REACTION PROBABILITIES IMPLIED BY MULTIPLE GAS-SURFACE INTERACTIONS

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FORWARD

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ABSTRACT

The probabilities of single and two successive gas surface collisions satisfying minimum interaction energy cutoffs are considered in the hard cube model sense. Comparison is made with atomic and molecular oxygen reaction probabilities with graphite corresponding to specific mass ratio examples of 4/3 and 8/3. Comparable peak probabilities result for multiple interactions but of much reduced sharpness with temperature for only a two collision constraint. Realistic activation energies are implied from a match of the energy cutoff probability model to the portion of the experimental data increasing with surface temperature. The rapidly decreasing reaction probabilities found experimentally at higher temperatures imply an appreciable increase in the required number of effective collisions.
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SYMBOLS

A, B
regions of \((u_0, v_0)\) plane used for \(P_{II}\) evaluations

A, B, C, D
integration limit parameters for \(I_j\), Equations (40), (41)

\(G\)
distribution function for surface particle speed \(|v_0|\), Equation (1)

\(H\)
distribution function for gas particle normal velocity component \(u_0\), Equation (2)

\(I_j(\bar{u}_0)\)
integral form, Equations (40), (42)

\(J_j(\bar{u}_0)\)
integral form, Equation (43)

\(K_j\)
Equations (45), (48)

\(k\)
Boltzmann constant

\(k_i\)
slope of border to \(B_i\) region

\(L_j\)
Equation (58)

\(L\)
surface "box" length; also integration limit for \(J_j\)

\(M_j\)
Equations (50), (51)

\(m_g\)
gas particle mass

\(m_s\)
surface particle mass

\(m^*\)
reduced mass, \(m_s m_g / (m_s + m_g)\)

\(N\)
Equation (49)

\(dP_{loc}\)
surface particle location probability, Equation (20)

\(P_D\)
directional probability that first collision is inward or outward for given \((u_0, v_0)\) pair, Equation (25)

\(P_2\)
probability that second collision follows from \((u_0, v_0)\) first collision

\(P_I(\epsilon)\)
probability of first collision with interaction energy in excess of \(\epsilon_1\)

\(P_{II}(\epsilon_1, \epsilon_2)\)
probability of second collision with interaction energy in excess of \(\epsilon_1, \epsilon_2\) for first, second collisions respectively
$P_i, Q_i, R_i$ coefficients in integrand of $B_i$ region contribution to $P_{II}$, Equation (65)

$T_g$ incident gas temperature

$T_s$ surface temperature

$\bar{T}$ $T_s/T_g \cos^2 \theta$, Equation (41)

$T^*$ $(\mu T)^{-1/2}$

$U$ integration limit for $J_j$

$u_0$ gas particle normal velocity component prior to first collision, positive inward

$u_1$ gas particle normal velocity component after first collision, positive inward

$\bar{u}_0, \bar{v}_0$ normalized particle velocities, $x_g^{1/2} u_0, x_s^{1/2} v_0$

$v_0$ surface particle velocity prior to both first collision and/or rebound, positive outward

$v_{0\text{coll}}$ surface particle velocity immediately prior to first collision, positive outward

$v_{0i}$ $v_0$ on cut-off energy, $\varepsilon_i$, locus, Equations (8), (17)

$v_1$ surface particle velocity after first encounter, positive inward

$x$ distance from outer edge of surface, positive inward

$x^*$ first collision location

$(x^*/L)_m$ minimum location (normalized) for first collision which allows second collision possibility, Equation (27)

$x_g$ $m_g/2kT_g \cos^2 \theta$

$x_s$ $m_s/2kT_s$

$y$ Equation (52)

$\varepsilon_i$ Cut-off energy for first collision

$\overline{\varepsilon}_i$ nondimensional cut-off energy, Equation (11)
\( \bar{\varepsilon}_I, \bar{\varepsilon}_{II} \)  
first, second collision relative integration energy, Equations (7), (12)

\( \langle \varepsilon_R \rangle_j \)  
reaction probability for oxygen on graphite, \( j = 0, O_2 \), Equation (72)

\( \varepsilon^* \)  
\[ [(1 + \mu)\bar{\varepsilon}_I]^{1/2} \]

\( \theta \)  
incidence angle measured from surface normal

\( \mu \)  
ratio of gas to surface particle masses, \( m_g/m_s \)
CHAPTER 1

INTRODUCTION

Measured reaction probabilities for both molecular and atomic oxygen reactions with carbon show anomalous (peaked) Arrhenius plots.\textsuperscript{1,2,3,4} Comparison of results for the molecular and atomic oxidation cases indicate the atom reactivity to be higher by an order of magnitude.\textsuperscript{1,4} It has been suggested that the increase may be related to a higher sticking probability,\textsuperscript{1} and that dissociative adsorption may be the limiting step in the reaction mechanism.\textsuperscript{2,4}

There is some evidence that primarily normal energy barriers are important in the dissociative adsorption and/or recombinative desorption processes.\textsuperscript{5-9} In simplest form, interactions with greater (normal) energies than some cut-off energy will then lead to adsorption in dissociated form. For a gas at rest with respect to an adjacent solid surface the intensity of all particles impinging on the surface varies as the cosine of the angle from the surface normal. However, the intensity is non-cosine for those particles with normal energies greater than some non-zero cut-off, and the distribution is then peaked in the normal direction.

Measured distributions for the desorbing CO product of the reaction in fact vary as $\cos^2 \theta$ \textsuperscript{3,4} and imply possible normal energy barriers for the recombinative desorption process as well. It has been argued that dissociation adsorption and recombinative desorption are reverse processes\textsuperscript{5,6} which suggests normal energy barriers for dissociative adsorption of CO.
These observations add to the importance of the normal interaction, which if dominant offers considerable simplification in modeling the events of gas and solid particle encounters. The hard-cube model has had some success with precisely that assumption assuming surface particles confined to a rigid box and interacting with the normal component of Maxwellian gas particles. However, a single collision restriction was imposed by limiting the mass ratio of gas and surface particles to be \( \mu < \frac{1}{3} \). Primary interest at the time was in inert gas scattering distributions, for which there was comparison data from experiment and with which there proved to be quite reasonable agreement.

For reaction encounters the natural extension of a hard-cube approach is an allowance for an energy barrier and removal of the \( \mu < \frac{1}{3} \) constraint. The former corresponds to imposing a reaction probability, while the latter introduces multiple collisions which may weight the probability according to changes in the residence time on the surface. Single and double collision interactions are considered here while retaining the simplifying assumption of negligible lateral interaction. Thus, near normal encounters are implied and with only one atom of the solid surface.
CHAPTER 2
COLLISION PROBABILITIES MODEL

2.1 Distribution Function Basis for Probability

Goodman\textsuperscript{11} points out that for \( \mu = m_g/m_s < 1/3 \) certain types of two successive collision sequences are ruled out, and that for \( \mu < 1 \) there will be two collisions at most. However, an oxygen-graphite encounter corresponds to \( \mu > 1 \), (i.e., either 16/12 or 32/12) and multiple collisions must be expected here.

We assume a constant speed surface particle moving normal to the surface and instantly reversing direction at the ends of its one dimensional box enclosure.\textsuperscript{10,11} Immediately prior to a first collision its velocity is \( v_{\text{coll}} \), positive in the outward direction, and of magnitude \( 0 \leq |v_{\text{coll}}| \leq \infty \). The probability of a first collision involving \( |v_{\text{coll}}| = |v_0| \) is the probability that the speed is \( |v_0| \) at the time the gas particle enters the box. From the one dimensional Maxwellian distribution this is

\[
G(v_0) dv_0 = 2 \left( \frac{k_b T_a}{\pi} \right)^{\frac{1}{2}} e^{-\frac{k_b T_a}{2} v_0^2} dv_0
\]

where \( \frac{k_b}{2} = m_s/2k_bT_a \).

An approaching gas particle in a Maxwellian beam oriented at an angle \( \theta \) from the surface normal has a normal velocity component \( u_0 \), positive inward, with the distribution \( (0 \leq u_0 \leq \infty) \):

\[
H(u_0) du_0 = 2 \frac{k_b}{2} u_0^3 e^{-\frac{k_b u_0^2}{2}} du_0
\]
where \( x_g = \frac{m_g}{2 E_g \cos \theta} \).

Classical encounters will be assumed between the surface and gas particles along the normal direction. Tangential interactions are neglected and clearly this is a progressively poorer approximation with an increasing number of collisions prior to departure of the gas particle from the box. Indeed, for a sufficiently large number of collisions it should be expected that a substantial part of the tangential momentum is absorbed for any one gas particle. Should a second collision occur it is assumed to be with the same surface particle. The tangential and second collision assumptions together effectively limit the range of \( \theta \) to "near" zero. The explicit appearance of \( \theta \) is then solely in \( x_g \), and in the combination \( T_g \cos^2 \theta \), and is effectively then a beam cooling parameter.

Consider the probability that a first collision occurs for a specific interaction energy. For such collisions to occur prior to rebound of the surfact particle

\[
\begin{align*}
\text{either,} & \quad v_{0,\text{coll}} = v_0 > 0, \quad \text{for an outward type} \\
\text{or,} & \quad v_{0,\text{coll}} = v_0 < 0, \quad \text{for an inward type} \quad (3)
\end{align*}
\]

whereas if rebound precedes the collision,

\[
\begin{align*}
\text{for an outward type} \quad (4)
\end{align*}
\]

This may be interpreted as an increased directional probability, \( P_D(u_0, v_0) \), for the \( v_0 > 0 \), i.e., outward type, collisions and a corresponding decrease in the probability for \( v_0 < 0 \), i.e., inward type, collisions. With that understanding, the probability of a first collision with \( v_0 \text{ coll} = v_0 \) is
at a point in the \((u_0, v_0)\) half plane. Only some fraction of these will result in a second collision. If \(P_2(u_0, v_0)\) is the probability of a second collision resulting from an initial \((u_0, v_0)\) pair for the first collision, then

\[
\frac{dP_2}{d\Omega} = P_2(u_0, v_0) d\Omega
\]

is the overall second collision probability for a point in the \((u_0, v_0)\) half plane. Integration of Equations (5) and (6) over portions of the half plane \((0 \leq u_0 \leq \infty; -\infty \leq v_0 \leq \infty)\) furnishes the proportional number of first and second collisions relative to the number of gas particles arriving at the surface.

In view of the \(P_D\) weighting in Equation (5) only Equation (3) (i.e. \(v_0\) coll = \(v_0\)) need be considered and the relative interaction energy for a first collision is

\[
\epsilon_I = \frac{m^*}{2} \left(\frac{m_0 + m_g}{m^*} \right)^2
\]

in terms of a reduced mass \(m^* = m_s m_g / (m_s + m_g)\). Here \(v_0 \geq -u_0\) since all lesser speeds correspond to Equation (4). For relative energies equal or greater than a cut-off energy \(\epsilon_1\), say, Equation (7) imposes the constraint (Fig. 1):

\[
v_0 > \sqrt{2\epsilon_1 \over m^*} - u_0 \quad (= v_0, \text{say})
\]

The relative number of such collisions is, then, from Equation (5),
\[ P_{\text{i}}(E) = \int_0^\infty \int_0^\infty P_0 H(\|v_0\|) \, dv_0 \, du_0 \]  

(9)

It is worth noting the limiting case of \( T_s \to 0 \) for which \( |v_0| \to 0 \) and \( u_0 \geq \sqrt{2E_i/m^*} \); i.e.,

\[ P_{\text{i}}(E_i) \bigg|_{T_s \to 0} = \int_0^\infty H \, du_0 = \left[ 1 + (\mu \bar{E}) \right] e^{-\frac{(\mu \bar{E})E_i}{E_0}} \]  

(10)

in which

\[ \bar{E}_i = \frac{E_i}{\sqrt{2m^* \omega_0^2 \sigma}} \]  

(11)

is the natural nondimensional interaction cut-off energy.

A second collision energy cut-off follows from the relative interaction energy

\[ E_{\text{II}} = \frac{m^*}{2} \left( u_1 + v_1 \right)^2 \quad \left[ u_1 > -v_1 \right] \]  

(12)

where \( u_1, v_1 \) are the particle speeds immediately after the first encounter. The sign choice in Equation (12) follows from \( v_1 > 0 \) for \( \mu > 1 \) since from momentum and energy considerations

\[ u_1 = \frac{(\mu - 1)u_0 - 2v_0}{\mu + 1} \]  

\[ v_1 = \frac{2\mu u_0 + (\mu - 1)v_0}{\mu + 1} \]  

(13)
if a collision occurs \((v_0 \geq -u_0)\). Here both \(u_1, v_1\) are positive inward.

As a consequence, the second collision occurs, if at all, after the surface particle rebounds from the bottom of the box, and the relative speed is then \((u_1 + v_1)\) for \(u_1 \geq 0\). From Equation (13) a second collision always \(i.e., P_2 = 1.0\) takes place for

\[
v_0 \leq \frac{\mu - 1}{2} u_0, \quad i.e., u_1 \geq 0, \tag{14}\]

and takes place for

\[
v_0 > \frac{\mu - 1}{2} u_0, \quad i.e., u_1 < 0, \tag{15}\]

only if the gas particle is overtaken prior to its departure from the surface \(i.e., P_2 < 1.0\).

A cut-off constraint for interaction energies \(\geq \varepsilon_2\) is then

\[
v_1 \geq \sqrt{\frac{2\varepsilon_2}{m^*}} - u_1\]

and corresponds to (Fig. 1):

\[
v_0 \leq \frac{1}{3-\mu} \left[ (3\mu - 1) u_0 - (\mu + 1) \sqrt{\frac{2\varepsilon_2}{m^*}} \right] \quad (= v_0, \text{ max}) \tag{17}\]

for \(\mu \leq 3\) on introducing Equation (13). The \(T_s \rightarrow 0\) limit here implies

\[
u_0 > \frac{\mu + 1}{3\mu - 1} \sqrt{\frac{2\varepsilon_2}{m^*}} \quad \tag{18}\]

for both \(\mu \leq 3\). Since \(P_2 = 1\) for \(|v_0| \rightarrow 0\) in view of Equation (14),
\[ P(\xi, \eta) = \int_{-\infty}^{\infty} Hdu_0 = \left[ \frac{1}{\mu_1^2 \xi^2} \right] e^{-(1+\mu)\xi^2} \]

where \( \mu_1 = 1, \mu_2 = \frac{\mu + 1}{3\mu - 1} \), and Equation (19) requires a lower limit of integration choice which is the larger of the \( u_0 \) minimums [either Equation (18) or that above Equation (10)].

2.2 Directional Probability, \( P_D \)

At the time that a gas particle enters the surface box the solid particle has neither preferred direction nor location. Thus the probability that the initial motion is either outward (\( v_0 > 0 \)) or inward (\( v_0 < 0 \)) is 1/2, and the initial location within a box of length \( L \) is

\[ dP_{box} = \frac{|dx|}{L} \]  

(20)

However, our interest is in the directional probability at the time of first impact, \( P_D \), such that the collision itself is of an outward (\( v_0^{coll} > 0 \)) or inward (\( v_0^{coll} < 0 \)) type for a given \( (u_0, v_0) \) pair. Clearly all \( v_0 > 0 \) will result in outward collisions. In addition some \( v_0 < 0 \) [see Equation (4)] also lead to outward collisions if rebound occurs prior to being overtaken by the gas particle. It follows that inward collisions will occur for only the remaining \( v_0 < 0 \) particles.

If \(-u_0 < v_0 < 0\) the times required for the surface and gas particles to reach the bottom of the box are \((L - x)/|v_0|\) and \(L/u_0\), respectively. The initial locations
\[
\frac{\nu}{L} \leq 1 - \frac{\nu_0}{u_0}
\] 

(21)

therefore imply overtaking (inward, \(v_{0 \text{ coll}} < 0\)) and rebound (outward, \(v_{0 \text{ coll}} > 0\)) collisions respectively when \(v_0 < 0\). Combining the initial direction probabilities with Equation (20) for the box regions of Equation (21):

\[
P_{\text{inward}} = \begin{cases} 
0 & (v_0 < u_0) \\
\int_0^\nu \left( 1 - \frac{v_{0 \text{ coll}}}{u_0} \right) \frac{dv}{L} = \frac{1}{2} \left( 1 - \frac{\nu_0}{u_0} \right) & (-u_0 < v_0 < 0) \\
0 & (0 < v_0)
\end{cases}
\] 

(22)

\[
P_{\text{outward}} = \begin{cases} 
\frac{1}{2} & (v_0 < u_0) \\
\int_0^\nu \left( 1 - \frac{v_{0 \text{ coll}}}{u_0} \right) \frac{dv}{L} = \frac{\nu_0}{2u_0} & (-u_0 < v_0 < 0) \\
\frac{1}{2} & (0 < v_0)
\end{cases}
\] 

(23)

for any \(v_0\) the probability of a first collision is \(P_{\text{in}} + P_{\text{out}}\) so that

\[
\beta = \int_{-\infty}^{0} \left[ \int_{0}^{u_0} -u_0 \right] \left[ \int_{0}^{u_0} +u_0 \right] \frac{G(\nu_0)}{\nu_0} d\nu_0
\] 

(24)

and indicates \(P_1\) to be unity for the entire half plane as would be expected.

With energy constraints such as Equation (8) it is more convenient to collect the inward and outward contributions together to obtain the equivalent directional probability as in Equation (5). I.e.,
to indicate outward type collisions ($v_0 \text{ coll} > 0$) are more probable for a given $|v_0|$, and that the overall probability for any $|v_0|$ is in fact unity for a first collision (Figure 2). E.g., for the entire half plane Equation (24) may then be written as

$$P_2 \left( u_0, v_0 \right) = \begin{cases} \frac{1}{2} (1 + \frac{|v_0|}{u_0}) & (u_0 < v_0) \\ \frac{1}{2} (1 - \frac{|v_0|}{u_0}) & (0 < v_0 < u_0) \\ 0 & (-u_0 < v_0 < 0) \end{cases}$$

(25)

2.3 Second Collision Probability, $P_2$

A second collision is certain if $v_0 \text{ coll} \leq (1 - 1)/2u_0$ in view of Equation (14). Thus

$$P_2 \left( u_0, v_{coll} \right) = 1 \quad \left[ v_{coll} \leq \frac{u_0 - 1}{2} \right]$$

(26)

The remaining portion of the $(u_0, v_0)$ half plane involves only outward $v_0 > 0$ first collisions with $u_1 < 0$ as in Equation (15). Some of these collisions take place at a location, $x^*$ say, which permits the surface particle to then rebound from the bottom of its box and overtake the gas particle for a second collision. $P_2$ follows from the probability that
a first collision will occur within the allowable $x^*$ range consistent with the arbitrary surface particle location, Equation (20), and direction initially.

Allowable $x^*$ locations are those for which the gas particle exit time, $x*/(-u_1)$, exceeds the equivalent time for the surface particle, $(2L - x*)/v_1$. Using Equation (13), overtaking second collisions correspond to first collisions in the range

$$1 \geq \frac{x^*}{L} \geq \frac{2\left(\frac{v_0\cos \theta}{u_0} - (\mu - 1)\right)}{(\mu + 1)(1 + \frac{v_0\cos \theta}{u_0})} [\epsilon = (\frac{x^*}{L})_m, \text{any}]$$

(27)

Larger $(v_0 \cos \theta/u_0)$ imply a smaller range of allowable locations. For $(v_0 \cos \theta/u_0) = (\mu - 1)/2$ the collision may occur anywhere in the box, consistent with Equation (26). For $\mu > 3$ there is an allowable range for all $(v_0 \cos \theta/u_0) < \infty$; for $\mu < 3$ a sufficiently large $v_0 \cos \theta/u_0$ [$> (3\mu - 1)/(3 - \mu)$] cannot result in an overtaking second collision, consistent with the energy constraint in Equation (17) when $\epsilon_2 = 0$.

Since outward $(v_0 \cos \theta > 0)$ first collisions arise from both $v_0 \preceq 0$ the collision site differs for each initial direction. Equating the times for the two particles, the first collision occurs at

$$\frac{x^*}{u_0} = \begin{cases} \frac{K - x^*}{v_0} & (v_0 \cos \theta = v_0 > 0) \\ \frac{L - x^*}{v_0} + \frac{L - x^*}{v_0} & (v_0 \cos \theta = -v_0 > 0) \end{cases}$$

(28)
or
\[
\frac{x^*}{L} = \begin{cases} 
\left(\frac{V}{L}\right)\left(1 + \frac{V_0}{u_0}\right)^{-1} & \text{for } V_0 > 0 \ (V_0 \geq 0) \\
\left(\frac{2-V}{L}\right)\left(1 + \frac{V_0}{u_0}\right)^{-1} & \text{for } V_0 < 0 \ (V_0 \leq 0) 
\end{cases}
\] (29)

The probability that the collision occurs within \(d(x*/L)\) is then
\[
d_P = \frac{1}{2} \frac{V_{0x}^{-1}}{L} = \frac{1}{2} \left(1 + \frac{V_0}{u_0}\right) \frac{V_{0x}^{-1}}{L} \quad (V_0 \leq 0)
\] (30)

and the collisions are in the ranges, from Equation (29),
\[
0 < \frac{x^*}{L} < \left(1 + \frac{V_0}{u_0}\right)^{-1} \quad (0 \leq \frac{x}{L} \leq 1)
\] (31)

The limitation on initial locations \(x/L\) for the inward directed surface particles with velocities between \(0 > v_0/u_0 > -1\) corresponds to those particles that rebound and thereafter undergo an outward \(v_0 > 0\) interaction (Equation 21). Note that there is an implicit influence of the mass ratio, \(\mu\), in Equation (31) by virtue of the present interest in \((v_0\text{ coll})/u_0 > (\mu - 1)/2\).

The second collision probability, \(P_2\), follows from Equation (30) applied to those portions of the first collision ranges, Equation (31), which fall within the allowable range for a second collision, Equation (27).

For the direct outward \((v_0 > 0)\) interactions: \([(\mu - 1)/2 < v_0/u_0]
If \( \left( \frac{\mu^*}{L} \right)_{\text{in}} < \left( 1 + \frac{\nu_0}{\nu_0} \right)^{-1} \), i.e. \( \frac{\mu^*}{L} < \frac{\nu_0}{\nu_0} < \frac{3\mu-1}{4} \)

\[
\rho_2 = \int \frac{1}{2} \left( 1 + \frac{\nu_0}{\nu_0} \right)^{-1} \left( 1 + \frac{\nu_0}{\nu_0} \right) d\left( \frac{\nu^*}{\nu} \right) = \frac{3\mu-1-4\nu_0}{2(\mu+1)}
\]

(32a)

If \( \left( \frac{\mu^*}{L} \right)_{\text{in}} > \left( 1 + \frac{\nu_0}{\nu_0} \right)^{-1} \), i.e. \( \frac{3\mu-1}{4} < \frac{\nu_0}{\nu_0} \)

\[ \rho_2 = 0 \]

(32b)

For the rebound outward (\( \nu_0 < 0 \)) interactions; for \( (\mu - 1)/2 < 1 < |\nu_0|/\nu_0 \):

If \( \left( \frac{\mu^*}{L} \right)_{\text{in}} < \left( 1 + \frac{\nu_0}{\nu_0} \right)^{-1} \), i.e. \( \frac{\mu^*}{L} < \frac{|\nu_0|}{\nu_0} < \frac{3\mu-1}{4} \)

\[
\rho_2 = \int \frac{1}{2} \left( 1 + \frac{\nu_0}{\nu_0} \right)^{-1} \left( 1 + \frac{\nu_0}{\nu_0} \right) d\left( \frac{\nu^*}{\nu} \right) = \frac{1}{2}
\]

(33a)

If \( \left( 1 + \frac{\nu_0}{\nu_0} \right)^{-1} < \left( \frac{\mu^*}{L} \right)_{\text{in}} < 2 \left( 1 + \frac{\nu_0}{\nu_0} \right)^{-1} \), i.e. \( \frac{3\mu-1}{4} < \frac{|\nu_0|}{\nu_0} < \mu \)

\[
\rho_2 = \int \frac{1}{2} \left( 1 + \frac{\nu_0}{\nu_0} \right)^{-1} \left( 1 + \frac{\nu_0}{\nu_0} \right) d\left( \frac{\nu^*}{\nu} \right) = 2 \left( \frac{\mu - \frac{|\nu_0|}{\nu_0}}{\mu+1} \right)
\]

(33b)

If \( 2 \left( 1 + \frac{\nu_0}{\nu_0} \right)^{-1} < \left( \frac{\mu^*}{L} \right)_{\text{in}} \), i.e. \( \mu < \frac{|\nu_0|}{\nu_0} \)

\[ \rho_2 = 0 \]

(33c)
and for $(\mu - 1)/2 < |v_0|^2/u_0 < 1$:

If

\[ \left( \frac{\mu}{L} \right) u_0 < \left( \frac{1+ |v_0|}{u_0} \right)^{-1}, \text{ i.e. } \frac{\mu - 1}{2} < \frac{1001}{u_0} < \frac{3\mu - 1}{4} \]

\[
\rho_2 = \int_{1+ |v_0|}^{(1+ |v_0|)^{-1}} \frac{1}{2} \left( 1 + \frac{1601}{u_0} \right) \mu \left( \frac{\mu}{L} \right) \frac{3 \mu - 1}{4} \frac{3 \mu - 1}{4} \]

(34a)

If

\[ \left( \frac{1+ |v_0|}{u_0} \right)^{-1} < \left( \frac{\mu}{L} \right) u_0, \text{ i.e. } \frac{3 \mu - 1}{4} < \frac{1001}{u_0} < 1 \]

\[
\rho_2 = \int_{\frac{1+ |v_0|}{u_0}}^{\left( \frac{\mu}{L} \right) u_0} \frac{1}{2} \left( 1 + \frac{1601}{u_0} \right) \mu \left( \frac{\mu}{L} \right) \frac{3 \mu - 1}{4} \frac{3 \mu - 1}{4} \]

(34b)

2.4 Overall Second Collision Probability, $P_{II}$

The mass ratio $\mu$ provides the basis for specific integration intervals when evaluating $P_{II}$ from Equation (6) with (25). Note that

\[ \frac{3 \mu - 1}{3 \mu} > \mu > \frac{3 \mu - 1}{4} > \frac{\mu - 1}{2}, \text{ if } \mu > 1 \]

(35)

and

\[ \frac{3 \mu - 1}{4} \geq 1 \text{ if } \mu \geq \frac{5}{3} \]

(36)

\[ \frac{\mu - 1}{2} \geq 1 \text{ if } \mu \geq 3 \]
Then for $1 < \mu < 5/3$, $5/3 < \mu < 3$, and $3 < \mu$ respectively:

\[
P_\Pi = \begin{cases} 
\int_0^\infty \left[ \int_{-\infty}^0 \left( \frac{\mu}{\mu_0} \right)^n \right] \frac{1}{\mu_0} \left( \frac{1 + \frac{\nu_0}{\mu_0}}{2} \right) \frac{7 \left( \frac{\nu_0}{\mu_0} \right) - 12 \frac{\nu_0}{\mu_0} + 7}{28} \right] \frac{d\mu_0}{\mu_0} \\
+ \int_{\nu_0}^{\infty} \frac{\nu_0}{\mu_0} \left( \frac{9 - \frac{\nu_0}{\mu_0}}{28} \right) \frac{4 \left( \frac{\nu_0}{\mu_0} \right) - 3 \frac{\nu_0}{\mu_0}}{7} \right] \frac{d\mu_0}{\mu_0}
\end{cases}
\]

(37a)

prior to any energy cut-off considerations. The essential point is the association of a proper $P_2$ with each $P_0$ term in accord with the latter's direct or rebound first collision origin.

For the specific interaction of atomic oxygen and carbon, i.e., $\mu = 16/12$, the contributions from Equations (25), (26) and (32) - (34) into (37) gives (for \( \epsilon_1 = \epsilon_2 = 0 \)):

\[
P_\Pi = \int_0^\infty \left[ \int_{-\infty}^0 \left( \frac{\mu}{\mu_0} \right)^n \right] \frac{1}{\mu_0} \frac{7 \left( \frac{\nu_0}{\mu_0} \right) - 12 \frac{\nu_0}{\mu_0} + 7}{28} \right] \frac{d\mu_0}{\mu_0} \\
+ \int_{\nu_0}^{\infty} \frac{\nu_0}{\mu_0} \left( \frac{9 - \frac{\nu_0}{\mu_0}}{28} \right) \frac{4 \left( \frac{\nu_0}{\mu_0} \right) - 3 \frac{\nu_0}{\mu_0}}{7} \right] \frac{d\mu_0}{\mu_0}
\]

(38)

Similarly for molecular oxygen, $\mu = 32/12$:

\[
P_\Pi = \int_0^\infty \left[ \int_{-\infty}^0 \left( \frac{\mu}{\mu_0} \right)^n \right] \frac{1}{\mu_0} \frac{11 \left( \frac{\nu_0}{\mu_0} \right) - 12 \frac{\nu_0}{\mu_0} + 21}{44} \right] \frac{d\mu_0}{\mu_0} \\
+ \int_{\nu_0}^{\infty} \frac{\nu_0}{\mu_0} \frac{8 \left( \frac{\nu_0}{\mu_0} \right) - 3 \frac{\nu_0}{\mu_0}}{11} \right] \frac{d\mu_0}{\mu_0}
\]

(39)
E.g., for \( \mu = 4/3 \) the \( v_0 \) integrals follow from

\[
\int_{v_0}^{u_0} (1)(0) + \int_{v_0}^{u_0} \frac{u_0}{(1+2v_0)} + \int_{v_0}^{u_0} \frac{3u_0-4u_0(1+(2v_0^2))}{2(\mu+1)} + (0)^2 + (0)^2
\]

where for each interval for which \( v_0 > u_0(\mu - 1)/2 \) the explicit

\[
[(P_2 P_D)_{\text{direct}} + (P_2 P_D)_{\text{rebound}}]
\]

terms are indicated.

The energy constraints \( v_0 \geq v_{01} \) and \( v_0 < v_{02} \) affect only the integration limits as in Figure 1. Explicit integral results are developed in the next section in appropriate nondimensional form.
CHAPTER 3
INTEGRAL FORMS

3.1. General Form

An evaluation of \( P_I \) and \( P_{II} \) involves solely the integral form

\[
\int_{A}^{B} C_{ij} \, d\xi \, d\eta \quad [j = 0, 1, 2]
\]  \hspace{1cm} (40)

Introducing the nondimensional parameters,

\[
\bar{u}_0 = \frac{r}{\mu} u_0, \quad \bar{B} = \frac{r}{\mu} B, \quad \bar{C} = \frac{C}{(\mu \bar{T})^n}
\]
\[
\bar{D} = \frac{D}{\mu \bar{T}}, \quad \bar{A} = \frac{A}{(\mu \bar{T})^n}
\]
\[
\bar{u} = \frac{u}{T \alpha / \bar{T}} = \frac{r}{\mu}
\]

then

\[
\int_{A}^{B} \frac{\bar{C} \, \bar{D}}{\bar{A} \, \bar{B}} = \frac{U}{TM} (\mu \bar{T})^{n/2} \int_{\bar{u}_0}^{\bar{u}} e^{-\bar{u}_0^2} J_0 d\bar{u}_0
\]

Here

\[
J_0(\bar{u}) = \int_{\bar{u}_0}^{\bar{u}} e^{-\bar{u}^2} d\bar{u}_0 = \begin{cases} \frac{\pi}{2} (\text{erf} \bar{U} - \text{erf} \bar{L}) & [j = 0] \\ \frac{1}{2} (e^{-\bar{L}^2} - e^{-\bar{U}^2}) & [j = 1] \\ \frac{1}{2} \left( e^{-\bar{L}^2} - Ue^{-\bar{U}^2} + \frac{\pi}{2} \left[ \text{erf} \bar{U} - \text{erf} \bar{L} \right] \right) & [j = 2] \end{cases}
\]  \hspace{1cm} (43)
and
\[ U = \tilde{C} \tilde{u}_0 + \tilde{D}, \quad \Sigma = \tilde{A} \tilde{u}_0 + \tilde{B} \]

I.e.,
\[ \mathbf{I}_I (\tilde{u}_0) = \mathbf{K}_I (U, \tilde{u}_0) = \mathbf{K}_f (L, \tilde{u}_0) \quad (44) \]

where
\[ K_0 (U; \tilde{u}_0) = \frac{2}{\pi} \int \tilde{u}_0 e^{-\tilde{u}_0} \sin f U d\tilde{u}_0 \]
\[ K_1 (U; \tilde{u}_0) = -\frac{2}{\pi} \left( \frac{\pi L}{\tilde{u}_0} \right) \int \tilde{u}_0 e^{-\tilde{u}_0} d\tilde{u}_0 \]
\[ K_2 (U; \tilde{u}_0) = \left( \mu \frac{\pi}{\tilde{u}_0} \right) \int \tilde{u}_0 e^{-\tilde{u}_0} \left( \cos f U - \frac{2U}{\pi} \sin f U \right) d\tilde{u}_0 \quad (45) \]

Integrating by parts, and noting that
\[ \int Z(\tilde{u}_0) e^{-\tilde{u}_0^2} d\tilde{u}_0 = \frac{\pi^{1/2}}{\alpha^{1/2}} \int Z(\alpha^{1/2} \tilde{u}) e^{-\tilde{u}^2} d\tilde{u} \quad (46) \]

where
\[ y = \alpha^{1/2} (\tilde{u}_0 + \frac{\beta}{\alpha}) \quad (47) \]

then
\[ K_0 (\tilde{C}, \tilde{D}; \tilde{u}_0) = \frac{2}{\pi^{1/2}} \left[ -(\alpha \tilde{u}_0) N + \tilde{C} (M_0 + M_2) \right] \]
\[ K_1 (\tilde{C}, \tilde{D}; \tilde{u}_0) = -\left( \frac{\mu \pi}{\tilde{u}_0} \right) \frac{1}{\pi} \tilde{M}_1 \]
\[ K_2 (\tilde{C}, \tilde{D}; \tilde{u}_0) = \frac{\pi^{1/2}}{\pi^{1/2}} \left[ -N + \tilde{C} \tilde{M}_1 + 2(M_0 + \tilde{C} \tilde{M}_2) \right] \quad (48) \]

in which
\[ N(\tilde{C}, \tilde{D}; \tilde{u}_0) = \frac{\pi^{1/2}}{2} e^{-\tilde{u}_0^2} \sin f (\tilde{C} \tilde{u}_0 + \tilde{D}) \quad (49) \]
and the
\[ M_\epsilon (\vec{C}, \vec{D}; \vec{u}_0) = \int \vec{u}_0 e^{-[\vec{u}_0 \cdot (\vec{C} + \vec{D})]^2} d\vec{u}_0 \] (50)

are
\[ M_0 = \frac{\exp \left( -\frac{\vec{d}_1 \cdot \vec{u}_0}{1 + \vec{c}_1} \right)}{2 (1 + \vec{c}_1)^{1/2}} \pi^{1/2} \text{erfi} \] (51)
\[ M_1 = -\frac{\exp \left( -\frac{\vec{d}_1 \cdot \vec{u}_0}{1 + \vec{c}_1} \right)}{2 (1 + \vec{c}_1)^{1/2}} \left( e^{\vec{y}_1} + \frac{\vec{c} \vec{d}}{(1 + \vec{c}_1)^{1/2}} \pi^{1/2} \text{erfi} \right) \]
\[ M_2 = \frac{\exp \left( -\frac{\vec{d}_1 \cdot \vec{u}_0}{1 + \vec{c}_1} \right)}{4 (1 + \vec{c}_1)^{3/2}} \left( 2 \left[ \vec{y}_1 + \frac{2 \vec{c} \vec{d}}{(1 + \vec{c}_1)^{1/2}} \right] e^{\vec{y}_1} \right. \\
+ \left[ 1 + 2 \left( \frac{\vec{c} \vec{d}}{(1 + \vec{c}_1)^{1/2}} \right)^2 \right] \pi^{1/2} \text{erfi} \left( \vec{y} \right) \) (51)

Here
\[ \vec{y} = \frac{(1 + \vec{C}) \vec{u}_0 + \vec{C} \vec{D}}{(1 + \vec{C}_1)^{1/2}} \] (52)

The special limit of \(|\vec{D}| \to \infty \) then corresponds to
\[ M_0 = M_1 = M_2 = K_1 = 0 \]
\[ N = \mp \frac{\pi^{1/2}}{2} e^{-\vec{u}_0 \cdot \vec{L}} \]
\[ K_0 = \mp (1 + \vec{u}_0 \cdot \vec{L}) e^{-\vec{u}_0 \cdot \vec{L}} \]
\[ K_2 = \mp \frac{\mu \vec{r}}{2} e^{-\vec{u}_0 \cdot \vec{L}} \] (53)
and application to Equation (24) yields

\[
P_1 = \int_0^\infty \int_{-\infty}^\infty \frac{HG}{2} \, d\bar{v}_0 \, d\bar{u}_0 = \frac{1}{2} \left[ I_\infty(w) - I_0(0) \right]
\]

\[
= \frac{1}{2} \left[ K_0(0, \infty, x) - K_0(0, \infty, 0) - K_0(0, x, 0) + K_0(0, x, x) \right]
\]

\[
= \frac{1}{2} \left[ 0 - 0 + 1 + 1 \right] = 1
\]  

(54)

as a simple example.

3.2 First Collision Probability, \( \epsilon_1 > 0 \)

For a finite energy cut-off as in Equation (8) the first collision probability is (compare Equation 25a)

\[
P_1 = \left[ \int_0^{\sqrt{\frac{2\epsilon_1}{\mu^*}}} + \int_{\sqrt{\frac{2\epsilon_1}{\mu^*}}}^{\infty} \left( \int_0^{u_0} \left( \frac{1}{2} \left( \frac{u_0^2}{2} + \int_{u_0}^\infty \right) \right) \right) H G \, d\bar{v}_0 \, d\bar{u}_0 \right]
\]  

(55)

in which \( v_{01} = \sqrt{\frac{2\epsilon_1}{\mu^*}} - u_0 \) is the cut-off locus. Now

\[
\sqrt{\frac{2\epsilon_1 \sqrt{\frac{\mu^*}{\nu^*}}} \mu^*} = \sqrt{(1+\mu) \frac{\epsilon_1}{\mu^*}} \quad (= \epsilon_1^{\nu^*/\mu^*})
\]

\[
\sqrt{\frac{2\epsilon_1 \sqrt{\frac{\mu^*}{\nu^*}}} \mu^*} = \sqrt{(1+\mu) \frac{\epsilon_1}{\mu^*}} \quad (= \epsilon_1^{\nu^*/\mu^*})
\]  

(56)

Equation (55) in nondimensional form analogous to Equation (42) then reduces to

\[
P_1 = \frac{4}{\pi \nu^*} \left[ \int_0^{\infty} + \int_{\sqrt{\frac{2\epsilon_1}{\mu^*}}}^{\infty} \left( \int_0^{u_0} \frac{1}{2} \left( L_0 + \frac{L_1}{\nu^*} \right) + \int_{u_0}^{\infty} \right) \right] d\bar{v}_0 \, d\bar{u}_0
\]  

(57)
where
\[ L_j = \frac{-\tilde{u}_0}{\tilde{u}_0} e^{-\tilde{u}_0} \tilde{u}_0^{-\gamma} e^{-\tilde{u}_0^\gamma} \]  
(58)
and
\[ \tilde{u}_{0j} = \tau^*(\tilde{u}_0 + \epsilon_j^*) \]  
(59)

Equivalently, for the three regions a, b, and c as shown in Figure 3:

\[ (P_{II})_a = \left[ I_0(\frac{\epsilon_j^*}{2}) - I_0(0) \right]^{0,\infty}_{-\tau_0^* \tau_0^*} \]
\[ (P_{II})_b = \frac{1}{2} \left\{ \left[ I_0(\alpha) - I_0(\frac{\epsilon_j^*}{2}) \right]^{\tau_0^*}_{-\tau^*} + \left[ I_0(\alpha) - I_0(\frac{\epsilon_j^*}{2}) \right]^{\tau_0^*}_{-\tau_0^*} \right\} \]
\[ (P_{II})_c = \left[ I_0(\alpha) - I_0(\frac{\epsilon_j^*}{2}) \right]^{0,\infty}_{\tau_0^* \tau_0^*} \]
(60)

and the \( P_{II}(\epsilon^*, \tau^*) \) dependence may be interpreted as \( P_{II}(\mu, \epsilon_1, \tau) \) on the basis of Equation (56). Thus the corresponding energy and temperature ratio pairs for specific collision probability are

\[ (\frac{\tilde{e}_1}{\bar{e}_1})_b = \frac{1 + \mu_a}{1 + \mu_b} \frac{\tilde{e}_1}{\bar{e}_1} \]  
(61)
\[ \frac{\tilde{T}_b}{\bar{T}_a} = \frac{\mu_a}{\mu_b} \frac{\tilde{T}_a}{\bar{T}_a} \]

for distinct particles \( \mu_a \) and \( \mu_b \).

3.3 Overall Second Collision Probability, \( \epsilon_1 > 0 \), \( \epsilon_2 > 0 \)

\( P_{II} \) follows from Equation (37) modified for finite cut-off energy levels. For example, for \( \mu < 3 \), i.e. for \( dv_{0z}/du_0 > 0 \) for the \( \epsilon_2 \) cut-off locus defined by Equation (17), two types of integration regions
provide contributions to $P_{II}$. These will be referred to as Type A and B regions, and in each case the result will be seen to depend upon the relative cut-off energy ratio, $\varepsilon_2/\varepsilon_1$, that implies limitations on the relevant portions of the regions (Figure 4). Region boundaries consist of the $v_0$ and $v_0$ cut-off locii, and the natural boundaries implied by $P_2$ as in Equation (37), i.e., $v_0 = k_1u_0$ with $k_1 = (\mu - 1)/2$, $(3\mu - 1)/4$, 1 and $\mu$.

The contribution from Region A, for which $k_1 = (\mu - 1)/2$ depends upon the relative positions

$$\frac{1}{2} \left( \sqrt{\frac{\varepsilon_2}{\varepsilon_1}} + \frac{3-\mu}{1+\mu} \right) e^x = (\bar{u}_0) \leq (\bar{u}_0)_{\varepsilon_1} = \frac{e^x}{1+k_1} \quad (62)$$

or, in general for any $k_1$,

$$\frac{\varepsilon_1}{\varepsilon_1} \leq \left( \frac{2}{1+k_1} - \frac{3-\mu}{1+\mu} \right)^2 \left[ \left( \frac{\varepsilon_2}{\varepsilon_1} \right)_{k_1}, -\bar{y} \right] \quad (63)$$

and for $k_1$, $(\varepsilon_2/\varepsilon_1) \leq 1$. Thus

$$\left( \frac{P_{II}}{P_{II}} \right)_{\text{Region A}} = \frac{4}{\pi^2} \int_{2e^x}^{\infty} \int_{0}^{\bar{u}_0} \frac{1}{2} \left( \sqrt{\frac{\varepsilon_2}{\varepsilon_1}} + \frac{3-\mu}{1+\mu} \right) d\bar{u}_0 d\bar{u} \quad (64a)$$

and

$$\left( \frac{P_{II}}{P_{II}} \right)_{\text{Region A}} = \frac{4}{\pi^2} \left[ \int_{\bar{u}_0}^{\infty} \int_{\bar{u}_0}^{\infty} \frac{1}{2} \left( \sqrt{\frac{\varepsilon_2}{\varepsilon_1}} + \frac{3-\mu}{1+\mu} \right) d\bar{u}_0 d\bar{u} \right] \quad (64b)$$
The contribution from a Region $B_i$ requires consideration of
\((\varepsilon_2/\varepsilon_1)\) relative to both \((\varepsilon_2/\varepsilon_1)_i\) and \((\varepsilon_2/\varepsilon_1)_{i+1}\). Thus

\[
\text{(Region } B_i\text{)} = \frac{\gamma}{\pi h} \left[ \left( \varepsilon_{t_{i}} + \varepsilon_{t_{i+1}} \right) \frac{t_{i+1}}{t_{i}} \int_{0}^{\infty} \int_{\varepsilon_{t_{i}}}^{\varepsilon_{t_{i+1}}} \text{d}u_{0} \text{d}t_{0} \right] (65a)
\]

\[
\left( \frac{\varepsilon_t}{\varepsilon_1} < \frac{\varepsilon_{t_{i+1}}}{\varepsilon_1} \right)
\]

and depending upon \((\bar{u}_0)_c \leq (\bar{u}_0)_b_1\) either:

\[
\text{(Region } B_i\text{)} = \frac{\gamma}{\pi h} \left[ \left( \varepsilon_{t_{i}} + \varepsilon_{t_{i+1}} \right) \frac{t_{i+1}}{t_{i}} \int_{0}^{\infty} \int_{\varepsilon_{t_{i}}}^{\varepsilon_{t_{i+1}}} \text{d}u_{0} \text{d}t_{0} \right] (65b)
\]

\[
\left( \frac{\varepsilon_t}{\varepsilon_1} < \frac{\varepsilon_{t_{i+1}}}{\varepsilon_1} \right)
\]

\[
\text{and } \frac{\varepsilon_t}{\varepsilon_1} < E
\]

or:

\[
\text{(Region } B_i\text{)} = \frac{\gamma}{\pi h} \left[ \left( \varepsilon_{t_{i}} + \varepsilon_{t_{i+1}} \right) \frac{t_{i+1}}{t_{i}} \int_{0}^{\infty} \int_{\varepsilon_{t_{i}}}^{\varepsilon_{t_{i+1}}} \text{d}u_{0} \text{d}t_{0} \right] (65c)
\]

\[
\left( \frac{\varepsilon_t}{\varepsilon_1} < \frac{\varepsilon_{t_{i+1}}}{\varepsilon_1} \right)
\]

\[
\text{and } \frac{\varepsilon_t}{\varepsilon_1} > E
\]

and

\[
\text{(Region } B_i\text{)} = \frac{\gamma}{\pi h} \left[ \left( \varepsilon_{t_{i}} + \varepsilon_{t_{i+1}} \right) \frac{t_{i+1}}{t_{i}} \int_{0}^{\infty} \int_{\varepsilon_{t_{i}}}^{\varepsilon_{t_{i+1}}} \text{d}u_{0} \text{d}t_{0} \right] (65d)
\]

\[
\left( \frac{\varepsilon_t}{\varepsilon_1} < \frac{\varepsilon_{t_{i+1}}}{\varepsilon_1} \right)
\]
where

\[ S = \frac{P}{T} L_2 + \frac{Q}{T} L_1 + R L_0 \]

\[ E = \left( \frac{(3 \mu - 1) - (3 \xi \lambda_{i+1})}{(1 + \mu)(1 + \lambda_s)} \right)^2 \]

Also, \( \bar{u}_{oa} \) and \( \bar{v}_{o_1} \) are given by Equations (62) and (59), and the remaining limits are:

\[ \bar{u}_{o_2} = \frac{T^*}{3 - \mu} \left[ (3 \mu - 1) u_{o_0} - (1 + \mu) \epsilon_{\text{max}} \frac{\sqrt{E_2}}{E_1} \right] \]

\[ \bar{u}_{oc} = \frac{(1 + \mu) \epsilon_{\text{max}} \sqrt{E_2/E_1}}{(3 \mu - 1) - (3 \xi \lambda_{i+1})} \]

\[ \bar{u}_{oe} = \frac{(1 + \mu) \epsilon_{\text{max}} \sqrt{E_2/E_1}}{(3 \mu - 1) - (3 \xi \lambda_{i+1})} \]

(66)

Since

\[ 1 = \frac{\frac{\bar{E}_2}{E_1}}{\frac{\bar{E}_1}{E_1}} > \left( \frac{\bar{E}_1}{E_1} \right)^j \quad (j > 1) \]

it follows that for \((\bar{E}_2/E_1) > 1\) only Equations (64b) and (65d) need be considered. The minimum energy criterion entering into Equation (65) is that corresponding to the \( v_0 = \mu u_0 \) upper bound (see Equations (38) and (39), e.g.) for which

\[ \left( \frac{\bar{E}_2}{E_1} \right)^j = \left( \frac{\mu - 1}{\mu + 1} \right)^2 = \begin{cases} 0.0204 & (\mu = 4/3) \\ 0.0207 & (\mu = 8/3) \end{cases} \]

(68)

from Equation (63). For lesser energy ratios only Equations (64a) and (65a) need be considered and the resulting \( P_{II} \neq P_{II}^{(\bar{E}_2)} \).

For the special case of \((\bar{E}_2/E_1) > 1\) Equations (64b) and (65d) reduce to:
From Equations (38), (39), and (63), for the specific mass ratios of interest here:

<table>
<thead>
<tr>
<th>Region</th>
<th>$\mu = 4/3$</th>
<th>$\mu = 8/3$</th>
</tr>
</thead>
<tbody>
<tr>
<td>$B_1$</td>
<td>$k_i$</td>
<td>$P_i$</td>
</tr>
<tr>
<td>$B_1$</td>
<td>$\mu - \frac{1}{2} = \frac{6}{2} = 3$</td>
<td>$1$</td>
</tr>
<tr>
<td>$B_2$</td>
<td>$3 \mu - \frac{1}{4} = \frac{3}{2}$</td>
<td>$-5$</td>
</tr>
<tr>
<td>$B_3$</td>
<td>$1$</td>
<td>$0$</td>
</tr>
</tbody>
</table>
In general, $P_{II} = P_{II} (e^*, T^*, \mu, \bar{\epsilon}_2/\bar{\epsilon}_1)$, and

$$P_{II} = P_{IIA} + \sum_i P_{IIB_i} \quad (71)$$

The results for $P_I$, Equation (60), and $P_{II}$, Equations (64), (65), were programmed for numerical evaluation over a range of $\bar{\epsilon}_i$ and $\bar{T}$ for $\mu = 4/3$ and $8/3$. The computer listings are provided in Appendix 1.
4.1 Reaction Rates and Energy Cut-offs.

A reaction probability may be defined as

\[ e_R = \frac{\text{carbon atom flux from surface}}{\text{incident oxygen flux to surface}} \]  \hspace{1cm} (72)

and implies maximum \((e_R)_0\), \((e_R)_{O_2}\) values equal to 1 and 2 respectively for atomic and molecular beams which result in CO as the reaction product. The results of Liu's measurements are given in Table 1 and \((e_R)_0\) and \((e_R)_{O_2}/2\) may be compared with the collision interaction energy probabilities outlined in Chapter 3 as a measure of the energy constraint importance for effective interactions.

Liu used an oxygen beam at \(\theta = 45^\circ\) incidence on a graphite target at temperatures in the range 900 \(\leq T_S\), 900 \(\leq T_S\), \(\leq 1800\). From beam temperatures of \(T_g = 300, 1475, 1875, \) and 2200 \(^\circ\) K, he inferred the separate atomic and molecular reaction probabilities, \((e_R)_0\) and \((e_R)_{O_2}\), the dissociated case being based on the highest beam temperature (2200 \(^\circ\) K) for which the Knudsen source provided a 15% equilibrium dissociation level. Table 1 indicates \((e_R)_{O_2}\) to have a primary dependence on surface temperature, \(T_S\), for the appreciable gas temperature, \(T_g\), range considered.

The probability of an encounter with minimum interaction energies \(\bar{e}_1, \bar{e}_j\), is of the form \(P(\bar{e}_1, \bar{T}, \nu, \bar{e}_j/\bar{e}_1)\), with the normalized energy referred to \(T_g\) as in Equation (11).

For \(\theta = 45^\circ\) this implies
\[ \epsilon_i = (0.993 \times 10^{-3})(T_g, \text{K}) \epsilon_i \approx (10^{-3}T_g)\bar{\epsilon}_1 \text{ kcal/mole} \]  
(73)

and therefore provides some indication of cutoff energy level, \( \bar{\epsilon}_1 \), variations with \( T_g \) if constant \( \epsilon_i \) are indeed governing for any collision.

### 4.2 Collision Probability Results.

The fundamental first collision probability, \( P_1(\bar{\epsilon}_1, T, \mu) \) is shown in Fig. 5 for \( \mu = 4/3 \). Tabulations of the calculated results for all figures are included in Tables 2 through 5. The probability of achieving a given energy cutoff decreases with increasing \( \bar{\epsilon}_1 \), and decreasing \( T \) as follows from the implicit (decreasing \( T_g \), fixed \( T_s \)) and (decreasing \( T_s \), fixed \( T_g \)) bases respectively. Of interest first is the resulting range \( 1.0 > P_1 > 0.1 \) corresponding to \( \bar{\epsilon}_1 \approx 5 \), or \( \bar{\epsilon}_1 \approx 0(10 \text{ kcal/mole}) \) from Equation (73). Secondly, the probability variation near \( T \approx 0(1) \) proves to be quite appreciable but provides no basis for a peak reaction level as is evident from the measurements.

However, increasing \( T \) implies relatively larger \( v_0 \) magnitudes for all interactions, and Equation (15) then implies that a lesser number of subsequent collisions may then be attainable. I.e., the larger negative velocity imparted to gas particles results in a larger number of escapes prior to being overtaken by a rebounding surface particle. Fig. 6 indicates the effect on \( P_{11} \) of a minimum combined constraint for two collisions, \((\bar{\epsilon}_1, \bar{\epsilon}_2) = (0, 0)\), and suggests a resulting basis for the appearance of a peak probability if a multiple interaction is in fact a necessity to affect a reaction. Similarly, the \((1,0)\) and \((10^{-3}, 1)\) constant \( \bar{\epsilon}_1 \) locii indicate the scale of the interaction to be as small as \( 0(1) \) kcal/mole in such cases if \( P > 0.1 \).
Figures 7 and 8 show the peak development explicitly. For small $\bar{T}$, $\varepsilon_1$ dominates, whereas for large $\bar{T}$, $\varepsilon_2$ dominates. For the two collision model peaking occurs for $\bar{T} \sim O(1)$ for energy levels up to $\varepsilon_1 \sim O(10)$ kcal/mole. Clearly an increasing number of collisions implies sharper peaks and their occurrence at lesser $\bar{T}$.

Figures 9 and 10 indicate similar $P_1$ and $P_{II}$ probabilities for $\mu = 8/3$. The lesser $(\varepsilon_R)_{O_2}$ values result in larger $\varepsilon_1$ choices but qualitatively similar behavior. Larger mass ratio, however, implies a shift of peak probabilities to relatively larger $\bar{T}$ levels, as must be expected from the increasing demands on $v_0$ in the inequality Equation (15).

### 4.3 Energy Levels.

Some activation energies reported for the atomic oxidation of graphite are as follows:

<table>
<thead>
<tr>
<th>°K</th>
<th>kcal/mole</th>
</tr>
</thead>
<tbody>
<tr>
<td>230-287</td>
<td>0-10</td>
</tr>
<tr>
<td>293-373</td>
<td>6.5</td>
</tr>
<tr>
<td>387-473</td>
<td>10</td>
</tr>
<tr>
<td>488-573</td>
<td>11-13</td>
</tr>
</tbody>
</table>

For molecular oxidation the comparable activation energy is in the range 55-65 kcal/mole.

Figure 5 shows a first collision, constant $\varepsilon_1$, probability variation with $\bar{T}$ that is consistent with the increasing experimental reactivity up to $T \approx 1300^\circ$ K., and Figures 6-8 indicate marked reductions in probability for any accompanying second collision. The fit with expe-
riment of the calculated $\bar{\varepsilon}_1 \approx 2.5$ locus corresponds to 5.5 kcal/mole and is in reasonable agreement with the above reported energies\textsuperscript{14} for $\mu = 4/3$.

The several data groups obtained for distinct $T_g$ beams in the $\mu = 8/3$ molecular case imply first collision energy level requirements of magnitude (Fig. 10):

<table>
<thead>
<tr>
<th>$T_g$</th>
<th>$\bar{\varepsilon}_1$</th>
<th>$\bar{\varepsilon}_1$, kcal/mole</th>
</tr>
</thead>
<tbody>
<tr>
<td>300</td>
<td>30</td>
<td>9.</td>
</tr>
<tr>
<td>1475</td>
<td>9</td>
<td>13.</td>
</tr>
<tr>
<td>1875</td>
<td>7</td>
<td>13.</td>
</tr>
</tbody>
</table>

The approximately constant $\varepsilon_1$ values above confirm the expected shift of experimental data with $\bar{T}$. This appears to add some importance to the first collision energy criteria.

For both atomic and molecular oxidation it is evident that multiple collisions are required for consistency with a decreased reactivity at higher surface temperatures. This is not totally inconsistent as well, with suggestions of surface annealing and activation\textsuperscript{2,16} as competing processes that tend to dominate the reaction control at high and low surface temperatures respectively. A less reactive surface at higher temperatures necessarily should require either a greater first collision energy cut-off or represent a longer dwell time on the surface during which further collisions can take place. These results indicate a marked decrease in probability must be expected if a second collision is required. For example, for $\mu = 4/3$ and near the peak reaction probability ($T_s \approx 1300^\circ K$) temperature, $P_1(2.5) \approx 2.2$ and $P_{11}(2.5, 0) \approx 0.065$. 
4.4 Conclusion.

The probability of encounters between gas and surface particles of sufficient energy provides some of the qualitative features found for oxygen-graphite measured reaction probabilities. Specifically, increased reactivity with increasing surface temperature may be attributed to constant interaction energy constraints, and sharply decreasing reactivity at higher temperatures may be associated with the very much reduced probabilities that follow if any additional collisions are necessary.
APPENDIX 1

COMPUTER LISTINGS

Listings for the individual programs for one and two collision probabilities are given below. In each case only a single input data card is shown (preceding the last deck card, "/*/") as an example of the multiple input cards called for by the \( J = 1, 4 \) or \( J = 1, 26 \) counters.

The symbol equivalence is:

\[
\begin{align*}
EB & = \varepsilon_1 \\
EBR & = \frac{\varepsilon_2}{\varepsilon_1} \\
EBRI & = \left(\frac{\varepsilon_2}{\varepsilon_1}\right)_i \\
EBRI1 & = \left(\frac{\varepsilon_2}{\varepsilon_1}\right)_i + 1 \\
ES & = \varepsilon^* \\
T & = T^* \\
T S & = T a \\
U A & = (u_0) a \\
U C & = (u_0) c \\
U E & = (u_0) e \\
V 2 I & = \overline{v}_{o_2}(0) \\
V 2 R & = \text{slope of } \overline{v}_{o_2}(u_0) \\
V 2 & = \overline{v}_{o_2}(0) \\
V 2 & = \overline{v}_{o_2}(0) \\
X M & = \mu \\
X M J & = M_j (j = 0,1,2) \\
X K & = K_j (k = 0,1,2) \\
Y & = y \\
Y N & = N \\
T B & = \overline{T}
\end{align*}
\]

The two collision listing specifically provides for either of two XMU (= \( \mu \)) by including the appropriate constants [see table after Equation (70)] in cards labeled 20 to 28.
C ONE COLLISION, BASIC TEST, FINITE D, U

C DIMENSION X(K), P(8)
COMMON X, T

C INPUT
DO 5 J = 1, 4
4 READ 5, XM, Y, T
5 FORMAT (FIN.T,2F7.3)
ES = SORT ((1., XM) + CT)
ES2 = ES/2.
TS = 1./SORT (XM + TH)
ET = ES * TS

C 7 CALL EYEL (-TS, ET = 0., U = ES2*1. * P(1))
8 CALL EYEL (-TS, ET = 0., U = 0.1 * P(2))
9 CALL EYEL (-TS, ET = 0., U = ES2*2. * P(3))
10 CALL EYEL (+TS, U = ES2*1. * P(4))
11 CALL EYEL (+TS, U = ES2*2. * P(5))
12 CALL EYEL (+TS, U = 0.1 * P(6))
13 CALL EYEL (+TS, U = 0.2 * P(7))
14 CALL EYEL (+TS, U = 0.3 * P(8))

C P1 = XP(1) * XP(3) * XP(5)/2. * XP(7)
P1 = P1 - XP(2) + XP(4) * XP(6)/2. - XP(8)
35 PRINT 37, XM, Y, T, ES, TS
37 FORMAT (1X,5F20.7)
38 PRINT 40, XP(1), XP(3), XP(5), XP(7)
40 FORMAT (1X,4E20.7)
50 PRINT 40, XP(2), XP(4), XP(6), XP(8)

C 60 SUBROUTINE EYEL (A, B, C, D, U, L, X)
COMMON X
CALL BASIC (C, D, U)
XI = X(K)
CALL BASIC (A, R, U)
XI = XI - X(K)
RETURN
END

C 350 SUBROUTINE BASIC (G, H, U)
DIMENSION X(K)
COMMON X, T
X2 = SORT (1., G) * G 
X3 = (G*H)/X2
Y = X2*U + X3
YN = (X2*2.693) * EXP(-U) * FRF(G*U + H)
X4 = EXP(-(H*U)/(X2*X2))/X2
X5 = (1.,7774539)*EDF(Y)
XMO = X4 * X5
IF (Y.GE.0.0) GO TO 355
XM1 = (-X3) * XMO/X2
XM2 = (X5*(1.+X3*X3)*X4) / (2.*X2*X2)
GO TO 360
355 XM1 = (-X4)*XMO/X2
XM2 = 2.*X2 - X3 - Y*EXP(-Y*Y)
XM3 = (X2 + X5*X4) / (2.*X2*X2)
360 XSS = (1.,1281792) * (G*XMO + XM2) - YN * (1., U) * U
XK(2) = -(XM2 + 56.4967)/T
XK(1) = (1.,564896) / (TS*TS) * (G*XMO - YN - 2.*H*XM1 + G*XMO)
RETURN
END

//,5,* 04,000 + 000
SAMPLE OUTPUT, ONE COLLISION PROGRAM

<p>| | | | | |</p>
<table>
<thead>
<tr>
<th></th>
<th></th>
<th></th>
<th></th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>0.2666667E+01</td>
<td>0.4000000E+01</td>
<td>0.2000000E+01</td>
<td>0.3829709E+01</td>
<td>0.4330126E+00</td>
</tr>
<tr>
<td>0.3161365E-01</td>
<td>0.6347336E+00</td>
<td>-0.3638910E+00</td>
<td>-0.5646836E+00</td>
<td></td>
</tr>
<tr>
<td>-0.7241398E-01</td>
<td>0.6181731E+00</td>
<td>-0.3703180E+00</td>
<td>-0.5865594E+00</td>
<td></td>
</tr>
<tr>
<td>0.1040276E+00</td>
<td>0.8280277E-02</td>
<td>0.3213495E-02</td>
<td>0.2189082E-01</td>
<td>0.1374122E+00</td>
</tr>
</tbody>
</table>
C TWO COLLISIONS, XMU INPUT FOR 4/3, 8/7
C
DIMENSION XK(3), XP(1A), P2B(3), PI(3), QI(3), RI(3), XKI(4)
COMMON XK, TS
C INPUT
DO 65 J = 1, 26
4 READ 200, XMU, EB, TR, EBR
ES = SQRT((1.0 * XMU) * EB)
ES2 = ES/2.
TS = 1.0 / SQRT(XMU * TR)
ET = ES * TS
EB2 = SQRT(EBR)
UA = ((3.0 - XMU) / (1.0 - XMU) * EB2) * ES * .5
V2R = ((3.0 - XMU - 1.0) * TS) / (3.0 - XMU)
V2I = -(1.0 - XMU) * ES * TS * EB2 / (3.0 - XMU)
P2B(3) = 0.
PRINT 204, XMU, EB, TR, ES, TS, EBR
PRINT 208
C
5 Z1 = (XMU - 1.0) * TS * .5
IF (EBR .LE. 1.) GO TO 10
UC = (2.0 * ES * EBR) / (1.0 * XMU)
CALL EVEL (-TS, ET, V2R, V2I, UC, 1, XP(1))
CALL EVEL (-TS, ET, V2R, V2I, UA, 1, XP(2))
CALL EVEL (-TS, ET, V2R, V2I, UC, 2, XP(3))
CALL EVEL (-TS, ET, Z1, 0.0, 8.0, 1, XP(5))
CALL EVEL (-TS, ET, Z1, 0.0, UC, 1, XP(6))
CALL EVEL (-TS, ET, Z1, 0.0, 8.0, 2, XP(7))
CALL EVEL (-TS, ET, Z1, 0.0, UC, 2, XP(8))
PRINT 206, XP(1), XP(3), XP(5), XP(7)
PRINT 206, XP(2), XP(4), XP(6), XP(8)
DO 6 K1 = 1, 7, 2
6 XP(K1) = XP(K1) - XP(K1+1)
PRINT 206, XP(1), XP(3), XP(5), XP(7)
P2A = (XP(1) * XP(3) * XP(5) * XP(7)) * .5
PRINT 210, P2A
GO TO 15
C
10 Z2 = (2.0 * ES) / (1.0 * XMU)
CALL EVEL (-TS, ET, Z1, 0.0, 8.0, 1, XP(1))
CALL EVEL (-TS, ET, Z1, 0.0, Z2, 1, XP(2))
CALL EVEL (-TS, ET, Z1, 0.0, 8.0, 2, XP(3))
CALL EVEL (-TS, ET, Z1, 0.0, Z2, 2, XP(4))
PRINT 212, XP(1), XP(3)
PRINT 212, XP(2), XP(4)
XP(1) = XP(1) - XP(2)
XP(3) = XP(3) - XP(4)
PRINT 212, XP(1), XP(3)
P2A = (XP(1) * XP(3)) * .5
PRINT 214, P2A
C
15 IF (XMU.GE.3.) GO TO 65
IF (XMU.LE.1.667) GO TO 25
20 PI(1) = 1.4
PI(2) = 0.
PI(3) = 0.
QI(1) = -3.111.
QI(2) = -3.111.
QI(3) = 0.
RI(1) = 21.444.
RI(2) = 8.711.
RI(3) = 0.
KXI(1) = 5.166.
KXI(2) = 1.8.
KXI(3) = 8.73.
KXI(4) = 0.
IL = 2.
GO TO 28
C

25 \( P1(1) = 1/4 \)
\( P1(2) = -5/28 \)
\( P1(3) = 0 \)
\( Q1(1) = -3/7 \)
\( Q1(2) = 9/28 \)
\( Q1(3) = -3/7 \)
\( R1(1) = 9/28 \)
\( R1(2) = 0 \)
\( R1(3) = 4/7 \)

\[ \begin{align*}
P1(1) &= 1/4, \\
P1(2) &= -5/28, \\
P1(3) &= 0, \\
Q1(1) &= -3/7, \\
Q1(2) &= 9/28, \\
Q1(3) &= -3/7, \\
R1(1) &= 9/28, \\
R1(2) &= 0, \\
R1(3) &= 4/7, \\
K1(1) &= 1/6, \\
K1(2) &= 3/4, \\
K1(3) &= 1,0 \\
K1(4) &= 4/3, \\
IL &= 3 \\
\end{align*} \]

C

28 PRINT 216, P1
PRINT 216, Q1
PRINT 216, R1
PRINT 208, K1
PRINT 208

C

DO 60 I = 1, IL
EBRI = ((2/((1+K1(I)))) - ((3.0*(X1)/((1+X1)))) ** 2
EBRI1 = ((2/((1+K1(I)))) - ((3.0*(X1)/((1+X1)))) ** 2
TSDK = T5 * K1(I)
TSDK1 = T5 * K1(I-1)
ESK = ES / (1+K1(I))
ESK1 = ES / (1+K1(I-1))
UC = ((1.0*(X1)+ES*EB2) / ( (3.0*(X1)-1.0) - (3.0*(X1))*K1(I-1) )
UE = ((1.0*(X1)*ES*EB2) / ( (3.0*(X1)-1.0) - (3.0*(X1))*K1(I) ))
PRINT 216, EBR11, EBR, EBR1
PRINT 212, UE, UC

C

30 IF (EBR,GE,EBRI1) GO TO 35
CALL EYEL (TSDK, 0., TSDK1, 0., A, 0, +3*XP(1) )
CALL EYEL (TSDK, 0., TSDK1, 0., ESK, +3*XP(2) )
CALL EYEL (TSDK, 0., TSDK1, 0., A, 0, +2*XP(3) )
CALL EYEL (TSDK, 0., TSDK1, 0., ESK, +2*XP(4) )
CALL EYEL (TSDK, 0., TSDK1, 0., A, 0, +1*XP(5) )
CALL EYEL (TSDK, 0., TSDK1, 0., ESK, +1*XP(6) )
CALL EYEL (-TSDK, ET, TSDK1, 0., ESK, +3*XP(7) )
CALL EYEL (-TSDK, ET, TSDK1, 0., ESK, +2*XP(8) )
CALL EYEL (-TSDK, ET, TSDK1, 0., ESK, +1*XP(9) )
CALL EYEL (-TSDK, ET, TSDK1, 0., ESK, +1*XP(10) )
CALL EYEL (-TSDK, ET, TSDK1, 0., ESK, +1*XP(11) )
CALL EYEL (-TSDK, ET, TSDK1, 0., ESK, +1*XP(12) )
PRINT 218
GO TO 39

C

35 IF (EBR,LE,EBRI1) GO TO 45
CALL EYEL (TSDK, 0., TSDK1, 0., A, 0, +3*XP(1) )
CALL EYEL (TSDK, 0., TSDK1, 0., UC, +3*XP(2) )
CALL EYEL (TSDK, 0., TSDK1, 0., A, 0, +2*XP(3) )
CALL EYEL (TSDK, 0., TSDK1, 0., UC, +2*XP(4) )
CALL EYEL (TSDK, 0., TSDK1, 0., A, 1, +XP(5) )
CALL EYEL (TSDK, 0., TSDK1, 0., UC, +1, +XP(6) )
CALL EYEL (TSDK, 0., V2R + V2I, UC, +3*XP(7) )
CALL EYEL (TSDK, 0., V2R + V2I, UC, +3*XP(8) )
CALL EYEL (TSDK, 0., V2R + V2I, UC, +2*XP(9) )
CALL EYEL (TSDK, 0., V2R + V2I, UC, +2*XP(10) )
CALL EYEL (TSDK, 0., V2R + V2I, UC, +1*XP(11) )
CALL EYEL (TSDK, 0., V2R + V2I, UC, +1*XP(12) )
PRINT 220

38 PRINT 204, XP(1), XP(2), XP(3), XP(4), XP(5), XP(6), XP(7), XP(8), XP(9), XP(10), XP(11), XP(12)
PRINT 204, XP(1), XP(2), XP(3), XP(4), XP(5), XP(6), XP(7), XP(8), XP(9), XP(10), XP(11), XP(12)
DO 40 K1 = 1,11,2
40 XP(K1) = XP(K1-1) - XP(K1)
PRINT 204, XP(1), XP(3), XP(5), XP(7), XP(9), XP(11)
DO 40 K1 = 1,17,6
XP(K1) = XP(K1-1) * PI(I)
XP(K1+2) = XP(K1+2) * QI(I)
42 XP(K1+4) = XP(K1+4) * RI(I)
\[ P28(1) = XP(1) \times XP(3) \times XP(5) \times XP(7) \times XP(9) \times XP(11) \]

GO TO 60

C

45 EBR = ((3.0XU-1.)-(3.-XUJ)\times K(I+1))

EBR = (EBR / ((1.+XUJ)\times (1.+XUJ)))) \times 2.

PRINT 222, EBR

IF (EBR.GE.EBR) GO TO 50

CALL (TSK, 0., TSK1, 0.0, 8.0, 3, XP(1))

CALL (TSK, 0., TSK1, 0.0, 8.0, 2, XP(2))

CALL (TSK, 0., TSK1, 0.0, 8.0, 1, XP(3))

CALL (TSK, 0., TSK1, 0.0, 8.0, 0, XP(4))

CALL (TSK, 0., TSK1, 0.0, 8.0, 1, XP(5))

CALL (TSK, 0., TSK1, 0.0, 8.0, 0, XP(6))

CALL (TSK, 0., TSK1, 0.0, 8.0, 1, XP(7))

CALL (TSK, 0., TSK1, 0.0, 8.0, 0, XP(8))

CALL (TSK, 0., TSK1, 0.0, 8.0, 1, XP(9))

CALL (TSK, 0., TSK1, 0.0, UC, 2, XP(10))

CALL (TSK, 0., TSK1, 0.0, UC, 1, XP(11))

CALL (TSK, 0., TSK1, 0.0, UC, 0, XP(12))

CALL (TSK, 0., V2R, V21, UC, 3, XP(13))

CALL (TSK, 0., V2R, V21, UA, 3, XP(14))

CALL (TSK, 0., V2R, V21, UC, 2, XP(15))

CALL (TSK, 0., V2R, V21, UA, 2, XP(16))

CALL (TSK, 0., V2R, V21, UC, 1, XP(17))

CALL (TSK, 0., V2R, V21, UA, 1, XP(18))

GO TO 55

C

50 CALL EYEL (TSK, 0., TSK1, 0.0, 8.0, 3, XP(1))

CALL EYEL (TSK, 0., TSK1, 0.0, UC, 3, XP(2))

CALL EYEL (TSK, 0., TSK1, 0.0, 8.0, 2, XP(3))

CALL EYEL (TSK, 0., TSK1, 0.0, UC, 2, XP(4))

CALL EYEL (TSK, 0., TSK1, 0.0, 8.0, 1, XP(5))

CALL EYEL (TSK, 0., TSK1, 0.0, UC, 1, XP(6))

CALL EYEL (TSK, 0., V2R, V21, UC, 3, XP(7))

CALL EYEL (TSK, 0., V2R, V21, UC, 2, XP(8))

CALL EYEL (TSK, 0., V2R, V21, UC, 1, XP(9))

CALL EYEL (TSK, 0., V2R, V21, UC, 2, XP(10))

CALL EYEL (TSK, 0., V2R, V21, UC, 1, XP(11))

CALL EYEL (TSK, 0., V2R, V21, UC, 2, XP(12))

CALL EYEL (TSK, 0., V2R, V21, UC, 1, XP(13))

CALL EYEL (TSK, 0., V2R, V21, UC, 2, XP(14))

CALL EYEL (TSK, 0., V2R, V21, UC, 1, XP(15))

CALL EYEL (TSK, 0., V2R, V21, UC, 2, XP(16))

DO 59 K1 = 1,17,2

57 XP(K1) = XP(K1) - XP(K1+1)

PRINT 204, XP(1), XP(3), XP(5), XP(7), XP(9), XP(11)

PRINT 216, XP(13), XP(15), XP(17)

DO 59 K1 = 1,13,6

XP(K1) = XP(K1) \times P1(I)

XP(K1+2) = XP(K1+2) \times Q1(I)

P28(I) = XP(1) \times XP(3) \times XP(5) \times XP(7) \times XP(9)

P28(I) = XP(11) \times XP(13) \times XP(15) \times XP(17) \times P28(I)

C

60 CONTINUE

P28(I) = P2A \times P28(I) \times P2B(2) \times P2B(3)

PRINT 224, P28

CONTINUE

STOP
200 FORMAT (F10.7,F6.3,E8.1,E11.4)
204 FORMAT (1X, 6E20.7)
206 FORMAT (1X, 4E20.7)
208 FORMAT (1X, //)
210 FORMAT (1X, 'P2A EBR GREATER THAN ONE =*', E20.7/)
212 FORMAT (1X, 2E20.7)
214 FORMAT (1X, 'P2A EBR LESS THAN ONE =*', E20.7/)
216 FORMAT (1X, 3E20.7)
218 FORMAT (1X, 'P2B EBR LESS THAN EBR1*/)
220 FORMAT (1X, 'P2B EBR GREATER THAN EBR1*/)
222 FORMAT (1X, 'P2B EBR BRACKETED EBR1 =', E 20.7 )
224 FORMAT (1X, 'TWO COLLISION PROBABILITY = ', E 20.7////////)
226 FORMAT (1X, 'P2B = ', E20.7//)

END

C 300 SUBROUTINE EYEL (A,B,C,D,U,L,X1)
DIMENSION XK (3)
COMMON XK
CALL BASIC (C,D,U)
X1 = XK(L)
CALL BASIC (A,B,U)
X1 = X1 - XK(L)
RETURN
END

C 350 SUBROUTINE BASIC (G,H,U)
DIMENSION XK(3)
COMMON XX, IS
X2 = SQRT (1. + G*G)
X3 = (G*H)/X2
Y = X2^U * X3
YN = (1.8882269) * EXP(-(U*U)) * ERFG*U + H)
X4 = EXP(-(H*H)/(X2*X2))/(2.*X2)
X5 = (1.7724539)*ERF(Y)
XMO = X4 * X5
IF (Y*LE.12.) GO TO 355
XM1 = -(X3 * XMO)/X2
XM2 = (X5*(1.*2.*X3*X3)*X4) / (2.*X2*X2)
GO TO 360
355 XM1 = -(X4*EXP(-Y*Y)) + X3*XMO/X2
XM2 = 2.*(2.*X3 - Y) * EXP(-Y*Y)
360 XK(1) = (1.1283792) * (G*XMO + XM2) - YN + (1. + U*U))
XK(2) = -(XM2 + .5641896)/X5
XK(3) = ((.5641896)/(X5*X5)) * (G*XMO - YN - 2.*(H*XM1 + G*XM2))
RETURN
END

//G.SYSIN DD *
*1.3333333*0.0005+5.0E-01+0.0000E+00
/
**SAMPLE OUTPUT, TWO COLLISION PROGRAM**

<p>| | | | | | |</p>
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**TWO COLLISION PROBABILITY =** 0.1361213E-01
**TABLE 1**
MEASURED REACTION PROBABILITIES
(From Reference 4)

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<tr>
<th>$T_s$, °K</th>
<th>$T_g$, °K = 300</th>
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<th>1875</th>
<th>2200</th>
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<td>.0034</td>
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TABLE 2
ONE COLLISION PROBABILITY, $\mu = 4/3$

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<td>.711</td>
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<tr>
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<td>.854</td>
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### TABLE 3
TWO COLLISION PROBABILITY, \( \mu = 4/3 \)

<table>
<thead>
<tr>
<th>( T )</th>
<th>( P_{II}(\bar{\varepsilon}_1, 0) )</th>
<th>( P_{II}(\bar{\varepsilon}_1, 1) )</th>
<th>( P_{II}(10^{-3}, \bar{\varepsilon}_2) )</th>
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</thead>
<tbody>
<tr>
<td>( \bar{\varepsilon}_1 ) =</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>0</td>
<td></td>
<td></td>
<td></td>
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<td>0.25</td>
<td>.791</td>
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<td>.150</td>
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<td>.623</td>
<td>.239</td>
<td>.148</td>
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<td>2.0</td>
<td>.517</td>
<td>.207</td>
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<td>.106</td>
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<tr>
<td>( \bar{\varepsilon}_2 ) =</td>
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<td>3</td>
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<td>.0158</td>
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<td>.0205</td>
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TABLE 4
ONE COLLISION PROBABILITY, \( \mu = \frac{8}{3} \)

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<tr>
<th>( \bar{\epsilon}_1 )</th>
<th>( P_1 )</th>
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<tr>
<td>( \bar{\epsilon}_1 = 4.0 )</td>
<td>.00780</td>
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<tr>
<td>( \bar{\epsilon}_1 = 8.0 )</td>
<td>.0431</td>
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<tr>
<td>( \bar{\epsilon}_1 = 16.0 )</td>
<td>.137</td>
</tr>
<tr>
<td>( \bar{\epsilon}_1 = 32.0 )</td>
<td>.284</td>
</tr>
<tr>
<td>( \bar{\epsilon}_1 = 4.0 )</td>
<td>.379</td>
</tr>
<tr>
<td>( \bar{\epsilon}_1 = 8.0 )</td>
<td>.445</td>
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TABLE 5
TWO COLLISION PROBABILITY, $\mu = 8/3$

<table>
<thead>
<tr>
<th>$\bar{\varepsilon}_1$</th>
<th>$p_{11}(\bar{\varepsilon}_1, 0)$</th>
<th>$p_{11}(\bar{\varepsilon}_1, \bar{\varepsilon}_2)$</th>
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<tbody>
<tr>
<td>0</td>
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$\bar{\varepsilon}_1 = 10^{-3}$

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<th>$\bar{\varepsilon}_2$</th>
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Fig. 1. Energy Constraints for \((u_0, v_0)\) Pairs
Fig. 2. Directional Probability

\[ P_D = \frac{1}{2} \left( 1 + \frac{v_o}{u_o} \right) \]
Fig. 3. Integration Regions for First Collision Probability, $P_1$
Fig. 4. Integration Regions for Second Collision Probability, $P_{ii}$
Fig. 5. Single Collision Probability, $\mu = 4/3$, Experiment (O) $T_g = 2200^\circ$ K.
Fig. 6. One and Two Collision Bounds, $\mu = 4/3$
Fig. 7. Two Collision Probability, $\bar{\epsilon}_2 = 0, \mu = 4/3$, Experiment (O) $T_g = 2200^\circ$ K.
Fig. 8. Two Collision Probability, $\bar{\varepsilon}_2 = 1, \mu = 4/3$, Experiment (O) $T_g = 2200^\circ$ K.
Collision Probabilities, $\mu = 8/3$, Experiment ($O, \blacktriangle, \square$) for $T_g = (300, 1475, 1875)^o$ K.
Collision Probabilities, \( \mu = 8/3 \), Experiment (\( \bigcirc, \triangle, \square \)) for \( T_g = (300, 1475, 1875) \)° K.
REFERENCES


The probabilities of single and two successive gas surface collisions satisfying minimum interaction energy cutoffs are considered in the hard cube model sense. Comparison is made with atomic and molecular oxygen reaction probabilities with graphite corresponding to specific mass ratio examples of 4/3 and 8/3. Comparable peak probabilities result for multiple interactions but of much reduced sharpness with temperature for only a two collision constraint. Realistic activation energies are implied from a match of the energy cutoff probability model to the portion of the experimental data increasing with surface temperature. The rapid
decreasing reaction probabilities found experimentally at higher temperatures imply an appreciable increase in the required number of effective collisions.