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Application of Thermochemical Modeling to Aircraft Interior Polymeric Materials II -Multilayered Seat

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Prepared by Jet Propulsion Laboratory California Institute of Technology Pasadena, California

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Technical Center

Final Report

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 Report No. DOT/FAA/CT-83/16 AD-A/34 UF/ Title and Subritle Application of Thermochemical Modeling to Aircraft Interior Polymeric Materials, II - Multilayered Seat Cushions Author(s) Won Dokko and Kumar Ramohalli Reforming Orgonization Name and Address Jet Propulsion Laboratory California Institute of Technology Pasadena, California 91109 Sponsoring Agency Name and Address U.S. Department of Transportation Federal Aviation Administration Technical Center Atlantic City Airport, New Jersey 08405 Supplementary Notes Supplementary Notes Net on a twelve-mon chemical modeling to multilayered polymeric material cushions. The use of fire-blocking layer(s) between covering fabric has been studied extensively to mini craft seats. The objectives of this work are to exp for the multilayered materials and to experimentally tions. First, the thermochemical model is extended materials, by applying the same analysis technique u The additional constraints of temperature and heat finterface are also applied. A computer program is d behavior of seat cushion systems with and without a a series of tests burning seat cushions with and without a 	3. Recipient's Catalog No. 5. Report Date October 1983 6. Performing Organization Code 8. Performing Organization Report No JPL D-955 10. Work Unit No. (TRAIS) 11. Contract or Grant No. 181-350-100 13. Type of Report and Period Covered June 1981 - May 1982 14. Sponsoring Agency Code 14. Sponsoring Agency Code 14. Sponsoring Agency Code 15. The form air- and the thermochemical model verify theoretical predic- to any number of multilayered sed in the previous work. 10. Vorking layer. Second, hout a fire-blocking layer
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PREFACE

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TABLE OF CONTENTS

INTRODUCTION . . 1. 1 2. THEORETICAL 2 (a) Applicability of the Steady-State Model 2 3 10 (d) The Overall Logic and Characteristics of the Model 10 3. EXPERIMENT . . . 10 (a) Thermogravimetry 13 14 23 4. 23 23 28 5. SUMMARY 33 REFERENCES 35

Appendix A - Derivation of r
Appendix B - Computer Program
Appendix C - Some Comments on Burning Tests Procedure
Appendix D - Photographs of Equipment and Burned Samples
Appendix E - Comparison With Factory Mutual's Numerical Model

inches and and

PAGE

LIST OF ILLUSTRATIONS

Figure	25	Page
1.	Geometry, Considered in the First Case Where $T = T_0$ at X = ∞	4
2.	Geometry Considered in the Second Case Where $T = T_{c}$ at $X = X_{L}$	6
3.	Geometry of the Third Case, a Three-Layered System	8
4.	Thermogravimetric Diagram of Seat Cushion Cover Fabric	15
5.	Thermogravimetric Diagrams of Three Fire Blocking Layers, LS-200, Vonar, and Preox	16
6.	Thermogravimetric Diagrams of Two Foams, Polyurethane Foam and Polyimide Foam	17
7.	Arrhenius Plot of Seat Cushion Cover Fabric, Deduced From TG Shown in Figure 4	18
8.	Arrhenius Plots of Two Fire Blocking Layers, Vonar and Preox, Deduced from TG Shown in Figure 5	19
9.	Arrhenius Plots of Polyurethane Foam and Polyimide Foam, Deduced from TG Shown in Figure 6	20
10.	Weight and Temperature Change Data Measured During the Burn Tests in a Modified NBS Smoke Density Chamber With and Without a Fire Blocking Layer	25-26
11.	Temperature Profiles of Multilayered Seat Cushion System With and Without a Fire Blocking Layer	29-32
12.	Comparison of Mass Losses Predicted by the Model and Experimental Data	34
	LIST OF TABLES	
Table		Page
1.	(a) Fundamental Properties of Seat Cushion Component Materials	21

and the second se

(b)	Kinetics Constants and Heat of Degradation of	
• •	Subdivided Thermochemical Lavers	21-22

NOMENCLATURE

В	Pre-exponential constant for thermal degradation of polymer (sec ⁻¹)
с	Specific heat (cal gm ⁻¹ °C ⁻¹)
D	Heat of degradation (cal gm^{-1})
E	Activation energy for thermal degradation of polymer (cal mole $^{-1}$)
FS	Remaining number of monomer units of degraded polymer
k	Thermal conductivity (ca) $cm^{-1} sec^{-1} c^{-1}$)
N	Remaining number of polymer bonds normalized with respect to the number in the unaffected state
q	Heat flux (cal cm ⁻² sec ⁻¹)
г	Linear regression rate of polymer solid (cm sec ⁻¹)
R	Universal gas constant (cal g-mole ⁻¹ K ⁻¹)
t	Time (sec)
T	Temperature (K)
w	Weight (gm)
x	1-dimensional coordinate, i.e., depth from surface (cm)
ρ	Density (gm cm ⁻³)
λ	The normalized regression rate eigenvalue ($\propto 1/r^2$)

Subscripts

- 1, 2, 3 Pertaining to each of multilayers
- s Front (hot side) surface of a layer
- c Rear (cold side) surface of a layer
- o Undisturbed state
- L Thickness of a layer

Superscripts

+, - A point immediately beyond and before, respectively

EXECUTIVE SUMMARY

The purpose of this effort was the evaluation of the fire performance of seat blocking layers for urethane seat cushions. Because urethane cushions have been demonstrated as highly flammable under aircraft post-crash fire conditions, seat blocking layers have been proposed as a method of protecting the urethane from involvement with the fire. Many hypothesis have been proposed to explain the manner in which fire blocking layers achieve their desired effect, but such hypothesis have not been verified in any quantitative sense.

A condensed phase thermochemical modeling theory previously developed for single component aircraft materials was modified to handle multi-layered materials. The model is based on analytical heat and mass transfer relationships. Required experimental inputs into the model are material thermal properties typical of heat transfer calculations. Additionally, the analysis technique known as thermogravimetric analysis is used to establish the parameters that describe the thermal breakdown of the plastic material when exposed to fire.

Parallel to this analytic work, experimental tests were conducted on sample blocking layer materials in a modified smoke chamber of National Bureau of Standards (NBS) variety. The samples were exposed to specified heating rates and both their weight loss and thermal behavior were measured. This measured behavior is compared with the predictions of the analytic evaluation.

The findings of the investigation are that the thermochemical model can predict the effectiveness of seat blocking layers within a limited range of fire exposure conditions. Additionally, the NBS Smoke Chamber is a useful small scale test for screening candidate seat blocking layers.

1. INTRODUCTION

Aircraft cabin fire safety has been one of the major research and development activities for the Federal Aviation Administration (FAA) for more than 30 years in the past (reference 1). The reasons for such a long-term commitment are the fatalities observed, unique environment in an aircraft (such as small space, high density of people, limited access for egress, etc.), and the complexity in fire phenomenon itself.

In general, fire potential in a passenger aircraft can be due to jet fuel and cabin interior materials. However, a fire involving only the interior materials can occur in-flight or on the ground. In the case of a postcrash fire, a jet fuel fire can ignite interior materials, and the hazards arise from interaction of the fuel fire with interior materials (reference 2).

As planned earlier (reference 4), this work is on the prediction of material burning phenomena through condensed phase thermochemical modeling, and does not involve gas dynamic modeling. These two approaches may well be bridged in the future (reference 3) with sufficient progress.

Among the cabin interior materials such as carpet, seats, window screens, sidewall panels, ceiling and partitions, one of the most flammable materials, seat cushion, is the subject of the current work. Previous work, last year, (reference 4) was on wool carpet and polyurethane foam treated as single layer materials.

In this work, the seat is treated as a system of multilayered polymeric materials, consisting of a seat cover fabric, polyurethane foam cushion, and a fire blocking layer.

In addition to analytically predicting the burning behavior of multilayered systems as a function of heat flux and layers' thicknesses, burning tests are conducted in a modified NBS density chamber to verify the temperature profile and weight losses predicted by the model.

- 1 -

2. THEORETICAL

The thermochemical model used in this work is an extension of the model developed earlier for a single-layered material. As a matter of fact, the approach used in predicting the burning rate of wool carpet with a char layer on its top (reference 4) is a particular form of such an extension. Another earlier work (reference 5) applied to the study of burning rate of composite materials is also referenced for the generalized extension.

(a) The Applicability of the Steady-State Model

There are two aspects of this time-independent model which show is be recognized. One is the coordinate transformation (a la Spaldi (reference 11)) that enables us to treat the regressing surface if it were stationary and the other is the more fundamental argum on the relative time scales.

- (1) The assumption of time-independent degradation enables one to see that d/dt can be written as $r \cdot d/dx$. This transformation, the authors believe, was first introduced in combustion theory by Brian Spalding (reference 11) in his treatment of laminar flame propagation in premixed gases. The point is that the "transient" propagation can be viewed as steady-state in another coordinate frame.
- (2) In any case, the assumption $\partial/\partial t \equiv 0$ implies that the characteristic heat transfer time is small compared to the characteristic "flow" or thermal wave propagation time. The time needed for the establishment of the fully developed temperature profile in the solid is, of course, infinity from the first order nature of the equation's; recall

$$(T_x - T_0)/(T_{xs} - T_0) = 1 - \exp(-\alpha t/x^2)$$

where T_x is the temperature at any depth x (from the surface), T_0 is the initial temperature, T_{xs} is the steady-state temperature at depth x, is the thermal diffusivity $(\equiv k/\rho c)$, t is the time. Thus the time-independence assumption in our treatment relies on a small value for the characteristic time L^2/α for its validity. In the case on hand, the following approximate numbers lend credence to this approach. Typically $\alpha = 10^{-3} \text{ cm}^2/\text{sec}$. The "characteristic" thermal depth, i.e., the distance from the surface to reach 1/e of the full temperature difference is 0.2 cm (for example from figure 11, first plot) at 120 seconds (2 min.); or

$$t - \frac{\alpha}{x^2} \approx 3.0$$

leading to

 $1 - \exp(-\alpha t/x^2) > 0.9$

Thus the steady-state assumption is valid in our case of heating rates.

(b) Mathematical Formulation

As in the previous work, the geometry under consideration is onedimensional, and the steady-state condition is assumed throughout the analysis. The following mathematical analyses for a multilayered system are divided into three parts: the first part is for the case where a single-layered material is so thick that the rear surface (cold side) temperature is undisturbed; the second part is the same as above except that the rear surface temperature has been raised substantially, and the third case is the extension for a layer which is on top of the above two layers and has as many colder layers underneath it as determined by the configuration.

-3-

In all three cases, the governing equation for each layer is the same, that is,

$$k \frac{d^2T}{dx^2} + \rho cr \frac{dT}{dx} = D\rho NB \exp(-E/RT)$$
(1)

The right-hand side of this equation is the heat sink term corresponding to the first-order degradation reaction described by

$$-\frac{1}{N}\frac{dN}{dt} = B \exp(-E/RT)$$
(2)

The symbol N represents the number of remaining bonds per unit mass, normalized with respect to that of fresh, unburned material. The symbol D, the heat of degradation, is "positive" when the degradation reaction is endothermic and "negative" when exothermic.

The boundary conditions, however, are different for each case.

(1) First case, i.e., $T = T_0$ at $x = \infty$

and here





-4-

When the rear surface (cold side) temperature is undisturbed, the boundary conditions for equations (2) and (3) are, respectively,

$$T = T_{S} \quad at \quad x = 0 \quad (3)$$

$$T = T_{O} \quad at \quad x = \infty$$

$$N = 1 - \frac{1}{FS} \quad at \quad x = 0 \quad (4)$$

$$N = 1 \quad at \quad x = \infty$$

The symbol FS stands for fragment size, an average number of monomer units of degraded polymer.

Assuming constant values of material properties, the solution for r, surface regression rate, has been obtained (reference 6) by singular perturbation methods with matching the solution for the inner (surface) and outer (deep solid) regions. For the steady state system, r is given as:

$$r = \sqrt{\frac{\left(\frac{E}{RT_{s}}\right)\left(\frac{T_{s}-T_{0}}{T_{s}}\right)\left[\left\{1 + \frac{D}{c(T_{s}-T_{0})}\right\} \ln\left(\frac{FS}{FS-1}\right) - \frac{D}{c(T_{s}-T_{0})FS}\right]} (5)$$

And the integration of equation (1) will show that the conductive heat flux at the hot surface is

$$q_{s} = \rho r [c (T_{s} - T_{o}) + D/F_{s}]$$
 (6)

if there is zero heat flux at the rear surface. Thus, all the thermal energy flowing into the layer can be said to be totally consumed within the layer.

-5-

(2) Second Case, i.e., $T = T_c$ (> T_o) at x > 0



Figure 2. Geometry considered in the second case where $T = T_C$ at $x = x_L$

This is the case when the front (hot) and rear surface temperatures of a layer are substantially higher than the unperturbed condition of T_0 . The governing equations are the same as above, i.e., equations (1), (2), and the boundary conditions are

$$T = T_s$$
 at $x = 0$ (7)
 $T = T_c$ at $x = x_1$

$$N = 1 - \frac{1}{FS_{s}} \text{ at } x = 0$$
(8)

$$N = 1 - \frac{1}{FS_{c}} \text{ at } x = x_{L}$$

Assuming that $\rm T_{C}$ is close to $\rm T_{S},$ that is

$$\frac{T_s - T_c}{T_s - T_o} < 0.1$$
⁽⁹⁾

the solution for r is given by

$$r = \sqrt{\frac{(k/\rho c) B \exp (-E/RT_s)}{\lambda \left(\frac{E}{RT_s}\right) \left(\frac{T_s - T_o}{T_s}\right)}}$$
(10)

-6-

The details of this derivation is shown in appendix A.

The integration of equation (1) in the region between the front and rear surfaces will give the net heat flux, that is, the incoming heat flux at the front surface minus the one at the rear surface.

$$q_{net} = \rho r \left[c \left(T_s - T_c \right) + D \left\{ \frac{1}{FS_s} - \frac{1}{FS_c} \right\} \right]$$
(11)

In other words, the above equation determines the net heat consumed within the layer volume, as sensible heat and heat of degradation.

(3) Third Case of a multilayered system

In the above two cases, the analyses are for a homogeneous single layer for which thermal, physical and chemical properties are constant and uniform in every part of the layer. If a system consists of more than one material, like the example of a cover fabric - blocking layer ~ foam, those properties are widely different and hence will be treated as a multilayered system.

Strictly speaking, even a single component material should be treated as a multilayered system, if the temperature drop from the front to the rear surface is so large that the thermal, physical and chemical properties can not be considered constant and uniform within the layer.

In other words, when a material of certain thickness is burning, it is a multilayered system from the view point of thermochemical behavior. As shown later, B and E of any single material employed in this study do not remain constant over the temperature range of interest but vary considerably.

-7-

For this reason, each component material of a multilayered seat cushion system is in itself considered multilayered, divided by temperatures at which any of the thermal and/or thermochemical properties changes substantially. Hence, the total number of layers, i.e., thermochemical layers, of a seat cushion system become equal to the sum of the thermochemical layers of each component material.

For the ease of analysis, first suppose that a system of three layers is undergoing pyrolysis reaction, as depicted in figure 3. The same analysis can be extended to any system with more than three layers with the change in subscripts.



Figure 3. Geometry of the third case, a three-layered system

The conductive heat flux at the top surface (x=0) should be the sum of all the energy, i.e., the sensible heat and the heat of degradation, required for the pyrolysis of all three layers in this 1-D approximation. The energy required for each individual layer is shown in equation (6) or equation (11). Then, the incoming heat flux at the top surface of the first, second, and the third layer is, respectively,

$$\left. \begin{array}{ccc} -k_{3} \left. \frac{dT}{dx} \right|_{x=0^{+}} &= & \rho_{3} r_{3} \left[c_{3} \left(T_{s_{3}} - T_{c_{3}} \right) + & D_{3} \left(\frac{1}{FS_{s_{3}}} - \frac{1}{FS_{c_{3}}} \right) \right] \\ &+ & \rho_{2} r_{2} \left[c_{2} \left(T_{s_{2}} - T_{c_{2}} \right) + & D_{2} \left(\frac{1}{FS_{s_{2}}} - \frac{1}{FS_{c_{2}}} \right) \right] (12) \\ &+ & \rho_{1} r_{1} \left[c_{1} \left(T_{s_{1}} - T_{c_{1}} \right) + & D_{1} \left(\frac{1}{FS_{s_{1}}} - \frac{1}{FS_{c_{1}}} \right) \right] \end{array} \right]$$

$$\frac{dT}{dx}\Big|_{x=x_{3}^{+}} = \rho_{2} r_{2} \left[c_{2} (T_{s_{2}} - T_{c_{2}}) + D_{2} \left(\frac{1}{FS_{s_{2}}} - \frac{1}{FS_{c_{2}}}\right)\right] \\ + \rho_{1} r_{1} \left[c_{1} (T_{s_{1}} - T_{c_{1}}) + D_{1} \left(\frac{1}{FS_{s_{1}}} - \frac{1}{FS_{c_{1}}}\right)\right]^{(13)}$$

and

$$-k_{1} \frac{dT}{dx} \bigg|_{x=x_{2}^{+}} = \rho_{1} r_{1} \left[c_{1} (T_{s_{1}} - T_{c_{1}}) + D_{1} \left(\frac{1}{FS_{s_{1}}} - \frac{1}{FS_{c_{1}}} \right) \right] (14)$$

There are however, two kinds of physical constraints applied at each interface. The first kind is that

$$T_{c_3} = T_{s_2}$$

$$T_{c_2} = T_{s_1}$$
(15)

and the second is that

$$\begin{array}{c|c} -k_{3} \frac{dT}{dx} \\ x=x_{3}^{-} \end{array} = \begin{array}{c} -k_{2} \frac{dT}{dx} \\ x=x_{3}^{+} \end{array}$$

$$\begin{array}{c|c} -k_{2} \frac{dT}{dx} \\ x=x_{2}^{-} \end{array} = \begin{array}{c} -k_{1} \frac{dT}{dx} \\ x=x_{2}^{+} \end{array}$$

$$(16)$$

-9-

The above two conditions are to satisfy the continuity of temperature and heat flux at interfaces.

In the expression of equations (12), (13), or (14), each r is implicitly defined. In other words, the heat fluxes at the upper and lower boundaries of each layer are used in obtaining the r of that layer (as shown in appendix A), and are in turn used to define the heat fluxes [as shown in equations (12), (13), and (14)]. This poses no problem in actual calculations since any iterative method leads to fairly rapidly converging answers. The significance of this fact is, however, that the calculation of r's should proceed layer by layer, from the rear (coldest) to the front (hottest), or from the front to the rear.

(c) Computer Program

Based on the theory discussed above, two computer programs are developed and listed in appendix B. The first program, (Computer Program A) takes the interfacial temperatures as the input data, and the second (Computer Program B) takes the temperature and the conductive heat flux at the front surface as its input. The choice among these two programs is dictated firstly by the available boundary conditions, and the other may be used as a complementary to obtain additional information.

(d) The Overall Logic and Characteristics of the Model

The model was originally developed to account for the host of inconsistencies in polymer degradation and burning rate data. The most important distinguishing characteristics are the recognition that the mean molecular weight of the vaporizing molecules need not be equal to the monomer molecular weight. Instead, a vapor pressure equilibrium criterion was used to unambiguously specify the mean molecular weight at the surface. This procedure not only removed an

-10-

annoying arbitrainess at the surface, but also checked well, in terms of results, with data from various classes of polymer combustion. Under the assumption of first order Arrhenius kinetics for the degradation of the subsurface polymer, the model, in its present state, cannot handle cases where extensive degradation beyond the monomer stage takes place <u>before</u> vaporization. This is not as severe a limitation as it may seem. It has been documented (e.g., see Stanley Martin, X Symposium (International) on Combustion) that many of the the degradation reactions actually take place after the major precursors (i.e., products of partial degradation) have left the surface.

The model also assumes that the thermal wave front moves at a uniform speed in the solid. This thermal wave can be associated with the mean surface (and hence the wave speed is indeed the burning rate or "regression rate") only when all of the solid (condensed phase) gets transformed into vapor. This is indeed the case in the burning of simple plastics (e.g., plexiglas, polyethylene, polyurethane . . .). In the case on hand, substantial charring occurs indicating that the familiar "regression rate" needs careful interpretation. For example, it is a familiar fact that in wood burning the reactive portion (cellulose and hemicellulose) leaves the solid framework of lignin char. In such cases of charring solids, the regression rate is more appropriately associated with the velocity of the thermal wave in the wake of which the reactive portion gets vaporized and leaves the char. Thus, it may appear that the surface is not really regressing in the physical sense, although the reactive portion is. Up to this point, the interpretation of the regression rate of this model is clear. The interpretation gets more complex as multilayered materials are concerned. Hence, clearly, the regression rate of the II layer, if interpreted literally, will give rise to a "void" or separation between the bottom of the I layer and the initial position of the IIlayer top surface. However, the complexity is removed when one recognizes that the regression meant here is really the movement of the thermal wave in the solid layers while the char framework (end

- 11 -

With this interpretation, the regression rate r is really always associated with the mass flux ρ r where ρ is not the density of the solid but is only the density of the reactive portion of the solid. However, in the evaluation of the properties such as the thermal diffusivity of the reacting solid, the full density has to be used. Also, the heat transfer through the char is characterized by a thermal conductivity coefficient of the char, and the volatiles convection (flow) through the char. This has been highly simplified in the present analysis which considers the temperature to be constant (and equal to the temperature of the bottom surface of the immediate layer above). Clearly, more work is needed to mathematically incorporate into the model, the realistic char formation, flow and heat transfer.

The "void" that is mentioned in the experimental portion is not the same on this void or separation. Experimentally it is found that the mechanical movement of the volatiles frequently causes a void in the assembly and physical separation of layers. This is a very complex process mathematically.

Considering all of these complexities, the agreement between the theory and experiments in this study is thought to be encouraging.

3. EXPERIMENTS

The experiments of this work consist of two parts; one is to obtain the kinetics constants using thermogravimetry (TG) and the other is, through actual burning tests, to determine the temperature profile established within a seat cushion assembly and the weight loss during pyrolysis. The experimental results are compared with the model predictions.

(a) Thermogravimetry

This experiment heats up a small sample suspended inside a furnace and records the weight change as a function of temperature. The kinetics constants, B and E, are deduced then from an Arrhenius plot, a plot of log $\{(1/w) \cdot -(dw/dt)\}$ versus reciprocal of absolute temperature $(1/_T)$.

It was known from previous work (reference 4) that the ambient gas composition, more precisely the oxygen concentration, affects the kinetics constants. This is a very important point. It has long been recognized that even minor ($\sim 1\%$) concentrations of certain oxidative species can significantly alter (by almost an order-of-magnitude) the degradation rates of polymers. And yet no detailed study seems to be available in this area. One study that addressed this question specifically stopped short of actually demonstrating the effects but had to make valid (but indirect) deductive arguments to point the importance. In any combustion situation, especially with flow of gases over the surface (a very common aircraft fire scenario), the oxygen concentration at the burning surface appears to be around 0.1%-1% (Wooldridge and Muzzy (reference 8), Kulgein (reference 9), Fennimore and Jones (reference 10). This measured non-zero concentration is significant, because earlier JPL work under FAA sponsorship (reference 4) indicated that actual aircraft interior materials exhibited a strong dependence of the kinetics constants of degradation on small concentrations of oxygen in a stream of oxygen in an inert. When it is realized that turbulent transport offers a mechanism for the availability of small concentrations of oxygen at the burning surface, it is easy to see that these oxygen concentration effects could be important in predicting full-scale burning behavior. Full-scale diffusion flames would be large enough to be fully turbulent.

Nevertheless, all the present TG experiments are performed in pure nitrogen except for seat cover fabric, due to the limited scope of the work.

-13-

The TG diagrams of the cover fabric, fire blocking layers (Vonar $^{\textcircled{O}}$, LS-200, and Preox $^{\textcircled{O}}$), and foams (polyurethane and polyimide) are shown in figures 4, 5, and 6, respectively. The Arrhenius plots of figures 7, 8, and 9 show that the pyrolysis of each of these materials cannot be described as a single first-order reaction. Thus, the entire reaction temperature range is subdivided into multiple segments in each of which the reaction rate is reasonably accurately described by the first-order Arrhenius expression. Figure 7 shows how it is done for the cover fabric, as an example, with six subdivided segments. Each segment comprises one thermochemical layer as discussed in the previous section.

Kinetics constants obtained along with other thermal and physical data of the materials used in this study are shown in Table I. These are used as input data for the computer program shown in Appendix B.

(b) Burning Tests

In order to measure the temperature profile and weight loss of samples, a modified NBS Smoke Density Chamber is used. The radiative heat flux was provided by a high heat flux furnace (Mellen furnace Model No. 10, with the maximum heat flux of 12 W/cm^2 or 10.6 BTU/ft² sec). The originally equipped furnace for the NBS chamber could provide 2.5 W/cm². For the temperature distribution measurement, nine thermocouples (Pt vs. Pt-10% Rh) were connected to a multichannel recorder (Leeds and Northrup's Speedomax Model 251, 12 channel, with resolution time of 1 second for each reading). The weight measuring device was a transducer-type cantilever beam, connected to an amplifier and then to a strip-chart recorder.

The furnace is heated up to, and maintained at, 850° C during the tests. After the furnace has reached a steady state, a radiation shield is removed and the seat cushion is exposed directly to the furnace. The radiative heat flux is around ~1.6 Btu/ft² sec.

-14-

<u>5</u> TEMPERATURE (°C) 00 NATERIAL =Cover Fabric ATM = Air HEATING RATE = 20°C/min <u>10</u> R S (%) MEIGHT LOSS



-15-





-16-





-17-



FIGURE 7. ARRHENIUS PLOT OF SEAT CUSHION COVER FABRIC, DEDUCED FROM TG SHOWN IN FIGURE 4.



FIGURE 8. ARRHENIUS PLOTS OF TWO FIRE BLOCKING LAYERS, VONAR AND PREOX, DEDUCED FROM TG SHOWN IN FIGURE 5.



FIGURE 9. ARRHENIUS PLOTS OF POLYURETHANE FOAM AND POLYIMIDE FOAM, DEDUCED FROM TG SHOWN IN FIGURE 6.

-20-

		с	k*	хL
	Density cm ³	Specific Heat 	Thermal Conductivity <u>cal</u> cm. sec.°C	Thickness cm
Seat Cover Fabric	0.4	0.3	0.0001	0.1
Fire Blocking Layer LS-200 Vonar (B) Preox (B)	0.12 0.146 0.62	0.3 0.345 0.3	0.0002 0.0002 0.00034	1.2 0.8 0.1
<u>Foam</u> Polyurethane Polyimide	0.03 0.023	0.4 0.2	0.00010~0.00034 0.0001	10.6

Table I(a). Fundamental Properties of Seat Cushion Component Material

(b) Kinetics Constants and Heat of Degradation of Subdivided Thermochemical Layers

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	T Tempera- ture Range (K)	B Pre Exponential Factor (1/sec)	E Activation Energy (cal/mole)	D Heat of Degrada- tion (cal/g)
Seat Cover Fabric	848-898 813-848 763-813 689-763 602-689 523-602	2.13 x 10 ⁴ 5.06 x 10-8 2.24 x 1013 2.51 x 10-1 7.30 x 10-6 1.57 x 10 ²	24800 -20300 56500 7800 -6500 13700	-2.9 -2.9 -2.9 -2.3 -16. -16.3

* Source: NASA Ames Research Center -- Handbook of Chemistry ond Physics

- 21 -

(b) Kinetics Constants and Heat of Degradation of Subdivided Thermochemical Layers (Cont'd)

	T	B	Ε	D
	Tempera- ture Range	Pre Exponential Factor	Activation Energy	Heat of Degrada- tion
	(K)	(1/sec)	(cal/mole)	(cal/g)
LS-200	877-923 767-877 723-767 698-723 673-698 648-673 623-648 598-623 573-598 548-573 473-548	1. 17×10^{-5} 7. 17×10^{-11} 5. 17×10^{1} 1. 04×10^{-4} 8. 28×10^{-14} 6. 56×10^{-3} 2. 46×10^{6} 1. 61×10^{-17} 2. 11×10^{7} 2. 01×10^{-1} 7. 83×10^{3}	-3857 -24770 16818 -2020 -31083 2475 27897 -38190 27800 6772 18279	0.5 9.6 -8.4 -19.1 -16.7 -8.4 -20.3 7.9 71.9 24.9 54.9
_{Vonar} ®	881-923 766-881 709-766 661-709 623-661 591-623 473-591	3. 37×10^{-4} 1. 66×10^{-7} 8. 94×10^{-2} 2. 37×10^{-12} 2. 75×10^{4} 3. 94×10^{-13} 2. 33×10^{4}	1043 -12285 7800 -26532 22078 -25933 19435	1.0 9.6 -23.9 -28.7 -28.7 50.9 107.5
Preox®	873-908 848-873 748-848 683-748 623-683 573-623 498-573	5. 42 x 10^{-6} 8. 55 x 1020 1. 55 x 10^2 9. 85 x 10^{-3} 1. 43 x 10^{-5} 5. 87 x 10^{-3} 5. 94 x 10^{0}	-13061 91578 18860 4500 -4365 3085 10963	0.5 0.5 -13.9 -13.9 -13.9 0.5
Polyurethane Foam	644-688 598-644 573-598 544-573 448-544	8.44 x 10 ¹⁹ 6.17 x 10 ¹ 1.77 x 10 ⁻² 2.60 x 10 ⁵ 6.00 x 10 ⁰	65760 12335 2646 21435 9900	3.1 11.9 2.5 2.5 0.5
Polyimide Foam	1006-1073 973-1006 915-973 854-915 796-854 711-796 673-711	3.06×10^{-5} 3.12×10^{-14} 1.80×10^{-7} 1.33×10^{1} 6.11×10^{-6} 1.07×107 7.63×10^{-3}	-4763 -46170 -16070 16860 -7890 36705 6938	0.5 0.5 0.5 -7.3 -7.3 -7.0 -0.3

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(c) A Note on the Heating Rates

The actual heat fluxes encountered in aircraft crash situations can approach 10 W/cm² -- a fact that led investigators to recognize that the NBS smoke-density chamber, with its original 2.5 W/m² furnace, may have to be modified for realistic simulations. In the present effort, the heat fluxes meant for the FAA samples were planned to be gradually increased from a low value (1 W/cm²) to the full value of $6-10 \text{ W/cm}^2$. It was the plan to verify the model at low heat fluxes first in a logical sequence of increasing heat fluxes. Thus, at the time of the present reporting the lower heat fluxes have been investigated. The capability exists for the higher heat fluxes. However, the actual tests await future research support.

Since the initial setup of equipment and trial runs, several modifications and improvements have been made on the sample preparation and run procedure, in order to obtain meaningful and reproducible test results. These are listed in appendix C for future reference.

A pair of thermocouples are placed at every interface, including the front surface. And additional five are placed within the polyurethane foam, 0.5 inches apart from each other along the axis. A test is always run in duplicate; one to measure the weight loss and the other for the temperature profile. This is because the thermocouples have such stiffness that their use would not allow free movement of the cantilever beam and the accurate measurement of weight change.

4. RESULTS AND DISCUSSION

(a) Burning Test Results

The pictures of the burned samples and the equipment used in this work are shown in appendix D for visual examination. The results of the burning tests in the modified NBS Smoke Density Chamber are

-23-

shown in figure 10. These are the temperature changes obtained from the multiple thermocouple readings placed at different positions within the seat cushion assemblies and the weight change with time, for the runs with and without the fire blocking layers.

The weight change data show that the sample without a fire blocking layer suffered the largest mass loss, and that the effectiveness of a fire blocking layer, if judged from the mass loss data, can be rated in increasing effectiveness on $Preox (\mathbb{R})$, $Vonar (\mathbb{R})$, and LS-200. The same order is observed when the burnt samples were examined visually.

The thermocouple reading at the sample surface facing the furnace, however, should be corrected for the caused by radiation and convection. The energy balance for the thermocouple bead can be shown as

 h_{rad} (T_{furnace} - T_{bead}) = h_{conv} (T_{bead} - T_{gas film})

where h_{rad} and h_{conv} are the radiative and convective heat transfer coefficient, respectively, and $T_{furnace}$, T_{bead} , T_{gas} film are the temperatures of the furnace, thermocouple bead, and film stagnant gas film surrounding the bead, respectively. The estimated error $(T_{actual} - T_{bead})$ is in the range of 25°C to 35°C under the current experimental conditions. Further detailed error analysis is not attempted here due to the lack of precise information on emittance, sample surface condition, and thermal contact between the bead and fabric fibers, which are necessary to estimate the heat transfer coefficients with sufficient precision.

Other than this, the thermocouple data lead to a few important observations.

 The first one is that the front surface temperature measured by the thermocouple No. 1 reaches a steady state in about three

-24-



(1 of 2 pages)

-25-


-26-

minutes, and then registers essentially the same temperature thereafter. When a fire blocking layer of Vonar \mathbb{R} or LS-200 is used, however, a very slight increase of the front surface temperature is observed. The temperature just behind the cover fabric, measured by the thermocouple No. 2, trails that of the front surface by about 15°C.

- (2) There is a further temperature drop across a fire blocking layer. This temperature drop is indicated by the vertical distance between the thermocouples No. 2 and No. 3 readings, as shown in figure 10. In the figure, LS-200 allows the largest temperature drop, Vonar \mathbb{R} medium, and Preox \mathbb{R} the least. Incidentally this is exactly the opposite order as the one observed in total mass loss.
- (3) The temperature-rise histories within the polyurethane foam, measured by thermocouples No. 4, 5, 6, and 7, hardly represent steady-states. Most of the measured temperatures simply keep rising steadily, as shown in figure 10.

In some cases, however, a plateau seems to appear after such monotonic increase in temperature, implying a steady-state condition attained within the foam. This behavior is believed to be caused by the formation and existence of a void or dome filled with relatively hot pyrolysis gas products in a convective flow motion. This explains why such a plateau shows up and why then these plateaus are close to each other - in other words, the temperature gradient is small. It is particularly serious when no fire blocking layer is used, because under the condition of high heat flux, the polyurethane foam liquefies and drips before complete burnout occurs. This eventually leaves an ever-growing void filled with the hot pyrolysis gas product. The hot gas of the void may flow freely and contribute to the rapid temperature rise at the rear surface.

-27-

(4) The use of a fire blocking layer helps maintain lower temperatures at the rear (colder) side essentially by reducing the heat transfer. This can be seen easily by comparing the temperatures measured by the thermocouples with the same number. For example, after 6 minutes of exposure, the thermocouple No. 5 reaches 321 K with LS-200, 323 K with Vonar (R), and 463 K with Preox (R), while it reaches 546 K without a fire blocking layer.

There is, however, a compensating effect near the front (hot) surface. That is, the front attains higher temperatures with the fire blocking layer than would be observed without the fire blocking layer. Figure 10 shows the front surface temperature is 701 K with LS-200, 695 K with Vonar \mathbb{R} , 691 K with Preox \mathbb{R} , and 643 K without one, respectively, after 10 minutes exposure. This leads to an observation that a fire blocking layer functions as if it is a reflector for heat flux.

This results in lowering the rear side temperature while maintaining the higher front side temperature.

(b) Predictions by Thermochemical Model

The temperature profiles obtained by the thermochemical model, in comparison with the experimental data, are shown in figure 11. The profile for each configuration (with and without a fire blocking layer) is calculated every 2 minutes using the Computer Program B of appendix B. (Note: The conductive heat flux at the front surface, needed as the input for the Computer Program B, was the one predicted by Computer Program A of appendix B.) In general, the calculated results are in good agreement with the experimental data. The difference is often less than 10°C, with a few exceptions of 25°C as the maximum difference. However, the predicted temperatures within the polyurethane foam show significant deviations from the measurements. The negative deviations, i.e., temperatures measured lower than predicted, are seen when the exposed time is small and/or



-29-

1



-30-



-31-



-32-

where the depth is large. The positive deviations, i.e., temperatures measured higher than predicted, are seen wherever the void is observed inside of the polyurethane foam. This is again believed to be caused by a convective flow of hot combustion gas within the void, which leads to increased heat transfer to the colder side.

The weight losses calculated by the model are compared with the experimental data, as shown in figure 12. The figure shows that the calculated weight loss is about 15 to 95 percent higher than the measured values in all cases, but the order of the mass loss is in agreement with the experimental data except for the control (without a fire blocking layer). This again is due to the presence of the void formed inside of the foam, which is clearly the physical state this thermochemical model is not aimed to be used for.

5. SUMMARY

The work on the prediction of thermochemical performance of multi-layered seat cushion materials leads to four conclusions:

- The concept of thermochemical sublayers is applicable to describe the complex pyrolysis behavior observed for many of the currently used and proposed aircraft interior polymeric materials.
- 2. The thermochemical model of the past can be extended to handle such rultilayered systems analytically.
- 3. The model predicts reasonably within a factor of 2 hoth the weight loss due to burning and the temperature profile established within the seat materials.
- 4. The thermochemical model, can be used with the minimum number of input data determined by experiments for the thermochemical performance prediction of other multilayered materials under fire conditions. This analysis coupled with experiments in the NBS Smoke Chamber offers a useful small-scale test procedure for evaluating candidate blocking layer materials.



FIGURE 12. COMPARISON OF MASS LOSSES PREDICTED BY THE MODEL AND THE EXPERIMENTAL DATA (AFTER 10 MINUTES OF BURNING IN THE NBS SMOKE CHAMBER): O - PREDICTED

• - EXPERIMENTAL

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APPENDIX A Derivation of r

The differential equations are

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$$\frac{d^2T}{dx^2} + \rho_{cr} \frac{dT}{dx} = D\rho_{NB} \exp(-E/RT)$$
(A-1)

$$-\frac{1}{N}\frac{dN}{dt} = B \exp(-E/RT)$$
 (A-2)

and the boundary conditions are

at
$$x = 0$$
 $T = T_s$ (A-3)
 $N = N_s = 1 - \frac{1}{FS_s}$

$$at x = x_{L} \qquad T = T_{C} \qquad (A-4)$$
$$N = N_{C} = 1 - \frac{1}{FS_{C}}$$

By introducing the dimensionless terms,

$$y = \frac{\rho_{\rm Cr} x}{k} \tag{A-5}$$

and

 $\tau = \frac{T - T_0}{T_s - T_0}$ (A-6)

Then Eqs. (A-1) and (A-2) become

$$\frac{d^2\tau}{dy^2} + \frac{d\tau}{dy} = \frac{kD\rho_{NB} \exp(-E/RT)}{(\rho_{cr})^2 (T_{s}-T_{o})}$$
(A-7)

$$\frac{dN}{d\tau} \cdot \frac{d\tau}{dy} = \frac{k c \rho NB exp (-E/RT)}{(\rho cr)^2}$$
(A-8)

A1

Further use of dimensionless variables,

$$p = \frac{d\tau}{dy}$$
(A-9)

$$h = \frac{D}{c(T_s - T_0)}$$
(A-10)

and
$$\Lambda = \frac{kB}{\rho_{cr}^2}$$
 (A-11)

convert the Eqs. (A-7) and (A-8) to

$$pp' + p = \Lambda h N exp (-E/RT)$$
(A-12)
$$pN' = \Lambda N exp (-E/RT)$$
(A-13)

$$pN' = \Lambda N \exp(-E/RT)$$
(A-13)

where a prime denotes $d/d\tau$.

The boundary conditions are accordingly

$$\tau = 1, \quad p = p_{S}, \quad N = N_{S} \tag{A-14}$$

$$\tau = \tau_{\rm C}, \ p = p_{\rm C}, \ N = N_{\rm C} \tag{A-15}$$

The Eq. (A-13) can be combined to Eq. (A-12) to yield

$$pp' + p - pN'h = 0$$
 (A-16)

Assuming $p \neq 0$, then Eq. (A-16) becomes essentially

$$\frac{dp}{d\tau} + 1 - \frac{dN}{d\tau} = 0$$
 (A-17)

If Eq. (A-17) is integrated using the boundary conditions of Eqs. (A-14) and (A-15), then

$$N = (p + \tau - p_{s} - 1 + h N_{s})/h$$
 (A-18)

or N =
$$(p + \tau - p_c - \tau_c + h N_c)/h$$
 (A-19)

Then substituting Eq. (A-19) into Eq. (A-12) gives

$$pp' + p = \Lambda(p + \tau - p_{S} - 1 + h N_{S}) exp(-E/RT)$$
 (A-20)

This is the differential equation to be solved with the boundary equations (A-14) and (A-15).

If the T is close to T_S , then as an approximation, Eq. (A-20) can be written as

$$pp' + p = (p + \tau - p_{S} - 1 + h N_{S}) \Lambda \exp \{-\Theta [1 + \chi (1-\tau)]\}$$
(A-21)

where
$$\Theta = \frac{E}{RT_s}$$
, (A-22)

and
$$\chi = \frac{T_s - T_o}{T_s}$$
 (A-23)

Let
$$\rightarrow = \Lambda \exp(-\Theta)$$
 (A-24)

and
$$g = -p_s - 1 + hN_s$$
 (A-25)

then, the Eq. (A-21) is

$$pp' + p = (p + \tau + g) \lambda exp [-\Theta \chi (1-\tau)]$$
 (A-26)

Further defining

$$\eta = \frac{1-\tau}{\varepsilon}$$
 (A-27)

$$\xi = \mathbf{p} + \boldsymbol{\tau} \tag{A-28}$$

$$\lambda = \varepsilon \mathbf{A} \tag{A-29}$$

whe

re
$$\varepsilon = \frac{1}{\Theta \chi}$$
 (A-30)

enables rewriting Eq. (A-26) as
-
$$(\xi - 1 + \epsilon \eta) \xi' = (\xi + g) \lambda \exp(-\eta)$$
 (A-31)

or

$$\frac{d\xi}{\xi + g} = \lambda e^{-\eta} d\eta \qquad (A-32)$$

The boundary conditions for this equation are

$$\eta = 0 \quad \xi = \xi_{S} = p_{S} + 1$$
 (A-33)

$$\eta = \eta_{\rm C} \quad \xi = \quad \xi_{\rm C} = p_{\rm C} + \tau_{\rm C} \tag{A-34}$$

Integrating now with respect to ξ and $\eta,$ in the region between two surfaces, s and c, is

$$\int_{c}^{s} \frac{d\xi}{\xi + g} - \int_{c}^{s} \left(\frac{\xi}{\xi + g}\right) d\xi = \int_{c}^{s} \lambda e^{-\eta} d\eta \qquad (A-35)$$

to give the result as

$$\ln\left(\frac{\xi_{s}+g}{\xi_{c}+g}\right) - \left\{ \left(\xi_{s}-\xi_{c}\right) - g \ln\left(\frac{\xi_{s}+g}{\xi_{c}+g}\right) \right\} = -\lambda \left(1-e^{-\eta_{c}}\right) \quad (A-36)$$

Then λ is simply given by

$$\lambda = \frac{1}{(1 - e^{-\eta c})} \cdot \left\{ (\xi_{s} - \xi_{c}) + (1 + g) - \ln \left(\frac{\xi_{c} + g}{\xi_{s} + g} \right) \right\}$$
(A-37)

By combining Eq. (A-29), (A-30), (A-22), (A-23) and (A-11), r can be shown in more familiar terms,

$$r = \sqrt{\frac{(k/\rho_c) B \exp(-E/RT_s)}{\lambda \left(\frac{E}{RT_s}\right) \left(\frac{T_s - T_o}{T_s}\right)}}$$
(A-38)

APPENDIX B Computer Program

1•	С	PROGRAM A
2+	C	THIS PROGRAM TAKES THE INTERFACIAL TEMPERATURES (B.C.+S) AS THE INPUT
3•	С	DATA AND THEN CALCULATES THE BURNING RATE. HEAT FLUX. TEMPERATURE PROFI
4 *	С	AND WEIGHT LOSS OF MULTI-LAVERED POLYMERIC MATERIALS (INCLUDING
5 +	С	THE FIRE PLOCKING LAYER) USED FOR AIRCRAFT SEAT CUSHIONS.
ŧ٠	C	THE DIMENSIONS USED IN THIS PROBAR ARE GRAME GRAMEMOLEE D-CELSIUSE
7+	С	CALORIF, CH, AND SECOND.
£ #	, c	
9 +		C1*EXS10* ARA*FT(7+2),**(7),*TFF0(7)+KT(7+11)+CPT(7+11)+
16+		- RHOT(7+11)+HT(7+11)+ET(7+11)+DT(7+11)+HEFTET(7+11)+
11+		- THL9(7,11),TKH1(7,11),TL(411),TAI(11),
12+		- FKL0(7,11)+FKH1(7+11)+FL0(11)+FH1(11)+
13+		- KKL0(7,11),KH1(7,11),KLC(11),MT1(11),
14+		- JSTAFT(11), JEND(11), PEADIM (14), ID(11),
15•		- NS(11),NC(11),AVAME(11),2),T(11),K(11),CP(11),KH((11),H(11))
16*		(11) + (11) + HEATKA(11) + FSS(11) + FSC(11) + FP(11) + GIN(11) + OPUT(11) +
17+		- ONET(11), TP(11,11), NSSP(11,11), TS(ART(11), TG(TV(11)))
18+		
10.		[] FFNSION 2(11-11)
50+		DIPERSION PETTY PSEITY PCETTY PICETTY FILETITY CONTRACTION
51+		DIMENSION ENCLIDED AND DATING CHICING AL PARTIN
55.		UIMENSION TICELISON ONLESSO BEELISOBELISOBELISOB
230	•	PLAL RIGRIMAGNSGN(GHKLUGNRHIGNLUGHHHIGLAMBUA
24 .		DEAC TH THE THEREDIEST HITCOLD DESCET IN DO BET OUT
20*		PEAL IN THE INGREDIEST FATEPIAL PROPERTY AND PRINT OUT
20* 774	L	45 AT 15. 0011 ITOTAL
21-		PERD 134-017 1-0786
<0. 784		
274		10 100 11-14110745 BEAT (5.902) 1.48.485781.43.48.4457(1.23.484(1.4.2).484(1.4.1).
31.		
32.		NTE-NTE-GIJ
31.		RFAD (5-503)(TKL0(1-3)-TKH1(1-3)-KT(1-3)-CPT(1-3)-RH0T(1-3)-
344		
35.		903 FORMAT (2F5.1. 7(10.3)
36+		100 CONTINUE
37.		WITE (6, 948)
38+		948 FORMAT (141. 257. BURNING HATE FREDICTION FOR MULTI-LAY FEED

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81

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- * AIRCRAFT SEAT CUSHION* / 26X, E3(*=*), /
                   5%, *MATERIAL PROPERTY*/ 21%, *TEMP, RANGE,
AT*, 6%,*DENSITY FREEXP, ACTIVATION HE
                                                                     THERMAL . SK.
                                                                HEAT OF
            - *HEAT*+ GX+*DENSITY
                                                                            HEAT .
             <del>-</del> 5λ
                      / 36%, "CONDUCTIV. CAPACITY",13%, "FACTOR
                                                                        ENERGY .
            - 7% . *DECOMPOS. RATE * . /
. •
            -26% + (K) + 5% + (CAL/CM-S.C) (CAL/GM.C) (GM/CM3) + 13%
 .
             -"(CAL/MOLE) (CAL/GM) (DEG C/SEC)" /)
18
             CO 102 I=1.ITOTAL
. .
'#
             NTR=NTREG(1)
              WRITE (6.951)
                             I.ANAMETEL.I.ANAMETEL.2)
1.
1.
         951 FOREAT (51,12,2),245)
)=
             W71TE (6,953) (TKLO(1.J), TKH1(1.J),
            .
۰.
                    J=1+\TR)
i+
         553 FORMAT 117x+ F7+1+ *-*+ F7+1+ 1*+
                   F10.5.37. F6.3. 3". F6.3. 27. E11.3. 4X. F9.0. F10.1. F10.2)
.
         102 CONTINUE
,#
, *
       C
          FIF CONSTANTS USED IN THIS PROGRAM
۲.
       C
...
       C
}+
             F=1.0
             P=1.98717
3.
1.*
             PE= 42.06
2+
             CF645=0.382
۹ø
             EPELON=0.9
             SIGMA=1.355E-12
4.0
5+
              10=225+0
5.
       C
          CALCULATE N AND FS. AT TWO END TEMPERATURES OF KINETIC REGIMES
1 .
       C
          THE CALCULATION SEQUENCE IS, BY LEGIC. FROM LOW TO HIGH TEMPERATURE
3.
       С
3.
       С
)•
              WRITE (6+955)
         955 FCRWAT (1H1, 10%, *NORMALIZED FF4GMENT SIZE AT END TEMPS UP*, 1%,
- *KINETIC RECIMES*, /, 11%, 564*=*), /)
1+
5.
             DO 200 I=1,110T#L
3•
                    WRITE (6.951) I. ANAMETEI. 11. ANAMETEI.2)
3.
                    NTR=NTREG(1)
٠.
5.
                    NKLO(I+NTR)=1.0
                    FKLO(I,NTP)=10600.
7.
             DU 205 JJ=1+NTR
3•
                 J=1.TR+1-JJ
3.
                    CALL FPAGMT (HT 41+J)+ET41+J)+TKL041+J)+TKH141+J)+
5.
                          HEATPT(I+J)+NKLO(I+J)+NKHI(I+J))
1.
                    IF (J.VE.1) NKL0(1,J-1)=NKH1(1,J)
2+
                    IF (NKLU(1.J).E0.1.0) NKL0(1.J)=0.9999599
3+
                    FKL0(1,J) = 1.0/(1.0-NKL0(1.J))
4.
                    IF (NKHI(1,J).EQ.1.0) MKHI(1,J)=0.9959599
5+
                    FKHI(I,J) = 1.0/(1.0-NKHI(I,J))
6+
                    WHITE (6.952) THLOCI, J), TKHICI, J), NKLOCI, J), NKHICI, J)
7+
         958 FORMAT (22%, F7.1, *-*, F7.1, 5%, F11.6, 2%, *--*, 1%, F11.6)
8+
         205 CONTINUE
9+
C+
         200 CONTINUE
1.
       С
       C
          INITIALLIZE A FEW VAFIABLES FOR COMPUTER PLOTTING
2*
3•
       C
              NPAGE = 9
4 .
              NPLOT=1
5+
```

```
5.8
              1PLOT=C
 7.
              CALL SGNPLT
 A .
        C
 4+
          READ IN TEMPERATURE DATA. I.E..
        С
          TEMP. OF TOP SURFACE OF EACH LAYER AND REAR SURFACE (COLD)
 Û+
        C
 1 .
        C
 2+
         1111 CONTINUE
              WRITE (6+898)
 3.
          898 FORMAT (1H1, 25%, +PURNING RATE PREDICTION FOR MULTI-LAYERED+
 4.
 5+
              - * #IRCRAFT SEAT CUSHIGH* / 26%, 63(*=*), /)
              PEAD (5,904,END=1112) TBACK, (HEADIN(1),1=1.14)
 6.
 7.
          904 FORFAT (F10.0.1445)
              WAITE (6,899) (HEADIN(I),1=1,14)
 8.
          199 FORMAT (40% 1445, ///
 9.
                                         7X+ *SURFACE
ACTIVATION HEAT O
 • ت
                   57. MATERIAL +
                                                                 THERMAL
                                                                             HEAT
 1+
              -
                 DENSITY PREEXP.
                                                      HEAT OF
                                                                   HEAT
                                                                                ...
                               CUNDUCTIV. CAPACITY
             -21X. *TEMP.
                                                                   FACTOR
                                                                              ENERGY
 2.
                       DECOMPOS. RATE ./
 3+
 4.
              -21% + (K) + 5x++(CAL/CM+S+C) (CAL/GM+C) (G4/CH3)+ 13%+
                                       (DEG C/SEC)+ //)
              -+(CAL/MOLE) (CAL/CM)
 5+
 6.
              1=1
          105 CONTINUE
 7+
              READ (5.906,ENO=11C) IC(I), THIETS
 4.
 ۰.
          906 FOFMAT (12+F10+2)
 Ú.
              1=1+1
 1.
              60 TO 105
          110 CUNTINUE
 2+
 3.
              NLAYG=1-1
 4.
        С
·5+
              NLAYG1=NLAYG-1
              00 115 I=1+NLAY61
16+
 7+
          115 TLO(I)=TH1(I+1)
              TLO INLAYG)=TBACK
16+
-5.
        r
.0.
           NOW CALCULATE THE FRAGMENT SIZES AT THE TWO BOUNDARIES OF LACH
        С
           GENERIC LAYER
11+
        C
12.
        C
13+
              DO 300 1=1+NLAYG
              IDENT=ID(1)
54.0
15+
              NTID=NTREG41DENT)
56+
              IF (TLO(I).GE.TKLO(IDENT.NTIG)) GO TO 3C1
               JSTART(1)=NTID
57+
18.0
              ALO(1)=1.0
59.
              FLO(1)=10000.
              60 TO 304
30.
11+
          301 CONTINUE
12+
                     DO 302 J=1.NTID
                     IF (TLO(I).GE.TKLO(IDENT.J).AND.TLO(I).LT.TKHI(IDEHT.J))
43+
...
                                 GO TO 305
45+
          302
                     CONT INUF
                     WRITE (6+961) J
46+
47+
                     FORMAT (54. FOR . 12. TH LAYER. TLO IS OUT OF BOUND')
          961
48+
                     GO TO 304
49.
          305
                     JSTART(I)=J
50+
                     CALL FPAGPT(PT(IDENT,J)+ET4IDENT,J)+TKLU(IDENT,J),TLO(I)+
                           HEATRICIDENT.J. NKLOCIDENT.J. NLOCIDI
51+
                     FLO(1) = 1.0/(1.0-NLO(1))
52.
```

83

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-3+
          304 CONTINUE
4+
               IF (THI(I).GE.THLO(IDENT.NTID)) GO TO 306
-5+
               JEND(1)=NTID
.6*
               NH1(1)=1.0
17+
               FH1(1)=10000.
+8+
               60 TO 300
:9+
          306 CONTINUE
5 G *
                     CO 307 J=1+*113
11+
                     IF CTHICLD.GE.TKLGCLDENT.J.AND.THICID.LT.TKHICIDENT.J.)
12+
                                  60 TO 308
13+
          307
                     CONTINUE
,4+
                     WRITE (6,962) 1
                     FORMAT (5%, *FUR +, 12, *TH LAYER, THI IS OUT OF BOUGD*)
.5+
          962
16+
          368
                     JEND(1)=J
.7 .
                     CALL FRAGET(ST(IDENT,J),ET(IDENT,J),TELD(IDENT,J),THI()),
....
                             FEATHT (IPENT+J)+NKLO(IDENT+J)+NH1(1))
.9+
                     FHI(I)=1.0/(1.0-NHI(1))
10+
          300 CONTINUE
11.
        С
2+
        ٢
           REPEFINE AND PENUMBER THE SUPDIVIDED MULTILAYERS, BASED ON THE
13+
        C
           TEMPERATURE RANGES OF THERMOCHEMICAL REGIMES AND ALSO ON THE GENERIC
14=
           DIVISIONS
        C
           SEQUENCE OF IJ IS FROM FRONT INSTTESTS LAVER TO PEAR (COLDEST) LAVER
15+
        C
16+
        C
17+
               1J=C
18-
               DO 310 I=1.VLAYG
19+
               IDENT=IC(I)
10+
               JENDI=JEND(1)
31+
               JSTAI=JSTART(])
12+
               1J=1J+1
134
               T(IJ)=THI(I)
               ANAME(1J+1)=ANAMET(1DERT+1)
540
               ANAME(IJ+2)=ANAMET(IDENT+2)
55.
<sup>1</sup>6+
               C(IJ)=DT(IDENT,JENDI)
P7.
               CP(1J)=CPT(ICENT.JENDI)
E8+
               RHU(1J)=RHOT(IDENT.JEND1)
65+
               E(JJ)=ET(JCENT,JENDI)
               EETJ)=BTIIDENT.JENDI)
80+
91+
               K(IJ)=XT(IDEFT,JENDI)
92+
               HEATRACIUS=HEATRTLIDENT.JENDIS
53+
               KS(IJ)=NHI(I)
....
               FSS(IJ)=FH1(1)
95+
               NC(IJ)=NKL0(IDENT,JENDI)
56+
               FSC (IJ)=FKLO(IDENT+JENDI)
               IF (JENDI.EQ.JSTAI) NC(1J)=NL0(1)
IF (JENDI.EQ.JSTAI) FSC(1J)=FL0(1)
57+
94.
               IF (JENDI.EQ.JSTAI) GO TO 310
99.
00+
               JENDI1=JENDI+I
01+
                     DO 311 J=JENDII+JSTAI
62+
                      1J=1J+1
                      T(IJ)=TKHI(IDENT.J)
03+
                      ANAMECTJ-1)=ANAMETCIDENT-1)
0.4+
05.
                      ANAME (1J+2)=ANAMET(IDENT+2)
06+
                      D(IJ)=CT(1DENT.J)
                      CP(1J)=CPT(IDENT,J)
07.
                     RHO(1J)=RHOT(IDENT+J)
0.6+
                     E(IJ)=ET(IDENT,J)
09+
```

10+ BIJJ)=PT(1DENT,J) 11+ KAIJ)=KTOJCENT.J) HEATRACIJ)=HEATRICIDENT.J) 12+ 13+ NS(JJ)=NKHI(IDENT.J) FSS(IJ)=FKH1(IDENT.J) 14+ 15+ NC(1J)=4KL0(1DENT,J) 16+ FSC(1J)=FKLU(1DENT+J) IF (J.EC.JSTAI) ACCIDIENLOCT) 17+ IF (J.FG.JSTAI) FSC(1J)=FLO(1) 18# 19+ 311 CONTINUE 310 CONTINUE 26+ 21+ NL FYER=IJ 22+ T(1J+1)=TE4CK 23. LPITE (6.500) (1.4NAME(1.1).ANAME(1.2). T(1). K(1). CP(1). RHO(1). - BOID, ECID, DOID, HEATRACID, 1=1-NLAYER) 24= 25+ 500 FORMAT (5% 1202% 245. F6.1. 14. F10.5. 44. F6.3. 5% F6.3. 26* E14.3. 3V. F9.0. 4%.F7.0.F9.3) WEITE (6.907) TEACK 27. 26* SCT FOPMAT (St. *KEAP SURF*, F6.0, //) 29+ WAITE (P. 410) 910 FORMAT (5/, *PREDICTED BURNING RATE* ///,20%, *BURN KATE* 30* - * SURFACE TEMP. HEAT* 31+ - . FLUX IN HEAT FLUX OUT HEAT CONSUMED FS-TOP SURFACE. 32+ - + FS-FGTTGM+ / 21x, +(CF/SEC)*, 7x, +(K)*, 7x, - 3(+(CAL/CM2SEC)*, 3%), //) 33+ 34+ 35+ NLAY1 = NLAYER - I 36+ ALC = D 37 • C CALCULATE THE PURNING RATES FOR THE FIRST TIME ASSUMING THAT THE BURN 30. C PATE ARE THE SAME FOR EVERY LAYFP 10. С 46+ С DO 333 J=1+ NLAYER 41+ 42+ TIO(1) = T(1) - TO43+ H(1) = O(1) / (CF(1) + (T(1)-TO))ALPHA(I) = K(I) / (RHO(I)+CP(I)) 44+ 4 ... CHI(I) = (T(I)-T() /T(I) 46+ THETACIN = E(1)/(R+T(1))XIS(I) = -H(I) /FSS(I)47. 48+ xIC(I) = -H(I) /FSC(I)49+ TAUC = (T(I+1) - TC) /(T(T) - TC) 50+ IF (T(1+1).LF.300.) GO TO 337 51+ ETAC(I) = THETA(I) + CHI(I) + (1.0 - TAUC) 52+ EVC(1) = EXP(-ETAC(1))HYICS=(H(1)+YIC(1)) / (H(1) + YIS(1))53+ LAMBDA(1) = (XIS(3)-YIC(3) + (1.C+H(1))+FLOG(HXICS))/(1.0-ENC(3)) 54+ PS0 = K(I) + E(I) + EYP(-THETA(I)) / (RHO(I)+CP(I)+ LAMBDA(I)+ 55+ THETA(1) + CHI(1))56+ 57-IF (RSG.LT.D.C) WRITE(6.950) 1.RSG.LAMBDA(1) IF (RSO.LT.C.C) RSG=-RSQ 54+ 59. 66 TO 500 60+ 337 CONTINUE RS0 = ALPHA())+U())+EXP(-THETA())/(THETA()+CHI()+C()+U())+ 61+ - ALGG(FSS(1)/(FSS(1)-1.0)) + XIS(1))) 62+ IF (RSQ.LT.0.0) WHITE (6.950)1.850.FSS(1) 63. JF (RSG.LT.0.0) RSG=-RSG 64+ 65+ SCD CONTINUE RB(1) = SCRT(PSQ) 66+

85

Б7• PS(1) = -1.0 + xIS(1)*P8(])*6H0(])*CP(]) 68+ GIR(I) = -FS(I) + TIO(I)FC(1) = -TAUC + VIC(I) €9+ QOUT(1) =-FC(1)+ T10(1) 76+ * RE(1) *RH0(1) * CP(1) QNET(I) = QIN(I) - COUT(I) 71. 333 CONTINUE 72+ 73+ С CALCULATE PUPKING RATES SIMULTANEOUSLY FOR ALL THE LAYERS С 74. 75+ ſ 379 CONTINUE 76 * IF (MCALC.E0.100) 60 TO 777 77+ DO 381 1=1.NLAYER 78* 79+ RET(I) = PR(I) ÷0+ R(RC(1) = RPT(1) + RHC(1) + CP(1)A(1) = ROQC(1)+(T(1)-T(1+1)) + D(1)*PHO(1)*PUT(1) P1. + (1./FSS(I) - 1./FSC(I)) 82. £3. 3F1 CONTINUE WRITE (6.524) %A+ С 924 FORMAT (//4).+1+. E'.+FS+.11'.+*XIS+. 11K.+PC+. 11).+*KIC+. 11*.+5*. 85. 1 11X+*ENC*+ 11Y+*CK*+ 11Y+*LAMBDA*) ê6+ С CO 382 11=1+NLAY1 ₹7• 1 = NLAY1 + 1 - 1I3.8 * FS(1) = 0+0 65. CG 384 JEI. MLAYER 50+ FS(I) = PS(I) + A(J)364 41. PS(1) = -PS(1) / (TIC(1)+RUPC(1)) 52. C(1) = - (FS(1) + 1.0 - H(1)+(1.0 - 1.0/FSS(1))) s 3+ $x_{1S(I)} = P_{S(I)} + 1.0$ 04+ FC(1) = PS(1) + A(1)/ (TIO(1)+RORC(1)) 95+ 56 · $v_{1}c(1) = Pc(1) + (T(1+1)-T0)/T(0)$ ETAC(1) = (E(1)/R) + (T(1) - T(1+1))/(T(1) + T(1))59. ENC(1) = EYP(-ETAC(1))58. GX1CS=(G(1)+(1C(1)) / (G(1) + XIS(1)) 99. IF (CXICS.LT.C.O) WRITE (6.918) I.CXICS.G(I).VIC(I).XIS(I) ¢01 IF (GYICS-LT-C-0) CO TO 777 01+ *18 FORMAT (10x+*FX1CS (*+12+*)=*+F15+6+ *XIC=*+E15+6+*VIS=*+E15+6) 02+ LAMEDA(1) = (/IS(1)-VIC(1) + (1.0+G(1))+ALOG(GKICS))/(1.0-ENC(1)) :3+ RSQ = ALPHA(I)+F(I)+FXP(-THETA(I))/(LAMHDA(I)+THETA(I)+CHI(I)) (4+ IF (RSG.LT.C.D) PSQ=-RSD 65+ 66. RB(1) = SORT(RSG) RORC(1) = RE (1) + RH0(1) + CP(1) 07. A(1) = RORC(1)+(T(1)-T(1+1)) + C(1)+RHO(1)+RB (1) CF+ :5. + (1./FSS(1) - 1./FSC(1)) WRITE (6+925) I.PS(I) +AIS(I) +PC(I) +XIC(I) +G(I) +ENC(I) + 10+ C - LAMBDA(I) 11+ С C 925 FORMAT (3%, 12. 8(3%,E10.5)) 12+ GIN(1) =- PS(1) + (T(1)-T0)+RH(1)+RH(1)+CP(1) 13. С QOUT(I) = -PC(I)+(T(I)-TC) + RP(I) +RHO(I) + CP(I) 14. C ONET(I) = OIN(I) - OCUT(I)15+ C 3R2 CONTINUE 16. WRITE (6,920) (1. ANAME(1.1). ANAME(1.2). RB(1). T(1). C 17+ QIN(I). QOUT(I). QNET(I). FSS(I). FSC(I).I=1.NLAVER) 18+ C 1 ... C STOP IF CONVERGED ENDUGH OF KEEP ITERATIVE CALCULATION 20+ C 21+ С DO 386 1=1.NLAYER 22+ IF (APS(RRT(1)/RP(1)-1.0).LT.0.01) GO TO 386 23+

A PROPERTY AND

. And

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24+ ACALC = NCALC + 1 25+ 66 10 379 ?E+ 386 CONTINUE 27+ 111 CONTINUE 28. WAITE (6.928) NCALC *28 FORMAT (27, *1.0 OF ITERATION = *, 13, /) DO 386 1=1.NLEYER 25+ 50+ 31. GIN (1) =+ ES(1) + (T(1)-TO)+RP(1)+RH0(1)+(*)); 32+ GOUTEID = -POILD+(TEID-TO) + REELD +RHOLL + ...PEED 33+ GNET(I) = GIN(I) - GOUT(I) 34+ 368 CONTINUE

 SEC CONTRACT

 SRSTE (6.920) (I. ANAME(I.1), ANAME(I.42), F. (3), T.1),

 - GIA(I), GOUT(I), ONET(I), FSSELS, FSCELS, I: 1, NLAYFH;

 - GIA(I), GOUT(I), ONET(I), FSSELS, FSCELS, I: 1, NLAYFH;

 - GIA(I), GOUT(I), ONET(I), FSSELS, FSCELS, I: 1, NLAYFH;

 - GIA(I), GOUT(I), ONET(I), FSSELS, FSCELS, FSCELS, I: 1, NLAYFH;

 - GIA(I), GOUT(I), ONET(I), FSSELS, F 35+ 36 * 37. 38+ F10.4. BY. E10.5. BY. E10.5 /) 35+ 10+ ALOSS=0.0 +1+ ED 393 1=1+NLAVER 3-3 4LCSS=WLCSS+P5(1)+PHC(1)+(1./+SS(1)-1./FSC(1)) +2+ WRITE (6.527) WLOSS .3. 427 FOFMAT (//. . WEIGHT LOSS PPEDICTED BY THE MUDILI. /.3 .34(+2+) -. /. 40%-F10.5. . GRAM/SEC.CM2 OF BURKING SUFFACE .) .4 . +5+ С 46+ \$7. C PRELIMINARY CALCULATION PEFORE PLOTTING T VS X 48+ 3 49. TO 1120 I=1+NLAYER 50+ TINTV(I)=(T())-T(1+1))/FLCAT(1C) . TP(J,1)=T(1) 51+ 52+ 00 1121 J=1+10 TP (] + J + 1) = T (]) - T1% TV(1) + FLOAT(J) 53+ 54+ 1121 CONTINUE 55+ 1120 CONTINUE 56* DO 1210 I=1.+LAYER 57+ NSSP(1,11)=NC(1) 59. DO 1220 JJ=2.11 J=12-JJ IF (TP(1,J).LE.TSTART(1)) ASSP(1,J)=1.0 60+ IFITPITADALE .TSTART(3)) 00 TO 1220 61. 62+ CALL FRAGMT (C (1) + E (1) + TH (1+11) + TH (1+J) + HEATRA (1) + NSSP (1+11) + 63+ - NSSPII.J)) 64 . 1220 CONTINUE 65+ 1216 CONTINUE 66+ RHS (NLAYER + 11) = 0.00001 DC 1310 11=1. VLAYER 67+ 68+ I=NLAYER+1-11 69+ RHORD SRHOLISTRREIS 76+ PO 1320 JJ=2+11 71+ J=12-JJ 72+ DTDXK=RHORB+(CF(1)+(TF(1+J)-TP(1+J+1)) 73+ + D(1)+(\SSP(1,J+1)-\SSP(1,J))) 74 . RHS(I,J)=CTDXK+FHS(I,J+1) 75+ DXDT11.JJ=K(I)/RHS(I.J) 76+ 1320 CONTINUE 77. RHS(1-1+11)=PHS(1+J) 1310 CONTINUE 71.0 75. SUM=0.C tū. 00 1330 11=1+NLAYER

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41+		1=NL &YFR+1-11
P2*		DG 1330 JJ=1.5.2
83+		J=12-JJ
84+		Dx(1,J-2)=(DYDT(1,J)+4.+DXDT(1,J-1)+DXDT(1,J-2))+TINTV(1)/3.
85*		SUM = SUM + DY (1 + J - 2)
86+		X(1,J-2) = SUM
87+	1330	CONTINUE
28+	C CLE	AR THE MENORIES OF XPLOT AND TPLOT
29+		D0 1336 I=1.121
50+		x PL 01(1) = 0.0
91.	1338	1PL01(I)=0.0
52+		00 1340 1=1. ALAYER
93*		CC 1340 JJ=1+5
94+		1\D=(1-1)+5 + JJ
95.		J=JJ2-1
96+)FLOT(IND)=SUM - X(I.J)
97.		TFLGT(IND)=TP(I+J)
98.	1740	CONTINUE
95+		1\0=1/0+1
0.0+		YPLOT(]ND)=SUM
01+		TFL 0T (1kn)=TP(1+J+2)
02+		1PL(T=1PL0T+1
03+		IF (IPLOT.GT.5) IPLOT=1
04+		IF (IPLOT.EG.1.AND.NPLOT.NE.NPAGE) CALL ADVPLT
05+		CALL PLFORM (*LINLIN*+2+5+2+75)
06+		CALL PLABEL ("TEMPERATURE PROFILE", 0.
07+	-	- *DEPTH FROM SURFACE (CM)** D* *TEMPERATUPE (K)** D)
08+		CALL PLSCAL ((0.0.10.0).2.100000.(300.0.800.0).2.100000)
09+		CALL PLCSIZ(0.06.0.0)
10.		60 TO (1341+1342+1343+1344+1345)+ IPLOT
11*	1341	CALL OPIGIN (1.C.4.3)
12•		CC TO 134P
13+	1342	CALL ORIGIN (3.0.0.0)
14•		GO TO 134E
15*	1343	CALL DRIGIN (3.0.0.0)
16*		CO TO 1344
17+	1344	CALL ORIGIN (-6.0,-3.1)
16+		60 TO 1349
19*	1345	CALL DRIGIN (3.0.0.0)
20+		NPLOT=NPLOT+1
21+	1348	CALL PLURAF
22•		CALL PLOURV (VPLOT+TPLOT+121+0+"+")
23+		50 10 1111
24 *	1112	CONTINUE
25+		CALL ENUPET
26+		STOP
27*		F 7 0

SUPROUTINE FRAGMT(SB+SE+STI+STF+SHR+SNI+SNF)
DOUBLE PRECISION SV1.SX2.SY1.SY2.DEI
STH I = SE / (1.92717 + STI)
STHF=SE/(1.98717+STF)
SA1=EXP(-STHF)*STF - EXP(-STHI)*STI
SX1 = -STHI
SX2 = -STHF
SY1=DEI(SX1)
SY2=DEI(SX2)
SA2= SY2-SY1
SNF=SH1+EXP(-(SA1+SE/1.9P717+SA2)+SB/SHR)
FETURN
END

BURNING RATE PREDICTION FOR MULTI-LAYERED AIRCRAFT SEAT CUSHION stratestrat

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MATERIAL	PROPERTY								
	TE NP.	RANGE.	THERMAL	HEAT	DENSITY	PREEXP.	ACTIVATION	HEAT OF	HEAT
		(X)	CAL/CH-S.C)	ICAL/6M	C) 16M/CM31		ENLINGTES	ICAL /GM)	RATE C/SEC
1 COVER	r FABR								
	8+8-2-	898.2	• 0 0 0 1 0	•600	. 400	•213+05	24P00.	-2.9	
	813.2-	848.2	.00010	.600	. • 00	.506-07	-20300.	-2.9	
	763.2-	813.2	.06010	.600	• • 00	.224+14	56500.	-2.9	• 33
	689.2-	763.2	.00010	.600	. • 00	•251+00	7900.	-2.9	.33
	602.2-	689.2	.00010	.600	004.	.730-05	-6500.	-16.3	. 53
2 VUNA	->•;>•;>•;	2•719	01000	••••		CO+/CI •	13700.	-16.3	• 33
	A81.0-	923.0	.00020	.345	.146	.337-03	1043-	1.0	
	766.0-	661.0	• 0 0 0 2 0	.345	.146	.166-06	-12285.	9.6	
	709.0-	766.0	.00020	.345	.146	10-468.	7A00.	-23.9	• 3 3
	661.0-	709-0	.00020	540.	.1.6	.237-11	-26532.	-28.7	.33
	623.0-	661.0	• 0 0 0 2 0	546.	•146	•275+05	22078.	-28.7	.33
	591.0-	623.0	.00020	5 4 7	•146	.344-12	-25933.	50.9	• 33
3 15 20	-13.0- 10	0•14C	02000*	040	• 146	•233+05	19435.	107.5	. 53
	876.6-	923.0	•00050	.300	.120	+D-211.	- 3857.	5	
	766.6-	876.6	.00020	.300	.120	.717-10	-24770.	9.6	
	723.0-	766.6	.00020	.300	•120	.517+02	16818.	.8.	
	678.0-	723.0	.00020	• 300	.120	.104-03	-2020-	-19.1	
	673.0-	698.0	.06020	• 300	.120	.H28-13	-31083.	-16.7	.33
	64R . 0-	673.0	.00020	.300	.120	.656-02	2475.	-8.4	• 33
	623.0-	648.0	.00020	.300	.120	.246+07	27897.	-20.3	
	598°0-	623.0	• 00050	- 300	.120	.161-16	-3A190.	7.9	
	573.0-	598°0	02000.	.300	•120	•211+08	27800.	71.7	•33
			• 0 0 0 2 0	000.	. 120	-201+00	6772.	24.9	.33
	-13.0-	0.840	• 02030	-300	• 120	.783+04	18279.	54.9	.33
								•	1
			• • • • •		029.		-13061-	.	
	748.0-		00010		.620	155601	18860.	n w	
	683.0-	748.0	.00010	300	.620	-945-02	- 200		
	623.0-	663.0	.00010	.300	. 620	.1.3-04	-1365.	-11-9	
	573.0-	623.0	.00010	.300	.620	.587-02	3085.	-13.9	
	498.0-	573.0	•00010	• 300	•620	.594+01	10963.	.	.33
	644.0-	688.0	.00010	. 4 00	.030		65757.	1-1	11.
	598.0-	644.0	.00010	. 4 0 0	- 030	•617+02	12335.	11.9	
	573.0-	598.0	.00010	• • 00	.030	.177-01	2646.	2.5	
	544.0-	573.0	.00010	. + 00	• 030	•260+06	21+35.	2.5	• 33
5 THT 5	448.0- -	244.0	•00010	.400	• 030	•600+01	•0066	°.	.33
	L FUAM 1002 0_							٩	
								•	
	915.0-					20-20-2		•	
	A54.0-	915.0	01000.	.200	.00.	-135+02	16862 .		
	796.0-	854 D	.00010	- 200	1021	-611-05	- 7849-		
	711.0-	796.0	.00010	-200	. 023	.107+0H	36705.		
	673.0-	711.0	.00010	.200	.023	.763-02	6938.	, .	

1	COVER FAUR					
		523.2-	602.2	1.000000		
		602.2-	689+2	• 8 2 1 3 2 2		•602153
		689.2-	163+2	6602183	•••	• 45 7 65 1
		163.2-	813+2	•467861		• 172770
		613.2-	648+2	.143770		.027366
		848+2-	898+2	• 059566		.007807
5	VONAR					
		473.0-	5-1.0	1.000000		•86787U
		591.0-	623.0	.865850		. 193606
		623+0-	661+C	. 793600		. /14568
		661.0-	769.0	• 718568		•644733
		709.0-	766.0	• 549 / 33		.575175
		766.0-	841+0	.54F173		.535456
		#81+0-	523.0	• 232426		• 522°00
3	LS 200					
		473.0-	54 P+0	1.000000		. 167082
		548.0-	573.0	. 767082		.934163
		573.0-	598.0	.934165		+471674
		598.0-	623.0	•R71674		.920223
		623.0-	648.0	.820223		.781499
		£48.0-	673.0	.781499		. 125286
		673+0-	678+0	.725286		•694445
		6ab*0-	723.0	.68/495	••	•666312
		723.0-	766.6	.666312		+615242
		766.6-	876.6	.615242		-220334
		+76+6-	923.0	• 550394	••	+ 54 26 58
•	PRFOX					
		494.0-	573.0	1.000000		• 40 31 37
		573.0-	623.0	.953137		.892503
		623.0-	683.0	.892503		•829126
		6P 3 • C =	748.0	•92P126		•76354R
		74R.0-	848.0	.763548		-542460
		848.0-	873.0	.542460		.36/852
_		873.0-	908.0	• 3E 7852		•146667
5	PU FOAM		••• •	1		010467
		444.0-	544.0	1.000000		+710002
		544.0-	573.0	• 319993	-	+0340/0
		573.0-	598.0	.834678		+ 12 19 32
		598 • D=	644.0	.121952		• • • 0 3 7 7
		644.0-	688.0	.490399	-	*00P216
6	INIDE FOAM					
		673.0-	711.0	1.000500		
		711.0-	746.0		**	•~1/053
		796.0-	854.0	• 1/053		+ 3037UE
		854.0-	912.0	. 403906		* P1 7423
		915.0-	973.0	• 67~ 453		+ 3/4FUE
		473.0-	1006.0	*3/4KUH		+ 34 P 1 24
		1006-0-	1073.0	.546/54	••	·213463

NORMALIZED FRAGMENT SIZE AT END TEMPS OF KINETIC REGIMES

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PURVING RATE PHEDICITUR FUR MULTI-LAYERED AIRCRAFT SEAT CUSMION Reconstructures constructions of the second se

and the second second

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RUN 10. 1º 0.5 14CH LS-200 T=6 MIV

MATFR JAL		SURFACE JEMP. (K)	THERMAL CONDUCTIV.	HF AT CAPACITY ICAL/GM+C	DEASTTY DEASTTY DEEM/CM3)	PKFL 1 P. F 1 7 7 7 7 7	ACTIVATION E52%67 *5al/Pulf)	HČAT OF Jecumpos. (Cal/GM)	HEAT Ratf Ise C/SEC)	
1 COVER	FAPR	0.169	.00010	• 60 0	• • 60	.251+00	7800.	- 3.	. 3 3 3	
Z CSVER		659.2	.00010	.600	. + 00	.730-05	-6500.	-16.		
3 LS 20	•	677.0	.000.20	• 36 0	.120	. 828-13	-51043.	-11.	.333	
A LS 20		673.0	.00020	.360	.120	.656-02	24 75.	. 4.	.333	
5 LS 20	0	64M.O	• 00020	. 10	.126	.246+07	27697.	-20-		
6 LS 20	<u>ل</u>	623.0	• 05250	. 30 C	.126	.161-16	-381 -0.	в.	. 3 3 3	
7 LS 20	0	598.0	.00020	.300	.120	•211•08	27800.	12.		
B LS 20	•	573.0	.00020	.300	.120	.261+00	6772.	25.		
9 LS 20	0	548.0	.00020	.300	.120	. 76.3+ 04	18279.	55.	.333	
10 PU FO	=	454°0	.00010		• 0 * 0	•600•01	9900	1.		
REAR	SURF	291.					2 - -			
PREDICTED	N N N	ING RATE								
		HURN NATE ICM/SEC)	SURFACE TEN	10. HEAT FI	LUX IN HO	AT FLUY DUT	HEAT CONSU ICAL/CM2SEC	HED FS-T(JP SURFACE	FS-80TTOP

COVE	R FABR	.00043	691.0	0000	.0396	•000 •	.24964+01	.25137+01
CUV	APA P	• 80 30 •	687.2	.0396	.0372	• 005 •	+ 25137+01	•26443+01
LS 2	00	.00160	677.0	.0369	. 6367	-000-	.35333+01	•36401+01
LS 2	00	.00181	673.0	.0367	.0352	.0615	•36401+01	.45766+01
L5 2	00	16100.	64R.O	•0352	.0337	.0015	.45766+31	• 556 24 • D]
LS 2	00	•6I0 0 •	623.0	.0337	.031R	.0018	.55624+01	• 77926•01
LS 2	0 0	.00215	598.0	.0319	•02F 6	.0031	.77926+01	.15189+02
Ls 2	00	.00234	573.0	. 0285	• 0264	.0023	15189+02	-30379+02
LS 2	JJ	.602 P 1	546.0	.0265	.016.F	.009	+ 303 79 + 05	•10000+C÷
PU F	DAM	.00036	•59.0	.0174	. 6004	.0168	.23784.03	• 10000 • 0

. UDUDH GRAF/SEC.C42 CF HURNING SURFACE

VEIGHT LOSS PREDICIED HY THE POPFL

1*	С	PROGRAM B
2*	С	THIS PROGRAM TAKES THE TEMPERATURE AND CONDUCTIVE HEAT FLUX AT THE
3+	С	FRONT (HOT) SUFFACE (I.E., B.C.+S) AS THE INPUT DATA AND THEY CALCULATES
4 #	C	THE PURNING RATE. TEMPERATURE PROFILE. AND WEIGHT LOSS OF MULTI-LAYERED
5+	С	POLYPERIC MATERIALS (INCLUDING FIRE BLOCKING LAYER) USED FOR AIRCRAFT
6*	C	SEAT CUSHIONS.
7+	C	THE DIMENSION'S USED IN THIS PROGRAM ARE GRAM. GPAM-HOLE. D-CELSIUS.
6 +	C	CALORIE. CM. AND SECOND.
5+	С	
10+		COMMON /TCM1/ANAMET(/,2),MW(7),TREG(7),KT(7,11),CPT(7,11),
11+		1 PHOT (7011)0+7(7011)0ET(7011)0DT(7011)0HEATPT(7011)0TKLU(7011)0
12+		2 TKH1(7+11)+TS(2(1)+N(201)+FS(2C1)+OS(201)+FP(201)+7(201)+
13+		3 NKL0(7,11)+"KH1(7,11)+KL0(7,11)+FKH1(7,11)+ID(8)+THICK(P)+
[4+		4 RMC42Cl)+LAMPC4+R+I+JUI+TC+J1
15*		FFAL KTO NO NKHIO NKLUO LAMADAOMW
16+		DIMENSION HEADIN (16) + WLOSS (7)
17*	C	
10+	C	READ IN THE INGREDIENT MATERIAL PROPERTY AND PRINT OUT
16*	С	
20+		READ (5,=01) ITOTAL
21+		YCI FORMAT (I2)
22+		00 100 11=1,110TAL
2.54	·	RIAD (59902) 1+ANAMET(1+1)+ANAMET(1+2)+MW(1)+NTRE((1)
18.		
2 0 •		NIR-NIR LUIIZ DEAD JE CORVINATION TAUNATION ATAL IN COTATION DUCTATION
(6* 13*		KCAU (20,50))(HEURIOJOINFILIOJOKILIOJOKILIOJOHUT(I))
20.		CONTROJECTALOJECTALOJECTALOJENCATRIALOJECTEATRI
20.4		JOD FONTANIE Ind Fontanie
17-		
11.		THE TOTAL TOTAL TOTAL THEORY TO RATE PREDICTION FOR AND THE AVERTON
12+		- + LIRCRAFT FLAT (USATON) / 2644 64(191) /
330		
14.		- HEATA AN FORMETY PREFAR ACTIVATION HEAT OF HEATA
35+		- 5X / 36X, FOUDUCTIV, CAPACITY, 13X, FACTOR FIFERY,
36+		- 7X. *DECOMPUS. RATE . /
37+		-26) • (K) • 5 × • (CAL/CM.S.C) (CAL/GM.C) (GM/CM3) • 13×.
38+		- (CAL/PULE) (CAL/54) (DEG C/SEC) /)
39+		DO 102 1=1.1TOTAL
NG+		NTR =NTREG(1)
¥1+		WRITE (6.951) I.ANAMET(1.1),ANAMET(1.2)
12+		951 FURMAT (5x+12+2x+245)
43+		WPITE (6,953) (TKLO(I,J),TKHI(I,J),
***		- KT(1,J),CFT(1,J),RHDT(1,J),RT(1,J),ET(1,J),DT(1,J),HEATRT(1,J),
۱5+		- J=10NTK)
¥ E +		953 FORMAT (1790 F7010 "-"0 F7010 1X0
47+		- F10.5.3', F6.3, 3', F6.3, 2', E11.3, 4X, F9.0, F10.1, F10.2)
\$P+		102 CONTINUE
19+	С	
50•	C	FIX CONSTANTS USED IN THIS PROGRAM

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51+
         С
 52+
               F=1.0
 53+
               R=1.98717
               R*=82.06
 54+
 55+
               C+645=0.382
 56+
               FFSLON=C.9
 57+
               SIGPA=1.355E-12
 58+
               TC=285.0
 59+
         C
            CALCULATE N AND FS AT TWO END TEMPERATURES OF KINETIC REGIMES
 60+
         C
            THE CALCULATION SEQUENCE IS. BY LOGIC. FROM LOW TO HIGH TEMPERATURE
 61+
         С
 62+
         C
631
               WPITE (6,955)
 64+
           955 FORMAT (1H1, 10%, *NURNALIZED FRAGMENT SIZE AT END TEMPS OF*, 1*,
 65+
               - *KINETIC REGIMES*. /. 11%. 56(*=*). /)
 66+
               DC 200 I=1.ITOTAL
 67+
                     WRITE (6,951) [. ANAMET(1, 1), ANAMET(1,2)
 68+
                     NTR=NTREG(1)
69*
                     NKL0([+NTR]=1.0
 70+
                     FKLO(I+NTR)=10000.
 71+
               D0 205 JJ=1.NTR
72+
                  J=NTR+1-JJ
73+
                     CALL FRAGMT (HT(1.J).ET(1.J).TKLD(1.J).TKH1(1.J).
74+
                            HEATPT(J.J), WKLD(J.J), NKHILI, J))
75+
                     IF (J.NE.1) NKLO(I.J+1)=NKHI(I.J)
76+
                     IF (NKL0(1+J)+ER+1+0) NKL0(1+J)=0.9999999
77+
                     FKL0(I+J) = 1.0/(1.0-NKL0(I+J))
78+
                     IF (NKH1(1,J).E0.1.0) NKHI(1,J)=0.9999999
79+
                     FKHI(I+J) = 1.0/(1.0-NKHI(I+J))
80+
                      WRITE (6.958) TKLC(I.J).TKHI(I.J).NKLC(I.J).NKHI(I.J)
           958 FORMAT (22"+ +7.1, *-*+ F7.1+ 5*+ F11.6+ 2*+ *--*+ 1X+ F11.6)
81.
82*
           205 CONTINUE
83+
           200 CONTINUE
84+
         C
95+
         С
            INITIALLIZE & FEW VAFIABLES FOR COPPUTER PLOTTING
         C
86*
87+
               NFAGE =9
88+
               NPLOT=1
95+
               IPL OT = 0
50*
               CALL PGNPLT
91+
         C
92+
            READ IN THE INFUT DATA. I.E.. TOTAL NUMBER OF LAYERS AND THEIR ORDER
         C
93+
         С
            THICKNESS OF EACH LAYER (CM), AND FRONT SURFACE HEAT FLUX
94+
            AND FRONT SURFACE TEMPERATURE (K)
         С
95+
         C
96+
          1112 CONTINUE
         C CLEAR THE MEMORIES OF PLOT
97+
               06 1336 I=1+201
98+
99.
               X(])=0.0
00.
          1338 TS(J)=0.0
01+
               PEAD (5,907,END=1110) (HEADIN(I),I=1,16)
02+
           907 FORMAT (1645)
03+
               READ (5.904) NLAYG.(ID(I),I=1.NLAYG)
04=
               READ (5.90E) (THICK(I).I=1.NLAYG)
05+
               READ (5.906) 051.751
06+
           904 FORMAT (1615)
107+
           405 FORMAT (BF10.0)
```

B14

To a later of

and the second second

```
18+
          906 FORMAT(2F10.0)
19+
               WRITE (6.955) (HEADIV(1),1=1,16)
.0+
          559 FORMAT (141.20X.1645/)
11+
               WRITE 16.960)
.2+
          960 FORMAT (10%, *MATERIAL / THICKNESS*,/)
13+
               D0 250 1=1.NLAY6
24=
               101=15(1)
15+
          250 WRITE (6.961) ANAMET(IDI.1), AVAMET(IDI.2), THICK(I)
16.
          961 FORMAT (204, 245, 5%, F5.1, * C***/)
17+
               WEITE (6,964) CS1.TS1
          964 FORMAT (10%, "FOUNDARY CONDITION", /, 20%, "HEAT FLUK (CONDUCTIVE)"
18.
19+
              - + = + F1C+4+ + CAL/CM2+SEC++ /+ 2DX+ +SURFACE TEMPERATURE
                                                                                 = * •
20+
              - F6.1. * K*. />
?1+
        C
:2+
        С
           THE DIMENSIONS FOR TSU KU GSU NU RP HAVE TO BE LARGE ENOUGH TO COVER THE
           WHOLE TEFFERATURE DRCP FROM THE FRONT TO REAR SURFACE IS COVERED
13+
        С
           WITH THE TEMPERATURE INTERVAL (TINTV) AS SHOWN BELOW
...
        С
15+
        С
               11"TV=4.0
:6+
7.4
               1=0
:8+
               TS(1)=TS1
29.
               65(1)=051
56+
               11120.0
31+
               STHICK=C.0
32+
               TWL T= D . C
33+
              00 251 ING=1.7
          251 &LOSS(ING)=0.0
34+
              WRITE (6.965)
35.
36+
          965 FORMAT (11X+1++ 9), "RE++ 5), "TS(1) - TS(1+1)++ 8X, "D#++
37 .
                Ex, *X(I) - V(I+1)*, 5*,*05(I) G5(I+1)*, 10X,*N(I) %(I+1)*/)
              DO 500 III=1.NLAYS
380
39.
               11=1+1
40+
               101=10(111)
               VPITE (6.970) III. AVEMETCIDI.1. ANAMETCIDI.2)
41+
          570 FORMAT 13% 13, 32,245)
424
               ATR=NTREG(1D1)
43+
44.
               STHICK=STHICK+THICK(III)
45+
          5C2 CONTINUE
               I=I+1
$6+
47 .
               TS(I+1)=TS(1) - TINTV
48+
          524 CONTINUE
....
               IF (TS(I+1).LT.TO) TS(I+1)=TO
50+
               IF(TS(I).LE.TKLO(IDI.NTR)) N(I)=1.0
51+
               IF(IS(I).LE.TKLO(IDI.HTR)) GO TO 310
               D0 3C7 J=1+NTR

IF (TS(I)+GT+TFLO(IDI+J)+AND+TS(I)+LE+TKH1(IDI+J)) G0 T0 300
52+
53+
54+
          307 CUNTINUE
55+
           308 J1=J
               JF (TS(1+1),LT.TKL0(101,J1)) TS(1+1)=TKL0(101,J1)
56.
               CALL FRAGMT (FT(1)1.J), ET(1), J), TKLO(1)1, J), TS(1),
57.
58.
              1
                           HEATRT(ID), J), NKLU(IDI, J), N(I))
               1F (K(1).FQ.1.0) FS(1)=10000.
59+
60+
               IF (N(I).NE.1.0) FS(I)=1.0/(1.0-N(I))
               CALL FRAGMT (ETGIDI.J), ETGIDI.J), TKLOGIDI.J), TSGI+1),
61.
62+
              1
                            HEATRT (IDI+J)+NKL0(JD1+J)+N(I+1))
               IF (N(I+1).EQ.1.0) FS(I+1)=10000.
63+
               IF (N(1+1).NE.1.0) FS(1+1)=1.(/(1.0-N(1+1))
64+
```

£5+	C				.												- .										
66+ 67+	C C		D THE	A S	THE. Sum	90 60	HE A	(% G (T	FLU	T L Y	R (√)		810 85	5 1 Aƙ	SU	. H IKF	L A A C	T I E	FLU TEP	IX IPEI	A T R A T	URE	FR)	GNI	ſ		
68* 68*	C		C & L 1	R II	RAD	Ŧ																					
70+			RHOL	1)=	RHO	T (T	014		>																		
71+			DIDY	= 2 5	(1)	/	(10)1.	J1)																		
72+			DX = (151	1)-	151	1+1	22	107	٥x																	
73•			60 T	0 4	05																						
74+		310	CONT	INU	Ε																						
75+			IF (TSC	1+1).L	£ • 1	KL	0(1	ЪI	• \1	FR))	N C I	+1)=	1.	0									
76*			DENO	M=K	нот	(1)	1.1	1TE) • C	ΡT	(1)	1.	ΝT	F) •	1 (T	S (D	- T	0 >								
77•			RPCI)=0	SEL	>/0	ENC) M.		_																	
764			£ • = K	Tel	DI +	9. T F.) • 1	ITS.	())	- T	S ()	+1	"	/([Et	٢ ١	1 R	H ()	[]])							
90- 199			- NFL (1)=	-00	1 1 1	114	- 13 I I 4 1 T	4) 4 1 7				EL										۰ – T	c , ,			
6 J 4		4.06	0201	* 1) * 1 1	- 63 r	(1)	- r r	101	• • •	1.4			КГ	• • •	Ψ	r i				- /		311	1-1	211	4111		
01- 82:		469	- 22.00	11=	r x f T) + D	•																				
ः २ - २ २ -			18 1	455	141	1+1	515	7 F	104	- 1	. 03		τ.	c	:1)	5	5	10	40	: a							
64+			1F (145	105	(1+	1))		[.1	<u>،</u>	E = 1	(4)	6	0 1	U	49	9										
85*			15 0	AES	(TS	(1+	1)/	110	-1.	0)	•11	• 6	. 0	035	5)	60	1	0 4	409)							
86+			IF C	xil	+1)	LT	• S T	14	cĸ.	4 V	n.,	S (1+	1).	6.1	• 0	• 0))	IK J	TE	(6	• 97	2)	1.0	S(I)	•	
87•			-	TS (1),	t s (1+1		•۲°	y ę	1),	5.0	I +	1).	ίúS	(1) • (as i	114	1),	NI	1).	NС	+1)			
88+		572	FORM	A T 6	1 C X	• 1	3.	FΣ	°.5		3)	(F 6	• 1 •	•	-	-•	• 1	6.	1+	5X	• F	ۥ2	, F	8.4.	* - * •	
89±		•	-		FR.	4 .	2, 1	F	10.	4	-F I	Û.	4.	21	•	F1	0 •1	5.1	-10	5)						
96+			IF (Y (]	+1)	• L T	• S 1	4 I I	C*•	<u> 1</u> 1	n•0	51	1+	1).	67	• 0	• 0) (60	10	50	2					
91•]F (Y (]	+1)	• C E	• 5 1	њI	C K)	- P	FAC	; 1 =	15	141	IC K	- ((1	>>	יח/	r 							
92+			IF (CS (1+1) • L	1.1)	C.F.	401	1=0	51]}/	(1	51	D	-0:	s ()	+1							
93.			1511	+1)	=15	(1)	- (1	50	11-	15	•1•	•1)	1.4	F # 6	C I												
74=		4 9 9		0 2 101	С. Г																						
94.		• • •	- C C 7 1	5 4	6.9	7 7 3	۲.			. T	< , 1	1.	1 <	() -	. 1 1												
57 *		•		r / .	x (1	1.1	e i •	1.		•		si	1+	1).	5.0	'n	• •	(1)	• 1 1								
9 µ •	С		нисте	***	NC	1+1) (Υ.F	AHO	VE	- N 3	ιĩ	Ē	E F	EP	LA	έċ	bΪ	ΡŸ	TH	F	IRS	то	FΥ	HE 1	JE XT	LAY!
39.			12=1																								
00+			00 6	0.0	J= 1	1.1	2																				
C1*			w11=	F. F. (J) =	r HO	(3)	• • •	•. (J	+1) = !	IJ))														
¢2•			15 (KL T	•LT	•[•	()	.	11.	f t	• • •	16)	J	• 6 6	. T •	¥L.	ΰS	S (111								
C 3 •		976	EC. P. M	4 T	110	۷.	• 5° L	, T C	••	12	• ');	••	F 1 (• 5	•	•		PL C	155	• •	F10	• 5)				
C4+		€00	WLOS	SII	11)	= Mr	CSS	5 (I	11)	* 5	LT																
05*			TWLT	= T ¥	LT+	wite	550	11	1)																		
06+		568		1.40							e . 1	• • •	•					• •									
0/= 0u+			- 6 -7 5 8	2 9 	614	221		111	• #L	() ()	211	14		111	1	•	1	1 U 4 V I	, , , ,		٢1	1.4	. 1				
05+		~~~~	- 4 3 4 4		13.		1 4 3		-+.	53	1.4		` •					~ ' '			- 4	1	• ·				
16+			2 	F 1	6.9	773	7.	<i>и</i> т	- •		4.0.	• • •															
11+		977	EGRM	AT	121	.51			•	T	61	:1	* £	55	LC	ss	=	• • •	E 1.)		. •	G M	/ SE	c.c	M2 (F .	
12+			- •P	URN	ING	su	6 F J	ICE.	• >		• ·				-							-					
13+			IPLO	1=1	PLC	T+1		• -																			
14+			75 6	IPL	01.	6T.	5)	IP	LOT	= 1																	
15•			16 6	IPL	.01.	EG.	1.4	•••	• NP	10	1.1	ŧ٤.	NΡ	AGE		C 4	LL		DVF	PL T							
16•			CALL	. FL	F 09	M (•L 1	P.L	IN•	• 2	•5	2.	75)													
17+			CALL	PL	APE	L (• 1 8	MP	ERA	τu	٩î	Př	υF	16	•		0.	_				_	_				
10.		•	- • 2	EPT	H F	PGM	SU	JRF	ACE	L	CMI	•		0.	• 1	EM	PEI	RA	TUP	E I	(K)	•	0)				
15.			CALL	PL	SCA		(0.	0.	16.	6)	• 2 •	10	СC	C C (. € 3	000	• 0	• * •	06.	C)	2.	100	000)			
20+			LALL	PL	CST	2 4 6	• 06	5 . [•0)		74.			<i>.</i> .		Di											
∠1*			60 T	0 (134	1 4 1	342	(• 1	3#3	•1	544	+ + 3	• د	516	, 1	٢L	וע										

120	1341	CALL CRIGIN	(1.0.4.3)
23.	••••	60 10 1348	
244	1342	CALL CRICIN	(3.0.0.0)
25.4		GO TO 1348	
264	1 3 4 3	CALL DRIGTS	13 0.0.01
10-	1040		
s / •		60 TO 1945	
58*	1344	CALL CPIGIN	(-6.(+-3.1)
29+		60 10 1348	
30+	1345	CALL CRIFIN	(3.0.0.6)
31+		NPECT=NPLOT+	1
82+	134P	CALL PLOPAF	
33-		CALL PLOURV	(x.TS.201.0.***)
54 *		60 TO 1112	
55+	1110	CUTTINUE	
36+		CALL ENDPLT	
37+		STOP	
380		E1 D	

4 AY

1+	SUPROUTINE FRACET(SH-SE-STIFSTF-SHR-SNIFSAF)
2•	DOUBLE PRECISICS SX1.SX2.SY1.SY2.DEI
3+	5TH1=SE/#1.5E717+57J)
4+	STHF=SE/(1+98717+STF)
51	SAI=EXP(-STHF)+STF - EXPI-STHI)+STI
6•	5×1 = -STH1
7.	5Y2 = -5THF
P 4	SY1=DEI(S(1)
G 🛊	SY2=DFI(SY2)
10.	SA2 = SY2-SY1
11*	SNF=S41+EXF(+(\$41+SE/1.90717+S42)+88/SHR)
12+	RETURN
13•	END

1+	SUBROUTINE EURNRT
2•	COMMON /TCM1/ANAMET67023.MW6 73.NTRE56 73.KT(70113.CPT(70113.
3+	1 FHUT (7+11)+(T (7+11)+CT (7+11)+CT (7+1,)+45 ATP T (7+11)+TKL 0(7+11)
4.	2 TKH147+11)+TS(201)+'.(201)+FS(201)+JS(201)+RP(201)+X(201)+
5×	3 NKL9(7+11)+NKHI(7+11)+FKL9(7+11)+FKHI(7+11)+ID(9)+THICK(R)+
£*	4 RH042013+L4MH0A+R+I+I01+T0+J1
7+	REAL RT. N. NHHI. NKLO, LAMBDA,MW
£ *	1F (1.FG.1) P+(1)=0.0G1
9+	1F (I.NE.1) RR(I)=RR(I-1)
10+	IF (N(I)-EC-1-0) 60 TO 378
11+	377 CONTINUE
12+	PHT=RR(I)
12+	PS=-GS(1)/((1S(1)-TO)*R((1)*RH(T(1C]*J1)*CFT(1D1*J1))
14+	H=DT(1D1+J1)/(CPT(ID1+J1)+(TS(I)-TG))
15+	G=-PS-1.0+H+N(])
16•	THE TA=ET(101+J1)/(P+TS61))
17+	ETAC=(ET(]L]+J])/R)+(TS(])-TS(]+])/(TS(])+TS(]))

18.	
15.	CHI=(TS(1)-TC)/TS(1)
20+	ALPH4=KT(101+J1)/(RH0T(101+J1)+CFT(101+J1))
21+	L 2MACA = (H+ (7113)-7(1+1))+(1+0+6)+AL 06(7(1+1)/7(1))/1(1+C-LFC)
22+	RSG=&LPH&+PT(1D1+J1)+EXP(+THETA)/(LAMPD&+THETA+CHI)
23+	IF (FSU.LT.C.C) WRITE (6,910) I.TS(I).TS(I+1).RSQ
24 .	918 FCRMAT (3X++1=++13+ + TS(1)=++F6+1+ + TS(1+1)=++ F6+1+ + RSQ=++
25+	1 E1C+4/)
26+	JF (FSG.LT.D.C) RS(=-RSQ
27.	FE(1)=SGRT(RSQ)
28+	IF (4HS(RHT/RB(I)-1.0).67.0.605) 60 TO 377
29+	376 CONTINUE
36+	(3(1+1)=QS(J)=RH(I)+RHQT(101+J1)+(CPT(101+J1)+(TS(1)=TS(1+1))
21+	1 + CT (10 1 + () (1 + 1) -) (1 >))
32+	Pr TUP*
33+	ENC

20.20

PUPVING RATE PREDICTION FOR MULTI-LATERCO AIRCRAFT SEAT CUSHION

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			1838 1888 1888 1888 1888 1888 1888 1888	****		······································	*******		
MA 1 E R	IAL PROVERTY								
		• 30.44			, rt vst i v	FULLER.	VOLTAVITA VOLTAVITA	HEAT OF	HT AT
		(*)	1CAL/C4.5.C)	EAL VEN	(CH2/28) (3.		ICAL/HOLF)	ICAL /6'1)	IPES C/SFC
	OVER FAME								
•	848.2-	898.2	-00010	.300	00**	•213•05	24800.	- 2.9	
	A13.2-	848.2	01000.	.300	• • 0 0	-506-07	-20360.	6 • 3 •	
	763.2-	P13.2	. 0010	.300	00	+1++22+	56530.	-2.9	. 33
	6A9.2-	163.2	.00010	.300	• • 60	•251+00	7500.	-2.4	. 5 3
	602.2-	669.2	01000.	.306	904 *	.730-05	-6503.	-16.3	6
•	- 7°C 7C		A T A A A A			CO+151.	• 0 2 / 5 1	-16.3	
>	U.A.F. AR1.0-	923.0	. 0 0 2 0	. 145	- 146	10-255-	1043.	1.0	
	766.0-	AUI,0	.0020	04%	441.	.165-06	-12285.	9°6	
	705.0-	766.0		540.	.146	. 494-01	7800.	-23.9	
	661.6-	109.0	.0000	-345	.1.6	.237-11	-26532.	-28.7	
	623.0-	661.0	.0020	.345	.146	•275+05	22079.	-28.7	.33
	-0-165	623.0	• 0 0 C 2 0	19 1 9 19	.146	.394-12	-25933.	50.9	
	473.0-	5°1.0	.00020	345	.146	•233+05	19935.	107.5	. 5 3
- -	5 200 417 6-		0.000					·	;
	766.60	1 2 2 2 B							
	72.0-0-0-0-0-0-0-0-0-0-0-0-0-0-0-0-0-0-0-	766.6	.0000		001.				
	696.00	721.0	.0000		121		- 2020-		
	673.0-	698 ° 0	-00020	300	.150		-11045-	-17.1	
	644°0-	673.0	.00520	COC.	120	.656-02	2115		
	623.0-	644.0	.00020	. 300	.120	.246+67	27497.	-20-5	
	596.0-	623.0	.00020	.300	.120	.161-16	-381-0.	1.9	
	573.0-	59.6.0	• C 6 N 2 C	.300	.120	•211+0P	27400.	71.7	. 33
	-0- K+0-	573.0		00.	.120	•201•60	6772.	24.9	.33
•	-0-C1+	0 • R • 1	*00050	• 300	.120	•783•0•	14279.	54.9	•33
,	R73.0-	906.0	- 000 34	100	027	543-05	-11061-	۲. ۲	
	-0 - 4 4	P73.0	+C000*	.300		• • • • • • • • • • • • • • • • • • • •	91579.	,	
	746.6-	84R.0	•2000 •	• 3 6 0	• 620	.155+63	19863.		.33
	6PJ.0-	748.0	•00034	001.	.620	.545-02	4500.	-13.9	. 33
	623.0-	663.0	.0003+	. 306	.620	.143-04	-4365.	-13.9	. 33
	-0.576	673.0	40004		• 6.20	.587-02 	3085.	-13.9	
ē.	U FOAM					70.26.04	16753.	•	
	644.0-	688.0	.000.	. 4 0 0	. 030	.644.20	65757 .	-3.1	.33
	-0	644.0	4 × 000 •	900	.030	•617+02	12335.	-11.9	• 3 3
	573.0-	148°0	.0034	- + 00	.030	.177-01	2646.	2•5	• 33
					. 030	•260+06	21435.	2•5	
9	MIDE FCAR				000.	• • • • • • • •	• 0 0 6 6	•	
	1006.0-	1073.0	. 60C1C	•200	• 02.5	.106-04	-4763.	-	. 13
	973.0-	1006.0	.00010	.200	. 023	.312-13	-46170.	<u>،</u>	
	915.0-	973.0	.66310	.260	. 023	.1.0-06	-16674.	. .	. 33
	8.4.0-	915.0	.00010	.200	. 623	•133•02	16862.	- 7 - 4	. 33
	796.6-	654 C	C C L J D •		.023	e11-05	- 7409.	-7.5	
	111.00	776.0	.00010	.200	- 023	.107+0A	36705.	- 7.0	
		1.11.0	91303.		.023	.763-02	6934.	•	

	NORMALIZED	FRAGMENT	SIZE AT E	40 TEMPS OF KI	NET10	PEGIMES -
		===========				
1	COVER FARR					
•	•••••	523.2-	602.2	1.006000		.821322
		602.2-	669.2	.821322		-602183
		689.2-	763.2	.602183		.467861
		763-2-	813.2	467861		. 193770
		813.2-	848.2	193770	'	.059566
		848-2-	896.2	. [59566		.007807
2	VONAR			•••••••		
-		473.0-	591.0	1.000606		.865850
		591.0-	623.0	.865850		.793600
		623.0-	661.0	.793600		.718568
		661.0-	709.0	.716566		.644733
		709.0-	766.0	.644735		.598173
		766.0-	881.0	.548173		+535456
		861.0-	523.0	.535456		.522900
3	LS 200					
		473.0-	548.0	1.000600		.967082
		549.0-	573.0	.907082		.934163
		573.0-	500.0	.734163		.871674
		59A . 0-	623.0	.871674		820223
		623.6+	648.0	. 620223		.781499
		648-0-	673.0	.781499		.7:5286
		673.0-	698.0	125286		-688495
		698.0-	723.0	. 688455		.666312
		723.0-	766.6	•666312		+615242
		766.6-	876.6	-615242		-550394
		876.6-	923.0	.550394		-542658
4	PREOX					
		498.0-	573.0	1.000000		.953137
		573.0-	623.0	•953137		-892563
	•	623.0-	683.0	.852503		.826126
		683.0-	748.0	. 828126		.763548
		748.0-	848.0	-16354A		.592460
		848.0-	873.0	.542460		.367852
		873.0-	908.0	. 3E7852		.146867
5	PU FOAM					
		448.0-	544.0	1.000000	••	.915893
		544.0-	573.0	.916883		.834678
		573.0-	598.0	.834678		. 12 7952
		598.0-	644.0	.7-1952		.490399
		644.0-	688.0	.490395		.006318
6	IMIDE FOAM					
		673.0-	711.0	1.000000		.994404
		711.0-	796.0	.994404		.917053
		796.0-	P54.0	•917053		.803906
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APPENDIX C Some Comments on Burning Tests Procedure

The burning tests of multi-layered seat cushion assemblies were conducted in an NBS Smoke Density Chamber. The orginial radiative heat source is replaced by a Mellen furnace Model 10, to provide a high heat flux, and further supplemented with a cantilever beam coupled to a transducer, to measure weight changes of suspended samples.

In addition to these, there were several modifications on the procedures of sample preparation and it's burning. These were instated following the detection of unsatisfactory conditions during the initial trial runs. These are summarized below.

Symptom	Diagnosis	Remedy
A) Uneven heating (Severe 2-D effect)	1. Misalignment of sample with furnace	1. Align correctly
	2. Use of stainless steel fram as sample holder	2. Eliminate the frame, i.e., large heat sink item
	3. Loss of heat on four sides	3. Use of light, thick insulator on four sides, also as sample holder
	4. Buoyancy of hot gas formed in foam (open cell)	4. No remedy unless sample positioned horizontally
B) Irregular surface shape (or disengaging the cover sheet in the worst case)	High pressure exerted by the gas product during intumescence	Cover fabric cut larger than the surface area, folded over the sides and stapled
C) Temperature rise at the sample surface before removing	Single sheet of radia- tion shield not enough	Use double layers (with air-gap in between) of shields

Symptom	Diagnosis	Remedy
D) Furnace temperature change on and after removing the shield	Change in Characteris- tics of reradiation from the shield and the sample surface	Adjust the power to the furnace manual' to maintain the same furnace temperature
E) Non-linear signal response of trans- ducer	Electronic feature of the equipment used	Make and use a calibration chart
F) Irreproducible weight measurement of sample	Irregularly distri- buted loading on the cantilever beam	Hold the entire sample by a single wire and load it at the fixed point of the beam
G) Step changes in weight measurement during the burning	Interference by the thermocouples inserted into the sample	Duplicate the run without thermo- couples purely for weight change measurement

APPENDIX D Photographs of Equipments and Burned Samples

Photo No.

Remarks

1. Thermogravimetric Analyzer (TG)

The thermogravimetric analyzer is shown in the right side of the picture, connected to a pair of flowmeters and a mixer, plus a vacuum gauge (vacuum pump not shown). The assemblage of a chart recorder, a microprocessor, a heater controller, and a weight suppressor are shown in the left side on the table.

2.

Differential Scanning Calorimeter (DSC) The DSC is shown here with a SAZ (Scanning Auto Zero) on its top. The sample heater unit is located on the deck on the left side. The DSC is not used extensively in this work.

3. NBS Smoke Density Chamber

The NBS Smoke Density Chamber is shown at left rear. A newly installed Mellen furnace is seen through its window. The multichannel recorder (Leeds & Northrop, Speedomax Model 251) is shown at the top of the front cabinet.

Sample Preparation 4.

This picture shows each seat cushion sample is prepared for the burning tests in the NBS Smoke Density Chamber. The front cover fabric is cut larger than the face area and folded over the sides and stapled. The entire sample is surrounded by four 1"-thick insulator panels (Kaowood-3000° Board, by Pyroengineering Co.).

Sample Arrangement Inside of the NBS Smoke Density Chamber 5.

Eight pairs of thermocouple (Pt vs. Pt-10% Rh) wires are shown here penetrating into the seat cushion assemblage, which is suspended by a cantilever beam. The beam is a part of a transducer-type weight measuring device. The Mellen furnace is shown at the lower left corner.

6. Visual Comparison of the Burned Seat Cushions

Each seat cushion system was suddenly exposed to the Mellen furnace preheated and maintained at 850°C and then remained so for 10 minutes. This pictures shows that the effectiveness of the fire blocking layer can be rated, based on this test only, in decreasing order: LS-200 $(1/2" \text{ thick}) > \text{Vonar } (3/16" \text{ thick}) > \text{Preox } (11 \text{ oz/yd}^2) > \text{No blocking}$ layer.

7. Comparison of the Degree of In-Depth Burning

Same as above, but this picture shows the degree of in-depth burning. with the samples oriented in the same way as in the NBS Smoke Density Chamber.

Comparison of LS-200 and Vonar as a Fire Blocking Layer 8. In this picture, LS-200 is shown more effective as a fire blocking layer than Vonar when tested with the same thickness.

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APPENDIX E Comparison With Factory Mutual's Numerical Model

The thermochemical model (TCM) of this work is compared with a numerical model developed by Factory Mutual Research Corporation. The numerical approach is described in Appendix A, "A Numerical Model for One-Dimensional Heat Conduction with Pyrolysis in a Slab of Finite Thickness," by F. Tamanini, in Technical Report, FMRC Serial No. 21011.7 dated November, 1976, entitled "The Third Full-Scale Bedroom Fire Test of the Home Fire Project," Vol. II, edited by A.T. Modak.

The computer calculations were actually performed with essentially the same program as listed in the Report, in order to enhance the familiarity with this numerical model (NM). The boundary value read-in instructions were rewritten to suit this comparison purpose, as suggested by the author.

The numerical model (NM) presented in the Report is for a single-layered material: however, the extension to a multi-layered system retaining its features seems possible without difficulty. The change of total weight due to pyrolysis is not included in the original program, but also can be added with ease. Since such extended/modified computer program is unavailable, the comparison of the result of the numerical model with the experimental data of this work is not attempted.

A) Model

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$$TCM : k \frac{d^2T}{dx^2} + \rho cr \frac{dT}{dx} = D\rho NB \exp(-E/RT)$$
$$\frac{dN}{dE} = -NB \exp(-E/RT)$$

$$NM : \frac{\partial}{\partial x} \left(k_{s} \frac{\partial T}{\partial x}\right) + \frac{\partial}{\partial x} \left(M_{g}h_{g}\right) = Q \frac{\partial \rho_{s}}{\partial t} + \frac{\partial \rho_{s}h_{s}}{\partial t}$$
$$\frac{\partial \rho_{s}}{\partial t} = -\rho_{a} B \exp \left(-E/RT\right)$$

For NM, M_g is the mass flux of volatiles, h is the enthalpy, while subscripts s and g refer to the solid matrix and pyrolysis gas respectiely. Another subscript a refers to the unpyrolyzed active material.

TCM	NM
Condensed phase reaction	Same
One - dimensional	Same
Steady state	Unsteady state
Moving coordinate system	Stationary coordinate system
First order Arrhenius reaction for thermal degradation	Same
Analytical solution available	Numerical solution technique presented

B) Dimensional Analysis

The dimensional analyses are performed for these two models to determine their relative strengths and shortcomings. The dimensionless groups needed to describe these models, according to Pi theorem, could be the following:

Dimensionless Group	JPL'S TCM	FMC'S NM
Temperature	$\frac{T - T_0}{T_s - T_0}$	$T - T_0$ $T_s - T_0$
Distance Coordinate	Prcx k	Mgcx k
Heat of Pyrolysis	$\frac{D}{c(T_s - T_0)}$	$\frac{Q}{c(T_{s} - T_{o})}$
Arrhenius Group (or Activation Temperature)	E RT	E RT
Preexponential Factor	Bx r	Bt
Fourier Number	-	$\frac{kt}{\rho_{cx}^2}$

The obvious difference is that Fourier number, which compares a characteristic length dimension with an approximate temperature-wave penetration depth for a given time t, is included in the NM while it is absent in TCM. Other than the capability of the NM to study unsteady-state behavior, essentially the same dimensionless groups are required for the analysis of the two models.

C) Input Data

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	TCM	NM
C _p , k	1. For one solid	 For two solids (active material and char) and one C_p for gas
	2. Stepwise variation with temperature	2. Linear variation with temperature
ρ	Stepwise variation with temperature	The initial and final densities (char fraction increases linearly from 0 to 100% of apparent density)
Β, Ε	Change often (as evidenced by TGA) depending on tempera- ture range	One set of constant values for the whole temperature range
Heat sink term	D: heat of degradation stepwise variation with temperature (to be obtained from DSC)	Q: heat of pyrolysis fixed at a reference temperature
Boundary Conditions	 Temperature B.C.'s at both boundaries or Temperature and heat flux B.C.'s at one boundary 	Either temperature or heat flux B.C. at each of two bounda- ries

D) <u>Calculation & Results</u>

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	ТСМ	NM
Calculation	Segment-wise application of analytical solution, progressively advancing from one boundary to the other	Finite difference representations of the partial differential equation for the entire region and solving the resultant tridiagonal matrix
Results	surface regression rate, temperature profile and total mass loss	temperature profile, density profile, and mass flux profile

E) Advantages & Disadvantages

	тсм	NM
Advantage	Parametric analysis is readily accessible - essential for basics understanding and estab- lishing design criteria	Time-dependent pyrolysis behavior can be pre- dicted (This may be useful for assuming and confirming time- dependent boundary conditions more realis- tically.)
Disadvantage	Steady state solution is not adequate for study of transient period	Parametric analysis may be not as clear as TCM

