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Final Technical Report

by

C.R. Wimberly

U.S. Army Research Office Grant Number DAA G29-77-G-0050

The University of Mississippi ;



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#### Forward

This document is the final report of the basic research project (P-14673-R-E) entitled "An Investigation of High Pressure Gaseous Injections into Liquid Propellants" granted by the U.S. Army Research Office (Grant number DAA G29-77-G-0050) for a 12-month post LRCP study from 1 January 1977 to 31 December, 1977. A 90 day nocost extension was obtained in December to allow completion of laboratory experiments that were completed on or about 31 March 1978.

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# Table of Contents

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1.	Introduction	1
11.	Analytical Effort	3
111.	Experimental Program	16
IV.	Conclusions	34
	References	35
	Appendix - Program Listing	37

Page

# List of Illustrations

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Figure		Page
1	Coordinates and Sketch of Gas Bubble Liquid Chamber Relation	7
2	Penetration Depth vs. time, $\gamma = 1.4$ , $T_0 = 294^{\circ}K$	8
3	Penetration Depth vs. Time, $\gamma = 1.4$ , $T_0 = 1000^{\circ}$ K	9
4	Penetration Depth vs. Time, $\gamma = 1.2$ , T <sub>0</sub> = 1000°K	10
5	Penetration Depth vs. Time, $\gamma = 1.2$ , T <sub>o</sub> = 3000°K	11
6	Penetration Rate vs. Time, $\gamma = 1.2$ , T <sub>o</sub> = 1000°K	12
7	Penetration Rate vs. Time, $\gamma = 1.2$ , $T_0 = 3000^{\circ}K$	13
8	Penetration Rate vs. Time, $\gamma = 1.4$ , T <sub>0</sub> = 294°K	14
9	Penetration Rate vs. Time, $\gamma = 1.4$ , T <sub>0</sub> = 1000°K	15
10	Schematic of Test Components and Arrangement	19
11	Photograph of Test Set-Up	20
12	Primary Injection Cylinder Arrangement	21
13	Photograph of Test Stand	22
14	Pressure-Time Curve from oscilloscope	23
15	Photograph of Penetration Sequence Test Case #1	25
16	Photograph of Penetration Sequence Test Case #2	27
17	Photograph of Penetration Sequence	29

Figure		Page
18	Photograph of Penetration Sequence Test Case #4	31
19	Penetration Depth for Cases as a Function of Time	32
20	Penetration Rate for Cases as a Function of Time	33

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#### Introduction

It has been recognized for many years that liquid propellants (LP) offer several advantages over solid propellants when applied to guns (References 1 through 8). Advantages which have become generally accepted include elimination of the cartridge case, high loading densities, reduced storage requirements, vulnerability reduction and increased flexibility in vehicle design due to weight and volume redistribution.<sup>9</sup> Because of these advantages much research and hardware development has been done at the BRL and at other places, including extensive test programs, which have in turn uncovered several problem areas. One of the concerns over the years has been the repeatibility of the pressure-time curves in the chamber, and occasional high pressure values that could exceed design limits of the casing.<sup>10</sup> It is apparent that several variables may be involved and that the liquid propellant is extremely sensitive to small changes in conditions at ignition and in the LP chamber. It was the attempt of this work reported here to provide more information on the ignition phase of the LP gun system operation.

The pyrotechnic igniter, which exhausts hot, high pressure gases into the LP is considered the prime candidate for causing variations in propellant decomposition, since it has not been greatly controlled in the past. The igniter itself is a solid capsule upstream of an orifice that decomposes very rapidly, sending the products of combustion through the orifice into the liquid propellant. These products are known and the conditions at injection are somewhat controlled. The phase of ignition that is very crucial and one that needs further investigation is the moment the gases exit the orifice, the subsequent

penetration until propellant ingition. From tests, this important phase of ignition lasts about 500 microseconds. But, during this time, the gas bubble has penetrated the liquid by several centimeters, and the shape and surface area have been developed which dictates decomposition rate. High pressures occur in the chamber during this period and it has been observed that the gases shear the liquid into many droplets. It is important to understand the conditions of the bubble at the instant of ignition of the LP because of its affect on the decomposition rate and therefore pressure variations with time in the chamber.

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Experimentally, an apparatus was used to provide high igniter chamber pressures and several orifice designs to investigate the effects on control of the injection. A high speed camera (on loan from the BRL) was used to provide a visual determination of the gas bubble formation, penetration depth and rate, gas-liquid entrainment and shock wave propagation. Tests were run to determine the sensitivity and effects of geometry (orifice size and shape), upstream initiation pressure, and the repeatability of tests.

Analytically, the gas bubble formation was mathematically modeled to simulate the sequence of events as the high pressure gaseous products penetrates the liquid propellant. The model includes the gas bubble formation, gas-liquid interactions, penetration depth and rate, plus the effects of pressure waves.

#### Analytical Effort

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The mathematical model developed under the grant consists of two separate models, 1) a simple model based on the equations of state of the gas and liquid, which will be designated Model I, and 2) a more sophisticated model based on linear continuity and the momentum equations (Model II). Model I utilizes estimates of the gas volume as a function of time, while Model II predicts the bubble shape, pressure, velocity distribution, and densities as functions of time and distance downstream in the chamber. References 12 through 16 were helpful in these developments.

The basic volume equation for Model I is

 $V_0 = V_g + V_L$ 

where  $V_g$  is the volume of the gas and  $V_L$  is the volume of the liquid. The volume of the gas is calculated using the real gas equation of state and the compressibility factor, Z, because of the high pressures involved. The volume of the liquid is expressed in terms of the bulk modulus, which is assumed to vary linearly with the pressure. Assuming choked flow and a constant mass flow into the chamber, the mass of the gas is known as a function of time. Equation (1) is then solved numerically for the chamber pressure at any time, t.

If the effect of the pressure wave is neglected,  $V_0$  is the total volume behind the wave front as it moves downstream and in front of the wave as it reflects from the far end of the chamber.

In Model I the shape of the gas bubble is assumed to be a prolate spheroid with various ratios of major diameters (see Figure 1). The proper ratio will be estimated from experimental results.

In Model II, the variable x is the distance downstream in the chamber and t is the time. It is desired to determine r, the radius of the bubble, as a function of x and t. Other quantities currently determined include velocities, pressure, and densities. The velocity is assumed parabolic over the cross section. This results in four velocity parameters (two for the gas and two for the liquid) which vary with x and t. The pressure and densities of the liquid and gas are also assumed to vary with x and t alone. The boundary of the liquid and gas is currently assumed perfect with no mixing, vapor pressure, or gas bubbles. The temperature is assumed constant throughout the chamber and for the length of the time considered.

For the eight dependent variables mentioned above, eight equations are needed. These consist of the state, continuity, and momentum equations for both the gas and the liquid, plus two boundary conditions at the gas-liquid interface.

Taking a volume element of width dx, the unsteady continuity equation can be derived which results in the following equation (see Figure 1).

 $\rho \frac{\partial A}{\partial t} + A \frac{\partial \rho}{\partial t} + \rho U_b \frac{A}{x} + (\rho \frac{\partial U}{\partial x} + U \frac{\partial \rho}{\partial x}) dA = 0$ 

Here,  $\rho$  is density, A is cross sectional area, U is velocity and U<sub>b</sub> is velocity at the boundary. Considering a momentum analysis of the same element under unsteady flow conditions produces

$$-A \frac{\partial p}{\partial x} - 2\pi r \mu \left[\frac{\partial u}{\partial \xi}\right]_{\xi=r} = \rho U_{b} \left(\frac{\partial A}{\partial t} + U_{b} \frac{\partial A}{\partial x}\right)$$
$$+ \left(\rho \frac{\partial U}{\partial t} + 2\rho U \frac{\partial U}{\partial x} + U \frac{\partial \rho}{\partial t} + U^{2} \frac{\partial \rho}{\partial x}\right) dA$$

Here,  $\rho$  is pressure  $\mu$  is viscosity, and  $\xi$  is the radial coordinate.

Substituting the assumed parabolic velocity distribution into equations (2) and (3) for both the gas and the liquid gives four partial differential equations. Two additional equations are obtained from equations of state.

For the gas

 $p = \rho RTZ$ 

and for the liquid

$$3 = \frac{-dp}{dv/v}$$

where the bulk modulus,  $\beta$ , is assumed to vary linearly with pressure. Two additonal equations are obtained by making the velocity and shear stress continuous at the gas-liquid boundary. Suitable initial conditions and boundary conditions at the entrance to the chamber complete the formulation of the problem.

The system of eight equations described above are highly nonlinear, first-order partial differential equations. To solve these, the method of lines was employed.<sup>17</sup> First, finite differences with respect to t were taken, producing a set of eight ordinary equations in terms of the independent bariable x. There equations were then solved by a standard Runge-Kutta technique. The resulting penetration depth and rate are shown in Figures 2 through 9.

The model currently assumes negligible gas-liquid entrainment, and temperature variations. It also does not provide for radial pressure waves or the effects of solid particles that may be injected into the LP. The model exhibits a particular problem of x=0. The solution is sensitive to the gas velocity of the entrance and an idealized model gives a singularity of the gas-liquid interface at x=0. In Model II, the shape of the bubble remains almost cylindrical in shape as x and t vary. In order to more accurately predict the shape of the bubble, it is believed that the momentum equation in the radial direction should be considered in addition to the equations already used. This involves introducing a radial velocity component. In order to maintain x and t as the only independent variables, the momentum equation in the radial direction will have to be integrated with respect to the radial coordinate,  $\xi$ . This could be accomplished by assuming a distribution of the radial velocity component over the cross section, leaving certain parameters as functions of x and t. At the gas-liquid interface, the radial velocity component will be equal to the time derivative of r.

Inclusion of the pressure wave was obtained into the model via the momentum equations, which were integrated in the x-direction in the usual Runge-Kutta fashion. If the wave front was encountered, the pressure change across the front was taken into account. This phenomenon was primarily encountered between 125 and 250 micro-seconds when the wave front was again traveling downstream.

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Penetration Depth vs. Time,  $\gamma = 1.4$ ,  $T_0 = 294^{\circ}K$ 

Figure 2.



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Figure 5.







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Figure 8. Penetration Rate vs. Time,  $\gamma = 1.4$ ,  $T_0 = 294^{\circ}K$ 

Figure 9. Penetration Rate vs. Time,  $\gamma = 1.4$ , T<sub>0</sub> = 1000°K



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### Experimental Effort

The experimental effort included the design, construction and testing of an apparatus for the simulation of the ignition phase. Figure 12 illustrates the primary injection cylinder. A mechanically sheared diaphram allowed control of the start of the gaseous injection into the liquid chamber for photographic purposes. Several materials were tested for use in the diaphram for repeatability. Aluminum disks were finally selected for the greatest repeatability. As the diaphram is sheared, the pieces and plug will necessarily have to have a cavity to fall into, thus increasing the volume above the plug, and reducing the pressure. By using different pin lengths, the pressure prior to diaphram rupture was controlled up to about 28000 psia, which was sufficient for a controlled design injection pressure of 200000 psia. Ballistics type piezoelectric pressure taps were used to monitor the pressure in the cylinder at all times during the injection sequence.

During a test, a high speed camera was used to observe bubble shape, injection distance and rate, and gas-liquid entrainment. Of primary interest was the sensitivity of geometry, gas type, injection velocities and upstream pressures on repeatability of the cavity formation. A square cross sectional liquid chamber was used to simplify the photography, although reflecting waves off this shape gave results somewhat different than what would occur for the cylindrical shape of a gun chamber. Future work could include piezocrystal monitoring of the square and round shapes to determine if there is any significant variations in results. If this is indicated, then a round section with appropriate optics could be developed for photographic purposes.

The experimental apparatus (shown in Figure 10) included a gas regulator, air filter, oil lubricator, solenoid switch, piston, 1/2 inch high pressure cylinder, water tank and gas supply system. The piston was made of cold roll stainless steel and was provided with two teflon o-rings to reduce gas leaks. The diaphrams (.048 and .078 inch) were designed to rupture at prescribed pressures, although in operation the rupture was controlled by a plunger at the end of the piston. The orifices were cut from brass (1mm & 2mm). Grease was used to block the orifice prior to each run to eliminate water seepage into the ullage upstream of the orifice. The water tank was of rectangular shape 8 x 1.25 x 1.25 inches and designed to withstand 2000 psia. Two piezoelectric crystals were used to measure pressure in the tank and one was placed in and fed into a four channel vertical amplifier oscilloscope containing a time base and delay generator. A polaroid camera was attached to the oscilloscope for impressions of pressure variations. A high speed camera was used to capture the injection sequence and had the capability of 11,000 frames per second, f of 1.9, zoom lense and usable distances of 4 to 100 feet. f of 4.0 was used for all tests and provided good resolution. The light source included two 1000 watts tungsten hallogen spotlights and three 500 watts super flood #2 photoflood bulbs. These light sources were kest about 1 foot from the tank. In order to get maximum light, one spot light and one super flood light were kept in between camera and tank. One spot light and two superflood lights were kept behind the water tank. Precaution was taken to prevent excessive light entering into camera, by shielding it with a black sheet. All of the lights were positioned at 45° angle with

the tank.

For pressure measurements peizo-electric crystals in the high pressure cylinder were calibrated at 500 psia per unit. Crystals in the tank (upstream) were calibrated at 100 psia per unit. A single sweep mode was used per crystal. On the time scale, one unit was one second. The total length of travel of the beam was about 7 units, hence, from the instance the beam was triggered, it took about 7 seconds for the beam to go from one end to another. The solenoid switch and camera motor was thus started within these seven seconds.

Figure 14 shows a typical pressure-time trace from the oscilloscope; this one from test case #1. Trace (1) represents the pressure in the cylinder while trace (2) represents that of the liquid chamber. For this case, the cylinder pressure increases rapidly from its initial pressure of 250 psia to 1450 psia just prior to diaphragm rupture. A time lag is shown from the diaphragm rupture time to the increase in chamber pressure.

A frame by frame analysis provided the information for calculating the depth and rate of penetration for the cases run and are plotted in Figures 19 and 20.











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Test		#1	
<sup>P</sup> cylinder	initial	=250	psia
P <sub>cylinder</sub>	final	=1,350	psia
Ptank	initial	=14.5	psia
Ptank	final	=180	psia
Diameter		≐lmm	
PPS		=2000	
F		=1.73	

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Frame No:	Time msec	Distance (Pictorial)	Distance (Actual) inches	Velocity Ft/sec
1	0.5	6/32	0.324	54.06
2 ′	1.0	12/32	0.649	54.06
3	1.5	16/32	0.865	37.5
4	2.0	20/32	1.08	34.45
5	2.5	23/32	1.24	27.07
6	3.0	26/32	1.40	27.07

Time =	0.5	msec
Disp.=	0.324	inch
vel. =	54.06	ft/sec
Time =	1.0	msec
Disp.=	0.649	inch
Vel. =	54.06	ft/sec
Time =	1.5	msec
Disp.=	0.865	inch
Vel. =	37.5	ft/sec
Time =	2.0	msec
Disp.=	1.08	inch
Vel. =	34.45	ft/sec
Time =	2.5	msec
Disp.=	1.24	inch
Vel. =	27.07	ft/sec
Time =	3.0	msec
Dist.=	1.40	inch
Vel. =	27.07	ft/sec

# Figure 15. Photograph of Penetration Sequence Test Case #1

lest		#2
<sup>P</sup> cylinder	initial	=600 psia
<sup>P</sup> cylinder	final	=2,600psia
Ptank	initial	=14.5 psia
Ptank	final	=280 psia
Diameter		= 1 mm
PPS		=2,000
F		=1.875

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Frame No:	Time msec	Distance (Pictorial)	Distance (Actual) inches	Velocity Ft/sec
1	0.5	33/63	.967	161.17
2	1.0	56/64	1.641	112.33
3	1.5	66/64	1.934	48.33
4	2.0	72/64	2.168	39.23
5	2.5	80/64	2.402	39.23
6	3.0	88/64	2.578	_29.22



Test		#3
<sup>P</sup> cylinder	initial	=500 psia
<sup>P</sup> cylinder	final	=2,500psia
Ptank	initial	=14.5 psia
Ptank	final	=260 psia
Diameter		=2mm
F		=1.915

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Frame No:	Time	Distance	Distance	Velocity
	msec	(Pictorial)	(Actual) inches	Ft/sec
1	0.67	20/32	1.197	149.63
2	1.33	33/32	1.975	97.25
3	2.00	41/32	2.45	59.75
4	2.67	47/32	2.81	44.94
5	3.33	52/32	3.11	37.31
6	4.00	56/32	3.35	30.00



		29	
	Time = Dist.= Vel. =	0.67 1.197 149.63	msec inch ft/sec
With Aller			
	Time = Dist.= Vel. =	1.33 1 1.975 2 97.25 2	msec inch ft/sec
	Time =	2.00	msec
	Dist.=	2.45	inch
	Vel. =	59.73	ft/sec
	Time -	7 77	
	Dist.=	3.33	msec
	Vel. =	37.31	ft/sec
	Time = Dist.= Vel. =	2.67 m 2.81 i 44.94 f	nsec inch ft/sec
	Time =	4.0	msec
	Vel. =	3.35 30.00	inch

Figure 17. Photograph of Penatration Sequence Test Case #3

Test		#4	
<sup>P</sup> cylinder	initial	=1,000	psia
P <sub>cylinder</sub>	final	=5,000	psia
Ptank	initial	= 14.5	psia
Ptank	final	=500	psia
Diameter		=2mm	
PPS		=3,000	
F		=1.957	

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Frame No:	Time msec	Distance (Pictorial)	Distance (Actual) inches	Velocity Ft/sec
1	0.33	46/64	1.40	350.0
2	0.67	76/64	2.32	230.8
3	1.00	91/64	2.78	114.8
4	1.33	104/64	3.18	99.3





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Figure 19. Penetration Depth for Cases as a Function of Time



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#### Conclusions

Simulation of the penetration sequence for gaseous injections into liquid propellants provided greater understanding of the significants of geometry and flow conditions on gas bubble formation, penetration depth and rate. Comparisons of test data and the simulation model indicated that the model is good to excellent in predicting the penetration depth and rate of the injection for the first 500 microseconds. The shape of the bubble obtained experimentally approached closely to the prolate spheroid assumed for model I. The excellent agreement between the test data and simulation program also provided support for the assumptions made for the program.

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L		THIS PAGE IS BEST QUALITY PRACTICABLE	
c		PROGRAM FUR ANALYSIS OF LIGUID PROPELLANT GUNS	38
c c	1	CUMMON V0, A, B, GFC, A1, A2, A3, VL, VG, GAM, TO, TEMP, KEY, DP, AFEA, CL, KEY2, FO DIMENSIUN RATIO(10), AXIS(10,200), AXMIN(10,200) DIMENSIUN VELG(200), VELL(200), PRESS(200), TIME(200), STEMP(200)	
C		INFUT GEOLETRY PALAMETERS	
	1	READ (2,1) DO,CD,CL FURMAT(4E)	
le l		INPUT GAS PARAMETINS	
Ic		READ (2,1) CAR, R, 10, 20, A1, A2, A3, TO	
1 C		INPUT LIQUIN LARAPETERS	
L		READ(2.1)A.F.C	
Ċ.		INPUT TIME LANAMETERS	
		READ (2,1) IT, TF	
Icc		IFFET PECLATE SPECIEL FAFAMETERS RATIO = MIMUR DIALETER/FAJOR DIAMETER N = NUMERN OF RATIOS TO BE CONSIDERED	
I	2	READ (2,2) P FORMAT(1)	
- C C		DATA CHECH	
I	4	WFITE(3,4)D0,CD,CL FURNAT('1',3E16.8) WFITE(3,5)GAM,R,P0,Z0,A1,A2,A3,T0	
•	5	WEITE(3,6)A,B,C,DT,TF,N	
lc	r	CONVERT INFUT DATA	
ľ		D0=D0/(25.4*12.) R=P#1545.33/8314.5 PD=D0/6.8943F3	
I		T0=T0+9./5. A0=3.14159#10+D0/4. DL=(2./(GAM+1.))**((GAM+1.)/(GAM-1.))	
I		DH=SOFT(GAM+32.174+DM/P) DL=DM+A0+P0+144./(ZO+SGET(TO)) DT=LT+1.E-6 TE=EF1.01.01	
1		CD = CD / (25, 4*12) $C = C / (25, 4*12)$ $C = CL / (25, 4*12)$	
I.		AFEA=3.14159*CD*CD/4.	
		INPUT SHAPE FACTORS	
Ľ		100 J=1,1	

	100	PFAD (2,1) (ATU(I) KIY = 1 KEY]=0	IS BEST QUALITY PRACTICABL.	39
		PHASE 1 - WAVE FOVING DOWESTREAM PHASE 2 - WAVE MOVING UPSTREAM - PHASE 3 - WAVE FOVING DOWESTPEAM PHASE 4 - WAVE FOVING UPSTREAM -	- KEY = 1 • KEY = 2 • - 2ND TIME - KEY = 3 • 2ND TIME - KEY = 4	· · · · · · · · · · · · · · · · · · ·
I		P1=0. NF=TF/DT+3. NT=(CL/C)/D1+1. NT1=DT		
1		T=0. PD=0. DU 101 J=1.1F IP2=1		
I	144	T=T+DT IF(T.J.E.(TF+0.5b=0))GOT0149 b6=d=1 G0 T0 117		
I	149	IF (J.EQ.NT)GOTO112 IF (J.KE.(NT+1))GOTO111 IF (KEY.NE.1)GOTO121 ATENT	· · · · · · · · · · · · · · · · · · ·	
I		T=AT+DT KEY=2 NT=(2.*(L/C)/DT+2. DD=P1		
Ι	121	V0=AREA*((L-C*(T-CL/C)) GU TO 114 IF(KEY,FE,2)GUTU122 AT=NT+N1F		
I		T=AT+DT KEY=3 KEY1=0 RT=(3.+(1/C=T)/DT+1.		
1		NT=NT+J NT3=NT V0=APEA+C+(T-2.+CL/C) GU TU 114		
1	122	T=TGAVE KLY=4 KFY1=0 V0=AREA*(CL=C*(T=3.*CL/C))		
1	112	GOTO114 IF(KEY.WE.1)GUTO123 T=CL/C V0=ARLA*C*T		
1	123	GU TU 114 IF (KEY.LE.2)GUTU124 T=2.*CL/C V0=AREA*CL		
1	124	KEY1=1 CU TU 1)4 TSAVE=T T=3.*CL/C		
1	111	KEY1=1 V0=AREA+CL GC TO 114 IF (KEY_KE_1)GCT0125		

		VO=AFIA*C*T	
_		GI TO 114 THIS PAGE IS BEST QUALITY PRACTICADE	40
	125	VO=ABLA+((L=C+(T=CL/C))	
		GOTG114	
	126	1F(KEX.NE.3)GOT0127	
•		V0=ARLA*C*(T=2.*CL/C)	
_	127	GC TO 114 VO=AFLA+(CL=C+(T=3_+CL/())	
1	114	GAC=DA+R+(T=0,5+5,0E=6)	
		IF(T.LT.5.0E-6)GMC=DM*R*0.5*(DM/5.0E-6)*T**2	
2		CLADCH FOR NDLDOVTHATE LOCATION OF LOOT	
lč		PEFEC = FUNCTION OF PRESSURE SET EQUAL TO ZERU	
Ċ			
T	152	F1=PRENC(P1)	
1		$1F(F1_{\bullet})_{\bullet} J_{\bullet} J_{\bullet} U_{\bullet} I_{\bullet} I$	
		IF(P1.GT.0.)GU TO 152	
T		WRITE (3,154) P1,F1	
1	154	FURMATC'0',5X, 'DEGATIVE PRESSURE'/6X, 'P1 =',E16.8,5X,	
-	•	NF=J	
1		CU TO 117	
- 29	151	P2=11	
77	150	P2=F2+100	
c		1.(1.5.0) 1.0000 ) (0.1014)	
C		IF FRESSURE EXCREDS 10,000 PS1 IN TIME INCREMENT	
C		DT IS HALVEL 4 TIMES	
C		11 2=11 2+1	
-		IF (IP2.1.T.5) GUTC 142	
1		WR1TE(3,143)DT	
-18	143	FURMATC'0', 10X, 'PRESSURE FXCEEDS 10,000 PSI IN TIME INCREMENT	=',
T	•		
1		GI TO 117	
	142		
T	141	+2=PRINC(P2)	
r		IF(F2.LT.0.)GO TO 150	
+ C	:	AND STARS INCOMPT - COMP TH PAULS NO. 14	
10		HALF FIFDS FFESSORE - SFIS IT EQUAL TO FI	
		CALL HALF(P1,F2,F1)	
TC			
TC		FIND FAJOR ANT FINOR AXES OF FROLATE SPHEROID	
		DU 102 1=1,1	
1		AX1S(I,J)=((3.+VG)/(4.+3.14159+EATIO(I)++2))++(1./3.)	
•		$A \times IS(I,J) = A \times IS(I,J) + 12 + 25 + 4$	
1		IF (KEY.PE.2) GOTO102	
1		11 (KEY1.FC.1) GOTO 102	
		11 (AXIS(1,J).GE.((CL-C*(T-CL/())*12.*25.4))KEY=9	
T	102	V01.G(4)=VC+1728.+25.4++3	
•		V(LL()=V1+1728.+25.4++3	
		PRESS(J)=1 1*6.8948E3	
1		TIME(C) = T	

	STEMP(J)=TELP*5./9.
	IF (KEY.NE. 9) GUTU101
	NT=J+1 THIS PAGE IS BEST QUALITY PRACTICABLE 41
	NT2=NT FROM COPY FURNISHED TO DDC
	NTF=(2.*CL/(-T)/DT+1.
	KLY=2
101	CONTINE
C	
2	OUTPUT FESULTS
C	
117	WRITE(3,7)
7	FORMAT(40%, 'LIQUID PROPELLANT GUN'//)
	WEITE (3,8)
8	FORMAT(44X, 'BUBDLE GPOWTH'///)
	WRITE (3,9)
9	FORMAT(8X, 'TIME', 9X, 'BREECH PRESSURE' , 3X, 'PROJECTILE PRESSURE',
r 1	3X, 'MAJOF AXIS', 5X, 'PEFSSURE FFONT', 6X, 'GAS VOLUME',
2	7X, 'TEMPEFATURE')
	WPITE(3,10)
r 10	FURNAT(3X, '(PICFO SECURES)', 6X, '(PASCALS)', 11X, '(PASCALS)',
1	11X, '(HE)', 13X, '(MM)', 12X, '(CU EM)', 11X, '(DEG-K)'//)
he .	DF=DF*6.8948E3
	T0=T0*5./9.
	DO 103 I=1,1
-	WEITE (3,11) HATIC(I)
11	FURHAT('0', 50%, 'PATIO =', F7.4)
	T=U.
	WRITE(3,12) T,T,T,T,T,T,T0
12	FURNAT(7(2X,E16.8))
*	DU 104 K=1,1G
	IF (K.CT.NTI)GOTO131
As .	PF=C*11/E(K)*12.*25.4
	PD=0.
1	GU TU 139
131	IF (K. GT. MT2) GOTU132
	Pf = (CL - C + (TIME(K) - CL/C)) + 12 + 25 + 4
T	PI = PFE SS(F) + DP
1	
132	
	$PF = ( + (1) ME (N) - 2_{0} + (1) (1 + 1) 2_{0} + 25_{0$
	PD=PPFSS(K)-DF
133	$P_1 = (1 - (+ (1 - 1) + 3 + (-1) + 12 + 25 + 4)$
1.20	
1.39	HIME () JETICI (NJET CO
	WE LET $V(r) = V(r) = $
r	$\frac{1}{1} \left( \frac{1}{1} + \frac{1}{1} + \frac{1}{1} + \frac{1}{1} + \frac{1}{1} \right)$
L	
12	FIDUATION, 20Y, IDEESSUDE MAVE DEMOURS DEDIFORTE EXCELLAS
. 14	ELEVATION, 20X, PRESSURE WAVE DEACHES DEPENDENTING
104	CONTINUE
103	CINTLUIF
105	67/10 <sup>2</sup>
	FLD
	ELECTION DRENC(D)
c	
Ic	CALCULATES I RESSURE FUNCTION
č	TOTAL VOLUME = LIGHTD VELOME + CAS VOLUME
c	
	CLAPDE VO.A.B. GLC.AL.A2.A3.VL.VC.CAM.TO.T.KEV.DU.
	CONTRACTOR DE CONT
•	

1	AFEA,(1,KEY2,F0 2=A1+A2*P+A3*P*P IF(2,LT.1,)2=1, PA=P+14.7 T=T0*(PA/P0)**((GAM-1.)/GAM) IF(KEY,EQ.1)GGT01 IF(KEY,EQ.1)GGT01 V1=AREA*CL-V0 IF(KEY.EC.3)DF=-DF VTMP=((A+B*(P+DF))/A)**(-1./ IF(KEY,EQ.3)DP=-DF	THIS PAGE IS BEST QUALITY PRA THIM COFY FURNISHED TO DDO	CTICA
1 2	VL1S=V1/VTMI VL0S=V0+V1-VL1S VTMP=((A+B*F)/A)**(-1./F) VL=VL0S*VTMP GC TO 2 VL=V0*((A+B*P)/A)**(-1./B) VC=GM(*Z*T/(PA*144.) PFFIC=VI.+VG PEFIC=V0-FRFHC RETURI END SUBEOFTINF LALF(X1,X2,X)		
5 1 3	CARCOLATES FOR TETT AAAF TATE         SEE P175 - SOUTHWERTH & DELE         F1=X         x=(X1+X2)/2.         FM=PRENC(X)         IF(F1*FF)1.2.3         X2=X         GP TO 4         F1=FM         X1=X	ELW	
4 2	IF((X2-X1).GT.1.0)GC TO 5 RETURN ENL		

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