QUALITATIVE ASSESSMENT OF THE IGNITION OF HIGHLY FLAMMABLE FUELS BY PRIMARY EXPLOSIVES (U) MATERIALS

RESEARCH LABS ASCOT VALE (AUSTRALIA)

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P.P. Elischer and Leo de Yong

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ABSTRACT

An assessment of the ignition of fuel/air mixtures and of fabrics soaked with different fuels (ethanol, n-hexane and diethyl ether) by primary explosives has been carried out.

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   2.2 Ignition of Flammable Environments by Primary Explosives
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FLAMMABLE FUELS BY PRIMARY EXPLOSIVES

1. INTRODUCTION

A search of the literature revealed that there are no standard tests which relate to the ignitability of flammable environments by primary explosives. Situations arise, both in civilian and military practice, where this information is either necessary or highly desirable.

In this report, the authors sought to establish a suitable test for assessing the ignitability of either fuel/air mixtures (vapour state ignition) or fuel-soaked woven fabrics (liquid-vapour state ignition) in the presence of either initiating primary explosives, a photoflash pyrotechnic composition or an electric fusehead.

2. RESULTS AND DISCUSSION

2.1 Flammable Environments

The limits of flammability, flash point and auto ignition temperature of several common organic solvents are given in Table 1.

As the most likely environment in which fuel-air mixtures are to be encountered in practice are those involving organic solvents or petroleum based fuels, it was decided to study the flammability of ethyl alcohol, diethyl ether and n-hexane (as a model for petroleum fuels).

Brown [1] has reported that a very small amount of a volatile contaminant in a relatively non flammable fuel may make it flammable, and Affens [2] has shown that small amounts of volatile n-alkanes present in a relatively non inflammable mixture of n-alkanes will yield a flammable mixture. As petrol contains predominantly C5-C10 n-alkanes, results obtained using n-hexane (C6) can probably be applied to petroleum.
2.2 Ignition of Flammable Environments by Primary Explosives

The study of ignition of flammable environments by primary explosives concentrated on two areas:

1. Ignition of fuel soaked fabrics
2. Ignition of fuel/air mixtures.

Depending on the nature of the ignition source and the fuel/air ratio, combustion of a fuel/air mixture can occur either by deflagration or detonation. Hooper [3] has reported that the mass of high explosive (tetryl) required to cause detonation in fuel/air mixtures ranged from less than 1 g (acetylene/air) to 10 kg (methane/air) for gas volumes up to 2 m$^3$. It is considered that the possibility of a fuel/air mixture detonating when exposed to the output from approximately 200 mg of primary explosive is remote.

2.2.1 Ignition of fuel soaked fabrics using primary explosive

These tests were carried out to screen the primary explosives and to ascertain whether they would ignite material soaked with fuel. Two series of tests were carried out:

(i) using a loose woven fabric (cheesecloth), and
(ii) using a tight woven fabric (nylon carpet).

Cheesecloth was chosen because its open weave would expose a large surface area to the output of the explosive and carpet was chosen because it is a common floor covering, representative of modern tight woven fabrics.

In all cases, 200 mg of primary explosive was used.

2.2.1.1 Loose woven fabrics

Selected primary explosives were confined in metal, perspex or bakelite tubes and fired onto hexane soaked cheesecloth using different igniter systems. Igniter tubes, filled with Photoflash Composition, MRL(X)206, were used as controls. These tubes consistently ignited the hexane-soaked fabric.

The results listed in Tables 2, 3 & 4 indicate that the ignition of hexane soaked cheesecloth depends on the ignition system, the primary explosive and on the igniter tube.

To test this, experiments were carried out using metal tubes (brass and aluminium) in place of the bakelite and perspex tubes (Table 4). Unlike
the tests summarized in Tables 2 & 3, the hexane soaked cheesecloth did not ignite. The ignition of the cheesecloth was attributed to the perspex or bakelite tubes bursting and exposing the flammable atmosphere to hot perspex or bakelite particles. The metal tubes did not burst nor did the force of the explosion significantly affect their physical dimensions.

2.2.1.2 Tight woven fabrics

A number of explosive units manufactured from metal tubing were fired over tightly woven fabric (carpet) soaked in ethanol. Ethanol was chosen as the fuel because of its wider flammability range compared to n-hexane. The results are summarised in Table 5.

Although there was some carbonaceous residue remaining on the carpet, the ethanol and carpet did not ignite. Electric fuseheads, used as controls, did ignite the ethanol which caused the carpet to burn.

2.2.2 Ignition of fuel/air mixtures

Investigations were also carried out to determine whether primary explosives filled into brass or aluminium tubes would ignite fuel/air mixtures. Metal tubes were filled with 200-400 mg of lead styphnate, 200 mg of lead dinitro-resorcinate or 200 mg of tetracene and ignited using nichrome bridgewires.

Three different fuels were assessed: ethanol, n-hexane and diethyl ether. These fuels have a range of auto-ignition temperatures and flammability limits (Table 1) corresponding to a range of flammabilities. Stoichiometric volumes ($V_1$) and volumes corresponding to the middle of the flammable range ($V_{2/3}$) were used. Controls to check ignition of the fuel/air mixture were fired into the test vessel prior to the commencement of the test.

The results summarized in Table 6 show that lead styphnate, tetracene and lead dinitro-resorcinate in the tubes used, did not ignite the fuel/air mixtures investigated.

3. CONCLUSION

For the arrangements examined, the primary explosives when confined in metal tubes did not ignite the fuel soaked fabrics nor the fuel/air mixtures. However, with perspex or bakelite tubes, ignition occurred possibly due to break up of the tube.

Examination of the reasons why the primary explosives did not ignite the flammable environments is complex and depends strongly on diffusive and chemical parameters such as intensity and duration of heat source and properties of the flammable mixture such as thermal conductivity and temperature of ignition. These parameters were not examined in this report.
as only a qualitative assessment of specific primary explosive arrangements was required.

4. EXPERIMENTAL

4.1 General

The hydrocarbon fuels, the cheesecloth and the carpet used for the flammability investigations were all commercially available. n-Hexane, diethyl ether and ethanol were analytical grade reagents. The primary explosives used in the investigation are listed below:

- basic lead azide
- lead azide
- lead styphnate
- barium styphnate
- potassium picrate
- lead dinitro-resorcinate (LDNR)
- and tetracene

They were prepared at the Materials Research Laboratories using established techniques. The Photoflash Composition, MRL (X) 206, was prepared at the Materials Research Laboratories using 40% magnesium, 59% potassium perchlorate and 1% acroid resin.

Electric Fuseheads

These were the standard ICI Vulcan Type A matchheads which contain a 0.9-1.6 Ω resistance nichrome bridgewire, 0.041 mm diameter.

Experimental Bridgewire Devices

These devices were manufactured at MRL (Fig. 1). They were fabricated from plastic, which was moulded around two phosphor bronze terminals. Platinum wire (0.04 mm diameter, 0.2 Ω resistance) or nichrome wire (0.015 mm diameter, 12-20 Ω resistance) was used as the bridgewire which was spot welded to the terminals.
4.2 Manufacture of Tubes Containing Primary Explosives

4.2.1 Bakelized paper tubes

These tubes were constructed using commercially available bakelized paper (9 mm OD, 7.2 mm ID, 30 mm L). They were fitted with fuseheads held in place with a rubber grummet and epoxy resin. The explosive compositions were loose filled into the tube and lightly consolidated with a 4 mm cork disc and a fillet of epoxy resin.

4.2.2 Perspex tubes

These tubes were prepared using perspex (12.5 mm OD, 4.1 mm ID, 15 mm L). Platinum or nichrome bridgewire devices were fitted to the tube using Eastman 910 adhesive. The primary explosive was pressed into the tubes using an Altor press and 35 kg dead load. These tubes were not sealed.

4.2.3 Metal tubes

The metal tubes were constructed from aluminium (15 mm OD, 9 mm ID, 15 mm L) or brass (9.5 mm OD, 7.2 mm ID, 14 mm L). They were fitted with bridgewire devices held in place with Eastman 910 adhesive. The explosive was loose filled into the tubes and lightly consolidated with a cork (4 mm thick) or cardboard disc (2 mm thick) and a fillet of epoxy resin (Fig. 1).

4.3 Ignition of Hydrocarbon Fuel/Air Mixtures

A test to assess the ignition characteristics of various fuel/air mixtures when exposed to different explosive stimuli was developed. The test apparatus consisted of a 20 litre metal container fitted with an air driven stirrer which was permanently attached through the side near the bottom. A plastic sheet was held securely over the top of the container with a large metal hose clip. The explosive tube or electric fusehead was inserted through a hole in the plastic so that it was suspended in the middle of the container. Hydrocarbon fuel was injected through the plastic using a syringe. Both holes were sealed with adhesive tape. The volume of fuel used for the tests was either a stoichiometric amount (Vs) or an amount corresponding to the middle of the flammability range (Vm). The relevant figures to calculate those volumes were obtained from the literature (4 & 5). A homogeneous fuel/air mixture was obtained by mixing the vapour for five minutes using the air driven stirrer. This was then turned off and the test units fired using a capacitor discharge firing box. The mixing of diethyl ether and n-hexane was carried out at ambient temperature. However, for ethanol, due to its lower volatility at ambient temperature, a homogeneous mixture was obtained by pre-heating the container to approx 30°C.

A firing was deemed successful when the fuel/air mixture did not ignite. Conversely ignition of the fuel/air mixture indicated by a fire ball, was classed as a failure. After each successful firing, the remaining
fuel was ignited using an electric fusehead.

4.4 Flammability of Loose Woven Fabric Saturated with Fuel

Cheesecloth material was cut into strips (63 mm x 200 mm), placed into an uncovered 20 litre container and thoroughly soaked with about 50 cm$^3$ of fuel. The primary explosive tubes or electric fuseheads were suspended 100 mm above the material and initiated. Again the firing was successful if the fuel soaked material did not ignite. If the cheesecloth ignited and was badly charred, it was replaced for subsequent tests. The cheesecloth was also replaced when the flammability characteristics of different hydrocarbon fuels were investigated.

4.5 Flammability of Tight Woven Fabric Saturated with Fuel

A sample of commercial nylon carpet (approximately 0.6 m$^2$) containing no fire retardant material, was saturated with 50 cm$^3$ of liquid fuel. The primary explosive tubes were orientated at different angles to the carpet and fired electrically. The following configurations were used:

(a) resting on the carpet, facing down at right angles to the plane of the carpet;
(b) resting on the carpet parallel to the plane of the carpet;
(c) held approximately 20 mm above the carpet facing down at right angles to the carpet.

Again a firing was successful if the carpet did not ignite.

5. ACKNOWLEDGEMENT

The helpful discussions and advice from Dr J. McRae on flammability of solvents is acknowledged.
6. REFERENCES


TABLE 1

LIMITS OF FLAMMABILITY OF GASES AND VAPOURS
IN AIR [4,5]

<table>
<thead>
<tr>
<th>Component</th>
<th>Boiling Point at 1 atm. (°C)</th>
<th>Vapour Density compared with air (air = 1)</th>
<th>Flash Point (°C)</th>
<th>Auto Ignition Temperature (°C)</th>
<th>Flammability Limits in air</th>
<th>Upper Vol %</th>
<th>Lower Vol %</th>
</tr>
</thead>
<tbody>
<tr>
<td>n-Butane</td>
<td>- 1</td>
<td>2.05</td>
<td>- 60</td>
<td>405</td>
<td>8.44</td>
<td>1.8</td>
<td></td>
</tr>
<tr>
<td>n-Pentane</td>
<td>36</td>
<td>2.48</td>
<td>&lt; - 20</td>
<td>258</td>
<td>7.8</td>
<td>1.4</td>
<td></td>
</tr>
<tr>
<td>n-Hexane</td>
<td>69</td>
<td>2.79</td>
<td>- 21</td>
<td>223</td>
<td>7.4</td>
<td>1.2</td>
<td></td>
</tr>
<tr>
<td>n-Heptane</td>
<td>98</td>
<td>3.46</td>
<td>- 4</td>
<td>223</td>
<td>6.7</td>
<td>1.05</td>
<td></td>
</tr>
<tr>
<td>n-Octane</td>
<td>126</td>
<td>-</td>
<td>-</td>
<td>220</td>
<td>-</td>
<td>0.95</td>
<td></td>
</tr>
<tr>
<td>n-Nonane</td>
<td>151</td>
<td>4.43</td>
<td>30</td>
<td>206</td>
<td>5.6</td>
<td>0.85</td>
<td></td>
</tr>
<tr>
<td>n-Decane</td>
<td>174</td>
<td>-</td>
<td>-</td>
<td>208</td>
<td>5.6</td>
<td>0.75</td>
<td></td>
</tr>
<tr>
<td>Ethanol</td>
<td>78</td>
<td>1.59</td>
<td>12</td>
<td>365</td>
<td>19.0</td>
<td>3.3</td>
<td></td>
</tr>
<tr>
<td>Diethyl ether</td>
<td>34.5</td>
<td>2.55</td>
<td>&lt; - 20</td>
<td>160</td>
<td>36.0</td>
<td>1.9</td>
<td></td>
</tr>
<tr>
<td>Dimethyl ether</td>
<td>- 25</td>
<td>1.59</td>
<td>-</td>
<td>350</td>
<td>27.0</td>
<td>3.4</td>
<td></td>
</tr>
</tbody>
</table>
**TABLE 2**

IGNITION OF HEXANE SOAKED CHEESECLOTH USING ELECTRIC FUSEHEADS IN BAKELITE TUBES

<table>
<thead>
<tr>
<th>Ignition System</th>
<th>Primary Explosive</th>
<th>Flammability</th>
</tr>
</thead>
<tbody>
<tr>
<td>Electric Fusehead in Bakelite Tubes</td>
<td>Basic Lead Azide</td>
<td>✓</td>
</tr>
<tr>
<td></td>
<td>Lead Styphnate</td>
<td>✓</td>
</tr>
<tr>
<td></td>
<td>LDNR</td>
<td>✓</td>
</tr>
<tr>
<td></td>
<td>Barium Styphnate</td>
<td>✓</td>
</tr>
<tr>
<td></td>
<td>Flash Composition</td>
<td>✓</td>
</tr>
<tr>
<td></td>
<td>Potassium Picrate</td>
<td>✓</td>
</tr>
</tbody>
</table>

**TABLE 3**

IGNITION OF HEXANE SOAKED CHEESECLOTH USING BRIDGEWIRES IN PERSPEX TUBES

<table>
<thead>
<tr>
<th>Ignition System</th>
<th>Primary Explosive</th>
<th>Flammability</th>
</tr>
</thead>
<tbody>
<tr>
<td>Platinum Bridgewire in Perspex Tubes</td>
<td>Potassium Picrate</td>
<td>✓</td>
</tr>
<tr>
<td></td>
<td>Basic Lead Azide</td>
<td>X</td>
</tr>
<tr>
<td></td>
<td>Lead Azide</td>
<td>X</td>
</tr>
<tr>
<td></td>
<td>Lead Styphnate</td>
<td>✓</td>
</tr>
<tr>
<td>Nichrome Bridgewire in Perspex Tubes</td>
<td>Lead Styphnate</td>
<td>✓</td>
</tr>
</tbody>
</table>

✓ denotes ignition of fuel soaked rags
X denotes no ignition of fuel soaked rags
<table>
<thead>
<tr>
<th>Ignition System</th>
<th>Primary Explosive</th>
<th>Flammability</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nichrome Bridgewire in Aluminium Tube</td>
<td>Potassium Picrate</td>
<td>X</td>
</tr>
<tr>
<td>&quot; &quot; &quot;</td>
<td>LDNR</td>
<td>X</td>
</tr>
<tr>
<td>&quot; &quot; &quot;</td>
<td>Barium Styphnate</td>
<td>X</td>
</tr>
<tr>
<td>&quot; &quot; &quot;</td>
<td>Tetracene</td>
<td>X</td>
</tr>
<tr>
<td>&quot; &quot; &quot;</td>
<td>200 mg Lead Styphnate</td>
<td>X</td>
</tr>
<tr>
<td>&quot; &quot; &quot;</td>
<td>400 mg Lead Styphnate</td>
<td>X</td>
</tr>
<tr>
<td>&quot; &quot; &quot;</td>
<td>600 mg Lead Styphnate</td>
<td>X</td>
</tr>
<tr>
<td>Nichrome Bridgewire in Brass Tube</td>
<td>Tetracene</td>
<td>X</td>
</tr>
<tr>
<td>&quot; &quot; &quot;</td>
<td>Lead Styphnate</td>
<td>X</td>
</tr>
</tbody>
</table>

✓ denotes ignition of fuel soaked rags

X denotes no ignition of fuel soaked rags

TABLE 4
IGNITION OF HEXANE SOAKED CHEESECLOTH USING ALUMINIUM AND BRASS TUBES
<table>
<thead>
<tr>
<th>Primary Explosive</th>
<th>Position</th>
<th>Results</th>
</tr>
</thead>
<tbody>
<tr>
<td>Lead Styphnate</td>
<td>Resting on carpet, facing onto carpet</td>
<td>No ignition of carpet, charring</td>
</tr>
<tr>
<td></td>
<td>Held down approx 20 mm from carpet</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Held down facing along surface of carpet</td>
<td></td>
</tr>
<tr>
<td>LDNR</td>
<td>Resting on carpet, facing onto carpet</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Held down approx 20 mm from carpet</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Held down facing along surface of carpet</td>
<td></td>
</tr>
<tr>
<td>Tetracene</td>
<td>Held down approx 20 mm from surface of carpet</td>
<td>No ignition, no charring</td>
</tr>
<tr>
<td></td>
<td>Resting on carpet facing along surface of carpet</td>
<td></td>
</tr>
<tr>
<td>Electric Fusehead</td>
<td>Held down approx 20 mm from surface of carpet</td>
<td>Fuel ignited and carpet burnt</td>
</tr>
<tr>
<td></td>
<td>Resting on carpet facing along surface of carpet</td>
<td></td>
</tr>
<tr>
<td>Flash Composition</td>
<td>Held down approx 20 mm from surface of carpet</td>
<td>No ignition, charring, hole in carpet</td>
</tr>
</tbody>
</table>
### TABLE 6
IGNITION OF FUEL/AIR MIXTURES USING BRIDGEWIRES AND METAL TUBING

<table>
<thead>
<tr>
<th>Primary Explosive</th>
<th>Fuel</th>
<th>Volume</th>
<th>Flammability</th>
</tr>
</thead>
<tbody>
<tr>
<td>Electric Fusehead</td>
<td>Hexane</td>
<td>Vs &amp; Vm</td>
<td>✓</td>
</tr>
<tr>
<td>&quot;</td>
<td>Ethanol</td>
<td>Vs</td>
<td>✓</td>
</tr>
<tr>
<td>&quot;</td>
<td>Ether</td>
<td>Vs</td>
<td>✓</td>
</tr>
<tr>
<td>200 mg Lead Styphnate (Aluminium Tube)</td>
<td>Hexane</td>
<td>Vs &amp; Vm</td>
<td>X</td>
</tr>
<tr>
<td>&quot;</td>
<td>Ethanol</td>
<td>Vm</td>
<td>X</td>
</tr>
<tr>
<td>&quot;</td>
<td>Ether</td>
<td>Vs &amp; Vm</td>
<td>X</td>
</tr>
<tr>
<td>300 mg Lead Styphnate</td>
<td>&quot;</td>
<td>Vm</td>
<td>X</td>
</tr>
<tr>
<td>400 mg Lead Styphnate</td>
<td>&quot;</td>
<td>Vm</td>
<td>X</td>
</tr>
<tr>
<td>200 mg LDNR</td>
<td>&quot;</td>
<td>Vm</td>
<td>X</td>
</tr>
<tr>
<td>&quot;</td>
<td>Ethanol</td>
<td>Vm</td>
<td>X</td>
</tr>
<tr>
<td>&quot;</td>
<td>Ether</td>
<td>Vs &amp; Vm</td>
<td>X</td>
</tr>
<tr>
<td>200 mg Tetracene (Aluminium tube)</td>
<td>Hexane</td>
<td>Vs</td>
<td>X</td>
</tr>
<tr>
<td>&quot;</td>
<td>Ethanol</td>
<td>Vs</td>
<td>X</td>
</tr>
<tr>
<td>&quot;</td>
<td>Ether</td>
<td>Vs &amp; Vm</td>
<td>X</td>
</tr>
<tr>
<td>200 mg Tetracene (Brass tube)</td>
<td>Ether</td>
<td>Vs &amp; Vm</td>
<td>X</td>
</tr>
</tbody>
</table>

✓ denotes ignition of fuel/air system  
X denotes no ignition of fuel/air system
MRL EXPERIMENTAL BRIDGEWIRE DEVICE

BRASS TUBE

PRIMARY EXPLOSIVE

CARDBOARD DISC

EPOXY PLUG

NICROME BRIDGEWIRE

LEAD WIRES

\[ \Phi 9.5 \]

\[ \Phi 7.2 \]

6.0

12.0

2.0

(DIMENSIONS IN mm)

FIG. 1 EXPLOSIVE TUBE
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