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MULTISHOT THERMAL BATTERY

by

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A summary of the development of a m battery is designed to power the AN cycles.	nultishot therma N/PRC-77 radio f	l battery is presented. The or at least four ten-minute						
Test regimens are described and tes battery will meet electrical specif preconditionings such as heat, cold engineering drawings have been prep	st results detai Fications when s d, wind, vibrati Dared (a set is	led showing that the ubjected to environmental on and dropping. Detailed included in this report)						

20. ABSTRACT CON'T

and a set of operating instructions has been written (Appendix B, this report).

A brief description of some of the tests which led to the choice of cell chemistry is included. A chronological description of design improvements is given, from the first working prototype which was capable of four cycles to the final version which is capable of at least five reliable cycles.

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INTRODUCTION

The object of this study was the design and development of an externally heated multishot thermal battery which would power the AN/PRC-77 radio. The battery was to operate the radio for at least four cycles, each cycle consisting of five minutes of transmission followed by five minutes of reception. The battery was to be heated by a readily available military fuel such as trioxane.

Such a battery has been developed. When heated by trioxane, it is capable of operating the radio for as many as ten cycles. Activation times for the battery range from five minutes for the first few cycles to eight minutes for the final cycles.

The development of the battery was divided into three phases. Phase I involved the investigation of cell components such as anode, cathode and electrolyte. This phase also included investigation of the heat source and design of a battery case which would allow the cells to be heated quickly and efficiently. The culmination of Phase I was a proposed prototype design.

Phase II efforts centered on modification of the prototype design so that the battery would perform satisfactorily from an electrical and an operational standpoint. Burner height was increased to insure complete combustion of the fuel, cell spacing was adjusted for even heating of all cells, and design changes were incorporated which would make the system easier to use. Batteries were then subjected to environmental preconditioning and testing.

Based on these environmental test results, some minor changes were effected resulting in the final battery design. Phase III efforts centered on the production of twenty-five units of this design.

RESEARCH AND DEVELOPMENT PROGRAM

Phase I

Cell Development

The following materials were studied during the investigation of cell components:

- 1. Mg and Ca anodes.
- 2. LiCl-KCl and LiI-KI electrolytes.
- 3. CaCrO₄, FeS₂ and CuO depolarizers.

The cell of choice used a Ca anode, LiCl-KCl electrolyte and a CaCrO_{Λ}

cathode (depolarizer). The choice was based primarily on electrical performance and secondarily on prior experience with the system. Tests show that both the Mg anode and FeS₂ cathode would also prove acceptable.

Based on initial test data and some thermal calculations, it was determined that, to achieve rapid activation, each individual cell would have to be exposed to the hot combustion gases of the burning fuel. This meant hermetically sealing each cell in a separate case. Conventional battery construction wherein all cells are sealed in a single case could not be used because activation times would be extremely long.

Several cell configurations were investigated to arrive at a reliable, complete cell. Major problems which were overcome were: (1) swelling due to moisture and gaseous reaction products and (2) formation of Li-Ca alloy at a rate which caused internal shorting of the cell. The swelling problems were eliminated by vacuum drying all components and by heating each cell to its operational temperature before final sealing. The shorting problems were overcome when a "Fiberfrax" gasket was included in the cell to absorb any excess Li-Ca alloy.

The resulting cell, shown in drawing 405756 proved to be very reliable. No failures have been encountered in the testing of more than 100 such cells. The cell is capable of delivering one ampere at 510°C for more than one hour, equivalent to twelve five-minute transmit cycles.

The final cell uses two calcium bimetal anodes and two depolarizer pellets made from a homogeneous mixture of LiCl-KCl electrolyte, $CaCrO_4$ depolarizer, and "Cab-O-Sil" binder. A screen is included in the depolarizer to provide better electrical contact with the pellet. These components are assembled and inspected according to Flow Chart 405756. For more detail concerning the components and assembly techniques, the full drawing package should be consulted (Appendix A).

Fuel Evaluation

Both the standard trioxane tablet and the US Army Land Warfare Laboratory (USALWL) Delrin tablet were examined as fuels. The former was the fuel of choice because of its even burning characteristics and ease of ignition. However, Delrin tablets could be used as an alternate heat source.

Although the heat contents of trioxane and Delrin are very nearly the same (3980 cal/g and 4300 cal/g respectively), the Delrin is a denser fuel. Thus, with Delrin, more heat can be packaged per unit volume. However, the denser Delrin is also more difficult to light, and its burning rate is uneven and dependent on physical configuration of the tablets. It is these unfavorable characteristics which prompted the choice of trioxane as the primary fuel.

First Working Prototype

Initial Phase II efforts centered on the modification of the Phase I prototype design to meet the performance goals, and the testing of this modified design. Burner height was increased to allow the fuel to burn completely. Modifications were made to the spacing between cells and to the end cell geometry so that the hot combustion gases would heat each cell evenly and at the same rate. A sketch of this modified design is given in Figure 1. It was this design which was first able to meet the electrical specifications for the battery.

One of the first successful tests using this design is shown in Figure 2. The lower graph shows the performance of the full six-cell battery during the first cycle while the upper graphs give the performance of two individual cells.

One and two-thirds bars of trioxane were ignited at time zero and the battery voltage was allowed to rise under no load. At 3 minutes (point A), a 100 ma constant current load was applied which drove the battery voltage negative. By 4 minutes the battery was able to sustain the 100 ma load showing a voltage greater than 15 volts. At four minutes (point B) a one ampere load was applied, slightly depressing the cell output. Not until the temperature of the top of the cell cases reached about 350° C was the battery able to deliver 10.8 volts at one ampere (1.8 volts/cell). This occurred at 4 1/2 minutes (point C) and was considered activation time for the battery. Cell voltage rose to 15 volts during the next five minutes under one ampere load. During this period the trioxane flame turned yellow (point D) and the cap was placed on the assembly when the flame went out entirely (point E). At 9 1/2 minutes the load was diminished to 100 ma (point F) and the battery was allowed to run an additional five minutes. At 14 1/2 minutes (point G), the one ampere load was reapplied for 30 seconds to determine whether transmission was possible at the end of a ten minute cycle. This was indeed the case; although battery voltage began to fall, the voltage at the end of the 30 second period was above 13 volts. At 15 minutes (point H), the 100 ma load was reapplied while the battery cooled and the electrolyte froze.

Since peak temperatures were not excessive during cycle 1, it was decided that two full bars of trioxane could be used in the following three cycles. Graphs of battery output for cycles 2, 3, and 4 are shown in Figure 3. As the number of cycles increased, the activation time increased. The time necessary to reach 10.8 volts under a one ampere load are as follows:

Cycle	1	4	1/2	min.
Cycle	2	5	1/4	min.
Cycle	3	5	1/4	min.
Cycle	4	6	1/4	min.

Reaction products accumulate as current is drawn from the cell, requiring higher and higher temperatures to maintain the load. Peak voltages under



FIGURE 1. Sketch of six-cell battery with modified burner.



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load also decrease as the number of cycles increases.

Thermal data for cycles 2, 3, and 4 are not given in Figure 3. These data are shown in Table 1 along with cycle 1 values.

The battery was able to supply the required one ampere for four 5-minute cycles at voltages above 11.5 volts. The total time the battery was loaded at one ampere was 25.75 minutes. Total time under the 100 ma load was at least 30 minutes. It was this design (Fig. 1) which was used as the basis for the development of the final design with more uniform heating and improved ruggedness.

Final Design

During the last report period, March 1 to June, 1974, the following was accomplished:

- 1. The mechanical design of the assembly was finalized and engineering drawings were completed.
- Complete assemblies were tested under various environmental conditions and a fuel loading table was generated.
- 3. Twenty-five assemblies were completed for shipment.

Mechanical Design

Previous work had yielded a general configuration for the cell array and battery assembly. In order to make the assembly more rugged and capable of being handled repeatedly, it was necessary to make several design modifications. The major modifications were:

- A. Change from .015" to .032" steel outer case.
- B. Addition of a .032" steel reinforcing ring to the top of the outer case at the point of cell array attachment.

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- C. Change from .060" to .020" outer case "Fiberfrax" insulator lining.
- D. Addition of a .005" nickel sleeve covering "Fiberfrax" liner in outer case to eliminate tearing of Fiberfrax when the inner and outer cases are collapsed.
- E. Reconfiguration of air vent holes in fuel can to eliminate catching on edge of outer can when collapsed and to make case more rigid.
- F. Installation of "Fiberfrax" insulating pads on array frame to help prevent shorting in the event of cell shifting.
- G. Adjustment of inter-cell spacing for more uniform heating.

After several mock-up assemblies were built to determine usableness, changes A thru F were incorporated and assembly FD-1 was tested (FD indicates final



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6.2

6.3

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TABLE 1	
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									ŀ	
Cycle	1 - 1 2	2/3 bar	s trio	xane	Cycle :	3 - 2	bars	trioxa	ne	
Time	Temp	. (°c)			Time	Temp. (°C)				
(min)	End Ce	11	Cente	r Cell	(min)	End	Cell	Cente	r Cell	
	Up	Low	Up	Low		Up	Low	Up	Low	
1	65	144	80	173	1	56	135	86	148	
2	139	239	166	269	2	120	242	144	248	
3	235	328	246	338	3	221	329	232	317	
4	318	398	328	385	4	308	405	302	378	
5	362	464	361	449	5	350	472	351	439	
6	424	519	400	505	6	428	529	408	497	
7	467	533	449	512	7	496	590	462	558	
8	484	511	472	494	8	511	574	491	550	
9	478	486	475	479	9	517	543	505	528	
10	465	464	466	467	10	503	512	504	508	
11	447	443	457	455	11	485	487	495	493	
12	429	424	445	444	12	466	464	483	478	
13	412	406	433	432	13	443	444	466	464	
14	396	390	421	420	14	429	425	451	449	
15	382	376	410	408	15	411	407	436	434	
16	368	363	398	397	16	395	391	423	421	
17	354	350	387	385	17	380	376	409	408	
18	346	344	376	375						
19	341	338	365	365						
20	335	331	356	355						
21	329	321	347	347						
22	315	310	340	341						
Cycl	e 2 - 2	2 bars	trioxa	ne	Cycle 4 - 2 bars trioxa					
1	63	120	95	162	1	58	133	74	163	
2	119	194	166	254	2	97	240	135	264	
3	198	275	250	332	3	224	329	234	339	
4	277	349	324	396	4	308	393	306	395	
5	340	420	391	451	5	352	452	358	443	
6	392	490	425	511	6	421	523	426	513	
7	452	558	469	570	7	474	582	463	582	
8	513	603	511	605	8	518	577	509	577	
9	534	578	537	588	9	521	550	522	546	
10	531	550	549	570	10	511	520	520	524	
11	521	527	543	554	11	494	495	509	507	
12	504	503	535	539	12	476	473	493	491	
13	485	481	523	525	13	456	452	478	475	
14	468	462	510	511	14	437	433	462	459	
15	450	444	495	495	15	420	415	445	444	
16	434	428	481	482	16	403	398	431	429	
					17	389	384	418	417	

acaperatar a startob for bin out miller	Temperature	Profiles	For	Six-Cell	Array
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design). The results of this test and others are reported in Table 2. In this table are listed the start times along with battery voltages and cell temperatures. Start time is defined as time to reach minimum battery voltage under the transmit load (See point C in Fig. 2 as an example). Battery voltages are given at the ends of the 5-minute transmit and 5-minute receive modes (See points F and G in Fig. 2 as examples).

It was evident from the results of FD-1 (long start and low temperature peak in center cell) that the cell separation needed adjustment. Unit FD-2 was constructed using the final, cell-separation configuration (modification G above). The test results indicate the assembly is very well balanced thermally. Ten cycles were obtained from this unit, the first five being shown in Table 2. Engineering drawings were prepared reflecting the final design. Drawings 405782 and 405785 have been included in this report to show the configuration of the final assembly. For more detail, the complete drawing package should be consulted (Appendix A).

Environmental Tests

The following tests were performed to determine the ruggedness and operability of the final design. In all tests involving operation, the assemblies were stabilized at the stated temperature prior to the start of each cycle. Experimental data for the environmental testing are given in Table 3.

A. Drop Test - Drop tests in accordance with Contract No. DAAD05-73-C-0555, section F, para. 5 were performed on a final design mock-up model. Shifting of cells and subsequent shorting to array frame resulted. The frame of the array was insulated with "Fiberfrax". Subsequent drop tests on unit FD-1 indicate the assembly will survive a drop of 2 1/2 ft. without causing excessive cell shift or shorting.

B. Cold Condition - Assembly ET-1 (Engineering Test Sample Number 1) was placed in a temperature chamber and allowed to stabilize at -65°F for two hours. The assembly was operated while it was in the temperature chamber. The assembly was cycled five times. The voltage on the third cycle was low due to insufficient fuel loading, all other cycles performed satisfactorily.

C. Hot Condition - Assembly ET-2 was placed in a temperature chamber and stabilized at +110°F. The assembly was operated while it was in the chamber. Four cycles were completed satisfactorily.

D. Wind - Assembly ET-3 was placed in a tube with a chimney for hot gases (see Fig. 4). Air was forced through the tube by means of a variable speed fan. Velocity was measured with a flowmeter. Three cycles were completed. The first was underheated but the second and third were satisfactory. It was then noted that the chimney was providing favorable air current for the assembly. The fourth cycle was tested on the bench with air blowing directly on the assembly. Almost no operational life was achieved. The flame came out the vent holes in the burner can. The assembly was shielded from the air flow and a fifth cycle completed successfully.

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TABLE 2 Engineering Tests of Finel Design

			Initial	Bars of	Start Time	Volts at end of	Volts at end of	Peak Cell	Temp. (^o C)
S/N	Cycle	Conditions	Temp.	Trioxane	(Sec.)	5 min. transmit	5 min. receive	End Cell	Center Cell
FD1	1	Static	25°C (77°F)	1 1/2	385	10.4	12.8	453	380
	2	=	r	82	400	12.7	14.2	546	489
	3	=	=	2 1/3	435	11.3	13.7	574	515
	4	=	=	2 1/2	480	11.0	13.3	546	481
FD2	Ч	Static	25 °C (77 °F)	1 1/2	335	12.6	15.7	479	414
	લ્ય	=	÷	1 2/3	335	14.6	16.3	557	513
	3	=	=	Q	376	14.1	16.6	584	556
11	4	-	=	Q	400	13.0	13.0	578	552
	5	-	=	2 1/2	. 430	14.1	15.4	5 95	577

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Temp. (^o C) Center Cell		505	T	T	1	T		491	494	569	585		515	594	507	411	545
Peak Cell End Cell		554	r	ı	ı	X		545	538	620	622		487	538	484	373	536
Volts at end of 5 min. receive		16.2	15.5	12.5	15.0	13.6		16.0	13.8	14.3	15.8		*2.50	*4.25	*3.75	*0.50	*4.50
Volts at end of 5 min. transmit	Tests	14.6	14.1	*4.5	13.8	13.0	a Tests	14.4	13.2	14.2	14.1		57.T*	14.1	*4.4	0.0	*4.75
Start Time (Seo.)	emperature	300	350	400	450	460	Temperatur	305	315	385	350	Wind Test	210	150	210	*	310
Bars of Trioxane	Low T	1 2/3	1 2/3	1 2/3	2 1/3	2 1/2	High	1 1/3	1 1/2	Q	02		1 1/3	1 2/3	2	Q	ୟ
Initial Temp.		-54°C (-65°F)	=	=	=	=		-41°G (+105°F)	z	z	=		25°C (77°F)	=	=	=	=
Conditions		Static	E	÷	F	=		=	z	z	Ξ		Wind 880 ft/min.	=	=	=	=
Cyole		Т	ଫ	3	4	5		Ч	02	3	4		Т	2	Ю	4	5
S/N		ET1						ET2		12			ET3				

TABLE 3

Environmental Tests

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TABLE 3 (cont.)

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Environmentel Tests

ak Cell Temp. (^O C) d Cell Center Cell		,								•		
En												
Volts at end of 5 min. receive	×	14.6	*4.0	*3.4	*2.0	*4.0		15.6	12.3	12.0	11.8	
Volts at end of 5 min. transmit		13.4	*4.0	*4.0	*1.5	*2.5	st	14.3	11.5	11.4	*4.5	
Start Time (Seo.)	Field Test	330	360	420	420	480	'ibration T∈	320	365	430	450	
Bars of Trioxane		1 1/2	1 2/3	Q	2 1/2	2 1/2		1 1/3	1 2/3	Q	2 1/3	
Initial Temp.	c	25 °C (77 °F)	=	=	=	=		25°C (77°F)	=	=	=	
<u>Conditions</u>		S.West Wind	=	=	=	=		Pre- Vibration	=	=	=	
Cyole		1	~	3	4	5		г	രു	3	4	
S/N		ET4						ET5	1	3		

* Represents minutes of running time, end voltage was less than 11.25.

** Did not reach minimum voltage.

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FIG 4

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WIND TEST SETUP

E. Field Test - Assembly ET-4 was tested outdoors with wind conditions of 5 - 15 mph. The assembly was shielded on three sides. Four cycles were completed. Life was marginal.

<u>F.</u> Vibration - Assembly ET-5 was vibrated one hour on each of three perpendicular axis. The frequency was swept from 10 to 55 Hz and back to 10 Hz, once every minute of testing. The assembly was subsequently tested for four cycles of electrical operation.

The results of the environmental tests show the unit is capable of being operated over the temperature range of $-65^{\circ}F$ to $+110^{\circ}F$. The assembly will survive transportation vibration and repeated drops of 2 1/2 ft. or less onto a 2" fir plank. The assembly, however, must be completely shielded from air currents. This may be due in part to the fact that the trioxane fuel must be shielded from wind.

A table of correct fuel loading, Table I, Appendix B, was evolved using the results of the development and environmental tests.

Phase III

Following the completion of the design and environmental tests, twenty-five production units were assembled incorporating all modifications. Minor cosmetic modifications were made at this point, modifications such as addition of locator marks to aid in opening and closing the array.

A set of operating instructions was prepared and is included as Appendix B.

CONCLUSIONS

The feasibility of a multishot thermal battery has been demonstrated. Although the unit does not meet all the specifications of the original program, the major design objectives have been accomplished.

The following is a list of the more important accomplishments of the project:

- 1. Electrical requirements have been met without exceeding the desired size and weight limitations. (3" diam x 3" high; 1 1/2 lb.)
- 2. A highly reliable cell design has been generated. More than one hundred and fifty cells have been tested without failure.
- 3. A battery package has been developed which demonstrates the practicality of the multishot concept.
- 4. The battery is capable of operating over the full military temperature range ($-65^{\circ}F$ to $+165^{\circ}F$), and will withstand moderate vibration and shock.
- 5. The fuel of choice (trioxane) is a standard military item.
- 6. No special skills are required to operate the battery.
- 7. No problems were encountered in interfacing the battery with the AN/PRC-77 radio.

RECOMMENDATIONS

Some of the areas which might be considered to improve the present design are:

- 1. Improved ruggedness.
- 2. Less sensitivity to wind.
- 3. Ability to operate continuously for periods greater than ten minutes.

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APPENDIX A

DRAWINGS

USED ON		REVISIONS		~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	5773.
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APPENDIX B

OPERATING INSTRUCTIONS

OPERATING INSTRUCTIONS

This device is a reusable power supply intended to power the AN/PRC-77 radio for a period of 10 minutes.

The major parts of the supply are shown in Figure 1.

TO OPERATE

A. Preparation:

- (1) Remove sealing tape from around end caps (save for resealing).
- (2) Remove and save end caps.
- (3) With a slow, steady, straight motion separate the burner can from the battery can by pulling on the pull ring on the bottom of the burner can.
- (4) Load trioxane fuel into the burner can. Break each bar into 2 or 3 pieces as required. The amount of fuel to be used varies with the temperature and the number of times the device has been used. Table 1 provides the correct fuel loadings.

Two Delrin tablets (interlocked in an X configuration) may be used as an alternate fuel supply when trioxane is not available.

- (5) Gently, insert the burner can with fuel into the battery can so that the triangular position indicator mark on the burner can aligns with the triangular mark on the battery can. (See Fig. 2). The lower end of the cell stack end plates will now rest on the cell stack support pads.
 - NOTE: Do not exert force on assembly as damage will result. Make certain support pads contact the cell stack support bracket and not the cells.
- (6) Plug leads into proper connector. The multipin jack (Power receptable J4) is inserted in the radio. The minature banana plug on the red wire is inserted in (+) terminal of the battery. The plug on the black wire is inserted in the (-) terminal of the battery.

NOTE: Improper connection of jacks may damage radio.

- (7) Place the power supply assembly on a level surface.
- (8) Completely shield assembly from wind.

NOTE: Do not overload fuel as this will result in serious damage to the power supply.

B. Operation:

- (1) Ignite fuel through burner can vent window.
- (2) After ignition of fuel the radio operator should listen for receiving noise. Receiving noise should be heard approx. three minutes after fuel ignition. The noise indicates the supply is beginning to supply power (power is not sufficient to transmit). Two minutes after receiver noise is acquired the power supply will be able to supply transmission power.
 - <u>CAUTION</u>: Power supply parts become extremely hot after fuel ignition. Parts remain too hot to touch for 40 minutes.
- (3) When fuel is first ignited it will burn with a clear or blue flame. After several minutes the flame will turn yellow and go out. Immediately after the yellow flame goes out the top end cap must be placed over the battery can to prevent excessive heat loss.
 - <u>NOTE</u>: Position top end cap so the connector notches align with connectors. Make sure cap does not touch connector pins as a short circuit will result.

C. Storage:

(1) After use, the top end cap should be removed to allow power supply to cool down faster.

<u>CAUTION</u>: Cap is extremely hot.

- (2) No attempt should be made to handle assembly for a period of 40 minutes following ignition of fuel. After 40 minutes the assembly should be sufficiently cooled for closing. No attempt to add fuel should be made while the unit is hot. If unit is to be reused while still warm to the touch, column 3 of the fuel loading chart must be used.
- (3) To close assembly pull out burner can and align the triangle with the circular position mark on the battery can. (See Fig. 3.) Gently push cans together. Replace end caps and re-tape. Top end cap should be rotated so the connector notches do not align with connectors.

TABLE 1

Fuel Loading (Trioxane)

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Number of Times Previously Used	Bars at Temp. Less Than O ^O F	Bars at Temp. OF to 90°F	Bars at Temp. Greater Than 90 ⁰ F
0	1-2/3	1-1/2	1-1/3
1	1-2/3	1-2/3	1-1/2
2	2	2	2
3	2-1/3	2-1/3	2
4	2-1/2	2-1/2	2-1/3
5	2-2/3	2-2/3	2-1/2





OPERATING POSITION FIG 2



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