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- 90 Equation 4-35 should read:

$$St_{gv} = St_0 \frac{t_{aw} - t_w}{t_r - t_w}$$

91 Equation 4-36 should read:

$$St_{gv} \doteq \frac{Y_w}{(\rho u)_e c_{p_e}} (t_r t_w)$$

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PURDUE RESEARCH FOUNDATION

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LIQUID-FILM COOLING, ITS PHYSICAL NATURE AND THEORETICAL ANALYSIS

by

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ABSTRACT

A discussion of the physical nature of liquid-film cooling is presented that gives special emphasis to those phenomena occurring at the gas-liquid interface. The interfacial mass and heat balances are discussed in detail, together with the phenomena of film instability and the entrainment of liquid by the gas stream.

A review is presented of the pertinent heat-mass transfer analyses for the wall region wetted by the liquid film. The limitations of those analyses are noted, and an alternate correlation procedure is suggested that is applicable to the case where the gas flow is compressible and subject to the influence of a streamwise pressure gradient. In order to implement that correlation procedure, a method is developed whereby the temperature at the gas-liquid interface can be approximated analytically. Calculated results are presented that are noted to agree favorably with the limited experimental data.

A heat transfer analysis is presented for the wall region downstream of the liquid film. That analysis is based on a relatively simple flow model, and is applicable to the case where the gas flow is compressible and subject to the influence of a streamwise pressure gradient. Two wall conditions are considered: (a) the case where the wall is adiabatic; and (b) the case where the wall is cooled externally.

Conclusions and recommendations for further experimental research are presented.

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1. INTRODUCTION

1.1 General Discussion

Liquid-film cooling refers to the injection of a thin continuous liquid film onto a given surface for the purpose of protecting that wetted surface from thermal damage from a hot gas flowing past the film.

The motivation for developing liquid-film-cooling technology for rocket motors has been the development of nuclear rocket engines, the use of higher energy chemical propellants, and the trend toward higher combustion chamber pressures. These newer developments are characterized by a large increase in the heat flux from the working fluid to the walls of the combustion chamber and the exhaust nozzle, such that the regenerative cooling of those walls becomes marginal (1)(2)(3).¹

A need, therefore, exists for developing a method of cooling which will supplement regenerative cooling. While porous-wall cooling techniques (transpiration cooling which utilizes gaseous coolants and <u>sweat</u> cooling which utilizes liquid coolants) are more effective than liquidfilm cooling from the standpoint of the degree of thermal protection realized per pound of coolant consumed, liquid-film cooling still appears to be quite attractive for the following reasons: (a) clogging of the porous wall is still a problem and leads to localized "burnout," (b) a reduction in the structural integrity of large components results ¹Numbers in parentheses refer to references listed in the Bibliography.

when they are constructed from porous materials, and (c) the fabrication problem is simpler in the case of a liquid-film-cooled system.

Considerable attention has also been devoted to studies concerned with employing a gaseous film coolant Gaseous-film cooling has the same advantage of simplicity as does liquid-film-cooling, but the absence of a phase change during the cooling process severely limits the cooling capacity of gaseous-film cooling in comparison to that of liquid-film cooling

1.2 Objectives of the Present Investigation

The objectives of the subject investigation were threefold: (a) to describe the physical nature of liquid-film conling; (b) to review the theoretical heat-mass transfer correlations for the liquid-film-cooled region, and to develop an alternate correlation for that region; and

(c) to develop a heat transfer correlation for the region downstream of the terminus of the liquid film.

The correlation procedures developed herein are generalized so as to be applicable to the flow of a compressible gas in the presence of a streamwise pressure gradient. Moreover, the developments consider boundary layer flow rather than fully developed pipe flow, as have previous analyses. The developments are limited, however, to the case where the gas flow is fully turbulent and the liquid-film coolant is non-reactive

Assessment of the validity of the correlation procedures developed herein awaits the existence of accurate experimental data. Such experimental work is being conducted currently at the Jet Propulsion Center, Purdue University, Lafayette, Indiana.

2. THE PHYSICAL NATURE OF LIQUID-FILM COOLING

2.1 Description of Liquid-Film Cooling

Figure 2-1 presents a schematic diagram illustrating the pertinent aspects of the liquid-film cooling process. There are two regions of the wall that are of principal interest: (a) the liquidfilm-cooled region, and (b) the gas-vapor-cooled region. The phenomena which characterize each of those regions are discussed in the following sections

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2.1 1 The Liquid-Film-Cooled Region

Figure 2-2 illustrates a physical boundary layer model for the liquid-film-cooled region The phenomena which relate to the development of that boundary layer region are discussed in the following paragraphs

<u>The Nature of Film Development</u> At station $x = x_0$, in Fig 2-1, the liquid coolant is injected onto the solid surface and establishes a spannise continuous liquid film. The liquid film interacts with and is accelerated by the hot gas stream flowing past it. The velocity distribution throughout the liquid film at any streamwise location is determined by the solution of the momentum equation for the twodimensional flow of an incompressible fluid; the latter can be written in the following form (4). Thus





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FIS. 2-2. A BOUNDARY LAYER MODEL FOR THE LIQUID-FILM-COOLED REGION

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$$\rho_{1}\left(u \frac{\partial u}{\partial x} + v \frac{\partial u}{\partial y}\right) = -\frac{\partial p}{\partial x} + \frac{\partial \tau}{\partial y} \qquad (2-1)^{1}$$

6

The acceleration terms in Eq. 2-1 are of considerable importance at and near the plane where the liquid is injected. However, for the largest portion of the liquid flow regime, because of the extreme thinness of the liquid film,² the acceleration terms and the pressure gradient term in Eq. 2-1 can be considered of negligible importance in comparison to the viscous term. Hence, if τ_s and τ_w denote the shear stress at the gas-liquid interface and at the liquid-solid interface, respectively, then one may write

$$\tau \equiv \rho_{1} \left(v + v_{1} \right)_{1} \frac{\partial u}{\partial y} = \text{constant} = \tau_{s} = \tau_{w}$$
 (2-2)

where v is the molecular contribution and v_t is the turbulent or eddy contribution to the "effective" kinematic viscosity. From Eq. 2-2 it follows that the velocity distribution for the liquid film is given by

$$u = \frac{\tau_{s}}{\rho_{1}} \int_{0}^{t} \frac{dy}{(v+v_{t})_{1}}$$
(2-3)

For the special case where the film is entirely laminar ($v_t = 0$), Eq. 2-3 reduces to

$$u = \frac{r_s}{(\rho v)} y$$
 (2-4)

 1 Refer to Appendix A for a definition of the notation.

² Typical film thicknesses are of the order of 1 to 5 X 10^{-3} in.

The problem of expressing the parameter $(v + v_t)_1$ as a function of the normal coordinate y, so that the integral term in Eq. 2-3 can be evaluated for the case of a turbulent film, can be circumvented by assuming that the universal "law of the wall" is valid for the lequidflow regime. Figure 2-3 presents the "law of the wall" in graphical form; the dimensionless parameters u⁺ and y⁺ presented in the figure are defined as follows:

$$u^{+} = \frac{u}{(-\tau_{s}/\rho)^{\frac{1}{2}}}$$
 (2-5)

$$y^{+} = \frac{y(\tau_{s}/\rho)^{\frac{1}{2}}}{v}$$
 (2-6)

Figure 2-3 represents, therefore, a means by which the velocity profile in the liquid film can be prescribed once the value of the interfacial shear stress $\frac{1}{s}$ is determined; the latter must result from the solution of the momentum equation for the gas-flow regime.

The mass flow rate of liquid per unit spanwise length, denoted by m', and the thickness of the liquid film, denoted by δ_{f} , are related due to the continuity condition Hence,

$$m' = \rho_1 \int_{C} u \, dy \tag{2-7}$$

For the case where the film is totally laminar, Eqs 2-4 and 2-7 can be combined to yield the following expression for the thickness



of the liquid film:

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$$\delta_{f} = \left| \frac{2m'}{\tau_{s}/\nu_{l}} \right|^{\frac{1}{2}}$$
(2-8)

To consider the relationship between m' and δ_{f} for the case where the liquid film is turbulent, it is convenient to rewrite Eq. 2-7 in the following non-dimensional form. Thus

$$W^{+} = \int_{0}^{\delta} u^{+} dy^{+}$$
(2-9)

where the parameter W^{\dagger} is termed the <u>dimensionless liquid flow rate</u>, and is defined by

$$W^{+} \equiv \frac{m'}{(\rho_{v})_{1}}$$
(2-19)

and where δ_{f}^{+} is the dimensionless film thickness defined by

$$\delta_{f}^{+} = \frac{\delta_{f} \left(\frac{1}{\sqrt{p_{1}}} \right)^{\frac{1}{2}}}{\sqrt{p_{1}}}$$
(2-11)

The parameters W^{\dagger} and δ_{f}^{\dagger} can be related by employing the "law of the wall," presented in Fig. 2-3, and graphically integrating Eq.2-9.

Figure 2-4 presents W^+ as a function of δ_f^+ , as determined by Kinney, et al. (5). The figure illustrates, as does Eq. 2-8, that the thickness of the liquid film is proportional to the liquid flow rate m' and the kinematic viscosity v_1 , and is inversely proportional to the interfacial shear stress τ_s . The significance of the parameter W^+ in determining the physical characteristics of the flow of a thin liquid film is discussed in Section 2.4.

Energy Transfer to the Liquid Film. Energy is transferred to the liquid film from the "hot" gas stream. A small percentage of that energy is utilized for increasing the sensible enthalpy of the liquid in the film. A significant percentage of that energy, however, causes the evaporation of liquid which was raised to its saturation state while flowing downstream.

If the solid surface wetted by the film is thermally insulated from the external environment (i.e., adiabatic), then, after that surface is heated to an equilibrium temperature which is approximately equal to that of the liquid film, all of the incident energy not utilized for increasing the sensible enthalpy of the liquid is utilized to evaporate it. On the other hand, if the solid surface is externally cooled, then a portion of the energy transferred to the liquid film will be transferred through the liquid film and through the solid surface to the external coolant. Thus, with external-regenerative cooling of the solid surface, longer film-cooled lengths will be realized for a given rate of coolant injection than is the case for an adiabatic solid surface.



FIG. 2-4, VARIATION OF THE LIQUID FLOW LATE PARAMETER W⁺ WITH THE DIMENSIONLESS FILM THICKNESS 8⁺_f

<u>Heat Transfer "Blocking."</u> The vapor produced by vaporizing the liquid film coolant is convected and diffused into the boundary layer of the hot gas wetting the liquid film and establishes in turn a concentration (or partial-pressure) boundary layer. In the process of being transported away from the gas-liquid interface, the vapor serves an important function, referred to as the "blocking" of the incident heat flux. In other words, as the vapor moves away from the gas-liquid interface it "absorbs" sensible heat, so that it thus represents a counter-convective heat transfer mechanism.

<u>Film termination</u>. At station $x = x_1$, in Fig. 2-1, the liquid film terminates, due to evaporation and possible entrainment of liquid droplets into the gas stream. An undesirably high degree of droplet entrainment is known to arise when large wave crests, termed disturbances, occur on the surface of the liquid film. The phenomenon of large disturbances on the surface of the film is termed <u>film</u> instability and is discussed in Section 2.4.

<u>Coupling of the Transport Processes and the Boundary Conditions</u>. The interdependence of the heat and mass transfer processes and the related boundary conditions for the evaporation process is illustrated by the following points:

(a) The rate at which energy is transferred to the liquid film and the physical properties of the liquid (particularly its latent heat of vaporization) determine the rate at which vapor is "injected" into the gaseous boundary layer.

(b) The rate of "vapor injection" determines the degree of the aforementioned "blockage" phenomenon and thus the net rate of energy transfer to the liquid film.

(c) The concentration of the vapor at the interface is dependent on the rate at which vapor is introduced into the boundary-layer region (see Section 2.2.1, Eq. 2-23).

(d) The temperature at the interface is determined by the concentration of the vapor at the interface (or, perhaps more correctly, the partial pressure of the vapor at the interface).

(e) The rate at which energy is transferred to the liquid film is dependent on the thermal potential which exists between the main stream and the gas-liquid interface.

The physics of the evaporation process thus require that for a given set of bulk flow conditions (i.e., working fluids, main-stream temperature, pressure, Reynolds number, Mach number, etc.), unique values for each of the following must result: the rate at which energy is transferred to the liquid film; the rate at which liquid is evaporated; and the concentration of the vapor and the temperature at the gas-liquid interface.

2.1.2 The Gas-Vapor-Cooled Region¹

Figure 2-5 illustrates schematically what is defined as the gasvapor-cooled region, together with representative profiles of the wall temperature for the cases where the wall is adiabatic and is

¹ The terminology "vapor-cooled region" introduced in Refs (6) and (7) would appear to be a misnomer in that the quantity of relatively cool gas which exists throughout the boundary layer at x=x₁ contributes substantially to the gas-vapor-cooled length.



FIG. 2-5. A SCHEMATIC REPRESENTATION OF THE PERTINENT WALL TEMPERATURE PROFILES FOR THE GAS-VAPOR-COOLED REGION

externally cooled. The length of the gas-vapor-cooled region, given by $x_2 - x_1$ in the figure, is defined as that in which the temperature of an adiabatic surface, denoted by t_{aw} , increases from the temperature of the liquid film at $x = x_1$, denoted by $t_{s,1}$, to the recovery temperature for the main stream

In the surface downstream from the end of the liquid film is cooled externally, then a wall temperature distribution such as that denoted by t_w in Fig 2-E results. In general, at any station $x \ge x_1$, the rate of heat transfer to the external coolant is given by

 $q_w = h_w^{\prime} (t_{aw} - t_w)$

where h'_W is the appropriate heat transfer film coefficient to be employed with the thermal potential $(t_{aW} - t_{W})$ in the calculation of the wall heat flux q_W . To either design an effective regenerativelycooled downstream region or determine the optimum location for introducing a second liquid-film-cooled region, it must be possible to determine the streamwise distribution of q_W for a prescribed distribution of t_W . Hence, the basic problem is that of determining the streamwise distribution of the adiabatic-wall temperature, together with the heat transfer film coefficient h'_W . An analysis for the gasvapor-cooled region is presented in Section 4 which considers those two points.

2.2 Conservation of Mass and Energy at the Gas-Liquid Interface

The conservation of mass and energy across the gas-liquid interface is considered in Sections 2.2.1 and 2.2.2, respectively. Consideration is given only to the case where the liquid film is nonreactive and is stable, free from the consequences of large disturbances (see Section 2.4). It is not assumed that the film is either smooth (free from waviness) or laminar. Moreover, mass diffusion in the boundary layer is treated as a binary process; i.e., the evaporated vapor and the gas entrained from the main stream into the boundary layer are considered as the two counter-diffusing species.

2.2.1 Interfacial Mass Balance

Figure 2-6 illustrates a general interfacial element for masstransfer, together with the several mass fluxes which are discussed in the following paragraphs. The S- and L-surfaces presented in Fig. 2-6 are defined as being placed in the gas-vapor phase and in the liquid phase, respectively, at an infinitesimal distance from the interface Additional definitions and subscript notation needed to clarify Fig 2-6 are as follows:

Definitions:

$$(\rho v)_{v} = \rho_{v} v_{v} = \rho_{v} (v + v_{d,v})$$
 (2-12)

$$(\rho v)_{g} = \rho_{g} v_{g} = \rho_{g} (v + v_{d,g})$$
 (2-13)

$$(\rho v) = (\rho_g + \rho_v) v$$
 (2-14)
 $\rho_i = \frac{p_i}{R_i t}$ = the partial density for the "i" species; i=g,v



FIG. 2-6. A MASS-TRANSFER INTERFACE ELEMENT FOR A NON-REACTIVE, STABLE LIQUID FILM

Subscripts

v - vapor

g - gas entrained from the main stream

L - evaluated at the L-surface

S - evaluated at the S-surface

d - diffusional component of the subscripted parameter

A term having neither "v," "g" nor "L" as a subscript refers to the gas-vapor mixture

Figure 2-6 illustrates that when the liquid coolant is evaporating into a medium other than its own vapor, mass transport is due to both diffusion and convection. The relationships governing the interdependence of the different convective and diffusive mass fluxes presented in Fig. 2-6 are derived in the following paragraphs

Fick's law relates the mass transfer by diffusion in the y direction to the concentration gradient in the y direction, and is expressed by

 $(\partial v_d)_i = -\rho (D + D_t) \left(\frac{\partial U_i}{\partial y}\right) (1 v, g)$ (2-15)

where

$$C = \frac{y_1}{y_1}$$
 = the concentration of the "i" species (i = g,v).

$$(D + D_t) =$$
 the "effective" diffusion coefficient

In the case of a binary medium, to which the subject discussion is restricted, the concentration gradients $\partial C_{v}^{}/\partial y$ and $\partial C_{q}^{}/\partial y$ are related by

$$\frac{\partial C_{v}}{\partial y} = -\frac{\partial C_{g}}{\partial y}$$
(2-16)

Equations 2-15 and 2-16 show that at any station y the rates of mass diffusion of the vapor, $(\wp v_d)_v$, and the gas, $(\wp v_d)_g$, are related by

$$(\wp v_d)_g = -(\wp v_d)_v$$
 (2-17)

Since the interface is impermeable to the gas, the diffusive flux of gas towards the liquid surface is exactly equal to the convective flux of gas away from the liquid surface; i e , the net transport of gas at the surface of the liquid film is zero. 1 Thus

$$(c v_d)_{g,S} = -(c_g v)_S$$
 (2-18)

Hence

$$v_{d,q,S} = -v_S$$
 (2-19)

Note, however, that the concentration of the gas is always greater than zero at the interface 1

Equations 2-17 and 2-19 can be combined to yield

$$(\rho v_d)_{v,S} = (1 - C_{v,S})(\rho v)_S$$
 (2-20)

Equations 2-12 and 2-20 can be combined to yield

$$(\rho v)_{S} = (\rho v)_{V,S}$$
(2-21)

Hence, Eqs. 2-20 and 2-21 show that at the interface, the diffusive rate of vapor transport, $(\rho v_d)_{v,S}$, and the total rate of vapor transport, $(\rho v)_{v,S}$, are related by

$$(\rho v_d)_{v,S} = (1 - C_{v,S})(\rho v)_{v,S}$$
 (2-22)

It follows from the conservation of the evaporating species across the interface and from Equations 2-21, 2-22, and 2-15, that

$$(\rho v)_{v,S} = (\rho v)_{L} = (\rho v)_{S} = - \left| \frac{\overline{\rho}(D+D_{t})}{1-C_{v}} \frac{\partial C_{v}}{\partial y} \right|_{S}$$
(2-23)

Equations 2-15, 2-17, 2-19, 2-22, and 2-23 describe, for the process of evaporation, the pertinent relationships between the several mass fluxes presented in Fig 2-6. Those equations also can be written in terms of the partial pressure of the vapor. Consider Eq. 2-15: from the definitions of the concentration of the vapor, C_v , and the partial pressure of the vapor, P_v , it follows that

$$\frac{\partial C_{\mathbf{v}}}{\partial \mathbf{y}} = \frac{\partial}{\partial \mathbf{y}} - \frac{\mathbf{p}_{\mathbf{v}} \mathbf{R}}{\mathbf{p} \mathbf{R}_{\mathbf{v}}} = \frac{\mathbf{R}}{\mathbf{p} \mathbf{R}_{\mathbf{v}}} - \frac{\partial}{\partial \mathbf{y}} + \frac{\mathbf{C}_{\mathbf{v}}}{\mathbf{R}} - \frac{\partial \mathbf{R}}{\partial \mathbf{y}}$$
(2-24)

The perfect gas constant for the gas-vapor mixture can be written as

$$R = C_v R_v + (1 - C_v) R_g$$
 (2-25)

Hence

$$\frac{\partial R}{\partial y} = (R_v - R_g) \frac{\partial C_v}{\partial y}$$
(2-26)

Substituting Eq. 2-26 into Eq. 2-24, solving for $\partial C_v/\partial y$; and utilizing Eq. 2-25 yields

$$\frac{\partial C_{v}}{\partial y} = \frac{R^{2}}{R_{v}} \frac{1}{R_{g}} \frac{\partial P_{v}}{\partial y}$$
(2.27a)

$$= \frac{1-C_{v}}{p-p_{v}} \frac{R}{R_{v}} = \frac{\partial p_{v}}{\partial y}$$
(2.27b)

Substituting Eq. 2-27a into Eq. 2-15 yields the following expression for the diffusive rate of vapor transport at any station y:

$$(\rho v_d)_v = -\rho (D+D_t) \frac{R^2}{R_g R_v} \frac{1}{p} \frac{2p_v}{\partial y}$$
 (2-28)

Equations 2-27b and 2-23 can be combined to yield the following expression for the conservation of mass across the interface:

$$(\rho v)_{S} = (\rho v)_{L} = (\rho v)_{V,S} = - \begin{vmatrix} \overline{\rho}(D+D_{t}) & R & \partial p_{v} \\ R_{v}(p-p_{v}) & \partial y \\ \hline R_{v}(p-p_{v}) & S \end{vmatrix}$$
(2-29)

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2.2.2 Interfacial Energy Balance

Figure 2-7 illustrates a general energy-transfer interface element placed at the surface of a stable, non-reactive liquid film. The different energy fluxes involved are defined as follows:

- (1) $q_S = (k \frac{at}{ay})_S$ = the rate at which energy is transferred across the S-surface by conduction
- (2) $q_L = (k \frac{\partial t}{\partial y})_L$ = the rate at which energy is transferred across the L-surface by conduction.
- (3) $(\rho v)_{v,S} h_{v,S}$ = the rate at which energy is convected across the S-surface by the evaporated vapor.
- (4) $(\rho v)_L h_L$ = the rate at which energy is convected across the Lsurface by the evaporating liquid.
- (5) q_{r,S} = the rate at which energy is transferred across the S-surface by radiation
- (6) q_{r,L} = the rate at which energy is transferred across the L-surface by radiation.
- (7) $(KE)_{S} = (vv)_{S} = \frac{vS}{2} + \frac{vS}{2} = the rate at which kinetic energy associated with the gas-vapor phase crosses the S-surface.$
- (8) $(KE)_{L} (Pv)_{L} (\frac{-v_{L}^{2}}{2} + \frac{-v_{L}^{2}}{2}) =$ the rate at which kinetic energy associated with the liquid phase crosses the L-surface.

Since the net transport of the gas at the surface of the liquid film is zero,¹ only the transport of enthalpy associated with the evaporated vapor (see Item (3) above) needs be considered on the <u>gas-vapor side of the interface. It is desirable, therefore, that</u> ¹See discussion leading to Eq. 2-18.



FIG. 2-7. AN ENERGY-TRANSFER INTERFACE ELEMENT FOR A STABLE, NON-REACTIVE LIQUID FILM

the vapor have a relatively large specific heat so that the vapor will cause a large amount of thermal "blockage."

In view of Eq. 2-23, the energy balance for the interface element illustrated in Fig. 2-7 can be written as follows:

$$q_{S} = q_{L} + \left| (KE)_{S} - (KE)_{L} \right| + (\rho v)_{S} (h_{v,S} - h_{L}) - (q_{r,S} - q_{r,L})$$
(2-30)
I II III IV V

Term IV in Eq. 2-30 can be rewritten in the form

(ρν)₅ ΔΗ_ν

where ΔH_v is the latent heat of vaporization for the liquid coolant which corresponds to the saturation conditions prescribed by the partial pressure of the vapor at the interface. If the partial pressure of the vapor at the interface is equal to or greater than the critical pressure for the liquid coolant, then term IV in Eq 2-30 is identically zero. In most instances, term III in Eq. 2-30 can be considered negligible in comparison to a term such as term I Thus, neglecting term III, Eq. 2-30 can be rewritten to yield

$$q_1 + q_{r,1} = q_s + q_{r,s} - (\rho v)_s \Delta H_v$$
 (2-31)

Equation 2-31 is the general energy balance for conditions at the interface. To obtain a better understanding of what is involved in evaluating the left-hand side of Eq. 2-31, the energy balance for the entire liquid film must be considered, and that energy balance is developed in the following section.

2.3 Energy Balance on the Liquid Film

2.3.1 General Energy Balance

Figure 2-8 illustrates diagrammatically a control volume placed within a stable, non-reactive liquid film. The following energy fluxes may be defined:

- m' h_{1,0} = the rate of enthalpy transfer into the control volume per unit spanwise length due to the continuous in-jection of film coolant.
- (2) $q_w \equiv (k \frac{\partial t}{\partial y})_w =$ the rate of conduction of heat from the liquid film to the wall.
- (3) $q_{r,w}$ = the rate of radiant energy transfer to the wall.

The rate of injection of liquid into the control volume illustrated in Fig. 2-8 per unit spanwise length is denoted by m'_0 . From the conservation of liquid film coolant it follows that

$$m_{0}^{\prime} = \int_{x_{0}}^{x_{1}} (\rho v)_{S} dx \qquad (2-32)$$

Employing Eq. 2-32, the energy balance for the control volume presented in Fig. 2-8 is given by

$$\frac{f_{1}}{x_{0}} \left(q_{L} + q_{r,L} \right) dx = \frac{f_{1}}{x_{0}} \left[(\rho v)_{S} (h_{L} - h_{1,0}) + q_{w} + q_{r,w} \right] dx$$
 (2-33)

Equation 2-33 is the general energy balance for the liquid film. For the special case where the summation of the terms under



GAS STREAM ---

FIG. 2-8. CONTROL VOLUME FOR ENERGY BALANCE ON A STABLE. NON-REACTIVE LIQUID FILM

each of the integral signs in Eq. 2-33 is independent of x, the integral signs can be removed and one obtains a local energy balance for the liquid film. Furthermore, the relationship obtained after removing the integral signs can be combined with Eq. 2-31, obtaining thereby a loca⁷ energy balance at the interface. Thus

$$q_{S} + q_{r,S} = q_{w} + q_{r,w} + (\rho v)_{S}(h_{L} - h_{1,o}) + (\rho v)_{S} \Delta H_{v}$$
 (2-34)

In many liquid-film cooling applications of interest, however, the terms under the integral signs in Eq. 2-33 are not independent of the streamwise coordinate x. Of particular interest is the problem of liquid-film cooling an exhaust nozzle wherein severe pressure gradients are common, and wherein the various terms in Eq. 2-33 are most certainly functions of x. In such a situation it is difficult to appraise the error involved when it is assumed that Eq. 2-34 is the local energy balance at the interface. In particular, it is difficult to appraise the error due to assuming that the portion of the incident heat flux utilized for increasing the sensible enthalpy of the liquid is equal to $(pv)_{\rm S}$ $(h_{\rm L} - h_{1,o})$, as given by Eq. 2-34. Since Eq. 2-34 is much more convenient to employ, however, than is Eq. 2-31, further consideration of the problem is worthwhile. A method for experimentally investigating that problem is suggested in Section 3.3.1.

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÷.'
2 3.2 Simplified Energy Balances Employed by Previous Investigators
 Knuth (8) simplified the general interfacial energy balance (see
 Eq 2-31) by assuming, in essence, that

 $q_{r,s} = q_L + q_{r,L}$

In other words, Knuth equated the radiant energy incident on the surface of the liquid film to the energy utilized for increasing the sensible enthalpy of the liquid coolant plus the energy transferred to the solid surface wetted by the film. (In Knuth's experimental investigation, the wetted surface was approximately adiabatic) Knuth, therefore, employed the following interfacial energy balance Thus

 $q_{S} = (\omega v)_{S} \Delta H_{v}$ (2-35)

Crocco (9) utilized the same interfacial energy balance as Knuth, but Crocco's reasoning was somewhat different. He assumed that the rate of radiant energy transfer to the liquid film was negligible and that the wetted surf ice was adiabatic. Crocco further assumed that in a so-called "first period of existence," all of the film coolant was heated to equilibrium saturation conditions, and that for the major portion of the liquid-flow regime, all of the energy incident on the liquid film was utilized for evaporating it. Moreover, the analysis was concerned only with that major portion of the liquid-flow regime. Graham (10), Sellers (11), Kinney (5), and Emmons (6) employed an interfacial energy balance which is a simplification of Eq. 2-34. First, it was assumed that the radiant energy transfer was negligible. Furthermore, in the experimental investigations conducted by Graham, Kinney, and Sellers, and in a portion of the experimental investigation conducted by Emmons, the wetted surface was essentially adiabatic. Thus, the interfacial energy balance employed by those investigators for the case where the wetted surface was adiabatic is

$$q_{S} = (\rho v)_{S} (h_{L} - h_{L,\rho}) + (\rho v)_{S} \Delta H_{v}$$
(2-36)

2.4 Gas-Liquid Interactions

The specification of the physical characteristics of the gasliquid interface is of primary importance in the analysis of liquidfilm cooling. Considerable experimental data are available that indicate that the interfacial structure has a considerable influence on the flow rate of film coolant required for establishing a specified liquid-film-cooled length (5)(8)(10)(11). Of particular significance is the phenomenon of <u>film instability</u> which causes the entrainment of liquid coolant droplets by the flowing gas stream. The physical nature of the gas-liquid interface is discussed in Section 2.4.1 and the phenomena of film instability and entrainment are discussed in Section 2.4.2.

2.4.1 Interfacial Structures

There appear to be three distinct interfacial structures associated with the process of liquid-film cooling that are of

interest. Figure 2-9 presents photographs of each of those interfacial structures. The photographs were taken during certain preliminary investigations conducted at the Jet Propulsion Center, Purdue University (12). The following features of the basic flow system and of the photographic apparatus are noteworthy: (a) the gaseous and liquid phases were air and deionized water, respectively; (b) the gaseous flow was turbulent; (c) a 35 mm camera equipped with a close-copy lens was employed in conjunction with a stroboscopic light source, which had a flash time of 2 microseconds, so that the gas side of the two-phase interface could be photographed; and (d) the optical line of sight was at an angle of 10⁰ with respect to the wetted test surface which was a highly polished, flat, stainless steel plate having a spanwise dimension of 4 ins..

The photographs presented in Fig. 2-9 were taken when the static pressure and the main-stream temperature had values of 220 psia and 1000 R, respectively. Flow parameters which were not common to all of the photographs are specified in the appropriate subtitles of Fig. 2-9; they are the main-stream velocity, u_e , the position downstream of the point of liquid-film coolant injection at which the photographs were taken, $x - x_0$, and the rate of liquid-film coolant injection per inch of injector slot width, m'_0 .

The interfacial structure presented in Fig. 2-9a is characterized by its extreme smoothness. In that particular photograph there is shown a $\frac{1}{2}$ in. diameter cylindrical plug which protrudes approximately 0.001 in. above the surface of the polished stainless steel surface.



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a. smooth ($u_e = 88 \text{ fps}$, $x - x_o = 9 \text{ ins}$, $m'_o = 0.001 \text{ lb/in-sec}$)



b."pebbled" ($u_e = 50 \text{ fps}, x - x_o = 5 \text{ ins}, m_o = 0.002 \text{ lb/in-sec}$)



c."pebbled" ($u_e = 130 \text{ fps}, x - x_o = 5 \text{ ins}, m_0 = 0.001 \text{ lb/in-sec}$).



d. unstable ($u_8 = 130 \text{ fps}, x - x_0 = 5 \text{ ins}, m_0^1 = 0.0045 \text{ lb/in-sec}$)

FIG. 2-9. CHARACTERISTIC INTERFACIAL STRUCTURES FOR LIQUID-FILM COOLING

The fortuitous presence of that surface irregularity serves to indicate that a liquid film which has a smooth interfacial structure is, in general, very thin and sensitive to relatively small surface roughnesses. The smooth character of the interfacial structure is apparently due to the fact that the liquid film is so thin and viscous that any disturbances imparted to it as a consequence of turbulent velocity fluctuations in the adjacent gas phase are completely suppressed or "damped out." Because of the aforementioned sensitivity of a thin film to surface roughnesses, it is highly improbable that the liquid films employed for liquidfilm cooling the walls of a rocket thrust chamber will have an extremely smooth interfacial structure.

The interfacial structure presented in Figs. 2-9b and 2-9c is quite common and is characterized by its "pebbled" appearance. The small disturbances which are evident on the surface of the liquid film are apparently a consequence of turbulent velocity fluctuations in the gas phase flowing past the liquid film (13). The liquid film is sufficiently thin and viscous that small disturbances which appear on its surface do not amplify and cause film instability. (It should not be implied, however, that there may not exist turbulent stresses in the liquid film.) The difference in the scale of the interfacial disturbances for the two "pebbled" interfaces presented in Fig. 2-9 is due primarily to a corresponding difference in the scale of the turbulence in the adjacent flowing gas; the turbulence scale of the gas phase decreases with increasing gas mass flow rate. That relationship between the gas mass flow rate and the nature of the "pebbled"

interfacial structure has been observed previously (5)(8)(13)(14).

The interfacial structure presented in Fig. 2-9d defines the phenomenon of <u>film instability</u>; the latter is characterized by the existence of large-scale disturbances which are turbulent in nature, and which are superimposed on a "pebbled" interfacial structure The phenomenon of film instability and the related phenomenon of liquid entrainment are discussed in the following section.

2.4.2 Film Instability and Entrainment

A large degree of liquid entrainment is known to occur when large-scale disturbances appear on the surface of a liquid film (15). Such entrainment reduces the effectiveness of the liquidfilm cooling process in either or both of two ways:

1. If the entrained coolant is convected out of the region which is to be liquid-film cooled before it is evaporated, then a corresponding loss in liquid-film cooling effectiveness arises due to the incomplete utilization of the available heat of vaporization associated with the flow of the liquid-film coolant.

2. If the entrained liquid droplets are evaporated immediately upon leaving the interface, then the corresponding multi-fold increase in the specific volume of the coolant upon evaporation will have a disturbing effect on the gaseous boundary layer structure which, in turn, might tend to promote the degree of turbulent heat, mass, and momentum transport near the gas-liquid interface. Luikov (16, 17) has termed such a phenomenon volumetric boiling.

Thus, a problem of primary importance is that concerned with the specification of (a) the physical conditions under which smallscale interfacial disturbances will amplify and cause film instability, and (b) the degree of and the effects of liquid coolant entrainment once transition has occurred. Intuitively, parameters which may be of importance in a formulation of the solution to the above problem are as follows:

(i) the interfacial shear stress, τ_s , which represents the degree to which the gaseous flow regime tends to deform the liquid film; (ii) the dynamic viscosity of the liquid, μ_1 , which is a measure of the resistance of the liquid film to deformation;

(iii) the thickness of the liquid film, δ_f , which represents the extent to which the gas-liquid interface experiences the stabilizing influence of the wetted solid surface;

(iv) the kinematic viscosity of the liquid, v_1 , which is a measure of the damping characteristics of the liquid film;

(v) the interfacial density ratio, ρ_S/ρ_1 , which represents the relative inertia characteristics of the two phases, and is a measure of the degree to which a turbulent velocity fluctuation in the gaseous phase will influence the liquid-flow regime;

(vi) the dynamic viscosity ratio μ_S / μ_l , which is a measure of the discontinuity in the velocity profile at the gas-liquid interface; and

(vii) the surface tension of the liquid phase.

References (5)(13)(14)(15) have experimentally investigated the phenomenon of film instability. Subsequent attempts to correlate the

inception of film instability have employed the dimensionless liquidflow rate W^+ (see Eq. 2-10). It is of interest to note that W^+ combines the individual influences of the parameters (i)-(iv) listed above into a single parameter.

The motivation for attempting to correlate the inception of film instability in terms of the single parameter W^+ arises from the following facts which characterize single phase turbulent flow: (a) a universal relationship between u^+ and y^+ (the "law-of-the-wall") characterizes the "inner" region (y^{+} < 1000) of the wall turbulent region¹; and (b) the distance normal to the wall at which turbulent forces are of the same order of importance as the viscous forces is defined approximately by $15 < y^+ < 20$ and $90 < W^+ < 150$ (5)(18). Thus, the transition state for the liquid-flow regime would be characterized by corresponding limits of δ_f^+ and W^+ if the following basic hypothesis is physically correct: (1) parameters (v)-(vii), presented in the foregoing, do not appreciably affect the transition characteristics of the liquid-flow regime; (2) the liquid-flow regime obeys the universal "law-of-the-wall" as established for single phase flow; and (3) transition occurs when the turbulent forces in the region of the liquid film close to the gas-liquid interface are of the same order of importance as the viscous forces.

While a large percentage of the experimental data tends to substantiate the foregoing simple hypothesis (5)(15), a limited amount of data has been obtained which indicates that the simple <u>hypothesis may not give a complete characterization of the phenomena.</u> ¹ See Fig. 2-3.

For example, experimental observations indicate that the critical value of W^+ increases slightly with decreasing surface tension (5), and that the parameter u_S/u_1 may also be significant in defining the transition state (13)(14). Furthermore, at relatively low values of the gas stream Reynolds number $1(<10^4)$ it is known that the critical value of W^+ can increase significantly (19); i.e., the transition state is a function of both the Reynolds number for the gas stream and W^+ . It would appear, however, that at values of the gas flow in a rocket thrust chamber $(>10^5)$, the dependence of the critical value of W^+ on the Reynolds number for the gas stream for the gas flow in a rocket thrust chamber $(>10^5)$, the dependence of the critical value of W^+ on the Reynolds number for the gas stream is negligible (5)(14)(19).

For the practical design of a liquid-film-cooled system which is characterized by gas-stream Reynolds numbers greater than 10^5 , the following generalized results given by Kinney, et al. (5) appear to be adequate until a more complete theory is formulated and substantially verified:

1. In the situation where the evaporation rate is negligible, and wherein m' is thus independent of the streamwise coordinate x, the inception of film instability is approximately defined by $W_0^+ = 90$ where

$$W_0^+ = \frac{m_0^+}{\mu_1}$$
 (2-37)

The viscosity μ_1 in Eq. 2-37 is evaluated at a streamwise-mean liquid-film temperature.

¹ The Reynolds number referred to on this page is R_d, defined on page 57.

2. In the case of liquid-film cooling, wherein the value of W^+ varies from a nonzero value at $x = x_0$ to zero at $x = x_1$, the onset of a high degree of liquid entrainment--which accompanies a corresponding decrease in liquid-film cooling effectiveness--is approximately defined by $W_0^+ = 360$.

Graham (10,20) experimentally investigated the relationship between the effectiveness of liquid-film cooling and the phenomenon of liquid entrainment. He introduced a <u>stability effectiveness</u> ϵ_s defined by

$$s = \frac{m'o,i}{m'o,a}$$
(2-38)

where

ε

 $m'_{0,i}$ = the ideal rate of coolant consumption (zero entrainment).

 $m'_{0,a}$ = the actual rate of coolant consumption.

Graham correlated ϵ_s in terms of W_0^+ utilizing the experimental data of References (5) and (10); the resultant correlation for smoothtube flow is presented in Fig. 2-10. It should be noted that the correlation was developed utilizing the assumption that the heat transfer film coefficient h_s^+ is independent of W_0^+ . There appear to exist, however, considerable experimental data which indicate that h_s^+ is significantly dependent on W_0^+ ; the latter phenomenon is discussed in Section 3.2.



FIG. 2-10. THE CORRELATION DUE TO GRAHAM(10) FOR THE STABILITY EFFECTIVENESS \in_s AS A FUNCTION OF THE LIQUID FLOW RATE PARAMETER W_0^+

The stability effectiveness $\varepsilon_{\rm s}$ represents a measure of an effect of the entrainment phenomenon rather than a direct measure of the actual quantity of coolant which is entrained by a flowing gas stream. The latter problem has been experimentally investigated by Hewitt, et al. (21) for the case of annular two phase flow. In that subject investigation the quantity of entrained liquid was determined by employing a sampling probe which traversed the central core of the two-phase flow field. It was observed that below a critical value of liquid flow rate, the entrainment was negligible; the critical state was approximately defined by W_0^+ = 90 (15). As the liquid flow rate increased above the critical value, the trainment of liquid into the central gas core increased correspondingly. It was observed, however, that the degree of entrainment increased with increasing rate of gas flow as well as with increasing rate of liquid flow. Furthermore, additional experiments conducted at different system pressures indicated that the degree of entrainment increased with decreased pressure, the mass flow rate of gas being held constant. Those two facts would indicate that the interfacial shear stress $\tau_{\rm s}$, in addition to the dimensionless liquid flow rate W_0^+ , might be of significance in determining the actual quantity of coolant which is entrained by a flowing gas stream.

The values of the experimental parameters investigated by Hewitt, et al. (21) were significantly different from those which would normally characterize an application of liquid-film cooling. In particular, the temperature and pressure employed were essentially ambient and the evaporative rate of mass transfer was zero. It is

impossible, therefore, to determine from those data the relationship which exists between the stability effectiveness ε_s and the actual quantity of coolant which is entrained by a flowing gas stream. It would appear that a need exists for additional experimental investigation of the phenomenon of liquid entrainment.

3. ANALYZING THE LIQUID-FILM-COOLED REGION

A review of the previous analyses for liquid-film cooling is presented in Section 3.1, and a review of some pertinent experimental data is presented in Section 3.2. The latter serves to illustrate the limitations of those analyses that are reviewed, and provides the motivation for the alternate correlation procedure that is presented in Section 3.3 for heat-mass transfer data that are obtained in liquidfilm cooling experiments

3.1 A Review of the Previous Analyses

Table 1 compares the analyses due to Knuth (8), Sellers (11)(22), Graham (10), Rannie (23)¹, and Emmons (6). Column I of that table shows that similar sets of equations are employed in the analyses due to Knuth, Sellers, Graham, and Rannie; consequently, those analyses are considered concurrently in Section 3.1 1. Because the analysis due to Emmons is one most recent analysis developed, and also since it differs fundamentally from the earlier analyses, that analysis is considered separately in Section 3.1.2

The fundamental steps that are common to each of the subject analyses are outlined briefly below:

Elementary boundary-layer theory is employed to analyze the Step 1. gaseous-flow regime above the liquid film. In the analyses due to Knuth.

As modified for liquid-film cooling by Sloop (24).

TABLE I

A COMPARISON OF THE PREVIOUS ANALYSE

COLUMN	1	II	
REFERENCE	EQUATIONS EMPLOYED	UNKNOWNS ELIMINATED FROM EGUATIONS EMPLOYED	THE FUNDAMENTAL ANALYTICAL SOLUTION(S) FOR eta \equiv
RANNIE (23)	3-4 [#] , 3-5 [#] , 3-6, 3-7, and terms (b) and (c) of 3-10	⁷ 8 · ⁷ 8 · ¹ 8 · ⁹ 8	$\beta + i = \left[1 + \frac{c_{p_v}}{c_p} \left(\frac{u_v}{u_g} - 1 \right) \left\{ i - \exp\left(- \frac{(\rho v)_s}{\mu} \delta \right) \right\} \right] \exp\left(1 - \frac{(\rho v)_s}{\mu} \delta \right]$
SELLERS (22)		τ ₈ , † ₈ , q ₈ , δ	$\beta + 1 = \left[1 + \frac{\frac{(\rho \cdot)_{S}}{(\rho \cdot)_{e}}}{\binom{u_{e}}{(u_{B})} \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}} Pr} \left[1 + \frac{\frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}} Pr} \left[1 + \frac{\frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}} Pr} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}} Pr} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}} Pr} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}} Pr} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}} Pr} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}} Pr} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}} Pr} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}} Pr} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}} Pr} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}}} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}}} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}}} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}}} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}}} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}}} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}}} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}}} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}}} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}}} \left[1 + \frac{C_{p}}{c_{p}} \left(\frac{u}{u}\right)}{1 + \left(\frac{\tau_{s}}{\rho \cdot u_{e}^{2}}\right)} \right]^{\frac{C_{p}}{c_{p}}} \left[1 + $
SELLERS (11)	3~6, 3~7, terms (b) and (c) of 3~10 and terms (a) and (c) of 3~14	τ _δ , t _δ , δ	$\beta + 1 = \left[1 + \frac{\frac{(\rho_v)_s}{(\rho_u)_e}}{\frac{u}{s}(\frac{\sigma_v}{\rho_u^2})} \right] \xrightarrow{c_{p_v}} x \left[\frac{1}{P_r} \left(\frac{u}{u_s} - 1 \right) + 1 \right]$
SELLERS(11,22) GRAHAM (10)	3-6,3-7; terms (b) and (c) of 3-0, and terms (b) and (c) of 3-14	· · · · · · · · · · · · · · · · · · ·	$\beta + 1 = \left[1 + \frac{\frac{(\rho v)_{S}}{(\rho u)_{\bullet}}}{\left(\frac{v}{u}_{B}\right)\left(\frac{\tau_{\bullet}}{\rho_{\bullet}}\right)^{2}}\right]^{\frac{C_{\rho_{v}}}{C_{\rho}}} + \left[\frac{\frac{(\rho v)_{S}}{(\rho u)_{\bullet}}}{\left(\frac{u}{u}_{B}\right)\left(\frac{\tau_{\bullet}}{\rho_{\bullet}}\right)^{2}}\right] \left(\frac{C_{\rho}}{c}\right]$
			(Graham assumed (C _{Pv} /C _p)Pr = 1)
KN (TH (8)	3-3 ^{**} , 3-4 ^{**} , 3-5 ^{**} , 3-6, 3-7, 3-8, 3-72, and data for P _{v,5} =f(t ₅)	^τ ε, † _δ , ^μ δ, ^μ ν,δ, α _δ , ^(ρν) δ	$\beta + 1 = \left[1 + \frac{(\rho_v)_S}{\tau_B} - u_e - \exp\left(\frac{(\rho_v)_S}{\mu}\delta\right)\right] \frac{C_{\rho_v}}{C_p} = \exp\left(\frac{C_{\rho_v}}{C_p}\right)$
			$\beta + \tau = \left[\frac{p - p_{v,e}}{p - p_{v,s}}\right] \frac{C_{p_v} R}{C_p R_v} = \exp\left\{\frac{C_{p_v}}{C_p} (Pr - Sc)\right\} \frac{(pv)_S}{\mu}$
EMMONS (6)	3-17 and 3-18	τ.	Equation due to Emmons [#] $\beta = \frac{C_{p_v}}{C_p} = \frac{(\rho v)_S}{(\rho u)_0} = (\frac{\rho u^2}{\tau_{s,0}}) = \exp\left[\frac{13.89}{b} - \frac{v_S}{(\tau_{s,0}/\rho)^{1/2}}\right]$
			As corrected by writer $\beta = \frac{C_{P_v}}{C_p} \frac{(\rho_v)_S}{(\rho_u)_e} \left(\frac{\rho_u^2}{\tau_{e,0}}\right) = \exp\left\{\frac{13.89}{b} \frac{v_S}{u_e}\right[\frac{1}{\sqrt{3}} \frac{(\rho_v)_S}{(\rho_v)_e}\right]$

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YSES FOR LIQUID-FILM COOLING

	IV	V
$\beta \equiv \frac{(\rho v)_{S} C_{P_{v}}(1_{e} - 1_{s})}{q_{S}}$	EMPIRICAL AND/OR EXPERIMENTAL EVALUATION OF THE REMAINING UNKNOWNS	THE INTERFACIAL ENERGY BALANCE EMPLOYED
$\exp\left(\Pr\left(\frac{c_{p_{v}}}{c_{p}}\frac{(\rho_{v})_{3}}{\mu}\delta\right)\right)$	$\frac{1}{10} \frac{1}{t_{g}} = 1_{gat}$ 2. $\delta = 5.6 \nu / (\tau_{g,0} / \rho)^{1/2}$ 3. $u_{g} = 5.6 (\tau_{g,0} / \rho)^{1/2}$	$\beta = \frac{C_{p_v}(t_e - t_e)}{(h_L - h_{1,0}) + \Delta H_v} $
$\frac{\frac{p_v}{p} \left(\frac{u_e}{u_g} - 1\right)}{\left(\frac{r_g}{u_g}\right) \left(\frac{u_e}{u_g}\right) \left(\frac{p_u}{p_v}\right) \left(\frac{p_u}{p_v}\right)_g}$ $+ 1 = \frac{1}{p_r} \left(\frac{u_e}{u_g} - 1\right)$ $\frac{\left(\frac{c_{p_v}}{c_p}\right) \left(\frac{u_e}{u_g} - 1\right)}{\left(\frac{c_{p_v}}{c_p}\right) \left(\frac{u_e}{u_g} - 1\right)}$	1. $t_a = t_{adt}$ 2. u_b and τ_a were determined from the data of Ref. 5; the results were $\frac{u_e}{u_b} = 4.5$ $\frac{u_e}{u_b} \frac{\tau_s}{\rho_e u_e^2} = 0.00933$	$\beta = \frac{C_{p_v}(t_e^{-t_a})}{(h_L^{-}h_{i_vo}) + \Delta H_v}$
) 	1. $t_s = t_{sat}$ 2. $u_s = 5.0 (\tau_s / p)^{1/2}$ 3. τ_s was determined from the data of Ref.10	•
$\frac{\frac{p_v}{p}}{\frac{p}{k}} \left(\frac{p_r}{p_r} - 1\right) - \frac{\left(\frac{p_v}{s}\right)}{\mu} \delta\right)$	1. $\tau_g = \tau_{s,o}$ 2. $\delta = 5.6 \nu / (\tau_{s,o} / \rho)^{1/2}$ 3. $p_{v,5}$ and t_s were evaluated by simultaneously satsiying both of the equations given in column III for β and the thermodynamic data for $p_{v,s}$ as a function of t_s	$\beta = \frac{C_{p_v}(t_e - t_e)}{\Delta H_v}$
$\frac{1}{\left(\frac{\rho_{v}}{\rho_{i}}\right)^{1/2}} \left[\frac{\left(\frac{\rho_{v}}{\rho_{i}}\right)_{S}}{\frac{\rho_{v}}{\rho_{i}}\left(\frac{\rho_{v}}{\rho_{i}}\right)_{e}} \frac{C_{\rho_{v}}}{C_{\rho}} \frac{\rho_{e}}{\rho_{i}} \right]^{-1/2} \right\}$	<pre>1 ts=tsat 2 b was determined from the data of Ref 6; the result was b=0.110</pre>	$\beta = \frac{C_{p_v}(t_e^{-t_e})}{(h_i^{-}h_{i_i}) + \Delta H_v}$
	#Wodification by Stoop (24)	Wodification by Sloop (24

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Rannie, Sellers, and Graham, the Prandtl-Taylor theory (18) is extended to include the effects of mass transfer. In the analysis due to Emmons, the theory developed by Turcotte (25) for the case of the transpired turbulent boundary layer on a flat plate is extended to the case of liquid-film cooling The result of those developments is basically an analytical expression for the following non-dimensional parameter^{1,2}. Thus

$$\beta = \frac{\left(\rho v\right)_{S} c_{p_{v}} \left(t_{e} - t_{s}\right)}{q_{S}}$$
(3-1)

where

- q_S = the rate at which energy is transferred across the S-surface³ by conduction.
- $(ov)_{S}$ = the rate at which mass is transferred across the S-surface by convection

$$c_{\rm p}$$
 = the specific heat for the vapor.

Step 2. Another expression is developed for the parameter B by considering the conservation of energy across the gas-liquid interface.⁴

¹ The parameter 8 was denoted by H in the original analyses due to Sellers, and is related to the Stanton number for heat transfer, defined by Eq. 3-25

- ² See Table 1. column III
- ³ See Section 2-2-1 for the definition of the S-surface.

⁴ See Table 1, column V.

That expression for β combined with the expression for β obtained in Step 1 yields, in principle, a single equation in the single unknown $(\rho v)_{S}$.

<u>Step 3</u>. Finally from the conservation of mass for the liquid-film coolant, the coolant flow rate m_0^4 is related to the liquid-film-cooled length L by

$$m_{0}^{\prime} = (\rho v)_{S} L$$
 (3-2)¹

Equation 3-2, which relates m_0^1 and L, is what is desired as the result of each analysis.

In completing Step 1, which constitutes the major part of each analysis, there arises more unknown quantities than there are independent equations. To circumvent that difficulty, one of the following methods is employed:²

(a) Empirical formulae, based on zero mass transfer, are employed to obtain approximate values for the unknown quantities needed for completing the solution. The appropriateness of the assumptions employed is determined by comparing the analytical predictions with experimental data (8)
 (23).

(b) Some of the unknown quantities needed for completing the solution are approximated by means of empirical formulae, and the remainder are determined experimentally. An obvious disadvantage of the latter method

¹ Equation 3-2 follows from Eq. 2-32 and Assumption (1.f) of the analytical model that is presented in the following section.
² See Table 1, column IV.

is that the final analytical expression cannot be applied with confidence to situations that differ significantly from that for which the unknowns were determined. Sellers determined two unknowns and Graham and Emmons each determined one unknown from experimental data.

3.1.1 The Analyses Due to Rannie, Sellers, Graham, and Knuth

Discussed in the present section are the fundamental developments that relate to the completion of Step 1 for the subject analyses.

The Physical and Analytical Models. Figure 3-1 illustrates schematically a physical model characteristic of those employed by References (8)(11)(22)(10)(23). The assumptions common to the different analytical models are as follows:

 $(1)^{1}$ The partial derivatives of the following parameters with respect to the streamwise coordinate x are zero:

(1.a) the streamwise component of velocity, u;

- (1.b) the static temperature of the gas stream, t;
- (1.c) the partial pressure of the vapor, p.;
- (1.d) the interfacial shear stress, τ_s ;
- (1.e) the rate at which heat is transferred across the S-surface by conduction, q_S ; and
- (1.f) the rate at which mass is transferred across the S-surface by convection $(\rho v)_{c}$.

(2) The gaseous flow regime is subdivided into an "effective" laminar sublayer, bounded by y = 0 and $y = \delta$, and a turbulent core, bounded by $y = \delta$ and $y = \delta$.

Assumption (1) is equivalent to either the assumption of fullydeveloped-pipe flow or the assumption of Couette flow.



FIG. 3-1. A SCHEMATIC REPRESENTATION OF THE PHYSICAL MODEL EMPLOYED IN THE SUBJECT ANALYSES (3) The wetted surface is adiabatic.

(4) The liquid-film coolant is non-reactive, and the interfacial surface is stable.

(5) The flowing gas is incompressible, and its properties may be evaluated at a suitable reference temperature.

(6) The effects of mass transfer upon heat transfer and momentum transfer are important only in the laminar sublayer (except in Knuth's analysis where said effects are also considered in the turbulent core).
(7) The turbulent Prandtl number and Schmidt number, defined by

$$Pr_{t} \equiv \frac{\mu_{t}}{k_{t}} \frac{c_{p}}{c_{t}}$$
 and $Sc_{t} \equiv \frac{\mu_{t}}{\rho D_{t}}$, respectively, are unity.

(8) The surface velocity of the liquid film is negligible compared to the main-stream velocity u_{a} .

(9) Across the "thim" laminar sublayer it can be assumed that

$$(9.a) (\rho v) = \frac{-DR\rho}{R_v(p-p_v)} - \frac{dp_v}{dy} = \text{constant} = (\rho v)_S \qquad (3-3)$$

$$(9 b) \quad \mu \frac{du}{dv} - (\rho v)_{S} u = \text{constant} = \tau_{S}$$
(3-4)

(9.c)
$$k \frac{dt}{dy} - (\rho v)_{S} c_{p_{v}} t = \text{constant} = q_{S} - (\rho v)_{S} c_{p_{v}} t_{S} (3-5)^{1}$$

¹ It is argued in Ref. (26), pp. 10-12, that the specific heat for the vapor in Eq. 3-5 should be replaced by the specific heat for the gas-vapor mixture. However, that argument is incorrect due to the fact that the diffusive mode of mass transport was not considered (see the developments leading to Eqs. 2-18 and 2-30).

The Resistance of the Laminar Sublayer to Momentum Transfer. It follows that by integrating Eq. 3-4 and substituting the limits u = 0at y = 0 and $u = u_{\delta}$ at $y = \delta$, one obtains an equation for the resistance of the laminar sublayer to momentum transfer. In dimensionless form, that equation is

$$\frac{u_{\delta}}{u_{e}} = \frac{\tau_{s}}{\rho_{e}u_{e}^{2}} \frac{(\rho u)_{e}}{(\rho v)_{s}} \quad \{\exp\left(\frac{(\rho v)_{s}\delta}{\mu}\right) - 1\}$$
(3-6)

<u>The Resistance of the Laminar Sublayer to Heat Transfer</u>. It follows that by integrating Eq. 3-5 and introducing the limits t = t, at y = 0 and $t = t_{\delta}$ at $y = \delta$, one obtains an equation for the resistance of the laminar sublayer to heat transfer. In non-dimensional form that equation is

$$\frac{t_{\delta} - t_{S}}{t_{e} - t_{S}} = \frac{1}{\beta} \left(\exp \left(\Pr \frac{c_{p_{v}}}{c_{p}} - \frac{(\rho v)_{S}}{\mu} - \delta \right) - 1 \right)$$
(3-7)

The Resistance of the Laminar Sublayer to Mass Transfer. Of the subject analyses, only that analysis due to Knuth includes a specification of the vapor-partial-pressure field (or its equivalent, the vapor-concentration field). In general, the specification of that field may be desirable for one or more of the following reasons: (a) the evaluation of the physical and transport properties; (b) the evaluation of the energy released by chemical reactions in the case where the liquid-film coolant is reactive; and (c) the determination of the

temperature at the gas-liquid interface--which is uniquely related to the partial pressure of the vapor at the interface. If that temperature must be known with a high degree of accuracy, the approximation¹ introduced by Graham, Sellers, and Emmons is inappropriate. Knuth considered the problem of specifying the partial-pressure field for the latter reason.

For completeness, the specification of the partial pressure distribution is considered herein. Integrating Eq. 3-3 and imposing the limits $p_v = p_{v,s}$ at y = 0 and $p_v = p_{v,\delta}$ at $y = \delta$, the equation is obtained for the resistance of the laminar sublayer to mass transport, which in non-dimensional form is given by

$$\frac{p - p_{v,\delta}}{p - p_{v,s}} = \exp\left(\operatorname{Sc}\frac{R_v}{R}\frac{(\rho v)_S}{\mu}\delta\right)$$
(3-8)

the Relative Resistance of the Turbulent Core to Mass, Momentum, and Heat Transfer (the Reynolds Analogy). A logical extension of the Reynolds analogy (18)(27) to heat, mass, and momentum transfer in the turbulent core of the gaseous-flow regime is

$$\frac{(\text{mass flux})_{y = y}}{(\text{mass flux})_{y = \delta}} = \frac{(\text{momentum flux})_{y = y}}{(\text{momentum flux})_{y = \delta}} = \frac{(\text{heat flux})_{y = y}}{(\text{heat flux})_{y = \delta}}$$

The above relationship can be written more explicitly in two ways: (a) based on the diffusive rates of transport only, and (b) based on the total rates of transport (diffusion plus convection). Rannie, Sellers, and Graham employed method (a); Knuth employed method (b).

The approximation being that the temperature of the liquid film equals the saturation temperature of the film coolant which corresponds to the static pressure p.

The Reynolds analogy which considers only the diffusive mode of transport can be written as follows. Thus

$$\frac{-D_{t} \frac{\rho}{R_{v}} \frac{R}{(p-p_{v})} \frac{dp_{v}}{dy}}{(\rho v)_{\delta}} = \frac{\mu_{t} \frac{du}{dy}}{\tau_{\delta}} = \frac{k_{t} \frac{dt}{dy}}{q_{\delta}}$$
(3-9)

where

 q_{δ} = the rate at which energy is transferred across the surface $y = \delta$ by conduction. $(\rho v)_{\varepsilon}$ = the rate at which mass is transferred across the surface $y = \delta$ by convection. τ_{δ} = the shear stress at $y = \delta$.

Integrating Eq 3-9, substituting the appropriate limits at $y = \delta$ and at $y = \Delta$, and introducing Assumption (7), yields

$$\frac{R}{R_{v}} \ln \left(\frac{p - p_{v,e}}{p - p_{v,\delta}}\right) = \frac{(\rho v)_{\delta}}{\tau_{\delta}} \left(u_{e} - u_{\delta}\right) = \frac{(\rho v)_{\delta}}{q_{\delta}} \left(t_{e} - t_{\delta}\right) c_{p} \quad (3-10)$$
(a)
(b)
(c)

The Reynolds analogy that considers both convective and diffusive rates of transport can be written in the following form (8):

$$\frac{-D_{t}\rho R}{R_{v}(p-p_{v})}\frac{dp_{v}}{dy} = \frac{u_{t}\frac{du}{dy} + \frac{D_{t}R\rho}{R_{v}(p-p_{v})}\frac{dp_{v}}{dy}u}{r_{\delta} - (\rho v)_{\delta}u_{\delta}} = \frac{(3-11)}{r_{\delta}}$$

$$\frac{k_{t}\frac{dt}{dy} + \frac{D_{t}\rho R}{R_{v}(p-p_{v})}\frac{dp_{v}}{dy}c_{p_{v}}t}{r_{\delta} - (\rho v)_{\delta}c_{p_{v}}t_{\delta}}$$

Integrating Eq. 3-11, substituting the appropriate limits at
$$y = \delta$$

and at $y = \Delta$, and employing Assumption (7), one obtains (8)

$$\frac{R}{R_{v}} \ln \left(\frac{p - p_{v,e}}{p - p_{v,\delta}}\right) = \ln \left\{1 + \frac{(\rho v)_{\delta}}{\tau_{\delta}} \left(u_{e} - u_{\delta}\right)\right\}$$

 $= \frac{c_{p}}{c_{p_{v}}} \ln \left(1 + \frac{(\rho v)_{\delta}}{q_{\delta}} (t_{e} - t_{\delta}) c_{p_{v}}\right)$

Equations 3-10 and 3-12 are the two ways in which the Reynolds analogy can be employed to describe the relative resistance of the turbulent core to the transfer of heat, mass, and momentum. It should be noted that Assumption (9) (see page 47), which was employed to derive the equations for the laminar sublayer, is a special case of the Reynolds analogy that considers both diffusive and convective transport (Eq. 3-11). To consider two of the analyses due to Sellers, the Reynolds analogy that considers only diffusive heat and momentum transport must also be written for the laminar sublayer; i.e.,

$$\frac{k \frac{dL}{dy}}{\frac{du}{dy}} = \frac{c_p}{Pr} \frac{dt}{du} = \frac{a \text{ constant across the}}{laminar \text{ sublayer}} = \frac{q_s}{\tau_s} = \frac{q_\delta}{\tau_\delta}$$
(3-13)

Integrating Eq. 3-13 across the sublayer and substituting the appropriate limits, one obtains¹

<u>ь</u>,

(3-12)

¹ Sellers states that Eq. 3-14 implies that the velocity and temperature profiles in the laminar sublayer are assumed to be linear (11, p. 20); however, since Eq. 3-13 needs not be so restricted, it would appear that qualifying statement by Sellers is unnecessary.

$$\frac{t_{\delta} - t_{s}}{u_{\delta}} = \frac{Pr}{c_{p}} \frac{q_{s}}{s} = \frac{Pr}{c_{p}} \frac{q_{\delta}}{s}$$
(3-14)

(a) (b) (c)

<u>Comparing the Analyses</u>. Equations 3-3 through 3-8 plus Eqs. 3-10, 3-12, and 3-14 represent the relationships from which the analyses due to Rannie, Sellers, Graham, and Knuth have been developed Table 1, p. 42, indicates the specific equations that are employed in each of the subject analyses (column I), together with the unknowns that are eliminated between those equations (column II) and the resultant analytical expression for the parameter 6 (column III) Also presented in that table are the empirical formulae and/or experimental data that are employed in each of the subject analyses (column IV), and the interfacial heat balance that is employed (column V)

3 1 2 The Analysis Due to Emmons¹

Physical and Analytical Models. The physical and analytical models presented in Section 3.1.1 are applicable to the analysic due to Emmons, with the exception of Assumptions (2) and (9). Assumption (2) chould be rewritten as follows: The gaseous-flow regime is subdivided into a turbulent core and a viscous sublayer region wherein the following eddy-viscosity relationship proposed by Turcotte (25) applies:

$$\frac{y_{t}}{u} = \sinh^{2}\left(\frac{by^{+}}{13.89}\right) \quad (y^{+} \circ \delta^{+}; \ \delta^{+} = 25.4) \tag{3-15}$$

¹ To improve the clarity, completeness, and correctness of the original analysis due to Emmons, the developments and the final solution presented herein differ from that originally presented by Emmons; however, the analytical and physical models are the same.

The parameter b in Eq. 3-15 must be determined experimentally, 1

Assumption (9) should be modified slightly in accordance with the new definition of the sublayer region; i.e., the molecular thermal conductivity k and dynamic viscosity μ should be replaced by their "effective" values, denoted by $k + k_t$ and $\mu + \mu_t$, respectively.

An additional significant assumption employed by Emmons is that the molecular Prandtl number, defined by $Pr = \frac{\mu C_p}{k}$, is equal to unity.

 τ_{s} Related to q_{s} (The Reynolds Analogy). To relate the interfacial shear stress τ_{s} to the heat flux q_{s} , Emmons assumed that the basic Reynolds analogy, written as follows, is valid for the entire gaseous-flow regime:

$$\frac{k + k_t}{\mu + \mu_t} \frac{dt}{dy} = \frac{\frac{\mu}{Pr} + \frac{\mu_t}{Pr}}{\mu + \mu_t} c_p \frac{dt}{du} = c_p \frac{dt}{du} = \frac{q_s}{\tau_s}$$
(3-16)

Integrating Eq. 3-16 and substituting the appropriate limits at y = 0and at $y = \Delta$, one obtains

$$c_{p} \frac{t_{e} - t_{s}}{u_{e}} = \frac{q_{s}'}{\tau_{s}}$$
 (3-17)

Evaluating s. To evaluate the interfacial shear stress τ_s , Emmons employs a relationship that was derived by Turcotte (25, Eq. 15) for the case of the transpired turbulent boundary layer on a flat plate.

¹ The parameter b was introduced by Turcotte into the relationship originally proposed by Rannie (28) for the case of zero mass transfer; the parameter b accounts for the fact that it is not completely clear as to whether τ_s is the correct shear stress to employ in the calculation of y.

Thus¹

$$\tau_{s} = \tau_{s,0} \exp \left[-\frac{v_{s} 13 89}{(\tau_{s}/\rho)^{2} b} \right]$$
 (3-18)

where

 $\tau_{s,0}$ = the shear stress at the wall in the absence of mass transfe

As indicated in Table 1, Eqs. 3-17 and 3-18 constitute an analytical solution for the parameter β . Since Emmons misquotes and misinterprets² Fq. 3-18, it is of value to outline Turcotte's derivation of that equation. By combining Eqs. 3-4³ and 3-15, Turcotte obtains the following equation for the shear stress distribution in the sublayer region. Thus

$$\tau \equiv u \{1 + \sinh^2(\frac{y^+ b}{13,89})\} \frac{du}{dy} = \tau_s + (\rho v)_s u \qquad (3-19)$$

Integrating Eq. 3-19 and substituting the limit u = 0 at y = 0, one obtains the equation due to Turcotte for the velocity profile in the viscous sublayer. Thus

$$u = \frac{\tau_{s}}{(\rho v)_{s}} \begin{bmatrix} -\frac{v_{s}}{(\tau_{s}/\rho)^{\frac{1}{2}}} & \frac{13.89}{b} \tanh(\frac{y^{+}b}{13.89}) - 1 \end{bmatrix}$$
(3-20)

Combining Eqs. 3-19 and 3-20, one obtains the following equation for the shear stress profile in the viscous sublayer:

¹ Emmons writes Eq. 3.18 incorrectly. He replaces the quantity
$$(\tau_s/\rho)^{\frac{1}{2}}$$
 by $(\tau_s, \rho/\rho)^{\frac{1}{2}}$ (see Ref. (6), Eq. B-11)
² See Ref. (6), pp. 84, 85, 91
³ With μ replaced by $\mu + \mu_t$

$$s exp = \frac{5}{13} = \frac{13}{89} tanh = \frac{y'}{13} \frac{b}{89}$$
 (3-21)

Turcotte then introduces the <u>assumption</u> that the turbulent core external to the viscous sublayer region is unaffected by the mass transfer at the wall Since, in the absence of mass transfer, the wall region defined by $y^+ = 1000$ is a constant-stress layer (18), it follows that

and that

limit
$$tanh \left(\frac{y^{+}}{13} \frac{b}{89}\right) = 1$$
 (3-23)
 $y^{+} + \frac{z^{+}}{5} = 25.4$

Equations 3-22 and 3-23 thus reduce Eq. 3-21 to Eq. 3-18.

Comparison of Emmons' Analysis with the Analyses Due to Sellers, Graham, Rannie, and Knuth Table 1, p 42, compares the analysis due to Emmons with those analyses discussed in Section 3 1.1 Column III of that table presents two analytical expressions for the paramber B under the analysis due to Emmons: (a) the expression due to Emmons (6, Eq B-19); and (b) the corrected expression derived by the writer

3.2 A Review of the Pertinent Experimental Data

The analyses reviewed in Section 3-1 are based on the premise that when the liquid film is stable, the analytical and physical mode's fur liquid-film cooling and transpiration cooling may be considered

I

substantially identical. That is, the surface of the liquid film may be treated as a zero-velocity, aerodynamically smooth porous surface through which the vapor is "injected" into the adjacent boundary layer. Moreover, the concept of the viscous sublayer is introduced, that concept having been employed with success in the analysis of transpiration cooling. A review of some pertinent experimental data is presented here to demonstrate that there is a serious question regarding the validity of that premise. Discussed in Section 3.2.1 are heatmass transfer data that indicate both the limitations of those theories reviewed in Section 3.1 and the complex nature of the phenomena that characterize the evaporation process. Data that present an explanation for anomalies that are noted in the heat-mass transfer data are discussed in Section 3.2.2.

3.2.1 Heat-Mass Transfer Data

The following items are discussed in the present section: (a) the agreement between the heat-mass transfer rates predicted by the theory due to Knuth (8) and the data reported by Refs. (5) and (8); (b) the physical significance of the correlation that was obtained by Emmons (6) for his experimental data; and

(c) the data obtained by Luikov (17) in an investigation of the basic phenomena that characterize the evaporation of a liquid in the presence of a "hot" gas stream.

A Comparison of Knuth's Theory with Experimental Data. Knuth observed that the average¹ heat-mass transfer rates predicted by his ¹ Averaged over the liquid-film-cooled length.

theory were significantly smaller than those he obtained experimentally, and those Kinney, et al. (5) reported; the difference was attributed by Knuth to "entrance length" effects.¹ These were not accounted for in any of the other analyses reviewed in Section 3.1. Knuth employed an empirical correction to bring the theoretical and experimental heatmass transfer rates into agreement; that empirical correction consisted of multiplying the theoretical heat and mass transfer rates by the following expression. Thus

- $(1 + \frac{1}{3}\frac{d}{L}R_{d}^{1})$
- R_d = the Reynolds number for the gas stream based on the diameter of the tube and the bulk flow parameters.

d = the diameter of the tube.

L = the liquid-film-cooled length.

Typical values of R_d and d/L for the data analyzed by Knuth were 10^6 and 10^{-1} , respectively. Hence, the empirical correction employed by Knuth was of the order of

 $1 + 1/3 \quad 10^{-1} (10^6)^{\frac{1}{4}} = 2.05$

The average heat-mass transfer rates predicted by Knuth's theory were, therefore, typically one half of those observed experimentally.

In the experimental investigations conducted by Knuth and Kinney, et al., the velocity profile at the plane of liquid injection was essentially fully developed, whereas the temperature and concentration profiles were flat.

By way of comparison, for the case where the mass transfer is zero and the thermal boundary conditions are such that the velocity profile at the entrance to the test section is fully developed while the temperature profile is flat, the empirical correction for the "entrance length" effects is given by the following expression (8):

$$1 + 0.1 \frac{d}{L} R_d^{\frac{1}{4}}$$

where L is the length of the test section over which heat transfer occurs. Thus, if $R_d = 10^6$ and $d/L = 10^{-1}$, the ratio of the average rate of heat transfer over the test section length L to the heat transfer rate that would obtain if the temperature profile also was fully developed is

 $1 + 10^{-1} (10^6)^{\frac{1}{2}} = 1.32$

It would appear, therefore, that the presence of the evaporating liquid film in the experimental investigations conducted by Knuth and Kinney, et al. resulted in a substantial increase in the average heat transfer rate over that which would obtain if the mass transfer was zero, that ratio being typically 2.05/1.32 = 1.55. Basic mass transfer theory, however, would predict that the "injection" of the vapor into the gas stream would act to "block" the heat transfer to the liquid film, thus reducing the observed rates of heat transfer to the film over those that would result for the case of zero mass transfer; the data for transpiration cooling presented in Fig. 3-3 in Section 3.3.2 illustrates that point. It would appear, therefore, that a significant anamolie exists in the data reported for liquid-film cooling by Refs. (5) and (8) The Physical Significance of the Correlation Due to Emmons (6). As noted in Table 1, the value of the parameter b with which Emmons was able to correlate his experimental data was 0.110 However, on a physical basis, Turcotte (25) argued that b must be bounded as follows:

 $1 < b < (\tau_{s,0}/\tau_{s})^{\frac{1}{2}}$

where, for transpiration cooling,

 $\left(\frac{1}{5},0/\frac{1}{5}\right)^{\frac{1}{2}} \rightarrow 1$

}

Turcotte was able to obtain satisfuctory correlation of experimental data for the transpired turbulent boundary layer by employing the median value of b; thus

 $b = \frac{1}{2} - (1 + (-s, 0) + s)^{\frac{1}{2}}$

Obviously, the value of b employed by Emmons for correlating his experimental data disagrees significantly with the values of b that are physically justified. Some of the discrepancy may be due to the incorrect quotation of Eq. 3-18 by Emmons, as discussed in Section 3.1.2. It would appear, however, that a portion of the discrepancy is attributable to an incorrect assumption in his analytical model. The discussion presented in the following section, Section 3.2.2, indicates that the eddy-viscosity law given by Eq. 3-15, which forms the basis for Emmons' correlation, does not generally apply to the gaseous-flow regime adjacent to the surface of the liquid film. <u>The Evaporation Data Reported by Luikov (17)</u> The following points concerning the experimental apparatus and the experimental procedure employed in the subject investigation are worthy of note:

(a) A dry body which was internally cooled and a shallow pan--having similar overall dimensions as the dry body--containing a liquid were placed side by side on the floor of a large wind tunnel.

(b) The liquid in the evaporating pan was continuously replenished so that the free surface of the liquid was always flush with the top of the pan.

(c) Both the free surface of the liquid in the evaporating pan and the surface of the internally cooled dry body were fully instrumented with thermocouples; in all of the experiments, the rate of coolant flow through the dry body was adjusted so that the temperature of the free liquid surface and the surface of the dry body were the same
(d) Four liquids were employed in the investigation: benzol, acetone, butanol, and water.

(e) The static pressure in the wind tunnel was essentially ambient and the temperature of the main stream varied from 40 C to 120 C, and the velocities from 5 meters/sec to 14 meters/sec.

The experimental results were as follows: in all cases, the observed average heat transfer coefficient for the free liquid surface was larger than the average heat transfer coefficient for the dry body. The ratio of the two heat transfer coefficients varied from 1.2 to 1.6--the larger values corresponding to the larger evaporation rates. As indicated in the foregoing paragraphs, that trend in the experimental

data is opposed to the trend which would be predicted by mass transfer theory. Moreover, the similarity in the thermal boundary condition that was established in the subject investigation for both the dry surface and the surface of the evaporating liquid discredits to a large degree the proposal by Knuth that the disagreement between his theory and experimental liquid-film cooling data is attributable solely to "entrance length" effects.

3.2.2 Correlation of Velocity Profile ta

In the present section the data reported by Refs (19) and (21) are noted; they relate to the phenomenon of gas-liquid shear interactions in the absence of evaporative mass transfer The experimental apparatus employed in each of the subject investigations were significantly different; for a complete description of either experimental apparatus the reader is referred to the references. The experimental results, however, were in qualitative agreement and may be summarized as follows When traverses were made with a pitot probe across the gas-flow regime adjacent to a "pebbled" gas-liquid interface, ¹ the resultant velocity profile data correlated in the same manner as velocity profile data for rough-pipe flow. It was observed that the "effective roughness" of the liquid film surface increased with increasing values of the dimensionless liquid flow rate W^{+} (defined by Eq. 2-10)

Because the viscous sublayer affords the bulk of the resistance of the boundary layer to the transport of heat from a hot gas stream to a

¹ Refer to Fig 2-9, presented in Section 2-4, for a photographic definition of the "pebbled" interface structure.

solid boundary (18), and since the sublayer region becomes less influential and less well defined as the surface roughness increases (4) (18), it would appear that the interfacial phenomenon discussed above may account to a large degree for the anomalies in the heatmass transfer data discussed in the previous section.

3.3 An Alternate Correlation for Heat-Mass Transfer Data

Consider the following recapitulation of the experimental information presented in Sections 3.2 and 2.4 For the case where the liquid film is stable, the effectiveness of liquid-film cooling apparently decreases with increasing values of the dimensionless liquid flow rate W^+ , due to a corresponding increase in the "effective roughness" of the gas-liquid interface At a critical value of W^+ the inception of film instability occurs, and a further increase in W^+ causes the phenomenon of coolant entrainment which results in an even more pronounced reduction in the effectiveness of liquid-film cooling Because of these points, and since the heat transfer phenomena characterizing liquid-film cooling and transpiration cooling would appear to become progressively more similar as W^+ is decreased, the following correlation is suggested for experimental heat-mass transfer data that are obtained for liquidfilm cooling linus.

$$\begin{pmatrix} St \\ St_0 \end{pmatrix} + \begin{pmatrix} W^+ \end{pmatrix} \begin{pmatrix} St \\ St_0 \end{pmatrix}$$
 (3-24)

where

limit
$$f(W^+) = 1$$

 $W^+ + o$
The notation and subscripts in Eq. 3-24 denote the following:

St = the Stanton number for heat transfer

 $f(W^{+})$ a function that must be determined experimentally.

lf - evaluated for the case of liquid film cooling

t - evaluated for the case of transpiration cooling

o - evaluated for the zero-mass transfer case

The Stanton number for heat transfer is defined by the following expression (29). Thus

$$st = \frac{q_{S}}{(\rho u)_{e} c_{p_{e}} (t_{r} - t_{s})}$$
(3-25)

where

$$t_{r} = t_{e} + r \frac{u_{e}^{2}}{2c_{p_{e}}}$$
 (3-26)

= the recovery temperature for the main stream.

r = the recovery factor.

2

As shown by Spalding (29), the recovery factor r can be determined in most instances with sufficient accuracy from the following empirical equation due to Squire. Thus

$$(3-27)$$

The subscript notation (), introduced in Eq. 3-27 means that the subscripted quantity is to be evaluated at an appropriate reference state. The reference state suggested herein is that defined by the reference

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temperature t_{*} due to Eckert (30), and the reference concentration $C_{v,*}$ due to Knuth (31). Thus

$$t_{\star} = \frac{1}{2} (t_e + t_s) + r \frac{u_e^2}{2 c_{p_e}}$$
 (3-28)

$$C_{v,*} = \frac{1}{2} (C_{v,e} + C_{v,s})$$
 (3-29)¹

The function $f(W^+)$ in Eq. 3-24 must be determined experimentally. Procedures are outlined in Section 3.3.4 for evaluating $f(W^+)$ given either (a) bulk heat transfer data (i.e., m'_0 as a function of L for various main-stream conditions and liquid-film coolants), or, preferably, (b) the streamwise distribution of $(St)_{1f}$ as determined from detailed measurements of the structure of the boundary-layer region adjacent to the surface of the liquid film.

to facilitate implementing the correlation procedure suggested by Eq 3-24, Section 3.3.1 presents a method for obtaining an approximate evaluation of the temperature at the gas-liquid interface, and Sections 3.3.2 and 3.3.3 suggest methods for evaluating $(St/St_0)_t$ and St_0 , respectively.

It should be noted that the correlation procedure developed herein differs from the semi-empirical correlations reviewed in Section 3.1 in that it is applicable to flows in the presence of pressure gradients, to the flow of a compressible fluid, and to either stable or unstable

¹ In the developments which follow, it is assumed that the concentration of the vapor in the main stream is zero. Hence, $C_{v,*} = \frac{1}{2} (C_{v,s})$.

liquid films Moreover, "entrance length" effects and streamwise variations in q_S and $(pv)_S$ are considered. The developments are limited, however, to the case where the gas flow is turbulent and the liquid-film coolant is non-reactive.

3.3.1 A Method for Predicting the Temperature at the Gas-Liquid Interface

The temperature at the gas-liquid interface, t_s, can be related to the bulk flow parameters by specifying (a) an interfacial energy balance, and (b) an analogy between the rates of heat and mass transfer. In the derivations that follow, the generalized energy balance given by the following equation is employed for convenience:

$$q_{S} = \phi \left(c v \right)_{S} \tag{3-30}$$

The parameter Φ is defined by Eqs. 3-30 and 2-31. Thus

$$\Rightarrow \exists \exists \exists H_v + (q_L + q_{r,L} - q_{r,S})/(\rho v)_S$$

The analogy between heat and mass transfer employed herein is that given by

$$St = St' \left(\frac{Pr}{Sc}\right)_{\star}^{-2/3}$$
 (3-31)

where St' is the Stanton number for mass transfer defined by

St'
$$\frac{(cv)}{(p_u)_e} \sum_{q}^{p-p_v} \frac{R_v}{R_g}$$
 (3-32)¹

Equation 2-32 also can be written in terms of the concentration of the vapor as follows (29). Thus St' = $\frac{(\rho v)_S}{(\rho u)_p} = \frac{1-C_{v,s}}{C_{v,s}}$

Experimental verification of the validity of Eq. 3-31 for incompressible flows is presented in Ref (32), and for compressible flows in Ref (29). It should be noted, however, that at and close to the plane where the liquid coolant is injected. Eq. 3-31 is strictly applicable only if the growths of both the thermal boundary layer and the partial-pressure (or vapor-concentration) boundary layer start at that plane.

Combining Eqs 3-30, 3-31, and 3-32 yields

$$\frac{\phi}{c_{p_{e}}(t_{r}^{-}-t_{s}^{-})} = \frac{p - p_{v,s}}{p_{v,s}} \frac{R_{v}}{R_{g}} \left(\frac{p_{r}}{Sc}\right)_{\star}^{-2/3}$$
(3-33)

In Eq. 3-33 there are three unknowns: ϕ , $p_{v,s}$ and t_s . Tabulated data presenting the saturation pressure of the liquid coolant as a function of its saturation temperature represents a second independent relationship for $p_{v,s}$ and t_s . Thus, by specifying ϕ , t_s (or its equivalent, $p_{v,s}$) can be related to the bulk flow parameters

Figure 3-2 presents, for an air/water system, calculated values of the interface temperature t_s as a function of the static pressure p and the main-stream temperature t_e^{-1} . The flow conditions assumed in the calculations are indicated in the figure. The saturation curve for water presented in the figure illustrates the approximate error introduced by the assumption that t_s is equal to the saturation temperature for the liquid coolant und r the prevailing static pressure, p, that assumption having been employed by Refs (6)(10)(11)(22)(24).

⁴ The calculations were performed by employing Eq. 3-33 in conjunction with the tabulated data of Refs (3?) and (33) A sample calculation is presented in Appendix B



FIG. 3-2. THE TEMPERATURE AT THE INTERFACE AS A FUNCTION OF THE STATIC PRESSURE AND THE MAIN-STREAM TEMPERATURE

The calculated results presented in Fig. 3.2 agree within \pm 10F with the data reported in Ref. (8) and that obtained in the preliminary investigations of liquid-film cooling conducted at the Jet Propulsion Center, Purdue University (12); the flow conditions presented in the figure apply closely to those experimental investigations. The values of p and t_e employed in the subject investigations were as follows:

Ref. (8): $t_e = 1000-230CR; p = 1 atm$ Ref (12): $t_e = 1000k; p = 220 psia$

It was stated in Section 2.2.2 that there is a question (especially for the case where a streamwise pressure gradient is imposed on the flow) regarding the fraction of the energy incident to the surface of the liquid film that is utilized for increasing the sensible enthalpy of the liquid coolant. The method presented herein for evaluating t_s affords a means by which that question can be investigated experimentally. The experimental measurement of the streamwise distribution of t_s can be related to a streamwise variation in the parameter ϕ by means of Eq. 3-33. Moreover, if the radiant and wall heat fluxes are reduced in importance, by proper design of the experimental apparatus and selection of the experimental parameters, then the definition of ϕ reduces to

 $\phi = \Delta H_V + (\text{the increase in the sensible enthalpy for the liquid coolant})$

Thus, for that case, the fraction of the incident heat flux utilized for increasing the sensible enthalpy of the liquid can be determined as a function of x

3.3.2 Correlation of $(St/St_0)_t$ for Transpiration Cooling

Analytically and/or experimentally it has been established satisfactorily that if transpiration cooling data are presented in the general form

 $\left(\frac{St}{St}\right)_{c}$ as a function of $\frac{(\rho v)_{S}}{(\rho u)_{e}} = \frac{1}{St_{o}}$

the resulting correlation is essentially independent of the Reynolds number for the main stream, based on the streamwise coordinate x, the flach number for the main stream, and the ratio of the main-stream temperature to the temperature at the wall (34)(35)(36). Furthermore, the correlation is independent of the streamwise pressure gradient imposed on the flow (36). Because of the significance of those facts, such a correlation for transpiration cooling data is employed in the present developments.

Figure 3-3 presents the correlation for transpiration cooling data due to Brunner (37). The curve presented in the figure is given by the following equation. Thus

$$\left(\frac{St}{St_{o}}\right)_{t} = \left[1 + \frac{I}{St_{o}} - \frac{(\rho v)_{S}}{(\rho u)_{e}}\right]^{-3}$$
 (3-34)

where

1

Marine and

 $\frac{2}{9} \left(\frac{p_v}{c_p}\right)$ (3-35)



FIG. 3-3, THE CORRELATION FOR TRANSPIRATION COOLING DATA DUE TO BRUNNER (37)

The data surveyed by Brunner were for the case where either nitrogen, or helium, or water vapor is injected into an uir stream. As a consequence, there is no assurance that the empirical correlation given by Eq. 3-34 is applicable to all flowing gases and injected fluids that might characterize a transpiration-cooled system. The fluids N_2 , H_e , and H_2O do, however, represent a significantly wide cross section of physical property values. It appears reasonable, therefore, that Eq. 3-34 can be extended with some degree of confidence to other fluids

3.3 3 A Method for Evaluating Sto

Ideally, in an investigation of liquid-film cooling, the streamwise distribution of the Stanton number for heat transfer in the absence of mass transer, denoted by St_0 , would be determined experimentally. There are, however, many instances where experimental determination of St_0 is impractical. In those cases, the method suggested herein can be employed for evaluating St_0 analytically.

The method suggested has been developed and/or employed by numerous investigators (36)(38)(39)(40)(41). It is based on the assumption that the empirical heat-transfer law for turbulent flow over a flat plate, when written in the following form, is also applicable to turbulent flows in the presence of streamwise pressure gradients. Thus

$$St_o = St_o (R_{\xi})$$

whe re

 $R_{\xi} = \frac{\rho_{e} - u_{e} - \xi}{\mu_{e}} = \text{the Reynolds number for the main stream based on}$ the characteristic length ξ . $\frac{t}{\omega} = \frac{t}{\omega} \frac{(z_e^0 - z_e^0)}{(z_e^0 - z_w^0)}$ $= \frac{t}{\omega} \frac{(z_e^0 - z_w^0)}{(z_e^0 - z_w^0)}$

$$t^0 = t + \frac{u^2}{2c_p}$$
 = the stagnation temperature.

The heat-transfer law suggested in Ref. (36), and employed in the following developments, is that given by

$$St_0 = 0.0128 (Pr)^{-3/4} R_{\xi}^{-1/4}$$
 (3-37)

Equation 3-37 and the developments that follow can be employed to analyze a compressible gas flow by introducing the concept of the reference state, and making the following transformations for the Stanton number St_o and the Reynolds number R_F (42). Thus

$$St_0 \rightarrow St_{0,*} \equiv \frac{\rho_e}{\rho_*} St_0$$
 (3-38)

$$R_{\xi} \rightarrow R_{\xi,s} = \frac{\mu_{e}}{\mu_{\star}} \frac{\rho_{\star}}{\rho_{e}} R_{\xi}$$
(3-39)

The energy integral equation is given by (38)

$$St_{0} = \frac{d\xi}{dx} + \xi \left[\frac{1}{u_{e}} \frac{du_{e}}{dx} + \frac{1}{(t_{r}-t_{w})} \frac{d(t_{r}-t_{w})}{dx} + \frac{1}{\rho_{e}} \frac{d\rho_{e}}{dx} \right]$$
(3-40)

Equation 3-37 substituted into Eq. 3-40 results in an explicit equation for St_o that can be solved stepwise by numerical integration. To start the numerical calculations, however, the value of x_0 and the value of

 St_0 at x = x_0 must be either known (determined experimentally) or approximated. In many applications of liquid-film cooling that are of interest, the temperature of the liquid film is significantly less than the temperature of the solid surface upstream of the plane at which film coolant is injected. In those cases, it is suggested that St_0 be calculated as if the growth of the thermal boundary layer began at x = x_0. In other words, assume that the boundary condition $St_0 = \infty$ at x = x_0 applies.

In the special case of an incompressible gas flow over an isothermal wall surface, Eq. 3-40 can be simplified to the form

$$St_{o} = \frac{\mu}{(\rho u)_{e}} \frac{dR_{\xi}}{dx}$$
 (3-41)

Substituting Eq 3-37 into Eq 3-41, integrating the resultant expression, and imposing the limit $St_0 = \infty$ at $x = x_0$; one obtains the following, relatively simple expression for St_0 . Thus

$$St_{0} = \frac{0.02923 \text{ Pr}^{-3/5} R_{x}^{-1/5}}{\left[\frac{1}{u_{e}x} + \frac{x}{v_{0}} u_{e}\right]} \frac{1}{u_{e}} \frac{1}{u_{e}} \frac{1}{v_{0}} \frac{1$$

where

 $R_{x} = \frac{e^{u}e^{x}}{e^{u}e^{u}e^{x}} = \text{the main-stream Reynolds number based on the characteristic length x.}$

Equation 3-42 reduces to the following expression for the case where the main-stream velocity u_e is constant in the streamwise direction. Thus

$$St_0 = 0.02923 Pr^{-3/5} R_x^{-1/5} (1 - \frac{x_0}{x})^{-1/5}$$
 (3-43)

The term $(1 - \frac{x_0}{x})^{-1/5}$ in Eq. 3-43 is the correction for "entrance length" effects due to the dynamic entrance length x_0 .

3.3.4 Methodology

The principal equations required for the correlation procedure suggested herein are summarized below. Thus, by definition

$$W^{+} = \frac{m'}{u_1} \tag{2-10}$$

From the conservation of film coolant one obtains

$$m' = m'_{0} - \int_{0}^{x} (\rho v)_{S} dx$$
 (3-44)

Combining Eqs 3-25 and 3-30, one obtains

$$St_{1f} = \frac{(\rho v)_{S}}{(\rho u)_{e}} J \qquad (3-45)$$

where

$$J \equiv \frac{\Phi}{c_{p_e}(t_r - t_s)}$$
(3-46)

Furthermore, combining Eqs 3-24, 3-34, and 3-45, yields

$$\left(\frac{\mathrm{St}}{\mathrm{St}_{0}}\right) = f(W^{+}) \begin{bmatrix} 1 + \frac{\mathrm{I}}{\mathrm{J}} & (\frac{\mathrm{St}}{\mathrm{St}_{0}}) \\ 0 & 1 & \mathrm{If} \end{bmatrix}$$
(3-47)

In the discussion that follows, it is assumed that the streamwise variations for the quantities listed in Table 2 are either determined experimentally, or can be approximated from knowledge of the experimental apparatus and parameters, or can be determined analytically from the procedures presented in Sections 3.3.1 and 3.3.3.

Table 2

The Parameters That Are Considered Known As a Function of x

In Outlining the Methodology

Symbol Definition °p_e the specific heat for the main-stream gas °pv the specific heat for the vapor non-dimensional parameter defined by Eq. 3-35 Ι J non-dimensional parameter defined by Eq. 3-46 the liquid-film-cooled length L the static pressure р Sto the Stanton number for heat transfer in the absence of mass transfer the main-stream temperature te the temperature at the gas-liquid interface ts ^ue the velocity of the main stream the plane at which liquid-film coolant is introduced ×o

A suggested procedure for evaluating the function $f(W^+)$ is outlined below for the following two cases: (a) the case where the streamwise distribution of $(St)_{|f}$ is determined from detailed measurements of the structure of the boundary-layer region above the liquid film; and (b) the case where only bulk heat transfer data are obtained (i.e., m'_o as a function of L for various main-stream conditions and liquid film coolants).

<u>Case (a): (St</u>)_{lf}<u>Determined Experimentally</u> The procedure for this case can be outlined as follows. Thus

(1) Determine the streamwise distribution of $(\rho v)_S$ from Eq. 3-45.

(2) Determine the streamwise distribution of m' and W^+ from Eqs. 3-44 and 2-10, respectively

(3) Determine the streamwise distribution of $f(W^+)$ from Eq. 3-47.

From steps (1), (2), and (3) one obtains, for each set of input data, a plot of $f(W^+)$ as a function of W^+ . Hence, for n sets of input data one obtains n such plots If the basic premise upon which the correlation suggested herein is correct, that is, if $f(W^+)$ is an universal function, each of those plots would be identical If, however, the curves differed significantly, it would indicate that the function $f(W^+)$ is not sufficient to characterize the difference between the heat transfer phenomena for liquid-film cooling and transpiration cooling However, even if that was found to be the case, if $f(W^+)$ differed from unity in any instance, the inadequacy of employing solely transpiration cooling theory in the analyses of liquid-film cooling would have been established

<u>Case (b):</u> m'_{0} and <u>L</u> <u>Determined Experimentally</u> In this case the functional form of $f(W^{+})$ is assumed; i.e., the function $f(W^{+})$ is assumed

to be linear, exponential, etc By studying several sets of input data, it can be determined whether or not the assumed form of $f(W^{\dagger})$ was satisfactory. The complete procedure is outlined below. Thus

(1) Assume the functional form of $f(W^{\dagger})$.

(2) Determine m' and W^{\dagger} at x = x

(3) Calculate $(St)_{1f}$ at $x = x_0$, employing Eq. 3-47.

(4) Calculate $(\rho v)_S$ at $x = x_0$ by means of Eq. 3-45.

(5) Determine m' and W⁺ at x = $x_0 + \Delta x$, employing Eqs. 3-44 and 2-10, respectively.

(6) Repeat steps (2) through (5), replacing $x = x_0$ by $x = x_0 + \Delta x$, and continue the reiteration until $x = x_1 = x_0 + L$.

(7) If $m' \neq 0$ at $x = x_1$, then changes must be made in step (1). To illustrate, if $f(W^+)$ was assumed to be linear in step (1), so that $f(W^+) = 1+CW^+$, then a different value must be assumed for the constant C and the computation repeated until m' = 0 when $x = x_1$.

For a single set of input data, regardless of the assumed functional form for $f(W^+)$, the condition m' = o at $x = x_1$ will be realized. By considering several sets of input data, however, it can be deduced whether that form for $f(W^-)$ satisfies the condition of universality. Note that if one experiences difficulty in obtaining a form for $f(W^+)$ that simultaneously satisfies several sets of input data, it is nearly impossible to determine whether that difficulty is due to a lack of insight into the physics of the problem (i.e., the inability \dagger 'guess" at the correct $f(W^+)$, or to the fact that $f(W^+)$ does not physically exist.

4. AN ANALYSIS FOR THE GAS-VAPOR-COOLED REGION

4 1 Introduction

It is difficult, if not impossible, to make a complete mathematical analysis of the boundary-layer and heat-transfer characteristics for the region immediately downstream from the end of the liquid film. Moreover, an involved analysis generally is not warranted because of the large number of assumptions that must be made concerning the velocity and temperature profiles, the form factor, the skin-friction law, the heat-transfer law, etc. As a consequence of this fact, the analysis presented herein is based on an idealized and relatively simple flow model

Both Emmons (6) and Guinn (7) developed semi-empirical analyses for the gas-vapor-cooled region; those analyses were based on the flow model suggested by Hatch and Papell (43) for gas-film cooling. The analysis presented here differs from those due to Emmons and to Guinn in the following respects: (a) boundary-layer flow rather than fullydeveloped-pipe flow is considered; and (b) the analysis is generalized so that it can be applied to a compressible gas flow in the presence of a streamwise pressure gradient. The analysis is dependent, however, on the analysis for the liquid-film-cooled region presented in Section 3.3.

As discussed in Section 2 1.2, a complete analysis for the gas-vaporcooled region requires the consideration of two cases: (a) the case where the wall is adiabatic; and (b) the case where the wall is externally cooled. Those two cases are considered in Sections 4.2 and 4.3, respectively.

4.2 The Case Where the Wall is Adiabatic

4.2.1 The Flow Model

The flow model employed for the analysis of the gas-vapor-cooled region is defined by the following conditions

(1) The wall is adiabatic.

(2) The boundary layer flow is fully turbulent

(3) The thermal, the diffusional, and the dynamic boundary layer thicknesses are equal; the latter is denoted by $y = \Delta r$.

(4) The stagnation enthalpy for the main stream, denoted by h_e^0 , is a constant in the streamwise direction; the stagnation enthalpy for the gas stream is defined, in general, by

$$n^{O} = c_{p}t + \frac{u^{2}}{2}$$
 (4-1)

(5) The velocity profile and the stagnation-enthalpy prefile for the boundary layer are of the functional forms

$$\frac{u_{e}}{u_{e}} = t_{1} \left(\frac{y}{2}\right)$$
 (4-2)

and

$$\frac{n^{0} - h}{aw} = f_{2} \left(\frac{y}{y}\right)$$

$$\frac{n^{0} - h}{aw} = f_{2} \left(\frac{y}{y}\right)$$

$$(4-3)$$

where h_{aw} is the orthalpy for the fluid adjacent to the adiabatic wall. To illustrate, f_1 could be assumed of the power-law form:

$$f_1 = \left(\frac{y}{\Delta}\right)^{\frac{1}{n}}$$
(4-4)

In addition, f_2 could be assumed of the form

$$f_2 = A + B \left(\frac{y}{\Delta}\right) + C \left(\frac{y}{\Delta}\right)^2 + D \left(\frac{y}{\Delta}\right)^2$$
(4-5)

The constants in Eq. 4-5 can be evaluated from the following boundary conditions that apply to f_2 . Thus

$$f_2(0) = 0$$

 $f_2(1) = 1$
 $\frac{\partial f_2}{\partial y}(0) = 0$
 $\frac{\partial f_2}{\partial y}(1) = 0$

Its is ultant expression for f_2 is

$$(\frac{1}{2})^2 - 2(\frac{y}{2})^3$$
 (4-6)

For the present analysis, however, the actual forms of f_1 and f_2 neer not be assumed, only the assumption that f_1 and f_2 exist for the gas-vapor-cooled region is required.

4 2.2 The Energy Balance

Figure 4-1 presents, schematically, a control volume for the gasvapor-cooled region; the control surface for that control volume is defined by $x = x_1$, x = x, y = 2, and y = 0. The following energy fluxes, written in accordance with the flow model defined in the foregoing section, are presented in the figure:

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(c) $(m'' - m'_1)n_e^0$ = the rate at which thermal and kinetic energy is transferred from the main stream into the control volume per unit spanwise length

The parameter m" in term (c) is defined as follows: m" _____ou dy (4-7)

> = the rate at which mass is transferred through the boundary layer per unit spanwise length at the station x = x.

The energy balance for the control volume in Fig. 4-1 can be written as follows. Thus

$$\int_{au}^{b} dy + (m''-m''_1) h_e^0 = \int_{au}^{bu} dy \qquad (4-8)$$

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MAIN STREAM ---

Employing Eq. 4-7, Eq. 4-8 can be written in the form

$$m_{1}^{u} \xrightarrow{\Delta}_{1} \rho u h^{0} dy + (m^{u} - m_{1}^{u}) h_{e}^{0} = m^{u} \frac{\Delta}{\beta} \rho u h^{0} dy$$

$$\int_{0}^{\Delta} \rho u dy \qquad (4-9)$$

In order to realize the desired simplicity of the present analysis, the following approximation is introduced concerning the integral terms in Eq. 4-9. Thus¹

Thus, Eq. 4-9 can be written in the following form by introducing that approximation and Assumption (5) of the flow model:

$$m_{1}^{"} \stackrel{\stackrel{\Delta_{1}}{\underset{o}{\circ}}{}^{f} f_{1} \{(h_{e}^{0} - h_{aw,1})f_{2} + h_{aw,1}\} dy}{\stackrel{\Delta_{1}}{\underset{o}{\circ}} f_{1} dy} + (m^{"} - m_{1}^{"}) h_{e}^{0} =$$

$$m^{"} \stackrel{\stackrel{\Delta_{1}}{\underset{o}{\circ}}{}^{f} f_{1} dy}{\stackrel{(h_{e}^{0} - h_{aw})f_{2} + h_{aw}\} dy}{\stackrel{\Delta_{1}}{\underset{o}{\circ}} f_{1} dy}$$

The latter expression can be rewritten in the form

¹ Note that this approximation is not equivalent to and is less restrictive than the assumption of constant density.

$$m_{1}^{"}(h_{e}^{0} - h_{aw,1}) - m_{1}^{"}(h_{e}^{0} - h_{aw,1}) \xrightarrow{\int_{0}^{1} f_{1} f_{2} - d(\frac{y}{\Delta})}{\int_{0}^{1} f_{1} - d(\frac{y}{\Delta})} =$$

$$m^{"}(h_{e}^{0} - h_{aw}) - m^{"}(h_{e}^{0} - h_{aw}) \frac{\int_{0}^{1} f_{1} f_{2} - d(\frac{y}{\Delta})}{\int_{0}^{1} f_{1} - d(\frac{y}{\Delta})}$$

$$(4-10)$$

Finally, the terms in Eq 4-10 can be combined to yield the following energy balance for the control volume presented in Fig. 4-1. Thus

$$\frac{h_{e} - h_{aw}}{h_{e}^{0} - h_{aw,1}} = \frac{m_{1}^{"}}{m_{1}^{"}}$$
(4-11)

The relatively simple form of Eq. 4-11 is due principally to Assumption (5) of the flow model.

Equation 4-11 enables the determination of the enthalpy for the fluid adjacent to the adiabatic wall as a function of x once h_e^0 and $h_{aw,1}$ and m'' as functions of x are known. The parameter h_e^0 is normally known from the statement of the problem, and the parameter $h_{aw,1}$ can be determined from the developments presented in Section 3.3.1 The evaluation of m'' as a function of x is considered in Section 4.2.3

To evaluate the temperature distribution for the adiabatic wall from the distribution of h_{aw} , the specific heat for the fluid adjacent to the wall must be known. Excluding the region close to the station $x = x_1$, it is sufficiently accurate to take that specific heat to be that for the main-stream gas because of the low concentration of the vapor in comparison to that of the main-stream gas. Moreover, since the temperature of the adiabatic wall at $x = x_1$ can be independently determined from the developments of Section 3.3.1, it follows that the adiabatic-wall temperature distribution for the entire gas-vapor-cooled region can be determined with sufficient accuracy once h_{aw} as a function of x is known.

4.2.3 The Evaluation of m"

Equation 4-5, which defines the parameter m", can be rewritten in the form

$$m'' = (\rho u)_{e} \left(\frac{\Delta}{\theta} - \frac{\Delta^{\star}}{\theta}\right) \theta \qquad (4-12)$$

where

$$\Delta^* = \int_{0}^{\Delta} (1 - \frac{\rho u}{(\rho u)_e}) dy = \text{the displacement thickness.}$$
(4-13)

$$\theta \equiv \int_{0}^{\infty} \frac{\rho u}{(\rho u)_{e}} (1 - \frac{u}{u_{e}}) dy = \text{the momentum thickness.}$$
(4-14)

Equation 4-12 shows that m" can be determined as a function of x if Δ/θ , Δ^*/θ , and θ are known as functions of x. The evaluation of those quantities is considered in the following paragraphs.

<u>The Evaluation of Δ/θ and Δ^*/θ </u>. Persh and Lee (44) have tabulated the values of Δ/θ and Δ^*/θ as functions of the main-stream Mach number and the wall-to-main-stream temperature ratio. It is suggested, therefore, that these tabulated data be employed for evaluating Δ/θ and Δ^*/θ when the variation in the density for the gas stream across the boundary layer is significant. If, however, that density variation is not significant, it is suggested that the subject parameters be calculated by applying the power-law velocity profile (see Eq. 4-4). The resultant expressions for Δ/θ and Δ^*/θ are (4, p. 536)

$$\frac{\Delta}{\theta} = \frac{(1+n)(2+n)}{n}$$
(4-15)

and

$$\frac{\Delta^{\star}}{\Theta} = \frac{(2+n)}{n} \tag{4-16}$$

<u>The Evaluation of θ </u>. To determine the value of the momentum thickness θ as a function of x, one must solve the momentum integral equation. That equation can be written in the form (45)

$$\frac{c_f}{2} = \frac{d\theta}{dx} + \theta \left[\left(2 + \frac{\Delta^*}{\theta} \right) \frac{1}{u_e} \frac{du_e}{dx} + \frac{1}{\rho_e} \frac{d\rho_e}{dx} \right]$$
(4-17)

Or in the form (46)

$$\frac{c}{2}f = \frac{d}{dx} + \theta \left[\frac{1}{M} - \frac{dM}{dx} - \frac{1}{(1 + (\frac{\lambda - 1}{2}) - M^2)}\right]$$
(4-18)

The notation for Eqs. 4-17 and 4-18 is as follows:

$$\frac{c_f}{2} = \frac{w_e}{v_e}^2 = \text{the skin-friction coefficient.}$$
(4-19)

M = the Mach number for the main-stream flow. $\lambda = \frac{c_p}{c_v}$ = the specific heat ratio. To solve¹ either Eq. 4-17 or Eq. 4-18 for θ , one must specify the form of the skin-friction law; the latter is denoted functionally by

$$\frac{c_{f}}{2} = \frac{c_{f}}{2} (R_{\theta})$$

where

and

$$R_{\theta} \equiv \frac{\rho_{e} - u_{e}^{\theta}}{\mu_{e}} = \text{the Reynolds number for the main stream based}$$
on the characteristic length θ . (4-20)

Furthermore, the boundary condition on θ at the station $x = x_1$, denoted by θ_1 , also must be specified before θ as a function of x can be determined from either Eq. 4-17 or Eq. 4-18.

Several empirical skin-friction laws are presented in Chapter 22 of Ref. (4) for the case where the gas flow is turbulent, incompressible, and over a flat plate. In the analysis of gas flows with pressure gradients (particularly favorable pressure gradients), it is common to assume that the skin-friction law is that established empirically for the flat-plate case (45)(46)(4). Furthermore, the case where the gas flow is compressible can be analyzed by introducing the following transformations into the form of the skin-friction law for incompressible flow (42):

$$\frac{c_f}{2} \div \left(\frac{c_f}{2}\right)_{\star} \equiv \frac{\rho_e}{\rho_{\star}} \frac{c_f}{2}$$
(4-21)

¹ A numerical solution is generally required for either Eq. 4-17 or 4-18.

$$R_{\theta} \rightarrow R_{\theta,\star} \equiv \frac{\rho_{\star}}{\rho_{e}} \frac{\mu_{e}}{\mu_{\star}} R_{\theta}$$
(4-22)

The reference state denoted by (), in Eqs. 4-21 and 4-22 was discussed in Section 3.3. It is suggested that, in evaluating the physical properties, the composition of the gas be assumed uniform throughout the boundary layer and equal to that for the main stream; i.e., assume that the reference temperature given by Eq. 3-28 is sufficient to define the reference state

A sufficiently accurate approximation to θ_1 can be obtained in the following manner First, assume that the Colburn analogy, given by the following equation, applies at $x = x_1$. Thus

$$(St)_{1f} = \frac{c_f}{2} (Pr)_{\star}$$
 (4-23)

Then, employing the skin-friction ¹aw which relates $\frac{c_f}{2}$ and θ , one can approximate the value of θ_1 once $(St)_{1f}$ at $x = x_1$ has been determined from the analysis presented in Section 3 3 for the liquid-film-cooled region

4 2 4 A Special Case

To illustrate the developments presented in the foregoing, consider the special case defined by the following conditions

1 The gas flow is incompressible, constant-density, and over an adiabatic flat plate

2

The skin-friction law is that due to Prandtl (4, Eq. 22.7). Thus

$$\frac{c_f}{2} = 0.0128 R_{\theta}^{-1/4}$$
(4-24)

3. The velocity profile for the boundary layer is given by

$$\frac{u}{u_e} = \left(\frac{y}{\Delta}\right)^{1/7}$$
(4-25)

4. The specific heat for the vapor and for the main-stream gas are equal and are not functions of temperature

For that special case, Eq 4-11 reduces to the form

$$\frac{t_{e} - t_{aw}}{t_{e} - t_{aw}, 1} = \frac{m_{1}^{"}}{m^{"}}$$
(4-26)

Equation 4-12 reduces to (refer to Eqs. 4-15 and 4-16)

$$\mathbf{m}^{\prime\prime} = 9 \rho \mathbf{u}_{\rho} \theta \tag{4-27}$$

Equation 4-17 simplifies to the form

$$\frac{c_{f}}{2} = \frac{d\theta}{dx} = \frac{dR_{\theta}}{dR_{x}}$$
(4-28)

where

 $R_{x} = \frac{\rho u}{\omega} e^{x}$ = the Reynolds number for the main stream based on the characteristic length x (4-29)

Upon combining Eqs 4-26 and 4-27, one obtains

$$\frac{t_e - t_{aw}}{t_e - t_{aw,1}} = \frac{\theta_1}{\theta} = \frac{R_{\theta_1}}{R_{\theta}}$$
(4-30)

In addition, substitution of Eq. 4-24 into Eq. 4-28 yields

$$0 \ 0128 \ dR_{x} = R_{\theta}^{L} \ dR_{\theta}$$
 (4-31)

r

Integrating Eq. 4-31 and substituting the limit $R_{\theta} = R_{\theta}$ at $x = x_1$, one obtains

$$R_{\theta}^{5/4} = 0.016 R_{\chi} (1 - \frac{x_1}{x}) + R_{\theta_1}^{5/4}$$
 (4-32)

It follows from Eq. 4-30 and Eq. 4-32 that for the special case in question, the adiabatic-wall temperature as a function of x is given in non-dimensional form by the following equation. Thus

$$\frac{t_e - t_{aw}}{t_e - t_{aw,1}} = \frac{\frac{R_{\theta_1}}{1}}{\left[0.016 R_x \left(1 - \frac{x_1}{x}\right) + R_{\theta_1}^{5/4}\right]^{4/5}}$$
(4-33)

Finally, to complete the solution, R_{θ_1} can be related to $(St)_{1f,1}$, the Stanton number for heat transfer at $x = x_1$, by combining Eqs. 4-21 and 4-24. The resultant expression is

$$R_{\theta_1} = \begin{vmatrix} -(St)_{1f,1} & 2/3 \\ 0 & 0128 \end{vmatrix}$$
(4-34)

4.3 The Case Where the Wall is Cooled Externally

For the subject case, the basic analytical problem is that of relating the local values for the wall temperature and the wall-heat flux to the main-stream conditions. It can be inferred from several mass-transfer cooling (and heating) studies reported in the literature (Refs. 47-51) that heat-transfer data that are obtained for the gasvapor-cooled region should correlate satisfactorily in the following manner Thus

$$St_{gv} = St_{o} \frac{t_{aw} - t_{w}}{t_{e}^{o} - t_{w}}$$
 (4-35)

where

$$St_{gv} = \frac{q_w}{(ou)_e (t_{aw} - t_w)}$$
(4-36)

= the Stanton number for heat transfer for the gas-vaporcooled region

Equation 4-35, together with the developments presented in Sections 4-2 and 3 3.3, which enable the determination of t_{aw} and St_0^1 as functions of x, respectively, thus represents the solution for the subject case

¹ It is important to note that in the evaluation of St in Eq. 4-35, the boundary condition St $_{0} = \infty$ at x = x₁ applies (47-51).

5 CONCLUSIONS AND RECOMMENDATIONS

Having studied the physical and theoretical nature of liquid-film cooling, together with the pertinent literature, the following general conclusions are drawn:

1 The feasibility of liquid-film cooling a rocket thrust chamber has been established experimentally (5)(6)(8)(10)(11)(52)(53)(54). However, the fundamental phenomena characterizing that process are not well understood. In particular, the interfactal phenomenu (film instability, droplet entrainment, and the "effective" film roughness)

is not well defined, especially for the case where the pressure and/or main-stream temperature is substantially greater than ambient. It is generally agreed, however, that those phenomena significantly influence the effectiveness of liquid-film cooling

2 A need exists for accurate heat-mass transfer data for the liquidfilm-cooled region, especially for the case characterized by one or more of the following conditions: (a) the gas flow is compressible; (b) a pressure gradient is imposed on the flow; and (c) a film coolant other than water is employed

3 The analyses presented to date for liquid-film cooling have either lacked agreement with the experimental data or have resulted in correlations that are not justified physically. Obviously, further experimental evaluation of the basic phenomena would enable the postulation of more

realistic analytical models and the development of more complete analyses

4 Additional experimental investigation of the gas-vapor-cooled region is needed. Boundary layer measurements would aid in defining the mixing characteristics for that region, and would enable the postulation of a more realistic flow model than those postulated previously (6)(7), and that presented in the subject report

In view of the aforementioned conclusions, and in accordance with the foregoing developments presented in the subject report, some recommendations for further research are presented below: 1. Conduct cold-flow studies (i.e., zero evaporative mass transfor) of film instability and entrainment to determine the influence, if any, of the dimensionless liquid flow rate W_0^+ , the dimensionless film thickness δ_f^+ , the Reynolds number for the gas stream, the pressure, etc on those phenomena. For these studies, and for those recommended below, several fluids in addition to water should be studied, and pressures significantly greater than ambient should be employed.

2 Conduct additional culd-flow studies utilizing a pitct probe to determine the velocity profile across the boundary layer is a function of x. Use those data to study the "effective" roughness of the gasliquid interface

3 Convuct heat-mass transfer studies to obtain data that can be employed to evaluate the correlation procedure suggested in Section 3.3 Initial studies should be conducted to obtain bulk heat-mass transfer

data for a significantly wide range of experimental parameters. Subsequently, more sophisticated studies should be conducted to obtain boundary layer data that would aid in the understanding of the phenomena occurring at the gas-liquid interface 4. Experimentally investigate the region downstream of the liquid film for the case where the wall is adiabatic. Initial studies should be conducted wherein only the wall temperature distribution and the bulk flow parameters are determined. Those data can be employed to evaluate the analysis presented in Section 4.2. Subsequently, experimental studies should be conducted to obtain boundary layer measurements that would aid in the postulation of a realistic flow model for the subject wall region.

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APPENDICES

APPENDIX A

NOTATION

C _i	specific concentration of "i" species
c _f 2	skin friction coefficient (Eq. 4-19)
с _р	specific heat at constant pressure
c _v	specific heat at constant volume
d	tube diameter
D	diffusion coefficient
f	denotes a function
h	specific enthalpy
h'	heat transfer coefficient
h ^O	stagnation enthalpy (Eq. 4-1)
ΔH	latent heat of vaporization
I	e Equation 3-35
J	see _quation 3-46
k	thermal conductivity
L	liquid-film-cooled length
m'	rate of liquid coolant flow/unit spanwise length (Eq. 2-7)
m''	rate of gas flow through the boundary layer/unit spanwise
	length (Eq. 4-7)

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n parameter in the power-law velocity profile (see Eq. 4-4)

- p static pressure
- p partial pressure for "i" species

Pr Prandtl number =
$$\frac{\mu}{k}$$

- $Pr_{t} \qquad turbulent Prandtl number = \frac{\mu_{t} c_{p}}{k_{t}}$
- q energy transfer due to conduction
- q_r energy transfer due to radiation
- r recovery factor (Eq. 3-26)
- R perfect gas constant
- R_d Reynolds number based on d
- Reynolds number based on $x = \frac{\rho_e u_e x}{\mu_e}$

R Reynolds number based on
$$\xi = -\frac{U}{\mu_e}$$

$$R_{\theta}$$
 Reynolds number based or $\theta = \frac{\rho e^{-\mu} e^{-\theta}}{\mu} e^{-\mu}$

- Sc Schmidt number = $\frac{u}{p D}$
- Sc_t turbulent Schmidt number = $\frac{\mu_t}{\rho D_t}$
- St Stanton number for heat transfer (Eq. 3-25)
- St_o St evaluated in the absence of mass transfer
- St' Stanton number for mass transfer (Eq. 3-32)
- t static temperature
- t recovery temperature for the main stream (Eq. 3-26)
- t⁰ stagnation temperature

Sat	saturation temperature for the liquid coolant which cor-
<pre>sdt responds to the static pressure p u streamwise component of velocity u⁺ dimensionless velocity (Eq. 2-5) v normal component of velocity W⁺ dimensionless liquid flow rare (Eq. 2-10) W⁺_i see Equation 2-37 distance downstream from the effective leading edge of the dynamic boundary layer y distance normal to the wall or the liquid film surface</pre>	
u	streamwise component of velocity
u ⁺	dimensionless velocity (Eq. 2-5)
v	normal component of velocity
W ⁺	dimensionless liquid flow rare (Eq. 2-10)
W,+	see Equation 2-37
:	distance downstream from the effective leading edge of the
	dynamic boundary layer
У	distance normal to the wall or the liquid film surface
y ⁺	dimensionless normal coordinate (Eq. 2-6)

Greek

Δ	dynamic boundary layer thickness
^ c	diffusional boundary layer thickness
Δt	thermal boundary layer thickness
7*	displacement thickness (Eq. 4-13)
	heat transfer parameter (Eq. 3-1)
	specific heat ratio
5	viscous sublayer thickness
⁵ f	liquia film thickness
3 ⁺	dimensionless viscous sublayer thickness
5 <mark>†</mark>	dimensionless liquid film thickness (Eq. 2-11)
۲ ک	stability effectiveness (Eq. 2-38)
t	momentum thickness (Eq. 4-14)
i.	dynamic viscosity

v	kinematic viscosity
ξ	energy thickness (Eq. 3-36)
Q	density
τ	shear stress
^τ s ,0	wall shear stress for zero mass transfer
Subscript	<u>s</u>

0	evaluated at the plane of liquid coolant injection
1	evaluated at the terminus of the liquid-film-cooled length
2	evaluated at the terminus of the gas-vapor-cooled length
aw	adiabatic wall
d	diffusional component
e	main-stream state
q	main-stream gas
gv	gas-vapor cooling
1	liquid
L	evaluated at the gas-liquid interface but in the liquid phase
lf	liquid-film cooling
S	evaluated at the gas-liquid interface
	evaluated at the gas-liquid interface but in the gas phase
t	transpiration cooling
t	turbulent contribution
v	vapor
W	wall
:	evaulated at y = 4

evaluated at the reference state defined by Eqs. 3-28 and
3-29

Superscripts

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APPENUIX B

A SAMPLE CALCULATION FOR THE INTERFACE TEMPERATURE

To illustrate the manner in which the calculated results presented in Fig. 3-2 were obtained, a sample calculation for the interface temperature t_s is presented below.

For the flow conditions presented in Fig. 3-2, Eq. 3-33 reduces to the form

$$c_{p_{air}} \frac{\Delta H_{v}}{(t_{e} - t_{s})} = \frac{R_{v}}{R_{g}} \left(\frac{p - p_{v,s}}{p_{v,s}} \right)$$
$$= \frac{83.35}{53.35} \left(\frac{p - p_{v,s}}{p_{v,s}} \right)$$
$$= 1.562 \left(\frac{p - p_{v,s}}{p_{v,s}} \right)$$
(B-1)

Solving Eq. B-1 for the partial pressure of the vapor at the interface, one obtains

$$P_{v,s} = \frac{p}{\frac{1}{1.562} \frac{\Delta H_{v}}{c_{\mu air}(t_{e} - t_{s})} + 1}$$
(B-2)

For purposes of illustration, assume that

 $t_e = 2000 R (c_p = 0.2772 B/1b)$ p = 500 psia For that case, Eq. B-2 reduces to

$$P_{v,s} = \frac{500}{(2.242)(\Delta H_v)} \text{ psia}$$
(B-3)
$$\frac{1}{(2^{\circ}00^{\circ}-t_s(R))} + 1$$

To initiate the trial-and-error calculations, assume that $t_s = 400F = 860R$ (the accuracy of the first approximation for t_s is not critical). For $t_s = 960^{\circ}R$, the data in Ref. 33 gives $\Delta H_v = 826.0$ B/lb and $p_{v,s} = 247.3$ psia. Substituting those data into Eq. B-3, one obtains

247.3 psia =
$$\frac{500}{1.524 + 1}$$
 psia

For the second calculation, assume that $p_{v,s} = 190.5$ psia. Corresponding to $p_{v,s} = 190.5$ psia, the data in Ref. 33 gives $\Delta H_v = 846.7$ B/lb and $t_s = 837.6R$. Substituting those data into Eq. B-3, one obtains

Continuing that procedure, assume for the next calculation that $p_{v,s} = 189.9$ psia. Corresponding to $p_{v,s} = 189.9$ psia, the data in Ref. 33 gives $\Delta H_v = 846.9$ B/lb and $t_s = 377.4F = 837.4R$. Substituting those data into Eq. B-3, one obtains

$$189.9 \stackrel{?}{=} \frac{500}{1.633 + 1} + 1 \text{ psia}$$
$$= 189.9 \text{ psia}$$

Thus, for $t_s = 2000$ R and p = 500 psia, and for the flow conditions presented in Fig. 3-2, the calculated value of the temperature at the gas-liquid interface is $t_s = 377.4F = 837.4R$.

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3 ABSTRACT A discussion of the p	hysical nature of li	nuid-fi	lm cooling is		
presented that gives special e	mphasis to those pher	nomena (occurring at the		
gas-liquid interface. The int	erfacial mass and he	at balar	nces are discussed		
in detail together with the ph	enomena of film insta	ability	and the entrainment		
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to agree favorably with the li	mited experimental da	ata.			
A heat transfer analysis	is presented for the	wall ro	egion downstream of		
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considered: (a) the case where		ic; and	(b) the case where		
the wall is cooled externally.					

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