

UNCLASSIFIED

AD NUMBER
AD836696
NEW LIMITATION CHANGE
TO Approved for public release, distribution unlimited
FROM Distribution authorized to U.S. Gov't. agencies and their contractors; Administrative/Operational Use; OCT 1963. Other requests shall be referred to Department of the Army, Fort Detrick, Attn: Technical Release Branch/TID, Frederick, MD 21701.
AUTHORITY
SMUFD D/A ltr, 8 Feb 1972

THIS PAGE IS UNCLASSIFIED

AD836696

TRANSLATION NO. 931

DATE: Oct. 19/63

DDC AVAILABILITY NOTICE

Reproduction of this publication in whole or in part is prohibited. However, DDC is authorized to reproduce the publication for United States Government purposes.

STATEMENT #2 UNCLASSIFIED

This document is subject to special export controls and each transmittal to foreign governments or foreign nationals may be made only with prior approval of Dept. of Army, Fort Detrick, ATTN: Technical Release Branch/TID, Frederick, Maryland 21701

20050623 119

Best Available Copy

DEPARTMENT OF THE ARMY
Fort Detrick

DESALETING AND FRACTIONATION OF SUBSTANCES WITH DEXTRAN GELS

Following is a translation of a report by E. J. Flodin and H. J. Forth of the Institute for Research of Virus Diseases of Animals of the Federal Institute of Bacteriology at Munchen, Germany, which appeared in the German language periodical Zeitschrift für Bakteriologie (Abteilung of Bacteriology), Vol 182, pages 169-174, 1961.

Recently Forth and Flodin (1) reported on a simple and reliable method of desalting biological material with the help of dextran gels. This method is based upon the well-known column chromatography technique in which the stationary phase is a new synthetic gel type. These dextran gels consist of hydrophilic polyanion chains which are interlaced. The degree of interlattice is very constant within the individual gels. They have no active ion groupings, are nonsoluble in water, but have a strong affinity for water. The degree of expansion of the gels in water is determined by the interlattice degree of the dextran. The polar character is brought about almost exclusively by the large number of hydroxyl groups of the gel.

The desalting as well as the fractionation of substances with dextran gels is based mainly upon the differences in their molecular size. When a substance mixture is filtered through a column packed with a dextran gel, the larger molecules migrate faster than those of smaller dimensions. In certain approximation and in the practical application of these methods, the molecular weight can be compared instead of molecular size. Therefore if the difference of the molecular weights of substances is sufficiently great, their complete separation is possible with this gel filtration. In this case, a molecule sieve effect is the main principle and the chromatographic process in its classical

of the medium, 1 liter of 0.1 M phosphate buffer, pH 7.5 (8,9).

In addition to various published methods for the purification of protein solutions, also used for the purification of virus suspensions and other colloidal media - the gel filtration technique (8-10). Further experiments, with various media, have undertaken a change of the buffer system in biological solutions with great time loss or difficulty, as is often desirable in ion exchange chromatography or in electrophoresis. The gel filtration method is particularly with regard to biological substances and biological media has sustained (4-6).

Knowledge of this gel filtration method led us to conduct the following experiments on the use of this technique for decanting of various virus suspensions. The method seemed to us to be particularly useful with respect to the concentration of virus suspensions with ammonium sulphate and in the adsorption of the foot and mouth disease virus on aluminium hydroxide and subsequent elution with 1/2 M phosphate buffer, $P_{H} = 7.5$ (11). We were looking for a full-scale replacement for the customary dialysis, which often leads to great losses in infectiveness and has certain other disadvantages as well. In order to obtain a broad basis, we examined the behavior of viruses of various origin, differing structuring and characteristics, and suspended in various saline solutions, during gel filtration. In addition to the Newcastle disease virus (NDV) and the virus of pigeonpox (TPV), our experiments dealt mainly with the contagious swine paralysis (polioencephalomyelitis enzootica suum) virus (Teschenvirus) and the virus of foot and mouth disease (FMD).

Materials and Methods

Virus material: For examination of the smaller types of virus, we used mouse and culture viruses of the MMS strain "O₂-Brescia" as well as culture viruses of contagious swine paralysis (Teschener Disease), strain Konratice.

The MMS mouse virus came from a 10% extract in culture medium No. 2 (12), which was produced from heart and skeletal muscles of new-born mice of the 36th serial passage of the virus in mice. The MMS culture virus used was the 20th serial passage of virus in cultures of swine kidney cells. The virus medium used here was VII 3a (12). The virus material was stored at -20°C until the start of the experiment. The melted mouse material was always partially purified by diluting it with 50% chloroform (p.A.), shaking for 30 minutes at + 4°C and followed by slow centrifugation. The culture liquids were only centrifuged for 10 minutes at 7000 r.p.m. in order to remove cell materials. In both cases,

The research virus was obtained in the manner of the MS virus from swine kidney cultures of the 4th serial passage of the virus.

Newcastle Disease virus (NDV) -- strain Italia -- and fowl pox virus (FPV) were selected to represent the medium and large size virus types. NDV was available in the form of allantoic liquid from incubated chicken eggs (79th serial passage). FPV in culture virus (62nd serial passage in chicken fibroblast cultures) was used. These viruses were also stored until use at -20°C and after thawing were cleared through slow centrifugation. Further details are mentioned in the text.

Preparation of dextran gel and column: In our experiments we used as the dextran gel Sephadex G 25 (Pharmacia, Uppsala, Sweden) with a tumeaction factor of 2.3 grams water/gram dry matter and a grain size (dry) of 50-270 mesh.

The dry Sephadex was suspended in a 1% solution of sodium chloride for steeping. After about one hour, we removed very fine grain material through repeated washing and decanting of the gel substance with distilled water. We then removed the small air bubbles clinging to the gel suspended in water through brief evacuation in a suction bottle. Chromatographic columns were then filled with this gel suspension. The column diameter was 1.5 cm and the column length 35-39 cm. In all experiments, we computed from the column diameter (d) and the column length (L) which the Sephadex gel filled, the total gel volume ($V_t = \pi \frac{d^2}{4} \cdot L$). This value was controlled by measuring the volume of the column filled only with distilled water. In the main experiments, the computed and the measured total gel volume was $V_t = 62-69$ milliliter. In comparison tests with NDV, FPV and MS viruses, we also used smaller columns with a value of $V_t = 25$ milliliter. We determined the empty volume (V_0) of the columns in prior tests with hemoglobin, which, as a high-molecular protein (molecular weight 68,000), reacts indifferently to the Sephadex gel. Because of its brown-red color, hemoglobin is also a good indicator for the elutriation of the virus.

The hemoglobin was dissolved to 0.1% in physiological, m/90 phosphate-buffered Sodium chloride solution, $p_H = 7.6$ (phys NaCl-solution). After slow centrifugation, we brought 5 milliliter of the clear 0.1% hemoglobin solution drop-wise into a column packed with gel, which had been well washed with NaCl solution. After the hemoglobin solution had set, we eluted the blood pigmentation material and took off the eluent in fractions of 3 milliliter. These fractions were then measured individual in an ultraviolet spectrophotometer at 280 millimicron. If the extinctions are plotted graphically to the elution volume in a graph, then

the ...
and a value of ...

Preparation of the virus suspension: ...
which contained 0.1 ...
units of ...
liter of the virus suspension ...
ed this with 5 milliliters of ...
necting the column ...
lution, we started the elution. ...
cm, we obtained a flow velocity of 1 to 1.5 milliliters per minute. We
collected the eluent in fractions of 5 to 5.5 milliliters. All experi-
ments were conducted at room temperature.

Disinfection of the Sephadex gel: The column ...
gel was filled with sterile, chloroform-saturated phys. NaCl solution
on the day preceding an experiment. Then under the ...
forced this out of the gel quantitatively using chloroform-saturated phys.
NaCl solution, again taking precautions for sterility. A ...
located at the lower end of our chromatographic column so that we could
receive our fraction primary in a sterile condition. By observing ster-
ile precautions, it was no difficulty to achieve a germ-free work.

We disinfected the Sephadex column following an experiment with
virus using a 5% Formalin solution which we allowed to remain in the
column over night.

Determining degree of infectiousness: We determined the infec-
tiousness of the MMS virus through intra-peritoneal immunization of 7-
day old mice. The Teschen virus was titrated in swine kidney cultures,
NDV and TPV in incubated chicken eggs by injection into the allantoic
cavity or the chorion membrane. All titrations were conducted with di-
lutions in stages of potentials of 10. We computed the titers according
to Behrens and Kürber (13). They are based upon 0.1 milliliter of the
starting material described under "experiments and results" ($MID_{50} =$
 $Micc-ID_{50}$; $KID_{50} = Cultur-ID_{50}$; $EID_{50} = Egg ID_{50}$).

Analytical determination: The ammonium sulphate in the individ-
ual fractions was determined by distillation in a half micro-Kjeldahl
apparatus. Titration was determined with n/70 hydrochloric acid in
boric acid.

We performed the phosphate determination according to the molyb-
date method of Fiske and Subbarow (14), but used ascorbic acid as the
reduction medium.

In the separation tests of the virus, the initial ID₅₀ value was stable throughout the separation process. The separation and stability of the individual fractions with respect to the initial ID₅₀ value.

Removal of virus from

Of particular interest to us is the separation of the virus from virus suspensions with 1/3 M phosphate buffer, pH = 7.5, obtained from aluminum hydroxide gel (AG), and the separation of aluminum sulphate. The experiments carried out in the laboratory of desalting on this problem. We studied the desalting of the virus from the precipitate which, after precipitation with the virus, was precipitated from the precipitate in distilled water, contained in phosphate buffer or analogous virus material which was suspended in 1/3 M phosphate buffer, pH = 7.5.

Separation of MKS and Teschen virus from ammonium sulphate:

Experiment MKS-1a. MKS mouse material purified with chloroform was diluted 1:100 with 1/3 M phosphate buffer, pH = 7.5, and to 300 milliliters of this suspension we added, drop-wise and during shaking, 300 milliliters of saturated ammonium sulphate, pH = 8.4. The pH value of the mixture (50% ammonium sulphate saturation) then amounted to 7.5. After one hour at + 4°C, there developed a strong clouding which sedimented after 90 minutes in rotor 21 of a preparation Spinco-Ultracentrifuge at 20,000 r.p.m. After this sediment had been absorbed in 6 milliliters of distilled water, we allowed it to sit over night at + 4°C and cleared it for 15 minutes in the laboratory centrifuge prior to the start of experiment in the morning. The clear virus-containing remnant (5 milliliters) with a MID₅₀ value of 10^{-5.90} was used as the starting material for the desalting tests with the Sephadex gel G 25.

Experiment Teschen-3a. To 2.5 milliliter Teschen culture virus we added, drop-wise, 1.5 milliliter saturated ammonium sulphate, pH = 7.4. This virus suspension was centrifuged slowly for ten minutes for clearing. 5 milliliter of the clear, virus-containing remnant with a MID₅₀ value of 10^{-6.5} was used as the starting material for the desalting.

The two experiments MKS-1a and Teschen-3a are graphically portrayed in Illustration 1. The separation of the virus components from the saline gradients can be clearly seen in the elution diagram. The initial value of ID₅₀ is reached in one fraction for each of the two types of virus. If the virus titer of the individual fractions is compared to the final volume, it can be seen that no loss of virus occurred in the desalting. The agreement between the two tests, which was also shown in numerous other tests, is clear. The peak of the virus titer is also identical

of the virus suspension
volume ().

$V_t = 25$

The results of the experiments are shown in Illustration 2. The results show that the virus suspension is not affected by the desalting process. The virus is not lost during the desalting process and used 5 milliliters of the eluate for desalting with Sephadex gel.

We determined the initial titer in virus suspension for the MS virus and a ND_{50} of $10^{7.5}$ for the MS virus. The Sephadex column from the separation column is used for the desalting and sterilization which will be described later.

The elution diagrams of experiment 5a and 5b in Illustration 2 again show the clear separation of the virus gradient from the saline gradients. No virus loss can be seen. The difference between the peaks of the virus gradient and the peaks of the corresponding saline gradients in tests 5a and 5b is identical with the difference in the described tests 1a and 3a. This means that the distribution coefficients for ammonium sulphate and the secondary sodium phosphate/primary sodium phosphate (phosphate buffer, $p_H = 7.5$) have the same values.

Comparison tests with IPV, IPV and MS virus

In order to determine that the small viruses such as MS virus and Teschen virus were no different, but that the lipid-containing large viruses also behaved indifferently to Sephadex gel and that they were also identical in this respect, we compared the MS virus with IPV and TPV in additional tests. For this we used columns of lesser volume ($V_t = 25$ ml). This time the pre-treatment and elution of the gel was done with distilled water which was brought to a p_H value of 7.5 with a few drops of 1/10 n NaOH. The saline gradients, which here are certain electrolyte mixtures (culture mediums and egg liquids), were determined through conductivity measurements of the individual fractions.

Illustration 3 shows the results of these comparison tests. All three types of virus have the same elution volume. Their peaks are identical and no loss of virus occurred during the desalting. This demonstrated that the desalting is independent of the virus size and virus structure.

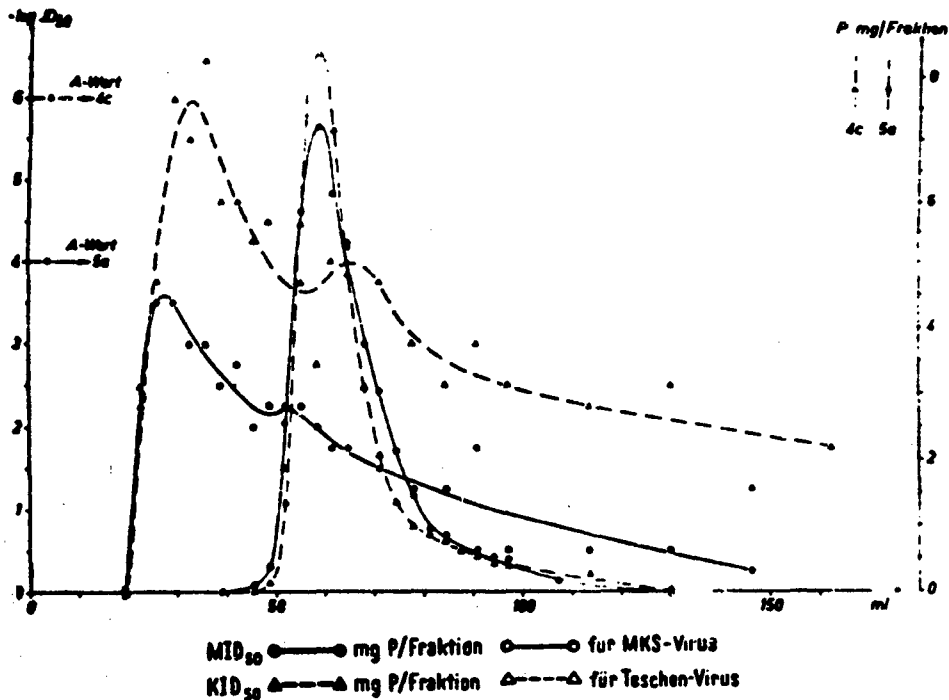


Illustration 2. Elution diagram of MKS and Teschen culture viruses in 1/3 m phosphate buffer, $p_H = 7.5$

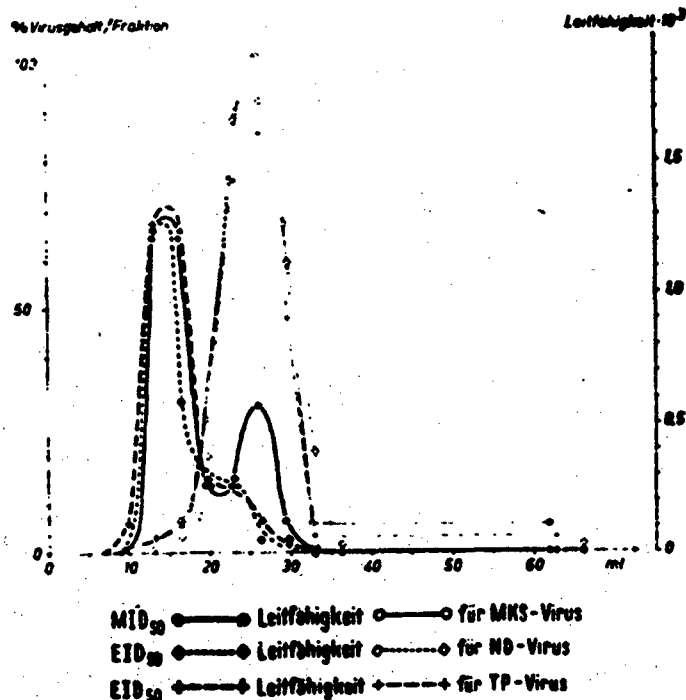


Illustration 3. Elution diagram of Newcastle disease, pigeon pox and MKS virus from physiological environments with distilled water, $p_H = 7.6$.

Legend: virusgehalt/Fraktion = virus content/fraction
Leitfähigkeit = conductivity

FIGURE APPENDIX

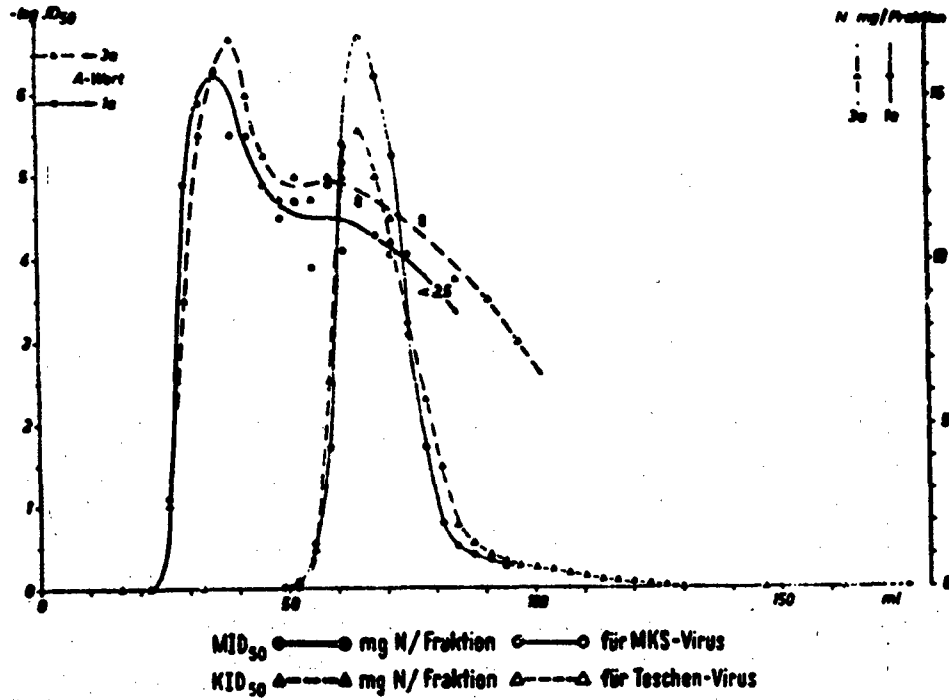


Illustration 1. Elution diagram of MKS mouse virus and Teschen culture virus in ~15% ammonium sulphate solution (experiments 1 a and 3 a).

Legend: A-Wert = initial value
 Fraktion = fraction
 für = for