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- Phys., vol. 44, pp. 548-555, 1966.
- 51. J. F. Bott and N. Cohen, "Temperature dependence of V-V and V-R,T energy transfer measurements in mixtures containing HF," J. Chem. Phys., vol. 59, pp. 4539-4549, 1973.
- J. F. Bott, "Vibrational relaxation of HF (V=1, 2, and
   in H<sub>2</sub>, N<sub>2</sub>, and CO<sub>2</sub>," <u>J. Chem. Phys.</u>, vol. 65, pp. 4239-4245, 1976.
- 53. H. L. Chen and C. B. Moore, "Vibration → vibration energy transfer in hydrogen chloride mixtures," J. Chem. Phys., vol. 54, pp. 4080-4084, 1971.
- 54. H. L. Chen, "Vibration-to-vibration energy transfer in hydrogen bromide mixtures," <u>J. Chem. Phys.</u>, vol. 55, pp. 5557-5560, 1971.
- 55. J. L. Ahl and T. A. Cool, "Vibrational relaxation in the HF-HCl, HF-HBr, HF-HI, and HF-DF systems," J. Chem. Phys., vol. 58, pp. 5540-5548, 1973.
- 56. C. B. Moore, R. E. Wood, B. L. Hu, and J. T. Yardley,
  "Vibrational energy transfer in CO<sub>2</sub> laser," <u>J. Chem.</u>
  <u>Phys.</u>, vol. 46, pp. 4222-4231, 1967.
- 57. H. L. Chen, J. C. Stephenson, and C. B. Moore, "Laser-excited vibrational fluorescence of HCl and HCl-CO<sub>2</sub> laser," <u>Chem. Phys. Lett.</u>, vol. 2, pp. 593-596, 1968.
- 58. J. C. Stephenson, J. Finzi, and C. B. Moore,

- "Vibration + vibration energy transfer in CO<sub>2</sub>-hydrogen halide mixtures," <u>J. Chem. Phys.</u>, vol. 56, pp. 5214-5221, 1972.
- 59. See a review paper; T. A. Cool, "The transfer chemical laser: a review of recent research," <a href="#IEEE J. Quantum Electron.">IEEE J. Quantum Electron.</a>, vol. JQE-1, pp. 72-83, 1973, and references cited therein.
- 60. L. M. Peterson, G. H. Lindquist, and C. B. Arnold, "Rotation relaxation measurements of laser-excited hydrogen fluoride," <u>J. Chem. Phys.</u>, vol. 61, pp. 3480-3482, 1974.
- 61. J. J. Hinchen and R. H. Hobbs, "Rotational relaxation studies of HF using i.r. double resonance," <u>J. Chem. Phys.</u>, vol. 65, pp. 2732-2739, 1976.
- 62. P. R. Bevington, <u>Data reduction and error analysis for</u>
  the physical sciences. New york: McGraw-Hill, 1969.
- 63. S. W. Provencher, "An eigenfunction expansion method for the analysis of exponential decay curves," J. Chem. Phys., vol. 64, pp. 2772-2777, 1976.
- 64. J. A. McGarvey, Jr., N. E. Friedman, and T. A. Cool, "Vibrational energy transfer in HF-HCN, DF-HCN, and H<sub>2</sub>-HCN mixtures," J. Chem. Phys., vol. 66, pp. 3189-3196, 1977.
- 65. A. Hariri, Ph. D. Thesis, University of Southern California, 1977.

- 66. D. R. Siebert and G. W. Flynn, "Mode to mode energy transfer in OCS directly pumped by a CO<sub>2</sub> laser," J. Chem. Phys., vol. 64, pp. 4973-4983, 1976.
- 67. F. J. Zeleznik and R. A. Svehla, "Rotational relaxation in polar gases. II," <u>J. Chem. Phys.</u>, vol. 53, pp. 632-646, 1970.
- 68. G. E. Hyde and D. F. Hornig, "The measurement of bond moments and derivatives in HCN and DCN from infrared intensities," <u>J. Chem. Phys.</u>, vol. 20, pp. 647-652, 1952.
- 69. R. J. Lovell and W. F. Herget, "Lorentz parameters and vibration-rotation interaction constants for the fundamental band of HF," J. Opt. Soc. Am., vol. 52, pp. 1374-1376, 1962.
- 70. H. D. Mettee, "Vapor-phase dissociation energy of (HCN)<sub>2</sub>," <u>J. Phys. Chem.</u>, vol. 77, pp 1762-1764, 1973.
- 71. R. K. Thomas, "Hydrogen bonding in the gas phase: the infrared spectra of complexes of hydrogen fluoride with hydrogen cyanide and methyl cyanide,"

  Proc. Roy. Soc. London, vol. A325, pp. 133-149, 1971.
- 72. H. K. Shin, "Deexcitation of molecular vibration on collision: vibration-to-rotation energy transfer in hydrogen halides," J. Phys. Chem., vol. 75, pp. 1079-1090, 1971; "Vibration-to-rotation energy transfer in hydrogen fluoride: Effects of the dipole-dipole and

- hydrogen-bond interactions, " <u>J. Chem. Phys.</u>, vol. 59, pp. 879-884, 1973.
- 73. A. Messiah, Quantum Mechanics. Amsterdam: North-Holland Publishing Co., 1958, Chapters XVII and XIX.
- 74. C. Zener, "Interchange of translational, rotational and vibrational energy in molecular collisions," <a href="Phys.organics.">Phys.</a>
  Rev., vol. 37, pp. 556-569, 1931.
- 75. J. M. Jackson and N. F. Mott, "Energy exchange between inert gas atoms and a solid surface," Proc.
  Roy. Soc. London, vol. Al37, pp. 703-717, 1932.
- 76. N. F. Mott and H. S. W. Massey, <u>Theory of atomic collision</u>. New York: Oxford University Press, 1965.
  3rd ed.
- 77. K. Takayanagi, "On the inelastic collision between molecules, I," <a href="Prog. Theoret. Phys.">Prog. Theoret.</a> <a href="Phys.">Phys.</a>, vol. 8, pp. 111-119, 1952.
- 78. R. N. Schwartz, Z. I. Slawsky, and K. F. Herzfeld, "Calculation of vibrational relaxation times in gases," J. Chem. Phys., vol. 20, pp. 1591-1599, 1952.
- 79. R. N. Schwartz and K. F. Herzfeld, "Vibrational relaxation times in gases (three-dimensional treatment)," J. Chem. Phys., vol. 22, pp. 767-773, 1954.
- 80. L. Landau and E. Teller, "Zur Theorie der Schalldispersion," Phys. Z. Sowietunion, vol. 10, pp.

34-43, 1936.

- 81. See 7.5 (a) of ref. 3.
- 82. K. Takayanagi, "On the Inelastic collision between molecules, II. Rotational transition of H<sub>2</sub>-molecule in collision with another H<sub>2</sub>-molecule," <u>Prog. Theoret.</u>
  Phys., vol. 8, pp. 497-508, 1952.
- 83. D. Rapp and T. E. Sharp, "Vibrational energy transfer in molecular collisions involving large transition probabilities," J. Chem. Phys., vol. 38, pp. 2641-2648, 1963.
- 84. D. Rapp and P. Englander-Golden, "Resonant and near-resonant vibrational-vibrational energy transfer between molecules in collisions," J. Chem. Phys., vol. 40, pp. 573-575, 1964.
- 85. R. Marriott, "Molecular collision cross sections and vibrational relaxation in carbon monoxide," Proc. Phys. Soc., vol. A83, pp. 159-169, 1964; "Molecular collision cross sections and vibrational relaxation in carbon dioxide," Proc. Phys. Soc., vol. A84, pp. 877-888, 1964.
- 86. D. Secrest and R. B. Johnson, "Exact quantum-mechanical calculation of a collinear collision of a particle with a harmonic oscillator," J. Chem. Phys., vol. 45, pp. 4556-4570, 1966.
- 87. D. Rapp and T. Kassal, "The theory of vibrational

- energy transfer between simple molecules in nonreactive collisions," <a href="Chem. Rev.">Chem. Rev.</a>, vol.69, pp. 61-102, 1969.
- 88. B. H. Mahan, "Resonant transfer of vibrational energy in molecular collisions," J. Chem. Phys., vol. 46, pp. 98-101, 1967.
- 89. R. D. Sharma and C. A. Brau, (a) "Near-resonant vibrational energy transfer in N<sub>2</sub>-CO<sub>2</sub> mixtures," Phys. Rev. Lett., vol. 19, pp. 1273-1275, 1967; (b) "Energy transfer in near-resonant molecular collisions due to long-range forces with application to transfer of vibrational energy from v<sub>3</sub> mode of CO<sub>2</sub> to N<sub>2</sub>," J. Chem. Phys., vol. 50, pp. 924-930, 1969.
- 90. B. C. Carlson and G. S. Rushbrooke, "On the expansion of a Coulomb potential in spherical harmonics," Proc. Cambridge Phil. Soc., vol. 46, pp. 626-633, 1950; C. G. Gray, "On the theory of multipole interactions," Can. J. Phys., vol. 46, pp. 135-139, 1968.
- 91. P. W. Anderson, "Pressure broadening in the microwave and infra-red regions," Phys. Rev., vol. 76, pp. 647-661, 1949.
- 92. R. D. Sharma and R. R. Hart, "Comparison of classical, distorted wave, and straight path methods of evaluating the integrals over the intermolecular trajectory during molecular collisions," J. Chem.

- Phys., vol. 63, pp. 5383-5389, 1975.
- 93. C. G. Gray and J. van Kranendonk, "Calculation of the pressure broadening of rotational Raman lines due to multipolar and dispersion interaction," <u>Can. J. Phys.</u>, vol. 44, pp. 2411-2430, 1966.
- 94. W. G. Tam, "Vibrational energy transfer probabilities due to Coulomb interaction," Can. J. Phys., vol. 50, pp. 2691-2697, 1972; "Long range force vibrational energy transfer," Chem. Phys. Lett., vol. 15, pp. 113-115, 1972.
- 95. M. Abramowitz and I. A. Stegun, <u>Handbook of</u>

  mathematical <u>functions</u>. New York: Dover Publications,
  Inc., 1965, pp. 1001-1003.
- 96. J.=Finzi, J. H. S. Wang, and F. N. Mastrup, "Transition moments and integrated intensities of HCN(v<sub>1</sub>+v<sub>3</sub>) and DCN(v<sub>1</sub>+v<sub>3</sub>) combination bands," preprint.
- 97. M. M. Shapiro and H. P. Gush, "The collision-induced fundamental and first overtone bands of oxygen and nitrogen," Can. J. Phys., vol. 44, pp. 949-963, 1966.
- 98. T. A. Dillon and J. C. Stephenson, "Multiquantum vibrational-energy exchange," Phys. Rev., vol. A6, pp. 1460-1468, 1972; "Calculation of vibrational and rotational energy transfer between HF, DF, HCl, and CO<sub>2</sub>," J. Chem. Phys., vol. 58, pp. 2056-2064, 1973.

99. J. M. Blatt and V. F. Weisskopf, <u>Theoretical nuclear</u>
physics. New York: John Wiley & Sons, 1952, Chapter
XII and Appendix B.

## APPENDIX: TRANSITION MOMENTS

Although the relation between the integrated absorption intensity and the transition rate is known for a long time, it has produced confusion among the workers in this laboratory. The relationship is derived below in a simple way and used to calculate the radiative lifetimes and the transition moments of HF and HCN.

Consider a train of light beam with spectral intensity  $I_{\nu}$  and spatial width  $\Delta\ell$ . When it propagates a distance  $\delta x$  through an absorbing medium, the intensity is attenuated by  $\delta I_{\nu}$  according to Beer's law

$$-\delta I_{v} = N\alpha_{v} I_{v} \delta x, \qquad (A1)$$

where N and  $\alpha_{\nu}$  are the density and the absorption coefficient of the medium at frequency  $\nu$ . The energy absorbed by the medium is determined by the overall transition coefficient of an absorption band. When the absorption band has a narrow spectral width, one has

$$(h_{V}) (B_{\rho_{V}}) (\Delta L/c) (N\delta x) = -\int \delta \rho_{V} \Delta L dv$$
 (A2)

where .

$$\rho_{v} = I_{v}/c.$$

By combining (Al) and (A2) one gets

$$B = \frac{c}{h\nu} \frac{1}{N} \int \alpha_{\nu} d\nu.$$

The spontaneous emission rate A can be obtained using the Einstein relation

$$A = \frac{8\pi h v^3}{c^3} B.$$

Finally one has

$$A = \frac{8\pi v^2}{c^2} \frac{1}{N} \int \alpha_{\nu} d\nu. \tag{A3}$$

The transition rates due to the dipole and the quadrupole moments can be obtained from Blatt and Weisskopf [99]. For the dipole transition

$$A = \frac{64\pi^4 v^3}{3hc^3} |\dot{\mu}|^2.$$
 (A4)

For the quadrupole transition

$$A = \frac{32\pi^6 v^5}{45hc^5} \operatorname{Tr}(Q^* \cdot Q), \tag{A5}$$

where

$$Q_{ij} = \int (3x_ix_j - r^2\delta_{ij})\rho(r)d^3x.$$

The quadrupole moment  $Q_2$  defined by (5.2.2) for an axially symmetric charge distribution is related to  $Q_{ij}$  by

$$Q_2^2 = \frac{1}{-\text{Tr}}(Q^* \cdot Q).$$

The radiative lifetimes and the dipole transition moments of HF and HCN are calculated using the integrated absorption-intensity measurements [68,69,96]. The 2-fold degeneracy of the bending mode of HCN is taken into account. The dipole transition moments are listed in Table 3.

The radiative lifetimes are

$$HCN(00^{\circ}1)$$
 13.4 msec.

If the  $(v_2+v_3)$  transition of HCN is caused by the quadrupole moment, then one gets

$$Q_2^2 = 1.07 \times 10^{-48}$$

which is obviously too big compared to the one obtained from the V-V rate, 1.61X10<sup>-53</sup>.