Because of our limited supply, you are requested to return this copy WHEN IT HAS SERVED YOUR PURPOSE so that it may be made available to other requesters. Your cooperation will be appreciated.

NOTICE: WHEN GOVERNMENT OR OTHER DRAWINGS, SPECIFICATIONS OR OTHER DATA ARE USED FOR ANY PURPOSE OTHER THAN IN CONNECTION WITH A DEFINITELY RELATED GOVERNMENT PROCUREMENT OPERATION, THE U. S. GOVERNMENT THEREBY INCURS NO RESPONSIBILITY, NOR ANY OBLIGATION WHATSOEVER; AND THE FACT THAT THE GOVERNMENT MAY HAVE FORMULATED, FURNISHED, OR IN ANY WAY SUPPLIED THE SAID DRAWINGS, SPECIFICATIONS, OR OTHER DATA IS NOT TO BE REGARDED BY IMPLICATION OR OTHERWISE AS IN ANY MANNER LICENSING THE HOLDER OR ANY OTHER PERSON OR CORPORATION, OR CONVEYING ANY RIGHTS OR PERMISSION TO MANUFACTURE, USE OR SELL ANY PATENTED INVENTION THAT MAY IN ANY WAY BE RELATED THERETO.

Reproduced by
DOCUMENT SERVICE CENTER
KNOTT BUILDING, DAYTON, OHIO

UNCLASSIFIED
AN EXPERIMENTAL MOLECULAR BEAM INVESTIGATION
OF THE SCATTERING OF MOLECULES FROM SURFACES

by

F. C. HURLBUT

SERIES NO. .......................... 20
ISSUE NO. .......................... 98
DATE ..................... DECEMBER 1, 1953
FLUID FLOW AND HEAT TRANSFER AT LOW PRESSURES AND TEMPERATURE

AN EXPERIMENTAL MOLECULAR BEAM INVESTIGATION OF THE SCATTERING OF MOLECULES FROM SURFACES

By

F. C. HURLBUT

FACULTY INVESTIGATORS:

S. A. Schaaf, Associate Professor of Engineering Science
R. M. Drake, Jr., Associate Professor of Mechanical Engineering
L. Talbot, Assistant Professor of Mechanical Engineering

Approved by:

Note: This report was condensed from a Ph.D. thesis, University of California Berkeley, 1953.
# TABLE OF CONTENTS

<table>
<thead>
<tr>
<th>Section</th>
<th>Page</th>
</tr>
</thead>
<tbody>
<tr>
<td>Abstract</td>
<td>5</td>
</tr>
<tr>
<td>Nomenclature</td>
<td>6</td>
</tr>
<tr>
<td>1.0 Introduction</td>
<td>10</td>
</tr>
<tr>
<td>1.1 General</td>
<td>10</td>
</tr>
<tr>
<td>1.2 Low Density Flow Investigation as a Technique of Gas-Surface Interaction Study</td>
<td>10</td>
</tr>
<tr>
<td>1.3 Detailed Investigations of Gas-Surface Interactions</td>
<td>12</td>
</tr>
<tr>
<td>1.4 Objectives of the Experiment</td>
<td>14</td>
</tr>
<tr>
<td>2.0 Description of the Apparatus</td>
<td>15</td>
</tr>
<tr>
<td>2.1 Introductory Remarks</td>
<td>15</td>
</tr>
<tr>
<td>2.2 The Physical Apparatus</td>
<td>16</td>
</tr>
<tr>
<td>2.3 The Molecular Beam Detection System</td>
<td>18</td>
</tr>
<tr>
<td>2.3.1 Preliminary Considerations</td>
<td>18</td>
</tr>
<tr>
<td>2.3.2 The Ionization Gage Detector</td>
<td>22</td>
</tr>
<tr>
<td>2.3.3 The Emission Regulator</td>
<td>24</td>
</tr>
<tr>
<td>2.3.4 The Difference Amplifier</td>
<td>27</td>
</tr>
<tr>
<td>3.0 The Experimental Program</td>
<td>29</td>
</tr>
<tr>
<td>3.1 Test Surfaces and Geometries</td>
<td>29</td>
</tr>
<tr>
<td>3.1.1 Cold Rolled Steel-Polished</td>
<td>29</td>
</tr>
<tr>
<td>3.1.2 Cold Rolled Steel-Etched</td>
<td>30</td>
</tr>
<tr>
<td>3.1.3 Aluminum-Polished</td>
<td>30</td>
</tr>
<tr>
<td>3.1.4 Unpolished Glass Surface</td>
<td>31</td>
</tr>
<tr>
<td>3.1.5 Polished Glass Surface</td>
<td>32</td>
</tr>
<tr>
<td>3.2 The Experimental Data</td>
<td>32</td>
</tr>
<tr>
<td>3.3 A Discussion of Errors</td>
<td>35</td>
</tr>
<tr>
<td>3.4 A Physical Interpretation of the Data</td>
<td>37</td>
</tr>
</tbody>
</table>
4.0 A Quantitative Examination of the Data .................. 40
5.0 Summary ................................................. 54
References .................................................... 57
Appendix A ..................................................... 61
Appendix B ..................................................... 63
Appendix C ..................................................... 65
Table I. Coefficients of Specular Reflection $f$ ............... 68
Table II. Thermal Accommodation Coefficients $\alpha$ .......... 69
Table III. Summary of Molecule - Surface Scattering Investiga-
		
tions ......................................................... 70

List of Figures

1. Geometry of Scattering Experiment
2. Schematic of Molecular Beam Model 2 Vacuum System
3. Molecular Beam Apparatus
4. Optical Bench with Beam Forming and Detecting Equipment
5. Schematic of Source Gas System
6. Schematic of Basic Rotation Elements of Detector Mechanism
7. Schematic for Estimation of Beam Pressure Increments in Detector
   Gage
8. Ionization Gage Molecular Beam Detector
9. Schematic of Molecular Beam Detection System
10. Block Diagram of Emission Regulator
11. Emission Regulator Circuit for Molecular Beam Detector
12. Power Supplies for Emission Regulator
13. Molecular Beam Detector Difference Amplifier
14. Power Supplies for Difference Amplifier
15. Experiment Beam Geometries
16. Test Surfaces and Holders

17. Raw Data Record

18, 19, 20, 21. Scattering Plots - Polished Steel Surfaces

22, 23, 24, 25. Scattering Plots - Polished Aluminum Surface

26, 27, 28, 29, 30, 31, 32. Scattering Plots and Representation of Smooched Scattering Shells - Unpolished Glass Surface.

33, 34, 35, 36, 37, 38, 39, 40. Scattering Plots - Polished Glass Surface

41. Coordinate System - Incident Momentum Calculations

42. Coordinate System - Issuing Momentum Calculations

43. Plots of $\frac{Gr(\Theta_i)}{m m \Omega_r(\Theta_i) V_m V_m}$ as a function of $\Theta_i$ for $s = 0.1, 1.0, 2.0, 10.0$

44. Plots of $gr(\Theta_i)$ as a Function of $\Theta_i$

45. Plot of $f$ as a Function of $s$ for Nitrogen on Unpolished Glass

46. Plot of $\int'$ as a Function of $b$

47. Plots of $\frac{G_1(\Theta_i)}{m m V_m}$ as a Function of $\Theta_i$ for $s = 0.1, 1.0, 2.0, 10.0$

48. Plots of $f$ as a Function of $\Theta_0$ for $s = 0.1, 1.0, 2.0, 10.0$

49. Diagram of Unit Hemisphere Area Division for Issuing Momentum Calculation
A molecular beam apparatus was used to investigate the scattering of air and \( \text{N}_2 \) molecules from surfaces of polished low carbon steel, etched low carbon steel, polished aluminum, unpolished window glass, and polished window glass. A beam of molecules was directed against the surface at representative angles of incidence. Polar flux distributions of the scattered molecules were measured by a movable detector. An ionization gage beam detection system was devised to accomplish this measurement.

Spatial polar plots of these measured flux distributions were found to approximate closely the form of the cosine scattering distribution for the cases of the steel and aluminum surfaces. Similar plots in the case of the glass surfaces were found to have well defined but small bulging deviations from the cosine scattering shape on the side away from the incident beam.

In order to illustrate one aerodynamic application of scattering data obtained by molecular beam techniques, a procedure was devised for the computation of the coefficient of specular reflection, \( f \) (\( f \) being that fraction of the total incident tangential momentum which is transferred to a surface in interaction with a flowing gas) and values of this coefficient were calculated for the case of an unpolished glass surface oriented tangentially to the direction of mass motion in a free molecule flow. These values were calculated as a function of the parameter \( s \), the molecular speed ratio (the ratio of the gas mass speed to the most probable thermal speed of the gas molecules). If all scattered molecules are assumed to have the distribution in speed in a particular direction which they would have had if they issued from a gas in equilibrium at the temperature of the surface, \( f \) takes on values of \( \sim 0.97 \) as \( s \) approaches 0 to \( \sim 0.99 \) at \( s = 10 \). Alternatively, if those molecules contributing to the deviations from cosine scattering are assumed to issue from the surface with their incident speed while the remainder are assumed to be diffusely scattered, (a modified Maxwell model), \( f \) takes on values of \( \sim 0.97 \) as \( s \) approaches 0 to \( \sim 0.93 \) at \( s = 10 \). These values do not correspond closely to the published values for air on glass of 0.89 (\( s \sim 0 \)).
\[ A = \text{tube or orifice area. (Subscripts 1, 2, 3, etc. denote association of } A \text{ with regions of the beam as required in section 2.31 and 2.32. Subscripts } a, b, \text{ etc., denote the } a \text{ or } b \text{ end of a certain tube as required in sections 2.31 and 2.32).} \]

\[ A = \text{constant} = \frac{\alpha^2}{\sqrt{\pi}} \text{ (Subscripts } r \text{ and } f \text{, if used, relate } A \text{ to reflected molecules or to molecules issuing from a fictitious gas respectively as described in section 4.2).} \]

\[ A(\theta; \theta', \phi') = \text{reflected beam signal amplitude in arbitrary units for incident molecules having direction } \theta \text{ and detector axis having direction } \theta' \phi'. \]

\[ A_D = \text{constant representing the diameter of the assumed cosine scattering shell.} \]

\[ A_s(\theta; \theta', \phi') = \text{beam signal amplitude differences defined by } A_s(\theta; \theta', \phi') = A(\theta; \theta', \phi') - A_D \cos \phi'. \]

\[ A_1, A_2 = \text{radii of tubes or orifices} \]

\[ b = s \sin \theta_f \]

\[ C = \text{magnitude of the total molecular velocity (molecular speed) (subscripts } i \text{ and } r \text{ denote incident and reflected molecules respectively. Subscript } f \text{ denotes molecules issuing from a fictitious gas as described in section 4.2).} \]

\[ C_x, C_y, C_z = \text{components of the total molecular velocity.} \]

\[ \overline{C} = \text{mean molecular speed referred to coordinate system fixed to the flowing gas } \sqrt{\frac{2 K T}{\pi M}} = \frac{2}{\sqrt{\pi}} \overline{C_r} \text{ (Subscripts } 1, 2, 3, \text{ etc., denote specific mean speeds as required in section 2.31 and 2.32).} \]

\[ \overline{C_r} = \text{mean molecular speed of molecules issuing diffusely from a surface } \frac{3}{4} \beta_r \]

\[ E_i, E_r, E_w = \text{molecular energy fluxes incident, reflected and diffusely reflected.} \]

\[ F_u = \text{Maxwellian distribution function, Eq. 4.3.} \]

\[ f = \text{coefficient of specular reflection, section 1.2 and Eq. 4.1} \]

\[ G_i = \text{magnitude of the net tangential momentum incident on unit surface per unit time.} \]
G_t(θ_t) = magnitude of the net tangential momentum incident on unit surface per unit time and unit solid angle having direction θ_t, Eq. 4.25.

G_r = magnitude of the net tangential momentum imparted by the surface to the issuing molecules per unit area per unit time, Eq. 4.1 (the addition of the subscript 2 indicates the application of assumption E-2 in the calculation of Gr).

G_r(θ_t) = issuing momentum function defined Eq. 4.15.

g_r(θ_t) = function defined in Eq. 4.10.

I_o, I_1 = modified Bessel function of zero and first order.

k = Boltzmann's constant

m = molecular mass

N = number of beam molecules entering gage per unit time.

N_i = number of molecules of all velocities striking unit surface area per unit time, Eq. 4.2.

N_i(C_i, θ_i, φ_i) = number of incident molecules per unit speed range and solid angle having the speed C_i and incident direction θ_i, φ_i striking unit surface area per unit time.

N_r = number of molecules per unit solid angle and unit speed range crossing a unit plane in a gas in a particular direction.

n = molecule number density (the absence of a subscript denotes the free-stream density. Subscripts 1, 2, 3, etc., denote specified regions. Subscript f denotes number density of a fictitious gas).

n(t) = instantaneous value of molecule number density in gage at time t, Eq. 2.321.

P_i(C_i, θ_i, φ_i, C_r, θ_r, φ_r) = distribution function denoting the probability per unit solid angle and speed range that a molecule with incident speed C_i and direction θ_i, φ_i will be scattered with speed C_r in direction θ_r, φ_r (If present, the subscript D identifies the probability distribution function with the diffusely scattered molecules while the subscript S identifies that associated with the specularly scattered molecules).
\[ P_2(\theta_i, \phi_i, \phi) = \text{distribution function denoting the probability per unit solid angle that a molecule with incident direction } \theta_i \text{ will be scattered in direction } \phi \text{ (the subscripts } D \text{ and } S \text{ perform the same function as above).} \]

\[ P_3(C_R) = \text{distribution function denoting the probability per unit speed range that a molecule will issue from a surface with the speed } C_R. \]

\[ P, \Delta P = \text{pressure and pressure increment (subscripts if used denote specified regions).} \]

\[ R = \text{surface - detector separation (Fig. 7)} \]

\[ \lambda = \text{beam path length (Fig. 7)} \]

\[ J' = \text{function defined Eq. 4.27} \]

\[ J'' = \frac{b e^{-b^2}}{2Vm} J' \]

\[ S = \text{molecular speed ratio} = \frac{U}{V_m} \]

\[ T = \text{free-stream temperature, Kelvin (subscripts 1, 2, 3, etc. if applied refer to specified regions. Subscript } f \text{ refers to fictitious gas. Subscript } w \text{ refers to wall temperature).} \]

\[ U = \text{magnitude of the mass velocity} \]

\[ V = \text{detector gage volume} \]

\[ V_m = \text{molecular most probable speed} = \sqrt{\frac{m}{2kT}} = \frac{1}{\beta} \]

(Subscripts r, f, if used, associate Vm with Tw, Tf respectively).

\[ W = \text{probability that a molecule striking the plane of the end of a tube of specified shape from a gas at rest will pass through without returning to the entrance plane (subscripts 1, 2, 3, etc., when used, denote specific tubes).} \]

\[ X, Y, Z = \text{coordinate system of incident momentum considerations, Fig. 41} \]

\[ X', Y', Z' = \text{coordinate system of issuing momentum considerations, Fig. 42.} \]

\text{GREEK SYMBOLS}

\[ \alpha = \text{accommodation coefficient, section 1.2} \]
\[ \beta = \frac{1}{V_m} \] (subscripts perform the same function as for \( V_m \))

\( \theta_i, \phi_i, \theta_r, \phi_r \) = incident and reflected molecule direction in unprimed coordinate system, Fig. 41.

\( \theta', \phi' \) = reflected molecule directions in primed coordinate system, section 4.4.

\( \lambda \) = mean free path

\( \mu \) = microns Hg = 0.001 mm Hg, a measure of pressure.

\( \xi, \eta, \zeta \) = coordinate system fixed in a gas.

\( \psi \) = polar angle as measured from \( \zeta \) axis in above coordinate system.

\( d\Omega \) = element of solid angle in above coordinate system.

\( d\omega, d\omega' \) = element of solid angle in unprimed and primed X,Y,Z, coordinate systems.

SYMBOLS USED IN SECTIONS 2.33 and 2.34 ONLY

\( A \) = amplifier loop gain

\( E_1 \) = potential across control resistor, emission regulator.

\( E_0 \) = reference potential, emission regulator.

\( e_\text{error} \) = error signal, emission regulator

\( e_\text{out} \) = signal to meter, difference amplifier.

\( e_\text{in} \) = signal current, difference amplifier.

\( R_1, R_2 \) etc. = circuit resistances.

\( T_1, T_2 \) etc. = circuit transformers.

\( V_1, V_2 \) etc. = circuit vacuum tubes.
1.0 **INTRODUCTION**

1.1 **General**

Since the time of Maxwell and extending through the first third of this century there have been many investigation in the field of rarefied gases. In recent times there has been a resurgence of interest, stimulated in part by the need for a wider understanding of aerodynamic conditions met by bodies in high speed travel through the upper atmosphere. Of the many investigations in this field only a small portion have been concerned directly with the interaction of gas and surface. It was the objective of the program of this report to investigate one aspect of these gas-surface interactions by molecular beam technique.

1.2 **Low Density Flow Investigation as a Technique of Gas-Surface Interaction Study**

Historically the problems of the interaction of gas and surface were first investigated following the realization by Kundt and Warburg that the gas was not at rest at the boundaries of rarefied gas flows (Ref. 1). This mass motion is determined, in part, by the detailed behavior of the molecules in interaction with the surface. Considerations relating to these phenomena, therefore, have been of great interest to aerodynamicists and have played an important role in the formulation of rarefied gas flow boundary conditions. Conversely, from experimental investigations of rarefied gas flows for which model boundary conditions have been formulated, some deductive understanding of the gas surface interactions has been gained. Summaries of the major concepts and results of rarefied gas flow theory and experiment are contained in references 2 and 3.

Maxwell in 1879, (Ref. 4) proposed a hypothetical model for the gas-surface interaction with respect to momentum transfer which was consistent with the available data and which has been used in part in the formulation of boundary conditions to this date. If some molecules are assumed to be scattered diffusely from the surface at some temperature as yet unspecified such that the scattered flux distribution is
in accordance with Lamberts law of diffuse scattering without regard to incident angle or speed, and if the remainder are supposed to be reflected specularly, then one can define \( f \), termed the "coefficient of specular reflection", as that fraction of all molecules which is diffusely scattered. Since only this fraction contributes to the tangential momentum transfer to the surface, one is led immediately to an alternative definition of \( f \),

\[
f = \frac{G_i - G_r}{G_i}
\]

where \( G_i \) is the total incident tangential momentum brought to the surface by all molecules and \( G_r \) is that carried from the surface. Only those molecules which are specularly reflected in the Maxwell model contribute to \( G_r \). The coefficient \( f \) can be restated, then, as being that fraction of the total incident tangential momentum which is transferred to the surface. It is in this sense that \( f \) is most reasonably defined since it can be applied to a macroscopic experiment without the making of assumptions regarding the detailed interaction. It is this definition which will be employed in this report.

In its past application \( f \) has been assumed to be constant of the surface and the gas independent of gas mass velocity or gas and surface temperature in the absence of more comprehensive data.

As a result of somewhat analogous reasoning applied to the development of a theory of heat conduction in gases, Knudsen in 1911 (Ref. 5) proposed a coefficient \( \alpha \), the accommodation coefficient, defined as

\[
\alpha = \frac{E_i - E_r}{E_i - E_w}
\]

where \( E_i \) is that energy brought up to the surface by the incident molecules, \( E_r \) is that which is carried away by them as they are scattered from the surface, and \( E_w \) is that energy which would be carried away by the issuing molecules if they had come from a gas in Maxwellian equilibrium at the temperature of the wall. The accommodation coefficient has been in use to the present in the formulation of boundary conditions. As with \( f \) it has been assumed to be constant of the surface and the gas.
There exists a small body of literature pertaining to theoretical and experimental determinations of $f$ as a result of studies of macroscopic flows. A table of all known values of $f$ from such determinations may be found in Table I.

Although the work of this report is applicable only in consideration of momentum transfer, a table of experimental values of $\alpha$ is included in Table II. Many important investigators whose names do not appear in the body of this report are referenced in these tables and so appear in the bibliography.

It can readily be seen from these tables that our knowledge of either of these quantities is not extensive. More objectionable than this is the likelihood that the values as reported are valid only for flow conditions approximating those of the experimental determinations. The chief deficiency of these determinations, however, lies in the fact that they yield very little insight into the physical nature of the gas-surface interaction beyond the essential fact that by and large the molecules "forget" their past history upon interaction with a surface.

1.3 Detailed Investigations of Gas-Surface Interactions

Soon after the first concepts of the gas-surface interactions were formulated investigations were undertaken to study these phenomena in more detail. Some variation of the molecular beam technique was almost invariably used. These investigations, that is, involved a directed stream of molecules in interaction with a plane surface element. Measurements of the flux of scattered molecules at one or more positions about the scattering surface were obtained. A considerable body of literature exists devoted to the description of these experiments and to theories of the interaction. Two very comprehensive surveys of the techniques and results of experiment are contained in Refs. 2, and 6. A summary table of the best known of these investigations can be found in Table III. Those investigators not specifically mentioned in the body of the report are referenced in that table. Beyond pointing out certain salient features of these investigations it would not further this report to elaborate on them in greater detail.
Of these many investigations the most definitive was clearly that of I. Estermann and O. Stern (Ref. 7) in their classical experiments on the diffraction of He and H₂ from cleavage planes of LiF₁ and NaCl. The location of the observed diffraction maxima was found to be in strict accordance with the prediction of the optics of cross gratings.

The majority of the remainder of the investigators also worked with cleaved surfaces of natural crystals. For the most part the mode of molecular beam detection dictated the use of various metallic atoms, more or less readily condensable at the temperature of the scattering surface, for the parent beam material. These experiments did not in general permit such a clear and unambiguous interpretation of results as did the work of Estermann and Stern.

Only one group of tests on surfaces other than crystal cleavage planes appears appropriate to this present report. These were performed by Knauer and Stern (Ref. 8) on polished speculum metal. An optical criterion for the specular reflection of transverse waves is that the irregularities of the reflecting surface be less than a wave-length in extent as measured normal to the axis of wave propagation. This condition was believed to occur at glancing angles of 0.001 radian and less for H₂ and the polished metal. Experiment verified these predictions. Above 0.001 radian no detectable amount of specular reflection was observed.

On the basis of the foregoing remarks and Table III the state of knowledge of the surface interaction may be summarized as follows:

1. He and H₂ interact with cleavage planes of simple crystals in a fashion predicted by wave mechanics.

2. Gases having high forces of interaction with the crystal surface apparently scatter diffusely from it. For example, Li, Na, Cs, from NaCl.

3. Apparently there is a tendency for some metallic gases to undergo a classical elastic specular reflection at a crystal surface where the interaction forces are low in a manner not readily understood in wave mechanical terms.

4. Gas-surface interactions involving surfaces which are "liquid"
micro-crystalline or man-formed by polishing have been studied hardly at all. The past findings in this area suggest that the scattering is largely diffuse at least wherever the criterion for optical specular reflection is not satisfied.

There is, however, experimental evidence from the determination of values of $f$ that some interactions are not entirely diffuse. For example, Millikan reported a value of $f = 0.89$ for air on glass. Clearly it is here in these areas of diffuse or semif-diffuse scattering that our understanding of the molecule-surface interaction is most meager.

1.4 Objectives of the Experiment

In most general terms it is the objective of the molecular beam program to study the fundamental nature of the gas-surface interactions by molecular beam techniques. More specifically the present experiment was designed to enable the measurement of the issuing flux distributions of atmospheric constituent molecules scattered from surfaces of certain ordinary materials oriented at arbitrary angles to the incident molecular beam.

It was recognized that the work herein reported could constitute only a beginning to a systematic inquiry into the nature of the molecule-surface interaction. Therefore, in establishing the choice of gas and surface, experiment geometry and resolution, an attempt was made to be guided by consideration of what should constitute a logical first step in such an investigation. The details of the experimental technique and of the test program will be dealt with in the appropriate sections of this report.

In addition to reporting the results of the experimental program and discussing these findings, it was a further objective of this paper to suggest by analysis the desirability of restating the rarefied gas flow boundary condition for momentum transfer in terms of a more general transfer ratio than the constant $f$. In connection with this analysis a technique was developed for computing the value of $f$, subject to certain assumptions to be discussed later in this report, for a Maxwellian gas with superimposed mass velocity tangentially directed relative to a surface. This computation was carried through for the $N_2$-glass interaction on the basis
of data obtained as a portion of the experimental work of this report.

2.0 DESCRIPTION OF THE APPARATUS

2.1 Introductory Remarks

The majority of molecular beam devices in existence, and those most familiar to physicists today, have been constructed for the study of the fundamental properties of the atom or nucleus as revealed by the interaction of these particles with electromagnetic fields. Since it might be expected that a molecular beam apparatus constructed for the study of the scattering of molecules from surfaces would differ slightly from these other equipments, a brief description of the equipment of this program will be given.

The molecular beam is formed by the thermal effusion of molecules through an orifice or from a small source chamber. The effusing molecules travel rectilinear diverging paths to the plane of the defining orifice. Those traveling in the proper direction pass on through and constitute the molecular beam. The remainder (the vast majority) are stopped and become removed by the source pump.

The molecular beam downstream from the defining orifice travels through a region of high vacuum \( P \sim 10^{-6} \text{ mm Hg} \); mean free path, \( \lambda \sim 50 \text{ meters} \) until it strikes the scattering surface oriented at an externally variable angle of incidence to the beam axis. Those molecules issuing from the surface into a small solid angle \( \Delta \Omega' \) at \( \theta' \) and \( \phi' \) measured from coordinate axes fixed in the surface (Figure 1) pass into the detector and create a beam signal by producing a small pressure increment within that gage. By moving the detector through all values of the coordinates one can obtain the relative distribution of molecular flux into all elements of the hemisphere at a fixed value of the incident angle. By repeating the measurements for representative values of the incident angle a description of the scattering of the molecules of a given gas from a given surface can be obtained. The data of this report were obtained in this fashion. Note, however, that this description gives a picture of scattering behavior which is by no means complete, since it applies to one incident gas temperature only, one surface temperature.
only, and makes no measurement of the distribution in speed of the scattered molecules.

A description of the molecular beam apparatus is most logically divided into two principle sections, one concerning the physical apparatus, and the other concerning the molecular beam detection system.

2.2 The Physical Apparatus

In briefest terms the physical apparatus consisted of the beam forming and detecting elements supported within vacuum envelopes which were in turn associated with an evacuating system. Figure 2 is a schematic of the pumping system showing the source and detector regions of the apparatus and attached pumps, valves and refrigeration components. Figure 3 is a photograph of the complete equipment. The principle vacuum envelopes (source region and detector region) were of stainless steel. Other parts were variously of cold rolled steel, aluminum, copper, or brass. Nearly all gasketing was by means of rubber "O" rings for fixed as well as rotating seals. With the exception of the quarter swing valves between the main envelopes and the refrigerated traps, all valves were equipped with Teflon seats. Under beam-on conditions typical detector chamber pressures were of the order of $10^{-6}$ mm Hg or less while the source chamber pressures were of the order of $2 \times 10^{-5}$ mm Hg or greater.

All parts of the molecular beam forming and detecting equipment were designed to be supported on an optical bench which in turn was supported by the plate dividing the source vacuum envelope from the detector envelope. The mounted elements were, in order of their appearance in the beam forming or detecting process, the source chamber and source orifice, the defining orifice (carried by the bench support plate), the shutter, the secondary defining orifice, the detector mechanism (with test surface and detector gage), and the comparator gage. Attached to the downstream end of the optical bench was a plate (the "apron") carrying coordinate counters and screw driver tipped fittings linking the moveable parts on the optical bench with the rotating shafts piercing the detector envelope and plate. Figure 4 shows the optical bench and all mounted components.
The source chamber was mounted on a slide providing adjustment of the source orifice center by means of lead screws and ways. The source "orifice" was a tube mounted so as to make possible precise alignment of the beam axis with the source tube axis. A pair of auxiliary optical bench carriages with transverse and vertical slides and coordinate counters and mounting a Point-o-lite and a telemicroscope respectively were used for the beam alignment throughout. The source chamber was connected to the source gas supply system and manometer system shown in Figure 5.

The most complex of the various components of the beam forming and detecting system was the detector mechanism. Basically, this mechanism was a mounting device for holding two elements of the molecular beam system; i.e., test surface, and detector gage, in the desired mutual geometrical relationship. The test surface was mounted normal to the axis of the azimuth rotation axis by means of a shaft along that axis. (See Fig. 6). The test surface and all other elements above the transverse carriage could be rotated in a horizontal plane about a vertical line element in the test surface face by means of exterior controls. Thus the incident angle of the beam with respect to the test surface was variable. The upper elements of the detector mechanism supported the detector gage. These elements were rotatably mounted to provide adjustment of the polar and azimuth angles of the gage entrance tube referred to the polar axis normal to the test surface. Azimuth angle control was made by internal adjustment. The polar angle was controlled externally by means of flexible shafting and swiveling miter gear linkages. The gage support system was terminated by a small ground bar way which carried the gage holder. The gage holder provided adjustment of gage-to-surface separation and rotational and vertical alignment. All motions not considered "set-up" were provided with coordinate indicators in the form of 5 place Veeder Root counters. Angular position settings were determined to have a reproducibility of 0.015°. On the basis of guaranteed minimum gear quality the total cumulative error in angular setting was estimated to be approximately 0.05°. For the incident and polar angle gear trains the counters rotated 38,519 units per degree.
Directly associated with the physical apparatus were certain items of general instrumentation including pressure monitors, source gas regulation and pressure measurement devices, and protective devices. Three Pirani type vacuum gages were employed to monitor pressures in the appropriate range, one being connected to the fore pressure line, one between the booster pump and the diffusion pumps, and one to the source pressure manifold. Both the source and detector sections of the main vacuum envelope were provided with Alpert type ionization gages for base pressure monitoring. Each was connected to a standard Low Pressures Project design power supply and metering circuit. This device is provided with an over-pressure protection circuit which removes filament power if the pressure should exceed the maximum pressure of the range selected. At the same time power is removed from a control line which may be connected to other devices. In this way the detector and comparator gages were also protected against the effects of over-pressure conditions.

The source pressure manometer indicated in Figure 5 was an early forerunner of the present Low Pressures design precision oil manometer (Ref. 25). While it lacked certain mechanical and optical refinements of the present models, oil column height differences were easily read to within 0.001 inches. The reliability of readings in calibration experiments with a precision McLeod have been found to be ±1 micron. A more detailed discussion of the design consideration leading to the present equipment may be found in Ref. 36.

2.3 The Molecular Beam Detection System

2.31 Preliminary Considerations

A portion of the molecular beam apparatus necessitated by the present experimental program was a beam detection system for atmospheric constituent molecules. Such a system must include a gage which collects a number of directed molecules and measures the resultant increment in molecule number density within its volume.

Referring to Figure 7, with the gage set to receive the direct beam and with the test surface withdrawn, let the molecular beam shutter be opened suddenly, the molecule number density within the gage having been at equilibrium with the
residual gas within the vacuum shell. Then $N$ directed molecules per second pass into the gage volume. At equilibrium with this new influx $\frac{m_3 c_3 A_{3b}}{4}$ molecules per second strike the area $A_{3b}$ from within the gage, where $c_3$ is the molecule mean speed at the temperature of the gas and $m_3$ is the incremental molecule number density. Of these $\frac{m_3 c_3 A_{3b} W_3}{4}$ succeed in emerging from the entrance tube, where $W_3$ is the probability that a molecule striking the area $A_{3b}$ will emerge through the area $A_{3b}$ without re-entering the gage. $\mathcal{W}_3$ depends upon the shape of the entrance tube only.

At equilibrium, then,

$$m_3 = \frac{4N}{c_3 A_{3b} \mathcal{W}_3} \quad (2.311)$$

Since this adjustment takes place under free molecule flow conditions it is assumed to be independent of other similar adjustments taking place simultaneously as a result of small changes in instrument base pressure, for example. It has also been assumed that no thermal gradients exist along the entrance tube.

In order to fix ideas of the orders of magnitude involved in the detection of a diffusely scattered beam, consider the idealized schematic of a scattering experiment, Figure 7.

When the length of the beam path, $\lambda$, very greatly exceeds $\alpha$, the radius of the source orifice, point source geometry affords a good approximation in the calculation of beam intensity. The number of molecules effusing from the source orifice under conditions of free molecule flow is $\frac{m_1 c_1 A_1}{4}$, where $A_1$ is the area of the source orifice, $c_1$ is the mean speed of the source molecules, and $m_1$ is the source molecule number density. Let the beam cross sectional area be chosen as $A_2$ at $\lambda$ and the defining slit size adjusted to permit this. The number of molecules crossing $A_2$ per second becomes

$$N_2 = \frac{m_1 c_1 A_1 A_2}{8 \pi \lambda^2} \quad (2.312)$$
Let these molecules be intercepted by the scattering surface. For an unambiguous determination of the flux of molecules through the area $A_3$ at angle $\Theta'$ as measured from the normal to the scattering surface, let $A_3$ be an orifice in the wall of the detector gage ($W=1$).

If there is perfectly diffuse scattering at the surface, the number of molecules crossing $A_3$ into the gage is

$$\frac{m_1C_1 A_3 \cos \Theta'}{16 \pi \lambda^2 R^2}.$$  

At equilibrium then

$$\frac{m_3 C_3 A_3}{4} = \frac{m_1 C_1 A_1 A_3 \cos \Theta'}{16 \pi \lambda^2 R^2}$$

since $C_3 = \sqrt{\frac{C_1}{C_1}}$, and this ratio is of order 1 for moderate gage temperatures. Thus:

$$m_3 = \frac{m_1 a_2^2 a_3^2 \cos \Theta'}{4 \lambda^2 R^2}$$  \hspace{1cm} (2.313)

where $m_3$ is the incremental molecule number density in the gage due to the scattered beam.

The desired angular resolution of the experiment, the physical size of various components of the beam forming and detecting system, and the available pumping speeds on source and detector sides of the plane of the defining orifice all play a part in limiting the choice of the parameters on the right hand side of equation 2.313. For the present experiment, typical values

- $a_1 = 0.25$ mm
- $a_2 = 1.5$ mm
- $\lambda = 25$ cm
- $R = 3$ cm
- $m_1 = 3.6 \times 10^{11}$ molecules cm$^{-3}$ at a maximum

then

$$m_3 = 2.25 \times 10^7 \cos \Theta'.$$

The pressure increment corresponding to this density increment is

$$\Delta P = 0.63 \times 10^{-9} \text{ mm Hg}.$$  

An entrance cone of taper such that no molecule can hit the side wall before entering the gage volume (Figure 7) can be constructed to have a value of $W_3$ of the order 0.2. The scattered beam pressure increment may thus be raised to the order of $3.1 \times 10^{-9}$ mm Hg.
A pressure increment of this small size would not be
difficult to detect were it not for the presence of the back-
ground gas. This residual gas, existing in the detector region
at a mean pressure of $10^{-6}$ mm Hg, suffers rapid random fluctu-
ations in pressure as a result of uneven diffusion pump operation.
These pressure fluctuations under operating conditions are of
the order of $3 \times 10^{-8}$ mm Hg. This gas also suffers slower
"drifts" in pressure in response to minor changes in diffusion
pump efficiency or system outgassing rate. These changes in
turn may be caused by changes in pump heater current, ambient
temperature, cooling water temperature or cold-trap temperature.
A background pressure change of $5 \times 10^{-8}$ mm Hg may occur in a
few seconds, of $10^{-7}$ mm Hg in several minutes or $5 \times 10^{-7}$ mm Hg
in the course of a few hours.

Thus the atmospheric constituent detection system must in-
clude a second gage known as the "comparator". This gage must
measure the background pressure, responding to its fluctuations
with the same sensitivity and with the same time of response as
that of the detector. If the comparator provided a perfect
replica of the signal from the detector in response to the back-
ground pressure fluctuations, a subtraction of the total signal
from each gage would leave a net signal representing only those
pressure increments due to the beam. This condition of perfect
compensation is never quite realized in practice.

An important requirement for good compensation is that the
time response of comparator and detector gages be equal. The
time response of either gage to a stepwise increment of mole-
cular flow into the gage, $N$ molecules per unit time, is given
by

$$m(t) = \frac{4N}{\bar{v}_3 W_3 A_3} \left( 1 - e^{-\frac{v}{4\bar{v}} t} \right)$$

(2.314)

where $m(t)$ is the incremental number density of molecules in
the gage volume at time $t$ after initiation of the incremental
flow, $\bar{v}_3$ is the mean molecular speed of the molecules within
the gage, $A_3$ is the area of the entrance tube opening, and
\( W_3 \) is the probability that a molecule will pass on out of the tube without returning to the gage volume. \( V \) is the volume of the gage. If \( N \) is due to a stepwise increase in molecule number density \( n_0 \) within the vacuum shell of the detector region,

\[
m(t) = m_0 \frac{c_3}{c_3} \left( 1 - e^{-\frac{c_3 W_3 A_3 t}{4V}} \right)
\]

Clearly, if the response times of detector and comparator are to be made equal, the factors \( \frac{c_3 W_3 A_3}{4V} \) must be made equal. This was most conveniently achieved by providing the comparator gage with a telescoping entrance tube which was then adjusted by trial and error for minimum time response difference.

Although the fundamental concepts of the detector-comparator form of measurement have been widely applied in other fields (and referred to in other terms) they are believed to have been first applied to a molecular beam detection system by O. Stern, I. Estermann and their associates.

2.32 The Ionization Gage Detector

As has been remarked earlier in this report, the ionization gage was chosen for development as a molecular beam detector. The Pirani gage appeared to have been brought close to the limit of its potential by earlier workers (again Estermann, Stern, et al). It did not seem that their work could be bettered by one or two orders of magnitude.

In its most common form the ionization gage employs three major elements, contained within and supported by a glass envelope. The heated filament, operating temperature-limited, supplies
electrons for the ionization of neutral molecules in the grid-anode region. The grid accelerates the electrons and ultimately collects them after they have made several passes from grid-cathode to grid-anode region and back. Such ions as are formed by electron collision in the grid-anode region are collected at the anode. Electrode potentials and field configuration are such as to permit this action. The magnitude of the ion current is linearly proportional to the number density of molecules in the gage at pressures below one micron Hg and to the electron emission current. It is also a function of grid potential and gage geometry, and is inherently very insensitive to changes in ambient temperature.

An emitting tungsten filament continuously loses mass to its surroundings. Ultimately, conducting films of tungsten become deposited on the gage walls. At the same time diffusion pump oils are broken down to elemental constituents or to resinous molecules and also become deposited on the walls. A complex film is built up having conduction properties of varied and sometimes rapidly fluctuating character. Leakage paths and fluctuating charge distributions have a very disturbing effect on the stability of the ion current when that current must be known to a few parts in $10^5$, although for ordinary pressure monitoring these deposits are of little interest.

In designing an ion gage for use as a molecular beam detector, particular care must be taken to insure that electrical leakage paths between the various elements or from them to ground have virtually no opportunity of formation. The gage must also be exceptionally rugged and physically compact.

By February of 1949, the major design features of a satisfactory ion gage for molecular beam detection were well established. A detailed record of this development is presented in Ref. 26. The ion gage had improved itself in two experimental programs (Refs. 27 and 28) and it had been established that signals at least as small as those encountered in the work of earlier experimenters could be resolved from the noise.
In 1952 a major change was made in the molecular beam detector as a result of the publication of the work of D. Alpert and others at Westinghouse Research Laboratory (Ref. 29). While the major advantage of the Alpert gage (i.e., the measurement of extremely low absolute pressures) was thought not to be of interest at this time its configuration allowed a cleanliness and ruggedness of structure which was regarded as very desirable. On the basis of the Alpert design and on the previously achieved design of this project, the ion gage detector of the present investigation was constructed (see Figure 8).

The sensitivity and other performance characteristics of these gages were discovered to be very similar to those of the Alpert gage. In spite of the attention given to the prevention of film formation this still constituted a serious problem. One month was found to be about the maximum running time before conducting films made the gage inoperable for molecular beam use. A thorough cleaning with dilute hydrofluoric acid was found to be necessary to restore a dirty gage to usefulness.

2.33 The Emission Regulator

For the proper operation of the ionization gage detector a stable source of gird potential must be supplied. Also a source of filament power must be provided which is so regulated as to permit an unvarying flow of emission current within the gage. It was a requirement of the experiment that this current be regulated to a few parts in $10^5$. Similarly the source of grid potential was required to possess a very high order of stability. The device which supplied both filament and grid was termed the emission regulator. Two such devices were constructed, one for the detector gage and one for the comparator. (Beam detector system schematic, Figure 9).

This filament emission regulator was of a type in which the total emission current develops a signal $E_1$ across the control resistor $R$. (Figure 10). The potential $E_1$ is compared with a reference potential $E_0$. The difference $e_0 (E_1 - E_0 = e_0)$ constitutes an error signal which is then minimised by the
regulator action. In a total feed-back circuit of this type
\[ \frac{E_1}{E_0} = \frac{A}{1 + A} \]
where \( A \) is the loop gain of the circuit. All minor
fluctuations in regulator performance, supply voltage, filament
emissivity, etc., can be considered equivalent to changes in loop
gain. In principle the ratio \( \frac{E_1}{E_0} \) can be made to remain as con-
stant as desired by providing a sufficiently high loop gain.
In particular if variations in effective loop gain of 10% \( A \) are
to be expected, and if the ratio \( \frac{E_1}{E_0} \) is to remain constant to
1 part in \( 10^5 \) then \( A \) must have the value of 10,000. Vari-
atations in the reference potential \( E_0 \) and variations in input
stage quiescent point are excluded from this argument, of course.

It is a matter of experimental record, however, that the
achievement of a loop gain of this order is not feasible in
the straight forward D.C. amplifier type of emission regulator
design. If the voltage output of the regulator is sinusoidal
the resultant emission current will have a ripple component
superimposed on the D.C. level. This ripple component is the
result of the response of the temperature limited emission
current to the periodic fluctuation of the impressed power.
A high gain D.C. amplifier normally becomes blocked by this
signal in a final stage. Attempts to reduce the magnitude of
this ripple component by brute force filtering are met with
very limited success because of the attendant phase shift around
the loop and the resultant low frequency hunting.

A point at which phase shift is unavoidably introduced is
at the filament of the ionization gage itself as a result of the
thermal capacity of the filament. This highly non-linear ele-
ment makes the realisation of extremely precise regulation
more difficult than in the case of a pure resistive load.

The design of the ion gage detector regulator followed
an approach suggested by Greenough, Williams and Taylor (Ref. 30)
and applied by them to a low voltage power supply for electro-
lysis. The essential feature of the Greenough, Williams ap-
proach was the use of a square wave as the regulator output
signal. The application of power in this form to the filament
results in a great reduction in the emission current ripple component and a consequent increase in the feasible loop gain.

Schematic diagrams of the complete regulator and power supply circuits are presented in Figures 11 and 12. Referring to the schematic of Figure 11 the emission current entering at point A develops a signal voltage across precision resistors R. This voltage is compared with reference battery $E_0$ and the resultant difference signal is applied to the grid of the first amplifier $V_2$. $V_1$ compensates the signal for small changes in contact potential originating from filament emission changes in $V_2$. $V_4$ and $V_5$ comprise a multivibrator pair serving as the plate load of the control tube $V_3$. A square wave of acceptable quality is electron coupled out through a loop gain control dual potentiometer to a high grain AC amplifier adapted from the Childs circuit (Ref. 31). The output transformer $T_1$ was chosen to have 1 db band pass of 20-20,000 cycles, to insure good transfer of the square wave to the ion gage filament. Provision was made for balancing the plate currents in the push-pull power output stage. It was found necessary to include the variable RC network $R_1 R_2 R_3 C_1 C_2$ as an anti-hunt control. Highest performance of the regulator was achieved by adjusting anti-hunt, loop gain, and various balance controls.

The power supply units (Figure 12) were rebuilt from Signal Corps. SCR 545A mobile Radar "standard" power supplies.

Using a precision voltage divider, biasing batteries and a 0-20 millivolt strip chart recorder, determinations of the stability of the regulator and of its loop gain were made. After a short initial warm up period the "drift" was of the order of one part in $10^4$ per minute. This was a matter of no concern since after warm up the drifts had been ascertained to be less than 1% per day except under conditions of excessive ambient temperature change. (The reference potentials were supplied by ordinary dry cells). The noise was of the order of 2 parts in $10^5$ of the total signal in amplitude, permitting adequate gage operation. The loop gain was variously checked at between 10,000 and 20,000 under good operating conditions.
While long time stability of the highest order was not a requirement of the program of this report, it could have been approached more closely by making two simple changes in the equipment. The reference potential could have been supplied by a mercury cell having a low temperature coefficient of e.m.f., and the heaters of the tubes of the input stage could have been powered from the regulated supply.

2.34 The Difference Amplifier

At some point in the detection process, in order that meaningful data can be obtained from the molecular beam detector, the signals from detector and comparator must be compared and a difference signal noted. The apparatus which performed this function in the molecular beam detector system was known as the difference amplifier.

In order to understand the nature of the problem it is useful to recall the orders of magnitude of the current involved. The background pressure current was of the order of \(10^{-8}\) to \(10^{-7}\) amperes. Random current fluctuations of the order of \(10^{-9}\) amperes in amplitude were superimposed on this as a result of the background pressure fluctuations. The beam signal current was of the order of \(10^{-10}\) amperes. It was thought desirable to resolve this signal to at least one part in 10. The conditions of the experiment required therefore that currents from detector and comparator be amplified and subtracted in such fashion as to yield difference signals of as little as one part in \(10^4\) of the total.

Systems for performing a similar function have been devised in the past for use in mass spectrometer isotope abundance studies (see for instance, Ref. 32). These systems normally used one complete amplifier for each signal, the subtraction being performed directly preceding the indicating device. It was thought, in view of the special circumstance of having nearly identical input signals, that the two amplifiers could be joined in a single push-pull circuit minimizing the opportunity for unequal phase shift through two amplifiers and for other inequalities of response possible with two complete amplifiers. This and more specialized design considerations led to the difference amplifier circuit shown in Figure 13. Power supplies for this
unit are shown in Figure 14.

If circuit sections of Figure 13 including $V_1$, $V_2$ and $V_2$ are considered alone, it can be seen that these together form a basic feed back electrometer circuit for which $e_{out} = \frac{i_{in} R_1}{1 + A}$ where $e_{out}$ is the signal potential to the metering circuit, $i_{in}$ is the signal current in, $R_1$ is the high resistance range determining resistor and $A$ is the effective gain of the amplifier section. As with the emission regulator, if gain variations of 10% $A$ are to be expected, $A$ must have a value of the order of $10^4$ for the ratio $\frac{e_{out}}{i_{in}}$ to remain constant to one part in $10^5$. If a shunting resistance to ground is present (as for instance $R_2$ in Figure 14) the product $A R_2$ must be very much greater than $R_1$ in order for this shunt to have a negligible effect on the output signal. The gain of the amplifier in the difference amplifier circuit was somewhat greater than $10^4$.

Referring again to the schematic of Figure 13, the signals from detector and comparator are conducted to $V_1$ and $V_5$ housed in the pick up head. This section of the unit was adapted with minor modification from the National Research Corporation Alpha-tron vacuum gage. Selected 954 type vacuum tubes connected as shown have been found to show grid currents of less than $10^{-14}$ amperes (Ref. 33). The pairs $V_2$, $V_6$ and $V_3$, $V_7$ were connected with common cathode resistors to minimize the effects of variation in plate supply voltage and to afford a means of balancing the gain of each stage. The balanced cathode follower output stage provided a low impedance drive for the strip chart recorder.

With an input range resistor of $10^8$ ohms the least output difference signal distinguishable on the strip chart recorder was of the order of $10^{-12}$ amperes. Drift and noise appeared to be absent at this sensitivity. Larger values of the input range resistor were not employed because of the magnitude of the voltage which would have been developed across them by the large background pressure current.

All heaters were supplied from the regulated supply of Figure 14. This supply was also modified from the Signal Corps.
3.0 THE EXPERIMENTAL PROGRAM

3.1 Test Surfaces and Geometries

In view of the survey nature of this experimental program, it was regarded as of first importance to choose test surface and beam geometries in such a way as to allow an overall determination of the scattering behavior with a minimum of ambiguity. The results of the first design choices are shown in Figure 15. The beam was axially symmetric as was the gage entrance cone. The surface was circular and of approximately the same area as that of the entrance end of the gage entrance cone. At normal incidence the beam umbra illuminated somewhat less than the total surface area. However, low intensity elements of the penumbra overlapped the surface slightly. The gage entrance cone was designed to accept approximately 1% of the total hemisphere solid angle. This was a "low resolution" geometry, chosen, on the basis of beam intensity calculations and available signal-to-noise ratio, so as to provide a reasonable expectation of yielding detectable signals for the case of diffuse scattering.

Three surfaces were tested using this test geometry. These surfaces and the test surface holder are illustrated in Figure 16. A summary of this portion of the testing program follows.

3.11 Cold Rolled Steel

This specimen was of common cold rolled steel of low carbon content, lathe turned from a 1/4" diameter rod. The specimen was mounted in lucite and polished using standard metallographic laboratory techniques. Examination revealed a uniform mat of fine polish marks of the order of 1/4 micron in depth. The surface appeared to the naked eye to have a high polish. After insertion into the specimen holder in the molecular beam apparatus, the surface was thoroughly washed with absolute alcohol and dried with tank nitrogen. Polar traverses were made both in the incident plane and out of the incident plane with the beam incident at a variety of representative incident angles. Room air served as the test gas. The surface temperature was very slightly above room temperature. About 250 data points were obtained in the period Feb. 12, 1953 to March 2, 1953.
Following the completion of this sequence the surface was heated to 250 degrees centigrade and a series of incident plane polar traverses were made at this temperature in order to discover any possible change in the scattering behavior as a result of the outgassing of organic films or other adsorbed layers.

A final polar traverse was made again at room temperature.

3.12 Cold Rolled Steel, Etched

A specimen of the same material as that of item 3.11 was polished in the same fashion. It was then given a light etch in a standard 10% nitric solution. Microscopic examination showed the characteristic grain structure of a low carbon steel. No polish marks were visible. Room air was again used as the background gas. This specimen was tested as before, first at room temperature, then at elevated temperature, and again at room temperature. Incident plane polar traverses only were made. It was thought not worthwhile to spend the time on the exhaustive tests of the first section unless significant changes in the scattering behavior developed. In all, about 200 data points were obtained in the period March 5, 1953 to March 13, 1953.

3.13 Aluminum

This specimen was lathe turned from 302, a commercially grade of aluminum. It was polished in the same manner as the steel specimen. This specimen was tested as in item 3.12. Together, some 120 data points were obtained in the period March 23, 1953 through April 8, 1953.

Examination of the data of these first tests revealed deficiencies of the experimental geometry which were corrected in part before proceeding with the glass surfaces. It was observed that at angles of incidence moderately close to glancing, the edge of the specimen intercepted a portion of the beam. Molecules scattered by the edge of the specimen contributed in part to the detector signal, making difficult an interpretation of the scattering data in those directions. Secondly, direct beam molecules were detected at incident and exit angles both close to grazing as a result.
of finite beam breadth. While such conditions are ultimately met in any beam geometry it was thought that in this geometry the limitation on large incident angle measurement was excessive.

The correction of both of these imperfections could have been accomplished by changing to a narrow ribbon geometry but only at the sacrifice of out-of-the-incident-plane information. It was thought better to retain the overall survey approach at this stage, but to narrow the beam by replacing the circular secondary defining orifice by a 1 mm x 5 mm slit. It had been ascertained that the signal to noise ratio was sufficiently high to permit this attenuation of the beam. At the same time, edge scattering was minimized by increasing the test surface length. This change also made the preparation and mounting of the glass surfaces less difficult. This beam geometry is illustrated in Figure 15. The glass test surfaces are included in Figure 16.

3.14 Unpolished Glass Surface

This specimen was obtained from a pane of single strength window glass. It was thoroughly cleaned, given a final rinse in absolute alcohol, and dried with tank nitrogen. The surface as viewed by the naked eye showed no defects and appeared free of film. The structure of the surface was presumably "as cooled".

It was this specimen which was chosen for the second exhaustive scattering examination, the results of which were later to be used in the determination of \( f \), the coefficient of specular reflection. This choice was in part determined by the reported value \( f = 0.89 \) for air on glass as remarked in Section 1 of this report. In all, during the period April 17, 1953 to June 21, 1953, more than 400 data points were obtained. Tank nitrogen (99.5% \( \text{N}_2 \)) was used as the test gas. The tests were run with the surface essentially at ambient temperature. Since it was mounted on a rod which was warmed by conduction through parts connected to the detector ion gage, the test surface was maintained at about 10°C above the temperature of the surrounding walls.
3.15 Polished Glass Surface

This specimen was cut from the same stock as the unpolished glass surface. It was then sent to a manufacturing optical laboratory for polishing. The polished specimen was cleaned and rinsed in the same fashion as before. For purposes of comparison with the unpolished glass plate it was necessary to observe scattering in the incident plane only. Approximately 100 data points were obtained. The experimental program of this report was completed with the conclusion of these tests July 23, 1953.

3.2 The Experimental Data

Each polar traverse made at fixed incident and azimuth angles yielded several data points representing the flux of scattered molecules into the detector gage. Each datum point resulted from the averaging of a number of beam on-off traces as recorded by the strip chart recorder. A reproduction of such a record is presented in Figure 17. For each day's run the beam source pressure was held constant as was the gage emission current, and hence the overall system sensitivity. When possible, in each run a record of the normal incidence scattering was taken and also a record of the direct beam strength. These last records afforded a means of comparison of one run with another.

It was found most convenient to represent the data in polar plot form, scaling the data of each run up or down to bring data of all runs of a given sequence into register.

In the case of the steel and aluminum surfaces two additional problems in data interpretation must be mentioned. As a result of the width of the shutter mechanism and of the detector ion gage the polar traverses could be continued to within about 40° of the reverse beam axis only. This condition was improved only slightly in the glass plate tests. In these it was possible for the detector axis to come within 30 degrees of the reverse beam axis. The second problem resulted as a consequence of the small test surface size in relation to the beam diameter. As the angle of incidence was increased, an increasing percentage of beam molecules failed to strike the surface. The intensity of illumination per unit of surface area thus
fell off and in consequence the signal amplitude.

In view of the bulk of the experimental data and its general similarity in the case of the steel and aluminum surfaces no attempt will be made to present it in its entirety. However, representative polar plots for the steel plate are given in Figures 18 through 21 and for the aluminum plate, Figures 22 through 25.

Referring to the figures for the steel surface one can see unfortunate gaps appearing in the data at each angle of incidence. However, these gaps are somewhat filled by data obtained at other angles of incidence. The most striking aspect of these plots is the remarkable adherence of the data to a cosine distribution. The plots of data taken out of the incident plane still exhibit the same degree of circularity. No discernible change in the character of the scattering was made by outgassing the surface or running traverses at elevated temperature. For these surfaces undistorted representations of forward scattering were obtained to incident angles of about 70 degrees. The back scattering data at this incident angle was, however, seriously distorted by edge scattering. At larger angles of incidence (to 82 degrees) no evidence appeared of marked specular scattering, although, has such occurred close to 90 degrees, detection would have been masked by the signal of the direct beam.

A simple comparison of these data with the reference circles (introduced to aid in bringing the plots into register) suggests that the amplitude resolution of the experiment under these conditions was of the order of two to three percent of the maximum diffusely scattered signal.

Considering now Figures 22 through 25 for the aluminum surface one can observe that most of the remarks applying to the discussion of steel surfaces might also be applied to the aluminum. This work was not as complete as that on the steel surfaces, being done in the incident plane only and at fewer angles of incidence. While there seemed to be some deterioration in the signal amplitude resolution over this period, one must infer that the data indicates only slight deviations if any from the cosine distributions. There does, however, seem to be an indication of small bulges in the plots of the data of March 23, 1953, incident angles 53.65° and 81.83°.
The plots of the distribution of flux from the glass surfaces show a somewhat different character. Representative data are illustrated in Figures 26 through 32 for the unpolished glass surface and Figures 33 through 40 for the polished glass surface. On these plots one can readily observe bulges in the forward scattering direction, increasing in importance as the angle of beam incidence increases. These bulges are of the same general magnitude and shape for the two surfaces. Thus there would appear to be some interesting but undramatic deviations from the pure cosine distribution. While scattering data was taken at an angle of beam incidence as great as 84.82° no large specular peaks were observed. The plots at this angle of incidence show the influence of edge scattering and of direct beam detection at the extremes of the polar traverse.

One further obstacle to the quantitative interpretation of the data appeared in the observed flattening of the scattering patterns at the poles. This flattening arises as a consequence of the elongation of the beam spot on the extended test surface at large angles of incidence. Where the gage entrance tube also makes a large angle with the test surface normal the illuminated spot is entirely "seen" by the gage. As the tube approaches normal incidence, however, the gage can only "see" a portion of the illuminated area. If completely diffuse scattering were present the entire pattern would lie within the scattering circle established at normal incidence and would be flattened at the pole but symmetrically disposed about the test surface normal.

In order to compare the phenomenological results of the scattering tests with the reported value of \( f \) for air on a glass surface, an attempt was made to organize the data for this surface into complete "shells", each shell representing the distribution of scattered flux for all values of the exit coordinates at one angle of incidence. This work resulted in the construction of the smooth figures of Figures 29 through 32. The outermost line of each figure represents the flux as measured in the incident plane. The inner lines represent the flux in the other meridian planes, taken at 10 degree intervals.

Since the data was quite incomplete in many areas it was necessary
to fill in the shells on the basis of certain reasonable assumptions. It was assumed that the shell area directly intersected by the incoming beam had no sharp cavity or protuberance. It was further assumed that the "back scattering" section of the shell followed the cosine distribution. This assumption was supported by observed data as far as it went. This assumption is further supported by Zahl in his observations of the scattering of Hg from NaCl cleavage planes. (Ref. 35). This means that the surplus of molecules forming the bulge on the "forward scattering" side were assumed to be withdrawn from the total number of molecules striking the surface, leaving the remainder to be scattered in totally diffuse fashion. Finally, it was assumed that the fall-off of the non-diffuse component away from the incident plane proceeded in the same gradual fashion as did the observed fall-off of this component about the specular ray axis. This assumption was again supported by such data as was obtained out of the incident plane.

Four such shells were completed in detail for the incident angles 53.65 degrees, 66.63 degrees, 79.62 degrees, and 84.82 degrees. Data for incident angles at 0.00 degrees, 29.10 degrees and 40.67 degrees yielded completely spherical shells or, in the 40.67 degree case, a shell so nearly spherical as to make uneconomical the rather laborious numerical computations to be described in section 4.0.

3.3 A Discussion of Errors

The usability of the data of this experimental program was not dependent upon a knowledge of the absolute value of a pressure or a current, but rather upon the stability of the experimental condition over the period of one data sequence. A data sequence would be any group of related polar traverses including some trace suitable for relating this sequence to another. It was also dependent upon the short time stability of the signal over the period of a beam on-off cycle.

The factors influencing the long time stability of the experimental conditions were the source pressure, the ion gage sensitivity, and the difference amplifier sensitivity. Of these, only the source pressure variation produced a measurable change in the ultimate signal amplitude over the period of a run. The source pressure responded somewhat to ambient temperature changes as a result of differential expansion in the source needle valve. In the operating range (400 to 800 microns
the slope of the source pressure-beam flux curve varied from \( \sim 0.3 \) to \( \sim 0.1 \) chart units per micron. However, only on rare occasions were the fluctuations within the period of a data sequence important enough to warrant a correction. A one percent variation in ultimate signal amplitude from one end of a run to the other, was typical and was regarded as of no significance for the quantitative or interpretive purposes of this program.

Less can be said about changes in inherent gage sensitivity except that these changes (apparently due to film formation on the gage walls) occurred slowly and did not significantly influence the self consistency of the data.

The ultimate assurance of satisfactory long time system stability arose from the comparison of direct beam signal strength at the beginning and end of a data sequence. Such checks showed the direct beam signal strength variation to be usually smaller than one percent of the total beam. Occasionally, the variation was no greater than this over a period of several days.

A much more serious problem was that of achieving a satisfactory short time fluctuation level. This had been discussed in section 2.3 on instrumentation. However, it is worth considering the experimentally observed signal-to-noise ratio in relation to the reliability of the data. In the normal observing range the difference amplifier sensitivity was \( 10^{-11} \) amperes per chart unit. One chart unit for this recorder was equivalent to 0.1 millivolt and occupied approximately 0.05 inches of the recorder scale. At normal gage sensitivity a deflection of one chart unit would be produced by a pressure change in one ion gage of the order \( 10^{-10} \) mm Hg. On this scale the typical direct beam signal was of the order of 900 chart units while the diffusely scattered signal produced a plotted amplitude circle of the order of 18 chart units in diameter. This is consistent with the acceptance solid angle of the detector entrance tube.

At times when the observing conditions were exceptionally good (for instance as represented in Fig. 17) repeated points frequently reproduced themselves to 0.1 chart units and polar traverses showing cosine scattering behavior appeared to have an internal consistency allowing a scatter of 0.1 to 0.3 chart units from the reference circle.
Root mean square deviations of the signal trace amplitude under these conditions were of the order of 0.2 chart units. Observing conditions were regarded as too poor for the taking of interpretable data when the r.m.s. deviations exceeded 2.0 chart units. The bulk of the data was obtained under conditions lying in between these extremes. From observation of the polar traverses it was judged that over 90% of the experimental points lay within one half chart unit of a faired curve. Least square curve fittings were not attempted, since it was clear that the shapes of the scattering plots were well defined. Unfortunately, there remained those whose scattered experimental points led to curve fitting judgements which may have been in error by as much as 10%.

3.4 Physical Interpretation of the Data

The salient geometrical features of the data may be reviewed as follows:

(1) The most characteristic shape of all the scattering flux shells was that of a sphere. In the case of the glass surfaces the shells showed a bulging distortion of the forward hemisphere. The spherical shape (cosine distribution) was nearly perfect in the case of the metal surfaces tested.

(2) No shell possessed a "lobe" or dramatic protuberance. There were no cusps, cavities or reversals of curvature, i.e., the local curvature was always inward.

(3) The back scattered hemisphere appeared to be spherical surfaces without distortions in so far as could be determined.

(4) The distortion of the shells in the case of the glass surfaces increased with increasing angle of incidence. The point of maximum distortion occurred close to the intersection point of the specular ray.

(5) The scattering behavior for the unpolished and polished glass surfaces was very similar.

(6) There was no discernable effect on the scattering distributions for the metal surfaces as a result of outgassing
the surfaces either during or following the period of
elevated surface temperature.

These data provide a confirmation in some degree of what is
already known or what might be guessed regarding the interaction of
air or nitrogen with the surfaces tested. That is, it is shown that
by far the greatest fraction of air molecules incident to these sur-
faces scatter from them according to a cosine distribution. Values
of \( f \) for air on steel and aluminum have not yet appeared in the
literature. On the basis of these measurements \( f = 1.0 \). If all
scattering shells in a sequence were figures of revolution about the
polar axis but not necessarily spheres, \( f \) would also turn out to be
1. It is intuitively satisfying to conclude that probably most other
surfaces for which \( f \) equals 1 would yield cosine distribution scatter-
ing shells in test.

Still, little understanding is gained of the basic mechanism of
the scattering. It was not discovered, for example, whether the
interactions of the single molecules at the surface were more elastic
or more inelastic in nature. A perfectly rough surface might well
give rise to the cosine scattering distribution even if every indi-
vidual molecule-surface interaction were perfectly elastic. One
would suspect from published values of the energy accommodation
coefficient that this is not the case. On the other hand, one would
also suspect from these published values of \( \alpha \) that the interactions are
not entirely inelastic. For example, \( \alpha \) for air on polished cast
iron has values of 0.87 to 0.88, although \( f \) for polished cold rolled
steel has the value 1.0. However, if multiple impacts with the sur-
face irregularities are not excluded, the individual interactions
might be quite elastic and still result in the above values of \( f \)
and \( \alpha \). (Ref. 34). Thus, while the results of these experiments
give issuing flux distributions, it must again be emphasized that
they tell us little about the energy exchange at the surface and
hence little about the individual molecule-surface interaction mech-
anism.

One important uncertainty in the physical interpretation of
these results is that relating to the adsorbed films which might
lie on the test surface. In the case of the metal surfaces the re-
moval of superficial surface films (if present) by heating had no
effect on the test results as noted in item (6) above. Since the metal surface scattering distributions differed from those of the glass surface one might imagine that some essential properties of the underlying surface were reproduced in the adsorbed films. However, the extent to which the basic properties of the surface were obscured by the surface films remained largely undetermined as did the composition or extent of the surface films themselves. Since oil diffusion pumps were employed, it is quite possible that oil monolayers formed on all test surfaces in spite of efficient cold trapping. It is also quite possible that nitrogen and other residual gas molecules were adsorbed on the surfaces. The point of view which was therefore adopted as a result of the preceding consideration was that the effects of surface contamination on the angular flux distributions were small (as suggested by the results of item (6) above) but that some reservation must be retained in predicting behavior for surfaces of absolute cleanness on the basis of the present tests. A modification of the apparatus which would make future experiment less ambiguous in this respect would be the substitution of mercury vapor pumps for the oil diffusion pumps. Mercury vapor does not possess, as strongly as oil vapor, the tendency to stick to metal or glass surfaces.

The scattering behavior evidenced by the glass surfaces suggests strongly that a not-negligible minority of the incident molecules suffered elastic impacts with tilted surface elements, thus smearing out the "specular" component. Since the majority of molecules might still be considered to have undergone a cosine scattering one might imagine that here and there the local field contours have joined to form elastic scattering regions, standing like islands in a jumbled sea of unjoined fields. This model of the inelastic interaction for the majority of incident molecules with a smeared out specular reflection for the remainder is interestingly close to the original Maxwell model. This particular picture will be referred to as the "modified Maxwell hypothesis".

This picture is by no means the only one permissible as an interpretation of the observed scattering distributions. As was remarked earlier in this section, many hypotheses concerning the
energy exchange between molecule and surface could be made which would be consistent with the observed scattered flux distributions. One such is that all molecules are completely accommodated in energy to that of the surface and are re-emitted from the surface with a speed distribution in a given direction as though they had originated from a gas in equilibrium at the temperature of the surface.

The tests on the polished glass surface suggest that the elementary scattering elements retain their basic character in spite of the polishing operation. Some substantiation of this is found in the observations of lens makers that the polishing operation results in some local melting and flow of the glass surface.

Certainly there is a vast amount of work to be done before much more understanding of the gas surface interaction with such surfaces as these could be claimed. However, one of the most promising avenues of future discovery would appear to be in the use of a velocity selector. Such a device could certainly tell whether or not the "specular" component was accommodated to the surface temperature, and at the same time of course, it could yield data leading to an evaluation of the accommodation coefficient.

There are other experiments also likely to yield interesting results. What are the effects of raising or lowering the mean incident momentum? At what angles of incidence on complex polished surfaces do sharp reflection (if any) occur? What can be deduced about complex surfaces and their associated gas surface interactions from careful experiments on pure single crystals? These questions suggest only a few of the problems. It is hoped that time and further experiment will provide an understanding that is still lacking with the completion of this report.

4.0 A QUANTITATIVE EXAMINATION OF THE DATA

4.1 Introduction

It is of interest to consider the application of the experimental scattering distribution to the calculation of $f$, the coefficient of specular reflection. In performing this analysis the surface under consideration will be regarded as bounding a hypothetical free molecule flow having mass velocity directed tangentially to the surface.
Incident and issuing momentum calculations will be made, subject to various assumptions enumerated below. The issuing momentum calculations will, in part, be dependent upon the form of the experimental scattering distributions. From the incident and issuing momentum calculations, values of $f$ will be obtained in accordance with the definition of that coefficient.

The following assumptions will be applied to these calculations:

A. For the molecule-surface interaction the element of surface is assumed to be isotropic, i.e., molecules entering at $\Theta$ and $\Phi$ exhibit a characteristic scattering behavior independent of $\Phi$ (see Fig. 41). This assumption does not apply to cleaved surfaces of regular single crystals. It would seem reasonable, however, to apply it to the present case of a cooled liquid surface or to the case of "man-finished" surfaces of micro crystalline or amorphous materials.

B. It is assumed in the absence of experimental evidence to the contrary that the form of the scattering distribution is dependent on gas, surface, and incident angle only and is independent of the speed of the incident molecules within the range of speeds possessed by molecules of gases at ordinary temperatures. This implies that the form of the distribution is independent of the distribution of speeds in the incident ray. It may very well be that this assumption will be proved inaccurate by later experiment. It is introduced here simply to help provide a model for the calculations.

C. It is assumed for this calculation that the incident gas is so rarefied that the motion of the impinging molecules is essentially unaltered by those scattered from the surface in question. The velocity distribution of the impinging molecules will be taken to be the Maxwell distribution with superimposed mass velocity.

D. The definition of $f$ will again be stated, i.e., that

$$f = \frac{G_i - G_r}{G_i}$$  \hspace{1cm} (4.1)$$

where $G_i$ is the magnitude of the net tangential momentum brought up to the surface per unit time by all incident molecules and $G_r$ is the magnitude of the tangential momentum imparted by the surface to the issuing molecules per unit time.

At this point two alternate assumptions relating to the mode
of energy accommodation at the surface must be introduced. Calculations made on the basis of each of these assumptions will be presented.

E-1 It is assumed that molecules issuing in any particular direction from the scattering surface possess the distribution in speed which they would have had they issued from a body of gas in thermal equilibrum at the surface. It follows that the thermal accommodation coefficient has the value 1.

E-2 Consider again the form of the scattering distribution of section 3.0. As was more fully discussed in that section, the scattering shells (represented in Figures 29 through 32) show a hemispheric shape as the "back scattering" section. The distortions from the cosine scattering sphere take the form of bulges protruding from the "forward scattering" hemisphere. It is assumed that those molecules contributing to the bulging deviation from the cosine scattering sphere have been perfectly elastically scattered and so possess the same distribution in speed as before impact. The remainder, i.e., all those contributing to the cosine scattering sphere are assumed to possess the distribution in speed given by assumption (E-1). Thus, for the purpose of this calculation, the great majority of molecules are assumed to have been diffusely scattered with accommodation coefficient 1 while the remainder are assumed to have been specularly reflected from tilted surface elements. This is a special case of the modified Maxwell hypothesis discussed earlier in this report.

4.2 Calculation of $f$ Under Terms of Assumption E-1

First consider the quantity of tangential momentum incident on the surface element. For a representation of surface and coordinate system, see Figure 41. Let the element of surface be placed at the origin of coordinates. Let the magnitude of the mass velocity be $U$. Let this mass velocity be directed along the positive $X$ axis. Consider those molecules so directed as to strike the surface from below. (This choice of coordinates is dictated by convenience since $C_z$ will be positive for those molecules striking the plate and $C_x$ and $C_y$ will be positive for molecules having a line of flight directed into the first octant). $C_x$, $C_y$, and $C_z$ are the $x$, $y$, and $z$ components of the total molecular velocity relative to the fixed
coordinate system.

The number of molecules of all velocities striking unit area per unit time can be expressed as

\[ N_i = m A \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_{0}^{\infty} C_2 F_u \, dC_x \, dC_y \, dC_z \]  \hspace{1cm} (4.2)

where \( F_u \) is the Maxwell distribution with superimposed mass velocity,

\[ F_u = e^{-\beta^2 \left\{ (C_x - U)^2 + C_y^2 + C_z^2 \right\} } \]  \hspace{1cm} (4.3)

where \( n \) is the molecule number density of the gas, \( A = \frac{\Theta^3}{\pi^{3/2}} \), \( \beta = \frac{1}{V_m} \frac{m}{2 A^{1/2}} \), and \( V_m \) is the most probable speed. Each molecule brings up to the surface a component of the tangential momentum in the direction of the \( x \) axis having the value \( m C_x \). Thus the total \( x \) component of tangential momentum brought up to a unit area of surface per unit time by molecules of all velocities is given by

\[ G_x = m n A \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_{0}^{\infty} C_2 C_x F_u \, dC_x \, dC_y \, dC_z \]  \hspace{1cm} (4.4)

Integration yields

\[ G_x = \frac{m n \bar{C} U}{4} = \frac{m n V_m U}{2 \sqrt{\pi}} \]  \hspace{1cm} (4.5)

The calculation of \( G_r \), the magnitude of the net tangential momentum transferred from the surface to the issuing molecules, requires a little more work since in some fashion the direct experimental data must be applied. Consider the expression

\[ G_r = \int m N_i (c_i \Theta_i \phi_i) P_i (c_i \Theta_i \phi_i ; c_r \Theta_r \phi_r) \frac{c_i \Theta_i \phi_i ; c_r \Theta_r \phi_r}{c_r \Theta_r \phi_r} C_r \cos \phi_r \sin \Theta_r dC_i dC_r d\phi_i d\Theta_i d\phi_r d\Theta_r d\omega_r \]  \hspace{1cm} (4.6)
Here $P_i(C_i, \theta_i, \phi_i, C_r, \theta_r, \phi_r)$ is a distribution function denoting the probability per unit solid angle and speed range that a molecule with incident speed $C_i$ and direction $\Theta_i, \Phi_i$ will be scattered with speed $C_r$ in direction $\Theta_r, \Phi_r$. Here also $N_i(C_i, \Theta_i, \Phi_i)$ represents the number of incident molecules per unit speed range and solid angle having the speed $C_i$ and incident direction $\Theta_i, \Phi_i$ striking unit surface area per unit time. Thus Eq. (4.6) states that the net tangential momentum transferred from the surface in the $x$ direction per unit time can be represented by the integral over all speeds and directions of the number per unit time of molecules incident with given speed and direction multiplied by the probability that these molecules will be scattered with a given speed and direction multiplied by the $x$ component of that speed and also by the molecular mass.

Now consider how (4.6) may be specialized by application of the assumptions. By the terms of assumption (E-1) $G_r$ must be determined by the same speed distribution in a particular direction as that holding for molecules issuing from a gas in equilibrium at the temperature of the surface. However, this implies that this distribution in speed is independent of the probability that an issuing molecule will take a given direction as determined from the measured flux distributions of scattering from a real surface. The issuing distribution in speed is also assumed by (E-1) to be independent of the incident speed distribution. Under the terms of assumption B the probability that an issuing molecule will take a given direction is also assumed to be independent of the incident speed. Thus the probability distribution function $P$ may be rewritten as

$$P_i(C_i, \Theta_i, \Phi_i, C_r, \Theta_r, \Phi_r) = P_2(\Theta_i, \Phi_i, \Theta_r, \Phi_r)P_3(C_r, \Theta_r, \Phi_r)$$

It will be shown at a later point in this section that $P_2$ is independent of the issuing direction, i.e., $P_3 = P_3(C_r)$.

It will be recalled that the representative scattering shells (see section 3.3 and Figure 42) give scattering information with respect to a primed coordinate system with the $X'$ coordinate axis oriented along the incident ray. From these shells a probability
distribution function of the form \( P(\Theta_i, \Theta'_i, \Phi') \) can be computed. The function \( P(\Theta_i, \Theta'_i, \Phi') \) is assumed (assumption A) to be independent of the orientation of \( x' \) axis in the scattering surface and hence independent of \( \Phi'_i \). A change of variables can now be introduced. Letting \( \Phi' = \Phi_r + \Phi_i \) and \( \Theta' = \Theta_r \), \( P_z \) can be rewritten as

\[
P_z(\Theta_i, \Phi_i, \Theta'_r, \Phi'_r) = P_z(\Theta_i, \Theta'_r, \Phi_r + \Phi_i) = P_z(\Theta_i, \Theta'_i, \Phi')
\]

and equation (4.6) becomes

\[
G_r = \int_0^\pi N(\cos \Phi' \cos \Phi_i + \sin \Phi' \sin \Phi_i) \sin^2 \Theta' d\Theta' d\Phi'
\]

in which the angular integrations are now to be carried out over the variables \( \Theta_i, \Phi_i, \Theta'_i, \Phi'_i \).

The probability distribution \( P_z \) may be expressed in terms of the measured flux distributions as

\[
P_z(\Theta_i, \Theta'_i, \Phi') = \frac{A(\Theta_i, \Theta'_i, \Phi')}{\int_0^{2\pi} \int_0^{\pi} A(\Theta_i, \Theta'_i, \Phi') \sin \Theta' d\Theta' d\Phi'}
\]

where \( A(\Theta_i, \Theta'_i, \Phi') \) is the beam signal amplitude in arbitrary units, and hence is proportional to the molecule number flux through unit area of the unit scattering hemisphere in the direction \( \Theta'_i, \Phi'_i \) i.e., \( A(\Theta_i, \Theta'_i, \Phi'_i) = K N(\Theta_i, \Theta'_i, \Phi'_i) \) where \( K \) is a constant of proportionality.

The terms in the primed coordinates may be collected after expansion of \( \cos(\Phi' - \Phi) \). The integral over the primed coordinates may be written

\[
\int_0^{2\pi} \int_0^{\pi} P_z(\Theta_i, \Theta'_i, \Phi')(\cos \Phi' \cos \Phi_i + \sin \Phi' \sin \Phi_i) \sin^2 \Theta' d\Theta' d\Phi'
\]

\[
= \int_0^{2\pi} \int_0^{\pi} A(\Theta_i, \Theta'_i, \Phi') (\cos \Phi' \cos \Phi_i + \sin \Phi' \sin \Phi_i) \sin^2 \Theta' d\Theta' d\Phi'
\]

(4.9)
It should be noted that the scattering shells are symmetric about the $X'$ axis, i.e., $A(\theta, \phi, \phi') = A(\theta, \phi', \phi)$, an even function in $\phi'$ about $\phi' = 0$. Hence the portion of the integral (4.9)
\[
\int_0^{2\pi} \int_0^\pi A(\theta, \phi, \phi') \sin \phi' \sin \phi \sin \theta' \sin \phi' d\theta' d\phi'
\]
must have the value zero.

The remainder of (4.9) may be written as
\[
\cos \phi \int_0^{2\pi} \int_0^\pi A(\theta, \phi, \phi') \cos \phi' \sin \theta' \sin \phi' d\theta' d\phi' = \cos \phi \, g_f(\theta_i) \tag{4.10}
\]
The integrals of $g_f(\theta_i)$, defined by equation (4.10) must be evaluated numerically. Details of the numerical integration are given in Appendix B and a plot of the results is shown in Figure 44.

The evaluation of $P_3(C_f)$ must be considered more fully at this point. Consider a coordinate system $\xi_j, \gamma_j, \zeta$ immersed in a body of gas at rest having molecule number density $n_f$. The number of molecules per unit time crossing unit area of the $\xi_j, \gamma_j$ plane into solid angle $d\Omega$ making angle $\zeta$ with the $\zeta$ axis and having speed lying within the range $C_f$ to $C_f + dC_f$ is given by
\[
N_{f,C_f} d\Omega = n_f A_f C^2_f e^{-\beta_f^2 C^2_f} C_f \cos \zeta dC_f d\Omega
\]
Then the probability per unit solid angle that a molecule will have a speed $C_f$ lying within unit speed range will be
\[
P(C_f) = \frac{n_f A_f e^{-\beta_f^2 C^2_f} C^2_f \cos \zeta}{\int_0^{\infty} n_f A_f e^{-\beta_f^2 C^2_f} C^2_f \cos \zeta dC_f}
\]
If this fictitious gas is assumed to be thermal equilibrium with the surface then
\[
P(C_f) = P_3(C_f) = \frac{e^{-\beta_f^2 C^2_f} C^3_f}{\int_0^{\infty} e^{-\beta_f^2 C^2_f} C^3_f dC_f}
\]
An integration of $P(C_r) C_r dC_r$ over all speeds may be performed immediately such that

$$
\int_{0}^{\infty} P(C_r) C_r dC_r = \overline{C_r} = \frac{3\sqrt{\pi}}{4\beta_r} \tag{4.11}
$$

Equation (4.7) may now be rewritten as

$$
G_r = \int_{C, \Theta, \Phi} \overline{C_r} g_r(\Theta_i) \cos \phi_i \sin \Theta_i \times \delta C_i d\Theta_i d\phi_i \tag{4.12}
$$

Equation (4.12) may be looked at in a very direct and intuitive fashion. A review of the integral of Eq. (4.10) defining $g_r(\Theta_i)$ will show that the quantity $m \overline{C_r} g_r(\Theta_i)$ might be thought of as being the magnitude of the mean $x'$ component of tangential momentum per molecule transferred from unit area of the surface per unit time in the primed coordinate system by molecules incident at angle $\Theta_i$. Multiplication by $\cos \phi_i$, the angle made by the $x'$ axis with the $x$ axis of the original coordinate system, gives the value of the aforementioned momentum in the $x$ direction of the original coordinate system. Multiplication of this quantity by the number of molecules incident per unit time of all speeds from all directions and an integration overall variables yields the total magnitude of tangential momentum transferred from the surface.

To return to the original argument, if the integrand of equation (4.2) is put into polar coordinate form, there results, as the number of molecules incident on the surface per unit time from direction $\Theta$ and $\Phi$ in the intervals $d\Theta_i$ and $d\Phi_i$ having the speed $C_i$ in the range $dC_i$, the expression

$$
N_i(C_i, \Theta_i, \Phi_i) dC_i d\Theta_i d\Phi_i = \rho A C_i^3 \sin \Theta_i \cos \Theta_i 
\times e^{-\beta^2 \left\{ & (C_i \sin \Theta_i \cos \phi_i - U)^2 + C_i^2 \sin^2 \Theta_i \sin^2 \phi_i + C_i^2 \cos^2 \Theta_i \right\}} \tag{4.13}
$$
which reduces to
\[ N_i(c_i \theta_i \phi_i) d c_i d \theta_i d \phi_i = \int \int \int \int c_i^3 \sin \theta_i \cos \theta_i \times e^{-\beta^2 \{c_i^2 - 2 c_i U \sin \theta_i \cos \phi_i + U^2 \}} d c_i d \theta_i d \phi_i \]

Then
\[ G_r = \frac{m \cdot m \cdot 3 \pi \cdot A}{4 \beta_r} \int \int \int c_i^3 g_r(\theta_i) \cos \theta_i \sin \theta_i \cos \phi_i \times e^{-\beta^2 \{c_i^2 - 2 c_i U \sin \theta_i \cos \phi_i + U^2 \}} d c_i d \theta_i d \phi_i \quad (4.14) \]

Performing the indicated integrations over \( c_i \) and \( \phi_i \) we have,
\[ G_r(\theta_i) = \frac{m \cdot m \cdot 3 \pi \cdot e^{1/2 - s^2}}{16 \beta_r \beta} \cos \theta_i \sin \theta_i g_r(\theta_i) \times \left[(3 b + 2 b^3) I_0(b) + (b + 2 b^3) I_1(b)\right] \quad (4.15) \]

where \( b = s \sin \theta_i \) and \( I_0 \) and \( I_1 \) are modified Bessel functions and \( s(s = \frac{U}{V}) \) is the molecular speed ratio.

The details of the integration above are given in Appendix A and \( G_r(\theta_i) \) is plotted as a function of \( \theta_i \) for \( s = 0.1, 1.0, 2.0, \) and \( 10.0 \) as given in Figure 43.

The final integration over \( \theta_i \) must be performed numerically of course. One can observe immediately, however, that for \( g_r(\theta_i) = 0 \) (diffuse reflection) \( G_r \) will equal 0 and \( f = 1 \).

Thus, subject to the limitations of the data and of the assumptions, \( f \) can be calculated for any molecular speed ratio, \( s \), from the direct measurement of the scattered flux. Such calculations were carried out for the case of the unpolished glass surface for values of \( s = 0.1, 1.0, 2.0, \) and \( 10.0 \) and \( \beta_r = \beta \).

A plot of the resultant findings showing the behavior of \( f \) as a function of \( s \) is shown in Figure 45.

4.3 Calculation of \( f \) under Terms of Assumption E-2
The main work of this section is again the calculation of $G_{r2}$, the magnitude of the tangential momentum transferred from the gas to the surface under the terms of assumption E-2. With the exception of this latter assumption, all assumptions are as outlined in section 4.1. The same flow situation is also assumed. $G_i$ will remain as calculated in Eq. 4.5.

Consider now again how the probability distribution function of Eq. (4.6) might be appropriately specialized. According to the terms of assumption E-2 two alternate classes of gas-surface interaction occur. The scattered molecules may be either diffusely scattered according to the cosine distribution with speed distribution as in assumption E-1 or they may be specularly scattered from tilted surface elements. The total probability distribution may be represented as the sum of the probability distributions for these two alternate events. Eq. 4.6 may be rewritten as follows:

$$G_{r2} = \sum_{m} N_i(C; \Theta; \Phi) \left\{ P_{1D}(C; \Theta; \Phi; C_r, \Theta_r, \Phi_r) C_r + P_{1S}(C; \Theta; \Phi; C_r, \Theta_r, \Phi_r) C_r \right\} \cos \Phi_r \sin^2 \Theta_r \sin \Theta_i \sin \Theta_i$$

where

$$P_{1D}(C; \Theta; \Phi; C_r, \Theta_r, \Phi_r)$$

identifies the probability function associated with the diffusely scattered molecules while the subscript $S$ identifies that associated with the specularly scattered molecules.

The probability distribution function $P_{1S}(C; \Theta; \Phi; C_r, \Theta_r, \Phi_r)$ can be specialized immediately to the form since the angular distribution of $C_i$ (assumption B) and for these molecules $C_r$ can have only the value $C_i$. Thus equation 4.16 can be rewritten as

$$G_{r2} = \sum_{m} N_i(C; \Theta; \Phi) \left\{ P_{2D}(C; \Theta; \Phi') P_{3D}(C_r) C_r + P_{2S}(\Theta; \Theta; \Phi') C_i \right\} \cos (\Phi' - \Phi_i) \sin^2 \Theta' \sin \Theta' \sin \Theta_i$$

where

$$P_{2D}(C; \Theta; \Phi')$$

identifies the probability function associated with the diffusely scattered molecules while the subscript $S$ identifies that associated with the specularly scattered molecules.
Recalling again assumption (E-2), it can be seen that \( A(\Theta, \Theta', \Phi') \), the beam signal amplitude (\( A(\Theta, \Theta', \Phi') = K N(\Theta, \Theta', \Phi') \) as before), can be written

\[
A(\Theta, \Theta', \Phi') = A_D \cos \Theta' + A_S(\Theta, \Theta', \Phi')
\]

where \( A_D \) is a constant representing the diameter of the assumed diffuse scattering sphere and \( A_S(\Theta, \Theta', \Phi') \) are the amplitude differences between this sphere and the total amplitudes of the specular bulges. From the data then, \( P_{2D}(\Theta, \Theta', \Phi') \) can be found as indicated below:

\[
P_{2D}(\Theta, \Theta', \Phi') = \frac{A_D \cos \Theta'}{\int \int A(\Theta, \Theta', \Phi') \sin \Theta' \, d\Omega' \, d\Phi'} \quad (4.18)
\]

while \( P_{2S}(\Theta, \Theta', \Phi') \) becomes

\[
P_{2S}(\Theta, \Theta', \Phi') = \frac{A_S(\Theta, \Theta', \Phi')}{\int \int A(\Theta, \Theta', \Phi') \sin \Theta' \, d\Omega' \, d\Phi'} \quad (4.19)
\]

Extracting the integrals over the primed coordinates which apply to \( P_{2S} \) from (4.17), we have

\[
\int \int \frac{A_D \cos \Theta'[\cos \Phi' \cos \Theta_i + \sin \Phi' \sin \Theta_i]}{\int \int A(\Theta, \Theta', \Phi') \sin \Theta' \, d\Omega' \, d\Phi'} \, d\Omega' \, d\Phi' \quad (4.20)
\]

which has the value zero.

The remainder of the integration over the primed coordinates may be written

\[
\int \int \frac{A_S(\Theta, \Theta', \Phi') \{\cos \Phi' \cos \Theta_i + \sin \Phi' \sin \Theta_i\}}{\int \int A(\Theta, \Theta', \Phi') \sin \Theta' \, d\Omega' \, d\Phi'} \, d\Omega' \, d\Phi' \quad (4.21)
\]
For the same reasons of symmetry as before, the integration over the second section of the numerator (i.e., over the section \( A_2(\Theta_i, \Theta', \Phi) \sin \Phi' \cos \Theta' \sin \Theta \)) makes no contribution to the total values of the integral. Further, we can add \( A_2 \cos \Theta' \) to \( A_2(\Theta_i, \Theta', \Phi) \) without changing the value of the integral. Thus the expression 4.21 can be re-expressed as

\[
\frac{\cos \Phi_i}{2\pi} \int_{0}^{2\pi} A(\Theta_i, \Theta', \Phi') \cos \Phi' \sin \Theta' \, d\Theta' \, d\Phi'
\]

\[
\int_{0}^{2\pi} A(\Theta_i, \Theta', \Phi') \sin \Theta' \, d\Theta' \, d\Phi' \tag{4.22}
\]

The \( q_r(\Theta) \) as defined here is the same as that defined by Eq. 4.10 of the previous section.

Equation 4.17 may now be rewritten as

\[
G_{r2} = \sum_{i} N_i (C_i, \Theta_i, \Phi_i) C_i q_r(\Theta_i) \cos \Phi_i \sin \Theta_i \, dC_i \, d\Theta_i \, d\Phi_i \tag{4.23}
\]

If now the function \( N_i (C_i, \Theta_i, \Phi_i) \) is expressed in polar coordinate form as in equation 4.13, and if the integrand of 4.23 is multiplied by \( \frac{\sin \Theta_i}{\sin \Theta_i} \), \( q_r(\Theta) \) becomes

\[
G_{r2} = \sum_{i} \left( \frac{q_r(\Theta_i)}{\sin \Theta_i} \right) \left[ -2 \frac{C_i^2}{U \sin \Theta_i \cos \Phi_i + U^2} \right] \frac{1}{c \beta^2} \sin \Theta_i \cos \Theta_i \, dC_i \, d\Theta_i \, d\Phi_i \tag{4.24}
\]

or

\[
G_{r2} = \int_{0}^{\pi} \frac{q_r(\Theta_i)}{\sin \Theta_i} \, d\Theta_i \tag{4.25}
\]

where \( q_r(\Theta_i) \) is the integral over \( C_i \) and \( \Phi_i \) of the quantity within the brackets of Eq. (4.24). Although this grouping might appear rather arbitrary it is justified by the fact that \( q_r(\Theta_i) \) has an interesting physical interpretation since the function represents
the net \( x \) component of momentum incident on the surface per unit time, unit area and unit polar angle at angle \( \Theta \) resulting from molecules of all speeds coming from all azimuth angles.

The result of integration over \( \Theta \) and \( \Phi \) can be written in the form

\[
G_i(\Theta) = \frac{m}{m} \frac{V_m^2}{\sin \Theta} \left( \frac{b}{\cos \Theta} \right)^2 \exp \left( \frac{-b^2}{\cos \Theta} \right)
\]

\[
\times \left[ (S''+1)(b^3 + 2b) \right]
\]

(4.26)

where

\[
S'' = S' b e^{b^2} \quad b = s \sin \Theta \quad s = \frac{U}{V_m}
\]

and

\[
S' = \int_0^{2\pi} \cos \Phi \text{ Erf}(b \cos \Phi) e^{-b^2 \sin^2 \Phi} d\Phi
\]

(4.27)

Details of the integration may be found in Appendix C and a plot of \( S'' \) may be found in Figure 46 giving \( S'' \) for values of \( b \) from 0 to 10. Also \( \frac{G_i(\Theta)}{m} \frac{V_m^2}{\sin \Theta} \) has been plotted as a function of \( \Theta \) for \( s = 0.1, 1.0, 2.0, \) and 10.0; see Figure 47.

Thus again, subject to the limitations of the assumptions, \( f \) can be calculated for any molecular speed ratio. Such calculations were performed for the unpolished glass surface for \( s = 0.1, 1.0, 2.0 \), and 10.0 and a plot of the results is shown in Figure 45. The final integration over \( \Theta \) in the evaluation of \( G_r^2 \) had to be performed numerically, of course.

4.4 Calculation of \( f \) for Pure Specular Reflection at Grazing Incidence

A consideration of Figure 47 will show that molecules incident at angles close to grazing, play an increasingly important part in the momentum transfer to the surface as the molecular speed ratio increases. It has already been argued that at some incident angle very close to grazing a condition of pure specular reflection may set in. (i.e., the reflected molecule has direction \( \Phi \) and \( \Pi - \Theta \).)
and has speed $C_r = C_4$. All molecules at some angle $\Theta_o$ on to $90^\circ$ will be assumed to be specularly reflected for the purpose of this speculative calculation while all molecules incident form $\Theta_o$ to $\Theta_1$ will be assumed to be diffusely reflected according to the cosine law and thus contribute nothing to the momentum transfer from the surface. $f$ for this situation can be very simply formulated as

$$f = 1 - \sqrt{\frac{2}{\pi}} \int_{\Theta_o}^{\Theta_1} G_i(\Theta_i) \, d\Theta_i$$

where the tangential momentum transferred from the surface is equal in magnitude to $G_i(\Theta_i)$ integrated from $\Theta_o$ to $\Theta_1$. Calculations of $f$ were performed for $\Theta_o$ assuming various values from $85^\circ$ to $90^\circ$ and for $s$, the molecular speed ratio taking on the values $s = 0.1, 1.0, 2.0$ and $10.0$. The results of these calculations are plotted in Figure 48.

4.5 **Discussion**

Figure 45 shows the behavior of $f$ for variations of the parameter $s$ for both assumed cases of molecular interaction E-1 and E-2. The diverging behavior of these plots lends plausibility to the intuition which had originally led to assumptions E-1 and E-2; namely that these assumptions were bracketing assumptions and that more complete knowledge of the scattering probability distribution function would probably lead to values of $f$ for this surface lying in between these extremes.

It is possible that the relative validity of assumptions E-1 and E-2 leading to the divergent results above might be assessed by the performance of a rotating cylinder experiment at molecular speed ratios of 2 or more. Unfortunately, such an experiment would be difficult to perform because of the high peripheral speeds involved and the delicate drag measurements necessary to resolve the rather small differences in $f$. A much more reasonable approach would be, as suggested in section 3.4, to introduce a velocity selector into the path of the scattered molecules in a molecular beam experiment and measure directly the reflected speed distributions.
It is interesting to note that neither calculation based on the alternate assumptions (3-1) and (3-2) yielded values of $f$ nearly low enough for agreement with the reported value of $f = 0.89$ for air on glass. Such a value is clearly beyond the limit of error of these present experiments since it could only result from much larger forward distortions of the scattering shells. Millikan, in arriving at his value of $f$ (above) re-examined the earlier work of Knudsen and of Warburg on slip in glass capillaries. Warburg worked with air on glass while Knudsen confined his investigations to $\text{H}_2$, $\text{O}_2$ and $\text{CO}_2$. It is barely possible that the somewhat higher pressures at which these early experiments were performed has some effect on the adsorbed gas layer thickness, which in turn may have allowed more "specular" scattering. It is also likely that some errors could be expected as a result of using the findings of experiments on gases other than air to make predictions for the air-glass surface interaction. However, no explanation has been found which properly accounts for the discrepancy between Millikan's results and that of this report.

The results of the calculation of section 4.4 are also quite interesting since they serve to show the importance of the gas-surface behavior at near glancing angles of incidence in determining the overall momentum transfer. These results also serve to show the importance of extending the molecular beam scattering experiments to include measurements at angles extremely close to glancing. It need hardly be stated that an engineering surface which exhibited some considerable degree of specular reflection at a few degrees from glancing incidence would find immediate application in high altitude, high speed aerodynamics.

It is hoped that the calculations of this section will suggest a new approach to the problems of evaluating the boundary conditions for rarefied gas flows. It seems possible that the old concepts of a constant $f$ and a constant $\alpha$, the energy transfer counterpart of $f$, might profitably be supplanted by the more general scattering probability function.

5.0 SUMMARY

A molecular beam apparatus was used to investigate the scattering of air and $\text{N}_2$ molecules from surfaces of polished low carbon steel,
etched low carbon steel, polished aluminum, unpolished window glass, and polished window glass. A beam of molecules was directed against the surfaces at representative angles of incidence. Polar flux distributions of the scattered molecules were measured by a movable detector. An ionization gage beam detector system was devised to accomplish this measurement.

Spatial polar plots of these measured flux distributions were found to approximate closely the form of the cosine scattering distribution for the cases of the steel and aluminum surfaces. Similar plots in the case of the glass surfaces were found to have well defined but small bulging deviations from the cosine scattering shape on the side away from the incident beam. These findings did not serve to distinguish the degree of elasticity of the interaction since both elastic and inelastic interaction models would serve to produce pure cosine scattering, and only a small degree of elasticity would suffice in some models to account for the observed deviation from cosine scattering. Among others, a modified Maxwell model would seem to satisfy the available data. In this, a fraction of the incident molecules undergo a diffuse scattering, being more or less accommodated in energy to the surface, while the remainder are specularly reflected from tilted surface elements.

In order to illustrate one aerodynamic application of scattering data obtained by molecular beam techniques, a procedure was devised for the computation of the coefficient of specular reflection, $f$, and values of this coefficient were calculated for the case of an unpolished glass surface oriented tangentially to the direction of mass motion in a free molecule flow. These values were calculated as a function of the parameter $s$, the molecular speed ratio. If all scattered molecules are assumed to possess the distribution in speed in a particular direction which they would have had had they issued from a gas in equilibrium at the temperature of the surface, $f$ takes on values of $\sim 0.97$ as $s$ approaches 0 to $\sim 0.99$ at $s = 10$. Alternatively, if the modified Maxwell surface interaction model is assumed, $f$ takes on values of $\sim 0.97$ as $s$ approaches 0 to $\sim 0.93$ at $s = 10$. These values do not agree with the published value of $f$ of 0.89 for air on glass. This lack of agreement could not be accounted for by a consideration of experimental errors.

A speculative calculation was performed showing the effect on $f$ for molecular speed ratios of 0.1, 1.0, 2.0, and 10.0, in the case where
a condition of pure specular reflection occurs from source angle, $\Theta_0$, close to grazing incidence on to 90° incidence, $f$ was shown to have the value 0.5 for $s = 10$ and $\Theta_0 = 85^\circ$.

Three points of departure for further experimental work were discussed. A. The substitution of mercury diffusion pumps for the present oil diffusion pumps would remove one important source of surface contamination. B. Higher resolution experiments using a slit-form beam geometry would yield interesting data on the behavior of the scattering at incident angles close to grazing. C. A velocity selector introduced into the scattered beam would afford a means of determining the relative validity of the assumptions leading to the computed values of $f$. This device would also afford a means of studying in detail the energy exchange between gas and surface. This latter work would provide a valuable basis for a better physical understanding of the basic scattering mechanisms.

ACKNOWLEDGMENT

I wish to acknowledge the assistance of those whose ideas and skills have contributed to this project. It would be impossible to mention all to whom I am indebted but I specifically want to thank Prof. L. B. Loeb for his guidance as chairman of my research committee and Dr. I. Estermann for his unfailing advice, interest and encouragement.
**REFERENCES**

<table>
<thead>
<tr>
<th></th>
<th>Author</th>
<th>Title</th>
<th>Publisher/Year</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>A. Kundt,</td>
<td>&quot;Uber Reibung und Wärmeleitung, verdunnter Gase&quot;, Phil. Mag. 50,</td>
<td></td>
</tr>
<tr>
<td></td>
<td>E. Warburg</td>
<td>(1875)</td>
<td></td>
</tr>
<tr>
<td>3</td>
<td>S. A. Schaaf,</td>
<td>&quot;The Flow of Rarefied Gases&quot;, High Speed Aerodynamics and Jet Propulsion</td>
<td></td>
</tr>
<tr>
<td>5</td>
<td>M. Knudsen</td>
<td>&quot;Die molikulare Wärmeleitung der Gase und der Akkommodations koef-</td>
<td></td>
</tr>
<tr>
<td></td>
<td>O. Stern</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>O. Stern</td>
<td>p 779.</td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td>(1915) p 1113.</td>
<td></td>
</tr>
</tbody>
</table>
12) E. Blankenstein
- "Coefficient of Slip and Momentum Transfer in H₂, He, air and O₂"
13) S. F. Chiang
- "Drag on a Rotating Cylinder at Low Pressures", Inst. of Engr. Res.,
14) M. Knudsen
15) R. N. Oliver,
    M. Farber
- "Experimental Determination of Accommodation Coefficients as Function
  of Temperature for Several Metals on Gases", Jet Propulsion Lab., Calif.
16) M. L. Weidmann
- "Thermal Accommodation Coefficients",
  Trans. ASME, 68, p 57 (1946).
17) J. K. Roberts
- "The Exchange of Energy Between Gas Atoms and Solid Surfaces",
  Proc. Roy. Soc. (London) A 129 p 146 (1930) and
  A 135 p 135 (1932).
18) W. C. Michels
- "Accommodation Coefficients of Helium and Argon against Tungsten",
19) J. B. Taylor
- "Reflection of Beams of Alkali Metals from Crystals",
20) A. Ellet,
    H. F. Olson
- "The Reflection of Atoms by a Crystal",
21) A. Ellet,
    H. F. Olson,
    H. A. Zahl
- "The Reflection of Atoms from Crystals",
22) H. A. Zahl

23) T. H. Johnson

24) I. Estermann, R. Frisch, O. Stern

25) G. J. Maslach

26) F. C. Hurlbut

27) F. C. Hurlbut

28) F. C. Hurlbut

29) R. T. Bayard, D. Alpert
<table>
<thead>
<tr>
<th></th>
<th>Authors</th>
<th>Title</th>
<th>Journal/Volume/Publication</th>
<th>Page</th>
</tr>
</thead>
<tbody>
<tr>
<td>30</td>
<td>W. L. Greenough, W. E. Williams, J. K. Taylor</td>
<td>&quot;Regulated Low Voltage Supply for Electrolysis and Other Uses&quot;</td>
<td>RSI 22 (1951)</td>
<td>484</td>
</tr>
<tr>
<td>32</td>
<td>A. Q. Mier, E. P. Ney, M. G. Ingraham</td>
<td>&quot;A Null Method for the Comparison of Two Ion Currents in a Mass Spectrometer&quot;</td>
<td>RSI 18 (1947)</td>
<td>294</td>
</tr>
<tr>
<td>33</td>
<td>G. L. Mellen</td>
<td>&quot;Radium Type Vacuum Gage&quot;</td>
<td>Electronics, April 1946</td>
<td></td>
</tr>
<tr>
<td>34</td>
<td>B. Baule</td>
<td>&quot;Theoretische Behandlung der Erscheinungen in Verdunnten Gasen&quot;</td>
<td>Ann. Physik 44 (1914)</td>
<td>145</td>
</tr>
</tbody>
</table>
APPENDIX A

To integrate over $C_i$ and $\phi_i$

$$G_r = \frac{3 \pi A}{4 \beta r} \int_0^{2\pi} \int_0^\infty C_i^3 G_r(\theta_i) \cos \theta_i \sin \theta_i \cos \phi_i$$

$$\times e^{-\beta \left\{ C_i^2 - 2C_i \mu \sin \theta_i \cos \phi_i + \mu^2 \right\}} dC_i d\theta_i d\phi_i$$  \hspace{1cm} (A-1)

(Eq. 4.14 = A-1)

let $P = 2 \mu \sin \theta \cos \phi$

and $P/\beta = \frac{\mu}{V_m} \sin \theta \cos \phi = b \cos \phi$

Making the above substitution, completing the square in the exponent, and letting

$$\frac{3 \pi A}{4 \beta r} \cos \theta \sin \theta \cos \phi = M$$

there results

$$G_r(\theta) d\theta = M d\theta \int_0^{2\pi} e^{-\beta^2 (U^2 - \frac{P^2}{4})} C^3 e^{-\beta^2 (C - \frac{P}{2})^2} dC$$

Integration over $C$ yields

$$G_r(\theta) d\theta = M d\theta \int_0^{2\pi} e^{-\beta^2 (U^2 - \frac{P^2}{4})} \left\{ \frac{P^2}{8 \beta^3} \left[ 1 + \text{ERF}(\frac{P}{2}) \right] \right\}$$

$$+ \left( \frac{P^2}{4} + \frac{1}{\beta^2} \right) \frac{1}{2 \beta^2} \left\{ e^{-\frac{P^2}{4 \beta^2}} + \frac{\pi \beta^2}{2} \left[ 1 + \text{ERF}(\frac{P}{2}) \right] \right\} d\phi$$  \hspace{1cm} (A-2)

Multiplying out and collecting terms it was found that the integrals over $\phi$ not equal to zero are:

$$\frac{3 \pi A}{4 \beta^4} \int_0^{2\pi} \cos^2 \phi e^{\frac{b^2 \cos^2 \phi}{\beta^4}} d\phi$$

$$= 3 \frac{\pi b^2}{4 \beta^4} b \left[ I_0 \left( \frac{b^2}{2} \right) + I_1 \left( \frac{b^2}{2} \right) \right]$$  \hspace{1cm} (A-3)

and

$$\frac{b^3 \pi A}{4 \beta^4} \int_0^{2\pi} \cos^4 \phi e^{\frac{b^2 \cos^2 \phi}{\beta^4}} d\phi$$

$$= \frac{\pi b^3}{4 \beta^4} \left[ I_0 \left( \frac{b^2}{2} \right) + \left( 1 - \frac{1}{b^2} \right) I_1 \left( \frac{b^2}{2} \right) \right]$$  \hspace{1cm} (A-4)
$I_0\left(\frac{b}{2}\right)$ and $I_1\left(\frac{b}{2}\right)$ are the zeroth and first order modified Bessel functions of the first kind.

A final collection of terms yields the result of Eq. 4.15:

$$G_r(\Theta_i) = \frac{m m 3 V \pi}{16 \pi r^3 \beta} e^{\frac{b^2}{\beta}} \cos \Theta_i \sin \Theta_i \ g_r(\Theta_i)$$

$$\times \left[ (3 b + 2 b^2) I_0\left(\frac{b^3}{2}\right) + (b + 2 b^3) I_1\left(\frac{b^3}{2}\right) \right]$$
APPENDIX B

In order to perform the indicated integration of Eq. 4.11 i.e.
\[ \int_{0}^{2\pi} \int_{0}^{\pi} A(\theta, \theta', \phi) \cos \phi' \sin^{2} \theta' \theta' \phi' \ d\theta' \ d\phi' \]
and
\[ \int_{0}^{2\pi} \int_{0}^{\pi} A(\theta, \theta', \phi) \sin \theta' \theta' \phi' \ d\theta' \ d\phi' \]
the indicated integrations must be replaced by summations and the summation carried out over the surface of a unit hemisphere. Thus the flux through a surface element is approximated by
\[ A_{m}(\theta, \theta', \phi') \Delta a_{m} \]
where
\[ \sum_{m=0}^{n} \Delta a_{m} = \alpha = 2 \pi (\text{unit sphere}) \]
A simple integration scheme was devised in which the total unit hemisphere was divided into 414 elementary surfaces of approximately equal area. Since the scattering distributions were assumed symmetrical about the incident plane, just half of this number were employed in the actual integration. Since all scattering shells were regarded as spherical on the incident ray side, this section of the integration was performed only once. For a representation of the division of the unit hemisphere surface see Figure 49.

The \( A_{m}(\theta, \theta', \phi') \) were taken from the representations of the scattering shells at the centerpoint of each elementary area. The resultant \( \eta \phi_{r}(\theta) \) are illustrated in Figure 44.

A table of the \( \Delta a_{m} \) of this integration scheme follows:

<table>
<thead>
<tr>
<th>BOUNDING POLAR ANGLES OF SPHERICAL ZONE</th>
<th>NUMBER OF ELEMENTARY SURFACES IN SPHERICAL ZONE</th>
<th>AREA OF SURFACE ELEMENT</th>
</tr>
</thead>
<tbody>
<tr>
<td>90° - 85°</td>
<td>36</td>
<td>0.015212</td>
</tr>
<tr>
<td>85° - 80°</td>
<td>36</td>
<td>0.015048</td>
</tr>
<tr>
<td>80° - 75°</td>
<td>36</td>
<td>0.014866</td>
</tr>
<tr>
<td>75° - 70°</td>
<td>36</td>
<td>0.014520</td>
</tr>
<tr>
<td>70° - 65°</td>
<td>36</td>
<td>0.014068</td>
</tr>
<tr>
<td>65° - 60°</td>
<td>36</td>
<td>0.013503</td>
</tr>
<tr>
<td>60° - 55°</td>
<td>36</td>
<td>0.012843</td>
</tr>
<tr>
<td>55° - 50°</td>
<td>36</td>
<td>0.012083</td>
</tr>
<tr>
<td>BOUNDING POLAR ANGLES OF SPHERICAL ZONE</td>
<td>NUMBER OF ELEMENTARY SURFACES IN SPHERICAL ZONE</td>
<td>AREA OF SURFACE ELEMENT</td>
</tr>
<tr>
<td>----------------------------------------</td>
<td>-----------------------------------------------</td>
<td>------------------------</td>
</tr>
<tr>
<td>50° - 40°</td>
<td>36</td>
<td>0.024586</td>
</tr>
<tr>
<td>40° - 30°</td>
<td>36</td>
<td>0.017455</td>
</tr>
<tr>
<td>30° - 20°</td>
<td>30</td>
<td>0.015425</td>
</tr>
<tr>
<td>20° - 10°</td>
<td>18</td>
<td>0.015752</td>
</tr>
<tr>
<td>10° - 0°</td>
<td>6</td>
<td>0.015909</td>
</tr>
</tbody>
</table>
APPENDIX C

To evaluate the net incident tangential momentum $G_i'(\Theta_i)$ as a function of $\Theta_i$ where

$$G_i(\Theta_i) = \frac{m m' G^3}{\pi \beta^3} \sin^2 \Theta_i \cos \Theta_i \int_0^{2\pi} \int_0^{\infty} c_i^+ \cos \phi \left\{ -\beta^3 \left[ c_i^2 - 2 c_i U \sin \Theta_i \cos \phi + U^2 \right] e^{-\beta^3 (U^2 - \frac{P^2}{4})} + c_i^+ \cos \phi \right\} \, dc_i \, d\phi_i \quad (C-1)$$

(From Eqs. 4.24 and 4.25)

let $P = U \sin \Theta \cos \phi$

and $P/\beta = U \sin \Theta \cos \phi = b \cos \phi$

as in Appendix A.

Making the above substitution, completing the square in the exponent and letting

$$\frac{m m' G^3}{\pi \beta^3} \sin^2 \Theta \cos \Theta = M'$$

there results

$$G_i(\Theta) = M' \int_0^{2\pi} \cos \phi \left\{ -\beta^3 (U^2 - \frac{P^2}{4}) \right\} \left[ + \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \nu \n
\[
\frac{\sqrt{2\pi}}{2\beta^3} e^{-s^2/2\beta^2} \int_0^{2\pi} \cos^2 \phi \left[ \text{erf}(b\cos \phi) e^{-b^2 \sin^2 \phi} \right] d\phi
\]

\[
= \frac{3\sqrt{2\pi}}{2\beta^3} e^{-s^2/2\beta^2} \left\{ \frac{(2\beta^2 - 1)}{2\beta^2} \int_0^{2\pi} 2b \sqrt{2\pi} e^{-b^2} \right\} \]

\[
= \frac{\sqrt{2\pi}}{2\beta^3} e^{-s^2/2\beta^2} \left\{ \int_0^{2\pi} \left[ 1 - \frac{1}{b^2} + \frac{3}{4b^4} \right] + \pi e^{-b^2} \left[ \frac{s}{4b} - \frac{3}{2b^3} \right] \right\}
\]

Equations (6-7) and (6-7) proceed by parts from (6-5). A collection and cancellation of terms yields

\[
G_{\theta}(\theta) = \frac{M}{\beta^3} e^{-s^2} \left[ \int_0^{2\pi} (b^2 + b^4 + \pi b^3 + 2\pi b) \right]
\]

If now \( S' \) is defined such that \( S' = \frac{b e^{-b^2}}{2\pi} \int_0^{2\pi} \cos \phi \left[ \text{erf}(b\cos \phi) e^{-b^2 \sin^2 \phi} \right] d\phi \)

a final substitution yields the results of Eq. 4.26:

\[
G_{\theta}(\theta) = \frac{M \sin^2 \theta}{\beta^3} e^{-s^2} \left[ (S' + 1)(b^3 + 2b) \right]
\]

There remains the task of evaluating \( S' \) or \( S \). This integral did not appear reducible to previously tabulated functions. A numerical integration was accomplished as shown below:

\[
S' = \int_0^{2\pi} \cos \phi \left[ \text{erf}(b\cos \phi) e^{-b^2 \sin^2 \phi} \right] d\phi
\]

Let \( x = b \cos \phi \), then \( d\phi = -\frac{dx}{\sqrt{b^2 - x^2}} \)

The appropriate transformation of the limits and an integration by parts makes possible the expression of \( S' \) as

\[
S' = \int_0^b \left[ (b^3 + \frac{4}{\sqrt{b^2 - x^2}} \text{erf}(x) e^{-x^2} \right] dx
\]

Applying the definition of \( S' \) there results

\[
S' = b^2 + \frac{4}{\sqrt{2\pi}} \int_0^b \sqrt{b^2 - x^2} \text{erf}(x) e^{-x^2} dx
\]

In this form \( S' \) was computed for values of \( b \) sufficient for the
further computation of \( \frac{G_i(t)}{\ln V_m} \). A table of the computed values of \( S' \) and \( \log S' \) follows and a plot of these values is given in Fig. 46.

<table>
<thead>
<tr>
<th>( b )</th>
<th>( S' )</th>
<th>( \log S' )</th>
</tr>
</thead>
<tbody>
<tr>
<td>0</td>
<td>0.0000</td>
<td>-\infty</td>
</tr>
<tr>
<td>0.005</td>
<td>0.00249</td>
<td>-2.6038</td>
</tr>
<tr>
<td>0.1</td>
<td>0.01005</td>
<td>-1.9978</td>
</tr>
<tr>
<td>0.15</td>
<td>0.02276</td>
<td>-1.6428</td>
</tr>
<tr>
<td>0.2</td>
<td>0.04081</td>
<td>-1.3892</td>
</tr>
<tr>
<td>0.25</td>
<td>0.06449</td>
<td>-1.1905</td>
</tr>
<tr>
<td>0.3</td>
<td>0.09417</td>
<td>-1.0261</td>
</tr>
<tr>
<td>0.35</td>
<td>0.13032</td>
<td>-0.8850</td>
</tr>
<tr>
<td>0.4</td>
<td>0.17351</td>
<td>-0.7607</td>
</tr>
<tr>
<td>0.45</td>
<td>0.22446</td>
<td>-0.6489</td>
</tr>
<tr>
<td>0.5</td>
<td>0.28402</td>
<td>-0.5466</td>
</tr>
<tr>
<td>0.55</td>
<td>0.35322</td>
<td>-0.4520</td>
</tr>
<tr>
<td>0.6</td>
<td>0.43330</td>
<td>-0.3632</td>
</tr>
<tr>
<td>0.65</td>
<td>0.52571</td>
<td>-0.2792</td>
</tr>
<tr>
<td>0.7</td>
<td>0.63219</td>
<td>-0.1992</td>
</tr>
<tr>
<td>0.75</td>
<td>0.75480</td>
<td>-0.1222</td>
</tr>
<tr>
<td>0.8</td>
<td>0.89598</td>
<td>-0.0477</td>
</tr>
<tr>
<td>0.85</td>
<td>1.05864</td>
<td>0.0248</td>
</tr>
<tr>
<td>0.9</td>
<td>1.24023</td>
<td>0.0956</td>
</tr>
<tr>
<td>0.95</td>
<td>1.46284</td>
<td>0.1652</td>
</tr>
<tr>
<td>1.0</td>
<td>1.72331</td>
<td>0.2338</td>
</tr>
<tr>
<td>1.25</td>
<td>3.772</td>
<td>0.5766</td>
</tr>
<tr>
<td>1.5</td>
<td>8.478</td>
<td>0.9283</td>
</tr>
<tr>
<td>1.75</td>
<td>20.37</td>
<td>1.309</td>
</tr>
<tr>
<td>2.0</td>
<td>53.58</td>
<td>1.729</td>
</tr>
<tr>
<td>4.0</td>
<td>8.891 \times 10^6</td>
<td>6.949</td>
</tr>
<tr>
<td>10.0</td>
<td>2.69 \times 10^{43}</td>
<td>43.43</td>
</tr>
</tbody>
</table>
**TABLE I**

**COEFFICIENT OF SPECULAR REFLECTION f**

<table>
<thead>
<tr>
<th>GAS</th>
<th>SURFACE</th>
<th>f</th>
<th>REFERENCE</th>
</tr>
</thead>
<tbody>
<tr>
<td>air</td>
<td>machined brass</td>
<td>1.00</td>
<td>11</td>
</tr>
<tr>
<td>CO₂</td>
<td>machined brass</td>
<td>1.00</td>
<td>11</td>
</tr>
<tr>
<td>air</td>
<td>old shellac</td>
<td>1.00</td>
<td>11</td>
</tr>
<tr>
<td>CO₂</td>
<td>old shellac</td>
<td>1.00</td>
<td>11</td>
</tr>
<tr>
<td>air</td>
<td>Hg</td>
<td>1.00</td>
<td>11</td>
</tr>
<tr>
<td>air</td>
<td>oil</td>
<td>0.90</td>
<td>11</td>
</tr>
<tr>
<td>CO₂</td>
<td>oil</td>
<td>0.92</td>
<td>11</td>
</tr>
<tr>
<td>H₂</td>
<td>oil</td>
<td>0.93</td>
<td>11</td>
</tr>
<tr>
<td>air</td>
<td>glass</td>
<td>0.89</td>
<td>11</td>
</tr>
<tr>
<td>He</td>
<td>oil</td>
<td>0.87</td>
<td>11</td>
</tr>
<tr>
<td>air</td>
<td>fresh shellac</td>
<td>0.79</td>
<td>11</td>
</tr>
<tr>
<td>air</td>
<td>Ag₂O</td>
<td>0.98</td>
<td>12</td>
</tr>
<tr>
<td>He</td>
<td>Ag₂O</td>
<td>1.00</td>
<td>12</td>
</tr>
<tr>
<td>H₂</td>
<td>Ag₂O</td>
<td>1.00</td>
<td>12</td>
</tr>
<tr>
<td>O₂</td>
<td>Ag₂O</td>
<td>0.99</td>
<td>12</td>
</tr>
<tr>
<td>air</td>
<td>oil on machined</td>
<td>0.90</td>
<td>13</td>
</tr>
<tr>
<td></td>
<td>aluminum</td>
<td></td>
<td></td>
</tr>
<tr>
<td>GAS</td>
<td>SURFACE</td>
<td>$\alpha$</td>
<td>REFERENCE</td>
</tr>
<tr>
<td>--------</td>
<td>--------------------------</td>
<td>--------------</td>
<td>-----------</td>
</tr>
<tr>
<td>$\text{H}_2$</td>
<td>bright Pt.</td>
<td>0.32</td>
<td>14</td>
</tr>
<tr>
<td>$\text{H}_2$</td>
<td>black Pt.</td>
<td>0.74</td>
<td>14</td>
</tr>
<tr>
<td>$\text{O}_2$</td>
<td>bright Pt.</td>
<td>0.81</td>
<td>14</td>
</tr>
<tr>
<td>$\text{O}_2$</td>
<td>black Pt.</td>
<td>0.93</td>
<td>14</td>
</tr>
<tr>
<td>$\text{N}_2$</td>
<td>Pt</td>
<td>0.50</td>
<td>15</td>
</tr>
<tr>
<td>$\text{N}_2$</td>
<td>tungsten</td>
<td>0.35</td>
<td>15</td>
</tr>
<tr>
<td>air</td>
<td>flat laquer on bronze</td>
<td>0.88 - 0.89</td>
<td>16</td>
</tr>
<tr>
<td>air</td>
<td>polished bronze</td>
<td>0.91 - 0.94</td>
<td>16</td>
</tr>
<tr>
<td>air</td>
<td>machined bronze</td>
<td>0.89 - 0.93</td>
<td>16</td>
</tr>
<tr>
<td>air</td>
<td>etched bronze</td>
<td>0.93 - 0.95</td>
<td>16</td>
</tr>
<tr>
<td>air</td>
<td>polished cast iron</td>
<td>0.87 - 0.93</td>
<td>16</td>
</tr>
<tr>
<td>air</td>
<td>machined cast iron</td>
<td>0.87 - 0.88</td>
<td>16</td>
</tr>
<tr>
<td>air</td>
<td>etched cast iron</td>
<td>0.89 - 0.96</td>
<td>16</td>
</tr>
<tr>
<td>air</td>
<td>polished aluminum</td>
<td>0.87 - 0.95</td>
<td>16</td>
</tr>
<tr>
<td>air</td>
<td>machined aluminum</td>
<td>0.95 - 0.97</td>
<td>16</td>
</tr>
<tr>
<td>air</td>
<td>etched aluminum</td>
<td>0.89 - 0.97</td>
<td>16</td>
</tr>
<tr>
<td>He</td>
<td>tungsten</td>
<td>0.025 - 0.057</td>
<td>17</td>
</tr>
<tr>
<td>He</td>
<td>nickel not flashed</td>
<td>0.20</td>
<td>17</td>
</tr>
<tr>
<td>He</td>
<td>nickel flashed</td>
<td>0.085</td>
<td>17</td>
</tr>
<tr>
<td>He</td>
<td>tungsten flashed</td>
<td>0.17</td>
<td>18</td>
</tr>
<tr>
<td>He</td>
<td>tungsten not flashed</td>
<td>0.53</td>
<td>18</td>
</tr>
<tr>
<td>argon</td>
<td>tungsten flashed</td>
<td>0.82</td>
<td>18</td>
</tr>
<tr>
<td>argon</td>
<td>tungsten not flashed</td>
<td>1.00</td>
<td>18</td>
</tr>
<tr>
<td>INVESTIGATOR</td>
<td>GAS</td>
<td>SURFACE</td>
<td>OBSERVATION TECHNIQUE</td>
</tr>
<tr>
<td>----------------------</td>
<td>--------</td>
<td>---------</td>
<td>-----------------------------</td>
</tr>
<tr>
<td>R. W. Wood</td>
<td>Hg, Cd</td>
<td>Glass</td>
<td>Condensation on cooled bulb</td>
</tr>
<tr>
<td>M. Knudsen</td>
<td>Hg</td>
<td>Glass</td>
<td>Condensation on cooled bulb</td>
</tr>
<tr>
<td>J. B. Taylor</td>
<td>Li, K, Cs</td>
<td>LiF, NaCl</td>
<td>Surface ionization detector</td>
</tr>
<tr>
<td>A. Ellet and H. F. Olson</td>
<td>Hg, Cd</td>
<td>NaCl</td>
<td>Condensation on cooled bulb</td>
</tr>
<tr>
<td></td>
<td>Na</td>
<td>NaCl</td>
<td>Condensation in cooled bulb</td>
</tr>
<tr>
<td></td>
<td>H</td>
<td>NaCl</td>
<td>Reduction of Mo O₃</td>
</tr>
<tr>
<td>A. Ellet, H. F. Olson, H. A. Zahl</td>
<td>Cd</td>
<td>NaCl</td>
<td>Double crystal molecular beam with disc velocity selector. Condensation detector. Fixed crystal angles of 45°.</td>
</tr>
<tr>
<td>H. A. Zahl</td>
<td>Zn</td>
<td>NaCl</td>
<td>Apparatus as above</td>
</tr>
<tr>
<td>T. H. Johnson</td>
<td>H</td>
<td>LiF</td>
<td>Reduction of Mo O₃</td>
</tr>
<tr>
<td>INVESTIGATOR</td>
<td>GAS</td>
<td>SURFACE</td>
<td>OBSERVATION TECHNIQUE</td>
</tr>
<tr>
<td>---------------------</td>
<td>------</td>
<td>---------</td>
<td>-----------------------</td>
</tr>
<tr>
<td>F. Knauer and O. Stern</td>
<td>H₂, He</td>
<td>Speculum metal</td>
<td>Pirani Gage</td>
</tr>
<tr>
<td>I. Estermann and O. Stern</td>
<td>H₂, He</td>
<td>LiF</td>
<td>Pirani gage</td>
</tr>
<tr>
<td>I. Estermann, R. Frisch and O. Stern</td>
<td>He</td>
<td>LiF</td>
<td>Pirani gage detector with double crystal molecular beam and disc velocity selector. Variable crystal angles.</td>
</tr>
<tr>
<td>H. A. Zahl and A. Ellet</td>
<td>Hg</td>
<td>NaCl</td>
<td>Ionization gage</td>
</tr>
</tbody>
</table>
LINES ABN LIE IN INCIDENT PLANE
LINES TCN LIE IN PLANE OF OBSERVATION
INCIDENT ANGLE $\theta$ MEASURED IN BON
POLAR OBSERVATION ANGLE $\theta'$ MEASURED IN NOT
AZIMUTH OBSERVATION ANGLE $\phi'$ MEASURED IN COA

INCIDENT ANGLE
ROTATION AXIS

TEST SURFACE

BLOCK OF TEST MATERIAL

AXIS OF INCIDENT BEAM

NORMAL TO TEST SURFACE

DETECTOR GAGE

FIG. 1 GEOMETRY OF SCATTERING EXPERIMENT

HYD 6530-150-0
FIG. 3 MOLECULAR BEAM APPARATUS

HYD 6532-150-0
FIG. 4 OPTICAL BENCH WITH BEAM FORMING AND DETECTING EQUIPMENT

HYD 6533-150-0
FIG. 5 SCHEMATIC OF SOURCE GAS AND CALIBRATION SYSTEM

HYD 6534-150-0
FIG. 6 SCHEMATIC OF BASIC ROTATION ELEMENTS OF DETECTOR MECHANISM
FIG. 7 SCHEMATICS FOR ESTIMATION OF BEAM PRESSURE INCREMENT IN DETECTOR GAGE

DETECTION GAGE

ENTRANCE AREA A₃

TEST SURFACE

SOURCE ORIFICE

DEFINING ORIFICE

SCATTERED BEAM DETECTION

DIRECT BEAM DETECTION

AREA A₃a

AREA A₃b

GAGE WITH TAPERED ENTRANCE TUBE

A₂ = AREA AT A

A₁ = AREA A₁
FIG. 8  IONIZATION GAGE-MOLECULAR BEAM DETECTOR

① SPUTTERING SHIELDS  ④ GRID
② FILAMENT SUPPORT POINTS  ⑤ HOUSEKEEPER SEAL
③ ION COLLECTOR  ⑥ GAGE ENTRANCE

HYD 6537-150-0
FIG. 10 BLOCK DIAGRAM OF EMISSION REGULATOR
FIG. II  MOLECULAR BEAM ION GAGE REGULATOR

CHASSIS "C"

ION GAGE

ION GAGE FLAMELET
AMP HEATING HEAT

ION GAGE

NOTE:
• 7 PIN CONNECTOR
• 2 PIN CONNECTOR 1654P

HYD 6540-150-0
CHASSIS A
MODIFIED D150 753 TYPE SCR 545 A
STANDARD POWER SUPPLY (382 V)

CHASSIS CONNECTIONS:
A1 - C1  A3 - LINE
A2 - C12  A4 - LINE

FIG. 12a

CHASSIS B
300 V POWER SUPPLY

CHASSIS CONNECTIONS:
B1 - C1  B4 - LINE
B2 - C2  B5 - C5
B3 LINE  B6 - C6

FIG. 12b

HYD 6541-150-0
POWER SUPPLY #1
MODIFIED DI50753 TYPE SCR 545A STANDARD POWER SUPPLY

POWER SUPPLY #2

FIG 14
POWER SUPPLIES MOLECULAR BEAM DIFFERENCE AMPLIFIER
1 SPECIMEN HOLDER METAL SURFACES
2 POLISHED CRS SURFACE
3 POLISHED ALUMINUM SURFACE
4 POLISHED GLASS SURFACE
5 ETCHED CRS SURFACE
6 SPECIMEN HOLDERS GLASS SURFACES
7 UNPOLISHED GLASS SURFACE

FIG. 16 TEST SURFACES & HOLDERS

HYD 6545-150-0
FIG. 17 RAW DATA RECORD
(FROM RUN OF JULY 22, 1953)

HYD 6546-150-0
FIG. 18 SCATTERING PLOT - POLISHED STEEL SURFACE

FIG. 19 SCATTERING PLOT - POLISHED STEEL SURFACE
FIG. 20 SCATTERING PLOT - POLISHED STEEL SURFACE

FIG. 21 SCATTERING PLOT - POLISHED STEEL SURFACE
INC ANGLE 0.00°  
AZ ANGLE 0.00°  
UNHEATED SPECIMEN  
○ - MARCH 23, 1953  
HEATED SPECIMEN  
○ - APRIL 7, 1953  
UNHEATED SPECIMEN  
△ - APRIL 8, 1953

FIG. 22 SCATTERING PLOT - ALUMINUM SURFACE

INC ANGLE 25.11°  
AZ ANGLE 0.00°  
UNHEATED SPECIMEN  
○ - MARCH 23, 1953  
HEATED SPECIMEN  
○ - APRIL 7, 1953  
UNHEATED SPECIMEN  
△ - APRIL 8, 1953

FIG. 23 SCATTERING PLOT - ALUMINUM SURFACE

HYD 6549-150-0
INC. ANGLE 53.65°
AZ. ANGLE 0.00°
UNHEATED SPECIMEN
○-MARCH 23, 1953
HEATED SPECIMEN
△-APRIL 7, 1953
UNHEATED SPECIMEN
△-APRIL 8, 1953

FIG. 24 SCATTERING PLOT - POLISHED ALUMINUM SURFACE

FIG. 25 SCATTERING PLOT - POLISHED ALUMINUM SURFACE

INC. ANGLE 71.83°
AZ. ANGLE 0.00°
UNHEATED RUN
○-MARCH 23, 1953

HYD 6550-150-0
FIG. 26 SCATTERING PLOT UNPOLISHED GLASS SURFACE

INC ANGLE 0.00°
AZ ANGLE 0.00°
- MAY 16, 1953

INC ANGLE 0.00°
AZ ANGLE 19.57°
- MAY 23, 1953

FIG. 27 SCATTERING PLOT UNPOLISHED GLASS SURFACE

INC ANGLE 25.11°
AZ ANGLE 0.00
- MAY 16, 1953
- MAY 5, 1953

INC ANGLE 25.11°
AZ ANGLE 19.57°
- MAY 17, 1953

HYD 6551-150-0
Inc angle 40.67°
Az angle 0.00°
- May 16, 1953

Inc angle 40.67°
Az angle 15.58°
- May 23, 1953

Inc angle 40.67°
Az angle 35.83°
- May 25, 1953

Fig. 28 Scattering plot unpolished glass surface

Inc angle 53.65°
Az angle 0.00°
- May 16, 1953
- June 18, 1953

θ'
95° & 85°
75°
65°
55°
45°
35°
25°
15°
5° & 0°

Fig. 29 Scattering plot unpolished glass surface

Hyd 6552-150-0
FIG. 30 SCATTERING PLOT UNPOLISHED GLASS SURFACE

INC ANGLE 66.63°
AZ ANGLE 0.00°
• JUNE 20, 1953

ϕ'
90°, 85° & 80°
65°, 55°
45°, 35°
25°, 15°
15°, 5°
0°

FIG. 31 SCATTERING PLOT UNPOLISHED GLASS SURFACE

INC ANGLE 79.62°
AZ ANGLE 0.00°
• JUNE 21, 1953

ϕ'
90°, 85° & 80°
65°, 55°
45°, 35°
25°, 15°
15°, 5°
0°

INC ANGLE 79.62°
AZ ANGLE 15.58°
• MAY 23, 1953

HYD 6553-150-0
FIG. 32 SCATTERING PLOT UNPOLISHED GLASS SURFACE
FIG. 33 SCATTERING PLOT - POLISHED GLASS SURFACE

FIG. 34 SCATTERING PLOT - POLISHED GLASS SURFACE
FIG. 35 SCATTERING PLOT - POLISHED GLASS SURFACE

INC. ANGLE 40.07°
AZ. ANGLE 0.00°
O - JULY 23, 1953

FIG. 36 SCATTERING PLOT - POLISHED GLASS SURFACE

INC. ANGLE 56.25°
AZ. ANGLE 0.00°
O - JULY 23, 1953

HYD 6556-150-0
FIG. 37 SCATTERING PLOT - POLISHED GLASS SURFACE

INC. ANGLE 71.83°  
AZ. ANGLE 0.00°  
◊ - JULY 22, 1953

FIG. 38 SCATTERING PLOT - POLISHED GLASS SURFACE

INC. ANGLE 79.62°  
AZ. ANGLE 0.00°  
◊ - JULY 22, 1953
INC. ANGLE 84.81°
AZ. ANGLE 0.00°
○-JULY 23, 1953

FIG. 39 SCATTERING PLOT - POLISHED GLASS SURFACE

INC. ANGLE 90.00°
AZ. ANGLE 0.00°
○-JULY 23, 1953

FIG. 40 SCATTERING PLOT - POLISHED GLASS SURFACE
FIG. 41 COORDINATE SYSTEM - INCIDENT MOMENTUM CALCULATIONS
FIG. 42  COORDINATE SYSTEM - ISSUING MOMENTUM CALCULATIONS
POINTS CALCULATED FROM EXPERIMENTAL DATA

\( g_r(\theta_i) \) AS A FUNCTION OF \( \theta_i \)
FOR NITROGEN ON UNPOLISHED GLASS

\( \frac{g_r(\theta_i)}{\sin \theta_i} \) AS A FUNCTION OF \( \theta_i \)
FOR NITROGEN ON UNPOLISHED GLASS

FIG. 44
FIG. 45  $f$ (COEFFICIENT OF SPECULAR REFLECTION) AS A FUNCTION OF $s$ (MOLECULAR SPEED RATIO) FOR NITROGEN ON UNPOLISHED GLASS
FIG. 46 PLOT OF $\log J'$ AS A FUNCTION OF $b$

HYD 6564-150-0
G(θ)

mnV/m²

AS A FUNCTION OF θ FOR s = 0.1, 1.0, 2.0, AND 10.0

FIG. 47

HYD 6561-150-0
FIG. 48 $f$ AS A FUNCTION OF $\theta_0$ FOR VARIOUS MOLECULAR SPEED RATIOS ($s$)
DISTRIBUTION LIST FOR ONR TECHNICAL REPORTS

Director, Aeronautical Res., NACA, 1724 F St. NW, Wash. 25, D. C. (1)
Commanding General, Air Res. and Development Command, PO Box 1395, Baltimore (2) Maryland, Attn: OSR(RDF)
Dr. Morton Alperin, Western Regional Office, Hdqrs, Air Res. & Development Command, PO Box 2035, Pasadena 2, Calif. (1)
Director, Ames Aeronautical Lab., NACA, Moffett Field, Calif. (1)
Applied Physics Lab., Johns Hopkins Univ., 8621 Georgia Ave, Silver Sprng, Md. (1)
Armed Forces Special Weapons Project, The Pentagon, Wash., 25, D. C. (1)
Commanding Officer, Arnold Engr. Devel. Center, Tullahoma, Tenn., Attn: ACCT, ACR & AEMCL (1)
Mr. C. K. Bauer, Hermes Proj., General Electric Co, Schenectady 5, N.Y. (1)
Dr. Seymour Bogdonoff, James Forrestal Res. Cen, Princeton Univ, Princeton, N. J. (1)
Prof. W. S. Bredfield, Dept. of Aeronautical Engr. Univ. of Minnesota Minneapolis, Minn. (1)
Chief, Bureau of Aeronautics, Dept. of the Navy, Wash., 25, D.C. - Attn: Aero and Hydro Branch (1)
Chief, Bureau of Aeronautics, Dept. of the Navy, Wash. 25, D.C. - Attn: Research Division (1)
Bureau of Aeronautics Representative, USN, Central District, Wright-Patterson AFB, Dayton, Ohio, Attn: WCOWS-3, James G. Law. (1)
Chief, Bureau of Ordnance, Dept. of the Navy, Wash. 25, D.C. - Attn: Re9a (1)
Dr. Francis H. Clauser, Johns Hopkins Univ. Dept. of Aeronautical Engr., Baltimore 18, Md. (1)
Commanding Officer and Director, David Taylor Model Basin, Wash. 7, D.C. - Attn: Technical Library (1)
Chief of Naval Res., Dept. of the Navy, Wash. 25, D.C. - Attn: Code 438 (3)
Chief of Staff, Hdqrs USAF, Directorate of Research and Development, Wash. D.C. - Attn: AFDRD-AC2 (1)
Chief, Documents Service Center, Armed Services Technical Information Agency, Knott Bldg., Dayton 2, Ohio (1 reproducible copy) (2)
Prof. Fred S. Eastman, Univ. of Wash., Dept. of Aero. Engr., Seattle 5, Wash. (1)
Dr. I. Estermann, Code 419, Office of Naval Research, Wash., D.C. (1)
Prof. K. O. Friedrichs, New York Univ. Inst. for Math. and Mechs., 45 Astor Place, New York, N. Y. (1)
Director, Jet Propulsion Lab., Calif. Inst. of Tech., Pasadena, Calif. (1)
Prof. A. Kantrowitz, Graduate School of Aeronautical Engr., Cornell Univ. Ithaca, New York. (1)
Dr. J. Kaplan, Dept. of Physics, Univ. of Calif. at Los Angeles, L.A., Calif. (1)
Prof. M. Z. Kraywoblocki, Univ. of Illinois, Urbana, Illinois


Dr. A. R. Kuhlthau, Univ. of Virginia, Dept. of Physics, Charlottesville, Va.


Director, Langley Memorial Aeronautical Lab., NACA, Langley Field, Virginia

Prof. Lester Lees, Dept. of Aero. Engr., Calif. Inst. of Tech., Pasadena

Director, Lewis Flight Propulsion Laboratory, NACA, Cleveland, Ohio

Prof. H. Lewy, Applied Math. Group, Math. Dept., Univ. of Calif., Berkeley

Dr. James E. Lipp, The Rand Corp., 1500 Fourth St., Santa Monica, Calif.


Dr. C. E. Willikan, Guggenheim Aeronautical Lab., Calif. Inst. of Tech., Pasadena, California

Dr. Henry T. Naganatsu, Director, Hypersonic Wind Tunnel, Calif. Inst. of Tech., Pasadena 4, Calif.

Commanding Officer, U.S. Naval Ordnance Test Station, Inyokern, China Lake

Director, Naval Research Lab., Wash. 20, D. C. - Attn: Code 2020

Commander, Naval Ordnance Lab., White Oak, Silver Spring 19, Md, Attn: Aeroballistic Res. Dept.

Director, Office of Naval Research, 346 Broadway, New York 13, N. Y.

Director, Office of Naval Research, 844 North Rush St., Chicago 11, Ill.

Director, Office of Naval Research, 1000 Geary St., San Francisco 9, Calif.

Director, Office of Naval Research, 1030 East Green St., Pasadena 1, Calif.

Officer in Charge, Office of Naval Research, Navy #100, FPO, New York, N.Y.

Radiation Lab., Univ. of Calif., Berkeley 4, Calif. - Attn: R. Wakerling

Chairman, Research and Development Board, The Pentagon, Wash. 25, D.C.

Mr. R. H. Shick, Consolidated Vultee Aircraft Corp., San Diego 12, Calif.

Mr. Joseph Sternberg, Ballistics Res. Lab., Ordnance Dept. Aberdeen, Md.

Sverdrup & Parcel, Inc., Consulting Engrs., Syndicate Trust Bldg, St. Louis

Dr. Theodore Theodorsen, Univ. of Maryland, Bldg. J, Rm. 314, College Pk, Md.

Dr. F. L. Wattendorf, A.G.A.R.D. of NATO, APO 58, c/o PM, New York

Western Coordination Office, NACA, 7660 Beverly Blvd. L. A., 36, Calif.

Dr. E. P. Williams, Missiles Div., Rand Corp., 1500 Fourth St., Santa Monica, California

Wright Air Development Center, Wright-Patterson Air Force Base, Dayton, Ohio

Attn: WCRS-3 (1) Attn: MCIDX (1)

Dr. J. H. Marchant, Div. of Engr., Brown Univ., Providence 12, Rhode Island