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ANALYTIC AND NUMERICAL CHEMISTRY ALGORITHMS FOR THE
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David W. Goetz, et al

Air Force Weapons Laboratory
Kirtland Air Force Base, New Mexico

June 1975

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Worry Document No. 2

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**Air Force Weapons Laboratory
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June 1975

Final Report for Period 1 January 1971 - 15 July 1974

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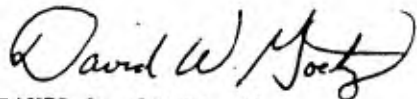
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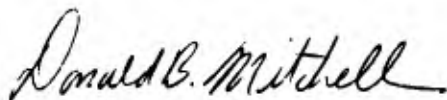
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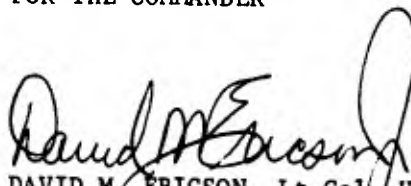


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SECTION I

INTRODUCTION

Nuclear weapon effects codes such as the Air Force Weapons Laboratory (AFWL) WORRY optical-IR systems code require the repetitive and fast computation of non-equilibrium atmospheric chemical behavior at numerous points in a disturbed environment. Such computations necessitate the solution of a set of nonlinear coupled differential equations, often stiff, which describe the kinetic behavior of the chemical species of interest. The algorithm which is chosen for solution of these differential equations must be fast in computational time, reliable, and versatile in handling various initial conditions of species, temperature, and density.

At the inception of WORRY in late 1970-early 1971 a limited but adequate set of chemical species, table I, and their reactions, table II, were chosen to describe the terrestrial atmosphere from an optical/infrared viewpoint under both ambient and disturbed conditions. Several reliable algorithms were available for numerical integration of the differential equations defined by these species and their reactions. However, none were fast enough even in the most favorable situations and under adverse conditions e.g., at altitudes below 30 km, they often become inoperative. The answer to the need for a better algorithm was found by devising an analytic solution to the set of differential equations.

An analytic solution requires the making of simplifying assumptions that decouple the differential equations to a degree which leaves the remaining equations amenable to analytic solution. These assumptions obviously limit the generality and applicability of the method. The key to choosing a useful algorithm lies in defining a working range for the solutions and balancing uncertainties - in reactions, reaction rates, and solution assumptions. The analytic

algorithm for WORRY was so conceived. Since the chemical behavior to be described by the solution was limited by the structure of the WORRY code to points outside any fireballs, the two basic limiting assumptions of the analytic algorithms, (1) the molecular temperature does not exceed $\sim 1200^\circ\text{K}$ and (2) the fractional ionization does not exceed ~ 0.1 percent, were acceptable.

The continued development of the WORRY analytic algorithm has required a standard against which to judge its effectiveness. Therefore, a total chemistry package was developed for WORRY, containing both the analytic algorithm and a numerical algorithm to solve the same set of differential equations "exactly" and serve as a check on the "correctness" of the analytic solution. To ensure the comparability of outputs, both algorithms share the same inputs and the same subroutines which calculate initial conditions and reaction rates.

The analytic algorithm to be described in section IV is undergoing continual change. The description in section IV and the code listing (with external driver) found in section VII reflect the status of the chemistry package at the time of the writing of this document.

Table 1

WORRY SPECIES

H, Li, Be, Mg, Al, Si, Ti, Fe, Cu, U
 OH, LiO, BeO, MgO, AlO, SiO, TiO, FeO, CuO, UO⁺
 NO⁺, NO, NO₂, N₂O, CO, CO₂, O₃, H₂O, CN, e⁻
 N₂, N₂(A), N₂(B), N(⁴S), N(²D), N(²P), O(³P), O(¹D), O(¹S), O⁻
 N₂⁺, O₂⁺, N⁺, O⁺, N₂⁺(A), N₂⁺(B), O₂, O₂(¹Δ), O₂(¹Σ), O₂(³Σ)
 H₂, UO, NO(A), NO(B), NO(C), NO(D), CN(A)

TABLE II

REACTIONS AND RATE CONSTANTS

$$k = a(T/300)^b \exp(-c/T) [M]^d$$

T = temperature (°K)

X = Li, Be, Mg, Al, Si, Ti, Fe, or Cu

	a 6.68 x 10 ¹²	b 0	c 3590	d 0
1. N(⁴ S) + O ₂ → NO + O N(² P) + O ₂ → NO + O				
2. N(² D) + O ₂ → NO + O	6.0 x 10 ⁻¹²	.5	0	0
3. N(⁴ S) + NO → N ₂ + O N(² P) + NO → N ₂ + O	5.1 x 10 ⁻¹¹	0	168	0
4. N(² D) + NO → N ₂ + O	6.0 x 10 ⁻¹¹	.5	0	0
5. O + O ₂ + M → O ₃ + M	2.8 x 10 ⁻³⁵	0	-900	1
6. O + O ₃ → O ₂ + O ₂	2.0 x 10 ⁻¹¹	0	2400	0
7. N ₂ ⁺ + e → 0.99 N(⁴ S) + 0.99 N(² D) + 0.02 N(² P)	2.8 x 10 ⁻⁷	-.2	0	0
8. N ₂ ⁺ + O ₂ → O ₂ ⁺ + N ₂	2.0 x 10 ⁻¹⁰	0	0	0
9. N ₂ ⁺ + O → NO ⁺ + 0.5 N(⁴ S) + 0.5 N(² D)	2.5 x 10 ⁻¹⁰	0	0	0

Table II (Cont'd)

	a	b	c	d
10. $O_2^+ + e \rightarrow 0.8 O(^3P) +$ $0.8 O(^1D) + 0.4 O(^1S)$	2.2×10^{-7}	-7	0	0
11. $O^+ + O_2 \rightarrow O_2^+ + O(^3P)$	2.0×10^{-11}	0	0	0
12. $NO^+ + e \rightarrow 0.75 O(^3P) +$ $0.75 N(^2D) + 0.25 O(^1D) +$ $0.25 N(^4S)$	5.0×10^{-7}	-1.2	0	0
13. $O_2 + N^+ + O_2^+ + 0.5 N(^4S) +$ $0.5 N(^2D)$	2.5×10^{-10}	0	0	0
14. $O_2 + N^+ + NO^+ + O(^3P)$	5.0×10^{-10}	0	0	0
15. $N_2 + O^+ \rightarrow NO^+ + N(^4S)$	1.3×10^{-12}	0	0	0
16. $O^+ + NO \rightarrow NO^+ + O(^3P)$	2.4×10^{-11}	0	0	0
17. $O_2^+ + NO \rightarrow NO^+ + O_2$	8.0×10^{-10}	0	0	0
18. $N_2^+ + NO \rightarrow NO^+ + N_2$	5.0×10^{-10}	0	0	0
19. $N^+ + NO \rightarrow NO^+ + 0.5 N(^4S) +$ $0.5 N(^2D)$	8.0×10^{-10}	0	0	0

Table II (Cont'd)

	a	b	c	d
20. $O_2^+ + N \rightarrow NO^+ + O(^3P)$	1.8×10^{-10}	0	0	0
21. $H + O_3 \rightarrow OH + O_2$	2.6×10^{-11}	0	0	0
22. $OH + O \rightarrow H + O_2$	5.0×10^{-11}	0	0	0
23. $O_2(^3\Sigma) + O_2 \rightarrow O_3 + O$	1.0×10^{-12}	0	0	0
24. $O_3 + NO \rightarrow NO_2 + O_2$	9.5×10^{-13}	0	1300	0
25. $O + NO + M \rightarrow NO_2 + M$	1.3×10^{-31}	-2.5	0	1
26. $O + NO_2 \rightarrow NO + O_2$	2.8×10^{-11}	0	550	0
27. $NO_2 + N \rightarrow N_2 + O_2$	1.5×10^{-11}	0	0	0
28. $NO + N + M \rightarrow N_2O + M$	1.0×10^{-34}	0	1000	1
29. $NO_2 + N \rightarrow N_2O + O$	1.5×10^{-11}	0	0	0
30. $O_2(^1\Delta) + N_2O \rightarrow N_2 + O_3$	1.0×10^{-13}	0	0	0
31. $N(^4S) + O_3 \rightarrow NO + O_2$	3.4×10^{-11}	.5	1200	0
32. $N(^2P) \rightarrow N(^2D) + h\nu$	7.78×10^{-2}	0	0	0
33. $O(^1S) \rightarrow O(^1D) + h\nu$	6.99×10^{-1}	0	0	0
34. $O(^1D) \rightarrow O(^3P) + h\nu$	6.25×10^{-3}	0	0	0

Table II (Cont'd)

		a	b	c	d
35.	$N^+ + O \rightarrow O^+ + N(^4S)$	1.0×10^{-12}	0	0	0
36 - 49	for future expansion				
50 - 83	for metals reactions				
	$X + O_2 \rightarrow XO + O$				
	$X + O + M \rightarrow XO + M$				
	$X + O_3 \rightarrow XO + O_2$				
	$XO + O \rightarrow X + O_2$				
	$XO + O_3 \rightarrow X + O_2 + O_2$				
84 - 89	for future expansion				
90	$O_3 + h\nu \rightarrow O_2 + O(^3P)$				

SECTION II

CHEMISTRY PACKAGE STRUCTURE

The WORKY Chemistry Package is structured as shown in the flow chart of figure 1. A requesting program calls DECAYR which either executes the analytic algorithm or transfers control to the numerical algorithm executive, DECAR, which employs subroutines FUNG and EVIL. The three routines and one entry point shared by both analytic and numerical solutions are ENTDEP, ENQDEP, RATE, and RRCRT. ENTDEP calculates the initial distribution of ions and neutrals produced by prompt energy deposition (ergs/cc), while the entry point ENQDEP partitions the continuous energy deposition (ion pairs/cc/sec) among the same set of ions and neutrals. For a given temperature and total density, the RATE subroutine fills the R array with the first-order, second-order, and pseudo-second-order reaction rate constants of the set of reactions in table II. RRCRT stores the species destruction rates and transforms the species formation rates as calculated by either algorithm into vibrational level formation rates to be used in the WORRY calculation of chemiluminescence. The remaining entries in figure 1 are functions used by DECAYR in its analytic solutions and will be elaborated in sections IV and V.

The chemistry package requires the following inputs for a problem:

1. The time, t_c , after instantaneous energy deposition for which the calculation is to be performed.
2. The initial, predeposition temperature ($^{\circ}\text{K}$).
3. The instantaneous energy deposition (ergs/cm³) at $t = 0$.
4. The continuous energy deposition (ion pairs/cm²/sec) at t_c .
5. The time of calculation, as day or night.
6. Concentrations for any metals present.

7. An initial set of species concentrations at $t = 0$ for ambient or previously insulted conditions. Ambient initialization may be accomplished by a prior call to the chemistry package with ambient inputs for 2, 3, and 4, a large value for t_c , and values for the concentrations of N_2 , O_2 , O , O_3 , NO , CO_2 , H , and OH .

The inputs as well as chemistry package outputs are accomplished through either the DECAYR calling arguments or COMMON blocks. The energy depositions (inputs), temperature (input/output), and time of calculation (input), are found in the DECAYR calling arguments. The variable DAY (input) is found in COMMON BURST and the DYMETL array (input) holds the metals concentrations (to be added to any preexisting concentrations). The CMD array (input/output) in COMMON XUX holds the concentrations of all species while the outputs of subroutine RRCRT are found in commoned arrays RR and RCRT.

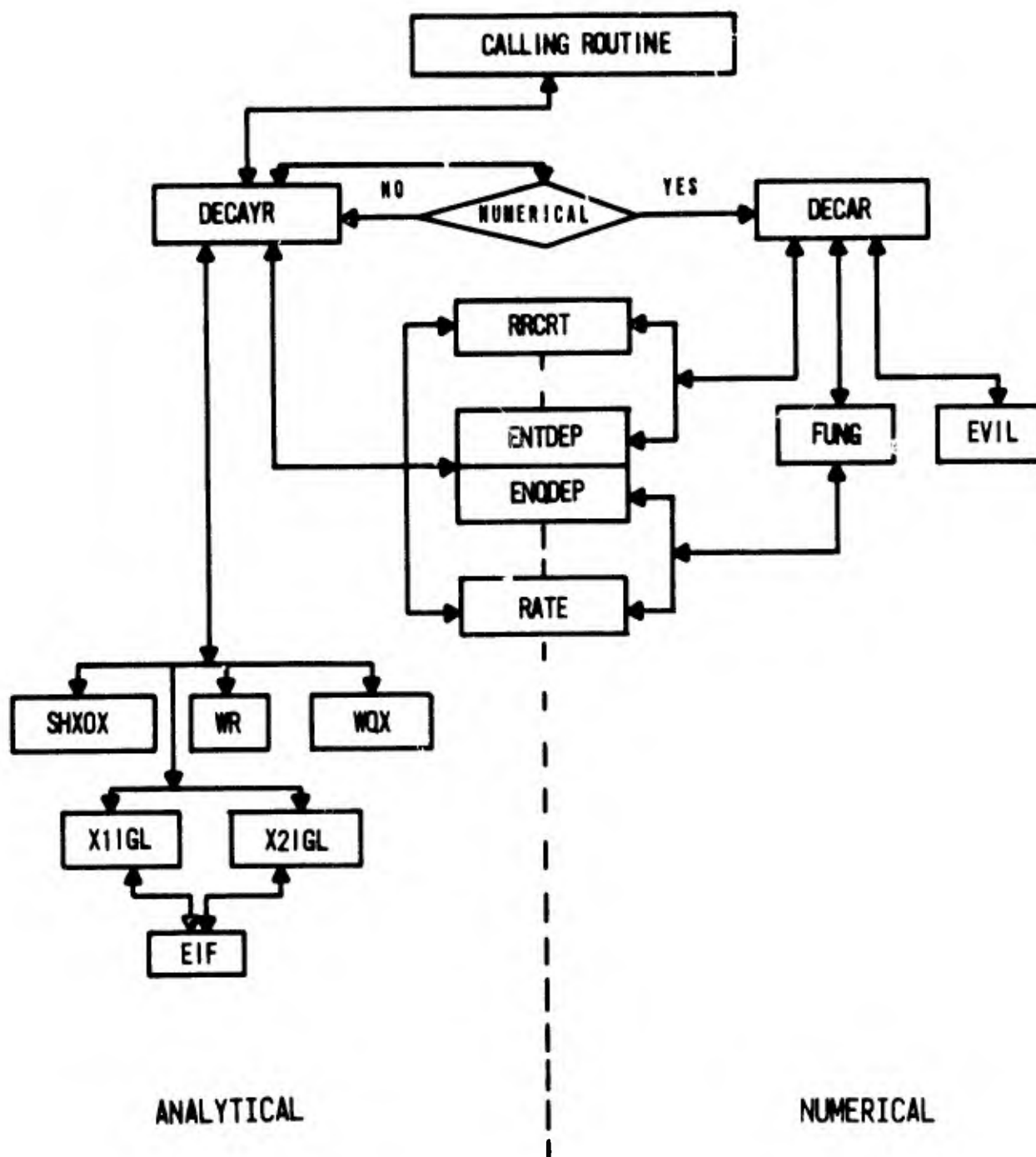


Figure 1. Flow Chart of Routines in WORRY Chemistry Package.

SECTION III

NUMERICAL CHEMISTRY

In DECAR, the numerical chemistry algorithm begins by calling ENTDEP to initialize the species concentrations and choosing an initial time step, DT, of $10^5/\text{CON}$, where CON is the total number density. The numerical integration is performed by a modified Kenneshea technique. The differential equation for each species is assumed to have the form $d[x]/dt = P - R[x]$, where $[x]$ is the species concentration, P is the production rate in $\text{cc}^{-1} \text{sec}^{-1}$, and R is the removal rate in sec^{-1} . P and R are assumed to be independent of $[x]$; and if they are assumed to be constant over the time interval DT, the solution may be written.

$$[x(T+DT)] = ([x(T)] - P/R)E^{-RT} + P/R \quad (1)$$

P and R are evaluated at time T; $[x]$ is then evaluated at $(T+DT)$ and used to evaluate P and R at $(T+DT)$. These values of P and R are then used with $[x(T)]$ to re-evaluate $[x]$ at $(T+DT)$. It is required that $|[x(T+DT)] - [x(T)]|/[x(T)]$ be less than ϵ for each species, where $\epsilon = 10^{-4}$ at the present time. The above process of requiring convergence is repeated three times, and if the requirement is not met, DT is halved. If the requirement is met, DT is doubled for the next step. This process is continued until the desired calculational time, END, is reached. When END is reached, RRCRT is called and the algorithm is complete.

The subroutine FUNG is used to evaluate P and R, and the subroutine EVIL is used to calculate $[x(T+DT)]$ from $[x(T)]$, P, and R. In addition to the species

concentrations, the internal kinetic energy of the gas is carried as a variable in the CMD array. The release and absorption of energy for each reaction is calculated in FUNG, and the energy is updated along with the other species. The updated energy is used to calculate a new temperature for each time step and a call to RATE is made to recalculate the reaction rate constants if necessary.

SECTION IV

ANALYTIC CHEMISTRY

As in the numerical chemistry, the analytic algorithm found in subroutine DECAVR begins by calling ENTDEP to update the CMD species array for the prompt energy deposition production of ions and neutrals at $t = 0$. The resultant updated temperature from ENTDEP is then used in a call to the RATE routine which fills the R array with all necessary rates constants for the reactions of table 2. As a final initialization step, the algorithm calculates the total energy of the sample volume using a table of heats of formation, the CMD array of species concentrations, and the total internal kinetic energy of the volume as calculated in ENTDEP.

The first assumption in the analytic algorithm is the separation of the fast-charged species from the slower neutral species chemistry. This separation and the following treatment are acceptable under (and lead to) the two basic assumptions of the analytic algorithm listed in section I. The faster charged species reactions are treated first and the resultant heating is calculated before the neutral species chemistry is begun. The charged species-- N^+ , O^+ , N_2^+ , O_2^+ , and NO^+ --react primarily through charge exchange and recombination reactions as found in reactions 7-20 and 35. The following description tracks the computer coding closely through the charged species and neutral species chemistries.

1. NITROGEN ION (N^+)

N^+ is at the top of the charge exchange chain and has no production from other species reactions. Assuming, as will be done successfully in many cases throughout the analytic treatment, that the concentrations of the species with

which N^+ reacts are constant, the N^+ loss mechanisms of reactions 13, 14, 19 and 35 result in the homogeneous differential equation for $[N^+]$.

$$d[N^+]/dt = - \{ k_{13}[O_2] + k_{14}[O_2] + k_{19}[NO] + k_{35}[O] \} [N^+] \quad (2)$$

The sum of constant quantities multiplying $[N^+]$ is defined as β_N and the equation is rewritten as

$$d[N^+]/dt = - \beta_N [N^+] \quad (3)$$

to which the solution is a simple exponential decay.

$$[N^+] = [N^+]_0 e^{-\beta_N t} \quad (4)$$

The full differential equation for $[N^+]$ including the continuous source is

$$d[N^+]/dt = Q_N + - \beta_N [N^+] \quad (5)$$

where Q_N is the production rate of $[N^+]$ from the continuous energy deposition. The particular solution to this inhomogeneous equation form is calculated in the Chemistry Package by FUNCTION WQX, which is described in Section V.

The final concentration of N^+ at the calculational time is taken as the sum of the solution, $[N^+]_t$, of Eq (4) and the particular solution for Eq (5) where $[N^+]_0$ in the later case is assumed to be zero. This procedure is general throughout the analytic treatment. Each species is treated twice - once without and once with the continuous energy deposition considered- and the results are either added or the maximum is

taken depending upon the species and the specific mathematical treatment and boundary conditions assumed.

2. OXYGEN ION (O^+)

O^+ is handled in a manner similar to N^+ except that the differential equation for $[O^+]$ includes a production term from the charge exchange of N^+ with O . Consideration of Reactions 11, 15, 16, and 35 leads to the inhomogeneous equation for the instantaneous solution.

$$d[O^+]/dt = - \{ k_{11}[O_2] + k_{15}[N_2] + k_{16}[NO] \} [O^+] + k_{35}[N^+][O] \quad (6)$$

The equation is rewritten by defining the constants β_0 and γ .

$$d[O^+]/dt = - \beta_0 [O^+] + \gamma [N^+] \quad (7)$$

Substituting Eq (4) for $[N^+]$ gives

$$d[O^+]/dt = - \beta_0 [O^+] + \gamma [N^+]_0 e^{-\beta_N t} \quad (8)$$

to which the general solution is of the form

$$[O^+] = \alpha e^{-\beta_0 t} + [\gamma [N^+]_0 / (\beta_0 - \beta_N)] e^{-\beta_N t} \quad (9)$$

The value of α is determined from initial conditions and the solution takes the form

$$[O^+] = \{ [O^+]_0 - \gamma [N^+]_0 / (\beta_0 - \beta_N) \} e^{-\beta_0 t} + \{ \gamma [N^+]_0 / (\beta_0 - \beta_N) \} e^{-\beta_N t} \quad (10)$$

If $y = \gamma [N^+]_0 / (\beta_0 - \beta_N)$ the final form of the instantaneous solution is

$$[O^+] = ([O^+]_0 - y) e^{-\beta_0 t} + y e^{-\beta_N t} \quad (11)$$

The solution for $[O^+]$ including the continuous source is found as for $[N^+]$ from the equation

$$d[O^+]/dt = Q - \beta_0[O^+] \quad (12)$$

where $Q = Q_{O^+} + \gamma[N^+]_t$ and $[O^+]_0$ is assumed to be zero.

The final concentration of O^+ at the calculational time is the sum of the instantaneous and continuous energy deposition solutions.

3. ELECTRON CONCENTRATION

Before proceeding with the positive molecular ions, an intermediate update is made to the electron concentration. The same calculational scheme is repeated after the positive molecular ions have been treated to give the final electron concentration. (At present, n_e is considered as the total negative charge concentration in an equal α approximation.)

Consideration of the continuous source production rate is postponed as usual, and Reactions 7, 10, and 12 define the electron concentration behavior

$$dn_e/dt = - \{k_7[N_2^+] + k_{10}[O_2^+] + k_{12}[NO^+]\} n_e \quad (13)$$

where the approximation has been made that atomic ions do not recombine—a valid approximation due to their very slow recombination rates compared to the molecular ions. An average recombination rate, α_r , is defined

$$\alpha_r = \frac{k_7[N_2^+] + k_{10}[O_2^+] + k_{12}[NO^+]}{[N_2^+] + [O_2^+] + [NO^+] + [O^+] + [N^+]}$$

and Eq (13) is replaced by

$$dn_e/dt = - \alpha_r n_e P \quad (14)$$

where P is the number of molecular positive ions,

$P = n_e - ([O^+] + [N^+])$. Defining $[A^+] = [O^+] + [N^+]$, Eq (14) is rewritten as

$$dn_e/dt = -\alpha_r n_e (n_e - [A^+]) \quad (15)$$

and if $[A^+]$ is treated as a constant the solution for n_e is

$$n_e = [A^+] + (n_o - [A^+]) / (1 + \alpha_r' n_o t) \quad (16)$$

where $\alpha_r' = \alpha_r [\exp(\alpha_r [A^+] t) - 1] / (\alpha_r [A^+] t)$ and n_o is the initial electron concentration at $t = 0$. This expression for n_e is exact in the limits of $[A^+]$ equal to zero and infinity, and from comparisons with numerical solutions, Eq (16) appears to be valid over a wide range of $[A^+]$ values.

The continuous source production of negative charge is handled simply by the solution of the equation

$$dn_e/dt = Q - \alpha_r n_e^2 \quad (17)$$

where the ion-pair production rate, Q , has been added to Eq (14) and the approximation, $P = n_e$, has been made.

The solution to Eq (17) is

$$n_e = \sqrt{\frac{Q}{\alpha_r}} (1 + ae^{-2\sqrt{\alpha_r Q} t}) / (1 - ae^{-2\sqrt{\alpha_r Q} t}) \quad (18)$$

where $a = (n_o - \sqrt{\frac{Q}{\alpha_r}}) / (n_o + \sqrt{\frac{Q}{\alpha_r}})$. This solution is contained in the Chemistry Package as FUNCTION WR.

The final electron concentration n_e at the calculational time is the maximum of the instantaneous and continuous source solutions.

4. MOLECULAR NITROGEN ION (N_2^+)

N_2^+ is considered next in the charge exchange chain of positive ions. The differential equation for $[N_2^+]$ is derived from Reactions 7, 8, 9, and 18

$$d[N_2^+]/dt = -k_7 n_e [N_2^+] - \{k_8[O_2] + k_9[O] + k_{18}[NO]\} [N_2^+] \quad (19)$$

Equation (19) may be rewritten by defining $\beta_{N_2} = k_8[O_2] + k_9[O] + k_{18}[NO]$ and substituting Eq (16) for n_e

$$d[N_2^+]/dt = -\{k_7[A^+] + k_7(n_o - [A^+])/(1 + \alpha'_r n_o t) + \beta_{N_2}\} [N_2^+] \quad (20)$$

If α'_r and β_{N_2} are treated as constants and Eq (4) and Eq (11) define

$[A^+] = [O^+] + [N^+]$, the instantaneous solution for N_2^+ is

$$\begin{aligned} [N_2^+] = [N_2^+]_o \exp\left\{-\beta_{N_2} t - k_7([O^+]_o - y)/\beta_o(1 - e^{-\beta_o t})\right. \\ \left. - k_7([N^+]_o + y)/\beta_N(1 - e^{-\beta_N t}) + (k_7/(\alpha'_r n_o))([O^+]_o - y)e^{A_1}(E_1(A_1) - E_1(B_1))\right. \\ \left. + ([N^+]_o + y)e^{A_2}(E_1(A_2) - E_1(B_2))\right\} / (1 + \alpha'_r n_o t)^{k_7/\alpha'_r} \end{aligned} \quad (21)$$

where $A_1 = \beta_o/(\alpha'_r n_o)$, $B_1 = A_1(1 + \alpha'_r n_o t)$, $A_2 = \beta_N/(\alpha'_r n_o)$, $B_2 = A_2(1 + \alpha'_r n_o t)$, and

$E_1(A)$ is the exponential integral $\int_A^\infty \frac{e^{-x}}{x} dx$ which is calculated by

FUNCTION EIF. Equation (21) has been modified during the development of the analytic treatment to assure the best possible agreement with numerical solutions. In particular, the denominator $(1 + \alpha'_r n_o t)^{k_7/\alpha'_r}$ has been replaced by $(1 + \alpha'_r n_o t)$, resulting in a more even performance of the solution over a wide range of α'_r values.

The inclusion of a continuous source term in Eq (19) results in the full equation

$$d[N_2^+]/dt = Q_{N_2^+} - k_7 n_e [N_2^+] - \beta_{N_2} [N_2^+] \quad (22)$$

which may be regrouped if n_e is assumed to be a constant

$$d[N_2^+]/dt = Q_{N_2^+} - \beta [N_2^+] \quad (23)$$

where $\beta = k_7 n_e + \beta_{N_2}$.

The solution to Eq (23) is found as for N^+ and O^+ using the FUNCTION WQX with $[N_2^+]_0 = 0$.

The final concentration of N_2^+ at t_c is considered to be the maximum of the instantaneous and continuous source solutions.

5. MOLECULAR OXYGEN ION (O_2^+)

Next in the charge exchange chain, O_2^+ is involved in Reactions 8, 10, 11, 13, 17, and 20. Of these, Reactions 8, 11 and 13 produce $[O_2^+]$ and are treated as increasing $[O_2^+]$ at $t = 0$ by an amount

$$x_{O_2^+} = k_8 [O_2] \int_0^t [N_2^+] dt' + k_{11} [O_2] \int_0^t [O^+] dt' + k_{13} [O_2] \int_0^t [N^+] dt' \quad (24)$$

where $I_{O^+} = \int_0^t [O^+] dt'$ and $I_{N^+} = \int_0^t [N^+] dt'$ can be found by integrating

Eq (11) and Eq (4) respectively.

The integral $I_{N_2^+} = \int_0^t [N_2^+] dt'$ can not be performed exactly, therefore, an integrable form is assumed

$$[N_2^+] = [N_2^+]_0 e^{-\beta_e^{N_2} t} / (1 + \alpha_r n_o t) \quad (25)$$

where $\beta_e^{N_2}$ is calculated by setting Eq (25) equal to the results of Eq (21). The integral may now be evaluated

$$I_{N_2^+} = [N_2^+]_0 (e^{A_3 / (\alpha_r n_o)}) [E_1(A_3) - E_1(B_3)] \quad (26)$$

where $A_3 = \beta_e^{N_2} / (\alpha_r n_o)$ and $B_3 = A_3 (1 + \alpha_r n_o t)$ and Eq (24) is condensed to

$$x_{O_2^+} = [O_2] \left\{ k_8 I_{N_2^+} + k_{11} I_{O^+} + k_{13} I_{N^+} \right\} \quad (27)$$

The remaining Reactions 10, 17, and 20 define the differential equation for $[O_2^+]$

$$d[O_2^+] / dt = -k_{10}[O_2^+] n_e - k_{17}[NO][O_2^+] - k_{20}[N][O_2^+] \quad (28)$$

or substituting Eq (16) and defining $\beta_{O_2} = k_{17}[NO] + k_{20}[N]$

$$d[O_2^+] / dt = - \left\{ k_{10} [A^+] + (n_o - [A^+]) / (1 + \alpha_r' n_o t) \right\} + \beta_{O_2} \left\{ [O_2^+] \right\} \quad (29)$$

Treating Eq (29) in the same manner as Eq (20), the final instantaneous solution for $[O_2^+]$ is

$$[O_2^+] = [O_2^+]_0 \exp \left\{ -\beta_{O_2} t - k_{10} I_{A^+} + k_{10} / (\alpha_r' n_o) [([O_2^+]_0 - y) e^{A_1} (E_1(A_1) - E_1(B_1)) \right\}$$

$$+ ([N^+]_0 + y)e^{A_2(E_1(A_2) - E_1(B_2))} \left\{ / (1 + \alpha'_r n_o t) \right\}^{k_{10}/\alpha'_r} \quad (30)$$

$$\text{where } I_{A^+} = I_{O^+} + I_{N^+} = (([O^+]_0 - y)/\beta_O)(1 - e^{-\beta_O t}) + (([N^+]_0 + y)/\beta_N)(1 - e^{-\beta_N t});$$

A_1 , B_1 , A_2 , and B_2 have been defined; and $[O_2^+]_0$ includes $X_{O_2^+}$.

The above treatment has been modified to take into account some of the errors introduced by the approximations made. Since $[O_2^+]_0$ is increased by an amount $X_{O_2^+}$ at $t = 0$ allowing that additional concentration to recombine from $t = 0$ and not from the actual time of creation, Eq (27) is modified to correct for the recombination of N_2^+ by the factor

$$C_{N_2} = [N_2^+]_0 / \left\{ [N_2^+] + (k_8[O_2] + k_9[O] + k_{18}[NO])I_{N_2^+} \right\} \quad (31)$$

where the latter term is the N_2^+ lost to all processes except recombination.

Equation (27) then has the form

$$X_{O_2^+} = [O_2] \left\{ k_8 C_{N_2} + I_{N_2^+} + k_{11}I_{O^+} + k_{13}I_{N^+} \right\} \quad (32)$$

Equation (30) has also been altered to improve agreement with numerically calculated solutions. If Eq (30) is rewritten as

$$[O_2^+] = [O_2^+]_0 \exp \left\{ -\beta_{O_2} t - k_{10}(I_{A^+} - I) \right\} / (1 + \alpha'_r n_o t)^{k_{10}/\alpha'_r} \quad (33)$$

the modification to Eq (33) is the denominator transformation

$$(1 + \alpha'_r n_o t)^{k_{10}/\alpha'_r} \rightarrow (1 + \alpha'_r n_o t (\exp(-k_{10}(I_{A^+} - I))))^{k_{10}/\alpha'_r}$$

The continuous source calculation for O_2^+ is handled in a similar manner to N_2^+ . The full differential equation for O_2^+

$$d[O_2^+]/dt = Q_{O_2^+} + \{k_8[N_2^+] + k_{11}[O^+] + k_{13}[N^+]\} [O_2] - \{k_{10}n_e + \beta_{O_2}\} [O_2^+] \quad (34)$$

can be regrouped into the familiar form

$$d[O_2^+]/dt = Q - \beta[O_2^+] \quad (35)$$

$$\text{where } Q = Q_{O_2^+} + \{k_8[N_2^+] + k_{11}[O^+] + k_{13}[N^+]\} [O_2]$$

and $\beta = k_{10}n_e + \beta_{O_2}$. Eq (35) is solved in the usual way using FUNCTION WQX with $[O_2^+]_0 = 0$.

The final concentration of O_2^+ at t_c is considered to be the maximum of the instantaneous and continuous source solutions.

6. NITRIC OXIDE ION (NO^+)

The final species in the charge exchange chain, NO^+ , is involved in Reactions 9, 12, 14, 15, 16, 17, 18, 19, and 20. All these reactions except 12 may be viewed as increasing $[NO^+]$ at $t = 0$ by an amount

$$X_{NO^+} = k_9[O]I_{N_2^+} + k_{14}[O_2]I_{N^+} + k_{15}[N_2]I_{O^+} + k_{16}[NO]I_{O^+} + k_{17}[NO]I_{O_2^+} + k_{18}[NO]I_{N_2^+} + k_{19}[NO]I_{N^+} + k_{20}[N]I_{O_2^+} \quad (36)$$

where $I_{O_2^+} = \int_0^t [O_2^+] dt'$. In analogy to the calculation of $I_{N_2^+}$,

an integrable form is assumed for $[O_2^+]$

$$[O_2^+] = [O_2^+]_0 e^{-\beta_e^{O_2} t} / (1 + \alpha_r n_o t) \quad (37)$$

where $\beta_e^{O_2}$ is calculated by setting Eq (37) equal to the results of Eq (30)

Thus

$$I_{O_2}^+ = [O_2^+]_0 e^{A_4 / (\alpha_r n_o)} (E_1(A_4) - E_1(B_4))$$

$$\text{where } A_4 = \beta_e^2 / (\alpha_r n_o) \text{ and } B_4 = A_4 (1 + \alpha_r n_o t). \quad (38)$$

This approach leaves the differential equation for the instantaneous solutions of $[NO^+]$ in the form

$$d[NO^+] / dt = -k_{12} n_e [NO^+] \quad (39)$$

Substituting Eq (16) for n_e and assuming α_r' a constant, the equation is integrated to give the solution

$$[NO^+] = [NO]_0^+ \exp[-k_{12} I_{A^+} + k_{12} / (\alpha_r' n_o) \{ ([O^+]_0 - y) e^{A_1} (E_1(A_1) - E_1(B_1)) + ([N^+]_0 + y) e^{A_2} (E_1(A_2) - E_1(B_2)) \}] / (1 + \alpha_r' n_o t)^{k_{12} / \alpha_r'} \quad (40)$$

where A_1 , B_1 , A_2 , and B_2 have been defined previously, and $[NO^+]_0$ includes X_{NO^+} .

Again, as for $[O_2^+]$, Eq (36) and Eq (40) must be modified to correct approximation errors. In Eq (36), all terms involving $I_{N_2^+}$ are multiplied by the factor C_{N_2} and all involving $I_{O_2^+}$, by the factor

$$C_{O_2} = [O_2^+]_0 / ([O_2^+] + (k_{17}[NO] + k_{20}[N]) I_{O_2^+}) \quad (41)$$

to correct for the fact that these concentrations did not recombine from $t = 0$.

Eq (40) is not only modified but is also passed through an iteration process along with the continuous source solution. First, if Eq (40) is rewritten as

$$[\text{NO}^+] = [\text{NO}^+]_0 \exp \left\{ -k_{12} (I_{A^+} - I) \right\} / (1 + \alpha'_r n_o t)^{k_{12}/\alpha'_r} \quad (42)$$

the denominator is modified to give the first iteration instantaneous solution form

$$[\text{NO}^+] = [\text{NO}^+]_0 \exp \left\{ -k_{12} (I_{A^+} - I) \right\} / (1 + \alpha'_r n_o t (\exp(-k_{12} (I_{A^+} - I)))) \quad (43)$$

The first iteration continuous source solution solves the full differential equation for $[\text{NO}^+]$ which is analogous to Eq (34) and may be reduced to

$$d[\text{NO}^+]/dt = Q - \beta[\text{NO}^+] \quad (44)$$

which is solved using FUNCTION WQX with $[\text{NO}^+]_0 = 0$ and $\beta = k_{12} n_e$.

The maximum of the instantaneous and continuous source solutions of Eq (43) and Eq (44) constitute the first iteration solution of $[\text{NO}^+]$.

Having calculated concentrations at t_c for all the positive molecular ions, a new α_r and α'_r are calculated and the formalism for the calculation of n_e in Section IV-3 is repeated to update the electron (i.e. negative ion) concentration at t_c .

The second iteration on $[\text{NO}^+]$ then uses the new α'_r to recalculate the instantaneous solution of Eq (42) modified as

$$[\text{NO}^+] = [\text{NO}^+]_0 \exp \left\{ -k_{12} (I_{A^+} + I) \right\} / (1 + \alpha'_r n_o t (\exp(-k_{12} (I_{A^+} - I))))^{k_{12}/\alpha'_r} \quad (45)$$

with I not being recalculated. Likewise, the continuous solution is recalculated using the new n_e . Finally, the maximum of the two solutions is taken to give the ultimate $[\text{NO}^+]$ concentration at t_c .

7. SPECIES AND TEMPERATURE UPDATING

Throughout the above treatment all neutral species concentrations were assumed to be constant. This approximation alone is insufficient, however, and intermediate updating of some neutral species concentrations is accomplished between the solutions for $[N_2^+]$ and $[O_2^+]$ (i.e., Sections IV-4 and IV-5). Likewise, the neutral species concentrations and the temperature are ultimately modified after the entire treatment to account for production by the fast ion chemistry.

a. Intermediate Updating of NO, N, and O

Intermediate updating is accomplished for $[N]$, $[O]$, and $[NO]$. $[N]$ and $[O]$ created or destroyed by an ion-neutral reaction are accounted for by additions to their initial concentrations, $[N]_0$ or $[O]_0$

$$[Y]_0 = [Y]_0 \pm bk[X]I_{m+} \quad (46)$$

where I_{m+} is either I_{N+} , I_{O+} , or I_{N_2+} , b is a branching ratio, k is the reaction rate constant, and X is the reacting neutral whose concentration is assumed to be constant. The increase in $[N]$ due to the dissociative recombination of N_2^+ additionally requires the more complex integral $k_7 \int_0^t n_e [N_2^+] dt'$ where neither term of the product can be assumed to be constant. This integral is calculated by substituting Eq (16) and Eq (25) for n_e and $[N_2^+]$ respectively and solving.

$$I_{N_2+}^{n_e} = \int_0^t [N_2^+] n_e dt' = \int_0^t \frac{[N_2^+]_0 e^{-\beta_e^{N_2} t'}}{1 + \alpha_r n_0 t'} \left\{ [A^+] + \frac{n_0 - [A^+]}{1 + \alpha_r' n_0 t'} \right\} dt' \quad (47)$$

$$I_{N_2+}^{n_e} = [N_2^+]_0 \left\{ \int_0^t \frac{[A^+] e^{-\beta_e^{N_2} t'}}{1 + \alpha_r n_0 t'} dt' + \int_0^t \frac{(n_0 - [A^+]) e^{-\beta_e^{N_2} t'}}{(1 + \alpha_r n_0 t')(1 + \alpha_r' n_0 t')} dt' \right\} \quad (48)$$

The denominator in the second integral is approximated by $(1 + \gamma n_o t')^2$

where the constant $\gamma = \sqrt{\alpha_r \alpha_r'}$. Integrating yields

$$\begin{aligned}
 I_{N_2}^{n_e+} = [N_2^+]_o & \left\{ \frac{([O^+]_o - y)}{\alpha_r n_o} e^{A_5(E_1(A_5) - E_1(B_5))} + \frac{([N^+]_o + y)}{\alpha_r n_o} e^{A_6(E_1(A_6) - E_1(B_6))} \right. \\
 & + \frac{1}{\gamma} \left[\left(1 - \frac{e^{-\beta_e N_2 t}}{1 + \gamma n_o t}\right) - A_7 e^{A_7(E_1(A_7) - E_1(B_7))} \right] - \frac{([O^+]_o - y)}{\gamma n_o} \left[\left(1 - \frac{e^{-\delta t}}{1 + \gamma n_o t}\right) \right. \\
 & \left. \left. - A_8 e^{A_8(E_1(A_8) - E_1(B_8))} \right] - \frac{([N^+]_o + y)}{\gamma n_o} \left[\left(1 - \frac{e^{-\sigma t}}{1 + \gamma n_o t}\right) - A_9 e^{A_9(E_1(A_9) - E_1(B_9))} \right] \right\} \quad (49)
 \end{aligned}$$

where $\delta = \beta_e N_2 + \beta_o$, $\sigma = \beta_e N_2 + \beta_N$, $A_5 = \frac{\delta}{\alpha_r n_o}$, $B_5 = A_5(1 + \alpha_r n_o t)$, $A_6 = \frac{\sigma}{\alpha_r n_o}$

$B_6 = A_6(1 + \alpha_r n_o t)$, $A_7 = \frac{\beta_e N_2}{\gamma n_o}$, $B_7 = A_7(1 + \gamma n_o t)$, $A_8 = \frac{\delta}{\gamma n_o}$, $B_8 = A_8(1 + \gamma n_o t)$,

$A_9 = \frac{\sigma}{\gamma n_o}$, and $B_9 = A_9(1 + \gamma n_o t)$.

Having calculated $I_{N_2}^{n_e}$, $[N]$ can be updated using the recombination

term $k_7 I_{N_2}^{n_e+}$. However, since several approximations were made in the calculation, a further modification is necessary. The total amount of N_2^+ that has been removed by the several reactions in which it participates is

$$\Delta N_2^+ = [N_2^+]_o - [N_2^+]_t \quad (50)$$

which if no approximations had been made in calculating $I_{N_2}^{n_e+}$ and $I_{N_2}^{n_e}$ would be equivalent to

$$A_{N_2} = (k_8[O_2] + k_9[O] + k_{18}[NO])I_{N_2}^{n_e+} + k_7 I_{N_2}^{n_e+} \quad (51)$$

Since the two terms will not be equal, the concentration of N_2^+ which has recombined is calculated as the total change in $[N_2^+]$ multiplied by the ratio of that which has recombined to the total that has reacted. Thus the concentration of N_2^+ which has recombined is

$$R_{N_2^+} = \Delta N_2^+ (k_7 I_{N_2^+}^n / A_{N_2^+}) \quad (52)$$

which is used along with the appropriate branching ratios to update the N concentrations.

The intermediate updating of [NO] is accomplished in a very approximate fashion. Reactions 1, 2, 3, 4, 24, 25, 26, 28, and 31 involve [NO], but only the first four are considered, resulting in the differential equation

$$\frac{d[NO]}{dt} = [NSP](k_1[O_2] - k_3[NO]) + [N(^2D)](k_2[O_2] - k_4[NO]) \quad (53)$$

where $[NSP] = [N(^4S)] + [N(^2P)]$. The number of variables in Eq (53) can be reduced by assuming that an increase in [NO] is reflected as a decrease in $[O_2]$ such that

$$[O_2] = [O_2]_0 - ([NO] - [NO]_0) \quad (54)$$

Substituting into Eq (53)

$$\begin{aligned} \frac{d[NO]}{dt} = [NSP] \{ k_1([O_2]_0 + [NO]_0) - (k_1 + k_3)[NO] \} + [N(^2D)] \{ k_2([O_2]_0 + [NO]_0) \\ - (k_2 + k_4)[NO] \} \end{aligned} \quad (55)$$

If reactions involving $N(^2D)$ are more rapid than those of $N(^4S)$ and $N(^2P)$, the two halves of Eq (55) may be solved sequentially.

First, the solution for $[N(^2D)]$ is assumed to have the form

$$[N(^2D)] = [N(^2D)]_0 e^{-\beta_1 t} \quad (56)$$

where $\beta_1 = k_2[O_2]_0 + k_4[NO]_0$. And the solution to the second half of Eq (55) is

$$[NO]_1 = k_2([O_2]_0 + [NO]_0)/(k_2+k_4) - \left\{ k_2([O_2]_0 + [NO]_0)/(k_2+k_4) - [NO]_0 \right\} \exp \left\{ -(k_2+k_4)[N(^2D)]_0 (1-e^{-\beta_1 t})/\beta_1 \right\} \quad (57)$$

The result of Eq (57) is then used as the initial condition for $[NO]$ in the solution to the first half of Eq (55). Thus, the intermediate updated concentration for NO is

$$[NO] = k_1([O_2]_0 + [NO]_0)/(k_1+k_3) - \left\{ k_1([O_2]_0 + [NO]_0)/(k_1+k_3) - [NO]_1 \right\} \exp \left\{ -(k_1+k_3)[NSP]_0 (1-e^{-\beta_2 t})/\beta_2 \right\} \quad (58)$$

where $\beta_2 = k_1[O_2]_0 + k_3[NO]_1$ and the time dependence of $[NSP]$ has the form of Eq (56).

b. FINAL UPDATING OF N AND O

After completion of the fast ion chemistry, the concentrations of N and O are updated to account for production by the reactions of O_2^+ and NO^+ . Procedures similar to those for the intermediate updating are followed. In particular for the recombination of O_2^+ , the integral is needed

$$I_{O_2^+}^{n_e} = \int_0^t [O_2^+] n_e dt' = [O_2^+]_0 \left\{ \int_0^t \frac{[A^+]_e e^{-\beta_{e, O_2^+} t'}}{(1 + \alpha_{r, O_2^+} t')^2} dt' + \int_0^t \frac{(n_e - [A^+]_e) e^{-\beta_{e, O_2^+} t'}}{(1 + \gamma_{n_e} t')^2} dt' \right\} \quad (50)$$

which can be solved in analogy to N_2^+ in Eq (48) to give the solution of Eq (49) where $[O_2^+]_0$ is substituted for $[N_2^+]_0$ and β_{e, O_2^+} for β_{e, N_2^+} . A similar integral for $[NO^+]$, however is not required since the only loss mechanism for NO^+ is recombination and the integral is simply

$$k_{12} \int_0^t n_e [NO^+] dt' = \Delta NO^+ = [NO^+]_0 - [NO^+]_t \quad (60)$$

Again in analogy to the N_2^+ , the recombination integral for O_2^+ is corrected for the approximations made throughout the calculation and the final concentration of O_2^+ which has recombined is calculated as

$$R_{O_2^+} = \Delta O_2^+ (k_{10} I_{O_2^+}^{n_e} / A_{O_2^+}) ([O_2^+]_0' / [O_2^+]_0) \quad (61)$$

where a second correction term $([O_2^+]_0' / [O_2^+]_0)$ has been added to account for the effects of $[O_2^+]_0$ having been increased by $X_{O_2^+}$, Eq (32), and the approximations made there. Similarly, the final concentration of NO^+ which has recombined is calculated as

$$R_{NO^+} = \Delta NO^+ ([NO^+]_0' / [NO^+]_0) \quad (62)$$

where the correction term has been added to account for the approximations made in updating $[NO^+]_0$ with X_{NO^+} .

The branching ratios of Reactions 10 and 12 are used with $R_{O_2^+}$ and R_{NO^+} respectively, to update the N and O concentrations.

c. FINAL UPDATING OF TEMPERATURE AND REACTION RATES

The final step in the fast ion chemistry treatment is the recomputation of the gas temperature and reaction rates. For the temperature calculation, the specific heat of the sample volume is recalculated as is the total heat of formation for all species in the volume. The internal energy of the volume is then recalculated as its former value plus the change in the total heat of formation for all species. This new internal energy is finally converted to an updated gas temperature using the new specific heat. The new temperature is then used in RATE to recalculate the reaction rates for all reactions of Table II.

8. NEUTRAL SPECIES CHEMISTRY

The slower neutral chemistry is much simpler in its mathematics, but is intricate in the ordering of species and reactions for solution. The subsequent discussion, therefore, follows the programmed logic rather than segmentation by species.

In general, the treatment follows four steps. In the first, the continuous source production rates are neglected and $[O]$, $[O_2(^3\Sigma)]$, $[N]$, $[NO]$, and $[O_3]$ are addressed. In the second the continuous source rates are included in the solutions for most species. The reactions involving $[OH]$ and $[H]$ are treated separately in the third. And the metal/metal oxide chemistry is handled in an uncoupled manner in the fourth.

a. Decay of Oxygen Excited States

Reactions 33 and 34 involve the decay of $[O(^1S)]$ and $[O(^1D)]$.

The $O(^1S)$ decay is simply expressed as

$$[O(^1S)] = [O(^1S)]_0 e^{-k_{33}t} \quad (63)$$

and the differential equation for $O(^1D)$

$$\frac{d[O(^1D)]}{dt} = -k_{34}[O(^1D)] - d[O(^1S)]/dt \quad (64)$$

has the solution

$$[O(^1D)] = \left\{ [O(^1D)]_0 - k_{33}[O(^1S)]_0 / (k_{34} - k_{33}) \right\} e^{-k_{34}t} + k_{33}[O(^1S)]_0 / (k_{34} - k_{33}) e^{-k_{33}t} \quad (65)$$

where the initial values $[O(^1D)]_0$ and $[O(^1S)]_0$ are those updated at the end of the charged species treatment. At the same time, $[O_2(^3\Sigma)]$ is decayed according to Reaction 23

$$[O_2(^3\Sigma)] = [O_2(^3\Sigma)]_0 e^{-k_{23}[O_2]t} \quad (66)$$

b. Nitrogen-Nitric Oxide

The interplay of $[N]$ and $[NO]$ is handled by dividing the solution for $[N]$ into two parts separated by an updating of $[NO]$.

The first set of nitrogen Reactions 1, 2, and 32 are faster and do not involve $[NO]$. $N(^2P)$ decays in a simple exponential manner according to Reactions 1 and 32, in analogy to $O(^1S)$ in Eq (63). The differential equation for $N(^2D)$, derived from Reactions 2 and 32, is solved as for $O(^1D)$ and the solution is of the same form as Eq (65). And $N(^4S)$ is exponentially decayed according to Reaction 1.

For the updating of $[NO]$, the pertinent reactions of Table II are divided into two groups: Reactions 1, 2, 3, 4 and Reactions 24, 25, 26, 28, and 31. The solution to the first group of four reactions is found in Eq (57) and Eq (58) as used in Section IV-7. The second group of five are

divided into a formation term, $F = k_{26} [O][NO_2] + k_{31}[N(^4S)][O_3]$, and a destruction term, $D = k_{24}[O_3] + k_{25}[O] + k_{28}[N]$, for $[NO]$ and FUNCTION WQX is used to evaluate $[NO]_t$. Once both solutions have been found, the maximum of the two is adopted as the final value of $[NO]$ at t_c .

With $[NO]$ updated, the second set of nitrogen Reactions 3, 4, 27, 28, 29, and 31 are considered. Each atomic state of nitrogen suffers a simple exponential decay with exponents derived from: Reactions 3, 27, 28, and 29 for $N(^2P)$, Reactions 4, 27, 28, and 29 for $N(^2D)$, and Reactions 3, 27, 28, 29, and 31 for $N(^4S)$. The total change in $[N]$ over the two part treatment is calculated and added to $[O]$ since the major reactions in the nitrogen treatment involve production of atomic oxygen.

C. Ozone and Atomic Oxygen

Ozone is involved in Reactions 5, 6, 21, 23, 24, 30, 31, and photo dissociation. Neglecting Reactions 21 and 30, the differential equation for $[O_3]$ may be written

$$\begin{aligned} d[O_3]/dt = & - [O_3] \left\{ k_6[O] + k_{24}[NO] + k_{31}[N(^4S)] + \phi \right\} + k_5[O_2][O] \\ & + k_{23}[O_2][O_2(^3\Sigma)] \end{aligned} \quad (67)$$

where ϕ is the photodissociation rate constant for ozone and Eq(66) can be substituted for $[O_2(^3\Sigma)]$. The value of $[O]$ in the second term is replaced by $[O]e^{-k_5[O_2]t}$ and Eq (67) is solved to give the final value for $[O_3]$

$$\begin{aligned} [O_3] = & \left\{ [O_3]_0 + \gamma[O_2(^3\Sigma)]_0/(\gamma-\beta) + \delta[O]/(\delta-\beta) \right\} e^{-\beta t} - \left\{ \gamma[O_2(^3\Sigma)]_0/(\gamma-\beta) \right\} e^{-\gamma t} \\ & - \left\{ \delta[O]/(\delta-\beta) \right\} e^{-\delta t} \end{aligned} \quad (68)$$

where $\beta = k_6[O] + k_{24}[NO] + k_{31}[N(^4S)] + \phi$, $\gamma = k_{23}[O_2]$, and $\delta = k_5[O_2]$.

With Reactions 33 and 34 already addressed, atomic oxygen is involved in Reactions 5, 6, 23, 25, 26, and 29 for which the differential equation is

$$\frac{d[O]}{dt} = - [O] \left\{ k_5[O_2] + k_6[O_3] + k_{25}[NO] + k_{26}[NO_2] \right\}$$

$$+ \left\{ k_{29}[NO_2][N] + \phi \right\} + k_{23}[O_2][O_2(^3\Sigma)] \quad (69)$$

If $[O_2(^3\Sigma)]$ is replaced by Eq (66), the solution to Eq (69) using the updated value for $[O_3]$ is

$$[O] = \left\{ [O]_0 + \gamma [O_2(^3\Sigma)]_0 / (\gamma - \beta) \right\} e^{-\beta t} + \delta / \beta (1 - e^{-\beta t}) - \left\{ \gamma [O_2(^3\Sigma)]_0 / (\gamma - \beta) \right\} e^{-\gamma t} \quad (70)$$

where $\beta = k_5[O_2] + k_6[O_3] + k_{25}[NO] + k_{26}[NO_2]$, $\gamma = k_{23}[O_2]$, and $\delta = k_{29}[NO_2][N] + \phi$. The above solution is applied to $[O(^3P)]$ and the values of $[O(^1S)]$ and $[O(^1D)]$ are adjusted in proportion to the resultant change in $[O(^3P)]$.

d. Inclusion of Continuous Source Production

The second step of the neutral chemistry includes continuous energy deposition contributions in the species time dependent solutions. The species and their reactions from step one are addressed again here. For each species, formation rates (sec^{-1}) and destruction rates ($\text{cc}^{-1}\text{sec}^{-1}$) are calculated using the continuous source rates from ENQDEP and the appropriate reaction rate constants and current species concentration values. Then, in sequence, the solutions for $O(^3P)$, $O(^1D)$, $O(^1S)$, $N(^4S)$, $N(^2D)$, $N(^2P)$, $O_2(^3\Sigma)$, and O_3 are systematically calculated using FUNCTION WQX, with the calculational time, t_c , and concentrations at $t = 0$ assumed to be zero. The final value for each species concentration at t_c is computed as the sum of the solutions from the first and second steps of the neutral species chemistry.

e. NO_2 , N_2O , and $\text{O}_2(^1\Delta)$

The NO_2 and N_2O concentrations are calculated using an approach to equilibrium with the FUNCTION WQX. NO_2 is formed in Reactions 24 and 25 and destroyed in Reaction 26, 27, and 29. N_2O is formed in Reactions 28 and 29 and destroyed in Reaction 30. Appropriate formation and destruction terms are computed for each species and FUNCTION WQX is applied to calculate their concentrations at t_c .

Having updated $[\text{N}_2\text{O}]$, $\text{O}_2(^1\Delta)$ is now addressed. It's only source of production is from the continuous energy deposition and its only form of destruction is by way of Reaction 30. FUNCTION WQX is utilized twice to solve for $[\text{O}_2(^1\Delta)]$ and the final value of the concentration is taken as the sum of the two results. In the first, both production and destruction terms are considered and $[\text{O}_2(^1\Delta)]_0$ is set equal to zero; and in the second, the production term is deleted and $[\text{O}_2(^1\Delta)]_0$ is its initial value.

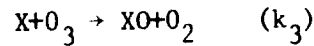
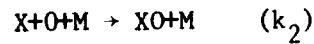
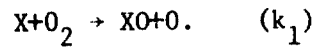
f. Hydrogen

The final neutral system to be addressed before the metal oxides is the H, OH equilibrium of Reactions 21 and 22. Again the FUNCTION WQX is employed in an ad hoc fashion to solve this cyclic set of equations. In this case, the value of $[\text{OH}]$ at t_c , is calculated using the production term $k_{21}[\text{O}_3] \{ [\text{H}] + [\text{OH}] \}$, the destruction term $k_{21}[\text{O}_3] + k_{22}[\text{O}(^3\text{P})]$, and the initial value $[\text{OH}]_0$. The value of $[\text{H}]$ is then taken as that free hydrogen which has not been converted to OH.

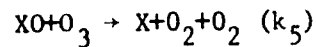
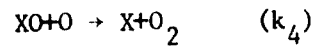
g. Metal Oxides

The metal and metal oxide reactions are uncoupled from the rest of the analytical algorithm. The light metals such as Li, Be, Mg, Al, Si, Ti, Fe, and Cu, presently carried in WORRY, are treated in the same manner; therefore, in the following discussion the symbol X will be used to describe

any one of these eight metals. The reactions and rate constants for the formation of the metal oxides are



The reactions and rate constants for the destruction of the metal oxides are



The differential equation for $[XO]$ is

$$\frac{d[XO]}{dt} = \left\{ k_1[O_2] + k_2[O][M] + k_3[O_3] \right\} [X] - [XO] \left\{ k_4[O] + k_5[O_3] \right\} \quad (71)$$

But the total number of metal atoms does not change

$$[X] + [XO] = A \quad (72)$$

and A is a constant, whose value is $[X]_0$ if all metal atoms are unoxidized at $t = 0$.

Thus Eq (71) can be rewritten as

$$\frac{d[XO]}{dt} = F(A-[XO]) - D[XO] \quad (73)$$

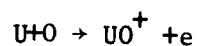
where $F = k_1[O_2] + k_2[O][M] + k_3[O_3]$ and $D = k_4[O] + k_5[O_3]$. F and D are computed for each species using the final values of $[O]$, $[O_2]$, and $[O_3]$, which are assumed to be constant.

Reordering Eq (73) gives

$$\frac{d[XO]}{dt} = FA - (F+D)[XO] \quad (74)$$

which may be solved for $[XO]$ at t_c using FUNCTION WQX with a formation term of FA, a destruction term of $F + D$, and $[XO]_0$.

Because of the unique chemistry of U, only the following reaction is treated and in a simple exponential fashion:



With all species concentrations now calculated and stored in the CMD array, the analytic algorithm is complete and new formation and destruction rates for all species are calculated where needed, to be used in subroutine RRCRT.

SECTION V

ACCESSORIAL SUBROUTINES AND FUNCTIONS

1. ROUTINES SHARED BY BOTH ALGORITHMS

a. ENTDEP and ENQDEP

The routine ENTDEP calculates the initial distribution of ions and neutrals produced at $t = 0$ by the prompt energy deposition, while its entry point ENQDEP partitions the continuous energy deposition (ion pairs/cc/sec) among the same set of species. ENTDEP also calculates the specific heat of the sample volume, the internal kinetic energy of the gas, and adjusts the gas temperature accordingly.

ENTDEP requires inputs of the gas temperature, the total density, and the total number of ion pairs produced by the prompt energy deposition. The latter two are converted to a fractional ionization and ionized fractions of $[N_2]$, $[O_2]$, and $[O]$ are calculated. Then the concentration for each of the following species is adjusted for its production or depletion by ionization of N_2 , O_2 , and O : N_2 , O_2 , $O(^3P)$, $O(^1D)$, $O(^1S)$, $N(^4S)$, $N(^2D)$, $N(^2P)$, $O_2(^1\Delta)$, $O_2(^3\Sigma)$, N^+ , O^+ , N_2^+ , O_2^+ , and e^- .

ENQDEP likewise calculates the continuous ionization rates of N_2 , O_2 , and O and converts these to formation and destruction rates for the same set of product species.

b. RATE

The RATE routine simply calculates the rate constants for the reactions of table II and stores them in the R array. RATE requires inputs of the gas temperature ($^{\circ}K$) and the total density (cc^{-1}). All three body reactions are provided with pseudo-second-order rate constants by including $[M]$ in the constant.

c. RRCRT

This routine stores the necessary species destruction rates in RCRT and transforms the species formation rates into vibrational level formation rates which are stored in the RR array and used by the WORRY calculation of chemiluminescence. Because the product vibrational level branching ratios for the various reactions are not known, equal partition to the highest accessible level is presently assumed.

2. FUNCTIONS OF THE ANALYTIC ALGORITHM

a. FUNCTION WQX (A, B, XO, T)

This function solves the differential equation

$$\frac{d[X]}{dt} = A - B[X] \quad (75)$$

to give $[x]$ at time, T , with $[x]_0 = X0$.

b. FUNCTION WR(A, B, XO, T)

This function solves the differential equation

$$\frac{d[x]}{dt} = A - B[x]^2 \quad (76)$$

to give $[x]$ at time, T , with $[x]_0 = X0$.

c. FUNCTION SHXOX (X)

This function evaluates the expression

$$\frac{e^x - 1}{x}$$

d. FUNCTION XLI GL (A, B, C, T)

This function evaluates the integral

$$\int_0^T \frac{e^{-Bt} dt}{(1 + ACt)}$$

e. FUNCTION X2IGL (A, B, C, T)

This function evaluates the integral

$$\int_0^T \frac{e^{-Bt} dt}{(1 + ACt)^2}$$

i. FUNCTION E1F (X)

This function evaluates the integral

$$\int_X^\infty \frac{e^{-x}}{x} dx$$

SECTION VI
COMPARISONS OF ALGORITHMS

Figures 2, 3, 4, and 5 demonstrate the agreement between the analytic and numerical algorithms. A large number of such comparisons were presented at the DNA 1973 Atmospheric Effects Symposium at San Diego (ref 1). The problem definitions for figures 2, 3, and 4 were taken from Ory and Gilmore (ref 2), while that for figure 5 was taken from Scheibe (ref 3).

100KM NIGHT SPECIES NUMBER DENSITY VS. TIME

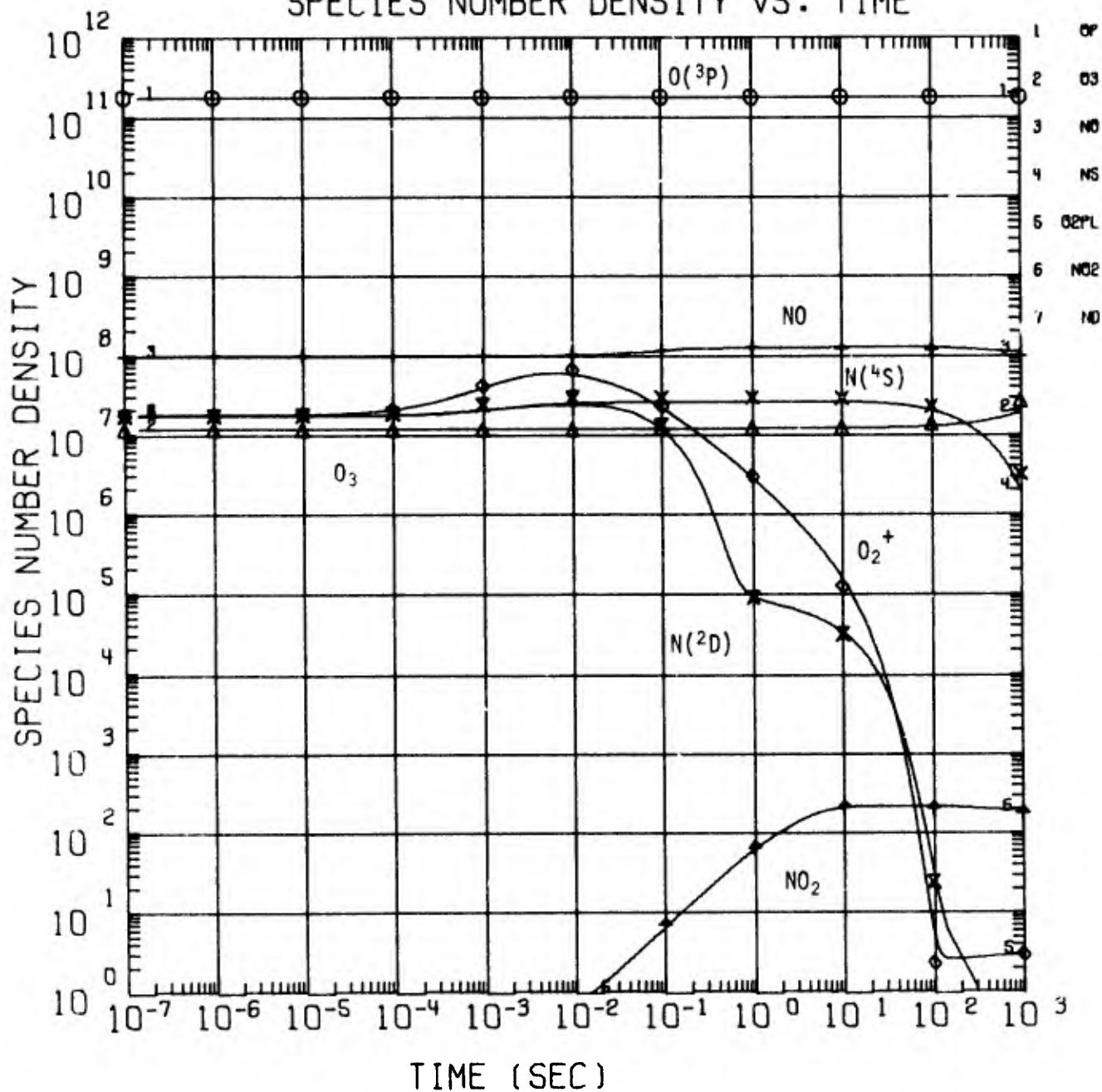


Figure 2. Comparison of Numerical (—) and Analytic (symbols) Algorithms (Altitude of 100 km, energy deposition of 10^8 ion pairs/cc during the night).

60KM NIGHT SPECIES NUMBER DENSITY VS. TIME

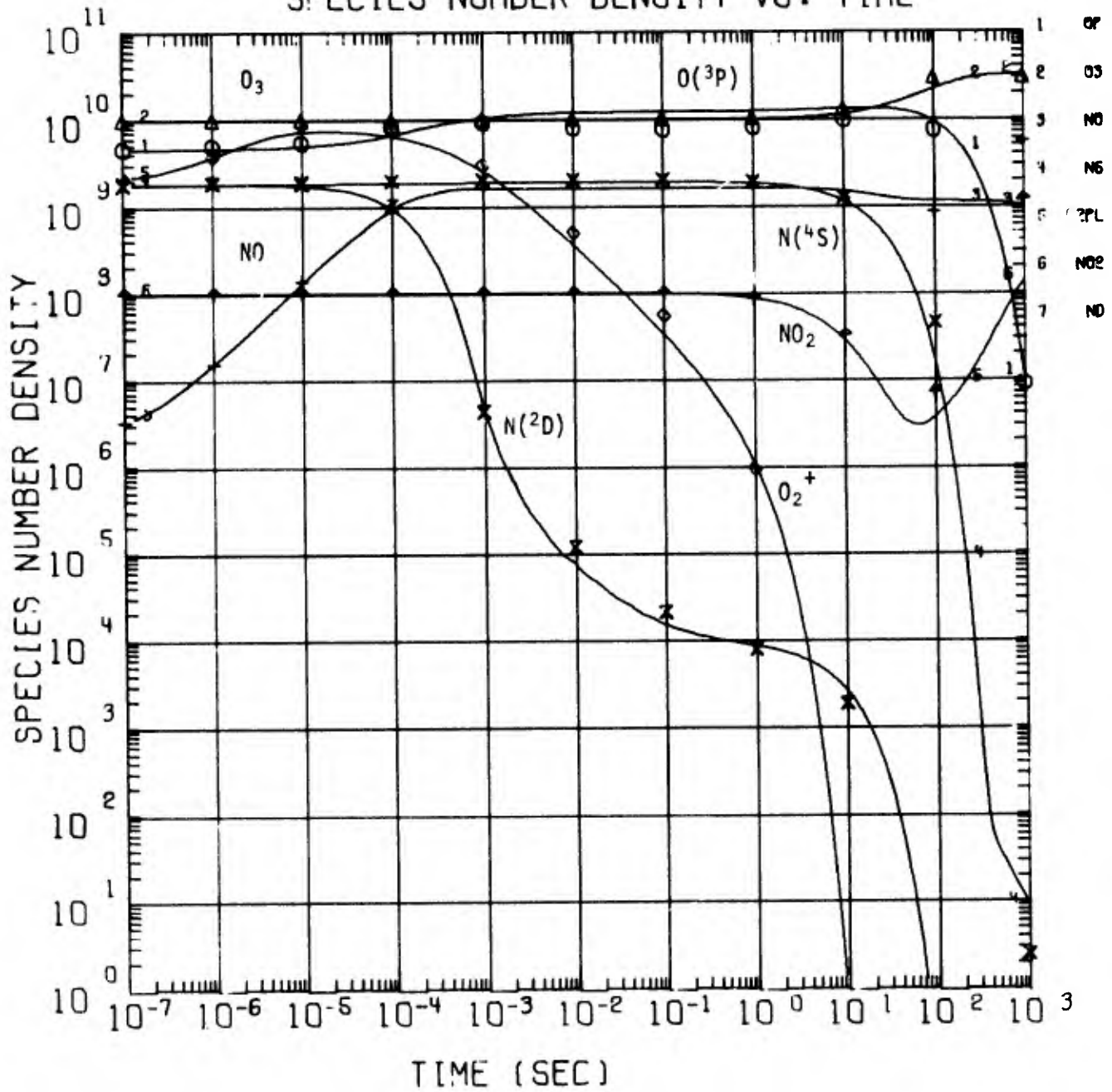


Figure 3. Comparison of Numerical and Analytic Algorithms (Altitude of 60 km, energy deposition of 10^{10} ion pairs/cc during the night).

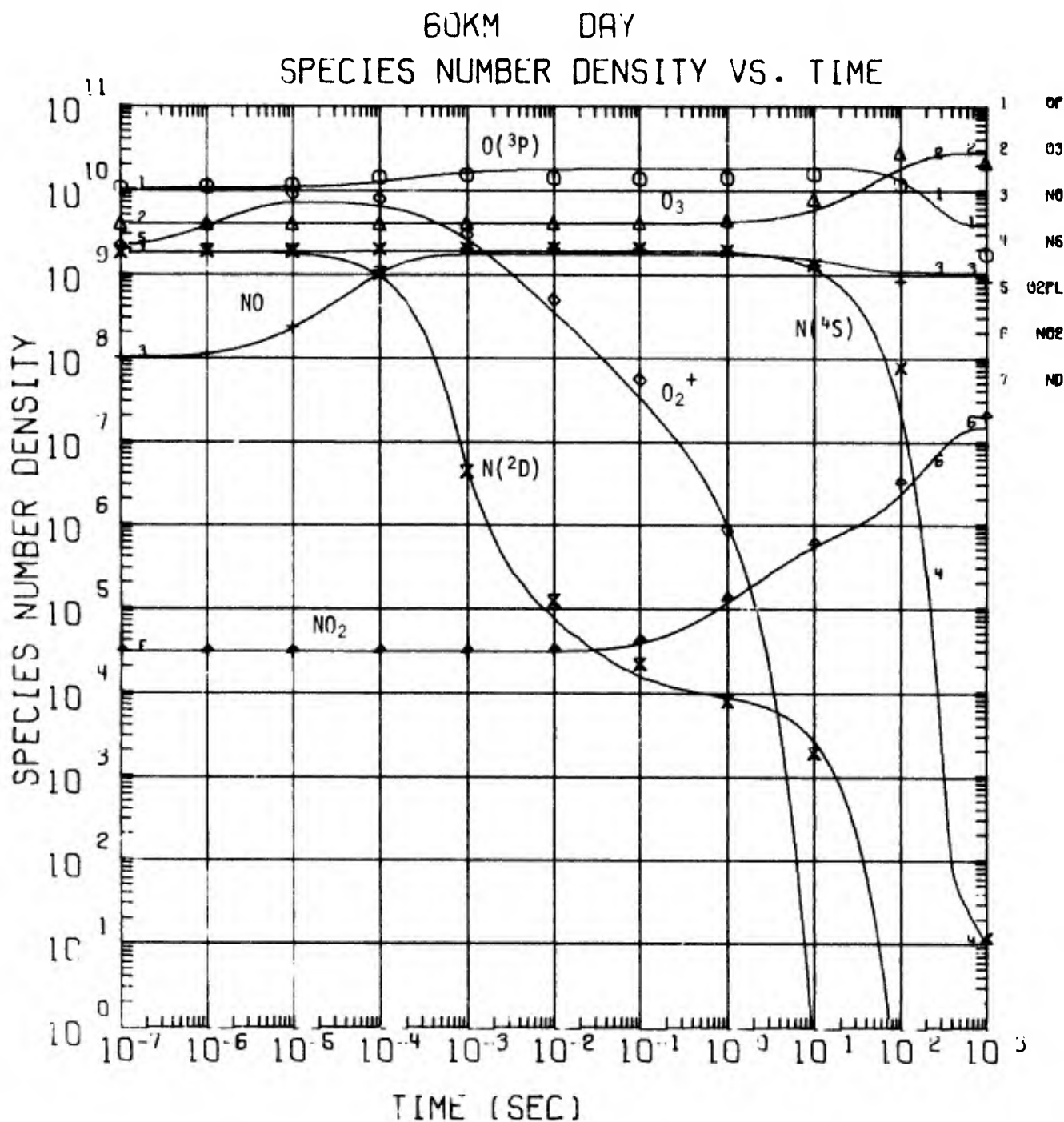


Figure 4. Comparison of Numerical and Analytic Algorithms (Altitude of 60 km, energy deposition of 10^{10} ion pairs/cc during the day).

40KM NIGHT SPECIES NUMBER DENSITY VS. TIME

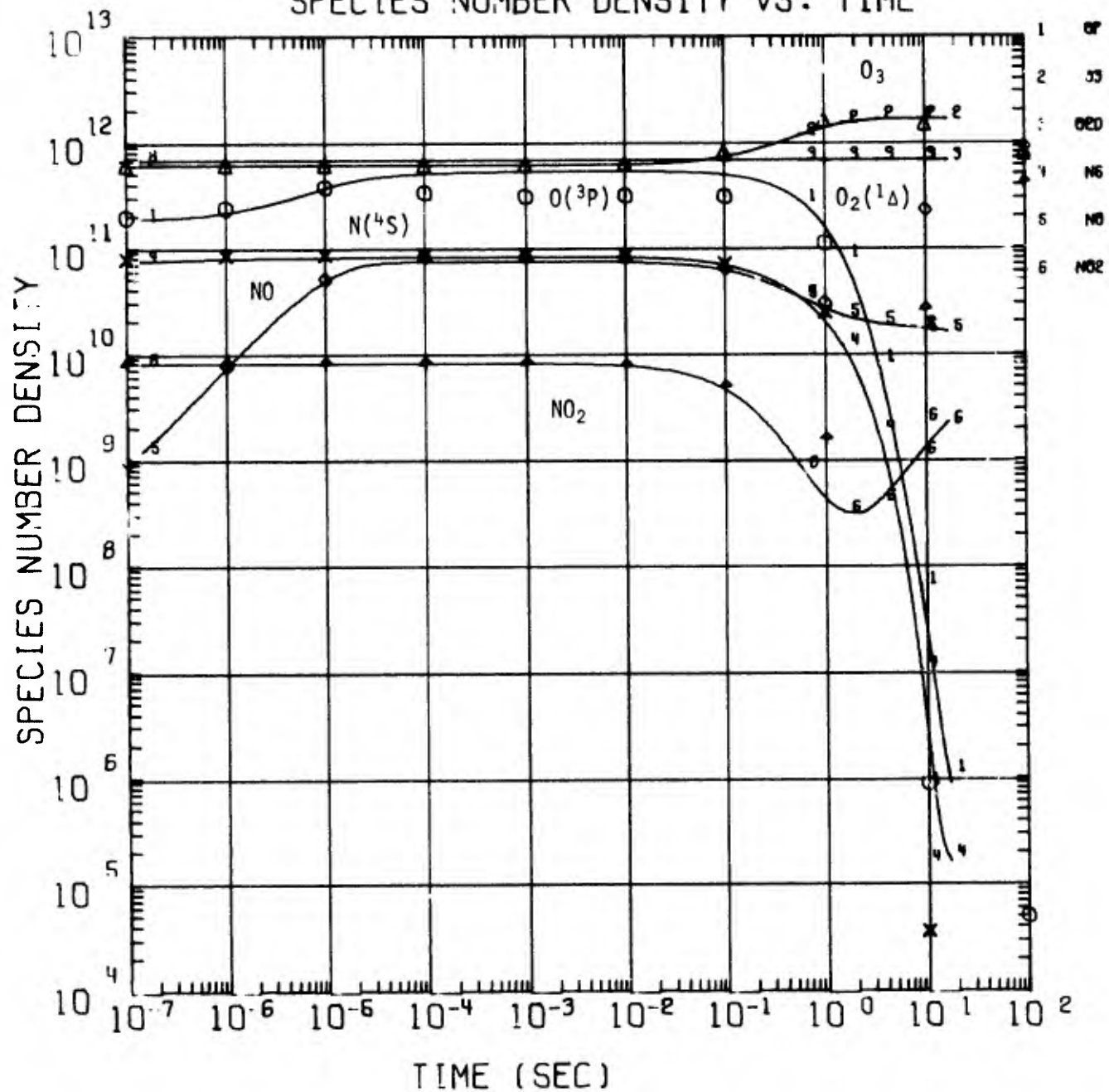


Figure 5. Comparison of Numerical and Analytic Algorithms (Altitude of 40 km, Energy Deposition of 4×10^{11} ion pairs/cc, and continuous ionization of 10^7 ion pairs/cc/sec during the night).

SECTION VII

CODE LISTING

This section lists the chemistry package as it is run outside WORRY. The first two routines constitute the external diver which decodes the users problem and prepares the necessary inputs for a chemistry package calculation by either the analytic or numerical algorithms. The diver also uses two plotting routines (not listed) to prepare comparison plots between the two methods. Since the external running of the chemistry package requires diagnostics which are otherwise unnecessary, the routines as listed contain some statements which are not standard in the operational version of WORRY. These statements are identified by a CH prefix in columns 73 and 74 of the card containing the statement.

When the chemistry package is used within WORRY, explicit phenomenology calculations provide a description of the continuous energy deposition as a function of time. When running in an external mode, the chemistry package as listed must assume a form for the time dependence of the continuous energy deposition. That form is

$$Q = QI*(1 + t)** 1.2$$

where t is the time after prompt energy deposition and QI is the energy deposition rate at $t = 0$.

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 1E5,TAPE6=OUTPUT,TAPE7=PUNCH,TAPE1,TAPE8)

EXTERNAL DRIVER FOR MORRY CHEMISTRY PACKAGE

NOTE THAT TAPE3=INPUT
 CALCOMP PLOT TAPE IS
 PROGRAM RETURNS 30TH MICROFILM AND CALCOMP

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 COMMON /BURST/ DAY(10)
 COMMON /BURST/ OYMET(10)
 COMMON /PLTS/ XPLOTS(7,54),NPPL(54),IPL,NO(7)
 COMMON /QIN/ QINQA
 COMMON /SMR/ SMRCH(80)
 COMMON /TITLE/ TITLE(7)
 COMMON /XUX/ RR(20),CMD(90),RCRT(90),NSPEC
 DIMENSION T1(70),TEST(10),YDMET(10)
 DIMENSION CH00(90),ANO(11),F03(11),FNO(11)
 DIMENSION X1(7),NSR(7)
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 0 (CMD(802),CMO781), (CMD(803),CMO782), (CMD(804),CMO783), (CMD(805),CMO784), (C
 1 (CMD(806),CMO785), (CMD(80

[illegible]

```

SWITCHES CARD
ONLY ONE CALL ARRAY PER TIME
IN CHD ARRAY AND DATA
BEFORE FIRST CARD TO BE READ. NI, AND
EXPLANATION OF SPECIES
ALL OTHER TO OCCUR. (58 ON)
FOR PLOTTING (SOLID LINES)
IN CHD ARRAY ON TAPE8 RUN (TAPE4)
IS READ ON TAPE5 NUMERICAL PLOTTING CAN NOT BE
USED, SINCE TAPE5 OVERLAYS TAPE4 PLOTTING.
TAPE6 TAPE7 (ANALYTIC CHEM) COMPARISON MAY
BE MADE BY USER. SWITCH(61) CARD
BEFORE FIRST CARD TO BE READ. NI, AND
FIRST CARD IS NO. OF TIMES TO BE READ. NI, AND
NEXT ONLY CARDS ARE IN TIME, WITH SWITCH(14)
USED ONLY FOR OVERLAYS. TAPE4 GREEN MADE PREV
PLOT TAPE5 NUML NOT BE USED IF SWITCH(15) IS USED.
PLOT TAPE6 MAY NOT BE USED IF SWITCH(15) IS USED.
SWITCH(61) MAY VARY IN NUML CHEMISTRY
DETAILED PRINT IN NUML CHEMISTRY
CHARGE AND PARTICAL NO. BEFORE AND AFTER E
EACH TIME. STRY CONCENTRATIONS VS TIME
PLOT CARDS ARE NECESSARY
PLOT CARDS OF PLOTS = TIME NUMBER OF PLOT CARDS
THE NUML CHEMISTRY
USED IN COMPARISON OF ANALYTIC AND NUML CHEMISTR
000 PLOT CARDS AND SWITCH(61) NEED ONLY BE INCLUDED
PLOT CARDS OF THE TWO INTERMEDIATE PLOTS WHICH
APPEAR IN THE SECOND. SWITCH(61) MUST BE ONLY ON THE PLACING
MUST BE CARD OF THE SET PROB WILL RESULT IN
SWITCH(61) IN THE FIRST(58) MUST BE ON IN THE
A SBADE PROBLEM. IT MAY OR MAY NOT BE ON IN THE
SECOND IF AN INDIVIDUAL PLOT IS DESIRED.
EITHER NUML OR ANALYTIC MAY BE RUN FIRST
EXAMPLE RUN
C 0 0 0 0 0 0
100KM NIGHT
1.0E-6 1.0E-5 1.0E-4 1.0E-3 1.0E-2
1.1 9.9E2
200. 0. 3.1E-1
0. 0. 0. 0. 0. 0.
0. 0. 0. 0. 0. 0.
1.2E7 1.0E3 1.0E4 1.0E11 0.0E12
1.99E12 100KM NIGHT
0. 0. 0. 0. 0. 0.
03 NO 3.1E-1 02PL NO2
200. 0. 0. 0. 0. 0.
0. 0. 0. 0. 0. 0.
1.2E7 1.0E3 1.0E4 1.0E11 0.0E12

```

SWTCH(12) = 1M	
SWTCH(13) = 1M	
SWTCH(14) = 1M	
SWTCH(15) = 1M	
SWTCH(55) = 1M	
SWTCH(56) = 1M	
SWTCH(57) = 1M	
SWTCH(58) = 1M	
SWTCH(59) = 1M	
SWTCH(60) = 1M	
SWTCH(61) = 1M	
SWTCH(62) = 1M	

[illegible]

C1 JUL 74 VERSION

PACKAGE CHEM74

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14  QA=X(4)
    QAY=OP 5
    GO TO 5
    CONTINUE
    CALL PLOT (XP,YP,40)
    STOP
    CONTINUE
15  METALS CARD M1
    IN=INM+1
    IF (INM.EQ.2) GO TO 16
    YDMETL(1)=X(2)
    YDMETL(2)=X(3)
    YDMETL(3)=X(4)
    YDMETL(4)=X(5)
    YDMETL(5)=X(6)
    YDMETL(6)=X(7)
    GO TO 5
16  METALS CARD M2
    YDMETL(7)=X(2)
    YDMETL(8)=X(3)
    YDMETL(9)=X(4)
    YDMETL(10)=X(5)
    GO TO 5
17  SPECIES TO BE PLOTTED
    JPL=9
    IPL=IPL+1
    READ (2,41) TYPE,(X(L),L=1,7)
    DO 18 IP=1,7
    IF (X(IP).EQ.BLANK) GO TO 18
    JPL=JPL+1
    XPLOIS(JPL,IPL)=X(IP)
    CONTINUE
    NPPL(IPL)=JPL
    GO TO 5
18  CONTINUE
19  PREPARE INPUTS FOR CHEMISTRY PACKAGE
    IF (SWITCH(14).EQ.1HR) GO TO 36
    TN=CN2+CO2+COP
    RHQ=TN+.522
    IF (ALTY+.100.) 20,25,21
    CONTINUE
    HQ=AMAX1((CALT-60.),0.)/20.
    HQ=EXP(-HQ)
    CC02=TN*.33E-4
    IHA=ALTY/20.10 IHA=3
    IF (IHA-1-IHA+10.)/10.
    IHA=IHA+1
    GO 3=ALOG(AO3(IHA))*(1.-IHA)+ALOG(AO3(IHA+1))*IHA
    GNO=ALOG(ANO(IHA))*(1.-IHA)+ALOG(ANO(IHA+1))*IHA
    GOM=ALOG(AOH(IHA))*(1.-IHA)+ALOG(AOH(IHA+1))*IHA
    CO3=EXP(GO3)
    GNO=EXP(GNO)
    GOM=EXP(GOM)
    GO TO 22
    CONTINUE
    TN=1.E-3E-4*TN
    CC02=3.3E-4*TN
    IHA=ALTY/10.-10
    IF (IHA-1-IHA+10.)/10.
    IHA=IHA+1
    GO 3=ALOG(FO3(IHA))*(1.-IHA)+ALOG(FO3(IHA+1))*IHA
    GNO=ALOG(FNO(IHA))*(1.-IHA)+ALOG(FNO(IHA+1))*IHA
    GOM=ALOG(FOM(IHA))*(1.-IHA)+ALOG(FOM(IHA+1))*IHA
    GO TO 21
20  CONTINUE
21  CONTINUE

```

```

369      GOJ=XP(GOJ)
370      QNO=EXP(QNO)
371      CONTINUE
372      IF (SWTCH(13).NE.1HQ) GO TO 23
373      R=QO(3,*) (CMD(I),I=1,NSPEC)
374      CONTINUE
375      IF (SWTCH(12).EQ.1HN) GO TO 24
376      SWTCH(6)=1. (CTIME)
377      CALL SECOND (CTIME)
378      PRINT 43, CTIME
379      TEMI=TEM
380      IF (TEMI.GE.0.) TEMI=-TEMI
381      INITIALIZATION CALL TO DECAYR
382      CALL DECAYR (I=-30,QA,OP,TEMI,10000.,RHO)
383      CALL SECOND (CTIME)
384      PRINT 43, CTIME
385      SWTCH(6)=0.
386      CONTINUE
387      DO 25 I=1,90
388      CMQ(I)=CMD(I)
389      LTM=0.
390      IF I=1,75
391      DO 26 II=1,75
392      DT=TI(II)
393      IF (DT.EQ.0.) GO TO 29
394      LTM=LTM+1
395      DO 26 I=1,75
396      CMQ(I)=CMD(I)
397      DO 27 I=1,75
398      DYMEIL(I)=DYMEIL(I)
399      WRITE(6,*) DT
400      TEM=TEM
401      CALL SECOND (CTIME)
402      PRINT 43, CTIME
403      CALCULATION OF CONTINUOUS ENERGY DEPOSITION AT TIME DT
404      Q=QI/(1.+DT)**1.2)
405      IF (SWTCH(55).EQ.1HQ) Q=QI
406      Q=Q+QA
407      CALL TO THE CHEMISTRY PACKAGE
408      CALL DECAYR (ED,Q,DP,TEM,DT,RHO)
409      CALL SECOND (CTIME)
410      PRINT 43, CTIME
411      PRINT 48
412      WRITE(6,47) ED,Q,QA,OP,TEM,DT,RHO
413      PRINT 48
414      WRITE(6,45) (CMD(I),I=1,54)
415      PRINT 48
416      WRITE(6,46) (RR(I),I=1,11)
417      CONTINUE
418      IF (SWTCH(50).NE.1HP) GO TO 35
419      START=1. (CTM)
420      CONTINUE
421      IF (SWTCH(14).NE.1HR) GO TO 33
422      S=ND(1,*)
423      IF (S=1. (CTM)
424      S=ND(3,4) NTI,(NSR(J),J=1,7)
425      DO 32 I=1,NTI
426      DO 32 J=1,NTI
427      DO 31 K=1,NTI
428      NSR(NSR(K))=X1(K)
429      WRITE(6,*) X1(K)
430      CONTINUE
431      DO 34 I=1,IPL
432      DO 34 J=1,IPL
433      CALL PLOTAK (START,END,I)
434      CONTINUE
435      CONTINUE

```


3450703234507330123
4444445555556555666
4444444444444444444

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01 JUL 74 VERSION

PACKAGE CHEM74

```

FUNCTION LOGIC (N,X,TEST)
  DIMENSION TEST(2)
  DO 1 K=1,N
    IF (X-TEST(K)) 1,2,1
    CONTINUE
  X=N+1
  LOGIC=X
  RETURN
END

```

0 1 2

467
469
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[illegible]

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DATA CONV/1.1657/
DATA (ENERG(1),4)=1.54/2.43,16.178,35.931,33.372,6.001,-1.179,-4.55
173,1.309,-2.476,0.33,0.3,0.167,7.354,4.890,7.264,6.459,3.00,3.55
28,4.525,6.748,15.583,12.363,19.412,16.175,16.583,16.752,5.00,3.97
37,1.627,4.340/
DATA CO2NG/0.0/

C PRINT 22, ED,Q,DP,TEM,DT,RHO
PRINT 23, CHD
Q=0
CON=CN2+CO2+COP
--- ADD METALS TO CHD ARRAY
DO 1 K=1,13
CHD(K)=CHD(K)+OYMETL(K)
OYMETL(K)=0.
DO 2 K=1,30
P(K)=0.
S(K)=0.
CONTINUE
CHIL=CL11+C3E+CHG+CAL+CSI+CPE+CCU+CII
CHILO=CL10+C8E0+CHG0+CAL0+CS10+CPE0+CCU0+CII0
IF (SWTCH(60).EQ.1.) GO TO 3
--- BRANCH TO NUMERICAL CHEMISTRY
IF (SWTCH(59).EQ.1HF) GO TO 21
CONTINUE
IF (SWTCH(57).NE.1HF) GO TO *
CHG=CN2PL+COPL+CNOPPL+CNPL+CO2PL+CUOPL+CN2APL+CN2BPL+CEL
PNUM=CN3+CN0+CNP+COP+COS+CU3+CNPL+COPL+CH+CHIL+CU+2.*(CO2+CO20+CO2
+3+CO2SG+CN2+CN2A+CN2B+CN2PL+CN2BPL+CN2BPL+CON+CO2+CO20)
2CHILO+CUOPL)+3.3*(CO3+UNO2+CN20+CCO2+CH20)
PRINT 24, CHG,PNUM
CONTINUE
--- CONVERT E0(ERGS/CM3) TO ION PAIRS/CM3
AND CALL ENQDEP TO ACCOUNT FOR PROMT ENERGY DEPOSITION EFFECTS
DEX=1.87E12*E0+1.E-30
CALL ENQDEP (DEX,CON,TEM)
--- CALCULATE TOTAL ENERGY FROM HEATS OF FORMATION AND TOTAL INTERNAL
KINETIC ENERGY
GENT=GEN
DO 5 I=1,NSPEC
GENT=GENT+ENERX(I)*CMD(I)
CONTINUE
--- FILL R ARRAY WITH RATE CONSTANTS
CALL RATE (TEM,CON)
--- CALL ENQDEP FOR CONTINUOUS SOURCE TERMS
CALL ENQDEP (DEX,CON,TEM)
--- BEGIN FAST ION CHEMISTRY
N+
CNPLJ=CNPL+COJ+CUJ
CN=CNPLNC+CN0+X02NPL*CG2+CNPLG*CO+X02NPL*CO2
EXPNT=EXP(-3N*UT)
UNPLJ=CNPLG*EXPNT
UNPLD=MAX(UNPL,3N*G.C,UT)
UNPL=CNPLT+CNPLJ
XN1GL=(1.-EXP3NT)/3N
---
0+
BO=XOPLU2+CO2+XOPLN0*CN0+RN2OPL*UN2

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EXPB0T=EXP(-30*DT)
COPLO=COPLO
GM=RNPL0*CO
YAE=GM*CNPL0/(80-BN)
COPLT=(COPLG-YAE)*EXPB0T+YAE*EXPBNT
XOGL=COPLT*GM*CNPL
COPLQ=MQA(QOPL,BO,G,G,DT)
COPL=COPLT+COPLT

      ATOMIC IONS

ATMPLI=COPLI+CNPLI
ATMPL0=COPLO+CNPL0
XOGL=1-EXP(-RO*DT))/BO
COPLO1=COPLO-YAE
COPLO2=COPLO+YAE
XOGL=COPLO1+XOGL+YAE*XNIGL
XOGLP=XOGL+XOGL+COPLO2*XNIGL
XOGLP2=XOGL+COPLO2+XOPL02

      NEGATIVE CHARGE

ALPHAR=(RN2PEL*CN2PL+RO2PEL*CO2PL+RNOPEL*CNOPL)/(CN2PL+CO2PL*CNOPL
+COPL*CNPL)
AB=ALPHAR*ATMPLI*DT
ALPHR=ALPHAR*SHXOX(AB)
CNEG=CEL+CO2NG
CNEG0=CNEG
ANGGP=ALPHR*CNEG
ANGGP1=ANGGP*DT
OANGP1=1+ANGGP1
CNEG=ATMPLI+(CNEG0-ATMPLI)/OANGP1
CNEG=MAX(CNEG,CNEG0)
XOGLP1=XNIGL(ALPHR,BO,CNEG0,DT)
XOGLP2=XNIGL(ALPHR,BN,CNEG0,DT)
XOGLP=COPLO1+XOGLP1+COPLO2+XOGLP2

      N2+

CN2PL0=CN2PL

      CN2PL0=CN2PL

      FACTOR OF 2 DUE TO SEPARATION OF REACTION 5 INTO TWO PARTS

BN2=BN2P02+CO2+XN2PNO*CO+RN2PLO*CO*2.G
CN2PLT=CN2PL0+EXP(-BN2*DT-RN2PEL*(XAGLP-XOGLP))/(1.+ALPHAR*CNEG0*0
1T)+1.E-50
BN2EFF=BN2+(RN2PEL*(XAGLP-XOGLP))/DT
IF (CN2PLT.LE.2.E-50) BN2EFF=BN2
GM=SQRT(ALPHAR*ALPHR)
XN2IGL=XNIGL(ALPHAR,BN2EFF,CNEG0,DT)
B=RN2PEL/CNEG0*BN2
CN2PLQ=MQX(CN2PL,B,J,0,DT)
CN2P02=MAX(CN2PLI,CN2PLQ)
CN2P03=CN2PL0+XN2IGL+XN2PNO*CO2
GN2PNO=CN2PLQ+XN2IGL+RN2PNO*CO
GN2PLO=CN2PLO+XN2IGL+RN2PLO*CO*2.G
DELTA1=BN2EFF+BO
DELTA2=BN2EFF+BN
GN2PEL=RN2PEL*(COPLO1+XNIGL(ALPHAR,DELTA1,CNEG0,DT)+COPLO2*
1XNIGL(ALPHAR,DELTA2,CNEG0,DT))+CNEG0*BN2EFF,CNEG0,DT)+C
3,DT))
AOELN2=GN2P02+GN2PNO+GN2PLO+GN2PEL+1.E-30
DELN2P=CN2PL0+GN2PLT
DN2TR=DELN2P/AOELN2

      UPDATE N AND O

GOPLN2=RN2OPL*CN2*XOIGL
GNPLO2=XO2NPL*CNPL0+CO2*XNIGL
GNPRO2=RO2NPL+CO2+XNIGL+CNPL0
GNPLO3=GM*CNPL0+XNIGL
GNPLNO=AMPLNO+LNO*CNPL0+XNIGL
GOPLNO=XOPLNO+LNO*XOIGL

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[illegible]

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775 QQNOPL=QNOPL+CN0*(CN2PL*KN2PNO+CO2PL*KN2PNO+COPL*KN2PNO+CNPL*KN2PNO+CN2P
776 101+2.0*CN2PL+CO*KN2PNO+COPL*KN2PNO+CNPL*KN2PNO+CN2P
777 2L*CN
778 3=KNOPEL*CNEL
779 CNCPLO=MAX(QNOPL,3.0*DT)
780 CNOPLO=MAX(QNOPL,3.0*DT)
781 --- REEVALUATE ALPHA AND ALPHRP AND THEN CEL AND CNOPL
782 1
783 ALPHA=(R2PEL*CN2PL+R2PEL*CO2PL+RNOPEL*CN2PL)/(CN2PL*CO2PL+CN2PL+CN2PL)
784 1+COPL+CNPL
785 ALPHRP=ALPHARP*SHXOX(AB)
786 OANOPT=1.0+ALPHARP*CNEL*DT
787 CNEGQ=ATMPLI*(CNEGQ-ATMPLI)/OANOPT
788 CNEGQ=HR(QQ,ALPHARP,0.0,DT)
789 CNEGQ=MAX(CNEGQ,CNEGQ)
790 CEL=CNEL*CNOPLO*EXPGL*(1.0+ALPHARP*CNEL*EXPGL*DT)*(-KNOPEL/ALPHARP)+1
791 CNOPL=CNOPL+CNEL
792 8=KNOPEL*CNEL
793 CNOPLQ=MAX(QNOPL,3.0*DT)
794 CNOPLQ=MAX(QNOPL,3.0*DT)
795 DNOTR=DNOTR-CNOPLQ
796 DNOTR=DNOTR-CNOPLQ
797 --- UPDATE N AND O
798 1
799 CNS=CNS+0.25*DNOTR-GO2PLN*(CNS/CN)
800 CND=CND+0.75*DNOTR-GO2PLN*(CND/CN)
801 CNE=CNE-GO2PLN*(CNE/CN)
802 CNE=CNE+CN2PL*CN2PL
803 CND=CND+0.8*GO2PLN*GO2PLN
804 CND=CND+0.8*GO2PLN*GO2PLN
805 CND=CND+0.4*GO2PLN
806 CND=CND+0.4*GO2PLN
807 CND=CND+0.4*GO2PLN
808 CND=CND+0.4*GO2PLN
809 IF (INTG.E1.0) GO TO 7
810 --- UPDATE TEMPERATURE AND RECALCULATE THE RATE CONSTANTS
811 1
812 CHT=1.5*(CNS+CND+CN2PL*CN2PL+CN2PL*CN2PL+CN2PL*CN2PL)+2.5*(CNOPL+CN2PL*CN2PL+CN2PL*CN2PL)
813 12+CN2PL*CN2PL+CN2PL*CN2PL+CN2PL*CN2PL
814 SUME=0.0
815 DO 6 I=1,NSPEC
816 SUME=SUME+CMU(I)*ENERX(I)
817 CONTINUE
818 CEN=ABS(CEN-SUME)
819 TGRV=CEN*CONV
820 TEM=1100.0
821 IF (TEMP.GT.1100.0) TEM=1100.0
822 CALL BAIL (TEM,CN)
823 CONTINUE
824 --- PREPARE PRODUCT TERMS FOR CONTINUOUS SOURCE CALCULATIONS
825 1
826 DOQX1=KN2PEL*CO*CN2PL
827 DOQX2=KN2PEL*CO*CN2PL
828 DOQX3=KN2PEL*CO*CN2PL
829 DOQX4=KN2PEL*CO*CN2PL
830 DOQX5=KN2PEL*CO*CN2PL
831 DOQX6=KN2PEL*CO*CN2PL
832 DOQX7=KN2PEL*CO*CN2PL
833 DOQX8=KN2PEL*CO*CN2PL
834 DOQX9=KN2PEL*CO*CN2PL
835 DOQX10=KN2PEL*CO*CN2PL
836 DOQX11=KN2PEL*CO*CN2PL
837 DOQX12=KN2PEL*CO*CN2PL
838 DOQX13=KN2PEL*CO*CN2PL
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840 DOQX15=KN2PEL*CO*CN2PL
841 DOQX16=KN2PEL*CO*CN2PL
842 DOQX17=KN2PEL*CO*CN2PL
843 DOQX18=KN2PEL*CO*CN2PL
844 DOQX19=KN2PEL*CO*CN2PL
845 DOQX20=KN2PEL*CO*CN2PL
846 DOQX21=KN2PEL*CO*CN2PL
847 DOQX22=KN2PEL*CO*CN2PL
848 DOQX23=KN2PEL*CO*CN2PL
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850 DOQX25=KN2PEL*CO*CN2PL
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854 DOQX29=KN2PEL*CO*CN2PL
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857 DOQX32=KN2PEL*CO*CN2PL
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859 DOQX34=KN2PEL*CO*CN2PL
860 DOQX35=KN2PEL*CO*CN2PL
861 DOQX36=KN2PEL*CO*CN2PL
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863 DOQX38=KN2PEL*CO*CN2PL
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995 DOQX170=KN2PEL*CO*CN2PL
996 DOQX171=KN2PEL*CO*CN2PL
997 DOQX172=KN2PEL*CO*CN2PL
998 DOQX173=KN2PEL*CO*CN2PL
999 DOQX174=KN2PEL*CO*CN2PL
1000 DOQX175=KN2PEL*CO*CN2PL

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U1 JUL 74 VERSION

PACKAGE CHEN74

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0000 --- 0 + U2 + U3 + M
0000 B1=U20*CO2
0000 --- 02 + 02*(S) 03 + 0
0000 B11=R02502*CO2
0000 --- 0(1S) 0(10)+ HV
0000 B13=R0SHV
0000 --- 0(10) 0(3P)+ HV
0000 B14=R0DHV
0000 --- N(4S)+ 03 NO** + 02
0000 B16=RNS03*CO3
0000 --- DELAY 0(1S) AND 0(10) AND 02*(S)
0000 XCOSD=COS*EXP(-B13*DT)
0000 COO=C00+B13/(B13-B14)*COS)*EXP(-B14*DT)-B13/(B13-B14)*XCOSD
0000 COB=C00-COS
0000 COE=C0250*EXP(-B11*DT)
0000 COZ50=C0250*EXP(-B11*DT)
0000 --- N(4S)+ 02 NO** + 0
0000 N(2P)+ 02 NO** + 0
0000 B3=RNS02*CO2
0000 --- N(20)+ 02 NO** + 0
0000 B4=RND02*CO2
0000 --- N(2P) N(20)+ HV
0000 B5=RMPHV
0000 ONP=CNP
0000 OND=CND
0000 ONS=CNS
0000 --- DECAY STATES OF N
0000 CNP=CNP*EXP(-(B5+B3)*DT)
0000 CND=CND+B3/(B5-B3)*ONP)*EXP(-B4*DT)-B5/(B5-B4)*CNP
0000 CNS=CNS*EXP(-B3*DT)
0000 CN=CNP+CND+CNS
0000 --- RECALCULATION OF NO
0000 BALL=R03NO*LO3+IONO*CO+IONO*CN
0000 CALL=B19*CN3+RONO2*CO*CN02
0000 CNO=MAX(CALL,BALL,CN02,DT)
0000 XKAY=RND02+RNDNO
0000 BTA=94+RNDNO*CN0
0000 RTIO=C02NO2+RND02/XKAY
0000 CNG=RTIO-(RTIO-CN02)*EXP(-XKAY*ZND*DT*SHXOX(STAT))
0000 CMA=33+RNSNO*LN0
0000 CMAI=34*DT
0000 ALLD=RNS02+RNSNO
0000 RT1=RNS02*CO2NO2/ALLD
0000 CNGRT1=RT1-CN02)*EXP(-ALLD*ZNS+ZND)*DT*SHXOX(GMAT))
0000 CNU=AHMA1(CNU,CNUG)
0000 --- DECAY OF STATES OF N
0000 BALL=(RNO2NA+RNO2N3)*CN02+TNON*CN0
0000 B3=RNSNO*CH0

```



```

99=RNDSNO*CNQ
CNS=CNS*EXP(-(B8+B16+BALL)*DT)
CND=CND*EXP(-(B9+BALL)*DT)
CNP=CNP*EXP(-(B6+BALL)*DT)
CNE=CNS*CNQ+CNQ
DEALLN=CNS+CNQ+ONP-CN
--- INCREASE O FOR LOSS OF N
COP=COP+DEALLN
CO=CO+DEALLN
XOC=COP
--- CALCULATION OF O3 AND O
BT03=RO30*CO+KNSO3*CNS*RO3NO*CNQ+POISO3
CO3=(CO3+311*OCOS/(B11-BT03))+B1*CO/(B1-BT03))*EXP(-(BT03*DT))-B11/(
1B11-BT03)*OCOS*EXP(-(B11*DT))-B1/(B1-BT03)*CO*EXP(-(B1*DT))
BT0=311*RO30*CO3+TONO*CNQ+RONO2*CNQ2
AYE=RNDSNO*CNQ2*CN+POISO3
PROO=(311*OCOS)/(B11-BT0)
EXPBT0=EXP(-(BT0*DT))
COP=COP+PROO
COS=COS*CO+XOC
CO=CO+CO*CO+XOC
CO=COP*CO3+CO
--- PREPARE USEFUL RATE TERMS FOR SUBSEQUENT CALCULATIONS
--- 0 + 03 02 + 02
B2=RO30*CO3
C2=B2*CO
--- 02 + 02*(S) 03 + 0
C11=311*CO2SG
--- N(4S) + 03 NO** + 02
B16=RMSO3*CO3
--- N(4S) + 02 NO** + 0
C1A=93*CNQ
--- N(2P) + 02 NO** + 0
C1B=93*CNQ
--- N(2D) + 02 NO** + 0
C2A=94*CNQ
--- N(4S) + NO N2 + 0
B3A=RNSNO*CNQ
C3A=B3A*CNQ
--- N(2P) + NO N2 + 0
C3B=93A*CNQ
--- N(2D) + NO N2 + 0
B4A=KNDNO*CNQ
C4A=B4A*CNQ
--- NO + U + h NO2 + M
B14J=TONO*CO
C14J=B14J*CO
--- NO + 03 NO2 + 02

```

[illegible]

C --- CALCULATION OF NO2 AND H2O

```

1074  FN02=C1+O+C141
1075  DN02=B145+3111+B112
1076  CN02=MOX(FM02, DN02, CN02, OT)
1077  C111=B111*CN02
1078  FN20=C111+C111
1079  DN20=9143
1080  CN20=MQX(FN20, DN20, CN20, OT)
1081  --- N20 + O2(10)
1082  ---
1083  ---
1084  B145=RO2N20*CN20
1085  ---
1086  --- CALCULATION OF O2(10)
1087  ---
1088  CO20Q=MQX(QO20, B145, O, C, OT)
1089  CO20=MQX(CO20Q, B145, CO20, OT)
1090  CO20=CO20Q+CO20
1091  ---
1092  --- H + O3
1093  ---
1094  B21=RM03*CO3
1095  ---
1096  --- OH + O
1097  ---
1098  B22=ROH0*CO
1099  ---
1100  --- CALCULATION OF H AND OH
1101  ---
1102  QX=B21
1103  QY=B22
1104  CHOH=CH+COH
1105  QZ=CHOH*QX
1106  COM=MOX(QZ, (QX+QY), COM, OT)
1107  CH=MAX1((CHOH-COM), 0.)
1108  ---
1109  --- BEGIN METAL-METAL OXIDE CALCULATION
1110  IF (CMTL.LE.0.) GO TO 17
1111  ---
1112  --- X + O3
1113  ---
1114  FMLOZ=RMTLO3*CO3
1115  ---
1116  --- LITHIUM
1117  ---
1118  NMT=1
1119  IF (CL1.LE.0.) GO TO 8
1120  ---
1121  --- LI + O + H
1122  ---
1123  B211=TLIO*CO3
1124  C211=B211*CL1
1125  ---
1126  --- LI + O2
1127  ---
1128  B212=RLIO2*CO2
1129  C212=B212*CL1
1130  ---
1131  --- LI0 + O
1132  ---
1133  B213=ROLIO*CO3
1134  ---
1135  --- LI0 + O3
1136  ---
1137  B214=RO3LIO*CO3
1138  ---
1139  --- LI + O3
1140  ---
1141  C215=FMLOZ*CL1
1142  QX=B211+B212+FMLOZ
1143  QY=B213+B214
1144  GO TO 15
1145  CONTINUE
1146  ---
1147  ---
1148  ---

```

| PACKAGE CHEM74 |   |                         | 01 JUL 74 VERSION |    |  | PAGE 10 |  |       |
|----------------|---|-------------------------|-------------------|----|--|---------|--|-------|
| ---            | 3 | CRYLLIUM                |                   |    |  |         |  | 11511 |
| ---            | 3 | NMT=NMT+1               |                   |    |  |         |  | 11512 |
| ---            | 3 | IF (CBE.LE.0.) GO TO 3  |                   |    |  |         |  | 11513 |
| ---            | 3 | BE + 0 + M              | BE0 +             | M  |  |         |  | 11514 |
| ---            | 3 | C221=B211*CB2           | BE0 +             | 0  |  |         |  | 11515 |
| ---            | 3 | BE + 02                 |                   |    |  |         |  | 11516 |
| ---            | 3 | B222=R322*CO2           |                   |    |  |         |  | 11517 |
| ---            | 3 | C222=B222*CB2           |                   |    |  |         |  | 11518 |
| ---            | 3 | BE0 + 0                 | BE0 +             | 02 |  |         |  | 11519 |
| ---            | 3 | BE0 + 03                | BE0 +             | 02 |  |         |  | 11520 |
| ---            | 3 | B224=R03610*CO3         |                   |    |  |         |  | 11521 |
| ---            | 3 | BE + 03                 | BE0 +             | 02 |  |         |  | 11522 |
| ---            | 3 | C245=FMLOZ*CB2          |                   |    |  |         |  | 11523 |
| ---            | 3 | QX=B211+B222+FMLOZ      |                   |    |  |         |  | 11524 |
| ---            | 3 | QY=B213+B224            |                   |    |  |         |  | 11525 |
| ---            | 3 | GO TO 15                |                   |    |  |         |  | 11526 |
| ---            | 3 | CONTINUE                |                   |    |  |         |  | 11527 |
| ---            | 3 | MAGNESIUM               |                   |    |  |         |  | 11528 |
| ---            | 3 | NMT=NMT+1               |                   |    |  |         |  | 11529 |
| ---            | 3 | IF (CMG.LE.0.) GO TO 11 |                   |    |  |         |  | 11530 |
| ---            | 3 | MG + 0 + M              | MG0 +             | M  |  |         |  | 11531 |
| ---            | 3 | C231=B211*CMG           | MG0 +             | 0  |  |         |  | 11532 |
| ---            | 3 | MG + 02                 |                   |    |  |         |  | 11533 |
| ---            | 3 | B232=RMG02*CO2          |                   |    |  |         |  | 11534 |
| ---            | 3 | C232=B232*CMG           |                   |    |  |         |  | 11535 |
| ---            | 3 | MG0 + 0                 | MG +              | 02 |  |         |  | 11536 |
| ---            | 3 | MG0 + 03                | MG +              | 02 |  |         |  | 11537 |
| ---            | 3 | B234=R03MG0*CO3         |                   |    |  |         |  | 11538 |
| ---            | 3 | MG + 03                 | MG0 +             | 02 |  |         |  | 11539 |
| ---            | 3 | C235=FMLOZ*CMG          |                   |    |  |         |  | 11540 |
| ---            | 3 | QX=B211+B232+FMLOZ      |                   |    |  |         |  | 11541 |
| ---            | 3 | QY=B213+B234            |                   |    |  |         |  | 11542 |
| ---            | 3 | GO TO 15                |                   |    |  |         |  | 11543 |
| ---            | 3 | CONTINUE                |                   |    |  |         |  | 11544 |
| ---            | 3 | ALUMINUM                |                   |    |  |         |  | 11545 |
| ---            | 3 | NMT=NMT+1               |                   |    |  |         |  | 11546 |
| ---            | 3 | IF (CAL.LE.0.) GO TO 11 |                   |    |  |         |  | 11547 |
| ---            | 3 | AL + 0 + M              | AL0 +             | M  |  |         |  | 11548 |
| ---            | 3 | C241=B211*CAL           | AL0 +             | 0  |  |         |  | 11549 |
| ---            | 3 | AL + 02                 |                   |    |  |         |  | 11550 |
| ---            | 3 | B242=RAL02*CO2          |                   |    |  |         |  | 11551 |
| ---            | 3 | C242=B242*CAL           |                   |    |  |         |  | 11552 |
| ---            | 3 | AL0 + 0                 | AL +              | 02 |  |         |  | 11553 |
| ---            | 3 | B243=R04AL0*CO3         |                   |    |  |         |  | 11554 |
| ---            | 3 | AL0 + 03                | AL +              | 02 |  |         |  | 11555 |

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[illegible]





PACKAGE CHEM74

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QNQ1=PFQD11*FRQ1
QNQ2PL=PFQ2PL*FRQ2
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[illegible]

21 JUL 74 VERSION

PACKAGE CHEM7+

FUNCTION X1IGL (A,B,C,T)

$$0.5 \exp(-8 \cdot T) \cdot DT / (1. + A \cdot C \cdot T)$$

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CC-0

AA=B/C/C  
IF (AA:LF:J:G) GO TO 1  
IF (AA:GT:700.) AA=700.

```
IF (AA.GT.750.)
  GO TO (1.+CC*I)
```

$$BB = A \cdot (1 + CC \cdot T) \cdot (EIF(A) - EIF(8B)) / CC$$

EXP

RETURN  
CONTINUE

X11GL=U. J  
R1 TURN

END

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RETURN

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[illegible][illegible]

החלטתו של בית דין זה היא סופית ובלתי ניתנת לערעור.

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FUNCTION X2IGL (A,B,C,T)
C --- INTEGRAL OF EXP(-B*T)*DT/(1.+A*C*T)**2
C
CC=A*C
AA=B/CC
IF (AA.GT.700.0) AA=700.0
BB=AA*(1.+CC*T)
XIGL=AA*EXP(AA)*(EIF(AA)-EIF(BB))
GO TO 2
CONTINUE
XIGL=0.0
CONTINUE
X2IGL=(1.-EXP(-B*T)/(1.+CC*T))-XIGL/CC
RETURN
END

```

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```

FUNCTION EIF (X)
  INTEGRAL OF EXP(-X)/X FROM X TO INF
  IF (X-1.)2
    Y=-ALOG(X)/(X*(X-0.976004)*X+0.0551968)*X-0.249910
    Y1=Y+0.9399193)*X-1.5721566
    GO TO 3
  GO TO 3
  IF GO TO 3
  RETURN
  RETEND

```

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PACKAGE CHEM74

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```

1      FUNCTION SHXOX (X)
2      --- EVALUATION OF (EXP(X)-1)/X
3      IF (ABS(X).LE.1.E-7) GO TO 1
4      IF (X.GT.70.) X=70.
5      SHXOX=(EXP(X)-1)/X
6      RETURN
7      CONTINUE
8      SHXOX=1.+X/2.
9      RETURN
10     END

```

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7 CALL EVIL (B,F,O,Y)
  DO 11 L=1,3
  CALL FUNG
  IF (NTG.EQ.1) GO TO 17
  CALL EVIL (B,P,S,Z)
  DO 8 I=1,N
  ZKR=ABS(Z(I)-Y(I))/(X(I)+1.0E-5)-1.0E-4
  IF (ERR.LE.0.) GO TO 6
  GO TO 9
  CONTINUE
  DO 10 I=1,N
  Y(I)=Z(I)
  CONTINUE
  DT=0.5*DT
  IF (DT.LE.1.0E-25) STOP
  GO TO 7
  CONTINUE
  LL=LL+1
  DO 13 J=1,N
  U(K,J)=Z(J)
  X(J)=Z(J)
  CONTINUE
  CEL=CN2PL+CUPL+CNPL+G02PL+CUOPL+CN2APL+CN28PL
  I=I+DT
  U(K,30)=CEL
  WRITE (4,27) T,(CMD(I),I=1,NSPEC)
  HRITE V*CONV
  SS(K,1)=P(87)
  SS(K,2)=P(86)
  SS(K,3)=P(83)
  SS(K,4)=P(83)
  SS(K,5)=P(23)
  SS(K,6)=P(24)
  IX(K)=IG
  DV(K)=DT
  V(K)=T
  DT=2.0*DT
  IF (I-END) 14,17,17
  IF ((T+DT)-CMD) 16,10,15
  DT=END-T+1.0E-10
  CONTINUE
  CONTINUE
  CONTINUE
  THE FOLLOWING ARE CHEM SPECIAL PRINTS *****
  IF (SWTCH156..NE.1HY) GO TO 22
  PRINT 24
  DO 18 K=1,LL
  PRINT 25, V(K),DV(K),U(K,33),U(K,40),U(K,25),(SS(K,I),I=1,6),TXX(K)
  1) CONTINUE
  PRINT 30
  DO 19 K=1,LL
  PRINT 34, V(K),U(K,34),U(K,35),U(K,51),U(K,22),U(K,31),U(K,38),U(K,
  1,33),U(K,41),U(K,42),U(K,43),U(K,44)
  CONTINUE
  PRINT 31
  DO 20 K=1,LL
  PRINT 34, V(K),U(K,21),U(K,27),U(K,1),U(K,11),U(K,36),U(K,23),U(K,
  124),U(K,13),U(K,20),U(K,52),U(K,54)
  CONTINUE
  IF (CHTCL.EC.) GO TO 22
  PRINT 32
  PRINT 33
  DO 21 K=1,LL
  PRINT 33, V(K),U(K,2),U(K,12),U(K,4),U(K,14),U(K,5),U(K,15),U(K,6)
  1,U(K,16),U(K,18),U(K,9),U(K,19)
  CONTINUE
  END OF SPECIAL PRINTS *****
  CONTINUE
  IF (I-END) 3,23,23

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SUBROUTINE _VIL (X,P,R,Y)
*****
      EVIL      SOLUTION OF DX/DT = P - R*X  FOR NUMERICAL ALGORITHM
*****
      DIMENSION X(1), P(1), R(1), Y(1)
      COMMON /KEN/ CONV,DT,N
      GO 3  I=1,N
      T=DT*R(I)
      IF (T.GT.1.E-3) GO TO 1
      Y(I)=X(I)*(1.-T)+DT*P(I)
      GO TO 3
1     CONTINUE
      RX=P(I)/R(I)
      IF (T.LE.100.) GO TO 2
      Y(I)=RX
      GO TO 3
2     Y(I)=(X(I)-RX)*EXP(-T)+RX
      CONTINUE
      RETURN
      END

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C
15 NO+
   RR(M+1)=FNOPL*LFZ(11)
   RR(M+2)=FNOPL*LFZ(11)
   RR(M+3)=FNOPL*LFZ(11)
   RR(M+4)=FNOPL*LFZ(11)
   GO TO 35
C
16 NO 17 L=1,7
   I=M+L
   RR(1)=FNOX*LFZ(12)+FNOG*LFZ(22)
   DO 18 L=8,N
   I=M+L
   RR(1)=FNOX*LFZ(12)
   GO TO 35
C
19 NO 20 L=1,N
   I=M+L
   RR(1)=FNO2*LFZ(13)
   GO TO 35
C
21 NO 22 L=1,N
   I=M+L
   RR(1)=FNO2*LFZ(14)
   GO TO 35
C
23 NO 24 L=1,N
   I=M+L
   RR(1)=0.C*LFZ(15)
   GO TO 35
C
25 NO 26 L=1,N
   I=M+L
   RR(1)=0.C*LFZ(16)
   GO TO 35
C
27 NO 28 L=1,7
   I=M+L
   RR(1)=F03X*LFZ(17)+F03S*LFZ(27)
   GO 29 L=8,N
C
30 NO 31 L=1,N
   I=M+L
   RR(1)=0.C*LFZ(18)
   GO TO 35
C
32 NO 33 L=1,N
   I=M+L
   RR(1)=0.C*LFZ(19)
   GO TO 35
C
34 CONTINUE
C
35 RR(199)=ROSHV*COO
   RR(200)=ROSHV*COS
   RR(11)=0H
   RR(12)=0L
   RR(13)=0J
   RR(14)=0K
   RR(15)=0M
   RR(16)=0N
   RR(17)=0O
   RR(18)=0P
   RR(19)=0Q
   RR(20)=0R
   RR(21)=0S
   RR(22)=0T
   RR(23)=0U
   RR(24)=0V
   RR(25)=0W
   RR(26)=0X
   RR(27)=0Y
   RR(28)=0Z

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0 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99

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PACKAGE CHM74  
RETURN  
END

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N 225  
N 226-

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SU2ROUTINE RATE (TG,CON)
*****
RATE      CALCULATES AND STORES REACTION RATE CONSTANTS IN R ARRAY
*****
COMMON /BURST/ DAY
COMMON /Y(93),P(43),S(93),R(93),CHT,TGEV,QQ,TNUM,NFLAS,NTG
COMMON /XCN/ CONV,DT,N
COMMON /XCN/ CM(10),
EQUIVALENCE (Y(93),XCN202), (Y(91),XCN201), (R(4),XCN200),
EQUIVALENCE (X(1),XCN201), (R(2),XCN202), (R(8),XCN203), (R(19),XCN204),
EQUIVALENCE (X(11),XCN205), (R(6),XCN206), (R(13),XCN207), (R(17),XCN208),
EQUIVALENCE (X(15),XCN209), (R(11),XCN210), (R(16),XCN211), (R(21),XCN212),
EQUIVALENCE (X(14),XCN213), (R(12),XCN214), (R(20),XCN215), (R(25),XCN216),
EQUIVALENCE (X(10),XCN217), (R(13),XCN218), (R(22),XCN219), (R(26),XCN220),
EQUIVALENCE (X(12),XCN221), (R(14),XCN222), (R(23),XCN223), (R(27),XCN224),
EQUIVALENCE (X(16),XCN225), (R(18),XCN226), (R(24),XCN227), (R(28),XCN228),
EQUIVALENCE (X(17),XCN229), (R(19),XCN230), (R(29),XCN231), (R(33),XCN232),
EQUIVALENCE (X(15),XCN233), (R(21),XCN234), (R(30),XCN235), (R(34),XCN236),
EQUIVALENCE (X(13),XCN237), (R(17),XCN238), (R(31),XCN239), (R(35),XCN240),
EQUIVALENCE (X(11),XCN241), (R(15),XCN242), (R(32),XCN243), (R(36),XCN244),
EQUIVALENCE (X(9),XCN245), (R(13),XCN246), (R(33),XCN247), (R(37),XCN248),
EQUIVALENCE (X(7),XCN249), (R(11),XCN250), (R(34),XCN251), (R(38),XCN252),
EQUIVALENCE (X(5),XCN253), (R(9),XCN254), (R(35),XCN255), (R(39),XCN256),
EQUIVALENCE (X(3),XCN257), (R(7),XCN258), (R(36),XCN259), (R(40),XCN260),
EQUIVALENCE (X(1),XCN261), (R(5),XCN262), (R(37),XCN263), (R(41),XCN264),
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EQUIVALENCE (X(1),XCN501), (R(5),XCN502), (R(97),
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2677	C	PACKAGE CHEN74	01 JUL 74	VERSION	PAGE	47	2677
2678	C	RETURN					2678
2679	C						2679
2680	30	FORMAT (E13.6,(6E13.6,/,))					2680
2681	31	FORMAT (1A,CHKTIME=,I5,3H					2681
2682		END					2682
2683							2683

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REFERENCES

1. Goetz, D. W. and S. H. Brecht, Vol IV, DNA 3131 P-4, June 1973.
2. Ory, H. A. and F. R. Gilmore, The Chemistry of Nitrogen Oxides and Ozone in the Disturbed D-Region, DNA 2835T, September 1971.
3. Scheibe, M., An Analytic Model for Nuclear Induced D-Region Chemistry, DNA 2920 F, October 1972.