

**UNCLASSIFIED**

---

---

**AD 285 338**

---

*Reproduced  
by the*

**ARMED SERVICES TECHNICAL INFORMATION AGENCY  
ARLINGTON HALL STATION  
ARLINGTON 12, VIRGINIA**



---

---

**UNCLASSIFIED**

NOTICE: When government or other drawings, specifications or other data are used for any purpose other than in connection with a definitely related government procurement operation, the U. S. Government thereby incurs no responsibility, nor any obligation whatsoever; and the fact that the Government may have formulated, furnished, or in any way supplied the said drawings, specifications, or other data is not to be regarded by implication or otherwise as in any manner licensing the holder or any other person or corporation, or conveying any rights or permission to manufacture, use or sell any patented invention that may in any way be related thereto.

CATALOGED BY ASTIA  
AS AD NO. 285338

285338

285338

ITEM OF INTEREST

Prepared by



Aerospace Information Division

SUBJECT: Combustion of Propellants: Ammonium Perchlorate

SOURCE: Romodanova, L. D., and V. I. Roshchupkin. On the combustion of ammonium perchlorate. Zhurnal fizicheskoy khimii, v. 36, no. 7, Jul 1962, 1554-1555. (S/076/62/036/007)

The article contains data on ammonium perchlorate (AP) combustion compiled from experiments at the Institute of Chemical Physics, Academy of Sciences USSR, and is apparently intended to augment work done in the field by non-Soviet researchers. (All four sources cited in the article are from English-language publications.) Applications of AP in aerospace technology are implied by the consideration in the article of the lower pressure limit for steady AP combustion. In addition, these experiments may be related to Soviet studies of the thermal decomposition of salts of ammonia and hydrazine. (See AID Reports 60-42 and 62-90.)

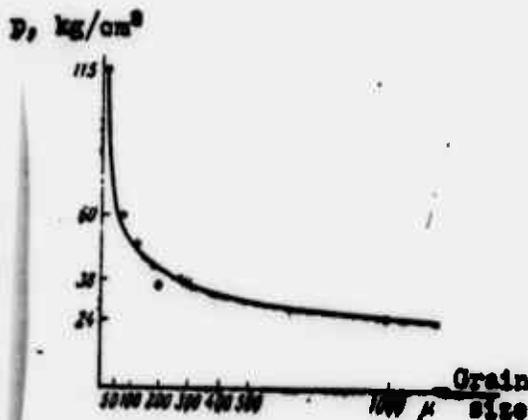


Fig. 1. Pressure limit versus grain size

AP heated in vacuum (~2 to 4 mm Hg) partially sublimed at ~200 to 210°C and after 30 to 40 minutes underwent partial endothermic decomposition (28 to 30%) without self-ignition. The 80 cm<sup>3</sup>/g of gaseous products formed consisted of NO; NO<sub>2</sub>; N<sub>2</sub>; N<sub>2</sub>O; Cl<sub>2</sub>; HCl; HClO<sub>4</sub>; O<sub>2</sub>; and H<sub>2</sub>O. At 300°C NOCl was also evolved.

Lower pressure limit (p) for steady AP combustion and its variation with the moisture content, density (ρ), and grain size (3 to 3000μ) of the charge were given primary consideration. The effect of moisture on AP combustion was determined from AP samples

moistened with water vapor and compressed to a density of 1.92 to 1.93 g/cm<sup>3</sup>. Moisture content of 0.1 to 0.7% did not affect p, but when it exceeded 2%, the samples did not burn at all in the pressure range studied (0 to 150 kg/cm<sup>2</sup>).

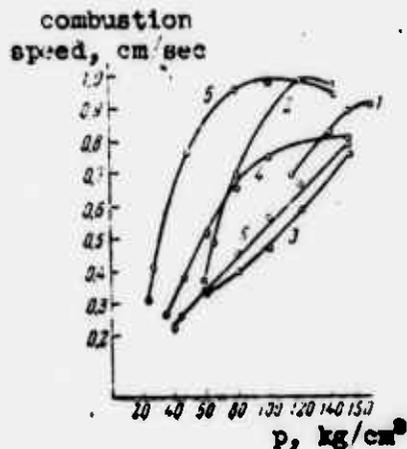


Fig. 2. Combustion speed versus pressure

- 1 - 3 to 7 $\mu$ ; 2 - 55 to 95 $\mu$ ;
- 3 - 5 to 250 $\mu$ ; 4 - 100 to 300 $\mu$ ;
- 5 - 250 to 400 $\mu$ ;
- 6 - 300 to 3000 $\mu$ .

centimeters per second were measured optically by means of photographic recording.) The combustion speeds of various grain size fractions of AP powders at all pressures (up to  $\sim 150$  kg/cm<sup>2</sup>) were dissimilar, but each fraction approached a constant combustion speed (Fig. 2).

The lower pressure limit was found to decrease with increasing charge density. The following values of p were found for the grain-size fraction from 200 to 350 $\mu$ :

- at  $\rho = 1.50$  g/cm<sup>3</sup>, p = 53 kg/cm<sup>2</sup>;
- at  $\rho = 1.67$  g/cm<sup>3</sup>, p = 48 kg/cm<sup>2</sup>;
- at  $\rho = 1.92$  g/cm<sup>3</sup>, p = 38 kg/cm<sup>2</sup>.

At constant density, values of p were inversely proportional to the grain size of the AP particles (Fig. 1). Steady combustion of large-crystal AP powders ( $\sim 1000\mu$ ) was observed at  $p \approx 25$  kg/cm<sup>2</sup>, whereas, fine-crystal AP powders ( $\sim 5\mu$ ) burned steadily at 115 to 118 kg/cm<sup>2</sup>. It was also found that p decreases inversely with sample diameter (at a charge density of 192 g/cm<sup>3</sup> and grain size of 200 to 350 $\mu$ ). For example at 8 mm sample diameter p = 38 kg/cm<sup>2</sup>; at 12.7 mm p = 33 kg/cm<sup>2</sup>.

Variations in combustion speed with pressure were also considered. (Combustion speeds in