 60-20
 FIELD # _____
20.3 °C
41 CONDUCTIVITY (umhos)
7.78 pH

SAMPLES TAKEN:
 VCA (60-1, 60-2)
 P.R.
 C.I. (60-2)



GROUNDWATER WELL LOG SHEET

Site MAFB: AFB

Samplers DB/ST

Field # _____

Sample I.D. MAFB-3

Date 5-22-85

Time 1703

TOTAL WELL DEPTH ~~120~~ 232.5 FT.

S.W.L. 110.08 FT.

WATER COLUMN 13.42 FT.

WATER VOLUME 8.8 G.

X3= 26.4 G.

FLOW RATE _____ G.P.M.

PUMPING TIME REQUIRED _____ MINS.

ACTUAL PUMPING TIME _____ MINS.

BAILED _____

Notes 24-100

SAMPLES TAKEN:

VCA
TCO
Oil - increase
PCB

_____ °C (Immediate)

139 Conductivity (umhos)

7.5 pH

20.8 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned _____

Field # _____

_____ °C (Immediate)

_____ Conductivity (umhos)

_____ pH

_____ °C (at pH Cond. Readings)

SAMPLES TAKEN:



GROUNDWATER WELL LOG SHEET

Site Mattings AFB Samplers DB/LT

Field # _____

Sample I.D. MAFB-2 Date 5-31-85 Time _____

TOTAL WELL DEPTH 123.7 ~~120~~ FT.

S.W.L. 109.7 FT.

WATER COLUMN 14.0 FT.

WATER VOLUME 7.5 G.

X3= 22.5 G.

FLOW RATE _____ G.P.M.

PUMPING TIME REQUIRED _____ MINS.

ACTUAL PUMPING TIME _____ MINS.

BAILED _____

correct 102-120

SAMPLES TAKEN:

VBA
TCC
Oil and Grease
PCB

_____ °C (Immediate)

134 Conductivity (umhos)

7.3 pH

20.1 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned MAFB-10

Field # _____

SAMPLES TAKEN:

VBA
TCC
Oil and Grease
PCB

_____ °C (Immediate)

12.8 Conductivity (umhos)

6.7 7.3 pH

17.0 °C (at pH Cond. Readings)



GROUNDWATER WELL LOG SHEET

Site Matier AFB Samplers DB/DJ

Field # _____

Sample I.D. MAFB-3 Date 5-31-85 Time 1:35

TOTAL WELL DEPTH 120 FT.
 S.W.L. 107.18 FT.
 WATER COLUMN 12.82 FT.
 WATER VOLUME 84 G.
 X3= 25.2 G.
 FLOW RATE 30 G.P.M.
 PUMPING TIME REQUIRED 25 MINS.
 ACTUAL PUMPING TIME 15 MINS.
 BAILED _____
SEASON 102-120

SAMPLES TAKEN:

VBA
T/C
Oil and Grease
P/B

_____ °C (Immediate)
120 Conductivity (umhos)
7.6 pH
20.5 °C (at pH Cond. Readings)

Duplicate Taken: Yes No
 I.D. Assigned _____
 Field # _____

SAMPLES TAKEN:

_____ °C (Immediate)
 _____ Conductivity (umhos)
 _____ pH
 _____ °C (at pH Cond. Readings)



GROUNDWATER WELL LOG SHEET

Site Mather AFB Samplers Bogner/PLNCS

Field # _____

Sample I.D. MAFB-4 Date 5-31-85 Time 0955

TOTAL WELL DEPTH 153.5 FT.
 S.W.L. 109.36 FT.
 WATER COLUMN 44.14 FT.
 WATER VOLUME 29 G.
 X3= 87 G.
 FLOW RATE 3 G.P.M.
 PUMPING TIME REQUIRED 29 MINS.
 ACTUAL PUMPING TIME 31 MINS.
 BAILED _____
SCREEN 135.5-153.5

SAMPLES TAKEN:
 VCA
 TOC
 Oil - Grease
 Metals
 DMN
 Pest/Herb

18 °C (Immediate)
205 Conductivity (umhos)
6.63 pH
18.3 °C (at pH Cond. Readings)

Duplicate Taken: Yes No
 I.D. Assigned _____
 Field # _____
 _____ °C (Immediate)
 _____ Conductivity (umhos)
 _____ pH
 _____ °C (at pH Cond. Readings)

SAMPLES TAKEN:



GROUNDWATER WELL LOG SHEET

Site Mother AFB

Samplers CB/JT

Field # _____

Sample I.D. MAFB-45

Date 5-31-85

Time 0955

TOTAL WELL DEPTH 128.5 ~~123.25~~ FT.

S.W.L. 89.18 FT.

WATER COLUMN 39.32 FT.

WATER VOLUME 25.7 G.

X3= 77.1 G.

FLOW RATE 3.3 G.P.M.

PUMPING TIME REQUIRED 25 MINS.

ACTUAL PUMPING TIME _____ MINS.

BAILED _____

Screen 110.5 - 128.5 FT

SAMPLES TAKEN:

VCA

TRC

Oil : Grease

Metals

DMN

Pesticide/Herbicide

_____ °C (Immediate)

_____ Conductivity (umhos)

7.1 ~~7.6~~ pH

_____ °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned MAFB-50

Field # _____

_____ °C (Immediate)

_____ Conductivity (umhos)

7.1 ~~7.6~~ pH

_____ °C (at pH Cond. Readings)

SAMPLES TAKEN:

VCA

TRC

Oil : Grease

Metals

DMN

Pesticide/Herb



GROUNDWATER WELL LOG SHEET

Site Walter AFB

Samplers DB DT

Field # _____

Sample I.D. MAFB-6

Date 5-30-85

Time 1000

TOTAL WELL DEPTH ^{129.5} ~~128.5~~ FT.

S.W.L. 71.27 FT.

WATER COLUMN ²⁸ ~~28~~ 23 FT.

WATER VOLUME ~~18.5~~ 18.5 G.

X3= ~~18.5~~ 55.5 G.

FLOW RATE 3.5 G.P.M.

PUMPING TIME REQUIRED 16 MINS.

ACTUAL PUMPING TIME 20 MINS.

BAILED _____

SAMPLES TAKEN:

VOA
 TOC
 O+G
 Metals
 DMN
 Pest/Herb

20.5 °C (Immediate)

253 Conductivity (umhos)

7.17 pH

21.1 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned _____

Field # _____

_____ °C (Immediate)

_____ Conductivity (umhos)

_____ pH

_____ °C (at pH Cond. Readings)

SAMPLES TAKEN:

_____ °C (Immediate)

_____ Conductivity (umhos)

_____ pH

_____ °C (at pH Cond. Readings)



GROUNDWATER WELL LOG SHEET

Site MAFEB AFB Samplers SAF

Field # -

Sample I.D. MAFEB-7 Date 9-30-85 Time 0930

TOTAL WELL DEPTH 110 FT.
 S.W.L. 72.55 FT.
 WATER COLUMN 37.45 FT.
 WATER VOLUME 375 G.
 X3= 73.5 G.
 FLOW RATE 0.0 G.P.M.
 PUMPING TIME REQUIRED 50 MINS.
 ACTUAL PUMPING TIME 05 MINS.
 BAILED _____
Bottom 92-110

SAMPLES TAKEN:
10A
10B
01E
02A
02B
02C
02D
02E
02F
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20Z

Duplicate Taken: Yes No
 I.D. Assigned NAED-70
 Field # _____
 _____ °C (Immediate)
 _____ Conductivity (umhos)
 _____ pH
 _____ °C (at pH Cond. Readings)

SAMPLES TAKEN:
20A
20B
20C
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GROUNDWATER WELL LOG SHEET

Site NAFB AFB Samplers DEPT

Field # _____

Sample I.D. NAFB-8 Date 5-29-85 Time 1455

TOTAL WELL DEPTH 157 FT.
S.W.L. 77.49 FT.
WATER COLUMN 29.51 FT.
WATER VOLUME 173 G.
X3= 579 G.
FLOW RATE 4 G.P.M.

PUMPING TIME REQUIRED 4.5 MINS.
ACTUAL PUMPING TIME 15 MINS.

BAILED _____

Sum 87-107

SAMPLES TAKEN:

1.0A
1.0B
1.0C
1.0D
1.0E
1.0F

14 °C (Immediate)
786 Conductivity (umhos)
6.6 pH
18.2 °C (at pH Cond. Readings)

Duplicate Taken: Yes No
I.D. Assigned NAFB 80
Field # _____

19 °C (Immediate)
987 Conductivity (umhos)
6.56 pH
18.2 °C (at pH Cond. Readings)

SAMPLES TAKEN:

1.0A
1.0B
1.0C
1.0D
1.0E
1.0F



GROUNDWATER WELL LOG SHEET

Site MAFB FEB

Samplers DAIS

Field # _____

Sample I.D. MAFB-91

Date 5-29-85

Time 30

TOTAL WELL DEPTH 110 FT.

S.W.L. 65.65 FT.

WATER COLUMN 44.35 FT.

WATER VOLUME 29 G.

X3= 27 G.

FLOW RATE 4.1 G.P.M.

PUMPING TIME REQUIRED 27 MINS.

ACTUAL PUMPING TIME _____ MINS.

BAILED _____

SINCE 92-110

SAMPLES TAKEN:

0.1-0.2

0.2-0.3

0.3-0.4

0.4-0.5

0.5-0.6

0.6-0.7

0.7-0.8

0.8-0.9

0.9-1.0

_____ °C (Immediate)

1012 Conductivity (umhos)

6.5 pH

24.1 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned MAFB-90

Field # _____

_____ °C (Immediate)

_____ Conductivity (umhos)

_____ pH

_____ °C (at pH Cond. Readings)

SAMPLES TAKEN:

1.0-1.1

1.1-1.2

1.2-1.3

1.3-1.4

1.4-1.5

1.5-1.6

1.6-1.7

1.7-1.8

1.8-1.9

1.9-2.0



GROUNDWATER WELL LOG SHEET

Site MARLBOROUGH Samplers 102107

Field # _____

Sample I.D. MAFB-10 Date 5-20-85 Time 1320

TOTAL WELL DEPTH 105 FT.

S.W.L. 74.71 FT.

WATER COLUMN 30.29 FT.

WATER VOLUME 20 G.

X3= 60 G.

FLOW RATE 3.3 G.P.M.

PUMPING TIME REQUIRED 18 MINS.

ACTUAL PUMPING TIME 20 MINS.

BAILED _____

Screen 84-105

SAMPLES TAKEN:

1/4
1/2
3/4
1.0
1.25
1.5

20 °C (Immediate)

207 Conductivity (umhos)

7.4 pH

18.4 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned MAFB-100

Field # _____

_____ °C (Immediate)

_____ Conductivity (umhos)

_____ pH

_____ °C (at pH Cond. Readings)

SAMPLES TAKEN:

1.0
1.25



GROUNDWATER WELL LOG SHEET

Site MAHAR AFB Samplers T.B./L.T.

Field # _____

Sample I.D. MAFB-11 Date 5-30-85 Time 1145

TOTAL WELL DEPTH 105 FT.

S.W.L. 73.37 FT.

WATER COLUMN 31.63 FT.

WATER VOLUME 20.7 G.

X3= 521 G.

FLOW RATE 3.5 G.P.M.

PUMPING TIME REQUIRED 18 MINS.

ACTUAL PUMPING TIME 20 MINS.

BAILED _____

MAFB-11

SAMPLES TAKEN:

VCA
TCC
S+G
Manganese
Organic
Metals

19.5 °C (Immediate)

58 Conductivity (umhos)

7.2 pH

21.3 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned MAFB-11D

Field # _____

_____ °C (Immediate)

57 Conductivity (umhos)

7.5 pH

_____ °C (at pH Cond. Readings)

SAMPLES TAKEN:

VCA
TCC
S+G
Manganese
Organic
Metals

ROUND 85-2



GROUNDWATER WELL LOG SHEET

Site MATHER AFB, CA Samplers DOE/INT

Field # _____

Sample I.D. FB-1 Date 6-27-55 Time 1515

TOTAL WELL DEPTH _____ FT.

S.W.L. _____ FT.

WATER COLUMN _____ FT.

WATER VOLUME _____ G.

X3= _____ G.

FLOW RATE _____ G.P.M.

PUMPING TIME REQUIRED _____ MINS.

ACTUAL PUMPING TIME _____ MINS.

BAILED _____

SAMPLES TAKEN:

_____ °C (Immediate)

103 Conductivity (umhos)

8.31 pH

11.2 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned _____

Field # _____

_____ °C (Immediate)

_____ Conductivity (umhos)

_____ pH

_____ °C (at pH Cond. Readings)

SAMPLES TAKEN:



PRODUCTION WELL

Site Alameda AFB

Samplers D. PERES

Field #

Sample I.D. FH-E

Date 6-27-85 Time 0900

STATIC

TOTAL WELL DEPTH FT.

S.W.L. FT.

WATER COLUMN FT.

WATER VOLUME G.
 X3= G.

FLOW RATE G.P.M.

PUMPING TIME REQUIRED MINS.

ACTUAL PUMPING TIME MINS.

SAMPLES TAKEN:
VOA
TOP
~~13M~~ OUT-INCREASE

OPERATING

WATER LEVEL 95 FT.

PUMPING RATE G.P.M.

TOTAL GALLONS PUMPED G.

70 °C (Immediate)

Conductivity (umhos)

7.45 pH

°C (at pH Cond. Readings)

DUPLICATE TAKEN YES NO

I.D. ASSIGNED

FIELD #

 °C

 CONDUCTIVITY (umhos)

 pH

SAMPLES TAKEN:



PRODUCTION WELL

Site MATHER DEP.

Samplers 225

Field # _____

Sample I.D. ACW

Date 5-22-50 Time 1:00

STATIC _____

TOTAL WELL DEPTH _____ FT.

S.W.C. _____ FT.

WATER COLUMN _____ FT.

WATER VOLUME _____ G.

 K# _____ G.

FLOW RATE _____ G.P.M.

PUMPING TIME RECEIVED _____ MINS.

ACTUAL PUMPING TIME _____ MINS.

SAMPLES TAKEN:

VCAD
100
100
100

OPERATING _____

WATER LEVEL 30 FT. _____

PUMPING RATE _____ G.P.M. _____

TOTAL GALLONS PUMPED _____ G. _____

DUPLICATE TAKEN YES NO

I.D. ASSIGNED ACW

FIELD # _____

11.3 °C

231 CONDUCTIVITY (umhos)

7.4 pH

SAMPLES TAKEN:

100
100
100

WESTERN

GROUNDWATER WELL LOG SHEET

Site Mather AFB CA Samplers 2B/ST

Field #

Sample I.D. MAFB-1 Date 4-27-85 Time 11:5

TOTAL WELL DEPTH 123.5 FT.
S.W.L. FT.
WATER COLUMN FT.
WATER VOLUME G.
X3= G.
FLOW RATE G.P.M.
PUMPING TIME REQUIRED MINS.
ACTUAL PUMPING TIME MINS.
BAILED

SAMPLES TAKEN:

 °C (Immediate)
 Conductivity
 (umhos)
7.33 ~~7.18~~ pH
 °C (at pH Cond.
Readings)

Duplicate Taken: Yes No
I.D. Assigned
Field #

SAMPLES TAKEN:

 °C (Immediate)
 Conductivity
 (umhos)
 pH
 °C (at pH Cond.
Readings)

WESTERN

GROUNDWATER WELL LOG SHEET

Site North AFB CA Samplers DDT

Field # _____

Sample I.D. MAFB-2 Date 6-7-85 Time 10:20

TOTAL WELL DEPTH 123.7 FT.
S.W.L. 110.6 FT.
WATER COLUMN 13.1 FT.
WATER VOLUME 8.6 G.
X3= 276 G.
FLOW RATE 33 $\frac{1}{2}$ G.P.M.
PUMPING TIME REQUIRED ? MINS.
ACTUAL PUMPING TIME 10 MINS.
BAILED _____

SAMPLES TAKEN:

VOA
VOC
PBB
PBB

22.0 °C (Immediate)
510 Conductivity (umhos)
~~6.75~~ 6.75 pH
12.1 °C (at pH Cond. Readings)

Duplicate Taken: Yes No
I.D. Assigned _____
Field # _____

_____ °C (Immediate)
_____ Conductivity (umhos)
_____ pH
_____ °C (at pH Cond. Readings)

SAMPLES TAKEN:

Very large quantity of samples
of samples



GROUNDWATER WELL LOG SHEET

Site MAFAC AFB, CA Samplers DB/CT

Field #

Sample I.D. MAFB-3 Date 6-27-85 Time 0920

TOTAL WELL DEPTH 123.6 FT.

S.W.L. 105.44 FT.

WATER COLUMN 15.76 FT.

WATER VOLUME 0.3 G.

X3= 24.24 G.

FLOW RATE 2.0 G.P.M.

PUMPING TIME REQUIRED 9 MINS.

ACTUAL PUMPING TIME 12 MINS.

BAILED

SAMPLES TAKEN:

22.5 °C (Immediate)

317 Conductivity (umhos)

7.55 pH

9.8 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned MAFB-30

Field #

 °C (Immediate)

233 Conductivity (umhos)

7.62 pH

 °C (at pH Cond. Readings)

SAMPLES TAKEN:

WESTERN

GROUNDWATER WELL LOG SHEET

Site MAFNB AFB Samplers DB/D

Field #

Sample I.D. MAFNB-4 Date 6-27-85 Time 13:00

TOTAL WELL DEPTH 156.4 FT.
S.W.L. 108.47 FT.
WATER COLUMN 47.93 FT.
WATER VOLUME 31 G.
X3= 93 G.
FLOW RATE 3.3 G.P.M.
PUMPING TIME REQUIRED 25 MINS.
ACTUAL PUMPING TIME MINS.
BAILED

SAMPLES TAKEN:

VOA
TOC
DTC
Metals
DMA
Pesticides

22 °C (Immediate)
163 Conductivity (umhos)
6.91 pH
27.1 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned

Field #

 °C (Immediate)
 Conductivity (umhos)
 pH
 °C (at pH Cond. Readings)

SAMPLES TAKEN:



GROUNDWATER WELL LOG SHEET

Site Maxwell AFB Samplers ZB/DJ

Field #

Sample I.D. MAFB-5 Date 6-27-85 Time 1430

TOTAL WELL DEPTH 127.9 FT.

S.W.L. 90.4 FT.

WATER COLUMN 37.5 FT.

WATER VOLUME 24.5 G.

X3= 13.5 G.

FLOW RATE 3.3 G.P.M.

PUMPING TIME REQUIRED 30 MINS.

ACTUAL PUMPING TIME MINS.

BAILED

SAMPLES TAKEN:

VOA

TSC

O+G

Metals

CMAS

pest/metals.

21 °C (Immediate)

132 Conductivity (umhos)

7.1 pH

26.6 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned

Field #

 °C (Immediate)

 Conductivity (umhos)

 pH

 °C (at pH Cond. Readings)

SAMPLES TAKEN:

WESTERN

GROUNDWATER WELL LOG SHEET

Site MAFBC AFB Samplers LB/ST

Field # -

Sample I.D. MAFB-6 Date 6-07-85 Time 1610

TOTAL WELL DEPTH 102.5 FT.

S.W.L. 72.11 FT.

WATER COLUMN 30.39 FT.

WATER VOLUME 20 G.

X3= 60 G.

FLOW RATE 3.3 G.P.M.

PUMPING TIME REQUIRED 18 MINS.

ACTUAL PUMPING TIME _____ MINS.

BAILED _____

SAMPLES TAKEN:

VOA
TCC
O&E
Metals
OMN
Pesticides

21 °C (Immediate)

242 Conductivity (umhos)

6.9 pH

25.3 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned MAFB-60

Field # _____

_____ °C (Immediate)

238 Conductivity (umhos)

6.9 pH

24.4 °C (at pH Cond. Readings)

SAMPLES TAKEN:



GROUNDWATER WELL LOG SHEET

Site Mather AFB, CA

Samplers DB/DT

Field #

Sample I.D. MAFB-7

Date 6-26-85

Time 1350

TOTAL WELL DEPTH 117.8 FT.

S.W.L. 74.83 FT.

WATER COLUMN 37.97 FT.

WATER VOLUME 24.8 G.

X3= 74.4 G.

FLOW RATE 3.3 G.P.M.

PUMPING TIME REQUIRED 22.5 MINS.

ACTUAL PUMPING TIME 22 MINS.

BAILED

SAMPLES TAKEN:

10A
 DC-
 OTC-
 Metals
 Phenol
 Cyanide

21.5 °C (Immediate)

1005 Conductivity (umhos)

10.7 pH

29.4 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned MAFB-70

Field #

_____ °C (Immediate)

_____ Conductivity (umhos)

_____ pH

_____ °C (at pH Cond. Readings)

SAMPLES TAKEN:

orange brown

Very dirty discharge at first
 Ran clear after 23-4 min.

Phenols
 Cyanide

WESTERN

GROUNDWATER WELL LOG SHEET

Site Mather AFB, CA Samplers D. Jones / D. Boyner
Field # _____
Sample I.D. MAFB-8 Date 6-26-85 Time 1220

TOTAL WELL DEPTH 109.8 FT.
S.W.L. 79.3 FT.
WATER COLUMN 30.49 FT.
WATER VOLUME 20 G.
X3= 60 G.
FLOW RATE 3.3 G.P.M.
PUMPING TIME REQUIRED _____ MINS.
ACTUAL PUMPING TIME _____ MINS.
BAILED _____

SAMPLES TAKEN:

VOAs
TOC
Oil & Grease
Metals
Phenolics
Cyanide

20 °C (Immediate)
1293 Conductivity (umhos)
6.89 pH
26.4 °C (at pH Cond. Readings)

Duplicate Taken: Yes No
I.D. Assigned MAFB-80
Field # _____

_____ °C (Immediate)
_____ Conductivity (umhos)
_____ pH
_____ °C (at pH Cond. Readings)

SAMPLES TAKEN:

Phenolics
Cyanide



GROUNDWATER WELL LOG SHEET

Site Mather AFB

Samplers DE/DT

Field # _____

Sample I.D. MAFB-9

Date 6-26-85

Time 1110
4:20 PM

TOTAL WELL DEPTH 113.8 ~~102.1~~ FT.

S.W.L. 69.21 FT.

WATER COLUMN 45.59 FT.

WATER VOLUME 29.78 G.

X3= 89.54 G.

FLOW RATE 3.3 G.P.M.

PUMPING TIME REQUIRED 27 MINS.

ACTUAL PUMPING TIME 27 MINS.

BAILED _____

SAMPLES TAKEN:

VDA
 TIC
 Oil Grease
 Phenolics
 Cyanide
 Metals

20.5 °C (Immediate)

1093 Conductivity (umhos)

6.7 pH

27.5 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned MAFB-90

Field # _____

SAMPLES TAKEN:

same as above

20.5 °C (Immediate)

1097 Conductivity (umhos)

6.7 pH

27.9 °C (at pH Cond. Readings)



GROUNDWATER WELL LOG SHEET

Site Mather AFB

Samplers DB 2

Field # _____

Sample I.D. MAFB-10

Date 10-20-85

Time 10:15

TOTAL WELL DEPTH 107.9 FT.

S.W.L. 74.32 FT.

WATER COLUMN 33.58 FT.

WATER VOLUME 22 G.

X3= 66 G.

FLOW RATE 3.3 G.P.M.

PUMPING TIME REQUIRED 20 MINS.

ACTUAL PUMPING TIME _____ MINS.

BAILED _____

SAMPLES TAKEN:

VOL
 TDC
 O+G
 Metals
 Phenolics
 Cyanide

20 °C (Immediate)

280 Conductivity (umhos)

7.6 pH

23.7 °C (at pH Cond. Readings)

Duplicate Taken: Yes No

I.D. Assigned MAFB-100

Field # _____

SAMPLES TAKEN:

Same as above

20 °C (Immediate)

283 Conductivity (umhos)

7.5 pH

23.6 °C (at pH Cond. Readings)

Cloudy and brown at first
slightly cloudy after 10 min

GROUNDWATER WELL LOG SHEET

Site MARLBOROUGH, CA Samplers TRP/J

Field # _____

Sample I.D. MARB-11 Date 6-26-85 Time 1530

TOTAL WELL DEPTH 108 FT.
S.W.L. 73.79 FT.
WATER COLUMN 34.2 FT.
WATER VOLUME 22.3 G.
X3= 67 G.
FLOW RATE 3.3 G.P.M.
PUMPING TIME REQUIRED 20 MINS.
ACTUAL PUMPING TIME 20 MINS.
BAILED 0+5

SAMPLES TAKEN:

VOA
TOC
OTG
Metals
Phenolics
Cyanide

20 °C (Immediate)
310 Conductivity (umhos)
7.5 pH
24.2 °C (at pH Cond. Readings)

Duplicate Taken: Yes No
I.D. Assigned MARB-110
Field # _____

SAMPLES TAKEN:

Very orange brown discharge at first. Cleared after a few minutes

_____ °C (Immediate)
_____ Conductivity (umhos)
_____ pH
_____ °C (at pH Cond. Readings)

Phenolics
Cyanide

APPENDIX H

SAMPLE CHAIN-OF-CUSTODY RECORDS

ROUND 85-1



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Richard Jones
Phone: _____

SHIP TO: Ray F Weston Inc.
2516 Wilson Rd. S.W.
Lisville, IA 19353

ATTENTION: Judi Pate
Phone No. 215-524-0180

SHIPPING INFORMATION

Location MALDEN AFB
Shipper Ray F Weston Inc.
Address 7720 Lorraine Ave Ste 105
Stockton CA 95210
Date Shipped 3 June 85
Shipment Service Federal Express
Airbill No. _____
Cooler No. _____

Relinquished by: (Signature)	Received by: (Signature)	Date/Time
<u>Richard Jones</u>		
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature)	Date/Time
	<u>Victoria R. Dinkens</u>	<u>6/4/85 10:00</u>

Analysis laboratory should complete "sample cond upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
—	1	FC-1	33,35	—	<u>good</u>

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) *Richard Jones*

Phone: _____

SHIP TO: _____

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION

Location *Mather AFB*

Shipper *Roy F. Weston Inc.*

Address *1720 Lorraine Ave Ste 115
Stockton CA 95210*

Date Shipped *31 Mar 85*

Shipment Service *Hand Carried*

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature)	Received by: (Signature)	Date/Time
<i>Richard Jones</i>		

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
		<i>3 FF-1</i>			<i>OK</i>
	<i>2</i>	<i>↓</i>	<i>↓</i>	<i>VOL (EPA 602)</i>	<i>↓</i>
	<i>1</i>	<i>↓</i>	<i>↓</i>	<i>PCB'S</i>	<i>↓</i>

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) R. V. Jones

Phone: 209-957-3405

SHIP TO:
Roy F. Weston Inc.
256 W. L. H. Pool Rd.
Loomville, PA 19353

ATTENTION: Judy Forta
 Phone No. 215-524-0181

SHIPPING INFORMATION

Location Mathor AFB PA

Shipper Roy F. Weston Inc.

Address 7720 Lorraine Ave #165
Stocketon PA

Date Shipped 3 June 85

Shipment Service Federal Express

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <u>R. V. Jones</u>	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received for laboratory by: (Signature) <u>Victoria R. D...</u>	Date/Time <u>4/4/85 1000</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	1	PS-2	5-20-85	TCC	<u>good</u> ↓
—	1	↓	↓	Oil and Grease	
—	1	↓	↓	Phenolics	
—	1	↓	↓	Cyanide	



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Richard J. Jones
 Phone: 204-957-3405
 SHIP TO:

 ATTENTION:
 Phone No. _____

SHIPPING INFORMATION
 Location Walter AFB CA
 Shipper Roy F. Nestor Inc.
 Address 7720 Lorraine Ave #105
Starkton CA 95510
 Date Shipped 30 May 85
 Shipment Service Hand Carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Richard J. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Ronald N. Schubert</u>	Date/Time <u>5/31 9:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
—	~	ES-2	5-25		
—	~	↓	↓	VOI (EMT 002)	
—	1	↓	↓	Metals (Cr Cd Pb Ni Ag)	

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca A. Jones
 Phone: 209-957-3405
 SHIP TO:

 ATTENTION: _____
 Phone No.: _____

SHIPPING INFORMATION

Location Mather AFB CA
 Shipper Ray F. Weston Inc.
 Address 7720 Lorraine Ave. #105
Stockton CA 95210
 Date Shipped 30 May 85
 Shipment Service Hand Carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Rebecca A. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>DOUBLA D. SCHULTZ</u>	Date/Time <u>5/31 9:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	1	535	5 30 85	<u>Mather AFB CA</u>	<u>OK</u>

Remarks: _____

H-5



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Reborah D. Jones
Phone: _____
SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE #105
STOCKTON, CA 95210
ATTENTION: _____
Phone No. _____

SHIPPING INFORMATION
Mather AFB
Location _____
Shipper ROY F. WESTON INC.
Address 7720 LORRAINE AVE #105
STOCKTON, CA 95210
Date Shipped 3 June 85
Shipment Service Hand Carried
Airbill No. _____
Cooler No. _____

Relinquished by: (Signature) <u>Reborah D. Jones</u>	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received for laboratory by: (Signature) <u>KATHIE N. SCHULTZ</u>	Date/Time <u>6/4 11:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
---	2	3-1	6-3-85	VDA TEPA 501	OK
---	2	"	"	VDA TEPA 1002	↓

Remarks: _____
H-6



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah L. Jones
Phone: _____

SHIP TO: Roy F. Weston Inc.
Weston Way
Nest Chester, PA 19380

ATTENTION: Judy Porta
Phone No. 215-524-0180

SHIPPING INFORMATION
Location Mather AFB

Shipper ROY F. WESTON INC.
Address 7720 LORRAINE AVE, #105
STOCKTON, CA 95210

Date Shipped 4 June 85
Shipment Service Fed Ex

Airbill No. _____
Cooler No. _____

Relinquished by: (Signature) <u>Deborah L. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Victoria B. Dondoro</u>	Date/Time <u>6/5/85 9:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
<u>—</u>	<u>1</u>	<u>cont-1</u>	<u>6-3-85</u>	<u>TCU</u>	<u>good</u>
<u>—</u>	<u>1</u>	<u>"</u>	<u>"</u>	<u>fat and release</u>	<u>good</u>

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah L. Jones
Phone:
SHIP TO: ROY E. WESTON INC.
7720 LORRAINE AVE #105
STOCKTON, CA 95210
ATTENTION:
Phone No.

SHIPPING INFORMATION
Location: Weather AFB
Shipper: ROY F. WESTON INC.
Address: 7720 LORRAINE AVE #105
STOCKTON, CA 95210
Date Shipped: 3 June 85
Shipment Service: Hand Carried
Airbill No.
Cooler No.

Table with 3 columns: Relinquished by: (Signature), Received by: (Signature), Date/Time. Includes signature of Deborah L. Jones and Philip D. Schuber.

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Table with 6 columns: Sample Number, No. Of Cont., Site Identification, Date Sampled, Analysis Requested, Sample Cond Upon Receipt. Contains handwritten data for four samples.

Remarks
H-8



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah L Jones
Phone:
SHIP TO: Roy F. Weston, Inc
Western Way
First Chester, PA 19380
ATTENTION: Judy Porta
Phone No. 215-524-0180

SHIPPING INFORMATION
Location: Mather AFB
Shipper: ROY F. WESTON INC
Address: 7720 LORRAINE AVE #1055
STOCKTON, CA 95210
Date Shipped: 4 JUN 85
Shipment Service: Federal EX
Airbill No.
Cooler No.

Table with 3 columns: Relinquished by: (Signature), Received by: (Signature), Date/Time. Includes signature of Deborah L Jones and Patrick K. Dindors.

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Table with 6 columns: Sample Number, No. Of Cont., Site Identification, Date Sampled, Analysis Requested, Sample Cond. Upon Receipt. Includes handwritten entries for TOC and Oil and Grease analysis.

Remarks:
H-9



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah L. Jones
 Phone: _____
 SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE. #105
STOCKTON, CA 95210
 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION

Location Mather AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE. #105
STOCKTON, CA 95210
 Date Shipped 3 June 85
 Shipment Service Hand Carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Deborah L. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Philip D. Schuler</u>	Date/Time <u>6/11/85</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	2	R-9	6-3-85	VCA (EPA 601)	OK
—	2	"	"	VCA (EPA 602)	✓

Remarks _____

H-10



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Reborah L. Jones
 Phone: _____
 SHIP TO:
Roy F. Weston, Inc.
Weston Way
West Chester PA 19380
 ATTENTION: Judy Porta
 Phone No. 215-521-0180

SHIPPING INFORMATION
 Location Melher AFB
ROY F. WESTON INC.
 Shipper 7720 LORRAINE AVE #105
 Address STOCKTON, CA 95210
 Date Shipped 4 June 85
 Shipment Service Fed Ex
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Reborah L. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>William K. Donders</u>	Date/Time <u>6/5/85 9:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	1	A-9	6-3-85	TC	good
—	1	"	"	Oil and Grease	—

Remarks:
 H-11



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Reborah L. Jones

SHIPPING INFORMATION *MATHEW PER*

Phone: _____

Location ROY F. WESTON INC.

SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE. #105
STOCKTON, CA 95210

Shipper 7720 LORRAINE AVE. #105

Address STOCKTON, CA 95210

Date Shipped June 3, 1985

Shipment Service Hand-Carried

ATTENTION: _____

Airbill No. _____

Phone No. _____

Cooler No. _____

Relinquished by: (Signature)	Received by: (Signature)	Date/Time
<u>Reborah L. Jones</u>		
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature)	Date/Time
	<u>H. Philip N. Schulz</u>	<u>6/4 11:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	<u>2</u>	<u>JTC</u>	<u>6-3-85</u>	<u>VIA (FA-601)</u>	<u>OK</u>
	<u>2</u>	<u>↓</u>	<u>↓</u>	<u>VIA (FA-602)</u>	<u>↓</u>
	<u>1</u>	<u>JTC</u>	<u>6-3-85</u>	<u>Metal: Cr, Pb, Cd, Ni, Ag</u>	<u>↓</u>

Remarks: _____

H-12



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca Jones
 Phone: _____
 SHIP TO:
ROY F WESTON
WESTON WAY
WEST CHESTER, PA 19380
 ATTENTION: JUDY PORTA
 Phone No. 215-579-0180

SHIPPING INFORMATION
 Location Mather AFB
 Shipper ROY E. WESTON INC.
 Address 7720 LORRAINE AVE #105
STOCKTON, CA 95210
 Date Shipped June 4 1985
 Shipment Service Fed - X
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Rebecca Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Victoria K. Dondoro</u>	Date/Time <u>6/5/85 9:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	1	JTC	6-3-85	TGC	<div style="font-size: 2em;">I</div> <div style="font-size: 1.5em;">good</div>
—	1	↓	↓	O+G	
—	1	↓	↓	Phenolics	
—	1	↓	↓	Cyanide	
—	1	JTC-1	6-3-85	Phenolics	
—	1	↓	↓	Cyanide	

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Robert Jones

Phone: _____

SHIP TO:

ROY F. WESTON INC.
7720 LORRAINE AVE. #109
STOCKTON, CA 95210

ATTENTION: _____

Phone No _____

SHIPPING INFORMATION

MATHER AFB

Location ROY F. WESTON INC.

Shipper 7720 LORRAINE AVE. #109

Address STOCKTON, CA 95210

Date Shipped June 3, 1985

Shipment Service Hand-carried

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) Robert Jones

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received for laboratory by: (Signature) Kathy N. Schell

Date/Time 6/4 11:30

Analysis laboratory should complete "sample cond upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
—	2	<u>WPA</u>	<u>6/3/85</u>	<u>NOA (EPA-602)</u>	<u>OK</u>
—	2	↓	↓	<u>Metals: Cr, Pb, Cd, Ni, As</u>	↓
—	1	↓	↓	<u>Dist/Herb</u>	↓
—	3	↓	↓		

Remarks _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca d. Jones

Phone: _____

SHIP TO: ROY F WESTON
WESTON WAY
WEST CHESTER, Pa 19380

ATTENTION: JUDY PORTA

Phone No. 215-524-0180

SHIPPING INFORMATION

Location MATHER AFB

Shipper ROY E. WESTON INC.

Address 5720 LORRAINE AVE. #105

STOCKTON, CA 95210

Date Shipped June 4 1985

Shipment Service Fed-X

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) Rebecca d. Jones

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received for laboratory by: (Signature) Peter K. Dando

Date/Time 6/5/85 9:30

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
<u>1</u>	<u>1</u>	<u>G-1</u>	<u>6-3-85</u>	<u>TLC</u>	<u>good</u>
<u>2</u>	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>O+G</u>	<u>1</u>
<u>3</u>	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>DMH</u>	<u>1</u>



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) *Richard A. Jones*

Phone: _____

SHIP TO: ROY F. WESTON INC.
7720 LORRAINE AVE #105
STOCKTON, CA 95210

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION
Location MATRIC FFB

Shipper ROY F. WESTON INC.

Address 7720 LORRAINE AVE #105

STOCKTON, CA 95210

Date Shipped June 3, 1985

Shipment Service hard packed

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <i>Richard A. Jones</i>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <i>Walter W. Schultze</i>	Date/Time <i>6/4 11:30</i>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
---	2	<i>0-2</i>	<i>5-30-85</i>	<i>VOA (EPA 101)</i>	<i>OK</i>
---	2			<i>VOA (EPA 602)</i>	
---	1			<i>Metals: Cr, Pb, Cd, Ni, Ag</i>	
---	3			<i>Pest Herbs</i>	

Remarks _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Selma H. Jones
 Phone: _____
 SHIP TO: ROY F WESTON
WESTON WAY
WEST CHESTER, PA 19380
 ATTENTION: JUDY PORTA
 Phone No. 215-524-0180

SHIPPING INFORMATION
 Location MATHER AFB
ROY F. WESTON INC.
 Shipper _____
7720 LORRAINE AVE #105
 Address _____
STOCKTON, CA 95210
 Date Shipped June 4, 1955
 Shipment Service Fed-X
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Selma H. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Victoria K. Dahlen</u>	Date/Time <u>6/8/55 9:50</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	1	B-2	5-28-55	T/C	grad
	1	↓	↓	U+G	I
	1	↓	↓	DMN	

Remarks: _____
 H-17



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) *Rebecca A. Jones*

Phone: _____

SHIP TO: ROY F. WESTON INC.
7720 LORRAINE AVE. #105
STOCKTON, CA 95210

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION
IIIHTHEIK AF-B

Location _____

Shipper ROY F. WESTON INC.

Address 7720 LORRAINE AVE. #105
STOCKTON, CA 95210

Date Shipped June 3 1985

Shipment Service hand carried

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) *Rebecca A. Jones*

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received for laboratory by: (Signature) *Philip D. Schell*

Date/Time 6/4 11:3

Analysis laboratory should complete "sample cond. upon receipt" section below. sign and return copy to Shipper

Sample Number	No Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
<u>1</u>	<u>3</u>	<u>R-3</u>	<u>5-30-85</u>	<u>VIA (FPA-602)</u>	<u>OK</u>
<u>2</u>	<u>2</u>	<u>↓</u>	<u>5-31-85</u>	<u>VIA (FPA-602)</u>	<u>↓</u>
<u>3</u>	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>Metal: Cr, Pb, Cd, Ni, Ag</u>	<u>↓</u>
<u>4</u>	<u>3</u>	<u>↓</u>	<u>↓</u>	<u>pest, Herb</u>	<u>↓</u>

Remarks _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah L. Jones

Phone: _____

SHIP TO:

ROY F WESTON
WESTON WAY
WEST CHESTER, PA 19380

ATTENTION: JUDY PORTA

Phone No. 715-574-0180

SHIPPING INFORMATION

Location MATHER AFB

Shipper ROY F. WESTON INC.

Address 7720 LORRAINE AVE. #105
STOCKTON, CA 95210

Date Shipped June 4, 1985

Shipment Service Fed-X

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <u>Deborah L. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Victoria K. Dondoro</u>	Date/Time <u>6/5/85 9:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	1	0-3	6-30-85	TOC	good
—	1	↓	↓	OTG	1
—	1	↓	↓	DMN	1

Remarks _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca Jones
 Phone: _____
 SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE #105
STOCKTON, CA 95210
 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION

Location MATHER AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE #105
STOCKTON, CA 95210
 Date Shipped JUNE 3, 1985
 Shipment Service hand carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Rebecca Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Philip N. Schultz</u>	Date/Time <u>6/4 11:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
—	2	B-4	5-2-85	VQA (EPA 601)	OK
—	2	↓	↓	VQA (EPA 602)	↓
—	1	↓	↓	VQA Metals: Cr, Pb, Cd, Ni, Ag	
—	3	↓	↓	Pest/Herb	
—	2	B-40	5-3-85	VQA (EPA 601)	
—	2	↓	↓	VQA (EPA 602)	
—	1	↓	↓	Metal: Cr, Pb, Cd, Ni, Ag	
—	3	↓	↓	Pest/Herb	

Remarks _____

H-20



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca A. Jones
 Phone: _____
 SHIP TO: ROY F WESTON
WESTON WAY
WEST CHESTER, PA 19380
 ATTENTION: JUDY PORTA
 Phone No. 215-524-0180

SHIPPING INFORMATION
Mather AFB
 Location ROY F. WESTON INC.
 Shipper 7720 LORRAINE AVE #105
 Address STOCKTON, CA 95210
 Date Shipped June 4, 1985
 Shipment Service Fed-X
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Rebecca A. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Victoria K. Donchin</u>	Date/Time <u>6/5/85 9:50</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	1	A4	5-28-85	TIC	good
—	1	↓	↓	O+G	
—	1	↓	↓	DMN	
—	1	B-40	5-28-85	TIC	
—	1	↓	↓	O+G	
—	1	↓	↓	DMN	
—	4				

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Reborah Jones
 Phone: _____
 SHIP TO:
Roy F. Weston Inc.
Weston Way
West Chester, PA 19380
 ATTENTION: Judy Porta
 Phone No. 215-524-0180

SHIPPING INFORMATION
MAKER AFFB
 Location: ROY F. WESTON INC.
 Shipper: 7720 LORRAINE AVE #105
 Address: STOCKTON, CA 95210
 Date Shipped: 4 June 85
 Shipment Service: Fed EX
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Reborah Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Victoria K. Jandino</u>	Date/Time <u>4/5/85 9:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
—	1	FH-1	6-3-85	Oil	good
—	1	↓	↓	Oil and Grease	good

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Reborah A. Jones

Phone: _____

SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE. #105
STOCKTON, CA 95210

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION
Mather AFB

Location _____

Shipper ROY F. WESTON INC.

Address 7720 LORRAINE AVE. #105

STOCKTON, CA 95210

Date Shipped 3 June 85

Shipment Service Hand Carried

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <u>Reborah A. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Halpin D. Schuller</u>	Date/Time <u>6/4 11:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	2	EH-1	6-3-85	WA (EPA 601)	OK
—	2	↓	↓	YCA (EPA 602)	↓

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Seborah Jones
 Phone: _____
 SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE. #105
STOCKTON, CA 95210
 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION

Location Mather AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE. #105
STOCKTON, CA 95210
 Date Shipped 5 June 85
 Shipment Service Hand Carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Seborah Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Arthur N. Schultz</u>	Date/Time <u>6/4 11/30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
—	2	PH-2	6-3-85	VCA (EPA 602)	OK
—	2	"	"	VCA (EPA 602)	↓

Remarks: _____

H-24



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebekah Jones
 Phone: _____
 SHIP TO:
Roy F. Weston Inc.
Weston Hwy
West Chester PA 19380
 ATTENTION: Judith Porta
 Phone No. 215-524-0180

SHIPPING INFORMATION
 Location Mather AFB
ROY F. WESTON INC.
 Shipper 7720 LORRAINE AVE #105
 Address STOCKTON, CA 95210
 Date Shipped 3 JUNE 85
 Shipment Service Fed EX
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature)	Received by: (Signature)	Date/Time
<u>Rebekah Jones</u>		
_____	_____	_____
_____	_____	_____
_____	_____	_____
_____	<u>Victoria J. Donadio</u>	<u>6/3/85 9:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	1	<u>FM-2</u>	<u>6-3-85</u>	<u>TIC</u>	<u>good</u>
—	1	<u>↓</u>	<u>↓</u>	<u>Oil and Grease</u>	<u>✓</u>

Remarks _____
 H-25



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Robert A. Jones
 Phone: _____
 SHIP TO:
ROY F. WESTON INC
7720 LORRAINE AVE #105
STOCKTON, CA 95210
 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION

Location Walton AFB
 Shipper _____
 Address ROY F. WESTON INC.
7720 LORRAINE AVE. #105
STOCKTON, CA 95210
 Date Shipped 3 June 85
 Shipment Service Hand Carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Robert A. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Arthur N. Schuler</u>	Date/Time <u>4/4</u> <u>11:00</u>

Analysis laboratory should complete "sample cond/upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
—	2	FH-30	6-3-85	VOA (EPA 602)	OK
—	2	↓	↓	VOA (EPA 602)	↓
—	2	FH-30	↓	VOA (EPA 601)	↓
—	2	↓	↓	VOA (EPA 602)	↓

Remarks: _____

H-26



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Reborah d Jones

Phone: _____

SHIP TO: Roy F. Weston Inc.
Weston Way
West Chester PA 19380

ATTENTION: Judy Porta

Phone No. 215 - 524-0180

SHIPPING INFORMATION

Location Mather AFB
ROY F. WESTON INC.

Shipper 7720 LORRAINE AVE #105

Address STOCKTON, CA 95210

Date Shipped 4th June 85

Shipment Service ~~HODSON~~ Fed Ex

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature)	Received by: (Signature)	Date/Time
<u>Reborah d Jones</u>		
	<u>Victor R Doolen</u>	<u>6/5 9:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below. sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
<u>---</u>	<u>1</u>	<u>FH-3</u>	<u>6-5-85</u>	<u>TOC</u>	<u>good</u>
<u>---</u>	<u>1</u>	<u>↓</u>	<u>"</u>	<u>Oil and grease</u>	<u>---</u>
<u>---</u>	<u>1</u>	<u>FH-30</u>	<u>"</u>	<u>TOC</u>	<u>---</u>
<u>---</u>	<u>1</u>	<u>↓</u>	<u>"</u>	<u>Oil and grease</u>	<u>---</u>



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Reborah L. Jones
 Phone: _____
 SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE #105
STOCKTON, CA 95210
 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION
7720 LORRAINE AVE
 Location ROY F. WESTON INC.
 Shipper 7720 LORRAINE AVE #105
 Address STOCKTON, CA 95210
 Date Shipped 5/19/95
 Shipment Service Hand Carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature)	Received by: (Signature)	Date/Time
<u>Reborah L. Jones</u>		
	<u>KATHIE N. SCHULTZ</u>	<u>6/4/95</u>

Analysis laboratory should complete "sample cond. upon receipt" section below. sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
1	2	EM-6	5-25-95	...	OK
2	2	"	"	WATER (2)	OK

Remarks: _____
 H-24



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Reborah Jones

Phone: _____

SHIP TO:
Roy F. Weston Inc
Dakota Hwy
West Chester, PA 19380

ATTENTION: Judy Porta

Phone No. 615-584-0180

SHIPPING INFORMATION
MAJOR AFB

Location: ROY F. WESTON INC.

Shipper: 7720 LORRAINE AVE. #105

Address: STOCKTON, CA 95210

Date Shipped: 4/1/85

Shipment Service: Fed Ex

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature)
Reborah Jones

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature)
[Signature]

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received for laboratory by: (Signature)
Victoria R. Jandura

Date/Time
4/1/85 9:30

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. of Containers	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
			6-3-85	TCE	<u>good</u>
			"	Oil and Grease	

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Robert D. Jones

Phone: _____

SHIP TO: _____

ATTENTION: _____

Phone No _____

SHIPPING INFORMATION

Location Maxwell AFB

Shipper Roy F. Weston Inc.

Address 51720 Lockhart Ave #100

Madison AL 36010

Date Shipped 31 May 85

Shipment Service Hand Carried

Airbill No _____

Cooler No _____

Relinquished by: (Signature)

Robert D. Jones

Received by (Signature)

Date/Time

Relinquished by: (Signature)

Received by: (Signature)

Date/Time

Relinquished by: (Signature)

Received by: (Signature)

Date/Time

Relinquished by: (Signature)

Received for laboratory by (Signature)

Date/Time

William Scheeley 6/3 82

Analysis laboratory should complete "sample cond upon receipt" section below sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond upon Receipt
	1	↓	↓	PCB	↓
	2	↓	↓	PCB	↓
					OK



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) [Signature]

Phone: 0

SHIP TO: Roy F. Weston Inc.
2200 Redwood Road
Walpole, MA 01553

ATTENTION July Purke

Phone No 615 524 0181

SHIPPING INFORMATION

Location Mather AFB

Shipper Roy F. Weston Inc.

Address 7720 Lorraine Ave #105
Stockton CA 95210

Date Shipped 3 June 55

Shipment Service Federal Express

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <u>[Signature]</u>	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received for laboratory by: (Signature) <u>[Signature]</u>	Date/Time <u>6/4/55 10:50</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
<u>1</u>	<u>1</u>	<u>✓</u>	<u>✓</u>	<u>[Signature]</u>	<u>good</u>



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Richard A. Jones
 Phone _____
 SHIP TO: _____

 ATTENTION: _____
 Phone No _____

SHIPPING INFORMATION
 Location Walter AFB
 Shipper Ruff F. Weston Inc.
 Address 7720 Lorraine Ave STE 105
Socaton GA 30210
 Date Shipped 31 May 85
 Shipment Service Hand Carried
 Airbill No _____
 Cooler No _____

Relinquished by: (Signature) <u>Richard A. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Walter N. Schiller</u>	Date/Time <u>6/3/85</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
—	2	MAFB RD	5-31-85	VOC (EPA 601)	✓
—	2	↓	↓	VOC (EPA 602)	
—	1	↓	↓	PCBS	
—	2	MAFB RD	5-31-85	VOC (EPA 601)	
—	2	↓	↓	VOC (EPA 602)	
—	1	↓	↓	PCBS	



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Richard Jones
 Phone: _____
 SHIP TO: ROY F WESTON INC.
520 WILSON ROAD
LITTLE ROCK AR 71220
 ATTENTION: Judy Porta
 Phone No 515-524-0180

SHIPPING INFORMATION
 Location Walter AFB
 Shipper ROY F. WESTON INC.
 Address 7720 HARRIS AVE. STE. 105
SOUTH LA 75210
 Date Shipped 5/15/85
 Shipment Service Priority
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Richard Jones</u>	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received for laboratory by: (Signature) <u>Victoria R. Dondos</u>	Date/Time <u>6/4/85 10:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below. sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
---	---	<u>NAFB-20</u>	<u>5-15-85</u>	<u>TOC</u>	<u>Good</u>
---	<u>1</u>	<u>NAFB-20</u>	<u>5-15-85</u>	<u>TOC</u>	
---	<u>1</u>	<u>NAFB-20</u>	<u>5-15-85</u>	<u>TOC</u>	
---	---	---	---	---	---
---	---	---	---	---	---



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) [Signature]

Phone: _____

SHIP TO: _____

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION

Location Mather AFB

Shipper Ray F. Weston Inc.

Address 7720 Lorraine Ave Ltr 105
Hickory CA 95010

Date Shipped 31 May 85

Shipment Service Freighters Hand Carried

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <u>[Signature]</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Kathryn N. Schullery</u>	Date/Time <u>6/3 18:25</u>

Analysis laboratory should complete "sample cond upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
---	<u>2</u>	<u>---</u>	<u>---</u>	<u>---</u>	<u>OK</u>
---	<u>2</u>	<u>---</u>	<u>---</u>	<u>PER ORD-2</u>	<u>---</u>
---	<u>1</u>	<u>---</u>	<u>---</u>	<u>PER 3</u>	<u>---</u>



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) *Nicholas S. Jones*

Phone: _____

SHIP TO:

ROY F. Weston Inc.
250 Nelson Pool Rd.
Ligonie PA 15353

ATTENTION: Judy Porta

Phone No. 315-524-0150

SHIPPING INFORMATION

Location MALDEN AFB

Shipper ROY F. Weston Inc.

Address 7770 LOURAINNE AVE STE 105
STOCKTON CA 95210

Date Shipped 3 JUN 85

Shipment Service FEDERAL EXPRESS

Airbill No. _____

Cooler No _____

Relinquished by: (Signature)	Received by: (Signature)	Date/Time
<u><i>Nicholas S. Jones</i></u>		
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature)	Date/Time
	<u><i>Mark R. Jones</i></u>	<u>6/4/85 10:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
<u>1</u>	<u>1</u>	<u>1</u>	<u>6-3-85</u>	<u>Oil + metal</u>	<u>good</u>

Remarks



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) *Deborah D. Jones*

Phone: _____

SHIP TO: _____

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION

Location *Walter AFB*

Shipper *Rev F. Weston, Inc.*

Address *7720 Lorraine Ave. #105
Stockton CA 95210*

Date Shipped *31 May 85*

Shipment Service *Air Mail*

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <i>Deborah D. Jones</i>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <i>Arthur N. Schulltz</i>	Date/Time <i>4/3 8:10</i>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
-	2	1-28-4	3-3-85		OK
-	2	↓	↓	(CONTAMINATED)	↓
-	1	↓	↓	Metals (Cd, Cr, Pb, Ni, Ag)	↓
-	3	↓	↓	Pest/Herb	↓



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah A Jones
Phone: _____

SHIP TO: Roy F Weston Inc
256 Welsh Pool Rd.
Lenoirville PA 17353

ATTENTION: Judy Perta
Phone No. 215-524-0180

SHIPPING INFORMATION
Location Mather AFB
Shipper Roy F. Weston Inc.
Address 7720 Lorraine Ave. St 105
Stockton, CA 95210
Date Shipped 3 June 85
Shipment Service Federal Express
Airbill No. _____
Cooler No. _____

Relinquished by: (Signature) <u>Deborah A Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory (Signature) <u>Victor R. Danduro</u>	Date/Time <u>6/4/85 10:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	1	1452-4	5-29-85	TC	<u>good</u>
—	1	↓	↓	<u>oil and grease</u>	<u>1</u>
—	1	↓	↓	<u>DMN</u>	—

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca Jones

Phone: _____

SHIP TO: _____

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION

Location Malibu AFB

Shipper Roy F. Weston, Inc.

Address 7720 Lorraine Ave. #H 105
Stockton CA 95210

Date Shipped 31 Mar 35

Shipment Service Hand Carried

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) Rebecca Jones

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received by: (Signature) _____

Date/Time _____

Relinquished by: (Signature) _____

Received for laboratory by: (Signature) Philip N. Schilling

Date/Time 6/3 | 8:2

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
—	2	MAFB-5	5-31-35	VCA (EPA 602)	OK
—	2	↓	↓	VCA (EPA 602)	
—	1	↓	↓	Metals (Cr, Cd, Pb, Ni, Ag)	
—	3	↓	↓	Pest/Herb	
—	2	MAFB-50	5-31-35	VCA (EPA 601)	
—	2	↓	↓	VCA (EPA 602)	
—	1	↓	↓	Metals (Cr, Cd, Pb, Ni, Ag)	
—	3	↓	↓	Pest/Herb	
—					
—					
—					
—					
—					
—					
—					
—					
—					
—					
—					

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Robert S. Jones
 Phone: _____
 SHIP TO:
Rev F Weston Inc.
2516 W. L. h Pool Rd.
Scottsdale PA
 ATTENTION: Judy Porta
 Phone No. 715-524-0180

SHIPPING INFORMATION
 Location Walter AFB
 Shipper Rev F Weston Inc.
 Address 7720 Lorraine Ave. #105
Stockton CA 95210
 Date Shipped 5-27-85
 Shipment Service Hand Carried Fed Ex
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Robert S. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Victoria H. ...</u>	Date/Time <u>4/85 10:10</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
—	1	—	5-27-85	—	good
—	1	↓	↓	Oil + Grease	
—	1	↓	↓	DMN	
—	1	MAFB-50	5-31-85	TEC	—
—	1	↓	↓	Oil and Grease	
—	1	↓	↓	DMN	

Remarks: _____
 H-39



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah A. Jones

Phone: _____

SHIP TO: _____

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION

Location Mather AFB

Shipper Ray F. Weston, Inc

Address 7720 Lorraine Ave #105
Stockton CA 95210

Date Shipped 30 May 85

Shipment Service Hand Carried

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature)	Received by: (Signature)	Date/Time
<u>Deborah A. Jones</u>		
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature)	Date/Time
	<u>Arthur N. Schutzy</u>	<u>3/31 9:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
<u>1</u>	<u>2</u>	<u>MAFB-6</u>	<u>5/30/85</u>	<u>Asst. (Pb, Cd, Fe, Ni, Ag)</u>	<u>OK</u>
<u>2</u>	<u>2</u>	↓	↓	<u>Asst. (Pb, Cd, Fe, Ni, Ag)</u>	↓
<u>3</u>	<u>1</u>	↓	↓	<u>Metals (Pb, Cd, Fe, Ni, Ag)</u>	↓
<u>4</u>	<u>3</u>	↓	↓	<u>Pesticides/Herbicides</u>	↓

Remarks: _____

H-40



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca A. Jones
 Phone: (609) 957-3405

SHIP TO:
Roy F. Weston Inc.
556 Welsh Pool Rd.
Lionville PA 19353

ATTENTION: Judy Porta
 Phone No. 215-524-0181

SHIPPING INFORMATION

Location Major AFB
 Shipper Roy F. Weston Inc.
 Address 7730 Lorraine Ave #105
Stanton PA 15310
 Date Shipped 5/14/85
 Shipment Service Federal Express
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature)	Received by: (Signature)	Date/Time
<u>Rebecca A. Jones</u>		
_____	_____	_____
_____	_____	_____
_____	_____	_____
_____	_____	_____

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
---	1	---	---	---	good
---	1	↓	↓	oil and grease	1
---	1	↓	↓	DMU	---

Remarks: _____

H-41



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca D. Jones
 Phone 209-957-3405
 SHIP TO: _____

 ATTENTION: _____
 Phone No _____

SHIPPING INFORMATION

Location Mothe AFB
 Shipper ROY F. WESTON, INC
 Address 7720 Westline Ave #101
Stockton CA 95215
 Date Shipped 5/31/85
 Shipment Service Hand Carried
 Airbill No _____
 Cooler No _____

Relinquished by: (Signature) <u>Rebecca D. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Philip N. Schultz</u>	Date/Time <u>5/31 9:00</u>

Analysis laboratory should complete "sample cond upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
—	2	↓	↓	VIA EPA 602	OK
—	2	↓	↓	↓	↓
—	—	↓	↓	↓	↓
—	1	↓	↓	Metals (Pb, Cr, Pb, Ni, Ag)	↓

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Richard J. Jones

Phone: 219-957-3405 U

SHIP TO: Ray F. Weston Inc
5550 West of Pool Rd.
Lawrence, PA 19353

ATTENTION: Luigi Forta

Phone No 215-524-0181

SHIPPING INFORMATION

Location Maxter AFB

Shipper Ray F. Weston Inc.

Address 7720 Lorraine Ave #105
Stockton CA 95210

Date Shipped 3 June 85

Shipment Service Federal Express

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) Richard J. Jones Received by: (Signature) _____ Date/Time _____

Relinquished by: (Signature) _____ Received by: (Signature) _____ Date/Time _____

Relinquished by: (Signature) _____ Received by: (Signature) _____ Date/Time _____

Relinquished by: (Signature) _____ Received for laboratory by: (Signature) Richard J. Jones Date/Time 6/4/85 10:00

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
—	1	MAFB-7	5-30-85	TOC	good
—	1	↓	↓	Oil and Grease	
—	1	↓	↓	Phenolic	
—	1	↓	↓	Cyanide	
—	1	MAFB-7D	↓	Phenolic	
—	1	↓	↓	Cyanide	

Remarks: _____
H-43



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) [Signature]

SHIPPING INFORMATION

Phone: _____

Location Mather AFB CA

SHIP TO:

Shipper _____

Roy F Weston Inc
7720 Lorraine Ave #105
Stockton CA 95210

Address _____

Date Shipped 27 Nov 85

Shipment Service Hand Carried

ATTENTION: _____

Airbill No. _____

Phone No. _____

Cooler No. _____

Relinquished by: (Signature) <u>[Signature]</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>[Signature]</u>	Date/Time <u>5:30 10/6</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper.

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	1	WAFB-8	5/27/85	Metal (Cd, Cr, Pb, Ni, Ag)	OK
—	2	↓	↓	VOC (EPA 601)	↓
—	2	↓	↓	VOC (EPA 602)	↓

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) A. Grandjean

Phone: _____

SHIP TO:

Roy F. Weston, Inc.
216 W. Nelson Road
Remondville, PA 19353

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION

Location: McKee AFB, CA

Shipper: Roy F. Weston, Inc.

Address: 770 Colma Ave

Stockton, CA 95210

Date Shipped: 30 May 85

Shipment Service: Federal Express

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <u>A. Grandjean</u>	Received by: (Signature) <u>[Signature]</u>	Date/Time _____
---	--	--------------------

Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
---------------------------------------	-----------------------------------	--------------------

Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
---------------------------------------	-----------------------------------	--------------------

Relinquished by: (Signature) _____	Received for laboratory by: (Signature) <u>[Signature]</u>	Date/Time <u>[Date]</u>
---------------------------------------	---	----------------------------

Analysis laboratory should complete sample cond. upon receipt section below, sign and return copy to Shipper.

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
_____	1	<u>HAFB-8</u>	<u>5-29-85</u>	<u>TOC</u>	<u>good</u>
_____	_____	_____	_____	<u>Oil and Grease</u>	
_____	_____	_____	_____	<u>Phenolics</u>	
_____	_____	_____	_____	<u>Cyanide</u>	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	
_____	_____	_____	_____	_____	

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) _____

Phone: _____

SHIP TO: _____

Roy F. Weston, Inc.
7720 Lorraine Ave. #105
Stockton CA 95210

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION

Location: *7720 Lorraine Ave #105*

Shipper: _____

Address: _____

Date Shipped: *29 May 85*

Shipment Service: *Hand*

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <i>[Signature]</i>	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <i>[Signature]</i>	Date/Time <i>5/30/10:00</i>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
---	<i>1</i>	<i>MAFB-80</i>	<i>5-29-85</i>	<i>Metals (Cd, Cr, Pb, Ni, Ag)</i>	<i>OK</i>
---	<i>2</i>	↓	↓	<i>VOA (EPA 601)</i>	↓
---	<i>2</i>	↓	↓	<i>VOA (EPA 602)</i>	↓

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) [Signature]
Phone: _____
SHIP TO: Dr. F. Winston En...
50 N. Shiloh Rd.
Hickoryville, PA 17353
ATTENTION: _____
Phone No: _____

SHIPPING INFORMATION

Location: Wether AFB CA
Shipper: Dr. F. Winston En...
Address: 1750 N. Main St. #105
Hickoryville, CA 95210
Date Shipped: 30 May 85
Shipment Service: Federal Express
Airbill No: _____
Cooler No: _____

Relinquished by: (Signature)	Received by: (Signature)	Date/Time
<u>[Signature]</u>	<u>[Signature]</u>	
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature)	Date/Time

Analysis laboratory should complete sample cond. upon receipt section below, sign and return copy to Shipper.

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
	<u>1</u>	<u>MAFB 80</u>	<u>5-29-85</u>	<u>TAC</u>	<u>Good</u>
				<u>oil and grease</u>	
				<u>phenolics</u>	
				<u>Cyanide</u>	

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) _____
 Phone _____
 SHIP TO _____

 ATTENTION: _____
 Phone No _____

SHIPPING INFORMATION

Location W. H. A. Co. A
 Shipper _____
 Address _____
 Date Shipped 11/20/01
 Shipment Service Express
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Michael A. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Dorothy D. Schellky</u>	Date/Time <u>8/30/01</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
1	1	W. H. A. Co. A	5/24/01	VOC (EPA 801)	OK
2	2	↓	↓	VOC (EPA 801)	↓
2	2	↓	↓	VOC (EPA 602)	↓



CHAIN OF CUSTODY RECORD

SHIPPING INFORMATION

SAMPLERS: (Signature) *Richard D. Jones*

Phone: 209-957-3405

Location: *McKer AFB, LA*

SHIP TO: *Roy F. Weston, Inc.*

Shipper: *Roy F. Weston, Inc.*

*256 W. 1st Street
Lumberville, PA 17353*

Address: *7705 Lorraine Ave #105*

Stockton, CA 95210

Date Shipped: *20 May 85*

Shipment Service: *Federal Express*

ATTENTION:

Airbill No:

Phone No:

Cooler No:

Relinquished by: (Signature) *Richard D. Jones* Received by: (Signature) _____ Date/Time _____

Relinquished by: (Signature) _____ Received by: (Signature) _____ Date/Time _____

Relinquished by: (Signature) _____ Received by: (Signature) _____ Date/Time _____

Relinquished by: (Signature) _____ Received for laboratory by: (Signature) *[Signature]* Date/Time *5/29/85*

Analysis laboratory should complete sample cond. upon receipt. Section below, sign and return copy to shipper.

Sample Number	No. of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	1	MAFB-9	5-29-85	TOL	<i>good</i>
—	1	↓	↓	Oil & Grease	↓
—	1	↓	↓	Phenolics	↓
—	1	↓	↓	Cyanide	↓
—	1	MAFB-90	5-29-85	Phenolics	↓
—	1	↓	↓	Cyanide	↓

Remarks:

AD-A184 581

INSTALLATION RESTORATION PROGRAM PHASE 2
CONFIRMATION/QUANTIFICATION STAGE 1 VOLUME 2 APPENDICES
(U) WESTON (ROY F) INC WEST CHESTER PA JUN 86
F33615-80-D-4006

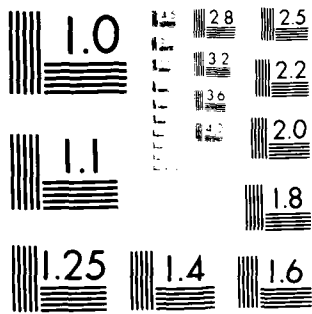
3/5

UNCLASSIFIED

F/G 24/4

NL





MERILEY RESOLUTION TEST CHART
NATIONAL BUREAU OF STANDARDS-1963-A



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) [Signature]
 Phone: _____
 SHIP TO: _____

 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION
 Location WATER AFB GA
 Shipper ROFF WESTON, INC.
 Address 7720 KENNEDY AVE #105
SPICERTON GA 31210
 Date Shipped 5/31/85
 Shipment Service Priority Mail
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>[Signature]</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>KATHY N. SCHUELLER</u>	Date/Time <u>5/31 9:0</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	2	VCA-10	5-30-85	VCA (EPA 601)	OIC
—	2	↓	↓	VCA (EPA 602)	↓
—	1	↓	↓	Metals (Cd, Cr, Pb, Ni, Ag)	↓

Remarks: _____
 H-50



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) [Signature]
 Phone: 909-957-3405

SHIP TO:
Roy F. Weston Inc.
2560 W. 13th St
Stockton, CA 95210

ATTENTION: Judy Porter
 Phone No. 215-524-0181

SHIPPING INFORMATION

Location 110-1100 AFB CA
 Shipper Roy F. Weston Inc
 Address 7720 Hollands Ave #105
Stockton CA 95210
 Date Shipped 3 June 85
 Shipment Service Federal Express
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>[Signature]</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>[Signature]</u>	Date/Time <u>6/4/85 10:50</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
					<u>good</u>
	<u>1</u>			<u>Soils</u>	
	<u>1</u>			<u>Fractures</u>	
	<u>1</u>			<u>Water</u>	
	<u>1</u>	<u>HAFF-100</u>		<u>Aluminum</u>	
	<u>1</u>			<u>Soils</u>	

Remarks: _____

H-51



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah A. Jones
 Phone: _____
SHIP TO: _____

ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION
 Location Walter AFB GA
 Shipper Roy F. Weston, Inc
 Address 7720 Lorraine Ave #105
Stockton MA 01520
 Date Shipped 30 March 85
 Shipment Service Hand Carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Deborah A. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Kathy N. Schuller</u>	Date/Time <u>3/31 9:10</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
—	2	SAFE 11	3.30.85	VIA EPA 602	OK
—	2	↓	↓	Metals (Cr, Cd, Pb, Ni, Ag)	↓

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Kelvin L. Jones
 Phone: 209-957-3405
 SHIP TO:
Roy F. Weston, Inc.
256 Welsh Pool Rd.
Lebanon PA 19353
 ATTENTION: Judy Porta
 Phone No. 215-524-0181

SHIPPING INFORMATION
 Location Mather AFB
 Shipper Roy F. Weston, Inc.
 Address 7720 Lorraine Ave. #105
Stockton, CA 95210
 Date Shipped 3 June 85
 Shipment Service Federal Express
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Kelvin L. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>William J. Porter</u>	Date/Time <u>6/3 10:02</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
---	1	MAFB-11	5-30-85	TOC	<u>good</u>
---	1	↓	↓	Oil and Grease	
---	1	↓	↓	Phenolic	
---	1	↓	↓	Cyanide	
---	1	MAFB-110	5-30-85	TOC	
---	1	↓	↓	Oil and Grease	
---	1	↓	↓	Phenolic	
---	1	↓	↓	Cyanide	

Remarks: _____

H-53



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca Jones
 Phone: _____
 SHIP TO:
Roy F. Weston, Inc.
7720 Lorraine Ave. Ste. 105
Stockton CA 95210
 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION
 Location Mather AFB
 Shipper Roy F. Weston, Inc.
 Address 7720 Lorraine Ave. Ste. 105
Stockton CA 95210
 Date Shipped 30 May 85
 Shipment Service Hand Carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Rebecca Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Kathleen N. Schilling</u>	Date/Time <u>6/3 8:2</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
1	2	MAFB-110	5-30-85	VCA (EPA 602)	OK
2	2	↓	↓	VCA (EPA 602)	↓
3	1	↓	↓	Metals (Cr, Cd, Pb, Ni, Ag)	↓

Remarks: _____
 H-54



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca L. Jones

Phone: 209-957-3405 U

SHIP TO: Roy F. Weston Inc.
256 Welsh Road
Linnville PA 19353

ATTENTION: Judy Porta

Phone No. 215-524-0181

SHIPPING INFORMATION

Location Mather AFB

Shipper Roy F. Weston Inc.

Address 7720 Lorraine Ave #105
Stockton MA 01520

Date Shipped 3 June 85

Shipment Service Federal Express

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <u>Rebecca L. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Victoria P. Jones</u>	Date/Time <u>6/4/85 10:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
—	1	USD	5-20-85	metals	good
—	1	USD		metals	
—	1	USD		Cyanide	
—	1	USD		metals (Cr, Cd, Pb, Ni, Hg)	
—	1	US		oil and grease	
—	1	US		phenols	
—	1	US		Cyanide	
—	1	US		metals (Cr, Cd, Pb, Ni, Hg)	
—	1	DS	5-20-85	oil and grease	
—	1	DS		phenols	
—	1	DS		Cyanide	
—	1	DS		metals (Cr, Cd, Pb, Ni, Hg)	

Remarks: SOIL SAMPLES



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca S. Jones

Phone: _____

SHIP TO: _____

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION

Location: Wether AFB MA
 Shipper: Ray F. Weston Inc.
 Address: 7720 Leaning Ave #105
Stockton MA 05210
 Date Shipped: 20 May 85
 Shipment Service: Hand Carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Rebecca S. Jones</u>	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Arthur J. Schuller</u>	Date/Time <u>5/31 9:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
					OK
	1	USD	↓	VOC, Metals (Cd, Pb, Ni, Hg)	↓
	1	DS	↓		

Remarks



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) _____

Phone: _____

SHIP TO:

Lionville Labs

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION

Location: _____

Shipper: ROY J. WESTON

Address: 4720 LORRAINE AVE SE 105
SILVERDALE, WA 98210

Date Shipped: 6-17-85

Shipment Service: AIRMAIL EXPRESS

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) _____

[Signature]

Received by: (Signature) _____

Date/Time

6/17/85

Relinquished by: (Signature) _____

Received by: (Signature) _____

Date/Time

Relinquished by: (Signature) _____

Received by: (Signature) _____

Date/Time

Relinquished by: (Signature) _____

Received for laboratory by: (Signature) _____

[Signature]

Date/Time

6/18/85

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
<u>3-41-01</u>	<u>1</u>	<u>U.S.</u>	<u>5/30/85</u>	<u>metals - Pb, Cr, Cd, Ni, Ag</u>	<u>good</u>
<u>3-41-02</u>	<u>1</u>	<u>U.S.-D.</u>	<u>↓</u>	<u>↓</u>	<u>↓</u>
<u>3-41-03</u>	<u>1</u>	<u>D.S.</u>	<u>↓</u>	<u>↓</u>	<u>↓</u>

Remarks: Case number 10000154 & 100

ROUND 85-2



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Robert Jones
Phone: _____

SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE. #105
STOCKTON, CA 95210

ATTENTION:
Phone No. _____

SHIPPING INFORMATION

Location Mather AFB
Shipper ROY F. WESTON INC.
Address 7720 LORRAINE AVE. #105
STOCKTON, CA 95210
Date Shipped 27-JUN-85
Shipment Service hard carried
Airbill No. _____
Cooler No. _____

Relinquished by: (Signature) <u>Robert Jones</u>	Received by: (Signature)	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>M. Maldonado</u>	Date/Time <u>6/28 7 AM</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
	<u>2</u>	<u>FB-1</u>	<u>6/27/85</u>	<u>EPA 601</u>	<u>OK</u>
	<u>2</u>	<u>↓</u>	<u>↓</u>	<u>EPA 602</u>	<u>↓</u>
	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>PCB</u>	<u>↓</u>
	<u>3</u>	<u>↓</u>	<u>↓</u>	<u>Pest/Herb</u>	<u>↓</u>

Remarks



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah Jones
 Phone: _____
 SHIP TO:
Roy F. Weston Inc
Lionsville Lab
256 Walnut Pool Rd.
Lionsville PA 19353
 ATTENTION: Judith Porta
 Phone No. 615-524-0180

SHIPPING INFORMATION
 Location Mathers AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE. #105
STOCKTON, CA 95210
 Date Shipped 1 July 85
 Shipment Service Fed Ex
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Deborah Jones</u>	Received by: (Signature)	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>John R. Jones</u>	Date/Time <u>7/1/85 9:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
	<u>1</u>	<u>FB-1</u>	<u>6-27-85</u>	<u>TOC</u>	<u>good</u>
	<u>1</u>	<u> </u>	<u> </u>	<u>Oil + Grease</u>	
	<u>1</u>	<u> </u>	<u> </u>	<u>Phenolics</u>	
	<u>1</u>	<u> </u>	<u> </u>	<u>Cyanide</u>	
	<u>1</u>	<u> </u>	<u> </u>	<u>Metals (Cd Cr Pb Ni Ag)</u>	
	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>DMN</u>	

Remarks metals field filtered + acidified
 H-60



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca L. Jones

Phone: _____

SHIP TO:

ROY F. WESTON INC.

7720 LORRAINE AVE #105

STOCKTON, CA 95210

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION

Location Mather AFB

Shipper **ROY F. WESTON INC.**

Address **7720 LORRAINE AVE #105**

STOCKTON, CA 95210

Date Shipped 27. JUN 85

Shipment Service hand carried

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <u>Rebecca L. Jones</u>	Received by: (Signature)	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>M. Maldonado</u>	Date/Time <u>6/28 7AM</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
	<u>2</u>	<u>FH-5</u>	<u>6-27-85</u>	<u>EPA 601</u>	<u>OK</u>
	<u>2</u>	<u>↓</u>	<u>"</u>	<u>EPA 602</u>	<u>↓</u>

Remarks:

H-61



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah L. Jones
Phone: _____
SHIP TO:
Roy F. Weston Inc.
Laboratory Lab
256 Walnut Pool Rd.
Lansdale PA 19353
ATTENTION: Julie Porta
Phone No 215-524-0180

SHIPPING INFORMATION

Location Melher AFB
Shipper ROY F. WESTON INC.
Address 7720 LORRAINE AVE. #105
STOCKTON, CA 95210
Date Shipped 1 July 85
Shipment Service Fed. Ex
Airbill No. _____
Cooler No. _____

Relinquished by: (Signature) <u>Deborah L. Jones</u>	Received by: (Signature)	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>[Signature]</u>	Date/Time <u>7/8/85 9:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
	1	FH-5	7/5/85	TDC	Good
	1	↓	"	Oil & Grease	L

Remarks

H-62



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah L. Jones
 Phone: _____
 SHIP TO:
Roy F. Weston, Inc
Lionville Lab
255 Watch Pool Rd.
Lionville PA 19353
 ATTENTION: Judy Porta
 Phone No. 215-524-0180

SHIPPING INFORMATION
 Location Walker AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE #105
STOCKTON, CA 95210
 Date Shipped 1 July 85
 Shipment Service Fed Ex
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Deborah L. Jones</u>	Received by: (Signature)	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>J. A. Dardano</u>	Date/Time <u>7/1/85 9:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
	<u>1</u>	<u>ACW</u>	<u>6-27-85</u>	<u>TOC</u>	<u>Good</u>
	<u>1</u>	<u>↓</u>	<u>"</u>	<u>Oil + Grease</u>	
	<u>1</u>	<u>ACW-1</u>	<u>6-27-85</u>	<u>TOC</u>	<u>I</u>
	<u>1</u>	<u>↓</u>	<u>"</u>	<u>Oil + Grease</u>	

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah L. Jones

SHIPPING INFORMATION

Phone: _____

Location Mather AFB

SHIP TO:

Shipper **ROY F. WESTON INC.**

ROY F. WESTON INC.

Address 7720 LORRAINE AVE. #105

7720 LORRAINE AVE #105

STOCKTON, CA 95210

STOCKTON, CA 95210

Date Shipped 27 June 85

Shipment Service hand carried

ATTENTION: _____

Airbill No. _____

Phone No. _____

Cooler No. _____

Relinquished by: (Signature) <u>Deborah L. Jones</u>	Received by: (Signature)	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>M. Maldonado</u>	Date/Time <u>6/28 7 AM</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	2	ACW	6-27-85	EPA 601	OK
	2	↓	↓	EPA 602	↓
	1	↓	↓	PCB	↓
	2	ACW-1	6-27-85	EPA 601	↓
	2	↓	↓	EPA 602	↓
	1	↓	↓	PCB	↓

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah S. Jones
 Phone: _____
 SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE #105
STOCKTON, CA 95210
 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION
 Location Mall. AFA
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE #105
STOCKTON, CA 95210
 Date Shipped 27 June 85
 Shipment Service Fed Ex
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Deborah S. Jones</u>	Received by: (Signature)	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>M. Maldonado</u>	Date/Time <u>4/28 7AM</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
	<u>2</u>	<u>MA-FB-1</u>	<u>6/27/85</u>	<u>EPA 601</u>	<u>OK</u>
	<u>2</u>	<u>↓</u>	<u>↓</u>	<u>EPA 602</u>	<u>↓</u>
	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>PCB</u>	<u>↓</u>

Remarks: _____
 H-65'



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Debrah L. Jones
 Phone: _____
 SHIP TO:
Roy F. Weston, Inc.
Lenoirville Lab
256 Welch Pool Rd
Lenoiville PA 19353
 ATTENTION: Field Point
 Phone No. 6215-524-0180

SHIPPING INFORMATION
 Location Mather AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE #105
STOCKTON, CA 95210
 Date Shipped 1-15-85
 Shipment Service Fed Ex
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Debrah L. Jones</u>	Received by: (Signature) _____	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>R. D. ...</u>	Date/Time <u>7/2/85 9:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	<u>1</u>	<u>MAFB-1</u>	<u>8/7/85</u>	<u>TOL</u>	<u>good</u>
	<u>↓</u>	<u>↓</u>	<u>"</u>	<u>Oil + Grease</u>	<u>L</u>

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah L. Jones

Phone: _____

SHIP TO:

ROY F. WESTON INC.
7720 LORRAINE AVE. #105
STOCKTON, CA 95210

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION

Location Walker AFB

Shipper ROY F. WESTON INC.

Address 7720 LORRAINE AVE. #105

STOCKTON, CA 95210

Date Shipped 27 June 85

Shipment Service hand carried

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <u>Deborah L. Jones</u>	Received by: (Signature)	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>M. Maldonado</u>	Date/Time <u>6/28 7AM</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
	<u>2</u>	<u>WAFB-2</u>	<u>6-27-85</u>	<u>EPA 601</u>	
	<u>2</u>	<u>↓</u>	<u>↓</u>	<u>EPA 602</u>	
	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>PCB</u>	

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca d. Jones
 Phone: _____

SHIPPING INFORMATION

Location McPherson AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE. #105
STOCKTON, CA 95210
 Date Shipped 1 July 85
 Shipment Service Fed Ex
 Airbill No. _____
 Cooler No. _____

SHIP TO: _____
Roy F. Weston, Inc.
Lionville Lab

256 Weigh Pool Rd.
Lionville, PA. 19353

ATTENTION: Judy Polta
 Phone No. 615-571-0180

Relinquished by: (Signature) <u>Rebecca d. Jones</u>	Received by: (Signature)	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>[Signature]</u>	Date/Time <u>[Signature]</u>

Analysis laboratory should complete "sample cond. upon receipt" section below. sign and return copy to Shipper

Sample Number	No. Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	<u>1</u>	<u>M&ER-2</u>	<u>5-27-85</u>	<u>TOC</u>	<u>Good</u>
	<u>↓</u>	<u>↓</u>	<u>"</u>	<u>Oil + Grease</u>	<u>L</u>

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah Jones
 Phone: _____
 SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE #105
STOCKTON, CA 95210
 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION
 Location MALDEN AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE #105
STOCKTON, CA 95210
 Date Shipped 27 June 85
 Shipment Service Hand carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Deborah Jones</u>	Received by: (Signature)	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>M. Maldonado</u>	Date/Time <u>6/28 7AM</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	<u>2</u>	<u>MAFB-3</u>	<u>6-27-85</u>	<u>EPA 601</u>	<u>OK</u>
	<u>2</u>	<u>↓</u>	<u>↓</u>	<u>EPA 602</u>	<u>↓</u>
	<u>21</u>	<u>↓</u>	<u>↓</u>	<u>PCB</u>	<u>↓</u>
	<u>2</u>	<u>MAFB-30</u>	<u>6-27-85</u>	<u>EPA 601</u>	<u>↓</u>
	<u>2</u>	<u>↓</u>	<u>↓</u>	<u>EPA 602</u>	<u>↓</u>
	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>PCB</u>	<u>↓</u>

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca R. Jones

Phone: _____

SHIP TO:

Roy E. Weston, Inc.

Linnville Lab

256 Welch Pool Rd.

Linnville, GA 19353

ATTENTION: John P. ...

Phone No. 215-574-0180

SHIPPING INFORMATION

Location Math... AFB

Shipper ROY E. WESTON INC.

Address 7720 LORRAINE AVE. #109

STOCKTON, CA 95210

Date Shipped 1 July 85

Shipment Service Fed Ex

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <u>Rebecca R. Jones</u>	Received by: (Signature) _____	Date/Time <u>6/27/85</u>
Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received for laboratory by: (Signature) <u>[Signature]</u>	Date/Time <u>7/28/85</u>

Analysis laboratory should complete "sample cond upon receipt" section below. sign and return copy to Shipper

Sample Number	No Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	↓	<u>MAFB-3</u>	<u>6-27-85</u>	<u>TOC</u>	<u>Good</u>
	↓	"	"	<u>Oil + Grease</u>	
	↓	<u>MAFB-30</u>	<u>6-27-85</u>	<u>TOC</u>	
	↓	"	"	<u>Oil + Grease</u>	

Remarks _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca S. Jones
 Phone: _____
 SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE #105
STOCKTON, CA 95210
 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION

Location Walker AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE #105
STOCKTON, CA 95210
 Date Shipped 27 JUN 85
 Shipment Service hand carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Rebecca S. Jones</u>	Received by: (Signature) <u>6/27/85</u>	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>M. Maldonado</u>	Date/Time <u>6/28 7AM</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	2	DGFR-4	6/27/85	EPA601	OK
	2	↓	↓	EPA602	↓
	3	↓	↓	Pest/Herb	↓

Remarks _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Nebram S. Jones
 Phone: _____
SHIP TO:
Roy F Weston, Inc.
Lionville Lab
256 Welsh Road
Lionville, PA. 19353
ATTENTION: Judy Porta
 Phone No. 315-524-0180

SHIPPING INFORMATION
 Location Walker AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE #108
STOCKTON, CA 95210
 Date Shipped 1-27-85
 Shipment Service Fed Ex
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Nebram S. Jones</u>	Received by: (Signature) _____	Date/Time <u>1/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for Laboratory by: (Signature) <u>[Signature]</u>	Date/Time <u>1/28/85</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	1	DAFB-4	1-27-85	TCC	good
	1	↓	↓	Oil + Grease	↓
	1	↓	↓	Metals (Cd Cr Pb Ni Ag)	↓
	1	↓	↓	DMN	↓

Remarks: Metals found filled + resampled



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Rebecca A. Jones
 Phone: _____
 SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE #105
STOCKTON, CA 95210
 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION
 Location Mather AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE #105
STOCKTON, CA 95210
 Date Shipped 27 June 85
 Shipment Service Wood carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Rebecca A. Jones</u>	Received by: (Signature)	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>M. Maldonado</u>	Date/Time <u>6/28 7AM</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
	<u>2</u>	<u>DAFB-5</u>	<u>---</u>	<u>EPA 601</u>	<u>OK</u>
	<u>2</u>	<u>↓</u>	<u>↓</u>	<u>EPA 602</u>	<u>↓</u>
	<u>3</u>	<u>↓</u>	<u>↓</u>	<u>Pest/Herb</u>	<u>↓</u>

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature)

Deborah A. Jones

Phone:

SHIP TO:

ROY F. Weston Inc

Linnville, Ore

256 Maple Pool Rd.

Linnville, OR 97353

ATTENTION:

Judy Porta

Phone No.

915-524-0180

SHIPPING INFORMATION

Location

Mather AFB

Shipper

ROY F. WESTON INC.

Address

7720 LORRAINE AVE. #105
STOCKTON, CA 95210

Date Shipped

1 July 85

Shipment Service

Fed. Ex.

Airbill No.

Cooler No.

Relinquished by: (Signature)

Deborah A. Jones

Received by: (Signature)

Date/Time

7/27/85

Relinquished by: (Signature)

Received by: (Signature)

Date/Time

Relinquished by: (Signature)

Received by: (Signature)

Date/Time

Relinquished by: (Signature)

Received for laboratory by: (Signature)

Date/Time

M. Jones

7/28/85

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	<u>1</u>	<u>MAFR-5</u>	<u>7/25</u>	<u>TOC</u>	<u>good</u>
	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>Oil + Grease</u>	<u>I</u>
	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>Metals (Cd Cr Pb Ni Ag)</u>	<u>I</u>
	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>DMN</u>	<u>I</u>

Remarks:

Metals field filtered & acidified

CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) *Debrah Jones*
 Phone: _____
 SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE #105
STOCKTON, CA 95210
 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION
 Location Malheur AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE. #105
STOCKTON, CA 95210
 Date Shipped _____
 Shipment Service hand carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u><i>Debrah Jones</i></u>	Received by: (Signature)	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u><i>M. Maldonado</i></u>	Date/Time <u>6/28 7 AM</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
	2	MAFB-6	6-27-85	EPA 601	OK
	2	↓	↓	EPA 602	↓
	3	↓	↓	Pest/Herb	↓
	2	MAFB-60	6-27-85	EPA 601	
	2	↓	↓	EPA 602	↓
	3	↓	↓	Pest/Herb	

Remarks: _____

H-75



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Reborah Jones
 Phone: _____
 SHIP TO:
Roy F. Weston, Inc.
Livermore Lab
356 Walsh Pool Rd.
Livermore, PA. 19353
 ATTENTION: Judy Porta
 Phone No. 215-524-0180

SHIPPING INFORMATION
 Location Walsh AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LIVERMORE AVE #103
STOCKTON, CA 95210
 Date Shipped 1 July 85
 Shipment Service Fed. Ex.
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Reborah Jones</u>	Received by: (Signature)	Date/Time <u>6/27/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for Laboratory by: (Signature) <u>PA Darius</u>	Date/Time <u>7/28/85 9:20</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	1	MAFB-6	6-27-85	TOC	good
	1	↓	↓	Oil + Grease	
	1	↓	↓	Metals (Cd Cr Pb Ni Ag)	
	1	↓	↓	DMN	
	1	MAFB-60	6-27-85	TOC	
	1	↓	↓	Oil + Grease	
	1	↓	↓	Metals (Cd Cr Pb Ni Ag)	
	1	↓	↓	DMN	

Remarks: Metals field filtered + acidified



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Abraham Jones
 Phone: _____
 SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE. #105
STOCKTON, CA 95210
 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION
 Location Malheur AFB
 Shipper **ROY F. WESTON INC.**
 Address **7720 LORRAINE AVE. #105**
STOCKTON, CA 95210
 Date Shipped 26 June 85
 Shipment Service hand carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Abraham Jones</u>	Received by: (Signature)	Date/Time <u>6/26/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Richard Schultz</u>	Date/Time <u>6/27/85</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
	<u>2</u>	<u>MAFB-7</u>	<u>6-26-85</u>	<u>EPA 601 (VDAS)</u>	<u>OK</u>
	<u>2</u>	<u>↓</u>	<u> </u>	<u>EPA 602 (VDAS)</u>	<u>↓</u>

Remarks:
 H-77



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah Jones

Phone: _____

SHIP TO:

Roy F. Weston, Inc.

Linnville Lab

256 Walsh Pool Rd.

Linnville, GA 19353

ATTENTION: Judy Porta

Phone No. 915-524-0180

SHIPPING INFORMATION

Location Mather AFB
ROY F. WESTON INC.

Shipper 7720 LORRAINE AVE #105

Address STOCKTON, CA 95210

Date Shipped 27 June 85

Shipment Service Fo. Ex

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature)	Received by: (Signature)	Date/Time
<u>Deborah Jones</u>		<u>6/26/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature)	Date/Time
	<u>Victoria D. Darden</u>	<u>6/26/85 9:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	<u>1</u>	<u>MAFB-7</u>	<u>6-26-85</u>	<u>TOC</u>	<u>good</u>
	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>Oil + Grease</u>	
	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>Phenolics</u>	
	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>Cyanide</u>	
	<u>1</u>	<u>↓</u>	<u>↓</u>	<u>Metals (Cd Cr Pb Ni Ag)</u>	
	<u>1</u>	<u>MAFB-70</u>	<u>6-26-85</u>	<u>Phenolics</u>	<u>↓</u>
	<u>1</u>	<u>↓</u>	<u>"</u>	<u>Cyanide</u>	

Remarks: Metals field filtered + acidified



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah d. Jones

Phone: _____

SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE. #105
STOCKTON, CA 95210

ATTENTION: _____

Phone No. _____

SHIPPING INFORMATION

Location McKees AFB

Shipper **ROY F. WESTON INC.**

Address 7720 LORRAINE AVE #105
STOCKTON, CA 95210

Date Shipped 26 Jun 85

Shipment Service hand carried

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <u>Deborah d. Jones</u>	Received by: (Signature)	Date/Time <u>6/26/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>KATHY N. SCHULTZ</u>	Date/Time <u>6/27/85</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	<u>2</u>	<u>MAFB-S</u>	<u>6-20-85</u>	<u>EPA 601</u>	<u>OK</u>
	<u>2</u>	<u>↓</u>	<u>"</u>	<u>EPA 602</u>	<u>↓</u>

Remarks: _____



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Richard Jones
 Phone: _____
 SHIP TO:
Roy F. Weston, Inc.
Linnville Lab
256 North P.O. Rd.
Linnville, GA. 19353
 ATTENTION: Julia Porta
 Phone No. 715-524-0180

SHIPPING INFORMATION
 Location Matheson AFB
 Shipper ROY E. WESTON INC.
 Address 7720 LORRAINE AVE #105
STOCKTON, CA 95210
 Date Shipped 6/26/85
 Shipment Service Fed. Ex
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Richard Jones</u>	Received by: (Signature)	Date/Time <u>6/26/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Julia Porta</u>	Date/Time <u>6/26/85 9:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
	1	MAFB-8		TCC	Good
	1	↓	↓	Oil + Grease	
	1	↓	↓	Phenolics	
	1	↓	↓	Cyanide	
	1	↓	↓	Metals (Cd Cr Pb Ni Ag)	
	1	MAFB-80	6-26-85	Phenolics	Good
	1	↓	"	Cyanide	

Remarks: _____
 H-80



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Richard Jones

SHIPPING INFORMATION

Phone: _____

Location Malibu AFB

SHIP TO: _____

Shipper ROY F. WESTON INC.

ROY F. WESTON INC.

Address 7720 LORRAINE AVE #105

7720 LORRAINE AVE #105

STOCKTON, CA 95210

STOCKTON, CA 95210

Date Shipped 4/26/85

Shipment Service hard received

ATTENTION: _____

Airbill No. _____

Phone No. _____

Cooler No. _____

Relinquished by: (Signature) <u>Richard L. Jones</u>	Received by: (Signature)	Date/Time <u>4/26/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Kathleen Schilling</u>	Date/Time <u>4/27/85 10:00</u>

Analysis laboratory should complete "sample cond. upon receipt" section below. sign and return copy to Shipper

Sample Number	No. Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond. Upon Receipt
	<u>2</u>	<u>MAFB-9</u>	<u>6-26-85</u>	<u>EPA 601</u>	<u>OK</u>
	<u>2</u>	<u>↓</u>	<u>6-26-85</u>	<u>EPA 602</u>	
	<u>2</u>	<u>MAFB-90</u>	<u>6-26-85</u>	<u>EPA 601</u>	<u>↓</u>
	<u>2</u>	<u>↓</u>	<u>6-26-85</u>	<u>EPA 602</u>	

Remarks: H-51



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Deborah Jones
Phone: _____
SHIP TO:
Roy F. Weston, Inc.
Lancaster Lab
256 Water Pool Rd.
Lancaster PA 19353
ATTENTION: Ludwig Porta
Phone No. 215-524-0180

SHIPPING INFORMATION

Location: M.H. - AER
Shipper: ROY F. WESTON INC.
Address: 7720 LORRAINE AVE #105
STOCKTON, CA 95210
Date Shipped: 27 JUNE 85
Shipment Service: Fed Ex.
Airbill No.: _____
Cooler No.: _____

Relinquished by: (Signature) <u>Deborah Jones</u>	Received by: (Signature) _____	Date/Time <u>6/26/85</u>
Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received by: (Signature) _____	Date/Time _____
Relinquished by: (Signature) _____	Received for laboratory by: (Signature) <u>Justin B. ...</u>	Date/Time <u>6/28/85 9:30</u>

Analysis laboratory should complete "sample cond upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	1	<u>MAFB-89</u>	<u>6-26-85</u>	<u>TOC</u>	<u>Good</u>
	1	↓	↓	<u>Oil + Grease</u>	
	1	↓	↓	<u>Phenolics</u>	
	1	↓	↓	<u>Cyanide</u>	
	1	↓	↓	<u>Metals (Cd Cr Pb Ni Ag)</u>	
	1	<u>MAFB-90</u>	<u>6-26-85</u>	<u>TOC</u>	<u>Good</u>
	1	↓	↓	<u>Oil + Grease</u>	
	1	↓	↓	<u>Phenolics</u>	
	1	↓	↓	<u>Cyanide</u>	
	1	↓	↓	<u>Metals (Cd Cr Pb Ni Ag)</u>	

Remarks: Metals field filtered + acidified



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Robert A. Jones
 Phone: _____
 SHIP TO:

ROY F. WESTON INC.
7720 LORRAINE AVE. #105
STOCKTON, CA 95210
 ATTENTION: _____
 Phone No. _____

SHIPPING INFORMATION
 Location Mather AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE. #105
STOCKTON, CA 95210
 Date Shipped 6/26/85
 Shipment Service hand carried
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Robert A. Jones</u>	Received by: (Signature)	Date/Time <u>6/26/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Myrleen Schuler</u>	Date/Time <u>6/27/85</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	2	MAFB-10		EPA 601	OK
	2	↓	"	EPA 602	
	2	MAFB-100	6-26-85	EPA 601	
	2	↓	"	EPA 602	✓

Remarks: _____

H-83



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Robert Jones
 Phone: _____
 SHIP TO:
R. F. Weston, Inc.
256 Walsh Pool Rd.
Lanville, PA. 19353
 ATTENTION: John Porta
 Phone No. 215-524-0180

SHIPPING INFORMATION
 Location Mather AFB
 Shipper ROY F. WESTON INC.
 Address 7720 LORRAINE AVE. #105
STOCKTON, CA 95210
 Date Shipped 27 June 85
 Shipment Service Fed Ex
 Airbill No. _____
 Cooler No. _____

Relinquished by: (Signature) <u>Robert D. Jones</u>	Received by: (Signature)	Date/Time <u>6/26/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Victoria P. Dondos</u>	Date/Time <u>6/25/85 9:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No Of Cont	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	1	MAFB-10	6-25-85	TOC	good
	1	↓	↓	Oil + Grease	
	1	↓	↓	Phenolics	
	1	↓	↓	Cyanide	
	1	↓	↓	Metals (Cd Cr Ni Pb Ag)	
	1	MAFB-100	6-26-85	TOC	
	1	↓	↓	Oil + Grease	
	1	↓	↓	Phenolics	
	1	↓	↓	Cyanide	
	1	↓	↓	Metals (Cd Cr Ni Pb Ag)	

Remarks: Metals field filtered + acidified



CHAIN OF CUSTODY RECORD

SAMPLERS..(Signature) Richard S. Jones
Phone _____
SHIP TO:
ROY F. WESTON INC.
7720 LORRAINE AVE. #105
STOCKTON, CA 95210
ATTENTION: _____
Phone No. _____

SHIPPING INFORMATION
Location Mather AFB
Shipper **ROY F. WESTON INC.**
Address **7720 LORRAINE AVE. #105**
STOCKTON, CA 95210
Date Shipped 6/26/85
Shipment Service Insurance
Airbill No. _____
Cooler No. _____

Relinquished by: (Signature) <u>Richard S. Jones</u>	Received by: (Signature)	Date/Time <u>6/26/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Richard Schuber</u>	Date/Time <u>6/27/85</u>

Analysis laboratory should complete "sample cond. upon receipt" section below. sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	2	WESTON ↓	6/26/85 "	EPA 601 EPA 602	OK ↓

Remarks. H-85



CHAIN OF CUSTODY RECORD

SAMPLERS: (Signature) Debra A. Jones

Phone: IT

SHIP TO:

Roy F. Weston, Inc.

Lodi, CA

256 West 1st St.

Lodi, CA 95253

ATTENTION: Debra Jones

Phone No 215-521-0180

SHIPPING INFORMATION

Location Mollo AFB

Shipper ROY F. WESTON INC.

Address 7720 LORRAINE AVE. #105

STOCKTON, CA 95210

Date Shipped 27 June 85

Shipment Service Fed Ex

Airbill No. _____

Cooler No. _____

Relinquished by: (Signature) <u>Debra A. Jones</u>	Received by: (Signature)	Date/Time <u>6/26/85</u>
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received by: (Signature)	Date/Time
Relinquished by: (Signature)	Received for laboratory by: (Signature) <u>Victor R. Jones</u>	Date/Time <u>6/26/85 9:30</u>

Analysis laboratory should complete "sample cond. upon receipt" section below, sign and return copy to Shipper

Sample Number	No. Of Cont.	Site Identification	Date Sampled	Analysis Requested	Sample Cond Upon Receipt
	1	MAFB-11	26-85	TCC	Good
	1	↓	↓	Oil & Grease	
	1	↓	↓	Phenolics	
	1	↓	↓	Cyanide	
	1	↓	↓	Metals (Cd, Cr, Ni, Pb, Ag)	
	1	MAFB-110	26-85	Phenolics	
	1	↓	"	Cyanide	

Remarks

APPENDIX I
ANALYTICAL METHODS AND REQUIRED DETECTION LIMITS

5586A

APPENDIX I

ANALYTICAL METHODS AND REQUIRED DETECTION LIMITS

Analyte	Water Samples		Soil Samples	
	Detection Limit	Method	Detection Limit	Method
Volatile organic compounds	Specified by compound in method	EPA Methods 601 and 602	Specified by compound in method	EPA Methods 8010 and 8020
Total organic carbon (TOC)	1 mg/L	EPA Method 415.1	---	---
Oils and grease	0.1 mg/L	EPA Method 413.2	100 ug/g	Standard Method 503D
Phenol (total)	1 ug/L	EPA Method 420.1	1 ug/g	EPA Method 420.1
Cadmium (Cd)	10 ug/L	EPA Method 213.2	1 ug/g	ICP Optical
Lead (Pb)	20 ug/L	EPA Method 239.2	2 ug/g	ICP Optical
Chromium (Cr)	50 ug/L	ICP Optical	5 ug/g	ICP Optical
Nickel (Ni)	100 ug/L	ICP Optical	10 ug/g	ICP Optical
Silver (Ag)	10 ug/L	EPA Method 272.2	1 ug/g	ICP Optical
Cyanide	10 ug/L	EPA Method 335.2	1 ug/g	EPA Method 9010
PCB	0.25 ug/L	EPA Method 608	---	---
DMN	1 ug/L	EPA Method 625	---	---
DDT	0.02 ug/L	EPA Method 608	---	---
Chlordane	0.02 ug/L	EPA Method 608	---	---
2,4-D	0.06 ug/L	EPA Method 608	---	---

APPENDIX J

LABORATORY QA/QC PLAN

APPENDIX J

LABORATORY QA/QC PLAN

J.1 QUALITY ASSURANCE PLAN

WESTON Analytical Services enforces a rigid QA/QC program toward maintenance of validity and reliability of all analytical data. The Laboratory QA/QC Manual (Table of Contents thereof is Attachment No. 1 to this appendix) outlines the specifics of the QA/QC plan. This plan is patterned after the EPA Handbook for Analytical Quality Control in Waste and Wastewater Laboratories (EPA-600/4-79-019, March 1979), augmented by general applicable experience and interaction with the QA/QC plan of the U.S. Army Toxic and Hazardous Materials Agency (USATHAMA). All methods and procedures followed by WESTON are either USEPA or ASTM-approved. Any variations from such procedures, regardless of cause, are documented by the responsible analyst(s) and are documentable, and, literature-traceable. A general review of this QA/QC plan is in the following paragraphs.

Although specific QA/QC measures for each method are designated in WESTON's Laboratory Quality Assurance Manual, the general QA/QC program normally includes:

- EPA-acceptable sample preparation and analytical methods.
- Instrument calibration via use of Standard Analytical Reference Materials (SARMS).
- Regular equipment maintenance and servicing.
- Use of SARMS and QA/QC samples (spikes, laboratory blanks, replicates, and splits) to ascertain overall precision.
- Statistical evaluation of data to delineate acceptable limits.
- Documentation of system/operator performance.
- Suitable chain-of-custody procedures.
- Maintenance and archiving of all records, charts, and logs generated in the above.
- Proper reporting.

Acceptable analyses at WESTON's Analytical Laboratory Services include, but are not limited to, the above.

In general, WESTON's QA/QC sequence follows the following diagram (Figure J-1). Documentation (as available from instrument recordings and technicians' notebooks) is sufficient to validate each step in the sequence.

J.2 CONTAINER PREPARATION

Another consideration in this, or any, analytical project is that of sample container preparation. Accordingly, all appropriate sample bottles shall be cleaned in a manner mandated by the U.S. EPA to insure maximal cleanliness (and minimal contamination) before the containers go to the field. Sufficient bottles to accommodate both laboratory and field blank requirements will be preferred in a single batch mode for each monthly sampling requirement.

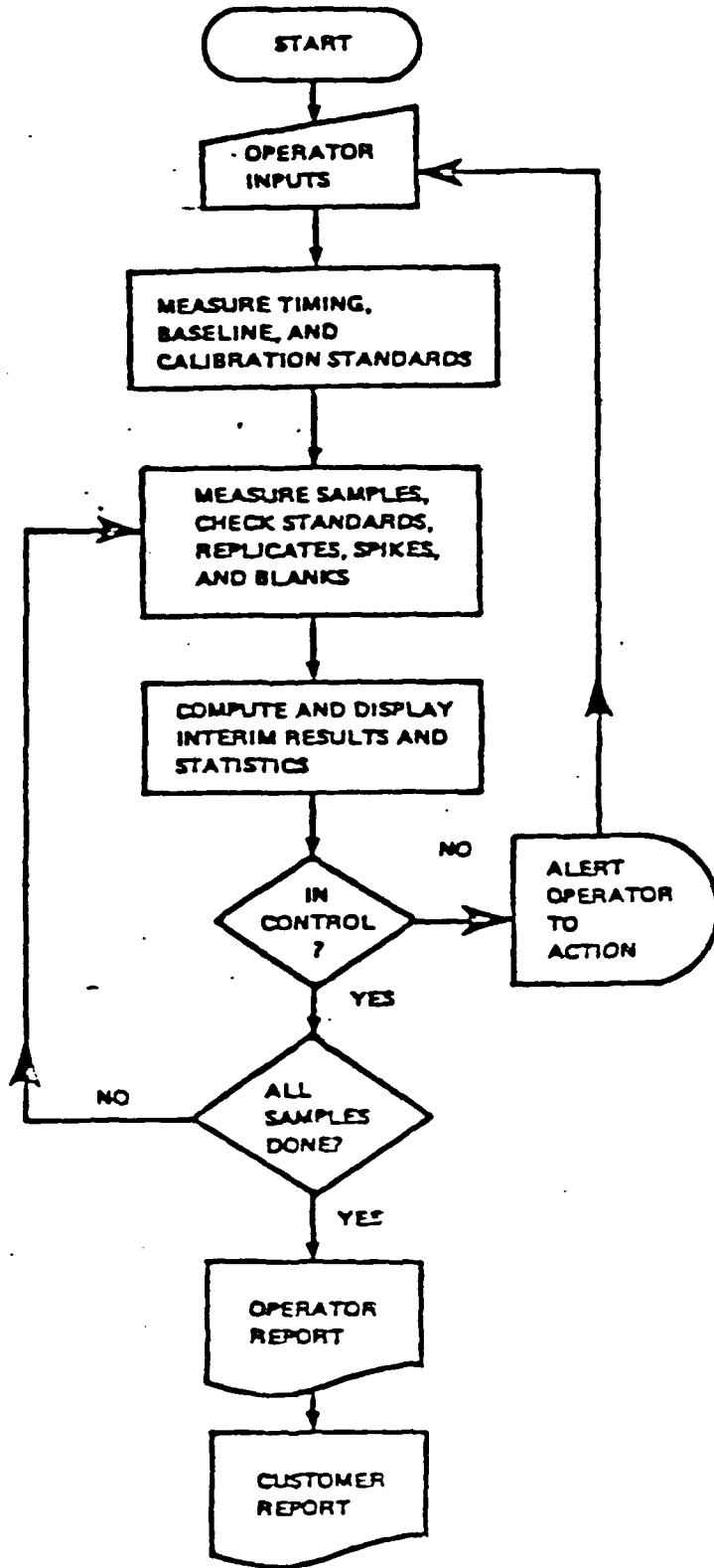


Figure J-1: Flow Chart of the Sequence of Events during a Controlled Series of Laboratory Measurements.



J.3 VERIFICATION/VALIDATION

In the laboratory, the analytical scheme begins with initial verification, which is comprised of:

- Lab Blanks - To insure that no background level of specific analytes is introduced by laboratory procedures.
- Standard Analytical Reference Materials (SARMS) - To determine the accuracy and precision of procedures.
- Spikes - To determine the percent recovery of analyte(s).

If the laboratory QA/QC program is extended to the field, it includes a fifth item:

- Field Blanks - To provide a check on contamination of containers and/or preservatives and to establish "practical" detection limits.

WESTON has used all of the above in this project. All data resulting from these verification media have been archived for future reference, retrieval, or processing. (QA/QC data from WESTON's above-described, internal QA/QC plan normally are not available to clientele without associated reimbursement to WESTON).

J.4 DATA HANDLING - LABORATORY

Use of any analytical data should be preceded by an assessment of its quality. The assessment should be based on accuracy, precision, completeness, representativeness, and comparability. These criteria are, in turn, assessed as follows:

- Accuracy - Is it acceptable for the planned use? QA/QC shall measure the accuracy of all data.

- Precision - Is it acceptable for the planned use? QA/QC shall reflect the reproducibility of the measurements.
- Completeness - Are the data sufficient for the planned use? QA/QC shall identify the quantity of data needed to match the goals.
- Representativeness - Do the data accurately reflect actual site conditions, sampling procedures, and analytical method? QA/QC shall ensure this.
- Comparability - Is the report self-consistent in format, units, and standardization of methods used to generate it? QA/QC shall ensure this.

Additionally, statistical methods outlined in the QA/QC program have been applicable to data evaluation.

The Laboratory Supervisor and the Laboratory QA/QC Officer have been responsible for the evaluation of the above criteria and for enforcement of analytical protocols that will necessarily lead to acceptable data quality. The signature of the Supervisor and QA/QC Officer accompany each laboratory analytical report and serve to ensure the overall validity of the reported data.

J.5 SAMPLE PLAN/LOG

Normal protocol demands client-and /or site-specific logging of all sample batches delivered to WESTON. Basic information -- such as client name, address, etc.; client phone number; reporting/invoicing instructions; site descriptions; and parameter-specifications and total requirements -- is initiated here. Additionally, sample storage/disposal instructions as well as turnaround requirements and sample collection requirements are addressed at this point.

The appropriate number of method blanks is also logged at this point, and in-house chain-of-custody documentation is initiated here.

J.6 SAMPLE RESULTS

WESTON's analytical protocols generally require five-point calibration curve plus a reagent blank s the basis for

quantification analytes from a linear calibration curve. (A three-point plus blank curve vs. the original five point one is acceptable if it falls within the QA/QC requirements of ± 3 standard deviation of the original curve.) Linear regression analysis is then performed. Method- and detection limit-specific data are accessed for quantitation and report-writing from each such data set. For reporting accuracy, the algorithm

$$\frac{\text{Linear-Regressed Raw Concentration from Calibration Curve} \times \text{Solid Sample Mass If Solid}}{\text{Solid Sample Extract Volume If Solid} \times \text{Concentration or Dilution Factor}} = \text{Final Concentration}$$

is used for all quantitations. (All such algorithm input data are archived for long-term storage.) Detection limits for solids are generated on a per-sample basis and calculated by replacing "LINEAR-REGRESSED RAW CONCENTRATION FROM CALIBRATION CURVE" with "DETECTION LIMIT OF ANALYTE IN LIQUID MATRIX" in the above equation.

J.7 CHAIN-OF-CUSTODY

Since they document the history of samples, chain-of-custody procedures are a crucial part of a sampling/analysis program. Chain-of-custody documentation enables identification and tracking of a sample from collection to analysis to reporting.

WESTON's chain-of-custody program necessitates the use of EPA-approved sample labels, secure custody, and attendant recordkeeping. Depending on the client's requirements, WESTON also offers container sealing during unattended transportation of samples.

In essence, WESTON considers a sample in custody if it: is in a WESTON employee's physical possession; it is in view of that WESTON employee; is secured by that WESTON employee to prevent tampering; or is secured by that WESTON employee in an area that is restricted to authorized personnel.

Each time a sample is relinquished from one analyst to another or from one major location to another, WESTON's analytical personnel are required to make appropriate entries. Personnel-specific initials are used as identifiers of analysts, as are location codes for various locations (refrigerators, extraction areas, analytical areas, etc.)



within the laboratory. Each transaction for each sample is accompanied by a specific reason for transfer. Chain-of-custody documentation is given in Appendix F.

J.8 QA/QC OFFICER

Toward maintenance of a rigid, credible QA/QC regimen, WESTON Analytical Services maintains a full-time, in-house QA/QC officer who retains independent authority to declare out-of-control situations, thereby precluding reporting of unacceptable data. The QA/QC officer has been available, as needed, on the project.



ATTACHMENT 1
LABORATORY QUALITY ASSURANCE MANUAL
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APPENDIX K
LABORATORY ANALYTICAL REPORTS

5586A

ROUND 85-1



inter-office memorandum

TO: [REDACTED]
Allison Dunn (Concord)
Kass Sheedy
Rich Johnson
cc: Les Eng (Memo Only)
Carter Nulton (Memo Only)

DATE: July 16, 1985

FROM: Judy Porta *jd*

RECEIVED

JUL 16 1985

GEOSCIENCES DEPT

SUBJECT: MATHER A.F.B. REPORT

W. O. No.: 0628-05-26

Enclosed are the reports of analysis for samples submitted May 31 to June 12, 1985 with the exception of results for soil samples US and USD which will follow tomorrow.

If you have any questions, please don't hesitate to call.

JP/eb

DATE OF REPORT: JULY 11, 1985

MATHER A.F.B.
WATER SAMPLES
INORGANIC SUMMARY REPORT
FOR
SAMPLES REC'D MAY 31 to JUNE 5, 1985
W.O. NO. 0628-05-26

I. TOTAL CYANIDE (CN⁻) ANALYSIS

a)

R.F.W. NO.	SAMPLE DESCRIPTION	DATE COLLECTED	DATE RECEIVED	DATE ANALYZED	TOTAL CYANIDE, mg/L
8506-579-0020	MAFB 7	5-30-85	6-4-85	6-12-85	<0.01
8505-564-0020	MAFB 8	5-29-85	5-31-85	6-3-85	<0.01
-0030	MAFB 9	5-29-85	5-31-85	6-3-85	<0.01
8506-579-0160	MAFB 10	5-30-85	6-4-85	6-12-85	<0.01
-0180	MAFB 11	5-30-85	6-4-85	6-12-85	<0.01
-0030	MAFB 70	5-30-85	6-4-85	6-12-85	<0.01
8505-564-0010	MAFB 80	5-29-85	5-31-85	6-3-85	<0.01
-0040	MAFB 90	5-29-85	5-31-85	6-12-85	<0.01
8506-579-0170	MAFB 100	5-30-85	6-4-85	6-12-85	<0.01
-0190	MAFB 110	5-30-85	6-4-85	6-12-85	<0.01
8506-579-0010	FB-2	5-30-85	6-4-85	6-12-85	<0.01
8506-583-0070	JTC	6-3-85	6-5-85	6-14-85	<0.01
-0080	JTC-1	6-3-85	6-5-85	6-14-85	<0.01

TOTAL CN⁻ ANALYSIS WAS NOT REQUESTED FOR SAMPLES IDENTIFIED AS MAFB 1-6, MAFB 20, MAFB 50, B1 to B4, B40, FB 1, FH 1-3, FH-6, FH 30, GC-1 to GC-2, GC-20 and K-9.

b) All samples were analyzed using EPA METHOD 335.2 within the EPA recommended holding time of 14 days. The requested detection limit of 0.01 mg/L (10 µg/L) was achieved.

WESTON

DATE OF REPORT: JULY 11, 1985

MATHER A.F.B. (CON'T.) PG. 2

II. OIL AND GREASE (O/G) ANALYSIS
a)

R.F.W. NO.	SAMPLE DESCRIPTION	DATE COLLECTED	DATE RECEIVED	DATE ANALYZED	O/G, mg/L
8506-579-0140	MAFB-1	5-31-85	6-4-85	6-8-85	0.74
-0080	MAFB-2	5-31-85	6-4-85	6-8-85	0.52
-0100	MAFB-3	5-31-85	6-4-85	6-8-85	0.26
-0110	MAFB-4	5-31-85	6-4-85	6-8-85	0.34
-0120	MAFB-5	5-31-85	6-4-85	6-8-85	0.33
-0150	MAFB-6	5-30-85	6-4-85	6-8-85	0.27
-0020	MAFB-7	5-30-85	6-4-85	6-8-85	0.44
8505-564-0020	MAFB-8	5-29-85	5-31-85	6-8-85	0.52
-0030	MAFB-9	5-29-85	5-31-85	6-8-85	0.55
8506-579-0160	MAFB-10	5-30-85	6-4-85	6-8-85	0.26
-0180	MAFB-11	5-30-85	6-4-85	6-8-85	0.17
-0090	MAFB-20	5-31-85	6-4-85	6-8-85	0.52
-0130	MAFB-50	5-31-85	6-4-85	6-8-85	0.52
8505-564-0010	MAFB-30	5-29-85	5-31-85	6-8-85	0.56
8506-579-0190	MAFB-110	5-30-85	6-4-85	6-8-85	0.17
8506-583-0010	B-1	6-3-85	6-5-85	6-8-85	0.10
-0020	B-2	6-3-85	6-5-85	6-8-85	0.10
-0030	B-3	6-3-85	6-5-85	6-8-85	0.13
-0040	B-4	6-3-85	6-5-85	6-8-85	0.10
-0050	B-40	6-3-85	6-5-85	6-8-85	0.13
8506-579-0070	FB-1	5-31-85	6-5-85	6-8-85	0.72
-0010	FB-2	5-30-85	6-5-85	6-8-85	0.20
8506-583-0130	FH-1	6-3-85	6-5-85	6-8-85	0.21
-0140	FH-2	6-3-85	6-5-85	6-8-85	0.12
-0150	FH-3	6-3-85	6-5-85	6-8-85	0.19
-0120	FH-6	6-3-85	6-5-85	6-8-85	0.31
-0160	FH-30	6-3-85	6-5-85	6-8-85	0.22
-0110	GC-1	6-3-85	6-5-85	6-8-85	0.63
-0090	GC-2	6-3-85	6-5-85	6-8-85	0.22
8506-583-0100	GC-20	6-3-85	6-5-85	6-8-85	0.22
-0070	JTC	6-3-85	6-5-85	6-8-85	0.10
-0060	K-9	6-3-85	6-5-85	6-8-85	0.10

OIL & GREASE ANALYSIS was not requested for samples identified as MAFB 70, MAFB 90, MAFB 100 and JTC-1.

b) All samples were analyzed using EPA METHOD 413.2 within the EPA recommended holding time of 28 days. The requested detection limit of 100 ug/L (0.100 mg/L) was achieved.

00000000

DATE OF REPORT: JULY 11, 1985

MATHER A.F.B. (CON'T.) PG. 3

III. TOTAL PHENOLICS ANALYSIS

a)

R.F.W. NO.	SAMPLE DESCRIPTION	DATE COLLECTED	DATE RECEIVED	DATE ANALYZED	TOTAL PHENOLICS mg/L
8506-579-0020	MAFB-7	5-30-85	6-4-85	6-10-85	0.006
8505-564-0020	MAFB-8	5-29-85	5-31-85	6-10-85	0.007
-0030	MAFB-9	5-29-85	5-31-85	6-10-85	0.005
-0030 DUP	MAFB-9 (LAB DUPLICATE)	5-29-85	5-31-85	6-10-85	0.005
8506-579-0160	MAFB-10	5-30-85	6-4-85	6-10-85	0.013
-0180	MAFB-11	5-30-85	6-4-85	6-10-85	0.006
-0030	MAFB-70	5-30-85	6-4-85	6-10-85	0.005
-0030 DUP	MAFB-70 (LAB DUPLICATE)	5-30-85	6-4-85	6-10-85	0.005
8505-564-0010	MAFB-80	5-29-85	5-31-85	6-10-85	0.005
8506-579-0190	MAFB-110	5-30-85	6-4-85	6-10-85	0.005
8506-579-0010	FB-2	5-30-85	6-5-85	6-10-85	0.006
8506-583-0070	JTC	6-3-85	6-5-85	6-10-85	0.005
-0080	JTC-1	6-3-85	6-5-85	6-10-85	0.005
-0080 DUP	JTC-1 (LAB DUPLICATE)	6-3-85	6-5-85	6-10-85	0.006
8505-564-0000	LAB BLANK	DNA	DNA	6-10-85	0.005
8505-564-SPIKE	BLANK SPIKE	DNA	DNA	6-10-85	84% RECOVERY
8506-579-0000	LAB BLANK	DNA	DNA	6-10-85	0.005
8506-579-SPIKE	BLANK SPIKE	DNA	DNA	6-10-85	76% RECOVERY
8506-583-0000	LAB BLANK	DNA	DNA	6-10-85	0.005
8506-583-SPIKE	BLANK SPIKE	DNA	DNA	6-10-85	79% RECOVERY

NOTE: No other samples required PHENOLICS analysis

b) All samples were analyzed using EPA METHOD 420.1 within the EPA recommended holding time of 28 days. As per the memo of March 5, 1985, this method is sensitive to 5 µg/L; therefore the requested detection limit of 1 µg/L was not achieved.

WESTON

Date of Revised Report: August 2, 1985
Date of Original Report: July 11, 1985

MATHER A.F.B.
ADD'N TOTAL PHENOLICS RESULTS
FOR
SAMPLES REC'D MAY 31 to JUNE 5, 1985
W.O. NO. 0628-05-26

R.F.W. NO.	SAMPLE DESCRIPTION	DATE COLLECTED	DATE RECEIVED	DATE ANALYZED	TOTAL PHENOLICS mg/L
8505-564-0040	MAFB-90	5-29-85	5-31-85	6-10-85	0.008
8506-579-0170	MAFB-100	5-30-85	6-4-85	6-10-85	0.005
E.P.A. METHOD:					420.1

COMPILED BY: Judith A. Porta
Judith A. Porta
Laboratory Operations Manager
WESTON Analytical Laboratories

APPROVED BY: Earl M. Hansen
Earl M. Hansen, Ph.D.
Manager
WESTON Analytical Laboratories

IV. TOTAL ORGANIC CARBON (TOC) ANALYSIS

a)

R.F.W. NO.	SAMPLE DESCRIPTION	DATE COLLECTED	DATE RECEIVED	DATE ANALYZED	TOC,mg/L
8506-579-0140	MAFB-1	5-31-85	6-4-85	6-11-85	<0.5
-0080	MAFB-2	5-31-85	6-4-85	6-11-85	2.4
-0100	MAFB-3	5-31-85	6-4-85	6-11-85	<0.5
-0110	MAFB-4	5-31-85	6-4-85	6-11-85	<0.5
-0120	MAFB-5	5-31-85	6-4-85	6-11-85	<0.5
-0150	MAFB-6	5-30-85	6-4-85	6-11-85	<0.5
-0020	MAFB-7	5-30-85	6-4-85	6-11-85	9.0
8505-564-0020	MAFB-8	5-29-85	5-31-85	6-6-85	4.4
-0020 DUP	MAFB-8 (LAB DUPLICATE)	5-29-85	5-31-85	6-6-85	4.6
8505-564-0030	MAFB-9	5-29-85	5-31-85	6-6-85	5.7
8506-579-0160	MAFB-10	5-30-85	6-4-85	6-6-85	<0.5
-0180	MAFB-11	5-30-85	6-4-85	6-6-85	0.7
-0090	MAFB-20	5-31-85	6-4-85	6-11-85	<0.5
-0130	MAFB-50	5-31-85	6-4-85	6-11-85	<0.5
8505-564-0010	MAFB-80	5-29-85	5-31-85	6-6-85	4.5
8506-579-0190	MAFB-110	5-30-85	6-4-85	6-11-85	0.7
-0190 DUP	MAFB-110 (LAB DUPLICATE)	5-30-85	6-4-85	6-11-85	1.0
-0190 SPIKE	MAFB-110 (MATRIX SPIKE)	5-30-85	6-4-85	6-11-85	104% RECOVERY
8506-583-0010	B-1	6-3-85	6-5-85	6-11-85	<0.5
-0020	B-2	6-3-85	6-5-85	6-11-85	<0.5
-0030	B-3	6-3-85	6-5-85	6-11-85	<0.5
-0040	B-4	6-3-85	6-5-85	6-11-85	<0.5
-0050	B-40	6-3-85	6-5-85	6-11-85	<0.5
8506-579-0070	FB-1	5-31-85	6-5-85	6-6-85	0.6
-0010	FB-2	5-30-85	6-5-85	6-6-85	0.6
8506-583-0130	FH-1	6-3-85	6-5-85	6-11-85	<0.5
-0140	FH-2	6-3-85	6-5-85	6-11-85	<0.5
-0150	FH-3	6-3-85	6-5-85	6-11-85	<0.5
-0120	FH-6	6-3-85	6-5-85	6-11-85	<0.5
-0160	FH-30	6-3-85	6-5-85	6-11-85	<0.5
-0110	GC-1	6-3-85	6-5-85	6-11-85	16.1
-0090	GC-2	6-3-85	6-5-85	6-11-85	17.8
-0100	GC-20	6-3-85	6-5-85	6-11-85	17.8
-0070	JTC	6-3-85	6-5-85	6-11-85	13.0
-0060	K-9	6-3-85	6-5-85	6-11-85	<0.5
8506-579-0000	LAB BLANK	DNA	DNA	6-6-85	<0.5
8506-579-SPIKE	BLANK SPIKE	DNA	DNA	6-6-85	98% RECOVERY
8506-583-0000	LAB BLANK	DNA	DNA	6-11-85	<0.5
8506-583-SPIKE	BLANK SPIKE	DNA	DNA	6-11-85	98% RECOVERY

WESTON

DATE OF REPORT: JULY 11, 1985

MATHER A.F.B. (CON'T.) PG. 5

IV. TOTAL ORGANIC CARBON (CON'T.)

NOTE: TOC analysis was not requested for samples identified as MAFB 70, MAFB 90, MAFB 100, and JTC-1.

b) All samples were analyzed by EPA METHOD 415.2 using a DOHRMANN DC 80 Carbon Analyzer within the EPA recommended holding time of 28 days. A detection limit of 500 µg/L was achieved.

COMPILED BY: Judith A. Porta
Judith A. Porta
Laboratory Support Manager
WESTON Analytical Laboratories

Approved By: Earl M. Hansen, Ph.D.
Earl M. Hansen, Ph.D.
Manager
WESTON Analytical Laboratories



DATE OF REPORT: July 15, 1985

MATHER A.F.B.
TOTAL METALS SUMMARY REPORT
FOR
SAMPLES RECEIVED JUNE 12, 1985

W.O. NO. 0628-05-26

DATE SAMPLES COLLECTED: May 29-June 3, 1985

SAMPLES SUBMITTED BY: Nancy Schultz

I.

R.F.W. NO.	SAMPLE DESCRIPTION	TOTAL				
		Cd ug/L	Cr ug/L	Pb ug/L	Ni ug/L	Ag ug/L
8506-611-0110	MAFB 4	<2.5	<50	<10	<100	<2.5
-0120	MAFB 5	<2.5	<50	<10	<100	<2.5
-0060	MAFB 6	<2.5	<50	<10	<100	<2.5
-0060 SPIKE	MAFB 6 (MATRIX SPIKE)	88%	75%	88%	75%	104%
-0070	MAFB 7	<2.5	<50	<10	<100	<2.5
-0010	MAFB 8	<2.5	60	<10	150	<2.5
-0020	MAFB 9	<2.5	<50	<10	<100	<2.5
-0020 DUP	MAFB 9 (LAB DUPLICATE)	<2.5	<50	<10	<100	---
-0080	MAFB 10	<2.5	<50	<10	<100	<2.5
-0090	MAFB 11	<2.5	<50	<10	<100	<2.5
-0130	MAFB 50	<2.5	<50	<10	<100	<2.5
-0030	MAFB 80	<2.5	<50	<10	<100	<2.5
-0030 DUP	MAFB 80 (LAB DUPLICATE)	---	---	---	---	<2.5
-0100	MAFB 110	<2.5	<50	<10	<100	<2.5
-0100 DUP	MAFB 110 (LAB DUPLICATE)	<2.5	<50	<10	<100	<2.5
-0140	B-1	<2.5	<50	<10	<100	<2.5
-0150	B-2	<2.5	<50	<10	<100	<2.5
-0160	B-3	<2.5	<50	<10	<100	<2.5
-0160 SPIKE	B-3 (MATRIX SPIKE)	92%	92%	148%	92%	108%



MATHER A.F.B. (CON'T) PG. 2

DATE OF REPORT: July 15, 1985

R.F.W. NO.	SAMPLE DESCRIPTION	TOTAL				
		Cd ug/L	Cr ug/L	Pb ug/L	Ni ug/L	Ag ug/L
8506-611-0170	B-4	<2.5	<50	<10	<100	<2.5
-0180	B-40	<2.5	<50	<10	<100	<2.5
-0040	FB-2	<2.5	<50	<10	<100	<2.5
-0050	FB-3	<2.5	<50	<10	<100	<2.5
-0190	JTC	<2.5	<50	<10	<100	<2.5
8506-611-0000	LAB BLANK	<2.5	<50	<10	<100	<2.5
8506-611-SPIKE	BLANK SPIKE	100% RECOVERY	92% RECOVERY	100% RECOVERY	90% RECOVERY	104% RECOVERY

II. All samples were analyzed within the EPA recommended holding time of six months from date of collection to date of analysis. The method of analysis and the requested and achieved detection limits are as follows:

METAL	METHOD	REQUESTED DETECTION LIMIT	ACHIEVED DETECTION LIMIT
CADMIUM	EPA 213.2	10 ug/L	2.5 ug/L
CHROMIUM	EPA 218.1	50 ug/L	50 ug/L
LEAD	EPA 239.2	20 ug/L	10 ug/L
NICKEL	EPA 249.1	100 ug/L	100 ug/L
SILVER	EPA 272.2	10 ug/L	2.5 ug/L

COMPILED BY: Judith A. Porta
Judith A. Porta
Laboratory Operations Manager
WESTON Analytical Laboratories

APPROVED BY: Earl M. Hansen, Ph.D.
Earl M. Hansen, Ph.D.
Manager
WESTON Analytical Laboratories

MATHER A.F.B.
DMN SUMMARY REPORT
W.D. NO. 0628-05-26

DIMETHYLNITROSAMINE (DMN) ANALYSIS

R.F.W. NO.	SAMPLE DESCRIPTION	DATE COLLECTED	DATE REC'D	DATE EXTRACTED	DATE ANALYZED	DMN, $\mu\text{g/L}$
8506-579-0110	MAFB-4	5-31-85	6-4-85	6-5-85	6-7-85	<0.2
-0120	MAFB-5	5-31-85	6-4-85	6-5-85	6-7-85	<0.2
-0130	MAFB-50	5-31-85	6-4-85	6-5-85	6-7-85	<0.2
-0150	MAFB-6	5-30-85	6-4-85	6-5-85	6-7-85	<0.2
8506-583-0010	B-1	6-03-85	6-5-85	6-5-85	6-7-85	<0.2
-0020	B-2	5-30-85	6-5-85	6-5-85	6-7-85	<1.0
-0030	B-3	5-30-85	6-5-85	6-5-85	6-7-85	<0.2
-0040	B-4	6-03-85	6-5-85	6-5-85	6-7-85	<0.2
-0050	B-40	6-03-85	6-5-85	6-5-85	6-7-85	<0.2
8506-579&583/	LAB BLANK	-----	-----	6-5-85	6-7-86	<0.2
8506-579&583/	SPIKE D.I. SPIKE	-----	-----	6-5-85	6-7-85	46% RECOVERY
8506-579&583/	S.D. D.I. SPIKE DUP.	-----	-----	6-5-85	6-7-85	46% RECOVERY

These samples were analyzed using EPA Method 607. All samples were extracted within seven days of collection and were analyzed within two days of extraction. The requested detection limit of $1\mu\text{g/L}$ was achieved.

*method 607 is correct, not 605
as specified in a memo to
the lab. verified by Carter
Boston*

Approved By: 
Earl M. Hansen, Ph.D.
Manager
WESTON Analytical Laboratories



DATE OF REPORT: July 11, 1985

MATHER A.F.B.
SOIL SAMPLES
INORGANIC SUMMARY REPORT
FOR
SAMPLES REC'D JUNE 4, 1985

W.O. NO. 0628-05-26

DATE SAMPLES COLLECTED: 5-30-85

R.F.W. NO.	SAMPLE DESCRIPTION	TOTAL CN ⁻ , ug/g	OIL & GREASE ug/g	TOTAL PHENOLICS ug/g
8506-579-0040	USD	<0.13	2,140	0.357
-0050	US	<0.13	3,840	0.395
-0050 DUP	US (LAB DUPLICATE)	---	3,800	---
-0060	DS	<0.13	302	0.234
8506-579-0000	LAB BLANK	---	12.0	---

NOTE: THE OIL & GREASE RESULTS ARE NOT BLANK CORRECTED.

DATE OF ANALYSIS:	6-13-85	6-12-85	6-10-85
METHOD OF ANALYSIS:	EPA 335.2	EPA 413.2	EPA 420.1
DETECTION LIMIT ACHIEVED:	0.13 ug/g	8.0 ug/g	0.123 ug/g
REQUESTED DETECTION LIMITS:	1.0 ug/g	100 ug/g	1.0 ug/g

COMPILED BY: Judith A. Porta
Judith A. Porta
Laboratory Operations Manager
WESTON Analytical Laboratories

APPROVED BY: Earl M. Hansen
Earl M. Hansen, Ph.D.
Manager
WESTON Analytical Laboratories



inter-office memorandum

TO: Katherine Sheedy

DATE: June 27, 1985

~~Bobbie Jones~~

FROM: David Ben-Hur, Stockton Laboratory *DB*

RECEIVED
JUL 3 1985
GEOSCIENCES DEPT
W. O. No.:

SUBJECT: Analytical results, Mather AFB, Sampling
Round of May 1985

Attached are the results of the analyses performed at the Stockton Laboratory on the samples collected in the first round of resampling at Mather AFB.

MATHER AFB
Sampling and Analysis Chronology

Sample ID	Date Sampled	EPA 601	EPA 602	Pesticides		Herbicides		PCB's	
				Extracted	Analyzed	Extracted	Analyzed	Extracted	Analyzed
MAFB-8	5/29/85	6/12/85	6/4/85	-	-	-	-	-	-
MAFB-9	"	"	"	-	-	-	-	-	-
MAFB-80	"	"	"	-	-	-	-	-	-
MAFB-6	5/30/85	"	"	6/5/85	6/7/85	6/6/85	6/12/85	-	-
MAFB-7	"	"	"	-	-	-	-	-	-
MAFB-10	"	"	"	-	-	-	-	-	-
MAFB-11	"	"	"	-	-	-	-	-	-
FB-2	"	"	"	-	-	-	-	-	-
US (soil)	"	6/18/85	6/20/85	-	-	-	-	-	-
USD (soil)	"	"	"	-	-	-	-	-	-
DS (soil)	"	"	"	-	-	-	-	-	-
MAFB-1	5/31/85	6/13/85	6/4/85	-	-	-	-	6/7/85	6/11/85
MAFB-2	"	"	"	-	-	-	-	"	"
MAFB-3	"	"	"	-	-	-	-	"	"
MAFB-4	"	"	"	6/5/85	6/7/85	6/6/85	6/12/85	-	-
MAFB-5	"	"	"	"	"	"	"	-	-
MAFB-20	"	"	"	-	-	-	-	6/7/85	6/11/85
MAFB-50	"	"	"	6/5/85	6/7/85	6/6/85	6/12/85	-	-
MAFB-110	"	"	"	-	-	-	-	-	-
FB-1	"	"	"	-	-	-	-	6/7/85	6/11/85

MATHER AFB
 Sampling and Analysis Chronology

Sample ID	Date Sampled	EPA 601		LVA602		Date Analyzed					
		Extracted	Analyzed	Extracted	Analyzed	Pesticides Extracted	Pesticides Analyzed	Herbicides Extracted	Herbicides Analyzed	PCB's Extracted	PCB's Analyzed
B-1	6/3/85	6/17/85	"	6/6/85	"	6/5/85	6/7/85	6/6/85	6/12/85	-	-
B-2	"	"	"	"	"	"	"	"	"	-	-
B-3	"	"	"	"	"	"	"	"	"	-	-
B-4	"	"	"	"	"	6/7/85	"	"	"	-	-
B-40	"	6/13/85	"	"	"	"	"	"	"	-	-
FH-1	"	"	"	"	"	-	-	-	-	-	-
FH-2	"	"	"	"	"	-	-	-	-	-	-
FH-3	"	"	"	"	"	-	-	-	-	-	-
FH-6	"	"	"	"	"	-	-	-	-	-	-
FH-30	"	6/14/85	"	"	"	-	-	-	-	-	-
GC-1	"	"	"	"	"	-	-	-	-	-	-
GC-2	"	"	"	"	"	-	-	-	-	-	-
GC-20	"	"	"	"	"	-	-	-	-	-	-
JTC	"	"	6/7/85	"	"	-	-	-	-	-	-
K-9	"	"	"	"	"	-	-	-	-	-	-

MATHER AFB
QA/QC Data

1. Second column confirmation for volatile compounds

The following samples have been subjected to a second column confirmation. The confirmation was performed qualitatively only. Compounds that were identified and quantitated in the primary column, but could not be confirmed, were reported as ND - not detected.

Sample ID

MAFB-8
MAFB-9
MAFB-80
MAFB-1
MAFB-2
MAFB-3
MAFB-20
GC-2
GC-20

2. Laboratory duplicates for volatile compounds analysis

Method 601

Sample ID: MAFB-9

<u>Compound</u>	<u>Concentration, ug/L</u>	
	<u>First</u>	<u>Second</u>
Trichloroethene	4.8	7.0
Tetrachloroethene	1.3	2.4

Sample ID: GC-20

1,1,1-Trichloroethane	9.5	ND
-----------------------	-----	----

Method 602

<u>Compound</u>	MAFB-8		MAFB-9		MAFB-80	
	<u>Concentration, ug/L</u>	<u>Concentration, ug/L</u>	<u>Concentration, ug/L</u>	<u>Concentration, ug/L</u>	<u>Concentration, ug/L</u>	<u>Concentration, ug/L</u>
	<u>First</u>	<u>Second</u>	<u>First</u>	<u>Second</u>	<u>First</u>	<u>Second</u>
Chlorobenzene	1.7	0.94	ND	ND	1.7	0.
1,3,-Dichlorobenzene	ND	ND	ND	ND	ND	ND
1,2-Dichlorobenzene	1.2	1.0	ND	ND	1.2	1.
1,4-Dichlorobenzene	0.32	ND	ND	ND	0.61	0.
Benzene	ND	ND	ND	ND	ND	ND
Toluene	0.47	0.33	ND	ND	0.40	0.
Ethylbenzene	ND	ND	ND	ND	ND	ND

MATHER AFB
QA/QC Data

3. Matrix Spikes for volatile compounds

Compound	Spike, ug/L	Percent Recovery			
		MAFB-9	MAFB-11	FH-3	B-1
1,1-Dichloroethene	1.2	NS	NS	82	NS
1,1-Dichloroethane	1.2	NS	NS	68	NS
trans-1,2-dichloroethene	1.2	NS	NS	68	NS
Trichloroethene	1.2	NS	NS	90	104
Tetrachloroethene	1.2	NS	NS	75	83
Chlorobenzene	1.2	101	98	NS	NS
1,2-Dichlorobenzene	1.2	86	98	NS	NS
1,3-Dichlorobenzene	1.2	94	106	NS	NS
1,4-Dichlorobenzene	1.2	95	98	NS	NS

NS = Not spiked

4. Pesticide and herbicide matrix spike

Compound	Spike, ug/L	Percent Recovery		
		B-2	B-40	Water
o,p'-DDT	0.15	NS	120	100
Chlordane	0.14	NS	80	71
2,4-D	0.18	98	NS	93

LAB NO. 85-05-044

Pesticide and Herbicide Analysis
Mather AFB
Sample: MAFB-6

<u>Compound</u>	<u>Detection Limit, ug/L</u>	<u>Found ug/L</u>
o,p'-DDT	0.02	ND
p,p'-DDT	0.02	ND
Chlordane	0.02	ND
2,4-D	0.06	ND

LAB NO. 85-06-001

Pesticide and Herbicide Analysis
Mather AFB

Compound	Detection Limit, ug/L	Concentration, ug/L		
		MAFB-4	MAFB-5	MAFB-50
o,p'-DDT	0.02	ND	ND	ND
p,p'-DDT	0.02	ND	ND	ND
Chlordane	0.02	ND	ND	ND
2,4-D	0.06	ND	ND	ND

LAB NO. 85-06-001

PCB Analysis
Mather AFB

Parameter	Detection Limit, ug/L	Concentration, ug/L				
		MAFB-1	MAFB-2	MAFB-3	MAFB-20	FB-1
PCB 1016	0.04	ND	ND	ND	ND	ND
PCB 1221	0.10	ND	ND	ND	ND	ND
PCB 1232	0.10	ND	ND	ND	ND	ND
PCB 1242	0.05	ND	ND	ND	ND	ND
PCB 1248	0.08	ND	ND	ND	ND	ND
PCB 1254	0.05	ND	ND	ND	ND	ND
PCB 1260	0.15	ND	ND	ND	ND	ND

LAB NO. 85-06-004

Pesticide and Herbicide Analysis
Mather AFB

Compound	Detection Limit, ug/L	Concentration, ug/L				
		B-1	B-2	B-3	B-4	B-40
o,p'-DDT	0.02	ND	ND	ND	ND	ND
p,p'-DDT	0.02	ND	ND	ND	ND	ND
Chlordane	0.02	ND	ND	ND	ND	ND
2,4-D	0.06	ND	ND	ND	ND	ND



inter-office memorandum

TO: Katherine Sheedy
cc: Alison Dunn, Concord Office

DATE: November 4, 1985

FROM: David Ben-Hur *DB*

SUBJECT: Mather AFB Volatiles Analysis Results W. O. No.:

Attached are the corrected results for the water samples collected at Mather AFB during May and June 1985.

These data are blank corrected.

Mather AFB - May 1985 Sampling
 Revised Report
 Volatiles Analysis by EPA Methods 601 (GC/Hall Detector) and 602 (GC/PID)

Component	Detection Limit, ug/L	Concentration, ug/L			
		GC-1	GC-2	GC-20	JTC K-9
Chloroethane	1.0	ND	ND	ND	ND
Bromoethane	1.2	ND	ND	ND	ND
Dichlorodifluoromethane	1.8	ND	ND	ND	ND
Vinyl chloride	0.2	ND	ND	ND	ND
Chloroethane	0.5	ND	ND	ND	ND
Methylene chloride	0.2	ND	ND	ND	ND
Trichlorofluoromethane	2.0	ND	ND	ND	ND
1,1-Dichloroethane	0.2	ND	ND	ND	ND
1,2-Dichloroethane	0.1	ND	ND	ND	ND
Trans-1,2-dichloroethane	0.1	ND	ND	ND	ND
Chloroform	0.1	ND	ND	ND	ND
1,2-Dichloroethane	0.02	ND	ND	ND	ND
1,1,1-Trichloroethane	0.1	ND	6.2	9.5	ND
Carbon tetrachloride	0.1	ND	ND	ND	ND
Bromodichloromethane	0.1	ND	ND	ND	ND
1,2-Dichloropropane	0.1	ND	ND	ND	ND
Cis-1,3-dichloropropene	0.3	ND	ND	ND	ND
Trichloroethene	0.1	ND	ND	ND	ND
Dibromochloromethane	0.1	ND	ND	ND	ND
1,1,2-Trichloroethane	0.05	ND	ND	ND	ND
Trans-1,3-dichloropropene	0.2	ND	ND	ND	ND
1,1-Dichloroethyl vinyl ether	0.2	ND	ND	ND	ND
Bromoform	0.2	ND	ND	ND	ND
1,1,2,2-Tetrachloroethane	0.05	ND	ND	ND	ND
Tetrachloroethene	0.05	ND	ND	ND	ND
Chlorobenzene	0.3	ND	ND	ND	ND
1,3-Dichlorobenzene	0.3	ND	ND	ND	ND
1,2-Dichlorobenzene	0.2	ND	ND	ND	ND
1,4-Dichlorobenzene	0.2	ND	ND	ND	ND
Benzene	0.2	ND	ND	ND	ND
Toluene	0.2	0.31	ND	0.84	1.2
1,1,1-Trichloroethane	0.2	ND	ND	ND	0.67
ND	0.2	ND	ND	ND	ND

ND = Not Detected

Mather AFB - May 1985 Sampling
 Revised Report
 Volatiles Analysis by EPA Methods 601 (GC/Hall Detector) and 602 (GC/PID)

Component	Detection Limit, ug/L	Concentration, ug/L																		
		MAFB-1	MAFB-2	MAFB-3	MAFB-4	MAFB-5	MAFB-20	MAFB-50	MAFB-110	FB-1										
Chloromethane	1.0	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	
Bromomethane	1.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Dichlorodifluoromethane	1.8	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Vinyl chloride	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Chloroethane	0.5	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Methylene chloride	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Trichlorofluoromethane	2.0	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1-Dichloroethene	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1-Dichloroethane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Trans-1,2-dichloroethene	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Chloroform	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,2-Dichloroethane	0.02	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1,1-Trichloroethane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Carbon tetrachloride	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Bromodichloromethane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,2-Dichloropropane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Cis-1,3-dichloropropene	0.3	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Trichloroethene	0.1	7.7	13.	33.	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Dibromochloromethane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1,2-Trichloroethane	0.05	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Trans-1,3-dichloropropene	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
2-Chloroethylvinyl ether	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Bromoform	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1,2-Tetrachloroethane	0.05	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Tetrachloroethene	0.05	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Chlorobenzene	0.3	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,3-Dichlorobenzene	0.3	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,2-Dichlorobenzene	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,4-Dichlorobenzene	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Benzene	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Toluene	0.2	2.0	0.74	0.94	1.2	0.69	0.84	0.84	0.84	0.84	0.84	0.84	0.84	0.84	0.84	0.84	0.84	0.84	0.84	0.84
Ethylbenzene	0.2	ND	ND	ND	0.35	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND

ND - Not Detected

Mather AFB - May 1985 Sampling
 Revised Report
 Volatiles Analysis by EPA Methods 601 (GC/Hall Detector) and 602 (GC/PID)

Component	Detection Limit, ug/L	Concentration, ug/L				
		MAFB-6	MAFB-7	MAFB-10	MAFB-11	FB-2
Chloromethane	1.0	ND	ND	ND	ND	ND
Bromomethane	1.2	ND	ND	ND	ND	ND
Dichlorodifluoromethane	1.8	ND	ND	ND	ND	ND
Vinyl chloride	0.2	ND	ND	ND	ND	ND
Chloroethane	0.5	ND	ND	ND	ND	ND
Methylene chloride	0.2	ND	ND	ND	ND	ND
Trichlorofluoromethane	2.0	ND	ND	ND	ND	ND
1,1-Dichloroethene	0.2	ND	ND	ND	ND	ND
1,1-Dichloroethane	0.1	ND	ND	ND	ND	ND
Trans-1,2-dichloroethene	0.1	ND	ND	ND	ND	ND
Chloroform	0.1	ND	ND	ND	ND	ND
1,2-Dichloroethane	0.02	ND	ND	ND	ND	ND
1,1,1-Trichloroethane	0.1	ND	ND	ND	ND	ND
Carbon tetrachloride	0.1	ND	ND	ND	ND	ND
Bromodichloromethane	0.1	ND	ND	ND	ND	ND
1,2-Dichloropropane	0.1	ND	ND	ND	ND	ND
Cis-1,3-dichloropropene	0.3	ND	ND	ND	ND	ND
Trichloroethene	0.1	ND	ND	ND	ND	ND
Dibromochloromethane	0.1	ND	ND	ND	ND	ND
1,1,2-Trichloroethane	0.05	ND	ND	ND	ND	ND
Trans-1,3-dichloropropene	0.2	ND	ND	ND	ND	ND
2-Chloroethylvinyl ether	0.2	ND	ND	ND	ND	ND
Bromoform	0.2	ND	ND	ND	ND	ND
1,1,2,2-Tetrachloroethane	0.05	ND	ND	ND	ND	ND
Tetrachloroethene	0.05	ND	ND	ND	ND	ND
Chlorobenzene	0.3	ND	ND	ND	ND	ND
1,3-Dichlorobenzene	0.3	ND	ND	ND	ND	ND
1,4-Dichlorobenzene	0.2	ND	ND	ND	ND	ND
1,4-Dichlorobenzene	0.2	ND	ND	ND	ND	ND
Benzene	0.2	ND	ND	ND	ND	ND
Toluene	0.2	ND	ND	ND	ND	ND
Ethylbenzene	0.2	ND	ND	ND	ND	ND

ND = Not Detected

Mather AFB - May 1985 Sampling
 Revised Report
 Volatiles Analysis by EPA Methods 601 (GC/Hall Detector) and 602 (GC/PID)

Component	Detection Limit, ug/L	Concentration, ug/L		
		MAFB-8	MAFB-9	MAFB-80
Chloromethane	1.0	ND	ND	ND
Bromomethane	1.2	ND	ND	ND
Dichlorodifluoromethane	1.8	ND	ND	ND
Vinyl chloride	0.2	ND	ND	ND
Chloroethane	0.5	ND	ND	ND
Methylene chloride	0.2	ND	ND	ND
Trichlorofluoromethane	2.0	ND	ND	ND
1,1-Dichloroethene	0.2	ND	ND	ND
1,1-Dichloroethane	0.1	ND	ND	ND
trans-1,2-dichloroethene	0.1	ND	ND	ND
Chloroform	0.1	ND	ND	ND
1,2-Dichloroethane	0.02	ND	ND	ND
1,1,1-Trichloroethane	0.1	ND	ND	ND
Carbon tetrachloride	0.1	ND	ND	ND
Bromochloromethane	0.1	ND	ND	ND
1,2-Dichloropropane	0.1	ND	ND	ND
Cis-1,3-dichloropropene	0.3	ND	ND	ND
Trichloroethene	0.1	47.	4.8	34.
1,1,1,1-Tetrachloroethane	0.1	ND	ND	ND
1,1,2-Trichloroethane	0.05	ND	ND	ND
trans-1,3-dichloropropene	0.2	ND	ND	ND
2-chloroethylvinyl ether	0.2	ND	ND	ND
Bromoform	0.2	ND	ND	ND
1,1,2,2-Tetrachloroethane	0.05	ND	ND	ND
Tetrachloroethene	0.05	20.	1.3	13.
Chlorobenzene	0.3	1.7	ND	1.7
1,3-Dichlorobenzene	0.3	ND	ND	ND
1,2-Dichlorobenzene	0.2	1.2	ND	1.2
1,4-Dichlorobenzene	0.2	0.32	ND	0.61
Benzene	0.2	ND	ND	ND
Toluene	0.2	ND	ND	ND
1-methylbenzene	0.2	ND	ND	ND

ND = Not Detected

March 85 - May 1985 Sampling
 Revised Report
 Volatiles Analysis by EPA Methods 601 (GC/Hall Detector) and 602 (GC/PID)

Component	Detection Limit, ug/L	Concentration, ug/L													
		B-1	B-2	B-3	B-4	B-40	FH-1	FH-2	FH-3	FH-6	FH-30				
Chloromethane	1.0	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Bromomethane	1.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Dichlorodifluoromethane	1.8	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Vinyl chloride	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Chloroethane	0.5	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Methylene chloride	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Trichlorofluoromethane	2.0	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1-Dichloroethane	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1-Dichloroethane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Trans-1,2-dichloroethane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Chloroform	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,2-Dichloroethane	0.02	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1,1-Trichloroethane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Carbon tetrachloride	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Bromodichloromethane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,2-Dichloropropane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Cis-1,3-dichloropropene	0.3	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Trichloroethene	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Dibromochloromethane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1,2-Trichloroethane	0.05	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Trans-1,3-dichloropropene	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
2-Chloroethylvinyl ether	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Bromoform	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1,2,2-Tetrachloroethane	0.05	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Tetrachloroethene	0.05	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Chlorobenzene	0.3	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,3-Dichlorobenzene	0.3	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,2-Dichlorobenzene	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,4-Dichlorobenzene	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Benzene	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Toluene	0.2	ND	0.84	ND	1.9	1.1	0.74	0.94	0.69	0.74	0.94	0.74	0.74	0.74	0.94
Triethylbenzene	0.2	ND	ND	0.35	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND

ND = Not Detected

Water AFB - May 1985 Sampling

Volatiles Analysis by EPA Methods 601 (GC/Hall Detector) and 602 (GC/FID)

Component	Detection Limit, ug/L	Concentration, ug/L						
		Blank 6/4	Blank 6/6	Blank 6/7	Blank 6/12	Blank 6/13	Blank 6/14	Blank 6/17
Chloromethane	1.0	-	-	-	ND	ND	ND	ND
Bromomethane	1.2	-	-	-	ND	ND	ND	ND
Dichlorodifluoromethane	1.8	-	-	-	ND	ND	ND	ND
Vinyl chloride	0.2	-	-	-	ND	ND	ND	ND
Chloroethane	0.5	-	-	-	ND	ND	ND	ND
Perchloroethylene	0.2	-	-	-	0.32	0.25	0.20	0.38
Trichloroethylene	2.0	-	-	-	ND	ND	ND	ND
1,1-Dichloroethane	0.2	-	-	-	ND	ND	ND	ND
1,1,1-Trichloroethane	0.1	-	-	-	ND	ND	ND	ND
1,1,2-Dichloroethane	0.1	-	-	-	ND	ND	ND	ND
Chloroform	0.1	-	-	-	ND	ND	ND	0.17
1,2-Dichloroethane	0.2	-	-	-	ND	ND	ND	ND
1,1,1-Trichloroethane	0.1	-	-	-	ND	ND	ND	ND
Carbon tetrachloride	0.1	-	-	-	ND	ND	ND	ND
Bromochloromethane	0.1	-	-	-	ND	ND	ND	ND
1,1-Dichloropropane	0.1	-	-	-	ND	ND	ND	ND
1,1,2,2-Tetrachloroethane	0.3	-	-	-	ND	ND	ND	ND
1,1,1,1-Tetrafluoroethane	0.1	-	-	-	ND	ND	ND	ND
1,1,2,2-Tetrachloroethane	0.1	-	-	-	ND	ND	ND	ND
1,1,1-Trichloroethane	0.05	-	-	-	ND	ND	ND	ND
1,1,2,2-Tetrachloroethane	0.2	-	-	-	ND	ND	ND	ND
1,1,1,2-Tetrachloroethane	0.2	-	-	-	ND	ND	ND	ND
1,1,1,2,2-Pentachloroethane	0.2	-	-	-	ND	ND	ND	ND
1,1,1,1,2-Pentachloroethane	0.2	-	-	-	ND	ND	ND	ND
1,1,1,2,2-Pentachloroethane	0.2	-	-	-	ND	ND	ND	ND
1,1,1,1,2,2-Hexachloroethane	0.2	0.21	0.23	ND	ND	ND	ND	ND
1,1,1,2,2,2-Hexachloroethane	0.2	ND	ND	ND	ND	ND	ND	ND

ROUND 85-2



inter-office memorandum

TO: ~~Bill Jones~~
Allison Dunn (Concord Office)
Kass Sheedy
Rich Johnson
cc: Earl Hansen (Memo Only)
Les Eng (Memo Only)

DATE: August 2, 1985

RECEIVED

OFFICE OF THE DIRECTOR

FROM: Judy Porta *JP*

SUBJECT: FINAL REPORT
MATHER A.F.B.

W. O. No.: 0628-09-05

The attached represents the final report for all samples currently in-house for MATHER A.F.B. If you have any questions, please don't hesitate to call.

JP/eb



Date of Report: August 2, 1985

MATHER A.F.B.
SOLUBLE METALS SUMMARY REPORT
FOR
WATER SAMPLES REC'D JUNE 28, 1985
W.O. NO. 0628-09-05

DATE SAMPLES COLLECTED: June 26, 1985
SAMPLES SUBMITTED BY: Debbie Jones

R.F.W. NO.	SAMPLE DESCRIPTION	SOLUBLE				
		Cd ug/L	Cr ug/L	Pb ug/L	Ni ug/L	Ag ug/L
8506-660-0010	MAFB-7	<2.5	<10	31.8*	<40	<2.5
-0020	MAFB-70	N.R.	N.R.	N.R.	N.R.	N.R.
-0030	MAFB-8	<2.5	<10	<10	<40	<2.5
-0040	MAFB-80	N.R.	N.R.	N.R.	N.R.	N.R.
-0050	MAFB-9	<2.5	<10	<10	55	<2.5
-0060	MAFB-90	<2.5	<10	<10	54	<2.5
-0070	MAFB-10	<2.5	<10	<10	41	<2.5
-0070	MAFB-10	<2.5	<10	<10	<40	<2.5
	DUP (LAB DUPLICATE)					
-0080	MAFB-100	<2.5	<10	<10	<40	<2.5
-0090	MAFB-11	<2.5	<10	<10	<40	<2.5
-0100	MAFB-110	N.R.	N.R.	N.R.	N.R.	N.R.
8506-660-0000	LAB BLANK	---	<10	---	---	---
-000K	BLANK SPIKE	---	118	---	---	---

RECOVERY

N.R. = NOT REQUESTED

EPA METHOD:	213.2	218.2	239.2	249.1	272.2
DATE OF ANALYSIS:	7-19-85	7-24-85	7-12-85	6-18-85	7-22-85

COMPILED BY: Judith A. Porta
Judith A. Porta
Laboratory Operations Manager
WESTON Analytical Laboratories

APPROVED BY: Earl M. Hansen
Earl M. Hansen, Ph.D.
Manager
WESTON Analytical Laboratories

* Suspended value per phone conversation with Leslie Eng, Inorganic Lab. Supervisor
7-22-85.



Date of Report: August 2, 1985

MATHER A.F.B.
INORGANICS SUMMARY REPORT
FOR WATER SAMPLES
REC'D JUNE 28, 1985

DATE SAMPLES COLLECTED: June 26, 1985
SAMPLES SUBMITTED BY: Debbie Jones

R.F.W. NO.	SAMPLE DESCRIPTION	CN mg/L	O/G mg/L	PHENOLICS mg/L	TOC mg/L
3506-660-0010	MAFB-7	<0.01	0.57	<0.005	10.2
-0020	MAFB-70	<0.01	N.R.	0.007	N.R.
-0030	MAFB-8	<0.01	0.68	<0.005	4.6
-0040	MAFB-80	<0.01	N.R.	0.007	N.R.
-0050	MAFB-9	<0.01	0.67	<0.005	5.6
-0050	MAFB-9	---	---	<0.005	---
	DUP (LAB DUPLICATE)				
-0060	MAFB-90	<0.01	0.57	0.006	5.7
-0060	MAFB-90	---	---	---	5.5
	DUP (LAB DUPLICATE)				
-006K	MAFB-90	---	---	---	106
	SPIKE (MATRIX SPIKE)				RECOVERY
-0070	MAFB-10	<0.01	0.33	0.007	0.6
-0080	MAFB-100	<0.01	0.30	<0.005	<0.5
-0090	MAFB-11	<0.01	0.33	<0.005	0.6
-0100	MAFB-110	<0.01	N.R.	<0.005	N.R.
-0010K	MAFB-110	---	---	91	---
	SPIKE (MATRIX SPIKE)				RECOVERY
-0000	LAB BLANK	---	---	<0.005	<0.5
	BLANK				
-000K	BLANK SPIKE	---	---	93	98
	SPIKE				RECOVERY
METHOD (EPA)		335.2	413.2	420.1	415.2
DATE OF ANALYSIS:		7-3-85	7-2-85	7-1-85	7-1-85

N.R. = NOT REQUIRED

PREPARED BY: Judith A. Porta
Judith A. Porta
Laboratory Operations Manager
WESTON Analytical Laboratories

APPROVED BY: Earl M. Hansen
Earl M. Hansen, Ph.D.
Manager
WESTON Analytical Laboratories



Date of Report: August 2, 1985

MATHER A.F.B.
SOLUBLE METALS SUMMARY REPORT
FOR
WATER SAMPLES REC'D JULY 2, 1985
W.O. NO. 0628-09-05

DATE SAMPLES COLLECTED: June 27, 1985
SAMPLES SUBMITTED BY: Debbie Jones

R.F.W. NO.	SAMPLE DESCRIPTION	SOLUBLE				
		Cd ug/L	Cr ug/L	Pb ug/L	Ni ug/L	Ag ug/L
8507-673-0010	FB-1	<2.5	<10	<10	51	<2.5
-0090	MAFB-6	<2.5	<10	<10	58	<2.5
-0100	MAFB-60	<2.5	<10	<10	61	<2.5
-0110	MAFB-5	<2.5	<10	<10	51	<2.5
-0120	MAFB-4	<2.5	<10	<10	62	<2.5

SOL. METALS ANALYSIS NOT REQUESTED FOR SAMPLES ACW, ACW-1, MAFB-1, MAFB-2, MAFB-3 AND MAFB-30.

E.P.A. METHOD: 213.2 218.2 239.2 249.1 272.2
DATE OF ANALYSIS: 7-19-85 7-24-85 7-12-85 6-18-85 7-22-85

COMPILED BY: Judith A. Porta
Judith A. Porta
Laboratory Operations Manager
WESTON Analytical Laboratories

APPROVED BY: Earl M. Hansen
Earl M. Hansen, Ph.D.
Manager
WESTON Analytical Laboratories



Date of Report: August 2, 1985

MATHER A.F.B.
INORGANICS SUMMARY REPORT
FOR
WATER SAMPLES REC'D JULY 2, 1985

W.O. NO. 0628-09-05

DATE SAMPLES COLLECTED: June 27, 1985
SAMPLES SUBMITTED BY: Debbie Jones

R.F.W. NO.	SAMPLE DESCRIPTION	CN ⁻ mg/L	O/G mg/L	PHENOLICS mg/L	TOC mg/L
8507-673-0010	FB-1	<0.01	0.24	0.009	0.5
-0020	FH-5	N.R.	0.20	N.R.	<0.5
-0030	ACW	N.R.	0.27	N.R.	<0.5
-0040	ACW-1	N.R.	0.28	N.R.	<0.5
-0050	MAFB-1	N.R.	0.76	N.R.	<0.5
-0060	MAFB-2	N.R.	0.29	N.R.	0.7
-0070	MAFB-3	N.R.	0.19	N.R.	1.0
-0080	MAFB-30	N.R.	0.21	N.R.	0.6
-0090	MAFB-6	N.R.	0.29	N.R.	<0.5
-0100	MAFB-60	N.R.	0.31	N.R.	<0.5
-0110	MAFB-5	N.R.	0.14	N.R.	0.5
-0120	MAFB-4	N.R.	0.38	N.R.	0.8
-0120	MAFB-4	---	---	---	0.8
	DUP (LAB DUPLICATE)				
-0000	LAB BLANK	---	---	---	<0.5
-000K	BLANK SPIKE	---	---	---	96 RECOVERY

E.P.A. METHOD: 335.2 413.2 420.1 415.2
DATE OF ANALYSIS: 7-9-85 7-2-85 7-16-85 7-5-85
N.R. = NOT REQUESTED

COMPILED BY: Judith A. Porta
Judith A. Porta
Laboratory Operations Manager
WESTON Analytical Laboratories

APPROVED BY: Earl M. Hansen
Earl M. Hansen, Ph.D.
Manager
WESTON Analytical Laboratories



Date of Report: August 2, 1985

MATHER A.F.B.
METALS SUMMARY REPORT
FOR
SOIL SAMPLES REC'D JULY 19, 1985
W.O. NO. 0628-05-26

DATE SAMPLES COLLECTED: June 30, 1985
SAMPLES SUBMITTED BY: Kathy Schultz

R.F.W. NO.	SAMPLE DESCRIPTION	TOTAL				
		Pb mg/Kg	Cr mg/Kg	Cd mg/Kg	Ni mg/Kg	Ag mg/Kg
S506-627-0010	US	14.4	53.8	4.20	26.0	0.730
-0020	USD	13.1	101	4.03	22.7	0.580
-0020 DUP	USD (LAB DUPLICATE)	48.4	23.3	3.37	20.6	0.330
-0030	DS	44.3	35.0	4.65	26.2	0.220
-0000	LAB BLANK	---	---	---	---	0.625
-000K	BLANK SPIKE	---	---	---	---	90 RECOVERY
DATE OF ANALYSIS:		6-22-85	6-29-85	6-29-85	8-1-85	7-25-85

COMPILED BY: Judith A. Porta
Judith A. Porta
Laboratory Operations Manager
WESTON Analytical Laboratories

APPROVED BY: Earl M. Hansen
Earl M. Hansen, Ph.D.
Manager
WESTON Analytical Laboratories



Date of Report: July 30, 1985

MATHER A.F.B.
DMN SUMMARY REPORT
FOR
SAMPLES REC'D JULY 2, 1985

W.O. NO.# 0628-09-05

DATE SAMPLES COLLECTED: June 27, 1985

DATE EXTRACTED: July 3, 1985

DATE ANALYZED: July 3, 1985

R.F.W. NO:	SAMPLE DESCRIPTION	DIMETHYLNITROSAMINE ug/L	SURROGATE RECOVERY (D ₅ -NITROBENZENE)
8507-673-0010	FB-1	<1	63%
-0090	MAFB-6	<1	49%
-0100	MAFB-60	<1	51%
-0110	MAFB-5	<1	73%
-0120	MAFB-4	<1	52%
8507-673/	Lab Blank	<1	65%
8507-673/Spike	Blank Spike	31% Recovery	55%
8507-673/Spike Dup.	Blank Spike Dup.	44% Recovery	76%

Compiled by: Judith A. Porta
Judith A. Porta
Lab Support Manager
WESTON Analytical Laboratories

Approved by: Earl M. Hansen, Ph.D.
Earl M. Hansen, Ph.D.
Manager
WESTON Analytical Laboratories

inter-office memorandum

TO: Katherine Sheedy

DATE: July 16, 1985

[REDACTED]

FROM: David Ben-Hur, Stockton Laboratory *DB*

SUBJECT: Analytical Results, Mather AFB, Second W. O. No.:
Sampling Round, June 1985

Attached are the results of the analyses performed at the Stockton Laboratory on the samples collected in the second round of resampling at Mather AFB

MATHER AFB
 Sampling and Analysis Chronology

Sample ID	Date Sampled	EPA 601	EPA 602	Pesticides		Herbicides		PCB's	
				Extracted	Analyzed	Extracted	Analyzed	Extracted	Analyzed
MAFB-7	6/26/85	7/10/85	7/9/85	-	-	-	-	-	-
MAFB-8	"	"	"	-	-	-	-	-	-
MAFB-9	"	"	"	-	-	-	-	-	-
MAFB-10	"	"	"	-	-	-	-	-	-
MAFB-11	"	"	"	-	-	-	-	-	-
MAFB-90	"	"	"	-	-	-	-	-	-
MAFB-100	"	"	"	-	-	-	-	-	-
MAFB-1	6/27/85	"	"	-	-	-	-	7/3/85	7/9/85
MAFB-2	"	"	"	-	-	-	-	"	"
MAFB-3	"	"	"	-	-	-	-	"	"
MAFB-4	"	"	"	7/3/85	7/9/85	7/1/85	7/8/85	-	-
MAFB-5	"	"	"	"	"	"	"	-	-
MAFB-6	"	"	"	"	"	"	"	-	-
MAFB-30	"	"	"	-	-	-	-	7/3/85	7/9/85
MAFB-60	"	"	"	7/3/85	7/9/85	7/1/85	7/8/85	-	-
FI-5	"	"	"	-	-	-	-	-	-
ACW	"	"	"	-	-	-	-	7/3/85	7/9/85
ACW-1	"	"	"	-	-	-	-	"	"
FB-1	"	"	"	7/3/85	7/9/85	7/1/85	7/8/85	"	"

MATHER AFB - Second Round, June 1985
QA/QC Data

1. Second Column Confirmation for Volatile Compounds

The following samples have been subjected to a second column confirmation. The confirmation was performed qualitatively only. Compounds that were identified and quantitated in the primary column, but could not be confirmed, were reported as ND - not detected.

Sample ID
MAFB-8
MAFB-9
MAFB-11
MAFB-90
MAFB-1
MAFB-2
MAFB-3
MAFB-5
MAFB-6
MAFB-30
MAFB-60
ACW
ACW-1

2. Laboratory Duplicate for Volatile Compounds

Sample ID: MAFB-8

<u>Compound</u>	<u>Concentration, ug/L</u>	
	<u>First</u>	<u>Second</u>
1,1-Dichloroethene	1.4	1.7
1,1-Dichloroethane	1.6	2.0
Trans-1,2-Dichloroethene	2.0	2.1
1,2-Dichloroethane	0.16	ND
1,1,1-Trichloroethane	0.71	0.53
Trichloroethene	100.	120.
Tetrachloroethene	5.4	7.1
Chlorobenzene	3.0	1.3
1,2-Dichlorobenzene	1.0	0.76
Benzene	0.99	0.52

3. Matrix Spike for Volatile Compounds

<u>Compound</u>	<u>Spike ug/L</u>	<u>Percent Recovery MAFB-7</u>
Chlorobenzene	2.0	92
1,2-Dichlorobenzene	2.0	90
1,3-Dichlorobenzene	2.0	86
1,4-Dichlorobenzene	2.0	85
Toluene	2.0	89

MATHER AFB - Second Round, June 1985
QA/QC Data

4. Water Spike for Pesticides and Herbicides

<u>Compound</u>	<u>Spike, ug/L</u>	<u>Percent Recovery</u>
o,p'-DDT	0.15	73
Chlordane	0.14	86
2,4-D	0.18	92

LAB NO. 85-06-035

MATHER AFB - Second Round, June 1985
PCB Analysis

Parameter	Detection Limit, ug/L	Concentration, ug/L						
		MAFB-1	MAFB-2	MAFB-3	MAFB-30	ACW	ACW-1	FB-1
PCB 1016	0.04	ND	ND	ND	ND	ND	ND	ND
PCB 1221	0.10	ND	ND	ND	ND	ND	ND	ND
PCB 1232	0.10	ND	ND	ND	ND	ND	ND	ND
PCB 1242	0.05	ND	ND	ND	ND	ND	ND	ND
PCB 1248	0.08	ND	ND	ND	ND	ND	ND	ND
PCB 1254	0.05	ND	ND	ND	ND	ND	ND	ND
PCB 1260	0.15	ND	ND	ND	ND	ND	ND	ND

LAB NO. 85-06-035

MATHER AFB - Second Round, June 1985
Pesticide and Herbicide Analysis

Compound	Detection Limit, ug/L	Concentration, ug/L				
		MAFB-4	MAFB-5	MAFB-6	MAFB-60	FB-1
o,p'-DDT	0.02	ND	ND	ND	ND	ND
p,p'-DDT	0.02	ND	ND	ND	ND	ND
Chlordane	0.02	ND	ND	ND	ND	ND
2,4-D	0.06	ND	ND	ND	ND	ND



inter-office memorandum

TO: Katherine Sheedy
cc: Alison Dunn, Concord Office

DATE: November 4, 1985

FROM: David Ben-Hur *DB*

SUBJECT: Mather AFB Volatiles Analysis Results W. O. No.:

Attached are the corrected results for the water samples collected at Mather AFB during May and June 1985.

These data are corrected for laboratory blanks.

K-42

Mather AFB - June 1985 Sampling
 Revised Report
 Volatiles Analysis by EPA Methods 601 (GC/Hall Detector) and 602 (GC/PID)

Component	Detection Limit, ug/L	Concentration, ug/L			
		FB-5	ACW	ACW-1	FB-1
Chloroethane	1.0	ND	ND	ND	ND
Bromomethane	1.2	ND	ND	ND	ND
Dichlorodifluoromethane	1.8	ND	ND	ND	ND
Vinyl chloride	0.2	ND	ND	ND	ND
Chloroethane	0.5	ND	ND	ND	ND
Methylene chloride	0.2	ND	ND	ND	ND
Trichlorofluoromethane	2.0	ND	ND	ND	ND
1,1-Dichloroethene	0.2	ND	ND	ND	ND
1,1-Dichloroethane	0.1	ND	ND	ND	ND
Trans-1,2-dichloroethene	0.1	ND	ND	ND	ND
Chloroform	0.1	ND	ND	ND	ND
1,2-Dichloroethane	0.02	ND	ND	ND	ND
1,1,1-Trichloroethane	0.1	ND	3.7	3.7	ND
Carbon tetrachloride	0.1	ND	ND	ND	ND
Bromodichloromethane	0.1	ND	ND	ND	ND
1,2-Dichloropropane	0.1	ND	ND	ND	ND
Cis-1,3-dichloropropene	0.3	ND	ND	ND	ND
Trichloroethene	0.1	ND	67.	76.	ND
Dibromochloromethane	0.1	ND	ND	ND	ND
1,1,2-Trichloroethane	0.05	ND	ND	ND	ND
Trans-1,2-dichloropropene	0.2	ND	ND	ND	ND
2-Chloroethylvinyl ether	0.2	ND	ND	ND	ND
Bromoform	0.2	ND	ND	ND	ND
1,1,2,2-Tetrachloroethane	0.05	ND	ND	ND	ND
Tetrachloroethene	0.05	ND	ND	ND	ND
Chlorobenzene	0.3	ND	ND	ND	ND
1,3-Dichlorobenzene	0.3	ND	ND	ND	ND
1,2-Dichlorobenzene	0.2	ND	ND	ND	ND
1,4-Dichlorobenzene	0.2	ND	ND	ND	ND
Benzene	0.2	ND	ND	0.48	0.28
Toluene	0.2	0.60	0.30	ND	ND
Benzofluorene	0.2	ND	ND	ND	ND

ND = Not Detected

Mather AFB - June 1985 Sampling
 Analytical Report
 Volatiles Analysis by EPA Methods 601 (GC/Hall Detector) and 602 (GC/PID)

Component	Detection Limit, ug/L	Concentration, ug/l					
		MAFB-1	MAFB-2	MAFB-3	MAFB-4	MAFB-5	MAFB-6
Chloroethane	1.0	ND	ND	ND	ND	ND	ND
Bromomethane	1.2	ND	ND	ND	ND	ND	ND
Dichlorodifluoromethane	1.8	ND	ND	ND	ND	ND	ND
Vinyl chloride	0.2	ND	ND	ND	ND	ND	ND
Chloroethane	0.5	ND	ND	ND	ND	ND	ND
Methylene chloride	0.2	ND	ND	ND	ND	ND	ND
Trichlorofluoromethane	2.0	ND	ND	ND	ND	ND	ND
1,1-Dichloroethene	0.2	ND	ND	ND	ND	ND	ND
1,1-Dichloroethane	0.1	ND	ND	ND	ND	ND	ND
Trans-1,2-dichloroethene	0.1	0.43	ND	ND	ND	ND	ND
Perchloroethene	0.1	ND	ND	ND	ND	ND	ND
1,1,1-Trichloroethane	0.02	ND	ND	ND	ND	ND	ND
1,1,2-Trichloroethane	0.1	4.5	ND	2.5	1.6	1.4	1.7
Tetra chloroethide	0.1	ND	ND	ND	ND	ND	ND
1,1,1,1-tetraethane	0.1	ND	ND	ND	ND	ND	ND
1,1,1,2-tetraethane	0.1	ND	ND	ND	ND	ND	ND
1,1,2,2-tetraethane	0.3	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.1	460.	36.	27.	ND	ND	120.
1,1,1,1,2-pentachloropropane	0.1	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.05	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.2	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.2	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.2	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.05	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.3	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.3	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.2	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.2	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.05	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.3	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.2	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.2	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.2	0.21	0.21	0.22	0.57	0.27	0.40
1,1,1,1,2-pentachloropropane	0.2	0.44	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.2	ND	ND	ND	ND	ND	ND
1,1,1,1,2-pentachloropropane	0.2	ND	ND	ND	ND	ND	ND

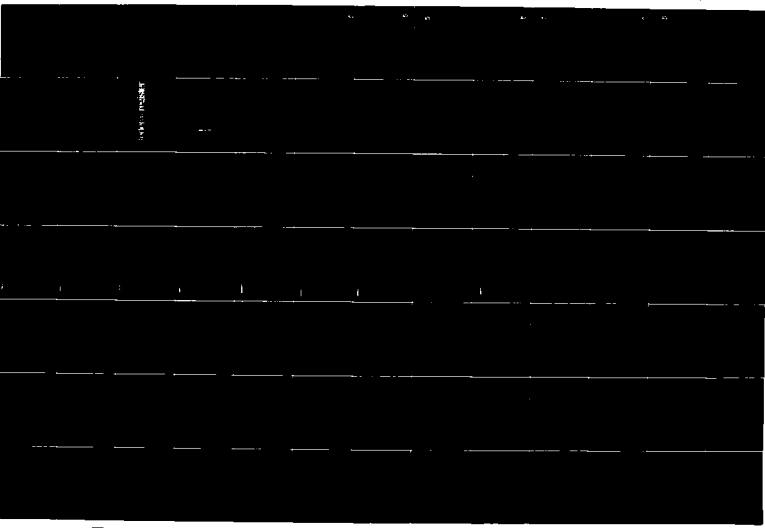
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INSTALLATION RESTORATION PROGRAM PHASE 2
CONFIRMATION/QUANTIFICATION STAGE 1 VOLUME 2 APPENDICES
(U) WESTON (ROY F) INC WEST CHESTER PA JUN 86
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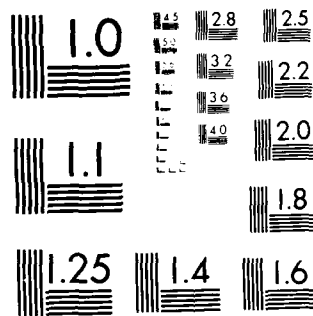
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Mather AFB - June 1985 Sampling
 Revised Report
 Volatiles Analysis by EPA Methods 601 (GC/Hall Detector) and 602 (GC/PID)

Component	Detection Limit, ug/L	Concentration, ug/L												
		MAFB-7	MAFB-8	MAFB-9	MAFB-10	MAFB-11	MAFB-90	MAFB-100						
Chloromethane	1.0	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Bromomethane	1.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Dichlorodifluoromethane	1.8	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Vinyl chloride	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Chloroethane	0.5	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Methylene chloride	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Trichlorofluoromethane	2.0	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1-Dichloroethene	0.2	ND	1.4	0.64	0.64	0.64	0.64	0.64	0.64	0.64	0.64	0.64	0.64	0.64
1,1-Dichloroethane	0.1	ND	1.6	0.34	0.34	0.34	0.34	0.34	0.34	0.34	0.34	0.34	0.34	0.34
Trans-1,2-dichloroethene	0.1	ND	2.0	0.15	0.15	0.15	0.15	0.15	0.15	0.15	0.15	0.15	0.15	0.15
Chloroform	0.1	ND	ND	ND	ND	ND	ND	ND	ND	0.32	ND	ND	ND	ND
1,2-Dichloroethane	0.02	ND	0.16	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1,1-Trichloroethane	0.1	0.15	0.71	0.89	0.89	0.89	0.89	0.89	0.89	0.52	3.7	ND	ND	ND
Carbon tetrachloride	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Bromodichloromethane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,2-Dichloropropane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Cis-1,3-dichloropropene	0.3	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Trichloroethene	0.1	0.79	100.	5.4	5.4	5.4	5.4	5.4	5.4	5.4	5.5	0.26	0.26	0.26
Dibromochloromethane	0.1	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1,2-Trichloroethane	0.05	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Trans-1,3-dichloropropene	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
2-Chloroethylvinyl ether	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Bromofom	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,1,2,2-Tetrachloroethane	0.05	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Tetrachloroethene	0.05	ND	5.4	0.95	0.95	0.95	0.95	0.95	0.95	0.95	0.98	0.98	0.98	0.98
Chlorobenzene	0.3	ND	3.0	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,3-Dichlorobenzene	0.3	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
1,2-Dichlorobenzene	0.2	ND	1.0	ND	ND	ND	ND	ND	ND	ND	0.55	0.55	0.55	0.55
1,4-Dichlorobenzene	0.2	0.54	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Benzene	0.2	ND	0.99	0.45	0.45	0.45	0.45	0.45	0.45	1.9	2.7	0.32	0.32	0.25
Toluene	0.2	ND	ND	ND	ND	ND	ND	ND	ND	0.32	ND	ND	ND	ND
Ethylbenzene	0.2	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND

ND = Not detected

Volatiles analysis by EPA Methods 601 (GC/Hall Detector) and 602 (GC/PID)

Component	Detection Limit ug/L	Concentration, ug/kg	
		US	DS
Chloromethane	1.0	ND	ND
Bromomethane	1.2	ND	ND
Dichlorodifluoromethane	1.8	ND	ND
Vinyl chloride	0.2	ND	ND
Chloroethane	0.5	ND	ND
Methylene chloride	0.2	ND	ND
Trichlorofluoromethane		ND	ND
1,1-Dichloroethane	0.2	ND	ND
1,1-Dichloroethane	0.1	ND	ND
Trans-1,2-dichloroethane	0.1	ND	ND
Chloroform	0.1	ND	ND
1,2-Dichloroethane	0.02	ND	ND
1,1,1-Trichloroethane	0.1	ND	ND
Carbon tetrachloride	0.1	ND	ND
Bromodichloromethane	0.1	ND	ND
1,2-Dichloropropane	0.1	ND	ND
Trans-1,3-dichloropropene	0.3	ND	ND
Trichloroethene	0.1	ND	ND
Dibromochloromethane	0.1	ND	ND
1,1,2-Trichloroethane	0.05	ND	ND
Cis-1,3-dichloropropene	0.2	ND	ND
2-Chloroethylvinyl ether	0.2	ND	ND
Bromoform	0.2	ND	ND
1,1,2,2-Tetrachloroethane	0.05	ND	ND
Tetrachloroethene	0.05	ND	ND
Chlorobenzene	0.3	ND	ND
1,3-Dichlorobenzene	0.3	ND	ND
1,2-Dichlorobenzene	0.2	ND	ND
1,4-Dichlorobenzene	0.2	1.1	1.9
Benzene	0.2	ND	ND
Toluene	0.2	ND	ND
Ethylbenzene	0.2	ND	ND

Mathar APB - June 1985 Sampling

Volatiles Analysis by EPA Methods 601 (GC/Hall Detector) and 602 (GC/PID)

Component	Detection Limit, ug/L	Concentration, ug/L	
		Blank 7/8	Blank 7/10
Chloromethane	1.0	-	ND
Bromomethane	1.2	-	ND
Dichlorodifluoromethane	1.8	-	ND
Vinyl chloride	0.2	-	ND
Chloroethane	0.5	-	ND
Methylene chloride	0.2	-	0.32
Trichlorofluoromethane	2.0	-	ND
1,1-Dichloroethane	0.2	-	ND
1,1-Dichloroethane	0.1	-	ND
Trans-1,2-dichloroethane	0.1	-	ND
Chloroform	0.1	-	ND
1,2-Dichloroethane	0.02	-	ND
1,1,1-Trichloroethane	0.1	-	0.17
Carbon tetrachloride	0.1	-	ND
Bromodichloromethane	0.1	-	ND
1,2-Dichloropropane	0.1	-	ND
Cis-1,3-dichloropropene	0.3	-	ND
Trichloroethene	0.1	-	ND
Dibromochloromethane	0.1	-	ND
1,1,2-Trichloroethane	0.05	-	ND
Trans-1,3-dichloropropene	0.2	-	ND
2-Chloroethylvinyl ether	0.2	-	ND
Bromoform	0.2	-	ND
1,1,2,2-tetrachloroethane	0.05	-	ND
Tetrachloroethane	0.05	-	ND
Chlorobenzene	0.3	ND	ND
1,3-Dichlorobenzene	0.3	ND	ND
1,2-Dichlorobenzene	0.2	ND	ND
1,4-Dichlorobenzene	0.2	ND	ND
Benzene	0.2	ND	ND
Toluene	0.2	0.23	-
Ethylbenzene	0.2	ND	-

(ND) = Not detected

APPENDIX L

FEDERAL AND STATE DRINKING WATER AND
HUMAN HEALTH STANDARDS APPLICABLE IN
THE STATE OF CALIFORNIA



**GUIDE TO GROUND-WATER STANDARDS
OF THE UNITED STATES**

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Prepared by
Ecological Analysts, Inc.
15 Loveton Circle
Sparks, Maryland 21152

3. FEDERAL PROTECTION OF GROUND-WATER QUALITY

The federal programs dealing with the protection of ground-water quality are administered largely by the Environmental Protection Agency (EPA). The federal programs which provide the framework for state regulations are summarized in this section.



3.1 GROUND-WATER PROTECTION POLICY

At this writing, February 1983, U.S. EPA's final policy on ground-water protection, scheduled for September 1982 release, has not been published. Based on the proposed strategy published by EPA in November 1980 and recent press releases, it appears that EPA will be implementing a policy that would give the states lead responsibility in the protection of ground-water quality. EPA's efforts apparently will be focused in three major areas:

1. Development of an internally consistent federal approach to ground-water protection
2. Monitoring, research and development efforts directed toward more comprehensive problem definition and new detection, controls, and clean-up technology development
3. Guidance, coordination, and assistance to states in the development of state policies

A significant component of EPA's policy is expected to be a ground-water classification system which could be used to determine the degree of protection needed for various types of ground water. Ground-water classification is discussed in Chapter 4.

3.2 CLEAN WATER ACT

This statute refers to ground-water protection in municipal waste water treatment, planning, and research programs. Its principal regulatory programs, however, focus on surface water. Section 303 empowers EPA to approve states water quality standards which are based on the states classification of rivers and streams. Many states have included ground water in their definition of "waters of the state" for purposes of this act (state summaries). On this basis the National (state) Pollutant Discharge Elimination System (NPDES/SPDES) permitting process may be invocable for purposes of ground-water protection. In addition the act empowers EPA to

1. Develop a comprehensive program for ground-water pollution control [Section 102(a)]
2. In cooperation with states, equip and maintain a surveillance system for monitoring ground-water quality [Section 104(a)(5)]
3. Provide grants to states and area-wide agencies to develop ground-water quality management plans to identify salt water intrusion and control disposal of pollutants in subsurface excavations, and control disposition of wastes (May include authority for comprehensive ground-water management plans, including conjunctive use with surface water) [Section 102(c), 208(b)]
4. Require development of Best Management Practices (BMP) to control nonpoint source pollution problems to ground-water quality [Section 208(b)]
5. Develop criteria for ground-water quality considering kind and extent of effects on health and welfare from the presence of pollutants [Section 304(a)]
6. Determine information necessary to restore and maintain chemical, physical, and biological integrity of ground water [Section 304(a)]
7. Issue information on the factors necessary to restore and maintain chemical, physical, and biological integrity of ground water [Sections 304(a)(2)]

3.3 SAFE DRINKING WATER ACT

This statute authorizes EPA to set maximum contaminant levels (MCLs) and monitoring requirements for public water systems and provides for the protection of underground sources of drinking water. The MCLs regulate the quality of "finished" water, i.e., water as delivered, not the quality of the source water. As discussed below, the MCLs have been utilized by EPA and the states as the basis for other regulations dealing with ground-water quality and protection.



3.3.1 National Interim Primary Drinking Water Regulations

EPA initiated a detailed study of the health effects of various contaminants in water soon after the Safe Drinking Water Act (SDWA) was signed into law. So that the regulations could include the findings of this and other studies, the primary drinking water regulations were to be developed in two stages: an interim version and a final version. The interim version of the regulation became effective 24 June 1977. SDWA provides for delegation of authority to the states. State Primary Drinking Water Regulations must be at least as stringent as the federal regulations.

The National Interim Primary Drinking Water Regulations define Maximum Contaminant Level as the maximum permissible level of a contaminant in water which is delivered to the free-flowing outlet of the ultimate user of a public water system, except in the case of turbidity (applicable to surface water only) where the maximum permissible level is measured at the point of entry to the distribution system. The MCLs are provided with the state summaries.

3.3.2 National Secondary Drinking Water Regulations

These regulations control contaminants in drinking water that primarily affect the aesthetic qualities relating to the public acceptance of drinking water. At considerably higher concentrations of these contaminants, health implications may also exist as well as aesthetic degradation. The National Secondary Drinking Water Regulations are not federally enforceable but are intended as guidelines for the states.

Secondary Maximum Contaminant Levels (SMCLs) are defined as the maximum permissible level of a contaminant in water which is delivered to the free-flowing outlet of the ultimate user of a public water system. Federal and state SMCLs are provided in the state summaries. The states may establish higher or lower levels which may be appropriate depending upon local conditions such as unavailability of alternate sources of water or other compelling factors, provided the public health and welfare are not adversely affected.

3.3.3 Sole Source Aquifer

The Sole Source Aquifer provisions of SDWA allow EPA to designate an aquifer as the sole source of drinking water for an area thereby guaranteeing protection from contamination by federally assisted activities. Local, regional, or state agencies can petition EPA for sole source designation. The EPA Administrator may designate an aquifer which is a sole or principal drinking water source if its contamination would create a significant hazard to public health. If the designation is made, no federal money or financial commitment may be made for any project which the Administrator determines may contaminate the designated aquifer through its recharge zone.

At this writing, February 1983, EPA has designated the following ten sole source aquifers:

Biscayne Aquifer - Florida	Nassau and Suffolk counties - New York
Bund Valley Aquifer - New Jersey	Cape Cod - Massachusetts
Edwards Aquifer - Texas	Fresno - California
Camano Island—Whidbey Island Aquifer - Washington	Ten Mile Creek - Maryland
Spokane-Rathdrum Aquifer - Washington and Idaho	Northern Guam Lens - Guam

The following eighteen are under consideration:

Arizona	New York
Santa Cruz, Upper Santa Cruz, Aura-Altar Basins	Kings and Queens counties
California	Sardinia
Scotts Valley	Schenectady
	Vestal
Delaware	Pennsylvania
New Castle County	Seven Valleys
Florida	Texas
Volusia - Floridan Aquifer	Carizo-Wilcox Aquifer
Idaho	Texas and New Mexico
Snake River Plain	Delaware Basin
Louisiana	Wisconsin
Baton Rouge	Niagara Aquifer
DeSota Parish	
New Jersey	
Coastal Plain	
Ridgewood	
Upper Rockaway	

3.3.4 Underground Injection Control

The Underground Injection Control (UIC) program regulates the uses of underground injection wells to protect an underground source of drinking water (USDW). USDW means an aquifer or its portion which

1. supplies any public water system or contains a sufficient quantity of ground water to supply a public water system;
2. currently supplies drinking water for human consumption or contains less than 10,000 mg/liter total dissolved solids; and
3. is not an exempted aquifer (40 CFR 146.04 provides criteria for exemption).

SDWA requires any state designated by EPA as requiring a UIC program to develop and submit a state UIC program for EPA approval. EPA has designated each of the fifty states.

The federal program classifies injection wells as follows:

Class I—Wells used to inject hazardous waste, or other industrial and municipal disposal wells which inject fluids beneath the lower-most formation containing a USDW within one-quarter mile of the well bore.

Class II—Wells that inject fluids

1. which are brought to the surface as part of conventional oil or natural gas production and may be mixed with production waste waters from gas plants, unless those waters are classified as a hazardous waste at the time of injection;
2. for enhanced recovery of oil or natural gas; and
3. for storage of hydrocarbons which are liquid at standard temperature and pressure.

Class III—Wells that inject for extraction of minerals including

1. mining of sulfur by the Frasch process;
2. in situ production of uranium or other metals. This category includes only in situ production from ore bodies which have not been conventionally mined. Solution mining of conventional mines such as stopes leaching is included in Class V; and
3. solution mining of salts or potash.

Class IV—Wells used to dispose of hazardous or radioactive waste into or above a formation which contains a USDW within one-quarter mile of the well. Also, wells used to inject hazardous waste that cannot be classified as Class I or Class IV under the above criteria are Class IV wells.

Class V—All other injection wells (40 CFR 146.05(e) and 146.51 provide specific information and exemptions).

Underground injection is controlled through the permitting process. Construction, operation, monitoring and reporting activities are controlled. Individual state programs are based upon, and must be essentially equivalent to, the federal criteria and standards (40 CFR 146).

3.4 TOXIC SUBSTANCE CONTROL ACT

This statute (TSCA) authorizes EPA to restrict or prohibit the manufacture, distribution, and use of products which may result in unreasonable risk to health and the environment. Although ground water is not specifically named in the Act, EPA has taken the position that the protection of health and the environment includes the protection of ground water.

3.5 FEDERAL INSECTICIDE, FUNGICIDE, RODENTICIDE ACT

This statute (FIFRA) gives EPA the responsibility to control the sale and use of all pesticides to prevent unreasonable adverse environmental and health effects. The use and disposal of pesticide packages and containers is also regulated. In deciding whether to register, cancel, suspend, or change the classification of a pesticide, EPA considers a broad range of environmental impacts including those affecting ground water.





3.6 RESOURCE CONSERVATION AND RECOVERY ACT

The Solid Waste Disposal Act and the Resource Recovery Act of 1970, as amended by the Resource Conservation and Recovery Act of 1976 (RCRA), require EPA to establish a national program to regulate the management of waste materials.

3.6.1 Solid Waste

Subtitle D of RCRA established a broad-based national program to improve solid waste management through the development of state and regional solid waste management plans. The act offered federal financial assistance to states interested in developing and implementing a solid waste management plan. The state plans, under federal guidelines, identify respective responsibilities of local, state, and regional authorities, and encourage resource recovery and conservations and the application and enforcement of environmentally sound disposal practices.

A major element of the Subtitle D program is the open dump inventory. Section 4005 of RCRA prohibits open dumping. Federal criteria for classifying solid waste management facilities are provided in 40 CFR 257. EPA cannot approve a state solid waste management program with less stringent criteria. Solid waste management facilities failing to satisfy the criteria are considered open dumps. In order to satisfy these criteria, a facility or practice (in addition to other environmental considerations) shall not contaminate an underground drinking water source beyond the solid waste boundary or beyond an alternative boundary established by the state or in court pursuant to the stipulations of 40 CFR 257.3-4. The federal criteria define contamination as an exceedence of the MCLs provided in the National Interim Primary Drinking Water Regulations or an increase in concentration of any parameter for which the ambient concentration exceed the MCL.

3.6.2 Hazardous Waste

EPA has issued a series of hazardous waste regulations under Subtitle C of RCRA (40 CFR 260 to 267 and 122 to 124). On 19 May 1980, EPA issued a comprehensive set of standards for generators and transporters of hazardous waste and "interim status" standards for facilities in existence on 19 November 1980, that treat, store, or dispose of hazardous waste. Such facilities were allowed to operate under interim status until they received an RCRA permit. Subsequently, EPA issued standards for granting RCRA permits to treatment and storage facilities. Standards for land disposal facilities were issued on 26 July 1982—virtually completing the program for controlling hazardous waste under RCRA.

The standards for permitting land disposal facilities were issued after a wide range of regulatory options were considered. Over a period of several years, EPA proposed two different sets of land disposal standards and solicited comments on various issues. On 13 February 1981, EPA issued temporary standards for new land disposal facilities. The 26 July regulations replace those temporary standards except for Class I underground injection wells. These will remain subject to the temporary standards until final standards are issued.

The regulations consist primarily of two complementary sets of performance standards:


1. A set of design and operating standards tailored to each of four types of facilities
2. Ground-water monitoring and response regulations applicable to all land disposal facilities

The design and operating standards implement a liquids management strategy that has two goals:

1. Minimize leachate generated at the facility
2. Remove leachate generated to minimize its chance of reaching ground water

The major requirements include

1. Liner
 - Requirement: design to prevent migration of waste out of the facility during its active life
 - Applicability: landfills, surface impoundments, and waste piles
2. Leachate collection and removal
 - Requirement: collect and remove leachate from the facility and ensure that leachate depth over the liner does not exceed 30 centimeters (1 foot)
 - Applicability: landfills and waste piles

- 
3. Run-on and runoff control systems
 - Requirement: design to control flow during at least 25-year storm
 - Applicability: landfills, waste piles, land treatment
 4. Wind dispersal controls
 - Requirement: cover waste or otherwise manage unit to control wind dispersal
 - Applicability: landfills, waste piles, and land treatment units that contain particulate matter
 5. Overtopping controls
 - Requirement: prevent overtopping or overfilling
 - Applicability: surface impoundments
 6. Disposal unit closure
 - Requirement: final cover (cap) over waste unit designed to minimize infiltration of precipitation
 - Applicability: landfills and surface impoundments (if used for disposal)
 7. Storage unit closure
 - Requirement: remove waste and decontaminate
 - Applicability: surface impoundments used for treatment or storage and waste piles
 8. Postclosure Care
 - Maintain effectiveness of final cover
 - Operate leachate collection and removal system
 - Maintain ground-water monitoring system (and leak detection system where double liner is used)
 - Continue 30 years after closure

The goal of the ground-water monitoring and response program is to detect and correct any ground-water contamination. There are four main elements:

1. A detection monitoring program which requires the permittee to install a system to monitor ground water in the uppermost aquifer to determine if a leachate plume has reached the edge of the waste management area.
2. A ground-water protection standard is set when a hazardous constituent is detected. The standard specifies concentration limits, compliance point, and compliance period.
3. A compliance monitoring program determines if the facility is complying with its ground-water protection standard.
4. Corrective action is required when the ground-water protection standard is violated. The permittee must either remove the contamination or treat it in place to restore ground-water quality.

Until hazardous waste management facilities are issued permits, existing facilities will continue to operate under interim status standards. Facilities operating under interim status will be required to file Part B applications for final permits.

Under Subtitle C of RCRA, EPA approves state hazardous waste management programs in two phases. Phase I authorization gives states the right to control transportation and generation of hazardous wastes within their borders and to regulate existing treatment, storage, and disposal facilities. Phase II authorization includes the permitting of new facilities.

3.7 COMPREHENSIVE ENVIRONMENTAL RESPONSE, COMPENSATION, AND LIABILITY ACT

This statute (CERCLA), commonly referred to as Superfund, authorizes EPA to respond to releases or threatened releases into the environment, including ground water, of any hazardous substance which may present an imminent and substantial danger to public health. The act provides funds for emergency action and has cost recovery provisions.



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Classification—Ground water is included in the definition of "Waters of the State" as found in the California Water Quality Act. Ground water has been included in beneficial use classes developed as part of Basin Management Programs of the Water Resources Control Board and the Regional Boards.

Quality Standards—The general policy is a nondegradation policy to protect the present and possible future use of ground water as a source of potable, industrial, and agricultural water supply. Quality standards are specific to each use class and Basin Program.

Drinking Water Standards—The California Water Resources Control Board has adopted the federal primary and secondary drinking water standards.

Appropriations—There are no state-wide permit requirements, however, see Controlled Use Areas below.

Controlled Use Areas—Several ground-water basins are being managed by local authorities in response to special legislative acts and court orders. These authorities regulate ground-water withdrawals within their jurisdictions. However, these areas account for less than five percent of all ground-water basins.

Well Construction—Local counties may adopt well construction standards and require drillers to be licensed. Approximately half of California's 58 counties have done so.

Underground Injection Control—California is in the process of submitting a UIC program for EPA approval. The Water Resources Control Board will be the lead agency in the program. Class II wells will be regulated by the Oil and Gas Division of the Department of Conservation.

Waste Management Facilities—The solid and hazardous waste management programs are administered by the Solid Waste Management Board. The Hazardous Waste Management Regulations are administered by the Department of Health Services.

Solid Waste—The California Solid Waste Management Regulations require a ground-water monitoring system for disposal sites. Monitoring requirements are on a case-by-case basis.

Hazardous Waste—California has received interim status authorization for its RCRA Phase I program and is seeking Phase II authority. Ground-water monitoring requirements are included in permit conditions and are generally equivalent to EPA requirements.

Sole Source Aquifers—The Fresno area aquifer has been designated as sole source by EPA. The Scotts Valley aquifer is under consideration by EPA.

Geological Surveys—

Division of Mines and Geology
Department of Conservation
1416 Ninth St.
Sacramento, CA 95814
916-445-1923
State Geologist:
Dr. James F. Davis

Water Resources Division
U.S. Geological Survey
Federal Bldg., Room W-2235
2800 Cottage Way
Sacramento, CA 95825
916-484-4606
District Chief:
T.J. Durbin

References—

California Water Quality Act
(California Water Code, Div. 7, Ch. 482)
California Solid Waste Management Regulations
(California Admin. Code, Title 14, Div. 7, Ch. 1-5
and 9)

California Hazardous Waste Management
Regulations
(California Admin. Code, Title 22, Div. 4, Ch. 30)

Contacts—

Ms. Helen Joyce Peters
Department of Water Resources
P.O. Box 388
Sacramento, CA 95802
916-445-2182

Mr. Evan Nossoff
Water Resources Control Board
P.O. Box 100
Sacramento, CA 95801
916-322-8353

Revisions provided by Ms. Helen Joyce Peters in a letter received 11 April 1983.

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Parameter (mg/l unless noted)	Drinking Water Standards		Quality Standards	Monitoring Requirements	
	Federal	State		Solid Waste	Hazardous Waste
Arsenic	0.05	0.05			
Barium	1.0	1.0			
Cadmium	0.010	0.010			
Chromium	0.05	0.05			
Lead	0.05	0.05			
Mercury	0.002	0.002			
Selenium	0.01	0.01			
Silver	0.05	0.05			
Fluoride	1.4-2.4	1.4-2.4			
Nitrate (as N)	10.0	10.0			
Endrin	0.0002	0.0002			
Lindane	0.004	0.004			
Methoxychlor	0.1	0.1			
Toxaphene	0.005	0.005			
2,4-D	0.1	0.1			
2,4,5-TP Silvex	0.01	0.01			
Trihalomethanes	0.1	0.1			
Turbidity (TU)	1.0	1.0			
Coliform bacteria — membrane filter test (#/100 ml)	1.0	1.0			
Gross alpha (pCi/l)	15.0	15.0			
Combined Radium 226 and Radium 228	5.0	5.0			
Beta and photon particle activity (mrem/yr)	4.0	4.0			
Sodium	M	M			
Chloride	250.0	250.0			
Color (units)	15.0	15.0			
Copper	1.0	1.0			
Corrosivity	Noncorrosive	Noncorrosive			
Foaming agents	0.5	0.5			
Iron	0.3	0.3			
Manganese	0.05	0.05			
Odor (threshold no.)	3.0	3.0			
pH (units)	6.5-8.5	6.5-8.5			
Sulfate	250.0	250.0			
Total dissolved solids	500.0	500.0			
Zinc	5.0	5.0			
Phenols					
Specific conductance					
Total organic carbon					
Total organic halogen					

Note: "M" denotes monitoring requirement. See Section 4.3

ENVIRONMENTAL PROTECTION AGENCY NATIONAL INTERIM PRIMARY DRINKING WATER REGULATIONS

(40 CFR 141; 40 FR 59565, December 24, 1975; Amended by 41 FR 28402, July 9, 1976; 44 FR 68641, November 29, 1979; Corrected by 45 FR 15542, March 11, 1980; 45 FR 57342, August 27, 1980)

Title 40—Protection of Environment CHAPTER I—ENVIRONMENTAL PROTECTION AGENCY

SUBCHAPTER D—WATER PROGRAMS

PART 141—NATIONAL INTERIM PRIMARY DRINKING WATER REGULATIONS

Subpart A—General

Sec.

- 141.1 Applicability.
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Subpart B—Maximum Contaminant Levels

- 141.11 Maximum contaminant levels for inorganic chemicals.
- 141.12 Maximum contaminant levels for organic chemicals.
- 141.13 Maximum contaminant levels for turbidity.
- 141.14 Maximum microbiological contaminant levels.
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Subpart C—Monitoring and Analytical Requirements

- 141.21 Microbiological contaminant sampling and analytical requirements.
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- 141.23 Inorganic chemical sampling and analytical requirements.
- 141.24 Organic chemicals other than total trihalomethanes, sampling and analytical requirements.
- 141.25 Analytical Methods for Radioactivity.
- 141.26 Monitoring Frequency for Radioactivity in Community Water Systems.
- 141.27 Alternative analytical techniques.
- 141.28 Approved laboratories.
- 141.29 Monitoring of consecutive public water systems.

Subpart D—Reporting Public Notification, and Record-keeping

- 141.31 Reporting requirements.
- 141.32 Public notification of variances, exemptions, and non-compliance with regulations.
- 141.33 Record maintenance.

Subpart E—Special Monitoring Regulations for Organic Chemicals

141.40 Special monitoring for organic chemicals.

Authority: Secs. 1412, 1414, 1445, and 1450 of the Public Health Service Act, 88 Stat. 1660 (42 U.S.C. 300g-1, 300g-3, 300g-4, and 300g-9).

Subpart A—General

§ 141.1 Applicability.

This part establishes primary drinking water regulations pursuant to section 1412 of the Public Health Service Act, as amended by the Safe Drinking Water Act (Pub. L. 93-523); and related regulations applicable to public water systems.

§ 141.2 Definitions.

As used in this part, the term:

(a) "Act" means the Public Health Service Act, as amended by the Safe Drinking Water Act, Pub. L. 93-523.

(b) "Contaminant" means any physical, chemical, biological, or radiological substance or matter in water.

(c) "Maximum contaminant level" means the maximum permissible level of a contaminant in water which is delivered to the free flowing outlet of the ultimate user of a public water system, except in the case of turbidity where the maximum permissible level is measured at the point of entry to the distribution system. Contaminants added to the water under circumstances controlled by the user, except those resulting from corrosion of piping and plumbing caused by water quality, are excluded from this definition.

(d) "Person" means an individual, corporation, company, association, partnership, State, municipality, or Federal agency.

(e) "Public water system" means a system for the provision to the public of piped water for human consumption, if such system has at least fifteen service connections or regularly serves an average of at least twenty-five individuals daily at least 60 days out of the year. Such term includes (1) any collection, treatment, storage, and distribution facilities under control of the operator of such system and used primarily in connection with such system, and (2) any collection or pretreatment storage facilities not under such control which are used primarily in connection with such system. A public water system is either

a "community water system" or a "non-community water system."

(i) "Community water system" means a public water system which serves at least 15 service connections used by year-round residents or regularly serves at least 25 year-round residents.

(ii) "Non-community water system" means a public water system that is not a community water system.

(f) "Sanitary survey" means an on-site review of the water source, facilities, equipment, operation and maintenance of a public water system for the purpose of evaluating the adequacy of such source, facilities, equipment, operation and maintenance for production and distribution of safe drinking water.

(g) "Standard sample" means an aliquot of finished drinking water that is examined for the presence of coliform bacteria.

(h) "State" means the agency of State government which has jurisdiction over public water systems. During any period when a State does not have primary enforcement responsibility pursuant to Section 1413 of the Act, the term "State" means the Regional Administrator, U.S. Environmental Protection Agency.

(i) "Supplier of water" means a person who owns or operates a public water system.

(j) "Dose equivalent" means the product of the absorbed dose from ionizing radiation and such factors as account for differences in biological effectiveness to the type of radiation and its distribution in the body as specified by the International Commission on Radiological Units and Measurements (ICRU).

(k) "Rem" means the unit of equivalent from ionizing radiation to total body or any internal organ or system. A "millirem (mrem)" is 1/1000 of a rem.

(l) "Picocurie (pCi)" means that quantity of radioactive material producing 2.22 nuclear transformations per minute.

(m) "Gross alpha particle activity" means the total radioactivity due to alpha particle emission as inferred from measurements on a dry sample.

(n) "Man-made beta particle and photon emitters" means all radionuclides emitting beta particles and/or photons.

listed in Maximum Permissible Body Burdens and Maximum Permissible Concentration of Radionuclides in Air or Water for Occupational Exposure, NIOS Handbook 69, except the daughter products of thorium-232, uranium-235 and uranium-238.

(o) "Gross beta particle activity" means the total radioactivity due to beta particle emission as inferred from measurements on a dry sample.

[41 FR 28402, July 9, 1976]
[141.2 (p)-(t) added by 44 FR 68641, November 29, 1979]

(p) "Halogen" means one of the chemical elements chlorine, bromine or iodine.

(q) "Trihalomethane" (THM) means one of the family of organic compounds, named as derivatives of methane, wherein three of the four hydrogen atoms in methane are each substituted by a halogen atom in the molecular structure.

(r) "Total trihalomethanes" (TTHM) means the sum of the concentration in milligrams per liter of the trihalomethane compounds (trichloromethane [chloroform], dibromochloromethane, bromodichloromethane and tribromomethane [bromoform]), rounded to two significant figures.

(s) "Maximum Total Trihalomethane Potential (MTP)" means the maximum concentration of total trihalomethanes produced in a given water containing a disinfectant residual after 7 days at a temperature of 25° C or above.

(t) "Disinfectant" means any oxidant, including but not limited to chlorine, chlorine dioxide, chloramines, and ozone added to water in any part of the treatment or distribution process, that is intended to kill or inactivate pathogenic microorganisms.

§ 141.3 Coverage.

This part shall apply to each public water system, unless the public water system meets all of the following conditions:

(a) Consists only of distribution and storage facilities (and does not have any collection and treatment facilities);

(b) Obtains all of its water from, but is not owned or operated by, a public water system to which such regulations apply;

(c) Does not sell water to any person; and

(d) Is not a carrier which conveys passengers in interstate commerce.

§ 141.4 Variances and exemptions.

Variances or exemptions from certain provisions of these regulations may be granted pursuant to Sections 1415 and 1416 of the Act by the entity with primary enforcement responsibility, Provisions under Part 142, National Interim Primary Drinking Water Regulations Implementation—subpart E (Variances)

and subpart F (Exemptions)—apply where EPA has primary enforcement responsibility.

§ 141.5 Siting requirements.

Before a person may enter into a financial commitment for or initiate construction of a new public water system or increase the capacity of an existing public water system, he shall notify the State, and, to the extent practicable, avoid locating part or all of the new or expanded facility at a site which:

(a) Is subject to a significant risk from earthquakes, floods, fires or other disasters which could cause a breakdown of the public water system or a portion thereof; or

(b) Except for intake structures, is within the floodplain of a 100-year flood or is lower than any recorded high tide where appropriate records exist. The U.S. Environmental Protection Agency will not seek to override land use decisions affecting public water systems siting which are made at the State or local government levels.

§ 141.6 Effective dates.

[141.6 revised by 44 FR 68641, November 29, 1979]

(a) Except as provided in paragraph (b) of this section, the regulations set forth in this part shall take effect on June 24, 1977.

(b) The regulations for total trihalomethanes set forth in § 141.12(c) shall take effect 2 years after the date of promulgation of these regulations for community water systems serving 75,000 or more individuals, and 4 years after the date of promulgation for communities serving 10,000 to 74,999 individuals.

(c) The regulations set forth in 141.11 (a), (c) and (d); 141.14(a)(1); 141.14(b)(1)(c); 141.14(b)(2)(i); 141.14(d); 141.21 (a), (c) and (i); 141.22 (a) and (e); 141.23 (a)(3) and (a)(4); 141.23(f); 141.24(a)(3); 141.24 (e) and (f); 141.25(e); 141.27(a); 141.28 (a) and (b); 141.31 (a), (c), (d) and (e); 141.32(b)(3); and 141.32(d) shall take effect immediately upon promulgation.

(d) The regulations set forth in 141.41 shall take effect 18 months from the date of promulgation. Suppliers must complete the first round of sampling and reporting within 12 months following the effective date.

(e) The regulations set forth in 141.42 shall take effect 18 months from the date of promulgation. All requirements in 141.42 must be completed within 12 months following the effective date.

[141.6 (c)-(e) added by 45 FR 57342, August 27, 1980]

Subpart B—Maximum Contaminant Levels
§ 141.11 Maximum contaminant level for inorganic chemicals.

(a) The MCL for nitrate is applicable to both community water systems and non-community water systems except as provided by in paragraph (d). The levels for the other organic chemicals apply only to community water systems. Compliance with MCLs for inorganic chemicals is calculated pursuant to § 141.23.

[141.11(a) amended by 45 FR 57342, August 27, 1980]

(b) The following are the maximum contaminant levels for inorganic chemicals other than fluoride:

Contaminant	Level, milligrams per liter
Arsenic	0.05
Barium	1
Cadmium	0.010
Chromium	0.05
Lead	0.05
Mercury	0.002
Nitrate (as N)	10
Selenium	0.01
Silver	0.05

(c) When the annual average of the maximum daily air temperatures for the location in which the community water system is situated is the following, the maximum contaminant levels for fluoride are:

Temperature Driest Fahrenheit	Driest Celsius	Level, milligrams per liter
33.7 and below	12.0 and below	0.4
33.8 to 34.4	12.1 to 14.6	0.2
34.5 to 35.0	14.7 to 17.6	0.0
35.1 to 35.6	17.7 to 21.4	1.4
35.7 to 36.2	21.5 to 25.2	1.6
36.3 to 36.8	26.3 to 31.5	1.4

(c) Fluoride at optimum levels in drinking water has been shown to have beneficial effects in reducing the occurrence of tooth decay.

[141.11 (c) amended by 45 FR 57342, August 27, 1980]

(d) At the discretion of the State, nitrate levels not to exceed 20 mg/l may be allowed in a non-community water system if the supplier of water demonstrates to the satisfaction of the State that:

(1) Such water will not be available to children under 6 months of age; and

(2) There will be continuous posting of the fact that nitrate levels exceed 10 mg/l and the potential health effects of exposure; and

(3) Local and State public health authorities will be notified annually of nitrate levels that exceed 10 mg/l; and

(4) No adverse health effects shall result.

[141.11 (d) added by 45 FR 57342, August 27, 1980]

§ 141.12 Maximum contaminant levels for organic chemicals.

[141.12 revised by 44 FR 68641, November 29, 1979]

The following are the maximum contaminant levels for organic chemicals. The maximum contaminant levels for organic chemicals in paragraphs (a) and (b) of this section apply to all community water systems. Compliance with the maximum contaminant levels in paragraphs (a) and (b) is calculated pursuant to § 141.24. The maximum contaminant level for total trihalomethanes in paragraph (c) of this section applies only to community water systems which serve a population of 10,000 or more individuals and which add a disinfectant (oxidant) to the water in any part of the drinking water treatment process. Compliance with the maximum contaminant level for total trihalomethanes is calculated pursuant to § 141.30.

Level,
milligrams
per liter

- (a) Chlorinated hydrocarbons:
 - Endrin (1,2,3,4,10, 10-hexachloro-6,7-epoxy-1,4, 4a,5,6,7,8,8a-octa-hydro-1,4-endo, endo-5,8-dimeth-ano naphthalene) 0.0002
0.2 ppb
 - Lindane (1,2,3,4,5,6-hexachlorocyclohexane, gamma isomer) 0.001
4 ppb
 - Methoxychlor (1,1,1-Trichloro-2, 2-bis [p-methoxy phenyl] ethane) 0.1
100 ppb
 - Toxaphene (C₁₂H₈Cl₄, Technical chlorinated camphene, 67-69 percent chlorine) 0.005
5 ppb

- (b) Chlorophenoxy acids:
 - 2,4-D, (2,4-Dichlorophenoxyacetic acid) 0.1
100 ppb
 - 2,4,5-TP Silvex (2,4,5-Trichlorophenoxy propionic acid) 0.01
10 ppb

(c) Total trihalomethanes (the sum of the concentrations of bromodichloromethane, dibromochloromethane, tri-bromomethane (bromoform) and tri-chloromethane (chloroform) 0.10 mg/l.

[141.12(c) added by 44 FR 68641, November 29, 1979]

§ 141.13 Maximum contaminant levels for turbidity.

The maximum contaminant levels for turbidity are applicable to both community water systems and non-community water systems using surface water sources in whole or in part. The maximum contaminant levels for turbidity in drinking water, measured at a representative entry point(s) to the distribution system, are

- (a) One turbidity unit (TU), as de-

termined by a monthly average pursuant to § 141.22, except that five or fewer turbidity units may be allowed if the supplier of water can demonstrate to the State that the higher turbidity does not do any of the following:

- (1) Interfere with disinfection;
 - (2) Prevent maintenance of an effective disinfectant agent throughout the distribution system; or
 - (3) Interfere with microbiological determinations.
- (b) Five turbidity units based on an average for two consecutive days pursuant to § 141.22.

§ 141.14 Maximum microbiological contaminant levels.

The maximum contaminant levels for coliform bacteria, applicable to community water systems and non-community water systems, are as follows:

- (a) When the membrane filter technique pursuant to § 141.21(a) is used, the number of coliform bacteria shall not exceed any of the following:

[141.14(a)(1) revised by 45 FR 57342, August 27, 1980]

- (1) One per 100 milliliters as the arithmetic mean of all samples examined per compliance period pursuant to § 141.21(b) or (c), except that, at the primary Agency's discretion systems required to take 10 or fewer samples per month may be authorized to exclude one positive routine sample per month from the monthly calculation if:

(i) as approved on a case-by-case basis the State determines and indicates in writing to the public water system that no unreasonable risk to health existed under the conditions of this modification. This determination should be based upon a number of factors not limited to the following: (A) the system provided and had maintained an active disinfectant residual in the distribution system, (B) the potential for

contamination as indicated by a sanitary survey, and (C) the history of the water quality at the public water system (e.g. MCL or monitoring violations); (ii) the supplier initiates a check sample on each of two consecutive days from the same sampling point within 24 hours after notification that the routine sample is positive, and each of these check samples is negative; and (iii) the original positive routine sample is reported and recorded by the supplier pursuant to § 141.31(a) and § 141.33(a). The supplier shall report to the State its compliance with the conditions specified in this paragraph and a summary of the corrective action taken to resolve the prior positive sample result. If a positive routine sample is not used for the monthly calculation, another routine

sample must be analyzed for compliance purposes. This provision may be used only once during two consecutive compliance periods.

- (2) Four per 100 milliliters in more than one sample when less than 20 are examined per month; or
- (3) Four per 100 milliliters in more than five percent of the samples when 20 or more are examined per month.

(b) (1) When the fermentation tub method and 10 milliliter standard portions pursuant to § 141.21(a) are used coliform bacteria shall not be present in any of the following:

[141.14(b)(1)(i) revised by 45 FR 57342, August 27, 1980]

(i) More than 10 percent of the portions (tubes) in any one month pursuant to § 141.21 (b) or (c) except that, at the State's discretion, systems required to take 10 or fewer samples per month may be authorized to exclude one positive routine sample resulting in one or more positive tubes per month from the monthly calculation if: (A) as approved on a case-by-case basis the State determines and indicates in writing to the public water system that no unreasonable risk to health existed under the conditions of this modification. This determination should be based upon a number of factors not limited to the following: (1) the system provided and had maintained an active disinfectant residual in the distribution system, (2) the potential for contamination as indicated by a sanitary survey, and (3) the history of the water quality at the public water system (e.g. MCL or monitoring violations); (B) the supplier initiates a check sample on each of two consecutive days from the sampling point within 24 hours after notification that the routine sample is positive, and each of these check samples is negative; and (C) the original positive routine sample is reported and recorded by the supplier pursuant to § 141.31(a) and § 141.33(a). The supplier shall report to the State its compliance with the conditions specified in this paragraph and report the action taken to resolve the prior positive sample result. If a positive routine sample is not used for the monthly calculation, another routine sample must be analyzed for compliance purposes. This provision may be used only once during two consecutive compliance periods.

- (ii) three or more portions in more than one sample when less than 20 samples are examined per month; or
- (iii) three or more portions in more than five percent of the samples when 20 or more samples are examined per month.

(2) When the fermentation tube

Friday
November 28, 1980

Federal Register

Part V

**Environmental
Protection Agency**

Water Quality Criteria Documents;
Availability

**ENVIRONMENTAL PROTECTION
AGENCY**
(FRL 1623-3)
**Water Quality Criteria Documents;
Availability**
AGENCY: Environmental Protection
Agency.

ACTION: Notice of Water Quality Criteria
Documents.

SUMMARY: EPA announces the availability and provides summaries of water quality criteria documents for 64 toxic pollutants or pollutant categories. These criteria are published pursuant to section 304(a)(1) of the Clean Water Act.

AVAILABILITY OF DOCUMENTS:

Summaries of both aquatic-based and health-based criteria from the documents are published below. Copies of the complete documents for individual pollutants may be obtained from the National Technical Information Service (NTIS), 5285 Port Royal Road, Springfield, VA 22161, (703-487-4650). A list of the NTIS publication order numbers for all 64 criteria documents is published below. These documents are also available for public inspection and copying during normal business hours at: Public Information Reference Unit, U.S. Environmental Protection Agency, Room 2404 (rear), 401 M St., S.W., Washington, D.C. 20460. As provided in 40 CFR Part 2, a reasonable fee may be charged for copying services. Copies of these documents are also available for review in the EPA Regional Office libraries.

Copies of the documents are not available from the EPA office listed below. Requests sent to that office will be forwarded to NTIS or returned to the sender.

1. Acenaphthene, PB81-117269.
2. Acrolein, PB81-117277.
3. Acrylonitrile, PB81-117285.
4. Aldrin/Dieldrin, PB81-117301.
5. Antimony, PB81-117319.
6. Arsenic, PB81-117327.
7. Asbestos, PB81-117335.
8. Benzene, PB81-117293.
9. Benzidine, PB81-117343.
10. Beryllium, PB81-117350.
11. Cadmium, PB81-117368.
12. Carbon Tetrachloride, PB81-117376.
13. Chlordane, PB81-117384.
14. Chlorinated benzenes, PB81-117392.
15. Chlorinated ethanes, PB81-117400.
16. Chloroalkyl ethers, PB81-117418.
17. Chlorinated naphthalene, PB81-117426.
18. Chlorinated phenols, PB81-117434.
19. Chloroform, PB81-117442.
20. 2-chlorophenol, PB81-117459.

21. Chromium, PB81-117467.
22. Copper, PB81-117475.
23. Cyanides, PB81-117483.
24. DDT, PB81-117491.
25. Dichlorobenzenes, PB81-117509.
26. Dichlorobenzidine, PB81-117517.
27. Dichloroethylenes, PB81-117525.
28. 2,4-dichlorophenol, PB81-117533.
29. Dichloropropanes/propenes, PB81-117541.
30. 2,4-dimethylphenol, PB81-117558.
31. Dinitrotoluene, PB81-117566.
32. Diphenylhydrazine, PB81-117731.
33. Endosulfan, PB81-117574.
34. Endrin, PB81-117582.
35. Ethylbenzene, PB81-117590.
36. Fluoranthene, PB81-117608.
37. Haloethers, PB81-117616.
38. Halomethanes, PB81-117624.
39. Heptachlor, PB81-117632.
40. Hexachlorobutadiene, PB81-117640.
41. Hexachlorocyclohexane, PB81-117657.
42. Hexachlorocyclopentadiene, PB81-117665.
43. Isophorone, PB81-117673.
44. Lead, PB81-117681.
45. Mercury, PB81-117699.
46. Naphthalene, PB81-117707.
47. Nickel, PB81-117715.
48. Nitrobenzene, PB81-117723.
49. Nitrophenols, PB81-117749.
50. Nitrosamines, PB81-117758.
51. Pentachlorophenol, PB81-117764.
52. Phenol, PB81-117772.
53. Phthalate esters, PB81-117780.
54. Polychlorinated biphenyls (PCBs), PB81-117798.
55. Polynuclear aromatic hydrocarbons, PB81-117806.
56. Selenium, PB81-117814.
57. Silver, PB81-117822.
58. Tetrachloroethylene, PB81-117830.
59. Thallium, PB81-117848.
60. Toluene, PB81-117855.
61. Toxaphene, PB81-117863.
62. Trichloroethylene, PB81-117871.
63. Vinyl chloride, PB81-117889.
64. Zinc, PB81-117897.

FOR FURTHER INFORMATION CONTACT:
Dr. Frank Gostomski, Criteria and Standards Division (WH-385), United States Environmental Protection

Agency, Washington, D.C. 20460 (202) 245-3022.

SUPPLEMENTARY INFORMATION:
Background

Pursuant to section 304(a)(1) of the Clean Water Act, 33 U.S.C. 1314(a)(1), EPA is required to periodically review and publish criteria for water quality accurately reflecting the latest scientific knowledge:

(A) on the kind and extent of all identifiable effects on health and welfare including, but not limited to, plankton, fish,

shellfish, wildlife, plant life, shorelines, beaches, esthetics, and recreation which may be expected from the presence of pollutants in any body of water, including groundwater. (B) on the concentration and dispersal of pollutants, or their byproducts, through biological, physical, and chemical processes, and (C) on the effects of pollutants on biological community diversity, productivity, and stability, including information on the factors affecting rates of eutrophication and rates of organic and inorganic sedimentation for varying types of receiving waters.

EPA is today announcing the availability of criteria documents for 64 of the 65 pollutants designated as toxic under section 307(a)(1) of the Act. The document on TCDD (Dioxin) will be published within the next month after review of recent studies. Criteria for the section 307(a)(1) toxic pollutants being published today will replace the criteria for those same pollutants found in the EPA publication, *Quality Criteria for Water*, (the "Red Book.") Criteria for all other pollutants and water constituents found in the "Red Book" remain valid. The criteria published today have been derived using revised methodologies for determining pollutant concentrations that will, when not exceeded, reasonably protect human health and aquatic life. Draft criteria documents were made available for public comment (44 FR 15926, March 15, 1979, 44 FR 43660, July 25, 1979, 44 FR 56628, October 1, 1979). These final criteria have been derived after consideration of all comments received.

These criteria documents are also issued in satisfaction of the Settlement Agreement in *Natural Resources Defense Council, et al. v. Train*, 8 E.R.C. 2120 (1976), modified, 12 E.R.C. 1833 (D.D.C. 1979). Pursuant to paragraph 11 of that agreement, EPA is required to publish criteria documents for the 65 pollutants which Congress, in the 1977 amendments to the Act, designated as toxic under section 307(a)(1). These documents contain recommended maximum permissible pollutant concentrations consistent with the protection of aquatic organisms, human health, and some recreational activities. Although paragraph 11 imposes certain obligations on the Agency, it does not create additional authority.

**The Development of Water Quality
Criteria**

Section 304(a)(1) criteria contain two essential types of information: (1) discussions of available scientific data on the effects of pollutants on public health and welfare, aquatic life and recreation, and (2) quantitative concentrations or qualitative assessments of the pollutants in water which will generally ensure water

quality adequate to support a specified water use. Under section 304(a)(1), these criteria are based solely on data and scientific judgments on the relationship between pollutant concentrations and environmental and human health effects. Criteria values do not reflect considerations of economic or technological feasibility.

Publication of water quality criteria of this type has been an ongoing process which EPA, and its predecessor Agency, the Federal Water Pollution Control Administration, have been engaged in since 1968. At that time the first Federal compilation of water quality criteria, the so-called "Green Book" (*Water Quality Criteria*), was published. As now, these criteria contained both narrative discussions of the environmental effects of pollutants on a range of possible uses and concentrations of pollutants necessary to support these uses. Since that time, water quality criteria have been revised and expanded with publication of the "Blue Book" (*Water Quality Criteria 1972*) in 1973 and the "Red Book" (*Quality Criteria for Water*) in 1978.

Since publication of the Red Book there have been substantial changes in EPA's approach to assessing scientific data and deriving section 304(a)(1) criteria. Previous criteria were derived from a limited data base. For many pollutants, an aquatic life criterion was derived by multiplying the lowest concentration known to have acute lethal effect on half of a test group of an aquatic species (the LC50 value) by an application factor in order to protect against chronic effects. If data showed a substance to be bioaccumulative or to have other significant long-term effects, a factor was used to reduce the indicated concentrations to a level presumed to be protective. Criteria for the protection of human health were similarly derived by considering the pollutants' acute, chronic, and bioaccumulative effects on non-human mammals and humans.

Although a continuation of the process of criteria development, the criteria published today were derived using revised methodologies (Guidelines) for calculating the impact of pollutants on human health and aquatic organisms. These Guidelines consist of systematic methods for assessing valid and appropriate data concerning acute and chronic adverse effects of pollutants on aquatic organisms, non-human mammals, and humans. By use of these data in prescribed ways, criteria are formulated to protect aquatic life and human health from exposure to the pollutants. For

some pollutants, bioconcentration properties are used to formulate criteria protective of aquatic life uses. For almost all of the pollutants,

bioconcentration properties are used to assess the relative extent of human exposure to the pollutant either directly through ingestion of water or indirectly through consumption of aquatic organisms. Human health criteria for carcinogens are presented as incremental risks to man associated with specific concentrations of the pollutant in ambient water. The Guidelines used to derive criteria protective of aquatic life and human health are fully described in appendices B and C, respectively, of this Notice.

The Agency believes that these Guidelines provide criteria which more accurately reflect the effects of these pollutants on human health and on aquatic organisms and their uses. They are based on a more rational and consistent approach for using scientific data. These Guidelines were developed by EPA scientists in consultation with scientists from outside the Agency and they have been subjected to intensive public comment.

Neither the Guidelines nor the criteria are considered inflexible doctrine. Even at this time, EPA is taking action to employ the resources of peer review groups, including the Science Advisory Board, to evaluate recently published data, and EPA is conducting its own evaluation of new data to determine whether revisions to the criteria documents would be warranted.

The criteria published today are based solely on the effect of a single pollutant. However, pollutants in combination may have different effects because of synergistic, additive, or antagonistic properties. It is impossible in these documents to quantify the combined effects of these pollutants, and persons using criteria should be aware that site-specific analysis of actual combinations of pollutants may be necessary to give more precise indications of the actual environmental impacts of a discharge.

Relationship of the Section 304(a)(1) Criteria to Regulatory Programs

Section 304(a)(1) criteria are not rules and they have no regulatory impact. Rather, these criteria present scientific data and guidance on the environmental effect of pollutants which can be useful to derive regulatory requirements based on considerations of water quality impacts. Under the Clean Water Act, these regulatory requirements may include the promulgation of water quality-based effluent limitations under section 302, water quality standards

under section 303, or toxic pollutant effluent standards under section 307. States are encouraged to begin to modify or, if necessary, develop new programs necessary to support the implementation of regulatory controls for toxic pollutants. As appropriate, States may incorporate criteria for toxic pollutants, based on this guidance, in their water quality standards.

Section 304(a)(1) criteria have been most closely associated with the development of State water quality standards, and the "Red Book" value have, in the past, been the basis for EPA's assessments of the adequacy of State requirements. However, EPA is now completing a major review of its water quality standards policies and regulations. After consideration of comments received on an Advance Notice of Proposed Rulemaking (43 F 29588, July 10, 1978) and the draft criteria documents, the Agency intends to propose, by the end of this year, a revised water quality standards regulation which will clarify the Agency's position on a number of significant standards issues.

With the publication of these criteria however, it is appropriate to discuss EPA's current thinking on standards issues relating to their use. This discussion does not establish new regulatory requirements and is intended as guidance on the possible uses of these criteria and an indication of future rulemaking the Agency may undertake. No substantive requirements will be established without further opportunity for public comment.

Water Quality Standards

Section 303 of the Clean Water Act provides that water quality standards be developed for all surface waters. A water quality standard consists basically of two parts: (1) A "designated use" for which the water body is to be protected (such as "agricultural," "recreation" or "fish and wildlife"); a (2) "criteria" which are numerical pollutant concentration limits or narrative statements necessary to preserve or achieve the designated use. A water quality standard is developed through State or Federal rulemaking proceedings and must be translated into enforceable effluent limitations in a point source (NPDES) permit or may form the basis of best management practices applicable to nonpoint sources under section 208 of the Act.

Relationship of Section 304(a)(1) Criteria to the Criteria Component of State Water Quality Standards

In the ANPRM, EPA announced a policy of "presumptive applicability" for

section 304(a)(1) criteria codified in the "Red Book." Presumptive applicability meant that a State had to adopt a criterion for a particular water quality parameter at least as stringent as the recommendation in the Red Book unless the State was able to justify a less stringent criterion based on: natural - background conditions, more recent scientific evidence, or local, site-specific information. EPA is rescinding the policy of presumptive applicability because it has proven to be too inflexible in actual practice.

Although the section 304(a)(1) criteria represent a reasonable estimate of pollutant concentrations consistent with the maintenance of designated water uses, States may appropriately modify these values to reflect local conditions. In certain circumstances, the criteria may not accurately reflect the toxicity of a pollutant because of the effect of local water quality characteristics or varying sensitivities of local populations. For example, in some cases, ecosystem adaptation may enable a viable, balanced aquatic population to exist in waters with high natural background levels of certain pollutants. Similarly, certain compounds may be more or less toxic in some waters because of differences in alkalinity, temperature, hardness, and other factors.

Methods for adjusting the section 304(a)(1) criteria to reflect these local differences are discussed below.

Relationship of Section 304(a)(1) Criteria to Designated Water Uses:

The criteria published today can be used to support the designated uses which are generally found in State standards. The following section discusses the relationship between the criteria and individual use classifications. Where a water body is designated for more than one use, criteria necessary to protect the most sensitive use should be applied.

1. *Recreation:* Recreational uses of water include such activities as swimming, wading, boating and fishing. Although insufficient data exist on the effects of toxic pollutants resulting from exposure through such primary contact as swimming, section 304(a)(1) criteria based on human health effects may be used to support this designated use where fishing is included in the State definition of "recreation." In this situation only the portion of the criterion based on fish consumption should be used.

2. *Protection and Propagation of Fish and Other Aquatic Life:* The section 304(a)(1) criteria based on toxicity to aquatic life may be used directly to support this designated use.

3. *Agricultural and Industrial Uses:* The section 304(a)(1) criteria were not specifically developed to reflect the impact of pollutants on agricultural and industrial uses. However, the criteria developed for human health and aquatic life are sufficiently stringent to protect these other uses. States may establish criteria specifically designed to protect these uses.

4. *Public Water Supply:* The drinking water exposure component of the human health effects criteria can apply directly to this use classification or may be appropriately modified depending upon whether the specific water supply system falls within the auspices of the Safe Drinking Water Act's (SDWA) regulatory control, and the type and level of treatment imposed upon the supply before delivery to the consumer. The SDWA controls the presence of toxic pollutants in finished ("end-of-tap") drinking water. A brief description of relevant sections of this Act is necessary to explain how the SDWA will work in conjunction with section 304(a)(1) criteria in protecting human health from the effects of toxics due to consumption of water.

Pursuant to section 1412 of the SDWA, EPA has promulgated "National Interim Primary Drinking Water Standards" for certain organic and inorganic substances. These standards establish "maximum contaminant levels" ("MCLs") which specify the maximum permissible level of a contaminant in water which may be delivered to a user of a public water system now defined as serving a minimum of 25 people. MCLs are established based on consideration of a range of factors including not only the health effects of the contaminants but also technological and economic feasibility of the contaminants' removal from the supply. EPA is required to establish revised primary drinking water regulations based on the effects of a contaminant on human health, and include treatment capability, monitoring availability, and costs. Under Section 1401(1)(D)(i) of the SDWA, EPA is also allowed to establish the minimum quality criteria for water which may be taken into a public water supply system.

Section 304(a)(1) criteria provide estimates of pollutant concentrations protective of human health, but do not consider treatment technology, costs and other feasibility factors. The section 304(a)(1) criteria also include fish bioaccumulation and consumption factors in addition to direct human drinking water intake. These numbers were not developed to serve as "end of tap" drinking water standards, and they have no regulatory significance under

the SDWA. Drinking water standards are established based on considerations, including technological and economic feasibility, not relevant to section 304(a)(1) criteria. Section 304(a)(1) criteria may be analogous to the recommended maximum contaminant levels (RMCLs) under section 1412(b)(1)(B) of the SDWA in which, based upon a report from the National Academy of Sciences, the Administrator should set target levels for contaminants in drinking water at which "no known or anticipated adverse effects occur and which allows an adequate margin of safety". RMCLs do not take treatment, cost, and other feasibility factors into consideration. Section 304(a)(1) criteria are, in concept, related to the health-based goals specified in the RMCLs. Specific mandates of the SDWA such as the consideration of multi-media exposure, as well as different methods for setting maximum contaminant levels under the two Acts, may result in differences between the two numbers.

MCLs of the SDWA, where they exist, control toxic chemicals in finished drinking water. However, because of variations in treatment and the fact that only a relatively small number of MCLs have been developed, ambient water criteria may be used by the States as a supplement to SDWA regulations. States will have the option of applying MCLs, section 304(a)(1) human health effects criteria, modified section 304(a)(1) criteria or controls more stringent than these three to protect against the effects of toxic pollutants by ingestion from drinking water.

For untreated drinking water supplies, States may control toxics in the ambient water through either use of MCLs (if they exist for the pollutants of concern), section 304(a)(1) human health effects criteria, or a more stringent contaminant level than the former two options.

For treated drinking water supplies serving less than 25 people, States may choose toxics control through application of MCLs (if they exist for the pollutants of concern and are attainable by the type of treatment) in the finished drinking water. States also have the options to control toxics in the ambient water by choosing section 304(a)(1) criteria, adjusted section 304(a)(1) criteria resulting from the reduction of the direct drinking water exposure component in the criteria calculation to the extent that the treatment procedure reduces the level of pollutants, or a more stringent contaminant level than the former three options.

For treated drinking water supplies serving 25 people or greater, States must control toxics down to levels at least as stringent as MCLs where they exist for

the pollutants of concern) in the finished drinking water. However, States also have the options to control toxics in the ambient water by choosing section 304(a)(1) criteria, adjusted section 304(a)(1) criteria resulting from the reduction of the direct drinking water exposure component in the criteria calculation to the extent that the treatment process reduces the level of pollutants, or a more stringent contaminant level than the former three options.

Inclusion of Specific Pollutants in State Standards:

To date, EPA has not required that a State address any specific pollutant in its standards. Although all States have established standards for most conventional pollutants, the treatment of toxic pollutants has been much less extensive. In the ANPRM, EPA suggested a policy under which States would be required to address a set of pollutants and incorporate specific toxic pollutant criteria into water quality standards. If the State failed to incorporate these criteria, EPA would promulgate the standards based upon these criteria pursuant to section 303(c)(4)(B).

In the forthcoming proposed revision to the water quality standard regulations, a significant change in policy will be proposed relating to the incorporation of certain pollutants in State water quality standards. This proposal will differ from the proposal made in the ANPRM. The ANPRM proposed an EPA-published list of pollutants for which States would have had to develop water quality standards. This list might have contained some (or all) of the 65 toxic pollutants. However, the revised water quality standards regulation will propose a process by which EPA will assist States in identifying specific toxic pollutants required for assessment for possible inclusion in State water quality standards. For these pollutants, States will have the option of adopting the published criteria or of adjusting those criteria based on site-specific analysis.

These pollutants would generally represent the greatest threat to sustaining a healthy, balanced ecosystem in water bodies or to human health due to exposure directly or indirectly from water. EPA is currently developing a process to determine which pollutants a State must assess for possible inclusion in its water quality standards. Relevant factors might include the toxicity of the pollutant, the frequency and concentration of its discharge, its geographical distribution, the breadth of data underlying the

scientific assessment of its aquatic life and human health effects, and the technological and economic capacity to control the discharge of the pollutant. For some of the pollutants, all States may be required to assess them for possible inclusion in their standards. For others, assessment would be restricted to States or limited to specific water bodies where the pollutants pose a particular site-specific problem.

Criteria Modification Process

Flexibility is available in the application of these and any other valid water quality criteria to regulatory programs. Although in some cases they may be used by the States as developed, the criteria may be modified to reflect local environmental conditions and human exposure patterns before incorporation into programs such as water quality standards. If significant impacts of site-specific water quality conditions in the toxicities of pollutants can be demonstrated or significantly different exposure patterns of these pollutants to humans can be shown, section 304(a)(1) criteria may be modified to reflect these local conditions. The term "local" may refer to any appropriate geographic area where common aquatic environmental conditions or exposure patterns exist. Thus, "local" may signify a Statewide, regional, river reach, or entire river basin area. On the other hand, the criteria of some pollutants might be applicable nationwide without the need for adaptation to reflect local conditions. The degree of toxicity toward aquatic organisms and humans characteristic of these pollutants would not change significantly due to local water quality conditions.

EPA is examining a series of environmental factors or water quality parameters which might realistically be expected to affect the laboratory-derived water quality criterion recommendation for a specific pollutant. Factors such as hardness, pH, suspended solids, types of aquatic organisms present, etc. could impact on the chemical's effect in the aquatic environment. Therefore, local information can be assembled and analyzed to adjust the criterion recommendation if necessary.

The Guidelines for deriving criteria for the protection of aquatic life suggest several approaches for modifying the criteria. First, toxicity data, both acute and chronic, for local species could be substituted for some or all of the species used in deriving criteria for the water quality standard. The minimum data requirements should still be fulfilled in calculating a revised criterion. Second,

criteria may be specifically tailored to a local water body by use of data from toxicity tests performed with that ambient water. A procedure such as this would account for local environmental conditions in formulating a criterion relevant to the local water body. Third, site-specific water quality characteristics resulting in either enhancement or mitigation of aquatic life toxicity for the pollutant could be factored into final formulation of the criterion. Finally, the criteria may be made more stringent to ensure protection of an individual species not otherwise adequately protected by any of the three modification procedures previously mentioned.

EPA does not intend to have States assess every local stream segment and lake in the country on an individual basis before determining if an adjustment is necessary. Rather, it is envisioned that water bodies having similar hydrological, chemical, physical, and biological properties will be grouped for the purpose of criteria adjustment. The purpose of this effort is to assist States in adapting the section 304(a) criteria to local conditions when needed, thereby precluding the setting arbitrary and perhaps unnecessarily stringent or underprotective criteria in a water body. In all cases, EPA will still be required, pursuant to section 303(c), to determine whether the State water quality standards are consistent with the goals of the Act, including a determination of whether State-established criteria are adequate to support a designated use.

Criteria for the Protection of Aquatic Life

Interpretation of the Criteria

The aquatic life criteria issued today are summarized in Appendix A of this Federal Register notice. Criteria have been formulated by applying a set of Guidelines to a data base for each pollutant. The criteria for the protection of aquatic life specify pollutant concentrations which, if not exceeded, should protect most, but not necessarily all, aquatic life and its uses. The Guidelines specify that criteria should be based on an array of data from organisms, both plant and animal, occupying various trophic levels. Based on these data, criteria can be derived which should be adequate to protect the types of organisms necessary to support an aquatic community.

The Guidelines are not designed to derive criteria which will protect all life stages of all species under all conditions. Generally some life stage or one or more tested species, and

probably some untested species, will have sensitivities below the maximum value or the 24-hour average under some conditions and would be adversely affected if the highest allowable pollutant concentrations and the worst conditions existed for a long time. In actual practice, such a situation is not likely to occur and thus the aquatic community as a whole will normally be protected if the criteria are not exceeded. In any aquatic community there is a wide range of individual species sensitivities to the effects of toxic pollutants. A criterion adequate to protect the most susceptible life stage of the most sensitive species would in many cases be more stringent than necessary to protect the overall aquatic community.

The aquatic life criteria specify both maximum and 24-hour average values. The combination of the two values is designed to provide adequate protection of aquatic life and its uses from acute and chronic toxicity and bioconcentration without being as restrictive as a one-number criterion would have to be to provide the same amount of protection. A time period of 24 hours was chosen in order to ensure that concentrations not reach harmful levels for unacceptably long periods. Averaging for longer periods, such as a week or a month for example, could permit high concentrations to persist long enough to produce significant adverse effects. A 24-hour period was chosen instead of a slightly longer or shorter period in recognition of daily fluctuations in waste discharges and of the influence of daily cycles of sunlight and darkness and temperature on both pollutants and aquatic organisms.

The maximum value, which is derived from acute toxicity data, prevents significant risk of adverse impact to organisms exposed to concentrations above the 24-hour average. Merely specifying the average value over a specified time period is insufficient because concentrations of chemicals higher than the average value can kill or cause irreparable damage in short periods. Furthermore, for some chemicals the effect of intermittent high exposures is cumulative. It is therefore necessary to place an upper limit on pollutant concentrations to which aquatic organisms might be exposed. The two-number criterion is intended to describe the highest average ambient water concentration which will produce a water quality generally suited to the maintenance of aquatic life while restricting the extent and duration of the excursions over that average to levels which will not cause harm. The only

way to assure the same degree of protection with a one-number criterion would be to use the 24-hour average as a concentration that is not to be exceeded at any time in any place.

Since some substances may be more toxic in freshwater than in saltwater, or vice versa, provision is made for deriving separate water quality criteria for freshwater and for saltwater for each substance. However, for some substances sufficient data may not be available to derive one or both of these criteria using the Guidelines.

Specific aquatic life criteria have not been developed for all of the 65 toxic pollutants. In those cases where there were insufficient data to allow the derivation of a criterion, narrative descriptions of apparent threshold levels for acute and/or chronic effects based on the available data are presented. These descriptions are intended to convey a sense of the degree of toxicity of the pollutant in the absence of a criterion recommendation.

Summary of the Aquatic Life Guidelines

The Guidelines for Deriving Water Quality Criteria for the Protection of Aquatic Life and its Uses were developed to describe an objective, internally consistent, and appropriate way of ensuring that water quality criteria for aquatic life would provide, on the average, a reasonable amount of protection without an unreasonable amount of overprotection or underprotection. The resulting criteria are not intended to provide 100 percent protection of all species and all uses of aquatic life all of the time, but they are intended to protect most species in a balanced, healthy aquatic community. The Guidelines are published as Appendix B of this Notice. Responses to public comments on these Guidelines are attached as Appendix D.

Minimum data requirements are identified in four areas: acute toxicity to animals (eight data points), chronic toxicity to animals (three data points), toxicity to plants, and residues. Guidance is also given for discarding poor quality data.

Data on acute toxicity are needed for a variety of fish and invertebrate species and are used to derive a Final Acute Value. By taking into account the number and relative sensitivities of the tested species, the Final Acute Value is designed to protect most, but not necessarily all, of the tested and untested species.

Data on chronic toxicity to animals can be used to derive a Final Chronic Value by two different means. If chronic values are available for a specified number and array of species, a final

chronic value can be calculated directly. If not, an acute-chronic ratio is derived and then used with the Final Acute Value to obtain the Final Chronic Value.

The Final Plant Value is obtained by selecting the lowest plant toxicity value based on measured concentrations.

The Final Residue Value is intended to protect wildlife which consume aquatic organisms and the marketability of aquatic organisms. Protection of the marketability of aquatic organisms is, in actuality, protection of a use of that water body ("commercial fishery"). Two kinds of data are necessary to calculate the Final Residue Value: a bioconcentration factor (BCF) and a maximum permissible tissue concentration, which can be an FDA action level or can be the result of a chronic wildlife feeding study. For lipid soluble pollutants, the BCF is normalized for percent lipids and then the Final Residue Value is calculated by dividing the maximum permissible tissue concentration by the normalized BCF and by an appropriate percent lipid value. BCFs are normalized for percent lipids since the BCF measured for any individual aquatic species is generally proportional to the percent lipids in that species.

If sufficient data are available to demonstrate that one or more of the final values should be related to a water quality characteristic, such as salinity, hardness, or suspended solids, the final value(s) are expressed as a function of that characteristic.

After the four final values (Final Acute Value, Final Chronic Value, Final Plant Value, and Final Residue Value) have been obtained, the criterion is established with the Final Acute Value becoming the maximum value and the lowest of the other three values becoming the 24-hour average value. All of the data used to calculate the four final values and any additional pertinent information are then reviewed to determine if the criterion is reasonable. If sound scientific evidence indicates that the criterion should be raised or lowered, appropriate changes are made as necessary.

The present Guidelines have been revised from the earlier published versions (43 FR 21506, May 18, 1978; 43 FR 29028, July 5, 1978; 44 FR 15926, March 15, 1979). Details have been added in many places and the concept of a minimum data base has been incorporated. In addition, three adjustment factors and the species sensitivity factor have been deleted. These modifications were the result of the Agency's analysis of public comments and comments received from the Science Advisory Board on earlier

versions of the Guidelines. These comments and the Resultant modifications are addressed fully in Appendix D to this notice.

Criteria for the Protection of Human Health

Interpretation of the Human Health Criteria

The human health criteria issued today are summarized in Appendix A of this Federal Register notice. Criteria for the protection of human health are presented for 62 of the 65 pollutants based on their carcinogenic, toxic, or organoleptic (taste and odor) properties. The meanings and practical uses of the criteria values are distinctly different depending on the properties on which they are based.

The objective of the health assessment portions of the criteria documents is to estimate ambient water concentrations which, in the case of non-carcinogens, prevent adverse health effects in humans, and in the case of suspect or proven carcinogens, represent various levels of incremental cancer risk.

Health assessments typically contain discussions of four elements: Exposure, pharmacokinetics, toxic effects, and criterion formulation.

The exposure section summarizes information on exposure routes: ingestion directly from water, indirectly from consumption of aquatic organisms found in ambient water, other dietary sources, inhalation, and dermal contact. Exposure assumptions are used to derive human health criteria. Most criteria are based solely on exposure from consumption of water containing a specified concentration of a toxic pollutant and through consumption of aquatic organisms which are assumed to have bioconcentrated pollutants from the water in which they live. Other multimedia routes of exposure such as air, non-aquatic diet, or dermal are not factored into the criterion formulation for the vast majority of pollutants due to lack of data. The criteria are calculated using the combined aquatic exposure pathway and also using the aquatic organism ingestion exposure route alone. In criteria reflecting both the water consumption and aquatic organism ingestion routes of exposure, the relative exposure contribution varies with the propensity of a pollutant to bioconcentrate, with the consumption of aquatic organisms becoming more important as the bioconcentration factor (BCF) increases. As additional information on total exposure is assembled for pollutants for which criteria reflect only the two specified

aquatic exposure routes, adjustments in water concentration values may be made. The Agency intends to publish guidance which will permit the States to identify significantly different exposure patterns for their populations. If warranted by the demonstration of significantly different exposure patterns, this will become an element of a process to adapt/modify human health-based criteria to local conditions, somewhat analogous to the aquatic life criteria modification process discussed previously. It is anticipated that States at their discretion will be able to set appropriate human health criteria based on this process.

The pharmacokinetics section reviews data on absorption, distribution, metabolism, and excretion to assess the biochemical fate of the compounds in the human and animal system. The toxic effects section reviews data on acute, subacute, and chronic toxicity, synergistic and antagonistic effects, and specific information on mutagenicity, teratogenicity, and carcinogenicity. From this review, the toxic effect to be protected against is identified taking into account the quality, quantity, and weight of evidence characteristic of the data. The criterion formulation section reviews the highlights of the text and specifies a rationale for criterion development and the mathematical derivation of the criterion number.

Within the limitations of time and resources, current published information of significance was incorporated into the human health assessments. Review articles and reports were used for data evaluation and synthesis. Scientific judgment was exercised in reviewing and evaluating the data in each criteria document and in identifying the adverse effects for which protective criteria were published.

Specific health-based criteria are developed only if a weight of evidence supports the occurrence of the toxic effect and if dose/response data exist from which criteria can be estimated.

Criteria for suspect or proven carcinogens are presented as concentrations in water associated with a range of incremental cancer risks to man. Criteria for non-carcinogens represent levels at which exposure to a single chemical is not anticipated to produce adverse effects in man. In a few cases, organoleptic (taste and odor) data form the basis for the criterion. While this type of criterion does not represent a value which directly affects human health, it is presented as an estimate of the level of a pollutant that will not produce unpleasant taste or odor either directly from water consumption or indirectly by consumption of aquatic

organisms found in ambient waters. Criterion developed in this manner is judged to be as useful as other types of criteria in protecting designated water uses. In addition, where data are available, toxicity-based criteria are also presented for pollutants with derived organoleptic criteria. The choice of criteria used in water quality standards for these pollutants will depend upon the designated use to be protected. In the case of a multiple use water body, the criterion protecting the most sensitive use will be applied. Finally, for several pollutants no criteria are recommended due to a lack of information sufficient for quantitative criterion formulation.

Risk Extrapolation

Because methods do not now exist to establish the presence of a threshold carcinogenic effects, EPA's policy is that there is no scientific basis for estimating "safe" levels for carcinogens. The criteria for carcinogens, therefore, state that the recommended concentration maximum protection of human health is zero. In addition, the Agency has presented a range of concentrations corresponding to incremental cancer risks of 10^{-7} to 10^{-9} (one additional case of cancer in populations ranging from ten million to 100,000, respectively). Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Summary of the Human Health Guidelines

The health assessments and corresponding criteria published today were derived based on *Guidelines on Methodology Used in the Preparation of Health Effect Assessment Chapters of the Consent Decree Water Criteria Documents* (the Guidelines) developed by EPA's Office of Research and Development. The estimation of health risks associated with human exposure to environmental pollutants requires predicting the effect of low doses for a lifetime in duration. A combination of epidemiological and animal dose/response data is considered the preferred basis for quantitative criterion derivation. The complete Guidelines are presented as Appendix C. Major issues associated with these Guidelines and responses to public comments are presented as Appendix E.

No-effect (non-carcinogen) or specified risk (carcinogen) concentrations were estimated by extrapolation from animal toxicity or

human epidemiology studies using the following basic exposure assumptions: a 70-kilogram male person (*Report of the Task Group on Reference Man*, International Commission for Radiation Protection, November 23, 1957) as the exposed individual; the average daily consumption of freshwater and estuarine fish and shellfish products equal to 6.5 grams/day; and the average ingestion of two liters/day of water (*Drinking Water and Health*, National Academy of Sciences, National Research Council, 1977). Criteria based on these assumptions are estimated to be protective of an adult male who experiences average exposure conditions.

Two basic methods were used to formulate health criteria, depending on whether the prominent adverse effect was cancer or other toxic manifestations. The following sections detail these methods.

Carcinogens

Extrapolation of cancer responses from high to low doses and subsequent risk estimation from animal data is performed using a linearized multi-stage model. This procedure is flexible enough to fit all monotonically-increasing dose response data, since it incorporates several adjustable parameters. The multi-stage model is a linear non-threshold model as was the "one-hit" model originally used in the proposed criteria documents. The linearized multi-stage model and its characteristics are described fully in Appendix C. The linear non-threshold concept has been endorsed by the four agencies in the Interagency Regulatory Liaison Group and is less likely to underestimate risk at the low doses typical of environmental exposure than other models that could be used. Because of the uncertainties associated with dose response, animal-to-human extrapolation and other unknown factors, because of the use of average exposure assumptions, and because of the serious public health consequences that could result if risk were underestimated, EPA believes that it is prudent to use conservative methods to estimate risk in the water quality criteria program. The linearized multistage model is more systematic and invokes fewer arbitrary assumptions than the "one-hit" procedure previously used.

It should be noted that extrapolation models provide estimates of risk since a variety of assumptions are built into any model. Models using widely different assumptions may produce estimates ranging over several orders of magnitude. Since there is at present no

way to demonstrate the scientific validity of any model, the use of risk extrapolation models is a subject of debate in the scientific community. However, risk extrapolation is generally recognized as the only tool available at this time for estimating the magnitude of health hazards associated with non-threshold toxicants and has been endorsed by numerous Federal agencies and scientific organizations, including EPA's Carcinogen Assessment Group, the National Academy of Sciences, and the Interagency Regulatory Liaison Group as a useful means of assessing the risks of exposure to various carcinogenic-pollutants.

Non-Carcinogens

Health criteria based on toxic effects of pollutants other than carcinogenicity are estimates of concentrations which are not expected to produce adverse effects in humans. They are based upon Acceptable Daily Intake (ADI) levels and are generally derived using no-observed-adverse-effect-level (NOAEL), data from animal studies although human data are used wherever available. The ADI is calculated using safety factors to account for uncertainties inherent in extrapolation from animal to man. In accordance with the National Research Council recommendations (*Drinking Water and Health*, National Academy of Sciences, National Research Council, 1977), safety factors of 10, 100, or 1,000 are used depending on the quality and quantity of data. In some instances extrapolations are made from inhalation studies or limits to approximate a human response from ingestion using the Stokinger-Woodward model (Journal of American Water Works Association, 1958). Calculations of criteria from ADIs are made using the standard exposure assumptions (2 liters of water, 6.5 grams of edible aquatic products, and an average body weight of 70 kg).

Dated: October 24, 1980.

Douglas M. Costle,
Administrator.

Appendix A—Summary of Water Quality Criteria

Acenaphthene

Freshwater Aquatic Life

The available data for acenaphthene indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 1,700 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of acenaphthene to sensitive freshwater aquatic animals but

toxicity to freshwater algae occur at concentrations as low as 520 µg/l.

Saltwater Aquatic Life

The available data for acenaphthene indicate that acute and chronic toxicity to saltwater aquatic life occur at concentrations as low as 870 and 710 µg/l, respectively, and would occur at lower concentrations among species that are more sensitive than those tested. Toxicity to algae occurs at concentrations as low as 500 µg/l.

Human Health

Sufficient data is not available for acenaphthene to derive a level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 20 µg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no demonstrated relationship to potential adverse human health effects.

Acrolein

Freshwater Aquatic Life

The available data for acrolein indicate that acute and chronic toxicity to freshwater aquatic life occurs at concentrations as low as 68 and 21 µg/l, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

The available data for acrolein indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 55 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of acrolein to sensitive saltwater aquatic life.

Human Health

For the protection of human health from the toxic properties of acrolein ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 320 µg/l.

For the protection of human health from the toxic properties of acrolein ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 780 µg/l.

Acrylonitrile

Freshwater Aquatic Life

The available data for acrylonitrile indicate that acute toxicity to freshwater aquatic life occurs at concentrations as

low as 7,350 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No definitive data are available concerning the chronic toxicity of acrylonitrile to sensitive freshwater aquatic life but mortality occurs at concentrations as low as 2,600 µg/l with a fish species exposed for 30 days.

Saltwater Aquatic Life

Only one saltwater species has been tested with acrylonitrile and no statement can be made concerning acute or chronic toxicity.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of acrylonitrile through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are .58 µg/l, .058 µg/l and .006 µg/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 6.5 µg/l, .65 µg/l and .065 µg/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Aldrin-Dieldrin

Dieldrin

Freshwater Aquatic Life

For dieldrin the criterion to protect fresh water aquatic life as derived using the Guidelines is 0.0019 µg/l as a 24-hour average and the concentration should not exceed 2.5 µg/l at any time.

Saltwater Aquatic Life

For dieldrin the criterion to protect saltwater aquatic life as derived using the Guidelines is 0.0019 µg/l as a 24-hour average and the concentration should not exceed 0.71 µg/l at any time.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of dieldrin through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold

assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 71 ng/l, .071 ng/l, and .0071 ng/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are .76 ng/l, .076 ng/l, and .0076 ng/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Aldrin

Freshwater Aquatic Life

For freshwater aquatic life the concentration of aldrin should not exceed 3.0 µg/l at any time. No data are available concerning the chronic toxicity of aldrin to sensitive freshwater aquatic life.

Saltwater Aquatic Life

For saltwater aquatic life the concentration of aldrin should not exceed 1.3 µg/l at any time. No data are available concerning the chronic toxicity of aldrin to sensitive saltwater aquatic life.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of aldrin through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are .74 ng/l, .074 ng/l, and .0074 ng/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are .79 ng/l, .079 ng/l, and .0079 ng/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Antimony

Freshwater Aquatic Life

The available data for antimony indicate that acute and chronic toxicity to freshwater aquatic life occur at concentrations as low as 8,000 and 1.6 µg/l, respectively, and would occur at lower concentrations among species that are more sensitive than those tested. Toxicity to algae occurs at concentrations as low as 610 µg/l.

Saltwater Aquatic Life

No saltwater organisms have been adequately tested with antimony, and no statement can be made concerning acute or chronic toxicity.

Human Health

For the protection of human health from the toxic properties of antimony ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 146 µg/l.

For the protection of human health from the toxic properties of antimony ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 45,000 µg/l.

Arsenic

Freshwater Aquatic Life

For freshwater aquatic life the concentration of total recoverable trivalent inorganic arsenic should not exceed 440 µg/l at any time. Short-term effects on embryos and larvae of aquatic vertebrate species have been shown occur at concentrations as low as 40 µg/l.

Saltwater Aquatic Life

The available data for total recoverable trivalent inorganic arsenic indicate that acute toxicity to saltwater aquatic life occurs at concentrations low as 508 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of trivalent inorganic arsenic to sensitive saltwater aquatic life.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of arsenic through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are

estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 22 ng/l, 2.2 ng/l, and .22 ng/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 175 ng/l, 17.5 ng/l, and 1.75 ng/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Asbestos

Freshwater Aquatic Life

No freshwater organisms have been tested with any asbestiform mineral and no statement can be made concerning acute or chronic toxicity.

Saltwater Aquatic Life

No saltwater organisms have been tested with any asbestiform mineral and no statement can be made concerning acute or chronic toxicity.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of asbestos through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 300,000 fibers/1,30,000 fibers/l, and 3,000 fibers/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Benzene

Freshwater Aquatic Life

The available data for benzene indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 5,300 $\mu\text{g/l}$ and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of benzene to sensitive freshwater aquatic life.

Saltwater Aquatic Life

The available data for benzene indicate that acute toxicity to saltwater aquatic life occurs at concentrations as

low as 5,100 $\mu\text{g/l}$ and would occur at lower concentrations among species that are more sensitive than those tested. No definitive data are available concerning the chronic toxicity of benzene to sensitive saltwater aquatic life, but adverse effects occur at concentrations as low as 700 $\mu\text{g/l}$ with a fish species exposed for 168 days.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of benzene through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 6.6 $\mu\text{g/l}$, .86 $\mu\text{g/l}$, and .066 $\mu\text{g/l}$, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 400 $\mu\text{g/l}$, 40.0 $\mu\text{g/l}$, and 4.0 $\mu\text{g/l}$, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Benzidine

Freshwater Aquatic Life

The available data for benzidine indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 2,500 $\mu\text{g/l}$ and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of benzidine to sensitive freshwater aquatic life.

Saltwater Aquatic Life

No saltwater organisms have been tested with benzidine and no statement can be made concerning acute and chronic toxicity.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of benzidine through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of

cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 1.2 ng/l, .12 ng/l, and .01 ng/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 5.3 ng/l, .53 ng/l, and .05 ng/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Beryllium

Freshwater Aquatic Life

The available data for beryllium indicate that acute and chronic toxicity to freshwater aquatic life occurs at concentrations as low as 130 and 5.3 $\mu\text{g/l}$, respectively, and would occur at lower concentrations among species that are more sensitive than those tested. Hardness has a substantial effect on acute toxicity.

Saltwater Aquatic Life

The limited saltwater data base available for beryllium does not permit any statement concerning acute or chronic toxicity.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of beryllium through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 37 ng/l, 3.7 ng/l, and .37 ng/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 641 ng/l, 64.1 ng/l, and 6.41 ng/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Cadmium

Freshwater Aquatic Life

For total recoverable cadmium the criterion (in $\mu\text{g/l}$) to protect freshwater aquatic life as derived using the Guidelines is the numerical value given

by $c(1-c)(\text{hardness})^{-0.63}$ as a 24-hour average and the concentration (in $\mu\text{g/l}$) should not exceed the numerical value given by $c(1-c)(\text{hardness})^{-0.73}$ at any time. For example, hardnesses of 50, 100, and 200 mg/l as CaCO_3 , the criteria are 0.012, 0.025, and 0.051 $\mu\text{g/l}$, respectively, and the concentration of total recoverable cadmium should not exceed 1.5, 3.0 and 6.3 $\mu\text{g/l}$, respectively, at any time.

Saltwater Aquatic Life

For total recoverable cadmium the criterion to protect saltwater aquatic life as derived using the Guidelines is 4.5 $\mu\text{g/l}$ as a 24-hour average and the concentration should not exceed 50 $\mu\text{g/l}$ at any time.

Human Health

The ambient water quality criterion for cadmium is recommended to be identical to the existing drinking water standard which is 10 $\mu\text{g/l}$. Analysis of the toxic effects data resulted in a calculated level which is protective of human health against the ingestion of contaminated water and contaminated aquatic organisms. The calculated value is comparable to the present standard. For this reason a selective criterion based on exposure solely from consumption of 6.5 grams of aquatic organisms was not derived.

Carbon Tetrachloride

Freshwater Aquatic Life

The available data for carbon tetrachloride indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 35,200 $\mu\text{g/l}$ and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of carbon tetrachloride to sensitive freshwater aquatic life.

Saltwater Aquatic Life

The available data for carbon tetrachloride indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 50,000 $\mu\text{g/l}$ and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of carbon tetrachloride to sensitive saltwater aquatic life.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of carbon tetrachloride through ingestion of contaminated water and contaminated aquatic organisms the ambient water concentration should be zero based on

the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-4} , and 10^{-2} . The corresponding criteria are 4.0 $\mu\text{g/l}$, 40 $\mu\text{g/l}$, and 400 $\mu\text{g/l}$, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 89.4 $\mu\text{g/l}$, 8.94 $\mu\text{g/l}$, and .89 $\mu\text{g/l}$, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Chlordane

Freshwater Aquatic Life

For chlordane the criterion to protect freshwater aquatic life as derived using the Guidelines is 0.0043 $\mu\text{g/l}$ as a 24-hour average and the concentration should not exceed 2.4 $\mu\text{g/l}$ at any time.

Saltwater Aquatic Life

For chlordane the criterion to protect saltwater aquatic life as derived using the Guidelines is 0.0040 $\mu\text{g/l}$ as a 24-hour average and the concentration should not exceed 0.09 $\mu\text{g/l}$ at any time.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of chlordane through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-4} , and 10^{-2} . The corresponding criteria are 4.8 ng/l , .48 ng/l , and .048 ng/l , respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 4.8 ng/l , .48 ng/l , and .048 ng/l , respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Chlorinated Benzenes

Freshwater Aquatic Life

The available data for chlorinated benzenes indicate that acute toxicity to freshwater aquatic life occurs at

concentrations as low as 250 $\mu\text{g/l}$ and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of the more toxic of the chlorinated benzene to sensitive freshwater aquatic life but toxicity occurs at concentrations as low as 50 $\mu\text{g/l}$ for a fish species exposed for 7.5 days.

Saltwater Aquatic Life

The available data for chlorinated benzenes indicate that acute and chronic toxicity to saltwater aquatic life occur at concentrations as low as 160 and 129 $\mu\text{g/l}$, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of hexachlorobenzene through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-4} , and 10^{-2} . The corresponding recommended criteria are 7.2 ng/l , .72 ng/l , and .072 ng/l , respectively. If the above estimates are made for consumption of aquatic organisms on excluding consumption of water, the levels are 7.4 ng/l , .74 ng/l , and .074 ng/l , respectively.

For the protection of human health from the toxic properties of 1,2,4,5-tetrachlorobenzene ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 38 $\mu\text{g/l}$.

For the protection of human health from the toxic properties of 1,2,4,5-tetrachlorobenzene ingested through contaminated aquatic organisms also, the ambient water criterion is determined to be 48 $\mu\text{g/l}$.

For the protection of human health from the toxic properties of pentachlorobenzene ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 74 $\mu\text{g/l}$.

For the protection of human health from the toxic properties of pentachlorobenzene ingested through contaminated aquatic organisms also, the ambient water criterion is determined to be 85 $\mu\text{g/l}$.

Using the present guidelines, a satisfactory criterion cannot be derived

at this time due to the insufficiency in the available data for trichlorobenzene.

For comparison purposes, two approaches were used to derive criterion levels for monochlorobenzene. Based on available toxicity data, for the protection of public health, the derived level is 488 µg/l. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 20 µg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no demonstrated relationship to potential adverse human health effects.

Chlorinated Ethanes

Freshwater Aquatic Life

The available freshwater data for chlorinated ethanes indicate that toxicity increases greatly with increasing chlorination, and that acute toxicity occurs at concentrations as low as 118,000 µg/l for 1,2-dichloroethane, 18,000 µg/l for two trichloroethanes, 9,320 µg/l for two tetrachloroethanes, 7,240 µg/l for pentachloroethane, and 980 µg/l for hexachloroethane. Chronic toxicity occurs at concentrations as low as 20,000 µg/l for 1,2-dichloroethane, 9,400 µg/l for 1,1,2-trichloroethane, 2,400 µg/l for 1,1,2,2-tetrachloroethane, 1,100 µg/l for pentachloroethane, and 540 µg/l for hexachloroethane. Acute and chronic toxicity would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

The available saltwater data for chlorinated ethanes indicate that toxicity increases greatly with increasing chlorination and that acute toxicity to fish and invertebrate species occurs at concentrations as low as 113,000 µg/l for 1,2-dichloroethane, 31,200 µg/l for 1,1,1-trichloroethane, 9,020 µg/l for 1,1,2,2-tetrachloroethane, 390 µg/l for pentachloroethane, and 940 µg/l for hexachloroethane. Chronic toxicity occurs at concentrations as low as 281 µg/l for pentachloroethane. Acute and chronic toxicity would occur at lower concentrations among species that are more sensitive than those tested.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of 1,2-dichloroethane through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this

chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 8.4 µg/l, 84 µg/l, and 0.084 µg/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 2,430 µg/l, 243 µg/l, and 24.3 µg/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

For the protection of human health from the toxic properties of 1,1,1-trichloroethane ingested through water and contaminated aquatic organism, the ambient water criterion is determined to be 18.4 mg/l.

For the protection of human health from the toxic properties of 1,1,1-trichloroethane ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 1.03 g/l.

For the maximum protection of human health from the potential carcinogenic effects due to exposure of 1,1,2-trichloroethane through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time.

Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 6.0 µg/l, 6 µg/l, and 0.6 µg/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 418 µg/l, 41.8 µg/l, and 4.18 µg/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

For the maximum protection of human health from the potential carcinogenic effects due to exposure of 1,1,2,2-tetrachloroethane through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} ,

and 10^{-4} . The corresponding criteria are 1.7 µg/l, .17 µg/l, and .017 µg/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 107 µg/l, 10.7 µg/l, and 1.07 µg/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

For the maximum protection of human health from the potential carcinogenic effects due to exposure of hexachloroethane through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time.

Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 19 µg/l, 1.9 µg/l, and .19 µg/l, respectively. If the above estimates are made for consumption of aquatic

organisms only, excluding consumption of water, the levels are 87.4 µg/l, 8.74 µg/l, and .87 µg/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Using the present guidelines, a satisfactory criterion cannot be derived at this time due to the insufficiency in the available data for monochloroethane.

Using the present guidelines, a satisfactory criterion cannot be derived at this time due to the insufficiency in the available data for 1,1-dichloroethane.

Using the present guidelines, a satisfactory criterion cannot be derived at this time due to the insufficiency in the available data for 1,1,1,2-tetrachloroethane.

Using the present guidelines, a satisfactory criterion cannot be derived at this time due to the insufficiency in the available data for pentachloroethane.

Chlorinated Naphthalenes

Freshwater Aquatic Life

The available data for chlorinated naphthalenes indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 1,800 µg/l and would occur at lower concentrations among species that are

more sensitive than those tested. No data are available concerning the chronic toxicity of chlorinated naphthalenes to sensitive freshwater aquatic life.

Saltwater Aquatic Life

The available data for chlorinated naphthalenes indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 7.5 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of chlorinated naphthalenes to sensitive saltwater aquatic life.

Human Health

Using the present guidelines, a satisfactory criterion cannot be derived at this time due to the insufficiency in the available data for chlorinated naphthalenes.

Chlorinated Phenols

Freshwater Aquatic Life

The available freshwater data for chlorinated phenols indicate that toxicity generally increases with increasing chlorination, and that acute toxicity occurs at concentrations as low as 30 µg/l for 4-chloro-3-methylphenol to greater than 500,000 µg/l for other compounds. Chronic toxicity occurs at concentrations as low as 970 µg/l for 2,4,6-trichlorophenol. Acute and chronic toxicity would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

The available saltwater data for chlorinated phenols indicate that toxicity generally increases with increasing chlorination and that acute toxicity occurs at concentrations as low as 440 µg/l for 2,3,5,6-tetrachlorophenol and 29,700 µg/l for 4-chlorophenol. Acute toxicity would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of chlorinated phenols to sensitive saltwater aquatic life.

Human Health

Sufficient data is not available for 3-monochlorophenol to derive a level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 0.1 µg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no

demonstrated relationship to potential adverse human health effects.

Sufficient data is not available for 4-monochlorophenol to derive a level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 0.1 µg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no demonstrated relationship to potential adverse human health effects.

Sufficient data is not available for 2,3-dichlorophenol to derive a level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is .04 µg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no demonstrated relationship to potential adverse human health effects.

Sufficient data is not available for 2,5-dichlorophenol to derive a level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is .5 µg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no demonstrated relationship to potential adverse human health effects.

Sufficient data is not available for 2,6-dichlorophenol to derive a level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is .2 µg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no demonstrated relationship to potential adverse human health effects.

Sufficient data is not available for 3,4-dichlorophenol to derive a level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is .3 µg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no demonstrated relationship to potential adverse human health effects.

Sufficient data is not available for 2,3,4,6-tetrachlorophenol to derive a

level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 1 µg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no demonstrated relationship to potential adverse human health effects.

For comparison purposes, two approaches were used to derive criterion levels for 2,4,5-trichlorophenol. Based on available toxicity data, for the protection of public health, the derived level is 2.6 mg/l. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 1.0 µg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no demonstrated relationship to potential adverse human health effects.

For the maximum protection of human health from the potential carcinogenic effects due to exposure of 2,4,6-trichlorophenol through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-5} , 10^{-6} , and 10^{-7} . The corresponding criteria are 12 µg/l, 1.2 µg/l, and .12 µg/l respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 36 µg/l, 3.6 µg/l, and .36 µg/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimate level is 2 µg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criterion have limitations and have no demonstrated relationship to potential adverse human health effects.

Sufficient data is not available for 2-methyl-4-chlorophenol to derive a level which would protect against any potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimate level is 1800 µg/l. It should be

recognized that organoleptic data as a basis for establishing a water quality criterion have limitations and have no demonstrated relationship to potential adverse human health effects.

Sufficient data is not available for 3-methyl-4-chlorophenol to derive a level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 3000 µg/L. It should be recognized that organoleptic data as a basis for establishing a water quality criterion have limitations and have no demonstrated relationship to potential adverse human health effects.

Sufficient data is not available for 3-methyl-6-chlorophenol to derive a level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 20 µg/L. It should be recognized that organoleptic data as a basis for establishing a water quality criterion have limitations and have no demonstrated relationship to potential adverse human health effects.

Chloroalkyl Ethers

Freshwater Aquatic Life

The available data for chloroalkyl ethers indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 238,000 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No definitive data are available concerning the chronic toxicity of chloroalkyl ethers to sensitive freshwater aquatic life.

Saltwater Aquatic Life

No saltwater organisms have been tested with any chloroalkyl ether and no statement can be made concerning acute and chronic toxicity.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of bis-(chloromethyl)-ether through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are .038 ng/L, .0038 ng/L and .00038 ng/L, respectively.

If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 18.4 ng/L, 1.84 ng/L, and .184 ng/L, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

For the maximum protection of human health from the potential carcinogenic effects due to exposure of bis (2-chloroethyl) ether through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are .3 µg/L, .03 µg/L, and .003 µg/L, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 13.6 µg/L, 1.36 µg/L, and .136 µg/L, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

For the protection of human health from the toxic properties of bis (2-chloroisopropyl) ether ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 34.7 µg/L.

For the protection of human health from the toxic properties of bis (2-chloroisopropyl) ether ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 4.36 mg/L.

Chloroform

Freshwater Aquatic Life

The available data for chloroform indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 28,900 µg/L and would occur at lower concentrations among species that are more sensitive than the three tested species. Twenty-seven-day LC50 values indicate that chronic toxicity occurs at concentrations as low as 1,240 µg/L and could occur at lower concentrations among species or other life stages that are more sensitive than the earliest life cycle stage of the rainbow trout.

Saltwater Aquatic Life

The data base for saltwater species is limited to one test and no statement can be made concerning acute or chronic toxicity.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of chloroform through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 1.90 µg/L, .19 µg/L, and .019 µg/L, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 157 µg/L, 15.7 µg/L, and 1.57 µg/L, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

2-Chlorophenol

Freshwater Aquatic Life

The available data for 2-chlorophenol indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 4,380 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No definitive data are available concerning the chronic toxicity of 2-chlorophenol to sensitive freshwater aquatic life but flavor impairment occurs in one species of fish at concentrations as low as 2,000 µg/l.

Saltwater Aquatic Life

No saltwater organisms have been tested with 2-chlorophenol and no statement can be made concerning acute and chronic toxicity.

Human Health

Sufficient data is not available for 2-chlorophenol to derive a level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 0.1 µg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no

demonstrated relationship to potential adverse human health effects.

Chromium

Freshwater Aquatic Life

For total recoverable hexavalent chromium the criterion to protect freshwater aquatic life as derived using the Guidelines is 0.29 µg/l as a 24-hour average and the concentration should not exceed 21 µg/l at any time.

For freshwater aquatic life the concentration (in µg/l) of total recoverable trivalent chromium should not exceed the numerical value given by " $e^{(1.08[\ln(\text{hardness})] + 3.48)}$ " at any time. For example, at hardnesses of 50, 100 and 200 mg/l as CaCO₃, the concentration of total recoverable trivalent chromium should not exceed 2,200, 4,700, and 9,900 µg/l, respectively, at any time. The available data indicate that chronic toxicity to freshwater aquatic life occurs at concentrations as low as 44 µg/l and would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

For total recoverable hexavalent chromium the criterion to protect saltwater aquatic life as derived using the Guidelines is 18 µg/l as a 24-hour average and the concentration should not exceed 1,260 µg/l at any time.

For total recoverable trivalent chromium, the available data indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 10,300 µg/l, and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of trivalent chromium to sensitive saltwater aquatic life.

Human Health

For the protection of human health from the toxic properties of Chromium III ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 170 mg/l.

For the protection of human health from the toxic properties of Chromium III ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 3433 mg/l.

The ambient water quality criterion for total Chromium VI is recommended to be identical to the existing drinking water standard which is 50 µg/l. Analysis of the toxic effects data resulted in a calculated level which is protective of human health against the ingestion of contaminated water and contaminated aquatic organisms. The

calculated value is comparable to the present standard. For this reason a selective criterion based on exposure solely from consumption of 6.5 grams of aquatic organisms was not derived.

Copper

Freshwater Aquatic Life

For total recoverable copper the criterion to protect freshwater aquatic life as derived using the Guidelines is 5.6 µg/l as a 24-hour average and the concentration (in µg/l) should not exceed the numerical value given by $e^{(0.94[\ln(\text{hardness})] - 1.23)}$ at any time. For example, at hardnesses of 50, 100, and 200 mg/l CaCO₃, the concentration of total recoverable copper should not exceed 12, 22, and 43 µg/l at any time.

Saltwater Aquatic Life

For total recoverable copper the criterion to protect saltwater aquatic life as derived using the Guidelines is 4.0 µg/l as a 24-hour average and the concentration should not exceed 23 µg/l at any time.

Human Health

Sufficient data is not available for copper to derive a level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 1 mg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no demonstrated relationship to potential adverse human health effects.

Cyanide

Freshwater Aquatic Life

For free cyanide (sum of cyanide present as HCN and CN⁻, expressed as CN) the criterion to protect freshwater aquatic life as derived using the Guidelines is 3.5 µg/l as a 24-hour average and the concentration should not exceed 52 µg/l at any time.

Saltwater Aquatic Life

The available data for free cyanide (sum of cyanide present as HCN and CN⁻, expressed as CN) indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 30 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. If the acute-chronic ratio for saltwater organisms is similar to that for freshwater organisms, chronic toxicity would occur at concentrations as low as 2.0 µg/l for the tested species and at lower concentrations among species

that are more sensitive than those tested.

Human Health

The ambient water quality criterion for cyanide is recommended to be identical to the existing drinking water standard which is 200 µg/l. Analysis of the toxic effects data resulted in a calculated level which is protective of human health against the ingestion of contaminated water and contaminated aquatic organisms. The calculated value is comparable to the present standard. For this reason a selective criterion based on exposure solely from consumption of 6.5 grams of aquatic organisms was not derived.

DDT and Metabolites

Freshwater Aquatic Life

DDT

For DDT and its metabolites the criterion to protect freshwater aquatic life as derived using the Guidelines is 0.0010 µg/l as a 24-hour average and the concentration should not exceed 1.1 µg/l at any time.

TDE

The available data for TDE indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 0.6 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of TDE to sensitive freshwater aquatic life.

DDE

The available data for DDE indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 1,050 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of DDE to sensitive freshwater aquatic life.

Saltwater Aquatic Life

DDT

For DDT and its metabolites the criterion to protect saltwater aquatic life as derived using the Guidelines is 0.0010 µg/l as a 24-hour average and the concentration should not exceed 0.13 µg/l at any time.

TDE

The available data for TDE indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 3.6 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the

chronic toxicity of TDE to sensitive saltwater aquatic life.

DDE

The available data for DDE indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 14 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of DDE to sensitive saltwater aquatic life.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of DDT through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 24 ng/l, .024 ng/l, and .0024 ng/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 24 ng/l, .024 ng/l, and .0024 ng/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment of an "acceptable" risk level.

Dichlorobenzenes

Freshwater Aquatic Life

The available data for dichlorobenzenes indicate that acute and chronic toxicity to freshwater aquatic life occurs at concentrations as low as 1,120 and 763 µg/l, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

The available data for dichlorobenzenes indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 1,970 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of dichlorobenzenes to sensitive saltwater aquatic life.

Human Health

For the protection of human health from the toxic properties of dichlorobenzenes (all isomers) ingested

through water and contaminated aquatic organisms, the ambient water criterion is determined to be 400 µg/l.

For the protection of human health from the toxic properties of dichlorobenzenes (all isomers) ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 2.6 mg/l.

Dichlorobenzidines

Freshwater Aquatic Life

The data base available for dichlorobenzidines and freshwater organisms is limited to one test on bioconcentration of 3,3'-dichlorobenzidine and no statement can be made concerning acute or chronic toxicity.

Saltwater Aquatic Life

No saltwater organisms have been tested with any dichlorobenzidine and no statement can be made concerning acute or chronic toxicity.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of dichlorobenzidine through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are .103 µg/l, .0103 µg/l, and .00103 µg/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are .204 µg/l, .0204 µg/l, and .00204 µg/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Dichloroethylenes

Freshwater Aquatic Life

The available data for dichloroethylenes indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 11,600 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No definitive data are available concerning the chronic toxicity of dichloroethylenes to sensitive freshwater aquatic life.

Saltwater Aquatic Life

The available data for dichloroethylenes indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 224,000 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity dichloroethylenes to sensitive saltwater aquatic life.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of 1,1-dichloroethylene through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are .33 µg/l, .033 µg/l, and .0033 µg/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 18.5 µg/l, 1.85 µg/l, and .185 µg/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level. Using the present guidelines, a satisfactory criterion cannot be derived at this time due to the insufficiency in the available data for 1,2-dichloroethylene.

2,4-Dichlorophenol

Freshwater Aquatic Life

The available data for 2,4-dichlorophenol indicate that acute and chronic toxicity to freshwater aquatic life occurs at concentrations as low as 2,020 and 365 µg/l, respectively, and would occur at lower concentrations among species that are more sensitive than those tested. Mortality to early life stages of one species of fish occurs at concentrations as low as 70 µg/l.

Saltwater Aquatic Life

Only one test has been conducted with saltwater organisms on 2,4-dichlorophenol and no statement can be made concerning acute or chronic toxicity.

Human Health

For comparison purposes, two approaches were used to derive criterion levels for 2,4-dichlorophenol.

Based on available toxicity data, for the protection of public health, the derived level is 3.09 mg/L. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 0.3 µg/L. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no demonstrated relationship to potential adverse human health effects.

Dichloropropanes/Dichloropropenes

Freshwater Aquatic Life

The available data for dichloropropanes indicate that acute and chronic toxicity to freshwater aquatic life occurs at concentrations as low as 23,000 and 5,700 µg/L, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

The available data for dichloropropenes indicate that acute and chronic toxicity to freshwater aquatic life occurs at concentrations as low as 6,060 and 244 µg/L, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

The available data for dichloropropanes indicate that acute and chronic toxicity to saltwater aquatic life occurs at concentrations as low as 10,300 and 3,040 µg/L, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

The available data for dichloropropenes indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 790 µg/L, and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of dichloropropenes to sensitive saltwater aquatic life.

Human Health

Using the present guidelines, a satisfactory criterion cannot be derived at this time due to the insufficiency in the available data for dichloropropanes.

For the protection of human health from the toxic properties of dichloropropenes ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 87 µg/L.

For the protection of human health from the toxic properties of dichloropropenes ingested through contaminated aquatic organisms alone,

the ambient water criterion is determined to be 14.1 mg/L.

2,4-Dimethylphenol

Freshwater Aquatic Life

The available data for 2,4-dimethylphenol indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 2.120 µg/L and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of dimethylphenol to sensitive freshwater aquatic life.

Saltwater Aquatic Life

No saltwater organisms have been tested with 2,4-dimethylphenol and no statement can be made concerning acute and chronic toxicity.

Human Health

Sufficient data are not available for 2,4-dimethylphenol to derive a level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 400 µg/L. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no demonstrated relationship to potential adverse human health effects.

2,4-Dinitrotoluene

Freshwater Aquatic Life

The available data for 2,4-dinitrotoluene indicate that acute and chronic toxicity to freshwater aquatic life occurs at concentrations as low as 330 and 230 µg/L, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

The available data for 2,4-dinitrotoluenes indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 590 µg/L and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of 2,4-dinitrotoluenes to sensitive saltwater aquatic life but a decrease in algal cell numbers occurs at concentrations as low as 370 µg/L.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of 2,4-dinitrotoluene through ingestion of contaminated water and contaminated

aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in an incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 1.1 µg/L, 0.11 µg/L, and 0.011 µg/L, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 91 µg/L, 9.1 µg/L, and 0.91 µg/L, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

1,2-Diphenylhydrazine

Freshwater Aquatic Life

The available data for 1,2-diphenylhydrazine indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 270 µg/L and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of 1,2-diphenylhydrazine to sensitive freshwater aquatic life.

Saltwater Aquatic Life

No saltwater organisms have been tested with 1,2-diphenylhydrazine and no statement can be made concerning acute and chronic toxicity.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of 1,2-diphenylhydrazine through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in an incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 422 ng/L, 42 ng/L, and 4 ng/L, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 5.8 µg/L, 0.56 µg/L, and 0.056 µg/L, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not

represent an Agency judgment on an "acceptable" risk level.

Endosulfan

Freshwater Aquatic Life

For endosulfan the criterion to protect freshwater aquatic life as derived using the Guidelines is 0.056 µg/l as a 24-hour average and the concentration should not exceed 0.22 µg/l at any time.

Saltwater Aquatic Life

For endosulfan the criterion to protect saltwater aquatic life as derived using the Guidelines is 0.0087 µg/l as a 24-hour average and the concentration should not exceed 0.034 µg/l at any time.

Human Health

For the protection of human health from the toxic properties of endosulfan ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 74 µg/L.

For the protection of human health from the toxic properties of endosulfan ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 159 µg/L.

Endrin

Freshwater Aquatic Life

For endrin the criterion to protect freshwater aquatic life as derived using the Guidelines is 0.0023 µg/l as a 24-hour average and the concentration should not exceed 0.18 µg/l at any time.

Saltwater Aquatic Life

For endrin the criterion to protect saltwater aquatic life as derived using the Guidelines is 0.0023 µg/l as a 24-hour average and the concentration should not exceed 0.037 µg/l at any time.

Human Health

The ambient water quality criterion for endrin is recommended to be identical to the existing drinking water standard which is 1 µg/L. Analysis of the toxic effects data resulted in a calculated level which is protective of human health against the ingestion of contaminated water and contaminated aquatic organisms. The calculated value is comparable to the present standard. For this reason a selective criterion based on exposure solely from consumption of 6.5 grams of aquatic organisms was not derived.

Ethylbenzene

Freshwater Aquatic Life

The available data for ethylbenzene indicate that acute toxicity to freshwater

aquatic life occurs at concentrations as low as 32,000 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No definitive data are available concerning the chronic toxicity of ethylbenzene to sensitive freshwater aquatic life.

Saltwater Aquatic Life

The available data for ethylbenzene indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 430 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of ethylbenzene to sensitive saltwater aquatic life.

Human Health

For the protection of human health from the toxic properties of ethylbenzene ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 1.4 mg/L.

For the protection of human health from the toxic properties of ethylbenzene ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 3.28 mg/L.

Fluoranthene

Freshwater Aquatic Life

The available data for fluoranthene indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 3980 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of fluoranthene to sensitive freshwater aquatic life.

Saltwater Aquatic Life

The available data for fluoranthene indicate that acute and chronic toxicity to saltwater aquatic life occur at concentrations as low as 40 and 16 µg/L respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Human Health

For the protection of human health from the toxic properties of fluoranthene ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 42 µg/l.

For the protection of human health from the toxic properties of fluoranthene ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 54 µg/l.

Haloethers

Freshwater Aquatic Life

The available data for haloethers indicate that acute and chronic toxicity to freshwater aquatic life occur at concentrations as low as 360 and 122 µg/L respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

No saltwater organisms have been tested with any haloether and no statement can be made concerning acute or chronic toxicity.

Human Health

Using the present guidelines, a satisfactory criterion cannot be derived at this time due to the insufficiency in the available data for haloethers.

Halomethanes

Freshwater Aquatic Life

The available data for halomethanes indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 11,000 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of halomethanes to sensitive freshwater aquatic life.

Saltwater Aquatic Life

The available data for halomethanes indicate that acute and chronic toxicity to saltwater aquatic life occur at concentrations as low as 12,000 and 6,400 µg/L respectively, and would occur at lower concentrations among species that are more sensitive than those tested. A decrease in algal cell numbers occurs at concentrations as low as 11,500 µg/L.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of chloromethane, bromomethane, dichloromethane, bromodichloromethane, tribromomethane, dichlorodifluoromethane, trichlorofluoromethane, or combination of these chemicals through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in an incremental increase of cancer risk, if the lifetimes are estimated at 10^4 , 10^5 , and 10^6 . The corresponding criteria

1.9 µg/L, 0.19 µg/L, and 0.019 µg/L, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 157 µg/L, 15.7 µg/L, and 1.57 µg/L, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Heptachlor

Freshwater Aquatic Life

For heptachlor the criterion to protect freshwater aquatic life as derived using the Guidelines is 0.0038 µg/l as a 24-hour average and the concentration should not exceed 0.52 µg/l at any time.

Saltwater Aquatic Life

For heptachlor the criterion to protect saltwater aquatic life as derived using the Guidelines is 0.0038 µg/l as a 24-hour average and the concentration should not exceed 0.053 µg/l at any time.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of heptachlor through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk, over the lifetimes are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 2.78 ng/L, 28 ng/L, and .028 ng/L, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 2.85 ng/L, .29 ng/L, and .029 ng/L, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Hexachlorobutadiene

Freshwater Aquatic Life

The available data for hexachlorobutadiene indicate that acute and chronic toxicity to freshwater aquatic life occur at concentrations as low as 90 and 9.3 µg/L, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

The available data for hexachlorobutadiene indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 32 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of hexachlorobutadiene to sensitive saltwater aquatic life.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of hexachlorobutadiene through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk, over the lifetimes are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 4.47 µg/L, 0.45 µg/L, and 0.045 µg/L, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 500 µg/L, 50 µg/L, and 5 µg/L, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Hexachlorocyclohexane

Lindane

Freshwater Aquatic Life

For Lindane the criterion to protect freshwater aquatic life as derived using the Guidelines is 0.080 µg/l as a 24-hour average and the concentration should not exceed 2.0 µg/l at any time.

Saltwater Aquatic Life

For saltwater aquatic life the concentration of lindane should not exceed 0.16 µg/l at any time. No data are available concerning the chronic toxicity of lindane to sensitive saltwater aquatic life.

BHC

Freshwater Aquatic Life

The available data for a mixture of isomers of BHC indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 100 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available

concerning the chronic toxicity of a mixture of isomers of BHC to sensitive freshwater aquatic life.

Saltwater Aquatic Life

The available data for a mixture of isomers of BHC indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 0.34 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of a mixture of isomers of BHC to sensitive saltwater aquatic life.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of alpha-HCH through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk, over the lifetimes are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 92 ng/L, 9.2 ng/L, and .92 ng/L, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 310 ng/L, 31.0 ng/L, and 3.1 ng/L, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

For the maximum protection of human health from the potential carcinogenic effects due to exposure of beta-HCH through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk, over the lifetimes are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 163 ng/L, 16 ng/L, and 1.63 ng/L, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 547 ng/L, 54.7 ng/L, and 5.47 ng/L, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not

represent an Agency judgment on an "acceptable" risk level.

For the maximum protection of human health from the potential carcinogenic effects due to exposure of tech-HCH through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk, over the lifetimes are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 123 ng/L, 12.3 ng/L, and 1.23 ng/L, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 414 ng/L, 41.4 ng/L, and 4.14 ng/L, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

For the maximum protection of human health from the potential carcinogenic effects due to exposure of gamma-HCH through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentrations should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time.

Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 186 ng/L, 18.6 ng/L, and 1.86 ng/L, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 625 ng/L, 62.5 ng/L, 6.25 ng/L, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Using the present guidelines, a satisfactory criterion cannot be derived at this time due to the insufficiency in the available data for delta-HCH.

Using the present guidelines, a satisfactory criterion cannot be derived at this time due to the insufficiency in the available data for epsilon-HCH.

Hexachlorocyclopentadiene

Freshwater Aquatic Life

The available data for hexachlorocyclopentadiene indicate that acute and chronic toxicity to freshwater

aquatic life occurs at concentrations as low as 7.0 and 5.2 $\mu\text{g/L}$, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

The available data to hexachlorocyclopentadiene indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 7.0 $\mu\text{g/L}$ and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of hexachlorocyclopentadiene to sensitive saltwater aquatic life.

Human Health

For comparison purposes, two approaches were used to derive criterion levels for hexachlorocyclopentadiene. Based on available toxicity data, for the protection of public health, the derived level is 206 $\mu\text{g/L}$. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 1.0 $\mu\text{g/L}$. It should be recognized that organoleptic data as a basis for establishing a water quality criterion have limitations and have no demonstrated relationship to potential adverse human health effects.

Isophorone

Freshwater Aquatic Life

The available data for isophorone indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 117,000 $\mu\text{g/L}$ and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of isophorone to sensitive freshwater aquatic life.

Saltwater Aquatic Life

The available data for isophorone indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 12,900 $\mu\text{g/L}$ and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of isophorone to sensitive saltwater aquatic life.

Human Health

For the protection of human health from the toxic properties of isophorone ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 5.2 $\mu\text{g/L}$.

For the protection of human health from the toxic properties of isophorone

ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 520 mg/L .

Lead

Freshwater Aquatic Life

For total recoverable lead the criterion (in $\mu\text{g/L}$) to protect freshwater aquatic life as derived using the Guidelines is the numerical value given by $e(2.35[\ln(\text{hardness})]-0.48)$ as a 24-hour average and the concentration (in $\mu\text{g/L}$) should not exceed the numerical value given by $e(3.22[\ln(\text{hardness})]-0.47)$ at any time. For example, at hardnesses of 50, 100, and 200 mg/L as CaCO_3 , the criteria are 0.75, 3.8, and 20 $\mu\text{g/L}$, respectively, as 24-hour averages, and the concentrations should not exceed 74, 170, and 400 $\mu\text{g/L}$, respectively, at any time.

Saltwater Aquatic Life

The available data for total recoverable lead indicate that acute and chronic toxicity to saltwater aquatic life occur at concentrations as low as 668 and 25 $\mu\text{g/L}$, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Human Health

The ambient water quality criterion for lead is recommended to be identical to the existing drinking water standard which is 50 $\mu\text{g/L}$. Analysis of the toxic effects data resulted in a calculated level which is protective to human health against the ingestion of contaminated water and contaminated aquatic organisms. The calculated value is comparable to the present standard. For this reason a selective criterion based on exposure solely from consumption of 8.5 grams of aquatic organisms was not derived.

Mercury

Freshwater Aquatic Life

For total recoverable mercury the criterion to protect freshwater aquatic life as derived using the Guidelines is 0.00057 $\mu\text{g/L}$ as a 24-hour average and the concentration should not exceed 0.0017 $\mu\text{g/L}$ at any time.

Saltwater Aquatic Life

For total recoverable mercury the criterion to protect saltwater aquatic life as derived using the Guidelines is 0.025 $\mu\text{g/L}$ as a 24-hour average and the concentration should not exceed 3.7 $\mu\text{g/L}$ at any time.

Human Health

For the protection of human health from the toxic properties of mercury

ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 144 ng/L.

For the protection of human health from the toxic properties of mercury ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 146 ng/L.

Note.—These values include the consumption of freshwater, estuarine, and marine species.

Naphthalene

Freshwater Aquatic Life

The available data to naphthalene indicate that acute and chronic toxicity to freshwater aquatic life occur at concentrations as low as 2,300 and 620 µg/L respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

The available data for naphthalene indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 2,350 µg/L and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of naphthalene to sensitive saltwater aquatic life.

Human Health

Using the present guidelines, a satisfactory criterion cannot be derived at this time due to the insufficiency in the available data for naphthalene.

Nickel

Freshwater Aquatic Life

For total recoverable nickel the criterion (in µg/L) to protect freshwater aquatic life as derived using the Guidelines is the numerical value given by $e(0.76[\ln(\text{hardness})] + 1.06)$ as a 24-hour average and the concentration (in µg/L) should not exceed the numerical value given by $e(0.76[\ln(\text{hardness})] + 4.02)$ at any time. For example, at hardnesses of 50, 100, and 200 mg/L as CaCO₃, the criteria are 58, 96, and 160 µg/L respectively, as 24-hour averages, and the concentrations should not exceed 1,100, 1,600, and 3,100 µg/L respectively, at any time.

Saltwater Aquatic Life

For total recoverable nickel the criterion to protect saltwater aquatic life as derived using the Guidelines is 7.1 µg/L as a 24-hour average and the concentration should not exceed 140 µg/L at any time.

Human Health

For the protection of human health from the toxic properties of nickel ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 13.4 µg/L.

For the protection of human health from the toxic properties of nickel ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 100 µg/L.

Nitrobenzene

Freshwater Aquatic Life

The available data for nitrobenzene indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 27,000 µg/L and would occur at lower concentrations among species that are more sensitive than those tested. No definitive data are available concerning the chronic toxicity of nitrobenzene to sensitive freshwater aquatic life.

Saltwater Aquatic Life

The available data for nitrobenzene indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 8,680 µg/L and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of nitrobenzene to sensitive saltwater aquatic life.

Human Health

For comparison purposes, two approaches were used to derive criterion levels for nitrobenzene. Based on available toxicity data, for the protection of public health, the derived level is 19.8 mg/L. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 30 µg/L. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have no demonstrated relationship to potential adverse human health effects.

Nitrophenols

Freshwater Aquatic Life

The available data for nitrophenols indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 230 µg/L and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of nitrophenols to sensitive freshwater aquatic life but toxicity to one species of algae occurs at concentrations as low as 150 µg/L.

Saltwater Aquatic Life

The available data for nitrophenols indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 4,850 µg/L and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of nitrophenols to sensitive saltwater aquatic life.

Human Health

For the protection of human health from the toxic properties of 2,4-dinitro-cresol ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 13.4 µg/L.

For the protection of human health from the toxic properties of 2,4-dinitro-cresol ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 785 µg/L.

For the protection of human health from the toxic properties of dinitrophenol ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 70 µg/L.

For the protection of human health from the toxic properties of dinitrophenol ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 143 mg/L.

Using the present guidelines, a satisfactory criterion cannot be derived at this time due to the insufficiency in the available data for mononitrophenol.

Using the present guidelines, a satisfactory criterion cannot be derived at this time due to the insufficiency in the available data for tri-nitrophenol.

Nitrosamines

Freshwater Aquatic Life

The available data for nitrosamines indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 5,850 µg/L and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of nitrosamines to sensitive freshwater aquatic life.

Saltwater Aquatic Life

The available data for nitrosamines indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 3,300,000 µg/L and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of nitrosamines to sensitive saltwater aquatic life.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of *n*-nitrosodimethylamine through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk, over the lifetimes are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 14 ng/l, 1.4 ng/l, and .14 ng/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 160,000 ng/l, 16,000 ng/l, and 1,600 ng/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

For the maximum protection of human health from the potential carcinogenic effects due to exposure of *n*-nitrosodiethylamine through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk, over the lifetimes are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 8 ng/l, 0.8 ng/l, and 0.08 ng/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 12,400 ng/l, 1,240 ng/l, and 124 ng/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

For the maximum protection of human health from the potential carcinogenic effects due to exposure in *n*-nitrosodimethylamine through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk, over the lifetimes are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are

64 ng/l, 6.4 ng/l, and .64 ng/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 5,868 ng/l, 587 ng/l, and 58.7 ng/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

For the maximum protection of human health from the potential carcinogenic effects due to exposure in *n*-nitrosodiphenylamine through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk, over the lifetimes are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 49,000 ng/l, 4,900 ng/l, and 490 ng/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 161,000 ng/l, 16,100 ng/l, and 1,610 ng/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

For the maximum protection of human health from the potential carcinogenic effects due to exposure in *n*-nitrosopyrrolidine through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk, over the lifetimes are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 180 ng/l, 18.0 ng/l, and 1.80 ng/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 919,000 ng/l, 91,900 ng/l, and 9,190 ng/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Pentachlorophenol**Freshwater Aquatic Life**

The available data for pentachlorophenol indicate that acute and chronic toxicity to freshwater aquatic life occur at concentrations as low as 55 and 3.2 $\mu\text{g/l}$, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

The available data for pentachlorophenol indicate that acute and chronic toxicity to saltwater aquatic life occur at concentrations as low as 53 and 34 $\mu\text{g/l}$, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Human Health

For comparison purposes, two approaches were used to derive criterion levels for pentachlorophenol. Based on available toxicity data, for the protection of public health, the derived level is 1.01 mg/l. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is 30 $\mu\text{g/l}$. It should be recognized that organoleptic data as a basis for establishing a water quality criterion have limitations and have no demonstrated relationship to potential adverse human health effects.

Phenol**Freshwater Aquatic Life**

The available data for phenol indicate that acute and chronic toxicity to freshwater aquatic life occur at concentrations as low as 10,200 and 2,580 $\mu\text{g/l}$, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

The available data for phenol indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 5,800 $\mu\text{g/l}$ and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of phenol to sensitive saltwater aquatic life.

Human Health

For comparison purposes, two approaches were used to derive criterion levels for phenol. Based on available toxicity data, for the protection of public health, the derived level is 3.5 mg/l. Using available organoleptic data, for controlling

undesirable taste and odor quality of ambient water, the estimated level is 0.3 mg/L. It should be recognized that organoleptic data as a basis for establishing a water quality criterion have limitations and have no demonstrated relationship to potential adverse human health effects.

Phthalate Esters

Freshwater Aquatic Life

The available data for phthalate esters indicate that acute and chronic toxicity to freshwater aquatic life occur at concentrations as low as 940 and 3 µg/L, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

The available data for phthalate esters indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 2944 µg/L and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of phthalate esters to sensitive saltwater aquatic life but toxicity to one species of algae occurs at concentrations as low as 3.4 µg/L.

Human Health

For the protection of human health from the toxic properties of dimethyl-phthalate ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 313 mg/L.

For the protection of human health from the toxic properties of dimethyl-phthalate ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 2.9 g/L.

For the protection of human health from the toxic properties of diethyl-phthalate ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 350 mg/L.

For the protection of human health from the toxic properties of diethyl-phthalate ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 1.8 g/L.

For the protection of human health from the toxic properties of dibutyl-phthalate ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 34 mg/L.

For the protection of human health from the toxic properties of dibutyl-phthalate ingested through

contaminated aquatic organisms alone, the ambient water criterion is determined to be 154 mg/L.

For the protection of human health from the toxic properties of di-2-ethylhexyl-phthalate ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 15 mg/L.

For the protection of human health from the toxic properties of di-2-ethylhexyl-phthalate ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 50 mg/L.

Polychlorinated Biphenyls

Freshwater Aquatic Life

For polychlorinated biphenyls the criterion to protect freshwater aquatic life as derived using the Guidelines is 0.014 µg/l as a 24-hour average. The available data indicate that acute toxicity to freshwater aquatic life probably will only occur at concentrations above 2.0 µg/l and that the 24-hour average should provide adequate protection against acute toxicity.

Saltwater Aquatic Life

For polychlorinated biphenyls the criterion to protect saltwater aquatic life as derived using the Guidelines is 0.030 µg/l as a 24-hour average. The available data indicate that acute toxicity to saltwater aquatic life probably will only occur at concentrations above 10 µg/l and that the 24-hour average should provide adequate protection against acute toxicity.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of PCBs through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-4} , and 10^{-2} . The corresponding criteria are 79 ng/l, 0.79 ng/l, and 0.079 ng/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 79 ng/l, 0.79 ng/l, and 0.079 ng/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not

represent an Agency judgment on an "acceptable" risk level.

Polynuclear Aromatic Hydrocarbons (PAHs)

Freshwater Aquatic Life

The limited freshwater data base available for polynuclear aromatic hydrocarbons, mostly from short-term bioconcentration studies with two compounds, does not permit a statement concerning acute or chronic toxicity.

Saltwater Aquatic Life

The available data for polynuclear aromatic hydrocarbons indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 30 µg/l and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of polynuclear aromatic hydrocarbons to sensitive saltwater aquatic life.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of PAHs through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-4} , and 10^{-2} . The corresponding criteria are 28 ng/l, 2.8 ng/l, and 28 ng/l, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 311 ng/l, 31.1 ng/l, and 3.11 ng/l, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Selenium

Freshwater Aquatic Life

For total recoverable inorganic selenite the criterion to protect freshwater aquatic life as derived using the Guidelines is 95 µg/l as a 24-hour average and the concentration should not exceed 260 µg/l at any time.

The available data for inorganic selenate indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 760 µg/l and would occur at lower concentrations among species that are more sensitive

than those tested. No data are available concerning the chronic toxicity of inorganic selenate to sensitive freshwater aquatic life.

Saltwater Aquatic Life

For total recoverable inorganic selenite the criterion to protect saltwater aquatic life as derived using the Guidelines is 54 $\mu\text{g/l}$ as a 24-hour average and the concentration should not exceed 410 $\mu\text{g/l}$ at any time.

No data are available concerning the toxicity of inorganic selenate to saltwater aquatic life.

Human Health

The ambient water quality criterion for selenium is recommended to be identical to the existing drinking water standard which is 10 $\mu\text{g/l}$. Analysis of the toxic effects data resulted in a calculated level which is protective of human health against the ingestion of contaminated water and contaminated aquatic organisms. The calculated value is comparable to the present standard. For this reason a selective criterion based on exposure solely from consumption of 6.5 grams of aquatic organisms was not derived.

Silver

Freshwater Aquatic Life

For freshwater aquatic life the concentration (in $\mu\text{g/l}$) of total recoverable silver should not exceed the numerical value given by $1.72[\ln(\text{hardness}) - 6.52]$ at any time. For example, at hardnesses of 50, 100, 200 mg/l as CaCO_3 the concentration of total recoverable silver should not exceed 1.2, 4.1, and 13 $\mu\text{g/l}$, respectively, at any time. The available data indicate that chronic toxicity to freshwater aquatic life may occur at concentrations as low as 0.12 $\mu\text{g/l}$.

Saltwater Aquatic Life

For saltwater aquatic life the concentration of total recoverable silver should not exceed 2.3 $\mu\text{g/l}$ at any time. No data are available concerning the chronic toxicity of silver to sensitive saltwater aquatic life.

Human Health

The ambient water quality criterion for silver is recommended to be identical to the existing drinking water standard which is 50 $\mu\text{g/l}$. Analysis of the toxic effects data resulted in a calculated level which is protective of human health against the ingestion of contaminated water and contaminated aquatic organisms. The calculated value is comparable to the present standard. For this reason a selective criterion based on exposure solely from

consumption of 6.5 grams of aquatic organisms was not derived.

Tetrachloroethylene

Freshwater Aquatic Life

The available data for tetrachloroethylene indicate that acute and chronic toxicity to freshwater aquatic life occur at concentrations as low as 5,280 and 840 $\mu\text{g/l}$, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Saltwater Aquatic Life

The available data for tetrachloroethylene indicate that acute and chronic toxicity to saltwater aquatic life occur at concentrations low as 10,200 and 450 $\mu\text{g/l}$, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of tetrachloroethylene through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 8 $\mu\text{g/l}$, 8 $\mu\text{g/l}$, and .08 $\mu\text{g/l}$, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 88.5 $\mu\text{g/l}$, 8.85 $\mu\text{g/l}$, and .88 $\mu\text{g/l}$, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Thallium

Freshwater Aquatic Life

The available data for thallium indicate that acute and chronic toxicity to freshwater aquatic life occur at concentrations as low as 1,400 and 40 $\mu\text{g/l}$, respectively, and would occur at lower concentrations among species that are more sensitive than those tested. Toxicity to one species of fish occurs at concentrations as low as 20 $\mu\text{g/l}$ after 2,600 hours of exposure.

Saltwater Aquatic Life

The available data for thallium indicate that acute toxicity to saltwater

aquatic life occurs at concentrations as low as 2,130 $\mu\text{g/l}$ and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of thallium to sensitive saltwater aquatic life.

Human Health

For the protection of human health from the toxic properties of thallium ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 13 $\mu\text{g/l}$.

For the protection of human health from the toxic properties of thallium ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 48 $\mu\text{g/l}$.

Toluene

Freshwater Aquatic Life

The available data for toluene indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 17,500 $\mu\text{g/l}$ and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of toluene to sensitive freshwater aquatic life.

Saltwater Aquatic Life

The available data for toluene indicate that acute and chronic toxicity to saltwater aquatic life occur at concentrations as low as 6,300 and 5,000 $\mu\text{g/l}$, respectively, and would occur at lower concentrations among species that are more sensitive than those tested.

Human Health

For the protection of human health from the toxic properties of toluene ingested through water and contaminated aquatic organisms, the ambient water criterion is determined to be 14.3 mg/l .

For the protection of human health from the toxic properties of toluene ingested through contaminated aquatic organisms alone, the ambient water criterion is determined to be 424 mg/l .

Toxaphene

Freshwater Aquatic Life

For toxaphene the criterion to protect freshwater aquatic life as derived using the Guidelines is 0.013 $\mu\text{g/l}$ as a 24-hour average and the concentration should not exceed 1.6 $\mu\text{g/l}$ at any time.

Saltwater Aquatic Life

For saltwater aquatic life the concentration of toxaphene should not exceed 0.070 $\mu\text{g/l}$ at any time. No data

are available concerning the chronic toxicity of toxaphene to sensitive saltwater aquatic life.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of toxaphene through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 7.1 ng/L, .71 ng/L, and .07 ng/L, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 7.3 ng/L, .73 ng/L, and .07 ng/L, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Trichloroethylene

Freshwater Aquatic Life

The available data for trichloroethylene indicate that acute toxicity to freshwater aquatic life occurs at concentrations as low as 45,000 $\mu\text{g/l}$ and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of trichloroethylene to sensitive freshwater aquatic life but adverse behavioral effects occurs to one species at concentrations as low as 21,900 $\mu\text{g/l}$.

Saltwater Aquatic Life

The available data for trichloroethylene indicate that acute toxicity to saltwater aquatic life occurs at concentrations as low as 2,000 $\mu\text{g/l}$ and would occur at lower concentrations among species that are more sensitive than those tested. No data are available concerning the chronic toxicity of trichloroethylene to sensitive saltwater aquatic life.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of trichloroethylene through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on

the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 27 $\mu\text{g/l}$, 2.7 $\mu\text{g/l}$, and .27 $\mu\text{g/l}$, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 807 $\mu\text{g/l}$, 80.7 $\mu\text{g/l}$, and 8.07 $\mu\text{g/l}$, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Vinyl Chloride

Freshwater Aquatic Life

No freshwater organisms have been tested with vinyl chloride and no statement can be made concerning acute or chronic toxicity.

Saltwater Aquatic Life

No saltwater organisms have been tested with vinyl chloride and no statement can be made concerning acute or chronic toxicity.

Human Health

For the maximum protection of human health from the potential carcinogenic effects due to exposure of vinyl chloride through ingestion of contaminated water and contaminated aquatic organisms, the ambient water concentration should be zero based on the non-threshold assumption for this chemical. However, zero level may not be attainable at the present time. Therefore, the levels which may result in incremental increase of cancer risk over the lifetime are estimated at 10^{-6} , 10^{-5} , and 10^{-4} . The corresponding criteria are 20 $\mu\text{g/l}$, 2.0 $\mu\text{g/l}$, and .2 $\mu\text{g/l}$, respectively. If the above estimates are made for consumption of aquatic organisms only, excluding consumption of water, the levels are 5,248 $\mu\text{g/l}$, 525 $\mu\text{g/l}$, and 52.5 $\mu\text{g/l}$, respectively. Other concentrations representing different risk levels may be calculated by use of the Guidelines. The risk estimate range is presented for information purposes and does not represent an Agency judgment on an "acceptable" risk level.

Zinc

Freshwater Aquatic Life

For total recoverable zinc the criterion to protect freshwater aquatic life as derived using the Guidelines is 47 $\mu\text{g/l}$ as a 24-hour average and the concentration (in $\mu\text{g/l}$) should not

exceed the numerical value given by $\frac{100}{\text{hardness} + 100}$ at any time. For example, at hardnesses of 50, 100, and 200 mg/l as CaCO_3 , the concentration of total recoverable zinc should not exceed 180, 320, and 570 $\mu\text{g/l}$ at any time.

Saltwater Aquatic Life

For total recoverable zinc the criterion to protect saltwater aquatic life as derived using the Guidelines is 58 $\mu\text{g/l}$ as a 24-hour average and the concentration should not exceed 170 $\mu\text{g/l}$ at any time.

Human Health

Sufficient data is not available for zinc to derive a level which would protect against the potential toxicity of this compound. Using available organoleptic data, for controlling undesirable taste and odor quality of ambient water, the estimated level is mg/l. It should be recognized that organoleptic data as a basis for establishing a water quality criteria have limitations and have not demonstrated relationship to potential adverse human health effects.

Appendix B—Guidelines for Deriving Water Quality Criteria for the Protection of Aquatic Life and Its Uses

Introduction

This version of the Guidelines provides clarifications, additional details, and technical and editorial changes in the last version published in the Federal Register [44 FR 15970 (March 15, 1979)]. This version incorporates changes resulting from comments on previous versions and from experience gained during U.S. EPA's use of the previous versions. Future versions of the Guidelines will incorporate new ideas and data as their usefulness is demonstrated.

Criteria may be expressed in several forms. The numerical form is commonly used, but descriptive and procedural forms can be used if numerical criteria are not possible or desirable. The purpose of these Guidelines is to describe an objective, internally consistent and appropriate way of deriving numerical water quality criteria for the protection of the uses of, as well as the presence of, aquatic organisms.

A numerical criterion might be thought of as an estimate of the highest concentration of a substance in water which does not present a significant risk to the aquatic organisms in the water and their uses. Thus the Guidelines are intended to derive criteria which will protect aquatic communities by protecting most of the species and their uses most of the time, but not

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Source:

California State Water Resources
Control Board, 1985,

Interim Guidance for Hazardous Substance
Site Clean-up, CWRCB Resolution No. 85-26

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CHEMICAL	EPA MCL	EPA RMCL	CA DOHS Action Level	acute 1 day	HEALTH ADVISORIES (SHARLS)		EPA NAWQC
					subchronic (no. of days)	chronic non-cancer cancer	
<u>INORGANICS</u>							
Aluminum		\$		35000*	5000(7)*		
Antimony		\$					146
Arsenic	50						2.2 ppt
Asbestos		\$					30,000 fibers/l
Barium	1000			6000*		4700*	
Beryllium		\$					6.8 ppt
Bromide				1400ppm*	224ppm(7)*	2.3ppm*	
Cadmium	10			150*	21(7)*	5*	10
Chloramine				1200*	125 (7)*		
Chlorate				125*	125 (7)*		
Chloride	250ppm††		100ppm (CA MCL)				
Chlorine Dioxide				1200*	125 (7)*		
Chlorite				125*	125 (7)*		
Chromium	50						170000(Cr ⁺³) 50(Cr ⁺⁶)
Copper	1000††	\$					1000
Cyanide		\$		1			200
Fluoride	1400-2400						
Iodide				115500	16500(7)*	1190*	

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CHEMICAL	EPA MCL	EPA RNCL	CA DOHS Action Level	HEALTH ADVISORIES (SNARLS)			EPA NAWQC
				acute 1 day	subchronic (no. of days)	chronic non-cancer	
				cancer	non-cancer	cancer	
Iron	300††						
Lead	50						50
Manganese	50††						
Mercury	2						144 ppt
Molybdenum		\$					
Nickel		\$					13.4
Nitrate (as N)	10.0ppm						
Selenium	10						10
Silver	50						50
Sodium		\$					
Strontium					8400(7)*		
Sulfate	250ppm††	\$					
Thallium		\$					13
Uranium						35*	
Vanadium		\$					
Zinc	5000††	\$					5000 (taste&odor)

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CHEMICAL	EPA MCL	EPA RMCL	CA DOHS Action Level	acute 1 day	HEALTH ADVISORIES (SNARLS)		EPA NAMQC non-cancer	cancer
					subchronic (no. of days)	chronic non-cancer		
<u>ORGANICS</u>								
Acenaphthene							20 (taste&odor)	
Acrolein						320		
Acrylamide	\$							
Acrylonitrile				35(10) 3(30)			0.77 *	0.05u
Adipates	\$							
Alachlor	\$							
Aldicarb	\$		10			10		
Aldrin			0.05					0.074ppt
Atrazine	\$							
Baygon			90					
Benzene		0	0.70	350(7)			0.67	0.66
Benzene hexachloride (BHC, Lindane)			0.70 (∞) 0.30 (p)	3500*	30(7)*	0.35	0.54*	9.2ppt (∞) 16.3ppt (p) 12.3ppt (r) 18.6ppt (d) 0.12ppt
Benzidine								
Benzo(a)pyrene					25(7)			
Boleto(thiobencarb)			10 (1 taste&odor)					0.19
Bromochloromethane 100**								0.19
Bromomethane (Methyl bromide)								

**Total Trihalomethanes

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CHEMICAL	EPA MCL	EPA RMCL	CA DOHS Action Level	acute 1 day	HEALTH ADVISORIES (SNARKS)		EPA NAMQC non-cancer	EPA NAMQC cancer
					subchronic (no. of days)	chronic non-cancer		
Captan			350					
Carbofuran		5						
Carbon tetrachloride		0	5.0	200†	20(10)†		0.4†	0.40
Catechol				2200*				
Chlordane		5	55	63	8(10)		0.023	0.46ppt
Chlorobenzene						72	4.7*	100 (20 taste&odor)
bis-(2-chloroethyl) ether								0.03
Chloroform (trichloromethane)				22000*	3200(7)*			0.19
bis-(2-chloroisopropyl) ether							34.7	
Chloromethane (Methyl chloride)								0.19
bis-(chloromethyl) ether								0.0038 ppt
2-Chlorophenol								0.1(taste&odor)
3-Chlorophenol								0.1(taste&odor)
4-Chlorophenol								0.1(taste&odor)
CIPC			350					
Dalapon		5						
DDT				1				0.024ppt
Diazinon			14					
Dibromochloromethane				18000*				0.19

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CHEMICAL	EPA MCL	EPA RMCL	CA DOHS Action Level	acute 1 day	HEALTH ADVISORIES (SNARLS)		EPA NAWQC		
					subchronic (no. of days)	chronic			
					non-cancer	cancer	non-cancer	cancer	
1,2-Dibromo-3-chloro-propane (DBCP)		\$	1.0			0.050		0.01	
1,2-Dibromothane (Ethylene Dibromide)		\$	LOQ(0.05)					0.67ppt	
Di-n-butyl phthalate									34000
1,2-Dichlorobenzene			130 ^{AA} (10 tasteodor)			300			400
1,3-Dichlorobenzene			130 ^{AA} (20 tasteodor)						400
1,4-Dichlorobenzene		750	130 ^{AA} (0.3 tasteodor)			13			400
Dichlorobenzidine									0.0103
Dichlorodifluoromethane									0.19
1,2-Dichloroethane		0	1.0					0.6†	0.94
1,1-Dichloroethylene		0	LOQ(.1-.4)	1000†		70†		0.24	0.033
cis-1,2-Dichloroethylene				4000†					
trans-1,2-Dichloroethylene				2700†					
Dichlorofluoromethane (Freon 21)				100ppm		43ppm(7)			1.6ppm
Dichloromethane									0.19
2,3-Dichlorophenol									0.04 (tasteodor)
2,4-Dichlorophenol						700*			
2,5-Dichlorophenol									0.5 (tasteodor)
2,6-Dichlorophenol									0.2 (tasteodor)
3,4-Dichlorophenol									0.3 (tasteodor)

**Action level is for a single isomer or sum of the three

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CHEMICAL	EPA MCL	EPA RMCL	CA DOHS Action Level	acute 1 day	HEALTH ADVISORIES (SHALES)		EPA NAWQC non-cancer	EPA NAWQC cancer
					subchronic (no. of days)	chronic cancer		
2,4-Dichlorophenoxy- acetic acid (2,4-D)	100							
1,2-Dichloropropane	\$		10					
Dichloropropene							0.07 ppt	
Dieldrin			100 (.05)					
Di-(2-ethylhexyl) phthalate or DEHP					210		15000	
Diethyl phthalate							350000	
Dimethoate			140					
2,4-dimethylphenol			400 (taste&odor)					
Dimethylphthalate							313000	
2,4-Dinitro-o-cresol							13.4	
2,4-Dinitrophenol					110*			
2,4-Dinitrotoluene								0.11
Dinoseb (2-sec-Butyl- 4,6-dinitrophenol)	\$					39*		
1,4-Dioxane					20(10)			
Diphenamide			40					0.042
1,2-Diphenylhydrazine								
Diquat	\$							
Endosulfan								74
Endothall	\$							

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CHEMICAL	EPA MCL	EPA RMCL	CA DOHS Action level	HEALTH ADVISORIES (SHARLS)			EPA HAHQ non-cancer	EPA HAHQ cancer
				acute 1 day	subchronic (no. of days)	chronic non-cancer		
Endrin	0.2							
Epichlorohydrin		5		840*	530(7)*			
Ethion			35					
Ethylbenzene							1400	
Ethylene dibromide (EDB)		5	LOQ(.05)			0.67ppt		
Ethylene glycol				19000†		5500†		
Fluoranthene							42	
Formaldehyde			30	30**				
Glyphosate		5	500					0.28
Heptachlor			0.02					
Heptachlor epoxide			0.10					
Hexachlorobenzene					30(7)*	0.35	0.54*	0.72
Hexachlorobutadiene								0.45
Hexachlorocyclo- pentadiene		5					206	
Hexachlorophene						0.35	1.0 (taste&odor)	
n-Hexane				12900†	4000(10)†			
Isophorone				†			5200	
Isopropyl alcohol (Isopropanol)				1000	1000(10)			
Lindane (gamma-BHC)	4							12.3ppt

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CHEMICAL	EPA MCL	EPA RMCL	CA DOHS Action Level	acute 1 day	HEALTH ADVISORIES (SNARLS)		EPA NAWQC non-cancer	EPA NAWQC cancer
					subchronic (no. of days)	chronic non-cancer		
Pentachlorobenzene							74	
Pentachlorophenol	\$		30					
Picloram	\$				1050*		3500	
Phenol			1.0				300 (tasteodor)	
Phthalates	\$							0.7 ppt
Polychlorinated Biphenyls (PCB)	\$			125	12.5 (10) 1 (30)			2.8 ppt
Polynuclear Aromatic Hydrocarbons	\$							
Resorcinol				11700*	500*			
Rotenone						14*		
Simazine	\$							
Styrene						1300		
Tetrachlor (pentachloronitrobenzene)								
1,2,4,5-Tetrachlorobenzene			0.9				38	
2,3,7,8-Tetrachlorodibenzo-p-dioxin	\$							1.3x10 ⁻⁸
1,1,2,2-Tetrachloroethane								0.17
Tetrachloroethylene	0		4.0	2300	175 (10)			0.8
2,3,4,6-Tetrachlorophenol								1 (tasteodor)
Toluene	\$		100	1000	1000 (10)	100	14300	
Toxaphene	5							0.7 ppt

DRINKING WATER STANDARDS AND HEALTH ADVISORIES

CHEMICAL	EPA MCL	EPA RMCL	CA DOHS Action Level	HEALTH ADVISORIES (SVANLIS)		EPA MCLC non-cancer	EPA MCLC cancer
				acute 1 day (no. of days)	subchronic chronic non-cancer cancer		
Malathion			160				
Methomyl (Lannate)	100				175*		
Methoxychlor						1800 (taste&odor)	
2-methyl-4-chloro-phenol						3000 (taste&odor)	
3-methyl-4-chloro-phenol						20 (taste&odor)	
3-methyl-6-chloro-phenol			40	1300†	1300-1500(10)†		150†
ethylene chloride				7500†	750(10)†		
Methyl ethyl ketone							35
Methyl methacrylate							
Methyl Parathion			30				
Mononitrophenol					290(7)*		
Nitrobenzene				35*	5(7)*	19800	30 (taste&odor)
n-Nitrosodi-n-butyl-amine						6.4 ppt	
n-Nitrosodiethylamine						0.8 ppt	
n-Nitrosodimethylamine						1.4 ppt	
n-Nitrosodiphenylamine						4.9	
n-Nitrosopyrrolidine				1			16.0 ppt
Ordram (Molinate)			20				
Parathion			30				

DRINKING WATER STANDARDS AND HEALTH ADVISORIES

10 of 10

CHEMICAL	EPA MCL	EPA RNCL	CA DQHS Action Level	HEALTH ADVISORIES (SHARLS)		HEALTH ADVISORIES (SHARLS) chronic		EPA NAWQC non-cancer	EPA NAWQC cancer
				acute 1 day	subchronic (no. of days)	non-cancer	cancer		
tribromomethane (Bromoform)	100**								0.19
1,1,1-Trichloroethane	200		200	14000	20000(10)	1000		18400	
1,1,2-Trichloroethane	\$								0.6
Trichloroethylene	0		5.0	2000	200(10)	75	2.8		2.7
Trichlorofluoromethane (Freon 11)				25000	2200(10)				0.19
2,4,5-Trichlorophenol								2600	1 (taste&odor)
2,4,6-Trichlorophenol				17500*	2500(7)*			2 (taste&odor)	1.2
2,4,5-Trichlorophenoxypropionic acid (2,4,5-TP Silvex)	10								
2,4,6-Trinitrophenol				4900*	200(7)*				
Trithion			7.0						
vinyl chloride	0		2.0				2.0		2.0
Vydate	\$								
Xylenes***	\$		620	12000†	1400(10)†	620†			
n-Xylene			620	6100	3200(10)				
o-Xylene			620						
p-Xylene			620						

APPENDIX M

RESULTS OF VOLATILE ORGANIC SAMPLING OF PRIVATE WELLS
NEAR MATHER AFB, PROVIDED BY
CRWQCB-CVR, AUGUST 1984

VOLATILE ORGANIC SAMPLING RESULTS OF PRIVATE WELLS NEAR MATHER AFB

Conducted by: - California Regional Water Quality Control Board
 - Sacramento County Health Department

WELL NO.	OWNER/ADDRESS	SAMPLE DATE	VOLATILE ORGANIC COMPOUND DETECTED	LEVEL DETECTED (ppb)	STATE ACTION LEVEL (ppb)
1	Camellia Mather Mobile Home Park (32 Units) 3000 Camellia Mather Drive	5 Feb 80	1,1,1 Trichloroethane	1.6	300.0
			Perchloroethylene	1.0	4.0
			Trichloroethylene	2.4	5.0
		29 Apr 80	None		
4		2 Jun 81	Chloroform	0.3	100.0
			Carbon Tetrachloride	1.5	5.0
			Tetrachloroethylene	1.4	4.0
			Trichloroethylene	5.1	5.0
			Trans-1,2-Dichloroethylene	1.5	None
		16 Sep 81	Chloroform	0.8	100.0
			Carbon Tetrachloride	1.8	5.0
			Tetrachloroethylene	1.7	4.0
			Trichloroethylene	5.6	5.0
		22 Aug 83	Carbon Tetrachloride	1.1	5.0

LEGEND: () State Action Level Exceeded

22 Aug 83	Tetrachloroethylene	2.3	4.0
	Trichloroethylene	4.0	5.0
	1,1,1, Trichloroethane	1.1	300.0
2 May 84	Carbon Tetrachloride	1.3	5.0
	Tetrachloroethylene	4.2	4.0
	Trichloroethylene (Sample taken at air valve leak)	6.3	5.0
	Carbon Tetrachloride	1.5	5.0
	Tetrachloroethylene	4.3	4.0
	Trichloroethylene (Sample taken after pressure tank)	6.6	5.0
	Carbon Tetrachloride	1.3	5.0
	Tetrachloroethylene	4.2	4.0
	Trichloroethylene (Sample taken in residents home)	7.0	5.0
Jun 84	Trichloroethylene	7.2	5.0

LINE	OWNER/ADDRESS	SAMPLE DATE	VOLATILE ORGANIC COMPOUND DETECTED	LEVEL DETECTED (ppb)	STATE ACTION LEVEL (ppb)
	Hernandez 3875 Happy Lane	5 Feb 80	None		
1	Matsumoto 3851 Happy Lane	22 Mar 82	None (Home) None (Irrigation water)		
		27 Apr 84	Carbon Tetrachloride	1.0	5.0
			Trichloroethylene (Sample #1)	0.9	5.0
			Carbon Tetrachloride	1.0	5.0
			Trichloroethylene (Sample #2)	1.1	5.0
2	Rand 3900 Happy Lane	29 Jan 82	1,1 Dichloroethane	1.0	None
			Trans-1,2-Dichloroethylene	1.5	None
			Trichloroethylene	9.3	5.0
		26 Apr 84	Trichloroethylene	3.6	5.0
3	KiKo Landscaping 3950 Happy Lane	28 Oct 83	Trichloroethylene	1.3	5.0
		26 Apr 84	Trichloroethylene	3.3	5.0

STI No.	OWNER/ADDRESS	SAMPLE DATE	VOLATILE ORGANIC COMPOUND DETECTED	CONCENTRATION DETECTED (ppb)	STATE ACTION LEVEL (ppb)
6	Sterling 3960 Happy Lane	26 Apr 84	None		
7	Gregory 3990 Happy Lane	22 Mar 82	Tetrachloroethylene	2.5	4.0
			Trichloroethylene	2.2	5.0
			Trans-1,2,-Dichloroethylene	4.9	None
			1,1,1, Trichloroethane	1.2	300.0
8	Furuike 4001 Happy Lane	5 Feb 80	Trichloroethylene	2.0	5.0
		29 Apr 80	None		
		18 Dec 81	Trichloroethylene	8.0	5.0
			Trans-1,2,-Dichloroethylene	22.0	None
			1,1-Dichloroethylene	4.2	0.1
			Tetrachloroethylene	1.0	4.0
			1,1-Dichloroethane	2.7	None
		29 Jan 82	Trichloroethylene	8.0	5.0
			Trans-1,2,-Dichloroethylene	22.0	None
			1,1,-Dichloroethylene	1.9	0.1

SUTI NO.	OWNER/ADDRESS	SAMPLE DATE	VOLATILE ORGANIC COMPOUND DETECTED	CONCENTRATION DETECTED (ppb)	STATE ACTION LEVEL (ppb)
		29 Jan 82	Tetrachloroethylene	0.9	4.0
			1,1-Dichloroethane	1.9	None
			1,1,1,-Trichloroethane	0.2	300.0
		22 Aug 83	1,1-Dichloroethylene	3.1	0.1
			1,1-Dichloroethane	2.6	None
			Trans-1,2-Dichloroethylene	28.0	None
			Trichloroethylene	8.4	5.0
			Tetrachloroethylene	1.5	4.0
		Jun 84	1,1-Dichloroethylene	9.0	0.1
			1,2-Dichloroethane	3.6	None
	Trichloroethylene	15.0	5.0		
9	Church of Godatsu 1016 Hupp Lane	22 Mar 82	Trichloroethylene	1.8	5.0
			Trans-1,2-Dichloroethylene	9.6	None
			1,1-Dichloroethylene	0.7	0.1
			1,1-Dichloroethane	1.5	None
10	Tanaka	22 Mar 82	None		

STF NO.	OWNER/ADDRESS	SAMPLE DATE	VOLATILE ORGANIC COMPOUND DETECTED	CONCENTRATION DETECTED (ppb)	STATE ACTION LEVEL (ppb)
11	Cordova Truck Dismantlers 4075 Happy Lane	29 Jun 82	Trichloroethylene	0.2	5.0
		22 Aug 83	None	0.4	None
12	Rancho Truck Dismantlers 4079 Happy Lane	29 Jan 84	Trichloroethylene	0.2	5.0
		22 Aug 83	None	0.4	None
13	Mather Auto Dismantlers	6 Feb 80	Trichloroethylene	1.0	5.0
		29 Jan 82	None	1.6	300.0
14	Kobata 6108 Happy Lane	14 Dec 81	None		
15	Brupper 9745 Old Placer- Village Road	28 Oct 83	None		

SITI NO.	OWNER/ADDRESS	SAMPLE DATE	VOLATILE ORGANIC COMPOUND DETECTED	CONCENTRATION DETECTED (ppb)	STATE ACTION LEVEL (ppb)
15	Carbon Tetrachloride	26 Apr 84	Carbon Tetrachloride	1.3	5.0
16	Fatyrynuik 9835 Old Placer- Ville Road	27 Apr 84	None		
17	Robel 9874 Old Placer- Ville Road	22 Mar 82	Carbon Tetrachloride	1.5	5.0
18	Corwin 9910 Old Placer- Ville Road	2 May 84	Carbon Tetrachloride	6.4	5.0
19	SMF Properties 9938 Old Placer- Ville Road	14 Dec 81	None		
20	Yokoi 9970 Old Placer- ville Road	22 Mar 82	Trichloroethylene (West Well)	1.4	5.0
			Carbon Tetrachloride (East Well)	2.6	5.0

WELL NO.	OWNER/ADDRESS	SAMPLE DATE	VOLATILE ORGANIC COMPOUND DETECTED	CONCENTRATION DETECTED (ppb)	STATE ACTION LEVEL (ppb)
		22 Aug 83	Trans-1,2-Dichloroethylene (West Well)	1.8	None
			1,2-Dichloroethane	0.5	0.1
		28 Oct 83	Carbon Tetrachloride	3.0	5.0
			Tetrachloroethylene (No Well Location Given)	3.1	4.0
		2 May 84	Carbon Tetrachloride (West Well)	7.8	5.0
			Tetrachloroethylene	9.2	4.0
			Carbon Tetrachloride	4.4	5.0
			Tetrachloroethylene (East Well)	3.4	4.0
21	Sutter 3590 Bradshaw Road	26 Apr 84	None		
22	Ogawa 9721 Farm Lane	6 Feb 80	None		
23	Panamura 9841 Farm Lane	5 Feb 80	None		

SITE NO.	OWNER/ADDRESS	SAMPLE DATE	VOLATILE ORGANIC COMPOUND DETECTED	CONCENTRATION DETECTED (ppb)	STATE ACTION LEVEL (ppb)
24	Sacramento County Office of Education 3408 Excelsior Rd	28 Oct 83	None		
25	Kiefer (Sacramento County) Dump East of Sunrise-Kiefer intersection	6 Feb 80	None		
26	Granite Construction Kiefer/Bradshaw Rd	26 Apr 84	None		
27	Site not yet evaluated				
28	Sacramento County Branch Center (Two Wells) 3701 Branch Center Road	28 Oct 83	None		
29	Chady Oaks Mobile Home Park	6 Feb 80	None		

STATE ACTION LEVEL
(ppb)

CONCENTRATION DETECTED
(ppb)

VOLATILE ORGANIC COMPOUND
DETECTED

SAMPLE
DATE

OWNER/ADDRESS

ID
No.

None

22 Aug 83

Satow
10122 Ellenwood

30

0.1

0.9

1,2 Dichloroethylene

Jun 84

Hayashi
3951 Happy Lane

31

5.0

15.0

Trichloroethylene

APPENDIX N

LABORATORY RESULTS - 1984 SAMPLE

NOTE: All data marked by * are invalid due to missed holding times or incorrect protocol.



inter-office memorandum

TO: Fred Bopp

DATE: May 22, 1984

FROM: Don Baker *DHB*

SUBJECT: Volatile Organic Analysis
Mather Air Force Base, CA

W. O. No.:

Please find attached the results of volatile organic analysis for Mather AFB, CA. These data were the results of analysis on groundwater samples and soil samples taken during the week of April 30, 1984. The analysis was completed by EPA methods 601 and 602.

The data reported are compounds found in detectable quantity all other 601/602 compounds are below the detection limit listed in table of 601/602 parameters attached hereto.

Note that well JTC is above the state action level of 0.1 ug/l for 1,2-Dichloroethane and well ACW approaches the TCE action level of 5.0 ug/l.

These data have been confirmed by a second column and five point calibrations were run. Appropriate blanks were run to prevent carryover.

These data have been transmitted to Dennis Korncinski at Mather per Pete Marks instructions.

DHB:mr

cc: Dennis Korncinski

MATHER AFB
 VOLATILE ORGANIC ANALYSIS
 METHOD EPA 601/602

<u>SAMPLE SITE</u>	<u>ANALYTE</u>	<u>RESULTS (ug/l)</u>
K9	No peaks found	
ACW	Trichloroethylene	3.6
	Unidentified peaks	3
JTC	1,2 Dichloroethane	3.6
FH3	No peaks found	
FH4	No peaks found	
FH5	No peaks found	
FH6	No peaks found	
GC1	No peaks found	
FH1	No peaks found	
MB4	No peaks found	
MB1	No peaks found	
MB2	No peaks found	
FH2	No peaks found	
DS	1,1 Dichloroethane	0.1
Soil Sample	1,2 Dichloroethane	0.4
	Trichloroethylene	0.1
	1,3 Dichlorobenzene	0.9

MATHER AFB
VOLATILE ORGANIC ANALYSIS
METHOD EPA 601/602

<u>SAMPLE SITE</u>	<u>ANALYTE</u>	<u>RESULTS (ug/l)</u>
Soil Sample	1,4 Dichlorobenzene	0.8
	Unidentified peaks	1
Soil Sample 15	1,3 Dichlorobenzene	1.4
	1,4 Dichlorobenzene	1.2
	Ethylbenzene	16
	Unidentified peaks	0
Unknown solvent mixture detected in sample		
DS-1	1,1 Dichloroethane	0.9
Soil Sample	1,2 Dichloroethane	0.3
	1,1,1 Trichloroethane	0.3
	1,3 Dichlorobenzene	2.5
	1,4 Dichlorobenzene	0.9
	Unidentified peaks	1
MB3	No peaks found	
Field Blank	No peaks found	
Field Blank	No peaks found	
TW-8	Methylene chloride	4.2
	1,1-Dichloroethane	2.3
	Trans-1,2-Dichloroethylene	1.1
	1,2-Dichloroethane	0.6
	1,1,1-Trichloroethane	3.2
	Trichloroethylene	17
	Tetra chloroethene	2.3
Chlorobenzene	0.9	

MATHER AFB
VOLATILE ORGANIC ANALYSIS
METHOD EPA 601/602

<u>SAMPLE SITE</u>	<u>ANALYTE</u>	<u>RESULTS (ug/l)</u>
	1,3-Dichlorobenzene	0.8
	1,2-Dichlorobenzene	0.4
	Unidentified Peaks	8
TW-9	Methylene chloride	3.2
	1,1-Dichloroethylene	5.8
	Trans 1,2 Dichloroethylene	1.4
	1,1,1-Trichloroethylene	1.4
	Trichloroethylene	40
	Unidentified Peaks	3
TW-10	Methylene chloride	2.4
TW-11	No peaks found	
TW-1	Trans-1,2-Dichloroethylene	0.5
	Chloroform	6.2
	1,2-Dichloroethane	0.3
	1,1,1-Trichloroethane	0.4
	Carbon tetrachloride	1.4
	Trichloroethylene	590
	Tetrachloroethene	0.1
	Unidentified Peaks	4
TW-2	Methylene chloride	1.0
	Chloroform	0.2
	Trichloroethylene	48
	Unidentified Peaks	1

MATHER AFB
VOLATILE ORGANIC ANALYSIS
METHOD EPA 601/602

<u>SAMPLE SITE</u>	<u>ANALYTE</u>	<u>RESULTS (ug/l)</u>
TW-3	Chloroform	0.2
	1,1,1-Trichloroethane	0.2
	Carbon tetrachloride	0.3
	Trichloroethylene	130
	Unidentified Peaks	2
TW-4	Unidentified Peaks	7
TW-5	Tetrachlorethylene	0.3
	Unidentified Peaks	1
TW-6	Unidentified Peaks	1
TW-7	Methylene chloride	2.2
	Trans-1,2-Dichlorethylene	0.3
	Trichloroethylene	0.8
	Unidentified Peaks	8



EPA METHOD 601 HALL DETECTOR

EPA METHOD 602 FID DETECTOR

<u>COMPOUND</u>	<u>DETECTION LIMIT ug/l</u>
Chloromethane	0.08
Bromomethane	1.18
Dichlorodifluoromethane	1.81
Vinyl chloride	0.18
Chloroethane	0.52
Methylene chloride	0.25
Trichlorofluoromethane	1.0
1,1 Dichloroethene	0.13
1,1 Dichloroethane	0.27
Trans 1,2 Dichloroethene	0.10
Chloroform	0.05
1,2 Dichloroethane	0.03
1,1,1-Trichloroethane	0.03
Carbon tetrachloride	0.12
Bromodichloromethane	0.10
1,2 Dichloropropane	0.04
Trans 1,3-Dichloropropene	0.34
Trichloroethene	0.12
Dibromochloromethane	0.09
1,1,2 Trichloroethane	0.02
Cis 1,3-Dichloropropene	0.20
2-Chloroethylvinylether	0.13
Bromoform	0.20
1,1,2,2-Tetrachloroethane	0.03
Tetrachloroethene	0.03
Chlorobenzene	0.25
1,3 Dichlorobenzene	0.32
1,2 Dichlorobenzene	0.15
1,4 Dichlorobenzene	0.24
Benzene	10
Toluene	10
Ethylbenzene	10

Notes: FID optional detector for Method 602

Reference: EPA Methods for Organic Chemical Analysis of
Municipal and Industrial Wastewater
EPA 600 14-82-057 - July 1982



inter-office memorandum

TO: FRED BOPP

DATE: JULY 30, 1984

FROM: DON BAKER *DB*

SUBJECT: ANALYSIS OF MATHER AFB
SAMPLES TAKEN MAY 1984

W. O. No.:

Dear Fred,

Enclosed please find the analysis for PCB's, Pesticides and Herbicides on the Mather AFB samples. Also enclosed are copies of reports we have already sent on WDA's and oil and grease.

Phencls were run on the following samples:

JTC, JTC-1, DS, DS-10S, TW6, TW7, TW8, TW9, TW10, TW11*

All were non-detectable at the following limits:

2-Chloro	2.0	<i>(ug/L)</i>
2-Nitro	2.5	
Phenol	1.4	
2,4-Dimethyl	1.7	
2,4-Dichloro	2.1	
2,4,6-Trichloro	5.0	
4-Chloro-3-methyl	8.3	
2,4-Dinitro	7.0	
2-methyl-4,6-dinitro	10	
Pentachloro	10	
4-Nitro	10	

PCB's were run on FB1, TW3, TW2, TW1 and ACW. All were non-detectable at the following limits:

FRED BOPP
Page Two
7-30-84

<u>PARAMETER</u>	<u>DETECTION LIMIT (µg/l)</u>
PCB - 1016	0.04
PCB - 1221	0.10
PCB - 1232	0.10
PCB - 1242	0.05
PCB - 1248	0.08
PCB - 1254	0.08
PCB - 1260	0.15

Field blanks and duplicates are identified as:

<u>FIELD #</u>	<u>SITE I.D. ASSIGNED</u>	<u>ACTUAL SITE I.D.</u>
0019W	T.W. 12	T.W. 4
0028W	MBW 5	MBW 3
0040W	LTC-1	J.T.C.
0043W	DS-1	D.S.
0015W	FB1	Field Blank
0017W	FB2	"

Soil samples were identified as D.S. for downstream and U.S. for upstream.

Pesticides and herbicides were run on MBW2, MBW5, TW6, TW5, TW4, MBW3, MBW1. All were non-detectable at the following limits: *

<u>PARAMETER</u>	<u>DETECTION LIMIT (µg/l)</u>
DBCP	0.1
EOB	1.0
ABHC	0.05
B-BHC	0.05
G-BHC	0.05
D-BHC	0.05
Heptachlor	0.05
Aldrin	0.05
Endosul I	0.05
Hept. Epox	0.05
4,4 DDE	0.05
Dieldrin	0.05
Endrin	0.05
4,4 DDT	0.05
Endrin Ald.	0.05
Endos Sul	0.05
Chlordane	0.5
Toxaphene	5.0



DATE OF FINAL REPORT: 3 January 1985

MATHER A.F.B. -1st ROUND

TOTAL METALS*

b.) R.F.W. NO.	SAMPLE DESCRIPTION	Ag ug/L	Cd ug/L	Cr ug/L	Pb ug/L	Ni ug/L
8405-273-0410	TW7	<2.5	12.6	<10	33	<100
-0420	TW4	<2.5	<2.5	<10	12	<100
-0430	MB2	<2.5	<2.5	<10	<10	<100
-0440	MB1	<2.5	<2.5	<10	<10	<100
-0450	TW5	<2.5	<2.5	<10	<10	<100
-0460	JTC	<2.5	<2.5	<10	<10	<100
-0470	TW11	<2.5	12.8	13	<10	<100
-0480	TW6	<2.5	<2.5	12	<10	<100
-0490	MB5	<2.5	<2.5	<10	15	<100
8405-273-0530	TW12	<2.5	<2.5	<10	21	<100
-0540	TW9	<2.5	<2.5	<10	36	<100
-0550	FB2	<2.5	10.0	<10	<10	<100
-0560	TW10	<2.5	9.9	<10	15	<100
-0570	TW8	<2.5	7.9	<10	47	<100
-0580	MB3	<2.5	<2.5	<10	<10	<100
-0590	FB1	<2.5	<2.5	<10	<10	<100

R.F.W. NO.	SAMPLE DESCRIPTION	TOTAL CN, ug/g
8405-273-0500	US (SOIL)	<0.1
-0510	DS (SOIL)	0.26
-0520	DS-1 (SOIL)	0.10

NAHTR AFB - Sampling of May 1984 and August 1984

Sample ID	Date Sampled	Date Extracted		Date Analyzed		DDT ug/L	Chlordane ug/L	2,4-D ug/L
		Pesticides	Herbicides	Pesticides	Herbicides			
Detection Limit for May sampling*								
MB1	5/7/84	5/7/84	5/8/84	5/8/84	5/9/84	ND	ND	ND
MB2	5/7/84	5/8/84	5/8/84	5/8/84	5/9/84	ND	ND	ND
MB3	5/4/84	5/7/84	5/8/84	5/8/84	5/9/84	ND	ND	ND
MB5	5/7/84	5/8/84	5/8/84	5/8/84	5/9/84	ND	ND	ND
TW4	5/3/84	5/7/84	5/8/84	5/8/84	5/9/84	ND	ND	ND
TW5	5/3/84	5/7/84	5/8/84	5/8/84	5/9/84	ND	ND	ND
TW6	5/3/84	5/7/84	5/8/84	5/8/84	5/9/84	ND	ND	ND
Detection Limit for August sampling								
MAFB4	8/16/84	8/21/84	8/20/84	8/21/84	8/22/84	ND	ND	ND
MAFB5	8/16/84	8/21/84	8/20/84	8/22/84	8/22/84	ND	ND	ND
MAFB6	8/16/84	8/21/84	8/20/84	8/22/84	8/22/84	ND	ND	ND

MAJLIER AFB - Extraction and analysis dates

Sample ID	Date Sampled	Pesticides Extracted	Pesticides Analyzed	Herbicides Extracted	Herbicides Analyzed	Extracted	ICB Analyzed
FB-1	8/14/84	-	-	-	-	8/17/84	8/20/84
MAFB-1	8/15/84	-	-	-	-	8/21/84	8/22/84
MAFB-2	"	-	-	-	-	"	"
MAFB-3	"	-	-	-	-	"	"
Mather 2	9/27/84	-	-	-	-	10/1/84	10/3/84
Mather 3	"	-	-	-	-	"	"
Mather 4	"	10/4/84	10/11/84	10/5/84	10/18/84	-	-
Mather 5	"	"	"	"	"	-	-
FB-1	10/2/84	10/5/84	10/11/84	10/5/84	10/18/84	10/5/84	10/11/84
MAFB-1	"	-	-	-	-	"	"
MAFB-6	"	10/5/84	10/11/84	10/5/84	10/18/84	-	-

ROUND 2



inter-office memorandum

TO: Katherine Sheedy
cc: Alison Dunn

DATE: October 16, 1985

FROM: David Ben-Hur *DB*

SUBJECT: Determination of vinyl chloride W. O. No.:
in a Mather AFB sample.

Sample MAFB-8 collected on August 15, 1984 at Mather AFB was reported to contain 170 ug/L of vinyl chloride. Reexamination of the chromatograms of that sample indicate that the identification was erroneous. The sample does not contain vinyl chloride.



inter-office memorandum

TO: Katherine Sheedy
cc: Alison Dunn

DATE: October 15, 1985

FROM: David Ben-Hur *DB*

SUBJECT: Mather AFB

W. O. No.:

The data from the first and second rounds of sampling at Mather AFB have been reviewed. In my estimation the finding of toluene in the first round and benzene in the second round is correct. Some of the values reported, however, are incorrect. The following table is a list of the corrections. Where a dash appears, there is no change from the previously reported value.

First Round

<u>Sample</u>	<u>Benzene</u>	<u>Toluene</u>	<u>Ethylbenzene</u>
MAFB-1	-	-	-
MAFB-2	-	-	-
MAFB-3	-	-	-
MAFB-4	-	-	-
MAFB-5	-	-	-
MAFB-6	-	-	-
MAFB-7	-	-	-
MAFB-8	-	ND	-
MAFB-9	-	-	-
MAFB-10	-	-	-
MAFB-11	-	-	-
MAFB-12	-	-	-
MAFB-13	-	-	-
MAFB-14	-	NI	-
MAFB-15	-	-	-
EB-1	ND	-	-
EB-2	-	-	-
E-1	-	-	-
E-2	-	-	-
E-3	-	-	-
E-4	-	-	-
E-40	-	-	-
EH-1	-	-	-
EH-2	-	-	-
EH-3	-	-	-
EH-6	-	-	-
EH-10	-	-	-

First Round (continued)

<u>Sample</u>	<u>Benzene</u>	<u>Toluene</u>	<u>Ethylbenzene</u>
GC-1	-	-	ND
GC-2	-	-	-
GC-20	-	-	-
JTC	-	-	-
K-9	-	-	-

Second Round

MAFB-1	0.21	-	-
MAFB-2	0.21	-	-
MAFB-3	-	-	-
MAFB-4	-	-	-
MAFB-5	-	-	-
MAFB-6	-	-	-
MAFB-7	-	-	-
MAFB-8	-	-	-
MAFB-9	0.48	-	-
MAFB-10	-	-	-
MAFB-11	-	-	-
MAFB-30	-	-	-
MAFB-50	-	-	-
MAFB-90	0.32	-	-
MAFB-100	0.20	-	-
FB-1	0.28	-	-
TH-5	-	-	-
AW	-	-	-
AW-1	0.42	-	-



inter-office memorandum

TO: FRED BOPP

DATE: 8-27-84

FROM:

DON H. BAKER

SUBJECT: MATHER AFB GROUNDWATER ANALYSIS
VOLATILE ORGANIC ANALYSIS (VOA)

W. O. No.:

Please find attached the results of the second round of Mather AFB Groundwater Samples for VOA's by EPA Method 601 and 602. We have reported only those compounds found above the detection limit. All other 601 and 602 parameters were not found above the attached detection limits.

It is noted that samples from Wells 1,2,3,8, and 9 indicate trichloroethylene concentrations above the 5.0 ug/l State action level and the Well 9 sample contained bromomethane. Well 8 results indicate a contamination of vinyl chloride and chlorobenzene. These results have been confirmed by second column conformation.

As the results of a phone conversation with Captain Ed Barnes in early July, I am sending copies of these data to Captain Barnes at OEHL and to Captain Jim Curran at Mather AFB.

No corrections have been made for the blanks.

cc: Capt. Ed Barnes
Capt. Jim Curran

MATHER WPI
 VOLATILE ORGANIC ANALYSIS FOR WATERB
 EPA METHOD 81602

LAB ID	FIELD NO.	UNIDENTIFIED PEAK(S)	PARAMETER	RESULTS (UG/L)
S 0981-82	MAFB-1	1	METHYLENE CHLORIDE	0.89
			CHLOROFORM	1.5
			1,2-DICHLOROETHANE	0.49
			CARBON TETRACHLORIDE	0.39
			TRICHLOROETHENE	1100
			TETRACHLOROETHENE	0.16
CHLOROBENZENE	2.2			
S 0982-82	MAFB-2	1	METHYLENE CHLORIDE	0.92
			CHLOROFORM	1.12
			1,2-DICHLOROETHANE	0.14
			TRICHLOROETHENE	17
S 0983-82	MAFB-3	1	METHYLENE CHLORIDE	0.92
			CHLOROFORM	0.16
			1,2-DICHLOROETHANE	0.17
			CARBON TETRACHLORIDE	0.12
			TRICHLOROETHENE	B1
S 0984	MAFB-4	1	METHYLENE CHLORIDE	2.1
			TRICHLOROETHENE	0.12
S 0985	MAFB-5	1	METHYLENE CHLORIDE	1.97
			TETRACHLOROETHENE	0.42
S 0986	MAFB-6	1	METHYLENE CHLORIDE	0.97
			TRICHLOROETHENE	1.15
S 0987-85	MAFB-7	1	METHYLENE CHLORIDE	2.7
			1,2-DICHLOROETHENE	1.29
			CHLOROFORM	0.18
			1,2-DICHLOROPROPANE	0.05
			CHLOROBENZENE	0.22
S 0988-85	MAFB-8	1	VINYL CHLORIDE	170
			METHYLENE CHLORIDE	1.71
			1,1-DICHLOROETHENE	0.6
			1,2-DICHLOROETHANE	0.8
			TRANS-1,2-DICHLOROETHENE	6.5
			CHLOROFORM	1.16
			1,1-DICHLOROETHANE	1.4
			1,1,1-TRICHLOROETHANE	1.89
			1,1-DICHLOROPROPANE	1.11
			TRICHLOROETHENE	1.2
			TETRACHLOROETHENE	1.1
			CHLOROBENZENE	8.1
			1,2-DICHLOROBENZENE	1.1
S 0989	MAFB-9	1	ETHANE METHANE	49
			METHYLENE CHLORIDE	1.4
			1,2-DICHLOROETHENE	1.1
			1,1-DICHLOROETHANE	1.1

10-107

MATHER AFB
 VOLATILE ORGANIC ANALYSIS FOR WATERS
 EPA METHOD 801.001

LAB ID	FIELD NO.	UNIDENTIFIED PEAKS	PARAMETER	RESULTS (UG/L)
B-100002	MAFB-9	1	TRANS-1,2-DICHLOROETHENE 1,1,1-TRICHLOROETHANE TRICHLOROETHENE TETRACHLOROETHENE CHLOROBENZENE	0.08 1.6 9.4 0.4 1.7
B-100004	MAFB-11	1	METHYLENE CHLORIDE 1,1-DICHLOROETHANE 1,2-DICHLOROETHANE TRICHLOROETHENE TETRACHLOROETHENE	1.4 0.19 0.05 0.94 0.00
B-100005	MAFB-10	1	METHYLENE CHLORIDE 1,1-DICHLOROETHANE TRICHLOROETHENE BROMOFORM	1.1 0.07 1.1 1
B-100006	MAFB-10	1	METHYLENE CHLORIDE 1,2-DICHLOROETHANE	1.7 0.07
B-100007	MAFB-11	1	METHYLENE CHLORIDE CHLOROFORM TRICHLOROETHENE	1.0 0.14 0.10
B-100008	MAFB-10	1	METHYLENE CHLORIDE	1.7
B-100009	FB-1	1	METHYLENE CHLORIDE 1,1-DICHLOROETHANE CHLOROFORM	1.0 0.05 0.05
	BLANK B-10	1	METHYLENE CHLORIDE CHLOROFORM	1.0 0.05



inter-office memorandum

TO: Fred Bopp

DATE: September 21, 1984

FROM: Maggie Neckels *MN*

SUBJECT: Analysis of Mather AFB Samples
taken August, 1984 (ROUND 2)

W. O. No.:

Dear Fred:

Enclosed please find the results for PCB, Pesticide, Herbicide, and Phenols (method 604) on the Mather AFB samples.

All data concerning field identification of samples, blanks and duplicates has been sent to Alison Dunn per her request.

September 21, 1984

PCB's were run on MAFB-1, -2, -3 and FB-1. All were not detectable at the following limits.

<u>PARAMETER</u>	<u>DETECTION LIMIT (ug/l)</u>
PCB - 1016	0.04
PCB - 1221	0.10
PCB - 1232	0.10
PCB - 1242	0.05
PCB - 1248	0.08
PCB - 1254	0.08
PCB - 1260	0.15

Pesticides and herbicides were run on MAFB-4, -5, -6 and FB-1. All were non detectable at the following limits.

<u>PARAMETER</u>	<u>DETECTION LIMIT (ug/l)</u>
DBCP	0.1
EDB	1.0
ABHC	0.05
B-BHC	0.05
G-BHC	0.05
D-BHC	0.05
Heptachlor	0.05
Aldrin	0.05
Endosul I	0.05
Hept. Epox	0.05
4,4 DDE	0.05
Dieldrin	0.05
Endrin	0.05
4,4 DDT	0.05
Endrin Ald.	0.05
Endos Sul	0.05
Chlordane	0.5
Toxaphene	5.0



Date of Final Report: January 2, 1985

MATHER A.F.B
2nd ROUND SAMPLING RESULTS
SAMPLES COLLECTED: AUGUST 14 TO AUGUST 20, 1984

I. TOC ANALYSIS

A] These samples were received by the laboratory on August 22, 1984 and analyzed on August 27, 1984. The detection limit for these samples was 1 mg/L. The found values follow:

R.F.W.NO.	SAMPLE NO.	SITE ID	TOC, mg/L
8408-588-0010	0129	MAFB-1	<1.0
8408-588-0020	0130	MAFB-2	<1.0
8408-588-0030	0131	MAFB-3	<1.0
8408-588-0040	0132	MAFB-7	10.3
8408-588-0050	0133	MAFB-8	6.3
8408-588-0060	0134	MAFB-9	7.8
8408-588-0070	0135	MAFB-10	1.4
8408-588-0080	0136	MAFB-11	<1.0
8408-588-0090	0137	MAFB-4	1.5
8408-588-0100	0138	MAFB-5	1.5
8408-588-0110	0139	MAFB-6	2.0
8408-588-0120	0140	FB-1	1.0
8408-588-0130	0144	FB-2	<1.0
8408-588-0140	0145	MAFB-4 (dup)	<1.0
8408-588-0150	0146	MAFB-6 (dup)	1.8
8408-588-0160	0147	MB-1	1.3
8408-588-0170	0148	MB-2	<1.0
8408-588-0180	0149	MB-3	10.3
8408-588-0190	0150	MB-4	<1.0
8408-588-0200	0151	FH-1	4.8
8408-588-0210	0152	FH-2	1.3
8408-588-0220	0153	FH-3	2.7
8408-588-0230	0155	FH-5	<1.0
8408-588-0240	0156	FH-6	<1.0
8408-588-0250	0157	GC-1	<1.0
8408-588-0260	0158	GC-2	9.0
8408-588-0270	0159	K-9	<1.0
8408-588-0280	0160	ACW	3.6
8408-588-0290	0161	JTC	2.0



Date of Final Report: January 2, 1985

Mather A.F.B. - 2nd Round (con't)

IV. DIMETHYLNITROSAMINE (DMN) ANALYSIS

a) These samples were received by the laboratory on August 21, 1984 and were extracted August 24, 1984. Analysis was completed October 3, 1984. There is a 40 day holding time between date of extraction and date of analysis. Holding times were not exceeded. As noted in the 1st round, the 1 ug/L requested detection limit was not met in some cases .

Sample concentration values follow:

R.F.W. NO:	SAMPLE NO.	SITE ID	DMN, ug/L
8408-588-0090	0137	MAFB-4	<1
8408-588-0100	0138	MAFB-5	<2 *
8408-588-0110	0139	MAFB-6	<1
8408-588-0120	0140	FB-1	<1
8408-588-0130	0144	FB-2	<1
8408-588-0160	0147	MB-1	<1
8408-588-0170	0148	MB-2	<1
8408-588-0180	0149	MB-3	<1
8408-588-0190	0150	MB-4	<2 *

OIL AND GREASE RESULTS *

MATHER AIR FORCE BASE

Site Identification	Oil & Grease * mg/L	Extraction Date	Weston Lab No.
MAFB-1	0.76	8/29/84	S-0984
MAFB-1 D	< 0.33	8/30	S-1128
MAFB-2	0.86	8/29	S-0989
MAFB-2 D	0.68	8/30	S-1129
MAFB-3	0.52	8/30	S-1130
MAFB-3 D	< 0.33	8/29	S-0994
MAFB-4	< 0.33	8/30	S-1024
MAFB-4 D	0.48	8/30	S-1131
MAFB-5	0.55	8/29	S-1014
MAFB-6	0.39	8/29	S-1018
MAFB-7	< 0.33	8/29	S-0977
MAFB-8	< 0.33	8/29	S-0999
MAFB-9	< 0.33	8/29	S-1004
MAFB-10	0.35	8/29	S-0979
MAFB-11	< 0.33	8/29	S-0975
MAFB-12	< 0.33	8/29	S-1020
MAFB-13	< 0.33	8/29	S-1019

mg/kg *

0141 Ditch upstream	600	8/29	S-0954
0142 Downstream	700	8/29	S-0956
0143 Downstream	550	8/29	S-0955

ROUND 3



inter-office memorandum

TO: FRED BOPP

DATE: 11-05-84

FROM: MAGGIE NECKELS *(initials)*

SUBJECT: MATHER AFB THIRD ROUND RESULTS

W. O. No.:

Enclosed please find the analytical results for the third round sampling at Mather A.F.B.

Wells 4, 5, 6, and FB1 were run for 4-4 DDT, 2,4-D and chlordanes. All were not detected at the following limits.*

4-4 DDT	0.02 $\mu\text{g/l}$
2,4-D	0.02 "
Chlordane	0.06 "

Wells 1, 2, 3, and FB1 were run for PCB's. All were not detected at the following limits:

PCB 1016	0.4 mg/l
PCB 1221	1.0
PCB 1232	1.0
PCB 1242	0.5
PCB 1245	0.8
PCB 1254	0.8
PCB 1260	1.5

WESTON

DATE OF INTERIM REPORT: 18 January, 1985

MATHER A.F.B
3rd ROUND SAMPLING RESULTS - SOIL SAMPLES
DATE SAMPLES COLLECTED: 27 SEPTEMBER, 1984

II. TOC ANALYSIS*

a. These samples were received by the laboratory on 4 October, 1984 and analyses was completed on 17 January, 1985. Sample concentrations follow:

b. R.F.W. NO:	SAMPLE DESCRIPTION	TOC ($\mu\text{g}/\text{kg}$)
8410-708-0010	0206 West Ditch - Upstream	6740
8410-708-0020	0207 West Ditch - Downstream	4270
8410-708-0030	0208 West Ditch - Downstream	821



MATIER A.F.B.-3RD ROUND WATER SAMPLES (cont.)

III. Soluble Metals Analysis

a. These samples were received by the laboratory on 6 October 1984 and analyzed within recommended holding times. Requested detection limits were met and are indicated by "less than" signs. These samples were field filtered prior to receipt by the laboratory. Sample concentrations follow.

b.

R.F.W. NO.	SAMPLE DESCRIPTION	SITE ID	SOLUBLE METALS				
			Cr µg/L	Pb µg/L	Cd µg/L	Ni µg/L	Ag µg/L
8410-720-0040	0212 MAFB-4		<50	<10	<2.5	<100	<2.5
-0050	0213 MAFB-5		<50	<10	<2.5	<100	<2.5
-0060	0214 MAFB-6		<50	<10	<2.5	<100	<2.5
-0070	0215 MAFB-7		<50	<10	<2.5	<100	<2.5
-0080	0216 MAFB-8		<50	<10	<2.5	<100	<2.5
-0090	0217 MAFB-9		<50	<10	<2.5	<100	<2.5
-0100	0218 MAFB-10		<50	<10	<2.5	<100	<2.5
-0110	0219 MAFB-11		<50	<10	<2.5	<100	<2.5
-0140	0222 FB-1		<50	<10	<2.5	<100	<2.5

No results for DMN reported as of the date of this report.



MATHER A.F.B.
3RD ROUND SAMPLING RESULTS-SOIL SAMPLES
DATE SAMPLES COLLECTED: 27 SEPTEMBER 1984

I Total Phenolics Analysis*

a. These samples were received by the laboratory on 4 October 1984. Sample number 0205 JTC was analyzed on 21 November 1984 and samples 0206, 0207, and 0208 on 31 December 1984. EPA Method 420.1 is applicable to the analysis of drinking, surface, and saline waters and specifies a holding time of 28 days. The method is sensitive to 5 µg/L for water samples. Therefore, sample 0205 JTC exceeded recommended holding times and does not meet the requested detection limit of 1 µg/L. Sample concentrations follow.

b.

<u>R.F.W. NO.</u>	<u>SAMPLE DESCRIPTION</u>	<u>CONCENTRATION OF PHENOLICS</u>
8410-708-0010	0206 West ditch-upstream	0.5 µg/g
8410-708-0020	0207 West ditch-downstream	<0.2 µg/g
8410-708-0030	0208 West ditch-downstream	0.3 µg/g
8410-708-0040	0205 JTC	<5 µg/L

NOTE: TOC, chromium, cadmium, lead, nickel and silver results have not been reported as of this date.

WESTON

MATHER A.F.B.
 3RD ROUND SAMPLING RESULTS-SOIL SAMPLES
 DATE SAMPLES COLLECTED: 27 SEPTEMBER, 1984

III TOTAL METALS ANALYSIS

- a) These samples were received by the laboratory on 4 October 1984 and analysis was completed on 23 January 1985. Requested detection limits and recommended EPA holding times were met. Sample concentrations follow.

b)

R.F.W. NO.	SAMPLE DESCRIPTION	TOTAL METALS				
		Ag ug/g	Cd ug/g	Cr ug/g	Pb ug/g	Ni ug/g
8410-708-0010	0206 West Ditch Upstream	4.3	5.2	70.4	184.8	10.2
-0020	0207 West Ditch Downstream	<0.07	1.9	28.0	35.0	26.8
-0030	0208 West Ditch Downstream	<0.07	3.4	43.0	60.1	32.4

ROUND 4



MATHER A.F.B.
4TH ROUND SAMPLING
SAMPLES COLLECTED: 14 to 19 NOVEMBER 1984

I. Total Metals Analysis

a. These samples were received by the laboratory on 27 November 1984 and were analyzed within the recommended holding times. Chain-of-Custody forms did not specify whether these samples were to be analyzed for total or soluble metals and they were logged in for total metals analyses. Since these samples were field filtered prior to receipt by the laboratory, this should not affect the scope of work. Detection limits are indicated by "less than" signs and all requested detection limits were met. Sample concentrations follow.

b.

R.F.W. NO.	SAMPLE DESCRIPTION	TOTAL METALS				
		Cr µg/L	Pb µg/L	Cd µg/L	Ni µg/L	Ag µg/L
8411-882-0010	MAFB-1	<50	<10	<2.5	<100	<2.5
-0020	MAFB-2	<50	<10	<2.5	<100	<2.5
-0030	MAFB-3	<50	<10	<2.5	<100	<2.5
-0040	MAFB-4	<50	<10	<2.5	<100	<2.5
-0050	MAFB-5	<50	<10	<2.5	<100	<2.5
-0060	MAFB-6	<50	<10	<2.5	<100	<2.5
-0070	MAFB-7	<50	<10	<2.5	<100	<2.5
-0080	MAFB-8	<50	<10	<2.5	<100	<2.5
-0090	MAFB-9	<50	<10	4	<100	<2.5
-0100	MAFB-10	<50	<10	<2.5	<100	<2.5
-0110	MAFB-11	<50	16	<2.5	<100	<2.5
-0120	MAFB-12 (MAFB-7 dup.)	<50	<10	<2.5	<100	<2.5
-0130	FB-1	<50	<10	<2.5	<100	<2.5

As of this date no results have been reported for TOC or total phenolics

Date of Report: 14 January 1985

WESTON

MATHER A.F.B.
4TH ROUND SAMPLING
SAMPLES COLLECTED: 14 to 19 NOVEMBER 1984

II. TOC ANALYSIS*

a) These samples were received by the laboratory on 27 November 1984 and analyzed on 4 January 1985 with a detection limit of 1 mg/L. The recommended EPA holding time of 28 days was therefore exceeded by 10 days. The requested detection limit of 1 mg/L was met. Sample concentrations follow.

b) R.F.W. NO.	SAMPLE DESCRIPTION	TOC, mg/L
8411-882-0010	MAFB-1	<1
-0020	MAFB-2	<1
-0030	MAFB-3	<1
-0040	MAFB-4	<1
-0050	MAFB-5	<1
-0060	MAFB-6	<1
-0070	MAFB-7	9
-0080	MAFB-8	4
-0090	MAFB-9	5
-0100	MAFB-10	1
-0110	MAFB-11	1
-0120	MAFB-12	1
-0130	FB-1	1

As of this date no total phenolics results have been reported.

III. DIMETHYLNITROSAMINE (DMN) ANALYSIS

a) These samples were received by the laboratory on 27 November 1984 and extracted on 29 November 1984. Analysis was completed on 13 January 1985. The detection limit of 1 mg/L using EPA METHOD 8171 was not met. The recommended holding times for extraction were exceeded and the requested detection limit was not met. Sample concentrations follow:

AD-A184 581

INSTALLATION RESTORATION PROGRAM PHASE 2
CONFIRMATION/QUANTIFICATION STAGE 1 VOLUME 2 APPENDICES
(U) WESTON (ROY F) INC WEST CHESTER PA JUN 86
F33615-80-D-4006 F/G 24/4

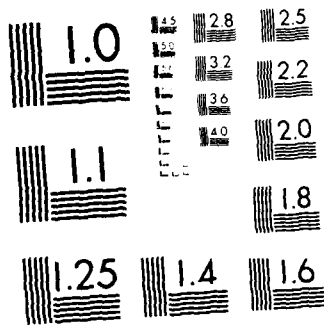
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WESTON

MATHER A.F.B. (4TH ROUND-cont.)

b. R.F.W. NO.	SAMPLE DESCRIPTION	DMN, mg/L
8411-882-0010	MAFB-1	<1
-0020	MAFB-2	<1
-0030	MAFB-3	<1
-0040	MAFB-4	<1
-0050	MAFB-5	<1
-0060	MAFB-6	<1
-0070	MAFB-7	<1
-0080	MAFB-8	<1
-0090	MAFB-9	<1
-0100	MAFB-10	<1
-0110	MAFB-11	<1
-0120	MAFB-12	<1
-0130	FB-1	<1

**DATA
FILM**