AD-759 217

### NOMEX ULTRAVIOLET INHIBITORS

John F. Krasny, et al

Gillette Company Research Institute

Prepared for:

Aeronautical Systems Division

December 1972

DISTRIBUTED BY:

National Technical Information Service U. S. DEPARTMENT OF COMMERCE 5285 Port Royal Road, Springfield Va. 22151

## AD 759217

ANT ANT ANT

THE PARTY OF THE P

SHALSON US

ASD-TR-72-104

### NOMEX ULTRAVIOLET INHIBITORS

JOHN F. KRASNY AND A. M. SCHWARTZ THE GILLETTE COMPANY RESEARCH INSTITUTE

**TECHNICAL REPORT ASD-TR-72-104** 

DECEMBER 1972



37

WITH STORE

Approved for public release; distribution unlimited.

Reproduced by NATIONAL TECHNICAL INFORMATION SERVICE U S Department of Commerce Springfield VA 22151

LIFE SUPPORT SYSTEM PROGRAM OFFICE AERONAUTICAL SYSTEMS DIVISION AIR FORCE SYSTEMS COMMAND WRIGHT-PATTERSON AIR FORCE BASE, OHIO 45433

### NOTICE

When Government drawings, specifications, or other data are used for any purpose other than in connection with a definitely related Government procurement operation, the United States Government thereby incurs no responsibility nor obligation whatsoever; and the fact that the Government may have formulated, furnished, or in any way supplied the said drawings, specifications, or other data, is not to be regarded by implication or otherwise as in any manner licensing the holder or any other person or corporation, or conveying any rights or permission to manufacture, use or sell any patented invention that may in any way be related thereto.

White Section Buff Section	
AMARIN GOS	_
	Patt Section

Copies of this report should not be returned unless return is required by security considerations, contractual obligations, or notice on a specific document.

AIR FORCE/56780/25 April 1973 - 100

DOCUMENT CONTR (Security classification of tills, body of abstract and indexing and 1. ORIGINATING ACTIVITY (Corporate author) The Gillette Company Research Institute 1413 Research Boulevard Rockville, Maryland 20850 3. REPORT TITLE Nomex Ultraviolet Inhibitors 4. DESCRIPTIVE NOTES (Type of report and inclusive dates)	notation must be entered	Unclassified
(Security classification of tills, body of abstract and indexing and 1. ORIGINATING ACTIVITY (Corporate author) The Gillette Company Research Institute 1413 Research Boulevard Rockville, Maryland 20850 3. REPORT TITLE Nomex Ultraviolet Inhibitors 4. DESCRIPTIVE NOTES (Type of report and inclusive dates)	notation must be entered	Unclassified
The Gillette Company Research Institute 1413 Research Boulevard Rockville, Maryland 20850 3. REPORT TITLE Nomex Ultraviolet Inhibitors	2 <b>8</b> . A 1	Unclassified
<ul> <li>1413 Research Boulevard Rockville, Maryland 20850</li> <li>REPORT TITLE</li> <li>Nomex Ultraviolet Inhibitors</li> <li>4. DESCRIPTIVE NOTES (Type of report and inclusive dates)</li> </ul>	2b. G	
Rockville, Maryland 20850 3. REPORT TITLE Nomex Ultraviolet Inhibitors 4. DESCRIPTIVE NOTES (Type of report and inclusive dates)	26. G	ROUP
3. REPORT TITLE Nomex Ultraviolet Inhibitors 4. DESCRIPTIVE NOTES (Type of report and inclusive dates)	l	
Nomex Ultraviolet Inhibitors 4. DESCRIPTIVE NOTES (Type of report and inclusive dates)		
4. DESCRIPTIVE NOTES (Type of roport and inclusive dates)		
4. DESCRIPTIVE NOTES (Type of report and inclusive dates)	• • • • • • • • • • • • • • • • • • • •	
Final Report, April 1970 to September 1972		
5. AUTHOR(5) (First name, middle initial, last name)		
John F. Krasny and Anthony M. Schwartz		
6. REPORT DATE	. TOTAL NO. OF PAGE	S 75. NO. OF REFS
December 1972	30	93
BR. CONTRACT OF GRANT NO.	. ORIGINATOR'S REPO	
F33657-70-C-0858		
A. PROJECT NO.	ASD-TR-72-104	1 m - 2
c	5. OTHER REPORT NO(S this report)	a) (Any other numbers that may be easigned
d.	None	
10. DISTRIBUTION STATEMENT	NOUG	
Approved for public release, distribution is	unlimited.	
11. SUPPLEMENTARY NOTES	SPONSORING MILITAR	ACTIVITY
	Life Support Sys	stem Program Office
	Aeronautical Sys	stems Division (AFSC)
None		n AFB, Ohio 45433
IS. ABSTRACT		
'/ Investigation has been made of several a resistance of Nomex fabric and webbing to act consisted in applying to the Nomex a material in the range 360 - 390 nm, which is known to the photochemical degradation. Several mater this primary requirement, but relatively few stability, safety in application and use, and on the fabric. A major problem proved to be bonding of the protective material to the Nom by applying a pigment that was opaque to the black) in a lacquer type polymeric "tie-coat" used as the major ingredient of the tie-coat. As a stipulated part of this investigati the technical literature relating to actinic is included as an appendix.	tinic degradation that absorbs a be the waveleng rials are availar meet the added i lack of undesi the durable and mex. Best result harmful radiati or binder. A	on. All of these actinic radiation gth involved in able that meet requirements of trable effects i satisfactory its were obtained on (carbon- vinyl resin was

DD . FORM .. 1473

いれていた

の言語を見たいないであるというない

Nomex* fiber	ROLE	WT	ROLE	WT	ROLE	*
Nomert fiber						
Nomey* fiber	1 1					Y
NUMER. LIDEL						
Harness Webbing						
Parachute Pack Cover Fabric						
Ultraviolet Inhibitor						
Actinic Degradation						
*Registered trade mark for E. I. du Pont						-
deNemours high temperature resistant aromatic polyamide						
			and the local division of the local division			

ASD-TR-72-104

### NOMEX ULTRAVIOLET INHIBITORS

•1

JOHN F. KRASNY AND A. M. SCHWARTZ THE GILLETTE COMPANY RESEARCH INSTITUTE

Approved for public release; distribution unlimited.

### FOREWORD

This final technical report was prepared by The Gillette Company Research Institute, 1413 Research Boulevard, Rockville, Maryland 20850, under Contract F33657-70-C-0858, Nomex Ultraviolet Inhibitors. The program was sponsored by Life Support System Program Office. Technical direction was provided by the Nonmetallic Materials Division, Air Force Materials Laboratory, Directorate of Laboratories, with Mr. P. C. Opt as Technical Monitor.

The work herein reported was performed during the period April 1970-September 1972. This report was submitted for approval November 1972.

This technical report has been reviewed and is approved.

Milliam M

WILLIAM M. QUINN, Colonel, USAF System Program Director Life Support System Program Office Deputy for Subsystems

### TAELE OF CONTENTS

		PAGE
I	Introduction	1
11	Methods and Materials	3
	A. Fabrics and Webbings	3 3 4
	B. Protective Systems	3
	C. Methods of Application	
	D. Exposure to Carbon Arc Weather-Ometer	4
	E. Evaluation	5
III	Experimental Results	5
	Systems Containing Extenders	6
	Systems Containing Elemental Iodine	8
	Pigmented Tie Coat Systems	9
	Semi-Plant Scale Applications	9
IV	Conclusions	10
v	Tables	11
VI	Appendix I	
	List of Products Used	18
VII	Appendix II	
	Literature Search	20
VIII	References	21

### Preceding page blank

### I. INTRODUCTION

The aromatic polyamide fiber, Nomex<sup>\*</sup>, is of particular interest for a number of Air Force applications because it exhibits some of the physical characteristics of nylon, but has much higher thermal stability; does not support combustion; and does not melt when in contact with a flame. However, Nomex is susceptible to serious actinic degradation from natural and artificial light sources. Items such as parachute packs discolor and lose strength from exposure to sunlight when stored in ready-to-use position in aircraft on alert. The present program was initiated to investigate the use of chemical systems to reduce actinic degradation of Nomex.

Basically, the program consisted of an extensive literature search (Appendix II) and an experimental phase. In the literature search emphasis was placed on the actinic degradation of Nomex and on means to overcome it. However, references to the actinic degradation of <u>all</u> polyamides, not only to the aromatic polyamide, Nomex, are included.

It became apparent in the beginning of this program, as a result of our literature search, discussions with duPont, and some of our preliminary experiments, that Nomex absorbs light at a higher wavelength than most other polymers. The degradation of the strength of Nomex occurs within the wavelength region of 310 - 390 nm with a peak between 360 - 390 nm. Since most commercial u.v. absorbers rapidly lose effectiveness at wavelengths above 340 - 365 nm, extension of the effective protection range was required. This might logically be accomplished by including with the u.v. absorber a material which absorbs strongly in the 360 - 390 nm region. We shall refer to such materials as extenders. Based on this information, a series of compositions was selected for application to Nomex fabrics and webbings. These consisted of commercial u.v. absorbers, alone and in combination with various extenders.

Methods for applying such compositions to Nomex have not previously been described, and their development was accordingly a pioneering effort. Various techniques analogous to dyeing were tried. These included applications from aqueous baths with and without carriers, and from swelling and nonswelling \*Registered trade mark for E.I. du Pont de Nemours high temperature resistant aromatic polyamide.

solvents. Best adhesion of the protective systems to Nomex was achieved by compounding them with resins, plasticizers, and heat stabilizers to form a paint which could be applied as a coating to the fabric. The binder portion of this paint (polymer, plasticizer, etc.) is referred to as the "tie coat" or "tie coating". Extenders were found which provided good protection against actinic light. These materials which (like the uv absorbers) were soluble in the vinyl resin tie coating mixture, bloomed out of it on drying. An attempt was then made to use insoluble pigment protective systems in the tie coating. The first pigment tried, carbon black, resulted in good adhesion, no blooming, and promising protection against actinic degradation.

The following properties were evaluated:

Effectiveness against actinic degradation of the Nomex, as measured by the retention of load-elongation properties of yarns after carbon arc exposure;

Load-elongation properties of the Nomex fabric and webbings after treatment: These do not always remain constant, e.g., application of uv absorbers from swelling solutions changed the load-elongation characteristics so extensively that fabrics and webbings would have to be reengineered;

Resistance to laundering, solvents, and abrasion or effectiveness of treatments (tumble drying and passing through buckles were employed as simulants of in-service abrasion);

Discoloration: compromises had to be made in this area because compounds which protect against radiation in the 360 - 390 nm region are yellow, and protective pigments also discolor the fabrics and webbings;

Flexibility: this was a problem with the resin coatings but could be minimized by the use of plasticizers;

Stability in storage: this presented great difficulties in some of the most successful protective systems applied with resin tie coatings;

Medical safety in both application and use.

This final report summarizes the technical effort leading to the p-lot scale application of the treatment using carbon black pigment in a tie coat. The treatments which were eliminated during the lengthy effort because of one or more deficiencies are described only briefly. The p-oducts used a this work are described in chemical terms in Appendix I.

### II. METHODS AND MATERIALS

### A. Fabrics and Webbings

The Nomex yarn required to accomplish this program was purchased from E. I. duPont de Nemours and Company, Inc. by our weaver subcontractors. The Nomex webbing was purchased from Alliance Webbing, Inc. I: conferms to MIL-W-38283 (U.S.A.F.) specification, Type XII, sage green, 200 denier, with the finish omitted. The Nomex fabric was purchased from Stern and Stern Textiles, Inc., according to MIL-C-38351 specification, Type II, Class 1, sage green, again with the finish omitted. A similar white fabric was obtained from the same manufacturer, and used in much of the screening effort, because the white fabric was more subject to actinic degradation than the sage green fabric and discoloration could be more easily detected. In addition to untreated Nomex fabrics and webbing, nylon fabric (MIL-C-7219, Type III) and webbing (MIL-W-4088, Type XXII) were used as controls in the carbon arc exposures.

### B. Protective Systems

The protective systems are listed in Table I, along with information pertaining to the method of application. Basically, they consisted of commercial u.v. absorbers; "extenders", i.e., compounds which offer protection in the wavelength range in which Nomex is most sensitive, 360 - 390 nm, and which is beyond the effective range for commercial u.v. absorbers; elemental iodine; protective pigments; and specially synthesized adducts of polyethyleneimine and commercial u.v. absorbers.

### C. Methods of Application

The original contract called for treatment and carbon arc exposure in yarn form. However, this was found difficult, because of excessive fraying of the yarns, and fabric samples were treated except at the inception of the work. In crossover tests, the yarns from the fabrics seemed to suffer at least as much actinic degradation as those exposed in yarn form. All fabrics and webbing were cleaned by repeated immersion in solvent.

Most applications of protective systems were made by padding, double dip-double nip, or triple dip, triple nip, followed by slow drying and exposing the sample in an oven to the melting temperature of the tie coating, u.v. absorber, or extender, to obtain better coverage and adhesion. In the case of the adduct of an uv absorber and a polyethyleneimine, samples were immersed in beakers and shaken. Other samples were exposed to iodine vapor in closed systems. In the padding treatments several different solvents were used. These included both swelling and non-swelling solvents and carrier solvents for the actinic absorbing materials. Times and temperatures of immersion were varied. In some of the treatments a resinous tie coat was used to improve retention of the actinic absorber by the fiber.

### D. Exposure of Nomex to Carbon Arc Weather-Ometer

For exposure to the carbon arc light (Fed. Std. 191 Textile Test Methods 4804 and 5804), the fabrics were cut into strips with the warp in the length direction. Most samples were exposed at both ambient conditions in the Weather-Ometer, as well as at approximately 100% r.h. This latter was accomplished by placing 30 ml of distilled water in Pyrex test tubes, 38 mm in diameter. The samples were suspended between two binder clips and lowered in the tubes so that they remained above water level. A copper wire was attached to the lower clamp to prevent the sample from twisting during exposure. The upper clamp was suspended from the wall of the tube. The tube was capped with aluminum foil, and attached to the Weather-Omegader with tape.

### E. Evaluation

The effect of exposure was assessed by removing yarns from the fabric and recording their load-elongation curve on an Instron Tester. Test conditions were two inch gauge length and 33%/minute extension rate. The breaking load was read from the recorder chart.

The shapes of the load-elongation curves for exposed and mexposed Nomex yarns were similar, except for the breaking point, so that load at break sufficiently characterizes the yarns. Yarns exposed to a swelling solvent, e.g., LiCl in DMF (dimethylformamide) showed changed load-elongation characteristics even before carbon arc exposure. The durability of the treatments was determined by one cycle of laundering and tumble drying in home launderinequipment according to AATCC Test Methods 124-1969, I.B. In some cases, webbing was passed five times through the buckle of a parachute harness, to test the effect of the abrasion under typical use conditions.

### III. EXPERIMENTAL AND RESULTS

The systems applied to Nomex fabric and webbing are summarized in Table I, along with information on their advantages and disadvantages.

Two systems, one containing u.v. absorbers and extenders in resin tie coatings, and the other elemental iodine, offered good protection against actinic degradation. However, neither system could be recommended because the protective agents proved to be toxic, and could not be affixed to the Nomex so that they would not rub or flake off. The extenders bloomed from the resin coating, i.e., they continued to emerge from the surface during prolonged storage even after being removed repeatedly by leaching or laundering. The iodine also bloomed, even when covered with a resin coating. Both systems could rub off on skin or could allow escape of noxious vapors into the atmosphere.

Application of u.v. absorbers and extenders from numerous solutions was tried, and was generally unsuccessful. Somewhat promising results were obtained when application was made from solutions which swell Nomex, hot

DMF and DMF with 1% LiCl. However, these treatments reduced the modulus and yield point, and increased the elongation at break of the yarn even before exposure, though they did not greatly affect the breaking load. This approach was abandoned, because it would require re-engineering of the Air Force fabrics and webbings to obtain the desired load-elongation characteristics.

A system consisting of a tie coating and dispersed carbon black has afforded the best all around result of any of the systems tried to date. It was fully stable, exhibiting no blooming or exudation. It gave reasonably good protection on webbing (Table II) although the protection on fabric was marginal. Carbon black was the first pigment chosen for trial when it was decided that pigmented resin systems should be examined. Since this decision was made late in the program, carbon black is the only pigment tried to date. It is used in the pigmented resin system being applied to 250 yards of webbing and 100 yards of fabric in fulfilling one of the requirements of this project.

### Systems Containing Extenders

The extenders were selected from tables in the "Handbook of Organic Structural Analysis"\* which lists the chemical composition as well as the log maximum absorption coefficient and the wavelength at which it occurs for a wide range of organic compounds. The extenders with which most work was performed were:

> DMAB - p-dimethylaminoazobenzene ADPH - acetone-2,4-dinitrophenylhydrazone PNA - p-nitroaniline PPAA - p-phenylazoaniline

Applications of DMAB and ADPH, together with a commercial uv absorber, from solvents and from carriers resulted in some improvement of the resistance to actinic degradation, but adhesion of the protective systems seemed \*Y. Yukawa, Editor, W. A. Benjamin Inc., New York, N. Y., 1965.

poor. Consequently, it was decided to apply these systems in a tie coating, i.e., from solutions containing the u.v. absorber, the extender, and a resin. Table III shows that these latter treatments resulted in quite effective protection. On the basis of the results, DMF was chosen as the preferred solvent, and VMCH polyvinyl chloride terpolymer as the preferred resin. Geon 222 coatings appeared to be less effective, and Butvar B98 polybutyral tie coatings are reported to be more flammable than polyvinyl chloride (however, this resin is presently applied to nylon webbing for the Air Force). VMCH was found, however, to have poor drycleaning resistance. ADPH used as the extender seemed to provide better adhesion than DMAB and to be somewhat less yellow. ADPH is also cheaper than DMAB, and was chosen as the preferred extender for further studies. The next objective was to overcome the stiffness of resin coated fabrics and webbings, by means of plasticizers. Phosphate-based plasticizers were chosen because of their known flame resistant characteristics. The stiffness of present nylon webbing, Mil-W-27265, CON R could be matched with these plasticized formulations. Addition of a heat stabilizer, dibutyltin-di-2-ethyl hexoate, eliminated tackiness. Results of carbon arc exposure of such systems are shown in Table IV.

The following was found:

Protection against the effects of carbon arc exposure was afforded •by these tie coating systems on both webbing and fabric, unlaundered and laundered, exposed at ambient conditions and at 100% r.h., and after abrasion by passing webbing repeatedly through a harness buckle.

Relatively high concentrations of Cyasorb uv 24 and ADPH were needed to provide protection, although recommendations for normal use of uv absorbers rarely exceed 1%.

Heating the bath did not improve the effectiveness of the treatments.

Choice of plasticizers or their concentration did not affect the results; concentration could then be chosen on the basis of the desired stiffness.

The unlaundered fabrics exhibited blooming, i.e. flaking off of the ADPH. The bloom was removed by laundering and did not recur immediately, but reappeared after prolonged storage. Because ADPH is toxic the treatment was not practical.

It was thought that perhaps a tie coating in which the extender was soluble in the plasticizer (ADPH was not) would not exhibit blooming. The solubility of several promising extenders in plasticizers was determined, and tie coatings containing two such products, PNA and PPAA, were applied. Table V shows the results; again, reasonable protection could be obtained but blooming persisted.

### Systems Containing Elemental Iodine

It is reported (by the personnel at the duPont company) that elemental iodine protects Nomex from actinic degradation. We applied iodine to test the concept, in the hope that we could firmly attach it to the fabrics either by treatment from a swelling solution or by applying a tie coating on top of it. However, all such treatments resulted in flaking off, sublimation, and/or blooming of the iodine. They could not, therefore, be considered practical because of iodine's toxicity. The results of the carbon arc exposure are given in Table VI. Application from hot DMF provided good protection but resulted in decreased modulus and yield point, and increased elongation at break, before and after carbon arc exposure. Obviously, this change in load-elongation properties would require re-engineering of the fabric and webbing; if it were not due to molecular reorientation, but due to retained solvent which would eventually evaporate, toxicity problems as well as undesirable aging effects would be encountered. Exposure to iodine vapor worked well in the laboratory but would present serious difficulties on a plant scale; we know of no installed equipment which could be used. In addition, iodine has permeated all of the subsequently applied tie coatings that have been tried to date, and it is evident that containing this reactive element would be a problem.

and the second second

### Pigmented Tie Coat Systems

The various extenders which were tried gave good uv protection, but failed in that they were not retained by the tie coating. It was decided to try a pigment to replace the extender, because pigments opaque to the harmful wavelengths of light could be found, and these pigments would presumably be retained by the tie coating. Carbon black was tried first. It is a common paint pigment, retained well by most polymeric coating materials, is opaque to visible and near ultraviolet light, and was therefore expected to provide protection.

The carbon black selected for use was available commercially as a dispersion with Vinylite VYHH in methyl ethyl ketone (see Appendix I). Three mixtures were made up based on this material, differing from one another in < ncentration of resin, plasticizer and pigment. These mixtures were applied to fabric and webbing in the same manner as previous tie-coating mixtures, i.e. by padding in three passes at room temperature. The treated goods were cured at 150° C for 5 minutes. The treatment appeared to be quite uniform, resulting in an even coating. No blooming or leaching in water occurred with either fabric or webbing. Test results are shown in Table II. On the basis of these results the composition of Sample # 1451-74 was chosen for semi-plant application. This composition contains 5% carbon black, 7.5% Vinylite VYHH (vinyl chloride-vinyl acetate copolymer), 20% plasticizer, 7% u.v. absorber and 1% stabilizer.

### Semi-Plant Scale Applications

The composition of Sample #1451-74 was applied to 250 yards of the webbing and 100 yards of the fabric. The apparatus consisted of two double roll padders in series separated by a knife scraper. The material was passed through at 10 feet per minute and a roll pressure of 3.5 pounds. It was taken up on a screen rack and passed into a solvent recovery chamber, held at 150°F to eliminate all the solvent, then passed to a curing oven, and finally cooled and rolled up. This was essentially identical to the procedure used in the laboratory applications.

The webbing took the treatment satisfactorily. The fabric was splotchy in appearance, and the finish was obviously distributed unevenly. The degree of unevenness was difficult to estimate and it is quite possible that the uv protective effect did not vary from point to point on the surface nearly so much as the appearance. In any event the finish obtained in this semi-plant run was much less uniform than that obtained with the identical formulation in the laboratory. This is ascribed to at least two factors, neither of which is inherent in the process. One is that the fabric was too wide for the padder that was used, and was put through folded. The second is that the solvent evaporated too fast for the speed of throughput; i.e. the rate of solvent evaporation was not matched to the fabric speed nor to the roll pressure. It is very probable that this could be corrected by adjustment and/or by modification of the solvent.

### IV. CONCLUSIONS

Materials which absorb strongly in the wavelength region of 360 -390 nm afford protection to Nomex against actinic degradation, when they are applied as a coating to the fabric surface. Of the numerous materials examined the only one that meets subsidiary requirements and appears to afford protection against actinic degradation is a coating consisting essentially of carbon black pigment in a plasticized vinyl resin vehicle. It is expected that this type of coating ultimately will prove optimal, but that it can be improved upon, presumably along the lines of more effective pigmentation and binders of greater all around suitability.

TABLE 1

ē,

# SUMMARY OF EXPERIMENTAL U.V. PROTECTIVE SYSTEMS FOR NOMEX

Tratective Agents	Solvent	Protection	Resistance to Laundering and Abrasion	Discoloration	Variables Investigated	Tree	and the second
A. Protective Figuents in Tie Costing	in The Conting						TIOGON DI ATTRACA
Cerbon black MEK Poor to modes B. U.V. Absorber and Extenders <sup>2</sup> in the Constant	HEK anders <sup>2</sup> in Ti	Poor to modest	70	ł	Cone. of carbon black resin and plasticizer.	Acceptable hand cen be attained by use of plasticiter.	Final Report
Cyaserb UV-24 + ADMI - + 1000 - + 110	Ħ	1	fair to good	ally	Conc. of Cyasorb extenders, resin, plasticizers; best cypes; plasticizer types; treatment time/tenperature	Bloowing and leathing in water of extenders makes systems unacceptable	Final and Moachly Reports No. 10-14 10, 11 15, 19 18, 19
C. Elemental lodine							
From solution Vapor freetweet	120	good poor good	1	brown to gold	time/temperature	Danger of sublimation makes fodine medically unacceptable; hot Dir changes load-elonge-	Firal and Hosthly Reports Ro. 10, 11
D. U.V. Absorber-PEI Adduct					after vapor treatment	Konex	
121-61-60	a a a a a a a a a a a a a a a a a a a	ł	1	1	PEL mol. vte.; degree of aubattu- tion of adduct. conc time/temo.	Not promising	Quatterly Report No. 3, Monthly Report No. 7
E. U.V. Absorbers and Extenders in Solvent	tendare in Sol	vent					
Cyssorb UV-24 and ADTE Cyssorb UV-26	warfouk swelling a non-svelling solv	various swelling and non-swelling solvente;	poor to slight	aller	Conc.; add"-ton of "quencher" time/temperature	Not presides	Quarterly Report N. 3 and Monthly Report No. 7.
Timeria 320 Dvieul DS-49	Various au Bon-swellfi carriers	various suelling and non-aveiling solvents;	Poor	1	conc.	prov f Nonex	~

Q.

FVC resine, phosphate plasticizers, heat stabilizer.
 Extenders: compounds with peak absorption between 360 - 390

TABLE II

•

EFFECT OF CARBON ARC EXPOSURE ON NOMEX TREATED WITH CARBON BLACK AND U.V. ABSORBER DISPERSED WITH PVC RESIN, PLASTICIZER, AND HEAT STABILIZER. 50 HOUR EXPOSURE. SAGE GREEN FABRIC AND WEBBING.

		+	Treatment				Strength		Yarn Breakine Lond/The.	a load	The.		Tinnestion at Brash	-	
No.	Substrate	Renta <sup>2</sup>	Plasticizer3	Carbon	2 sh	Hand	not Laundered DRY Only	Not	Not Laundered Dry 100% RH	Laund	Laundered + TD Dry 1001 8H	Not L	Not Laundered Dry 1002 Bil	Dry	Laundered + TD Dry 1002 RH
89:15	Tabric	11.5	10	2.5	ĩ	Very	5	2.8	2.7	I	1	-	16	:	1
21-78	Vebbing				ñ	Sette	55	19.8	1	1	-1	43	1	:	:
1451-70	Pabric	-	2	5.5	8	SHEE	28	3.2	2.8	2.6	2.7	52	51	1	16
62-15	Vebbing				2		69	19.0	1	1	1	40	1	:	1
1451-74	Pabric	1.3	8	•		Cood	26	1.1	2.8	9.0	2.7	4	15	20	51
21-80	Vebbing				=	2	55	20.0	1	1	1	42	1	1	•
1451-58	Fabric					•	11	2.9	2.5	2.8	2.6		9		1
29-15	Webbing				1	1	2	15.5	1	1	1	20	•	;	1
1451-81	<b>Tabele</b>			1			100	1.4	1	1	1	95	•	1	1
29-12	Vebbing				1	1	100	21.5	I	1	1		1	1	1
1	1) Treatment Includes:	K H	Cyssorb UV-24 fibury1 cin-di-	di-2-athyl herests, heat -	Textoor D	- 14	stabilizer								

A bakelite reain VYEN-copolymer of vinyl chloride and vinyl acetate (67:13)
 Plasticizer was Santicizer 148 - isodecyl diphenyl phosphate

TABLE III

行けつ

ş

EFFECT OF CARBON ARC EXPOSURE ON NOMEX TREATED WITH U.V. ABSORBERS, EXTENDERS, AND RESINS. 50 HOUR EXPOSURE; WHITE FABRIC, SAGE GREEN WEBBING.

		Treatment												
Sample No.	Extenders	Polyner Couring	Solvent	Add-on	Not Laundered	-	Not laundered Laundered	Laund	Laundered	Not L	undered	Not Laundered Laundered	1	
1.111	52 U.V. 24.				T				NA TOO	E	LOUT BH	DEV 1002 RH	HI TO	Itaria
: :	ST Dras	1	Ttheno1	1.0	53	2.0	2.7	4.2	3.8	•			-	
1	57 1548	I	Ithanol	1.4	1	1.1	3.2	2.2	3.0		1			Gold, stiff, finish
1-121	51 U.V. 24. 2.51 ADPH	11 NOW	2	=	R		**	2	;	5				
(2-10)	72 U.V. 24.	II YOU	-	8	104		,		1					1111 'Addition
1-61	51 U.V. 24.	72 Com 222	Ħ	*	3	9.0	3.0	9.0	1					
(J-cn	51 U.V. 26, 2,51 ADPH	31 Cam-222	¥	*		=				8				11104, stift
124-2	51 U.V. 24.	52 Butwar 298	Ithese 1	2	2			-						Tellow cast, stiff
131-3	51 U.V. 24, 51 Dian	MEYNE STZA	Ethenol.	2	R				1 1		R 1			Gold, very stiff, yerns glued togsther,
Controlas													-	tinitia flakas off.
Witte Pabrie Sage Green Vi	white Pabric Sage Green Vebbing				3 8									
- YZ A	1) UV 24 - Cyasorb U.V. 24. a converted	and a state		0										

(2,2"-dihydroxy-4-matheny benachhan

ADPP- stetons'2,4-dinitrephanylhy DAB-- p-dimethyl aminoscobentens

padded at arbient conditions

Sign gras webbing; all other samples were white persebute fabrie.
 Unaven treatment; add-on figures would be seeninglass.

### TABLE IV

### EFFECT OF CARBON ARC EXPOSURE ON NOMEX TREATED WITH U.V. ABSORBER, EXTENDER, PVC RESIN, PLASTICIZER, AND HEAT STABILIZER. 50 HOUR EXPOSURE: SAGE GREEN FABRIC AND WEBBING.

3.3					Strength Retention				Lord, 1as			Flong	tion	at Break, 1	
Sample					Nat Loundered		aundered		ered = TB	Abrad.		Liundered		dered + TD	Abrad.
No.		<u>Substrate</u>	Added	Temp.	Dry only	Dry	1004 RH	Dry	100 . RH	Dry	Dry	1007, AH	Dry	1003, 88	Dry
A. 51	PN, 101	Santleizer	148												
24-14		Tabric		150	87	3.4	3.1	2.9	3.0	4===	20	16	14	15	
14-23		Pabric	20.9	75	74	2.9	2.9	2.8			15	14	14	**	
24-78		Webbing		150	92	20.8	-			16.2	40			**	25
26-88		Webblog		150	78	17.6	***	***	***	19.2	28		4.	**	54
24-64		Webbing		150	90	20.4	***			20.0	35		-		34
16-54		Wabbing	****	150	96	21.6	***			18.9	41				30
16-4		Webbing	20.0	75	84	18.9					32	••	••		
20-2		Webbing	19.2	75	86	19.4		***	444	****	34				
1. 22	PNL 107.	Santicizer	161												
16-13		Pabric	23.1	75	100	4.1	3.2	3.0			39	17	19		
16-3		Webbing	19.2	75	85	19.3		***	***		34				
20-1		Webbing	17.4	75	84	18.9	4	444	***		33		**	44	**
c. 11	PPAL 101	<u>Santicizer</u>	148												
18-3		Pabric	13.7	150	. 79	3.1	2.6	3.0			17	14	16	·	
18-4		Webbing	14.3	150	44	20.0					34				
		Santicizer													
p. 51	PPAA, 107	<u>Santicizer</u>	141												
18-1		<b>Fabric</b>	13.2	150	100	3.9	3.1	2.9			33	16	15		
18-3		Vebbirg	15.3	150	80	18.1	•••		***	4450	34			-	**
20-3		Webbing	12.9	130	76	17.2		wat	•••	****	30		••		
<u>1. In</u>	ored Cont	rols													
36-1	Hones	Pabric		**	64	2.5	2.6	3.9	7.7	****	11	12	13	12	
36-2	Scent.	Webbing			66	13.0			***	14.2	16				13
36-3	Byles	Fabric				5.4	5.8	6.2	5.7		29	32	54	31	
58-4	Byles	Vebbing		***	84	29.5				30.3	32	-	-		35
P. Dne	rposed Ca	ntrole													
36-5	Lonez	Fabric				3.9			***	****	39				
36-6	Tonex	Webbing	****		***	22.6				****	39		**	~	**
36-7	Hylen	Fabric			400	6.3					37	_	-	-	+-
36-8	Tyles	Vebbing			***	9.3 35.2	***				-	**	••	**	••
	aktad	ALL DULING				33.6			-		52	••			••

Remaining treatment bath composition was: 107 VNCH-Bakelite Resin 107 Cystorb UV-26 13 dikutyl-tin-di-2-ethyl-hereata methyl athyl ketona q.a. cs 1002

TABLE V

EFFECT OF CARBON ARC EXPOSURE ON NOMEX FABRIC AND WEBBING TREATED WITH U.V. ABSORBER, EXTENDER, PVC RESIN, PLASTICIZER, AND HEAT STABILIZER. 50 AND 100 HOURS EXPOSURE; WHITE AND SAGE GREEN FABRICS; SAGE GREEN WEBBING.

Vermints on

	CYACOTA	Verm	latter.														
Seals No.	UV-26 Absorber Z	AUT	PACH Continue	Plasticizar X/type	i.	<u>Substrate</u>		Dipeeure		Net Laundered		Tern Breaking Land, 1bs.	Abrad.	Por La	Elengari t Laundered L Z 1001 AM D	Ziengation at Break, 2 Merce Laundered + TD DUL MH Dry 1001 BH	Abra.
. Experimental Treatments	Tretwart	-															1
171-16	222	ή τή τη	222	5/148 5/148 5/148	***	hh) 234	5.01	885	2 2 5 6 2 5	11	21			ងដ	= 1		::
11-1/1	91	-	91	3/148	2	- An - 31	1	8 9		:	: ;	ł	16.2	5	1		1
171-12	01	*	10	1/148	2	1	10.5	5		ţ	Ĵ	ł	ļ	2	17 13	:	1
171-7	91		9	E 1949	ł			ł		•	1	ļ	:	2	:	:	;
	22	) #1 IP	122	5/141	<b>1</b>	2 2 j	44	88	2.2	31	8 N N			3:	11	:	;
	10	-m	9	5/141	13	2 32		85	0.2	ł	ł		17.0	22	= ;		18
1-701	2	vi	2	5/141	46			ş :		;	ŧ	I	12	20	: ;		12
171-0					:		1.41	8	1.0	3.5	2.9 7	3.0	i	61	15 17		1
			22	191/6	**	14 22 23 41	16.0	22	0.0	1.5	2.5	Ľ	ł	-	CI 51	:	:
171-9	10		:		;			ł		ł	•	19-7	i	1	1	2	ł
	2	'n	22	1/141	11			22	5	31	21		ł	2	16 13		• 1
1-5/1	10	-	10	1111		:				1	ł			N	:	22	:
	2	103	2	171/6	:3	1. J.	7 11	25			-	:	i	16	10 11	1	1
	99	••••	8	141/6	-	3	10.5	ខ្មន	2.0	: :				2		1	: 1
	1	n	2	191/6	2	2	10.3	8	1.91						:	:	2
2-341	-		••••	191/2	2	4.2	10.0	8	2.5	2.5	1.5					ł	5
			4 4		2	2	10.0	8	5.5	•					Z] Z]	:	1
			<b>n</b> #			20 1	4-9	8	1	1	1			= =		1	:
	•	.	•	144/2	\$	-	4-4	20	17.2	1		I	1.2	18	::	::	ដូ
	• •	••	22	141/6	22	N 3 3 H		25	5.6		2.2			1	11	1	: :
1-1-11			•		;			\$		1	ł		1	2	·	#	1
			h	167/0	\$	A.,	•••	2	0.0	3.1	0.0		-		16 15	91	i

- CONTINUED -

TABLE V (Concluded)

-

clon at Ereak, t Liumder d + 12 Abrad. Dry IGOT AH Dry			: :		    B	
<u>Rive Laundered Liunderid a 1</u> Dry 1002 NH Dry 1007 F		1ev41 1v411	12 12		:#9	:   : :   :
		52544	92	*2193	352	388
Arrad.						
Yarn Breaking Lasd, Iba. mdared 1 Lundered + TD Abrad. 002 M Dry 1002 AH Dry			2.6		3	
2			13		135	
Yafn Bi Mot Laundered Dry 1002 NH		12221	2.6		133	<b>[]</b> ]
		*****	12	22222	333	35.2
Erposure hra.		0 2 2 2 2 0 0 5 2 2 2 0 0	e ç	82202 e	• 58	• 98
<u>Add-on</u> 1	None					
<u>Substrate</u>		22323	**		8. 6. 6. 19 93 43	<b>* * *</b>
100	71ber	Nonex Konex Konex Konex	Nomex Nomex	Котех Котех Иотех Котех Котех	Nylon Nylon Nylen	Nylon Nylon Nylon
<u>Planticizer</u> 2/type						
Formulation PH VNCK Inder Costing						
Farmulation Cystorb ADPH WACK Absorber Extender Costing 2						
Cyasorb UV-24 Absorber Z						
Samle No.	. Controle					

A thread by possible 2 T - angle green fabries 2 T - angle green vabiling.
 A breaded by possible five times through bothle of percentua harmons.

TABLE VI

4

1

.

NATA N

EFFECTS OF CARBON ARC EXPOSURE ON NOMEX TREATED WITH ELEMENTAL IODINE. 50 HOURS EXPOSURE; WHITE FABRIC AND DAGE GREEN WEBBING.

	<u>Titme/Temp.</u> ain. 'C	<u>Ites/Teep. U.V. Absorber Extenter</u> sig. <sup>2</sup> C	Extenter	Rents	uo-ppy	Substrate		Not Lat	Laundered	Tarn Breaking Load	의 이희	5.1	Abr. Not 1 Dry Dry	100		Elengation at Break, ered Laundered + TD 3. RH Dry 1002, RH	- 50	AMA N
A. Sample No.																1		
1-141	:	Iod. Vapor	ł	1	•••	•	12	3.2	3.2	2.6 0.2		. 1.6	00 9.5	:		;		
141-2	ł	Iod. Yapor	1	ł	10.3		19	12.9									-	2
1-641	ł	Tod. Vapor + 5% UV-24	2.5% ADFT	St hos	1		2					1 0	8 8			1 2	1 2	1 2
143-2	ł	Iod. Vapor + 57 UV-24	2.51 ADFN	SK VHCE	3		*	1.4	1	1	1	1		1	1		1	1
2-141	1	Iod. Vapor + 52 DHAB	ł	SIL THCE	;		#			3.1 3.2	1.6 5.1				2	#	.2	2
141-4	ł	Iod. Yapor + 52 DHAB	ł	ECHA IS	20.0	•	5	13.0	1	1	1	1	5	,		1	1	:
2-111	60/100	Todine in Diff	ł	1	1	•		2	5.5	2.9 3.2	1	1	2	\$	*	*	-1	1
1-111	40/24	Todine is	ł	1	1		\$	5	1	-	1	!	9	,	=	1	:	1
111-10	15/100	Iodine fa Xylene	1	;	1		\$		1	1	1	1	•	•		,	;	1
9-111	15/100	Todim In CCI4	ŧ	1	1		\$		1	-	1	!	•	1		1		:
J. Patrested Centrele	entrele																	
1						:		12		1	•1	1	•	•	1	1	•	1
						:	-	2.2	1	1	1	1	2		1	1	:	1
and los							8			1	1	1					1	

APPENDIX I

記録の

LIST OF PRODUCTS USED

Trade Name or Abbreviation

Chemical Name

### A. U.V. Absorbers

Cyasorb UV-24	2,2'-dihydroxy-4-methoxybenzophenone	Am
Tinuvin-320	2-hydroxy-3,5-ditertiarybutylphenyl benzotriazole	e
Tinuvin-P	2(2'-hydroxy-5'-methylphenylbenzotriazole)	Ge
Uvinul DS-49	2,2'-dihydroxy-4,4'-dimethoxy-5-sulfobenzophenone	GAI
B. Quencher		
AM-105	a nickel complex of a thiobisphenol compound	Fei
C. Extenders		
ADPH	acetone-2,4-dinitrophenylhydrazone	FL
		Fis
DMAB	p-dimethylaminoazobenzene	FIE
PNA	p-nitroaniline	Fis
PPAA	p-phenylazoaniline	FIE
D. Solvents, Car	D. Solvents, Carriers and Adduct Materials	
TLF-2791	benzophenotetracarboxylic acid (carrier)	Ind
M-Pycol	H-methyl-2-pyrrolidone	GAI
Dynex N	acetophenone derivative (carrier)	Arl

Source

American Cyanamid Co. Geigy Chemical Co. Geigy Chemical Co. GAF Corporation

Ferro Corporation

Fine Organics, Inc.

isher Scientific Chem. Co. Isher Scientific Chem. Co. Isher Scientific Chem. Co. Isher Scientific Chem. Co.

DuPont GAF Corporation Arkansas Co. Dow Chemical Co. Dow Chemical Co.

> polymer of polyethyleneimine (mol wt 1800) polymer of polyethyleneimine (mol wt 60,000)

PE1-18 PE1-600

	APPENDIX I (Concluded)	
Trade Name or Abbreviation	Chemical Name	Source
E. Plasticizers		
Santicizer 140	cresyl diphenyl phosphate	Monsanto Co.
Santicizer 141	2-ethylhexyldiphenyl phosphate	Monsanto Co.
. Santicizer 148	isodecyldiphenyl phosphate	Monsanto Co.
Santicizer 160	butylbenzyl phthalate	Monsanto Co.
F. Heat Stabilizers		
.K&K #2101	dibutyltin-di-2-ethylhexoate	K and K Labs. Inc.
G. Resins		
HHXA	copolymer of vinyl chloride/vinyl acetate (87:13)	Bakelite Co.
VMCH	terpolymer of vinyl chloride, vinyl acetate and maleic acid, (86:13:1)	Bakelite Co.
Geon 222	vinyl chloride/vinylidene chloride copolymer (low mol. wt.)	B.F. Goodrich Chem. Co.
Butvar B-72A	polyvinyl butyral resin (mol. wt. 180 - 270,000)	Monsanto Co.
H. Pigments		
Black Shield 1611 (carbon black dispersion)	15% carbon black, 22.5% VYHH in methyl ethyl ketone	CDI Dispersions

### APPENDIX II

### REFERENCES

### A. Literature Searches

The literature search consisted of a survey of Chemical Abstracts by our own staff (1948 to present) and searches ordered from the Defense Documentation Center (1965-1972), the Textile Information Center, Institute of Textile Technology, Charlottesville, Virginia (1965-1970), NASA and Science Information Exchange. The emphasis was primarily on information on the degradation of Nomex due to uv radiation which would help in the selection of the most promising treatments. However, references to the actinic degradation of all polyamides are also listed.

One of the most important references for our immediate purpose is the description of the work by Stephenson and Wilcox (75) who determined the tensile properties of several polymeric materials, including Nomex, after exposure at five wavelengths of light under nitrogen. Nomex, (at the time of this work was still called HT-1) lost strength and elongation at break most rapidly at 340 and 369 nm, less so at 244 nm and least at 214 and 437 nm.

Johnson <u>et</u>. <u>al</u>. (38) compared the photodegradative wavelength dependence of Nomex with that of experimental, thermally resistant fibers; they indicate that degradation of the strength and color of Nomex occurs within the region from 310 to 390 nm with the peak between 360 and 370 nm. Nomex thus absorbs at higher wavelength than most other polymers.

Data on the deterioration of Nomex materials in carbon arc and outdoor exposure have been published by Ross and Opt (62), Little and McGrath (47), and Hargraeves (29-32); we have also received such data from DuPont (70). DuPont also indicated some success with a commercial uv absorber, Tinuvin P (Geigy Chemical Company).

In Summary, the literature search helped In establishing the wavelength range in which Nomex shows greatest sensitivity (this was confirmed by obtaining the uv spectra of Nomex in our laboratory using a KBr disc of

ground fiber). Furthermore, it indicated that there seems to be no published work on the application of uv absorbers to Nomex. We, thus had to rely primarily on the experimental approach of chosing absorbers on the bases of the ranges in which they absorb and their expected substantivity to aromatic polyamide.

- Achhammer, Bernard G., et al., "Mechanism of the Degradation of Polyamides," (Natl. Bureau of Standards, Washington, D. C.). J. <u>Research</u> <u>Natl. Bur. Standards, 46</u>, 391-421 (1951) (Research Paper No. 2210); <u>J. Applied Chem</u>. (London) <u>1</u>, 301-20.
- Anon., "Effects of U.V. Absorbers on the Sunlight Resistance of Various Types of Nylon," <u>Am. Dyestuff Reptr.</u>, <u>55</u>, 1065-1071, 1966.
- Anon., "Weathering Degradation of Nylon," <u>Textile Chem. and Color.</u>,"
   p. 57-61, February, 1972.
- Asmussen, R. W. and Buchmann-Olsen, B., "Investigations on Tensile Strength Losses Produced in Nylon Yarns by Irradiation," <u>Trans. Danish</u> <u>Acad. Tech. Sci.</u>, 4:1948, Contribs. Danish Inst. Textile Research, No. 5 (12 pp).
- 5. Bhavsar, M. D., "Yellowing of Fibers and Inhibitors of Yellowing," (Silk and Art Silk Mills' Res. Assoc., Bombay, India.) <u>Silk Rayon</u> <u>Ind. India 1968, 11(1), 67-77 (Eng.).</u>
- Bogdanov, M. N., et al., "Crosslinking of Polyamides Containing Aromatic Amino Groups," <u>Vysokomol. Soedin.</u>, Ser. A 1968, <u>10</u>(6), 1251-5 (Russ.).
- Bruck, Stephen D., "Ultraviolet Stability of Cross-Linked Polycaprolactam Systems," (John Hopkins Univ., Silver Springs, Md.). J. <u>Res.</u> <u>Natl. Bur. Std.</u> 66A, 489-95 (1962).
- Chu, C. C. and R. A. Kenney, "The Measurement of Ultraviolet Reflectance as a Criterion of Nylon Fabric Deterioration," November, 1961 AD-273-105.

and and an end of the construction of the the the the the the the second s

- Gostain, W., et al., "Process for the Stabilization of Delustered Nylon," (to Imp. Chem. Inds. Ltd. and Brit. Nylon Spinners, Ltd.) BP997316, July 7, 1965, Appl. September 3, 1962.
- Dierkes, Gerhard, "Chemistry, Production, and Properties of the Nomex Polyamide Fiber," (Ingenieusch. Textilwesen Krefeld, Krefeld, Ger.).
   <u>Spinner Weber Textilwered1</u>. 1969, 87(1), 43-4 (Ger.).
- Dolezel, Bretislav, "Technique of Accelerated Aging Tests of Plastics. Determining the Importance of the Effect of Some Simulated Atmopheric Factors on the Aging of Polyamide 6," (State Res. Inst. Protect. Matters., Prague). <u>Chem. Prumysl</u>, <u>13</u>,(7), 388-91 (1963).
- Dorset, B. C. M., "Recent Developments in Processes to Improve Manmade Fibers," <u>Textile Mfr.</u>, <u>93</u>, 70-75 (1967).
- Doshi, S. M., "Photodegradation of Polyamides and Its Possible Prevention," (Bombay Text. Res. Assoc., Bombay, India). <u>Colourage</u> 1968, <u>15</u>(1), 75-86, 110 (Eng.).
- Dunlap, E. S., et al., "Degradation of Polyamide Fibers Exposed to Various Sources of Radiation," Textile Chem. & Color. 1, 99-104, 1969.
- 15. Egerton, G. S., "The Photochemical Degradation of Textile Fibers with Near Ultraviolet and Visible Light," <u>De Tex</u> 11, No. 1, 28, 30-6 (1952).
- Egerton, G. S. and Shah, K. M., "Effect of Temperature on the Photosensitized Degradation of Textile Fibers," (Univ. of Manchester, Engl.). <u>Nature, 202</u> (4927), 81-2 (1964).
- Evans, B. A., Shumm, R. W., "Dyeing Nomex Type 450 Nylon," (Org. Chem. Dept., du Pont, Wilmington, Del.). <u>Text. Chem. Color.</u>, 1970, <u>2</u>(15), 262-65 (Eng.).
- Fedotora, O., et al., "New Fiber-forming Polyamides," AD-684-487, 1968.
- Fester, W. and Kehren, M. L., "Effect of Disperse Dyes on the Lightfastness of Nylon 6 Fibers," (Textilforschungsanstalt Krefeld, Ger.).
   <u>Z. Ges. Textil.-Ind.</u>, 67(2), 115-18 (1965) (Ger.).

: 1

22

- Fester, W. and Kehren, M. L., "Photochemical Degradation of Polyamide Fibers," <u>Melliand Textilchem</u>. No. 2, 56-60 (1965) (Ger.).
- Figucia, F. and R. D. Wells, "Strength Losses in Nylon Parachute Materials with Time, Exposure, and Use," U. S. Army Natick Labs., AD-668-901, March, 1968.
- Flynn, Joseph H., "An Equation for Calculating the Number of Chain Scissions in the Photochemical Degradation of Solid Polymers," (Natl. Bur. of Standards, Washington, D. C.). J. Polymer Sci. 27, 83-6 (1958).
- Fomenko, A. S., et al., "Radiation Stability of Isomeric Aromatic Polyamides," (Inst. Fiz. Khim., Kiev, Russia). <u>Khim Prom. Ukr</u>. (Russ. Ed.) 1968, (1) 31-4 (Russ.).
- Gerashimova, L. S., et al., "The Effect of Crosslinked Chemical Bonds on Stress Relaxation in Polyamide Fibers," <u>Mekh. Polim.</u> 1968 (1), 53-8 (Russ.).
- 25. Gnip, N. P., "Light Stabilization of Polyamides by Graft Copolymerization," <u>Khimicheskaya</u>, <u>No. 4</u> (1964).
- Gnip, N. P., et al., "Light Stabilization of Polyamides by Graft Copolymerization," <u>Khim. Prom., Inform. Nauk.-Tekhn. Zb.</u>, 1964 (4), 9-10 (Ukrain).
- 27. Goldfore, I. J., "Research on Polymer Degradation Mechanisms," S.I.E. No: ZQF-240740, July 1970.
- 28. Hall, J., et al., "Surface Treatment of Polymers with Activated Gas Plasma for Adhesive Bonding," January 1969, AD-680-957.
- 29. Hargreaves, G., "Molecular Changes in Polyamides and Elastomeric Polymers due to Nuclear, Ultraviolet and Thermal Radiation," December 1961, AD-272-964L, NAEC-AML, Philadelphia, Pa.
- 30. Hargreaves, G., "Molecular Changes in Polyamide and Elastomeric Fortyn. .s due to Nuclear, Ultraviolet and Thermal Radiation," December 1962, AD-404-764.

31. Hargreaves, G., "Molecular Changes in Polyamide and Elastomeric Polymers due to Nuclear, Ultraviolet and Thermal Radiation," June 1963, AD-406-793.

adamente en principarses in perio de

- 32. Hargreaves, G., "Molecular Changes in Polyamide and Elastomeric Polymers due to Nuclear, Ultraviolet and Thermal Radiation," December 1963, AD-434-216.
- 33. Hays, Margaret B., et al., "Effects of UV Absorbers on the Sunlight Resistance of Various Types of Nylon," <u>Amer. Dyest. Rep. 55</u>(25), P1065-P1071(1966) (Eng.).
- 34. Hemingway, E., "Ultraviolet Absorbers for Stabilizing Organic Polymers," Dyestuffs Div., ICI Ltd., Eng., <u>Rep. Proc. Appl. Chem</u>. 1969, 54, 1959-63 (Eng.).
- 35. Hearst, P. J., "Degradation of Organic Coatings by Irradiation with Light, IM. Volatile Products from Simulated Solar Irradiation in Air," June 1963, AD-617-244.
- 36. Holsten, J. R., et al., "Fiber of an All-Para-Oriented Ordered Aromatic Copolyamide," (Chemstrand Res. Center, Inc., Durham, N. C.). <u>Appl.</u> <u>Polym. Symp.</u> 1969, No. 9, 63-74 (Eng.).
- 37. Hujii, Shigeru, "Degradation of Polyamide Resins and Its Prevention," <u>Sekiyu to Sekiyu Kagaku</u> 1968, <u>12(4)</u>, 42-7 (Japan).
- Johnson, L. D., et al., "Photodegradation Wavelength Dependence of Thermally Resistant Organic Polymers," (Chemstrand Res. Center, Inc., Durham, N. C.). J. <u>Appl. Polym. Sci.</u> 1969, 13(9), 1825-32 (Eng.).
- Kamogawa, Hiroyoshi, "Responsive Polymers, V. Preparation of Some Polymeric Phenolic Ultraviolet Absorbers," (Govt. Text. Ind. Res. Inst., Yokohama, Japan). <u>Kogyo Gijutsuin Sen'i Kogyo Shikensho Kenkyu Hokoku</u> 1969, No. 86, 95-6 (Japan).
- Kinlock, J. C., "Correlation of Dyestuff Color Change with Actinic Degradation of Nylon," April 1968, AD-834-011L, A34-531-036/2071/F102-10-01, NASC-Air 604, Washington, D. C. 20360.

- Korshak, K. K., et al., "The Effect of Low Molecular Weight Compounds on the Photochemical Destruction of the Polyamide Acid G-699," <u>Vysokomolekulyarnye Soedineniya</u>, 1, 990-7 (1959).
- Korshak, K. K., et al., "Control of Some Properties of Polyarylates and Aromatic Polyamides," (Inst. Elementoorg. Soedin., Moscow, USSR). <u>Vysokomol. Soedin., Ser A</u> 1968, 10(9), 2058-69 (Russ.).
- Kratzsch, Eberhard and Hans Hendrix, "Differentiation Between Light and Acid Destruction of Polyamide Fibers," (Vereinigte Glanzstoff Fabriken A.-G., Wuppertal-Eberfeld, Ger.). <u>Melliand Textilber</u>. <u>45</u>(10), 1129-33 (1964).
- 44. Krema, L., "Combined Stiffening, Water-repellent and Flameproof Finishing of Polyamide Industrial Fabrics," <u>Textil 21</u>, No. 6 230-233, 1966.
- 45. Lauttman, R. G., "Evaluation of Thermal Kinetic Parameters of Nylon 610 Without Degradation," August 1963, AD-417406.
- 46. Leclercq, B., "Modification of Polyamides by Chemical and (or) Physical Actions. III. Thermal Degradation Mechanisms of Secondary Diamides as Models for Nylon 66," (Inst. Text. France, Fr.). <u>Bull. Inst. Text.</u> <u>Fr. 1968, 22(139), 915-22 (Fr.).</u>
- 47. Little, C. O. and McGrath, J. C., Symposium on Fibrous Materials, Ross, J. H., Ed., Wright-Patterson Air Force Base, Ohio, 1963.
- 48. Long, W. C., "'Nomex' Nylon A New High Temperature Medium for Gas Filtration," Filtration Separation 2(3), 198-202 (1965).
- Moore, R. F., "Photochemical Degradation of Polyamides and Related Model N-Alkylamides," (Imp. Chem. Inds. Ltd., Manchester, England). <u>Polymer 4</u>(4), 493-513 (1963).
- 50. Opt, P. C. and J. H. Ross, "Effects of Combined Gamma Radiation and Thermal Exposure on Some Mechanical Properties of Three Polyaromatic Yarns," <u>Textile Res. J.</u>, <u>37</u>, 1050-1055, December 1967.
- 51. P. autuas, A. P. and A. A. Visokinskas, "Changes in Some Properties of Polyamide Fibers and Fabrics Due to Prolonged Exposure to Sunlight," <u>Tekstil. Prom. 25</u>, 66-70, 1965.

- 52. Prati, Giovanni, "Physicochemical Tests and Viscometry of Heat Resistant Polyamides," (Centro Stud. Ind. Tess., C. N. R., Milan, Italy). <u>Ric.</u> <u>Doc. Tess</u>. 1968, 5(4), 185-9 (Ital.).
- Pavlov, N. N., et al., "Light-Stabilizing Effect of Some Compounds, Containing Sulfur and Nitrogen During Ultraviolet Irradiation," <u>Plast</u>. <u>Massy</u>, 1967(1), 56-8 (Russ.).
- 54. Rabek, Jan F., "Effect of Ultraviolet Light on Polymers," (Katedry Technol. Tworzyw Sztucznych, Polytec., Wroclaw, Poland). <u>Wiado-mosci</u> <u>Chem.</u> 20(7), 435-47 (1966) (Pol.).
- 55. Rafikov, S. R. and Hsu, Chih-P'ing., "Chemical Changes in Polymers. VI. The Effect of Ultraviolet Irradiation on Polyamides in the Presence of Oxygen and Water Vapor," <u>Vysokomolekul</u>. <u>Soedin</u>. <u>4</u>, 851-9 (1962).
- 56. Rader, C. and A. Schwartz, "Accelerated Deterioration of Textiles," June 1963, AD-406-304.
- 57. Rader, C. and A. Schwartz, "Accelerated Deterioration of Textiles," October 1963, AD-419-966.
- 58. Rizzo, F. J. and B. Natseos, "Absorbers of Near Infrared Radiation," June 1964, AD-672-512.
- 59. Rode, V. V., et al., "Thermal Degradation of Aromatic Polyamides at High Temperatures," (Inst. Elementoorg., Soedin., Moscow, USSR). <u>Vysokomol. Soedin., Ser A</u> 1968, <u>10</u>(11), 2550-4 (Russ.).
- Rode, V. V., et al., "Structural Transformations in Some Aromatic Polyamides at Elevated Temperatures," (Inst. Elementoorg. Soedin. Moscow, USSR). <u>Vysokomol. Soedin.</u>, <u>Ser A</u> 1969, <u>11</u>(1), 57-68 (Russ.).
- Ross, Jack H., "Translation of Fibers and Yarns into Flexible Structures," (Dept. of the Air Force, Wright-Patterson AFB, Ohio). <u>Text</u>. <u>Res. J. 36(7), 662-72 (1966) (Eng.).</u>

- Ross, Jack H. and Opt, Preston C., "Behavior of Polyaromatic Fibers and Yarns in Unique Environments," (Fibrous Mater. Br., Wrightratterson AFB, Ohio). <u>Appl. Polym. Symp</u>. 1969, No. 9, 23-47 (Eng.).
- Roszkowski, W., "Effects of Oils and Solvents on Mechanical Properties of Propylene and Aromatic Polyamide Fabrics," Zakl. Ochr. Osobistych, Cent. Inst. Ochr. Pracy, Poland, <u>Pr. Cent Inst. Ochr. Pr</u>. 1970, 20(64) 33-42, (Pol.).
- 64. Roth, W. and R. Schroth, "A Study of Different Degradation Processes in Polyamide Filaments," <u>Deut</u>. <u>Textiltech</u>., <u>15</u>, 661-663, 1965.
- 65. Schwemmer, Martin, "Photochemical Degradation of Polyamide Fibers," <u>Textil-Rundschau</u>, <u>11</u>, 1 (1956).
- 66. Samoilov, G. G., et al., "Photomechanical Destruction and Life of Irradiated Stressed Polymers," (Fiz.-Tekh. Inst. im. Ioffe, Leningrad, USSR). <u>Fiz. Tverd. Tela</u> 1968, <u>10</u>(10), 3039-45 (Russ.).
- Sbrolli, Walter and Capaccioli, Tarcisio, "Photodegradation of Polycaprolactam Fibers," (Centro Studi Chimico Soc. Brombrini Parodi-Delfino, Rome). <u>Ann. Chim</u>. (Rome) <u>53</u>(8-9), 1184-98 (1963).
- Scala, L. C., et al., "Effect of Long-Term Oxidation at 200-300° on Six Types of Aromatic Amide and Imide Resins," (Res. and Develop. Center, Westinghouse Elec. Corp., Pittsburgh, Pa.). J. <u>Appl. Polym.</u> <u>Sci</u>. 1968, <u>12</u>(10), 2339-57 (Eng.).
- Schlimper, R., "Stability of Macromolecules. XII. Stabilization of Poly(vinyl chloride) and Polyamide Against Degradation Due to Weathering," (Inst. Chem. Technol. Plaste, Leipzig, Ger.). <u>Plaste Kautschuk,</u> <u>11</u>(8), 466-71 (1964) (Ger.).
- 70. Schuum, A. W., duPont Co., personal communication, 1970.
- 71. Schwenmer, Martin, "Photochemical Decomposition of Polyamide Fibers," Eidgenoss, Materialpruf.-u. Versuchsanstalt Ind., Bauw. u. Gewerbe.
- 2 Zurich, Ber. No. 180, 39 pp. (1955).

- 72. Shah, C. J., et al., "Photodegradation of Indigenous and Imported Undyed and Dyed Polyamide Yarns," <u>Silk Rayon Ind.</u>, (India) <u>9</u>(6), 191-200 (1966) (Eng.).
- 73. Shah, S. F. and Subramanian, R. V. R., "Photodegradation of Polyamides," <u>Silk Rayon Ind.</u>, (India) <u>10(2)</u>, 128-32 (1967) (Eng.).
- 74. Sippel, A., "Degradation of Textile Fibers by High Energy Radiations and by Heat," Melliand Textilber, 38, 898-905 (1957).
- 75. Stephenson, C. V. and Wilcox, W. S., "Ultraviolet Irradiation of Plastics. IV. Further Studies of Environmental Effects of Films and Fibers," (Southern Res. Inst., Birmingham, Ala.). J. Polymer. Sci. Pt. A, 1, 2741-52 (1963).
- 76. Vachon, Raymond N., et al., "Oxidative Degradation of Nylon 66 Filaments," (Princeton Univ., Princeton, N. J.). <u>Text. Res. J.</u> 1968, <u>38</u>(7), 716-28 (Eng.).
- 77. Vershinina, M. P., et al., "Effect of Ultraviolet Irradiation on the Strength of Polymers," <u>Vysokomolekul. Soedin.</u> 6(8), 1450-7 (1964).
- 78. Weiss, J. O., et al., "A New High-Temperature Fiber, Its Properties and Performance," (Chemstrand Res. Center, Inc., Durham, N. C.). J. Polymer Sci., Part C 1967 (19), 41-8 (Eng.).
- 79. Willis, C. A., "Treatment of Nylon Webbing to Increase Resistance to Abrasion," CADO (Central Air Documents Office, Navy-Air Force), <u>Tech</u>. <u>Data Digest</u>, <u>14</u>: No. 15 (15-25).

B. Patent Search

The search of U. S. Patent Office files from 1950 to 1970 by IF1/Plenum Data Corporation produced no references to agents that may be candidate UV inhibitors for aromatic polyamides (Nomex).

One of us (GA), on a personal visit to the U. S. Patent Office, updated the above patent search from 1970 to March 1972. This again produced no references to our subject. Consultation with two U. S. Patent examiners assured us that the correct classes and subclasses were being searched.

- a. Specific Patent Search on Related Topics.
   No patents available.
- b. General Subject Patent Search:
- 80. "Nylon Fibers with Improved Light and Heat Stability," E. I. du Pont de Nemours & Co., Inc., B.P. 1,073,505, June 28, 1967, Appl. USA April 29, 1964.
- 81. "Finish for Improved Nylon Filaments," W. Roth and R. Schroth, assignor to VEB Chemiefaserwerk Schwarza Wilhelm Pieck, B.P. No. 1,001,318, August 11, 1965.
- 82. "Stabilization of Nylon Against Photochemical Degradation with Manganous Hypophosphite or Acetate and Cerous Oxalate," T. F. Carbin and L. M. Ownbey, assignor to Am. Enka. Corp., U. S. Patent No. 3,206,430, September 14, 1965.
- "Light Stabilization of Nylon Fabrics," P. B. Roth, assignor to Am. Cyanamid Co., U. S. Patent No. 3,207,620, September 21, 1965.
- 84. "Improved Light Stability of Polyamides," W. Henssler, Ger. (East) 46,543, March 20, 1966.
- 85. "Stabilization of Nylon Textiles," Soc. de la Viscose Suisse, Belgian 674,888, May 3, 1966.
- 86. "Stabilization Agent for Nylons," Snia Viscosa SpA. Belgian 678,915, October 3, 1966.
- 87. "Stabilized Nylon 6 Filaments," Soc de le Viscose Suisse, BP 1,060,710, March 8, 1967.
- 88. "Process for Applying UV Absorbers to Nylon Fibers and Fabrics,"
  B. P. Roth, L. B. Hallows, and W. N. Nakajima, assignor to Am. Cyanamid, Co., U. S. Patent No. 3,320,207, May 16, 1967.
- "Dyed Nylon Yarns Resistant to Light and Gas Fading," J. A. Brooks, assignor to E. I. du Pont de Nemours & Co., Inc., U. S. Patent No. 3,363,969, January 16, 1968.

- 90. "Nylon Fibers with Improved Stability, Whiteness, and Dyeability," A. Anton, assignor to E. I. du Pont de Nemours & Co., Inc., U. S. Patent No. 3,377,314, April 9, 1968.
- 91. "Light Stabilized Polyamide and Process Therefor," L. J. Exner, assignor to Rohm and Haas Company, U. S. Patent No. 3,407,170, October 22, 1968.
- 92. "Light Stabilized Delustered Nylon Fibers," L. J. Exner, assignor to Rohm and Haas Co., U. S. Patent No. 3,407,170, October 22, 1968.
- 93. "Light-stabilized Nylon," P. V. Papero, assignor to Allied Chemical Co., Germ. Offen. 1,495,018, June 4, 1969, U. S. Appl. October 20, 1961.