

The public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing the collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden, to Washington Headquarters Services, Directorate for Information Operations and Reports, 1215 Jefferson Davis Highway, Suite 1204, Arlington VA, 22202-4302. Respondents should be aware that notwithstanding any other provision of law, no person shall be subject to any penalty for failing to comply with a collection of information if it does not display a currently valid OMB control number.  
PLEASE DO NOT RETURN YOUR FORM TO THE ABOVE ADDRESS.

1. REPORT DATE (DD-MM-YYYY) 26-09-2018	2. REPORT TYPE Final Report	3. DATES COVERED (From - To) 1-May-2015 - 30-Apr-2018
---	--------------------------------	--

4. TITLE AND SUBTITLE Final Report: Exploring Mechanophore and Polymer Design Rules for Mechanochemical Transduction	5a. CONTRACT NUMBER W911NF-15-1-0139
	5b. GRANT NUMBER
	5c. PROGRAM ELEMENT NUMBER 611102

6. AUTHORS	5d. PROJECT NUMBER
	5e. TASK NUMBER
	5f. WORK UNIT NUMBER

7. PERFORMING ORGANIZATION NAMES AND ADDRESSES University of Washington Office of Sponsored Programs 4333 Brooklyn Ave NE Box 359472 Seattle, WA 98195 -9472	8. PERFORMING ORGANIZATION REPORT NUMBER
--	--

9. SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS (ES) U.S. Army Research Office P.O. Box 12211 Research Triangle Park, NC 27709-2211	10. SPONSOR/MONITOR'S ACRONYM(S) ARO
	11. SPONSOR/MONITOR'S REPORT NUMBER(S) 62317-CH.15

12. DISTRIBUTION AVAILABILITY STATEMENT Approved for public release; distribution is unlimited.
--

13. SUPPLEMENTARY NOTES The views, opinions and/or findings contained in this report are those of the author(s) and should not be construed as an official Department of the Army position, policy or decision, unless so designated by other documentation.
---

14. ABSTRACT
--------------

15. SUBJECT TERMS
-------------------

16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF ABSTRACT UU	15. NUMBER OF PAGES	19a. NAME OF RESPONSIBLE PERSON Andrew Boydston
a. REPORT UU	b. ABSTRACT UU	c. THIS PAGE UU			19b. TELEPHONE NUMBER 206-616-8195

**RPPR Final Report**  
as of 22-Oct-2018

Agency Code:

Proposal Number: 62317CH

**Agreement Number: W911NF-15-1-0139**

**INVESTIGATOR(S):**

**Name:** Ph.D Andrew Jackson Boydston  
**Email:** boydston@chem.washington.edu  
**Phone Number:** 2066168195  
**Principal:** Y

Organization: **University of Washington**

Address: Office of Sponsored Programs, Seattle, WA 981959472

Country: USA

DUNS Number: 605799469

EIN: 916001537

**Report Date:** 31-Jul-2018

Date Received: 26-Sep-2018

**Final Report** for Period Beginning 01-May-2015 and Ending 30-Apr-2018

**Title:** Exploring Mechanophore and Polymer Design Rules for Mechanochemical Transduction

**Begin Performance Period:** 01-May-2015

**End Performance Period:** 30-Apr-2018

**Report Term:** 0-Other

Submitted By: Ph.D Andrew Boydston

Email: boydston@chem.washington.edu

Phone: (206) 616-8195

**Distribution Statement:** 1-Approved for public release; distribution is unlimited.

**STEM Degrees:** 3

**STEM Participants:** 6

**Major Goals:** At the outset of our proposed work, we had established the following goals:

1) Computational Modeling and Identification of Target Flex Activated Mechanophores.

For each model mechanophore, relaxed potential energy surface scans will be conducted in a method consistent with the envisioned effects of mechanical force upon the molecule. We will use the computational results in combination with experimental reactivities to establish guidelines for predicting mechanochemical transformations via flex activation.

2) Synthesis of Flex Activated Mechanophores and Incorporation into Polymers and Crosslinked Networks.

We will synthesize flex activated mechanophores designed to release small molecules without degradation of the polymer main chain. These will be incorporated into linear polymers and crosslinked networks for solid state and solution-based studies. Particular focus will be placed on incorporating mechanophores into elastomeric materials capable of multiple cycles of mechanical activation with shape recovery following each cycle.

3) Evaluation of Flex Activated Mechanophores in Solution and Solid State.

We will use a combination of solution techniques and solid state methods to evaluate the mechanochemical reactivity of the mechanophores. Quantitative assessment of percent activation as a function of applied load will be determined, and correlations between material type, method of applying stress, mechanophore structure, and calculated activation energies (from Aim 1) will be investigated.

During the course of the project, we added the following goals:

A1) Evaluation of mechanophores as tools in additive manufacturing.

A2) Fundamental investigations of metal-free ring-opening metathesis polymerization

**Accomplishments:** Please see attached document.

**Training Opportunities:** Participants in the research program had multiple opportunities for professional training, including participation in group meetings, attendance at conferences, enrollment in advanced chemistry courses, and contributions to scientific writing. Notably, students were directly involved in drafting patents as co-inventors on technologies that were borne out during the funding period. As the projects matured, students and postdocs took part in collaborative research, which exposed them to joint group meetings with experts in mechanical engineering and materials science. During the joint group meetings, researchers were responsible for presenting and discussing their recent results, strategizing to address challenges, and corresponding with collaborators to advance the project goals. Students also took classes outside of chemistry to enhance their training, including courses in materials science, mechanical engineering, and bioengineering.

## RPPR Final Report as of 22-Oct-2018

**Results Dissemination:** Several talks were presented on the funded research topic. These were largely at conferences of professional societies (e.g., ACS National Meetings, Gordon Research Conferences, Fusion Conference). Additionally, researchers in the program participated in several outreach activities. These included assisting in hands-on lab activities at Sammamish High School (Problem-Based Learning Curriculum), Yakima Valley Science Fair, Bellevue School District Science Fair, the University of Washington Math Science Upward Bound program, and the University of Washington "Research Exposed" seminar series.

Formal presentations at departmental seminars, conferences, and symposia are listed below:

The Ohio State University, departmental seminar, March 2018  
Virginia Tech, departmental seminar, February 2018  
University of Wisconsin–Madison, departmental seminar, February 2018  
University of Massachusetts–Amherst, departmental seminar, February 2018  
ExxonMobil, invited seminar, December 2017  
University of Chicago, Institute for Molecular Engineering seminar, November 2017  
University of Michigan, student-invited seminar, November 2017  
University of Minnesota, departmental seminar, November 2017  
Virginia Tech, departmental seminar, October 2017  
Florida State University, College of Engineering Seminar, October 2017  
Florida State University, departmental seminar (Chemistry), October 2017  
University of Southern Mississippi, departmental seminar, October 2017  
Tulane University, departmental seminar, October 2017  
Case Western Reserve University, departmental seminar (student invite), September 2017  
254th National Meeting of the American Chemical Society, Washington, D.C., August 2017. Seminar Title: Expanded Materials Space for Digital Light Processing Additive Manufacturing.  
Polymers Gordon Research Conference, South Hadley, MA, June 2017. Invited talk: Integration of Mechanoresponsive Materials and Additive Manufacturing.  
International Conference of Photopolymer Science and Technology, Chiba, Japan, June 2017  
Lawrence Livermore National Labs, Livermore, CA, May 2017.  
Society of Polymer Science, Japan, 66th Annual Meeting of SPSJ, Chiba, Japan, May 2017  
253rd National Meeting of the American Chemical Society, San Francisco, CA, April 2017. Seminar Titles: Polymers for Thermal and Mechanochemical Release of Nitroxyl (HNO) and Other Small Molecule Payloads; Redox-Mediated Metal-Free Ring-Opening Metathesis Polymerization  
Boston College, departmental seminar, March 2017  
Air Force Research Labs, Dayton, OH, December 2016  
Materials Research Symposium, Boston, MA, December 2016  
12th International Symposium on Organic Reactions, Kyoto, Japan, April 2016  
Stanford University, departmental seminar, March 2016  
IBM Almaden Research Center, invited seminar, February 2016  
University of Akron, departmental seminar, February 2016  
University of Colorado, departmental seminar, January 2016  
Colorado State University, departmental seminar, January 2016  
Air Force Academy, invited seminar, January 2016  
14th Pacific Polymer Conference, Kauai, HI, December 2015. Seminar Titles: 1) Metal-Free Ring-Opening Metathesis Polymerization. 2) Integration of Mechanochemistry and Additive Manufacturing.  
University of Massachusetts–Amherst, departmental seminar, November 2015  
2015 International Symposium on Stimuli-Responsive Materials, Santa Rosa, CA, October 2015. Seminar Title: Integration of Polymer Mechanochemistry and Additive Manufacturing.  
University of California, Santa Barbara, departmental seminar, September 2015  
Washington University in St. Louis, departmental seminar, September 2015  
Aberdeen Proving Ground, Army Research Labs, August 2015  
2015 Polymers for Advanced Technologies, Hangzhou, China, June 2015. Seminar Title: Metal-Free Ring-Opening Metathesis Polymerization.

# RPPR Final Report

## as of 22-Oct-2018

### Honors and Awards: Boydston:

Camille Dreyfus Teacher-Scholar Award

Thieme Chemistry Journal Award

Distinguished Teaching Award for Innovation with Technology, University of Washington

Dr. Laura Pascual:

NDSEG Fellowship

Miss Johanna Schwartz:

Eastman Chemical Graduate Student Award (finalist)

### Protocol Activity Status:

**Technology Transfer:** Five US patent applications and one European patent application were filed (each currently pending). We have also launched a startup company based upon our metal-free ROMP technology. That same project also transitioned into a cooperative research agreement with the ARL APG (main point-of-contact: Dr. Dan Knorr, Jr.). Additional interactions have included visits and material transfer with New Balance, Henkel Corporation, ExxonMobil, and 3-D Systems.

### PARTICIPANTS:

**Participant Type:** PD/PI

**Participant:** Andrew Jackson Boydston

**Person Months Worked:** 2.00

**Funding Support:**

Project Contribution:

International Collaboration:

International Travel:

National Academy Member: N

Other Collaborators:

**Participant Type:** Graduate Student (research assistant)

**Participant:** Laura Pascual

**Person Months Worked:** 10.00

**Funding Support:**

Project Contribution:

International Collaboration:

International Travel:

National Academy Member: N

Other Collaborators:

**Participant Type:** Graduate Student (research assistant)

**Participant:** Victoria Kensy

**Person Months Worked:** 6.00

**Funding Support:**

Project Contribution:

International Collaboration:

International Travel:

National Academy Member: N

Other Collaborators:

**Participant Type:** Graduate Student (research assistant)

**Participant:** Bo Cao

**Person Months Worked:** 10.00

**Funding Support:**

Project Contribution:

International Collaboration:

International Travel:

National Academy Member: N

Other Collaborators:

**RPPR Final Report**  
as of 22-Oct-2018

**Participant Type:** Postdoctoral (scholar, fellow or other postdoctoral position)

**Participant:** Adam Goetz

**Person Months Worked:** 2.00

**Funding Support:**

Project Contribution:

International Collaboration:

International Travel:

National Academy Member: N

Other Collaborators:

**Participant Type:** Graduate Student (research assistant)

**Participant:** Pengtao Lu

**Person Months Worked:** 8.00

**Funding Support:**

Project Contribution:

International Collaboration:

International Travel:

National Academy Member: N

Other Collaborators:

**Participant Type:** Graduate Student (research assistant)

**Participant:** Ashlynn Vander Meer

**Person Months Worked:** 2.00

**Funding Support:**

Project Contribution:

International Collaboration:

International Travel:

National Academy Member: N

Other Collaborators:

**Participant Type:** Postdoctoral (scholar, fellow or other postdoctoral position)

**Participant:** Damian Dunford

**Person Months Worked:** 4.00

**Funding Support:**

Project Contribution:

International Collaboration:

International Travel:

National Academy Member: N

Other Collaborators:

**Participant Type:** Undergraduate Student

**Participant:** Thomas Bearrood

**Person Months Worked:** 2.00

**Funding Support:**

Project Contribution:

International Collaboration:

International Travel:

National Academy Member: N

Other Collaborators:

**RPPR Final Report**  
as of 22-Oct-2018

**PATENTS:**

**Intellectual Property Type:** Patent Date Received: **26-Sep-2018**  
**Patent Title:** METATHESIS POLYMERIZATION METHODS  
**Patent Abstract:** The present disclosure is directed to methods of making a polymer, including exposing a reactio  
**Patent Number:** 20170240695  
Patent Country: USA  
Application Date: 24-Aug-2017 Application Status: 2  
Date Issued:

**Intellectual Property Type:** Patent Date Received: **26-Sep-2018**  
**Patent Title:** THREE-DIMENSIONAL PRINTED MECHANORESPONSIVE MATERIALS AND RELATED METHODS  
**Patent Abstract:** The present disclosure provides additive manufacturing methods that include depositing onto a :  
**Patent Number:** 20170225395  
Patent Country: USA  
Application Date: 10-Aug-2017 Application Status: 2  
Date Issued:

**Intellectual Property Type:** Patent Date Received: **26-Sep-2018**  
**Patent Title:** POLYMERS FOR THERMALLY-REGULATED RELEASE OF HNO AND OTHER MOLECULES  
**Patent Abstract:** Multivalent scaffolds configured to facilitate drug release upon exposure to a stimulus, such as f  
**Patent Number:** 20170202874  
Patent Country: USA  
Application Date: 20-Jun-2017 Application Status: 2  
Date Issued:

**Intellectual Property Type:** Patent Date Received: **26-Sep-2018**  
**Patent Title:** CONTROLLED RELEASE OF CARBON DIOXIDE FROM SELF-IMMOLATIVE POLYMERS  
**Patent Abstract:** The present disclosure is directed to a carbon dioxide-releasing device, including a chamber ha  
**Patent Number:** 20160066553  
Patent Country: USA  
Application Date: 10-Mar-2016 Application Status: 2  
Date Issued:

**Intellectual Property Type:** Patent Date Received: **26-Sep-2018**  
**Patent Title:** THERMALLY-ACTIVATED SELF-IMMOLATIVE MATERIALS  
**Patent Abstract:** A polymer including a self-immolative polymer segment and a thermally-activated trigger moiety  
**Patent Number:** 20150368403  
Patent Country: USA  
Application Date: 24-Dec-2015 Application Status: 2  
Date Issued:

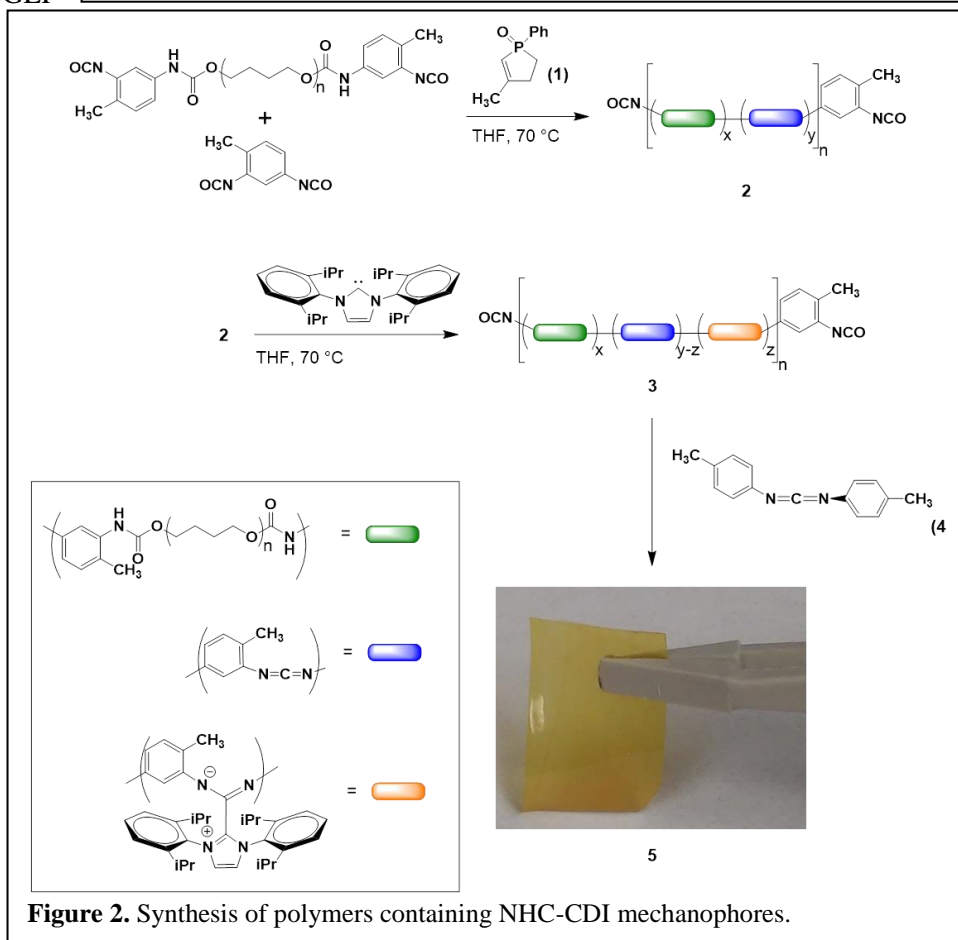
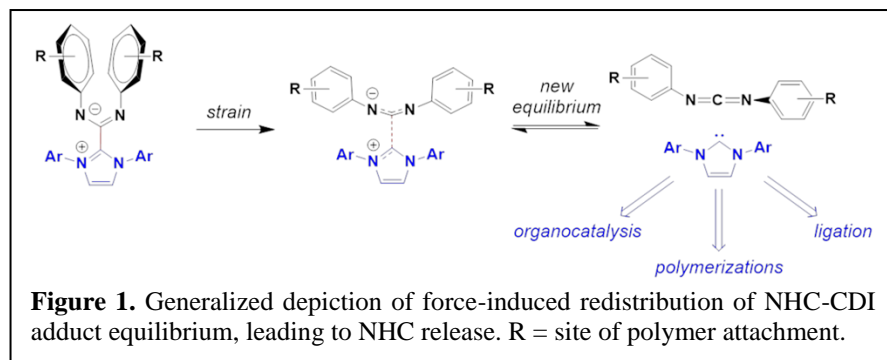
## ACCOMPLISHMENTS

## 1. FLEX-ACTIVATED MECHANOPHORES:

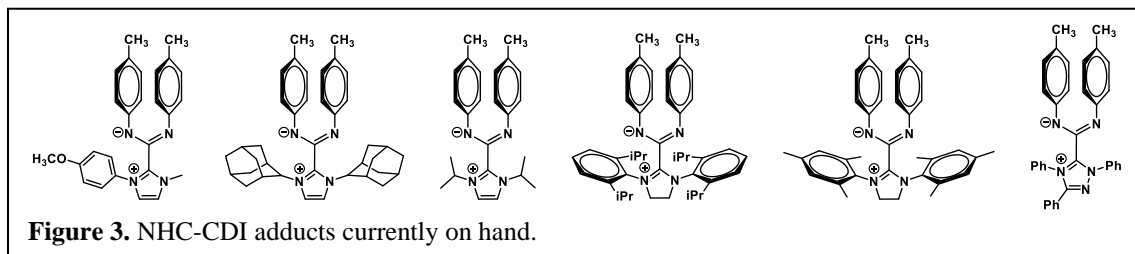
## 1.1. NHC-CDI adducts as flex-activated mechanophores

Our recent efforts have centered on mechanochemical activation of adducts occurring between N-heterocyclic carbenes (NHCs) and carbodiimides (CDIs). The NHC-CDI mechanophore structures and envisioned reactivity are depicted in Figure 1. Following CoGEF modeling, we next confirmed the mechanochemical activation and release of NHC species from bulk polyurethane materials (each described in our previous IPR). As depicted in Figure 2, the chain extenders and NHC structure are expected to be modular in nature. We have prepared samples with PEG, PPG, and alkyl linkages that display a range of mechanical properties and swelling behaviors.

Variations in the NHC reactivity have also been the focus of our current funding period. Toward this end, we have prepared a series



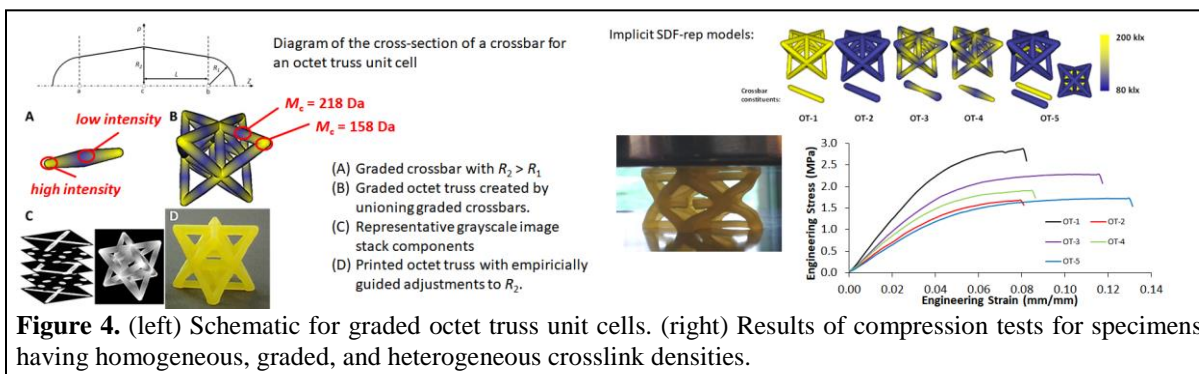
of small molecule NHC-CDI adducts (Figure 3). These small molecules adducts serve multiple purposes for our studies: 1) we are determining the thermal equilibrium parameters for each adduct, 2) the adducts are used as internal standards for LS-MS analysis of activated NHC species from polymeric networks, and 3) they confirm the product species spectroscopic signatures for crossover experiments. Moreover, the NHCs in this series represent a broad range of nucleophilicity and basicity. Therefore, we expect to find correlations between fundamental physical organic parameters and the efficiency of mechanochemical reactivity.



## 2. FLEX ACTIVATED MECHANOPHORES IN ADDITIVE MANUFACTURING

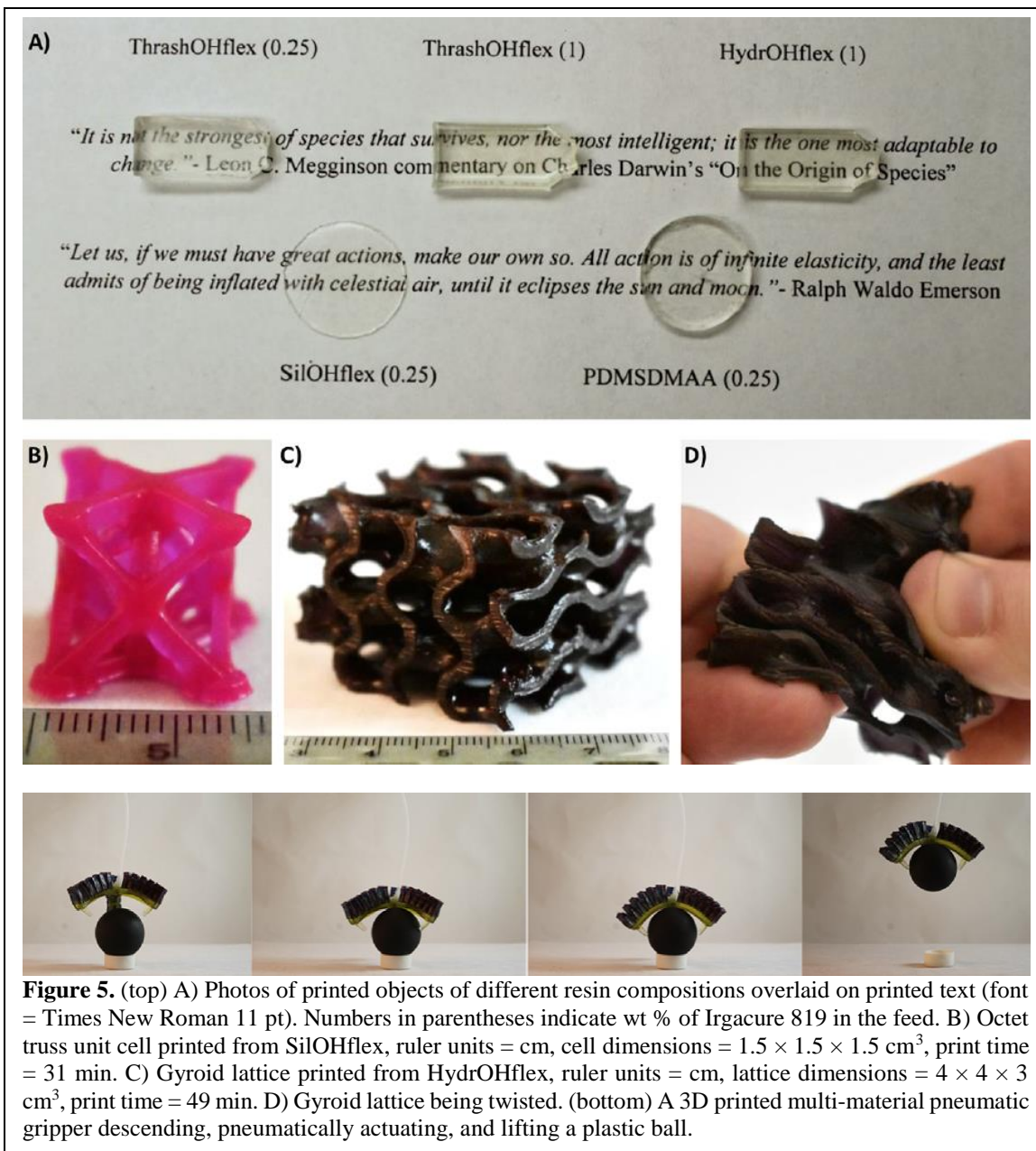
### 2.1. Chemical control over each voxel in 3D printed objects.

Our desire to integrate mechano-responsive materials with additive manufacturing (AM) began inspired us to consider the interplay between object geometry and chemical composition. For example, the unit cell for an octet truss lattice (Figure 4) compensates for uniaxial compression through a combination of non-uniform local stress and strain fields. This prompted questions about the potential to modulate mechanical responses as a function of crosslink density (chemistry to enhance engineering designs) and the use of localized strain to activate mechanophores (engineering designs that enhance chemical responses). We first explored modifications to digital light processing (DLP) AM. The research goal was to assess the feasibility of using grayscale input to control crosslink density in prescribed regions of the test specimen, and therefore achieve heterogeneous or graded material properties. Digital image stacks were created based upon implicit sign-distance field representations of the octet truss crossbars. This enabled an overlay of geometric and materials functions (Figure 4) such that each voxel within the digital object was assigned a grayscale light intensity based upon location within the object (visually depicted as yellow and blue for 200 to 80 klux intensities). The results were exciting in that different stress-strain profiles were obtained, which included increased mechanical toughness and strain to break for graded and heterogeneous specimens in comparison with the homogeneous counterparts.



Before integrating mechanophores into lattice structures via DLP-AM we wanted to address two challenges: 1) thermoplastics are not ideal for evaluating mechanochemistry since permanent plastic deformation is required before mechanophores are activated, and 2) quantifying of mechanochromic changes would be difficult in a complex, small feature size lattice. The first challenge was addressed by developing new elastomer photoresin formulations, which were recently accepted for publication. This work required a combination of custom synthesis and formulations science to establish low viscosity photoresins with good “printability.” Ultimately, we found the mechanical properties and swelling behavior of printed parts to be easily tuned. These materials also enabled rapid access to a multi-material pneumatic gripper (an air-driven soft robot) as depicted in Figure 5 (bottom).

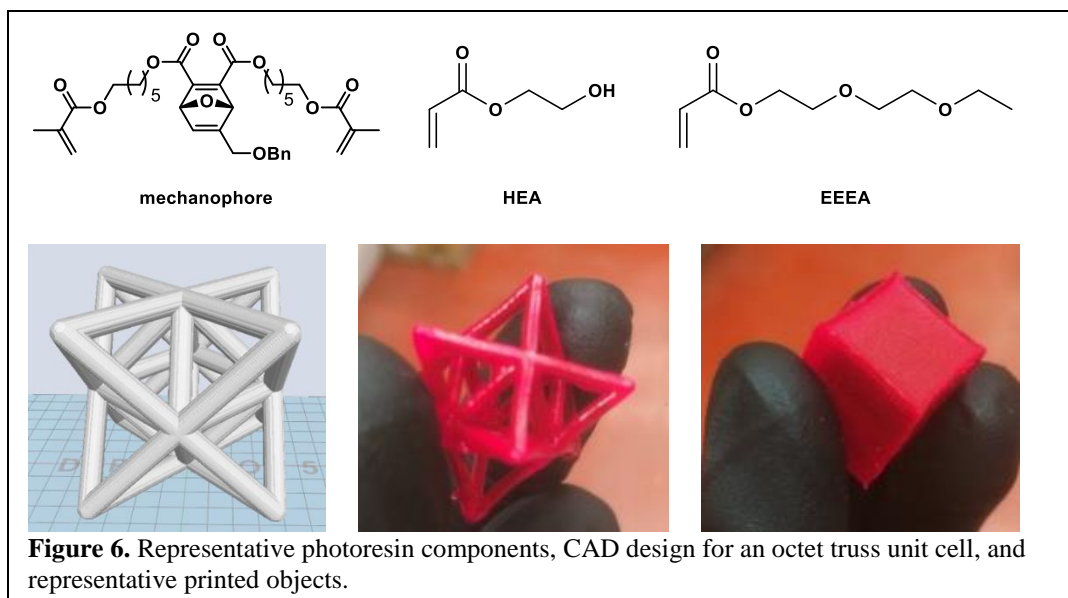




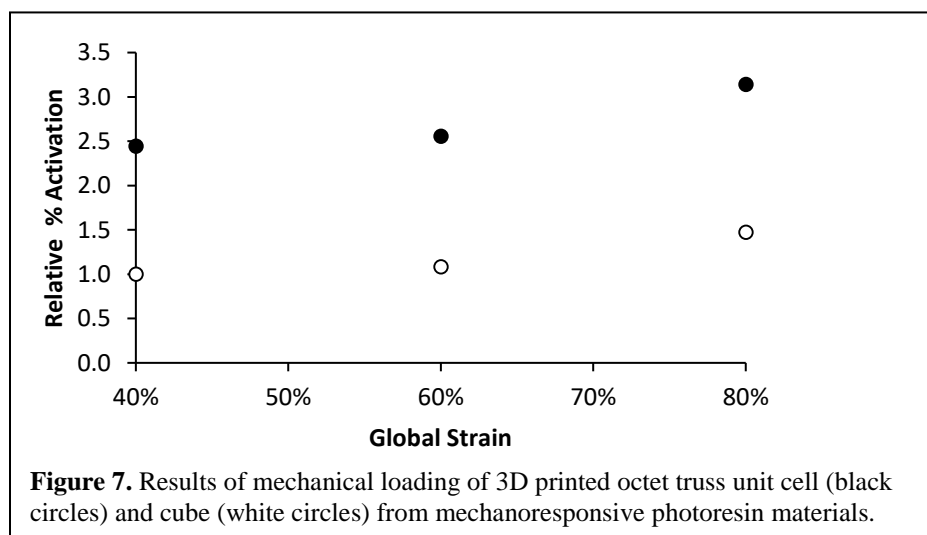
The second challenge, quantifying mechanophore activation in lattice structures, was addressed by using our flex activated mechanophore design. This enables us to extract and quantify the amount of small molecule that is released upon mechanophore activation in a way that is nondestructive to the printed object. The results of these studies thus far are currently being prepared for publication.

## 2.2. DLP-AM and mechanical loading of flex activated mechanophores in truss structures.

We have printed a number of geometries using a combination of the mechanophore, HEA, EEEA, Irgacure 819 photoinitiator, and Nile Red as a dye to enhance resolution (representative structures shown in Figure 6). Examples of a cube and an octet truss unit cell are shown in Figure 6.



Our goal was to obtain preliminary results on how an object's geometry could translate into different mechanochemical activation efficiencies at the molecular level. Upon compression of the printed test specimens, we found that the octet truss unit cell gave equal or greater percent activation of mechanophores in comparison with the cube. This is consistent with the greater local strains experienced in the truss structure at the same global strain as the cube. Figure 7 shows the results of loading each specimen type in a hydraulic press to the same global strain. As can be seen, the truss consistently gives roughly 2-3 times the percent activation of the cube.



### 3. METAL-FREE RING-OPENING METATHESIS POLYMERIZATION.

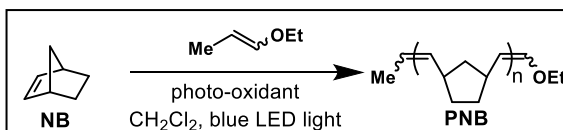
Building from our initial success with organocatalyzed anodic oxidations, we recently discovered a method for “metal-free” ROMP (MF-ROMP). Our first major advance came when we applied visible light photoredox catalysis. The general polymerization scheme (Figure 8) involves use of a photo-oxidant to reversibly oxidize an electron rich vinyl ether initiator. The resulting radical cation reacts with strained cycloalkene monomers to facilitate ring opening metathesis and generation of a new radical cation chain

end. Since our initial discovery in 2015, we have worked to increase our depth of understanding for the mechanism and broaden the potential fields of use for MF-ROMP.

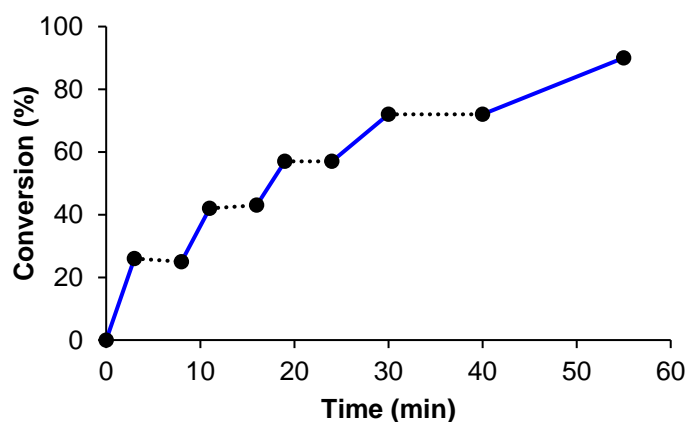
The reversibility of the oxidation and dynamic interplay between the chain end and photo-oxidant enable temporal control over the polymerization. The results presented in Figure 9 demonstrate the ability to switch the polymerization on and off using the photoredox-mediated method.

Notably, no significant changes in conversion and molecular weight were observed during the dark periods, and each re-exposure to light resulted in increased conversion and molecular weight while maintaining monomodal molecular weight distributions. In this regard, the MF-ROMP shows characteristics of a living polymerization. This level of temporal control is distinct from many of the metal-based externally-controlled initiators.

Efforts parallel to our mechanistic studies have focused on expanding functional group compatibility (in collaboration with Dr. Dan Knorr, Jr. at ARL APG), synthesis of complex polymer architectures (e.g., cyclic, block, and graft copolymers), production of thermally processable linear polydicyclopentadiene for commercial applications, and collaborative work with the ARL APG (Cooperative Research Agreement via Dr. John La Scala and Dr. Dan Knorr, Jr.). For example, initial studies identified esters, ethers, carbonates, silyl ethers, halides, and arenes as compatible functional groups for homo- and copolymerizations via MF-ROMP. Those results enabled us to then target block and random copolymers that can function as surfactants and film formers for glass fiber sizing packages by controlling the hydrophile-lipophile balance. Copolymers were obtained initially by direct copolymerization of two or more monomer species. However, we also capitalized on the ease with which vinyl ether initiator moieties could be installed to produce multitopic macroinitiators. We prepared ditopic (bidirectional) initiators from polyethylene glycol and polypropylene glycol. Subsequent MF-ROMP produced ABA triblock copolymers in good yield. "Grafting from" approaches were also successful, such as grafting via MF-ROMP from vinyl ether functionalized polystyrenes. Separately, we discovered that dicyclopentadiene, the most widely commercial resin feedstock for ROMP, can be polymerized via MF-ROMP to give a linear (non-crosslinked) product. This offers unique processing capabilities and potential for broadened fields-of-use in commercial applications. The production of linear polydicyclopentadiene and composites thereof are currently the centerpiece for our commercialization efforts through our recently formed startup company.



**Figure 8.** Generalized MF-ROMP using a photoredox catalyst and visible light.



**Figure 9.** Plot of % conversion of NB over time. Solid blue line = time under blue LED light. Dashed black line = time in the dark.