

AD-A179 395

INVESTIGATION OF GASEOUS CHLORINE COMPOUNDS BY X-RAY
ABSORPTION SPECTROSCOPY(U) BOEING AEROSPACE CO SEATTLE
WR F W LYTLE ET AL. 28 MAR 87 TR-5 N00014-85-C-0207

1/1

UNCLASSIFIED

F/G 7/4

ML





MI

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE

DTIC FILE COPY

2

REPORT DOCUMENTATION PAGE

AD-A179 395

1. REPORT SECURITY CLASSIFICATION Unclassified		1b. RESTRICTIVE MARKINGS	
2a. SECURITY CLASSIFICATION AUTHORITY		3. DISTRIBUTION/AVAILABILITY OF REPORT Approved for public release and Sale. Distribution unlimited.	
2b. DECLASSIFICATION/DOWNGRADING SCHEDULE			
4. PERFORMING ORGANIZATION REPORT NUMBER(S) ONR Technical Report #5		5. MONITORING ORGANIZATION REPORT NUMBER(S)	
6a. NAME OF PERFORMING ORGANIZATION Boeing Aerospace Company	6b. OFFICE SYMBOL (If applicable)	7a. NAME OF MONITORING ORGANIZATION	
6c. ADDRESS (City, State and ZIP Code) Box. 3999 M/S 87-08 Seattle, WA 98124-2499		7b. ADDRESS (City, State and ZIP Code)	
8a. NAME OF FUNDING/SPONSORING ORGANIZATION Office of Naval Research	8b. OFFICE SYMBOL (If applicable)	9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER N00014-85-C-0207	
10. SOURCE OF FUNDING NOS.		PROGRAM ELEMENT NO.	PROJECT NO.
11. TITLE (Include Security Classification) Investigation of Chlorine Compounds by X-ray Absorption Spectroscopy		TASK NO.	WORK UNIT NO.
PERSONAL AUTHOR(S) Arrel, W. Lytle, Robert B. Greeger, Edward C. Marques, and Donald R. Sandstrom			
13a. TYPE OF REPORT Technical	13b. TIME COVERED FROM Sept. '86 to Aug. '87	14. DATE OF REPORT (Yr., Mo., Day) 3/20/87	15. PAGE COUNT 10
16. SUPPLEMENTARY NOTATION Submission to 1986 SSRL Activity Report			
17. COSATI CODES		18. SUBJECT TERMS (Continue on reverse if necessary and identify by block number)	
FIELD	GROUP	X-ray absorption spectroscopy, chlorine edge, e-yield detection, fluorescence gaseous chlorine, carbon tetrachloride, ethylene chloride, double resonance (Π, σ), C-Cl band distance.	
	SUB. GR.		
19. ABSTRACT (Continue on reverse if necessary and identify by block number) See attached.			
20. DISTRIBUTION/AVAILABILITY OF ABSTRACT UNCLASSIFIED/UNLIMITED <input checked="" type="checkbox"/> SAME AS RPT. <input type="checkbox"/> DTIC USERS <input type="checkbox"/>		21. ABSTRACT SECURITY CLASSIFICATION Unclassified	
22a. NAME OF RESPONSIBLE INDIVIDUAL		22b. TELEPHONE NUMBER (Include Area Code)	22c. OFFICE SYMBOL

DTIC
ELECTE
APR 21 1987
E

OFFICE OF NAVAL RESEARCH

Contract N00014-85-C-0207

R&T Code 413a001---01

Technical Report No. 5

INVESTIGATION OF GASEOUS CHLORINE COMPOUNDS
BY X-RAY ABSORPTION SPECTROSCOPY

by

Farrel W. Lytle, Robert B. Greigor,
Edward C. Marques and Donald R. Sandstrom

Boeing Aerospace Company
Seattle, Washington 98124

published in

1986 SSRL Activity Report
Stanford Synchrotron Radiation Laboratory

March 1987

Reproduction in whole or in part is permitted for
any purpose of the United States Government

*This document has been approved for public release and sale;
its distribution is unlimited.

Accession For	
NTIS GRA&I	<input checked="" type="checkbox"/>
DTIC TAB	<input type="checkbox"/>
Unannounced	<input type="checkbox"/>
Justification	
By _____	
Distribution/	
Availability Codes	
Dist	Avail and/or Special
A-1	



INVESTIGATION OF GASEOUS CHLORINE COMPOUNDS BY X-RAY ABSORPTION SPECTROSCOPY

F. W. Lytle, R. B. Gregor, E. C. Marques and D. R. Sandstrom

The Boeing Co., Seattle, WA 98124

G. P. Huffman, F. E. Huggins, University of Kentucky, Lexington, KY 40506

carbon tetrachloride
chlorine
ethylene chloride

We measured the x-ray absorption spectra of gaseous Cl_2 , CCl_4 and $H_4C_2Cl_2$. The samples were mixed with He at approximately 1000 ppm concentration and flowed through the detector/sample cell. This consisted of a cavity in a 1.3 cm Lucite block covered front and back with 6 μ m aluminized Mylar windows. At the center was a thin electron-collecting grid of Ni mesh. The windows were connected to -45 V while the positive battery terminal was connected to the ground of the electrometer. The e-yield signal was collected at 10^8 gain from the center mesh. Absorption and fluorescent mode data were also collected but were much inferior in quality. The Si(111) double crystal monochromator was detuned 80% to reduce harmonics. A 1 mm entrance slit gave an energy resolution $\Delta E/E=0.5$ eV(2). Early data for Cl_2 gas was published by Stephenson et al.(3). Their data was obtained in the absorption mode, cranking the spectrometer and recording the data by hand. Although their first peak was attenuated by the thickness effect, the spectra are comparable with ours to 10 eV. In the region from 10-24 eV we found an interesting double series resonance which is blown up in scale in the inset to Fig. 1. By analogy to N_2 data(4) these features are due to transitions to unfilled orbitals of the molecule in its various charged states.

The data of Fig. 1 were placed absolutely in energy by noting the impurity Ar 1s resonance at 3203.3 eV(5) at the end of each scan. The zero of energy of Fig. 1. is 2833.4 eV. The π resonance peaks were located at -2.6, 0.0 and 0.2 ± 0.2 eV for Cl_2 , CCl_4 , and $H_4C_2Cl_2$, respectively. The σ resonance energy as defined by Sette et al.(6) moves with bond distance as noted for smaller molecules(6). The Ar K-edge spectrum is shown for comparison, $E_0=3203.3$ eV. Note that no feature similar to the Ar double electron ionization (1s3p) at 23 eV (the vertical arrow) occurs in the Cl spectra. The EXAFS of Fig. 2 was terminated by the ubiquitous Ar impurity in the x-ray path. Considerably more EXAFS could be measured if this could be corrected. The Cl-Cl phase-corrected

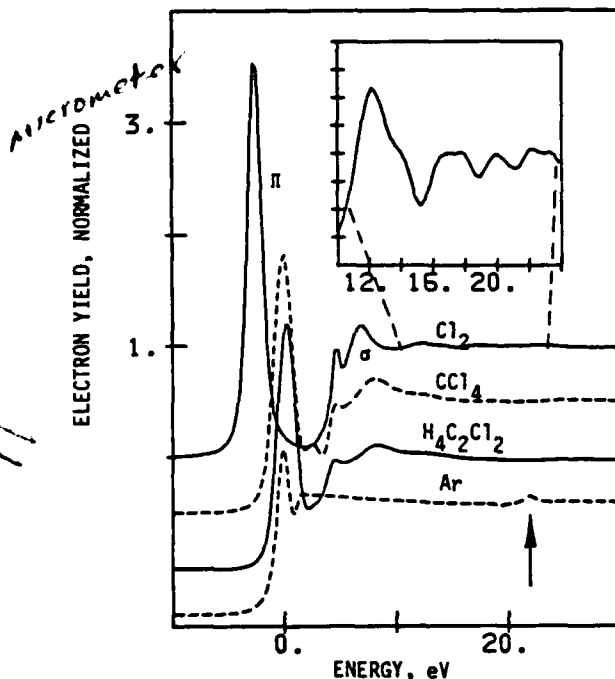


Fig. 1 Near edge spectra of gas phase Cl_2 , CCl_4 , $H_4C_2Cl_2$ and Ar all normalized to unit edge jump. The π and σ maxima are indicated. The Ar 1s3p edge is marked by the vertical arrow. A region of the Cl_2 spectrum is blown up in the inset to illustrate a double series resonance.

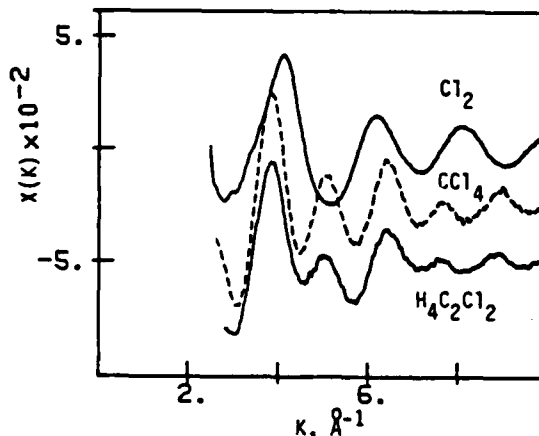


Fig. 2. Normalized EXAFS of gas phase Cl_2 , CCl_4 and $H_4C_2Cl_2$.

1

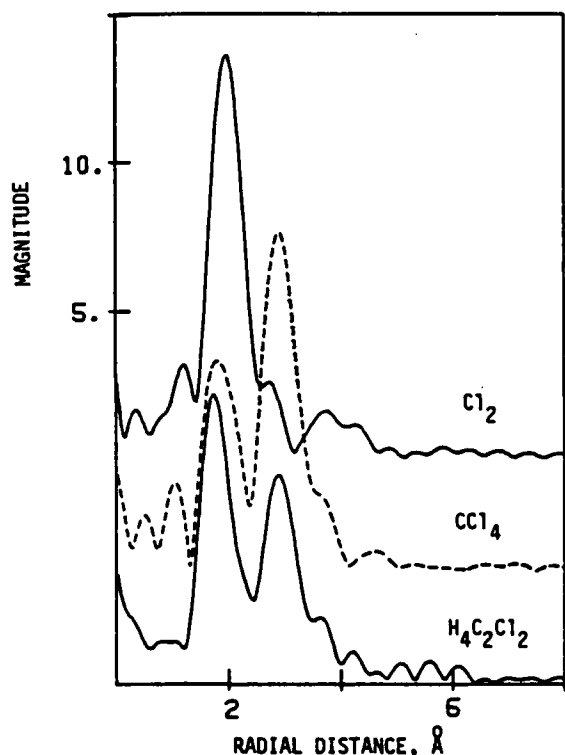


Fig. 3 K^1 , Cl-Cl phase corrected Fourier transforms of gas phase Cl_2 , CCl_4 and $H_4C_2Cl_2$ all plotted to the same scale.

Fourier transforms are given in Fig. 3. The Cl peaks were found at the expected distances of 1.99, 2.90 and 2.97 Å, top to bottom. The shorter Cl-C bonds are clearly resolved and could easily be analyzed. Any sample with appreciable vapor pressure may be introduced into an ion chamber with a diluent gas. With a long beam path a sensitivity of 1 ppm is possible. This remarkable sensitivity occurs because of the nominal 4π collecting efficiency.

Research funded by NSF and ONR

References

1. F. W. Lytle et al., "GAS PHASE X-RAY ABSORPTION SPECTROSCOPY WITH AN ELECTRON YIELD DETECTOR", presented at IVth Int. EXAFS Conf., Fontevraud, France, July 1986.
2. F. W. Lytle et al., Nucl. Inst. Methods 226, 542 (1984).
3. S. T. Stephenson et al., Phys. Rev. 84, 806 (1951).
4. M. Nakamura et al., Phys. Rev. 178, 80 (1969).
A. Bianconi et. al., Chem. Phys. Lett. 58, 263 (1978).
A. P. Hitchcock et al., J. Elec. Spec. 18, 1 (1980).
5. R. D. Deslattes et al., Phys. Rev. A27, 923 (1983)
6. F. Sette et al., Chem. Phys. Lett. 110, 517 (1984)

TECHNICAL REPORT DISTRIBUTION LIST, GEN

	<u>No. Copies</u>		<u>No. Copies</u>
Office of Naval Research Attn: Code 1113 800 N. Quincy Street Arlington, Virginia 22217-5000	2	Dr. David Young Code 334 NORDA NSTL, Mississippi 39529	1
Dr. Bernard Douda Naval Weapons Support Center Code 50C Crane, Indiana 47522-5050	1	Naval Weapons Center Attn: Dr. Ron Atkins Chemistry Division China Lake, California 93555	1
Naval Civil Engineering Laboratory Attn: Dr. R. W. Drisko, Code L52 Port Hueneme, California 93401	1	Scientific Advisor Commandant of the Marine Corps Code RD-1 Washington, D.C. 20380	1
Defense Technical Information Center Building 5, Cameron Station Alexandria, Virginia 22314	12 high quality	U.S. Army Research Office Attn: CRD-AA-IP P.O. Box 12211 Research Triangle Park, NC 27709	1
DTNSRDC Attn: Dr. H. Singerman Applied Chemistry Division Annapolis, Maryland 21401	1	Mr. John Boyle Materials Branch Naval Ship Engineering Center Philadelphia, Pennsylvania 19112	1
Dr. William Tolles Superintendent Chemistry Division, Code 6100 Naval Research Laboratory Washington, D.C. 20375-5000	1	Naval Ocean Systems Center Attn: Dr. S. Yamamoto Marine Sciences Division San Diego, California 91232	1

ABSTRACTS DISTRIBUTION LIST, 359/627

Dr. Paul Delahay
Department of Chemistry
New York University
New York, New York 10003

Dr. J. Driscoll
Lockheed Palo Alto Research
Laboratory
3251 Hanover Street
Palo Alto, California 94304

Dr. D. N. Bennion
Department of Chemical Engineering
Brigham Young University
Provo, Utah 84602

Dr. R. A. Marcus
Department of Chemistry
California Institute of Technology
Pasadena, California 91125

Dr. J. J. Auburn
Bell Laboratories
Murray Hill, New Jersey 07974

Dr. Joseph Singer, Code 302-1
NASA-Lewis
21000 Brookpark Road
Cleveland, Ohio 44135

Dr. P. P. Schmidt
Department of Chemistry
Oakland University
Rochester, Michigan 48063

Dr. Roger Belt
Litton Industries Inc.
Airtron Division
Morris Plains, NJ 07950

Dr. Ulrich Stimming
Department of Chemical Engineering
Columbia University
New York, NY 10027

Dr. Manfred Breiter
Institut für Technische Elektrochemie
Technischen Universität Wien
9 Getreidemarkt, 1160 Wien
AUSTRIA

Dr. E. Yeager
Department of Chemistry
Case Western Reserve University
Cleveland, Ohio 44106

Dr. C. E. Mueller
The Electrochemistry Branch
Naval Surface Weapons Center
White Oak Laboratory
Silver Spring, Maryland 20910

Dr. Sam Perone
Chemistry & Materials
Science Department
Lawrence Livermore National Laboratory
Livermore, California 94550

Dr. Royce W. Murray
Department of Chemistry
University of North Carolina
Chapel Hill, North Carolina 27514

Dr. Adam Heller
Bell Laboratories
Murray Hill, New Jersey 07974

Dr. A. B. Ellis
Chemistry Department
University of Wisconsin
Madison, Wisconsin 53706

Dr. Steven Greenbaum
Department of Physics and Astronomy
Hunter College
695 Park Ave.
New York, NY 10021

ABSTRACTS DISTRIBUTION LIST, 359/627

Dr. M. Wrighton
Chemistry Department
Massachusetts Institute
of Technology
Cambridge, Massachusetts 02139

Dr. B. Stanley Pons
Department of Chemistry
University of Utah
Salt Lake City, Utah 84112

Donald E. Mains
Naval Weapons Support Center
Electrochemical Power Sources Division
Crane, Indiana 47522

S. Ruby
DOE (STOR)
Room 5E036 Forrestal Bldg., CE-14
Washington, D.C. 20595

Dr. A. J. Bard
Department of Chemistry
University of Texas
Austin, Texas 78712

Dr. Janet Osteryoung
Department of Chemistry
State University of New York
Buffalo, New York 14214

Dr. Donald W. Ernst
Naval Surface Weapons Center
Code R-33
White Oak Laboratory
Silver Spring, Maryland 20910

Mr. James R. Moden
Naval Underwater Systems Center
Code 3632
Newport, Rhode Island 02840

Dr. Bernard Spielvogel
U.S. Army Research Office
P.O. Box 12211
Research Triangle Park, NC 27709

Dr. Aaron Fletcher
Naval Weapons Center
Code 3852
China Lake, California 93555

Dr. Michael J. Weaver
Department of Chemistry
Purdue University
West Lafayette, Indiana 47907

Dr. R. David Rauh
EIC Laboratories, Inc.
Norwood, Massachusetts 02062

Dr. Aaron Wold
Department of Chemistry
Brown University
Providence, Rhode Island 02192

Dr. Martin Fleischmann
Department of Chemistry
University of Southampton
Southampton SO9 5MH UNITED KINGDOM

Dr. R. A. Osteryoung
Department of Chemistry
State University of New York
Buffalo, New York 14214

Dr. John Wilkes
Air Force Office of Scientific
Research
Bolling AFB
Washington, D.C. 20332

Dr. D. Rolison
Naval Research Laboratory
Code 6171
Washington, D.C. 20375-5000

Dr. D. F. Shriver
Department of Chemistry
Northwestern University
Evanston, Illinois 60201

Dr. Edward M. Eyring
Department of Chemistry
University of Utah
Salt Lake City, UT 84112

Dr. M. M. Nicholson
Electronics Research Center
Rockwell International
3370 Miraloma Avenue
Anaheim, California

ABSTRACTS DISTRIBUTION LIST, 359/627

Dr. Hector D. Abruna
Department of Chemistry
Cornell University
Ithaca, New York 14853

Dr. A. B. P. Lever
Chemistry Department
York University
Downsview, Ontario M3J1P3

Dr. Stanislaw Szpak
Naval Ocean Systems Center
Code 633, Bayside
San Diego, California 95152

Dr. Gregory Farrington
Department of Materials Science
and Engineering
University of Pennsylvania
Philadelphia, Pennsylvania 19104

M. L. Robertson
Manager, Electrochemical
and Power Sources Division
Naval Weapons Support Center
Crane, Indiana 47522

Dr. T. Marks
Department of Chemistry
Northwestern University
Evanston, Illinois 60201

Dr. Micha Tomkiewicz
Department of Physics
Brooklyn College
Brooklyn, New York 11210

Dr. Lesser Blum
Department of Physics
University of Puerto Rico
Rio Piedras, Puerto Rico 00931

Dr. Joseph Gordon, II
IBM Corporation
5600 Cottle Road
San Jose, California 95193

Dr. Nathan Lewis
Department of Chemistry
Stanford University
Stanford, California 94305

Dr. D. H. Whitmore
Department of Materials Science
Northwestern University
Evanston, Illinois 60201

Dr. Alan Bewick
Department of Chemistry
The University of Southampton
Southampton, SO9 5NH UNITED KINGDOM

Dr. E. Anderson
NAVSEA-56Z33 NC #4
541 Jefferson Davis Highway
Arlington, VA

Dr. Bruce Dunn
Department of Engineering &
Applied Science
University of California
Los Angeles, California 90024

Dr. Elton Cairns
Energy & Environment Division
Lawrence Berkeley Laboratory
University of California
Berkeley, California 94720

Dr. Richard Pollard
Department of Chemical Engineering
University of Houston
Houston, Texas 77004

Dr. M. Philpott
IBM Corporation
5600 Cottle Road
San Jose, California 95193

Dr. Donald Sandstrom
Boeing Aerospace Co.
P.O. Box 3999
Seattle, Washington 98124

Dr. Carl Kannewurf
Department of Electrical Engineering
and Computer Science
Northwestern University
Evanston, Illinois 60201

Dr. Joel Harris
Department of Chemistry
University of Utah
Salt Lake City, Utah 84112

ABSTRACTS DISTRIBUTION LIST, 359/627

Dr. Robert Somoano
Jet Propulsion Laboratory
California Institute of Technology
Pasadena, California 91103

Dr. Johann A. Joebstl
USA Mobility Equipment R&D Command
DRDME-EC
Fort Belvoir, Virginia 22060

Dr. Judith H. Ambrus
NASA Headquarters
M.S. RTS-6
Washington, D.C. 20546

Dr. Albert R. Landgrebe
U.S. Department of Energy
M.S. 68025 Forrestal Building
Washington, D.C. 20595

Dr. J. J. Brophy
Department of Physics
University of Utah
Salt Lake City, Utah 84112

Dr. Charles Martin
Department of Chemistry
Texas A&M University
College Station, Texas 77843

Dr. H. Tachikawa
Department of Chemistry
Jackson State University
Jackson, Mississippi 39217

Dr. Farrell Lytle
Boeing Engineering and
Construction Engineers
P.O. Box 3707
Seattle, Washington 98124

Dr. Robert Gotscholl
U.S. Department of Energy
MS G-226
Washington, D.C. 20545

Dr. Edward Fletcher
Department of Mechanical Engineering
University of Minnesota
Minneapolis, Minnesota 55455

Dr. John Fontanella
Department of Physics
U.S. Naval Academy
Annapolis, Maryland 21402

Dr. Martha Greenblatt
Department of Chemistry
Rutgers University
New Brunswick, New Jersey 08903

Dr. John Wasson
Syntheco, Inc.
Rte 6 - Industrial Pike Road
Gastonia, North Carolina 28052

Dr. Walter Roth
Department of Physics
State University of New York
Albany, New York 12222

Dr. Anthony Sammells
Eltron Research Inc.
4260 Westbrook Drive, Suite 111
Aurora, Illinois 60505

Dr. C. A. Angell
Department of Chemistry
Purdue University
West Lafayette, Indiana 47907

Dr. Thomas Davis
Polymer Science and Standards
Division
National Bureau of Standards
Washington, D.C. 20234

ABSTRACTS DISTRIBUTION LIST, 359/627

Dr. John Owen
Department of Chemistry and
Applied Chemistry
University of Salford
Salford M5 4WT ENGLAND

Dr. J. O. Thomas
University of Uppsala
Institute of Chemistry
Box 531
S-751 21 Uppsala, Sweden

Dr. Boone Owens
Department of Chemical Engineering
and Materials Science
University of Minnesota
Minneapolis, Minnesota 55455

Dr. O. Stafsudd
Department of Electrical Engineering
University of California
Los Angeles, California 90024

Dr. Menahem Anderman
W.R. Grace & Co.
Columbia, MD 20144

END

5-87

DTIC