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A LONG LIFE CO2 LASER

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# A LONG LIFE CO, LASER

by Cai Lijuan, Yu Linxiu and Zhang Kenian (Institute of Laser Technology, Shanghai)

#### Abstract

It is shown experimentally that three major factors must be considered to obtain  $\in CO_2$  laser with long life. They are: 1) elimination of slow gas leakage; 2) selection and oxidation of electrode material; 3) decreasing the decomposition of  $CO_2$ molecules. Our method was: the laser tube is not dried but filled with water vapor, the electrode material must be degassed in a vacuum and oxidized by filling oxygen gas. When pure silver electrodes are employed, the gas mixture constitutions are separately 1)  $C'_{2}$ -N<sub>2</sub>-He-Xe-H<sub>2</sub>O; 2)CO<sub>2</sub>-CO-He-Xe-H<sub>2</sub>O; 3)CO<sub>2</sub>-N<sub>2</sub>-CO-He-Xe-H<sub>2</sub>O, the operating lives reach 1) 10,476 hours; 2) 9,185 hours (still in operation); 3) 9,207 hours (still in operation). When pure nickel electrodes are used, the gas composition is  $CO_2$ -N<sub>2</sub>-He-Xe-H<sub>2</sub>O and the operating life reaches 1,644 hours.

The operating life of a sealed CO, laser is a problem in

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need of fast solution. In the last ten or more years a great deal of domestic and foreign testing has been done on sealed  $CO_2$  lasers and there have been noticeable achievements. Foreign nations have reported that when a silver-copper (internal oxidation) alloy is used as the electrode material in a  $CO_2$ -CO-He-Xe gas mixture, the  $CO_2$  laser output power is 1 watt and its life reaches 12,000 hours.

We drew from foreign tests and at the same time based on the actual circumstances of existing laboratory conditions and operating  $CO_2$  lasers, tests were carried out on lasers with 25-30 watt output power and an 800 millimeter discharge length whereupon an operating life of 10,000 hours was able to be obtained.

#### I. Test Conditions

The serial numbers were A, B, C, D, E, F and G. The effective discharge length was 800 millimeters, the inner diameter of the discharge tube was 8 millimeters, the cavity structure was a plane-spherical cavity, the total reflector curvature radius was 3 meters, the output window plate was a germanium wafer, and there was non-coated uniform output. Fig. 1 shows the structure of the  $CO_2$  laser. The material used was GG17 hard glass.

# Fig. 1 Structure of the CO, Laser

Key: 1. Output window (germanium wafer)

- 2. Cathode
  - 3. Cathode cover
  - 4. Outfall
  - 5. Receiver
  - 6. Water-cooled tube
- 7. Discharge tube (8 millimeter inner diameter)
- 8. Gas return tube
- 9. Water intake
- 10. Anode cover
- 11. Anode
- 12. Reflector place (R is 3 meters, gold plated)

There are two types of electrode material, nickel and silver hollow cylinders, and both must be degassed in a vacuum and oxidized by filling oxygen gas. All of the manufacturing techniques for the lasers are the same. The laser tube is not dried but filled twice with He gas for discharge degassing and  $\sim$  adding H<sub>2</sub>O. Vacuity is 5 x 10<sup>-5</sup> torr. The gas compositions of these seven lasers are listed separately in table 1.

/	-	-	气体比例的					270	ß	
	5	材料	CO3 :	N,	:00:	He	// <u>、</u> • (托)	日 <sub>-</sub> つ (作)	E (托)	
		<b>11</b> 3	1	2		7	0.5	0.2	25	•
	B	<b>#</b> 1	1		1.5	7	0.5		24	
	C	<b>#5</b>	1	8		7	0.5	0.2	25	
	D		1		1.5	7	0.3	J.2	27	
	E		1	1	1	7	0.5	0.2	28	
	7	<b>R</b> 1	1		1.5	7	0.5		24	
•	G		1		1.5	7	0.5		24	

#### Table 1

Key: 1. Serial number
2. Electrode material
3. Nickel
4. Silver
5. Silver
6. Silver
7. Silver
8. Silver
9. Nickel
10. Gas ratio
11. (Torr)
12. (Torr)
13. Total gas pressure (torr)

The discharge is: when the direct current discharge (whole wave integral flow) operates in a discharge current with  $N_2$  in the mixed gas it is 18 to 20 milliamperes and when it is in a CO (when there is no  $N_2$ ) discharge current it is about 16 milliamperes. The operating mode is: it operates continually 24 hours each day, the water coooling temperature is a constant temperature (room temperature), the reflector plate and output window both use water cooling, and when the laser is operating it has a water and electric breaking controller. So-called laser

life is the output power of the component which changes with the operating time. When the laser's output power declines to 70% of the initial power, the operating life is terminated.

## II. Test Results

See table 2 for the long-life test results of various materials used for the sealed  $CO_2$  laser. Fig. 2 shows that the cutput power changes with the time.

		2				
1	11 4	电压 打井	工作寿命 夕(小时)		4	往
		5 .	1,644	寿余终	<u>止(1978</u>	.3.18-5.31) /2
	8	4	2,514	因新水	经新路外	ミズ南非泉 /
	O	<b>×</b>	10,676	***	<u>ا</u> ل ال	→ 1978.8.8) /4 →1979.6.23)
	D	6	9,185	仍在运	<b># (19</b> 7	8.8.31 → 現在) <i>15</i>
	E	74	9,207	のを達	<b># (1</b> 97	8.8.29→現在) 16
	7	84	1,200	***	与纯粹的	的连线力输出/2
		1	T I	(197	7.5.26	→1977.9.21)
	G	19 m.	43	80	j ni s	化学反应的第具/8

Table 2

(ey:	1.	Serial number
•	-	Electrode material
	3.	Nickel
	4.	Silver
	5.	Silver
		Nickel
		Operating life (hours)
		Remarks
		Life terminated (1978.3.18 to 5.31)

13. Blew up because water breaking controller was not working (1977.12.9 to 1978.8.8)

- 14. Life terminated (1978.3.22 to 1978.6.23) 15. Still in operation (1978.8.31 to present)
- 16. Still in operation (1978.8.29 to present)
- 17. Connector wire of silver electrode and tungsten
- rod is nickel wire (1977.5.26 to 1977.9.21)
- 18. Result of chemical reaction caused by CO and Ni





Fig. 2 Output Power Changes With Time

Key: 1.	Output power (watts)
2.	(A) Nickei electrode
	Hours
	Output power (watts)
5.	(B) Silver electrode
6.	Hours
7.	Output power (watts)
_	(C) Silver electrode
	Hours
10.	Output power (watts)
11.	Output power (watts) x 10 hours
	(D) Silver electrode
13.	Output power (watts) x 10 hours
14.	x 10 <sup>°</sup> hours
15.	(E) Silver electrode
16.	Output power (watts)
17	$x 10^2$ hours
	(F) Silver electrode
19.	Output power (watts)
20.	(G) Nickel electrode
	Hours
- 21.	nout a

## III. Discussion

It can be seen from fig. 2 which shows ouput power changes with the time that the operating life varies with different electrode materials and different gas compositions. Based on test results, we think that there are three main factors which must be considered to obtain a sealed CO<sub>2</sub> laser with long-life.

1. Elimination of Slow Gas Leakage

Manufacturing technology for sealed CO<sub>2</sub> lasers must be strict. This includes the conditions that the manufacture of glass semi-finished products cannot have weak links and tungsten sintering. To guarantee all technological aspects when gluing the wafer we can eliminate slow gas leakage.

2. Selection and Oxidation of Electrode Material

As regards the cold cathode of the CO<sub>2</sub> laser, we select a material with very low sputtering rate for the cathode. This type of material does not separate the gas because of the sputtering and avoids undergoing a chemical reaction and consuming any gas compounds. This is very important.

Nickel is a widely used electrode material in sealed  $CO_2$ lasers. Although its sputtering is less than other metals yet when it reactes with the  $CO_2$  and is decomposed into CO the formed carbonyl nickel (Ni(CO)<sub>5</sub>) can spoil the electrode. Moreover, because of the decomposed O which continually consumes  $CO_2$ molecules the CO cannot again form  $CO_2$ . Therefore, when a nickel electrode is used this makes it more difficult for the

sealed CO, laser to obtain long-life.

When we change to use pure silver as the electrode material, because the silver and CO do not cause a reaction and the compound of O and silver is very reactive, when in 200-300°C, the oxide of silver can be reduced to form pure silver. Therefore, the use of a silver electrode can cause the sealed  $CO_2$  laser to reach long-life. Because the sputtering of silver is relatively strong it can be suitable for lengthening the distance from the window to the electrode. In this way, we can decrease or eliminate the sputtering substances on the window plate.

Furthermore, the oxidation of the electrode is very important. If the nickel electrode does not undergo oxidation, its operating life will only be 500-600 hrs but if it undergoes oxidation its operating life will be 1644 hrs.

oxygen gas so that a uniform oxidation layer will first form on the surface of the electrode. The aim is that during the discharge process as little oxygen as possible will be consumed and at the same time this can cause the decomposed product to again synthesize CO<sub>2</sub>. Besides this, after oxidizing the electrode there is little escaped energy of the electrons and there is also a decrease in sputtering. Because there is less absorbed gas caused by sputtering, operation life increases.

3. Decreasing the Decomposition of CO<sub>2</sub> Molecules

During the discharge process, a portion of the CO<sub>2</sub> molecules decompose:



idd≏nly go out. Therefore, when CO is used to replace N<sub>2</sub> and the Ke-H<sub>2</sub>O.

er replacing N<sub>2</sub> with CO, its the current voltage of adding a entron temperature is also se of the decomposition of the b extending long-life. After CO erating life exceeds 5,514 hours already reached 9,185 hours and The output power which changes and (D) of fig. 2.

ng-life and not cause output power ver electrode. There is CO in the that is, CO<sub>2</sub>-N<sub>2</sub>-CO-He-Xe-H<sub>2</sub>O. The 1. Operating life has reached perating normally.

ng pure silver electrodes with a  $N_2$ -He-Xe-H<sub>2</sub>O we can also obtain to 10,476 hours. The output power s shown in (B) of fig. 2. The 2 laser with added  $N_2$  is 30 watts 6 hours the output power is still 21 watts. By using CO to rep operating life close to 10,( after using  $N_2$  the operating 10,000 hours. This explains oxide of the nitrogen is sti using silver electrodes.

When manufacturing  $CO_2$ H<sub>2</sub>O is added. The effect of catalytic effect in the disc noticeable effects on output is advantageous to the rever the  $CO_2$  laser's energy level quickens the composite of th the quantity of  $CO_2$  decompos long-life of the sealed  $CO_2$ 

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