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for six weeks retained 30 square meters per gram with a coating of 2.2 mg/h/, or 1.5 monolayers.

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#### Problem Studied

Missile propellants are fast-burning dispersions of ammonium perchlorate (AP) in an organic binder. Some of these dispersions are made with finely ground AP with average particle diameters less than one micron. The burning rate of these ultrafine AP propellants is quite dependent on the specific surface area of the AP. At the MIRADCOM Propulsion Directorate of Redstone Arsenal accurate specific surface areas of "ultra-fine" AP have been measured and it has been determined that humidity can cause a rapid reduction in the specific surface area of the AP. This effect is considered responsible for the significant decrease in burning rate (and thrust) of such propellants upon storage under humid conditions.

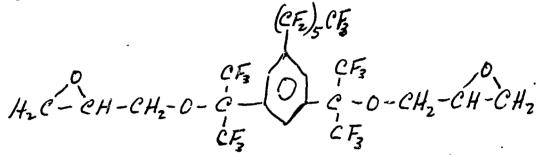
We have studied the use of a novel water-repellant grinding aid, introduced at the time of grinding ammonium perchlorate to the ultrafine state. This grinding aid is the highly fluorinated epoxy system developed by J. R. Griffith of the Naval Research Laboratory.<sup>1-3)</sup> It is soluble in the Freon liquid now in use for grinding ammonium perchlorate, adsorbs and polymerizes the surface of the particles, aids in grinding, and forms a cross-linked adsorbed monolayer which has enough alcohol groups to be bonded into the urethane rubber matrix.

We have tested the coated ultra fine powders for their sensitivity to moisture by storing the powders in dessicators with controlled humidity (35%, 51%, and 79% R.H.), and by measuring their specific surface acids with argon adsorption, using the BET method.<sup>4)</sup>

### Experimental Details

#### Additives.

The fluoro-epoxy used in this work is the NRL "C-6" diepoxide of mole weight 837:

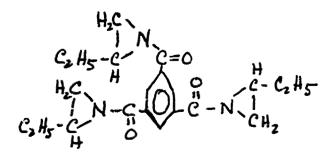


which was polymerized with a silicone diamine (bis y-aminopropyl tetramethyl disiloxane from Bergstrom Associates in Cincinnatti, Ohio):

 $(H_2 N - (CH_2)_3 - S_i - O - S_i - (CH_2)_3 - NH_2$  $(H_2 N - (CH_2)_3 - S_i - O - S_i - (CH_2)_3 - NH_2$ 

The amine and epoxy form a cloudy solution in Freon 112, but on heating five minutes at  $50^{\circ}$ C. the solution becomes clear as the prepolymer forms.

Since the current process at Redstone Arsenal involves using 3M's aziridine HX-868 as a grinding aid, we used it in our process, also.



Molecular models indicate the aziridine occupies 135  $Å^2$  per molecule while the epoxy and silicone occupy about 100  $Å^2$  per molecule, when specific groups are all in contact with the surface. It is expected that the aziridine and amine will react quickly and adsorb on the salt surface with the fluoro-epoxy in contact with the Freon. A complete monolayer of fluoroepoxy (at 100  $Å^2$  per molecule) will require 1.4 mg per square meter of salt surface. <u>Milling Procedure</u>. Ammonium sulfate was used as a safe substitute for ammonium sulfate was used as a safe substitute for ammonium perchlorate, and each batch of 5 g of salt was ground in 55 ml of 1,1,2-trichloro-1,2,-trifluoroethane, either at 25°C. in an oven at 50°C. The jar and cylindrical grinding media were of a

# Results

ceramic material, and the 8.5 cm diameter jar was rolled at 150 rpm.

Quite a few batchs were made, but data on the 24 more significant batches are shown in Table 1. Variations in amounts of additives, in milling time and temperature, and in storage times and humidity were explored. The surface areas were determined on all of these batches, using argon adsorption and the BET method.

The fluoro-epoxy system is shown to be an excellent grinding aid, providing surface areas of 50 square meters per gram, and usually allowing the powder to grind down until only one monolayer of fluoro-epoxy is left  $(1.4 \text{ mg/m}^2)$ .

In some batches good resistance to loss of surface during exposure to humidity was abserven, especially in batches 18-24. Poorer resistance was observed when the time or temperature allowed for epoxy polymerization was insufficient. Milling at room temperature for several hours was successful, but only 15 minutes was disastrous, and 120 minutes much better. Milling at 50°C.

was sometimes better than at room temperature.

The weight ratio of epoxy to amine turned out to be quite important. The exact stoichiometric weight ratio (two diepoxide for each diamine) was 6, but better results were obtained when there was excess amine (weight ratios of 3.6 or 3.8). The excess amine can help in adsorption, and might be necessary because of reaction with the aziridine.

#### Recommendations

1. The fluoroepoxy system is an excellent grinding aid. The aziridine may not be needed.

2. Excess amine appears desirable to promote better properties. It probably assists in the adsorption.

3. Grinding times and temperatures after epoxy addition should be so chosen that the polymerization is complete and so that only a monolayer of the fluoroepoxy results.

4. If there is interest in this process, the next stop, binding of the epoxy into the urethane system, needs to be tested.

#### Publications

None of this work has been published, and there are no plans to do so at present.

#### Personnel

The experimental work has been done be two graduate students, Yale West and Joseph Sohara. Mr. Sohara has completed most of his M. S. degree studies on this project.

SURFACE AREAS OF AMMONIUM SULFATE POWDERS STORED IN HUMID ATMOSPHERE AT 23°C. TABLE I.

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Mg Epoxy Ŧ 1.3 1.9 5.6 1.8 2.5 2.4 7.5 10.9 12.0 3.5 4.2 5.6 1.4 1.8 5.1 1.4 2.5 2.2 00 00 Poor to fair Reliability Excellent Good Excellent Excellent Excellent Excellent Excellent Excellent Excellent V. Good V. Good V. Poor Good Poor Fair Good Poor Poor Fair Good Good Good BET Results M<sup>2</sup>/g <u>C</u> F 9.7 8.4 9.3 10 48 12.9 37 13.5 12.7 8.5 25 61 12.3 9.7 183 20 29 450 8 22 8 14 3.4 11.3 17.9 12.6 15.0 4.3 6.9 6.3 1.0 1.0 17.9 15.0 11.3 53.3 42.6 14.6 47.7 52.4 28.6 32.2 Storage Conditions IRII. 51% 51% 51**%** -51**\$** 35**%** 51**%** 79**%** 35**%** 51**%** 79**%** 35% 51% 79% 35**%** 35**%** 51**%** 2.5 weeks 3 weeks 5 weeks 2 weeks 3.5 weeks one week one week 3 weeks weeks weeks weeks weeks 6 weeks 6 weeks 6 weeks 6 weeks 2 weeks weeks weeks weeks weeks 4 weeks Time none none none 20 ເດເຕ S ν Milling Conditions 2000 2000 50°C. 50°C. Temp RT **R**1 **R**1 RT RT 222 **R R T R T** RTRT hours hours hours hours hours hours min nin min min min min min min hours min Min Time min nin min min N N 120 120 120 120 120 120 120 155 60 60 24 24 48 48 73 88 73 Ratio 5.2 3.6 3.5 0 0 0 0 രര **Amine** 0000 4444 0.00 0.2 1.2 2.1 0.1.0 00 00 Additives **U**Epoxy 4444 ~~~ ~~~ 6.3 6.3 6.3 7.5 ..... 00 ဂဝ HX-868 0.2 1.2 N N N N 1.3 0.6 0.6 2.3 0.2 4444 Batch ŝ 521 1212 15 17 2018 5000 2005 0000

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## References

- James R. Griffith and John D. Bultman, Ind. Eng. Chem. Prod. Res. Dev. <u>17</u>, 8 (1978)
- Donald L. Hunston, James R. Griffith, and Robert C. Bowers, Ind. Eng. Chem.
  Prod. Res. Dev. <u>17</u>, 10 (1978).
- 3. Don Dagani, C & EN, p.30 (July 14, 1980).