

60 AD A 0 4 890 OFFICE OF NAVAL RESEARCH Contract N00014-78-0015 Task No. NR 356-522 Technical Kepert, #7 COPV PIEZO- AND PYROELECTRICITY OF POLY (VINYLIDENE FLUORIDE) FROM PLASMA POLING . by John E. McKinney 😅 G. Thomas Davis Prepared for Publication in ACS Polymer Preprints 1 Nec 77 National Bureau of Standards Bulk Properties Section Washington, D. C. 20234 330

December 1, 1977

Reproduction in whole or in part is permitted for any purpose of the United States Government.

Approved for Public Release; Distribution Unlimited.

JAN 23 1978 B

240 800

BEFORE COMPLETING FORM RECIPIENT'S CATALOG NUMBER TYPE OF REPORT & PEHIOD COVE echnical Report #7 PERFORMING ORG. REPORT NUMBER CONTRACT OR GRANT NUMBER(*) DO014-78-0015 PROGRAM ELEMENT. PROJECT. TA AREA & WORK UNIT NUMBERS NR 356-522 REPORT DATE December 1, 1977 NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified DECLASSIFICATION/DOWNGRADIN
Achnical Report #7 PERFORMING ORG. REPORT NUMBER CONTRACT OR GRANT NUMBER(*) DO014-78-0015 PROGRAM ELEMENT. PROJECT. T AREA & WORK UNIT NUMBERS NR 356-522 REPORT DATE December 1, 1977 NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified
Achnical Report #7 PERFORMING ORG. REPORT NUMBER CONTRACT OR GRANT NUMBER(*) DO014-78-0015 PROGRAM ELEMENT. PROJECT. T AREA & WORK UNIT NUMBERS NR 356-522 REPORT DATE December 1, 1977 NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified
PERFORMING ORG. REPORT NUMBER CONTRACT OR GRANT NUMBER(*) DO014-78-0015 F PROGRAM ELEMENT. PROJECT. TA AREA & WORK UNIT NUMBERS NR 356-522 REPORT DATE December 1, 1977 NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified
PERFORMING ORG. REPORT NUMBER CONTRACT OR GRANT NUMBER(*) DO014-78-0015 F PROGRAM ELEMENT. PROJECT. TA AREA & WORK UNIT NUMBERS NR 356-522 REPORT DATE December 1, 1977 NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified
CONTRACT OR GRANT NUMBER(*) DO014-78-0015 PROGRAM ELEMENT. PROJECT. TA AREA & WORK UNIT NUMBERS NR 356-522 REPORT DATE December 1, 1977 NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified
DOO14-78-0015 PROGRAM ELEMENT, PROJECT, TA AREA & WORK UNIT NUMBERS NR 356-522 REPORT DATE December 1, 1977 NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified
PROGRAM ELEMENT. PROJECT. TAREA & WORK UNIT NUMBERS NR 356-522 REPORT DATE December 1, 1977 NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified
NR 356-522 REPORT DATE December 1, 1977 NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified
NR 356-522 REPORT DATE December 1, 1977 NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified
NR 356-522 REPORT DATE December 1, 1977 NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified
REPORT DATE December 1, 1977 NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified
December 1, 1977 NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified
NUMBER OF PAGES 18 SECURITY CLASS. (of this report) Unclassified
18 SECURITY CLASS. (of this toport) Unclassified
SECURITY CLASS. (of this report) Unclassified
Unclassified
DECI / SCIEICATION/DOWNCOM
SCHEDULE
roved for public release; Distribution Unlimited
port)
Ŷ
ing:
ALDHA
HUP
05-01.
BETT .
BETA
s essentially instant-
s essentially instant- at very high fields.
s essentially instant- at very high fields. sing essentially pure
s essentially instant- at very high fields. sing essentially pure nformation) were polari
s essentially instant- at very high fields. sing essentially pure nformation) were polari determined from current
s essentially instant- at very high fields. sing essentially pure nformation) were polari
s essentially instant- at very high fields. sing essentially pure nformation) were polari determined from current pelectric coefficients
s esso at vo sing e nforma determ oelect

•.

.

1

.

•

•

•

1

LUNITY CLASSIFICATION OF THIS PASE(When Date Entered)

# 20.

theory of Mopsik and Broadhurst, which expresses these coefficients as linear functions of polarization. The theory gives an accurate representation of the piezoelectric coefficients, but underestimates the pyroelectric coefficients considerably.

In addition, x-ray measurements from various techniques were made on both drawn and undrawn PVDF samples. For the unpoled samples the results are in good agreement with those in the literature. Some changes in spacings and intensities were observed with noling for both the drawn and undrawn samples. As yet we do not have a clear interpretation of these results in terms of crystal structure changes resulting from polarization.

ACCESS		-
NTIS	· White Section	•
DDC	Buff Section	
UNANN	DUNCED	
JUSTIFI	CATION	
BY		
BY DISTR Dist.	BUTION/AVAILABILITY CODE AVAIL and/or SPEC	S

SECURITY CLASSIFICATION OF THIS PAGE(When Date Entered

Piezo- and Pyroelectricity of Poly(vinylidene fluoride) from Plasma Poling

J. E. McKinney and G. T. Davis

## Abstract

A plasma poling device is described which allows essentially instantaneous poling of polymeric films at room temperature at very high fields. Using poly(vinylidene fluoride) both undrawn (comprising essentially pure  $\alpha$  conformation) and biaxially drawn (both  $\alpha$  and  $\beta$  conformation) were polarized in this manner. The polarization of the sample was determined from current integration measurements. The piezoelectric and pyroelectric coefficients were then measured and compared with those estimated from the polarization theory of Mopsik and Broadhurst, which expresses these coefficients as linear functions of polarization. The theory gives an accurate representation of the piezoelectric coefficients, but underestimates the pyroelectric coefficients considerably.

In addition, x-ray measurements from various techniques were made on both drawn and undrawn PVDF samples. For the unpoled samples the results are in good agreement with those in the literature. Some changes in spacings and intensities were observed with poling for both the drawn and undrawn samples. As yet we do not have a clear interpretation of these results in terms of crystal structure changes resulting from polarization.

## PIEZO- AND PYROELECTRICITY OF POLY(VINYLIDENE FLUORIDE) FROM PLASMA POLING By

J. E. McKinney and G. T. Davis National Bureau of Standards Washington, D.C. 20234

## I. Introduction

t typing

his line

This paper is a preliminary report which includes a description of the plasma poling process, the consequences of this kind of poling in terms of piezo- and pyroelectric response, and an attempt to interpret the results in terms of molecular structure.

Conventional poling procedure involves heating the polymer sample to an elevated temperature, applying an electric field for a brief period of time, and then cooling to room temperature under the field. Recently, Southcate<sup>1</sup> has demonstrated that by applying corona discharge at high potential in the vicinity of a poly(vinylidene fluoride) (PVDF) sample, with the opposite side at ground potential, the poling process is essentially completed in several seconds at room temperature. There is also the advantage that higher fields may be obtained by corona or plasma poling. Since poling times are so short it is relatively easy to measure the total charge on the sample from poling, as demonstrated by Southgate. After the applied field is removed, the residual charge is a measure of the irreversible or "permanent" polarization from which it is possible to derive expressions for the piezoand pyroelectric responses.

The piezoelectric and pyroelectric response from polymers has been related to polarization, which is linear with applied field for amorphous polymers<sup>2</sup>. Since the charge has been measured during the poling process in our work, these data give us the opportunity to evaluate the theory in terms of experimental quantities for semicrystalline polymers as well. It will be shown that the model predicts the piezoelectric response within experimental accuracy; however, some refinement appears to be necessary for the pyroelectric response. On the other hand, some of the experimental values of quantities appearing in the theory are somewhat dubious, it is not appropriate to pass judgement at this time.

There has been considerable recent interest and some speculation on the nature of apparent phase transitions in PVDF resulting from poling at high fields. For example, such behavior is indicated by changes in infrared spectra as observed by Southgate<sup>1</sup>. We are presently conducting investigations along these lines employing various X-ray and IR-techniques.

## II. <u>Theoretical Relationships between Piezo- and Pyroelectric Coefficients</u> and Polarization

Using an Onsager type reaction field model, Mopsik and Broadhurst<sup>2</sup> have derived the relation for the polarization

$$P = (\epsilon_{m} + 2)(N \mu_{0}/3V) < \cos \theta >$$

(1)

where  $\epsilon_{\rm o}$  is the limiting high frequency dielectric constant of the medium, N is the number of dipole moments each of strength  $\mu_0$  in total volume V, and <cos 0> is the average cosine of the dipole orientation angle 0. The above relation gives the permanent polarization in a homogeneous, nonconducting, polarizable medium.

t type

With semicrystalline polymers some modification is necessary to take into account the fact that polarization exists only within the crystalline region and the amorphous regions can be slightly conductive. Although the conductivity of poly(vinylidene fluoride) is very small, a gradual decay of polarization results from migration of real charge through the amorphous regions to the crystal surfaces<sup>3</sup>.

In order to adapt the continuum model<sup>2</sup> to a semicrystalline medium we assume<sup>4</sup>

 $P = \chi \frac{3\varepsilon_s}{2\varepsilon_s + \varepsilon_c} P_c$  (2)

where  $\chi$  is the crystalline fraction based on volume,  $P_{C}$  is the total polarization of the crystalline region, and  $\varepsilon$  is the relative permittivity with subscripts s and c pertaining to sample and crystal respectively. The dipole moments in the amorphous region are assumed to be randomized at temperatures above  $T_{g}$  after removal of the external poling field. To account for the real charge redistribution mentioned above, we include an additional term in eq.(1) which becomes

$$P(t) = \chi \frac{3\varepsilon_s}{2\varepsilon_s + \varepsilon_c} \left[ \frac{N\mu_0}{3} \frac{\varepsilon_c + 2}{V_c} <\cos \theta > - \frac{Q'(t)}{A_c} \right]$$
(3)

A is the total area of the crystals, and 0' is a time dependent charge at the crystal surfaces which satisfies the boundary conditions

$$Q'(0) = 0$$
 and  $\lim_{t\to\infty} P(t) = 0$ 

From application of the experimental definitions

$$d_{p} = (1/A) [\Im(AP)/\partial P]_{T}$$
(4a)  

$$\kappa = -(1/A) [\Im(AP)/\partial T]_{D}$$
(4b)

where d is the hydrostatic piezoelectric coefficient,  $\kappa$  is the pyroelectric coefficient, A is the electrode area, and T and p are the temperature and pressure. The final expressions become

$$d_{p} = P_{0}\beta_{c}\left[\frac{2\varepsilon_{c}+1}{6} + \phi \frac{J_{1}(\phi)}{J_{0}(\phi)}v\right]$$
 (5a)

$$\kappa = P_0 \alpha_c \left[ \frac{2\varepsilon_c + 1}{6} + \phi \frac{J_1(\phi)}{J_0(\phi)} \left( v + \frac{1}{2T\alpha_c} \right) \right]$$
(5b)

\* where  $P_0$  is the initial value of P(t) after poling,  $\beta_c$  and  $\alpha_c$  are the crystal compressibility and thermal expansion,  $\phi$  is the dipole fluctuation angle, v is the Gruneisen constant, and  $J_0$  and  $J_1$  are Bessel functions of first kind of order 0 and 1, respectively.

Ing

this line

More explicit detail of the development of eqs.(5) will be included in a later publication. In brief, it is assumed that the c cell dimension (along the chain axis) change with temperature or pressure is negligible. Pressure and temperature dependence in  $\epsilon_c$  are taken from the Clausius Mossotti equation. Considerable simplification in the expressions for  $d_p$ and  $\kappa$  ensues from evaluating these quantities at the ultimate polarization [P(t) = 0] when crystal polarization is compensated by real charge at the crystal liquid interface.

#### III. Experimental

## A. Apparatus

The plasma chamber used to generate plasma for the poling process was adapted from a commercial plasma cleaner. The electrode assembly is shown schematically in Fig. 1. An aluminum electrode of 2.5 cm diameter is evaporated on one side of each polymer film sample. The sample is clamped as shown with the bare side exposed to the plasma. The chamber (containing air) is evacuated to and maintained at about 200 mT during the poling process. Water vapor and other condensable gases are removed with a liquid nitrogen trap.

Voltages up to 10 kV may be applied. The corresponding charge transfer is measured with a charge amplifier in series with the high voltage circuit. Electrical leakage was found to be negligible over the short poling times used. From the charge transfer the polarization of the sample was determined.

After poling, a graphite electrode was painted on the bare side of each sample. The electrodes were shorted and the samples stored for at least twelve hours for surface charges to dissipate. The piezoelectric coefficients were then measured using a pressure cell<sup>5</sup> imposing nearly constant rate of pressure change. From the slopes of the pressure-time curves and corresponding currents,  $d_p$  is determined. In an analogous manner the pyroelectric coefficient  $\kappa$  is determined from the measurement of current resulting from nearly constant rate of temperature change.

X-ray measurements were made with an automatic scanning diffractometer and a powder camera using Nickel-filtered-Cu Ka radiation in both cases. For the powder camera, the sample film was cut into thin strips with dimensions approximately  $3 \times 0.3 \times .025$  mm. These were inserted and sealed in a capillary of 0.3 mm diameter.

#### B. Samples

Two kinds of commercial PVDF semicrystalline samples were studied. The first is a biaxially stretched (blow molded) film for which the polymer chains tend to orient in the plane of stretch. The stretch process converts a portion of the non-polar  $\alpha$  (TGTG' conformation) phase to the polar B (planar zig-zag) phase. The second sample is undrawn and contains essentially only the  $\alpha$  phase. Both kinds of film are about 25 µm in thickness.

#### IV. Results and Discussion

Fig. 2 gives the initial polarization  $P_{0}$  (as measured from the charge remaining after the applied field is removed) plotted against the applied field, or applied voltage, for the biaxially drawn sample. About three seconds are required to reach 90% of the ultimate value of charge at room temperature. As indicated, the data are somewhat erratic. Subsequent hysteresis loops, however, reveal smooth responses with applied field.

typing is line

In all cases, except the one indicated by the triangle where the field was reversed, the plasma was at negative potential with respect to the electrode. As seen from the figure, there is essentially no polarization below one MV/cm. There is no indication of saturation in the polarization and dielectric breakdown prevents attaining fields much above 3.5 MV/cm. The numbers on the data points indicate correspondence to those on figures which follow.

According to the theory, for which the results are given in eqs.(5), the piezo- and pyroelectric coefficients are linear functions of the initial polarization P<sub>0</sub>. The experimental values applied to these equations are given in Table 1. Values of  $\alpha_c$  and  $\beta_c$  have been taken for the  $\alpha$  (non-polar) phase since these quantities for the  $\beta$  (polar) phase are apparently not available in the literature. The value of  $\alpha_c$  was obtained from the expansions of the unit cell dimensions<sup>6</sup>. The value of  $\beta_c$  was estimated by somewhat devious means from macroscopic compressibility measurements of a 68% crystalline fraction sample<sup>7</sup>.

In Fig. 3 the piezoelectric coefficient  $d_p$  is plotted against the initial polarization  $P_0$ . The straight line is a plot of the equation

$$d_p = 2.11 \times 10^{-10} P_0$$
 (SI units)

which is evaluated from eq.(5a) taking the values given in Table I. In this case the agreement between theory and experiment are essentially within data scatter.

The corresponding pyroelectric response is shown in Fig. 4. The straight line represents the equation

$$\kappa = 2.97 \times 10^{-4} P_0$$
 (SI units)

evaluated from eq.(5b) again using the experimental values from Table I. There is considerable underestimation by the theory for which the ratio, theoretical/experimental, is about 2/3. At this time we do not know why  $\kappa$ is underestimated.

We also polarized an undrawn PVDF sample composed of the "non-polar"  $\alpha$  phase. At an applied field of 1.75 MV/cm the responses are  $d_p = 13 \text{ pC/N}$  and  $\kappa = 3.1 \text{ nC/cm}^2\text{K}$ . A charge meter malfunction prevented us from obtaining a polarization value on this sample.

There is considerable interest in apparent morphological changes during the poling process. (See for example, references 1 and 8.) From reflection x-ray diffractometer measurements on biaxially drawn PVDF (comprising both  $\alpha$  and  $\beta$  phases), we have observed changes in structure at discrete intervals during the poling process. Fig. 5 shows the diffraction intensities with angle 20 before and after poling at 3.4 MV/cm. Note that the (100), (020) and (120) reflections from the  $\alpha$  phase essentially disappear after poling, whereas the (110) of  $\alpha$  and the combined (110) and (200) of  $\beta$  appear to increase slightly. At this time it is not clear if these increases are significant. At 36° 20 some changes occur, but, unfortunately, it is not possible to distinguish between contributions from  $\alpha$  and  $\beta$  phases in this region. typing s line

When the same specimens are examined in transmission using photographic techniques, the (100) and (020) reflections are still readily apparent so that only certain crystal orientations with respect to poling field are affected in this manner.

In looking at the undrawn ( $\alpha$  phase) sample, the situation appears to be slightly different, as revealed by Fig. 6. The (110), (020), and (100) intensities seem to diminish slightly with poling; however, none of these disappear as with the (020) and (100) for the biaxially stretched sample. The important result is that an additional reflection seems to appear at 20.7°, which would coincide with the (110) and (200) of 3 phase shown on Fig. 5. Based on this evidence alone, one might speculate that there is a partial conversion from  $\alpha$  to 8 phase during poling. However, we have not observed any of the other reflections to be expected from 8 phase either by reflection in the diffractometer, by transmission in a flat plate camera or in a powder camera. The spacings obtained by us using the powder camera for the unpoled samples are in good agreement with those obtained in Refs. 11 and 12.

Our tentative conclusion is that the non-polar  $\alpha$  phase is modified under the influence of high electric fields to yield a polar crystal phase which persists after removal of the external field. The unit cell of the  $\alpha$  phase is non-polar but the TGTG' configuration within a given chain traverse of the crystal has a large component of dipole moment normal to the chain axis. The mechanism proposed would involve rotation of polymer chains about their own axes to align the dipole with the applied field resulting in a metastable state. Those crystallites oriented with the chain axis parallel to the surface of the film would be subjected to the largest interaction energy. It is clear that the applied field has altered the orientation of crystallites and perhaps the crystal structure, but the details are not yet resolved.

#### References

1. P. D. Southgate, Appl. Phys. Letters., 28, 250 (1976).

- 2. F. I. Mopsik and M. G. Broadhurst, J. Appl. Phys., <u>46</u>, 4204 (1975).
- 3. M. G. Broadhurst, G. T. Davis, S. C. Roth, and R. E. Collins, Conference on Electrical Insulation and Dielectric Phenomena, Buck Hill Falls, Pa., October 1976.
- 4. Y. Wada and R. Hayakawa, Jap. J. Appl. Phys., 15, 2041 (1976).
- M. G. Broadhurst, C. G. Malmberg, F. I. Mopsik, and W. P. Harris, in Electrets, Charge Storage and Transport in Dielectrics, M.M. Perlman, ed. Electrochem. Soc. Princeton, N.J., p.492 (1973).
- 6. K. Nakagawa and Y. Ishida, Kolloid Z. u. Z. Polymere, 251, 103 (1973).
- 7. W. W. Doll and J. B. Lando, J. Macromol. Sci.-Phys., B2, 219 (1968).
- 8. D. K. Das Gupta and K. Doughty, Appl. Phys. Letters, <u>31</u>, 585 (1977).

9. H. Kakutani, J. Polym. Sci., A-2, 8, 1177 (1970).

10. M. G. Broadhurst and F. I. Mopsik, J. Chem. Phys., 52, 3634 (1970).

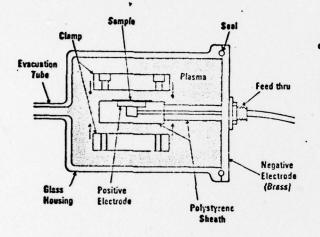
- R. Hasegawa, Y. Takahashi, Y. Chatani, and H. Tadakoro, Polym. J., 3, 600 (1972).
- 12. W. W. Doll and J. B. Lando, J. Macromol. Sci., B-4, 309 (1970).

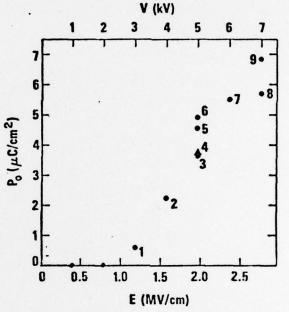
Experimen Quantiti	Table I tal Values in S es Appearing in	I Units of Eqs. (5)
Quantity	Value	Reference
°c	$1.69 \times 10^{-4}$	6
β <sub>c</sub>	1.56 X 10 <sup>-10</sup>	6,7
εc	3	.9
v	4.55	10
	0.284	2

300

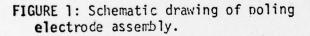
1

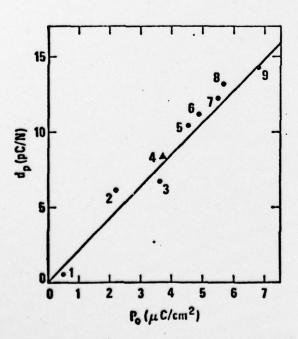
T





7





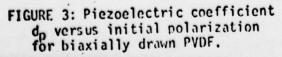
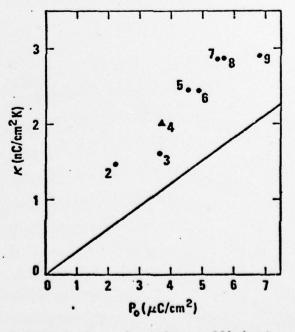
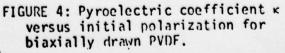


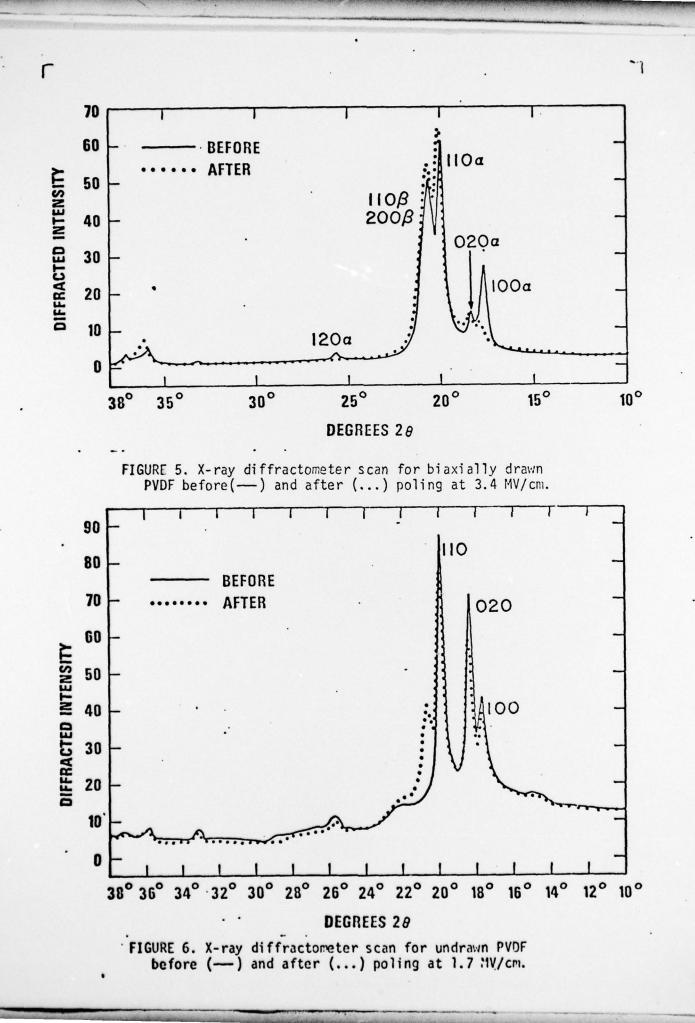
FIGURE 2: Initial polarization versus applied field (or voltage) for biaxially drawn PVDF.





1





#### TECHNICAL REPORT DISTRIBUTION LIST

Dr. D. R. Uhlman

#### No. Copies

2

Dr. Stephen H. Carr Department of Materials Science Northwestern University Evanston, Illinois 60201 1

Dr. M. Broadhurst Bulk Properties Section National Bureau of Standards U.S. Department of Commerce Washington, D.C. 20234

Dr. C. H. Wang Department of Chemistry University of Utah Salt Lake City, Utah 84112 1

Dr. T. A. Litovitz Department of Physics Catholic University of America Washington, D.C. 20017 1

Dr. R. V. Submaranian Washington State University Department of Materials Science Pullman, Washington 99103 1

Dr. M. Shen Department of Chemical Engineering University of California Berkeley, California 94720 1

Dr. R. S. Porter Polymer Research Institute and Polymer Science and Engineering University of Massachusetts Amherst, Massachusetts 01002 1

Dr. Morton Litt 1 Dept. of Macromolecular Science Case Western Reserve University Cleveland, Ohio 44106

Dr. V. Stannett Department of Chemical Engineering North Carolina State University Raleigh, North Carolina 27607 1

Department of Metallurgy and Material Science Center for Materials Science and Engineering Massachusetts Institute of Technology Cambridge, Massachusetts 02139 1 Naval Surface Weapons Center White Oak Silver Spring, Maryland 20910 1 Attn: Dr. J. M. Augl Dr. B. Hartmann Dr. C. Goodman Globe Union Inc. 5757 North Green Eay Avenue Milwaukee, Wisconsin 53201 1 Picatinny Arsenal SMUPA-FR-M-D Dover, New Jersey 07801 Attn: A. M. Anzalone 1 Bldg. 3401 Dr. J. K. Gillham Princeton University Department of Chernistry Princeton, New Jersey 08540 1 Douglas Aircraft Co. 3855 Lakewood Boulevard Long Beach, California 90846 Attn: Technical Library

Dr. E. Baer Department of Macromolecular Science Case Western Reserve University Cleveland, Onio 44106

C1 290/35-84

AUTO-Sutton

Dr. K. D. Pae Department of Mechanics and Materials Science Rutgers University New Brunswick, New Jersey 08903 1

1

1

No. Copies

# TECHNICAL REPORT DISTRIBUTION LIST

No. C	opies		No. Copie
Office of Naval Research .		Defense Documentation Center	
Arlington, Virginia 22217		Building 5, Cameron Station	
Attn: Code 472	2	Alexandria, Virginia 22314	12
Office of Naval Research		U.S. Army Research Office	
Arlington, Virginia 22217		P.O. Box 12211	
Attn: Code 102IP	6	Research Triangle Park, North Carolin	na 27709
		Attn: CRD-AA-IP	
ONR Branch Office		•	
536 S. Clark Street		Commander	
Chicago, Illinois 60605		Naval Undersea Research & Development	Ł
Attn: Dr. George Sandoz	1	Center	
	•	San Diego, California 92132	•
ONR Branch Office		Attn: Technical Library, Code 133	1
715 Broadway		termieat biblaif, tode 155	-
New York, New York 10003		Naval Weapons Center	
Attn: Scientific Dept.	1	China Lake, California 93555	
		Attn: Head, Chemistry Division	1
ONR Branch Office			-
1030 East Green Street		Naval Civil Engineering Laboratory	
Pasadena, California 91106		Port Hueneme, California 93041	
Attn: Dr. R. J. Marcus	1	Attn: Mr. W. S. Haynes	1
ONR Branch Office		Professor O. Heinz	
760 Market Street, Rm. 447		Department of Physics & Chemistry	
San Francisco, California 94102		Naval Postgraduate School	
Attn: Dr. P. A. Miller	1	Monterey, California 93940	
		•, , , , , , , , , , , , , , , , ,	
ONR Branch Office	•	Dr. A. L. Slafkosky	
495 Summer Street		Scientific Advisor	
Boston, Massachusetts 02210		Commandant of the Marine Corps (Code	RD-1)
Attn: Dr. L. H. Peebles	1	Washington, D.C. 20380	1
Director, Naval Research Laboratory	•		
Washington, D.C. 20390			
Attn: Library, Code 2029 (ONRL)	6		
Technical Info. Div.	1	· ·	
Code 6100, 6170	1		
The Asst. Secretary of the Navy (R&	D)		
Department of the Navy			
Room 4E736, Pentagon			
Washington, D.C. 20350 .	1		
Commander, Naval Air Systems Comman	d		
Department, of the Navy			
Washington, D.C. 20360			
Attn: Code 310C (H. Rosenwasser)	1		

I

#### No. Copies

1

1

1

NASA-Levis Research Center 21000 Brookpark Road Cleveland, Ohio 44135 Attn: Dr. T. T. Serofini, MS-49-1 1

Dr. Charles H. Sherman, Code TD 121 Naval Underwater Systems Center New London, Connecticut 1

Dr. William Risen Department of Chemistry Brown University Providence, Rhode Island 02912

Dr. Alan Gent Department of Physics University of Akron Akron, Ohio 44304

Mr. Robert W. Jones Advanced Projects Manager Hughes Aircraft Company Mail Station D 132 Culver City, California 90230 1

Dr. C. Giori IIT Research Institute 10 West 35 Street Chicago, Illinois 60616

Dr. Jerome B. Lando 1 Dept. of Macromolecular Science Case Western Reserve University Cleveland, Ohio 44106 Dr. David Roylance Department of Materials Science and Engineerin Massachusetts Institute of Technology Cambridge, Massachusetts 02039 1 Dr. W. A. Spitzig United States Steel Corporation Research Laboratory Monroeville, Pennsylvania 15146 1

Dr. T. P. Conlon, Jr., Code 3622 Sandie Laboratories Sandia Corporation Albuquerque, New Mexico 87115

Dr. Martin Kaufmann, Head Materials Research Branch, Code 4542 Naval Weapons Center China Lake, California 93555

Dr. Charles Hicks Naval Undersea Center San Diego, California 92132

Mr. A. Johnson N43-12 National Highway Traffic Safety Admin. 1 2100 2nd Street, S.W. Washington, D.C. 20590

#### to. Conies

1

1

1