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Chemical Effects in the Corrosion of Aluminum and Aluminum Alloys

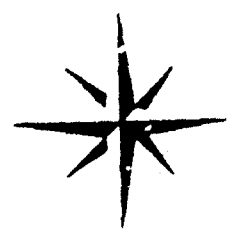
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By
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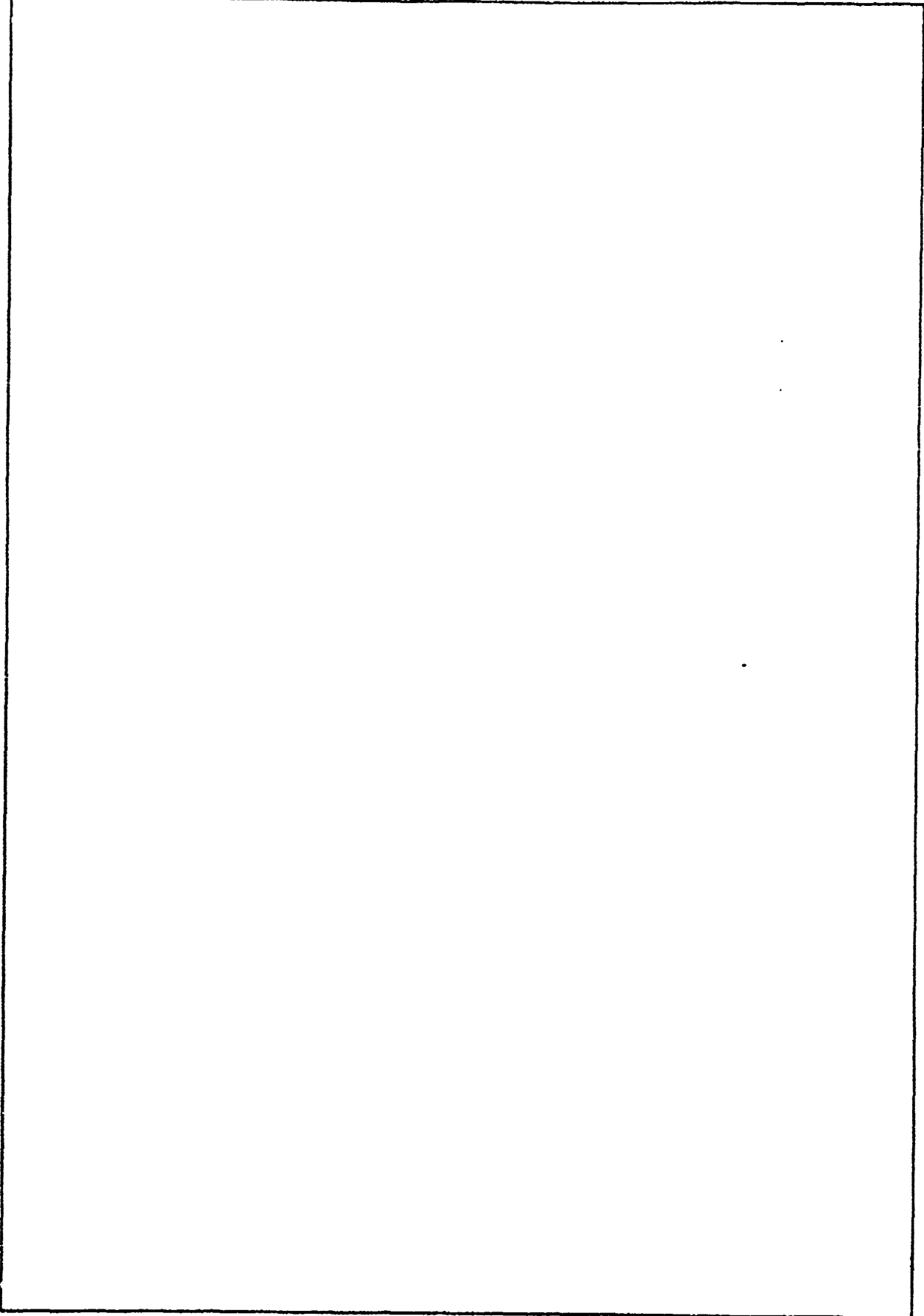
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"CHEMICAL EFFECTS IN THE CORROSION OF ALUMINUM AND
ALUMINUM ALLOYS"

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Bibliography of Literature on Chemical Effects
in the Corrosion of Aluminum and Aluminum Alloys

Introduction

In 1969 a bibliography of the literature on chemical effects in the corrosion of aluminum and aluminum alloys was issued by this Laboratory. In that first survey the literature from the year 1913 and through the year 1968 was critically examined, specifically with reference to information on the influence of anions and other chemical species on the corrosion of aluminum. This first search served the basis for experimental investigations that have been conducted in our Laboratory from that year up to the present.

With this bibliography the literature is covered from 1968 through 1975. Thus the present search overlaps the first.

In the present bibliographic review, considerable attention is given to corrosion inhibitors, again to support experimental work being conducted in the Laboratory. An effort has been made to do some evaluation, that is, not all of the available references were included, only those, in the Author's opinion that offer pertinent information or would lead to sources that would be of value to the corrosion program.

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 The corrosion of Al in a closed ethylene glycol/water system.
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 Hazards of traces of Hg on the corrosion of metal and alloys in the process industries.
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 Development of a corrosivity scale based on the theoretical treatment of anions in solution.
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 Effects of moderately-severe industrial and marine atmosphere on four Al alloys.
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 The corrosion-immunity-passivation diagram, the metal-potential
 of aq. system-pH diagram, and the Al-potential of aq. system-
 pH diagram are discussed.
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 Corrosion resistance of Al and its alloys in a NaCl medium.
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 Study of corrosion rates of Al and Al alloys 2S, 3S, and 5S in 1M NaOH with the addition of CaO and Na citrate.
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 Evaluation of the effectiveness of 11 complex-forming agents in inhibiting corrosion of Armco-Fe, M-1 Type Cu, Al and Type Ts 1 Zn.
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 Corrosion rate of Al in deaerated 3% NaCl solutions at pH 1-7.
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The influence of tartrate ions in alk. solns. during the anodic polarization and corrosion behavior of com. Al.
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Potentiostatic studies and chemical analysis of hexacyanoferrate (III) on 52S, 257S aluminum alloys and pure aluminum (99.50%).

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 Dissolution of Al in 2-4N HCl and 0.5-3N NaOH in the presence alkylamine and alkylammonium ions.
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 Corrosion of Al and its alloys in underground areas in the presence of water, oil or steam at 150° .
- 1972-22 E.Y. Lyublinskii and N.N. Bibikov
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 Intergranular Corrosion And Stress-Corrosion Cracking"
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 of restricted diffusion.
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 Inhibitive action of 2-mercaptobenzothiazole against
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 A review of the corrosion resisting characteristics
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 A review showing the importance of Al, composition and
 performance of anode installation, principles and design
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 Compilation of data from the immersion of 1150 spe-
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 water for 12 and 18 months.
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 "Effect Of Temp. On The Pit Formation Potential Of
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- 1972-30 R.M. Saleh and A.A. El Hosary
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 "Corrosion Inhibition By Naturally Occurring Substances.II. Effect Of Pomegranate Juice And The Aqueous Extract Of Pomegranate Fruits And Tea Leaves On The Corrosion Of Aluminum"
 The effect of the juices and tea leaves on the dissolution of Al in 2N HCl and 1.5N NaOH.
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 Adsorption of Bu₂S by Cu, mild steel, Zn and Al in N H₂SO₄ using radioactive S.
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 The effect of sucrose, glucose, fructose, mannose, and mannitol on the corrosion and anodic polarization of Al-2S in M NaOH in the presence of Ca.
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 "Corrosion Of Aluminum And A Mg-3 and A Mts Alloys In Ch⁺ Soln Containing Aqueous Solutions Of Ethylene Glycol"
 An electrochemical evaluation of the corrosion resistance of Al and Al alloys in aq. ethylene glycol solns containing chloride ions.
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 Use of gas chromatography to study the corrosion of metals in water at high temperatures.
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 A review w/43 refs.
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 A review is given with 44 refs. on the corrosion of steels, Ti and Al alloys.
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 Use of alicyclic compounds as inhibitors for Al 3S in HCl.
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 An evaluation is given of various inhibitors, mostly amines. for Al alloys in HCl under the influence of ext. anodic pol.
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 "Colloidal Substances As Inhibitors Of Corrosion Of Aluminum 2S And 57S In Acetic Acid And Chloro-Substituted Acetic Acids"
 Use of colloidal substances as corrosion inhibitors for Al in acetic and chloro-substituted acetic acids.

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 Current-voltage curves of an Al-0.28% Fe alloy and highly pure Al in buffered M NaCl solns.
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 "Pitting Corrosion Of Al In Synthetic Supplemented Water"
 Corrosion behavior of pure Al in water containing Ca^{2+} , Mg^{2+} , HCO_3^- , Cl^- , SO_4^{2-} and free Cl.
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 Effects of some ions in water on the number of corrosion pits and the depths of these pits on Al.
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 The mechanism of acid attack, the mechanism of corrosion inhibition and different types of corrosion inhibitors are discussed.
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 Presentation of the general aspects of scientific and technical problems of metal corrosion in the USSR.
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 The influence of some organic thiocompounds on the corrosion
 rate of 3S Al in 0.5 N HCl solutions.
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 Behavior Of A Columnar Al-Cu Alloy"
 Evaluation of the morphology and kinetic nature of corrosion
 of directionally solidified Al-4.5 wt% Cu.
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 Of Aluminum Alloys"
 The mechanism of anodic dissoln. of various heat treated Al
 alloys under plastic deformation was examd. in a 1% NH₄ borate
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 The inhibitor effect of some phenols on the corrosion of Al-
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 A review.

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The potentials of binary Al alloys (Al-0.5% Fe, Al-0.5% Ni,
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The types of corrosion to which Al alloys are subject are re-
viewed. Methods of protection are described with special ref.
to anodizing. Applications of Al alloys in the food industry
are given. No refs.

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 "Corrosion behaviour of aluminum in phosphate solutions"
 The potential of the Al (7429 - 90 - 5) electrode was measured at 30° in 5×10^{-4} M aq. phosphate (14265 -44 -2) solns. as a function of pH and electrolyte concn.
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 "Influence of Copper Bicarbonate Ions on the Corrosion of Aluminum Alloy in Saline Solutions"
 Corrosion rate studies of 3 Al. alloys (1100, 5052, 6063) were conducted in saline solns.
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 "Localized Corrosion of Al and its Alloys. I. Critical Potential, E_p , with respect to Pitting "
 Determination of the critical potential for Al and its alloys with Si and Zn in 0.5 M NaCl at pH 2 and pH 6.
- 1974-4 P.L. Bonora, G.P. Ponzano and V. Lorenzelli
Brit. Corros. J., 9, 112, (1974)
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 Corrosion of Al and its alloys in 0.5 M HCl and 0.5 M NaCl.
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 "N-Substituted Anilines as Corrosion Inhibitors for Aluminum -3S in HCl"
 Study of N-substituted anilines as corrosion inhibitors for Indal 3S(11146-15-9) in HCl solns.
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Corrosion 30, 427, (1974)
 "Use of Galvanokinetic Methods for the Determination of Characteristic Potentials for Pitting Corrosion on Aluminum"
 Use of galvanokinetic methods to det. the pitting potential and protection potential against pitting for unalloyed Al in a deaerated soln of 3% NaCl at 0° and 30°.
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 "Effect of the Nature of a Solvent on the Inhibiting Effect of Some Organic Compounds in HCl-Alcohol Systems"
 Use of pyridine derivatives or perylium perchlorates to inhibit corrosion of Fe, Al and Zn in 0.5-4.0 M HCl.
- 1974-8 F.E. Faller
Korrosion, 25, 128-32, (1974)
 "Corrosion behaviour of aluminum in sea water and brackish waters with special reference to shipbuilding conditions"
 The importance of the low electro-chem. potential of Al. & phy. chem. properties of Al. oxide on the corrosion resistance of Al. are reviewed with 8 refs.

- 1974-9 H. Fisher
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 "Corrosion inhibition mechanisms as compared to the inhibition mechanisms of other electrode reactions"
 A review with 3 refs.
- 1974-10 R.J. Gesl and A. Troiano
Corrosion, 3C, 274-9, (1974)
 "Stress Corrosion and Hydrogen Embrittlement in an Aluminum Alloy"
 A strain aging type of reversible H embrittlement was demonstrated for a high strength Al alloy in a 3% NaCl soln.
- 1974-11 W. Gruhl and F.E. Faller
Z. Werkstofftech., 5, 274-9, (1974)
 "Corrosion & stress corrosion, problems of aluminum structural alloys"
 The corrosion behaviour of several structural alloys of Al. in different media is discussed.
- 1974-12 E.S. Ivanov
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 "Mechanism of the Action of Aminochromates as Inhibitors of the Corrosion of Some Metals"
 Mechanism of the action of aminochromates as inhibitors of the corrosion of steel, Al, Cu, and Cd.
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 "Behaviour of Al in NaOCl Solns"
 Behaviour of aluminum (7429-90-5) in NaOCl solutions.
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 A review on the corrosion mechanism of Al and classification of corrosion inhibitors.
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 "Pitting Corrosion of Aluminum"
 A new method was developed to determine the so-called breakdown pot. V in the pitting corrosion of Al (7429 - 90 - 5) and Al alloys in NaCl solns. contg. various additives.
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 "Behaviour of Some Ketones and Ethers as Inhibitors of Corrosion of Aluminum in HCl and in NaOH Solution"
 The effectiveness of some ketones and ethers as inhibitors of corrosion of Al(7429 -90 -5) in hydroxide soln and in HCl both in the presence and absence of Ca.

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 "Effect Of Magnesium On The Anodic Behavior Of Aluminum Magnesium Alloys In Alkaline Nitrate Solutions"
 Mg effect on the corrosion rate of Al-Mg alloys in alk.-nitrate solns was determined by studying the anodic behavior of this system.
- 1974-18 B.S. Lee, M. Seno and T. Asahara
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 "Vapor Phase Corrosion Inhibitors. 4. Effects Of Various Sub-Components On The Corrosion-Inhibiting Action Of Hexamerhylenetetramine On Aluminum"
 Investigation of vapor-phase corrosion-inhibiting papers, powders, tables and oils for Al.
- 1974-19 R.T. Lawson
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 "Al Corrosion Studies. I. Potential-pH-Temperature Diagrams For Aluminum"
 The potential-pH-temp. relation for the Al-H₂O system were calcd. by the methods of Bethune, Khodakovskiy, Criss & Cobble and Nelgeson & a crit. comparison made.
- 1974-20 T.J. Lennox, M.H. Peterson, J.A. Smith and R.E. Groover
Mater. Performance, 13, 31, (1974)
 "Corrosion And Cathodic Protection Of 5086-H32 Aluminum Coupled To Dissimilar Metals"
 Study of corrosion of 5086-H32 aluminum in various environments.
- 1974-21 H. Lommel
Korrosion, 25, 29-34, (1974)
 "Corrosion Behavior And Corrosion Protection Of Light Metal Alloys In Shipbuilding & Similar Applications"
 The corrosion behavior of Al and its alloys in shipbuilding is reviewed with 7 refs. The effect of structure, compn., heat treatment, welding & contact with other metals is considered.
- 1974-22 V.A. Makavov
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 A review w/194 refs.
- 1974-23 A. Maitra and S. Barua
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 "Dicyandiamide. Inhibitor For Acid Corrosion Of Pure Aluminum"
 Inhibition efficiency of dicyandiamide on Al in 0.5-2.0 N HCl at 27° for 1-6 hrs.

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 "Stimulated Crevice Corrosion Expt. For pH And Solution Chemistry Determination"
 Crevice corrosion in dental amalgam
- 1974-25 T.L. Rama Char
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 "Aluminum. Corrosion And Metal Finishing Bibliography Of Publications"
 A list of 83 papers published from the electrochem. labs of the world covering various aspects of the use of Al(7429-90-5).
- 1974-26 B. Sanyal, et. al.
Indian Chem. Manuf., 12, 13, (1974)
 "Corrosion Of Metal In Different Chemical Environments And Its Protection"
 Corrosion of Al, steels and bronzes in acids, water, sea water, cutting oil emulsions, detergents, petroleum and org. solvents is reviewed.
- 1974-27 R.K. Shah, B.B. Patel and N.K. Patel
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 "Azoles As Corrosion Inhibitors For 3S Aluminum In Local Supply Water"
 Use of 2-mercaptobenzothiazole and 2-mercaptobenzimidazole as corrosion inhibitors for 3S Al in Gujarat Univ supply water.
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 Statistical study of the corrosion of Al alloys in sea water and in an industrial atm for 5 yrs.
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 "Protection Of Metals And Light Alloys"
 A review is given of industrial Al and Mg alloys and of corrosion and its prevention.
- 1974-30 N. Subramanyan and K. Ramakrishnaiah
 Proc. Semin. Electrochem. 14th 1973, 375, (1974)
 "Effect Of Some Amino Acids In The Corrosion Of Al In 1M HCl"
 The influence of 10 amino acids on the corrosion of Al (7429-90-5) in 1 N HCl both in the presence and absence of Ca (7440-90-2).

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 The influence of amino acids on the corrosion of Al (7429-90-5) in 1 N HCl both in the presence and absence of Ca (7440-90-2).
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 "Corrosion Inhibitors"
 Types of inhibitors and their applications in various chem. and environmental processes are reviewed.
- 1974-32 J.D. Talati and J.M. Pandya
Anti-Corros. Method Mater., 21, 7, (1974)
 "Amines As Corrosion Inhibitors For B26S Al in H₃PO₄"
 The inhibition of corrosion of Al-Cu (4%) alloy B26S in 0.1 N solns of H₃PO₄ (7664-38-2) by different amines.
- 1974-33 S. Terai, Z. Tanabe and M. Hagiwara
Suiyokai-Shi, 18, 80, (1974)
 "Corrosion And Corrosion Control Of Aluminum And Its Alloys. I"
 A review w/46 refs.
- 1974-34 S. Terai, Z. Tanabe and T. Suzuk
Suiyokai-Shi, 18, 80, (1974)
 "Corrosion And Corrosion Control In Aluminum And Its Alloys. II"
 A review w/84 refs is given on cathodic protection of Al, corrosion inhibitors and corrosion protection by surface treatment.
- 1974-35 D.E. Taylor and R.B. Waterhouse
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 "Electrochemical Investigation Of Fretting Corrosion Of A Number Of Pure Metals In 0.5M Sodium Chloride"
 Free potential measurement supplemented by transient linear polarization was used to study the effect of fretting corrosion on Zn, Al, Ag, Ta, Cu, Cr, and Ni in 0.5 M NaCl.
- 1974-36 E.D. Verink, Jr.
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 Work at an exptl. desalting plant shows that Al alloys can handle saline water with min. corrosion.

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A review on corrosion inhibitors for Al in acid, basic,
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"Anodic Polarization Behavior And Pitting Corrosion Mechanism
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Anodic polarization studies of 99.999% Al in 0.5 N NaCl soln
of pH 4,6,8.5,10 and 12.
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"Dihydroxy-benzenes As Corrosion Inhibitors For Al-Cu Alloy
In NaOH"
Investigation of dihydroxy-benzenes as corrosion inhibitors
of Al-4% Cu alloys in NaOH.