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Structural Effects on Electrical Properties in Amorphous Semiconductors

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ABSTRACT

During this report period, work has been concentrated on a survey of the structure, electrical and magnetic properties of transition metal oxide-phosphate glasses and glasses in the $\text{As}_2\text{Te}_3-\text{As}_2\text{Se}_3$ system which possess electrical or magnetic device potential. Results of magnetic and electrical observations in several transition metal-phosphate glasses have revealed a high degree of magnetic and structural order. The pronounced influence of glass-glass phase separation and compositional segregation has also been noted.

Examination of a series of FeO-P$_2$O$_5$ glasses has revealed a glass forming range which extends to 80 mole % FeO. Susceptibility and Mössbauer spectroscopy results on these glasses are reported herein. Results on a CuO-P$_2$O$_5$ glass indicate that the conduction mechanism in this system is partially ionic and thermal treatments can change the ionic mechanism to electronic. These glasses exhibit microstructural features resembling crystals but no crystals have been detected using x-ray techniques.

Detailed studies of the $\text{As}_2\text{Te}_3-\text{As}_2\text{Se}_3$ system has shown switching behavior which can be controlled with compositional variation. The compositional variation of the microstructure is noted in micrographs included in this report.
Statement of Problem

The device potential in amorphous semiconducting materials is a largely unexploited area, despite extensive research in this area. This is the result of a lack of systematic structure-property oriented research in these materials. A fundamental understanding of the structural features of this class of materials will allow rational interpretation and control of relationships between glass preparation variables and important electrical and magnetic properties.

Electronic conduction in amorphous solids has become the subject of interest to a number of theoreticians and has been reviewed by Mort, Gubanov and numerous others. Virtually all of these works have begun with an assumption that amorphous solids are uniformly random, even though they recognize glasses are generally heterogenous. These theoreticians have developed analytical descriptions of several systems which have been experimentally verified in some cases. Attempts to extend this approach to microscopically heterogenous systems have had notably little success. There remains a considerable body of experimental results, including Hall and Seebeck coefficients, which are not rationalized by present theory.

Pearson has suggested that heterogenous structure in these materials may explain these anomalies if the separated phase is crystalline. It appears that heterogenous transport analysis similar to that of Volger or Bube is required to ascertain the transport behavior in each phase.

Another important anomaly between theory and observation concerns the theoretically predicted insensitivity of amorphous semiconductors to doping. Early experimental observations by Kolimets, et. al., conformed to the theoretical predictions, but recent work by Mackenzie clearly conflicts with the theory and the early work. It appears that the above anomalies are the result
of inadequate structural characterization, rather than fundamental theoretical problems.

Further evidence that structural heterogeneities lie at the root of these anomalies can be inferred from work by Kinser, et al., in $K_2O-P_2O_5-V_2O_5$ glasses. This work has shown that marked changes in dielectric behavior occur during thermal treatments customarily used to stress relieve glasses. These changes have been shown to be the result of structural changes involving precipitation of small amounts of crystals.

Wilson and Kinser have observed similar, but somewhat more complex, behavior in $FeO-P_2O_5$ glasses after thermal treatments corresponding to annealing. Electron spin resonance (ESR) results have shown the onset of structural changes during thermal treatment prior to their observation by other commonly employed techniques.

It is thus apparent that homogenous glasses, semiconducting or otherwise, are the exception rather than the rule.

**General Methodology**

The electrical and magnetic property changes accompanying structural modifications during glass processing are of prime interest in the present work. The above questions can only be answered with detailed structural characterization of representative glasses from the oxide and chalcogenide groups. The initial oxide glass examined was the $55FeO-45P_2O_5$ glass along with glasses from the $V_2O_5-P_2O_5$, $CuO-P_2O_5$, $TiO_2-P_2O_5$, and $MnO-P_2O_5$ systems. The initial chalcogenide glasses are from the $As_2Te_3-As_2Se_3$ system with $Ag-As-s$ glasses in preliminary stages of study.

Structural characterization of these systems is being accomplished using electron microscopy, Guinier-DeWolff x-ray, electron spin resonance spectroscopy, magnetic susceptibility, electron microprobe, dielectric
relaxation, Mössbauer spectroscopy and differential thermal analysis techniques.

In conjunction with the structural tools, it is necessary that the conductivity, switching behavior and Seebeck coefficient be monitored to allow direct structure-property correlations.

**Transition-Metal Phosphate Glasses**

Experimental results during this report period have justified our previous assumption that the interactions between ions in the amorphous metal oxide-phosphates are similar to those in crystalline oxides containing the same ions. Such interactions are principally negative superexchanges. Interesting magnetic observations have led to the conclusion that the magnetic susceptibility is a result of both interacting and non-interacting (partially isolated) ions. Such an assumption can be used to explain the unique downward curvatures of the reciprocal magnetic susceptibility versus temperature curves at low temperatures.

**Iron-oxide-phosphate**

Previously published data on the iron-oxide phosphate glass system has been restricted to 55 mole% FeO-45 mole% P₂O₅. Recently a glass forming study of the FeO-P₂O₅ system has shown that the glass forming range extends to 80 mole% FeO when the glasses are melted in an oxidizing atmosphere. Glasses above 80 mole% FeO spontaneously crystallized to α-Fe₂O₃ while intentionally devitrified glasses of compositions below 80% crystallized to FePO₄.

The reciprocal magnetic susceptibility versus temperature curves for four representative glasses are shown in Figure 1. These data indicate that all the glasses antiferromagnetically coupled and the magnetic interactions are attributable to negative superexchange. The downward curvature of the
curves are a direct consequence of the amorphous nature of the glasses.\textsuperscript{13} The matrix of interacting ions is responsible for the negative Curie temperature associated with the linear high temperature relation. Isolated ions contribute increasingly to the total susceptibility at low temperatures and, hence, cause the downward curvature of the reciprocal susceptibility. The existence of "isolated" ferric ions in the 55 mole% glass has previously been reported.\textsuperscript{14} Analyses of these data are continuing.

The ratio of the Curie temperature, $\Theta$, to the temperature at which the reciprocal susceptibility appears to curve downward, $T'$, gives an important contact between experimental results and theory. The fraction of interacting ions can be uniquely determined from experimentally observed values of $\Theta/T'$ since these have been shown to be essentially independent of the coordination number of the ions in an amorphous matrix.\textsuperscript{15} At present susceptibility data is incomplete and exact values of $T'$ have not been obtained. It is anticipated that a thesis containing a theoretical analysis of these glasses will be completed in the next report period.

It should be noted that no pronounced Néel temperature is observed in the susceptibility data for the iron glasses. Electron spin resonance data previously reported does indicate distinct onset of antiferromagnetic ordering however, these data clearly distinguish between $\text{Fe}^{2+}$-$\text{Fe}^{2+}$, $\text{Fe}^{2+}$-$\text{Fe}^{3+}$, and $\text{Fe}^{3+}$-$\text{Fe}^{3+}$ coupling. The bulk susceptibility of the iron phosphate glasses is therefore not related to one specific crystalline phase but to possibly three. The principal value of the susceptibility data for these glasses clearly seems to be as another method to determine the number of interacting ions.

Mössbauer effect spectroscopy of the 55 mole% FeO glass at 77° and 300°K has revealed that the interacting $\text{Fe}^{2+}$ and $\text{Fe}^{3+}$ ions are on octahedral sites. In addition, the $\text{Fe}^{3+}/\text{Fe}^{2+}$ ratio for this glass and for intentionally reduced
glasses are in good agreement with previously reported ESR data.\(^\text{14}\) Figures 3, 4, and 5 show the spectra at 300\(^\circ\)K for these glasses with known Fe\(^{3+}\) concentration. A spectrum is composed of two doublets, one from Fe\(^{3+}\) ions and one from the Fe\(^{2+}\). The "stick diagram" below the actual spectrum is a computer analysis of the experimental data. The ratio of height of the integrated intensity line represents the Fe\(^{3+}\)/Fe\(^{TOT}\) ratio.

Figures 6, 7, and 8 are the spectral parameters of the 55 mole\% FeO glass as a function of Fe\(^{3+}\)/Fe\(^{TOT}\). An analysis of these data is currently being conducted and will be included in a M.S. thesis to be published during the next report period.\(^\text{16}\)

**Manganese-oxide phosphate**

A study of the structural and magnetic properties of the manganese-phosphate glass system is currently in progress. As indicated in a preliminary study, the magnetic behavior of a phase separated 55-45 mole\% Mn\(_2\)O\(_5\)-P\(_2\)O\(_5\) glass appears to be a superposition of the magnetic behavior of the phosphate-rich phase, which contains isolated, paramagnetic manganese ions, and the magnetic behavior of the manganese-rich phase, which contains exchange-coupled manganese ions. The magnetic behavior of the latter phase is remarkably similar to that of crystalline MnO.

In order to further investigate the interrelationship of composition, microstructure, and magnetic behavior, glasses over the range \(x\) MnO\(_2\)-(1-\(x\)) P\(_2\)O\(_5\) with \(x = 20, 30, 40, 50, 55,\) and 60 have poured. The limit of the glass forming region of the manganese-phosphate system has been determined to be at \(x = 60\); beyond this the melt spontaneously crystallizes into Mn\(_2\)P\(_2\)O\(_7\). Crystalline Mn\(_2\)P\(_2\)O\(_7\), was identified by vacuum Guinier-DeWolff x-ray techniques in glasses with \(x = 55\) and \(x = 60\).

At present, the low temperature magnetic susceptibility of the as-cast
glass melts is being investigated using a permeability bridge and a nitrogen gas flow system. The bridge will be adapted for use with a hydrogen cryotip so that the magnetic susceptibility down to 200 K can be measured. It is anticipated that the results of these studies and variable temperature electron spin resonance studies on the as-cast samples will establish a relationship between the composition and magnetic behavior of the manganese-phosphate glass system. Further studies on heat treated samples will be conducted and a correlation between the magnetic behavior as observed in susceptibility and electron spin resonance experiments and the microstructure as determined by replication electron microscopy will be established.

**Copper Oxide-Phosphate**

Since the last report, attempts have been made to further characterize microstructural and electrical features of the CuO-P2O5 system. A glass of 55% CuO-45% P2O5 nominal composition was used in this investigation. D.C. electrical resistivity vs. time for varying applied potentials was studied using glasses of varying Cu2+/CuTOT ratios and thermal histories. Direct replication techniques were used to study microstructural features of freshly fractured surfaces etched in 10% HCl for varying lengths of time.

As mentioned in the last report, high activation energies in the range of 0.80 to 1.0 ev/atom were observed for glasses of the same nominal composition as that used in this experiment, although Cu2+/CuTOT ratios differed. These activation energies were higher than those observed for the FeO-P2O5 glasses (approximately 0.5 ev/atom). This fact, coupled with an observation of ionic conduction in a 50% CuO-50% P2O5 glass by Hansen, led to suspicion of a more complex conduction mechanism than originally suspected, although Hansen's activation energies of conduction were higher (in the range of 1.4 ev/atom). Activation energies for electronic conduction in the same range as
our CuO-P_2O_5 glasses have been reported for other glass forming systems by Mackenzie.\textsuperscript{8} To date, no significant time dependence of conduction has been observed for our specimens. Further experiments utilizing non-blocking electrodes and electromotive force measurements are being performed in order to further characterize the charge carrier. It should be pointed out that electronic conduction has been observed in a CuO-CaO-P_2O_5 glass.\textsuperscript{19} It is felt on the basis of this evidence that the conduction mechanism changes from ionic to electronic depending upon thermal history of the specimens and the amount of glass former (i.e., CaO, Na_2O, etc.) added to the melt. The glass former constricts the structure, thereby decreasing considerably the mobility of the cations.

\textbf{As_2Te_3-As_2Se_3 Glasses}

\textbf{Electrical}: The bulk D.C. conductivity data for the 40As_2Te_3:60As_2Se_3 to 80As_2Te_3:20As_2Se_3 glasses are plotted in Figure 9 as log $\sigma$ vs. 1/T. The activation energies derived from these plots are listed in Table 1. These data agree within experimental error with those of Roilos\textsuperscript{20} but not with the Kolomiets\textsuperscript{7} data.

Room temperature A.C. measurements from 0.2 kc to 2000 kc were taken on the series of glasses. Plots of tan $\delta$ vs. frequency show no dispersions, indicating that conditions for Maxwell-Wagner-Sillars heterogenous losses are not present in these glasses.

Behavior of both the D.C. conductivity and the tan $\delta$ factor of heat treated glasses should prove informative and this work is to be undertaken during the next report period.

\textbf{Structural}: Replica electron microscopy on the chalcogenides indicates that all the glasses studied are heterogenous and, thus their structure is sensi-
tive to thermal history.

An electron micrograph of a fresh fracture surface of arsenic triselenium, Figure 10, clearly shows the characteristic droplike structure of liquid-liquid phase separation. This phase separation in As$_2$Se$_3$ was surprising and, to our knowledge, has not been previously noted in the literature although this glass has been previously examined.

The other glass compositions examined also exhibit evidence of phase separation. This is evident with compositional changes and within different samples of the same composition. This is evidenced by the series of micrographs in Figures 11 through 15. Figure 11 shows large, well-formed drops, indicating well-separated phases. Figure 12 shows the glass in the beginning of the drop forming process with many small, irregular droplets. The glass in Figure 13 is in an advanced stage of phase separation as indicated by the droplets coalescence. Two coherent phases are shown in Figure 14, suggesting a spinodal process. Figure 15 shows an interesting structure which may be indicative of the formation of secondary segregations in the primary separated phases.

These E.M. studies indicate that these glasses are extremely structure sensitive to thermal history. The variations in droplet formation seen over a cross-section of an individual boule of glass shows that these diffusion controlled phase separation processes are sensitive to cooling rates. The different structures are noticeable between the rapidly cooled middle of the boule. The number and stages of phase separation processes occurring in these glasses conclusively point out the importance of control of melting variables such as maximum temperature, quench rate and subsequent heat treatment.

The variation on the stages of phase separation in the samples studied
makes accurately correlating the amount of phase separation, and therefore the structure, with composition difficult. The trend is to a larger degree of phase separation, i.e., more of the second phase, with increased tellurium content. Electron microscopic studies of glasses which have been heat treated under carefully controlled conditions are needed, and would be complimentary to the necessary electrical studies of heat treated samples.

Thus the next phase of our research on these glasses will be to study heat treated samples of the same composition. A report of these studies will be included in the next semi-annual report.

**UHF and Microwave Dielectric Properties**

The UHF and microwave dielectric properties of the arsenic-tellurium-selenium glasses have been determined at room temperature. The dielectric constant and loss tangent for a given glass composition were found to be essentially constant of the range 100-18,000 MHz. The dielectric constant varies from approximately 7 to 10 and increased with tellurium content. The loss tangent also increased with tellurium and had a maximum value of approximately $5.5 \times 10^{-2}$. The dielectric properties of these glasses compare favorably with those for other commonly used dielectrics for microwave integrated circuits, e.g., Y.I.G. and silicon. A paper on this phase of the program by Pearson, O'Reilly and Wilson is appended to this report.
Recommendations

1. Our principal recommendation is to continue the present program in its present direction to allow the synthesis of each of the results in a unified theory along the lines which are now clear in the chalcogenide system.

2. As in our previous recommendations, we continue to recommend the survey preparation of new glasses. We anticipate that the transition metal oxide-phosphate, silicate, borate and germanate survey presently in progress will be continued.

   It is also anticipated that results on a new system Ag-As-S will be most helpful in developing switching models. A detailed ternary phase diagram was recently published (30) and our analyses should be simplified with this as a basis. It is further anticipated that Cu and Au substituted in the above system will be quite informative from an atomistic and microstructural model point of view.

3. The Mössbauer studies should be continued to examine 57 Fe and 127 Te in each of the systems presently under examination using other techniques. This will significantly aid in atomic structure model development in these systems as an addition to the present tools.

4. We recommend that the far infrared "conductivity spectra" be obtained to facilitate in theoretical analysis of the conductivity/loss spectra. This will allow the loss behavior to be explicitly attributed (31) to each mechanism thus reinforcing both atomic and microstructural analyses.
References


References (Continued)


18. K. Hansen, private communication.


Table 1. Summary of DC conductivity from low field DC results and switching results.

<table>
<thead>
<tr>
<th>Composition</th>
<th>ΔE(ev) (low field)</th>
<th>ΔE(ev) switching analysis</th>
</tr>
</thead>
<tbody>
<tr>
<td>% As$_2$Te$_3$</td>
<td>% As$_2$Se$_3$</td>
<td>(low field)</td>
</tr>
<tr>
<td>80</td>
<td>20</td>
<td>0.43</td>
</tr>
<tr>
<td>70</td>
<td>30</td>
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</tr>
<tr>
<td>0</td>
<td>100</td>
<td>0.73</td>
</tr>
</tbody>
</table>
Figure 1.
Figure 2.
Figure 3.
Figure A.

**Folded Spectra of 50% Fe$^{3+}$/Fe Sample at 300 K**
Figure 5.

Folded spectra of 15% Fe$^{3+}$/Fe sample at 300 K.
Fig. 6. Isomer shift of the Fe$^{3+}$ and Fe$^{2+}$ ion as a function of the concentration ratio, Fe$^{3+}$/Fe.
Fig. 7. Line positions as a function of the concentration ratio, Fe$^{3+}$/Fe, for the four lines of the 77 K Kelvin sample.
Fig. 8. Quadrupole splitting of the $\text{Fe}^{3+}$ and $\text{Fe}^{2+}$ ion as a function of the concentration ratio, $\text{Fe}^{3+}/\text{Fe}$. 
Figure 9
Figure 10. $\text{As}_2\text{Se}_3$ Fracture Surface

Figure 11. 40 $\text{As}_2\text{Te}_3$ - 60 $\text{As}_2\text{Se}_3$
Fracture Surface from Center of Boule
Figure 12. 40 $\text{As}_2\text{Te}_3$ - 60 $\text{As}_2\text{Se}_3$ KOH Etched Fracture Surface from Peripheral Region of Boule

Figure 13. 70 $\text{As}_2\text{Te}_3$ - 30 $\text{As}_2\text{Se}_3$ KOH Etched Fracture Surface
Figure 14. 80 As₂Te₃ - 20 As₂Se₃ KOH Etched Fracture Surface

Figure 15. 40 As₂Te₃ - 60 As₂Se₃ Fracture Surface from same area as Figure 12.
Electrical, Thermal & Structural Properties
of As$_2$Te$_3$-As$_2$Se$_3$ Glasses

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Electrical conductivity, switching, D.T.A., and electron microstructures are presented for a series of the chalcogenide glasses As$_2$Te$_3$-As$_2$Se$_3$.

DTA observations were conducted on each of these glasses to determine the relative glass stability. Bulk threshold switching observations were conducted on a compositional series of glasses.

Replica electron microscopy observations indicate the presence of liquid-liquid immiscibility in the As-Te-Se ternary system. The evidence indicates that the As$_2$Te$_3$-As$_2$Se$_3$ join is not a quasi binary.

The immiscibility observed suggests that this system, like several analogous silicate ternary systems, is quite sensitive to thermal treatment.
I. INTRODUCTION

This paper reports results of electrical conductivity, switching, DTA and electron microstructural observations on a series of As$_2$Te$_3$-As$_2$Se$_3$ glasses. Rollos (1) and Kolomets and Nazarova (2) have previously reported electrical conductivity in this system but their results are not in agreement. The present electrical conductivity results generally agree with the observations of Rollos although some disparities are present. The DTA observations indicate that the high As$_2$Se$_3$ glasses are more stable glasses. All of the glasses in this system exhibit bulk threshold memory switching with the critical voltage for switching increasing with As$_2$Se$_3$ content. The switching stability increases with As$_2$Se$_3$ content although the switching voltage rises to around 15 kilovolts in As$_2$Se$_3$. The microstructure exhibits a liquid-liquid separation in all glasses with a matrix structural change with composition. The more heterogenous microstructures occur in As$_2$Se$_3$ rich glasses in which better switching stability is observed. The switching process observed in these glasses appears to occur by a thermal mechanism as previously reported by Warren (3) in a single composition As$_2$Se$_2$Te. This conclusion is further substantiated by the compositional and microstructural trends noted above.

II. EXPERIMENTAL

All glasses examined in this study were prepared by fusing the appropriate materials in evacuated Vycor ampoules at 800°C in a rocking furnace. After heating for one hour the ampoules were rapidly quenched in water. Six compositions of the As$_2$Te$_3$-As$_2$Se$_3$ glass system were prepared with As$_2$Te$_3$/As$_2$Se$_3$ ratios of 0.8, 0.7, 0.6, 0.5, 0.4 and 0.0. Reagent grade raw materials were employed after an initial study revealed essentially no purity effects between reagent materials and 99.9999 materials. All samples were formed into small
platelets by briefly remelting on a graphite plate and quenching with a second plate. Samples for all observations had an area of about 1 cm² and a thickness of about 0.1 cm with similar thermal history. No samples were subjected to further thermal treatments.

Conductivity measurements were conducted using silver electrodes in a guard ring configuration with applied fields of approximately 5 volts/cm. Switching observations were conducted using a special fixture with spring loaded electrodes with appropriate external current limiting circuitry (4). DTA observations were conducted on powdered samples in a Modified Fisher Model 260 thermal analyzer.

Electron microscopy observations were conducted on freshly fractured and on etched fracture surfaces utilizing platinum/rhodium shadowed indirect carbon replicas. The etchant used was a dilute aqueous KOH solution.

III. RESULTS

The DC conductivity at 21°C as a function of composition is plotted in Figure 1 with previous data of Roilos and Kolomiets. The general compositional trends are similar although pronounced differences exist between the Roilos/present work and Kolomiets results. An example of the conductivity versus inverse temperature behavior for a 50/50 glass is given in Figure 2 along with Roilos' data. It is clear that the temperature dependence of both samples is different although the room temperature conductivity is similar in this case. Results of the threshold switching behavior as a function of composition are given in Figure 3. These data indicate an increasing breakdown voltage at higher As₂Se₃ content with somewhat larger scatter from sample to sample in the higher As₂Se₃ glasses. Observations on a single sample of the higher As₂Se₃ glasses do not exhibit changes in switching voltage through numerous switching cycles; hence the stability of this material as a switch is considerably higher than glasses with high As₂Te₃ content.
The DTA observations, summarized in Table 1, indicate that the crystallization temperature of these glasses increases with As$_2$Se$_3$ content. As one might predict, the selenium is a much stronger glass former, thus, the high selenium glasses crystallize at higher temperatures, if in fact they crystallize at all.

Electron microstructures of five representative glasses presented in Figures 4-8 exhibit a primary liquid-liquid drop-like separation. The drop size and volume fraction is sensibly unchanged throughout the composition range including the As$_2$Se$_3$ glass. The drop phase in the 70/30 glass exhibits a peculiar structure unlike any known to the authors. A pronounced change in the matrix phase with composition is apparent, and the 60/40 matrix exhibits a structure similar to that previously attributed to a spinodal decomposition process. (5)

IV. DISCUSSION

The disagreement between the present work and Roilos' versus the early Kolomiets work is surprising and no rationalization of this disparity is advanced. The more subtle differences between Roilos and the present work are probably the result of differing thermal histories leading to microstructural differences. The difference in activation energy for the 50/50 glass is undoubtedly a consequence of compositional differences in the matrix phase resulting from different degrees of compositional segregation between the two laboratories.

The switching behavior exhibits first a smooth compositional trend along with an opposite trend in switching stability. The increase in switching voltage with selenium is a consequence of combined Joule heating/thermal conductivity trends with composition. The change in switch stability appears to be a consequence of increased microstructural heterogeneity with higher selenium content.
The phase separation observed in the \( \text{As}_2\text{Se}_3 \) glass indicates that a metastable immiscibility gap is present in the binary As-Se system and similar observations in tellurium containing glasses indicate that the immiscibility extends into the ternary system. Since the lines connecting phases in metastable equilibria do not lie in the \( \text{As}_2\text{Se}_3-\text{As}_2\text{Te}_3 \) join the system under examination is not a true quasibinary. This appears to be the only conclusion that can be reached from observations of constant phase fractions over the compositional range examined.

It thus appears that the As-Se-Te ternary system exhibits metastable immiscibility similar to the \( \text{Na}_2\text{O-B}_2\text{O}_3-\text{SiO}_2 \) system (6) and that the physical properties are similarly sensitive to thermal history.

CONCLUSIONS

1. Disagreement in electrical conduction observations among various investigators can be partially rationalized by the microstructure.

2. Switching stability in this system is enhanced by the presence of a more heterogenous microstructure.

3. The As-Se-Te ternary system exhibits metastable immiscibility on the join \( \text{As}_2\text{Te}_3-\text{As}_2\text{Se}_3 \).

ACKNOWLEDGEMENT

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**Figure Captions**

1. DC conductivity as a function of composition with data from Rollos (1) and Kolomiets and Nazarova (2).
2. DC conductivity as a function of inverse temperature with data from Rollos (1).
3. Threshold switching voltage as a function of composition. Large error bars are result of numerous samples—single samples exhibit smaller scatter in higher $\text{As}_2\text{Se}_3$ glasses.
4. Replica electron micrograph of stoichiometric $\text{As}_2\text{Se}_3$. Unetched fracture surface.
5. Replica electron micrograph of 40 $\text{As}_2\text{Te}_3$–60 $\text{As}_2\text{Se}_3$ glass. Unetched fracture surface.
6. Replica electron micrograph of matrix region of 60 $\text{As}_2\text{Te}_3$–40 $\text{As}_2\text{Se}_3$ glass. KOH etched fracture surface.
7. Replica electron micrograph of 70 $\text{As}_2\text{Te}_3$–30 $\text{As}_2\text{Se}_3$ glass. KOH etched fracture surface.
8. Replica electron micrograph of droplet/matrix interface region in 80 $\text{As}_2\text{Te}_3$–20 $\text{As}_2\text{Se}_3$ glass. KOH etched fracture surface.
Graph showing the relationship between log conductivity (cm·cm⁻¹) and 10³/T (°K⁻¹) for 50 As₂Te₃ - 50 As₂Se₃. The graph includes two lines: one labeled 'Roilos 0.57 ev' and another labeled 'Present work 0.44 ev'.
Breakdown voltage vs. composition for the As$_2$Te$_3$:As$_2$Se$_3$ system. Error bars indicate results of measurement of several samples. Single sample observations show little scatter.
4. $\text{As}_2\text{Se}_3$

FRACTURE SURFACE

5. 40 $\text{As}_2\text{Te}_3$ - 60 $\text{As}_2\text{Se}_3$

FRACTURE SURFACE
60 As$\textsubscript{2}$Te$\textsubscript{3}$ - 40 As$\textsubscript{2}$Se$\textsubscript{3}$
KOH Etched Fracture Surface

70 As$\textsubscript{2}$Te$\textsubscript{3}$ - 30 As$\textsubscript{2}$Se$\textsubscript{3}$
KOH Etched Fracture Surface
80 As$_2$Te$_3$ - 20 As$_2$Se$_3$
KOH Etched Fracture Surface
Summary

This paper considers the Ultra High Frequency and microwave dielectric properties of a class of amorphous semiconductors at room temperature. Complex permittivity measurements are presented for the glass system (x)As2Te3(1-x)As2Se3 for x = 0.4, 0.5, 0.6, 0.7 and 0.8, over the frequency range 100-18,000 MHz. The dielectric constant was found to be essentially frequency independent over the frequency range studied, and was found to decrease with decreasing Te content. The loss tangents observed compare favorably with values encountered for commonly used microwave substrates such as YIG or silicon.

Introduction

The last several years has evidenced an increase in research concerning the electrical and physical properties of amorphous semiconductors. In an attempt to test various density-of-state models for the amorphous materials, considerable experimental study has been conducted on the temperature dependence of the d.c. conductivity, the frequency dependence of the conductivity in the audio and high frequency range, and the optical absorption spectrum. The switching and electrical, thermal, and structural properties of the system (x)As2Te3(1-x)As2Se3 has been extensively investigated at Vanderbilt University. However, there has been no published research characterizing the high frequency dielectric properties of this glass system.

relations used for 100-500 MHz, to yield the complex permittivity of the sample.

10,000 MHz-18,000 MHz. For this frequency range, a K_0-band slotted waveguide, terminated with the sample and a variable short circuit was used. The rectangular sample, slightly smaller than waveguide dimensions, was placed just inside the variable waveguide short. A tightly fitting styrofoam spacer held the sample erect in the guide. Sample thickness was approximately 0.22". Measurements taken for this method were wavelength, position of the voltage minimum, and VSWR. The two-point method was again used to obtain the VSWR. A graphical technique given by Von Hippel was used to solve the equation for the wave impedance in the sample.

This latter method could also have been applied to X-band frequencies, but data was not taken in this range because samples of sufficient height and thickness were not available at the time of testing. Errors involved in the experimental setups include errors in slotted line null measurements, VSWR, and sample height and position. These led to a statistical error of about ± ten per cent.
Experimental

Preparation of Sample

The materials were prepared by fusing a mixture of As$_2$Te$_3$ and As$_2$Se$_3$ in evacuated Vycor ampoules at 800°C in a rocking furnace. After heating for one hour the molten material was quenched in water. The samples were then shaped and polished according to the experimental requirements.

Permittivity Measurements

100 Mhz-500 Mhz. For this frequency range the materials were shaped into circular disks approximately 0.24" in diameter and 0.03" thick and placed at the end of a shorted air-filled coaxial line. A Thurston Bridge (G. R. Model 1602-B) was used to determine the complex admittance of the sample. The sample arrangement was considered as a parallel plate capacitor and the corresponding field equations were used to calculate the complex permittivity from the measured admittance.

500 Mhz-2000 Mhz. In this frequency range a slotted coaxial line, terminated in the sample and a variable short, was used to measure null shift and VSWR. The VSWR was obtained by the two point method, since high VSWR values were encountered. The frequency was measured using a transfer oscillator and a 10 Mhz Electronic Counter. The same disk-shaped samples used in the frequency range 100-500 Mhz were used in this range. The null shift and VSWR data were substituted into the transmission line equations for input admittance to yield the sample admittance. The sample admittance was then used with the same capacitance.

Conclusions

Figures 1 and 2 show the compositional and frequency variation of the dielectric constant and loss tangent of the glass system.

The permittivities of the samples were found to be essentially frequency independent over the frequency range studied, with slight variations attributed to differences in experimental techniques. By increasing the Te composition, it was found that the dielectric constant can be varied from approximately 7.0 to 10.0, for the compositions studied. The values for dielectric constant found here compare favorably with data recently given for As$_2$Se$_3$ by Taylor, et al. The loss tangent was found to also increase with increasing Te content, and varied between approximately 0.02 and 0.07. Crevecoeur and de Wit have presented the dielectric properties of As$_2$Se$_3$ glass over the frequency range $10^2$ to $10^4$ Hz. In the U. H. F. and microwave regions, they give an average loss tangent of $10^{-4}$. This value may be considered as a low limit to the data given in the present paper, since it represents glass having no Te content. The dielectric properties of this glass system compares favorably with those for dielectric materials used for microwave substrates, such as YIG or silicon.

Acknowledgment

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References

See for example Proceedings of the 4th International Amorphous Semiconductor Conference, Ann Arbor, Michigan, August, 1971 (to be published).


FIG. 1. The frequency dependence of the dielectric constant for 
\((x)\text{As}_2\text{Te}_3(1-x)\text{As}_2\text{Se}_3\) for \(x=0.4, 0.5, 0.6, 0.7,\) and 0.8.
FIG. 1. The frequency dependence of the dielectric constant for 
\((x)\text{As}_2\text{Te}_3(1-x)\text{As}_2\text{Se}_3\) for \(x=0.4, 0.5, 0.6, 0.7,\) and 0.8.

FIG. 2. The frequency dependence of the loss tangent for 
\((x)\text{As}_2\text{Te}_3(1-x)\text{As}_2\text{Se}_3\) for \(x=0.4, 0.5, 0.6, 0.7,\) and 0.8.
Abstract presented to
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U. H. F. AND MICROWAVE DIELECTRIC PROPERTIES OF
AMORPHOUS SEMICONDUCTORS

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This paper considers the Ultra High Frequency and
microwave dielectric properties of some amorphous
semiconductors at room temperature. Complex permittivity
measurements are presented for the glass system \(x)As_2Te_3(1-x)As_2Se_3\)
for \(x=0.4, 0.5, 0.6, 0.7\) and 0.8, over the frequency range
100-18,000 MHz. The dielectric constant was found to vary
slightly with frequency and to decrease with decreasing As_2Te_3
content. This research was sponsored by the U. S. Army Research
Office-Durham.

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Recent interest in amorphous semiconductors has focused primarily on the electrical switching behavior of these materials. (1) Considerable experimental study has been conducted on the temperature dependence of the d.c. conductivity, the frequency dependence of the conductivity in the audio and high frequency range, and the optical absorption spectrum, in an attempt to test various density-of-states models for the amorphous materials. Very little work has been directed, however, toward characterizing the dielectric properties of these materials in the U. H. F. and microwave range.

This paper considers the compositional variation in the dielectric properties of the glass system \( (x)\text{As}_2\text{Te}_3(1-x)\text{As}_2\text{Se}_3 \) \( x=0.4, 0.5, 0.6, 0.7 \) and 0.8 over the frequency range 100-18,000 Mhz, at room temperature \( (25^\circ C) \).

The materials were prepared by fusing a mixture of \( \text{As}_2\text{Te}_3 \) and \( \text{As}_2\text{Se}_3 \) in a rocking furnace. After heating for several hours, the molten material was quenched in ice water.

Complex permittivity measurements were made by three techniques. For the frequency range 100-500 Mhz, the materials were shaped into circular disks and placed in an air-filled coaxial line. A Thurston Bridge (G.R. Model 1602-B) was used to determine the complex impedance of the sample, which was then used to determine \( \varepsilon' \) and \( \varepsilon'' \). In the frequency range 500 Mhz-2 Gzh, a slotted coaxial line was used for measurement of null shift and VSWR. For 2-6 Gzh this procedure was repeated using a greater precision coaxial slotted line. In the range 12-18 Gzh, rectangular shaped samples were used in conjunction with a Ku-band slotted waveguide. Permittivity was obtained by measuring VSWR, wavelength, and position of voltage minimum and calculating the proper order solution to a complex transcendental equation. (2)

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The average complex permittivities measured over the entire frequency band are given in Table I.

**TABLE I. Average Complex Permittivity Variation with Composition.**

<table>
<thead>
<tr>
<th>Composition</th>
<th>Permittivity</th>
</tr>
</thead>
<tbody>
<tr>
<td>80% As$_2$S$_2$-20% As$_2$Se$_3$</td>
<td>10.0-10.4</td>
</tr>
<tr>
<td>70% As$_2$-30% As$_2$Se$_3$</td>
<td>9.1-10.3</td>
</tr>
<tr>
<td>60% As$_2$-40% As$_2$Se$_3$</td>
<td>8.7-10.25</td>
</tr>
<tr>
<td>50% As$_2$Te$_3$-50% As$_2$Se$_3$</td>
<td>8.4-10.2</td>
</tr>
<tr>
<td>40% As$_2$Te$_3$-60% As$_2$Se$_3$</td>
<td>7.2-10.4</td>
</tr>
</tbody>
</table>

Slight variations of permittivity with frequency were noted and the dielectric constant decreased as the As$_2$Te$_3$ content decreased. The values for dielectric constant found here compare favorably with data recently given for As$_2$Se$_3$ by Taylor et al. (3).

Variations of permittivity with frequency can be attributed to variations in experimental techniques. Errors involved in the experimental setup include errors in slotted line null measurements, VSWR, and sample height and position. These lead to a statistical error of ±10%. All measurements were within this statistical error.

**References**

1. See for example PROCEEDINGS OF THE 4TH INTERNATIONAL AMORPHOUS SEMICONDUCTOR CONFERENCE, Ann Arbor, Michigan, August 1971 (to be published).
